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DESCRIPTION

LUBRICATING OIL FOR REFRIGERATOR, WORKING FLUID
COMPOSITION FOR REFRIGERATOR, AND METHOD FOR LUBRICATING
5 REFRIGERATOR

TECHNICAL FIELD

The present invention relates to a lubricating oil
10 for compressors such as those of automotive air
conditioners, refrigerators, room air conditioners, and
large-scale refrigerators for industrial use, and is
intended to provide a low-viscosity lubricating oil for
refrigerators comprising a branched-chain carboxylic acid
15 monoalkyl ester with outstanding hydrolytic stability.

BACKGROUND ART

Unlike the properties required of ordinary
20 lubricating oils such as engine oils, transmission oils,
hydraulic oils and turbine oils, considerably different
performance is required of oils used in the compressors of
refrigerators because of the use of refrigerants in
refrigerators. Particularly, in cases where esters are
25 used as the lubricating oil for refrigerators, the

requisite properties include lubricity as well as miscibility with refrigerants, electrical insulating property and a high level of hydrolytic stability.

Conventionally used refrigerants for refrigerators
5 are fluorocarbons, such as those containing one or more chlorine atoms within the molecule, such as CFC-11 (trichloromonofluoromethane), CFC-12 (dichlorodifluoromethane) and HCFC-22 (monochlorodifluoromethane). However, it has become increasingly evident that such volatile
10 substances are destroying the ozone layer and are a source of global warming, leading to various adverse effects on humans and the earth's environment, so that the manufacture of CFC's is now prohibited, and there is an international commitment to step-wise reduction in HCFC
15 level. Natural refrigerants such as hydrocarbons, carbon dioxide and ammonia, and replacement fluorocarbons such as HFC-134a, are increasingly being used instead as chlorine-free refrigerants that will avoid destroying the ozone layer.

20 Because of good miscibility that conventional chlorofluorocarbons have with lubricating oils based on mineral oils, such mineral oil lubricating oils were used as lubricating oils for refrigerators. Lubricating oils for refrigerators circulate in the refrigerator system
25 along with refrigerant discharged from the compressor, and

then the good miscibility between chlorofluorocarbons and mineral oil-based lubricating oils prevented the refrigerant and lubricating oil from separating.

However, replacement fluorocarbons (hydrofluoro-
5 carbons) such as HFC-134a have much higher molecular polarity than chlorofluorocarbons, they have poor miscibility with mineral oil-based lubricating oils. Poor miscibility between refrigerants and lubricating oils can result in the separation of the refrigerant and
10 lubricating oil during operation, leading to seizure as a result of insufficiently lubricated compressors.

This has led to the use of polyglycols (polyalkylene glycols, polyethers) or esters, which have good miscibility, as lubricating oils to be used with
15 replacement fluorocarbons. However, the electrical insulating property of polyglycols is a problem, and although they can be used in automotive air conditioners, they are unsuitable for use in hermetic refrigerator compressors.

20 Because esters have good electrical insulating properties, they can be used in various applications. Particularly, esters of neopentyl polyol and monovalent carboxylic acids (polyol esters) are actually being used.

However, among the polyol esters, esters of
25 straight-chain carboxylic acids have poor hydrolytic

stability and are decomposed through hydrolysis, and therefore polyol esters of mixtures of branched and straight-chain carboxylic acids, or polyol esters of branched-chain carboxylic acids are being used. Even so, 5 the hydrolytic stability thereof is still poor, requiring complicated operations, such as the need to monitor air and moisture during assembly and maintenance or repair of refrigerators. Thus, there is a need to develop esters with better hydrolytic stability. There is a particular 10 need to develop esters having better electric insulating properties and miscibility with refrigerants, and better hydrolytic stability.

On the other hand, aliphatic carboxylic acid monoesters such as coconut oil methyl ester, 2-ethylhexyl 15 laurate, 2-ethylhexyl palmitate and isobutyl stearate have conventionally been used as base oils in metal processing oils and the like. However, such aliphatic straight-chain carboxylic acid monoesters suffer from poor hydrolytic stability, and have poor miscibility with replacement 20 fluorocarbons, making them unusable as lubricating oils for refrigerators.

Furthermore, in light of the recent problem of global warming, there has been development of high-efficiency equipment for the purposes of energy 25 conservation in various types of refrigerators in order to

reduce carbon dioxide emission, which is a major cause of warming. To minimize energy loss in such equipment, there have been improvements in the materials for the parts that are used, and it has become essential to lower the

5 viscosity of lubricating oils to reduce friction loss during operation. Examples of such low-viscosity oils include those with a kinematic viscosity (at 40 °C) of 32 mm²/s (VG32), 15 mm²/s (VG15), 10 mm²/s (VG10), and some with even lower viscosity properties. There is thus

10 increasing demand for esters with lower viscosity and better hydrolytic stability.

DISCLOSURE OF THE INVENTION

15 In view of the foregoing, an object of the present invention is to provide a lubricating oil for refrigerators that overcomes such drawbacks. A particular object of the present invention is to provide a low-viscosity lubricating oil for refrigerators with better

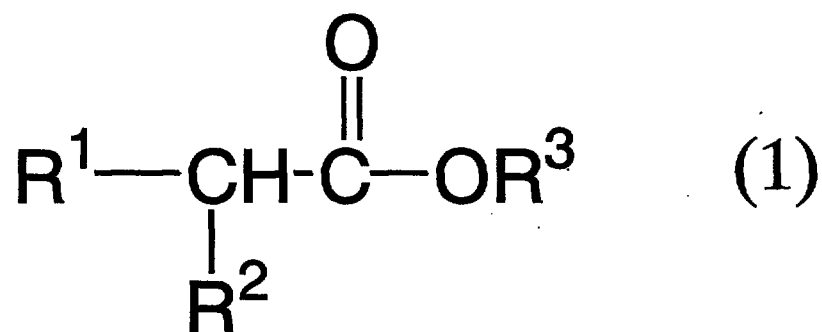
20 hydrolytic stability.

As a result of extensive research to achieve the aforementioned object, the present inventors has found that certain aliphatic branched-chain carboxylic acid monoalkyl esters, prepared by the esterification of a

25 specific aliphatic saturated branched-chain carboxylic

acid and a specific monohydric alcohol, are excellent in hydrolytic stability, electrical insulating properties and miscibility with refrigerants, and also reduce the viscosity of lubricating oils for refrigerators, and these
5 findings have led to the completion of the present invention.

Thus, the lubricating oil for refrigerators according to the present invention comprises at least one aliphatic saturated branched-chain carboxylic acid
10 monoalkyl ester represented by the formula (1)



wherein R^1 is C_1 - C_{18} straight-chain alkyl or C_3 - C_{18} branched-chain alkyl; R^2 is hydrogen, C_1 - C_{18} straight-chain alkyl or C_3 - C_{18} branched-chain alkyl; with the proviso that the
15 total number of carbon atoms contained in the alkyls represented by R^1 and R^2 is 2 to 18 and that when R^2 is hydrogen, R^1 is a branched-chain alkyl; and R^3 is C_1 - C_{20} straight-chain alkyl, C_3 - C_{20} branched-chain alkyl or C_3 - C_{10} cycloalkyl.

It has also been found that (I) at least one aliphatic saturated branched-chain carboxylic monoalkyl ester represented by the aforementioned formula (1) achieve(s) excellent advantages when used in combination with (II) a conventional lubricating oil for refrigerators, such as a fatty acid polyol ester, a phthalic acid ester, an alicyclic dicarboxylic acid ester, a polyvinyl ether, a hydrocarbon oil, a polyalkylene glycol or the like. For example, better hydrolytic stability is achieved in lubricating oils based on both components (I) and (II), and the reduction in the viscosity is achieved in lubricating oils based on components (I) and (II).

Thus, the present invention also provides a lubricating oil for refrigerators, characterized in that it comprises a mixture of

- a) at least one aliphatic saturated branched-chain carboxylic acid monoalkyl ester represented by the formula (1) and
- b) at least one member selected from the group consisting of
 - i) fatty acid polyol esters,
 - ii) phthalic acid esters,
 - iii) alicyclic dicarboxylic acid esters,
 - iv) polyvinyl ethers,

v) hydrocarbon oils, and
vi) polyalkylene glycols,
wherein the weight ratio of component a) : component b) is
0.5 : 99.5 to 99.5 : 0.5.

5

It has also been found that the lubricating oil for
refrigerators of the present invention, particularly its
component, namely at least one aliphatic saturated
branched-chain carboxylic acid monoalkyl ester represented
10 by the aforementioned formula (1), has good miscibility
with refrigerants, particularly hydrofluorocarbon
refrigerants, and that a mixture of at least one such
ester and a refrigerant, particularly a hydrofluorocarbon
refrigerant, acts as a working fluid for refrigerators.

15 Accordingly, the present invention also provides a
working fluid composition for refrigerators, comprising
(I) at least one aliphatic saturated branched-chain
carboxylic acid monoalkyl ester represented by the formula
(1) of the present invention and (II) a refrigerant
20 (particularly at least one hydrofluorocarbon refrigerant).

Additionally, the present invention provides a
working fluid composition for refrigerators, comprising
(I) a) at least one aliphatic saturated branched-chain
25 carboxylic acid monoalkyl ester represented by the

above formula (1), and

- b) at least one member selected from the group consisting of i) fatty acid polyol esters, ii) phthalic acid esters, iii) alicyclic dicarboxylic acid esters, iv) polyvinyl ethers, v) hydrocarbon oils and vi) polyalkylene glycols, and

(II) a refrigerant (particularly, at least one hydrofluorocarbon refrigerant).

10

Furthermore, the present invention provides a method for lubricating a refrigerator, particularly a method for lubricating refrigerators using at least one hydrofluorocarbon refrigerant, characterized by the use of the aforementioned lubricating oil for refrigerators.

Furthermore, the present invention provides use of an aliphatic saturated branched-chain carboxylic acid monoalkyl ester represented by the formula (1) as a lubricating oil for refrigerators, particularly as a lubricating oil for refrigerators using a refrigerant (particularly at least one hydrofluorocarbon refrigerant).

Similarly, the present invention provides use of a mixture of

a) at least one aliphatic saturated branched-chain

carboxylic acid monoalkyl ester represented by the formula (1) and

- b) at least one member selected from the group consisting of i) fatty acid polyol esters,
5 ii) phthalic acid esters, iii) alicyclic dicarboxylic acid esters, iv) polyvinyl ethers, v) hydrocarbon oils and vi) polyalkylene glycols,
as a lubricating oil for refrigerators, particularly as a lubricating oil for refrigerators using a refrigerant
10 (particularly at least one hydrofluorocarbon refrigerant).

BRIEF DESCRIPTION OF THE DRAWING

Figure 1 is a schematic diagram of a refrigerator
15 used in the present invention.

DETAILED DESCRIPTION OF THE INVENTION

The components of the lubricating oil for
20 refrigerators (hereinafter referred to as "refrigerator oil") according to the present invention will be described in further detail below.

Aliphatic saturated branched-chain carboxylic
25 acid monoalkyl ester represented by the formula (1)

The aliphatic saturated branched-chain carboxylic acid monoalkyl ester of the present invention is represented by the foregoing formula (1) above.

5 In the formula (1), the total number of carbon atoms of the alkyls represented by R^1 and R^2 is preferably 2 to 13, and in such cases R^3 is preferably C_3 to C_{12} straight- or branched-chain alkyl.

Also preferred are esters of the formula (1) wherein
10 R^1 and R^2 are C_1 to C_{17} straight-chain alkyls or C_3 to C_{18} branched-chain alkyls.

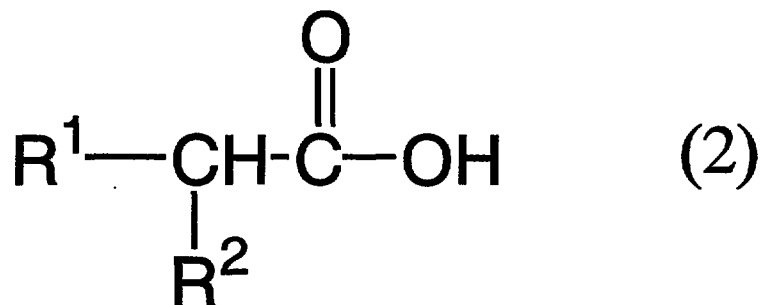
Especially, esters of the formula (1) wherein one of R^1 and R^2 is C_1 to C_5 straight-chain alkyl or C_3 to C_5 branched-chain alkyl, and the other of R^1 and R^2 is C_1 to
15 C_{17} (particularly C_1 to C_9) straight-chain alkyl or C_3 to C_{17} (particularly C_3 to C_9) branched-chain alkyl are also preferred.

The aliphatic saturated branched-chain carboxylic acid monoalkyl ester represented by the formula (1) above
20 (hereinafter referred to as "Present ester") of the present invention can be produced by various methods. For example, such an ester can be produced by an esterification reaction of a corresponding carboxylic acid and a corresponding alcohol (dehydration condensation), a
25 de-hydrohalogenation reaction of a corresponding

carboxylic acid halide and a corresponding alcohol, an ester interchange reaction or the like.

Present ester can also be produced by hydrogenating a corresponding alkyl ester of an aliphatic unsaturated
5 branched-chain carboxylic acid, alkenyl ester of an aliphatic saturated branched-chain carboxylic acid or alkyl or alkenyl ester of an aliphatic unsaturated branched-chain carboxylic acid.

Generally, Present ester is produced by esterification of a corresponding carboxylic acid and a
10 corresponding alcohol. More specifically, Present ester can be produced by esterifying an aliphatic saturated branched-chain carboxylic acid represented by the formula (2)



15 wherein R^1 and R^2 are as defined above and a monohydric alcohol represented by the formula (3)



20 wherein R^3 is as defined above by a known method,

preferably with heating and stirring in the presence or absence of an esterification catalyst in an atmosphere of an inert gas such as nitrogen.

In the present invention, R^1 and R^2 are groups
5 derived from an aliphatic saturated branched-chain
carboxylic acid represented by the formula (2), and R^1
represents C_1 to C_{18} straight-chain alkyl or C_3 to C_{18}
branched-chain alkyl, R^2 represents hydrogen, C_1 to C_{18}
straight-chain alkyl or C_3 to C_{18} branched-chain alkyl,
10 with the proviso that the total number of carbon atoms
contained in the alkyls represented by R^1 and R^2 is 2 to 18.

Aliphatic saturated branched-chain carboxylic acid
represented by the formula (2)

15 Examples of the aliphatic saturated branched-chain
carboxylic acid represented by the formula (2) include
isobutanoic acid (2-methylpropanoic acid), 2-methyl-
butanoic acid, 3-methylbutanoic acid, isohexanoic acid, 2-
ethylbutanoic acid, 2,3-dimethylbutanoic acid, 2-methyl-
20 pentanoic acid, isoheptanoic acid, 2-methylhexanoic acid,
2-ethylpentanoic acid, 2,4-dimethylpentanoic acid, 3-
methylhexanoic acid, 5-methylhexanoic acid, isooctanoic
acid, 2-ethylhexanoic acid, 2-methylheptanoic acid, 2-
propylpentanoic acid, isononanoic acid, 2-methyloctanoic
25 acid, 2-ethylheptanoic acid, 3,5,5-trimethylhexanoic acid,

isodecanoic acid, 2-ethyloctanoic acid, 2-methylnonanoic acid, isoundecanoic acid, isododecanoic acid, isotridecanoic acid, isotetradecanoic acid, 2-pentyl-
nonanoic acid, isopentadecanoic acid, isohexadecanoic acid,
5 2-hexyldecanoic acid, isoheptadecanoic acid, isooctadecanoic acid, isononadecanoic acid, and isoicosanoic acid.

Preferably, recommended are C₄ to C₁₅ branched-chain carboxylic acids such as isobutanoic acid, 2-methyl-
10 butanoic acid, 3-methylbutanoic acid, isohexanoic acid, 2-ethylbutanoic acid, 2,3-dimethylbutanoic acid, 2-methylpentanoic acid, isoheptanoic acid, 2-methylhexanoic acid, 2-ethylpentanoic acid, 2,4-dimethylpentanoic acid, 3-methylhexanoic acid, 5-methylhexanoic acid, isooctanoic
15 acid, 2-ethylhexanoic acid, 2-methylheptanoic acid, 2-propylpentanoic acid, isononanoic acid, 2-methyloctanoic acid, 2-ethylheptanoic acid, 3,5,5-trimethylhexanoic acid, isodecanoic acid, 2-ethyloctanoic acid, 2-methylnonanoic acid, isoundecanoic acid, isododecanoic acid,
20 isotridecanoic acid, isotetradecanoic acid, 2-pentyl-nonanoic acid, and isopentadecanoic acid.

Furthermore, mixtures of fatty acids such as C₈ acid (product of Exxon Chemical), C₁₀ acid (product of Exxon Chemical), and C₁₃ acid
25 (product of Exxon Chemical) may also be used as the

aliphatic saturated branched-chain carboxylic acids represented by the formula (2).

The aforementioned aliphatic saturated branched-chain carboxylic acids represented by the formula (2) can be used singly or at least two of them may be used.

In order to obtain an ester with better hydrolytic stability and refrigerant miscibility, it is preferable to use an aliphatic saturated branched-chain carboxylic acid with two alkyls at the α position of the carboxyl group, namely a branched-chain carboxylic acid in which R^1 and R^2 represent C_1 to C_{17} straight-chain alkyl or C_3 to C_{17} branched-chain alkyl.

It is particularly preferable to use branched-chain carboxylic acids in which one of R^1 and R^2 is a C_1 to C_5 straight-chain alkyl or C_3 to C_5 branched-chain alkyl, and the other of R^1 and R^2 is a C_1 to C_{17} , more preferably C_1 to C_9 , straight-chain alkyl or C_3 to C_{17} , more preferably C_3 to C_9 , branched-chain alkyl.

Specifically, isobutanoic acid, 2-methylbutanoic acid, 2-ethylbutanoic acid, 2,3-dimethylbutanoic acid, 2-methylpentanoic acid, 2-methylhexanoic acid, 2-ethylpentanoic acid, 2,4-dimethylpentanoic acid, 2-ethylhexanoic acid, 2-methylheptanoic acid, 2-propylpentanoic acid, 2-ethyloctanoic acid, 2-methylnonanoic acid, and 2-pentylnonanoic acid are

preferable. In particular, 2-ethylhexanoic acid is recommendable because of its low metal corrosion property.

Monohydric alcohols represented by the formula (3)

5 On the other hand, R^3 in the present invention is a group derived from a monohydric alcohols represented by the formula (3). Examples thereof include C_1 to C_{20} (particularly C_3 to C_{12}) straight-chain alkyl, C_3 to C_{20} (particularly C_3 to C_{12}) branched-chain alkyl, or C_3 to C_{10} (particularly C_6 to C_8) cycloalkyl.

Specific examples of R^3 include isopropyl, isobutyl, sec-butyl, tert-butyl, isopentyl, isohexyl, isoheptyl, 2-methylhexyl, 2-ethylhexyl, 1-methylheptyl, 2-methylheptyl, isooctyl, isononyl, 3,5,5-trimethylhexyl, 2,6-dimethyl-4-heptyl, isodecyl, isoundecyl, isododecyl, isotridecyl, isotetradecyl, isopentadecyl, isohexadecyl, isoheptadecyl, isooctadecyl, isononadecyl, isoicosyl, methyl, ethyl, n-propyl, n-butyl, n-pentyl, n-hexyl, n-heptyl, n-octyl, n-nonyl, n-decyl, n-undecyl, n-dodecyl, n-tridecyl, n-tetradecyl, n-pentadecyl, n-hexadecyl, n-heptadecyl, n-octadecyl, n-nonadecyl, n-icosyl, cyclohexyl, methylcyclohexyl, and dimethylcyclohexyl. Particularly preferred examples of R^3 include C_3 to C_{12} straight- or branched-chain (particularly branched) alkyl or C_6 to C_8 cycloalkyl (particularly C_3 to C_{12} straight- or branched-

chain alkyl), such as isobutyl, sec-butyl, cyclohexyl, n-heptyl, isoheptyl, 2-ethylhexyl, 1-methylheptyl, isooctyl, isononyl, 3,5,5-trimethylhexyl, 2,6-dimethyl-4-heptyl, isodecyl and isoundecyl.

- 5 Examples of monohydric alcohols represented by the formula (3) include C_1 to C_{20} straight-chain alcohols, C_3 to C_{20} branched-chain alcohols, and C_3 to C_{10} cycloalcohols.

Specific examples of straight-chain alcohols include methanol, ethanol, n-propanol, n-butanol, n-pentanol, n-
10 hexanol, n-heptanol, n-octanol, n-nonanol, n-decanol, n-undecanol, n-dodecanol, n-tridecanol, n-tetradecanol, n-pentadecanol, n-hexadecanol, n-octadecanol, n-nonadecanol, n-icosanol and the like.

Specific examples of branched-chain alcohols include
15 isopropanol, isobutanol, sec-butanol, tert-butanol, isopentanol, isohexanol, isoheptanol, 2-methylhexanol, 2-ethylhexanol, 1-methylheptanol, 2-methylheptanol, isooctanol, isononanol, 3,5,5-trimethylhexanol, 2,6-dimethyl-4-heptanol, isodecanol, isoundecanol,
20 isododecanol, isotridecanol, isotetradecanol, isopentadecanol, isohexadecanol, isooctadecanol, isononadecanol, iso-icosanol and the like.

Specific examples of cycloalcohols include cyclohexanol, methylcyclohexanol, dimethylcyclohexanol and
25 so on.

Among the above alcohols, straight-chain alcohols and cycloalcohols are preferred in terms of better lubricity, and preferred examples thereof include C_3 to C_{12} straight-chain alcohols and C_6 to C_8 cycloalcohols, such as
5 n-propanol, n-butanol, n-pentanol, n-hexanol, n-heptanol, n-octanol, n-nonanol, n-decanol, n-undecanol, n-dodecanol, and cyclohexanol.

In terms of better refrigerant miscibility and hydrolytic stability, on the other hand, branched-chain
10 alcohols are preferred. Preferred examples thereof include C_3 to C_{12} branched-chain alcohols such as isopropanol, isobutanol, sec-butanol, isoheptanol, 2-ethylhexanol, 1-methylheptanol, isooctanol, isononanol, 3,5,5-trimethylhexanol, 2,6-dimethyl-4-heptanol,
15 isodecanol, and isoundecanol.

In view of the balance of the physical properties required of refrigerator oils, such as lubricity, refrigerant miscibility, hydrolytic stability and electrical insulating properties, more preferable as the
20 alcohols represented by the formula (3) are C_3 to C_{12} straight-chain or branched-chain (preferably branched-chain) alcohols or C_6 to C_8 cycloalcohols, such as isopropanol, isobutanol, sec-butanol, cyclohexanol, n-heptanol, isoheptanol, 2-ethylhexanol, 1-methylheptanol,
25 isooctanol, isononanol, 3,5,5-trimethylhexanol, 2,6-

dimethyl-4-heptanol, isodecanol and isoundecanol.

Alcohols represented by the formula (3) can be subjected to esterification reaction individually or in the form of a mixture of at least two of them.

5 Specific examples of such a mixture of alcohols include isobutanol/2-ethylhexanol, isobutanol/n-octanol, isobutanol/isononanol, isobutanol/3,5,5-trimethylhexanol, isobutanol/isodecanol, isobutanol/n-decanol, isobutanol/n-undecanol, isobutanol/isoundecanol, isobutanol/n-dodecanol,
10 2-ethylhexanol/n-octanol, 2-ethylhexanol/isononanol, 2-ethylhexanol/3,5,5-trimethylhexanol, 2-ethylhexanol/isodecanol, 2-ethylhexanol/n-decanol, 2-ethylhexanol/n-undecanol, 2-ethylhexanol/isoundecanol, 2-ethylhexanol/n-dodecanol, n-octanol/isononanol, n-octanol/3,5,5-
15 trimethylhexanol, n-octanol/isodecanol, n-octanol/n-decanol, n-octanol/n-undecanol, n-octanol/isoundecanol, n-octanol/n-dodecanol, isononanol/3,5,5-trimethylhexanol, isononanol/isodecanol, isononanol/n-decanol, isononanol/n-undecanol, isononanol/isoundecanol, isononanol/n-dodecanol,
20 3,5,5-trimethylhexanol/isodecanol, 3,5,5-trimethylhexanol/n-decanol, 3,5,5-trimethylhexanol/n-undecanol, 3,5,5-trimethylhexanol/isoundecanol, and 3,5,5-trimethylhexanol/n-dodecanol.

When at least two alcohols, particularly two alcohols,
25 are used in admixture, they are mixed in a molar ratio of

95:5 to 5:95, more preferably 90:10 to 10:90, and even more preferably 70:30 to 30:70.

When a balance of refrigerant miscibility, hydrolytic stability and lubricity is particularly
5 important, the combined use of a straight-chain alcohol and a branched-chain alcohol is preferred.

When a straight-chain alcohol is used in combination with a branched-chain alcohol, it is recommended that the content of a C_{12} to C_{20} straight-chain alcohols be 5 to 50
10 mol% of the total alcohol for the sake of better refrigerant miscibility and hydrolytic stability.

Esterification

In carrying out the esterification reaction, the
15 amounts of the starting materials are such that the alcohol represented by the formula (3) is used, for example, in an amount of 1 to 1.5 equivalents, preferably 1.05 to 1.2 equivalents, per equivalent of the carboxylic acid represented by the formula (2).

20 The present ester can also be obtained by an ester interchange reaction in which, in place of the carboxylic acid represented by the formula (2) or the alcohol represented by the formula (3), a lower alcohol ester of said carboxylic acid and/or a C_1 to C_4 lower alkanolic acid
25 ester of an alcohol represented by the formula (3) such as

an acetic acid ester or a propionic acid ester of said alcohol.

It is preferable that the acid and alcohol components to be used in the esterification reaction are as free of impurities as possible. Impurities include metal compounds, hetero atom-containing compounds (such as sulfur-containing compounds, nitrogen-containing compounds, and phosphorus-containing compounds), and oxygen-containing compounds (such as peroxides and carbonyl compounds).

Examples of esterification catalyst include Lewis acids, alkali metals, and sulfonic acids. Specific examples of Lewis acids include aluminum derivatives, tin derivatives, and titanium derivatives. Specific examples of alkali metals include sodium alkoxides and potassium alkoxides. Specific examples of sulfonic acids include para-toluenesulfonic acid, methanesulfonic acid, and sulfuric acid. Acidic ion exchange resins can also be used. Among them, C_3 to C_8 tetraalkyl titanates, titanium oxide, sodium C_1 to C_4 alkoxides, sodium hydroxide, C_3 to C_{12} tin fatty acids, and tin oxide are preferred. These may be used, for example, in an amount of about 0.05 to 1 wt% relative to the total amount of the carboxylic acid represented by the formula (2) and the alcohol represented by the formula (3).

The esterification temperature may be 100 to 230°C, and the reaction is normally completed within 3 to 30 hours.

In the esterification, a water-entraining agent such
5 as benzene, toluene, xylene and cyclohexane can be used to accelerate the distillation of the water produced by the reaction.

Since the water absorption, hydrolytic stability, and electrical insulating properties will be adversely
10 affected in the event that oxygen-containing compounds such as oxides, peroxides and carbonyl compounds are formed as a result of the oxidative degradation of the starting materials, the ester formed and the organic solvent (water-entraining agent) during esterification,
15 the reaction is preferably carried out in a nitrogen or other inert gas atmosphere or in an inert gas stream.

After the completion of the reaction, the excess starting materials are distilled off at reduced or normal pressure. The ester formed can then be purified by a
20 conventional purification method, such as liquid-liquid extraction, distillation under reduced pressure, adsorption by treatment with activated carbon and so forth.

The ester that is obtained may sometimes be contaminated by various impurities depending on the
25 production process. Such impurities include acidic

compounds, metal compounds (organometal compounds and inorganic metal compounds), hetero atom-containing compounds (such as sulfur-containing compounds, nitrogen-containing compounds and phosphorus-containing compounds),
5 and oxygen-containing compounds (such as peroxides and carbonyl compounds). Such impurities lead to poor hydrolytic stability, electric insulating properties and thermal stability, and can also cause corrosion of metal components or production of sludge.

10 Examples of processes for removing the acidic compounds include separation of unreacted starting acids by distillation, thorough neutralization with alkalis and thorough washing with water, and purification with adsorbents that are effective for removing acids. As to
15 the processes for removing metal compounds, metal catalysts are removed, optionally after having been hydrolyzed, by filtration, optionally followed by purification treatment of the product with adsorbents that are effective for removing metals. Alkali metals used for
20 neutralization can be removed by thorough washing with water, optionally followed by purification of the product with an adsorbent that is effective for removing metals. As to the processes for removing hetero atom compounds, there may be mentioned a sufficient neutralization with
25 use of alkalis and washing with water for removing sulfur-

containing catalysts and phosphorus-containing catalysts, and purification with an adsorbent that is effective for removing hetero atoms. Examples of processes for removing peroxides and carbonyl compounds include reduction

5 treatment with metal hydrides such as sodium borohydride and lithium aluminum hydride.

Specific examples of the adsorbents used in such purification include natural and synthetic adsorbents, such as activated carbon, activated alumina, silica gel,
10 silica alumina, activated clay, zeolites, magnesia, calcia, diatomaceous earth and hydrotalcite, as well as non-sulfonic acid-based ion exchange resins, synthetic hydrotalcites and similar synthetic adsorbents. Although the amount thereof to be used will vary depending on the
15 type of adsorbent, an amount of 0.01 to 5 wt%, preferably 0.05 to 5 wt%, relative to the theoretical yield of the ester is usually recommended.

Particularly, an ester purified by using one or more adsorbents selected from the group consisting of activated
20 carbon, activated alumina, silica gel, activated clay, zeolites, hydrotalcite, non-sulfonic acid-based ion exchange resins and synthetic hydrotalcites at room temperature to 110°C, preferably room temperature to 90°C, has the following properties a) through h) and are
25 especially suitable as lubricating oils for refrigerators.

a) The total acid number of the present esters is recommendably not more than 0.1 mgKOH/g, preferably not more than 0.05 mgKOH/g, and even more preferably not more than 0.03 mgKOH/g. A total acid number of not more than
5 0.1 mgKOH/g will make it easier to achieve good hydrolytic stability, with minimal metal corrosion property.

b) Ash content or sulfated ash content as defined in JIS-K-2272 of the present esters is recommendably not more than 30 ppm, preferably not more than 10 ppm, and even
10 more preferably not more than 5 ppm. An ash content or sulfated ash content of not more than 30 ppm will easily allow better hydrolytic stability and thermal stability to be achieved.

c) The present esters recommendably have a sulfur
15 content of not more than 50 ppm, preferably not more than 30 ppm, and even more preferably not more than 10 ppm. A sulfur content of not greater than 50 ppm will readily allow better hydrolytic stability and thermal stability to be achieved.

20 d) The present esters recommendably have a phosphorus content of not more than 50 ppm, preferably not more than 30 ppm, and even more preferably not more than 10 ppm. A phosphorus content not greater than 50 ppm will readily allow better hydrolytic stability and thermal
25 stability to be achieved.

e) The present esters recommendably have a hydroxyl value of not more than 5 mgKOH/g, preferably not more than 3 mgKOH/g. A hydroxyl value of not greater than 5 mgKOH/g will readily result in an ester with less hygroscopicity, and will allow better hydrolytic stability and electrical insulating properties to be achieved.

f) The present esters recommendably have a peroxide value of not more than 1.0 meq/kg, preferably not more than 0.5 meq/kg. A peroxide value of not greater than 1.0 meq/kg will tend to result in better thermal stability.

g) The present ester recommendably have a carbonyl value of not more than 5.0, preferably not more than 3.0, and even more preferably not more than 1.0. A carbonyl value of not greater than 5.0 will result in better thermal stability.

h) The hue of the present esters are recommendably such that they have a Hazen number of not greater than 100, preferably not greater than 50, and even more preferably not greater than 30. A Hazen number of not greater than 100 will result in better thermal stability.

Present ester

Specific examples of present esters obtained in the manner described above include at least one member selected from the group consisting of isopropyl

- isobutanoate, isobutyl isobutanoate, (sec-butyl)
isobutanoate, isopentyl iso-butanoate, cyclohexyl
isobutanoate, isoheptyl isobutanoate, (2-ethylhexyl)
isobutanoate, (1-methylheptyl) isobutanoate, isooctyl
5 isobutanoate, isononyl isobutanoate, (3,5,5-
trimethylhexyl) isobutanoate, (2,6-dimethyl-4-heptyl) iso-
butanoate, isodecyl isobutanoate, isoundecyl isobutanoate,
(n-propyl) isobutanoate, (n-butyl) isobutanoate, (n-
pentyl) isobutanoate, (n-hexyl) isobutanoate, (n-heptyl)
10 isobutanoate, (n-octyl) isobutanoate, (n-nonyl)
isobutanoate, (n-decyl) isobutanoate, (n-undecyl)
isobutanoate, (n-dodecyl) isobutanoate, isopropyl 2-
methylbutanoate, isobutyl 2-methylbutanoate, (sec-butyl)
2-methylbutanoate, isopentyl 2-methylbutanoate, cyclohexyl
15 2-methylbutanoate, isoheptyl 2-methylbutanoate, (2-
ethylhexyl) 2-methylbutanoate, (1-methylheptyl) 2-
methylbutanoate, isooctyl 2-methylbutanoate, isononyl 2-
methylbutanoate, (3,5,5-trimethylhexyl) 2-methylbutanoate,
(2,6-dimethyl-4-heptyl) 2-methylbutanoate, isodecyl 2-
20 methylbutanoate, isoundecyl 2-methylbutanoate, (n-propyl)
2-methylbutanoate, (n-butyl) 2-methylbutanoate, (n-pentyl)
2-methylbutanoate, (n-hexyl) 2-methylbutanoate, (n-heptyl)
2-methylbutanoate, (n-octyl) 2-methylbutanoate, (n-nonyl)
2-methylbutanoate, (n-decyl) 2-methylbutanoate, (n-
25 undecyl) 2-methylbutanoate, (n-dodecyl) 2-methylbutanoate,

isopropyl 2-ethylbutanoate, isobutyl 2-ethylbutanoate,
(sec-butyl) 2-ethylbutanoate, isopentyl 2-ethylbutanoate,
cyclohexyl 2-ethylbutanoate, isoheptyl 2-ethylbutanoate,
(2-ethylhexyl) 2-ethylbutanoate, (1-methylheptyl) 2-
5 ethylbutanoate, isooctyl 2-ethylbutanoate, isononyl 2-
ethylbutanoate, (3,5,5-trimethylhexyl) 2-ethylbutanoate,
(2,6-dimethyl-4-heptyl) 2-ethylbutanoate, isodecyl 2-
ethylbutanoate, isoundecyl 2-ethylbutanoate, (n-propyl) 2-
ethylbutanoate, (n-butyl) 2-ethylbutanoate, (n-pentyl) 2-
10 ethylbutanoate, (n-hexyl) 2-ethylbutanoate, (n-heptyl) 2-
ethylbutanoate, (n-octyl) 2-ethylbutanoate, (n-nonyl) 2-
ethylbutanoate, (n-decyl) 2-ethylbutanoate, (n-undecyl) 2-
ethylbutanoate, (n-dodecyl) 2-ethylbutanoate, isopropyl
2,3-dimethylbutanoate, isobutyl 2,3-dimethylbutanoate,
15 (sec-butyl) 2,3-dimethylbutanoate, isopentyl 2,3-
dimethylbutanoate, cyclohexyl 2,3-dimethylbutanoate,
isoheptyl 2,3-dimethylbutanoate, (2-ethylhexyl) 2,3-
dimethylbutanoate, (1-methylheptyl) 2,3-dimethylbutanoate,
isooctyl 2,3-dimethylbutanoate, isononyl 2,3-dimethyl-
20 butanoate, (3,5,5-trimethylhexyl) 2,3-dimethylbutanoate,
(2,6-dimethyl-4-heptyl) 2,3-dimethylbutanoate, isodecyl
2,3-dimethylbutanoate, isoundecyl 2,3-dimethylbutanoate,
(n-propyl) 2,3-dimethylbutanoate, (n-butyl) 2,3-dimethyl-
butanoate, (n-pentyl) 2,3-dimethylbutanoate, (n-hexyl)
25 2,3-dimethylbutanoate, (n-heptyl) 2,3-dimethylbutanoate,

(n-octyl) 2,3-dimethylbutanoate, (n-nonyl) 2,3-dimethylbutanoate, (n-decyl) 2,3-dimethylbutanoate, (n-undecyl) 2,3-dimethylbutanoate, (n-dodecyl) 2,3-dimethylbutanoate, isopropyl 2-methylpentanoate, isobutyl
5 2-methylpentanoate, (sec-butyl) 2-methylpentanoate, isopentyl 2-methylpentanoate, cyclohexyl 2-methylpentanoate, isoheptyl 2-methylpentanoate, (2-ethylhexyl) 2-methylpentanoate, (1-methylheptyl) 2-methylpentanoate, isooctyl 2-methylpentanoate, isononyl 2-methylpentanoate, (3,5,5-trimethylhexyl) 2-methyl-
10 pentanoate, (2,6-dimethyl-4-heptyl) 2-methylpentanoate, isodecyl 2-methylpentanoate, isoundecyl 2-methylpentanoate, (n-propyl) 2-methylpentanoate, (n-butyl) 2-methylpentanoate, (n-pentyl) 2-methylpentanoate, (n-hexyl) 2-methylpentanoate, (n-heptyl) 2-methylpentanoate, (n-octyl) 2-methylpentanoate, (n-nonyl) 2-methylpentanoate, (n-decyl) 2-methylpentanoate, (n-undecyl) 2-methylpentanoate, (n-dodecyl) 2-methylpentanoate, isopropyl 2-methylhexanoate, isobutyl 2-methylhexanoate,
15 (sec-butyl) 2-methylhexanoate, isopentyl 2-methylhexanoate, cyclohexyl 2-methylhexanoate, isoheptyl 2-methylhexanoate, (2-ethylhexyl) 2-methylhexanoate, (1-methylheptyl) 2-methylhexanoate, isooctyl 2-methylhexanoate, isononyl 2-methylhexanoate, (3,5,5-trimethylhexyl) 2-methylhexanoate, (2,6-dimethyl-4-heptyl) 2-methylhexanoate, isodecyl 2-

25

methylhexanoate, isoundecyl 2-methylhexanoate, (n-propyl)
2-methylhexanoate, (n-butyl) 2-methylhexanoate, (n-pentyl)
2-methylhexanoate, (n-hexyl) 2-methylhexanoate, (n-heptyl)
2-methylhexanoate, (n-octyl) 2-methylhexanoate, (n-nonyl)
5 2-methylhexanoate, (n-decyl) 2-methylhexanoate, (n-
undecyl) 2-methylhexanoate, (n-dodecyl) 2-methylhexanoate,
isopropyl 2-ethylpentanoate, isobutyl 2-ethylpentanoate,
(sec-butyl) 2-ethylpentanoate, isopentyl 2-ethylpentanoate,
cyclohexyl 2-ethylpentanoate, isoheptyl 2-ethylpentanoate,
10 (2-ethylhexyl) 2-ethylpentanoate, (1-methylheptyl) 2-
ethylpentanoate, isooctyl 2-ethylpentanoate, isononyl 2-
ethylpentanoate, (3,5,5-trimethylhexyl) 2-ethylpentanoate,
(2,6-dimethyl-4-heptyl) 2-ethylpentanoate, isodecyl 2-
ethylpentanoate, isoundecyl 2-ethylpentanoate, (n-propyl)
15 2-ethylpentanoate, (n-butyl) 2-ethylpentanoate, (n-pentyl)
2-ethylpentanoate, (n-hexyl) 2-ethylpentanoate, (n-heptyl)
2-ethylpentanoate, (n-octyl) 2-ethylpentanoate, (n-nonyl)
2-ethylpentanoate, (n-decyl) 2-ethylpentanoate, (n-
undecyl) 2-ethylpentanoate, (n-dodecyl) 2-ethylpentanoate,
20 isopropyl 2,4-dimethylpentanoate, isobutyl 2,4-
dimethylpentanoate, (sec-butyl) 2,4-dimethylpentanoate,
isopentyl 2,4-dimethylpentanoate, cyclohexyl 2,4-
dimethylpentanoate, isoheptyl 2,4-dimethylpentanoate, (2-
ethylhexyl) 2,4-dimethylpentanoate, (1-methylheptyl) 2,4-
25 dimethylpentanoate, isooctyl 2,4-dimethylpentanoate,

isononyl 2,4-dimethylpentanoate, (3,5,5-trimethylhexyl)
2,4-dimethylpentanoate, (2,6-dimethyl-4-heptyl) 2,4-
dimethylpentanoate, isodecyl 2,4-dimethylpentanoate,
isoundecyl 2,4-dimethylpentanoate, (n-propyl) 2,4-
5 dimethylpentanoate, (n-butyl) 2,4-dimethylpentanoate, (n-
pentyl) 2,4-dimethylpentanoate, (n-hexyl) 2,4-
dimethylpentanoate, (n-heptyl) 2,4-dimethylpentanoate, (n-
octyl) 2,4-dimethylpentanoate, (n-nonyl) 2,4-
dimethylpentanoate, (n-decyl) 2,4-dimethylpentanoate, (n-
10 undecyl) 2,4-dimethylpentanoate, (n-dodecyl) 2,4-
dimethylpentanoate, isopropyl 2-ethylhexanoate, isobutyl
2-ethylhexanoate, (sec-butyl) 2-ethylhexanoate, isopentyl
2-ethylhexanoate, cyclohexyl 2-ethylhexanoate, isoheptyl
2-ethylhexanoate, (2-ethylhexyl) 2-ethylhexanoate, (1-
15 methylheptyl) 2-ethylhexanoate, isooctyl 2-ethylhexanoate,
isononyl 2-ethylhexanoate, (3,5,5-trimethylhexyl) 2-
ethylhexanoate, (2,6-dimethyl-4-heptyl) 2-ethylhexanoate,
isodecyl 2-ethylhexanoate, isoundecyl 2-ethylhexanoate,
(n-propyl) 2-ethylhexanoate, (n-butyl) 2-ethylhexanoate,
20 (n-pentyl) 2-ethylhexanoate, (n-hexyl) 2-ethylhexanoate,
(n-heptyl) 2-ethylhexanoate, (n-octyl) 2-ethylhexanoate,
(n-nonyl) 2-ethylhexanoate, (n-decyl) 2-ethylhexanoate,
(n-undecyl) 2-ethylhexanoate, (n-dodecyl) 2-ethylhexanoate,
isopropyl 2-methylheptanoate, isobutyl 2-methylheptanoate,
25 (sec-butyl) 2-methylheptanoate, isopentyl 2-

methylheptanoate, cyclohexyl 2-methylheptanoate, isoheptyl
2-methylheptanoate, (2-ethylhexyl) 2-methylheptanoate, (1-
methylheptyl) 2-methylheptanoate, isooctyl 2-
methylheptanoate, isononyl 2-methylheptanoate, (3,5,5-
5 trimethylhexyl) 2-methylheptanoate, (2,6-dimethyl-4-
heptyl) 2-methylheptanoate, isodecyl 2-methylheptanoate,
isoundecyl 2-methylheptanoate, (n-propyl) 2-
methylheptanoate, (n-butyl) 2-methylheptanoate, (n-pentyl)
2-methylheptanoate, (n-hexyl) 2-methylheptanoate, (n-
10 heptyl) 2-methylheptanoate, (n-octyl) 2-methylheptanoate,
(n-nonyl) 2-methylheptanoate, (n-decyl) 2-methylheptanoate,
(n-undecyl) 2-methylheptanoate, (n-dodecyl) 2-
methylheptanoate, isopropyl 2-propylpentanoate, isobutyl
2-propylpentanoate, (sec-butyl) 2-propylpentanoate,
15 isopentyl 2-propylpentanoate, cyclohexyl 2-
propylpentanoate, isoheptyl 2-propylpentanoate, (2-
ethylhexyl) 2-propylpentanoate, (1-methylheptyl) 2-
propylpentanoate, isooctyl 2-propylpentanoate, isononyl 2-
propylpentanoate, (3,5,5-trimethylhexyl) 2-
20 propylpentanoate, (2,6-dimethyl-4-heptyl) 2-
propylpentanoate, isodecyl 2-propylpentanoate, isoundecyl
2-propylpentanoate, (n-propyl) 2-propylpentanoate, (n-
butyl) 2-propylpentanoate, (n-pentyl) 2-propylpentanoate,
(n-hexyl) 2-propylpentanoate, (n-heptyl) 2-
25 propylpentanoate, (n-octyl) 2-propylpentanoate, (n-nonyl)

2-propylpentanoate, (n-decyl) 2-propylpentanoate, (n-undecyl) 2-propylpentanoate, (n-dodecyl) 2-propylpentanoate, isopropyl 2-ethyloctanoate, isobutyl 2-ethyloctanoate, (sec-butyl) 2-ethyloctanoate, isopentyl 2-ethyloctanoate, cyclohexyl 2-ethyloctanoate, isoheptyl 2-ethyloctanoate, (2-ethylhexyl) 2-ethyloctanoate, (1-methylheptyl) 2-ethyloctanoate, isooctyl 2-ethyloctanoate, isononyl 2-ethyloctanoate, (3,5,5-trimethylhexyl) 2-ethyloctanoate, (2,6-dimethyl-4-heptyl) 2-ethyloctanoate, isodecyl 2-ethyloctanoate, isoundecyl 2-ethyloctanoate, (n-propyl) 2-ethyloctanoate, (n-butyl) 2-ethyloctanoate, (n-pentyl) 2-ethyloctanoate, (n-hexyl) 2-ethyloctanoate, (n-heptyl) 2-ethyloctanoate, (n-octyl) 2-ethyloctanoate, (n-nonyl) 2-ethyloctanoate, (n-decyl) 2-ethyloctanoate, (n-undecyl) 2-ethyloctanoate, (n-dodecyl) 2-ethyloctanoate, isopropyl 2-methylnonanoate, isobutyl 2-methylnonanoate, (sec-butyl) 2-methylnonanoate, isopentyl 2-methylnonanoate, cyclohexyl 2-methylnonanoate, isoheptyl 2-methylnonanoate, (2-ethylhexyl) 2-methylnonanoate, (1-methylheptyl) 2-methylnonanoate, isooctyl 2-methylnonanoate, isononyl 2-methylnonanoate, (3,5,5-trimethylhexyl) 2-methylnonanoate, (2,6-dimethyl-4-heptyl) 2-methylnonanoate, isodecyl 2-methylnonanoate, isoundecyl 2-methylnonanoate, (n-propyl) 2-methylnonanoate, (n-butyl) 2-methylnonanoate, (n-pentyl) 2-methylnonanoate, (n-hexyl) 2-methylnonanoate, (n-heptyl)

2-methylnonanoate, (n-octyl) 2-methylnonanoate, (n-nonyl) 2-methylnonanoate, (n-decyl) 2-methylnonanoate, (n-undecyl) 2-methylnonanoate, and (n-dodecyl) 2-methylnonanoate.

5 Among the above esters, recommendable in view of better hydrolytic stability and lower metal corrosion property are C₃ to C₁₂ straight-chain or branched-chain alkyl esters or C₆ to C₈ cycloalkyl esters of 2-ethylhexanoic acid, such as isopropyl 2-ethylhexanoate,
10 isobutyl 2-ethylhexanoate, (sec-butyl) 2-ethylhexanoate, isopentyl 2-ethylhexanoate, cyclohexyl 2-ethylhexanoate, isoheptyl 2-ethylhexanoate, (2-ethylhexyl) 2-ethylhexanoate, (1-methylheptyl) 2-ethylhexanoate, isooctyl 2-ethylhexanoate, isononyl 2-ethylhexanoate,
15 (3,5,5-trimethylhexyl) 2-ethylhexanoate, (2,6-dimethyl-4-heptyl) 2-ethylhexanoate, isodecyl 2-ethylhexanoate, isoundecyl 2-ethylhexanoate, (n-propyl) 2-ethylhexanoate, (n-butyl) 2-ethylhexanoate, (n-pentyl) 2-ethylhexanoate, (n-hexyl) 2-ethylhexanoate, (n-heptyl) 2-ethylhexanoate,
20 (n-octyl) 2-ethylhexanoate, (n-nonyl) 2-ethylhexanoate, (n-decyl) 2-ethylhexanoate, (n-undecyl) 2-ethylhexanoate, (n-dodecyl) 2-ethylhexanoate and the like.

Further, among such 2-ethylhexanoate esters, it is recommended, in view of excellent refrigerant miscibility
25 and electrical insulating properties, to use at least one

member selected from the group consisting of isopropyl 2-ethylhexanoate, isobutyl 2-ethylhexanoate, (sec-butyl) 2-ethylhexanoate, cyclohexyl 2-ethylhexanoate, (n-heptyl) 2-ethylhexanoate, isooctyl 2-ethylhexanoate, (2-ethylhexyl) 2-ethylhexanoate, (1-methylheptyl) 2-ethylhexanoate, isooctyl 2-ethylhexanoate, isononyl 2-ethylhexanoate, (3,5,5-trimethylhexyl) 2-ethylhexanoate, (2,6-dimethyl-4-heptyl) 2-ethylhexanoate, isodecyl 2-ethylhexanoate and isoundecyl 2-ethylhexanoate.

10 The esters of the present invention recommendably have a volume resistivity of $1 \times 10^{11} \Omega \cdot \text{cm}$ or more, preferably $1 \times 10^{12} \Omega \cdot \text{cm}$ or more, and more preferably $1 \times 10^{13} \Omega \cdot \text{cm}$ or more. A volume resistivity of less than $10^{11} \Omega \cdot \text{cm}$ is less likely to result in good electrical insulating
15 properties.

Preferably, the ester of the present invention separates from refrigerant at a temperature of 10°C or lower, particularly 0°C or lower, and even more preferably -10°C or lower.

20 The esters of the present invention have a kinematic viscosity of 0.5 to $15 \text{ mm}^2/\text{s}$ at 40°C , preferably 0.5 to $10 \text{ mm}^2/\text{s}$ at 40°C .

In general, exposing the ester to elevated temperatures in the presence of water will result in
25 decomposition into the corresponding carboxylic acid and

alcohol. It is known that refrigerator lubricating oils are exposed to very high temperatures caused by friction during the operation of compressors. The presence of water in the refrigerator system might result in the

5 hydrolysis of the ester serving as the lubricating oil.

It is considered that minimal increase in the total acid number following heat treatment of the ester is indicative of suppressed hydrolysis and higher stability. When compared by this method, polyol esters which can be used
10 as low-viscosity refrigerator lubricating oils suffer from greater increase in the total acid number, greater change in the weight of iron piece immersed in the oil, and changes in appearance due to surface corrosion of metal such as iron, copper and aluminum.

15 By contrast, the present esters are sufficiently resistant to hydrolysis so that there is less increase in acid number and less change in appearance due to surface corrosion of iron and copper.

20 Combined use of present ester and other base oil

The lubricating oil for refrigerators according to the present invention can be mixed with at least one other base oil (henceforth, "combination base oil" to prepare a
25 lubricating oil for refrigerators, provided that the

effects of the present invention are not thereby compromised.

The aliphatic saturated branched-chain carboxylic acid monoalkyl ester (a) represented by the formula (1) and the combination base oil (b) are used in a weight ratio of (a):(b) = 0.5:99.5 to 99.5:0.5, preferably (a):(b) = 5:95 to 95:5, and even more preferably (a):(b) = 15:85 to 85:15.

Examples of the combination base oils include one or more members selected from the group consisting of (b1) hydrocarbon oils, (b2) organic acid esters other than the present ester, (b3) polyalkylene glycols, (b4) polyvinyl ethers, (b5) polyphenyl ethers, (b6) alkyl phenyl ethers, and (b7) silicone oils. These will be described below.

(b1) hydrocarbon oils

Examples of hydrocarbon oils include refined petroleum products such as (b1-1) hydrocarbon oils (mineral oils), (b1-2) poly- α -olefins, (b1-3) polybutenes, (b1-4) alkylbenzenes, (b1-5) alkylnaphthalenes, (b1-6) isomerized hydrocarbons obtained by the Fischer-Tropsch process, and similar synthetic hydrocarbon oils.

(b1-1) Examples of mineral oils include solvent-refined mineral oils, hydrotreated mineral oils, and wax isomerized oils, which usually have a kinematic viscosity

(at 100°C) of 1.0 to 40 mm²/s, and preferably 2.0 to 30 mm²/s.

(b1-2) Examples of poly- α -olefins include C₂ to C₁₆ α -olefin (such as ethylene, propylene, 1-butene, 1-hexene, 5 1-octene, 1-decene, 1-dodecene, 1-tetradecen or 1-hexadecene, etc.) polymers and copolymers having a kinematic viscosity (at 100°C) of 1.0 to 40 mm²/s and a viscosity index of at least 100. Particularly, those having a kinematic viscosity (at 100°C) of 2.0 to 30 mm²/s 10 and a viscosity index of at least 120 are preferred.

(b1-3) Examples of polybutenes include those obtained by the polymerization of isobutylene and those obtained by the copolymerization of isobutylene with n-butylene. There may be mentioned a wide range of those 15 having a kinematic viscosity (at 100°C) of between 2.0 to 6000 mm²/s.

(b1-4) Examples of alkyl benzenes include C₁ to C₄₀ straight-chain or branched-chain alkyl group-substituted monoalkyl benzenes, dialkyl benzenes, trialkyl benzenes, 20 and tetraalkyl benzenes, etc. which have a molecular weight of 200 to 450. Those having one or two alkyl groups wherein the alkyl has 1 to 20 carbon atoms are preferred, because of their longer-lasting stability as a refrigerator oil. Branched-chain alkyl groups are 25 preferred for refrigerator oils with longer-lasting

stability. Specific alkyl groups include methyl, ethyl, n-propyl, n-butyl, n-pentyl, n-hexyl, n-heptyl, n-octyl, n-nonyl, n-decyl, n-undecyl, n-dodecyl, n-tridecyl, n-tetradecyl, n-pentadecyl, n-hexadecyl, n-heptadecyl, n-octadecyl, n-nonadecyl, n-icosyl, isopropyl, isobutyl, isopentyl, isohexyl, isoheptyl, isooctyl, isononyl, isodecyl, isoundecyl, isodecyl, isotridecyl, isotetradecyl, isopentadecyl, isohexadecyl, isoheptadecyl, isooctadecyl, isononadecyl and isoeicosyl.

10 (b1-5) Examples of alkylnaphthalenes include monoalkylnaphthalenes and dialkylnaphthalenes wherein the alkyl is C₁ to C₃₀ straight- or branched-chain alkyl.

(b2) Organic acid esters other than the present ester

Examples of organic acid esters other than the present ester include (b2-1) aliphatic saturated straight-chain carboxylic acid monoesters, (b2-2) aliphatic dibasic acid esters, (b2-3) aromatic polycarboxylic acid esters, (b2-4) alicyclic dicarboxylic acid esters, (b2-5) fatty acid polyol esters, and (b2-6) other esters.

20 (b2-1) Examples of aliphatic saturated straight-chain carboxylic acid monoesters include full esters of C₃ to C₁₈ straight-chain monocarboxylic acids and C₃ to C₂₂ straight- or branched-chain saturated or unsaturated aliphatic alcohols.

25 (b2-2) Examples of aliphatic dibasic acid esters

include full esters of C_6 to C_{12} aliphatic dicarboxylic acid such as adipic acid, azelaic acid, sebacic acid or dodecanedioic acid, and a C_3 to C_{22} straight- or branched-chain saturated or unsaturated aliphatic alcohol.

- 5 (b2-3) Examples of aromatic polycarboxylic acid esters include full esters of an aromatic polycarboxylic acid or its anhydride such as phthalic acid, isophthalic acid, terephthalic acid or trimellitic acid, or pyromellitic acid, and a C_3 to C_{22} straight- or branched-
10 chain saturated or unsaturated aliphatic alcohol.

Preferred examples include full esters of phthalic acid and a C_1 to C_{18} straight- or branched-chain saturated or unsaturated aliphatic alcohol or a C_3 to C_{10} cycloalcohol. Examples of phthalic acid esters include
15 dimethyl phthalate, diethyl phthalate, di(n-propyl) phthalate, di(n-butyl) phthalate, di(n-pentyl) phthalate, di(n-hexyl) phthalate, di(n-heptyl) phthalate, di(n-octyl) phthalate, di(n-nonyl) phthalate, di(n-decyl) phthalate, di(n-undecyl) phthalate, di(n-dodecyl) phthalate, di(n-
20 tridecyl) phthalate, di(n-tetradecyl) phthalate, di(n-pentadecyl) phthalate, di(n-hexadecyl) phthalate, di(n-heptadecyl) phthalate, di(n-octadecyl) phthalate, diisopropyl phthalate, diisobutyl phthalate, (sec-butyl) phthalate, dicyclohexyl phthalate, diisoheptyl phthalate,
25 di(2-ethylhexyl) phthalate, diisononyl phthalate,

di(3,5,5-trimethylhexyl) phthalate, di(2,6-dimethyl-4-heptyl) phthalate, diisodecyl phthalate, diisoundecyl phthalate, didodecyl phthalate, diisotridecyl phthalate, diisotetradecyl phthalate, diisopentadecyl phthalate, 5 diisohexadecyl phthalate, diisoheptadecyl phthalate, and diisooctadecyl phthalate. Especially preferred are di(n-propyl) phthalate, di(n-butyl) phthalate, di(n-pentyl) phthalate, di(n-hexyl) phthalate, di(n-heptyl) phthalate, di(n-octyl) phthalate, di(n-nonyl) phthalate, di(n-decyl) 10 phthalate, di(n-undecyl) phthalate, diisopropyl phthalate, diisobutyl phthalate, (sec-butyl) phthalate, dicyclohexyl phthalate, diisoheptyl phthalate, di(2-ethylhexyl) phthalate, diisononyl phthalate, di(3,5,5-trimethylhexyl) phthalate, di(2,6-dimethyl-4-heptyl) phthalate, diisodecyl 15 phthalate, and diisoundecyl phthalate.

(b2-4) Examples of alicyclic dicarboxylic acid esters include those represented by the formula (4)



20

wherein A is a cyclohexane ring or cyclohexene ring; X is hydrogen or methyl; and R⁴ and R⁵ are the same or different

and each represents a C_3 to C_{22} straight- or branched-chain saturated or unsaturated aliphatic monovalent hydrocarbon group, which may be obtained, for example, by esterification of a corresponding aliphatic dicarboxylic acid and a C_3 to C_{22} straight- or branched-chain saturated or unsaturated aliphatic monohydric alcohol.

Examples of the aforementioned alicyclic dicarboxylic acid include 1,2-cyclohexanedicarboxylic acid, 4-cyclohexene-1,2-dicarboxylic acid, 1-cyclohexene-1,2-dicarboxylic acid, 1,3-cyclohexanedicarboxylic acid, 1,4-cyclohexanedicarboxylic acid, 3-methyl-1,2-cyclohexanedicarboxylic acid, 4-methyl-1,2-cyclohexanedicarboxylic acid, 3-methyl-4-cyclohexene-1,2-dicarboxylic acid and/or 4-methyl-4-cyclohexene-1,2-dicarboxylic acid and anhydrides of these.

Examples of particularly desirable alicyclic dicarboxylic acid esters include esters of the formula (4) in which R^4 and R^5 are the same and each represents a C_3 to C_{11} straight- or branched-chain alkyl. Specific examples thereof include di(n-propyl) 1,2-cyclohexanedicarboxylate, di(n-butyl) 1,2-cyclohexanedicarboxylate, di(n-pentyl) 1,2-cyclohexanedicarboxylate, di(n-hexyl) 1,2-cyclohexanedicarboxylate, di(n-heptyl) 1,2-cyclohexanedicarboxylate, di(n-octyl) 1,2-cyclohexanedicarboxylate, di(n-nonyl) 1,2-cyclohexanedicarboxylate,

di(n-decyl) 1,2-cyclohexanedicarboxylate, di(n-undecyl)
1,2-cyclohexanedicarboxylate, diisopropyl 1,2-cyclohexane-
dicarboxylate, diisobutyl 1,2-cyclohexanedicarboxylate,
di(sec-butyl) 1,2-cyclohexanedicarboxylate, dicyclohexyl
5 1,2-cyclohexanedicarboxylate, diisoheptyl 1,2-cyclohexane-
dicarboxylate, di(2-ethylhexyl) 1,2-cyclohexane-
dicarboxylate, diisononyl 1,2-cyclohexanedicarboxylate,
di(3,5,5-trimethylhexyl) 1,2-cyclohexanedicarboxylate,
di(2,6-dimethyl-4-heptyl) 1,2-cyclohexanedicarboxylate,
10 diisodecyl 1,2-cyclohexanedicarboxylate, diisoundecyl 1,2-
cyclohexanedicarboxylate, di(n-propyl) 4-cyclohexene-1,2-
dicarboxylate, di(n-butyl) 4-cyclohexene-1,2-dicarboxylate,
di(n-pentyl) 4-cyclohexene-1,2-dicarboxylate, di(n-hexyl)
4-cyclohexene-1,2-dicarboxylate, di(n-heptyl) 4-
15 cyclohexene-1,2-dicarboxylate, di(n-octyl) 4-cyclohexene-
1,2-dicarboxylate, di(n-nonyl) 4-cyclohexene-1,2-
dicarboxylate, di(n-decyl) 4-cyclohexene-1,2-dicarboxylate,
di(n-undecyl) 4-cyclohexene-1,2-dicarboxylate, diisopropyl
4-cyclohexene-1,2-dicarboxylate, diisopropyl 4-
20 cyclohexene-1,2-dicarboxylate, diisobutyl 4-cyclohexene-
1,2-dicarboxylate, di(sec-butyl) 4-cyclohexene-1,2-
dicarboxylate, dicyclohexyl 4-cyclohexene-1,2-
dicarboxylate, diisoheptyl 4-cyclohexene-1,2-dicarboxylate,
di(2-ethylhexyl) 4-cyclohexene-1,2-dicarboxylate,
25 diisononyl 4-cyclohexene-1,2-dicarboxylate, di(3,5,5-

trimethylhexyl) 4-cyclohexene-1,2-dicarboxylate, di(2,6-dimethyl-4-heptyl) 4-cyclohexene-1,2-dicarboxylate, diisodecyl 4-cyclohexene-1,2-dicarboxylate, dioundecyl 4-cyclohexene-1,2-dicarboxylate and the like.

5 On the other hand, the alicyclic dicarboxylic acid esters include those having R^4 and R^5 which are different from each other, such as esters of the formula (4) in which R^4 and R^5 are different and represent C_3 to C_{11} straight- or branched-chain alkyl. Specific examples
10 thereof include (isopropyl) (2-ethylhexyl) 1,2-cyclohexanedicarboxylate, (isopropyl) (isononyl) 1,2-cyclohexanedicarboxylate, (isopropyl) (3,5,5-trimethylhexyl) 1,2-cyclohexanedicarboxylate, (isobutyl) (2-ethylhexyl) 1,2-cyclohexanedicarboxylate, (isobutyl) (isononyl) 1,2-cyclohexanedicarboxylate, (isobutyl) (3,5,5-trimethylhexyl) 1,2-cyclohexanedicarboxylate, (isopropyl) (2-ethylhexyl) 4-cyclohexene-1,2-dicarboxylate, (isopropyl) (isononyl) 4-cyclohexene-1,2-dicarboxylate, (isopropyl) (3,5,5-trimethylhexyl) 4-cyclohexene-1,2-dicarboxylate, (isobutyl) (2-ethylhexyl) 4-cyclohexene-1,2-dicarboxylate, (isobutyl) (isononyl) 4-cyclohexene-1,2-dicarboxylate, and (isobutyl) (3,5,5-trimethylhexyl) 4-cyclohexene-1,2-dicarboxylate.

 More desirable examples of alicyclic dicarboxylic
25 acid esters are alicyclic dicarboxylic acid esters of the

formula (4) in which -COOR^4 and -COOR^5 are present on the adjacent carbons of the cyclohexane ring or cyclohexene ring, R^4 and R^5 are each independently a C_3 to C_9 straight- or branched-chain alkyl, and X is hydrogen. Examples

5 thereof include di(n-propyl) 1,2-cyclohexanedicarboxylate, di(n-butyl) 1,2-cyclohexanedicarboxylate, di(n-pentyl) 1,2-cyclohexanedicarboxylate, di(n-hexyl) 1,2-cyclohexanedicarboxylate, di(n-heptyl) 1,2-cyclohexanedicarboxylate, di(n-octyl) 1,2-cyclohexanedicarboxylate, di(n-nonyl) 1,2-

10 cyclohexanedicarboxylate, diisopropyl 1,2-cyclohexanedicarboxylate, diisobutyl 1,2-cyclohexanedicarboxylate, di(sec-butyl) 1,2-cyclohexanedicarboxylate, dicyclohexyl 1,2-cyclohexanedicarboxylate, diisoheptyl 1,2-cyclohexanedicarboxylate, di(2-ethylhexyl) 1,2-cyclohexane-

15 dicarboxylate, diisononyl 1,2-cyclohexanedicarboxylate, di(3,5,5-trimethylhexyl) 1,2-cyclohexanedicarboxylate, di(2,6-dimethyl-4-heptyl) 1,2-cyclohexanedicarboxylate, di(n-propyl) 4-cyclohexene-1,2-dicarboxylate, di(n-butyl) 4-cyclohexene-1,2-dicarboxylate, di(n-pentyl) 4-

20 cyclohexene-1,2-dicarboxylate, di(n-hexyl) 4-cyclohexene-1,2-dicarboxylate, di(n-heptyl) 4-cyclohexene-1,2-dicarboxylate, di(n-octyl) 4-cyclohexene-1,2-dicarboxylate, di(n-nonyl) 4-cyclohexene-1,2-dicarboxylate, diisopropyl 4-cyclohexene-1,2-dicarboxylate, diisobutyl 4-cyclohexene-

25 1,2-dicarboxylate, di(sec-butyl) 4-cyclohexene-1,2-

dicarboxylate, dicyclohexyl 4-cyclohexene-1,2-
dicarboxylate, diisohexyl 4-cyclohexene-1,2-dicarboxylate,
di(2-ethylhexyl) 4-cyclohexene-1,2-dicarboxylate,
diisononyl 4-cyclohexene-1,2-dicarboxylate, di(3,5,5-
5 trimethylhexyl) 4-cyclohexene-1,2-dicarboxylate, di(2,6-
dimethyl-4-heptyl) 4-cyclohexene-1,2-dicarboxylate,
(isobutyl) (2-ethylhexyl) 1,2-cyclohexanedicarboxylate,
(isobutyl) (isononyl) 1,2-cyclohexanedicarboxylate,
(isobutyl) (3,5,5-trimethylhexyl) 1,2-cyclohexane-
10 dicarboxylate, (isobutyl) (2-ethylhexyl) 4-cyclohexene-
1,2-dicarboxylate, (isobutyl) (isononyl) 4-cyclohexene-
1,2-dicarboxylate, and (isobutyl) (3,5,5-trimethylhexyl)
4-cyclohexene-1,2-dicarboxylate.

The alicyclic dicarboxylic acid esters represented
15 by the formula (4) include isomers (cis and trans) with
respect to the ester groups, and both isomers can be used
as a lubricating oil for refrigerators.

(b2-5) Examples of fatty acid polyol esters include
full esters of a C_2 to C_{10} aliphatic polyhydric alcohol
20 having 2 to 6 hydroxyl groups, and preferably C_5 to C_{10}
aliphatic polyhydric alcohols having 2 to 6 hydroxyl
groups, such as neopentyl glycol, trimethylol propane,
pentaerythritol, ditrimethylolpropane, dipentaerythritol,
or a similar neopentyl polyol, and a C_3 to C_{22} straight- or
25 branched-chain saturated fatty acid.

Specific examples of C₃ to C₂₂ straight- or branched-chain saturated fatty acids serving as a constituent of such polyol esters include propionic acid, n-butanoic acid, isobutanoic acid, n-pentanoic acid, isopentanoic acid, n-
5 hexanoic acid, n-heptanoic acid, isoheptanoic acid, n-octanoic acid, 2-ethylhexanoic acid, 3,5,5-trimethylhexanoic acid, n-nonanoic acid, n-decanoic acid, and n-undecanoic acid. Such monovalent fatty acids may be used alone or in mixtures of two or more.

10 Specific examples of fatty acid polyol esters include neopentyl glycol fatty acid diesters, such as neopentyl glycol (n-butanoate), neopentyl glycol (isobutanoate), neopentyl glycol (n-pentanoate), neopentyl glycol (isopentanoate), neopentyl glycol (n-pentanoate),
15 neopentyl glycol (isoheptanoate), neopentyl glycol (n-octanoate), neopentyl glycol (2-ethylhexanoate), neopentyl glycol (3,5,5-trimethylhexanoate), neopentyl glycol (n-nonanoate), neopentyl glycol (n-butanoate/2-ethylhexanoate) mixed ester, neopentyl glycol
20 (isobutanoate/2-ethylhexanoate) mixed ester, neopentyl glycol (n-pentanoate/2-ethylhexanoate) mixed ester, neopentyl glycol (isopentanoate/2-ethylhexanoate) mixed ester, neopentyl glycol (n-heptanoate/2-ethylhexanoate) mixed ester, neopentyl glycol (isoheptanoate/2-
25 ethylhexanoate) mixed ester, neopentyl glycol (n-

octanoate/2-ethylhexanoate) mixed ester, neopentyl glycol
(2-ethylhexanoate/3,5,5-trimethylhexanoate) mixed ester,
and neopentyl glycol (2-ethylhexanoate/n-nonanoate) mixed
ester. Examples of trimethylol propane fatty acid
5 triesters include trimethylol propane (n-butanoate),
trimethylol propane (isobutanoate), trimethylol propane
(n-pentanoate), trimethylol propane (isopentanoate),
trimethylol propane (n-heptanoate), trimethylol propane
(isoheptanoate), trimethylol propane (n-octanoate),
10 trimethylol propane (2-ethylhexanoate), trimethylol
propane (3,5,5-trimethylhexanoate), trimethylol propane
(n-nonanoate), trimethylol propane (n-butanoate/2-
ethylhexanoate) mixed esters, trimethylol propane
(isobutanoate/2-ethylhexanoate) mixed esters, trimethylol
15 propane (n-pentanoate/2-ethylhexanoate) mixed esters,
trimethylol propane (isopentanoate/2-ethylhexanoate) mixed
esters, trimethylol propane (n-heptanoate/2-
ethylhexanoate) mixed esters, trimethylol propane
(isoheptanoate/2-ethylhexanoate) mixed esters, trimethylol
20 propane (n-octanoate/2-ethylhexanoate) mixed esters,
trimethylol propane (2-ethylhexanoate/3,5,5-
trimethylhexanoate) mixed esters, and trimethylol propane
(2-ethylhexanoate/n-nonanoate) mixed esters. Examples of
pentaerythritol fatty acid pentaesters include
25 pentaerythritol (n-butanoate), pentaerythritol

(isobutanoate), pentaerythritol (n-pentanoate),
pentaerythritol (isopentanoate), pentaerythritol (n-
heptanoate), pentaerythritol (isoheptanoate),
pentaerythritol (n-octanoate), pentaerythritol (2-
5 ethylhexanoate), pentaerythritol (3,5,5-
trimethylhexanoate), pentaerythritol (n-nonanoate),
pentaerythritol (n-butanoate/2-ethylhexanoate) mixed
esters, pentaerythritol (isobutanoate/2-ethylhexanoate)
mixed esters, pentaerythritol (n-pentanoate/2-
10 ethylhexanoate) mixed esters, pentaerythritol
(isopentanoate/2-ethylhexanoate) mixed esters,
pentaerythritol (n-heptanoate/2-ethylhexanoate) mixed
esters, pentaerythritol (isoheptanoate/2-ethylhexanoate)
mixed esters, pentaerythritol (n-octanoate/2-
15 ethylhexanoate) mixed esters, pentaerythritol (2-
ethylhexanoate/3,5,5-trimethylhexanoate) mixed esters, and
pentaerythritol (2-ethylhexanoate/n-nonanoate) mixed
esters.

Among them, branched fatty acid esters are preferred.

20 Specific examples include neopentyl glycol (isobutanoate),
neopentyl glycol (isopentanoate), neopentyl glycol
(isoheptanoate), neopentyl glycol (2-ethylhexanoate),
neopentyl glycol (3,5,5-trimethylhexanoate), neopentyl
glycol (isobutanoate/2-ethylhexanoate) mixed ester,
25 neopentyl glycol (isopentanoate/2-ethylhexanoate) mixed

ester, neopentyl glycol (isoheptanoate/2-ethylhexanoate)
mixed ester, neopentyl glycol (2-ethylhexanoate/3,5,5-
trimethylhexanoate) mixed ester, trimethylol propane
(isobutanoate), trimethylol propane (isopentanoate),
5 trimethylol propane (isoheptanoate), trimethylol propane
(2-ethylhexanoate), trimethylol propane (3,5,5-trimethyl-
hexanoate), trimethylol propane (isobutanoate/2-
ethylhexanoate) mixed esters, trimethylol propane
(isopentanoate/2-ethylhexanoate) mixed esters, trimethylol
10 propane (isoheptanoate/2-ethylhexanoate) mixed esters,
trimethylol propane (2-ethylhexanoate/3,5,5-trimethyl-
hexanoate) mixed esters, pentaerythritol (isobutanoate),
pentaerythritol (isopentanoate), pentaerythritol
(isoheptanoate), pentaerythritol (2-ethylhexanoate),
15 pentaerythritol (3,5,5-trimethylhexanoate),
pentaerythritol (isobutanoate/2-ethylhexanoate) mixed
esters, pentaerythritol (isopentanoate/2-ethylhexanoate)
mixed esters, pentaerythritol (isoheptanoate/2-
ethylhexanoate) mixed esters, and pentaerythritol (2-
20 ethylhexanoate/3,5,5-trimethylhexanoate) mixed esters.

(b2-6) Examples of other esters include carboxylic
acid esters such as esters of dimer acid or a hydrogenate
thereof (saturated acids) and C₃ to C₂₂ straight- or a
branched-chain saturated or unsaturated aliphatic alcohol.

25 (b3) Polyalkylene glycols

Examples of polyalkylene glycols include a ring-opening polymerization product (adduct) of C_1 to C_{18} alcohol and C_2 to C_4 straight- or branched-chain alkylene oxide. Examples of alkylene oxides include ethylene oxide, propylene oxide, and butylene oxide. Polymers (adducts) obtained by using one of these, or copolymers (co-adducts) obtained by using a mixture of two or more can be used. Also usable are those wherein hydroxyl group or groups at one or both ends has/have been etherified or esterified.

10 The kinematic viscosity of the polymer should be 5.0 to 1000 mm^2/s (40°C), and preferably 5.0 to 500 mm^2/s (40°C).

(b4) Polyvinyl ethers

Examples of polyvinyl ethers include compounds obtained by the polymerization of vinyl ether monomers.

15 Examples of such monomers include methyl vinyl ether, ethyl vinyl ether, isopropyl vinyl ether, n-butyl vinyl ether, isobutyl vinyl ether, sec-butyl vinyl ether, tert-butyl vinyl ether, n-pentyl vinyl ether, n-hexyl vinyl ether, 2-methoxyethyl vinyl ether, and 2-ethoxyethyl vinyl

20 ether. Polymers obtained by using one of these, or copolymers obtained by using two or more can be used. The kinematic viscosity should be 5.0 to 1000 mm^2/s (40°C), and preferably 5.0 to 800 mm^2/s (40°C).

(b5) Polyphenyl ethers

25 Examples of polyphenyl ethers include compounds with

a structure in which two or more aromatic rings are linked by ether bonds or thioether bonds at the meta positions. Specific examples include bis(m-phenoxyphenyl) ether, m-bis(m-phenoxyphenoxy)benzene, and the corresponding

5 thioethers (so-called C-ethers) wherein one or more oxygen atoms has/have been replaced by one or more sulfur atoms.

(b6) Alkyl phenyl ethers

Examples of alkyl phenyl ether include compounds in which polyphenyl ethers are substituted with one or more

10 C₆ to C₁₈ straight- or branched-chain alkyls, particularly alkyl diphenyl ethers substituted with one or more alkyl groups.

(b7) Silicone oils

Examples of silicone oils include dimethyl silicone

15 and methyl phenyl silicone, as well as long-chain alkyl silicone, fluorosilicone and similar modified silicones.

Preferred combination base oils

Among the aforementioned combination base oils, the

20 following are recommended for their hydrolytic stability and practical use: (i) fatty acid polyol esters (b2-5), (ii) phthalic acid esters (b2-3), (iii) alicyclic dicarboxylic acid esters (b2-4), (iv) polyvinyl ethers (b4), (v) hydrocarbon oils (b1), and (vi)

25 polyalkylene glycol (b3). Particularly recommended are

(i) fatty acid polyol esters (b2-5), (iii) alicyclic dicarboxylic acid esters (b2-4), and (iv) polyvinyl ethers (b4).

The aforementioned combination base oils have a high viscosity on their own and are sometimes not usable for refrigerator oils, depending on the type of refrigerator. However, the use thereof in combination with the aliphatic saturated branched-chain carboxylic acid monoalkyl ester of the formula (1) of the present invention allows the viscosity of the refrigerator oil to be readily adjusted in order to provide refrigerator oils of varying viscosity.

When a combination of the present ester and the aforementioned combination base oil is used as the refrigerator oil, it is preferable that the refrigerator oil that is obtained has a kinematic viscosity (at 40°C) of 4.0 to 36 mm²/s.

Also it is recommended that the lubricating oil for refrigerators containing the present ester has a water content not greater than 100 ppm, and preferably not greater than 50 ppm. When the water content is not greater than 100 ppm, the lubricating oil is excellent in hydrolytic stability, electrical insulating properties, and thermal stability.

Additives

The lubricating oil for refrigerators according to the present invention can contain at least one additive as
5 needed to improve the base oil performance, such as antioxidants, metal deactivators, antiwear agents, defoaming agents, and hydrolysis depressants. Such additives are not limited, provided that the intended effects are achieved. The following are specific examples.

10 Examples of antioxidants include 2,6-di-tert-butyl-p-cresol, 4,4'-methylenebis-2,6-di-tert-butylphenol or other such phenol-based compounds, N-phenyl- α -naphthylamine, p,p'-dioctyldiphenylamine or other such amine-based compounds, and phenothiazine or other such
15 sulfur-based compounds. When any antioxidants are used, they may be added, for example, in an amount of 0.01 to 5 wt%, and preferably 0.1 to 2 wt%, relative to the refrigerator oil.

Examples of metal deactivators which can be used
20 include benzotriazole-based and thiadiazole-based compounds. When any such metal deactivators are used, they can be added, for example, in an amount of 0.01 to 0.4 wt% relative to the refrigerator oil.

Examples of antiwear agents generally include
25 tricresyl phosphate, cresyl diphenyl phosphate, alkyl

phenyl phosphate, tributyl phosphate, dibutyl phosphate and other such phosphate esters, tributyl phosphite, dibutyl phosphite, triisopropyl phosphite and other such phosphite esters, as well as amine salts thereof. When

5 any such antiwear agents are used, it is recommended that the amount thereof is 0.01 to 5 wt%, preferably 0.01 to 2 wt%, relative to the refrigerator oil.

Liquid silicones are suitable defoaming agents.

When used, they may be added in an amount of 0.0005 to
10 0.01 wt% relative to the base oil.

In order to further improve thermal and hydrolytic stabilities, the refrigerator oil composition of the present invention can also include at least one epoxy compound selected from the group consisting of:

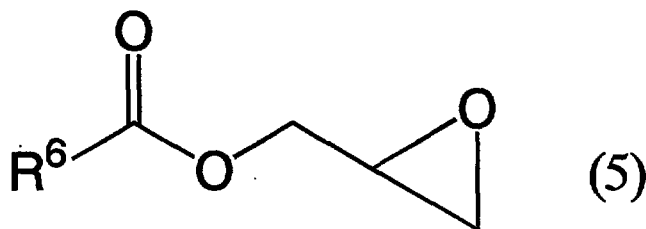
- 15 (1) phenyl glycidyl ether-based epoxy compounds;
(2) alkyl glycidyl ether-based epoxy compounds;
(3) glycidyl ester-based epoxy compounds;
(4) allyloxirane compounds;
(5) alkyloxirane compound;
20 (6) alicyclic epoxy compounds;
(7) epoxidized fatty acid monoesters; and
(8) epoxidized vegetable oils.

(1) Specific examples of phenyl glycidyl ether-based
25 epoxy compounds include phenyl glycidyl ethers and alkyl

phenyl glycidyl ethers. Herein, the alkyl phenyl glycidyl ethers include those with 1 to 3 C₁ to C₁₃ alkyl groups, and preferably those with one C₄ to C₁₀ alkyl group, such as n-butylphenyl glycidyl ether, i-butylphenyl glycidyl ether, sec-butylphenyl glycidyl ether, tert-butylphenyl glycidyl ether, pentylphenyl glycidyl ether, hexylphenyl glycidyl ether, heptylphenyl glycidyl ether, octylphenyl glycidyl ether, nonylphenyl glycidyl ether, and decylphenyl glycidyl ether.

10 (2) Specific examples of alkyl glycidyl ether-based epoxy compounds include decyl glycidyl ether, undecyl glycidyl ether, dodecyl glycidyl ether, tridecyl glycidyl ether, tetradecyl glycidyl ether, 2-ethylhexyl glycidyl ether, neopentyl glycol diglycidyl ether, trimethylol propane triglycidyl ether, pentaerythritol tetraglycidyl ether, 1,6-hexandiol diglycidyl ether, sorbitol polyglycidyl ether, polyalkylene glycol monoglycidyl ether, polyalkylene glycol diglycidyl ether, and the like.

20 (3) Specific examples of glycidyl ester-based epoxy compounds are those represented by the formula (5) below:



wherein R^6 is a C_1 to C_{18} hydrocarbon group.

In the aforementioned formula, R^6 is a C_1 to C_{18} hydrocarbon group. Examples of such hydrocarbon groups include C_1 to C_{18} alkyls, C_2 to C_{18} alkenyls, C_5 to C_7 cycloalkyls, C_6 to C_{18} alkylcycloalkyls, C_6 to C_{10} aryls, C_7 to C_{18} alkylaryls, and C_7 to C_{18} arylalkyls. Of these, C_5 to C_{15} alkyls, C_2 to C_{15} alkenyls, phenyl, and alkylphenyls with C_1 to C_4 alkyls are preferred.

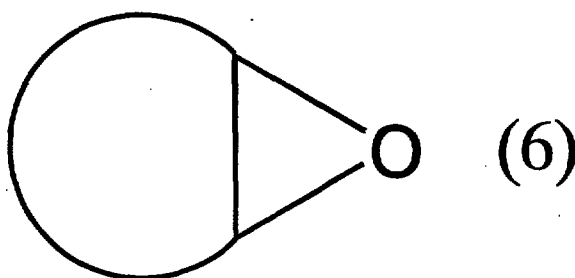
Specific preferred examples among the glycidyl ester-based epoxy compounds include glycidyl 2,2-dimethyloctanoate, glycidyl benzoate, glycidyl tert-butyl benzoate, glycidyl arylate, and glycidyl methacrylate.

(4) Specific examples of allyloxirane compounds include 1,2-epoxystyrene and alkyl-1,2-epoxystyrene.

(5) Specific examples of alkyloxirane compounds include 1,2-epoxybutane, 1,2-epoxypentane, 1,2-epoxyhexane, 1,2-epoxyheptane, 1,2-epoxyoctane, 1,2-epoxynonane, 1,2-epoxydecane, 1,2-epoxyundecane, 1,2-epoxydodecane, 1,2-epoxytridecane, 1,2-epoxytetradecane, 1,2-epoxypentadecane, 1,2-epoxyhexadecane, 1,2-epoxyheptadecane, 1,1,2-

epoxyoctadecane, 2-epoxynonadecane, and 1,2-epoxyicosane.

(6) Examples of alicyclic epoxy compounds include compounds in which the alicyclic ring is formed directly by the carbon atoms forming the epoxy groups, such as
5 compounds represented by the formula (6) below.



Examples of alicyclic epoxy compounds include epoxy compounds having C₃ to C₁₀ alicyclic alkyl. Specific
10 examples include 1,2-epoxycyclohexane, 1,2-epoxycyclopentane, 3,4-epoxycyclohexylmethyl-3,4-epoxycyclohexanecarboxylate, bis(3,4-epoxycyclohexylmethyl) adipate, exo-2,3-epoxynorbornane, bis(3,4-epoxy-6-methylcyclohexylmethyl) adipate, 2-(7-
15 oxabicyclo[4.1.0]hept-3-yl)-spiro(1,3-dioxane-5,3'-[7]oxabicyclo[4.1.0]heptane, 4-(1'-methylepoxyethyl)-1,2-epoxy-2-methylcyclohexane, and 4-epoxyethyl-1,2-epoxycyclohexane.

(7) Specific examples of the epoxidized fatty acid
20 monoesters include esters of an epoxidized C₁₂ to C₂₀ fatty

acid and a C₁ to C₈ alcohol, phenol or an alkylphenol. Particularly desirable examples include butyl, hexyl, benzyl, cyclohexyl, methoxyethyl, octyl, phenyl and butylphenyl esters of epoxystearic acid.

- 5 (8) Specific examples of epoxidized vegetable oils include epoxy compounds of soybean oil, linseed oil, and cottonseed oil.

Of these epoxy compounds, preferred for improving thermal and hydrolytic stability include the phenyl
10 glycidyl ether-based epoxy compounds, the glycidyl ester-based epoxy compounds, the alicyclic epoxy compounds, and the epoxidized fatty acid monoesters. The glycidyl ester-based epoxy compounds and the alicyclic epoxy compounds are especially preferred.

- 15 When such epoxy compounds are blended with the refrigerator oil composition of the present invention, the amount is not particularly limited, but it is usually desirable that the epoxy compound is blended in such an amount that the content thereof will be 0.1 to 5.0 mass%,
20 more preferably 0.2 to 2.0 mass%, based on the total amount of the refrigerator oil composition (relative to the total amount of the base oil and all the additives added).

- 25 The aforementioned phosphorus compounds and epoxy compounds can, of course, be used in combinations of two

or more.

Method for producing lubricating oil for refrigerators
according to the present invention

5

The lubricating oil for refrigerators according to the present invention can be produced using the esters represented by the formula (1) above on their own or by mixing such esters with the aforementioned combination
10 base oils, and if desired, uniformly mixing the aforementioned additives.

The process for producing the aforementioned lubricating oil for refrigerators may include dehydration treatment of the lubricating oil. Such a dehydration
15 treatment may be performed at any point before or after the combination base oil is mixed, or at any point before or after the additives are dissolved. Dehydration can be carried out, for example, by treating the lubricating oil for 0.1 to 10 hours at ordinary or reduced pressure,
20 preferably at a reduced pressure of 0.13 MPa to 66.6 MPa, at a temperature ranging from room temperature to 150°C.

The lubricating oils for refrigerators having a water content of not more than 100 ppm, preferably not more than 50 ppm, are preferred, because they are excellent
25 in hydrolytic stability, electrical insulating properties,

and thermal stability.

Refrigerators in which the lubricating oil for
refrigerators according to the present invention is used

5

The lubricating oil for refrigerators according to the present invention can be used as lubricating oil for refrigerators using various types of refrigerants.

Examples of such refrigerants of these refrigerators
10 include hydrocarbon refrigerants, halogen-containing hydrocarbon refrigerants, perfluoroethers and similar fluoroether refrigerants, dimethyl ether and similar fluorine-free ether refrigerants, carbon dioxide, ammonia, or the like, as well as mixtures thereof.

15 Preferred examples of hydrocarbon refrigerants are those which is in the state of gas at a temperature of 25°C and at a pressure of 1 atm. Specific examples include C₁ to C₅, and preferably C₁ to C₄, alkanes, cycloalkanes, alkenes and other such saturated or
20 unsaturated hydrocarbons, alkyl ethers and other such ether bond-containing hydrocarbons, and mixtures thereof. More specific examples include methane, ethylene, ethane, propylene, propane, cyclopropane, butane, isobutane, cyclobutane, methylcyclopropane, or mixtures of two or
25 more of these.

Examples of halogen-containing hydrocarbon refrigerants include hydrofluorocarbon, hydrochlorofluorocarbon and other such fluorine-containing hydrocarbon refrigerants, hydrochlorocarbons and similar fluorine-free hydrocarbon refrigerants. These refrigerants may be used either singly or in combination.

Among the above refrigerants, at least one member selected from the group consisting of hydrocarbon refrigerants and halogen-containing hydrocarbons is preferred. For protecting the ozone layer, hydrocarbon refrigerants or hydrofluorocarbon refrigerants are recommended, among which C_1 to C_3 , particularly C_1 to C_2 , hydrofluorocarbon refrigerants are most preferred.

Examples of hydrofluorocarbon refrigerants include C_1 to C_3 , preferably C_1 to C_2 , hydrofluorocarbons. Specific examples include difluoromethane (HFC-32), trifluoromethane (HFC-23), pentafluoroethane (HFC-125), 1,1,2,2-tetrafluoroethane (HFC-134), 1,1,1,2-tetrafluoroethane (HFC-134a), 1,1,1-trifluoroethane (HFC-143a), 1,1-difluoroethane (HFC-152a) and other such HFC refrigerants, or mixtures of two or more of these. Such refrigerants may be selected as desired according to the intended application or required performance. Preferred examples include HFC-32 alone; HFC-23 alone; HFC-134a alone; HFC-125 alone; HFC-134a/HFC-32 = 60 to 80 mass%/40

to 20 mass% mixtures; HFC-32/HFC-125 = 40 to 70 mass%/60
to 30 mass% mixtures; HFC-125/HFC-143a = 40 to 60 mass%/60
to 40 mass% mixtures; HFC-134a/HFC-32/HFC-125 = 60
mass%/30mass%/10 mass% mixture; HFC-134a/HFC-32/HFC-125 =
5 40 to 70 mass%/15 to 35 mass%/5 to 40 mass% mixtures; and
HFC-125/HFC-134a/HFC-143a = 35 to 55 mass%/1 to 15
mass%/40 to 60 mass% mixtures. More specifically, there
is mentioned HFC-134a/HFC-32 = 70/30 mass% mixture; HFC-
32/HFC-125 = 60/40 mass% mixture; HFC-32/HFC-125 = 50/50
10 mass% mixture (R410A); HFC-32/HFC-125 = 45/55 mass%
mixture (R410B); HFC-125/HFC-143a = 50/50 mass% mixture
(R507C); HFC-32/HFC-125/HFC-134a = 30/10/60 mass% mixture;
HFC-32/HFC-125/HFC-134a = 23/25/52 mass% mixture (R407C);
HFC-32/HFC-125/HFC-134a = 25/15/60 mass% mixture (R407E);
15 and HFC-125/HFC-134a/HFC-143a = 44/4/52 mass% mixture
(R404A).

HFC-134a, R404A, R407C, R407E and R410A are
particularly recommended.

Examples of refrigerators in which the refrigerator
20 oil of the present invention is usable include automotive
air conditioners, room air conditioners, air conditioners
for packages, dehumidifiers, refrigerator freezers,
automatic vending machines, refrigerated display cases,
warehouse refrigerators, and cooling units in chemical
25 plants. Particularly preferred are refrigerators using at

least one member selected from the group consisting of the
aforementioned hydrocarbon refrigerants and halogen-
containing hydrocarbon refrigerants. More preferably,
refrigerators using the aforementioned hydrofluorocarbon
5 refrigerants are recommended.

Working fluid composition for refrigerators

The present invention is also intended to provide a
10 working fluid composition for refrigerators comprising the
lubricating oil for refrigerators according to the present
invention and a refrigerant as described above.

That is, researches by the inventors revealed that
the lubricating oil for refrigerators according to the
15 present invention, particularly the ester represented by
the formula (1), has good miscibility with refrigerants,
such as the aforementioned hydrocarbon refrigerants,
halogen-containing hydrocarbon refrigerants,
perfluoroether and other fluoroether refrigerants,
20 dimethyl ether and other such fluorine-free ether
refrigerants, carbon dioxide, ammonia, and similar
refrigerants, especially at least one such refrigerant
selected from the group consisting of the aforementioned
hydrocarbon and halogen-containing hydrocarbon
25 refrigerants (especially hydrofluorocarbon refrigerants).

It was thus discovered that a mixture of (I) the lubricating oil for refrigerators according to the present invention and (II) a refrigerant, such as the aforementioned hydrocarbon refrigerant, halogen-containing hydrocarbon refrigerant, perfluoroether and other 5 fluoroether refrigerant, dimethyl ether and other such fluorine-free ether refrigerant, carbon dioxide, ammonia, and similar refrigerant, especially at least one refrigerant selected from the group consisting of the 10 aforementioned hydrocarbon and halogen-containing hydrocarbon refrigerants (especially hydrofluorocarbon refrigerants) function as a working fluid for refrigerators.

That is, the present invention provides a working 15 fluid composition comprising:

- (I) an aliphatic saturated branched-chain carboxylic acid monoalkyl ester represented by the formula (1), and
- (II) a refrigerant, such as the aforementioned 20 hydrocarbon refrigerant, halogen-containing hydrocarbon refrigerant, perfluoroether and other fluoroether refrigerant, dimethyl ether and other such fluorine-free ether refrigerant, carbon dioxide, ammonia, and similar refrigerant; particularly at least one 25 refrigerant selected from the group consisting of the

aforementioned hydrocarbon and halogen-containing hydrocarbon refrigerants (especially hydrofluorocarbon refrigerants).

The ratio of the aforementioned component (I) to component (II) is generally (I):(II) = 1:99 to 85:15, preferably 2:98 to 80:20, by weight.

The present invention also provides a working fluid composition comprising:

- (I) (a) at least one aliphatic saturated branched-chain carboxylic acid monoalkyl ester represented by the formula (1), and
(b) at least one member selected from the group consisting of (i) the fatty acid polyol esters, (ii) the phthalic acid esters, (iii) the alicyclic dicarboxylic acid esters, (iv) the polyvinyl ethers, (v) the hydrocarbon oils and (vi) the polyalkylene glycols, and
- (II) a refrigerant, such as the aforementioned hydrocarbon refrigerant, halogen-containing hydrocarbon refrigerant, perfluoroether and other fluoroether refrigerant, dimethyl ether and other such fluorine-free ether refrigerant, carbon dioxide, ammonia, and similar refrigerant, particularly at least one refrigerant selected from the group consisting of the aforementioned

hydrocarbon and halogen-containing hydrocarbon refrigerants (especially hydrofluorocarbon refrigerants).

The ratio of components (a) to (b) in component (I) above is, for example, (a):(b) = 0.5:99.5 to 99.5:0.5, preferably (a):(b) = 5:95 to 95:5, more preferably (a):(b) = 15:85 to 85:15, by weight. The ratio of component (I) to component (II) is generally (I):(II) = 1:99 to 85:15, preferably 2:98 to 80:20, by weight.

When such a mixture is used as working fluid, an advantage is that the lubricating oil of the present invention will be miscible over a wide temperature range, from low to elevated temperatures, with refrigerants such as the aforementioned hydrocarbon refrigerants, halogen-containing hydrocarbon refrigerants, perfluoroether and other fluoroether refrigerants, dimethyl ether and other such fluorine-free ether refrigerants, carbon dioxide, ammonia, and similar refrigerants, particularly at least one refrigerant selected from the group consisting of the aforementioned hydrocarbon and halogen-containing hydrocarbon refrigerants (especially hydrofluorocarbon refrigerants). The working fluid is less likely to cause phase separation.

Method for lubricating refrigerators

The lubricating oil of the present invention can be used to lubricate the aforementioned types of
5 refrigerators. Figure 1 is a schematic diagram of such a refrigerator.

The refrigerator is composed primarily of a compressor 1, condenser 2, expansion valve 3, evaporator 4, and tubing 5 connecting the various components. A dryer
10 (not shown) may also be included as needed.

The method for lubricating refrigerators is described below with reference to a refrigerator using a hydrofluorocarbon refrigerant, but the method for lubricating refrigerators using other refrigerants is
15 basically the same.

A working fluid composition comprising the lubricating oil of the present invention and a hydrofluorocarbon refrigerant is introduced into the compressor 1 of the refrigerator. At this point, the
20 working fluid composition may be used in an amount effective for operating the refrigerator. In general, the amount may be determined in accordance with the performance required of the refrigerator, the intended application, and so forth. The amount of the working
25 fluid composition can be suitably selected in accordance

with the knowledge of those having ordinary skill in the art.

The refrigerant is primarily compressed by the compressor 1 and is passed through the tubing 5, and the compressed refrigerant is liquefied by the condenser 2, and is guided through the expansion valve 3 to the evaporator 4, where the refrigerant is vaporized to effect heat exchange through the evaporator 4. At this point, most of the refrigerator oil remains in the compressor 1, but the hydrofluorocarbon refrigerant is present as dissolved in the refrigerator oil. During this refrigerator cycle, a small amount of the refrigerator oil circulates in the device along with the refrigerant.

The compressor 1 illustrated here is a compressor with an open, semi-open, or hermetic compressor, depending on the method by which the compression parts and the motor driving them are assembled. Due to differences in the compression method, there are refrigerators with reciprocal, rotary, scroll or screw compressors, as well as refrigerators with centrifugal compressors, but any of them can be used.

The lubricating oil of the present invention results in better operation of the compressor 1 because of good miscibility between the lubricating oil of the present invention and the hydrofluorocarbon in the compressor 1.

Since the working fluid composition is less likely to cause phase separation, the lubricating oil is less likely to localize in areas other than the compressor 1.

Furthermore, the lubricating oil of the present invention
5 has good hydrolytic stability and causes a small increase in the total acid number, so that less corrosion is encountered in the compressor 1, condenser 2, expansion valve 3, evaporator 4, or tubing 5 connecting such parts. At the same time, the lubricating oil of the present
10 invention has low viscosity, leading to less energy loss as a result of friction during the operation of the refrigerator. The working fluid composition also has excellent insulating properties, and thus serves as an insulating oil to prevent leakage in the compressor motors
15 of refrigerators having hermetic or semi-hermetic compressors, making the oil of the invention especially desirable for use in refrigerators with hermetic compressors.

Examples of compressor 1 include (1) high pressure
20 container compressors, in which a hermetic container holding the refrigerator oil houses a motor comprising a rotor and stator, a rotating shaft fixed to the rotor, and a compressor device connected to the motor via the rotating shaft and a high pressure refrigerant gas
25 discharged from the compressor device is held in the

hermetic container; and (2) low pressure container compressors, in which a hermetic container holding the refrigerator oil houses a motor comprising a rotor and stator, a rotating shaft fixed to the rotor, and a
5 compressor device connected to the motor via the rotating shaft and a high pressure refrigerant gas discharged from the compressor device is directly discharged out of the hermetic container.

As an insulating film serving as a material in the
10 electrical insulation system of the motor, preferably usable are crystalline plastic films with a glass transition point of 50°C or higher, specifically an insulating film made of at least one member selected from the group consisting of polyethylene terephthalate,
15 polybutylene terephthalate, polyphenylene sulfide, polyether ether ketone, polyethylene naphthalate, polyamideimide and polyimide, or a composite film comprising a film with a low glass transition temperature covered by a resin layer with a higher glass transition
20 temperature, because of lower susceptibility to loss of tensile strength or electrical insulating properties. The magnet wire used in the motor is preferably coated with an enamel having a glass transition temperature of 120°C or higher, such as a single layer of polyester, polyester
25 imide, polyamide or polyamideimide, or an enamel coating

comprising a composite coating composed of a lower layer with a low glass transition temperature and an upper layer with a higher glass transition temperature. Examples of enamel wires coated with such composite coating include
5 those having a polyester imide lower layer and a polyamideimide upper layer (AI/EI), and those having a polyester lower layer and a polyamideimide upper layer (AI/PE).

The dessicant to be used in the dryer is preferably
10 a synthetic zeolite comprising a silicate and aluminate alkali metal complex salt with a pore diameter of not more than 3.3 Å and a carbon dioxide gas absorption capacity of not more than 1.0% at a carbon dioxide partial pressure of 250 mmHg at 25°C. Specific examples include XH-9, XH-10,
15 XH-11, and XH-600, manufactured by Union Showa.

PREFERRED MODE FOR CARRYING OUT THE INVENTION

Examples of preferred embodiments of the
20 refrigerator oil of the present invention are given below:
• refrigerator oil containing (2-ethylhexyl) 2-ethylhexanoate,
• refrigerator oil containing (isononyl) 2-ethylhexanoate,
• refrigerator oil containing (3,5,5-trimethylhexyl) 2-
25 ethylhexanoate,

- refrigerator oil containing (3,5,5-trimethylhexyl) 2-ethylbutanoate,
- refrigerator oil containing (2-ethylhexyl) 2-ethylhexanoate and (n-dodecyl) 2-ethylhexanoate,
- 5 • refrigerator oil containing (2-ethylhexyl) 2-ethylhexanoate and a C₃ to C₁₈ fatty acid triester of trimethylolpropane,
- refrigerator oil containing (2-ethylhexyl) 2-ethylhexanoate and a C₃ to C₁₈ fatty acid tetraester of
- 10 pentaerythritol,
- refrigerator oil containing (2-ethylhexyl) 2-ethylhexanoate and a di(C₃ to C₁₈ alkyl) ester of 4-cyclohexene-1,2-dicarboxylic acid,
- refrigerator oil containing (2-ethylhexyl) 2-
- 15 ethylhexanoate and a di(C₃ to C₁₈ alkyl) ester of 1,2-cyclohexanedicarboxylic acid,
- refrigerator oil containing (2-ethylhexyl) 2-ethylhexanoate and a di(C₃ to C₁₈ alkyl) ester of phthalic acid,
- 20 • refrigerator oil containing (3,5,5-trimethylhexyl) 2-ethylbutanoate and a di(C₃ to C₁₈ alkyl) ester of 1,2-cyclohexanedicarboxylic acid.

Examples of preferred embodiments of the working fluid composition of the present invention are given below.

- 25 The proportions of the components of the working fluid

composition are the same as in the section entitled

"Working fluid composition for refrigerators."

- working fluid compositions containing (2-ethylhexyl) 2-ethylhexanoate and HFC-134a,
- 5 • working fluid compositions containing (isononyl) 2-ethylhexanoate and HFC-134a,
- working fluid compositions containing (3,5,5-trimethylhexyl) 2-ethylhexanoate and HFC-134a,
- working fluid compositions containing (3,5,5-trimethylhexyl) 2-ethylbutanoate and HFC-134a,
- 10 • working fluid compositions containing (2-ethylhexyl) 2-ethylhexanoate, (n-dodecyl) 2-ethylhexanoate and HFC-134a,
- working fluid compositions containing (2-ethylhexyl) 2-ethylhexanoate, a trimethylolpropane C_3 to C_{18} fatty acid
- 15 triester and HFC-134a,
- working fluid compositions containing (2-ethylhexyl) 2-ethylhexanoate, a pentaerythritol C_3 to C_{18} fatty acid tetraester and HFC-134a,
- working fluid compositions containing (2-ethylhexyl) 2-
- 20 ethylhexanoate, a dialkyl (C_3 to C_{18}) ester of 4-cyclohexene-1,2-dicarboxylic acid and HFC-134a,
- working fluid compositions containing (2-ethylhexyl) 2-ethylhexanoate, a dialkyl (C_3 to C_{18}) ester of 1,2-cyclohexanedicarboxylic acid and HFC-134a,
- 25 • working fluid compositions containing (2-ethylhexyl) 2-

ethylhexanoate, a dialkyl (C_3 to C_{18}) ester of phthalic acid and HFC-134a,

- working fluid compositions containing (3,5,5-trimethylhexyl) 2-ethylbutanoate, a dialkyl (C_3 to C_{18}) ester of 1,2-cyclohexanedicarboxylic acid and HFC-134a, and
- working fluid compositions containing (3,5,5-trimethylhexyl) 2-ethylhexanoate, a dialkyl (C_3 to C_{18}) ester of phthalic acid and HFC-134a,

10

EFFECTS OF THE INVENTION

The use of the present ester for refrigerator oil achieves the following advantages.

- 15 1) The present ester has better hydrolytic stability.
- 2) The present ester has better miscibility with hydrofluorocarbon refrigerants.
- 3) The present ester has a higher volume resistivity, and better electrical insulating properties.
- 20 4) The present ester has a far lower viscosity than conventionally used refrigerator oils, and can minimize energy loss.
- 5) Compositions obtained by mixing the present ester with a conventional refrigerator oil can be adjusted to a
- 25 desired viscosity by adjusting the proportions thereof.

6) A lubricating oil for refrigerators comprising the aliphatic saturated branched-chain carboxylic acid monoalkyl ester of the present invention has excellent hydrolytic stability and entails less increase in the total acid number under stringent conditions, causing less change in metal surfaces.

7) The lubricating oil of the present invention is excellent in electrical insulating properties and miscibility with fluorocarbons at low temperatures, and can be used as a lubricating oil for compressors in which the refrigerant is a hydrofluorocarbon containing no chlorine atom within the molecule, and is tremendously beneficial for industrial purposes.

8) The present ester has a low viscosity and can thus be used with other lubricating oils, thereby enabling to provide refrigerator lubricating oils of varying viscosity grade.

EXAMPLES

20

The present invention will be described in further detail below with reference to examples and comparative examples. The properties of the lubricating oils in the examples and comparative examples were assessed in the following manner.

25

Total acid number

Measured in accordance with JIS-K-2501.

Kinematic viscosity

Measured in accordance with JIS-K-2283 using an

- 5 Ubbelohde viscometer.

Water content

Measured using a Karl Fischer's moisture meter (MKC-510, manufactured by Kyoto Denshi) in accordance with JIS-K-2275.

- 10 Volume resistivity

Measured at 25°C in accordance with JIS-C-2101.

Two-layer separation temperature

- In accordance with JIS-K-2211, a sample oil was added to a hydrofluorocarbon refrigerant HFC-134a so as to adjust the proportion of the sample oil to 20 wt%, and the temperature at which the mixture separated into two phases was determined between -50 and 38°C. The lower the temperature, the better the miscibility between the sample oil and hydrofluorocarbon.
- 15

- 20 Hydrolytic stability test

- An iron wire, a copper wire and an aluminum wire, each having a length of 4 cm, were placed in glass test tube having a length of 30 cm and an inside diameter of 6.6 mm, and 2.0 g of sample ester and 0.2 g of distilled water were placed therein. The test tubes were sealed
- 25

while being degassed with an aspirator, and the test tube was heated for 20 or 40 hours at 175°C in an oven. The sample ester was then taken out to measure the total acid number. The surfaces of the metal wires were visually
5 inspected and evaluation was made according to the following three-grade criteria. The state of the ester after the test was also inspected.

O: no change

Δ: some discoloration

10 x: turned black or deep brown

Manufacturing Example 1.

144 g (1 mol) of 2-ethylhexanoic acid and 143 g (1.1 mol) of 2-ethylhexanol were fed to a four-necked flask equipped with a stirrer, a thermometer and a water
15 separator having a condenser tube, and the mixture was heated to 200°C under reduced pressure in the presence of a tetraisopropyl titanate catalyst. Esterification reaction was carried out for about 9 hours while removing the formed water by the water separator. After the
20 reaction, the excess 2-ethylhexanol was distilled off, and the product was neutralized with excess sodium hydroxide and then washed with water to neutrality. The product was then treated with activated carbon (0.3 wt.% relative to the starting materials fed) at 90°C, and was then filtered,
25 giving 246 g of (2-ethylhexyl) 2-ethylhexanoate, which was

then dehydrated for 5 hours at a temperature of 100°C and a reduced pressure of 13.3 MPa. The resulting ester had a total acid number of 0.01 mgKOH/g, a water content of 11 ppm, a kinematic viscosity of 2.7 mm²/s (40°C) and 1.1 mm²/s (100°C), a volume resistivity of $3.0 \times 10^{13} \Omega \cdot \text{cm}$, and a two-layer separation temperature of -24°C.

Manufacturing Example 2

Following the procedure of Manufacturing Example 1, (3,5,5-trimethylhexyl) 2-ethylbutanoate was obtained from 2-ethylbutanoic acid and 3,5,5-trimethylhexanol. The resulting ester had a total acid number of 0.01 mgKOH/g, a water content of 18 ppm, a kinematic viscosity of 2.6 mm²/s (40°C) and 1.1 mm²/s (100°C), a volume resistivity of $2.1 \times 10^{13} \Omega \cdot \text{cm}$, and a two-layer separation temperature of -50°C or below.

Manufacturing Example 3

Isononyl 2-ethylhexanoate was obtained from 2-ethylhexanoic acid and isononanol in the same manner as in Manufacturing Example 1. The resulting ester had a total acid number of 0.01 mgKOH/g, a water content of 25 ppm, a kinematic viscosity of 3.4 mm²/s (40°C) and 1.1 mm²/s (100°C), a volume resistivity of $8.0 \times 10^{13} \Omega \cdot \text{cm}$, and a two-layer separation temperature of -8°C.

Manufacturing Example 4

(3,5,5-Trimethylhexyl) 2-ethylhexanoate was obtained

from 2-ethylhexanoic acid and 3,5,5-trimethylhexanol in the same manner as in Manufacturing Example 1. The resulting ester had a total acid number of 0.01 mgKOH/g, a water content of 16 ppm, a kinematic viscosity of 3.5 mm²/s (40°C) and 1.3 mm²/s (100°C), a volume resistivity of $1.0 \times 10^{14} \Omega \cdot \text{cm}$ and a two-layer separation temperature of -31°C.

Manufacturing Example 5

An ester mixture was obtained, in the same manner as in Example 1, from 2-ethylhexanoic acid and an alcohol mixture of 2-ethylhexanol and n-dodecanol (molar ratio 82:18). The resulting ester had a total acid number of 0.01 mgKOH/g, a water content of 22 ppm, a kinematic viscosity of 3.1 mm²/s (40°C) and 1.2 mm²/s (100°C), a volume resistivity of $7.7 \times 10^{13} \Omega \cdot \text{cm}$, and a two-layer separation temperature of -8°C.

Comparative Manufacturing Example 1

2-Ethylhexyl palmitate was prepared following the procedure of Example 1. The resulting ester had a total acid number of 0.01 mgKOH/g, a water content of 13 ppm, a kinematic viscosity of 8.1 mm²/s (40°C) and 2.7 mm²/s (100°C), and a volume resistivity of $6.2 \times 10^{13} \Omega \cdot \text{cm}$. However, when the ester was subjected to the measurement of the two-layer separation temperature, it was found that the ester was not miscible with the fluorocarbon at the

measurement temperature range of -50 to 38°C.

Comparative Manufacturing Example 2

2-Ethylhexyl ester of coconut fatty acid (C_8 to C_{20} aliphatic straight-chain monocarboxylic acid mixture) was prepared in the same manner as in Example 1. The resulting ester had a total acid number of 0.01 mgKOH/g, a water content of 15 ppm, a kinematic viscosity of 5.3 mm²/s (40°C) and 1.9 mm²/s (100°C), and a volume resistivity of $5.7 \times 10^{13} \Omega \cdot \text{cm}$. However, when the ester was subjected to measurement of the two-layer separation temperature, it was found that the ester was not miscible with the fluorocarbon at the measurement temperature range of -50 to 38°C.

Oil mixtures were prepared using the esters of Manufacturing Examples 1 and 2. The details will be given in Manufacturing Examples 6 through 14.

Manufacturing Example 6

The (2-ethylhexyl) 2-ethylhexanoate obtained in Manufacturing Example 1 and trimethylol propane tri(3,5,5-trimethylhexanoate) prepared in the same manner as in Manufacturing Example 1 were mixed in a weight ratio of 17:83, giving an ester mixture. The mixture was then dehydrated for 5 hours at a temperature of 100°C and a reduced pressure of 13.3 MPa. The resulting ester had a

total acid number of 0.01 mgKOH/g, a water content of 24 ppm, a kinematic viscosity of 15.1 mm²/s (40°C) and 3.1 mm²/s (100°C), a volume resistivity of 8.3×10^{13} Ω·cm, and a two-layer separation temperature of -32°C.

5 Manufacturing Example 7

The (2-ethylhexyl) 2-ethylhexanoate obtained in Manufacturing Example 1, and an ester of pentaerythritol and a branched acid mixture (2-ethylhexanoic acid/3,5,5-trimethylhexanoic acid = 50/50) prepared in the same
10 manner as in Manufacturing Example 1 were mixed in a weight ratio of 33:67, giving an ester mixture. The mixture was then dehydrated for 5 hours at a temperature of 100°C and a reduced pressure of 13.3 MPa. The resulting ester had a total acid number of 0.01 mgKOH/g, a
15 water content of 32 ppm, a kinematic viscosity of 15.3 mm²/s (40°C) and 3.3 mm²/s (100°C), a volume resistivity of 3.3×10^{14} Ω·cm, and a two-layer separation temperature of -19°C.

Manufacturing Example 8

20 The (2-ethylhexyl) 2-ethylhexanoate obtained in Manufacturing Example 1 and diisodecyl phthalate (tradename "Sansocizer DIDP", manufactured by New Japan Chemical Co., Ltd.) were mixed in a weight ratio of 40:60, giving an ester mixture. The mixture was then dehydrated
25 for 5 hours at a temperature of 100°C and a reduced

pressure of 13.3 MPa. The resulting ester had a total acid number of 0.01 mgKOH/g, a water content of 11 ppm, and a kinematic viscosity of 10.2 mm²/s (40°C) and 2.3 mm²/s (100°C).

5 Manufacturing Example 9

The (2-ethylhexyl) 2-ethylhexanoate obtained in Manufacturing Example 1 and di(2-ethylhexyl) 4-cyclohexene-1,2-dicarboxylate prepared in the same manner as in Manufacturing Example 1 were mixed in a weight ratio
10 of 15:85, giving an ester mixture. The mixture was then dehydrated for 5 hours at a temperature of 100°C and a reduced pressure of 13.3 MPa. The resulting ester had a total acid number of 0.01 mgKOH/g, a water content of 19 ppm, a kinematic viscosity of 11.8 mm²/s (40°C) and 2.7
15 mm²/s (100°C), a volume resistivity of $4.7 \times 10^{13} \Omega \cdot \text{cm}$, and a two-layer separation temperature of 6°C.

Manufacturing Example 10

The (2-ethylhexyl) 2-ethylhexanoate obtained in Manufacturing Example 1 and diisononyl 4-cyclohexene-1,2-
20 dicarboxylate prepared in the same manner as in Manufacturing Example 1 were mixed in a weight ratio of 30:70, giving an ester mixture. The mixture was then dehydrated for 5 hours at a temperature of 100°C and a reduced pressure of 13.3 MPa. The resulting ester had a
25 total acid number of 0.01 mgKOH/g, a water content of

15 ppm, and a kinematic viscosity of 9.6 mm²/s (40°C) and 2.5 mm²/s (100°C).

Manufacturing Example 11

The (2-ethylhexyl) 2-ethylhexanoate obtained in Manufacturing Example 1 and di(3,5,5-trimethylhexyl) 4-cyclohexene-1,2-dicarboxylate prepared in the same manner as in Manufacturing Example 1 were mixed in a weight ratio of 40:60, giving an ester mixture. The mixture was then dehydrated for 5 hours at a temperature of 100°C and a reduced pressure of 13.3 MPa. The resulting ester had a total acid number of 0.01 mgKOH/g, a water content of 28 ppm, a kinematic viscosity of 9.1 mm²/s (40°C) and 2.2 mm²/s (100°C), a volume resistivity of $5.6 \times 10^{13} \Omega \cdot \text{cm}$, and a two-layer separation temperature of -12°C.

Manufacturing Example 12

The (3,5,5-trimethylhexyl) 2-ethylbutanoate obtained in Manufacturing Example 2 and diisodecyl 1,2-cyclohexane-dicarboxylate prepared in the same manner as in Manufacturing Example 1 were mixed in a weight ratio of 30:70, giving an ester mixture. The mixture was then dehydrated for 5 hours at a temperature of 100°C and a reduced pressure of 13.3 MPa. The resulting ester had a total acid number of 0.01 mgKOH/g, a water content of 23 ppm, and a kinematic viscosity of 11.5 mm²/s (40°C) and 2.7 mm²/s (100°C).

Manufacturing Example 13

The (3,5,5-trimethylhexyl) 2-ethylhexanoate obtained in Manufacturing Example 2 and diisobutyl phthalate and di(2-ethylhexyl) phthalate prepared in the same manner as
5 in Manufacturing Example 1 were mixed in a weight ratio of 30:35:35, giving an ester mixture. The mixture was then dehydrated for 5 hours at a temperature of 100°C and a reduced pressure of 13.3 MPa. The resulting ester had a total acid number of 0.01 mgKOH/g, a water content of
10 15 ppm, a kinematic viscosity of 10.2 mm²/s (40°C) and 2.3 mm²/s (100°C), a volume resistivity of $4.5 \times 10^{12} \Omega \cdot \text{cm}$, and a two-layer separation temperature of -35°C.

Manufacturing Example 14

The (2-ethylhexyl) 2-ethylhexanoate obtained in
15 Manufacturing Example 1 and diisobutyl 1,2-cyclohexanedicarboxylate and diisononyl 1,2-cyclohexanedicarboxylate prepared in the same manner as in Manufacturing Example 1 were mixed in a weight ratio of 15:50:35, giving an ester mixture. The mixture was then
20 dehydrated for 5 hours at a temperature of 100°C and a reduced pressure of 13.3 MPa. The resulting ester had a total acid number of 0.01 mgKOH/g, a water content of 24 ppm, a kinematic viscosity of 8.9 mm²/s (40°C) and 2.3 mm²/s (100°C), a volume resistivity of $6.2 \times 10^{12} \Omega \cdot \text{cm}$,
25 and a two-layer separation temperature of -25°C.

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Comparative Manufacturing Example 3

Neopentyl glycol di(3,5,5-trimethylhexanoate) was prepared in the same manner as in Manufacturing Example 1.

The resulting ester had a total acid number of

- 5 0.01 mgKOH/g, a water content of 18 ppm, and a kinematic viscosity of 12.9 mm²/s (40°C) and 3.1 mm²/s (100°C).

Comparative Manufacturing Example 4

Neopentyl glycol di(2-ethylhexanoate) and neopentyl glycol di(3,5,5-trimethylhexanoate) were mixed in a weight
10 ratio of 40:60, giving an ester mixture. The mixture was then dehydrated for 5 hours at a temperature of 100°C and a reduced pressure of 13.3 MPa. The resulting ester had a total acid number of 0.01 mgKOH/g, a water content of
17 ppm, and a kinematic viscosity of 10.2 mm²/s (40°C) and
15 2.6 mm²/s (100°C).

Comparative Manufacturing Example 5

Neopentyl glycol di(2-ethylhexanoate) and trimethylol propane tri(2-ethylhexanoate) were mixed in a weight ratio of 35:65, giving an ester mixture. The mixture was then
20 dehydrated for 5 hours at a temperature of 100°C and a reduced pressure of 13.3 MPa. The resulting ester had a total acid number of 0.01 mgKOH/g, a water content of 28 ppm, a kinematic viscosity of 15.6 mm²/s (40°C) and 3.2 mm²/s (100°C), a volume resistivity of $9.5 \times 10^{13} \Omega \cdot \text{cm}$,
25 and a two-layer separation temperature of -50°C or below.

Examples 1 through 14

The hydrolytic stability of the esters obtained in Manufacturing Examples 1 through 14 was evaluated. The
5 results are given in Table 1.

Comparative Examples 1 through 5

The hydrolytic stability of the esters obtained in Comparative Manufacturing Examples 1 through 5 was
10 evaluated. The results are given in Table 2.

Table 1 Hydrolytic stability of Esters of Manufacturing Examples

Example	Ester	After 20 hours					After 40 hours				
		Total acid number [mgKOH/g]	Steel	Copper	Alumi- num	Color of oil	Total acid number [mgKOH/g]	Steel	Copper	Alumi- num	Color of oil
1	Ester of Manufacturing Example 1	0.08	○	Δ	○	White	0.23	○	Δ	○	white
2	Ester of Manufacturing Example 2	0.10	○	Δ	○	White	0.51	Δ	Δ	○	white
3	Ester of Manufacturing Example 3	0.14	○	Δ	○	White	0.58	○	Δ	○	white
4	Ester of Manufacturing Example 4	0.12	○	Δ	○	White	0.47	○	Δ	○	white
5	Ester of Manufacturing Example 5	0.14	○	Δ	○	White	0.38	○	Δ	○	white
6	Ester of Manufacturing Example 6	0.22	○	Δ	○	White	1.26	Δ	Δ	○	white
7	Ester of Manufacturing Example 7	0.28	○	Δ	○	White	0.98	Δ	Δ	○	white
8	Ester of Manufacturing Example 8	0.55	○	Δ	○	White	1.42	Δ	Δ	○	white
9	Ester of Manufacturing Example 9	0.20	○	Δ	○	White	0.59	○	Δ	○	white
10	Ester of Manufacturing Example 10	0.13	○	Δ	○	White	0.60	○	Δ	○	white
11	Ester of Manufacturing Example 11	0.13	○	Δ	○	White	0.55	○	Δ	○	white
12	Ester of Manufacturing Example 12	0.11	○	Δ	○	White	0.62	○	Δ	○	white
13	Ester of Manufacturing Example 13	0.49	○	Δ	○	White	1.36	Δ	Δ	○	white
14	Ester of Manufacturing Example 14	0.12	○	Δ	○	White	0.44	○	Δ	○	white

Table 2 Hydrolytic stability of Esters of Comparative Manufacturing Examples

Compa- rative Example	Ester	After 20 hours					After 40 hours				
		Total acid number [mgKOH/g]	Steel	Copper	Alumi- num	Color of oil	Total acid number [mgKOH/g]	Steel	Copper	Alumi- num	Color of oil
1	Ester of Comparative Manufacturing Example 1	7.25	×	Δ	○	Pale yellow	14.32	×	Δ	○	Pale yellow
2	Ester of Comparative Manufacturing Example 2	5.72	×	Δ	○	Pale yellow	11.54	×	×	○	Pale yellow
3	Ester of Comparative Manufacturing Example 3	6.32	×	Δ	○	Pale yellow	18.22	×	×	○	Pale yellow
4	Ester of Comparative Manufacturing Example 4	3.26	×	Δ	○	Pale yellow	9.42	×	×	○	Pale yellow
5	Ester of Comparative Manufacturing Example 5	0.53	○	Δ	○	white	1.45	Δ	×	○	white

Examples 1 through 5 show that the present esters have good hydrolytic stability, have a small increase in total acid number, and cause substantially no change on metal surfaces. Their refrigerant miscibility and
5 electrical insulating properties are also excellent. They also have an extremely low kinematic viscosity at 40°C, are capable of reducing energy loss during the operation of refrigerators, and are effective for saving energy.

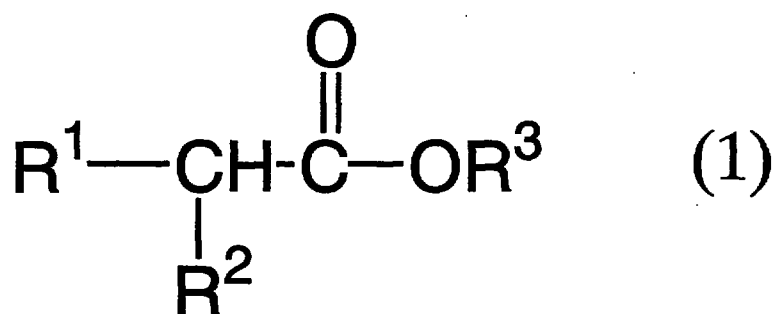
On the other hand, Comparative Examples 1 and 2 show
10 that aliphatic straight-chain monocarboxylic acid esters have poor hydrolytic stability and refrigerant miscibility.

In Examples 6 through 14, the ester was mixed with various other esters. The oil mixtures with a kinematic viscosity at 40°C adjusted to between 10 and 15 mm²/s also
15 have good hydrolytic stability, and therefore the present esters can be used as viscosity regulators to reduce the viscosity.

Comparative Examples 3 through 5, on the other hand, show that branched-chain carboxylic acid esters of
20 neopentyl glycol suffered a large increase in total acid number, with a tendency to cause a greater level of metal corrosion.

CLAIMS

1. A lubricating oil for refrigerators,
characterized in that it comprises at least one aliphatic
5 saturated branched-chain carboxylic acid monoalkyl ester
represented by the formula (1)



- wherein R^1 is C_1 to C_{18} straight-chain alkyl or C_3 to C_{18}
10 branched-chain alkyl; R^2 is hydrogen, C_1 to C_{18} straight-
chain alkyl or C_3 to C_{18} branched-chain alkyl; with the
proviso that the total number of carbon atoms contained in
the alkyls represented by R^1 and R^2 is 2 to 18 and that
when R^2 is hydrogen, R^1 is branched-chain alkyl; and R^3 is
15 C_1 to C_{20} straight-chain alkyl, C_3 to C_{20} branched-chain
alkyl or C_3 to C_{10} cycloalkyl.

2. The lubricating oil according to Claim 1,
wherein the total number of carbon atoms of the alkyls
20 represented by R^1 and R^2 is 2 to 13.

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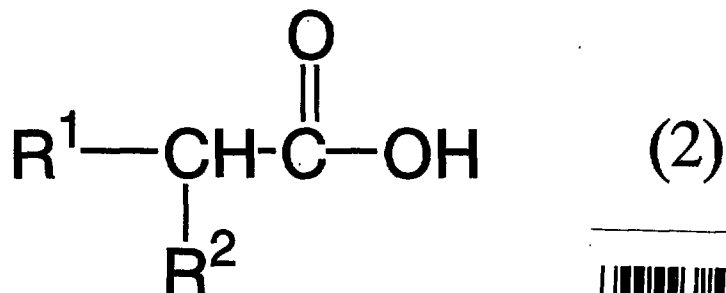
3. The lubricating oil according to Claim 2,
wherein R^3 is C_3 to C_{12} straight- or branched-chain alkyl.

5 4. The lubricating oil according to Claim 1,
wherein R^1 and R^2 are C_1 to C_{17} straight-chain alkyl or C_3
to C_{18} branched-chain alkyl.

5. A lubricating oil according to Claim 1,
10 wherein one of R^1 and R^2 is C_1 to C_5 straight-chain alkyl
or C_3 to C_5 branched-chain alkyl, and the other of R^1 and
 R^2 is C_1 to C_{17} straight-chain alkyl or C_3 to C_{17} branched-
chain alkyl.

15 6. The lubricating oil according to Claim 1,
wherein the aliphatic saturated branched-chain carboxylic
acid monoalkyl ester represented by the formula (1) is
obtained by subjecting an aliphatic saturated branched-
chain carboxylic acid represented by the formula (2):

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wherein R^1 and R^2 are as defined in Claim 1 and an aliphatic monohydric alcohol represented by the formula (3)



5 wherein R^3 is as defined in Claim 1 to an esterification reaction.

7. The lubricating oil according to Claim 6, wherein the aliphatic saturated branched-chain carboxylic acid represented by the formula (2) is at least one member
10 selected from the group consisting of isobutanoic acid, 2-methylbutanoic acid, 2-ethylbutanoic acid, 2,3-dimethylbutanoic acid, 2-methylpentanoic acid, 2-methylhexanoic acid, 2-ethylpentanoic acid, 2,4-dimethylpentanoic acid,
15 2-ethylhexanoic acid, 2-methylheptanoic acid, 2-propylpentanoic acid, 2-ethyloctanoic acid, 2-methylnonanoic acid and 2-pentylnonanoic acid.

8. The lubricating oil according to Claim 6,
20 wherein the aliphatic monohydric alcohol represented by the formula (3) is at least one member selected from the group consisting of isopropanol, isobutanol, sec-butanol, cyclohexanol, n-heptanol, isoheptanol, 2-ethylhexanol, 1-methylheptanol, isooctanol, isononanol, 3,5,5-trimethyl-
25 hexanol, 2,6-dimethyl-4-heptanol, isodecanol and

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* C L M S *

isoundecanol.

9. The lubricating oil according to Claim 1,
wherein the aliphatic saturated branched-chain carboxylic
5 acid monoalkyl ester represented by the formula (1) is at
least one member selected from the group consisting of
isopropyl 2-ethylhexanoate, isobutyl 2-ethylhexanoate,
(sec-butyl) 2-ethylhexanoate, isopentyl 2-ethylhexanoate,
cyclohexyl 2-ethylhexanoate, isoheptyl 2-ethylhexanoate,
10 (2-ethylhexyl) 2-ethylhexanoate, (1-methylheptyl) 2-
ethylhexanoate, isooctyl 2-ethylhexanoate, isononyl 2-
ethylpentanoate, (3,5,5-trimethylhexyl) 2-ethylhexanoate,
(2,6-dimethyl-4-heptyl) 2-ethylhexanoate, isodecyl 2-
ethylhexanoate, isoundecyl 2-ethylhexanoate, (n-propyl) 2-
15 ethylhexanoate, (n-butyl) 2-ethylhexanoate, (n-pentyl) 2-
ethylhexanoate, (n-hexyl) 2-ethylhexanoate, (n-heptyl) 2-
ethylhexanoate, (n-octyl) 2-ethylhexanoate, (n-nonyl) 2-
ethylhexanoate, (n-decyl) 2-ethylhexanoate, (n-undecyl) 2-
ethylhexanoate and (n-dodecyl) 2-ethylhexanoate.

20

10. The lubricating oil according to Claim 1,
comprising two or more aliphatic saturated branched-chain
carboxylic acid monoalkyl esters represented by the
formula (1), wherein the proportion of C_{12} to C_{20} straight-
25 chain alkyls relative to all alkyl groups represented by

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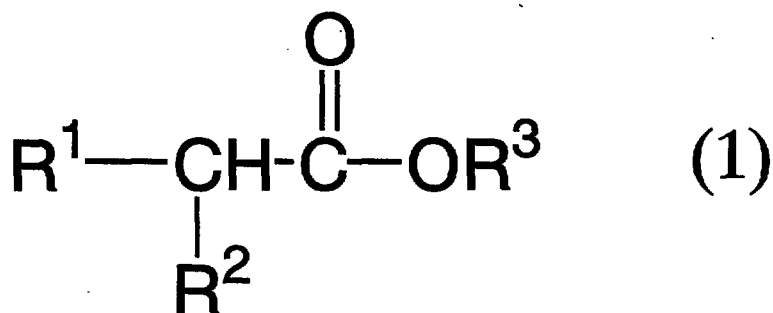


R³ is 5 to 50 mol%.

11. The lubricating oil according to Claim 1,
wherein the aliphatic saturated branched-chain carboxylic
5 acid monoalkyl ester represented by the formula (1) has a
kinematic viscosity of 0.5 to 15 mm²/s at 40°C.

12. A lubricating oil for refrigerators,
characterized in that it comprises a mixture of:

10 a) at least one aliphatic saturated branched-chain
carboxylic acid monoalkyl ester represented by the formula
(1)



15 wherein R¹ is C₁ to C₁₈ straight-chain alkyl or C₃ to C₁₈
branched-chain alkyl; R² is hydrogen, C₁ to C₁₈ straight-
chain alkyl or C₃ to C₁₈ branched-chain alkyl; with the
proviso that the total number of carbon atoms contained in
the alkyls represented by R¹ and R² is 2 to 18 and that
20 when R² is hydrogen, R¹ is branched-chain alkyl; and R³ is

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C₁ to C₂₀ straight-chain alkyl, C₃ to C₂₀ branched-chain alkyl or C₃ to C₁₀ cycloalkyl; and

b) at least one member selected from the group consisting of:

- 5 i) fatty acid polyol esters,
 - ii) phthalic acid esters,
 - iii) alicyclic dicarboxylic acid esters,
 - iv) polyvinyl ethers,
 - v) hydrocarbon oils, and
 - 10 vi) polyalkylene glycols,
- the weight ratio of component a) : component b) being
0.5 : 99.5 to 99.5 : 0.5.

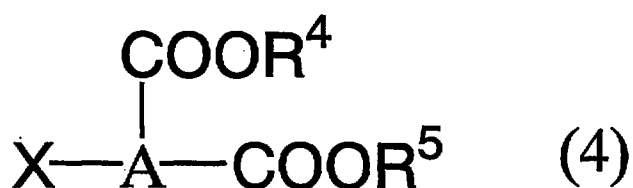
13. The lubricating oil according to Claim 12,
15 which has a kinematic viscosity of 4 to 36 mm²/s at 40°C.

14. The lubricating oil according to Claim 12,
wherein component b) is at least one member selected from
the group consisting of i) fatty acid polyol esters, iii)
20 alicyclic dicarboxylic acid esters and iv) polyvinyl
ethers.

15. The lubricating oil according to Claim 12,
wherein the fatty acid polyol ester is a full ester of a
25 C₅ to C₁₀ aliphatic polyhydric alcohol with 2 to 6 hydroxyl



groups and a C₃ to C₂₂ straight- or branched-chain saturated or unsaturated fatty acid, and the alicyclic dicarboxylic acid ester is represented by the formula (4)



5

wherein A is a cyclohexane ring or a cyclohexene ring; X is hydrogen or methyl; and R⁴ and R⁵ are the same or different and each independently represents a C₃ to C₂₂ straight- or branched-chain saturated or unsaturated
10 aliphatic monovalent hydrocarbon group.

16. A working fluid composition for refrigerators, comprising (I) the aliphatic saturated branched-chain carboxylic acid monoalkyl ester represented by the formula
15 (1) according to Claim 1, and (II) a refrigerant.

17. A working fluid composition for refrigerators, comprising:

(I) a) at least one aliphatic saturated branched-chain
20 carboxylic acid monoalkyl ester represented by the formula (1), and
b) at least one member selected from the group



consisting of i) fatty acid polyol esters, ii) phthalic acid esters, iii) alicyclic dicarboxylic acid esters, iv) polyvinyl ethers, v) hydrocarbon oils and vi) polyalkylene glycols, and

5 (II) a refrigerant.

18. A method for lubricating a refrigerator, comprising using the lubricating oil for refrigerators according to any one of Claims 1 to 11.

10

19. A method for lubricating a refrigerator, comprising using the lubricating oil for refrigerators according to any one of Claims 12 to 15.

15 20. The use of an aliphatic saturated branched-chain carboxylic acid monoalkyl ester represented by the formula (1) according to any of Claims 1 to 11 as a lubricating oil for refrigerators.

20 21. The use of a mixture of a) at least one aliphatic saturated branched-chain carboxylic acid monoalkyl ester represented by the formula (1) according to Claim 1; and
b) at least one member selected from the group consisting
25 of i) fatty acid polyol esters, ii) phthalic acid esters,



iii) alicyclic dicarboxylic acid esters, iv) polyvinyl ethers, v) hydrocarbon oils and vi) polyalkylene glycols, as a lubricating oil for refrigerators.

