

METHOD FOR PRODUCING RENEWABLE KEROSENE OR DIESEL BY TWO-STEP HYDROPROCESSING USING A SPECIFIC CATALYST FOR THE HYDROCONVERSION STEP, WITHOUT RECYCLE

Field of the invention

The search for new renewable energy sources for the production of fuels is a major challenge both for meeting fuel demand and for addressing environmental concerns.

In this respect, the upgrading as fuels of feedstocks derived from renewable sources has received substantial renewed interest in recent years. Among these feedstocks, examples that may be mentioned include plant oils (for example, palm, rapeseed, soybean), animal fats, used cooking oils, oils of microbial origin (for example, derived from algae), fish oils, long-chain paraffins (waxes) derived from the Fischer-Tropsch process, whether crude or pretreated, and also mixtures thereof. These feedstocks contain, for the most part, chemical structures of triglyceride or fatty acid or ester type, the structure and length of the hydrocarbon chain of these feedstocks being compatible with the hydrocarbons present in gas oils and kerosenes.

One possible route is catalytic transformation of the feedstock obtained from a renewable source into paraffinic fuel deoxygenated in the presence of hydrogen (hydrotreating). There are many metal or sulfide catalysts that are known to be active for this type of reaction.

These processes for the hydrotreating of a feedstock derived from renewable sources are already well known and are described in numerous patents. Mention may be made, for example, of the following patents: US 4 992 605, US 5 705 722, EP 1 681 337 and EP 1 741 768.

The use of solids based on transition metal sulfides allows the production of paraffins from molecules of ester type according to two reaction pathways:

- hydro-deoxygenation leading to the formation of water by consumption of hydrogen and to the formation of hydrocarbons with a carbon number (C_n) equal to that of the fatty acid chains,
- decarboxylation/decarbonylation leading to the formation of carbon oxides (carbon monoxide and dioxide: CO and CO₂) and to the formation of hydrocarbons containing one carbon less (C_{n-1}) relative to the initial fatty acid chains.

The liquid effluent obtained from these hydrotreating processes, after separation, consists essentially of n-paraffins and is substantially free of sulfur-based, nitrogen-based and oxygen-based impurities. After hydrotreating and separation of the gases, the sulfur content is typically between 1 and 20 ppm by weight, the nitrogen content is generally between 0.2 and 30 ppm by weight and the oxygen content is generally less than 2000 ppm by weight. The paraffins have a number of carbon atoms typically between 9 and 25, which is mainly dependent on the composition of the feedstock to be hydrotreated.

However, this liquid effluent generally cannot be incorporated as such into the kerosene or gas oil pool notably on account of insufficient cold properties and/or excessively high boiling points. Specifically, the paraffins present lead to high pour points and thus to setting phenomena for uses at low temperatures. For example, eicosane (linear paraffin of 20 carbon atoms, $C_{20}H_{42}$) has a boiling point equal to 340°C and a melting point of 37°C. The boiling point of eicosane is thus compatible with its incorporation in a gas-oil pool but its melting point may give rise to setting problems and limit its use. By way of illustration, the limit filterability temperature for a winter gas oil is a maximum of -15°C. Moreover, the boiling point of eicosane renders it ineligible for incorporation into a kerosene pool, for which the final temperature of the distillation curve must be less than 300°C.

Depending on the degree of incorporation and the favoured fuel pool (gas oil or kerosene) targeted, it may be necessary to perform a hydroconversion step (hydroisomerization and/or hydrocracking reactions) to transform the linear paraffins of the hydrotreated liquid effluent. Hydroisomerization makes it possible to convert a linear paraffin into a branched paraffin with conservation of the number of carbon atoms of the molecule. This makes it possible to improve the cold properties of the effluent since branched paraffins have better cold properties than linear paraffins. For example, nonadecane has a melting point of 32°C, whereas one of its singly branched isomers, 7-methyloctadecane, has a melting point of -16°C. Hydrocracking enables a linear paraffin to be converted into linear or branched paraffins of smaller molecular weight. This makes it possible to adjust, as need be, the distillation curve of the effluent to render it compatible with the kerosene pool. By way of illustration, the hydrocracking of an eicosane molecule can lead to the production of two 2-methylnonane molecules. The boiling point of 2-methylnonane is 167°C, which is compatible with incorporation into the kerosene pool. The hydroconversion step is performed on a bifunctional catalyst containing both a hydro/dehydrogenating function and a Brønsted acid function. The operating conditions may be adapted to promote the hydroisomerization or hydrocracking reactions as need be. In any

case, it is desirable to minimize the production of cracking products that are too light to be incorporated into the kerosene or gas oil pool.

The appropriate choice of the acid phase makes it possible to promote the isomerization of long linear paraffins and to minimize the cracking. Thus, the form selectivity of medium-pore (10 MR) one-dimensional zeolites such as zeolites ZSM-22, ZSM-23, NU-10, ZSM-48 and ZBM-30 makes them particularly suitable for use for obtaining catalysts that are selective toward isomerization. Other acidic phases of zeolite or non-zeolite type such as halogenated (notably chlorinated or fluorinated) aluminas, phosphorus-based aluminas, silica-aluminas or siliceous aluminas may also be used.

However, it is well known that factors other than the acid phase have an impact on the activity and selectivity of a bifunctional catalyst. Hydroisomerization and hydrocracking of normal paraffins have thus been the subject of numerous academic studies since the original investigations in the 1960s by Weisz or Coonradt and Garwood. The most commonly accepted mechanism first involves dehydrogenation of the n-paraffin to an n-olefin on the hydrogenating-dehydrogenating phase and then, after diffusion to the acid phase, protonation to a carbenium ion. After structural rearrangement and/or β -scission, the carbenium ions are desorbed from the acidic phase in the form of olefins after deprotonation. Next, after diffusion to the hydro-dehydrogenating phase, the olefins are hydrogenated to form the final reaction products. It is then appropriate to have a hydro/dehydrogenating function that is sufficiently active toward the acid function firstly to rapidly supply the acidic phase with olefins and secondly to rapidly hydrogenate the olefinic intermediates after they have reacted on the acidic phase. This makes it possible firstly to maximize the activity of the catalyst and secondly to promote hydroisomerization relative to hydrocracking when the first reaction is desired, or to limit the production of excessively light cracking products when the hydrocracking reaction is desired. The use of a sufficiently active hydrogenating function is also desirable so as to limit the deactivation of the bifunctional catalyst by coking during the hydroconversion of n-paraffins (Alvarez et al., *Journal of Catalysis*, 162, 2, 179-189) for a range of set operating conditions.

The proximity between the two functions on the catalyst may also have an impact on the performance of the difunctional catalyst. Thus, Zecevic et al. (*Nature*, 2015, 528, 245-254) recently studied the impact of the localization of platinum on the hydroisomerization performance of long paraffins (n-decane, n-nonadecane, pristane) by a difunctional catalyst using the zeolite USY as acidic phase and an alumina matrix. It is observed that the difunctional catalyst for which platinum is deposited on alumina is systematically more selective towards

isomerization than the catalyst for which platinum is deposited in the zeolite. In the light of these results, a person skilled in the art is thus inclined to favour localization of the hydrogenating function on the alumina matrix rather than on the acidic phase to improve the selectivity towards isomerization. From the point of view of the activity, localizing platinum on the aluminium matrix has a variable impact depending on the long paraffin under consideration: a positive impact as regards n-decane, a marginal impact as regards n-nonadecane and finally a negative impact as regards pristane.

Noble metals (Pt, Pd) or transition metals from group VIA (Mo, W) combined with transition metals from group VIII (Ni, Co) can act as hydrogenating function for the catalyst. The noble metals are used in their reduced form whereas the transition metals are used in a sulfide form. For the latter, there is a known synergistic effect between the transition metals of group VIA and the transition metals of group VIII, generally attributed to the decoration of the sulfide phases of group VIA with the transition metals of group VIII. These are then referred to as molybdenum or tungsten sulfide phases promoted with nickel or cobalt ("CoMoS", "NiMoS", "NiWS"). This synergistic effect is reflected by an increase in the catalytic activity of the promoted phase in comparison with an unpromoted phase.

The choice of the nature of the hydrogenating function, of noble metal or sulfide type, depends on various criteria, of economic nature (the price of noble metals is markedly higher than that of transition metals of groups VIA and VIII) or of chemical nature (impact of the presence of contaminants). Thus, the hydrogenating activity of noble metals is higher than that of transition metal sulfides when the partial pressure of hydrogen sulfide (H_2S) in the reaction medium is low or even zero. Conversely, the hydrogenating activity of transition metal sulfides is higher than that of noble metals when the partial pressure of H_2S in the reaction medium becomes high (C. Marcilly, *Catalyse acido-basique [Acid-base catalysis]*, volume 2, 2003, published by Technip).

Patent US2022/0127537 discloses a process for hydrotreating a renewable feedstock. Said process comprises a step of hydrotreating the feedstock in the presence of hydrogen and a hydrotreating catalyst to deoxygenate said feedstock and thus produce a hydrotreated effluent. Said process comprises a step of hydroisomerization, in the presence of hydrogen and a hydroisomerization catalyst, of an effluent derived from the hydrotreated effluent to obtain a hydroisomerized effluent. The hydroisomerization catalyst used may comprise a Group VIII metal chosen from Pt and Pd, alone or in combination, and a support which may be amorphous or crystalline, chosen from alumina, amorphous alumina-silica, ferrierite, ALPO-31, SAPO-11,

SAPO-31, SAPO-37, SAPO-41, SM-3, MgAPSO-31, FU-9, NU-10, NU-23, ZSM-12, ZSM-22, ZSM-23, ZSM-35, ZSM-48, ZSM-50, ZSM-57, MeAPO-11, MeAPO-31, MeAPO-41, MgAPSO-11, MgAPSO-31, MgAPSO-41, MgAPSO-46, ELAPSO-11, ELAPSO-3, EMAPSO-41, ELAPSO-11, ELAPSO-31, ELAPSO-41, alone or in combination.

In a first embodiment claimed in claim 1, said process comprises a step of separating the hydrotreated effluent and/or the isomerized effluent to obtain a vapour effluent and a liquid effluent; a step of distilling the liquid effluent or the hydroisomerized effluent to produce a kerosene cut and a gas oil cut; and a step of hydrocracking the gas oil cut to obtain a hydrocracked effluent comprising a kerosene cut. In contrast with the present invention, said process thus allows the co-production of gas oil and kerosene cuts, which are separated via a distillation step, and the gas oil cut obtained then undergoes a hydrocracking step.

In a second embodiment claimed in claim 16, said process involves obtaining a gas oil cut from the hydroisomerized effluent, said gas oil cut being characterized in that the concentration of normal paraffin of a given carbon number is at least twice that of the corresponding normal paraffin concentration in said hydroisomerized effluent. Said gas oil cut undergoes a hydrocracking or hydroisomerization step to obtain a hydrocracked effluent comprising a kerosene cut. Unlike the present invention, said process thus comprises either a step of hydrocracking the gas oil cut obtained, or a step of recycling said gas oil cut into the hydroisomerization step.

In a third embodiment claimed in claim 19, said process comprises a step of separating said hydroisomerized effluent into a vapour effluent and a liquid effluent; a step of distilling the liquid effluent or the hydroisomerized effluent to obtain a kerosene cut and a gas oil cut, and a step of hydrocracking said gas oil cut to obtain a hydrocracked effluent comprising a kerosene cut. In contrast with the present invention, said process thus allows the co-production of gas oil and kerosene cuts, which are separated via a distillation step, and the gas oil cut obtained then undergoes a hydrocracking step.

Patent US 8 324 439 B2 discloses a process for treating renewable feedstocks of plant or animal origin. Said process comprises a step of hydrotreating the renewable feedstock, and a step of separating the hydrotreated effluent to obtain hydrogen, other gases and at least one effluent containing hydrocarbons. Said process then comprises a step of hydroisomerizing at least a portion of said hydrocarbon-based effluent in the presence of a selective hydroisomerization catalyst, said catalyst comprising at least one 10 MR one-dimensional

zeolite and at least one metal from Group VIII and/or Group VIB. Finally, said process comprises a step of separating the hydroisomerized effluent so as to obtain hydrogen, other gases and at least one gas oil type cut. Said patent also discloses a process comprising a step of hydroisomerization of an effluent obtained from the step of hydrotreating a renewable feedstock of animal or plant origin over a hydroisomerization catalyst. Said catalyst comprises at least one 10 MR one-dimensional zeolite of TON or EUO structure code, or a zeolite ZSM-48, ZBM-30, IZM-1, COK-7, EU-2 and EU-11, alone or as a mixture. The structural framework codes are defined in the International Zeolite Association classification (IZA: <http://www.iza-structure.org/databases/>).

US patent Application 2022/0403252 describes a process for producing gas oil and kerosene, comprising the hydrotreating of renewable feedstocks such as plant oils, followed by a step of hydroisomerization of the hydrotreated effluent, and then fractionation of the intermediate paraffinic effluent produced by hydroisomerization into gas oil and kerosene cuts. No separation step is described between the hydrotreating and hydroisomerization steps. This is a “one-step” process, meaning that the hydrotreating and hydroisomerization can be performed in a single reactor or in different reactors without intermediate separation. The hydroisomerization catalysts comprise a Group VIII metal, preferably platinum, and optionally a zeolite chosen from the zeolites SAPO-11, SAPO-41, ZSM-22, ZSM-23 or ferrierite.

For the purposes of the present invention, the various embodiments presented can be used alone or in combination with each other, without any limit to the combinations.

For the purposes of the present invention, the various ranges of parameters for a given step, such as the pressure ranges and the temperature ranges, can be used alone or in combination. For example, for the purposes of the present invention, a preferred range of pressure values may be combined with a more preferred range of temperature values.

In the remainder of the text, the term “naphtha” or “naphtha cut” means the hydrocarbon fraction with a boiling point lower than the middle distillate cut. The middle distillate cut generally has an initial cut point of between 120°C and 160°C, preferentially 120°C. The naphtha cut may have boiling points between that of the hydrocarbon compounds containing five carbon atoms per molecule (or 36°C boiling point) up to 216°C and includes the gasoline cut.

Throughout the remainder of the text, the term “kerosene” or “kerosene cut” means the cut having initial and final boiling points of between 120°C and 300°C and the term “gas-oil” or “gas-oil cut” means the cut having initial and final boiling points in a range from 120°C to 400°C and preferably between 120°C and 380°C.

In the remainder of the text, the groups of chemical elements are given according to the CAS classification (CRC Handbook of Chemistry and Physics, published by CRC Press, editor-in-chief D.R. Lide, 81st edition, 2000-2001). For example, group VIII according to the CAS classification corresponds to the metals from columns 8, 9 and 10 according to the new IUPAC classification, and group VIB corresponds to the metals from column 6.

In the continuation of the text, the expressions “of between ... and ...” and “between ... and ...” are equivalent and mean that the limiting values of the range are included in the range of values described. If such were not the case and if the limiting values were not included in the range described, such a detail will be provided by the present invention.

In the present description, the term “greater than...” is understood as strictly greater than, and is symbolized by the sign “>”, and the term “less than...” is understood as strictly less than, and is symbolized by the sign “<”.

ADVANTAGES OF THE PROCESS ACCORDING TO THE INVENTION

Whilst attempting to develop a process for treating feedstock from a renewable source, the Applicant discovered that the use of a specific bifunctional catalyst comprising a noble metal (Pt)-based hydrogenating/dehydrogenating phase and a support comprising a specific zeolite chosen from zeolites with the structure code MTW, and IZM-2 zeolite, either alone or as a mixture, for the hydroconversion step (hydroisomerization and/or hydrocracking reactions) of linear paraffins derived from the hydrodeoxygenation step of renewable feedstocks, was of great interest.

Subject of the invention

More specifically, the present invention relates to a process for treating a feedstock from a renewable source to produce alternatively a gas oil cut or a kerosene cut, said process comprising at least the following steps and preferably consisting of the following steps:

a) a step of hydrotreating said feedstock in the presence of a catalyst in a fixed bed, said catalyst comprising a hydrogenating function and an oxide support, at a temperature of between 200 and 450°C, at a pressure of between 1 MPa and 10 MPa, at an hourly space velocity of between 0.1 h⁻¹ and 10 h⁻¹ and in the presence of a total amount of hydrogen mixed with the feedstock such that the hydrogen/feedstock ratio is between 70 and 2000 Nm³ of hydrogen/m³ of feedstock,

b) a step of separating at least a portion of the effluent obtained from step a) into at least a light gaseous fraction, at least a hydrocarbon liquid effluent, and at least an aqueous liquid effluent,

c) a step of hydroconversion of at least a portion of the hydrocarbon liquid effluent obtained from step b) in the presence of at least one difunctional hydroconversion catalyst in a fixed bed, said catalyst comprising a hydrogenating phase comprising at least one noble metal from group VIII of the Periodic Table chosen from platinum and palladium, and a support comprising at least one zeolite chosen from zeolites with the structure code MTW and the zeolite IZM-2, either alone or as a mixture, and at least one binder, said hydroconversion step being performed at a temperature of between 250°C and 500°C, at a pressure of between 1 MPa and 10 MPa, at an hourly space velocity of between 0.1 and 10 h⁻¹ and in the presence of a total amount of hydrogen mixed with the feedstock such that the hydrogen/feedstock ratio is between 70 and 1000 Nm³ of hydrogen/m³ of feedstock,

d) a step of separating at least a portion of the effluent obtained from step c) making it possible to separate at least a gaseous fraction and at least a hydrocarbon liquid effluent,

e) a step of stabilizing at least a portion of the liquid hydrocarbon effluent obtained from step d), so as to separate three hydrocarbon cuts: a light gaseous fraction, a naphtha hydrocarbon cut and, alternatively, a gas oil cut or a kerosene cut,

no step of recycling, into the hydrotreating step a) and/or the hydroconversion step c), of all or some of the effluents obtained from the hydroconversion step c), of the effluents from the separation step d) and/or of the effluents from the stabilization step e) being performed.

An advantage of the process according to the present invention is thus that it provides a process for treating a feedstock derived from a renewable source which undergoes hydrotreatment before being sent to a hydroconversion step using a specific difunctional catalyst comprising a noble metal (Pt)-based hydrogenating/dehydrogenating phase and a

support comprising a specific zeolite chosen from zeolites of the MTW structure type, and zeolite IZM-2, either alone or as a mixture, thereby enabling an increase in the activity and selectivity of the hydroconversion catalyst.

Performing the hydroconversion step with the specific catalyst in accordance with the invention makes it possible, all other things being equal, to reduce the temperature required to obtain a target cold property value for the middle distillate cut (measured, for example, by a cloud point value).

Another advantage of the present invention is that it provides a process for treating a feedstock derived from a renewable source which undergoes hydrotreating before being sent to a hydroconversion step using a specific bifunctional catalyst, thereby also improving the yield of middle distillate cut for a target cold property value (measured, for example, by a cloud point value).

Another advantage of the present invention is that it provides a process for treating a feedstock derived from a flexible renewable source, in which, depending on the preferred fuel pool (gas oil or kerosene), the operating conditions used in the hydroconversion step can advantageously be adjusted to produce either a gas oil cut or a bio-gas oil cut, or a kerosene cut or a bio-kerosene cut.

Compliance with product targets is achieved by adapting the operating conditions of step c) and does not require an additional distillation or fractionation step to obtain the liquid effluent from step d). This results in a simplification of the process and an economic benefit (easier operation, reduced equipment requirements and lower unit cost). The use of the specific catalyst in the hydroconversion step c) allows the yields of either bio-gas oil or bio-kerosene to be maximized for target cold properties.

Detailed description of the invention

Feedstocks

The present invention is particularly dedicated to the preparation of gas oil and/or kerosene fuel bases which meet the new environmental standards, starting with feedstocks obtained from renewable sources.

The feedstocks derived from renewable sources used in the process according to the present invention are advantageously chosen from plant oils (for example palm, rapeseed, soybean) and animal fats, used cooking oils, oils of microbial origin (for example derived from algae), fish oils, long-chain paraffins (waxes) derived from the Fischer-Tropsch process, whether crude or pretreated, or mixtures of such feedstocks, containing triglycerides and/or free fatty acids and/or esters. The vegetable oils may advantageously be crude or wholly or partially refined and obtained from the following plants: rapeseed, sunflower, soybean, palm, palm kernel, olive, coconut, jatropha, this list not being limiting. Algal oils or fish oils are also relevant. The animal fats are advantageously chosen from lard or fats composed of residues from the food industry or resulting from the catering industries.

These feedstocks essentially contain chemical structures of triglyceride type that a person skilled in the art also knows under the name fatty acid triester, and also free fatty acids. A fatty acid triester is thus composed of three fatty acid chains. These fatty acid chains in triester form or in free fatty acid form have a number of unsaturations per chain, also known as the number of carbon-carbon double bonds per chain, generally between 0 and 3, but which may be higher notably for oils derived from algae which generally contain from 5 to 6 unsaturations per chain.

The molecules present in the feedstocks obtained from renewable sources used in the present invention thus have a number of unsaturations, expressed per triglyceride molecule, advantageously between 0 and 18. In these feedstocks, the degree of unsaturation, expressed as the number of unsaturations per hydrocarbon fatty chain, is advantageously between 0 and 6.

The feedstocks obtained from renewable sources generally also include various impurities and notably heteroatoms such as nitrogen. The nitrogen contents in plant oils are generally between 1 ppm and 1000 ppm by weight approximately, depending on their nature.

Process and catalysts

Advantageously, the feedstock may undergo, prior to step a) of the process according to the invention, a pretreatment or pre-refining step so as to remove, via a suitable treatment, contaminants such as metals, for instance alkali metal compounds, for example on ion-exchange resins, alkaline-earth metal compounds and phosphorus. Suitable treatments may be, for example, heat treatments and/or chemical treatments that are well known to those skilled in the art.

In accordance with step a) of the process according to the invention, the optionally pretreated feedstock is placed in contact with a catalyst in a fixed bed at a temperature of between 200 and 450°C, preferably between 220 and 350°C, preferably between 220 and 320°C and even more preferably between 220 and 310°C. The pressure is between 1 MPa and 10 MPa, preferably between 1 MPa and 6 MPa and even more preferably between 1 MPa and 4 MPa. The hourly space velocity, i.e. the volume of feedstock per volume of catalyst and per hour, is between 0.1 h⁻¹ and 10 h⁻¹. The feedstock is placed in contact with the catalyst in the presence of hydrogen. The total amount of hydrogen mixed with the feedstock is such that the hydrogen/feedstock ratio is between 70 and 2000 Nm³ of hydrogen/m³ of feedstock and preferably between 150 and 1000 Nm³ of hydrogen/m³ of feedstock.

In step a) of the process according to the invention, the fixed-bed catalyst is advantageously a hydrotreating catalyst comprising a hydro-dehydrogenating function comprising at least one metal from group VIII and/or from group VIB, taken alone or as a mixture, and a support chosen from the group formed by alumina, silica, silica-aluminas, magnesia, clays and mixtures of at least two of these minerals. This support may also advantageously contain other compounds, for example oxides chosen from the group consisting of boron oxide, zirconia, titanium oxide and phosphorus pentoxide. The preferred support is an alumina support and very preferably η , δ or γ alumina.

Said catalyst is advantageously a catalyst comprising metals from group VIII preferably chosen from nickel and cobalt, taken alone or as a mixture, preferably in combination with at least one metal from group VIB preferably chosen from molybdenum and tungsten, taken alone or as a mixture.

The content of oxides of metals from group VIII and preferably of nickel oxide is advantageously between 0.5% and 10% by weight of nickel oxide (NiO) and preferably between 1% and 5% by weight of nickel oxide, and the content of oxides of metals from group VIB and preferably of molybdenum trioxide is advantageously between 1% and 30% by weight of molybdenum trioxide (MoO₃), preferably from 5% to 25% by weight, the percentages being expressed as weight percentages relative to the total mass of the catalyst.

The total content of oxides of metals from groups VIB and VIII in the catalyst used in step a) is advantageously between 5% and 40% by weight and preferably between 6% and 30% by weight relative to the total mass of the catalyst.

Said catalyst used in step a) of the process according to the invention must advantageously be characterized by a high hydrogenating power so as to orient the reaction selectivity as much as possible toward a hydrogenation conserving the number of carbon atoms in the fatty chains, i.e. the hydro-deoxygenation route, so as to maximize the yield of hydrocarbons included in the distillation range of kerosenes and/or gas oils. This is why the process is preferably performed at a relatively low temperature. Maximizing the hydrogenating function also makes it possible to limit the polymerization and/or condensation reactions leading to the formation of coke which would degrade the stability of the catalytic performance. A catalyst of Ni or NiMo type is preferably used.

Said catalyst used in the hydrotreating step a) of the process according to the invention may also advantageously contain a doping element chosen from phosphorus and boron, taken alone or as a mixture. Said doping element may be introduced into the matrix or, preferably, may be deposited on the support. Silicon may also be deposited on the support, alone or with phosphorus and/or boron and/or fluorine.

The weight content of oxide of said doping element is advantageously less than 20% by weight and preferably less than 10% by weight and is advantageously at least 0.001% by weight.

Preferred catalysts are the catalysts described in patent application FR 2 943 071 that describes catalysts having a high selectivity for hydro-deoxygenation reactions.

Other preferred catalysts are the catalysts described in patent application EP 2 210 663 that describes supported or bulk catalysts comprising an active phase consisting of a group VIB sulfide element, in which the group VIB element is molybdenum.

The metals of the catalysts used in the hydrotreating step a) of the process according to the invention are metal sulfides or metallic phases and preferably metal sulfides.

It would not constitute a departure from the context of the present invention to use in step a) of the process according to the invention, simultaneously or successively, a single catalyst or several different catalysts. This step may be performed industrially in one or more reactors with one or more catalytic beds, preferably with a descending flow of liquid.

Said hydrotreating step a) allows the hydrogenation, hydro-deoxygenation, hydro-denitrogenation and hydro-desulfurization of said feedstock.

In accordance with step b) of the process according to the invention, a step of separating at least a portion and preferably all of the effluent obtained from step a) is performed. Said step b) enables the separation of at least one hydrogen-rich "light" gas fraction, at least one hydrocarbon liquid effluent composed of n-paraffins, and at least one liquid aqueous effluent.

Said light gaseous fraction comprises at least the hydrogen not converted in the reactions performed in step a), at least the gases containing one or more oxygen atoms resulting from the decomposition of oxygenated compounds in step a), and at least the C_4^- compounds, i.e. C_1 to C_4 compounds preferably having a final boiling point below 20°C. The aim of this step is to separate the gases from the liquids. More particularly, the aim is to recover at least the hydrogen-rich gases, which may also contain compounds such as CO and CO_2 , at least one hydrocarbon liquid effluent consisting of n-paraffins, and at least one liquid aqueous effluent containing the water produced in the reactions performed in step a). Said hydrocarbon liquid effluent preferably has a sulfur content of less than 10 ppm by weight and a nitrogen content of less than 2 ppm by weight.

The separation step b) may advantageously be performed via any method known to those skilled in the art, for instance the combination of one or more high-pressure and/or low-pressure separators operated under hot or cold conditions, and/or high-pressure and/or low-pressure stripping.

Step b) also allows the separation of at least one liquid aqueous effluent, preferably water. The removal of at least a portion of the water and preferably all of the water may be performed via any method or technique known to those skilled in the art. Preferably, the removal of water is performed by decantation into a separating vessel, by drying, by passing through a desiccant, by flash distillation, or via a combination of at least two of these techniques. The atomic oxygen content of the hydrocarbon liquid effluent containing the paraffinic hydrocarbons obtained from step b) of the process according to the invention, expressed in parts per million by weight (ppm), is preferably less than 10 000 ppm, preferably less than 6000 ppm, very preferably less than 1000 ppm and more preferably less than 500 ppm by weight. The content in ppm by weight of atomic oxygen in said hydrocarbon liquid effluent is measured via the infrared absorption technique, for instance the technique described in patent application US 2009/0018374 A1.

In accordance with step c) of the process according to the invention, at least part and preferably all of the liquid hydrocarbon effluent obtained from step b) of the process according to the

invention is converted in the presence of at least one difunctional hydroconversion catalyst in a fixed bed, said catalyst comprising a hydrogenating phase comprising at least one noble metal from group VIII of the Periodic Table chosen from platinum and palladium, alone or as a mixture, and a support comprising at least one zeolite chosen from zeolites with the structure code MTW and IZM-2, either alone or as a mixture, and at least one binder, said hydroconversion step being performed at a temperature of between 250°C and 500°C, at a pressure of between 1 MPa and 10 MPa, at an hourly space velocity of between 0.1 and 10 h⁻¹ and in the presence of a total amount of hydrogen mixed with the feedstock such that the hydrogen/feedstock ratio is between 70 and 1000 Nm³ of hydrogen/m³ of feedstock, and preferably between 150 and 750 Nm³ of hydrogen/m³ of feedstock.

The operating conditions of the hydroconversion step c) are adjusted to promote the hydroisomerization or hydrocracking reactions as need be. Preferably, the hydroconversion step c) of the process according to the invention is a hydroisomerization step and advantageously operates at a temperature of between 250°C and 450°C and very preferably between 250°C and 400°C, at a pressure of between 2 MPa and 10 MPa and very preferably between 3 MPa and 9 MPa, at an hourly space velocity advantageously between 0.2 and 7 h⁻¹ and very preferably between 0.5 and 5 h⁻¹, at a hydrogen flow rate such that the hydrogen/feedstock volume ratio is advantageously between 100 and 1000 Nm³ of hydrogen/m³ of feedstock and preferably between 150 and 1000 Nm³ of hydrogen/m³ of feedstock.

Advantageously, the operating conditions used in the hydroconversion step c) may be adjusted so as to produce either a gas oil cut or a bio-gas oil cut, or a kerosene or bio-kerosene cut. Generally, the total pressures are fixed, and the operating conditions are usually adjusted by varying the temperature of the hydroconversion step and/or the hourly space velocity. The requirements in terms of cold properties or cut-off points are higher for the production of the kerosene cut than for the gas oil cut. All other things being equal, the switch from a “gas oil” production mode to a “kerosene” production mode can be made by increasing the reaction temperature and/or by reducing the hourly space velocity.

The process according to the invention is not intended for the simultaneous co-production of a gas oil cut and a kerosene cut. The process according to the invention is a flexible process which, by applying adjusted operating conditions in the hydroconversion step c), combined with the use of a specific catalyst in step c), enables the production alternatively of a gas oil

cut or a kerosene cut. The process thus does not require a final fractionation step to separate the two cuts, as is the case with a process that allows the co-production of gas oil and kerosene.

In accordance with the invention, the catalyst used in step c) is a difunctional catalyst comprising a hydrogenating phase comprising at least one noble metal from group VIII of the Periodic Table chosen from platinum and palladium, alone or as a mixture, and a support comprising at least one zeolite chosen from zeolites of the MTW structure type, and IZM-2 zeolite, either alone or as a mixture, and at least one binder.

The hydro-dehydrogenating function

Preferably, the Group VIII metal of the catalyst used in step c) is platinum.

Advantageously, the hydro-dehydrogenating (metal) element, preferably platinum, may be introduced onto the catalyst support by any method known to those skilled in the art, for instance co-kneading, dry impregnation or impregnation by exchange.

According to one or more embodiments, the content of group VIII metal, and preferably the content of platinum, in the catalyst used in step c) is between 0.01% by weight and 4% by weight, preferably between 0.05% by weight and 2% by weight, relative to the total weight of said catalyst.

The catalyst used in step c) may also advantageously comprise at least one additional metal chosen from the group formed by metals from groups IIIA, IVA and VIIB of the Periodic Table of the Elements, preferably chosen from gallium, indium, tin and rhenium. Said additional metal is preferably chosen from indium, tin and rhenium.

Preferably, the content of at least one additional metal in the catalyst used in step c) is between 0.01% by weight and 2% by weight, preferably between 0.05% by weight and 1% by weight, relative to the total weight of said catalyst.

According to one embodiment, the sulfur content in the hydroconversion catalyst is such that the ratio of the number of moles of sulfur to the number of moles of at least one group VIIB metal is between 0.3 and 3. According to one or more embodiments, the presence of sulfur in the catalyst results from an optional sulfurization step of the hydroconversion catalyst. According to one or more embodiments, the presence of sulfur in the catalyst results from

impurities that may be present, for instance in the alumina binder. According to another embodiment, the catalyst does not contain any sulfur.

The acid function

In accordance with the invention, the catalyst used in step c) comprises a support comprising at least one zeolite chosen from zeolites of the MTW structure type and zeolite IZM-2, either alone or as a mixture, and at least one binder.

The acid function of the difunctional catalyst used in step c) is provided by the zeolite chosen from zeolites of the MTW structure type and zeolite IZM-2, either alone or as a mixture.

Preferably, the zeolites of structure type MTW are chosen from ZSM-12, TPZ-12, Theta-3, NU-13, CZH-5 zeolites, alone or as a mixture, and preferably the zeolite of structure type MTW is ZSM-12.

According to a preferred embodiment, the catalyst used in step c) comprises a support comprising an IZM-2 zeolite alone, or a support comprising a ZSM-12 zeolite alone.

IZM-2 zeolite is a crystalline microporous solid, the crystal structure and preparation process of which are described in patent application FR2918050A1. The structure code of IZM-2 zeolite is unknown to date. ZSM-12 zeolite is a crystalline microporous solid, the crystal structure of which is described on the International Zeolite Association website (<http://www.iza-structure.org/>). It is a monodimensional 12 MR zeolite; its structure code is MTW. A process for preparing this zeolite is described, for example, in the article "Synthesis of zeolite ZSM-12 in the system $(\text{MTEA})_2\text{O}-\text{Na}_2\text{O}-\text{SiO}_2-\text{Al}_2\text{O}_3-\text{H}_2\text{O}$ " by S. Ernst et al. (Zeolites, 7, 5, 458-462, DOI10.1016/0144-2449(87)90015-7).

The zeolites are preferably essentially in acid form, i.e. the atomic ratio between the monovalent compensation cation (for example sodium) and the aluminium inserted into the crystal lattice of the solid is advantageously less than 0.1, preferably less than 0.05 and very preferably less than 0.01. According to one or more embodiments, the zeolites included in the composition of said hydroisomerization catalyst are advantageously calcined. According to one or more embodiments, said zeolites are exchanged by means of at least one treatment with a solution of at least one ammonium salt so as to obtain the ammonium form of the zeolites which, once calcined, leads to the acid form of said zeolites.

In a preferred embodiment, the catalyst used in step c) comprises a hydrogenating phase comprising platinum and a support comprising IZM-2 zeolite alone and an alumina binder.

In another preferred embodiment, the catalyst used in step c) comprises a hydrogenating phase comprising platinum and a support comprising ZSM-12 zeolite alone and an alumina binder.

Preferably, the catalyst used in step c) comprises a zeolite content of between 1% by weight and 90% by weight, preferably between 3% by weight and 80% by weight, more preferentially between 4% by weight and 60% by weight, preferably between 4% and 30% by weight and more preferably between 4% and 20% by weight relative to the total weight of said catalyst.

The binder

Preferably, the binder of the catalyst support of step c) is amorphous or crystalline. Preferably, the binder used in the catalyst support of step c) is advantageously chosen from the group formed by alumina, silica, silica-alumina, clays, titanium oxide, boron oxide, zirconia and aluminates, taken alone or as a mixture. Preferably, the binder is alumina. Preferably, said binder may contain alumina in all its forms known to those skilled in the art, for instance aluminas of alpha, gamma, eta and delta type.

Preferably, the catalyst used in step c) comprises a binder content of between 10% by weight and 99% by weight relative to the total weight of said catalyst, i.e. so as provide the balance up to 100% by weight of the elements constituting the catalyst used in step c).

According to the invention, the catalyst support comprises the zeolite as a mixture with a binder. The support is preferably formed into a mixture by co-kneading, extrusion and then heat treatment of the zeolite with the binder or a precursor of the binder, for instance boehmite, which is transformed into alumina by heat treatment.

A preferred catalyst for step c) comprises, and preferably consists of, platinum, and a support comprising, and preferably consisting of, a ZSM-12 zeolite and an alumina binder.

Another preferred catalyst for step c) comprises, and preferably consists of, platinum, and a support comprising, and preferably consisting of, an IZM-2 zeolite and an alumina binder.

According to a preferred embodiment, the catalyst used in step c) more particularly comprises, and preferably consists of:

- from 1% to 90% by weight, preferably from 3% to 80% by weight and even more preferably from 4% to 60% by weight of zeolite;
- from 0.01% to 4% by weight and preferably from 0.05% to 2% by weight of at least one metal from group VIII B, preferably platinum;
- optionally from 0.01% to 2% by weight and preferably from 0.05% to 1% by weight of at least one additional metal chosen from the group formed by metals from groups III A, IV A and VII B;
- optionally a sulfur content, preferably such that the ratio of the number of moles of sulfur to the number of moles of the group VIII B metal(s) is between 0.3 and 3; and
- optionally at least one binder, preferably alumina, providing the balance up to 100% by weight in the catalyst, relative to the total weight of the catalyst of step c).

Preferably, the catalyst used in step c) is formed into cylindrical or polylobal extrudates such as straight or twisted bilobal, trilobal or polylobal extrudates. According to one or more embodiments, the catalyst used in step c) is formed into crushed powders, tablets, rings, beads or wheels. Techniques other than extrusion, such as pelletizing or coating, may advantageously be used.

Preferably, the noble metal contained in said catalyst used in step c) may advantageously be reduced. One of the preferred methods for reducing the metal is treatment at a temperature of between 150°C and 650°C and a total pressure of between 0.1 and 25 MPa. For example, a reduction may comprise a steady stage at 150°C for two hours, then a rise in temperature up to 450°C at a rate of 1°C/min and then a steady stage of two hours at 450°C; during the reduction step, the hydrogen flow rate may be 1000 Nm³ of hydrogen per m³ of catalyst and the total pressure may be kept constant at 0.1 MPa. Any reduction method can advantageously be envisaged, either in situ (the reduction of the catalyst is carried out in the unit where the catalytic reaction is carried out), or ex situ (the reduction is carried out outside the unit where the catalytic reaction is carried out, therefore before the catalyst is loaded into the unit).

In accordance with the invention, the process comprises a step d) of separating at least a portion and preferably all of the effluent obtained from step c).

Said step d) makes it possible to separate at least a “light” gaseous fraction and at least a hydrocarbon liquid effluent. Optionally, step d) also makes it possible to separate at least a portion of the residual water and preferably all of the residual water.

Said “light” gaseous fraction comprises at least the hydrogen not converted by the reactions described in step c) and at least a portion of the C_4^- cracking products, that is to say C_1 to C_4 compounds preferably having a final boiling point below 20°C. The aim of this step is to separate the gases from the liquids. More particularly, the aim is to recover at least the hydrogen-rich gases, at least one hydrocarbon liquid effluent rich in branched paraffins and optionally an aqueous liquid effluent containing a small amount of residual water (residual water generated either by the hydrodeoxygenation of the residual oxygen compounds present in the feedstock of step c) or by slight entrainment of water from step b) to step c)).

The separation step d) described may advantageously be performed via any method known to those skilled in the art, for instance the combination of one or more high-pressure and/or low-pressure separating vessels operated under hot or cold conditions, and/or high-pressure and/or low-pressure stripping. Said step d) does not require the use of a distillation column. Said step d) is advantageously not performed in a distillation column.

In accordance with the invention, the process comprises a step e) of stabilizing the liquid hydrocarbon effluent obtained from step d).

The stabilization step e) may be performed via any method known to those skilled in the art, for instance, a stripping step, advantageously using steam, and/or separation in separating vessels, and/or a combination of these options.

Preferably, said stabilization step is a step of stripping at least a portion, and preferably all, of the hydrocarbon liquid effluent obtained from step d).

According to the invention, said stripping step enables the separation of a light gas fraction, a naphtha hydrocarbon cut and alternatively a gas oil cut or a kerosene cut.

More particularly, the cut-off point between the middle distillates and naphtha cuts is adjustable within the 80–160°C range, preferably about 120°C. The naphtha cut thus contains compounds with a boiling point below 120°C, and the middle distillate cut contains compounds with a boiling point above 120°C. In the presence of the specific catalyst in accordance with the invention, the operating conditions of hydroconversion step c) may advantageously be

adjusted to produce a middle distillate cut oriented alternatively towards either bio-kerosene production or bio-gas oil production.

The upgrading of the bio-naphtha cut is not the subject of the present invention, but this cut may advantageously be sent to a steam cracking or reforming unit or else upgraded as a gasoline base in admixture with other gasoline bases.

The process according to the invention advantageously does not comprise a hydrocracking step downstream of step c) of hydroconversion of part or all of the effluent obtained from step c).

Preferably, the process according to the invention does not comprise a step of final fractionation or distillation of the effluent obtained from the hydroconversion step c).

According to the invention, the process according to the invention does not comprise recycling, into the hydrotreating step a) and/or hydroconversion step c), of all or part of the effluents obtained from the hydroconversion step c), of the effluents obtained from the separation step d) and of the effluents obtained from the stabilization step e).

Description of the figures

Figure 1 represents the various steps in the two-step process for producing renewable kerosene or gas oil, using a specific catalyst for the hydroconversion step c).

The feedstock from renewable sources is sent via pipe 1, in admixture with supplemental and/or recycled hydrogen (pipe 2), into a hydrotreating step a) (enabling the hydrogenation, hydrodeoxygenation, hydrodenitration and hydrodesulfurization of the feedstock). The hydrotreated effluent obtained from the hydrotreating step a) is drawn off via pipe 3 and sent to the three-phase separation step b), which allows the separation of at least a hydrogen-rich gaseous effluent (pipe 4) and at least a hydrocarbon liquid effluent (pipe 5). Step b) also allows the removal of at least a portion of the water produced by the hydrodeoxygenation reactions and, preferably, all of this water (pipe 6).

The hydrocarbon liquid effluent (5) is sent to a hydroconversion (hydroisomerization and/or hydrocracking) step c), in the presence of a supplemental and/or recycled hydrogen stream (pipe 7) to produce an effluent (pipe 8) which is sent to a three-phase separation step d) which allows the separation of a hydrogen-rich gaseous effluent (9) which may also contain light

fractions such as the C₁-C₄ cut, and at least one hydrocarbon liquid effluent (10). Said step also allows the removal of at least a portion of the water, and preferably all of the residual water (11).

The liquid hydrocarbon effluent obtained from step d) (10) is sent to a step e) involving stabilization by stripping, which allows the separation of a light gas fraction (12), a naphtha hydrocarbon cut (13) and a middle distillate cut (14), i.e. alternatively a gas oil cut or a kerosene cut. No recycling step is performed for the effluents from steps c), d) and e).

The examples below illustrate the invention without limiting the scope thereof.

EXAMPLES

Example 1: Preparation of a hydrotreating catalyst (C1)

The catalyst is an industrial catalyst based on nickel, molybdenum and phosphorus on alumina with contents of molybdenum oxide MoO_3 of 22% by weight, of nickel oxide NiO of 4% by weight and of phosphorus oxide P_2O_5 of 5% by weight relative to the total weight of the finished catalyst.

Example 2: Preparation of a hydroconversion catalyst not in accordance with the invention (C2)

The silica-alumina powder is prepared according to the synthetic protocol described in patent EP 1 415 712 A. The amounts of orthosilicic acid and of aluminium hydrate are chosen so as to have a composition containing 70% by weight of alumina Al_2O_3 and 30% by weight of silica SiO_2 in the final solid.

This mixture is rapidly homogenized in a commercial colloidal mill in the presence of nitric acid so that the nitric acid content of the suspension leaving the mill is 8% relative to the silica-alumina mixed solid. Next, the suspension is dried conventionally in an atomizer conventionally from 300°C to 60°C. The powder thus prepared is formed in a Z-shaped blade mixer in the presence of 8% nitric acid relative to the anhydrous product. The extrusion is performed by passing the paste through a die equipped with orifices 1.4 mm in diameter. The extrudates thus obtained are oven-dried at 140°C, then calcined under a stream of dry air at 550°C, and then calcined at 850°C in the presence of steam.

The characteristics of the support thus prepared are as follows:

- a mean mesopore diameter, measured by mercury porosimetry, of 7.7 nm,
- a total pore volume of 0.49 ml/g,
- a mesopore volume of 0.48 ml/g,
- a volume of macropores, the diameter of which is greater than 50 nm, of less than 0.02 ml/g,
- a BET surface area of 240 m^2/g .

50 g of silica-alumina extrudates are then subjected to a step of dry impregnation with an aqueous solution of platinum tetramine nitrate, left to mature in a water maturator for 24 hours at ambient temperature, then calcined for two hours in dry air in a traversed bed at 450°C (temperature increase ramp of 5°C/minute). The content by weight of platinum in the finished catalyst after calcination is 0.57%, its dispersion measured by H_2/O_2 titration is 26%, and its coefficient of distribution measured by Castaing microprobe is 0.87.

Example 3: Preparation of a hydroconversion catalyst in accordance with the invention (C3)

Synthesis of the IZM-2 zeolite

The IZM-2 zeolite was synthesized in accordance with the teaching of patent FR 2 918 050 B. A colloidal silica suspension known under the trade name Ludox HS-40, sold by Aldrich, is incorporated into a solution composed of sodium hydroxide (Prolabo), 1,6-bis(methylpiperidinium)hexane dibromide structuring agent, aluminum hydroxide (Aldrich) and deionized water. The molar composition of the mixture is as follows: 1 SiO_2 ; 0.0060 Al_2O_3 ; 0.1666 Na_2O ; 0.1666 1,6-bis(methylpiperidinium)hexane; 33.3333 H_2O . The mixture is stirred vigorously for 30 minutes. The mixture is then transferred, after homogenization, into a Parr autoclave. The autoclave is heated for 5 days at 170°C with spindle stirring (30 rpm). The product obtained is filtered, washed with deionized water to reach neutral pH and then dried overnight at 100°C in an oven. The solid is then introduced into a muffle furnace and calcined therein so as to remove the structuring agent. The calcination cycle comprises a temperature increase up to 200°C, a hold of two hours at this temperature, a temperature increase up to 550°C, followed by a hold of eight hours at this temperature and finally a return to ambient temperature. The temperature rises are performed at a rate of 2°C/minute. The solid thus obtained is then refluxed for 2 hours in an aqueous ammonium nitrate solution (10 ml of solution per gram of solid, ammonium nitrate concentration of 3M) so as to exchange the sodium alkaline cations with ammonium ions. This refluxing step is performed four times with fresh ammonium nitrate solution, and the solid is then filtered off, washed with deionized water and dried in an oven overnight at 100°C. Finally, to obtain the zeolite in its acid (protonated H^+) form, a step of calcination is performed at 550°C for 10 hours (temperature increase ramp of 2°C/minute) in a traversed bed under dry air (2 normal litres per hour and per gram of solid). The solid thus obtained was analysed by X-ray diffraction and identified as consisting of IZM-2 zeolite. Characterizations via the ^{27}Al isotope NMR, x-ray fluorescence and ICP methods afford the following results for IZM-2:

- weight percentage of hexacoordinated aluminium atoms Al^{VI} : 5%,
- ratio of the number of moles of silicon divided by the number of moles of lattice aluminium, in mol/mol, Si/Al: 72
- ratio of the number of moles of sodium divided by the number of moles of lattice aluminium, in mol/mol, Na/Al: 0.03.

Preparation of the IZM-2/alumina support

The IZM-2/alumina support is obtained by blending and extrusion of the zeolite IZM-2 with a Pural SB3 alumina gel. The blended paste is extruded through a quadrilobal die 1.8 mm in diameter. After drying in an oven overnight at 110°C, the extrudates are calcined at 500°C for two hours (temperature increase ramp of 5°C/min) in a crossed bed under dry air (2 normal litres per hour and per gram of solid). The weight content of IZM-2 zeolite in the support after calcination is 13% by weight.

Platinum impregnation onto the IZM-2/alumina support

Platinum is impregnated by dry impregnation of the support with an aqueous solution containing platinum tetramine nitrate $Pt(NH_3)_4(NO_3)_2$. 50 g of support are used, and are dry-impregnated in a pan. After impregnation, the solid is left to mature for at least five hours in the laboratory air and is then dried overnight in an oven at 110°C. Finally, a calcination step is performed under a flow of dry air (2 normal litres per hour and per gram of solid) in a tubular oven under the following conditions:

- temperature rise from room temperature to 450°C at 5°C/min;
- steady stage of two hours at 450°C;
- decrease to room temperature.

The Pt content measured by XRF on the calcined catalyst C3 is 0.29% by weight, its dispersion measured by H_2/O_2 assay is 52%, and its coefficient of distribution measured by Castaing microprobe is 0.91.

Example 4: Preparation of a hydroconversion catalyst in accordance with the invention (C4)

ZSM-12 Zeolite

ZSM-12 zeolite is supplied by the company Zeolyst. The solid was analysed by X-ray diffraction and identified as consisting of ZSM-12 zeolite. Characterizations via the ^{27}Al isotope NMR, x-ray fluorescence and ICP methods afford the following results for ZSM-12:

- weight percentage of hexacoordinated aluminium atoms Al^{VI} : 0%;
- ratio of the number of moles of silicon divided by the number of moles of lattice aluminium, in mol/mol, Si/Al: 43;
- ratio of the number of moles of sodium divided by the number of moles of lattice aluminium, in mol/mol, Na/Al: 0.009.

Preparation of the ZSM-12/alumina support

The ZSM-12/alumina support is obtained by blending and extrusion of the zeolite ZSM-12 with a Pural SB3 alumina gel. The blended paste is extruded through a quadrilobal die 1.8 mm in diameter. After drying in an oven overnight at 110°C, the extrudates are calcined at 500°C for two hours (temperature increase ramp of 5°C/min) in a crossed bed under dry air (2 normal litres per hour and per gram of solid). The weight content of ZSM-12 zeolite in the support after calcination is 6% by weight.

Platinum impregnation onto the ZSM-12/alumina support

Platinum is impregnated by dry impregnation of the support with an aqueous solution containing platinum tetramine nitrate $\text{Pt}(\text{NH}_3)_4(\text{NO}_3)_2$. 50 g of support are used, and are dry-impregnated in a pan. After impregnation, the solid is left to mature for at least five hours in the laboratory air and is then dried overnight in an oven at 110°C. Finally, a calcination step is performed under a flow of dry air (2 normal litres per hour and per gram of solid) in a tubular oven under the following conditions:

- temperature rise from room temperature to 450°C at 5°C/min;
- steady stage of two hours at 450°C;

- decrease to room temperature.

The Pt content measured by XRF on the calcined catalyst C4 is 0.31% by weight, its dispersion measured by H_2/O_2 assay is 38%, and its coefficient of distribution measured by Castaing microprobe is 0.87.

Example 5: Preparation of a hydroconversion catalyst not in accordance with the invention (C5)

EU-2 Zeolite

EU-2 zeolite is supplied by the company Zeolyst. The solid was analysed by X-ray diffraction and identified as consisting of EU-2 zeolite. Characterizations via the ^{27}Al isotope NMR, x-ray fluorescence and ICP methods afford the following results for EU-2:

- weight percentage of hexacoordinated aluminium atoms Al^{VI} : 9%,
- ratio of the number of moles of silicon divided by the number of moles of lattice aluminium, in mol/mol, Si/Al: 82
- ratio of the number of moles of sodium divided by the number of moles of lattice aluminium, in mol/mol, Na/Al: < 0.005.

Preparation of the EU-2/alumina support

The EU-2/alumina support is obtained by blending and extrusion of the EU-2 zeolite with a Pural SB3 alumina gel. The blended paste is extruded through a quadrilobal die 1.8 mm in diameter. After drying in an oven overnight at 110°C, the extrudates are calcined at 500°C for two hours (temperature increase ramp of 5°C/min) in a crossed bed under dry air (2 normal litres per hour and per gram of solid). The weight content of EU-2 zeolite in the support after calcination is 38% by weight.

Platinum impregnation onto the EU-2/alumina support

Platinum is impregnated by dry impregnation of the support with an aqueous solution containing platinum tetramine nitrate $\text{Pt}(\text{NH}_3)_4(\text{NO}_3)_2$. 50 g of support are used, and are dry-impregnated in a pan. After impregnation, the solid is left to mature for at least five hours in the laboratory air and is then dried overnight in an oven at 110°C. Finally, a calcination step is

performed under a flow of dry air (2 normal litres per hour and per gram of solid) in a tubular oven under the following conditions:

- temperature rise from room temperature to 450°C at 5°C/min,
- steady stage of two hours at 450°C,
- decrease to room temperature.

The Pt content measured by XRF on the calcined catalyst C5 is 0.31% by weight, its dispersion measured by H_2/O_2 titration is 56%, and its coefficient of distribution measured by Castaing microprobe is 0.91.

Example 6: Hydrotreating of a feedstock obtained from a renewable source according to a process in accordance with the invention

Pre-refined rapeseed oil with a mass per unit volume at 15°C of 920 kg/m³, having an oxygen content of 11% by weight, is hydrotreated in a reactor which is temperature-regulated so as to ensure isothermal functioning and with a fixed bed containing 190 ml of hydrotreating catalyst C1, the catalyst being sulfide-treated beforehand. The cetane number is 35 and the fatty acid distribution of the rapeseed oil is detailed in table 1. Prior to the hydrotreating step, said feedstock is supplemented with dimethyl disulfide so as to adjust its sulfur content to 50 ppm by weight.

Table 1

Fatty acid composition	(% m/m)
14:0	0.1
16:0	5.0
16:1	0.3
17:0	0.1
17:1	0.1
18:0	1.5
18:1 trans	< 0.1
18:1 cis	60.1
18:2 trans	< 0.1
18:2 cis	20.4
18:3 trans	< 0.1
18:3 cis	9.6
20:0	0.5
20:1	1.2
22:0	0.3
22:1	0.2
24:0	0.1
24:1	0.2

Table 1: Characteristics of the rapeseed oil renewable feedstock used as feedstock for the hydrotreating step

Before hydrotreating the feedstock, the catalyst is sulfide-treated *in situ* in the unit, with a distillation gas oil supplemented with 2% by weight of dimethyl disulfide, under a total pressure of 5.1 MPa, a hydrogen/supplemented gas oil ratio of 700 Nm³ per m³. The volume of supplemented gas oil per volume of catalyst and per hour is set at 1 h⁻¹. Sulfurization is performed for 12 hours at 350°C, with a temperature increase ramp of 10°C per hour.

After sulfurization, the operating conditions of the unit are adjusted so as to perform the hydrotreating of the feedstock:

- HSV (volume of feedstock/volume of catalyst/hour): 1 h⁻¹,

- total working pressure: 5.1 MPa,
- hydrogen/feedstock ratio: 700 Nm³ of hydrogen/m³ of feedstock,
- temperature: 310°C.

The hydrogen employed is supplied by Air Products and has a purity of greater than 99.999% by volume.

Step b): separation of the effluent obtained from step a)

All of the hydrotreated effluent obtained from step a) is separated using a gas/liquid separator so as to recover a light fraction predominantly containing hydrogen, propane, water in vapor form, carbon oxides (CO and CO₂) and ammonia, and a liquid hydrocarbon effluent predominantly consisting of linear hydrocarbons. The water present in the hydrocarbon liquid effluent is removed by decantation. The hydrocarbon liquid effluent thus obtained has an atomic oxygen content of less than 80 ppm by weight, said atomic oxygen content being measured via the infrared adsorption technique described in patent application US 2009/0018374, and a sulfur content of 2 ppm by weight and a nitrogen content of less than 1 ppm by weight, said nitrogen and sulfur contents being measured, respectively, by chemiluminescence and by UV fluorescence. Said hydrocarbon liquid effluent has a mass per unit volume at 15°C of 791 kg/m³. The hydrocarbon liquid effluent is composed of paraffins; its composition, measured by gas chromatography, is given in table 2.

Table 2

Distribution of the n-paraffins by carbon number	(% m/m)
nC8	0.01
nC9	0.00
nC10	0.01
nC11	0.01
nC12	0.01
nC13	0.02
nC14	0.06
nC15	0.97
nC16	4.09
nC17	17.70
nC18	72.77
nC19	0.67
nC20	1.56
nC21	0.18
nC22	0.56
nC23	0.09
nC24	0.23
nC25	0.02
nC26	0.02

Table 2: Composition of the hydrocarbon liquid effluent used as feedstock for the hydroconversion

Non-compliant Example 7 Hydroconversion of the liquid hydrocarbon effluent obtained from Example 6 according to a compliant process using the hydroconversion catalyst C2 not in accordance with the invention

Example 7 is not in accordance with the invention in that it illustrates a process that is in accordance with the invention but using a C2 catalyst that is not in accordance with the invention in the hydroconversion step c).

The hydroconversion catalyst C2 was evaluated in a pilot plant, representative in terms of implementation (reaction operating conditions, effluent separation quality), of the industrial process according to the invention. The various steps and unit operations of the pilot plant, which mimic the industrial implementation of the process according to the invention, are described hereinbelow.

The hydroconversion of the hydrocarbon liquid effluent obtained from Example 6 is performed in a reactor which is temperature-regulated so as to ensure isothermal functioning and with a fixed bed containing 50 ml of hydroconversion catalyst C2, the catalyst being preactivated by reduction. Given the nature of the catalyst C2 (noble metal), any injection of sulfur is to be avoided.

The catalyst C2 undergoes a reduction step under hydrogen performed at 5.1 MPa and 400°C (2-hour steady stage).

After reduction, the operating conditions of the unit are adjusted so as to perform the hydroconversion of the hydrocarbon liquid effluent under the following range of operating conditions:

- HSV (volume of feedstock/volume of catalyst/hour): 0.5 h^{-1} ,
- total working pressure: 5.1 MPa,
- hydrogen/feedstock ratio: 350 Nm^3 of hydrogen/ m^3 of feedstock.

The hydrogen used and entering the hydroconversion step is supplied by Air Products: it has a purity of greater than 99.999% by volume and is free of hydrogen sulfide.

On leaving the hydroconversion reactor, the reaction effluent is sent to a gas-liquid separation step performed using a flash vessel operated at a pressure comparable to that of the hydroconversion reactor. The hydrogen-rich gas phase is sent to the unit's gas outlet. The liquid hydrocarbon phase is depressurized and sent to a stripper for stabilizing the liquid effluent; the gas phase collected at the top of the stripper is also sent to the unit's gas outlet.

At the unit's gas outlet, inline analysis by gas chromatography and a gas meter makes it possible to calculate the mass of light hydrocarbons produced (essentially hydrocarbons containing 1 to 5 carbon atoms) and present in the hydrogen stream. The liquid effluent is weighed separately, topped off at 120°C to remove the naphtha fraction, and the 120°C+ liquid

effluent is then reweighed and analysed, notably by measuring the cold filter plugging point (CFPP) (NF EN 116) targeting gas oil and by measuring the crystal disappearance point (ASTM D5972) targeting kerosene.

Temperature steady stages in the 250-400°C range were applied so as to adjust the severity of the hydroconversion. Measurement (typically daily) of the cold filter plugging point (when the process is operating targeting diesel) or the crystal disappearance point (when the process is operating targeting kerosene) of the 120°C⁺ liquid effluent allows the performance of the catalyst to be monitored at each temperature steady stage. For each temperature, the test duration is extended until a stable cold filter plugging point (for the process operating targeting gas oil) or a stable crystal disappearance point (for the process operating targeting kerosene) is obtained.

Once the stable cold filter plugging point or stable crystal disappearance point has been reached, the yield of 120°C⁺ cut (middle distillates) is determined by means of the following calculation:

120°C⁺ (middle distillate) yield = [(mass of 120°C⁺ liquid effluent) / (mass of feedstock)] × 100, where the feedstock corresponds here to the hydrocarbon effluent obtained from Example 6.

The mass of 120°C⁺ liquid effluent corresponds to the amount of 120°C⁺ liquid accumulated over a certain period of time, typically 24 hours, and the mass of feedstock corresponds to the amount of feedstock injected into the hydroconversion reactor over the same period of time.

The temperature was adjusted in the reactor to achieve the following cold property targets for the fuels: summer gas oil, winter gas oil and kerosene. For each target, the features obtained for the 120°C⁺ cut of interest and according to Example 7, along with the associated operating conditions, are reported in the Summary Table 3. The various cold property targets are achieved by varying the reaction temperature, all other things being equal. The reference temperature is the temperature required to achieve the summer gas oil target (CFPP of 0°C); it is marked as “base” in Table 3. A 10°C increase in the “base” temperature thus allows the winter gas oil target to be achieved, and a 25°C increase in the “base” temperature allows the kerosene target to be achieved. Logically, the yield of the 120°C⁺ cut decreases as the reaction temperature increases. The yield of the 120°C⁺ cut is thus 97% by weight for the summer gas oil target, 91% by weight for the winter gas oil target, and 67% for the kerosene target.

Non-compliant Example 8 Hydroconversion of the liquid hydrocarbon effluent obtained from Example 6 according to a process using the hydroconversion catalyst C5 not in accordance with the invention

Example 8 is not in accordance with the invention in that it illustrates a process that is in accordance with the invention but using a C5 catalyst that is not in accordance with the invention in the hydroconversion step c).

The hydroconversion catalyst C5 was evaluated in a pilot plant, representative in terms of implementation (reaction operating conditions, effluent separation quality), of the industrial process according to the invention. The various steps and unit operations of the pilot plant, which mimic the industrial implementation of the process according to the invention, are described hereinbelow.

The hydroconversion of the hydrocarbon liquid effluent obtained from Example 6 is performed in a reactor which is temperature-regulated so as to ensure isothermal functioning and with a fixed bed containing 50 ml of hydroconversion catalyst C5, the catalyst being preactivated by reduction. Given the nature of the catalyst C5 (noble metal), any injection of sulfur is to be avoided.

The catalyst C5 undergoes a reduction step under hydrogen performed at 5.1 MPa and 400°C (2-hour steady stage).

After reduction, the operating conditions of the unit are adjusted so as to perform the hydroconversion of the hydrocarbon liquid effluent under a range of operating conditions identical to that of Example 7 and in the strict absence of sulfur, i.e.:

- HSV (volume of feedstock/volume of catalyst/hour) = 0.5 h^{-1} ,
- total working pressure: 5.1 MPa,
- hydrogen/feedstock ratio: 350 Nm^3 of hydrogen/ m^3 of feedstock.

The hydrogen used and entering the hydroconversion step is supplied by Air Products: it has a purity of greater than 99.999% by volume and is free of hydrogen sulfide.

The daily measurements taken and the operating procedure are identical to those detailed in Example 7.

The temperature steady stages are adjusted within the 250-400°C range so as to achieve the same fuel targets as in Example 7. For each target, the features obtained for the 120°C⁺ cut of interest and according to Example 8 are reported in the Summary Table 3. The reference temperature is the temperature required to achieve the summer gas oil target for the non-compliant catalyst C2 (CFPP of 0°C). As with catalyst C2, an increase in temperature allows the various cold property targets to be achieved.

For a given cold property target, catalyst C5 is consistently less active than catalyst C2.

For the summer gas oil target, the temperature used with catalyst C5 is 5°C higher than that used for catalyst C2.

For the winter gas oil target, the temperature used with catalyst C5 is 6°C higher than that used for catalyst C2.

For the kerosene target, the temperature used with catalyst C5 is 4°C higher than that used for catalyst C2.

However, catalyst C5 affords yields of 120°C⁺ cut greater than or equal to those of catalyst C2 for a given cold property target.

For the summer gas oil target, the yield of the 120°C⁺ cut with catalyst C5 is equal to that obtained with catalyst C2.

For the winter gas oil target, the yield of 120°C⁺ cut with catalyst C5 is 3 percentage points higher than that obtained with catalyst C2.

For the kerosene target, the temperature used with catalyst C5 is 9 degrees higher than that obtained with catalyst C2.

Compliant Example 9: Hydroconversion of the liquid hydrocarbon effluent obtained from Example 6 according to a compliant process using the hydroconversion catalyst C3 in accordance with the invention

The hydroconversion catalyst C3 was evaluated in a pilot plant, representative in terms of implementation (reaction operating conditions, effluent separation quality), of the industrial process according to the invention. The various steps and unit operations of the pilot plant,

which mimic the industrial implementation of the process according to the invention, are described hereinbelow.

The hydroconversion of the hydrocarbon liquid effluent obtained from Example 6 is performed in a reactor which is temperature-regulated so as to ensure isothermal functioning and with a fixed bed containing 50 ml of hydroconversion catalyst C3, the catalyst being preactivated by reduction. Given the nature of the catalyst C3 (noble metal), any injection of sulfur is to be avoided.

The catalyst C3 undergoes a reduction step under hydrogen performed at 5.1 MPa and 400°C (2-hour steady stage).

After reduction, the operating conditions of the unit are adjusted so as to perform the hydroconversion of the hydrocarbon liquid effluent under a range of operating conditions identical to that of Example 7 and in the strict absence of sulfur, i.e.:

- HSV (volume of feedstock/volume of catalyst/hour) = 0.5 h^{-1} ,
- total working pressure: 5.1 MPa,
- hydrogen/feedstock ratio: 350 Nm³ of hydrogen/m³ of feedstock.

The hydrogen used and entering the hydroconversion step is supplied by Air Products: it has a purity of greater than 99.999% by volume and is free of hydrogen sulfide.

The daily measurements taken and the operating procedure are identical to those detailed in Example 7.

The temperature steady stages are adjusted within the 250-400°C range so as to achieve the same fuel targets as in Example 7. For each target, the features obtained for the 120°C⁺ cut of interest and according to Example 9 are reported in the Summary Table 3.

The reference temperature is the temperature required to achieve the summer gas oil target for the non-compliant catalyst C2 (CFPP of 0°C). As with catalyst C2, an increase in temperature allows the various cold property targets to be achieved.

For a given cold property target, catalyst C3 is consistently more active than catalyst C2.

For the summer gas oil target, the temperature used with catalyst C3 is 3°C lower than that used for catalyst C2.

For the winter gas oil target, the temperature used with catalyst C3 is 6°C lower than that used for catalyst C2.

For the kerosene target, the temperature used with catalyst C3 is 10°C lower than that used for catalyst C2.

Furthermore, catalyst C3 affords yields of 120°C⁺ cut greater than or equal to those of catalyst C2 for a given cold property target.

For the summer gas oil target, the yield of the 120°C⁺ cut with catalyst C3 is equal to that obtained with catalyst C2.

For the winter gas oil target, the yield of 120°C⁺ cut with catalyst C3 is 5 percentage points higher than that obtained with catalyst C2.

For the kerosene target, the temperature used with catalyst C3 is 16 degrees higher than that obtained with catalyst C2.

The use of a catalyst C3 in accordance with the invention thus allows both an increase in activity and an increase in yield of 120°C⁺ cut for a given cold property target.

Compliant Example 10: Hydroconversion of the liquid hydrocarbon effluent obtained from Example 6 according to a process using the hydroconversion catalyst C4 in accordance with the invention

The hydroconversion catalyst C4 was evaluated in a pilot plant, representative in terms of implementation (reaction operating conditions, effluent separation quality), of the industrial process according to the invention. The various steps and unit operations of the pilot plant, which mimic the industrial implementation of the process according to the invention, are described hereinbelow.

The hydroconversion of the hydrocarbon liquid effluent obtained from Example 6 is performed in a reactor which is temperature-regulated so as to ensure isothermal functioning and with a fixed bed containing 50 ml of hydroconversion catalyst C4, the catalyst being preactivated by

reduction. Given the nature of the catalyst C4 (noble metal), any injection of sulfur is to be avoided.

The catalyst C4 undergoes a reduction step under hydrogen performed at 5.1 MPa and 400°C (2-hour steady stage).

After reduction, the operating conditions of the unit are adjusted so as to perform the hydroconversion of the hydrocarbon liquid effluent under a range of operating conditions identical to that of Example 7 and in the strict absence of sulfur, i.e.:

- HSV (volume of feedstock/volume of catalyst/hour) = 0.5 h⁻¹,
- total working pressure: 5.1 MPa,
- hydrogen/feedstock ratio: 350 Nm³ of hydrogen/m³ of feedstock.

The hydrogen used and entering the hydroconversion step is supplied by Air Products: it has a purity of greater than 99.999% by volume and is free of hydrogen sulfide.

The daily measurements taken and the operating procedure are identical to those detailed in Example 7.

The temperature steady stages are adjusted within the 250-400°C range so as to achieve the same fuel targets as in Examples 7, 8 and 9.

For each target, the features obtained for the 120°C⁺ cut of interest and according to Example 10 are reported in the Summary Table 3.

The reference temperature is the temperature required to achieve the summer gas oil target for the non-compliant catalyst C2 (CFPP of 0°C). As with catalyst C2, an increase in temperature allows the various cold property targets to be achieved.

For a given cold property target, catalyst C4 is consistently more active than catalyst C2.

For the summer gas oil target, the temperature used with catalyst C4 is 10°C lower than that used for catalyst C2.

For the winter gas oil target, the temperature used with catalyst C4 is 13°C lower than that used for catalyst C2.

For the kerosene target, the temperature used with catalyst C4 is 16°C lower than that used for catalyst C2.

Furthermore, catalyst C4 affords yields of 120°C⁺ cut greater than or equal to those of catalyst C2 for a given cold property target.

For the summer gas oil target, the yield of 120°C⁺ cut with catalyst C4 is 1 percentage point higher than that obtained with catalyst C2.

For the winter gas oil target, the yield of 120°C⁺ cut with catalyst C4 is 6 percentage points higher than that obtained with catalyst C2.

For the kerosene target, the temperature used with catalyst C4 is 16 degrees higher than that obtained with catalyst C2.

The use of a catalyst C4 in accordance with the invention thus allows both an increase in activity and an increase in yield of 120°C⁺ cut for a given cold property target.

Table 3

		Non-compliant Example 7	Non-compliant Example 8	Compliant Example 9	Compliant Example 10
Target summer gas oil					
Operating conditions					
Temperature	°C	Base	Base + 5	Base - 3	Base - 10
Total pressure (MPa)	MPa	5.1	5.1	5.1	5.1
H ₂ /feedstock ratio	Nm ³ /m ³	350	350	350	350
Volume feedstock/volume catalyst/h	h ⁻¹	0.5	0.5	0.5	0.5
	weight%/feedstock	97	97	97	98
Yield of 120°C⁺ middle distillate					
Properties of the 120°C⁺ cut					
Cold filter plugging point	°C	0	0	0	0
Density at 15°C	g/cm ³	0.783	0.784	0.784	0.785
Cetane number	-	> 60	> 60	> 60	> 60
Sulfur content	ppm by weight	< 10	< 10	< 10	< 10
Nitrogen content	ppm by weight	< 1	< 1	< 1	< 1
Target winter gas oil					
Operating conditions					
Temperature	°C	Base + 10	Base + 16	Base + 4	Base - 3
Total pressure	MPa	5.1	5.1	5.1	5.1
H ₂ /feedstock ratio	Nm ³ /m ³	350	350	350	350
Volume feedstock/volume catalyst/h	h ⁻¹	0.5	0.5	0.5	0.5
	weight%/feedstock	91	94	96	97
Yield of 120°C⁺ middle distillate					
Properties of the 120°C⁺ cut					
Cold filter plugging point	°C	-14	-15	-16	-17
Density at 15°C	g/cm ³	0.776	0.780	0.782	0.784
Cetane number	-	> 60	> 60	> 60	-19
Sulfur content	ppm by weight	< 10	< 10	< 10	< 10
Nitrogen content	ppm by weight	< 1	< 1	< 1	< 1
Target kerosene					
Operating conditions					
Temperature	°C	Base + 25	Base + 29	Base + 15	Base + 9
Total pressure	MPa	5.1	5.1	5.1	5.1
H ₂ /feedstock ratio	Nm ³ /m ³	350	350	350	350
Volume feedstock/volume catalyst/h	h ⁻¹	0.5	0.5	0.5	0.5
	weight%/feedstock	67	76	83	83
Yield of 120°C⁺ middle distillate					
Properties of the 120°C⁺ cut					

Crystal disappearance point	°C	-50	-48	-52	-51
Density at 15°C	g/cm ³	0.750	0.768	0.771	0.772
Smoke point	mm	> 25	> 25	> 25	> 25
Sulfur content	ppm by weight	< 10	< 10	< 10	< 10
Nitrogen content	ppm by weight	< 1	< 1	< 1	< 1

Table 3: Yields and properties of the summer and winter gas oil cuts and kerosene cuts obtained for the various examples of the invention.

CLAIMS

1. Process for treating a feedstock from a renewable source to produce alternatively a gas oil cut and a kerosene cut, said process comprising at least the following steps and preferably consisting of the following steps:

a) a step of hydrotreating said feedstock in the presence of a catalyst in a fixed bed, said catalyst comprising a hydrogenating function and an oxide support, at a temperature of between 200 and 450°C, at a pressure of between 1 MPa and 10 MPa, at an hourly space velocity of between 0.1 h⁻¹ and 10 h⁻¹ and in the presence of a total amount of hydrogen mixed with the feedstock such that the hydrogen/feedstock ratio is between 70 and 2000 Nm³ of hydrogen/m³ of feedstock,

b) a step of separating at least a portion of the effluent obtained from step a) into at least a light gaseous fraction, at least a hydrocarbon liquid effluent, and at least an aqueous liquid effluent,

c) a step of hydroconversion of at least a portion of the hydrocarbon liquid effluent obtained from step b) in the presence of at least one difunctional hydroconversion catalyst in a fixed bed, said catalyst comprising a hydrogenating phase comprising at least one noble metal from group VIII of the Periodic Table chosen from platinum and palladium, and a support comprising at least one zeolite chosen from zeolites with the structure code MTW and the zeolite IZM-2, either alone or as a mixture, and at least one binder, said hydroconversion step being performed at a temperature of between 250°C and 500°C, at a pressure of between 1 MPa and 10 MPa, at an hourly space velocity of between 0.1 and 10 h⁻¹ and in the presence of a total amount of hydrogen mixed with the feedstock such that the hydrogen/feedstock ratio is between 70 and 1000 Nm³ of hydrogen/m³ of feedstock,

d) a step of separating at least a portion of the effluent obtained from step c) making it possible to separate at least a gaseous fraction and at least a hydrocarbon liquid effluent,

e) a step of stabilizing at least a portion of the hydrocarbon effluent obtained from step d), so as to separate three hydrocarbon cuts: a light gaseous fraction, a naphtha hydrocarbon cut and, alternatively, a gas oil cut or a kerosene cut, no step of recycling, into the hydrotreating step a) and/or the hydroconversion step c), of all or some of the effluents obtained from the hydroconversion step c), of the effluents from the separation step d) and/or of the effluents from the stabilization step e), being performed.

2. Process according to Claim 1, in which the feedstock derived from renewable sources is chosen from oils and fats of plant or animal origin, used cooking oils, oils of microbial origin, fish oils, long-chain paraffins derived from the Fischer-Tropsch process, whether crude or pretreated, or mixtures of such feedstocks, containing triglycerides and/or free fatty acids and/or esters.
3. Process according to either of Claims 1 and 2, in which, in step a), the feedstock is placed in contact with a catalyst in a fixed bed at a temperature of between 220 and 350°C, at a pressure of between 1 MPa and 6 MPa, and at an hourly space velocity of between 0.1 h⁻¹ and 10 h⁻¹. The feedstock is placed in contact with the catalyst in the presence of hydrogen and in the presence of a total amount of hydrogen mixed with the feedstock such that the hydrogen/feedstock ratio is between 150 and 1000 Nm³ of hydrogen/m³ of feedstock.
4. Process according to one of Claims 1 to 3, in which the hydrotreating catalyst used in step a) comprising a hydro-dehydrogenating function comprising at least one metal from group VIII and/or from group VIB, taken alone or as a mixture, and a support chosen from the group formed by alumina, silica, silica-aluminas, magnesia, clays and mixtures of at least two of these minerals.
5. Process according to one of Claims 1 to 4, in which the separation step b) is performed by combining one or more high-pressure and/or low-pressure separators, and/or high-pressure and/or low-pressure stripping.
6. Process according to one of Claims 1 to 5, in which the hydroconversion step c) operates at a temperature of between 250°C and 450°C and very preferably between 250°C and 400°C, at a pressure of between 2 MPa and 10 MPa and very preferably of between 3 MPa and 9 MPa, at an hourly space velocity advantageously between 0.2 and 7 h⁻¹ and very preferably between 0.5 and 5 h⁻¹, at a hydrogen flow rate such that the hydrogen/feedstock volume ratio is advantageously between 100 and 1000 Nm³ of hydrogen per m³ of feedstock and preferably between 150 and 1000 Nm³ of hydrogen per m³ of feedstock.
7. Process according to one of Claims 1 to 6, in which the Group VIII metal of the catalyst used in step c) is platinum.
8. Process according to one of Claims 1 to 7, in which the zeolite of structure code MTW used in the catalyst of step c) is ZSM-12.

9. Process according to one of Claims 1 to 8, in which the catalyst used in step c) comprises a hydrogenating phase comprising platinum and a support comprising an IZM-2 zeolite alone and an alumina binder.

10. Process according to one of Claims 1 to 8, in which the catalyst used in step c) comprises a hydrogenating phase comprising platinum and a support comprising a ZSM-12 zeolite alone and an alumina binder.

11. Process according to one of Claims 1 to 10, in which the stabilization step e) is a steam stripping step and/or separation in separating vessels and/or a combination of these options.

12. Process according to one of the preceding claims, in which the process does not comprise a step of final fractionation or distillation of the effluent obtained from the hydroconversion step c).

13. Process according to one of Claims 1 to 12, in which the process does not comprise a hydrocracking step downstream of step c) of hydroconversion of part or all of the effluent obtained from step c).