

"THIXOTROPIC AQUEOUS SUSPENSIONS
CONTAINING LONG CHAIN SATURATED FATTY
ACID STABILIZERS"

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The present invention relates to thixotropic aqueous suspension with improved physical stability. More specifically the invention relates to the use of long chain fatty acids as physical stabilizers for
10 thixotropic aqueous suspensions.

The present invention specifically relates to automatic dishwashing detergent compositions having thixotropic properties, improved chemical and physical stability, and with increased apparent viscosity, and
15 which are readily dispersible in the washing medium to provide effective cleaning of dishware, glassware, china and the like.

Commercially available household-machine dishwasher detergents provided in powder form have several
20 disadvantages, e.g. non-uniform composition; costly operations necessary in their manufacture; tendency to cake in storage at high humidities, resulting in the formation of lumps which are difficult to disperse; dustiness, a source of particular irritation to users
25 who suffer allergies; and tendency to cake in the dishwasher machine dispenser.

Recent research and development activity has focussed on the gel or "thixotropic" form of such compositions, e.g. scouring cleansers and automatic-
30 dishwasher products characterised as thixotropic pastes. Dishwasher products so provided are primarily objectionable in that they are insufficiently viscous to remain "anchored" in the dispenser cup of the dish-

washer, and moreover yield spotty residues on dishware, glassware, china and the like. Ideally, thixotropic cleansing compositions should be highly viscous in a quiescent state, Bingham plastic in nature, and have relatively high yield values. When subjected to shear stresses, however, such as being shaken in a container or squeezed through an orifice, they should quickly fluidize and, upon cessation of the applied shear stress, quickly revert to the high viscosity/Bingham plastic state. Stability is likewise of primary importance, i.e. there should be no significant evidence of phase separation or leaking after long standing.

U.K. Patent Application Serial No. GB-A-2176495 (Colgate-Palmolive Co) is directed to thixotropic clay aqueous suspension dishwashing detergent compositions containing metal salts of long chain fatty acids, such as aluminium stearate as physical stabilizing agents. The application GB-A-2176495 compositions show improvement in the physical stability of the detergent composition and improvement against phase separation over those clay containing compositions that do not contain the aluminium stearate.

The provision of automatic-dishwasher compositions in gel form having the aforescribed properties, other than for the improvements described in the above mentioned application GB-A-2176495 has thus far proven problematical, particularly as regards compositions for use in home dishwasher machines. For effective use, it is generally recommended that the automatic dishwashing detergent, hereinafter also designated ADD, contain (1) sodium tripolyphosphate (NaTPP) to soften or tie up hard-water minerals and to emulsify and/or peptize

soil; (2) sodium silicate to supply the alkalinity necessary for effective detergency and to provide protection for fine china glaze and pattern; (3) sodium carbonate, generally considered to be optional, to
5 enhance alkalinity; (4) a chlorine-releasing agent to aid in the elimination of soil specks which lead to water spotting; and (5) defoamer/surfactant to reduce foam, thereby enhancing machine efficiency and supplying requisite detergency. See, for example, SDA
10 Detergents in Depth, "Formulations Aspects of Machine Dishwashing", Thomas Oberle (1974). Cleansers approximating to the aforescribed compositions are mostly liquids or powders. Combining such ingredients in a gel form effective for home-machine use has proved
15 difficult. Generally, such compositions omit hypochlorite bleach, since it tends to react with other chemically active ingredients, particularly surfactant. Thus, U.S. Patent No. 4,115,308 discloses thixotropic automatic dishwasher pastes containing a suspending
20 agent, e.g. CMC, synthetic clays or the like; inorganic salts including silicates, phosphates and polyphosphates; a small amount of surfactant and a suds depressor. Bleach is not disclosed. U.S. Patent No. 4,147,650 is somewhat similar, optionally including Cl-
25 (hypochlorite) bleach but no organic surfactant or foam depressant. The product is described, moreover, as a detergent slurry with no apparent thixotropic properties.

U.S. Patent No. 3,985,668 describes abrasive
30 scouring cleansers of gel-like consistency containing (1) suspending agent, preferably the Smectite and attapulgite types of clay; (2) abrasive, e.g. silica sand or perlite; and (3) filler comprising light

density powdered polymers, expanded perlite and the like, which has a bouyancy and thus stabilizing effect on the composition in addition to serving as a bulking agent, thereby replacing water otherwise available for undesired supernatant layer formation due to leaking and phase destabilization. The foregoing are the essential ingredients. Optional ingredients include hypochlorite bleach, bleach stable surfactant and buffer, e.g. silicates, carbonates, and monophosphates. Builders, such as NaTPP, can be included as further optional ingredients to supply or supplement building function not provided by the buffer, the amount of such builder not exceeding 5% of the total composition, according to the patent. Maintenance of the desired (greater than) pH 10 levels is achieved by the buffer/builder components. High pH is said to minimize decomposition of chlorine bleach and undesired interaction between surfactant and bleach. Foam killer is not disclosed.

In U.K. Patent Application GB 2,116,199A and GB 2,140,450A, both of which are assigned to Colgate-Palmolive, liquid ADD compositions are disclosed which have properties desirably characterising thixotropic, gel-type structure and which include each of the varioyus ingredients necessary for effective detergency with an automatic dishwasher. The normally gel-like aqueous automatic dishwasher detergent composition having thixotropic properties includes the following ingredients, on a weight basis:

- (a) 5 to 35% alkali metal tripolyphosphate;
- (b) 2.5 to 20% sodium silicate;
- (c) 0 to 9% alkali metal carbonate;
- (d) 0.1 to 5% chlorine bleach stable, water

dispersible organic detergent active material;

(e) 0 to 5% chlorine bleach stable foam depressant;

(f) chlorine bleach compound in an amount to
5 provide about 0.2 to 4% of available chlorine;

(g) thixotropic thickener in an amount sufficient
to provide the composition with thixotropy index of
about 2.5 to 10;

(h) sodium hydroxide, as necessary, to adjust pH;
10 and

(i) balance water.

ADD compositions so formulated are low-foaming;
are readily soluble in the washing medium and most
effective at pH values best conducive to improved
15 cleaning performance, viz, pH 10.5-14. The
compositions are normally of gel consistency, i.e. a
highly viscous, opaque jelly-like material having
Bingham plastic character and thus relatively high
yield values. Accordingly, a definite shear force is
20 necessary to initiate or increase flow, such as would
obtain within the agitated dispenser cup of an
energized automatic dishwasher. Under such conditions,
the composition is quickly fluidized and easily
dispersed. When the shear force is discontinued, the
25 fluid composition quickly reverts to a high viscosity,
Bingham plastic state closely approximating its prior
consistency.

U.S. Patent 4,511,487 dated April 16, 1985
describes a low-foaming detergent paste for dish-
30 washers. The patented thixotropic cleaning agent has a
viscosity of at least 30 Pa.s at 20°C as determined
with a rotational viscometer at a spindle speed of 5
revolutions per minute. The composition is based on a

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mixture of finely divided hydrated sodium metasilicate, an active chlorine compound and a thickening agent which is a foliated silicate of the hectorite type. Small amount of nonionic tensides and alkali metal carbonates and/or hydroxides may be used.

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The formation of organoclays by the interaction of clays (such as bentonite and hectorite) with organic compounds such as quaternary ammonium salts, has also been described (W.S. Mardis, JAOCS, Vol. 61, No. 2, p.382 (1984)).

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While these previously disclosed liquid ADD formulations are not subject or are subject to a lesser degree to one or more of the above described deficiencies, it has been found that further improvements in physical stability at lower costs are desired to increase the shelf-life of the product and thereby enhance consumer acceptance.

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At the same time it would be highly desirable to increase the physical stability of other clay based thixotropic liquid formulations, such as scouring cleansers; dental pastes, "liquid" soaps, and the like.

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Accordingly, it is an objective of the invention to provide anti-settling additives for thixotropic clay aqueous suspensions.

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It is another object of the invention to provide liquid ADD compositions having thixotropic properties with improved physical stability and rheological properties at lower costs by using fatty acids in place of the more expensive metal salts of fatty acids.

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It is still another object of the invention to provide thixotropic liquid ADD compositions having reduced levels of thixotropic thickener without adversely effecting the generally high viscosities at

low shear rates and lower viscosities at high shear rates which are characteristic of the desired thixotropic properties.

5 More broadly, it is an object of this invention to improve the stability of aqueous thixotropic compositions, especially liquid automatic dishwasher detergent pastes or gels, by incorporating in the aqueous suspension a minor amount of a fatty acid effective to increase the apparent viscosity of the
10 formulation and to inhibit the settling of the suspended particles and to prevent phase separation.

These and other objects of the invention which will become more readily understood from the following detailed description of the invention and preferred
15 embodiments thereof are achieved by incorporating in a normally gel-like aqueous liquid composition a small but effective amount of a physical stabilizer which is a long chain fatty acid. More particularly, according to a preferred and specific embodiment of the
20 invention, there is provided a normally gel-like automatic dishwasher detergent composition in which is incorporated an amount of a long chain fatty acid which is effective to increase the apparent viscosity of the formulation and to inhibit settling of the suspended
25 particles, such as thixotropic agent.

In accordance with this particular aspect, the present invention provides a normally gel-like aqueous automatic dishwasher detergent composition having thixotropic properties which include, on a weight
30 basis:

- (a) 5 to 35% alkali metal tripolyphosphate;
- (b) 2.5 to 20% sodium silicate;
- (c) 0 to 9% alkali metal carbonate;

(d) 0.1 to 5% chlorine bleach stable, water dispersible organic detergent active material;

(e) 0 to 5% chlorine bleach stable foam depressant;

5 (f) chlorine bleach compound in an amount to provide about 0.2 to 4% of available chlorine;

(g) thixotropic thickener in an amount sufficient to provide the composition with thixotropy index of about 2 to 10; and

10 (h) 0 to 8% sodium hydroxide;

(i) a long chain fatty acid in an amount effective to increase apparent viscosity and the physical stability of the composition; and

(j) balance water.

15 Also related to this specific aspect, the invention provides a method for cleaning dishware in an automatic dishwashing machine with an aqueous wash bath containing an effective amount of the liquid automatic dishwasher detergent (LADD) composition as described
20 above. According to this aspect of the invention, the LADD composition can be readily poured into the dispensing cup of the automatic dishwashing machine and will, within just a few seconds, promptly thicken to its normal gel-like or pasty state to remain securely
25 within the dispensing cup until shear forces are again applied thereto, such as by the water spray from the dishwashing machine.

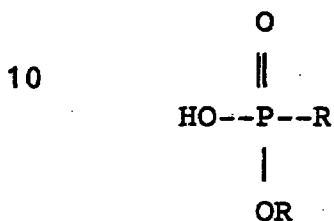
Generally, LADD effectiveness is directly related to (a) available chlorine levels; (b) alkalinity; (c)
30 solubility in washing medium; and (d) foam inhibition. It is preferred herein that the pH of the LADD composition be at least about 9.5, more preferably from about 10.5 to 14.0 e.g. 10.5 to about 13.5 and most preferably at least about 11.5.

The presence of carbonate is also often needed herein, since it acts as a buffer helping to maintain the desired pH level. Excess carbonate is to be avoided, however, since it may cause the formation of needle-like crystals of carbonate, thereby impairing the stability, as well as impairing the dispensibility of the product from, for example, squeeze tube bottles. Caustic soda (NaOH) serves the further function of neutralizing the phosphoric or phosphonic acid ester foam depressant when present. About 0.5 to 6 wt% of NaOH and about 2 to 9 wt% of sodium carbonate in the LADD composition are typical, although it should be noted that sufficient alkalinity may be provided by the NaTPP and sodium silicate.

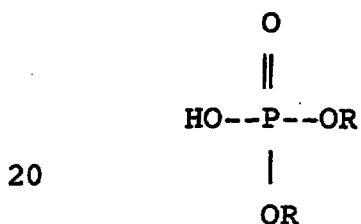
The NaTPP employed in the LADD composition in a range of about 8 to 35 wt%, preferably about 20 to 30 wt%, should preferably be free of heavy metal which tends to decompose or inactivate the preferred sodium hypochlorite and other chlorine bleach compounds. The NaTPP may be anhydrous or hydrated, including the stable hexahydrate with a degree of hydration of 6 corresponding to about 18% by weight of water or more. Especially preferred LADD compositions are obtained, for example, when using a 0.5:1 to 2:1 weight ratio of anhydrous to hexahydrated NaTPP, values of about 1:1 being particularly preferred.

Foam inhibition is important to increase dishwasher machine efficiency and minimize destabilizing effects which might occur due to the presence of excess foam within the washer during use. Foam may be sufficiently reduced by suitable selection of the type and/or amount of detergent active material, the main foam-producing component. The degree of foam is also

somewhat dependent on the hardness of the wash water in the machine whereby suitable adjustment of the proportions of NaTPP which has a water softening effect may aid in providing the desired degree of foam inhibition. However, there may optimally be included a chlorine bleach stable foam depressant or inhibitor where a low foam LADD is desired. Particularly effective are the alkyl phosphonic acid esters of the formula



available for example from BASF-Wyandotte (PCUK-PAE), and especially the alkyl acid phosphate esters of the formula



available, for example, from Hooker (SAP) and Knapsack (LPKn-158), in which one or both R groups in each type of ester may represent independently a C₁₂₋₂₀ alkyl group. Mixtures of the two types, or any other chlorine bleach stable types, or mixtures of mono- and di-esters of the same type, may be employed. Especially preferred is a mixture of mono- and di-C₁₆₋₁₈ alkyl acid phosphate esters such as monostearyl/di-stearyl acid phosphates 1.2/1 (Knapsack) or 4/1 (UGINE KULHPLAN). When employed, proportions of 0.1 to 5 wt%, preferably about 0.1 to 0.5 wt%, of foam depressant in the composition is typical, the weight ratio of

detergent active component (d) to foam depressant (e) generally ranging from about 10:1 to 1:1 and preferably about 5:1 to 1:1. Other defoamers which may be used include, for example, the known silicones. In addition, it is an advantageous feature of this invention that many of the stabilizing long chain fatty acids, such as stearic acid and behenic acid also act as foam killers.

Although any chlorine bleach compound may be employed in the compositions of this invention, such as dichloro-isocyanurate, dichloro-dimethyl hydantoin, or chlorinated TSP, alkali metal, e.g. potassium lithium, magnesium and especially sodium, hypochlorite is preferred. The composition should contain sufficient chlorine bleach compound to provide about 0.2 to 4.0% by weight of available chlorine, as determined, for example, by acidification of 100 parts of the composition with excess of hydrochloric acid. A solution containing about 0.2 to 4.0% by weight of sodium hypochlorite contains or provides roughly the same percentage of available chlorine. About 0.8 to 1.6% by weight of available chlorine is especially preferred. For example, sodium hypochlorite (NaOCl) solution of from about 11 to about 14% available chlorine in amounts of about 3 to 20%, preferably about 7 to 12%, can be advantageously used.

The sodium silicate, which provides alkalinity and protection of hard surfaces, such as fine china glaze and pattern, is employed in an amount ranging from about 2.5 to 20 wt%, preferably about 5 to 15 wt%, in the composition. The sodium silicate is generally added in the form of an aqueous solution, preferably having an $\text{Na}_2\text{O}:\text{SiO}_2$ ratio of about 1:2 to 1:2.8.

Detergent active material useful herein must be stable in the presence of chlorine bleach, especially hypochlorite bleach, and those of the organic anionic, amine oxide, phosphine oxide, sulfoxide or betaine water dispersible surfactant types are preferred, the first mentioned anionics being most preferred. They are used in amounts ranging from about 0.1 to 5%, preferably about 0.3 to 2.0%. Particularly preferred surfactants herein are the linear or branched alkali metal mono- and/or di-(C₈₋₁₄) alkyl diphenyl oxide mono and/or disulphates or disulphonates, commercially available for example as DOWFAX (Registered Trademark) 3B-2 and DOWFAX 2A-1. In addition, the surfactant should be compatible with the other ingredients of the composition. Other suitable surfactants include the primary alkylsulphates, alkylsulphonates, alkylaryl-sulphonates and sec.-alkylsulphates. Examples include sodium C₁₀-C₁₈ alkylsulphates such as sodium dodecylsulphate and sodium tallow alcoholsulphate; sodium C₁₀-C₁₈ alkanesulphonates such as sodium hexadecyl-1-sulphonate and sodium C₁₂C₁₈ alkylbenzenesulphonates such as sodium dodecylbenzenesulphonates. The corresponding potassium salts may also be employed.

As other suitable surfactants or detergents, the amine oxide surfactants are typically of the structure R_2R^1N-O , in which each R represents a lower alkyl group, for instance, methyl, and R^1 represents a long chain alkyl group having from 8 to 22 carbon atoms, for instance a lauryl, myristyl, palmityl or cetyl group. Instead of an amine oxide, a corresponding surfactant phosphine oxide R_2R^1PO or sulfoxide RR^1SO can be employed. Betaine surfactants are typically of the structure $R_2R^1N-R'COO^-$, in which each R represents a

lower alkylene group having from 1 to 5 carbon atoms. Specific examples of these surfactants are lauryl-dimethylamine oxide, myristyldimethylamine oxide, the corresponding phosphine oxides and sulphoxides, and the
5 corresponding betaines, including dodecyldimethylammonium acetate, tetradecyldiethylammonium pentanoate, hexadecyldimethylammonium hexanoate and the like. For biodegradability, the alkyl groups in these surfactants should be linera, and such compounds are preferred.

10 Surfactants of the foregoing type, all well known in the art, are described, for example, in U.S. Patents 3,985,668 and 4,271,030.

Thixotropic thickeners, i.e. thickeners or
suspending agents which provide an aqueous medium with
15 thixotropic properties, are known in the art and may be organic or inorganic water soluble, water dispersible or colloid-forming, and monomeric or polymeric, and should of course be stable in these compositions, e.g. stable to high alkalinity and chlorine bleach
20 compounds, such as sodium hypochlorite. Those especially preferred generally comprise the inorganic, colloid-forming clays of smectite and/or attapulgite types. These materials were generally used in amounts of about 1.0 to 10, preferably 1.2 to 5 wt%, to confer
25 the desired thixotropic properties and Bingham plastic character in the assignee's prior disclosed LADD formulations of the aforementioned GB 2,116,199A and GB 2,140,450A. It is one of the advantages of the LADD
formulations of the present invention that the desired
30 thixotropic properties and Bingham plastic character can be obtained in the presence of the fatty acid stabilizers with lesser amounts of the thixotropic thickeners. For example, amounts of the inorganic

colloid-forming clays of the smectite and/or attapul-
gite types in the range of from about 0.1 to 3%,
preferably 0.1 to 2.5%, especially 0.1 to 2%, are
generally sufficient to achieve the desired thixotropic
properties and Bingham plastic character when used in
combination with the physical stabilizer.

Smectite clays include mont morillonite
(bentonite), hectorite, attapulgite, smectite,
saponite, and the like. Montmorillonite clays are
preferred and are available under tradenames such as
Thixogel (Registered trademark) No. 1 and Gelwhite
(Registered Trademark) GP, H, etc., from Georgia Kaolin
Company; and ECCAGUM (Registered Trademark) GP, H,
etc., from Luthern Clay Products. Attapulgite clays
include the materials commercially available under the
tradename Attagel (Registered Trademark), i.e. Attagel
40, Attagel 50 and Attagel 150 from Engelhard Minerals
and Chemicals Corporation. Mixtures of smectite and
attapulgite types in weight ratios of 4:1 to 1:5 are
also useful herein. Thickening or suspending agents of
the foregoing types are well known in the art, being
described, for example, in U.S. Patent No. 3,985,668
referred to above. Abrasives or polishing agents
should be avoided in the LADD compositions as they may
mar the surface of the fine dishware, crystal and the
like.

The amount of water contained in these composi-
tions should, of course, be neither so high as to
produce unduly low viscosity and fluidity, nor so low
as to produce unduly high viscosity and low
flowability, thixotropic properties in either case
being diminished or destroyed. Such amount is readily
determined by routine experimentation in any particular

instance, generally ranging from about 30 to 75 wt%, preferably about 35 to 65 wt%. The water should also be preferably deionized or softened.

5 So far, the description of the LADD product, except as otherwise noted, conforms to the compositions as disclosed in the aforementioned U.K. Patent Applications on GB 2,116,199A and GB 2,140,450A, which are in the name of Colgate-Palmolive Co.

10 The LADD products of the prior U.K. Patent Application GB 2,116,199A and GB 2,140,450 exhibit improved rheological properties as evaluated by testing product viscosity as a function of shear rate. The compositions exhibited higher viscosity at a low shear rate and lower viscosity at a high shear rate, the data
15 indicating efficient fluidization and gellation well within the shear rates extant within the standard dishwasher machine. In practical terms, this means improved pouring and processing characteristics as well as less leaking in the machine dispenser-cup, compared
20 to prior liquid or gel ADD products. For applied shear rates corresponding to 3 to 30 rpm, viscosities (Brookfield) correspondingly ranged from about 10,000 to 30,000 cps to about 3,000 to 7,000 cps, as measured at room temperature by means of an LVT Brookfield
25 viscometer after 3 minutes using a No.4 spindle. A shear rate of 7.4 sec^{-1} corresponds to a spindle rpm of about 3. An approximate ten-fold increase in shear rate produces about a 3- to 9-fold reduction in viscosity. With prior ADD gels, the corresponding
30 reduction in viscosity was only about two-fold. Moreover, with such compositions, the initial viscosity taken at about 3 rpm was only about 2,500 to 2,700 cps. The compositions of the prior patent application thus

exhibit threshold fluidizations at lower shear rates and of significantly greater extent in terms of incremental increases in shear rate versus incremental decrease in viscosity. This property of the LADD products of the prior invention is summarized in terms of a thixotropic index (TI) which is the ratio of the apparent viscosity at 3 rpm and at 30 rpm. The prior compositions have a TI of from 2 to 10. The LADD compositions tested exhibited substantial and quick return to prior quiescent state consistency when the shear force was discontinued.

The present invention is based upon the discovery that the physical stability, i.e. resistance to phase separation, settling, etc., of the U.K. Patent Applications GB 2,116,199A and GB 2,140,450 and GB-A-2176495 liquid aqueous ADD compositions can be significantly improved or not adversely affected while at the same time significantly increasing the apparent viscosity and improving the physical stability of the formulations and at lower cost, by adding to the composition a small but effective amount of a long chain fatty acid.

As an example of the improvement in rheological properties, it has been found that the viscosities at low shear rates, e.g. at a spindle rpm of about 3, apparent viscosities may often be increased from two- to three-fold with the incorporation of as little as 0.2% or less, e.g. 0.16%, of the fatty acid stabilizer. At the same time, the physical stability may be improved to such an extent that even after a long time, the compositions containing the fatty acid stabilizers do not undergo any visible phase separation.

The preferred long chain fatty acids are the

higher aliphatic fatty acids having from about 8 to about 24 carbon atoms, more preferably from about 10 to 24 carbon atoms, and especially preferably from about 12 to 22 carbon atoms, inclusive of the carbon atom of the carboxyl group of the fatty acid. The aliphatic radical may be saturated or unsaturated and may be straight or branched. Straight chain saturated fatty acids are preferred. Mixtures of fatty acids may be used, such as those derived from natural sources, such as tallow fatty acid, coco fatty acid, soya fatty acid, etc., or from synthetic sources available from industrial manufacturing processes.

Thus, examples of the fatty acids which can be used as stabilizers include, for example, decanoic acid, dodecanoic acid, palmitic acid, myristic acid, stearic acid, behenic acid, oleic acid, eicosanoic acid, tallow fatty acid, coco fatty acid, soya fatty acid, mixtures of these acids, etc. Behenic acid, stearic acid and mixed fatty acids are preferred, with behenic acid being the most preferred.

Naturally, for LADD compositions, as well as any other applications where the invention composition will or may come into contact with articles used for the handling, storage or serving of food products or which otherwise may come into contact with or be consumed by people or animals, the use of the fatty acids as the stabilizing agent are of particular advantage because of their known low toxicity. For this purpose, the stearic acid and behenic acid are especially preferred as generally safe food additives. Another distinct advantage of the use of the fatty acids as stabilizers is their lower cost as compared to the fatty acid metal salts.

Many of these fatty acids are commercially available. For example, the stearic acid and behenic acid are readily available.

5 Mixed fatty acids, such as the naturally occurring acids, e.g. coco acid, as well as mixed fatty acids resulting from the commercial manufacturing process are also advantageously used as an inexpensive but effective source of long chain fatty acids.

10 The amount of the fatty acid stabilizers to achieve the desired enhancement of physical stability and apparent viscosity increase will depend on such factors as the nature of the fatty acid, the nature and amount of the thixotropic agent, detergent active compound, inorganic salts, especially TPP, other LADD
15 ingredients, as well as the anticipated storage and shipping conditions.

Generally, however, amounts of the fatty acid stabilizing agents in the range of from about 0.02 to 1%, preferably from about 0.06 to 0.8%, especially
20 preferably from about 0.08 to 0.4%, provide the increase in apparent viscosity and the long term stability and absence of phase separation upon standing or during transport at both low and elevated temperatures as are required for a commercially acceptable
25 product.

From the examples to be given below, it will be seen that, depending on the amounts, proportions and types of physical stabilizers and thixotropic agents, the addition of the fatty acids not only increases
30 physical stability but also provides a simultaneous increase in apparent viscosity. Ratios of fatty acid to thixotropic agent in the range of from about 0.08 to 0.4 weight percent fatty acid and from about 1.3 to 2.5

weight percent thixotropic agent are usually sufficient to provide these simultaneous benefits and, therefore, the use of these ingredients in these ratios is most preferred.

5 According to one preferred method of making these compositions, one should dissolve or disperse first all the inorganic salts, i.e. carbonate (when employed), silicate and tripolyphosphate, in the aqueous medium. Thickening agent is added last. The foam depressor
10 (when employed) is preliminarily provided as an aqueous dispersion, as is the thickening agent. The foam depressant dispersion, caustic soda (when employed) and inorganic salts are first mixed at elevated temperatures in aqueous solution (deionized water) and, there-
15 after, cooled, using agitation throughout. Bleach, surfactant, fatty acid stabilizer and thickener dispersion at room temperature are thereafter added to the cooled (25-35°C) solution. Excluding the chlorine bleach compound, total salt concentration (NaTPP, sodium silicate and carbonate) is generally about 20 to
20 50 weight percent, preferably about 30 to 40 weight percent in the composition.

Another highly preferred method for mixing the ingredients of the LADD formulations involves first
25 forming a mixture of the water, foam suppressor (when employed), detergent, physical stabilizer (fatty acid) and thixotropic agent, e.g. clay. These ingredients are mixed together under high shear conditions, preferably starting at room temperature, to form a
30 uniform dispersion. Thus this premixed portion, the remaining ingredients are introduced under low shear mixing conditions. For instance, the required amount of the premix is introduced into a low shear mixer and

thereafter the remaining ingredients are added, with mixing, either sequentially or simultaneously. Preferably, the ingredients are added sequentially, although it is not necessary to complete the addition of all of one ingredient before beginning to add the next ingredient. Furthermore, one or more of the ingredients can be divided into portions and added at different times. Good results have been obtained by adding the remaining ingredients in the following sequence: sodium hydroxide, alkali metal carbonate, sodium silicate, alkali metal tripolyphosphate (hydrated), alkali metal tripolyphosphate (anhydrous or up to 5% water), bleach (preferably, sodium hypochlorite) and sodium hydroxide.

Other conventional ingredients may be included in these compositions in small amounts, generally less than about 3 weight percent, such as perfume, hydro-tropic agents such as the sodium benzene, toluene, xylene and cumene sulphonates, preservatives, dyestuffs and pigments and the like, all of course being stable to chlorine bleach compound and high alkalinity (properties of all the components). Especially preferred for colouring are the chlorinated phthalocyanines and polysulphides of aluminosilicate which provide, respectively, pleasing green and blue tints. TiO_2 may be employed for whitening or neutralizing off-shades.

The liquid ADD compositions of this invention are readily employed in known manner for washing dishes, other kitchen utensils and the like in an automatic dishwasher, provided with a suitable detergent dispenser, in an aqueous wash bath containing an effective amount of the composition.

While the invention has been particularly described in connection with its application to liquid automatic dishwasher detergents it will be readily understood by one of ordinary skill in the art that the benefits which are obtained by the addition of the long chain fatty acids, namely increased apparent viscosity and increased physical stability of the clay based thixotropic suspension, will apply equally well to other clay based thixotropic suspensions, such as the scouring paste formulations described in the aforementioned U.S. Patent 3,985,668.

The invention may be put into practice in various ways and a number of specific embodiments will be described to illustrate the invention with reference to the accompanying examples.

All amounts and proportions referred to herein are by weight of the composition unless otherwise indicated.

EXAMPLE 1

In order to demonstrate the effect of the fatty acid stabilizer a liquid ADD formulation is prepared as follows:

	<u>Percent</u>
Deionized water	41.10
Caustic soda solution (50% NaOH)	2.20
Sodium carbonate, anhydrous	5.00
Sodium silicate, 47.5% solution of	15.74
Na ₂ O:SiO ₂ ratio of 1:2.4	
Sodium TPP (substantially anhydrous-	12.00
i.e. 0-5%, especially 3%, moisture)	
(Thermphos NW)	
Sodium TPP (hexahydrate) (Thermphos N	12.00
hexa)	

The mixture is cooled at 25-30°C and agitation maintained throughout, and the following ingredients at room temperature are added thereto:

		<u>Percent</u>
5	Sodium hypochlorite solution (11% available chlorine)	9.00
	Monostearyl phosphate	0.16
	DOWFAX 3B-2 (45% Na monodecyl/didecyl diphenyl oxide disulphonate- aqueous solution)	0.80
10	Physical stabilizer (fatty acid or fatty acid salt)	X
	Pharmagel H ¹	2.00

¹ Pharmagel H is a thixotropic agent.

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There are three formulations prepared in which X = 0%, X = 0.10% calcium stearate and X = 0.16% behenic acid.

The monostearyl phosphate foam depressant (when employed) and Dowfax 3B-2 detergent active compound
20 fatty acid stabilizer are added to the mixture just before the Pharmagel H thickener.

The Run 1 is a control formulation which includes the monostearyl phosphate anti-foam agent, but which does not contain a fatty acid stabilizer.

25 The Run 2 is a control formulation of Run 1 to which has been added a calcium stearate stabilizing agent of application GB-A-2176495.

The Run 3 is a formulation of the present invention in which behenic acid $(CH_3(CH_2)_{20}COOH)$ is used
30 as the stabilizing agent and the monostearyl phosphate anti-foam agent is optionally omitted.

Each of the resulting liquid ADD formulations as shown in the Table are measured for apparent viscosity

at 3 and 30 rpm. The results obtained are also shown in Table.

From the data reported in the Table the following conclusions are reached:

5 The incorporation of 0.1% calcium stearate in a 2.0% Pharmagel H containing formula Run 2 (control) leads to an increase in the apparent viscosity Table, Run 1 (control).

10 The incorporation of 0.16% benzoic acid in a 2% Pharmagel H containing formula Run 3 (invention) leads to a significant increase in the apparent viscosity as compared to both the control Runs 1 and 2.

TABLE

			BROOK LVT VISCOSITY	
			(KCPS)	(1)
	<u>RUN</u>	<u>FORMULATION</u>	<u>3RPM</u>	<u>30RPM</u>
		H ₂ O = 41.10%		
	1	Monostearyl Phosphate = 0.16%	18	4.9
20	(control)	Stabilizer = 0%		
		Pharmagel H = 2.0%		
		H ₂ O = 41.0%		
	2	Monostearyl Phosphate = 0.16%	24	3.8
25	(control)	Ca Stearate = 0.1%		
		Pharmagel H = 2.0%		
		H ₂ O = 41.0%		
	3	Monostearyl Phosphate = 0%	87	10.2
30	(inven-	Behenic Acid = 0.16%		
	tion)	Pharmagel H = 2.0%		

(1) Measured with spindle 4 after three minutes at 3 and 30 rpm on 24 hour old samples.

EXAMPLE 2

The following gel-like thixotropic liquid ADD is prepared following the same general procedures as in Example 1:

5	<u>Ingredient</u>	<u>Amount (A.I.) Wt%</u>
	Sodium silicate (47.5% solution of Na ₂ O:SiO ₂ ratio of 1:2.4)	7.48
	Monostearyl phosphate	0.16
10	Dowfax 3B-2	0.36
	Thermphos NW	12.0
	Thermphos N hexa	12.0
	Behenic Acid	0.1
	Sodium carbonate, anhydrous	5.0
15	Caustic soda solution (50% NaOH)	3.1
	Pharmagel H	1.5
	Sodium hypochlorite solution (11%)	1.0
	Water	balance

20 Minor amounts of perfume, colour, etc., can also be added to formulation.

EXAMPLE 3

The following gel-like thixotropic liquid ADD is prepared following the same general procedures as in Example 1:

25	<u>Ingredient</u>	<u>Amount (A.I.) Wt%</u>
	Sodium silicate (47.5% solution of Na ₂ O:SiO ₂ ratio of 1:2.4)	7.48
	Monostearyl phosphate	0.16
30	Dowfax 3B-2	0.36
	Thermphos NW	12.0
	Thermphos N hexa	12.0
	Stearic Acid	0.2

	Sodium carbonate, anhydrous	5.0
	Caustic soda solution (50% NaOH)	3.1
	Pharmagel H	1.0
	Sodium hypochlorite solution (11%)	1.0
5	Water	balance

Minor amounts of perfume, colour, etc., can also be added to formulation.

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CLAIMS

1. An aqueous thixotropic liquid composition
5 comprising a clay thixotropic agent,

an amount of at least one long chain fatty acid effective to increase the apparent viscosity and the physical stability of the composition, water, and optionally at least one additional ingredient selected
10 from the group consisting of organic detergents, pH modifying agents, chlorine bleach detergent builder, sequestering agent, foam inhibitors, abrasive particles and mixtures thereof.

15 2. A composition as claimed in Claim 1 wherein the fatty acid comprises a long chain fatty acid having from 8 to 24 carbon atoms or a mixture of two or more of such fatty acids.

20 3. A composition as claimed in Claim 1 wherein the fatty acid contains 12 to 24 carbon atoms.

4. A composition as claimed in Claim 1, 2 or 3 wherein the fatty acid comprises stearic acid or
25 behenic acid.

5. An aqueous thixotropic automatic dishwasher composition comprising approximately by weight:
(a) 5 to 35% alkali metal tripolyphosphate;
30 (b) 2.5 to 20% sodium silicate;
(c) 0 to 9% alkali metal carbonate;
(d) 0.1 to 5% chlorine bleach stable, water-dispersible organic detergent active material;



(e) 0 to 5% chlorine bleach stable foam depressant;

(f) chlorine bleach compound in an amount to provide about 0.2 to 4% of available chlorine;

5 (g) thixotropic thickener in an amount sufficient to provide the composition with a thixotropy index of about 2 to 10;

(h) 0 to 8% of sodium hydroxide;

10 (i) a long chain fatty acid as a physical stabilizer in an amount effective to increase the apparent viscosity and to increase the physical stability of the composition; and

(j) balance water.

15 6. A composition as claimed in Claim 5, wherein the physical stabilizer (i) is an aliphatic fatty acid having from 10 to 24 carbon atoms.

20 7. A composition as claimed in Claim 5 or 6, wherein the acid has from 12 to 22 carbon atoms.

8. A composition as claimed in Claim 5, 6 or 7, wherein the physical stabilizer (i) is stearic acid.

25 9. A composition as claimed in Claim 5, 6 or 7, wherein the physical stabilizer (i) is behenic acid.

30 10. A composition as claimed in any one of Claims 5 to 9, wherein the physical stabilizer (i) is present in an amount of from about 0.02 to 1%.

11. A composition as claimed in any one of Claims 5 to 9, wherein the physical stabilier (i) is present



in an amount of from about 0.06 to 0.8%.

5 12. A composition as claimed in any one of Claims 5 to 9, wherein the physical stabilizer (i) is present in an amount of from about 0.08 to 0.4%.

13. A composition as claimed in any one of Claims 5 to 12, wherein the thixotropic thickener (g) is an inorganic, colloid-forming clay.

10 14. A composition as claimed in Claim 13 wherein the clay is a montmorillonite clay, an attapulgite clay, a hectorite clay or a smectite clay.

15 15. A composition as claimed in Claim 13 or 14 wherein the amount of the clay thickener is in the range of from about 0.1 to 3%.

20 16. A composition as claimed in Claim 13, 14 or 15, wherein the amount of the clay thickener is in the range of from about 0.1 to 2.5%.

25 17. A composition as claimed in any one of Claims 13 to 16, which contains from about 0.06 to 0.8% of the physical stabilizer (i) and from about 0.1 to 2% by weight of an inorganic, colloid-forming clay as the thixotropic thickener (g).

30 18. A composition as named in any one of Claims 13 to 16, which contains from about 0.08 to 0.4% of the physical stabilizer (i) and about 0.25% to 1.0% of an inorganic colloid-forming clay as the thixotropic thickener (g).



19. A composition as claimed in any one of Claims 5 to 18, in which the chlorine bleach compound (f) is sodium hypochlorite.

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20. A composition as claimed in any one of Claims 5 to 19, which contains about 0.1 to 0.5% of the foam depressant (e).

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21. A composition as claimed in Claim 20 in which the foam depressant is an alkyl acid phosphate ester or an alkyl phosphonic acid ester containing one or two C₁₂₋₂₀ alkyl groups, or a mixture thereof.

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22. A composition as claimed in any one of Claims 5 to 21 having a pH of 10.5 to about 13.5.

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23. A method for cleaning soiled dishware in an automatic dishwashing machine which comprises contacting the soiled dishware in an automatic dishwashing machine in an aqueous washbath having dispersed therein an effective amount of a composition as claimed in any one of Claims 1 to 22.

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24. A method for improving the stability of an aqueous, gel-like thixotropic composition having a small but effective amount of a clay-thixotropic agent which comprises including in the composition long chain fatty acid having from 10 to 24 carbon atoms.

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25. An aqueous thixotropic liquid composition substantially as herein described with reference to any one of the Examples.



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26. A method for cleaning soiled dishware substantially as herein described with reference to any one of the Examples.

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27. A method for improving the stability of an aqueous, gel-like thixotropic composition substantially as herein described with reference to any one of the Examples.

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