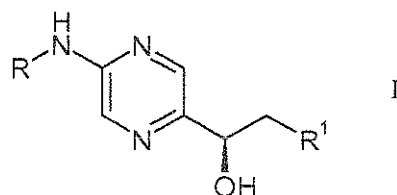


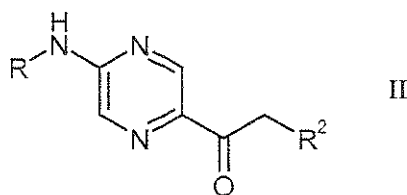
MICROBIAL REDUCTION OF AN ACETOXYKETONE

The present invention relates to a process for the preparation of 2-amino-[5-(1-hydroxy-2-hydroxy or halogen-ethyl)]-pyrazine derivatives of the formula



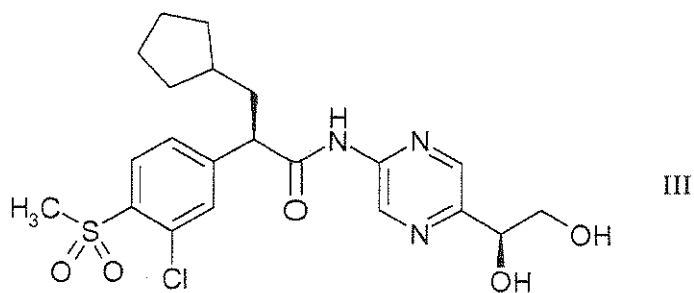
wherein R is lower alkylcarbonyl or an amino protecting group and R¹ is hydroxy or
5 halogen,

by enzymatic hydrolysis and/or enzymatic asymmetric reduction of a ketone of the
formula



wherein R² is lower alkylcarbonyloxy or halogen.

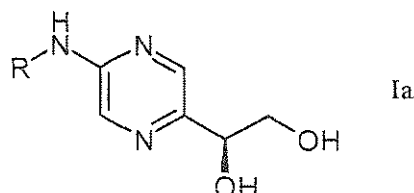
10 This process is especially useful in the preparation of enantiomerically pure (S)-N-[5-(1,2-dihydroxy-ethyl)-pyrazinyl]-2,2-dimethyl-propionamide. This compound is an intermediate for a glucokinase activator, 2-(3-chloro-4-methanesulfonyl-phenyl)-3-cyclopentyl-N-[5-(1,2-dihydroxy-ethyl)-pyrazin-2-yl]-propionamide, of the formula III which is useful for the treatment and/or prophylaxis of type II diabetes:



15

The compound of formula III is disclosed in PCT International Patent Application No. WO 2004/052869 A1.

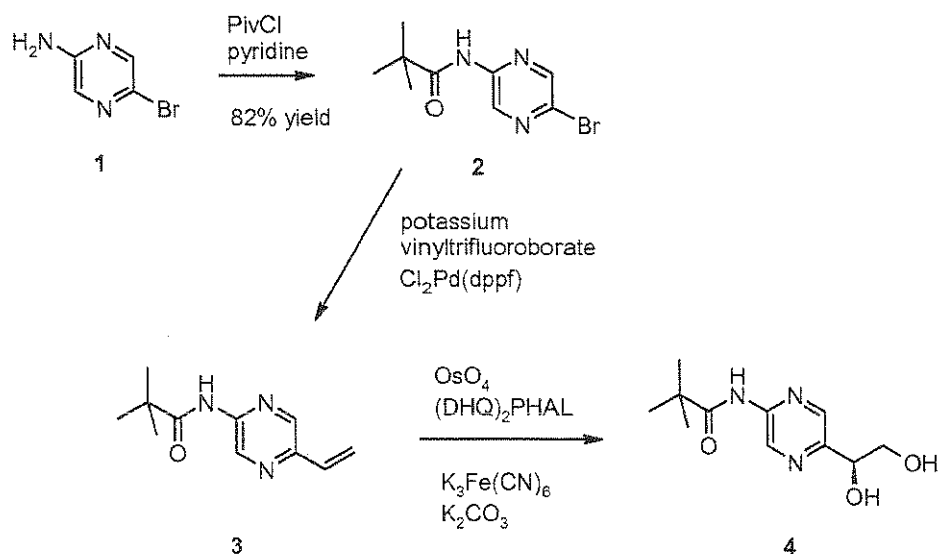
One of the key building blocks used in the synthesis of the compound of formula III are enantiomerically pure 2-amino-[5-(1,2-dihydroxy-ethyl)]-pyrazine derivatives of
5 the formula



wherein R is lower alkylcarbonyl or an amino protecting group. For the preparation of active pharmaceutical ingredients (APIs) it is absolutely necessary to use isomerically pure building blocks and/or highly stereoselective procedures, because side components
10 in APIs may have adverse effects in the treatment of illnesses. Therefore, a high purity is requested for all APIs.

Optically active 1,2-diols are versatile synthetic intermediates and difficult to obtain in enantiomerically pure form. The method described in WO 2004/052869 A1 for preparing (S)-N-[5-(1,2-dihydroxy-ethyl)-pyrazinyl]-2,2-dimethyl-propionamide
15 involved Sharpless oxidation of the corresponding vinyl pyrazine precursor in a reaction comprising osmium tetroxide (see scheme 1). This reaction is not possible on multi-kg scale due to the toxicity of the osmium tetroxide catalyst. Thus, the problem to be solved was to find a suitable process alternative which is free of toxic reagents and can be carried out on large technical scale.

Scheme 1



There are few literature examples reporting microbial hydrolysis / reduction of alkoxy ketones to the corresponding diols in a single step. G. Egri et al., *Tetrahedron* 5 *Asymmetry* 1998, 9, 271-283, describe the biotransformation of a series of 1-acetoxy-3-aryloxypropan-2-ones by bakers yeast. Out of 13 ketones tested only two were transformed directly to the diol with no formation of the intermediate monoacetate. In most cases a mixture of monoacetate and diol was seen which is undesirable for the process for the preparation of compounds of formula I. In addition these reactions were 10 only carried out on the 0.5 g scale at a substrate concentration of 0.25% w/w; far below that to be used for the manufacture of compounds of formula I.

T. Kometani et al, *J. Bioscience. Bioeng.* 2001, 91, 525-527, describe the preparation of (S)-1,2-propanediol by reduction of 1-acetoxy-2-propanone using bakers yeast. Although conversion to the diol was complete at 1% w/v substrate concentration the ee 15 was 88%. This value could only be improved by suppressing the hydrolysis of the acetoxy ketone.

Z-L Wei et al, *Bioorganic and Medicinal Chemistry* 2000, 8, 1129-1137, describe the preparation of S-diols from the corresponding 2-acetoxy-1-arylethanones but again the selectivities were relatively low and the monoacetates were in most cases present.

20 The ee of the (S)-diol formed by the microbial reduction/hydrolysis reaction is a crucial value as it influences the final yield of API. In case of the API as prepared according to the present invention the subsequent ketalization and crystallization steps led to an increase of the enantiomeric excess to >99%.

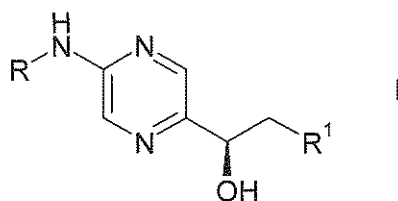
Beside the described microbial biotransformation for the hydrolysis / reduction of an alkoxy ketone one might apply only the necessary isolated biocatalyst – enzymes (e.g. a hydrolase, a ketoreductase, a glucose dehydrogenase) – in a one pot approach. The asymmetric reduction by isolated ketoreductases in combination with the enzymatic cofactor recycling using glucose dehydrogenase and enzymatic hydrolysis of ester moieties are state of the art. S. Kambourakis et al., *Tetrahedron Asymmetry* 2005, 16, 3682-3689, describe the reduction of 2-hydroxy-1-phenyl-ethanone using different ketoreductases. Enzymatic transformations employing two or more enzyme types are frequently successful as described in V. Kren et al., *Angew. Chem. Int. Ed. Engl.* 1995, 34 (8), 893. Therefore, an upstream enzymatic hydrolysis of an acetoxy ketone into a hydroxyl ketone and a downstream reduction of this in situ generated hydroxyl ketone mimics the potential microbial - hydrolysis / reduction – biotransformation. The multi enzymatic reaction using isolated enzymes display some advantages such as i) standard equipment might be used; ii) high reaction rates, iii) no side activities in comparison with whole cell systems, iv) simple reaction control, and v) higher yield in the subsequent ketalization reaction due to the higher ee of the (S)-diol produced.

The terminal position of 2-amino- [5-(acetyl)]-pyrazine derivatives might have different substituents, which are convertible into a hydroxyl function after the asymmetric reduction of the ketone moiety. For a halogen substituents, or more specifically, the chloro substituent, one potential candidate, several biocatalytic asymmetric reductions of different aryl ketones are described in literature, L. Hua et al., *Organic & Biomolecular Chemistry* 2006, 4, 2690-2695 and L. Hua et al. *Tetrahedron Asymmetry* 2005, 16, 3275-3278. The synthesis of enantiomerically pure 1,2-diols starting from aromatic chlorinated alcohols is described via corresponding enantiomerically pure epoxide in T. Ikaraiya et al., *Tetrahedron* 2004, 60, 7411-7417 and subsequent epoxide hydrolysis either via biocatalytic hydrolysis as described in Z. Li et al., *Tetrahedron Asymmetry* 2006, 17, 47-52 or via hydrolysis using metal catalysis as described in G.-J. Kim et al., *Tetrahedron Letters* 2005, 46, 2263-2266.

With the biotransformation processes according to the disclosure an efficient procedure for the preparation of enantiomerically pure 2-amino-[5-(1,2-dihydroxy-ethyl)]-pyrazine derivatives has been found.

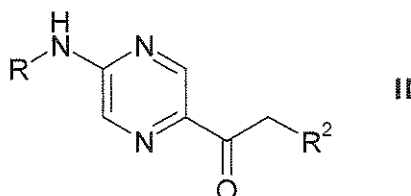
In a first aspect, there is provided a process for the preparation of compounds of the formula

4a



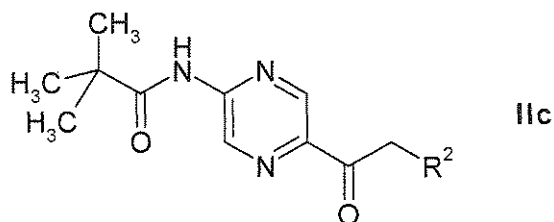
wherein R is lower alkylcarbonyl or an amino protecting group and R¹ is hydroxy or halogen,

- 5 by enzymatic hydrolysis and/or enzymatic asymmetric reduction of a ketone of the formula



wherein R² is lower alkylcarbonyloxy or halogen.

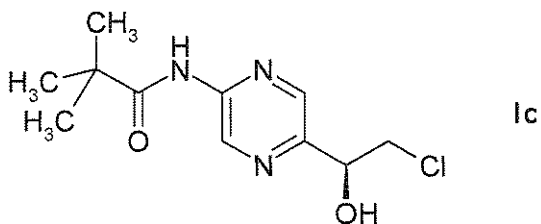
In a second aspect, there is provided compounds of the formula



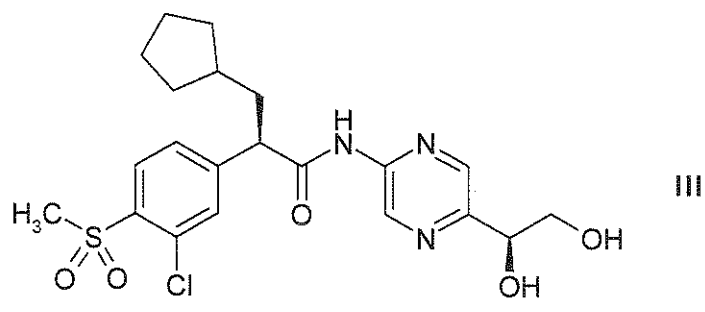
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wherein R² is lower alkylcarbonyloxy or halogen.

In a third aspect, there is provided the compound of the formula

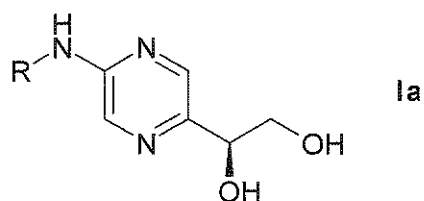


In a fourth aspect, there is provided a process for the preparation of a compound of the formula

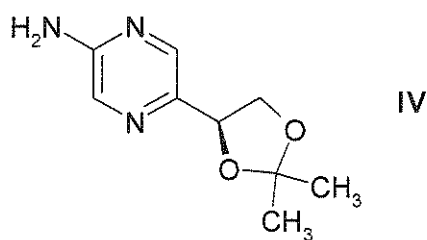


comprising the process as defined above, followed by

5 a) reaction of the diol of formula Ia

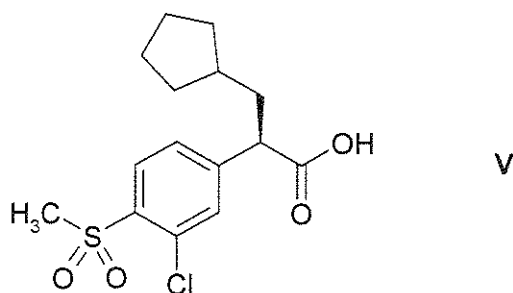


wherein R is lower alkylcarbonyl or an amino protecting group, with 2,2-dimethoxypropane to form an acetal and deprotection of the amine under basic conditions to obtain a compound of formula



10

b) condensation of the amine of formula IV with the carboxylic acid of formula



or an activated derivative thereof to obtain the amide; and

c) hydrolysis of the acetal under acidic conditions.

Unless otherwise indicated, the following definitions are set forth to illustrate and define the meaning and scope of various terms used to describe the invention herein.

- 5 In this specification the term “lower” is used to mean a group consisting of one to six, preferably of one to four carbon atom(s).

The term "halogen" refers to fluoro and chloro, with chloro being preferred.

The term "alkyl", alone or in combination with other groups, refers to a branched or straight-chain monovalent saturated aliphatic hydrocarbon radical of one to twenty carbon atoms, preferably one to sixteen carbon atoms, more preferably one to ten carbon
5 atoms.

The term "lower alkyl" or "C₁-C₆-alkyl", alone or in combination with other groups, refers to a branched or straight-chain monovalent alkyl radical of one to six carbon atoms, preferably one to four carbon atoms. This term is further exemplified by radicals such as methyl, ethyl, n-propyl, isopropyl, n-butyl, s-butyl, isobutyl, tert-butyl,
10 n-pentyl, 3-methylbutyl, n-hexyl, 2-ethylbutyl and the like. Preferable lower alkyl residues are methyl, ethyl and tert-butyl, with tert-butyl being especially preferred.

The term "lower alkylcarbonyl" refers to the group -C(O)-R', wherein R' is a branched or straight-chain monovalent alkyl radical of one to six carbon atoms, preferably one to four carbon atoms. Preferred "lower alkylcarbonyl" or "C₁-C₆-
15 alkylcarbonyl" groups are acetyl, propionyl, butyryl, pivaloyl, pentanoyl and hexanoyl. More preferred are acetyl and pivaloyl (tert-butylcarbonyl), with tert-butylcarbonyl being most preferred.

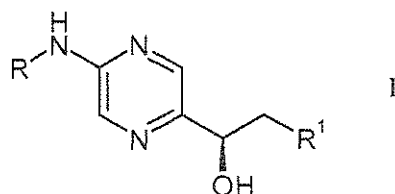
The term "amino protecting group" as used herein refers to a substituent commonly employed to block or protect the amino functionality while reacting other
20 functional groups on the compound. Suitable amino protecting groups are selected from the group consisting of the formyl group, the benzyl group, ester groups such as benzyloxycarbonyl ("Cbz"), 9-fluorenylmethoxycarbonyl ("Fmoc"), tert-butoxycarbonyl ("BOC") and allyloxycarbonyl, and arylsulfonyl derivatives such as para-toluenesulfonyl, benzyloxysulfonyl and phenylsulfonyl. The selection and use (addition and
25 subsequent removal) of amino protecting groups is well known to the skilled in the art. Further examples of groups referred to by the above terms are described by T. W. Greene and P. G. M. Wuts, Protective Groups in Organic Synthesis, 3rd edition, John Wiley and Sons, New York, NY, 1999. A preferred amino protecting group is BOC.

The term "lower alkylcarbonyloxy" refers to the group -O-C(O)-R", wherein R" is
30 a straight-chain monovalent alkyl radical of one to six carbon atoms, preferably one to four carbon atoms. Preferred "lower alkylcarbonyloxy" or "C₁-C₆-alkylcarbonyloxy" groups are acetyloxy, propionyloxy, butyryloxy, pentanoyloxy and hexanoyloxy. Especially preferred "lower alkylcarbonyloxy" is acetyloxy.

The term "enantiomerically pure" refers to a composition that comprises at least 90%, preferably about 95% to 100%, more preferably 98% to 100%, and most preferably 99% to 100% of a single enantiomer of that composition.

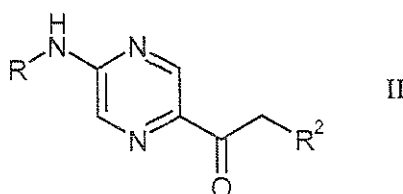
- 5 The term "enantiomeric excess" (abbreviated "ee"), as used herein is defined as $[F(+)-F(-)]$, wherein $F(+)$ refers to the mole or weight fraction of the (+) - enantiomer and $F(-)$ refers to the mole or weight fraction of the (-) -enantiomer. Correspondingly, the the term "percent enantiomeric excess" or "% ee" is defined as $100 \times [F(+)-F(-)]$. Alternatively, the percent enantiomeric excess can be calculated as $100 \times ([R]-[S])/[R]+[S]$.

- 10 The present invention refers to a process for the preparation of compounds of the formula



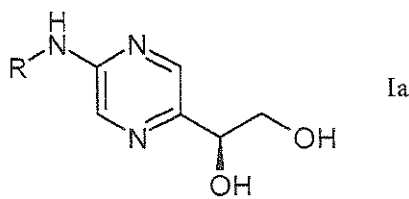
wherein R is lower alkylcarbonyl or an amino protecting group and R^1 is hydroxy or halogen,

- 15 by enzymatic hydrolysis and/or enzymatic asymmetric reduction of a ketone of the formula



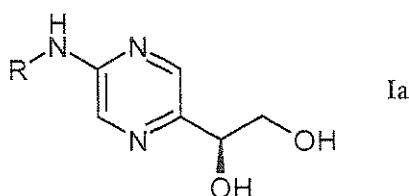
wherein R^2 is lower alkylcarbonyloxy or halogen.

- 20 In a preferred embodiment of the invention, the process is characterized in that R^1 is hydroxy and R^2 is acetyloxy, meaning a compound of formula

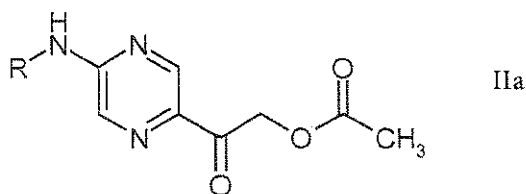


wherein R is lower alkylcarbonyl or an amino protecting group, is obtained.

In one embodiment of the invention, the process is characterized in that the enzymatic hydrolysis and the enzymatic asymmetric reduction is performed together
5 with a yeast of the species *Candida parapsilosis*. Thus, the invention refers to a process for the preparation of 2-amino-[5-(1,2-dihydroxy-ethyl)]-pyrazines of the formula



wherein R is lower alkylcarbonyl or an amino protecting group,
by enzymatic hydrolysis and enzymatic asymmetric reduction with a yeast of the species
10 *Candida parapsilosis* of a keto compound of the formula



By using a strain of the yeast *Candida parapsilosis* the desired product Ia is produced by hydrolysis and asymmetric reduction of the corresponding acetoxy ketone IIa at a technically relevant substrate concentration. The process can be carried out
15 without addition of a hydrolyzing enzyme such as a lipase because the strain can catalyze both hydrolysis and asymmetric reduction.

In detail, the invention relates to a scalable biocatalytic process comprising hydrolysis and asymmetric microbial reduction of the compound of formula IIa using the yeast *Candida parapsilosis* to obtain an enantiomerically pure (S)-diol of the formula
20 Ia, comprising the steps

- a) Growing a culture of *Candida parapsilosis* at 27 to 30 °C for 1 to 2 days in flasks or fermenters containing a rich media comprising; yeast extract (1% w/v), soytone (1% w/v), yeast nitrogen base (0.67% w/v) and glucose (2% w/v);
- b) Adding of NH_4OH 16 to 20 h after inoculation in order that the pH is maintained in the range of 6.5 to 7.0 and feeding of ethanol equivalent to 3 to 5% (v/v) per 24h in order to provide reducing equivalents for growth and for the asymmetric reduction;
- c) After a further 2 to 4 hours adding of 175 g of the acetoxy ketone substrate of formula IIa to the fermentation broth as a suspension in 875ml water to give a final concentration of 1 to 5% (w/v);
- d) Hydrolysis and reduction of the acetoxy ketone of formula IIa to the corresponding (S)-diol within 2 to 5 days;
- e) Isolation of the (S)-diol by separation of the biomass (centrifugation) followed by extraction of the (S)-diol with ethyl acetate (3 times with 2 volume equivalents) and concentration.

The reaction must also proceed to completion, i.e. all the substrate must be converted, at a substrate concentration of 5 % (w/v), which is considerably higher than the concentration quoted in the literature examples.

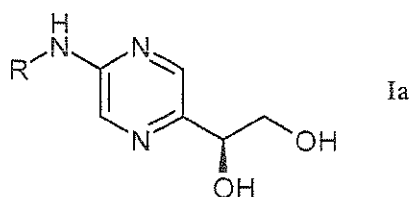
By biotransformation of the acetoxy ketone of formula IIa with *C. parapsilosis* enantiomeric pure (S)-diol with an ee in the range of 91.4% to 95.6% is obtained.

As used herein, *Candida parapsilosis* is a strain isolated at Roche and deposited under the Budapest treaty on March 9, 2007, at DSMZ (Deutsche Sammlung von Mikroorganismen und Zellkulturen GmbH, Inhoffenstrasse 7B, 38124 Braunschweig, Germany) under accession number DSM 19155. Several other strains of *C. parapsilosis* also catalyzed the described biotransformation resulting in (S)-diol with ee of 92 to 95% indicating that any *C. parapsilosis* strain can potentially be used. In addition, strains of the yeasts *Candida kefyr*, *Kluyveromyces marxianus* and the fungus *Calonectria rigidiuscula* can be used.

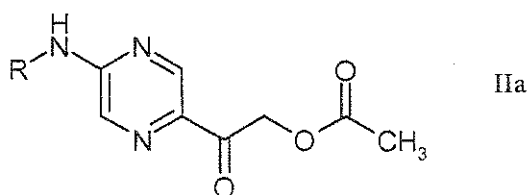
In a preferred embodiment, the invention relates to a scalable biocatalytic process comprising asymmetric microbial reduction of acetic acid 2-[5-(2,2-dimethyl-propionylamino)-pyrazin-2-yl]-2-oxo-ethyl ester (the compound of formula IIa, wherein R is tert-butylcarbonyl) and hydrolysis using the yeast *Candida parapsilosis* to obtain

enantiomerically pure (S)-N-[5-(1,2-dihydroxy-ethyl)-pyrazinyl]-2,2-dimethyl-propionamide.

In a further embodiment of the invention, the process is characterized in that the enzymatic hydrolysis is performed by means of a hydrolase (EC 3.1.1) selected from the group consisting of an esterase, a protease or a lipase and that the subsequent enzymatic asymmetric reduction is performed by means of one or more oxidoreductases (EC 1.1.1). Thus, the invention also refers to a process for the preparation of 2-amino-[5-(1,2-



wherein R is lower alkylcarbonyl or an amino protecting group, by enzymatic hydrolysis by means of an enzyme selected from the group consisting of a hydrolase, an esterase, a protease or a lipase and enzymatic asymmetric reduction by means of one or more oxidoreductases of a keto compound of the formula



Preferably, the enzymatic hydrolysis is performed by means of a lipase. More preferably, the lipase is obtained from *Candida Antarctica*, *Alcaligenes sp.* or *Burkholderia cepacia*.

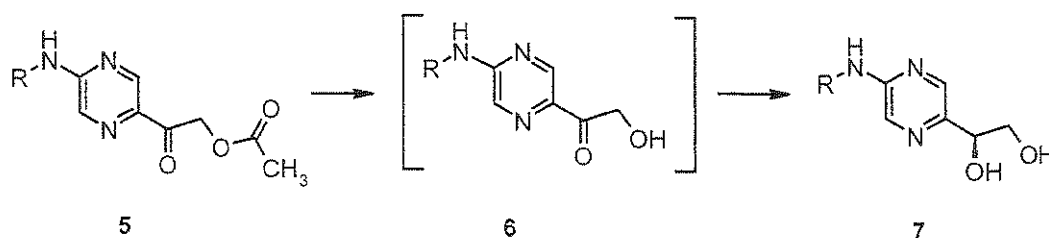
In a preferred embodiment, a ketoreductase or an alcohol dehydrogenase is used as oxidoreductase in the enzymatic asymmetric reduction.

The multi enzymatic biotransformation is transformed as one pot reaction. The hydrolysis – deacetylation - is carried out by contacting a hydrolase with the acetoxy ketone of formula IIa suspended in a biphasic reaction media. The reduction is carried out by contacting an oxidoreductase with the in situ formed α -hydroxy ketone. Due to the low stability of the in situ formed α -hydroxy ketone the applied reactivity of the reductases has to be in excess of the applied reactivity of the hydrolase. The required

reducing equivalents are applied in catalytic amounts and are recycled in situ. The desired product of formula Ia is produced at technical reaction conditions.

In a preferred embodiment, the invention relates to a scalable biocatalytic process comprising hydrolysis of acetic acid 2-[5-(2,2-dimethyl-propionylamino)-pyrazin-2-yl]-2-oxo-ethyl ester (the compound of formula IIa, wherein R is tert-butylcarbonyl) by means of an enzyme selected from the group consisting of a hydrolase, an esterase, a protease or a lipase and enzymatic asymmetric reduction by means of one or more oxidoreductases to obtain enantiomerically pure (S)-N-[5-(1,2-dihydroxy-ethyl)-pyrazinyl]-2,2-dimethyl-propionamide.

Scheme 2



The first step, the *in situ* generation of the hydroxyl ketone (6) by deacetylation of the acetoxy ketone (5) is carried out by hydrolases, esterases, proteases or lipases, preferably by lipases; even more preferred by a lipase from *Candida Antarctica* [e.g. CALB L (Novozyme)], by a lipase from *Alcaligenes sp.* [e.g. QLM (Meito Sangyo)], by a lipase from *Burkholderia cepacia* (lipase PS) and its mutant lipase AH. The subsequent asymmetric reduction is carried out by oxidoreductases, preferably by ketoreductases or alcohol dehydrogenases, more preferred by the ketoreductases KRED 101, 107, 111, 112, 113 and 114, A1F, B1D and B1E [BioCatalytics]. The required reducing equivalents might be recycled in situ by all state of the art methods; preferably by enzymes; more preferred by glucose dehydrogenase GDH 102 [BioCatalytics].

Suitable buffers are the conventional buffers commonly used in biochemistry in the range of pH 5 to 8, preferably of pH 6 to 7. In the course of the reaction the pH of the reaction mixture is kept constant at the selected value by the addition of a base, preferentially NaOH or KOH-solution. One equivalent is required to neutralize the formed acetic acid and a further equivalent is needed to neutralize the formed gluconic acid.

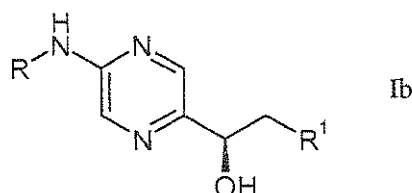
In the case of the enzyme combination KRED 101, lipase AH and GDH 102 the use of 2-(4-morpholino)-ethanesulfonic acid buffer (e.g. pH 6.25) in the presence of a

non-polar organic solvent such as n-heptane or tert-butyl methyl ether (TBME) (e.g. 20% v/v) and D-glucose (e.g. 0.5M) positively influences overall reactivity.

The reaction temperature may be in a range of 25 to 45 °C, preferably 30 to 40 °C. The substrate concentration may range from 1 – 20% (w/w), preferably 5% (w/w).

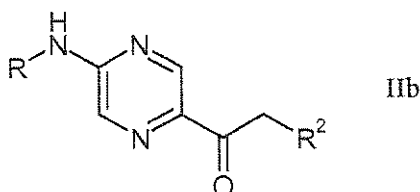
5 The low stability of the *in situ* formed hydroxy ketone (6) requires a catalytic reducing activity in excess of the deacetylation activity. The *in situ* concentration of the hydroxy ketone (6) has to be high enough enabling high turnover rates for its asymmetric reduction. The ketoreductases display a significantly lower activity towards the direct reduction of the acetoxy ketone (5) in combination with a significantly lower
10 enantiomeric purity for the generated acetoxy alcohol – a potential intermediate towards the diol (7). The process conditions have to suppress the direct reduction of the acetoxy ketone (5) or to enhance the high turnover rates of the asymmetric reduction of the *in situ* formed hydroxy ketone (6) by triggering its *in situ* concentration in order to maintain the enantiomerically pure (S)-diol (7).

15 In another embodiment, the invention refers to a process for the preparation of 2-amido-[5-(1-hydroxy-2-halo-ethyl)]-pyrazines of the formula



wherein R is lower alkylcarbonyl or an amino protecting group and R¹ is halogen,

by enzymatic asymmetric reduction of a ketone of the formula



20

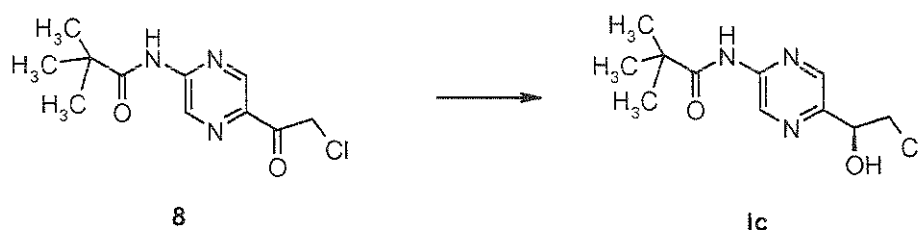
wherein R² is halogen.

Preferably, R¹ and R² are chlorine.

Preferably, the enzymatic asymmetric reduction is performed by means of one or more oxidoreductases. More preferably, a ketoreductase or an alcohol dehydrogenase is used as oxidoreductase.

In a preferred embodiment, the invention relates to the enzymatic reduction of N-[5-(2-chloro-acetyl)-pyrazinyl]-2,2-dimethyl-propionamide (4) to (S)-N-[5-(2-chloro-1-hydroxy-ethyl)-pyrazin-2-yl]-2,2-dimethyl-propionamide (5).

Scheme 3



The asymmetric reduction is carried out by contacting an oxidoreductase with the chloro ketone (8). The required reducing equivalents are applied in catalytic amounts and are recycled *in situ*.

Preferred oxidoreductases are ketoreductases or alcohol dehydrogenases, more preferred are the ketoreductases KRED 101, KRED 111, KRED 112, KRED 113 and KRED 114 [BioCatalytics]. The required reducing equivalents may be recycled *in situ* by conventional methods; preferably by enzymes; more preferred by glucose dehydrogenase GDH 102 [BioCatalytics].

Suitable buffers are the conventional buffers commonly used in biochemistry in the range of pH 5 to 8, preferably in the range of 6 to 7. In the course of the reaction the pH of the reaction mixture is kept constant at the selected value by the addition of a base, preferentially NaOH or KOH-solution. One equivalent is required to neutralize the formed gluconic acid.

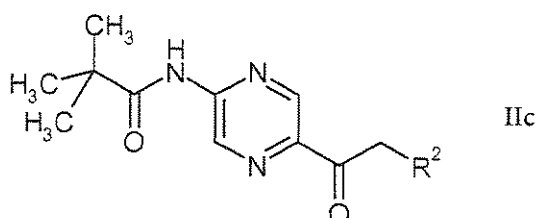
In the case of the enzyme combination KRED 101 and GDH 102 the use of potassium phosphate buffer (e.g. pH 6.5) in the presence of D-glucose (e.g. 0.06M) at higher substrate concentration and addition of higher D-glucose concentration influences overall reactivity positively.

The reaction temperature may be in a range of 25 to 45 °C, preferably in a range of 30 to 35 °C. The substrate concentration may range from 0.1 to 10% (w/w), preferably 5% (w/w), more preferably 0.5% (w/w).

Advantageously, the asymmetric reduction of chloro ketone (8) towards the enantiomerically pure (S)-N-[5-(2-chloro-1-hydroxy-ethyl)-pyrazin-2-yl]-2,2-dimethyl-propionamide (Ic) did not need a third hydrolyzing enzyme (e.g. lipase) and did not produce *in situ* a potential unstable intermediate. The absolute configuration was
5 determined by crystal structure. Subsequently, the enantiomerically chlorinated alcohol (Ic) has to be converted into the desired enantiomerically pure (S)-diol (7) by nucleophilic substitution of chloro against a hydroxyl group.

As already described above, R is preferably tert-butylcarbonyl, i.e. the processes as defined herein before are preferably carried out starting from compounds of formula II,
10 wherein R is tert-butylcarbonyl.

Thus, in another embodiment, the present invention relates to new compounds of the formula



wherein R² is lower alkylcarbonyloxy or halogen.

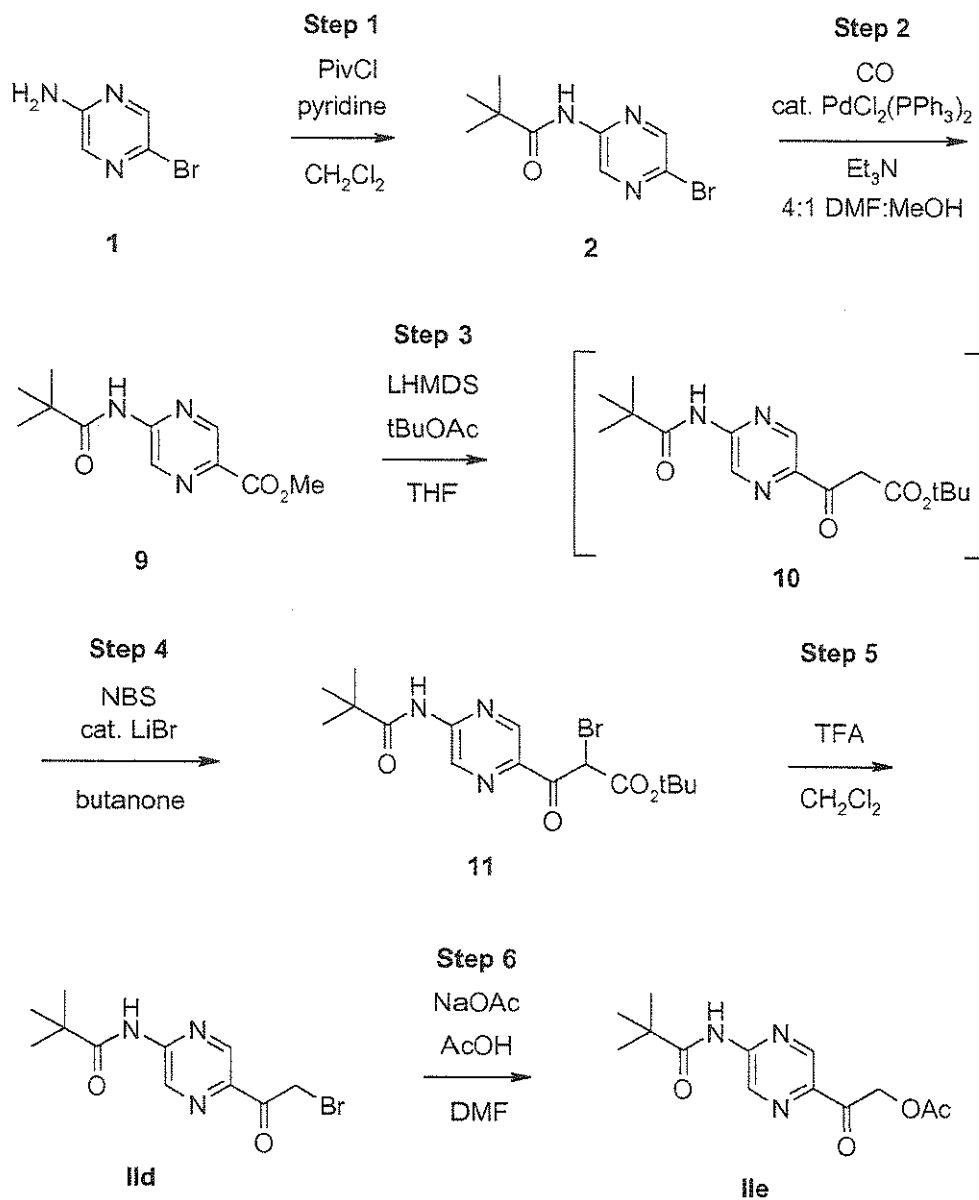
15 The preparation of compounds of formula IIc can be performed according to schemes 4 and 5 below.

In Step 1, the amino group of 2-amino-5-bromopyrazine (1) was protected with trimethylacetyl chloride (pivaloyl chloride; PivCl) in dichloromethane to give amide 7 in 90% yield. Palladium catalyzed carbomethoxylation of amide 2 (step 2) was carried out
20 in a mixed solvent of dimethylformamide:methanol 4:1 under 500 psi of carbon monoxide in a Parr reactor to give methyl ester 9 in 84% yield. In Step 3, Claisen condensation of methyl ester 9 with the enolate generated from *tert*-butyl acetate by treatment with lithium bis(trimethylsilyl)amide (LHMDS) gave keto ester 10. After extractive workup and solvent exchange, the resulting ethanol solution of 10 was treated
25 with *N*-bromosuccinimide (NBS) in the presence of a catalytic amount of lithium bromide to give bromide 11 in 95% overall yield from 9. Treatment of 11 with trifluoroacetic acid (TFA) in dichloromethane afforded α -bromo ketone IIId in 97% yield (Step 5), the decarboxylation was complete after stirring at room temperature for 40h. The α -bromo ketone IIId was converted to acetoxy ketone IIe by substitution reaction

with sodium acetate in DMF at room temperature (step 6). After crystallization from ethyl acetate/heptane, the acetoxy ketone **Ile** was obtained in 90% yield. Subsequently, it was found that **Ile** can be precipitated directly from the reaction mixture by the addition of water.

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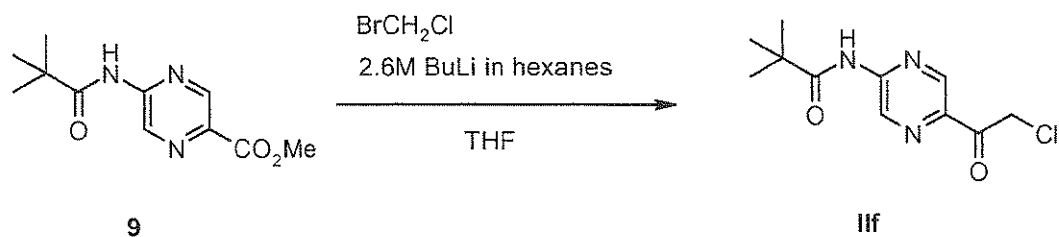
Scheme 4



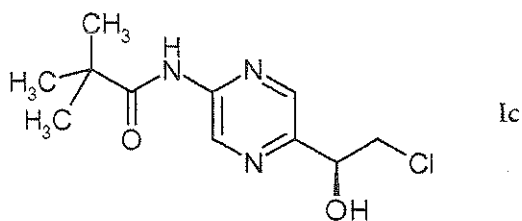
N-[5-(2-Chloro-acetyl)-pyrazin-2-yl]-2,2-dimethyl-propionamide (**IIf**) can be directly obtained from the methyl ester **9** by reaction with bromochloromethane and activation with butyllithium (see scheme 5).

- 15 -

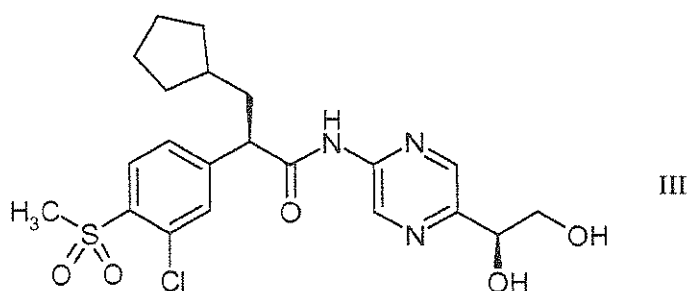
Scheme 5



In another embodiment, the invention relates to a new compound of the formula

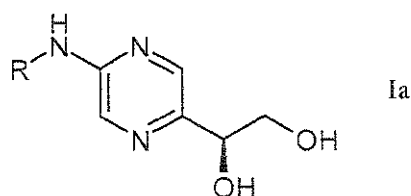


- 5 In a further embodiment, the invention provides a process for the preparation of a compound of the formula



comprising the process according to claims 1 to 11, followed by

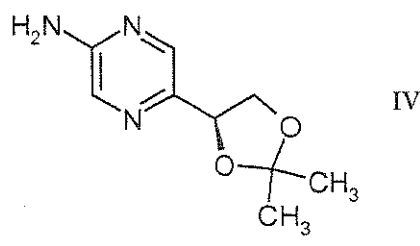
- a) reaction of the diol of formula Ia



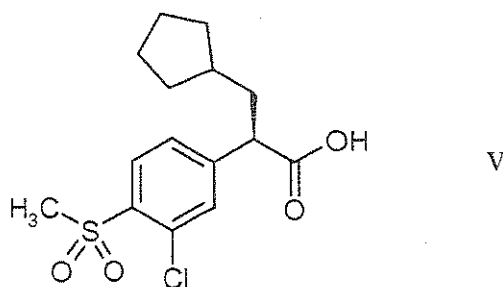
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wherein R is lower alkylcarbonyl or an amino protecting group, with 2,2-dimethoxypropane to form an acetal and deprotection of the amine under basic conditions to obtain a compound of formula

- 16 -



b) condensation of the amine of formula IV with the carboxylic acid of formula



or an activated derivative thereof to obtain the amide; and

5 c) hydrolysis of the acetal under acidic conditions.

2R-(3-Chloro-4-methanesulfonyl-phenyl)-3-cyclopentyl-N-[5-(1S,2-dihydroxy-ethyl)-pyrazin-2-yl]-propionamide (the compound of formula III) was found to be a potent glucokinase activator. Compounds that activate GK, and thereby increase the sensitivity of the GK sensor system, are useful in the treatment of the hyperglycemia
 10 characteristic of all type II diabetes. Glucokinase activators will increase the flux of glucose metabolism in β -cells and hepatocytes, which will be coupled to increased insulin secretion. Such agents are thus useful for treating type II diabetes and other metabolic disorders.

In step a), the 1,2-diol group is protected in form of a cyclic acetal. Reaction of the
 15 1,2-diol with dimethoxypropane provides a 1,3-dioxolane. Preferably, the reaction is carried out in the presence of an acid catalyst such as p-toluenesulfonic acid (PTSA) or camphorsulfonic acid (CSA). The acetals are stable to most reaction conditions except protic acids such as aqueous acetic acid, aqueous trifluoroacetic acid and hydrochloric acid and Lewis acids. Thus the acetal will not be attacked by a base such as potassium
 20 carbonate which is used for the subsequent deprotection of the amine.

For the condensation in step b), an activated derivative of the carboxylic acid of formula V may be employed, for example a protected ester or acid chloride thereof which may be prepared by methods known to those skilled in the art. Preferably, an acid

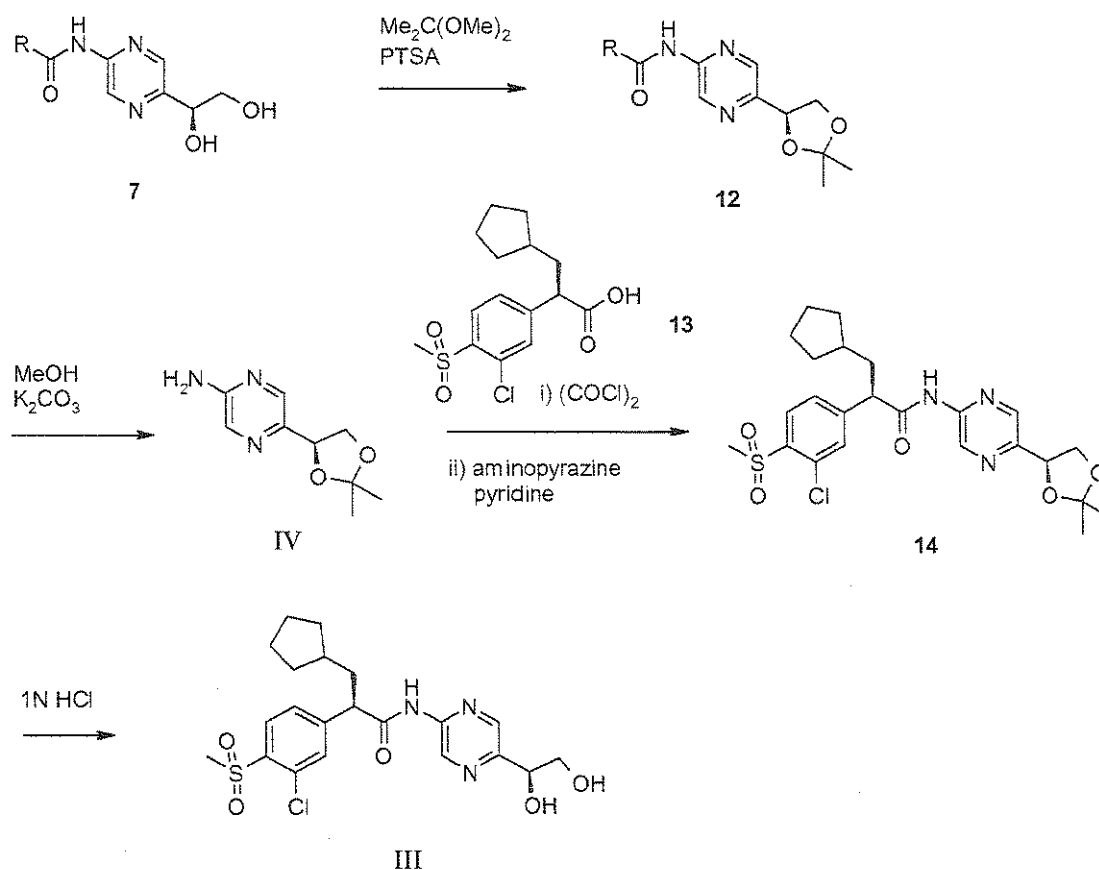
chloride of the acid of formula V may be used and the coupling will be carried out in the presence of a base such as pyridine or aminopyrazine. The acid chloride can be prepared by reaction of the compound of formula V with oxalyl chloride or thionyl chloride in a suitable solvent such as dichloromethane.

- 5 In step c), the acetal protecting group is cleaved under acidic conditions, for example by using hydrochloric acid to obtain the 1,2-diol of formula III.

In scheme 6 below, the process for the preparation of the compound of formula III starting from a compound of formula Ia as prepared by the enzymatic processes as defined herein before is illustrated.

10

Scheme 6



The following examples shall illustrate the invention without limiting it.

Examples

Example 1

Preparation of acetic acid 2-[5-(2,2-dimethylpropionylamino)-pyrazin-2-yl]-2-oxo-ethyl ester

5 Step 1: Preparation of N-(5-bromo-pyrazin-2-yl)-2,2-dimethylpropionamide (2)

A 3-necked 1L round bottomed flask equipped with a magnetic stirrer, thermometer, condenser and nitrogen inlet/outlet was charged with 50.00 g (287.4 mmol) of 2-amino-5-bromopyrazine (1), 218 mL of dichloromethane and 30.50 mL (377.1 mmol) of pyridine. Then, 39.30 mL (319.1 mmol) of trimethylacetyl chloride
10 (PivCl) was added dropwise over 5 min. An exotherm ensued that raised the temperature of the mixture from 22 °C to 44 °C. After stirring at *ca.* 40 °C for 2 h, HPLC analysis indicated complete reaction. The reaction mixture was diluted with 200 mL of ethanol, then concentrated by distillation at atmospheric pressure. After 240 mL of distillate had collected and the temperature of the mixture reached 68 °C, 100 mL of water was added
15 slowly, while maintaining the temperature of the mixture at *ca.* 68 °C. After the addition was complete, the resulting suspension was allowed to cool to room temperature and stirred overnight. The solid was collected by filtration, washed with 100 mL of ethanol:water 1:1 and dried by suction to give 67.08 g (90.4% yield) of the title compound as a light beige solid; 98.21% pure as determined by HPLC analysis (HPLC
20 column Zorbax Eclipse XDB-C8, 4.6 x 50 mm, 1.8 µm, eluent 5-100% acetonitrile/water + 01.% TFA over 5 min at 1 mL/min, detection at UV 250 nm, retention time 4.22 min).

Step 2: Preparation of 5-(2,2-dimethyl-propionylamino)-pyrazine-2carboxylic acid methyl ester (9)

A 300 mL Parr reactor was charged with 15.00 g (58.11 mmol) of the compound
25 prepared in step 1, 16.80 mL (414.8 mmol) of methanol, 67.20 mL of dimethylformamide (DMF), 61.20 mg (0.0872 mmol) of bis(triphenylphosphine) palladium dichloride and 8.900 mL (63.92 mmol) of triethylamine. The reactor was purged twice with nitrogen (by pressurizing it, followed by venting it to atmospheric pressure), then twice with carbon monoxide. The mixture was heated to 92 °C, with
30 stirring at 500 rpm, then pressurized with CO to 500 psi for 18 h. HPLC analysis indicated complete reaction. After cooling to 65 °C, the reactor was depressurized and the contents were transferred to a 500 mL round bottom flask. The reactor was rinsed with 30 mL of DMF and the rinse was also transferred to the flask. Then, 80 mL of water was

added. After cooling to room temperature, the resulting solid was collected by filtration, washed with 50 mL of DMF:water 1:1 and 50 mL of water and dried by suction to give 11.56 g (83.8% yield) of the title compound as a light beige solid; 100% pure as determined by HPLC analysis (same conditions as in step 1, retention time 3.46 min).

5 Step 3: Preparation of 3-[5-(2,2-dimethyl-propionylamino)-pyrazin-2-yl]-3-oxo-propionic acid-*tert*-butyl ester (10)

A 3-necked 1-L round bottomed flask equipped with a magnetic stirrer, addition funnel, thermocouple probe and nitrogen inlet/outlet was charged with 25.00 mL (185.5 mmol) of *tert*-butyl acetate, 20.00 g (84.30 mmol) of the compound prepared in step 2
10 and 20 mL of THF. After cooling to -20 °C, a solution of 261.4 mL (261.4 mmol) of 1.0M lithium bis(trimethylsilyl)amide (LHMDS) in THF was added dropwise, while maintaining the temperature of the reaction mixture between -20 °C and 0 °C. The resulting red solution was stirred at -20 °C for 40 min. HPLC analysis indicated complete reaction. The mixture was allowed to warm to 0 °C, then quenched by the addition of
15 pre-cooled 200 mL (260.3 mmol) of 25wt% citric acid solution. The organic layer was separated, washed with 2 × 200 mL of saturated sodium chloride solution and concentrated at 30 °C/60 mmHg to a volume of *ca.* 50 mL. The concentrated solution was diluted with 200 mL of butanone and again concentrated at 30 °C/60 mmHg to *ca.* 50 mL. The concentrate was again diluted with 200 mL of butanone and concentrated at
20 30 °C/60 mmHg to a volume of *ca.* 100 mL. NMR analysis indicated the absence of THF. The resulting butanone solution of the title compound was used directly in the next step.

Step 4: Preparation of 2-bromo-3-[5-(2,2-dimethyl-propionylamino)-pyrazin-2-yl]-3-oxo-propionic acid *tert*-butyl ester (11)

A 1-L round bottomed flask equipped with a magnetic stirrer was charged with
25 73.00 mg (0.841 mmol) of lithium bromide and the butanone solution obtained in step 3 (*ca.* 100 mL), which *theoretically* contained 27.09 g (84.30 mmol) of 3-[5-(2,2-dimethyl-propionylamino)-pyrazin-2-yl]-3-oxo-propionic acid-*tert*-butyl ester and *ca.* 73 mL of butanone. To the resulting mixture was added a total of 15.16 g (85.17 mmol) of *N*-bromosuccinimide portionwise with careful reaction monitoring by HPLC. After stirring
30 at room temperature for 1 h, HPLC analysis indicated complete reaction. The reaction mixture was concentrated at 25 °C/25 mmHg to a volume of *ca.* 70 mL, then diluted with 130 mL of ethyl acetate and washed with 3 × 100 mL L of water. After concentration at 35 °C/60 mmHg to a volume of *ca.* 90 mL, the resulting suspension was diluted with 200 mL of heptane, and re-concentrated to a volume of *ca.* 90 mL. Then, 200 mL of heptane
35 was added, and the suspension was again concentrated to a volume of *ca.* 150 mL. The

solid was then collected by filtration, washed with 2×50 mL of heptane and dried by suction to give 32.16 g of the title compound as a light yellow solid; 98.7% pure as determined by HPLC analysis (HPLC column Zorbax XDB-C8, 3×100 mm, $3.5 \mu\text{m}$, eluent 20-100% acetonitrile/water + 0.1% TFA over 10 min at 0.5 mL/min, detection at UV 254 nm, retention time 9.52 min).

Step 5: Preparation of N-[5-(2-bromo-acetyl)-pyrazin-2-yl]-2,2-dimethyl-propionamide (IIId)

A 500 mL round bottomed flask equipped with a magnetic stirrer and nitrogen inlet/outlet was charged with 32.10 g (80.19 mmol) of the compound prepared in step 4, 90 mL of dichloromethane and 56.20 mL (756.6 mmol) of trifluoroacetic acid and the reaction mixture was stirred at room temperature for 40 h. HPLC analysis indicated complete reaction. The reaction mixture was concentrated at $30^\circ\text{C}/30$ mmHg to a volume of *ca.* 40 mL, diluted with 200 mL of toluene, and concentrated to a volume of *ca.* 50 mL. The resulting slurry was diluted with 100 mL of toluene and again concentrated to a volume of *ca.* 50 mL. After diluting with 100 mL of heptane, the solid was collected by filtration and dried by suction to give 23.30 g (96.8% yield) of the title compound as a yellow solid; 99.15% pure as determined by HPLC analysis (same conditions as in step 4, retention time 7.76 min).

Step 6: Preparation of acetic acid 2-[5-(2,2-dimethylpropionylamino)-pyrazin-2-yl]-2-oxo-ethyl ester (IIe)

A 500 mL round bottomed flask equipped with a magnetic stirrer, addition funnel, thermocouple probe and nitrogen inlet/outlet was charged with 4.40 mL (76.86 mmol) of acetic acid, 140 mL of DMF and 7.000 g (85.33 mmol) of sodium acetate. Then, 23.30 g (77.62 mmol) of the compound obtained in step 5 was added portionwise over 45 min. After stirring at room temperature for an additional 1 h, HPLC analysis indicated complete reaction. The reaction mixture was diluted with 350 mL of ethyl acetate and 100 mL of saturated sodium bicarbonate was added with stirring. The organic layer was separated, washed with 3×100 mL of water and concentrated at $30^\circ\text{C}/60$ mmHg to a volume of *ca.* 60 mL. The resulting slurry was diluted with 200 mL of heptane, concentrated at $30^\circ\text{C}/60$ mmHg to a volume of *ca.* 150 mL, and stirred at 50°C for 30 min. After cooling to room temperature, the solid was collected by filtration, washed with 40 mL of 10% ethyl acetate in heptane and dried by suction, then under reduced pressure (house vacuum) for 24 h, to give 19.52 g (90.0% yield) of the title compound as

an off-white solid; 98.81% pure as determined by HPLC analysis (same conditions as in step 4, retention time 6.78 min).

¹H-NMR (DMSO-d₆): 10.71 (s, 1H), 9.37 (d, 1H), 8.89 (d, 1H), 5.48 (s, 2H), 2.14 (s, 3H), 1.25 ppm (s, 9H).

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Example 2

Preparation of N-[5-(2-chloro-acetyl)-pyrazin-2-yl]-2,2-dimethyl-propionamide (IIIf)

A 3-necked 1-L round bottomed flask equipped with a mechanical stirrer, addition funnel, thermocouple probe and nitrogen inlet/outlet was charged with 24.10 g (102 mmol) 5-(2,2-dimethyl-propionylamino)-pyrazine-2-carboxylic acid methyl ester (9) as prepared in example 1, step 2, 300 mL of THF, and 24.0 mL (369 mmol) of bromochloromethane was added. After cooling to -78 °C using a -90 °C heptane-liquid nitrogen cooling bath, a solution of 100.0 mL (260 mmol) of 2.6M butyllithium in hexanes was added dropwise, while maintaining the temperature of the reaction mixture at -77 ± 2 °C. Then another 15.0 mL (231 mmol) of bromochloromethane was added, followed by the dropwise addition of another 55.0 mL (143 mmol) of 2.6M butyllithium in hexanes, maintaining the temperature of the reaction mixture at -77 ± 2 °C. HPLC analysis indicated complete reaction. The cold mixture was poured slowly into 300 mL (300 mmol) of 1N hydrochloric acid and stirred to warm up to ambient temperature. The mixture was the partially concentrated under vacuum and the precipitated solids were isolated by filtration, washed with water and dried by suction to give 19.42 g of crude product as a light orange solid, 91.5% pure as determined by HPLC analysis. This crude product, 19.0 g, was slurried in 100 mL of ethyl acetate and the resulting suspension was diluted with 50 mL of heptane. The solid was then collected by filtration, washed with 2 × 50 mL of heptane-ethyl acetate 1:1 and dried by suction to give 13.5 g (53% yield) of the chloroketone as an off-white solid, >99% pure as determined by HPLC analysis.

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¹H-NMR (DMSO-d₆): 10.71 (s, 1H), 9.35 (d, 1H), 8.94 (d, 1H), 5.20 (s, 2H), 1.25 ppm (s, 9H).

Example 3

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Fermentation and Biotransformation

2x 500ml baffled flasks containing 100ml of 508S medium comprising per L de-ionized water: glucose 20 g, yeast extract 10 g and soytone 10 g was inoculated with 1 mL

of a frozen stock of *C. parapsilosis* R 2599. The flask is then incubated at 27 °C for 72 hours on an orbital shaker set to 220 rpm. The contents of the flask are then pooled into a suitable inoculation flask and inoculated into a 7.5L fermenter containing 5 L of YSD medium comprising per L de-ionized water: glucose 20 g, yeast extract 10 g, soytone 10 g and yeast nitrogen base without amino acids 6.7 g and Shell Aseol antifoam 0.3 ml. The fermentation parameters were set as follows: Temperature 27 °C, dissolved oxygen was maintained above 50% by automatic adjustment of aeration and stirring speed, pH was maintained at 6.5 by automatic addition of 25% w/v ammonium hydroxide, ethanol (100%) was fed using a dosimeter at a rate of 4-5% v/v per day. After 20 h cultivation 1.5 L of broth was removed and ethanol feeding started. After a further 2 hours 175.5 g of acetic acid 2-[5-(2,2-dimethylpropionylamino)-pyrazin-2-yl]-2-oxo-ethyl ester (Example 1) was added as a suspension in 875 mL of water to give a final concentration of 5% (w/v). Periodically samples were removed and analyzed by HPLC to determine the titre of (S)-N-[5-(1,2-dihydroxy-ethyl)-pyrazin-yl]-2,2-dimethyl-propionamide (7) and also the enantiomeric purity of this product. When the reaction was judged complete after 68 hours the *C. parapsilosis* was inactivated by heating the broth in situ to 70 °C for 30 minutes.

Table 1

Reaction Time (h)	Diol (g/L)	R-Diol (%)	S-Diol (%)
1	1.0		
3	3.4		
20	17.0		
24	17.0		
26	17.6		
44.5	27.2		
51	32.3		
68	30.7	3.6	96.4

Isolation

The heat inactivated broth as obtained above was used for product isolation. 4.23 L broth was centrifuged on a laboratory centrifuge with swing-out rotors (3500 rpm, 15 min). The opalescent supernatant (3.60 L) was removed. The pellet was re-suspended in 0.8 L water and centrifuged, giving 0.76 L turbid supernatant. The unified aqueous solutions were extracted three times with ethyl acetate (each 9L). At the first extraction a spontaneous phase separation occurred. At the second extraction an emulsion was obtained. The emulsion was broken by mixing in 250 g dicalite speed plus (Acros Organics 123380010) and filtering the mixture in vacuo (Filtrox filter plate AF 50/8427). At the third extraction a fast phase separation was obtained. The obtained organic extracts were pooled and concentrated in vacuo on a laboratory rotavap. The concentrate was mixed with two spoons of dicalite speed plus, filtered and made up to 1.00 L concentrate with ethyl acetate. The concentrate contained 114.4 g (s)-diol (7).

A sample of the concentrate was dried and showed following analytical data: (HPLC): 99.4% by area purity by HPLC (column SupelcoSil ABZ+, 4.6 x 250 mm, 5 μ m, eluent 20-90% acetonitrile/water + 0.1% TFA over 10 min at 1 mL/min, detection at UV 300 nm, retention time 4.26 min), 92.0% ee by chiral HPLC (column Chiralpak AD-H, eluent 20% ethanol/ 80% acetonitrile at 1 mL/min, 40 °C, detection at UV 237 nm, retention time: 18.14 min (R-diol) and 20.86 min (S-diol)).

^1H -NMR (DMSO- d_6): 10.15 (s, 1H), 9.18 (s, 1H), 8.44 (s, 1H), 5.54 (d, 1H), 4.72 (t, 1H), 4.63 (dd, 1H), 3.69 (m, 1H), 3.53 (m, 1H), 1.25 ppm (s, 9H).

MS (Ion Spray): m/z 240.1 (M+H for M 239.1).

Example 4

Large scale multi-enzymatic reaction

50 g of acetic acid 2-[5-(2,2-dimethylpropionylamino)-pyrazin-2-yl]-2-oxo-ethyl ester (178 mmol) was stirred in 150 ml tert-butyl methyl ether (TBME). Subsequently the reaction buffer, 674 mL 20 mM of 2-(4-morpholino)-ethansulfonic acid, and 100.1 g of D-glucose (658 mmol) were added. The temperature was adjusted to 29 °C and the pH to 6.25. The reaction – deacetylation – was started by the addition of 2.01 g lipase AH. Directly afterwards, 40 mg of glucose dehydrogenase GDH 102, 201 mg of ketoreductase KRED 101 and 202 mg of cofactor NADP were added to initiate the asymmetric reduction. The reaction temperature was increased up to 37 °C. The stirred suspension was maintained at pH 6.25 (and 37 °C) by the controlled addition (pH-stat) of 1.0 N

sodium hydroxide solution. After 11.2 h, after a total consumption of 354.1 mL of 1.0 N sodium hydroxide, and after complete conversion, the reaction mixture was stirred for further 10.5 h. For the product extraction 300 g of sodium chloride were added into the reaction mixture and the pH was adjusted to 7.5. Subsequently, the reaction mixture was extracted 5 times with 1L of ethyl acetate. The phase separation occurred spontaneously. The combined organic phases were dried over anhydrous sodium sulfate, evaporated and dried on HV over night. 44.76g (S)-N-[5-(1,2-dihydroxy-ethyl)-pyrazin-yl]-2,2-dimethyl-propionamide (4) (96.4% HPLC purity, [SupelcoSil ABZ+, 250 x 4.6 mm, eluent 20-90% acetonitrile/water + 0.1% TFA over 10 min at 1 mL/min, detection at UV 300 nm, retention time 5.3 min], ee >99.9% [Chiralpak IA, 250 x 4.6 mm, 5 µm, eluent 50% Heptane 50% Ethanol/Methanol 1:1 over 20 min at 1 mL/min, detection at UV 240 nm, retention time enantiomers 9.3 and 10.9 min]) were isolated as a light orange, highly viscous oil.

Example 5

Small scale reductions of acetic acid 2-[5-(2,2-dimethylpropionylamino)-pyrazin-2-yl]-2-oxo-ethyl ester

2 mg of acetic acid 2-[5-(2,2-dimethylpropionylamino)-pyrazin-2-yl]-2-oxo-ethyl ester were dissolved in 20 µl DMSO and added into reaction vials containing 20 µl of 2-propanol, 1.5 ml of 100 mM 2-(4-morpholino)-ethanesulfonic acid, pH 6.0, 3 mg of NADPH and 3 mg of a ketoreductase (see table 1). After 2h the reactions were extracted with 0.5ml ethyl acetate and analyzed via chiral HPLC ([Chiralcel OD-H, 250 x 4.6 mm, Nr.146, eluent 65% heptane 20% heptane + 0.1% TFA 15% *iso*-propanol, 40 °C over 15 min at 1 mL/min, detection at UV 210 nm, retention time enantiomers 6.2 and 7.2 min], results see table 2).

Table 2: Selected analytical results of the formation of the corresponding acetoxy alcohol (acetic acid 2-[5-(2,2-dimethylpropionylamino)-pyrazin-2-yl]-2-hydroxy-ethyl ester)

KRED ketoreductase	Conversion Area%	ee acetoxy alcohol %
101	33	73.7
107	0	n.d.

111	39.6	68.4
112	46.7	70.8
113	28.5	70.4
114	42.7	65.4
A1F	1.0	n.d.
B1D	13.6	79.5
B1E	0	n.d.

Example 6

Small scale multi enzymatic reactions

1 mg of acetic acid 2-[5-(2,2-dimethylpropionylamino)-pyrazin-2-yl]-2-oxo-ethyl ester (IIe) were dissolved in 20 μ l DMSO and added into reaction vials containing 20 μ l of 2-propanol, 1.5 ml of 100mM potassium phosphate, pH 7.2, 3mg of NADPH, 30 μ l of Lipozyme CALB L [Novozyme] and 2 mg of a ketoreductase (see table 2). After 16 h the reactions were extracted with 0.5ml ethyl acetate and analyzed via chiral HPLC (Chiralcel AD-H, Nr.417, eluent 90% ethanol 10% methanol, 40°C over 30 min at 1 mL/min, detection at UV 210 nm, retention time enantiomers 9.2 and 10.2 min], result see table 3).

Table 3: Selected analytical results of the formation of the corresponding (S)-diol (4)

KRED ketoreductase	Conversion Area%	ee diol %
101	97.6	99
107	97.3	>99
111	97.9	99.5
112	97.8	>99
113	97.9	>99
114	97.3	98
A1F	97.5	>99
B1D	97.4	>99
B1E	97.4	>99

Example 7

Enzymatic reduction of N-[5-(2-chloro-acetyl)-pyrazin-2-yl]-2,2-dimethyl-
propionamide (IIf)

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1.5 g of N-[5-(2-chloro-acetyl)-pyrazin-2-yl]-2,2-dimethyl-propionamide (example 2, 5.8 mmol) was placed into a reactor equipped with a pH electrode, a pH controlled dosing pump and a stirrer. Subsequently the reaction buffer, 300 ml of 100mM potassium phosphate buffer and 3.5 g of D-glucose (17.7mmol) were added. The temperature was adjusted to 30 °C and the pH to 6.5. The asymmetric reduction was started by the addition of 25 mg of glucose dehydrogenase GDH 102, 100 mg of ketoreductase KRED 101 and 250 mg of cofactor NADP. The pH was maintained at pH 6.5 (and 30 °C) by the controlled addition (pH-stat) of 1.0 N sodium hydroxide solution. After 46 h, after a total consumption of 5.79 mL of 1.0 N sodium hydroxide, the reaction was clarified by filtration. Subsequently, the product was extracted with 0.4 L of ethyl acetate. The phase separation occurred spontaneously. The organic phase was dried over anhydrous sodium sulfate, evaporated and dried on HV over night. 1.42g (S)-chloro

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alcohol (97.7% HPLC purity, [Suplecosil Abz+, 250*4.6 mm, eluent 35-90% acetonitrile/water + 0.1% TFA, 25°C over 10.9 min at 1 mL/min, detection at UV 300 nm, retention time 5.6 min], ee >99.9% [Chiralcel OD-H, 250*4.6 mm, eluent 85% heptane 15% ethanol + 0.01 M ammonium acetate, 25 °C over 25 min at 0.8 mL/min, detection at UV 302 nm, retention time enantiomers 6.6 and 7.5 min]) was isolated as light yellow crystals.

¹H-NMR (DMSO-d₆): 10.21 (s, 1H), 9.2 (d, 1H), 8.50 (d, 1H), 6.10 (d, 2H), 4.94 ppm (d/tr, 1H), 3.92 (d/d, 2H), 1.25 ppm (s, 9H).

MS (Ion Spray): m/z 257.8 (M+H for M 257.1).

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Example 8

Preparation of 2(R)-(3-chloro-4-methanesulfonyl-phenyl)-3-cyclopentyl-N-[5-(1(S),2-dihydroxy-ethyl)-pyrazin-2-yl]-propionamide (III)

Step 1: Preparation of N-[5-((S)-2,2-dimethyl-[1,3]dioxolan-4-yl)-pyrazin-2-yl]-2,2-dimethyl-propionamide (12)

15 A solution of N-[5-(1(S),2-dihydroxy-ethyl)-pyrazin-2-yl]-2,2-dimethyl-propionamide (46 g slightly wet with solvent, ~170 mmol) in tetrahydrofuran (275 mL) was treated with 2,2-dimethoxypropane (225 mL, 1.88 mol) and *p*-toluenesulfonic acid monohydrate (3.4 g, 17.9 mmol). The reaction mixture was stirred at 25 °C for 16.5 h. Thin layer chromatography showed that the reaction was complete to form a less polar product. The reaction mixture was concentrated *in vacuo*, and the residue was dissolved in methylene chloride (600 mL). The organic layer was washed with a saturated aqueous sodium chloride solution (250 mL) and a saturated aqueous sodium bicarbonate solution (250 mL). Each aqueous layer was back-extracted with methylene chloride (250 mL). The combined organic layers were stirred with sodium sulfate (35 mg) and Norit A
20 Charcoal (8 g) and then filtered through a pad of celite. The filtrate was concentrated *in vacuo* to a weight of about 250 g. The material was treated with diethyl ether (300 mL), and the mixture again was concentrated *in vacuo* to a weight of about 350 g, at which time, crystallization began. The mixture was stored in a refrigerator (4 °C) for 4 h and filtered. The solids were dried in a vacuum oven at 30 °C for 16 h to afford white crystals
25 (32.3 g, 68%), mp 144-144.5 °C. Collection of an additional crop from the mother liquor afforded white crystals (9.5 g, 20%) which were comparable in purity to the first crop. High-performance liquid chromatography analysis with a chiral column indicated both crops were 100% ee as compared to an authentic racemate sample. The two crops were
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combined to afford the desired N-[5-((S)-2,2-dimethyl-[1,3]dioxolan-4-yl)-pyrazin-2-yl]-2,2-dimethyl-propionamide.

Step 2: Preparation of 5-((S)-2,2-dimethyl-[1,3]dioxolan-4-yl)-pyrazin-2-ylamine (13)

A mixture of N-[5-((S)-2,2-dimethyl-[1,3]dioxolan-4-yl)-pyrazin-2-yl]-2,2-dimethyl-propionamide (8.4 g, 30.7 mmol) and potassium carbonate (4.32 g, 31.2 mmol) in methanol (150 mL) was stirred at 25 °C for 16.5 h, at which time, thin layer chromatography suggested partial conversion to a more polar product. In an effort to avoid epimerization at the stereogenic center, the reaction was discontinued before completion. Therefore, the solvent was removed under reduced pressure at 25 °C. The resulting residue was again concentrated *in vacuo* from ethyl acetate (50 mL). The material was purified using Biotage chromatography (FLASH 40L, Silica, ethyl acetate). The early fractions collected allowed for the recovery of unreacted starting pivaloylamide as a white solid (2.0 g, 24 %). The later fractions were concentrated *in vacuo* to provide 5-((S)-2,2-dimethyl-[1,3]dioxolan-4-yl)-pyrazin-2-ylamine (3.7 g, 63 %) as a pale yellow oil. High-performance liquid chromatography analysis with a chiral column indicated 100% ee.

Step 3: Preparation of 2(R)-(3-chloro-4-methanesulfonyl-phenyl)-3-cyclopentyl-N-[5-((S)-2,2-dimethyl-[1,3]dioxolan-4-yl)-pyrazin-2-yl]-propionamide (14)

A solution of 2(R)-(3-chloro-4-methanesulfonyl-phenyl)-3-cyclopentyl-propionic acid (prepared as in Example 1, 6.29 g, 19.01 mmol) and *N,N*-dimethylformamide (2 drops) in methylene chloride (70 mL) was stirred at 2°C and then treated with oxalyl chloride (4.15 mL, 45.7 mmol). The mixture was stirred at 2 °C for 5 min and at 25 °C for 15 min. The reaction mixture was then concentrated *in vacuo*. The residue was dissolved in benzene (25 mL), and the evaporation was repeated. The resulting acid chloride was dissolved in methylene chloride (40 mL), cooled to at 0 °C, and then treated with a solution composed of 5-((S)-2,2-dimethyl-[1,3]dioxolan-4-yl)-pyrazin-2-ylamine (3.65 g, 18.95 mmol), pyridine (4.6 mL, 56.9 mmol) and methylene chloride (40 mL). The mixture was stirred for 16 h without replenishing the cooling bath. The reaction mixture was then treated with a 1N aqueous hydrochloric acid solution (100 mL). The layers were separated, and the aqueous layer was extracted with methylene chloride (75 mL). The organic layers were washed with a saturated aqueous sodium bicarbonate solution (100 mL) and a saturated aqueous sodium chloride solution. The combined organic layers were dried over sodium sulfate, filtered, and concentrated *in vacuo*. Biotage chromatography (FLASH 40L, Silica, 1/1 ethyl acetate/hexanes) afforded 2(R)-(3-chloro-4-methanesulfonyl-phenyl)-3-cyclopentyl-N-[5-((S)-2,2-dimethyl-

[1,3]dioxolan-4-yl)-pyrazin-2-yl]-propionamide (8.9 g, 92%) as a white foam: (ES)⁺-HRMS m/e calcd for C₂₄H₃₀ClN₃O₅S (M+H)⁺ 508.1668, found 508.1671.

Step 4: Preparation of 2(R)-(3-chloro-4-methanesulfonyl-phenyl)-3-cyclopentyl-N-[5-(1(S),2-dihydroxy-ethyl)-pyrazin-2-yl]-propionamide (III)

- 5 A solution of 2(R)-(3-chloro-4-methanesulfonyl-phenyl)-3-cyclopentyl-N-[5-((S)-2,2-dimethyl-[1,3]dioxolan-4-yl)-pyrazin-2-yl]-propionamide (8.85 g, 17.4 mmol) in tetrahydrofuran (50 mL) was treated with a 1N aqueous hydrochloric acid solution (50 mL). The resulting milky reaction mixture was stirred at 25 °C, and within 15 min, the milky reaction mixture became clear. The stirring was continued at 25 °C for 16 h. The
- 10 reaction was concentrated *in vacuo*, and the residue was extracted with methylene chloride (1 x 100 mL then 2 x 50 mL). Each organic extract was washed with a saturated aqueous sodium bicarbonate solution (50 mL) and a saturated aqueous sodium chloride solution (50 mL). The combined organic layers were dried over sodium sulfate, filtered, and concentrated *in vacuo*. Biotage chromatography (FLASH 40L, Silica, 1/1 ethyl
- 15 acetate/hexanes then 100% ethyl acetate) afforded 2(R)-(3-chloro-4-methanesulfonyl-phenyl)-3-cyclopentyl-N-[5-(1(S),2-dihydroxy-ethyl)-pyrazin-2-yl]-propionamide (7.15 g, 88%) as a colorless foam: (ES)⁺-HRMS m/e calcd for C₂₁H₂₆ClN₃O₅S (M+H)⁺ 468.1355, found 468.1360.

BUDAPEST TREATY ON THE INTERNATIONAL
RECOGNITION OF THE DEPOSIT OF MICROORGANISMS
FOR THE PURPOSES OF PATENT PROCEDURE

INTERNATIONAL FORM

F. Hoffmann-La Roche Ltd.
Bldg. 62/136
4070 Basel
Switzerland

RECEIPT IN THE CASE OF AN ORIGINAL DEPOSIT
issued pursuant to Rule 7.1 by the
INTERNATIONAL DEPOSITARY AUTHORITY
identified at the bottom of this page

I. IDENTIFICATION OF THE MICROORGANISM	
Identification reference given by the DEPOSITOR: R2599	Accession number given by the INTERNATIONAL DEPOSITARY AUTHORITY: DSM 19155
II. SCIENTIFIC DESCRIPTION AND/OR PROPOSED TAXONOMIC DESIGNATION	
The microorganism identified under I above was accompanied by: () a scientific description (X) a proposed taxonomic designation (Mark with a cross where applicable).	
III. RECEIPT AND ACCEPTANCE	
This International Depositary Authority accepts the microorganism identified under I. above, which was received by it on 2007-03-09 (Date of the original deposit) ¹	
IV. RECEIPT OF REQUEST FOR CONVERSION	
The microorganism identified under I above was received by this International Depositary Authority on (date of original deposit) and a request to convert the original deposit to a deposit under the Budapest Treaty was received by it on (date of receipt of request for conversion).	
V. INTERNATIONAL DEPOSITARY AUTHORITY	
Name: DSMZ-DEUTSCHE SAMMLUNG VON MIKROORGANISMEN UND ZELLKULTUREN GmbH Address: Inhoffenstr. 7 B D-38124 Braunschweig	Signature(s) of person(s) having the power to represent the International Depositary Authority or of authorized official(s): Date: 2007-03-14

¹ Where Rule 6.4 (d) applies, such date is the date on which the status of international depositary authority was acquired.

BUDAPEST TREATY ON THE INTERNATIONAL
RECOGNITION OF THE DEPOSIT OF MICROORGANISMS
FOR THE PURPOSES OF PATENT PROCEDURE



INTERNATIONAL FORM

F. Hoffmann-La Roche Ltd.
Bldg. 62/136
4070 Basel
Switzerland

VIABILITY STATEMENT
issued pursuant to Rule 10.2 by the
INTERNATIONAL DEPOSITARY AUTHORITY
identified at the bottom of this page

I. DEPOSITOR	II. IDENTIFICATION OF THE MICROORGANISM
Name: F. Hoffmann-La Roche Ltd. Bldg. 62/136 Address: 4070 Basel Switzerland	Accession number given by the INTERNATIONAL DEPOSITARY AUTHORITY: DSM 19155 Date of the deposit or the transfer: 2007-03-09
III. VIABILITY STATEMENT	
The viability of the microorganism identified under II above was tested on 2007-03-09 On that date, the said microorganism was ☒ viable ☐ no longer viable	
IV. CONDITIONS UNDER WHICH THE VIABILITY TEST HAS BEEN PERFORMED ¹	
V. INTERNATIONAL DEPOSITARY AUTHORITY	
Name: DSMZ-DEUTSCHE SAMMLUNG VON MIKROORGANISMEN UND ZELLKULTUREN GmbH Address: Inhoffenstr. 7 B D-38124 Braunschweig	Signature(s) of person(s) having the power to represent the International Depositary Authority or of authorized official(s): Date: 2007-03-14

¹ Indicate the date of original deposit or, where a new deposit or a transfer has been made, the most recent relevant date (date of the new deposit or date of the transfer).

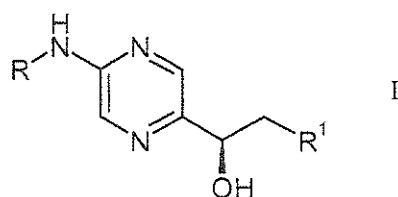
² In the cases referred to in Rule 10.2(a) (ii) and (iii), refer to the most recent viability test.

³ Mark with a cross the applicable box.

⁴ Fill in if the information has been requested and if the results of the test were negative.

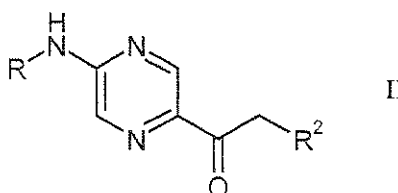
Claims

1. Process for the preparation of compounds of the formula



wherein R is lower alkylcarbonyl or an amino protecting group and R¹ is hydroxy or
5 halogen,

by enzymatic hydrolysis and/or enzymatic asymmetric reduction of a ketone of the
formula



wherein R² is lower alkylcarbonyloxy or halogen.

10 2. Process according to claim 1, characterized in that R¹ is hydroxy and R² is
acetyloxy.

3. Process according to claim 2, characterized in that the enzymatic hydrolysis and
the enzymatic asymmetric reduction is performed with a yeast of the species *Candida*
parapsilosis.

15 4. Process according to claim 3, characterized in that the enzymatic hydrolysis is
performed by means of a hydrolase selected from the group consisting of an esterase, a
protease and a lipase, and that the subsequent enzymatic asymmetric reduction is
performed by means of one or more oxidoreductases.

5. Process according to claim 4, characterized in that the enzymatic hydrolysis is
20 performed by means of a lipase.

6. Process according to claim 5, characterized in that the lipase is obtained from *Candida Antarctica*, *Alcaligenes sp.* or *Burkholderia cepacia*.

7. Process according to claim 4, characterized in that a ketoreductase or an alcohol dehydrogenase is used as oxidoreductase.

5 8. Process according to claim 4 or claim 7, characterized in that one of the ketoreductases selected from the group consisting of KRED 101, KRED107, KRED111, KRED112, KRED113, KRED114, A1F, B1D and B1E in combination with glucose dehydrogenase GDH 102 are used as oxidoreductases.

9. Process according to claim 1, characterized in that R^1 and R^2 are chlorine.

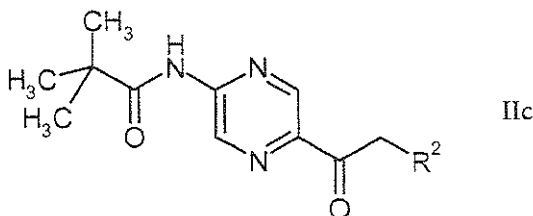
10 10. Process according to claim 9, characterized in that the enzymatic asymmetric reduction is performed by means of one or more oxidoreductases.

11. Process according to claim 10, characterized in that a ketoreductase or an alcohol dehydrogenase is used as oxidoreductase.

12. Process according to any one of claims 1 to 11, characterized in that R is lower
15 alkylcarbonyl.

13. Process according to any one of claims 1 to 12, characterized in that R is tert-butylcarbonyl.

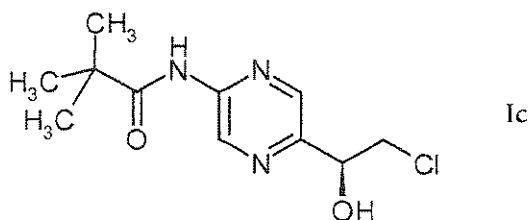
14. Compounds of the formula



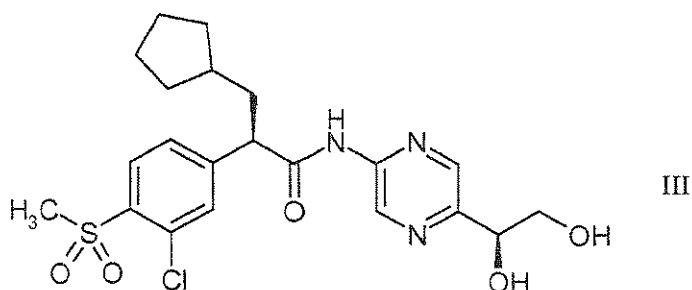
20 wherein R² is lower alkylcarbonyloxy or halogen.

15. Compounds of the formula IIc according to claim 14, wherein R² is acetyloxy or halogen.

16. The compound of the formula

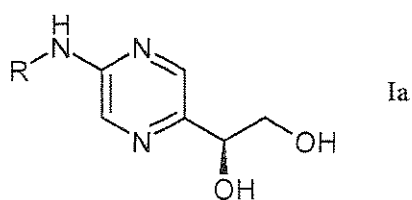


5 17. Process for the preparation of a compound of the formula

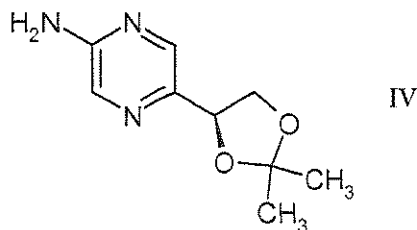


comprising the process according to claims 1 to 12, followed by

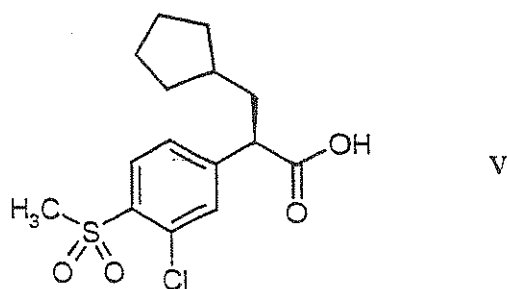
a) reaction of the diol of formula Ia



10 wherein R is lower alkylcarbonyl or an amino protecting group, with 2,2-dimethoxypropane to form an acetal and deprotection of the amine under basic conditions to obtain a compound of formula



b) condensation of the amine of formula IV with the carboxylic acid of formula



or an activated derivative thereof to obtain the amide; and

c) hydrolysis of the acetal under acidic conditions.