

POLYMERS PRODUCED VIA USE OF VINYL TRANSFER AGENTS

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PRIORITY AND CROSS REFERENCE TO RELATED APPLICATIONS

[0001] This invention claims priority to and the benefit of USSN 62/212,405, filed August 31, 2015. This application also relates to USSN 62/332,940, filed May 6, 2016 and USSN 62/332,921, filed May 6, 2016.

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FIELD OF THE INVENTION

[0002] The invention relates to the use of pyridyldiamido transition metal complexes and catalyst systems with an activator and a metal hydrocarbenyl chain transfer agent, such as an aluminum vinyl-transfer agent (AVTA).

BACKGROUND OF THE INVENTION

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[0003] Pyridyldiamido transition metal complexes are disclosed in US 2015/0141601, US 2014/0316089, US 2012/0071616, US 2011/0301310, US 2011/0224391, and US 2010/0022726, where such complexes are useful as catalyst components for the polymerization of olefins.

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[0004] US 2015/0141596, US 2015/0141590, and US2014/0256893 describe the production of polyolefins using pyridyldiamido catalysts in the presence of chain-transfer agents that do not feature transferrable vinyl groups.

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[0005] Macromolecules 2002, 35, 6760-6762 discloses propene polymerization with tetrakis(pentafluorophenyl)borate, 7-octenyldiisobutylaluminum, and racMe₂Si(2-Me-indenyl)₂ZrCl₂ or Ph₂C(cyclopentadienyl)(fluorenyl)ZrCl₂ to produce polypropylene with octenyldiisobutylaluminum incorporated as a comonomer.

[0006] Japanese Kokai Tokkyo Koho (2004), JP 2004-83773-A describes the preparation of polypropylene in the presence of trialkenylaluminum using metallocene and Ziegler-Natta catalysts.

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[0007] Macromolecules 1995, 28, 437-443 describes the formation of isotactic polypropylene containing vinyl end groups by the Ziegler-Natta catalyzed polymerization of propylene in the presence of dialkenylzincs.

[0008] Macromolecules 2002, 35, 3838-3843 describes the formation of long-chain branched polypropylene via the insertion of in situ formed vinyl-terminated polypropylene into growing polymer chains.

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[0009] Macromolecules 2002, 35, 9586-9594 describes the formation of long-chain

branched copolymers of ethylene and alpha olefins via the insertion of in situ formed vinyl-terminated polymer into growing polymer chains.

[0010] Eur. Pat. Appl. (2012), EP 2436703 A1 describes the production of comb architecture branch block copolymers in a process that uses dual catalysts and a zinc-based polymerizable chain shuttling agent.

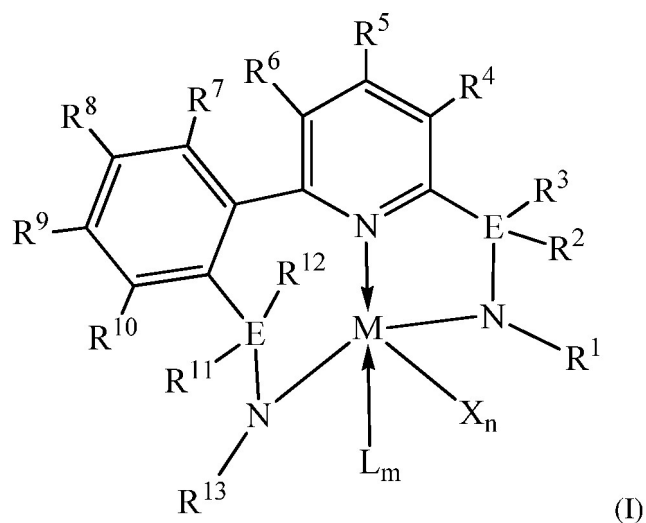
[0011] WO 2007/035492 describes the production of long-chain branched and branch block copolymers by polymerization of alkene monomers in the presence of a zinc-based polymerizable shuttling agent.

[0012] References of interest also include: 1) Vaughan, A; Davis, D. S.; Hagadorn, J. R. in Comprehensive Polymer Science, Vol. 3, Chapter 20, "Industrial catalysts for alkene polymerization"; 2) Gibson, V. C.; Spitzmesser, S. K. Chem. Rev. 2003, 103, 283; and 3) Britovsek, G. J. P.; Gibson, V. C.; Wass, D. F. Angew. Chem. Int. Ed. 1999, 38, 428; 4) US 2002/0142912; 5) US 6,900,321; 6) US 6,103,657; 7) WO 2005/095469; 8) US 2004/0220050A1; 9) WO 2007/067965; 10) Froese, R. D. J. et al., J. Am. Chem. Soc. 2007, 129, pp. 7831-7840; 11) WO 2010/037059; 12) US 2010/0227990 A1; 13) WO 2002/38628 A2; 14) US 2014/0256893; 15) Guerin, F.; McConville, D. H.; Vittal, J. J., Organometallics, 1996, 15, p. 5586.

[0013] There is still a need in the art for new and improved catalyst systems for the polymerization of olefins, in order to achieve specific polymer properties, such as long chain branching, high vinyl content, to increase conversion or comonomer incorporation, or to alter comonomer distribution without impacting the resulting polymer's properties.

SUMMARY OF THE INVENTION

[0014] This invention relates to catalyst systems comprising an activator, metal hydrocarbenyl chain transfer agent (such as an aluminum vinyl transfer agent, and single site catalyst complex, such as a pyridyldiamido transition metal complex represented by the formula (A):



wherein:

M is a Group 3, 4, 5, 6, 7, 8, 9, 10, 11, or 12 metal;

E is selected from carbon, silicon, or germanium;

5 X is an anionic leaving group;

L is a neutral Lewis base;

R¹ and R¹³ are each independently selected from the group consisting of hydrocarbyls, substituted hydrocarbyls, and silyl groups;

10 R², R³, R⁴, R⁵, R⁶, R⁷, R⁸, R⁹, R¹⁰, R¹¹, and R¹² are each independently selected from the group consisting of hydrogen, hydrocarbyls, alkoxy, silyl, amino, aryloxy, substituted hydrocarbyls, halogen, and phosphino;

n is 1 or 2;

m is 0, 1, or 2; and

two X groups may be joined together to form a dianionic group;

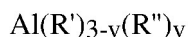
15 two L groups may be joined together to form a bidentate Lewis base;

an X group may be joined to an L group to form a monoanionic bidentate group;

R⁷ and R⁸ may be joined to form a ring;

R¹⁰ and R¹¹ may be joined to form a ring.

[0015] This invention further relates to catalyst systems comprising activator, transition
20 metal catalyst complex, and aluminum vinyl transfer agent (AVTA) represented by formula:



wherein each R', independently, is a C₁-C₃₀ hydrocarbyl group; each R'', independently, is a C₄-C₂₀ hydrocarbenyl group having an end-vinyl group; and v is from 0.1 to 3.

[0016] This invention further relates to processes to produce the above catalyst systems

and methods to polymerize olefins using the above catalyst systems.

[0017] This invention further relates to processes to produce the above catalyst systems and methods to produce vinyl terminated polyolefins, using the above catalyst systems.

BRIEF DESCRIPTION OF THE DRAWINGS

5 [0018] Figure 1 is a plot of 100,000/M_n versus nmols of AVTA1. Key: circles = complex 3; triangles = complex 1; stars = complex 2; diamonds = complex 4.

[0019] Figure 2 is a drawing of a selection of species produced in the Experimental Section.

DETAILED DESCRIPTION OF THE INVENTION

10 [0020] The specification describes transition metal complexes and catalyst systems that include the transition metal complexes. The term complex is used to describe molecules in which an ancillary ligand is coordinated to a central transition metal atom. The ligand is bulky and stably bonded to the transition metal so as to maintain its influence during use of the catalyst, such as polymerization. The ligand may be coordinated to the transition metal
15 by covalent bond and/or electron donation coordination or intermediate bonds. The transition metal complexes are generally subjected to activation to perform their polymerization or oligomerization function using an activator which is believed to create a cation as a result of the removal of an anionic group, often referred to as a leaving group, from the transition metal.

20 [0021] As used herein, the numbering scheme for the Periodic Table groups is the new notation as set out in Chemical and Engineering News, 63(5), 27 (1985).

[0022] "Catalyst productivity" is a measure of how many grams of polymer (P) are produced using a polymerization catalyst comprising W g of catalyst (cat), over a period of time of T hours; and may be expressed by the following formula: $P/(T \times W)$ and expressed in
25 units of gP/gcat⁻¹hr⁻¹. Conversion is the amount of monomer that is converted to polymer product, and is reported as mol% and is calculated based on the polymer yield and the amount of monomer fed into the reactor. Catalyst activity is a measure of how active the catalyst is and is reported as the mass of product polymer (P) produced per mole of catalyst (cat) used (kgP/molcat).

30 [0023] An "olefin," alternatively referred to as "alkene," is a linear, branched, or cyclic compound of carbon and hydrogen having at least one double bond. For purposes of this specification and the claims appended thereto, when a polymer or copolymer is referred to as comprising an olefin, the olefin present in such polymer or copolymer is the polymerized form of the olefin. For example, when a copolymer is said to have an "ethylene" content of

35 wt% to 55 wt%, it is understood that the mer unit in the copolymer is derived from ethylene in the polymerization reaction and said derived units are present at 35 wt% to 55 wt%, based upon the weight of the copolymer. A “polymer” has two or more of the same or different mer units. A “homopolymer” is a polymer having mer units that are the same.

5 A “copolymer” is a polymer having two or more mer units that are different from each other. A “terpolymer” is a polymer having three mer units that are different from each other. “Different” as used to refer to mer units indicates that the mer units differ from each other by at least one atom or are different isomerically. Accordingly, the definition of copolymer, as used herein, includes terpolymers and the like. An “ethylene polymer” or “ethylene

10 copolymer” is a polymer or copolymer comprising at least 50 mol% ethylene derived units, a “propylene polymer” or “propylene copolymer” is a polymer or copolymer comprising at least 50 mol% propylene derived units, and so on. An oligomer is typically a polymer having a low molecular weight, such as an M_n of less than 25,000 g/mol, or in an embodiment less than 2,500 g/mol, or a low number of mer units, such as 75 mer units or less or 50 mer units

15 or less. An “ethylene polymer” or “ethylene copolymer” is a polymer or copolymer comprising at least 50 mol% ethylene derived units, a “propylene polymer” or “propylene copolymer” is a polymer or copolymer comprising at least 50 mol% propylene derived units, and so on.

[0024] For the purposes of this invention, ethylene shall be considered an α -olefin.

20 [0025] For purposes of this invention and claims thereto, the term “substituted” means that a hydrogen group has been replaced with a heteroatom, or a heteroatom-containing group. For example, a “substituted hydrocarbyl” is a radical made of carbon and hydrogen where at least one hydrogen is replaced by a heteroatom or heteroatom-containing group.

[0026] A metallocene catalyst is defined as an organometallic compound with at least one

25 π -bound cyclopentadienyl moiety (or substituted cyclopentadienyl moiety) and more frequently two π -bound cyclopentadienyl moieties or substituted cyclopentadienyl moieties.

[0027] As used herein, M_n is number average molecular weight, M_w is weight average molecular weight, and M_z is z average molecular weight, wt% is weight percent, and mol% is mole percent. Molecular weight distribution (MWD), also referred to as polydispersity index

30 (PDI), is defined to be M_w divided by M_n . Unless otherwise noted, all molecular weight units (e.g., M_w , M_n , M_z) are g/mol.

[0028] Unless otherwise noted all melting points (T_m) are DSC second melt.

[0029] The following abbreviations may be used herein: dme is 1,2-dimethoxyethane, Me is methyl, Ph is phenyl, Et is ethyl, Pr is propyl, iPr is isopropyl, n-Pr is normal propyl, Bu is

butyl, cPR is cyclopropyl, iBu is isobutyl, tBu is tertiary butyl, p-tBu is para-tertiary butyl, nBu is normal butyl, sBu is sec-butyl, TMS is trimethylsilyl, TIBAL is triisobutylaluminum, TNOAL is tri(n-octyl)aluminum, MAO is methylalumoxane, p-Me is para-methyl, Ph is phenyl, Bn is benzyl (i.e., CH₂Ph), THF (also referred to as thf) is tetrahydrofuran, RT is
5 room temperature (and is 23°C unless otherwise indicated), tol is toluene, EtOAc is ethyl acetate, Cy is cyclohexyl and AVTA is an aluminum-based vinyl transfer agent.

[0030] A “catalyst system” comprises at least one catalyst compound and at least one activator. When “catalyst system” is used to describe such the catalyst compound/activator combination before activation, it means the unactivated catalyst complex (pre-catalyst)
10 together with an activator and, optionally, a co-activator. When it is used to describe the combination after activation, it means the activated complex and the activator or other charge-balancing moiety. The transition metal compound may be neutral as in a pre-catalyst, or a charged species with a counter ion as in an activated catalyst system. For the purposes of this invention and the claims thereto, when catalyst systems are described as comprising
15 neutral stable forms of the components, it is well understood by one of ordinary skill in the art, that the ionic form of the component is the form that reacts with the monomers to produce polymers.

[0031] In the description herein, the catalyst may be described as a catalyst precursor, pre-catalyst compound, catalyst compound, transition metal complex, or transition metal
20 compound, and these terms are used interchangeably. A polymerization catalyst system is a catalyst system that can polymerize monomers to polymer. An “anionic ligand” is a negatively charged ligand which donates one or more pairs of electrons to a metal ion. A “neutral donor ligand” is a neutrally charged ligand which donates one or more pairs of electrons to a metal ion. Activator and cocatalyst are also used interchangeably.

[0032] A scavenger is a compound that is typically added to facilitate polymerization by scavenging impurities. Some scavengers may also act as activators and may be referred to as
25 co-activators. A co-activator, that is not a scavenger, may also be used in conjunction with an activator in order to form an active catalyst. In some embodiments a co-activator can be pre-mixed with the transition metal compound to form an alkylated transition metal
30 compound.

[0033] Non-coordinating anion (NCA) is defined to mean an anion either that does not coordinate to the catalyst metal cation or that does coordinate to the metal cation, but only weakly. The term NCA is also defined to include multicomponent NCA-containing activators, such as N,N-dimethylanilinium tetrakis(pentafluorophenyl)borate, that contain an

acidic cationic group and the non-coordinating anion. The term NCA is also defined to include neutral Lewis acids, such as tris(pentafluorophenyl)boron, that can react with a catalyst to form an activated species by abstraction of an anionic group. An NCA coordinates weakly enough that a neutral Lewis base, such as an olefinically or acetylenically unsaturated monomer can displace it from the catalyst center. Any metal or metalloid that can form a compatible, weakly coordinating complex may be used or contained in the noncoordinating anion. Suitable metals include, but are not limited to, aluminum, gold, and platinum. Suitable metalloids include, but are not limited to, boron, aluminum, phosphorus, and silicon. Activators containing non-coordinating anions can also be referred to as stoichiometric activators. A stoichiometric activator can be either neutral or ionic. The terms ionic activator and stoichiometric ionic activator can be used interchangeably. Likewise, the terms neutral stoichiometric activator and Lewis acid activator can be used interchangeably. The term non-coordinating anion activator includes neutral stoichiometric activators, ionic stoichiometric activators, ionic activators, and Lewis acid activators.

[0034] For purposes of this invention and claims thereto in relation to catalyst compounds, the term “substituted” means that a hydrogen group has been replaced with a hydrocarbyl group, a heteroatom, or a heteroatom-containing group. For example, methyl cyclopentadiene (Cp) is a Cp group substituted with a methyl group.

[0035] For purposes of this invention and claims thereto, “alkoxides” include those where the alkyl group is a C₁ to C₁₀ hydrocarbyl. The alkyl group may be straight chain, branched, or cyclic. The alkyl group may be saturated or unsaturated. In some embodiments, the alkyl group may comprise at least one aromatic group.

[0036] The terms “hydrocarbyl radical,” “hydrocarbyl,” “hydrocarbyl group,” “alkyl radical,” and “alkyl” are used interchangeably throughout this document. Likewise, the terms “group”, “radical”, and “substituent” are also used interchangeably in this document. For purposes of this disclosure, “hydrocarbyl radical” is defined to be C₁-C₁₀₀ radicals, that may be linear, branched, or cyclic, and when cyclic, aromatic or non-aromatic. Examples of such radicals include, but are not limited to, methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, sec-butyl, tert-butyl, pentyl, iso-amyl, hexyl, octyl cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl, cyclooctyl, and the like including their substituted analogues. Substituted hydrocarbyl radicals are radicals in which at least one hydrogen atom of the hydrocarbyl radical has been substituted with at least one halogen (such as Br, Cl, F or I) or at least one functional group such as NR₂*, OR*, SeR*, TeR*, PR₂*, AsR₂*, SbR₂*, SR*, BR₂*, SiR₃*, GeR₃*, SnR₃*, PbR₃*, and the like, or where at least one heteroatom has been inserted within

a hydrocarbyl ring.

[0037] The term "alkenyl" or "hydrocarbenyl" means a straight-chain, branched-chain, or cyclic hydrocarbon radical having one or more double bonds that are not part of an aromatic ring. These alkenyl radicals may, optionally, be substituted. Examples of alkenyl radicals include, but are not limited to, ethenyl, propenyl, allyl, 1,4-butadienyl cyclopropenyl, cyclobutenyl, cyclopentenyl, cyclohexenyl, cycloctenyl, and the like, including their substituted analogues.

[0038] The term "alkoxy" or "alkoxide" means an alkyl ether or aryl ether radical wherein the term alkyl is as defined above. Examples of suitable alkyl ether radicals include, but are not limited to, methoxy, ethoxy, n-propoxy, iso-propoxy, n-butoxy, iso-butoxy, sec-butoxy, tert-butoxy, phenoxy, and the like.

[0039] The term "aryl" or "aryl group" means a six carbon aromatic ring and the substituted variants thereof, including but not limited to, phenyl, 2-methyl-phenyl, xylyl, 4-bromo-xylyl. Likewise, heteroaryl means an aryl group where a ring carbon atom (or two or three ring carbon atoms) has been replaced with a heteroatom, preferably N, O, or S. As used herein, the term "aromatic" also refers to pseudoaromatic heterocycles which are heterocyclic substituents that have similar properties and structures (nearly planar) to aromatic heterocyclic ligands, but are not by definition aromatic; likewise the term aromatic also refers to substituted aromatics.

[0040] Where isomers of a named alkyl, alkenyl, alkoxide, or aryl group exist (e.g., n-butyl, iso-butyl, sec-butyl, and tert-butyl) reference to one member of the group (e.g., n-butyl) shall expressly disclose the remaining isomers (e.g., iso-butyl, sec-butyl, and tert-butyl) in the family. Likewise, reference to an alkyl, alkenyl, alkoxide, or aryl group without specifying a particular isomer (e.g., butyl) expressly discloses all isomers (e.g., n-butyl, iso-butyl, sec-butyl, and tert-butyl).

[0041] The term "ring atom" means an atom that is part of a cyclic ring structure. By this definition, a benzyl group has six ring atoms and tetrahydrofuran has 5 ring atoms.

[0042] A heterocyclic ring is a ring having a heteroatom in the ring structure as opposed to a heteroatom substituted ring where a hydrogen on a ring atom is replaced with a heteroatom. For example, tetrahydrofuran is a heterocyclic ring and 4-N,N-dimethylamino-phenyl is a heteroatom substituted ring.

[0043] The term "continuous" means a system that operates without interruption or cessation. For example, a continuous process to produce a polymer would be one where the reactants are continually introduced into one or more reactors and polymer product is

continually withdrawn.

[0044] A solution polymerization means a polymerization process in which the polymer is dissolved in a liquid polymerization medium, such as an inert solvent or monomer(s) or their blends. A solution polymerization is typically homogeneous. A homogeneous polymerization is one where the polymer product is dissolved in the polymerization medium. Such systems are preferably not turbid as described in J. Vladimir Oliveira, C. Dariva and J. C. Pinto, Ind. Eng. Chem. Res., 29, 2000, 4627.

[0045] A bulk polymerization means a polymerization process in which the monomers and/or comonomers being polymerized are used as a solvent or diluent using little or no inert solvent as a solvent or diluent. A small fraction of inert solvent might be used as a carrier for catalyst and scavenger. A bulk polymerization system contains less than 25 wt% of inert solvent or diluent, preferably less than 10 wt%, preferably less than 1 wt%, preferably 0 wt%.

[0046] This invention relates to catalyst systems comprising a pyridyldiamido transition metal complex represented by the formula (A), an activator (such as an alumoxane or a non-coordinating anion), and metal hydrocarbenyl chain transfer agent, typically represented by the formula: $\text{Al}(\text{R}')_{3-v}(\text{R}'')_v$ or $\text{E}[\text{Al}(\text{R}')_{2-y}(\text{R}'')_y]_2$, wherein each R' , independently, is a C_1 to C_{30} hydrocarbyl group; each R'' , independently, is a C_4 to C_{20} hydrocarbenyl group having an allyl chain end; E is a group 16 element (such as O or S); v is from 0.01 to 3 (such as 1 or 2), and y is from 0.01 to 2 (such as 1 or 2). Preferably the metal hydrocarbenyl chain transfer agent is an aluminum vinyl-transfer agent (AVTA) represented by the formula: $\text{Al}(\text{R}')_{3-v}(\text{R})_v$ with R defined as a hydrocarbenyl group containing 4 to 20 carbon atoms and featuring an allyl chain end, R' defined as a hydrocarbyl group containing 1 to 30 carbon atoms, and v is 0.1 to 3 (such as 1 or 2).

[0047] The catalyst/activator combinations are formed by combining the transition metal complex with activators in any manner known from the literature, including by supporting them for use in slurry or gas phase polymerization. The catalyst/activator combinations may also be added to or generated in solution polymerization or bulk polymerization (in the monomer). The metal hydrocarbenyl chain transfer agent (preferably an aluminum vinyl transfer agent) may be added to the catalyst and or activator before, during or after the activation of the catalyst complex or before or during polymerization. Typically, the metal hydrocarbenyl chain transfer agent (preferably the aluminum vinyl-transfer agent) is added to the polymerization reaction separately, such as before, the catalyst/activator pair.

[0048] The polymer produced from the polymerization using the catalyst systems described herein preferably contains at least 0.05 allyl chain ends per polymer chain, 0.1 allyl

chain ends per polymer chain, at least 0.2 allyl chain ends per polymer chain, at least 0.3 allyl chain ends per polymer chain, at least 0.4 allyl chain ends per polymer chain, at least 0.5 allyl chain ends per polymer chain, at least 0.6 allyl chain ends per polymer chain, at least 0.7 allyl chain ends per polymer chain, at least 0.8 allyl chain ends per polymer chain, at least 0.8 allyl chain ends per polymer chain, at least 1.0 allyl chain ends per polymer chain. Ethylene homopolymers and copolymers are particularly preferred products. If the catalyst complex chosen is also capable of incorporating bulky alkene monomers, such as C₆ to C₂₀ alpha olefins, into the growing polymer chain, then the resulting polymer (typically an ethylene copolymer) may contain long chain branches formed by the insertion of an allyl terminated polymer chain formed in situ (also referred to as a "vinyl-terminated macromonomer") into the growing polymer chains. Process conditions including residence time, the ratio of monomer to polymer in the reactor, and the ratio of transfer agent to polymer will affect the amount of long chain branching in the polymer, the average length of branches, and the type of branching observed. A variety of branching types may be formed, which include comb architectures and branch on branch structures similar to those found in low-density polyethylene. The combination of chain growth and vinyl-group insertion may lead to polymer with a branched structure and having one or fewer vinyl unsaturations per polymer molecule. The absence of significant quantities of individual polymer molecules containing greater than one vinyl unsaturation prevents or reduces the formation of unwanted crosslinked polymers. Polymers having long chain branching typically have a g'vis of 0.95 or less, alternately 0.90 or less, alternately 0.85 or less, alternately 0.80 or less, alternately 0.75 or less, alternately 0.70 or less, alternately 0.60 or less.

[0049] If the catalyst chosen is poor at incorporating comonomers such as C₂ to C₂₀ alpha olefins, then the polymer obtained is largely linear (little or no long chain branching). Likewise, process conditions including the ratio of transfer agent to polymer will affect the molecular weight of the polymer. Polymers having little or no long chain branching typically have a g'vis of more than 0.95, alternately 0.97 or more, alternately 0.98 or more.

[0050] Alkene polymerizations and co-polymerizations using one or more transfer agents, such as an AVTA, with two or more catalysts are also of potential use. Desirable products that may be accessed with this approach includes polymers that have branch block structures and/or high levels of long-chain branching.

[0051] The transfer agent to catalyst complex equivalence ratio can be from about 1:100 to 500,000:1. Preferably the molar ratio of transfer agent to catalyst complex is greater than one. Alternately, the molar ratio of transfer agent to catalyst complex is greater than 30. The

AVTA to catalyst complex equivalence ratio can be from about 1:100 to 500,000:1. Preferably the molar ratio of AVTA to catalyst complex is greater than one. More preferred the molar ratio of AVTA to catalyst complex is greater than 30.

[0052] The AVTA can also be used in combination with other chain transfer agents that are typically used as scavengers, such as trialkylaluminum compounds (where the alkyl groups are selected from C₁ to C₂₀ alkyl groups, preferably methyl, ethyl, propyl, butyl, pentyl, hexyl, heptyl, octyl, nonyl, decyl, undecyl, dodecyl, or an isomer thereof). Usefully the ATVA can be used in combination with a trialkyl aluminum compound such as tri-n-octylaluminum and triisobutylaluminum. The ATVA can also be used in combination with a dialkyl zinc compound, such as diethylzinc, dimethylzinc, or dipropylzinc.

[0053] The transfer agent can also be used in combination with oxygen-containing organoaluminums such as bis(diisobutylaluminum)oxide, MMAO-3A, and other alumoxanes. Certain of these oxygen-containing organoaluminums are expected to serve as scavengers while remaining significantly less prone to hydrocarbyl group chain-transfer than typical organoaluminums, such as trimethylaluminum or tri(n-octyl)aluminum.

[0054] The production of di-end-functionalized polymers is possible with this technology. One product, prior to exposure to air, from an alkene polymerization performed in the presence of AVTA is the aluminum-capped species $\text{Al}(\text{R}')_{3-v}(\text{polymer-CH=CH}_2)_v$, where v is 0.1 to 3 (alternately 1 to 3, alternately 1, 2, or 3). The Al-carbon bonds will react with a variety of electrophiles (and other reagents), such as oxygen, halogens, carbon dioxide, and the like. Thus, quenching the reactive polymer mixture with an electrophile prior to exposure to atmosphere would yield a di-end-functionalized product of the general formula: $\text{Z}-(\text{monomers})_n-\text{CH=CH}_2$, where Z is a group from the reaction with the electrophile and n is an integer, such as from 1 to 1,000,000, alternately from 2 to 50,000, alternately from 10 to 25,000. For example, quenching with oxygen yields a polymer functionalized at one end with a hydroxy group and at the other end with a vinyl group. Quenching with bromine yields a polymer functionalized at one end with a Br group and at the other end with a vinyl group.

[0055] Functional group terminated polymers can also be produced using functional group transfer agents (FGTA). In this embodiment of the invention, the FGTA is represented by the formula $\text{M}^{\text{FGTA}}(\text{R}')_{3-v}(\text{FG})_v$, with R' and v defined as above, M^{FGTA} a group 13 element (such as B or Al), and with FG defined as a group containing 1 to 20 carbon atoms and a functional group Z. The choice of FG is such that it is compatible with the catalyst system being used. Preferred Z groups include, but are not limited to, non-vinyl olefinic groups such

as vinylidene, vinylene or trisubstituted olefins, cyclics containing unsaturation such as cyclohexene, cyclooctene, vinyl cyclohexene, aromatics, ethers, and metal-capped alkoxides.

[0056] In another embodiment of the invention, the polymer product of this invention are of the formula: polymer-(CH₂)_nCH=CH₂ where n is from 2 to 18, preferably from 6 to 14, more preferably 6, and where “polymer” is the attached polymeryl chain. Polymers of this formula are particularly well suited in making branch polymer combs. The polymer combs can be made by any number of methods. One method would be to use a catalyst system to make the vinyl terminated polymer, and then use a second catalyst system to incorporate the vinyl terminated polymer into a polymer backbone made from the second catalyst. This can be done sequentially in one reactor by first making the vinyl terminated polymer and then adding a second catalyst and, optionally, different monomer feeds in the same reactor. Alternatively two reactors in series can be used where the first reactor is used to make the vinyl terminated polymer which flows into a second reactor in series having the second catalyst and, optionally, different monomer feeds. The vinyl terminated polymer can be a soft material, as in an ethylene propylene copolymer (such as ethylene propylene copolymer rubber), low density polyethylene, or a polypropylene, or a harder material, as in an isotactic polypropylene, high density polyethylene, or other polyethylene. Typically, if the vinyl terminated polymer is soft, it is preferred that the polymer backbone of the comb be hard; if the vinyl terminated polymer is hard, it is preferred that the polymer backbone of the comb be soft, however any combination of polymer structures and types can be used.

[0057] In another embodiment of the invention, the vinyl-terminated polymers (VTPs) of this invention are of formula: polymer-(CH₂)_nCH=CH₂ where n is from 2 to 18, preferably from 6 to 14, more preferably 6, and where “polymer” is the attached polymeryl chain. VTPs of this formula are particularly well suited in making branch block polymers. The branch block polymers can be made by any number of methods. One method involves using the same catalyst that is used to make the VTP, and then changing polymerization conditions (such as, but not limited to, by changing monomer composition and/or type and/or the amount or presence of AVTA) in the same or different reactor (such as two or more reactors in series). In this case, the branch will have a different polymeric composition vs. the polymer backbone created under the different polymerization conditions. Another method would be to use a catalyst system to make the VTP, then use a second catalyst system to incorporate the VTPs into a polymer backbone made from the second catalyst. This can be done sequentially in one reactor by first making the VTP and then adding a second catalyst and, optionally, different monomer feeds in the same reactor. Alternatively, two reactors in

series can be used where the first reactor is used to make the VTP which flows into a second reactor in series having the second catalyst and, optionally, different monomer feeds. The branched block polymers can be of any composition, however, typically a combination of soft and hard polymers (relative to one another) are preferred. For example, an iPP VTP
5 could be produced in a reactor, and then ethylene added to the existing propylene feed to make a rubber EP that would have iPP branches. Or an iPP VTP could be produced in a first reactor, and then sent to a second reactor containing ethylene (or additional ethylene for a propylene ethylene copolymer and, optionally, additional propylene monomer (and the same or different catalyst) to make a rubber EP that would have iPP branches (or propylene
10 ethylene copolymer branches).

[0058] Useful metal hydrocarbenyl chain transfer agents (preferably the aluminum vinyl transfer agents) are typically present at from 10 or 20 or 50 or 100 equivalents to 600 or 700 or 800 or 1000 equivalents relative to the catalyst complex. Alternately, the metal hydrocarbenyl chain transfer agents are present at a catalyst complex-to-transfer agent molar
15 ratio of from about 1:3000 to 10:1; alternatively 1:2000 to 10:1; alternatively 1:1000 to 10:1; alternatively, 1:500 to 1:1; alternatively 1:300 to 1:1; alternatively 1:200 to 1:1; alternatively 1:100 to 1:1; alternatively 1:50 to 1:1; alternatively 1:10 to 1:1.

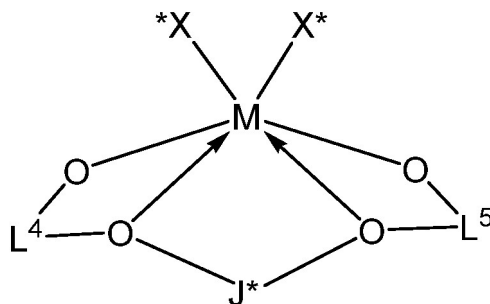
[0059] In any embodiment of this invention where the aluminum vinyl transfer agent is present, the aluminum vinyl transfer agent is present at a catalyst complex-to-aluminum vinyl
20 transfer agent molar ratio of from about 1:3000 to 10:1; alternatively 1:2000 to 10:1; alternatively 1:1000 to 10:1; alternatively, 1:500 to 1:1; alternatively 1:300 to 1:1; alternatively 1:200 to 1:1; alternatively 1:100 to 1:1; alternatively 1:50 to 1:1; alternatively 1:10 to 1:1, alternately from 1:1000 or more.

Transition Metal Complex

25 [0060] Transition metal complexes useful herein are certain “non-metallocene” olefin polymerization catalysts that undergo alkyl group transfer with the AVTA at a rate that is much higher than the rate at which they undergo typical termination processes, such as beta hydride elimination or chain-transfer to monomer. The term “non-metallocene catalyst”, also known as “post-metallocene catalyst” describe transition metal complexes that do not feature any pi-
30 coordinated cyclopentadienyl anion donors (or the like) and are useful the polymerization of olefins when combined with common activators. Families of non-metallocene complexes that may be useful for the present invention are described in Angew. Chem. Int. Ed. 2014, 53, 2-25; Chem. Rev. 2003, 103, 283-315; Angew. Chem. Int. Ed. 1999, 38, 428-447; ACS

Catal. 2011, 1, 887-900; Polymer J. 2007, 39:3, 193-207; and US Patent Nos. 6,841,502, 7,256,296, 7,018,949, 7,964,681.

[0061] Transition metal complexes useful herein include chelated transition metal complexes, such as those represented by the following formula (4):

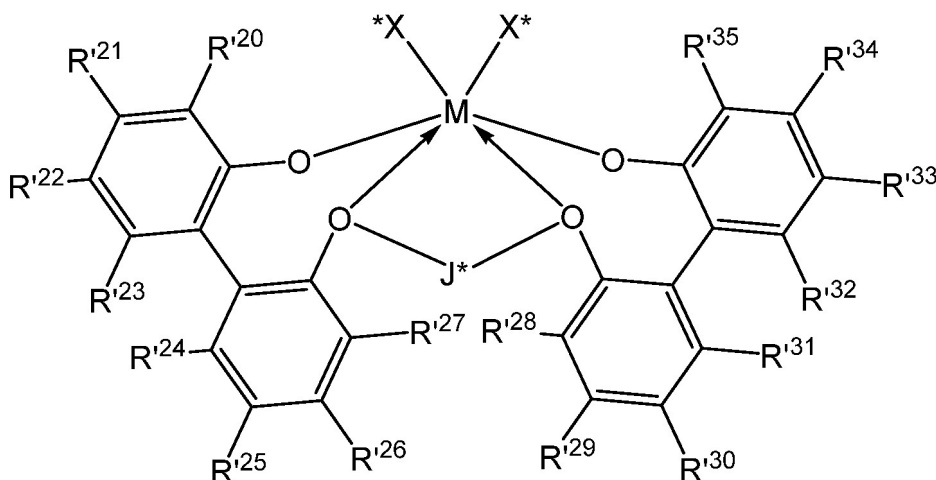


5 where: (1) J* is a divalent bridging group comprising C, Si, or both; (2) M is a group 4 metal, preferably hafnium or zirconium; (3) O is oxygen; (4) each X* is independently a univalent anionic ligand, or two X*s are joined and bound to the metal atom to form a metallocycle ring, or two X*s are joined to form a chelating ligand, a diene ligand, or an alkylidene ligand; and (5) L⁴ and L⁵ are independently, a substituted monocyclic or polycyclic aromatic groups.

[0062] In some embodiments, J* is a divalent substituted or unsubstituted C₃₋₆ aliphatic or cycloaliphatic group.

[0063] In some embodiments, L⁴ and L⁵ are independently a monocyclic or polycyclic aromatic group substituted with any combination alkyl, aryl, alkoxy, or amino substituents which may optionally be substituted with halogens.

[0064] In yet further embodiments, suitable catalyst compounds useful herein include biphenyl phenol transition metal complexes, such as those having the following structural formula (4a):



20 where: (1) J* is a divalent bridging group comprising C, Si, or both; (2) M is a group 4 metal,

preferably hafnium or zirconium; (3) O is oxygen; (4) each X* is independently a univalent anionic ligand, or two X*s are joined and bound to the metal atom to form a metallocycle ring, or two X*s are joined to form a chelating ligand, a diene ligand, or an alkylidene ligand; and (5) each R²⁰, R²¹, R²², R²³, R²⁴, R²⁵, R²⁶, R²⁷, R²⁸, R²⁹, R³⁰, R³¹, R³², R³³, R³⁴,
 5 R³⁵ is independently hydrogen, halo, C₁-C₅₀ hydrocarbyl, substituted hydrocarbyl, halocarbyl or substituted halocarbyl.

[0065] In particular embodiments, each R²⁰ and R³⁵ may be or may comprise a bulky substituent, such as substituted or unsubstituted aryl, carbazolyl, fluorenyl and/or anthracenyl.

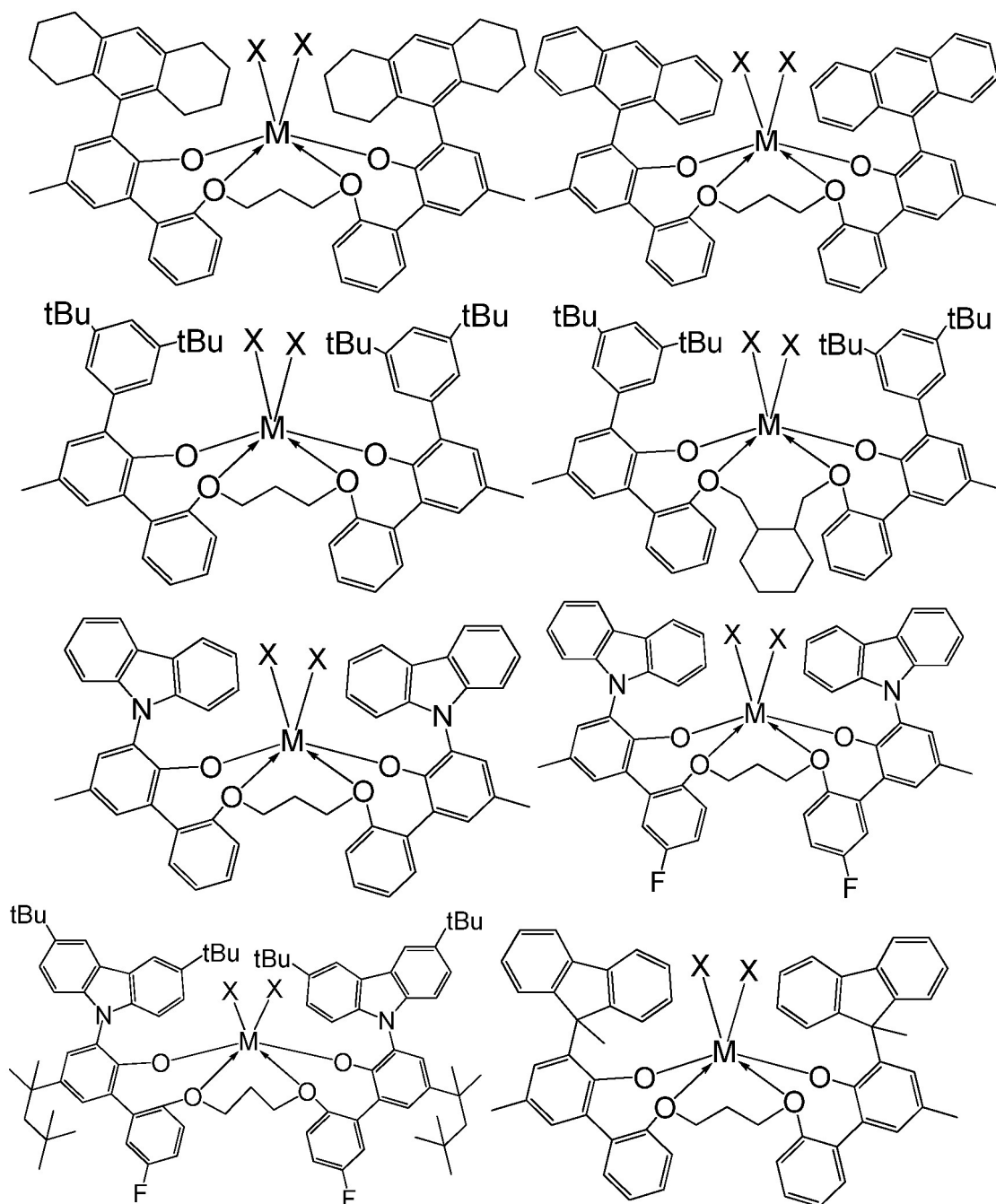
[0066] In other particular embodiments, each R²⁰ and R³⁵ independently may be 3,5-
 10 di(isopropyl)phenyl, 3,5-di(isobutyl)phenyl, 3,5-di(*tert*-butyl)phenyl, carbazol-9-yl, 3,6-di-*tert*-butylcarbazol-9-yl, 2,3,4,5,6,7,8,9-octahydrocarbazol-1-yl, anthracen-9-yl, 1,2,3,4,5,6,7,8-octahydroanthracen-9-yl, naphthyl, fluoren-9-yl, 9-methylfluoren-9-yl, 1,2,3,4,5,6,7,8-octahydrofluoren-9-yl, or 9-methyl-1,2,3,4,5,6,7,8-octahydrofluoren-9-yl.

In other particular embodiments, R²² and R³³ are independently C₁-C₁₀ hydrocarbyl,
 15 alternatively C₁-C₁₀ alkyls such as methyl, ethyl and all isomers of propyl, butyl, pentyl, hexyl, heptyl, octyl, nonyl and decyl.

[0067] In other particular embodiments, R²⁵ and R³⁰ are independently C₁-C₁₀ substituted or unsubstituted hydrocarbyl, halo, C₁-C₁₀ alkoxy, and C₂-C₂₀ dialkylamino. Alternatively R²⁵ and R³⁰ are independently C₁-C₁₀ alkyls such as methyl, ethyl and all
 20 isomers of propyl, butyl, pentyl, hexyl, heptyl, octyl, nonyl and decyl (including cyclic and linear or branched cyclic combinations); halogens such as fluoro, chloro, bromo; C₁-C₁₀ alkoxy such as methoxy, ethoxy, and all isomers of propoxy, butoxy, pentoxy, hexoxy, heptoxy, octoxy, nonoxy, and decoxy (including cyclic and linear or branched cyclic combinations); C₂-C₂₀ dialkylamino such as dimethyl amino, diethyl amino, and all isomers
 25 of dipropylamino, dibutylamino, dipentylamino, dihexylamino, diheptylamino, dioctylamino, dinonylamino, didecylamino (including cyclic and linear or branched cyclic combinations) and mixed alkyls such as methylethylamino, methylbutyl amino and the like.

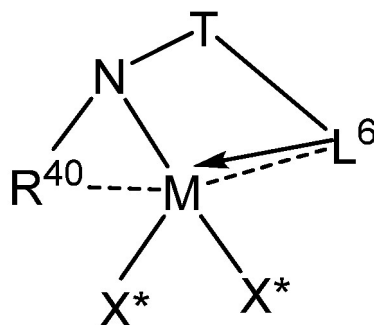
[0068] In other particular embodiments, J* may be propan-1,3-diyl, butan-1,4-diyl, cyclohexanediyl, cyclohexen-4,5-diyl, or bis(methylene)cyclohexan-1,2-diyl.

[0069] Non-limiting examples of particularly useful biphenyl phenol transition metal complexes are illustrated below, wherein M is hafnium or zirconium and X is methyl, benzyl,
 30 or chloro:



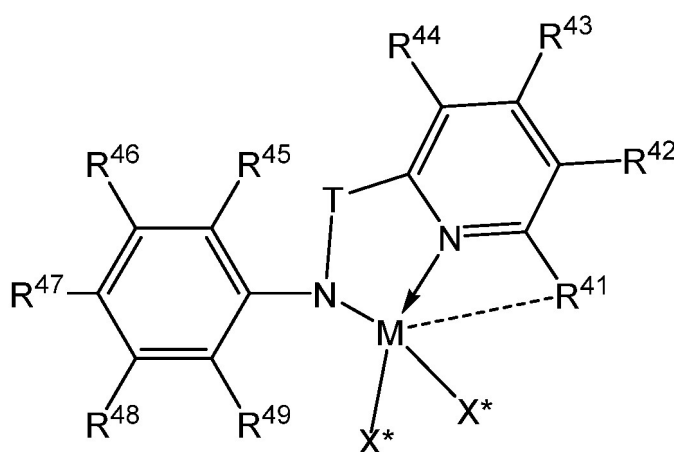
5 [0070] Additional particularly useful biphenyl phenol transition metal complexes are described in WO 2003/091262; WO 2005/108406; US 2006/0025548; US 2006/0052554; WO 2007/136494; WO 2007/136496; WO 2007/136495; WO 2009/064482; and WO2013/096573, and are incorporated by reference.

[0071] In yet further embodiments, transition metal complexes useful herein include
 10 those represented by the following formula (5):



wherein (1) M is a group 4 metal, preferably hafnium; (2) L⁶ is a C₅-C₂₀ heteroaryl group containing a Lewis base functionality, especially pyridine-2-yl or substituted pyridine-2-yl group or a divalent derivative thereof; (3) R⁴⁰ is selected from a C₁-C₃₀ alkyl, cycloalkyl, heteroalkyl, cycloheteroalkyl, aryl and substituted derivatives thereof or a divalent derivative thereof; (4) T is a divalent bridging group comprising carbon and or silicon, preferably a C₁-C₂₀ hydrocarbyl substituted methylene or silane group; (5) each X* is independently a univalent anionic ligand, or two X*s are joined and bound to the metal atom to form a metallocycle ring, or two X*s are joined to form a chelating ligand, a diene ligand, or an alkylidene ligand; (6) N is nitrogen; and (7) bonds, optional bonds and dative bonds are represented by lines, dotted lines and arrows, respectively.

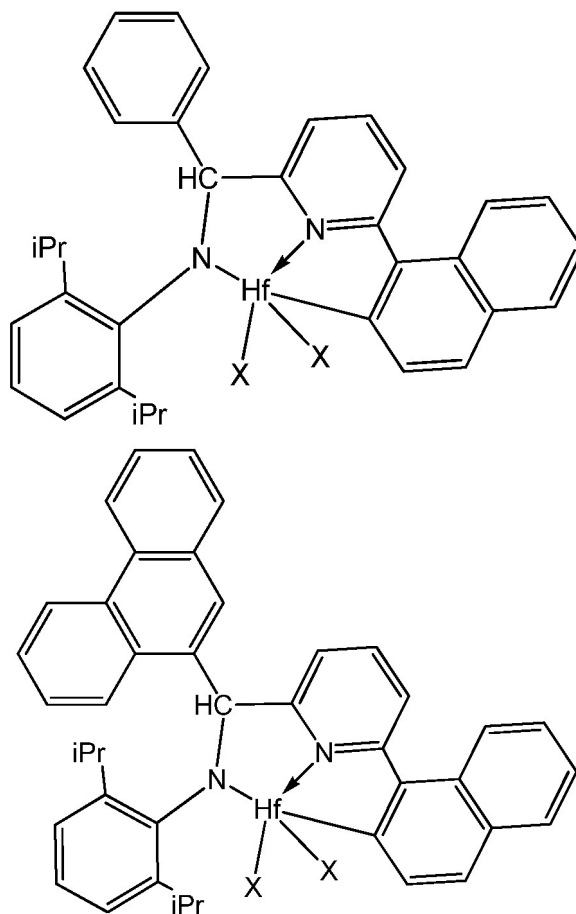
[0072] In yet further embodiments, catalyst compounds useful herein are pyridyl amide metal complexes, such as those having the following formula (5a):



wherein M, T, N and X* are as previously defined as in formula (5); each R⁴¹, R⁴², R⁴³, and R⁴⁴ are independently hydrogen, halo, or an alkyl, cycloalkyl, heteroalkyl, heterocycloalkyl, aryl or silyl group, or one or more adjacent R⁴¹-R⁴⁴ may be joined together to form a fused ring derivative; R⁴⁵-R⁴⁹ are independently hydrogen, or C₁-C₁₀ alkyl, most preferably R⁴⁵ and R⁴⁹ are alkyl such as isopropyl or *tert*-butyl; T is preferably CR⁵⁰R⁵¹ where R⁵⁰ and

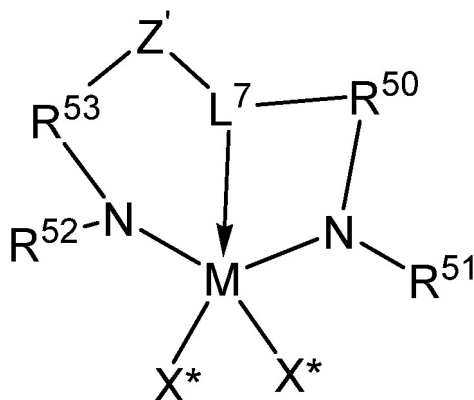
R^{51} are independently hydrogen, halogen, a C_1 - C_{20} hydrocarbyl, most preferably, one of R^{50} and R^{51} is hydrogen and the other is a C_6 - C_{20} aryl group, especially 2-isopropyl, phenyl or a fused polycyclic aryl group, most preferably anthracenyl; and bonds, optional bonds and dative bonds are represented by lines, dotted lines and arrows, respectively.

- 5 **[0073]** Non-limiting examples of pyridyl amide catalysts useful herein are illustrated below, wherein X is preferably methyl, benzyl or chloro:



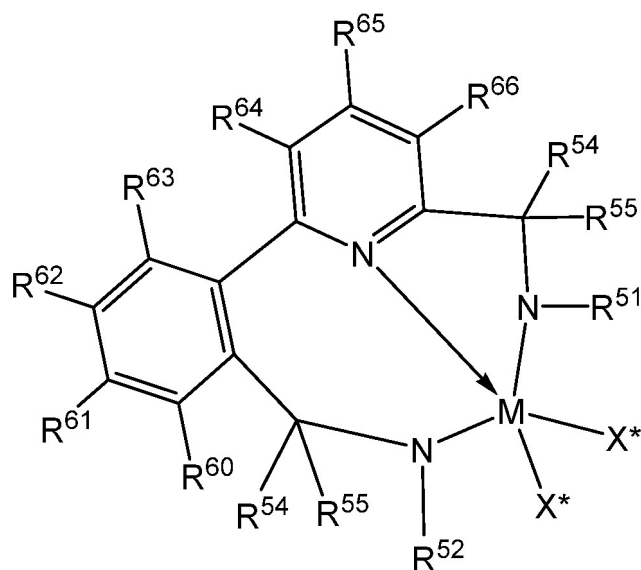
- [0074]** Additional pyridyl amide transition metal complexes particularly useful herein are described in WO 2010/0227990; US 2004/0220050; WO 2004/026925; WO 2004/024740; WO 2004/024739; WO 2003/040201; WO 2002/046249; and WO2002/038628, and are incorporated by reference.

[0075] In yet further embodiments, useful catalyst compounds include those having the following structural formula (6):



wherein (1) M is a group 4 metal, preferably hafnium; (2) N is nitrogen; (3) L^7 is a group that links R^{50} to Z' by a three atom bridge with the central of the three atoms being a group 15 or 16 element that preferably forms a dative bond to M, and is a C_5 - C_{20} heteroaryl group containing a Lewis base functionality, especially a divalent pyridinyl or substituted pyridinyl group; (4) Z' is a divalent linker group, $(R^{56})_pC-C(R^{57})_q$, where R^{56} and R^{57} are independently selected from the group consisting of hydrogen, hydrocarbyls, substituted hydrocarbyls, and wherein adjacent R^{56} and R^{57} groups may be joined to form an aromatic or saturated, substituted or unsubstituted hydrocarbyl ring, wherein the ring has 5, 6, 7 or 8 ring carbon atoms and where the substituents on the ring can join to form additional rings, and p is 1 or 2 and q is 1 or 2; (5) R^{50} and R^{53} are each, independently, $ER^{54}R^{55}$ with E being carbon, silicon or germanium, and each R^{54} and R^{55} being independently selected from the group consisting of hydrogen, hydrocarbyls, substituted hydrocarbyls, alkoxy, silyl, amino, aryloxy, halogen and phosphino, and R^{54} and R^{55} may be joined to form a saturated heterocyclic ring, or a saturated substituted heterocyclic ring where substitutions on the ring can join to form additional rings; (6) R^{51} and R^{52} are independently selected from the group consisting of hydrocarbyls, substituted hydrocarbyls, silylcarbyls and substituted silylcarbyl groups; and (7) each X^* is independently a univalent anionic ligand, or two X^* s are joined and bound to the metal atom to form a metallocycle ring, or two X^* s are joined to form a chelating ligand, a diene ligand, or an alkylidene ligand.

[0076] In yet further embodiments, useful catalyst compounds include pyridyl diamide metal complexes represented by the following formula (6a):



wherein M, X*, N, R⁵¹, R⁵², R⁵⁴, and R⁵⁵ are as previously defined as in formula (6); R⁶⁰, R⁶¹, R⁶², R⁶³, R⁶⁴, R⁶⁵, R⁶⁶ are independently selected from the group consisting of hydrogen, hydrocarbyls, substituted hydrocarbyls, alkoxy, aryloxy, halogen, amino, and silyl, and wherein any one or more adjacent R⁶⁰-R⁶⁶ may be joined to form a substituted or unsubstituted hydrocarbyl or heterocyclic ring, wherein the ring has 5, 6, 7, or 8 ring atoms and where substitutions on the ring can join to form additional rings.

[0077] In an embodiment of the invention, R⁶⁰ to R⁶⁶ are hydrogen.

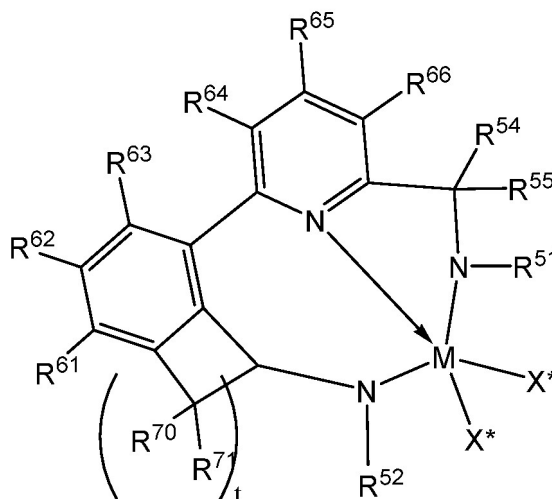
[0078] In an embodiment of the invention, R⁶² is joined with R⁶³ to form a phenyl ring fused to the existing phenyl ring (e.g., a naphthyl group), and R⁶⁰, R⁶¹, R⁶⁴, R⁶⁵, and R⁶⁶ are independently hydrogen or an alkyl group, preferably hydrogen.

[0079] In an embodiment of the invention, each R⁵⁴ and R⁵⁵ are independently hydrogen, an alkyl group or an aryl group or substituted aryl group; preferably one or both R⁵⁴ or R⁵⁵ is hydrogen, or one R⁵⁴ or R⁵⁵ is hydrogen and the other is an aryl group or substituted aryl group. Preferred but non limiting aryl groups for R⁵⁴ or R⁵⁵ include phenyl, 2-methylphenyl, 2-ethylphenyl, 2-isopropylphenyl, and naphthyl.

[0080] In an embodiment of the invention, R⁵² and R⁵¹ are independently aryl or substituted aryl; preferably R⁵¹ is a substituted phenyl group such as, but not limited to 2,6-diisopropylphenyl, 2,6-diethylphenyl, 2,6-dimethylphenyl, mesityl, and the like, and preferably R⁵² is phenyl or a substituted phenyl group such as, but not limited to 2-tolyl, 2-ethylphenyl, 2-propylphenyl, 2-trifluoromethylphenyl, 2-fluorophenyl, mesityl, 2,6-

diisopropylphenyl, 2,6-diethylphenyl, 2,6-dimethylphenyl, 3,5-di-*tert*-butylphenyl, and the like.

[0081] In yet further embodiments, useful catalyst compounds include pyridyl diamide metal complexes, such as those represented by the following formula (6b):



5

wherein M, X*, N, R⁵¹, R⁵², R⁵⁴, R⁵⁵, R⁶¹-R⁶⁶ are as previously defined as in formulae (6) and (6a); each R⁷⁰-R⁷¹ are independently selected from the group consisting of hydrogen, hydrocarbyls, substituted hydrocarbyls, alkoxy, aryloxy, halogen, amino, and silyl, and wherein any one or more adjacent R⁷⁰-R⁷¹ may be joined to form a substituted or unsubstituted hydrocarbyl or heterocyclic ring, wherein the ring has 5, 6, 7, or 8 ring atoms and where substitutions on the ring can join to form additional rings, and t is 2 or 3 (corresponding to cyclopentyl and cyclohexyl rings, respectively).

10

[0082] In an embodiment of the invention, R⁶¹-R⁶⁶ are hydrogen.

[0083] In an embodiment of the invention, each R⁷⁰ and R⁷¹ are independently hydrogen, and t is 2 or 3, preferably 2.

15

[0084] In an embodiment of the invention, each R⁵⁴ and R⁵⁵ are independently hydrogen, an alkyl group or an aryl group or substituted aryl group; preferably one or both R⁵⁴ or R⁵⁵ is hydrogen, or one R⁵⁴ or R⁵⁵ is hydrogen and the other is an aryl group or substituted aryl group. Preferred but non limiting aryl groups include phenyl and 2-methylphenyl, 2-ethylphenyl, 2-isopropylphenyl and naphthyl.

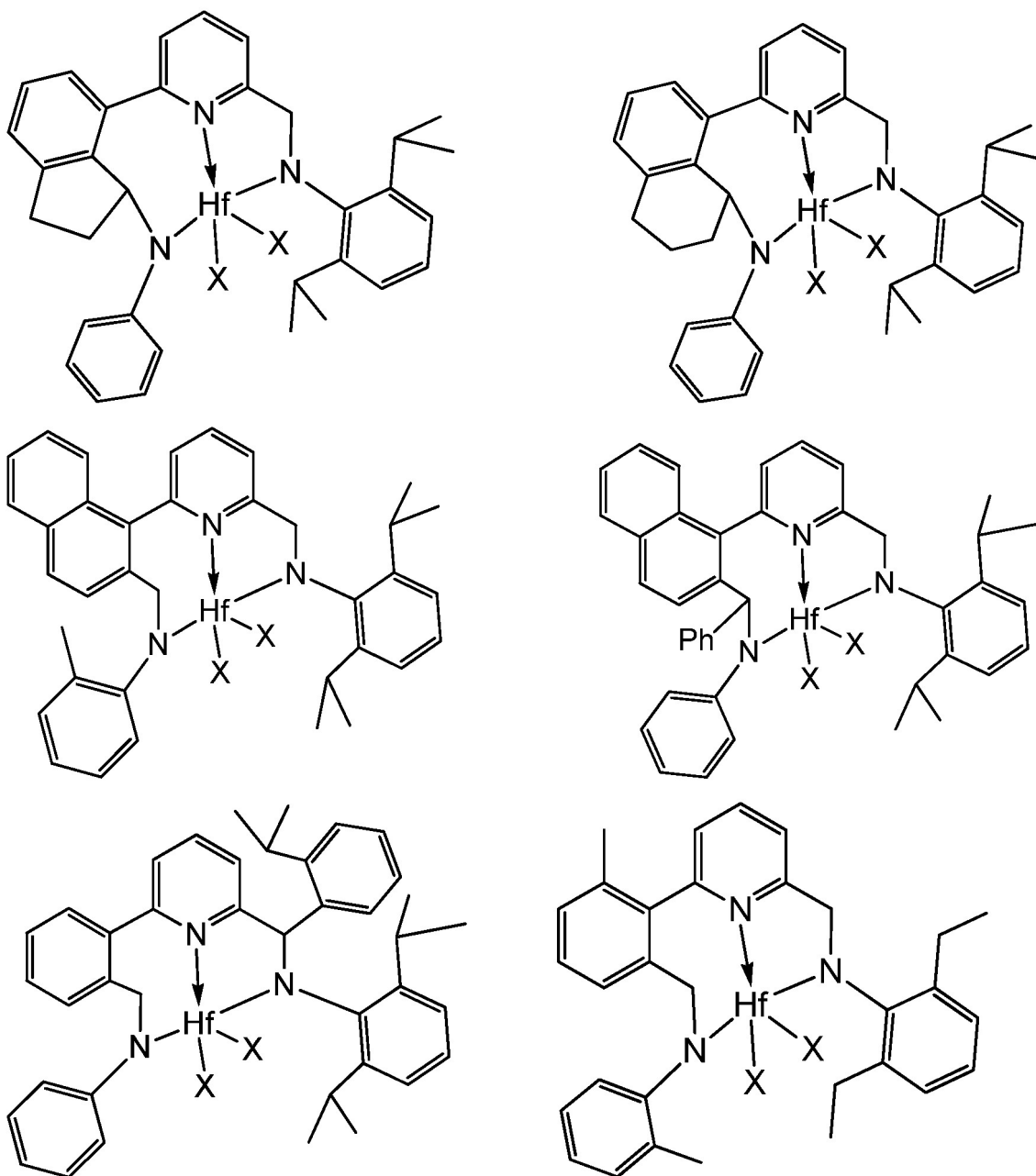
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[0085] In an embodiment of the invention, R⁵² and R⁵¹ are independently aryl or substituted aryl; preferably R⁵¹ is a substituted phenyl group such as, but not limited to 2,6-diisopropylphenyl, 2,6-diethylphenyl, 2,6-dimethylphenyl, mesityl, and the like, and preferably R⁵² is phenyl or a substituted phenyl group such as, but not limited to 2-tolyl, 2-

ethylphenyl, 2-propylphenyl, 2-trifluoromethylphenyl, 2-fluorophenyl, mesityl, 2,6-diisopropylphenyl, 2,6-diethylphenyl, 2,6-dimethylphenyl, 3,5-di-tert-butylphenyl, and the like.

[0086] In an embodiment of the invention, R^{54} , R^{55} , R^{61} - R^{66} , each R^{70} - R^{71} are hydrogen, R^{52} is phenyl, R^{51} is 2,6-diisopropylphenyl and t is 2.

[0087] Non-limiting examples of pyridyl diamide catalysts that are useful herein are illustrated below, wherein X is methyl, benzyl, or chloro:

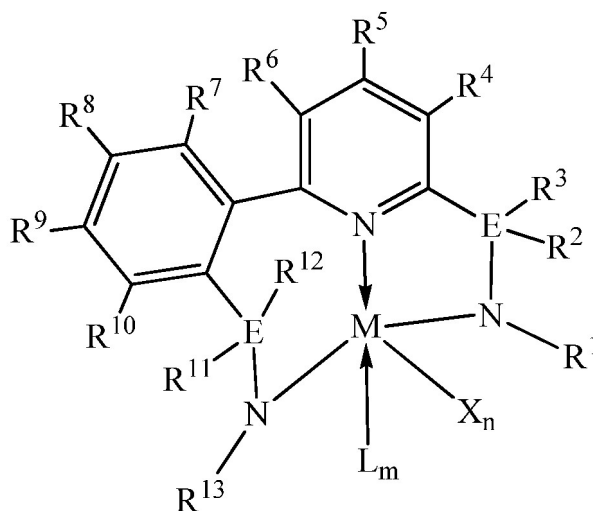


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[0088] Additional pyridyl diamide transition metal complexes useful herein are described in US 2014/0316089; WO 2012/134614; WO 2012/134615; WO 2012/134613; US

2012/0071616; US 2011/0301310; and US 2010/0022726 and are incorporated by reference.

[0089] Transition metal complexes (also referred to as catalyst complexes or pre-catalyst complexes) useful herein include pyridyldiamido transition metal complexes represented by the formula (A):



5

wherein:

M is a Group 3, 4, 5, 6, 7, 8, 9, 10, 11, or 12 metal;

E is selected from carbon, silicon, or germanium, preferably carbon;

10 X is an anionic leaving group, preferably alkyl, aryl, hydride, alkylsilane, fluoride, chloride, bromide, iodide, triflate, carboxylate, alkylsulfonate;

L is a neutral Lewis base, preferably ether, amine, thioether;

R¹ and R¹³ are each independently selected from the group consisting of hydrocarbyls, substituted hydrocarbyls, and silyl groups, preferably aryl;

15 R², R³, R⁴, R⁵, R⁶, R⁷, R⁸, R⁹, R¹⁰, R¹¹, and R¹² are each independently selected from the group consisting of hydrogen, hydrocarbyls, alkoxy, silyl, amino, aryloxy, substituted hydrocarbyls, halogen, and phosphino;

n is 1 or 2;

m is 0, 1, or 2; and

two X groups may be joined together to form a dianionic group;

20 two L groups may be joined together to form a bidentate Lewis base;

an X group may be joined to an L group to form a monoanionic bidentate group;

R⁷ and R⁸ may be joined to form a ring, preferably an aromatic ring, a six-membered aromatic ring with the joined R⁷ and R⁸ group being -CH=CHCH=CH-;

R¹⁰ and R¹¹ may be joined to form a ring, preferably a five-membered ring with the joined

25 R¹⁰ and R¹¹ group being -CH₂CH₂-, a six-membered ring with the joined R¹⁰ and R¹¹ group

being -CH₂CH₂CH₂-.

[0090] In a preferred embodiment, R⁴, R⁵, and R⁶ are independently selected from the group consisting of hydrogen, hydrocarbyls, substituted hydrocarbyls, alkoxy, aryloxy, halogen, amino, and silyl, and wherein adjacent R groups (R⁴ & R⁵ and/or R⁵ & R⁶) may be
5 joined to form a substituted or unsubstituted hydrocarbyl or heterocyclic ring, where the ring has 5, 6, 7, or 8 ring atoms and where substitutions on the ring can join to form additional rings.

[0091] In another preferred embodiment, R⁷, R⁸, R⁹, and R¹⁰ are independently selected from the group consisting of hydrogen, hydrocarbyls, substituted hydrocarbyls, alkoxy,
10 halogen, amino, and silyl, and wherein adjacent R groups (R⁷ & R⁸, and/or R⁹ & R¹⁰) may be joined to form a saturated, substituted or unsubstituted hydrocarbyl or heterocyclic ring, where the ring has 5, 6, 7, or 8 ring carbon atoms and where substitutions on the ring can join to form additional rings.

[0092] In still another preferred embodiment, n + m is not greater than 4.

15 [0093] In yet another preferred embodiment, R² and R³ are each, independently, selected from the group consisting of hydrogen, hydrocarbyls, and substituted hydrocarbyls, alkoxy, silyl, amino, aryloxy, halogen, and phosphino, R² and R³ may be joined to form a saturated, substituted or unsubstituted hydrocarbyl ring, where the ring has 4, 5, 6, or 7 ring carbon atoms and where substitutions on the ring can join to form additional rings, or R² and R³ may
20 be joined to form a saturated heterocyclic ring, or a saturated substituted heterocyclic ring where substitutions on the ring can join to form additional rings.

[0094] In still yet another preferred embodiment, R¹¹ and R¹² are each, independently, selected from the group consisting of hydrogen, hydrocarbyls, and substituted hydrocarbyls, alkoxy, silyl, amino, aryloxy, halogen, and phosphino, R¹¹ and R¹² may be joined to form a
25 saturated, substituted or unsubstituted hydrocarbyl ring, where the ring has 4, 5, 6, or 7 ring carbon atoms and where substitutions on the ring can join to form additional rings, or R¹¹ and R¹² may be joined to form a saturated heterocyclic ring, or a saturated substituted heterocyclic ring where substitutions on the ring can join to form additional rings.

[0095] In a preferred embodiment, R¹ and R¹³ may be independently selected from
30 phenyl groups that are variously substituted with between zero to five substituents that include F, Cl, Br, I, CF₃, NO₂, alkoxy, dialkylamino, aryl, and alkyl groups having 1 to 10 carbons, such as methyl, ethyl, propyl, butyl, pentyl, hexyl, heptyl, octyl, nonyl, decyl, and isomers thereof.

[0096] Preferred R^3 -E- R^2 groups and preferred R^{12} -E- R^{11} groups include CH_2 , CMe_2 , $SiMe_2$, $SiEt_2$, $SiPr_2$, $SiBu_2$, $SiPh_2$, $Si(aryl)_2$, $Si(alkyl)_2$, $CH(aryl)$, $CH(Ph)$, $CH(alkyl)$, and $CH(2\text{-isopropylphenyl})$, where alkyl is a C_1 to C_{40} alkyl group (preferably C_1 to C_{20} alkyl, preferably one or more of methyl, ethyl, propyl, butyl, pentyl, hexyl, heptyl, octyl, nonyl, decyl, undecyl, dodecyl, and isomers thereof), aryl is a C_5 to C_{40} aryl group (preferably a C_6 to C_{20} aryl group, preferably phenyl or substituted phenyl, preferably phenyl, 2-isopropylphenyl, or 2-tertbutylphenyl).

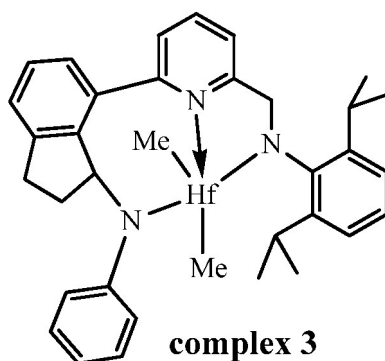
[0097] In a preferred embodiment, each X may be independently selected from halide, alkyl, aryl, alkoxy, amido, hydrido, phenoxy, hydroxy, silyl, allyl, alkenyl, triflate, alkylsulfonate, arylsulfonate, and alkynyl. The selection of the leaving groups depends on the synthesis route adopted for arriving at the complex and may be changed by additional reactions to suit the later activation method in polymerization. For example, alkyl is preferred when using non-coordinating anions such as N,N-dimethylanilinium tetrakis(pentafluorophenyl)-borate or tris(pentafluorophenyl)borane. In another embodiment, two L groups may be linked to form a dianionic leaving group, for example, oxalate.

[0098] In another embodiment of the invention, each L is independently selected from the group consisting of ethers, thio-ethers, amines, nitriles, imines, pyridines, and phosphines, preferably ethers.

[0099] In any embodiment of the invention described herein, M is preferably a Group 4 metal, preferably Zr or Hf.

[00100] In any embodiment of the invention described herein, each E is preferably carbon.

[00101] In any embodiment of the invention described herein, the transition metal complex is represented by the formula:



[00102] The pyridyl diamine ligands described herein are generally prepared in multiple steps in accordance with the disclosure of US 9,290,519. An important step involves the preparation of a suitable "linker" group(s) containing both an aryl boronic acid (or acid ester)

and an amine group. Examples of these include compounds of the general formula: 7-(boronic acid)-2,3-dihydro-1H-inden-1-(amine), 7-(boronic acid ester)-2,3-dihydro-1H-1-(amine), 7-(boronic acid)-1,2,3,4-tetrahydronaphthalen-1-(amine), 7-(boronic acid ester)-1,2,3,4-tetrahydronaphthalen-1-(amine), which include various boronic acids, boronic acid esters, and amines. The linker groups may be prepared in high yield from arylhalide precursors containing amine functionality by first deprotonation of the amine group with 1.0 molar equivalents of n-BuLi, followed by transmetalation of an arylhalide with t-BuLi and subsequent reaction with a boron-containing reagent. This amine-containing linker is then coupled with a suitable pyridine containing species, such as 6-bromo-2-pyridinecarboxaldehyde. This coupling step typically uses a metal catalyst (e.g., Pd(PPh₃)₄) in less than 5 mol% loading. Following this coupling step, the new derivative, which can be described as amine-linker-pyridine-aldehyde, is then reacted with a second amine to produce the imine derivative amine-linker-pyridine-imine in a condensation reaction. This can then be reduced to the pyridyl diamine ligand by reaction with a suitable aryl anion, alkyl anion, or hydride source. This reaction is generally performed in etherial solvents at temperatures between -100°C and 50°C when aryllithium or alkyl lithium reagents are employed. This reaction is generally performed in methanol at reflux when sodium cyanoborohydride is employed.

[00103] The preparation of pyridyl diamide metal complexes from pyridyl diamines may be accomplished using typical protonolysis and methylation reactions. In the protonolysis reaction the pyridyl diamine is reacted with a suitable metal reactant to produce a pyridyldiamide metal complex. A suitable metal reactant will feature a basic leaving group that will accept a proton from the pyridyl diamine and then generally depart and be removed from the product. Suitable metal reactants include, but are not limited to, HfBn₄ (Bn = CH₂Ph), ZrBn₄, TiBn₄, ZrBn₂Cl₂(OEt₂), HfBn₂Cl₂(OEt₂)₂, Zr(NMe₂)₂Cl₂(dimethoxyethane), Hf(NMe₂)₂Cl₂(dimethoxyethane), Hf(NMe₂)₄, and Hf(NEt₂)₄. Pyridyldiamide metal complexes that contain metal-chloride groups, such as the PDA dichloride complex, can be alkylated by reaction with an appropriate organometallic reagent. Suitable reagents include organolithium and organomagnesium, and Grignard reagents. The alkylations are generally performed in etherial or hydrocarbon solvents or solvent mixtures at temperatures typically ranging from -100°C to 50°C.

[00104] Another route to pyridyl diamide and other complexes of interest as catalysts involves the insertion of an unsaturated molecule into a covalent metal-carbon bond where the covalently bonded group is part of a multidentate ligand structure, such as that described

by Boussie et al. in US 6,750,345. The unsaturated molecule will generally have a carbon-X double or triple bond where X is a group 14 or group 15 or group 16 element. Examples of unsaturated molecules include alkenes, alkynes, imines, nitriles, ketones, aldehydes, amides, formamides, carbon dioxide, isocyanates, thioisocyanates, and carbodiimides. Examples
5 showing the insertion reactions involving imines and carbonyls are found in US 7,973,116 and US 2012/0071616.

[00105] In a preferred embodiment of the invention, the transition metal complex is not a metallocene. A metallocene catalyst is defined as an organometallic compound with at least one π -bound cyclopentadienyl moiety (or substituted cyclopentadienyl moiety) and more
10 frequently two π -bound cyclopentadienyl moieties or substituted cyclopentadienyl moieties.

[00106] Usefully, the single site catalyst compounds useful herein are preferably tridentate ligands bound to the transition metal (such as a group 4 metal), preferably tridentate N,N,N ligands bound to a transition metal (such as a Zr, Hf or Ti).

Activators

[00107] The catalyst systems typically comprise a transition metal complex as described above and an activator such as alumoxane or a non-coordinating anion. Activation may be performed using alumoxane solution including methyl alumoxane, referred to as MAO, as well as modified MAO, referred to herein as MMAO, containing some higher alkyl groups to improve the solubility. Particularly useful MAO can be purchased from Albemarle, typically
20 in a 10 wt% solution in toluene. The catalyst system employed in the present invention may use an activator selected from alumoxanes, such as methyl alumoxane, modified methyl alumoxane, ethyl alumoxane, iso-butyl alumoxane, and the like.

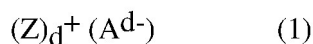
[00108] When an alumoxane or modified alumoxane is used, the complex-to-activator molar ratio is from about 1:3000 to 10:1; alternatively 1:2000 to 10:1; alternatively 1:1000 to
25 10:1; alternatively, 1:500 to 1:1; alternatively 1:300 to 1:1; alternatively 1:200 to 1:1; alternatively 1:100 to 1:1; alternatively 1:50 to 1:1; alternatively 1:10 to 1:1. When the activator is an alumoxane (modified or unmodified), some embodiments select the maximum amount of activator at a 5000-fold molar excess over the catalyst precursor (per metal catalytic site). The preferred minimum activator-to-complex ratio is 1:1 molar ratio.

[00109] Activation may also be performed using non-coordinating anions, referred to as NCA's, of the type described in EP 277 003 A1 and EP 277 004 A1. NCA may be added in the form of an ion pair using, for example, [DMAH]⁺ [NCA]⁻ in which the N,N-dimethylanilinium (DMAH) cation reacts with a basic leaving group on the transition metal complex to form a transition metal complex cation and [NCA]⁻. The cation in the precursor
30

may, alternatively, be trityl. Alternatively, the transition metal complex may be reacted with a neutral NCA precursor, such as $B(C_6F_5)_3$, which abstracts an anionic group from the complex to form an activated species. Useful activators include N,N-dimethylanilinium tetrakis (pentafluorophenyl)borate (i.e., $[PhNMe_2H]B(C_6F_5)_4$) and N,N-dimethylanilinium tetrakis (heptafluoronaphthyl)borate, where Ph is phenyl, and Me is methyl.

[00110] Additionally preferred activators useful herein include those described in US 7,247,687 at column 169, line 50 to column 174, line 43, particularly column 172, line 24 to column 173, line 53.

[00111] In an embodiment of the invention described herein, the non-coordinating anion activator is represented by the following formula (1):



wherein Z is (L-H) or a reducible Lewis acid; L is a neutral Lewis base; H is hydrogen and (L-H)⁺ is a Bronsted acid; A^{d-} is a non-coordinating anion having the charge d-; and d is an integer from 1 to 3.

[00112] When Z is (L-H) such that the cation component is (L-H)^{d+}, the cation component may include Bronsted acids such as protonated Lewis bases capable of protonating a moiety, such as an alkyl or aryl, from the catalyst precursor, resulting in a cationic transition metal species, or the activating cation (L-H)^{d+} is a Bronsted acid, capable of donating a proton to the catalyst precursor resulting in a transition metal cation, including ammoniums, oxoniums, phosphoniums, silyliums, and mixtures thereof, or ammoniums of methylamine, aniline, dimethylamine, diethylamine, N-methylaniline, diphenylamine, trimethylamine, triethylamine, N,N-dimethylaniline, methyldiphenylamine, pyridine, p-bromo N,N-dimethylaniline, p-nitro-N,N-dimethylaniline, phosphoniums from triethylphosphine, triphenylphosphine, and diphenylphosphine, oxoniums from ethers, such as dimethyl ether diethyl ether, tetrahydrofuran, and dioxane, sulfoniums from thioethers, such as diethyl thioethers and tetrahydrothiophene, and mixtures thereof.

[00113] When Z is a reducible Lewis acid, it may be represented by the formula: (Ar₃C⁺), where Ar is aryl or aryl substituted with a heteroatom, or a C₁ to C₄₀ hydrocarbyl, the reducible Lewis acid may be represented by the formula: (Ph₃C⁺), where Ph is phenyl or phenyl substituted with a heteroatom, and/or a C₁ to C₄₀ hydrocarbyl. In an embodiment, the reducible Lewis acid is triphenyl carbenium.

[00114] Embodiments of the anion component A^{d-} include those having the formula

[M_k+Q_n]d⁻ wherein k is 1, 2, or 3; n is 1, 2, 3, 4, 5, or 6, or 3, 4, 5, or 6; n - k = d; M is an element selected from Group 13 of the Periodic Table of the Elements, or boron or aluminum, and Q is independently a hydride, bridged or unbridged dialkylamido, halide, alkoxide, aryloxy, hydrocarbyl radicals, said Q having up to 20 carbon atoms with the proviso that in not more than one occurrence is Q a halide, and two Q groups may form a ring structure. Each Q may be a fluorinated hydrocarbyl radical having 1 to 20 carbon atoms, or each Q is a fluorinated aryl radical, or each Q is a pentafluoryl aryl radical. Examples of suitable Ad⁻ components also include diboron compounds as disclosed in US 5,447,895, which is fully incorporated herein by reference.

- 10 **[00115]** In an embodiment in any of the NCA's represented by Formula 1 described above, the anion component Ad⁻ is represented by the formula [M^{*}k^{*}+Q^{*}n^{*}]d⁻ wherein k^{*} is 1, 2, or 3; n^{*} is 1, 2, 3, 4, 5, or 6 (or 1, 2, 3, or 4); n^{*} - k^{*} = d^{*}; M^{*} is boron; and Q^{*} is independently selected from hydride, bridged or unbridged dialkylamido, halogen, alkoxide, aryloxy, hydrocarbyl radicals, said Q^{*} having up to 20 carbon atoms with the proviso that
- 15 in not more than 1 occurrence is Q^{*} a halogen.

[00116] This invention also relates to a method to polymerize olefins comprising contacting olefins (such as propylene) with a catalyst complex as described above and an NCA activator represented by the Formula (2):



- 20 where R is a monoanionic ligand; M^{**} is a Group 13 metal or metalloid; ArNHal is a halogenated, nitrogen-containing aromatic ring, polycyclic aromatic ring, or aromatic ring assembly in which two or more rings (or fused ring systems) are joined directly to one another or together; and n is 0, 1, 2, or 3. Typically the NCA comprising an anion of Formula 2 also comprises a suitable cation that is essentially non-interfering with the ionic
- 25 catalyst complexes formed with the transition metal compounds, or the cation is Z_d⁺ as described above.

- [00117]** In an embodiment in any of the NCA's comprising an anion represented by Formula 2 described above, R is selected from the group consisting of C₁ to C₃₀ hydrocarbyl radicals. In an embodiment, C₁ to C₃₀ hydrocarbyl radicals may be substituted with one or
- 30 more C₁ to C₂₀ hydrocarbyl radicals, halide, hydrocarbyl substituted organometalloid, dialkylamido, alkoxy, aryloxy, alkylsulfido, arylsulfido, alkylphosphido, arylphosphide, or other anionic substituent; fluoride; bulky alkoxides, where bulky means C₄ to C₂₀ hydrocarbyl radicals; --SRa, --NRa₂, and --PRa₂, where each Ra is independently a

monovalent C₄ to C₂₀ hydrocarbyl radical comprising a molecular volume greater than or equal to the molecular volume of an isopropyl substitution or a C₄ to C₂₀ hydrocarbyl substituted organometalloid having a molecular volume greater than or equal to the molecular volume of an isopropyl substitution.

- 5 **[00118]** In an embodiment in any of the NCA's comprising an anion represented by Formula 2 described above, the NCA also comprises cation comprising a reducible Lewis acid represented by the formula: (Ar₃C⁺), where Ar is aryl or aryl substituted with a heteroatom, and/or a C₁ to C₄₀ hydrocarbyl, or the reducible Lewis acid represented by the formula: (Ph₃C⁺), where Ph is phenyl or phenyl substituted with one or more heteroatoms,
10 and/or C₁ to C₄₀ hydrocarbyls.

- [00119]** In an embodiment in any of the NCA's comprising an anion represented by Formula 2 described above, the NCA may also comprise a cation represented by the formula, (L-H)d⁺, wherein L is a neutral Lewis base; H is hydrogen; (L-H) is a Bronsted acid; and d is 1, 2, or 3, or (L-H)d⁺ is a Bronsted acid selected from ammoniums, oxoniums,
15 phosphoniums, silyliums, and mixtures thereof.

[00120] Further examples of useful activators include those disclosed in US 7,297,653 and US 7,799,879, which are fully incorporated by reference herein.

[00121] In an embodiment, an activator useful herein comprises a salt of a cationic oxidizing agent and a noncoordinating, compatible anion represented by the Formula (3):



- wherein OX^{e+} is a cationic oxidizing agent having a charge of e+; e is 1, 2, or 3; d is 1, 2, or 3; and A^{d-} is a non-coordinating anion having the charge of d- (as further described above). Examples of cationic oxidizing agents include: ferrocenium, hydrocarbyl-substituted ferrocenium, Ag⁺, or Pb⁺². Suitable embodiments of A^{d-} include
25 tetrakis(pentafluorophenyl)borate.

- [00122]** Activators useful in catalyst systems herein include: trimethylammonium tetrakis(perfluoronaphthyl)borate, N,N-dimethylanilinium tetrakis(perfluoronaphthyl)borate, N,N-diethylanilinium tetrakis(perfluoronaphthyl)borate, triphenylcarbenium tetrakis(perfluoronaphthyl)borate, trimethylammonium tetrakis(perfluorobiphenyl)borate,
30 N,N-dimethylanilinium tetrakis(perfluorobiphenyl)borate, triphenylcarbenium tetrakis(perfluorobiphenyl)borate, and the types disclosed in US 7,297,653, which is fully incorporated by reference herein.

[00123] Suitable activators also include:

N,N-dimethylanilinium tetrakis(perfluoronaphthyl)borate, N,N-dimethylanilinium tetrakis(perfluorobiphenyl)borate, N,N-dimethylanilinium tetrakis(3,5-bis(trifluoromethyl)phenyl)borate, triphenylcarbenium tetrakis(perfluoronaphthyl)borate, triphenylcarbenium tetrakis(perfluorobiphenyl)borate, triphenylcarbenium tetrakis(3,5-bis(trifluoromethyl)phenyl)borate, triphenylcarbenium tetrakis(perfluorophenyl)borate, $[\text{Ph}_3\text{C}^+][\text{B}(\text{C}_6\text{F}_5)_4^-]$, $[\text{Me}_3\text{NH}^+][\text{B}(\text{C}_6\text{F}_5)_4^-]$; 1-(4-(tris(pentafluorophenyl)borate)-2,3,5,6-tetrafluorophenyl)pyrrolidinium; and tetrakis(pentafluorophenyl)borate, 4-(tris(pentafluorophenyl)borate)-2,3,5,6-tetrafluoropyridine.

10 [00124] In an embodiment, the activator comprises a triaryl carbonium (such as triphenylcarbenium tetraphenylborate, triphenylcarbenium tetrakis(pentafluorophenyl)borate, triphenylcarbenium tetrakis-(2,3,4,6-tetrafluorophenyl)borate, triphenylcarbenium tetrakis(perfluoronaphthyl)borate, triphenylcarbenium tetrakis(perfluorobiphenyl)borate, triphenylcarbenium tetrakis(3,5-bis(trifluoromethyl)phenyl)borate).

15 [00125] In an embodiment, two NCA activators may be used in the polymerization and the molar ratio of the first NCA activator to the second NCA activator can be any ratio. In an embodiment, the molar ratio of the first NCA activator to the second NCA activator is 0.01:1 to 10,000:1, or 0.1:1 to 1000:1, or 1:1 to 100:1.

20 [00126] In an embodiment of the invention, the NCA activator-to-catalyst ratio is a 1:1 molar ratio, or 0.1:1 to 100:1, or 0.5:1 to 200:1, or 1:1 to 500:1 or 1:1 to 1000:1. In an embodiment, the NCA activator-to-catalyst ratio is 0.5:1 to 10:1, or 1:1 to 5:1.

[00127] In an embodiment, the catalyst compounds can be combined with combinations of alumoxanes and NCA's (see for example, US 5,153,157; US 5,453,410; EP 0 573 120 B1; WO 94/07928; and WO 95/14044 which discuss the use of an alumoxane in combination with an ionizing activator, all of which are incorporated by reference herein).

25 [00128] In a preferred embodiment of the invention, when an NCA (such as an ionic or neutral stoichiometric activator) is used, the complex-to-activator molar ratio is typically from 1:10 to 1:1; 1:10 to 10:1; 1:10 to 2:1; 1:10 to 3:1; 1:10 to 5:1; 1:2 to 1.2:1; 1:2 to 10:1; 1:2 to 2:1; 1:2 to 3:1; 1:2 to 5:1; 1:3 to 1.2:1; 1:3 to 10:1; 1:3 to 2:1; 1:3 to 3:1; 1:3 to 5:1; 1:5 to 1:1; 1:5 to 10:1; 1:5 to 2:1; 1:5 to 3:1; 1:5 to 5:1; 1:1 to 1:1.2.

30 [00129] Alternately, a co-activator or chain transfer agent, such as a group 1, 2, or 13 organometallic species (e.g., an alkyl aluminum compound such as tri-n-octyl aluminum), may also be used in the catalyst system herein. The complex-to-co-activator molar ratio is from 1:100 to 100:1; 1:75 to 75:1; 1:50 to 50:1; 1:25 to 25:1; 1:15 to 15:1; 1:10 to 10:1; 1:5

to 5:1; 1:2 to 2:1; 1:100 to 1:1; 1:75 to 1:1; 1:50 to 1:1; 1:25 to 1:1; 1:15 to 1:1; 1:10 to 1:1; 1:5 to 1:1; 1:2 to 1:1; 1:10 to 2:1.

Metal Hydrocarbenyl Transfer Agents (Aluminum Vinyl Transfer Agents)

[00130] The catalyst systems described herein further comprise a metal hydrocarbenyl transfer agent (which is any group 13 metal agent that contains at least one transferrable group that has an allyl chain end), preferably an aluminum vinyl-transfer agent, also referred to as an AVTA, (which is any aluminum agent that contains at least one transferrable group that has an allyl chain end). An allyl chain end is represented by the formula $\text{H}_2\text{C}=\text{CH}-\text{CH}_2-$. "Allylic vinyl group," "allyl chain end," "vinyl chain end," "vinyl termination," "allylic vinyl group," "terminal vinyl group," and "vinyl terminated" are used interchangeably herein and refer to an allyl chain end. An allyl chain end is not a vinylidene chain end or a vinylene chain end. The number of allyl chain ends, vinylidene chain ends, vinylene chain ends, and other unsaturated chain ends is determined using ^1H NMR at 120°C using deuterated tetrachloroethane as the solvent on an at least 250 MHz NMR spectrometer.

[00131] Useful transferable groups containing an allyl chain end are represented by the formula $\text{CH}_2=\text{CH}-\text{CH}_2-\text{R}^*$, where R^* represents a hydrocarbyl group or a substituted hydrocarbyl group, such as a C_1 to C_{20} alkyl, preferably methyl, ethyl, propyl, butyl, pentyl, hexyl, heptyl, octyl, nonyl, decyl, undecyl, dodecyl, or an isomer thereof.

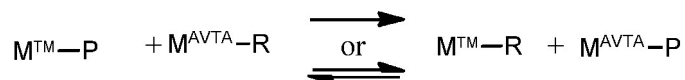
[00132] In the catalyst system described herein, the catalyst undergoes alkyl group transfer with the transfer agent, which enables the formation of polymer chains containing one or more allyl chain ends.

[00133] Useful transferable groups containing an allyl chain end also include those represented by the formula $\text{CH}_2=\text{CH}-\text{CH}_2-\text{R}^{**}$, where R^{**} represents a hydrocarbenyl group or a substituted hydrocarbenyl group, such as a C_1 to C_{20} alkylene, preferably methylene (CH_2), ethylene [$(\text{CH}_2)_2$], propandiyl [$(\text{CH}_2)_3$], butandiyl [$(\text{CH}_2)_4$], pentandiyl [$(\text{CH}_2)_5$], hexandiyl [$(\text{CH}_2)_6$], heptandiyl [$(\text{CH}_2)_7$], octandiyl [$(\text{CH}_2)_8$], nonandiyl [$(\text{CH}_2)_9$], decandiyl [$(\text{CH}_2)_{10}$], undecandiyl [$(\text{CH}_2)_{11}$], dodecandiyl [$(\text{CH}_2)_{12}$], or an isomer thereof. Useful transferable groups are preferably non-substituted linear hydrocarbenyl groups. Preferably, at least one R^{**} is a C_4 - C_{20} hydrocarbenyl group.

[00134] The term "hydrocarbenyl" refers to a hydrocarb-di-yl divalent group, such as a C_1 to C_{20} alkylene (i.e., methylene (CH_2), ethylene [$(\text{CH}_2)_2$], propandiyl [$(\text{CH}_2)_3$], butandiyl [$(\text{CH}_2)_4$], pentandiyl [$(\text{CH}_2)_5$], hexandiyl [$(\text{CH}_2)_6$], heptandiyl [$(\text{CH}_2)_7$],

octandiyl [(CH₂)₈], nonandiyl [(CH₂)₉], decandiyl [(CH₂)₁₀], undecandiyl [(CH₂)₁₁], dodecandiyl [(CH₂)₁₂], and isomers thereof).

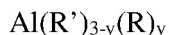
[00135] AVTA's are alkenylaluminum reagents capable of causing group exchange between the transition metal of the catalyst system (MTM) and the metal of the AVTA (M^{AVTA}). The reverse reaction may also occur such that the polymeryl chain is transferred back to the transition metal of the catalyst system. This reaction scheme is illustrated below:



wherein MTM is an active transition metal catalyst site and P is the polymeryl chain, M^{AVTA} is the metal of the AVTA, and R is a transferable group containing an allyl chain end, such as a hydrocarbyl group containing an allyl chain end, also called a hydrocarbenyl or alkenyl group.

[00136] Catalyst systems of this invention preferably have high rates of olefin propagation and negligible or no chain termination via beta hydride elimination, beta methyl elimination, or chain transfer to monomer relative to the rate of chain transfer to the AVTA or other chain transfer agent, such as an aluminum alkyl, if present. Pyridyldiamido catalyst complexes (see US 7,973,116; US 8,394,902; US 8,674,040; US 8,710,163; US 9,102,773; US 2014/0256893; US 2014/0316089; and US 2015/0141601) activated with non-coordinating activators such as dimethylanilinium tetrakis(perfluorophenyl)borate and/or dimethylanilinium tetrakis(perfluoronaphthyl)borate are particularly useful in the catalyst systems of this invention. Compound **3**, described above is particularly preferred.

[00137] In any embodiment of the invention described herein, the catalyst system comprises an aluminum vinyl transfer agent, which is typically represented by the formula (I):



where R is a hydrocarbenyl group containing 4 to 20 carbon atoms having an allyl chain end, R' is a hydrocarbyl group containing 1 to 30 carbon atoms, and v is 0.1 to 3, alternately 1 to 3, alternately 1.1 to less than 3, alternately v is 0.5 to 2.9, 1.1 to 2.9, alternately 1.5 to 2.7, alternately 1.5 to 2.5, alternately 1.8 to 2.2. The compounds represented by the formula Al(R')_{3-v}(R)_v are typically a neutral species, but anionic formulations may be envisioned, such as those represented by formula (II): [Al(R')_{4-w}(R)_w]⁻, where w is 0.1 to 4, R is a hydrocarbenyl group containing 4 to 20 carbon atoms having an allyl chain end, and R' is a hydrocarbyl group containing 1 to 30 carbon atoms.

[00138] In any embodiment of any formula for a metal hydrocarbenyl chain transfer agent,

such as formula I or II, described herein, each R' is independently chosen from C₁ to C₃₀ hydrocarbyl groups (such as a C₁ to C₂₀ alkyl groups, preferably methyl, ethyl, propyl, butyl, pentyl, hexyl, heptyl, octyl, nonyl, decyl, undecyl, dodecyl, or an isomer thereof), and R is represented by the formula:



where n is an integer from 2 to 18, preferably between 6 to 18, preferably 6 to 12, preferably 6. In any embodiment of the invention described herein, particularly useful AVTAs include, but are not limited to, tri(but-3-en-1-yl)aluminum, tri(pent-4-en-1-yl)aluminum, tri(oct-7-en-1-yl)aluminum, tri(non-8-en-1-yl)aluminum, tri(dec-9-en-1-yl)aluminum, dimethyl(oct-7-en-1-yl)aluminum, diethyl(oct-7-en-1-yl)aluminum, dibutyl(oct-7-en-1-yl)aluminum, diisobutyl(oct-7-en-1-yl)aluminum, diisobutyl(non-8-en-1-yl)aluminum, diisobutyl(dec-8-en-1-yl)aluminum, diisobutyl(dodec-10-en-1-yl)aluminum, and the like. Mixtures of one or more AVTAs may also be used. In some embodiments of the invention, isobutyl-di(oct-7-en-1-yl)-aluminum, isobutyl-di(dec-9-en-1-yl)-aluminum, isobutyl-di(non-8-en-1-yl)-aluminum, isobutyl-di(hept-6-en-1-yl)-aluminum are preferred.

[00139] Useful aluminum vinyl transfer agents include organoaluminum compound reaction products between aluminum reagent (AlR₃) and an alkyl diene. Suitable alkyl dienes include those that have two "alpha olefins", as described above, at two termini of the carbon chain. The alkyl diene can be a straight chain or branched alkyl chain and substituted or unsubstituted. Exemplary alkyl dienes include but are not limited to, for example, 1,3-butadiene, 1,4-pentadiene, 1,6-heptadiene, 1,7-octadiene, 1,8-nonadiene, 1,9-decadiene, 1,10-undecadiene, 1,11-dodecadiene, 1,12-tridecadiene, 1,13-tetradecadiene, 1,14-pentadecadiene, 1,15-hexadecadiene, 1,16-heptadecadiene, 1,17-octadecadiene, 1,18-nonadecadiene, 1,19-eicosadiene, 1,20-heneicosadiene, etc. Exemplary aluminum reagents include triisobutylaluminum, diisobutylaluminumhydride, isobutylaluminumdihydride and aluminum hydride (AlH₃).

[00140] In any embodiment of the invention described herein, R" is butenyl, pentenyl, heptenyl, or octenyl. In some embodiments R" is preferably octenyl.

[00141] In any embodiment of the invention described herein, R' is methyl, ethyl, propyl, isobutyl, or butyl. In any embodiment of the invention described herein, R' is isobutyl.

[00142] In any embodiment of the invention described herein, v is about 2, or v is 2.

[00143] In any embodiment of the invention described herein, v is about 1, or v is 1, preferably from about 1 to about 2.

[00144] In any embodiment of the invention described herein, v is an integer or a non-integer, preferably v is from 1.1 to 2.9, from about 1.5 to about 2.7, e.g., from about 1.6 to about 2.4, from about 1.7 to about 2.4, from about 1.8 to about 2.2, from about 1.9 to about 2.1 and all ranges there between.

5 [00145] In preferred embodiments of the invention described herein, R' is isobutyl and each R'' is octenyl, preferably R' is isobutyl, each R'' is octenyl, and v is from 1.1 to 2.9, from about 1.5 to about 2.7, e.g., from about 1.6 to about 2.4, from about 1.7 to about 2.4, from about 1.8 to about 2.2, from about 1.9 to about 2.1.

[00146] The amount of v (the aluminum alkenyl) is described using the formulas: $(3-v) + v = 3$, and $Al(R')_{3-v}(R'')_v$ where R'' is a hydrocarbenyl group containing 4 to 20 carbon atoms having an allyl chain end, R' is a hydrocarbyl group containing 1 to 30 carbon atoms, and v is 0.1 to 3 (preferably 1.1 to 3). This formulation represents the observed average of organoaluminum species (as determined by 1H NMR) present in a mixture, which may include any of $Al(R')_3$, $Al(R')_2(R'')$, $Al(R')(R'')_2$, and $Al(R'')_3$. 1H NMR spectroscopic studies
10 are performed at room temperature using a Bruker 400 MHz NMR. Data is collected using samples prepared by dissolving 10-20 mg the compound in 1 mL of C_6D_6 . Samples are then loaded into 5 mm NMR tubes for data collection. Data is recorded using a maximum pulse width of 45° , 8 seconds between pulses and signal averaging either 8 or 16 transients. The spectra are normalized to protonated tetrachloroethane in the C_6D_6 . The chemical shifts (δ)
15 are reported as relative to the residual protium in the deuterated solvent at 7.15 ppm.

[00147] In still another aspect, the aluminum vinyl-transfer agent has less than 50 wt% dimer present, based upon the weight of the AVTA, preferably less than 40 wt%, preferably less than 30 wt%, preferably less than 20 wt%, preferably less than 15 wt%, preferably less than 10 wt%, preferably less than 5 wt%, preferably less than 2 wt%, preferably less than 1
25 wt%, preferably 0 wt% dimer. Alternately dimer is present at from 0.1 to 50 wt%, alternately 1 to 20 wt%, alternately at from 2 to 10 wt%. Dimer is the dimeric product of the alkyl diene used in the preparation of the AVTA. The dimer can be formed under certain reaction conditions, and is formed from the insertion of a molecule of diene into the Al-R bond of the AVTA, followed by beta-hydride elimination. For example, if the alkyl diene used is 1,7-octadiene, the dimer is 7-methylenepentadeca-1,14-diene. Similarly, if the alkyl diene is 1,9-
30 decadiene, the dimer is 9-methylenonadeca-1,18-diene.

[00148] Useful compounds can be prepared by combining an aluminum reagent (such as alkyl aluminum) having at least one secondary alkyl moiety (such as triisobutylaluminum) and/or at least one hydride, such as a dialkylaluminum hydride, a monoalkylaluminum

dihydride or aluminum trihydride (aluminum hydride, AlH_3) with an alkyl diene and heating to a temperature that causes release of an alkylene byproduct. The use of solvent(s) is not required. However, non-polar solvents can be employed, such as, as hexane, pentane, toluene, benzene, xylenes, and the like, or combinations thereof.

5 [00149] In an embodiment of the invention, the AVTA is free of coordinating polar solvents such as tetrahydrofuran and diethylether.

[00150] After the reaction is complete, solvent if, present can be removed and the product can be used directly without further purification.

10 [00151] The AVTA to catalyst complex equivalence ratio can be from about 1:100 to 500,000:1. More preferably, the molar ratio of AVTA to catalyst complex is greater than 5, alternately greater than 10, alternately greater than 15, alternately greater than 20, alternately greater than 25, alternately greater than 30.

[00152] The metal hydrocarbenyl chain transfer agent may be represented by the formula: $\text{E}[\text{Al}(\text{R}')_{2-v}(\text{R})_v]_2$ wherein E is a group 16 element (such as O or S, preferably O); each R',
15 independently, is a C_1 to C_{30} hydrocarbyl group (such as a C_1 to C_{20} alkyl group, preferably methyl, ethyl, propyl, butyl, pentyl, hexyl, heptyl, octyl, nonyl, decyl, undecyl, dodecyl, or an isomer thereof); each R'', independently, is a C_4 to C_{20} hydrocarbenyl group having an allyl chain end (such as a C_1 to C_{20} alkenyl group, preferably methenyl, ethenyl, propenyl, butenyl, pentenyl, heptenyl, octenyl, nonenyl, decenyl, undecenyl, dodecenyl, or an isomer thereof); v is from 0.01 to 3 (such as 1 or 2).

[00153] In another embodiment of the invention, the metal hydrocarbenyl chain transfer agent is an alumoxane formed from the hydrolysis of the AVTA. Alternatively, the alumoxane can be formed from the hydrolysis of the AVTA in combination with other aluminum alkyl(s). The alumoxane component is an oligomeric compound which is not well
25 characterized, but can be represented by the general formula $(\text{R}-\text{Al}-\text{O})_m$ which is a cyclic compound, or may be $\text{R}'(\text{R}-\text{Al}-\text{O})_m-\text{AlR}'_2$ which is a linear compound where R' is as defined above and at least one R' is the same as R (as defined above), and m is from about 4 to 25, with a range of 13 to 25 being preferred. Most preferably all R' are R. An alumoxane is generally a mixture of both the linear and cyclic compounds.

30 Supports

[00154] The complexes described herein may be supported (with or without an activator and with or without a transfer agent) by any method effective to support other coordination catalyst systems, effectively meaning that the catalyst so prepared can be used for polymerizing olefin(s) in a heterogeneous process. The catalyst precursor, activator, optional

transfer agent, co-activator if needed, suitable solvent, and support may be added in any order or simultaneously. Typically, the complex, activator, and optional transfer agent may be combined in solvent to form a solution. Then the support is added, and the mixture is stirred for 1 minute to 10 hours. The total solution volume may be greater than the pore volume of the support, but some embodiments limit the total solution volume below that needed to form a gel or slurry (about 90% to 400%, preferably about 100% to 200% of the pore volume). After stirring, the residual solvent is removed under vacuum, typically at ambient temperature and over 10-16 hours. But greater or lesser times and temperatures are possible.

[00155] The complex may also be supported absent the activator, and in that case, the activator (and co-activator if needed) is added to a polymerization process' liquid phase. Additionally, two or more different complexes may be placed on the same support. Likewise, two or more activators or an activator and co-activator may be placed on the same support. Likewise the transfer agent may be added to the polymerization reaction separately from the supported catalyst complex and/or activator.

[00156] Suitable solid particle supports are typically comprised of polymeric or refractory oxide materials, each being preferably porous. Preferably, any support material that has an average particle size greater than 10 μm is suitable for use in this invention. Various embodiments select a porous support material, such as for example, talc, inorganic oxides, inorganic chlorides, for example magnesium chloride and resinous support materials such as polystyrene polyolefin or polymeric compounds or any other organic support material and the like. Some embodiments select inorganic oxide materials as the support material including Group-2, -3, -4, -5, -13, or -14 metal or metalloid oxides. Some embodiments select the catalyst support materials to include silica, alumina, silica-alumina, and their mixtures. Other inorganic oxides may serve either alone or in combination with the silica, alumina, or silica-alumina. These are magnesia, titania, zirconia, and the like. The support can optionally double as the activator component; however, an additional activator may also be used.

[00157] The support material may be pre-treated by any number of methods. For example, inorganic oxides may be calcined, chemically treated with dehydroxylating agents such as aluminum alkyls and the like, or both.

[00158] As stated above, polymeric carriers will also be suitable in accordance with the invention, see, for example, the descriptions in WO 95/15815 and US 5,427,991. The methods disclosed may be used with the catalyst complexes, activators or catalyst systems of this invention to adsorb or absorb them on the polymeric supports, particularly if made up of porous particles, or may be chemically bound through functional groups bound to or in the

polymer chains.

[00159] Useful supports typically have a surface area of from 10-700 m²/g, a pore volume of 0.1-4.0 cc/g and an average particle size of 10-500 μm. Some embodiments select a surface area of 50-500 m²/g, a pore volume of 0.5-3.5 cc/g, or an average particle size of 20-200 μm. Other embodiments select a surface area of 100-400 m²/g, a pore volume of 0.8-3.0 cc/g, and an average particle size of 30-100 μm. Useful supports typically have a pore size of 10-1000 Angstroms, alternatively 50-500 Angstroms, or 75-350 Angstroms.

[00160] The catalyst complexes described herein are generally deposited on the support at a loading level of 10-100 micromoles of complex per gram of solid support; alternately 20-80 micromoles of complex per gram of solid support; or 40-60 micromoles of complex per gram of support. However, greater or lesser values may be used provided that the total amount of solid complex does not exceed the support's pore volume.

Polymerization

[00161] Invention catalyst complexes are useful in polymerizing unsaturated monomers conventionally known to undergo coordination-catalyzed polymerization such as solution, slurry, gas-phase, and high-pressure polymerization. Typically, one or more of the complexes described herein, one or more activators, one or more transfer agents (such as an aluminum vinyl transfer agent) and one or more monomers are contacted to produce polymer. The complexes may be supported and, as such, will be particularly useful in the known, fixed-bed, moving-bed, fluid-bed, slurry, gas phase, solution, or bulk operating modes conducted in single, series, or parallel reactors.

[00162] One or more reactors in series or in parallel may be used in the present invention.

[00163] The complexes, activator, transfer agent, and, when required, co-activator, may be delivered as a solution or slurry, either separately to the reactor, activated in-line just prior to the reactor, or pre-activated and pumped as an activated solution or slurry to the reactor. Polymerizations are carried out in either single reactor operation, in which monomer, comonomers, catalyst/activator/co-activator, optional scavenger, and optional modifiers are added continuously to a single reactor or in series reactor operation, in which the above components are added to each of two or more reactors connected in series. The catalyst components can be added to the first reactor in the series. The catalyst component may also be added to both reactors, with one component being added to the first reaction and another component to other reactors. In one preferred embodiment, the complex is activated in the reactor in the presence of olefin and transfer agent.

[00164] In a particularly preferred embodiment, the polymerization process is a continuous

process.

[00165] Polymerization process used herein typically comprises contacting one or more alkene monomers with the complexes, activators and transfer agents described herein. For purpose of this invention alkenes are defined to include multi-alkenes (such as dialkenes) and alkenes having just one double bond. Polymerization may be homogeneous (solution or bulk polymerization) or heterogeneous (slurry -in a liquid diluent, or gas phase -in a gaseous diluent). In the case of heterogeneous slurry or gas phase polymerization, the complex and activator may be supported. Silica is useful as a support herein. Chain transfer agents (such as hydrogen or diethyl zinc) may be used in the practice of this invention.

[00166] The present polymerization processes may be conducted under conditions preferably including a temperature of about 30°C to about 200°C, preferably from 60°C to 195°C, preferably from 75°C to 190°C. The process may be conducted at a pressure of from 0.05 to 1500 MPa. In a preferred embodiment, the pressure is between 1.7 MPa and 30 MPa, or in another embodiment, especially under supercritical conditions, the pressure is between 15 MPa and 1500 MPa.

[00167] If branching (such a g'vis of less than 0.95) is desired in the polymer product, then, among other things, one may increase the moles of metal hydrocarbenyl chain transfer agent added to the reactor relative to the amount of polymer produced (such as grams of polymer/mols of AVTA being less than 500,000; 400,000 or less; 200,000 or less; 100,000 or less; 50,000 or less; 25,000 or less; 10,000 or less), and/or increase the temperature of the polymerization reaction (such as above 80°C), and/or increase the solids content in the polymerization reaction mass (such as 10 weight % or more, based on the weight of the components entering the reactor), and/or increase the residence time of the polymerization (such as 10 minutes or more). Likewise, if a more linear polymer is desired (such as a g'vis of more than 0.95), then, among other things, one may reduce the moles of metal hydrocarbenyl chain transfer agent added to the reactor relative to the amount of polymer produced (such as grams of polymer/mols of AVTA being 500,000 or more), and/or reduce the temperature of the polymerization reaction (such as 80°C or less), and/or reduce the solids content in the polymerization reaction mass such as 10 volume % or less), and/or reduce the residence time of the polymerization (such as 10 minutes or less). One of ordinary skill in the art will readily appreciate that 1, 2, 3 or 4 of the above conditions may be varied above or below the suggested conditions above to obtain a desired result. For example a lower catalyst/AVTA molar ratio can be used if the catalyst activity is higher.

Monomers

[00168] Monomers useful herein include olefins having from 2 to 40 carbon atoms, alternately 2 to 12 carbon atoms (preferably ethylene, propylene, butylene, pentene, hexene, heptene, octene, nonene, decene, and dodecene) and, optionally, also polyenes (such as
5 dienes). Particularly preferred monomers include ethylene, and mixtures of C₂ to C₁₀ alpha olefins, such as ethylene-propylene, ethylene-hexene, ethylene-octene, propylene-hexene, and the like.

[00169] The catalyst systems described herein are also particularly effective for the polymerization of ethylene, either alone or in combination with at least one other olefinically
10 unsaturated monomer, such as a C₃ to C₂₀ α-olefin, and particularly a C₃ to C₁₂ α-olefin. Likewise, the present complexes are also particularly effective for the polymerization of propylene, either alone or in combination with at least one other olefinically unsaturated monomer, such as ethylene or a C₄ to C₂₀ α-olefin, and particularly a C₄ to C₂₀ α-olefin. Examples of preferred α-olefins include ethylene, propylene, butene-1, pentene-1, hexene-1,
15 heptene-1, octene-1, nonene-1, decene-1, dodecene-1, 4-methylpentene-1, 3-methylpentene-1, 3, 5, 5-trimethylhexene-1, and 5-ethylnonene-1.

[00170] In some embodiments, the monomer mixture may also comprise one or more dienes at up to 10 wt%, such as from 0.00001 to 1.0 wt%, for example from 0.002 to 0.5 wt%, such as from 0.003 to 0.2 wt%, based upon the monomer mixture. Non-limiting
20 examples of useful dienes include, cyclopentadiene, norbornadiene, dicyclopentadiene, 5-ethylidene-2-norbornene, 5-vinyl-2-norbornene, 1,4-hexadiene, 1,5-hexadiene, 1,5-heptadiene, 1,6-heptadiene, 6-methyl-1,6-heptadiene, 1,7-octadiene, 7-methyl-1,7-octadiene, 1,9-decadiene, and 9-methyl-1,9-decadiene.

[00171] Where olefins are used that give rise to short chain branching, such as propylene,
25 the catalyst systems may, under appropriate conditions, generate stereoregular polymers or polymers having stereoregular sequences in the polymer chains.

[00172] In a preferred embodiment, the catalyst systems described herein are used in any polymerization process described above to produce ethylene homopolymers or copolymers, or propylene homopolymers or copolymers.

Scavengers

[00173] In some embodiments, when using the complexes described herein, particularly when they are immobilized on a support, the catalyst system will additionally comprise one or more scavenging compounds. Here, the term scavenging compound means a compound that removes polar impurities from the reaction environment. These impurities adversely

affect catalyst activity and stability. Typically, the scavenging compound will be an organometallic compound such as the Group-13 organometallic compounds of US 5,153,157; US 5,241,025; WO 1991/09882; WO 1994/03506; WO 1993/14132; and that of WO 1995/07941. Exemplary compounds include triethyl aluminum, triethyl borane, tri-iso-butyl aluminum, methyl alumoxane, iso-butyl alumoxane, tri-n-octyl aluminum, bis(diisobutylaluminum)oxide, modified methylalumoxane. (Useful modified methylalumoxane include cocatalyst type 3A (commercially available from Akzo Chemicals, Inc. under the trade name Modified Methylalumoxane type 3A) and those described in US 5,041,584). Those scavenging compounds having bulky or C₆-C₂₀ linear hydrocarbyl substituents connected to the metal or metalloid center usually minimize adverse interaction with the active catalyst. Examples include triethylaluminum, but more preferably, bulky compounds such as tri-iso-butyl aluminum, tri-iso-prenyl aluminum, and long-chain linear alkyl-substituted aluminum compounds, such as tri-n-hexyl aluminum, tri-n-octyl aluminum, or tri-n-dodecyl aluminum. When alumoxane is used as the activator, any excess over that needed for activation will scavenge impurities and additional scavenging compounds may be unnecessary. Alumoxanes also may be added in scavenging quantities with other activators, e.g., methylalumoxane, [Me₂HNPh]+[B(pfp)₄]- or B(pfp)₃ (perfluorophenyl = pfp = C₆F₅).

[00174] In embodiments, the transfer agent, such as the aluminum vinyl transfer agent, may also function as a scavenger.

[00175] In a preferred embodiment, two or more catalyst complexes as described herein are combined with a chain transfer agent, such as diethyl zinc or tri-n-octylaluminum, in the same reactor with monomer. Alternately, one or more complexes are combined with another catalyst (such as a metallocene) and a chain transfer agent, such as diethyl zinc and/or tri-n-octylaluminum, in the same reactor with monomer.

25 Polymer Products

[00176] While the molecular weight of the polymers produced herein is influenced by reactor conditions including temperature, monomer concentration and pressure, the presence of chain terminating agents and the like, the homopolymer and copolymer products produced by the present process may have an Mw of about 1,000 to about 2,000,000 g/mol, alternately of about 30,000 to about 600,000 g/mol, or alternately of about 100,000 to about 500,000 g/mol, as determined by Gel Permeation Chromatography. Preferred polymers produced herein may be homopolymers or copolymers. In a preferred embodiment, the comonomer(s) are present at up to 50 mol%, preferably from 0.01 to 40 mol%, preferably 1 to 30 mol%, preferably from 5 to 20 mol%.

[00177] The polymers produced by the process of the invention can be used in a wide variety of products and end-use applications. The polymers produced can be homo- and co-polymers of ethylene and propylene and include linear low density polyethylene, elastomers, plastomers, high-density polyethylenes, medium density polyethylenes, low density polyethylenes, polypropylene and polypropylene copolymers. Polymers, typically ethylene based copolymers, have a density of from 0.86g/cc to 0.97g/cc; density being measured in accordance with ASTM-D-1238. Propylene based polymers produced include isotactic polypropylene, atactic polypropylene and random, block or impact copolymers.

[00178] The polymers of embodiments of the invention may have an M_n (number-average molecular weight) value from 300 to 1,000,000, or between from 700 to 300,000 g/mol. For low weight molecular weight applications, such as those copolymers useful in lubricating and fuel oil compositions, an M_n of 300 to 20,000 is contemplated, or less than or equal to 10,000 g/mol. Additionally, copolymer of embodiments of the invention will comprise a molecular weight distribution (M_w/M_n) in the range of ≥ 1 , or ≥ 1.5 or ≤ 6 , or ≤ 4 or ≤ 3 , preferably from greater than 1 to 40, alternatively from 1.5 to 20, alternatively from 1.5 to 10, alternatively from 1.6 to 6, alternatively from 1.5 to 4, or alternatively from 2 to 3.

[00179] Preferred propylene polymer, preferably homopolymer, produced herein has an M_w of 20,000 up to 2,000,000 g/mol.

[00180] For higher molecular weight applications, preferred polymer, preferably homopolymer, produced herein has an M_w of 20,000 up to 2,000,000 g/mol, alternately 50,000 to 1,500,000 g/mol, alternately 100,000 to 1,300,000 g/mol, alternately 300,000 to 1,300,000 g/mol, alternately 500,000 to 1,300,000 g/mol.

[00181] For higher molecular weight applications, preferred polymer, preferably homopolymer, produced herein has an M_w of 20,000 up to 2,000,000 g/mol and a g' of a g'_{vis} of more than 0.90, alternately 0.95 or more, alternately 0.98 or more.

[00182] For lower molecular weight applications, preferred polymer, preferably homopolymer, produced herein has an M_w of less than 100,000 g/mol and a g'_{vis} of 0.95 or less, alternately 0.90 or less, alternately 0.85 or less, alternately 0.80 or less, alternately 0.75 or less, alternately 0.70 or less, alternately 0.60 or less.

End Uses

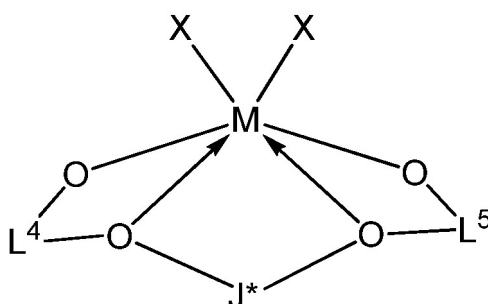
[00183] The polymers of this invention may be blended and/or coextruded with any other polymer. Non-limiting examples of other polymers include linear low density polyethylenes, elastomers, plastomers, high pressure low density polyethylene, high density polyethylenes, isotactic polypropylene, ethylene propylene copolymers and the like.

[00184] Articles made using polymers produced herein may include, for example, molded articles (such as containers and bottles, e.g., household containers, industrial chemical containers, personal care bottles, medical containers, fuel tanks, and storageware, toys, sheets, pipes, tubing) films, non-wovens, and the like. It should be appreciated that the list of applications above is merely exemplary, and is not intended to be limiting.

[00185] In particular, polymers produced by the process of the invention and blends thereof are useful in such forming operations as film, sheet, and fiber extrusion and co-extrusion as well as blow molding, injection molding, roto-molding. Films include blown or cast films formed by coextrusion or by lamination useful as shrink film, cling film, stretch film, sealing film or oriented films.

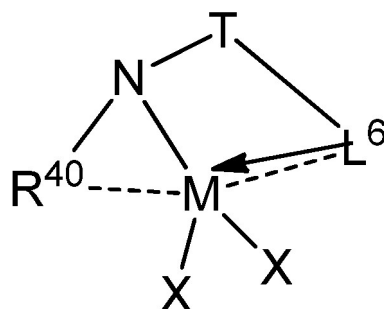
[00186] This invention further relates to:

1. A catalyst system comprising an activator; a metal hydrocarbenyl transfer agent, preferably an aluminum vinyl-transfer agent; and a transition metal complex represented by the formula:



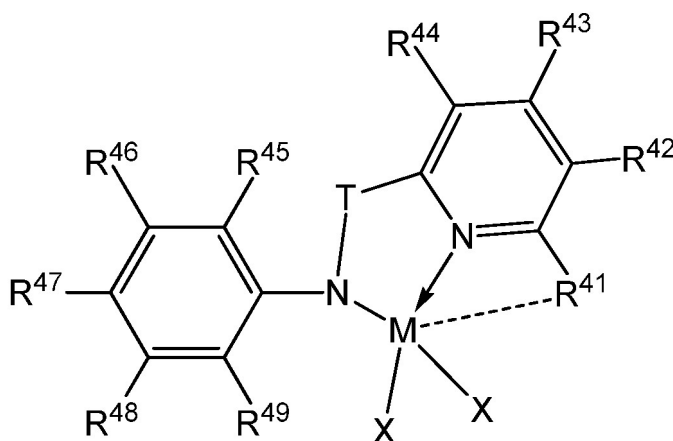
where: (1) J^* is a divalent bridging group comprising C, Si, or both; (2) M is a group 4 metal, preferably hafnium and zirconium; (3) O is oxygen; (4) each X is independently a univalent anionic ligand, or two Xs are joined and bound to the metal atom to form a metallocycle ring, or two Xs are joined to form a chelating ligand, a diene ligand, or an alkylidene ligand; and (5) L^4 and L^5 are independently, a substituted monocyclic or polycyclic aromatic groups.

2. A catalyst system comprising an activator; a metal hydrocarbenyl chain transfer agent, preferably an aluminum vinyl-transfer agent; and a transition metal complex represented by the formula:



wherein (1) M is a group 4 metal, preferably hafnium; (2) L⁶ is a C₅-C₂₀ heteroaryl group containing a Lewis base functionality, especially pyridine-2-yl or substituted pyridine-2-yl group or a divalent derivative thereof; (3) R⁴⁰ is selected from a C₁-C₃₀ alkyl, cycloalkyl, heteroalkyl, cycloheteroalkyl, aryl and substituted derivatives thereof or a divalent derivative thereof; (4) T is a divalent bridging group comprising carbon and or silicon, preferably a C₁-C₂₀ hydrocarbyl substituted methylene or silane group; (5) each X is independently a univalent anionic ligand, or two Xs are joined and bound to the metal atom to form a metallocycle ring, or two Xs are joined to form a chelating ligand, a diene ligand, or an alkylidene ligand; (6) N is nitrogen; and (7) bonds, optional bonds and dative bonds are represented by lines, dotted lines and arrows, respectively.

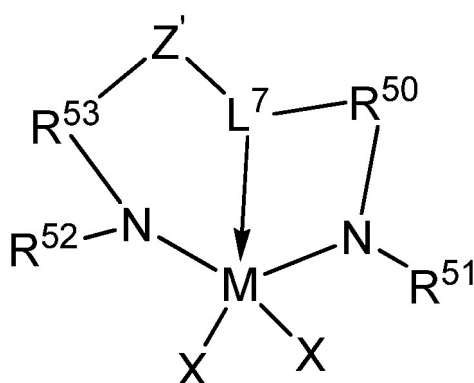
3. A catalyst system comprising an activator; a metal hydrocarbenyl chain transfer agent, preferably an aluminum vinyl-transfer agent; and a transition metal complex represented by the formula (5a):



wherein M, T, N and X are as previously defined as in formula (5); each R⁴¹, R⁴², R⁴³, and R⁴⁴ are independently hydrogen, halo, or an alkyl, cycloalkyl, heteroalkyl, heterocycloalkyl, aryl or silyl group, or one or more adjacent R⁴¹-R⁴⁴ may be joined together to form a fused ring derivative; R⁴⁵-R⁴⁹ are independently hydrogen, or C₁-C₁₀ alkyl, most preferably R⁴⁵

and R^{49} are alkyl such as isopropyl or *tert*-butyl; T is preferably $CR'^{50}R'^{51}$ where R'^{50} and R'^{51} are independently hydrogen, halogen, a C_1 - C_{20} hydrocarbyl, most preferably, one of R'^{50} and R'^{51} is hydrogen and the other is a C_6 - C_{20} aryl group, especially 2-isopropyl, phenyl or a fused polycyclic aryl group, most preferably anthracenyl; and bonds, optional bonds and
 5 dative bonds are represented by lines, dotted lines and arrows, respectively.

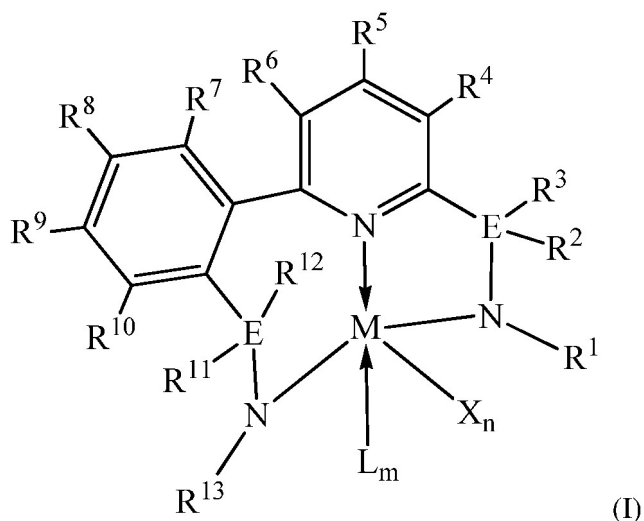
4. A catalyst system comprising an activator; a metal hydrocarbenyl chain transfer agent, preferably an aluminum vinyl-transfer agent; and a transition metal complex represented by the formula (6):



10 wherein (1) M is a group 4 metal, preferably hafnium; (2) N is nitrogen; (3) L^7 is a group that links R^{50} to Z' by a three atom bridge with the central of the three atoms being a group 15 or 16 element that preferably forms a dative bond to M, and is a C_5 - C_{20} heteroaryl group containing a Lewis base functionality, especially a divalent pyridinyl or substituted pyridinyl group; (4) Z' is a divalent linker group, $(R^{56})_pC-C(R^{57})_q$, where R^{56} and R^{57} are
 15 independently selected from the group consisting of hydrogen, hydrocarbyls, substituted hydrocarbyls, and wherein adjacent R^{56} and R^{57} groups may be joined to form an aromatic or saturated, substituted or unsubstituted hydrocarbyl ring, wherein the ring has 5, 6, 7 or 8 ring carbon atoms and where the substituents on the ring can join to form additional rings, and p is 1 or 2 and q is 1 or 2; (5) R^{50} and R^{53} are each, independently, $ER^{54}R^{55}$ with E being carbon,
 20 silicon or germanium, and each R^{54} and R^{55} being independently selected from the group consisting of hydrogen, hydrocarbyls, substituted hydrocarbyls, alkoxy, silyl, amino, aryloxy, halogen and phosphino, and R^{54} and R^{55} may be joined to form a saturated heterocyclic ring, or a saturated substituted heterocyclic ring where substitutions on the ring can join to form additional rings; (6) R^{51} and R^{52} are independently selected from the group consisting of
 25 hydrocarbyls, substituted hydrocarbyls, silylcarbyls and substituted silylcarbyl groups; and

(7) each X is independently a univalent anionic ligand, or two Xs are joined and bound to the metal atom to form a metallocycle ring, or two Xs are joined to form a chelating ligand, a diene ligand, or an alkylidene ligand.

5. A catalyst system comprising a pyridyldiamido transition metal complex represented by the formula:



wherein:

M is a Group 3, 4, 5, 6, 7, 8, 9, 10, 11, or 12 metal;

E is selected from carbon, silicon, or germanium;

10 X is an anionic group;

L is a neutral Lewis base;

R¹ and R¹³ are each independently selected from the group consisting of hydrocarbyls, substituted hydrocarbyls, and silyl groups;

15 R² through R¹² are each independently selected from the group consisting of hydrogen, hydrocarbyls, alkoxy, silyl, amino, aryloxy, substituted hydrocarbyls, halogen, and phosphino;

n is 1 or 2;

m is 0, 1, or 2; and

two X groups may be joined together to form a dianionic group;

20 two L groups may be joined together to form a bidentate Lewis base;

an X group may be joined to an L group to form a monoanionic bidentate group;

R⁷ and R⁸ may be joined to form a ring;

R¹⁰ and R¹¹ may be joined to form a ring;

an activator; and

25 a metal hydrocarbenyl chain transfer agent, preferably an aluminum vinyl-transfer agent.

6. The catalyst system of any of paragraphs 1 to 5, wherein M is Ti, Zr, or Hf.
7. The catalyst system of any of paragraphs 1 to 6, wherein E is carbon.
8. The catalyst system of any of paragraphs 1 to 7, wherein X or X* is an alkyl, aryl, hydride, alkylsilane, fluoride, chloride, bromide, iodide, triflate, carboxylate, or alkylsulfonate.
9. The catalyst system of paragraph 5, wherein L is an ether, amine, or thioether.
10. The catalyst system of any of paragraphs 1 through 9, wherein the activator comprises an alumoxane.
11. The catalyst system of any of paragraphs 1 through 10, wherein the activator comprises a non-coordinating anion.
12. The catalyst system of paragraph 11, wherein the activator comprises one or more of: trimethylammonium tetrakis(perfluoronaphthyl)borate, N,N-dimethylanilinium tetrakis(perfluoronaphthyl)borate, N,N-diethylanilinium tetrakis(perfluoronaphthyl)borate, triphenylcarbenium tetrakis(perfluoronaphthyl)borate, trimethylammonium tetrakis(perfluorobiphenyl)borate, N,N-dimethylanilinium tetrakis(perfluorobiphenyl)borate, triphenylcarbenium tetrakis(perfluorobiphenyl)borate, N,N-dimethylanilinium tetrakis(3,5-bis(trifluoromethyl)phenyl)borate, triphenylcarbenium tetrakis(perfluoronaphthyl)borate, triphenylcarbenium tetrakis(perfluorobiphenyl)borate, triphenylcarbenium tetrakis(3,5-bis(trifluoromethyl)phenyl)borate, triphenylcarbenium tetrakis(perfluorophenyl)borate, $[\text{Ph}_3\text{C}^+][\text{B}(\text{C}_6\text{F}_5)_4^-]$, $[\text{Me}_3\text{NH}^+][\text{B}(\text{C}_6\text{F}_5)_4^-]$, 1-(4-(tris(pentafluorophenyl)borate)-2,3,5,6-tetrafluorophenyl)pyrrolidinium, tetrakis(pentafluorophenyl)borate, 4-(tris(pentafluorophenyl)borate)-2,3,5,6-tetrafluoropyridine, triphenylcarbenium tetraphenylborate, and triphenylcarbenium tetrakis-(2,3,4,6-tetrafluorophenyl)borate.
13. The catalyst system of any of paragraphs 1 through 11, wherein the metal hydrocarbenyl chain transfer agent is represented by the formula: $\text{E}[\text{Al}(\text{R}')_{2-y}(\text{R}'')_y]_2$ and the aluminum vinyl-transfer agent is represented by the formula: $\text{Al}(\text{R}')_{3-v}(\text{R}'')_v$; wherein each R', independently, is a C₁ to C₃₀ hydrocarbyl group; each R'', independently, is a C₄ to C₂₀ hydrocarbenyl group, having an end-vinyl group; E is a group 16 element; v is from 0.01 to 3 (preferably 0.1 to 3); and y is from 0.01 to 2.
14. The catalyst system of paragraph 13, wherein R'' is butenyl, pentenyl, heptenyl, or octenyl and or wherein R' is methyl, ethyl, propyl, isobutyl, or butyl.
15. The catalyst system of any of paragraphs 1 to 14, wherein the aluminum vinyl transfer agent comprises one or more of tri(but-3-en-1-yl)aluminum, tri(pent-4-en-1-yl)aluminum,

- tri(oct-7-en-1-yl)aluminum, tri(non-8-en-1-yl)aluminum, tri(dec-9-en-1-yl)aluminum, dimethyl(oct-7-en-1-yl)aluminum, diethyl(oct-7-en-1-yl)aluminum, dibutyl(oct-7-en-1-yl)aluminum, diisobutyl(oct-7-en-1-yl)aluminum, diisobutyl(non-8-en-1-yl)aluminum, diisobutyl(dec-8-en-1-yl)aluminum, and diisobutyl(dodec-10-en-1-yl)aluminum.
- 5 16. The catalyst system of any of paragraphs 13 through 15, wherein $v = 2$.
 17. A polymerization process comprising contacting one or more alkene monomers with the catalyst system of any of paragraphs 1 to 16.
 18. The process of paragraph 17, wherein the activator comprises an alumoxane.
 19. The process of paragraph 17 or 18, wherein the activator comprises a non-
10 coordinating anion.
 20. The process of paragraph 17, 18, or 19 wherein the monomer comprises ethylene or propylene.
 21. The process of any of paragraphs 17 to 20, wherein the monomers comprise at least two of ethylene, propylene, and an alkyl diene.
 - 15 22. The process of any of paragraphs 17 through 21, wherein the pyridyldiamido transition metal complex is supported.
 23. The process of paragraph 22, wherein the support is silica.
 24. The process of any of paragraphs 17 through 23, wherein the metal hydrocarbenyl chain transfer agent is represented by the formula: $E[Al(R')_{2-y}(R'')_y]_2$ and the aluminum
20 vinyl-transfer agent is represented by the formula: $Al(R')_{3-v}(R'')_v$; wherein E is a group 16 element; y is 2; R'' is butenyl, pentenyl, heptenyl, or octenyl, R' is methyl, ethyl, propyl, isobutyl, or butyl; and $v = 2$.
 25. The process of any of paragraphs 17 through 24, further comprising the step of adding a metallocene catalyst in a second step.
 - 25 26. The process of paragraph 25, wherein the second step of the process is conducted in a second reactor.
 27. The process of either paragraphs 25 or 26, wherein the steps are performed in a continuous stirred tank reactor.
 28. The process of any of paragraphs 17 to 24 wherein the transition metal complex is not
30 a metallocene.
 29. The polymerization process of any of paragraphs 17 to 28 wherein 1) the residence time is less than 10 minutes and/or the catalyst complex-to-aluminum vinyl transfer agent molar ratio is 1:50 or more, and 2) a polyolefin having allyl chain ends and a g'vis of 0.95 or more is obtained.

30. The polymerization process of any of paragraphs 17 to 28 wherein 1) the residence time is 10 minutes or more and/or the catalyst complex-to-aluminum vinyl transfer agent molar ratio is 1:50 or less, and 2) a polyolefin having a g'vis of 0.95 or less is obtained.

31. The polymerization process of any of paragraphs 17 to 28 wherein a polyolefin
5 having allyl chain ends, and optionally a g'vis of 0.95 or more, is produced and wherein the moles of metal hydrocarbenyl chain transfer agent added to the reactor relative to the amount of polymer produced is 500,000 or more.

32. The polymerization process of any of paragraphs 17 to 28 wherein a polyolefin
10 having having a g'vis of 0.95 or less is produced and wherein the moles of metal hydrocarbenyl chain transfer agent added to the reactor relative to the amount of polymer produced is less than 500,000.

EXPERIMENTAL

[00187] All manipulations were performed under an inert atmosphere using glove box techniques unless otherwise stated. Benzene-*d*₆ (Cambridge Isotopes) (Sigma Aldrich) was
15 degassed and dried over 3Å molecular sieves prior to use.

[00188] Diisobutylaluminum hydride (DIBAL-H) was purchased from Akzo Nobel Surface Chemistry LLC and used as received. Triisobutyl aluminum (TIBAL) was purchased from Akzo Nobel and was used as received. 1,7-octadiene were purchased from Sigma Aldrich and purified by the following procedure prior to use. The diene was purged under
20 nitrogen for 30 minutes and then this was stored over 3Å molecular sieves for overnight. Further this was stirred with NaK (sodium-potassium alloy) for overnight and then filtered through basic alumina column prior to use.

Preparation of 7-octenyldiisobutylaluminum (AVTA1)

[00189] Diisobutylaluminum hydride (8.54 g, 60 mmol) was added to 1,7-octadiene
25 (39.7 g, 360 mmol) in a round bottomed flask. The mixture was heated to 50°C for 16 hours. The volatiles were evaporated under a stream of nitrogen. Further removal of volatiles was done under reduced pressure. The product was isolated as a colorless oil. Yield 13.2 g. Evaluation of ¹H-NMR spectroscopic data indicated a molar ratio of 1.8:1.2:1.1 for Al-ⁱBu, Al-CH₂(CH₂)_n-, and vinyl groups, respectively. This indicates an approximate average composition of Al(ⁱBu)_{1.8}(oct-7-en-1-yl)_{1.2}. ¹H NMR (D₈-toluene, 400 MHz): δ 5.78 (m, 1.05H, vinyl), 4.99 (m, 2.1H, vinyl), 1.9-2.03 (m, 3.9H), 1.55 (m, 2.4H), 1.34 (m, 6.9H), 1.04
30 (d, 10.6H), 0.49 (m, 2.37H, Al-CH₂(octenyl group)), 0.25 (3.63H, Al-CH₂(isobutyl group)).

Preparation of tri(oct-7-en-1-yl)aluminum (AVTA2)

[00190] The preparation was performed using the procedure described by Shiono and co-

workers in JP2004-83772-A. ^1H -NMR spectroscopic analysis indicated that the organoaluminum product was not pure tri(oct-7-en-1-yl)aluminum, but instead a mixture with an average composition of $\text{Al}(\text{iBu})_{0.45}(\text{oct-7-en-1-yl})_{2.55}$. The ratio of octenyl to isobutyl groups is expected to be a function of reaction temperature, pressure, and time. The product was used as the isolated mixture.

Preparation of $\text{Al}(\text{iBu})_{0.73}(\text{oct-7-en-1-yl})_{2.27}$ (AVTA3)

[00191] Triisobutylaluminum (6.02 g, 30.3 mmol) and 1,7-octadiene (66.9 g, 607 mmol) were combined in a thick walled glass bottle. This was sealed and the stirred mixture was heated in an oil bath to 150°C. After 15 minutes the flask was removed from the oil bath and allowed to cool to ambient temperature. The volatiles were removed by vacuum distillation at ambient temperature. ^1H NMR spectroscopy of the crude product indicated an average composition of $\text{Al}(\text{iBu})_{0.80}(\text{octen-7-en-1-yl})_{2.20}$ with 0.095 molar equivalents of byproduct $\text{CH}_2=\text{CH}(\text{CH}_2)_6\text{C}(\text{=CH}_2)(\text{CH}_2)_4\text{CH}=\text{CH}_2$. This mixture was subjected to further vacuum distillation at 70-90°C for an hour using a Kugelrohr distillation apparatus in an attempt to remove the byproduct triene. ^1H NMR spectroscopy of the remaining product indicated an average composition of $\text{Al}(\text{iBu})_{0.73}(\text{octen-7-en-1-yl})_{2.27}$ with 0.067 molar equivalents of byproduct $\text{CH}_2=\text{CH}(\text{CH}_2)_6\text{C}(\text{=CH}_2)(\text{CH}_2)_4\text{CH}=\text{CH}_2$ remaining. Isolated yield ~ 6 g. ^1H NMR (D_8 -toluene, 400 MHz): δ 5.78 (m, 2.30H, vinyls), 4.99 (m, 4.78H, vinyls), 4.82 (m, 0.13H, vinylidene of byproduct triene), 1.9-2.1 (m, 5.88H), 1.2-1.8 (m, 22.81H), 1.10 (d, 2.37H), 0.45-0.60 (m, 4.50H, Al- CH_2 (octenyl group)), 0.30 (d, 1.44H, Al- CH_2 (isobutyl group)).

Preparation of Diisobutyl(oct-7-en-1-yl) aluminum, $\text{iBu}_2\text{Al}(\text{Oct=})$ (AVTA4).

[00192] A neat 1,7-octadiene (16.53 g, 150 mmol) was added drop wise to DIBAL-H (3.56 g, 25 mmol) at room temperature over a period of 5 minutes. The reaction mixture was either stirred at 45°C for overnight or refluxed at 110°C for 1 hour to form the title compound, as indicated below. The excess 1,7-octadiene from the reaction mixture was removed under the flow of dry nitrogen at room temperature. The residual 1,7-octadiene was then removed in vacuo for 30 min to obtain a colorless viscous oil of AVTA4 (5.713 g, 91%). The product formation was confirmed by ^1H NMR spectroscopy and based on the relative integration the molecular formula was assigned as $(\text{C}_4\text{H}_9)_{2.1}\text{Al}(\text{C}_8\text{H}_{15})_{0.9}$. ^1H NMR (400 MHz, benzene- d_6): δ = 5.78 (m, 1H, =CH), 5.01 (m, 2H, =CH₂), 1.95 (m, 4H, -CH₂), 1.54 (m, 2H, iBu-CH), 1.34 (m, 6H, -CH₂), 1.04 (d, 12H, iBu-CH_3), 0.49 (t, 2H, Al-CH₂), 0.27 (d, 4H, iBu-CH_2) ppm.

Preparation of Isobutyldi(oct-7-en-1-yl)aluminum, $\text{iBuAl}(\text{Oct=})_2$ (AVTA5).

[00193] A neat 1,7-octadiene (9.00 g, 81.8 mmol) was added drop wise to DIBAL-H (0.81

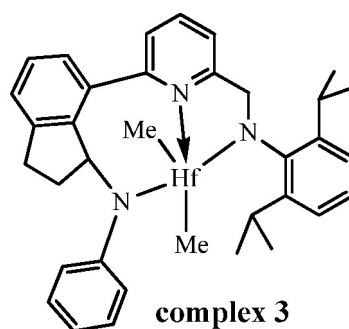
g, 7.2 mmol) at room temperature over 5 minutes. The resulting mixture was stirred under reflux at 110°C for 60 minutes and then continuously stirred at 70°C overnight. The excess 1,7-octadiene from the reaction mixture was removed under the flow of dry nitrogen at room temperature. The residual 1,7-octadiene was then removed in vacuo for 30 minutes to obtain a colorless viscous oil of **AVTA5** (1.689 g, 79 %). The product formation was confirmed by ¹H NMR and based on the relative integration the molecular formula of was assigned as (C₄H₉)_{1.15}Al(C₈H₁₅)_{1.85}. ¹H NMR (400 MHz, benzene-*d*₆): δ = 5.81 (m, 2H, =CH), 5.05 (m, 4H, =CH₂), 2.03 (m, 8H, -CH₂), 1.59 (m, 1H, ⁱBu-CH), 1.38 (m, 12H, -CH₂), 1.09 (d, 6H, ⁱBu-CH₃), 0.51 (t, 4H, Al-CH₂), 0.31 (d, 2H, ⁱBu-CH₂) ppm.

10 **Preparation of Isobutyldi(oct-7-en-1-yl)aluminum, ⁱBuAl(Oct=)₂ (AVTA6).**

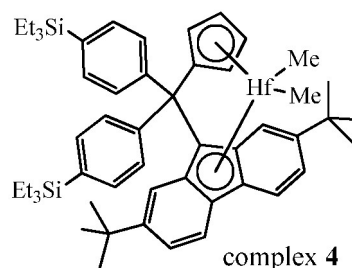
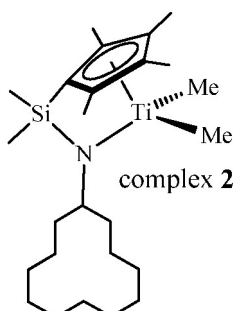
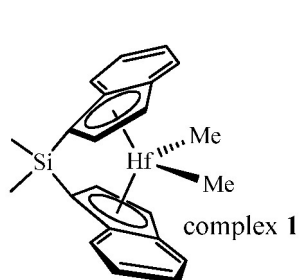
[00194] A neat 1,7-octadiene (22.91 g, 207.92 mmol) was added drop wise to DIBAL-H (2.612 g, 18.4 mmol) at room temperature over 5 minutes. The resulting mixture was stirred under reflux at 110°C for 60 minutes and then continuously stirred at 70°C overnight. The excess 1,7-octadiene from the reaction mixture was removed under the flow of dry nitrogen at room temperature. The residual 1,7-octadiene was then removed in vacuo for 30 minutes to obtain a colorless viscous oil of **AVTA6** (5.530 g, 97 %). The product formation was confirmed by ¹H NMR and based on the relative integration the molecular formula of was assigned as (C₄H₉)_{0.95}Al(C₈H₁₅)_{2.05}. ¹H NMR (400 MHz, benzene-*d*₆): δ = 5.81 (m, 2H, =CH), 5.05 (m, 4H, =CH₂), 2.03 (m, 8H, -CH₂), 1.59 (m, 1H, ⁱBu-CH), 1.38 (m, 12H, -CH₂), 1.09 (d, 6H, ⁱBu-CH₃), 0.51 (t, 4H, Al-CH₂), 0.31 (d, 2H, ⁱBu-CH₂) ppm.

[00195] General considerations in the synthesis of AVTA's. Reaction of DIBAL-H with 1,7-octadiene (as shown in Figure 2) under mild conditions produces the mono-alkenyl aluminum complex **1** (like **AVTA4**) without the formation of triene side product, **3**. More robust reaction conditions were required to form the di-alkenyl aluminum complex **2** (like **AVTA5**), and care was taken to exclude the formation of the triene, **3**. Intermediate temperatures such at 70°C for overnight or 110°C for 1 hr produced the di-alkenyl aluminum complex **2**, without formation of the triene. Higher temperatures or prolonged heating at intermediate temperatures resulted in the formation of the triene.

[00196] Complex 3 was prepared as described in US 9,290,519.



[00197] Complexes 1, 2, and 4 are used for comparative purposes.



Polymerization Examples

[00198] Solutions of the pre-catalysts (complexes 1-4) were made using toluene (ExxonMobil Chemical—anhydrous, stored under N₂) (98%). Pre-catalyst solutions were typically 0.5 mmol/L.

[00199] Solvents, polymerization grade toluene and/or isohexanes were supplied by ExxonMobil Chemical Co. and are purified by passing through a series of columns: two 500 cc Oxyclear cylinders in series from Labclear (Oakland, Calif.), followed by two 500 cc columns in series packed with dried 3 Å mole sieves (8-12 mesh; Aldrich Chemical Company), and two 500 cc columns in series packed with dried 5 Å mole sieves (8-12 mesh; Aldrich Chemical Company).

[00200] 1-octene (C₈; 98%, Aldrich Chemical Company) was dried by stirring over NaK overnight followed by filtration through basic alumina (Aldrich Chemical Company, Brockman Basic 1).

[00201] Polymerization grade ethylene (C₂) was used and further purified by passing it through a series of columns: 500 cc Oxyclear cylinder from Labclear (Oakland, Calif.) followed by a 500 cc column packed with dried 3 Å mole sieves (8-12 mesh; Aldrich Chemical Company), and a 500 cc column packed with dried 5 Å mole sieves (8-12 mesh; Aldrich Chemical Company).

[00202] Polymerization grade propylene (C₃) was used and further purified by passing it through a series of columns: 2250 cc Oxyclear cylinder from Labclear followed by a 2250

cc column packed with 3 Å mole sieves (8-12 mesh; Aldrich Chemical Company), then two 500 cc columns in series packed with 5 Å mole sieves (8-12 mesh; Aldrich Chemical Company), then a 500 cc column packed with Selexsorb CD (BASF), and finally a 500 cc column packed with Selexsorb COS (BASF).

5 [00203] For polymerization runs collected in Tables 5 and 8, bis(diisobutylaluminum)oxide (DiBAI-O, AkzoNobel 0.92M in hexane) was used as a scavenger prior to introduction of the AVTA, activator and pre-catalyst into the reactor. Scavenger was typically used as a 5 mmol/L solution in toluene.

[00204] Tri-n-octyl aluminum (TnOAl, neat, AkzoNobel) was used as a comparative to the
10 AVTA's in Tables 5 and 8 and was typically used as a 5 or 10 mmol/L solution in toluene. The AVTA's were typically also used as a 5 or 10 mmol/L solution in toluene. AVTA# in Tables 5 and 8, corresponds to the AVTA's prepared in the examples above.

Reactor Description and Preparation:

[00205] Polymerizations were conducted in an inert atmosphere (N₂) drybox using
15 autoclaves equipped with an external heater for temperature control, glass inserts (internal volume of reactor=23.5 mL for C2 and C2/C8; 22.5 mL C2/C3 runs), septum inlets, regulated supply of nitrogen, ethylene and propylene, and equipped with disposable PEEK mechanical stirrers (800 RPM). The autoclaves were prepared by purging with dry nitrogen at 110°C or 115°C for 5 hours and then at 25°C for 5 hours.

20 **Ethylene Polymerization (PE) or Ethylene/1-octene Copolymerization (EO):**

[00206] The reactor was prepared as described above, and then purged with ethylene. Isohexanes, 1-octene (50 µL when used in experiments in Table 5 and 100 uL in experiments in Table 1), and additional DiBAI-O scavenger (0.05 umol, used in experiments in Table 5 only) were added via syringe at room temperature and atmospheric pressure. The reactor was
25 then brought to process temperature (values indicated below) and charged with ethylene to process pressure (100 psig = 790.8 kPa for experiments in Table 5 and 50 psig = 446 kPa for experiments in Table 1) while stirring at 800 RPM. The AVTA solution (or control TnOAl solution for experiments in Tables 5) was injected via syringe to the reactor at process conditions. Next, the activator solution, followed by the pre-catalyst solution, was injected
30 via syringe to the reactor at process conditions. Ethylene was allowed to enter (through the use of computer controlled solenoid valves) the autoclaves during polymerization to maintain reactor gauge pressure (+/-2 psig). Reactor temperature was monitored and typically maintained within +/-1°C. Polymerizations were halted by addition of approximately 50 psi O₂/Ar (5 mole % O₂) gas mixture to the autoclave for approximately 30 seconds. The

polymerizations were quenched after a predetermined cumulative amount of ethylene had been added (maximum quench value in psid) or for a maximum of 30 minutes polymerization time. Afterwards, the reactors were cooled and vented. Polymers were isolated after the solvent was removed in-vacuo. Yields reported include total weight of polymer and residual catalyst. Catalyst activity is reported as grams of polymer per mmol transition metal compound per hour of reaction time (g/mmol•hr). Ethylene homopolymerization runs and ethylene/1-octene copolymerization runs are summarized in Tables 1 and 5. Additional reaction details can be found above Tables 1 and 5.

Ethylene-Propylene Copolymerization (EP):

10 [00207] The reactor was prepared as described above, then heated to 40°C. and then purged with ethylene gas at atmospheric pressure. Ethylene was then added at 125 psid (790.8 kPa) to the reactor. Isohexanes and scavenger (DiBAI-O, 0.5 μ mol) were added via syringe. The stirrers were then started and maintained at 800 RPM. Liquid propylene (0.5 ml) was then injected into the reactor. The reactor was then brought to process temperature 15 (100°C). The AVTA or control TnOAl solutions were next injected into the reactor at process temperature. Next, the activator solution, followed by the pre-catalyst solution, was injected via syringe to the reactor at process conditions. Reactor temperature was monitored and typically maintained within $\pm 1^\circ\text{C}$. Polymerizations were halted by addition of approximately 50 psi O₂/Ar (5 mole % O₂) gas mixture to the autoclaves for approximately 20 30 seconds. The polymerizations were quenched based on a predetermined pressure loss (quench value) of approximately 4 psid pressure loss or for a maximum of 30 minutes polymerization time. The reactors were cooled and vented. The polymer was isolated after the solvent was removed in-vacuo. The quench times are reported in Table 8 for each run. Yields reported include total weight of polymer and residual catalyst. Catalyst activity is 25 reported as grams of polymer per mmol transition metal compound per hour of reaction time (g/mmol•hr). Ethylene/propylene copolymerization examples are collected in Table 8 and additional information is located above the table.

Polymer Characterization

30 [00208] For analytical testing, polymer sample solutions were prepared by dissolving polymer in 1,2,4-trichlorobenzene (TCB, 99+% purity from Sigma-Aldrich) containing 2,6-di-tert-butyl-4-methylphenol (BHT, 99% from Aldrich) at 165°C in a shaker oven for approximately 3 hours. The typical concentration of polymer in solution was between 0.1 to 0.9 mg/mL with a BHT concentration of 1.25 mg BHT/mL of TCB. Samples were cooled to 135°C for testing.

[00209] High temperature size exclusion chromatography was performed using an automated "Rapid GPC" system as described in U.S. Patent Nos. 6,491,816; 6,491,823; 6,475,391; 6,461,515; 6,436,292; 6,406,632; 6,175,409; 6,454,947; 6,260,407; and 6,294,388; each of which is incorporated herein by reference. Molecular weights (weight average molecular weight (Mw) and number average molecular weight (Mn)) and molecular weight distribution (MWD = Mw/Mn), which is also sometimes referred to as the polydispersity (PDI) of the polymer, were measured by Gel Permeation Chromatography using a Symyx Technology GPC equipped with dual wavelength infrared detector and calibrated using polystyrene standards (Polymer Laboratories: Polystyrene Calibration Kit S-M-10: Mp (peak Mw) between 580 and 3,039,000). Samples (250 μ L of a polymer solution in TCB were injected into the system) were run at an eluent flow rate of 2.0 mL/minute (135°C sample temperatures, 165°C oven/columns) using three Polymer Laboratories: PLgel 10 μ m Mixed-B 300 x 7.5mm columns in series. No column spreading corrections were employed. Numerical analyses were performed using Epoch® software available from Symyx Technologies or Automation Studio software available from Freeslate. The molecular weights obtained are relative to linear polystyrene standards. Molecular weight data is reported in Tables 1, 5 and 8 under the headings Mn, Mw and PDI (or Mw/Mn) as defined above.

[00210] Differential Scanning Calorimetry (DSC) measurements were performed on a TA-Q100 instrument to determine the melting point of the polymers. Samples were pre-annealed at 220°C for 15 minutes and then allowed to cool to room temperature overnight. The samples were then heated to 220°C at a rate of 100°C/minute and then cooled at a rate of 50°C/minute. Melting points were collected during the heating period. The results are reported in the Tables 5 and 8 under the heading, Tm (°C).

[00211] Samples for infrared analysis were prepared by depositing the stabilized polymer solution onto a silanized wafer (Part number S10860, Symyx). By this method, approximately between 0.12 and 0.24 mg of polymer is deposited on the wafer cell. The samples were subsequently analyzed on a Bruker Equinox 55 FTIR spectrometer equipped with Pikes' MappIR specular reflectance sample accessory. Spectra, covering a spectral range of 5000 cm^{-1} to 500 cm^{-1} , were collected at a 2 cm^{-1} resolution with 32 scans.

[00212] For ethylene-1-octene copolymers, the wt% octene in the copolymer was determined via measurement of the methyl deformation band at $\sim 1375 \text{ cm}^{-1}$. The peak height of this band was normalized by the combination and overtone band at $\sim 4321 \text{ cm}^{-1}$, which corrects for path length differences. The normalized peak height was correlated to individual

calibration curves from ^1H NMR data to predict the wt% octene content within a concentration range of ~2 to 35 wt% for octene. Typically, R^2 correlations of 0.98 or greater are achieved. These numbers are reported in Table 5 under the heading C8 wt%).

[00213] Unless otherwise indicated, proton NMR spectra are collected using a 400 MHz Varian pulsed fourier transform NMR spectrometer equipped with a variable temperature proton detection probe operating at 120°C. The polymer sample is dissolved in 1,1,2,2-tetrachloroethane-d2 (TCE-d2) and transferred into a 5 mm glass NMR tube. Typical acquisition parameters are sweep width = 10 KHz, pulse width = 30 degrees, acquisition time = 2 s, acquisition delay = 5 s and number of scans = 120. Chemical shifts are determined relative to the TCE-d2 signal which is set to 5.98 ppm.

[00214] The chain end unsaturations are determined as follows. The vinyl resonances of interest are between from about 5.0 to 5.1 ppm (VRA), the vinylidene resonances between from about 4.65 to 4.85 ppm (VDRA), the vinylene resonances from about 5.31 to 5.55 ppm (VYRA), the tri-substituted unsaturated species from about 5.11 to 5.30 ppm (TSRA) and the aliphatic region of interest between from about 0 to 2.1 ppm (IA). The number of vinyl groups/1000 Carbons is determined from the formula: $(\text{VRA} * 500) / ((\text{IA} + \text{VRA} + \text{VYRA} + \text{VDRA})/2) + \text{TSRA}$. Likewise, the number of vinylidene groups / 1000 Carbons is determined from the formula: $(\text{VDRA} * 500) / ((\text{IA} + \text{VRA} + \text{VYRA} + \text{VDRA})/2) + \text{TSRA}$, the number of vinylene groups / 1000 Carbons from the formula $(\text{VYRA} * 500) / ((\text{IA} + \text{VRA} + \text{VYRA} + \text{VDRA})/2) + \text{TSRA}$ and the number of tri-substituted groups from the formula $(\text{TSRA} * 1000) / ((\text{IA} + \text{VRA} + \text{VYRA} + \text{VDRA})/2) + \text{TSRA}$. VRA, VDRA, VYRA, TSRA and IA are the integrated normalized signal intensities in the chemical shift regions defined above. Proton NMR data is reported in Tables 1, 6 and 9. For many examples, end-group unsaturation was below the detection limit, or noise level, and could not be determined.

[00215] GPC3D data is reported in Tables 4, 7 and 10 and is described below after the small scale polymerization experiments.

[00216] For the polymerization results collected in Tables 5-10 below, “Ex#” stands for example number. Under the Ex# column heading, the following abbreviations are defined: PE = polyethylene, EO = ethylene-1-octene copolymer, and EP=ethylene-propylene copolymer, T(°C) is the polymerization temperature which was typically maintained within +/- 1°C. “Yield” is polymer yield, and is not corrected for catalyst residue. “Quench time (s)” is the actual duration of the polymerization run in seconds. “Quench Value (psid)” for ethylene based polymerization runs (no propylene) is the set maximum amount of ethylene

uptake (conversion) for the experiment. If a polymerization quench time is less than the maximum time set, then the polymerization ran until the set maximum value of ethylene uptake was reached. For ethylene-propylene copolymerization runs, quench value indicates the maximum set pressure loss (conversion) of ethylene and propylene combined during the polymerization. Activity is reported at grams polymer per mmol of catalyst per hour.

High-Throughput Ethylene-Octene Copolymerization.

[00217] The copolymerization of ethylene and 1-octene using four different catalyst systems in the presence of AVTA1 was performed using the parallel polymerization reactor. Specifics of run conditions are collected in Table 1 and in the description below. Characterization data for the polymers produced are shown in Tables 1 and 3. Four different complexes were evaluated, with complex **3** being inventive. The catalyst formed by the activation of complex **3** was found to readily undergo chain transfer with AVTA1 under the conditions explored. This is best observed in the plot of 100,000/Mn vs nmols of AVTA1 (FIG. 1). In this plot it is seen that the runs performed using complex **3** (runs 9-12) demonstrate an inverse relationship between number average molecular weight and the amount of AVTA1 added. The runs performed using complexes **1**, **2**, and **4** do not demonstrate a dependence of molecular weight on the amount of AVTA1 used (runs 1-8 and 13-15), so are less effective at transferring the octenyl group (or other alkyl groups) from the AVTA1 to a growing polymer chain. ¹H-NMR spectroscopic analysis of the polymer from runs 9, 10, 11 and 12 indicate the presence of 0.02, 0.06, 0.20 and 0.41 vinyls/1000 carbon atoms, respectively (Table 3). This provides additional evidence that the octenyl group is being transferred from the added AVTA1 to the polymer chains (via the catalyst formed from complex **3**).

Table 1. High-throughput copolymerization of ethylene and 1-octene.

Conditions: 80°C, 5 mL volume, 0.1 mL 1-octene, 50 psi ethylene, isohexane solvent, 20 nmol complex, 20 nmol [PhNMe₂H]B(C₆F₅)₄ activator. Reaction was quenched after 18 psi of ethylene uptake.

run	complex	AVTA1 (nmol)	time (s)	yield (mg)	Rapid GPC Mw (g/mol)	Rapid GPC Mn (g/mol)	Mw/Mn
1*	1	500	36	80	231553	91702	2.5
2*	1	1000	45	74	285113	139647	2.0
3*	1	2000	64	69	328360	158189	2.1
4*	1	4000	78	57	331039	159898	2.1
5*	2	500	63	92	390539	146644	2.7
6*	2	1000	62	91	407889	181015	2.3
7*	2	2000	58	82	442646	202590	2.2

8*	2	4000	96	71	443012	197328	2.2
9	3	500	33	75	260282	157236	1.7
10	3	1000	27	69	126433	81244	1.6
11	3	2000	28	58	66661	40379	1.7
12	3	4000	30	55	32547	19279	1.7
13*	4	500	118	88	1228247	623589	2.0
14*	4	1000	133	89	1276615	514084	2.5
15*	4	2000	125	85	1534274	734052	2.1

* denotes data for comparative purposes

Small scale batch polymerizations.

[00218] Olefin polymerizations were performed in a thick walled glass reactor. Details of polymerization conditions and the products formed are described in Tables 2, 3, and 4. All solvents, reactants, and gases were purified by passing through multiple columns containing 3 angstrom molecular sieves and oxygen scavenger. Typically, the solvent was added to the reactor and the reactor was allowed to equilibrate to the targeted temperature. Then, optional comonomer was added followed by optional scavenger and AVTA2. Then N,N-dimethylanilinium tetrakis(pentafluorophenyl)borate activator (1.1 molar equivalents relative to transition metal complex) was added as a toluene solution followed by a toluene solution of the transition metal complex. Immediately, the reactor was pressurized with ethylene to the desired pressure and the mixture was rapidly stirred. At the indicated time the pressure was vented and the polymer solution was poured into a large excess of acidic methanol (typically a 80:1 mixture of methanol and concentrated hydrochloric acid, by volumes) to precipitate the polymer product. The resulting solid was collected by filtration and washed thoroughly with water and methanol before being dried in a vacuum oven at 60°C.

Table 2. Polymerization Conditions and Results.

Run	complex	complex (nmol)	1-hexene (mL)	ethylene (psi)	toluene (mL)	AVTA2 (mmol)	T _{polym} (°C)	time (min)	yield (g)
196	3	1500	0	50	75	0.18	90	10	2.5
197	3	1500	0	35	100	0.12	90	30	2.6
198	3	1500	0	50	100	0.06	90	7	0.2

Table 3. Characterization of polymers produced.

-59-

run	Me per 1000 C (HNMR)	vinyls per 1000 C (HNMR)	vinylenes per 1000 C (HNMR)	vinylidenes per 1000 C (HNMR)	trisubs per 1000 C (HNMR)
9	39.8	0.02	0.00	0.00	0.00
10	32.1	0.06	0.00	0.01	0.01
11	27.1	0.20	0.01	0.00	0.02
12	28.5	0.41	0.00	0.00	0.00
196	2.7	0.50	none det.	0.08	none det.
197	1.6	0.41	none det.	none det.	none det.
198	1.8	0.78	none det.	0.10	none det.

Table 4. Characterization of polymers produced. GPC data are relative to linear PE.

Run	Mn_LS (kg/mol)	Mw_LS (kg/mol)	Mz_LS (kg/mol)	g'vis	Mn_DRI (kg/mol)	Mw_DRI (kg/mol)	Mz_DRI (kg/mol)	Mw/Mn (DRI)
196	10.0	28.6	390.5	0.87	11.2	26.4	68.8	2.4
197	18.6	46.8	381.9	0.98	17.7	41.8	115.2	2.4
198	17.0	37.5	205.2	0.99	14.0	37.2	95.9	2.7

Table 5. High throughput ethylene (PE) and ethylene-octene (EO) runs comparing tri-n-octyl aluminum (TNOAl)

Conditions: For all examples in Table 5, 0.025 μmol of catalyst 3 was used and activated with 1.1 molar tetrakis(pentafluorophenyl)borane. For all EO runs, 50 μL of 1-octene was added, and the total reagent and solvent volume utilized 0.05 μmol of bis(diisobutyl aluminum)oxide as scavenger. Ethylene (100 psid) was added before catalyst injection pressure. The run was quenched after 20 psid of ethylene uptake.

Type of Run	AVTA or TnOAl	AVTA or TnOAl (μmol)	AVTA/Cat or TnOAl/Cat (molar ratio)	Iso-hexane (μL)	Toluene (μL)	T ($^{\circ}\text{C}$)	yield (g)	quench time (s)	Activity (g P/ μmol cat.hr)	Rapid Mn	Rapid Mw
PE-1	TnOAl	0.05	2	4895	105	70	0.117	29	580,966	611,234	1,534,783
PE-2	TnOAl	0.05	2	4895	105	70	0.115	27	613,333	821,401	1,434,952
PE-3	AVTA3	0.25	10	4845	155	70	0.114	20	820,800	723,037	2,178,739
PE-4	AVTA3	0.25	10	4845	155	70	0.108	22	706,909	690,424	1,961,751
PE-5	AVTA3	0.50	20	4795	205	70	0.103	74	200,432	365,061	995,570
PE-6	AVTA3	0.50	20	4795	205	70	0.072	49	211,592	250,659	1,369,867
PE-7	AVTA3	1.00	40	4695	305	70	0.103	30	494,400	150,570	417,288
PE-8	AVTA3	1.00	40	4695	305	70	0.088	52	243,692	123,782	738,015
PE-9	TnOAl	0.05	2	4895	105	85	0.094	87	155,586	612,982	965,485
PE-10	AVTA3	0.25	10	4845	155	85	0.086	38	325,895	639,484	1,228,787
PE-11	AVTA3	0.25	10	4845	155	85	0.090	31	418,065	619,083	1,056,722
PE-12	AVTA3	0.50	20	4795	205	85	0.088	56	226,286	299,624	598,659
PE-13	AVTA3	0.50	20	4795	205	85	0.096	33	418,909	355,564	720,242
PE-14	AVTA3	1.00	40	4695	305	85	0.080	26	443,077	106,172	159,543
PE-15	AVTA3	1.00	40	4695	305	85	0.087	27	464,000	130,138	197,198
PE-16	TnOAl	0.05	2	4895	105	100	0.079	293	38,826	563,425	760,348
PE-17	TnOAl	0.05	2	4895	105	100	0.076	78	140,308	519,070	687,477
PE-18	AVTA3	0.25	10	4845	155	100	0.058	70	119,314	537,961	1,052,060
PE-19	AVTA3	0.25	10	4845	155	100	0.074	56	190,286	435,578	741,860
PE-20	AVTA3	0.50	20	4795	205	100	0.070	36	280,000	202,235	315,204
PE-21	AVTA3	0.50	20	4795	205	100	0.067	35	275,657	181,659	268,361
PE-22	AVTA3	1.00	40	4695	305	100	0.067	28	344,571	97,996	137,432

PE-23	AVTA3	1.00	40	4695	305	100	0.069	29	342,621	98,229	147,216
EO-1	TnOA1	0.05	2	4845	105	70	0.140	101	199,604	699,239	1,000,073
EO-2	AVTA3	0.25	10	4795	155	70	0.130	24	780,000	613,941	1,225,435
EO-3	AVTA3	0.25	10	4795	155	70	0.120	56	308,571	621,889	1,170,603
EO-4	AVTA3	0.50	20	4745	205	70	0.115	82	201,951	335,885	755,194
EO-5	AVTA3	0.50	20	4745	205	70	0.052	38	197,053	177,525	303,355
EO-6	AVTA3	1.00	40	4645	305	70	0.114	37	443,676	177,501	300,150
EO-7	AVTA3	1.00	40	4645	305	70	0.082	25	472,320	134,329	242,503
EO-8	TnOA1	0.05	2	4845	105	85	0.120	181	95,470	549,028	808,311
EO-9	TnOA1	0.05	2	4845	105	85	0.115	130	127,385	443,501	1,052,865
EO-10	AVTA3	0.25	10	4795	155	85	0.113	96	169,500	395,938	1,213,520
EO-11	AVTA3	0.25	10	4795	155	85	0.104	33	453,818	450,414	849,932
EO-12	TnOA1	0.25	10	4810	140	85	0.121	71	245,408	205,259	316,386
EO-13	AVTA4	0.25	10	4760	190	85	0.122	124	141,677	165,770	402,164
EO-14	AVTA5	0.25	10	4760	190	85	0.110	47	337,021	219,643	426,475
EO-15	AVTA3	0.50	20	4745	205	85	0.107	65	237,046	298,521	562,949
EO-16	AVTA3	0.50	20	4745	205	85	0.101	28	519,429	259,826	527,495
EO-17	TnOA1	0.50	20	4810	140	85	0.125	37	486,486	160,324	220,213
EO-18	AVTA4	0.50	20	4710	240	85	0.117	51	330,353	163,071	301,301
EO-19	AVTA5	0.50	20	4710	240	85	0.112	49	329,143	183,979	323,797
EO-20	AVTA3	1.00	40	4645	305	85	0.098	23	613,565	156,718	253,205
EO-21	AVTA3	1.00	40	4645	305	85	0.087	26	481,846	122,922	200,078
EO-22	TnOA1	1.00	40	4809	140	85	0.117	23	732,522	100,600	127,448
EO-23	AVTA4	1.00	40	4610	340	85	0.114	32	513,000	118,903	215,431
EO-24	AVTA5	1.00	40	4610	340	85	0.107	28	550,286	125,780	215,005
EO-25	TnOA1	1.50	60	4809	140	85	0.115	17	974,118	70,613	91,660
EO-26	AVTA4	1.50	60	4510	439	85	0.113	26	625,846	100,956	171,649
EO-27	AVTA5	1.50	60	4510	439	85	0.109	23	682,435	98,506	170,689
EO-28	TnOA1	2.00	80	4809	140	85	0.116	14	1,193,143	58,901	76,808
EO-29	AVTA4	2.00	80	4410	539	85	0.107	22	700,364	80,193	133,598
EO-30	AVTA5	2.00	80	4410	539	85	0.107	20	770,400	87,528	133,709
EO-31	TnOA1	2.50	100	4809	140	85	0.112	14	1,152,000	50,989	65,163
EO-32	AVTA4	2.50	100	4310	639	85	0.108	19	818,526	68,664	119,550
EO-33	AVTA5	2.50	100	4310	639	85	0.106	21	726,857	78,826	120,684

EO-34	TnOAl	0.05	2	4845	105	100	0.100	58	248,276	308,415	557,381	
EO-35	TnOAl	0.05	2	4845	105	100	0.075	73	147,945	527,364	717,343	
EO-36	AVTA3	0.25	10	4795	155	100	0.094	76	178,105	320,792	601,663	
EO-37	AVTA3	0.25	10	4795	155	100	0.090	47	275,745	335,360	638,845	
EO-38	TnOAl	0.25	10	4810	140	100	0.094	35	386,743	173,587	246,764	
EO-39	AVTA4	0.25	10	4760	190	100	0.091	32	409,500	161,234	282,821	
EO-40	AVTA5	0.25	10	4760	190	100	0.088	32	396,000	167,509	284,855	
EO-41	AVTA3	0.50	20	4745	205	100	0.084	44	274,909	200,438	317,900	
EO-42	AVTA3	0.50	20	4745	205	100	0.088	39	324,923	240,124	414,829	
EO-43	TnOAl	0.50	20	4810	140	100	0.100	24	600,000	122,961	172,818	
EO-44	AVTA4	0.50	20	4710	240	100	0.090	26	498,462	114,065	213,077	
EO-45	AVTA5	0.50	20	4710	240	100	0.091	27	485,333	142,588	228,232	
EO-46	AVTA3	1.00	40	4645	305	100	0.078	25	449,280	109,700	173,250	
EO-47	AVTA3	1.00	40	4645	305	100	0.079	25	455,040	119,665	179,796	
EO-48	TnOAl	1.00	40	4809	140	100	0.098	19	742,737	79,958	103,513	
EO-49	AVTA4	1.00	40	4610	340	100	0.090	24	540,000	100,060	156,084	
EO-50	AVTA5	1.00	40	4610	340	100	0.087	24	522,000	106,284	159,047	
EO-51	TnOAl	1.50	60	4809	140	100	0.092	15	883,200	57,647	72,663	
EO-52	AVTA4	1.50	60	4510	439	100	0.087	21	596,571	76,639	119,797	
EO-53	AVTA5	1.50	60	4510	439	100	0.083	22	543,273	76,289	119,640	
EO-54	TnOAl	2.00	80	4809	140	100	0.087	15	835,200	44,450	56,907	
EO-55	AVTA4	2.00	80	4410	539	100	0.085	18	680,000	65,794	99,659	
EO-56	AVTA5	2.00	80	4410	539	100	0.082	18	656,000	64,297	99,337	
EO-57	TnOAl	2.50	100	4809	140	100	0.087	14	894,857	37,830	49,843	
EO-58	AVTA4	2.50	100	4310	639	100	0.086	17	728,471	54,455	89,639	
EO-59	AVTA5	2.50	100	4310	639	100	0.083	20	597,600	56,310	87,178	

Table 6. Additional characterization of high throughput PE and EO runs comparing tri-n-octyl aluminum (TNOAl) vs. AVTA's.

Type of Run	¹ H NMR mol% C8	¹ H NMR Wt% C8	¹ H NMR vinyl- enes/ 1000C	¹ H NMR trisubs/ 1000C	¹ H NMR vinyls/ 1000C	¹ H NMR vinyli- denes/ 1000C	¹ H NMR % vinyl
PE-15			0.05	0	0.11	0	69
PE-23			0.04	0	0.12	0	75
EO-2	3.0	11.1					
EO-6	1.8	6.7	0.07	0.05	0.15	0.01	54
EO-9	5.4	18.6					
EO-11	3.7	13.3					
EO-15	3.5	12.8					
EO-17	4.2	14.9					
EO-18	4.8	16.7					
EO-19	4.7	16.6					
EO-20	2.1	8.0	0.02	0	0.05	0	71
EO-22	3.8	13.5					
EO-23	4.0	14.3					
EO-24	3.4	12.2					
EO-28	2.6	9.8					
EO-29	3.1	11.4					
EO-30	2.7	10.1	0.01	0.01	0.08	0	80
EO-31	2.7	10.0					
EO-32	2.8	10.5					
EO-33	2.1	8.0					
EO-34	5.6	19.2					
EO-36	4.1	14.6					
EO-42	2.9	10.7					
EO-43	3.9	13.9					
EO-44	4.0	14.3					
EO-45	2.8	10.4					
EO-47	1.3	5.0	0.02	0	0.1	0	83
EO-48	3.7	13.3					
EO-49	3.5	12.6					
EO-50	2.7	10.2					
EO-54	3.0	11.1					
EO-55	2.4	8.9					
EO-56	1.8	6.8	0.01	0	0.11	0	92
EO-57	2.9	10.6					
EO-58	2.2	8.4					
EO-59	1.7	6.6	0.02	0	0.13	0	87

Table 7. Additional characterization of high throughput PE and EO runs comparing tri-n-octyl aluminum (TNOA)

Type of Run	GPC3D Malls Mn	GPC3D Malls Mw	GPC3D Malls Mz	GPC3D Malls Mv	GPC3D Malls Mw/Mn	GPC3D Malls g'(vis)	GPC DRI Mn	GPC DRI Mw	GPC DRI Mz
PE-9	269,178	338,477	398,608	329,351	1.26	1.03	185,665	275,765	392,111
PE-11	310,636	619,218	1,191,134	555,911	1.99	0.853	207,539	435,813	902,111
PE-13	180,217	362,977	736,974	323,882	2.01	0.851	146,397	289,819	552,111
PE-15	59,181	91,274	143,068	85,152	1.54	0.906	54,249	89,616	142,111
EO-9	201,715	332,847	435,524	316,952	1.65	1.053	142,147	295,436	452,111
EO-11	210,890	475,673	892,771	426,204	2.26	0.893	157,214	386,919	742,111
EO-15	155,773	313,479	586,661	282,195	2.01	0.863	123,650	265,499	512,111
EO-20	71,054	119,356	207,320	109,664	1.68	0.879	60,770	111,813	192,111
EO-22	41,978	50,836	59,683	49,529	1.21	0.99	36,186	49,322	62,111
EO-23	53,014	93,185	138,500	86,769	1.76	0.98	43,265	92,101	122,111
EO-24	64,906	112,463	177,004	104,127	1.73	0.916	52,225	105,237	182,111
EO-29	33,442	56,228	86,569	52,325	1.68	0.976	28,258	54,308	82,111
EO-30	35,778	59,345	95,959	55,023	1.66	0.925	29,143	55,352	92,111
EO-32	30,245	49,759	73,249	46,514	1.65	0.959	25,648	49,378	82,111
EO-33	29,738	50,351	84,155	46,486	1.69	0.913	24,314	46,758	72,111
EO-48	34,535	42,379	50,451	41,207	1.23	0.981	29,745	41,519	52,111
EO-49	41,380	64,626	93,670	60,751	1.56	0.968	34,611	62,946	102,111
EO-50	41,429	69,968	111,177	64,917	1.69	0.928	35,247	66,489	112,111
EO-55	26,239	40,908	60,500	38,373	1.56	0.965	22,391	39,289	62,111
EO-56	25,391	40,801	64,127	38,011	1.61	0.924	22,023	38,657	62,111
EO-58	22,944	36,453	53,146	34,170	1.59	0.952	19,189	36,130	52,111
EO-59	23,465	36,442	56,964	34,006	1.55	0.912	19,426	34,280	52,111

Table 8. High throughput ethylene-propylene (EP) runs comparing tri-n-octyl aluminum (TNOAl) vs. AVTA's.

Conditions: For all examples in Table 8, 0.025 μmol of catalyst 3 was used and activated with 1.1 molar equivalent tetrakis(pentafluorophenyl)borane. Propylene (0.5 ml) was added, and the total reagent and solvent volume in the reactor was 1.0 ml. The reactor was pressurized with ethylene to 125 psid and utilized 0.05 μmol of bis(diisobutyl aluminum)oxide as scavenger. Ethylene (125 psid) was added before catalyst. The reactor was vented and ethylene was added to the reactor. The run was quenched after 4 psi of reactor pressure loss.

Type of Run	AVTA or TnOAl	AVTA or TnOAl (μmol)	AVTA/Cat or TnOAl/Cat (molar ratio)	Iso-hexane (uL)	Toluene (uL)	T (C)	yield (g)	actual quench time (s)	Activity (g P/mmol cat.hr)	Rapid Mn	Rapid Mw
EP-1	TnOAl	0.25	10	4153	222	85	0.1043	7	2,145,600	67,972	109,040
EP-2	TnOAl	1.00	40	4152	222	85	0.1031	5	2,969,280	43,729	64,880
EP-3	TnOAl	2.00	80	4152	222	85	0.1089	4	3,920,400	33,502	48,170
EP-4	TnOAl	2.50	100	3903	471	85	0.0960	5	2,764,800	26,660	37,700
EP-5	TnOAl	3.50	140	3802	571	85	0.0965	5	2,779,200	18,816	28,340
EP-6	TnOAl	4.00	160	3752	621	85	0.0914	6	2,193,600	16,654	27,030
EP-7	TnOAl	0.25	10	4153	222	100	0.0735	6	1,764,000	64,713	103,320
EP-8	TnOAl	1.00	40	4152	222	100	0.0747	6	1,792,800	41,146	59,700
EP-9	TnOAl	2.00	80	4152	222	100	0.0846	5	2,436,480	29,388	42,600
EP-10	TnOAl	2.50	100	3903	471	100	0.0723	8	1,301,400	21,546	31,550
EP-11	TnOAl	3.50	140	3802	571	100	0.0733	6	1,759,200	16,987	25,780
EP-12	TnOAl	4.00	160	3752	621	100	0.0800	6	1,920,000	16,001	23,210
EP-13	AVTA6	0.25	10	4103	272	85	0.1204	6	2,889,600	78,151	127,910
EP-14	AVTA6	0.25	10	4103	272	85	0.1183	6	2,839,200	71,587	125,150
EP-15	AVTA6	0.50	20	4053	322	85	0.1133	5	3,263,040	59,429	105,050
EP-16	AVTA6	0.50	20	4053	322	85	0.1114	5	3,208,320	75,246	117,980
EP-17	AVTA6	1.00	40	3953	422	85	0.1137	6	2,728,800	61,453	95,950
EP-18	AVTA6	1.00	40	3953	422	85	0.1055	5	3,038,400	66,359	101,980
EP-19	AVTA6	1.50	60	3852	522	85	0.1124	4	4,046,400	52,308	82,500
EP-20	AVTA6	1.50	60	3852	522	85	0.1100	5	3,168,000	51,255	80,950
EP-21	AVTA6	2.00	80	3752	622	85	0.1094	4	3,938,400	47,065	76,380

EP-22	AVTA6	2.00	80	3752	622	85	0.1096	4	3,945,600	45,201	72,63
EP-23	AVTA6	2.50	100	3902	472	85	0.1266	12	1,519,200	42,865	64,74
EP-24	AVTA6	2.50	100	3902	472	85	0.1090	6	2,616,000	37,994	63,47
EP-25	AVTA6	3.00	120	3853	521	85	0.1016	6	2,438,400	36,321	55,95
EP-26	AVTA6	3.00	120	3853	521	85	0.0858	4	3,088,800	29,387	46,84
EP-27	AVTA6	3.50	140	3803	571	85	0.0977	4	3,517,200	33,698	50,95
EP-28	AVTA6	3.50	140	3803	571	85	0.0913	8	1,643,400	31,413	49,24
EP-29	AVTA6	4.00	160	3752	621	85	0.1030	5	2,966,400	31,398	47,63
EP-30	AVTA6	4.00	160	3752	621	85	0.0855	8	1,539,000	27,437	42,51
EP-31	AVTA6	0.25	10	4103	272	100	0.0752	4	2,707,200	63,687	110,21
EP-32	AVTA6	0.25	10	4103	272	100	0.0743	8	1,337,400	79,827	127,04
EP-33	AVTA6	0.50	20	4053	322	100	0.0807	6	1,936,800	74,873	112,53
EP-34	AVTA6	0.50	20	4053	322	100	0.0740	8	1,332,000	75,379	117,12
EP-35	AVTA6	1.00	40	3953	422	100	0.0958	4	3,448,800	50,724	91,42
EP-36	AVTA6	1.00	40	3953	422	100	0.0747	6	1,792,800	61,174	95,59
EP-37	AVTA6	1.50	60	3852	522	100	0.0827	6	1,984,800	52,151	76,52
EP-38	AVTA6	1.50	60	3852	522	100	0.0770	5	2,217,600	48,206	74,53
EP-39	AVTA6	2.00	80	3752	622	100	0.0815	4	2,934,000	43,859	64,59
EP-40	AVTA6	2.00	80	3752	622	100	0.0724	5	2,085,120	41,366	66,13
EP-41	AVTA6	2.50	100	3902	472	100	0.0816	4	2,937,600	35,439	54,23
EP-42	AVTA6	2.50	100	3902	472	100	0.0827	6	1,984,800	36,506	56,25
EP-43	AVTA6	3.00	120	3853	521	100	0.0706	6	1,694,400	29,872	47,71
EP-44	AVTA6	3.00	120	3853	521	100	0.0589	6	1,413,600	24,302	39,72
EP-45	AVTA6	3.50	140	3803	571	100	0.0758	4	2,728,800	27,566	42,19
EP-46	AVTA6	3.50	140	3803	571	100	0.0687	5	1,978,560	24,509	39,21
EP-47	AVTA6	4.00	160	3752	621	100	0.0863	6	2,071,200	26,086	41,38
EP-48	AVTA6	4.00	160	3752	621	100	0.0736	6	1,766,400	25,416	39,21

Table 9. Additional characterization of high throughput EP runs comparing tri-n-octyl aluminum (TNOAl) vs. AVTA's.

Type of Run	¹ H NMR mol % C3	¹ H NMRW t% C3	¹ H NMR vinylenes/ 1000C	¹ H NMR trisubs/ 1000C	¹ H NMR vinyls/ 1000C	¹ H NMR vinylidene/ 1000C	¹ H NMR % vinyl
EP-2	15.9	22.1					
EP-6	14.8	20.6					
EP-13	16.2	22.5					
EP-15	16.4	22.7					
EP-17	16.0	22.2					
EP-19	16.1	22.4	0.00	0.00	0.22	0.02	92
EP-22	16.2	22.5	0.00	0.01	0.32	0.02	91
EP-23	20.5	27.8	0.03	0.09	0.36	0.03	71
EP-25	16.2	22.5	0.00	0.01	0.59	0.01	97
EP-27	15.5	21.6	0.04	0.02	0.69	0.02	90
EP-29	16.7	23.1	0.00	0.14	0.85	0.12	77
EP-35	15.5	21.5	0.02	0.00	0.16	0.00	89
EP-39	14.5	20.3	0.00	0.00	0.40	0.01	98
EP-43	14.0	19.6	0.05	0.07	0.94	0.02	87
EP-45	14.8	20.6	0.00	0.05	1.11	0.07	90

Table 10. Additional characterization of high throughput EP runs comparing tri-n-octyl aluminum (TNOAl) vs. AVTA's.

Type of Run	GPC3D Malls Mn	GPC3D Malls Mw	GPC3D Malls Mz	GPC3D Malls Mv	GPC3D Malls Mw/Mn	GPC3D Malls g'(vis)	GPC DRI Mn	GPC DRI Mw	GPC DRI Mz	GPC DRI Mw/Mn
EP-2	14,489	22,885	32,096	21,453	1.58	1.008	14,281	26,613	39,609	1.86
EP-6	7,297	9,962	14,592	9,361	1.37	0.924	6,289	15,722	194,650	2.50
EP-15	26,294	45,960	71,298	42,534	1.75	1.01	25,456	48,885	76,945	1.92
EP-17	24,575	39,981	59,048	37,257	1.63	0.993	21,352	44,047	70,220	2.06
EP-19	19,973	32,240	48,552	29,971	1.61	0.982	18,642	35,173	54,576	1.89
EP-22	16,907	28,253	41,941	26,274	1.67	0.97	16,225	31,332	49,800	1.93
EP-23	18,364	24,235	35,861	22,514	1.32	0.988	13,362	27,525	44,982	2.06
EP-25	12,927	19,373	27,632	18,061	1.50	0.968	11,163	22,696	37,230	2.03
EP-27	11,643	17,388	25,818	16,001	1.49	0.997	10,721	20,825	33,102	1.94
EP-29	10,897	17,203	24,844	16,066	1.58	0.957	10,645	19,239	29,507	1.81
EP-35	23,038	34,891	50,878	32,530	1.51	0.989	20,759	39,299	64,023	1.89
EP-39	16,047	25,018	37,105	23,167	1.56	0.993	14,304	28,148	43,491	1.97
EP-43	11,529	16,828	23,965	15,738	1.46	0.955	9,072	19,065	30,144	2.10
EP-45	9,906	14,955	22,297	13,866	1.51	0.978	9,068	17,299	26,908	1.91

[00219] Unless otherwise indicated, molecular weight (weight-average molecular weight, M_w , number-average molecular weight, M_n , and molecular weight distribution, M_w/M_n or MWD, and branching index (g'vis)) are determined using a High Temperature Size Exclusion Chromatograph (either from Waters Corporation or Polymer Laboratories), equipped with a differential refractive index detector (DRI), an online light scattering (LS) detector, and a viscometer. Experimental details not described below, including how the detectors are calibrated, are described in: T. Sun, P. Brant, R. R. Chance, and W. W. Graessley, *Macromolecules*, Volume 34, Number 19, 6812-6820, (2001).

[00220] Three Polymer Laboratories PLgel 10mm Mixed-B columns are used. The nominal flow rate was $0.5 \text{ cm}^3/\text{min}$, and the nominal injection volume is $300 \text{ }\mu\text{L}$. The various transfer lines, columns and differential refractometer (the DRI detector) are contained in an oven maintained at 135°C . Solvent for the SEC experiment is prepared by dissolving 6 grams of butylated hydroxy toluene as an antioxidant in 4 liters of Aldrich reagent grade 1, 2, 4 trichlorobenzene (TCB). The TCB mixture is then filtered through a $0.7 \text{ }\mu\text{m}$ glass pre-filter and subsequently through a $0.1 \text{ }\mu\text{m}$ Teflon filter. The TCB is then degassed with an online degasser before entering the SEC. Polymer solutions are prepared by placing dry polymer in a glass container, adding the desired amount of TCB, then heating the mixture at 160°C with continuous agitation for about 2 hours. All quantities are measured gravimetrically. The TCB densities used to express the polymer concentration in mass/volume units are 1.463 g/ml at room temperature and 1.324 g/ml at 135°C . The injection concentration ranges from 1.0 to 2.0 mg/ml , with lower concentrations being used for higher molecular weight samples. Prior to running each sample the DRI detector and the injector are purged. Flow rate in the apparatus is then increased to 0.5 ml/minute , and the DRI is allowed to stabilize for 8-9 hours before injecting the first sample. The LS laser is turned on 1 to 1.5 hours before running samples.

[00221] The concentration, c , at each point in the chromatogram is calculated from the baseline-subtracted DRI signal, I_{DRI} , using the following equation:

$$c = K_{\text{DRI}} I_{\text{DRI}} / (dn/dc)$$

where K_{DRI} is a constant determined by calibrating the DRI, and (dn/dc) is the same as described below for the light scattering (LS) analysis. Units on parameters throughout this description of the SEC method are such that concentration is expressed in g/cm^3 , molecular weight is expressed in g/mole , and intrinsic viscosity is expressed in dL/g .

[00222] The light scattering detector used is a Wyatt Technology High Temperature mini-DAWN. The polymer molecular weight, M , at each point in the chromatogram is determined

by analyzing the LS output using the Zimm model for static light scattering (M.B. Huglin, LIGHT SCATTERING FROM POLYMER SOLUTIONS, Academic Press, 1971):

$$\frac{K_o c}{\Delta R(\theta)} = \frac{1}{MP(\theta)} + 2A_2 c$$

[00223] Here, $\Delta R(\theta)$ is the measured excess Rayleigh scattering intensity at scattering angle θ , c is the polymer concentration determined from the DRI analysis, A_2 is the second virial coefficient [for purposes of this invention and the claims thereto, $A_2 = 0.0006$ for propylene polymers and 0.001 otherwise], $P(\theta)$ is the form factor for a monodisperse random coil (M.B. Huglin, LIGHT SCATTERING FROM POLYMER SOLUTIONS, Academic Press, 1971), and K_o is the optical constant for the system:

$$K_o = \frac{4\pi^2 n^2 (dn/dc)^2}{\lambda^4 N_A}$$

in which N_A is Avogadro's number, and (dn/dc) is the refractive index increment for the system. The refractive index, $n = 1.500$ for TCB at 135°C and $\lambda = 690$ nm. For purposes of this invention and the claims thereto $(dn/dc) = 0.104$ for propylene polymers and 0.1 otherwise.

[00224] A high temperature Viscotek Corporation viscometer, which has four capillaries arranged in a Wheatstone bridge configuration with two pressure transducers, is used to determine specific viscosity. One transducer measures the total pressure drop across the detector, and the other, positioned between the two sides of the bridge, measures a differential pressure. The specific viscosity, η_s , for the solution flowing through the viscometer is calculated from their outputs. The intrinsic viscosity, $[\eta]$, at each point in the chromatogram is calculated from the following equation:

$$\eta_s = c[\eta] + 0.3(c[\eta])^2$$

where c is concentration and is determined from the DRI output.

[00225] The branching index, g' (also referred to as g'_{vis}), is calculated using the output of the SEC-DRI-LS-VIS method as follows. The average intrinsic viscosity, $[\eta]_{avg}$, of the sample is calculated by:

$$[\eta]_{avg} = \frac{\sum c_i [\eta]_i}{\sum c_i}$$

where the summations are over the chromatographic slices, i , between the integration limits.

[00226] The branching index g'_{vis} is defined as:

$$g' = \frac{[\eta]_{avg}}{kM_v^\alpha}$$

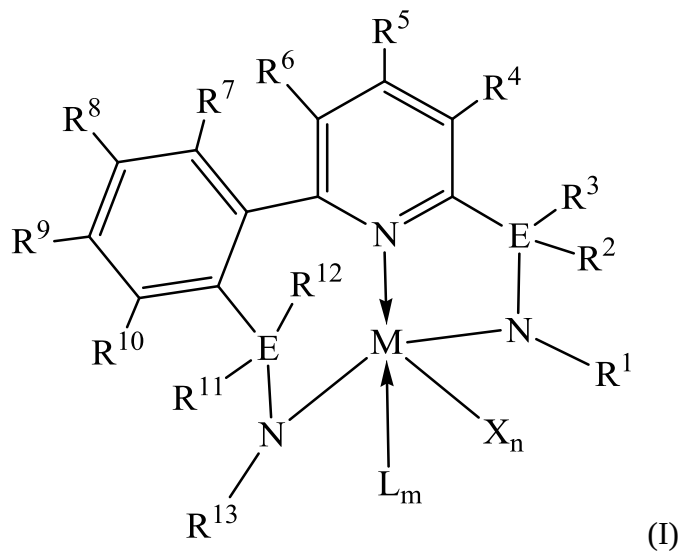
where, for purpose of this invention and claims thereto, $\alpha = 0.695$ for ethylene, propylene, and butene polymers; and $k = 0.000579$ for ethylene polymers, $k = 0.000262$ for propylene polymers, and $k = 0.000181$ for butene polymers. M_v is the viscosity-average molecular weight based on molecular weights determined by LS analysis.

- 5 [00227] All documents described herein are incorporated by reference herein for purposes of all jurisdictions where such practice is allowed, including any priority documents and/or testing procedures to the extent they are not inconsistent with this text. As is apparent from the foregoing general description and the specific embodiments, while forms of the invention have been illustrated and described, various modifications can be made without departing
- 10 from the spirit and scope of the invention. Accordingly, it is not intended that the invention be limited thereby. Likewise, the term “comprising” is considered synonymous with the term “including.” Likewise whenever a composition, an element or a group of elements is preceded with the transitional phrase “comprising,” it is understood that we also contemplate the same composition or group of elements with transitional phrases “consisting essentially
- 15 of,” “consisting of,” “selected from the group of consisting of,” or “is” preceding the recitation of the composition, element, or elements and vice versa.

CLAIMS

What is claimed is:

1. A catalyst system comprising a pyridyldiamido transition metal complex; an activator; a metal hydrocarbenyl chain transfer agent; and optionally, a support, wherein the pyridyldiamido transition metal complex is represented by the formula:



wherein:

M is a Group 4 metal;

E is selected from carbon, silicon, or germanium;

X is an anionic leaving group;

L is a neutral Lewis base;

R¹ and R¹³ are each independently selected from the group consisting of hydrocarbyls, substituted hydrocarbyls, and silyl groups;

R², R³, R⁴, R⁵, R⁶, R⁷, R⁸, R⁹, R¹⁰, R¹¹, and R¹² are each independently selected from the group consisting of hydrogen, hydrocarbyls, alkoxy, silyl, amino, aryloxy, substituted hydrocarbyls, halogen, and phosphino;

n is 1 or 2;

m is 0, 1, or 2; and

two X groups may be joined together to form a dianionic group;

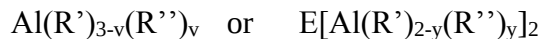
two L groups may be joined together to form a bidentate Lewis base;

an X group may be joined to an L group to form a monoanionic bidentate group;

R⁷ and R⁸ may be joined to form a ring; and

R¹⁰ and R¹¹ may be joined to form a ring.

2. The catalyst system of claim 1, wherein the metal hydrocarbenyl chain transfer agent is represented by the formula:



wherein each R', independently, is a C₁-C₃₀ hydrocarbyl group;

each R'', independently, is a C₄-C₂₀ hydrocarbenyl group having an allyl chain end;

E is a group 16 element;

v is from 0.01 to 3; and

y is from 0.01 to 2.

3. The catalyst system of claim 1, wherein M is Ti, Zr, or Hf;

E is carbon;

X is an alkyl, aryl, hydride, alkylsilane, fluoride, chloride, bromide, iodide, triflate, carboxylate, or alkylsulfonate; and

L is an ether, amine, or thioether.

4. The catalyst system of claim 1, wherein the activator comprises an alumoxane and/or a non-coordinating anion, preferably the non-coordinating anion comprises one or more of:

trimethylammonium	tetrakis(perfluoronaphthyl)borate,	N,N-dimethylanilinium
tetrakis(perfluoronaphthyl)borate,		N,N-diethylanilinium
tetrakis(perfluoronaphthyl)borate,		triphenylcarbenium
tetrakis(perfluoronaphthyl)borate,		trimethylammonium
tetrakis(perfluorobiphenyl)borate,		N,N-dimethylanilinium
tetrakis(perfluorobiphenyl)borate,		triphenylcarbenium
tetrakis(perfluorobiphenyl)borate,	N,N-dimethylanilinium	tetrakis(3,5-
bis(trifluoromethyl)phenyl)borate,		triphenylcarbenium
tetrakis(perfluoronaphthyl)borate,		triphenylcarbenium
tetrakis(perfluorobiphenyl)borate,	triphenylcarbenium	tetrakis(3,5-
bis(trifluoromethyl)phenyl)borate,		triphenylcarbenium
tetrakis(perfluorophenyl)borate, [Ph ₃ C ⁺][B(C ₆ F ₅) ₄ ⁻], [Me ₃ NH ⁺][B(C ₆ F ₅) ₄ ⁻], 1-(4-		
(tris(pentafluorophenyl)borate)-2,3,5,6-tetrafluorophenyl)pyrrolidinium,		
tetrakis(pentafluorophenyl)borate,	4-(tris(pentafluorophenyl)borate)-2,3,5,6-	
tetrafluoropyridine, triphenylcarbenium tetraphenylborate, and triphenylcarbenium		

tetrakis-(2,3,4,6-tetrafluorophenyl)borate.

5. The catalyst system of claim 2, wherein R'' is butenyl, pentenyl, heptenyl, or octenyl and/or R' is methyl, ethyl, propyl, isobutyl, or butyl, and preferably v = 2.

6. The catalyst system of claim 1, wherein the catalyst system does not comprise a metallocene.

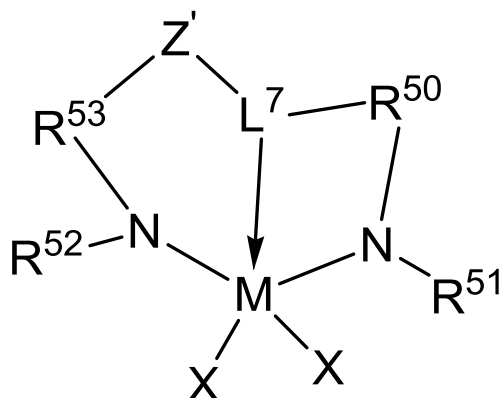
7. The catalyst system of claim 1, wherein the metal hydrocarbenyl chain transfer agent is an aluminum vinyl transfer agent (AVTA), wherein the aluminum vinyl transfer agent comprises one or more of tri(but-3-en-1-yl)aluminum, tri(pent-4-en-1-yl)aluminum, tri(oct-7-en-1-yl)aluminum, tri(non-8-en-1-yl)aluminum, tri(dec-9-en-1-yl)aluminum, dimethyl(oct-7-en-1-yl)aluminum, diethyl(oct-7-en-1-yl)aluminum, dibutyl(oct-7-en-1-yl)aluminum, diisobutyl(oct-7-en-1-yl)aluminum, diisobutyl(non-8-en-1-yl)aluminum, diisobutyl(dec-8-en-1-yl)aluminum, and diisobutyl(dodec-10-en-1-yl)aluminum.

8. A polymerization process comprising contacting one or more alkene monomers with the catalyst system of any one of claims 1 to 7.

9. The process of claim 8, wherein the monomer comprises ethylene or propylene.

10. The process of claim 8, further comprising the step of adding a metallocene in a second step.

11. A catalyst system comprising a transition metal complex; an activator; and an aluminum vinyl transfer agent, wherein the transition metal complex is represented by the formula:



wherein (1) M is a group 4 metal, preferably hafnium; (2) N is nitrogen; (3) L^7 is a group that links R^{50} to Z' by a three atom bridge with the central of the three atoms being a group 15 or 16 element that preferably forms a dative bond to M, and is a C_5 - C_{20} heteroaryl group containing a Lewis base functionality, especially a divalent pyridinyl or substituted pyridinyl group; (4) Z' is a divalent linker group, $(R^{56})_pC-C(R^{57})_q$, where R^{56} and R^{57} are independently selected from the group consisting of hydrogen, hydrocarbyls, substituted hydrocarbyls, and wherein adjacent R^{56} and R^{57} groups may be joined to form an aromatic or saturated, substituted or unsubstituted hydrocarbyl ring, wherein the ring has 5, 6, 7 or 8 ring carbon atoms and where the substituents on the ring can join to form additional rings, and p is 1 or 2 and q is 1 or 2; (5) R^{50} and R^{53} are each, independently, $ER^{54}R^{55}$ with E being carbon, silicon or germanium, and each R^{54} and R^{55} being independently selected from the group consisting of hydrogen, hydrocarbyls, substituted hydrocarbyls, alkoxy, silyl, amino, aryloxy, halogen and phosphino, and R^{54} and R^{55} may be joined to form a saturated heterocyclic ring, or a saturated substituted heterocyclic ring where substitutions on the ring can join to form additional rings; (6) R^{51} and R^{52} are independently selected from the group consisting of hydrocarbyls, substituted hydrocarbyls, silylcarbyls and substituted silylcarbyl groups; and (7) each X is independently a univalent anionic ligand, or two Xs are joined and bound to the metal atom to form a metallocycle ring, or two Xs are joined to form a chelating ligand, a diene ligand, or an alkylidene ligand,

wherein the aluminum vinyl transfer agent is represented by the formula: $Al(R')_{3-v}(R'')_v$, wherein each R' , independently, is a C_1 to C_{30} hydrocarbyl group;

each R'', independently, is a C₄ to C₂₀ hydrocarbenyl group having an allyl chain end; and v is from 0.01 to 3.

12. A polymerization process to produce polyolefin having allyl chain ends, and optionally a g'vis of 0.95 or more, comprising contacting one or more alkene monomers with the catalyst system of claim 1 where grams of polymer produced/mols of metal hydrocarbenyl chain transfer agent added to the reactor is 500,000 or more.

13. A polymerization process to produce polyolefin having a g'vis of 0.95 or less comprising contacting one or more alkene monomers with the catalyst system of claim 1 where grams of polymer produced/mols of metal hydrocarbenyl chain transfer agent added to the reactor is less than 500,000.