

PROCESS FOR THE PREPARATION OF ALKANEDIOL

The present invention relates to a process for the preparation of alkanediol from alkylene oxide.

Background of the invention

5 It is well known to contact alkylene oxide with carbon dioxide in the presence of a suitable catalyst to obtain a cyclic alkylene carbonate. Such process has been described for example in EP-A-119840. Further, it is known that the cyclic alkylene carbonate can be converted further. A conventional further process comprises
10 hydrolysis of the cyclic alkylene carbonate to produce a diol as is described in US-A-5,847,189.

US-A-6,407,279 teaches to integrate the above steps which makes that cyclic carbonate can be fed to the transesterification reaction zone without purification of
15 the cyclic carbonate. US-A-6,407,279 describes that the elimination of the cyclic carbonate purification step causes elimination of two vacuum distillation columns and gives a yield benefit by eliminating losses of cyclic carbonate attributable to the purification steps. The
20 carbonation reaction of US-A-6,407,279 is carried out at a pressure in the range from about 500 psia (about 34×10^5 N/m²) to about 1000 psia (about 69×10^5 N/m²) while the transesterification reaction is carried out at a pressure of about 100 psia (about 7×10^5 N/m²) to
25 300 psia (about 21×10^5 N/m²).

EP-A-0776890 describes a process for producing ethylene glycol from ethylene oxide wherein in a second step ethylene oxide is allowed to react with carbon dioxide and in a third step ethylene carbonate is

subjected to hydrolysis. The reaction solution obtained
after completion of the hydrolysis is introduced to a
carbon dioxide separator wherein the pressure is lowered.
The separated carbon dioxide can be recycled to the
5 carbonation reaction, which carbonation reaction is
carried out at higher pressure.

Summary of the invention

We have now found that an integrated process can be
improved further. The improvement eliminates the need to
10 increase the pressure of a gas or a mixture of gas and
liquid.

Increasing the pressure of fluids can be done in many ways. The means employed usually is centrifugal force or transfer of momentum. In the chemical industry, usually centrifugal pumps or compressors are used which are based on the principle of producing kinetic energy by the action of centrifugal force and then converting this energy into pressure by efficiently reducing the velocity of the flowing fluid. If transfer of momentum is used, one fluid is decelerated in order to transfer its momentum to a second fluid. Jets and eductors make use of transfer of momentum.

A common phenomenon in increasing the pressure of fluids is that increasing the pressure of a gas or a mixture of gas and liquid is more difficult than increasing the pressure of a liquid only. Apparatus for increasing the pressure of a gas or a mixture of gas and liquid have the disadvantage that they more easily fail in operation and are expensive.

The present invention makes it possible to operate while increasing the pressure of a liquid instead of having to increase the pressure of a gas or a mixture of gas and liquid.

The present invention relates to a process for the preparation of alkanediol, which process comprises:

- (a) contacting alkylene oxide with carbon dioxide in the presence of catalyst to obtain a first liquid reaction mixture containing cyclic carbonate,
- (b) optionally removing carbon dioxide and/or alkylene oxide,
- (c) increasing the pressure of the liquid reaction mixture obtained in step (a) and/or (b),
- (d) contacting the pressurized first reaction mixture obtained in step (c) with water in the presence of

catalyst to obtain a second reaction mixture containing alkanediol and carbon dioxide,

(e) separating the second reaction mixture into a liquid effluent and a gaseous effluent containing carbon dioxide, and

(f) recycling at least part of the gaseous effluent containing carbon dioxide to step (a), in which process the pressure in step (d) is higher than the pressure in step (a).

Detailed description of the invention

In process step (a) of the present invention, the alkylene oxide is contacted with carbon dioxide in the presence of catalyst. Several catalysts are known to be suitable for such process. Preferably, the catalyst is a homogeneous phosphorus containing catalyst. Well known phosphorus containing compounds which are suitable catalysts are phosphine compounds and phosphonium compounds. The catalyst preferably is a homogeneous phosphonium catalyst, more specifically a phosphonium halide catalyst. It was found especially advantageous to employ a tetraalkylphosphonium halide catalyst, more specifically a tributyl-methyl phosphonium iodide.

The catalyst can be either added as such or can be formed in-situ. The reaction mixture which is added to step (a), preferably is substantially free of water. The amount of water is preferably less than 5 %wt, more preferably less than 1 %wt, most preferably less than 0.1 %wt, based on total amount of reaction mixture added.

The alkylene oxide for use in the present invention preferably is propylene oxide. The use of propylene oxide starting compound in the present invention, makes that propylene carbonate is formed in step (a) which leads to the manufacture of 1,2-propanediol or monopropylene

glycol in step (b). In such process, the alkanediol is 1,2-propanediol and the alkylene oxide is propylene oxide.

5 The carbon dioxide can be either pure carbon dioxide
or carbon dioxide containing further compounds. Carbon
dioxide which is especially suitable for use in the
present invention, is carbon dioxide which has been
separated off in subsequent steps of the present process.
Carbon dioxide can either be separated off directly after
10 the alkylene oxide has reacted with carbon dioxide or at
a later stage.

Carbon dioxide is produced in the reaction of the
cyclic alkylene carbonate with water. Therefore, it is
especially attractive to separate carbon dioxide and
15 recycle the carbon dioxide thus obtained to step (a)
either as such or after having been purified. The extent
to which the carbon dioxide is purified depends on the
nature and the amounts of contaminants present in the
carbon dioxide. These again depend on the exact reaction
20 conditions and purification steps of the process.

Operating conditions for step (a) are well known in
the art. Preferred operating conditions will generally
comprise a temperature of from 50 to 200 °C, more
specifically 100 to 150 °C, and a pressure of at least
25 $2 \times 10^5 \text{ N/m}^2$, more specifically a pressure of from 2 to
 $100 \times 10^5 \text{ N/m}^2$, most specifically of from 3 to
 $40 \times 10^5 \text{ N/m}^2$. Preferably, process step (a) is carried
out at a pressure of from 10 to $30 \times 10^5 \text{ N/m}^2$.

30 The catalyst can be added to the reactor in any form
known to be suitable to someone skilled in the art.
Generally, the catalyst will be added as such or as a
solution of the catalyst preferably in an inert solvent
such as a cyclic carbonate or alkylene glycol. The

catalyst can be added either to the alkylene oxide or to the carbon dioxide or to the mixture of both. Preferably, the catalyst solution is added to the reactor containing the mixture of alkylene oxide and carbon dioxide.

5 The reaction mixture obtained in step (a) is preferably used without further purification in the manufacture of alkanediol. However, some purification of the reaction mixture can be carried out. It can be advantageous to remove carbon dioxide and/or alkylene
10 oxide from the first reaction mixture. The carbon dioxide and/or unconverted alkylene oxide which are separated off can be recycled to step (a). Separation of carbon dioxide and/or unconverted alkylene oxide can substantially
15 reduce the volume of the reaction mixture to be subjected to steps (c) and (d). Additionally, the removal of carbon dioxide and/or unconverted alkylene oxide reduces the risk of cavitation in a pump.

 In step (c), the pressure of the first reaction mixture obtained in step (a) and/or (b) is increased. The
20 increase in pressure can be done in any way known to someone skilled in the art. Generally, a pump will be used which is known to be suitable for this specific application such as a centrifugal pump.

 The pressurized liquid obtained in step (c) will
25 generally have a pressure of from 1 to $100 \times 10^5 \text{ N/m}^2$, more specifically of from 3 to $60 \times 10^5 \text{ N/m}^2$, more specifically of from 5 to $50 \times 10^5 \text{ N/m}^2$.

 In a preferred embodiment, the feed subjected to step (d) contains a substantial amount of alkylene oxide.
30 The alkylene oxide can be due to the fact that part of the alkylene oxide has not been converted in step (a) and/or to the fact that alkylene oxide is added either before or during step (d). Preferably, the amount of

alkylene oxide and cyclic carbonate in step (d) is such that the molar ratio of alkylene oxide to cyclic carbonate is of from 0.01:1 to 1:1, more preferably of from 0.02:1 to 0.6:1, more preferably of from 0.03:1 to 0.4:1, more preferably of from 0.04:1 to 0.3:1, more preferably of from 0.05:1 to 0.2:1. Most preferably, the molar ratio of alkylene oxide to cyclic carbonate is of from 0.08:1 to 0.15:1.

In the process according to the present invention, it is preferred that process step (a) is carried out with the help of a homogeneous catalyst and step (d) is carried out with the help of a heterogeneous catalyst. More preferably, the homogeneous catalyst present in step (a) is not removed until after step (d). This has the advantage that the homogeneous catalyst of step (a) can be active in the conversion of the alkylene oxide if alkylene oxide is still present in step (d). Removal of a limited amount of the homogeneous catalyst such as in distillation of part or all of the reaction mixture, will generally still leave sufficient homogeneous catalyst in the reaction mixture to serve its purpose in step (d) of the present process.

In step (d), at least part of the pressurized reaction mixture obtained in step (c) is hydrolysed. Although it is possible to process only part of the product of step (c), generally substantially all of the pressurized liquid obtained will be subjected to step (d).

In the present process, water is added in step (d) to the pressurized reaction mixture obtained in step (c). Although alcohol such as methanol can be added in addition to water, it is preferred to add water only. The presence of water only makes that a more limited number

of products is obtained which is advantageous in the subsequent separation of products.

5 The catalysts for use in the present invention are well known in the art. Preferably, the catalysts are heterogeneous catalysts. Examples of such catalysts comprise solid inorganic compounds such as alumina, silica-alumina, alumina carrying a copper compound, silica-alumina carrying a copper compound, silica-magnesia, aluminosilicate, gallium silicate, zeolites, 10 metal-exchanged zeolites, ammonium-exchanged zeolites, zinc on a support, lanthanum on a support, a mixture of aluminium and magnesium (hydr)oxide and ion-exchange resins.

15 Preferably, the catalyst employed in step (d) is chosen from the group consisting of a mixture of aluminium and magnesium (hydr)oxide, zinc on a support, lanthanum on a support and alumina. These catalysts will be described hereinafter in more detail.

20 The mixture of aluminium and magnesium (hydr)oxide preferably has a magnesium to aluminium molar ratio in the range of from 3 to 50, more preferably of from 4 to 20. In the preparation of the catalyst, generally a so-called mixed magnesium/aluminium hydroxide is formed. However, it might be that under working conditions mixed 25 magnesium/aluminium oxides are present. Our reference to a mixture of aluminium and magnesium (hydr)oxide covers both mixtures of aluminium and magnesium hydroxide and mixtures of aluminium and magnesium oxide and a combination of both mixtures. These mixtures were found 30 to give the highest activity at a molar ratio of more than 3, preferably more than 4. A preferred range was found to be of from 4 to 20, more specifically of from 5 to 15, most specifically of from 5 to 10. Preferred

catalysts are described in International application No. PCT/EP02/12640 (our TS 1067).

5 In another preferred embodiment of the present invention, the catalyst comprises a lanthanum compound on a support. A preferred catalyst comprises at least 7 %wt of lanthanum supported on a support. The lanthanum compound preferably is La_2O_3 or a precursor thereof. Under reaction conditions this lanthanum compound may be temporarily and/or reversibly converted due to the
10 reaction conditions into lanthanum hydroxide ($\text{La}(\text{OH})_3$), lanthanumoxyhydroxide ($\text{LaO}(\text{OH})$) and/or corresponding alcoholate species such as ($\text{La}(\text{OR})_3$ or $\text{LaO}(\text{OR})$).

As a support for the lanthanum containing catalyst any suitable support may be used. The support preferably
15 is substantially inert under the reaction conditions and is provided with sufficient mechanical strength. Potential supports comprise clay minerals, inorganic supports such as Al_2O_3 , SiO_2 , MgO , TiO_2 , ZrO_2 , ZnO and mixtures thereof. Other examples are a kaolinite, a
20 hallosyte, a chrysotile, a montmorillonite, a beidellite, a hectorite, a sauconite, a muscovite, a phlogopite, a biotite, a hydrotalcite and talc. Particularly preferred are the inorganic supports selected from the group consisting of Al_2O_3 , SiO_2 , MgO , TiO_2 , ZrO_2 , ZnO and
25 mixtures thereof.

The lanthanum containing catalyst preferably comprises at least 7 %wt of lanthanum, more specifically in the range of from 7 to 40 %wt of lanthanum based on total amount of catalyst. The lanthanum containing
30 catalyst may be produced using any suitable method. A preferred method comprises impregnating a support with a lanthanum containing salt, and subsequently drying and calcining the impregnated support. After impregnation the

impregnated support can be dried and subsequently calcined. Calcination is generally carried out at a calcination temperature from between 120 to 700 °C. The catalyst activity can be increased even further if the catalyst is calcined at a temperature in the range of from 350 to 600 °C. Preferred catalysts are described in PCT patent application PCT/EP02/12638 (our TS 1144).

A further catalyst which is especially suitable for use in step (d) of the present invention is a zinc supported catalyst. The support preferably is selected from the group consisting of Al_2O_3 , SiO_2 , MgO , TiO_2 , ZrO_2 , Cr_2O_3 , carbon and mixtures thereof. The zinc supported catalyst can be prepared by impregnation of silica, alumina or mixtures of aluminium and magnesium (hydr)oxide with a zinc nitrate solution. Preferably, the zinc supported catalysts comprise at least 15 %wt of zinc on a support having a surface area of at least 20 m^2/g , more preferably at least 40 m^2/g . Preferred catalysts are described in the patent applications claiming priority of European patent application No. 02256347.2 (our TS 1199, not yet published).

A further catalyst which is preferably used is a catalyst consisting of alumina. Preferably, the alumina is gamma-alumina.

Step (d) of the present process is preferably carried out at a temperature of from 50 to 300 °C, preferably of from 80 to 250 °C, more specifically of from 100 to 200 °C. The pressure can vary widely, and preferably is at least $1 \times 10^5 \text{ N/m}^2$, more specifically at least $2 \times 10^5 \text{ N/m}^2$. The pressure generally will be at most $100 \times 10^5 \text{ N/m}^2$, more specifically at most $70 \times 10^5 \text{ N/m}^2$, most specifically at most $60 \times 10^5 \text{ N/m}^2$. Preferably, the

pressure in step (d) is of from 5 to $50 \times 10^5 \text{ N/m}^2$, more specifically of from 15 to $35 \times 10^5 \text{ N/m}^2$.

The pressure at which step (d) is operated is higher than the pressure at which step (a) is operated.

5 Generally, the pressure in step (d) will be at least sufficient to compensate for any pressure drop in the process as a whole. Preferably, the pressure in step (d) will be at least $0.1 \times 10^5 \text{ N/m}^2$ higher than the pressure in step (a), more specifically at least $1 \times 10^5 \text{ N/m}^2$
10 higher, preferably at least $2 \times 10^5 \text{ N/m}^2$ higher. The pressure in step (d) generally will be at most $50 \times 10^5 \text{ N/m}^2$ higher than the pressure in step (a), more specifically at most $20 \times 10^5 \text{ N/m}^2$, more specifically at most $15 \times 10^5 \text{ N/m}^2$, most specifically at most
15 $10 \times 10^5 \text{ N/m}^2$.

In step (e), the second reaction mixture is separated into a liquid effluent and a gaseous effluent containing carbon dioxide. Such separation can be carried out by sending the second reaction mixture to an empty vessel.
20 The reaction mixture separates into a carbon dioxide rich gaseous phase and a liquid phase. The gaseous effluent containing carbon dioxide obtained in step (e) preferably is recycled directly to step (a), optionally after having been combined with fresh carbon dioxide. A direct recycle
25 is considered to be a recycle in which no further compounds are separated off. However, the gas which is recycled directly can have been subjected to treatments causing a change in temperature and/or pressure such as heat exchange with one or more other streams either of
30 the present process or of a different process.

The liquid effluent obtained in step (e) can be separated further. Preferably, the separation comprises

(g) separating the liquid effluent obtained in step (e) further into a gaseous effluent rich in carbon dioxide and a liquid effluent, (h) contacting with water the gaseous effluent rich in carbon dioxide obtained in step (g), and (i) sending the water used in step (h) to step (d). The separation of step (g) preferably is carried out at a pressure of from 0.1 to 20×10^5 N/m², more specifically of from 1 to 15×10^5 N/m², most specifically of from 1 to 2×10^5 N/m². Alkanediol can be separated from the liquid effluent obtained in step (g).

In step (f), at least part of the gaseous effluent containing carbon dioxide is recycled to step (a). The carbon dioxide can be recycled to step (a) without the need to increase its pressure. Advantageously, the process according to the present invention can be carried out without increasing the pressure of any of the streams present in step (a) to (f) besides the increase in pressure attained in step (c).

Preferably, alkanediol is separated from the second reaction mixture. The alkanediol can be separated from the reaction mixture obtained in step (d), (e) and/or (g) in any way known in the art. A preferred separation comprises distillation of the second reaction mixture, preferably by vacuum distillation, optionally followed by further distillation of one or more of the distillate fractions and/or bottom fractions. One or more of the fractions separated will have a high content of alkylene glycol. Alkylene glycol obtained by distillation will usually be sufficiently pure to use as such. If required, small amounts of by-products can be removed separately.

C L A I M S

1. Process for the preparation of alkanediol, which process comprises:
 - (a) contacting alkylene oxide with carbon dioxide in the presence of catalyst to obtain a first liquid reaction mixture containing cyclic carbonate,
 - (b) optionally removing carbon dioxide and/or alkylene oxide,
 - (c) increasing the pressure of the liquid reaction mixture obtained in step (a) and/or (b),
 - (d) contacting the pressurized first reaction mixture obtained in step (c) with water in the presence of catalyst to obtain a second reaction mixture containing alkanediol and carbon dioxide,
 - (e) separating the second reaction mixture into a liquid effluent and a gaseous effluent containing carbon dioxide, and
 - (f) recycling at least part of the gaseous effluent containing carbon dioxide to step (a),in which process the pressure in step (d) is higher than the pressure in step (a).
2. Process according to claim 1, in which process step (d) is carried out at a pressure of from 5 to $50 \times 10^5 \text{ N/m}^2$.
3. Process according to claim 1 and/or 2, in which process step (a) is carried out at a pressure of from 3 to $40 \times 10^5 \text{ N/m}^2$.
4. Process according to any one of claims 1 to 3, in which process the pressure in step (d) is at least $0.1 \times 10^5 \text{ N/m}^2$ higher than the pressure in step (a).

5. Process according to any one of claims 1 to 4, in which process step (a) is carried out with the help of a homogeneous catalyst and step (d) is carried out with the help of a heterogeneous catalyst.
- 5 6. Process according to any one of claims 1 to 5, in which process the gaseous effluent obtained in step (e) is recycled directly to step (a) optionally after having been combined with fresh carbon dioxide.
7. Process according to any one of claims 1 to 6, which
10 process further comprises separating alkanediol from the second reaction mixture.
8. Process according to any one of claims 1 to 7, which process further comprises:
- 15 (g) separating the liquid effluent obtained in step (e) further into a gaseous effluent rich in carbon dioxide and a liquid effluent,
- (h) contacting with water the gaseous effluent rich in carbon dioxide obtained in step (g), and
- (i) sending the water used in step (h) to step (d).
- 20 9. Process according to claim 8, which process further comprises separating alkanediol from the liquid effluent obtained in step (g).