

PROCESS FOR SEPARATING LINEAR ALPHA OLEFINS FROM A STREAM
CONTAINING SATURATED HYDROCARBONS, INTERNAL OLEFINS,
BRANCHED OLEFINS AND LINEAR ALPHA OLEFINS

The present invention relates to a process for separating olefins from saturated hydrocarbons, and more particularly, to a process for separating olefins from saturated hydrocarbons in a Fisher-Tropsch (FT) stream.

5 Many industrial processes produce olefin/saturated hydrocarbon streams that are mixtures of olefins, saturated hydrocarbons, and oxygenates. Olefins are frequently used in the manufacture of polymers such as polyethylene, as drilling mud additives, or as intermediates for the
10 production of oil additives and detergents. Some industrial processes manufacture olefin streams by oligomerizing ethylene over an alpha olefin catalyst to produce mixtures of alpha and internal olefins having a broad range of carbon numbers. However, these streams rely
15 on the use of ethylene as a feedstock material, which adds a significant cost to the manufacture of the olefin. On the other hand, the Fischer-Tropsch (FT) process starts with an inexpensive feedstock, syngas, generally derived from natural gas, coal, coke, and other carbonaceous
20 compounds to make oligomers comprised of olefins, aromatics, saturates, and oxygenates.

The FT process, however, is not very selective to the production of olefins. While reaction conditions and catalysts can be tuned to manufacture a stream rich in the
25 desired species within the FT product stream, a large percentage of the FT stream contains other types of compounds which must be separated from the olefins, which olefins are purified, and then sold into different markets. For example, a typical commercial FT stream will

contain a mixture of saturated hydrocarbons having a broad spectrum of molecular weights, olefins, aromatics, and oxygenates such as organic carboxylic acids, alcohols, ethers, esters, ketones, and aldehydes. All these compounds must be separated from the crude FT stream before a particular composition may be offered commercially. To further complicate the separation operation, the FT stream contains compounds having a wide spectrum of carbon numbers, as well as a wide variety of olefins, ranging from C₂-C₂₀₀ species, internal linear olefins, alpha linear olefins, internal branched olefins, alpha branched olefins, and cyclic olefins, many of which have similar molecular weights. Separating and isolating these species is no easy task. Conventional distillation methods are frequently inadequate to separate species having closely related boiling points.

Various processes have been proposed to efficiently separate the different species in an FT stream with sufficient purity that a particular composition is acceptable in the intended application. These processes for separating out different species in an FT stream include the use of molecular sieves, which are restricted to a feed having an average carbon number range which is more limited than a composition containing a broad spectrum of average carbon numbers ranging from C₅-C₂₀, the use of exchange resins, the use of super-fractionators often operated at high pressure, and the use of oligomerization catalysts or etherification techniques to alter the boiling points of the species in the FT stream. Many reactive methods for separating species in an FT stream do not, however, selectively react with olefins while simultaneously rejecting paraffins.

It would be desirable to conduct a separation operation on an FT stream in which the activity and life of the

separating agent is not diminished by the presence of impurities in the stream, such as oxygenates; which remains active under a wide band of average carbon numbers ranging from C5-C20, and which distinguishes
5 between olefins and paraffins in an FT stream.

US Patent No. 4,946,560 describes a process for the separation of internal olefins from alpha olefins by contacting a feedstock with an adducting compound such as anthracene to form an olefin adduct, separating the
10 adduct from the feedstock, dissociating the olefin adduct through heat to produce anthracene and an olefin composition enriched in alpha olefin, and separating out the anthracene from the alpha olefin. This reference does not suggest the desirability or the capability of
15 anthracene to separate olefins from saturated hydrocarbons or linear alpha olefins from saturated hydrocarbons.

WO 98/57911 describes a process for the separation of linear alpha-olefins from a feedstock comprising 2-
20 branched alpha-olefins and/or 3-branched alpha-olefins. The process comprises contacting the feedstock with a linear polyaromatic compound having three or more fused aromatic rings, such as anthracene, to form a linear alpha-olefin adduct, separating the adduct from the
25 feedstock, dissociating the adduct to produce a linear polyaromatic compound and an olefin composition enriched in linear alpha-olefins and separating out the linear polyaromatic compound from the linear alpha-olefin. This reference does not suggest the desirability or the
30 capability of anthracene to separate olefins from saturated hydrocarbons or linear alpha olefins from saturated hydrocarbons.

The present invention relates to a process for separating olefins from saturated hydrocarbons, and
35 thereafter treating the olefins to separate linear alpha

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olefins from internal olefins. The process of the present invention is well suited to treating an FT stream.

In particular, there is provided a process for treating a feedstock composition comprising linear alpha olefins,
5 internal olefins, and saturated hydrocarbons, comprising:

(a) contacting the feedstock composition with a linear
polyaromatic compound in a first reaction zone under
conditions effective to form a reaction mixture comprising
linear polyaromatic compound-olefin adducts and saturated
10 hydrocarbons;

(b) separating the linear polyaromatic compound-olefin
adducts from the saturated hydrocarbons in the reaction
mixture to form a first adducted olefin stream and a
saturated hydrocarbon stream;

15 o(i) dissociating linear polyaromatic compound-olefin
adducts in the first adducted olefin stream to form linear

polyaromatic compounds and an olefin composition comprising alpha olefins and internal olefins;

o(ii) contacting the olefin composition with linear polyaromatic compounds in an alpha olefin reaction zone under conditions effective to form a reaction mixture comprising linear polyaromatic compound-linear alpha olefin adducts and an internal olefin composition;

o(iii) separating the linear polyaromatic compound-linear alpha olefin adducts from the reaction mixture obtained from the alpha olefin reaction zone to form a linear alpha olefin adducted stream and an internal olefin stream; and

o(iv) dissociating the linear polyaromatic compound-linear alpha olefin adducts in the linear alpha olefin adducted stream to form linear polyaromatic compounds and a linear alpha olefin composition;

whereby the concentration of linear alpha olefins in the linear alpha olefin composition is enriched over the concentration of linear alpha olefins in the olefin composition, and the concentration of linear alpha olefins in the olefin composition is enriched over the concentration of linear alpha olefins in the feedstock.

According to the present invention there is also provided a process for contacting a linear polyaromatic compound with a feedstock composition comprising linear alpha olefins, internal olefins, and saturated hydrocarbons, separating the olefins from the saturated hydrocarbons in the feedstock composition to form an olefin composition and a saturated hydrocarbon stream, subsequently contacting a linear polyaromatic compound with the olefin composition comprising linear alpha olefins and internal olefins, and separating the linear alpha olefins from the internal olefins to form a linear alpha olefin stream, whereby the concentration of linear alpha olefins in the linear alpha olefin stream is enriched over the

concentration of linear alpha olefins in the feedstock and in the olefin composition.

The feed stream to be treated comprises at least olefins and saturated hydrocarbons. The class of saturated hydrocarbons as used herein includes at least a paraffin. The class of saturated hydrocarbons may also include other molecules such as cycloparaffins.

An olefin means any compound containing at least one carbon-carbon double bond. The olefins may be linear, branched, conjugated, contain multiple double bonds anywhere along the chain, substituted, unsubstituted, contain aryl or alicyclic groups, or contain heteroatoms.

The olefins may contain aryl moieties along with an aliphatic or cycloaliphatic moiety within the same compound, or may consist solely of an aliphatic, cycloaliphatic, or cycloaliphatic with aliphatic moieties on the compound. Preferably, the olefin is an aliphatic compound.

The olefin may be branched or linear. Examples of branching include alkyl, aryl, or alicyclic branches. The number of unsaturation points along the chain is also not limited. The olefin may be a mono-, di-, tri-, etc. unsaturated olefin, optionally conjugated. The olefin may also contain acetylenic unsaturation.

An alpha olefin is an olefin whose double bond is located on both of α and β carbon atoms. An α carbon atom is any terminal carbon atom, regardless of how long the chain is relative to other chain lengths in a molecule. The alpha olefin may be linear or branched. Branches or functional groups may be located on double bond carbon atoms, on carbon atoms adjacent to the double bond carbon atoms, or anywhere else along the carbon backbone. The alpha olefin may also be a poly-ene, wherein two or more points of unsaturation may be located anywhere along the

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molecule, so long as at least one double bond is in the alpha position.

5 An internal olefin is an olefin whose double bond is located anywhere along the carbon chain except at any terminal carbon atom. The internal olefin may be linear or branched. The location of a branch or substitution on the internal olefin is not limited. Branches or functional groups may be located on the double bond carbon atoms, on carbon atoms adjacent to the double bond carbon atoms, or
10 anywhere else along the carbon backbone.

The olefin may also be substituted with chemically reactive functional groups. These types of compounds include those identified as oxygenates. Examples of chemically reactive functional groups are carboxyl, aldehyde, keto, thio, ether, hydroxyl, and amine. The
15 number of functional groups on a molecule is not limited. The functional groups may be located anywhere along the carbon backbone.

The feedstock is generally produced by commercial processes such as the oligomerization of ethylene, optionally followed by isomerization and disproportionation. Alternatively, the feedstock may be produced by the Fischer-Tropsch process, which typically contains a high proportion of paraffins. A Fischer-Tropsch
20 process catalytically hydrogenates carbon monoxide to produce compositions containing aliphatic molecular chains. Other processes for making feedstocks which may contain mixtures of olefins and paraffins include the dehydrogenation of paraffin, such as those made by the PacolTM processes of UOP, and the cracking of waxes. The
25 most preferred feedstock is that obtained from a Fischer-Tropsch (FT) synthesis.

FT catalysts and reaction conditions can be selected to provide a particular mix of species in the reaction product stream. For example, the particular catalyst and reaction
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conditions may be tuned to enhance the amount of olefins and decrease the amount of paraffins and oxygenates in the stream. Alternatively, the catalyst and reaction conditions may be tuned to enhance the amount of paraffins and decrease the amount of olefins and oxygenates in the stream.

Generally, the reaction conditions will vary depending on the type of equipment employed. The FT reaction temperatures vary from 100°C to 500°C, an inlet gas pressure to the reactor from atmospheric to 10.3 MPa (1500 psig), and an hydrogen/carbon monoxide ratio from 0.5:1 to 5:1, preferably from 1.8:1 to 2.2:1, and gas hourly space velocity ranging from 1 to 10,000 v/v/hour. A variety of reactor vessel configurations can be used, including a fluidized(entrained) bed, a fixed bed, and a slurried bed. The temperature in these beds can be adjusted by those of ordinary skill to optimize the formation of FT products, including hydrocarbons, and particularly, olefins and types of olefins. To illustrate without limitation, in fluidized (entrained) bed(s), the temperature of reaction is generally high- e.g. ranging from 280° to 350°C, preferably 310° to 340°C. If a fixed bed reactor(s) is used, the reaction temperature generally ranges from 200°C to 250°C, preferably from 210° to 240°C, and when a slurry bed reactor(s) is used, the temperature is generally within the range of 190°C to 270°C.

The catalyst used in the FT process is any known in the art, but preferably from among molybdenum, tungsten, and Group VIII compounds, including iron, cobalt, ruthenium, rhodium, platinum, palladium, iridium, osmium, combinations of the foregoing, combinations with other metals, and each being in the free metal form or as alloys, or as an oxide or carbide or other compound, or as a salt. Iron based and cobalt based catalysts have found common commercial use,

and ruthenium has gained importance as a metal for the catalyst which favours the formation of high melting waxy species under high pressure conditions. Those skilled in the art will recognize which catalysts and combinations will favour the manufacture of desired species in the FT reaction composition. For example, fused iron containing a promoter such as potassium or oxides on a silica, alumina, or silica-alumina support are known as FT synthetic catalysts. Another example is the use of cobalt metal. Cobalt has the advantage of producing less methane during synthesis over the older nickel based catalysts, and produces a wide spectrum of species. With the proper selection of supports, promoters, and other metal combinations, the cobalt catalyst can be tuned to manufacture a composition rich in the desired species. Other catalysts, such as iron-cobalt alloy catalysts, are known for their selectivity toward olefins under certain process conditions.

The catalysts may be fused or precipitated, or sintered, cemented, impregnated, kneaded or melted on to a suitable support.

The catalysts may also contain promoters to promote the catalyst's activity, stability, or selectivity. Suitable promoters include alkali or alkaline earth metals, in free or combined form as an oxide, hydroxide, salt, or combinations thereof.

An FT stream generally contains virtually no sulphur or nitrogen compounds, which may be deleterious to other catalysts which derivatize the olefins or catalyze the reaction of olefins in other oligomerization or polymerization processes. Regardless of the method used, however, the FT process is not very selective to a particular species, and yields a wide variety of species within a composition.

The linear polyaromatic compound used in the process of the invention, however, is particularly well adapted for the separation of olefins from saturated hydrocarbons in an FT stream in the presence of oxygenates since oxygenates do not significantly impair the performance of the linear polyaromatic compound.

While reference is made to a FT stream, it is to be understood that any stream made by any process containing olefins and saturated hydrocarbons are suitable feedstocks for the process of the present invention. Most crude FT streams contain from 5 wt.% to 95 wt.% olefins, the remainder being saturated hydrocarbons comprising paraffins and cycloparaffins, and optionally other compounds such as aromatics optionally containing saturated or unsaturated alkyl branches, and oxygenates, based on the weight of all ingredients in the feedstock stream to the process of the invention.

The preferred amount of olefins present in the FT stream ranges from 15 wt.% to 70 wt.%, based on the weight of the FT stream. The amount of linear alpha olefin in the FT stream is not limited, but preferably ranges from 15 wt.% to 65 wt.%, based on the weight of the FT stream. The amount of other olefins, including branched alpha olefins and internal olefins, both linear and branched, is also not limited, but preferably ranges from 1 wt.% to 55 wt.%, more preferably from 5 wt.% to 45 wt.%, based on the weight of the FT stream. The amount of paraffin in most FT streams ranges from 5 wt.% to 95 wt.% based on the weight of all ingredients in the feedstock. In some FT streams, the FT catalyst is tuned to enhance the olefin concentration and decrease the paraffin concentration. In these streams, the amount of paraffin generally ranges from 5 to 65 wt.% of the stream. In other FT streams where the FT catalyst is tuned to enhance the amount of paraffin, the amount of paraffin in the stream ranges from 65 wt.% to 95 wt.%. The

amounts of other compounds in a FT stream, such as oxygenates and aromatics, make up most of the remainder of the FT stream, and are generally present in amounts ranging from 5 wt.% to 40 wt.%. Minor amounts of other by-products and impurities, less than 5 wt.%, may be present in most FT streams.

The feedstock may be a processed FT stream which has been fractionated and/or purified by a conventional distillation, extraction, or other separation operation to obtain a desired carbon number cut, including a composition containing a mixture of carbon numbers or a single carbon cut composition, and to remove high and low boiling compounds, including olefins, paraffins, aromatics, and oxygenates from the crude stream. When the separation operation is conducted by distilling the reaction mixture containing the adduct, it is preferred that the feedstock used in the process of the invention contain an average carbon number ranging from C₅-C₂₀ and wherein the predominant olefin species in the feedstock is within the range of C₅-C₂₀, inclusive. The linear polyaromatic compound efficiently separates the saturated hydrocarbons from the olefins when the average carbon number of the feedstock and the predominant olefinic species is within this range, inclusive. When the average carbon number of the feedstock exceeds C₂₀, the polyaromatic compound-olefin adduct boils at a lower temperature than many of the species in the C₂₀ + feedstock composition, thereby leaving these high boiling species in the reaction mixture bottoms containing the adduct. Accordingly, the particular linear polyaromatic compound and the particular feedstock composition should be so selected that the linear polyaromatic compound-olefin adduct composition in the reaction mixture boils at a higher temperature than the amount of unreacted paraffin species in the feedstock one

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desires to separate. Therefore, the feedstock stream is preferably one which contains an average carbon number ranging from 5 to 20, and more preferably ranging from 6 to 18, and wherein the predominant olefin species is within these ranges, inclusive. These types of FT streams are generally processed by one of the techniques identified above to substantially remove cuts containing ingredients below or exceeding the range of C₅-C₂₀.

In addition to mixtures of olefins within this range, one may also employ what are known as single carbon cuts of olefins as feedstocks, wherein the single cut is within this range. For example, the feedstock employed may be a single C₆ carbon cut, C₈ carbon cut, or C₁₀ carbon cut, C₁₂ carbon cut, C₁₄ carbon cut, and C₁₆ carbon cut. These carbon cuts have utility as comonomers for polyethylene, poly alpha olefins, alpha olefin sulfonates, and as drilling fluids.

In the event that one desires to employ a feedstock outside the range of C₅-C₂₀, other separation techniques can be used to separate the adduct from the unreacted reaction mixture, including the selection of higher boiling polyaromatic compounds and/or other separation techniques such as liquid/liquid extraction or crystallization. These techniques, of course, can also be used with feedstocks within the range of C₅-C₂₀, inclusive.

The linear polyaromatic compound is utilized in the process of the present invention to form the adduct with the olefins in the feed stream. As used herein, "linear polyaromatic compound" refers to a linear polyaromatic compound having at least three fused aromatic rings, which may be unsubstituted or substituted and possess similar adducting properties as the unsubstituted molecule, and mixtures thereof. The linearity should extend to all three of the fused rings if a three fused ring compound is used

and to at least four consecutively fused cyclic rings if a four or more fused ring compound is used. The linear polyaromatic compound also refers to mixtures of compounds containing as one of their ingredients the linear
5 polyaromatic compound, including but not limited to coal tars, anthracene oil, and any crude mixtures containing cuts separated from naphthalene. The linear polyaromatic compound also includes aromatic molecules linked together by a bridging group, such as a hydrocarbon chain, an ether
10 linkage, or a ketone group-containing chain so long as at least three fused rings are present in a linear arrangement; as well as those containing a heteroatom which do not interfere in the separation of olefins from saturated hydrocarbons.

15 The linear polyaromatic compound has a preferential selectivity toward adducting with linear alpha olefin compounds, and secondly with other olefins, and last with paraffins, with which the compound is virtually unreactive under any operating condition outside of cracking
20 conditions. The linear polyaromatic compound of choice is one which has a selectivity for linear alpha olefin compounds over other olefins greater than 1:1 by mole, preferably 2:1 or more, more preferably 4:1.

25 Non-limiting examples of the linear polyaromatic compound include anthracene, 2,3-benzanthracene, pentacene, and hexacene. Suitable examples of substituents in substituted linear polyaromatic compounds include, but are not limited to, lower alkyl, e.g., methyl, ethyl, butyl; halo, e.g., chloro, bromo, fluoro; nitro; sulphato;
30 sulphonyloxy; carboxyl; carbo-lower-alkoxy, e.g., carbomethoxy, carbethoxy; amino; mono- and di-lower-alkylamino, e.g., methylamino, dimethylamino, methylethylamino; amido; hydroxy; cyano; lower-alkoxy, e.g., methoxy, ethoxy; lower-alkyanoxyloxy, e.g., acetoxyl;
35 monocyclic aryls, e.g., phenyl, xylyl, tolyl, benzyl, etc.

The particular substituent size, their number, and their location, should be selected so that they are relatively inert under the reaction conditions and not so large as to block the formation of the Diels-Alder adduct. Suitable substituted linear polyaromatic compounds can be determined by routine experimentation. Examples of suitable linear polyaromatic compounds include 9,10-dimethylantracene, 9,10-dichloroanthracene, 9-methylantracene, 9-acetylantracene, 9-(methylaminomethyl)anthracene, 2-chloroanthracene, 2-ethyl-9,10-dimethoxyanthracene, anthrarobin, and 9-anthryl trifluoromethyl ketone. The preferred linear polyaromatic compounds are substituted or unsubstituted anthracene and/or benzanthracene, particularly unsubstituted anthracene and 2,3-benzanthracene.

In a first reaction zone in step (a), the feedstock composition, preferably an FT feedstock stream having an average carbon number ranging from C₆-C₁₈, is contacted with a linear polyaromatic compound. In the Alpha Olefin (AO) reaction zone in step o(ii), the product of the first reaction zone and separation, the olefin composition, is also contacted with a linear polyaromatic compound. In each reaction zone, the Diels-Alder adduct forming reaction is carried out in a conventional fashion. Examples of suitable equipment in which the reactions are carried out include a continuously stirred tank reactor, configured as a single unit, or units in parallel or in series, wherein feedstock or an olefin composition, and linear polyaromatic compound, are added continuously to a stirred tank to form a liquid reaction mixture under heat, and the reaction mixture is continuously withdrawn from the stirred tank. Alternatively, the reaction may be carried out in a bubble column, or in a batch reactor, or utilizing a plug flow reaction scheme.

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The feedstock and olefin composition adducting reactions are typically carried out over a range of temperatures from 150° to 290°C, preferably from 200° to 280°C, and most preferably from 240° to 265°C. Pressures typically run from 101.33 kPa to 10.13 MPa (atmospheric pressure to 100 atmospheres). The reactions can be carried out in the gas phase under vacuum or liquid phase or mixed gas-liquid phase, depending on the volatility of the feedstock, but generally in the liquid phase.

Stoichiometric ratios or an excess of either olefin or linear polyaromatic compound can be used to form the adducts. The molar ratio of olefins to linear polyaromatic compound is preferably from 0.25:1 to 10:1. Preferably, a molar excess of linear polyaromatic compounds is used to ensure a complete and large recovery of all olefins in the first and subsequent adduction zones. However, in the succeeding adducting reaction zone where greater selectivity towards forming adducts with linear alpha olefins is desired, the molar ratio of linear polyaromatic compounds to olefins may be moderated, for example, desirably closer towards a 1.5:1 to 0.5:1 molar ratio. The residence time is for a time sufficient to adduct the desired amount of linear polyaromatic compound with the olefin. Typical residence times range from 30 minutes to 4 hours in a batch reaction.

An inert solvent can be utilized to dissolve the feedstock olefins or the linear polyaromatic compound or both in the reactor. Preferred solvents are the hydrocarbon solvents which are liquid at reaction temperatures and in which the olefins, linear polyaromatic compound and olefin-linear polyaromatic compound adducts are soluble. Illustrative examples of useful solvents include the alkanes such as pentane, iso-pentane, hexane, heptane, octane and nonane; cycloalkanes such as

cyclopentane and cyclohexane; and aromatics such as benzene, toluene, ethylbenzene and diethylbenzene. The amount of solvent to be employed can vary over a wide range without a deleterious effect on the reaction.

5 Preferably, the feedstock adduction, and particularly, the linear polyaromatic compound-olefin adduct formation is carried out in the absence of a solvent to improve the rate or reaction and avoid additional equipment and process steps for separating the solvent.

10 After formation of the linear polyaromatic compound-olefin adduct in step (a), the adduct stream flows to a separation vessel effective for separating the saturated hydrocarbons from the linear polyaromatic compound-olefin adduct to form a saturated hydrocarbon stream and an olefin
15 adducted stream in step (b). Due to the large molecular weight and structural difference between the adducts and the other ingredients in the reaction mixtures, such as saturated hydrocarbons and internal olefins, conventional separation techniques are quite suitable for removing the
20 unreacted saturated hydrocarbons in step (b) and internal olefins in step o(iii) from the respective adducts. For example, the saturated hydrocarbons in step (b) and the internal olefins in step o(iii) may be removed at the overhead or in fractions, by partial vacuum or flash
25 distillation of the reaction mixture to leave the adducts and unreacted linear polyaromatic compounds as a liquid bottoms. It is desirable to raise the temperature at the bottom of the distillation column sufficient to retain the bottoms in liquid state, while keeping the temperature and
30 residence time as low as possible to avoid dissociating the adducts. Suitable temperatures at the bottom of the separation vessel range from 210°C to 280°C, more preferably from 230°C to 270°C. While the pressure is not particularly limited, and the separation can be conducted

under atmospheric pressure, it is preferred to conduct the separation under slight vacuum, e.g. 200 mmHg to 700 mmHg, to reduce the operating temperature and the residence time within the separation vessel. The residence time within the vessel should be short to avoid excessive dissociation of the adducts, such as from 1 to 30 minutes.

In step (b), the unreacted saturated hydrocarbon stream distillate includes paraffins and may include, if present in the feedstock composition, aromatics and oxygenates such as the alcohols, ketones, acids, along with internal and branched olefins which failed to adduct with the linear polyaromatic compound.

Alternatively, the adducts may be separated by cooling the reaction mixture until the adducts crystallize out, followed by filtration or centrifugation to remove the unreacted saturated hydrocarbons in step (b) or the unreacted internal olefins in step o(iii).

In most cases, any unreacted linear polyaromatic compound will separate out with the adduct in the adducted olefin stream and the adducted linear alpha olefin stream. Other ingredients, such as small amounts of higher molecular weight unreacted olefins, internal olefins, and branched olefins, may remain in the adducted olefin stream and the adducted linear alpha olefin stream.

The process of the invention affords the flexibility for adjusting the recovery of a stream at each adducting and separation step to optimize the desired stream yield and concentration of species in the desired stream. For example, if one desires an alpha olefin stream highly concentrated in alpha olefin, the recovery of olefins from the feedstock will be moderate to avoid entraining excessive amounts of other olefins, part of which would otherwise be entrained in successive separations and dilute the linear alpha olefin concentration. Highly concentrating the linear alpha olefins, however, results in

lower linear alpha olefins stream yields than could be had if the recovery of olefin levels from the feedstock were set higher. On the other hand, if the linear alpha olefin stream yield is more desirable than attaining high concentrations of linear alpha olefins in the linear alpha olefin stream, then the recovery of olefins from the feedstock should be set a high level to ensure that larger quantities of olefins, including linear alpha olefins, are entrained in the olefin composition during the separation step, thereby resulting downstream in a larger alpha olefin stream yield, but at lower alpha olefin concentrations.

The recovery of a stream in a separation operation is determined by the molar ratio of linear polyaromatic compound to olefins, the adducting residence time, the temperature within the separation vessel, and most importantly, the residence time (rate of separation) of the reaction mixture in the separation vessel. To obtain a large olefin composition recovery, any one or a combination of the following variables are adjusted: a high linear polyaromatic compound to olefin molar ratio, e.g., >1 , long residence times to ensure complete adduction, and moderate distillation temperatures to avoid dissociating the adducts. To obtain a smaller olefin composition recovery and highly concentrate the linear alpha olefins in the linear alpha olefins composition, any one or a combination of the following variables are adjusted: moderate and close to 1:1 molar ratio of linear polyaromatic compounds to olefins in the feedstock and shorter residence times to selectively adduct the linear alpha olefins in the feedstock. In either case, however, the concentration of the linear alpha olefin, or any other desired species, is enriched in the ultimate stream compared to the concentration of linear alpha olefin, or the other desired species, in the preceding composition treated and in the feedstock.

The rate of olefin recovery from the feedstock is not limited, and generally will depend upon the amount of olefin present in the feedstock. In one embodiment, the rate of recovery of olefin adducts from the first separation vessel, in moles/unit time, range from 0.10 to 0.40, more preferably from 0.15 to 0.35, each based upon a feedstock rate of 1.00. At these rates, from 45% to 100% of the olefins in the feedstock may be recovered into the olefin composition. In another embodiment, the rate of recovery ranges from 0.20 to 0.30, based upon a feedstock rate of 1.00.

In general, when the desired product is a highly concentrated linear alpha olefin, from 50% to 70% of the linear alpha olefins in the feedstock may be recovered into the olefin composition, and when the emphasis is on quantity of linear alpha olefin with a slight reduction in linear alpha olefin concentration in the linear alpha olefin stream, the recovery of olefins from the feedstock ranges from 70% to 100%. As noted above, in either case, the concentration of the desired species in the ultimate stream will be enriched over the concentration of the desired species in preceding feeds.

The rate of recovery of linear alpha olefin adducts in the separation vessel of step o(iii) is also variable and not limited. In general, the percentage of linear alpha olefin recovered from the olefin composition is set such that from a total of 30 to 60% linear alpha olefins are recovered into the linear alpha olefin stream, based upon the amount of linear alpha olefin present in the feedstock. If emphasis is placed upon recovering larger amounts of linear alpha olefin into the linear alpha olefin stream, the percentage of linear alpha olefin recovered from the olefin composition is set such that from a total of greater than 60% to 95% linear alpha olefins based upon the amount

of alpha olefins in the feedstock are recovered into the linear alpha olefin stream.

Based on the above as an example of optimizing the concentration or the quantity of linear alpha olefins in a linear alpha olefin stream, those of ordinary skill can set the rates of recovery and percentage of a desired species recovered at each separation step to optimize the concentration or quantity of other species in the feedstock one desires to recover.

The next step o(i) of the instant process is to dissociate the linear polyaromatic compound-olefin adduct in a dissociation zone. The dissociation process can be accomplished by feeding the adducted olefin stream to a dissociation vessel where the adducted olefin stream is heated and pyrolyzed at a temperature of from 200° to 500°C, preferably from 300° to 350°C, for a time sufficient to dissociate the adducts. The pyrolysis frees the olefins from the linear polyaromatic compound. One or more dissociation vessels may be used in series to conduct the dissociation, and the dissociation vessels may also be operated under a partial vacuum up to superatmospheric pressures.

The linear polyaromatic compound is optionally, but preferably, separated from the resulting mixture by any conventional means, which may occur simultaneously with the pyrolysis operation, such as by vacuum or flash distilling off the olefins along with any impurities at the pyrolysis temperatures, and removing the linear polyaromatic compound as a bottoms from the dissociation zone. The dissociation vessel is operated under slight vacuum to lower the boiling point of the dissociated linear alpha olefin and at a temperature sufficient to dissociate the adduct. Other separation techniques include filtration and centrifugation. Alternatively, the dissociated composition

comprising the olefin composition and the linear
polyaromatic compound may be recycled and fed to the first
adducting zone and/or the AO adducting reaction zone
wherein the dissociated linear polyaromatic compounds
5 become a source of the linear polyaromatic compounds used
for the adducting reaction in the first adducting zone
and/or the AO reaction zone, optionally with a fresh source
of linear polyaromatic compound derived from linear
polyaromatic compounds obtained from separation operations
10 elsewhere in the process or from virgin stock.

The olefin composition, whether separated or in mixture
with the dissociated linear polyaromatic compounds, is now
enriched in the concentration of olefins over the
concentration of olefins in the feedstock. Since the
15 linear polyaromatic compound exhibits a preference towards
adducting with linear alpha olefins, the linear alpha
olefin concentration in the olefin composition is enriched
over the concentration of linear alpha olefins present in
the feedstock, based on the weight of all ingredients in
20 the feedstock and the olefin composition. In the case that
the feedstock comprises branched olefins, the concentration
of branched olefins in the olefin composition may be
reduced over the concentration of branched olefins in the
feedstock. In the event that the olefin composition is not
25 separated from the linear polyaromatic compounds prior to
feeding the olefin composition to the AO adducting reaction
zone, the concentration of the ingredients in the olefin
composition exclusive of the weight and amount of linear
polyaromatic compounds, is enriched over the concentration
30 of olefins in the feedstock composition. Further, the
concentration of saturated hydrocarbons and the
concentration of paraffins in the olefin composition is
reduced over that of the feedstock.

Likewise, when the saturated hydrocarbons are separated
35 from the linear polyaromatic compound-olefin adduct in the

separation vessel as a saturated hydrocarbon stream, the saturated hydrocarbon stream is enriched in its concentration of saturated hydrocarbons over the concentration of saturated hydrocarbons in the feedstock to the adduct reaction zone, and the concentration of olefins in the saturated hydrocarbon stream is reduced over the concentration of olefins in the feedstock entering the adduct reaction zone. The saturated hydrocarbon stream may be recovered and isolated for use into other applications or as an intermediate in other reactive processes.

In the next step of the process, step o(ii), the olefin composition is contacted with linear polyaromatic compounds in an AO reaction zone under conditions effective to form a reaction mixture comprising linear polyaromatic compound-linear alpha olefin adducts and internal olefins. Suitable reaction conditions and vessels include those used in the adducting reaction zone for the feedstock. Since the olefin composition used as the feed is essentially, if not completely, free of saturated hydrocarbons which would otherwise have an effect of diluting a feedstock, the conversion of olefins in the olefin composition toward linear polyaromatic compound-linear alpha olefin adducts is higher than the conversion of the feedstock toward linear polyaromatic compound-linear alpha olefin adducts. The preferential selectivity of the linear polyaromatic compound toward linear alpha olefins makes possible the separation between the linear alpha olefins and other species in the olefin composition, such as linear internal olefins, branched internal olefins, and branched alpha olefins.

Once the linear polyaromatic compound-linear alpha olefin adduct has been formed in the AO reaction zone in step o(ii), the adduct stream flows to a separation vessel in step o(iii) effective for separating the internal olefins and other unreacted olefins from the linear

polyaromatic compound-linear alpha olefin adducts to form an internal olefin stream and an alpha olefin adducted stream. Suitable methods and conditions for separating the adducts from the reaction mixture include any of the methods used to remove the adducted olefins from the reaction mixture in the first separation zone. Preferably, the reaction mixture is distilled and the internal olefin stream is removed at the overhead of the distillation column, while the linear alpha olefin adducts are removed from the column as a liquid bottoms stream. The separated unreacted internal olefin stream distillate includes linear internal olefins, branched internal olefins, and branched alpha olefins. The concentration of the linear internal olefins, branched internal olefins, and branched alpha olefins in the internal olefin stream is enriched over the concentration of these olefins in the olefin composition and in the feedstock.

The linear polyaromatic compound-linear alpha olefin adducts in the linear alpha olefin adduct stream removed from step o(iii) are dissociated in step o(iv) in a dissociation zone to form linear polyaromatic compounds and a linear alpha olefin composition. Suitable methods and conditions for dissociating the adducts in the linear alpha olefin adduct stream include any of the methods mentioned as suitable for dissociating the adducts in the adducted olefin stream. This stream is enriched in the concentration of linear alpha olefin over the concentration of linear alpha olefin in the olefin composition. The linear alpha olefin composition comprises a high concentration of linear alpha olefins, and minor amounts of other olefins such as linear internal olefin, branched internal olefin, and branched alpha olefins. The concentration of these other olefins are reduced in the linear alpha olefin composition over the concentration of these other olefins in the olefin composition.

Optionally, and preferably, in step o(v), the olefin composition is separated and isolated from the dissociated linear polyaromatic compounds. The linear alpha olefin composition at this step, or at the last step of dissociation in the event that multiple addition/separation/dissociation steps are carried out, is removed from the dissociation vessel to form a linear alpha olefin stream. The linear alpha olefin may be removed from the dissociation vessel through the overhead of a cracking vessel operated under slight vacuum and at a temperature sufficient to vaporize the linear alpha olefin adduct and dissociate the adduct. In a more preferred embodiment, the removal of linear alpha olefin is carried out in the same vessel as used to contain the dissociation reaction in step o(iv).

The saturated hydrocarbon stream formed in step (b) may be subjected to a second and repeating adducting reactions with linear polyaromatic compounds, separations, and dissociations to further enrich the concentration of saturated hydrocarbons and separate out any olefins, such as internal olefins, which may have been present in the saturated hydrocarbon stream. The internal olefins separated from the saturated hydrocarbons may be combined with the internal olefins separated from the olefin composition. Thus, the overall process of the present invention recovers and isolates a concentrated saturated hydrocarbon stream, a linear alpha olefin stream, and an internal olefin stream from a crude feedstock comprising a mixture of saturated hydrocarbons, linear alpha olefins, and internal olefins.

The concentration of each desired species increases with each successive addition, separation, and dissociation step in the process, and the concentration of each undesired species in a particular stream decreases with each successive addition, separation, and

dissociation step in the process. For purposes of measuring the percentage reduction of a species in a stream, the concentration (all concentrations determined on the basis of the total weight of all ingredients present in the stream in question) of the species or series of species in question contained in the product stream is subtracted from the concentration of the species or series of species in question contained in the predecessor stream in question, the difference then divided by the concentration of the same species in the feedstock multiplied by 100. For purposes of measuring the % enrichment of a species in a stream, the concentration of the species or series of species in the predecessor or feedstock stream is subtracted from the concentration of species or series of species in question contained in the product stream, the difference then divided by the concentration of those same species present in the feedstock stream and multiplied by 100. For purposes of adding together a series of species, the sum total of the series in the predecessor stream is added, and then the sum total of the species in the product stream are added. The total in the product stream is then compared to the total in the predecessor stream to determine whether the total of the series in the product stream was enriched or reduced over the sum total in the predecessor stream. The appropriate calculation mentioned above is then applied depending on whether the series in the product stream were reduced or enriched.

The process of the present invention will enrich the total concentration of olefins, enrich the concentration of linear alpha olefin, and reduce the concentration of saturated hydrocarbons, each in the olefin composition over the concentration of all olefins, linear alpha olefin and saturated hydrocarbons contained in the feedstock stream. The process of the present invention will also enrich the concentration of saturated hydrocarbons present in the

saturated hydrocarbon stream over the concentration of saturated hydrocarbon in the feedstock stream. The process of the present invention will enrich the concentration of linear alpha olefin and reduce the concentration of internal olefins in the linear alpha olefin composition over the concentration of linear alpha olefins and internal olefins present in the olefin composition.

The concentration of all olefins in the saturated hydrocarbon stream are preferably reduced through the process of the invention in only one pass by at least 15%, more preferably at least 30%, most preferably at least 40%, over the concentration of all the olefins in the feedstock.

Since the linear polyaromatic compound is more selective towards adducting with linear alpha olefins relative to other olefins, the concentration of linear alpha olefins in the saturated hydrocarbon stream is preferably reduced in one pass by at least 30%, more preferably by at least 40%, most preferably by at least 50%, over the concentration of linear alpha olefins present in the feedstock stream.

The amount of excess linear polyaromatic compound present in the adducting reaction zone, the residence time, and temperature, will affect the amount of internal or branched olefins adducting with the linear polyaromatic compound, and therefore, the amount of internal or branched olefins left unreacted and passing into the saturated hydrocarbon stream. While the linear polyaromatic compound preferentially adducts with a linear alpha olefin, the presence of a large excess of the polyaromatic compound relative to the amount of linear alpha olefins present in the feedstock coupled with long residence times, will leave un-adducted linear polyaromatic compounds free to adduct with the internal and branched olefins, thereby enhancing the reduction of these olefins in the saturated hydrocarbon stream over the concentration of these olefins in the

feedstock stream. The concentration of internal olefins present in the saturated hydrocarbon stream is preferably reduced by from 1 to 50% over the concentration of linear internal olefins present in the feedstock.

5 With respect to the concentration of branched olefins, their concentration in the saturated hydrocarbon stream generally ranges from a slight reduction to an enrichment relative to the concentration of the branched olefins present in the feedstock. The concentration of branched
10 olefins may be reduced by only 1 to 30%, or enriched by 1 to 30%, or remain unchanged, over the concentration of these branched olefins present in the feedstock.

The concentration of saturated hydrocarbon in the saturated hydrocarbon stream is enriched over the
15 concentration of saturated hydrocarbon in the feedstock stream. The concentration is preferably enriched by at least 5%, more preferably by at least 20%, and can be enriched by 100-400%, especially when the concentration of saturated hydrocarbon in the feedstock is low. Generally,
20 the degree of enrichment of saturated hydrocarbon in the saturated hydrocarbon stream varies inversely with the concentration of the saturated hydrocarbons in the particular feedstock employed.

The concentration of saturated hydrocarbons in the
25 olefin composition is preferably reduced through the process of the invention in only one pass by at least 80%, more preferably by at least 90%, most preferably by at least 95% over the concentration of saturated hydrocarbon in the feedstock, and most preferably by 100%.

30 As above, the percentage reduction or enrichment of branched olefins and internal olefins in the olefin composition depends upon the amount of linear polyaromatic compound, temperature, and residence time of the feedstock in the adducting reaction zone. The concentration of
35 branched olefins in the olefin composition is preferably

reduced over the concentration of branched olefins in the feedstock.

The concentration of linear alpha olefins in the olefin composition is enriched over the concentration of linear alpha olefins present in the feedstock stream. The concentration of linear alpha olefins present in the olefin composition is preferably enriched by at least 30%, more preferably by at least 40%, most preferably by at least 60%, over the concentration of linear alpha olefins present in the feedstock composition.

The concentration of all olefins in the olefin composition is enriched over the concentration of all olefins in the feedstock stream. The degree of olefin enrichment varies inversely with the concentration of olefins present in the feedstock. Preferably, the concentration of all olefins in the olefin composition is enriched by at least 40%, preferably by at least 60%.

The process of the present invention is capable of separating olefins from saturated hydrocarbons in a feedstock consisting essentially of saturated hydrocarbons and olefins, resulting in a concentration of olefins in the olefin composition ranging from 90% to 100%.

The concentration of linear alpha olefins in the linear alpha olefin composition is enriched over the concentration of linear alpha olefins in the olefin composition and in the feedstock. In one embodiment, the concentration of linear alpha olefins in the linear alpha olefin composition is enriched over the concentration of linear alpha olefins in the olefin composition by at least 15%, more preferably by at least 20%, most preferably by at least 30%. The concentration of all other olefins in the linear alpha olefin stream are reduced, collectively, by at least 20%, more preferably by at least 30%, most preferably by at least 40%, over the concentration of all other olefins collectively in the olefin composition. Specifically, the

concentration of branched olefins in the linear alpha olefin stream may be reduced by 60%, more preferably by 75%, and as high as 90% over the concentration of the branched olefins in the feedstock and olefin composition.

5 The concentration of internal olefins in the internal olefin composition is enriched over the concentration of internal olefins in the olefin composition and in the feedstock. The concentration of internal olefins in the internal olefin composition is preferably enriched over the
10 concentration of internal olefins in the olefin composition by at least 10%, more preferably by at least 15%, and generally up to 40%. The concentration of branched olefins in the internal olefin composition is preferably enriched by at least 30%, more preferably by at least 50%, most
15 preferably by at least 70%. The concentration of linear alpha olefins in the internal olefin stream is preferably reduced by at least 20%, more preferably by at least 30% over the concentration of linear alpha olefin in the olefin composition.

20 The present invention will now be described by way of example with reference to the single figure of the accompanying drawing, which is a block flow diagram representing the process of adducting/separating/dissociating a crude feedstream of olefin and saturated
25 hydrocarbon and subjecting the separated olefin stream to a further adducting/separation/dissociation treatment to obtain a purified linear alpha olefin stream.

30 Block 1 represents the first adduction zone, separation zone, and dissociation zone. Block 2 represents the AO adduction zone, separation zone, and dissociation zone. Line 1 represents the composition of the feedstock, Line 2 represents the composition of the saturated hydrocarbon stream, Line 3 represents the olefin composition stream, Line 4 represents the internal olefin composition stream,

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and Line 5 represents the composition of the linear alpha olefin stream.

The mass balances tabulated below illustrate as a model one of the embodiments of the present invention wherein the recovery of a high concentration of linear alpha olefins in the linear alpha olefin stream is desirable. Table A tabulates the mass balance based upon the quantity of each species in a feed and product stream, while Table B presents a mass balance based upon the concentration of each species in a feed and product stream. The mass balances are on a calculated basis to illustrate the concept of the present invention, and are based upon the use of anthracene as the linear polyaromatic compound and upon the assumptions noted below Table B. Table A results are reported as moles/unit time, and Table B results are reported as a mole percent composition in each stream. The assumptions upon which the models are based are reported beneath each table.

Table A

	1	2	3	4	5
Paraffins (linear/branched)	0.15	0.15	0.00	0.00	0.00
Saturated alkyl aromatics	0.15	0.15	0.00	0.00	0.00
Saturated oxygenates	0.15	0.15	0.00	0.00	0.00
Linear alpha olefins	0.20	0.06	0.14	0.06	0.08
Linear 2-olefins	0.10	0.05	0.05	0.03	0.02
2-methyl 1-olefins	0.25	0.22	0.03	0.02	0.00
Total	1.00	0.79	0.21	0.11	0.10

Table B

	1	2	3	4	5
Paraffins (linear/branched)	15%	19%	0%	0%	0%
Saturated alkyl aromatics	15%	19%	0%	0%	0%
Saturated oxygenates	15%	19%	0%	0%	0%
Linear alpha olefins	20%	8%	66%	51%	82%
Linear 2-olefins	10%	7%	22%	27%	16%
2-methyl 1-olefins	25%	28%	12%	22%	2%

Assumptions:

Block 1 recovery set at 70%. Block 2 recovery set at 60%.
 5 Equilibrium is assumed at each stage. Assume the equipment
 is capable of perfect rejection of saturated hydrocarbons,
 aromatics, and oxygenates. The ratio of equilibrium
 constants between linear alpha olefins and linear 2-olefins
 is set at 2.7. The ratio of equilibrium constants between
 10 linear alpha olefins and 2-Methyl 1-Olefins is set at 20.
 In block 1, the percentage of linear 2-olefin extracted
 from block 1 is 46%, and of the 2-methyl 1-olefin is 10%.
 In block 2, 36% of the linear 2-olefins are extracted, and
 7% of the 2-methyl 1-olefin are extracted, each based upon
 15 the amount of material entering block 2.

The mass balances tabulated below illustrate another
 embodiment of the present invention wherein the recovery of
 higher quantities, albeit at lower concentrations relative
 to the embodiment above, of linear alpha olefins in the
 20 linear alpha olefin stream is desirable. Table C tabulates
 the mass balance based upon the quantity of each species in
 a feed and product stream, while Table D presents a mass
 balance based upon the concentration of each species in a
 feed and product stream. The mass balances are on a
 25 calculated basis to illustrate the concept of the present
 invention, and are based upon the use of anthracene as the
 linear polyaromatic compound and upon the assumptions noted
 below Table D.

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Table C

	1	2	3	4	5
Paraffins (linear/branched)	0.15	0.15	0.00	0.00	0.00
Saturated alkyl aromatics	0.15	0.15	0.00	0.00	0.00
Saturated oxygenates	0.15	0.15	0.00	0.00	0.00
Linear alpha olefins	0.20	0.03	0.17	0.04	0.13
Linear 2-olefins	0.10	0.03	0.07	0.03	0.04
2-methyl 1-olefins	0.25	0.19	0.06	0.05	0.01
Total	1.00	0.71	0.29	0.12	0.17

Table D

	1	2	3	4	5
Paraffins (linear/branched)	15%	21%	0%	0%	0%
Saturated alkyl aromatics	15%	21%	0%	0%	0%
Saturated oxygenates	15%	21%	0%	0%	0%
Linear alpha olefins	20%	4%	58%	35%	75%
Linear 2-olefins	10%	5%	23%	26%	21%
2-methyl 1-olefins	25%	28%	19%	39%	4%

5 Assumptions:

Block 1 recovery set at 85%. Block 2 recovery set at 75%. Equilibrium is assumed at each stage. Assume the equipment is capable of perfect rejection of saturated hydrocarbons, aromatics, and oxygenates. The ratio of equilibrium constants between linear alpha olefins and linear 2-olefins is set at 2.7. The ratio of equilibrium constants between linear alpha olefins and 2-Methyl 1-Olefins is set at 20. In block 1, the percentage of linear 2-olefin extracted from block 1 is 68%, and of the 2-methyl 1-olefin is 22%. In block 2, 53% of the linear 2-olefins are extracted, and 13% of the 2-methyl 1-olefin are extracted, each based upon the amount of material entering block 2.

An advantage of a Fischer-Tropsch stream is that it contains a mixture of both even and odd carbon, and the process of the present invention produces a stream having

even and odd carbon number olefin species at very low to zero amount of saturated hydrocarbons, with high concentrations of linear alpha olefins. The process of the present invention can also provide a Fischer-Tropsch olefin composition having a mixture of internal olefins and/or
5 branched olefins, and linear alpha olefins with low amounts of saturated hydrocarbons.

The process of the present invention further provides a composition, preferably Fischer-Tropsch derived, comprising
10 odd and even numbered olefins, and the composition has an average carbon number ranging from C₅ to C₂₀, preferably C₆ to C₁₈, or more preferably in the C₆ to C₁₂ range, comprising:

(a) at least two linear alpha olefin species having
15 different carbon chain lengths;

(b) the two most predominant (on a mole basis) linear alpha olefin species of said at least two linear alpha olefin species are each within the range of C₅ to C₂₀, or in the case of using a C₆ to C₁₈ feedstock, within that
20 range, or in the case of using a C₆ to C₁₂ feedstock, within that range, inclusive;

(c) said two most predominant linear alpha olefin species are present in an amount of at least 20 wt%, preferably at least 30 wt.%, more preferably at least 40
25 wt.%, based on the weight of the olefins in the composition;

(d) cumulatively, the total amount of linear alpha olefins present in the composition within said range, inclusive, is at least 40 wt.%, preferably at least 60
30 wt.%, more preferably at least 70 wt.%, and even 90 wt.% or more, based on the weight of the olefins in the composition;

(e) one or more odd numbered olefins within said range present in an amount of at least 10 wt.%, preferably at

least 20 wt.%, more preferably at least 30 wt.%, and even 40 wt.% or more, cumulative;

5 (f) a cumulative amount of aromatics, saturated hydrocarbons, and oxygenates of 5 wt.% or less, preferably 2 wt.% or less, more preferably 1 wt.% or less, most preferably 0.5 wt.% or less, each based on the weight of the composition; and preferably

10 (g) 6 wt.% or less of branched olefins having branching at the C2 or C3 position relative to the most proximate double bond, more preferably 4 wt.% or less, based on the weight of the composition.

Said above mentioned composition preferably has as one of the two most predominant olefin species an odd carbon number linear alpha olefin.

15 In another embodiment of the present invention, there is provided a composition, preferably Fischer-Tropsch derived, having an average carbon number ranging from C6 to C18 comprising at least two linear alpha olefin species having different carbon chain lengths within said range, inclusive, at least 50 wt.% of linear alpha olefins, where
20 the composition has a most predominant olefin species represented by n carbon numbers, wherein the next most predominant olefin species has either n+1 or n-1 carbon numbers; and wherein said composition comprises 2 wt.% or
25 less of saturated hydrocarbons; and preferably wherein said composition has branched olefins containing branching at the C2 or C3 positions, relative to the most proximate double bond, in an amount of 6 wt.%, more preferably 4 wt.% or less, based on the weight of the composition.

30 Said composition preferably comprises 1 wt.% or less of saturated hydrocarbons. It also preferably comprises 4 wt.% or less of branched olefins. It further preferably comprises at least 70 wt.% linear alpha olefins.

The process of the present invention advantageously provides an olefin stream which is highly concentrated in olefins, wherein the concentration of olefins in the olefin composition may be at least 90% and up to 100% olefin purity in the olefin composition.

The olefin composition stream of the present invention is useful as a component in drilling fluids, to react with elemental sulphur to make sulphurized products as extreme pressure agents in metal working fluids, as a co-monomer for the polymerization of polyethylene, as an intermediate in making poly alpha olefins (PAO) used as a lubricant, as a chlorination feed to make polychlorinated hydrocarbons in PVC applications, to react with hydrogen sulphides to make primary and secondary mercaptans as pharmaceutical intermediates and as additives to modify the properties of rubber, as solvents, and as a precursor for the manufacture of plasticizer alcohols and detergent range alcohols and surfactants, which may be derivatized into detergent range sulphates or alkoxysulphates for laundry liquids and powders, dishwashing powders and liquids, bar soap, shampoo, liquid hand soap, and hard surface cleaners.

The ranges and limitations provided in the present specification and claims are those which are believed to particularly point out and distinctly claim the present invention. It is, however, understood that other ranges and limitations that perform substantially the same function in substantially the same manner to obtain the same or substantially the same result are intended to be within the scope of the present invention. The present invention will now be illustrated by reference to the following Examples.

EXAMPLE 1

A Fischer-Tropsch stream comprised of the composition set forth in Table 1 was used as a feedstock. The FT composition was derived by passing syngas over a FT

catalyst and subsequently distilling products in the boiling point range of hexyl to undecyl hydrocarbons. This composition was used as the feed. Hydrocarbons in the C₇-C₁₀ range were the most abundant.

5 0.24 moles (42.2g) of anthracene having a 95% purity and 150g of the feedstock were placed in an autoclave. The total olefin content of the charged feed was about 0.5 moles (55.9g). The autoclave was sealed and then purged with nitrogen. The autoclave was heated to 255°C for four
10 hours to form the Diels-Alder adduct between the olefin and the anthracene. The autoclave contents were stirred during heating.

 Once the reaction was complete, the autoclave was cooled to 20°C. The product mixture was transferred to a
15 glass flask and the unreacted olefin, saturated hydrocarbons, and unreacted oxygenates were removed by distillation. The material remaining in the flask consisted of some entrained saturated hydrocarbons, unreacted anthracene, and the anthracene-olefin adduct.
20 The flask and its contents were then heated to a temperature ranging from 310-350°C to dissociate the adduct to anthracene and Product A described in Table 1 below. Product A was separated and isolated from the anthracene by distillation. The compositions of each Product stream were
25 determined by gas chromatographic analysis.

 As can be seen from the data in Table 1, Product A is greatly enriched in alpha olefin content and overall olefin content over the concentration of alpha olefin and overall
30 olefin content in the feedstock stream. Product A is enriched in alpha olefin content by 205%, and in overall olefin content, was enriched by 155% $\left(\frac{[(86.5 + 8.9) - (28.3 + 9.0)]}{(9.0 + 28.3)} \times 100 \right)$.

 Further, the concentration of saturated hydrocarbon (exclusive of oxygenates) in Product A stream was greatly

reduced; by 91% . The presence of saturated hydrocarbons in Product A is due to its incomplete removal upon distillation of the unreacted material from the adduct before the dissociation step. The concentration of internal olefin in Product A stream was only marginally reduced over the concentration of internal olefin present in the feedstock.

Product B represents the saturated hydrocarbon stream at the overhead on the distillation column taken from the reaction mixture. As can be seen from Table 1, Product B is enriched in saturated hydrocarbons over the concentration of saturated hydrocarbons in the feedstock stream, by 38%. The concentration of alpha olefin in the saturated hydrocarbon stream was reduced by 67% over the concentration of alpha olefin in the feedstock.

TABLE 1
SEPARATION OF SATURATED HYDROCARBONS FROM OLEFINS

Composition	Total Weight (g)	Saturated Hydrocarbons (wt.%) ¹	Internal Olefins (wt.%)	Alpha Olefins (wt.%)	Oxygenates (wt.%)
Feedstock	150	56.7	9.0	28.3	6.0
Product A	18.2	4.6	8.9	86.5	Trace
Product B	62	78.2	7.2	9.3	5.3

¹Exclusive of oxygenates

The olefin concentration and the concentration of alpha olefin in the olefin Product A stream was further enhanced by subjecting olefin stream Product A to a second adducting cycle. Product A, containing 0.16 moles of olefin, was treated with 13.5 g (0.076 moles) of anthracene for 4 hours at 255°C in the equipment noted above. The object of using an approximate 2:1 molar ratio of olefin to anthracene was to obtain a highly purified stream of linear alpha olefins. Unreacted material was removed by distillation as an

internal olefin stream. 8.2g of Product D distillate were recovered. The bottoms of the distillation column was thermally dissociated at 310-350°C as described above. 6.7 g of the resulting Product C linear alpha olefin stream was removed from anthracene and isolated as described above.

Product C was analyzed by gas chromatography as follows: 0.4 wt.% saturated hydrocarbons, 3.3 wt.% of internal olefins, 96.3 wt.% of alpha olefins, and trace amounts of oxygenates. Thus, Product C, compared to the feedstock stream, was enriched in linear alpha olefin content by 11% over the preceding Product A olefin composition feed, and by 240% over the concentration of linear alpha olefins in the feedstock. The concentration of internal olefin in Product C, compared to the feedstock stream, was decreased by 63%, and compared to the Product A feed, was decreased by 62%.

Product D was analyzed by gas chromatography as follows: 2 wt.% saturated hydrocarbons, 16 wt.% internal olefins, and 82 wt.% linear alpha olefins.

EXAMPLE 2

In this example, an F-T stream having the composition set forth in Table 2 was treated with anthracene. This F-T stream was derived by passing syngas over an FT catalyst and subsequently distilling and collecting the products in the boiling point range of pentyl to nonyl (C₅-C₉).

0.6 moles (112g) of anthracene having a 95% purity and 96 g of the feedstock were placed in a 300 ml autoclave. The autoclave was sealed and then purged with nitrogen. The autoclave was heated to 255°C for seven hours to form the Diels-Alder adduct between the olefin and the anthracene. The autoclave contents were stirred during heating.

Once the reaction was complete, the autoclave was cooled to 20°C. The product mixture was transferred to a glass flask and the unreacted olefin, saturated hydrocarbons, and unreacted oxygenates were removed from the reaction mixture by distillation as Product Stream B. The removed Product Stream B analyzed by GLC techniques had the following composition:

Unreacted olefins: 11.1 wt.% internal olefins and 1.4 wt.% linear alpha olefins, 81.9 wt.% saturated hydrocarbons and 5.6 wt.% oxygenates.

The material remaining in the flask was heated to a temperature ranging from 300-350°C to dissociate the adduct to anthracene and Product A described in Table 2 below. Product A was separated and isolated from the anthracene by distillation. The compositions of each Product stream were analyzed by gas chromatography.

As can be seen from the data in Table 2, Product A is greatly enriched in linear alpha olefin content and overall olefin content over the concentration of alpha olefin and overall olefin content in the feedstock stream. Product A is enriched in linear alpha olefin content by 579%, and in overall olefin content, Product A was enriched by 348%. The olefin purity in Product A olefin stream was 100%. The concentration of internal olefin in Product A stream was increased over the concentration of internal olefin present in the feedstock by an amount of 163%.

TABLE 2
SEPARATION OF SATURATED HYDROCARBONS FROM OLEFINS

Composition	Total Weight (g)	Saturated Hydrocarbons (wt.%) ¹	Internal Olefins (wt.%)	Linear Alpha Olefins (wt.%)	Oxygenates (wt.%)
Feedstock	96	72	12.4 ^a	9.9	5.7
Product A	9.0	0	32.7	67.3	0

¹Exclusive of oxygenates

5 a: 66 wt.% of 2-olefins and 35 wt.% of other internal olefins

The olefin concentration and the concentration of alpha olefin in the olefin Product A stream was further enhanced by subjecting the olefin composition Product A to an AO
10 adducting cycle. 9 grams of Product A was treated with anthracene at a 2:1 molar ratio of olefin to anthracene in like manner as described above in this Example at 255°C for 7 hours. The unreacted materials were removed from the anthracene adduct mixture by distillation. 4 g of Product D
15 distillate were recovered as an internal olefin stream. The bottoms of the distillation column were dissociated at 310°C to 350°C for 1 hour in a glass flask. The dissociated anthracene was separated from the olefin composition by distillation.

2 2.9 grams of linear alpha olefin stream Product C were recovered and analyzed by gas chromatography as follows: 0 wt.% saturated hydrocarbon, 13.4 wt.% of internal olefins, 86.6 wt.% of linear alpha olefins, and 0 wt.% of oxygenates. Thus, Product C, compared to the feedstock
25 stream, was enriched in linear alpha olefin content by 774%, and was enriched by 28% in linear alpha olefin content over the preceding Product A feed. The concentration of internal olefin in Product C, compared to the feedstock stream, was decreased by 59% compared to the
30 concentration of internal olefin in the Product A feed.

Product D was analyzed as a mixture of internal and linear alpha olefins as follows:

53.4 wt.% linear alpha olefins and 46.6 wt.% of internal olefins.

5 Example 3

10 A Fischer-Tropsch stream comprised of the composition set forth in Table 1 was used as a feedstock. The FT composition was derived by passing syngas over a FT catalyst and subsequently distilling products in the boiling point range of hexyl to undecyl hydrocarbons. This composition was used as the feed. Hydrocarbons in the C7-C10 range were the most abundant.

15 0.14 moles of anthracene having a 95% purity and 62.5 g of the feedstock were placed in an autoclave. The total olefin content of the charged feed was about 0.15 moles (19.8g), for an olefin/anthracene molar ratio of 1.1:1. The autoclave was sealed and then purged with nitrogen. The autoclave was heated to 255°C for 5.6 hours to form the Diels-Alder adduct between the olefin and the anthracene.
20 The autoclave contents were stirred during heating.

25 Once the reaction was complete, the autoclave was cooled to 20°C. The product mixture was transferred to a glass flask and the unreacted olefin, saturated hydrocarbons, and unreacted oxygenates were removed by distillation as Sat.Str. 1. The composition of Sat.Str. 1 was determined by gas chromatographic analysis.

30 The material remaining in the flask consisted of some entrained saturated hydrocarbons, unreacted anthracene, and the anthracene-olefin adduct. The flask and its contents were then heated to a temperature ranging from 250-280°C to dissociate the adduct to anthracene and Olefin 1 product described in Table 3 below. Nitrogen gas was swept over the anthracene-olefin adduct during the dissociation step to facilitate olefin removal and recovery. Olefin 1 product

was separated and isolated from the anthracene by distillation. 9.3 g of Olefin 1 product was recovered, of which 8.7 grams was olefin. The composition of Olefin 1 product was determined by gas chromatographic analysis.

5 The results indicate that Sat.Str. 1 is enriched in saturated hydrocarbons (alkanes) over the concentration of saturated hydrocarbons in the feedstock stream, by 24%. The concentration of alpha olefin in the Sat.Str. 1 stream was reduced by 55% over the concentration of alpha olefin
10 in the feedstock.

Olefin 1 product is greatly enriched in alpha olefin content and overall olefin content over the concentration of alpha olefin and overall olefin content in the feedstock stream. Olefin 1 product is enriched in alpha olefin
15 content by 202%, and in overall olefin content, was enriched by 197% $\left(\frac{[(88.21 + 5.77) - (27.18 + 4.43)]}{(27.18 + 4.43)} \times 100 \right)$.

Further, the concentration of saturated hydrocarbon (alkane) in Olefin 1 stream was greatly reduced; by 95%.
20 The presence of saturated hydrocarbons in Olefin 1 product is due to its incomplete removal upon distillation of the unreacted material from the adduct before the dissociation step.

TABLE 3
SEPARATION OF SATURATED HYDROCARBONS FROM OLEFINS

Composition	Total Weight (g)	Alkanes (wt.%)	Internal Olefins (wt.%)	Alpha Olefins (wt.%)	Oxygenates (wt.%)	Unknowns (wt.%)
Feedstock	62.5	63.8	4.43	27.18	3.06	1.45
Sat.Str.1	44.3	78.25	4.6	12.23	3.01	0.91
Olefin 1	9.3	3.31	5.77	88.21	2.0	0.81

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The Olefin 1 product was also treated in the following manner to enhance the concentration of alpha olefin. 0.055 moles of anthracene having a 95% purity and 9.3 g of the Olefin 1 product were placed in an autoclave. The total
5 olefin content of the charged feed was about 0.068 moles (8.7g), for an olefin/anthracene molar ratio of 1.2:1. The autoclave was sealed and then purged with nitrogen. The autoclave was heated to 255°C for 6 hours to form the Diels-Alder adduct between the olefin and the anthracene.
10 The autoclave contents were stirred during heating. Once the reaction was complete, the autoclave was cooled to 20°C. The product mixture was transferred to a glass flask and the unreacted olefin, saturated hydrocarbons, and unreacted oxygenates were removed by distillation as
15 Internal Olefin product. The composition of Internal Olefin product was determined by gas chromatographic analysis, and reported below in Table 4.

The material remaining in the flask consisted of some unreacted anthracene and the anthracene-olefin adduct. The
20 flask and its contents were then heated to a temperature ranging from 250-280°C to dissociate the adduct to anthracene and Alpha Olefin product described in Table 4 below. Nitrogen gas was swept over the anthracene-olefin adduct during the dissociation step to facilitate olefin
25 removal and recovery. Alpha Olefin product was separated and isolated from the anthracene by distillation. 2.6 g of Alpha Olefin product was recovered. The composition of Alpha Olefin product was determined by gas chromatographic analysis.

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TABLE 4
SEPARATION OF LINEAR ALPHA OLEFINS FROM THE OLEFIN 1 PRODUCT

Composition	Total Weight (g)	Alkanes (wt.%)	Internal Olefins (wt.%)	Alpha Olefins (wt.%)	Oxygenates (wt.%)	Unknowns (wt.%)
Olefin 1 Feedstock	9.3	3.31	5.77	88.21	2.0	0.81
Alpha Olefin Comp.	2.61	0.00	2.88	96.93	0.05	0.14
Internal Olefin Comp.	1.78	8.12	12.25	78.53	0.9	0.2

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Alpha Olefin product, compared to the feedstock stream Olefin 1, was enriched in alpha olefin by 10%. The concentration of internal olefin in Alpha Olefin product, compared to the feedstock Olefin 1 stream, was decreased by 50%.

5

Internal Olefin product was enriched in internal olefins over the concentration of internal olefins in the Olefin 1 feedstock by 112%.

C L A I M S

1. A process for treating a feedstock composition comprising linear alpha olefins, internal olefins, and saturated hydrocarbons, comprising:

(a) contacting the feedstock composition with a linear polyaromatic compound in a first reaction zone under conditions effective to form a reaction mixture comprising linear polyaromatic compound-olefin adducts and saturated hydrocarbons;

(b) separating the linear polyaromatic compound-olefin adducts from the saturated hydrocarbons in the reaction mixture to form a first adducted olefin stream and a saturated hydrocarbon stream;

o(i) dissociating linear polyaromatic compound-olefin adducts in the first adducted olefin stream to form linear polyaromatic compounds and an olefin composition comprising alpha olefins and internal olefins;

o(ii) contacting the olefin composition with linear polyaromatic compounds in an alpha olefin reaction zone under conditions effective to form a reaction mixture comprising linear polyaromatic compound-linear alpha olefin adducts and an internal olefin composition;

o(iii) separating the linear polyaromatic compound-linear alpha olefin adducts from the reaction mixture obtained from the alpha olefin reaction zone to form a linear alpha olefin adducted stream and an internal olefin stream; and

o(iv) dissociating the linear polyaromatic compound-linear alpha olefin adducts in the linear alpha olefin adducted stream to form linear polyaromatic compounds and a linear alpha olefin composition;

whereby the concentration of linear alpha olefins in the linear alpha olefin composition is enriched over the concentration of linear alpha olefins in the olefin composition, and the concentration of linear alpha olefins in the olefin composition is enriched over the concentration of linear alpha olefins in the feedstock.

2. The process of claim 1, wherein the feedstock is contacted with a linear polyaromatic compound at a temperature ranging from 150° to 290°C.

3. The process of claim 1 or 2, wherein the linear polyaromatic compound-olefin adduct is dissociated by heating the linear polyaromatic compound-olefin adduct to a temperature ranging from 200°C to 500°C.

4. The process of claim 1, 2 or 3, wherein the feedstock comprises a stream derived from a Fisher-Tropsch process.

5. The process of any one of claims 1 to 4, wherein the feedstock comprises from 15 wt.% to 70 wt.% olefin, based on the weight of all ingredients in the feedstock.

6. The process of any one of claims 1 to 5, wherein the feedstock comprises from 5 wt.% to 95 wt.% paraffin, based on the weight of all ingredients in the feedstock.

7. The process of any one of claims 1 to 6, wherein the linear polyaromatic compound comprises substituted or unsubstituted anthracene and/or benzanthracene.

8. The process of any one of claims 1 to 7, wherein the total concentration of olefins and the concentration of linear alpha olefins are enriched in the olefin composition over the concentration of olefins and linear alpha olefins in the feedstock stream, and the concentration of saturated hydrocarbons are reduced in the olefin composition over the concentration of saturated hydrocarbons in the feedstock stream.

9. The process of any one of claims 1 to 8, wherein the feedstock comprises a single carbon cut composition.