

ION SOURCE CLEANING IN SEMICONDUCTOR PROCESSING SYSTEMS

FIELD OF THE INVENTION

[0001] The present invention relates to monitoring, control and cleaning of material deposition on components of semiconductor processing systems, in particular to ion implantation systems.

DESCRIPTION OF THE RELATED ART

[0002] Ion implantation is used in integrated circuit fabrication to accurately introduce controlled amounts of dopant impurities into semiconductor wafers and is a crucial process in microelectronic/semiconductor manufacturing. In such implantation systems, an ion source ionizes a desired dopant element gas and the ions are extracted from the source in the form of an ion beam of desired energy. Extraction is achieved by applying a high voltage across suitably shaped extraction electrodes, which incorporate apertures for passage of the extracted beam. The ion beam is then directed at the surface of the workpiece, such as a semiconductor wafer, in order to implant the workpiece with the dopant element. The ions of the beam penetrate the surface of the workpiece to form a region of desired conductivity.

[0003] Several types of ion sources are commonly used in commercial ion implantation systems, including the Freeman and Bernas types using thermoelectrodes and powered by an electric arc, a microwave type using a magnetron, indirectly heated cathode sources, and RF plasma sources, all of which typically operate in a vacuum. The ion source generates ions by introducing electrons into a vacuum chamber filled with the dopant gas (commonly referred to as the "feedstock gas"). Collisions of the electrons with dopant atoms and molecules in the gas results in the creation of ionized plasma consisting of positive and negative dopant ions. An extraction electrode with a negative or positive bias will respectively allow the positive or negative ions to pass through the aperture and out of the ion source as a collimated ion beam, which is

accelerated towards the workpiece. Feedstock gases include, but are not limited to, BF_3 , $\text{B}_{10}\text{H}_{14}$, $\text{B}_{18}\text{H}_{22}$, PH_3 , AsH_3 , PF_5 , AsF_5 , H_2Se , N_2 , Ar, GeF_4 , SiF_4 , O_2 , H_2 , and GeH_4 .

[0004] Presently, there are upwards of 10-15 implantation steps in the fabrication of state of the art devices. Increasing wafer sizes, decreasing critical dimensions, and growing circuit complexity are placing greater demands on ion implant tools, with respect to better process control, the deliverance of high beam currents at low energies, and a decrease in the mean time between failures (MTBF).

[0005] The parts of the ion implanter tool that require the most maintenance include: the ion source, which must be serviced after approximately 100 to 300 hours of operation, depending on its operating conditions; the extraction electrodes and the high voltage insulator, which usually require cleaning after a few hundred hours of operation; the forelines and vacuum pumps of the ion implant vacuum system, including the ion source turbo pump and its associated forelines. Additionally, the filament of the ion source can require replacement after operation.

[0006] In the ideal case, all feedstock molecules would be ionized and extracted, but in reality a certain amount of feedstock decomposition occurs, which results in the deposition on and contamination of the ion source region. For example, phosphorous residue (*e.g.*, deriving from use of a feedstock gas such as phosphine) readily deposits on surfaces in the ion source region. The residue can form on low voltage insulators in the ion source, causing electrical short circuits, which can interrupt the arc required to produce thermionic electrons. This phenomenon is generally known as "source glitching," and it is a major contributor to ion beam instability, and may eventually cause premature failure of the source. The residue also forms on the high voltage components of the ion implanter, such as the source insulator or the surfaces of the extraction electrodes, causing energetic high voltage sparking. Such sparks are another contributor to beam instability, and the energy released by these sparks can damage sensitive electronic components, leading to increased equipment failures and poor MTBF.

[0007] Another common problem can occur with antimony (Sb^+) implant using Sb_2O_3 as the solid doping material, which can be exacerbated by running boron (B)

even after only a few hours of Sb⁺ implantation. The boron beam current can significantly deteriorate with performance and lifetime of ion source significantly compromised. The cause of this degradation in performance is due to the excessive Sb deposited on the source chamber and its components. Failure of the ion source significantly reduces implanter productivity as throughput is reduced due to more frequent preventive maintenance or less beam current. Since Sb implantation is widely used in analog bipolar devices, and also used as n-type doping for shallow junction formation for MOS (metal oxide semiconductor) devices, there is a need in the art to develop a method which can remove Sb deposited on the source chamber and components thereof when Sb⁺ is used as dopant, particularly when switching to B after Sb implant.

[0008] Furthermore, dopant atoms such as B, Ge, Si, P, and As may deposit in the ion source turbo pump, and downstream in its associated vacuum forelines. Over time, these deposits build up and require cleaning, which, historically, has been done manually. But some of the deposits (e.g. solid phosphorus) are pyrophoric and can ignite during the manual maintenance operations. Not only is this a fire hazard, but toxic compounds also can be liberated. There is thus a need in the art to develop an improved method which can desirably clean the deposits in-situ with a gaseous cleaning agent.

[0009] In another cause of ion source failure, various materials (e.g. tungsten, W) can accumulate on the cathode during extended ion implantation processes. Once these reach a critical level, the cathode power can no longer maintain temperature sufficient to meet the beam current setpoint. This causes loss of ion beam current requiring replacement of the ion source. The resultant performance degradation and short lifespan of the ion source reduces productivity of the ion implanter system.

[0010] Yet another cause of ion source failure is the erosion (or sputtering) of cathode material. For example, metallic materials (e.g. W, Mo etc.) from the cathode are sputtered by ions in the plasma of the arc chamber. Because sputtering is dominated by the heaviest ions in the plasma, as ion mass increases the sputtering effect may worsen. In fact, continued sputtering of material "thins" the cathode

eventually leading to formation of a hole in the cathode ("cathode punch-through"). Performance and lifetime of the ion source are greatly reduced as a result. The art thus continues to seek methods which can maintain a balance between the accumulation and erosion of material on the cathode to prolong the ion source life.

[0011] Further residue may result from reaction of the source material with the components of the ion implantation system, depending on the conditions within the system. Such reaction may result in deposition of residue on additional components of the system.

[0012] Deposits are common on ion source elements such as the filament and the repeller electrode. Such internal deposits are generally composed of arc chamber material, and are most commonly seen when running high plasma power with a fluoride source feedstock in conjunction with an arc chamber constructed of tungsten or molybdenum. While the ion source life expectancy for ion implantation systems using non halide-containing source materials is generally around 100 to 300 hours, with some halide-containing materials such as GeF_4 , the ion source life can be as low as 10 to 50 hours due to the detrimental effects of internal deposits on source operation.

[0013] In addition to the operational difficulties caused by residues in the ion implanter, there are also significant personnel safety issues due to the emission of toxic or corrosive vapors when components are removed for cleaning. The safety issues arise wherever residues are present, but are of particular concern in the ion source region because the ion source is the most frequently maintained component of the ion implanter. To minimize down time, contaminated ion sources are often removed from the implanter at temperatures significantly above room temperature, which increases the emission of vapors and exacerbates the safety issue.

[0014] Prior methods of dealing with the above difficulties have included attempts to prevent formation of the deposits and cleaning of the resulting deposits on the extraction electrode and the ion source (*i.e.* on the extraction electrode, as discussed in published U.S. Patent Application 2006/0272776, published U.S. Patent Application 2006/0272775 and published International Patent Application WO 2005/059942 A2).

However, a need for additional processes of cleaning of all elements of an ion implantation system remains.

[0015] It would therefore be desirable in the art of ion implantation to provide an *ex situ* cleaning process with a separate cleaning station whereby contaminated components that have been removed from the implanter can be cleaned safely without any mechanical abrasion which might damage delicate components such as graphite electrodes. It would therefore also be a significant advance in the art of ion implantation to provide an off-line cleaning station that could be used to selectively and non-destructively clean components following removal from the implant system and with minimal down time.

[0016] It would also be a significant advance in the art of ion implantation to provide an *in situ* cleaning process for the effective, selective removal of unwanted residues deposited throughout the implanter, particularly in the ion source region, during implantation. Such *in situ* cleaning would enhance personnel safety and contribute to stable, uninterrupted operation of the implantation equipment.

[0017] An *in situ* cleaning process is performed without disassembly of the process chamber. For *in situ* processes, a gaseous agent is flowed through the process chamber to remove accumulated films, in either a continuous, pulsed, or hybrid continuous-pulsed fashion. Depending on the situation, a plasma may or may not be generated during such cleaning.

[0018] Plasmaless or dry cleaning processes using chlorine trifluoride (ClF_3) and other fluorine source materials (e.g., CF_4 , NF_3 , C_2F_6 , C_3F_8 , SF_6 and ClF_3) can be used for removing solid residues from semiconductor processing chambers, e.g., by reacting with solid residues to form volatile reaction products that are removable from the processing chamber by vacuum or other removal conditions, and in such instances these cleaning reagents may require elevated temperature cleaning conditions. See Y. Saito et al., "Plasmaless Cleaning Process of Silicon Surface Using Chlorine Trifluoride", *APPLIED PHYSICS LETTERS*, vol. 56(8), pp. 1119-1121 (1990); also see D.E. Ibbotson et al., "Plasmaless Dry Etching of Silicon with Fluorine-Containing Compounds", *JOURNAL OF APPLIED PHYSICS*, vol. 56(10), pp. 2939-2942 (1984).

[0019] U.S. Patent No. 4,498,953 describes an *in-situ* cleaning method in which an interhalogen, such as BrF₅, BrF₃, ClF₃, or IF₅ is continuously flowed through the processing chamber while maintaining a predetermined pressure within the chamber. At the end of the treatment, the flow of the interhalogen gas is terminated. Such processes can produce byproduct containing Cl, Br, or I, as well as byproduct containing fluorine, thereby yielding significant amounts of hazardous waste material that require treatment or other disposition. In addition, such continuous flow cleaning is performed under very low pressure conditions, at which cleaning efficiency is substantially reduced.

[0020] In some ion source applications, strategic sequencing of BF₃, PH₃, and/or AsH₃ has been performed in order to achieve longer ion source lifetimes.

[0021] Additionally, GeF₄, SiF₄ and/or BF₃ have been used as cleaning gases, but are always used with the plasma struck in the source chamber.

[0022] The use of fluorine radicals or fluorine-containing interhalogens for cleaning of semiconductor processing equipment has associated deficiencies that have constrained its commercial viability. Fluorine radicals and fluorine-containing interhalogens, including ClF₃, are highly corrosive in character. Further, interhalogen compounds are severe irritants to the human respiratory tract. For example, the threshold human tolerance levels of ClF₃ vapor can be as low as 100 ppb, with an LC50 on the order of 1 hour at 300 ppm.

[0023] The art continues to seek new cleaning reagents and *ex situ* and *in situ* systems and processes, and associated monitoring and control apparatus and methods.

SUMMARY OF THE INVENTION

[0024] The present invention relates generally to apparatus and methods for monitoring, control, and cleaning of ion implantation systems or components thereof, as well as to compositions usefully employed for such cleaning.

[0025] In one aspect the invention provides a method of monitoring the condition of the filament of an ion implantation system during operation of the system comprising: (a) energizing a filament in an arc chamber of an ion source with an initial current sufficient to generate a plasma in said arc chamber; (b) measuring

current input to the filament to maintain the plasma in the arc chamber at a predetermined time of continued plasma generation; (c) comparing the current input measured at the predetermined time to the initial current, and (d) determining from such comparing whether material has been deposited on the filament or whether etching of the filament has occurred, wherein a larger current at the predetermined time, in relation to the initial current, is indicative of deposition of material on the filament, and a smaller current at the predetermined time, in relation to the initial current, is indicative of etching of the filament.

[0026] In another aspect the invention provides a method of controlling condition of a filament of an ion implantation system during operation of the system comprising: (a) energizing a filament in an arc chamber of an ion source with an initial current sufficient to generate a plasma in said arc chamber; (b) measuring current input to the filament to maintain the plasma in the arc chamber at a predetermined time of continued plasma generation; (c) comparing the current input measured at the predetermined time to the initial current, (d) determining from such comparing whether material has been deposited on the filament or whether etching of the filament has occurred, wherein a larger current at the predetermined time, in relation to the initial current, is indicative of deposition of material on the filament, and a smaller current at the predetermined time, in relation to the initial current, is indicative of etching of the filament, and (e) in response to the determining, removing deposited material from the filament or depositing additional material on the filament, to an extent reestablishing the initial current input or a current input within a predetermined range of said initial current input.

[0027] In still another aspect the invention provides a method of operating an ion implantation system including a filament in an arc chamber of an ion source, to maintain operating efficiency of the ion source, said method comprising contacting the filament with a tungsten reagent under conditions selected from the group consisting of:

- (a) conditions effecting deposition of tungsten on the filament; and
- (b) conditions effecting etching of deposited material from the filament.

[0028] The invention in a further aspect relates to a method of cleaning one or more components of an ion implantation system for at least partial removal of ionization-related deposits from said one or more components, said method comprising flowing a cleaning gas through the system, under conditions selected from the group consisting of:

- (a) conditions effecting deposition of material on the filament; and
- (b) conditions effecting etching of deposited material from the filament.

[0029] A further aspect of the invention relates to a method of maintaining a predetermined electrical resistance of a filament of an ion source in an arc chamber, comprising contacting the filament with a reagent effective to deposit material on the filament or to etch material from the filament depending on temperature in the arc chamber, and controlling temperature in the arc chamber to effect deposition or etching of material at the filament, so as to maintain said predetermined electrical resistance.

[0030] In another aspect, the invention relates to a method of cleaning an ion implantation system or one or more components thereof, to remove ionization-related deposits therefrom, comprising contacting the ion implantation system or one or more components thereof with BrF_3 under conditions in which said BrF_3 is chemically reactive with the deposits to effect at least partial removal thereof.

[0031] In another aspect, the invention relates to a method of cleaning a foreline of an ion implant system to remove an ionization-related deposit therefrom comprising contacting the foreline of an ion implantation system with a cleaning gas under conditions in which said cleaning gas is chemically reactive with the deposit to at least partially removal same. Such method can improve the performance and extend the lifetime of an ion implant system.

[0032] In another aspect, the invention relates to a method of improving the performance and extending the lifetime of an ion implant system, comprising contacting the cathode with a gas mixture comprising at least one cleaning gas and at least one deposition gas wherein said gas mixture balances the deposition of material on the cathode and erosion of the same or other material from the cathode.

[0033] Other aspects, features and embodiments of the invention will be more fully apparent from the ensuing disclosure and appended claims.

BRIEF DESCRIPTION OF THE DRAWINGS

[0034] Figure 1 is a graph of source lifetime data before and after introduction of *in situ* cleaning processes, showing the increase in lifetime due to the process.

[0035] Figure 2 is a graph showing effect of XeF_2 on the suppressor leakage current, as detailed in Example 1.

[0036] Figures 3A and 3B are photographs showing evidence of the cleaning effects *in situ* cleaning, as detailed in Example 1.

[0037] Figures 4A and 4B show cleaning effects *in situ* cleaning, as detailed in Example 5.

[0038] Figures 5A and 5B are graphs of the increasing filament weight (Figure 5A) and filament current (Figure 5B) over an elapsed time period with XeF_2 flow.

[0039] Figure 6 is a graph of the filament weight change as a function of filament current, with regard to tungsten transport within the system with XeF_2 flow.

DETAILED DESCRIPTION OF THE INVENTION

[0040] The present invention relates to apparatus and methods for monitoring, control and cleaning of semiconductor processing systems and/or components thereof, and to compositions for such cleaning.

[0041] In one aspect the invention relates to removal of deposits from semiconductor processing systems or components of semiconductor processing systems, in which the system or system component is contacted with a cleaning composition including a gas-phase reactive material.

[0042] As used herein, the term “gas-phase reactive material” is intended to be broadly construed as meaning material comprising halide compound(s) and/or complex(es), in gaseous or vapor form, ionic and plasma forms of such compound(s) and/or complex(es), and elements and ions derived from such compound(s), complex(es) and ionic and plasma forms. A gas-phase reactive material as utilized in

the broad practice of the present invention may also be variously referred to, without limitation, as a “gas-phase reactive composition,” a “cleaning agent,” a “cleaning gas,” an “etchant gas,” a “gaseous halide,” a “gaseous cleaning agent,” a “reactive halide,” a “cleaning compound,” a “cleaning composition,” a “cleaning vapor,” an “etchant vapor” or any combination of such terms.

[0043] As used herein, “ion source region” in reference to an ion implanter, includes the vacuum chamber, the source arc chamber, the source insulators, the extraction electrodes, the suppression electrodes, the high voltage insulators, the source bushing, the filament and the repeller electrode.

[0044] The invention contemplates the cleaning of semiconductor processing systems and components thereof, as well as other substrates and apparatus that are susceptible to formation of deposits thereon during their normal processing operation.

[0045] The present invention in various aspects provides an ion implantation system with the capability to grow/etch the filament in the ion source of the arc chamber, by appropriate control of temperature in the arc chamber to effect the desired filament growth or alternative filament etching.

[0046] Additional aspects of the invention relate to use of reactive gases such as WF_x , AsF_x , PF_x and TaF_x , wherein x has a stoichiometrically appropriate value or range of values, for cleaning regions of ion implanters, or components of implanters in *in situ* or *ex situ* cleaning arrangements, under plasma or elevated temperature conditions.

[0047] A still further aspect of the invention relates to use of BrF_3 for cleaning of ion implant systems or component(s) thereof, in *in situ* or *ex situ* cleaning arrangements, under ambient temperature, elevated temperature, or plasma conditions.

[0048] The operation of an ion implantation system results in deposition of ionization-related materials in the system and its components. The invention contemplates monitoring, controlling and/or cleaning of the ion implantation system or one or more components thereof, for at least partial removal of such ionization-related deposits from the system and/or components thereof. The cleaning

method involves contacting the system and/or components thereof with a cleaning composition comprising a gas-phase reactive material, under conditions enabling reaction of the gas-phase reactive material with the deposits to effect their at least partial removal.

[0049] In addition to the ionization-related deposits resulting from feedstock gases per se, it has also been found that deposits or residue formed within an ion implantation system may result from reactivity of feedstock gases with materials from which the system components are constructed. For example, the vacuum chamber of an ion implantation system may be constructed utilizing stainless steel or aluminum. System components within the vacuum chamber may be constructed utilizing graphite (*e.g.*, standard or vitreous), insulator materials (*e.g.*, boron nitride) and/or sealant materials such as Teflon®, Kel-F™, PEEK™, Delrin™, Vespel™, Viton™, Buna-N, silicon, etc. Other materials that can be present in the ion implantation system, and susceptible to deposit-producing chemical reactions therein, include, without limitation, ceramics, epoxy compositions containing lead oxide, aluminum nitride, aluminum oxide, silicon dioxide, and boron nitride.

[0050] The ion source itself can be constructed of tungsten, graphite, molybdenum or tantalum, sometimes with minor amounts of copper and silver. The ion source arc chamber is usually constructed of tungsten or molybdenum, or with a graphite body lined with tungsten or molybdenum. In such circumstance, a fluoride source feed material, *e.g.*, BF₃, GeF₄, SiF₄, AsF₅, AsF₃, PF₅, and/or PF₃, reacts at operating temperatures with the material of the arc chamber, *e.g.*, tungsten or molybdenum from the chamber or the lining of the chamber, to form an intermediate byproduct that in turn can migrate in the system and decompose to deposit tungsten or molybdenum and liberate fluorine.

[0051] For example, a feedstock gas such as GeF₄ will dissociate in the ion source chamber and the resulting free fluoride will attack the material of the arc chamber, such as tungsten. This reaction will occur with tungsten on a colder surface, so if the plasma is struck and the filament is therefore hot, the fluoride will react with the tungsten on the walls of the arc chamber, etching the walls and forming WF₆ gas. The

WF₆ will then deposit tungsten on the hot filament, causing it to grow in size.

[0052] While GeF₄ produces large amounts of free fluorine, feedstock gases such as BF₃ or SiF₄ produce lesser amounts of free fluorine and correspondingly lower levels of tungsten deposition on the filament, which although lower, are still significant.

[0053] Non-fluorine-containing feedstock gases, such as PH₃ and AsH₃, can be problematic in causing metal from the filament to deposit onto walls of the arc chamber, with resultant thinning of the filament.

[0054] The present invention therefore contemplates cleaning of an ion implantation system or component(s) thereof, for at least partial removal of ionization-related deposits that are the same as the material of the arc chamber.

[0055] Cleaning in accordance with the invention can be carried out in ion implantation systems in which multiple feedstock gases are concurrently introduced to the system. Feedstock gases can also be used concurrently with one or more gas-phase reactive materials, or can be pulsed alternately with one or more gas-phase reactive materials into the system.

[0056] The ionization-related deposits to which the cleaning methods of the invention are directed, include a variety of materials that may interfere with normal operation of the ion implantation system, such as by forming and accumulating in the ion source or other ionization process equipment. The deposited material can variously comprise, consist, or consist essentially of, silicon, boron, phosphorous, germanium, arsenic, tungsten, molybdenum, selenium, antimony, indium, carbon, aluminum and/or tantalum.

[0057] Ionization-related deposits in the ion source arc chamber and on extraction electrodes can flake and form small particles. These particles, once formed, can be transported in an ion beam, *e.g.*, a beam of dopant ions being implanted into a wafer. If such transported particles reach the wafer, the resulting particle contamination on the wafer can severely reduce the yield of useful devices that can be manufactured on the wafer. The cleaning processes of the present invention remove such ionization-related deposits before they are able to form flakes and particles, and

thereby achieve reduction of particles on product wafers and increased yield of semiconductor devices.

[0058] The gas-phase reactive material, or cleaning gas, used for cleaning in accordance with the invention can include any material that is effective to at least partially remove the ionization-related deposits in the ion implantation system.

[0059] The invention also contemplates the use of gas-phase reactive materials to remove ionization-related deposits from undesired locations, and/or to deposit material in desired locations, by appropriate control of the reaction. In specific embodiments of the invention, tungsten constitutes the material being removed as an undesired deposit, and in other embodiments, tungsten is desirably deposited on surfaces benefiting from its presence. Accordingly, a gas that is reactive to form a tungsten fluoride intermediate product, such as XeF_2 , GeF_4 , SiF_4 , BF_3 , AsF_5 , AsF_3 , PF_5 , and/or PF_3 , can be used in the controlling and cleaning methods of the invention. Additionally, tungsten fluoride gases such as WF_6 , WF_5 , and/or WF_4 may be utilized directly in the controlling and cleaning methods of the invention. As such, gas-phase reactive materials of the invention include, but are not limited to, XeF_2 , GeF_4 , SiF_4 , BF_3 , AsF_5 , AsF_3 , PF_5 , PF_3 , F_2 , TaF_3 , TaF_5 , WF_6 , WF_5 , and/or WF_4 .

[0060] In various specific embodiments, the gas-phase reactive material can be co-administered with a "cleaning enhancement agent" or "co-reactant" that will increase the volatility of the gas-phase reactive material, resulting in removal of more deposits than with the gas-phase reactive material without the cleaning enhancement agent or co-reactant. For example, removal of iridium deposits by XeF_2 can be enhanced by co-administration of Lewis bases and electron back-bonding species. In specific applications, carbon monoxide, trifluorophosphine, and trialkylphosphines can be employed.

[0061] As a further example, in an ion implantation system in which feed gases are ionized in plasma sustained in an arc chamber having tungsten walls on which a filament on one side and a repeller on the other side are mounted and separated from the walls by ceramic insulators, the components of the arc chamber may become contaminated with decomposition products of feed gases, elements of the arc chamber,

and carbon.

[0062] In such circumstance, a cleaning agent useful for removal of metal contaminants such as tungsten that form volatile fluorides, such as XeF_2 , can be combined with an oxygen-containing additive that is effective for removing contaminant carbon by converting it to CO , CO_2 , and/or COF_2 . Oxygen-containing additive components useful for such purpose, in specific embodiments of the invention, include, without limitation, NO , N_2O , NO_2 , CO_2 and/or O_2 .

[0063] The invention therefore contemplates cleaning compositions including both a cleaning agent effective for removing a metal contaminant by reaction forming a volatile (gaseous) fluoride compound of such metal, and a cleaning agent effective for removing carbon contaminant by forming a volatile oxide or oxyfluoride therefrom. These cleaning reagents can be flowed into the arc chamber concurrently or sequentially.

[0064] In one embodiment, these reagents are concurrently flowed into the arc chamber under ionization conditions so that both cleaning agents are ionized to convert metal and carbon contaminants into volatile compounds that are easily removed from the chamber by mechanical pumping thereof.

[0065] The conditions enabling reaction of the gas-phase reactive material and the deposits may include any suitable conditions of temperature, pressure, flow rate, composition, etc. under which the gas-phase reactive material contacts and chemically interacts with the contaminant so as to remove such material from the substrate, e.g., the surface of the implanter equipment contaminated with deposited material.

[0066] Examples of various conditions that may be employed include, without limitation, ambient temperatures, temperatures in excess of ambient temperature, presence of plasma, absence of plasma, sub-atmospheric pressure, atmospheric pressure, and superatmospheric pressure.

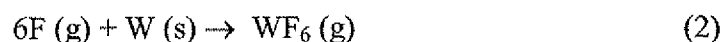
[0067] Specific temperatures for gas-phase reactive material contacting to remove deposits in various embodiments can range from about 0°C to about 1000°C . Contacting can involve delivery of the gas-phase reactive material in a carrier gas, or in a neat form, or in admixture with a further cleaning agent, dopant, etc. The

gas-phase reactive material can be heated for chemical reaction with deposits that are at ambient temperature, in order to increase the kinetics of the reaction.

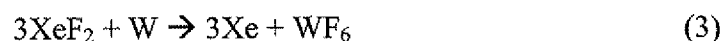
[0068] Reaction between the gas-phase reactive material and contaminant deposits can be monitored and/or regulated, based on varying characteristics of the reaction between the cleaning agent and the contaminant. Such reaction characteristics can include pressure, time, temperature, concentration, presence of a particular species, rate of pressure change, rate of concentration change (of a particular species), change of current, etc. Accordingly, the introduction of the gas-phase reactive material to the system can be terminated based on attainment of a predetermined characteristic of the reaction, such as a predetermined pressure in the vacuum chamber, passage of a predetermined amount of time, or a predetermined temperature, concentration of a specific element in the system, presence of a particular by-product, reaction product or other species in the system, or realization of a predetermined current condition in the monitoring operation.

[0069] Tungsten deposits can result from reaction of feed gases with the arc chamber of an implanter system. The method used in cleaning such deposits may be dependent on the temperature gradient of the system and/or the current flowing to and through the filament, and/or any other characteristics that are usefully determined and able to be monitored.

[0070] For example, fluorine from the feed material may react with the arc chamber at a first temperature to form WF_6 by reactions (1) or (2):



There may also be reaction between the cleaning gas and tungsten material of the arc chamber, such as:



Alternatively, WF_6 (or WF_5 or WF_4) may be provided directly to the system.

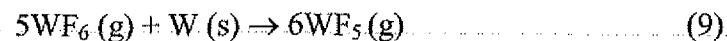
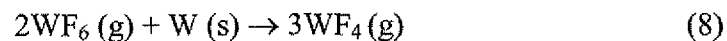
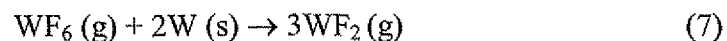
[0071] The tungsten fluoride once formed or otherwise present in the system can then migrate to another location in the system. Depending on the temperature of that

other location, the tungsten fluoride will etch or deposit tungsten at that location. At the filament, the temperature will primarily depend on the actual current flux passing through it. Temperatures at other locations in the arc chamber may vary depending on the specific location and design of the arc chamber, the filament current, as well as other non-filament electrical currents.

[0072] If the second location is at high temperature, the tungsten fluoride is decomposed, tungsten is deposited and fluorine is released, with the tungsten deposit growing in size as long as tungsten fluoride continues to be present. Deposition reactions can include the following reactions (4), (5) and/or (6):



[0073] Contrarily, if the second location is at moderate temperature, the tungsten fluoride can etch that location, removing tungsten and retaining fluorine in the reaction product, so that the etched location shrinks as etching continues. Such etching reactions can include the following reactions (7), (8) and/or (9):



[0074] Therefore, for removal of tungsten deposits, the temperature of the components bearing the deposits can be selected to maximize the rate and extent of removal.

[0075] In other embodiments of the invention, boron and/or molybdenum deposits in the arc chamber are removed in a corresponding manner.

[0076] The contacting of the cleaning agent with the process equipment in the process of the invention can be carried out with monitoring of the pressure change during the contacting, with the contacting being terminated when the pressure change goes to zero.

[0077] Alternatively the contacting may be conducted with monitoring of partial pressure of the gas-phase reactive material, or reactants deriving therefrom, or reaction products produced in the contacting, with the contacting being terminated when the partial pressure reaches a predetermined value, *i.e.*, an end point. Such end point monitoring can for example be carried out with a suitable end point monitor, *e.g.*, an end point monitor of a type as more fully described in U.S. Patent No. 6,534,007 and U.S. Patent Application Nos.: 10/273,036; 10/784,606; 10/784,750; and 10/758,825, or a thermopile infrared (TPIR) or other infrared detector.

[0078] In another embodiment, the contacting may be conducted by controlled flow of the gas-phase reactive material using components of the process equipment system that allow regulation of the partial pressure of gas-phase reactive material and therefore control of the reaction rate.

[0079] In still another embodiment, a continuous flow of the gas-reactive material, at a pre-determined flow rate, is employed to carry out the cleaning operation.

[0080] As discussed hereinabove with reference to reactions (1) – (9), ionization-related deposits of tungsten can be deposited at very high temperatures and etched at low to moderate temperatures. Where the location of the deposition or etching is the filament of the implanter system, the temperature and current flux are directly related to one another. When the filament is being etched, the filament becomes thinner and the resistance to electrical current will increase as the cross-section of the filament decreases, so that current flow through the filament is reduced. If the conditions at the filament promote deposition thereon, the resistance to electrical current will decrease with continued deposition, as the cross-section of the filament increases and the filament becomes thicker, with correspondingly increased current flow therethrough.

[0081] In another aspect, the invention relates to a method of monitoring the deposition on the source filament and resulting growth of the filament, involving monitoring of the current flow through the filament. As the filament cross-section increases due to deposition, the resistance to the electrical current will decrease and the current will increase in order to maintain the filament at the temperature required

to support the plasma in the arc chamber. A monitored increase in current can therefore be used to indicate a need for filament cleaning.

[0082] In a further aspect, the invention relates to a method of monitoring etching or cleaning of the filament, by monitoring current flow through the filament. As the filament cross-section decreases due to etching, sputtering or evaporation, the resistance to the electrical current will increase and the current will decrease in order to maintain the filament at the temperature required to support the plasma in the arc chamber. A monitored decrease in such current can therefore be used to indicate a need for deposition of additional material to an etched filament, or for termination of a cleaning or ionization process.

[0083] Another implementation of the invention comprises a method of controlling the condition of the filament, based on monitoring of current flowed through the filament as detailed above.

[0084] In one embodiment, a decrease in the monitored filament current provides an indication that the filament is close to breaking, in response to which a gas-phase reactive material is flowed into the system, *e.g.*, while the plasma is struck, or alternatively with the plasma off but the filament still hot (for example, $\sim 200^{\circ}\text{C}$), to induce reaction producing deposition of metal on the filament, *e.g.*, tungsten from the arc chamber walls. Such reaction may be allowed to proceed until the current is within a predetermined range for effective operation of the ion implantation system, indicating that the filament has been "regrown" to a satisfactory extent.

[0085] In another embodiment, an increase in the monitored filament current provides an indication that the filament is growing, due to material deposition. In response, a gas-phase reactive material is flowed into the system after allowing the filament to cool for a predetermined period of time, or to a predetermined temperature (which may for example be in a range of from room temperature up to about 1200°C) so that the filament is cooled sufficiently to allow etching of the filament. The subsequent etching reaction mediated by the gas-phase reactive material may thereafter be allowed to proceed until the current is within a predetermined range for effective operation of the ion implantation system, indicating that the filament has

been thinned to an appropriate level.

[0086] The method of the invention can therefore be used for removing from a substrate a deposit including at least one of boron, silicon, arsenic, phosphorus, germanium, tungsten, molybdenum, selenium, antimony, indium, tantalum and carbon, by contacting the substrate with a gas-phase reactive material for sufficient time to at least partially remove the deposit from the substrate. The gas-phase reactive material for such purpose may include one or more of XeF_2 , XeF_4 , XeF_6 , GeF_4 , SiF_4 , BF_3 , AsF_5 , AsF_3 , PF_5 , PF_3 , F_2 , TaF_3 , TaF_5 , WF_6 , WF_5 , WF_4 , NF_3 , IF_5 , IF_7 , KrF_2 , SF_6 , C_2F_6 , CF_4 , Cl_2 , HCl , ClF_3 , ClO_2 , N_2F_4 , N_2F_2 , N_3F , NFH_2 , NH_2F , HOBr , Br_2 , BrF_3 , C_3F_8 , C_4F_8 , C_5F_8 , CHF_3 , CH_2F_2 , CH_3F , COF_2 , HF , C_2HF_5 , $\text{C}_2\text{H}_2\text{F}_4$, $\text{C}_2\text{H}_3\text{F}_3$, $\text{C}_2\text{H}_4\text{F}_2$, $\text{C}_2\text{H}_5\text{F}$, C_3F_6 , COCl_2 , CCl_4 , CHCl_3 , CH_2Cl_2 , and CH_3Cl .

[0087] Fluorinated xenon compounds can be employed as cleaning agents and plasma source reagents in the practice of the invention, and can include any suitable number of fluorine atoms. A higher ratio of F to Xe enables relatively faster and more efficient cleaning, in relation to lower F/Xe compounds. Higher vapor pressures increase delivery rates of the cleaning agent and enable delivery of more material.

[0088] In one embodiment of the invention, xenon hexafluoride is employed as a cleaning agent or plasma source reagent. Although XeF_6 vapor pressure at room temperature is approximately seven times higher than that of XeF_2 , XeF_6 , as well as XeF_4 , is very reactive with water. XeF_6 is most advantageously employed in cleaning environments not involving the presence or generation of water, hydrocarbons, hydrogen or reducing agents. However, when cleaning compounds with a lower vapor pressure are used, adjustments to the flow circuitry may be required in order to avoid undue pressure drops in the flow path and maintain suitably high delivery rates of the cleaning agent.

[0089] Apparatus to effectuate the methods of the invention may be constructed and arranged in any suitable manner to accommodate the cleaning with the gas-phase reactive material.

[0090] In one embodiment, the invention provides an ion implantation and cleaning assembly, comprising (i) an ion implantation system including one or more

components accumulating ionization-related deposits thereon during ion implantation processing in the system, (ii) a cleaning assembly including a cleaning composition source containing a cleaning composition including a gas-phase reactive material, *e.g.*, a halide compound that is reactive with the deposits to effect at least partial removal thereof from the one or more components under cleaning conditions involving contact of the cleaning composition with the deposits, (iii) flow circuitry adapted to transport the cleaning composition from the cleaning composition source to the one or more components for contacting therewith under the cleaning conditions, and (iv) flow componentry adapted to control flow of the cleaning composition through the flow circuitry during the cleaning conditions to effect the at least partial removal of the deposits from the one or more components.

[0091] The flow componentry in the above-described assembly can be of any suitable type, including for example valves, valve actuators, flow restrictors, regulators, pumps, mass flow controllers, pressure gauges, residual gas analyzers, central processing units, membranes, etc. Such flow componentry is suitably adapted to operate under the specific cleaning conditions employed.

[0092] The one or more components in the implanter apparatus that accumulate ionization-related deposits thereon during ion implantation processing in the system, can be of any suitable type, *e.g.*, vacuum chambers, arc chambers, electrodes, filaments, high voltage bushings, magnet waveguides, wafer handling components, clamp rings, wheels, discs, etc. In one embodiment, the component is a vacuum chamber or a component contained therein.

[0093] The cleaning composition source may comprise a material storage and dispensing package containing the cleaning composition. The material storage and dispensing package includes a vessel that may, for example, be of generally cylindrical shape, defining an interior volume therein. In a specific embodiment, the cleaning composition may be a solid at ambient temperature conditions and such cleaning composition may be supported on an enhanced surface area within the vessel. Such enhanced surface area may include structures therein, such as trays, as described in U.S. Patent No. 6,921,062, or porous inert foam, *e.g.*, of anodized aluminum,

stainless steel, nickel, bronze, etc., to provide a consistent rate of evaporation of the cleaning material and in turn provide sufficient vapor pressure for dispensing and ionizing steps of the associated cleaning process. Where trays are utilized, the cleaning composition may be supported on surfaces of trays having flow passage conduits associated therewith, for flow of vapor upwardly in the vessel to a dispensing port thereof, in dispensing operation.

[0094] The flow circuitry in the above-described equipment arrangement is adapted to transport the cleaning composition from the cleaning composition source to the arc chamber under cleaning conditions. Such adaptation may be based on various properties of the cleaning composition. For example, high conductance may be used to avoid unnecessary pressure drops in the flow path when the cleaning composition has a low vapor pressure. Methods of maximizing flow conductance and minimizing flow constriction are well known in the art.

[0095] In all cleaning methods of the invention, the cleaning may optionally be utilized with additional methods and apparatus for extending the lifetime of the ion implantation system, in particular the ion source. Such lifetime-extending approaches may include modification of an ion implantation system to accommodate specific substrate, deposited materials and/or gas-phase reactive materials. System apparatus modifications may include, without limitation, the provision of: extraction electrodes with active thermal control systems, actively heated extraction electrodes that reduce the frequency/occurrence of electrical discharges, extraction electrodes comprising metals, preferably aluminum, molybdenum or alumina (Al_2O_3), remote plasma sources, association of extraction electrodes with heaters, association of extraction electrodes with cooling devices, smooth, featureless extraction electrodes, plasma chambers arranged to receive source gases capable of being disassociated by plasma to produce flows of reactive gas through chamber outlets and conduits for transporting reactive gases to the ionization chamber, temperature detectors designed to detect substantial termination of exothermic reactions of the reactive gas with contamination on surfaces in process systems, protection of components in the process equipment that are susceptible to harm by gas-phase reactive materials (*e.g.*, provision of shields

against gas-phase reactive materials, around components that are sensitive to such materials), and/or use of system components comprising aluminum or alumina.

[0096] Methods to extend the lifetime of the process equipment may include, without limitation: actively heating extraction electrodes to reduce the frequency and occurrence of electrical discharges, heating extraction electrodes above the condensation temperature of source materials that are delivered to the ion source, actively controlling temperature of extraction electrodes adapted to the specific type of ion source employed (*e.g.*, heating or cooling the electrode, in combination with a heated or cooled ion source), and/or maintenance of extraction electrodes at elevated temperature during extraction. Such additional apparatus modifications and methods are more fully described in U.S. Patent Application Publication Nos. 2006/0272776 and 2006/0272775 and International Patent Publication No. WO05/059942, hereby incorporated by reference in their entirety.

[0097] The ion implantation system in a specific embodiment includes an arc chamber and a dopant source, in which the dopant source may for example comprise BF_3 , XeF_2 , AsH_3 , PH_3 , GeF_4 , SiF_4 , H_2Se , AsF_5 , AsF_3 , PF_5 , PF_3 or other boron, silicon, arsenic, phosphorus or germanium-containing dopant sources.

[0098] In another implementation, the invention relates to an ion implantation method, involving plasma generation in an arc chamber of an ion implantation system from a dopant source gas flowed through the arc chamber to form dopant source ions for implantation, wherein during at least part of the time during which the dopant source gas is flowed through the arc chamber, gas-phase reactive material is flowed through the arc chamber concurrently with the dopant source gas, to effect cleaning in the ion implantation system.

[0099] In general, although concurrent flow of the dopant source gas and gas-phase reactive material may be effected to achieve *in situ* cleaning, it typically is preferred to effect the cleaning operation in a sequential manner, *e.g.*, when the ion source is generating a first plasma from a first dopant source, and subsequently the ion source is generating a second plasma from a second dopant source, an intervening cleaning step is employed, in which gas-phase reactive material is flowed through the

ion source, with or without plasma generation.

[0100] In one embodiment, the invention provides a method of forming a doped silicon substrate, comprising implanting Xe^+ ions in a silicon substrate, and thereafter implanting dopant ions in the silicon substrate. In such process, implanting of the Xe^+ ions serves to amorphize the crystal structure of the substrate.

[0101] In the generation of a fluorinated xenon plasma, *e.g.*, an XeF_2 plasma, for cleaning, Xe^+ ions may do some low energy sputter cleaning of the source itself. After extraction, the Xe^+ ions may do some high energy sputtering of components downstream of the ion source, such as vacuum walls, ion optical components, wafer discs and wafer holders.

[0102] Similarly, where tungsten fluoride species such as WF_6 , WF_5 , and/or WF_4 are utilized, free fluoride may sputter clean various components of the ion source and/or tungsten may be deposited on various components of the ion source. The action that occurs, as between cleaning and deposition, is dependent on the temperature of the individual components in the system.

[0103] The present invention in various aspects relates to a method and apparatus for cleaning the ion source region of an ion implantation system used in the fabrication of a microelectronic device. The ion source region may for example include an indirectly heated cathode source, a Freeman source or a Bernas source.

[0104] The present invention in one embodiment relates to *in situ* removal of residue from the vacuum chamber of the ion implanter and components contained therein by contacting the vacuum chamber and/or components with a gas-phase reactive halide composition for sufficient time and under sufficient conditions to at least partially remove the residue from the vacuum chamber and/or components, and to do so in such a manner that when the residue and the material from which the vacuum chamber and/or components are constructed is different, the gas-phase reactive material is selectively reactive with the residue and minimally reactive (*e.g.* substantially non-reactive, and preferably completely non-reactive) with the materials from which the vacuum chamber and/or components of the ion implanter are constructed; and when the residue and the material from which the vacuum chamber

and/or components are constructed is the same, then the gas-phase reactive material may be reactive with both the residue and the vacuum chamber and/or components.

[0105] As used herein, the term “selectively,” as applied to the reactivity of the gas-phase reactive halide with a residue, is used to describe a preferential reaction between the gas-phase reactive halide and a residue. While remaining essentially non-reactive with the materials from which the vacuum chamber and/or components of the ion implanter are constructed, the gas-phase reactive halide may react with some with the materials from which the vacuum chamber and/or components of the ion implanter are constructed if the vacuum chamber and/or components comprise elements that are the same as or similar to those of the residue itself. For example, while being selectively reactive and removing tungsten deposits from a component, the gas-phase reactive material might also react with tungsten in the component itself. The residue and the components do not have to be exactly the same material for such co-reaction to occur, but will comprise some materials in common.

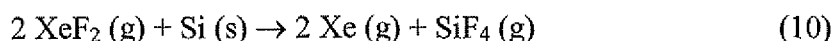
[0106] In another embodiment, the ion implanter components are cleaned *ex situ*, in a separate dedicated chamber to which the components are removed from an ion implanter.

[0107] Considering *in situ* cleaning in further detail, such cleaning is primarily dependent on three factors: the reactive nature of the cleaning precursor, the volatility of the cleaning reaction by-products, and the reaction conditions employed in the chemical cleaning. The cleaning composition must remove unwanted residue while minimizing wear of the materials of construction of the implanter. The byproducts generated by the cleaning reaction must be volatile enough to facilitate their removal by the vacuum system of the ion implanter or other pumping apparatus.

[0108] The cleaning of residue formed from the same material as the component(s) of the implanter does result in some wear of the component itself. Specifically, use of XeF_2 as a cleaning agent to remove tungsten deposits from a system utilizing a tungsten arc chamber will result in removal of some tungsten from the interior of the arc chamber. However, in the interest of maximizing system

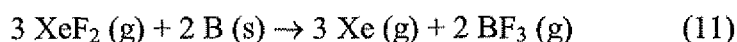
efficiency, loss of some of the interior material of the arc chamber is not significant when viewed in light of the decreased system performance if the system is not cleaned and the tungsten deposits are allowed to accumulate in the system.

[0109] The gas-phase reactive material may for example include a fluorinated xenon compound vapor, such as XeF₂ vapor. XeF₂ is a preferred reactive halide gas, and will sublime at room temperature, but may be heated using a heater to increase the rate of sublimation. XeF₂ is known to be an effective silicon etchant and has been used as a silicon selective etchant in Micro Electro Mechanical System (MEMS) device processing. Specifically, XeF₂ reacts with silicon according to the following reaction.

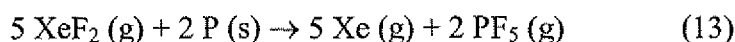
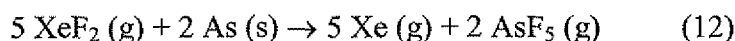


The silicon/XeF₂ reaction can occur without activation, *i.e.*, without plasma or thermal heating. The reaction rate of XeF₂ with Si is much greater than the reaction rate of XeF₂ with SiO₂, making XeF₂ selective for reaction with Si.

[0110] XeF₂ or other fluorinated xenon compound is usefully employed in the practice of the invention as an etchant for metallic boron. Although not wishing to be bound by theory, it is thought that the boron is etched according to the following reaction (11):



The use of XeF₂ as an etchant for arsenic, phosphorus and germanium is contemplated by the invention, and may involve the following reactions:



Such reactions may be carried out with or without energetic activation.

[0111] The method and apparatus of the invention are useful to at least partially remove residue from the components of the ion implanter, *e.g.*, with removal of at least 25%, more preferably at least 50% and most preferably at least 75% of such residue, and to do so in such a manner that residue is removed selectively with respect

to the materials from which the components of the ion implanter are constructed, *e.g.*, aluminum, tungsten, molybdenum, graphite, insulator materials, sealant materials, etc., when the residue material differs from those materials.

[0112] When the residue and the component material of construction are the same material, similar levels of residue removal are desired, while maintaining removal of material from the component at low levels, such as in the range of microns or tens of microns, so as not to significantly affect the performance of the component. Further, since deposits generally do not have a uniform thickness or deposition, they may be more reactive than the material of the component itself in the cleaning process, so that the gas-phase reactive material composition is more selective to reaction with the residue than to reaction with the component part.

[0113] Several modes of delivery of the gas-phase reactive material composition to the ion source region for *in situ* cleaning may be employed, including a stagnant mode, a continuous mode, and a direct introduction mode. Such modes of cleaning are more fully described in International Publication WO07/127865, together with apparatus and methodology that are usefully employed in the practice of the present invention. The disclosure of International Publication WO07/127865 is hereby incorporated herein by reference, in its entirety. Although use of XeF₂ as a cleaning composition is described herein in connection with various embodiments of the invention, it will be understood that other fluorinated compounds, such as WF₆, WF₅, and/or WF₄, may be used, alternatively to or in combination with XeF₂, or other and additional fluorinated compounds can be employed. For example, BrF₃ can be used to etch tungsten without a plasma. In another aspect, the invention relates to a method of improving the performance and extending the lifetime of an ion implant system that employs solid doping material which comprises using XeF₂ or N₂F₄ as a carrier gas for said solid doping material. The solid doping material includes, but is not limited to, Sb₂O₃ and InCl₃. As contemplated by the present invention, the use of XeF₂ or N₂F₄ as a carrier gas for Sb₂O₃, InCl₃ or other solid doping material, removes Sb, In and other dopants deposited on the source chamber and components thereof. The instant method has utility even if a switch over to boron is made after Sb implanting.

Advantages gained by the method of the invention are at least two-fold: first, it provides real time source cleaning to prevent or reduce dopant accumulation on the ion source chamber and components thereof, thus improving ion source performance with prolongation of ion source life; second, it enhances and/or stabilizes the plasma and/or beam current.

[0114] In another aspect, the invention relates to a method of improving the performance and extending the lifetime of an ion implant system that employs gaseous doping material which comprises using XeF_2 or N_2F_4 as a co-flow gas with said gaseous doping material. The gaseous doping material includes, but is not limited to, GeH_4 . As contemplated by the present invention, the use of XeF_2 or N_2F_4 as a co-flow gas with GeH_4 or other gaseous doping material, removes Ge or other dopants deposited on the source chamber and components thereof. Advantages obtained by this practice of the invention are at least two-fold: first, it provides real time source cleaning to prevent or reduce dopant accumulation on the ion source chamber and components thereof, thereby improving ion source performance and prolonging ion source life; second, it enhances and/or stabilizes the plasma and/or beam current.

[0115] In another aspect, the invention relates to a method of cleaning a foreline of an ion implant system to remove an ionization-related deposit therefrom comprising contacting the foreline of an ion implantation system with a cleaning gas under conditions in which said cleaning gas is chemically reactive with the deposit to effect at least partial removal thereof. Deposits include, but are not limited to those comprising B, Ge, Si, P, and As, or mixtures of same. The cleaning gas includes, but is not limited to XeF_2 , N_2F_4 , F_2 , and other fluorinated species which are reactive with a deposit as constituted aforesaid. As appreciated by those in the art the amount of cleaning gas needed is dependent on the amount of deposit present. Similarly, the amount of heat given off during the reaction of the cleaning gas and the deposit is dependent on the flow rate of the cleaning gas. The identity and concentration of byproduct species that result from the cleaning process depends on cleaning gas flow rate, the compositional make-up of the deposits, and the pump purge flow rate. For

non-limiting illustrative purposes only, an example of using XeF₂ to clean phosphorus from a foreline is described below:

[0116] The chemical reaction used to determine the quantity of XeF₂ required in the cleaning process is: $5 \text{ XeF}_2 (\text{g}) + 2 \text{ P} (\text{s}) \rightarrow 5 \text{ Xe} (\text{g}) + 2 \text{ PF}_5 (\text{g})$. The enthalpies of formation (in kJ/mol) are taken from Lange's Handbook of Chemistry (14th ed) and are listed here for the determination of the heat liberated during the reaction: XeF₂ (-164); Xe (0); P (0); and PF₅ (-1594.4). The flow rate of XeF₂ determines the length of time required for the cleaning process as well as the heat liberated. Without providing a means of heating the XeF₂ cylinder, the maximum sustained flow rate is approximately 50 sccm, assuming adequate delivery tubing conductance. This flow rate can be increased to 100 sccm or greater if the cylinder is maintained at room temperature by using a heating jacket. The amount of XeF₂ needed to clean the phosphorus deposit is shown in Table 1, and the amount of heat given off during the cleaning reaction is shown in Table 2.

Table 1.

Mass of Phosphorus Deposits (g)	Mass of XeF ₂ Required (g)
10	137
100	1367
1000	13,666

Table 2.

Flow Rate of XeF ₂ (sccm)	Rate of Heat Generation (watts)
50	17.6
100	35
200	70

[0117] The maximum production rate of various byproducts from the above-described cleaning reaction is shown in Table 3.

Flow Rate of XeF ₂ (sccm)	AsF ₃	PF ₃	BF ₃	GeF ₄	SiF ₄
50	0.20 g/min (33.3 sccm)	0.13 g/min (33.3 sccm)	0.10 g/min (33.3 sccm)	0.17 g/min (25 sccm)	0.12 g/min (25 sccm)

[0118] As appreciated by those in the art, because the composition of the residues may vary, the data shown in Table 3 is based upon the assumption that the byproduct quantities are determined for each element assuming 100% compositional make-up of that element. Furthermore, the maximum concentrations of these species are dependent upon the dilution flow rates in the exhaust systems. For example, just downstream of the roughing pump, the maximum steady state concentration of PF₃ is 3330 ppm if the pump has a nitrogen purge of 10 slpm. This value can increase if the flow rate of XeF₂ is higher than 50 sccm.

[0119] In one embodiment of the above-described method, the cleaning gas flows into the implant source chamber, with the turbo pump off, but the roughing pump on. This practice improves the flow rate of the cleaning gas over foreline deposits thus providing a faster cleaning process. Cleaning gas flow rate can be further improved by heating the gas cylinder wherein the cleaning gas is stored to up to room temperature or above. Preferably, in this practice, the delivery lines from the gas cylinder to the ion implanter are similarly heated.

[0120] In another embodiment of the above-described method, the cleaning gas flows into the implant source chamber in a pulsed flow in which the implant source chamber, the pump and the forelines are filled to a certain pressure and then pumped to lower pressure. This process is repeated until the deposit on the foreline of the ion implant system is removed. This practice preferably employs an isolation valve on the inlet of the roughing pump.

[0121] In a preferred practice, the above-described embodiment further comprises heating the gas cylinder wherein the cleaning gas is stored to room temperature or above.

[0122] For all embodiments, the method preferably further comprises a scrubber at the roughing pump outlet to remove the volatile byproducts produced from the cleaning process.

[0123] Each embodiment preferably further comprises a Xe recovery system as commercially available from Air Products and Chemicals, Inc. (PA, USA), and is described in

http://www.fabtech.org/product_briefings/_a/new_product_air_products_offers_on_site_xenon_recovery., incorporated herein.

[0124] Another embodiment of the method of cleaning a foreline of an ion implant system comprises providing the cleaning gas downstream of a turbo pump and flowing said cleaning gas continuously to the foreline of the ion implant system. This practice preferably cleans the deposits on the foreline even when the implant processes are running, thereby reducing the disruption of the operation of ion implanting.

[0125] In the above-described embodiment, the cleaning gas is preferably stored in a gas cylinder; the method preferably further comprises heating the gas cylinder wherein the cleaning gas is stored at room temperature or above.

[0126] The above-described embodiment preferably further comprises providing a scrubber at the roughing pump outlet to remove the volatile byproducts produced from the cleaning process.

[0127] The above-described embodiment further comprises providing a Xe recovery system as commercially available from Air Products and Chemicals, Inc. (PA, USA), and is described in http://www.fabtech.org/product_briefings/_a/new_product_air_products_offers_on_site_xenon_recovery., incorporated herein.

[0128] In another aspect, the invention relates to a method of improving the performance and extending the lifetime of an ion implant system having a cathode comprising contacting the cathode with a gas mixture comprised of at least one cleaning gas and at least one deposition gas wherein said gas mixture balances the deposition of material on the cathode and the stripping off of the same or other material from the cathode. The cleaning gas of the gas mixture removes dopant deposited on the cathode and the material of the cathode, whereas the deposition gas of the gas mixture directly or indirectly causes depositing dopant onto the cathode. Such a gas mixture maintains a balance between the accumulation of dopant material and stripping off of the same or other material on the cathode, and therefore prolongs the ion source life. Storage and dispensing of the gas mixture into ion source implanter may be accomplished by using an adsorption-desorption apparatus (referred to as SDS - safe delivery source) described in US Patent No. 5,518,528, and the contents of which are incorporated herein by reference; a fluid storage and dispensing system comprising a vessel for holding a fluid at a desired pressure (referred to as VAC Vacuum Actuated Cylinder) described in US Patent No. 6,101,816, and the contents of which are incorporated herein by reference; or a hybrid fluid storage and dispensing system of SDS and VAC (referred to as VAC-Sorb) described in US Patent No. 6,089,027, and the contents of which are incorporated herein by reference. These fluid storage and dispensing systems provide sub-atmospheric pressure delivery of the gases, and thereby are safer and more efficient than the high pressure fluid storage and dispensing systems. Moreover, some of the gases of the gas mixture, which are not compatible to co-exist under high pressure fluid storage and dispensing system, can be stored and dispensed together under the SDS, VAC or VAC-Sorb systems.

[0129] In one embodiment of the above-described method, gases of the gas mixture flow simultaneously to contact the cathode.

[0130] In another embodiment of the above-described method, gases of the gas mixture flow sequentially to contact the cathode.

[0131] In another embodiment of the above-described method, the gas mixture comprising a combination of at least one hydrogen-containing gas and at least one

fluorine-containing gas wherein the hydrogen-containing gas functions as the cleaning gas, and the fluorine-containing gas functions as the deposition gas.

[0132] In another embodiment of the above-described method, the gas mixture comprises a combination of at least one non-dopant gas, i.e., gas does not contain As, P, Ge, B, Si, or C, and at least one dopant gas wherein the non-dopant gas functions as the cleaning gas, and the dopant gas functions as the deposition gas.

[0133] Examples of cleaning gas are, but not limited to, Xe/H₂, Ar/H₂, Ne/H₂, Xe/NH₃, Ar/NH₃, Ne/NH₃, Ar/Xe, and Ar/Xe/H₂.

[0134] Examples of deposition gas are, but not limited to, F₂, N₂F₄, ClF₃, WF₆, MoF₆, and NF₃.

[0135] Examples of the gas mixture are, but not limited to, AsH₃/AsF₃, AsH₃/AsF₅, PH₃/PF₃, PH₃/PF₅, SiH₄/SiF₄, H₂/Xe/SiF₄, GeH₄/GeF₄, H₂/Xe/GeF₄, H₂/GeF₄, B₂H₆/BF₃, H₂/BF₃, F₂/BF₃, CO₂/F₂, CO₂/CF₄, CO/F₂, CO/CF₄, COF₂/F₂, COF₂/CH₄, COF₂/H₂.

[0136] The features and advantages of the invention are more fully shown by the following non-limiting examples.

EXAMPLE 1

[0137] This example shows the improvement in ion source lifetime and implanter utilization that is achievable by use of a chemical cleaning agent to remove deposits. Preferably, the deposits are removed at regular intervals to prevent buildup of contaminant flakes and conductive films in the implanter.

[0138] *In-situ* cleaning was carried out by introducing XeF₂ at regular intervals from a supply vessel thereof positioned in the gas box of the ion implanter, with XeF₂ cleaning vapor introduced into the ion source for 10-15 minutes twice a day. A high current implanter was utilized for testing to assess flow dynamics of the cleaning reagent. XeF₂ cleaning properties were determined and it was verified that the cleaning agent had no adverse effect on the implanter's beamline components. The

cleaning process using the XeF_2 reagent then was qualified for use with a medium current implanter apparatus.

[0139] Fig. 1 is a graph of compiled source lifetime data from such medium current implanter, prior and subsequent to implementation of the *in-situ* cleaning process. The data were developed for a dopant composition including arsine and phosphine. Before cleaning, the ion source had an average operating life of approximately 250 ± 90 hours, limited by two common failure modes.

[0140] The predominant mode of failure was excessive leakage from a suppressor voltage supply. For successful extraction of a stable ion beam, suppressor voltage is applied to an electrode positioned outside of the arc chamber. The electrode is electrically isolated by a number of small insulators, and buildup of a conductive film on one or more of these insulators can cause excessive suppressor leakage.

[0141] A secondary mode of failure was shorting out of components in the arc chamber attributable to flakes of deposited material.

[0142] These failure modes were found to be minimized by the *in-situ* chemical cleaning process. Regular twice-daily cleaning increased source lifetime on the production.

[0143] The effect of XeF_2 on the suppressor leakage current is further shown in Fig. 2, which is a graph of the leakage current for the medium current tool before and after introduction of *in-situ* cleaning. Each data point represents an average suppressor current during the time required to implant a wafer lot, and points have been plotted over the lifetime of several ion sources. The magnitude of the leakage depends on the elapsed time since replacement of insulators at the last preventive maintenance. The data show that regular *in-situ* cleaning greatly reduced leakage current so that it never reached the upper control limit of 1.5 mA at which an unscheduled source maintenance is required.

[0144] *In-situ* cleaning effects were also evaluated, using an implant dopant mix including BF_3 and PH_3 . The source ran for 497 hours under these conditions and failed for an arc limited condition involving tungsten deposits on the filament, attributable to

the BF_3 chemistry. The single source life of 497 hours on the test system compares favorably with the long term historical average of 299 hours on the same system. This is a single data point, but it fits the established pattern. The source life improvement in this case appears to be due to etching by XeF_2 of tungsten deposits in the source arc chamber.

[0145] The Figs. 3A and 3B photographs provide additional evidence of the effect of the cleaning agent. In both photographs the appearance of the ion source housing after removal of the ion source assembly for periodic preventive maintenance is shown, after about 98 days of production in each instance. For the photograph in Fig. 3A, *in situ* cleaning was conducted twice daily, while for the photograph in Fig 3B no cleaning had been performed.

[0146] In the absence of cleaning, substantial quantities of deposited material were present, some of which had begun to delaminate and flake. During regular maintenance activity, manual scrubbing was used to remove deposited material from internal surfaces of the housing. The housing appeared much cleaner with *in situ* cleaning, with little or no time being spent on manual cleaning. Deposits were removed by unreacted XeF_2 flowing out of the arc chamber, and passing to the vacuum chamber walls, dopant and other deposits were removed by chemical reaction.

[0147] Deposits in and around the ion source give rise to the so-called "implanter memory effect." When changing from one dopant source gas to another, ions from elements of the first dopant continue to be extracted from the ion source plasma long after flow of the first dopant gas has been terminated. This effect in some instances causes severe contamination of the desired ion beam and resulting degradation of the implant process.

[0148] One example of the implanter memory effect is P contamination in a BF_2 implant. The consequences of this contamination to yield of the process are so severe that many semiconductor manufacturing facilities avoid scheduling phosphorus and boron implants on the same tool. This is a substantial impediment in scheduling

implant operations. P / BF₂ contamination results from phosphorus deposits in the source from implantation with PH₃. Upon change-over to BF₃ gas for BF₂⁺ implantation, some of the fluorine reacts to form ³¹P¹⁹F⁺. The mass of ³¹P¹⁹F⁺ is 50. This is sufficiently close enough to the desired mass for ¹¹B¹⁹F₂ of 49, so that PF⁺ is co-implanted with the BF₂⁺ ions. As a result, BF₂⁺ implantation is restricted on some high current systems with marginal mass resolution capability at specific mass-energy ranges.

[0149] XeF₂ cleaning was assessed to determine its affect on the implanter memory effect, using a high current implanter operated for approximately 200 hours in simulated production with a P⁺ ion beam from PH₃ dopant gas. The system was switched to BF₃ gas and immediate implantation of a bare silicon monitor wafer was carried out using a high dose (5 x 10¹⁵ ions/cm²) of BF₂⁺. During BF₂⁺ implantation, the resolving aperture of the system's analyzing magnet was opened more than usual to ensure that the contamination effect would be sufficiently large for convenient measurement using secondary ion mass spectrometry (SIMS) analysis.

[0150] Cleaning effects of BF₃, argon and XeF₂ were compared by running each of the 3 gases and then periodically monitoring the amount of remaining contamination by implanting monitor wafers with BF₂⁺. The amount of P co-implanted with the BF₂ was measured by SIMS. A typical SIMS spectrum of implanted phosphorus is shown in Fig. 4A, in which the peak in the phosphorus spectrum corresponds to the implanted depth of PF⁺ ions extracted from the ion source, and the dose corresponds to a contamination level of about 3% PF in BF₂.

[0151] Fig. 4B is a plot of the contamination level as a function of cleaning time with either BF₃ or XeF₂, in which the plot is normalized to the contamination level immediately after change-over from PH₃ to BF₃. There was little effect on the PF contamination even after 2 hours when running BF₃ plasma. Similar results (not shown) obtained when argon plasma was used. By contrast, PF contamination was reduced by a factor of two after only 15 minutes of *in-situ* cleaning with XeF₂, and by a factor of almost five after 30 minutes of *in-situ* cleaning with XeF₂.

[0152] Prior to use of *in-situ* cleaning, the medium current implanter unit averaged 3.3 source changes per tool per month, with average source change-over and subsequent qualification testing requiring approximately 5 hours, equivalent to nearly 200 hours annually of lost production time per tool. The source life was effectively doubled by *in-situ* cleaning, yielding about 100 hours of additional production time for each medium current tool. The resulting savings of test wafers, as well as savings in fabrication time and metrology tools needed for post processing of qualification wafers (with up to 40 qualifications annually being carried out for each medium current implanter), demonstrated the effectiveness of the *in-situ* cleaning.

EXAMPLE 2

[0153] This example demonstrates the control of filament growth, in an ion source of an illustrative ion implanter system.

[0154] Figure 5A is a graph showing the effects of XeF₂ flow and arc power variation, with respect to increased filament current and weight. The graph shows filament weight in grams, as a function of elapsed time, in hours, of operation of the implanter system. The upper line in the graph represented operation at an XeF₂ flow of 2.2 standard cubic centimeters per minute (sccm) and arc power of 100 volts/0.05 amps, for which a filament weight gain of 319 milligrams/hour was determined after 3 hours of operation. The lower line in the graph reflected an XeF₂ flow of 0.5 sccm and arc power of 40 volts/0.05 amps, which produced a filament weight gain of 63 milligrams/hour for the 3 hour duration of operation.

[0155] Figure 5B is a graph showing the effects of XeF₂ flow and arc power variation, with respect to filament current. The graph shows filament current in amperes, as a function of time of operation of the implanter system. The upper line in the graph represented operation at an XeF₂ flow of 2.2 standard cubic centimeters per minute (sccm) and arc power of 100 volts/0.05 amps, for which a filament current rise of 16 amps/hour was determined. The lower line in the graph reflected an XeF₂ flow of 0.5 sccm and arc power of 40 volts/0.05 amps, which produced a filament current rise of 2.3 amps/hour for the 3 hour duration of operation.

[0156] Figure 6 is a graph of filament weight change, in milligrams per hour, as a function of average filament current, in amperes. The graph shows the effect of hot flow (no plasma) and plasma conditions for tungsten transport, with data for low flow and high flow hot filament conditions, and for low flow and high flow plasma conditions. These data show that the transport of tungsten in the system can be selectively adjusted by selection of appropriate process conditions to effect deposition, or alternatively etching, of material at the filament.

[0157] While the invention has been described herein with reference to various specific embodiments, it will be appreciated that the invention is not thus limited, and extends to and encompasses various other modifications and embodiments, as will be appreciated by those ordinarily skilled in the art. Accordingly, the invention is intended to be broadly construed and interpreted, in accordance with the ensuing claims.

THE CLAIMS

What is claimed is:

1. A method of improving the performance and extending the lifetime of an ion implant system including an ion source having a cathode, the method comprising:
 - (a) providing dopant gas to an ion source chamber during ion implantation processing;
 - (b) monitoring one or more conditions to effect deposition onto the cathode and etching from the cathode;
 - (c) responding to the monitoring, contacting the cathode with a gas mixture comprised of at least one cleaning gas and at least one fluorinated deposition gas and adjusting temperature of the ion source to temperature sufficient to cause deposition or etching, wherein the gas mixture performs deposition of materials onto the cathode and stripping off of the same or other material from the cathode based on the temperature,
wherein the cleaning gas comprising a gas-phase reactive material including one or more gases or gas mixtures selected from AsH_3 , PH_3 , SiH_4 , H_2/Xe , GeH_4 , H_2 , B_2H_6 , CO , CH_4 , Xe/NH_3 , Ar/NH_3 , Ne/NH_3 , and Ar/Xe , the cleaning gas controlling the temperature in the ion source to be effective in the etching of the deposits onto the cathode, and
wherein, when the fluorinated deposition gas comprises at least one gas selected from F_2 , N_2F_4 , AsF_3 , PF_3 , PF_5 , SiF_4 , GeF_4 , BF_3 , CF_4 , COF_2 , ClF_3 , WF_6 , MoF_6 , and NF_3 and the dopant gas comprises a fluorinated dopant gas, the fluorinated deposition gas is effective in regrowing the cathode by controlling the temperature in the ion source on a condition that the fluorinated dopant gas and the fluorinated deposition gas may be the same.
2. The method of improving the performance and extending the lifetime of the ion implant system according to claim 1, wherein gases of the gas mixture flow simultaneously to contact the cathode.
3. The method of improving the performance and extending the lifetime of the ion implant system according to claim 1, wherein the gas mixture comprises a combination of at least one non-dopant gas and at least one dopant gas.
4. The method of improving the performance and extending the lifetime of the ion implant system according to claim 1 or 2, wherein the gas mixture is stored and dispensed by

a gas storage and dispensing system selected from an adsorption-desorption gas storage and dispensing system, pressure adjustment gas storage and dispensing system, and a hybrid adsorption-desorption pressure adjustment gas storage and dispensing system.

5. The method of improving the performance and extending the lifetime of the ion implant system according to claim 1, wherein gases of the gas mixture flow simultaneously to contact the cathode.

6. The method of improving the performance and extending the lifetime of the ion implant system according to claim 1, wherein the gas mixture is stored and dispensed by a gas storage and dispensing system selected from an adsorption-desorption gas storage and dispensing system, pressure adjustment gas storage and dispensing system, and a hybrid adsorption-desorption pressure adjustment gas storage and dispensing system.