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**Sintering material on the basis of silver-tin oxide for
electrical contacts and process for its manufacture**

Description:

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The invention relates to a sintering material manufactured by powder metallurgy on the basis of silver-tin oxide with added amounts of indium oxide and bismuth oxide for electrical contacts which are used for switching rated
10 currents between 20 and 100 Ampere and also to a process for its manufacture.

Silver/metal and silver/metal-oxide composite materials have proved useful for the manufacture of electrical
15 contacts in low-voltage switching devices. By way of silver/metal composite material use is most frequently made of silver/nickel, the main field of application of which relates to relatively low currents. In the case of high currents, up until a few years ago use was made almost
20 exclusively of silver-cadmium oxide. Due to environmental pollution, however, attempts have increasingly been made to replace the cadmium oxide with other oxides. Meanwhile, tin oxide has gained acceptance in many fields as an alternative to cadmium oxide.

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By virtue of the higher thermal stability of tin oxide the silver-tin-oxide composite material displays a clearly reduced burn-off rate in comparison with silver-cadmium oxide, resulting in a longer operational life in the
30 switching device. The disadvantage of AgSnO_2 is that it has a tendency to form a covering layer and consequently tends to give rise to greater heating effects in the switching devices. By means of certain additives such as WO_3 or MoO_3 , it has been possible to cope with this problem. These
35 materials have proved outstandingly valuable in the case of switching devices that need to withstand high thermal loads. AgSnO_2 with these additives has proved particularly

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effective in switching devices with rated currents of more than 100 Amp and under so-called AC4 loading, though in the case of lower switching currents the operational life of these materials is relatively short.

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The $\text{AgSnO}_2\text{WO}_3/\text{MoO}_3$ material is manufactured by powder metallurgy using extrusion technology. Manufacture by powder metallurgy has the advantage that use can be made of additives of any type and in any quantity. Hence the material can be optimised selectively with a view to achieving particular properties such as welding strength or degree of heating. In addition, the combination of powder metallurgy with extrusion technology permits a particularly high degree of economy in the production of the contact pieces.

An internally oxidised $\text{AgSnO}_2/\text{In}_2\text{O}_3$ material also finds application. This material, which is described in DE-OS 24 28 147, contains, besides 5-10 % SnO_2 , also 1-6 % In_2O_3 . However, internally oxidised materials have the disadvantage that the additives have to be selected with due regard to the oxidation kinetics of the materials. A selective change in the concentrations of the oxide additives in order to influence particular switching properties is frequently not possible by reason of the oxidation kinetics. However, the $\text{AgSnO}_2/\text{In}_2\text{O}_3$ has the disadvantage that it results in high excess temperatures in the course of switching.

In DE-OS 27 54 335 a contact material is described which besides silver contains 1.6 to 6.5 Bi_2O_3 and 0.1 to 7.5 SnO_2 . This material can be manufactured both by means of internal oxidation and also by powder metallurgy. However, such high contents of Bi_2O_3 result in embrittlement, so that the material can only be manufactured by individual sintering and not by means of the more economic extrusion technology.

- From US-PS 4,680,162 an internally oxidised AgSnO_2 material is known which in the case of tin contents of more than 4.5 % may contain additions of 0.1-5 indium and 0.01-5 bismuth. The metal-alloy powder is compacted and then
- 5 internally oxidised. By means of these additives the inhomogeneous oxide precipitations that are customary in the course of internal oxidation are prevented. However, this material does not display optimal contact properties.
- 10 The publication "Investigation into the Switching Behaviour of new Silver-Tin-Oxide Contact Materials, in Proc. of the 14th Int. Conf. on El. Contacts, Paris 1988, June 20-24, pages 405-409" gives an account of the switching behaviour of electrical contacts made of silver-tin oxide and
- 15 manufactured by powder metallurgy which may contain two additional oxides selected from the group comprising bismuth oxide, indium oxide, copper oxide, molybdenum oxide and tungsten oxide, there being no statement regarding the precise composition of these materials.
- 20 In US-PS 4,695,330 a special process is described for the manufacture of an internally oxidised material with 0.5-12 tin, 0.5-15 indium and 0.01-1.5 bismuth. However, this process is very cost-intensive.
- 25 The manufacture by powder metallurgy of contact materials on the basis of silver-tin oxide by mixing of the powders, cold isostatic pressing, sintering and extrusion so as to form a semi-finished product is known from, for example,
- 30 DE-OS 43 19 137 and DE-OS 43 31 526.
- From US-PS 4,141,727 contact materials made from silver are known which contain bismuth-tin oxide by way of mixed-oxide powder. In addition, in DE-PS 29 52 128 the tin-oxide
- 35 powder is annealed prior to being mixed with the silver powder at 900 to 1600°C.

In medium current ranges from 20 to 100 Amp none of the known AgSnO_2 materials has hitherto been able to replace the toxic material AgCdO entirely, since in this field of application AgCdO exhibits very good switching lives that
5 have not quite been achievable by AgSnO_2 .

The object of the present invention was therefore to develop a sintering material manufactured by powder metallurgy on the basis of silver-tin oxide with added
10 amounts of indium oxide and bismuth oxide for electrical contacts, said material displaying a welding tendency that is as low as possible and excess temperatures that are as low as possible in the course of switching rated currents between 20 and 100 Ampere and, in the case of AC3 loading
15 in switching devices, having an operational life which is similar to that of silver-cadmium oxide. In addition, a process for the manufacture of said sintering material was to be found that is economical and brings about further improvements in the material.

20 This object is achieved in accordance with the invention in that said material consists of 3.2 to 19.9 wt-% tin oxide, 0.05 to 0.4 wt-%, in each case, of indium oxide and bismuth oxide, the remainder being silver.

25 In the current range from 20 to 100 Ampere this material displays an outstanding operational life with excess temperatures lying clearly below 100°C .

30 Particularly good material properties are achieved in the course of manufacture of the material by mixing of the powders, cold isostatic pressing of the powder mixture, sintering at temperatures from 500 to 940°C and extrusion so as to form wires or profiles if more than 60 wt-% of the
35 tin-oxide powder employed exhibits a particle size of more than $1\ \mu\text{m}$ prior to the mixing with the silver powder and the other oxide powders.

It has proved particularly useful to react the bismuth oxide, prior to the mixing with the silver powder and the indium-oxide powder, with the tin-oxide powder so as to form the mixed-oxide powder $\text{Bi}_2\text{Sn}_2\text{O}_7$, which should also exhibit a particle size of more than $1\text{ }\mu\text{m}$ in a proportion amounting to more than 60 wt-%.

Since commercial tin oxide normally has a particle size of less than $1\text{ }\mu\text{m}$ in a proportion amounting to more than 70 wt-%, it is necessary to coarsen this powder. This is preferably effected by annealing the tin-oxide powder or the tin-oxide powder together with the bismuth-oxide powder at temperatures from 700 to 1400°C until more than 60 wt-% of the tin oxide or of the mixed-oxide powder exhibits a particle size of more than $1\text{ }\mu\text{m}$.

The use of these coarsened oxide powders produces, after sintering of the pressed articles, a material that is substantially less brittle than materials having commercial oxide-particle sizes and can therefore be deformed more easily.

The following Examples are intended to elucidate the invention in more detail:

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1. A material with the composition $\text{Ag}_{90}\text{SnO}_2$, 9.4 In_2O_3 , 0.4 Bi_2O_3 , 0.2 was produced by annealing commercial SnO_2 powder, the particle sizes of which pertained to the class of $<1\text{ }\mu\text{m}$ in a proportion amounting to 82 %, in air for 20 h at 1000°C so that the SnO_2 powder exhibited particle sizes pertaining to the class of $<1\text{ }\mu\text{m}$ in a proportion amounting to only 25 %. These powders were mixed together with In_2O_3 and Bi_2O_3 powders and Ag powder, the particle sizes of which were $<63\text{ }\mu\text{m}$ in each case. The mixture was pressed by a cold isostatic method so as to form a billet and sintered for 2 h at 750°C . The billet was then extruded to give it a

profile. In a commercial switching device with a rated current of about 50 A the material achieved an operational life of 2 million switching cycles. This operational life lies clearly above that of the AgSnO_2 materials known hitherto. The excess temperature shows non-critical values of, on average, clearly below 100°C .

2. A material with the composition $\text{Ag}88\text{SnO}_2$, $11.4 \text{ In}_2\text{O}_3$, $0.3 \text{ Bi}_2\text{O}_3$, 0.3 was produced in accordance with Example 1. In a commercial switching device with a rated current of about 50 A this material also achieved an operational life of 2 million switching cycles. The excess temperature showed non-critical values of, on average, clearly below 100°C .
3. A material with the composition $\text{Ag}88\text{SnO}_2$, $11.4 \text{ In}_2\text{O}_3$, $0.3 \text{ Bi}_2\text{O}_3$, 0.3 was produced by mixing commercial SnO_2 powder, the particle sizes of which pertained to the class of $<1 \mu\text{m}$ in a proportion amounting to 82 %, with Bi_2O_3 powder with particle sizes $<32 \mu\text{m}$ and annealing it in air for 15 h at 1000°C so that an SnO_2 - Bi_2O_3 mixed oxide was formed with particle sizes that pertained to the class of $<1 \mu\text{m}$ in a proportion amounting to only 20 %. This powder was mixed with Ag powder with a particle size of $<63 \mu\text{m}$ and In_2O_3 powder and pressed by a cold isostatic method so as to form a bolt. The bolt was then sintered (750°C , 2 h) and extruded to give it a profile. In a commercial switching device with a rated current of about 50 A the material achieved an operational life of over 2.2 million switching cycles. The excess temperature shows non-critical values of, on average, far below 100°C .
4. A material with the composition $\text{Ag}90\text{SnO}_2$, $8.7 \text{ In}_2\text{O}_3$, $0.5 \text{ Bi}_2\text{O}_3$, 1.6 was produced by annealing commercial SnO_2

powder, the particle sizes of which pertained to the class of $<1 \mu\text{m}$ in a proportion amounting to 82 %, for 60 h at 1000°C so that the SnO_2 powder exhibited a particle size pertaining to the class of $<1 \mu\text{m}$ in a proportion amounting to only 5 %. This powder was processed further as described in Example 1. The material, the composition of which does not lie within the range in accordance with the invention, could only be processed with difficulty, and the switching life was below the values of the material according to the invention.



Sintering material on the basis of silver-tin oxide for electrical contacts and process for its manufacture

5 Claims:

1. Sintering material manufactured by powder metallurgy on the basis of silver-tin oxide with added amounts of indium oxide and bismuth oxide for electrical contacts, characterised in that it consists of 3.2 to 19.9 wt-% tin oxide, 0.05 to 0.4 wt-%, in each case, of indium oxide and bismuth oxide, the remainder being silver.
2. Process for the manufacture of sintering materials according to Claim 1, by mixing of the powders, cold isostatic pressing of the powder mixture, sintering at temperatures from 500 to 940°C and extrusion so as to form wires or profiles, characterised in that prior to the mixing with the silver powder and the other oxide powders more than 60 wt-% of the tin-oxide powder exhibits a particle size of more than 1 μm .
3. Process according to Claim 2, characterised in that the bismuth-oxide powder is caused to react thermally with the tin-oxide powder so as to form $\text{Bi}_2\text{Sn}_2\text{O}_7$, mixed-oxide powder which exhibits a particle size of more than 1 μm in a proportion amounting to more than 60 wt-% and this mixed-oxide powder is mixed with the silver powder and the indium-oxide powder.
4. Process according to Claim 2, characterised in that commercial tin oxide is annealed at temperatures from



700 to 1400°C until more than 60 wt-% of the powder exhibits a particle size of more than 1 μm .

5. Process according to Claim 3,
5 characterised in that
commercial tin oxide is annealed together with
commercial bismuth oxide at temperatures from 700 to
1400°C until more than 60 wt-% of the mixed-oxide
powder exhibits a particle size of more than 1 μm .