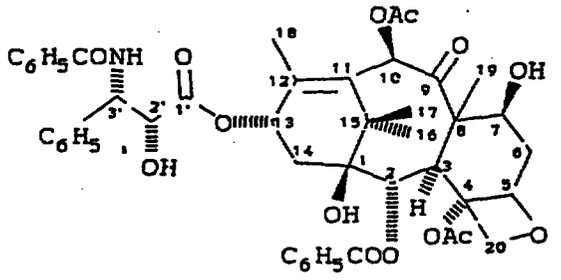
SEMI-SYNTHESIS OF TAXANE DERIVATIVES USING METAL ALKOXIDES AND OXAZINONES

BACKGROUND OF THE INVENTION

The present invention is directed to a semi-synthesis for the preparation of taxane derivatives such as taxol, taxotere and other biologically active derivatives involving the use of metal alkoxides and oxazinones.

The taxane family of terpenes, of which taxol is a member, has attracted considerable interest in both the biological and chemical arts. Taxol is a promising cancer chemotherapeutic agent with a broad spectrum of antileukemic and tumor-inhibiting activity. Taxol has the following structure:



(I)

wherein Ac is acetyl. Because of this promising activity, taxol is currently undergoing clinical trials in both France and the United States.

The supply of taxol for these clinical trials is presently being provided by the bark from Taxus brevifollia (Western Yew). However, taxol is found only in minute quantities in the bark of these slow growing evergreens, causing considerable concern that the limited supply of

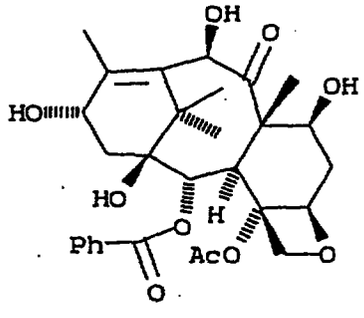
This Sheet Has Been Replaced

REPLACED SHEETS SG
☑ Description : paged 1 to paged 24
☑ Claim(s) : paged 27 to paged 29
☐ Drawing(s) : paged ..... to paged .....
☑ Abstract : paged 30 to paged 31
DATE OF RECEIPT .....

taxol will not meet the demand. Consequently, chemists in recent years have expended their energies in trying to find a viable synthetic route for the preparation of taxol. So far, the results have not been entirely satisfactory.

One synthetic route that has been proposed is directed to the synthesis of the tetracyclic taxane nucleus from commodity chemicals. A synthesis of the taxol congener taxusin has been reported by Holton, et al. in JACS 110, 6558 (1988). Despite the progress made in this approach, the final total synthesis of taxol is, nevertheless, likely to be a multi-step, tedious, and costly process.

A semi-synthetic approach to the preparation of taxol has been described by Greene, et al. in JACS 110, 5917 (1988), and involves the use of a congener of taxol, 10-deacetyl baccatin III which has the structure of formula II shown below:



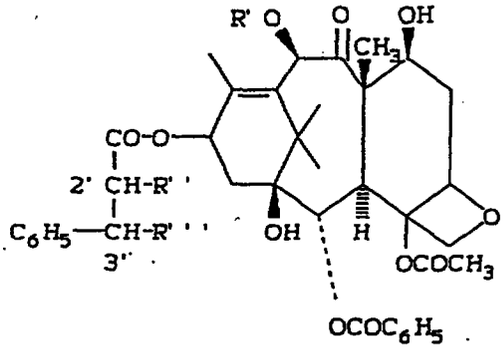
(II)

10-deacetyl baccatin III is more readily available than taxol since it can be obtained from the needles of Taxus baccata. According to the method of Greene et al., 10-deacetyl baccatin III is converted to taxol by attachment of the C-10 acetyl group and by attachment of

This Sheet Has Been Replaced

the C-13 ß-amido ester side chain through the esterification of the C-13 alcohol with a ß-amido carboxylic acid unit. Although this approach requires relatively few steps, the synthesis of the ß-amido carboxylic acid unit is a multi-step process which proceeds in low yield, and the coupling reaction is tedious and also proceeds in low yield. However, this coupling reaction is a key step which is required in <u>every</u> contemplated synthesis of taxol or biologically active derivative of taxol, since it has been shown by Wani, et al. in JACS <u>93</u>, 2325 (1971) that the presence of the ß-amido ester side chain at C13 is required for anti-tumor activity.

More recently, it has been reported in Colin et al. U.S. Patent No. 4,814,470 that taxol derivatives of the formula III below, have an activity significantly greater than that of taxol (I).



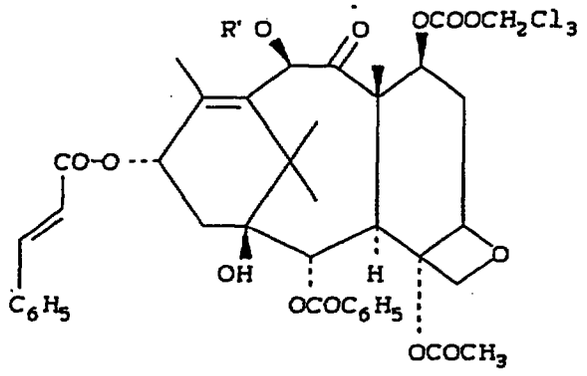
(III)

R' represents hydrogen or acetyl and one of R" and R"' represents hydroxy and the other represents tert-butoxy-carbonylamino and their stereoisomeric forms, and mixtures thereof.

According to Colin et al., U.S. Patent 4,418,470, the products of general formula (III) are obtained by the

This Sheet Has Been Replaced

action of the sodium salt of tert-butyl N-chlorocarbamate on a product of general formula:

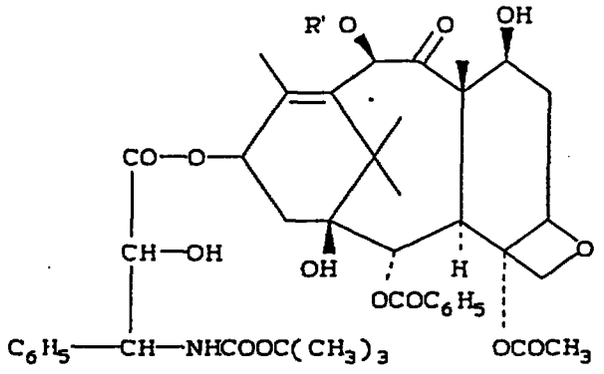


(IV)

in which R' denotes an acetyl or 2,2,2-trichloroethoxycarbonyl radical, followed by the replacement of the 2,2,2-trichloroethoxycarbonyl group ~~or groups~~ by hydrogen. It is reported by Denis et al. in U.S. Patent No. 4,924,011, however, that this process leads to a mixture of isomers which has to be separated and, as a result, not all the baccatin III or 10-deactylbaccatin III employed for the preparation of the product of general formula (IV) can be converted to a product of general formula (III).

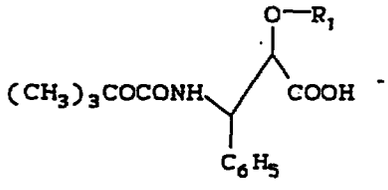
In an effort to improve upon the Colin et al. process, Denis et al. disclose a different process for preparing derivatives of baccatin III or of 10-deactylbaccatin III of general formula

This Sheet Has Been Replaced



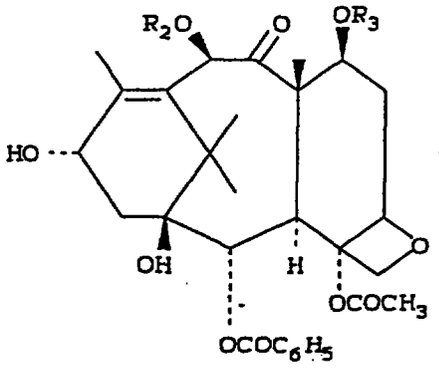
(V)

in which R' denotes hydrogen or acetyl wherein an acid of general formula:



(VI)

in which R_1 is a hydroxy-protecting group, is condensed with a taxane derivative of general formula:



(VII)

in which R_2 is an acetyl hydroxy-protecting group and R_3 is a hydroxy-protecting group, and the protecting groups R_1,

This Sheet Has Been Replaced

R_3 and, where appropriate, R_2 are then replaced by hydrogen. However, this method employs relatively harsh conditions, proceeds with poor conversion, and provides less than optimal yields.

A major difficulty remaining in the synthesis of taxol and other potential anti-tumor agents is the lack of a readily available method for easy attachment, to the C-13 oxygen, of the chemical unit which provides the ß-amido ester side chain. Development of such a process for its attachment in high yield would facilitate the synthesis of taxol as well as related anti-tumor agents having a modified set of nuclear substituents or a modified C-13 side chain. This need has been fulfilled by the discovery of a new, efficient process for attachment, to the C-13 oxygen, of the chemical unit which provides the ß-amido ester side chain.

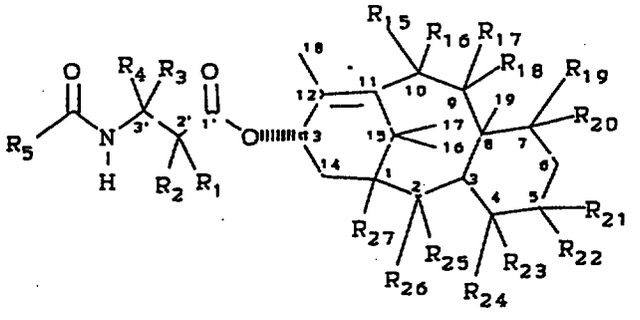
SUMMARY OF THE INVENTION

Among the objects of the present invention, therefore, is the provision of a side chain precursor for the synthesis of taxane derivatives, and the provision of a process for the attachment of the side chain precursor in relatively high yield to provide an intermediate which is readily converted to the desired taxane derivative.

In accordance with the present invention, a process is provided for preparing taxol, taxotere and other biologically active taxane derivatives having the following structural formula:

This Sheet Has Been Replaced





(1)

wherein

R_1 is $-OR_6$;

R_2 is hydrogen, alkyl, alkenyl, alkynyl, aryl, or heteroaryl;

R_3 and R_4 are independently hydrogen, alkyl, alkenyl, alkynyl, aryl, heteroaryl, or acyl, provided, however, that R_3 and R_4 are not both acyl;

R_5 is alkyl, alkenyl, alkynyl, aryl, alkoxy, aryloxy or heteroaryloxy;

R_6 is hydrogen, alkyl, alkenyl, alkynyl, aryl, heteroaryl, or hydroxy protecting group,

R_{15} and R_{16} are independently hydrogen, hydroxy, lower alkanoyloxy, alkenoyloxy, alkynoyloxy, aryloyloxy or R_{15} and R_{16} together form an oxo;

R_{17} and R_{18} are independently hydrogen or lower alkanoyloxy, alkenoyloxy, alkynoyloxy, or aryloyloxy or R_{17} and R_{18} together form an oxo;

R_{19} and R_{20} are independently hydrogen or hydroxy or lower alkanoyloxy, alkenoyloxy, alkynoyloxy, or aryloyloxy;

This Sheet Has Been Replaced

R_{21} and R_{22} are independently hydrogen or lower alkanoyloxy, alkenoyloxy, alkynoyloxy, or aryloyloxy or R_{21} and R_{22} together form an oxo;

R_{24} is hydrogen or hydroxy or lower alkanoyloxy, alkenoyloxy, alkynoyloxy, or aryloyloxy; or

R_{23} and R_{24} together form an oxo or methylene or

R_{23} and R_{24} together form an oxirane ring or

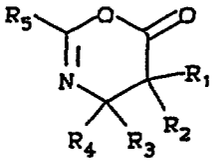
R_{23} and R_{22} together form an oxetane ring;

R_{25} is hydrogen, hydroxy, or lower alkanoyloxy, alkenoyloxy, alkynoyloxy, or aryloyloxy or

R_{26} is hydrogen, hydroxy, or lower alkanoyloxy, alkenoyloxy, alkynoyloxy, or aryloyloxy; or R_{26} and R_{25} taken together form an oxo; and

R_{27} is hydrogen, hydroxy or lower alkoxy, alkanoyloxy, alkenoyloxy, alkynoyloxy, or aryloyloxy.

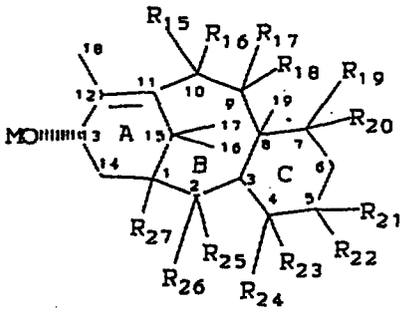
Briefly, therefore, the present invention is directed to a process for the preparation of a taxane derivative in which oxazinone (2) is reacted with a metal alkoxide having the bi-, tri- or tetracyclic taxane nucleus to form a ß-amido ester intermediate. The intermediate is then converted to the taxane derivative. Oxazinone (2) has the general formula:



(2)

wherein R_1 - R_5 are as previously defined. Preferably, the metal alkoxide has the tricyclic taxane nucleus corresponding to the general formula:

This Sheet Has Been Replaced



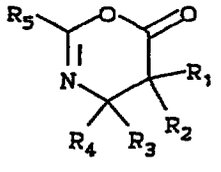
(3)

wherein M is a metal, and R_{15}-R_{27} are as previously defined. Most preferably, the metal alkoxide has the tetracyclic taxane nucleus corresponding to metal alkoxide (3) wherein R_{22} and R_{23} together form an oxetane ring.

Other objects and features of this invention will be in part apparent and in part pointed out hereinafter.

DETAILED DESCRIPTION

The present invention is directed to a process for preparing taxol, taxotere and other taxane derivatives which are biologically active using oxazinone (2), the structure of which is depicted hereinbelow:



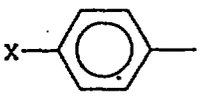
(2)

wherein R_1, R_2, R_3, R_4 and R_5 are as previously defined.

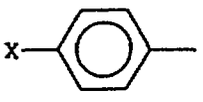
In accordance with the present invention, R_5 of oxazinone (2) is preferably aryl, p-substituted phenyl, or lower alkoxy, and most preferably phenyl, methoxy, ethoxy, tert-butoxy ("tBuO"; $(CH_3)_3CO$-), or

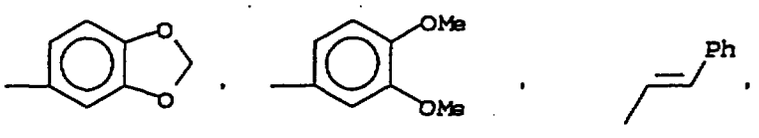
This Sheet Has Been Replaced

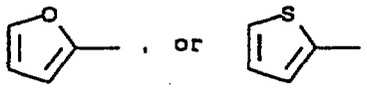




wherein X is Cl, Br, F, CH_3O-, or NO_2-. Preferably R_2 and R_4 are hydrogen or lower alkyl. R_3 is preferably aryl, most preferably, naphthyl, phenyl,







wherein X is as previously defined, Me is methyl and Ph is phenyl. Preferably, R_1 is $-OR_6$ wherein R_6 is 1-ethoxyethyl ("EE") methoxymethyl, or 2,2,2-trichloroethoxymethyl.

The oxazinone alkyl groups, either alone or with the various substituents defined hereinabove are preferably lower alkyl containing from one to six carbon atoms in the principal chain and up to 15 carbon atoms. They may be straight or branched chain and include methyl, ethyl, propyl, isopropyl, butyl, isobutyl, tert-butyl, aryl, hexyl, and the like.

This Sheet Has Been Replaced



The oxazinone alkenyl groups, either alone or with the various substituents defined hereinabove are preferably lower alkenyl containing from two to six carbon atoms in the principal chain and up to 15 carbon atoms. They may be straight or branched chain and include ethenyl, propenyl, isopropenyl, butenyl, isobutenyl, aryl, hexenyl, and the like.

The oxazinone alkynyl groups, either alone or with the various substituents defined hereinabove are preferably lower alkynyl containing from two to six carbon atoms in the principal chain and up to 15 carbon atoms. They may be straight or branched chain and include ethynyl, propynyl, butynyl, isobutynyl, aryl, hexynyl, and the like.

The oxazinone aryl moieties described, either alone or with various substituents, contain from 6 to 15 carbon atoms and include phenyl, α-naphthyl or ß-naphthyl, etc. Substituents include alkanoxy, protected hydroxy, halogen, alkyl, aryl, alkenyl, acyl, acyloxy, nitro, amino, amido, etc. Phenyl is the more preferred aryl.

As noted above, R_1 of oxazinone (2) is $-OR_6$ with R_6 being alkyl, acyl, 1-ethoxyethyl ("EE"), 2,2,2-trichloroethoxymethyl, or other hydroxyl protecting group such as acetals and ethers, i.e., methoxymethyl ("MOM"), benzyloxymethyl; and esters, such as acetates; carbonates, such as methyl carbonates. A variety of protecting groups for the hydroxyl group and the synthesis thereof may be found in "Protective Groups in Organic Synthesis" by T. W. Greene, John Wiley and Sons, 1981. The hydroxyl protecting group selected should be easily removed under conditions that are sufficiently mild, e.g., in 48% HF, acetonitrile, pyridine, or 0.5% HCl/water/ethanol, and/or zinc, acetic acid so as not to disturb the ester linkage or other substituents of the taxol intermediate. However, R_6 is

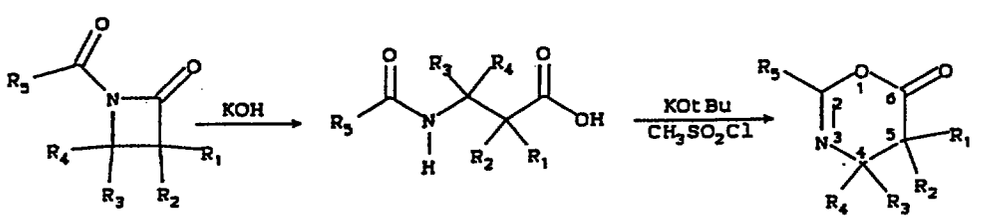
This Sheet Has Been Replaced



preferably 1-ethoxyethyl or 2,2,2-trichloroethoxymethyl, and most preferably 1-ethoxyethyl.

Since oxazinone (2) has several asymmetric carbons, it is known to those skilled in the art that the compounds of the present invention having asymmetric carbon atoms may exist in diastereomeric, racemic, or optically active forms. All of these forms are contemplated within the scope of this invention. More specifically, the present invention includes enantiomers, diastereomers, racemic mixtures, and other mixtures thereof.

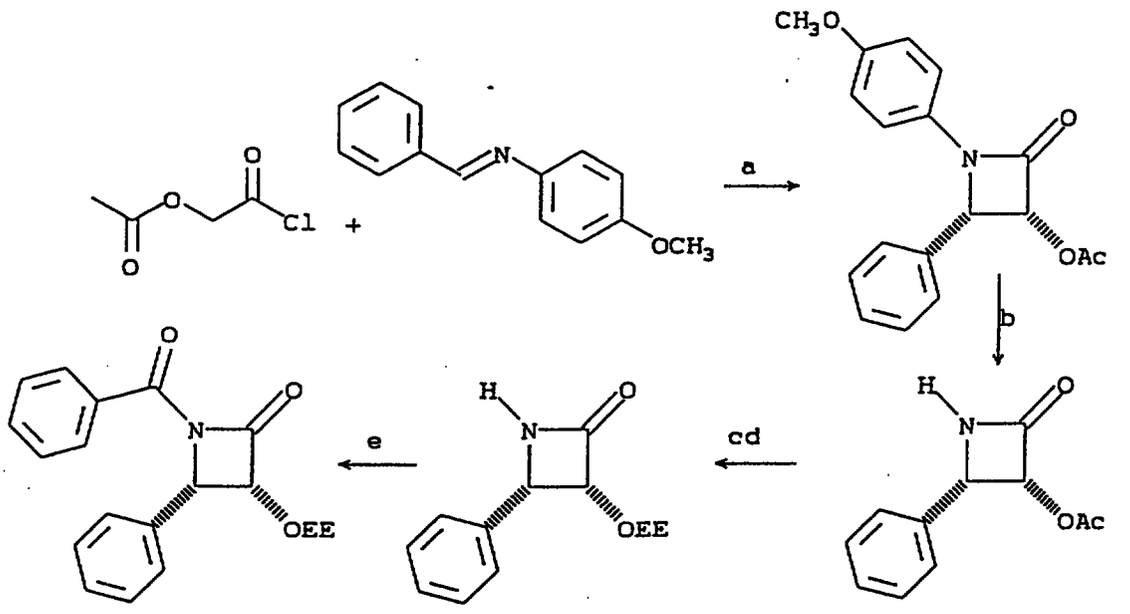
Oxazinone (2) can be prepared from readily available materials according to the following reaction scheme:



4 5 2

Carboxylic acid 5 may alternatively be prepared according to the method described in Greene et al., JACS 110, 5917 (1988). ß-lactams 4 can be prepared from readily available materials, as illustrated in the following reaction scheme in which R_3 and R_5 are phenyl, R_2 and R_4 are hydrogen, and R_2 $-OR_6$ wherein R_6 is is ethoxyethyl:

This Sheet Has Been Replaced



reagents: (a) triethylamine, CH_2Cl_2, 25°C, 18h; (b) 4 equiv ceric ammonium nitrate, CH_3CN, -10 °C, 10 min; (c) KOH, THF, H2O, O °C, 30 min; (d) ethyl vinyl ether, THF, toluene sulfonic acid (cat.), O °C, 1.5h; (e) CH_3Li, ether, -78 °C, 10 min; benzoyl chloride, -78 °C, 1h.

The starting materials are readily available. α-Acyloxy acetyl chloride is prepared from glycolic acid, and, in the presence of a tertiary amine, it cyclocondenses with imines prepared from aldehydes and p-methoxyaniline to give 1-p-methoxyphenyl-3-acyloxy-4-arylazetidin-2-ones.

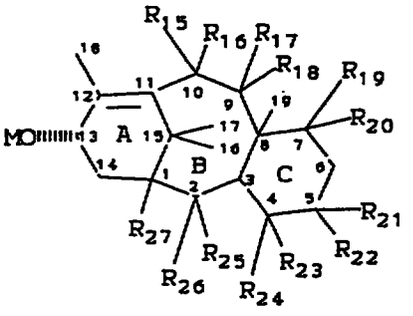
The p-methoxyphenyl group can be readily removed through oxidation with ceric ammonium nitrate, and the acyloxy group can be hydrolyzed under standard conditions familiar to those experienced in the art to provide 3-hydroxy-4-arylazetidin-2-ones.

The 3-hydroxyl group may be protected with a variety of standard protecting groups such as the 1-ethoxyethyl group. Preferably, the racemic 3-hydroxy-

This Sheet Has Been Replaced

4-arylazetidin-2-one is resolved into the pure enantiomers prior to protection by recrystallization of the corresponding 2-methoxy-2-(trifluoromethyl) phenylacetic esters and only the dextrorotatory enantiomer is used in the preparation of taxol. In any event, the 3-(1-ethoxy-ethoxy)-4-phenylazetidin-2-one can be converted to ß-lactam 4, by treatment with a base, preferably n-butyllithium, and an aroyl chloride at -78 °C or below.

As noted above, the metal alkoxides used in the process of the present invention have the bi-, tri- or tetracyclic taxane nucleus. As used herein, a metal alkoxide having the bicyclic taxane nucleus corresponds to a compound containing rings A and B of metal alkoxide (3):



(3)

wherein M and R_{15}-R_{27} are as previously defined. A metal alkoxide having the tricyclic taxane nucleus corresponds to a compound containing rings A, B and C of metal alkoxide (3). A metal alkoxide having the tetracyclic taxane nucleus corresponds to a compound containing rings A, B and

This Sheet Has Been Replaced

C of metal alkoxide (3) and in which R_{22} and R_{23} together form an oxetane ring.

Preferably, the metal alkoxide used in the process of the present invention is metal alkoxide (3). Most preferably, R_{15} is $-OT_2$ or $-OCOCH_3$; R_{16} is hydrogen; R_{17} and R_{18} together form an oxo; R_{19} is $-OT_1$; R_{20} and R_{21} are hydrogen; R_{22} and R_{23} together form an oxetane ring; R_{24} is CH_3COO-; R_{25} is $PhCOO-$; R_{26} is hydrogen; R_{27} is hydroxy; and T_1 and T_2 are independently hydrogen or hydroxy protecting group. The metal alkoxide hydroxy protecting groups may be the same as those previously identified for R_6 of the oxazinone, and preferably are trialkyl or triaryl silyl, and most preferably, triethyl silyl ("TES").

Metal substituent, M, of metal alkoxide (3) is a Group IA, IIA, IIIA, lanthanide or actinide element or a transition, Group IIIA, IVA, VA or VIA metal. Preferably, it is a Group IA, IIA or transition metal, and most preferably, it is lithium, magnesium, sodium, potassium or titantium.

The metal alkoxide alkyl groups, either alone or with the various substituents defined hereinabove are preferably lower alkyl containing from one to six carbon atoms in the principal chain and up to 10 carbon atoms. They may be straight or branched chain and include methyl, ethyl, propyl, isopropyl, butyl, isobutyl, tert-butyl, aryl, hexyl, and the like.

The metal alkoxide alkenyl groups, either alone or with the various substituents defined hereinabove are preferably lower alkenyl containing from two to six carbon atoms in the principal chain and up to 10 carbon atoms. They may be straight or branched chain and include ethenyl, propenyl, isopropenyl, butenyl, isobutenyl, aryl, hexenyl, and the like.

This Sheet Has Been Replaced

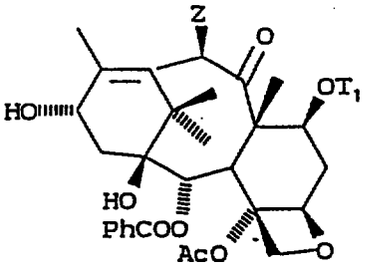


The metal alkoxide alkynyl groups, either alone or with the various substituents defined hereinabove are preferably lower alkynyl containing from two to six carbon atoms in the principal chain and up to 10 carbon atoms. They may be straight or branched chain and include ethynyl, propynyl, butynyl, isobutynyl, aryl, hexynyl, and the like.

Exemplary alkanoyloxy include acetate, propionate, butyrate, valarate, isobutyrate and the like. The more preferred alkanoyloxy is acetate.

The metal alkoxide aryl moieties, either alone or with various substituents contain from 6 to 10 carbon atoms and include phenyl, α-naphthyl or ß-naphthyl, etc. Substituents include alkanoxy, hydroxy, halogen, alkyl, aryl, alkenyl, acyl, acyloxy, nitro, amino, amido, etc. Phenyl is the more preferred aryl.

Metal alkoxides (3) are prepared by reacting an alcohol having two to four rings of the taxane nucleus and a C-13 hydroxyl group with an organometallic compound in a suitable solvent. Preferably, the alcohol is derivative of baccatin III or 10-deacetyl baccatin III having the structure



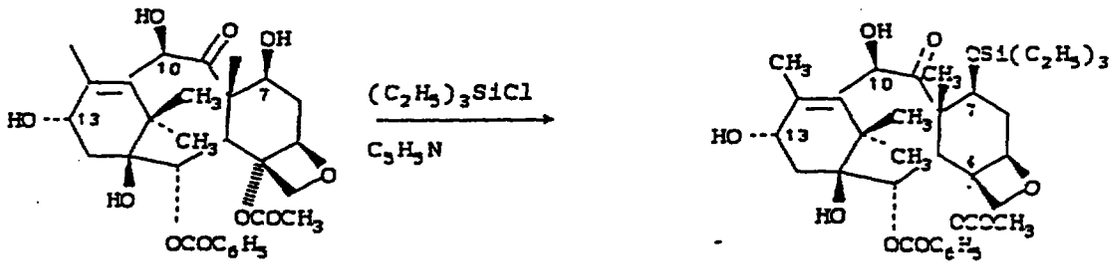
(6)

wherein T_1 is a hydroxy protecting group, and Z is $-OT_2$ wherein T_2 is acyl, preferably acetyl, or hydroxy protecting group. Most preferably, the alcohol is a

This Sheet Has Been Replaced

protected baccatin III, in particular, 7-O-triethylsilyl baccatin III (which can be obtained as described by Greene, et al. in JACS 110, 5917 (1988) or by other routes) or 7,10-bis-O-triethylsilyl baccatin III.

As reported in Greene et al., 10-deacetyl baccatin III is converted to 7-O-triethylsilyl-10-deacetyl baccatin III according to the following reaction scheme:



(7) **(8a)**

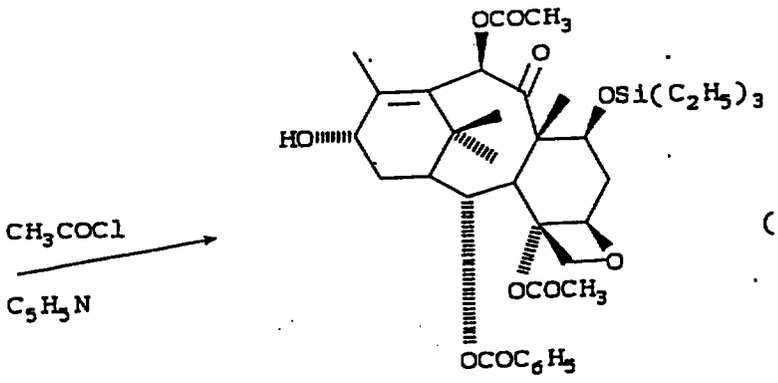
Under what is reported to be carefully optimized conditions, 10-deacetyl baccatin III is reacted with 20 equivalents of $(C_2H_5)_3SiCl$ at 23°C under an argon atmosphere for 20 hours in the presence of 50 ml of pyridine/mmol of 10-deacetyl baccatin III to provide 7-O-triethylsilyl-10-deacetyl baccatin III (8a) as a reaction product in 84-86% yield after purification. The reaction product is then acetylated with 5 equivalents of CH_3COCl and 25 mL of pyridine/mmol of (8a) at 0 °C under an argon atmosphere for 48 hours to provide 86% yield of 7-O-triethylsilyl baccatin III (8b). Greene, et al. in JACS 110, 5917 at 5918 (1988).

This Sheet Has Been Replaced

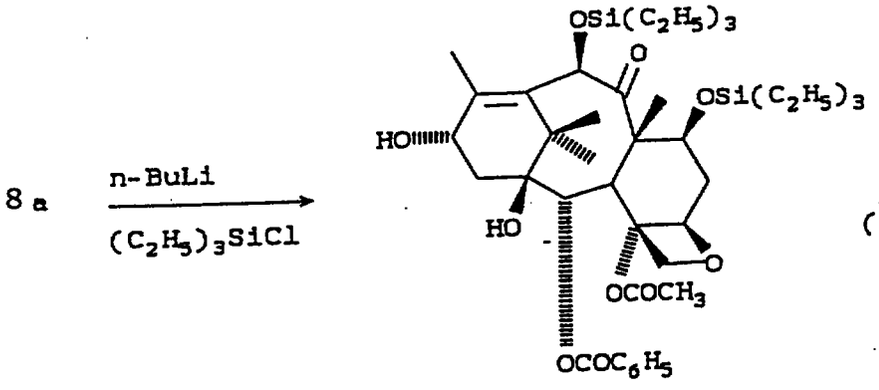
Alternatively, 7-triethylsilyl-10-deacetyl baccatin III (8a) can be protected at C-10 oxygen with an acid labile hydroxyl protecting group. For example, treatment of (8a) with n-butyllithium in THF followed by treithylsilyl chloride (1.1 mol equiv.) at 0°C gives 7,10-bis-O-triethylsilyl baccatin III (8c) in 95% yield. Also, (8a) can be converted to 7-O-triethylsilyl-10-(1-ethoxyethyl) baccatin III (8d) in 90% yield by treatment with excess ethyl vinyl ether and a catalytic amount of methane sulfonic acid. These preparations are illustrated in the reaction scheme below.

This Sheet Has Been Replaced

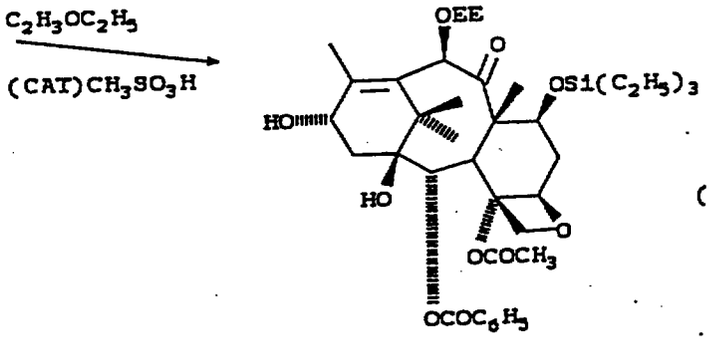




(8 b)



(8 c)

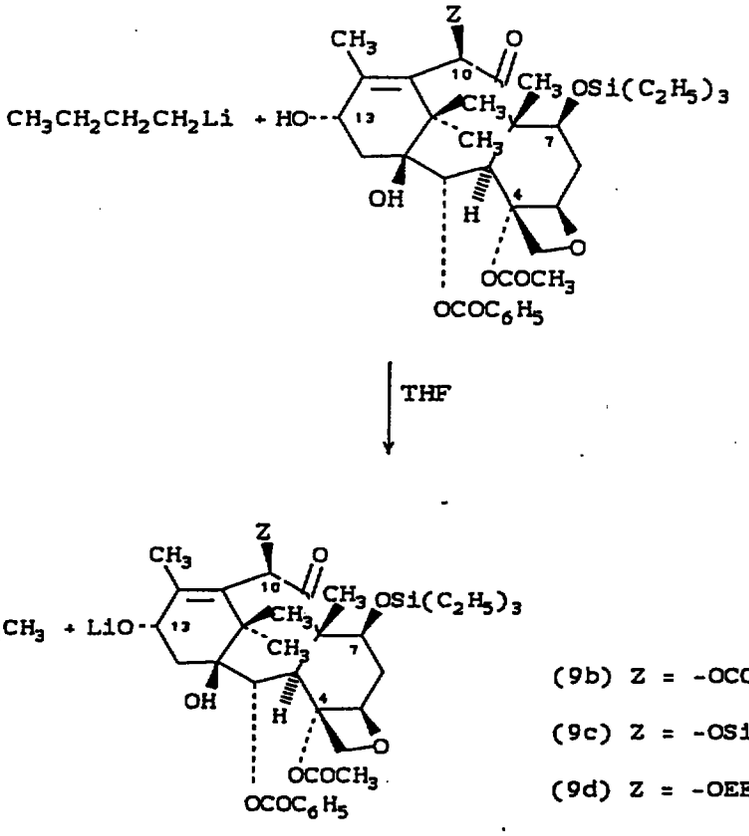


(8d)

7-O-triethylsilyl baccatin III **(8b)**, 7,10-bis-O-triethylsilyl baccatin III **(8c)**, or 7-O-triethylsilyl-10-(1-ethoxyethyl) baccatin III **(8d)** is reacted with an organometallic compound such as n-butyllithium in a solvent such as tetrahydrofuran (THF), to form the metal alkoxide

This Sheet Has Been Replaced

13-O-lithium-7-O-triethylsilyl baccatin III (9b) 13-O-lithium-7,10-bis-O-triethylsilyl baccatin III (9c), or 13-O-lithium-7-O-triethylsilyl-10- (1-ethoxyethyl) baccatin III (9d) as shown in the following reaction scheme:



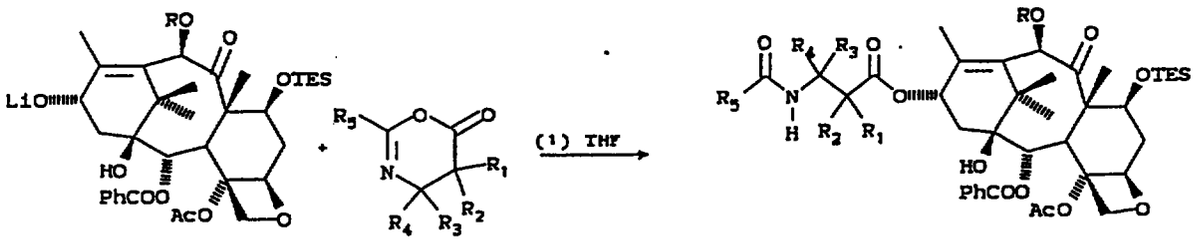
(9b) Z = -OCOCH$_3$

(9c) Z = -OSi(C$_2$H$_5$)$_3$

(9d) Z = -OEE

As illustrated in the following reaction scheme, a suitable metal alkoxide of the present invention such as 13-O-lithium-7-O-triethylsilyl baccatin III derivative (**9b**, **9c**, or **9d**) reacts with an oxazinone of the present invention to provide an intermediate (**10b**, **10c**, or **10d**) in which the C-7 hydroxyl group is protected with a triethylsilyl or 1-ethoxyethyl group.

This Sheet Has Been Replaced



9 b-d (2) 10 b-d

b, R = $-COCH_3$
c, R = $-Si(C_2H_5)_3$
d, R = -EE

Intermediate compound **(10b)** readily converts to taxol when R_1 is $-OR_6$, R_2 and R_3 are hydrogen, R_4 is phenyl, R_5 is benzoyl and R_6 is a hydroxy protecting group such as 1-ethoxyethyl. Intermediate compound **(10c)** readily converts to taxotere when R_1 is $-OR_6$, R_2 and R_3 are hydrogen, R_4 is phenyl, R_5 is tertbutoxycarbonyl and R_6 is a hydroxy protecting group such as 1-ethoxyethyl. Intermediate compound **(10d)** readily converts to 10-deacetyl taxol when R_1 is $-OR_6$, R_2 and R_3 are hydrogen, R_4 is phenyl, R_5 is benzoyl, and R_6 is a hydroxy protecting group such as 1-ethoxyethyl. Intermediate compounds **(10b, 10c and 10d)** may be converted to the indicated compounds by hydrolyzing the triethylsilyl and 1-ethoxyethyl groups under mild conditions so as not to disturb the ester linkage or the taxane derivative substituents.

10b $\xrightarrow{HF,\ C_5H_5N,\ CH_3CN}$ TAXOL

10c $\xrightarrow{HF,\ C_5H_5N,\ CH_3CN}$ TAXOTERE

10d $\xrightarrow{0.1\%\ HCl,\ EtOH}$ 10-DEACETYL TAXOL

This Sheet Has Been Replaced

Other taxane derivatives may readily be prepared by selection of the proper substituents R_1 - R_5 of oxazinone (2) or R_{15} - R_{27} of metal alkoxide (3). The preparation of such other compounds is illustrated in the examples which follow.

Both the conversion of the alcohol to the metal alkoxide and the ultimate synthesis of the taxol can take place in the same reaction vessel. Preferably, the oxazinone is added to the reaction vessel after formation therein of the metal alkoxide.

The organometallic compound n-butyllithium is preferably used to convert the alcohol to the corresponding metal alkoxide, but other sources of metallic substituent such as lithium diisopropyl amide, other lithium or magnesium amides, ethylmagnesium bromide, methylmagnesium bromide, other organolithium compounds, other organomagnesium compounds, organosodium, organotitanium or organopotassium may also be used. Organometallic compounds are readily available, or may be prepared by available methods including reduction of organic halides with metal. For example, butyl bromide can be reacted with lithium metal in diethyl ether to give a solution of n-butyllithium in the following manner:

$$CH_3CH_2CH_2CH_2Br + 2Li \xrightarrow[Et_2O]{-10^\circ C} CH_3CH_2CH_2CH_2Li + LiBr$$

Although THF is the preferred solvent for the reaction mixture, other ethereal solvents, such as dimethoxyethane, or aromatic solvents may also be suitable. Certain solvents, including some halogenated solvents and some straight-chain hydrocarbons in which the

This Sheet Has Been Replaced

reactants are too poorly soluble, are not suitable. Other solvents are not appropriate for other reasons. For example, esters are not appropriate for use with certain organometallic compounds such as n-butyllithium due to incompatibility therewith.

Although the reaction scheme disclosed herein is directed to the synthesis of certain taxane derivatives, it can be used with modifications in either the oxazinone or the tetracyclic metal alkoxide. Therefore metal alkoxides other than 13-O-lithium-7-O-triethylsilyl baccatin III may be used to form an intermediate according to the method of this invention. The oxazinone and the tetracyclic metal alkoxide can be derived from natural or unnatural sources, to prepare other synthetic taxols, taxol derivatives, 10-deacetyltaxols, and the enantiomers and diastereomers thereof contemplated within the present invention.

The water solubility of compounds of formula (1) may be improved if R_1 is $-OR_6$ and R_{19} is $-OT_1$, and R_6 and/or T_1 are a functional group which increases solubility, such as $-COGCOR^1$ wherein

G is ethylene, propylene, -CHCH-, 1,2-cyclohexane, or 1,2-phenylene,

R^1 = OH base, NR^2R^3, OR^3, SR^3, $OCH_2CONR^4R^5$, OH

R^2 = hydrogen, methyl

R^3 = $(CH_2)_nNR^6R^7$; $(CH_2)_nN^{\oplus}R^6R^7R^8X_1^{\ominus}$

n = 1 to 3

R^4 = hydrogen, lower alkyl containing 1 to 4 carbons

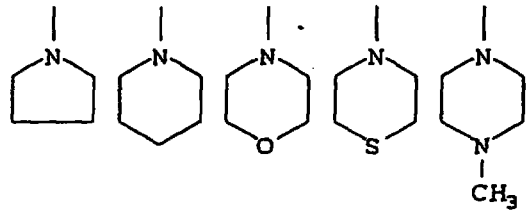
R^5 = hydrogen, lower alkyl containing 1 to 4 carbons, benzyl, hydroxyethyl, CH_2CO_2H, dimethylaminoethyl

R^6R^7 = lower alkyl containing 1 or 2 carbons, benzyl or R^6 and

This Sheet Has Been Replaced



R^7 together with the nitrogen atom of NR^6R^7 form the following rings



R^8 = lower alkyl containing 1 or 2 carbons, benzyl

$X_1^{\ominus}$ = halide

base = NH_3, $(HOC_2H_4)_3N$, $N(CH_3)_3$, $CH_3N(C_2H_4OH)_2$, $NH_2(CH_2)_6NH_2$, N-methylglucamine, NaOH, KOH.

The preparation of compounds in which R_6 or T_1 is $-COGCOR^1$ is set forth in Hangwitz U.S. Patent 4,942,184 ~~which is incorporated herein by reference~~.

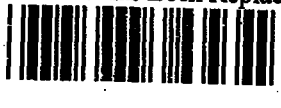
The following examples illustrate the invention.

EXAMPLE 1

Preparation of Taxol

To a solution of 7-triethylsilyl baccatin III (100 mg, 0.143 mmol) in 1 mL of THF at -45°C was added dropwise 0.087 mL of a 1.63M solution of nBuLi in hexane. After 1 h at -45°C, a solution of 92 mg (0.286 mmol) of (-)-cis-2,4-diphenyl-5-(1-ethoxyethoxy)-4,5-dihydro-1,3-oxazin-6-one in 1 mL of THF was added dropwise to the mixture. The solution was warmed to 0°C and kept at that temperature for 2 h before 1 mL of a 10% solution of AcOH in THF was added. The mixture was partitioned between saturated aqueous $NaHCO_3$ and 60/40 ethyl acetate/hexane. Evaporation of the organic layer gave a residue which was purified by flash chromatography followed by recrystallization to give 120 mg (80%) of (2'R,3'S)-2'-ethoxyethyl-7-triethylsilyl taxol.

This Sheet Has Been Replaced



A 5 mg sample of (2'R,3'S)-2'-ethoxyethyl-7-triethylsilyl taxol was dissolved in 2 mL of ethanol and 0.5 mL of 0.5% aqueous HCl solution was added. The mixture was stirred at 0°C for 30 h and diluted with 50 mL of ethyl acetate. The solution was extracted with 20 mL of saturated aqueous sodium bicarbonate solution, dried over sodium sulfate and concentrated. The residue was purified by flash chromatography to provide 4.5 mg (ca. 90%) of taxol, which was identical with an authentic sample in all respects.

EXAMPLE 2

Preparation of N-debenzoyl-N-tertbutoxycarbonyl-10-deacetyl Taxol (Taxotere)

2-tertbutoxy-4-phenyl-5-(1-ethoxyethoxy)-4,5-dihydro-1,3-oxazin-6-one. To a solution of 409 mg (1.16 mmol) of N-tertbutoxycarbonyl-O-(1-ethoxyethyl)-3-phenylisoserine (3) in 20 mL of THF was added 261 mg (2.33 mmol) of solid potassium tert-butoxide and the mixture was stirred at 25°C for 30 min. A solution of 134 mg (1.16 mmol) of methanesulfonyl chloride in 3.2 mL of THF was added and the mixture was stirred at 25°C for 1.5 h. The mixture was diluted with 80 mL of hexane and ethyl acetate and this solution was extracted with 20 mL of saturated aqueous sodium bicarbonate solution and 10 mL of brine. The organic phase was dried over sodium sulfate and concentrated to give 235 mg (70%) of 2-tertbutoxy-4-phenyl-5-(1-ethoxyethoxy)-4,5-dihydro-1,3-oxazin-6-one as a colorless oil.

N-debenzoyl-N-tertbotoxycarbonyl-10-deacetyl-Taxol (Taxotere). To a solution of 7,10-bis-triethylsilyl baccatin III (100 mg, 0.124 mmol)) in 1 mL of THF at -45°C was added dropwise 0.087 of a 1.63M solution of nBuLi in

This Sheet Has Been Replaced



hexane. After 1 h at -45°C, a solution of 88 mg (0.286 mmol) of (-)-cis-2-tertbutoxy-4-phenyl-5-(1-ethoxyethoxy)-4,5-dihydro-1,3-oxazin-6-one in 1 mL of THF was added dropwise to the mixture. The solution was warmed to 0°C and kept at that temperature for 2 h before 1 mL of a 10% solution of AcOH in THF was added. The mixture was partitioned between saturated aqueous $NaHCO_3$ and 60/40 ethyl acetate/hexane. Evaporation of the organic layer gave a residue which was purified by flash chromatography followed by recrystallization to give 120 mg (80%) of (2'R,3'S)-N-debenzoyl-N-tertbutoxycarbonyl-2'-ethoxyethyl-7,10-bis-treithylsilyl taxol.

A 5 mg sample of (2'R,3'S)-N-debenzoyl-N-tert-butoxycarbonyl-2'-ethoxyethyl-7,10-bis-triethylsilyl taxol was dissolved in 2 mL of ethanol and 0.5 mL of 0.5% aqueous HCl solution was added. The mixture was stirred at 0°C for 30 h and diluted with 50 mL of ethyl acetate. The solution was extracted with 20 mL of saturated aqueous sodium bicarbonate solution, dried over sodium sulfate and concentrated. The residue was purified by column chromatography on silica gel eluted with ethyl/hexane to provide 3.8 mg (ca. 90%) of N-debenzoyl-N-tertbutoxy-carbonyl-10-deacetyl Taxol (Taxotere). All analytical and spectral data were identical with that reported for taxotere in U.S. Patent 4,814,470.

In view of the above, it will be seen that the several objects of the invention are achieved.

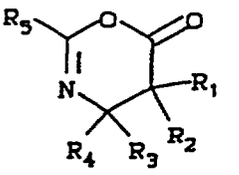
This Sheet Has Been Replaced



* C L M S *

CLAIMS

1. A process for the preparation of a taxane derivative characterized in that a metal oxide substituted taxane having a bi-, tri or tetracylic nucleus is reacted with an oxazinone wherein the oxazinone has the formula



(2)

and

R_1 is $-OR_6$;

R_2 is hydrogen, alkyl, alkenyl, alkynyl, aryl, or heteroaryl;

R_3 and R_4 are independently hydrogen, alkyl, alkenyl, alkynyl, aryl, heteroaryl, or acyl, provided, however, that R_3 and R_4 are not both acyl;

R_5 is alkyl, alkenyl, alkynyl, aryl, alkoxy, aryloxy or heteroaryloxy; and

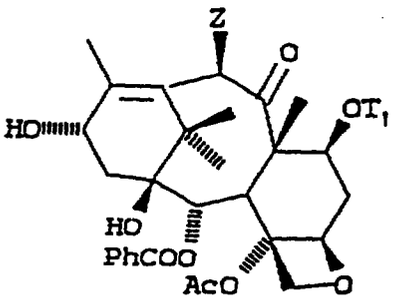
R_6 is hydrogen, alkyl, alkenyl, alkynyl, aryl, heteroaryl, or hydroxy protecting group.

2. A process according to claim 1, wherein R_2 and R_4 are hydrogen or lower alkyl and R_3 is aryl.

3. A process according to claim 1 or claim 2, wherein the metal alkoxide has a tetracylic taxane nucleus.

4. A process according to any one of claims 1 to 3, wherein the metal alkoxide is derived from an alcohol having the formula:

This Sheet Has Been Replaced



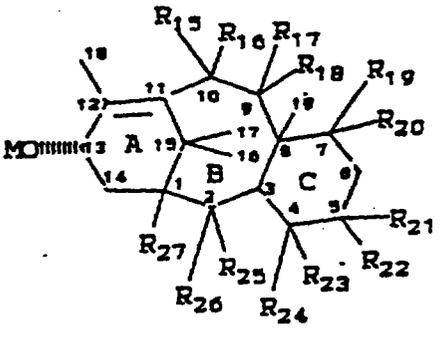
(6)

wherein T_1 us a hydroxy protecting group, Z is $-OT_2$, and T_2 is acetyl or hydroxy protecting group.

5. A process according to any one of claims 1 to 4, wherein the metal alkoxide is a metal alkoxide of 7-protected baccatin III.

6. A process according to claim 5, wherein the metal alkoxide is a metal alkoxide of 7-O-triethylsilyl baccatin III or a metal alkoxide at 7,10-bis-O-triethylsilyl baccatin III or a metal alkoxide of 7-O-triethylsilyl-10-deacetyl baccatin III or a metal alkoxide of 7-O-triethylsilyl-10-(1-ethoxyethyl) baccatin III.

7. A process according to claim 1 or claim 2, wherein the metal alkoxide has the formula:



(3)

wherein:

R_{15} and R_{16} are independently hydrogen, hydroxy, lower alkanoyloxy, alkenoyloxy, alkynoyloxy, aryloyloxy or R_{15} and R_{16} together form an oxo;

R_{17} and R_{18} are independently hydrogen or lower

This Sheet Has Been Replaced

alkanoyloxy, alkenoyloxy, alkynoyloxy, or aryloyloxy or R_{17} and R_{18} together form an oxo;

R_{19} and R_{20} are independently hydrogen or hydroxy or lower alkanoyloxy, alkenoyloxy, alkynoyloxy, or aryloyloxy;

R_{21} and R_{22} are independently hydrogen or lower alkanoyloxy, alkenoyloxy, alkynoyloxy, or aryloyloxy or R_{21} and R_{22} together form an oxo;

R_{24} is hydrogen or hydroxy or lower alkanoyloxy, alkenoyloxy, alkynoyloxy, or aryloyloxy; or

R_{23} and R_{24} together form an oxo or methylene or

R_{23} and R_{24} together form an oxirane ring or

R_{23} and R_{22} together form an oxetane ring;

R_{25} is hydrogen, hydroxy, or lower alkanoyloxy, alkenoyloxy, alkynoyloxy, or aryloyloxy or

R_{26} is hydrogen, hydroxy, or lower alkanoyloxy, alkenoyloxy, alkynoyloxy, or aryloyloxy; or R_{26} and R_{25} taken together form an oxo; and

R_{27} is hydrogen, hydroxy or lower alkoxy, alkanoyloxy, alkenoyloxy, alkynoyloxy, or aryloyloxy.

8. A process according to claim 7, wherein R_{16} is hydrogen, R_{15} is $-OT_2$ or $-OCOCH_3$, T_2 is a hydroxy protecting group and M is selected from the group comprising Li, Mg, Na, K and Ti.

9. A process according to any one of claims 1 to 8, wherein the metal oxide substituent is selected from the group consisting of LiO-, MgO-, NaO-, KO-, and TiO-.

This Sheet Has Been Replaced