

CORE MATERIAL, ELECTRONIC COMPONENT AND
METHOD FOR FORMING BUMP ELECTRODE

Technical Field

5 [0001]

The present embodiment relates to a core material, an electronic component having a bump using this core material, and a method for forming a bump electrode.

[0002]

10 The present application claims the priority of Japanese Patent Application No. 2020-151742 filed on September 10, 2020, the contents of which are entirely incorporated by reference.

15 Background Art

[0003]

In recent years, with development of a small information device, size of an electronic component mounted on the information device is rapidly being reduced. A ball
20 grid array (BGA) with an electrode disposed on a rear surface of the BGA is applied as the electronic component in order to correspond to narrowing a connection terminal and reducing mounting area according to a demand for miniaturization.

25 [0004]

An example of the electronic component to which the BGA is applied includes a semiconductor package. The

semiconductor package is configured by sealing a semiconductor chip having an electrode with a resin. A solder bump is formed on the electrode of the semiconductor chip. The solder bump is formed by bonding a solder ball
5 to the electrode of the semiconductor chip. The semiconductor package to which the BGA is applied is mounted on a printed circuit board by bonding the solder bump melted by heating to a conductive land of the printed circuit board. In recent years, in order to meet the
10 demand for higher-density mounting, a three-dimensional high-density mounting in which semiconductor packages are stacked in a height direction has also been developed.
[0005]

When the semiconductor package obtained with the
15 three-dimensional high-density mounting is the BGA and the solder ball is placed on the electrode of the semiconductor chip and reflowed, the solder ball may be crushed due to weight of the semiconductor package itself. When such a situation occurs, there is a problem that the solder
20 extrudes from electrodes to bring the electrodes into contact with each other, and a short-circuit between the electrodes is caused.
[0006]

In order to prevent such a short-circuit from
25 occurring, a solder bump that is not crushed due to the weight of the semiconductor package itself and is not deformed when the solder is melted is proposed as a solder

ball. Specifically, it is proposed to use a ball molded of metal or a resin as a core, and to use a core material obtained by coating the core with a solder as a solder bump.

5 [0007]

Japanese Patent No. 5367924 discloses a solder bump in which a Cu ball is used as a core and a Sn-based solder alloy formed of Sn and Bi is plated on the Cu ball. A plating treatment is performed at a concentration gradient
10 in which a content of Bi contained in a solder plating layer in Japanese Patent No. 5367924 is high on an inner side (inner circumferential side) and is lower as being directed toward an outer side (outer circumferential side).
[0008]

15 As in Japanese Patent No. 5367924, when a solder is heated and melted in a state in which a concentration of Bi in the solder plating layer is high on the inner circumferential side and is low on the outer circumferential side, a density of Bi on the inner
20 circumferential side is high. Therefore, the solder begins to be melted from a Bi region on the inner circumferential side. Even when the Bi region on the inner circumferential side is melted, the Bi region on the outer circumferential side has yet to begin to be melted. Therefore, a volume
25 expansion in the Bi region on the inner circumferential side occurs early. Due to a slow volume expansion on the inner and outer circumferential sides, a situation occurs

in which a pressure difference is generated between the inner circumferential side and the outer circumferential side (outside air) of Bi, and the pressure difference due to the volume expansion on the inner circumferential side causes the Cu ball, which is a core, to burst open when the outer circumferential side of Bi begins to be melted. The occurrence of the situation has to be avoided. As described above, a defect has been generated in a Cu core ball having a solder plating layer formed of the Sn-based solder alloy formed of Sn and Bi when Bi in the solder plating layer has a concentration gradient.

[0009]

Japanese Patent No. 6217836 is proposed in order to solve the above-described problems caused in Japanese Patent No. 5367924. In Japanese Patent No. 6217836, in a core material having a solder plating layer formed by plating a (Sn-Bi)-based solder alloy formed of Sn and Bi on a core surface, the core material is provided in which Bi contained in the solder plating layer is distributed in a concentration ratio within a predetermined range of 91.4% to 106.7% in the solder plating layer.

SUMMARY OF INVENTION

PROBLEM TO BE SOLVED BY INVENTION

[0010]

When a relatively large amount of Ag and Cu are contained in the surface of the solder plating layer, Ag

and Cu are locally discolored to black, and the appearance is mottled, which may cause deteriorated appearance quality. As a result of the color becoming black from the viewpoint of yellowness, the yellowness increases, which
5 may cause defect determination in appearance inspection.
[0011]

As described above, when a relatively large amount of Ag and Cu are contained in the surface of the solder plating layer, the solder plating layer easily reacts with
10 a sulfur component of the plating liquid and oxygen in the air. As a result, oxidation easily proceeds, which tends to easily cause an increased oxide film thickness. When the oxide film thickness increases as described above, dissolution behavior at the time of substrate reflow
15 deteriorates, which may cause a state (ball missing) where the solder ball is detached from the electrode pad or positional deviation.
[0012]

The present invention provides a core material and so
20 on that is unlikely to be determined to be defective in a visual inspection and can prevent the oxide film thickness from increasing and suppress a bonding defect.

Solution to Problem

[0013]

25 A core material according to the present invention may comprise:

a core;

a solder layer provided outside the core and being a solder alloy containing Sn and at least any one element of Ag, Cu, Sb, Ni, Co, Ge, Ga, Fe, Al, In, Cd, Zn, Pb, Au, P, S, Si, Ti, Mg, Pd, and Pt; and

5 a Sn layer provided outside the solder layer,
wherein

the solder layer may have a thickness of 1 μm or more on one side,

the Sn layer may have a thickness of 0.1 μm or more
10 on one side, and

a thickness of the Sn layer may be 0.215% or more and 36% or less of the thickness of the solder layer.

[0014]

In the core material according to the present
15 invention, the solder layer may be a (Sn-Ag-Cu)-based solder.

[0015]

The core material, according to the present invention, may further comprise a base plating layer made
20 of any one of Ni, Co, Ni-Co, Ni-P and Ni-B between the core and the solder layer.

[0016]

In the core material according to the present invention, Cu ball or Cu column may be used as the core.

25 [0017]

An electronic component according to the present invention may use the any one of the above core materials

as a solder bump.

[0018]

A method for forming bump electrode according to the present invention may comprise steps of:

5 mounting any one of the above core materials on an electrode;

forming a bump electrode by heating the mounted core material.

[0019]

10 According to the present invention, a core material and so on that is unlikely to be determined to be defective in a visual inspection and can prevent the oxide film thickness from increasing and suppress a bonding defect can be provided.

15

BRIEF DESCRIPTION OF DRAWINGS

[0020]

Fig. 1 is a cross-sectional view illustrating an example of a Cu core ball according to an embodiment of the present invention.

20

Fig. 2 is a cross-sectional view illustrating another example of a Cu core ball according to an embodiment of the present invention.

25

EMBODIMENT FOR CARRYING OUT INVENTION

[0021]

Hereinafter, preferred embodiments of the present

embodiment will be described in detail. In the embodiment, a core material and an electronic component including a semiconductor package using the core material are provided. [0022]

5 The core material of the present embodiment may include a core; a solder layer provided outside the core and being a solder alloy containing Sn and at least any one element of Ag, Cu, Sb, Ni, Co, Ge, Ga, Fe, Al, In, Cd, Zn, Pb, Au, P, S, Si, Ti, Mg, Pd, and Pt; and a Sn layer
10 provided outside the solder layer. The thickness of the solder layer may be 1 μm or more on one side. The thickness of the Sn layer is considered to be 0.1 to 12 μm on one side, but the upper limit value thereof may exceed 12 μm as long as the thickness of the Sn layer is 0.215% or
15 more and 36% or less of that of the solder layer. [0023]

 The Sn layer may be formed to have a thickness of preferably 0.26% or more and 33.3% or less of that of the solder layer. In the present embodiment, as an example, a
20 solder plating layer is described as the solder layer to be used, and a Sn plating layer is described as the Sn layer to be used, but the present invention is not limited thereto. The solder layer may be formed by a method other than plating, and similarly, the Sn layer may be formed by
25 a method other than plating. The Sn layer may contain a component other than Sn as an impurity such as Ag, Cu, Bi, Sb, Ni, Co, Ge, Ga, Fe, Al, In, Cd, Zn, Pb, Au, P, S, or

Si.

[0024]

The solder alloy may be a Sn-based solder alloy. The Sn-based solder alloy can include other additive elements in addition to (Sn-Ag)-based and (Sn-Cu)-based solder alloys. The element that can be added to the (Sn-Ag)-based and (Sn-Cu)-based solder alloys is one or two or more elements of Ag, Cu, Ni, Bi, Ge, Ga, In, Zn, Fe, Pb, Sb, Au, Pd, and Co and the like. For example, a (Sn-Ag-Cu-Ni)-based solder alloy and a (Sn-Ag-Sb)-based solder alloy and the like are considered. A Pb-based solder alloy may be used; the content (main component) of Pb may be the largest in mass; a (Pb-Sn)-based solder alloy may be used; and the content of Pb may be more than that of Sn. In Examples to be described later, SAC 305 (Ag 3.0 Cu 0.5 Sn balance), SnAg (Ag 3.5 Sn balance), SnCu (Cu 0.7 Sn balance), SnSb (Sb 10 Sn balance), SnIn (In 45 Sn balance), and SnPb (Pb 37 Sn balance) are used, but the present invention is not limited thereto. For example, for SAC, the content of Ag and the content of Cu may be changed, and Ag 0.1 Cu 0.1 Sn balance to Ag 10 Cu 10 Sn balance may be used. For SnAg, Ag 0.1 Sn balance to Ag 10 Sn balance may be used by changing the content of Ag. For SnCu, Cu 0.1 Sn balance to Cu1Sn balance may be used by changing the content of Cu. For SnSb, Sb 0.1 Sn balance to Sb 15 Sn balance may be used by changing the content of Sb. For SnIn, In 0.1 Sn balance to In 95 Sn balance may be used by changing the content of

In. For SnPb, Pb 0.1 Sn balance to Pb 95 Sn balance may be used by changing the content of Pb. In the present application, the numerical value of the addition amount of each element is represented by % by mass. For example, "Ag 3.0 Cu 0.5 Sn balance" indicates that 3.0% by mass of Ag and 0.5% by mass of Cu are contained, the balance being Sn. [0025]

The Sn plating layer is formed after the solder plating layer is formed. As an example, a solder plating layer containing Sn and at least any one element of Ag, Cu, Sb, Ni, Co, Ge, Ga, Fe, Al, In, Cd, Zn, Pb, Au, P, S, Si, Ti, Mg, Pd, and Pt on the core is formed. The core material on which the solder plating layer is formed may be pulled up from a molten liquid containing Sn and at least any one element of Ag, Cu, Sb, Ni, Co, Ge, Ga, Fe, Al, In, Cd, Zn, Pb, Au, P, S, Si, Ti, Mg, Pd, and Pt, and then immersed in a molten liquid of Sn to form the Sn plating layer. Both the solder plating layer and the Sn plating layer may be formed using barrel electroplating.

Electroless plating may be used as a plating method. [0026]

When the Sn plating layer is not provided, and the solder plating layer contains 0.1% by mass or more of a component other than Sn, the problem of discoloration occurs. When the Sn plating layer is not provided, and the solder plating layer is 100% by mass of Sn, the problem of discoloration does not occur, but the problem of

discoloration is confirmed to occur when the solder plating layer is made of the following components. In the case of electroplating, if Ag or the like remains in the plating liquid just before the end of solder plating, substitution
5 deposition-based discoloration of the surface is considered to occur.

Ag 0.05 Cu 0.05 Sn balance, discolored

Ag 0.1 Sn balance, discolored

Cu 0.1 Sn balance, discolored

10 Sb 0.1 Sn balance, discolored

In 0.1 Sn balance, discolored

Pb 0.1 Sn balance, discolored

[0027]

Metal or a resin is conceivable as the core, and a
15 shape of the core is conceivable to be a sphere or other shape (columnar column, sheet shape, and the like). In the embodiment, a case of a Cu core ball which is a sphere and uses a ball which is, particularly, formed of Cu as the core (also referred to as "Cu ball") will be described.
20 The Cu core ball in the embodiment only needs to contain Cu in the core, and other configurations are not particularly limited.

[0028]

A particle diameter (sphere diameter) of the core
25 differs depending on size of the BGA, but in the following example, the core has a spherical shape of about 300 μm and a thickness of the solder plating layer on one side in

a diameter direction is 1 μm to 100 μm . A thickness of the Sn plating layer on one side in the diameter direction may be determined based on the thickness of the solder plating layer, and as described above the thickness of the Sn

5 plating layer on one side in the diameter direction is 0.215% or more and 36% or less of the thickness of the solder plating layer, preferably 0.26% or more and 33.3% or less of the thickness of the solder plating layer. The particle diameter of the Cu core ball is appropriately

10 selected according to a density and size of the electronic components to be used. Therefore, the core having the particle diameter in a range of 1 μm to 1000 μm can be used, and the plating thickness can be appropriately selected according to the particle diameter of the core to

15 be used. A typical particle diameter when being used as a ball is 200 μm to 300 μm . When the particle diameter is further minimized in the future, a mainstream particle diameter may be 100 μm to 160 μm , and the present

application can be used without any problem even with this particle diameter. As a plating apparatus that performs the plating treatment, an electroplating apparatus may be used.

[0029]

Subsequently, an example of the Cu core ball using
25 the Cu ball as a core will be described. Figs. 1 and 2 are cross-sectional views illustrating an example of a Cu core ball 10 according to the embodiment of the present

invention.

[0030]

As illustrated in Fig. 2, in the Cu core ball 10, a core 12 containing Cu and a solder plating layer 16 formed of the solder alloy are formed via a Ni base plating layer 14 in the example, and a Sn plating layer 20 is formed on the solder plating layer 16. The Ni base plating layer 14 functions as a base plating (functions as a barrier layer) for preventing a composition change of the solder plating layer 16 due to a metal diffusion between the core 12 and the solder plating layer 16, and a thickness of the Ni base plating layer 14 on one side is about 1 μm to 4 μm . The Ni base plating layer 14 is not an indispensable requirement, and the solder plating layer 16 can be formed directly on the surface of the core 12 as illustrated in Fig. 1. When forming the base plating layer 14, the base plating layer 14 may be configured of a layer formed of one element or more selected from the group consisting of Ni, Co, P and B. More specifically, the base plating layer made of any one of Ni, Co, Ni-Co, Ni-P and Ni-B may be provided.

[0031]

Cu used in the core 12 may be pure copper or an alloy of copper.

[0032]

When the core 12 having an alloy composition containing Cu as a main component is used, purity of Cu is not particularly limited, but from a viewpoint of

suppressing deterioration of electric conductivity and thermal conductivity of the Cu core ball due to a decrease in purity and suppressing α -rays as necessary, the purity of Cu is preferably 99.9% by mass or more.

5 The core may be formed of a metal simple substance of Ni, Ag, Bi, Pb, Al, Sn, Fe, Zn, In, Ge, Sb, Co, Mn, Au, Si, Pt, Cr, La, Mo, Nb, Pd, Ti, Zr, Mg, an alloy made from two or more of these components, a metal oxide, or a metal mixture oxide in addition to Cu, and may be formed of a
10 resin material. As the resin material, a resin material formed of an amino resin, an acrylic resin, an ethylene-vinyl acetate resin, a styrene-butadiene block copolymer, a polyester resin, a melamine resin, a phenol resin, an alkyd resin, a polyimide resin, a urethane resin, an epoxy resin,
15 a crosslinked resin, or the like is exemplified. Of these, it is preferable to use conductive plastic such as polyacetylene, polypyrrole, polythiophene, and polyaniline. When the core is formed of the resin material, the Cu core ball can be formed by a resin core, a Cu plating layer
20 coating an outer side of the resin core, a base plating layer such as Ni coating a surface of the Cu plating layer, and a solder plating layer coating a surface of the base plating layer. A laminated structure of the Cu core ball is not limited to the above example.

25 [0033]

When the core 12 is spherical, from a viewpoint of controlling a stand-off height, sphericity of the core 12

is preferably 0.95 or more, and more preferably 0.990 or more. When the sphericity of the core 12 is less than 0.95, the core 12 has an indefinite shape. Therefore, a bump having a height which is not uniform is formed at the time of forming the bump, and the possibility of causing defective bonding increases. Further, when the Cu core ball 10 is mounted on the electrode and reflowed, and the sphericity is low, the Cu core ball 10 is misaligned and a self-alignment property deteriorates.

10 [0034]

Here, the sphericity represents a deviation from the sphere. The sphericity is obtained by various methods such as a least squares center method (LSC method), a minimum zone center method (MZC method), a maximum inscribed center method (MIC method), and a minimum circumscribed center method (MCC method). Specifically, the sphericity is an arithmetic average value calculated when the diameter of each of 500 cores is divided by a long diameter, and the closer the value is to the upper limit of 1.00, the closer to the sphere. The length of the long diameter is a length measured by ULTRA QV350-PRO measurement device (ultra quick vision manufactured by MITUTOYO CORPORATION).

[0035]

It is preferable that the diameter of the entire Cu core ball 10 including the solder plating layer 16 is 1 μm to 1000 μm . Within this range, the spherical Cu core ball 10 can be stably manufactured, and by selecting the

particle diameter, it is possible to suppress a connection short circuit when there is a narrow pitch between the electrode terminals.

[0036]

5 The solder plating layer 16 is made of a solder alloy, and includes Sn and at least any one element of Ag, Cu, Sb, Ni, Co, Ge, Ga, Fe, Al, In, Cd, Zn, Pb, Au, P, S, Si, Ti, Mg, Pd, and Pt in the present embodiment.

[0037]

10 The thickness of the solder plating layer 16 varies depending on the particle diameter of the core 12, but the thickness of the solder plating layer 16 on one side in the diameter direction is preferably 100 μm or less. For example, when the core 12 has a particle diameter of 300
15 $\mu\text{m}\phi$, the solder plating layer 16 is formed so as to have a thickness of 1 μm to 100 μm on one side. This is for securing a sufficient amount of solder bonding.

As the plating liquid, a known plating liquid can be appropriately used according to the solder composition to
20 be formed.

[Example]

[0038]

Hereinafter, the embodiment will be described in detail with reference to Examples and Comparative Examples.
25 The embodiment is not limited to these Examples.

[0039]

As an evaluation of the core ball having a

composition shown in each Table which will be described later, an evaluation of yellowness and an evaluation of a defect at the time of bonding were performed based on the following criteria.

5 <Yellowness>

The yellowness of the core ball was measured by using a CM-2600d spectrophotometer manufactured by KONICA MINOLTA. The yellowness of 8.5 or less in the L*a*b* table color system was set to "Good" and indicated as "o" in each Table. The yellowness of more than 8.5 in the L*a*b* table color system was determined to be "defective" and indicated as "x" in each Table.

 <Defect at the time of bonding>

Ten core balls (samples) having a composition in each Example and each Comparative Example, which are produced in the same production batch as each sample of which yellowness was measured, were prepared, and each core ball was bonded to a substrate by normal reflow processing. Regarding defects at the time of bonding, a case where a bonding defect was not measured in all the ten samples was indicated as "Good" and indicated as "o" in each Table. On the other hand, a case where misalignment or a ball missing is caused at the time of bonding in even one sample or a case where a core bursts open as the time of bonding in even one sample was determined to be "defective", and indicated as "x" in Table.

[0040]

Core balls 10 in Examples and Comparative Examples were prepared under the following conditions.

- Diameter of core 12: 300 μm

- Film thickness of Ni base plating layer 14: 2 μm on
5 one side

- Film thickness of solder plating layer 16: 1 μm on one side, 18 μm or 38 μm on one side

- Diameter of core ball 10 excluding Sn plating layer 20: 306 μm , 340 μm or 380 μm

10 As a method for plating the Ni base plating layer 14, an electroplating method (barrel electroplating) was used for a sample having a core made of Cu. An electroless plating method was used for a sample having a core made of a resin. Aluminum as a core was subjected to a double
15 zincate treatment for a sample having a core made of aluminum. Ni plating was thinly applied to the treated aluminum by an electroless plating method, and the Ni plating was then grown to a desired thickness by an electroplating method.

20 The solder plating layer 16 was plated by an electroplating method (barrel electroplating).

[0041]

The Sn plating layer 20 was formed by pulling up the core ball including the solder plating layer 16 from a
25 molten liquid containing Sn, Ag and Cu, Cu, Ag, Sb, In or Pb, and then immersing the core ball in a molten liquid of Sn. More specifically, the sample on which the solder

plating layer 16 was formed was taken out from a plating liquid containing Sn, Ag and Cu, Cu, Ag, Sb, In or Pb, washed, and dried.

After that, the sample was put into a plating liquid containing methane sulfonic acid Sn, an organic acid and a surfactant, and the barrel electroplating was performed until the sample has the thickness of the Sn plating layer 20 described in each Example and each Comparative example (excluding Comparative example in which the Sn plating layer 20 is not formed).

[0042]

In the following Table 1, aspects in which copper is used as the material of the core and the solder layer composition is made of SAC 305 are shown as Examples 1 to 9. Also in the following Table 2, aspects in which copper is used as the material of the core and the solder layer composition is made of SAC 305 are shown as Comparative Examples 1 to 9. In Examples 1 to 3 and Comparative Examples 1 to 3, the thickness of the solder layer on one side is 18 μm , but the thicknesses of the Sn layer one side are different from each other. In Examples 4 to 6 and Comparative Examples 4 to 6, the thickness of the solder layer one side is 38 μm , but the thicknesses of the Sn layer on one side are different from each other. In Examples 7 to 9 and Comparative Examples 7 to 9, the thickness of the solder layer on one side is 1 μm , but the thicknesses of the Sn layer on one side are different from

each other. The "Sn layer: solder layer" shown in each Table is calculated by an expression of Sn layer thickness (one side)/solder layer (one side) $\times 100$, and is a value obtained by dividing the thickness of the Sn layer on one side by the thickness of the solder layer on one side, and expressed in %. It has the same meaning in the other Tables. The phrase "less than the lower limit value" shown in Tables of Comparative Examples means that the thickness of the Sn layer is less than 0.215% of that of the solder layer, and the phrase "exceeding the upper limit value" shown in Tables of Comparative Examples means that the thickness of the Sn layer exceeds 36% of that of the solder layer. These have the same meanings in the other Tables. In all of Examples 1 to 9, good results could be confirmed to be obtained in each of yellowness and a defect at the time of bonding. Meanwhile, in all of Comparative Examples 1 to 9, any of yellowness and the time of bonding could be confirmed to be poor. In particular, it could be confirmed that when the Sn layer is not provided or when the thickness of the Sn layer is less than 0.215% of that of the solder layer, a problem in yellowness occurs. Meanwhile, it could be confirmed that when the thickness of the Sn layer exceeds 36% of that of the solder layer, a defect at the time of bonding occurs.

[Table 1]

	EXAMPLE 1	EXAMPLE 2	EXAMPLE 3	EXAMPLE 4	EXAMPLE 5	EXAMPLE 6v	EXAMPLE 7	EXAMPLE 8	EXAMPLE 9
CORE MATERIAL	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER
THICKNESS OF SOLDER LAYER	18 µm ON ONE SIDE	18 µm ON ONE SIDE	18 µm ON ONE SIDE	38 µm ON ONE SIDE	38 µm ON ONE SIDE	38 µm ON ONE SIDE	1 µm ON ONE SIDE	1 µm ON ONE SIDE	1 µm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305
THICKNESS OF Sn LAYER (µm)	0.1	0.5	6	0.1	0.5	12	0.1	0.23	0.36
Sn LAYER: SOLDER LAYER	0.556	2.778	33.333	0.263	1.316	31.579	10.000	23.000	36.000
YELLOWNESS	○	○	○	○	○	○	○	○	○
DEFECT AT THE TIME OF BONDING	○	○	○	○	○	○	○	○	○

[Table 2]

	COMPARATIVE EXAMPLE 1	COMPARATIVE EXAMPLE 2	COMPARATIVE EXAMPLE 3	COMPARATIVE EXAMPLE 4	COMPARATIVE EXAMPLE 5	COMPARATIVE EXAMPLE 6	COMPARATIVE EXAMPLE 7	COMPARATIVE EXAMPLE 8	COMPARATIVE EXAMPLE 9
CORE MATERIAL	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER
THICKNESS OF SOLDER LAYER	18 µm ON ONE SIDE	18 µm ON ONE SIDE	18 µm ON ONE SIDE	38 µm ON ONE SIDE	38 µm ON ONE SIDE	38 µm ON ONE SIDE	1 µm ON ONE SIDE	1 µm ON ONE SIDE	1 µm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305
THICKNESS OF Sn LAYER (µm)	0	0.03	10	0	0.03	15	0	0.002	0.5
Sn LAYER: SOLDER LAYER	0.000	0.167	55.556	0.000	0.079	39.474	0.000	0.200	50.000
	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE
YELLOWNESS	×	×	○	×	×	○	×	×	○
DEFECT AT THE TIME OF BONDING	○	○	×	○	○	×	○	○	×

[0043]

In the following Table 3, aspects in which copper is used as the material of the core and the solder layer composition is made of SnCu (Cu 0.7 Sn balance) are shown as Examples 10 to 18. Also in the following Table 4, aspects in which copper is used as the material of the core and the solder layer composition is made of SnCu are shown as Comparative Examples 10 to 18. In Examples 10 to 12 and Comparative Examples 10 to 12, the thickness of the solder layer on one side is 18 μm , but the thicknesses of the Sn layer on one side are different from each other. In Examples 13 to 15 and Comparative Examples 13 to 15, the thickness of the solder layer on one side is 38 μm , but the thicknesses of the Sn layer on one side are different from each other. In Examples 16 to 18 and Comparative Examples 16 to 18, the thickness of the solder layer on one side is 1 μm , but the thicknesses of the Sn layer on one side are different from each other. In all of Examples 10 to 18, good results could be confirmed to be obtained in each of yellowness and a defect at the time of bonding. Meanwhile, in all of Comparative Examples 10 to 18, any of yellowness and the time of bonding could be confirmed to be poor. In particular, it could be confirmed that when the Sn layer is not provided or when the thickness of the Sn layer is less than 0.215% of that of the solder layer, a problem in yellowness occurs. Meanwhile, it could be confirmed that when the thickness of the Sn layer exceeds 36% of that of

the solder layer, a defect at the time of bonding occurs.

[Table 3]

	EXAMPLE 10	EXAMPLE 11	EXAMPLE 12	EXAMPLE 13	EXAMPLE 14	EXAMPLE 15	EXAMPLE 16	EXAMPLE 17	EXAMPLE 18
CORE MATERIAL	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu
THICKNESS OF Sn LAYER (μm)	0.1	0.5	6	0.1	0.5	12	0.1	0.23	0.36
Sn LAYER: SOLDER LAYER	0.556	2.778	33.333	0.263	1.316	31.579	10.000	23.000	36.000
YELLOWNESS	○	○	○	○	○	○	○	○	○
DEFECT AT THE TIME OF BONDING	○	○	○	○	○	○	○	○	○

[Table 4]

	COMPARATIVE EXAMPLE 10	COMPARATIVE EXAMPLE 11	COMPARATIVE EXAMPLE 12	COMPARATIVE EXAMPLE 13	COMPARATIVE EXAMPLE 14	COMPARATIVE EXAMPLE 15	COMPARATIVE EXAMPLE 16	COMPARATIVE EXAMPLE 17	COMPARATIVE EXAMPLE 18
CORE MATERIAL	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu
THICKNESS OF Sn LAYER (μm)	0	0.03	10	0	0.03	15	0	0.002	0.5
Sn LAYER: SOLDER LAYER	0.000	0.167	55.556	0.000	0.079	39.474	0.000	0.200	50.000
	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE
YELLOWNESS	×	×	○	×	×	○	×	×	○
DEFECT AT THE TIME OF BONDING	○	○	×	○	○	×	○	○	×

[0044]

In the following Table 5, aspects in which copper is used as the material of the core and the solder layer composition is made of SnAg (Ag 3.5 Sn balance) are shown as Examples 19 to 27. Also in the following Table 6, aspects in which copper is used as the material of the core and the solder layer composition is made of SnAg are shown as Comparative Examples 19 to 27. In Examples 19 to 21 and Comparative Examples 19 to 21, the thickness of the solder layer one side is 18 μm , but the thicknesses of the Sn layer one side are different from each other. In Examples 22 to 24 and Comparative Examples 22 to 24, the thickness of the solder layer on one side is 38 μm , but the thicknesses of the Sn layer on one side are different from each other. In Examples 25 to 27 and Comparative Examples 25 to 27, the thickness of the solder layer on one side is 1 μm , but the thicknesses of the Sn layer on one side are different from each other. In all of Examples 19 to 27, good results could be confirmed to be obtained in each of yellowness and a defect at the time of bonding. Meanwhile, in all of Comparative Examples 19 to 27, any of yellowness and the time of bonding could be confirmed to be poor. In particular, it could be confirmed that when the Sn layer is not provided or when the thickness of the Sn layer is less than 0.215% of that of the solder layer, a problem in yellowness occurs. Meanwhile, it could be confirmed that when the thickness of the Sn layer exceeds 36% of that of

the solder layer, a defect at the time of bonding occurs.

[Table 5]

	EXAMPLE 19	EXAMPLE 20	EXAMPLE 21	EXAMPLE 22	EXAMPLE 23	EXAMPLE 24	EXAMPLE 25	EXAMPLE 26	EXAMPLE 27
CORE MATERIAL	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg
THICKNESS OF Sn LAYER (μm)	0.1	0.5	6	0.1	0.5	12	0.1	0.23	0.36
Sn LAYER: SOLDER LAYER	0.556	2.778	33.333	0.263	1.316	31.579	10.000	23.000	36.000
YELLOWNESS	○	○	○	○	○	○	○	○	○
DEFECT AT THE TIME OF BONDING	○	○	○	○	○	○	○	○	○

[Table 6]

	COMPARATIVE EXAMPLE 19	COMPARATIVE EXAMPLE 20	COMPARATIVE EXAMPLE 21	COMPARATIVE EXAMPLE 22	COMPARATIVE EXAMPLE 23	COMPARATIVE EXAMPLE 24	COMPARATIVE EXAMPLE 25	COMPARATIVE EXAMPLE 26	COMPARATIVE EXAMPLE 27
CORE MATERIAL	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg
THICKNESS OF Sn LAYER (μm)	0	0.03	10	0	0.03	15	0	0.002	0.5
Sn LAYER: SOLDER LAYER	0.000	0.167	55.556	0.000	0.079	39.474	0.000	0.200	50.000
	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE
YELLOWNESS	×	×	○	×	×	○	×	×	○
DEFECT AT THE TIME OF BONDING	○	○	×	○	○	×	○	○	×

[0045]

In the following Table 7, aspects in which copper is used as the material of the core and the solder layer composition is made of SnSb (Sb 10 Sn balance) are shown as
5 Examples 28 to 36. Also in the following Table 8, aspects in which copper is used as the material of the core and the solder layer composition is made of SnSb are shown as Comparative Examples 28 to 36. In Examples 28 to 30 and Comparative Examples 28 to 30, the thickness of the solder
10 layer on one side is 18 μm , but the thicknesses of the Sn layer on one side are different from each other. In Examples 31 to 33 and Comparative Examples 31 to 33, the thickness of the solder layer on one side is 38 μm , but the thicknesses of the Sn layer on one side are different from
15 each other. In Examples 34 to 36 and Comparative Examples 34 to 36, the thickness of the solder layer on one side is 1 μm , but the thicknesses of the Sn layer on one side are different from each other. In all of Examples 28 to 36, good results could be confirmed to be obtained in each of
20 yellowness and a defect at the time of bonding. Meanwhile, in all of Comparative Examples 28 to 36, any of yellowness and the time of bonding could be confirmed to be poor. In particular, it could be confirmed that when the Sn layer is not provided or when the thickness of the Sn layer is less
25 than 0.215% of that of the solder layer, a problem in yellowness occurs. Meanwhile, it could be confirmed that when the thickness of the Sn layer exceeds 36% of that of

the solder layer, a defect at the time of bonding occurs.

[Table 7]

	EXAMPLE 28	EXAMPLE 29	EXAMPLE 30	EXAMPLE 31	EXAMPLE 32	EXAMPLE 33	EXAMPLE 34	EXAMPLE 35	EXAMPLE 36
CORE MATERIAL	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnSb	SnSb	SnSb	SnSb	SnSb	SnSb	SnSb	SnSb	SnSb
THICKNESS OF Sn LAYER (μm)	0.1	0.5	6	0.1	0.5	12	0.1	0.23	0.36
Sn LAYER: SOLDER LAYER	0.556	2.778	33.333	0.263	1.316	31.579	10.000	23.000	36.000
YELLOWNESS	○	○	○	○	○	○	○	○	○
DEFECT AT THE TIME OF BONDING	○	○	○	○	○	○	○	○	○

[Table 8]

	COMPARATIVE EXAMPLE 28	COMPARATIVE EXAMPLE 29	COMPARATIVE EXAMPLE 30	COMPARATIVE EXAMPLE 31	COMPARATIVE EXAMPLE 32	COMPARATIVE EXAMPLE 33	COMPARATIVE EXAMPLE 34	COMPARATIVE EXAMPLE 35	COMPARATIVE EXAMPLE 36
CORE MATERIAL	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnSb	SnSb	SnSb	SnSb	SnSb	SnSb	SnSb	SnSb	SnSb
THICKNESS OF Sn LAYER (μm)	0	0.03	10	0	0.03	15	0	0.002	0.5
Sn LAYER: SOLDER LAYER	0.000	0.167	55.556	0.000	0.079	39.474	0.000	0.200	50.000
	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE
YELLOWNESS	×	×	○	×	×	○	×	×	○
DEFECT AT THE TIME OF BONDING	○	○	×	○	○	×	○	○	×

[0046]

In the following Table 9, aspects in which copper is used as the material of the core and the solder layer composition is made of SnIn (In 45 Sn balance) is shown as Examples 37 to 45. Also in the following Table 10, aspects in which copper is used as the material of the core and the solder layer composition is made of SnIn are shown as Comparative Examples 37 to 45. In Examples 37 to 39 and Comparative Examples 37 to 39, the thickness of the solder layer on one side is 18 μm , but the thicknesses of the Sn layer on one side are different from each other. In Examples 40 to 42 and Comparative Examples 40 to 42, the thickness of the solder layer on one side is 38 μm , but the thicknesses of the Sn layer on one side are different from each other. In Examples 43 to 45 and Comparative Examples 43 to 45, the thickness of the solder layer on one side is 1 μm , but the thicknesses of the Sn layer on one side are different from each other. In all of Examples 37 to 45, good results could be confirmed to be obtained in each of yellowness and a defect at the time of bonding. Meanwhile, in all of Comparative Examples 37 to 45, any of yellowness and the time of bonding could be confirmed to be poor. In particular, it could be confirmed that when the Sn layer is not provided or when the thickness of the Sn layer is less than 0.215% of that of the solder layer, a problem in yellowness occurs. Meanwhile, it could be confirmed that when the thickness of the Sn layer exceeds 36% of that of

the solder layer, a defect at the time of bonding occurs.

[Table 9]

	EXAMPLE 37	EXAMPLE 38	EXAMPLE 39	EXAMPLE 40	EXAMPLE 41	EXAMPLE 42	EXAMPLE 43	EXAMPLE 44	EXAMPLE 45
CORE MATERIAL	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn
THICKNESS OF Sn LAYER (μm)	0.1	0.5	6	0.1	0.5	12	0.1	0.23	0.36
Sn LAYER: SOLDER LAYER	0.556	2.778	33.333	0.263	1.316	31.579	10.000	23.000	36.000
YELLOWNESS	○	○	○	○	○	○	○	○	○
DEFECT AT THE TIME OF BONDING	○	○	○	○	○	○	○	○	○

[Table 10]

	COMPARATIVE EXAMPLE 37	COMPARATIVE EXAMPLE 38	COMPARATIVE EXAMPLE 39	COMPARATIVE EXAMPLE 40	COMPARATIVE EXAMPLE 41	COMPARATIVE EXAMPLE 42	COMPARATIVE EXAMPLE 43	COMPARATIVE EXAMPLE 44	COMPARATIVE EXAMPLE 45
CORE MATERIAL	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn
THICKNESS OF Sn LAYER (μm)	0	0.03	10	0	0.03	15	0	0.002	0.5
Sn LAYER: SOLDER LAYER	0.000	0.167	55.556	0.000	0.079	39.474	0.000	0.200	50.000
	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE
YELLOWNESS	×	×	○	×	×	○	×	×	○
DEFECT AT THE TIME OF BONDING	○	○	×	○	○	×	○	○	×

[0047]

In the following Table 11, aspects in which copper is used as the material of the core and the solder layer composition is made of SnPb (Pb 37 Sn balance) are shown as
5 Examples 46 to 54. Also in the following Table 12, aspects in which copper is used as the material of the core and the solder layer composition is made of SnPb are shown as Comparative Examples 46 to 54. In Examples 46 to 48 and Comparative Examples 46 to 48, the thickness of the solder
10 layer on one side is 18 μm , but the thicknesses of the Sn layer on one side are different from each other. In Examples 49 to 51 and Comparative Examples 49 to 51, the thickness of the solder layer on one side is 38 μm , but the thicknesses of the Sn layer on one side are different from
15 each other. In Examples 52 to 54 and Comparative Examples 52 to 54, the thickness of the solder layer on one side is 1 μm , but the thicknesses of the Sn layer on one side are different from each other. In all of Examples 46 to 54, good results could be confirmed to be obtained in each of
20 yellowness and a defect at the time of bonding. Meanwhile, in all of Comparative Examples 46 to 54, any of yellowness and the time of bonding could be confirmed to be poor. In particular, it could be confirmed that when the Sn layer is not provided or when the thickness of the Sn layer is less
25 than 0.215% of that of the solder layer, a problem in yellowness occurs. Meanwhile, it could be confirmed that when the thickness of the Sn layer exceeds 36% of that of

the solder layer, a defect at the time of bonding occurs.

[Table 11]

	EXAMPLE 46	EXAMPLE 47	EXAMPLE 48	EXAMPLE 49	EXAMPLE 50	EXAMPLE 51	EXAMPLE 52	EXAMPLE 53	EXAMPLE 54
CORE MATERIAL	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnPb	SnPb	SnPb	SnPb	SnPb	SnPb	SnPb	SnPb	SnPb
THICKNESS OF Sn LAYER (μm)	0.1	0.5	6	0.1	0.5	12	0.1	0.23	0.36
Sn LAYER: SOLDER LAYER	0.556	2.778	33.333	0.263	1.316	31.579	10.000	23.000	36.000
YELLOWNESS	○	○	○	○	○	○	○	○	○
DEFECT AT THE TIME OF BONDING	○	○	○	○	○	○	○	○	○

[Table 12]

	COMPARATIVE EXAMPLE 46	COMPARATIVE EXAMPLE 47	COMPARATIVE EXAMPLE 48	COMPARATIVE EXAMPLE 49	COMPARATIVE EXAMPLE 50	COMPARATIVE EXAMPLE 51	COMPARATIVE EXAMPLE 52	COMPARATIVE EXAMPLE 53	COMPARATIVE EXAMPLE 54
CORE MATERIAL	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnPb	SnPb	SnPb	SnPb	SnPb	SnPb	SnPb	SnPb	SnPb
THICKNESS OF Sn LAYER (μm)	0	0.03	10	0	0.03	15	0	0.002	0.5
Sn LAYER: SOLDER LAYER	0.000	0.167	55.556	0.000	0.079	39.474	0.000	0.200	50.000
	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE
YELLOWNESS	×	×	○	×	×	○	×	×	○
DEFECT AT THE TIME OF BONDING	○	○	×	○	○	×	○	○	×

[0048]

In the following Tables 13 to 24, the solder layer composition was changed to SAC 305, SnCu (Cu 0.7 Sn balance), SnAg (Ag 3.5 Sn balance), SnSb (Sb 10 Sn balance), SnIn (In 45 Sn balance), and SnPb (Pb 37 Sn balance), and the thickness of the solder layer on one side and the thickness of the Sn layer on one side were changed in the same manner as in Examples 1 to 54 and Comparative Examples 1 to 54 shown in Tables 1 to 12 except that aluminum was used as the material of the core. In all of Examples 55 to 108, good results could be confirmed to be obtained in each of yellowness and a defect at the time of bonding. Meanwhile, in all of Comparative Examples 55 to 108, any of yellowness and the time of bonding could be confirmed to be poor. In particular, it could be confirmed that when the Sn layer is not provided or when the thickness of the Sn layer is less than 0.215% of that of the solder layer, a problem in yellowness occurs. Meanwhile, it could be confirmed that when the thickness of the Sn layer exceeds 36% of that of the solder layer, a defect at the time of bonding occurs.

[Table 13]

	EXAMPLE 55	EXAMPLE 56	EXAMPLE 57	EXAMPLE 58	EXAMPLE 59	EXAMPLE 60	EXAMPLE 61	EXAMPLE 62	EXAMPLE 63
CORE MATERIAL	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305
THICKNESS OF Sn LAYER (μm)	0.1	0.5	6	0.1	0.5	12	0.1	0.23	0.36
Sn LAYER: SOLDER LAYER	0.556	2.778	33.333	0.263	1.316	31.579	10.000	23.000	36.000
YELLOWNESS	○	○	○	○	○	○	○	○	○
DEFECT AT THE TIME OF BONDING	○	○	○	○	○	○	○	○	○

[Table 14]

	COMPARATIVE EXAMPLE 55	COMPARATIVE EXAMPLE 56	COMPARATIVE EXAMPLE 57	COMPARATIVE EXAMPLE 58	COMPARATIVE EXAMPLE 59	COMPARATIVE EXAMPLE 60	COMPARATIVE EXAMPLE 61	COMPARATIVE EXAMPLE 62	COMPARATIVE EXAMPLE 63
CORE MATERIAL	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305
THICKNESS OF Sn LAYER (μm)	0	0.03	10	0	0.03	15	0	0.002	0.5
Sn LAYER: SOLDER LAYER	0.000	0.167	55.556	0.000	0.079	39.474	0.000	0.200	50.000
	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE
YELLOWNESS	×	×	○	×	×	○	×	×	○
DEFECT AT THE TIME OF BONDING	○	○	×	○	○	×	○	○	×

[Table 15]

	EXAMPLE 64	EXAMPLE 65	EXAMPLE 66	EXAMPLE 67	EXAMPLE 68	EXAMPLE 69	EXAMPLE 70	EXAMPLE 71	EXAMPLE 72
CORE MATERIAL	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu
THICKNESS OF Sn LAYER (μm)	0.1	0.5	6	0.1	0.5	12	0.1	0.23	0.36
Sn LAYER: SOLDER LAYER	0.556	2.778	33.333	0.263	1.316	31.579	10.000	23.000	36.000
YELLOWNESS	○	○	○	○	○	○	○	○	○
DEFECT AT THE TIME OF BONDING	○	○	○	○	○	○	○	○	○

[Table 16]

	COMPARATIVE EXAMPLE 64	COMPARATIVE EXAMPLE 65	COMPARATIVE EXAMPLE 66	COMPARATIVE EXAMPLE 67	COMPARATIVE EXAMPLE 68	COMPARATIVE EXAMPLE 69	COMPARATIVE EXAMPLE 70	COMPARATIVE EXAMPLE 71	COMPARATIVE EXAMPLE 72
CORE MATERIAL	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu
THICKNESS OF Sn LAYER (μm)	0	0.03	10	0	0.03	15	0	0.002	0.5
Sn LAYER: SOLDER LAYER	0.000	0.167	55.556	0.000	0.079	39.474	0.000	0.200	50.000
	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE
YELLOWNESS	×	×	○	×	×	○	×	×	○
DEFECT AT THE TIME OF BONDING	○	○	×	○	○	×	○	○	×

[Table 17]

	EXAMPLE 73	EXAMPLE 74	EXAMPLE 75	EXAMPLE 76	EXAMPLE 77	EXAMPLE 78	EXAMPLE 79	EXAMPLE 80	EXAMPLE 81
CORE MATERIAL	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg
THICKNESS OF Sn LAYER (μm)	0.1	0.5	6	0.1	0.5	12	0.1	0.23	0.36
Sn LAYER: SOLDER LAYER	0.556	2.778	33.333	0.263	1.316	31.579	10.000	23.000	36.000
YELLOWNESS	○	○	○	○	○	○	○	○	○
DEFECT AT THE TIME OF BONDING	○	○	○	○	○	○	○	○	○

[Table 18]

	COMPARATIVE EXAMPLE 73	COMPARATIVE EXAMPLE 74	COMPARATIVE EXAMPLE 75	COMPARATIVE EXAMPLE 76	COMPARATIVE EXAMPLE 77	COMPARATIVE EXAMPLE 78	COMPARATIVE EXAMPLE 79	COMPARATIVE EXAMPLE 80	COMPARATIVE EXAMPLE 81
CORE MATERIAL	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg
THICKNESS OF Sn LAYER (μm)	0	0.03	10	0	0.03	15	0	0.002	0.5
Sn LAYER: SOLDER LAYER	0.000	0.167	55.556	0.000	0.079	39.474	0.000	0.200	50.000
	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE
YELLOWNESS	×	×	○	×	×	○	×	×	○
DEFECT AT THE	○	○	×	○	○	×	○	○	×

YELLOWNESS	×	×	○	×	×	○	×	×	○
DEFECT AT THE TIME OF BONDING	○	○	×	○	○	×	○	○	×

[Table 21]

	EXAMPLE 91	EXAMPLE 92	EXAMPLE 93	EXAMPLE 94	EXAMPLE 95	EXAMPLE 96	EXAMPLE 97	EXAMPLE 98	EXAMPLE 99
CORE MATERIAL	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn
THICKNESS OF Sn LAYER (μm)	0.1	0.5	6	0.1	0.5	12	0.1	0.23	0.36
Sn LAYER: SOLDER LAYER	0.556	2.778	33.333	0.263	1.316	31.579	10.000	23.000	36.000
YELLOWNESS	○	○	○	○	○	○	○	○	○
DEFECT AT THE TIME OF BONDING	○	○	○	○	○	○	○	○	○

[Table 22]

	COMPARATIVE EXAMPLE 91	COMPARATIVE EXAMPLE 92	COMPARATIVE EXAMPLE 93	COMPARATIVE EXAMPLE 94	COMPARATIVE EXAMPLE 95	COMPARATIVE EXAMPLE 96	COMPARATIVE EXAMPLE 97	COMPARATIVE EXAMPLE 98	COMPARATIVE EXAMPLE 99
CORE MATERIAL	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn
THICKNESS OF Sn LAYER (μm)	0	0.03	10	0	0.03	15	0	0.002	0.5
Sn LAYER: SOLDER LAYER	0.000	0.167	55.556	0.000	0.079	39.474	0.000	0.200	50.000
	0:100	LESS THAN LOWER LIMIT	EXCEEDING UPPER LIMIT	0:100	LESS THAN LOWER LIMIT	EXCEEDING UPPER LIMIT	0:100	LESS THAN LOWER LIMIT	EXCEEDING UPPER LIMIT

		VALUE	VALUE		VALUE	VALUE		VALUE	VALUE
YELLOWNESS	×	×	○	×	×	○	×	×	○
DEFECT AT THE TIME OF BONDING	○	○	×	○	○	×	○	○	×

[Table 23]

	EXAMPLE 100	EXAMPLE 101	EXAMPLE 102	EXAMPLE 103	EXAMPLE 104	EXAMPLE 105	EXAMPLE 106	EXAMPLE 107	EXAMPLE 108
CORE MATERIAL	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnPb	SnPb	SnPb	SnPb	SnPb	SnPb	SnPb	SnPb	SnPb
THICKNESS OF Sn LAYER (μm)	0.1	0.5	6	0.1	0.5	12	0.1	0.23	0.36
Sn LAYER: SOLDER LAYER	0.556	2.778	33.333	0.263	1.316	31.579	10.000	23.000	36.000
YELLOWNESS	○	○	○	○	○	○	○	○	○
DEFECT AT THE TIME OF BONDING	○	○	○	○	○	○	○	○	○

[Table 24]

	COMPARATIVE EXAMPLE 100	COMPARATIVE EXAMPLE 101	COMPARATIVE EXAMPLE 102	COMPARATIVE EXAMPLE 103	COMPARATIVE EXAMPLE 104	COMPARATIVE EXAMPLE 105	COMPARATIVE EXAMPLE 106	COMPARATIVE EXAMPLE 107	COMPARATIVE EXAMPLE 108
CORE MATERIAL	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM	ALUMINUM
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnPb	SnPb	SnPb	SnPb	SnPb	SnPb	SnPb	SnPb	SnPb
THICKNESS OF Sn LAYER (μm)	0	0.03	10	0	0.03	15	0	0.002	0.5

Sn LAYER: SOLDER LAYER	0.000	0.167	55.556	0.000	0.079	39.474	0.000	0.200	50.000
	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE
YELLOWNESS	×	×	○	×	×	○	×	×	○
DEFECT AT THE TIME OF BONDING	○	○	×	○	○	×	○	○	×

[0049]

In the following Tables 25 to 36, the solder layer composition was changed to SAC 305, SnCu (Cu 0.75 Sn balance), SnAg (Ag 3.5 Sn balance), SnSb (Sb 10 Sn balance), SnIn (In 45 Sn balance), and SnPb (Pb 37 Sn balance), and the thickness of the solder layer on one side and the thickness of the Sn layer on one side were changed in the same manner as in Examples 1 to 54 and Comparative Examples 1 to 54 shown in Tables 1 to 12, and Examples 55 to 108 and Comparative Examples 55 to 108 shown in Tables 13 to 24 except that a styrene-butadiene block copolymer as a resin was used as the material of the core. In all of Examples 109 to 162, good results could be confirmed to be obtained in each of yellowness and a defect at the time of bonding. Meanwhile, in all of Comparative Examples 109 to 162, any of yellowness and the time of bonding could be confirmed to be poor. In particular, it could be confirmed that when the Sn layer is not provided or when the thickness of the Sn layer is less than 0.215% of that of the solder layer, a problem in yellowness occurs. Meanwhile, it could be confirmed that when the thickness of the Sn layer exceeds 36% of that of the solder layer, a defect at the time of bonding occurs.

[Table 25]

	EXAMPLE 109	EXAMPLE 110	EXAMPLE 111	EXAMPLE 112	EXAMPLE 113	EXAMPLE 114	EXAMPLE 115	EXAMPLE 116	EXAMPLE 117
CORE MATERIAL	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305
THICKNESS OF Sn LAYER (μm)	0.1	0.5	6	0.1	0.5	12	0.1	0.23	0.36
Sn LAYER: SOLDER LAYER	0.556	2.778	33.333	0.263	1.316	31.579	10.000	23.000	36.000
YELLOWNESS	○	○	○	○	○	○	○	○	○
DEFECT AT THE TIME OF BONDING	○	○	○	○	○	○	○	○	○

[Table 26]

	COMPARATIVE EXAMPLE 109	COMPARATIVE EXAMPLE 110	COMPARATIVE EXAMPLE 111	COMPARATIVE EXAMPLE 112	COMPARATIVE EXAMPLE 113	COMPARATIVE EXAMPLE 114	COMPARATIVE EXAMPLE 115	COMPARATIVE EXAMPLE 116	COMPARATIVE EXAMPLE 117
CORE MATERIAL	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305
THICKNESS OF Sn LAYER (μm)	0	0.03	10	0	0.03	15	0	0.002	0.5
Sn LAYER: SOLDER LAYER	0.000	0.167	55.556	0.000	0.079	39.474	0.000	0.200	50.000
	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE
YELLOWNESS	×	×	○	×	×	○	×	×	○
DEFECT AT THE TIME OF BONDING	○	○	×	○	○	×	○	○	×

[Table 27]

	EXAMPLE 118	EXAMPLE 119	EXAMPLE 120	EXAMPLE 121	EXAMPLE 122	EXAMPLE 123	EXAMPLE 124	EXAMPLE 125	EXAMPLE 126
CORE MATERIAL	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu
THICKNESS OF Sn LAYER (μm)	0.1	0.5	6	0.1	0.5	12	0.1	0.23	0.36
Sn LAYER: SOLDER LAYER	0.556	2.778	33.333	0.263	1.316	31.579	10.000	23.000	36.000
YELLOWNESS	○	○	○	○	○	○	○	○	○
DEFECT AT THE TIME OF BONDING	○	○	○	○	○	○	○	○	○

[Table 28]

	COMPARATIVE EXAMPLE 118	COMPARATIVE EXAMPLE 119	COMPARATIVE EXAMPLE 120	COMPARATIVE EXAMPLE 121	COMPARATIVE EXAMPLE 122	COMPARATIVE EXAMPLE 123	COMPARATIVE EXAMPLE 124	COMPARATIVE EXAMPLE 125	COMPARATIVE EXAMPLE 126
CORE MATERIAL	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu	SnCu
THICKNESS OF Sn LAYER (μm)	0	0.03	10	0	0.03	15	0	0.002	0.5
Sn LAYER: SOLDER LAYER	0.000	0.167	55.556	0.000	0.079	39.474	0.000	0.200	50.000
	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE
YELLOWNESS	×	×	○	×	×	○	×	×	○

DEFECT AT THE TIME OF BONDING	○	○	×	○	○	×	○	○	×
-------------------------------	---	---	---	---	---	---	---	---	---

[Table 29]

	EXAMPLE 127	EXAMPLE 128	EXAMPLE 129	EXAMPLE 130	EXAMPLE 131	EXAMPLE 132	EXAMPLE 133	EXAMPLE 134	EXAMPLE 135
CORE MATERIAL	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg
THICKNESS OF Sn LAYER (μm)	0.1	0.5	6	0.1	0.5	12	0.1	0.23	0.36
Sn LAYER: SOLDER LAYER	0.556	2.778	33.333	0.263	1.316	31.579	10.000	23.000	36.000
YELLOWNESS	○	○	○	○	○	○	○	○	○
DEFECT AT THE TIME OF BONDING	○	○	○	○	○	○	○	○	○

[Table 30]

	COMPARATIVE EXAMPLE 127	COMPARATIVE EXAMPLE 128	COMPARATIVE EXAMPLE 129	COMPARATIVE EXAMPLE 130	COMPARATIVE EXAMPLE 131	COMPARATIVE EXAMPLE 132	COMPARATIVE EXAMPLE 133	COMPARATIVE EXAMPLE 134	COMPARATIVE EXAMPLE 135
CORE MATERIAL	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg	SnAg
THICKNESS OF Sn LAYER (μm)	0	0.03	10	0	0.03	15	0	0.002	0.5
Sn LAYER: SOLDER LAYER	0.000	0.167	55.556	0.000	0.079	39.474	0.000	0.200	50.000
	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE

YELLOWNESS	×	×	○	×	×	○	×	×	○
DEFECT AT THE TIME OF BONDING	○	○	×	○	○	×	○	○	×

[Table 31]

	EXAMPLE 136	EXAMPLE 137	EXAMPLE 138	EXAMPLE 139	EXAMPLE 140	EXAMPLE 141	EXAMPLE 142	EXAMPLE 143	EXAMPLE 144
CORE MATERIAL	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnSb	SnSb	SnSb	SnSb	SnSb	SnSb	SnSb	SnSb	SnSb
THICKNESS OF Sn LAYER (μm)	0.1	0.5	6	0.1	0.5	12	0.1	0.23	0.36
Sn LAYER: SOLDER LAYER	0.556	2.778	33.333	0.263	1.316	31.579	10.000	23.000	36.000
YELLOWNESS	○	○	○	○	○	○	○	○	○
DEFECT AT THE TIME OF BONDING	○	○	○	○	○	○	○	○	○

[Table 32]

	COMPARATIVE EXAMPLE 136	COMPARATIVE EXAMPLE 137	COMPARATIVE EXAMPLE 138	COMPARATIVE EXAMPLE 139	COMPARATIVE EXAMPLE 140	COMPARATIVE EXAMPLE 141	COMPARATIVE EXAMPLE 142	COMPARATIVE EXAMPLE 143	COMPARATIVE EXAMPLE 144
CORE MATERIAL	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnSb	SnSb	SnSb	SnSb	SnSb	SnSb	SnSb	SnSb	SnSb
THICKNESS OF Sn LAYER (μm)	0	0.03	10	0	0.03	15	0	0.002	0.5
Sn LAYER:	0.000	0.167	55.556	0.000	0.079	39.474	0.000	0.200	50.000

SOLDER LAYER	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE
YELLOWNESS	×	×	○	×	×	○	×	×	○
DEFECT AT THE TIME OF BONDING	○	○	×	○	○	×	○	○	×

[Table 33]

	EXAMPLE 145	EXAMPLE 146	EXAMPLE 147	EXAMPLE 148	EXAMPLE 149	EXAMPLE 150	EXAMPLE 151	EXAMPLE 152	EXAMPLE 153
CORE MATERIAL	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn
THICKNESS OF Sn LAYER (μm)	0.1	0.5	6	0.1	0.5	12	0.1	0.23	0.36
Sn LAYER: SOLDER LAYER	0.556	2.778	33.333	0.263	1.316	31.579	10.000	23.000	36.000
YELLOWNESS	○	○	○	○	○	○	○	○	○
DEFECT AT THE TIME OF BONDING	○	○	○	○	○	○	○	○	○

[Table 34]

	COMPARATIVE EXAMPLE 145	COMPARATIVE EXAMPLE 146	COMPARATIVE EXAMPLE 147	COMPARATIVE EXAMPLE 148	COMPARATIVE EXAMPLE 149	COMPARATIVE EXAMPLE 150	COMPARATIVE EXAMPLE 151	COMPARATIVE EXAMPLE 152	COMPARATIVE EXAMPLE 153
CORE MATERIAL	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN	RESIN
THICKNESS OF SOLDER LAYER	18 μm ON ONE SIDE	18 μm ON ONE SIDE	18 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn	SnIn
THICKNESS OF Sn	0	0.03	10	0	0.03	15	0	0.002	0.5

SOLDER LAYER									
THICKNESS OF Sn LAYER (μm)	0	0.03	10	0	0.03	15	0	0.002	0.5
Sn LAYER: SOLDER LAYER	0.000	0.167	55.556	0.000	0.079	39.474	0.000	0.200	50.000
	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE
YELLOWNESS	×	×	○	×	×	○	×	×	○
DEFECT AT THE TIME OF BONDING	○	○	×	○	○	×	○	○	×

[0050]

The present embodiment also provides a method for forming a bump electrode, the method including the steps of: mounting a core material on an electrode; and forming
5 the bump electrode by heating the mounted core material.

[0051]

The shape of the core material also includes not only a spherical body but also a columnar column and a sheet and the like. For example, a core column in which a Ni base
10 plating layer and Co base plating layer and the like of 1 to 4 μm on one side are provided on the surface of a Cu column having diameters of top and bottom surfaces of 1 to 1000 μm and a height of 1 to 3000 μm , and in the same condition as that of Examples, the Cu column is covered
15 with a solder plating layer can provide the same effects as those in Examples described above, and good results could be obtained in each of yellowness and a defect at the time of bonding.

[0052]

20 As an example, the results of using an aspect in which a column having a diameter ϕ of 0.30 mm and a height of 0.53 mm is used and the Ni plating thickness is 2 μm on one side are shown below. The thickness of the solder layer is a value not including the Ni plating thickness.

[Table 37]

	EXAMPLE 163	EXAMPLE 164	EXAMPLE 165	EXAMPLE 166	EXAMPLE 167	EXAMPLE 168	EXAMPLE 169	EXAMPLE 170	EXAMPLE 171
CORE MATERIAL	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER
THICKNESS OF SOLDER LAYER	5.5 μm ON ONE SIDE	5.5 μm ON ONE SIDE	5.5 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305
THICKNESS OF Sn LAYER (μm)	0.1	0.5	1.8	0.1	0.5	12	0.1	0.23	0.36
Sn LAYER: SOLDER LAYER	1.818	9.091	32.727	0.263	1.316	31.579	10.000	23.000	36.000
YELLOWNESS	○	○	○	○	○	○	○	○	○
DEFECT AT THE TIME OF BONDING	○	○	○	○	○	○	○	○	○

[Table 38]

	COMPARATIVE EXAMPLE 163	COMPARATIVE EXAMPLE 164	COMPARATIVE EXAMPLE 165	COMPARATIVE EXAMPLE 166	COMPARATIVE EXAMPLE 167	COMPARATIVE EXAMPLE 168	COMPARATIVE EXAMPLE 169	COMPARATIVE EXAMPLE 170	COMPARATIVE EXAMPLE 171
CORE MATERIAL	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER	COPPER
THICKNESS OF SOLDER LAYER	5.5 μm ON ONE SIDE	5.5 μm ON ONE SIDE	5.5 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	38 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE	1 μm ON ONE SIDE
COMPOSITION OF SOLDER LAYER	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305	SAC305
THICKNESS OF Sn LAYER (μm)	0	0.01	2.5	0	0.03	15	0	0.002	0.5
Sn LAYER: SOLDER LAYER	0.000	0.182	45.455	0.000	0.079	39.474	0.000	0.200	50.000
	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE	0:100	LESS THAN LOWER LIMIT VALUE	EXCEEDING UPPER LIMIT VALUE
YELLOWNESS	×	×	○	×	×	○	×	×	○
DEFECT AT THE TIME OF BONDING	○	○	×	○	○	×	○	○	×

[0053]

The description of the embodiments, the description of Examples, and the disclosure of the drawings described above are merely examples for describing the invention
5 described in claims. The invention described in claims is not limited by the description of the embodiments or the disclosure of the drawings described above.

[Explanation of Reference]

[0054]

- 10 10 Cu core ball
- 12 Core
- 14 Base plating layer
- 16 Solder plating layer (solder layer)
- 20 Sn plating layer (Sn layer)

15

What is claimed is:

1. A core material comprising:

a core;

a solder layer provided outside the core and being a
5 solder alloy containing only Sn and at least any one
element of Ag, Cu, Sb, Ni, Co, Ge, Ga, Fe, Al, In, Cd, Zn,
Pb, Au, P, S, Si, Ti, Mg, Pd, and Pt; and

a Sn layer provided outside the solder layer,
wherein

10 the solder layer has a thickness of 1 μm or more on
one side,

the Sn layer has a thickness of 0.1 μm or more on one
side, and

a thickness of the Sn layer is 0.215% or more and 36%
15 or less of the thickness of the solder layer.

2. The core material according to claim 1, wherein the
solder layer is a (Sn-Ag-Cu)-based solder.

20 3. The core material according to claim 1 or 2, further
comprising a base plating layer made of any one of Ni, Co,
Ni-Co, Ni-P and Ni-B between the core and the solder layer.

4. The core material according to any one of claims 1 to
25 3, wherein Cu ball or Cu column is used as the core.

5. An electronic component using the core material

according to any one of claims 1 to 4 as a solder bump.

6. A method for forming bump electrode comprising steps of:

5 mounting the core material according to any one of claims 1 to 5 on an electrode;

forming a bump electrode by heating the mounted core material.