

POLYMER COMPOSITIONS AND ADHESIVES, COATINGS, AND SEALANTS MADE THEREFROM

FIELD OF THE INVENTION

[0001] The present invention relates to new polymer compositions for preparing high performance adhesives, coatings, extrusions, and sealants.

BACKGROUND

[0002] High performance adhesives, coatings, extrusions, and sealants typically have been based on organic polymer compositions or compositions comprising a silicone polymer. Products based on organic polymers tend to have lower weather stability than products based on silicone polymers. Products based on silicone polymers typically have a higher affinity for dirt pickup and staining than products based on organic polymers. The latter also tend to have relatively poor adhesion to certain substrates and are difficult to paint. Organic polymers and silicone polymers are immiscible and, as a result, blends combining these two types of polymers phase separate. There is a need for new polymer compositions for producing adhesives, coatings and sealants.

SUMMARY OF THE INVENTION

[0003] Provided herein are polymer compositions for preparing sealants, coatings, adhesives, and extrusions with improved weather stability and/or performance. The polymer compositions of the present invention comprise from 1% to 99% by weight (of polymer) of an organic polymer (referred to hereinafter as a "compatibilizer") that comprises silane groups, from 1% to 45% by weight (of polymer) of a reactive or non-reactive organopolysiloxane polymer, and from 0% to 98% by weight (of polymer) of an organic polymer that does not comprise silane groups. In certain embodiments, the compatibilizer is an organic polymer that is partially or fully end-capped with silane groups. The present polymer compositions are based, at least in part, on the inventors' discovery that sealants based on polymer compositions that comprise both a silicone polymer and an organic polymer that comprises silane groups do not phase separate as readily as compositions lacking the present compatibilizer.

[0004] Upon cure, the present polymer compositions can be used as coatings, sealants, or adhesives. The present polymer compositions may further comprise other additives to modify the properties of such coatings, sealants, and adhesives.

[0005] The present sealant, adhesive, and coating compositions may be 1-part curable compositions or 2-part curable compositions. The present sealant, adhesive, and coating

compositions may be either thermally curing systems or room temperature curing systems. The present sealant compositions may be extrusion or *in situ* cured systems.

[0006] Additional objects and advantages of the invention will be set forth in part in the description which follows, and in part will be obvious from the description, or may be learned by practice of the invention. The objects and advantages of the invention will be realized and attained by means of the elements and combinations particularly pointed out in the appended claims.

[0007] It is to be understood that both the foregoing general description and the following detailed description are exemplary and explanatory only and are not restrictive of the invention, as claimed.

DETAILED DESCRIPTION OF THE INVENTION

[0008] The present invention will now be described by reference to more detailed embodiments. This invention may, however, be embodied in different forms and should not be construed as limited to the embodiments set forth herein. Rather, these embodiments are provided so that this disclosure will be thorough and complete, and will fully convey the scope of the invention to those skilled in the art.

[0009] Unless otherwise defined, all technical and scientific terms used herein have the same meaning as commonly understood by one of ordinary skill in the art to which this invention belongs. The terminology used in the description of the invention herein is for describing particular embodiments only and is not intended to be limiting of the invention. As used in the description of the invention and the appended claims, the singular forms “a,” “an,” and “the” are intended to include the plural forms as well, unless the context clearly indicates otherwise. All publications, patent applications, patents, and other references mentioned herein are expressly incorporated by reference in their entirety.

[0010] Unless otherwise indicated, all numbers expressing quantities of ingredients, reaction conditions, and so forth used in the specification and claims are to be understood as being modified in all instances by the term “about.” Accordingly, unless indicated to the contrary, the numerical parameters set forth in the following specification and attached claims are approximations that may vary depending upon the desired properties sought to be obtained by the present invention. At the very least, and not as an attempt to limit the application of the doctrine of equivalents to the scope of the claims, each numerical parameter should be construed in light of the number of significant digits and ordinary rounding approaches.

[0011] Notwithstanding that the numerical ranges and parameters setting forth the broad scope of the invention are approximations, the numerical values set forth in the specific examples are reported as precisely as possible. Any numerical value, however, inherently contains certain errors necessarily resulting from the standard deviation found in their respective testing measurements. Every numerical range given throughout this specification will include every narrower numerical range that falls within such broader numerical range, as if such narrower numerical ranges were all expressly written herein.

POLYMER COMPOSITIONS

[0012] The polymer compositions of the present invention are comprised of 1% to 99% by weight of at least one organic polymer that comprises silane groups, i.e., a compatibilizer, and from 1% to 45% by weight of at least one reactive or non-reactive organopolysiloxane polymer, preferably a reactive organopolysiloxane. In certain embodiments, the silane groups are end-cappers, i.e., the compatibilizer is partially or fully end-capped with silane groups. In other embodiments, the silane groups are incorporated into the backbone of the compatibilizer or are pendant groups, i.e., are attached to the backbone of the compatibilizer. The compatibilizers used in the present compositions have a molecular weight greater than 1000 g/mole, preferably greater than 10,000 g/mole. The present polymer compositions also comprise from 0% to 98% by weight of an organic polymer that does not comprise silane groups.

[0013] Both the compatibilizer and the organic polymer that does not comprise silane groups can be a homopolymer, a copolymer, a multipolymer, or any mixture thereof. The compatibilizer and the organic polymer that lacks silane groups can be chosen from aromatic and aliphatic urethane polymers, polyethers, polyesters, acrylic polymers, polystyrene, styrene butadiene polymers, polybutadiene, and butyl rubber, and any mixture or blend thereof. The organic polymers used in the present compositions may have different backbones such as polyester, polyether, polyacrylate, polybutadiene, polycarbonate or any mixture of such backbones. Furthermore, the organic polymers used in the present compositions could also have other functional groups, such as acrylates, amides, maleic anhydride, and so forth, attached thereto.

[0014] In certain embodiments of the present polymer composition, at least one of the compatibilizers is an aromatic or aliphatic polyurethane polymer that is partially or fully end-capped with silane groups. In certain embodiments, at least one of the compatibilizers is a polyurethane polymer that is end capped with silane groups and other end-cappers. Examples of other suitable end-cappers include but are not limited to alcohols, and monomers or

polymers containing epoxy, hydroxy, and mercapto functional groups, or any combination thereof. The silane end-capped polyurethane polymer may be derived by the steps of a) reacting a diisocyanate compound with a polyol to form an intermediate, wherein the intermediate is selected from isocyanates or hydroxyl terminated polyurethane prepolymers; and b) silylating the intermediate. Preferably, the intermediate is silylated with an organo functional silane having one or more hydrolyzable groups.

[0015] Suitable hydroxy-terminated polymeric materials for preparing the polyurethane polymers include, but are not limited to di, tri, and tetra functional polyols, including polyether polyols, polyester polyols, acrylic polyols, and polyols comprising two or more hydroxyl groups and a straight or branched chain hydrocarbon.

[0016] Suitable polyether diols and triols include polyethylene ether diols or triols, polypropylene ether diols or triols, polybutylene ether diols or triols, polytetramethylene ether diols or triols, and block copolymers of such diols and triols.

[0017] Suitable hydroxy-terminated polyesters include any hydroxy-terminated polyester prepared from poly-basic acids or anhydrides (for example, adipic acid and phthalic anhydride) and polyols in which the hydroxyl functionality of the polyester prepolymer is greater than 2, preferably over 2.3. Polylactone containing hydroxyl groups are also suitable for making the prepolymer, particularly polycaprolactone diol and triol.

[0018] Suitable acrylic polyols include hydroxyl-terminated polyacrylate. Acrylates include, but are not limited to, butylacrylate, methylacrylate, methylmethacrylate, ethyl acrylate, 2-ethylhexyl acrylate or the mixture of above. Suitable polyols comprising two or more hydroxyl groups and a straight or branched hydrocarbon chain include hydroxyl functionalized polybutadiene. Other suitable polyols include polycarbonates having hydroxyl groups.

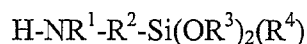
[0019] Preferably, the polyol has a number average molecular weight of from 500 to 18,000. For polymers that are used to make sealant compositions, it is preferred that the polyol have a number average molecular weight of from 2,000 to 8,000. For polymers that are used to make coating or adhesive compositions, it is preferred that the polyol have a number average molecular weight of from 500 to 4000.

[0020] The isocyanates that are reacted with the hydroxy-terminated backbone polymer are organic isocyanates having 1 or more isocyanate groups or a mixture of such organic isocyanates. The isocyanates are aromatic or, preferably, aliphatic isocyanates. Examples of suitable aromatic di- or triisocyanates include p,p',p''-triisocyanato triphenyl methane, p,p'-diisocyanato diphenyl methane, naphthalene-1,5-diisocyanate, 2,4-tolylene

diisocyanate, 2,6-tolylene diisocyanate, and mixtures thereof. Examples of preferred aliphatic isocyanates are isophorone diisocyanate, dicyclohexyl methane- 4,4'- diisocyanate, and mixtures thereof.

[0021] The polyurethane polymer may be prepared by mixing the hydroxy-terminated polymer and organic isocyanate together at ambient temperature and pressure, although the speed of the reaction is significantly increased if the temperature of the reaction mixture is raised to a higher temperature, for example, a temperature between 60°-100°C. A stoichiometric excess of the isocyanate or polyol is used to ensure that the polyurethane polymer has NCO or OH terminal groups.

[0022] In certain embodiments, the polymer having terminal NCO groups is then reacted with silane capping agents so that 50 to 100% of the NCO groups are blocked or end-capped with silane groups. Examples of suitable silane capping agents include, but are not limited to, silanes corresponding to the formula I.



wherein R^1 represents hydrogen, a substituted aliphatic, cycloaliphatic, and/or aromatic hydrocarbon radical containing 1 to 10 carbon atoms, a second $\text{-R}^2\text{-Si(OR}^3\text{)}_2\text{(R}^4\text{)}$, or $\text{-CHR}^5\text{-CHR}^6\text{COOR}^7$ where R^5 and R^6 are H or C_{1-6} organic moiety, and R^7 is C_{1-10} organic moiety.

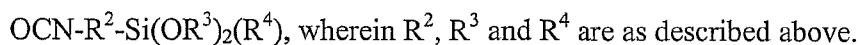
R^2 represents a linear or branched alkylene radical containing 1 to 8 carbon atoms.

R^3 represents a C_{1-6} alkyl group.

$\text{R}^4 = \text{-CH}_3, \text{-CH}_2\text{CH}_3, \text{ or OR}^3$.

[0023] Examples of suitable aminosilanes corresponding to formula I include N-phenylaminopropyltrimethoxysilane, N-ethylaminoisobutyltrimethoxysilane, gamma-aminopropyltrimethoxysilane, gamma-aminopropyltriethoxysilane, and the reaction product of an aminosilane (such as gamma-aminopropyltrimethoxysilane, gamma-aminopropyltriethoxysilane, gamma-aminopropylmethyldimethoxysilane) with an acrylic monomer (such as ethyl acrylate, butyl acrylate, 2-ethylhexyl acrylate, methyl acrylate, methyl methacrylate, and glycidal acrylate).

[0024] Examples of other suitable silanes include mercaptosilane, the reaction product of a mercaptosilane with a monoepoxide, and the reaction product of an epoxysilane with a secondary amine and silanes with the following structure:



[0025] Methods of preparing polyurethane polymers, and silylated polyurethane polymers are well known in the art. See, e.g., U.S. Pat. Nos. 3,627,722, 3,632,557, 3,979,344, and 4,222,925, which are incorporated herein by reference.

[0026] In certain embodiments, the resulting partially silylated, polyurethane polymer is then combined with a sufficient amount of a non-hindered aromatic or aliphatic alcohol to block or end-cap any unreacted, i.e., non-silylated, NCO groups. Examples of suitable aliphatic alcohols include, but are not limited to, methanol, ethanol, iso-propanol, butanol. Preferably, the aromatic alcohol has a molecular weight less than 2000. Examples of suitable aromatic alcohols include phenol, 3-methoxyphenol, 4-methoxyphenol, nonylphenol, meta-cresol, para-cresol, 4-chlorophenol, meta-hydroxybenzaldehyde, ortho-hydroxybenzaldehyde, para-hydroxybenzaldehyde, hydroquinone, 1-hydroxy-2-propanone, 3-hydroxyacetophenone, and 4-hydroxyacetophenone. The polyurethane polymer that is partially end-capped with silane groups is reacted with the aromatic alcohol at 80°C for 90 minutes to provide a composition comprising a silane and aromatic alcohol end capped, polyurethane prepolymer and an excess of aromatic alcohol.

[0027] Optionally, the polymer composition also comprises a moisture scavenger which may be added after the reaction. Examples of moisture scavengers for inclusion in the prepolymer composition are vinyltrimethoxysilane, methyltrimethoxysilane, hexamethyldisilazane, paratoluene sulfonyl isocyanate (PTSI), toluene diisocyanate (TDI), diphenyl methane diisocyanate (MDI), and polymeric MDI. The moisture content of the polymer composition preferably is 0.1 to 10%, more preferably 0.2 to 3%.

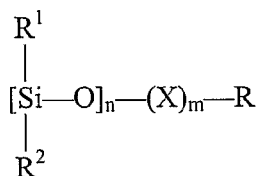
[0028] In certain embodiments of the present polymer composition, at least one of the compatibilizers is a silylated allylic terminated linear or branched polyether. The polyether backbone may have other functional groups, such as acrylates, amides, maleic anhydride, and so forth. The polymers' terminal ends could be partially or fully end-capped either with silane groups or with a combination of silane groups and other end-capping groups. The silylated allylic terminated polyether is derived from the reaction of a vinyl alkyl terminated polyol with a hydride functional silane. Preferably, the hydride functional silane is selected from the group consisting of triethoxysilane, trimethoxysilane, methyldiethoxysilane, methyldimethylsilane and combinations thereof.

[0029] Another suitable compatibilizer is a reactive silylated polyol. This polyol may be selected from polyester, polyether, polyacrylate, polybutadiene, polycarbonate, and so forth. The silylation may be done as discussed above.

[0030] The compatibilizer may also be a polyether having silicone functional groups. Preferably, when the organic polymer is a polyether having a silicone functional group, the silicone functional group is a hydrolyzable silane group. Preferable polyethers having a silicone functional group are those having a molecular weight in the range from about 2,000 to about 50,000 g/mole.

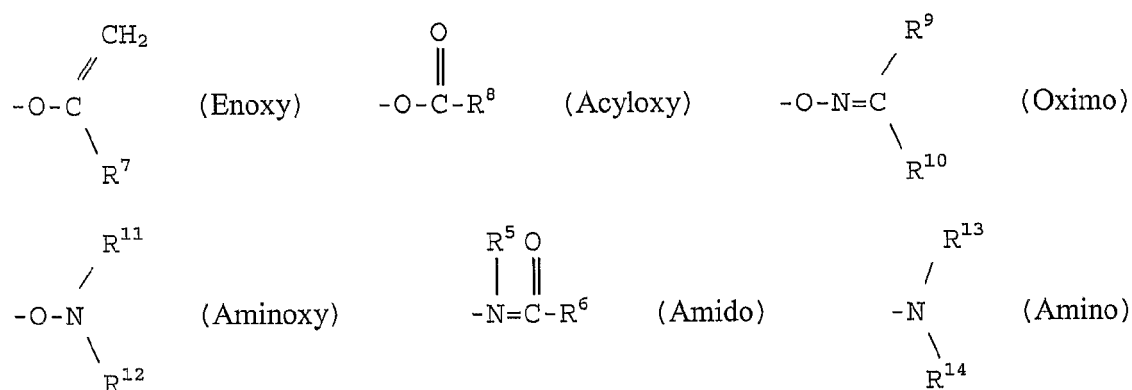
[0031] The organopolysiloxane used in the present composition may be a non-reactive organopolysiloxane, i.e., a polysiloxane that contains no reactive functional groups. In other embodiments, the organopolysiloxane is a reactive organopolysiloxane that contains reactive functional groups, preferably two reactive functional groups on the polymer chain, preferably at the terminal portion thereof, i.e., preferably the reactive functional groups are end-groups. Organopolysiloxanes useful in this invention include, but are not limited to, those which contain a condensable functional group which can be an hydroxyl group, or hydrolyzable group such as a silicon-bonded alkoxy group, acyloxy group, ketoximo group, amino group, amido group, aminoxy group, an alkenoxy group, and so forth. The reactive functional groups may be hydroxyl, alkoxy, silicone alkoxy, acyloxy, ketoximo, amino, amido, aminoxy, alkenoxy, alkenyl, or enoxy groups or any combination thereof. The reactive functional groups are end groups, pendant groups, or a combination thereof. In certain embodiments, the organopolysiloxanes used in the present invention preferably have a molecular weight in the range from 20,000 to 100,000 grams/mole.

[0032] In one embodiment, the reactive organopolysiloxane polymer is of the formula:



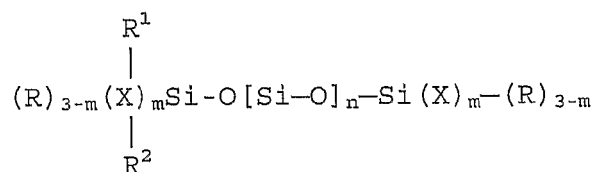
where R^1 and R^2 , independently, are an alkyl having from 1 to 8 carbon atoms, desirably from 1 to 4 carbon atoms with methyl being preferred, or is an aromatic group or substituted aromatic group having from 6 to 10 carbon atoms with phenyl being preferred, and "n" is such that the weight average molecular weight of the organopolysiloxane is from about 10,000 to about 200,000 and desirably from about 20,000 to about 100,000 grams/mole. It is to be understood that the above polymers also contain, as noted above, two or more reactive functional groups (X) therein. The functional groups, independently, can be OH, or OR^3 , or

$N(R^4)_2$, enoxy, acyloxy, oximo, or aminoxy, wherein these functional groups may have substituents at any substitutable location. For example,



wherein R^3 through R^{14} are, independently, an alkyl or cycloalkyl having from about 1 to about 8 carbon atoms.

[0033] In one embodiment, the reactive organopolysiloxane of the present polymer composition may be depicted as

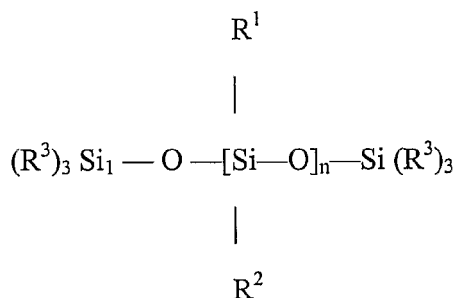


The one or more R groups, independently, is an alkyl having from 1 to 8 carbon atoms or an aromatic or an alkyl-aromatic having from 6 to 20 carbon atoms and optionally containing one or more functional groups thereon, such as amine, hydroxyl, alkene, alkoxy, and so forth. The amount of the functional groups, *i.e.*, m , is 1, 2 or 3.

[0034] The reactive functional group (X), can be OH, or OR', or $N(R')$, or enoxy, or acyloxy, or oximo, or aminoxy, or amido, wherein the reactive functional group may have substitutions, R' , at any substitutable C or N, and which is selected from the group consisting of an alkyl having from about 1 to about 8 carbon atoms, an aromatic, an alkyl-aromatic having from 6 to 20 carbon atoms, and wherein R' may optionally contain one or more functional groups thereon such as amine, hydroxyl, and so forth. An organopolysiloxane fluid can furthermore contain a blend of two or more different polysiloxanes and/or organopolysiloxanes having different molecular weights. The polysiloxanes are generally a

viscous liquid and are commercially available from several silicone manufacturers such as Wacker Corporation, General Electric, Dow Corning and Rhone-Poulenc.

[0035] In another embodiment, the present polymer composition comprises a non-reactive organopolysiloxane, i.e., the organopolysiloxane lacks functional groups. The non-reactive organopolysiloxane may be depicted as



where R^1 , R^2 , and R^3 independently, are an alkyl having from 1 to 8 carbon atoms, desirably from 1 to 4 carbon atoms with methyl being preferred, or is an aromatic group or substituted aromatic group having from 6 to 10 carbon atoms with phenyl being preferred, and “n” is such that the weight average molecular weight of the organopolysiloxane is from about 100 to about 100,000 and desirably from about 3,000 to about 50,000 grams/mole.

[0036] The polymer compositions of the present invention are generally made by blending the compatibilizer, the organopolysiloxane polymer, and optionally an organic polymer that does not contain silane groups.

ADHESIVE, COATING AND SEALANT COMPOSITIONS

[0037] The sealant, adhesive, and coating compositions of the present invention comprise a polymer composition of the present invention. In certain embodiments, the sealant, adhesive and coating compositions further comprise a silicone cross-linker. The silicone crosslinker used in the present adhesive, coating and sealant compositions may be selected from such crosslinkers as oximes, alkoxysilanes, epoxyalkylalkoxysilanes, amido silanes, aminosilanes, enoxysilanes, tetraethoxysilanes, methyltrimethoxysilane, vinyltrimethoxysilane, glycidoxypropyltrimethoxysilane, vinyltris-isopropenoxysilane, methyltris-isopropenoxysilane, methyltris-cyclohexylaminosilane, methyltris-secondarybutylaminosilane, condensation cure catalysts, and combinations thereof. Preferred oxime crosslinkers include vinyltrismethylethylketoximosilane, methyltrismethylethylketoximosilane, and combinations thereof. The compatibilizer contains silyl terminal groups that can enter into a vulcanization reaction with the silicone functional crosslinking system attached to the reactive organopolysiloxane polymer. Alternatively, the

compatibilizer can form an interpenetrating network with the organopolysiloxane. The sealant, coating and adhesive compositions of the present invention may also include a plasticizer, an adhesion promoter, a reinforcement agent, a rheology modifier, a chain extender, a UV stabilizer, catalyst, water scavenger, and/or surface modifier, such as wax, minifiber, glass bead, plastic bead and ceramic bead, and so forth.

[0038] The adhesive, coating and sealant compositions of the present invention generally are cured by generally subjecting them to moisture or a curative with or without heat. Either a conventional one-component or two-component cure system can be utilized. In a conventional one-component cure, the organopolysiloxane is converted to a compound having an alkoxy, an oxime, an enoxy, an amido, an amino, or an acetoxymethyl blocking group in a manner well known to the art and to the literature. Conventional condensation catalysts may be utilized, such as an organotin, for example, dibutyltin dilaurate, dibutyltin diacetate, dimethyltin di-2-ethylhexanoate, or dimethyltin hydroxyoctanoate, or an organotitanate.

[0039] Optionally, the blocked one-component cure system can contain crosslinking agents as set forth herein. In a two-component cure system, crosslinking agents, such as a multi-functional alkoxy silane or oligomers thereof and catalysts are kept separate from the organopolysiloxane until reaction. The amount of catalyst for the two-component system is generally higher than in the one-component system. Such cure systems are well known to the art, *see, e.g.*, Maurice Morton, *et al.*, *Rubber Technology*, 3rd Ed., pp. 406-407, (1987), which is hereby fully incorporated by reference. When a two part cure is used, the two-part cure may either be a room-temperature curable or a thermally curing material.

[0040] While a variety of conventional crosslinkers are suitable, oxime and alkoxy crosslinkers are preferred, such as, for example, vinyltris-methylethylketoximosilane, and methyltris-methylethylketoximosilane, and alkoxy silanes such as methyltrimethoxysilane and vinyltrimethoxysilane. Methyltrimethoxysilane is available, for example, under the trade name A-1630 and vinyltrimethoxysilane is available, for example, under the trade name A-171 from General Electric-OSI Specialties. Methyltris-methylethylketoximosilane (MOS) is available under the trade name OS-1000, for example, and vinyl tris-methylketoximosilane under the trade name of OS-2000 by Honeywell Corporation, for example. Other crosslinkers are also suitable, such as alkoxy silanes, epoxyalkylalkoxy silanes, amidosilanes, aminosilanes, enoxysilanes and the like, as well as tetraethoxysilanes, glycidoxypropyltrimethoxysilane, vinyltris-isopropenoxysilane, methyltris-isopropenoxysilane, methyltris-cyclohexylaminosilane, and methyltris-secondarybutylaminosilane. Mixtures of crosslinkers may also be employed. The amount of

the crosslinking agent is generally from about 0.2 to about 20 parts by weight, desirably from about 1 to about 10 parts by weight, and preferably from about 1.5 to about 6.5 parts by weight for every 100 parts by weight of said copolymer-organopolysiloxane. Addition cure crosslinkers may also be used in conjunction with alkenyl functional organopolysiloxane polymers. These crosslinkers may be pre-reacted to the polyorganosilane polymers.

[0041] The crosslinker is used in amounts which are conventionally used for making curable silicone elastomeric compositions. Those skilled in the art may determine the proper amounts for room-temperature curable and thermally curable crosslinkers. The amounts used will vary depending upon the particular crosslinker chosen and the properties of the cured elastomer desired, and may readily be determined by those of ordinary skill in the art.

[0042] A mixture of polyorganosiloxane and crosslinker will usually cure at room temperature when exposed to moisture, however, sometimes it is desirable to accelerate the cure rate, *i.e.*, reduce the time to cure composition. In these situations a catalyst may be used. Preferred catalysts include metal salts of carboxylic acids such as dibutyltindilaurate, dibutyltindiacetate, dibutyltin-bis(acetylacetonate) and dimethyltindi-2-ethylhexanoate; organotitanates such as tetrabutyltitanate, tetra-n-propyltitanate, diisopropoxydi(ethoxyacetoacetyl)titanate, and bis(acetylacetonyl)diisopropyltitanate. Alternatively, thermally curable materials may be used in addition to or in place of the room-temperature curable systems.

[0043] In certain embodiments, the sealant, coating and adhesive compositions comprise a plasticizer. The plasticizer generally has a number average molecular weight of from about 100 to about 800 and desirably from about 200 to about 400. Preferably the plasticizer has a boiling point greater than about 100° C. at 1 atmospheric pressure. Suitable plasticizers are organic oils used as silicone extenders, phthalate esters, dibenzoate esters, phosphate esters, and adipate esters. Examples of organic oils used as silicone extenders include aliphatic hydrocarbon oils and aromatic hydrocarbon oils such as alkylbenzene. Examples of phthalate esters are dibutyl phthalate, dioctyl phthalate, dimethyl phthalate, diisodecyl phthalate, butyl benzyl phthalate, texanol benzyl phthalate, C₇ benzyl phthalate, and C₉ benzyl phthalate. Examples of dibenzoate esters are dipropylene glycol dibenzoate, polypropylene glycol dibenzoate, diethylene glycol, and triethylene glycol dibenzoate. Examples of adipate esters are dioctyl adipate, ditiidecyl adipate, and dialkyl adipate. Examples of phosphate esters are tricryesyl phosphate, trioctyl phosphate, triphenyl phosphate, and t-butylphenyl diphenyl phosphate. Other suitable plasticizers are for example terephthalates such as 1,4-benzenedicarboxylates, and trimellitics such as 1,2,4-

benzenedicarboxylates. A suitable butyl benzyl phthalate, a phthalate ester of butyl alcohol and benzyl alcohol, is commercially available as Santicizer 160. Optionally, an adhesion promoter may be added to the present sealant, coating, and adhesive compositions. The amount of adhesion promoter may readily be determined by those of ordinary skill in the art. The amount of the adhesion promoter in the sealant is generally from 0 to about 10, desirably from about 1 to about 8, preferably from about 2 to about 6, and more preferably from about 1.5 to about 3 parts by weight per 100 parts of organosilicone polymer.

[0044] Suitable classes of adhesion promoters are aminoalkyl, mercaptoalkyl, ureidoalkyl, carboxy, acrylate and isocyanurate functional silanes. Examples of suitable adhesion promoters are mercaptopropyltrimethoxysilane, N-ethylaminoisobutyltrimethoxysilane, glycidoxypentyltrimethoxysilane, aminopropyltriethoxysilane, aminopropyltrimethoxysilane, aminoethylaminopropyltrimethoxysilane, ureidopropyltrimethoxysilane, bis- γ -trimethoxysilylpropylurea, 1,3,5-tris- γ -trimethoxysilylpropylisocyanurate, bis- γ -trimethoxysilylpropylmaleate and fumarate and γ -methacryloxypropyltrimethoxysilane.

[0045] Reinforcing agents may be added to the present sealant and adhesive compositions. The amount of the reinforcing agent is generally from 0 to about 250, desirably from about 30 to about 200 parts by weight of 100 parts of the polymer.

[0046] Reinforcing agents increase tensile strength in the cured sealant and reduce sag of the uncured sealant. The reinforcing agent also functions as a thixotrope. Such reinforcing agents are finely divided particulates and include both the conventionally known reinforcing agents and semi-reinforcing agents, typically having a particle size less than about 10 microns, preferably about 5 microns or less, more preferably about 0.1 microns or less. Suitable reinforcing agents include hydrophobic treated fumed silicas, such as TS 720 from Cabot Corporation, or R-972 from Degussa Corporation, hydrophobic precipitated calcium carbonates, talc, aluminum silicate, zinc oxides, polyvinyl chloride powders, and soft acrylates such as those of US Patent No. 6,403,711B1, incorporated herein by reference. Other ingredients can also be utilized in the sealant formulation in amounts up to about 20 parts by weight and desirably from about 0.01 to about 15 parts by weight per 100 parts by weight of the copolymer and the organopolysiloxane. Such ingredients include fungicides. Moreover, extender fillers such as ground calcium carbonates and diatomaceous earth are optionally employed. Such extenders have minimal or no reinforcing effect and/or minimal or no thixotropic effect.

[0047] UV stabilizers may also optionally be added to the present sealant and adhesive compositions. Pigments or colorants such as titanium dioxide, iron oxide, carbon black are optionally employed to impart color to the sealant and/or to act as ultraviolet stabilizer. LTV inhibitors, anitozonates are also optionally added.

[0048] The sealant in its uncured state may optionally contain solvents such as organic solvents to reduce the viscosity.

APPLICATION OF THE SEALANT AND COATING COMPOSITIONS

[0049] The sealant and adhesive compositions may be used to seal or bond buildings, highways, bridges, trucks, trailers, buses, recreational vehicles, cars, utilities, window applications, etc. Dissimilar materials that may, for example, be sealed or bonded with the present sealant composition include cement containing products, metals, plastics, glass, EIFS materials, vinyls, painted metals, coated glasses, Kynar, marble, granite, and brick. The sealant composition may be applied using standard equipment, such as a caulking gun or pump.

[0050] The coating composition may be applied to a surface by spraying or poured onto the surface and spread across the surface with a roller or other similar device.

EXAMPLES

[0051] The following compositions can be compounded using conventional mixing equipment such as a twin screw extruder or a change can mixer.

[0052] EXAMPLE 1

	<u>Weight, %</u>
1. Silane and aromatic alcohol endcapped polyurethane polymer	27.86
2. 80,000 cps. silanol terminated dimethylpolysiloxane	5
3. 20,000 cps. silanol terminated dimethylpolysiloxane	5
4. Organic plasticizer	9
5. White tint paste	6.5
6. Coated precipitated calcium carbonate	41
7. Rheology control agent	0.5
8. Fume silica	0.5
9. UV stabilizer	0.4
10. Hexamethyldisilazane	1
11 Vinyltrimethoxysilane	1
12. Methyltris-methylethylketoximosilane	0.8
13. Vinyltris-methylethylketoximosilane	0.8

14. Aminopropyltrimethoxysilane	0.1
15. N-ethylaminoisobutyltrimethoxysilane	0.5
16. Dibutyltin diacetate	0.04

[0053] In this example, about 26% (by weight, based on total polymer) of silicone polymer was used to prepare a medium modulus sealant. The organic polymer comprised a silane and aromatic alcohol endcapped polyurethane polymer. The composition upon curing had a tack free time of 180 minutes accompanied by a tooling time of 60 minutes with a shore A hardness of about 20. The sealant underwent more than 10,000 hours of weathering test in Xenon Arc weatherometer without any observable surface changes taking place.

[0054] EXAMPLE 2

	<u>Weight, %</u>
1. Silane and aromatic alcohol endcapped polyurethane polymer	27.86
2. 80,000 cps. silanol terminated dimethylpolysiloxane	10
3. 20,000 cps. silanol terminated dimethylpolysiloxane	10
4. White tint paste	6.5
5. Coated precipitated calcium carbonate	40
6. Rheology control agent	0.5
7. Fume silica	0.5
8. UV stabilizer	0.4
9. Hexamethyldisilazane	1
10 Vinyltrimethoxysilane	1
11. Methyltris-methylethylketoximosilane	0.8
12. Vinyltris-methylethylketoximosilane	0.8
13. Aminopropyltrimethoxysilane	0.1
14. N-ethylaminoisobutyltrimethoxysilane	0.5
15. Dibutyltin diacetate	0.04

[0055] In this example, about 42% (by weight, based on total polymer) of silicone polymer was used to prepare a faster cure sealant. The organic polymer comprised a silane and aromatic alcohol endcapped polyurethane polymer. The composition upon curing had a tack free time of 100 minutes accompanied by a tooling time of 30 minutes. The sealant underwent more than 10,000 hours of weathering test in Xenon Arc weatherometer without any observable surface changes taking place.

[0056] EXAMPLE 3

	<u>Weight, %</u>
1. Silane and aromatic alcohol endcapped polyurethane polymer	27.12
2. 1,000 cps. trimethylsiloxy terminated polydimethylsiloxane	6
3. Organic plasticizer	10
4. White tint paste	6.5
5. Coated precipitated calcium carbonate	40
6. Coated ground calcium carbonate	7
7. Rheology control agent	0.5
8. Fume silica	0.5
9. UV stabilizer	0.4
10. Hexamethyldisilazane	0.2
11. Vinyltrimethoxysilane	1.1
12. Aminopropyltrimethoxysilane	0.1
13. N-ethylaminoisobutyltrimethoxysilane	0.5
14. Dibutyltin diacetate	0.08

[0057] In this example, about 6% (by weight, based on total polymer) of non-reactive silicone polymer was used to prepare a sealant. The organic polymer comprised a silane and aromatic alcohol endcapped polyurethane polymer. The composition upon curing had a tack free time of 120 minutes accompanied by tooling time of 40 minutes. The sealant did not show any sign of staining on marble.

[0058] EXAMPLE 4

	<u>Weight, %</u>
1. Silane and aromatic alcohol endcapped polyurethane polymer	2.82
2. 20,000 cps. silanol terminated dimethylpolysiloxane	2.82
3. Vinyltris-methylethylketoximosilane	0.19
4. Polyvinylchloride	16.45
5. Organic plasticizer	19.95
6. Color tint paste	2.95
7. Fungicide	0.04
8. UV stabilizer	0.43
9. Calcium oxide	1.90
10. Calcium carbonate	18.32
11. Xylene	40.77

12. Mineral Spirits	4.2
13. NCO terminated urethane prepolymer	26.2
14. Glycidoxypropyltrimethoxysilane	0.22
15. Hardener OZ	2.43
16. Dimer acid	0.31

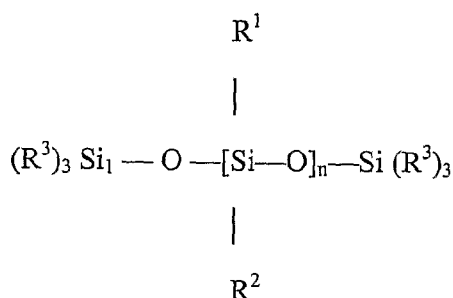
[0059] In this example, about 8.9% (by weight, based on total polymer) of silicone polymer, 8.9% of silane and aromatic alcohol endcapped polyurethane polymer (by weight, based on total polymer), and 82.3% of a NCO terminated urethane prepolymer were used to prepare a sealant that did not show phase separation.

[0060] The examples included herein are for illustration and are not meant to limit the scope of the invention.

WE CLAIM:

1. A polymer composition comprising:
 - a) from 1 to 99% by weight (based on total polymer) of at least one organic polymer that comprises silane groups, wherein said organic polymer has a molecular weight greater than 1000 g/mole;
 - b) from 1 to 45% by weight (based on total polymer) of at least one organopolysiloxane polymer; and
 - c) from 0 to 98 % by weight (based on total polymer) of an organic polymer that lacks silane groups, wherein said organic polymer has a molecular weight greater than 1000g/mole.
2. The polymer composition of claim 1, wherein at least one of the organic polymers that comprises silane groups is partially or fully end capped with silane groups.
3. The polymer composition of claim 1, wherein the organic polymers that comprise silane groups are chosen from urethane polymers, polyethers, polyesters, acrylic polymers, styrene polymers, styrene butadiene polymers, butadiene polymers, and butyl rubber, or any mixture or blend thereof.
4. The polymer composition of claim 1, wherein at least one of the organic polymers that comprises silane groups is a polyurethane polymer that is partially or fully end-capped with silane groups.
5. The polymer composition of claim 1, wherein at least one of the organic polymers that comprises silane groups is a polyurethane polymer that is partially end-capped with silane groups and partially end-capped with alcohol groups.
6. The polymer composition of claim 1, wherein the organopolysiloxane polymer has at least two reactive functional groups on the polymer chain.
7. The polymer composition of claim 6, wherein the reactive functional groups are selected from the group consisting of hydroxyl, alkoxy, silicone alkoxy, acyloxy, ketoximo, amino, amido, aminoxy, alkenoxy, alkenyl, enoxy and mixtures thereof.
8. The polymer composition of claim 7, wherein the reactive functional groups are end groups or pendant groups, or both.
9. The polymer composition of claim 1, wherein the organopolysiloxane has a molecular weight in the range from 10,000 to 200,000 grams/mole.
10. The polymer composition of claim 1, wherein the organopolysiloxane is a non-reactive organopolysiloxane.

11. The polymer composition of claim 10, wherein the non-reactive organopolysiloxane has the following formula:



where R^1 , R^2 , and R^3 independently, are an alkyl having from 1 to 8 carbon atoms, an aromatic group or substituted aromatic group having from 6 to 10 carbon atoms, with, and “n” is such that the weight average molecular weight of the organopolysiloxane is from about 100 to about 100,000.

12. The polymer composition of claim 1, wherein the organic polymers that lack silane groups are chosen from urethane polymers, polyethers, polyesters, acrylic polymers, styrene polymers, styrene butadiene polymers, butadiene polymers, and butyl rubber, or any mixture or blend thereof
13. The polymer composition of claim 1, wherein the composition further comprises a silicone functional cross-linker.
14. The polymer composition of claim 13, wherein the amount of cross-linker in the composition is from 0.2% to 4% by weight, based on organopolysiloxane.
15. The polymer composition of claim 13, wherein the silicone crosslinker is selected from the group consisting of oximes, alkoxysilanes, epoxyalkylalkoxysilanes, amido silanes, aminosilanes, enoxysilanes, tetraethoxysilanes, methyltrimethoxysilane, vinyltrimethoxysilane, glycidoxypropyltrimethoxysilane, vinyltris-isopropenoxysilane, methyltris-isopropenoxysilane, methyltris-cyclohexylaminosilane, methyltris-secondarybutylaminosilane, condensation cure catalysts, and combinations thereof.
16. The polymer composition of claim 15, wherein the crosslinker is an oxime crosslinker.
17. The polymer composition of claim 15, wherein the oxime crosslinker is selected from vinyltris-methylethylketoximosilane, methyltris-methylethylketoximosilane, and combinations thereof.
18. The polymer composition of claim 1, wherein the polymer composition is a 1-part curable compositions, a 2-part curable composition, or a combination thereof.

19. The polymer composition of claim 1, wherein the polyorganosiloxane is selected from the group consisting of thermally curing systems and room temperature curing systems.
20. The polymer composition of claim 1, further comprising from 0.01% to 2% (by total weight) catalyst.
21. The polymer composition of claim 20, wherein the catalyst is selected from the group consisting of metal salts of carboxylic acids, organotitanates, platinum complexes, peroxides, and combinations thereof.
22. The polymer composition of claim 20, wherein the catalyst comprises metal salts of carboxylic acids selected from the group consisting of dibutyltindilaurate, dibutyltindiacetate, dibutyltin-bis(acetylacetonate), dimethyltindi-2-ethylhexanoate, and combinations thereof.
23. The curable silicone composition of claim 20, wherein the catalyst comprises organotitanates selected from the group consisting of tetrabutyltitanate, tetra-n-propyltitanate, diisopropoxydi(ethoxyacetoacetyl)titanate, bis(acetylacetonyl)diisopropyl titanate and combinations thereof.
24. The polymer composition of claim 1, further comprising a plasticizer.
25. The polymer composition of claim 24, wherein the plasticizer has a number average molecular weight of from 100 to 1000.
26. The polymer composition of claim 24, wherein the plasticizer is an aliphatic or aromatic hydrocarbon oil.
27. The polymer composition of claim 1, further comprising from 3% to 60% (by total weight) reinforcement agents, semi-reinforcing agents, or combinations thereof.
28. The polymer composition of claim 27, wherein the reinforcement agent is selected from the group consisting of hydrophobic treated fumed silicas, untreated fumed silicas, hydrophobic precipitated calcium carbonates, ground calcium carbonates, talc, zinc oxides, polyvinyl chloride powders, soft acrylic polymers, aluminum silicate, and combinations thereof.
29. A sealant, coating or adhesive composition comprising:
 - a) from 1 to 99% by weight (based on total polymer) of at least one organic polymer that comprises silane groups, wherein said organic polymer is a polyether polymer or a polyacrylate polymer;
 - b) from 1 to 45% by weight (based on total polymer) of at least one organopolysiloxane polymer;

- c) from 0 to 98 % by weight (based on total polymer) of an organic polymer that lacks silane groups, wherein said organic polymer has a molecular weight greater than 1000g/mole and optionally
 - d) a plasticizer;
 - e) an adhesion promoter;
 - f) a crosslinker, and
 - g) a catalyst.
30. A sealant, coating or adhesive composition comprising
- a) from 1 to 99% by weight (based on total polymer) of at least one organic polymer that comprises silane groups, wherein said organic polymer is chosen from urethane polymers, polyethers, polyesters, acrylic polymers, styrene polymers, styrene butadiene polymers, butadiene polymers, and butyl rubber, or any mixture or blend thereof;
 - b) from 1 to 45% by weight (based on total polymer) of at least one organopolysiloxane polymer
 - c) 0 to 98 % by weight (based on total polymer) of an organic polymer that lacks silane groups, wherein said organic polymer has a molecular weight greater than 1000g/mole; and
 - d) an aliphatic or aromatic hydrocarbon that acts as a silicone extender; and optionally
 - e) an adhesion promoter;
 - f) a crosslinker, and
 - g) a catalyst.