

DESCRIPTION

TITLE OF THE INVENTION: RESIST UNDERLAYER FILM FORMING
COMPOSITION CONTAINING SILICON CONTAINING CYCLIC ORGANIC
5 GROUP HAVING HETERO ATOM

TECHNICAL FIELD

[0001] The present invention relates to a composition for forming an underlayer
10 film between a substrate and a resist (for example, a photoresist, an electron beam resist,
and an EUV resist) that are used in the production of semiconductor devices. More in
detail, the present invention relates to a resist underlayer film forming composition for
lithography for forming an underlayer film used for an underlayer of a photoresist in a
lithography process of the production of semiconductor devices. In addition, the present
15 invention relates to a forming method of a resist pattern using the underlayer film
forming composition.

BACKGROUND ART

20 [0002] Conventionally, in the production of semiconductor devices, fine
processing by lithography using a photoresist has been performed. The fine processing
is a processing method for forming fine convexo-concave shapes corresponding to the
following pattern on the surface of a substrate by: forming a thin film of a photoresist on
a semiconductor substrate such as a silicon wafer; irradiating the resultant thin film with
25 active rays such as ultraviolet rays through a mask pattern in which a pattern of a
semiconductor device is depicted; developing the thin film; and subjecting the substrate
to etching processing using the resultant photoresist pattern as a protecting film.
However, in recent years, high integration of semiconductor devices has progressed and
the adopted active rays tend to have a shorter wavelength, such as an ArF excimer laser

(193 nm) and EUV light (13.5 nm), replacing a KrF excimer laser (248 nm). Following such a tendency, the influence of reflection of active rays on a semiconductor substrate has become a large issue.

[0003] As an underlayer film between the semiconductor substrate and the photoresist, the use of a film known as a hardmask containing a metal element such as silicon (see, for example, Patent Document 1) has been performed. In this case, the resist and the hardmask have components largely different from each other, so that the removal rates of the resist and the hardmask by dry etching largely depend on the type of a gas used for dry etching. By appropriately selecting the type of a gas, the hardmask can be removed by dry etching without a large decrease in the film thickness of the photoresist. Thus, in the production of semiconductor devices in recent years, for achieving various effects such as the reflection preventing effect, a resist underlayer film has become disposed between the semiconductor substrate and the photoresist. Then, also conventionally, studies of a composition for a resist underlayer film have been performed; however, due to the diversity of characteristics required for the composition and so on, development of a novel material for the resist underlayer film is desired.

A resist underlayer film forming composition containing silicon having a urea group (see Patent Document 2), a resist underlayer film forming composition containing silicon having a cyclic amino group (see Patent Document 3), and a resist underlayer film forming composition containing silicon having an isocyanurate group (see Patent Documents 4 and 5) are also described.

Prior Art Documents

Patent Documents

[0004] Patent Document 1: International Publication No. WO 2011/033965 pamphlet

Patent Document 2: International Publication No. WO 2009/088039 Pamphlet

Patent Document 3: International Publication No. WO 2009/104552 Pamphlet

Patent Document 4: International Publication No. WO 2009/034998 Pamphlet

Patent Document 5: International Publication No. WO 2011/102470 Pamphlet

SUMMARY OF THE INVENTION

Problem to be Solved by the Invention

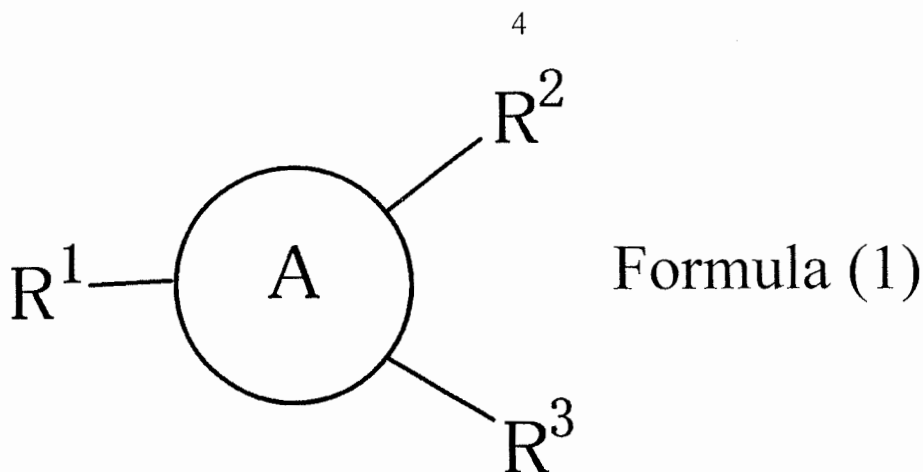
5 [0005] It is an object of the present invention to provide a resist underlayer film forming composition for lithography capable of performing fine processing of the substrate by utilizing a rectangular resist pattern and capable of being used in the production of semiconductor devices. More in detail, it is an object of the present invention to provide: a resist underlayer film for lithography capable of being used as a
10 hardmask and an anti-reflective coating, causing no intermixing with a resist, and having a dry etching rate higher than that of the resist; and a resist underlayer film forming composition for forming the underlayer film.

It is another object of the present invention to provide a method for producing a semiconductor device produced by using a film formed from the resist underlayer film
15 forming composition on a semiconductor substrate as a resist underlayer film.

Means for Solving the Problem

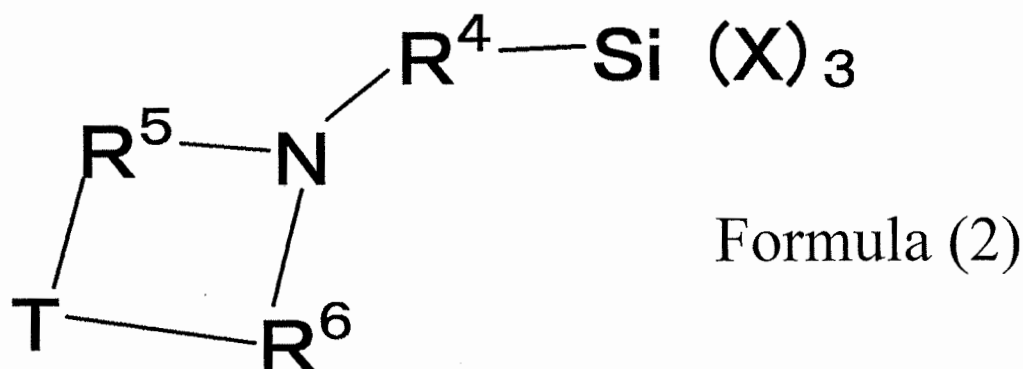
[0006] The present invention relates to, according to a first aspect, a resist underlayer film forming composition for lithography, comprising: as a silane: at least one
20 among a hydrolyzable organosilane, a hydrolysis product thereof, and a hydrolysis-condensation product thereof, in which the silane includes a silane having a cyclic organic group containing as atoms making up the ring, a carbon atom, a nitrogen atom, and a hetero atom other than a carbon atom and a nitrogen atom;

according to a second aspect, the resist underlayer film forming composition
25 according to the first aspect, in which the hydrolyzable organosilane is a hydrolyzable organosilane of Formula (1):



(in Formula (1), at least one group among R^1 , R^2 , and R^3 is a group in which a $-\text{Si}(\text{X})_3$ group is bonded to a C_{1-10} alkylene group, and the other group(s) among R^1 , R^2 , and R^3 is(are) a hydrogen atom, a C_{1-10} alkyl group, or a C_{6-40} aryl group; a ring A is a cyclic organic group of a 5- to 10-membered ring containing as atoms making up the ring, a carbon atom, at least one nitrogen atom, and further containing a sulfur atom or an oxygen atom; and X is an alkoxy group, an acyloxy group, or a halogen atom);

according to a third aspect, the resist underlayer film forming composition according to the first aspect, in which the hydrolyzable organosilane is a hydrolyzable organosilane of Formula (2):



(in Formula (2), each of R^4 , R^5 , and R^6 is a C_{1-6} alkylene group; T is a sulfur atom or an oxygen atom; a cyclic organic group made up with a nitrogen atom, R^5 , R^6 , and T is a 5- to 10-membered ring; and X is an alkoxy group, an acyloxy group, or a halogen atom);

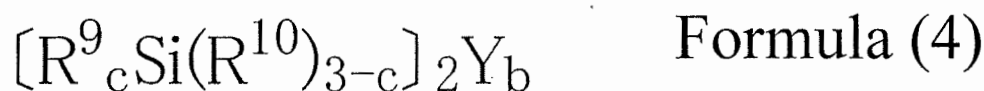
according to a fourth aspect, the resist underlayer film forming composition

according to the first aspect, in which the silane includes any one of or both of: a combination of at least one selected from the group consisting of at least one of organic silicon compound (B) selected from the group consisting of an organic silicon compound of Formula (3):



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(in Formula (3), R^7 is a group that is an alkyl group, an aryl group, an aralkyl group, a halogenated alkyl group, a halogenated aryl group, a halogenated aralkyl group, an alkenyl group, or an organic group having an epoxy group, an acryloyl group, a methacryloyl group, a mercapto group, an alkoxyaryl group, an acyloxyaryl group, an isocyanurate group, a hydroxy group, a cyclic amino group, or a cyano group, or a combination of these groups, and that is bonded to a silicon atom through a Si-C bond; R^8 is an alkoxy group, an acyloxy group, or a halogen atom; and a is an integer of 0 to 3) and an organic silicon compound of Formula (4):



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(in Formula (4), R^9 is an alkyl group; R^{10} is an alkoxy group, an acyloxy group, or a halogen atom; Y is an alkylene group or an arylene group; b is an integer of 0 or 1; and c is an integer of 0 or 1), a hydrolysis product of (B), and a hydrolysis-condensation product of (B) with at least one selected from the group consisting of a hydrolyzable organosilane (A) of Formula (1) or Formula (2), a hydrolysis product of (A), and a hydrolysis-condensation product of (A); and a hydrolysis-condensation product of (A) and (B);

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according to a fifth aspect, the resist underlayer film forming composition according to any one of the first aspect to the fourth aspect, in which any one of or both of: a hydrolysis-condensation product of a hydrolyzable organosilane (A) of Formula (1) or Formula (2); and a hydrolysis-condensation product of a hydrolyzable organosilane (A) of Formula (1) or Formula (2) with an organic silicon compound of Formula (3) is contained as a polymer;

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according to a sixth aspect, the resist underlayer film forming composition according to any one of the first aspect to the fifth aspect, further comprising an acid;

according to a seventh aspect, the resist underlayer film forming composition according to any one of the first aspect to the sixth aspect, further comprising water;

5 according to an eighth aspect, a resist underlayer film obtained by applying the resist underlayer film forming composition as described in any one of the first aspect to the seventh aspect onto a semiconductor substrate, and baking the composition;

according to a ninth aspect, a method for producing a semiconductor device, the method comprising: applying the resist underlayer film forming composition as described
10 in any one of the first aspect to the seventh aspect onto a semiconductor substrate and baking the composition to form a resist underlayer film; applying a composition for a resist onto the underlayer film to form a resist film; exposing the resist film to light; developing the resist film after the exposure to obtain a patterned resist film; etching the resist underlayer film according to a pattern of the patterned resist film to obtain a
15 patterned resist underlayer film; and processing the semiconductor substrate according to a pattern of the patterned resist film and the patterned resist underlayer film; and

according to a tenth aspect, a method for producing a semiconductor device, the method comprising: forming an organic underlayer film on a semiconductor substrate; applying the resist underlayer film forming composition as described in any one of the
20 first aspect to the seventh aspect onto the organic underlayer film and baking the composition to form a resist underlayer film; applying a composition for a resist onto the resist underlayer film to form a resist film; exposing the resist film to light; developing the resist film after the exposure to obtain a patterned resist film; etching the resist underlayer film according to a pattern of the patterned resist film to obtain a patterned
25 resist underlayer film; etching the organic underlayer film according to a pattern of the patterned resist film and the patterned resist underlayer film to obtain a patterned organic underlayer film; and processing the semiconductor substrate according to a pattern of the patterned resist film, the patterned resist underlayer film, and the patterned organic underlayer film.

Effects of the Invention

[0007] The resist underlayer film forming composition of the present invention contains a silane having a cyclic organic group containing as atoms making up the ring, a carbon atom, a nitrogen atom, and a hetero atom other than these atoms, so that a resist underlayer film obtained from the resist underlayer film forming composition can allow a cross-sectional shape of a patterned resist film formed as an upper layer of the resist underlayer film to be rectangle, which makes it possible to form a fine pattern in a semiconductor substrate.

The resist underlayer film of the present invention contains a polyorganosiloxane structure, so that the resist underlayer film can function as a hardmask or an anti-reflective coating and is unlikely to cause intermixing with a resist film.

According to the method for producing a semiconductor device of the present invention, a semiconductor substrate in which a fine pattern is formed can be preferably obtained.

MODES FOR CARRYING OUT THE INVENTION

[0008] In the present invention, the resist underlayer film is formed by an coating method either on a substrate or on an organic underlayer film formed on a substrate and, on the resist underlayer film, a resist film (for example, a photoresist, an electron beam resist, and an EUV resist) is formed. Then, a resist pattern is formed by exposure and development, and either by dry etching the resist underlayer film according to the resist pattern to transfer the pattern, the substrate is processed according to the pattern, or by etching the organic underlayer film to transfer the pattern, the substrate is processed using the etched organic underlayer film.

In forming a fine pattern, for preventing a pattern collapse, the resist film thickness tends to become smaller. Due to the thinning of the resist, the dry etching for transferring the pattern to a film existing as an underlayer of the resist cannot be used to

transfer the pattern unless the etching rate of the underlayer film is higher than that of the upper layer film. In the present invention, the substrate is coated in such an order that the substrate is coated with the resist underlayer film (containing an inorganic silicon-based compound) of the present specification either with or without an organic underlayer film interposed therebetween, and then the resist underlayer film is coated with a resist film (an organic resist film). An organic component film and an inorganic component film have dry etching rates largely different from each other depending on the selection of the etching gas. The dry etching rate of the organic component film is enhanced by an oxygen-based gas and the dry etching rate of the inorganic component film is enhanced by a halogen-containing gas.

For example, a resist pattern is formed and the pattern is transferred to the resist underlayer film of the present specification existing as an underlayer of the resist pattern by dry etching the resist underlayer film with a halogen-containing gas, and the substrate is processed with a halogen-containing gas according to the pattern which has been transferred to the resist underlayer film. Alternatively, by dry etching the organic underlayer film existing as an underlayer of the resist underlayer film to which the pattern has been transferred with an oxygen-based gas using the etched resist underlayer film, the pattern is transferred to the organic underlayer film, and the substrate is processed with a halogen-containing gas using the organic underlayer film to which the pattern has been transferred.

In the present invention, the resist underlayer film functions as a hardmask. A hydrolyzable group in the structures of Formulae (1), (2), (3), and (4) such as an alkoxy group, an acyloxy group, or a halogen atom is hydrolyzed or partially hydrolyzed to form a silanol group and then, by a condensation reaction of the silanol group, a polymer having a polyorganosiloxane structure is formed. This polyorganosiloxane structure has a satisfactory function as a hardmask.

Then, the polyorganosiloxane structure (intermediate film) is effective as a hardmask for etching an organic underlayer film existing as an underlayer of the intermediate film or for processing (etching) the substrate. That is, the intermediate film

has satisfactory dry etching resistance during the substrate processing or against an oxygen-based dry etching gas for etching the organic underlayer film.

The resist underlayer film of the present invention possesses enhancement of the dry etching rate of the resist underlayer film relative to the upper layer resist and dry etching resistance of the resist underlayer film during the substrate processing or the like.

The resist underlayer film formed from the resist underlayer film forming composition of the present invention contains a silane compound containing an organic group containing a hetero atom-containing cyclic organic group, a hydrolysis product thereof, or a hydrolysis-condensation product thereof, so that the shape of the resist pattern after exposure and development becomes a rectangular shape. Thus, fine processing of the substrate by using a rectangular pattern becomes possible.

[0009] The present invention relates to a resist underlayer film forming composition for lithography containing as a silane, at least one among a hydrolyzable organosilane, a hydrolysis product thereof, and a hydrolysis-condensation product thereof, in which the silane includes a silane having a cyclic organic group containing as atoms making up the ring, a carbon atom, a nitrogen atom, and a hetero atom other than these atoms.

The resist underlayer film forming composition of the present invention contains at least one of the hydrolyzable organosilane, a hydrolysis product thereof, and a hydrolysis-condensation product thereof, and a solvent. Then, the resist underlayer film forming composition of the present invention may contain as optional components, acids, water, alcohols, curing catalysts, acid generators, other organic polymers, light absorptive compounds, surfactants, and the like.

The solid content in the resist underlayer film forming composition of the present invention is, for example, 0.1 to 50% by mass, 0.1 to 30% by mass, or 0.1 to 25% by mass. Here, the solid content is a component remaining after removing a solvent component from all components of the resist underlayer film forming composition.

The content of the hydrolyzable organosilane, a hydrolysis product thereof, or a hydrolysis-condensation product thereof in the solid content is 20% by mass or more, for

example, 50 to 100% by mass, 60 to 100% by mass, or 70 to 100% by mass.

[0010] In the resist underlayer film forming composition of the present invention, the hydrolyzable organosilane is preferably a hydrolyzable organosilane of Formula (1).

In Formula (1), at least one group among R^1 , R^2 , and R^3 is a group in which a
 5 -Si(X)₃ group is bonded to a C₁₋₁₀ alkylene group, and the other group(s) among R^1 , R^2 , and R^3 is(are) a hydrogen atom, a C₁₋₁₀ alkyl group, or a C₆₋₄₀ aryl group; a ring A is a cyclic organic group of a 5- to 10-membered ring containing as atoms making up the ring, a carbon atom, at least one nitrogen atom, and further a sulfur atom or an oxygen atom; and X is an alkoxy group, an acyloxy group, or a halogen atom.

10 In the resist underlayer film forming composition of the present invention, the hydrolyzable organosilane can be a hydrolyzable organosilane of Formula (2).

In Formula (2), each of R^4 , R^5 , and R^6 is a C₁₋₆ alkylene group; T is a sulfur atom or an oxygen atom; a cyclic organic group made up with a nitrogen atom, R^5 , R^6 , and T is a 5- to 10-membered ring; and X is an alkoxy group, an acyloxy group, or a halogen atom.

15 [0011] Examples of the C₁₋₁₀ alkyl group include a methyl group, an ethyl group, an n-propyl group, an isopropyl group, a cyclopropyl group, an n-butyl group, an isobutyl group, a sec-butyl group, a tert-butyl group, a cyclobutyl group, a 1-methyl-cyclopropyl group, a 2-methyl-cyclopropyl group, an n-pentyl group, a 1-methyl-n-butyl group, a 2-methyl-n-butyl group, a 3-methyl-n-butyl group, a 1,1-dimethyl-n-propyl group, a
 20 1,2-dimethyl-n-propyl group, a 2,2-dimethyl-n-propyl group, a 1-ethyl-n-propyl group, a cyclopentyl group, a 1-methyl-cyclobutyl group, a 2-methyl-cyclobutyl group, a 3-methyl-cyclobutyl group, a 1,2-dimethyl-cyclopropyl group, a 2,3-dimethyl-cyclopropyl group, a 1-ethyl-cyclopropyl group, a 2-ethyl-cyclopropyl group, an n-hexyl group, a 1-methyl-n-pentyl group, a 2-methyl-n-pentyl group, a
 25 3-methyl-n-pentyl group, a 4-methyl-n-pentyl group, a 1,1-dimethyl-n-butyl group, a 1,2-dimethyl-n-butyl group, a 1,3-dimethyl-n-butyl group, a 2,2-dimethyl-n-butyl group, a 2,3-dimethyl-n-butyl group, a 3,3-dimethyl-n-butyl group, a 1-ethyl-n-butyl group, a 2-ethyl-n-butyl group, a 1,1,2-trimethyl-n-propyl group, a 1,2,2-trimethyl-n-propyl group, a 1-ethyl-1-methyl-n-propyl group, a 1-ethyl-2-methyl-n-propyl group, a cyclohexyl

group, a 1-methyl-cyclopentyl group, a 2-methyl-cyclopentyl group, a 3-methyl-cyclopentyl group, a 1-ethyl-cyclobutyl group, a 2-ethyl-cyclobutyl group, a 3-ethyl-cyclobutyl group, a 1,2-dimethyl-cyclobutyl group, a 1,3-dimethyl-cyclobutyl group, a 2,2-dimethyl-cyclobutyl group, a 2,3-dimethyl-cyclobutyl group, a 2,4-dimethyl-cyclobutyl group, a 3,3-dimethyl-cyclobutyl group, a 1-n-propyl-cyclopropyl group, a 2-n-propyl-cyclopropyl group, a 1-isopropyl-cyclopropyl group, a 2-isopropyl-cyclopropyl group, a 1,2,2-trimethyl-cyclopropyl group, a 1,2,3-trimethyl-cyclopropyl group, a 2,2,3-trimethyl-cyclopropyl group, a 1-ethyl-2-methyl-cyclopropyl group, a 2-ethyl-1-methyl-cyclopropyl group, a 2-ethyl-2-methyl-cyclopropyl group, and a 2-ethyl-3-methyl-cyclopropyl group. Among them, a methyl group, an ethyl group, and an n-propyl group are preferably used.

[0012] Examples of the C_{1-10} alkylene group include alkylene groups corresponding to the above alkyl groups. And then, a methylene group, an ethylene group, and a propylene group are preferably used.

Examples of the C_{6-40} aryl group include a phenyl group, an o-methylphenyl group, an m-methylphenyl group, a p-methylphenyl group, an o-chlorophenyl group, an m-chlorophenyl group, a p-chlorophenyl group, an o-fluorophenyl group, a p-fluorophenyl group, an o-methoxyphenyl group, a p-methoxyphenyl group, a p-nitrophenyl group, a p-cyanophenyl group, an α -naphthyl group, a β -naphthyl group, an o-biphenyl group, an m-biphenyl group, a p-biphenyl group, a 1-anthryl group, a 2-anthryl group, a 9-anthryl group, a 1-phenanthryl group, a 2-phenanthryl group, a 3-phenanthryl group, a 4-phenanthryl group, and a 9-phenanthryl group.

Among them, a phenyl group is preferably used.

Examples of the C_{6-40} arylene group include arylene groups corresponding to the above aryl groups. And then, a phenylene group is preferably used.

Examples of the hydrolyzable group in the hydrolyzable silane include an alkoxy group, an acyloxy group, and a halogen atom.

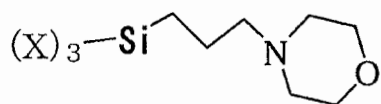
[0013] Examples of the alkoxy group include alkoxy groups having a linear,

branched, or cyclic alkyl moiety having a carbon atom number of 1 to 20 such as a methoxy group, an ethoxy group, an n-propoxy group, an isopropoxy group, an n-butoxy group, an isobutoxy group, a sec-butoxy group, a tert-butoxy group, an n-pentyloxy group, a 1-methyl-n-butoxy group, a 2-methyl-n-butoxy group, a 3-methyl-n-butoxy group, a 1,1-dimethyl-n-propoxy group, a 1,2-dimethyl-n-propoxy group, a 2,2-dimethyl-n-propoxy group, a 1-ethyl-n-propoxy group, an n-hexyloxy group, a 1-methyl-n-pentyloxy group, a 2-methyl-n-pentyloxy group, and a 3-methyl-n-pentyloxy group.

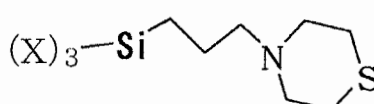
Examples of the C₂₋₂₀ acyloxy group include a methylcarbonyloxy group, an ethylcarbonyloxy group, an n-propylcarbonyloxy group, an isopropylcarbonyloxy group, an n-butylcarbonyloxy group, an isobutylcarbonyloxy group, a sec-butylcarbonyloxy group, a tert-butylcarbonyloxy group, an n-pentylcarbonyloxy group, a 1-methyl-n-butylcarbonyloxy group, a 2-methyl-n-butylcarbonyloxy group, a 3-methyl-n-butylcarbonyloxy group, and a 1,1-dimethyl-n-propylcarbonyloxy group.

Examples of the halogen atom include a fluorine atom, a chlorine atom, a bromine atom, and an iodine atom.

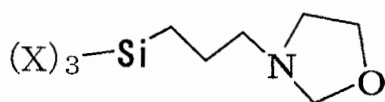
[0014] The hydrolyzable organosilanes of Formula (1) and (2) can be exemplified below. In the formulae below, X is the above alkoxy group, the above acyloxy group, or the above halogen atom.



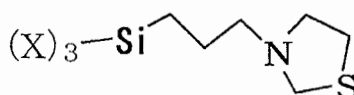
Formula (1-1)



Formula (1-2)



Formula (1-3)



Formula (1-4)

[0015] The resist underlayer film forming composition of the present invention contains any one of or both of: a combination of at least one selected from the group consisting of at least one of organic silicon compound (B) selected from the group

consisting of an organic silicon compound of Formula (3) and an organic silicon compound of Formula (4), a hydrolysis product of (B), and a hydrolysis-condensation product of (B) with at least one selected from the group consisting of a hydrolyzable organosilane (A) of Formula (1) or Formula (2), a hydrolysis product of (A), and a hydrolysis-condensation product of (A); and a hydrolysis-condensation product of (A) and (B).

Then, a preferred aspect of the present invention is a resist underlayer film forming composition in which a hydrolysis-condensation product of a hydrolyzable organosilane (A) of Formula (1) or Formula (2), or a hydrolysis-condensation product of a hydrolyzable organosilane (A) of Formula (1) or Formula (2) with an organic silicon compound of Formula (3) is contained as a polymer.

Further, in the present invention, a hydrolyzable organosilane (A) of Formula (1) or Formula (2) can be used together with at least one of organic silicon compound (B) among an organic silicon compound of Formula (3) and an organic silicon compound of Formula (4), and it is preferred that a hydrolyzable organosilane (A) of Formula (1) or Formula (2), a hydrolysis product thereof, or a hydrolysis-condensation product thereof is used together with an organic silicon compound of Formula (3), a hydrolysis product thereof, or a hydrolysis-condensation product thereof.

[0016] The ratio between a hydrolyzable organosilane (A) of Formula (1) or Formula (2) and at least one of organic silicon compound (B) among an organic silicon compound of Formula (3) and an organic silicon compound of Formula (4) in a mole ratio can be used in a range of 1:0 to 1:200. In order to obtain a favorable resist shape, the ratio between the hydrolyzable organosilane (A) and the organic silicon compound (B) in a mole ratio can be used in a range of 1:199 to 1:2.

These are preferably used as a hydrolysis-condensation product (polymer of polyorganosiloxane), and a hydrolysis-condensation product (polymer of polyorganosiloxane) of the hydrolyzable organosilane (A) of Formula (1) or Formula (2) and the organic silicon compound of Formula (3) is preferably used.

[0017] In Formula (3), R^7 is a group that is an alkyl group, an aryl group, an

aralkyl group, a halogenated alkyl group, a halogenated aryl group, a halogenated aralkyl group, an alkenyl group, or an organic group having an epoxy group, an acryloyl group, a methacryloyl group, a mercapto group, an alkoxyaryl group, an acyloxyaryl group, an isocyanurate group, a hydroxy group, a cyclic amino group, or a cyano group, or a combination thereof, and is bonded to a silicon atom through a Si-C bond; R⁸ is an alkoxy group, an acyloxy group, or a halogen atom; and a is an integer of 0 to 3.

Examples of an alkyl group, an aryl group, an aralkyl group, a halogenated alkyl group, a halogenated aryl group, a halogenated aralkyl group, an alkenyl group, or an organic group having an epoxy group, an acryloyl group, a methacryloyl group, a mercapto group, an alkoxyaryl group, an acyloxyaryl group, an isocyanurate group, a hydroxy group, a cyclic amino group, or a cyano group, and further an alkoxy group, an acyloxy group, and a halogen atom contained in a hydrolyzable group of R⁷ and R⁸, include examples thereof described above with respect to Formula (1) and Formula (2).

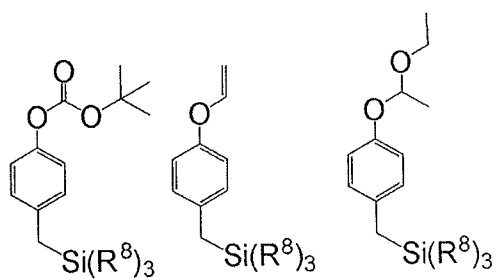
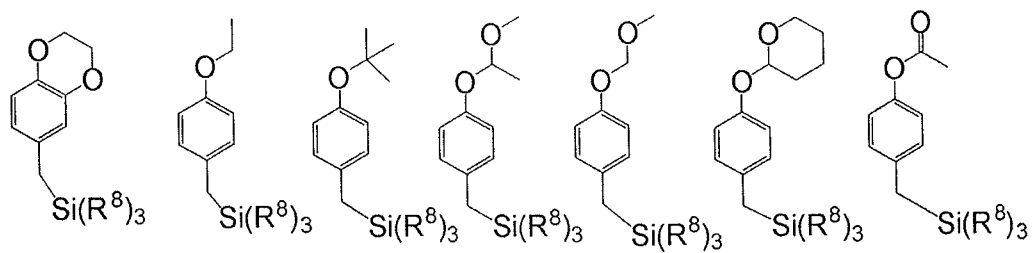
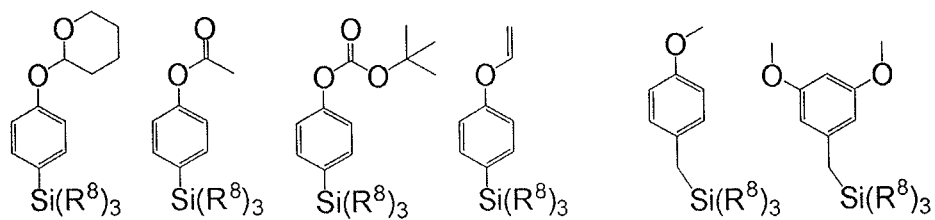
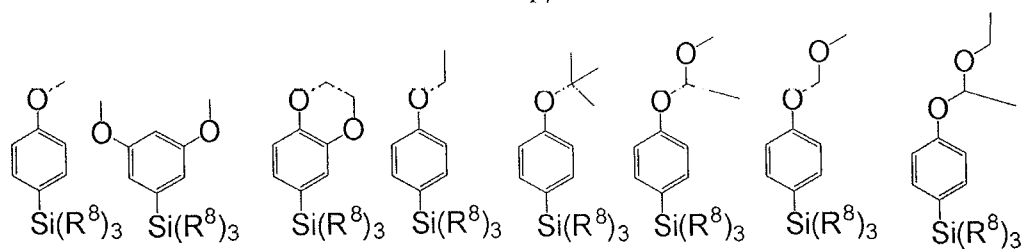
[0018] Examples of the organic silicon compound of Formula (3) include tetramethoxysilane, tetrachlorosilane, tetraacetoxysilane, tetraethoxysilane, tetra-n-propoxysilane, tetraisopropoxysilane, tetra-n-butoxysilane, tetraacetoxysilane, methyltrimethoxysilane, methyltriethoxysilane, methyltrichlorosilane, methyltriacetoxysilane, methyltripropoxysilane, methyltriacetoxysilane, methyltributoxysilane, methyltripropoxysilane, methyltriamploxysilane, methyltriphenoxysilane, methyltribenzyloxysilane, methyltriphenethyloxysilane, glycidoxymethyltrimethoxysilane, glycidoxymethyltriethoxysilane, α-glycidoxyethyltrimethoxysilane, α-glycidoxyethyltriethoxysilane, β-glycidoxyethyltrimethoxysilane, β-glycidoxyethyltriethoxysilane, α-glycidoxypropyltrimethoxysilane, α-glycidoxypropyltriethoxysilane, β-glycidoxypropyltrimethoxysilane, β-glycidoxypropyltriethoxysilane, γ-glycidoxypropyltrimethoxysilane, γ-glycidoxypropyltriethoxysilane, γ-glycidoxypropyltripropoxysilane, γ-glycidoxypropyltributoxysilane, γ-glycidoxypropyltriphenoxysilane, α-glycidoxybutyltrimethoxysilane, α-glycidoxybutyltriethoxysilane, β-glycidoxybutyltriethoxysilane,

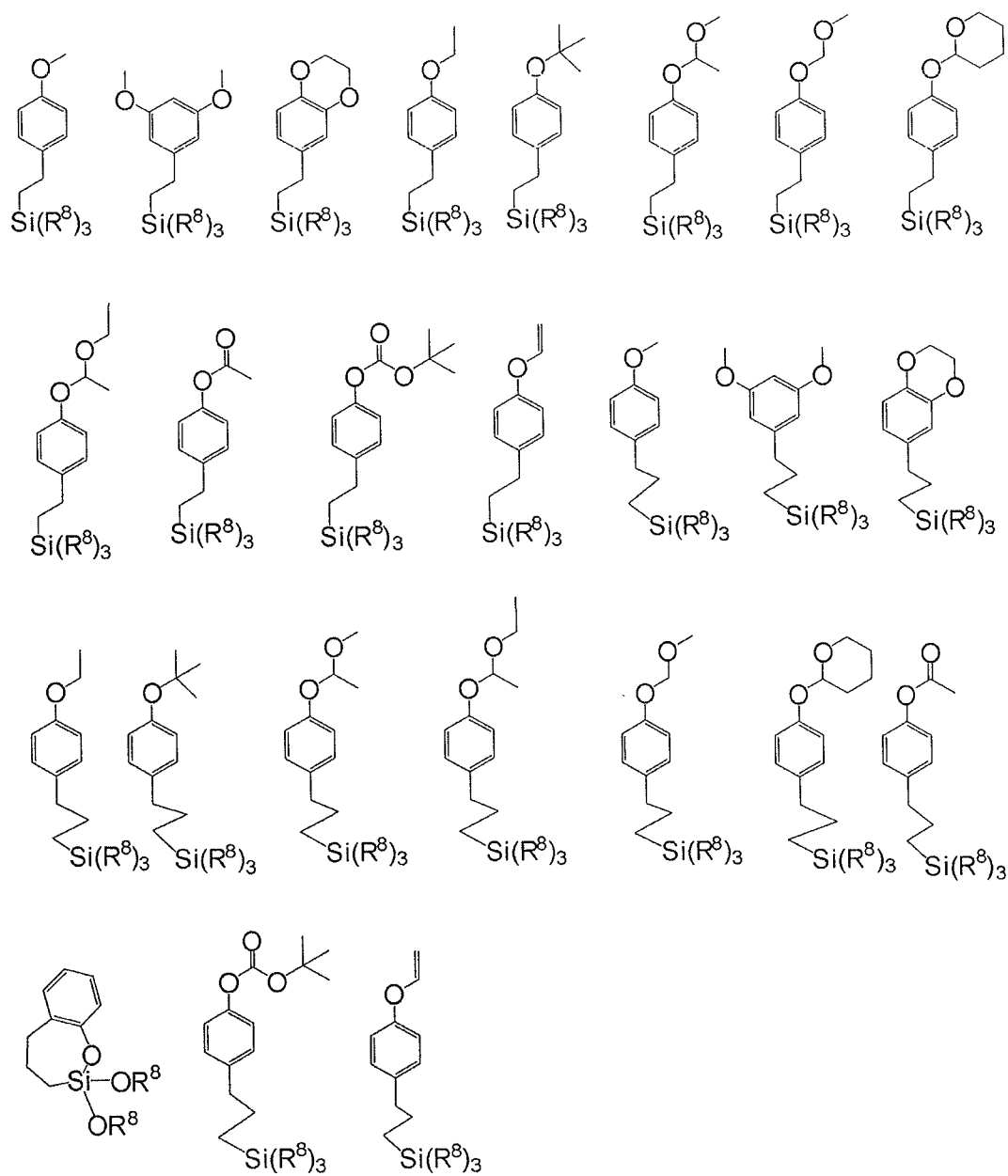
- γ -glycidoxybutyltrimethoxysilane, γ -glycidoxybutyltriethoxysilane,
 δ -glycidoxybutyltrimethoxysilane, δ -glycidoxybutyltriethoxysilane,
 (3,4-epoxycyclohexyl)methyltrimethoxysilane,
 (3,4-epoxycyclohexyl)methyltriethoxysilane,
 5 β -(3,4-epoxycyclohexyl)ethyltrimethoxysilane, β -(3,4-epoxycyclohexyl)
 ethyltriethoxysilane, β -(3,4-epoxycyclohexyl)ethyltripropoxysilane,
 β -(3,4-epoxycyclohexyl)ethyltributoxysilane,
 β -(3,4-epoxycyclohexyl)ethyltriphenoxysilane,
 γ -(3,4-epoxycyclohexyl)propyltrimethoxysilane,
 10 γ -(3,4-epoxycyclohexyl)propyltriethoxysilane,
 δ -(3,4-epoxycyclohexyl)butyltrimethoxysilane,
 δ -(3,4-epoxycyclohexyl)butyltriethoxysilane, glycidoxymethylmethyldimethoxysilane,
 glycidoxymethylmethyldiethoxysilane, α -glycidoxyethylmethyldimethoxysilane,
 α -glycidoxyethylmethyldiethoxysilane, β -glycidoxyethylmethyldimethoxysilane,
 15 β -glycidoxyethylethyldimethoxysilane, α -glycidoxypropylmethyldimethoxysilane,
 α -glycidoxypropylmethyldiethoxysilane, β -glycidoxypropylmethyldimethoxysilane,
 β -glycidoxypropylethyldimethoxysilane, γ -glycidoxypropylmethyldimethoxysilane,
 γ -glycidoxypropylmethyldiethoxysilane, γ -glycidoxypropylmethyldipropoxysilane,
 γ -glycidoxypropylmethyldibutoxysilane, γ -glycidoxypropylmethyldiphenoxysilane,
 20 γ -glycidoxypropylethyldimethoxysilane, γ -glycidoxypropylethyldiethoxysilane,
 γ -glycidoxypropylvinylmethoxysilane, γ -glycidoxypropylvinylmethyldiethoxysilane,
 ethyltrimethoxysilane, ethyltriethoxysilane, vinyltrimethoxysilane, vinyltrichlorosilane,
 vinyltriacetoxysilane, vinyltriethoxysilane, vinyltriacetoxysilane, phenyltrimethoxysilane,
 phenyltrichlorosilane, phenyltriacetoxysilane, phenyltriethoxysilane,
 25 phenyltriacetoxysilane, γ -chloropropyltrimethoxysilane, γ -chloropropyltriethoxysilane,
 γ -chloropropyltriacetoxysilane, 3,3,3-trifluoropropyltrimethoxysilane,
 γ -methacryloxypropyltrimethoxysilane, γ -mercaptopropyltrimethoxysilane,
 γ -mercaptopropyltriethoxysilane, β -cyanoethyltriethoxysilane,
 chloromethyltrimethoxysilane, chloromethyltriethoxysilane, dimethyldimethoxysilane,

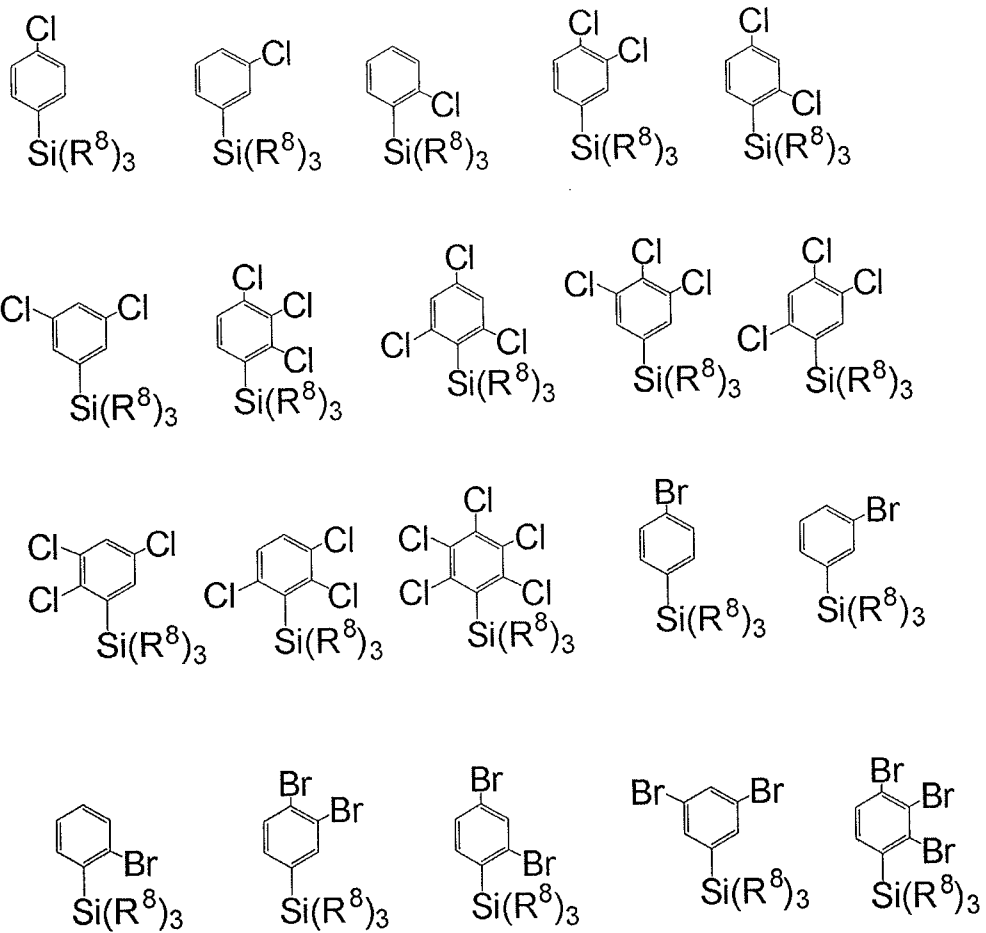
phenylmethyldimethoxysilane, dimethyldiethoxysilane, phenylmethyldiethoxysilane,
γ-chloropropylmethyldimethoxysilane, γ-chloropropylmethyldiethoxysilane,
dimethyldiacetoxysilane, γ-methacryloxypropylmethyldimethoxysilane,
γ-methacryloxypropylmethyldiethoxysilane, γ-mercaptopropylmethyldimethoxysilane,
5 γ-mercaptopmethyldiethoxysilane, methylvinylmethyldimethoxysilane,
methylvinylmethyldiethoxysilane, phenylsulfonylaminopropyltriethoxysilane,
methylsulfonylaminopropyltriethoxysilane, phenylsulfonylaminopropyltrimethoxysilane,
and methylsulfonylaminopropyltrimethoxysilane.

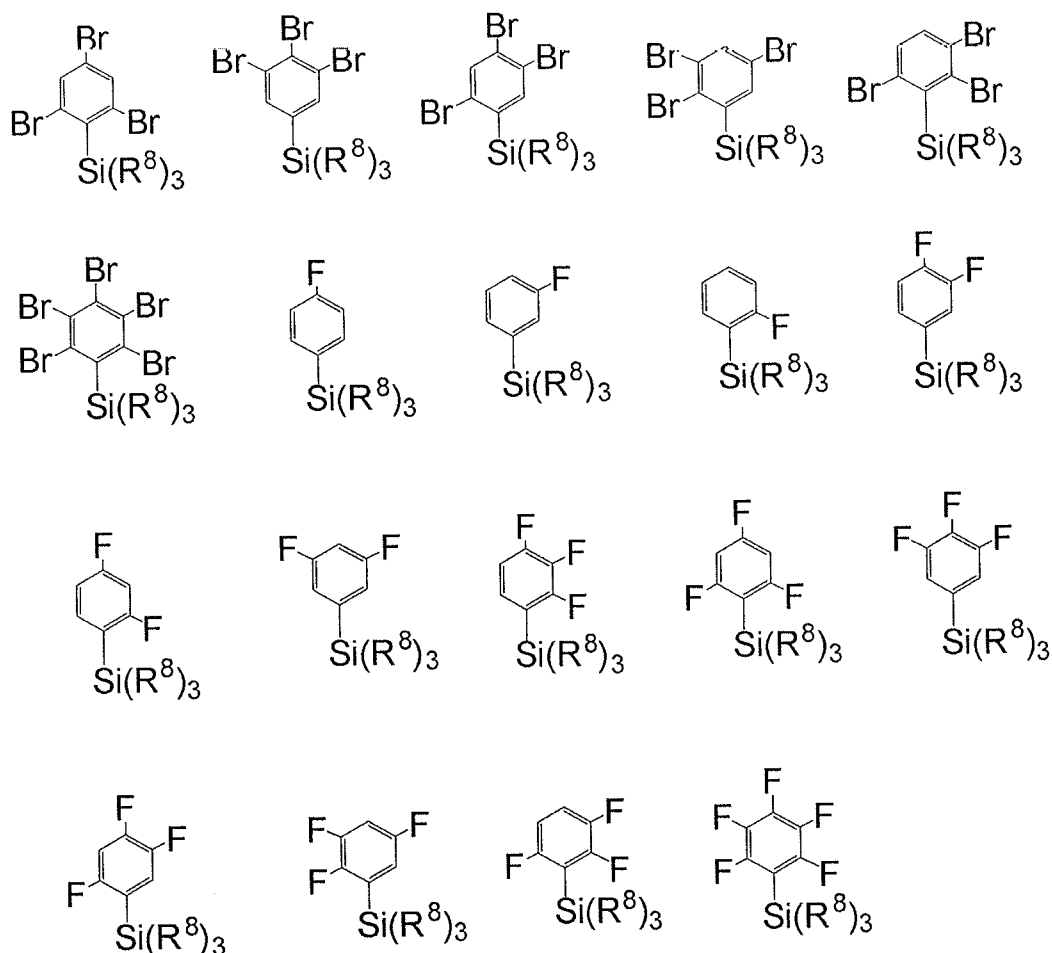
[0019] Particularly preferred is a combination of a tetraalkoxysilane such as
10 tetramethoxysilane and tetraethoxysilane with a phenyltrialkoxysilane such as
phenyltrimethoxysilane and phenyltriethoxysilane. Further preferred is a combination
of the above combination with an alkyltrialkoxysilane such as methyltrimethoxysilane
and methyltriethoxysilane.

[0020] Examples of the organic silicon compound of Formula (3) also include
15 compounds having structures of formulae below. R^8 in the structures is the same as
defined as R^8 in Formula (3).









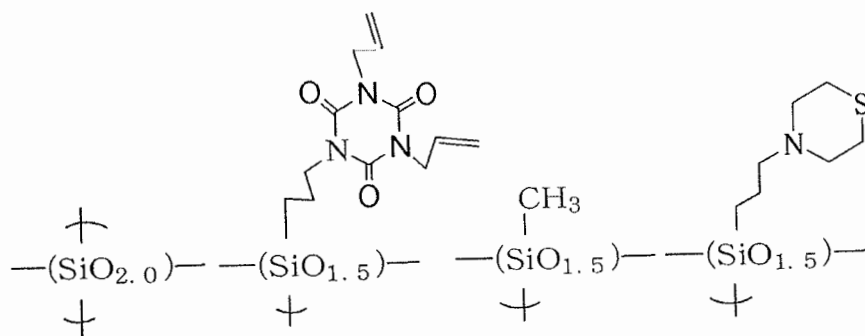
[0021] In Formula (4), R^9 is an alkyl group; R^{10} is an alkoxy group, an acyloxy group, or a halogen atom; Y is an alkylene group or an arylene group; b is an integer of 0 or 1; and c is an integer of 0 or 1.

5 As the alkyl group, the alkoxy group, the acyloxy group, or the halogen atom, those exemplified above with respect to Formula (1) and Formula (2) can be used. Examples of the alkylene group and the arylene group include divalent organic groups corresponding to the above examples of the alkyl group and the above examples of the aryl group, respectively.

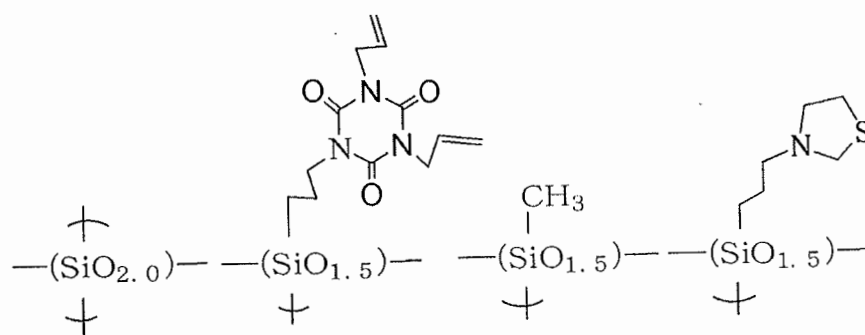
10 Examples of the organic silicon compound of Formula (4) include methylenebis(trimethoxysilane), methylenebis(trichlorosilane), methylenebis(triacetoxysilane), ethylenebis(triethoxysilane), ethylenebis(trichlorosilane), ethylenebis(triacetoxysilane), propylenebis(triethoxysilane), butylenebis(trimethoxysilane), phenylenebis(trimethoxysilane),

phenylenebis(triethoxysilane), phenylenebis(methyldiethoxysilane), phenylenebis(methyldimethoxysilane), naphthylenebis(trimethoxysilane), bis(trimethoxydisilane), bis(triethoxydisilane), bis(ethyldiethoxydisilane), and bis(methyldimethoxydisilane).

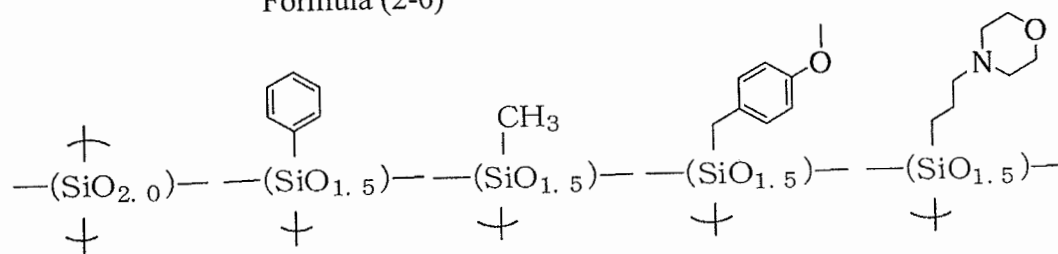
- 5 [0022] Specific examples of the hydrolysis-condensation product of the hydrolyzable organosilane of Formula (1) or Formula (2) and the organic silicon compound of Formula (3) include compounds of formulae below.



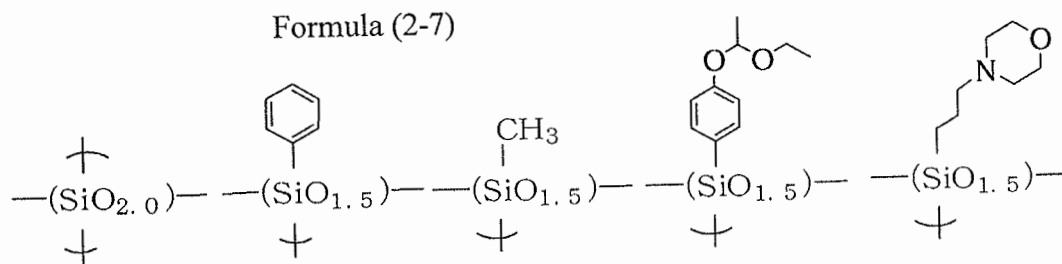
Formula (2-5)



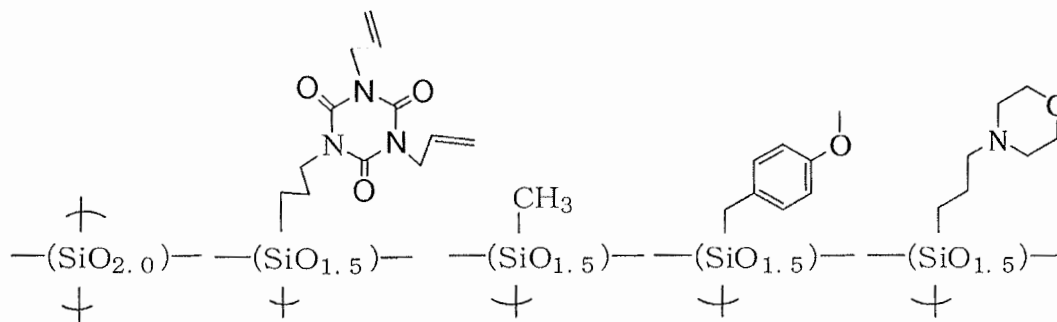
Formula (2-6)



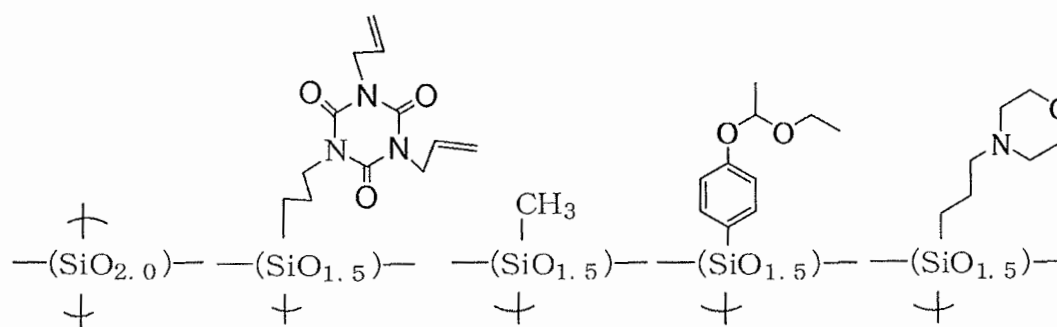
Formula (2-7)



Formula (2-8)



Formula (2-9)



Formula (2-10)

[0023] As the hydrolysis-condensation product (polyorganosiloxane) of the hydrolyzable organosilane of Formula (1) or Formula (2) and the organic silicon compound of Formula (3) or the hydrolysis-condensation product (polyorganosiloxane) of the hydrolyzable organosilane of Formula (1) or Formula (2), the organic silicon compound of Formula (3), and the organic silicon compound of Formula (4), a condensation product having a weight average molecular weight of 1,000 to 1,000,000 or 1,000 to 100,000 can be obtained. This molecular weight is a molecular weight obtained by a GPC analysis in terms of polystyrene.

Examples of the conditions for the GPC measurement include: using a GPC apparatus (trade name: HLC-8220GPC; manufactured by Tosoh Corporation); using a GPC column (trade names: Shodex KF803L, KF802, and KF801; manufactured by Showa Denko K.K.); using a column temperature of 40°C; using tetrahydrofuran as the eluting liquid (eluting solvent); using a flow amount (flow rate) of 1.0 mL/min; and using polystyrene (manufactured by Showa Denko K.K.) as the standard sample.

For the hydrolysis of an alkoxysilyl group, an acyloxysilyl group, or a halogenated silyl group, water is used in an amount of 0.5 to 100 mol, preferably 1 to 10 mol, relative

to 1 mol of a hydrolyzable group.

[0024] A catalyst for the hydrolysis can be used in an amount of 0.001 to 10 mol, preferably 0.001 to 1 mol, relative to 1 mol of a hydrolyzable group.

The reaction temperature for performing the hydrolysis and the condensation is
5 usually 20 to 80°C.

The hydrolysis may be performed either perfectly or partially. That is, in the hydrolysis-condensation product thereof, a hydrolysis product or a monomer may remain.

During the hydrolysis and the condensation, a catalyst can be used.

Examples of the catalyst for the hydrolysis include metal chelate compounds,
10 organic acids, inorganic acids, organic bases, and inorganic bases.

[0025] Examples of the metal chelate compound as the catalyst for the hydrolysis include: titanium chelate compounds such as triethoxy-mono(acetylacetonate)titanium, tri-n-propoxy-mono(acetylacetonate)titanium, triisopropoxy-mono(acetylacetonate)titanium,
15 tri-n-butoxy-mono(acetylacetonate)titanium, tri-sec-butoxy-mono(acetylacetonate)titanium, tri-tert-butoxy-mono(acetylacetonate)titanium, diethoxy-bis(acetylacetonate)titanium, di-n-propoxy-bis(acetylacetonate)titanium, di-isopropoxy-bis(acetylacetonate)titanium, di-n-butoxy-bis(acetylacetonate)titanium, di-sec-butoxy-bis(acetylacetonate)titanium,
20 di-tert-butoxy-bis(acetylacetonate)titanium, monoethoxy-tris(acetylacetonate)titanium, mono-n-propoxy-tris(acetylacetonate)titanium, mono-isopropoxy-tris(acetylacetonate)titanium, mono-n-butoxy-tris(acetylacetonate)titanium, mono-sec-butoxy-tris(acetylacetonate)titanium,
25 mono-tert-butoxy-tris(acetylacetonate)titanium, tetrakis(acetylacetonate)titanium, triethoxy-mono(ethylacetoacetate)titanium, tri-n-propoxy-mono(ethylacetoacetate)titanium, tri-isopropoxy-mono(ethylacetoacetate)titanium, tri-n-butoxy-mono(ethylacetoacetate)titanium,

- tri-sec-butoxy-mono(ethylacetoacetate)titanium,
tri-tert-butoxy-mono(ethylacetoacetate)titanium, diethoxy-bis(ethylacetoacetate)titanium,
di-n-propoxy-bis(ethylacetoacetate)titanium,
di-isopropoxy-bis(ethylacetoacetate)titanium, di-n-butoxy-bis(ethylacetoacetate)titanium,
5 di-sec-butoxy-bis(ethylacetoacetate)titanium,
di-tert-butoxy-bis(ethylacetoacetate)titanium, monoethoxy-tris(ethylacetoacetate)titanium,
mono-n-propoxy-tris(ethylacetoacetate)titanium,
mono-isopropoxy-tris(ethylacetoacetate)titanium,
mono-n-butoxy-tris(ethylacetoacetate)titanium,
10 mono-sec-butoxy-tris(ethylacetoacetate)titanium,
mono-tert-butoxy-tris(ethylacetoacetate)titanium, tetrakis(ethylacetoacetate)titanium,
mono(acetylacetonate)tris(ethylacetoacetate)titanium,
bis(acetylacetonate)bis(ethylacetoacetate)titanium, and
tris(acetylacetonate)mono(ethylacetoacetate)titanium; zirconium chelate compounds such
15 as triethoxy-mono(acetylacetonate)zirconium,
tri-n-propoxy-mono(acetylacetonate)zirconium,
tri-isopropoxy-mono(acetylacetonate)zirconium,
tri-n-butoxy-mono(acetylacetonate)zirconium,
tri-sec-butoxy-mono(acetylacetonate)zirconium,
20 tri-tert-butoxy-mono(acetylacetonate)zirconium, diethoxy-bis(acetylacetonate)zirconium,
di-n-propoxy-bis(acetylacetonate)zirconium,
di-isopropoxy-bis(acetylacetonate)zirconium, di-n-butoxy-bis(acetylacetonate)zirconium,
di-sec-butoxy-bis(acetylacetonate)zirconium,
di-tert-butoxy-bis(acetylacetonate)zirconium, monoethoxy-tris(acetylacetonate)zirconium,
25 mono-n-propoxy-tris(acetylacetonate)zirconium,
mono-isopropoxy-tris(acetylacetonate)zirconium,
mono-n-butoxy-tris(acetylacetonate)zirconium,
mono-sec-butoxy-tris(acetylacetonate)zirconium,
mono-tert-butoxy-tris(acetylacetonate)zirconium, tetrakis(acetylacetonate)zirconium,

- triethoxy-mono(ethylacetoacetate)zirconium,
 tri-n-propoxy-mono(ethylacetoacetate)zirconium,
 tri-isopropoxy-mono(ethylacetoacetate)zirconium,
 tri-n-butoxy-mono(ethylacetoacetate)zirconium,
 5 tri-sec-butoxy-mono(ethylacetoacetate)zirconium,
 tri-tert-butoxy-mono(ethylacetoacetate)zirconium,
 diethoxy-bis(ethylacetoacetate)zirconium, di-n-propoxy-bis(ethylacetoacetate)zirconium,
 di-isopropoxy-bis(ethylacetoacetate)zirconium,
 di-n-butoxy-bis(ethylacetoacetate)zirconium,
 10 di-sec-butoxy-bis(ethylacetoacetate)zirconium,
 di-tert-butoxy-bis(ethylacetoacetate)zirconium,
 monoethoxy-tris(ethylacetoacetate)zirconium,
 mono-n-propoxy-tris(ethylacetoacetate)zirconium,
 mono-isopropoxy-tris(ethylacetoacetate)zirconium,
 15 mono-n-butoxy-tris(ethylacetoacetate)zirconium,
 mono-sec-butoxy-tris(ethylacetoacetate)zirconium,
 mono-tert-butoxy-tris(ethylacetoacetate)zirconium, tetrakis(ethylacetoacetate)zirconium,
 mono(acetylacetonate)tris(ethylacetoacetate)zirconium,
 bis(acetylacetonate)bis(ethylacetoacetate)zirconium, and
 20 tris(acetylacetonate)mono(ethylacetoacetate)zirconium; and aluminum chelate
 compounds such as tris(acetylacetonate)aluminum and tris(ethylacetoacetate)aluminum.

- [0026] Examples of the organic acid as the catalyst for the hydrolysis include
 acetic acid, propionic acid, butanoic acid, pentanoic acid, hexanoic acid, heptanoic acid,
 octanoic acid, nonanoic acid, decanoic acid, oxalic acid, maleic acid, methylmalonic acid,
 25 adipic acid, sebacic acid, gallic acid, butyric acid, mellitic acid, arachidonic acid,
 shikimic acid, 2-ethylhexanoic acid, oleic acid, stearic acid, linolic acid, linoleic acid,
 salicylic acid, benzoic acid, p-aminobenzoic acid, p-toluenesulfonic acid,
 benzenesulfonic acid, monochloroacetic acid, dichloroacetic acid, trichloroacetic acid,
 trifluoroacetic acid, formic acid, malonic acid, sulfonic acid, phthalic acid, fumaric acid,

citric acid, and tartaric acid.

Examples of the inorganic acid as the catalyst for the hydrolysis include hydrochloric acid, nitric acid, sulfuric acid, hydrofluoric acid, and phosphoric acid.

Examples of the organic base as the catalyst for the hydrolysis include pyridine,
 5 pyrrole, piperazine, pyrrolidine, piperidine, picoline, trimethylamine, triethylamine, monoethanolamine, diethanolamine, dimethylmonoethanolamine, monomethyldiethanolamine, triethanolamine, diazabicyclo octane, diazabicyclo nonane, diazabicyclo undecene, and tetramethylammonium hydroxide. Examples of the
 inorganic base as the catalyst for the hydrolysis include ammonia, sodium hydroxide,
 10 potassium hydroxide, barium hydroxide, and calcium hydroxide. Among these catalysts, metal chelate compounds, organic acids, and inorganic acids are preferred and these catalysts may be used singly or in combination of two or more types thereof.

[0027] Examples of the organic solvent used for the hydrolysis include: aliphatic hydrocarbon solvents such as n-pentane, isopentane, n-hexane, isohexane, n-heptane,
 15 isoheptane, 2,2,4-trimethylpentane, n-octane, isooctane, cyclohexane, and methylcyclohexane; aromatic hydrocarbon solvents such as benzene, toluene, xylene, ethylbenzene, trimethylbenzene, methylethylbenzene, n-propylbenzene, isopropylbenzene, diethylbenzene, isobutylbenzene, triethylbenzene, di-isopropylbenzene, n-amyl naphthalene, and trimethylbenzene; monoalcohol solvents such as methanol,
 20 ethanol, n-propanol, isopropanol, n-butanol, isobutanol, sec-butanol, tert-butanol, n-pentanol, isopentanol, 2-methylbutanol, sec-pentanol, tert-pentanol, 3-methoxybutanol, n-hexanol, 2-methylpentanol, sec-hexanol, 2-ethylbutanol, sec-heptanol, heptanol-3, n-octanol, 2-ethylhexanol, sec-octanol, n-nonyl alcohol, 2,6-dimethylheptanol-4, n-decanol, sec-undecyl alcohol, trimethylnonyl alcohol, sec-tetradecyl alcohol,
 25 sec-heptadecyl alcohol, phenol, cyclohexanol, methylcyclohexanol, 3,3,5-trimethylcyclohexanol, benzyl alcohol, phenylmethylcarbinol, diacetone alcohol, and cresol; polyalcohol solvents such as ethylene glycol, propylene glycol, 1,3-butylene glycol, pentanediol-2,4, 2-methylpentanediol-2,4, hexanediol-2,5, heptanediol-2,4, 2-ethylhexanediol-1,3, diethylene glycol, dipropylene glycol, triethylene glycol,

tripropylene glycol, and glycerin; ketone solvents such as acetone, methyl ethyl ketone, methyl-n-propyl ketone, methyl-n-butyl ketone, diethyl ketone, methyl-isobutyl ketone, methyl-n-pentyl ketone, ethyl-n-butyl ketone, methyl-n-hexyl ketone, di-isobutyl ketone, trimethylnonanone, cyclohexanone, methylcyclohexanone, 2,4-pentanedione,

5 acetonylacetone, diacetone alcohol, acetophenone, and fenchone; ether solvents such as ethyl ether, isopropyl ether, n-butyl ether, n-hexyl ether, 2-ethylhexyl ether, ethylene oxide, 1,2-propylene oxide, dioxolane, 4-methyldioxolane, dioxane, dimethyldioxane, ethylene glycol monomethyl ether, ethylene glycol monoethyl ether, ethylene glycol diethyl ether, ethylene glycol mono-n-butyl ether, ethylene glycol mono-n-hexyl ether,

10 ethylene glycol monophenyl ether, ethylene glycol mono-2-ethylbutyl ether, ethylene glycol dibutyl ether, diethylene glycol monomethyl ether, diethylene glycol monoethyl ether, diethylene glycol diethyl ether, diethylene glycol mono-n-butyl ether, diethylene glycol di-n-butyl ether, diethylene glycol mono-n-hexyl ether, ethoxy triglycol, tetraethylene glycol di-n-butyl ether, propylene glycol monomethyl ether, propylene

15 glycol monoethyl ether, propylene glycol monopropyl ether, propylene glycol monobutyl ether, dipropylene glycol monomethyl ether, dipropylene glycol monoethyl ether, dipropylene glycol monopropyl ether, dipropylene glycol monobutyl ether, tripropylene glycol monomethyl ether, tetrahydrofuran, and 2-methyltetrahydrofuran; ester solvents such as diethyl carbonate, methyl acetate, ethyl acetate, γ -butyrolactone, γ -valerolactone,

20 n-propyl acetate, isopropyl acetate, n-butyl acetate, isobutyl acetate, sec-butyl acetate, n-pentyl acetate, sec-pentyl acetate, 3-methoxybutyl acetate, methylpentyl acetate, 2-ethylbutyl acetate, 2-ethylhexyl acetate, benzyl acetate, cyclohexyl acetate, methylcyclohexyl acetate, n-nonyl acetate, methyl acetoacetate, ethyl acetoacetate, ethyleneglycol monomethyl ether acetate, ethyleneglycol monoethyl ether acetate,

25 diethyleneglycol monomethyl ether acetate, diethyleneglycol monoethyl ether acetate, diethyleneglycol mono-n-butyl ether acetate, propyleneglycol monomethyl ether acetate, propyleneglycol monoethyl ether acetate, propyleneglycol monopropyl ether acetate, propyleneglycol monobutyl ether acetate, dipropyleneglycol monomethyl ether acetate, dipropyleneglycol monoethyl ether acetate, glycol diacetate, methoxytriglycol acetate,

ethyl propionate, n-butyl propionate, isoamyl propionate, diethyl oxalate, di-n-butyl oxalate, methyl lactate, ethyl lactate, n-butyl lactate, n-amyl lactate, diethyl malonate, dimethyl phthalate, and diethyl phthalate; nitrogen-containing solvents such as N-methylformamide, N,N-dimethylformamide, N,N-diethylformamide, acetamide, N-methylacetamide, N,N-dimethylacetamide, N-methylpropionamide, and N-methylpyrrolidone; and sulfur-containing solvents such as dimethyl sulfide, diethyl sulfide, thiophene, tetrahydrothiophene, dimethyl sulfoxide, sulfolane, and 1,3-propane sultone. These solvents may be used singly or in combination of two or more types thereof.

Particularly preferred are ketone solvents such as acetone, methyl ethyl ketone, methyl-n-propyl ketone, methyl-n-butyl ketone, diethyl ketone, methyl-isobutyl ketone, methyl-n-pentyl ketone, ethyl-n-butyl ketone, methyl-n-hexyl ketone, di-isobutyl ketone, trimethylnonanone, cyclohexanone, methylcyclohexanone, 2,4-pentanedione, acetonylacetone, diacetone alcohol, acetophenone, and fenchone

(1,1,3-trimethyl-2-norbornene) in terms of the preservation stability of the solution.

[0028] From the hydrolysis-condensation product (polymer) obtained by hydrolyzing and condensing a hydrolyzable organosilane in a solvent using a catalyst, an alcohol as a by-product, the used catalyst for the hydrolysis, and the used water can be simultaneously removed by distilling them under reduced pressure or the like. An acid catalyst or a base catalyst used for the hydrolysis can be removed by neutralization or ion exchange. Then, with respect to the resist underlayer film forming composition for lithography of the present invention, in the resist underlayer film forming composition containing the hydrolysis-condensation product thereof, an acid (for example, organic acid), water, an alcohol, or a combination thereof can be blended to stabilize the composition.

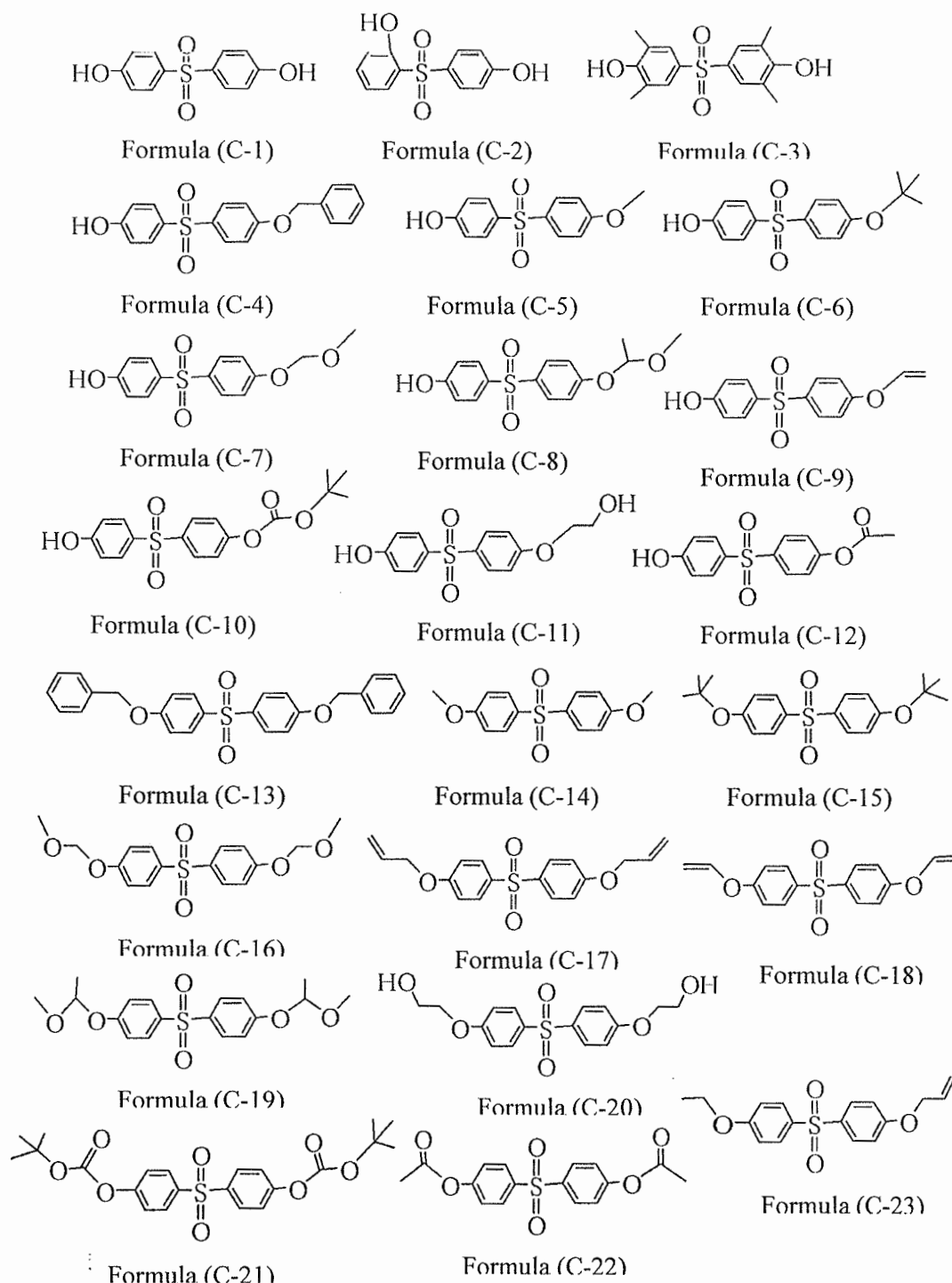
Examples of the organic acid include oxalic acid, malonic acid, methylmalonic acid, succinic acid, maleic acid, malic acid, tartaric acid, phthalic acid, citric acid, glutaric acid, citric acid, lactic acid, and salicylic acid. Among them, oxalic acid and maleic acid are preferred. The amount of the organic acid to be blended in the composition is 0.5 to 5.0

parts by mass, relative to 100 parts by mass of the condensation product (polyorganosiloxane). As the water to be blended in the composition, pure water, ultrapure water, ion-exchanged water, or the like can be used and the blending amount thereof can be 1 to 20 parts by mass, relative to 100 parts by mass of the resist underlayer film forming composition.

The alcohol to be blended in the composition is preferably an alcohol that can be easily diffused by heating the composition after the application of the composition, and examples thereof include methanol, ethanol, propanol, isopropanol, and butanol. The blending amount of the alcohol can be 1 to 20 parts by mass, relative to 100 parts by mass of the resist underlayer film forming composition.

As an additive, bisphenol S or a bisphenol S derivative can be blended in the composition. The blending amount of bisphenol S or a bisphenol S derivative is 0.01 to 20 parts by mass, or 0.01 to 10 parts by mass, or 0.01 to 5 parts by mass, relative to 100 parts by mass of the polyorganosiloxane.

[0029] Preferred examples of bisphenol S or a bisphenol S derivative include the compounds below.



The underlayer film forming composition for lithography of the present invention may contain, besides the components described above, if necessary, organic polymer compounds, photoacid generators, surfactants, and the like.

- 5 By using an organic polymer compound, the dry etching rate (a decreased amount of the film thickness per unit time), the attenuation coefficient, the refractive index, and the like of a resist underlayer film formed from the underlayer film forming composition

for lithography of the present invention can be controlled.

[0030] The organic polymer compound is not particularly limited and various organic polymers such as condensation polymerization polymers and addition polymerization polymers can be used. As the organic polymer compound, addition polymerization polymers and condensation polymerization polymers such as polyesters, polystyrenes, polyimides, acrylic polymers, methacrylic polymers, polyvinylethers, phenolnovolacs, naphtholnovolacs, polyethers, polyamides, and polycarbonates can be used. Organic polymers having an aromatic ring structure functioning as a light absorbing moiety such as a benzene ring, a naphthalene ring, an anthracene ring, a triazine ring, a quinoline ring, and a quinoxaline ring are preferably used.

Examples of such an organic polymer compound include addition polymerization polymers containing as a structure unit thereof, an addition polymerizable monomer such as benzyl acrylate, benzyl methacrylate, phenyl acrylate, naphthyl acrylate, anthryl methacrylate, anthrylmethyl methacrylate, styrene, hydroxystyrene, benzyl vinyl ether, and N-phenylmaleimide, and condensation polymerization polymers such as phenolnovolacs and naphtholnovolacs.

When an addition polymerization polymer is used as the organic polymer compound, the polymer compound may be either a homopolymer or a copolymer. For producing the addition polymerization polymer, an addition polymerizable monomer is used. Examples of such an addition polymerizable monomer include acrylic acid, methacrylic acid, acrylic acid ester compounds, methacrylic acid ester compounds, acrylamide compounds, methacrylamide compounds, vinyl compounds, styrene compounds, maleimide compounds, maleic anhydride, and acrylonitrile.

[0031] Examples of the acrylic acid ester compound include methyl acrylate, ethyl acrylate, n-hexyl acrylate, isopropyl acrylate, cyclohexyl acrylate, benzyl acrylate, phenyl acrylate, anthrylmethyl acrylate, 2-hydroxyethyl acrylate, 3-chloro-2-hydroxypropyl acrylate, 2-hydroxypropyl acrylate, 2,2,2-trifluoroethyl acrylate, 2,2,2-trichloroethyl acrylate, 2-bromoethyl acrylate, 4-hydroxybutyl acrylate, 2-methoxyethyl acrylate, tetrahydrofurfuryl acrylate,

5-acryloyloxy-6-hydroxynorbornene-2-carboxylic-6-lactone,

3-acryloxypropyltriethoxysilane, and glycidyl acrylate.

Examples of the methacrylic acid ester compound include methyl methacrylate, ethyl methacrylate, n-hexyl methacrylate, isopropyl methacrylate, cyclohexyl

- 5 methacrylate, benzyl methacrylate, phenyl methacrylate, anthrylmethyl methacrylate, 2-hydroxyethyl methacrylate, 2-hydroxypropyl methacrylate, 2,2,2-trifluoroethyl methacrylate, 2,2,2-trichloroethyl methacrylate, 2-bromoethyl methacrylate, 4-hydroxybutyl methacrylate, 2-methoxyethyl methacrylate, tetrahydrofurfuryl methacrylate, 5-methacryloyloxy-6-hydroxynorbornene-2-carboxylic-6-lactone,
- 10 3-methacryloxypropyltriethoxysilane, glycidyl methacrylate, 2-phenylethyl methacrylate, hydroxyphenyl methacrylate, and bromophenyl methacrylate.

Examples of the acrylamide compound include acrylamide, N-methylacrylamide, N-ethylacrylamide, N-benzylacrylamide, N-phenylacrylamide, N,N-dimethylacrylamide, and N-anthrylacrylamide.

- 15 Examples of the methacrylamide compound include methacrylamide, N-methylmethacrylamide, N-ethylmethacrylamide, N-benzylmethacrylamide, N-phenylmethacrylamide, N,N-dimethylmethacrylamide, and N-anthrylacrylamide.

Examples of the vinyl compound include vinyl alcohol, 2-hydroxyethyl vinyl ether, methyl vinyl ether, ethyl vinyl ether, benzyl vinyl ether, vinylacetic acid,

20 vinyltrimethoxysilane, 2-chloroethyl vinyl ether, 2-methoxyethyl vinyl ether, vinylnaphthalene, and vinylanthracene.

Examples of the styrene compound include styrene, hydroxystyrene, chlorostyrene, bromostyrene, methoxystyrene, cyanostyrene, and acetylstyrene.

- Examples of the maleimide compound include maleimide, N-methylmaleimide,
- 25 N-phenylmaleimide, N-cyclohexylmaleimide, N-benzylmaleimide, and N-hydroxyethylmaleimide.

When a condensation polymerization polymer is used as a polymer, examples of such a polymer include condensation polymerization polymers of a glycol compound and a dicarboxylic acid compound. Examples of the glycol compound include diethylene

glycol, hexamethylene glycol, and butylene glycol. Examples of the dicarboxylic acid compound include succinic acid, adipic acid, terephthalic acid, and maleic anhydride. Examples of the polymer also include polyesters, polyamides, and polyimides such as polypyromellitimide, poly(p-phenyleneterephthalamide), polybutylene terephthalate, and polyethylene terephthalate.

[0032] When the organic polymer compound contains a hydroxy group, the hydroxy group can effect a crosslinking reaction with a polyorganosiloxane.

As the organic polymer compound, a polymer compound having a weight average molecular weight of, for example, 1,000 to 1,000,000, or 3,000 to 300,000, or 5,000 to 200,000, or 10,000 to 100,000 can be used.

The organic polymer compounds may be used singly or in combination of two or more types thereof.

When the organic polymer compound is used, the content thereof is 1 to 200 parts by mass, or 5 to 100 parts by mass, or 10 to 50 parts by mass, or 20 to 30 parts by mass, relative to 100 parts by mass of the condensation product (polyorganosiloxane).

The resist underlayer film forming composition of the present invention may contain an acid generator.

Examples of the acid generator include thermoacid generators and photoacid generators.

The photoacid generator generates an acid during exposure of the resist. Therefore, the acidity of the underlayer film can be controlled. This is one method for adjusting the acidity of the underlayer film to that of the resist as an upper layer of the underlayer film. By adjusting the acidity of the underlayer film, the pattern shape of the resist formed in the upper layer can be controlled.

Examples of the photoacid generator contained in the resist underlayer film forming composition of the present invention include onium salt compounds, sulfonimide compounds, and disulfonyl diazomethane compounds.

Examples of the onium salt compound include: iodonium salt compounds such as diphenyliodoniumhexafluorophosphate, diphenyliodoniumtrifluoromethanesulfonate,

diphenyliodoniumnonafluoro normal butane sulfonate, diphenyliodoniumperfluoro normal octane sulfonate, diphenyliodoniumcamphorsulfonate, bis(4-tert-butylphenyl)iodoniumcamphorsulfonate, and bis(4-tert-butylphenyl)iodoniumtrifluoromethanesulfonate; and sulfonium salt compounds such as triphenylsulfoniumhexafluoroantimonate, triphenylsulfoniumnonafluoro normal butane sulfonate, triphenylsulfoniumcamphorsulfonate, and triphenylsulfoniumtrifluoromethanesulfonate.

Examples of the sulfonimide compound include

N-(trifluoromethanesulfonyloxy)succinimide, N-(nonafluoro normal butane sulfonyloxy)succinimide, N-(camphorsulfonyloxy)succinimide, and N-(trifluoromethanesulfonyloxy)naphthalimide.

Examples of the disulfonyldiazomethane compound include

bis(trifluoromethylsulfonyl)diazomethane, bis(cyclohexylsulfonyl)diazomethane, bis(phenylsulfonyl)diazomethane, bis(p-toluenesulfonyl)diazomethane, bis(2,4-dimethylbenzenesulfonyl)diazomethane, and methylsulfonyl-p-toluenesulfonyldiazomethane.

These photoacid generators may be used singly or in combination of two or more types thereof.

When the photoacid generator is used, the content thereof is 0.01 to 5 parts by mass, or 0.1 to 3 parts by mass, or 0.5 to 1 part by mass, relative to 100 parts by mass of the condensation product (polyorganosiloxane).

[0033] The surfactant is effective in suppressing the formation of a pin hole, a striation, and the like when the resist underlayer film forming composition for lithography of the present invention is applied onto a substrate.

Examples of the surfactant contained in the resist underlayer film forming composition of the present invention include: nonionic surfactants, for example, polyoxyethylene alkyl ethers such as polyoxyethylene lauryl ether, polyoxyethylene stearyl ether, polyoxyethylene cetyl ether, and polyoxyethylene oleyl ether, polyoxyethylene alkylaryl ethers such as polyoxyethylene octylphenol ether and

polyoxyethylene nonylphenol ether, polyoxyethylene-polyoxypropylene block copolymers, sorbitan fatty acid esters such as sorbitan monolaurate, sorbitan monopalmitate, sorbitan monostearate, sorbitan monooleate, sorbitan trioleate, and sorbitan tristearate, polyoxyethylene sorbitan fatty acid esters such as polyoxyethylene sorbitan monolaurate, polyoxyethylene sorbitan monopalmitate, polyoxyethylene sorbitan monostearate, polyoxyethylene sorbitan trioleate, and polyoxyethylene sorbitan tristearate; fluorinated surfactants, for example, EFTOP EF301, EF303, and EF352 (trade name; manufactured by Tohkem Products Corp.), MEGAFAC F171, F173, R-08, and R-30 (trade name; manufactured by Dainippon Ink & Chemicals Inc.), Fluorad FC430 and FC431 (manufactured by Sumitomo 3M Limited), AsahiGuard AG710 and Surflon S-382, SC101, SC102, SC103, SC104, SC105, and SC106 (trade names; manufactured by Asahi Glass Co., Ltd.); and Organosiloxane polymer KP341 (manufactured by Shin-Etsu Chemical Co., Ltd.). These surfactants may be used singly or in combination of two or more types thereof. When the surfactant is used, the content thereof is 0.0001 to 5 parts by mass, or 0.001 to 1 part by mass, or 0.01 to 0.5 part by mass, relative to 100 parts by mass of the condensation product (polyorganosiloxane).

[0034] In the resist underlayer film forming composition of the present invention, a rheology controlling agent and an adhesion assistant may be blended. The rheology controlling agent is effective in enhancing the fluidity of the underlayer film forming composition. The adhesion assistant is effective in enhancing the adhesion of the underlayer film to the semiconductor substrate or the resist.

[0035] The solvent used for the resist underlayer film forming composition of the present invention is not particularly limited to be used so long as the solvent can dissolve the solid content. Examples of such a solvent include methyl cellosolve acetate, ethyl cellosolve acetate, propylene glycol, propylene glycol monomethyl ether, propylene glycol monoethyl ether, methylisobutylcarbinol, propylene glycol monobutyl ether, propylene glycol monomethyl ether acetate, propylene glycol monoethyl ether acetate, propylene glycol monopropyl ether acetate, propylene glycol monobutyl ether acetate, toluene, xylene, methyl ethyl ketone, cyclopentanone, cyclohexanone, ethyl

2-hydroxypropionate, ethyl 2-hydroxy-2-methylpropionate, ethyl ethoxyacetate, ethyl hydroxyacetate, methyl 2-hydroxy-3-methylbutanoate, methyl 3-methoxypropionate, ethyl 3-methoxypropionate, ethyl 3-ethoxypropionate, methyl 3-ethoxypropionate, methyl pyruvate, ethyl pyruvate, ethylene glycol monomethyl ether, ethylene glycol monoethyl ether, ethylene glycol monopropyl ether, ethylene glycol monobutyl ether, ethylene glycol monomethyl ether acetate, ethylene glycol monoethyl ether acetate, ethylene glycol monopropyl ether acetate, ethylene glycol monobutyl ether acetate, diethylene glycol dimethyl ether, diethylene glycol diethyl ether, diethylene glycol dipropyl ether, diethylene glycol dibutyl ether, propylene glycol monomethyl ether, propylene glycol dimethyl ether, propylene glycol diethyl ether, propylene glycol dipropyl ether, propylene glycol dibutyl ether, ethyl lactate, propyl lactate, isopropyl lactate, butyl lactate, isobutyl lactate, methyl formate, ethyl formate, propyl formate, isopropyl formate, butyl formate, isobutyl formate, amyl formate, isoamyl formate, methyl acetate, ethyl acetate, amyl acetate, isoamyl acetate, hexyl acetate, methyl propionate, ethyl propionate, propyl propionate, isopropyl propionate, butyl propionate, isobutyl propionate, methyl butyrate, ethyl butyrate, propyl butyrate, isopropyl butyrate, butyl butyrate, isobutyl butyrate, ethyl hydroxyacetate, ethyl 2-hydroxy-2-methylpropionate, methyl 3-methoxy-2-methylpropionate, methyl 2-hydroxy-3-methylbutyrate, ethyl methoxyacetate, ethyl ethoxyacetate, methyl 3-methoxypropionate, ethyl 3-ethoxypropionate, ethyl 3-methoxypropionate, 3-methoxybutyl acetate, 3-methoxypropyl acetate, 3-methyl-3-methoxybutyl acetate, 3-methyl-3-methoxybutyl propionate, 3-methyl-3-methoxybutyl butyrate, methyl acetoacetate, toluene, xylene, methyl ethyl ketone, methyl propyl ketone, methyl butyl ketone, 2-heptanone, 3-heptanone, 4-heptanone, cyclohexanone, N,N-dimethylformamide, N-methylacetamide, N,N-dimethylacetamide, N-methylpyrrolidone, 4-methyl-2-pentanol, and γ -butyrolactone. These solvents may be used singly or in combination of two or more types thereof.

[0036] Hereinafter, the use of the resist underlayer film forming composition of the present invention will be described.

The resist underlayer film forming composition of the present invention is applied onto a substrate used in the production of semiconductor devices (for example, silicon wafer substrates, substrates coated with silicon/silicon dioxide, silicon nitride substrates, glass substrates, ITO substrates, polyimide substrates, substrates coated with a low dielectric constant material (low-k material)) by an appropriate coating method such as a spinner and a coater and then, is baked to form a resist underlayer film. The baking conditions are appropriately selected from baking temperatures of 80°C to 250°C and baking time of 0.3 to 60 minutes. Preferably, the baking temperature is 150°C to 250°C and the baking time is 0.5 to 2 minutes. Here, the formed underlayer film has a film thickness of, for example, 10 to 1,000 nm, or 20 to 500 nm, or 50 to 300 nm, or 100 to 200 nm.

[0037] Next, on the resist underlayer film, for example, a photoresist layer is formed. The formation of the photoresist layer can be performed by a known method, that is, by applying a photoresist composition solution onto the underlayer film and by baking the composition solution. The photoresist has a film thickness of, for example, 50 to 10,000 nm, or 100 to 2,000 nm, or 200 to 1,000 nm.

In the present invention, after the organic underlayer film is formed on the substrate, the resist underlayer film of the present invention can be formed on the organic underlayer film and further, the resist underlayer film can be coated with the photoresist. Thus, even when the pattern width of the photoresist becomes smaller and the resist underlayer film is coated thinly with the photoresist for preventing a pattern collapse, the processing of the substrate becomes possible by selecting an appropriate etching gas. For example, the resist underlayer film of the present invention can be processed with a fluorine-based gas having an etching rate of the resist underlayer film satisfactorily higher than that of the photoresist as an etching gas, and the organic underlayer film can be processed with an oxygen-based gas having an etching rate of the organic underlayer film satisfactorily higher than that of the resist underlayer film of the present invention as an etching gas. Furthermore, the substrate can be processed with a fluorine-based gas having an etching rate of the substrate satisfactorily higher than that of the organic

underlayer film as an etching gas.

The photoresist formed on the resist underlayer film of the present invention is not particularly limited so long as the photoresist is sensitive to light used for exposure, and both a negative-type photoresist and a positive-type photoresist can be used. Examples of the photoresist include: a positive-type photoresist made of a novolac resin and 1,2-naphthoquinonediazide sulfonic acid ester; a chemical amplification-type photoresist made of a binder having a group elevating the alkali dissolving rate by being decomposed by an acid, and a photoacid generator; a chemical amplification-type photoresist made of a low molecular compound elevating the alkali dissolving rate of the photoresist by being decomposed by an acid, an alkali-soluble binder, and a photoacid generator; and a chemical amplification-type photoresist made of a binder having a group elevating the alkali dissolving rate by being decomposed by an acid, a low molecular compound elevating the alkali dissolving rate of the photoresist by being decomposed by an acid, and a photoacid generator. Examples of the photoresist include trade name: APEX-E manufactured by Shipley Company, L.L.C., trade name: PAR710 manufactured by Sumitomo Chemical Co., Ltd., and trade name: SEPR430 manufactured by Shin-Etsu Chemical Co., Ltd. The examples also include fluorine atom-containing polymer-based photoresists described in Proc. SPIE, Vol. 3999, 330-334 (2000), Proc. SPIE, Vol. 3999, 357-364 (2000), and Proc. SPIE, Vol. 3999, 365-374 (2000).

[0038] Next, the exposure is performed through a predetermined mask. For the exposure, a KrF excimer laser (wavelength: 248 nm), an ArF excimer laser (wavelength: 193 nm), a F₂ excimer laser (wavelength: 157 nm), and the like can be used. After exposure, if necessary, post exposure bake can also be performed. The post exposure bake is performed under conditions appropriately selected from baking temperatures of 70°C to 150°C and baking time of 0.3 to 10 minutes.

In the present invention, as the resist, a resist for electron beam lithography or a resist for EUV lithography can be used instead of the photoresist. As the electron beam resist, both a positive type and a negative type can be used. Examples of the electron beam resist include: a chemical amplification-type resist made of an acid generator and a

binder having a group changing the alkali dissolving rate by being decomposed by an acid; a chemical amplification-type resist made of an alkali-soluble binder, an acid generator, and a low molecular compound changing the alkali dissolving rate of the resist by being decomposed by an acid; a chemical amplification-type resist made of an acid generator, a binder having a group changing the alkali dissolving rate by being decomposed by an acid, and a low molecular compound changing the alkali dissolving rate of the resist by being decomposed by an acid; a non-chemical amplification-type resist made of a binder having a group changing the alkali dissolving rate by being decomposed by an electron beam; and a non-chemical amplification-type resist made of a binder having a moiety changing the alkali dissolving rate by being broken by an electron beam. Also in the case of using the electron beam resist, a resist pattern can be formed in a manner similar to that in the case of using a photoresist, by using an electron beam as the irradiation source.

As the EUV resist, a methacrylate resin-based resist can be used.

[0039] Next, the development is performed by a developer (for example, an alkaline developer). Consequently, for example when a positive-type photoresist is used, the photoresist of an exposed part is removed to form a photoresist pattern.

Examples of the developer include alkaline aqueous solutions such as: aqueous solutions of alkali metal hydroxides such as potassium hydroxide and sodium hydroxide; aqueous solutions of quaternary ammonium hydroxides such as tetramethylammonium hydroxide, tetraethylammonium hydroxide, and choline; and aqueous solutions of amines such as ethanolamine, propylamine, and ethylenediamine. Furthermore, in these developers, a surfactant and the like may also be blended. The conditions for development are appropriately selected from temperatures of 5 to 50°C and time of 10 to 600 seconds.

In the present invention, as the developer, an organic solvent can be used. After exposure, development is performed by a developer (solvent). By development, for example when a positive-type photoresist is used, the photoresist of an unexposed part is removed to form a photoresist pattern.

Examples of the developer include methyl acetate, butyl acetate, ethyl acetate, isopropyl acetate, amyl acetate, isoamyl acetate, ethyl methoxyacetate, ethyl ethoxyacetate, propylene glycol monomethyl ether acetate, ethylene glycol monoethyl ether acetate, ethylene glycol monopropyl ether acetate, ethylene glycol monobutyl ether acetate, ethylene glycol monophenyl ether acetate, diethylene glycol monomethyl ether acetate, diethylene glycol monopropyl ether acetate, diethylene glycol monoethyl ether acetate, diethylene glycol monophenyl ether acetate, diethylene glycol monobutyl ether acetate, diethylene glycol monoethyl ether acetate, 2-methoxybutyl acetate, 3-methoxybutyl acetate, 4-methoxybutyl acetate, 3-methyl-3-methoxybutyl acetate, 3-ethyl-3-methoxybutyl acetate, propylene glycol monomethyl ether acetate, propylene glycol monoethyl ether acetate, propylene glycol monopropyl ether acetate, 2-ethoxybutyl acetate, 4-ethoxybutyl acetate, 4-propoxybutyl acetate, 2-methoxypentyl acetate, 3-methoxypentyl acetate, 4-methoxypentyl acetate, 2-methyl-3-methoxypentyl acetate, 3-methyl-3-methoxypentyl acetate, 3-methyl-4-methoxypentyl acetate, 4-methyl-4-methoxypentyl acetate, propylene glycol diacetate, methyl formate, ethyl formate, butyl formate, propyl formate, ethyl lactate, butyl lactate, propyl lactate, ethyl carbonate, propyl carbonate, butyl carbonate, methyl pyruvate, ethyl pyruvate, propyl pyruvate, butyl pyruvate, methyl acetoacetate, ethyl acetoacetate, methyl propionate, ethyl propionate, propyl propionate, isopropyl propionate, methyl 2-hydroxypropionate, ethyl 2-hydroxypropionate, methyl 3-methoxypropionate, ethyl 3-methoxypropionate, ethyl 3-ethoxypropionate, and propyl 3-methoxypropionate. Further, in these developers, a surfactant or the like may also be blended. The conditions for development are appropriately selected from temperatures of 5 to 50°C and time of 10 to 600 seconds.

[0040] Then, using the pattern of the photoresist (upper layer) thus formed as a protecting film, the removal of the resist underlayer film (intermediate layer) of the present invention is performed and next, using the film composed of the patterned photoresist and the patterned resist underlayer film (intermediate layer) of the present invention as a protecting film, the removal of the organic underlayer film (underlayer) is

performed. Finally, using the patterned resist underlayer film (intermediate layer) of the present invention and the patterned organic underlayer film (underlayer) as a protecting film, the processing of the semiconductor substrate is performed.

First, the resist underlayer film (intermediate layer) of the present invention at the part where the photoresist is removed is removed by dry etching to expose the semiconductor substrate. For dry etching the resist underlayer film of the present invention, gases such as tetrafluoromethane (CF_4), perfluorocyclobutane (C_4F_8), perfluoropropane (C_3F_8), trifluoromethane, carbon monoxide, argon, oxygen, nitrogen, sulfur hexafluoride, difluoromethane, nitrogen trifluoride and chlorine trifluoride, chlorine, and trichloroborane and dichloroborane can be used. For dry etching of the resist underlayer film, a halogen-based gas is preferably used. By dry etching with a halogen-based gas, fundamentally, a photoresist that is composed of organic substances is difficult to be removed. On the contrary, the resist underlayer film of the present invention containing a large amount of silicon atoms is immediately removed by a halogen-based gas. Therefore, the decrease in the film thickness of the photoresist due to dry etching of the resist underlayer film can be suppressed. As the result thereof, the photoresist can be used as a thin film. The resist underlayer film is dry-etched preferably with a fluorine-based gas and examples of the fluorine-based gas include tetrafluoromethane (CF_4), perfluorocyclobutane (C_4F_8), perfluoropropane (C_3F_8), trifluoromethane, and difluoromethane (CH_2F_2).

[0041] Subsequently, using the film composed of the patterned photoresist and the patterned resist underlayer film of the present invention as a protecting film, the removal of the organic underlayer film is performed. The removal of the organic underlayer film (underlayer) is performed by dry etching preferably with an oxygen-based gas. This is because the resist underlayer film of the present invention containing a large amount of silicon atoms is difficult to be removed by dry etching with an oxygen-based gas.

Finally, the processing of the semiconductor substrate is performed. The processing of the semiconductor substrate is performed by dry etching preferably with a

fluorine-based gas.

Examples of the fluorine-based gas include tetrafluoromethane (CF_4), perfluorocyclobutane (C_4F_8), perfluoropropane (C_3F_8), trifluoromethane, and difluoromethane (CH_2F_2).

5 [0042] As an upper layer of the resist underlayer film of the present invention, an organic anti-reflective coating can be formed before the formation of the photoresist. The anti-reflective coating composition used here is not particularly limited and can be optionally selected from the compositions commonly used in a conventional lithography process to be used. The formation of the anti-reflective coating can be performed by a
10 commonly used method, for example, by applying an anti-reflective coating composition by a spinner or a coater and by baking the composition.

The substrate onto which the resist underlayer film forming composition of the present invention is applied may also be a substrate having an organic or inorganic anti-reflective coating formed by a CVD method on its surface and, on the anti-reflective
15 coating, the underlayer film of the present invention can also be formed.

A resist underlayer film formed from the resist underlayer film forming composition of the present invention may absorb light used in a lithography process depending on the wavelength of light. Then, in such a case, the resist underlayer film can function as an anti-reflective coating having the effect of preventing light reflected on the substrate.
20 Furthermore, the underlayer film of the present invention can also be used as a layer for preventing an interaction between the substrate and the photoresist, a layer having a function of preventing an adverse action of a material used in the photoresist or of a substance generated during exposure of the photoresist against the substrate, a layer having a function of preventing the diffusion of a substance generated in or on the
25 substrate during heating and baking to the upper layer photoresist, a barrier layer for reducing a poisoning effect to the photoresist layer by a semiconductor substrate dielectric layer, and the like.

A resist underlayer film formed from the resist underlayer film forming composition can be applied to a substrate in which a via hole used in the dual damascene process is

formed to be used as an embedding material capable of filling the hole without any void. The resist underlayer film can also be used as a planarizing material for planarizing the surface of a semiconductor substrate having unevenness.

The above resist underlayer film can be used as an underlayer film of an EUV resist also for the following purpose besides the purpose of the function as a hardmask. The above resist underlayer film forming composition can be used as an EUV resist underlayer anti-reflective coating capable of preventing a reflection of exposure light undesirable during EUV exposure (wavelength: 13.5 nm) such as UV and DUV (ArF light and KrF light) on the substrate or the interface without causing intermixing with the EUV resist. The above resist underlayer film forming composition can efficiently prevent light reflection as an underlayer of an EUV resist. When the above resist underlayer film forming composition is used as the EUV resist underlayer film, the process can be performed in a manner similar to that in the case of an underlayer film for a photoresist.

Examples

[0043] (Synthesis of Compound 1)

Into a 100 mL eggplant-shaped flask, 15.00 g (0.17 mol) of morpholine and 13.82 g (0.057 mol) of 3-chloropropyltriethoxysilane were charged. The resultant mixture was heated to 120°C with stirring using a magnetic stirrer to be reacted for 4 hours. The hydrochloride generated by the reaction was removed by filtration. The filtrate was distilled under reduced pressure to obtain Compound 1 as the target product (Corresponding to Formula (1-1). X is an ethoxy group).

¹H-NMR (500MHz): 0.55 ppm (t, 2H), 1.13 ppm (t, 9H), 1.45 ppm (quint, 2H), 2.22 - 2.30 ppm (m, 6H), 3.55 ppm (t, 4H), 3.74 ppm (q, 6H)

[0044] (Synthesis of Compound 2)

A synthesis was performed in a similar manner to that of Compound 1 except that morpholine was changed to thiomorpholine of the same number of moles to obtain Compound 2 as the target product (Corresponding to Formula (1-2). X is an ethoxy group).

¹H-NMR (500MHz): 0.52 ppm (t, 2H), 1.15 ppm (t, 9H), 1.45 ppm (quint, 2H), 2.28 ppm (t, 2H), 2.50 - 2.62 ppm (m, 8H), 3.74 ppm (q, 6H)

[0045] (Synthesis of Compound 3)

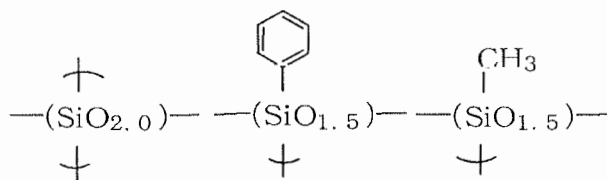
A synthesis was performed in a similar manner to that of Compound 1 except that morpholine was changed to thiazolidine of the same number of moles to obtain Compound 3 as the target product (Corresponding to Formula (1-4). X is an ethoxy group).

¹H-NMR (500MHz): 0.58 ppm (t, 2H), 1.14 ppm (t, 9H), 1.48 ppm (quint, 2H), 2.26 ppm (t, 2H), 2.77 ppm (t, 2H), 2.97 ppm (t, 2H), 3.75 ppm (q, 6H), 3.99 ppm (s, 2H)

[0046] (Synthesis Example 1)

25.70 g (70 mol%) of tetraethoxysilane, 7.86 g (25 mol%) of methyltriethoxysilane, 1.75 g (5 mol%) of phenyltrimethoxysilane, and 52.95 g of acetone were charged into a 200 mL flask. While the resultant mixed solution was stirred with a magnetic stirrer, 11.75 g of a 0.01 mol/L hydrochloric acid was added dropwise into the mixed solution.

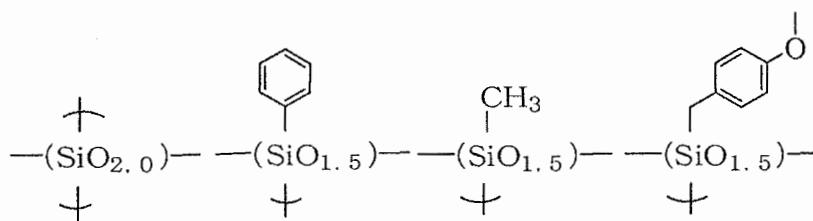
After the addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 35 g of propylene glycol monomethyl ether acetate was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a hydrolysis-condensation product (polymer) propylene glycol monomethyl ether acetate solution. To the obtained solution, propylene glycol monoethyl ether was added to adjust the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80. The resultant polymer corresponded to Formula (3-1) and had a weight average molecular weight measured by GPC of Mw 1,600 in terms of polystyrene.



Formula (3-1)

[0047] (Synthesis Example 2)

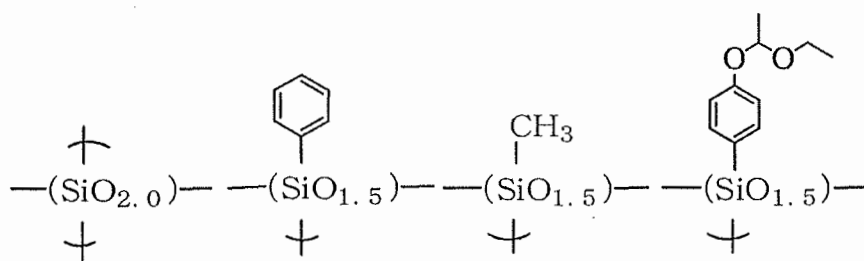
- 25.59 g (70 mol%) of tetraethoxysilane, 7.35 g (23.5 mol%) of
- 5 methyltriethoxysilane, 1.74 g (5 mol%) of phenyltrimethoxysilane, 0.64 g (1.5 mol%) of (4-methoxybenzyl)trimethoxysilane, and 52.98 g of acetone were charged into a 200 mL flask. While the resultant mixed solution was stirred with a magnetic stirrer, 11.70 g of a 0.01 mol/L hydrochloric acid was added dropwise into the mixed solution. After the addition, the flask was transferred into an oil bath adjusted to 85°C, and under
- 10 warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 35 g of propylene glycol monomethyl ether acetate was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a
- 15 hydrolysis-condensation product (polymer) propylene glycol monomethyl ether acetate solution. To the obtained solution, propylene glycol monoethyl ether was added to adjust the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80. The resultant polymer corresponded to Formula
- 20 (3-2) and had a weight average molecular weight measured by GPC of Mw 1,600 in terms of polystyrene.



Formula (3-2)

[0048] (Synthesis Example 3)

25.52 g (70 mol%) of tetraethoxysilane, 7.33 g (23.5 mol%) of methyltriethoxysilane, 1.73 g (5 mol%) of phenyltrimethoxysilane, 0.75 g (1.5 mol%) of (4-ethoxyethoxy)trimethoxysilylbenzene, and 53.27 g of acetone were charged into a 200 mL flask. While the resultant mixed solution was stirred with a magnetic stirrer, 11.67 g of a 0.01 mol/L hydrochloric acid was added dropwise into the mixed solution. After the addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 35 g of propylene glycol monomethyl ether acetate was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a hydrolysis-condensation product (polymer) propylene glycol monomethyl ether acetate solution. To the obtained solution, propylene glycol monoethyl ether was added to adjust the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80. The resultant polymer corresponded to Formula (3-3) and had a weight average molecular weight measured by GPC of Mw 1,600 in terms of polystyrene.

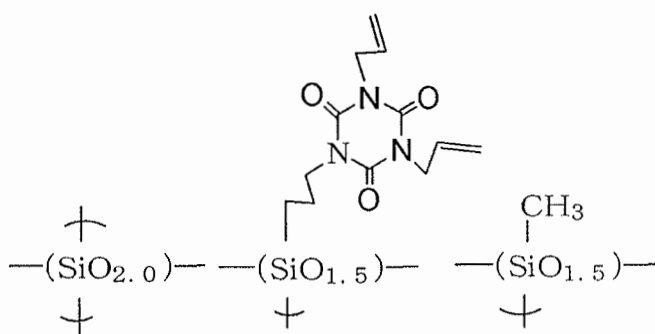


Formula (3-3)

[0049] (Synthesis Example 4)

24.53 g (70 mol%) of tetraethoxysilane, 7.50 g (25 mol%) of methyltriethoxysilane, 3.48 g (5 mol%) of 3-(triethoxysilylpropyl)diallyl isocyanurate, and 53.27 g of acetone were charged into a 200 mL flask. While the resultant mixed solution was stirred with a

magnetic stirrer, 11.22 g of a 0.01 mol/L hydrochloric acid was added dropwise into the mixed solution. After the addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 32 g of propylene glycol monomethyl ether acetate was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a hydrolysis-condensation product (polymer) propylene glycol monomethyl ether acetate solution. To the obtained solution, propylene glycol monoethyl ether was added to adjust the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80. The resultant polymer corresponded to Formula (3-4) and had a weight average molecular weight measured by GPC of Mw 1500 in terms of polystyrene.



Formula (3-4)

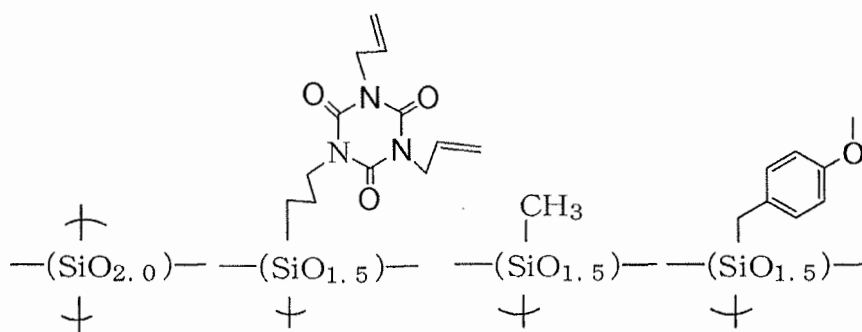
[0050] (Synthesis Example 5)

24.44 g (70 mol%) of tetraethoxysilane, 7.02 g (23.5 mol%) of methyltriethoxysilane, 3.47 g (5 mol%) of 3-(triethoxysilylpropyl)diallyl isocyanurate, 0.61 g (1.5 mol%) of (4-methoxybenzyl)trimethoxysilane, and 53.27 g of acetone were charged into a 200 mL flask. While the resultant mixed solution was stirred with a magnetic stirrer, 11.17 g of a 0.01 mol/L hydrochloric acid was added dropwise into the mixed solution. After the addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution

was cooled down to room temperature, and to the reaction solution, 36 g of propylene glycol monomethyl ether acetate was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a

5 hydrolysis-condensation product (polymer) propylene glycol monomethyl ether acetate solution. To the obtained solution, propylene glycol monoethyl ether was added to adjust the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80. The resultant polymer corresponded to Formula

10 (3-5) and had a weight average molecular weight measured by GPC of M_w 1600 in terms of polystyrene.



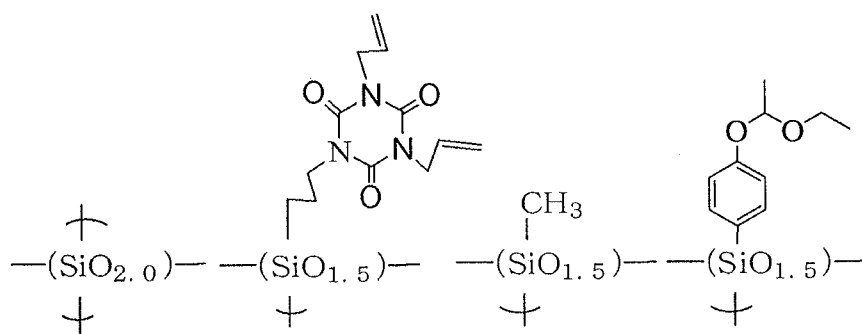
Formula (3-5)

[0051] (Synthesis Example 6)

15 24.37 g (70 mol%) of tetraethoxysilane, 7.00 g (23.5 mol%) of methyltriethoxysilane, 3.46 g (5 mol%) of 3-(triethoxysilylpropyl)diallyl isocyanurate, 0.72 g (1.5 mol%) of (4-ethoxyethoxy)trimethoxysilylbenzene, and 53.32 g of acetone were charged into a 200 mL flask. While the resultant mixed solution was stirred with a magnetic stirrer, 11.14 g of a 0.01 mol/L hydrochloric acid was added dropwise into the

20 mixed solution. After the addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 32 g of propylene glycol monomethyl ether acetate was added. From the resultant reaction

solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a hydrolysis-condensation product (polymer) propylene glycol monomethyl ether acetate solution. To the obtained solution, propylene glycol monoethyl ether was added to adjust the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80. The resultant polymer corresponded to Formula (3-6) and had a weight average molecular weight measured by GPC of Mw 1600 in terms of polystyrene.



Formula (3-6)

[0052] (Synthesis Example 7)

25.67 g (70 mol%) of tetraethoxysilane, 7.78 g (24.8 mol%) of methyltriethoxysilane, 1.75 g (5 mol%) of phenyltrimethoxysilane, 0.10 g (0.02 mol%) of Compound 1, and 52.96 g of acetone were charged into a 200 mL flask. While the resultant mixed solution was stirred with a magnetic stirrer, 11.74 g of a 0.1 mol/L hydrochloric acid was added dropwise into the mixed solution. After the addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 35 g of propylene glycol monoethyl ether was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a hydrolysis-condensation product (polymer)

propylene glycol monoethyl ether solution. To the obtained solution, propylene glycol monomethyl ether acetate was added to adjust the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80. The resultant polymer corresponded to Formula (2-1) and had a weight average molecular weight measured by GPC of Mw 1600 in terms of polystyrene.

[0053] (Synthesis Example 8)

25.67 g (70 mol%) of tetraethoxysilane, 7.78 g (24.8 mol%) of methyltriethoxysilane, 1.75 g (5 mol%) of phenyltrimethoxysilane, 0.11 g (0.02 mol%) of Compound 2, and 52.96 g of acetone were charged into a 200 mL flask. While the resultant mixed solution was stirred with a magnetic stirrer, 11.73 g of a 0.1 mol/L hydrochloric acid was added dropwise into the mixed solution. After the addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 35 g of propylene glycol monoethyl ether was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a hydrolysis-condensation product (polymer) propylene glycol monoethyl ether solution.

To the obtained solution, propylene glycol monomethyl ether acetate was added to adjust the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80. The resultant polymer corresponded to Formula (2-2) and had a weight average molecular weight measured by GPC of Mw 1600 in terms of polystyrene.

[0054] (Synthesis Example 9)

25.67 g (70 mol%) of tetraethoxysilane, 7.78 g (24.8 mol%) of methyltriethoxysilane, 1.75 g (5 mol%) of phenyltrimethoxysilane, 0.10 g (0.02 mol%) of Compound 3, and 52.96 g of acetone were charged into a 200 mL flask. While the

resultant mixed solution was stirred with a magnetic stirrer, 11.74 g of a 0.1 mol/L hydrochloric acid was added dropwise into the mixed solution. After the addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 35 g of propylene glycol monoethyl ether was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a hydrolysis-condensation product (polymer) propylene glycol monoethyl ether solution.

- 10 To the obtained solution, propylene glycol monomethyl ether acetate was added to adjust the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80. The resultant polymer corresponded to Formula (2-3) and had a weight average molecular weight measured by GPC of Mw 1600 in terms of polystyrene.

[0055] (Synthesis Example 10)

- 24.51 g (70 mol%) of tetraethoxysilane, 7.43 g (24.8 mol%) of methyltriethoxysilane, 3.48 g (5 mol%) of 3-(triethoxysilylpropyl)diallyl isocyanurate, 0.10 g (0.02 mol%) of Compound 1, and 53.28 g of acetone were charged into a 200 mL flask. While the resultant mixed solution was stirred with a magnetic stirrer, 11.21 g of a 0.1 mol/L hydrochloric acid was added dropwise into the mixed solution. After the addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 36 g of propylene glycol monoethyl ether was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a hydrolysis-condensation product (polymer) propylene glycol monoethyl ether solution. To the obtained solution, propylene glycol monomethyl ether acetate was added to adjust

the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80. The resultant polymer corresponded to Formula (2-4) and had a weight average molecular weight measured by GPC of Mw 1600 in terms of polystyrene.

[0056] (Synthesis Example 11)

24.51 g (70 mol%) of tetraethoxysilane, 7.43 g (24.8 mol%) of methyltriethoxysilane, 3.48 g (5 mol%) of 3-(triethoxysilylpropyl)diallyl isocyanurate, 0.10 g (0.02 mol%) of Compound 2, and 53.28 g of acetone were charged into a 200 mL flask. While the resultant mixed solution was stirred with a magnetic stirrer, 11.21 g of a 0.1 mol/L hydrochloric acid was added dropwise into the mixed solution. After the addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 36 g of propylene glycol monoethyl ether was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a hydrolysis-condensation product (polymer) propylene glycol monoethyl ether solution. To the obtained solution, propylene glycol monomethyl ether acetate was added to adjust the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80. The resultant polymer corresponded to Formula (2-5) and had a weight average molecular weight measured by GPC of Mw 1600 in terms of polystyrene.

[0057] (Synthesis Example 12)

24.51 g (70 mol%) of tetraethoxysilane, 7.43 g (24.8 mol%) of methyltriethoxysilane, 3.48 g (5 mol%) of 3-(triethoxysilylpropyl)diallyl isocyanurate, 0.10 g (0.02 mol%) of Compound 3, and 53.28 g of acetone were charged into a 200 mL flask. While the resultant mixed solution was stirred with a magnetic stirrer, 11.21 g of

a 0.1 mol/L hydrochloric acid was added dropwise into the mixed solution. After the addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 36 g of propylene glycol monoethyl ether was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a hydrolysis-condensation product (polymer) propylene glycol monoethyl ether solution. To the obtained solution, propylene glycol monomethyl ether acetate was added to adjust the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80. The resultant polymer corresponded to Formula (2-6) and had a weight average molecular weight measured by GPC of Mw 1700 in terms of polystyrene.

[0058] (Synthesis Example 13)

25.56 g (70 mol%) of tetraethoxysilane, 7.28 g (23.3 mol%) of methyltriethoxysilane, 1.74 g (5 mol%) of phenyltrimethoxysilane, 0.64 g (1.5 mol%) of (4-methoxybenzyl)trimethoxysilane, 0.10 g (0.02 mol%) of Compound 1, and 52.96 g of acetone were charged into a 200 mL flask. While the resultant mixed solution was stirred with a magnetic stirrer, 11.69 g of a 0.1 mol/L hydrochloric acid was added dropwise into the mixed solution. After the addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 35 g of propylene glycol monoethyl ether was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a hydrolysis-condensation product (polymer) propylene glycol monoethyl ether solution. To the obtained solution, propylene glycol monomethyl ether acetate was added to adjust the resultant solution to contain a solid residue in a proportion

of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80. The resultant polymer corresponded to Formula (2-7) and had a weight average molecular weight measured by GPC of Mw 1600 in terms of polystyrene.

5 [0059] (Synthesis Example 14)

25.49 g (70 mol%) of tetraethoxysilane, 7.26 g (23.3 mol%) of methyltriethoxysilane, 1.73 g (5 mol%) of phenyltrimethoxysilane, 0.75 g (1.5 mol%) of (4-ethoxyethoxy)trimethoxysilylbenzene, 0.10 g (0.02 mol%) of Compound 1, and 53.00 g of acetone were charged into a 200 mL flask. While the resultant mixed solution was stirred with a magnetic stirrer, 11.65 g of a 0.1 mol/L hydrochloric acid was added dropwise into the mixed solution. After the addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 35 g of propylene glycol monoethyl ether was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a hydrolysis-condensation product (polymer) propylene glycol monoethyl ether solution. To the obtained solution, propylene glycol monomethyl ether acetate was added to adjust the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80. The resultant polymer corresponded to Formula (2-8) and had a weight average molecular weight measured by GPC of Mw 1600 in terms of polystyrene.

[0060] (Synthesis Example 15)

25 24.41 g (70 mol%) of tetraethoxysilane, 6.95 g (23.3 mol%) of methyltriethoxysilane, 3.46 g (5 mol%) of 3-(triethoxysilylpropyl)diallyl isocyanurate, 0.61 g (1.5 mol%) of (4-methoxybenzyl)trimethoxysilane, 0.10 g (0.02 mol%) of Compound 1, and 53.30 g of acetone were charged into a 200 mL flask. While the resultant mixed solution was stirred with a magnetic stirrer, 11.16 g of a 0.1 mol/L

hydrochloric acid was added dropwise into the mixed solution. After the addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 36 g of propylene glycol monoethyl ether was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a hydrolysis-condensation product (polymer) propylene glycol monoethyl ether solution. To the obtained solution, propylene glycol monomethyl ether acetate was added to adjust the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80. The resultant polymer corresponded to Formula (2-9) and had a weight average molecular weight measured by GPC of Mw 1700 in terms of polystyrene.

[0061] (Synthesis Example 16)

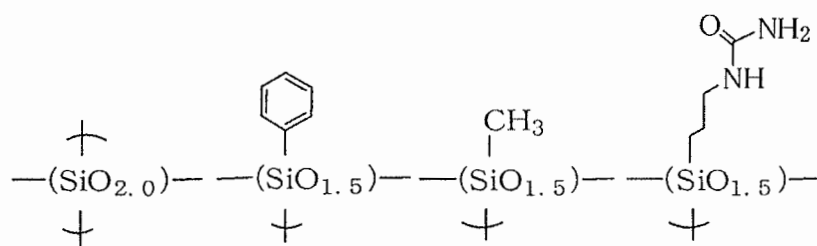
24.35 g (70 mol%) of tetraethoxysilane, 6.94 g (23.3 mol%) of methyltriethoxysilane, 3.46 g (5 mol%) of 3-(triethoxysilylpropyl)diallyl isocyanurate, 0.71 g (1.5 mol%) of 4-ethoxyethoxy)trimethoxysilylbenzene, 0.10 g (0.02 mol%) of Compound 1, and 53.32 g of acetone were charged into a 200 mL flask. While the resultant mixed solution was stirred with a magnetic stirrer, 11.13 g of a 0.1 mol/L hydrochloric acid was added dropwise into the mixed solution. After the addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 36 g of propylene glycol monoethyl ether was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a hydrolysis-condensation product (polymer) propylene glycol monoethyl ether solution. To the obtained solution, propylene glycol monomethyl ether acetate was added to adjust the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene

glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80.

The resultant polymer corresponded to Formula (2-10) and had a weight average molecular weight measured by GPC of Mw 1600 in terms of polystyrene.

[0062] (Comparative Synthesis Example 1)

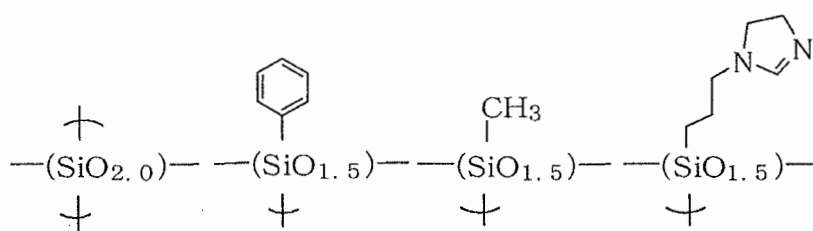
5 25.69 g (70 mol%) of tetraethoxysilane, 7.79 g (24.8 mol%) of methyltriethoxysilane, 1.75 g (5 mol%) of phenyltrimethoxysilane, 0.08 g (0.02 mol%) of 1-[3-(trimethoxysilyl)propyl]urea, and 52.95 g of ethanol were charged into a 200 mL flask. While the resultant mixed solution was stirred with a magnetic stirrer, 11.74 g of a 0.1 mol/L hydrochloric acid was added dropwise into the mixed solution. After the
10 addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 35 g of propylene glycol monoethyl ether was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced
15 pressure, and the resultant reaction mixture was concentrated to obtain a hydrolysis-condensation product (polymer) propylene glycol monoethyl ether solution. To the obtained solution, propylene glycol monomethyl ether acetate was added to adjust the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol
20 monoethyl ether was 20/80. The resultant polymer corresponded to Formula (3-7) and had a weight average molecular weight measured by GPC of Mw 5000 in terms of polystyrene.



Formula (3-7)

25 [0063] (Comparative Synthesis Example 2)

25.68 g (70 mol%) of tetraethoxysilane, 7.79 g (24.8 mol%) of methyltriethoxysilane, 1.75 g (5 mol%) of phenyltrimethoxysilane, 0.10 g (0.02 mol%) of N-(3-triethoxysilylpropyl)-4,5-dihydroimidazole, and 52.96 g of acetone were charged into a 200 mL flask. While the resultant mixed solution was stirred with a magnetic stirrer, 11.74 g of a 0.1 mol/L hydrochloric acid was added dropwise into the mixed solution. After of the addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 35 g of propylene glycol monoethyl ether was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a hydrolysis-condensation product (polymer) propylene glycol monoethyl ether solution. To the obtained solution, propylene glycol monomethyl ether acetate was added to adjust the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80. The resultant polymer corresponded to Formula (3-8) and had a weight average molecular weight measured by GPC of Mw 1600 in terms of polystyrene.

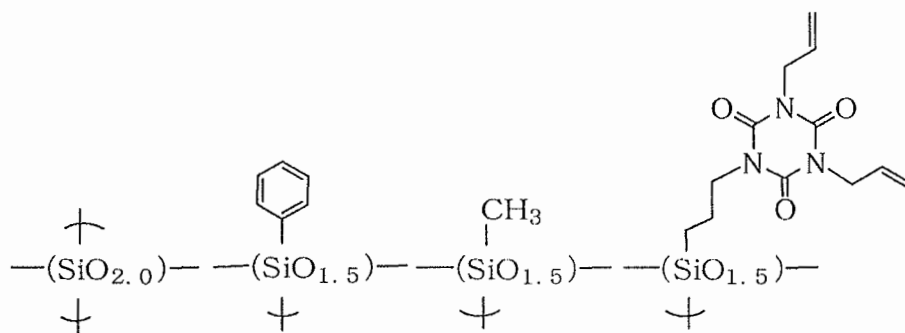


Formula (3-8)

[0064] (Comparative Synthesis Example 3)

25.65 g (70 mol%) of tetraethoxysilane, 7.78 g (24.8 mol%) of methyltriethoxysilane, 1.74 g (5 mol%) of phenyltrimethoxysilane, 0.15 g (0.02 mol%) of 3-(triethoxysilylpropyl)diallyl isocyanurate, and 52.97 g of acetone were charged into a 200 mL flask. While the resultant mixed solution was stirred with a magnetic stirrer,

11.72 g of a 0.1 mol/L hydrochloric acid was added dropwise into the mixed solution. After the addition, the flask was transferred into an oil bath adjusted to 85°C, and under warming-reflux, the reaction was effected for 240 minutes. Then, the reaction solution was cooled down to room temperature, and to the reaction solution, 35 g of propylene glycol monoethyl ether was added. From the resultant reaction solution, ethanol as a reaction by-product, water, and hydrochloric acid were distilled off under reduced pressure, and the resultant reaction mixture was concentrated to obtain a hydrolysis-condensation product (polymer) propylene glycol monomethyl ether acetate solution. To the obtained solution, propylene glycol monoethyl ether was added to adjust the resultant solution to contain a solid residue in a proportion of 15% by mass at 140°C while the solvent ratio of propylene glycol monomethyl ether acetate/propylene glycol monoethyl ether was 20/80. The resultant polymer corresponded to Formula (3-9) and had a weight average molecular weight measured by GPC of Mw 1600 in terms of polystyrene.



Formula (3-9)

[0065] (Production of Resist Underlayer Film)

Each of the silicon-containing polymers obtained in Synthesis Examples 1 to 16 and Comparative Synthesis Examples 1 to 3, an acid, a curing catalyst, an additive, a solvent, and water were mixed in ratios shown in Table 1 to Table 4 and the resultant mixture was filtered with 0.1 μm fluorine resin filter to prepare each solution of the resist underlayer film forming compositions.

In Table 1 to Table 4, maleic acid is abbreviated as MA;

N-(3-triethoxysilylpropyl)-4,5-dihydroimidazole is abbreviated as IMIDTEOS; bisphenol S is abbreviated as BPS; triphenylsulfonium trifluoromethane sulfonate is abbreviated as TPS105; monotriphenylsulfonium maleate is abbreviated as TPSMA; propylene glycol monomethyl ether acetate is abbreviated as PGMEA; and propylene

- 5 glycol monoethyl ether is abbreviated as PGEE. As the water, ultrapure water was used. Each blending amount is expressed in parts by mass. The blending ratio of the polymer is indicated not as the mass of the polymer solution, but as the mass of the polymer itself.

Table 1

	Polymer	Acid	Curing catalyst	Additive	Solvent	Water	
Example 1	Synthesis Example 1	MA	Compound 1		PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012		7	80	13
Example 2	Synthesis Example 1	MA	Compound 2		PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012		7	80	13
Example 3	Synthesis Example 1	MA	Compound 3		PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012		7	80	13
Example 4	Synthesis Example 2	MA	Compound 1		PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012		7	80	13
Example 5	Synthesis Example 3	MA	Compound 1		PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012		7	80	13
Example 6	Synthesis Example 4	MA	Compound 1		PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012		7	80	13
Example 7	Synthesis Example 4	MA	Compound 2		PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012		7	80	13
Example 8	Synthesis Example 4	MA	Compound 3		PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012		7	80	13
Example 9	Synthesis Example 5	MA	Compound 1		PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012		7	80	13
Example 10	Synthesis Example 6	MA	Compound 1		PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012		7	80	13

Table 2

	Polymer	Acid	Curing catalyst	Additive	Solvent	Water	
Example 11	Synthesis Example 7	MA			PGMEA	PGEE	Water
(parts by mass)	2	0.06			7	80	13
Example 12	Synthesis Example 8	MA			PGMEA	PGEE	Water
(parts by mass)	2	0.06			7	80	13
Example 13	Synthesis Example 9	MA			PGMEA	PGEE	Water
(parts by mass)	2	0.06			7	80	13
Example 14	Synthesis Example 10	MA			PGMEA	PGEE	Water
(parts by mass)	2	0.06			7	80	13
Example 15	Synthesis Example 11	MA			PGMEA	PGEE	Water
(parts by mass)	2	0.06			7	80	13
Example 16	Synthesis Example 12	MA			PGMEA	PGEE	Water
(parts by mass)	2	0.06			7	80	13
Example 17	Synthesis Example 13	MA			PGMEA	PGEE	Water
(parts by mass)	2	0.06			7	80	13
Example 18	Synthesis Example 14	MA			PGMEA	PGEE	Water
(parts by mass)	2	0.06			7	80	13
Example 19	Synthesis Example 15	MA			PGMEA	PGEE	Water
(parts by mass)	2	0.06			7	80	13
Example 20	Synthesis Example 16	MA			PGMEA	PGEE	Water
(parts by mass)	2	0.06			7	80	13

Table 3

	Polymer	Acid	Curing catalyst	Additive	Solvent		Water
Example 21	Synthesis Example 1	MA	Compound 1	BPS	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.10	7	80	13
Example 22	Synthesis Example 1	MA	Compound 1	TPS105	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.02	7	80	13
Example 23	Synthesis Example 1	MA	Compound 1	TPS-MA	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.02	7	80	13
Example 24	Synthesis Example 2	MA	Compound 1	BPS	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.10	7	80	13
Example 25	Synthesis Example 2	MA	Compound 1	TPS105	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.02	7	80	13
Example 26	Synthesis Example 2	MA	Compound 1	TPS-MA	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.02	7	80	13
Example 27	Synthesis Example 3	MA	Compound 1	BPS	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.10	7	80	13
Example 28	Synthesis Example 3	MA	Compound 1	TPS105	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.02	7	80	13
Example 29	Synthesis Example 3	MA	Compound 1	TPS-MA	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.02	7	80	13
Example 30	Synthesis Example 4	MA	Compound 1	BPS	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.10	7	80	13

Table 4

	Polymer	Acid	Curing catalyst	Additive	Solvent		Water
Example 31	Synthesis Example 4	MA	Compound 1	TPS105	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.02	7	80	13
Example 32	Synthesis Example 4	MA	Compound 1	TPS-MA	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.02	7	80	13
Example 33	Synthesis Example 5	MA	Compound 1	BPS	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.10	7	80	13
Example 34	Synthesis Example 5	MA	Compound 1	TPS105	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.02	7	80	13
Example 35	Synthesis Example 5	MA	Compound 1	TPS-MA	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.02	7	80	13
Example 36	Synthesis Example 6	MA	Compound 1	BPS	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.10	7	80	13
Example 37	Synthesis Example 6	MA	Compound 1	TPS105	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.02	7	80	13
Example 38	Synthesis Example 6	MA	Compound 1	TPS-MA	PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012	0.02	7	80	13
Comparative Example 1	Synthesis Example 1	MA			PGMEA	PGEE	Water
(parts by mass)	2	0.06			7	80	13
Comparative Example 2	Synthesis Example 1	MA	BTEAC		PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012		7	80	13
Comparative Example 3	Synthesis Example 1	MA	IMIDTEOS		PGMEA	PGEE	Water
(parts by mass)	2	0.06	0.012		7	80	13
Comparative Example 4	Comparative Synthesis Example 1	MA			PGMEA	PGEE	Water
(parts by mass)	2	0.06			7	80	13
Comparative Example 5	Comparative Synthesis Example 2	MA			PGMEA	PGEE	Water
(parts by mass)	2	0.06			7	80	13
Comparative Example 6	Comparative Synthesis Example 3	MA			PGMEA	PGEE	Water
(parts by mass)	2	0.06			7	80	13

[0066] (Solvent Resistance Test)

The Si-containing resist underlayer film forming compositions prepared in Example 1 to Example 38 and Comparative Example 1 to Comparative Example 6 were each applied onto corresponding silicon wafers by a spin coating method. The composition was baked on a hot plate at 240°C for 1 minute to form a Si-containing resist underlayer film. Then, the resist underlayer film was immersed in propylene glycol monomethyl ether acetate used as a solvent for an over coating resist composition for 1 minute. When the change in the film thickness of the resist underlayer film between before the immersion and after the immersion was 1 nm or less, it was determined as "favorable" and indicated as "○" whereas when the change in the film thickness was more than 1 nm, it was determined as "failed" and indicated as "×". The obtained results are shown in Tables 5 and 6.

[0067] (Measurement of Optical Constants)

The Si-containing resist underlayer film forming compositions prepared in Examples 1 to 38 and Comparative Examples 1 to 6 were each applied onto corresponding silicon wafers using a spinner. The composition was heated on a hot plate at 240°C for 1 minute to form a Si-containing resist underlayer film (film thickness: 0.05 μm). Then, the refractive index (n value) and the optical absorptivity (k value; also called the attenuation coefficient) of the resist underlayer film at a wavelength of 193 nm were measured using a spectro-ellipsometer (VUV-VASE VU-302; manufactured by J.A.Woollam Co., Inc.). The obtained results are shown in Tables 5 and 6.

Table 5

	Refractive index	Optical absorptivity	Solvent resistance	Fluorine-based gas etching rate	Oxygen-based gas resistance
Example 1	1.63	0.13	○	19.9	0.01
Example 2	1.63	0.13	○	20.1	0.01
Example 3	1.63	0.13	○	20.0	0.01
Example 4	1.62	0.15	○	20.3	0.01
Example 5	1.62	0.15	○	20.6	0.01
Example 6	1.64	0.08	○	21.8	0.02
Example 7	1.64	0.08	○	21.7	0.02
Example 8	1.64	0.08	○	21.8	0.02
Example 9	1.64	0.10	○	22.6	0.02
Example 10	1.64	0.10	○	22.9	0.02
Example 11	1.63	0.13	○	20.2	0.01
Example 12	1.63	0.13	○	21.0	0.01
Example 13	1.63	0.13	○	21.2	0.01
Example 14	1.64	0.08	○	21.9	0.02
Example 15	1.64	0.08	○	22.2	0.02
Example 16	1.64	0.08	○	22.5	0.02
Example 17	1.62	0.15	○	20.6	0.01
Example 18	1.62	0.15	○	20.9	0.01
Example 19	1.64	0.10	○	22.9	0.02
Example 20	1.64	0.10	○	23.0	0.02

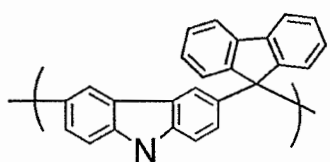
Table 6

	Refractive index	Optical absorptivity	Solvent resistance	Fluorine-based gas etching rate	Oxygen-based gas resistance
Example 21	1.61	0.18	○	21.5	0.01
Example 22	1.62	0.15	○	20.5	0.01
Example 23	1.62	0.15	○	20.7	0.01
Example 24	1.60	0.20	○	22.0	0.01
Example 25	1.61	0.17	○	20.7	0.01
Example 26	1.61	0.17	○	20.9	0.01
Example 27	1.60	0.20	○	22.1	0.01
Example 28	1.61	0.17	○	20.9	0.01
Example 29	1.61	0.17	○	21.0	0.01
Example 30	1.62	0.13	○	23.8	0.02
Example 31	1.63	0.10	○	22.4	0.02
Example 32	1.63	0.10	○	22.6	0.02
Example 33	1.61	0.15	○	24.5	0.02
Example 34	1.62	0.12	○	23.2	0.02
Example 35	1.62	0.12	○	23.4	0.02
Example 36	1.61	0.15	○	24.6	0.02
Example 37	1.62	0.12	○	23.5	0.02
Example 38	1.62	0.12	○	23.5	0.02
Comparative Example 1	1.62	0.13	○	20.7	0.01
Comparative Example 2	1.63	0.13	○	19.7	0.01
Comparative Example 3	1.63	0.13	○	19.8	0.01
Comparative Example 4	1.62	0.13	○	20.4	0.01
Comparative Example 5	1.63	0.13	○	19.7	0.01
Comparative Example 6	1.63	0.13	○	20.1	0.01

[0068] (Production of Organic Resist Underlayer Film)

In nitrogen, into a 100 mL four-neck flask, carbazole (6.69 g, 0.040 mol,
5 manufactured by Tokyo Chemical Industry Co., Ltd.), 9-fluorenone (7.28 g, 0.040 mol,
manufactured by Tokyo Chemical Industry Co., Ltd.), and p-toluenesulfonic acid
monohydrate (0.76 g, 0.0040 mol, manufactured by Tokyo Chemical Industry Co., Ltd.)
were charged and thereto, 1,4-dioxane (6.69 g, manufactured by Kanto Chemical Co.,
Inc.) was added, followed by stirring the resultant reaction mixture. The temperature of
10 the reaction mixture was elevated to 100°C to dissolve the reaction mixture and the

polymerization was initiated. After 24 hours, the reaction mixture was left to be cooled down to 60°C, and then, to the reaction mixture, chloroform (34 g, manufactured by Kanto Chemical Co., Inc.) was added to dilute the reaction mixture, followed by reprecipitating the resultant reaction mixture in methanol (168 g, manufactured by Kanto Chemical Co., Inc.). The obtained precipitate was filtered and was dried using a vacuum drier at 80°C for 24 hours to obtain 9.37 g of the objective polymer (Formula (4-1), hereinafter abbreviated as PCzFL).



Formula (4-1)

[0069] The measurement result of $^1\text{H-NMR}$ of PCzFL was as follows.

$^1\text{H-NMR}$ (400 MHz, DMSO- d_6): δ 7.03-7.55 (br, 12H), δ 7.61-8.10 (br, 4H), δ 11.18 (br, 1H)

PCzFL had a weight average molecular weight M_w of 2,800 that was measured by GPC in terms of polystyrene and a polydispersity M_w/M_n of 1.77.

With 20 g of the obtained resin, 3.0 g of tetramethoxymethyl glycoluril (trade name: POWDER LINK 1174; manufactured by Mitsui Cytec Ltd.) as a crosslinking agent, 0.30 g of pyridinium *p*-toluenesulfonate as a catalyst, and 0.06 g of MEGAFAC R-30 (trade name; manufactured by Dainippon Ink & Chemicals Inc.) as a surfactant were mixed.

The resultant mixture was dissolved in 88 g of propylene glycol monomethyl ether acetate to prepare a solution. The solution was then filtered using a polyethylene microfilter having a pore diameter of 0.10 μm and was further filtered using a polyethylene microfilter having a pore diameter of 0.05 μm to prepare a solution of an organic resist underlayer film forming composition to be used for a lithography process with a multilayer film.

[0070] (Resist Patterning Evaluation)

The organic underlayer film (layer A) forming composition obtained in the above

was applied onto a silicon wafer, and the composition was baked on a hot plate at 240°C for 60 seconds to obtain an organic underlayer film (layer A) having a film thickness of 200 nm. The Si-containing resist underlayer film (layer B) forming compositions obtained in Example 1 to Example 38 and Comparative Example 1 to Comparative Example 6 were each applied onto the corresponding organic underlayer films (layers A). The composition was baked on a hot plate at 240°C for 45 seconds to obtain a Si-containing resist underlayer film (layer B). The Si-containing resist underlayer films (layers B) had a film thickness of 35 nm.

Onto each of the layers B, a commercially available photoresist solution (trade name: AR 2772; manufactured by JSR Corporation) was applied by a spinner, and the solution was baked on a hot plate at 110°C for 60 seconds to form a photoresist film (layer C) having a film thickness of 120 nm. The patterning of the resist was performed using an ArF exposing machine S-307E, manufactured by Nikon Corporation (wavelength: 193 nm, NA, σ : 0.85, 0.93/0.85 (Dipole), immersion liquid: water). The target was a photoresist after the development having a line width and a width between lines of 0.065 μm and a line width and a width between lines of 0.060 μm , which is what is called lines and spaces (dense lines), and the exposure was performed through a mask set to form such a photoresist.

Then, the resultant product was baked on a hot plate at 110°C for 60 seconds, was cooled down, and was developed with a tetramethylammonium hydroxide aqueous solution (developer) having a concentration of 2.38% by mass in a 60-second single paddle process. With respect to the obtained photoresist patterns, a photoresist pattern in which no large pattern peeling, no undercut, or no widening (footing) at a line bottom was caused was evaluated as favorable. The obtained results are shown in Table 7 and

8.

Table 7

	Width and Interval in Pattern 0.065 μm	Width and Interval in Pattern 0.060 μm
Example 1	Favorable	Favorable
Example 2	Favorable	Favorable
Example 3	Favorable	Favorable
Example 4	Favorable	Favorable
Example 5	Favorable	Favorable
Example 6	Favorable	Favorable
Example 7	Favorable	(partial footing)
Example 8	Favorable	Favorable
Example 9	Favorable	(partial footing)
Example 10	Favorable	Favorable
Example 11	Favorable	Favorable
Example 12	Favorable	Favorable
Example 13	Favorable	Favorable
Example 14	Favorable	Favorable
Example 15	Favorable	(partial footing)
Example 16	Favorable	Favorable
Example 17	Favorable	(partial footing)
Example 18	Favorable	Favorable
Example 19	Favorable	Favorable
Example 20	Favorable	Favorable

Table 8

	Width and Interval in Pattern 0.065 μm	Width and Interval in Pattern 0.060 μm
Example 21	Favorable	Favorable
Example 22	Favorable	Favorable
Example 23	Favorable	Favorable
Example 24	Favorable	Favorable
Example 25	Favorable	Favorable
Example 26	Favorable	Favorable
Example 27	Favorable	Favorable
Example 28	Favorable	Favorable
Example 29	Favorable	Favorable
Example 30	Favorable	Favorable
Example 31	Favorable	Favorable
Example 32	Favorable	Favorable
Example 33	Favorable	Favorable
Example 34	Favorable	Favorable
Example 35	Favorable	Favorable
Example 36	Favorable	Favorable
Example 37	Favorable	Favorable
Example 38	Favorable	Favorable
Comparative	Failed	Failed
Example 1	(large peeling)	(large peeling)
Comparative	Favorable	Failed
Example 2	(partial footing)	(footing)
Comparative	Favorable	Failed
Example 3	(partial footing)	(footing)
Comparative	Failed	Failed
Example 4	(large peeling)	(large peeling)
Comparative	Favorable	Failed
Example 5	(partial footing)	(footing)
Comparative	Failed	Failed
Example 6	(large peeling)	(large peeling)

INDUSTRIAL APPLICABILITY

- 5 [0071] The resist underlayer film forming composition can be utilized as a resist underlayer film forming composition for lithography for forming a resist underlayer film capable of being used as a hardmask. The resist underlayer film forming composition causes no intermixing with an over coating resist. The resist underlayer film forming composition has a dry etching rate larger than that of a resist with respect to a
- 10 fluorine-based etching gas, so that a resist pattern can be transferred to the resist

underlayer film of the present application, and exhibits etching resistance against an oxygen-based etching gas, so that a resist pattern can be transferred to an organic underlayer film. A resist underlayer film forming composition for forming a resist underlayer film with which a substrate can be processed by using such a rectangular pattern is provided.

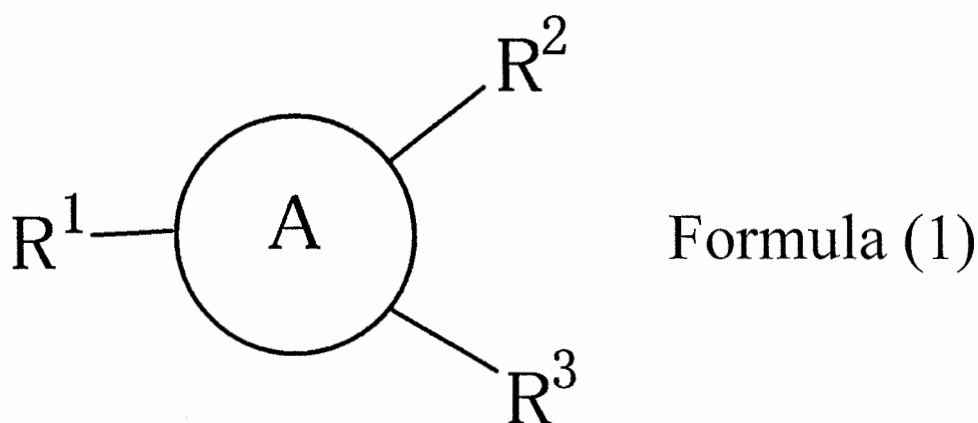
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1. A resist underlayer film forming composition for lithography, comprising:
as a silane:

5 at least one among a hydrolyzable organosilane, a hydrolysis product thereof, and
a hydrolysis-condensation product thereof, wherein

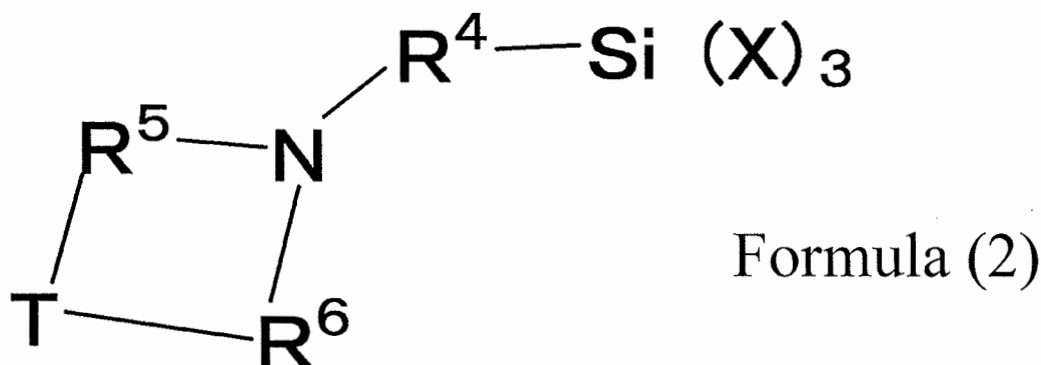
the silane includes a silane having a cyclic organic group containing as atoms
making up the ring, a carbon atom, a nitrogen atom, and a hetero atom other than a
carbon atom and a nitrogen atom.

10 2. The resist underlayer film forming composition according to claim 1, wherein
the hydrolyzable organosilane is a hydrolyzable organosilane of Formula (1):



15 (in Formula (1), at least one group among R^1 , R^2 , and R^3 is a group in which a $-\text{Si}(\text{X})_3$
group is bonded to a C_{1-10} alkylene group, and the other group(s) among R^1 , R^2 , and R^3
is(are) a hydrogen atom, a C_{1-10} alkyl group, or a C_{6-40} aryl group; a ring A is a cyclic
20 organic group of a 5- to 10-membered ring containing as atoms making up the ring, a
carbon atom, at least one nitrogen atom, and further containing a sulfur atom or an
oxygen atom; and X is an alkoxy group, an acyloxy group, or a halogen atom).

3. The resist underlayer film forming composition according to claim 1, wherein the hydrolyzable organosilane is a hydrolyzable organosilane of Formula (2):

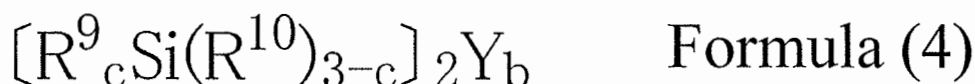


(in Formula (2), each of R^4 , R^5 , and R^6 is a C_{1-6} alkylene group; T is a sulfur atom or an oxygen atom; a cyclic organic group made up with a nitrogen atom, R^5 , R^6 , and T is a 5- to 10-membered ring; and X is an alkoxy group, an acyloxy group, or a halogen atom).

4. The resist underlayer film forming composition according to claim 1, wherein the silane includes any one of or both of: a combination of at least one selected from the group consisting of at least one of organic silicon compound (B) selected from the group consisting of an organic silicon compound of Formula (3):



(in Formula (3), R^7 is a group that is an alkyl group, an aryl group, an aralkyl group, a halogenated alkyl group, a halogenated aryl group, a halogenated aralkyl group, an alkenyl group, or an organic group having an epoxy group, an acryloyl group, a methacryloyl group, a mercapto group, an alkoxyaryl group, an acyloxyaryl group, an isocyanurate group, a hydroxy group, a cyclic amino group, or a cyano group, or a combination of these groups, and is bonded to a silicon atom through a Si-C bond; R^8 is an alkoxy group, an acyloxy group, or a halogen atom; and a is an integer of 0 to 3) and an organic silicon compound of Formula (4):



(in Formula (4), R^9 is an alkyl group; R^{10} is an alkoxy group, an acyloxy group, or a halogen atom; Y is an alkylene group or an arylene group; b is an integer of 0 or 1; and c is an integer of 0 or 1), a hydrolysis product of (B), and a hydrolysis-condensation product of (B) with at least one selected from the group consisting of a hydrolyzable organosilane (A) of Formula (1) or Formula (2), a hydrolysis product of (A), and a hydrolysis-condensation product of (A); and a hydrolysis-condensation product of (A) and (B).

5. The resist underlayer film forming composition according to any one of claims

10 1 to 4, wherein

any one of or both of: a hydrolysis-condensation product of a hydrolyzable organosilane (A) of Formula (1) or Formula (2); and a hydrolysis-condensation product of a hydrolyzable organosilane (A) of Formula (1) or Formula (2) with an organic silicon compound of Formula (3) is contained as a polymer.

15

6. The resist underlayer film forming composition according to any one of claims

1 to 5, further comprising an acid.

7. The resist underlayer film forming composition according to any one of claims

20 1 to 6, further comprising water.

8. A resist underlayer film obtained by applying the resist underlayer film forming composition as claimed in any one of claims 1 to 7 onto a semiconductor substrate, and baking the composition.

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9. A method for producing a semiconductor device, the method comprising:

applying the resist underlayer film forming composition as claimed in any one of claims 1 to 7 onto a semiconductor substrate and baking the composition to form a resist underlayer film;

applying a composition for a resist onto the underlayer film to form a resist film;
exposing the resist film to light;
developing the resist film after the exposure to obtain a patterned resist film;
etching the resist underlayer film according to a pattern of the patterned resist film

5 to obtain a patterned resist underlayer film; and

processing the semiconductor substrate according to a pattern of the patterned resist film and the patterned resist underlayer film.

10. A method for producing a semiconductor device, the method comprising:

10 forming an organic underlayer film on a semiconductor substrate;

applying the resist underlayer film forming composition as claimed in any one of claims 1 to 7 onto the organic underlayer film and baking the composition to form a resist underlayer film;

15 applying a composition for a resist onto the resist underlayer film to form a resist film;

exposing the resist film to light;

developing the resist film after the exposure to obtain a patterned resist film;

etching the resist underlayer film according to a pattern of the patterned resist film to obtain a patterned resist underlayer film;

20 etching the organic underlayer film according to a pattern of the patterned resist film and the patterned resist underlayer film to obtain a patterned organic underlayer film; and

processing the semiconductor substrate according to a pattern of the patterned resist film, the patterned resist underlayer film, and the patterned organic underlayer film.

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