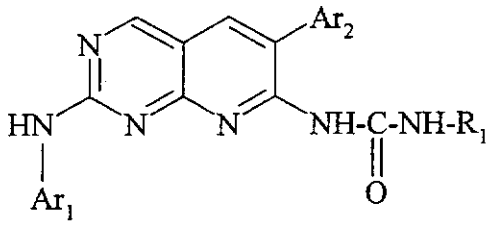
PYRIDO-PYRIMIDINE DERIVATIVES, PREPARATION THEREOF, THERAPEUTIC USE THEREOF

The present invention relates to pyrido[2,3-d]pyrimidine derivatives, to their preparation and to their therapeutic application.

Compounds derived from pyrido[2,3-d]pyrimidine have been described in International patent applications WO 01/55 147 and WO 03/000 011 and in European and United States patents EP-B-790 997 and US 5 733 913 respectively. Those compounds are of potential use in treating cellular proliferation disorders.

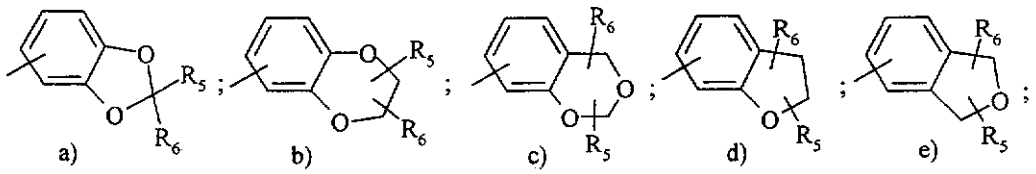
In a first aspect, the present invention concerns compounds with formula (I):



(I)

in which:

- R_1 is selected from the group constituted by (C_1-C_6)alkyl, (C_3-C_7)cycloalkyl, CH_2COR_4, phenyl or phenyl substituted with hydroxyl and/or halogen and/or (C_1-C_6)alkyl;
- R_4 represents a hydroxyl, (C_1-C_4)alkoxy, amino, (C_1-C_4)alkylamino or di(C_1-C_4)alkylamino group;
- Ar_1 represents a radical selected from:



- R5 represents a cyano, hydroxy(C1-C4)alkyl or (C1-C6)alkoxy(C1-C6)alkyl group, or a (CH2)nNR7R8, CO2R7, CONHNR7R8, CONR7R8, CONR8OR9, (CH2)nNR7COR8, (CH2)nNR7COOR8 group;

- R_6 represents a hydrogen atom, a (C_1-C_4)alkyl group or one of the values given for R_5;
- or R_5 and R_6, as defined above, are bonded together to form a cycle containing 4 to 7 chain links comprising 0 to 2 heteroatoms selected from N and O, said cycle containing 4 to 7 chain links optionally being substituted with one or more substituents independently selected from halogen, (C_1-C_4)alkyl, halogenated (C_1-C_4)alkyl, hydroxy(C_1-C_4)alkyl, (C_1-C_4)alkoxy(C_1-C_4)alkyl, $(CH_2)_mNR_7R_8$, or a *tert*-butoxycarbonyl group;
- R_7 and R_8 each independently represent a substituent selected from H, (C_1-C_4)alkyl, (C_1-C_4)alkyl-OH, (C_3-C_7)cycloalkyl, (C_3-C_7)cycloalkyl-NH_2, (C_1-C_4)alkyl-(C_3-C_7)cycloalkyl, $C(=NH)NH_2$, $SO_2(C_1-C_6)$alkyl and SO_2-phenyl; R_8 may also represent a *tert*-butoxycarbonyl or benzyloxycarbonyl group;
- or R_7 and R_8 together with the nitrogen atom to which they are bonded constitute an azetidinyl, pyrrolidinyl, piperidinyl, piperazinyl or morpholinyl radical, said radical being unsubstituted or substituted one or more times with a (C_1-C_6)alkyl, (C_1-C_4)alkyl-OH, $COO(C_1-C_6)$alkyl or F group;
- R_9 represents a hydrogen atom or a (C1-C4)alkyl group;
- Ar2 represents a phenyl group which is unsubstituted or substituted 1 to 5 times with similar or different substituents selected from a halogen atom and a (C1-C4)alkyl, trifluoromethyl or (C1-C4)alkoxy group;
- n represents 1, 2 or 3;
- m represents 0, 1, 2 or 3.

The compounds with formula (I) may comprise one or more asymmetrical carbon atoms. Thus, they may exist in the form of enantiomers or diastereoisomers. Said enantiomers, diastereoisomers and their mixtures including racemic mixtures fall within the scope of the invention.

The compounds with formula (I) may exist as bases or as addition salts with acids. When compounds with formula (I) comprise free acid functions, for example carboxylic, sulphonic or phosphonic, said acid functions may be transformed into salts using bases to form addition salts. Said addition salts fall within the scope of the invention.

The addition salts with acids or bases are advantageously prepared with pharmaceutically acceptable acids or bases respectively, but salts of other acids or bases for use, for example, in purifying or isolating compounds with formula (I) also fall within the scope of the invention.

The compounds with formula (I) may also exist in the form of hydrates or solvates, namely in the form of associations or combinations with one or more water molecules or with a solvent. Said hydrates and solvates also fall within the scope of the invention.

The following terms have the following meanings within the context of the invention:

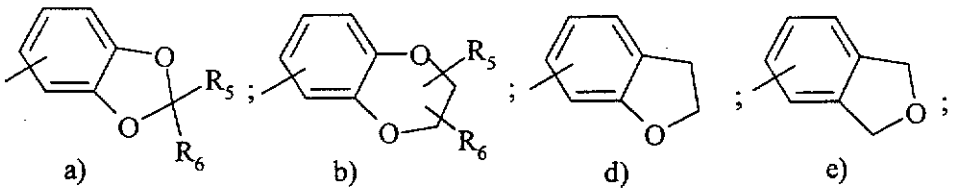
- a halogen atom: a fluorine, chlorine, bromine or iodine atom;
- an alkyl group: a linear or branched saturated aliphatic group. The following groups may be cited as examples: methyl, ethyl, *n*-propyl, *n*-butyl, *n*-pentyl, *n*-hexyl, *n*-heptyl, 1-methylethyl, 1-methylpropyl, 2-methylpropyl, 1,1-dimethylethyl, 1-methylbutyl, 2-methylbutyl, 3-methylbutyl, 1,1-dimethylpropyl, 1,2-dimethylpropyl, 2,2-dimethylpropyl, 1-methylpentyl, 2-methylpentyl, 3-methylpentyl, 4-methylpentyl, 1,1-dimethylbutyl, 1,2-dimethylbutyl, 1,3-dimethylbutyl, 2,2-dimethylbutyl, 2,3-dimethylbutyl, 3,3-dimethylbutyl, 1,1,2-trimethylpropyl, 1,2,2-trimethylpropyl, 1-ethyl-1-methylpropyl, 1-ethyl-2-methylpropyl, 1-ethylbutyl, 2-ethylbutyl, 1-methylhexyl, 2-methylhexyl, 3-methylhexyl, 4-methylhexyl, 5-methylhexyl, 1,1-dimethylpentyl, 1,2-dimethylpentyl, 1,3-dimethylpentyl, 1,4-dimethylpentyl, 2,2-dimethylpentyl, 2,3-dimethylpentyl, 2,4-

dimethylpentyl, 3,3-dimethylpentyl, 3,4-dimethylpentyl, 4,4-dimethylpentyl, 1,1,2-trimethylbutyl, 1,1,3-trimethylbutyl, 1,2,2-trimethylbutyl, 1,2,3-trimethylbutyl, 1,3,3-trimethylbutyl, 2,2,3-trimethylbutyl, 2,3,3-trimethylbutyl, 1,1,2,2-tetramethylpropyl, 1-ethylpentyl, 2-ethylpentyl, 3-ethylpentyl, 1-ethyl-1-methylbutyl, 1-ethyl-2-methylbutyl, 1-ethyl-3-methylbutyl, 2-ethyl-1-methylbutyl, 2-ethyl-2-methylbutyl, 2-ethyl-3-methylbutyl, 1-propylbutyl, 1-(1-methylethyl)butyl, 1-(1-methylethyl), 2-methylpropyl;

- a cycloalkyl group: cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cycloheptyl, bicyclo[2.2.1]heptyl, cyclooctyl, bicyclo[2.2.2]octyl, bicyclo[3.2.1]octyl, adamantyl.

Preferred compounds with formula (I) in accordance with the invention which may be cited are defined as follows:

- R_1 represents a *tert*-butyl, ethyl or phenyl group;
- and/or Ar_1 represents a radical selected from:



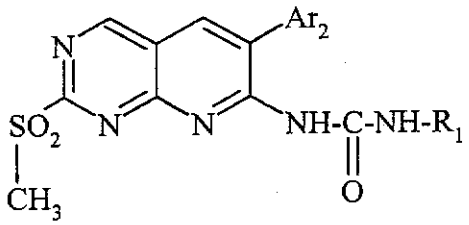
- and/or R_5 represents a $(CH_2)_nNR_7R_8$, $CONHNR_7R_8$, $CONR_7R_8$, hydroxy(C_1-C_4)alkyl or $(CH_2)_nNR_7COR_8$ group;
- and/or R6 represents a hydrogen atom or a methyl, (CH2)nNR7R8 or hydroxymethyl group;
- and/or Ar2 represents an aryl group substituted with 1 or 2 substituents independently selected from a halogen atom or a (C1-C4)alkyl or (C1-C4)alkoxy group;
- n, m, R_7 and R_8 being as defined above for a compound with formula (I);

as a base or as an addition salt with an acid, and also as a hydrate or as a solvate.

The products of the invention will advantageously have a substituent R_5 selected from $(CH_2)_nNR_7R_8$, $CONR_7R_8$ and $(CH_2)_nNR_7COR_8$.

A product in accordance with the invention may be present in a non chiral or racemic form or enriched in one stereoisomer or enriched in one enantiomer; it may optionally be in the solvated or hydrated form, and may optionally be in the form of a salt.

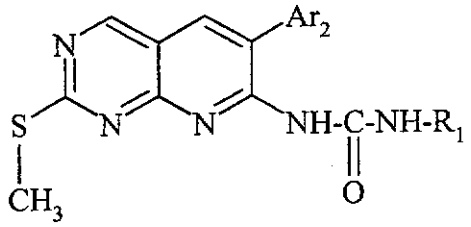
In a second aspect, the invention concerns the preparation of synthesis intermediates used in preparing products in accordance with the first aspect of the invention, said intermediates having the following general formula:



(II)

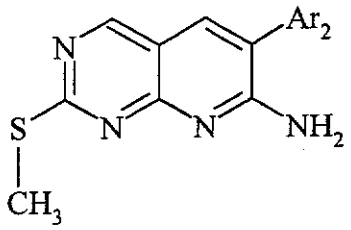
in which R_1 and Ar_2 are as defined above.

In a third aspect, the invention concerns the preparation of intermediates in accordance with the first and second aspect of the invention, having the following general formula:



in which R1 and Ar2 are as defined above.

In a fourth aspect, the invention concerns the preparation of intermediates in accordance with the first, second and third aspect, having the following general formula:



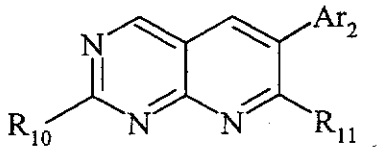
in which Ar_2 is as defined above.

Synthesis intermediates in accordance with the second to fourth aspects of the invention comprise a substituent Ar2 selected from phenyl, 2-methoxyphenyl, 2,6-dichlorophenyl, 3,5-dimethoxyphenyl, 3,4-dimethoxyphenyl, 2,6-dibromophenyl, 2-bromo-6-chlorophenyl, 2,4-dichlorophenyl and 3,5-dichlorophenyl.

Synthesis intermediates in accordance with the second and third aspects of the invention comprise a substituent R1 selected from ethyl, tertiobutyl and phenyl.

In accordance with the invention, compounds with formula (I) may be prepared using a process which is characterized in that the following are reacted:

(i) a compound with formula:



(IIb)

in which R_{10} is a leaving group such as: (a) halogen, in particular Cl or Br or (b) alkyl-S(O)m-, in which m = 0, 1, or 2; R_{11} is NHC(=R_{12})-NH-R_1, in which R_{12} = O or S; and

(ii) an amine with formula Ar'_1NH_2 (III), in which Ar'_1 represents Ar_1 as defined for (I) or a precursor of Ar_1; if appropriate, the group Ar'_1 in the compound obtained is transformed into a group Ar_1.

When R_{10} is a halogen or alkyl-S(O)m- in which m = 2, the reaction is carried out in a solvent, preferably a polar solvent:

(i) for example terahydrofuran, dimethylsulphoxide or ethanol, optionally in the presence of a trace of an acid such as hydrochloric acid; or

(ii) in dimethylsulphoxide in the presence of a strong base such as tBuOK;

at a temperature in the range from ambient temperature to the reflux temperature of the solvent.

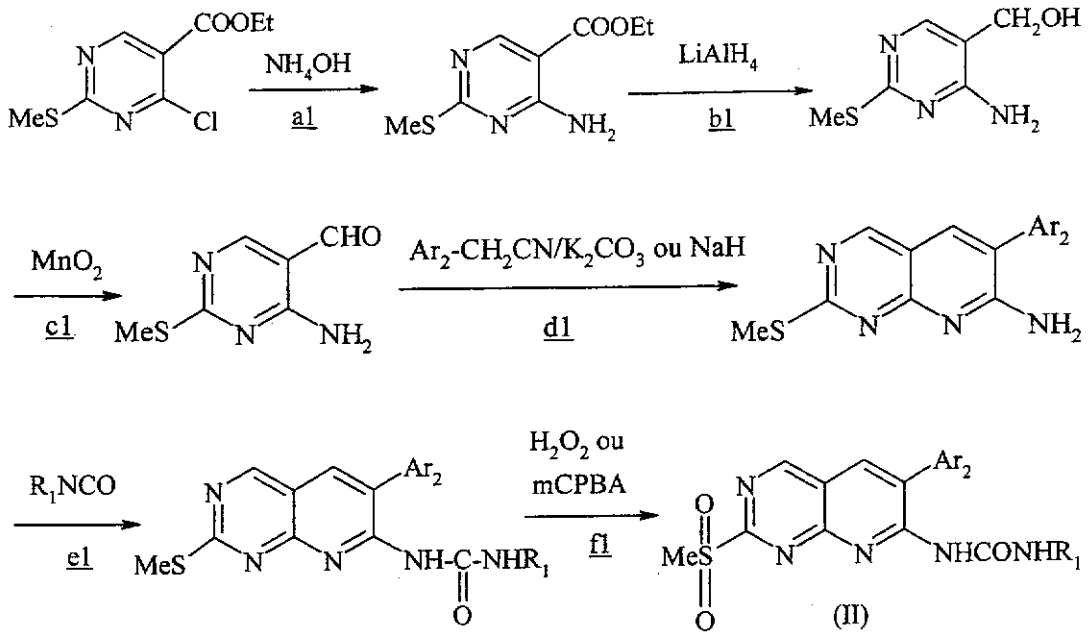
When R_{10} is alkyl-S(O)m- in which m = 0 or 1, the reaction may be carried out with molten Ar'_1NH_2 (III), preferably at a temperature close to 200°C, with no catalyst.

If appropriate, the amine functions present in the Ar'_1 group of compound (III) may have been transformed into salts or may have been protected.

The term "Ar1 precursor" means a group a), b), c), d) or e) as defined above for (I), in which substituents R_5 and/or R_6 are as defined above for (I) or are precursors of R_5 and/or R_6.

The compounds with formula (II) are prepared using the operating protocol described in European patent EP-A-0 790 997 and United States patent US-A-5 733 913, as described in Scheme 1 below:

SCHEME 1



mCPBA: *meta*-chloroperbenzoic acid.

Amines with formula (III) are known or prepared using known methods from the corresponding nitro compounds Ar'_1NO_2 (IV), by reduction either (i) in an acid medium in the presence of a metal such as iron or powdered zinc, or (ii) using hydrogen in the presence of a catalyst such as Pd/C.

Compounds with formula (IV) are known or prepared using known methods.

Thus, 5-nitro-1,3-benzodioxoles monosubstituted in the 2- position by a R_5 = methoxycarbonyl group may be prepared by the action of methyl dichloroacetate on 4-nitro-catechol (4-nitrobenzene-1,2-diol).

5-nitro-1,3-benzodioxoles gem-disubstituted in the 2- position may be prepared in accordance with Pharmazie, 2003, 58 (1), 13-17 by the action of ethyl dibromomalonate on 4-nitrocatechol (4-nitrobenzene-1,2-diol).

7-nitro-2,3-dihydro-1,4-benzodioxines substituted in the 2- or 3- positions with R_5 and R_6 may be prepared using the method described in International patent WO 01/021 577 from 4-nitrocatechol or using known chemical transformations.

1,3-dihydro-2-benzofuran-5-amine is described in J. Med. Chem., 1978, 21, 965-978; 4*H*-1,3-benzodioxin-6-amine is described in J. Org. Chem., 1994, 59 (4), 754-757; 4*H*-1,3-benzodioxin-7-amine is described in Chimie Therapeutique, 1972, 7, 443-449.

Known methods such as those described in March's Advanced Organic Chemistry, 5th Edition, 2005, ISBN 0471585890 can be used to transform the R_5 and/or R_6 group of compounds with formula (IV), depending on the substituents desired R_5 and/or R_6 for the compounds with formula (I). Group R_5 and/or R_6 in compounds with formula (I) may also be transformed to obtain novel compounds with formula (I) carrying the desired substituents R_5 and/or R_6.

Thus, the group R_5 = CO_2Me can be used to prepare compounds with formula (IV) or (I) in which R5 represents a CO2H, CN, CH2OH, CONR7R8. CONHNR7R8, CONR8OR9 or CH2NR7R8 group using methods which are known in the art.

Starting from a compound with formula (IV) or (I) comprising a group R_5 = $(CH_2)_n$-OH, in which n = 1, 2, or 3, a compound with formula (IV) or (I) can be prepared in which R_5 = mesyloxymethyl by the action of

mesyl chloride, followed by a compound with formula (IV) or (I) in which R_5 = $-CH_2NR_7R_8$ by the action of HNR_7R_8, R_7 and R_8 being as defined for compounds with formula (I).

The compounds of the invention are obtained in the racemic form; optically pure isomers may then be prepared using resolution methods which are known to the skilled person, such as crystallization by forming salts with chiral agents. It is also possible to prepare compounds in accordance with the invention in their optically pure form using asymmetric or stereospecific synthesis methods, or chromatographic techniques using a chiral phase. Further, the products of the invention may be separated by forming diastereoisomers, separating them then decomposing the pharmacologically useful diastereoisomer into its enantiomerically pure active product. Enzymatic techniques may also be used. Known supplemental separation techniques may be used. They include those discussed in: Enantiomers, Racemates, and Resolutions, John Wiley and Sons, New York (1981).

The compounds of the invention may also be prepared in a form which is enriched in one stereoisomer while preparing the synthesis intermediates. Hence, the resolution of enantiomers of amines with formula (III) or nitro precursors (IV) may be carried out using known methods.

The following examples describe the preparation of certain intermediates and compounds in accordance with the invention. These examples are not limiting and serve solely to illustrate the present invention.

The examples use the following abbreviations:

MP: melting point

Boc: tert-butoxycarbonyl

BOP: benzotriazol-1-yloxytris(dimethylamino) phosphonium hexafluorophosphate.

THF: tetrahydrofuran

AT: ambient temperature

DCM: dichloromethane

MeOH: methanol.

DCCI: dicyclohexylcarbodiimide

DIPEA:diisopropylethylamine

KHSO4 / K2SO4: 5% solution of KHSO4/K2SO4

Z: benzyloxycarbonyl

Proton nuclear magnetic resonance (NMR) spectra were recorded at 200 MHz or at 250 MHz in DMSO-d_6, unless otherwise indicated. The DMSO-d_6 signal was at 2.5 ppm and acted as a reference. The following abbreviations were used to interpret the spectra: s: singlet, d: doublet, t: triplet, m: broad signal, mt: multiplet, se: broad singlet, dd: double doublet, qd: quadruplet, qt: quintuplet.

Preparation of a compound with formula (II).

Preparation 1

N-(t-Butyl)-N'-[6-(2,6-dichlorophenyl)-2-(methylsulphonyl)pyrido[2,3-d]pyrimidin-7-yl]urea.

1.1 Ethyl 4-amino-2-(methylthio)pyrimidine-5-carboxylate.

140 ml of a 20% NH_4OH solution was added to a suspension of 50.7 g of ethyl 4-chloro-2-(methylthio)pyrimidin-5-carboxylate in 400 ml of EtOH over 20 minutes, keeping the temperature at about 20°C. After stirring for 20 hours at ambient temperature, the reaction medium was concentrated under vacuum almost to dryness, then the residue was taken up in 350 ml of water, stirred for 20 minutes, filtered, and washed with 3 x 60 ml of water then vacuum dried in the presence of P_2O_5. A white solid was obtained: MP = 134-135°C, m = 39.9 g.

1.2 [4-Amino-2-(methylthio)pyrimidin-5-yl]methanol.

210 ml of a 1M solution of $LiAlH_4$ in THF was added to 39.68 g of the ester obtained in the preceding step dissolved in 1 litre of THF over 45 minutes, keeping the temperature below 30°C. It was stirred for a 1 hour longer, then the temperature was reduced to 5°C and 9 ml of water, 6.5 ml of 5N sodium hydroxide then 32 ml of water were added, dropwise in succession.

After stirring for 10 minutes, the solid was filtered then rinsed with THF. The filtrate was concentrated under vacuum and the residue was re-dissolved in 600 ml of boiling toluene, rapidly hot filtered to eliminate some of the insolubles and the filtrate was allowed to cool overnight. The white crystals obtained were filtered, washed with a little toluene then with ether and dried. MP = 124-127°C, m = 23.9 g.

1.3 4-Amino-2-(methylthio)pyrimidine-5-carbaldehyde.

79.5 g of active MnO2 was added over 2 minutes to a suspension of 23.8 g of alcohol obtained from the preceding step in 1600 ml of chloroform then stirred overnight at ambient temperature; the solid was filtered, washed with 3 x 75 ml of CHCl3 and the filtrate was concentrated under vacuum to dryness: the white solid residue was taken up in ether, filtered and dried. MP = 184-186°C, m = 21.05 g.

1.4 6-(2,6-dichlorophenyl)-2-(methylthio)pyrido[2,3-d]pyrimidin-7-amine.

5.47 g of 60% NaH was added over 5 minutes to 21 g of the aldehyde obtained in the preceding step dissolved in 240 ml of DMF cooled to 5°C, then 29.05 g of 2,6-dichlorophenylacetonitrile was added over 20 minutes. Stirring was continued for 30 minutes at 5°C then overnight at ambient temperature. The reaction medium was cooled to 5°C and 65 ml of a saturated solution of NH_4Cl was added followed by 500 ml of a water/ice mixture; a red precipitate was formed which was filtered, washed twice with water, completely drained, washed with ether using 100 ml of chloroform, then again with ether; after drying, a beige solid was obtained. MP = 250-253°C, m = 29.92 g.

The ether and chloroform wash phases were concentrated to dryness and taken up in a little chloroform to which ether had been added: a second quantity of 3.15 g was obtained. Total m = 33.07 g.

1.5 N-(t-Butyl)-N'-[6-(2,6-dichlorophenyl)-2-(methylthio)pyrido[2,3-d]pyrimidin-7-yl]urea.

4.6 g of 60% NaH was added over 10 minutes to 29.9 g of the amine obtained above, in solution in 300 ml of DMF, keeping the temperature below 25°C; it was stirred for a further 20 minutes, then 12.2 ml of tertiobutyl isocyanate was added over 20 minutes followed by stirring overnight. The reaction medium was slowly poured onto 800 ml of a water/ice mixture + 100 ml of 6N HCl; the precipitate formed was filtered, washed with water, drained then stirred for 1 hour in 300 ml of ether, followed by filtering, washing with ether and drying. A beige solid was obtained. MP = 195-196°C (dec.), m = 26.5 g.

1.6 N-(t-Butyl)-N'-[6-(2,6-dichlorophenyl)-2-(methylsulphonyl)pyrido[2,3-d]pyrimidin-7-yl]urea.

27 g of metachloroperbenzoic acid was added to 21.95 g of the urea obtained above, in solution in 300 ml of chloroform, over 25 minutes, keeping the temperature to below 25°C. A precipitate was formed. After 2 hours, the reaction medium was diluted with 1 litre of dichloromethane and Na_2SO_4 was added, followed by 14 g of $Ca(OH)_2$; after stirring for 30 minutes, the solid was filtered, washed with dichloromethane then the filtrate was concentrated to dryness; the residue was triturated into 80 ml of hot ether; it was allowed to cool then the white solid was filtered, washed with ether and dried. MP = 138-140°C, m= 20.5 g.

The following compounds with general formula (II) could be prepared in the same manner as that described for the compound described in preparation 1:

Ar_2	R_1	NMR
2,6-dichlorophenyl	**tert-butyl**	**1.40: s: 9H; 3.50: s: 3H; 7.50-7.70: m: 3H; 8.55: s: 1H; 9.10: s: 1H; 9.60: s: 1H; 9.95: s: 1H.**
2,6-dichlorophenyl	phenyl	3.50 ppm: s: 3H; 7.10 ppm: t: 1H; 7.40 ppm: t: 2H; 7.55-7.75 ppm: m: 5H; 8.60 ppm: s: 1H; 9.60 ppm: s: 1H; 9.80 ppm: s: 1H; 11.90 ppm: s: 1H.

3,5-dimethoxyphenyl	*tert*-butyl	1.40 ppm: s: 9H; 3.50 ppm: s: 3H; 3.80 ppm: s: 6H; 6.65-6.80 ppm: mt: 3H; 7.75 ppm: s: 1H; 8.45 ppm: s: 1H; 9.60 ppm: s: 1H; 9.80 ppm: s: 1H.
2,6-dichlorophenyl	ethyl	1.20 ppm: t: 3H; 3.40 ppm: qd: 2H; 3.50 ppm: s: 3H; 7.50 ppm - 7.75 ppm: m: 3H; 8.55 ppm: s: 1H; 9.40 ppm: s: 1H; 9.60 ppm: s: 1H; 9.70 ppm: s: 1H.
3,4-dimethoxyphenyl	*tert*-butyl	1.40 ppm: s: 9H; 3.45 ppm: s: 3H; 3.80 ppm: s: 3H; 3.90 ppm: s: 3H; 7.10 - 7.20 ppm: m: 3H; 7.75 ppm: s: 1H; 8.45 ppm: s: 1H; 9.55 ppm: s: 1H; 9.80 ppm: s: 1H.
phenyl	*tert*-butyl	1.40 ppm: s: 9H; 3.50 ppm: s: 3H; 7.60 ppm: se: 6H; 8.45 ppm: s: 1H; 9.40 ppm: s: 1H; 9.80 ppm: s: 1H.
2-methoxyphenyl	*tert*-butyl	1.40 ppm: s: 9H; 3,.50 ppm: s: 3H; 3.80 ppm: s: 3H; 7.10 - 7.40 ppm: mt: 4H; 7.60 ppm: t: 1H; 8.40 ppm: s: 1H; 9.60 ppm: s: 1H; 9.80 ppm: s: 1H.
2,6-dibromophenyl	*tert*-butyl	1.40 ppm: s: 9H; 3.50 ppm: s: 3H; 7.40 ppm: t: 1H; 7.85 ppm: d: 2H; 8.50 ppm: s: 1H; 9.00 ppm: s: 1H; 9.60 ppm: s: 1H; 10.00 ppm: s: 1H.
2-bromo-6-chlorophenyl	*tert*-butyl	1.40 ppm: s: 9H; 3.45 ppm: s: 3H; 7.50 ppm: t: 1H; 7.65 ppm: d: 1H; 7.80 ppm: d: 1H; 8.50 ppm: s: 1H; 9.00 ppm: s: 1H; 9.50 ppm: s: 1H; 9.90 ppm: s: 1H.
2,6-dibromophenyl	ethyl	1.15 ppm: t: 3H; 3.30 ppm: qd: 2H (masked by DOH); 3.50 ppm: s:3H; 7.40 ppm: t: 1H; 7.85 ppm: d: 2H; 8.50 ppm: s: 1H; 9.25 ppm: s: 1H; 9.60 ppm: s: 1H; 9.70 ppm: s: 1H.

2-bromo-6-chlorophenyl	Phenyl	(DMSO + TFA) 3.55 ppm: s: 3H; 7.10 ppm: t: 1H; 7.30-7.90 ppm: m: 7H; 8.60 ppm: s: 1H; 9.65 ppm: s: 1H.
2,6-dibromophenyl	Phenyl	3.55 ppm: s: 3H; 7.10 ppm: t: 1H; 7.35 ppm: qd: 3H; 7.60 ppm: d: 2H; 7.85 ppm: d: 2H; 8.60 ppm: s: 1H; 9.70 ppm: s: 1H; 9.80 ppm: s: 1H; 12.00 ppm: s: 1H.
2,4-dichlorophenyl	*tert*-butyl	1.35 ppm: s: 9H; 3.50 ppm: s: 3H; 7.45-7.60 ppm: mt: 2H; 7.80 ppm: s: 1H; 8.40 ppm: s: 1H; 8.80 ppm: s: 1H; 9.55 ppm: s: 1H; 9.80 ppm: s: 1H.

Preparation of compounds with formula (III).

The preparation numbers relate to the numbers of the compounds in Tables 1 and 2 below. When they contain an asymmetric carbon atom, these compounds were obtained in the racemic form unless otherwise indicated.

Preparation 2

2.1 Methyl 5-nitro-1,3-benzodioxole-2-carboxylate.

31.0 g of 4-nitrocatechol was added to 17.6 g of 60% NaH in suspension in 300 ml of DMF over 1 hour, with cooling to keep the temperature below 30°C. Stirring was continued for 15 minutes then 104 ml of methyl dichloroacetate was added over 1 hour, and stirring was continued for 4 hours at 90°C. The reaction medium was poured over a mixture of 2 litres of ice/water then extracted 4 times with 400 ml of AcOEt. The combined organic phases were washed once with a saturated NaCl solution then dried and concentrated under vacuum (DMF evaporated off). The residue was taken up in a AcOEt/H_2O mixture and the pH was brought to 8.6 using Na_2CO_3; the organic phase was decanted, washed with saturated $NaHCO_3$, H_2O, 5%

$KHSO_4/K_2SO_4$, H_2O, saturated NaCl then dried and vacuum evaporated; a semi-solid residue was obtained which was taken up then triturated in heptane to produce a solid. m = 27.7 g, MP = 90-92°C.

2.2 Methyl 5-amino-1,3-benzodioxole-2-carboxylate.

3.92 g of powdered zinc was added to 900 mg of the ester from the preceding step, dissolved in 30 ml of THF and, after cooling to -5°C, 4 ml of acetic acid diluted with 4 ml of THF was added over 30 minutes, then the temperature was allowed to rise. After 1½ hours, filtration was carried out and the solid was washed with a little THF and methanol. The filtrate was diluted with AcOEt and washed with H_2O, saturated $NaHCO_3$, H_2O, saturated NaCl; after drying and concentration under vacuum, a yellow wax was obtained which was identified by NMR. m = 800 mg.

Preparation 3

3.1 5-Nitro-1,3-benzodioxole-2-carboxamide.

20 ml of a 2M ammoniacal solution in methanol was poured onto 1.12 g of the methyl ester from preparation 2.1. After 25 minutes, it was concentrated under vacuum, the solid residue was taken up in Et2O, filtered and dried. m = 0.99 g; MP = 202-207°C.

3.2 5-Amino-1,3-benzodioxole-2-carboxamide.

4.57 g of powdered zinc was added to 0.98 g of the amide from preparation 3.1 in 35 ml of THF; after cooling to -5°C, 5 ml of acetic acid diluted in 5 ml of THF was added over 30 minutes. Following addition, the temperature was allowed to rise. After 1 hour, the solid was filtered, washed with a little THF, methanol, AcOEt. The filtrate was diluted with AcOEt, water was added thereto and the pH was brought to 6 with saturated NaHCO3. The precipitate formed was eliminated by filtering, the filtrate was decanted and then the organic phase was washed with saturated NaHCO3, H2O, saturated NaCl, then dried and evaporated; a wax which hardened on cooling was obtained. m = 0.63 g.

Preparation 4

4.1 5-Nitro-(1,3-benzodioxol-2-yl)methanol.

22.3 ml of a 1M solution of LiAlH4 in THF was added to 5.02 g of the methyl ester obtained in preparation 2.1, dissolved in 25 ml of THF, at -5°C over 1 hour 15 minutes; 20 minutes after addition was complete, 20 ml of AcOEt then 9 ml of 1N NaOH was added dropwise; the precipitate formed was eliminated by filtration, washed with AcOEt; the filtrate was diluted with AcOEt and washed with H2O, 5% KHSO4/K2SO4, H2O, saturated NaCl; after drying and concentration under vacuum, a wax was obtained which crystallized. m = 2.74 g, MP = 80-82°C.

In the next step, the nitro derivative from preparation 4.1 was reduced using the methods described above to produce the amine with formula (III) described in Preparation 4.2.

Preparation 5

5.1 (5-Nitro-1,3-benzodioxol-2-yl)methylsulphonate.

3 ml of triethylamine was added at 5°C to 4.12 g of the alcohol obtained in preparation 4.1 dissolved in 30 ml of CH2Cl2, followed by 1.85 g of mesyl chloride over 15 minutes. After 15 minutes, the ice bath was removed. After 55 minutes, the reaction medium was diluted with CH2Cl2 and water; the organic phase was decanted, washed with H2O, saturated NaCl, dried, evaporated. After trituration in heptane, a brown solid was obtained. m = 5.20 g, MP = 112-115°C.

5.2 [(5-Nitro-1,3-benzodioxol-2-yl)methyl]diethylamine.

2.19 g of diethylamine was added to 2.91 g of the mesylate obtained in the preceding step in 18 ml of DMF, then heated to 80°C. 0.73 g of diethylamine was added after 15 hours followed by a further 0.73 g after 8 hours. After a total of 48 hours, the reaction medium was diluted with AcOEt, washed with H2O then with saturated NaCl; after drying, the AcOEt had evaporated off and the residue was taken up in 40 ml of

Et2O + 10 ml of AcOEt and extracted twice with 60 ml of 0.25N HCl; the acid phases were mixed, brought into contact with AcOEt and the pH was brought to 9 with 10N NaOH; the organic phase was decanted off, washed with H2O then with saturated NaCl, then dried and evaporated. An oil was obtained. m = 1.55 g.

5.3 [(5-Amino-1,3-benzodioxol-2-yl)methyl]diethylamine.

7.45 g of powdered zinc was added to 1.93 g of the nitro compound obtained in the preceding step, dissolved in 70 ml of THF, then at -5°C, 7.6 ml of AcOH was added over 25 minutes and stirring was continued between 0°C and 5°C. After 1½ hours, the solid was filtered, washed with THF and a little methanol; the filtrate was diluted with AcOEt + H2O and the pH was brought to 9 with 10N NaOH; the precipitate formed was eliminated by filtering; the filtrate was decanted; the organic phase was washed with H2O, saturated NaCl, dried and evaporated off to obtain a black oil. m = 1.75 g.

Preparation 6

6.1 1-Methyl-4-((5-nitro-1,3-benzodioxol-2-yl)methyl)piperazine.

The reaction followed operating protocol 5.2; the product was isolated in the form of the dihydrochloride.

6.2 2-((Methylpiperazin-1-yl)methyl)-1,3-benzodioxol-5-amine.

The reaction followed operating protocol 5.3.

Preparation 7

7.1 5-nitro-1,3-benzodioxol-2-carboxylic acid.

1.5 ml of 5N NaOH was added over 30 minutes to 1.12 g of the ester obtained in preparation 2.1 in 12 ml of methanol. 35 minutes after addition was complete, the reaction medium was diluted with AcOEt and H2O and the pH was brought to 2 with 2N HCl; the organic phase was decanted, washed with H2O and

saturated NaCl, then dried and evaporated to obtain 1.25 g of oil.

7.2 N,N-Dimethyl-5-nitro-1,3-benzodioxol-2-carboxamide.

0.36 g of dimethylamine hydrochloride, 0.77 ml of DIPEA then 0.91 g of DCCI were added to 0.84 g of the acid obtained in the preceding step dissolved in 15 ml of dichloromethane. After stirring for 3 hours at ambient temperature, the reaction medium was filtered and the filtrate was diluted with CH2Cl2 and washed in succession with a solution of saturated NaHCO3, H2O, 5% KHSO4/K2SO4, H2O, saturated NaCl; after drying, the solvent was evaporated off then the residue was chromatographed on silica using a 99/1 dichloromethane/methanol mixture. 0.5 g of solid product was obtained. MP = 109°C.

Preparation 9

9.1 5-Nitro-1,3-benzodioxol-2-carbonitrile.

3.7 ml of POCl3 was added to 45 ml of DMF cooled to 5°C. After stirring for 30 minutes at 5°C, 1.67 g of the amide obtained in preparation 3.1 was added all at once. After stirring for 3 hours at ambient temperature, the reaction mixture was poured onto a mixture of 250 ml of water/ice. The precipitate formed was filtered, washed with water then dried. m = 1.32 g; MP = 105-110°C.

9.2 5-Amino-1,3-benzodioxol-2-carbonitrile.

Reduction of the nitro group of the product obtained in step 9.1 using NH2 was carried out employing the method described above, using a Zn/AcOH mixture.

Preparation 11

11.1 Ethyl 5-nitro-1,3-benzodioxol-2,2-dicarboxylate.

This compound was prepared using the operating protocol described in Pharmazie 2003 58 (1) 13-17.

11.2 5-Nitro-1,3-benzodioxol-2,2-dicarboxamide.

1.24 g of the diester obtained in the preceding step was added all at once to 14 ml of a 2M ammoniacal solution in methanol. After stirring for 30 minutes, the reaction medium was concentrated to dryness then the solid residue was taken up in ether, filtered and dried. m = 1.01 g; MP = 231-233°C.

Preparation 12

12.1 (5-Nitro-1,3-benzodioxol-2,2-diyl)dimethanol.

1.50 g of NaBH4 was added to 1.87 g of the diester obtained in preparation 11.1 in 60 ml of THF, at ambient temperature and over 1 hour; 25 minutes after addition was complete, it was diluted dropwise with 250 ml of AcOEt then 5 ml of methanol then 40 ml of water. The organic phase was decanted, washed with H2O, 5% KHSO4/K2SO4 solution, H2O, then saturated NaCl solution. After evaporating to dryness, the residue was chromatographed on silica with a chloroform/methanol mixture (98/2) and 0.64 g of an oil was obtained which set. MP = 111-113°C.

Preparation 13

13.1 Ethyl 3-(2-methyl-5-nitro-1,3-benzodioxol-2-yl)propanoate.

22.4 g of phosphoric anhydride was added to 15.51 g of 4-nitrocatechol in suspension in 22.10 g of ethyl acetoacetate at 70°C over 15 minutes. After 1 hour 45 minutes, the reaction medium was cooled then extracted with 4 x 150 ml of tepid toluene. The toluene phases were combined, washed with H2O, 1N NaOH, H2O, 5% KHSO4/K2SO4, H2O and a saturated NaCl solution. After drying and evaporating, the product was purified by silica chromatography, eluting with chloroform, to obtain 2.44 g of solid. MP = 76-78°C.

13.2 3-(2-methyl-5-nitro-1,3-benzodioxol-2-yl)propan-1-ol.

8 ml of 1M LiAlH4 in THF was added to 2.33 g of the ester obtained above, dissolved in 40 ml of THF, at

-5°C over 45 minutes. After 35 minutes, 8 ml of ethyl acetate was added dropwise, followed by 1 ml of water then 1 ml of 1N NaOH. The solid was eliminated by filtration; the filtrate was diluted with AcOEt, washed with H2O, 5% KHSO4/K2SO4, saturated NaCl; after drying, the organic phase was concentrated under vacuum; an oil was obtained. m = 1.90 g.

13.3 3-(2-methyl-5-nitro-1,3-benzodioxol-2-yl)propylmethanesulphonate.

1.01 g of triethylamine then 1.14 g of mesyl chloride were added to 1.89 g of the alcohol obtained in the preceding step 13.2, in 40 ml of dichloromethane at 5°C, over 20 minutes. Following addition, the ice bath was removed and stirring was continued for 1 hour; the reaction medium was diluted with CH2Cl2 and washed with H2O, then with a saturated NaCl solution; after drying, the solvent was evaporated off; 2.46 g of wax was obtained which solidified when cold.

13.4 N,N-diethyl-3-(2-methyl-5-nitro-1,3-benzodioxol-2-yl)propan-1-amine.

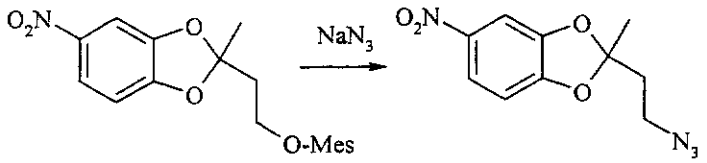
0.87 g of diethylamine was added to 1.21 g of mesylate in 20 ml of DMF then heated to 80°C. After 8½ hours, 0.44 g of diethylamine was added and heating was continued for 14 hours. The reaction medium was concentrated under vacuum, the residue was re-dissolved in AcOEt and the pH was brought to 9.5 with 1N NaOH; the organic phase was decanted, washed with H2O, a saturated NaCl solution, then dried; after evaporation, the crude product was obtained which was re-dissolved in 20 ml of AcOEt plus 20 ml of Et2O and extracted twice with 50 ml of 0.5N HCl; the two aqueous phases were mixed, brought into contact with AcOEt and the pH was brought to 9 using 10N NaOH; the organic phase was decanted, washed again with H2O then with a saturated NaCl solution, dried, evaporated, and 0.66g of oil was obtained.

13.5

The NO2 substituent of the product obtained in step 13.4 was reduced to NH2 using the method described

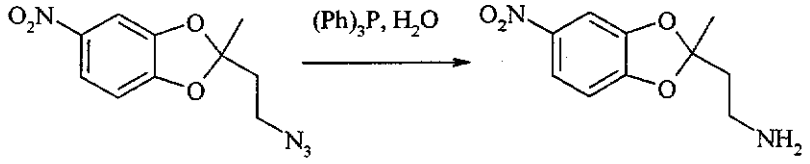
above using a Zn/AcOH mixture, to obtain a reduced product 13.5.

13.6



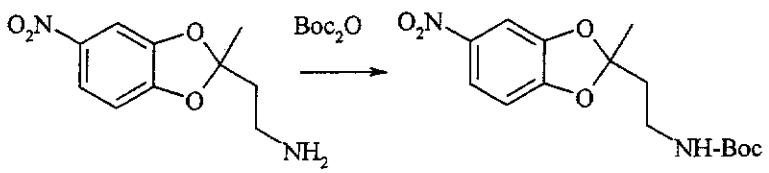
0.85 g of sodium nitride was added to 2.00 g of the product obtained in preparation 13.3 in 20 ml of DMF and stirred for 5 days at ambient temperature, extracted with ether and the organic phase was washed with water then with a saturated NaCl solution. An oil was obtained. m = 1.50 g. NMR in agreement.

13.7



A mixture of 1.49 g of the product obtained in preparation 13.6, 1.71 g of triphenylphosphine and 0.12 g of water in 25 ml of THF was stirred for 24 hours. The reaction medium was then extracted with AcOEt and washed with water. The crude product obtained was dissolved in a AcOEt/Et_2O mixture and extracted with an aqueous solution of 1N HCl. The aqueous acid phase was brought into contact with AcOEt and the pH was brought to 9 with 10N NaOH. The organic phase was isolated, washed with water, then with a saturated NaCl solution. The organic phase was concentrated under reduced pressure to obtain 1.10 g of oil.

13.8



1.09 g of the amine obtained in step 13.7 was dissolved in 10 ml of dichloromethane (DCM) then 0.20 g of triethylamine and 1.18 g of Boc_2O were added. After 5 hours, the reaction medium was diluted with DCM then washed in succession with a 5% $KHSO_4/K_2SO_4$ solution, water, and a saturated NaCl solution. After drying and evaporating the isolated organic phase under reduced pressure, 1.36 g of oil was obtained.

13.9 (precursor for Example 55)

1.35 g of the product obtained in step 13.8 was treated with Zn/AcOH as described in preparation 2.2 to reduce the NO_2 group to NH_2. 1.13 g of wax was obtained.

Preparation 14

14.1 2-(2-Methoxymethyl)-5-nitro-1,3-benzodioxole.

1.45 g of hydroxymethyl obtained in preparation 4.1 was dissolved in 25 ml of THF; after cooling to 5°C, 353 mg of 60% NaH was added in small fractions; after 30 minutes, 0.92 ml of methyl iodide was added followed by stirring overnight at ambient temperature; 1 ml of methyl iodide was added and stirring was continued for 5 hours; 30 ml of saturated NH_4Cl solution was added to the reaction medium followed by water and ethyl acetate; the organic phase was decanted, washed again with H_2O then with saturated NaCl, dried and evaporated. The crude product was chromatographed on silica, eluting with 9/1 $CHCl_3$/heptane. 595 mg of oil was obtained and identified using NMR.

Preparation 16

16.1 Tert-butyl 2-((5-nitro-1,3-benzodioxol-2-yl)carbonyl)hydrazinecarboxylate.

A mixture of 900 mg of the ester (preparation 2.1) and 2.114 g of tertiobutylcarbazate in 40 ml of methanol was heated at 60°C for 60 hours; 600 mg of tertiobutylcarbazate was added and heating was continued for 3 hours. The methanol was evaporated off, the residue was taken up in ethyl acetate and washed with water, 0.2N hydrochloric acid, a saturated

sodium bicarbonate solution, water and a saturated NaCl solution. A wax was obtained which set. m = 1.27 g.

16.2 Tert-butyl 2-((5-amino-1,3-benzodioxol-2-yl)carbonyl)hydrazinecarboxylate.

3.92 g of powdered zinc was added to 1.30 g of the product obtained in the preceding step dissolved in 25 ml of THF then, at -5°C, 4.8 g of acetic acid was added over 30 minutes; the ice bath was removed and stirring was continued for 2 hours; the solid was filtered, washed with a little THF then with AcOEt; water was added to the filtrate and the pH was brought to 6.5 with a 15% Na2CO3 solution; the AcOEt was decanted, washed with saturated NaHCO3, H2O, saturated NaCl, dried and evaporated. The expected compound was obtained in the form of a black oil. m = 1.12 g.

Preparation 21

21.1

1.51 g of sodium azide was added to 2.14 g of the mesylate obtained in step 5.1 dissolved in 17 ml of DMF, and heated to 70°C for 3 hours. The reaction mixture was extracted with AcOEt then washed with water then with a saturated NaCl solution. An oil was obtained. m = 1.71 g.

21.2

3.41 g of triphenylphosphine was added in small portions to 1.7 g of the product obtained in step 21.1 dissolved in 20 ml of AcOEt, then after 10 minutes, 2.34 ml of water was added and heated to 60°C. After 1 hour, the reaction mixture was evaporated to dryness, and taken up in Et2O. The insolubles were eliminated then an excess of a solution of saturated HCl in ether was added. The solid formed was filtered, washed with ether then dried to obtain the expected product in the form of the hydrochloride. The corresponding amine was obtained by liberating the hydrochloride.

21.3

The product obtained in step 21.2 could be separated into its two enantiomers:

1.97 g of (S)(+)mandelic acid was added to 2 g of the amine obtained in step 21.2 dissolved in 70 ml of water and 7 ml of dioxane at 70°C. The reaction medium was then allowed to return slowly to 30°C with magnetic stirring. The precipitate obtained was filtered, dissolved in 40 ml of water and 4 ml of dioxane at 70°C, then was allowed to return slowly to 30°C with stirring. The solid which formed during cooling was filtered and dried. m = 0.49 g. The product obtained was taken up in 20 ml of water and 100 ml of AcOEt then was brought to a pH of 9.5 by adding 1N NaOH. The mixture was decanted, the organic phase was isolated, washed several times with water then with a saturated NaHCO3 solution, with water, and finally with a saturated NaCl solution. The organic phase was collected, dried and the solvent was evaporated off under reduced pressure. 0.27 g of a wax was obtained which slowly hardened. [α]D = +94.7°, at 25°C; C = 0.5 (MeOH); optical purity: chiral HPLC: 96/4 (rotatory power of 100% purified enantiomer = 102°).

21.4

1.49 ml of triethylamine was added to the product obtained in step 21.2 in 30 ml of DCM, then 2.53 g of Boc2O was added in small portions. After 1 hour, the reaction medium was washed with 5% KHSO4/K2SO4, H2O, saturated NaCl. After drying, the organic phase was concentrated to dryness then the residue was triturated in heptane; 2 g of solid was obtained.

21.5

1.79 g of the product obtained in step 21.4 dissolved in 15 ml of THF was added to 480 mg of NaH in 20 ml of THF at 5°C over 30 minutes. The mixture was stirred for 45 minutes at ambient temperature then 1.2 ml of iodomethane was added over 10 minutes. After 3 hours, the reaction medium was poured onto 60 ml of a saturated aqueous solution of citric acid then extracted with ethyl acetate. The organic phase was isolated, washed with water then with a saturated NaCl

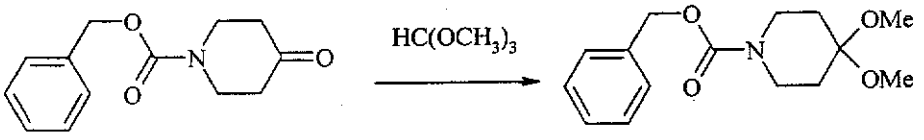
solution, dried and evaporated to dryness. The crude product obtained was purified by flash chromatography with a dichloromethane gradient in cyclohexane. A white solid was obtained. m = 1.4 g.

21.6

The product from preparation 21.5 was reduced using the usual method employing Zn/AcOH.

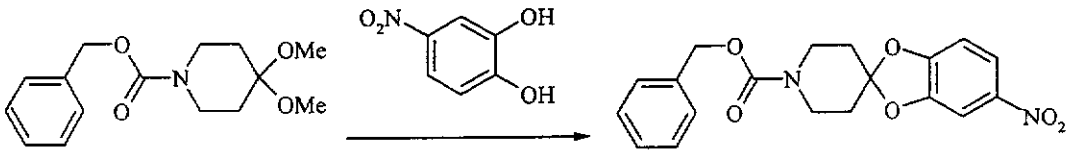
Preparation 22

22.1



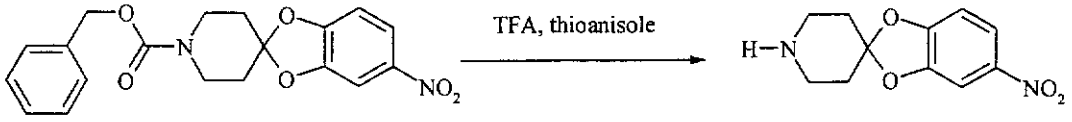
A mixture of 11.65 g of 4-N-Z-piperidone, 6.35 g of methyl orthoformate and 40 mg of para-toluenesulphonic acid was heated in a Claisen reactor at 60°C for 1 hour then at 70°C for 1 hour, allowing the methyl formate to distill. The residue was diluted with AcOEt + H2O and several drops of 1N sodium hydroxide were added to bring the pH to 7. The organic phase was decanted, isolated, washed with water then with a saturated NaCl solution, dried and evaporated off. 14 g of colourless oil was obtained.

22.2



16.09 g of the ketal obtained in step 22.1, 10.80 g of 4-nitrocatechol and 60 mg of para-toluenesulphonic acid were heated in 120 ml of toluene with slow distillation of the toluene. After 4h30, the reaction medium was diluted with toluene and cooled, the insolubles were eliminated by filtration, the filtrate was washed with 1N NaOH, H2O, a 5% KHSO4/K2SO4 solution, H2O and saturated NaCl. After drying and evaporating off, the crude product was chromatographed

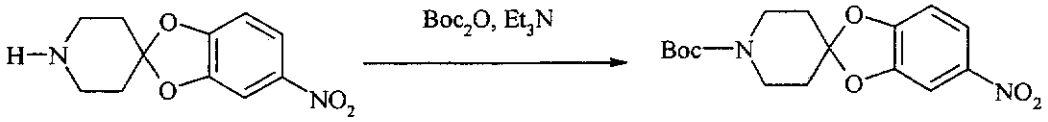
on silica, eluting with 98/2 CHCl3/AcOEt. 0.65 g of the expected product was obtained.



22.3

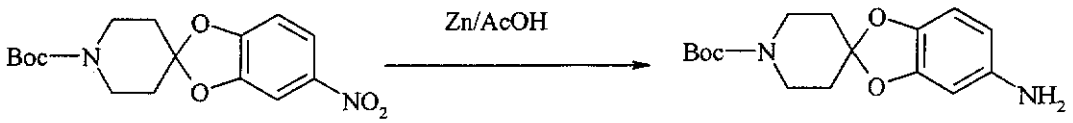
0.5 ml of thioanisole was added to 0.60 g of the product obtained above in 5 ml of trifluoroacetic acid. After 3 h, the reaction medium was concentrated under vacuum; the residue was taken up in CH2Cl2 with H2O and brought to a pH of 9 with 1N NaOH. After decanting, the organic phase was washed again with water then with a saturated NaCl solution. The organic phase was dried and evaporated. The recovered crude product was chromatographed on silica with 95/5/0.1 CHCl3/MeOH/NH4OH. 100 mg of the expected product was obtained as a solid.

22.4



95 mg of the amine obtained in step 22.3 was treated with 98 mg of Boc2O and 20 mg of triethylamine in 3.5 ml of dichloromethane for 1 h. After the usual reaction and treatment, a white solid was obtained. m = 130 mg.

22.5



The NO2 group was reduced to amine using Zn/AcOH employing the operating protocol described above.

23.1

0.46 ml of pyridine was added to 985 mg of the product produced in preparation 4.1 dissolved in 25 ml of DCM. 0.84 ml of triflic anhydride dissolved in 3 ml

of DCM was then added over 20 minutes at 5°C. After 1 h at 5°C, the reaction medium was washed with ice water then with a saturated NaCl solution. The organic phase was dried and concentrated under vacuum. 1.42 g of solid was obtained.

23.2

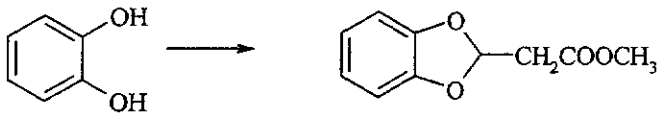
75 mg of diethanolamine was added to 1.15 g of the product obtained in step 23.1 dissolved in 10 ml of DCM + 0.5 ml of DMF. After stirring overnight, the reaction medium was diluted with 100 ml of DCM, washed with water, washed with a saturated NaCl solution, dried and concentrated under reduced pressure. The residue was purified by flash chromatography using a methanol gradient of 0 to 15% in chloroform. 730 mg of solid was obtained.

23.3

The NO2 group was reduced to the amine by Zn/AcOH as described above. Starting from 720 mg of the product obtained in step 23.2, 400 mg of the expected product was obtained in the form of a gum.

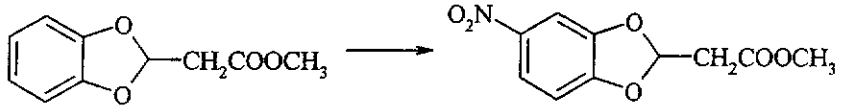
Preparation 24

24.1



The product was prepared using the operating protocol described in Org. Lett. 2001, 3(9), 1399-1402.

Preparation 24.2



Racemic

5 ml of 69% nitric acid was added over 2 minutes to 2.73 g of the product obtained in 24.1 in 30 ml of DCM. After stirring for 2h 30 mins, the reaction medium was diluted with Et2O; the organic phase was washed twice with H2O, twice with a 7% iced

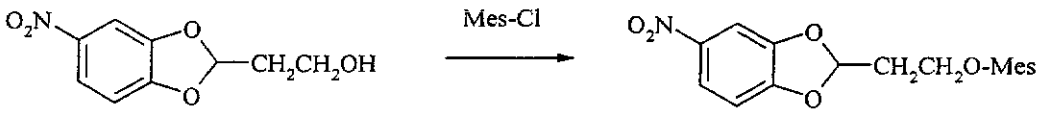
Na2CO3 solution, once with water, once with a 5% KHSO4/K2SO4 solution, once with water then once with a saturated NaCl solution. After drying and evaporating, a white solid was obtained. m = 3.20 g.

Preparation 24.3



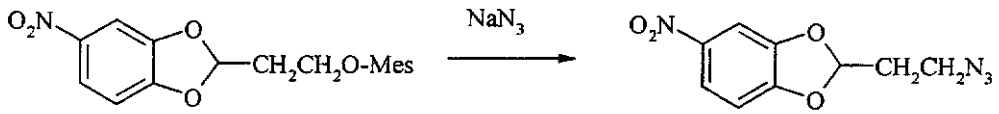
3.63 g of the product obtained using the operating protocol described above in 80 ml of THF was treated with 532 mg of LiAlH4 at -5°C for 1 h. After the usual treatment, 2.60 g of product was isolated in the form of a thick oil.

Preparation 24.4



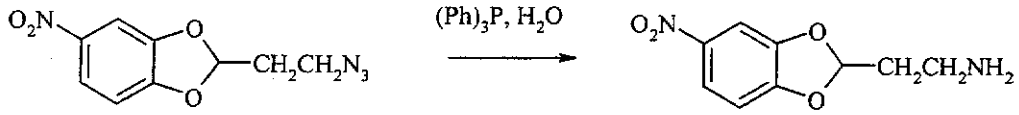
2.59 g of the alcohol obtained in step 24.3 was treated with mesyl chloride as described in the operating protocol described in preparation 13.3 to give 3.52 g of mesylate. It was identified by NMR.

Preparation 24.5



3.51 g of the product obtained in step 24.4 was treated with 1.97 g of sodium azide using the operating protocol described in preparation 13.6. 2.60 g of the expected product was obtained.

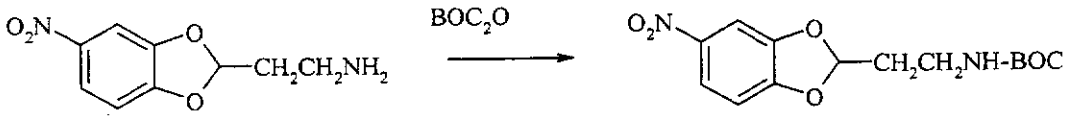
Preparation 24.6



2.59 g of the product obtained in step 24.5 was treated with 4.90 g of triphenylphosphine and 2 ml of water using the operating protocol described in

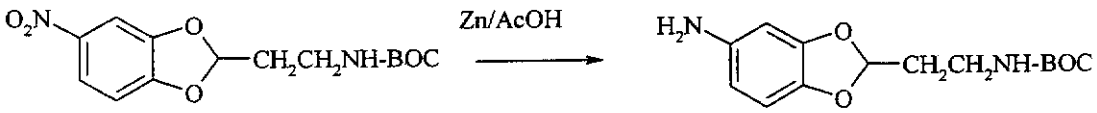
preparation 13.7. 2 g of the expected product was obtained as an oil.

Preparation 24.7



1.99 g of the product obtained in step 24.6 was treated with BOC_2O using the operating protocol described in preparation 13.8. 2.41 g of solid was obtained.

Preparation 24.8



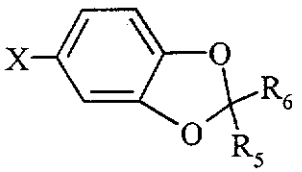
The product obtained in step 24.7 was treated with Zn/AcOH using the usual protocol to reduce the nitro group to an amino group. 0.84 g of the expected product, in the form of a wax, was obtained from 0.93 g of starting product.

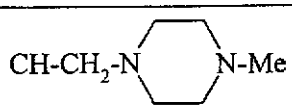
The compounds with formula (III) and intermediates with formula (IV), all benzodioxole derivatives, are characterized in the table below.

Transformation of the compound with formula (IV) into a compound with formula (III) was carried out using preparation 5.3 for the following compounds: 7.2, 9.1, 11.1, 12.1, 13.4 and 14.1.

TABLE 1

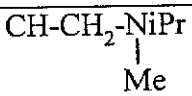
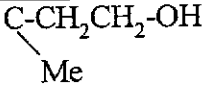
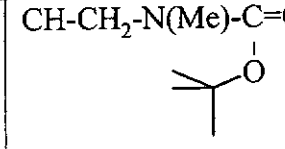
Preparations of compounds with formula (III).

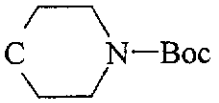


Prepa-rations	CR_5R_6	X = NO_2 (IV) NMR	X = NH_2 (III) NMR
2	CH-CO_2Me	2.1 3.80 ppm: s: 3H; 6.95 ppm: s: 1H; 7.25 ppm: d: 1H 7.85-8.00 ppm: m: 2H	2.2 3.80 ppm: s: 3H; 4.90 ppm: se: 2H; 6.10 ppm: dd: 1H; 6.35 ppm: d: 1H; 6.55 ppm: s: 1H; 6.75 ppm: d: 1H
3	CH-$CONH_2$	3.1 6.60 ppm: s: 1H; 7.25 ppm: d: 1H; 7.80-8.30 ppm: m: 4H	3.2 4.80 ppm: se: 2H; 6.00 ppm: dd: 1H; 6.10 ppm: s: 1H; 6.25 ppm: d: 1H; 6.60 ppm: d: 1H; 7.65 ppm: se: 1H; 7.90 ppm: se: 1H
4	CH-CH_2OH	4.1 3.75 ppm: m: 2H; 5.40 ppm: t: 1H; 6.55 ppm: t: 1H; 7.10 ppm: d: 1H; 7.75 ppm: d: 1H; 7.90 ppm: dd: 1H	4.2 3.65 ppm: m: 2H; 4.70 ppm: se: 2H; 5.25 ppm: t: 1H; 5.90-6.05 ppm: m: 2H; 6.20 ppm: d: 1H; 6.55 ppm: t: 1H
5	CH-CH_2NEt_2	5.2 0.95 ppm: t: 6H; 2.55 ppm: qd: 4H; 2.95 ppm: d: 2H; 6.55 ppm: t: 1H; 7.10 ppm: d: 1H; 7.70 ppm: d: 1H; 7.90 ppm: dd: 1H	5.3 0.85 ppm: t: 6H; 2.45 ppm: qd: 4H; 2.65 ppm: d: 2H; 4.55 ppm: se: 2H; 5.75-5.90 ppm: m: 2H; 6.05 ppm: d: 1H; 6.40: d: 1H
6		6.1 2.60 ppm: s: 3H; 2.70-3.40 ppm: m: 10H; 6.65 ppm: t: 1H;	6.2 2.30-2.70 ppm: m: 11H; 2.80 ppm: d: 2H; 4.80 ppm: se: 2H; 6.00 ppm: dd: 1H;

Prepa-rations	CR_5R_6	X = NO_2 (IV) NMR	X = NH_2 (III) NMR
		7.05 ppm: d: 1H; 7.65 ppm: d: 1H; 7.80 ppm: dd: 1H	6.15 ppm: t: 1H; 6.20 ppm: d: 1H; 6.60 ppm: d: 1H
7	CH-$CONMe_2$	7.2 2.85 ppm: s: 3H; 3.10 ppm: s: 3H; 7.15 ppm: d: 1H; 7.35 ppm: s: 1H; 7.80 ppm: d: 1H; 7.90 ppm: dd: 1H	7.3 2.85 ppm: s: 3H; 3.05 ppm: s: 3H; 4.80 ppm: se: 2H; 6.00 ppm: dd: 1H; 6.20 ppm: d: 1H; 6.60 ppm: d: 1H; 6.75 ppm: s: 1H
8	CH-CONHEt	8.1 1.00 ppm: s: 3H; 3.10 ppm: qt: 2H; 6.50 ppm: s: 1H; 7.10 ppm: d: 1H; 7.70 ppm: d: 1H; 7.85 ppm: dd: 1H; 8.70 ppm: te: 1H	8.2 1.00 ppm: t: 3H; 3.05 ppm: qt: 2H; 4.70 ppm: se: 2H; 5.95 ppm: dd: 1H; 6.05 ppm: s: 1H; 6.15 ppm: d: 1H; 6.50 ppm: d: 1H; 8.40 ppm: te: 1H
9	CH-CN	9.1 7.30 ppm: d: 1H; 7.45 ppm: s: 1H; 8.05-8.85 ppm: m: 2H	9.2 5.10 ppm: se: 2H; 6.20 ppm: dd: 1H; 6.45 ppm: d: 1H; 6.85 ppm d: 1H; 7.25 ppm: s: 1H
10	CH-$CONHNMe_2$	10.1 2.50 ppm: s: 6H; 6.50 ppm: s: 1H; 7.10-7.20 ppm: m: 1H; 7.75-7.80 ppm: m: 1H; 7.85-7.95 ppm: m: 1H; 9.80 ppm: se: 1H	10.2 2.50 ppm: s: 6H; 4.75 ppm: se: 2H; 5.95-6.50 ppm: m: 4H; 9.50 ppm: se: 1H

Prepa-rations	CR_5R_6	X = NO_2 (IV) NMR	X = NH_2 (III) NMR
11	C-$CONH_2$ $CONH_2$	11.1 7.05 ppm: d: 1H; 7.65 ppm: d: 1H; 7.80 ppm: dd: 1H; 7.90 ppm: se: 2H; 8.05 ppm: se: 2H	11.2 4.80 ppm: se: 2H; 6.00 ppm: dd: 1H; 6.20 ppm: d: 1H; 6.55 ppm: d: 1H; 7.70 ppm: de: 4H
12	C-CH_2OH CH_2OH	12.1 3.80 ppm: d: 4H; 5.40 ppm: t: 2H; 7.05 ppm: d: 1H; 7.65 ppm: d: 1H; 7.85 ppm: dd: 1H	12.2 3.45 ppm: d: 4H; 4.50 ppm: se: 2H; 5.05 ppm: t: 2H; 5.80 ppm: dd: 1H; 6.05 ppm: d: 1H; 6.35 ppm: d: 1H
13	C-CH_2CH_2-NEt_2 Me	13.4 0.90 ppm: t: 6H; 1.70 ppm: s: 3H; 2.10-2.60 ppm: m: 8H; 7.05 ppm: d: 1H; 7.70 ppm: d: 1H; 7.85 ppm: dd: 1H.	13.5 0.90 ppm: t: 6H; 1.50 ppm: s: 3H; 1.85-2.00 ppm: m: 2H; 2.30-2.60 ppm: m: 6H; 4.60 ppm: se: 2H; 5.90 ppm: dd: 1H; 6.10 ppm: d: 1H; 6.45 ppm: d: 1H
13a	C-CH_2CH_2-NH-Boc Me	13.8 1.35 ppm: s: 9H; 1.45 ppm: s: 3H; 2.10-2.20 ppm: m: 2H; 3.0-3.10 ppm: m: 2H; 6.80-6.90 ppm: m:1H; 7.10 ppm: d:1H; 7.65 ppm: d: 1H; 7.85 ppm: dd: 1H.	13.9 1.35 ppm: s: 9H; 1.45 ppm: s: 3H; 1.90-2.05 ppm: m: 2H; 2.95-3.10 ppm: m: 2H; 4.60 ppm: se: 2H; 5.95 ppm: dd:1H; 6.15 ppm: d: 1H; 6.45 ppm: d: 1H; 6.75 ppm: t:1H.
14	CH-CH_2OMe	14.1 3.30 ppm: s: 3H; 3.75 ppm: d: 2H; 6.60 ppm: t: 1H;	14.2 3.30 ppm: s: 3H; 3.60 ppm: d: 2H; 4.60 ppm: se: 2H;

Prepa-rations	CR_5R_6	X = NO_2 (IV) NMR	X = NH_2 (III) NMR
		7.10 ppm: d: 1H; 7.70 ppm: d: 1H; 7.85 ppm: dd: 1H	5.95 ppm: dd: 1H; 6.10 ppm: t: 1H; 6.20 ppm: d: 1H; 6.50 ppm: d: 1H
15		15.1 1.30-1.40 ppm: 2d: 6H; 2.70 ppm: d: 3H; 3.50-4.00 ppm: m: 3H; 7.10 ppm: t: 1H; 7.20-7.30 ppm: m: 1H; 7.80-8.00 ppm: m: 2H.	15.2
16	CH-CO-NH-NHBoc	16.1 1.40 ppm: s: 9H; 6.70 ppm: s: 1H; 7.20 ppm: d: 1H; 7.80 ppm: d: 1H; 7.95 ppm: dd: 1H; 9.00 ppm: s: 1H; 10.60 ppm: s: 1H.	16.2 1.40 ppm: s: 9H; 4.70 ppm: se: 2H; 6.05 ppm: dd: 1H; 6.30 ppm: d+s: 2H; 6.65 ppm: d: 1H 8.90 ppm: s: 1H; 16.20 ppm: s: 1H.
17		17.1 1.60 ppm: s: 3H; 2.15 ppm: t: 2H; 3.60 ppm: qd: 2H; 4.60 ppm: t: 1H; 7.00 ppm: d: 1H; 7.70 ppm: d: 1H; 7.85 ppm: dd: 1H.	17.2 1.50 ppm: s: 3H; 2.00 ppm: t: 2H; 3.55 ppm: qd: 2H; 4.50 ppm: t: 1H; 4.60 ppm: se: 2H; 5.90 ppm: dd: 1H; 6.10 ppm: d: 1H; 6.50 ppm: d: 1H.
21		21.5 1.35 ppm: s: 9H; 2.85 ppm: se: 3H; 3.70 ppm: d: 2H; 6.60 ppm: te: 1H;	21.6 1.40 ppm: s: 9H; 2.80 ppm: se: 3H; 3.50 ppm: d: 2H; 4.70 ppm: s: 2H;

Prepa-rations	CR_5R_6	X = NO_2 (IV) NMR	X = NH_2 (III) NMR
		7.10 ppm: d: 1H; 7.70 ppm: d: 1H; 7.85 ppm: dd: 1H.	5.95 ppm: dd: 1H; 6.10-6.20 ppm: m: 1H; 6.20 ppm: d: 1H; 6.50 ppm: d: 1H.
22		22.4 1.40 ppm: s: 9H; 2.00 ppm: te: 4H; 3.55 ppm: te: 4H; 7.10 ppm: d: 1H; 7.75 ppm: d: 1H; 7.90 ppm: dd: 1H.	
23	CH-CH_2N($(CH_2)_2$OH)$_2$	23.2 2.65 ppm: t: 4H; 3.05 ppm: d: 2H; 3.40 ppm: t: 4H; 4.35 ppm: se: 2H; 6.50 ppm: t: 1H; 7.05 ppm: d: 1H; 7.70 ppm: d: 1H; 7.85 ppm: dd: 1H.	
24	CH-CH_2CH_2-NH-Boc	24.7 1.35 ppm: s: 9H; 2.10 ppm: qd: 2H; 3.10 ppm: qd: 2H; 6.50 ppm: t: 1H; 6.95 ppm: t: 1H; 7.10 ppm: d: 1H; 7.70 ppm: d: 1H; 7.90 ppm: dd: 1H.	24.8 1.35 ppm: s: 9H; 1.95 ppm: qd: 2H; 3.10 ppm: qd: 2H; 4.70 ppm: se: 2H; 5.90-6.05 ppm: mt: 2H; 6.15 ppm: d: 1H; 6.50 ppm: d: 1H; 6.90 ppm: t:1H.

Preparation 18

18.1 ((7-Nitro-2,3-dihydro-1,4-benzodioxin-2-yl)methyl) diethylamine hydrochloride.

800 µl of diethylamine was added to 1.50 g of 7-nitro-2,3-dihydro-1,4-benzodioxin-2-yl methanesulphonate, described in International patent

application WO 01/021 577, in 30 ml of DMF, and heated to 80°C, then 3 times 800 µl of diethylamine were added at 12 hour intervals. After 48 hours, the reaction medium was evaporated to dryness; the residue was taken up in 100 ml of AcOEt then 5 ml of 4N NaOH; the organic phase was decanted, washed with water then with saturated NaCl; after drying and evaporating off the AcOEt, the residue was taken up in 20 ml of AcOEt then 50 ml of 0.5N HCl and stirred then decanted; the aqueous phase was evaporated to dryness and the residue was triturated in ether, filtered and dried. m = 0.95 g, MP = 192-194°C.

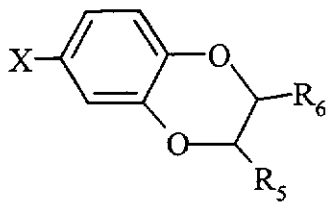
18.2 7-Amino-2,3-dihydro-1,4-benzodioxin-2-yl)methyl)diethylamine.

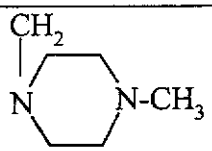
3.05 g of powdered zinc was added to 0.94 g of the nitro derivative obtained in the preceding step in suspension in 40 ml of THF. After cooling to 0°C, 3.1 ml of acetic acid was added over 20 minutes; after 10 minutes the ice bath was removed and stirring was continued for 2 hours. The solid was filtered, washed with THF then with a little methanol. The filtrate was evaporated to dryness, taken up in ethyl acetate, water was added and the pH was brought to 8 using 10N NaOH; the precipitate formed was eliminated by filtration, washed with AcOEt; the filtrate was decanted, washed with saturated NaCl, dried and evaporated; a brown oil was obtained. m = 0.51 g.

The compounds with formula (III) and intermediates with formula (IV), all benzodioxine derivatives, were prepared in the racemic form and are characterized in Table 2 below:

TABLE 2

Preparation of compounds with formula (III)



Prepa-rations	R_5	R_6	X = NO_2 (IV), HCl NMR	X = NH_2 (III) NMR
18	CH_2-NEt_2	H	18.1 1.00-1.10 ppm: m: 6H; 3.00-3.40 ppm: m: 6H; 4.00-4.50 ppm: m: 2H; 4.80-4.95 ppm: m: 1H; 7.10 ppm: d: 1H; 7.65-7.80 ppm: m: 2H	18.2 1.00 ppm: t: 6H; 2.45-2.65 ppm: m: 6H; 3.80-4.20 ppm: m: 3H; 4.70 ppm: se: 2H; 6.05-6.20 ppm: m: 2H; 6.60 ppm: d: 1H
19		H	19.1 2.85 ppm: s: 3H; 3.30-3.80 ppm: m: 10H; 4.15-4.50 ppm: m: 2H; 4.90-5.10 ppm: m: 1H; 7.15 ppm: d: 1H; 7.75-7.90 ppm: m: 2H	19.2 2.15 ppm: s: 3H; 2.30-2.60 ppm: m: 10H; 3.70-4.30 ppm: m: 3H; 4.60 ppm: se: 2H; 6.00-6.10 ppm: m: 2H; 6.50 ppm: d: 1H.
20	H	CH_2-NEt_2	20.1 1.20-1.30 ppm: m: 6H; 3.15-3.60 ppm: m: 6H; 4.10-4.60 ppm: m: 2H; 5.05 ppm: te: 1H; 7.15 ppm: d: 1H; 7.75-7.90 ppm: m: 2H	20.2 0.90 ppm: t: 6H; 2.40-2.60 ppm: m: 6H; 3.75-4.25 ppm: m: 3H; 4.60 ppm: se: 2H; 5.95-6.10 ppm: m: 2H; 6.50 ppm: d: 1H.

The numbers of the compounds shown relate to those given in Table 3 below which illustrates the chemical structures and physical properties of some compounds of the invention. When they contain an asymmetric carbon, these compounds were obtained in the racemic form.

EXAMPLE 1: Compound N° 2

Methyl 5-((7-(((tert-butylamino)carbonyl)amino)-6-(2,6-dichlorophenyl)pyrido[2,3-d] pyrimidin-2-yl)amino)-1,3-benzodioxole-2-carboxylate.

A mixture of 0.97 g of the compound from preparation 2.2 and 1.40 g of the compound from preparation 1 was heated to reflux in 15 ml of THF. After 6 hours, the reaction medium was concentrated under vacuum. The product was purified by silica chromatography with AcOEt/toluene (2/3) then re-chromatographed with 98/2 CHCl3/MeOH. 0.45 g of a yellow solid was obtained which was identified by mass spectrometry. MH^+ = 583.

EXAMPLE 2: Compound N° 4

5-((7-(((Tert-butylamino)carbonyl)amino)-6-(2,6-dichlorophenyl)pyrido [2,3-d]pyrimidin-2-yl)amino)-1,3-benzodioxole-2-carboxylic acid

0.3 ml of 2N sodium hydroxide was added to 0.25 g of the ester obtained in the above example in 10 ml of methanol. After 1 hour 10 minutes of stirring at ambient temperature, the reaction medium was diluted with AcOEt + H2O and the pH was brought to 4 using 1N HCl. The organic phase was decanted, washed with H2O then saturated NaCl, dried and evaporated off. The solid yellow residue was taken up in ether, triturated, filtered and dried. 205 mg of the product was obtained, identified by mass spectrometry. MH^+ = 569.

EXAMPLE 3: Compound N° 3

5-((7-(((tert-Butylamino)carbonyl)amino)-6-(2,6-dichlorophenyl)pyrido [2,3-d] pyrimidin-2-yl)amino)-1,3-benzodioxole-2-carboxamide.

A mixture of 0.62 g of the compound from preparation 3.2 and 1.40 g of the sulphone derivative from preparation 1 was heated to reflux in 20 ml of THF. After 3 hours, the reaction medium was concentrated under vacuum, then the residue was chromatographed on silica with 97/3 v/v CHCl3/MeOH. 315 mg of a yellow solid was obtained, identified by mass spectrometry. MH^+ = 568.

EXAMPLE 4: Compound N° 9 SSR 105451

N-(Tert-butyl)-N'-(6-(2,6-dichlorophenyl)-2-((2-hydroxymethyl)-1,3-benzodioxol-5-yl)amino)pyrido[2,3-d]pyrimidin-7-yl)urea.

43 µl of concentrated HCl was added to a mixture of 350 mg of the compound from preparation 4.2 and 937 mg of the compound from preparation 1 in 15 ml of EtOH; it was heated to 55°C for 6 hours. The reaction medium was concentrated under vacuum, then the residue was chromatographed on silica; 490 mg of a yellow solid was obtained and identified by mass spectrometry. MH^+ = 555.

EXAMPLE 5: Compound N° 6 SSR 104788

5-((7-(((Tert-butylamino)carbonyl)amino)-6-(2,6-dichlorophenyl)pyrido[2,3-d] pyrimidin-2-yl)amino)-1,3-benzodioxole-2-carbonitrile.

50 µl of concentrated HCl was added to 1.00 g of the compound from preparation 1 and 450 mg of the compound from preparation 9.2 in 15 ml of EtOH, then brought to a gentle reflux. After 1½ hours, the reaction medium was concentrated under vacuum. The residue was chromatographed on silica with CHCl3/ AcOEt (90/10, v/v). 465 mg of a yellow solid was obtained, identified by mass spectrometry. MH^+ = 550.

EXAMPLE 6: Compound N° 12 SSR 107159

N-(2-((Aminomethyl)-1,3-benzodioxol-5-yl)amino)-6-(2,6-dichlorophenyl)pyrido [2,3-d]pyrimidin-7-yl)-N'-(tert-butyl)urea.

1.9 ml of 1M LiAlH4 solution in THF was added to 530 mg of the nitrile derivative of the preceding example dissolved in 25 ml of THF, at -10°C over 30 minutes. 15 minutes after addition was complete, 1.5 ml of AcOEt then 10 ml of a saturated NH4Cl solution was added and the temperature was allowed to rise. After dilution with AcOEt, it was washed with water then with saturated NaCl. After drying, the organic phase was concentrated under vacuum and the residue was chromatographed on silica with CHCl3/MeOH (95/5; v/v). 165 mg of a yellow solid was obtained which was identified by mass spectrometry. MH^+ = 554.

EXAMPLE 7: Compound N° 19 SSR 109194

Tert-butyl 2-((5-((7-(((tert-butylamino)carbonyl)amino)-6-(2,6-

dichlorophenyl)pyrido[2,3-d]pyrimidin-2-yl)amino)-1,3-benzodioxol-2-yl)carbonyl)hydrazine carboxylate.

0.025 ml of concentrated HCl was added to 0.645 g of the product obtained in preparation 1 and 0.63 g of the product obtained in preparation 16.2 in 25 ml of ethanol, then stirred for 5 hours at 70°C. The reaction medium was evaporated and the residue was taken up in CHCl3 then washed with water, a saturated NaHCO3 solution, water, saturated NaCl, then dried and evaporated under vacuum; the residue was chromatographed on silica. 450 mg of a yellow solid was obtained and was identified by mass spectrometry. MH^+ = 683.

EXAMPLE 8: Compound N° 20 SSR 109195

N-(tert-Butyl)-N'-(6-(2,6-dichlorophenyl)-2-((2-hydrazinocarbonyl)-1,3-benzodioxol-5-yl)amino)pyrido[2,3-d]pyrimidin-7-yl)urea.

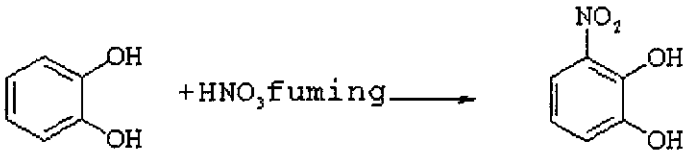
360 mg of the compound from the preceding example was stirred for 45 minutes in a mixture of 4 ml of CH2Cl2 and 14 ml of TFA ; the reaction medium was evaporated under vacuum ; the residue was taken up in CHCl3 and washed with H2O, 15% Na2CO3 in water, H2O, and a saturated NaCl solution; after drying, the chloroform was evaporated under vacuum then the residue was triturated in ether, filtered and dried. 225 mg of yellow solid was obtained. MH^+ = 583.

EXAMPLE 9: Compound N° 34

431 mg of compound N° 33 was stirred for 30 minutes at ambient temperature in 5 ml of DCM and 5 ml of TFA. After evaporation, the residue was taken up in a chloroform/water mixture and the pH was brought to 9 by adding an aqueous 15% Na2CO3 solution. The organic phase was decanted, washed with water then with a saturated NaCl solution, dried and concentrated under reduced pressure. 116 mg of a solid was obtained. $[M + H]^+$ = 568

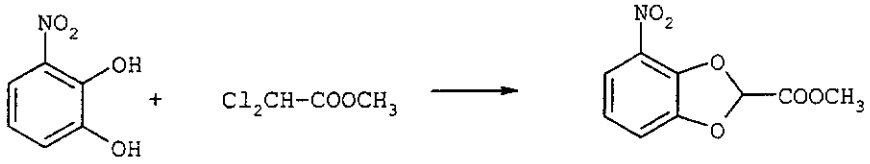
EXAMPLE 10: Compound N° 35

Step 1:



Step carried out in accordance with the operating protocol described in J.Organometall. Chem. 1996, 507, 1-21.

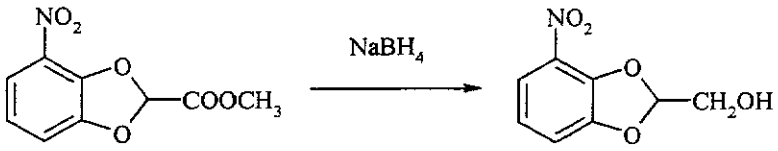
Step 2:



Racemic

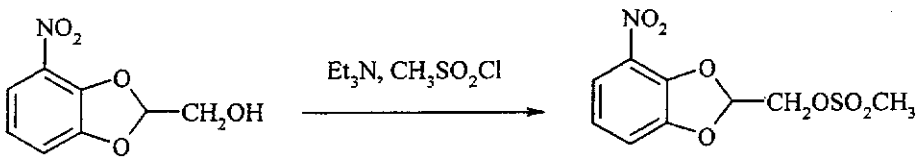
Step carried out in accordance with J. Med. Chem. 1988, 31, 84-91.

Step 3:



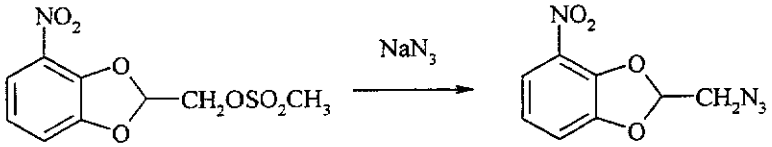
800 mg of sodium borohydride was added to 2.91 g of the product obtained in the preceding step in 40 ml of methanol, at 8°C over 35 minutes. After 50 minutes, the reaction medium was poured onto 150 ml of water/ice plus 400 ml of ethyl acetate; after stirring for 5 minutes, decanting, washing the organic phase with a 5% KHSO4/K2SO4 solution, water, a saturated NaCl solution, drying and evaporating the organic phase, a brown solid was isolated. m = 2.25 g.

Step 4:



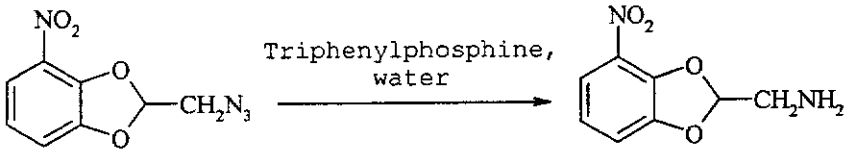
1.43 g of triethylamine was added to 2.24 g of the preceding alcohol in 60 ml of DCM, followed by 1.67 g of methane sulphonyl chloride over 25 minutes. After 45 minutes, the reaction medium was diluted with 100 ml of DCM; the organic phase was washed twice with ice water and once with a saturated NaCl solution, then dried and concentrated under vacuum. A brown solid was obtained. m = 3.01 g.

Step 5



A mixture of 3.0 g of the product obtained in step 4 and 1.82 g of sodium nitride was heated for 7h 30 mins in 20 ml of DMF at 65°C. The reaction medium was then poured onto 75 ml of ice water and 300 ml of ether. The organic phase was isolated and washed several times with water then with a saturated NaCl solution. The organic phase was dried and concentrated under vacuum to produce a brown solid. m = 2.20 g.

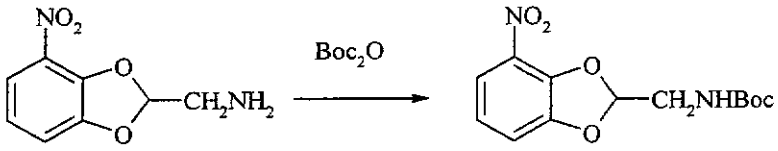
Step 6:



4.33 g of triphenylphosphine was added to 2.19 g of the product obtained in step 5, dissolved in 50 ml of ethyl acetate, over 15 minutes then, after 10 minutes, 1.8 ml of water was added over 2 minutes. The reaction medium was stirred for 1h 40 mins at 60°C then was diluted with 150 ml of ethyl acetate. The solution obtained was washed twice with water, once with a saturated NaCl solution, dried and evaporated. The residue from evaporation was dissolved in a mixture of 50 ml of ethyl acetate and 50 ml of ethyl ether and extracted twice with 50 ml of 1N HCl. The acid aqueous phases were combined and extracted using a mixture of

25 ml of ethyl acetate and 25 ml of ethyl ether, then brought into contact with 300 ml of ethyl acetate and the pH was brought to 9 with 10N NaOH. After decanting, the organic phase was washed with water, a saturated NaHCO3 solution, water, and then a saturated NaCl solution. The residual organic phase was dried then concentrated under vacuum. An oil was obtained. m = 1.40 g.

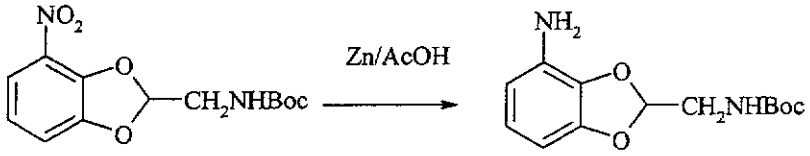
Step 7:



0.35 g of triethylamine then 1.75 g of Boc2O were added at 5°C to 1.39 g of the amine obtained in step 6 in 35 ml of DCM over a period of 10 minutes. After stirring overnight at ambient temperature, the reaction medium was diluted with 150 ml of DCM, washed with water, a 5% KHSO4/K2SO4 solution, water, then with a saturated NaCl solution. After drying and evaporating off the DCM, the solid obtained was dissolved in a minimum of ethyl ether then heptane was added to total precipitation. A solid was obtained. m = 1.90 g.

NMR: 1.30 ppm:s:9H; 3.40-3.55 ppm:mt:2H; 6.55 ppm:t:1H; 7.00 ppm:t:1H; 7.15-7.30 ppm:mt:2H; 7.55 ppm:d:1H.

Step 8:



1.96 g of powdered zinc was added to 0.60 g of the product obtained in step 7 in 25 ml of THF then, at -3°C, 2 ml of AcOH was added over a period of 25 minutes. When addition was complete, the reaction medium was allowed to return to ambient temperature. After 1h 15 mins, the reaction medium was filtered, the

filtrate was diluted with 150 ml of ethyl acetate and 30 ml of water and the pH was brought to 9 by adding a 15% Na_2CO_3 solution. After decanting, the organic phase was isolated, washed with a saturated $NaHCO_3$ solution, water, then with a saturated NaCl solution. The organic phase was isolated, dried, and then concentrated under reduced pressure. 0.53 g of a viscous yellow oil was obtained.

NMR: 1.35 ppm:s:9H; 3.30 ppm:mt:2H; 4.80 ppm:s:2H; 6.05 ppm:t:1H; 6.10-6.25 ppm:mt:2H; 6.50 ppm:t:1H; 7.10 ppm:t:1H.

Step 9:

Coupling to obtain compound 35 under standardized conditions as described above.

EXAMPLE 11: Compound N° 45

1.035 ml of DIPEA then 0.675 g of formamidinylsulphonic acid ($H_2N-C(=NH)-SO_3H$) were added to 1 g of compound 12 dissolved in 15 ml of methanol and 2 ml of DMF. After stirring overnight at 25°C, the reaction medium was supplemented with water and the precipitate formed was filtered, washed with water then dried under reduced pressure. The crude product was flash chromatographed on silica gel with a 10% to 50% MeOH gradient in DCM. 170 mg of solid was obtained, transformed into a salt using one mole/mole of H_2SO_4.
MS: $MH^+ = 596$

EXAMPLE 12: Compound N° 39

Preparation 39.1

0.11 ml of pyridine was added to 555 mg of compound 9 (Example 4) in 10 ml of DCM at 5°C, followed by 0.20 ml of triflic anhydride diluted in 2 ml of DCM over 15 minutes. After 45 minutes, the reaction medium was diluted with 50 ml of DCM, washed in succession with ice water, water, a 5% $KHSO_4/K_2SO_4$ solution, water, then a saturated NaCl solution. The organic phase was then dried and concentrated under reduced pressure. 593 mg of solid was recovered.

1H NMR: 1.40 ppm: s: 9H; 4.65 ppm: se: 2H; 6.50 ppm: t: 1H; 6.90 ppm: d: 1H; 7.40 ppm: de: 1H;

7.50 - 7.70 ppm: m: 3H; 8.00 ppm: se: 1H; 8.05 ppm: s: 1H; 8.20 ppm: se: 1H; 9.00 ppm: s: 1H; 10.15 ppm: s: 1H; 10.70 ppm: s: 1H.

Preparation 39.2

0.17 ml of morpholine was added to 579 mg of the compound obtained in step 39.1 in 10 ml of DCM. The reaction medium was stirred for 3h 30 mins at ambient temperature, and then evaporated under reduced pressure. The residue was purified by flash chromatography on silica gel with a gradient of 0 to 20% AcOEt in DCM. 300 mg of a yellow powder was obtained. MS: MH^+ = 624.

EXAMPLE 13: Compounds N°s 40-43

Compounds 40 to 43 were prepared in the same manner as compound 39 cited in Example 12, using the compound obtained in step 39.1 and replacing the morpholine in step 39.2 by tert-butylamine, 1-Boc-piperazine; cyclopropylamine cis-2,6-dimethylpiperidine respectively.

EXAMPLE 14: Compound N° 44

Compound N° 44 was obtained by de-protecting compound 41 obtained in Example 13 using the method described above employing TFA.

EXAMPLE 15: Compounds N°s 46-48

Compound 12 was reacted with an acid in its activated form, for example the anhydride, acid chloride, acid + coupling agent, for example DCCI, BOP.

Thus, compounds 46 to 48 were produced by reacting compound 12 with acetic anhydride, benzoyl chloride and cyclopropanecarbonyl chloride respectively.

EXAMPLE 16: Compound N° 49

444 mg of compound 12 in 12 ml of acetonitrile was cooled to 5°C. 0.14 ml of triethylamine was added followed by 0.075 ml of methanesulphonyl chloride. After stirring for 40 minutes at 25°C, the reaction medium was taken up in AcOEt. The organic phase was washed with water, washed with a saturated NaCl solution, dried and evaporated under reduced pressure.

The residue was purified by flash chromatography on silica gel with a gradient of 0 to 5% methanol in chloroform. 220 mg of solid was obtained. MS: MH^+ = 632.

EXAMPLE 17: Compound N° 50

40 mg of tertiobutylamine, 71 mg of diisopropylamine and 176 mg of O-(1H-benzotriazol-1-yl)-N,N,N',N'-tetramethyluronium tetrafluoroborate (TBTU) were added to 284 mg of compound 4 (Example 2) in 5 ml of DMF. The reaction medium was stirred for 1 hour at 25°C then diluted with AcOEt. The organic phase was washed successively with water, a saturated NaHCO3 solution, water, and then a saturated NaCl solution. The organic phase was dried, evaporated under reduced pressure, and the residue was purified by flash chromatography on silica gel with a gradient of 0 to 5% methanol in DCM. 196 mg of a yellow solid was obtained. MS: MH^+ = 624.

EXAMPLE 18: Compounds Nos 51-54

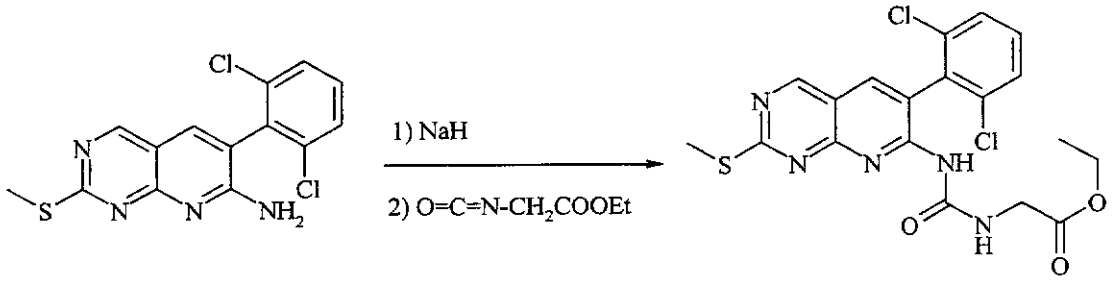
Compounds 51 to 54 were prepared in the same manner as described for compound 50 starting from compound 4 and replacing the tertiobutylamine with cyclopropylamine, pyrrolidine, N-isopropyl-methylamine and methylamine respectively.

EXAMPLE 19: Compound N° 62

819 mg of the compound described in preparation 1 and 0.314 mg of 6-amino-2,3-dihydro-benzo[b]furan, which may be prepared in accordance with Eur. J. Med. Chem. Chimica Therapeutica, 1977, Vol. 12, 231-235, were stirred for 3 hours at 70°C in 25 ml of absolute alcohol containing 0.02 ml of concentrated HCl. After cooling, the precipitate was filtered, washed with tepid MeOH then with Et2O. 637 mg of a yellow solid melting at 197°C was obtained. MS: MH^+ = 523.

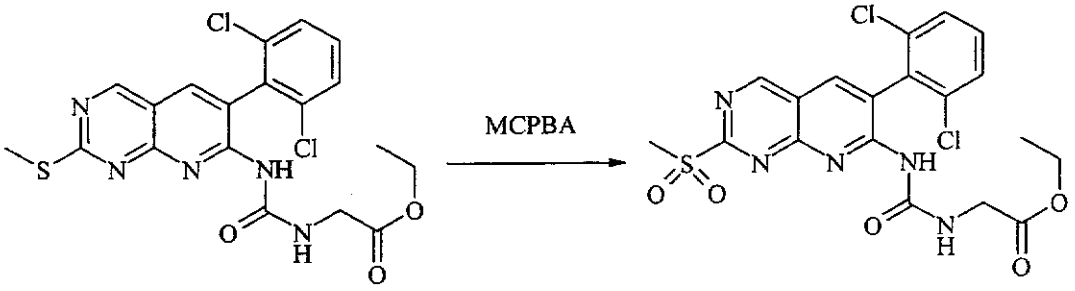
EXAMPLE 20: Compound N° 69

Step 1



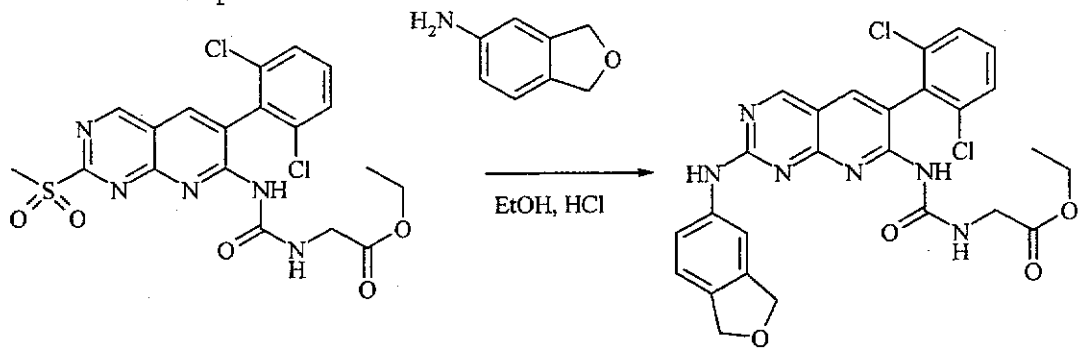
432 mg of 60% NaH was added over 10 minutes to 3.0 g of the product obtained in step 1.4 in 25 ml of DMF. After stirring for 30 minutes, 1.40 g of ethyl isocyanatoacetate was added over 10 minutes then the reaction medium was stirred for 3h 30 mins at ambient temperature. The reaction medium was extracted with AcOEt, washed successively with water, a 5% KHSO4/K2SO4 solution, water, and with a saturated NaCl solution. The organic phase was dried, concentrated under vacuum and the crude product obtained was purified by silica gel chromatography, eluting with 85/15 v/v CHCl3/AcOEt to obtain 1.52 g of the expected product.

Step 2



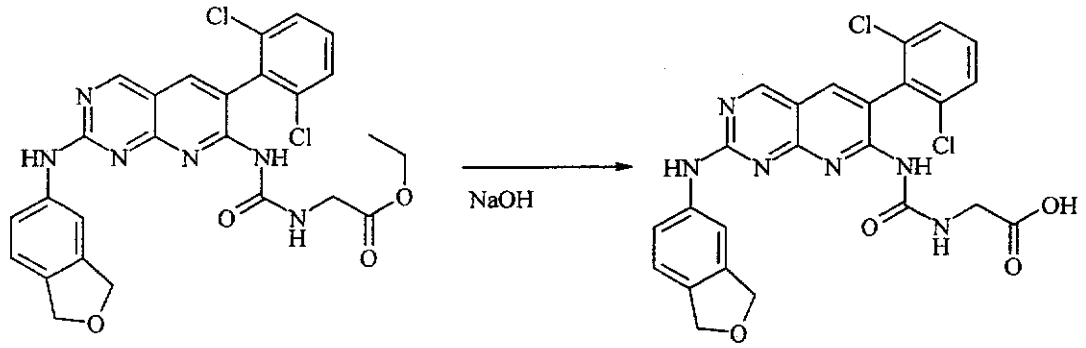
The product obtained in step 1 was oxidized with metachloroperbenzoic acid (MCPBA) using the method described in preparation 1.6. 900 mg of the expected product was obtained in the form of a beige solid.

Step 3



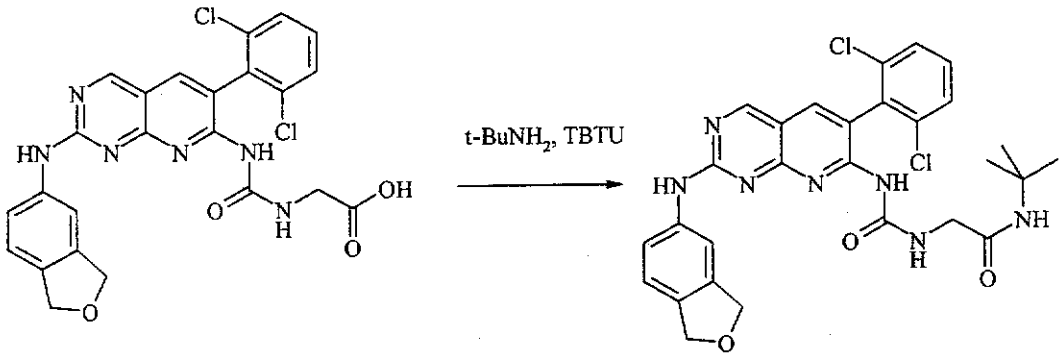
798 mg of the product from step 2 and 295 mg of 1,3-dihydro-2-benzofuran-5-amine were heated at 65°C for 6 hours in the presence of 15 ml of EtOH and 0.06 ml of concentrated HCl. The reaction medium was diluted with CHCl3 and water, then the pH of the aqueous phase was brought to 9 by adding a saturated NaHCO3 solution. After decanting, the organic phase was isolated, washed with water and then with a saturated NaCl solution, dried and concentrated under reduced pressure. The product was crystallized in AcOEt. 0.67 g of a yellow solid was obtained.

Step 4



0.66 g of the ester obtained in step 3 in 25 ml of ethanol and 2 ml of DMF were treated with 1.5 ml of 2N NaOH for 5 hours. The reaction medium was diluted with CHCl3, and then the pH was brought to 4 with 1N HCl and decanted. The organic phase was isolated, washed with water then with a saturated NaCl solution, dried and concentrated under reduced pressure to obtain 0.61 g of a yellow powder.

Step 5



105 mg of the product obtained in step 4, 16 mg of tert-butylamine, 28 mg of DIPEA (diisopropylethylamine) and 70 mg of TBTU were stirred for 1h 15 mins in 2.5 ml of DMF. The reaction medium was then extracted with CHCl3, the organic phase was washed successively with water, a 5% KHSO4/K2SO4 solution, a saturated NaCl solution and then was dried and evaporated under reduced pressure. The crude product was purified by silica chromatography, eluting with CHCl3/MeOH, 94/6 v/v. 90 mg of the expected product was obtained in the form of a yellow solid. MS: MH^+ = 580.

EXAMPLE 21: Compounds 31 and 32

The enantiomers of compound 12, prepared in Example 6, were separated by chiral stationary phase chromatography as follows:

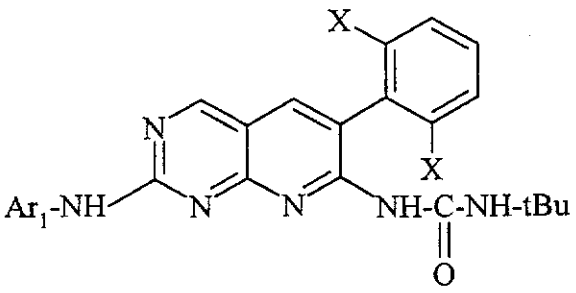
Technique	Berger Prep SFC
Detection	UV 230 nm
Stationary phase	Chiralpack AD-H 250 x 21 mm (5µm)
Mobile phase	60% / 40% CO_2 / (Ethanol + 0.5% isopropylamine)
Flow rate	50 ml/minute
Pressure	100 bar
Number of injections	Stacking, 350 injections of 28 mg

Starting from 10 g of the racemic mixture, following separation, 4.147 g of the dextrogyratory enantiomer (compound 31) and 4.077 g of the laevogyratory enantiomer (compound 32) were obtained.

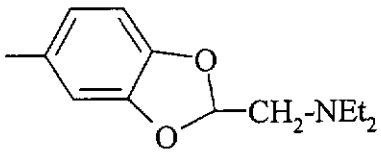
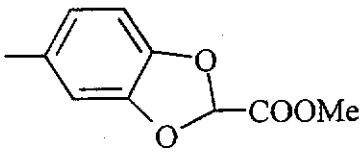
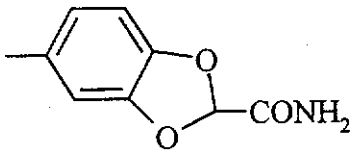
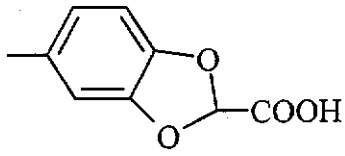
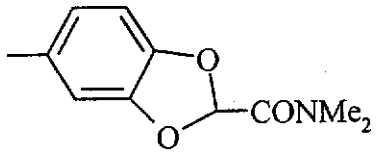
Tables 3 and 4 below illustrate the chemical structures and physical properties of some examples in

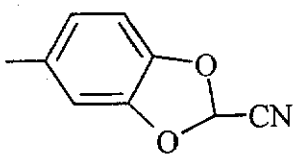
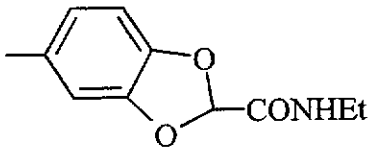
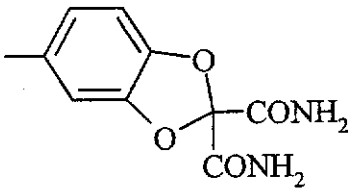
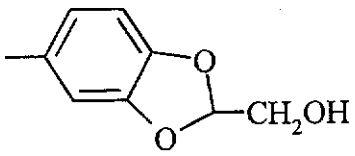
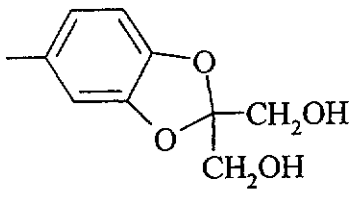
accordance with the invention. In these tables, Me, Et, iPr and tBu respectively represent methyl, ethyl, isopropyl and *tert*-butyl groups, and Boc represents the *tert*-butoxycarbonyl group. Unless otherwise indicated, products comprising an asymmetric carbon atom were obtained in the racemic form.

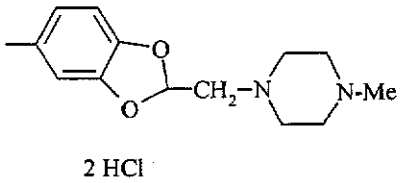
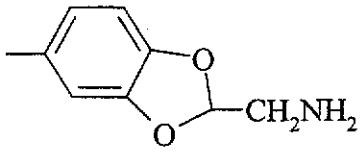
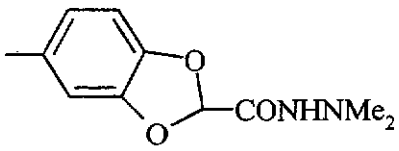
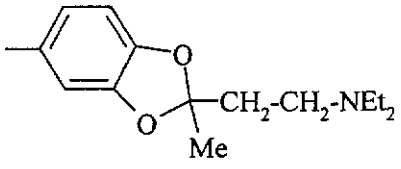
TABLE 3

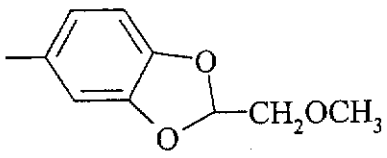
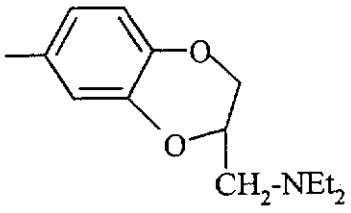
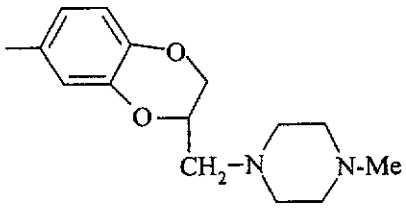
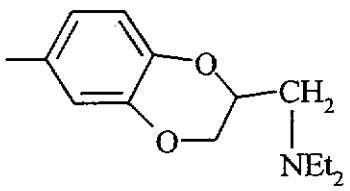


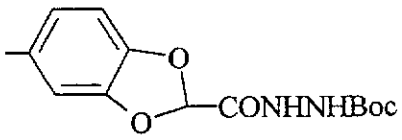
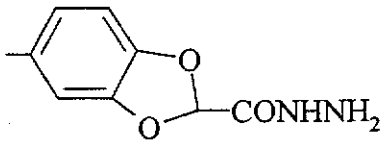
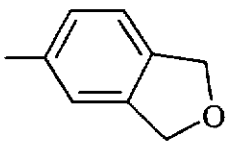
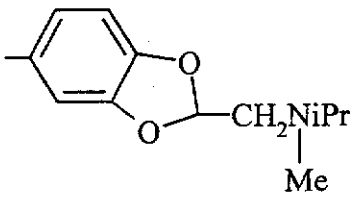
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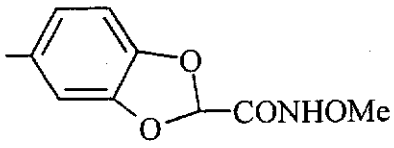
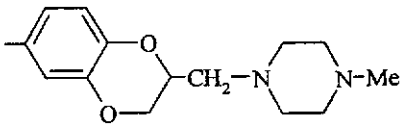
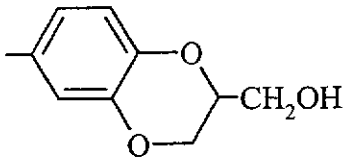
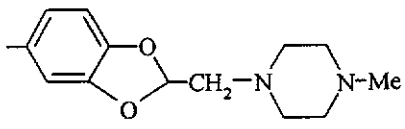
Compounds	Ar_1	X	NMR characterization
1		Cl	0.90 ppm: t: 6H; 1.40 ppm: s: 9H; 2.55 ppm: qd: 4H; 2.80 ppm: de: 2H; 6.15 ppm: t: 1H; 6.70 ppm: d: 1H; 7.20 ppm: de: 1H; 7.40-7.65 ppm: m: 3H; 7.95 ppm: se: 1H; 8.00 ppm: s: 1H; 8.10 ppm: s: 1H; 9.00 ppm: s: 1H; 10.05 ppm: s: 1H; 10.60 ppm: s: 1H.
2		Cl	1.40 ppm: s: 9H; 3.75 ppm: s: 3H; 6.65 ppm: s: 1H; 6.90 ppm: d: 1H; 7.25-7.65 ppm: m: 4H; 8.00 ppm: se: 2H; 8.20 ppm: s: 1H; 9.00 ppm: s: 1H; 10.15 ppm: s: 1H; 10.65: s: 1H.
3		Cl	1.35 ppm: s: 9H; 6.20 ppm: s: 1H; 6.75 ppm: d: 1H; 7.25-8.05 ppm: m: 9H; 9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.55 ppm: s: 1H.
4		Cl	1.45 ppm: s: 9H; 6.60 ppm: s: 1H; 7.00 ppm: d: 1H; 7.40-7.75 ppm: m: 4H; 8.10 ppm: s: 1H; 8.15 ppm: s: 1H; 8.25 ppm: s: 1H; 9.15 ppm: s: 1H; 10.20 ppm: s: 1H; 10.75 ppm: s: 1H; 14 ppm: se: 1H
5		Cl	1.45 ppm: s: 9H; 2.95 ppm: s: 3H; 3.15 ppm: s: 3H; 6.90 ppm: d: 1H; 7.05 ppm: s: 1H; 7.35 ppm: de: 1H; 7.50-7.75 ppm: m: 3H;

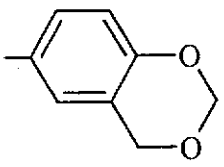
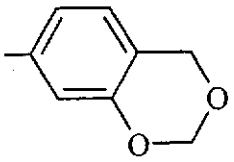
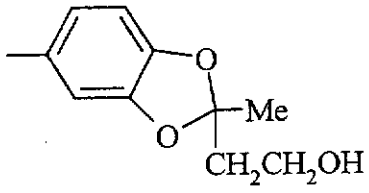
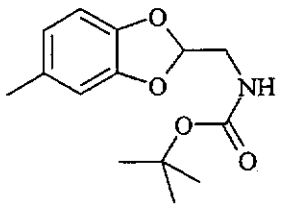
			8.05 ppm: se: 1H; 8.10 ppm: s: 1H; 8.20 ppm: s: 1H; 9.15 ppm: s: 1H; 10.20 ppm: s: 1H; 10.70 ppm: s: 1H.
6		Cl	1.40 ppm: s: 9H; 7.10 ppm: d: 1H; 7.35-7.65 ppm: m: 5H; 8.05 ppm: s: 1H; 8.20 ppm: s: 1H; 8.30 ppm: s: 1H; 9.05 ppm: s: 1H; 10.25 ppm: s: 1H; 10.75 ppm: s: 1H.
7		Cl	1.05 ppm: t: 3H; 1.40 ppm: s: 9H; 3.10 ppm: qt: 2H; 6.30 ppm: s: 1H; 6.90 ppm: d: 1H; 7.30-7.65 ppm: m: 4H; 7.85 ppm: s: 1H; 8.05 ppm: s: 1H; 8.15 ppm: s: 1H; 8.55 ppm: t: 1H; 9.05 ppm: s: 1H; 10.10 ppm: s: 1H; 10.65 ppm: s: 1H.
8		Cl	1.45 ppm: s: 9H; 6.85 ppm: d: 1H; 7.50-7.90 ppm: m: 9H; 8.05 ppm: s: 1H; 8.10 ppm: s: 1H; 9.10 ppm: s: 1H; 10.10 ppm: s: 1H; 10.60 ppm: s: 1H.
9		Cl	1.50 ppm: s: 9H; 3.80 ppm: m: 2H; 5.35 ppm: t: 1H; 6.25 ppm: t: 1H; 6.85 ppm: d: 1H; 7.40-8.25 ppm: m: 7H; 9.10 ppm: s: 1H; 10.15 ppm: s: 1H; 10.75 ppm: s: 1H.
10		Cl	1.40 ppm: s: 9H; 3.60 ppm: d: 4H; 5.15 ppm: t: 2H; 6.60 ppm: d: 1H; 7.30-7.60 ppm: m: 5H; 7.95 ppm: s: 1H; 8.05 ppm: s: 1H; 9.00 ppm: s: 1H; 9.95 ppm:

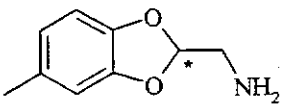
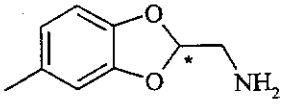
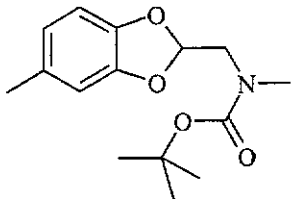
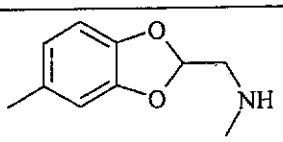
			s: 1H; 10.55 ppm: s: 1H.
11	 2 HCl	Cl	1.40 ppm: s: 9H; 2.80 ppm: s: 3H; 3.20-3.70 ppm: m: 10H; 6.65 ppm: te: 1H; 6.95 ppm: d: 1H; 7.35 ppm: de: 1H; 7.50-7.75 ppm: m: 3H; 7.95 ppm: se: 1H; 8.25 ppm: s: 1H; 9.15 ppm: s: 1H; 10.40 ppm: se: 2H; 11.50 ppm: se: 1H.
12		Cl	1.40 ppm: s: 9H; 1.55 ppm: se: 2H; 2.90 ppm: d: 2H; 6.10 ppm: t: 1H; 6.75 ppm: d: 1H; 7.30 ppm: de: 1H; 7.45-7.70 ppm: m: 3H; 7.85 ppm: se: 1H; 8.05 ppm: s: 1H; 8.10 ppm: se: 1H; 9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.70 ppm: s: 1H.
13		Cl	1.40 ppm: s: 9H; 2.45 ppm: s: 6H; 6.20 ppm: s: 1H; 6.80-6.95 ppm: m: 2H; 7.30-7.70 ppm: m: 4H; 7.85-8.15 ppm: m: 3H; 9.00 ppm: s: 1H; 10.10 ppm: s: 1H; 10.70 ppm: s: 1H.
14		Cl	0.95 ppm: t: 6H; 1.40 ppm: s: 9H; 1.60 ppm: s: 3H; 1.95-2.05 ppm: m: 2H; 2.35-2.70 ppm: m: 6H; 6.70 ppm: d: 1H; 7.20 ppm: de: 1H; 7.40-7.70 ppm: m: 3H; 7.90 ppm: se: 1H; 8.00 ppm: s: 1H; 8.10 ppm: s: 1H; 9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.60 ppm: s: 1H.

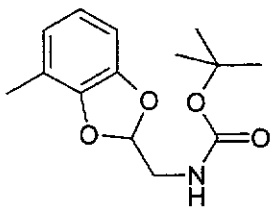
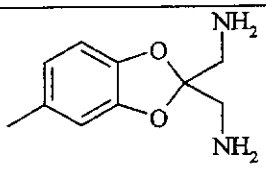
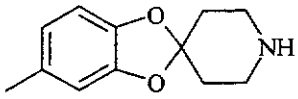
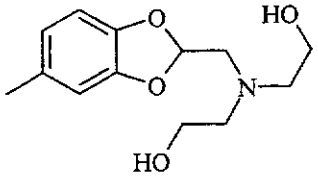
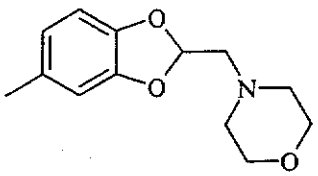
15		Cl	1.40 ppm: s: 9H; 3.40 ppm: s: 3H; 3.65 ppm: d: 2H; 6.30 ppm: t: 1H; 6.75 ppm: d: 1H; 7.30 ppm: dd: 1H; 7.45-7.60 ppm: m: 3H; 7.85 ppm: s: 1H; 8.05 ppm: s: 1H; 8.10 ppm: s: 1H; 9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.70 ppm: s: 1H.
16		Cl	0.85 ppm: t: 6H; 1.35 ppm: s: 9H; 2.40-2.60 ppm: m: 6H; 3.80-4.20 ppm: m: 3H; 6.65 ppm: d: 1H; 7.15 ppm: de: 1H; 7.40-7.60 ppm: m: 3H; 7.80 ppm: se: 1H; 7.95 ppm: s: 1H; 8.00 ppm: se: 1H; 8.95 ppm: s: 1H; 9.95 ppm: s: 1H; 10.50 ppm: s: 1H.
17		Cl	1.50 ppm: s: 9H; 2.20 ppm: s: 3H; 2.40-2.60 ppm: m: 10H; 3.90-4.40 ppm: m: 3H; 6.80 ppm: d: 1H; 7.35 ppm: de: 1H; 7.55-7.75 ppm: m: 3H; 8.00 ppm: se: 1H; 8.10 ppm: s: 1H; 8.20 ppm: s: 1H; 9.10 ppm: s: 1H; 10.10 ppm: s: 1H; 10.65 ppm: s: 1H.
18		Cl	0.95 ppm: t: 6H; 1.40 ppm: s: 9H; 2.40-2.70 ppm: m: 6H; 3.90-4.35 ppm: m: 3H; 6.75 ppm: d: 1H; 7.35-7.65 ppm: m: 4H; 7.90 ppm: se: 1H; 8.05 ppm: s: 1H; 8.10 ppm: s: 1H; 9.05 ppm: s: 1H; 10 ppm: s: 1H; 10.60 ppm: s: 1H.

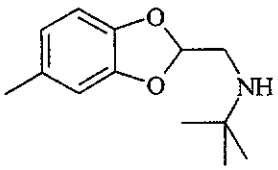
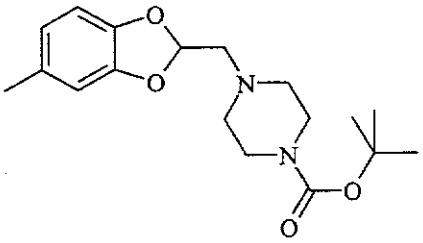
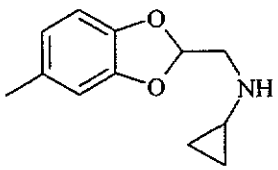
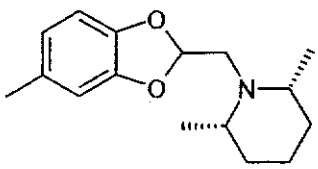
19		Cl	1.40 ppm: s: 9H; 1.45 ppm: s: 9H; 6.45 ppm: s: 1H; 6.85 ppm: s: 1H; 7.40-7.70 ppm: m: 4H; 7.90 ppm: se: 1H; 8.05 ppm: s: 1H; 8.15 ppm: s: 1H; 8.90 ppm: se: 1H; 9.05 ppm: s: 1H; 10.10 ppm: s: 1H; 10.40 ppm: s: 1H; 10.70 ppm: s: 1H.
20		Cl	1.40 ppm: s: 9H; 4.50 ppm: se: 2H; 6.45 ppm: s: 1H; 6.60 ppm: d: 1H; 7.30-7.70 ppm: m: 4H; 7.85 ppm: se: 1H; 8.00 ppm: s: 1H; 8.10 ppm: s: 1H; 9.00 ppm: s: 1H; 9.80 ppm: s: 1H; 10.10 ppm: s: 1H; 10.70 ppm: s: 1H.
21		Cl	1.45 ppm: s: 9H; 5.00 ppm: s: 4H; 7.25 ppm: d: 1H; 7.45-7.75 ppm: m: 4H; 8.05 ppm: s: 1H; 8.10 ppm: se: 1H; 8.35 ppm: se: 1H; 9.10 ppm: s: 1H; 10.10 ppm: se: 1H; 10.70 ppm: se: 1H.
22		Cl	0.95 ppm: d: 6H; 1.40 ppm: s: 9H; 2.25 ppm: s: 3H; 2.70-2.85 ppm: m: 3H; 6.20 ppm: t: 1H; 6.75 ppm: d: 1H; 7.25 ppm: de: 1H; 7.45-7.65 ppm: m: 3H;7.95 ppm: se: 1H; 8.05 ppm: s: 1H; 8.10 ppm: se: 1H; 9.00 ppm: s: 1H; 10.10 ppm: se: 1H; 10.70 ppm: se: 1H.

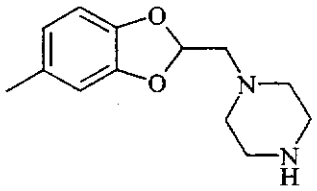
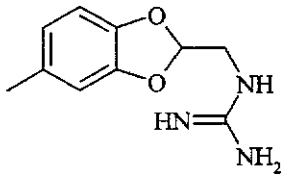
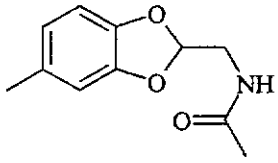
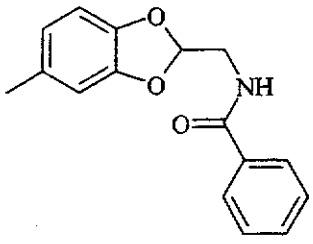
23		Cl	1.45 ppm: s: 9H; 3.60 ppm: s: 3H; 6.35 ppm: s: 1H; 6.75 ppm: d: 1H; 7.35-7.65 ppm: m: 4H; 7.90 ppm: se: 1H; 8.05 ppm: s: 1H; 8.15 ppm: se: 1H; 9.00 ppm: s: 1H; 10.10 ppm: se: 1H; 10.60 ppm: se: 1H; 11.90 ppm: se: 1H.
24		Cl	1.40 ppm: s: 9H; 2.20-2.60 ppm: m: 13H;3.80-4.30 ppm: m: 3H; 6.75 ppm: d: 1H; 7.30 ppm: de: 1H; 7.45-7.70 ppm: m: 3H; 7.85 ppm: se: 1H; 8.00 ppm: s: 1H; 8.10 ppm: se: 1H; 9.00 ppm: s: 1H; 10.05 ppm: se: 1H; 10.70 ppm: se: 1H.
25		Cl	1.45 ppm: s: 9H; 3.55-3.65 ppm: m: 2H;3.90-4.30 ppm: m: 3H; 5.05 ppm: t: 1H; 6.75 ppm: d: 1H; 7.35 ppm: dd: 1H; 7.45-7.70 ppm: m: 3H; 7.85 ppm: se: 1H; 8.05 ppm: s: 1H; 8.10 ppm: se: 1H; 9.00 ppm: s: 1H; 10.00 ppm: se: 1H; 10.70 ppm: se: 1H.
26		Me	1.40 ppm: s: 9H; 2.00 ppm: s: 6H; 2.20 ppm: s: 3H; 2.40-2.75 ppm: m: 10H; 6.25 ppm: t: 1H; 6.45 ppm: s: 1H; 6.75 ppm: d: 1H; 7.15-7.40 ppm: m: 4H; 7.95 ppm: s: 1H; 8.05 ppm: se: 1H; 9.00 ppm: s: 1H; 10.00 ppm: se: 1H; 10.50 ppm: se: 1H.

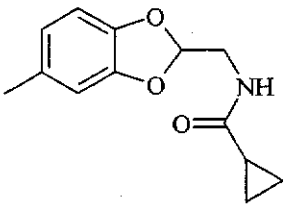
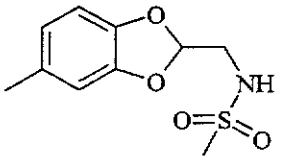
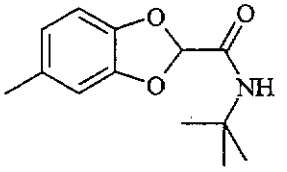
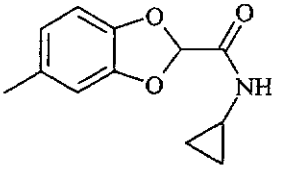
27		Cl	1.40 ppm: s: 9H; 4.80 ppm: s: 2H; 5.25 ppm: s: 2H; 6.70 ppm: d: 1H; 7.45-7.70 ppm: m: 4H; 7.90-8.10 ppm: m: 3H; 9.00 ppm: s: 1H; 10.00 ppm: se: 1H; 10.50 ppm: se: 1H.
28		Cl	1.40 ppm: s: 9H; 4.80 ppm: s: 2H; 5.25 ppm: s: 2H; 6.90 ppm: d: 1H; 7.40-7.70 ppm: m: 4H; 7.90 ppm: se: 1H; 8.05 ppm: s: 1H; 8.15 ppm: se: 1H; 9.10 ppm: s: 1H; 10.20 ppm: se: 1H; 10.70 ppm: se: 1H.
29		Cl	1.40 ppm: s: 9H; 1.60 ppm: s: 3H; 2.05 ppm: t: 2H; 3.60 ppm: t: 2H; 3.60 ppm: se: 1H; 6.70 ppm: d: 1H; 7.20 ppm: de: 1H; 7.40-7.70 ppm: m: 3H; 7.80 ppm: se: 1H; 8.00 ppm: s: 1H; 8.20 ppm: se: 1H; 9.00 ppm: s: 1H; 10.00 ppm: se: 1H; 10.70 ppm: se: 1H.
30		Cl	1.30 ppm: s: 9H; 1.40 ppm: s: 9H; 3.30 ppm-3.40 ppm: mt: 2H; 6.15 ppm: t: 1H; 6.75 ppm: d: 1H; 7.15 ppm: t: 1H; 7.35 ppm: de: 1H; 7.45-7.65 ppm: m: 3H; 7.85 ppm: se: 1H; 8.05 ppm: s: 1H; 8.15 ppm: se: 1H; 9.00 ppm: s: 1H; 10.05 ppm: s: 1H; 10.65 ppm: s: 1H.

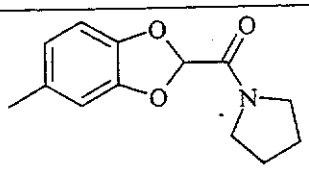
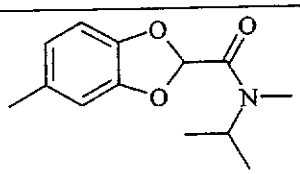
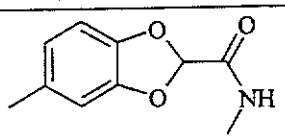
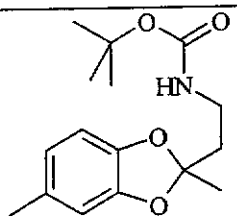
31	 Dextrogyratory: $[\alpha]_D$ = +67,6°, (C =0,5; MeOH)	Cl	1.40 ppm: s: 9H; 1.55 ppm: se: 2H; 2.90 ppm: d: 2H; 6.10 ppm: t: 1H; 6.75 ppm: d: 1H; 7.30 ppm: de: 1H; 7.45-7.70 ppm: m: 3H; 7.85 ppm: se: 1H; 8.05 ppm: s: 1H; 8.10 ppm: se: 1H; 9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.70 ppm: s: 1H.
32	 Laevogyratory: $[\alpha]_D$ = -67 (C =0,5; MeOH)	Cl	1.40 ppm: s: 9H; 1.55 ppm: se: 2H; 2.90 ppm: d: 2H; 6.10 ppm: t: 1H; 6.75 ppm: d: 1H; 7.30 ppm: de: 1H; 7.45-7.70 ppm: m: 3H; 7.85 ppm: se: 1H; 8.05 ppm: s: 1H; 8.10 ppm: se: 1H; 9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.70 ppm: s: 1H.
33		Cl	1.35 ppm: s: 9H; 1.40 ppm: s: 9H; 2.85 ppm: se: 3H; 3.60 ppm: d: 2H; 6.30 ppm: t: 1H; 6.80 ppm: d: 1H; 7.30 ppm: de: 1H; 7.45-7.70 ppm: m: 3H; 8.00 ppm: se: 1H; 8.05 ppm: s: 1H; 8.15 ppm: se: 1H; 9.00 ppm: s: 1H; 10.10 ppm: s: 1H; 10.70 ppm: s: 1H.
34		Cl	1.40 ppm: s: 9H; 1.80 ppm: se: 1H; 2.30 ppm: s: 3H; 2.85 ppm: d: 2H; 6.20 ppm: t: 1H; 6.75 ppm: d: 1H; 7.30 ppm: dd: 1H; 7.45-7.65 ppm: m: 3H; 7.85 ppm: se: 1H; 8.00 ppm: s: 1H; 8.10 ppm: se: 1H; 9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.70 ppm: s: 1H.

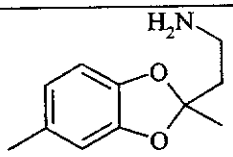
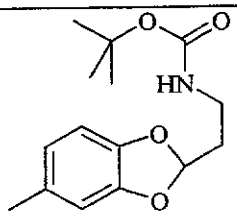
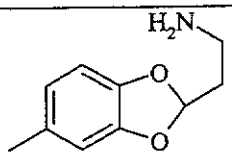
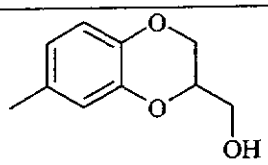
35		Cl	1.35 ppm: s: 9H; 1.40 ppm: s: 9H; 3.25 ppm-3.40 ppm: m: 2H; 6.25 ppm: t: 1H; 6.65 ppm-6.80 ppm: mt: 2H; 7.10 ppm: t: 1H; 7.50 ppm-7.65 ppm: m: 4H; 8.05 ppm: s: 1H; 8.10 ppm: s: 1H; 9.05 ppm: s: 1H; 9.20 ppm: s: 1H; 10.40 ppm: s: 1H.
36		Cl	1.40-1.60 ppm: m + s: 13H; 2.90 ppm: s: 4H; 6.70 ppm: d: 1H; 7.30 ppm: de: 1H; 7.45 ppm-7.60 ppm: se: 4H; 8.05 ppm: s: 1H; 8.20 ppm: se: 1H; 9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.65 ppm: s: 1H.
37		Cl	LC/MS: MH^+ = 594
38		Cl	1.45 ppm: s: 9H; 2.70 ppm: t: 4H; 3.00 ppm: d: 2H; 3.45 ppm: qd: 4H; 4.35 ppm: t: 2H; 6.20 ppm: t: 1H; 6.75 ppm: d: 1H; 7.25 ppm: d: 1H; 7.50 ppm-7.70 ppm: m: 3H; 7.90 ppm: se: 1H; 8.05 ppm: s: 1H; 8.10 ppm: se: 1H; 9.05 ppm: s: 1H; 10.10 ppm: s: 1H; 10.70 ppm: s: 1H.
39		Cl	1.40 ppm: s: 9H; 2.60 ppm: te: 4H; 2.80 ppm: te: 2H; 3.55 ppm: t: 4H; 6.30 ppm: t: 1H; 6.80 ppm: d: 1H; 7.25 ppm: de: 1H; 7.45 ppm-7.70 ppm: m: 3H; 8.00 ppm: se: 1H; 8.05 ppm:

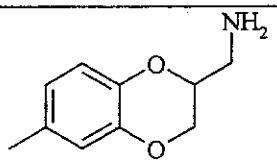
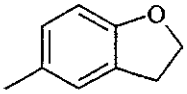
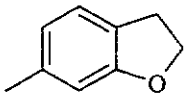
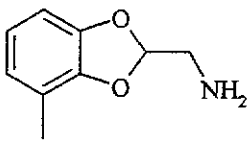
			s: 1H; 8.15 ppm: se: 1H; 9.00 ppm: s: 1H; 10.10 ppm: s: 1H; 10.75 ppm: s: 1H.
40		Cl	1.00 ppm: s: 9H; 1.45 ppm: s: 9H; 2.90 ppm: d: 2H; 6.10 ppm: t: 1H; 6.75 ppm: d: 1H; 7.30 ppm: de: 1H; 7.40 ppm-7.65 ppm: m: 3H; 7.85 ppm: se: 1H; 8.05 ppm: s: 1H; 8.15 ppm: se: 1H; 9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.65 ppm: s: 1H.
41		Cl	1.35 ppm: s: 9H; 1.45 ppm: s: 9H; 2.45 ppm-2.60 ppm: mt: 4H; 2.85: te: 2H; 3.25-3.35 ppm: mt: 4H; 6.30 ppm: t: 1H; 6.80 ppm: d: 1H; 7.25 ppm: de: 1H; 7.45 ppm-7.70 ppm: m: 3H; 8.05 ppm: s: 2H; 8.15 ppm: se: 1H; 9.05 ppm: s: 1H; 10.10 ppm: s: 1H; 10.70 ppm: s: 1H.
42		Cl	0.20 ppm: mt: 2H; 0.35 ppm: mt: 2H; 1.40 ppm: s: 9H; 2.20 ppm: mt: 1H; 2.35 ppm: se: 1H; 2.95 ppm: d: 2H; 6.20 ppm: t: 1H; 6.75 ppm: d: 1H; 7.25 ppm: de: 1H; 7.45 ppm-7.60 ppm: m: 3H; 7.85 ppm: se: 1H; 8.05 ppm: s: 1H; 8.10 ppm: se: 1H; 9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.65 ppm: s: 1H.
43	 Cis	Cl	1.05 ppm: s: 3H; 1.10 ppm: s: 3H; 1.40 ppm: s: 9H; 1.10-1.60 ppm: m: 6H; 2.45-2.65 ppm: m: 2H; 3.00 ppm: d: 2H; 6.15 ppm: t: 1H;

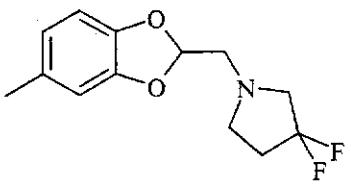
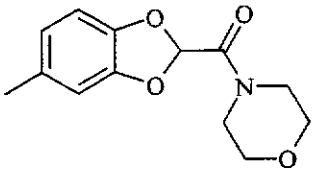
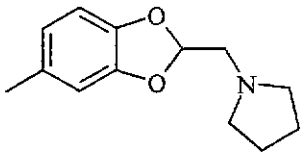
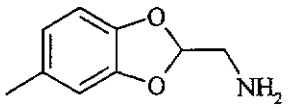
			6.75 ppm: d: 1H; 7.25 ppm: de: 1H; 7.45 ppm-7.65 ppm: m: 3H; 7.95 ppm: se: 1H; 8.05 ppm: s: 1H; 8.10 ppm: se: 1H; 9.00 ppm: s: 1H; 10.05 ppm: s: 1H; 10.70 ppm: s: 1H.
44		Cl	(DMSO + TFA) 1.30 ppm: s: 9H; 3.40-3.90 ppm: m: 10H; 6.75 ppm: t: 1H; 6.95 ppm: d: 1H; 7.25 ppm: de: 1H; 7.55 ppm-7.75 ppm: m: 4H; 8.55 ppm: se: 1H; 9.20 ppm: s: 1H.
45		Cl	(DMSO + TFA) 1.35 ppm: s: 9H; 3.75 ppm: d: 2H; 6.35 ppm: t: 1H; 6.90 ppm: d: 1H; 7.35 ppm: de: 1H; 7.45 ppm-7.75 ppm: m: 4H; 8.40 ppm: s: 1H; 9.25 ppm: s: 1H.
46		Cl	1.40 ppm: s: 9H; 1.80 ppm: s: 3H; 3.50 ppm: de: 2H; 6.15 ppm: t: 1H; 6.80 ppm: d: 1H; 7.35 ppm: de: 1H; 7.45 ppm-7.65 ppm: m: 3H; 7.85 ppm: se: 1H; 8.00 ppm: s: 1H; 8.10 ppm: se: 1H; 8.20 ppm: t: 1H; 9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.70 ppm: s: 1H.
47		Cl	1.40 ppm: s: 9H; 3.70 ppm: te: 2H; 6.35 ppm: t: 1H; 6.80 ppm: d: 1H; 7.30 ppm: de: 1H; 7.35 ppm-7.70 ppm: m: 6H; 7.85 ppm: d: 2H; 8.00 ppm: se: 1H; 8.05 ppm: s: 1H; 8.10 ppm: se: 1H; 8.80 ppm: t: 1H; 9.05 ppm:

			s: 1H; 10.10 ppm: s: 1H; 10.65 ppm: s: 1H.
48		Cl	0.60 - 0.75 ppm: m: 4H; 1.40 ppm: s: 9H; 1.60 ppm: mt: 1H; 3.55 ppm: mt: 2H; 6.20 ppm: t: 1H; 6.70 ppm: d: 1H; 7.35 ppm: de: 1H; 7.45 ppm-7.70 ppm: m: 3H; 7.85 ppm: se: 1H; 8.00 ppm: se: 1H; 8.10 ppm: se: 1H; 8.40 ppm: t: 1H; 9.05 ppm: s: 1H; 10.05 ppm: s: 1H; 10.70 ppm: s: 1H.
49		Cl	1.45 ppm: s: 9H; 2.95 ppm: s: 3H; 3.40 ppm: mt: 2H; 6.25 ppm: t: 1H; 6.80 ppm: d: 1H; 7.40 ppm: de: 1H; 7.45 ppm-7.65 ppm: m: 4H; 7.85 ppm: se: 1H; 8.05 ppm: s: 1H; 8.15 ppm: se: 1H; 9.00 ppm: s: 1H; 10.10 ppm: s: 1H; 10.60 ppm: s: 1H.
50		Cl	1.25 ppm: s: 9H; 1.40 ppm: s: 9H; 6.25 ppm: s: 1H; 6.80 ppm: d: 1H; 7.30 ppm: de: 1H; 7.45 ppm-7.65 ppm: m: 3H; 7.85 ppm: se: 1H; 8.05 ppm: s: 1H; 8.10 ppm: se: 2H; 9.00 ppm: s: 1H; 10.10 ppm: s: 1H; 10.75 ppm: s: 1H.
51		Cl	0.40 - 0.70 ppm: mt: 4H; 1.40 ppm: s: 9H; 2.70 ppm: mt: 1H; 6.30 ppm: s: 1H; 6.80 ppm: d: 1H; 7.40 ppm: de: 1H; 7.45 ppm-7.65 ppm: m: 3H; 7.85 ppm: se: 1H; 8.05 ppm: s: 1H; 8.15 ppm: se: 1H; 8.60 ppm: d: 1H;

			9.05 ppm: s: 1H; 10.10 ppm: s: 1H; 10.65 ppm: s: 1H.
52		Cl	1.40 ppm: s: 9H; 1.70-1.95 ppm: m: 4H; 3.35 ppm: t: 2H; 3.60 ppm: t: 2H; 6.75 ppm: s: 1H; 6.85 ppm: d: 1H; 7.30 ppm: de: 1H; 7.45 ppm-7.65 ppm: m: 3H; 8.00 ppm: se: 1H; 8.05 ppm: s: 1H; 8.15 ppm: se: 1H; 9.05 ppm: s: 1H; 10.10 ppm: s: 1H; 10.65 ppm: s: 1H.
53		Cl	1.05 ppm: d: 3H; 1.20 ppm: d: 3H; 1.40 ppm: s: 9H; 2.70 ppm and 2.90 ppm: 2s: 3H; 4.20-4.60 ppm: m: 1H; 6.80-7.05 ppm: mt: 2H; 7.30 ppm: de: 1H; 7.45 ppm-7.65 ppm: m: 3H; 8.00 ppm: se: 1H; 8.05 ppm: s: 1H; 8.15 ppm: se: 1H; 9.05 ppm: s: 1H; 10.15 ppm: s: 1H; 10.60 ppm: s: 1H.
54		Cl	1.40 ppm: s: 9H; 2.65 ppm: d: 3H; 6.35 ppm: s: 1H; 6.85 ppm: d: 1H; 7.40 ppm: de: 1H; 7.45 ppm-7.65 ppm: m: 3H; 7.85 ppm: se: 1H; 8.00 ppm: s: 1H; 8.15 ppm: se: 1H; 8.50 ppm: qd: 1H; 9.05 ppm: s: 1H; 10.10 ppm: s: 1H; 10.60 ppm: s: 1H.
55		Cl	1.35 ppm: s: 9H; 1.40 ppm: s: 9H; 1.60 ppm: s: 3H; 2.65 ppm: t: 2H; 3.00-3.20 ppm: mt: 2H; 6.70 ppm: d: 1H; 6.80 ppm: t: 1H; 7.20 ppm: de: 1H; 7.45 ppm-7.70 ppm: m: 3H;

			7.80 ppm: se: 1H; 8.05 ppm: s: 2H; 9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.60 ppm: s: 1H.
56		Cl	1.40 ppm: s: 9H; 1.60 ppm: s: 3H; 2.00 ppm: t: 2H; 2.70 ppm: t: 2H; 6.70 ppm: d: 1H; 7.20 ppm: de: 1H; 7.50 ppm-7.70 ppm: m: 4H; 7.80 ppm: se: 1H; 8.05 ppm: s: 1H; 9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.70 ppm: s: 1H.
57		Cl	1.35 ppm: s: 9H; 1.45 ppm: s: 9H; 2.00 ppm: qd: 2H; 3.10: qd: 2H; 6.20 ppm: t: 1H; 6.75 ppm: d: 1H; 6.95 ppm: t: 1H; 7.30 ppm: de: 1H; 7.45 ppm-7.65 ppm: m: 3H; 7.90 ppm: se: 1H; 8.05 ppm: s: 1H; 8.10 ppm: se: 1H; 9.05 ppm: s: 1H; 10.10 ppm: s: 1H; 10.65 ppm: s: 1H.
58		Cl	(DMSO + TFA) 1.25 ppm: s: 9H; 2.20 ppm: qd: 2H; 3.00 ppm: t: 2H; 6.30 ppm: t: 1H; 6.75 ppm: d: 1H; 7.25 ppm: de: 1H; 7.50-7.70 ppm: m: 4H; 8.50 ppm: s: 1H; 9.15 ppm: s: 1H.
59		Cl	1.45 ppm: s: 9H; 3.50-4.35 ppm: m: 5H; 5.05 ppm: t: 1H; 6.75 ppm: d: 1H; 7.30 ppm: de: 1H; 7.45 ppm-7.70 ppm: m: 3H; 7.90 ppm: se: 1H; 8.05 ppm: s: 1H; 8.10 ppm: se: 1H;

			9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.60 ppm: s: 1H.
60		Cl	1.40 ppm: s: 9H; 2.70-2.85 ppm: m: 2H; 3.90-4.05 ppm: m: 2H; 4.35 ppm: d: 1H; 6.75 ppm: d: 1H; 7.35 ppm: de: 1H; 7.45 ppm-7.70 ppm: m: 4H; 7.85 ppm: se: 1H; 8.05 ppm: s: 1H; 9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.60 ppm: s: 1H.
61		Cl	1.40 ppm: s: 9H; 3.15 ppm: t: 2H; 4.50 ppm: t: 2H; 6.65 ppm: d: 1H; 7.45 ppm-7.65 ppm: m: 4H; 8.00 ppm: s: 1H; 8.05 ppm: se: 1H; 8.30 ppm: se: 1H; 9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.70 ppm: s: 1H.
62		Cl	1.40 ppm: s: 9H; 3.10 ppm: t: 2H; 4.50 ppm: t: 2H; 7.10 ppm: d: 1H; 7.35 ppm: dd: 1H; 7.45 ppm-7.70 ppm: m: 3H; 7.75 ppm: s: 1H; 8.05 ppm: s: 1H; 8.15 ppm: s: 1H; 9.05 ppm: s: 1H; 10.10 ppm: s: 1H; 10.65 ppm: s: 1H.
63		Cl	1.35 ppm: s: 9H; 1.40-1.75 ppm: m: 2H; 2.90 ppm: d: 2H; 6.15 ppm: t: 1H; 6.60 ppm-6.80 ppm: mt: 2H; 7.40 ppm-7.65 ppm: mt: 4H; 8.05 ppm: s: 2H; 9.05 ppm: s: 1H; 9.35 ppm: s: 1H; 10.45 ppm: s: 1H.

64		Cl	1.40 ppm: s: 9H; 2.10-2.35 ppm: mt: 2H; 2.80 ppm-3.15 ppm: m: 6H; 6.30 ppm: t: 1H; 6.80 ppm: d: 1H; 7.25 ppm: de: 1H; 7.45 ppm-7.70 ppm: m: 3H; 8.05 ppm: s: 2H; 8.15 ppm: s: 1H; 9.05 ppm: s: 1H; 10.10 ppm: s: 1H; 10.60 ppm: s: 1H.
65		Cl	1.40 ppm: s: 9H; 3.40-3.70 ppm: m: 8H; 6.85 ppm: d: 1H; 7.00 ppm: s: 1H; 7.30 ppm: dd: 1H; 7.45 ppm-7.70 ppm: m: 3H; 8.05 ppm: s: 2H; 8.15 ppm: s: 1H; 9.05 ppm: s: 1H; 10.10 ppm: s: 1H; 10.65 ppm: s: 1H.
66		Cl	1.40 ppm: s: 9H; 1.60-1.75 ppm: m: 4H; 2.50 ppm-2.65 ppm: m: 4H; 2.75 ppm-3.00 ppm: mt: 2H; 6.25 ppm: t: 1H; 6.75 ppm: d: 1H; 7.25 ppm: de: 1H; 7.45 ppm-7.70 ppm: m: 3H; 7.95 ppm: s: 1H; 8.05 ppm: s: 1H; 8.15 ppm: s: 1H; 9.05 ppm: s: 1H; 10.10 ppm: s: 1H; 10.70 ppm: s: 1H.
67		Br	1.40 ppm: s: 9H; 1.60 ppm: se: 2H; 2.90 ppm: d: 2H; 6.10 ppm: t: 1H; 6.75 ppm: d: 1H; 7.20 ppm-7.40 ppm: m: 2H; 7.70 ppm-7.85 ppm: d: 4H; 8.00 ppm: s: 1H; 9.05 ppm: s: 1H; 10.05 ppm: s: 1H; 10.70 ppm: s: 1H.

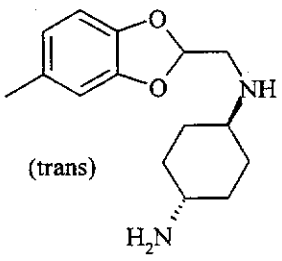
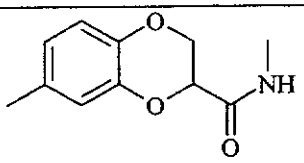
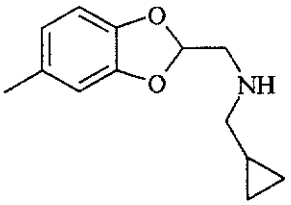
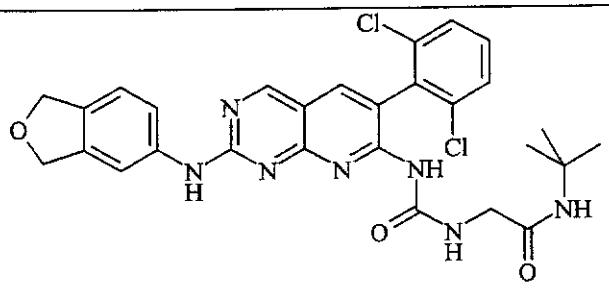
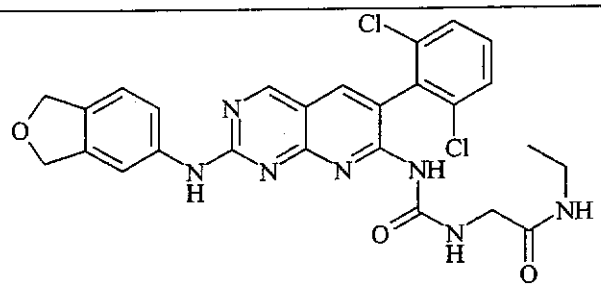
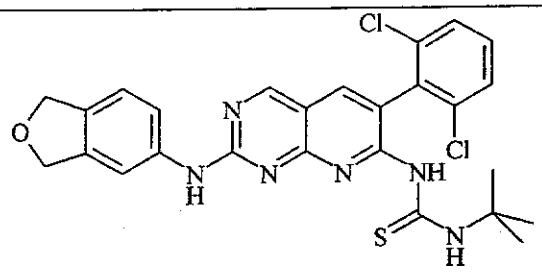
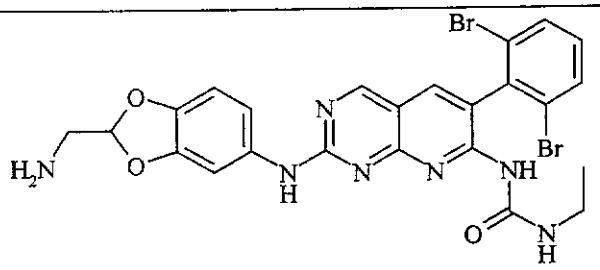
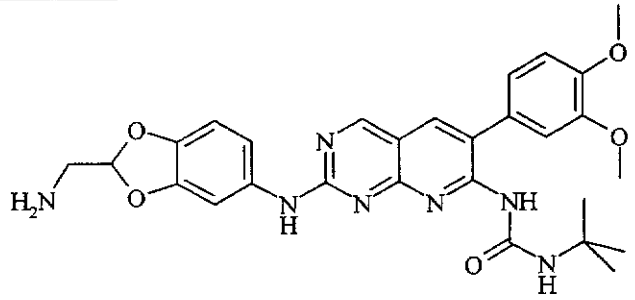
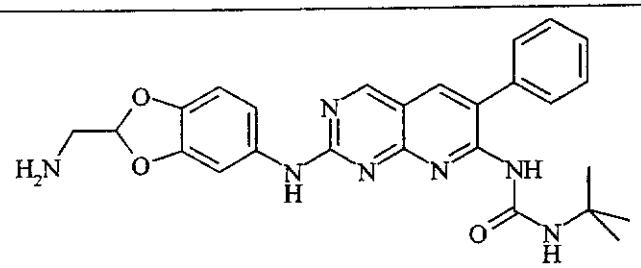
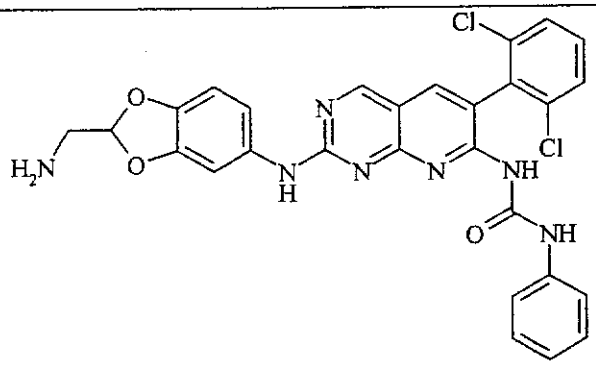
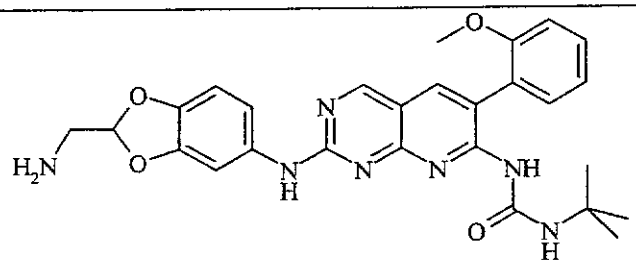
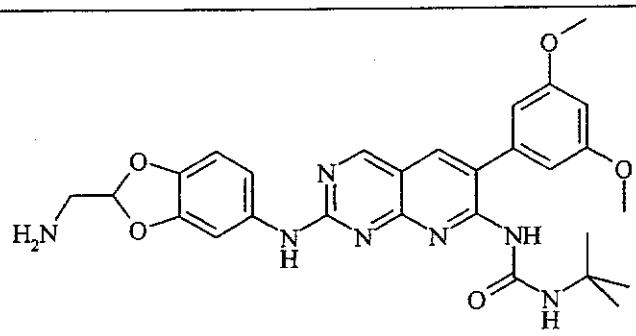
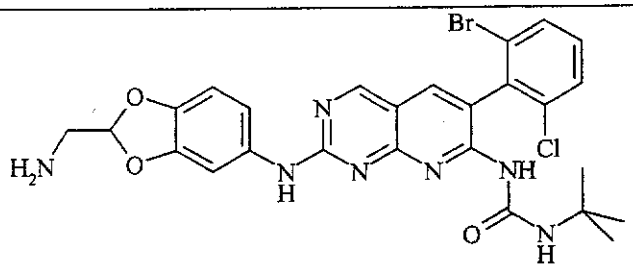
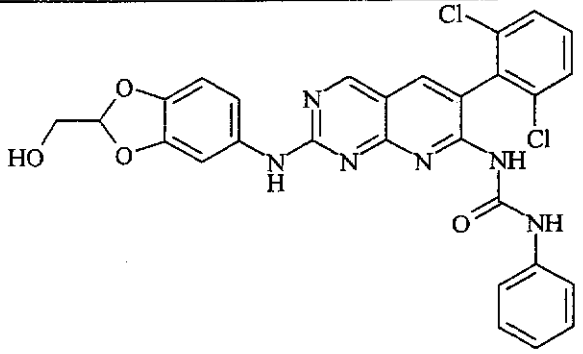
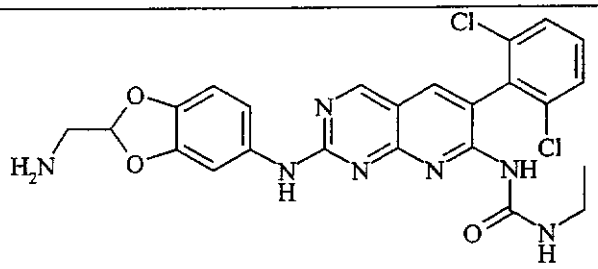
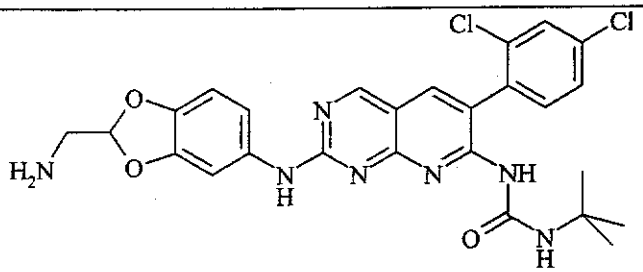
68	 (trans)	Cl	0.95 ppm: t: 4H; 1.35 ppm: s: 9H; 1.60-1.80 ppm: m: 4H; 2.25 ppm-2.45 ppm: m: 2H; 2.85 ppm: d: 2H; 6.10 ppm: t: 1H; 6.70 ppm: d: 1H; 7.25 ppm: d: 1H; 7.40 ppm-7.60 ppm: m: 3H; 7.85 ppm: se: 1H; 8.00 ppm: s: 1H; 9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.65 ppm: s: 1H.
80		Cl	1.40 ppm: s: 9H; 2.65 ppm: d: 3H; 4.10-4.35 ppm: mt: 2H; 4.65-4.75 ppm: mt: 1H; 6.80 ppm: d: 1H; 7.35 ppm: dd: 1H; 7.45-7.70 ppm: mt: 3H; 8.00-8.15 ppm: m: 4H; 9.05 ppm: s: 1H; 10.10 ppm: s: 1H; 10.60 ppm: s: 1H.
81		Cl	0.00-0.10 ppm: m: 2H; 0.30-0.40 ppm: m: 2H; 0.80-0.90 ppm: m: 1H; 1.40 ppm: s: 9H; 1.85 ppm: se: 1H; 2.40-2.50 ppm: m: 2H; 2.95 ppm: d: 2H; 6.20 ppm: t: 1H; 6.75 ppm: d: 1H; 7.25 ppm: de: 1H; 7.45-7.70 ppm: m: 3H; 7.90 ppm: s: 1H; 8.05 ppm: s: 1H; 8.20 ppm: s: 1H; 9.00 ppm: s: 1H; 10.05 ppm: s: 1H; 10.70 ppm: s: 1H.

TABLE 4

Com-pounds	Structure	NMR characerization
69		1.30 ppm: s: 9H; 3.90 d: 2H; 5.00 ppm: d: 4H; 7.25 ppm: d: 1H; 7.40 ppm-7.65 ppm: m: 4H; 7.90 ppm: d: 1H; 8.00 ppm: s: 1H; 8.10 ppm: s: 1H; 8.50 ppm: s: 1H; 9.10 ppm: s: 1H; 10.10 ppm: s: 1H; 10.25 ppm: s: 1H.
70		1.00 ppm: mt: 3H; 3.10 ppm: qt: 2H; 3.90 d: 2H; 5.00 ppm: d: 4H; 7.25 ppm: d: 1H; 7.40 ppm-7.65 ppm: m: 3H; 7.75 ppm: d: 1H; 8.05 ppm: s: 1H; 8.10 ppm: s: 1H; 8.25 ppm: s: 1H; 8.60 ppm: s: 1H; 9.10 ppm: s: 1H; 10.20 ppm: s: 1H; 10.30 ppm: s: 1H.
71		1.65 ppm: s: 9H; 5.00 ppm: s: 4H; 7.20 ppm: d: 1H; 7.50 ppm-7.70 ppm: m: 5H; 8.25 ppm: s: 1H; 8.30 ppm: s: 1H; 9.20 ppm: s: 1H; 10.40 ppm: s: 1H; 12.80 ppm: s: 1H.

72		(DMSO + TFA) 1.10 ppm: t: 3H; 3.20 ppm: qd: 2H; 3.45 ppm: d: 2H; 6.45 ppm: t: 1H; 7.00 ppm: d: 1H; 7.25 ppm: de: 1H; 7.45 ppm: t: 1H; 7.70 ppm: se: 1H; 7.85 ppm: s: 1H; 7.90 ppm: s: 1H; 8.50 ppm: s: 1H; 9.20 ppm: s: 1H.
73		1.40 ppm: s: 9H; 1.60-1.80 ppm: se: 2H; 2.95 ppm: d: 2H; 3.80 ppm: s: 3H; 3.85 ppm: s: 3H; 6.10 ppm: t: 1H; 6.75 ppm: d: 1H; 6.95 ppm-7.35 ppm: m: 5H; 7.80 ppm: s: 1H; 8.05 ppm: s: 1H; 9.05 ppm: s: 1H; 9.95 ppm: s: 1H; 10.55 ppm: s: 1H.
74		1.40 ppm: s: 9H; 1.50-1.70 ppm: se: 2H; 2.90 ppm: d: 2H; 6.10 ppm: t: 1H; 6.75 ppm: d: 1H; 7.10 ppm: se: 1H; 7.30 ppm: d: 1H; 7.50 ppm: se: 5H; 7.85 ppm: s: 1H; 8.10 ppm: s: 1H; 9.05 ppm: s: 1H; 9.95 ppm: s: 1H; 10.50 ppm: s: 1H.

75		1.30 ppm-1.80 ppm: se: 2H; 3.00 ppm: d: 2H; 6.20 ppm: t: 1H; 6.85 ppm: d: 1H; 7.10 ppm: t: 1H; 7.20 ppm: d: 1H; 7.40 ppm: t: 2H; 7.45 ppm-7.70 ppm: m: 6H; 8.10 ppm: s: 2H; 9.10 ppm: s: 1H; 10.15 ppm: s: 1H; 13.20 ppm: s: 1H.
76		1.40 ppm: s: 9H; 1.50-1.65 ppm: se: 2H; 2.90 ppm: d: 2H; 3.80 ppm: s: 3H; 6.10 ppm: t: 1H; 6.75 ppm: d: 2H; 7.10 ppm-7.40 ppm: m: 4H; 7.50 ppm: t: 1H; 7.85 ppm: s: 1H; 8.00 ppm: s: 1H; 9.05 ppm: s: 1H; 10.00 ppm: s: 1H; 10.50 ppm: s: 1H.
77		1.40 ppm: s: 9H; 1.50-1.65 ppm: s: 2H; 2.90 ppm: d: 2H; 3.80 ppm: s: 6H; 6.10 ppm: t: 1H; 6.60 ppm: s: 3H; 6.75 ppm: d: 1H; 7.20 ppm-7.35 ppm: mt: 2H; 7.80 ppm: s: 1H; 8.10 ppm: s: 1H; 9.05 ppm: s: 1H; 10.00 ppm: s: 1H; 10.50 ppm: s: 1H.

78		1.35 ppm: s: 9H; 1.50-1.70 ppm: se: 2H; 2.85 ppm: d: 2H; 6.05 ppm: t: 1H; 6.70 ppm: d: 1H; 7.20 ppm-7.80 ppm: mt: 5H; 7.95 ppm: se: 1H; 8.00 ppm: s: 1H; 9.00 ppm: s: 1H; 10.00 ppm: s: 1H; 10.60 ppm: s: 1H.
79		LC/MS: MH^+ = 575
82		(DMSO + TFA) 1.05 ppm: t: 3H; 3.20 ppm: qd: 2H; 3.40 ppm: d: 2H; 6.45 ppm: t: 1H; 6.95 ppm: d: 1H; 7.20 ppm: de: 1H; 7.50-7.75 ppm: m: 4H; 8.55 ppm: s: 1H; 9.20 ppm: s: 1H.
83		1.40 ppm: s: 9H; 1.60 ppm: se: 2H; 2.90 ppm: d: 2H; 6.05 ppm: t: 1H; 6.75 ppm: d: 1H; 7.30 ppm: d: 1H; 7.45 ppm-7.60 ppm: mt: 2H; 7.75-7.90 ppm: mt: 3H; 8.00 ppm: s: 1H; 9.00 ppm: s: 1H; 10.05 ppm: s: 1H; 10.60 ppm:

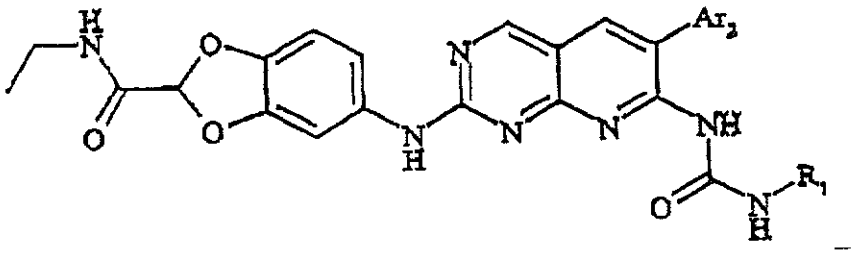
		s: 1H.

Compound 34 was separated into its two enantiomers by chiral chromatography. Dextrogyratory enantiomer : $[\alpha]_D$ = +72.1°; t = 25°C; C = 0.5 (MeOH). Laevogyratory enantiomer: $[\alpha]_D$ = -73.0°; t = 25°C; C = 0.5 (MeOH).

Databases of products according to the invention were also prepared by employing conventional techniques of combinatorial chemistry and applying the preparation procedures described in the present invention. The structures of the products prepared and their characterization are presented here.

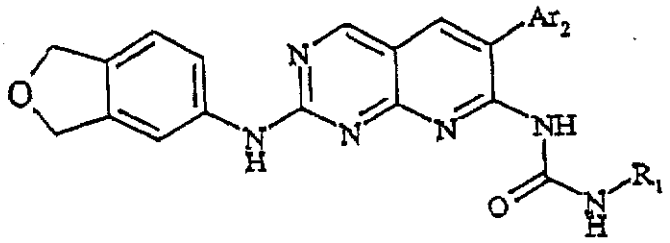
The LC/MS analyses were carried out on a Micromass model LCT device, coupled to an HP1100 model device. The extent of the compounds was measured with a G1315A diode array detector in the wavelength region from 200 to 600 nm and using a Sedex model 65 light scattering evaporative detector (LSED). The mass spectra were recorded in the range from 180 to 800 (m/z). The data were analysed using Micromass MassLynx software. The separations were performed on a Hypersil BDS C18 column (50 × 4.6 mm), particle size 3 µm. Elution with a gradient from 5 to 90% of acetonitrile comprising 0.05% (v/v) of trifluoroacetic acid (TFA) in water comprising 0.05% (v/v) of TFA over 3.5 minutes at a flow rate of 1 ml/minute. The total duration of the procedure, including the reequilibrating of the column, is 7 minutes.

TABLE 5



Compounds	R_1	Ar_2	LC/MS (retention time (minutes): MH^+; MH^-)
1	Phenyl	2,6-Dichlorophenyl	4.42; 616; 614
2	*t*-Butyl	3,5-Dimethoxyphenyl	4.24; 588; 586
3	Ethyl	2,6-Dichlorophenyl	3.77; 568; 566
4	*t*-Butyl	3,4-Dimethoxyphenyl	3.93; 588; 586
5	*t*-Butyl	Phenyl	3.80; 528; 526
6	*t*-Butyl	2-Methoxyphenyl	4.06; 558; 556
7	*t*-Butyl	2,6-Dibromophenyl	4.32; 685; 684
8	*t*-Butyl	2-Bromo-6-chlorophenyl	4.29; 640; 638
9	Ethyl	2,6-Dibromophenyl	3.83; 658; 656
10	Phenyl	2-Bromo-6-chlorophenyl	4.46; 660; 658
11	Phenyl	2,6-Dibromophenyl	4.49; -; 704

TABLE 6



Compounds	R_1	Ar_2	LC/MS (retention time (minutes): MH^+)
1	Phenyl	2,6-Dichlorophenyl	4.75; 543.28
2	*t*-Butyl	3,5-Dimethoxyphenyl	4.52; 515.42
3	Ethyl	2,6-Dichlorophenyl	4.95; 495.30
4	*t*-Butyl	3,4-Dimethoxyphenyl	4.21; 515.43
5	*t*-Butyl	Phenyl	4.42; 455.41
6	*t*-Butyl	2-Methoxyphenyl	4.34; 485.41
7	*t*-Butyl	2,6-Dibromophenyl	4.63; 611.25
8	*t*-Butyl	2-Bromo-6-chlorophenyl	4.72; 567.09
9	Ethyl	2,6-Dibromophenyl	4.14; 582.96
10	Phenyl	2-Bromo-6-chlorophenyl	4.88; 587.06
11	Phenyl	2,6-Dibromophenyl	4.91; 630.97

The compounds of the invention underwent pharmacological tests in order to determine their anti-cancer activity.

The compounds with formula (I) of the present invention were tested in vitro on a panel of tumour lines of human origin deriving from the following:

- breast cancer: MDA-MB231 (American Type Culture Collection, Rockville, Maryland, USA, ATCC-HTB26), MDA-A1 or MDA-ADR (known as the multi-drug resistant line, MDR, and described by E. Collomb et al., in Cytometry, 12(1):15-25, 1991), and MCF7 (ATCC-HTB22),
- prostate cancer: DU145 (ATCC-HTB81) and PC3 (ATCC-CRL1435),
- colon cancer: HCT116 (ATCC-CCL247) and HCT15 (ATCC-CCL225),

- lung cancer: H460 (described by Carmichael in Cancer Research 47 (4):936-942, 1987 and filed by the National Cancer Institute, Frederick Cancer Research and Development Center, Frederick, Maryland, USA),
- glioblastoma (SF268 described by Westphal in Biochemical & Biophysical Research Communications 132 (1): 284-289, 1985 and filed by the National Cancer Institute, Frederick Cancer Research and Development Center, Frederick, Maryland, USA),
- leukaemia (CMLT1 described by Kuriyama et al. in Blood, 74: 1989, 1381-1387, by Soda et al. in the British Journal of Haematology, 59: 1985, 671-679 and by Drexler, in Leukemia Research, 18: 1994, 919-927 and filed by DSMZ, Mascheroder Weg 1b, 38124 Brunswick, Germany).

Proliferation and cellular viability were determined by means of a test using 3-(4,5-dimethylthiazol-2-yl)-5-(3-carboxymethoxyphenyl)-2-(4-sulphophenyl)-2*H*-tetrazolium (MTS) in accordance with Fujishita T. et al., Oncology, 2003, 64 (4), 399-406. In this test, the mitochondrial capacity of living cells to transform MTS into a coloured compound after incubating a compound with formula (I) in accordance with the invention for 72 hours is measured. The concentrations of the compound of the invention which result in a 50% drop in proliferation and cellular viability (CI_{50}) were in the range 1 nM to 10 µM depending on the tumour line and the test compound.

Thus, in accordance with the present invention, it appears that compounds with formula (I) cause a drop in proliferation and viability of tumour cells. It thus appears that the compounds of the invention have an anti-cancer activity and an activity in the treatment of other proliferative diseases such as psoriasis, restenosis, arteriosclerosis, AIDS, for example, and in diseases caused by the proliferation of cells of the smooth vascular muscle cells and in rheumatoid polyarthritis.

Thus, in accordance with a further aspect, the invention concerns medicaments which comprise a compound with formula (I) or an addition salt of the latter with a pharmaceutically acceptable acid or a hydrate or a solvate of the compound with formula (I).

Said medicaments are of use in therapy, in particular in the treatment or prevention of diseases caused or exacerbated by the proliferation of cells, in particular tumour cells.

As an inhibitor of tumour cell proliferation, said compounds may be used in the prevention and treatment of leukaemias, both primary and metastatic solid tumours, carcinomas and cancers, in particular: breast cancer; lung cancer; cancer of the small intestine; cancer of the colon and rectum; cancer of the respiratory tracts, of the oropharynx and the hypopharynx; cancer of the oesophagus; cancer of the liver, stomach cancer, cancer of the biliary canals, cancer of the biliary vesicle, cancer of the pancreas; cancers of the urinary tracts including the kidney, urothelium and bladder; cancers of the female genital tract including cancer of the uterus, the neck of the uterus, the ovaries, chloriocarcinoma and trophoblastoma; cancers of the male genital tract including cancer of the prostate, of the seminal vesicles, the testicles, germinal cell tumours; cancers of the endocrinal glands including cancer of the thyroid, the pituitary, of the adrenal glands; skin cancers, including haemangiomas, melanomas, sarcomas, including Kaposi's sarcoma; tumours of the brain, nerves, eyes, meninges, including astrocytomas, gliomas, glioblastomas, retinoblastomas, neurinomas, neuroblastomas, schwannomas, meningiomas, malignant haematopoietic tumours; leukaemias (Acute Lymphocytic Leukemia (ALL), Acute Myeloid Leukemia (AML), Chronic Myeloid Leukemia (CML), Chronic lymphocytic leukemia (CLL)), chloromas, plasmocytomas, T or B cell leukaemias, non Hodgkins or Hodgkins lymphomas, myelomas, and various malignant haemopathies.

In a further aspect, the present invention concerns pharmaceutical compositions comprising a compound of the invention as an active principle. Said pharmaceutical compositions contain an effective dose of at least one compound in accordance with the invention or a pharmaceutically acceptable salt, a hydrate or a solvate of said compound, and at least one pharmaceutically acceptable excipient.

The choice of said excipients is dependent on the pharmaceutical form and the desired mode of administration; they are selected from the usual excipients, which are known to the skilled person.

In pharmaceutical compositions of the present invention for oral, sublingual, subcutaneous, intramuscular, intravenous, topical, local, intratracheal, intranasal, transdermal or rectal administration, the active principle with formula (I) above, or possibly its optional salt, solvate or hydrate, may be administered in a unitary administration form, mixed with conventional pharmaceutical excipients, to animals or to human beings for prophylactic treatment or for the treatment of the diseases or disorders cited above.

Appropriate unitary forms of administration comprise forms for oral administration such as tablets, soft or hard gelules, powders, granules and oral solutions or suspensions, sublingual, buccal, intratracheal, intraocular, intranasal forms of administration, forms for inhalation, forms for topical, transdermal, subcutaneous, intramuscular or intravenous administration, forms for rectal administration and implants. For topical administration, the compounds of the invention may be used in creams, gels, pomades or lotions.

An example of a unitary form of administration for a compound of the invention in the form of a tablet may comprise the following components:

Compound of the invention	50.0 mg
Mannitol	223.75 mg

Croscarmellose, sodium form	6.0 mg
Corn starch	15.0 mg
Hydroxypropylmethyl cellulose	2.25 mg
Magnesium stearate	3.0 mg

The above compounds with formula (I) may be used in daily doses of 0.002 to 2000 mg per kilogram of body weight of the mammal to be treated, preferably in daily doses of 0.1 mg to 300 mg/kg. In humans, the dose may preferably be from 0.02 to 10000 mg per day, more particularly 1 to 3000 mg depending on the age of the subject to be treated and the type of treatment: prophylactic or curative.

Particular cases may arise for which higher or lower doses are appropriate; such doses also fall within the scope of the invention. The usual practice is for the physician to determine the dosage appropriate for each patient depending on the mode of administration and the patient's weight and response.

In a further aspect, the present invention also concerns a method for treating the diseases indicated above which comprises administering to a patient an effective dose of a compound of the invention or one of its pharmaceutically acceptably salts, hydrates or solvates.

In accordance with the present invention, the compound or compounds with formula (I) may be administered in association with one or more anticancerous active principles, in particular antitumour compounds such as alkylating agents, for example alkylsulphonates (busulfan), dacarbazine, procarbazine, nitrogen mustards (chlormethine, melphalan, chlorambucil), cyclophosphamide, ifosfamide; nitrosoureas such as carmustine, lomustine, semustine, streptozocine; antineoplasic alkaloids such as vincristine, vinblastine; taxanes such as paclitaxel or taxotere; antineoplasic antibiotics such as actinomycin; intercalating agents, antineoplasic antimetabolites, folate antagonists, methotrexate; purine synthesis inhibitors; purine analogues such as

mercaptopurine, 6-thioguanine; pyrimidine synthesis inhibitors, aromatase inhibitors, capecitabine, pyrimidine analogues such as fluorouracil, gemcitabine, cytarabine and cytosine arabinoside; brequinar; topoisomerase inhibitors such as camptothecine or etoposide; anticancer hormonal agonists and antagonists such as tamoxifen; kinase inhibitors, imatinib; growth factor inhibitors; antiinflammatories such as pentosane polysulphate, corticosteroids, prednisone, dexamethasone; antitopoisomerases such as etoposide, antracyclins including doxorubicin, bleomycin, mitomycin and methramycin; anticancerous metallic complexes, platinum complexes, cisplatinum, carboplatinum, oxaliplatinum; alpha interferon, triphenylthiophosphoramide, altretamine; antiangiogenic agents; thalidomide; immunotherapeutic adjuvants; vaccines.

According to the present invention, the compounds with the formula (I) may also be administered in association with one or more other active principles used in one of the diseases indicated above, for example an anti-emitic agent, a pain killer, an anti-inflammatory, an anti-cachexy agent.

A product of the invention may be used to manufacture a drug for use in treating a pathological condition, in particular cancer.

The present invention also concerns therapeutic compositions containing a compound of the invention, in association with a pharmaceutically acceptable excipient which depends on the selected mode of administration. The pharmaceutical composition may be in the form of a solid, liquid or liposomes.

Examples of solid compositions which may be cited are powders, capsules and tablets. The oral forms also include solid forms protected from the acidic medium of the stomach. The supports used for the solid forms are constituted in particular by mineral supports such as phosphates or carbonates, or organic supports such as lactose, celluloses, starch or

polymers. The liquid forms are constituted by solutions, suspensions or dispersions. They contain, as the dispersant, either water or an organic solvent (ethanol, NMP or others) or mixtures of surfactants and solvents or complexing agents and solvents.

The liquid forms are preferably injectable and for this reason will have a formulation suitable for such a use.

Acceptable injection modes include intravenous, intra-peritoneal, intramuscular and sub-cutaneous; the intravenous route is preferred.

The dose of the compounds of the invention which is administered will be adapted by the physician as a function of the method of administration to the patient and the condition of the patient.

The compounds of the present invention may be administered alone or as a mixture with other anticancer agents. Possible combinations which may be cited are as follows:

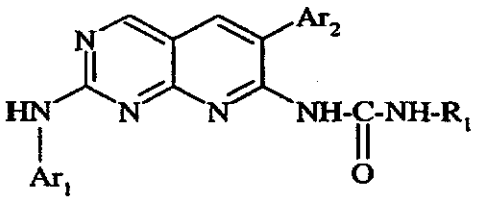
- alkylating agents, in particular cyclophosphamide, melphalan, ifosfamide, chlorambucil, busulfan, thiotepa, prednimustine, carmustine, lomustine, semustine, steptozotocin, decarbazine, temozolomide, procarbazine and hexamethylmelamine;
- platinum derivatives such as cisplatinum, carboplatinum or oxaliplatinum;
- antibiotic agents, in particular bleomycin, mitomycin and dactinomycin;
- antimicrotubule agents, in particular vinblastine, vincristine, vindesine, vinorelbine or taxoids (paclitaxel and docetaxel);
- antracyclins such as doxorubicin, daunorubicin, idarubicin, epirubicin, mitoxantrone or losoxantrone;
- group I and II topoisomerases such as etoposide, teniposide, amsacrine, irinotecan, topotecan and tomudex;
- fluoropyrimidines such as 5-fluorouracile, UFT or floxuridine;

- cytidine analogues such as 5-azacytidine, cytarabine, gemcitabine, 6-mercaptomurine, 6-thioguanine ;
- adenosine analogues such as pentostatin, cytarabine or fludarabine phosphate ;
- methotrexate and folinic acid ;
- various enzymes and compounds, such as L-asparaginase, hydroxyurea, trans-retinoic acid, suramine, dexrazoxane, amifostine, herceptin and oestrogenic or androgenic hormones;
- antivasculaires agents, such as combretastatin or colchicine derivatives and their prodrugs.

It is also possible to combine the compounds of the present invention with radiation treatment. Said treatments may be administered simultaneously, separately or sequentially. The treatment will be adapted by the physician to the patient to be treated.

What is claimed is:

1. A compound of formula (I):



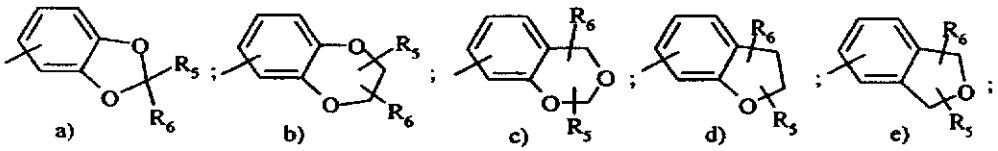
(I)

wherein:

R_1 is selected from the group consisting of $(C_1$-$C_6)$alkyl, $(C_3$-$C_7)$cycloalkyl, CH_2COR_4, phenyl and phenyl substituted with one or more substituents independently selected from hydroxy, halogen, or $(C_1$-$C_6)$alkyl;

R_4 represents a hydroxyl, $(C_1$-$C_4)$alkoxy, amino, $(C_1$-$C_4)$alkylamino or di$(C_1$-$C_4)$alkylamino group;

Ar_1 represents a radical selected from:



R_5 represents a cyano, hydroxy$(C_1$-$C_4)$alkyl, $(C_1$-$C_6)$alkoxy$(C_1$-$C_4)$alkyl group or a $(CH_2)_nNR_7R_8$, CO_2R_7, $CONHNR_7R_8$, $CONR_7R_8$, $CONR_8OR_9$, $(CH_2)_nNR_7COR_8$ or $(CH_2)_nNR_7COOR_8$ group;

R_6 represents a hydrogen atom, a (C1-C4)alkyl group or one of the values given for R5;

or R_5 and R_6 are bonded together to form a cycle comprising 0 to 2 heteroatoms selected from N and O, said cycle containing 4 to 7 chain links optionally being substituted with one or more substituents independently selected from halogen, $(C_1$-$C_4)$alkyl, halogenated $(C_1$-$C_4)$alkyl, hydroxy$(C_1$-$C_4)$alkyl, $(C_1$-$C_4)$alkoxy$(C_1$-$C_4)$alkyl, $(CH_2)_mNR_7R_8$, or a *tert*-butoxycarbonyl group;

R_7 and R_8 each independently represent a substituent selected from H, phenyl, $(C_1$-$C_4)$alkyl, $(C_1$-$C_4)$alkyl-OH, $(C_3$-$C_7)$cycloalkyl, $(C_3$-$C_7)$cycloalkyl-NH_2, $(C_1$-$C_4)$alkyl-$(C_3$-$C_7)$cycloalkyl, $C(=NH)NH_2$, $SO_2(C_1$-$C_6)$alkyl, and SO_2-phenyl;

R_8 may also represent a *tert*-butoxycarbonyl or benzyloxycarbonyl group;

or R_7 and R_8 together with the nitrogen atom to which they are bonded constitute an azetidinyl, pyrrolidinyl, piperidinyl, piperazinyl or morpholinyl radical, said radical being unsubstituted or substituted one or more times with a (C_1-C_6)alkyl, (C_1-C_4)alkyl-OH, $COO(C_1-C_6)$alkyl or F group;

R_9 represents a hydrogen atom or a (C_1-C_4)alkyl group;

Ar_2 represents a phenyl group which is unsubstituted or substituted 1 to 5 times with substituents independently selected from a halogen atom or a (C_1-C_4)alkyl, trifluoromethyl or (C_1-C_4)alkoxy group;

n represents 1, 2 or 3;

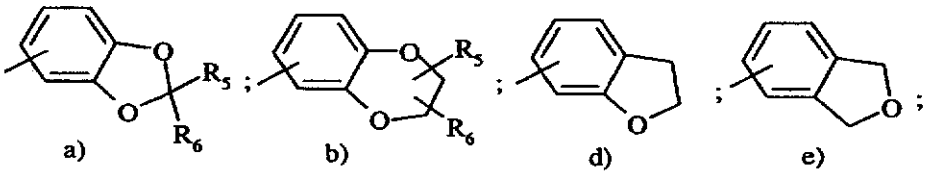
m represents 0, 1, 2 or 3;

or an acid addition salt thereof, or a hydrate or solvate of said compound or said salt.

2. A compound of formula (I) according to claim 1, wherein:

R_1 represents a *tert*-butyl, ethyl or phenyl group;

or Ar_1 represents a radical selected from:



or R_5 represents a $(CH_2)_nNR_7R_8$, $CONHNR_7R_8$, $CONR_7R_8$, hydroxy(C_1-C_4)alkyl or $(CH_2)_nNR_7COR_8$ group;

or R_6 represents a hydrogen atom or a methyl, $(CH_2)_nNR_7R_8$ or hydroxymethyl group;

or Ar_2 represents a phenyl group substituted with 1 or 2 substituents independently selected from a halogen atom or a (C_1-C_4)alkyl or (C_1-C_4)alkoxy group;

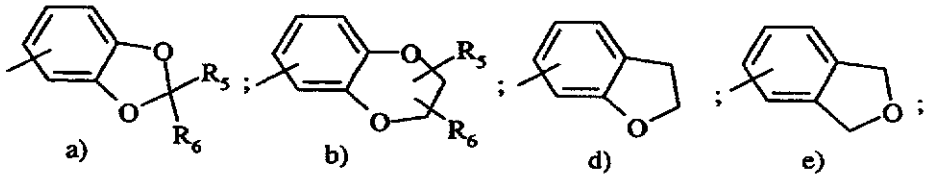
n, R_7 and R_8 are as defined in claim 1;

or an acid addition salt thereof, or a hydrate or solvate of said compound or said salt.

3. A compound of formula (I) according to claim 1, wherein:

R_1 represents a *tert*-butyl, ethyl or phenyl group;

Ar_1 represents a radical selected from:



R_5 represents a $(CH_2)_nNR_7R_8$, $CONHNR_7R_8$, $CONR_7R_8$, hydroxy(C_1-C_4)alkyl or $(CH_2)_nNR_7COR_8$ group;

R_6 represents a hydrogen atom or a methyl, $(CH_2)_nNR_7R_8$ or hydroxymethyl group;

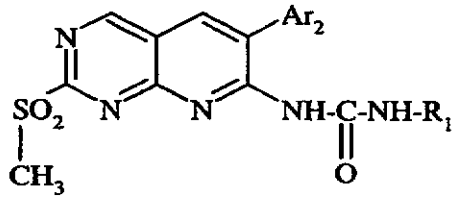
Ar_2 represents a phenyl group substituted with 1 or 2 substituents independently selected from a halogen atom or a (C_1-C_4)alkyl or (C_1-C_4)alkoxy group;

n, R_7 and R_8 being as defined above for a compound of formula (I);
or an acid addition salt thereof, or a hydrate or solvate of said compound or said salt.

4. A compound according to claim 1, wherein R5 is selected from $(CH_2)_nNR_7R_8$, $CONR_7R_8$ and $(CH_2)_nNR_7COR_8$.

5. A compound according to claim 1, wherein it is in the following form:
a non chiral; or
racemic; or
enriched in one stereoisomer; or
enriched in one enantiomer;
and optionally in the form of a solvate or hydrate, and optionally in the form of a salt.

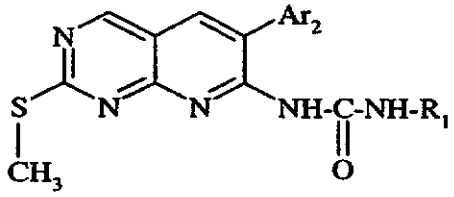
6. A compound of formula (II):



(II)

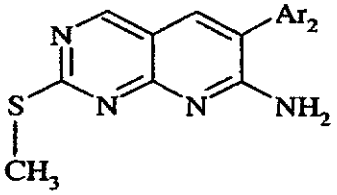
wherein R_1 and Ar_2 are as defined in claim 1.

7. A compound of the formula:



in which R_1 and Ar_2 are as defined in claim 1.

8. A compound of the formula:



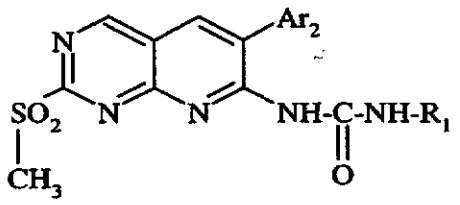
in which Ar2 is as defined in claim 1.

9. A compound according to claim 6 wherein Ar_2 is selected from phenyl, 2-methoxyphenyl, 2,6-dichlorophenyl, 3,5-dimethoxyphenyl, 3,4-dimethoxyphenyl, 2,6-dibromophenyl, 2-bromo-6-chlorophenyl, 2,4-dichlorophenyl and 3,5-dichlorophenyl.

10. A compound according to claim 6 wherein R_1 is selected from ethyl, tertiobutyl and phenyl.

11. A compound according to claim 7 wherein R_1 is selected from ethyl, tertiobutyl and phenyl.

12. A process for preparing a compound of formula (I) according to claim 1, wherein a compound of formula (II):

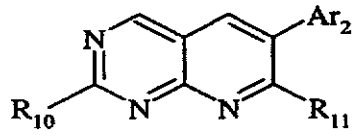


(II)

in which R_1 and Ar_2 are as defined for the compounds of formula(I) in claim 1, is reacted with an amine of the formula Ar'_1NH_2 (III) in which Ar'_1 represents Ar_1, as defined for the compounds of formula (I) in claim 1 or a precursor of Ar_1; if necessary, the group Ar'_1 of the compound obtained is transformed into a group Ar_1.

13. A process for preparing a compound of formula (I) according to claim 1, comprising reacting:

(i) a compound of formula:



(IIb)

in which Ar_2 is as defined in claim 1 and R_{10} is a leaving group selected from: (a) halogen or (b) alkyl-S(O)m- in which m = 0, 1, or 2; in which R_{11} is $NHC(=R_{12})-NH-R_1$, in which R_{12} = O or S; and

(ii) an amine of formula Ar'_1NH_2 (III) in which Ar'_1 represents Ar_1 as defined for a compound of formula (I) in claim 1 or a precursor of Ar_1; if necessary, the group Ar'_1 of the compound obtained is transformed into a group Ar_1;

wherein

a) when R_{10} is a halogen or alkyl-S(O)m- in which m = 2, the reaction is carried out in a solvent;

at a temperature in the range from ambient temperature to the reflux temperature of the solvent;

b) when R_{10} is alkyl-S(O)m- in which m = 0 or 1, the reaction is carried out with molten Ar'_1NH_2 (III) in the molten state at 200°C;
if necessary, the amine functions present in the Ar'_1 group of compound (III) have been transformed into salts or have been protected.

14. The process according to claim 13 wherein the solvent is selected from:

(i) terahydrofuran, dimethylsulphoxide or ethanol, optionally in the presence of a trace amount of an acid; or

(ii) dimethylsulphoxide in the presence of a strong base.

15. A pharmaceutical composition, comprising a compound of formula (I) according to claim 1, or a pharmaceutically acceptable salt thereof, or a hydrate or a solvate of said compound, and at least one pharmaceutically acceptable excipient.

16. A pharmaceutical composition according to claim 15, further comprising one or more other anticancer active principle(s).

17. A method of treating or preventing a disease caused or exacerbated by cell proliferation comprising administering to a patient in need thereof a compound of formula (I) according to claim 1.

18. The method according to claim 17, wherein the disease is selected from leukaemias, primary and metastatic solid tumours, carcinomas and cancers.

19. The method according to claim 17 wherein the disease is selected from breast cancer, prostate cancer, colon cancer, lung cancer, glioblastoma, and leukaemia.