DESCRIPTION

Analyte Assay Using Particulate Labels

Related Application

This application claims priority to Yguerabide et al., United States Provisional Application No. 60/016,383, entitled "Analyte Assay Using Particulate Labels", filed April 25th, 1996, which is hereby incorporated by reference in its entirety, including drawings.

Background of the Invention

The following is an outline of relevant existing detection methods. It is also a summary of relevant science to aid the reader in understanding the details of the claimed invention. It should not be taken as an admission that any of the cited art is prior art to the claims. The cited art is hereby incorporated herein by reference so that the general procedures and methods in that art that are of use to practice of the present invention need not be rewritten herein. In particular, applicant incorporates those sections related to general methods of "binding-pair" methodology, and methods for measurement of light scattering herein.

Sensitive Analyte Assays

Binding-pair (also known as ligand-receptor, molecular recognition binding and the like) techniques play an important role in many applications of biomedical analysis and are gaining importance in the fields of environmental science, veterinary medicine, pharmaceutical research, food and water quality control and the like. For the detection of analytes at low concentrations (less than about 1 picomole analyte/sample volume analyzed) the use of fluorescent, luminescent, chemiluminescent, or electrochemiluminescent labels and detection methods are often used.

For the detection of low concentrations of analytes in the field of diagnostics, the methods of chemiluminescence and electrochemiluminescence are gaining wide-spread use. These methods of chemiluminescence and electro-chemiluminescence provides a means to detect low concentrations of analytes by amplifying the number of luminescent molecules or photon generating events many-fold, the resulting "signal amplification" then allowing for detection of low concentration analytes.

In addition, the method of Polymerase Chain Reaction (PCR) and other related techniques have gained wide use for amplifying the number of nucleic acid analytes in the sample. By the addition of appropriate enzymes, reagents, and temperature cycling methods, the number of nucleic acid analyte molecules are amplified such that the analyte can be detected by most known detection means. The high level of commercial activity in the development of new signal generation and detection systems, and the development of new types of test kits and instruments utilizing signal and analyte molecule amplification attests to the importance and need for sensitive detection methods.

However, the above mentioned methods of signal and analyte molecule amplification have associated limitations which makes the detection of analytes by these methods complicated, not easy to use, time consuming, and costly. Problems of interference of chemical or enzymatic reactions, contamination, complicated and multi-step procedures, limited adaptability to single step "homogeneous" (non-separation) formats, and the requirement of costly and sophisticated instrumentation are areas that those in the art are constantly trying to improve.

Thus, there is a tremendous need for easy to use, quantitative, multi-analyte, and inexpensive procedures and instruments for the detection of analytes. Such procedures, test kits, and instruments would overcome the

disadvantages and limitations of the current methods of signal and analyte molecule amplification, and would be useful in research, individual point of care situations (doctor's office, emergency room, out in the field, etc.), and in high throughput testing applications.

It is the object of the present invention to provide a new means to more easily detect one or more analytes in a sample to low concentrations than was previously possible. The present invention can detect low concentrations of analytes without the need for signal or analyte molecule amplification.

The present invention provides a signal and detection system for the detection of analytes where the procedures can be simplified and the amount and types of steps and reagents reduced. The present invention provides for the quantitative detection of single or multiple analytes in a sample. The present invention also provides for substantial reductions in the number of different tests and amounts of sample material that are analyzed. Such reduction in the number of individual tests leads to reduced cost and waste production, especially medically-related waste that must be disposed of.

Light Scattering Detection Methods and Properties of Light Scattering Particles

There is a large body of information concerning the phenomenon of light scattering by particles, the use of particulate labels in diagnostic assays, and the use of light scattering methods in diagnostic assays which are now presented in the following discussion of relevant art none of which is admitted to be prior art to the pending claims. This art is provided as a background for understanding of the novelty and utility of the claimed invention.

The general study of light scattering comprises a very large field. The phenomena of light scattering has been studied intensely for about the last one hundred or

so years and the applications of the knowledge of light scattering to different aspects of human endeavor are wide and varied.

The classical theory of light scattering by small, homogeneous, non light absorbing, spherical particles of a size of about 1/20 or less the wavelength of the incident radiation was initially developed by Rayleigh. Later a more general phenomenological theory of light scattering by homogeneous, spherical particles of any size and composition was developed by Mie. The Mie theory applies both to light absorbing and nonabsorbing particles. It has also been shown from Mie theory that the expressions of Rayleigh can easily be generalized so as to apply to particles which absorb light as long as the particles are much smaller than the wavelength of incident light. For these small diameter particles, Mie theory and the generalized Rayleigh theory give similar results. Light scattering (elastic) can be viewed from a classical or quantum mechanical point of view. An excellent quantitative description can be obtained through the classical point of view.

A historical background as well as a description of the basic theories of scattered light and other electromagnetic radiation is provided in the following references; Absorption and Scattering of Light By Small Particles (1983), C.F. Bohren, D.R. Huffman, John Wiley and Sons; The Scattering of Light and Other Electromagnetic Radiation (1969), M. Kerker, Academic Press.

Further background information of the phenomenon of light scattering can be found in the following publications.

Zsigmondy, Colloids and the Ultramicroscope - A Manual of Colloid Chemistry and Ultramicroscopy,1914, John Wiley & Sons, Inc. is described various light scattering properties of gold particles and other types of particles.

Hunter, Foundation of Colloid Science, Vol, I, 105, 1991, describes use of optical microscopes, ultramicroscopes, and electron microscopes in observation of particles.

Shaw et al., Introduction to Colloid and Surface Chemistry, 2nd ed., 41, 1970, describe optical properties of colloids and the use of electron microscopy, and dark field microscopy e.g., the ultramicroscope.

Stolz, SpringerTracts, Vol. 130, describes time resolve light scattering methodologies.

Klein and Metz, 5 Photographic Science and Engineering 5-11, 1961, describes the color of colloidal silver particles in gelatin.

Eversole and Broida, 15 Physical Review 1644-1654, 1977, describes the size and shape effects on light scattering from various metal particles such as silver, gold, and copper.

Kreibig and Zacharias, 231 Z. Physik 128-143, 1970, describe surface plasma resonances in small spherical silver and gold particles.

Bloemer et al., 37 Physical Review 8015-8021, 1988, describes the optical properties of submicrometer-sized silver needles and the use of such needles is described in Bloemer, U.S. Patent 5,151,956,where a surface plasmon resonance of small particles of metal to polarize light propagating in a wave guide is described.

Wiegel., 136 Zeitschrift fur Physik, Bd., 642-653, 1954, describes the color of colloidal silver and the use of electron microscopy.

Use of Particles, Light Scattering and Other Methods for Detection of Analytes

For about the last thirty-five years, metal particles including gold and silver have been used as both contrast enhancement agents or light absorption labels in many different types of analytic and/or diagnostic applications. The great majority of these applications fall under

the category of cytoimmunochemistry studies which have used gold or silver enhanced gold particles as markers to study structural aspects of cellular, subcellular, or tissue organization. In these studies, metal particles are usually detected and localized by electron microscopy, including scanning, transmission, and BEI (backscattered electron imaging). These methods take advantage of the electron dense nature of metals or the high atomic number of metals to facilitate the detection of the gold particles by virtue of the large numbers of secondary and backscattered electrons generated by the dense metal (see; Hayat, Immunogold-silver staining reference Page 1 and Chapters 1, 6, 15; and Hayat, Colloid Gold reference Chapters 1, 5, 7 and others).

There have been a few reports of the use of gold and silver enhanced gold particles in light microscopic studies. For example, in 1978 gold particles were used as an immunogold stain with detection by light microscopy. A review of the use of gold particles in light microscopy (See, Hayat, Immunogold-Silver Staining Reference Page 3) published in 1995 discusses this 1978 work and presents the following analysis:

> "Geoghehan et al. (1978) were the first to use the red or pink color of colloidal gold sols for light microscopical immunogold staining using paraffin sections. In semithin resin sections red color of light scattered from gold particles as small as 14 nm was seen in cell organelles containing high concentrations of labeled antigens in the light microscope (Lucocq and Roth, 1984). Since the sensitivity of immunogold staining in light microscopy is inferior in comparison with other immunocytochemical techniques, the former did not gain general acceptance; the pinkish color of the gold deposit is difficult to visualize."

This paragraph is an indication of the state of understanding of the light scattering properties of gold and other metal particles for diagnostic and analytic studies. The paragraph specifically states "In semithin resin sections red color of light scatter from gold particles as small as 14 nm was seen in organelles containing high concentrations of labeled antigens in the light microscope."

However, with white light illumination, the scattered light from 14 nm gold particles is predominantly green. Since the particles appear red in the light microscope this indicates that some interactions other than pure light scattering are being detected. It is probable that the red color observed in the light microscope is predominantly transmitted light and not scattered light. When the gold particles accumulate sufficiently at the target site in cells, tissue sections or some other surface the red color due to transmitted light will be seen (see also; J. Roth (1983) Immunocytochemistry 2 p217; and Dewaele et al (1983) in Techniques in Immunochemistry Vol 2 p1, Eds. Bullock and Petrusz, Academic Press).

As mentioned in the above quote, it appears that the sensitivity of immunogold staining in light microscopy was believed to be inferior to that of other methods, and the use of gold particles as markers for light microscope detection did not gain general acceptance. In the 1995 review book in Chapter 12, p198 by Gao and Gao is the following quote on the same subject.

> "Colloidal gold was initially used only as a marker for electron microscopy (EM), because of its electron dense nature and secondary electron emission feature (Horisberger, 1979). Direct visualization of colloidal gold in light microscopy (LM) was limited. The size of colloidal gold is too small to be detected at the light microscope level, although using highly concentrated immunogold cells may be

stained red by this reagent (Geoghegan et al., 1978; Roth, 1982; Holgate et al., 1983."

As mentioned in both of the above, the sensitivity of detection of colloidal gold with light microscopy was believed to be low. The method of silver enhancement of gold particles was developed to overcome this perceived drawback. The following is another quote from the 1995 review book.

> "The real breakthrough for immunogold staining for light microscopy came with the introduction of silver-enhancement of colloidal gold particles (20nm) bound to immunoglobin in paraffin sections 5 microns (Holgate et al., 1983). This approach significantly enhanced the sensitivity, efficiency, and accuracy of antigen detectability in the light microscope. Using IGSS, gold particles as small as 1nm in diameter can be visualized in the light microscope. Thin section subjected to IGSS can also be viewed with the light microscope, especially by using phase contrast or epi-polarization illumination (Stierhof et al., 1992)."

The method of silver enhancement of gold particles is widely used. The enhancement method transforms the marker gold particle into a larger metal particle or an even larger structure which is microns or greater in dimensions. These structures are composed primarily of silver, and such enlarged particles can be more readily detected visually in the bright field optical microscope. Individual enlarged particles have been visualized by high resolution laser confocal and epipolarization light microscopy. Id at 26 and 203.

However, even with the use of silver enhancement techniques, those in the art indicate that this will not achieve the sensitivity and specificity of other methods. For example, in the publication of Vener, T. I. et. al., Analytical Biochemistry 198, p308-311 (1991) the authors

discuss a new method of sensitive analyte detection called Latex Hybridization Assay (LHA). In the method they use large polymer particles of 1.8 microns in diameter that are filled with many highly fluorescent dye molecules as the analyte tracer, detecting the bound analytes by the fluorescent signal. The following excerpt is from this publication:

> "To assess the merits of LHA we have compared our technique with two other indirect nonradioactive techniques described in the literature. The most appropriate technique for comparison is the streptavidin colloid gold method with silver enhancement of a hybridization signal, since this is a competing corpuscular technique. However, this method is not very sensitive even with the additional step of silver enhancement: 8pg of λ-phage DNA is detected by this method as compared to 0.6 pg or 2 X 10^4 molecules of λ DNA detected by LHA on the nylon membrane."

Stimpson et al., 92 Proc. Natl. Acad. Sci. USA 6379-6383, July 1995, a real time detection method for detection of DNA hybridization is described. The authors describe use of a particulate label on a target DNA which acts as a

> "light-scattering source when illuminated by the evanescent wave of the wave guide and only the label bound to the surface generates a signal. ... The evanescent wave created by the wave guide is used to scatter light from a particulate label adsorbed at multiple DNA capture zones placed on the wave guide surface. Since an evanescent wave only extends a few hundred nanometers from the wave guide surface, the unbound/dissociated label does not scatter light and a wash step is not required. The signal intensity is sufficient to allow

> measurement of the surface binding and desorption of the light-scattering label can be studied in real time; i.e., detection is not rate limiting. The hybridization pattern on the chip can be evaluated visually or acquired for quantitative analysis by using a standard CCD camera with an 8-bit video frame grabber in 1/30 of a second."

Experiments were performed with 70 nanometer diameter gold particles and 200 nanometer diameter selenium particles. More intense signals were observed with the selenium particles. The authors indicate

> "A wave guide signal sufficient for single-base discrimination has been generated between 4 and 40 nm DNA and is, therefore, comparable to a fluorescence signal system."

This method uses waveguides and evanescent type illumination. In addition, the method is about as sensitive as current fluorescence-based detection systems. Particles of 70nm diameter and larger are said to be preferred.

Schutt et al., U.S. Patent 5,017,009, describes an immunoassay system for detection of ligands or ligand binding partners in a heterogenous format. The system is based upon detection of

> "back scattered light from an evanescent wave disturbed by the presence of a colloidal gold label brought to the interface by an immunological reaction. ... Placement of the detector at a back angle above the critical angle insures a superior signal-to-noise ratio.

The authors explain that the immunoassay system described utilizes scattered total internal reflectance, i.e., propagation of evanescent waves. They indicate that the presence of colloidal gold disrupts propagation of the evanescent wave resulting in scattered light which may be detected by a photomultiplier or other light sensor to

provide a responsive signal. They indicate that an important aspect of their invention is the physical location of the detector.

> "The detector is ideally placed at an angle greater than the critical angle and in a location whereby only light scattered backward toward the light source is detected. This location thereby ideally avoids the detection of superior scattered light within the bulk liquid medium."

Total internal reflection of the incident beam is used to create the evanescent wave mode of illumination and the detection is performed on an optically-transmissive surface. The use of specialized apparatus is preferred.

Leuvering, U.S. Patent 4,313,734, describes a method for detection of specific binding proteins by use of a labeled component obtained by coupling particles "of an aqueous dispersion of a metal, metal compound, or polymer nuclei coated with a metal or metal compound having a diameter of at least 5 nm." The process is said to be especially suited for estimation of immunochemical components such as haptens, antigens and antibodies. The metal particles are said to have already been used as contrast-enhancing labels in electron microscopy but their use in immunoassays had apparently

> "not previously been reported and has surprisingly proved to be possible.
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> The metal sol particle, immunochemical technique, according to the instant invention which has been developed can be not only more sensitive than the known radio- and enzyme-immunotechniques, but renders it furthermore possible to demonstrate and to determine more than one immunological component in the same test medium simultaneously by utilizing sol

particles of different chemical compositions as labels."

Examples of metals include platinum, gold, silver, and copper or their salts.

"The measurement of the physical properties and/or the concentration of the metal and/or the formed metal containing agglomerate in a certain phase of the reaction mixture may take place using numerous techniques, which are in themselves known. As examples of these techniques there may be cited the colorimetric determination, in which use is made of the intense colour of some dispersions which furthermore change colour with physicochemical changes; the visual method, which is often already applicable to qualitative determinations in view of the above-noted fact that metal sols are coloured; the use of flame emission spectrophotometry or another plasma-emission spectrophotometric method which renders simultaneous determination possible, and the highly sensitive method of flame-less atomic absorption spectrophotometry."

Two or more analytes in a sample are preferably detected by using flame emission spectrophotometry or another plasma-emission spectrophotometric method. The preferred method of detection for greatest sensitivity is by flame-less atomic absorption spectrophotometry.

Swope et al., U.S. Patent 5,350,697 describes apparatus to measure scattered light by having the light source located to direct light at less than the critical angle toward the sample. The detector is located to detect scattered light outside the envelope of the critical angle.

Craig et al., U.S. Patent 4,480,042 describes use of high refractive index particle reagents in light scattering immunoassays. The preferred particles are

composed of polymer materials. The concentration of compounds of biological interest was determined by measuring the change in turbidity caused by particle agglutination or inhibition of agglutination. The preferred particles are of a diameter less than approximately 0.1 μ and greater than 0.03 μ. "Shorter wavelengths, such as 340 nm, give larger signal differences than longer wavelengths, such as 400nm."

Cohen et al., U.S. Patent 4,851,329 and Hansen, U.S. Patent 5,286,452, describe methods for detection of agglutinated particles by optical pulse particle size analysis or by use of an optical flow particle analyzer. These systems are said to be useful for determination of antigen or antibody concentrations. These methods use sophisticated apparatus and specialized signal processing means. Preferred particle diameters are of about 0.1 to 1 micron in diameter for the method of Cohen and about 0.5 to about 7.0 microns in diameter for the method of Hansen.

Okano et al., 202 Analytical Biochemistry 120, 1992, describes a heterogenous sandwich immunoassay utilizing microparticles which can be counted with an inverted optical microscope. The microparticles were of approximately 0.76 microns in diameter, and were carboxylated microparticles made from acrylate.

Other particle detection methods are described by Block, U.S. Patent 3,975,084, Kuroda, U.S. Patent 5,274,431, Ford, Jr., U.S. Patent 5,305,073, Furuya, U.S. Patent 5,257,087, and by Taniguchi et al., U.S. Patent 5,311,275.

Geoghegan et al., 7 Immunological Communications 1-12, 1978, describes use of colloidal gold to label rabbit anti-goat IgG for indirect detection of other antibodies. A light and electron microscope were used to detect labeled particles. The gold particles had an average size of 18-20 nanometers and bright field light microscopy was used. For electron microscopy, Araldite silver-gold thin sections were used. "Similar percentages of surface

labeled cells were noted by immunofluorescence and the colloidal gold bright field method." 1-5 particles per cell could be detected by electron microscopy but the authors state that:

> "Such small quantities of label were not detected by fluorescence or by brightfield microscopy and may represent either non-specific and Fc receptor bound GAD and GAM, where a low level of surface immunoglobulin (S.Ig) on the GAD and GAM treated cells."

Hari et al., U.S. Patent 5,079,172, describes use of gold particles in antibody reactions and detection of those particles using an electron microscope. 15 nanometer gold particles were exemplified. In the preferred method, electron microscopy is used.

DeMey et al., U.S. Patent 4,420,558, describes the use of a bright field light microscopic method for enumerating cells labeled with gold-labeled antibodies. The method uses a light microscope in the bright field arrangement and magnifications of 500 or greater with immersion oil lenses are used to count gold-labeled peroxidase negative cells. The visualization of the labeled-surfaces is based on the aggregate properties of the gold particles, which, under the indicated circumstances, undergo extensive patching, these patches on the cell surface being resolvable with the method described. 40 nanometer gold was found to give optimal results.

De Mey et al., U.S. Patent 4,446,238, describes a similar bright field light microscopic immunocytochemical method for localization of colloidal gold labeled immunoglobulins as a red colored marker in histological sections. The method of Immuno Gold Staining (IGS) as described by the authors

> "In both procedures the end-product is an accumulation of large numbers of gold granules

over antigen-containing areas, thus yielding the typical reddish colour of colloidal gold sols."

DeBrabander et al., U.S. Patent 4,752,567 describes a method for detecting individual metal particles of a diameter smaller than 200nm by use of bright field or epi-polarization microscopy and contrast enhancement with a video camera is described. The inventors state:

"Typically, in the above mentioned procedures, the employed metal particles have a diameter of from about 10 to about 100nm. This is well below the resolution limit of bright field microscopy, which is generally accepted to lie around 200 nm. It is therefore quite logical that all previously known visual light microscopic methods are limited in their applications to the detection of immobilized aggregates of metal particles. Individual particles could be observed with ultramicroscopic techniques only, in particular with electron microscopy.

It has now quite surprisingly been found that individual metal particles of a diameter smaller than 200 nm can be made clearly visible by means of bright field light microscopy or epi-polarization microscopy in the visible spectrum, provided that the resulting image is subjected to electronic contrast enhancement."

In subsequent sections the authors state:

"Compared with existing diagnostic methods based on sol particle immuno assays, the present method has a much greater sensitivity. Indeed, existing methods are in general based on light absorption or scattering by the bulk of absorbed or suspended metal particles. Obviously, the observation of colour, e.g. on a blotting medium, requires the presence of massive numbers of particles. In contrast therewith, the present method makes it possible to observe and

count single particles. Hence, the present method will largely facilitate the development of diagnostic blots for applications where existing, e.g. visual or colorimetric, techniques are too less sensitive, e.g. for the detection of Hepatitis."

Schafer et al., 352 Nature 444-448, 1991, describes use of nanometer size particles of gold which could be observed using video enhanced differential interference contrast microscopy. A 40 nanometer diameter gold particle was used.

DeBrabander et al., 6 Cell Motility and the Cytoskeleton 105-113, 1986, (and U.S. Patent 4,752,567) describe use of submicroscopic gold particles and bright field video contrast enhancement. Specifically, the cells were observed by bright field video enhanced contrast microscopy with gold particles of 5-40 nanometers diameters. They also state that

"individual gold particles, having a size smaller than plus or minus 100 nanometers, adsorbed under glass or cells or microinjected in cells are not visible in the light microscope. They are, however, easily visualized when using the capacity of a video camera to enhance contrast electronically."

The authors describe use of epi-illumination with polarized light and collection of reflected light or by use of a "easier and apparently more sensitive way" with a transmitted bright field illumination using monochromatic light and a simple camera. The authors indicate that the gold particles can be easily detected with phase contrast microscopy.

"Unlike that which is possible with larger gold (usually 20-40 nm), even dense accumulations of 5-nm gold, e.g., on structures such as microtubules, are not visible in the light microscope. They do not produce a detectable

red colour. Recently, this has been corrected by a physical development with silver salts, which increases the size of the particles to produce an easily visible black stain.

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We have described a method for localizing ligands almost at the molecular level. The method is new because it enables one for the first time to do this in the light microscope with discrete individual markers that are unambiguously discernible from background structures. Because it is applicable even in living cells, one can thus follow the dynamic behaviour of individual proteins. The method is because it combines two well developed techniques: gold labelling and video microscopy. Most of the applications can be done with inexpensive video equipment the price of which is less than most good 100X oil objectives. Still, many more possibilities arise when combining this with modern digital image manipulations. Some additional advantages are worth noting. Because the label consists of individual discrete markers, both manual and automatic (computer assisted) counting is easy and reliable. The small size of the marker minimizes problems of penetration and diffusion. The possibility of changing the charge of the marker almost at will is helpful in diminishing nonspecific binding in any particular application."

This method was termed by the authors "nanoparticle video ultramicroscopy or short nanovid ultramicroscopy." A similar technology is described in "Geerts et al., 351 Nature, 765-766, 1991.

The preceding discussions of the state of the art of light scattering methods, and the use of light scattering

particles and methods in the field of diagnostics clearly shows the limits of current methods of analyte detection and the novelty and great utility of the present invention. It is the purpose of this invention not only to overcome the present day limitations and disadvantages of light scattering-based diagnostic assays, but to also overcome the limitations and disadvantages of other non-light scattering methods such as signal and analyte molecule amplification. This invention as described herein is easier to use, has greater detection sensitivity, and is capable of measuring analytes across wider analyte concentration ranges than was previously possible. The present invention is broadly applicable to most sample types and assay formats as a signal generation and detection system for analyte detection.

Summary of the Invention

The present invention features a new method for the detection and measurement of one or more analytes in a sample. The method is based on the use of certain particles of specific composition, size, and shape and the detection and/or measurement of one or more of the particle's light scattering properties. The detection and/or measurement of the light-scattering properties of the particle is correlated to the presence, and/or amount, or absence of one or more analytes in a sample. The present invention is versatile and has utility in one form or another to detect and measure one or more analytes in a sample.

The invention features a method for detection of one or more analytes in a sample by binding those analytes to at least one detectable light scattering particle with a size smaller than the wavelength of the illumination light. This particle is illuminated with a light beam under conditions where the light scattered from the beam by the particle can be detected by the human eye with less than 500 times magnification. The light that is scattered

from the particle is then detected under those conditions as a measure of the presence of those one or more analytes. Applicant has surprisingly determined, by simply ensuring appropriate illumination and ensuring maximal detection of specific scattered light, that an extremely sensitive method of detection can result. The method of light illumination and detection is named "DLASLPD" (direct light angled for scattered light only from particle detected) by applicant.

The method and associated apparatus are designed to maximize detection of only scattered light from the particles and thus is many times more sensitive than use of fluorophores, or the use of such particles in methods described above. Such particles can be detected by using a low magnification microscope (magnifying at 2 to 500 times, e.g. 10 to 100 times) without the need for any electronic amplification of the signal. In addition, methods are provided in which no microscope or imaging system is necessary but rather one or more of the light scattering properties are detected of a liquid or solid-phase sample through which light is scattered. These scattered light properties can be used to determine the presence, absence or amount of analyte present in any particular sample.

The source of light in general need not be treated in any particular manner (e.g., polarized, or laser or high intensity) but need only be directed such that it allows scattered light from the particles to be detected. Spatial filtering can be used to ensure reduction of non-specific light scatter. Such filtering can be supplemented by other instrumental components and sample chambers which reduce stray light.

The direct light can be polychromatic or monochromatic, steady-state or pulsed, and coherent or not coherent light. It need not be polarized and can be generated from a low power light source such as a Light Emitting Diode (LED), or 12 watt filament light bulb and

the like. The light is not evanescent light, as described by Stimpson, supra. The light is directed at a sample which may contain the particles at an angle such that the direct light itself will not be observed by the detector unless it is scattered by the particles. The method and apparatus differs from that of Swope, supra in that such scatter can be observed by eye within the critical angle, preferably within the angle of illumination. It can, however, also be detected at greater than the critical angle and outside of the intensity envelope of the forward direction of scattered light. When used with an imaging apparatus, e.g., a microscope, the present method preferentially uses a detector perpendicular to the plane of the sample.

Unlike the diagnostic art that has been described in the "Background of the Invention", Applicant has found that specific types of particles can be detected and measured to very low concentrations, to a high degree of specificity, and across wide concentration ranges with easier to use and less costly methods and apparatus. The invention provides for detection of analytes with greater ease of use, sensitivity, sensitivity, specificity, and is less costly than known methods of analyte detection.

Applicant has determined by methods of theoretical modeling and physical experimentation, that coated metal-like particles have similar light scattering properties as compared to uncoated metal-like particles, both of which have superior light scattering properties as compared to non-metal-like particles. By "metal-like" particles is meant any particle or particle-like substance that is composed of metal, metal compounds, metal oxides, semiconductor(SC), superconductor, or a particle that is composed of a mixed composition containing at least 0.1% by weight of metal, metal compound, metal oxide, semiconductor, or superconductor material. By "coated" particle is meant a particle has on it's surface a layer of additional material. The layer is there to chemically

stabilize the particle in different sample environments, and/or to bind specific analytes by molecular recognition means. Such coatings are for example, inorganic and/organic compounds, polymers, proteins, peptides, hormones, antibodies, nucleic acids, receptors, and the like. By "non-metal-like" particles is meant particles that are not composed of metal, metal compounds, superconductor, metal oxides, semiconductor, or mixed compositions that are_not composed of at least 0.1% by weight of metal, metal compound, metal oxide, superconductor, or semiconductor material.

Applicant has also determined the following: (1) one or more analytes in a sample can be detected and measured by detection and/or measurement of one or more of the specific light scattering properties of metal-like particles. These light scattering properties include the intensity, wavelength, color, polarization, angular dependence, and the RIFSLIW (rotational individual fluctuations in the scattered light intensity and/or wavelengths) of the scattered light. One or more of these properties of particle scattered light can be used to provide information regarding the analytes in the sample; (2) by varying the size, and/or shape and/or composition of a metal-like particle in various combinations, one or more of the light scattering properties can be adjusted to generate more easily detectable and measurable light scattering signals; (3) illumination and detection of the metal-like particles of certain size, shape, and composition with DLASLPD provides a highly sensitive and easy to use method to detect and measure metal-like particles by their light scattering properties. The method provides for single particle detection with easy to use and inexpensive apparatus means; (4) the DLASLPD methods can be used with particle counting and/or integrated light intensity measurements to provide for detection and measurement of the particles across wide concentration ranges; (5) the use of refractive index

enhancement methods provides for enhancement of a particle's light scattering properties, and/or decreases in non-specific light background; (6) the use of DLASLPD video contrast enhancement methods can provide for more sensitive detection in many different types of samples and diagnostic assay formats; (7) for sensitive detection of analytes in a small solid-phase area such as commonly used in microarray and array chip formats, certain types of metal-like particles are more preferred to use than others. Metal-like particles in microarray and array chip formats can be most easily and inexpensively detected by using DLASLPD methods. Such particles in these formats can also be detected by methods of laser scanning confocal microscopy, brightfield or epi-polarization microscopy, and reflection contrast and differential interference contrast microscopy. However these methods and apparatus are not as easy to use and inexpensive as detection by DLASLPD methods and apparatus; and (8) useful apparatus and particle types for specific test kits can be constructed. These different test kits, and associated apparatus are useful for applications to consumer use, portable field use, point of care applications such as doctor's offices, clinics, emergency rooms and the like, research laboratories, and centralized high throughput testing. The above aspects of the present invention provide for detection of one or more analytes in many different types of samples and diagnostic assay formats.

As will be discussed in more detail below, there are many variations of the type of particle, and of the light source and light detection mechanisms. In addition, many variations can be made on the types of particles used.

In preferred embodiments, the particle has a size, composition and shape suitable for producing specific light scattering signal(s) of specific wavelength(s), color(s), polarization, angular dependence, and RIFSLIW of the scattered light that is detectable by eye or photodetector means; the detection includes the methods of

counting the particles, and/or measurement of the intensity of scattered light as a measurement of the concentration of the particles; the particle is formed from metal-like materials or is formed from a mixed composition including non-metal-like materials, the particles are spherical, oval, or asymmetrical (by asymmetrical is meant not roughly spherical in shape); the particles are coated with binding agents, polymers, reactive chemical groups, base material molecules, inorganic and organic compounds; the change in scattered light properties when two or more particles are brought into close contact with each other in assay formats is used; particle reagents comprised of metal-like material and coated with base material molecules adapted to bind to a binding agent are used; assay formats wherein two or more particles are brought sufficiently close together so that the light scattering property of the two or more particles can be resolved from single particles is used; assay formats wherein two or more particles that are held in close proximity to one another are caused to be separated so that the light scattering property of any one particle is altered is used; assay formats wherein two or more particles are linked together by one or more molecular interactions such that when the molecular interaction holding the particles together is disrupted, one or more particles are released from the molecular interaction is used; assay formats wherein the amplified detection of analytes is accomplished by cross-linking two or more particles together using chemical or biological cross-linking agents is used; the particles are composed of additional materials to allow them to be oriented in an electric, magnetic or related electromagnetic field (EMF); the particles are attached to other particles of magnetic or ferro-electric properties; and the illuminating light beam has a wavelength selected to reduce background as compared to other wavelengths.

In other embodiments, the illumination light is steady-state or pulsed; the illumination light is coherent or not coherent; the illumination light is polarized or unpolarized; two or more different wavelengths either from the same light source or from two or more different light sources are used to illuminate the sample, and the scattered light signals are detected.

In other embodiments, the method involves using a plurality of different particles each having one or more scattered light properties which can be detected by eye or photodetector means; and/or a plurality of different wavelengths of light are used in the illumination or detection step; refractive index enhancement methods are used to reduce non-specific light background; the detector is placed at angles outside of the envelope of the forward direction of sample and light beam scattered light; spatial filtering methods are used, optical filters such as cutoff and/or narrow band pass filters are used in the detecting step to reduce non-specific background light.

In yet other embodiments, the particle is increased in size by autometallography prior to detection; the illuminating light beam lacks infra-red radiation; the analyte is present in serum; the particles are released into solution prior to detection; the particles are concentrated into a small volume or solid-phase area prior to detection; the particles are detected by time-dependent binding to a surface or flowing the particles pass a detector or set of detectors; multiple analytes are detected on a solid-phase in a microarray; the microarray is covered with liquid or is dry; single or multiple analytes are detected on a cell surface, cell lysate, or chromosome preparation; the illuminating light beam is polychromatic white light or monochromatic light; the analyte is present in solution or solid-phase, or is present on a microscope slide, or a microtitre plate or another plastic container; the particle is a gold or silver particle, having a size between 1 and 500 nm,

preferably between 10 and 200 nm; the detecting step does not include amplification of the light scattered by electronic means; and the illuminating light beam is directed toward the particle by a prism or other light guiding systems.

In addition, the detection may include observing the particle through at least a x10 objective; the method of DLASLPD video contrast enhancement is used; fiber optic illumination and detection is used; bright-field, laser confocal scanning, reflection contrast or differential interference contrast microscopy detection methods are used; the detection and purification of combinatorial synthesized molecules is performed; particles and/or specialized coatings are used as solid-phase synthetic supports for combinatorial or other synthesized molecules; specially designed sample chambers are used; antireflective coatings on optical components and sample chambers are used; apparatus for field use, doctor's office, clinics and hospital care units are used; and specific particle types are provided in appropriate test kits.

The high sensitivity and ease of use of the signal generation and detection system of the present invention means that one skilled in the art can by inexpensive means, detect and measure one or more analytes in a sample to extremely low concentrations without need of signal (label) or target analyte molecule amplification methods.

The wide range of specific light scattering signals from different particle types in the present invention means that one skilled in the art can detect and measure to a high degree of specificity one or more analytes in a sample.

The high optical resolvability of two or more different particles types in the present invention means that very simple multi-analyte detection (i.e. two or more different analytes) in a sample is possible without the need for complex apparatus.

Those in the art will recognize that applicant has discovered novel methods and apparatus with broad utility. The present invention can be applied in one form or another to most situations where it is desirable to use a signal generation and detection system as part of an assay system to quantitate and/or detect the presence or absence of an analyte. Such analytes include industrial and pharmaceutical compounds of all types, proteins, peptides, hormones, nucleic acids, lipids, and carbohydrates, as well as biological cells and organisms of all kinds. One or another mode of practice of this invention can be adapted to most assay formats which are commonly used in diagnostic assays of all kinds. For example, these include heterogeneous and homogeneous assay formats which are of the sandwich type, aggregation type, indirect or direct and the like. Sample types can be liquid-phase, solid-phase, or mixed phase.

Other features and advantages of the invention will be apparent from the following description of the preferred embodiments, and from the claims.

Description of the Preferred Embodiments

The drawings will first briefly be described.

Drawings

Fig. 1 illustrates illumination of a sample from below. L is a lens; D is the diameter of the lens; O is the area on surface S being detected; C is the cone showing the angles at which L collects light; LB is the illuminating light beam.

Fig. 2 illustrates the collecting angle of a lens. D is the diameter of the lens; f is the focal length; O is the area detected; θ_H is the planar half angle of the collection cone.

Fig. 3 is a diagram which defines angles used to describe reflection and refraction at a surface. S is the surface; ni and nt are the refractive indices of incident

medium and surface medium respectively; RFRB and RFLB are the refracted and reflected light beams respectively; IB is the incident light beam; θi, θr, and θt are the angles of incidence, reflection, and refraction of the light beam.

Figs. 4A, 4B, and 4C are light reflection graphs for $n_i < n_t$ taken from various texts.

Figs. 5A, and 5B are light reflection graphs for $n_i > n_t$ taken from various texts.

Fig. 6 illustrates the refraction and reflection involved in the illumination of particles on a dry surface in air.

Fig. 7 is a graph of plot of $\theta i2$ vs. θil for $n_2=1.5$ + $n_3=1$ (see Fig. 6).

Fig. 8 illustrates an angular distribution of light scattered by surface artifacts and particles; the dashed lines represent light scattered by the particle; the solid line with an arrow is incident white light beam, and one beam of scattered light by surface artifacts, and the circle is an intensity envelope for the light scattered in the forward direction.

Fig. 9 shows a sample in a thin film of water that is on a microscope slide and covered with a cover glass. The illuminating beam encounters four media interfaces; S1 (air to glass); S2 (glass to water); S3(water to glass); S4 (glass to air). The particles are at 0 on surface S2 or freely moving above surface S2. Incident light strikes surface S1.

Fig. 10 shows illumination of a sample from above. L is the lens; C is the collection cone.

Fig. 11 shows illumination using a prism arrangement (illumination from below). S1 is surface of prism where light is incident; S2 and S3 are the bottom and top surface of a plastic piece substrate respectively.

Figs. 12A, 12B, 12C, 12D, and 12E represent a pore prism (12A), an equilateral prism (12B), a home made prism (12C and 12D), and a plane convex lens, respectively.

Fig. 13 shows an illuminating light beam as viewed with a rhodamine plastic block.

Fig. 14 shows a surface and associated planes for descriptive purposes for Figure 15; S1 is solid substrate optically or not optically transmissive; SP1 is the 3-dimensional space above the plane of surface S1; SP2 is the 3-dimensional space below plane of surface S1. Light Scattering particles or material is in the SP1 plane at or near the surface S1.

Fig 15 summarizes the different methods of DLASLPD illumination and detection.

Fig. 16 shows the experimentally measured scattered light intensity versus incident wavelength spectrum for coated and uncoated 100nm diameter gold particles.

Figs. 17, 18, and 19 show various sample chamber designs to reduce the level of non-specific light background. These sample chambers can test both liquid and immobilized samples. For Fig. 17, S1 is the surface where the light beam impinges on the sample chamber; S2 is the surface that contains the light scattering material(for immobilized samples). S3 is another beveled surface; the surfaces S1 and S3 are beveled at angles of from about 20 degrees to 70 degrees depending on the angle of illumination, the face of surface S1 should be angled so that the light beam strikes S1 at 0 degrees with respect to the perpendicular; S4 is a optically transmissive surface with or without an opening; S5 is the opposite side of the surface S2. If the chamber is enclosed(i.e. S4 is solid with no opening), a small opening is placed at one of the surfaces for the introduction of sample and washing if required.

Fig. 18 is similarly designed as Figure 17 except the beveled sides are replaced with curved sides. Everything else is the same as the design of Figure 17.

Figure 19, S1 is a flat beveled optically transmissive surface where the light beam impinges on the sample chamber. The face of surface S1 should be angled so

that the light beam strikes S1 at 0 degrees; S2 is the surface that contains the light scattering material if the material is immobilized; S3 is another curved or beveled surface. S4 is an optically transmissive surface for a sample chamber that is enclosed; Alternatively, S4 has an opening of various size and shape for introduction and washing of sample and detection; S5 is the opposite side of the surface S2. If the chamber is enclosed, then a small opening is required at one of the surfaces for the introduction and/or washing of sample.

Fig. 20 shows a coordinate system that is used to describe the interaction of particles with polarized light. Light travels along y and is polarized in the z direction. D is the detector of scattered light intensity. γ is the direction of observation. θ and ϕ are the core angle and polar angle respectively.

Fig. 21 shows a schematic of an instrument for analysis of liquid samples. Light from a filament or discharge lamp is focused by a lens system, represented by lens L1, onto the entrance slit of a monochromator; monochromatic light which exits the monochromator is collected by lens L2 and focused by lens L3 onto the center of the transparent sample cuvet (ST); the sample cuvet contains a liquid solution of fluorescent molecules or suspension of light scattering particles; the sample cuvet is angled or inclined so that light reflected from its walls is deflected downward and away from the photodetector; light scattered or emitted by the sample is collected by the lens L4 which forms at the plane M a magnified image of the sample cuvet and liquid contents; in the plane M is positioned a small aperture which selectively allows light emitted or scattered by the liquid at the center of the cuvet to reach the photodetector but blocks the photodetector from light reflected or scattered from the side walls of the sample cuvet; the magnified image at M of the center of the liquid contents is displaced from the optic axis of lens

L4 by refractive index effects of the wall of the inclined sample cuvet through which the emitted or scattered light is detected; the photodetector and aperture are positioned to one side of the optic axis of lens L4 so that the displaced image of the liquid center falls on the aperture and photodetector; holders for introducing optical filters and/ or polarizers into the illuminating light and scattered light paths are provided (H1 and H2); a light shutter is positioned in front of the photodetector. If the photodetector has a small light sensitive area, a lens L5 is used to focus the light crossing the aperture at M unto the light sensitive area. If monochromatic light is not required, the monochromator can be easily removed and light from the filament or discharge lamp can be delivered directly by lenses L1, L2 and L3 to the center of the sample container.

Figs. 22, 23, and 24 outline methods for using light scattering particles and specific DLASLPD methods, which leads to specific test kits and apparatus.

Figs. 25 outlines an apparatus and assay development process.

Figs. 26 A, B, C, D, E, and F show the calculated scattered light intensity versus incident wavelength profiles for spherical 10 nm diameter gold, silver, aluminum, copper, selenium, and polystyrene particles respectively. L_k is the wavelength; C_{sca} is the light scattering cross section.

Figs. 27 A, B, C, D, E, and F show the calculated scattered light intensity versus wavelength of incident light for gold particles of various sizes. A, B, C, D, E, and F correspond to spherical gold particles of 10, 20, 40, 60, 80, and 100 nm in diameter respectively. REL CSCA is the relative light scattering cross section; WAVE,NM is the wavelength.

Fig. 28 is a diagram of a spherical coated particle. (1) is a coat of polymer, binding agent, or other

substance on the surface of the particle; (2) is the core particle.

Figs. 29 A, B, and C show diagrams of MLSP (Manipulatable Light Scattering Particle) mixed composition particles. A(1) is a core magnetic or ferroelectric material coated with (2)the desired light scattering material; B shows (4) a light scattering material core coated with (3) magnetic or ferroelectric material; C shows a mixture of (5) light scattering material with (6) magnetic or ferroelectric material.

Figs. 30 A, B, and C show dimer, tetramer, and higher order particle constructs respectively for orientable MLSP particles. (1) are light scattering detectable particles and (2) are magnetic or ferroelectric particles. The line (3) is the linkage chemical, ionic, or other that binds the particles together in the multi-particle construct.

The following abbreviations are used herein.

E-EMR - Emitted Electromagnetic Radiation

I-EMR - Incident Electromagnetic Radiation

EMF - Electromagnetic Field

SC - Semiconductor

Sec - Second

Q_f - Fluorescence Quantum Yield

I_{abs} - Incident Light Absorption (Photons absorbed per sec)

I_o - Incident Light Intensity (Photons per sec)

M - Molarity (Moles per liter)

ml - Milliliter

mM - Millimolar

g - gram

mg - milligram

mm - millimeter

μl - microliter

pI - Isoelectric point

E or e - Molar Decadic Extinction Coefficient (M^{-1} cm^{-1})

C - Molar Concentration (M)

X - Optical Path Length (cm)

I_f - Fluorescence Intensity (Photons per sec)

S_{eff} - Scattering efficiency of a particle

C_{abs} - Absorption Cross Section (cm^2)

A_{CSR} - Ratio of Particle Absorbance Cross Section Over the Particle Physical Cross Section Area

C_{sca} - Scattering Cross Section (cm^2)

S_{CSR} - Ratio of Particle Scattering Cross Section Over the Particle Physical Cross Section Area

a - Radius of a Particle

C_{ext} - Scattering Extinction Cross Section of Particle (cm^2)

I - Photons per sec which exit a solution after passing through a solution thickness X

N - Particle Concentration (particles/cm^3)

t - Turbidity of Suspension

I_s - Scattering Intensity (Photons/sec)

n_2 - Refractive Index of Material

n_2Rel - Real Component of n_2

n_2Im - Imaginary Component of n_2

n_1 - Refractive Index of Medium

m - The ratio of Refractive Index of Particle Material to Refractive Index of Medium

l_o - Incident Light Wavelength (nm)

RI - Refractive Index Factor

Refmed - Refractive Index of Medium (n_1)

e_m - Dielectric constant of medium

n_m - Refractive index of medium

a - Determines the polarizability of the coated particle

nm - nanometer

cm - centimeter

μ - micron

Description of the Invention

This invention features a method for the detection and measurement of one or more analytes in a sample. The method is based on the use of specific types of particles of certain composition, size, and shape and the detection and/or measurement of one or more of the particle's light scattering properties.

The present invention is easier to use, has greater sensitivity and specificity, and is capable of detection and measurement across a wider concentration range of analyte(s) than was previously possible. The present invention has many advantages over the use of the methods of signal and target analyte amplification (e.g. chemiluminescence and PCR), fluorescence labels and fluorescence methods, and previous particle-based assays and light scattering methods. The method is versatile and has broad application to the field of diagnostics as well as other fields. The method can be used in most, if not all standard binding-pair type assays such as immunoassay and nucleic acid assays and the like for samples in the liquid-phase, mixed-phase, solid-phase, and solid-phase microarray assay formats.

Rather than illustrate the broad utility by explicit illustrations of each specific practice of a particular form of the invention, applicant describes the key elements and considerations for one of average skill in the art to practice this invention to fit most if not all analyte detection needs. Such practice leads to specific apparatus and test kits.

The disclosure presented herein enables one of average skill in the art to practice the present invention in many different forms to achieve a desired analyte or particle detection capability to suit most if not all sample types, analyte types, diagnostic assay format types, and apparatus types. The present invention is so versatile that it can be practiced to detect one or more analytes in the field (away from a lab), or in a small

medical or analytical lab, at the bedside, emergency rooms, specialized hospital care units (such as cardiac care, intensive care, trauma unit and the like), a research lab, or the capability to process many samples a day. Different types of inexpensive apparatus and test kits can be made by practice of the invention in one form or another to suit a specific analytic diagnostic need.

There are several aspects of the invention which when practiced in various combinations with each other define the analyte detection capabilities for a specific practice of the invention. Two of these aspects are (1) the use of specific particle types that possess highly measurable and detectable light scattering properties in a defined assay format and sample type, and (2) use of specific particle types with a preferred method of DLASLPD illumination and detection. In certain applications, refractive index enhancement methods and DLASLPD video contrast enhancement methods are also used.

Determination of Useful Light Scattering Properties of Metal-like Particles

The following provides information helpful to fully understand the claimed invention. These formulae are useful in practice and optimization of the invention, but are not admitted to be prior art to the claims.

In the development of the novel signal generation and detection system for the detection of analytes of the present invention, we found it useful to develop new formulae which allowed us to evaluate various light scattering attributes of different particle types in terms of fluorescence parameters. This allowed us to study ϵ, Qf, fluorescence and excitation spectra, dependence of the emitted light intensity on the angle of observation, and state of polarization of the emitted light (these are defined below). These novel formulae allows one of skill in the art to select the specific particle parameters such as composition, size, and shape to embody desirable light

scattering property(s) that can be detected and measured when used in diagnostic assays or any other application. Equations 1 through 7 are presented as background information so the reader will understand the new formulations of Equations 8 through 15. It should not be taken as an admission that any of the formulae or light scattering parameters described is prior art to the claims.

Applicant developed an analytical method based on certain modifications of the art known light scattering theories of Rayleigh and Mie to evaluate many different types of particles with different parameters of size, shape, composition, and homogeneity to determine what specific configurations of particle parameters result in desirable light scattering signals that are easily detected and measured in analytical and diagnostic assays.

Definitions of Fluorescence Parameters

For fluorescent materials, the fluorescence intensity is determined by the product of the number of photons which are absorbed per second and the fraction of absorbed photons that are re-emitted as light (the Qf) as shown by equation 1.

$$Iabs(\lambda) = 2.303\ Io(\lambda)e(\lambda)Cx \quad (1)$$

where $I_o(\lambda)$ is the incident light intensity (photons/sec) wavelength λ, $I(\lambda)$ is the molar decadic extinction coefficient in units of M-1 cm-1 at wavelength λ and C is the molar concentration (in units of M) of the fluorophore and x is the optical path length in cm.

The integrated fluorescence intensity $I(\lambda_f)$ (sum of photons emitted in all direction per sec) at the emission wavelength λ_f and excitation wavelength λ_e is given by (for low fluorophore concentrations)

$$I(\lambda_f) = 2.303\ I_o(\lambda_e)Q_f(\lambda_f)e(\lambda_f)Cx \quad (2)$$

The assessment of the usefulness of a fluorescent compound in an assay application in terms of the listed parameters is a well known procedure. Use of fluorescent molecules

and fluorescent techniques is limited by the photo-stability of the fluorescent molecule, and the ability to detect the specific fluorescence emission signal in samples with high levels of non-specific fluorescence, phosphorescence, and scattered light. For sensitive detection of fluorescent molecules or other fluorescent substances such as particles composed of fluorescent dye molecules, more sophisticated instrumentation is required.

Definitions of Light Scattering Parameters

Absorption Cross Section (C_{abs}) of a Particle

Consider a particle that is illuminated by a monochromatic beam of light of wavelength λ. The absorption cross section C_{abs} of the particle is defined as an area (usually expressed in units of cm^2 or μ^2) surrounding the particle, such that any photon falling on this area will be irreversibly absorbed by the particle. The value of C_{abs} depends on the particle size, composition, shape and homogeneity. It also depends on the wavelength of light and a plot of Cabs vs. wavelength gives the pure absorption profile of the particle. The Cabs vs. wavelength profile for any spherical particle with a homogeneous composition can be calculated with Rayleigh or Mie theory. In our terminology, C_{abs} is related to irreversible light absorption. The nature of C_{abs} can be better understood by reference to the section below where we define the extinction cross section C_{ext}.

Relative Absorption Cross Section A_{csr}

The relative absorption cross section A_{csr} is defined as the ratio of the particle's C_{abs} divided by the physical cross sectional area of the particle πa^2 where a is the radius of the particle, i.e., $A_{csr} = C_{abs}/\pi a^2$. A_{csr} provides a measure of the particle's ability to irreversibly absorb photons falling on the area surrounding the particle. A_{csr} can have values in the range of 0 to 6 depending on the composition, shape, and size of particle and the

wavelength of light. A value greater than one means that a particle can reach beyond its physical dimensions to attract photons to it and absorb them. In the physical literature, A_{csr} is called the absorption efficiency factor of the particle. This nomenclature is misleading since A_{csr} can have values greater than 1, uncharacteristic of an efficiency.

Light Scattering Cross Section (C_{sca}) of a Particle

There is a finite probability that a photon of light absorbed (absorbed here includes reversible and irreversible absorption) by a scattering particle is re-emitted at the same wavelength as the absorbed photon (quantum mechanical point of view). The re-emitted photon can be emitted in directions different from the direction of the incident photon. That is, the incident photons are scattered by absorption and re-emission. The scattering cross section of a particle (C_{sca}) at the incident wavelength is defined as an area surrounding the particle such that every photon which falls on that area is scattered (that is absorbed and then re-emitted in the quantum mechanical view). C_{sca} is usually expressed in units of cm^2 or μ^2 and depends on the composition, shape, size and homogeneity of the particle and the wavelength. The light scattering profile C_{sca} versus wavelength can be calculated for any spherical particle of homogeneous composition using Rayleigh or Mie theory.

Ratio of C_{sca} to Physical or Geometric Cross Sectional Area of a Particle (S_{csr})

The ratio of the particle's C_{sca} divided by the physical or geometrical cross sectional area of the particle πa^2, (where a is the spherical radius of the particle) provides a measure of the particle's ability to attract, absorb, and reemit photons from the area surrounding the particle). That is $S_{csr} = C_{sca} / \pi a^2$. In the

physical literature, S_{csr} is called the scattering efficiency factor.

Experimental and theoretical results show that the value of S_{csr} can be in the range of 1 to 5 or greater depending on particle composition, shape, homogeneity and size and wavelength of light. A S_{csr} value greater than one means that a particle can reach beyond its physical dimensions to attract photons to it and absorb and then re-emit them. This is possible because an electrical interaction of the particle with the electromagnetic wave of the photon can occur at distances larger than the radius of the particle. In general, S_{csr} increases with particle size. For small particles (less than ~40nm) the S_{csr} is less than one while for larger particles Scsr equals greater than one and can reach a value of five for the larger particles.

Extinction Cross Section (C_{ext}) of a Particle

The extinction cross section C_{ext} of a light scattering particle is defined as the sum of the scattering cross section (C_{sca}) and absorption cross section (C_{abs}) of the particle.

$$C_{ext} = C_{sca} + C_{abs} \quad (3)$$

C_{ext} is usually expressed in terms of cm^2 or μ^2.

The extinction cross section C_{ext} of any particle can be readily measured at any given wavelength in a regular absorption spectrometer. Let I_o (photons/sec) be the intensity of a light beam falling on suspension of particles which are at a concentration of N particles/cm^3. X(cm) is the thickness of the solution and I (photons/sec) is the amount of light which exits the solution after traversing the distance x. The intensities are related to C_{ext} by the expression:

$$I_{(\lambda)} = I_{o(\lambda)} e^{-N C_{ext(\lambda)} x} \quad (4)$$

This expression shows explicitly the parameters depend on λ. It is assumed that the photodetector is positioned such that it does not detect scattered light.

When the particles are pure scatterers, that is, do not irreversibly absorb any light, then $c_{ext} = C_{sca}$ and the above equation is written as

$$I=I_o e^{-NC_{sca}x} \quad (5)$$

$$=I_o e^{-\tau x} \quad (6)$$

where $t = NC_{sca}$ is the turbidity of the suspension.

Molar Decadic Extinction Coefficient

In the field of chemistry, the strength with which a substance in solution absorbs light at a given wavelength is expressed in terms of the molar decadic extinction e which has units of M^{-1} cm^{-1} (M stands for moles/liter). This coefficient is related to the experimentally determined absorbance by the expression

$$A_{(\lambda)}=e_{(\lambda)}Cx \quad (7)$$

Applicant Developed Formulae for Studying Particle Light Scattering Parameters

Applicant now briefly presents his own theoretical methods used. One skilled in the art can use the following methods to evaluate, modify, and adjust specific particle parameters of composition, size, shape, and homogeneity to derive one or more desirable light scattering properties that are easily detected and measured. Considerations must be made with regard to sample types, diagnostic formats, and limitations of apparatus means. For example, in one application, multi-analyte detection may be performed on a solid-phase sample that contains a high non-specific light background on a high throughput testing apparatus, while in another application, single analyte detection in solution is performed in the doctors office.

Applicant's major interest was in optimizing particle types for use in analytical and diagnostic assays. In most of these applications, the particles must be coated with a macromolecular substance such as polymer, protein, or the like to confer suitable chemical stability in various

mediums. This is known in the art. Applicant also places binding agents such as antibodies, receptors, peptides, and the like on the surface of the particle so that the coated particle can be used in an analytic or diagnostic format. In some applications, the binding agent serves a dual function in that it stabilizes the particle in solution and provides the specific recognition binding component to bind the analyte. The coating of particles with proteins such as antibodies is known in the art. However, applicant was interested in measuring one or more specific parameters of the light scattering signals of different types of particles which in some cases are of similar size and/or shape and/or composition and it was not clear if such optical resolvability of one or more of the specific light scattering properties of coated particles was possible.

Applicant determined by physical experimentation and theoretical modeling that the presence of thin coats of binding agents, non-optically absorbing polymers (in the visible region of the spectrum), or other materials on the particle surface does not noticeably alter the light scattering properties specific for that type of particle which is not coated with these types of materials. By "thin coat" is meant monolayer(s) of different amounts and compositions of the above materials coated on the surface of the particle.

Applicant determined that a molar decadic extinction coefficient can be determined at any wavelength for a coated or uncoated particle suspension by measuring its absorbance at that wavelength. The molar decadic extinction coefficient at that wavelength can then be calculated with Eq.(7) and the following expression to convert particle concentration from N(particles/cm^3) to C(M). M is moles/liter.

$$C(M) = 1000\ N(particles/cm^3)/6.03x10^{23} \qquad (8)$$

The molar decadic extinction coefficient can be related to the extinction cross section Cext (or vice versa) by the expression

$$e(M^{-1}cm^{-1}) = [C_{ext}(cm^2/particle)(6.03x10^{23})]/2.303x1000 \quad (9)$$

$$= 2.63x10^{20}C_{ext}(cm^2/particle) \quad (10)$$

or

$$C_{ext}(cm^2/particle) = 2.303\ (M^{-1}cm^{-1})x1,000/6.03x10^{23} \quad (11)$$

$$= 3.82x10^{-21}\ e(M^{-1}cm^{-1}) \quad (12)$$

With Eq. (9)or (10) we can calculate ϵ from C_{ext}.

As described previously, it is well known in the art that for particles, the extinction cross section (C_{ext}) is equal to the sum of the scattering cross section (C_{sca}) and the absorption cross section (C_{abs}). The extinction coefficient e reflects the loss of photons from the incident beam by irreversible absorption as well as by scattering (absorption and re-emission). Applicant has determined that the molar decadic extinction coefficient of a particle, evaluated experimentally or by calculation from the extinction cross section, can be used to compare the absorption power of a particle with, for example, that of a fluorophore as shown later.

Light Scattering Efficiency (S_{eff})

Applicant determined that a light scattering efficiency S_{eff} can be defined for a coated or uncoated particle, by analogy to fluorescence efficiency Q_f, as the fraction of photons absorbed (reversible plus irreversible absorption)by a particle that are re-emitted as scattered light. Mathematically, applicant defines the scattering efficiency by the expression

$$S_{eff} = C_{sca}/C_{ext} \quad (13)$$

$$= C_{sca}/(C_{sca} + C_{abs})$$

For particles which are pure scatters, that is, particles composed of a material which does not irreversibly absorb photons but only absorbs and re-

emits photons, C_{abs} is equal to zero and S_{eff} is equal to one. Small polystyrene particles behave as pure light scatters in the visible region of the spectrum and S_{eff} is 1 for these particles. For particles composed of materials which reversibly and irreversibly absorb photons, S_{eff} is less than one. Gold particles display the latter type of behavior in the visible region of the spectrum.

Intensity of Light Scattered by a Particle

Applicant determined that the intensity of light scattered by a coated or uncoated particle is determinable by the product of the number of photons which are absorbed (reversibly and irreversibly) per second and the fraction of the absorbed photons that are re-emitted (quantum mechanical point of view). Light scattering intensity measurements are usually done in dilute solutions where the amount of light absorbed, I_{abs} (photons absorbed per second) is given by

$$I_{abs} = I_o\, 2.303\, e\, C\, x \qquad (14)$$

I_o is the incident light intensity (photons/sec), ϵ is the molar decadic extinction coefficient of the particles in terms of $M^{-1}cm^{-1}$, C is the molar concentration of the particles and x is the optical path length in cm.

Applicant further realizes that the total scattered light intensity I_s, intensity integrated over all light scattering angles, is then given by the relation:

$$I_{s(\lambda)}=2.303\, I_{o(\lambda)}\, S_{eff(\lambda)} e_{(\lambda)}\, (C)\, (x) \qquad (15)$$

where $I_{o(\lambda)}$ is the intensity of the incident light. This equation is comparable to Eq.(2) for a fluorophore.

Note that when the expressions for ϵ and S_{eff} in terms of C_{sca} and C_{ext} are inserted in the above equation, the result shows that the scattered light intensity is directly proportional to and completely determined by the magnitude of the scattering cross section (C_{sca}). This means that the relative scattering intensities of different particles can be predicted from their scattering cross sections.

Specific Light Scattering Properties of Particles

Applicant now briefly summarizes some of the most important light scattering properties that can be used to detect analytes in various sample types using a variety of different assay formats. The measured light scattering properties that are detected are one or more of the following: the intensity, the wavelength, the color, the polarization, the angular dependence, and the RIFSLIW (rotational individual fluctuations in the scattered light intensity and/or wavelengths) of the scattered light.

Coated and uncoated metal-like particles have similar light scattering properties and both have superior light scattering properties as compared to non-metal-like particles. In addition, applicant has determined that it is relatively easy to adjust the types of light scattering properties in metal-like particles by varying in one form or another, the size, shape, composition, and homogeneity such that the specific light scattering attributes can be measured from the metal-like particle in various sample types.

Metal-like particles can be detected to extreme sensitivity. The individual particles are easily detected to the single particle limit using DLASLPD illumination and detection methods with inexpensive and easy to use apparatus.

One or more types of metal-like particles are detected in a sample by measuring their color under white light or similar broad band illumination with DLASLPD type illumination and detection methods. For example, roughly spherical particles of gold (for example, coated with binding agent, bound to analyte, released into solution or bound to a solid-phase) of 40, 60, and 80 nm diameters and a particle of silver of about 30nm diameter can easily be detected and quantitated in a sample by identifying each particle type by their respective unique scattered light color and/or measuring the intensity. This can be done on a solid phase such as a microtitier well or microarray

chip, or in solution. The measurement in solution is more involved, because the particles are not spatially resolved as in the solid-phase format. For example, one can detect the different types of particles in solution by flowing the solution past a series of detectors each set to measure a different wavelength or color region of the spectrum and the intensity at these different wavelengths is measured. Alternatively, a series of different wavelengths of illumination and/or detection can be used with or without the flow system to detect the different particle types.

For solid-phase analytical applications, a very wide range of concentrations of metal-like particles is detectable by switching from particle counting to integrated light intensity measurements depending on the concentration of particles. The particles can be detected from very low to very high particle densities per unit area.

In other assay applications, the particles which are bound to a solid substrate such as a bead, surface such as the bottom of a well, or the like can be released into solution by adjusting the pH, ionic strength, or other liquid property. Higher refractive index liquids can be added, and the particle light scattering properties are measured in solution. Similarly, particles in solution can be concentrated by various means into a small volume or area prior to measuring the light scattering properties. Again, higher refractive index liquids can be added prior to the measurement.

Both theoretical evaluation and physical experimentation indicate that for spherical particles of any composition up to about 120 nm in diameter and somewhat greater, a large proportion of the light scattered by the particle is radiated outside the envelope of the forward direction of the scattered light (see Figure 8). Applicant determined that detection and measurement of the scattered light at angles outside of

the envelope of the forward direction of scattered light provides for a significant decrease in non-specific scattered light from the light beam and other light scattering constituents and objects being detected. This significantly increases the specific light scattering signal/non-specific light scattering ratio for many samples.

The intensity of light scattered by a particle in different directions and the state of polarization of the scattered light depends on the wavelength and state of polarization of the incident light and on the particle size, shape and homogeneity. Below we summarize some of the most important facts concerning the intensity and state of polarization of light emitted in different directions by certain types of particles.

Smaller particles of spherical shape (smaller being about 1/20 or smaller as compared to the wavelength of light) behave as isotropic dipole scatterers or emitters, that is, the light is highly polarized. This is very different to fluorescent molecules which usually behave as linear dipole emitters. For example, such particles when illuminated with unpolarized light, the light scattered in the direction ϕ=0, θ=90 (see Figure 20) is one hundred percent linearly polarized (P=1). This property allows for the more specific and more sensitive detection of analytes by measurement of the light scattering properties as compared to fluorescent molecules in many different types of samples.

For even larger particles (> 1/20 wavelength of light) there are certain ranges of particle sizes where the degree of polarization of light P decreases and becomes dependent on the wavelength as the size increases. As the particles become very large the degree of polarization approaches 0 for the direction θ=0, φ=90°. There appears to be certain size ranges where the change in polarization changes the most, that is, the slope of degree of polarization versus size is at a maximum. Those

regions where the slope is changing are used in certain analytic applications as for example, agglutination or aggregation types of assays to detect and measure one or more analytes in a sample.

For larger spherical particles in certain size ranges, for example from about 200 nm to about 1.2 microns in diameter, the intensity of light oscillates (for monochromatic incident light) between relative values of 1 to 0 as the angle ϕ is changed from 90° to -90° for θ=0 with reference to figure 20. That is, if one views the scattered light in the horizontal plane (θ=0), the light intensity oscillates from bright to dark as the eye is moved from θ = 90° to θ=-90°. For illumination with white light, the light changes color as the eye is moved from θ = 90° to θ=-90°. That is, the particles behave as diffraction gratings. Such light scattering properties are very useful to detect more specifically and to greater sensitivity one or more analytes in many different types of samples.

Small nonspherical particles behave somewhat as linear dipole scatters with the absorption and emission moments along the long axis of the particle. Applicant has observed the following under DLASLPD illumination and detection conditions in an ordinary light microscope. When the illuminating light is linearly polarized, the non-spherical particles flicker as they rotate. The particles are most intense when their orientation is such that their long axis is oriented in the direction of polarization and is at a minimum when the moment is perpendicular to this direction. In contrast, small spherical particles do not flicker when illuminated by polarized light. For nonspherical particles of certain compositions, the color of the scattered light (with white light illumination) changes with the degree of asymmetry. As the asymmetry is increased, the color shifts towards longer wavelengths. For example, asymmetric particles of silver were observed by applicant to change colors as the particles were

rotating in solution when viewed with an ordinary light microscope under DLASLPD like conditions. This property termed "RIFSLIW" by applicant (rotational individual fluctuations in the scattered light intensity and or wavelengths) is used in many different aspects of the current invention to more specifically and more sensitively detect and or measure one or more analytes or particles in a sample.

Applicant has also determined that certain mixed compositions of particles made from metal-like materials, and non-metal-like and metal-like materials provides for additional light scattering properties and/or additional physical properties. These properties include the ability to manipulate the particles by applying an EMF field. This property of the particles can be used in many different ways with the practice of one or more aspects of this invention. Applicant now provides further illustrative discussions of particle-dependent light scattering properties and the use of these properties to detect one or more analytes in a sample.

It will be useful to first describe the present invention in terms of the light scattering properties of homogeneous, spherical particles of different sizes and compositions. However, the basic aspects of the invention apply as well to non-spherical particles as one in the art can determine. In addition, it will be useful to describe the present invention in terms of the incident light wavelengths in the range 300nm to 700nm. However, the basic aspects of the invention apply as well to electromagnetic radiation of essentially all wavelengths. By "light" is meant ultraviolet, visible, near infrared, infrared, and microwave frequencies of electromagnetic radiation. It will be further useful for the description of the present invention to use polystyrene particles to represent non-metal-like particles of various types. Other non-metal-like particle types include those composed of glass and many other polymeric compounds. Such

particles have roughly similar light scattering characteristics as compared to polystyrene particles.

The relative intensities of scattered light obtained from different particles irradiated with the same intensity and wavelengths of incident light can be directly compared by comparing their C_{sca}s. The higher the C_{sca}, the greater the scattering power (light scattering intensity) of the particle. In the following sections we use the words "scattering power" to mean C_{sca} or scattered light intensity.

We have calculated the light scattering powers, in water, of small spherical particles identical in size, and different in composition, for incident wavelengths over the wavelength ranges of 300 to 700 nanometers (nm). In these calculations, we have used values of refractive index vs. wavelength tabulated in standard handbooks for different bulk materials in vacuum.

For some particle compositions, the light scattering power decreases continuously from 300 to 700 nm while for other compositions the scattering power vs. wavelength profile shows peaks or bands. When these peaks or bands are in the visible region of the spectrum the light scattered by the particles is colored when the incident light is white.

For illustrative purposes we show in some of the following tables various comparisons of light scattering properties for different types of 10 nm diameter particles. The general trends with regard to the relative magnitudes and wavelengths of scattered light that is demonstrated for these 10 nm diameter particles is generally the same for larger particles up to about 100 nm. For example, the calculations of Table 1 were done with particles of ten nanometer diameter. However, for small particles(less than about 1/20 wavelength of the light) the light scattering intensity vs. wavelength profiles do not change in shape as the particle size is increased as long as one remains in the small particle

limit. The apparent effect of increase in particle size is to increase the amplitude of the profile graph. It is well known in the art of theoretical physics and light scattering that the scattering power of small particles increases with the sixth power of the radius. One skilled in the art can calculate the relative scattering power of any small particle of diameter d from the value obtained for the 10 nm diameter particle by multiplying the light scattering power of the 10 nm particle by $(d/10)^6$ where d is in nm. This method can be used by one skilled in the art to determine the usefulness of certain particle sizes in various diagnostic assay applications where the intensity of the scattered light of a particle is used to detect the presence of an analyte.

From our theoretical and physical experimentation we were surprised to find that this general relationship does also apply to larger particles outside of the Rayleigh limit, that is, for particles with diameters larger than about 30 nm.

Table 1 presents the calculated C_{sca} values (light scattering power) and their respective approximate wavelengths in the visible range where the particles scatter light most intensely. The data of Table 1 suggest that metal-like particles are much more powerful light scatterers than, for example, polystyrene particles.

Figure 26 shows selected calculated light scattering intensity vs. wavelength profiles for certain types of 10 nm spherical particles. Small particles composed of gold or silver exhibit very prominent scattering and absorption peaks in the visible wavelength region, while copper particles exhibit a small scattering and absorption peak in this region. The gold and silver scattered light maxima occur at about 530nm and 380nm respectively. Even at incident wavelengths far removed from the light scattering maxima, the light scattering power of the gold, silver, and other metal-like particles is much greater than that of non-metal-like polystyrene particles of similar size.

Table 2 presents the calculated light scattering power (C_{sca}) values for metal-like particles and polystyrene(non-metal-like) of 10 nm diameter when the incident (illumination) wavelength has been shifted to much longer wavelengths. In many different analytic and diagnostic assays, it is preferable to work at longer wavelengths. Table 2 indicates that one skilled in the art may use illumination wavelengths at much longer wavelengths and that the metal-like particles are far superior to non-metal-like particles as for example, polystyrene for applications to analytical or diagnostic methods. For example, at an incident light wavelength of 700nm, a wavelength far from the light scattering maximum at 530nm for gold particles, the data suggest that a gold particle's scattered light intensity is about 220 times more than a polystyrene particle of similar size and shape. We have experimentally observed that indeed the light scattering power(intensity) of the metal-like particles is much greater than non-metal-like particles across the visible wavelengths of the spectrum.

These results indicate that metal-like particles have much greater light scattering power than non-metal-like particles of comparable size and shape and are broadly applicable as analytical and diagnostic tracers for use in most areas where it is desirable to use a signal generation and detection system. For example, in any assay designed to detect the presence or absence of an analyte substance by detecting the scattered light from the particles.

Table 1

CALCULATED (C_{sca}) VALUES FOR TEN NANOMETER SPHERICAL PARTICLES OF DIFFERENT COMPOSITION IN WATER

Particle Composition	Wavelength (nm) [a]	Maximum C_{sca} (cm^2)	Relative C_{sca}
Polystyrene	300	$1.32x10^{-17}$	1
Selenium	300	$8.6x10^{-16}$	~65
Aluminum	300	$1.95x10^{-15}$	~148
Copper	300	$7.8x10^{-16}$	~59
Gold	530 [b]	$1.24x10^{-15}$	~94
Silver	380 [b]	$1.1x10^{-14}$	~833

(a) Incident wavelength used and at which maximum value occurs in the visible range of the EM spectrum (300-700 nm).

(b) Some particles display peaks in certain regions. See Figure 27.

Table 2

CALCULATED VALUES OF (C_{sca}) AT 700 NM ILLUMINATION FOR TEN NANOMETER SPHERICAL PARTICLES OF DIFFERENT COMPOSITIONS IN WATER

Particle Composition	Incident Wavelength (nm)	Maximum C_{sca} (cm^2)	Relative C_{sca}
Polystyrene	700	$\sim 3.1x10^{-19}$	1
Selenium	700	$\sim 1.4x10^{-17}$	~45
Aluminum	700	$1.4x10^{-17}$	~45
Copper	700	$4.7x10^{-17}$	~152
Gold	700	$7x10^{-17}$	~225

Table 3 shows a comparison of the molar decadic extinction coefficients (ϵ) calculated and experimentally

measured for spherical gold particles of different diameter.

We calculated the ϵ values at the wavelength of maximum value using the expressions previously described. Measured ϵ values were obtained by measuring the optical absorption in a standard spectrophotometer at the calculated wavelength of maximum absorption. The agreement between calculated and experimentally measured ϵ values while not perfect, is quite good. The approximately two-fold differences observed between the observed and calculated results may reflect inaccuracies in the stated diameters of the gold particles. Details of the experimental methods are given in the Example Section.

TABLE 3

CALCULATED AND MEASURED MOLAR DECADIC EXTINCTION COEFFICIENT(S) AND WAVELENGTHS OF MAXIMUM ABSORPTION FOR GOLD PARTICLES OF DIFFERENT SIZE IN WATER

	CALCULATED[b]		MEASURED[b]	
PARTICLE DIAMETER[a,b]	e (M^{-1} cm^{-1})	WAVELENGTH AT MAXIMUM ABSORPTION	ϵ (M^{-1} cm^{-1})	WAVELENGTH AT MAXIMUM ABSORPTION
10nm	$1.8x10^{8}$	~530nm	$2.3x10^{8}$	~525nm
16nm	$9x10^{8}$	~530nm	$5.2x10^{8}$	~522nm
20nm	$1.8x10^{9}$	~528nm	$1.2x10^{9}$	~525nm
40nm	$1.6x10^{10}$	~532nm	$8.5x10^{9}$	~528nm
60nm	$5.6x10^{10}$	~542nm	$2.6x10^{10}$	~547nm
80nm	$1.1x10^{11}$	~550nm	$5.8x10^{10}$	~555nm
100nm	$1.6x10^{11}$	~574nm	$9.2x10^{10}$	~575nm

(a) Denotes exact diameter of particle for calculating values.

(b) Denotes approximate diameter of gold particles used for measurements. Actual diameters were slightly higher or slightly lower than the diameters noted.

At visible wavelengths of incident light, the light scattering power (i.e., C_{sca}) of metal-like particles is much greater than for a comparable non-metal-like particle such as polystyrene. Another important distinction between the light scattering properties of metal-like and non-metal-like particles is that for metal-like particles, the profile of scattered light intensity versus incident light wavelength for metal-like particles of same composition but varying size can be very different. This is in contrast to non-metal-like particles where in the size ranges of about 10nm diameter to a few hundred nm diameter the profile is essentially the same. These differences are extremely useful to more specifically and more sensitively detect metal-like particles in various samples. The incident wavelength at which maximum light scattering (C_{sca}) occurs for various diameter particles of silver, gold, copper, and aluminum are presented in Table 4.

Figure 16 shows the experimentally measured scattered light intensity vs. incident light wavelength profile for roughly spherical 100nm diameter gold particles coated with polyethylene compound (MW=20,000) and without the polyethylene compound. The data show that the wavelength dependent light scattering intensity properties of the coated and uncoated 100nm diameter gold particle are very similar.

Figure 27 shows the calculated scattered light intensity versus incident light wavelength spectra profiles for spherical gold particles of varying diameter. The scattered light intensity peak wavelengths shift to longer wavelengths as the size of the gold particles is increased. We have directly observed these light scattering properties for coated or uncoated gold particles of 40, 60, 80, 100 nm diameters and they appear

as green, yellow-green, orange, and orange-red particles when illuminated with a white light source in solution or in a light microscope using DLASLPD illumination methods. Small spherical silver particles appear blue. Thus, metal-like particles coated with various types of binding agents can be used in numerous ways in analytic type assays. The color properties of the scattered light of different types of metal-like particles allows for multi-analyte visual detection. For example, spherical gold particles of 40, 60, 80, and 100 nm diameter and 20 nm diameter silver particles, each coated with a different type of binding agent, can be used in the same sample to detect five different analytes in the sample. In one format, five different types of cell surface receptors, or other surface constituents present on the cell surface can be detected and visualized. Detection of the scattered light color of the differently coated particles that are bound to the surface of the cell under DLASLPD conditions with a light microscope with white light illumination makes this possible. The number and types of analytes are identified by the number of green, yellow, orange, red, and blue particles detected. Similarly, chromosome and genetic analysis such as <u>in situ</u> hybridization and the like can also be done using the method as described above where the different types of metal-like particles are used as "chromosome paints" to identify different types of nucleic acid sequences, nucleic acid binding proteins, and other similar analytes in the sample by the color of the scattered light of the different types of metal-like particles. These examples are provided as illustrative examples, and one skilled in the art will recognize that the color of the scattered light of different types of metal-like particles can be used in many different assay formats for single or multi-analyte detection.

Thus, adjusting the size of certain types of spherical metal-like particles is a useful method to increase their detectability in various samples by using

the color and/or other properties of their scattered light. By using a white light source, two or more different types of particles are easily detectable to very low concentrations.

Table 5 shows that modest increases in gold particle size results in a large increase in the light scattering power of the particle (the C_{sca}). The incident wavelength for the maximum C_{sca} is increased significantly with particle size and the magnitude of scattered light intensity is significantly increased. For example, the incident wavelength for maximum C_{sca} is around 535nm, 575 nm and 635nm for gold particles 40nm, 100nm, and 140nm in diameter, respectively. When illuminated with white light, the 40nm gold particles strongly and preferentially scatter the wavelengths around 535nm and the particles appear green, while the 100nm particles appear orange-red and the 140nm particles appear red in color. This further shows that when illuminated with white light, certain metal-like particles of identical composition but different size can be distinguished from one another in the same sample by the color of the scattered light. The relative magnitude of the scattered light intensity can be measured and used together with the color or wavelength dependence of the scattered light to detect different particles in the same sample more specifically and sensitively, even in samples with high non-specific light backgrounds.

In contrast, for non-metal-like particles, these particles do not possess these specific types of light scattering properties and thus, it is more difficult to detect the non-metal-particles in most types of sample medium as compared to the metal-like particles.

TABLE 4

CALCULATED INCIDENT VISIBLE WAVELENGTH AT WHICH MAXIMUM C_{sca} IS OBSERVED IN WATER

PARTICLE MATERIAL	PARTICLE DIAMETER	WAVELENGTH OF INCIDENT LIGHT AT WHICH MAXIMUM C_{sca} OCCURS
Silver	10nm	~380nm
	40nm	~400nm
	100nm	~475nm
Gold	10nm	~528nm
	40nm	~535nm
	100nm	~575nm
	140nm	~635nm
Copper	100nm	~610nm
	150nm	~644nm
Aluminum	100nm	~377nm
Selenium	130nm	~660nm
	200nm	~702nm

TABLE 5

CALCULATED VALUES FOR LIGHT SCATTERING CHARACTERISTICS OF SPHERICAL GOLD PARTICLES OF DIFFERENT SIZES

PARTICLE DIAMETER (nm)	WAVELENGTH AT MAXIMUM C_{csa} (nm)	MAXIMUM C_{sca} (cm^2)	CALCULATED RELATIVE SCATTERING POWER
10	~528	1.26×10^{-15}	1
20	~525	8.4×10^{-14}	67.5
30	~530	1.03×10^{-12}	817
40	~535	6×10^{-12}	4.8×10^{3}
60	~545	6.3×10^{-11}	5×10^{4}
80	~555	2.3×10^{-10}	1.8×10^{5}
100	~575	4.6×10^{-10}	3.6×10^{5}
120	~605	6.9×10^{-10}	5.5×10^{5}
140	~635	8.8×10^{-10}	7×10^{5}
160	~665	1×10^{-9}	7.9×10^{5}
200	~567	1.4×10^{-9}	1.1×10^{6}
300	~670	2.9×10^{-9}	2.3×10^{6}
600	~600	1.01×10^{-8}	8×10^{6}
1,000	~620	2.5×10^{-8}	1.8×10^{7}
1,000	~670	2.5×10^{-8}	1.8×10^{7}

The relative light scattering powers of particles of the same shape and size, but of different composition, can be directly compared experimentally by comparing the light scattering intensities at right angles to the path of the incident light. We experimentally compared the relative light scattering powers of gold and polystyrene particles of similar size and shape, using a light

scattering instrument we built designed to measure scattered light at right angles to the path of the incident light and which is described elsewhere. Table 6 shows that the experimentally measured scattering power of a particle composed of gold is much greater than the scattering power of a particle composed of polystyrene when both particle types are compared at the same incident visible wavelength. The experimentally measured values of Table 6 are a factor of two to three lower than the calculated values. A large part of this difference can be attributed to the approximately two-fold lower values obtained for the experimentally measured molar decadic extinction coefficients of gold particles relative to the calculated values (see Table 3). In addition, there is a certain level of uncertainty in particle sizes (e.g. for a polystyrene particle preparation 21nm±1.5nm the actual size could be about 1.5 nm larger or smaller). This uncertainty makes the quantitative values less certain for both the polystyrene and gold particles but does not change the basic conclusion concerning the relative scattering powers. Even at the greatest level of uncertainty, Table 6 indicates that at a minimum, the scattering power of a gold particle is 100 to 200 times greater than that of a polystyrene particle of comparable size and shape.

Table 7 compares the relative light scattering power of spherical gold and polystyrene particles of similar size and shape, at different wavelengths of visible light. Table 7 indicates that even at illumination wavelengths far away from the wavelength of maximum light scattering intensity, the light scattering power of a gold particle is much greater than that of a polystyrene particle of comparable size and shape. These experimental results agree with our calculated results(see Table 2).

Table 8 shows that the experimentally determined light scattering power of spherical gold particles is

much greater than comparable polystyrene particles using white incandescent light illumination conditions.

Overall, the agreement between our calculated and experimentally determined results presented herein is quite good. This validates the calculated results as well as the use of the calculation process for identifying potentially useful particle materials and compositions and for evaluating the utility of the light scattering properties of such particles. Most types of light sources which produce polychromatic and/or monochromatic light, steady-state and/or pulsed light, and coherent or not coherent light can be used for illumination. Our results indicate that more specific and more intense light scattering signals can be can be obtained from metal-like particles as compared to non-metal-like particles of comparable size and shape. Our results indicate that the present invention provides a means to detect lesser amounts of particles, and to more specifically detect lesser and greater amounts of particles than was previously possible.

TABLE 6

CALCULATED AND MEASURED RELATIVE SCATTERING POWER IN WATER OF POLYSTYRENE AND GOLD PARTICLES OF SIMILAR SIZE AND SHAPE AT THE INCIDENT WAVELENGTH AT WHICH MAXIMUM SCATTERING OCCURS FOR THE GOLD PARTICLES

PARTICLE COMPOSITION	PARTICLE SIZE(b)	INCIDENT WAVELENGTH	RELATIVE SCATTERING POWER (a) CALCULATED	MEASURED
PST[c]	21 ± 1.5nm	~525nm	1	1
Gold	19.8±<1.9nm	~525nm	~664	~220
PST	32 ± 1.3nm	~527nm	1	1
Gold	29.5 ±<3.5nm	~527nm	~663	~318
PST	41 ± 1.8nm	~530nm	1	1
Gold	39.5 ± <6nm	~530nm	~985	~461
PST	83 ± 2.7nm	~560nm	1	1
Gold	76.4 ± 15nm	~560nm	~708	~211

(a) Calculated for perfectly spherical particles

(b) Manufacturer's measured particle sizes. The manufactured particles are spherical in shape but not perfectly spherical. This will have little effect on the quantitative aspects of this comparison.

(c) PST-polystyrene

TABLE 7

MEASURED RELATIVE SCATTERING POWER IN WATER OF POLYSTYRENE AND GOLD PARTICLES OF SIMILAR SIZE AND SHAPE AT INCIDENT WAVELENGTHS AWAY FROM THE GOLD PARTICLE ABSORPTION BAND

PARTICLE COMPOSITION	PARTICLE SIZE[a]	INCIDENT WAVELENGTH	VISIBLE WAVELENGTH OF MAXIMUM SCATTERING	RELATIVE SCATTERING POWER	
				DIRECT COMPARISON	[c]CORRECTED FOR PARTICLE SIZE
PST[b]	41.1 ± 1.8nm	~530nm ~460nm ~400nm	~300nm	1 1 1	1 1 1
Gold	39.9 ±<6nm	~530nm ~460nm ~400nm	~530nm	~377 ~96 ~80	~443 ~113 ~ 94
PST	83 ± 2.7nm	~560nm ~430nm ~380nm	~300nm	1 1 1	1 1 1
Gold	76.4 ±<15nm	~560nm ~430nm ~380nm	~560nm	~251 ~ 36 ~ 27	~412 ~ 55 ~ 41

(a) Manufacturer's measured particle size

(b) PST-Polystyrene

(c) Adjust the relative light scattering values for the gold particles for any difference in size between the PST and gold particles by using the relationship that scattering power increases as the sixth power of the radius. Although in these size ranges, this is not quite accurate, the corrected figures are a good approximation.

TABLE 8

MEASURED RELATIVE SCATTERING POWER IN WATER OF WHITE LIGHT ILLUMINATED POLYSTYRENE (PST) AND GOLD PARTICLES OF A SIMILAR SIZE AND COMPOSITION

[a]PARTICLE COMPOSITION	[b]PARTICLE SIZE	RELATIVE SCATTERING POWER AT SAME CONCENTRATION	[c]RELATIVE SCATTERING POWER ADJUSTED FOR PARTICLE SIZE
PST	21nm	1	1
GOLD	19.8nm	~40	~58
PST	38nm	1	1
GOLD	39.9nm	~212	~158
PST	37.9nm	1	1
GOLD	39.9nm	~105	~77
PST	59nm	1	1
GOLD	59.6nm	~100	~94
PST	79nm	1	1
GOLD	76.4nm	~108	~132

(a) Polystyrene particles obtained from interfacial Dynamics, Inc., Portland, Oregon or Duke Scientific, Inc., Palo Alto, CA. While the gold particles were obtained from Goldmark Biologicals, Phillipsburg, N.J. a distributor for British Biocell LTD., Cardiff, UK.

(b) Manufacturers represented particle diameter.

(c) Done using the relationship that scattering power increases approximately as the sixth power of the particle radius.

Particle Light Generation Power Compared to Fluorescence

Fluorescence is currently being used in many assays designed to detect the presence or absence of an analyte substance.

Fluorescein is one of the best understood and most widely used fluorescent compounds. Many studies have been conducted with the purpose of detecting as few fluorescein molecules as possible. Fluorescein has a high molar decadic extinction coefficient (about $6 \times 10^4 M^{-1} cm^{-1}$) and has a very high fluorescent quantum yield of about 0.8.

Table 9 compares the calculated signal generating power of certain particles to fluorescein. Clearly, a single gold or silver particle is a much more intense light source than a single fluorescence molecule. Under ideal conditions and using appropriate optical filters, a good fluorimeter can detect fluorescein at a lower concentration of about 10^{-10}M to 10^{-11}M. The comparison presented in Table 9 indicates that this same fluorimeter should be able to detect a lower concentration of a 60nm gold particle of around 10^{-15}M - 10^{-16}M. We have verified these observations experimentally.

Table 9 indicates that the total scattered light output from a single 60nm gold particle is equivalent to the output of about 350,000 fluorescein molecules. While one fluorescein molecule cannot be directly visualized in the light microscope, we are able to directly visualize individual metal-like particles in many different types of samples and assay formats. The light is directed at the sample with such an angle so that the light scattered from the particle is maximally visualized or measured by the eye or photodetector. This broadly applicable method of illumination and detection as we have developed in one form or another for use in analytic and diagnostic applications is called DLASLPD (direct light angled for scattered light of particle only to be detected). These methods are described in greater detail elsewhere. This allows the detection of single particles and the quantitation of such particles by particle counting methods including image analysis, photon correlation spectroscopy, light microscopy and other methods. In contrast, only very large particles of polystyrene can be seen in the light microscope using DLASLPD techniques.

Table 10 presents results comparing the experimentally measured relative signal generating power of fluorescein and gold particles of various sizes using white light illumination. These results are similar to those presented in Table 8 and show that the light

generation power of a gold particle is much greater than a fluorescein molecule. For example, gold particles of a diameter of 39.9nm and 59.6nm emit a light intensity equivalent to that given off by about $2x10^4$ and $2.3x10^5$ fluorescein molecules respectively, when illuminated with white light.

The scattered light emitted by gold particles illuminated with white light is composed of all of the wavelengths present in the incident white light, but the efficiency of light scattering at any particular wavelength varies such that one or more bands of scattered light wavelengths are scattered more intensely. The actual wavelength composition and the scattered light wavelength versus scattered light intensity profile obtained when white incident light is used, depends on a number of variables which include the type of light source used and the method of light detection. The results of Table 10 were obtained with an incandescent light source or color temperature of 2,800°Kelvin and the light was passed through a simple filter to reduce the infrared component before passing through the sample. The scattered light intensity was measured with a standard photomultiplier tube. The results of Table 10 have not been corrected for phototube or light source properties. Any such corrections would not affect the conclusions discussed herein.

TABLE 9

CALCULATED RELATIVE SIGNAL GENERATING POWER OF FLUORESCEIN AND SPHERICAL PARTICLES OF VARIOUS COMPOSITIONS AND SIZES

PARTICLE COMPOSITION	PARTICLED IAMETER	PARTICLE VOLUME (micron)^3	NUMBER OF FLUORESCEIN MOLECULES NECESSARY TO MATCH THE TOTAL LIGHT INTENSITY FROM ONE PARTICLE[a]
Polystyrene	10nm	$5.23x10^{-7}$	~0.07
	20nm	$4.2x10^{-6}$	~5
	40nm	$3.35x10^{-5}$	~280
	60nm	$1.13x10^{-4}$	~2800
	100nm	$5.23x10^{-4}$	~42,000
Silver	10nm	$5.23x10^{-7}$	~46
	20nm	$4.2x10^{-6}$	~3,500
	40nm	$3.35x10^{-5}$	~150,000
	60nm	$1.13x10^{-4}$	~770,000
	100nm	$5.23x10^{-4}$	~2,300,000
Gold	10nm	$5.23x10^{-7}$	~7
	20nm	$4.2x10^{-6}$	~455
	40nm	$3.35x10^{-5}$	~35,000
	60nm	$1.13x10^{-4}$	~350,000
	100nm	$5.23x10^{-4}$	~3,100,000

(a) Fluorescein and the particle are illuminated with the same light intensity at wavelengths which generate the maximum fluorescent or light scattering signal. For polystyrene the incident light wavelength used was 300nm, while the incident light wavelength used for each gold or silver particle was the wavelength at maximum C_{sca} for the various sizes.

Results from measurement of the relative signal generating powers of fluorescein and roughly spherical gold particles of different size when illuminated with incident monochromatic light are presented in Table 11. The fluorescein sample was illuminated with monochromatic light of an incident wavelength (490nm) and the resulting

emitted light was not monochromatic or polarized and was composed of the wavelengths characteristic of fluorescein emission. Different sized spherical gold particles were illuminated with monochromatic light of an incident wavelength at which maximum scattering of incident light occurs and the resulting scattered light was either completely or partially polarized, depending on the size of the particle.

TABLE 10

MEASURED RELATIVE SIGNAL GENERATING POWER OF FLUORESCEIN VERSUS GOLD PARTICLES WHEN ILLUMINATED WITH WHITE LIGHT[a]

SIGNAL SOURCE	PARTICLE DIAMETER (nm)	[c]MEASURED NUMBER OF FLUORESCEIN MOLECULES NECESSARY TO MATCH THE LIGHT INTENSITY FROM ONE GOLD PARTICLE
GOLD	10.1± < 1.2	~9
GOLD	19.8± < 2	~2.3×10^2
GOLD	29.5± < 3.5	~4.1×10^3
GOLD	39.9± < 6	~2×10^4
GOLD	49.2± < 9.8	~7.3×10^4
GOLD	59.6± < 11.9	~2.3×10^5
GOLD	76.4± < 15.3	~9×10^5
PST[b]	80± < 6.6	~8.3×10^3

(a) Incident light from an incandescent Leica microscope light source with a color temperature of 2,800° K, and the emitted light was passed through a glass lens to decrease the infrared component.

(b) PST - polystyrene

(c) Results not corrected.

TABLE 11

MEASUREMENT OF RELATIVE SIGNAL GENERATING POWER OF FLUORESCEIN VERSUS GOLD PARTICLES WHEN ILLUMINTED WITH MONOCHROMATIC LIGHT

PARTICLE TYPE	[a] PARTICLE DIAMETER	[b] INCIDENT WAVELENGTH	[c] MEASURED NUMBER OF FLUORESCEIN MOLECULES NECESSARY TO MATCH THE LIGHT SIGNAL FROM ONE GOLD PARTICLE
GOLD	10.1nm	~525nm	~4
GOLD	19.8nm	~530nm	~120
GOLD	29.5nm	~532nm	~1,400
GOLD	39.9nm	~535nm	~7,800
GOLD	49.2nm	~550nm	~25,000
GOLD	59.6nm	~542nm	~78,000
GOLD	76.4nm	~565nm	~190,000
GOLD	~100nm	~560nm	~550,000

(a) Measurements represented size

(b) Incident monochromatic light composed of a significant fraction of horizontally polarized light arising from the monochrometer. Vertically polarized incident light would yield a significantly larger particle signal intensity but would not affect the signal intensity from fluorescein.

(c) Results not corrected.

Table 11 illustrates that the scattered light signal intensity generated from various sized individual gold particles illuminated with incident monochromatic light is much more intense relative to the light signal intensity from a single fluorescein molecule. These results further

illustrate that individual metal-like particles, as for example, gold particles when illuminated with incident monochromatic light can be detected to very low concentrations. Such detectabilty is extremely useful for using such particles with appropriate detection methods as extremely sensitive light scattering labels in diagnostic assays and analytical applications.

Non-metal-like particles as for example, polystyrene which have 100's-1000's of highly fluorescent molecules incorporated into the body of the particle are well known in the art. An example of such a particle is a 110nm diameter particle into which has been incorporated a fluorescent compound which has excitation and emission wavelength maxima, 490nm and 515nm respectively, which are similar to fluorescein. Each particle contains an average of 4,400 highly fluorescent molecules and the volume of such a particle is about $7x10^{-16}cm^3$ and the fluorescent molecule concentration in the particle is about $10^{-5}M$. Table 12 presents the experimentally measured results for the light generation power of 110 nm diameter polystyrene, polystyrene particles loaded with many molecules of highly fluorescent compound, and 100 nm diameter gold particles. The light generating power of these are directly compared against a solution of fluorescein which gives the same light generation power. It is interesting to note that the total scattered light signal from the 110nm polystyrene particle alone is equivalent to the light signal from about 12,000 fluorescein molecules. The presence of the fluorescent molecules in the polystyrene particle only increases the total light signal by about 1.5 fold of the particle. The fluorescence signal from this particle can be separated from the light scattering signal by using the proper filter between the sample and the detector which excludes incident light wavelengths and passes the wavelengths characteristic of the fluorescence emission. Such a filter results in this particle generating a fluorescent signal intensity equivalent to

about 3,000 fluorescein molecules. The 100nm diameter gold particle was clearly far superior in emitted light generation power as compared to these particles.

TABLE 12

MEASURED RELATIVE SIGNAL GENERATING POWER OF FLUORESCEIN, POLYSTYRENE PARTICLES, POLYSTYRENE PARTICLES CONTAINING FLUORESCENT MOLECULES AND GOLD PARTICLES

PARTICLE TYPE	[a] PARTICLE DIAMETER	FLUORESCENT MOLECULES PER PARTICLE	INCIDENT WAVELENGTH	[c] MEASURED NUMBER OF FLUORESCEIN MOLECULES NECESSARY TO MATCH THE LIGHT SIGNAL FROM ON PARTICLE
(d) PST	110nm	0	490nm	~12,000
PST	110nm	~4400	490nm	~19,000
(e) Gold	100nm	0	555nm (b)	[e] ~1.3×10^6

(a) Measurements represented size
(b) See Table 11(b)
(c) Results not corrected.
(d) Polystyrene
(e) The 100nm diameter particle used herein was from a different production batch than that used in Table 11.

Mixed Composition Particles

Spherical particles of mixed compositions were evaluated by theoretical and physical experimentation to assess their possible utility in various diagnostic and analytic applications. For theoretical evaluations, a gold "core" particle coated with different thickness of silver and a silver core particle coated with different thickness of either gold or polystyrene were studied. By "core" is meant a spherical particle upon which an additional layer or thickness of different light scattering material is placed, resulting in a mixed composition of certain proportions. Direct physical experimentation was done for particles composed of a mixed composition where an additional thickness of silver was added to a core gold

particle of 16nm diameter. In these illustrative examples, gold and silver are representative of metal-like materials and polystyrene is representative of non-metal-like materials. These examples are only a few of a larger number of different possible combinations which involve particles composed of mixtures of one or more different metal-like and/or non-metal-like materials.

Results from calculations for the light scattering properties of the above illustrative examples are presented in Tables 13 and 14. Table 13 section A shows that for series of spherical 10 nm diameter particles which are composed of increasing proportions of a silver coat on a gold core, the light scattering properties are changing to those more like a pure gold particle. Most importantly, we observed in these calculations and by physical experimentation, that certain proportions of a silver coated gold particle can exhibit two intense light scattering maxima at incident wavelengths close to those characteristic of pure gold and pure silver particles of this rough size.

Direct experimental observation of 16 nm diameter gold particles coated with silver using white light illumination under DLASLPD conditions with a simple light microscope showed that there were new colors of scattered light from these particles which had not been previously seen in pure gold or pure silver particle preparations. Many of the particles had a brilliant purple to magenta color of scattered light.

Table 13 section B presents a comparison of the calculated results for mixed composition particles composed of a 10nm diameter gold sphere coated with different thickness of silver. The results show similar trends as seen in Table 13 Section A for the light scattering properties of these mixed compositions as the proportion of silver to gold is varied. In additional calculations (Table 14), where a silver core particle is coated with varying gold thickness, the light scattering

properties show similar trends in their changes as the proportion of gold and silver is changed as seen in Table 13.

TABLE 13

CALCULATED SCATTERING PROPERTIES OF SPHERICAL PARTICLES COMPOSED OF MIXED COMPOSITION - A GOLD CORE COATED WITH SILVER

	PARTICLE DIAMETER	GOLD CORE DIAMETER	SILVER COAT THICKNESS	VOL GOLD TOTAL VOL	C_{sca} AT WAVELENGTH MAXIMUM (cm^2)	INCIDENT WAVELENGTH AT SCATTERING MAXIMUM(S)
A	10nm	20nm	0	1	1.26×10^{-15}	~530nm
	10nm	9nm	0.5	0.73	7.3×10^{-16} 8×10^{-16}	~340nm ~516nm
	10nm	8.4nm	0.8nm	0.59	9×10^{-16} 6.8×10^{-16}	~340nm ~510nm
	10nm	4nm	3nm	0.064	5.7×10^{-15}	~380nm
B	10nm	10nm	0	1	1.26×10^{-25}	~530nm
	11nm	10nm	0.5nm	0.75	1.25×10^{-15} 1.45×10^{-15}	~340nm ~518nm
	12nm	10nm	1nm	0.58	2.8×10^{-25} 2×10^{-15}	~340nm ~505nm
	20nm	10nm	5nm	0.125	2.4×10^{-13}	~375nm
Polystyrene Particle						
C	10nm	0	0	--	1.3×10^{-17}	~300nm
	20nm	0	0	--	8.3×10^{-16}	~300nm

We have determined from our combined theoretical and physical experimentation the following. For particles composed of certain mixed compositions of metal-like materials, as for example, mixed compositions of gold and silver, new light scattering properties appear which are useful in many different sample types and specific diagnostic and analytic applications. Particles with two or more optically distinct and resolvable wavelengths of

high scattering intensities can be made by varying the composition of the metal-like-materials.

In contrast, particles composed of mixed compositions of non-metal-like and metal-like materials generally exhibit light scattering properties similar to the metal-like materials at equal proportions or less of non-metal-like materials to metal-like materials. Only at very high proportions of non-metal-like to metal-like materials do the light scattering properties of the mixed composition particle resemble that of the non-metal-like material as the results of Table 14 section B indicate.

Both the mixed silver-gold compositions and the silver-polystyrene compositions exhibit the high light scattering power and visible wavelength scattering bands which are characteristic of particles composed of pure metal-like materials. Particles of certain mixed compositions are detectable by specifically detecting the scattered light from one or both of the scattering intensity peaks and or by the color or colors of these mixed composition type particles. Such mixed composition type particles enhances the capability for detecting lesser amounts of particles and more specifically, detecting lesser and greater amounts of particles than was previously possible.

Asymmetric Particles

The physical orientation of asymmetric particles with regard to a light beam allows for additional scattered light properties to be used in the detection of these particles. The property of RIFSLIW can be used in many different aspects of the current invention to more specifically and more sensitively detect and or measure one or more analytes or particles in a sample. For example, the flickering of the scattered light intensity and/or change in color provides additional detection means to determine which particles are bound to a surface and which particles are not. This allows for non-separation

type of assays (homogeneous) to be developed. All that is required is to detect by particle counting, intensity measurements or the like the particles that do not flicker and/or change color. Unbound particles in solution will flicker and/or change color while those bound to the surface will not. Additional image processing means such as video recorders and the like allow for additional methods of detection to be used with both asymmetric and spherical (symmetric particles). For example, In either a separation or non-separation format, the bound particles are detected by focusing the collecting lens at the surface and only recording those scattered light signals per unit area which are constant over some period of time. Particles free in solution undergoing brownian motion or other types of motion results in variable scattered light intensity per unit area per unit time for these particles. Bound light scattering particles are fixed in space and are not moving. By using image-processing methods to separate the "moving" light-scattering particles from the "bound" light scattering particles, the amount of bound particles is determined and correlated to the amount of analyte in the sample. One of skill in the art will recognize there are many other image processing methods that can be used to discriminate between bound particles to a surface and unbound spherical or asymmetric particles in solution.

Addition of Other Materials to the Surface or Core of the Particle to Provide Additional Physical Attributes not Related to the Light Scattering Properties

In certain applications and with the use of certain types of compositions, it may be useful to "coat" the surface of a particle to further chemically stabilize the particle, or to add additional surface binding attributes which can be very important in specific applications to analytical diagnostic assays. For example, it is well known that silver rapidly oxidizes. For use of silver

particles or particles of mixed composition which contain silver, one can chemically stabilize the silver-containing particle by applying a thin coat of gold or other substance on the surface such that the silver is no longer susceptible to environmental effects on it's chemical stability.

In another example, one may want to coat the surface with another material such as a polymer containing specifically bound binding agents, or other materials useful for attaching binding agents, or the binding agents themselves to the particles. In each of these examples, these "thin" coats do not significantly alter the light scattering properties of the core material. By "thin" coats is meant a monolayer or similar type of coating on the surface of the particle.

Manipulatable Light Scattering Particles (MLSP's) are particles which in addition to having one or more desirable light scattering properties, these particles can also be manipulated in one-, two- or three-dimensional space by application of an EMF. A MLSP particle can be made in many different ways. For example, a MLSP particle is made by coating a small diameter "core" ferroelectric, magnetic or similar material with a much greater proportion of a material that has the desirable light scattering properties, for example a 10nm diameter core of magnetic or ferroelectric material is coated with enough gold to make a 50, 70, or 100 nm diameter particle. This is shown in Figure 29 A.

Another method of making such a particle is to coat the material that has the desirable light scattering properties with a thin coat of the magnetic or ferroelectric material. For example, a gold or silver particle of about 50 nm is coated with a 1-2 nm thick coat of the magnetic or ferroelectric material. This is shown in Figure 29 B.

Alternatively, the MLSP particles are made by mixing in the appropriate proportions the light scattering

desirable materials and the ferroelectric or magnetic materials such that as the particle is formed, the appropriate proportions of light scattering desirable material to magnetic or ferroelectric material per particle ratio is attained. This is shown in Figure 29 C.

An alternative to the above MLSP particles is to link or assemble one or more types of particles with desirable light scattering properties to one or more particles that can be moved by a EMF. Such multi-particle structures can then have similar properties to the MLSP's. For example, small particles of magnetic or ferroelectric material are linked to one or more particles who's light scattering properties are detected. The linking is by ionic, chemical or any other means that results in a stable multi-particle structure. For example, the different particles are coated with appropriate polymers so that when mixed in the proper portion, a defined distribution of discreet multi-particle structures are achieved by crosslinking the different types of individual particles together. There many different ways to link the particles together to achieve the desired multi-particle structure(s). For illustrative purposes, a few of the possible multi-particle structures are shown in Figure 30. Figures 30 A, B, and C show dimer, tetramer, and higher order particle constructs respectively for orientable MLSP particles. One skilled in the art will recognize that these are just a few of the many different types of multi-particle structures possible and there are numerous methods to make such structures.

These examples of particles composed of mixtures of one or more material are but a few of a very large number of different compositions of different materials which are possible, and which would be apparent to one of skill in the art.

Particle Size and Shape Homogeneity

Depending on how the light scattering properties of particles are detected, the approximate size and distribution of particle sizes in the particle population can be extremely important. As an example, many of the commercially available gold particle preparations quote the particle size distributions anywhere from about <10 to about <20 percent coefficient of variation. Percent coefficient of variation is defined as the standard deviation of the particle size distribution divided by the mean of the particle preparation. Thus, for a 60nm particle preparation with a coefficient of variation of 20%, one standard deviation unit is about ±12nm. This means that about 10% of the particles are smaller than 48nm or greater than 72nm. Such variation in size has significant effects on the intensity of scattered light and the color of scattered light depending on the approximate "mean" size of the particles in the preparation.

We have developed a particle growing procedure which seems to give narrower size distributions than those available commercially. The procedure involves first making a preparation of "seed" gold particles which is then followed by taking the "seed" particle preparation and "growing" different size gold (see examples 11 and 15) or silver particles (see Example 13) by chemical methods. For example, 16nm diameter gold particles are used as the "seed" particle and larger diameter gold particles are made by adding the appropriate reagents (see Example 15). This method is also very useful for making mixed composition particles.

Particle Homogeneity - Detection of Analytes by Scattered Light Color of Individual Particles

In certain applications, the color of the individual particles are used to identify and quantitate specific types of analytes. For example, in image cytometry

applications, it may be of interest to identify and count different types of cell surface antigens or the like by detecting the number and color of different types of particles attached to the surface. For this or any other related type of multi-analyte detection, the size distributions of the different particles need to be kept as tight as possible. The average particle diameter of the particle preparation should be chosen to provide the desired color of scattered light under white light illumination, using an average or "mean" particle size that is as close to the size midpoint between the mean particle sizes of smaller and larger particles which will be used in the same application to produce different colors of scattered light. In this fashion, the resolvability of the different types of particles by their color of scattered light is maximized.

Particle Homogeneity-Integrated Light Intensity Measurement

In other sections we have described how the intensity of scattered light can vary greatly as particle size is increased or decreased. This variation in the intensity must be taken into consideration especially when integrated light intensity measurements are being performed. Using the 60nm particle preparation described above with a 20% coefficient of variation, this means that 10% of the particles have intensities about 3 times greater or less than a 60 nm particle. In addition, the particles within the remaining 90% of the population have quite varying intensities. In applications where there are many particles being measured, the "average" integrated light intensity should approximate a 60nm particle. However, at lower concentrations of particles, the statistics of such a variation may affect the accuracy of the reading from sample to sample, and correction algorithms may be needed. By using the narrowest

distribution of particles possible, the accuracy and ease of measurement is enhanced.

Useful Metal-like Particles for Detection of Analytes by their Light Absorption Color

For some types of analyte assays, analytes are at concentrations where detection of the analytes by the light absorption properties can be accomplished. For example, a current problem in the art of immunochromatographic assays and the like is that the use of gold particles of the sizes typically used (4 to 50 nm diameter) only provides for particles that can not be optically resolved by their light absorption color. These particles have a pink to red color when observed on filter paper or similar diagnostic assay solid-phase media. By varying the size and/or shape of silver particles and other metal-like particles many different colors of light absorption can be achieved. These different colors of the particles by light absorption can be used to detect different analytes by the light absorption color of a particle. These colors which can be detected by the eye are very useful in many types of solid-phase assays such as immunochromatographic flow assays, panel type, and microarray or larger solid-phase single or multi-analyte assays. Spherical and asymmetrical particles of silver and certain mixed compositions of other metal-like particles allow for a wide range of colors by light absorption.

Autometallographic Enhancement of Light Scattering Properties of Particles

It is well known in the art that autometallography and related techniques can be used to enlarge the size of existing metal-like particles by small or large factors. The light absorbing power of particles composed of metal and/or semiconductor material, and in particular, gold and silver particles has often been used to quantitate and or detect the presence of these particles, by using either

the eye or an instrument designed to measure light absorbance. Such a method is inferior to the light scattering detection methods of the present invention in its ability to detect small numbers of particles enlarged by metallography.

As an example, it has been reported (see Immunogold-Silver Staining, Principles, Methods and Applications, CRC Press, 1995 M.A. Hayat Ed.) that one nanometer diameter gold particles were enlarged by metallographic methods, coating the 1nm diameter gold particles with silver to an average diameter of about 110nm in about twenty minutes. The particles in this preparation ranged in size from about 40nm to 200nm diameter and were roughly spherical in shape. Surprisingly, our calculations show that enlarging the diameter of the 1nm core tracer particle to 110nm results in an increase in scattering power of roughly 10^{10} while the light absorption power is only increased by roughly 10^{5}.

By increasing the diameter of a small particle, the incident wavelength at which maximum light scattering occurs shifts to much longer wavelengths, as compared to a small core particle of the same material. Thus, enlarged particles are easily detected in the presence or absence of small 1nm particles by measuring the light scattering signal from the enlarged particles. The utilization for detection of the enlarged particles of incident light of the wavelength at which maximum scattering occurs for the enlarged particles allows the more specific detection of the enlarged particles relative to the smaller particles which constitute the major source of non-specific light scattering background.

TABLE 14

CALCULATED SCATTERING PROPERTIES OF SPHERICAL MIXED COMPOSITION PARTICLES - SILVER CORE PARTICLE COATED WITH GOLD OR POLYSTYRENE (PST)

	PARTICLE DIAMETER	SILVER CORE DIAMETER	COAT COMPOSITION	COAT THICKNESS	VOL SILVER TOTAL VOL	C_{sca} AT WAVELENGTH MAXIMA (cm^2)	INCIDENT WAVELENGTH AT SCATTERING MAXIMA
A	10nm	10nm	--	0	--	$1.1x10^{-14}$	~384nm
	14nm	10nm	Gold	2nm	0.36	$4.5x10^{-15}$	~372nm
	20nm	10nm	Gold	5nm	0.125	$6.5x10^{-15}$	~525nm
B	10nm	10nm	--	0	1	$1.1x10^{-14}$	~384nm
	10nm	9nm	Gold	0.5nm	0.73	$1.9x10^{-15}$	~384nm
	10nm	7nm	Gold	1.5nm	0.34	$5.6x10^{-16}$ $6.8x10^{-16}$	~300nm ~ 520nm
C	(a)10nm PST	0	--	0	--	$1.3x10^{-17}$	~300nm
	10nm	10nm	--	0	0	$1.1x10^{-14}$	~384nm
	(a)20nm PST	0	--	0	--	$8.3x10^{-16}$	~300nm
	20nm	20nm	--	0	1	$7x10^{-13}$	~382nm
	40nm	20nm	PST	10nm	0.125	$9.3x10^{-13}$	~412nm
	60nm	20nm	PST	20nm	0.037	$1.25x10^{-12}$	~418nm
	20nm	12nm	PST	4nm	0.216	$4x10^{-14}$	~408nm
	20nm	10nm	PST	5nm	0.125	$1.4x10^{-14}$	~410nm
	20nm	8nm	PST	6nm	0.064	$4.3x10^{-13}$	~414nm

(a) Particle composed of polystyrene only

Table 15 provides additional data with regard to the light scattering properties of small particles that are increased in size by metallographic or related methods. Our calculated data show that enlarging the size of a particle results in a greater increase in the particle's light scattering power as compared to it's light absorption power. For example, a twenty percent increase in particle diameter increases the small particle light

scattering power by $(1.2)^6$ or about three-fold. An increase of two and ten-fold in small particle diameter will result in a scattering power increase of about sixty-four and one million-fold respectively, while the light absorbing power is increased by only eight-fold and one thousand-fold respectively.

Thus, when the method of the present invention is used to quantitate and or detect the presence of particles which have been enlarged by metallography (that is, the deposition of a coat of a metal-like material onto a small diameter composed of either metal-like or non-metal-like materials), it is possible to detect lesser amounts of such particles, and to more specifically detect lesser amounts of such particles, than was previously possible.

The above is an illustration of the combination of metallographic enlargement of a metal-like particle core by the deposition of a metal-like coat on the core, followed by the detection of the enlarged particles by the method of the present invention. Such methods can be used for enlarging particles free in solution and or particles attached to a surface. The preceding example is only one of the many different permutations of this combination method which include the use of the many different strategies and methods discussed herein for detecting particles by light scattering as well as different combinations of core and coat compositions and different degrees of enlargement. These alternate combinations will be readily apparent to one of skill in the art. Such a combination approach can be applied to almost any situation where it is desirable to use a signal generation and detection system to detect an analyte.

TABLE 15

2nm DIAMETER GOLD TRACER CORE PARTICLE - DIFFERENT THICKNESS OF A SILVER COAT OF UNIFORM THICKNESS - CALCULATED LIGHT SCATTERING PROPERTIES

DIAMETER OF PARTICLE	THICKNESSOF SILVER COAT	PARTICLE LIGHT SCATTERING POWER (C_{sca}) (cm^2)	WAVELENGTH AT WHICH SCATTERING MAXIMUM OCCURS	PARTICLES RELATIVE SCATTERED LIGHT INTENSITY	RELATIVE DECADIC MOLAR EXTINCTION COEFFICIENT OF PARTICLE
2nm	0nm	$\sim 8\times10^{-20}$	~520nm	1	1
10nm	4nm	$\sim 10^{-14}$	~382nm	$\sim 1.25\times10^{5}$	$\sim 3\times10^{2}$ (a)
20nm	9nm	$\sim 6.5\times10^{-13}$	~384nm	$\sim 8.1\times10^{6}$	$\sim 2.3\times10^{3}$
40nm	19nm	$\sim 2.8\times10^{-11}$	~400nm	$\sim 3.5\times10^{8}$	$\sim 1.6\times10^{4}$
80nm	39nm	$\sim 2.9\times10^{-10}$	~447nm	$\sim 3.6\times10^{9}$	$\sim 5.8\times10^{4}$
100nm	49nm	$\sim 4.3\times10^{-10}$	~481nm	$\sim 5.4\times10^{9}$	$\sim 7.6\times10^{4}$
150nm	74nm	$\sim 7.9\times10^{-10}$	~432nm	$\sim 9.9\times10^{9}$	$\sim 1.5\times10^{5}$
150nm	74nm	$\sim 7.6\times10^{-10}$	~600nm	$\sim 9.5\times10^{9}$	$\sim 1.2\times10^{5}$

(a) The molar decadic extinction coefficient, the C_{sca} and the incident wavelength at which maximum light scattering occurs are essentially identical to those for a pure 10nm diameter silver particle.

Method of Refractive Index Enhancement

The use of refractive index matching techniques in light microscopy, telecommunications, and other related fields is well known in the art. This technique is generally used to decrease the nonspecific light scattering and reflections that occur as a light beam passes from one medium or device to the other as for example, from the surface of one material to the surface of another different material.

We have determined that the light scattering power (C_{sca}) of a specific type of particle is affected by the medium in which the particle resides. Altering the refractive index of the medium results in a change in a particle's light scattering properties.

Table 16 provides an illustrative example of medium refractive index effects on selected particles. Calculated refractive index medium effects for gold, silver, and polystyrene spherical particles of 10nm diameter are presented.

The effects of the refractive index of the medium are quite different for metal-like particles as compared to non-metal-like particles as Table 16 shows. Increasing the refractive index of the medium for metal-like particles as for example gold, results in increasing the intensity and wavelength maximum of the light scattered from the particle while for a non-metal-like particle, as for example polystyrene, the light scattering power is decreased.

The unique light scattering properties of metal-like particles as compared to non-metal-like particles as an effect of the refractive index of the sample medium can be used to more specifically and with greater sensitivity detect metal-like particles in samples including those which have high non-specific light scattering backgrounds. This is important for many different types of diagnostic analytical assays.

TABLE 16

CALCULATED MEDIUM REFRACTIVE INDEX EFFECTS FOR TEN NANOMETER DIAMETER PARTICLES OF DIFFERENT COMPOSITION. WAVELENGTH AND INTENSITY EFFECTS

GOLD			SILVER		POLYSTYRENE	
N_1	(A)	(B)	(A)	(B)	(A)	(B)
1	1	520nm	1	355nm	1	400nm
1.1	1.9	525nm	1.6	360nm	0.9	400nm
1.2	3.9	525nm	2.3	370nm	0.75	400nm
1.3	7.7	530nm	2.9	380nm	0.52	400nm
1.4	15.1	535nm	3.9	390nm	0.27	400nm
1.5	27.7	540nm	5.3	400nm	0.084	400nm
1.6	45.4	550nm	7.3	415nm	~0	--
1.7	71.5	555nm	9.7	425nm	0.1	400nm

(A) = Relative scattering power at different medium refractive indices

(B) = Wavelength at which scattering maximum occurs

N_1 = refractive index of medium

In many types of samples and diagnostic assay formats, the problem of non-specific scattered, reflected, and other background light from sample containers and non-analyte sample constituents are well known. These non-specific light backgrounds make it difficult, if not impossible to perform sensitive to ultrasensitive detection of analytes by detection and/or measurement of the scattered light properties of a particle.

We have determined that metal-like particles can be detected to much greater specificity and sensitivity as compared to non-metal-like particles when the method of refractive index enhancement is used. The method is now

described. The effect of the refractive index of the particle and medium on the scattered light intensity can be evaluated by using the following expression (RI is refractive index factor)

$$RI = refmed^4 \left| \frac{m^1 - 1}{m^2 + 2} \right|^2 \qquad (16)$$

where refmed is the refractive index of the medium, and m is equal to the refractive index of the particle/refmed. m depends implicitly on wavelength but the exact dependence varies with particle composition and medium. The refractive index of most solvents which have no color is usually independent of wavelength, at least in the visible region of the spectrum.

It is of interest for the use of light scattering particles in sensitive assays to determine which values of refractive index leads to higher light scattering intensities. This is determined from the refractive index factor (RI) of Eq.(16). This factor has it's highest value when the denominator of Eq.(16) is zero. For this condition, the refractive index factor has an infinite value. Thus, the condition for high light scattering is

$$m^2+2=0 \qquad (17)$$

Solving above equation for m, we get

$$m = \sqrt{-2} \qquad (18)$$

$$= 1.41i \qquad (19)$$

where $i = \sqrt{-1}$. The above equation indicates that the refractive index factor has its highest value and light scattering from the particle is at a maximum when the refractive index is a pure imaginary number with a value of 1.41. The calculated data presented in Table 16 do follow the expected trends. In addition, the method of refractive index enhancement works very well at incident wavelengths far removed from the incident wavelength at which maximum light scattering occurs for the metal-like particles.

An illustrative example of the use of the refractive index enhancement method is now provided. In highly scattering samples, such as samples where there is a high level of non-specific light scattering background, metal-like particles and the method of refractive index enhancement are used as follows.

One skilled in the art increases the refractive index of the sample medium as for example, placing a film of water or other liquid on top of a dry or wet sample. This increases the refractive index of the medium. In another example, a serum or other type of highly scattering sample is diluted with a high refractive index liquid which substantially increases the refractive index of the medium.

For the above mentioned examples, the following processes occur. The specific light scattering signal of the metal-like particles increases and the non-specific light scattering background decreases as the refractive index of the sample is increased. The largest increases in particle light scattering/non-specific scatter background ratio is achieved when the refractive index of the sample medium approaches the refractive index of the metal-like particle as demonstrated in Table 16. This means that at the proper medium refractive index values, the non-specific light scatter from serum proteins or similar constituents can be significantly reduced or eliminated while the specific light scattering intensity of the particles is increased. This results in superior analyte detection signal/background ratios when the light scattering properties of metal-like particles are used as the analytical tracer. These methods can be applied to samples such as dry surfaces, surfaces covered by solutions, or solutions.

These index matching methods can also be used with longer wavelengths for the metal-like particles to increase the specific light scatter signal/non-specific scatter background ratio even further. While Table 16 only

shows the effects for gold and silver particles, particles composed of other metal-like materials can also be used to detect lesser amounts of particles using the methods we have described. The description of the method of refractive index enhancement described herein presents only a few of many possible variations of this practice of the invention. Many other variations of the method are possible and will be apparent to one of skill in the art. One or another of these variations can be effectively utilized in most diagnostic formats to determine the presence or absence of an analyte. This aspect of the present invention provides a means for the detection of lesser amounts of particles, and for the more specific detection of lesser amounts of particles, and for the more specific detection of lesser and greater amounts of particles than was previously possible.

A method of the present invention which combines refractive index enhancement with the narrow band pass filter approach described earlier has great utility for detection of lesser and greater amounts of particles than was previously possible. These approaches are complimentary. The refractive index enhancement method is used to decrease non-specific light scattering background while the narrow filter is used to reduce and minimize other sources of non-specific light background such as fluorescence and the like. A combination of these methods results in highly optimized particle specific scattering signal/non-specific light background ratios and allows for the more specific and more sensitive detection of particles.

The above example is only one of many possible variations of this combined method. Other variations will be apparent to one of skill in the art.

Detection of Light Scattering Particles in Highly Scattering and Fluorescent Samples - Serum

Mammalian serum contains many medically important substances whose quantitation and or presence is determined in the clinical laboratory as well as elsewhere. Many different signal generation and detection systems are used to determine the presence of these analytes in serum and these include light signal generation methods such as fluorescence, light scattering, and chemiluminescence, as well as colorimetric methods which are used in formats involving both direct labeling and signal amplification methods. Natural serum contains a variety of substances which are capable of producing a non-specific light signal by either fluorescent, chemiluminescent or light scattering mechanisms. In addition, the serum often contains substances which interfere with the generation and or detection of the specific light signal from the tracer entity. These difficulties make it difficult if not impossible to conduct analyte detection in pure or nearly pure serum samples.

In order to be able to effectively employ most, if not all, of the existing test systems for detection of serum analytes, it is almost always necessary to pre-process the serum in some way to make it suitable for testing. Many such serum processing methods exist and perhaps the simplest is dilution of the serum into some appropriate solution which is usually aqueous in nature. Another commonly used approach is to conduct the actual test in such a manner that the undesirable serum components are removed before the presence of the specific light producing tracer is determined. From a cost and labor point of view, the less effort and reagents needed to conduct the test, the better. It is highly desirable not to pre-process the sample at all. Such capability could also be beneficial to the performance of the test. The method of the present invention provides a means to

conduct such analyte tests in almost pure serum and further provides a means, to detect lesser or greater amounts of particles more specifically in high concentrations of serum than was previously possible.

For example, it is common to dilute serum samples to a final concentration of about 5 percent serum before analyzing it with a fluorescent tracer such as fluorescein. The serum sample is illuminated with monochromatic light at 490nm, and optical filters are used to minimize non-specific scattered light background. The non-specific light signal is equivalent to a highly pure liquid sample of fluorescein which contains 10^{-8}M to 10^{-9}M fluorescein. Thus, in the 5% serum sample, one can detect 10^{-8}M to 10^{-9}M fluorescein at a signal to noise ratio of 2. In a 95 percent serum sample, the lower limit of detection of fluorescein would be about 19 times higher, or about 1.9×10^{-7} M to 1.9×10^{-8} M. Thus, in 95 percent serum with optical filters, the lower limit of detection of fluorescein is about 1.9×10^{-7} M to 1.9×10^{-8} M and this amount of fluorescein light signal results in a (total light signal) to (non-specific light signal) ratio of about 2 to 1.

Table 17 presents the experimentally measurable detection limits, at a (total light signal) to (non-specific light signal) ratio of 2 to 1, of fluorescein at very high serum concentration. At this high serum concentration, and in the absence of optical filtration to remove non-specific light signal due to scattered incident light, the lower limit of detection of fluorescein is about 6×10^{-7}M.

Table 18 presents the lower limit of detection of fluorescein at very high serum concentration when an optical filter which eliminates the non-specific light signal due to scattering of incident light is placed between the sample and the photomultiplier tube. In this situation, the lower limit of detection of fluorescein in high serum concentration is about 2×10^{-8}M.

In contrast to fluorescein, the results presented in Table 17 Section B and Table 18 demonstrate that in the absence of optical filtration the presence of 59.6nm diameter gold particles in 95 percent serum can be detected with a (total light signal) to (non-specific light signal) ratio of 2 to 1 at a concentration of about $1.8x10^{-12}$ M. The non-specific light signal observed from the serum was equivalent to that from about $5x10^{-7}$ M fluorescein. Under these same conditions 60nm diameter polystyrene particles in high serum concentration can only be detected at a lower limit of about $6x10^{-9}$M (see Table 18).

TABLE 17

DETECTION OF 59.6nm DIAMETER GOLD PARTICLES AT HIGH SERUM CONCENTRATION

	PERCENT SERUM	GOLD PARTICLE CONCENTRATION	INCIDENT WAVELENGTH	FLUORESCEIN CONCENTRATION	RELATIVE LIGHT INTENSITY
A	97.8%	0	490nm	0	102
	95.7%	0	490nm	$8.7X10^{-6}$M	15.8
	97.8%	0	545nm	0	0.52
	95.7%	0	545nm	$8.7X10^{-6}$M	0.56
B	97.8%	0	490nm	0	0[(c)]
	95.8%	$1.77X10^{-12}$M	40nm	0	1.1
	97.8%	0	543nm	0	0.55
	95.8%	$1.77X10^{-12}$M	543nm	0	105

Fetal bovine serum purchased from Biowhitaker, Walkerville, MD catalog number 14-501F. Serum was passed through a one micron filter before sale and was clear but straw colored. Serum pH was adjusted to about ph9 to 9.5. No wavelength filtration of the emitted light was done

(a) The light signal obtained here represents a value of 1. This signal was equivalent to $3.7x10^{-7}$M fluorescein.

TABLE 18

LOWER LIMIT OF DETECTION OF FLUORESCEIN, GOLD AND POLYSTYRENE PARTICLES AT 92.8% SERUM CONCENTRATION

PERCENT SERUM	FLUORESCEIN CONCENTRATION	TYPE AND CONCENTRATION OF PARTICLES	INCIDENT WAVELENGTH	FILTER FOR EMISSION	RELATIVE[c] SIGNAL INTENSITY
92.8%	0	0	554nm	NO	1 (~520mv)[b]
92.8%	0	Gold $1.8x10^{-12}$M	554nm	NO	2.1 (~1100mv)
92.8%	0	0	496nm	YES[a]	1 (~39mv)
92.8%	$2.3X10^{-9}$M	0	496nm	YES	~2 (~79mv)
92.8%	0	0	554nm	NO	1 (~468mv)
92.8%	0	PST $6X10^{-9}$M	554nm	NO	~2 (~960mv)

Polystyrene (PST) and gold particles had a measured diameter of 60nm and 59.6nm respectively. The pH of the solutions containing fluorescein was adjusted to pH 9-10 before measurement. The maximum light intensity in serum was observed at an incident wavelength of about 496nm for fluorescein and about 554nm for the 59.6nm gold particle.

(a) The light signal emitted from the sample was passed through a No. 16 Wratten filter before encountering the photomultiplier tube.

(b) The instrument measurements were all obtained at identical instrument settings and are directly comparable to one another (mv = millivolts).

(c) The light detection instrument detects fluorescein emissions slightly more efficiently than the particle emitted light. In addition the incident monochromatic light is enriched for horizontally polarized light and this reduces the particle results due to a lower level of light scattering from the particles but does not affect the fluorescein light intensity. The total instrument bias towards the fluorescein signal is roughly 1.5-2 fold.

Results presented in Table 19 present a comparison of the relative detection limits (at total light signal to non-specific light signal ratio of about 2 to 1) of 100nm diameter gold particles, 110nm diameter polystyrene particles, and 110nm diameter polystyrene particles containing 4,400 molecules of highly fluorescent compound per particle, in 95.7 percent serum. These results further demonstrate that the 100nm diameter gold particles can be detected at a much lower concentration than 110nm diameter particles composed of polystyrene or polystyrene containing many molecules of highly fluorescent compound. The gold particles can be detected in serum at about 230

times lower concentration than the other non-metal-like particles.

Table 20 compares the amount of scattered light measured from identical concentrations of 59.6nm gold particles in a solution containing a high concentration of serum and a solution containing only water under the same illumination conditions. Under these conditions, a gold particle concentration of 1.8×10^{-12}M was detectable at a signal/background ratio of about 3. These results indicate that the presence of serum or any of the common constituents does not appear to have any direct effect on the light scattering power of the gold particles. Such stability and inertness of the light scattering properties of metal-like particles make them extremely useful in samples such as serum and other related samples which contain many other constituents.

The detection of 100nm diameter spherical polystyrene or gold particles in serum provides a further illustrative example.

Mammalian serum contains around 3.7 gram percent of protein, of which about two-thirds is serum albumin. The detection of polystyrene particles in serum is hampered by the non-specific light scattering which originates from protein and other substances in serum, as well as many other sources. The similarity of the light scattering intensity versus the incident visible wavelength profiles for polystyrene particles and the proteins and other substances in serum severely limits the ability to detect the polystyrene particles in serum or any other highly scattering medium.

TABLE 19

DETECTION OF PARTICLES COMPOSED OF GOLD, POLYSTYRENE (PST), AND POLYSTYRENE CONTAINING A FLOURESCENT COMPOUND AT HIGH SERUM CONCENTRATION

PERCENT SERUM	PARTICLE DIAMETER AND COMPOSITION	PARTICLE MOLARITY	INCIDENT WAVELENGTH	RELATIVE LIGHT INTENSITY[d]
100%	0	0	490nm 580nm	1[e] 0.27
95.7%	110nmPST[b]	$1.9x10^{-11}M$	490nm 580nm	1.9 0.54
95.7%	110nmPST[a] +fluor	$1.9x10^{-11}M$	490nm 580nm	2.2 0.54
95.7%	100nm gold[c]	$8.2x10^{-14}M$	496nm 580nm	1.1 0.59

(a) Obtained from Interfacial Dynamics Corp., Portland, Oregon. Each particle contains an average about 4400 fluorescent molecules. The excitation and emission maxima are 490nm and 515Nnm respectively for the fluorescent molecules. The fluorescent compound concentration in the particle is about $3x10^{-2}M$.

(b) Obtained from Interfacial Dynamics Corp.

(c) Produced by art known methods.

(d) No wavelength filtration of emitted light was done.

(e) The light signal observed here represent a value of one. All other values are relative to this value. This signal is equivalent to that from about $2x10^{-7}M$ fluorescein.

TABLE 20

SIGNAL GENERATION FROM 59.6nm DIAMETER GOLD PARTICLES AT HIGH SERUM CONCENTRATION AND AT ZERO SERUM CONCENTRATION

PERCENT SERUM	GOLD PARTICLE CONCENTRATION	INCIDENT WAVELENGTH	RELATIVE TOTAL SIGNAL SIZE
0	$1.8X10^{-12}M$	543nm	1
95.7%	0	543nm	0.58
95.7%	$1.8X10^{-12}M$	543nm	1.38

Gold particles had a measured diameter of 59.6nm respectively. 95.7% serum is straw colored and has an optical density at 1cm pathlength and wavelength of 543nm of about 0.14, The limit scattering measurements were made in a 6mm by 50mm glass tube with an inner diameter of about 5mm. It is estimated that absorbance of light by the serum reduces the scattered light signal by about 15 percent.

The use of an incident wavelength of 575nm instead of 300nm to illuminate the serum results in an about 13 fold reduction in the non-specific light scattering signal, but also results in about the same extent of reduction for the specific scattering signal from the polystyrene particles. Increasing the wavelength of illumination for polystyrene or other non-metal-like particles does not appear to significantly increase the specific signal to background ratio, that is, the detectability of the polystyrene particles in the sample.

In contrast, metal-like particles are detected to greater signal to background ratios as compared to non-metal-like particles by increasing the visible wavelength of illumination and/or detection. A 100 nm diameter gold particle maximally scatters light around wavelengths of about 575nm in aqueous media similar to water.

Illumination of the sample with monochromatic light of wavelengths around 575nm results in the generation of the maximum light scattering signal from the gold particles and significantly reduces the non-specific light scattering signal. For example, under these conditions, the total non-specific light scattering is reduced by about thirteen-fold as compared to an illumination wavelength of 300nm relative to an incident wavelength of 300nm.

The illumination of the serum sample with incident white light and appropriate optical filters which minimize the amount of light outside of the wavelengths of interest (less than and or greater than a specified band centered at about 575nm) provides another means to detect lesser amounts of metal-like particles in serum. Under these conditions the incident visible wavelength which produces the maximum light scattering intensity from the gold particle is utilized, and the non-specific light scattering signal originating from serum protein and other substances as well as other sources is greatly reduced. Multiple types of different metal-like particles are detectable in serum samples when illuminated by white light (or several different wavelengths) and using an appropriate array of optical filters. This method makes use of each type of particle having a different incident wavelength at which maximum light scattering occurs.

Another approach involves filtering the total light signal from the sample through a proper polarization filter and/or a bandpass filter. Use of the proper polarization filter will result in the effective removal of unpolarized fluorescence background but will have little effect on the non-specific light-scattering background since it is mostly polarized. When using broad band illumination, as for example, white light illumination, using optical band pass filters of higher wavelength allows for significant reduction of the non-specific light scattering and fluorescence background.

Many of the metal-like particles have high light scattering intensities at longer wavelengths and this property can be utilized in combination with the bandpass filter and/or polarization filter approach. For example, a spherical gold particle of 300nm diameter has near maximum scattering efficiency at a wavelength of about 700nm and it's scattered light intensity is about six times than a 100nm diameter gold particle. Using the 300nm particle and a bandpass filter centered at 700nm decreases the non-specific light by half and increases the gold particle scattering power by a factor of 6 (as compared to the 100nm particle). Thus, the signal to background ratio in this system has been increased by a factor of 12. Use of this approach with non-metal particles, for example, polystyrene of comparable size, does not significantly increase the signal to background ratio but may actually lower it. The use of anti-reflective coating on the optical components of the apparatus, and/or sample chamber may also improve the signal to background ratio. Many other schemes and approaches are also possible and these would be apparent to one of skill in the art.

This aspect of the invention results in improved discrimination between the specific light scattering signal and the non-specific light scattering background signal of a diagnostic assay system over that attainable by other methods which use the detection of scattered light as part of a test system format. In addition, the availability of different types of metal-like particles which exhibit different colors when illuminated by white light makes it possible to detect the presence of multiple types of particles in one sample, which has utility for detecting multiple analyte types in one sample.

A further advantage of particles of metal-like particles is the chemical inertness of these particles, relative to fluorescent compounds. Such particles do not photobleach and their signal generation capacity is not affected by such mechanisms.

The above approaches are just a few examples of the many possible approaches for using particles composed of metal-like materials to improve discrimination between the specific light scattering signal due to the particle, and the non-specific light scattering signal which can originate from a variety of sources. For example, a large number of schemes are possible in which such particles are specifically detected at a wavelength different from the wavelength at which maximum light scattering occurs for the particle being used. Many other schemes or approaches are also possible and these would be apparent to one of skill in the art.

The detection of one or more analytes in a solid-phase or related sample by detection of one or more of a light scattering particle's properties is now discussed.

Solid-Phase Detection Methods

In the previous sections we have described various aspects of the invention as they relate to certain light scattering properties of metal-like particles, and the detection of these particles in a solution. We now describe our methods for detection of particles that are on a surface or very close to a surface.

We have determined that by using gold, silver, and other metal-like particles with our methods of DLASLPD illumination and detection, we are able to detect very low quantities of particles and particle-labeled binding agents (coated particles) per unit area, being able to detect single particles and particles coated with binding agents on or near a surface using simple illumination and detection means. These methods can be used on either optically transmissive or non-optically transmissive surfaces.

We have determined that with the use of certain combinations of particles and methods of illumination and detection, we can detect a wide range of particle densities from about 0.001 to 10^3 particles per square

micron (μ^2) in a sample. By using the proper type(s) of particles, different types of analytes can be detected to very low levels and across very wide concentration ranges in the same sample, as for example, in microarrays. This is accomplished on one apparatus by utilizing both particle counting (at low particle densities) and integrated light intensity measurements (at high densities) on the same sample. For example, if a sample is to be analyzed for two or more different analytes by using solid-phase related means such as array chips or other solid-phase methods, different types of analytes exist at different concentrations in the samples. That is, some analytes may be at higher or lower concentrations from a couple to a few orders of magnitude as compared to other analytes in the sample. The selection of the proper types of particles is extremely important in achieving the desired analyte detection sensitivity and range of concentrations the method will work for. Our methods as we describe herein provides for detection of analytes in such samples. Even wider detection ranges and greater sensitivities are possible if more powerful light sources such as lasers are used, and more sophisticated detection methods such as confocal imaging are added to our basic illumination and detection methods.

We have determined that we can detect high densities of particles more specifically and easily, that is, with very good signal to background ratios using simple methods. In some aspects of the present invention, a collection lens(imaging lens, mirror or similar device) is used and in other aspects, a collection lens is not used.

The scattered light from the particles is detected by a photodetector as for example, a photodiode or photodiode array, photomultiplier tube, camera, video camera or other CCD device, or the human eye. The amount of particles is determined by counting the number of particles per unit area and/or measuring the total integrated light intensity per unit area. The specific scattered light properties

detected and measured are one or more of the following: the scattered light intensity at one or more wavelengths, the color, the polarization, the RIFSLIW, and/or the angular dependence of the particle scattered light per unit area. This is then correlated to the presence, absence, and/or amount of the analyte(s) in the sample.

In some assays where one or more analytes is to be determined, one or both of the particle counting or integrated light intensity measurements can be used. It should be noted that with proper selection of particles and the use of DLASLPD illumination and detection methods, there is usually so much optically resolvable and detectable scattered light intensity available that more sophisticated light sources, and spatial and optical filtering techniques are not necessary. However, in some samples where there may be significant amounts of non-specific light background, the ultimate signal to background is improved by using optical filters, confocal imaging, or other aperture type spatial filtering techniques to increase the particle scattered light signal(s) to total non-specific light background ratio.

In some analytical and diagnostic applications, the scattered light intensity can be detected and measured using our basic methods without the use of a collection lens or mirror. In these samples, one or more properties of the scattered light is detected and measured in the same manner as described above without the use of a collection lens. The methods are now discussed in more detail.

<u>Detection of scattered light from light scattering particles using a collection lens or mirror</u>

We have found that we can use various types of light collection optical devices to collect the scattered light of the particles. We have used both particle counting and intensity measurements (integrated intensity per unit surface area) to detect one or more of the specific light

scattering properties of the particles in a given area with our methods of DLASLPD illumination and detection. We have found that in most of the experiments we have done, that it is generally useful to use the counting measurement method when the particle densities are about 0.1 particles per μ^2 are greater than about 0.1 particles per μ^2, we find that measuring the total integrated light intensity is a useful measurement method. It should be noted however, that one can use the counting measurement or integrated light intensity measurement methods at particle densities greater than or less than about 0.1 particles per μ^2.

The use of a specific type or types of lens to collect and/or image the scattered light from the sample we have found useful relates to the field or area of the surface we are interested in measuring, the type of sample container that is being measured, and the upper limit of particle densities that are to be measured by particle counting. For example, if we are interested in measuring larger areas to detect the scattered light, a x10 or even smaller microscope objective or lens or mirror can be used to collect the scattered light from the sample. Similarly, if a smaller areas of the sample is to be measured, a x20, x40, x100, or greater microscope objective lens or similar lens or mirror can be used to collect the light. If the method of particle counting is to be used at higher particle densities, greater power objective lenses allow for better resolution of the particles at high densities. It should be noted that when larger objectives are used, additional requirements and limitations come into play. For example, the working distance becomes very small and immersion oil may be needed to be added to the sample. When a camera, video camera, or similar CCD type photodetector is used, the total scattered light from the sample area is detected. This information can then be processed by simple hardware and/or software means to analyze the scattered light

measurements. This is a powerful capability, because many different analytes in a sample can be detected and quantitated by use of a solid-phase microarray, array chip, or similar format. In the microarray format, small areas of the surface are each covered by a different type of binding agent in a spatially distinct region that specifically binds a particular analyte. We describe later specific applications of the present invention to solid-phase multi-analyte microarrays and the like.

The method of particle counting is usually more instrumentally demanding than the method of integrated light intensity measurement. However, for very sensitive detection of one or more of the light scattering properties of a particle, there are many advantages to using the counting technique. For example, fluctuations and inhomogeneities in the light source or sample chamber do not effect the particle counting measurement whereas these problems can cause severe problems when the method of integrated light intensity measurement is used. In addition, there are many software and hardware options to enhance the quality and signal/background ratios of the measured particles by counting techniques.

Detection of light scattering particles without the use of a collection lens or mirror

We have also developed methods where the use of a collection lens is not necessary to detect the scattered light of the particles at or near a surface. In this arrangement we usually detect the scattered light coming from the area of interest by the integrated light intensity. This can be done by the naked eye, or a photodetector as previously described. We have found that when we use metal-like particles that are about 120nm in diameter or less, we can significantly increase the particle light scattering signal to total non-specific light background ratio by placing our detector (either eye

or photodetector) at angles outside of the envelope of the forward direction of the scattered light.

Key Concepts for Increasing Signal/Background Ratios

Before we describe the DLASLPD methods of illumination and detection in detail, it is useful to summarize the key concepts that when used in one form or another determine the signal and signal to background ratio limits for the detection of the light scattering particles. These methods are in addition to adjusting or changing various apparatus components such as using a more powerful light source, a more highly collimated light source, a smaller wavelength band light source, a different wavelength light source, a more sensitive photodetector, optical and/or spatial filters between the illumination source and the sample and/or between the sample and the detector, and/or confocal or similar imaging techniques. These strategies and methods are outlined below.

(1) by the use of larger diameter metal-like particles, the light scattering power of the particle can be significantly increased. The increase in size may also change the scattered light intensity versus incident wavelength profile. These properties can be adjusted to suit the need of any particular assay such that one or more of the scattered light properties are easily detectable. For example, for the measurement of analytes in samples with high non-specific light backgrounds, a larger gold particle, about 80-120nm or greater in diameter is useful. The maximum wavelength at which maximum light scattering occurs shifts to higher wavelengths and the intensity of the scattered light also increases as compared to a 40 nm diameter gold particle. The combination of these two effects significantly increases the signal/background ratio as compared to the 40nm diameter gold particle.

(2) by measuring the scattered light of the sample at angles outside the envelope of the forward direction of the scattered light, the signal/background ratio in either the intensity or counting mode is substantially increased. We have observed that the detector can be placed either above or below the surface plane of the sample as well as on the same side or opposite side of the sample plane where the illumination beam is located. In these various orientations, the specific light scattering signals from the particles are detected outside the envelope of the forward direction of the scattered light, while most of the non-specific scattered light from optical aberrations in the sample chamber and other constituents in the sample are within this envelope of the forward direction of scattered light. This allows for more sensitive and specific detection of the particle scattered light.

(3) the amount of non-specific reflected light also affects the sensitivity of detection as we have previously described. We have found that the amount of reflected light can be substantially reduced by moving the incident light surface as far as possible away from the collection area that is being detected. This can be accomplished in many different ways, including the proper design of the sample chamber (discussed later). For example, we noticed that if we put a thin layer of immersion oil on the bottom of a glass slide, through which the light beam illuminates the particles on the opposite surface, we saw highly improved results. In another experiment, when we glued a small plastic light guide to the bottom of a plastic sample chamber with a microarray of bound particles on the opposite side of the surface, we saw very improved results. We have also used much larger optical alignment means such as an equilateral prism and/or other types of prisms or optical light guides and placed immersion oil at the surface where the sample container interfaces with the prism. We have concluded that these superior results are

a result of (i) having the light incident surface removed a greater distance away from the area of detection that contains the light scattering particles; (ii) having an angle of incidence of 0 degrees on the surface of the light guide such as a prism face and the like (with respect to the perpendicular) and (iii) that much of the reflected light which occurs in the system is guided out of the system and away from the collection point. All of these methods are useful in increasing the signal/background ratios for the detection of light scattering particles in various samples. There are several light guiding strategies that can be used to effectively remove the reflected light out of the system to improve signal to background.

(4) refractive index enhancement methods are also extremely useful in increasing the signal to background ratios in many different types of samples. We have found several methods to increase the signal to background, a few of which are now discussed. The use of liquid to cover the surface containing the particles, where the closer the liquid's refractive index is to the refractive index of the surface that contains the particles, the better the signal/background ratio. We have found that for detecting analytes on a dry solid-phase, the signal/background ratio is improved by placing a liquid layer on top of the surface. For example, when an aqueous buffer solution of refractive index of about 1.33 is used to cover the sample surface we get much improved results as compared to measuring the particles on the same surface in air. Even better signal/background ratios are obtained by using liquids which more closely match the refractive index of the solid-phase. For example, an assay can be performed by first binding the analytes with light scattering particles coated with binding agent to the solid-phase in the sample medium or other appropriate reaction mixture or buffer. The solution in the sample is then diluted or replaced with a solution of chosen refractive index that

covers the solid-phase prior to detection of the particles. In this fashion, highly sensitive results can be achieved.

In addition to the above methods, the further use of narrow band pass optical filters, cutoff optical filters, spatial filtering such as apertures either between the illumination beam and the sample and/or between the sample and the photodetector or eye will also increase the signal/background ratio. Use of confocal imaging techniques may also be useful in certain analytical assay applications where the cost and sophistication of such techniques and apparatus are not an issue. The use of longer wavelength sources either optically filtered or not are also ways to increase the signal/background ratio. Guiding the excess non-specific light out of the system by using specifically designed sample chambers to remove the excess light is another useful method. General sample chamber designs that are useful are described later. All of these variations on one or more aspects of the current invention provide for increased signal/background ratios and thus provides for the more specific and sensitive detection of one or more analytes in a sample.

There are many different ways the DLASLPD illumination and detection methods can be specifically applied to a sample and these are outlined in Figure 15. Figure 14 provides a diagram for orientation and description of the outline of the DLASLPD methods shown in Figure 15. One skilled in the art will recognize the utility the method affords when used with certain metal-like particles to detect one or more analytes in a solid-phase or similar sample.

The detail of the methods is now described.

Illuminating and Light Collection Optics

1. General Concepts

The solid-phase methods we now describe can be applied to the detection of light scattering particles. The detection and measurement of one or more light scattering properties is then correlated to the presence, absence, or concentration of one or more analytes in a sample. These methods can be used with most, if not all known solid-phase analytic methods including microarray, array chip, or similar formats. The method is designed to have a wide range of sensitivities (from low sensitivity to the ultra sensitive range). This range of sensitivities is achieved with easy to use and inexpensive apparatus.

In the technology, the number or relative number of particles on a surface is determined through methods that depend on the light scattering properties of particles. The detection system consists fundamentally of (1) a magnifying lens (also called an imaging or collection lens) that forms a magnified image of the light scattering particle patch or a portion of the patch and (2) an illuminating system that makes the particles appear as bright objects on a dark background (the DLASLPD method). The method can be performed without the need for the collection lens also. The number of particles in the magnified image can be quantified by particle counting or by measuring the scattered light intensity (which is proportional to particle number or density). Particle counting can be done by (a) eye (unaided or with an ocular lens, depending on particle size), (b) an electronic imaging system (for example video camera, CCD camera, image intensifier) or (c) a photosensitive detector with a field limiting aperture and a scanning light beam arrangement. Scattered light intensity can be measured with an electronic imaging system or photosensitive detector. At low particle surface

densities (less than about 0.1 particles per μ^2), the particle counting method is preferred while at higher surface densities (especially, where the individual particles are closer than the spatial resolution capabilities of the magnifying lens) the steady light scattering intensity measurement is preferred. The technology is designed to easily shift between these two methods of detection, that is, between particle counting and intensity measurements and can be used with particle diameters down to about 20 nm depending on the light scattering power of the particles and specific hardware components of the detection apparatus.

Light Illumination Systems

The illuminating system is a key element in the technology. The illuminating systems are designed to illuminate a particle patch or group of particle dots with high light intensity in such a manner that the individual particles appear as bright objects on a dark background. This allows visualization of particles attached to a surface or free in a fluid film above the surface. Free particles are distinguished from attached particles by their Brownian motion which is absent in attached particles. In the following sections we describe the details and logic of the illuminating systems.

Applicant has experimented with many different illumination systems including an expensive commercial dark field illuminator called an ultracondenser (Zeiss). Two fundamental methods of illumination and several versions of these two methods can be used. These methods are simpler and seem to produce higher illuminating light intensities than, for example, the ultracondenser.

General description of the fundamental illuminating methods

The illuminating systems are designed to (1) deliver a beam of high light intensity to a patch (or group of

dots) of light scattering particles and (2) minimize the amount of the illuminating light that enters the detecting system directly or through reflections. This is achieved by constraining the light beam and its reflections to angles that are outside the light collecting angles of the detecting system. In one illuminating method, the collecting lens and the light source are on opposite sides of the solid-phase surface (illumination from below) and in the other method, the illuminating light source and magnifying lens are on the same side of the surface.

Direct Illumination from below the Magnifying Lens

Figure 1 presents a schematic diagram of one of the basic methods of illumination used. In this method, the light impinges on the solid-phase surface S from below the surface. It is assumed that S is transparent (although it could have some color). O is a region on the surface that contains light scattering particles. The magnifying or light collecting lens L is located above S. The angles at which L collects light are shown as a shaded cone C (light collecting cone of lens L) with an apex at the surface S (where the light scattering particles are located) and a base determined by the diameter D of the lens. The illuminating light beam (LB) is angled so that is does not enter the light collecting cone of L. The arrows show the direction of travel of LB.

The solid-phase can be, for example, a microscope slide, a microtiter plate or other types of transparent solid-phases used in clinical diagnostics. The light source can be any type of light source, such as filament lamps, discharge lamps, LEDs, lasers and the like. Light is collected from the illumination light beam using optical light fibers and light collecting lenses, and then focused onto the scattering light particles using a condenser lens. The mean angle θ which the light beam makes with the surface S is adjusted so that the beam of light does not enter the lens L as explained above. The

adjustment of the angle θ can easily be done by visual observation of the light scattering particles through the magnifying lens and occular (compound microscope arrangement), adjusting the angle so that the particles appear as bright objects on a dark background. This angle also serves well for light scattering intensity measurements although at high particle densities, the focusing requirement is not as stringent.

The magnitude of the angle θ can be deduced from the numerical aperture of the magnifying lens. For ultrasensitive detection, a microscope objective is used as the magnifying or imaging lens. A microscope objective usually has its numerical aperture inscribed on its housing. Numerical aperture can be defined in terms of the diagram of figure 2. This figure shows a magnifying lens (with focal length f) that is focused on a patch of light scattering particles at O. The distance between the lens and O is equal to f. The lens (L) collects all light scattered from O into the solid cone whose base is the diameter D of the lens. The angle θ_H is defined as the planar half angle of this solid cone. The numerical aperture (N.A.) of the cone is related to θ_H by the expression

$$N.A. = n \sin(\theta_H) \qquad (36)$$

where n is the refractive index of the medium between the lens and the point O. The medium, for example, can be air (n=1), water (n=1.33) or immersion oil (n=1.5). For small values of θ_H, N.A. is approximately equal to D/2f where D is the diameter of the lens and f is focal length.

The following table gives typical values for the numerical apertures and θ_H of objectives that can be commonly used (for n=1).

Magnification	N.A.	θ_u in degrees
x10	0.25	14.47
x20	0.5	30
x40	0.65	40.54
x63	0.85	58.2

As already mentioned, the exciting light beam must be angled so that it is outside of the light collecting solid cone of the magnifying lens. For high magnifications, the angle at which the exciting beam impinges on the solid-phase surface must be large. For example, for a x40 objective, the incident angle must be larger than 40°. Since the fraction of light reflected at a surface increases with incident angle, we must consider whether the angles that have to be used in our illuminating system result in a large loss of light due to reflection. Furthermore, we must consider whether critical reflections (total internal reflections) are involved at high incidence angles. The following is a brief discussion of the fundamental laws of refraction and reflection needed in the subsequent discussion of the effects of reflections in this illuminating system.

Snells Law of Refraction

We describe Snells law of refraction in terms of the diagram of Fig. 3. This figure shows a light beam that travels along a medium of refractive index ni (i for incident medium) and impinges on the surface S of a medium of refractive index nt (t for transmission medium). Part of the incident light is transmitted into medium t (the refracted beam) and part is reflected (the reflected beam) back into medium i. If the angle of incidence is θi then the angle of the refracted beam is given by Snells Law which can be written as

$$ni \text{ Sin } (\theta i) = nt \text{ Sin } (\theta t) \quad (37)$$

If $ni < nt$ then $\theta i < \theta t$. If $ni > nt$ then $\theta i > \theta t$. Note that angles are measured with respect to a line that is perpendicular to the surface S. The reflected beam makes an angle $\theta r = \theta i$ (that is, the reflected and incident angles are equal).

Laws of Reflection. Fraction of Incident Light that is Reflected at a Surface.

The fraction R of incident light intensity which is reflected for different incident angles θi can be calculated using Fresnels equations of reflection. (It should be noted that intensity is here defined as energy per unit time per unit area. Intensity is also called irradiance). However for simplification, we present our discussion in terms of plots relating R to θi. The exact dependence of R on θi is determined by the values of ni and nt and the state of polarization of the incident light. Important facts concerning reflectance are as follows.

i. Reflectance for the case where the light beam travels from a medium of low refractive index to one of high refractive index ($ni < nt$).

Figure 4 shows plots of R vs. θi ($\omega = \theta i$) for the case where ni = 1 (air) and nt = 1.5 (the latter is close to the refractive index of glass or plastic) and for light polarized parallel (rp) and perpendicular (rs) to the plane of incidence. The plane of incidence is defined as the plane which contains the incident light beam and the line perpendicular to the surface (see Fig. 3). The reflectance R of unpolarized light is given by the average of the graphs for light polarized parallel and perpendicular to the plane of incidence. In Fig. 4, the reflectance graph for unpolarized light is labeled Ord (for ordinary). The graphs of Fig. 4, show that

a. rs increases continuously with increasing ω (ω in Fig. 3 is the same as θi as used herein). The increases in rs is small up to about 70° (where the reflectance is only about 15%) and then increases much more rapidly reaching 100% reflectance at 90°. Thus, the fraction of light that is reflected is less than 20% up to incidence angles of 60°.

b. rp decreases with increasing ω up to about 57° where rp is zero. The angle at which rp=0 is called the Brewster angle or polarizing angle. The Brewster angle θb can be calculated with the expression

$$\text{Tan}\ (\theta b) = nt \qquad (38)$$

assuming that ni =1 (air). For nt = 1.5, above equation gives $\theta b = 56.3$. It should be noted that at the Brewster angle, $\theta i + \theta t = 90°$. Thus for nt=1.5, $\theta i = \theta b = 56.3°$ and $\theta t = 33.7°$. For angles greater than the Brewster angle, rp increases rapidly with increase in ω and reaches a value of 100% at 90°.

c. For unpolarized (ordinary light), the reflectance increased gradually with increase in ω up to about 70° and then increases rapidly reaching 100% at 90°. Less than 20% of the incident light is reflected for θi <70°.

d. It should be noted that the intensities of the reflected and transmitted light do not add up to the intensity of the incident light. This seems to violate the law of conservation of energy. This apparent violation is actually due to the definition of intensity as energy per unit time per unit area. Because of refraction, the incident and transmitted light do not have the same cross sectional area. If the differences in cross sectional areas are taken into consideration, then it can be shown that the energy per unit time in the reflected and transmitted beams add up to the energy per unit time in the incident beam.

ii. Reflectance for the case where the light beam travels from a medium of high refractive index to one of low refractive index (ni > nt).

Figure 5 shows plots of reflectance of polarized light vs. angle of incidence ($\omega = \theta i$) for ni = 1.54 and nt = 1. The plots are quite different than those of Fig. 4 for ni < nt. The most significant difference is that at an incident greater than about 41 ° all of the light is reflected (100% reflection, total reflection). The smallest incident angle at which total internal reflection occurs is called the critical reflection angle θc. The value of this angle depends on the values of ni and nt. An expression for calculating θc from values of ni and nt can be derived by considering the angles of the incident and transmitted light beams at the critical angle. At the critical angle θc, the reflected beam contains most of the incident light and makes an angle θc with the respect to a line perpendicular to the surface as required by the laws for specular reflection. The transmitted light has low intensity and its angle θt with respect to perpendicular line is 90°. That is, the transmitted light beam travels parallel to the surface. We can therefore obtain the value of θc by inserting θt =90° in Snell's equation. This insertion gives

$$nt \text{ Sin } (90) = ni \text{ Sin } (\theta c) \qquad (39)$$

Since Sin (90) =1, we can write

$$\text{Sin } (\theta c) = ni/nt \qquad (40)$$

For nt = 1.54 and ni =1 (air), the above equation gives θc = 40.5°. It should be noted that the critical angle is the same for unpolarized light and light polarized perpendicular or parallel to the incident plane. That is, θc is independent of whether the light is unpolarized or plane polarized.

iii. Effects of reflectance and refraction on the illumination of a patch of light scattering particles.

We first consider the simple case where the particles are on the surface of a dry microscope slide in air. That is, the particles are dry and air is the medium on both sides of the microscope slide. Figure 6 shows a schematic diagram of the reflections and refractions involved in this case. The first reflection occurs at the surface S1 (ni<nt, ni=1 and nt=1.5). Figure 4 shows that the fraction of light reflected is below 20% for incident angles up to 70°. Therefore, reflections at S1 are not problematic in this method of illumination. Surface 2 could be problematic because the light beam passes from a low to a high refractive index and the possibility exists for total internal reflection at this surface. The critical angle for total internal reflection at a surface where ni=1.5 and nt=1 (air) is about 42° (calculated with Eq. (40)). The question now is whether this critical angle is attained when the incident light beam at surface 1 has a large angle.

Figure 7 shows a plot of $\theta i2$ [θtj] vs. $\theta i1$ [θij] calculated with Snells Eq.((37)) and using $\theta t1 = \theta i2$. As can be seen from the plot, $\theta i2$ rapidly increases with increase in $\theta i1$ up to about $\theta i1 = 70°$. The increase in $\theta i2$ then levels off and does not reach the critical angle until $\theta i1 = 90°$. However at $\theta i1 = 90°$, no light is transmitted across S1. We thus conclude that for the arrangement of Figure 4, critical illumination is never achieved at any practical angle of $\theta i1$. Furthermore, reflections do not significantly diminish the amount of light delivered to the particles on S2 for $\theta i1$ values less than about 70°.

We have verified the above conclusions experimentally. In our experiments, however, we have found that light scattered by smudges, dirt, scratches and other irregularities or artifacts on surfaces S1 and S2 (non-specific light scattering) can become comparable to the

light scattered by the particles on S2 and thus significantly increases the background light and diminishes the sensitivity of particle detection. However, the non-specific light scattered by artifacts on S1 and S2 is concentrated in the forward direction of the illuminating light beam whereas the light scattered by the particles (for small particles) is in all directions. These effects are demonstrated in Fig. 8.

In Figure 8, the intensity of non-specific light scattered by surface artifacts in any direction θ is given by the length of the line (with angle θ) extending from the origin O to the intensity envelope. The light scattered by the particles is shown as dashed lines which go out in all directions from O. The effects of non-specific light scattering on the detection of specific light scattering can be demonstrated experimentally using a microscope slide containing 60 nm gold particles on the surface. In air, these particles scatter green light, and in the absence of non-specific light scattering, an illuminated patch of particles appears as a green patch on a dark background. Surface artifacts scatter white light and when this type of non-specific light scattering is superimposed on the specific particle light scattering, the light scattered from a patch of particles has a greenish-white color instead of a pure green color. The effects of preferential forward scattering by surface artifacts can be seen by viewing the scattered light by eye positioned at different angles θv as shown in Fig. 8. When the eye is placed at $\theta v = 0$, the scattered light has a greenish white color. As the angle of observation θv is increased, the white color decreases and, at θv greater than 30°, only the pure green color of gold particles is seen. Thus, in the present invention, it is useful to observe by eye or detect with photodetector means at an angle greater than θv of 30°. As we show later, non-specific light scattering due to surface artifacts can be

further reduced by illuminating through light guides as for example, a prism type of arrangement.

We next consider the case where the light scattering particles are in a thin film of water that is on a microscope slide and covered with a cover glass as shown in Fig. 9. The illuminating beam encounters four surfaces where changes in refractive index occur, namely S1 (air to glass), S2 (glass to water), S3(water to glass), S4 (glass to air). Consideration of the reflection and refraction at each surface as done in previous paragraphs leads to the conclusion that in the system of Fig. 9, reflections do not significantly reduce the amount of light energy delivered to the scattering light particles and that critical reflection does not occur at any surface. Non-specific light scattering due to surface artifacts on surfaces S1 and S4 are present as in the case of Fig. 8. However, the presence of water greatly reduces non-specific light scattering for surfaces S2 and S3.

Direct Illumination from the Same Side as the Magnifying Lens

This method of illumination is shown in Fig. 10. The meaning of S, L, C, and LB are the same as in Fig. 2. In this Figure, the exciting light beam impinges on the surface S from above the surface. The light collecting lens is also above S. The exciting light beam is angled so that neither the incident or reflected light enter the light collecting cone C of the lens L. In this method of illumination, it is necessary to keep the light reflected from the different surfaces in the path of the beam from being reflected into the light collecting cone C of the magnifying or imaging lines L. Since the angle of reflection is the same as the angle of incidence at each surface, collection of unwanted reflected light by L can be minimized by confining the illuminating light beam to those angles which are outside of the cone C. It can be shown, as in the previous section, that reflections do not

significantly reduce the amount of energy delivered to the light scattering particles and that critical reflections do not occur in dry or water covered particles. Nonspecific light scattering due to surface artifacts are the same as discussed in the previous section.

Illumination through a Prism Arrangement.

i. Illumination from below.

Figure 11 presents a schematic diagram of a prism setup. In one of its simplest forms, it consists of a triangular prism on which can sit microtiter wells, glass slides, microarrays on plastic or glass substrates and the like which contain the light scattering particles to be detected. The sample chamber or slide that contains the light scattering particles is located on the upper surface S2 of the prism, the surface containing the particles is at the focal point of the lens L. Immersion oil is placed between the sample chamber or slide and the prism to minimize non-specific light by refractive index matching. The particles are dry in air. If the index matching at S2 is exact or almost exact, then the light beam does not undergo significant refraction or reflection at S2. Thus, an illuminating light beam travels in an approximately straight line as it transverses the prism and surface containing the light scattering particles. However, there is refraction at the air-prism interface S1 and at S3. Consider an illuminating light beam which enters the prism in a direction which is perpendicular to S1 (angle of incidence is 0°) and suppose that the prism is a 45° prism (angle γ = 45°). Because the beam travels in a straight line from S1 to the point O on S3, it strikes the surface S1 at an angle of 45°. The beam will then undergo total internal reflection since the critical angle for glass to air is about 42°. Thus, in contrast to illumination from below without a prism (see Fig.6), the prism arrangement allows critical reflection. Recall, that in the absence of a

prism (Fig.6), critical reflection cannot be achieved because of the refraction of light at S1 as shown in Fig. 6 (also see Fig.7). In order to deliver a high energy light beam when using a light guide such as a prism, the illuminating light beam must be directed so that it strikes S3 (Fig. 11) at an angle less than 42°. The question now is whether incident angles less than 42° at S3 allow one to satisfy the condition that the light which exits at S3 must be outside the collecting cone of the lens L. Suppose that the angle of incidence at S3 is 35°. From Snells law (with ni =1.5 and nt=1) we calculate that the exit angle is 62° which is outside the collecting cone of our most demanding objective, namely the x40 objective with θH = 41°. We conclude that the prism arrangement of Fig. 11 permits delivery of high light energy to dry light scattering particles in air while maintaining a dark background.

We now consider the prism arrangement of Fig. 11 in which the light scattering particles are covered with water and a cover glass. As described before, an exciting light beam travels in a straight line from S1 to O where it encounters the glass-water interface. It then travels through the water and cover glass, finally encountering the glass-air interface at the upper surface of the cover glass. It is of interest to consider the reflections that occur at the glass-water and glass-air interfaces. The angle for critical reflection at the air water interface is equal to 62.5° (using ni=1.5 and nt=1.33 in Eq.(40)). The introduction of water at the surface S3 thus permits illumination at much higher angles, than in air, without encountering total internal reflection. Furthermore, at angles less than 62.5°, reflectance is low at the glass-water interface. We now consider the reflection at the cover glass-air interface. Consider a beam of light which enters perpendicular to the surface at S1. If the prism is a 45° prism, then the beam strikes S3 at an angle of 45° where it undergoes total internal reflection. Refraction

at S3 (glass to water) changes the angle of the beam to 55°. However, refraction at the water-cover glass interface bends the beam back to 45°. The beam thus strikes the cover glass-air interface at 45°. The prism arrangement with particles in water and a cover glass thus permits efficient delivery of light energy to the light scattering particles (attached to the surface S3 or free in water) but the incident light is totally reflected at the cover glass.

From the above discussion we conclude that there are no reflections that seriously affect the delivery of light energy to light scattering particles in a film of water. Total reflection does occur at the cover glass-air interface but these reflections do not affect the delivery of incident light energy to the scattering particles.

From the above discussions, it is evident that both prism and non-prism arrangements permit efficient delivery of energy by angled illumination of particles attached to a surface or free in solution. By efficient delivery we mean that the beam of light does not undergo total internal reflection at any pertinent surface and the collection of non-specific light is minimized. However, we have found experimentally that superior detection capabilities occur with the prism arrangement in certain applications because it eliminates or considerably reduces artifacts due to scattering from irregularities on glass or plastic surfaces close to the light scattering particles. The immersion oil used to couple the solid-phase to the prism, eliminates almost completely non-specific light scattering from irregularities at S2 (Fig. 11). These effects seem to be due to an index matching mechanism. The non specific light scattering at the point of entry of the light beam at S1 is far removed from the specific scattering particles at point O (if the prism is sufficiently large) and does not contribute to the scattered light collected by the magnifying lens. Furthermore, if the specific scattering particles are in

water, index matching by the water film considerably reduces non-specific scattering on the surface S3. Moreover, in the presence of water and a cover glass, the illuminating light beam undergoes total reflection at the cover glass-air interface. This reflection considerably diminishes non-specific scattering from irregularities at the cover glass-air interface because only a small amount of light energy reaches the irregularities on this surface. In addition, the total internal reflection diminishes or eliminates the probability that illuminating light can enter directly the light collecting cone of the magnifying lens. It should be noted that total internal reflection can also affect the collection angle of the lens L because particle scattered light which makes an angle greater than 42° at the cover glass-air interface is totally reflected. This effect however is not a serious one.

Microscopic Observations

In the previous section we discussed our illumination and detection (magnifying lens) systems with emphasis on the factors (reflections and refractions) that govern the efficient delivery of light energy to light scattering particles (which are attached to a surface or free just above the surface) while minimizing the non-specific light that is collected. In this section we present experimental details obtained by visual observation, through an ocular, of the magnified image produced by the magnifying lens L.

a. Details of Light Sources and Optical Fibers

We have used three different types of light sources as follows.

i. Leica Microscope Illuminator

This is a standard, commercial microscope illuminator. It uses a tungsten filament light bulb and

lenses that produce a 5 x 7 mm focused image of the filament at a distance of about 22 mm from the tip of the illuminator. To produce a more focused light beam, we attached a x10 objective lens to the microscope. The lens is about 6.5 cm from the tip of the illuminator. The objective produces a focused spot of light of about 4 to 5 mm in diameter at a distance of about 7 mm from the objective lens.

ii. Bausch and Lomb Fiber Lite Illuminator

This is also a commercial illuminator. It uses a 150 watt tungsten filament lamp that is mounted on a parabolic reflector. The reflector produces a beam of almost parallel light with a diameter of about 25 mm. An 11 mm diameter optical light guide (consisting of many small optical fibers bundled together) is positioned close to the filament lamp. The optical light guide is then split into two equal light guides, each with a diameter of about 5.3 mm and a length of about 2 feet. We use one of the light guides. To produce a focused light spot, we collimate the light from the optical fiber using a 25 mm focal length lens (20 mm diameter) positioned about 25 mm from the end of the optical fiber. The collimated light is then focused, into a 5 mm diameter light spot, by a x10 objective which is positioned about 50 mm from the collimating lens. The collimating and objective lenses are in a compact holder that is rigidly attached to the optical fiber. The flexibility of the optical fiber makes this type of light source system much easier to use than the rigid Leica Microscope Illuminator.

iii. Custom Illuminator

This is an illuminator which we constructed. It uses a 12 V, 28 Watt tungsten filament lamp that is mounted on a parabolic reflector. The reflector produces an almost parallel beam of light which is focused into an optical guide that has a diameter of 0.125 inches. The optical

guide is 36 inches long. Light from the reflector is focused unto the optical fiber with a lens (23 mm focal length, 9 mm diameter) that is positioned close to the lamp. The light guide has a numerical aperture of 0.55 and accepts a cone of light with a planar half angle of 60 °. Light which exits at the other end of the light guide is collimated by a 12 mm focal length lens (diameter 11 mm) positioned about 17 mm from the end of the fiber. The collimated light is finally focused into a bright spot of 5 mm diameter by a x 10 objective. The objective lens is positioned at a distance of about 26 mm from the collimating lens. The 5mm bright spot is at a distance of about 12 mm from the end of the focusing lens.

b. Prisms (light guides)

Figure 12 shows diagrams of some light guides such as prisms which we have found useful in our illuminating systems. Some of these are not actually prisms in the classical sense but may rather be called light guiders or light guides. The light guider allows light to be delivered efficiently at an angle while allowing the reflected light beam to exit at the face opposite to the incident light face. It must have some minimum dimensions for the following reasons. The spots where the light beam enters and reflected light exits can produce significant non-specific light scattering due to surface irregularities. These spots must be sufficiently removed from the patch of specific light scattering particles to minimize their non-specific light scattering contribution. The light guides can be molded into one piece with the sample chamber thus eliminating the need to use immersion oil between the light guider and analysis piece. We have proto-typed such a device by gluing a small light guide to the bottom of a plastic chamber which has a microarray of streptavidin spots coated on the inner surface of the sample chamber well. Our detection and measurement of bound light scattering particles to the individual

microspots of the microarray using this device were essentially the same as our measurements by particle counting and intensity measurements with the sample chamber placed on a prism with immersion oil between the two surfaces.

c. Microscopic Observations

Using an ocular for visual microscopic observations, we have evaluated several illuminating arrangements with special emphasis on brightness of particles, darkness of background, and usefulness in different types of clinical assay formats. We have found several arrangements that give good results. Here we will limit our discussion to one of the easiest to use and least expensive arrangements which gives excellent results. By excellent results we mean bright particles on a dark background using x10 and x40 objectives as magnifying lenses.

The imaging system is an inexpensive microscope from Edmund Scientific. The microscope consists of an objective (x10 or x40) and an ocular in a standard 160 mm tube. The reason for using a microscope, instead of just simply an objective and an ocular, is the convenience provided by the fine/course focusing mechanism of the microscope. However, we have modified the stage of the microscope to adapt it to our method of illumination and have replaced the microscope condenser with a light guide (prism type) as shown in Fig. 12(d). The cylinder below the prism fits into the condenser holder. The modified stage and illuminator make it possible to work with microscope slides, microtiter plates and other plastic plates. However, the x40 objective cannot be used with the thick plastic plates because of the small working distance (about 0.45 mm) of this objective. x40 objectives with longer working distance are available. The custom Illuminator is used for illumination.

To setup the microscope system for the DLASLPD method, a microscope slide containing free or surface

bound gold particles (60 nm gold particles in a thin water film covered with a cover glass) is placed on the microscope stage. The prism is positioned so that its surface is almost in contact with the microscope slide. The slide and prism surfaces are coupled with immersion oil. The x10 focusing objective of the illuminating system is positioned so that it is almost in contact with the side of the prism that is illuminated (see Fig. 13). The objective is angled so that the light enters perpendicular to the illuminating surface and strikes the surface S (surface in contact with the microscope slide) at an angle of about 45°. If the gold particle film on the slide has a sufficiently high concentration of particles (about 6 x 10^9 particles/ml or higher) the spot where the light crosses the particle film displays a strong yellow-green color due to light scattering by the particles. The position of the x10 focusing objective is adjusted so that the yellow-green spot is centered with respect to the microscope (magnifying) objective. The microscope is then focused on the spot so that the particles appear as sharp objects when viewed through the ocular. The position and angle of the x10 illuminating objective is than repositioned to produce bright objects on a dark background. This adjustment is repeated with the x40 objective of the microscope. There is a narrow range of positions which produce bright objects on a dark background for both x10 and x40 objectives.

It should be noted that none of the illuminating light is transmitted into the air space above the microscope slide when the illuminating light beam strikes the prism surface at greater than about 42° (critical angle for total internal reflection at the plastic or glass-air interface). In our arrangement, the angle of incidence is 45°. This can be verified by placing a piece of white paper above the microscope slide. No illuminating light falls on the paper. However, it is of interest to visualize the illuminating beam to determine its shape or

profile in space. This can be done by placing a rhodamine plastic block on top of the slide using immersion oil for coupling. The rhodamine block is a transparent plastic block that contains the fluorescent molecule rhodamine. A beam traveling through the block can be seen from the rhodamine fluorescence that it produces. The immersion oil eliminates the air space and allows the illuminating beam to enter the plastic block. The profile of the illuminating beam seen by fluorescence inside of the block is shown in Fig. 13.

Once the illuminating system has been properly positioned, gold particles greater than about 30 nm can easily be seen on microscope slides, plastic wells, and solid-phase microarrays or array chips. The x10 objective permits detection of particles densities less that 0.005 particles per μ^2. The x10 objective has a working distance of about 8.5 mm so that it can be used with plastic pieces that are about 8 mm thick.

Method of DLASLPD Video Contrast Enhancement

We have determined that metal-like particles and non-metal-like particles can be detected to greater sensitivities in samples by using the method of DLASLPD Video Contrast Enhancement. This method involves adjusting the imaged non-specific light background electronically such that it is essentially removed from the imaged field while keeping individual particles visible. The method works extremely well with the other methods as described herein. We have observed improved specificity and sensitivity results using this method.

Sample Chamber Improvements

In the spirit of taking certain aspects of the present invention and now further improving it in terms of ease of use, adaptability to different testing environments and conditions, we have found that several aspects of the present invention can be embodied in the

design of sample chambers that are used to conduct the assay and detect the analytes.

For example, based on our observations and conceptualizations we can apply our principles to the general design of the sample chamber, that is the container which contains the sample to be analyzed. These improvements can facilitate the ease of use and applicability to the types of tests and testing conditions briefly outlined above. It should be stated however, that the invention as described herein can be practiced equally well without use of the following sample chamber improvements. These improvements are for the means of increasing the practical applicability of the present invention to specific testing conditions and environments and described elsewhere herein.

We have found that by moving or displacing the surface S1 (incident light surface) as far as possible away from the area that contains the particles to be measured, the signal/background ratio is significantly increased. We have previously described the use of an optical alignment means such as a prism or similar optical light guide which is used to assist in orienting the delivery of the illumination beam to the surface S1. Usually one will use immersion oil between a surface of the light guide (prism etc.) and a surface of the sample chamber. There are numerous conditions in analyte testing where it may be preferable to not have immersion oil as a component in the analytical methodology. We have determined that by increasing the thickness of the surface upon which the particles are on or nearby significantly reduces the level of non-specific light as described earlier.

We have applied this and other aspects to the design of sample chambers for the detection of one or more analytes in a sample. These general sample chamber designs are diagrammed in Figures 17, 18, and 19. Figure 17 shows a sample chamber that has beveled flat sides. The

degree of angle of the beveled sides is matched to the angle of illumination such that the illumination beam strikes the face of the beveled side at an angle as close to 0 degrees as possible (with respect to the perpendicular). In this fashion non-specific reflected and scattered light is minimized. Figure 18 shows a sample chamber that has curved sides instead of the flat beveled sides as described for Figure 17. In this sample chamber, for the exit beam which is diverging, the curved surface allows for the more efficient removal of this non-specific light. Figure 19 shows a sample chamber that utilizes both concepts of moving the incident surface where the light beam strikes the sample further away from the area to be measured and the curved sides to allow for more efficient removal of non-specific light. Thus, this sample chamber has an increased thickness of material below the bottom surface of the well, beveled flat surfaces below the plane of the surface for illumination, and curved sides above the surface plane of the bottom of the well to allow for the efficient removal of non-specific light. The sample chambers as shown in Figure 17, 18, and 19 are useful for measuring immobilized samples as well as solution samples.

Practice of the Present Invention to Analytical Diagnostic Assays - Apparatus Types and Test Kits

It is well known in the art that there is a wide range of analyte types. These analytes exist in different sample environments such as water, urine, blood, sputum, tissue, soil, air, and the like. Depending on the requirements of a particular type of analytic assay, one may want to get semi-quantitative or quantitative information, or both with regard to the analytes of interest. There are conditions where it is desirable to perform the analysis with a small, inexpensive, and highly portable instrument. For example, consumer use, use in the field (away from a lab), or at bedside in the hospital. One wants to be able to quickly get either semi-

quantitative and/or quantitative measurements on the analytes in question. In other applications, it is desirable to have a small and inexpensive instrument for analyte detection in a small lab where from a few to several samples are tested a day, capable of quantitative results. For example, doctor's office, clinic, satellite testing lab, research labs and the like. There are also conditions where one wants to test several hundred to thousand samples per day, such as high throughput testing. Each of the above testing conditions and environments thus require different types of apparatus means. The advantages and disadvantages in terms of ease of use and cost of such apparatus can only be determined in detail when the exact requirements of testing for the analyte(s) in a sample is well defined.

We have determined that the use of certain metal-like particles with certain variations of the DLASLPD methods of detection allow for the development of specific test kits and apparatus for the above mentioned testing environments and applications. There are numerous different combinations of analytes, testing environments, sample types, assay formats, analyte detection requirements, and apparatus cost and size requirements. One of average skill in the art will recognize the tremendous utility of the present invention in that the practice of the current invention in one form or another leads to easy to use and inexpensive apparatus and test kits to solve most analytical or diagnostic testing needs.

There are many different configurations and combinations of the DLASLPD methods, particle types, and sample types that are used together to achieve a specific analyte detection capability. In any specific diagnostic assay application, the sample type(s), and method(s) of illumination and detection are usually fixed, as for example, assay format, sample chamber type, and detection apparatus. The metal-like particles have unique scattered light properties which vary by size, shape, composition,

and homogeneity of the particles. The specific properties of light scattering that can be detected and/or measured from the particles is set by the previous mentioned particle properties, and the method and apparatus used to detect and measure the scattered light properties. Therefore, the ultimate utility and practice of the current invention in one form or another is achieved by combining various aspects of the illumination and detection means, and sample type, with the appropriate type(s) of light scattering particles. This results in specific apparatus and test kits.

One skilled in the art can practice many different aspects of this invention by using various particle types, assay formats, and apparatus in many different configurations to achieve many different resultant diagnostic analytic detection capabilities. Figure 22 diagrams the various aspects of the invention which, when configured in a specific combination, yields apparatus and test kits to suit a specific diagnostic analytic testing need. The resulting apparatus and test kits are made by choosing the appropriate components of the Methodology/ Apparatus Type Configuration (Figure 23) and the Particle Type Configuration (Figure 24). Figure 23 shows that one skilled in the art chooses the illumination source, method and other apparatus components, the assay and sample type, and the detection method and apparatus components. Figure 24 shows that one skilled in the art chooses the appropriate particle composition, shape, size and homogeneity to detect the desired light scattering properties of the particle. These processes as outlined in Figs 23 and 24, and summarized in the diagram of Figure 22, lead to specific apparatus and test kits. The diagram in Figure 25 shows one of the general methods we have used to develop specific apparatus and test kits for a particular diagnostic testing need. One skilled in the art does not need to practice the method in Figure 25 to practice the present invention in one form or another.

The remarkable signal generation and detection capabilities of combining metal-like particles with the DLASLPD methods of illumination and detection as described herein allows for a wide range of analyte detection sensitivities. With regard to the general types of testing environments and the Figures 22-25 briefly described above, one skilled in the art can easily develop apparatus and test kits where in some diagnostic testing applications one only need to use the naked eye for detection and/or measurement, and in other cases a simple light source such as an LED (light emitting diode) or low power filament lamp are used and a photodiode or photodiode array can be used to detect and/or measure the signal. In other analytical testing applications, a laser, laser diode, filament lamp, or the like can be used with a camera, video camera or other CCD device (charge-coupled device), and with simple image processing means, the scattered light from the particles in a microarray format, or any other format can be detected and measured. These examples are not meant to be limiting but rather to generally show the versatility and broad utility of the present invention to detect one or more analytes of interest in a sample.

For example, a small hand-held apparatus for portable use that can measure one or more analytes in a sample can be built by using a low power filament bulb, LED, or laser-diode as a light source. A photodiode or photodiode array is used as a detector. Depending on the sensitivity of detection required, certain types of metal-like particles can be used with this apparatus to satisfy the analyte detection requirements. Test kits are constructed for multi-analyte or single analyte detection in liquid or solid-phase samples. For example, for liquid samples, different particle types, each having different easily detectable scattered light properties are used. In solid-phase samples and formats, such as a microarray, one particle type can be used for all the different analytes

or various combinations of particle types could be used (depending on the concentrations of the different analytes in the sample).

In another example, an inexpensive apparatus and test kits capable of measuring to low analyte concentrations can be constructed as follows. A low or high power light source is used with a photomultiplier tube, photodiode array, or video camera. A lens is used to collect the scattered light from the surface(s) containing the particles. A microprocessor or external desktop computer is used to collect and analyze the scattered light data. Test kits for multi-analyte solid-phase analysis are made by using the proper particle type(s) with appropriate microarray sample chambers to achieve the concentration ranges and detection limits as required. This type of apparatus and test kits may be useful in research labs, doctor's offices, satellite clinics, environmental testing labs, and high throughput testing labs.

The above examples of apparatus and test kits are given as illustrative examples and should not be interpreted as the only practices of the present invention. One skilled in the art will recognize the broad utility of the present invention. By practicing one or more aspects of the present invention to suit a specific analyte(s) detection need, one skilled in the art will recognize the wide range of apparatus and test kits that can be made.

<u>Assays Involving the Association or Aggregation of Two or More Particles by Interaction of Analyte and Specific Analyte Recognition Reagents.</u>

It is known in the art that by using the appropriate binding agent(s) and concentration of binding agents and analyte, agglutination, aggregation, cross-linking, networking and similar binding events can occur and that these events can be used to detect one or more analytes in a sample. In some immunoassays, visible precipitates are

formed if the antigen is soluble and multivalent while agglutinated or clumped particles are formed if the antigen is particulate and multivalent. In some nucleic acid assays, one specific single-stranded probe can "crosslink" two or more single-stranded targets and propagate networks. Alternatively, two or more different unique single-stranded nucleic acid probes can be used to bind to different sites on the same target single-stranded nucleic acid. In this approach, cross-linking of two or more targets can be accomplished, or, by just having the two unique probe sequences bound to the same target allows for detection.

The present invention allows for easier to use, more sensitive, and more versatile detection of analytes than was previously possible. In specific assay formats, the submicroscopic particles of this invention can form different types of aggregates that can be detected visually or instrumentally in a microscope or through macroscopic observation or measurements without having to separate free from analyte bound particles. The type of aggregates formed depends on the size of the cross-linking agent or agents and their valency and on the type of binding agent attached to the particle. Aggregates can range from two particles to many.

The particles used in a homogeneous, liquid phase, aggregation detection assay can be labeled directly or indirectly. In a direct labeling assay, an agent that can bind directly to the analyte is attached to the signal generating particle. For example, in a direct labeling nucleic acid analyte assay, the DNA probe is attached to the light scattering particle. In an indirect assay, the analyte detecting agent is labeled with a chemical group A and the particle is labeled or coated with an agent that can recognize the group A. Using direct or indirect labeling, an assay can be formatted so that interaction of the analyte recognition binding agents with the analyte (and with group A in the case of indirect labeling) lead

to aggregation of the particles. The aggregates can be composed of two or more particles.

We have found that if the aggregating or cross-linking agent(s) in an assay are small in size so that the particles in the aggregate are in very close proximity, then aggregates containing two or at most a few particles appear as a single particle (submicroscopic aggregate) in the microscope. However, this submicroscopic aggregate displays different light scattering properties than unaggregated particles due to particle-particle perturbations. Depending on particle composition, size and shape, we have observed changes in the color, intensity RIFSLIW, and polarization of the scattered light. These changes can be used to measure the amount of analyte without having to separate free from analyte bound particles. In a microscope the changes can be used to distinguish submicroscopic aggregates from nonaggregated particles even though both may appear as single particles. If the aggregating or cross-linking agents are large in size, such as a long DNA chain where the distance between the particles in the aggregate is larger than the resolution of the microscope, then the particles in the aggregate can be seen individually and distinguished from unaggregated particles by the fact that they stay or move together. These microscopic aggregates can readily be seen in the microscope when as few as two particles are in the aggregate. If the distance is sufficiently large so that particle-particle perturbations are small, than the particles in the aggregate retain their original light scattering properties. There are also particle-particle separation distances which are between the two general cases discussed above. We have observed in some specific cases that the particles in a submicroscopic aggregate do not perturb their light scattering properties presumably because they are not close enough for particle-particle-perturbation to occur. The aggregate can nevertheless be distinguished from nonaggregated particles because their

intensity is n times that of the unaggregated particle where n is the number of particles in an aggregate and/or the particles are "fixed" into position relative to one another.

From the above discussion, it can be seen that liquid, phase homogeneous assays based on macroscopic measurements or visual observations can readily be achieved if aggregates produced by the presence of analyte have different light scattering properties than free unaggregated particles. If particle-particle perturbations in the aggregates are small so that the light scattering properties of aggregated and free particles are similar, homogeneous assays are still possible using methods of detection which allow visualization or measurement of the light scattering intensities of the individual particles and aggregates. In the situation where the individual particles in an aggregate can be seen, then the aggregates can be easily distinguished from free particles and quantified as described above by visual observations or computerized image analysis. Aggregates can also be distinguished from free particles and quantified in a flow cytometer or similar apparatus or device since aggregates would have a higher light intensity than individual particles. In situations where the individual particles in an aggregate cannot be seen in the microscope and particle-particle-perturbations are absent, the free particles and aggregates can be distinguished by their differences in intensities and the number of particles in an aggregate established from the scattered light intensity of the aggregate (assuming that the intensities are additional). This can be done by image analysis or flow cytometry and includes the development of an image by laser scanning or other methods which can spatially analyze an area or volume of the sample.

As the number of particles in a submicroscopic aggregate increases, the aggregate can be seen as an

enlarged particle or large particle even though the individual submicroscopic particles in the aggregate may not visible through the microscope. In the case of microscopic aggregates, increase in the number of particles in the aggregate produces visible networks and the particles in the network can be counted. Large networks and particle aggregates produce macroscopic entities that can be observed with the unaided eye and can form precipitates or agglutinates.

One of average skill in the art will recognize that the different aggregation phenomena described in the preceding paragraph can be exploited to develop many different types of homogeneous assays, some of which employ a microscope or other image analysis techniques and others which involve macroscopic observations or measurements.

The following are selected illustrative examples of how a homogeneous type or other types of assays can be performed.

Examples of Assay Formats Using Light Scattering Particles

Below are given a few illustrative examples which demonstrate the broad versatility and great utility of the present invention in different assay formats. One of average skill in the art will recognize that there are many variations of the present invention which allows for the more specific, easier to use, and more sensitive detection of one or more analytes in a sample than was previously possible.

i. Assay Formats involving the association of two or more particles by molecular recognition based binding events. General Principles

In one set of experiments, we biotinylated the surface of a preparation of 40 nm diameter gold particles using the method of base material molecule attachment. After purification by centrifugation and washing, we placed a

drop of this material on a glass slide and covered it with a coverslip, and observed with the light microscope using the DLASLPD method of light illumination and detection. The material appeared homogeneous, the particles moving very fast in Brownian motion with a green color. We then removed the coverslip, and placed a drop of a solution of streptavidin onto the slide and recovered it with the coverslip. After a period of time, new yellow-orange and orange, and orange-white colored particle structures appeared in solution which had much greater intensity and moved much slower than the green particles. Some of these new particle structures also appeared to be asymmetrical, as they flickered as they rotated in solution. After some time, many of the green particles had disappeared and there were many of these yellow-orange and orange particle aggregates. When the edge of the coverslip was examined under the microscope, it was coated with a layer of orange, yellow-orange, and white-orange particle aggregates which were very bright in color. We have observed similar phenomena in a homopolymer nucleic acid system. These observations show that in various forms of the present invention, changes in the particle scattered light properties can be used to detect molecular binding events either by visualization of the aggregates, decrease in the number of "free" single particles, or in bulk solution by using other methods. For example, for detection in bulk solution or a flow system, the increase in number of new particle forms with unique scattered light properties and/or the decrease in the amount of particles with the original properties by illuminating a part of the solution under appropriate conditions and looking for changes in the scattered light emanating from the solution. Alternatively, by using a flow-based system, the material in the sample can be more specifically analyzed. For example, a microchannel, capillary, or flow cytometry apparatus or device is used such that a portion or the entire sample solution can be analyzed on a

particle by particle basis. The solution flows by an illumination source(s) and detector(s) or alternatively, the solution is captured in a microchannel or capillary tube and then all or part of the microchannel or tube is analyzed by moving either the sample container, light source or detector (or some combination of these) along the length of the sample.

For example, a certain nucleic acid analyte is composed of about 100 nucleic acid bases and is present in a sample. The sample is prepared so that this nucleic acid is in single-strand form. Then two or more unique single-stranded "probe" nucleic acid sequences are added to the sample where these different probe nucleic acids bind to different regions of the target strand. Each of these probe nucleic acids also has attached to by indirect or direct labeling means one or more particles. Following incubation, the sample is placed in a flow cytometer apparatus or similar flow device where the solution containing the sample can be analyzed. If the target sequence is present, there will be two or more particles which are "bound" together in close proximity. Depending on the separation distance of the particles, particle-particle perturbations may or may not be occur. These molecular structures which contain two or more particles as a result of the hybridization of the probe strands to the target strand are detected using appropriate means as described earlier.

ii. Assays Involving the Release of Molecular Entities

There are assay format applications where the present invention can be used to detect the presence of an analyte as the result of a molecular, chemical, or other event. For example, intra or inter molecular bonds, linkages, or other molecular structures can be altered such that the overall molecular geometry changes, or, a molecular piece(s) is dissociated as a result of the process. For example, peptides, proteins, nucleic acids, or pharma-

ceutical agents and the like can be attached to the surface of a sample container by various means that are known in the art. In some of these substances, there is one or more intramolecular linkage or bonding sites which can be cleaved or otherwise altered by chemical, biological, or other processes. For example, the presence of a specific enzyme or ribozyme can be detected by monitoring the amount of cleavage products that are released as a result of it's activity. A light scattering particle(s) is attached directly or indirectly to areas of the molecular substrate such that the cleavage process is minimally affected. The presence and amount of free particles in solution or alternatively, the decrease of bound particles attached to the sample container, or to other particles, can be related to the presence, amount, and activity of the enzyme. In another example, light scattering particles have been coated with an antigenic substance and mixed with an antibody such that all of the particles are bound together by the antibody-antigen bond in a multivalent fashion. This network or agglutinated material is placed in or if desired attached to a sample container. A sample is placed into the container which may contain the analyte (which could be either the same antibody or antigen or a competing antibody or antigen of somewhat similar structure). Depending on the presence and amount of antigen or antibody specific analyte present in the sample, some fraction of the antibodies and particle coated antigens will become dissociated from the network structure by competition. The amount of analyte present can be detected by measuring the amount of particles in solution and/or the amount of particles that remain in the agglutinated network. This variation on the method can also be used by coating the antibodies on the particles, or using other binding agents as for example, nucleic acids, peptides, receptors, pharmaceutical agents, hormones and the like.

iii. Detection and Characterization of Molecular Binding Events

In another illustrative example, the Brownian motion of a particle that is coated with a binding agent can be used in an image analysis format to detect the presence and amount of an analyte present. This method can also be used to study the molecular binding event and properties in a binding pair where one partner is attached to the particle and the other is free in solution. Such capabilities are extremely important in characterizing the binding properties of antibodies, antigens, pharmaceutical agents, receptors and any substance where it's molecular binding properties are important. For example, a 40 nm gold particle preparation is made to contain either antigen, pharmaceutical agent, or antibody on the surface. These particle-binding agents are then placed on a microscope slide and viewed on a microscope using the methods of DLASLPD illumination and detection. Their Brownian motion properties and quantified. Then a sample solution which may contain an analyte that may bind to the attached binding agent on the particle is added. If the added solution contains the binding agent partner it will bind to the bound binding agent on the particle and a change in the Brownian motion can be observed. Alternatively, for characterization applications, known concentrations of the substance whose molecular properties are being characterization are titrated in at known concentrations to determine it's binding properties. In this fashion the molecular binding events of most any molecular recognition binding pair can be studied.

iv. Amplified Detection of Analytes

In certain analytical and diagnostic assays, it may be preferable to increase the detectability of the scattered light properties of the particles so that very simplified or no detection instrumentation is required. By use of the appropriate molecular recognition binding-pairs

and particles it is possible to significantly increase the level of detection sensitivity. Single-stranded homopolymer sequences, avidin-biotin, streptavidin-biotin, and other binding-pair systems can be used to "chain-together" and "build-up" many particles. As an example, a solid-phase assay is designed where a sandwich antibody-antigen-antibody structure is formed. One antibody is attached to a solid-phase so that the antigen analyte is captured. Additional antibody is then added which contains a biotin group. Then particles that are coated with streptavidin and free biotin are added to the solution. From the (solid-phase-antibody)-antigen-(antibody-biotin) complex grows a ...(streptavidin-particle)-biotin-(streptavidin-particle)-... structure which contains many particles bound together. Such a particle aggregate or network structure produces a high level of intensity which is much easier to detect than one particle. As another example, polydeoxyadenylic acid (Poly dA) and polythymidylic acid (Poly dT) or other homopolymer single-stranded nucleic acids can be used where Poly dA homopolymer sequence is incorporated into a region of the single-stranded "probe" molecule. Particles are coated with the complementary dT sequence to this homopolymer and are added to the sample with additional "free" dA single-strands to produce the structure containing many particles. The above examples are for illustrative purposes and one of ordinary skill in the art will recognize that there are many variations of this aspect of the invention possible depending on the analytical and diagnostic conditions and requirements.

Improved Particle-binding agent Reagents

The attachment of binding agents that are proteinacious such as antibodies to metal-like and non-metal like particles and other surfaces by the method of adsorption are well known in the art (see M. Horisberger, Scanning Electron Microscopy (1981),2, p9-31). The method

of adsorption can be used as for example, with antibody molecules to attach substances which have a binding property to the particle. In the case of antibodies, the attachment of the antibody molecules to the particle also confers partial chemical stability to the particle. If the adsorption conditions are carefully controlled, some of the antibody molecules will still possess binding activity towards it's respective antigen. The use of certain synthetic and biological polymers as chemical stabilizers for metal particles is also known in the art (see Heller et al.(1960), Journal of Polymer Science, 47, p203-217). The methods of adsorption of substances to particles and other surfaces are hereby incorporated by reference herein.

The exact mechanism(s) and nature of the adsorption of substances to particles and other surfaces is not well understood. When antibody molecules are adsorbed to a particle or other surface, the density and orientation of the adsorbed antibodies seem to be related to the level of binding activity. Due to the lack of control of the adsorption process, it is likely that many of the bound antibody molecules have become attached in such a manner that the molecular recognition region of the molecular structure has been altered such that the binding activity may become significantly reduced or possess no activity at all.

While the method of adsorption provides for the attachment of proteinacious binding agents and other substances which may or may not be useful in analyte assays to particles, it is difficult to attach some types of substances which may be of interest in analyte testing and other fields. For example, nucleic acids, smaller protein and protein-like substances such as peptides, and other non-proteinacious substances such as low molecular weight antigenic substances, hormones, pharmaceutical agents and the like can not effectively be attached to particles by the adsorption process. Further limitations

of the adsorption technique are that there are unique adsorption conditions for each type of substance which must be carefully controlled. Even when such procedures are followed rigorously, there can be significant variation in the amount of protein and the integrity and binding properties of the substance that has been adsorbed to the surface. In many cases the binding activity (affinity and specificity) of the adsorbed binding agent is significantly reduced as compared to the unadsorbed form.

Our experience with attaching various proteinacious binding agents such as antibodies to the surface of the particle by using the method of adsorption have shown us that there is great variability in the binding properties and stability of the resulting particle-binding agent materials. The binding affinities of the adsorbed antibodies or other binding agents are highly sensitive to the labeling conditions and can also vary significantly from batch to batch. A significant decrease in the binding activity of antibodies, avidin, streptavidin and other binding agents that have been adsorbed to the particle is common. In some of the preparations, it appears that some fraction of the adsorbed binding agent is prone to dissociating from the particle. This can pose serious problems as this dissociated material will compete against the particle-binding agent for analyte in an analytical or diagnostic assay.

Such lack of control of the attachment process, variability in binding activity, and limitations as to what types of substances that can be attached to particles by adsorption methods poses many problems for the production and use of such materials for analytic diagnostic purposes. Perhaps most importantly, particle-binding agent conjugates prepared by the adsorption technique may not be of sufficient quality for many analytical applications where very or ultra low concentrations of analytes are being detected.

It would be of great use in the art to have a method where any type of substance including binding agents of varying size and composition could be attached specifically to a particle or surface whereby the binding activity of the attached substance is minimally affected. It would also be of great use in the art to have a method for achieving a desired density of agent per particle (or, in general any surface). In addition, it would be desirable for these methods to allow binding of more than one type of agent. From a manufacturing and cost standpoint, it would be of great utility if the synthetic procedures are easy and inexpensive to perform such that a wide variety of different types of substances can be attached to particles using the same basic procedures.

We have developed new methods that allow for the specific attachment of binding agents and most other substances to metal-like particles and other surfaces. The particle reagents produced by these new methods are highly stable and possess high binding affinities with low non-specific binding properties. These new methods overcome many of the limitations of the prior art adsorption procedures with the additional benefit that the procedures are easy to perform at low cost. In some embodiments, these new procedures allow for a universal linker chemistry platform where almost any type of substance can be quickly and simply attached to a particle or surface using many of the same materials and procedures. This is extremely important in the day to day manufacture of such particle-binding agent reagents for use in analyte testing.

The following procedures apply to any substance which includes binding agents or other substances as for example, antigens, antibodies, lectins, carbohydrates, biotin, avidin, streptavidin, nucleic acids, peptides and proteins, receptors, pharmaceutical agents and the like. The methods can be used to attach most substances to metal, metal-like, and some non-metal like particles and

macroscopic surfaces. For example, non-metal-like surfaces and particles include materials that may be composed of organic or inorganic material such as glass, plastics, and the like.

Methods of Attachment of Substances to Particles and Other Surfaces

i. Base Material Molecule Method

In this method of attaching substances to particles or other surfaces, a two step approach which involves the use of base material molecules is used. Suitable base material molecules are any substance which can approach and interact with the surface by adsorption or other chemical process, and have accessible functional groups to which additional substances, as for example, binding agents can be attached. The base material molecule may also have the additional property of conferring chemical stability to the particle. Generally the base material molecule is of a macromolecular form (size >1000 MW) but it can be smaller or much larger. Preferred base material molecules are those which attach to the particle with high affinity, confer some level of physical stability to the particle, and possess accessible chemical groups which are easy to conjugate most any substance thereto. The chemical groups allow for binding agents or other substances to be linked either through chemical, covalent, or non-covalent attachment. For example, covalent attachment could involve photochemical or chemical attachment. Non-covalent attachment could involve cross-linking with molecules such as streptavidin, or by adsorption through hydrophobic, hydrogen bonding, or electrostatic interactions. The base material molecule may also contain one or more chemical groups which can be used to cross-link several base unit molecules together across the surface of the particle utilizing the appropriate chemical or cross-linking agents.

The following are selected examples of how the method of base material molecule attachment can be used to create particle-binding agent reagents which are highly stable, possess high binding affinities for the entity(ies) they bind to, and provide for a highly flexible, easy to use and low cost method of attaching most any substance to particles or other surfaces. One of ordinary skill in the art will recognize that there are many variations of the general technique to synthesize particle-binding agent reagents for most any purpose. Using this new method, antibodies, peptides, proteins, nucleic acids, pharmaceutical agents and most any other substance can be attached to the particle in a highly controlled and predictable fashion.

As an example, we have used a derivative of a polyethylene glycol compound of approximately MW 20,000. The properties of this molecule (bis(Polyoxyethylene bis[3-Amino-2-hydroxypropyl])) allow for it's use as a base material molecule. Each molecule of this polymer has four amine groups which can serve as linkage sites for the conjugation of additional substances. The hydrophobic backbone of the polyethylene derivative interacts with the particle and is attached to the particle surface by adsorption or some other process. This interaction is very strong, as we have not detected any of this material dissociating from the particle surface following labeling and use in analytical and diagnostic assays. The amine groups do not appear to interact with the particle surface and are accessible as conjugation sites for the attachment of additional substances as for example, binding agents. Using this polymer as the base molecule we have prepared two different types of particle-binding agent reagents. One reagent contains biotin groups as binding agents and the other particle-binding agent reagent was made to contain single-stranded nucleic acids as binding agents. The biotin used for attachment was a chemically modified form where it will covalently link to amine groups. For

the nucleic acids, the 5' ends were chemically modified so that they would chemically react with the amine groups. In our use of these reagents in various assay formats we have observed that both of these particle-binding agent reagents demonstrated a high degree of stability in low and high salt aqueous solutions with exceptional binding activities. In experiments where the particle-biotin reagent was used no effect upon the binding affinities was detected. This was determined by placing the concentration of the particle-biotin reagent at concentrations of 6×10^{-14}M in suspension and submerging a plastic solid-phase that was coated with avidin into this solution. After a couple of hours of incubation the solid-phase was removed and washed. When examined under the light microscope using DLASLPD methods of illumination and detection, particles were detected specifically bound to the avidin coated solid-phase while the control solid-phase (which contains no avidin) showed no particle binding. At these working concentrations of particle-biotin reagent, if the binding properties of the biotin attached to the particles was substantially decreased, no binding would have been visible.

In another example, gelatin is used as a base material and the gelatin can be cross-linked on the particle surface by use of chromate ion or other cross-linking agents to minimize the chance of desorption. Binding agents or other substances are then linked to the particle by using the appropriate conjugation chemistries for attachment of these substances to accessible amine, carboxylic or other chemical groups to which attachment can be accomplished.

In another example, streptavidin or avidin can be used as a base material. Substances such as binding agents and the like are attached to the particle by using a chemically modified version of the molecule which contains at least one biotin group.

In a further example, polymer-like materials and other materials which possess polymer-like properties, as for example, carbohydrates, polyamino acids, proteins, and the like can also be polymerized from co-polymer units in solution right onto the surface of the particle under appropriate conditions.

For all of the above examples, one can also first conjugate the binding agents or other substances to the base material and then apply this material to the particle surface with or without chemically cross-linking the base materials together. In addition, two or more different types of base material molecules, or one or more base material molecules can be used with other chemical stabilizer molecules such that the amount of chemically reactive groups available for conjugation and the chemical stability of the particle-binding agent conjugate can be adjusted to suit most any analytical need.

In the examples above, available materials were used and selected for use as base material molecules. One skilled in the art can synthesize new types of base material molecules to further optimize their use for attachment of substances to particles and other surfaces. The following improvements allow for particle-binding agent reagents which are more chemically stable, and optimization of the conjugation process with enhanced performance with regard to binding affinities of the attached binding agents or other substances. For example, additional chemical groups can be added to the backbone structure of the polymer which increases the stability of the binding of the base material molecule to the surface of the particle. Linker arms of various lengths with appropriate reactive chemical groups at the end or close thereto can be added to increase the distance from the particle at which the binding agent or other substance is attached and ultimately resides. Different types of reactive chemical groups can be added to the base material to further improve the ability to cross-link or otherwise

attach the individual base material molecules together across the surface of the particle.

ii. Direct Attachment of Substances to Particles or Other Surfaces By Means of Chemical Groups Which Adsorb to Metal Surfaces.

We have developed additional methods which allow for direct attachment of many different types of substances, including binding agents, to be attached to metal and metal-like particles and surfaces. In the art of material science and related fields, it is known that certain types of small molecules (<1000 MW) can be attached to metal surfaces and the like. For most of these small molecules there are certain types of chemical groups at specific locations within the molecule which allow for one part of the small molecule to become bound to the metal surface while other parts are not bound to the surface. For example, the adsorption of thiol and disulfide containing substances, and amphiphilic substances, such as n-alkonic acids and certain detergent molecules onto metal surfaces is known in the art of material science (see Nuzzo et al. (1983), Journal of the American Chemical Society, 105, p4481-4483; Allara et al. (1984), Langmuir, 1, p45-52; and Bain et al. (1989), Journal of the American Chemical Society, 111, p321-335). The methods of adsorption of substances to metal surfaces are hereby incorporated by reference herein. Therefore, the properties which allow for attachment of the above substances can be conferred onto binding agents and other substances by incorporating the appropriate chemical into specific location(s) within the molecular structure of the substance that is to be attached. Certain types of substances will be easier to attach with this method than others. For example, substances whose molecular structure is charged or ionic, or is polarized such that at one end of the molecular structure it is hydrophobic while at the other end it is

hydrophilic will generally be useful in this particular variation of the method.

For example, nucleic acids contain a phosphate backbone which contains a high negative charge. A single-stranded nucleic acid is end labeled with a thiol or disulfide at the 3' or 5' end with or without additional hydrophobic groups incorporated into the same region of the molecule. This modified nucleic acid will bind to the metal surface or particle at the end labeled with these groups. The ionic part of the nucleic acid's molecular structure away from the surface such that it is accessible for molecular interactions with most any substance that can specifically bind to it.

Other substances such as biotin, peptides, pharmaceutical agents, polymers, and the like can be attached to the particle using this method. The method is generally useful for most substances which do not interact significantly with the particle or surface in their native form. For substances that may interact with the particle or surface additional methods are required. For example, certain small molecules, proteins and the like may interact with the particle or surface such that their binding activity is diminished. In one variation of the method, the particle is first labeled with, for example, a polymer stabilizing agent. Following this labeling, there are usually open areas on the surface of the particle to which small molecule entities can bind. The appropriately modified substance is then added to the chemically stabilized particle to confer a desired binding activity or other property. Alternatively, the chemical stabilizer and chemically modified binding agents can be mixed together in a desired ratio prior to mixing with the particle or surface. By using these methods, the amounts and types of substances that are attached to the particle or surface can be controlled to yield a coated surface or

particle with the desired chemical stability and binding activity properties.

Linker arms of various lengths and composition can also be incorporated into the molecular structure. For example, a small molecular weight base material molecule can be used where it's molecular structure is optimized for attachment to the particle or surface, attachment of most any substance to it with any desired orientation, and with a high level of binding activity. As an example, a linear polypeptide twenty amino acids in length is chemically modified at one terminus by the addition of disulfide or thiol chemical groups. The native polypeptide is composed of amino acids such that the polypeptide chain will not interact with the surface except through the chemically modified end. At the other terminus a free amino group exists, or alternatively, has been chemically modified for a desired conjugation process such that most any substance can be attached at this position. This low molecular weight base material molecule then is used in one or more variations of the methods as described herein.

The method of base material molecule conjugation and the method of direct attachment as described herein allows for the more specific control of the amounts, types, and orientations of substances that can be attached to particles and other surfaces. A further advantage is that these methods provide for the synthesis of particle-binding agent reagents where the binding affinity of the attached binding agent remains at high levels.

An important feature of the base material molecule method utilizing either small molecular weight or larger molecular weight base material molecules is that with the proper selection and utilization of base material molecules, the base material molecule can serve as a universal linker platform to which most any substance can be attached to a particle or surface. This capability becomes extremely important in the day to day manufacturing of particle based reagents for testing of

analytes. One of ordinary skill in the art will recognize the many different variations of these new attachment methods that can made by varying the chemical groups, molecular weights, molecular structure, labeling reaction conditions, and the type of conjugation chemistry (i.e. cross-linking, covalent attachment, etc.) that is used.

Microarray or Micropattern Assays with Light Scattering Particles

The microarray or micropattern method of analysis uses discreet spatially addressable areas of a solid-phase to detect different types of analytes. For example, each spatially addressable area or microspot may contain a different type of antibody, receptor, nucleic acid or the like. The arrangement of the spatially addressable areas on the solid-phase is dictated by the size of the solid-phase, the number of analytes or different areas that will be used, and the method of detection. Each of the spatially addressable microspots which contain a particular type of binding agent may be shaped as a square, circle, or any pattern depending on the methods used to make the microarray. The dimensions may be from a few square microns to square millimeters or even larger. The microarray method can be implemented using any one of the many solid-phase formats that are used for single analyte detection and in which the final quantification is done by measuring a solid-phase signal that is related to the amount of analyte bound to the solid-phase. In practice, the general analytical steps for the microarray method are as follows. The microarray is exposed to an analyte sample, e.g. serum, and after a suitable incubation period the array is washed and exposed to a second analyte binding entity. In one format, the second analyte binding entity is bound to the light scattering particles whose light scattering properties are detected. The number of light scattering particles attached to each microspot is then a measure of the amount of analyte

present in each microspot and can be correlated with the concentration of the analyte in the sample. In another format, the second specific analyte binding entity is not bound to the light scattering particles. In this latter format, a third entity, that binds specifically to the second specific binding entity, is bound to the light scattering particles. This third entity, for example, can be streptavidin which binds specifically to biotin covalently attached to the second entity. There are many other assay methods which can be used for detecting the second entity with the third entity bound to the light scattering particles. In any of these formats, the amount of analyte bound to each microspot is established by measuring a light scattering signal that is related to the number of light scattering particles bound to each microspot.

Different methods can be used to detect the number of light scattering particles on each microspot in a microarray. The amount of analyte bound to each spot is established from the number of light scattering particles attached to each spot in the final assay step. In general, some type of imaging system is needed to separate the light scattering signals from the different areas in the array. Many different types of formats can be used for imaging and particle quantification. The method of choice depends on the precision that is required and the number of samples to be analyzed per day. The needed precision can range from low as in the cases where only a positive or negative type of answer is needed to very high precision in cases where the amount of analyte has to be determined with a precision of a few percent. Examples of different imaging and particle quantification formats are now described.

Special features can be introduced into the microarray for any imaging method for example, the chemistry of some of the microspots in the array can be formulated to yield a background signal, or the chemistry

of some of the microspots in the array can be formulated to serve as calibrating spots containing known amounts of analyte. The signals from these spots can be used to correct for variations in incident light intensity, light transmission between multi-microspot array carriers, light collection efficiency and photodetector sensitivity from one sample to the next.

Some specific imaging and light-scattering particle quantification methods for applications to microarray and array chips are now described.

a. The DLASLPD Method with Simple Light Microscope

i. Low particle surface density (less than 0.1 particles per μ^2) on a spot.

If the number of samples to be examined is not high, the number of particles in each spot can be determined by visual or other counting methods of the number of particles on each spot. A background count is also made. The counting can be done on liquid covered or dry microarrays. The number of particles per microspot which is considered to be positive is defined by previous test experiments. If many samples are to be examined, the counting can be done automatically using a simple video detector and object counting software.

ii. High particle surface density (greater than 0.1 particles per μ^2) on a spot.

For positive or negative types of analysis, the intensity from each spot can be detected by visual observation or photodetection. A Result is positive if the intensity is higher than that of the background. If a quantitative result is needed and there are not too many samples to be examined (for example, bedside, field, small clinic, or research lab testing) , a manual technique with a two observation port microscope can be used as follows. A single microspot is illuminated with a narrow beam of

light. The beam is positioned on the spot by visual observation through one observation port and the intensity is measured quantitatively through a photosensitive device with or without spatial filtering aperture depending on the level of stray light signal. The scattered light intensity from each spot is measured by manually scanning each spot through the beam. Alternatively, the beam could be scanned manually and the light detected from each spot detected by a large area photodetector or a small area detector in which the detector area is kept confocal with the illuminating spot. This could also be automated. If many samples are to be analyzed, the microarray can be illuminated with a broad beam of light and an image of the microspot array digitized through a video camera and frame grabber. The intensity of each microspot is then determined by software image analysis. We have determined that these methods allow for very sensitive and wide concentrations of one or more analytes in a sample to be detected. One skilled in the art will appreciate that many other variations of the method are possible.

Use of Certain Types of Metal-like Particles in Microarray and Array Chip Detection of Analytes

In our work with microarrays, we have found that metal-like particles are preferred light scattering particles. The size, shape, composition, and homogeneity of the particle type(s) used in a specific microarray application depend mainly on the following; the amount of non-specific light background in the sample; if the microarray is dry or covered with liquid; the dimensions of the discreet solid-phase binding areas; amount and concentration ranges of the analyte(s) that are detected; detection by eye or by photodetector, and measurement by particle counting and/or intensity measurements.

As an example, we were easily able to detect the binding of individual 60nm diameter gold particles coated with BSA-biotin to 80 micron diameter spots containing

streptavidin on a plastic solid-phase in a microarray format covered with buffer solution. We used our custom illuminator under DLASLPD conditions and the inexpensive microscope system we developed. In microarray streptavidin microspots with lower densities of bound 60nm diameter gold particles coated with BSA-biotin, we counted the number of particles bound. At higher densities, we measured the intensity of the scattered light arising from the particles bound to the individual streptavidin microspots. We detected particle densities down to about 0.06 particles per μ^2 at a signal to background ratio of 13. This implies that for this type of assay, densities down to about 0.015 particles per μ^2 can be detected at a signal to background ratio of about 3. Very high densities of bound particles were also detected (saturation of available binding sites per individual microspot of 80 micron diameter). To perform the same type of microarray assay in a dry form (not covered with liquid), the use of larger diameter gold particles or other metal-like particles with greater light scattering powers may be required to achieve the same sensitivity. Also, using a longer wavelength light source such as a HeNe laser with illumination above 600nm and spatial filtering may also be useful.

For detection of samples by use of a small handheld or other type of portable device, even larger particles may be needed to be used depending on the level of sensitivity required as typically one skilled in the art must use low power light sources in such a device.

For multi-analyte detection in the microarray format, the concentrations of different analytes may exist at very different levels with differences of 1,000 to 1,000,000 or even greater in concentration. In such situations, the light scattering power and the relative size of the particles become very important. For example, if one is analyzing multi-analytes on an array chip or microarray where the individual discreet binding areas are about 100

square microns, the number of particles that can be bound to this 100 micron square area is highly dependent on the size of the particle used. For example, if a 40 nm particle is used, at binding saturation, about 79,600 particles can be bound to this area. However, if a particle of 120nm is used, only about 8,800 particles can be bound to the area. Depending on the amount of non-specific light background and non-specific binding of the particles, the minimum number of particles that must be bound to the area for a reliable measurement can be quite variable. For example, in certain situations a few thousand or more particles may be needed to be bound to the binding site area on the microarray in order to get a positive detection result. Using larger particles thus limits the detectability of the analyte. In binding site areas of small dimensions, the smallest particle that can be used which gives adequate signal/background should be used. In addition, optical and spatial filtering, confocal imaging, more powerful light sources and other instrumental components can be optimized to increase the detection limit. Similarly, if two or more of the analytes exist at very different concentrations, then different types of particles with the appropriate size and light scattering power may be needed to be used.

These examples are not meant to be limiting but to show how in various applications, the selection of certain types of metal-like particles leads to specific test kits for microarray analysis and detection of multi-analytes. One skilled in the art will recognize that there are many other variations of the method of the present invention to detect multi-analytes on array chips and microarrays.

Use of Certain Aspects of the Present Invention with other Illumination and Detection Methods

This discovery means that you can use various aspects of the present invention in the art existing diagnostic detection methods and apparatus even without using the

optimal light and detection methods and systems as the present invention has disclosed. For example, laser confocal microscopy methods, brightfield and epi-illumination microscopy methods, and the methods of reflection contrast and differential interference contrast microscopy can be used with certain types of metal-like particles for measurement of multiple analytes on microarray chips and the like.

For example, a confocal microscope as described by Ekins in U.S. Patent 5,432,099 (hereby incorporated by reference) can be used. Generally, such confocal microscopy relies on point illumination rather than field illumination and usually works in a fluorescence mode with epi-illumination. Usually the detector is a photomultiplier because the signal to be detected is very low. The light source is frequently a laser. In the present invention even though the signal is extremely high (compared to normal confocal microscopy) and the light source need not be a laser, an apparatus as complex as the confocal microscope can still be used. Clearly, the use of such an apparatus provides even more sensitive detection of particles as described in this invention as we have found, and minimizes stray light problems.

Thus, in another example, the methodology of Fodor et al. in 364 Nature 555, 1993 for detection of target molecules using biological chips can also be used.

These methods when combined with the use of one or more aspects of the present invention are useful in certain microarray analysis applications where the cost and ease of use of the method and apparatus are not of concern. We are not aware of anyone using these above mentioned art existing techniques with metal-like particles and/or the method of refractive index enhancement and/or autometallography and/or any other aspects of the current invention. We thus claim use of these previously described art existing detection methods

and apparatus with one or more aspects of the present invention.

Other Adaptations of the Invention for Applications to Microarrays

The methods of the present invention provide an excellent means for detection of one or more analytes using a microarray format. The following methods provide additional variations which are useful in certain analytical applications.

We can miniaturize our illumination and detection methods such that a single or multi-optical fiber based apparatus is constructed. This provides an alternative to using the imaging methods for detection.

One problem in using a two-dimensional array, or other type of solid-phase spatially addressable system is the problem of signal crosstalk between different areas of the microarray. Crosstalk, glare or other similar problems can arise from several sources, for example, (1) the individual areas containing the light scattering or fluorescent material(s) are so close together that they appear as one area, or (2) one area contains a high amount of light scattering particles or fluorescence while other adjacent areas contain very low amounts of light scattering particles or fluorescent materials. Depending on how close the areas are to each other, some portion of light coming from the highly intense areas will be picked up by the detector in the areas from which lower light intensities are coming from.

One potential solution in the art is to illuminate each spatially addressable solid-phase site separately by using a scanning process and recording the light signals coming from each spatially addressable site separately when it is illuminated. This can be accomplished by scanning the different areas one at a time by moving the illumination beam, or the sample. However, these scanning mechanisms are usually complicated and add a tremendous

amount of cost and complicated procedures to the analytical method, and may be too costly and not robust enough for the everyday rigors of a clinical testing laboratory or highly active research laboratory.

An example of a further variation of the present invention is now described. An optical fiber at one end is beveled to a suitable angle and is used as a discreet illumination source such that when brought close to an area to be measured, the emitted fluorescent or scattered light from the area is detected from the other side of the sample surface. This configuration allows for the specific illumination of an area and eliminates the above mentioned problems of crosstalk. It also removes the requirement for an imaging-type of detector such as a video camera, and any type of photodetector can be used. As an example, for an array of 24 microspots or distinct areas to be measured, 24 individual illumination fibers are used, one for each spot. All that is required is that the individual spots are illuminated at different times with respect to one another. Several small spatially addressable areas, down to about ½ the diameter of the optical fiber can be measured in this fashion.

In another embodiment of the method, where the use of epi-illumination or similar methods are desired, as for example, confocal imaging, one could miniaturize the system by placing at the end on an optical fiber a very small imaging lens and then achieve confocal conditions where the scattered or fluorescence light can be measured from the desired region of the microarray. For any area to be measured on the microarray surface, a single optical fiber is used with a microlens to deliver the incident light and collect the emitted fluorescent or scattered light that is to be detected. Multiple optical fibers could be used as described above if desired to detect more than one area of the surface at a time.

One of skill in the art will recognize that the preceding illustrative examples are only a few of the many variations of the present invention that are possible.

Screening of combinatorial synthesized molecule libraries

A tremendous problem currently in the burgeoning field of combinatorial synthesis of important molecules is the lack of highly sensitive, practical, and easy to use signal and detection technology and assay formats to detect the few copies of newly synthesized combinatorial molecule(s).

We have determined that our signal and detection technology is easily used on solid-phases that contain spatially addressable sites such as 2-dimensional arrays or any spatially addressable solid-phase. Therefore, our methods in one form or another are directly applicable to the screening and detection of one or more than one class of combinatorial or biocombinatorial molecule(s) in this type of format. The assay method can be any of the known procedures in the art.

The invention as we have described herein can also be used to detect and quantify one or more specific combinatorial, biocombinatorial, or otherwise synthesized molecules on a spatially addressable solid-phase. For example, it is well known in the art that a wide diversity of biocombinatorial and combinatorial molecules can be synthesized by using the methods of "split synthesis", "parallel synthesis", and related methods (all of these methods are incorporated herein). Typically, a wide diversity of combinatorial molecules are synthesized on small particles or other pieces of solid substrate where each particle or substrate piece contains one unique set of combinatorial synthesized molecules. There are problems of identifying and purifying those pieces or particles that contain the "active" sets of synthesized molecules in the art.

There are several ways to utilize our signal and detection system to detect these specific and desirable combinatorial product(s). In one assay method, a binding agent(that is specific for the desired analyte) is coated on a selected type of metal-like particle. When the coated particle is added to the sample, it binds to the analyte. Alternatively, an indirect method involving the use of biotin labeled binding agent first binds the analyte, which is then detected by adding streptavidin coated metal-like particles prior to detection. The light scattering particle is bound in one form or another to the desired analyte of interest which resides on the synthetic solid phase. In this manner, the desired molecules are identified, isolated, and purified from the sample by filtration, centrifugation, dialysis or any other art known technique. Alternatively, binding agents labeled with particles coated with binding agents can be added such that aggregates or networks are formed between the specific molecule-containing synthetic particles and the metal-like particles. Similar means as described above are used to identify and purify the desired molecules.

Multi-analyte analysis for different combinatorial synthesized molecules is accomplished by using two or more types of metal-like particles each coated with a different type of binding agent. Refractive index enhancement and DLASLPD video-enhanced contrast methods can also be used.

In another assay method, the metal-like particles also contain a composition of a ferro-electric or magnetic composition such that these particles can be manipulated in three dimensional space by using an applied EMF to the reaction container. In this manner, the substrate particles containing the "active" combinatorial molecules can be easily purified and detected from all the other material. It should be noted that a mixed composition of ferro-electric or magnetic and other metal-like compositions of specific particles are also very useful in many other fields including diagnostic assays, and for

isolation and purification of desired molecules. The use of refractive index enhancement methods in combination with the above methods increases the sensitivity of detection.

Metal-like Particles Used as Solid-Phase Synthetic Supports

Metal like particles when coated with appropriate substances, are excellent substrates for conducting chemical or biochemical synthesis as for example, combinatorial synthesis. The specific coating of metal-like particles can be made to consist of for example, polymers, antibodies, proteins, nucleic acids, amino acids, reactive chemical groups, and the like. For example, metal-like particles are coated with a polyethylene glycol compound containing chemically reactive amine groups. Synthesis is then initiated on these amine groups which are numerous on the surface of metal-like coated particle. Other reactive chemical groups or groups that can be specifically activated can be used instead of the amine group. In another example, amino acids, or small peptides are coated directly on the surface of the metal or metal-like particle, or are chemically linked to polymer or other type of macromolecules that are coated on the surface of the metal-like particle. Synthesis is then initiated on the metal-like coated particles. In yet another example, reactive groups are attached to the surface of the metal-like particle so that protein, nucleic acid, chemical, or biochemical synthesis can be performed. The number of reactive groups on the surface of the particle can be also modified as follows. A mixture of polyethylene glycol compounds (MW 20,000) with and without reactive amine groups (or other reactive groups) are mixed in an appropriate ratio to achieve a desired number of reactive groups on the surface per particle. In this manner, the metal-like particle is coated with a specific amount of

chemical synthetic sites, or binding sites per metal-like particle. The specific number of sites and the type of reactive groups sites can be varied to suit any particular need as for example, further chemical synthesis or for diagnostic reagent purposes. For example, for diagnostic-type applications, adding a discreet number of specific binding agent molecules per metal-like particle may be important to achieve the desired assay performance. In addition, two or more different types of reactive synthetic or binding sites can be placed on the same metal-like particle in specific amounts utilizing the same approach as described for the polyethylene compound above by mixing in appropriate ratios the desired substances (i.e. different binding agents or chemical groups, etc.). These types of coated particles may be useful for example, to isolate, purify and detect two or more different molecules using the same particle. The high densities (grams/cm^3) of many types of metal-like particles also offers many advantages in the purification, isolation, and identification of molecules of interest. MLSP type particles offer the further advantage of more easily manipulating the particles within the medium. The above examples are only a few of the many possible variations of this method. Other variations will be apparent to one skilled in the art.

Practice of Various Aspects of the Invention Outside the Field of Analytical Diagnostics

The present invention features methods to detect one or more analytes in a sample by detection of the scattered light properties of a particle. It should be noted that various aspects of the invention as disclosed herein are directly applicable to many other specific applications outside of the diagnostic field. One skilled in the art or the art of other fields such as optical information and storage, image formation and processing, electrical-optical signal transduction and switches, telecommuni-

cations, information transducers, and many other related applications has been enabled by this disclosure to practice various aspects of the present invention specifically to solve problems and create new products in fields outside of analytic diagnostic assays.

One aspect of the current invention that is very useful for applications to other fields is the ability to identify specific metal-like particles of certain size, shape, composition, and homogeneity by a unique optical signature which is characteristic of that type of particle. Embodying such specific optical signatures in a very small structure allows for these particle signal agents to be used in numerous fields. For example, they can be used in industrial quality control, markers or labels to identify or trace any product, material, substance, or object that contains the particle. The particles in one form or another can be used as a means for identification and the like similar to the art known method of "barcoding". For example, a coating containing one or more types of particles can be applied to consumer products to identify the authenticity, date, or other relevant information. Similarly, paper currency, stock and bond certificates and the like can have coated on the surface or embedded within the paper material itself certain particle types that can be detected to determine the authenticity of the object. Other examples include placing small amounts of a specific type of particle inside prescription or over the counter drugs to authenticate or trace of drug. In addition, the particles can be used as environmental, industrial, pharmaceutical, or biological tracers to study the physical properties of a system such as disposition of fluids, materials, and the like. One skilled in the art will recognize that these are just a few of the many possibilities.

Another aspect of the current invention which is directly applicable to other fields is the use of light scattering particles which can be physically manipulated

by electric, magnetic, or related fields. We name such particles Manipulatable Light Scattering Particles (MLSP) and these are described in detail later. Such MLSP particles can be oriented into various arrangements in one-, two-, or three-dimensional space by using a magnetic, electric, or related electromagnetic field (EMF). In this way the unique light scattering properties of the particles can be used to form certain patterns, images, or colors. The specific orientations of one or more MLSP particles are used to store or transduce information via the light scattering properties of an individual particle, or the resulting light scattering information, that is, optical signature of two or more particles arranged in a particular orientation. For example, three different types of particles that scatter blue, red, and green light are placed inside a small area or volume such as a "pixel" in a screen that contains a specified number of pixels in a two-dimensional array. The screen forms a color or black and white image, or moving picture that is similar to the appearance of a television image, video image, motion picture image and the like when the screen is illuminated with white light. Each pixel or groups of pixels is spatially addressable by an electric or magnetic field (EMF) such that by applying the appropriate EMF, the individual particles that scatter blue, red, and green light are oriented to produce a specific color with a certain hue and intensity when appropriately illuminated. For example, at one applied EMF, the red and green particles are concentrated to a very tiny spot while the blue scattering particles are freely dispersed within the inner volume of the pixel. This pixel will then appear blue. A different EMF can then be applied to cause the same effect for the red or green light scattering particles. In this way, by specifically orienting the different particles in each pixel, the desired color image is produced. This method and apparatus

offers many attractive advantages to the current cathode ray tube based image formation technology and the like.

In another example, MLSP particles are switched from one specific orientation to another by appropriately adjusting the EMF. For example, asymmetrical silver particles which can produce green or red scattered light and/or also have two different levels of intensity of scattered light are used as follows. One or more particles are placed in a specific location in either a liquid type or solid type of material where the particle is able to rotate, that is, re-orient itself an EMF field is applied to the material or device containing the particles. Depending on how many particles are used and the desired function of the device, the different orientations of the particles will signify different types of information. For example, in one orientation, the light scattering properties of the asymmetric MLSP particle(s) indicate the "off" or number 0 in a binary code system, while in a different position or orientation the light scattering properties indicate the "on" or number 1 in a binary numeric system. The orientation of the asymmetric MLSP particles is changed by varying the EMF to achieve the desired orientation of the particles in the material or device. When light interacts with the particles in a specific orientation, the properties of the scattered light signify a specific type of information as described above. In this manner, simple and multi-component optical switches that are useful in telecommunications and related fields can be made. Similarly, a series of these switches can be assembled in a serial or parallel fashion for more complex information storage and handling.

New types of information storage devices can be made by encrypting or storing the information by using different types and/or orientations of light scattering particles and MLSP particles. For example, an optical storage disk can be made that is similar to what is known in the art as a "Compact Disk" or "CD-ROM" Disk or the

like. Instead of using bumps which project above the surface to encode the information, light scattering particles are used. The particles can be placed on or in any material from which the light scattering properties of the particles can be detected. In this manner much more specific information and storage of higher densities of information are possible.

One skilled in the art will recognize the many different types of devices that can be built using various light scattering particles in a particular application. The above examples are just a few of the many ways such metal-like and MLSP particles are used outside the field of analytic and diagnostic detection. These applications are enabled by the disclosure herein and applicant hereby claims right to the practice of various parts of the invention as described herein to fields outside the field of diagnostic analytic assays.

Description of the SpectraMetrix Spectrophotometer Instrument for Liquid Samples and Theory of Operation

The SpectraMetrix Photometer is a right angle photometer that measures the intensity of light which is scattered or emitted light at right angles to the exciting light beam. A schematic diagram of the instrument is given in Figure 21. The light source is a microscope illuminator or any other type of light source. The instrument can be used with or without the monochromator. Adapters for connecting different light sources are provided with the photometer. Scattered or emitted light is detected by a photomultiplier (PM) tube. The photometer has a manual light shutter to keep light from reaching the PM tube while changing samples. Optical filters or polarizers are introduced in the incident or emitted light path as required. Cylindrical cuvets (e.g. test tubes) of different diameters are used as sample cuvets. However, any type of optically transmissive sample container may also be used with the appropriate holder. Test tubes of a

diameter of 6 mm by 50 mm were used and a microscope illuminator with an infrared (heat) filter were used to obtain the data reported herein.

The illuminator can be connected directly to the spectrophotometer or through a monochromator. The monochromator used for the measurements reported herein is a diffraction grating monochromator. The illuminator is powered by a regulated DC power supply at 6 V and 3 Amps.

In this paragraph we describe the optics of the instrument that we used without monochromator operation. A x10 objective focuses the light from the illuminator unto the sample tube. A light collecting lens (23 mm focal length, 19 mm diameter), that is positioned at right angles to the exciting light beam (about 1.5 inches from center of sample cuvet), focuses an image of the sample tube at a distance of about 106 mm from the center of the sample tube. This distance allows a light shutter and filter holder to be placed between the collecting lens and the PM tube. A diaphragm with a 3.25 mm diameter hole (made with a #20 drill bit) is placed at the image plane. The PM tube is placed behind the diaphragm. The diaphragm blocks the light reflected from the walls of the cuvet and allows only light scattered from the center of the sample volume to reach the PM tube. The diaphragm, while reducing the amount of scattered light that reaches the PM tube, maximizes the signal to background ratio. In order to further minimize the detection of light reflected from the sample tube, the tube is positioned at an angle of about 40-50 degrees with respect to the vertical direction so that reflected light does not reach the collecting lens. Because of this angle and refractive index effects, the light emerging from the tube does not travel along the center axis of the collecting lens and the scattered light beam at the image plane is displaced downward from the center axis of the collecting lens. This requires that the 3.25 mm aperture and PM tube be displaced downwards from the collecting lens axis. The instrument is constructed

such that the downward displacement can be manually adjusted for the most efficient scattered light detection.

When the monochromator is used, the optics are the same as above except that an additional lens (23 mm focal length, 19 mm diameter) is positioned between the 10x objective and the monochromator exit slit. The lens is 4 inches from the center of the sample cuvet. The exit slit of the monochromator is 5.6 inches from center of sample cuvet. The illuminator is connected to the adapter at the entrance slit of the monochromator.

Adjustment of the photometer optics

a. Place a 60 nm, 4 x 10^{-12} M gold sol in a 6 x 50 mm culture tube in the sample holder of the spectrometer. Adjust the angle of the tube with respect to the vertical so that it is between 40 and 50 degrees. Position the angled tube so that the focused exciting light beam crosses through the center of the cuvet. Do not allow the exciting light to strike the front surface of the tube (surface towards collecting lens) as this will increase the amount of light reflected towards the detecting system.

b. The distance of the collecting lens from the center of the sample tube is adjusted so as to form a sharp image of the walls of the tube at a distance of 106 mm from the tube center. The image of the scattered light beam and the walls of the sample tube can be seen clearly on a piece of white paper placed at the distance of about 106 mm from the tube center. The tube image has a diameter of about 8 to 10 mm at the image plane. The lens should be positioned so as to obtain a sharp image of the walls of the cuvet. The scattered light beam appears a little fuzzy at the image plane because of its finite width. The best position of the lens is about 1.5 inches from the center of the sample cuvet. The exciting beam can clearly be seen as it crosses a scattering solution.

c. The above adjustments of the collecting lens position are performed with the structure that contains the shutter, filter holder and diaphragm holder removed from the instrument. After the lens is correctly positioned, the latter structure is replaced and the light blocking diaphragm with 3.25 mm opening is inserted. The PM tube is inserted into place.

d. After steps a, b, and c, the position of the aperture with respect to the collecting lens optics is adjusted as follows. Place a piece of white paper in the place where plane of the PM photocathode will be positioned when the PM is inserted. With light scattering gold particles in the sample compartment, adjust the position of the aperture until the maximum amount of light on the PM tube is seen. When the aperture is properly positioned, the light on the paper appears as a 0.32 inch (8 mm) diameter spot.

EXAMPLES

Examples 1 through 10 involve the measurement of light scattered from particles, or emitted from fluorescent molecules, or both. The instrument used to measure the light signal was a photometer built by SpectraMetrix as described previously.

For examples 1 through 3 the polystyrene particles used for these measurements were NIST traceable certified nanospheres obtained from Duke Scientific Corp., Palo Alto, CA. The gold particles were obtained from Goldmark Biologicals, Phillipsburg, N.J., a distributor for British Biocell Intl., Cardiff UK.

For examples 4 through 10 the fluorescein was obtained from Molecular Probes Inc., Eugene Oregon, the gold particles were obtained from Goldmark Biologicals, Phillipsburg, N.J., a distributor for British Biocell Intl., Cardiff UK., and the Polystyrene particles were obtained from Interfacial Dynamics Inc., Portland, Oregon.

The relative light scattering powers of particles of the same shape and size, but of different composition, can be directly compared by comparing the light scattering intensities at right angles to the path of the incident light. If the light scattering intensity measurement for a known concentration of each particle of interest is done at the right angle of observation, the light scattering intensities for identical concentrations of particles of the same size and shape but of different composition, can be directly compared and the relative total light scattering powers of the different particles determined.

Examples 1, 2, and 3- Calculated and Measured Relative Scattering Power of Comparable Polystyrene and Gold Particles

The results are presented in Tables 6,7, and 8. Calculations were performed using known light scattering relationships and our newly defined relationships as previously described. Experimental measurements were done for particles in water by detection of the light scattered by particles free in solution at a given illumination intensity and wavelength using the SpectraMetrix Photometer. The following steps were performed.

(a) Illuminate the control and comparable sized particle samples with the same incident light composition and intensity.

(b) Determine the light signal emitted from a control tube containing water but no particles.

(c) Determine the light signal emitted from a tube containing particles at known concentration.

(d) Subtract the control light signal value (b) from the light signal value of (c).

(e) Compare light signals from equal concentration of gold and polystyrene particles.

Example 4- Measured Relative Signal Generating Power of Fluorescein and Gold Particles-White Light Illumination.

The results are shown in Table 10. The same method of light detection was practiced to determine the light signal emitted from all samples in a 6 mm by 50 mm glass tube. No optical filters were used in the measurement of the light signal from either the gold particles or the fluorescein.

All measurements were made in water. The solution containing the fluorescein had a pH of 8-9. The light signal value of a tube containing only water was subtracted from the gold particle or fluorescein value in order to obtain the light signal due to just the fluorescein or gold particles.

The following steps were performed for the measurement of the light signal from particles.

A. (a) Illuminate all samples with the same incident light composition and intensity.

(b) Determine the light signal emitted from a control tube containing water but no particles.

(c) Determine the light signal emitted from a tube containing particles at known concentration.

(d) Subtract the control light signal value (b) from the light signal value of (c).

The following steps were performed for the measurement of the light signal from fluorescein.

B. (a) Illuminate the samples with incident light of the same intensity and composition as above in

(b) Determine the light signal emitted from the control tube.

(c) Determine the light signal emitted form a known concentration of fluorescein in a tube.

(d) Subtract the control light signal value (b) form the light signal value of (c).

C. (a) Compare light signals obtained from known concentrations of particles and fluorescein molecules.

Example 5 - Measured Relative Signal Generating Power of Fluorescein and Gold Particles-Monochromatic Illumination.

The results are given in Table 11. These results have not been corrected for differences in incident light intensity. Monochromatic incident light at wavelengths where maximum light emission occurs from fluorescein (490 nm) and where maximum light scattering occurs form the gold particles was used. The incident light intensity at 490 nm was slightly lower than the intensities used for the gold particles and ranged from about 86 percent of the intensity at 520 nm to about 80 percent of the intensity used at 565 nm. On the other hand, the quantum efficiency of the photomultiplier tube ranged from 0.34 at the primary emission wavelength of fluorescein (520 nm) while it was about 0.18 at 560 nm.

Except for the incident wavelength, the same method of light detection was used on all samples in a 6 mm by 50 mm glass tube. No optical filters were used in the measurement of the light signal from either the gold particles or the fluorescein.

All measurements were made in water. The solution containing the fluorescein had a pH of 8-9. The light signal value of a tube containing only water was subtracted the gold particle or fluorescein value in order to obtain the light signal due to just the fluorescein or gold particles.

The following steps were performed for the measurement of the light signal from particles.

A. (a) Determine the light signal emitted from a control tube containing water but no particles.
   (b) the light signal emitted from a tube containing particles at known concentration.
   (c) Subtract the control light signal value (a) from the light signal value of (b).

The following steps were performed for the measurement of the light signal from fluorescein.

B. (a) Determine the light signal emitted from the control tube.

(b) Determine the light signal emitted form a known concentration of fluorescein in a tube.

(c) Subtract the control light signal value (a) from the light signal value of (b).

C. (a) Compare light signals obtained form known concentrations of particle sand fluorescein molecules.

Example 6- Measured Relative Signal Generating Power of Fluorescein, Polystyrene, Polystyrene-fluorescent Compound, and Gold Particles

The results are given in Table 12. These results have not been corrected for differences in incident light intensity.

All samples were illuminated with monochromatic incident light. The 100nm diameter gold particle was illuminated with incident monochromatic light of a wavelength near where maximum light scattering form the particle occurs. The polystyrene-fluorescent compound particle sample was illuminated with monochromatic incident light of a wavelength where maximum fluorescence excitation occurred (490 nm). The maximum fluorescence emission for this fluorescent compound occurred at 515 nm. The incident light intensity at 490 nm was about 80 percent of that at 555 nm. The quantum efficiency of the photomultiplier tube at 555 nm was about 60 percent of that at 515 nm.

Except for the incident wavelength, the same method of light detection was used on all samples in 6 mm by 50 mm glass tubes. No optical filters were used in the measurement of the light signal from either the gold particles or the fluorescent particles. All measurements were made in water. The light signal value of a tube containing only water was subtracted form the gold particle or polystyrene particle value in order to obtain

the light signal due to just the polystyrene or gold particles.

The following steps were performed for the measurement of the light signal from particles.

A. (a) Determine the light signal emitted from a control tube containing water but no particles.

(b) Determine the light signal emitted from a tube containing particles at known concentration.

(c) Subtract the control light signal value (a) from the light signal value of (b).

The following steps were performed for the measurement of the light signal from the fluorescent particles.

B. (a) Determine the light signal emitted form the control tube.

(b) Determine the light signal emitted from a known concentration of fluorescent particles in a tube.

(c) Subtract the control light signal value (a) from the light signal value of (b).

C. (a) Compare light signals obtained from known concentrations of particles.

Example 7- Detection of 59.6nm Diameter Gold Particles and Fluorescein at High Serum Concentration

The results are given in Table 17. The serum was obtained from Biowhittaker Inc., Walkerville, MD. The serum had been filtered through a one micron filter before sale and was clear and straw colored in appearance. For fluorescein measurements the serum was adjusted to about pH 9 to 9.5. The solution containing the gold particles was illuminated with monochromatic incident light of 543 nm wavelength, a wavelength near that where maximal light scattering from the particle occurs. The solution containing the fluorescein was illuminated at 490 nm where the maximal fluorescence excitation occurs.

Except for the incident wavelength ,the same method of light detection was used for all samples in 6 mm by 50

mm glass tubes. No optical filters were used in the measurement of the light signal from either the gold particles or the fluorescein.

Measurements were made in the stated concentration of serum. The light signal value of a tube containing only serum at the proper concentration was subtracted from the gold particle or fluorescein value in order to obtain the light signal due to just the fluorescein or gold particles.

The following steps were performed for the measurement of the light signal from particles.

A. (a) Determine the light signal emitted from a control tube containing serum at the proper concentration but no particles.

(b) Determine the light signal emitted from a tube containing particles at known concentration.

(c) Subtract the control light signal value (a) from the light signal value of (b).

The following steps were performed for the measurement of the light signal from the fluorescent solution.

B. (a) Determine the light signal emitted from the control tube.

(b) Determine the light signal emitted from a known concentration of fluorescein in a tube.

(c) Subtract the control light signal value (a) from the light signal value of (b).

C. (a) Compare light signals obtained from known concentrations of particles.

<u>Example 8- Lower Limit of Detection of Fluorescein, Gold and Polystyrene Particles at 92.8% Serum Concentration</u>

The results are given in Table 18. For the fluorescein measurement, the light signal emitted from the sample containing fluorescein was passed through a Kodak No. 16 Wratten filter before encountering the photomultiplier tube. The maximum light intensity from

the fluorescein solution was observed at an incident monochromatic wavelength of 498 nm, while the maximum light scattering from the gold particles was observed at 554 nm. No optical filters were used in the measurement of the light signal from the gold or polystyrene particles. For fluorescence measurements the pH of the serum was adjusted to about pH 9.

Except for the incident wavelength, the same method of light detection was used for all samples in 6 mm by 50 mm glass tubes. The serum is described in Example 7.

Measurements were made in the stated concentration of serum. The light signal value of a tube containing only serum at the proper concentration was subtracted from the gold particle or fluorescein value in order to obtain the light signal due to just the fluorescein or gold particles.

The following steps were performed for the measurement of the light signal from particles.

A. (a) Determine the light signal emitted from a control tube containing serum at the proper concentration but no particles.

(b) Determine the light signal emitted from a tube containing particles at known concentration.

(c) Subtract the control light signal value (a) from the light signal value of (b).

The following steps were performed for the measurement of the light signal from the fluorescent solution.

B. (a) Determine the light signal emitted from the control tube.

(b) Determine the light signal emitted from a known concentration of fluorescein in a tube.

(c) Subtract the control light signal value (a) from the light signal value of (b).

C. (a) Compare light signals obtained from known concentrations of particles.

Example 9- Detection Limits for Polystyrene, Polystyrene-fluorescent compound, and Gold Particles at High Serum Concentration

The results are given in Table 19. Measurements were made in the stated concentration of serum. The light signal value of a tube containing only serum at the proper concentration was subtracted from the gold particle or polystyrene particle value in order to obtain the light signal due to just the polystyrene or gold particles. No optical filtration was done.

The following steps were performed for the measurement of the light signal from particles.

A. (a) Determine the light signal emitted from a control tube containing serum at the proper concentration but no particles.

(b) Determine the light signal emitted from a tube containing particles at known concentration.

(c) Subtract the control light signal value (a) from the light signal value of (b).

(d) Compare light signals from known concentrations of particles.

Except for the incident wavelength, the same method of light detection was used for all samples in 6 mm by 50 mm glass tubes. The serum is described in Example 7.

Example 10- At Low concentrations of Gold Particles Serum has no effect on Light Scattering Properties of the Gold Particles

The results are presented in Table 20. The serum at 95.7 percent concentration is clear and straw colored and has an absorbance of 0.14 at one cm pathlength and incident light wavelength of 543 nm. The light scattering measurements were made in 6 mm by 50 mm glass tubes with an inner diameter of about 5 mm. On the basis of the difference in the absorption of both incident and scattered light of a wavelength of 543 nm the light scattering signal from gold particles present in the serum

sample should be roughly 80 percent of the signal from the same concentration of gold particles present in water. No optical filters are used in this example.

The following steps were performed.

(a) Illuminate all samples with the same incident light composition and intensity.

(b) Determine the light signal emitted from a control tube containing water or a proper concentration of serum but no particles.

(c) Determine the light signal emitted from a tube containing particles at a known concentration.

(d) Subtract the control light signal value (b) from the light signal value of (c).

(e) Compare light signals from equal concentrations of gold serum and water.

Example 11 - Preparation of a 16nm Gold particle Suspension

2.5 ml of sterile water was added to 0.1g $HAuCl_4 \cdot 3H_2O$ to form a 4% $HAuCl_4 \cdot 3H_2O$ solution. The solution was centrifuged to remove particulate matter. In a separate flask, 10ml of sterile water was added to 0.1g. of sodium citrate to form a 1% sodium citrate solution. The citrate solution was filtered through a 0. 4 μ polycarbonate membrane filter to remove particulate matter. To a very clean 250 ml Erlenmeyer flask, 100ml of sterile water and 0.25ml of the 4% $HAuCl_4 \cdot 3H_2O$ was added. The flask was placed on a stir hot plate at a setting of 4 and covered with a 100ml beaker. When the mixture started boiling, 2 ml of the 1% sodium citrate was added. The solution turned a black color within a minute after adding the citrate. It then turned purple and finally a deep red. The red color was achieved after about 2 minutes after adding the citrate solution. The mixture solution was boiled for 30 more minutes and then cooled to room temperature and sterile water was added to bring the

total volume to 100ml. The final gold concentration is about 0.005% and particle concentration is 1.2×10^{12} particles/ml, assuming that all the gold salt was converted to gold particles.

Example 12- Stabilization of Metal Particles with Polyethylene Compound

1 gram of the PEG compound (MW 20,000) was added to 100ml of sterile water to form a 1% PEG compound solution and the solution was filtered through a 0.4 μ polycarbonate filter using a 50 ml syringe. To stabilize a given volume of particles, add the volume of particle solution to a volume of 1% PEG compound solution that gives a final PEG concentration of 0.1%.

Example 13 - Preparation of 30 nm Silver Coated Particles from 5 nm Diameter Gold Particles

10 ml of sterile water was brought to a boil in a 30ml beaker. 2mg of gelatin was then added slowly and the solution was allowed to continue to boil with stirring until all of the gelatin was dissolved. The solution was then cooled to room temperature. 2 ml of a 47% citrate buffer pH 5 was added. 0. 18 ml of a solution containing 5 nm gold particles (at a concentration of about 0.005% gold, 3.8×10^{13} gold particles/ml) was added followed by the addition of 3 ml of a 5.7% hydroquinone solution. The mixture was mixed well, followed by addition of sterile water for a final volume of 20 ml. 50μl of a 4% silver lactate solution was added in 10μl increments and the mixture was stirred rapidly by hand. The final silver concentration is about 0.005% and the final silver coated particle concentration was about 3.4×10^{11} particles/ml. Assuming that all of the added silver had deposited equally on each gold particle, the particle size was calculated to be 30 nm. After the final addition, the sol appeared bright yellow in room lights. In bulk solution, the light scattered by a diluted volume of the sol

contained in a 6 x 50 mm glass tube was blue when illuminated by a narrow beam of white light. When a dilution of the silver sol was examined microscopically with the SpectraMetrix microscope under DLASLPD conditions with a 10x objective and 12.5 eyepiece, a mixture of bright particles with different colors could easily be seen. The particles dominant in number were purple-blue particles. Yellow, green and red particles were also present. By adjusting the concentration of the 5 nm diameter gold particles that we use in the procedure described here, we made silver coated particles with diameters in the range 20 to 100 nm.

Example 14 -Scattered Light Properties of Nonspherical Silver Particles Formed and Examined on a Microscope Glass Slide

A small drop of a diluted, silver particle sol prepared as described in example 13 was placed on a microscope glass slide and covered with a cover glass to form a very thin film of sol between the cover glass and microscope slide. When a spot of the thin silver sol film was illuminated by a very narrow beam of light and viewed by the naked eye, at an angle which prevented the incident light from entering the eye, the illuminated spot had a blue scattered light appearance. The silver sol film was then examined microscopically with a light microscope under DLASLPD conditions with 10x objective and 12.5x eyepiece. It was observed that in a few minutes most of the particles became attached to and immobilized on the surface of the glass slide and cover glass. Blue colored particles were the most abundant. We then discovered that when a point on the cover glass was pressed with the point of a fine needle probe, the particles in the pressed area permanently changed color from their original blue (scattered light detection). At the center of the pressed area the particles were red. This center spot was surrounded by concentric circles of

different colors. From the center on out, the colors changed from red, to green to yellow to blue. The red, green and yellow particles were very bright. Theoretical calculations which we have done indicate that small silver particles have a blue color. The effect of pressing seems to be to change the shape of the particles. Our results therefore show that small silver particles can be converted from their original blue scattered light color to other colors of scattered light by changing their shapes. By moving the cover glass we found that we could disperse the differently colored particles in the pressed area into the liquid phase. In this phase the particles underwent Brownian motion and the light scattered by green and red particles flickered, which is expected for nonspherical particles.

Example 15 - Preparation of Larger Diameter Gold Particles from 16nm Diameter Particles

A 2.4% solution of hydroxylamine hydrochloride was made by adding 24 mg. of hydroxylamine hydrochloride to 1 ml of sterile water, mixing and then filtering through a 4 μ polycarbonate membrane filter attached to a 10 ml syringe. A solution of 4% $HAuCl_4 \cdot 3H_2O$ was made by adding 2.5 ml of sterile water to 0.1g $HAuCl_4 \cdot 3H_2O$ in a test tube mixing and then centrifuging to remove particulate matter. 25ml of sterile water was added to a 250 ml Erlenmeyer flask, followed by addition of the volume of 16 nm gold particles shown in Table 1 depending on the desired particle size. Next we added the volume of the 4% $HAuCl_4 \cdot 3H_2O$ solution specified in Table 1. Finally we added sterile water to bring the total volume to 100ml. Then the volume of the hydroxylamine hydrochloride solution specified in Table 1 was added with rapid hand stirring and the mixture was allowed to sit for 30 minutes. Within seconds after adding the hydroxylamine hydrochloride solution, the solution turned from a clear, slightly pink color to a final clear red or murky brown

color, depending on particle size. The smaller sizes give red colored solutions.

TABLE 1

Desired Au Particle Diameter, nm	16 nm Gold Sol, ml	$HAuCL_4 \cdot 3H_2O$ solution, ml	Hydroxyl-amine solution ml
40	6.4	0.234	1.25
60	1.9	0.245	1.25
80	0.8	0.248	1.25
100	0.41	0.249	1.25

Larger diameter particles were prepared following the same procedure as described above, but using the specified volumes of solutions as described in Table 2 and using the 100nm diameter Au particle solution instead of the 16nm gold solution.

TABLE 2

Desired Au Particle Diameter, nm	16 nm Gold Sol, ml	4% $HAuCl_4 \cdot 3H_2O$ Solution, ml	Hydroxylamine Solution ml
200	12.5	0.219	1.25
400	1.56	0.246	1.25
600	0.436	0.249	1.25
800	0.195	0.25	1.25
1000	0.1	0.25	1.25
2000	0.012	0.25	1.25

Example 16 - Preparation of a Silver Coated Particle from 16 nm Gold Particles

25 ml of sterile water was added to a 250ml Erlenmeyer flask followed by the addition of 6.4 ml of a 0.005% 16nm gold particle sol and the resulting solution was mixed. 0.234 ml of a 40 mg/ml L(+) Lactic Acid silver salt solution was then added. A deep purple color was immediately seen. Enough sterile water was then added to bring the total volume to 100ml. While rapidly stirring by hand, 1.25ml of a 24mg/ml solution of Hydroxylamine Hydrochloride was added and the resulting sol appeared lavender silver in color. A small drop of the sol was placed on a glass slide, covered with a cover glass and examined with a light microscope under DLASLPD conditions. Red, green, purple, and yellow particles were seen. The scattered light color of a dilute solution of these particles in a test tube with white light illumination was ice blue.

Example 17 - Preparing BSA Coated Glass Slides

A model system was setup to study the signal and detection parameters for various combinations of particles and illumination and detection methods for detecting particles on a solid phase as in a solid phase assay.

This system involved first coating glass slides with bovine serum albumin (BSA) and then depositing different amounts of gold particles in specified areas to study the parameters. Here we discuss the method which we use to coat the slides with BSA.

A 10% BSA in water solution was made by adding 1.5 g BSA to 15 ml of sterile ultrapure water mixing and then filtering the solution with a 0.44 mm polycarbonate membrane. A 0.02% BSA (200 μg BSA/ml) solution was made by adding 20 μl of the 10% BSA solution to 10 ml of sterile water and filtering the BSA solution through a 0.4 mm polycarbonate membrane.

Ordinary microscope glass slides were cleaned by scrubbing with a brush dipped in methanol. After brushing, the slide was then cleaned by squirting the slide with sterile water using a plastic squirt bottle. A glass slide was coated with BSA by submerging the slide in a beaker containing 0.02 % BSA in sterile water and incubating for 1 hour. The slide was then removed from the beaker and rinsed by squirting sterile water from a squirt bottle. Both sides of the slide were rinsed. The slide was then submerged in 150 ml beaker full of sterile water for about 10 min. It was rinsed again by squirting water. It is most important to remove free BSA from the slide because free BSA hinders the binding of metal particles to the coated slide. The BSA coated glass slide is then dried and stored dry in a clean plastic box.

Example 18 - Depositing Gold Particles on a BSA Slide

Small circles (about 8 mm diameter) were scribed on the BSA coated glass slides using a diamond scriber to mark the areas where gold particles were to be deposited. 3 μl of an unprotected gold particle solution with the desired gold particle concentration is deposited on one of the marked areas of the slide. The gold particle solution is deposited on the opposite side of the actual scribe marks to prevent interaction of the gold particle solution with the scribe marks.

To prepare a series of patches of gold particles, on a glass slide, in which the gold particle density is systematically decreasing, it is desirable to have the series of patches on a line in the center of the slide. To achieve this alignment we mount the slide on a holder, which we made, which allows us to deposit the gold patches in the correct alignment. It should be noted that the patches cannot be seen in room lights (that is, the particle density is so low that they have no color in room lights). We thus make a mark on the side of the slide to identify the position of the patches. The marks are made

as we deposit the particles. To form a patch of particles, we deposit 3 µl of the unprotected gold solution diluted to the desired gold particle concentration. The slide is then incubated for a specified time in a closed plastic box. The interior walls and bottom of the box are covered with a wet paper towel to prevent evaporation of the gold sol on the slide. The slide is then removed and rinsed by squirting sterile water with a pasteur pipette. We have found that for the most efficient binding of gold particles to the BSA on the slide, the pH of the gold particle solution should be adjusted to the pI of BSA (pI= 4.58 - 5.45).

Example 19 - Microarray Analytical Assay - Binding 60nm Diameter Gold Particles Coated with BSA - Biotin to Discreet Individual 80 Micron Diameter Streptavidin Patches on a Plastic Substrate

The following solutions were prepared. A 1mg/ml BSA-Biotin solution was made by adding 2mg of BSA-Biotin to 2ml of sterile water and dialyzing against distilled water in a 500ml Erlenmeyer flask for several hours at room temperature. The water was changed 4 times. The last water change was sterile water. A 20mM Tris-Saline, 0.1% PEG Compound, 0.02% Na Azide pH 7.4 buffer was also prepared. All solutions were filtered through a 0.4 µ polycarbonate membrane filter. Polystyrene test tubes were washed very well with sterile water using a squirt bottle and filled with 4ml of the 60nm diameter gold particle solution and centrifuged in the clinical centrifuge for half an hour.

The particles were then washed as described elsewhere. The soft pellets were resuspended in 10 ml of sterile water.

The pH of the gold particle solution was adjusted as follows: 100 µl of 1% PEG compound solution was added to a clean polycarbonate test tube. To this 1 ml of the 60 nm gold sol was added and incubated for 2 minutes. 0.02M K_2CO_3 was added to the gold sol in increments of 2µl until

pH 6.6 was achieved. The number of μl of 0.02M K_2CO_3 needed to adjust the pH is then calculated to add to the remaining ml of the gold sol, in this case it was 80 μl. 9.5 ml of the pH 6.6 gold sol was then added to 1.15 ml of a 1mg/ml BSA-Biotin solution in a polycarbonate tube, and incubated for 5 minutes at room temperature. The solution was then centrifuged for half an hour in the clinical centrifuge and the supernate was then decanted. The remaining soft pellet was resuspended in 3 ml of sterile water and then centrifuged as previously described, supernate decanted, and then resuspended in 0.1% PEG compound solution and centrifuged again. The supernate was decanted and the pellet was resuspended in 20mM Tris-Saline, 0.1% PEG compound solution and centrifuged The supernate was then decanted leaving behind about a 200 μl soft pellet. 50 μl of this solution was added to the plastic wells that contained the 80 μ diameter streptavidin spots and the wells were incubated overnight in a humid chamber. The wells were then washed several times with sterile water using a pasteur pipette to squirt and remove water from the well. For detection with the microscope the wells were filled with 60 μl of sterile water.

Example 20 - Detection of Bound 60nm diameter Gold Particles Coated with BSA-biotin to a Microarray of 80 Micron Diameter Streptavidin Coated Binding Site Spots

A suspension of the BSA-biotin-Au binding agent (60 nm diameter gold particle) was added to plastic wells which contain the microarray of streptavidin 80 μ diameter spots on the bottom surface of the well. After a suitable incubation time, the wells were washed and viewed with our developed light microscope system under DLASLPD conditions. We observed the BSA-biotin-Au labeled particles bound to the individual 80u individual spots. The 80 μ streptavidin spots which were not visible prior to the addition of the particles, appeared as bright

fairly circular spots. Individual spots containing different BSA-biotin-Au particle surface densities were obtained by incubating for different times or by using different BSA-biotin-Au concentrations. We were easily able to detect by eye with our microscope at magnification of about 200X, individual particles bound to the streptavidin spots at low binding densities. To automate the counting and integrated light intensity measurements from individual 80u spots, we tested video image processing software which we had on 24 hour loan from a Company here in San Diego. We captured video images using an inexpensive black and white video camera, a video frame grabber, and a simple desktop computer images which contained the 25 individual spots that were in the array device well. The software was able to measure the integrated light intensity of each spot as well as the number of particles per individual spot. For example, a streptavidin spot that was labeled with a low density of BSA-biotin-Au binding agent was analyzed with the video imaging system using a particle counting mode. To get some idea of the signal to background, an 80u diameter spot of solid-phase area not coated with streptavidin was analyzed to determine the background. In this preliminary model system, the signal/background was 317/25 ~ 13 with a labeling density on the spot of about 0.06 particles/u^2 under non-optimized illumination and detection conditions. Based on this non-optimized preliminary data, these data imply that at a signal/background of 3/1 particle densities of 0.015 particles/u^2 are detectable. Under more optimized conditions, the lower level of sensitivity may be much lower.

Example 21 - Detection Sensitivity of Gold Particles in Thin Films

60 nm gold sol was diluted by factors of 10 and 20 μl of each dilution was deposited as spots on a BSA coated slide. The slide, with no cover glass was then placed on

a Porro prism with immersion oil. Each gold sol spot had a diameter of about 4 mm. The following table gives pertinent information on each spot.

Au Sol M	Particles/ml	SPOT Diameter (mm)	# of Particles	Observation*
$3x\ 10^{-11}$	2.3×10^{10}	12.4	4.6×10^{8}	Very Intense Yellow
$I.x\ 10^{-12}$	2.3×10^{9}	12.8	4.6×10^{7}	Intense Yellow Green
3.8×10^{-13} M	2.3×10^{8}	11.4	4.6×10^{6}	Weak but detectable green

* As detected by eye

The table below expresses the above data in units that are more meaningful in clinical assay applications. The height of the liquid in the spot can be calculated with the expression

$$h = V/A$$

where V is the volume of liquid in the spot (20 ml = 0.02 cm^3) and A is the area (in cm^2) of the spot. Using $A = 1.2\ cm^2$ we get $h = 0.016$ cm = 160 m. This height is much smaller than the depth of field of the eye or our electronic and optical methods of detection and thus each spot behaves (from a geometrical point of view) as if all of the particles were on the surface of the slide and the sensitivity reported in the table are similar to the sensitivities expected for particle deposited on the surface.

Au M	Spot Area, cm^2	Spot Area, μ^2	Particles per Spot	#Particles per μ^2	*Intensity of Spot
3.8×10^{-11}	1.2	1.2×10^8	4.6×10^8	3.83	Very Intense, Yellow
3.8×10^{-12}	1.3	1.3×10^8	4.6×10^7	0.35	Intense Yellow Green
3.8×10^{-11}	1.2	1.2×10^8	4.6×10^6	0.038	Weak but Detectable green

*As detected by eye.
This column was calculated by dividing the number of particles per spot by the area of the spot.

Example 22 - 60 nm Gold Particles Deposited on the Surface of a BSA Coated Glass Slide

A series of 60 nm gold particle solution dilutions were formed and 3μl of each dilution was deposited as a small spot on a BSA coated slide. The spots were in a row on the same slide. The slide was incubated in a humid chamber for 6.5 hours, then rinsed with sterile water. The particle density on each spot was determined by particle counting under DLASLPD conditions in a light microscope which has an eyepiece with a calibrated reticle. The following table shows our results.

Area Number	Deposit concentration* particles/ml	Total number of particles deposited	#Particles per 100 micron^2
1	2.3×10^{10}	2.3×10^{8}	460
2	2.1×10^{9}	2.1×10^{7}	41.8
3	7×10^{8}	7×10^{6}	13.9
4	3.5×10^{8}	3.5×10^{6}	7
5	1.75×10^{8}	1.75×10^{6}	3.48
6	8.75×10^{7}	8.75×10^{5}	1.74

*Deposit concentration is the concentration of unprotected gold sol solution that was placed on top on a spots.

Particles per micron^2 - This gives the number of particles per 100 micron square if all of the particles in the solution deposited on a specified are became attached to the slide. The area covered by the solution has a diameter of about 8 mm. The area is $A=3.1416*(.4)^2$ cm^2 = 0.5 cm^2 = 0.5×10^8 micron^2.

After 6.5 hours, the slide was washed by gently squirting sterile water on each gold containing area on the slide. The slide was dried with heat gun on cold setting. The dried slide was viewed under DLASLPD conditions with a light microscope and the particle density in each area was determined using the calibrated reticle in the microscope eyepiece to count particles/ reticle square. The following table shows the results.

Sample	Eye*	Particles deposited /100 micron^2	Particles counted using reticle (area, micron^2)#	Particles counted per 100 micron^2
1	Very intense yellow green	460	20** (39)	51.2
2	Intense yellow green	41.8	7 (39)	18
3	Fairly intense yellow green	13.9	13 (100)	13
4	No intensity detected	7	0	0
5	No intensity detected	3.48	0	0
6	No intensity detected	1.74	0	0

* Eye - This is the light intensity excited by the microscope illuminator under DLASLDP conditions and viewed by eye from the gold on the specified area of the slide.

** - The particles in this area were too numerous to count and the count listed here may be in error by as much as 2x.

- The area listed in parenthesis is the area of 1 reticule square for the objective and optovar setting used to count particles.

Sample	Particles mounted per micron^2	Eye*
1	0.512	Very intense yellow green
2	0.18	Intense yellow green
3	0.13	Fairly intense yellow green
4	0	No intensity detected
5	0	No intensity detected
6	0	No intensity detected

Example 23 - Sensitivity for Visual Detection of Intensity from a Small Liquid Spot of 60nm Gold Particle

Two dilutions of 3.4×10^{-12} M (0.005% gold) 60 nm gold sols were prepared and 2 mL of each dilution was deposited

in a separate spot on a glass slide. Each spot had a diameter of about 4 mm (Area =6.28 x 10^6 m^2). The different spots were in a row in the middle of the same slide. Particle concentrations and densities in each spot are shown in the following table.

Gold Sol M	Particles/ml	Particles/μl	Particles/μ^2
3.4 x 10^{-12} M	2.1 x 10^9	2.1 x 10^6	0.31
1.05 x 10^{-12} M	1.05 x 10^9	1.05 x 10^6	0.155
0.5 x 10^{-12} M	0.5 x 10^9	0.5 x 10^6	0.077
0.25 x 10^{-12} M	0.25 x 10^9	0.25 x 10^6	0.0385

To determine the lowest particle density from scattered light intensity as detected by the unaided eye, we placed the slide on a Porro prism by means of immersion oil. Each spot, still in liquid form, was sequentially illuminated by light from the Baush-Lomb Illuminator with a x10 objective at the end of the fiber. The spot produced by the illuminator was about 4 mm in diameter. In the dark room, at night, we could see down to 0.0385 although the latter could just barely be seen.

Example 24 - Sensitivity for Photodiode Detection of 60 nm Gold Particles (in suspension) at Different Concentrations in Immulon Plastic microtitier Wells

Different dilutions of 60 nm gold sol were placed in different Immulon Wells (200 μl in each well). To measure scattered light intensity, the bottom of a well was illuminated with white light from a Leica Microscope Illuminator that was equipped with x10 objective. The bottom of the well was a few mm from the objective. The light from the objective produced a beam that was focused on the center of the well. The beam diameter at the focal point was about 5 mm. Scattered light was detected by a photodiode that was positioned to detect light

through the side wall of the well (right angle detection). The scattered light was detected through a small hole (diameter about 1 mm) that was positioned in front of the photodiode to limit background light detection. The wells containing the different gold sol dilutions were attached to each other and each well can be sequential positioned in the illumination and detection paths. The output of the photodiode is measured with an operational amplifier that is operated in the current mode. The feedback resistor of the op amp determines the sensitivity of the amplifier. The photodiode is operated in the photovoltaic mode. Two sets of 60 nm gold dilutions were prepared and the intensities measured with the photodiode.

a. First set of dilutions

The master solution (3.8×10^{-11} M) was diluted by factors of two. The following readings were obtained. Readings were made with a 5 megohm resistor in the feedback loop of the op amp.

Gold Particle Concentration	Intensity (Volts)
1.9×10^{-11} M	3.27
0.95×10^{-11} M	1.6
4.75×10^{-12} M	0.89
2.38×10^{-12} M	0.44
1.2×10^{-12} M	0.24
0	0.075

b. Second dilution

A x11 dilution solution (3.4×10^{-12} M) was diluted by factors of x2. Results are as follows.

Gold Particle Concentration	Intensity (Volts)
3.4×10^{-12} M	0.614
1.7×10^{-12} M	0.378
8.5×10^{-13} M	0.198
4.25×10^{-13} M	0.147
2.13×10^{-13} M	0.100
1.06×10^{-13} M	0.086
0	0.075

The above results show that, in wells, we can detect 60 nm diameter gold particles in the range 1.9×10^{-11} M to 1×10^{-13} M. The upper range can be extended.

Example 25 - Reproducibility for Depositing and Visually Detecting (Integrated Scattered Light Intensity) 60 nm Gold Particles Deposited on a BSA Coated Glass Slide

3 µl of a 2x dilution of 0.005% gold (60 nm) solution was deposited in each of 5 spots on a BSA coated slide. The slide was incubated for five minutes and then introduced into a beaker containing 150 ml of distilled water. The water washed the unbound gold off the slide. The spots were then illuminated with our illuminator (white light SpectraMetrix Illuminator). The gold particles in each spot were in the form of a ring (that is the particles were not homogeneously distributed in the spot but were confined to a ring) which scattered green light and could be clearly seen in the dark room with the unaided eye.

The experiment was repeated with a newly coated BSA slide except that during the incubation of the gold dots, the slide was gently tapped on the side with the finger so as to stir the liquid in the gold particle spots. After 5 minutes, the slide was introduced into 150 ml of sterile water in a beaker and the light scattered by each

spot was viewed alternatively using the white light SpectraMetrix Illuminator. The illuminator produced a spot of light of about 5 mm diameter on the slide. Gold spots could clearly be seen through scattered light where the gold sol was deposited. The spots were viewed with the slide submerged in water which reduced scattering by imperfection on the slide. All of the spots scattered green light and had about the same intensity as evaluated by visual detection. A small, non-light scattering spot (dark spot) appeared in the center of each spot.

Example 26 - Color of Light Scattered by 60 nm Gold Sols at Different Gold Particle Concentrations

Six 8 x 50 mm (1.6 mL) polystyrene tubes were washed by rinsing with sterile from a squirt bottle. Excess water was removed from each tube by shaking but the tubes were not dried. A gold particle solution (60 nm, 0.005%) was then serially diluted by factors of 1,2,4,8,16 and 32. Each tube had 500 mL of gold particle solution. The diluted gold sol was stable (scattered light did not change color on standing) in the polystyrene tubes. No evidence of aggregation. The light scattered by the different dilutions had the following colors. The gold sol used in these observations had been washed several times with sterile water to remove salts (that are used to form the gold sol) which seem to destabilize the gold particles.

Dilution	Color
1	yellow green
2	yellow green
4	light green
8	light green
16	light green
32	light green

Example 27 - Stabilization of Gold Particles with BSA

We found that 900 mg of BSA were required per ml of 60 nm, 0.005% gold sol to stabilize the gold sol against agglutination by 1% NaCl.

Example 28 - Deposition of 60 nm Gold Particles at High Particle Densities on BSA Coated Glass Slides

This example shows how to deposit spots of different surface densities (25 to 100 particles/μ^2) of gold particles on a glass slide. These spots are used in examples 30 and 31 to determine the intensity, color and homogeneity of light scattered from these spots (white light illumination) as seen by the naked eye and in a light microscope using DLASLPD methods.

4 ml of 60 nm gold sol (0.005%, 3 x 10^{10} particles per ml) were centrifuged in the clinical centrifuge at maximum speed until all of the gold particles sedimented to the bottom of the tube (about 30 min). The supernatant was removed and the soft pellet was diluted by factors of 1, 2 and 4. We estimate that the soft pellet had a particle concentration of about 3 x 10^{11} particles per ml. 4 ml of each dilution was deposited on separate spots on a BSA coated glass slide and the liquid in each spot was allowed to evaporate at room Temperature. The number of particles deposited in each spot (assuming that the x1 gold sol had a concentration of 3 x 10^{11} particles per ml) is shown in the following table. It should be noted that the maximum particle density which can be achieved with 60 nm particles (saturated monolayer of particles) is 354 particles/μ^2.

Dilution	Particles Deposited#	Particles/μ^2 *
1	1.2 x 10^9	100
2	6 x 10^8	50
4	3 x 10^8	25

*Particle Density in particles/μ^2 is calculated for a spot with a diameter of 4 mm (4 x 10^3 μ) and area of 12.6 x 10^6 μ^2.

#Calculated for a deposit of 4 μl of a 3 x 10^{11} particles per ml 60 nm gold sol.

After solvent evaporation, each spot was examined for its appearance in room lights and under DLASLPD illumination (as viewed with unaided eye). DLASLPD illumination was with a Leica microscope illuminator that had a x10 objective focusing lens and produced a narrow beam of light that focused to a small spot at a distance of about 10 mm from the objective. The spots were also examined by light microscopy using DLASLPD methods with x2.5 , x10 , x25 and x40 objectives and plus additional magnification of x1.25, x1.6, and x2. Only a small area of each spot could be seen at a time with the x10 and x20 objectives. However, the whole spot could be seen with the x2.5 objective. To determine the particle surface density on each spot, we counted the particles seen through the microscope on a given area and divided by this number by the area. The area was determined with a reticle positioned in the ocular of the microscope which had been calibrated with slide micrometer. As an example, when particle counting was done with the x40 objective and additional magnification of 1.25 and 2 (before the ocular), the unit square in the ocular reticle used for particle counting were respectively 6.25 μ x 6.25 μ (area of square = 39.1 μ^2) and 10 μ x 10 μ (area = 100 μ^2). These are the areas in the object plane.

Example 29 - Observations of High Surface Density Gold Particle Spots in Air

In this example, the gold particle spots prepared as described in example 29 are examined visually and microscopically using DLASLPD illumination. The spots were dry and the gold particles were thus in air.

a. x1 dilution spot

In room lights the spot has a dark purple appearance with a lighter spot of less than 1 mm diameter in the center. Under DLASLPD illumination the spot had a rather uniform whitish orange appearance. The spot was highly intense. Under DLASLPD conditions in the microscope (x10 objective, extra magnification x2, 12.5 ocular) the spot as viewed through the ocular had a highly intense orange color. Individual particles could easily be seen. Some particles could be seen very close to each other or even overlapping. The majority or particles are orange but some are green. Two or more particles which are separated from each other by less than about the spatial resolution of the microscope appear as single particles. If the space between the particles are close enough to perturb their light scattering properties, the particle group appears as a single particle of a color that is different than that of the single particle. At high particle density, it is expected from theoretical calculations that many particles will be separated by distances which are smaller than the resolution of the microscope. The appearance of the spot did not change much when viewed with a x10 or x20 objective. The area on the slide which is outside of the spot (background) was very dark compared to the intense spot. With the x2.5 objective, the whole spot could be seen. It had an intense orange appearance with a small green ring at the periphery. The color seemed to be fairly uniform in the orange area although same patches seemed to be lighter than others. The particle surface density in the green ring was much lower than in the orange area.

b. x 2 dilution spot

In room lights, the spot has a medium purple outer ring with a dark purple spot of about 2 mm in the center. Under DLASLPD illumination the spot had a rather uniform whitish yellow appearance. The spot was highly intense.

Under DLASLPD conditions in the light microscope (x10 and x20 objectives, x2 extra magnification, 12.5 ocular) the spot as seen through the ocular had a highly intense orange color but the color was not as uniform as x1 dilution. There are patches which have a greenish appearance. Very closely spaced particles could be seen. The majority of the particles are orange but there are some green which are in high abundance in the green patches. The appearance of the spot did not very much when viewed with a x10 or x 20 objective. The area outside of the spot was very dark compared to the intense spot. With the 2.5 objective the whole spot could be seen. It had an intense orange appearance with a small green ring at the periphery. The color seemed to be fairly uniform in the orange area although same patches seemed to be lighter than others. Some of this non-uniformity is due to inhomogeneities in the illuminating system which had not been optimized.

c. x 4 dilution spot

In room lights, the spot had a very light purple color with a small dark spot (about 1 mm diameter) displaced to one side of the circle. Under DLASLPD illumination the spot had a rather uniform whitish green appearance. The spot was highly intense. Under DLASLPD conditions in the light microscope (x10 objective, x2 extra magnification, 12.5 ocular) the spot appeared to be highly non-uniform probably to uneven evaporation of solvent. The center of the spot had a highly intense orange or lavender color. Particles are very close to each other or even overlapping in this central area, with most particles having an orange color and some a green color. Away from the center, the spot had a greenish appearance with a predominance of green particles with some orange particles. There are alternating rings of green and yellow color as one goes from the center of the spot to the periphery. The area outside of the spot

(background) was very dark compared to the intense spot. The whole spot can be seen with x2.5 objective. The spot had an oval appearance with a orange or lavender spot (about 1.5 mm diameter) towards the center. This was surrounded by intense circles of alternating yellow green and green area. A small ring at the periphery had a less intense (but still intense) color which was distinctly green. In this peripheral area, almost all of the particles are green particles and could be counted with x40 objective, extra magnification x2. The particle surface density in the green particle area was about 20 particles/ 39.1 μ^2 or 0.5 particles per μ^2. At the very periphery the particle count dropped rapidly and we counted about 7 particles per 100 μ^2 or 0.07 particles/μ^2. The gradient of particles on this spot allows us to count the particles up to the counting limit of about 1 particle/μ^2.

Conclusions for immobilized particles in air

a. The procedure described above allows us to deposit gold particles at high surface densities on small (4 mm) spots. The deposits are not completely uniform when viewed in room lights as expected for the evaporation process which we use to form the spots.

b. Under DLASLPD conditions in the light microscope, the x1 and x2 dilution spots are fairly uniform. For the x4 dilution, the spot displayed more non-uniformity.

c. The particle density of the spots seems too high for particle counting to be meaningful (particles to close to be resolved as individual particles). However, at the periphery of the x4 dilution spot the particles could be counted and the density here was around 0.5 particles/ μ^2. This particle density is close to the maximum particle density which can be counted with the resolution of our microscope.

Example 30 - Observations with High Surface Density Gold Particle Spots in Water

The slide used in example 30 was submerged in 150 ml of sterile water in a beaker. The particles did not seem to come off the slide. Using the microscope illuminator with 10x objective, the gold spots in the submerged slide could each be separately illuminated with a narrow beam of light. The color of the spots (observation with unaided eye) did not seem to change in going from air to water. The glass slide was removed from the beaker of water and covered with a cover glass. A thin film of water surrounded the gold particles. Microscope observations were as follows.

a. x1 dilution spot

When viewed under DLASLPD conditions with a light microscope with x2.5 objective, x1.25 extra magnification and x12.5 ocular, the spot had a fairly uniform orange lavender appearance. The periphery of the spot contained bright, yellow green particles. The particles at the periphery can be easily seen with 10x to 40x objectives. The particle surface density was very high throughout the spot even at the periphery except at a very thin ring at the very periphery where the individual particles can easily be seen as bright objects.

b. x2 dilution spot

The whole spot could be seen through the 12.5 x ocular using the x2.5 objective and x1.25 extra magnification. The spot had a very intense yellow color overall and seemed fairly uniform. Most of the spot was yellow but towards the periphery the spot had a yellow green color. With a x40 objective and x1.25 extra magnification, the particles at a very high surface density could be seen. Most of the particles have a yellow green color. A few have a green or red color. The spot had a very fairly uniform intensity except at the

periphery where the particle density drops off very rapidly to zero (dark background). The individual particles with dark spaces between then can easily be seen and counted at the periphery where the particle surface density is low.

c. x4 dilution spot

The whole spot could be seen with the 2.5 x objective plus 1.25x extra magnification. The spot had a very intense yellow green color and the color was very uniform in contrast to the observations in air where the spot had many green patches. Individual particles could easily be seen with x40 objective and x1.25 extra magnification. The particles in water are much more intense or brilliant than in air. The particles at the periphery were not predominantly green but in water most of the particles were yellow green with some red and orange particles. Most of the spot had a very intense yellow color. Individual particles can be seen with x40 objective but the particles are very dense and overlap. In the most intense area of the spot, the particles seen with x40 objective and x2 extra magnification, were at a density of about 25 particles / 39.1 μ^2 or about 0.6 particle/μ^2. This number may not represent the true particle surface density because of limitations microscope resolution.

<u>Conclusions for immobilized particles covered by water</u>

Placing the gold spots in water seems to give them a more uniform appearance. Spots x2 and x4 dilutions seem to both have a yellow color when viewed by the unaided eye with the illumination of the light microscope under DLASLPD conditions (slide sitting on prism and coupled to the prism with immersion oil). The x1 dilution spot appears to be orange to the eye.

Example 31 - Binding of 60nm Gold-BSA-biotin Reagent to Magnetic Beads

This example demonstrates our ability to detect and quantify the specific binding of gold particles to magnetic beads by light scattering intensity measurements in suspension and to detect individual gold particles bound to magnetic beads by light microscopy under DLASLPD conditions.

60 nm gold particles were coated with BSA that had been covalantly labeled with biotin (BSA-biotin-Au). 500ml of a phosphate buffered saline solution, pH 7.4, containing 0.1% BSA solution was added to each of 5 microcentrifuge tubes. The tubes were labeled 0, 1, 2, 3, 4. A solution of BSA-biotin-Au at a gold particle concentration of 3.8×10^{-11} M was added to each tube and the tubes were shaken. Additional amounts of the BSA-biotin-Au solution were added to bring the concentration of gold BSA-biotin particles = 8×10^{-13} M in each tube. The following amounts in μl of Dyna beads M280 Streptavidin (2.8 μ diameter beads with streptavidin covalantly attached to the surface of the bead) suspension containing 6.7×10^{8} Dyna beads/ml (10mg/ml) or about 1×10^{-12} M bead molar concentration, dissolved in phosphate buffered saline (PBS), pH 7.4, containing 0.1% BSA and 0.02% NaN_3 was added to each tube:

Tube #	μl	Conc. of mag. beads	Conc. of biotin	Gold BSA-Biotin Conc.
0	0	0	0	
1	5	1×10^{-14} M	3×10^{-8} M	8×10^{-13} M
2	10	2×10^{-14} M	6×10^{-8} M	8×10^{-13} M
3	15	3×10^{-14} M	9×10^{-8} M	8×10^{-13} M
4	20	4×10^{-14} M	12×10^{-8} M	8×10^{-13} M

The tubes were incubated for 30 minutes at room temperature and then were placed, one at a time starting with tube zero, in a MPC-E/E-1 magnetic particle concentrator to separate magnetic beads from solution. After 2 minutes the supernatant solution was carefully removed with the tube still in the magnetic separator. The supernatant was placed in a 1ml microculture tube. Tube 5 was left for 5 minutes in the magnetic concentrator while the other tubes were left for 2 minutes. The scattered light intensity of the supernates were then measured in the SpectraMetrix photometer with the following settings: Resister PM out = 0.1 Meg ohm, Filter on excitation side = orange filter CS 3-67, x10 neutral density attenuator = out.

The following scattered light intensities were measured:

Tube #	Intensity
0	1.21 Volts
1	1.04
2	0.644
3	0.49
4	0.362

The supernatants were returned to their respective tubes containing the magnetic beads and allowed to incubate for an additional 2 hours. After 2 hours the tubes were then processed as previously described and the following scattered light intensities for the supernates were obtained:

Tube #	Intensity (Volts)	Normalized Intensity	Frac. of Gold Particle Bound
0	0.855	1.21	0
1	0.604	0.85	0.3
2	0.382	0.54	0.55
3	0.326	0.46	0.61
4	0.206	0.29	0.76

The 2 hour extra incubation did not result in greater binding of gold BSA -Biotin to magnetic beads.

A drop of the magnetic beads with attached gold particles was deposited on a microscope glass slide and covered with a cover glass. The slide was then examined under DLASLPD conditions with a light microscope. The magnetic beads could easily be seen as strongly scattering objects but the gold particles on the beads were more difficult to see because of the strong scattering by the large magnetic beads. However, if the water medium was replaced by a bathing medium with a refractive index around 1.4 to 1.5, the particles could be seen more clearly. Also if the illuminating light beam was inclined at a higher angle with respect to a line perpendicular to the slide, the gold particles could be seen more clearly.

Example 32 - Detection of Nucleic Acid Hybridization With Nucleic acid-Labeled 40 nm Diameter Gold (Au) Particles

1. Preparation of Chemically Activated Polyethylene Glycol-Amine Coated Au Particles

Reactive amine groups for conjugation of nucleic acids to the Au particles was accomplished as follows. 40nm Au particles were coated with bis(Polyoxyethylene bis[3-Amino-2-hydroxypropyl]) Polyethylene compound using the procedure described in Example 12. This results in 40

nm diameter Au particles with a thin coat of polyethylene compound that has several chemically reactive amine groups for conjugation of the nucleic acids to the particles.

2. Preparation of nucleic acids for conjugation to the 40 nm diameter Au particles

Homopolymers of polycytidylic acid(Poly (C)) and polyinosinic acid (Poly(I)) were chemically modified as follows. 0.8 mg and 1.3 mg of Poly(I) and Poly(C) were placed in separate tubes. To each of the tubes was added 1.0 ml of a 0.1M solution of 1-ethyl-3, 3-dimethylamino-propylcarbodiimide (CDI) in imidizole buffer pH 8.5 and incubated for one hour. The nucleic acids were then precipitated by ethanol precipitation and resuspended in hybridization buffer (20 mM Tris-HCl, 100 mM NaCl, 1 mM EDTA pH 7.6).

3. Conjugation of Activated Nucleic Acids to Activated Au Particles

The same protocol was used for both the Poly(I) and Poly(C) preparations. To 50 ul of nucleic acid solution was added 20 ul of 40nm-Au-PEG activated particle solution, and 100 ul of Hepes 0.2 M pH 8.0 and incubated for 1 hour at 50° C. Following the reaction, the Poly(C) and Poly(I) 40nm Au-nucleic acid conjugates were collected by centrifugation, washed, and resuspended in hybridization buffer.

4. Hybridization Experiments

The hybridization properties of the nucleic acid-40 nm Au particle conjugates (40nm diameter gold particles with nucleic acids covalently attached to the polymer coated surface of the particle) were studied as follows. A light microscope using DLASLPD methods was used. The glass slide-liquid-cover slip experimental setup as shown in Figure 9 was used. A drop of the Poly(C)-Au particle conjugate preparation was placed on the slide and covered

with a cover slip. A drop of immersion oil was placed on the microscope condenser and the slide then placed on top of the condenser. The 10X objective was used. The solution appeared fairly homogeneous, and the Poly(C)-Au particle conjugates could be seen floating across the field. Their brownian motion appeared to be less than what we have previously observed for 40 nm Au particles not labeled with nucleic acids. A few Poly(C)-Au particles appeared to be stuck to the surface of the glass slide. There were a few aggregates of Poly(C)-Au particles, mostly of two to four Poly(C)-Au particles. These aggregate structures moved as one unit as they floated across the field of view. We noticed that the particle density of Poly(C)-Au particles attached to the surface of the slide was increasing as we observed this slide for several minutes. The color of the scattered light coming from the Poly(C)-Au particles was green. The cover slide was removed, and a drop of the Poly(I)-Au preparation was then placed next to the wet area of the slide containing the Poly(C)-Au drop. Contact between the two spots was achieved by using a metal probe to drag a line of liquid from the Poly(I)-Au drop to the wet part of the slide containing the poly(C)-Au. A cover slide was then placed on top, and the slide was then placed back on the microscope and viewed. We observed the formation of increasing numbers of multiple Poly(I)-Au-Poly(C)-Au particle aggregates over time. After about 20 minutes, we scanned the slide and observed that there were very few single particles, and most of the particles were in aggregates of several particles, many of them stuck to the glass slide. The aggregates appeared to have defined shapes, that is, there seemed to be a particular way these particle aggregates were assembled, some appearing as a spool of randomly wound up string, and others appeared as branched chain networks. The appearance of these multiple aggregates was very different as compared to the few aggregates we observed on the control Poly(C)-Au slide. We

switched to the 40X objective and in some of the aggregates, some of the particles appeared more yellow in color than green. We then removed the cover slip of the slide containing the Poly(Au)-Poly(I)-Au reaction and added a drop of 10^{-5} M ethidium bromide onto the slide and covered with a cover slip. We observed a faint orange color coming from the aggregates of particles on the slides, that is, it appeared like the green and yellow-green color of the particles was placed on a background of a faint orange color. No orange color was observed in the areas away from the aggregates. This faint orange color indicated to us that there was double-stranded structures of nucleic acids near and within the particle aggregates. We interpret this as the hybridization of the Poly(C)-Au conjugates to Poly(I)-Au conjugates. This slide was removed and a control slide containing a drop of the Poly(I)-Au was observed under the microscope. We made similar observations as compared to the control Poly(C)-Au slide but that this Poly(I)-Au preparation seemed to have about twice as many small aggregates as compared to the Poly(C)-Au control slide. The color of scattered light appeared green with some aggregates appearing yellow-green.

In this particular format, both of the complementary strands were labeled with gold particles. As the complementary strands hybridize, more particle aggregates appeared. The binding of nucleic acids can be detected by detecting the scattered light from gold or similar particles. It also appears that when two or more Au particles are in close proximity to each other, the color of the scattered light can change. This change in color of the scattered light can be used also as a way to detect a binding event. It should be noted that the Au particles, or any other particle which scatters light sufficiently can be used in numerous formats to detect the binding of nucleic acids or any other ligand-receptor pair in a separation or non-separation assay format.

Example 33 - Detection of Bound Gold Particles to Large Polystyrene Beads

We placed a drop of a solution of spherical polystyrene particles of about 2 microns in diameter coated with biotin on a glass microscope slide and viewed in the light microscope under DLASLPD conditions. The polystyrene particles were easily seen as bright white light point sources. We then place a drop of a preparation of 60nm gold particles coated with streptavidin onto the drop of polystyrene particles and viewed this preparation in the microscope. The bright white polystyrene particles could be seen and a faint halo of yellow-green color was observed surrounding the polystyrene particle. We evaporated the solution from the slide and then placed a drop of microscope immersion oil on the preparation and then viewed under the microscope. Individual gold particles and large circular ring areas of yellow-green gold particles were easily visible. The polystyrene particles appeared as almost a dark or black spot surrounded by a halo or ring of yellow-green color. This method can be used to detect bound gold particles or other metal-like particles to the surface of solid particulate matter and small solid-phases such as glass or other beads, and biological cells and the like.

Example 34 - Light Scattering Properties of Gold Particles Coated with Polyethylene Compound

Gold particles approximately 100nm in diameter made by the citrate procedure were used. A portion of this solution was placed in a separate container and the particles were coated with polyethylene glycol compound(MW = 20,000) using the procedure described elsewhere.

For scattered light comparisons of coated and uncoated particles, the samples were diluted in water until each solution had a faint tinge of pinkish-red color. The scattered light intensity versus incident

wavelength profiles for the samples were collected using the SpectraMetrix Photometer.

For these measurements, a monochromator was placed between the light source and sample Scattered light data was collected at 10 nm increments from 400 nm to 700 nm by adjusting the monochromator setting. The data were corrected for wavelength dependent monochromator and photodetector variation as a function of wavelength using a calibration graph that was made by using 12 nm silica particles. The data was analyzed using the calibration graph. The data are shown in Figure 16.

The data show that the coated and uncoated 100nm gold particles have very similar scattered light intensity vs. Incident wavelength profiles. Therefore, many different types of macromolecular substances such antibodies, nucleic acids, receptors, or similar can be coated on the surface of the particles without significantly altering the scattered light properties.

Other embodiments are within the following claims.

This Sheet Has Been Replaced

Claims

1. Method for specific detection of one or more analytes in a sample, comprising the steps of:

specifically associating any said one or more analytes in said sample with a scattered-light detectable particle,

illuminating any said particles associated with said analytes with light under conditions which produce scattered light from said particle and in which light scattered from one or more said particles can be detected by a human eye with less than 500 times magnification and without electronic amplification, and

detecting said light scattered by any said particles under said conditions as a measure of the presence of said one or more analytes.

2. The method of claim 1 wherein said particle has a size suitable for producing a specific colored light when observed by said human eye and illuminated with white light.

3. The method of claim 2 wherein the color of said specific colored light provides a measure of the presence or amount of said one or more analytes.

4. The method of claim 1 wherein said detecting comprises measurement of the intensity of scattered light as a measurement of the concentration of said one or more analytes.

5. The method of claim 1 wherein said detecting comprises measurement of the color of scattered light as a measurement of the concentration of said one or more analytes.

* C L M S *

REPLACED SHEETS SG

☐ Description : paged ..... to paged .....
☑ Claim(s) : paged 212 to paged 219
☐ Drawing(s) : paged ..... to paged .....
☐ Abstract : paged ..... to paged .....

6. The method of claim 1 wherein said particle has a composition suitable for producing a specific colored light when observed by said human eye and illuminated with white light.

7. The method of claim 1 wherein said particles are associated with a solid phase bound analyte.

8. The method of claim 1 wherein said particles are in a liquid phase during said detecting step.

9. The method of claim 1 wherein said analyte is bound to a solid phase.

10. The method of claim 1 wherein said analyte is free in liquid solution.

11. The method of claim 1 wherein said sample is a microarray or array chip comprising discrete areas each of which may contain said one or more analytes.

12. The method of claim 1 wherein said light is polychromatic white light.

13. The method of claim 1 wherein a monochromatic light illumination source is used to provide said light.

14. The method of claim 1 wherein said method comprises providing a plurality of different particles each having a different visual appearance when observed by said human eye.

15. The method of claim 1 wherein said particles are used in a homogenous assay and wherein two or more particles are brought sufficiently close together so that the light scattering property of any one particle is

altered, wherein said alteration is a measure of the presence of said one or more analytes.

16. The method of claim 1 wherein said particles are used in a assay and wherein two or more particles are brought sufficiently close together so that the light scattering property of the two or more particles can be resolved from single particles and said light scattering is a measure of the presence of said one or more analytes.

17. The method of claim 1 wherein said particles are used in a homogeneous assay and wherein two or more particles are brought sufficiently close together so that the light scattering property of the two or more particles can be resolved from single particles and said light scattering is a measure of the presence of said one or more analytes.

18. The method of claim 1 wherein said particles are used in a homogeneous assay and wherein two or more particles that are held in close proximity to one another are caused to be separated so that the light scattering property of any one particle is altered, wherein said alteration is a measure of the presence of said one or more analytes.

19. The method of claim 1 wherein said particles are used in a homogeneous assay and wherein two or more particles are linked together by one or more molecular interactions, wherein the molecular interaction holding the particles together is disrupted so that one or more particles is released from the molecular interaction, wherein said release is a measure of the presence of said one or more analytes.

20. The method of claim 1 wherein said particle is a gold or silver particle.

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21. The method of claim 1 wherein said particle has a size between 1 and 500 nanometers inclusive.

22. The method of claim 1 wherein said light is directed toward said particle by a prism or other light guide system.

23. A population of specifically detectable metal like light scattering particles formed from at least one light scattering material, said particle comprising at least one additional material on its surface to provide chemical stability and an ability to said particle to bind to an analyte, wherein said population of particles is adapted to sufficient homogeneity in size so that one or more specific light scattering properties of the individual particles in said population are similar from particle to particle.

24. The particle of claim 23 wherein said particles are formed from a material selected from the group consisting of metal, metal compound, metal oxide, semiconductor, and superconductor.

25. The population of claim 23 wherein said particles are formed of a mixed composition comprising a metal-like material as one part of said composition.

26. A specifically detectable light scattering particle reagent formed from at least one light scattering material selected from the group consisting of a metal, a metal compound, a metal oxide, a semiconductor, and a superconductor having a diameter between 1 and 500 nanometers wherein said particle reagent comprises a base molecule adapted to bind to the surface of the particle and adapted to bind to a binding agent, said particle reagent capable of binding to an analyte.

* C L M S *

27. The population or particle of claim 23 or 26 wherein said particles have a coat selected from the group consisting of a polymer, an inorganic material, an organic material, a proteinaceous material, a base molecule material, and a binding agent.

28. The population or particle of claim 23 or 26 wherein said particles are spherical.

29. The population or particle of claim 28 wherein said particles are oval or ellipsoidal.

30. The population or particle of claim 29 wherein said particles are asymmetrical.

31. The population or particles of claim 23 or 26 wherein said particles have a size distribution with a coefficient of variation less than 5%.

32. The population or particle of claim 23 or 26 wherein said particles have a size distribution with a coefficient of variation of less than 10%.

33. The population or particle of claim 23 or 26 wherein said particles have a size distribution with a coefficient of variation of less than 15%.

34. The population of claim 23 wherein said particles comprise gold.

35. The population of claim 23 wherein said articles comprise a mixed composition of gold and silver.

36. The population of claim 23 wherein said particles are composed of silver and a magnetic or ferro electric material.

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37. The population of claim 23 wherein said particles are composed of gold and a magnetic or ferro electric material.

38. The population of claim 23 wherein said particles are composed of a mixture of metal-like materials and a magnetic or ferro electric material.

39. The population of claim 23 wherein said particles are composed of gold with a surface coating selected from the group consisting of polymer, protein, nucleic acid inorganic compound and organic compound, base material molecule, binding agent, and wherein said particles have a diameter of between 10 and 50 nanometers inclusive and produce a green scattered light when illuminated with white light.

40. The population of claim 23 wherein said particles are composed of gold with a surface coating selected from the group consisting of polymer, protein, inorganic compound and organic compound, base material molecule, binding agent, and the diameter of said particles are between 50 and 70 nanometers inclusive and produces a yellow-green to yellow-scattered light color when illuminated with white light.

41. The population of claim 23 wherein said particles are composed of gold with a surface coating selected from the group consisting of polymer, protein, inorganic compound, an organic compound, base material molecule, binding agent, and said particles have a diameter between about 70 and 120 nanometers and produces an orange to orange-red scattered light color when illuminated with white light.

42. The population of claim 23 wherein said particles are composed of gold with a surface coating





selected from the group consisting of polymer, protein, inorganic compound, an organic compound, base material molecule, binding agent, and said particles have a diameter of greater than 120 nanometers and less than one micron and produces an orange to orange-red scattered light color when illuminated with white light.

43. The population of claim 23 wherein said particles are composed of silver with a surface coating selected from the group consisting of polymer, protein, inorganic compound, an organic compound, base material molecule, binding agent, and said particles have a diameter between about 5 and 50 nanometers and produces a blue scattered light color when illuminated with white light.

44. Apparatus for analysis of a solid phase sample, comprising:

a light source angled such that light is delivered to said sample and the scattered light from any particles bound to said sample can be maximally detected, wherein said apparatus is constructed and arranged such that any said particles associated with said sample can be illuminated with said light under conditions which produce scattered light from said particles and in which light scattered from one or more said particles can be detected by a human eye with less than 500 times magnification and without electronic amplification.

45. The apparatus of claim 44 wherein a collection lens detector is placed outside the intensity envelope of the forward direction of scattered light from said sample.

46. The apparatus of claim 44 comprising a particle counter having necessary computer software or firmware configured and arranged to detect said scattered light.



47. The apparatus of claim 44 wherein a collecting lens is provided in said apparatus essentially perpendicular to the surface upon which said sample is to be applied.

48. The apparatus of claim 44 wherein said apparatus is constructed and arranged to allow assay of a microarray wherein each separate area of said microarray has a dimension between ten square microns and one square millimeter.

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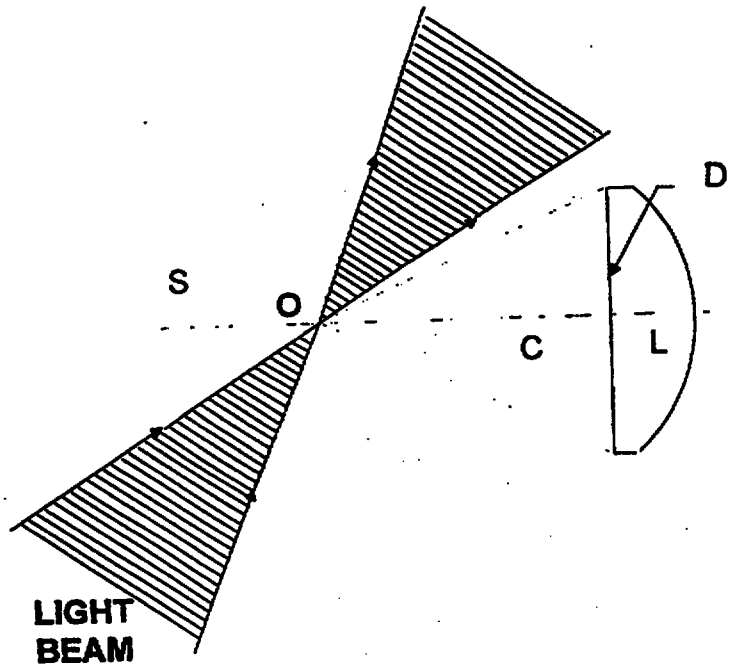


FIGURE 1

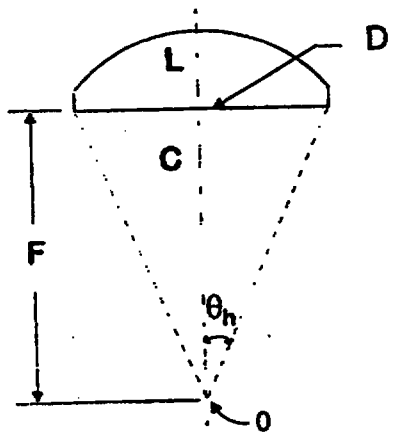


FIGURE 2

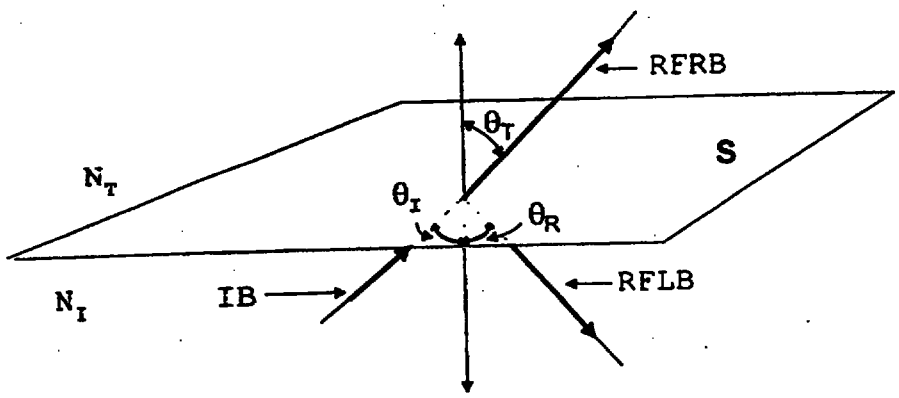


FIGURE 3

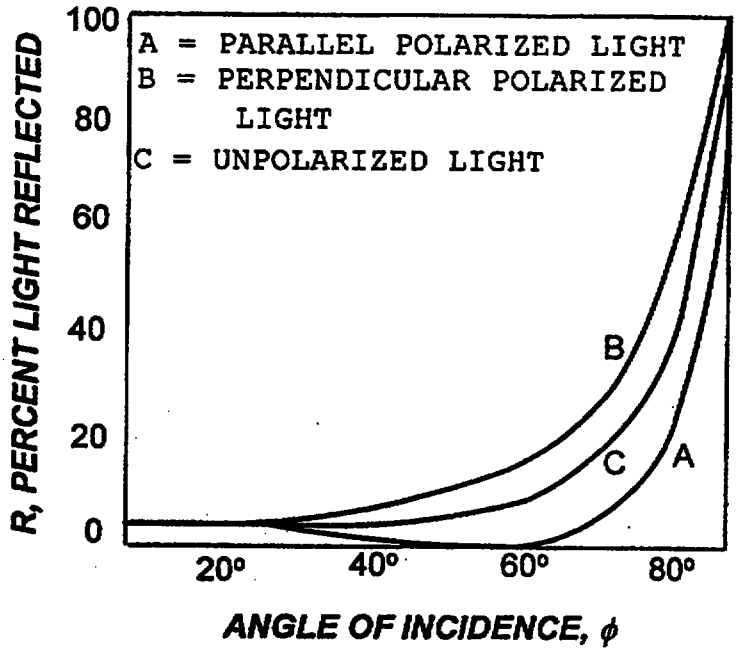


FIGURE 4A

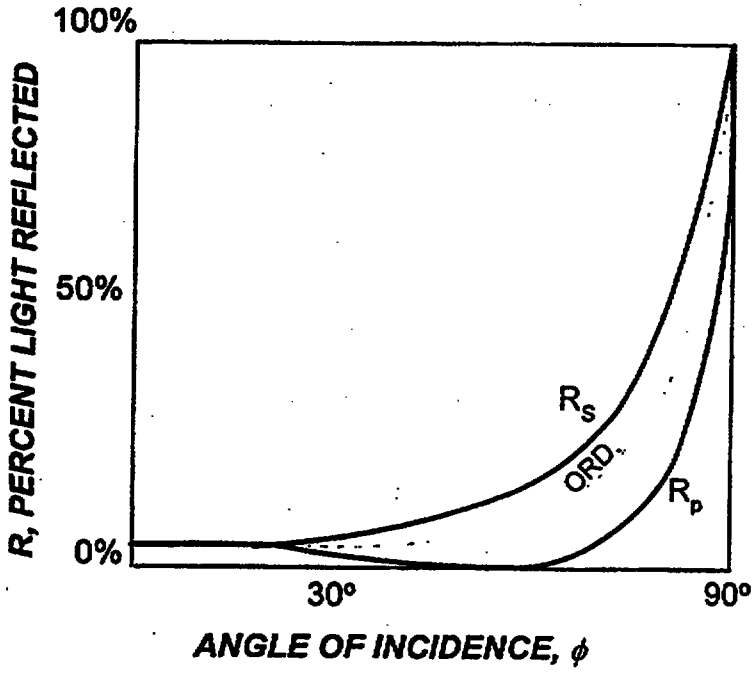


FIGURE 4B

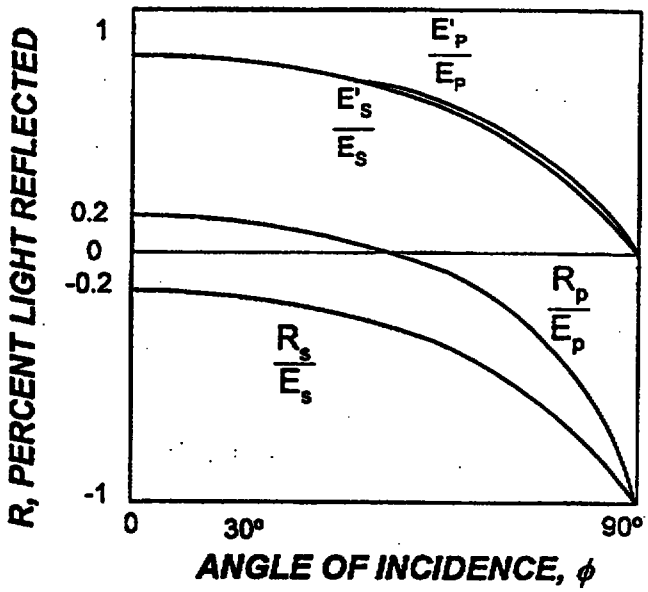


FIGURE 4C

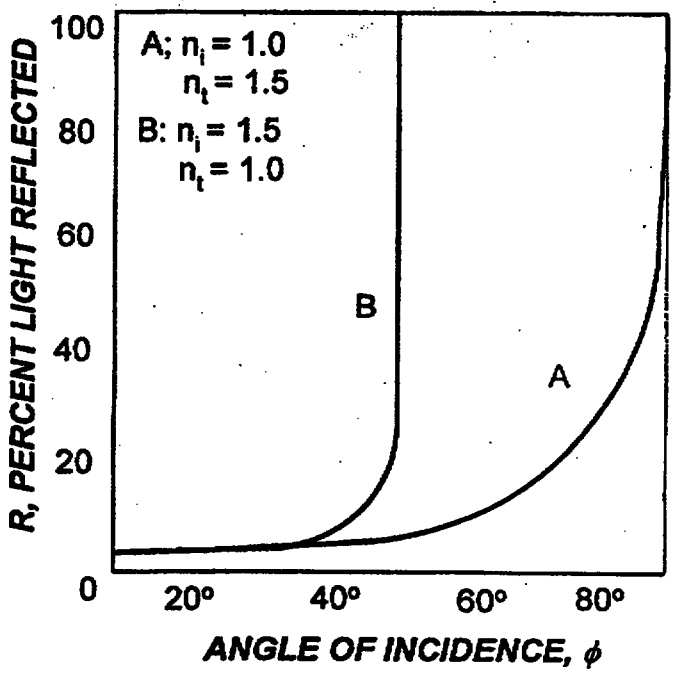


FIGURE 5A

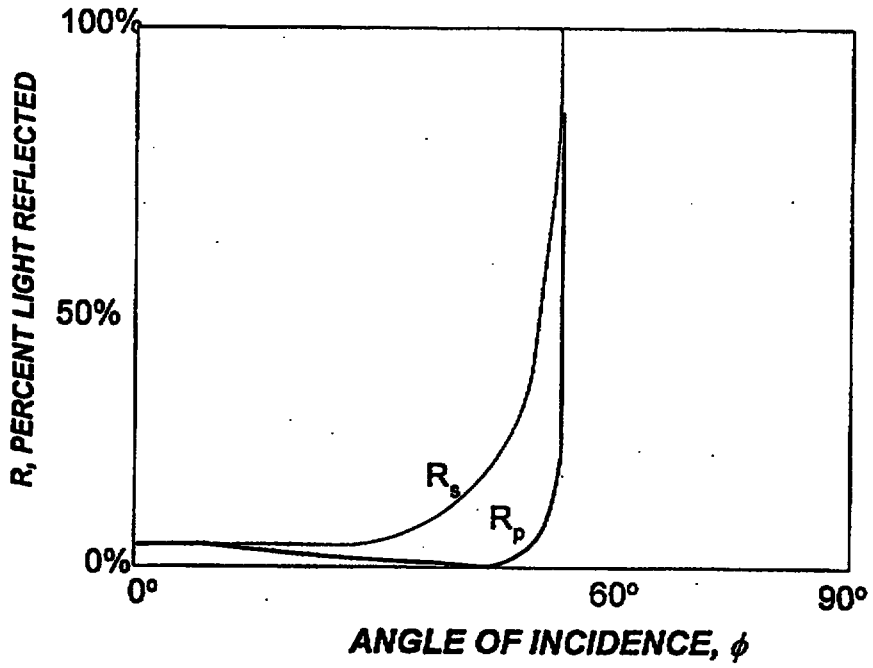


FIGURE 5B

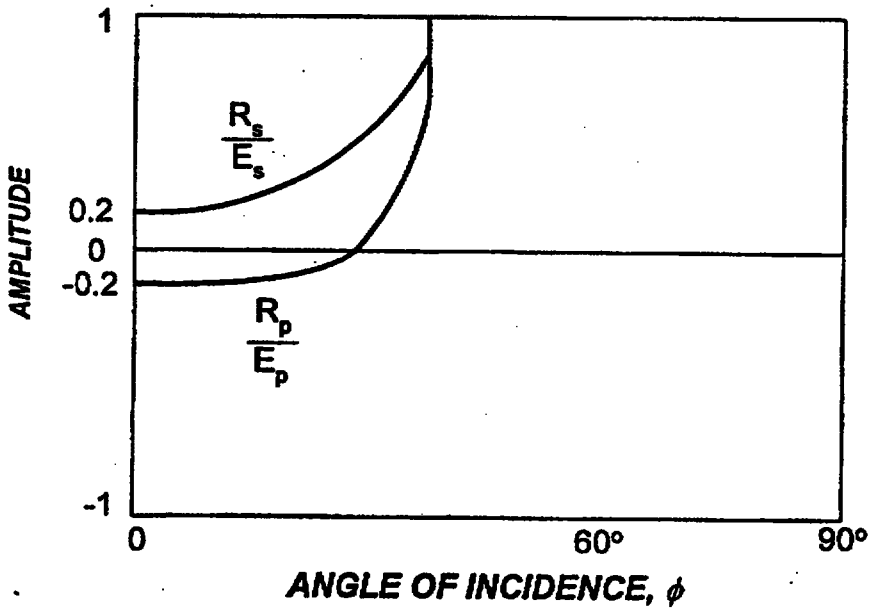


FIGURE 5C

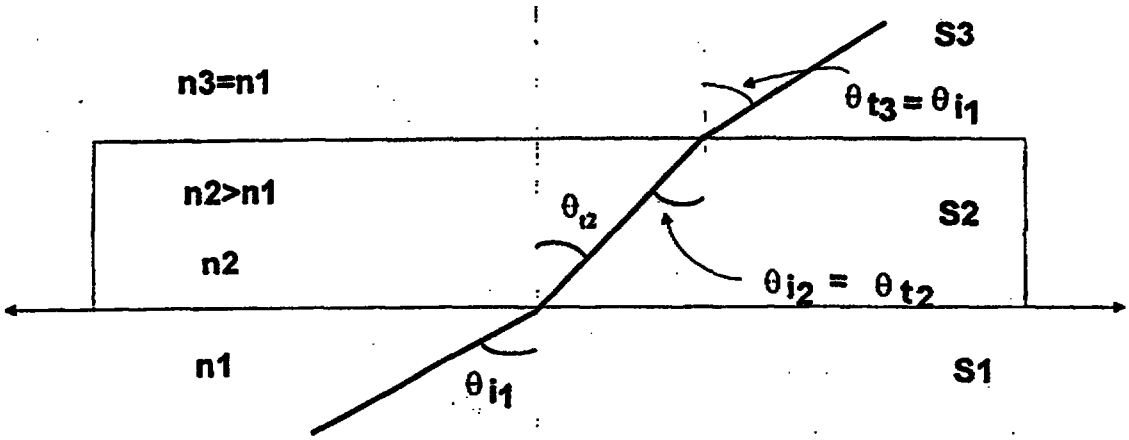


FIGURE 6

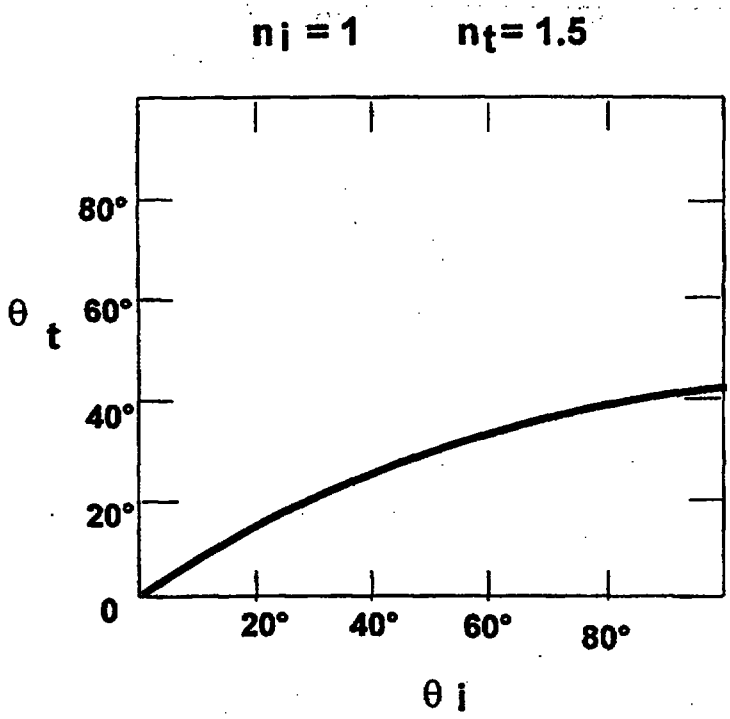


FIGURE 7

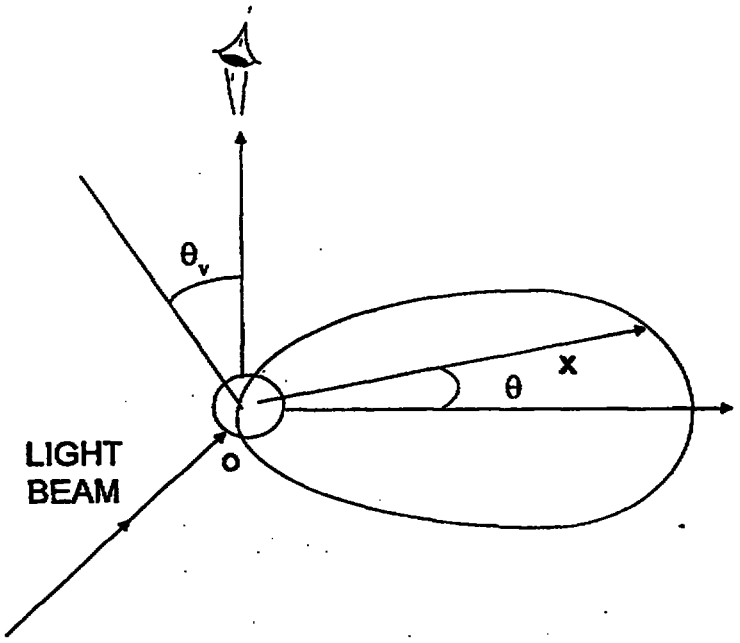


FIGURE 8

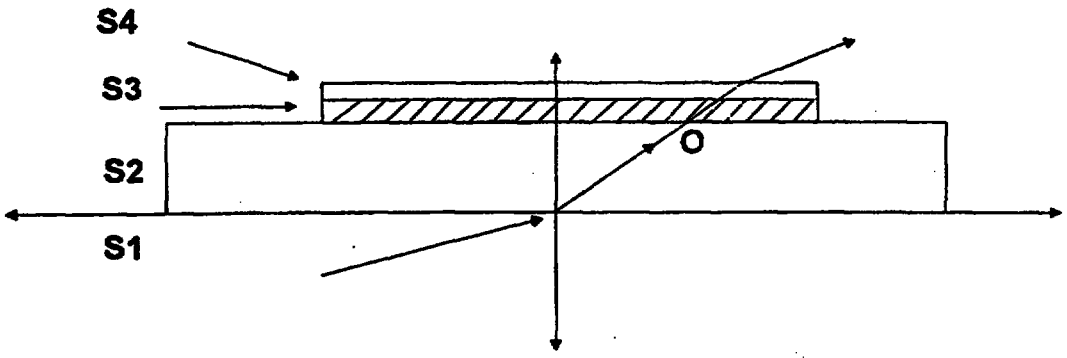


FIGURE 9

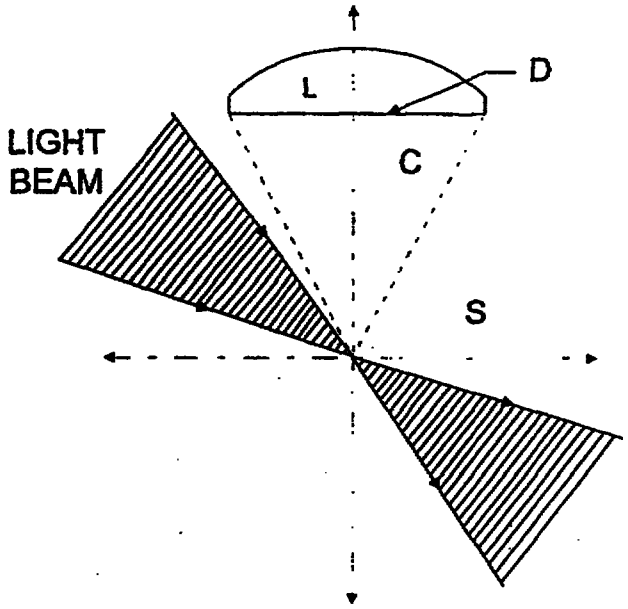


FIGURE 10

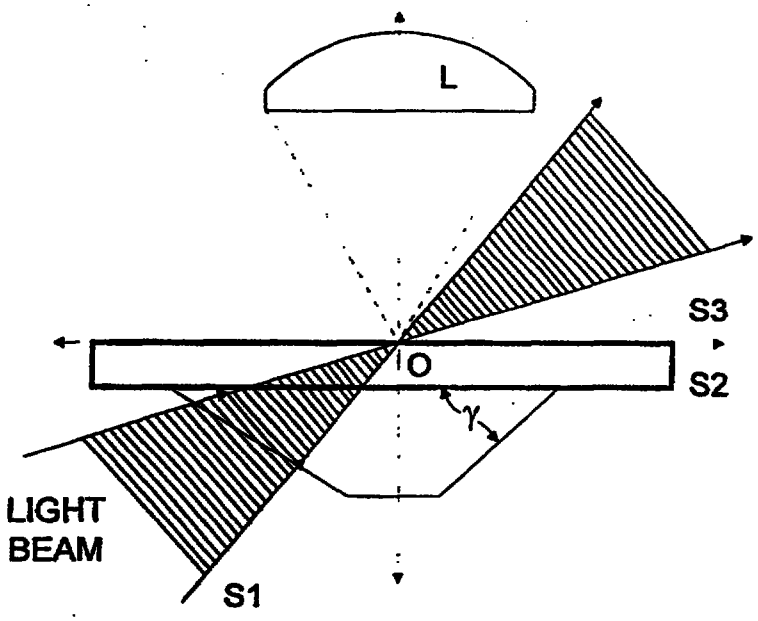


FIGURE 11

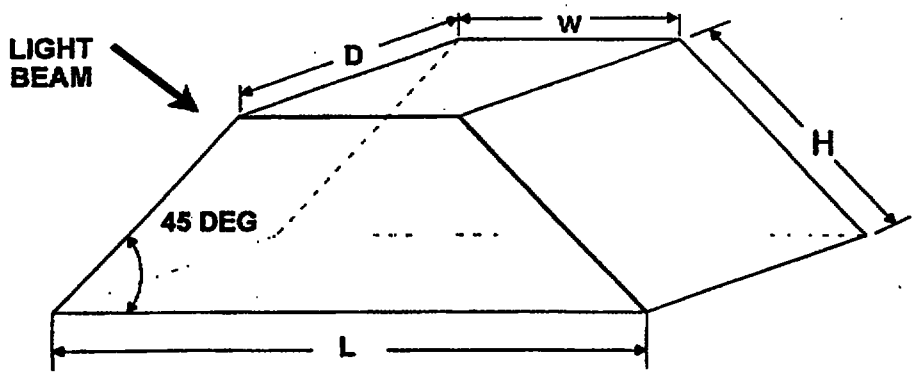


FIGURE 12A

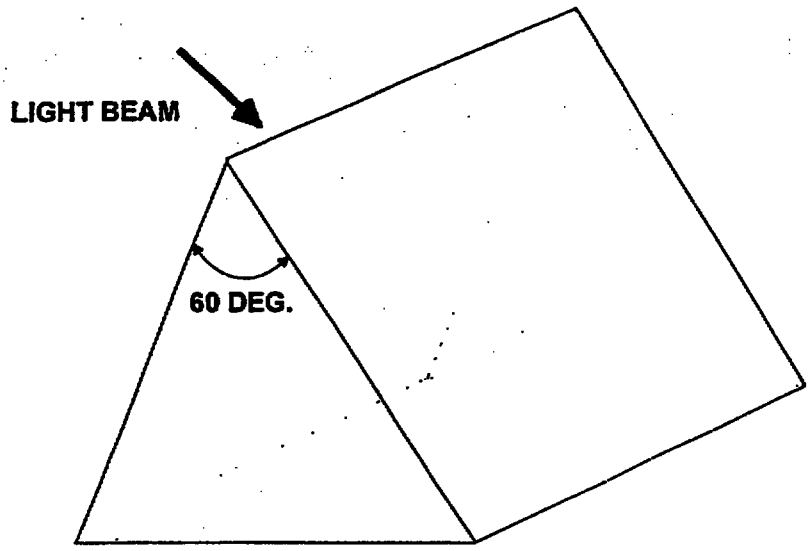


FIGURE 12B

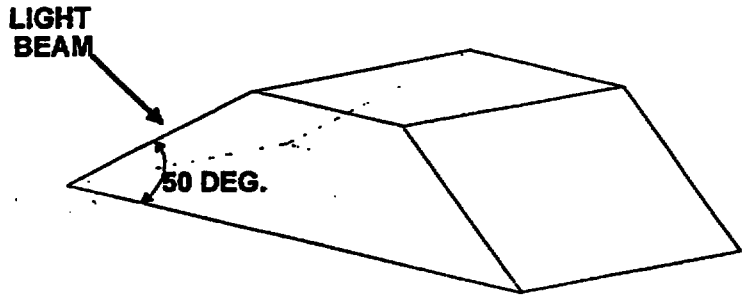


FIGURE 12C

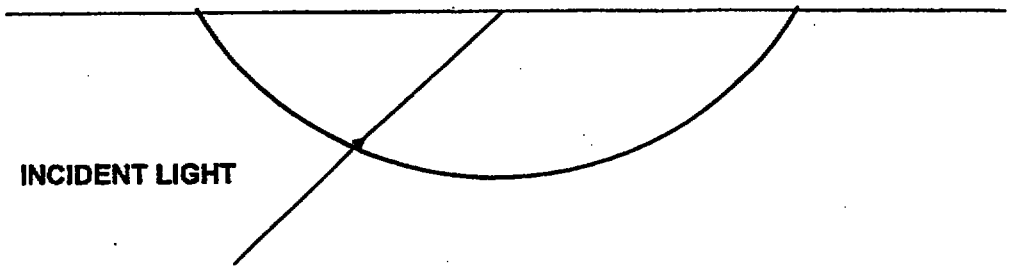
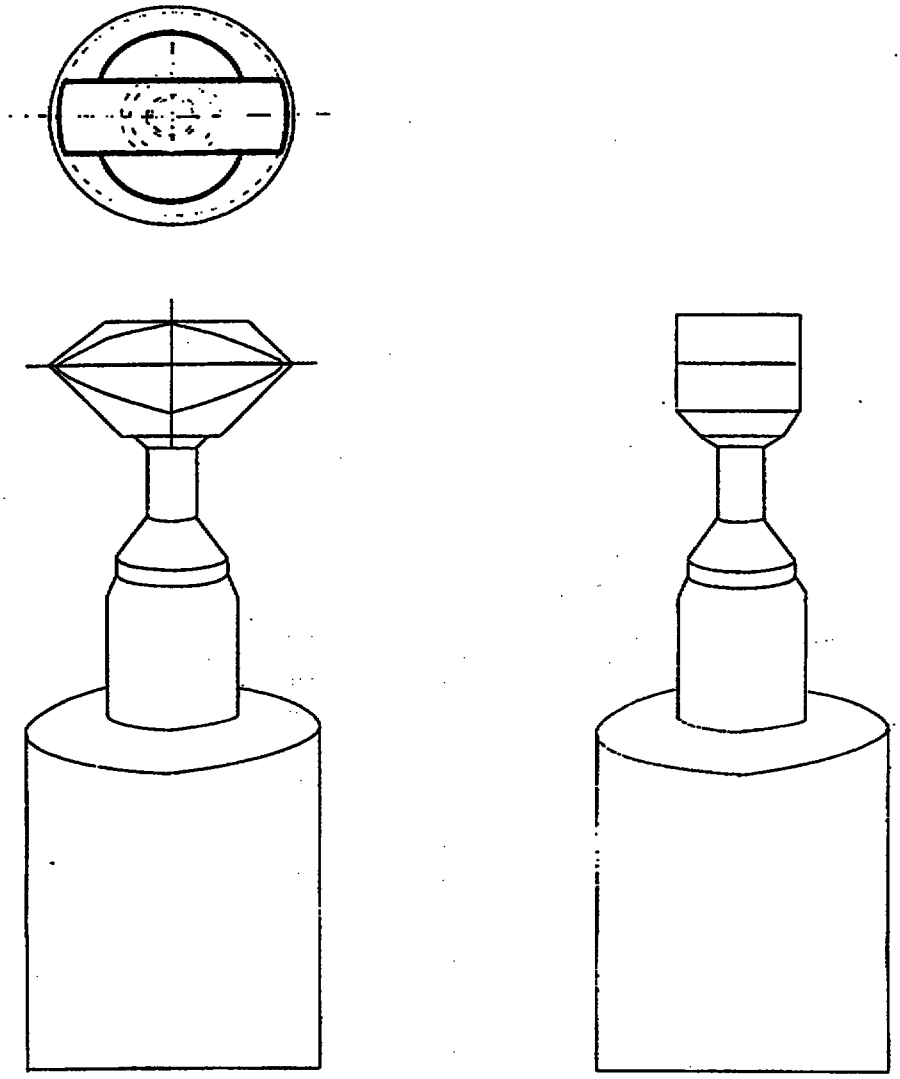


FIGURE 12D



NOT TO SCALE

FIGURE 12E

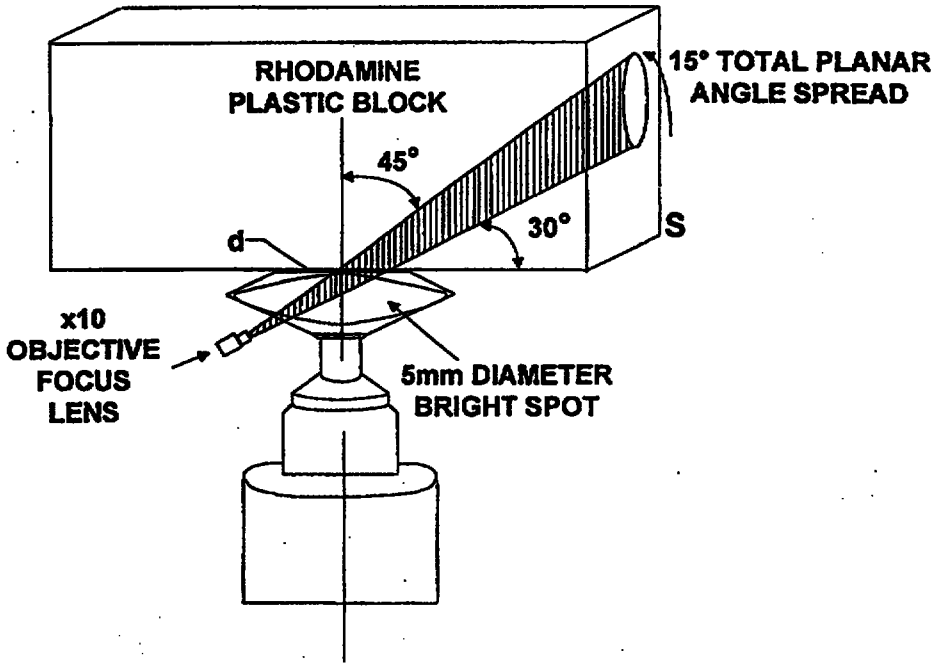


FIGURE 13

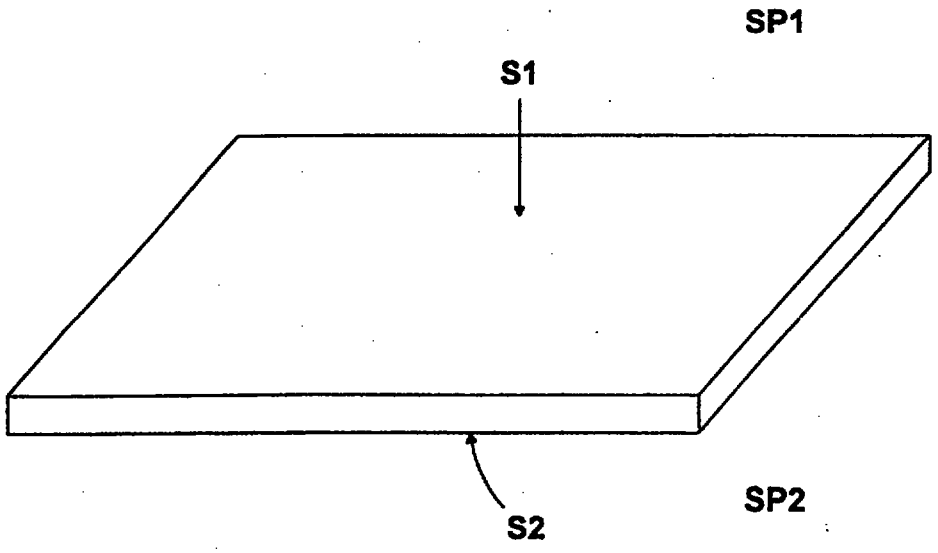


FIGURE 14

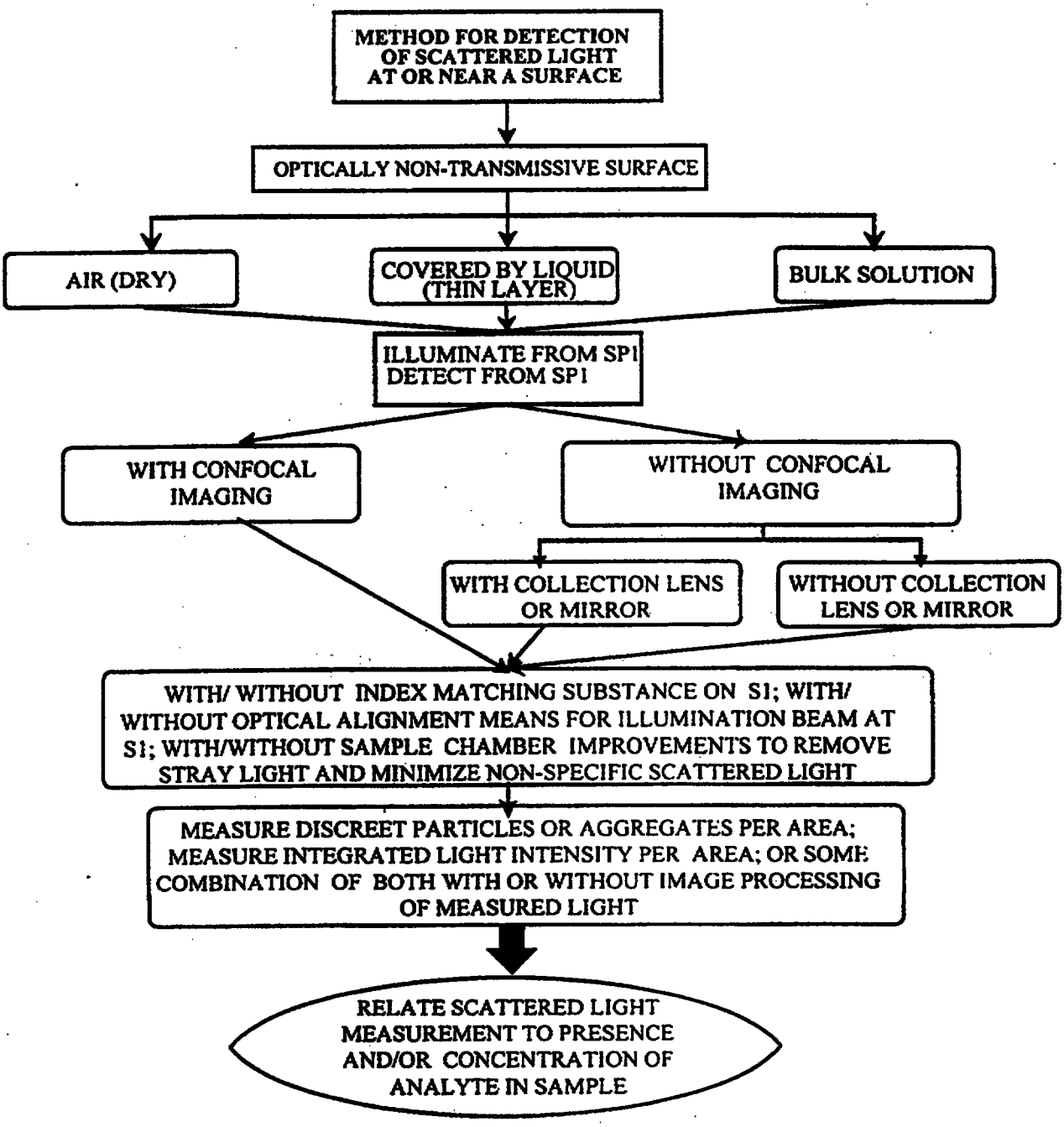

METHOD FOR DETECTION OF SCATTERED LIGHT AT OR NEAR A SURFACE
OPTICALLY NON-TRANSMISSIVE SURFACE
AIR (DRY)
COVERED BY LIQUID (THIN LAYER)
BULK SOLUTION
ILLUMINATE FROM SP1 DETECT FROM SP1
WITH CONFOCAL IMAGING
WITHOUT CONFOCAL IMAGING
WITH COLLECTION LENS OR MIRROR
WITHOUT COLLECTION LENS OR MIRROR
WITH/ WITHOUT INDEX MATCHING SUBSTANCE ON S1; WITH/ WITHOUT OPTICAL ALIGNMENT MEANS FOR ILLUMINATION BEAM AT S1; WITH/WITHOUT SAMPLE CHAMBER IMPROVEMENTS TO REMOVE STRAY LIGHT AND MINIMIZE NON-SPECIFIC SCATTERED LIGHT
MEASURE DISCREET PARTICLES OR AGGREGATES PER AREA; MEASURE INTEGRATED LIGHT INTENSITY PER AREA; OR SOME COMBINATION OF BOTH WITH OR WITHOUT IMAGE PROCESSING OF MEASURED LIGHT
RELATE SCATTERED LIGHT MEASUREMENT TO PRESENCE AND/OR CONCENTRATION OF ANALYTE IN SAMPLE

FIGURE 15A

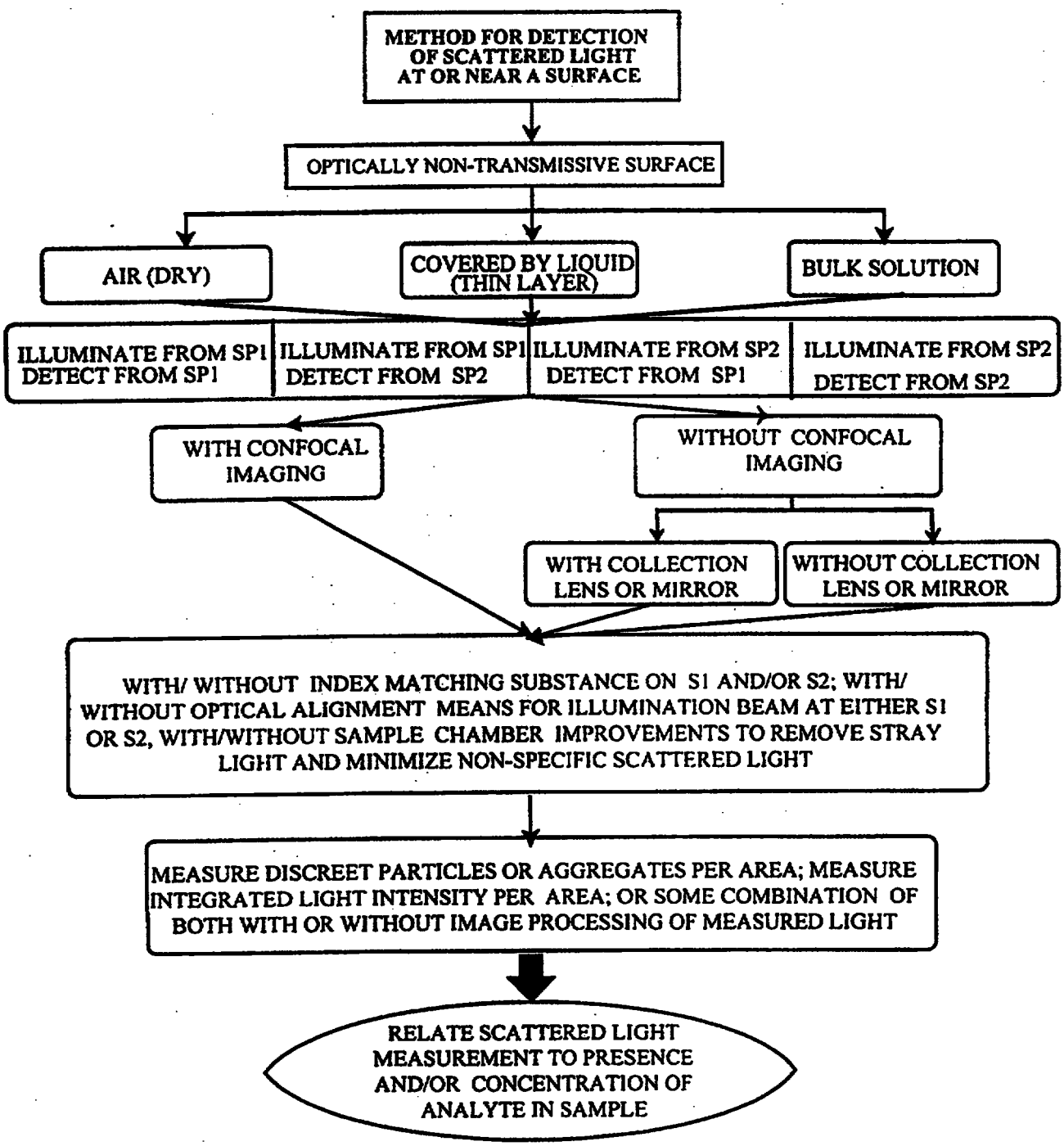

METHOD FOR DETECTION OF SCATTERED LIGHT AT OR NEAR A SURFACE
OPTICALLY NON-TRANSMISSIVE SURFACE
AIR (DRY)
COVERED BY LIQUID (THIN LAYER)
BULK SOLUTION
ILLUMINATE FROM SP1 DETECT FROM SP1
ILLUMINATE FROM SP1 DETECT FROM SP2
ILLUMINATE FROM SP2 DETECT FROM SP1
ILLUMINATE FROM SP2 DETECT FROM SP2
WITH CONFOCAL IMAGING
WITHOUT CONFOCAL IMAGING
WITH COLLECTION LENS OR MIRROR
WITHOUT COLLECTION LENS OR MIRROR
WITH/ WITHOUT INDEX MATCHING SUBSTANCE ON S1 AND/OR S2; WITH/ WITHOUT OPTICAL ALIGNMENT MEANS FOR ILLUMINATION BEAM AT EITHER S1 OR S2, WITH/WITHOUT SAMPLE CHAMBER IMPROVEMENTS TO REMOVE STRAY LIGHT AND MINIMIZE NON-SPECIFIC SCATTERED LIGHT
MEASURE DISCREET PARTICLES OR AGGREGATES PER AREA; MEASURE INTEGRATED LIGHT INTENSITY PER AREA; OR SOME COMBINATION OF BOTH WITH OR WITHOUT IMAGE PROCESSING OF MEASURED LIGHT
RELATE SCATTERED LIGHT MEASUREMENT TO PRESENCE AND/OR CONCENTRATION OF ANALYTE IN SAMPLE

FIGURE 15B

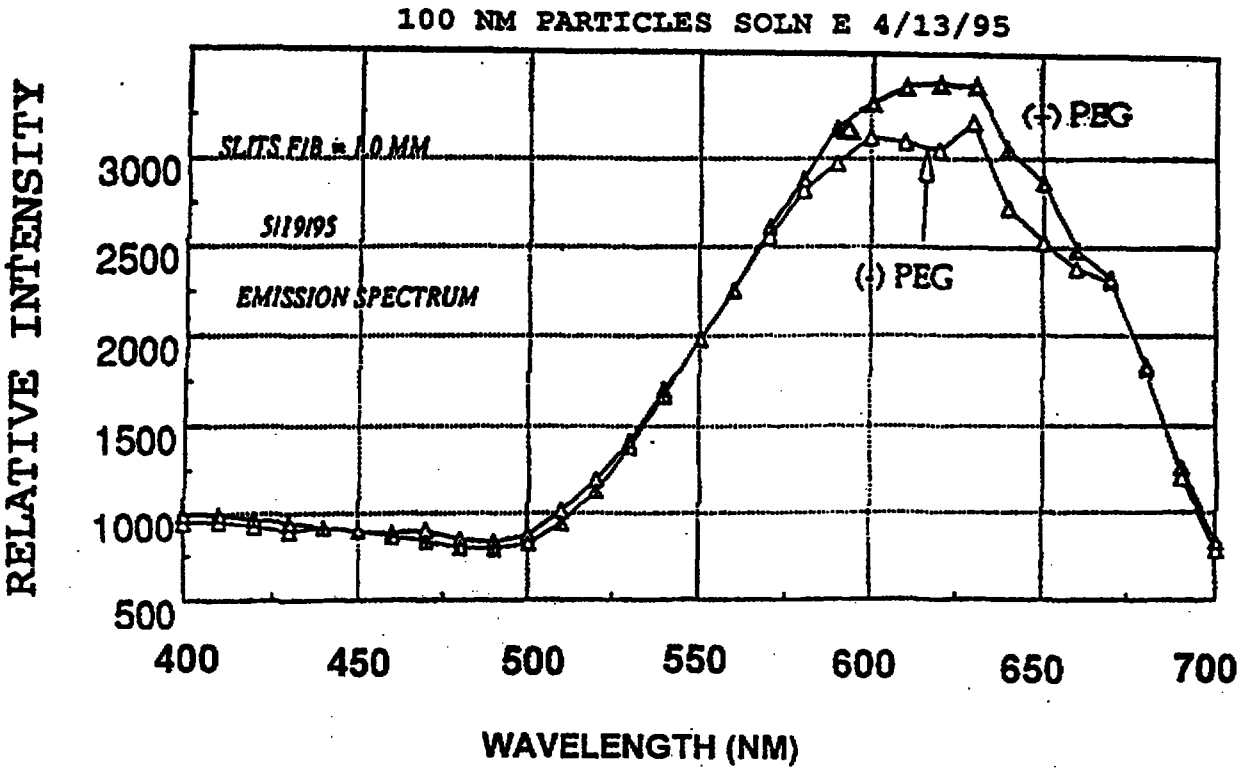

100 NM PARTICLES SOLN E 4/13/95
RELATIVE INTENSITY
3000
2500
2000
1500
1000
500
400
450
500
550
600
650
700
WAVELENGTH (NM)
SLITS F/B = 1.0 MM
5/19/95
EMISSION SPECTRUM
(+) PEG
(-) PEG

FIGURE 16

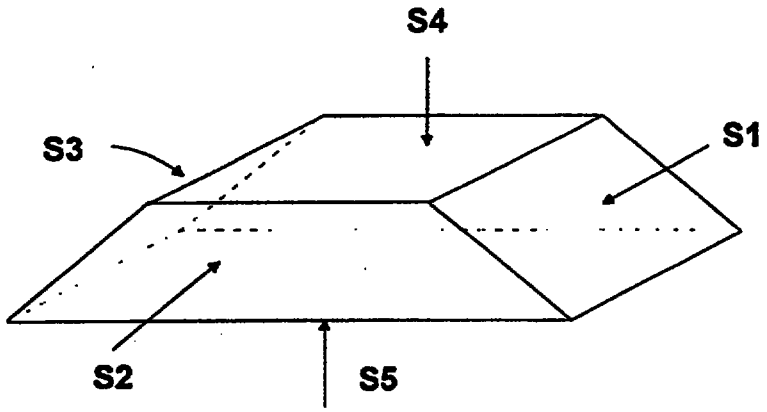


FIGURE 17

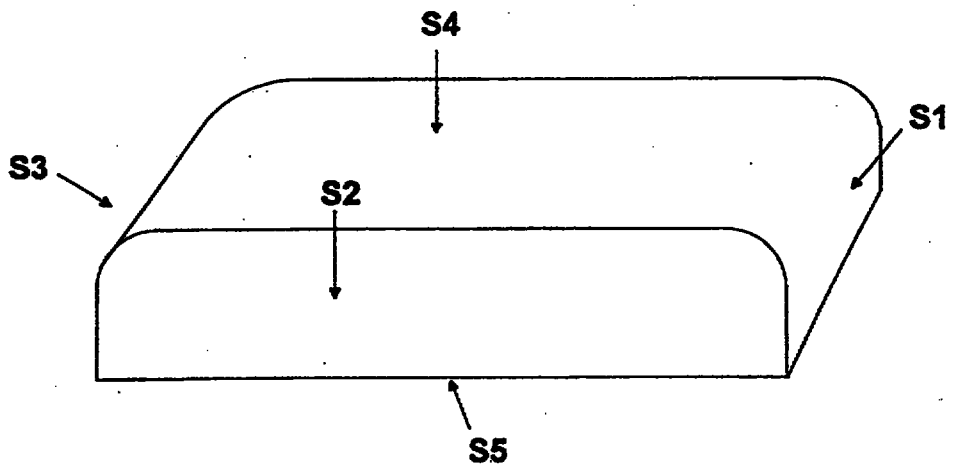


FIGURE 18

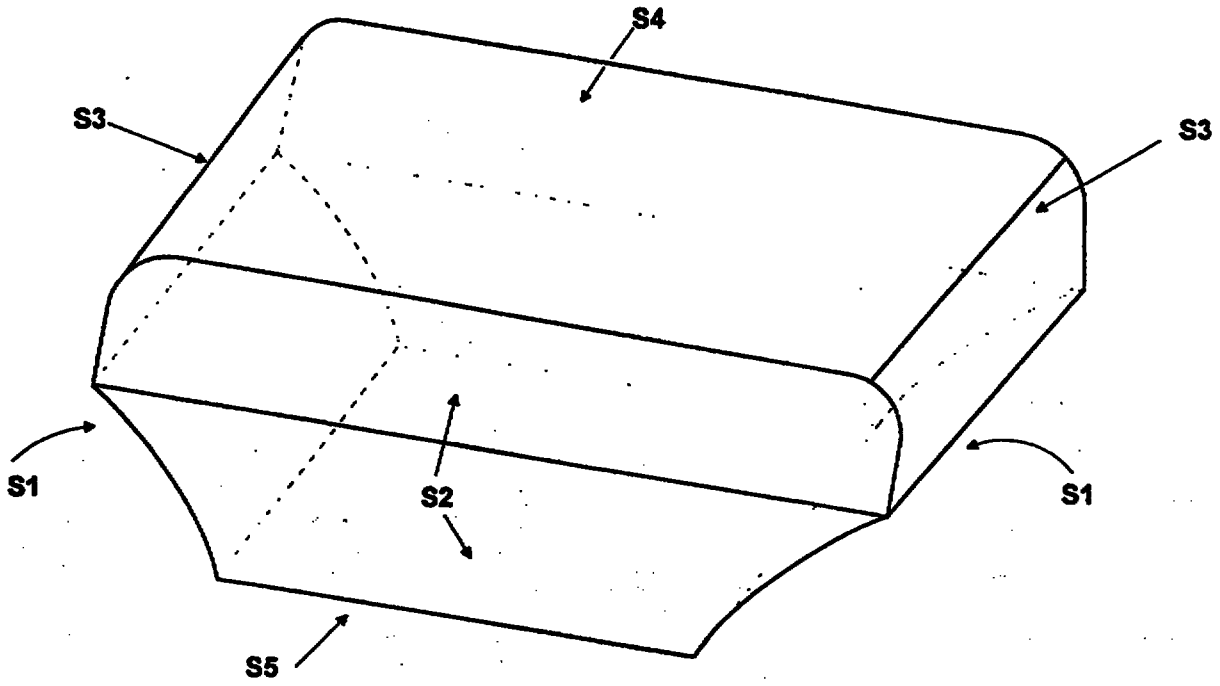


FIGURE 19

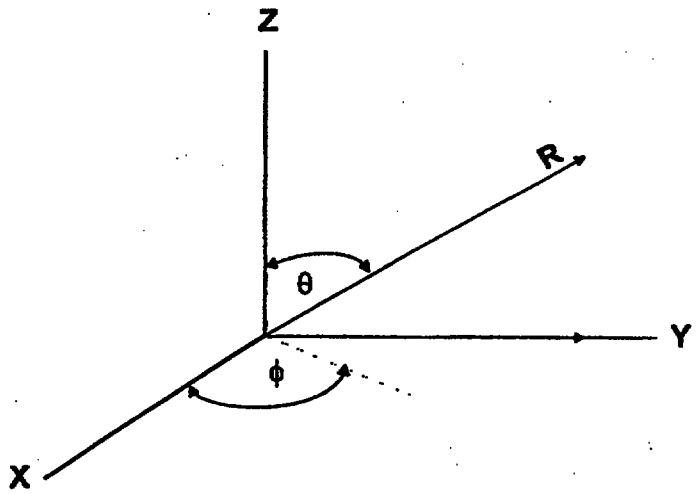


FIGURE 20

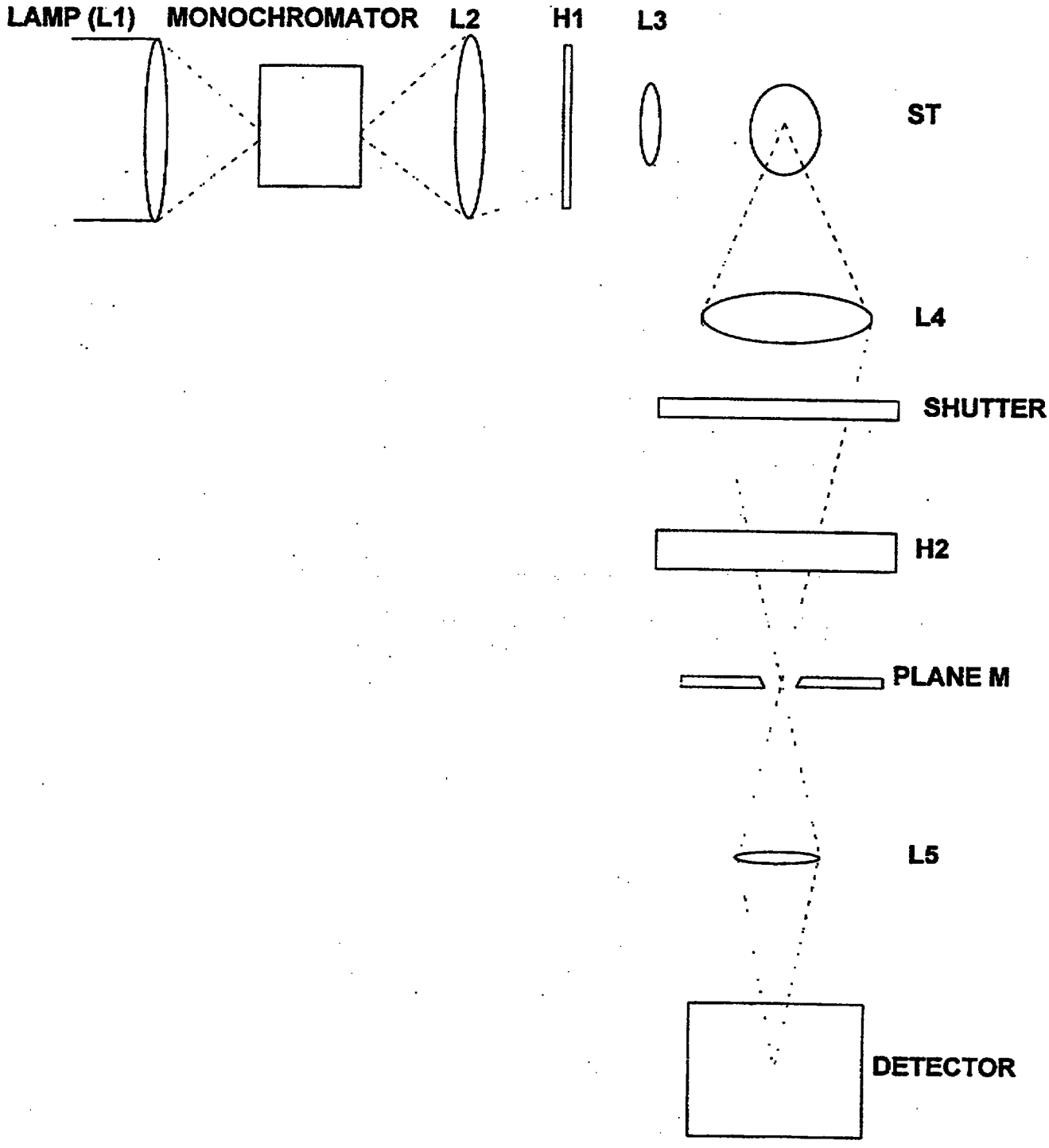

LAMP (L1)
MONOCHROMATOR
L2
H1
L3
ST
L4
SHUTTER
H2
PLANE M
L5
DETECTOR

FIGURE 21

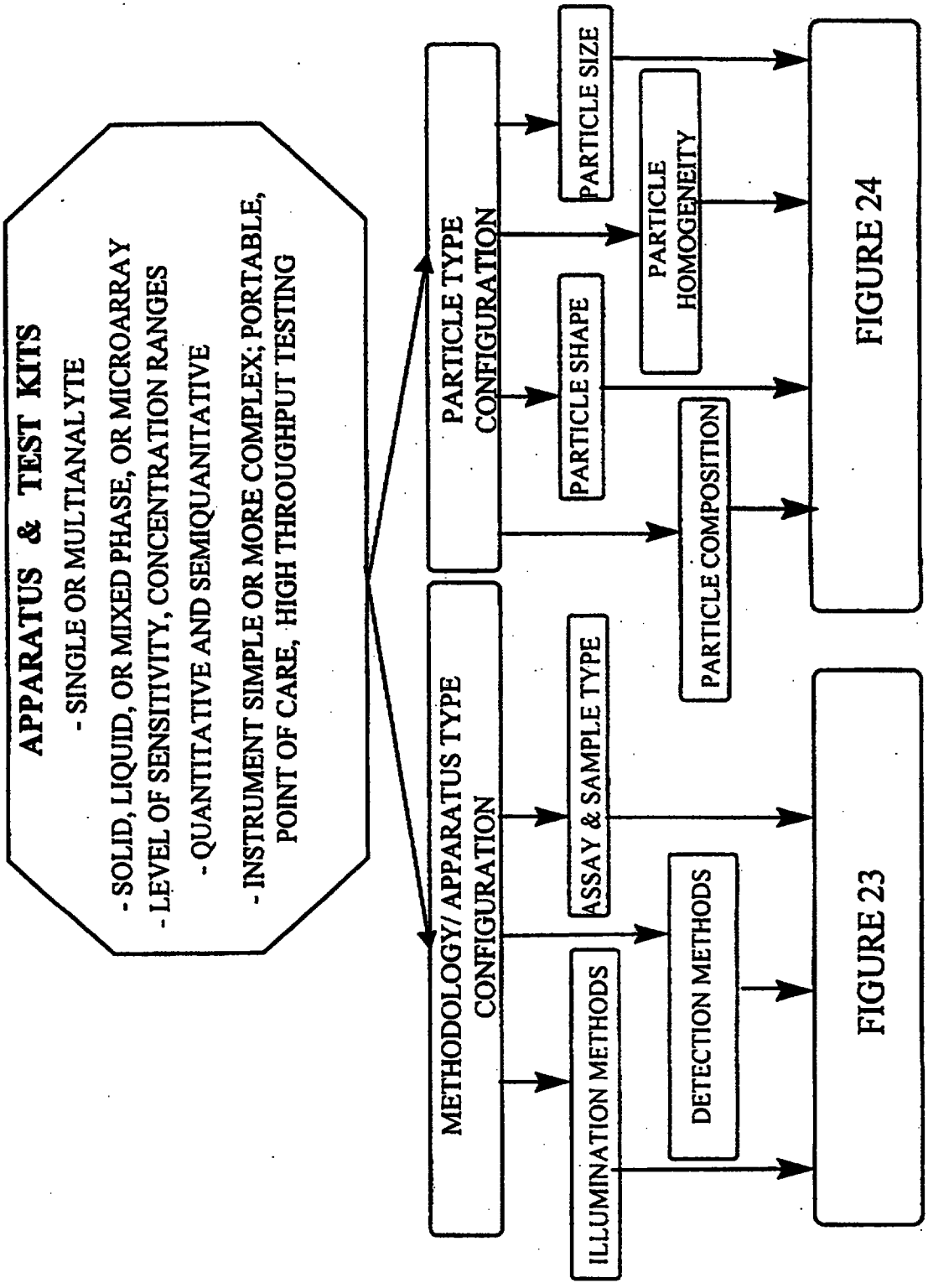

APPARATUS & TEST KITS
- SINGLE OR MULTIANALYTE
- SOLID, LIQUID, OR MIXED PHASE, OR MICROARRAY
- LEVEL OF SENSITIVITY, CONCENTRATION RANGES
- QUANTITATIVE AND SEMIQUANITATIVE
- INSTRUMENT SIMPLE OR MORE COMPLEX; PORTABLE, POINT OF CARE, HIGH THROUGHPUT TESTING
PARTICLE TYPE CONFIGURATION
METHODOLOGY/ APPARATUS TYPE CONFIGURATION
PARTICLE SIZE
PARTICLE SHAPE
PARTICLE HOMOGENEITY
PARTICLE COMPOSITION
ASSAY & SAMPLE TYPE
ILLUMINATION METHODS
DETECTION METHODS
FIGURE 24
FIGURE 23

FIGURE 22

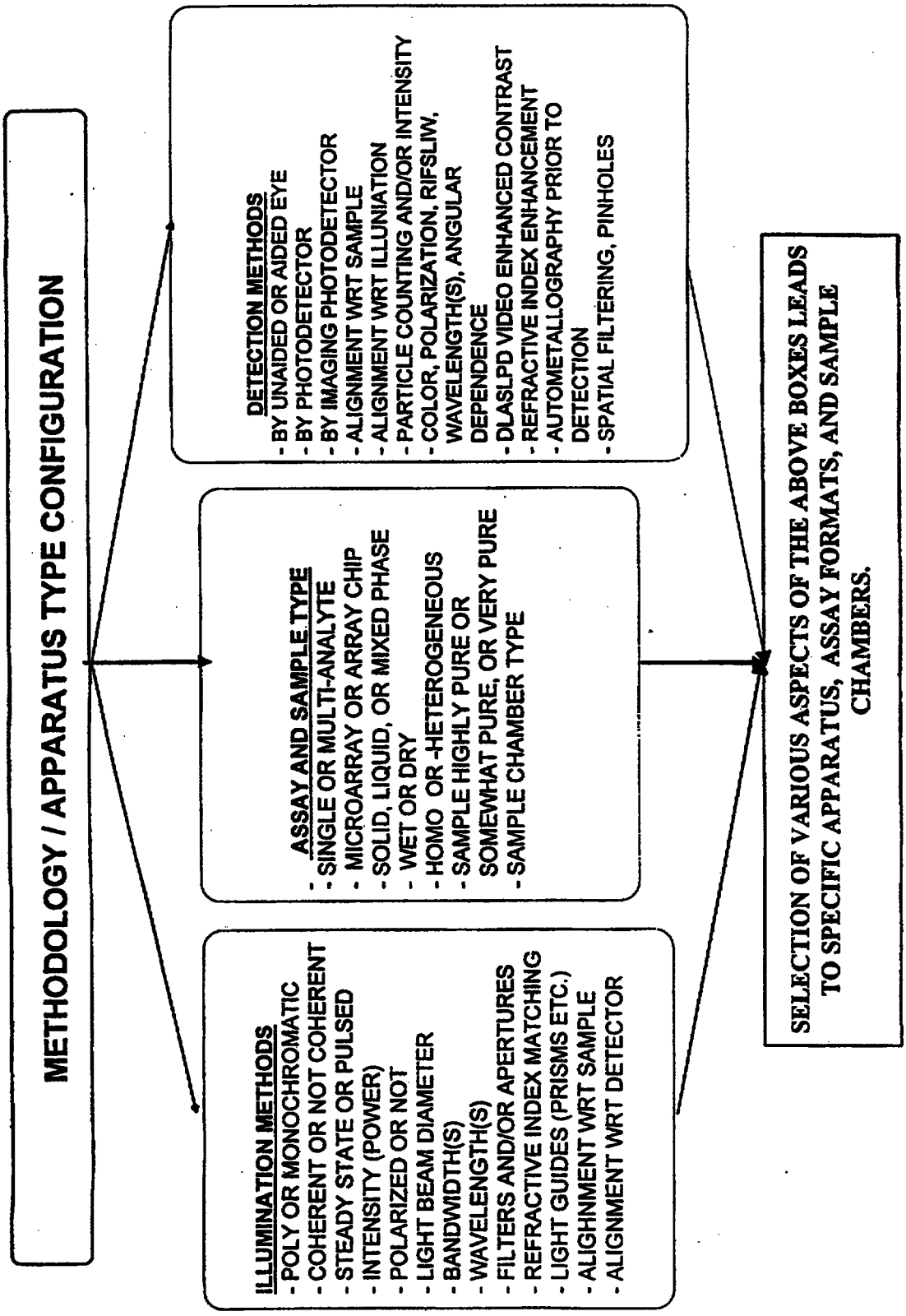

METHODOLOGY / APPARATUS TYPE CONFIGURATION
ILLUMINATION METHODS
- POLY OR MONOCHROMATIC
- COHERENT OR NOT COHERENT
- STEADY STATE OR PULSED
- INTENSITY (POWER)
- POLARIZED OR NOT
- LIGHT BEAM DIAMETER
- BANDWIDTH(S)
- WAVELENGTH(S)
- FILTERS AND/OR APERTURES
- REFRACTIVE INDEX MATCHING
- LIGHT GUIDES (PRISMS ETC.)
- ALIGHNMENT WRT SAMPLE
- ALIGNMENT WRT DETECTOR
ASSAY AND SAMPLE TYPE
- SINGLE OR MULTI-ANALYTE
- MICROARRAY OR ARRAY CHIP
- SOLID, LIQUID, OR MIXED PHASE
- WET OR DRY
- HOMO OR -HETEROGENEOUS
- SAMPLE HIGHLY PURE OR SOMEWHAT PURE, OR VERY PURE
- SAMPLE CHAMBER TYPE
DETECTION METHODS
- BY UNAIDED OR AIDED EYE
- BY PHOTODETECTOR
- BY IMAGING PHOTODETECTOR
- ALIGNMENT WRT SAMPLE
- ALIGNMENT WRT ILLUNIATION
- PARTICLE COUNTING AND/OR INTENSITY
- COLOR, POLARIZATION, RIFSLIW, WAVELENGTH(S), ANGULAR DEPENDENCE
- DLASLPD VIDEO ENHANCED CONTRAST
- REFRACTIVE INDEX ENHANCEMENT
- AUTOMETALLOGRAPHY PRIOR TO DETECTION
- SPATIAL FILTERING, PINHOLES
SELECTION OF VARIOUS ASPECTS OF THE ABOVE BOXES LEADS TO SPECIFIC APPARATUS, ASSAY FORMATS, AND SAMPLE CHAMBERS.

FIGURE 23

SUBSTITUTE SHEET (RULE 26)

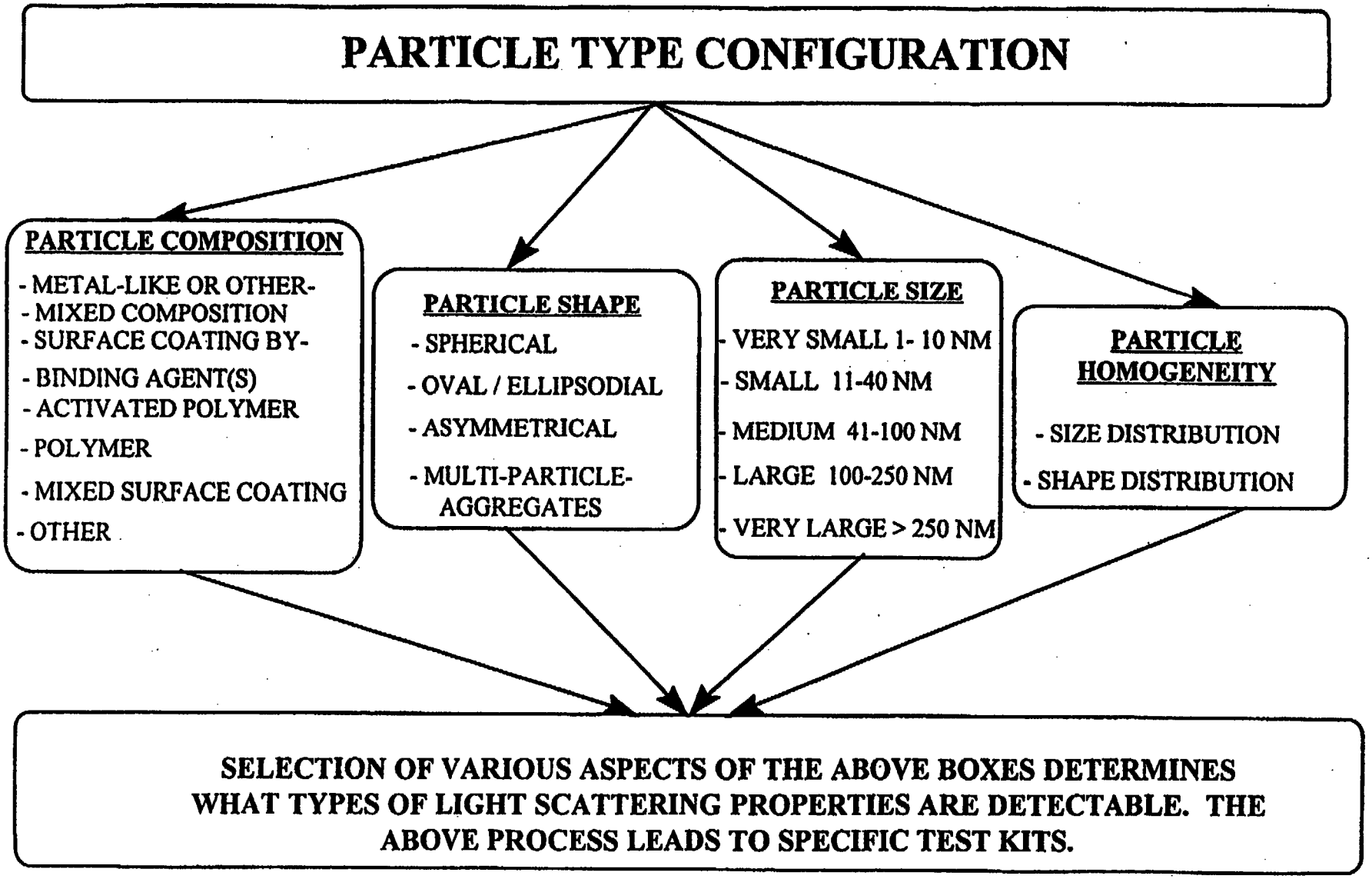


FIGURE 24

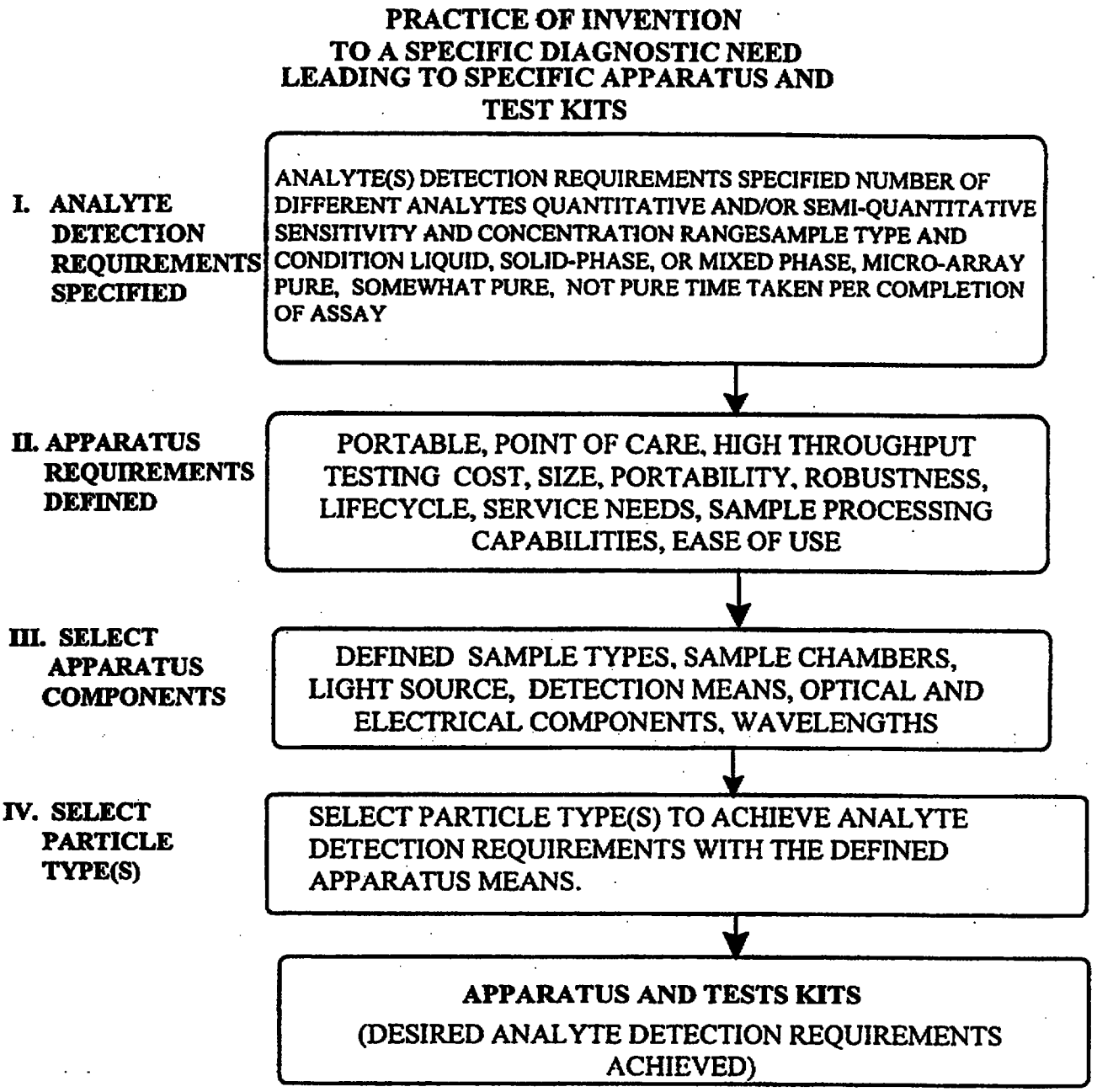

PRACTICE OF INVENTION
TO A SPECIFIC DIAGNOSTIC NEED
LEADING TO SPECIFIC APPARATUS AND
TEST KITS
I. ANALYTE DETECTION REQUIREMENTS SPECIFIED
ANALYTE(S) DETECTION REQUIREMENTS SPECIFIED NUMBER OF DIFFERENT ANALYTES QUANTITATIVE AND/OR SEMI-QUANTITATIVE SENSITIVITY AND CONCENTRATION RANGESAMPLE TYPE AND CONDITION LIQUID, SOLID-PHASE, OR MIXED PHASE, MICRO-ARRAY PURE, SOMEWHAT PURE, NOT PURE TIME TAKEN PER COMPLETION OF ASSAY
II. APPARATUS REQUIREMENTS DEFINED
PORTABLE, POINT OF CARE, HIGH THROUGHPUT TESTING COST, SIZE, PORTABILITY, ROBUSTNESS, LIFECYCLE, SERVICE NEEDS, SAMPLE PROCESSING CAPABILITIES, EASE OF USE
III. SELECT APPARATUS COMPONENTS
DEFINED SAMPLE TYPES, SAMPLE CHAMBERS, LIGHT SOURCE, DETECTION MEANS, OPTICAL AND ELECTRICAL COMPONENTS, WAVELENGTHS
IV. SELECT PARTICLE TYPE(S)
SELECT PARTICLE TYPE(S) TO ACHIEVE ANALYTE DETECTION REQUIREMENTS WITH THE DEFINED APPARATUS MEANS.
APPARATUS AND TESTS KITS
(DESIRED ANALYTE DETECTION REQUIREMENTS ACHIEVED)

FIGURE 25

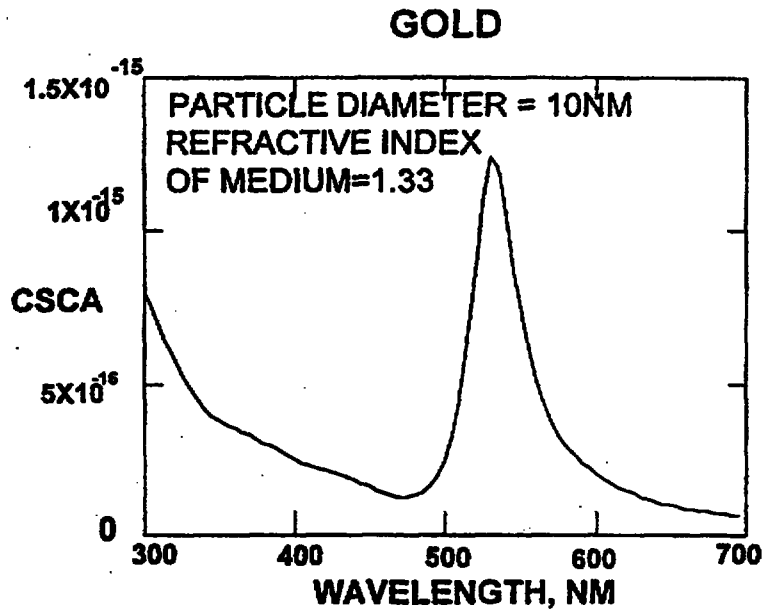


FIGURE 26A

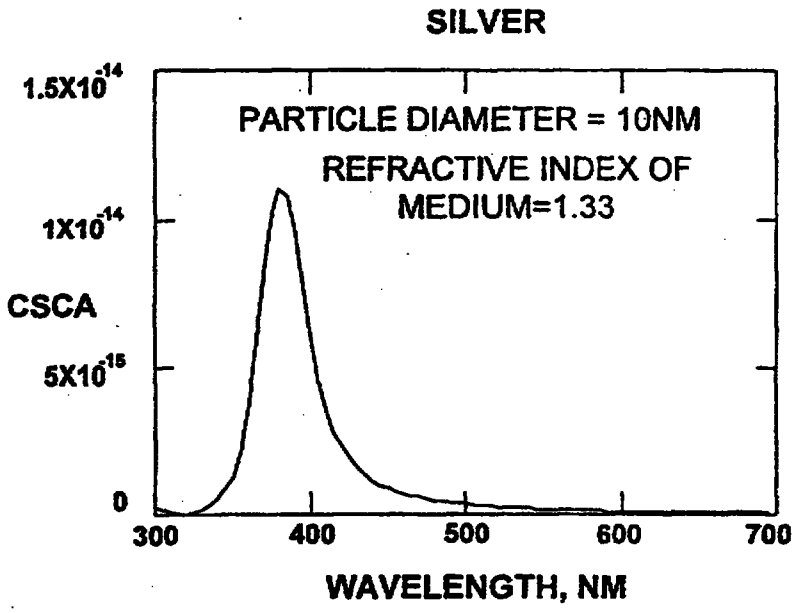


FIGURE 26B

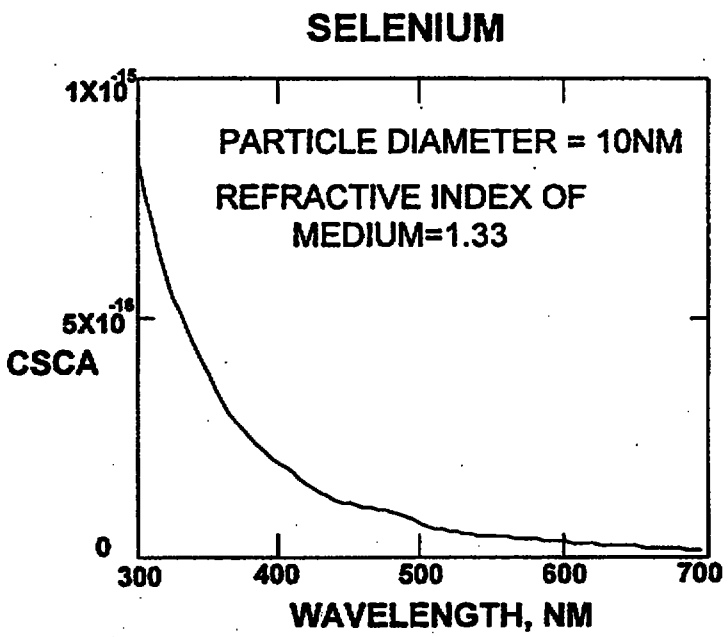


FIGURE 26C

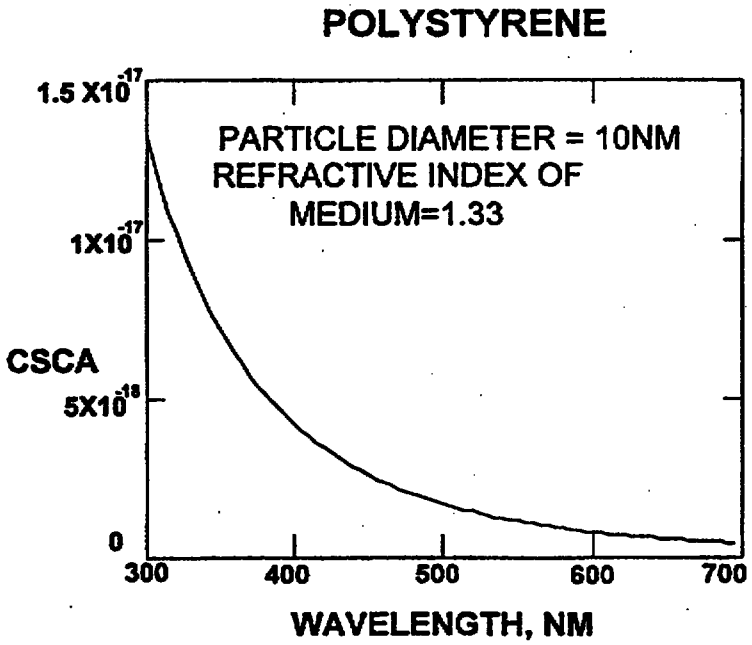


FIGURE 26D

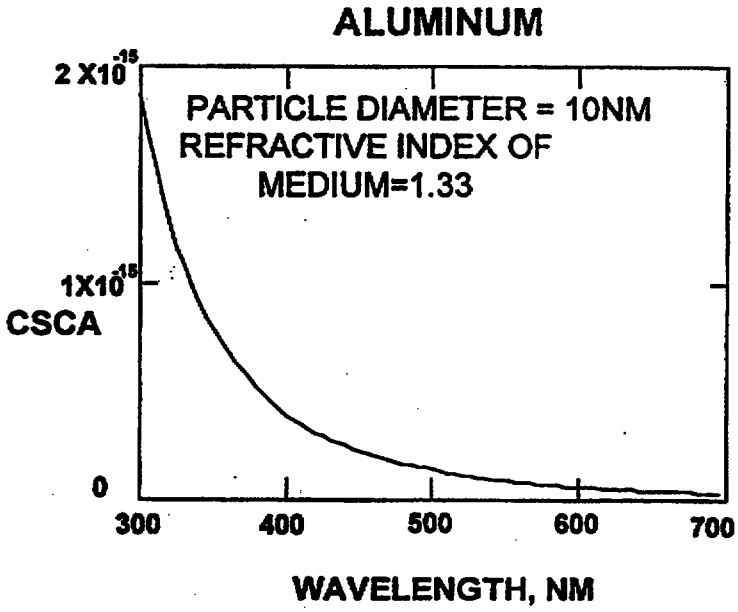


FIGURE 26E

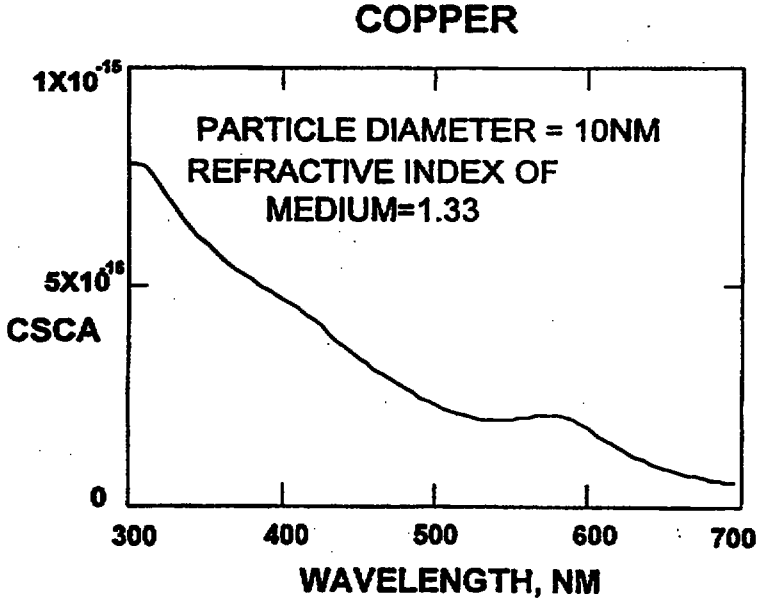


FIGURE 26F

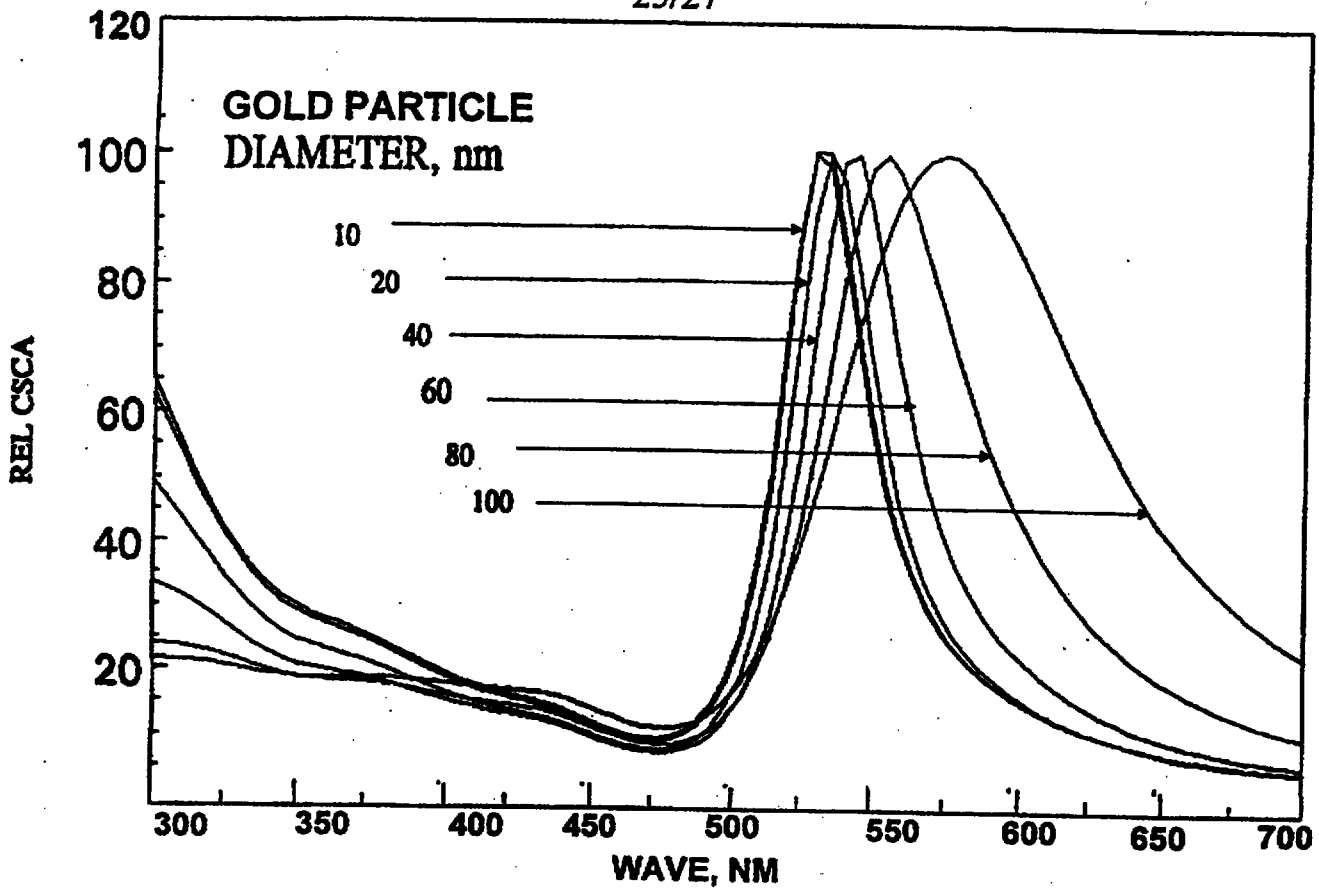


FIGURE 27

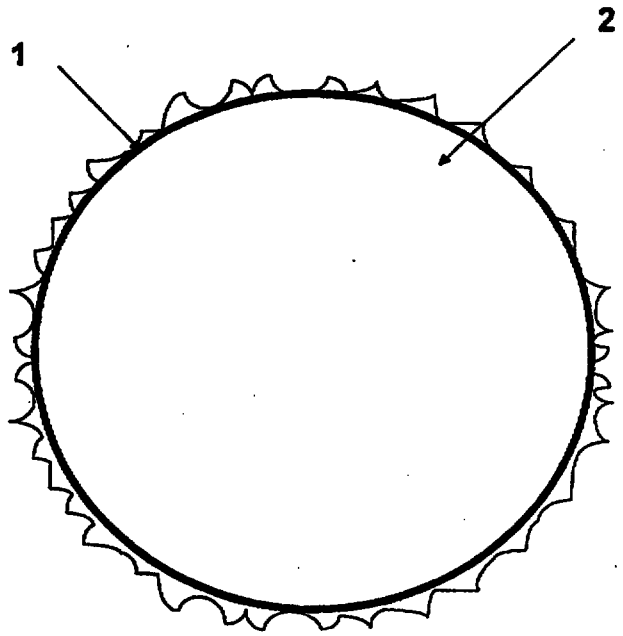


FIGURE 28

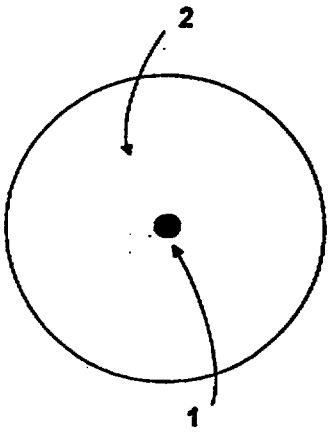


FIGURE 29A

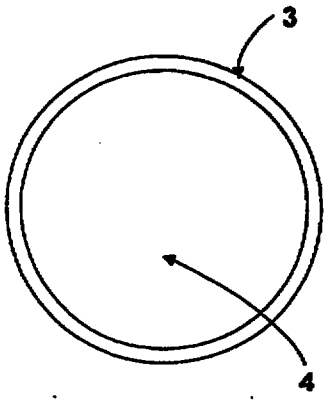


FIGURE 29B

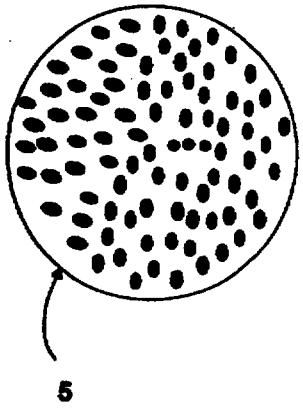


FIGURE 29C

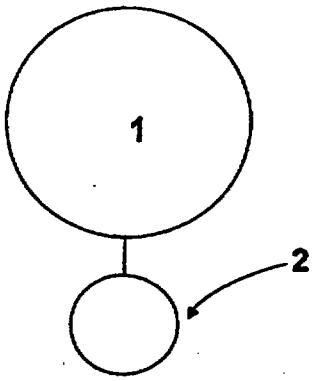


FIGURE 30A

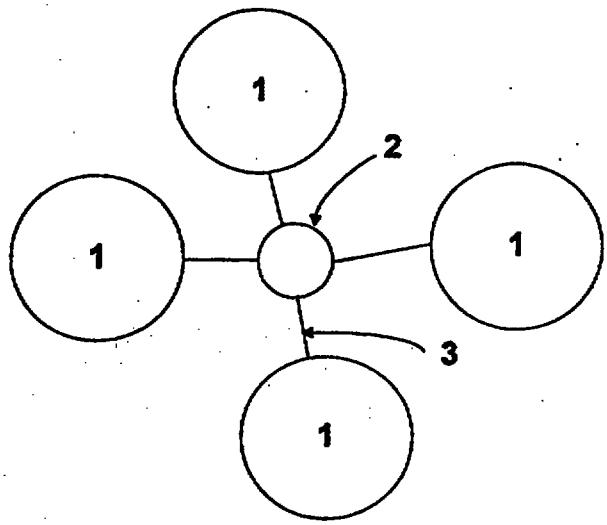


FIGURE 30B

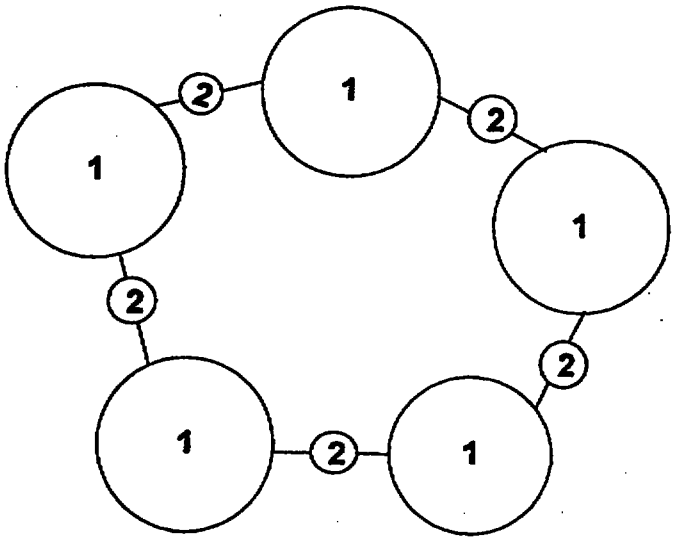


FIGURE 30C

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CLAIMS

1. Method for specific detection of one or more analytes in a sample, comprising the steps of:

specifically ~~[associating]~~ binding any said one or more analytes in said sample with ~~[a]~~ at least one scattered-light detectable particle of a size between 1 and 500 nm inclusive,

illuminating any said particles ~~[associated]~~ bound with said analytes with non-evanescent wave light under conditions which produce scattered light from said particle and in which light scattered from one or more said particles can be detected by a human eye with less than 500 times magnification and without electronic amplification, and

detecting said light scattered by any said particles under said conditions as a measure of the presence of said one or more analytes.

2. The method of claim 1 wherein said particle has a size ~~[suitable for producing]~~ which produces a specific colored light when observed by said human eye and illuminated with ~~[white]~~ polychromatic light.

3. The method of claim 2 wherein the color of said specific colored light provides a measure of the presence or amount of said one or more analytes.

4. The method of claim 1, wherein said detecting comprises measurement of the intensity of scattered light as a measurement of the ~~[concentration]~~ presence or amount of said one or more analytes.

5. The method of claim 1, wherein said detecting comprises measurement of the color of scattered light as a measurement of the ~~[concentration]~~ presence or amount of said one or more anlytes.

REPLACEMENT SHEETS SG
☐ Description : paged ..... to paged ...
☑ Claim(s) : paged ~~212~~ to paged ~~22~~
☐ Drawing(s) : paged ..... to paged ...
☐ Abstract : paged ..... to paged ...
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6. The method of claim 1 wherein said particle has a composition ~~[suitable for producing]~~ which produces a specific colored light when observed by said human eye and illuminated with ~~[white]~~ polychromatic light.

7. The method of claim 1 wherein said particles are ~~[associated with]~~ bound to a solid phase bound analyte.

8. The method of claim 1 wherein said particles are suspended in a liquid phase during said detecting step.

9. The method of claim 1 wherein said analyte is bound to a solid phase.

10. The method of claim 1 wherein said analyte is free in liquid solution.

11. The method of claim 1 wherein said ~~[sample]~~ one or more analytes is bound onto a microarray or array chip comprising discrete areas each of which may contain said one or more analytes.

12. The method of claim 1 wherein said light is polychromatic ~~[white]~~ light.

13. The method of claim 1 wherein a monochromatic light illumination source is used to provide said light.

14. The method of claim 1 wherein said method comprises providing a plurality of different particles each having a different visual appearance when observed by said human eye.

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15. The method of claim 1, wherein said particles are used in a homogeneous assay and wherein two or more particles are brought sufficiently close together so that the light scattering property of ~~[any]~~ at least one particle is altered, wherein said alteration is a measure of the presence or amount of said one or more analytes.

16. The method of claim 1, wherein said particles are used in an assay and wherein two or more particles are brought sufficiently close together so that the light scattering property of the two or more particles can be resolved from single particles and said light scattering is a measure of the presence or amount of said one more analytes.

17. The method of claim 1, wherein said particles are used in a homogeneous assay and wherein two or more particles are brought sufficiently close together so that the light scattering property of the two or more particles can be resolved from single particles and said light scattering is a measure of the presence or amount of said one or more analytes.

18. The method of claim 1, wherein said particles are used in a homogeneous assay and wherein two more particles that are held in close proximity to one another are caused to be separated so that the light scattering property of any one particle is altered, wherein said alteration is a measure of the presence or amount of said one or more analytes.

19. The method of claim 1, wherein said particles are used in a homogeneous assay and wherein two or more particles are linked together by one or more molecular interactions, wherein the molecular interaction holding the particles together is disrupted so that one or more particles is released from the molecular interaction, wherein said released particle or particles is a measure of the presence or amount of said one or more analytes.

20. The method of claim 1 wherein said particle is a gold or silver particle.

21. The method of claim 1, wherein said particle is coated with gold.

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22. The method of claim 1 wherein said light is directed toward said particle by a prism or other light guide system.

23. A population of specifically detectable metal-like light scattering particles formed from at least one light scattering material, said particle comprising at least one additional material on its surface to provide chemical stability and an ability to said particle to bind to an analyte, wherein said population of particles is adapted to sufficient homogeneity in size so that one or more specific light scattering properties of the individual particles in said population are similar from particle to particle.

24. The particle of claim 23 wherein said particles are formed from a material selected from the group consisting of metal, metal compound, metal oxide, semiconductor, and superconductor.

25. The population of claim 23 wherein said particles are formed of a mixed composition comprising a metal-like material as one part of said composition.

26. A specifically detectable light scattering particle reagent formed from at least one light scattering material selected from the group consisting of a metal, a metal compound, a metal oxide, a semiconductor, and a superconductor having a diameter between 1 and 500 nanometers wherein said particle reagent comprises a base molecule adapted to bind to the surface of the particle and adapted to bind to a binding agent, said particle reagent capable of binding to an analyte.

27. The population or particle of claim 23 or 26 wherein said particles have a coat selected from the group consisting of a polymer, an inorganic material, an organic material, a proteinaceous material, a base molecule material, and a binding agent.

28. The population or particle of claim 23 or 26 wherein said particles are spherical.

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29. The population or particle of claim 28 wherein said particles are oval or ellipsoidal.

30. The population or particle of claim 29 wherein said particles are asymmetrical.

31. The population or particles of claim 23 or 26 wherein said particles have a size distribution with a coefficient of variation less than 5%.

32. The population or particle of claim 23 or 26 wherein said particles have a size distribution with a coefficient of variation of less than 10%.

33. The population or particle of claim 23 or 26 wherein said particles have a size distribution with a coefficient of variation of less than 15%.

34. The population of claim 23 wherein said particles comprise gold.

35. The population of claim 23 wherein said articles comprise a mixed composition of gold and silver.

36. The population of claim 23 wherein said particles are composed of silver and a magnetic or ferro electric material.

37. The population of claim 23 wherein said particles are composed of gold and a magnetic or ferro electric material.

38. The population of claim 23 wherein said particles are composed of a mixture of metal-like materials and a magnetic or ferro electric material.

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39. The population of claim 23 wherein said particles are composed of gold with a surface coating selected from the group consisting of polymer, protein, nucleic acid inorganic compound and organic compound, base material molecule, binding agent, and wherein said particles have a diameter of between 10 and 50 nanometers inclusive and produce a green scattered light when illuminated with white light.

40. The population of claim 23 wherein said particles are composed of gold with a surface coating selected from the group consisting of polymer, protein, inorganic compound and organic compound, base material molecule, binding agent, and the diameter of said particles are between 50 and 70 nanometers inclusive and produces a yellow-green to yellow-scattered light color when illuminated with white light.

41. The population of claim 23 wherein said particles are composed of gold with a surface coating selected from the group consisting of polymer, protein, inorganic compound, an organic compound, base material molecule, binding agent, and said particles have a diameter between about 70 and 120 nanometers and produces an orange to orange-red scattered light color when illuminated with white light.

42. The population of claim 23 wherein said particles are composed of gold with a surface coating selected from the group consisting of polymer, protein, inorganic compound, an organic compound, base material molecule, binding agent, and said particles have a diameter of greater than 120 nanometers and less than one micron and produces an orange to orange-red scattered light color when illuminated with white light.

43. The population of claim 23 wherein said particles are composed of silver with a surface coating selected from the group consisting of polymer, protein, inorganic compound, an

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organic compound, base material molecule, binding agent, and said particles have a diameter between about 5 and 50 nanometers and produces a blue scattered light color when illuminated with white light.

44. Apparatus for analysis of a solid phase sample, comprising:

a light source angled such that light is delivered to said sample and the scattered light from any particles bound to said sample can be maximally detected, wherein said apparatus is constructed and arranged such that any said particles associated with said sample can be illuminated with said light under conditions which produce scattered light from said particles and in which light scattered from one or more said particles can be detected by a human eye with less than 500 times magnification and without electronic amplification.

45. The apparatus of claim 44 wherein a collection lens detector is placed outside the intensity envelope of the forward direction of scattered light from said sample.

46. The apparatus of claim 44 comprising a particle counter having necessary computer software or firmware configured and arranged to detect said scattered light.

47. The apparatus of claim 44 wherein a collecting lens is provided in said apparatus essentially perpendicular to the surface upon which said sample is to be applied.

48. The apparatus of claim 44 wherein said apparatus is constructed and arranged to allow assay of a microarray wherein each separate area of said microarray has a dimension between ten square microns and one square millimeter.

49. The method of claim 1, wherein said particle is spherical and has a size between 10 and 200 nm inclusive.

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50. The method of claim 1, wherein said particle is spherical and has a size between 20 and 200 nm inclusive.

51. The method of claim 1, wherein said particle spherical and has a size between 40 and 120 nm inclusive.

52. The method of claim 1, wherein said particle spherical and has a size between 80 and 120 nm inclusive.

53. The method of claim 1, wherein said particle spherical and has a size between 1 and 10 nm inclusive.

54. The method of claim 1, wherein said particle spherical and has a size between 11 and 40 nm inclusive.

55. The method of claim 1, wherein said particle spherical and has a size between 100 and 250 nm inclusive.

56. The method of claim 1, wherein said particle spherical and has a size greater than 250 nm.

57. The method of claim 1, wherein said particle is spherical.

58. The method of claim 1, wherein said particle is ellipsoidal.

59. The method of claim 1, wherein said particle is asymmetric.

60. The method of claim 1, wherein said particle comprises a multiple particle aggregate.

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61. The method of claim 1, wherein said illuminating and said detecting are on opposite sides of a solid phase surface.

62. The method of claim 61, wherein said illuminating is from below a solid phase surface and said detecting is from above said solid phase surface.

63. The method of claim 1, wherein said illuminating and said detecting are on the same side of a solid phase surface.

64. The method of claim 63, wherein said illuminating and said detecting are from above said solid phase surface.

65. The method of claim 63, wherein said detecting is at an angle to said surface outside the cone of the illuminating light and outside the angle of reflection of the illuminating light.

66. The method of claim 1, wherein said detecting is at angles outside the angles of incident, transmitted, refracted and reflected light from the illuminating light.

67. The method of claim 1, wherein said illuminating is at an angle to an interface of a solid phase surface less than the critical angle, and said detecting is at an angle less than said critical angle.

68. The method of claim 1, wherein said detecting is at right angles to the surface of said sample.

69. The method of claim 1, wherein said detecting is at angles outside the envelope of forward scattered light.

70. The method of claim 1, wherein said analyte comprises a protein or peptide.

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71. The method of claim 70, wherein said protein comprises a cell surface constituent.

72. The method of claim 70, wherein said protein comprises a receptor.

73. The method of claim 70, wherein said protein comprises an antibody.

74. The method of claim 1, wherein said one or more analytes comprises one or more analytes on a cell surface, in a cell lysate, or in a chromosome preparation.

75. The method of claim 1, wherein said analyte comprises an antigen.

76. The method of claim 1, wherein said analyte comprises a pharmaceutical agent.

77. The method of claim 1, wherein said one or more analytes comprise combinatorial molecules.

78. The method of claim 1, wherein said analyte comprises a hormone.

79. The method of claim 1, wherein said analyte comprises a lipid or carbohydrate.

80. The method of claim 1, wherein said analyte comprises a nucleic acid molecule.

81. The method of claim 1, wherein two or more different nucleic acid probe molecules are each bound to a said particle, and each said probe will bind to a different target sequence in a nucleic acid analyte.

82. The method of claim 1, wherein a binding agent is attached to said

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particles.

83. The method of claim 82, wherein said binding agent is adsorbed to said particle.

84. The method of claim 82, wherein said binding agent is covalently attached.

85. The method of claim 82, wherein said binding agent is attached to a base material.

86. The method of claim 85, wherein a plurality of different base materials are used.

87. The method of claim 82, wherein said binding agent is proteinaceous.

88. The method of claim 87, wherein said binding agent is an antibody.

89. The method of claim 87, wherein said binding agent is a receptor.

90. The method of claim 82, wherein said binding agent is a nucleic acid.

91. The method of claim 82, wherein said binding agent is selected from the group consisting of antigens, lectins, carbohydrates, biotin, avidin, streptavidin, and pharmaceutical agents.

92. The method of claim 82, wherein said particles are coated with a stabilizing layer.

93. The method of claim 92, wherein said stabilizing layer is a thin metal coating.

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94. The method of claim 1, wherein said particles comprise coated particles.

95. The method of claim 94, wherein the coating of said coated particles comprises a stabilizing coating.

96. The method of claim 95, wherein the coating comprises polymers, proteins, peptides, hormones, antibodies, nucleic acids, or receptors.

97. The method of claim 95, wherein said stabilizing layer is a thin metal coating.

98. The method of claim 1, wherein the particles detected are detected following metallographic enlargement.

99. The method of claim 98, wherein said metallographic enlargement is with gold or silver.

100. The method of claim 9, further comprising a wash step.

101. The method of claim 1, further comprising refractive index enhancement of the sensitivity and specificity of detection of said particles.

102. The method of claim 1, further comprising use of one or more narrow band pass filters in detection of said scattered light.

103. The method of claim 1, further comprising filtering scattered light through a polarization filter or a bandpass filter or both.

104. The method of claim 1, wherein said detecting comprises magnification with a microscope 2 to 500 times.

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105. The method of claim 104, wherein said magnification is 10 to 100 times.

106. The method of claim 104, wherein said detecting comprises use of confocal microscopy.

107. The method of claim 1, wherein said detecting is by eye.

108. The method of claim 1, wherein said detecting uses a photodetector.

109. The method of claim 108, wherein said photodetector comprises a photodiode or photodiode array.

110. The method of claim 108, wherein said photodetector comprises a photomultiplier tube.

111. The method of claim 108, wherein said photodetector comprises a camera or video camera.

112. The method of claim 108, wherein said photodetector comprises a charge-coupled device.

113. The method of claim 11, wherein said illuminating comprises scanning with a light beam.

114. The method of claim 113, wherein said scanning is performed by moving an illuminating beam.

115. The method of claim 113, wherein said scanning is performed by moving the sample.

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116. The method of claim 11, wherein separate spatially addressable sites on said array are separately illuminated.

117. The method of claim 11, wherein the array is illuminated with a broad light beam.

118. The method of claim 1, wherein said illuminating light is non-polarized.

119. The method of claim 1, wherein said illuminating light is polarized.

120. The method of claim 1, wherein said illuminating light is from a filament lamp source.

121. The method of claim 1, wherein said illuminating light is from a discharge lamp source.

122. The method of claim 1, wherein said illuminating light is from a laser.

123. The method of claim 1, wherein said illuminating light is from a light emitting diode.

124. The method of claim 1, wherein said illuminating light is pulsed. P19 L22

125. The method of claim 1, wherein said illuminating light is constant.

126. The method of claim 1, wherein said illuminating light is coherent.

127. The method of claim 1, wherein said illuminating light is noncoherent.

128. The method of claim 1, wherein said analyte is in a sample selected from the group consisting of water, urine, blood, sputum, and tissue.

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129. The method of claim 1, wherein said one or more analytes comprises a plurality of different analytes and said one or more particles comprises a plurality of different particles.

130. The method of claim 1, wherein said one or more analytes are in a highly scattering medium.

131. The method of claim 1, wherein said particles are mixed composition particles.

132. The method of claim 1, wherein spatial filtering is utilized, thereby reducing non-specific background light.

133. The method of claim 1, wherein cutoff filters, narrow bandpass filters, or both are utilized, thereby reducing non-specific background light.

134. The method of claim 1, wherein scattered light from individual particles is detected.

135. The method of claim 1, wherein scattered light from a plurality of particles is detected.

136. The method of claim 1, wherein said particle has a size that produces a specific intensity level when illuminated under said conditions.

137. The method of claim 1, wherein said particle has a size that produces a specific polarization when illuminated under said conditions.

138. The method of claim 1, wherein said particle has a size that produces a specific angular dependency to said scattered light.

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139. The method of claim 1, wherein said particle has a composition that produces a specific intensity level when illuminated under said conditions.

140. The method of claim 1, wherein said particle has a composition that produces a specific polarization when illuminated under said conditions.

141. The method of claim 1, wherein said particle has a composition that produces a specific angular dependency to said scattered light.

142. The method of claim 1, wherein said one or more analytes is in a microtiter plate.

143. The method of claim 1, wherein said one or more anlytes is in a flow-based system.

144. The method of claim 1, wherein said one or more analytes is in a device selected from the group consisting of a flow cytometry apparatus, a microchannel, and a capillary.

145. The method of claim 1, wherein said particle is a metal or metal-like particle.

146. The method of claim 1, wherein the light path of said illuminating or scattered light passes through an immersion oil or other refractive index matching material.

147. The method of claim 1, wherein said method is performed in a homogeneous format.

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(57) Abstract

Method for specific detection of one or more analytes in a sample. The method includes specifically associating any one or more analytes in the sample with a scattered-light detectable particle, illuminating any particle associated with the analytes with light under conditions which produce scattered light from the particle and in which light scattered from one or more particles can be detected by a human eye with less than 500 times magnification and without electronic amplification. The method also includes detecting the light scattered by any such particles under those conditions as a measure of the presence of the analytes.