

Description

BACKGROUND OF THE INVENTION

This invention relates to long wavelength ultraviolet ray absorbers (hereinafter referred to simply as UV absorbers) and more particularly, to UV absorbers comprising dibenzoylmethane derivatives having one carboxylic acid, a carboxylate or a carboxylic acid ester group in each benzene ring thereof with or without hydrous oxides of polyvalent metallic elements.

Ultraviolet rays are known to cause various dermatological effects on the skin. Dermatologically, ultraviolet rays are divided into a long wavelength ultraviolet ray having a wavelength of from 400 to 320 nm, a medium wavelength ultraviolet ray having a wavelength of from 320 to 290 nm, and a short wavelength ultraviolet ray having a wavelength not larger than 290 nm, which are, respectively, called UV-A, UV-B and UV-C.

Sunlight is the main ultraviolet ray source to which men are ordinarily exposed. Ultraviolet rays in the sunlight which arrive at the earth are UV-A and UV-B. UV-C is absorbed in the ozone layer and does rarely reach the earth. Of the UV rays which arrive at the earth, UV-B serves to form erythema and blisters when irradiated on the skin over a certain level of exposure. In addition, melanogenesis, which leads to pigmentation, is accelerated. In contrast, UV-A was believed not to cause any substantial changes in the skin. In recent years, however, it has been revealed through electron-microscopic and histological techniques that the skin suffers dermatological changes by irradiation of UV-A. In particular, the energy of UV-A arrives even at the inner skin as is different from UV-B and thus causes elastic fibers in the blood vessel walls and connective-tissue membranes to be affected slightly chronically. These changes are considered to lead to the promotion of senility of the skin. Moreover, it is also known that UV-A has the action of melanising the skin immediately after the irradiation thereof (immediate pigmentation) and increases the degenerating action of UV-B on the skin. Thus, UV-A is considered to be one of main factors of producing or exacerbating moth-patches or freckles on the face.

As will be appreciated from the above, it is important to protect the skin not only from UV-B, but also from UV-A in order to prevent aging, moth-patches, and freckles on skin being produced or exacerbated.

However, studies on the action of UV-A on the skin have just started and thus few substances of effectively absorbing UV-A upon application to the skin have been known. Currently known substances are only dibenzoylmethane derivatives [CH-A5-623229, FR-A-2506156, DE-A1-3129493 and CH-A5-595840] and cinnamic acid derivatives, most of which are liposoluble (West Germany Published Patent Nos. 2728241 and 2728243, Japanese Laid-open Patent Application Nos. 51-61641, 52-46056 and 57-197209). Only a few are soluble in water (Japanese Laid-open Patent Application No. 57-59840).

If it is intended to add these UV-A absorbers to cosmetics, limitations are placed on the properties of a cosmetic substrate. Accordingly, there is a demand for a development of more widely usable UV-A absorbers. Such UV-A absorbers should satisfy the following requirements.

- (1) A maximum absorption wavelength in the vicinity of 350 nm.
- (2) A sufficiently high molar extinction coefficient (ϵ) at the above-indicated wavelength.
- (3) A small absorbance in the visible light range because coloration of cosmetic compositions is not desirable, i.e. $\epsilon = 0$ over 400 nm.
- (4) Good stability against heat and light.
- (5) No toxic, irritative and other noxious actions on the skin.
- (6) Good compatibility with cosmetic substrates.
- (7) Difficulty in percutaneous absorption and in removal such as by perspiration upon application to the skin, with the effects of the absorber being effectively lasted.
- (8) Inexpensiveness.

Moreover, there is also a demand of a development for UV absorbers which are effective not only for UV-A, but also for UV-B. Such UV-absorbers should satisfy, aside from the above requirements (1) to (8), the requirement that they have a maximum absorption in the vicinity of 290 nm.

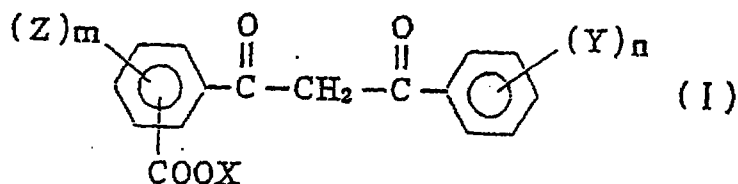
SUMMARY OF THE INVENTION

Under these circumstances, the present inventors made intensive studies and, as a result, found that dibenzoylmethane derivatives having specific substituents can satisfy all the requirements with regard to UV-A absorbers.

It was also found that the dibenzoylmethane derivatives used in combination with hydrous oxides of

polyvalent metallic elements were able to absorb not only UV-A, but also UV-B.

In accordance with one embodiment of the invention, there is provided a cosmetic composition long wavelength ultraviolet ray absorber which comprises a dibenzoylmethane derivative of the general formula (I)



in which X represents a hydrogen atom, a monovalent metal cation, an organic cation, a linear or branched alkyl group having from 1 to 24 carbon atoms, or a polyoxyalkylene oxide having from 1 to 10 alkylene oxide units with 2 or 3 carbon atoms in each unit, each Y and each Z independently represent a hydroxyl group, a linear or branched alkyl group having from 1 to 24 carbon atoms, or an alkoxy group having from 1 to 24 carbon atoms, or a polyoxyalkylene oxide group having from 1 to 10 alkylene oxide units with 2 or 3 carbon atoms in each unit, and m and n are independently an integer of from 0 to 3 as a cosmetic composition long wavelength ultraviolet ray absorber and the usual cosmetically acceptable ingredients.

The cosmetic composition may further comprise, aside from the dibenzoylmethane derivative (I), hydrous oxides of polyvalent metallic elements. Further contemplated by the invention is the use of the dibenzoylmethane derivative (I) for shielding the skin from UV rays.

BRIEF DESCRIPTION OF THE DRAWINGS

Fig. 1 is a graph showing SPF values of the absorber products 1 to 6 of the invention and comparative products 1 and 2;

Fig. 2 is a graph showing a molar extinction coefficient of a mixture of 4'-methoxydibenzoylmethane-2-carboxylic acid (hereinafter referred to simply as MDMC) and alumina monohydrate in relation to the weight ratio of the alumina monohydrate to MDMC;

Fig. 3 is a graph showing a molar extinction coefficient of a mixture of sodium 4'-methoxydibenzoylmethane-2-carboxylate (hereinafter referred to simply as SMDMC) and alumina monohydrate in relation to the weight ratio of the alumina monohydrate to SMDMC;

Fig. 4 is a graph showing an absorbance of mixtures of SMDMC with alumina monohydrate in relation to pH;

Fig. 5 is a graph showing an absorbance of SMDMC, mixtures of SMDMC with active alumina and aluminum hydroxide in relation to wavelength;

Fig. 6 is a graph showing a molar extinction coefficient of SMDMC mixed with hydrous iron oxide or hydrous titanium oxide in relation to ratio by weight of each oxide to SMDMC;

Fig. 7 is a graph showing a molar extinction coefficient of SMDMC mixed with hydrous iron oxide or hydrous titanium oxide in relation to wavelength; and

Fig. 8 is a graph showing an absorbance of emulsion A of the invention or emulsion B, indicated in Table 1, in relation to wavelength.

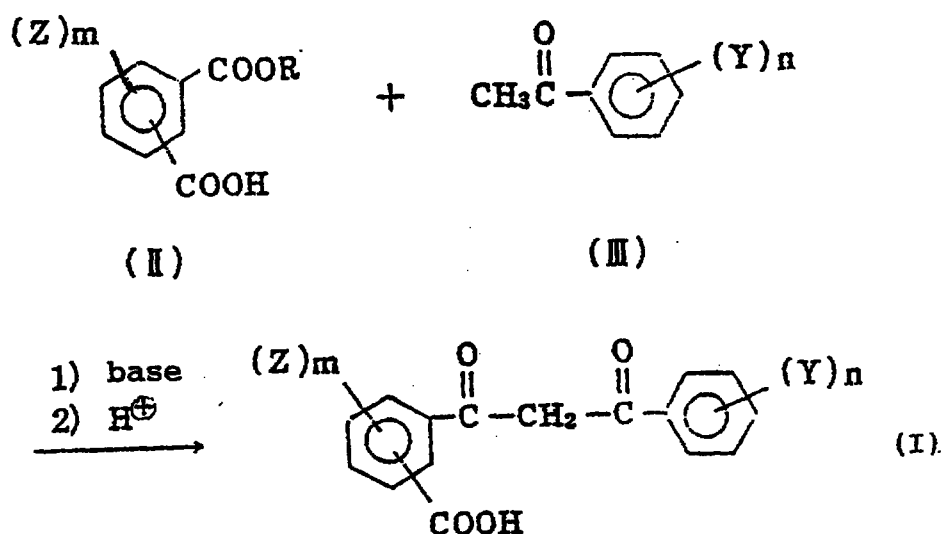
DETAILED DESCRIPTION OF THE INVENTION AND PREFERRED DESCRIPTION OF THE EMBODIMENTS

In the general formula (I), X represents a hydrogen atom, a monovalent metal cation, or an organic cation. Examples of the monovalent metal cation include sodium, potassium or lithium, and examples of the organic cation include arginine or trimethylethanolamine.

The dibenzoylmethane derivatives of the general formula (I) are prepared by known processes including, for example, processes described in Ann. Chim. (Rome), 48, 762(1958), J. Chem. Soc., 2063-(1952), Pestic. Sci., 4, 473(1973), and United States Patent No. 4381360.

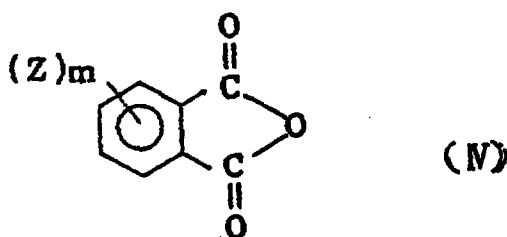
The simplest process is a process which comprises subjecting a monoester of substituted phthalic acid

(including all isomers such as ortho, meta and para isomers) having the following formula (II) to condensation reaction with substituted acetophenone of the formula (III) in the presence of a base according to the following reaction formula, thereby obtaining a compound of the formula (Ia), and treating the compound (Ia) by a usual manner to obtain an intended salt or ester



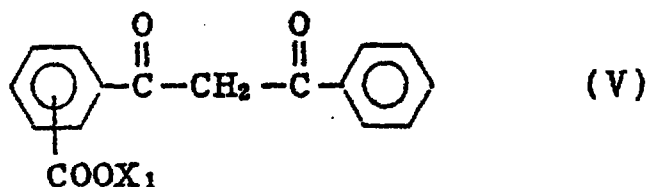
in which R represents a methyl group or an ethyl group, and Y, Z, m and n have, respectively, the same meanings as defined before.

If ortho-phthalic acid is used in the above process, an acid anhydride of the following general formula (IV) may be used instead of the monoester (II)

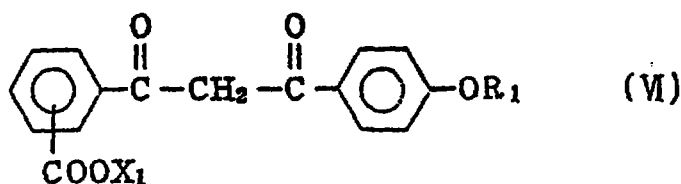


The dibenzoylmethane derivatives of the general formula (I) may be divided into the following groups.

(1) Dibenzoylmethanecarboxylic acids or salts of the general formula (V)

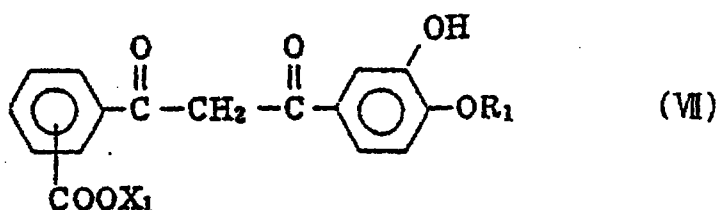


(2) 4'-alkoxydibenzoylmethanecarboxylic acids or salts of the general formula (VI)



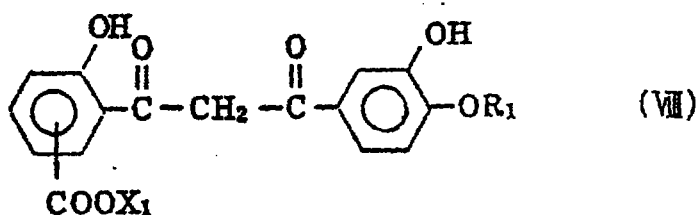
in which R₁ represents a hydrogen atom, or a hydrocarbon group having from 1 to 18 carbon atoms, preferably a linear or branched alkyl or alkylene group having from 1 to 4 carbon atoms, and the position of the substituent, -COOX₁, and X₁ are, respectively, the same as defined before.

(3) 3'-hydroxy-4'-alkoxydibenzoylmethane-carboxylic acids or salts of the general formula (VII)



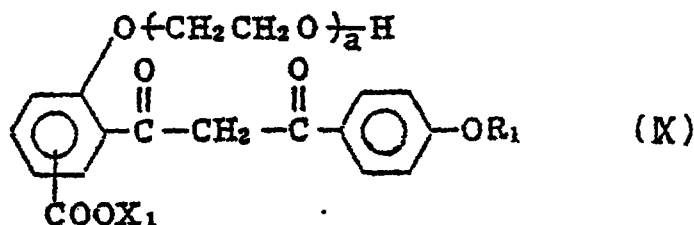
in which the position of the substituent, -COOX₁, X₁, and R₁ are, respectively, the same as defined before.

(4) 2,3'-dihydroxy-4'-alkoxydibenzoylmethanecarboxylic acids or salts of the general formula (VIII)



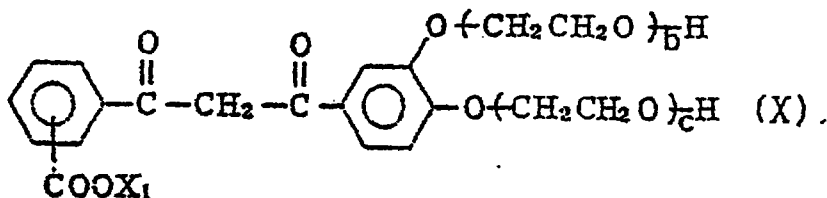
in which the position of the substituent, -COOX₁, X₁, and R₁ have, respectively, the same meanings as defined before.

(5) 4'-alkoxy-2-polyoxyethyleneoxydibenzoyl-methanecarboxylic acids or salts of the general formula (IX)



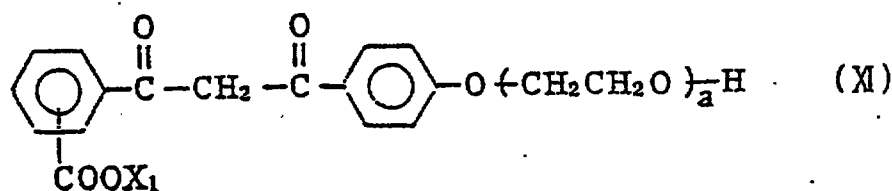
in which the position of the substituent, -COOX₁, X₁, and R₁ are, respectively, the same as defined before, and a is an integer of from 1 to 10, preferably from 1 to 6.

(6) 3',4'-bis(polyoxyethyleneoxy)-dibenzoyl-methanecarboxylic acids or salts of the general formula (X)



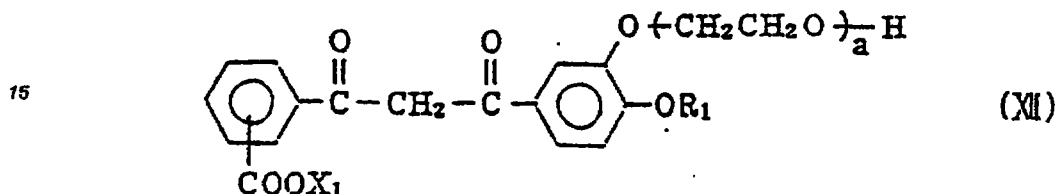
in which the position of the substituent, -COOX₁, and X₁ are, respectively, the same as defined before, and b and c are independently an integer of from 1 to 6, preferably from 1 to 3.

(7) 4'-polyoxyethyleneoxydibenzoylmethane-carboxylic acids or salts of the general formula (XI)



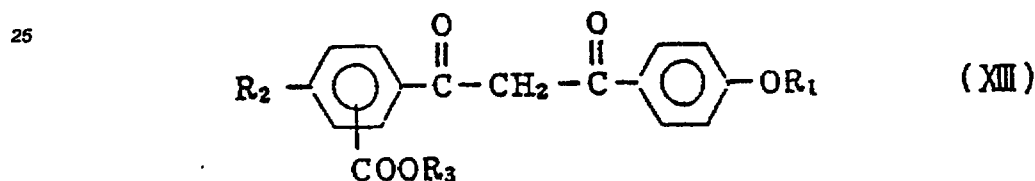
in which the position of the substituent, -COOX₁, X₁, and a are, respectively, the same as defined before.

(8) 4'-alkoxy-3'-polyoxyethyleneoxydibenzoyl-methanecarboxylic acids or salts of the general formula (XII)



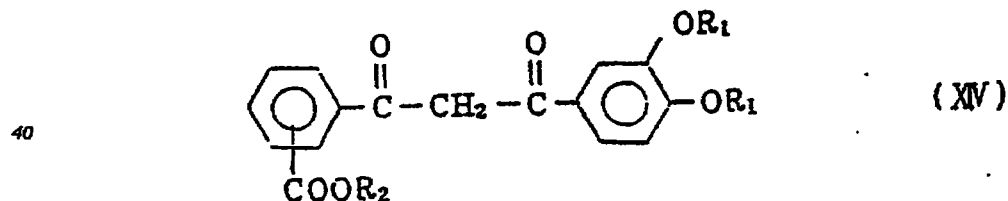
in which the position of the substituent, -COOX₁, X₁, R₁, and a are, respectively, the same as defined before.

(9) 4'-alkoxy-4-alkyldibenzoylmethane-carboxylic acid alkyl esters of the general formula (XIII)



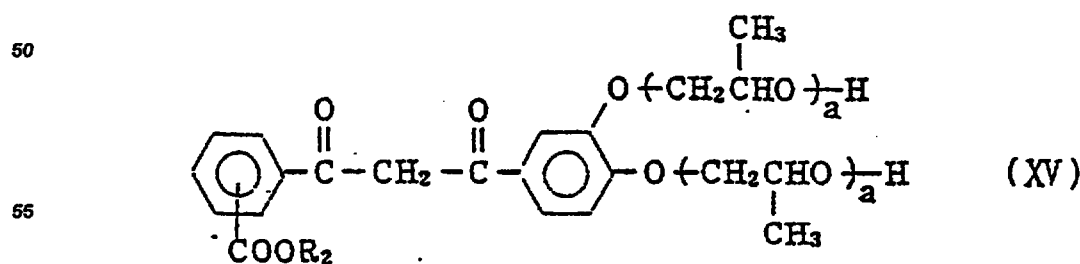
in which R₂ and R₃ independently represent a linear or branched alkyl or alkylene group having from 1 to 24 carbon atoms, preferably from 1 to 18 carbon atoms, the position of the substituent, -COOR₃, is an ortho or meta position, and R₁ has the same meaning as defined before.

(10) 3',4'-dialkoxydibenzoylmethane-carboxylic acid alkyl esters of the general formula (XIV)



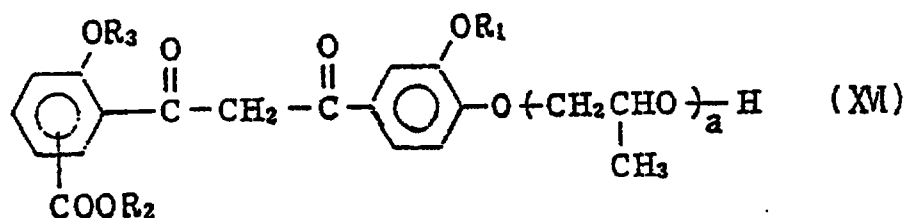
in which the position of the substituent, -COOR₂, is an ortho, meta or para position, and R₁ and R₂ have, respectively, the same meanings as defined before.

(11) 3',4'-bis(polyoxypropyleneoxy)dibenzoyl-methanecarboxylic acid alkyl esters of the general formula (XV)



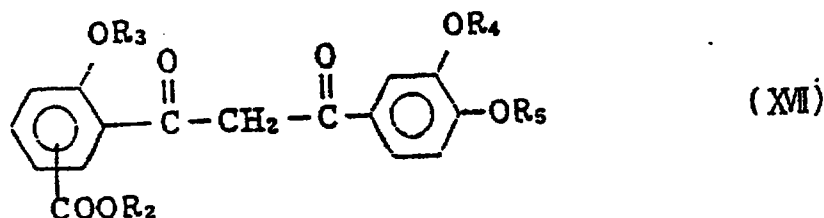
in which the position of the substituent, $-\text{COOR}_2$, and a have, respectively, the same meaning as defined before.

(12) 2,3'-dialkoxy-4'-polyoxypropyleneoxy-dibenzoylmethanecarboxylic acid alkyl esters of the general formula (XVI)



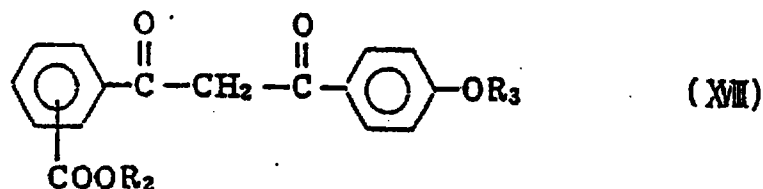
in which the position of the substituent, $-\text{COOR}_2$, R_1 , R_2 , R_3 , and a have, respectively, the same meanings as defined before.

(13) 2,3',4'-trialkoxydibenzoylmethane-carboxylic acid alkyl esters of the general formula (XVII)



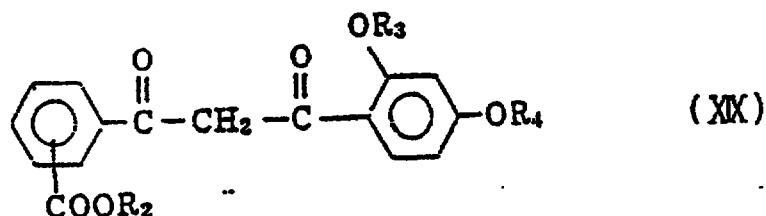
in which R_4 and R_5 independently represent a linear or branched alkyl or alkylene group having from 1 to 24 carbon atoms, preferably from 1 to 18 carbon atoms, and the position of the substituent, $-\text{COOR}_2$, R_2 , and R_3 have, respectively, the same meanings as defined before.

(14) 4'-alkoxydibenzoylmethanecarboxylic acid alkyl esters of the general formula (XVIII)



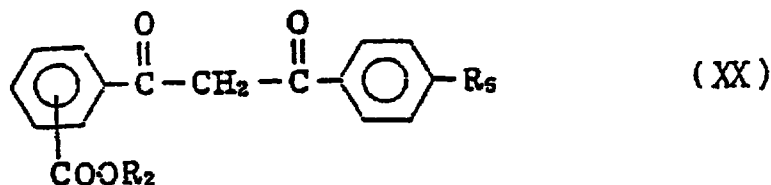
in which the position of the substituent, $-\text{COOR}_2$, R_2 , and R_3 have, respectively, the same meanings as defined before.

(15) 2',4'-dialkoxydibenzoylmethanecarboxylic acid alkyl esters of the general formula (XIX)



in which the position of the substituent, $-\text{COOR}_2$, R_2 , R_3 , and R_4 have, respectively, the same meanings as defined before.

(16) 4'-alkyldibenzoylmethanecarboxylic acid alkyl esters of the general formula (XX)



in which R_5 represents a hydrogen atom, or a linear or branched alkyl group having from 1 to 24 carbon atoms, and the position of the substituent, $-\text{COOR}_2$, and R_2 are, respectively, the same as defined before.

Among the dibenzoylmethane derivatives, compounds which are soluble in water include, for example, dibenzoylmethane-2-carboxylic acid trimethylethanolamine salt, sodium dibenzoylmethane-3-carboxylate, sodium 4'-methoxydibenzoylmethane-2-carboxylate, sodium 4'-methoxydibenzoylmethane-3-carboxylate, 3'-hydroxy-4'-methoxydibenzoylmethane-2-carboxylic acid, sodium 2,3'-dihydroxy-4'-methoxybenzoylmethane-5-carboxylic acid, potassium 4'-methoxy-2-polyoxyethyleneoxydibenzoylmethane-5-carboxylate, sodium 3',4'-bis(polyoxyethyleneoxy)-dibenzoylmethane-3-carboxylate, potassium 4'-methoxydibenzoylmethane-4-carboxylate, sodium 4'-polyoxyethyleneoxydibenzoylmethane-4-carboxylate, and potassium 4'-methoxy-3'-polyoxyethyleneoxydibenzoylmethane-4-carboxylate. Examples of the derivative compounds which are fat-soluble include 4'-methoxy-4-methyldibenzoylmethane-2-carboxylic acid isostearyl ester, 3',4'-dimethoxydibenzoylmethane-2-carboxylic acid oleyl ester, 3',4'-bis(polyoxypropoxy)dibenzoylmethane-2-carboxylic acid methyl ester, 2,3'-dimethoxy-4'-polyoxypropyleneoxydibenzoylmethane-5-carboxylic acid hexyl ester, 3',4'-dihexyloxy-2-methoxydibenzoylmethane-5-carboxylic acid butyl ester, 4'-tetracosyloxydibenzoylmethane-3-carboxylic acid methyl ester, 2',4'-dimethoxydibenzoylmethane-4-carboxylic acid stearyl ester, 4'-tert-butylidibenzoylmethane-4-carboxylic acid-2-ethylhexyl ester and 4'-palmityldibenzoylmethane-4-carboxylic acid isopropyl ester.

The hydrous oxides of polyvalent metallic elements used according to another embodiment of the invention are hydrous oxides of di- through hexa-valent metallic elements such as, for example, magnesium, beryllium, gallium, indium, iron (III), aluminum, lanthanum, manganese (III), antimony (III), bismuth (III), titanium (IV), tin (IV), zirconium, cerium (IV), thorium, silicon, manganese (IV), niobium, tantalum, antimony (V), molybdenum and tungsten.

The term "hydrous oxide" used herein may be also called hydrated oxides and means hydrates of metal oxides including so-called hydroxides as expressed by the general formula, $M_nO_m \cdot xH_2O$ in which M represents a metal, n and m are independently an integer, and x is a positive value. Examples of the hydrous oxides of polyvalent metallic elements include $\text{BeO} \cdot 1.7H_2O$, $\text{Mg}(\text{OH})_2$, $\text{Ga}_2\text{O}_3 \cdot 2.4H_2O$, $\text{In}_2\text{O}_3 \cdot 3.5H_2O$, $\text{Sb}_2\text{O}_3 \cdot 3H_2O$, $\text{Bi}_2\text{O}_3 \cdot 3H_2O$, $\text{La}_2\text{O}_3 \cdot 3H_2O$, $\text{Fe}_2\text{O}_3 \cdot 2H_2O$, $\text{beta-Fe}_2\text{O}_3 \cdot H_2O$, $\text{Al}_2\text{O}_3 \cdot 3H_2O$, $\text{Al}_2\text{O}_3 \cdot 1.2H_2O$, $\text{Al}_2\text{O}_3 \cdot H_2O$, $\text{MnO} \cdot \text{OH}$, $\text{MnO}_2 \cdot 0.6H_2O$, $\text{SiO}_2 \cdot 1.2H_2O$, $\text{alpha-TiO}_2 \cdot 1.6H_2O$, $\text{beta-TiO}_2 \cdot 0.8H_2O$, $\text{TiO}_2 \cdot 0.5H_2O$ (rutile), $\text{ZrO}_2 \cdot 2.8H_2O$, $\text{alpha-SnO}_2 \cdot 1.8H_2O$, $\text{beta-SnO}_2 \cdot 1.6H_2O$, $\text{CeO}_2 \cdot 2.6H_2O$, $\text{ThO}_2 \cdot 3.5H_2O$, $\text{Nb}_2\text{O}_5 \cdot 4.6H_2O$, $\text{Ta}_2\text{O}_5 \cdot 3.4H_2O$, $\text{Sb}_2\text{O}_5 \cdot 4H_2O$, HfO_2 , $\text{MoO}_3 \cdot 2H_2O$, $\text{WO}_3 \cdot 3H_2O$ and $\text{WO}_2 \cdot 1.8H_2O$. Where these hydrous oxides are added to cosmetic compositions, hydrous oxides of aluminum, titanium or silicon are preferred.

The UV-A absorber according to one embodiment of the invention is made by adding dibenzoylmethane derivatives (I) to a suitable carrier. The carrier should be inert to the dibenzoylmethane derivatives (I), and may be in the form of solid, liquid, emulsion, aerosol or gel. Typical carriers include water, alcohols, oils and fats such as, for example, hydrocarbon oils, aliphatic acid esters, long-chain alcohols or silicone oils, fine powders of starch or talc, and low boiling hydrocarbons or halogenated hydrocarbons used as aerosol propellants. The UV-A absorber of the invention may further comprise other additives such as preservatives, perfumes, colorants or surfactants, in amounts not impeding the UV-A absorbing action of the derivatives (I).

The dibenzoylmethane derivatives (I) whose carboxylic acid, $-\text{COOH}$, is suitably substituted with a group, X , have good affinity for various cosmetic substrates without impeding the UV-A absorption activity, so that the derivatives can be added to such substrates in order to impart UV-A absorbance. If X is a monovalent metal cation or an organic cation, the resulting derivatives are improved in solubility in water. On the other hand, when X is a hydrocarbon group, i.e. the resulting derivatives are esters, the derivatives may have a controlled HLB balance which depends on the number of carbon atoms in the hydrocarbon group. As a result, the derivatives can be controlled so that they have an optimum affinity for an intended cosmetic substrate. In addition, the HLB balance can appropriately be controlled by suitably changing the substituents, Y and 2 , in the general formula (I). For instance, when a hydroxyl group or a polyethyleneoxide group is introduced, the resulting derivatives become soluble in water. The introduction of an alkoxy

group or a polypropylene-oxide group results in an increase of fat solubility.

The UV-A and UV-B absorber according to the second embodiment of the invention is prepared, for example, by the following procedure. First, a volatile acid is added to an aqueous dispersion of a hydrous metal oxide to obtain a colloidal dispersion. Then, an aqueous solution of a dibenzoylmethane derivative is gradually added to the dispersion at room temperature and agitated. Alternatively, a powder or oil of the derivatives may be gradually added to the dispersion at room temperature and agitated. The resulting aqueous dispersion of the UV absorber composition may be used as it is, or may be dialyzed and/or washed with water, dried and ground into pieces, if necessary.

In the above procedure, the pH of the composition is preferred to be in the range of from 1 to 7, under which the composition is transparent to visible light and has thus small hiding power, without involving any problems such as on use as cosmetics, e.g. whitening. The pH of the composition is preferably adjusted by the use of a volatile acid such as hydrochloric acid, nitric acid, acetic acid or formic acid. It should be noted, however, that hydrous metal oxides exhibit the buffer action in a certain pH range, depending upon the type of metal oxide. The amount of the acid is preferred to be in the range of from 0.001 to 5 moles per mole of the hydrous metal oxide. Within the above-defined pH range of from 1 to 7, hydrous metal oxides are dispersed in colloidal form, with a colloid particle size of from 0.001 to 10 μm . In this condition, the colloid particles are positively charged on the surface thereof. The rate of travel of the colloid particles is preferred to be in the range of from 10 to 70 $\times 10^5$ cm/second.

In the UV-A absorber composition of the invention, the amount of dibenzoylmethane derivatives (I) may vary depending on the type of composition and is not critically determined, but an effective amount is necessary. In general, the derivative is used in an amount of from 0.1 to 20 wt% (hereinafter referred to simply as %), preferably from 0.5 to 10%.

In the composition of the second embodiment in which a hydrous oxide of a polyvalent metallic element is further contained, the amount of the dibenzoylmethane derivative (I) may also vary depending on the type of final product and is not critically limited. In general, the amount is in the range of from 0.1 to 20%, preferably from 0.5 to 10%, of the total composition. The ratio by weight of the hydrous metal oxide to the dibenzoylmethane derivative is preferably in the range of from 0.01 to 100:1, most preferably from 0.5 to 10:1.

The long wavelength ultraviolet ray absorbers of the invention may be used in cosmetics. For the preparation of cosmetics, a dibenzoylmethane derivative (I) and/or a hydrous oxide of a polyvalent metallic element which has good affinity for a selected cosmetic substrate is first provided, followed by adding to the cosmetic substrate by a usual manner to obtain a cream, solution, oil, spray, stick, emulsion, foundation, or ointment.

In other words, the dibenzoylmethane derivative (I) and/or the hydrous oxide is suitably selected depending on the type of cosmetic substrate. In doing so, various types of cosmetics having the UV-A absorbance and/or UV-B absorbance can be prepared, including base cosmetics such as cosmetic oils using oil substrates, oily creams or oil rich emulsions, water rich creams or emulsions, and water-base lotions, foundations using oils as substrates, and makeup cosmetics such as lip sticks. The substrates and solvents suitable for the above purposes include, for example: hydrocarbons such as solid or liquid paraffin, (crystal oil), ceresin, ozokerite and montan wax, plant oils or animal fats, oils and waxes such as olive oil, earth wax, carnauba wax, lanolin and spermaceti; fatty acids and esters thereof such as stearic acid, palmitic acid, oleic acid, glycerine monostearate, glycerine distearate, glycerine monooleate, isopropyl myristate, isopropyl stearate and butyl stearate; and alcohols such as ethyl alcohol, isopropyl alcohol, cetyl alcohol, stearyl alcohol, palmityl alcohol and hexyldodecyl alcohol. Additionally, polyvalent alcohols such as glycol, glycerine or sorbitol may also be used.

The cosmetics may further contain effective UV-B absorbers, if desired. Examples of such other UV-B absorbers include p-methylbenzylidene-D,L-camphor and sodium sulfonate thereof, sodium 2-phenylbenzimidazole-5-sulfonate, sodium 3,4-dimethylphenylglyoxylate, 4-phenylbenzophenone, isooctyl 4-phenylbenzophenone-2'-carboxylates, p-methoxycinnamates, 2-phenyl-5-methylbenzoxazole and p-dimethylaminobenzoates.

The cosmetics of the invention may comprise, aside from the above ingredients, various additives. Appropriate additives include, for example, W/O type or O/W type emulsifiers. Such emulsifiers may be any commercially available emulsifiers. Thickeners such as methyl cellulose, ethyl cellulose, carboxymethyl cellulose, polyacrylic acid, gum tragacanth, agar-agar and gelatin may be used as additives. If necessary, perfumes, preservatives, humectants, emulsification stabilizers, medical ingredients and/or physiologically acceptable colorants may be added.

As described hereinabove, the long wavelength UV absorbers of the invention can effectively absorb ultraviolet rays which will cause such harmful action as described before, and thus protect the skin from

sunburning and are stable against light and heat over a long term. The UV absorbers of the present invention are not irritant, not percutaneously absorbable and are not easily removable by water or by sweat. In other words, the composition shows continued effects.

The present invention is described in more detail by way of examples.

Test Example 1

Creams comprising 2% of dibenzoylmethane derivatives according to the invention were used to determine the effect of the derivatives serving to protect the skin from UV-A irradiation. In this test, a cream composition of Example 4 was used except that there were used, instead of 4'-methoxydibenzoylmethane-2-carboxylic acid, sodium 4'-methoxydibenzoylmethane-2-carboxylate (product 1 of the invention), sodium 3',4'-bis(polyoxyethyleneoxy)dibenzoylmethane-3-carboxylate ($n_1 = 3$, which indicates the moles of added oxyalkylene group, product 2 of the invention), sodium 4'-methoxy-3'-polyoxyethyleneoxydibenzoylmethane-4-carboxylate ($n_1 = 4$, product 3 of the invention), isostearyl 4'-methoxy-4-methyldibenzoylmethane-2-carboxylate (product 4 of the invention), methyl 3',4'-bis(polyoxypropyleneoxy)-dibenzoylmethane-2-carboxylate ($n_1 = 3$, product 5 of the invention), and isopropyl 4'-palmitoyldibenzoylmethane-4-carboxylate (product 6 of the invention). The test was carried out by the method as described by Gschnait et al. [Archives of Dermatological Research 263, 181-188 (1978)]. More particularly, guinea pigs were each shaved on the back thereof to expose the skin, followed by peritoneal administration of 8-methoxypsoralen in order to enhance susceptibility to UV-A. Subsequently, the exposed skin was applied with each of the products 1 to 6 of the invention in an amount of 2 mg/cm², followed by UV-A irradiation 15 minutes after the application. Twenty four hours after completion of the irradiation, the skin was visually observed to determine whether or not erythema appeared, and a minimum irradiation time before appearance of erythema on the skin was determined. This time was compared with a minimum UV-A irradiation time before erythema appeared on non-applied skin, and a sunlight protecting factor (hereinafter referred to simply as SPF) was calculated according to the following equation to determine the skin protection effect of each compound. The results are shown in Fig. 1. For comparison, a cream base of Example 4 (comparative product 1), and vaseline (comparative product 2) were used.

$$\text{SPF} = \frac{\text{Minimum UV-A irradiation time before erythema appears on the skin applied with a product of the invention}}{\text{Minimum UV-A irradiation time before erythema appears in the non-applied skin}}$$

From the results of this test, it will be seen that with the application of the cream base alone and vaseline, the skin cannot be protected from UV-A, whereas the creams comprising 2% of the dibenzoylmethane derivatives of the invention exhibit SPF values as high as 5 to 6, thus serving to effectively protect the skin from the UV-A light.

Example 1

Lotion:

The following ingredients were mixed as usual to obtain a lotion.

Formulation:

	Sodium 3',4'-bis(polyoxyethyleneoxy)- dibenzoylmethane-3-carboxylate ($n_1=3$)	2.0 wt%
5	Glycerine	2.0
	Propylene glycol	5.0
	Ethanol	10.0
10	Polyoxyethylene (20) hexadecyl ether	1.0
	Purified water	79.0
15	Preservative	suitable amount
	Perfume	suitable amount
	Colorant	very small amount

Example 2

25 Milky Lotion:

The following ingredients were mixed as usual to obtain a milky lotion.

Formulation:

30	Potassium 4'-methoxy-3'-polyoxyethyleneoxy- dibenzoylmethane-4-carboxylate ($n_1=4$)	5.0 wt%
	Stearic acid	1.5
	Cetanol	1.0
35	Oleic monoglyceride	0.5
	Vaseline	2.0
40	Liquid paraffin	6.0
	Squalane	6.0
	Purified water	69.0
45	Triethanolamine	0.8
	Dipropylene glycol	5.0
50	Glycerine	2.0
	Carboxyvinyl polymer	0.2
	Preservative	suitable amount
55	Perfume	"

Example 3

Milky Lotion:

The following ingredients were mixed as usual to obtain a milky lotion.

Formulation:

5	Isostearyl 4'-methoxy-4-methyldibenzoyl-methane-2-carboxylate	1.5 wt%
10	Stearic acid	2.4
	Stearyl alcohol	0.5
	Cetanol	1.0
15	Isopropyl myristate	10.0
	Liquid paraffin	5.0
20	Vaseline	2.0
	Purified water	71.0
	Triethanolamine	1.0
25	Glycerine	5.0
	Preservative	suitable amount
30	Perfume	suitable amount

Example 4

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Cream:

The following ingredient were mixed as usual to obtain a cream.

Formulation:

40	4'-Methoxybenzoylmethane-2-carboxylic acid	2.0 wt%
	Polyoxyethylene (20) sorbitan monooleate	2.0
45	Sorbitan monooleate	2.0
	Isopropyl myristate	18.0
50	Squalane	7.0

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	2-Ethylhexanoic triglyceride	12.0
	Antioxidant	suitable amount
5	Purified water	51.0
	Propylene glycol	5.0
	Preservative	suitable amount
10	Perfume	"

15 Example 5

Oily Foundation:

The following ingredients were mixed as usual to obtain an oily foundation.

20 Formulation:

	Methyl 3',4'-bis(polyoxypropyleneoxy)- dibenzoylmethane-2-carboxylate ($n_1=3$)	5.0 wt%
25	2-Ethylhexyl p-methoxycinnamate	5.0
	Liquid paraffin	15.0
	Isopropyl palmitate	16.0
30	Lanoline alcohol	3.0
	Microcrystalline wax	7.0
35	Ozokerite	8.0
	Candelilla wax	0.5
	Preservative	suitable amount
40	Antioxidant	"
	Perfume	"
45	Titanium oxide	15.0
	Kaolin	15.0
50	Talc	6.0
	Blood red	1.5
	Yellow iron oxide	2.0
55	Black iron oxide	0.5

Example 6

Lip Stick:

The following ingredients were mixed as usual to obtain a lip stick.

Formulation:

	Isopropyl 4'-palmitoyldibenzoylmethane-4-carboxylate	2.0 wt%
10	p-Dimethylaminobenzoic acid	2.0
	Castor oil	41.5
15	Hexadecyl alcohol	25.0
	Lanoline	4.0
	Beeswax	5.0
20	Candelilla wax	7.0
	Carnauba wax	6.0
25	Antioxidant	suitable amount
	Preservative	"
	Titanium oxide	2.0
30		
	Red No. 202	0.5
	Red No. 204	2.5
35	Red No. 227 Al lake	2.5
	Orange No. 201	0.2
40	Perfume	suitable amount

Test Example 2

Long wavelength ultraviolet light absorbers were prepared using 4'-methoxydibenzoylmethane-2-carboxylic acid and a salt thereof and alpha-alumina monohydrate according to the invention, and the absorption effect thereof was determined.

Alpha-alumina monohydrate was prepared by hydrolyzing aluminum triisopropoxide and adjusted in pH with hydrochloric acid, followed by mixing it with 4'-methoxydibenzoylmethane-2-carboxylic acid (MDMC) or a sodium salt thereof (SMDMC) in different weight ratios. The mixtures were each dispersion in water to determine an absorbance, from which a molal absorption factor of each UV absorber was determined. The results are shown in Figs. 2 and 3, in which the abscissa axis indicates a ratio by weight of alpha-alumina monohydrate to an ultraviolet absorber and the ordinate axis indicates the molal absorption factor. In the figures, the a indicates an absorption in an UV-A region at about 360 nm; and curve b indicates an absorption at about 290 nm, i.e. an absorption in an UV-B region. Fig. 4 shows a variation of the molal absorption factor in relation to pH of an absorber system using sodium 4'-dibenzoylmethane-carboxylate and alpha-alumina monohydrate. In the figure, curves e, g indicate absorptions in the UV-A region, and

curves f, h indicate absorptions in the UV-B region. Moreover, curves e, f are the case where a ratio by weight of alpha-alumina monohydrate was kept at 1:1, whereas curves g, h are the case where the weight ratio of alpha-alumina monohydrate was changed from 0 to 4.2:1.

The absorptions in the UV-A and UV-B regions can be arbitrarily changed by changing the weight ratio of the dibenzoylmethane derivatives and the polyvalent hydrous metal oxide as will be clearly seen from Figs. 2 and 3 and also by changing the pH of the system as will be seen from Fig. 4. Thus, if the weight ratio is defined in a certain range and/or the pH of the system is also defined, it becomes possible to effectively protect the skin from a total range of ultraviolet rays which give adverse influences on the skin.

Not only alpha-alumina monohydrate, but also alumina hydrate having the formula, $\text{Al}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ in which $x = 0.5$ to 3 can change the absorption in the UV-A and UV-B regions when mixed with the dibenzoylmethane derivatives. In Fig. 5, there are shown variations in absorbance of sodium 4'-methoxydibenzoylmethane-2-carboxylate, and mixtures of the carboxylate, with aluminum hydroxide or commercially available alumina ($\text{Al}_2\text{O}_3 \cdot x\text{H}_2\text{O}$) in a ratio by weight of 1:1. For the figure, it will be appreciated that the mixtures have similar absorptions in both UV-A and UV-B regions, and are thus useful in UV protection over a whole range of UV rays.

The result of a test using, as the hydrous metal oxide, hydrous iron oxide and hydrous titanium oxide are shown in Fig. 6. From the results, it will be seen that when these hydrous metal oxides are used in combination with a dibenzoylmethane derivative (i.e. sodium 4'-methoxydibenzoylmethane-2-carboxylate) in varying weight ratios, the absorbance is also varied. The variation in absorbance of a composition having a ratio by weight of the hydrous metal oxide and the derivative at 1:1 is as shown in Fig. 7, revealing that a whole range of ultraviolet rays which are harmful against the skin can be shut out.

In order to check the UV-protective effect in more detail, a dibenzoylmethane derivative and a hydrous polyvalent metal oxide were added to an O/W-type emulsion by a usual manner to determine an absorbance. The measurement of the absorbance was effected as follows: the emulsion was applied onto a transparent quartz plate in a thickness of 30 μm , and subjected to spectrophotometer. The results are shown in Fig. 8, from which it will be seen that the whole range of UV rays which give harmful influences on the skin can be effectively protected.

Moreover, the emulsion comprising 4 wt% of an UV absorber composition according to the invention was used to determine the effect of protecting the skin from UV-A and UV-B irradiation. The UV-A protective effect was determined as follows: guinea pigs which had been peritoneally dosed with 8-methoxypsoralen and were thus sensitive UV-A were applied with the emulsion comprising the absorber of the invention. A minimum energy necessary for development of erythema one day after the UV-A irradiation was determined and compared with the case where no emulsion was applied. On the other hand, the UV-B protective effect was determined as follows: the emulsion comprising the absorber of the invention was applied to healthy men on the back thereof, and a minimum energy required for development of erythema one day after the UV-B irradiation, and compared with the case where no emulsion was applied. The comparison was made according to the following equation concerning sun protective factor (SPF).

Minimum energy necessary for developing
erythema on the skin applied with the
emulsion comprising an absorber of the
invention

SPF = $\frac{\text{Minimum energy necessary for developing erythema on the one-applied skin}}{\text{Minimum energy necessary for developing erythema on the skin applied with the emulsion comprising an absorber of the invention}}$

Minimum energy necessary for developing
erythema on the one-applied skin

The formulations of the O/W-type emulsions are shown in Table 1 with the results shown in Table 2.

Table 1

	A (wt%)	B (wt%)
Stearic acid	2.0	2.0
Cetyl alcohol	1.0	1.0
Liquid paraffin	20.0	20.0
Polyoxyethylene(20) sorbitan stearate	1.0	1.0
Stearic monoglyceride	1.0	1.0
Alpha-Al ₂ O ₃ ·H ₂ O	4.0	-
Sodium 4'-methoxydibenzoylmethane-2-carboxylate	4.0	-
Propylene glycol	5.0	5.0
Purified water	balance	balance
Perfume, preservative	suitable amounts	suitable amounts

Table 2

	SPF	
	UV-A	UV-B
Emulsion A	10.8	3.9
Emulsion A	0.8	0.9

Test Example 3

The stability of the present invention was determined as follows: an aqueous dispersion comprising 0.5 wt% of a composition of the invention was applied onto a transparent quartz plate in a thickness of 30 μ m and subjected to measurement of an absorbance. In addition, an accelerated test was effected in such a way that a decomposition rate at 50° C after 55 days was determined from a rate of reduction of the absorbance at about 360 nm. The results are shown in Table 3. Dibenzoylmethane derivatives having, for example, carboxyl groups therein are soluble in water and when such derivatives are dissolved in water and particularly under basic conditions, they undergo hydrolysis. However, when such derivatives are used in

combination with hydrous oxides of polyvalent metallic elements, they become more stable. No discoloration such as blackening by application of light took place when the derivatives were exposed to sunlight.

Table 3

	Rate of Decomposition (50°C, 55 days)
Sodium 4'-methoxydibenzoylmethane-2-carboxylate + alpha $\text{Al}_2\text{O}_3 \cdot \text{H}_2\text{O}$	3 %
Arginine dibenzoylmethane-2-carboxylate	70 %
Arginine 4'-methylbenzoylmethane-2-carboxylate	63 %
Potassium 4'-methoxydibenzoylmethane-2-carboxylate	48 %
Arginine 4'-ethoxydibenzoylmethane-2-carboxylate	59 %

Example 7

Lotion:

Formulation:

Sodium 3'-hydroxy-4'-methoxydibenzoylmethane-2-carboxylate	3%
Alpha-alumina monohydrate	3%
Glycerine	4%
Polyoxyethylene hardened castor oil	2%
Ethanol	10%
Preservative	suitable amount
Perfume	very small amount
Colorant	"
Purified water	balance

Example 8**Milky Lotion:****Formulation:**

5	4'-Methoxy-3'-polyoxyethyleneoxydibenzoyl-methane-4-carboxylic acid	3%
10	Alumina monohydrate	3%
	Stearic acid	1%
15	Cetanol	2%
	Vaseline	2.5%
	Squalane	4%
20	Hardened palm oil	2%
	Polyoxyethylene (20) sorbitan monostearate	1.5%
25	Oleophilic glycerine monostearate	1.2%
	Glycerine	3%
30	Potassium hydroxide	0.2%
	Carboxyvinyl polymer	0.2%
	Preservative	suitable amount
35	Perfume	very small amount
	Purified water	balance

Example 9**O/W-type Cream:**

45	Arginine 4'-methoxydibenzoylmethane-2-carboxylate	5%
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5	Alumina monohydrate	5%
	Stearic acid	2%
	Cetanol	4%
	Vaseline	5%
10	Squalane	8%
	Hardened palm oil	4%
15	Polyoxyethylene (20) sorbitan monostearate	2%
	Sorbitan monostearate	2%
	Glycerine	5%
20	Preservative	suitable amount
	Perfume	very small amount
25	Purified water	balance

Example 10

30 Foundation Stick:

Formulation:

35	Sodium 4'-polyoxyethylenoxydibenzoyl-methane-4-carboxylate	5%
	Titanium oxide hydrate	5%
40	Hydrophobic titanium oxide	10%
	Kaolin	15%
	Talc	10%
45	Liquid paraffin	18%
	Isopropyl myristate	15%

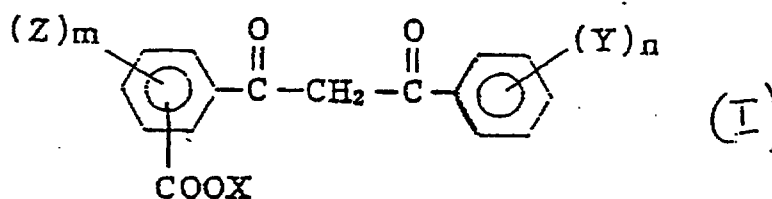
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Liquid lanoline	4.5%
Microcrystalline wax	4.5%
Ceresin wax	10%
Carnauba wax	2%
Sorbitan sesquioleate	1%

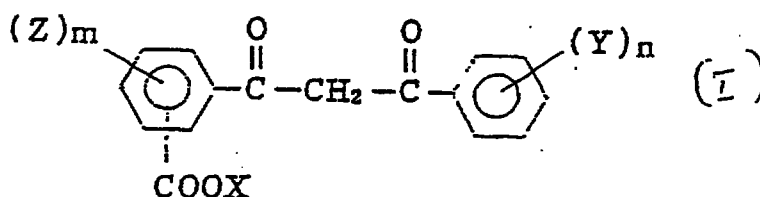
15 Claims

1. A cosmetic composition comprising a dibenzoylmethane derivative of the general formula (I)



in which X represents a hydrogen atom, a monovalent metal cation, an organic cation, a linear or branched alkyl group having from 1 to 24 carbon atoms, or a polyoxyalkylene oxide group having from 1 to 10 alkylene oxide units with 2 or 3 carbon atoms in each unit, each Y and each Z independently represent a hydroxyl group, a linear or branched alkyl group having from 1 to 24 carbon atoms, or an alkoxy group having from 1 to 24 carbon atoms, or a polyoxyalkylene oxide group having from 1 to 10 alkylene oxide units with 2 or 3 carbon atoms in each unit, and m and n are independently an integer of from 0 to 3 as a long wavelength ultraviolet ray absorber, and the usual cosmetically acceptable ingredients.

2. A cosmetic composition comprising a dibenzoylmethane derivative of the general formula (I)



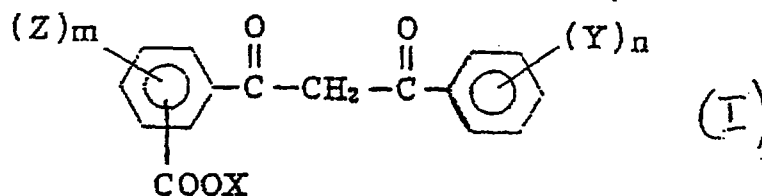
in which X represents a hydrogen atom, a monovalent metal cation, an organic cation, a linear or branched alkyl group having from 1 to 24 carbon atoms, or a polyoxyalkylene oxide group having from 1 to 10 alkylene oxide units with 2 or 3 carbon atoms in each unit, each Y and each Z independently represent a hydroxyl group, a linear or branched alkyl group having from 1 to 24 carbon atoms, or an alkoxy group having from 1 to 24 carbon atoms, or a polyoxyalkylene oxide group having from 1 to 10 alkylene oxide units with 2 or 3 carbon atoms in each unit, and m and n are independently an integer of from 0 to 3, a hydrous oxide of a polyvalent metallic element, and other usual cosmetically acceptable ingredients.

3. Use of a dibenzoylmethane derivative of the general formula (I)

Liquid lanoline	4.5%
Microcrystalline wax	4.5%
Ceresin wax	10%
Carnauba wax	2%
Sorbitan sesquioleate	1%

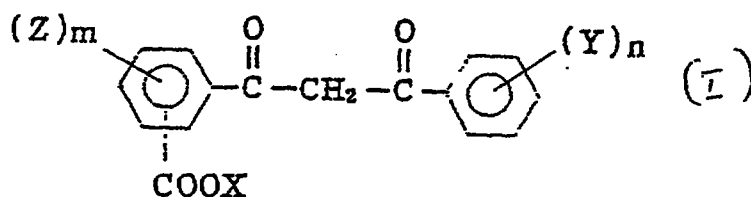
15 Claims

1. A cosmetic composition comprising a dibenzoylmethane derivative of the general formula (I)



in which X represents a hydrogen atom, a monovalent metal cation, an organic cation, a linear or branched alkyl group having from 1 to 24 carbon atoms, or a polyoxyalkylene oxide group having from 1 to 10 alkylene oxide units with 2 or 3 carbon atoms in each unit, each Y and each Z independently represent a hydroxyl group, a linear or branched alkyl group having from 1 to 24 carbon atoms, or an alkoxy group having from 1 to 24 carbon atoms, or a polyoxyalkylene oxide group having from 1 to 10 alkylene oxide units with 2 or 3 carbon atoms in each unit, and m and n are independently an integer of from 0 to 3 as a long wavelength ultraviolet ray absorber, and the usual cosmetically acceptable ingredients.

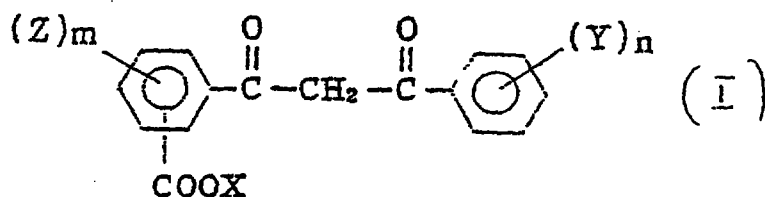
2. A cosmetic composition comprising a dibenzoylmethane derivative of the general formula (I)



in which X represents a hydrogen atom, a monovalent metal cation, an organic cation, a linear or branched alkyl group having from 1 to 24 carbon atoms, or a polyoxyalkylene oxide group having from 1 to 10 alkylene oxide units with 2 or 3 carbon atoms in each unit, each Y and each Z independently represent a hydroxyl group, a linear or branched alkyl group having from 1 to 24 carbon atoms, or an alkoxy group having from 1 to 24 carbon atoms, or a polyoxyalkylene oxide group having from 1 to 10 alkylene oxide units with 2 or 3 carbon atoms in each unit, and m and n are independently an integer of from 0 to 3, a hydrous oxide of a polyvalent metallic element, and other usual cosmetically acceptable ingredients.

3. Use of a dibenzoylmethane derivative of the general formula (I)



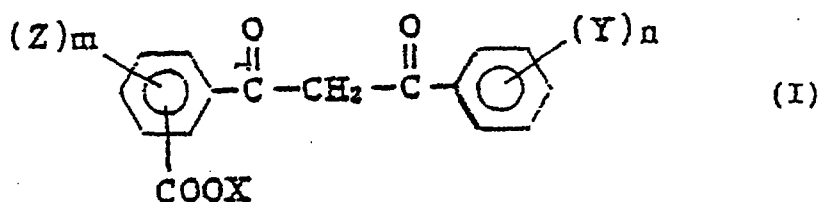


wherein X, Y, Z, m, and n are as defined in claim 1 for shielding the skin from UV rays.

Revendications

1. Composition cosmétique comprenant :

- un dérivé de dibenzoylméthane de la formule générale (I) :

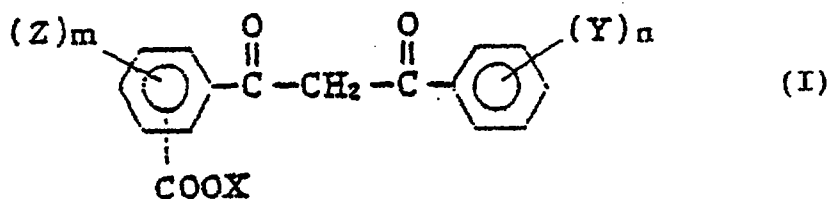


dans laquelle :

- X représente un atome d'hydrogène, un cation métallique monovalent, un cation organique, un groupe alkyle linéaire ou ramifié ayant de 1 à 24 atomes de carbone, ou un groupe polyoxyalkylène oxyde ayant de 1 à 10 unités oxyde d'alkylène avec 2 ou 3 atomes de carbone dans chaque unité ;
- chaque Y et chaque Z représentent indépendamment un groupe hydroxyle, un groupe alkyle linéaire ou ramifié ayant de 1 à 24 atomes de carbone, ou un groupe alcoxy ayant de 1 à 24 atomes de carbone, ou un groupe polyoxyalkylène oxyde ayant de 1 à 10 unités oxyde d'alkylène avec 2 ou 3 atomes de carbone dans chaque unité ; et
- m et n représentent indépendamment un nombre entier allant de 0 à 3, en tant qu'agent absorbant le rayonnement ultraviolet de grande longueur d'onde ; et
- les ingrédients habituels, cosmétiquement acceptables.

2. Composition cosmétique comprenant :

- un dérivé de dibenzoylméthane de la formule générale (I) :



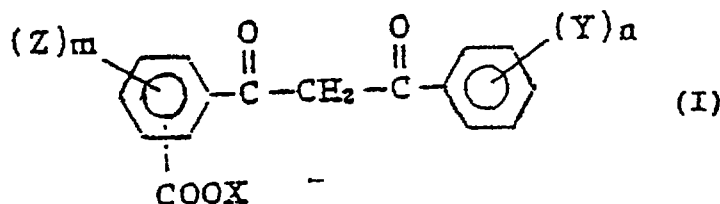
dans laquelle :

- X représente un atome d'hydrogène, un cation métallique monovalent, un cation organique, un groupe alkyle linéaire ou ramifié ayant de 1 à 24 atomes de carbone, ou un groupe polyoxyalkylène oxyde ayant de 1 à 10 unités oxyde d'alkylène avec 2 ou 3 atomes de carbone dans chaque unité ;
- chaque Y et chaque Z représentent indépendamment un groupe hydroxyle, un groupe alkyle linéaire ou ramifié ayant de 1 à 24 atomes de carbone, ou un groupe alcoxy ayant de 1 à 24 atomes de carbone, ou un groupe polyoxyalkylène oxyde ayant de 1 à 10 unités oxyde d'alkylène avec 2 ou 3 atomes de carbone dans chaque unité ; et
- m et n représentent indépendamment un nombre entier allant de 0 à 3 ; - un oxyde hydraté d'un



élément métallique polyvalent ; et
- d'autres ingrédients habituels cosmétiquement acceptables.

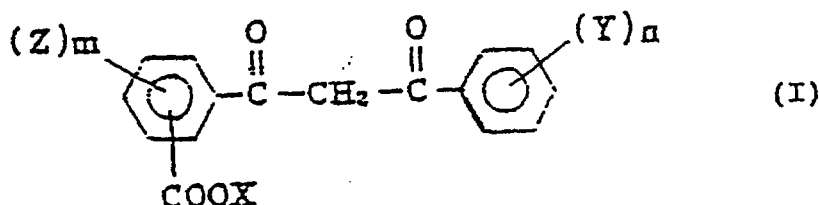
3. Utilisation d'un dérivé de dibenzoylméthane de la formule générale (I) :



dans laquelle X, Y, Z, m et n sont tels que définis à la revendication 1, pour protéger la peau des rayons UV.

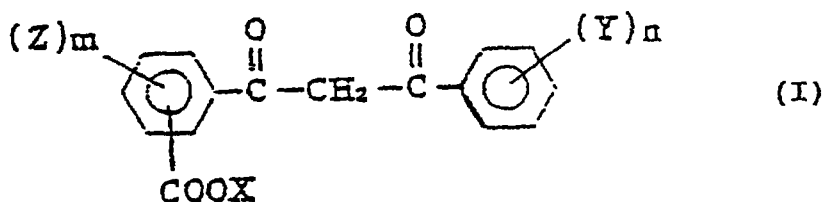
Ansprüche

1. Kosmetische Zusammensetzung, umfassend ein Dibenzoylmethanderivat der allgemeinen Formel (I)



in der X für ein Wasserstoffatom, ein monovalentes Metallkation, ein organisches Kation, eine lineare oder verzweigte Alkylgruppe mit 1 bis 24 Kohlenstoffatomen oder eine Polyoxyalkylenoxidgruppe mit 1 bis 10 Alkylenoxideinheiten mit 2 oder 3 Kohlenstoffatomen in jeder Einheit steht, jedes Y und jedes Z unabhängig für eine Hydroxylgruppe, eine lineare oder verzweigte Alkylgruppe mit 1 bis 24 Kohlenstoffatomen oder eine Alkoxygruppe mit 1 bis 24 Kohlenstoffatomen oder eine Polyoxyalkylenoxidgruppe mit 1 bis 10 Alkylenoxideinheiten mit 2 oder 3 Kohlenstoffatomen in jeder Einheit steht und m und n unabhängig eine ganze Zahl von 0 bis 3 bedeuten, als Absorptionsmittel für langwellige Ultraviolettstrahlen, und die üblichen kosmetisch akzeptablen Bestandteile.

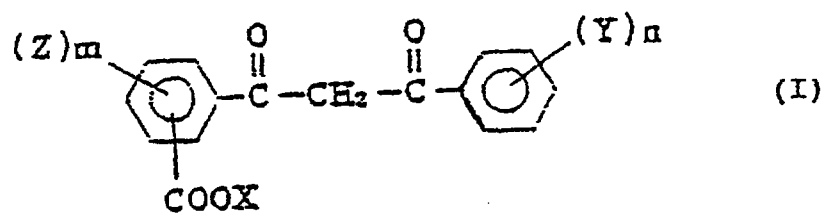
2. Kosmetische Zusammensetzung, umfassend ein Dibenzoylmethanderivat der allgemeinen Formel (I)



in der X für ein Wasserstoffatom, ein monovalentes Metallkation, ein organisches Kation, eine lineare oder verzweigte Alkylgruppe mit 1 bis 24 Kohlenstoffatomen oder eine Polyoxyalkylenoxidgruppe mit 1 bis 10 Alkylenoxideinheiten mit 2 oder 3 Kohlenstoffatomen in jeder Einheit steht, jedes Y und jedes Z unabhängig für eine Hydroxylgruppe, eine lineare oder verzweigte Alkylgruppe mit 1 bis 24 Kohlenstoffatomen oder eine Alkoxygruppe mit 1 bis 24 Kohlenstoffatomen oder eine Polyoxyalkylenoxidgruppe mit 1 bis 10 Alkylenoxideinheiten mit 2 oder 3 Kohlenstoffatomen in jeder Einheit steht und m und n unabhängig eine ganze Zahl von 0 bis 3 sind, ein wässriges Oxid eines polyvalenten metallischen Elements und andere übliche kosmetisch akzeptable Bestandteile.

3. Verwendung eines Dibenzoylmethanderivats der allgemeinen Formel (I)





10 wobei X, Y, Z, m und n wie in Anspruch 1 definiert sind, zur Abschirmung der Haut vor UV-Strahlen.

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