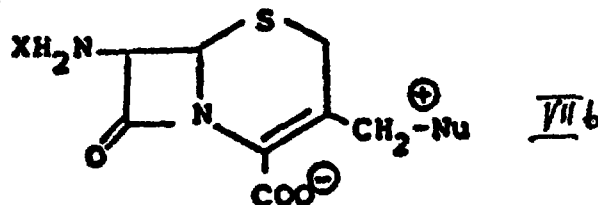


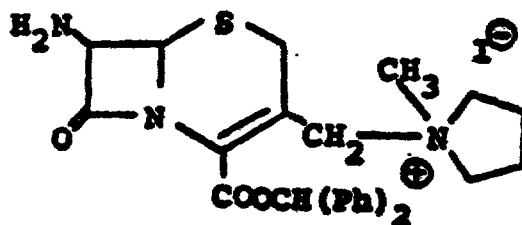
Description

This invention relates to temperature stable, crystalline salts of a cephalosporin intermediate having the formula VIIb

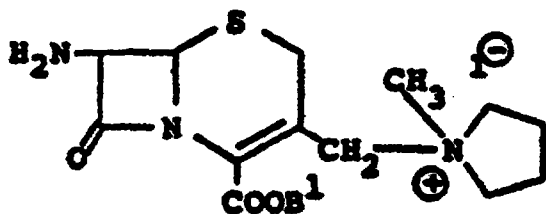


which are substantially free of the Δ^2 isomer, and which are convertible into broad spectrum cephalosporin antibiotics without the necessity of a carboxyl group deblocking step.

US-A-4,406,899, issued September 27, 1983 to Aburaki, et al., discloses compounds of the formulae

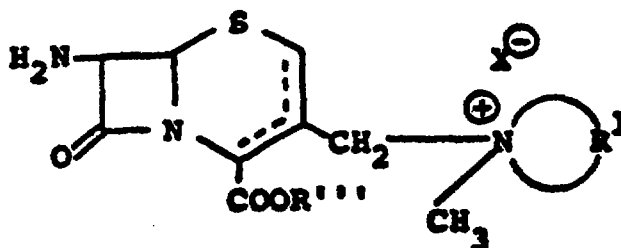


wherein Ph is phenyl, and



or an N-silyl derivative thereof, wherein B¹ is hydrogen or a conventional carboxyl-protecting group. These compounds are not exemplified in the patent, but are disclosed by structural formulae only, as intermediates in an alternate reaction scheme for the preparation of certain cephalosporins (by acylation and then deblocking of the protected carboxyl group). The reaction scheme actually exemplified did not utilize these compounds (and also required deblocking of the protected carboxyl group as the final step). Each of the final products of US-A-4,406,899 required a chromatographic purification step to separate the mixture of Δ^2 and Δ^3 isomers which were produced.

US-A-4,168,309, issued September 18, 1979 to Barry E. Ayres, discloses compounds of the formula

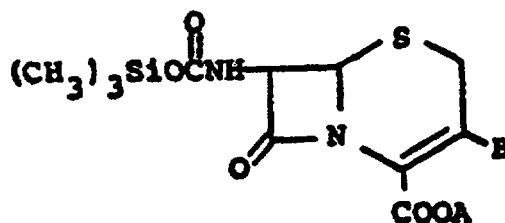


wherein R''' is a carboxyl-protecting group; R¹, taken together with the carbon atom to which it is attached, forms an optionally substituted saturated or partially saturated 4-10 membered heterocyclic ring which may contain one or more further hetero-atoms selected from O, N and S; the dotted line indicating a ceph-2-em or a ceph-3-em compound; and X⁻ representing an anion; and acid addition salts or an N-silylated derivative thereof.

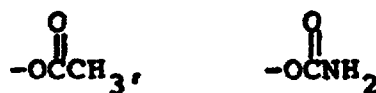
These compounds are not exemplified but are disclosed by structural formula only as intermediates in an alternative synthesis of certain cephalosporins (by acylation and then deblocking of the protected carboxyl group). The reaction scheme actually exemplified did not utilize these compounds (and also required deblocking of the protected carboxyl group as a final step).

The use of blocked carboxyl groups in the processes of the above two patents has the disadvantage of requiring deblocking after acylating whereby acyl group is lost in the deblocking step due to less than 100% yield in the deblocking step.

US-A-4,223,135, issued September 16, 1980 to Derek Walker *et al.*, discloses compounds of the formula



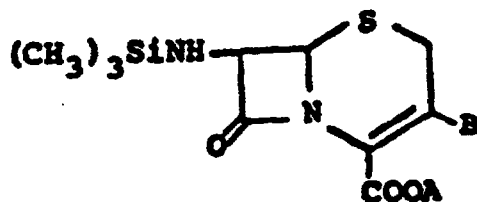
in which B is chloro, methoxy or -CH₂E, E is hydrogen,



or -SZ

in which Z is an optionally substituted 5- or 6- membered heterocyclic ring containing 2-4 N atoms and zero or one atom selected from O and S, the S atom of -SZ being connected to a carbon atom of the heterocyclic ring Z, and A is trimethylsilyl or an easily cleavable ester-protecting group.

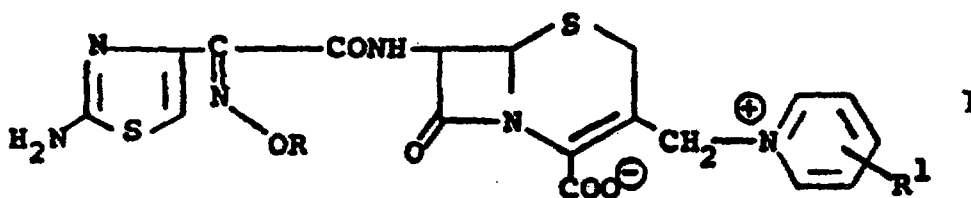
The compounds are prepared by adding dry carbon dioxide to a solution of a compound of the formula



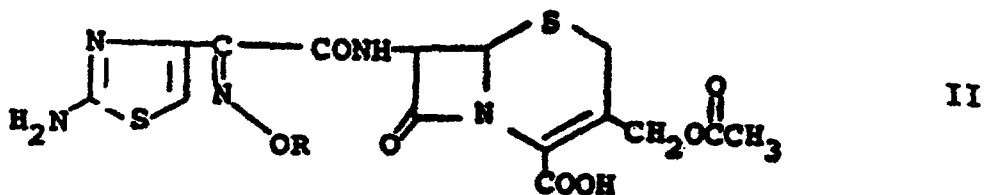
wherein A and B are as described above, in an inert organic solvent which is preferably methylene chloride.

US-A-4,316,017, which is a continuation-in-part of a division of US-A-4,223,135, has a substantially identical disclosure.

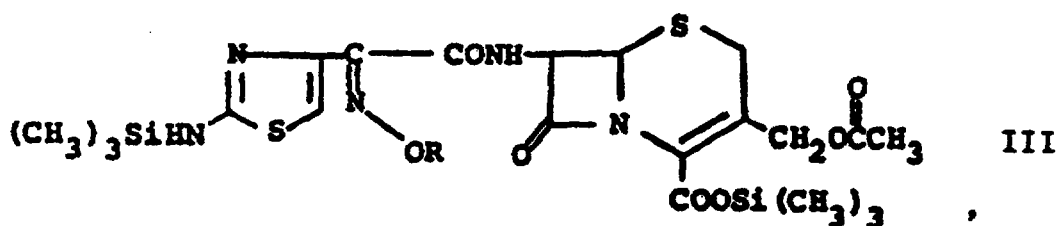
US-A-4,336,253, issued June 22, 1982 to William H.W. Lunn, discloses compounds of the formula



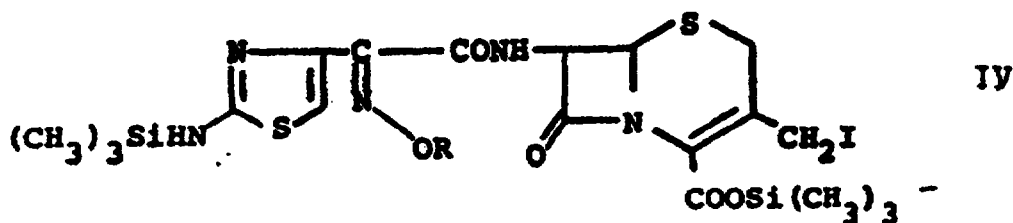
and their preparation by reaction of a compound of the formula



with a silylating agent such as trimethylsilylacetamide, bistrimethylsilylacetamide, or the like, to produce a compound of the formula



reacting the compound of formula III with trimethylsilyl iodide to produce a compound of the formula



and then reacting the compound Formula IV with

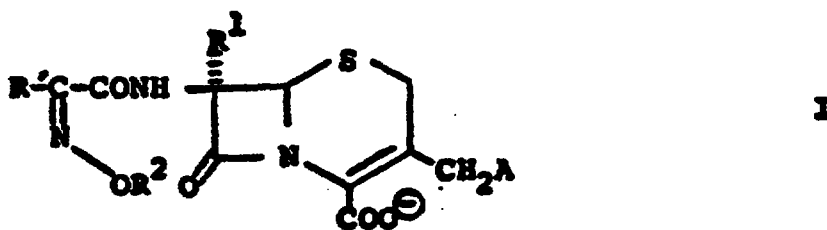


to produce a compound of Formula I.

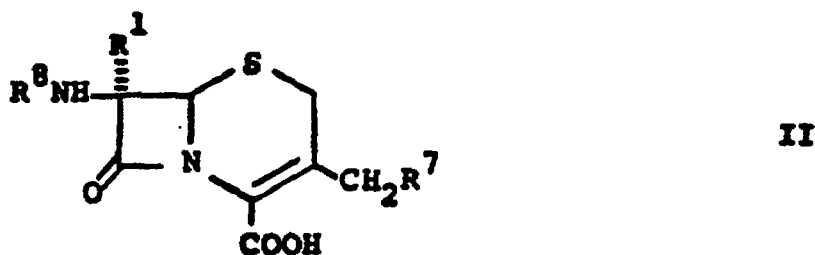
Solvents suitable for the reactions to produce Compounds III and IV are stated to be chlorinated hydrocarbons and lower alkyl nitriles.

25 US-A-4,379,787, US-A-4,382,931 and US-A-4,382,932 to William H. W. Lunn et al. are to analogous compounds and processes; wherein R is, for example, an amino-substituted oxazole, oxadiazole or isoxazole ring, and the 3-substituent is, for example, optionally substituted pyridine, quinoline or isoquinoline.

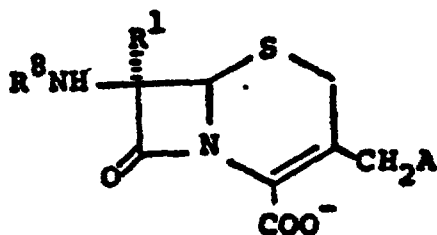
30 South African Patent 84/3343, issued October 25, 1984, corresponding to DE-A-33 16 798 discloses a process for the preparation of cephalosporins of the formula



45 in which R is thiazolyl or 1,2,4-thiadiazolyl radical, R¹ is hydrogen or methoxy, R² is one of numerous substituents known in the cephalosporin art, and A is optionally substituted quinolinium, isoquinolinium or pyridinium. The compounds of Formula I are prepared by reacting a compound of the formula

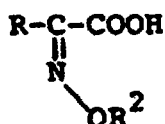


in which R⁸ is hydrogen or an amino-protecting group and R⁷ is "a group which can be replaced by the base corresponding to radical A", with the base corresponding to radical A in the presence of a trialkylsilyl silane, to form a compound of the formula



III

which (after removal of any amino-protective group) is subsequently acylated with an acid of the formula



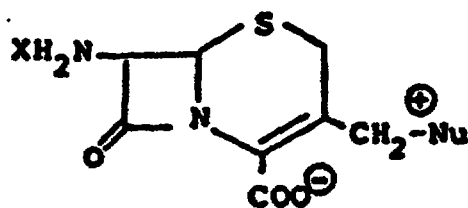
IV

Suitable R^7 groups which are mentioned include acetoxy, propionyloxy, chloroacetoxy, acetylacetoxy and carbamoyloxy. Suitable solvents which are mentioned include methylene chloride, chloroform, dichloroethane, trichloroethane, carbon tetrachloride, acetonitrile, propionitrile and frigands. Methylene chloride is most preferred.

It is acknowledged in the patent that the general process is known, but it is stated that the inventors have found that the yields are surprisingly high if the nucleophilic displacement reaction on Compound II (to form Compound III) is carried out by first adding an excess of the base corresponding to radical A (up to a 20-fold excess) and then adding the trialkylsilyl silane (up to a 10-fold excess).

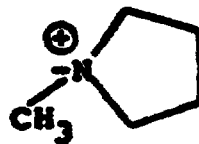
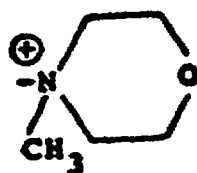
In the W. German priority application (P3316798.2, filed May 7, 1983) for this S. African application, the solvents listed are the same as above, except that "frigand" is called "frigen". The Merck Index, 10th Edition, lists Frigen 11, Frigen 12 and Frigen 114, and shows them as other names for Freon 11, Freon 12, and Freon 114, which are trichlorofluoromethane, dichlorodifluoromethane and 1,2-dichloro-1,1,2,2-tetrafluoroethane, respectively.

The present invention relates to a process for the preparation of a stable, crystalline compound selected from the group having the formula VIb



VIb

wherein X is HCl or HI, and $-Nu^+$ is

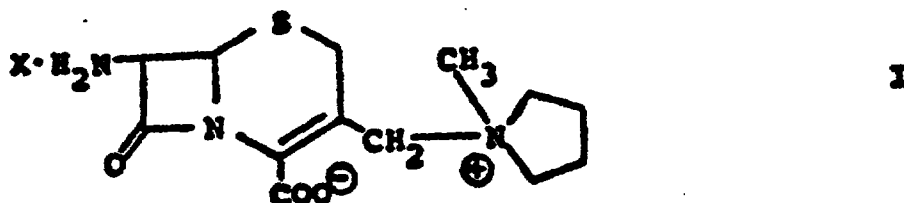


or



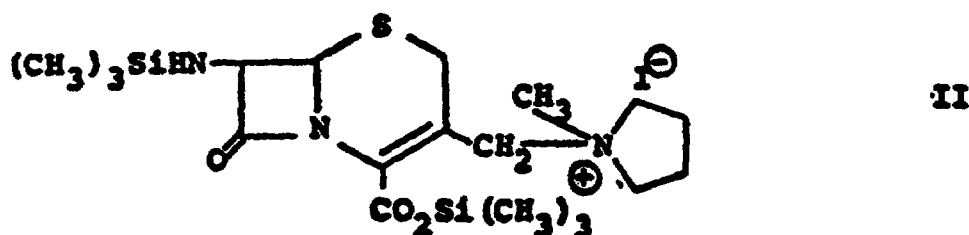
in which $-\text{Nu}$ is as defined above, in 1,1,2-trichlorotrifluoroethane (Freon TF) or 1,1,1-trichlorotrifluoroethane with a lower alkanol at a temperature of from -10 to 25°C to remove the silyl groups, followed by acidification with HCl or HI to form the hydrochloride or hydroiodide salt of the compound of formula VIIb.

According to one embodiment the compounds of the formula I



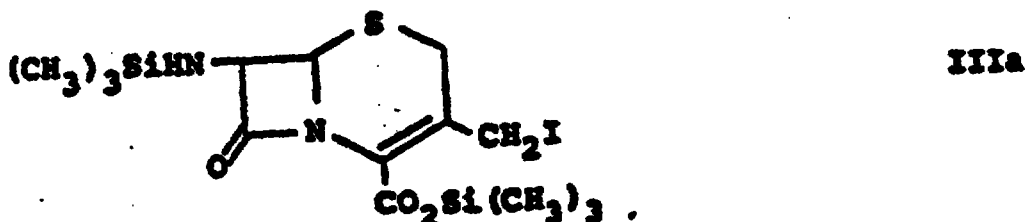
wherein X is HI or HCl are prepared which are crystalline, temperature stable and substantially free of the corresponding Δ^2 isomer. As a result of being substantially free of the Δ^2 isomer, they are convertible (by acylation) to broad spectrum cephalosporins which themselves are substantially free of the Δ^2 isomer, without the need for chromatographic separation of the Δ^2 and Δ^3 isomers. As a result of their temperature stability, they may be isolated and stored, and converted to the end products when desired. An additional advantage of the intermediates of Formula I is that they do not require blocking (protection) of the carboxyl group prior to acylation or deblocking (deprotection) of the carboxyl group after acylation, thus offering process efficiency.

The compounds of Formula I may be prepared by treating a solution of the compound of the formula II



in 1,1,2-trichlorotrifluoroethane (Freon TF) or 1,1,1-trichlorotrifluoroethane with a lower alkanol to remove the trimethylsilyl groups, followed by HCl or HI to form the hydrochloride or hydroiodide salt. It is preferred to use a lower alkanol for removal of the trimethylsilyl groups, and most preferably methanol. The reaction is conducted at a temperature of from -10°C to 25°C , and preferably at a temperature of from 0°C to 10°C . From 2 to 5 equivalents of methanol are used per equivalent of Compound II, and preferably 3 to 4 equivalents of methanol.

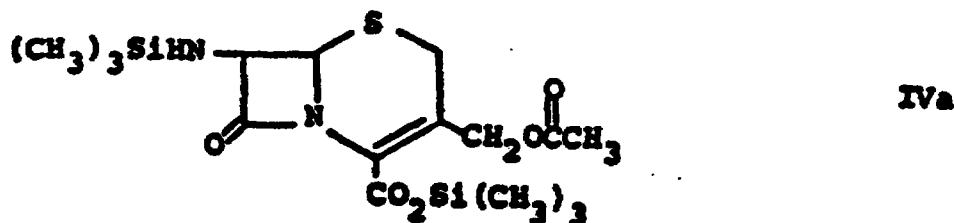
The compound of Formula II may be prepared by reacting a solution of the compound of the formula IIIa



in Freon TF or 1,1,1-trichlorotrifluoroethane with N-methylpyrrolidine (NMP). It has surprisingly been found that the use of Freon TF or 1,1,1-trichlorotrifluoroethane as solvent produces Compound II which is substantially free of the Δ^2 isomer, while commonly used solvents such as methylene chloride, carbon tetrachloride, chloroform or dioxane produce Compound II which contains large amounts of the undesirable Δ^2 isomer (e.g. 50% Δ^2 isomer).

The reaction is conducted at a temperature of from -10°C to 25°C , and preferably from 0°C to 10°C . Although it is possible to use greater or lesser amounts of N-methylpyrrolidine, we obtain highest purity of product when about 1 equivalent of NMP is utilized per equivalent of Compound IIIa.

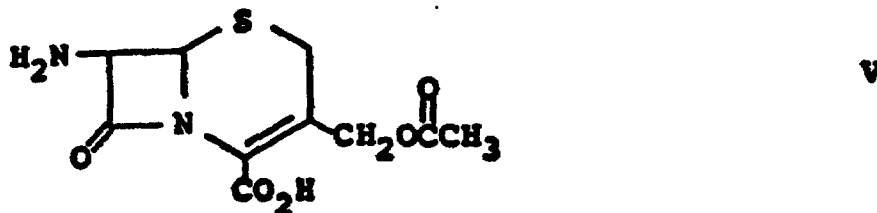
The compound of Formula IIIa may be prepared by reaction of a solution of the compound of the formula IVa



in Freon TF or 1,1,1-trichlorotrifluoroethane with trimethylsilyl iodide (TMSI). Surprisingly, these solvents yield Compound IIIa which is substantially free of the Δ^2 isomer, while similar common solvents (e.g. (1,2-dichloroethane) give Compound IIIa containing significant amounts of the undesirable Δ^2 isomer (e.g. 25%). Other common solvents such as chlorobenzene, dioxane, carbon tetrachloride, and the like, also give significant amounts of the undesirable Δ^2 isomer.

The reaction is conducted at a temperature of from 5°C to 45°C , but preferably is conducted at ambient temperature for convenience. At least one equivalent of TMSI is required to convert all of Compound IVa to Compound IIIa, we prefer to utilize an amount of from 0.9 to 1.5 equivalents per equivalent of Compound IVa. More preferably, we utilize from 1.0 to 1.2 equivalents of TMSI.

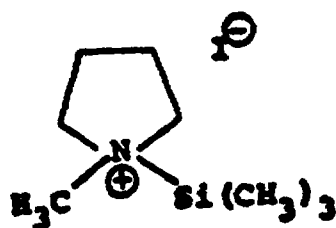
The compound of Formula IVa may be prepared by reacting 7-aminocephalosporanic acid (7-ACA), i.e. the compound of the formula V



with hexamethyldisilazane (HMDS) in the presence of from 0.01 to 0.1 equivalents of TMSI per equivalent of 7-ACA, in Freon TF, at a temperature from room temperature to the boiling point of the solvent. Preferably, the reaction is conducted at reflux. The HMDS may be used in an amount of from 0.95 to 1.4 equivalents per equivalent of 7-ACA, and preferably from 1.0 to about 1.3 equivalents of HMDS per equivalent of 7-ACA. We most prefer to utilize 1.2 equivalents of HMDS.

In an alternate preparation of Compound II, a solution of compound IVa in Freon TF or 1,1,1-trichlorotrifluoroethane is first treated with N-methylpyrrolidine followed by the addition of at least one equivalent of TMSI. The reaction can be conducted at a temperature of from 5°C to 45°C . For convenience, we prefer to conduct the reaction at about 35°C . The N-methylpyrrolidine may be used in an amount of from 1.0 to 2.0 equivalents per equivalent of compound IVa, and preferably from 1.2 to 1.5 equivalent. The TMSI may be used in an amount of from 1.0 to 3.0 equivalents per equivalent of compound IVa, and preferably from 1.5 to 2.0 equivalents.

In another alternate preparation of Compound II, a solution of Compound IVa in Freon TF or 1,1,1-trichlorotrifluoroethane is reacted with N-methyl-N-trimethylsilylpyrrolidinio iodide having the formula VI



VI

at a temperature of from 5°C to 45°C. For convenience, we prefer to conduct the reaction at 35°C. The compound of Formula VI may be used in an amount of from 1.0 to 2.0 equivalents per equivalent of Compound IVa, and preferably from 1.2 to 1.5 equivalents of Compound VI per equivalent of compound IVa. If desired, a small amount of imidazole (e.g. 0.1 equivalents) may be added to the reaction mixture to shorten the reaction time. We have found that, when preparing Compound II by the reaction of Compound IVa with Compound VI, only small amounts of Compound IIIa can be detected in the reaction mixture during the reaction (by NMR analysis).

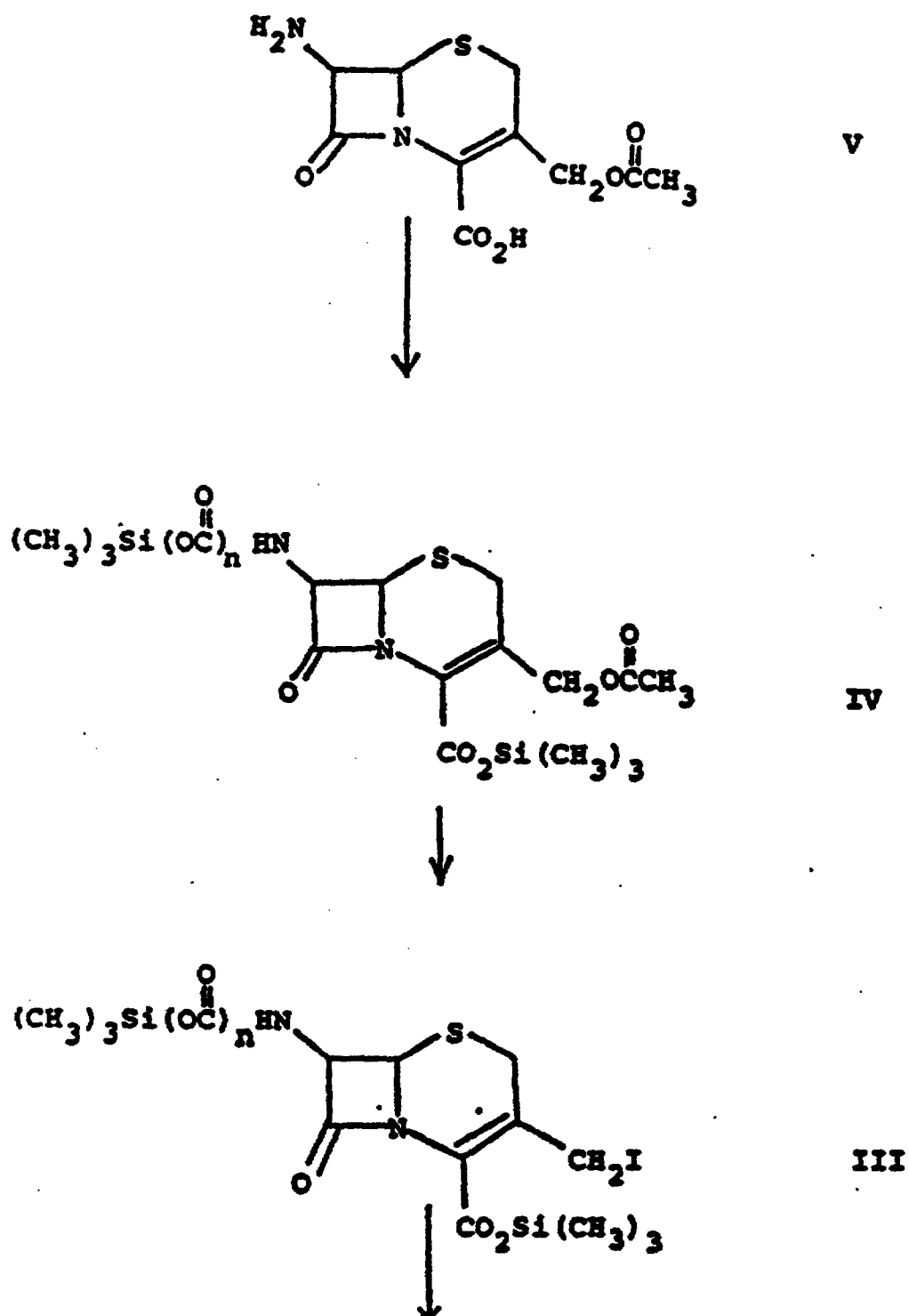
The compound of Formula VI may be prepared by reacting N-methylpyrrolidine with about an equimolar amount of TMSI in Freon TF or 1,1,1-trichlorotrifluoroethane as solvent, at a temperature of from -10°C to 25°C. We prefer to conduct the reaction at a temperature of from 0°C to 5°C. The reaction ratio of NMP and TMSI may be varied, but we obtain excellent results by utilizing equimolar amounts.

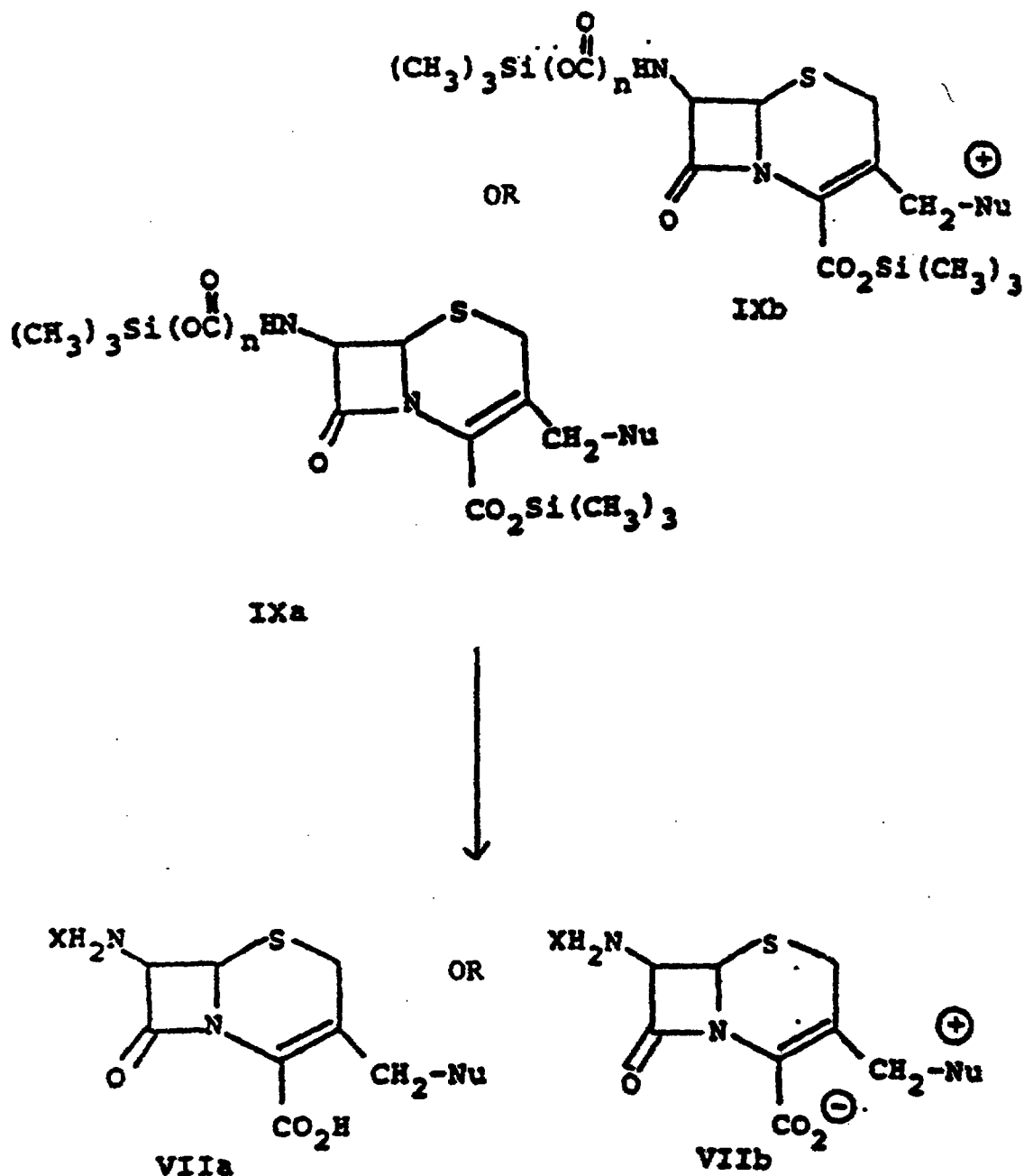
In a preferred reaction scheme, the compounds of Formula I are prepared from 7-ACA in a "one pot" reaction, i.e. without the isolation of any intermediates. Although certain of the individual reaction steps may be conducted in solvents other than Freon TF or 1,1,1-trichlorotrifluoroethane, as indicated above, others require the use of Freon TF or 1,1,1-trichlorotrifluoroethane. Accordingly, when conducting the "one pot" reaction, the reaction is preferably conducted in Freon TF.

In conducting the reaction sequence described herein, we have found it necessary to proceed via the 3-iodomethyl Compound IIIa, rather than the corresponding 3-chloromethyl or 3-bromomethyl analog. We have surprisingly found that the reaction of Compound IVa with trimethylsilyl chloride or trimethylsilyl bromide gives little of the expected 3-chloromethyl or 3-bromomethyl analog of Compound IIIa, e.g. a maximum of about 5-15% even after refluxing for 10 days.

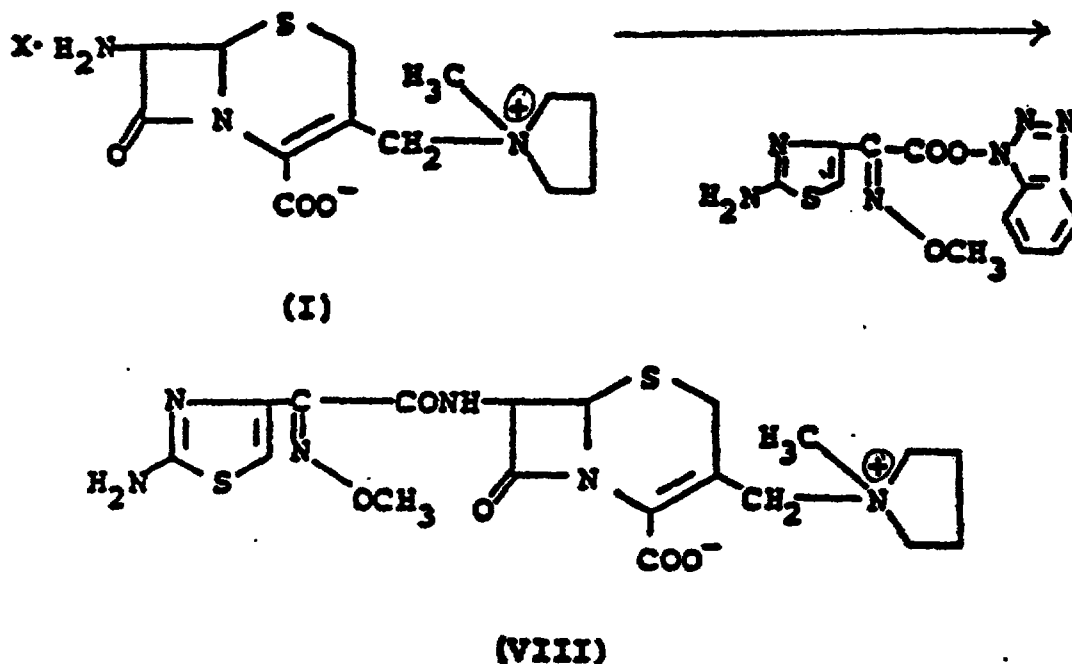
Although the quaternization of Compound IIIa with NMP to produce Compound II (and, subsequently, a compound of Formula I), must be conducted in Freon TF or 1,1,1-trichlorotrifluoroethane in order to obtain Compound II (and I) which are substantially free of the Δ^2 isomer, Compound IIIa may be reacted with other nucleophilic agents in Freon TF or 1,1,1-trichlorotrifluoroethane solution to produce analogs of Compound II. After treatment with a lower alkanol and optional salt formation with, HCl or HI, there are obtained analogs of Compound I which have the formula VIIb.

The preparation of the compounds of Formula VIIb can be described by the general procedure shown below in Reaction Scheme 1. The procedure may be conducted in a stepwise manner as shown in Scheme 1, or when desired, the order of addition of the reactants such as TMSI and the desired nucleophile may be altered as described herein. The preparation of compounds of the Formula VIIb may also, more preferably, be conducted in a "one pot" reaction.

Reaction Scheme 1



The compounds of Formula I and VIIb are readily converted to broad spectrum antibiotics by acylation with the appropriate side-chain acid. For example, a compound of Formula I (X = HCl or HI) is converted to 7-[α -(2-aminothiazol-4-yl)- α -(Z)methoxyiminoacetamido]-3-[(1-methyl-1-pyrrolidinio)methyl]-3-cephem-4-carboxylate (VIII) by N-acylating with 1-benzotriazolyl (Z)-2-(2-aminothiazol-4-yl)-2-methoxyiminoacetate ester. The reaction equation is set forth below.



25 This reaction is readily carried out in the presence of N,N-dimethylaniline in dimethylformamide at room temperature over a period of 10-20 hours; or by dissolving (I) in water and dimethylformamide and adding sodium bicarbonate with ice cooling, and reacting at room temperature for 30 minutes to 5 hours; or by dissolving (I) in water, cooling to 5-15°C, adding NaOH dropwise to pH 5.5-6, adding tetrahydrofuran, adding sodium hydroxide to adjust the pH to 6.7-6.9, adding the active ester reactant and reacting for 1 to 5 hours at room temperature. The active ester is a known compound and is described in Hoechst, Japan Kokai 54-95593 (7/28/79) and German Application 2758000.3 (12/24/77). The utility of the compound (VIII) is shown in Aburaki et al., US-A-4,406,899.

The compounds of Formula VIIb may be acylated in a similar manner to produce broad spectrum cephalosporins.

Example 1

(6R,7R)-Trimethylsilyl 7-(Trimethylsilyl)amino-3-acetoxymethylceph-3-em-4-carboxylate

Method A

45 An oven-dried flask and Friedrich's condenser were cooled to ambient temperature under a stream of dry nitrogen. The flask was then charged with 50.0 g (184 mmoles) of 7-ACA (97.2% purity) and 400 ml of dry 1,1,2-trichlorotrifluoroethane (Freon TF, dried over molecular sieves). To the resulting slurry was added 46.5 ml (222 mmoles, 1.2 equivalents) of 98% 1,1,1,3,3,3-hexamethyldisilazane (HMDS) and 0.80 ml (5.6 mmoles, 0.03 equivalents) of iodotrimethylsilane (TMSI) with good stirring and with protection from moisture. The slurry was vigorously heated under reflux for 16-18 hours, after which time it was cooled to ambient temperature. A ¹H NMR spectrum of an aliquot of the slightly hazy reaction mixture showed >95% conversion to the desired product. NMR (CD₂Cl₂, 360 MHz) δ 0.23 (s, 9H, N-Si(CH₃)₃), 0.38 (s, 9H, -COOSi(CH₃)₃), 1.51 (d, 1H, J=13.6 Hz, NH-), 2.09 (s, 3H, -COCH₃), 3.41 (d, 1H, J=18.3 Hz, -SCH₂-), 3.61 (d, 1H, J=18.3 Hz, -SCH₂), 4.80 (dd, 1H, J=4.5, 13.6 Hz, -COCH(NHSi(CH₃)₃)), 4.83 (d, 1H, J=13.2 Hz, -CH₂OCOCH₃), 4.91 (d, 1H, J=4.5 Hz, -COCH(NHSi(CH₃)₃CH-)), 5.11 (d, 1H, J=13.2 Hz, -CH₂OCOCH₃).

Method B

55 An oven-dried flask and Friedrich's condenser were cooled to ambient temperature under a stream of dry nitrogen. The flask was charged with 10.0 g (36.7 mmoles) of 7-ACA (97.2% purity) and 80 ml of dry Freon TF (dried over molecular sieves). To the resulting slurry was added 9.3 ml (44.1 mmoles, 1.2

equivalents) of 98% HMDS and 44 ml (1.1 mmoles, 0.03 equivalent) of a 0.025M solution of HI in Freon TF (prepared by bubbling HI into dry Freon TF and titrating the resulting saturated solution to a phenolphthalein endpoint). The slurry was vigorously heated under reflux with good stirring and with protection from moisture for 22 hours, after which time it was cooled to ambient temperature. A ¹H NMR spectrum (CD₂Cl₂, 360 MHz showed >95% conversion to the desired product.

Example 2

(6R,7R)-Trimethylsilyl 7-(Trimethylsilyl)amino-3-iodomethylceph-3-em-4-carboxylate

Method A

A total of 30 ml (210 mmoles, 1.15 equivalents) of TMSI was added in a slow stream at ambient temperature to a reaction mixture of (6R,7R)-trimethylsilyl 7-(trimethylsilyl)amino-3-acetoxymethylceph-3-em-4-carboxylate in Freon TF (prepared according to Method A of Example 1; 50.0 g input 7-ACA) under a blanket of dry nitrogen. The progress of the reaction was monitored by ¹H NMR (acetate region). After 1 hour, the slurry was filtered through a Schlenk tube under positive nitrogen pressure. The collected solid was washed with fresh Freon TF (1 x 100 ml). An aliquot of the filtrate gave the following ¹H NMR spectrum in support of the in situ generated (6R,7R)-trimethylsilyl 7-(trimethylsilyl)amino-3-iodomethylceph-3-em-4-carboxylate: NMR(CD₂Cl₂, 360 MHz) δ 0.16 (s, 9H, NHSi(CH₃)₃), 0.40 (s, 9H, -COOSi(CH₃)₃), 1.51 (d, 1H, J=13.4 Hz, NH), 3.54 (d, 1H, J=17.9 Hz, -SCH₂-), 3.80 (d, 1H, J=17.9 Hz, -SCH₂-), 4.37 (d, 1H, J=9.2 Hz, -CH₂I), 4.49 (d, 1H, J=9.2 Hz, -CH₂I), 4.75 (dd, 1H, J=4.6, 13.4 Hz, -COCH(NHSi(CH₃)₃)), 4.89 (d, 1H, J=4.6 Hz, -COCH(NHSi(CH₃)₃)CH-).

Example 3

N-Methyl-N-trimethylsilylpyrrolidino iodide

An oven-dried flask was cooled to ambient temperature under a stream of positive nitrogen. The flask was charged with 25 ml of dry Freon TF (dried over molecular sieves) and 1.42 ml (10.0 mmoles, 1.0 equivalent) of TMSI. The resulting solution was cooled to 0-5°C under a blanket of dry nitrogen. A total of 1.04 ml (10.0 mmoles, 1.0 equivalent) of dry, 97% N-methylpyrrolidine (dried over molecular sieves) was added dropwise and the resulting slurry was stirred at 0-5°C under a nitrogen blanket for 30 minutes. After this time, the slurry was filtered under positive nitrogen pressure through a Schlenk funnel, and the collected solid was washed with fresh, dry Freon TF (2 x 25 ml). The filter cake was partially dried under a positive stream of nitrogen for 15 minutes. Further drying at ambient temperature for 12 hours at 0.05 mm Hg afforded 2.51 g (89%) of N-methyl-N-trimethylsilylpyrrolidino iodide as a colorless, extremely air sensitive solid: C₈H₂₀INSi requires 44.49% iodide; found, 44.40% iodide (iodide ion chromatography analysis).

In a separate experiment, the reaction was conducted generally as above, except that 0.40 ml (10.0 mmoles, 1.0 equivalent) of methanol was added dropwise to the slurry of the salt in Freon TF, at 0-5°C. The resulting heterogeneous mixture was stirred at 0-5°C under a nitrogen blanket for an additional 30 minutes. The slurry was filtered under anhydrous conditions through a Schlenk funnel. The filter cake was washed with fresh Freon TF (2 x 25 ml) and was dried at ambient temperature for 3 hours at 0.05 mm Hg. The isolated solid (1.93 g, 91%) was identified as N-methylpyrrolidino hydriodide, m.p. 80-82°C(uncorrected). An authentic sample of this salt was prepared independently by gassing a solution of N-methylpyrrolidine in Freon TF with hydrogen iodide. The isolated solid had mp 83.5-85.5°C (uncorrected). The 360 MHz ¹NMR spectrum of this material (D₂O) was consistent with the spectrum observed for the salt isolated from the methanol quenched reaction.

The 360 MHz ¹NMR spectrum of the filtrate from the methanol quenched reaction showed, as the major components, methoxytrimethylsilane and methyl iodide (integration ratio 17/1), as well as a small amount of hexamethyldisiloxane.

Example 4

(6R,7R)-7-Amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylate Monohydrochloride

5 Method A

To a solution of (6R,7R)-trimethylsilyl 7-(trimethylsilyl)amino-3-iodomethylceph-3-em-4-carboxylate in Freon TF (prepared according to Method A of Example 2; 50.0 g input 7-ACA) at 0-5°C under a blanket of dry nitrogen was added 19.0 ml (183 mmoles, 1.0 equivalent) of dry 97% N-methylpyrrolidine (NMP, dried
 10 over molecular sieves) dropwise at a rate sufficient to maintain a reaction temperature of <10°C. The resulting slurry was stirred vigorously at 0-5°C for 15 minutes following complete addition. After this time, an additional 100 ml of dry Freon TF was added to facilitate agitation. Next, 25 ml (615 mmoles, 3.35 equivalents) of methanol was added dropwise at a rate sufficient to maintain a reaction temperature of <10°C. The slurry was vigorously stirred at 0-5°C for an additional 15 minutes. The reaction mixture was
 15 filtered and the collected solid was washed with fresh Freon TF (1 x 100 ml) followed by partial drying under suction for 15 minutes. Further drying in vacuo at ambient temperature for 16 hours gave 71.3 g (>100%) of crude product.

To this material was added 200 ml of water. The pH of the slurry (2.40) was lowered to 0.50 by dropwise addition of concentrated HCl. A total of 10.0 g of decolorizing carbon was added, and the slurry
 20 was stirred at ambient temperature for 15 minutes. The decolorizing carbon was removed by filtration through a celite pad (10.0 g), and the pad was washed with fresh deionized water (1 x 25 ml). The product was precipitated from the aqueous solution by dropwise addition of 5 volumes of acetone. The slurry was cooled to 0-5°C and was maintained at this temperature for 30 minutes. The slurry was vacuum filtered, sequentially washed with 50 ml of cold (0-5°C) 5/1 acetone/water and acetone (2 x 50 ml), and was partially
 25 dried under suction for 15 minutes. Further drying in vacuo afforded 23.3 g (39%) of snow white, crystalline (6R,7R)-7-amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylate monohydrochloride: HPLC as-sayed as 96.3% versus the standard lot; NMR (D₂O, 360 MHz, D₂O/H₂O suppressed) δ2.14-2.32 (envelope, 4H, -N(CH₃)CH₂CH₂-), 3.00 (s, 3H, NCH₃), 3.46-3.67 (m, 5H, -N(CH₃)CH₂CH₂-SCH₂), 3.96 (d, 1H, J=16.9 Hz, -SCH₂), 4.09 (d, 1H, J=13.9 Hz, =CCH₂N-), 4.73 (d, 1H, J=13.9 Hz, =CCH₂N-), 5.21 (d, 1H, J=5.1 Hz, -COCHCHS-), 5.41 (d, 1H, J=5.1 Hz, -COCHCHS-).
 30

Method B

An oven-dried flask was cooled to ambient temperature under a stream of dry nitrogen. The flask was
 35 charged with 15.7 ml (110mmoles, 1.5 equivalents) of TMSI and 140 ml of dry Freon TF (dried over molecular sieves). The resulting solution was cooled to 0-5°C, and 10.7 ml (103 mmoles, 1.4 equivalents) of 97% NMP was added dropwise so as to maintain a reaction temperature of <10°C. Following the addition, the slurry was stirred at 0-5°C for 30 minutes under a nitrogen blanket. After this time, the slurry was allowed to warm to room temperature.

A reaction mixture containing (6R,7R)-trimethylsilyl 7-(trimethylsilyl)amino-3-acetoxymethylceph-3-em-4-carboxylate (prepared according to Method A of Example 1; 20.0 g input 7-ACA) was added by cannula as quickly as possible to the NMP/TMSI slurry at ambient temperature. The resulting heterogeneous mixture was stirred at room temperature for 14 days with periodic monitoring of the reaction progress by HPLC. At
 40 a point where the 7-ACA area % was <2% of the total peak area, methanol (6.0 ml, 147 mmoles, 2.0 equivalents) was added dropwise and stirring was continued for an additional 30 minutes. The crude product was filtered, washed with fresh Freon TF (1 x 300 ml), and was partially dried under suction for 15 minutes. The product was further dried in vacuo at ambient temperature for 16 hours.

The crude product was slurried in 80 ml of deionized water. The pH was lowered to 0.50 by dropwise addition of concentrated HCl with good stirring. Decolorizing carbon (20% by weight of the crude product)
 50 was added, and the mixture was stirred for an additional 45 minutes. The slurry was vacuum filtered through diatomaceous earth, and the pad was washed with deionized water. Isopropyl alcohol (IPA, 900 ml) was added dropwise over a period of 1 hour to the aqueous phase with moderate agitation. The resulting slurry was stirred at 25°C for 1 hour, cooled to 0-5°C and was stirred for an additional 1 hour. Filtration, washing with cold (0-5°C) 9/1 IPA/water (1 x 200 ml) and acetone (1 x 200 ml) and drying at 25°C in vacuo for 16
 55 hours gave 15.5 g (63%) of crystalline (6R,7R)-7-amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylate monohydrochloride.

Method C

As an adjunct to the procedure of Method B, imidazole (0.50 g, 7.3 mmoles, 0.10 equivalent) was added to the N-methylpyrrolidine/iodotrimethylsilane/Freon TF slurry at 0-5°C. This modification resulted in a modest increase in reaction rate as the 7-ACA level reached <2% of the total HPLC peak area after 10 days. Processing of the reaction mixture as described in Method B above afforded 14.7 g (60%) of crystalline (6R,7R)-7-amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylate monohydrochloride.

Method D

To a slurry of (6R,7R)-trimethylsilyl 7-(trimethylsilyl)amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylate iodide in Freon TF (prepared according to Method A of Example 4; 20.0 g input 7-ACA) at 0-5°C under a blanket of dry nitrogen was added 10 ml (246 mmoles, 3.35 equivalents) of methanol dropwise over 5 minutes (Temperature <9°C). The resulting slurry was stirred at 0-5°C for 15 minutes following the addition. Next, 50 ml of 3N HCl (prepared by adding 250 ml of concentrated HCl to 756 ml of deionized water) was added dropwise over a period of 10 minutes. Following the addition, the cooling bath was removed and the mixture was stirred for 15 minutes. The phases were separated and the aqueous phase (brought to a volume of 100 ml) was stirred for 30 minutes at ambient temperature with 4.0 g (20% of weight of input 7-ACA) of decolorizing carbon. The slurry was filtered through 4.0 g of diatomaceous earth, and the pad was washed with deionized water (1 x 10 ml). The aqueous volume was brought to 100 ml.

Crystallization Method 1: To 50 ml of the rich aqueous was added 250 ml (5 volumes) of acetone dropwise to precipitate the product. The resulting slurry was stirred with ice water cooling for 1 hour, after which time it was filtered under suction, washed with cold (0-5°C) 5/1 acetone/water (2 x 40 ml) and acetone (1 x 40 ml) and was partially dried under suction for 15 minutes. The product was further dried in vacuo at ambient temperature for 15 hours to give 4.48 g (37%, based on 50% of the theoretical yield for 20.0 g input 7-ACA) of colorless, crystalline (6R,7R)-7-amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylate monohydrochloride.

Crystallization Method 2: To 50 ml of the rich aqueous was added 150 ml (3 volumes) of isopropyl alcohol (IPA) dropwise to precipitate the product. The resulting slurry was stirred with ice water cooling for 1 hour, after which time it was filtered under suction, washed with cold (0-5°C) 9/1 IPA/water (2 x 40 ml) and acetone (1 x 40 ml), and was partially dried under suction for 15 minutes. The product was further dried in vacuo at ambient temperature for 15 hours to give 5.46 g (45%, based on 50% of the theoretical yield for 20.0 g input 7-ACA) of slightly off-white, crystalline (6R,7R)-7-amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylate monohydrochloride.

Example 5

(6R,7R)-7-Amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylate Hydrochloride or Hydroiodide
Via a "One Pot" Reaction Scheme

The following is a compendium of experiences and observations derived from several repeated reactions using the same amounts of reactants, temperatures, etc.

PROCEDURE:

1. 7-ACA (50.0 g, 0.184 mole) was added into CCl₂FCClF₂ (Freon TF) (350 ml) under a nitrogen atmosphere (Note 1).
2. HMDS (46.5 ml, 0.22 mole, 1.2 equivalents) was added to this stirred suspension in one portion.
3. TMSI (0.78 ml, 6.0 mmoles, 0.03 equivalent) was added in one portion by means of a syringe (Note 2).
4. The resulting mixture was heated and refluxed for 7-10 hours (Notes 3, 4). The reaction was monitored by NMR (Note 5).
5. The silylation mixture was cooled to ambient temperature, further diluted with Freon TF (150 ml) and was cooled to 5°C under a nitrogen blanket.
6. With good stirring, N-methylpyrrolidine (26.83 ml, 0.25 mole, 1.4 equivalents) was added over 10 minutes while maintaining a reaction temperature below 10°C (Note 6).

7. TMSI (47.1 ml, 0.33 mole, 1.8 equivalents) was added at a slow rate over a period of 10-15 minutes by syringe into the well-stirred mixture from Step 6 at 5°C. A small exothermic temperature rise was observed during the TMSI addition. The reaction slurry was stirred at 5°C for 30 minutes under a blanket of dry nitrogen.

8. The resulting slurry was carefully heated and stirred at 35°C to 36°C for a 45-55 hour period. The course of the displacement reaction was monitored by HPLC (Note 7).

9. Both the volume of the reaction mixture and the rate of agitation were checked regularly. An additional 100 ml of Freon TF was added as necessary (Note 8).

10. Upon completion of the reaction (7-ACA) less than 2 area % by HPLC), the slurry was cooled to 5°C under nitrogen protection and methanol (25 ml, 0.615 mole) was added dropwise over an 8 minute period at 5°C (Note 9).

11. The slurry thinned considerably and was stirred for an additional 15 minutes at 5-10°C after methanol was added.

12. The cooling bath was then removed, and 125 ml of 3N HCl (prepared by adding 250 ml of concentrated HCl solution to 756 ml of water) was added over 2 minutes with good stirring. The reaction temperature rose to 12-15°C.

13. The hydrolysis mix was warmed to 20-25°C as fast as possible (without exceeding 25°C) and was further stirred at 20-25°C for 15 minutes.

14. The phases were separated, and the organic phase (bottom phase) was back extracted with water (1 x 50 ml). This aqueous phase was used as the wash during the polish filtration of the rich aqueous solution.

15. Diatomaceous earth (2.5 g) was added to the rich aqueous phase from Step 14 and it was filtered through a precoated diatomaceous earth filter (7.5 g). The diatomaceous earth cake was washed with the aqueous solution from Step 14 (back extraction) and then with 25 ml of deionized H₂O.

16. The combined rich aqueous solution and wash (volume ca. 270 ml, reddish brown in color) was stirred for 30 minutes at 21-23°C with decolorizing carbon (10 g). Diatomaceous earth (2.5 g) was added to the mixture and stirring was continued for 5 minutes.

17. The carbon was removed by filtration through a precoated diatomaceous earth filter (7.5 g). The carbon cake was washed with water (1 x 75 ml), and was further dried under suction for 5 minutes.

18. If necessary, additional amounts of Freon TF found at the bottom of the rich aqueous solution were separated from the aqueous phase.

19. To the clear, orange colored aqueous solution (volume ca. 350 ml; pH 0.9-1.15) was added isopropyl alcohol (Note 10) dropwise to the cloud point.

20. The isopropyl alcohol addition was stopped, and the crystallization was allowed to proceed at 21-23°C for 15 minutes.

21. After this time, additional isopropyl alcohol was added into the slurry over a period of 45 to 60 minutes (a total of 1.2 liters of isopropyl alcohol was added), and then the slurry was stirred and cooled at 0-5°C for 60 minutes.

22. The product was collected by filtration, and the cake was washed with cold (0-5°C) 9/1 isopropyl alcohol/water (2 x 100 ml, Note 11) and acetone (1 x 100 ml). The product (Note 12) was dried under suction for an additional 15 minutes. The product was then dried *in vacuo* to constant weight to give 46-51 g (75-83%) of a crude HCl/HI salt mixture of the title product (Compound I), as a slightly off-white to white crystalline solid. The activity yield was 60-63.4%.

23. The product purities were >95% based on NMR analysis. The HPLC potencies were 750-800 mcg/mg versus an analytical sample of the Δ^3 form of Compound I HCl. The area % purities were >95%.

Note 1. 7-ACA is an extremely dusty solid. It should be weighed out in a hood or other area with adequate ventilation. A dust respirator and disposable outer garment should be worn as protection from the 7-ACA dust.

Note 2. All operations with TMSI should be performed under anhydrous conditions as much as possible.

Note 3. The reaction time has proven to be variable and is probably dependent on the presence of TMSI (or its reactive equivalent) catalyst. If the reaction appears to be sluggish, additional TMSI hastens the completion of the reaction.

Note 4. It is important that vigorous reflux is maintained, as a major driving force for reaction is the removal of the ammonia that is produced.

Note 5. It is important that anhydrous conditions are maintained when samples are removed from the reaction mixture. This holds true for all reactions in this sequence.

- Note 6. A rise in temperature to over 10°C during the addition of NMP resulted in an enhanced amount of the undesired Δ^2 isomer of 1/26.
- Note 7. Samples should be analyzed every 4-6 hours by HPLC. The reaction temperature, time and concentration and the number of equivalents of TMSI and the medium basicity are critical in this reaction.
- Note 8. The slurry will become very thick and some dilution of $\text{CCl}_2\text{FCClF}_2$ will be required to facilitate agitation.
- Note 9. Prior to the addition of CH_3OH , the slurry is so thick that chunks of material cling to the side of the reaction vessel, making complete mixing with CH_3OH difficult. It should be determined by visual observation that good and completed mixing is occurring throughout the reaction vessel.
- Note 10. This usually required 0.5-1.0 volume of isopropanol.
- Note 11. This wash was prepared by mixing 90 ml of isopropanol and 10 ml of water and cooling to 0-5°C in an ice water bath.
- Note 12. In a separate experiment, the product was isolated as the pure HI salt by treating the slurry from Step 11 with 125 ml of 3N HI instead of 125 ml of 3N HCl (as in Step 12). Processing of the aqueous phase as described (cf. Steps 13-24) yielded 44.3 g of white, crystalline HI salt. The corrected HPLC potency was 105% versus an analytical standard HCl salt. The activity yield from 7-ACA was 56.7%.

Example 6

Recrystallization of (6R,7R)-7-Amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylate Monohydrochloride (Compound I HCl)

PROCEDURE:

1. Crude I HCl (15.0 g, .045 moles) was added in one portion to 1N HCl (125 ml, 3.5 moles, 3.50 equivalents) with good stirring.
 2. The resulting mixture was stirred at ambient temperature for 5 minutes.
 3. A total of 8.0 g of decolorizing carbon was added in one portion with continued good stirring. The slurry was agitated for an additional 45 minutes.
 4. The carbon slurry was filtered under suction through a pad of diatomaceous earth (8.0 g). The pad was washed with water (1 x 35 ml), and was dried under suction for 5 minutes.
 5. The slightly cloudy filtrate was polish filtered through a 5 μm Millipore filter to give a sparkling clear, water-white aqueous filtrate (volume = 170 ml).
 6. Isopropyl alcohol (125 ml) was added dropwise to the cloud point over a period of 25 minutes with good stirring. At this point, the addition of isopropanol was stopped. The slurry was stirred at ambient temperature for 15 minutes, during which time a good seed bed was established.
 7. Additional isopropyl alcohol (475 ml, Note 3) was added dropwise over a period of 25 minutes with good stirring.
 8. The resulting slurry was stirred with ice water bath cooling for 1 hour.
 9. The slurry was filtered, and was sequentially washed with cold (0-5°C) 9/1 isopropanol/water (2 x 120 ml, Note 4) and acetone (1 x 120 ml).
 10. The filter cake was partially dried under suction for 15 minutes. Further drying in *vacuo* (steam injector suction) for 15 hours at 40°C gave 7.87 g (52%) of snow-white, electrostatic, crystalline I HCl (Note 5).
- Note 1. The number of moles of input I HCl is based on 100% purity.
- Note 2. The 1N hydrochloric acid solution was prepared by adding 83 ml of concentrated hydrochloric acid to 920 ml of distilled water.
- Note 3. The total volume of isopropyl alcohol used for the crystallization was 600 ml, which was 3.5 times the volume of the polished aqueous filtrate from Step 5.
- Note 4. The isopropanol/water washes consisting of 108 ml isopropanol and 12 ml distilled water were cooled to 0-5°C in an ice bath.
- Note 5. The analytical data compiled on the recrystallized I HCl are shown below:

ANALYSIS	THEORY	FOUND	CORRECTED FOR KF
%C	46.77	46.02	46.71
%H	6.04	6.17	6.10
%N	12.59	12.31	12.49
%S	9.61	9.50	9.64
KF(H ₂ O)	-	1.47	
Residue (sulfated ash)	-	<0.1	

This material was assayed at 99.5% potency versus an HPLC standard lot of I HCl. The Klett number was 3 (100.0 mg sample diluted to 10 ml in a volumetric flask with Milli-Q water; filtered through & Millex HPLC sample preparation filter; blue light; path length ca. 1.2 cm).

In general, it has been found that the "crude" compound I HCl formed by the addition of HCl to the final mixture, contains some I HI (formed from the iodide present in the precursor intermediate II). Thus, although it is of high antibacterial purity, it normally must be recrystallized as above to remove Compound I HI.

On the other hand, the initially crystallized Compound I HI formed by the addition of HI to the final reaction mixture, is free of Compound I HCl. Accordingly, the I HI is normally of high purity and need not be recrystallized.

Example 7

Conversion of I, (X = HCl) to (VIII)

A sample of Compound I (X = HCl), (21.72 g, 0.0612 mole), was dissolved in water (190 ml) at 25°C with stirring. The mixture was then cooled to 8-10°C, and the pH adjusted from 2.5 to 5.8 (range 5.7-5.9) by the dropwise addition of sodium hydroxide solution (2N, 30.5 ml, 0.061 mole, 1.0 equivalent). Total volume was 214 ml.

Tetrahydrofuran (THF, 555 ml) was then added in three portions. The temperature of the mixture after each addition rose to 12-13°C and was allowed to return to 8-10°C before the next portion was added. The total addition time was 10 minutes. The pH of the mixture was 5.8-6.1.

The pH of the mixture was then adjusted to 6.8 (range 6.7-6.9) by the dropwise addition of sodium hydroxide solution (2N, 2.0 ml, 0.004 mole).

A sample of the 1-benzotriazol (Z)-2-(2-aminothiazol-4-yl)-2-methoxyiminoacetate active ester (29.5 g, 0.0927 mole) was added to the reaction mixture in five equal portions over 45 minutes. The cooling bath was removed after the first portion of the active ester had been added. The pH of the reaction mixture was readjusted to 6.5 (range 6.5-6.7) 5-10 minutes after each addition of the active ester by the dropwise addition of sodium hydroxide solution (2N).

The clear, pale orange reaction mixture was stirred for 2-3 hours at 25°C. In the initial 30 minutes, the pH was readjusted to 6.5 (range 6.5-6.7) every 5-10 minutes by the dropwise addition of 2N sodium hydroxide solution. In the remaining reaction time, the pH was readjusted to 6.5 every 15 minutes (total 2N NaOH: 29.5 ml, 0.059 mole, 0.97 equivalent). The completion of the reaction was judged by HPLC analysis.

Solids present in the reaction mixture were then removed by filtration and washed with water (2 x 5 ml). The filtrate was extracted with methyl isobutyl ketone (MIBK, 790 ml) and the aqueous layer separated. The organic phase was washed with water (64 ml), and the aqueous phases combined and stirred with Dicalite (5.1 g) for 10 minutes. The solids were removed by suction filtration and washed with water (2 x 5 ml).

The resulting clear orange solution (volume 314 ml) was acidified with good stirring to pH 3.7 (range 3.5-4.0) by the dropwise addition of sulfuric acid (4N, 14.5 ml). At this point the mixture became cloudy and crystallization of the sulfuric acid addition salt of (VIII) began.

The crystallization was allowed to proceed for 10-15 minutes, and then the pH was adjusted to 3.0 (range 2.9-3.1) by the dropwise addition of sulfuric acid (4N, 7.5 ml). The mixture was cooled to 0 to 5°C and the remaining sulfuric acid (4N, 63.5 ml) was added over 20-30 minutes (resulting pH: 1.3-1.5). After the addition of sulfuric acid was complete, the slurry was stirred for 1 hour at 0 to 5°C.

The white crystalline product was removed by suction filtration and washed with sulfuric acid (0.5N, 63.5 ml). The solids were partially dried under suction for 15 minutes and then washed with acetone (2 x 100 ml). The solids were again partially dried under suction for 10 minutes and then slurried for 1 hour in acetone (400 ml) with good agitation. The solids were removed by suction filtration, washed with acetone (2 x 100 ml) and dried in vacuo (10-15 mm Hg) at 35-40°C to constant weight (3-6 hours).

The product, the sulfuric acid addition salt of (VIII), was recovered as a slightly electrostatic, white crystalline solid (28.79 g, 81.4%).

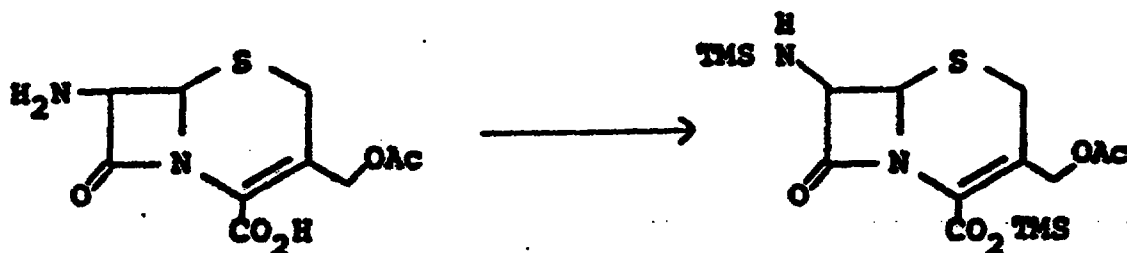
Example 8

Conversion of I (X = HI) to (VIII)

The general procedure of Example 7 is repeated except that the I (X = HCl) starting material is replaced by an equimolar amount of I (X = HI), and the title compound is thereby produced.

Example 9

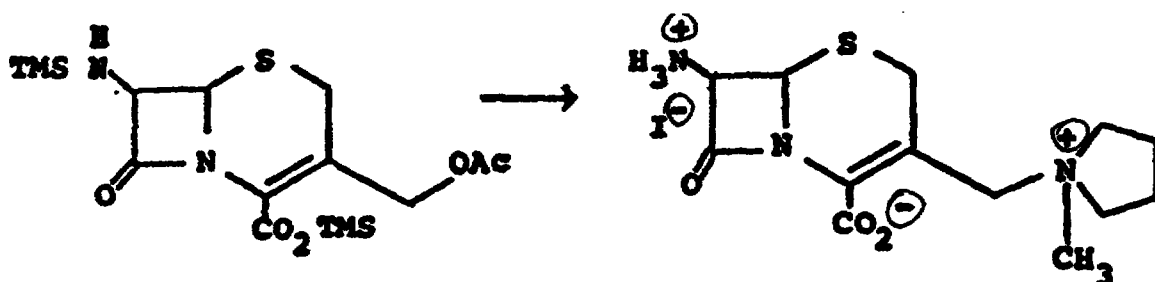
(6R,7R)-Trimethylsilyl 7-(Trimethylsilyl)amino-3-acetoxy methylceph-3-em-4-carboxylate



An oven-dried flask and Friedrich's condenser were cooled to ambient temperature under a stream of dry nitrogen. The flask was then charged with 10.0 g (36.7 mmoles) of 7-ACA (97.2% purity) and 70 ml of dry 1,1,2-trichloro-trifluoroethane (Freon TF, dried over molecular sieves). To the resulting slurry was added 9.3 ml (44.1 mmoles, 1.2 equivalents) of 98% 1,1,1,3,3,3-hexamethyldisilazane (HMDS) and 0.16 ml (1.1 mmoles, 0.03 equivalents) of iodotrimethylsilane (TMSI) by syringe with good stirring and with protection from moisture. The slurry was vigorously heated under reflux for 7-10 hours with a slight nitrogen sweep through the system. The reaction was then cooled to ambient temperature under a blanket of dry nitrogen, and was diluted with 30 ml of fresh Freon TF. A ^1H NMR spectrum of an aliquot of the slightly hazy reaction mixture showed >95% conversion to the desired product which is identical to the product of Example 1, and the ^1H NMR data is the same as reported in Example 1, Method A.

Example 10

(6R,7R)-7-Amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylate monohydroiodide



To a slightly hazy solution of (6R,7R)-trimethylsilyl 7-(trimethylsilyl)-amino-3-acetoxymethylceph-3-em-4-carboxylate in Freon TF (prepared according to Example 18) at 0-5°C under a blanket of dry nitrogen was added 5.35 ml (51.4 mmoles, 1.4 equivalents) of dry 97% N-methylpyrrolidine (dried over molecular sieves) dropwise over 1-2 minutes with good stirring. Next, 9.40 ml (66.1 mmoles, 1.8 equivalents) of TMSI was added by syringe over about 5 minutes with continued good stirring. The reaction temperature was kept below 10°C during the addition. The resulting slurry was stirred at 0-5°C for an additional 30 minutes. After this time, the slurry was placed into an oil bath carefully maintained at 35-36°C. The progress of the reaction was monitored by HPLC. After 45-48 hours, the reaction was complete (<2 area % 7-ACA) and it

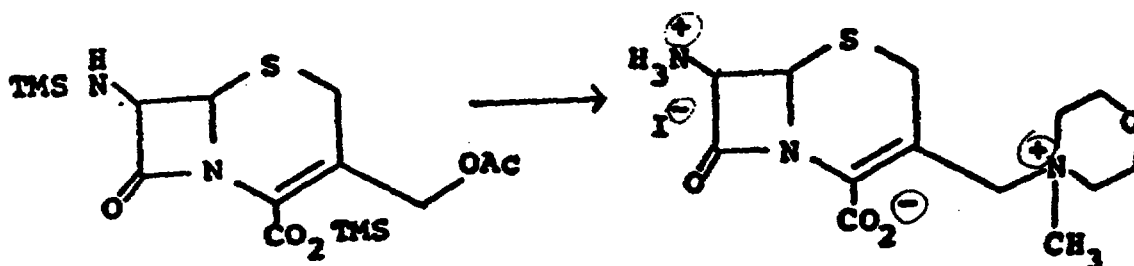
was cooled to 0-5°C under a blanket of dry nitrogen. A total of 5.0 ml (123 mmoles, 3.35 equivalents) of methanol was added dropwise with good stirring. The reaction temperature was maintained at <10°C during the addition. The resulting slurry was stirred at 0-5°C for an additional 15 minutes. Next, 25 ml (75 mmoles, 2.0 equivalents) of 3N aqueous HI solution was added in one portion. Following the addition, the cooling bath was removed, and the 2 phase mixture was rapidly warmed to 20-25°C. Vigorous stirring was continued for 15 minutes. The phases were separated, and the organic phase was back extracted with water (1 x 10 ml). This back wash was saved for later use.

The main aqueous phase was stirred at 20-25°C for 10 minutes with 0.5 g of diatomaceous earth. The slurry was filtered through 1.5 g of diatomaceous earth (pre-washed with 50 ml water). The pad was washed with the aqueous back wash from above, then with water (1 x 5 ml). The cake was partially dried under suction for 5 minutes. A total of 2.0 g of decolorizing carbon was added and the slurry was stirred at 20-25°C for 30 minutes. After this time, 0.5 g of diatomaceous earth was added and stirring was continued for an additional 5 minutes. The slurry was filtered through 1.5 g of diatomaceous earth (pre-washed with 50 ml water), and the pad was washed with water (1 x 5 ml). The diatomaceous earth pad was partially dried under suction for 5 minutes. The filtrate was polish filtered through a 5 µm Millipore filter.

Precipitation of the product was achieved by dropwise addition of 3.5 volumes of isopropanol to the clear, amber-colored aqueous phase at 20-25°C. The resulting slurry was cooled to 0-5°C and was allowed to stand for 1 hour. The slurry was filtered and washed with cold (0-5°C) isopropanol/water (4/1;v/v) (2 x 20 ml) and acetone (2 x 20 ml). The cake was partially dried under suction for 5 minutes. Further drying in vacuo at 20-25°C to constant weight afforded 8.94 g (57%) of white, crystalline (6R,7R)-7-amino-3-(1-methyl-1-pyrrolidinio) methylceph-3-em-4-carboxylate monohydroiodide. HPLC showed the salt to be 97 area % pure; and the 360 MHz ¹H-NMR spectrum observed for the title compound as a monohydroiodide is the same as the ¹H NMR data reported for the HCl salt of the identical product in Example 6, Method A.

Example 11

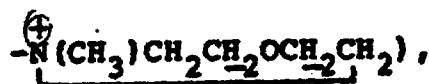
(6R,7R)-7-Amino-3-(4-methyl-4-morpholinio)methylceph-3-em-4-carboxylate monohydroiodide



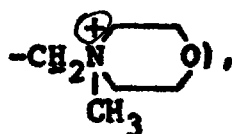
The title compound was prepared from a Freon TF solution of (6R,7R)-trimethylsilyl 7-(trimethylsilyl)-amino-3-acetoxymethylceph-3-em-4-carboxylate (from 50.0 g input 7-ACA) as described in Example 10, except that 28.3 ml (257 mmoles, 1.4 equivalents) of dry N-methylmorpholine (dried over molecular sieves) was substituted for the N-methylpyrrolidine. The progress of the reaction was monitored by HPLC and was found to be complete after 7-8 hours at 35-36°C. The reaction was worked up as described in Example 19 (amounts of materials are multiplied by 5X due to increased batch size) to afford 36.0 g (41%) of slightly off-white crystalline (6R,7R)-7-amino-3-(4-methyl morpholinio)methylceph-3-em-4-carboxylate monohydroiodide. HPLC area % purity was >95%; ¹H NMR (360 MHz, D₂O)

δ 3.30 (s, 3H, N-CH₃),

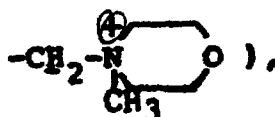
3.60(m, 4H

3.68(d, 1H, J=10 Hz, -SCH₂-), 4.04(d, 1H, J=10 Hz, -SCH₂-), 4.2 (m, 4H,

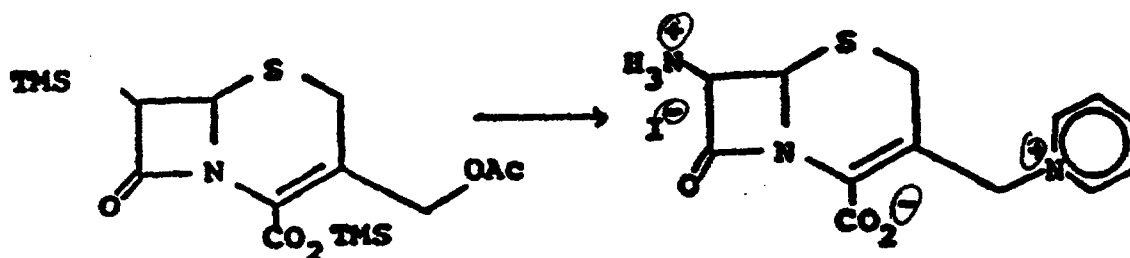
4.25 (d, 1H, J=14 Hz,



4.93 (d, 1H, J=14 Hz,

5.30 (d, 1H, J=5Hz, C-6 β-lactam), 5.53 (d, 1H, J=5Hz, C-7 β-lactam);
IR (KBr) 3480, 1795 and 1600 cm⁻¹.

Anal. Calc'd for C ₁₉ H ₁₉ N ₃ O ₄ S·HI:	C, 35.40;	H, 4.34;	N, 9.53
Found:	C, 34.99;	H, 4.38;	N, 9.35

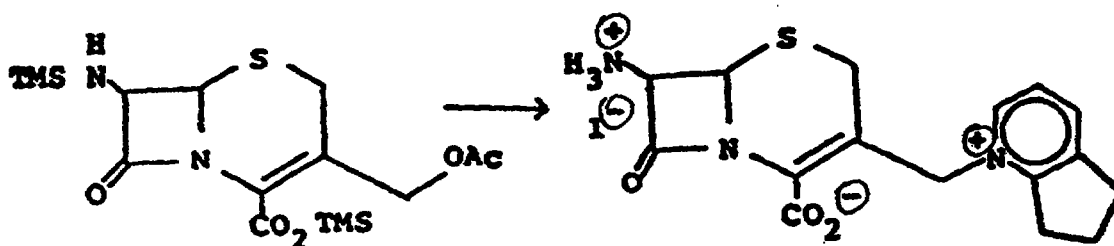
Example 12**(6R,7R)-7-Amino-3-(1-pyridinio)methylceph-3-em-4-carboxylate monohydroiodide**

The title compound was prepared from a solution of (6R,7R)-trimethylsilyl 7-(trimethylsilyl)amino-3-acetoxymethylceph-3-em-4-carboxylate in Freon TF (10.0 g input 7-ACA) as described in Example 10, except that 4.2 ml (51.4 mmol, 1.4 equivalents) of dry pyridine (dried over KOH) was substituted for the N-methyl-pyrrolidine. The progress of the reaction was monitored by HPLC and was found to be complete after 51 hours at 35-36°C. The slurry was cooled to 20-25°C under a blanket of dry nitrogen. The solid was

removed by filtration through a funnel under positive nitrogen pressure. The collected solid was washed with fresh Freon TF (2 X 100 ml). The filtercake was quickly added to 50 ml of dry dichloromethane which was pre-cooled to 0-5°C in an ice-water bath. To the resulting dark solution was added 5.0 ml (123 mmol, 3.35 equivalents) of methanol dropwise with good stirring and with a reaction temperature of <10°C. The resulting slurry was stirred at 0-5°C for an additional 15 minutes. The solid was collected by filtration and was washed with fresh CH₂Cl₂ (2 X 50 ml). The filtercake was reslurried in 150 ml CH₂Cl₂ for 1 hour. The solid was filtered, washed with CH₂Cl₂ (2 X 50 ml) and dried in vacuo at 20-25°C to constant weight to afford 14.7 g (95%) of crude (6R,7R)-7-amino-3-(1-pyridinio)methylceph-3-em-4-carboxylate monohydroiodide. The 360 MHz ¹H NMR spectrum and HPLC chromatogram (85 area % purity; major contaminant was pyridine hydroiodide, 13 area %) of this material were consistent with an authentic sample prepared independently; see also Smith, G.C.D., EP-70706, January 26, 1983; pg. 21).

Example 13

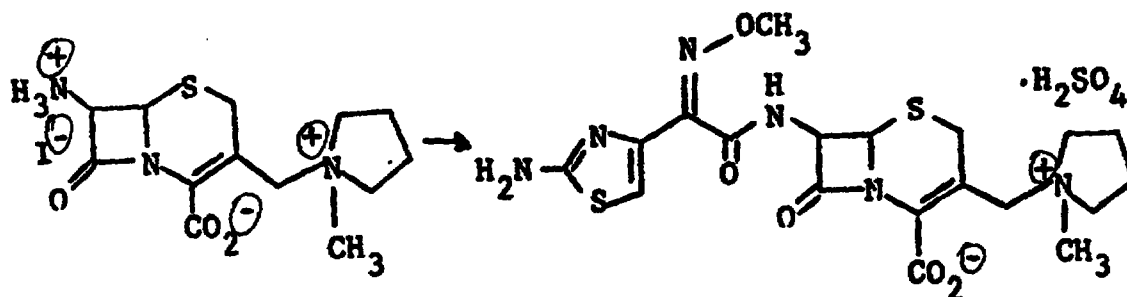
(6R,7R)-7-Amino-3-[1-(2,3-cyclopenteno)pyridinio]methylceph-3-em-4-carboxylate monohydroiodide



The title compound was prepared from a solution of (6R,7R)-trimethylsilyl 7-(trimethylsilylamino)-3-acetoxymethylceph-3-em-4-carboxylate in Freon TF (10.0 g input 7-ACA) as described in Example 10, except that 6.02 ml (51.4 mmol, 1.4 equivalents) of 2,3-cyclopentenopyridine (dried over molecular sieves) was substituted for the N-methylpyrrolidine. The progress of the reaction was monitored by HPLC and was found to be complete after 52 hours at 35-36°C. Work-up of the reaction mixture as described previously in Example 12 afforded 13.2 g (78%) of crude (6R,7R)-7-amino-3-[1-(2,3-cyclopenteno)pyridinio]methylceph-3-em-4-carboxylate monohydroiodide. The 360 MHz ¹H NMR spectrum and HPLC chromatogram (92 area % purity) of this material were consistent with an authentic sample of this material prepared independently.

Example 14

7-[α-(2-Aminothiazol-4-yl)-α-(Z)-methoxyiminoacetamido]-3-[(1-methyl-1-pyrrolidinio)methyl]-3-cephem-4-carboxylate

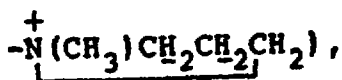


A total of 12.76 g (30 mmol) of (6R,7R)-7-amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylate monohydroiodide (prepared in Example 10) was suspended in 87 ml of water with good stirring at 20-25°C. The slurry was cooled to 8-10°C, and the pH was raised to 5.80 at 7°C by dropwise addition of 13.0 ml (26 mmol, 0.87 equivalents) of 2N sodium hydroxide solution over a period of 35 minutes. Next, 555 ml of tetrahydrofuran was added, and the pH of the resulting solution was increased to 6.8 at 10°C by

dropwise addition of 1.9 ml (3.8 mmoles, 0.13 equivalents) of 2N sodium hydroxide solution. The cooling bath was removed, and 14.75 g (92.7 mmoles, 1.5 equivalents) of the HOBT active ester of syn-2-(2-aminothiazol-4-yl)-2-methoxyiminoacetic acid was added in two equal portions of 7.38 g each over 30 minutes. Following each addition, the pH was readjusted to 6.5 every 5-10 minutes, by dropwise addition of 2N sodium hydroxide solution. In the remaining reaction time, the pH was readjusted to 6.5 by dropwise addition of 2N sodium hydroxide solution every 15 minutes (total 2N sodium hydroxide used was 28.6 ml (57.2 mmoles, 1.91 equivalents). The completion of the reaction was determined by HPLC analysis.

After 1.75 hours the dark solution was poured into 365 ml of methyl isobutyl ketone, and the lower aqueous phase was separated. The organic phase was back extracted with water (1 X 30 ml). The combined aqueous phases were stirred 10 minutes at 20-25°C with 2.35 g of diatomaceous earth. The insoluble material was removed by suction filtration, and the filtercake was washed with water (1 X 5 ml). The amber colored aqueous phase was polish filtered through a 5 m Millipore filter. The pH of the solution was 6.40 at 17°C.

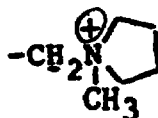
A total of 6.7 ml of 4N sulfuric acid was added dropwise with good stirring to give a turbid solution of pH 3.82 at 18°C. Crystallization of the product was allowed to proceed for 5 minutes with good stirring. An additional 3.5 ml of 4N sulfuric acid was added to give a slurry of pH 3.09 at 20°C. The slurry was cooled to 0-5°C and a total of 30 ml of 4N sulfuric acid was added dropwise over 20 minutes. The resulting slurry was stirred for 1 hour at 0-5°C. The precipitate was filtered and washed with 0.5N sulfuric acid (1 X 30 ml). The filtercake was partially dried under suction for 15 minutes. After this time, the cake was washed with acetone (2 X 50 ml), and was again partially dried under suction for 15 minutes. The cake was reslurried in 200 ml of acetone at 20-25°C for 1 hour. The salt was filtered, washed with acetone (2 X 50 ml) and was partially dried under suction for 15 minutes. Further drying in vacuo at 40°C to constant weight gave 12.41 g (72%) of crystalline, slightly off-white title compound as its sulfate salt: ¹H NMR (360 MHz, D₂O with solvent suppression) δ2.16-2.33 (envelope, 4H,



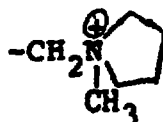
3.01 (s, 3H, $\hat{\text{N}}\text{-CH}_3$), 3.45-3.64 (m, 5H,



-SCH₂-), 3.95 (d, 1H, J = 17 Hz, -SCH₂-), 4.04 (d, 1H, J = 14 Hz,



), 4.08 (s, 3H, -OCH₃), 4.75 (d, 1H, J = 14 Hz,

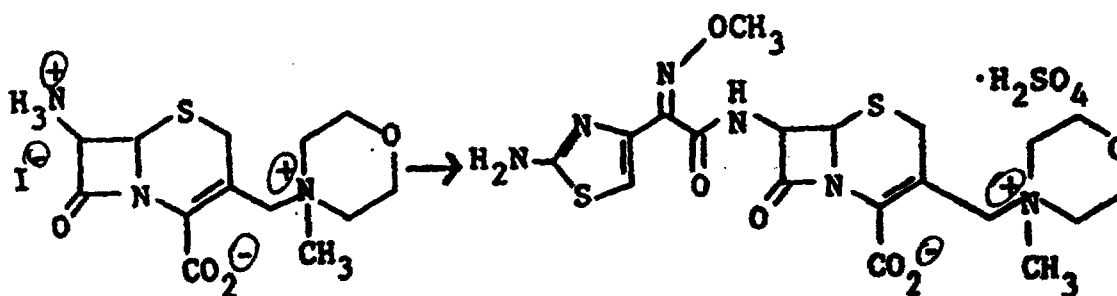


), 5.37 (d, 1H, J = 5 Hz, C-6 β-lactam), 5.86 (d, 1H, J = 5 Hz, C-7 β-lactam), 7.16 (s, 1H, C-5 thiazole).

Anal. Calc'd for $C_{19}H_{24}N_6O_5S_2 \cdot H_2SO_4$:	C, 39.43; N, 14.53;	H, 4.53; S, 16.63
Found:	C, 39.40; N, 14.39;	H, 4.47; S, 16.60

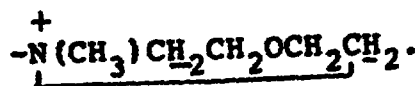
Example 15

7-[α -(2-Aminothiazol-4-yl)- α -(Z)-methoxyiminoacetamido]-3-[(4-methyl-4-morpholinio)methyl]-3-cephem-4-carboxylate

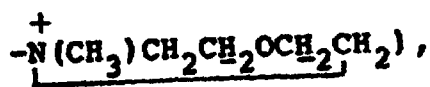


The title compound was prepared from 54.0 g (122 mmoles) of (6R,7R)-7-amino-3-(4-methyl-4-morpholinio)methylceph-3-em-4-carboxylate monohydroiodide (prepared in Example 11) and 59.0 g (184 mmoles, 1.5 equivalents) of syn-2-(2-aminothiazol-4-yl)-2-methoxyiminoacetic acid HOBT active ester by the procedure described in Example 14. A total of 60.1 g (84%) of crystalline, white title compound as its sulfate salt was prepared by this procedure:

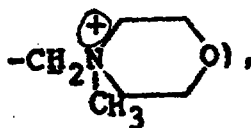
^1H NMR (360 MHz, $D_2O/NaDCO_3$) δ 3.30 (s, 3H, $\dot{N}CH_3$), 3.55 (m, 5H,



-SCH₂-), 4.05 (d, 1H, J = 16 Hz, -SCH₂-), 4.10 (s, 3H, -OCH₃), 4.1 (m, 4H,



4.21 (d, 1H, J = 14 Hz,



4.95 (d, 1H, J = 14 Hz,



5.47 (d, 1H, J=5 Hz, C-6 β -lactam), 5.96 (d, 1H, J=5 HZ, C-7 -lactam), 7.10 (s, 1H, C-5 thiazole).

Anal. Calc'd for $C_{15}H_{24}N_6O_6S_2 \cdot H_2SO_4$:	C, 38.37; N, 14.13;	H, 4.41; S, 16.18
Found:	C, 38.16; N, 14.08;	H, 4.32; S, 16.14

Example 16

The following Table provides a summary of results for some of the Examples in the present invention. With the exception of compounds of Examples 10 and 11, the reported data are for crude HI salts isolated by filtration of the slurry obtained from the methanol quench. Compounds of Examples 10 and 11 were isolated after quenching the methanol slurry with aqueous 3N HI.

TABLE

Compound of Example No.	% Yield	Δ^3 -HPLC Area % Purity	Δ^3/Δ^2 Ratio	
			HPLC ^a	NMR ^b
10	57	97	97/0	-
11	41	>95	>95/0	-
12	95	65 ^c	65/6	-
13	78	92	92/0	-

a. Reported data is area % for each isomer in the HPLC chromatograms of the isolated HI salt.

b. Reported data is the isomer ratio obtained by comparative integration of a baseline resolved peak(s) for each isomer.

c. The major contaminant was the HI salt of pyridine (24 area %).

Example 17

(6R,7R)-7-amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylate monohydroiodide

An oven-dried flask and Friedrich's condenser were cooled to ambient temperature under a stream of positive nitrogen. The flask was charged with 10.0g (36.7 mmoles) of 7-ACA and 80 ml of 1,1,1-trichlorotrifluoroethane (dried over molecular sieves). With good stirring, a total of 9.3 ml (44.1 mmoles, 1.2 equivalents) of HMDS was added by syringe in one portion. Immediately thereafter, 0.16 ml (1.1 mmoles, 0.03 equivalents) of TMSI was added in one portion by syringe. The resulting slurry was vigorously heated under reflux with good agitation and with protection from moisture for 6.0 hours. Examination of an aliquot by NMR (CD_2Cl_2) showed >95% conversion to the desired trimethylsilyl 7-(trimethylsilyl)amino-3-acetoxymethylceph-3-em-4-carboxylate.

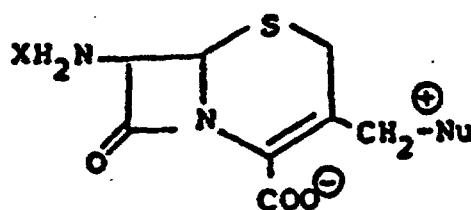
The silylation mixture from above was cooled to ambient temperature. Next, 6.0 ml (42.2 mmoles, 1.15 equivalents) of TMSI was added by syringe in a slow stream over 2-3 minutes. The resulting slurry was stirred at ambient temperature for 1.0 hour. After this time examination of an aliquot by NMR showed >95% conversion to the desired 3-iodomethylcephem. After 1.25 hour the slurry was cooled to 0-5°C and was held at this temperature for 15 minutes. The slurry was filtered under positive nitrogen pressure through a Schlenk funnel into an ice-water cooled receiver flask. The collected solid was washed with fresh 1,1,1-trichlorotrifluoroethane (1 x 17 ml).

To the filtrate containing the trimethylsilyl 7-(trimethylsilyl)amino-3-iodomethylceph-3-em-4-carboxylate at 0-5°C under an inert atmosphere of dry nitrogen was added 3.82 ml (36.7 mmoles, 1.0 equivalent) of 97% N-methylpyrrolidine (dried over molecular sieves) dropwise by syringe at such a rate so as to maintain a reaction temperature of less than 10°C. The resulting slurry was stirred an additional 15 minutes at 0-5°C.

After this time, 5.0 ml (123 mmoles, 3.35 equivalent) of methanol was added dropwise while maintaining a reaction temperature of less than 10°C. The resulting thin slurry was stirred an additional 15 minutes at 0-5°C. The solid was suction filtered, washed with Freon TF (1 x 100 ml) and partially dried under suction for 15 minutes. Further drying in vacuo at ambient temperature to constant weight afforded 10.27g (66%) of (6R,7R)-7-amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylate monohydroiodide. The 360 MHz ¹H NMR spectrum of this crude material (D₂O) showed a Δ³/Δ² isomer ratio of 5.4/1; the HPLC chromatogram of the salt showed a Δ³/Δ² area % ratio of 8.1/1. The activity yield for the conversion of 7-ACA to 7-amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylate was 43.5% (determined by HPLC quantitation of the salt versus a reference standard of the desired Δ³ isomer).

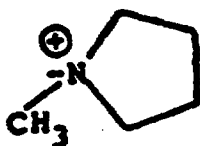
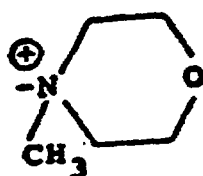
Claims

1. A process for the preparation of a stable, crystalline compound selected from the group having the formula VIIb



VIIb

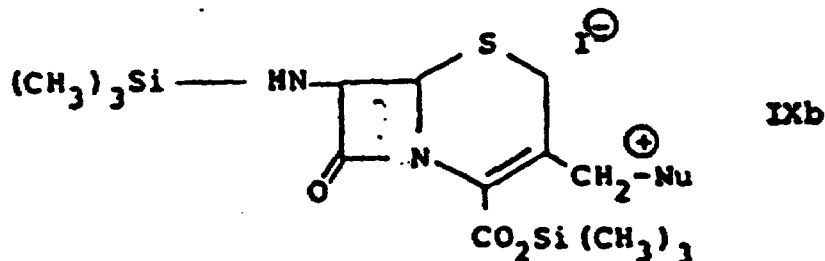
wherein X is HCl or HI, and -Nu⁺ is



or



which is substantially free of the Δ² isomer, which comprises treating a compound of the formula IXb



IXb

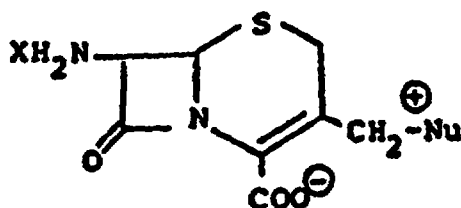
in which -Nu⁺ is as defined above, in 1,1,2-trichlorotrifluoroethane (Freon TF) or 1,1,1-trichlorotrifluoroethane with a lower alkanol at a temperature of from -10 to 25 °C to remove the silyl groups, followed by acidification with HCl or HI to form the hydrochloride or hydroiodide salt of the compound of formula VIIb.

2. The process of Claim 1 wherein the lower alkanol is methanol.

After this time, 5.0 ml (123 mmoles, 3.35 equivalent) of methanol was added dropwise while maintaining a reaction temperature of less than 10°C. The resulting thin slurry was stirred an additional 15 minutes at 0-5°C. The solid was suction filtered, washed with Freon TF (1 x 100 ml) and partially dried under suction for 15 minutes. Further drying in vacuo at ambient temperature to constant weight afforded 10.27g (66%) of (6R,7R)-7-amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylate monohydroiodide. The 360 MHz ¹H NMR spectrum of this crude material (D₂O) showed a Δ³/Δ² isomer ratio of 5.4/1; the HPLC chromatogram of the salt showed a Δ³/Δ² area % ratio of 8.1/1. The activity yield for the conversion of 7-ACA to 7-amino-3-(1-methyl-1-pyrrolidinio)methylceph-3-em-4-carboxylate was 43.5% (determined by HPLC quantitation of the salt versus a reference standard of the desired Δ³ isomer).

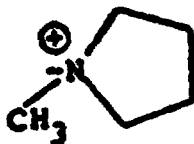
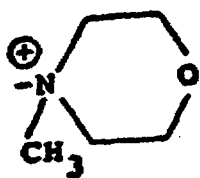
Claims

1. A process for the preparation of a stable, crystalline compound selected from the group having the formula VIIb



VIIb

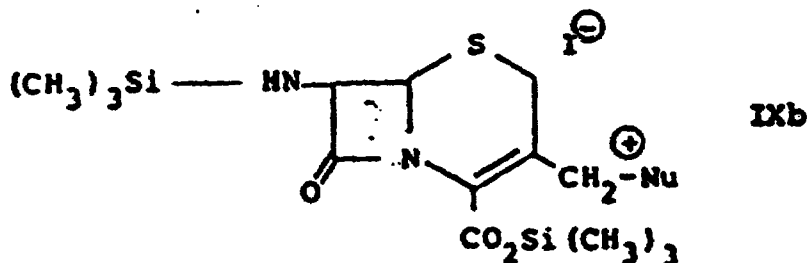
wherein X is HCl or HI, and -Nu⁺ is



or



which is substantially free of the Δ² isomer, which comprises treating a compound of the formula IXb



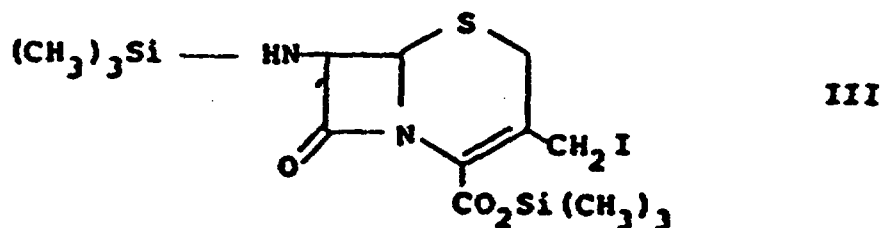
IXb

in which -Nu⁺ is as defined above, in 1,1,2-trichlorotrifluoroethane (Freon TF) or 1,1,1-trichlorotrifluoroethane with a lower alkanol at a temperature of from -10 to 25 °C to remove the silyl groups, followed by acidification with HCl or HI to form the hydrochloride or hydroiodide salt of the compound of formula VIIb.

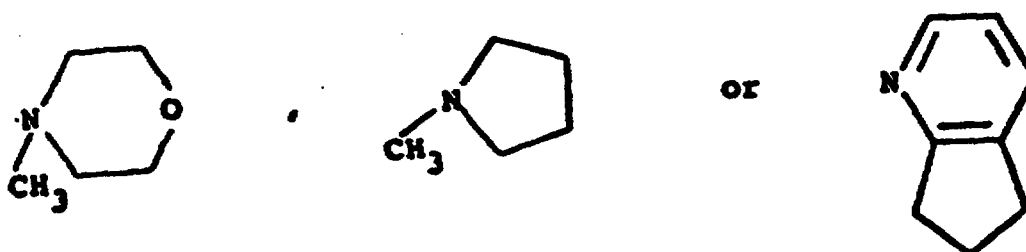
2. The process of Claim 1 wherein the lower alkanol is methanol.



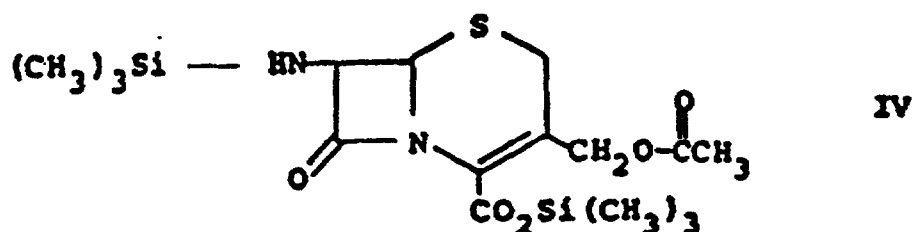
3. The process of claim 1 wherein the compound of formula IXb is prepared by reacting the compound of the formula III



15 in 1,1,2-trichlorotrifluoroethane (Freon TF) or 1,1,1-trichlorotrifluoroethane at a temperature of from -10 to 25 °C with a compound of the formula

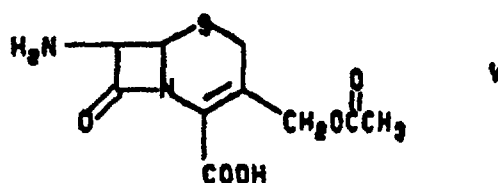


4. The process of Claim 3 wherein the compound of formula III is prepared by treating the compound of the formula IV



45 in 1,1,2-trichlorotrifluoroethane (Freon TF) or 1,1,1-trichlorotrifluoroethane at a temperature of from to 45° C with at least one equivalent of iodotrimethylsilane per equivalent of compound IV.

5. The process of Claim 4 wherein the compound of formula IV is prepared by treating the compound of formula V



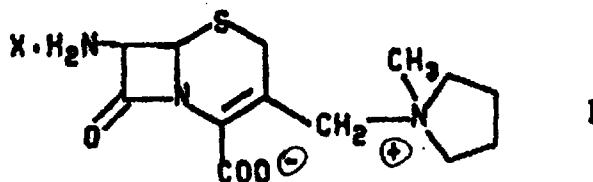
in 1,1,2-trichlorotrifluoroethane (Freon TF) with



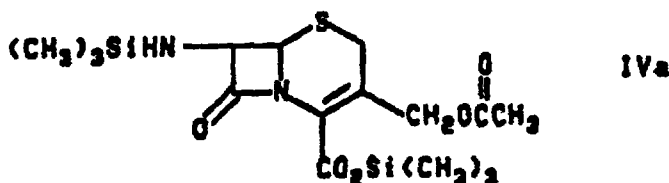
hexamethyldisilazane and 0.01 to 0.1 equivalents of iodotrimethylsilane at a temperature of from 5 to 45° C.

6. The process of anyone of the preceeding claims wherein -Nu is 1-methyl-1-pyrrolidinio.

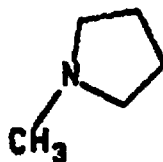
7. A process for the preparation of a stable, crystalline compound of the formula I



wherein X is HCl or HI which is substantially free of the Δ^2 isomer, which comprises treating a solution of the compound of formula IVa

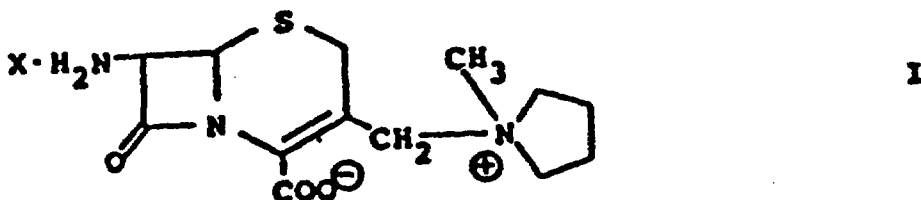


in 1,1,2-trichlorotrifluoroethane (Freon TF) or 1,1,1-trichlorotrifluoroethane with at least one equivalent of a compound of the formula



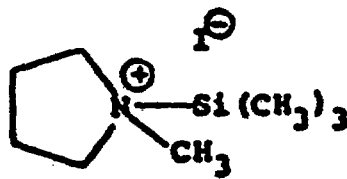
then with at least one equivalent of iodotrimethylsilane per equivalent of compound IVa at a temperature of from 5 to 45° C, followed by the treatment with a lower alkanol at a temperature of from -10 to 25° C to remove the silyl groups, and acidification with HCl or HI

8. A process for the preparation of a stable, crystalline compound of the formula I

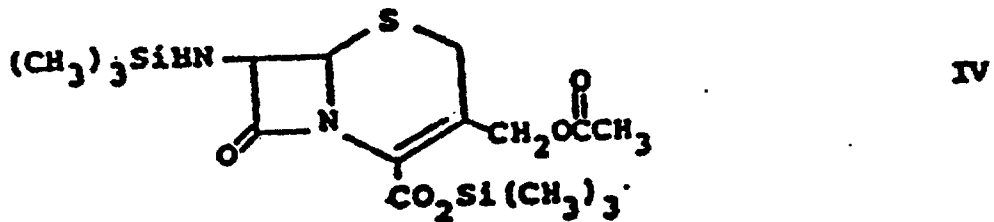


wherein X is HCl or HI, which is substantially free of the Δ^2 isomer, which comprises treating a compound of the formula





with a solution of the compound of formula IV

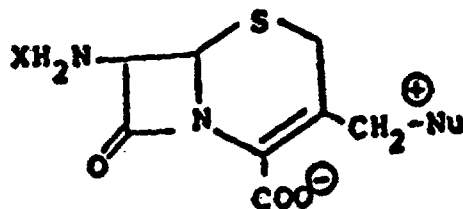


in 1,1,2-trichlorotrifluoroethane (Freon TF) or 1,1,1-trichlorotrifluoroethane at a temperature of from 5 to 45 °C, followed by treatment with a lower alkanol at a temperature of from - 10 to 25 °C to remove the silyl groups, and acidification with HCl or HI.

9. The process of Claim 8 wherein the compound of formula IV is prepared according to Claim 5.

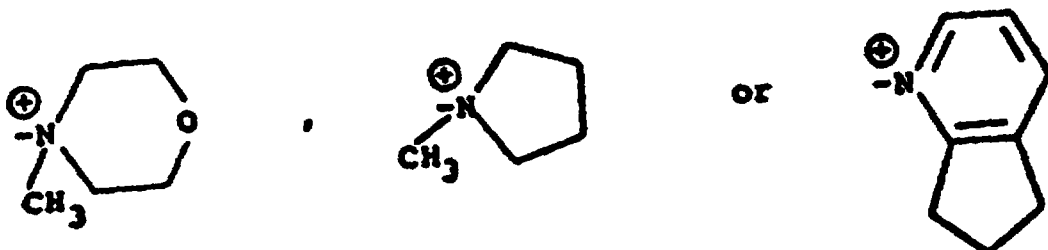
Patentansprüche

1. Verfahren zur Herstellung einer stabilen, kristallinen Verbindung der Formel



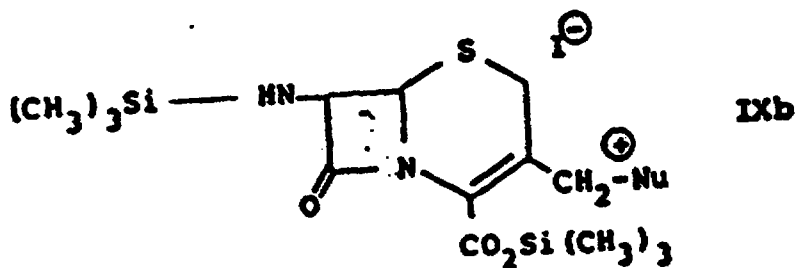
VIIB

,worin für HCl oder HI steht, und -Nu⁺ für



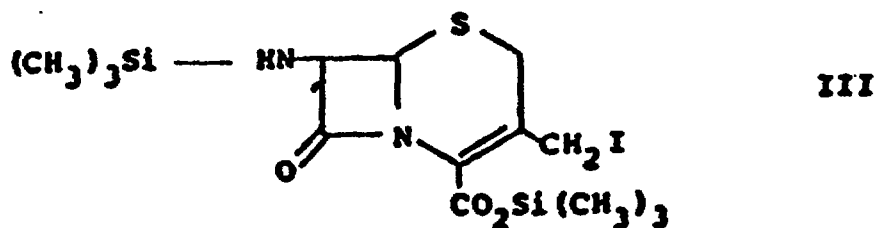
steht, die im wesentlichen frei an Δ^2 Isomer ist, wobei man eine Verbindung der Formel



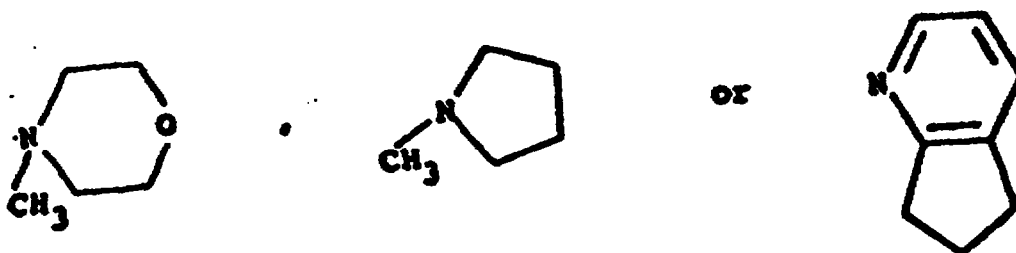


in der -Nu wie oben definiert ist, in 1,1,2-Trichlortrifluorethan (Freon TF) oder 1,1,1-Trichlortrifluorethan mit einem Niedrigalkanol bei einer Temperatur von -10 bis 25°C zur Entfernung der Silylgruppen behandelt und anschließend mit HCl oder HI ansäuert, um das Hydrochlorid oder Hydrojodid der Verbindung der Formel VIIb zu bilden.

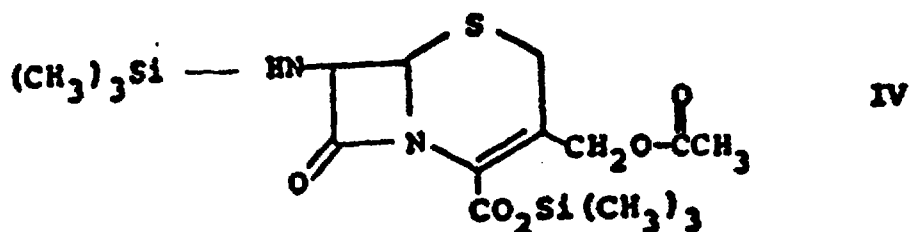
2. Verfahren nach Anspruch 1, worin das Niedrigalkanol Methanol ist.
3. Verfahren nach Anspruch 1, wobei die Verbindung der Formel IXb hergestellt wird durch Umsetzung der Verbindung der Formel III



in 1,1,2-Trichlortrifluorethan (Freon TF) oder in 1,1,1-Trichlortrifluorethan bei einer Temperatur von -10 bis 25°C mit einer Verbindung der Formel

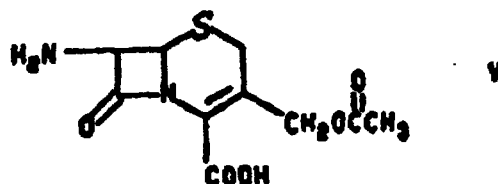


4. Verfahren nach Anspruch 3, wobei die Verbindung der Formel III hergestellt wird durch Behandlung der Verbindung der Formel



in 1,1,2-Trichlortrifluorethan (Freon TF) oder 1,1,1-Trichlortrifluorethan bei einer Temperatur von 5 bis 45°C mit mindestens einem Äquivalent Jodtrimethylsilan pro Äquivalent der Verbindung IV.

5. Verfahren nach Anspruch 4, wobei die Verbindung der Formel IV hergestellt wird durch Behandlung der Verbindung der Formel V

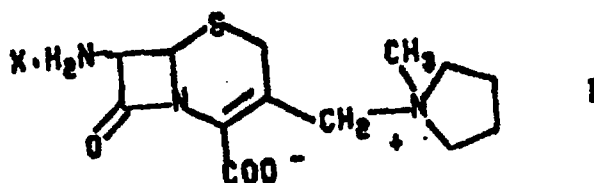


in 1,1,2-Trichlortrifluorethan (Freon FT)

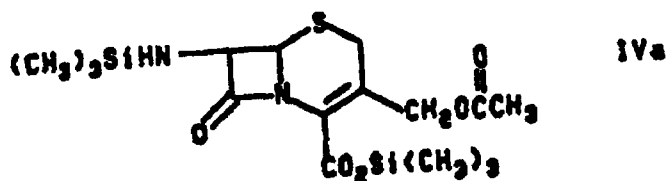
mit Hexamethyldisilazan und etwa 0,01 bis etwa 0,1 Äquivalenten Jodtrimethylsilan bei einer Temperatur von 5 bis 45 °C.

6. Verfahren nach einem der vorhergehenden Ansprüche, worin -Nu für 1-Methyl-1-pyrrolidinio steht.

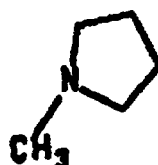
7. Verfahren zur Herstellung einer stabilen, kristallinen Verbindung der Formel I



worin X für HCl oder HI steht, die im wesentlichen frei an Δ^2 Isomer ist, wobei man eine Lösung der Verbindung der Formel IVa



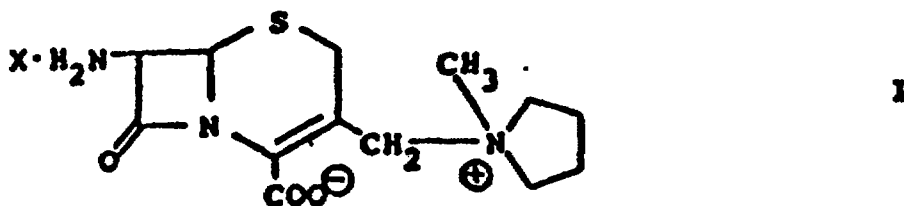
in 1,1,2-Trichlortrifluorethan (Freon TF) oder 1,1,1-Trichlortrifluorethan mit mindestens einem Äquivalent der Verbindung der Formel



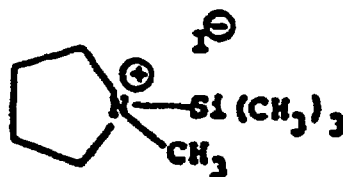
dann mit mindestens einem Äquivalent Jodtrimethylsilan pro Äquivalent der Verbindung IVa bei einer Temperatur von 5 bis 45°C behandelt, anschließend zur Beseitigung der Silylgruppen mit einem Niedrigalkanol bei einer Temperatur von -10 bis 25°C behandelt und mit HCl oder HI zur Bildung des Hydrochlorids oder Hydrojodids der Verbindung der Formel I ansäuert.



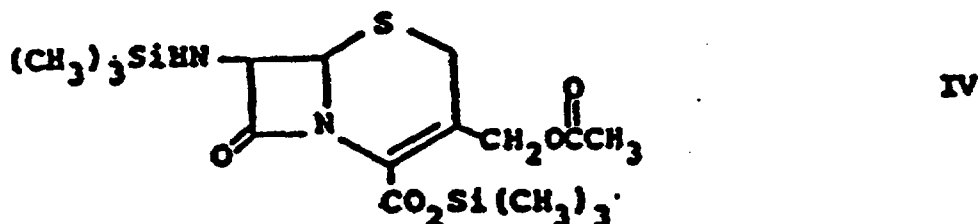
8. Verfahren zur Herstellung einer stabilen, kristallinen Verbindung der Formel I



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worin X für HCl oder HI steht, die im wesentlichen frei an Δ^2 Isomer ist, wobei man die Verbindung der Formel



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mit einer Lösung der Verbindung der Formel IV

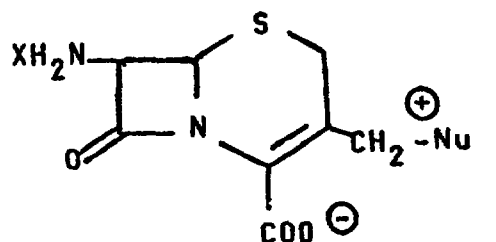


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in 1,1,2-Trichlortrifluorethan (Freon TE) oder 1,1,1-Trichlortrifluorethan bei einer Temperatur von 5 bis 45°C behandelt und anschließend zur Beseitigung der Silylgruppen mit einem Niedrigalkanol bei einer Temperatur von -10 bis 25°C behandelt und mit HCl oder HI zur Bildung des Hydrochlorids oder Hydrojodids ansäuert.

9. Verfahren nach Anspruch 8, wobei die Verbindung der Formel IV gemäß Anspruch 5 hergestellt wird.

Revendications

- 45 1. Procédé pour la préparation d'un composé stable, cristallin, choisi dans le groupe ayant la formule VIIb



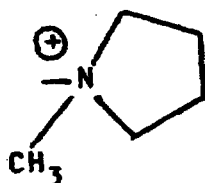
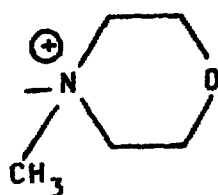
VIIb



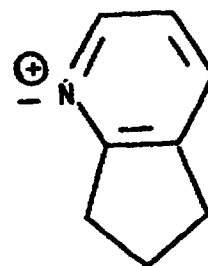
où X est HCl ou HI, et $-\text{Nu}^{\oplus}$ est

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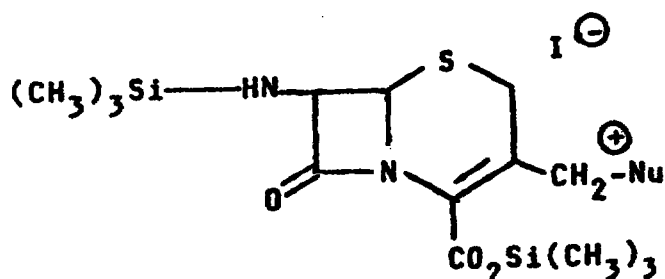
ou



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qui est sensiblement exempt de l'isomère Δ^2 , qui comprend le traitement d'un composé de la formule IXb

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IXb

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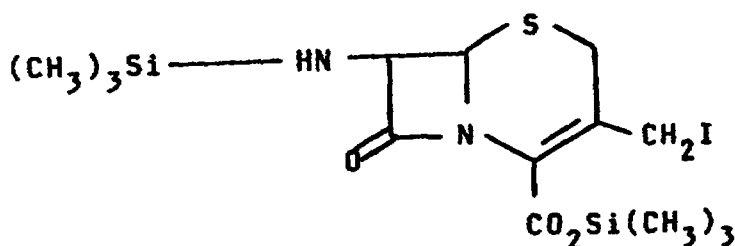
où $-\text{Nu}^{\oplus}$ est tel que défini ci-dessus, dans du 1,1,2-trichlorotrifluoroéthane (Fréon TF) ou du 1,1,1-trichlorotrifluoroéthane avec un alcool inférieur à une température de -10 à 25°C pour éliminer les groupes silyles, avec ensuite acidification avec HCl ou HI pour former le sel de chlorhydrate ou d'iodhydrate du composé de formule VIIb.

2. Procédé de la revendication 1, où l'alcool inférieur est le méthanol.

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3. Procédé de la revendication 1, où le composé de formule IXb est préparé par réaction du composé de formule III

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III

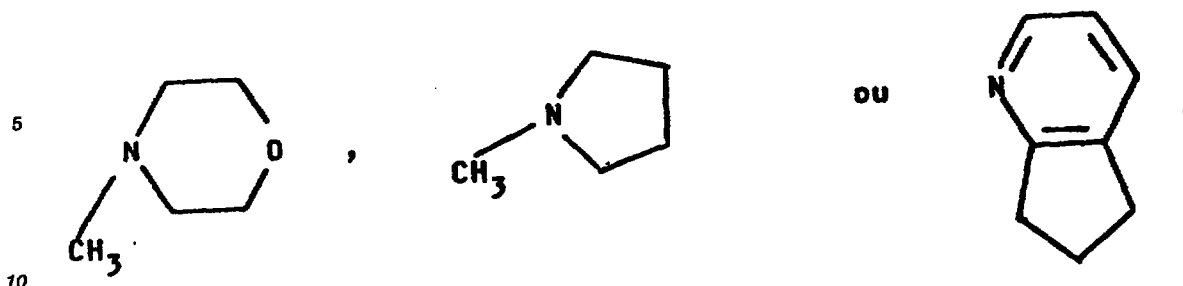
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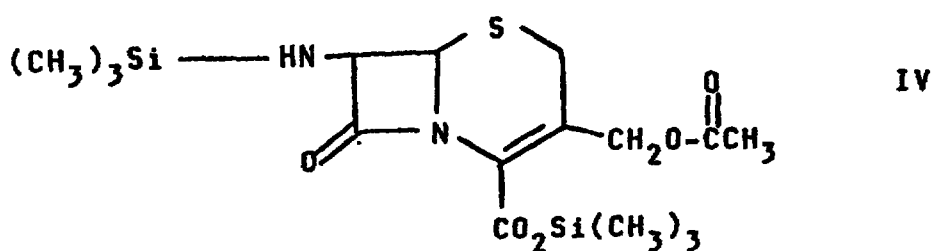
dans le 1,1,2-trichlorotrifluoroéthane (Fréon TF) ou le 1,1,1-trichlorotrifluoroéthane à une température de -10 à 25°C avec un composé de la formule

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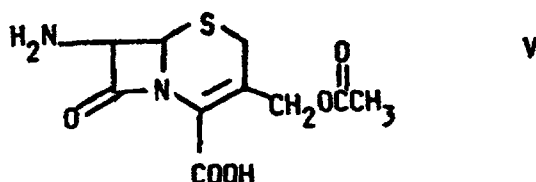


- 15 4. Procédé de la revendication 3, où le composé de formule III est préparé par traitement du composé de formule IV



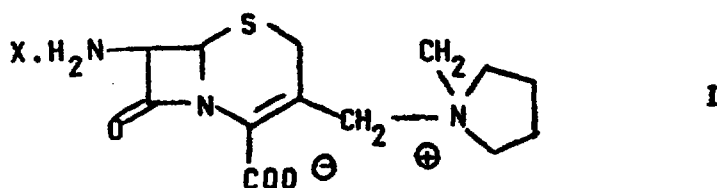
dans le 1,1,2-trichlorotrifluoroéthane (Fréon TF) ou le 1,1,1-trichlorotrifluoroéthane à une température de 5 à 45 °C avec au moins un équivalent d'iodotriméthylsilane par équivalent du composé IV.

- 30 5. Procédé de la revendication 4, où le composé de formule IV est préparé par traitement du composé de formule V

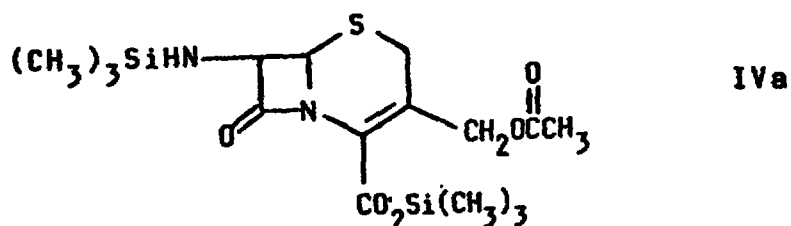


dans le 1,1,2-trichlorotrifluoroéthane (Fréon TF) avec de l'hexaméthylidisilazane et 0,01 à 0,1 équivalent d'iodotriméthylsilane à une température de 5 à 45 °C.

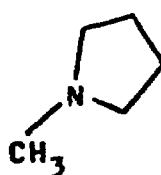
- 45 6. Procédé selon l'une quelconque des revendications précédentes où -Nu est 1-méthyl-1-pyrrolidinio.
- 50 7. Procédé de préparation d'un composé stable et cristallin de la formule I



où X est HCl ou HI, qui est sensiblement exempt de l'isomère Δ^2 , qui comprend le traitement d'une solution du composé de formule IVa

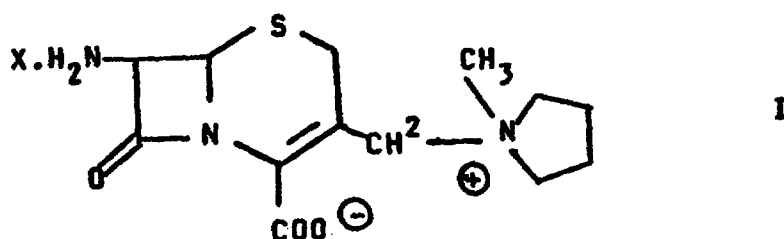


15 dans le 1,1,2-trichlorotrifluoroéthane (Fréon TF) ou le 1,1,1-trichlorotrifluoroéthane avec au moins un équivalent d'un composé de la formule

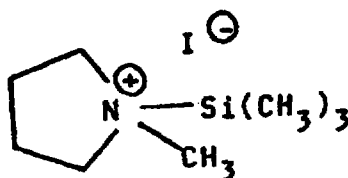


25 puis avec au moins un équivalent d'iodotriméthylsilane par équivalent du composé IVa à une température de 5 à 45° C avec ensuite le traitement avec un alcool inférieur à une température de -10 à 25° C pour éliminer les groupes silyles, et acidification avec HCl ou HI.

8. Procédé pour la préparation d'un composé stable et cristallin de formule I

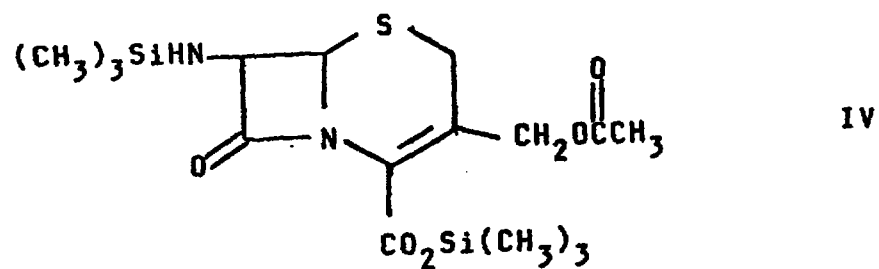


45 où X est HCl ou HI, qui est sensiblement exempt de l'isomère Δ^2 , qui comprend le traitement d'un composé de la formule



avec une solution du composé de formule IV





dans le 1,1,2-trichlorotrifluoroéthane (Fréon TF) ou le 1,1,1-trichlorotrifluoroéthane à une température de 5 à 45°C, avec ensuite traitement avec un alcool inférieur à une température de -10 à 25°C pour éliminer les groupes silyles, et acidification avec HCl ou HI.

9. Procédé selon la revendication 8, où le composé de formule IV est préparé selon la revendication 5.

