

**PROCESS AND INTERMEDIATES FOR MAKING SUBSTITUTED ASPARTIC
ACID ACETALS**

Related Applications

This application claims priority to US Provisional Patent Application 60/199,329 filed April 24, 2000.

FIELD OF THE INVENTION

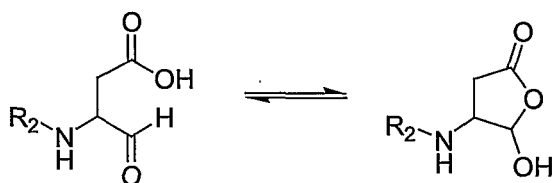
5 This invention relates to a process for the
synthesis of substituted aspartic acid acetals. The
process is useful for preparing biologically active
compounds, particularly certain caspase inhibitors, or
prodrugs thereof, such as inhibitors of interleukin-1 β
10 converting enzyme ("ICE").

BACKGROUND OF THE INVENTION

Caspases are a family of cysteine protease
enzymes that are key mediators in the signaling pathways
15 for apoptosis and cell disassembly (Thornberry, *Chem.*
Biol., 1998, **5**, R97-R103). Apoptosis, or programmed cell
death, is a principal mechanism by which organisms
eliminate unwanted cells. The deregulation of apoptosis,
either excessive apoptosis or the failure to undergo it,
20 has been implicated in a number of diseases such as
cancer, acute inflammatory and autoimmune disorders, and
certain neurodegenerative disorders (see generally
Science, 1998, **281**, 1283-1312; Ellis et al., *Ann. Rev.*
Cell. Biol., 1991, **7**, 663). Caspase-1, the first
25 identified caspase, is also known as interleukin-1 β
converting enzyme or "ICE." Caspase-1 converts precursor
interleukin-1 β ("pIL-1 β ") to the pro-inflammatory active

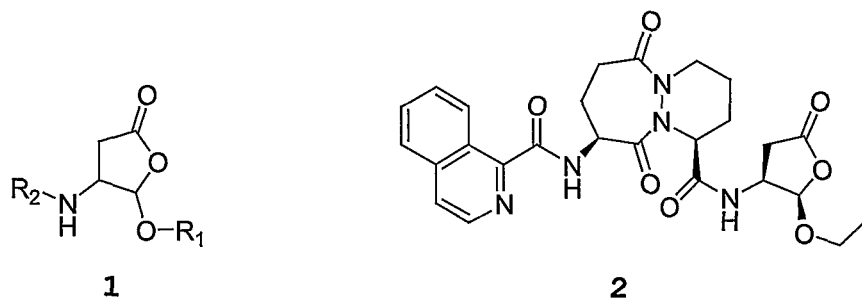
form by specific cleavage of pIL-1 β between Asp-116 and Ala-117. Besides caspase-1 there are also eleven other known human caspases which have been classified into families based on their biological function.

5 A number of useful caspase inhibitors has been reported that contain an aspartic acid aldehyde moiety, which will exist in equilibrium with its cyclic hemiacetal form as shown below:



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where R₂ represents the rest of the caspase inhibitor molecule. Based on the hemiacetal, orally available prodrugs of these inhibitors have been developed having the acetal structure **1**, where R₁ is alkyl or aralkyl, as exemplified by **2**. The ICE inhibitor **2** is a prodrug being developed as a treatment for rheumatoid arthritis (see US Patent 5,716,929).

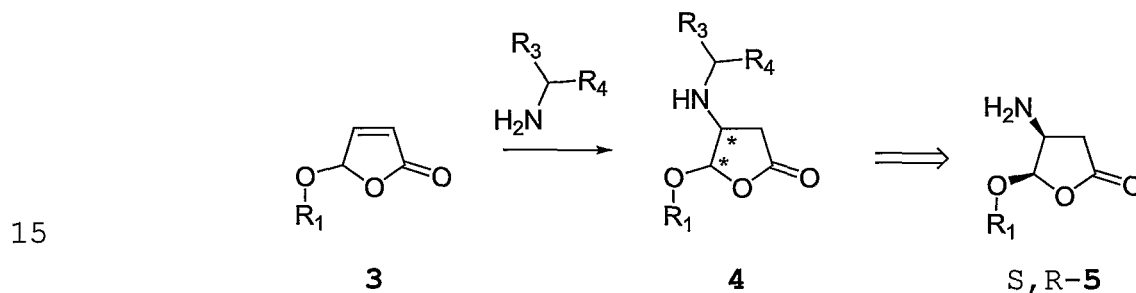


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A process for the preparation of a peptidic caspase inhibitor prodrug of formula **1** where R₁ is benzyl and R₂ is the amino acid sequence Ac-Y-V-A has been

described by Chapman et al. (*Bioorg. Med. Chem. Lett.* 1992, **2**(6), 613). However, this route has significant disadvantages, especially if one wishes to obtain chiral compounds. For example, the process requires an
5 expensive starting material and chromatographic separation of diastereomers (see discussion in PCT application WO/9903852).

More recently, a shorter process for the preparation of compounds of formula **1** where R₁ is ethyl
10 has been described (PCT patent application WO/9903852). The process involves the conjugate addition of an aralkylamine to an alkoxyfuranone **3** to provide diastereomeric compounds **4** as shown below:

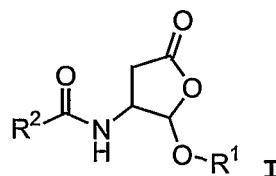


where R₃ is an alkyl group having one to four carbons and R₄ is an optionally substituted aryl group. The
20 diastereomers of **4**, or their addition salts, are reportedly separable by crystallization. The aralkyl group on the amine may then be removed by hydrogenolysis to liberate **5**, a useful synthetic intermediate for preparing caspase inhibitors. One limitation to this
25 approach is in the hydrogenolysis conditions used to remove R₃R₄CH- when R₁ is benzyl. Under such conditions, R₁ will also be removed.

It would be desirable to have a synthetic route to aspartic acetal caspase inhibitors, or prodrugs thereof, that is amenable to large-scale and overcomes the aforementioned shortcomings or otherwise improves upon the current methods.

DESCRIPTION OF THE INVENTION

This invention provides a process for making a compound of formula **I**:



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wherein R¹ is an optionally substituted group selected from an aliphatic group, aralkyl group, heterocyclalkyl group, or aryl group, and R² is an organic radical. The process is particularly useful for obtaining compounds **I** where R² is a P₂-P₄ moiety of a caspase inhibitor, or portion thereof.

Certain compounds of formula **I** are prodrugs of caspase inhibitors, particularly ICE inhibitors. R² is preferably any moiety that, when attached to the rest of the molecule of formula **I**, provides such an inhibitor. Portions of R² are specifically referred to in the art as a P₂, P₃ or P₄ moiety or site. These P_x terms are references to the amino acid sequence next to the aspartyl cleavage site of a particular caspase substrate. P₁ refers to the aspartyl residue of the substrate where caspase-induced cleavage occurs in the natural substrate. In the design of new, nonpeptidic caspase inhibitors, the P_x designation is often retained to show which portion of

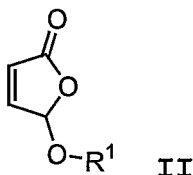
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the amino acid sequence has been replaced by the non-peptidic moiety. As used herein, the term "P₂-P₄" moiety refers to either the amino acid sequence described above or a chemical moiety known to replace such a sequence for the purpose of being a caspase substrate, and in particular an ICE substrate.

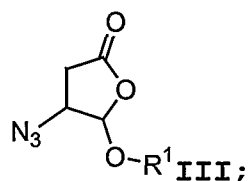
Examples of P₂-P₄ moieties that are non-peptidic are described in US 5,919,790 (Allen et al.); US 5,874,424 (Batchelor et al.); US 5,847,135 (Bemis et al.); US 5,843,904 (Bemis et al.); US 5,756,466 (Bemis et al.); US 5,716,929 (Bemis et al.); US 5,656,627 (Bemis et al.); WO 99/36426 (Warner-Lambert); Dolle et al., *J. Med. Chem.*, 40, 1941 (1997); WO 98/10778 (Idun); WO 98/11109 (Idun); WO 98/11129 (Idun) and WO 98/16502 (Warner Lambert), all of which are incorporated by reference.

One method of the present process for making **I**, referred to herein as Method A, comprises the steps of:
(a) providing a butenolactone of formula **II**:

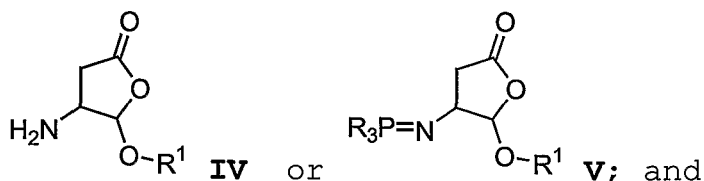


wherein R¹ is as described above;

(b) treating **II** with an azide N_3-Y , where Y is hydrogen, a silyl group, or a counterion, to form an azidolactone **III**:



(c) converting **III** to an aminolactone **IV** or an iminophosphorane **V**:



15 (d) coupling **IV** or **V** with R^2COOH or reactive equivalent thereof, to form **I**. It will be understood that the R^2 group may be selected from any organic radical that is stable to conditions of the coupling reaction, such as
20 those conditions described herein. Preferably R^2 is a P_2- , P_2-P_3- , or $P_2-P_3-P_4-$ moiety.

As used herein, the following definitions shall apply unless otherwise indicated. The terms "lactone" and "furanone" may be used interchangeably as will be
25 understood by one skilled in the art. The term "aliphatic" as used herein means straight chained, branched or cyclic C_1-C_{12} hydrocarbons which are completely saturated or which contain one or more units of unsaturation. For example, suitable aliphatic groups

include substituted or unsubstituted linear, branched or cyclic alkyl, alkenyl, alkynyl groups and hybrids thereof such as (cycloalkyl)alkyl, (cycloalkenyl)alkyl or (cycloalkyl)alkenyl. The term "alkyl" and "alkoxy" used
5 alone or as part of a larger moiety refers to both straight and branched chains containing one to twelve carbon atoms. The terms "alkenyl" and "alkynyl" used alone or as part of a larger moiety shall include both straight and branched chains containing two to twelve
10 carbon atoms. The term "aryl", used alone or as part of a larger moiety as in "aralkyl", refers to aromatic ring groups having five to fourteen members, such as phenyl, benzyl, 1-naphthyl, 2-naphthyl, 1-anthracyl and 2-anthracyl, and heterocyclic aromatic groups or heteroaryl
15 groups such as 2-furanyl, 3-furanyl, N-imidazolyl, 2-imidazolyl, 4-imidazolyl, 5-imidazolyl, 3-isoxazolyl, 4-isoxazolyl, 5-isoxazolyl, a 1,3,4-oxadiazolyl, a 1,2,4-oxadiazolyl, 2-oxadiazolyl, 5-oxadiazolyl, 2-oxazolyl, 4-oxazolyl, 5-oxazolyl, 2-pyrrolyl, 3-pyrrolyl, 2-pyridyl,
20 3-pyridyl, 4-pyridyl, 2-pyrimidyl, 4-pyrimidyl, 5-pyrimidyl, 3-pyridazinyl, 2-thiadiazolyl, 5-thiadiazolyl, 2-thiazolyl, 4-thiazolyl, 5-thiazolyl, 5-tetrazolyl, 2-triazolyl, 5-triazolyl, 2-thienyl, or 3-thienyl. The term "aryl ring" also refers to rings that are optionally
25 substituted. Aryl groups also include fused polycyclic aromatic ring systems in which a carbocyclic aromatic ring or heteroaryl ring is fused to one or more other rings. Examples include tetrahydronaphthyl, benzimidazolyl, benzothienyl, benzofuranyl, indolyl,
30 quinolinyl, benzothiazolyl, benzooxazolyl, benzimidazolyl, isoquinolinyl, isoindolyl, acridinyl,

benzoxisoxazolyl, and the like. Also included within the scope of the term "aryl", as it is used herein, is a group in which one or more carbocyclic aromatic rings and/or heteroaryl rings are fused to a cycloalkyl or non-
5 aromatic heterocyclic ring, for example, indanyl or tetrahydrobenzopyranyl. The term "heterocyclic group" refers to saturated and unsaturated monocyclic or polycyclic ring systems containing one or more heteroatoms and a ring size of three to eight such a
10 piperidinyl, piperazinyl, tetrahydrofuranlyl, pyrrolidinyl, tetrahydropyranyl, morpholinyl, and the like.

An aliphatic, alkyl, aryl, heterocyclic, or a carbocyclic group may contain one or more substituents.
15 The substituents are selected from those that will be stable under the reaction conditions of the present process, as would be generally known to those skilled in the art. Examples of substituents include halogen, -R, -OR, -OH, protected OH (such as acyloxy), phenyl (Ph),
20 substituted Ph, -OPh, substituted -OPh, -NO₂, -CN, -NHR, -N(R)₂, -NHCOR, -NHCONHR, -NRCONHR, -NHCON(R)₂, -NRCON(R)₂, -NRCOR, -NHCO₂R, -NRCO₂R, -CO₂R, -COR, -CONHR, -CON(R)₂, -S(O)₂R, -SONH₂, -S(O)R, -SO₂NHR, -SO₂N(R)₂, -NHS(O)₂R, -NRS(O)₂R, =O, =S, =NNHR, =NNR₂, =N-OR,
25 =NNHCOR, =NNRCOR, =NNHCO₂R, =NNRCO₂R, =NNHSO₂R, =NNRSO₂R, or =NR where R is an optionally substituted aliphatic, aryl or aralkyl group.

A substitutable nitrogen on a heterocyclic ring may be optionally substituted. Suitable substituents on
30 the nitrogen include R, COR, S(O)₂R, and CO₂R, where R is an aliphatic group or a substituted aliphatic group.

Unless otherwise stated, structures depicted herein are also meant to include all stereochemical forms of the structure; i.e., the R and S configurations for each asymmetric center. Therefore, single stereochemical isomers as well as enantiomeric and diastereomeric mixtures of the present compounds are within the scope of the invention. Unless otherwise stated, structures depicted herein are also meant to include compounds which differ only in the presence of one or more isotopically enriched atoms. For example, compounds having the present structures except for the replacement of a hydrogen by a deuterium or tritium, or the replacement of a carbon by a ^{13}C - or ^{14}C -enriched carbon are within the scope of this invention.

Butenolactone **II** is readily available and inexpensive. Preferred R^1 groups include methyl, ethyl, propyl, 2-propyl, butyl, pentyl, hexyl, 4-methylpentyl, 2-methylpropyl, cyclopentyl, cyclohexyl, cyclopentylmethyl, cyclohexylmethyl, phenylethyl, phenylpropyl, phenylbutyl, (d)-menthyl, (l)-menthyl, 1-adamantyl, 2-adamantyl, 1-indanyl, 2-indanyl, bornyl, 3-tetrahydrofuranyl, benzyl, α -methylbenzyl, 4-chlorobenzyl, 4-fluorobenzyl, 4-methylbenzyl, 4-(2-propyl)benzyl, and 4-trifluoromethylbenzyl. Most preferred R^1 groups include ethyl and benzyl.

The azidolactone **III** may be obtained by the conjugate or Michael addition of an N_3 group to **II** according to methods that are generally known in the art for analogous compounds (see S. J. Miller et al., 1999, *Org. Lett.*, **1**(7), 1107). For example, **III** may be formed by adding **II** to a premixed solution of $\text{N}_3\text{-Y}$ and an acid

catalyst in a suitable solvent, followed by the addition of a Lewis base. The azide may be any nucleophilic azide known in the art to be suitable. Examples of such azides include alkali or alkaline earth salts of azide
5 such as NaN_3 or LiN_3 , tetralkylammonium azide, azidotrialkyl-, azidotriaryl-, azidoalkyldiaryl-, or azidodialkylarylsilanes such as trimethylsilylazide, triphenylsilylazide, or diphenylmethylsilylazide, or azidotrialkyltins such as azidotrimethyltin or
10 azidotributyltin.

When $\text{N}_3\text{-Y}$ is a trisubstituted silylazide such as trimethylsilylazide, the following conditions and reagents may be used. Suitable acid catalysts include carboxylic acids such as formic acid, acetic acid,
15 propanoic acid, and benzoic acid, and halogenated carboxylic acids such as trifluoroacetic acid and trichloroacetic acid, and Lewis acids such as $\text{BF}_3\cdot\text{OEt}_2$, aluminum trichloride, zinc chloride and titanium trichloride. Acetic acid and $\text{BF}_3\cdot\text{OEt}_2$ are preferred acids.
20 Suitable solvents include ethereal solvents such as tetrahydrofuran, DME, diethyl ether, methyl tert-butyl ketone, or dioxane; aromatic hydrocarbons such as benzene, toluene or xylene; halogenated hydrocarbons such as chloroform, carbon tetrachloride, dichloromethane, or
25 dichloroethane. A preferred solvent is dichloromethane. The Lewis base need only be present in a catalytic amount. Suitable Lewis bases include aliphatic tertiary amines such as triethylamine, diisopropylamine, 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU), 1,5-diazabicyclo[4.3.0]non-5-ene (DBN); heteroaromatic bases
30 such as an N-alkyl imidazole (which may be bound to a

resin) or a pyridine. A preferred base is DBU. The reaction may be carried out at a temperature in the range of about 0° to 100° C, preferably between about 0° to 40° C, and most preferably between about 20° to 40° C. The concentration of **II** will be in the range of about 0.01M to 10M, preferably about 0.1 to 1.0M. The amounts of N₃-Y reactant such as trimethylsilylazide and Lewis acid will each generally be in the range of about 1.0 to 10 equivalents per equivalent of **II**.

The above conditions and reagents for converting **II** to **III** may vary depending on the nature of N₃-Y. When N₃-Y is an azide where the counterion is an alkaline earth metal such as lithium, sodium, barium or calcium, the following conditions and reagents may be used. The amount of azide will again be in the range of about 1.0 to 10.0 equivalents. The reaction temperature will be as described for trimethylsilylazide. Suitable acids include formic acid, acetic acid, benzoic acid and buffered acids such as tetrabutylammonium bisulfate, ammonium chloride, ammonium acetate, and ammonium formate. Preferred acids are acetic acid, tetrabutylammonium bisulfate, and ammonium chloride. The amount of acid will generally be in the range of about 1.0 to about 10.0 equivalents. Suitable solvents include nonprotic organic solvents such as acetone, N-methylpyrrolidone, methyl ethylketone, tetrahydrofuran (THF), dioxane, dimethoxyethane (DME), dimethylformamide (DMF) and the halogenated hydrocarbons described above.

When N₃-Y is HN₃, it is preferred that an excess of azide be used, generally about 5 to 25 equivalents of azide per equivalent of butenolactone **II** (see

Lakschmipathi et al., 1997, *Tetrahedron Lett.*, **38**(14), 2551). Only a catalytic amount of the base is required, generally in the range of about 0.01 to 0.25 equivalents, preferentially at least about 0.10 equivalents. Suitable
5 bases include tertiary amine bases such as triethylamine, diisopropylamine, DBU, DBN or aromatic N-heterocycles such as pyridine, alkylpyridines and N-alkylimidazole (which may be resin bound), preferably triethylamine. Suitable solvents include aromatic hydrocarbons such as
10 benzene, toluene or xylene, preferably toluene. The reaction temperature will generally be in the range of about 20°C to about 110°C, preferably about 70°C to about 90°C.

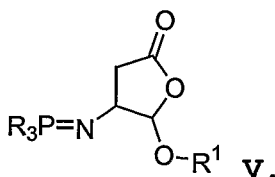
When N₃-Y is Et₂AlN₃, it is preferred to use
15 about 1.0 to 3.0 equivalents of the azide per equivalent of butenolactone **II**. (Chung, et al., 1998, *Bull. Korean Chem. Soc.*, **9**, 269) Suitable solvents include aprotic organic solvents such as diethyl ether, methyl *tert*-butyl ether, dioxane, tetrahydrofuran, hexane, benzene, and
20 toluene which is a preferred solvent. The reaction temperature will be in the range of about -20°C to about 40°C, preferably about 20°C to about 40°C.

Azidolactone **III** may be converted to the corresponding aminolactone **IV** by hydrogenation or by a
25 reaction with triphenylphosphine. Hydrogenation is more suitable when R¹ is a group that is stable to the hydrogenation conditions such as an alkyl group. Standard hydrogenation conditions may be used, such as using hydrogen gas at 1-4 atmospheres of pressure.
30 Alternatively, the hydrogen may be generated *in situ* by

known methods, such as from ammonium formate under phase transfer conditions.

Azidolactones **III** containing benzylic and other R^1 groups that are not stable to hydrogenation may be reduced to aminolactone **IV** with triphenylphosphine via the known Staudinger reaction. Similar reducing reagents include trimethyl-, triethyl-, or tributylphosphine, or an alkyl diphenylphosphinite such as methyl- or ethyldiphenylphosphinite. For this reduction, suitable solvents include aqueous organic solvents such as THF, dioxane, acetonitrile, acetone, and DMF containing about 1% to 50% water, preferably about 5% to 10% water. A preferred organic solvent is THF. The reaction temperature may be in the range of about 0°C to about 60°C, preferably between about 20°C to about 40°C.

Alternatively, the azidolactone **III** may be treated with triphenylphosphine or a similar reducing agent under anhydrous conditions to provide the iminophosphorane **V**, which is a useful intermediate:



Therefore, one embodiment of Method A proceeds through intermediate **V** and another embodiment of Method A proceeds through intermediate **IV**.

The aminolactone **IV**, obtained as described above, may be used with or without isolation from the reaction mixture. The desired caspase inhibitor prodrug **I** is derived from **IV** by attaching the appropriate P_2 , P_2-P_3 , or P_2-P_4 moiety. A coupling of **IV** with such a moiety may

be carried out using the corresponding carboxylic acid, or reactive equivalent thereof, under standard amide bond-forming or coupling conditions. A typical coupling reaction includes a suitable solvent, **IV** in a concentration ranging from about 0.01 to 10M, preferably about 0.1 to 1.0M, the requisite carboxylic acid, a base and a peptide coupling reagent.

If **IV** is used without isolation, the coupling may be carried out *in situ* in the solvent of the reaction mixture used in the preparation of **IV**, or in a different solvent. To this reaction mixture, the requisite carboxylic acid may be added and the reaction maintained at a temperature in the range of about 0° to 100°C, preferably between about 20° to 40°C. The base and peptide coupling reagent are then added to the mixture, which is maintained at a temperature in the range of about 0° to 60°C, preferably between about 20° to 40°C. The base is typically a tertiary amine base, such as triethylamine, diisopropylethylamine, N-methylmorpholine, DBU, DBN, N-methylimidazole, preferably triethylamine or diisopropylethylamine. The amount of base used is generally up to about 20 equivalents per equivalent of **IV**, preferably at least about 3 equivalents of base. Examples of peptide coupling reagents include DCC (dicyclohexylcarbodiimide), DIC (diisopropylcarbodiimide), di-*p*-toluoylcarbodiimide, BDP (1-benzotriazole diethylphosphate-1-cyclohexyl-3-(2-morpholinylethyl)carbodiimide), EDC (1-(3-dimethylaminopropyl-3-ethyl-carbodiimide hydrochloride), cyanuric fluoride, cyanuric chloride, TFFH (tetramethyl fluoroformamidinium hexafluorophosphate), DPPA

(diphenylphosphorazidate), BOP (benzotriazol-1-yloxytris(dimethylamino)phosphonium hexafluorophosphate), HBTU (O-benzotriazol-1-yl-N,N,N',N'-tetramethyluronium hexafluorophosphate), TBTU (O-benzotriazol-1-yl-N,N,N',N'-tetramethyluronium tetrafluoroborate), TSTU (O-(N-succinimidyl)-N,N,N',N'-tetramethyluronium tetrafluoroborate), HATU (N-[(dimethylamino)-1-H-1,2,3-triazolo[4,5,6]-pyridin-1-ylmethylene]-N-methylmethanaminium hexafluorophosphate N-oxide), BOP-Cl (bis(2-oxo-3-oxazolidinyl)phosphinic chloride), PyBOP ((1-H-1,2,3-benzotriazol-1-yloxy)-tris(pyrrolidino)phosphonium tetrafluorophosphate), BrOP (bromotris(dimethylamino)phosphonium hexafluorophosphate), DEPBT (3-(diethoxyphosphoryloxy)-1,2,3-benzotriazin-4(3H)-one) PyBrOP (bromotris(pyrrolidino)phosphonium hexafluorophosphate). EDC, HOAT, BOP-Cl and PyBrOP are preferred peptide coupling reagents. The amount of peptide coupling reagent is in the range of about 1.0 to about 10.0 equivalents. Optional reagents that may be used in the amide bond-forming reaction include DMAP (4-dimethylaminopyridine) or active ester reagents, such as HOBT (1-hydroxybenzotriazole), HOAT (hydroxyazabenzotriazole), HOSu (hydroxysuccinimide), HONB (endo-N-hydroxy-5-norbornene-2,3-dicarboxamide), in amounts ranging from about 1.0 to about 10.0 equivalents.

Alternatively, one may treat either **IV** or **V** with a reactive equivalent of the R²COOH carboxylic acid, such as P₂-, P₂-P₃-, or P₂-P₃-P₄-C(=O)X, where C(=O)X is a group that is more reactive than COOH in the coupling reaction. Examples of -C(=O)X groups include groups

where X is Cl, F, OC(=O)R (R = aliphatic or aryl), SH, SR, SAr, or SeAr. When **V** is the intermediate, rather than **IV**, it is preferred that the acid fluoride (X is F) be used in the coupling reaction. Suitable conditions for using these reactive equivalents are known in the art.

A number of chemical groups are known that may be used as the P₃-P₂- portion of the ICE or caspase inhibitor prodrug. Examples of such P₃-P₂- groups are shown in Table 1 as part of a P₄-P₃-P₂- moiety.

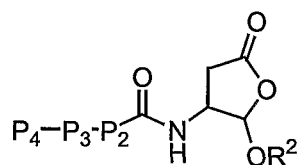
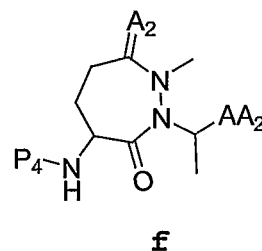
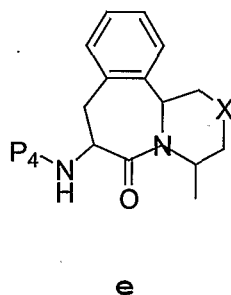
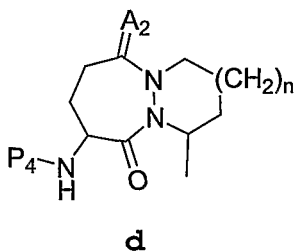
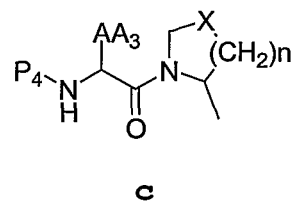
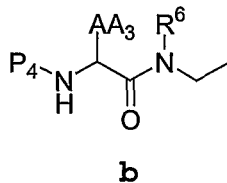
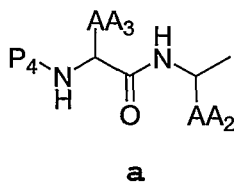
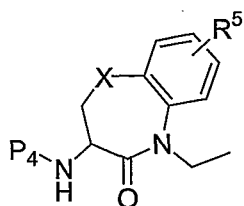
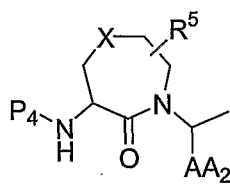
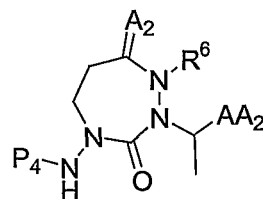
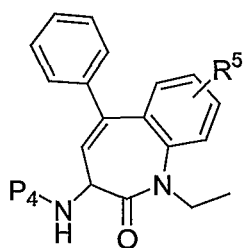
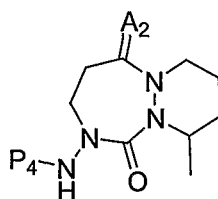
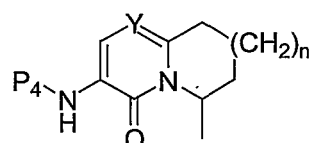
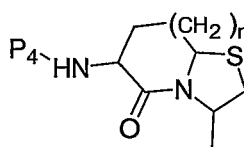
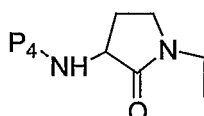
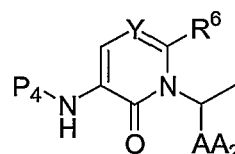
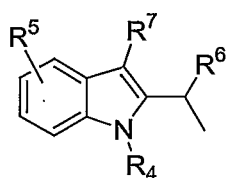
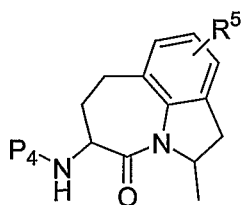
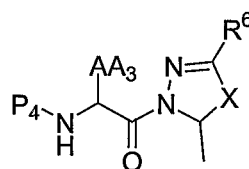


Table 1. P_4 - P_3 - P_2 - Groups



**g****h****i****j****k****l****m****n****o****p****q****r**

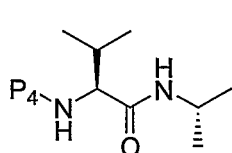
10

where n is zero to three; AA refers to an amino acid side
 15 chain; X is N, O, S, SO, SO₂, CHF, CF₂, C(R³)₂, C=O, or
 C=NOR; A is O, S or H₂; Y is N or CH; R is hydrogen, C₁₋₁₂
 alkyl group, aryl group, or heteroaryl group, the R
 groups being optionally substituted with one or halogen;
 R³ is an alkyl having one to six carbons; and R⁵ is

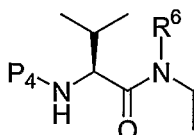
hydrogen, halo, alkyl, alkoxy, haloalkyl, haloalkoxy,
 amino, phenyl, phenoxy, hydroxy, alkoxycarbonyl,
 carboxyl, alkylcarbonyl, alkylsulfonyl, alkylsulfoxyl,
 alkylcarbonylamino, alkylcarbonylalkylamino, alkylamino,
 5 dialkylamino, aminosulfonyl, or cyano; and R^6 and R^7 are
 independently selected from R^3 , aryl, heteroaryl, $(C_{1-12}$
 alkyl)aryl, (C_{1-12}) benzocycloalkyl, or $(C_{1-12}$
 alkyl)heteroaryl.

Preferred P_4 - P_3 - P_2 - groups are shown in Table 2.

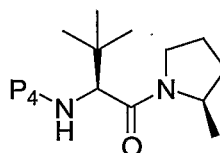
Table 2. Preferred P_4 - P_3 - P_2 - Groups



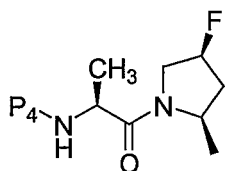
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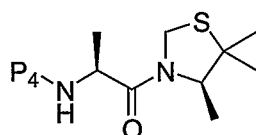
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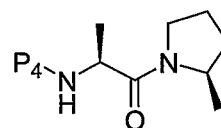
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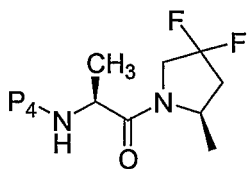
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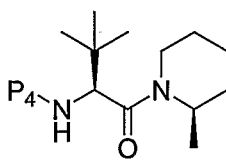
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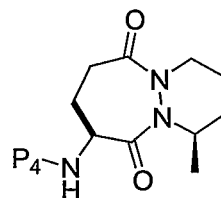
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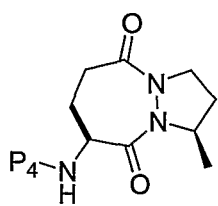
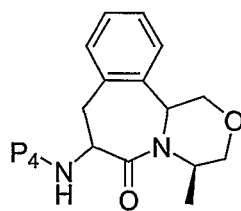
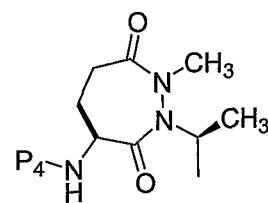
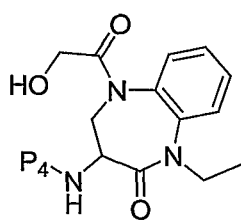
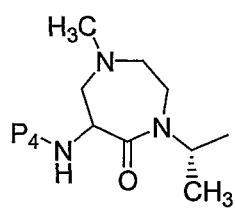
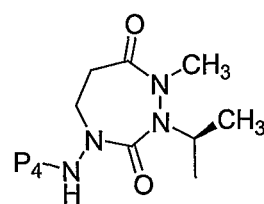
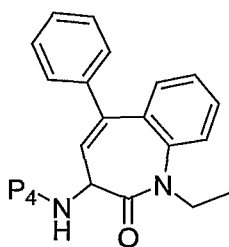
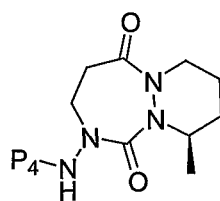
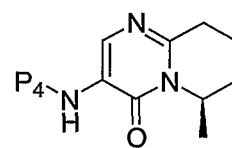
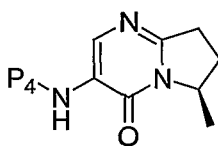
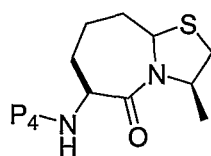
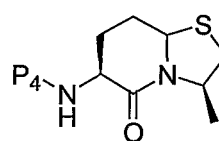
c-5



c-6

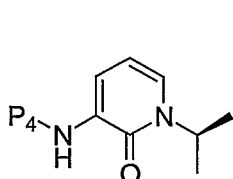
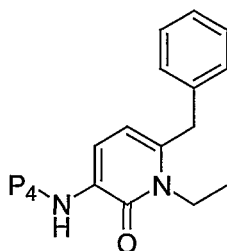
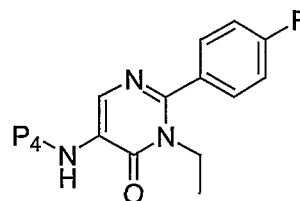
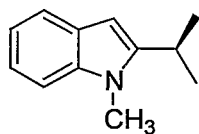
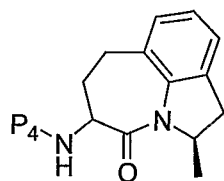
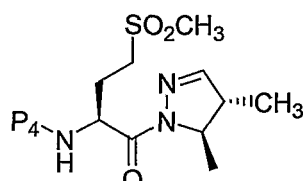
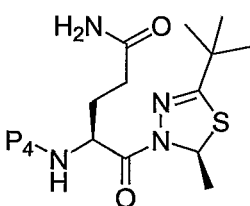
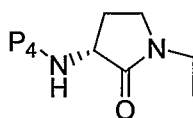


d-1

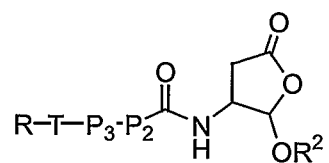
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5

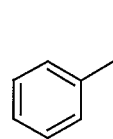
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**o-1****o-2****o-3****p-1****q-1****r-1****r-2****n-1**

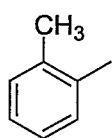
- 10 where R⁶ is an optionally substituted benzyl as described below or 2-indanyl, and the P₄ moiety is represented by R-T-, where R-T is R-CO, ROC=O, RNHC=O, RC(O)C=O, or RSO₂. Preferred R groups of P₄ are shown in Table 3.



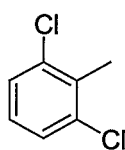
5 Table 3. Preferred R Groups of P₄



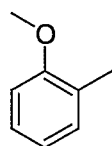
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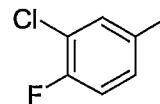
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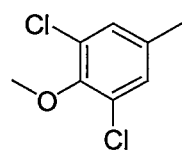
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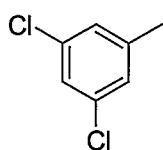


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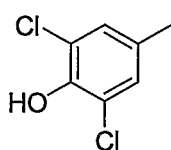


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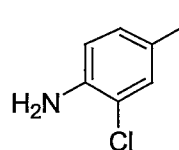
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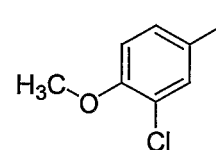
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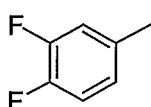
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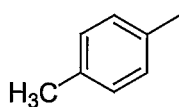


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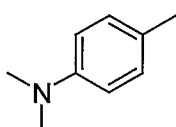


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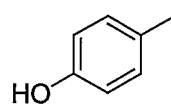
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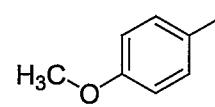
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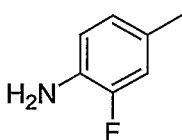
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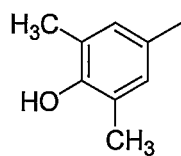
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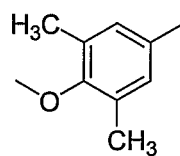
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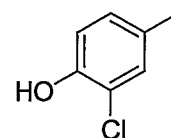
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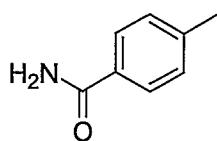
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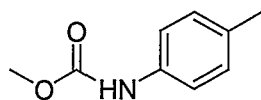
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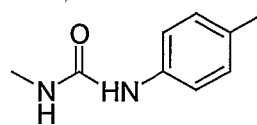
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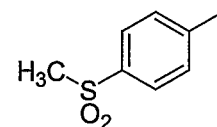
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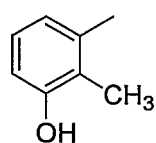
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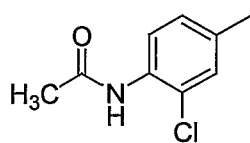
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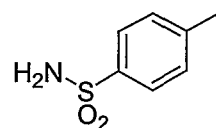
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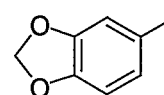
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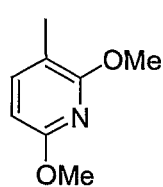
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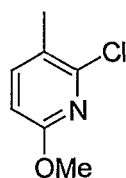
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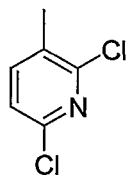
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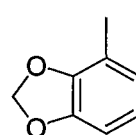
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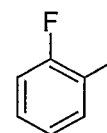
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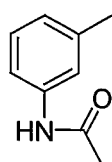
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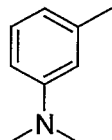
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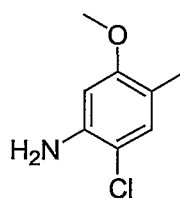
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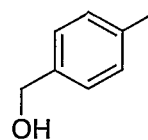
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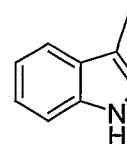
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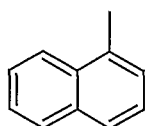
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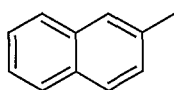
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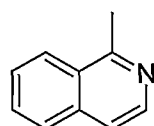
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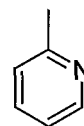
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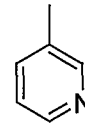
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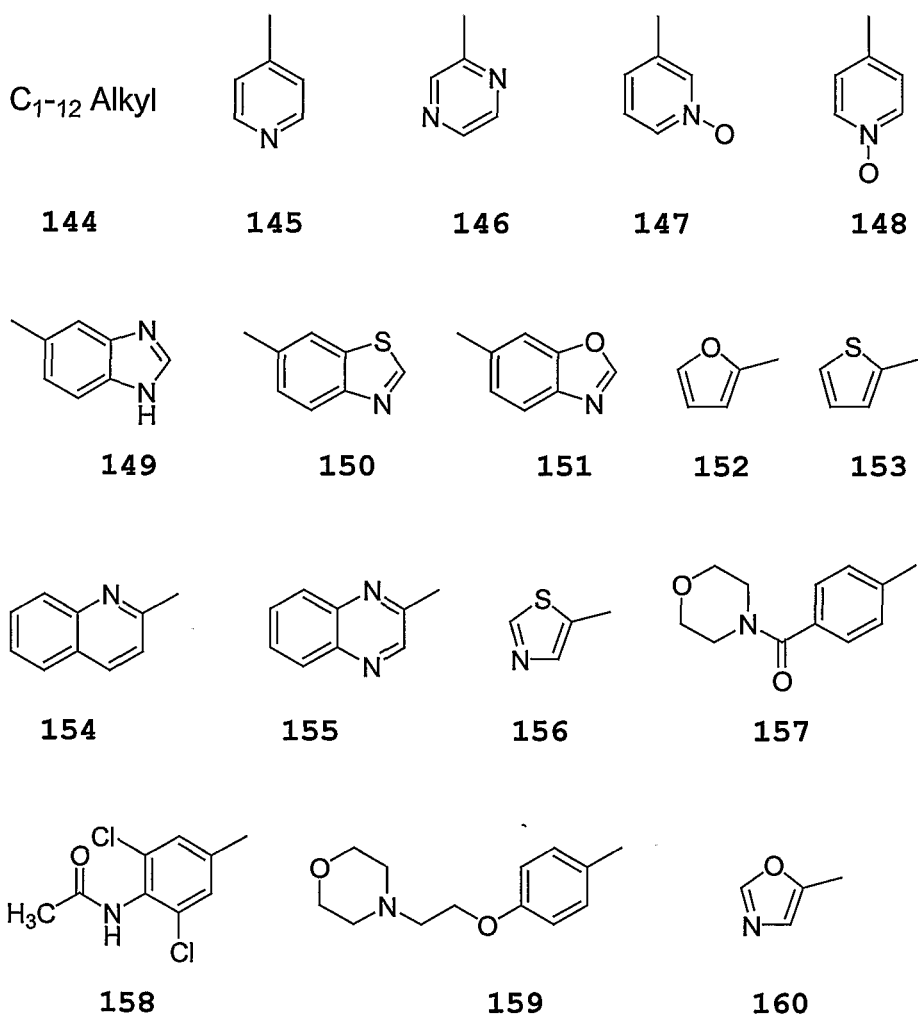
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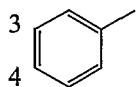
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143

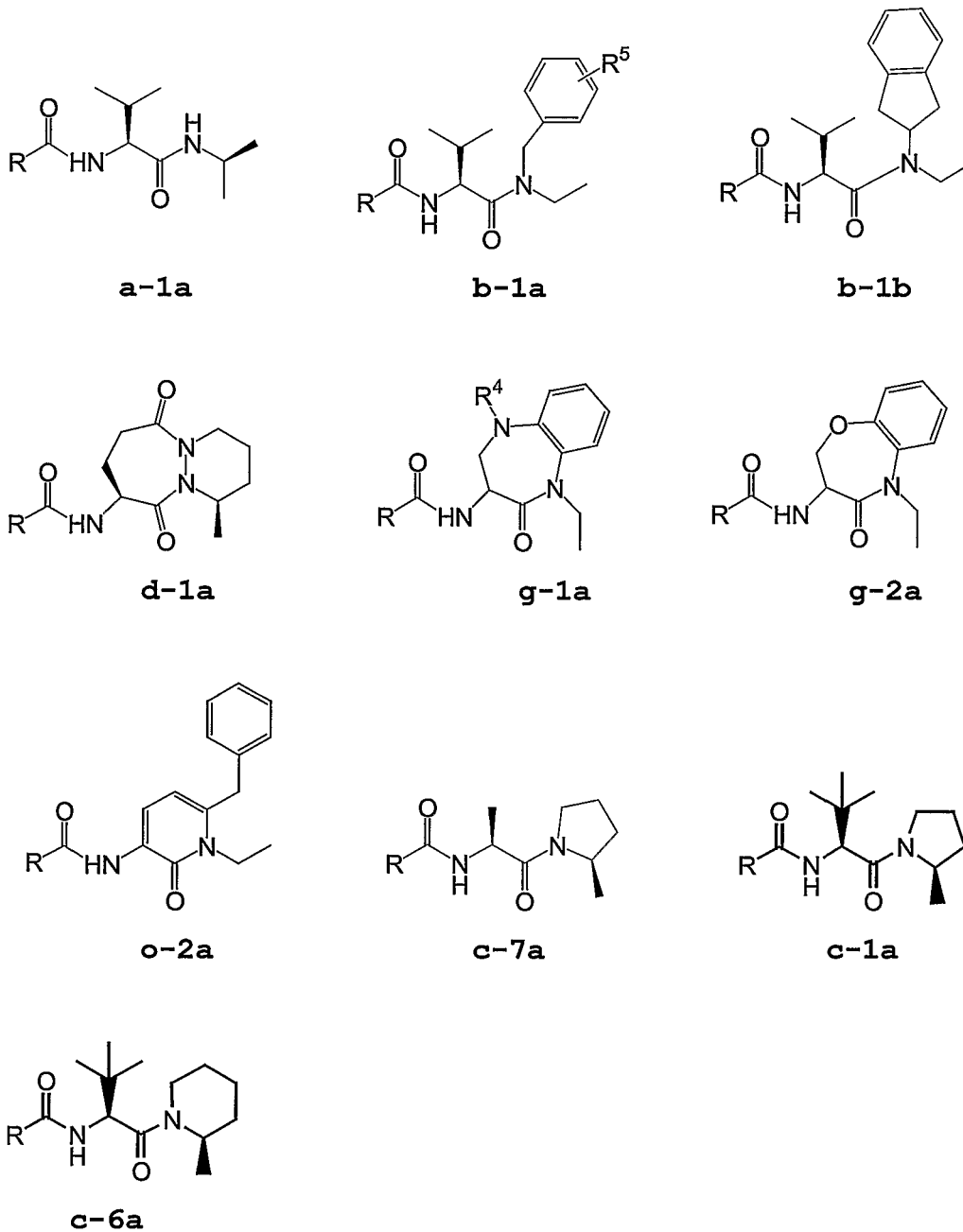


Most preferably, R-T- is R-CO where R is 1-naphthyl, 2-naphthyl, 1-isoquinolinyl, or



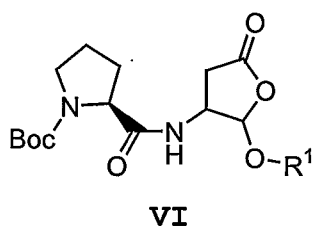
where positions 3 and 5 of R are independently and optionally substituted by halogen, preferably chloro, or C₁₋₃ alkyl, and position 4 is optionally substituted by amino, acetamido, hydroxy or methoxy.

The most preferred P₄-P₃-P₂- groups are shown in Table 4.

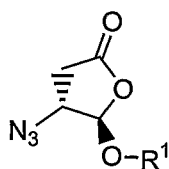
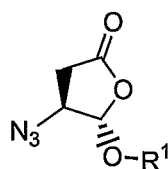
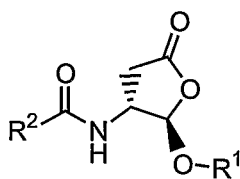
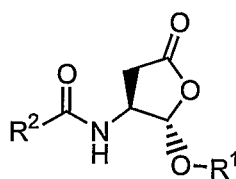
Table 4. Most Preferred P₄-P₃-P₂- Groups

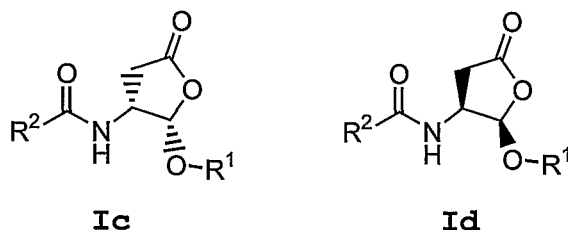
15 where R is, referring to Table 3, one of the following groups: 100, 105, 107, 108, 114, 117, 119, 126, 136, 139, 140, and 141.

In attaching the P₄-P₃-P₂ moiety, or portion thereof, the moiety may be attached in one piece as described above or subunits of the moiety may be added in a sequential manner. For example, Boc-protected proline
 5 may be coupled to **IV** to provide **VI**.



10 After removal of the Boc group, a P₃ or P₃-P₄ moiety may be attached by alkylation or acylation of the proline nitrogen.

(4R, 5R)-**IIIa**(4S, 5S)-**IIIb****Ia****Ib**



The reaction of N_3 -Y on racemic **II** generally produces the anti isomers **IIIa** and **IIIb**, shown above.

5 These may be converted by the present process to the anti products **Ia** and **Ib**. In many cases R^2 will be a chiral fragment and **Ia** and **Ib** will be diastereomeric. To separate the diastereomers, one may use chromatography and/or crystallization, depending on the nature of R^1 and

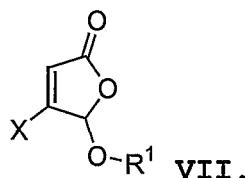
10 R^2 . Epimerization of either **Ia** or **Ib** provides the syn isomers **Ic** or **Id**, respectively. The epimerization reaction is performed in the presence of a protic acid or Lewis acid (French patent application 97/08932).

Suitable Lewis acids include ferric chloride, titanium tetrachloride, boron trichloride, boron trifluoride and tin tetrachloride and suitable protic acids include

15 organic acids such as methanesulfonic acid, trifluoroacetic acid and para toluenesulfonic acid and mineral acids such as hydrochloric acid and sulfuric

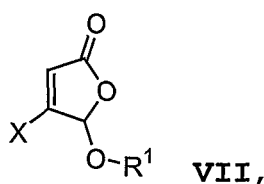
20 acid.

Another method of the present process for making a compound **I** proceeds through the butenolactone **VII** where X is chloro, bromo or iodo:



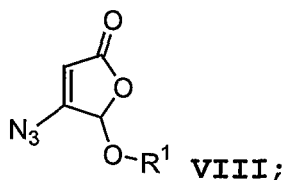
A preferred starting butenolactone **VII** is the bromofuranone (X = Br), which may be obtained according to Escobar et al., *An. Quim.*, 1971, **67**, 43. This process, referred to herein as Method B, comprises the

(a) providing a butenolactone **VII**:

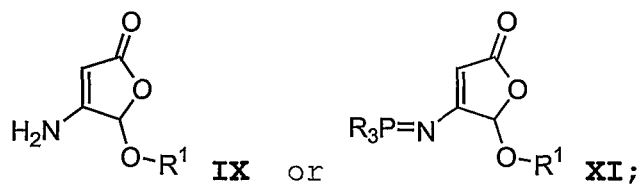


wherein R¹ is as described above, and X is chloro, bromo or iodo;

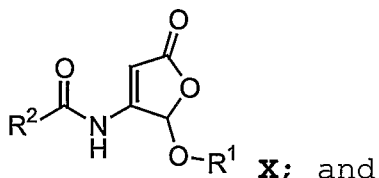
(b) treating **VII** with an azide N₃-Y, where Y is a silyl group or a counterion, to form an azidobutenolactone **VIII**:



(c) converting **VIII** to an aminobutenolactone **IX** or iminophosphorane **XI**:



(d) coupling either **IX** or **XI** with R^2COOH , or a reactive equivalent thereof, to form **X**:



5

(e) reducing the furanone ring double bond in **X** to provide **I**. R^2COOH is an organic radical, preferably a P_2 -, P_2 - P_3 -, or P_2 - P_4 carboxylic acid.

It will be apparent that steps b-d of the above Method B process are analogous to those described earlier with respect to Method A, and may be carried out in a similar manner.

Also within the scope of this invention, another embodiment of the coupling reaction of amine **IX** to form **I** proceeds by acylation of the anion of **IX** using a reactive equivalent of the carboxylic acid, such as P_2 -, P_2 - P_3 -, or P_2 - P_3 - P_4 - $C(=O)X$, where $C(=O)X$ is as described above. The anion of **IX** is first generated by treating **IX** in a solvent with any suitable base. Examples of solvents that may be used include ethereal solvents such as THF, DME, dioxane, diethyl ether, methyl-*tert*-butyl ether; aromatic hydrocarbons, such as benzene, toluene, xylene; halogenated hydrocarbons, such as dichloromethane, carbon tetrachloride, dichloroethane; or other organic solvents, such as acetonitrile. Preferred solvents include THF, DME, toluene or dichloromethane. Suitable bases for generating the anion include organic bases such as an alkali metal hydride, an alkali metal *tert*-butoxide, an alkyl or aryl lithium, such as methyl-, butyl- or phenyllithium; an alkali metal amide, such as lithium-,

sodium- or potassium bis(trimethylsilyl)amide, diisopropylamide, or tetramethylpiperidine. Preferred bases include lithium bis(trimethylsilyl)amide, lithium diisopropylamide, or lithium tetramethylpiperidine. The anion of **IX** is treated with the carboxylic acid equivalent at a reaction temperature that may be in the range of about -78°C to 120°C, preferably between about 0°C to 60°C.

The azidobutenolactone **VIII** may be obtained from **VII** by the displacement of its substituent X with an azide. For the reaction of **VII** where its substituent X is bromo, **VII** is preferably treated with an alkali or alkaline earth salt of azide, such as NaN₃ or LiN₃. A silylazide (N₃-Y where Y is a silyl group as described above) may be used in the presence of a fluoride reagent, such as tetrabutylammonium fluoride, cesium fluoride, potassium fluoride, sodium fluoride or the like, to generate the nucleophilic azide anion. Suitable solvents include non-protic organic solvents, such as acetone, NMP, MEK, THF, DME, and dioxane, and halogenated hydrocarbons, such as chloroform, carbon tetrachloride, dichloromethane, and dichloroethane. The reaction is run at a temperature in the range of about 0°C to 100°C, preferably between about 20°C to 40°C.

The reduction of the furanone ring double bond in **X** to provide **I** may be accomplished with a hydride reducing agent, especially a borohydride. Examples of such borohydrides include sodium or lithium borohydride, sodium or lithium triacetoxymethylborohydride, sodium or lithium cyanoborohydride, tetrabutylammonium cyanoborohydride, sodium or lithium trialkylborohydride,

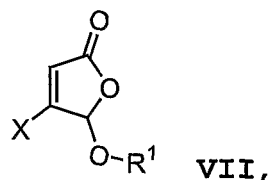
preferably sodium cyanoborohydride. Typically the reaction mixture is adjusted to be mildly acidic, preferably at a pH between 3.0 and 6.0 with acids such as HCl, HBr, acetic acid, formic acid, trifluoroacetic acid, 5 BF₃.OEt₂, aluminum trichloride, zinc chloride, or titanium tetrachloride. Optionally, the reaction may be buffered with 1.0-5.0 equivalents of sodium acetate. Optionally, the reaction may be catalyzed by the addition of 1-5% CoCl₂/semicorrin, ZnCl₂, or 1-2 equivalents of 10 chlorotrimethylsilane. Chiral hydride reducing agents are known such as R- or S-Alpine Hydride® (lithium B-isopinocampheyl-9-bora-bicyclo[3.3.1]nonyl hydride to provide asymmetric reduction.

Reduction of the ring double bond in **X** to 15 provide **I** may also be accomplished by hydrogenation. This is useful when R¹ is stable to the hydrogenation conditions, such as when R¹ is alkyl. Typical hydrogenation conditions include hydrogen gas at a pressure in the range of about one to 100 atmospheres, 20 usually between about 15 to 70 atmospheres, and a catalyst present in the range of about 0.01 to 0.5 equivalents per equivalent of **X**. Suitable catalysts include Pd/C, Pd(OH)₂, PdO, Pt/C, PtO₂, preferentially Pt/C or Pd/C. Suitable solvents include ethyl acetate, 25 alcohols, such as methanol, ethanol, isopropanol, aromatic hydrocarbons, such as benzene, toluene, xylene, ethereal such as THF, DME, dioxane, preferentially ethanol or THF. When R¹ is alkyl or aralkyl, such as benzyl, a rhodium (I) or ruthenium (II) catalyst is 30 preferred for stereoselective reduction. Such catalyst is formed by reacting the metal as one of its various

complexes with chiral forms of ligands such as methyl- or ethyl-DuPHOS (1,1-bis-2,5-dialkylphospholano)benzene, DIOP (2,3-O-isopropylidene-2,3-dihydroxy-1,4-bis(diphenylphosphino)butane), BINAP (2,2'-bis(diphenylphosphino)-1,1'-binaphthyl), CHIRAPHOS (5 bis(diphenylphosphino)butane), BPPM (N-t-butoxycarbonyl-2-(diphenylphosphino)methyl-4-(diphenylphosphino)pyrrolidine), BPPFA (N,N-dimethyl-1-[1',2-bis(diphenylphosphino)ferrocenyl]ethylamine), 10 DEGPPOS (N-benzyl-3,4-bis(diphenylphosphino)pyrrolidine), or alkyl-BPE (bisphospholanoethane). Many other suitable ligands are known in the art. Preferred catalysts are 1,2-bis(2,5-dialkylphospholano)benzene(cyclooctadiene)rhodium(I) 15 trifluoromethanesulfonate, where alkyl is a straight chain or branched alkyl group of 1-8 carbons, optionally substituted with an aromatic hydrocarbon such as phenyl. Use of the (R,R) isomer of these ligands will lead to the (S)-configuration of the α -amino carbon in the product 20 and use of the (S,S) isomer will lead to the (R)-configuration. Suitable solvents include ethyl acetate, alcohols, such as methanol, ethanol, or isopropanol, aromatic hydrocarbons, such as benzene, toluene, or xylene, ethers such as THF, DME, or dioxane. Preferred 25 solvents are toluene or methanol. The reaction concentration of **X** will typically be in the range of about 0.01M to 1.0M, preferably about 0.1M to 1.0M. The reaction temperature is usually in the range of about 0°C to 60°C, preferably between about 20°C to 40°C. (For the 30 use of rhodium catalysts see: G. Zhu, Z. Chen, X. Zhang; *J. Org. Chem.* (1999) **64**, 6907-6910; M.J.Burk, J.G.Allen,

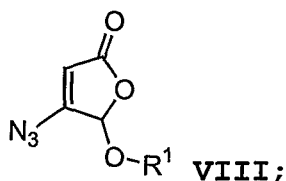
W.F.Kiesman; *J. Amer. Chem. Soc.*, (1998), **120**, 657-663;
M.J.Burk, J.E.Feaster, W.A.Nugent, R.L.Harlow; *J. Amer.
Chem. Soc.*, (1993), **115**, 10125-10138; For the use of
ruthenium catalysts see: J.M.Brown, M.Rose, F.I.Knight,
5 A.Wienand; *Recl Trav Chim Pays-Bas*, (1995), **114**, 242-251;
M.Saburi, M.Ohnuki, M.Ogasawara, T.Takahashi, Y.Uchida;
Tetrahedron Lett. (1992), **33**, 5783-5786; U Matteoli,
V.Beghetto, A.Scrivanti; *J Molecular Catalysis A:
Chemical* **140** (1999) 131-137)

10 Method B above describes a sequence in which
the aminobutenolactone **IX** is first coupled to a caspase P_x
or P_{x-y} moiety and then the ring double bond is reduced.
Alternatively, the reduction and coupling may be
performed in reverse order (Method C). Method C of the
15 present process comprises the steps of:
(a) providing a butenolactone **VII**:



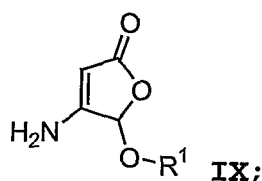
20 wherein R¹ is as described above, and X is chloro, bromo
or iodo;

(b) treating **VII** with an azide N₃-Y, where Y is a silyl
25 group or a counterion, to form an azidobutenolactone
VIII:

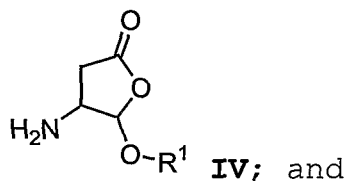


(c) converting **VIII** to an aminobutenolactone **IX**:

5

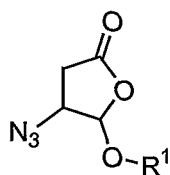
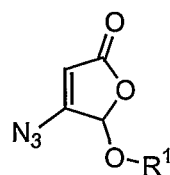


(d) reducing the ring double bond of **IX** to provide the
10 aminolactone **IV**:



(e) coupling **IV** with an organic acid R^2COOH , or reactive
15 equivalent thereof, to form **I**, where R^2COOH is preferably
a P_2 -, P_2 - P_3 -, or P_2 - P_4 carboxylic acid. In Method C,
steps (a)-(c) are the same as the corresponding steps in
Method B, and steps (d) and (e) are the same as the
corresponding steps in Method A. Therefore, Method C may
20 be carried out in a like manner with respect to the
corresponding steps.

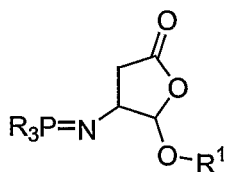
Within the scope of this invention are certain
intermediates described herein that are useful in the
preparation of the caspase inhibitors and prodrugs
25 thereof. Therefore, one aspect of this invention relates
to compounds of formula **III** or **VIII**:

**III****VIII**

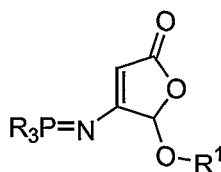
wherein R¹ is selected from an optionally substituted aliphatic group, aralkyl group or aryl group. Examples
 5 of R¹ include methyl, ethyl, propyl, 2-propyl, butyl, pentyl, hexyl, 4-methylpentyl, 2-methylpropyl, cyclopentyl, cyclohexyl, cyclopentylmethyl, cyclohexylmethyl, phenylethyl, phenylpropyl, phenylbutyl, (d)-menthyl, (l)-menthyl, 1-adamantyl, 2-adamantyl, 1-
 10 indanyl, 2-indanyl, bornyl, 3-tetrahydrofuranyl, benzyl, α-methylbenzyl, 4-chlorobenzyl, 4-fluorobenzyl, 4-methylbenzyl, 4-(2-propyl)benzyl, or 4-trifluoromethylbenzyl. Particularly useful are **III** and **VIII** where R¹ is benzyl or a C₁₋₆ alkyl such as ethyl.

15

Another aspect of this invention relates to compounds of formula



V or

**XI,**

20

wherein R¹ is selected from an optionally substituted aliphatic group, aralkyl group or aryl group, and in particular, the R¹ groups described above.

25

In order that this invention be more fully understood, the following preparative examples are set forth. These examples are for the purpose of

illustration only and are not to be construed as limiting the scope of the invention in any way.

Synthetic Examples

5 Example 1. Preparation of 4-azido-5-ethoxy-dihydrofuran-2-one (III, R¹=Et)

This procedure was carried out in a manner similar to that described by D. J. Guerin, et al., *Org. Lett* (1999), **1**, 1107-1109. To a solution of
10 azidotrimethylsilane (25.8 mL, 0.32 mol) in dichloromethane (400 mL) at room temperature under nitrogen was added acetic acid (18.1 mL, 0.32 mol), and the reaction was stirred for 20 min. 5-Ethoxy-5H-furan-2-one (II, R¹=Et) (8.10 g, 0.063 mol) was added dropwise,
15 followed by 1,8-diazabicyclo[5.4.0]undec-7-ene (1.9 mL, 0.013 mol). The reaction was stirred for 24h, washed with sodium bicarbonate, dried over sodium sulfate and evaporated. Purification by flash chromatography (SiO₂) eluted with 1:9 ethyl acetate:hexanes afforded 4-azido-5-
20 ethoxy-dihydrofuran-2-one (7.85 g, 73% yield) as a colorless oil.

¹H-NMR (500 MHz, CDCl₃) δ 5.17 (s, 1H), 4.00 (dd, J=7.0, 1.0Hz, 1H), 3.71 (m, 1H), 3.49 (m, 1H), 2.77 (dd, J=17.0, 6.0 Hz, 1H), 2.32 (dd, J=8.0, 2.2Hz, 1H), 1.08 (t, J=7.1Hz, 1.5H), 1.07 (t, J=7.1Hz, 1.5H) ppm.
25

In a manner similar to that described above, except starting with 5-benzyloxy-5H-furan-2-one (II, R¹=Bn), 4-azido-5-benzyloxy-dihydrofuran-2-one (III, R¹=Bn) was prepared as a white solid, 1.62 g (72% yield).
30 ¹H-NMR (500 MHz, CDCl₃) δ 7.19 (m, 5H), 5.25 (s, 1H), 4.71 (d, J=11.5Hz, 1H), 4.48 (d, J=11.4Hz, 1H), 4.07 (dd,

J=6.9, 0.9 Hz, 1H), 2.82 (ddd, J=18.1, 7.1, 1.0 Hz, 1H), 2.36 (ddd, J=18.1, 4.3, 1.4 Hz, 1H) ppm.

Example 2. Preparation of 4-

5 [(triphenylphosphoranylidene)-amino]-5-ethoxy-
dihydrofuran-2-one (V, R¹=Et)

A solution of 4-azido-5-ethoxy-dihydrofuran-2-one (0.05 g, 0.29 mmol) and triphenylphosphine (0.078 g, 0.29 mmol) in toluene (5 mL) was stirred at room
10 temperature under nitrogen for 5h. The solvent was evaporated to afford 4-
[(triphenylphosphoranylidene)amino]-5-ethoxy-dihydrofuran-2-one (0.12 g, 100% yield) as an off-white, waxy solid. ¹H-NMR (500 MHz, CDCl₃) δ 7.50 (m, 6H), 7.40
15 (m, 3H), 7.33 (m, 6H), 5.08 (d, J=3.1 Hz, 1H), 3.63 (m, 1H), 3.23 (m, 1H), 2.50 (dd, J=17.1, 5.9 Hz, 1H), 2.27 (dd, J=17.1, 5.9 Hz, 1H), 0.91 (t, J=7.0 Hz, 3H) ppm.

Example 3. Preparation of (R)-2-(2-ethoxy-5-oxo-tetrahydrofuran-3-ylcarbamoyl)-pyrrolidine-1-carboxylic acid tert-butyl ester (I, R¹=Et)

5

Method 1. From 4-azido-5-ethoxy-dihydrofuran-2-one via hydrogenation:

A mixture of 4-azido-5-ethoxy-dihydrofuran-2-one (1.06 g, 6.2 mmol), (S)-pyrrolidine-1,2-dicarboxylic acid 1-*tert*-butyl ester (1.33 g, 6.2 mmol), and 10% palladium on carbon (0.50 g) in ethyl acetate previously degassed with N₂ (50 mL) was stirred under 1 atm hydrogen at room temperature for 1h. The mixture was diluted with dichloromethane, filtered (Celite) and evaporated. The crude mixture was dissolved in dichloromethane (100 mL), was treated with diisopropylethylamine (5.4 mL, 30.8 mmol), EDC (1.48 g, 7.71 mmol), and HOBt (1.04 g, 7.71 mmol) and was stirred at room temperature under nitrogen for 24h. The reaction was diluted with ethyl acetate, was washed with 10% sodium bisulfate, saturated sodium bicarbonate, and brine, was dried over sodium sulfate, and was evaporated. Purification by flash chromatography (SiO₂) eluted with 1:1 ethyl acetate:hexanes provided (R)-2-(2-ethoxy-5-oxo-tetrahydrofuran-3-ylcarbamoyl)-pyrrolidine-1-carboxylic acid *tert*-butyl ester (1.19 g, 56% yield) as a very viscous, pale yellow oil. ¹H-NMR (500 MHz, CDCl₃) δ 7.61 (br, 0.6H), 5.29 (s, 0.6H), 5.25 (br s, 0.4H), 4.29 (br, 1.2H), 4.20 (br s, 0.8H), 3.78 (m, 1H), 3.57 (m, 1H), 3.34 (br, 1.4H), 3.25 (br, 0.6H), 2.94 (br dd, J=14.9, 3.8Hz, 1H), 2.31 (dd, J=18.0, 1.4Hz,

1H), 2.1-2.3 (br 1H), 1.82 (br s, 1H), 1.39 (s, 9H), 1.17 (m, 3H) ppm.

MS (ES+): m/e=343 (M+H).

- 5 Method 2. From 4-[(triphenylphosphoranylidene)amino]-5-ethoxy-dihydrofuran-2-one:

A solution of (S)-pyrrolidine-1,2-dicarboxylic acid 1-*tert*-butyl ester (0.11g, 0.5 mmol), di-
isopropylethylamine (0.18 mL, 1.0 mmol), and
10 tetramethylfluoroformamidinium hexafluorophosphate (TFFH) (0.13 g, 0.5 mmol) in dichloromethane (3 mL) was stirred at room temperature under nitrogen for 3h. A solution of 4-[(triphenylphosphoranylidene)amino]-5-ethoxy-
dihydrofuran-2-one (0.20 g, 0.5 mmol) in dichloromethane
15 (3 mL) was added and the mixture was stirred for 24h. The reaction was diluted with ethyl acetate, was washed with 10% sodium bisulfate, saturated sodium bicarbonate, and brine, was dried over sodium sulfate, and was evaporated. Purification by flash chromatography (SiO₂) eluted with
20 1:1 ethyl acetate:hexanes provided (R)-2-(2-ethoxy-5-oxo-tetrahydrofuran-3-ylcarbamoyl)-pyrrolidine-1-carboxylic acid *tert*-butyl ester (0.11 g, 65% yield) as a very viscous, pale yellow oil.

- 25 Method 3. From (R)-2-(2-ethoxy-5-oxo-2,5-dihydrofuran-3-ylcarbamoyl)-pyrrolidine-1-carboxylic acid *tert*-butyl ester (X, R¹=Et) via hydrogenation:

To a solution of (R)-2-(2-ethoxy-5-oxo-2,5-dihydrofuran-3-ylcarbamoyl)-pyrrolidine-1-carboxylic acid
30 *tert*-butyl ester (X, R¹=Et) (0.09 g, 0.27 mmol) in toluene previously degassed with N₂ (20 mL) in a high pressure

reactor in a nitrogen filled glove bag, was added (-)-
1,2-bis((2R,5R)-2,5-diethyl-phospholano)benzene-
(cyclooctadiene)rhodium(I) trifluoromethanesulfonate (5-
15 mg). The reactor was sealed and pressurized with
5 hydrogen (950 psi, 65 atm) and was let stand at room
temperature for 2 d. Solvent was evaporated and the
residue was purified by flash chromatography (SiO₂) eluted
with 1.5:98.5 methanol:dichloromethane to provide (R)-2-
(2-ethoxy-5-oxo-tetrahydrofuran-3-ylcarbamoyl)-
10 pyrrolidine-1-carboxylic acid *tert*-butyl ester (0.092 g,
quant yield) as a colorless oil. Chiral HPLC (chiralpak-
AD column, eluted with 1:9 ethanol:hexanes): isomer Ib-
35.4%, isomer Id-56.4%, (mixture of isomers Ia and Ic)-
8.2%. ¹H-NMR (500 MHz, CDCl₃) δ 7.60 (br s, 0.25H), 7.40
15 (br s, 0.25H), 6.5 (br m, 0.25H), 5.38 (d, J=5Hz, 0.5H),
5.27 (s, 0.5H), 4.65 (br, 0.5H), 4.20 (br m, 1.5H), 3.85
(m, 0.5H), 3.77 (m, 0.5H), 3.57 (m, 1H), 3.30 (m, 2H),
2.95 (m, 0.5H), 2.80 (br m, 0.5H), 2.30 (br m, 2H), 1.85
(br s, 3H), 1.37 (s, 9H), 1.20 (t, J=7Hz, 1.5H), 1.15 (t,
20 J=7Hz, 1.5H) ppm.

Example 4. Preparation of (R)-2-(2-benzyloxy-5-oxo-
tetrahydrofuran-3-ylcarbamoyl)-pyrrolidine-1-carboxylic
acid *tert*-butyl ester (I, R¹=Bn)

25 A solution of (S)-pyrrolidine-1,2-dicarboxylic
acid 1-*tert*-butyl ester (0.13g, 0.6 mmol), 4-azido-5-
benzyloxy-dihydrofuran-2-one (0.14 g, 0.6 mmol) and
triphenylphosphine (0.28 g, 1.0 mmol) in tetrahydrofuran
(5mL) and water (5 drops) was stirred under nitrogen for
30 0.5h at room temperature and for 2h at 65°C. The reaction
was cooled to room temperature, was treated with with di-

isopropylethylamine (0.52 mL, 5.0 mmol), EDC (0.15 g, 0.75 mmol), and HOBT (0.10 g, 0.75 mmol) and was stirred at room temperature under nitrogen for 20h. The reaction was diluted with ethyl acetate, was washed with 10% sodium bisulfate, saturated sodium bicarbonate, and brine, was dried over sodium sulfate, and was evaporated. Purification by flash chromatography (SiO₂) eluted with 2:3 ethyl acetate:hexanes provided (R)-2-(2-benzyloxy-5-oxo-tetrahydrofuran-3-ylcarbamoyl)-pyrrolidine-1-carboxylic acid *tert*-butyl ester (0.12 g, 49% yield) as a sticky resin. ¹H-NMR (500 MHz, CDCl₃) δ 7.63 (br d, J=7.6Hz, 0.7H), 7.28 (m, 5H), 6.50 (br, 0.3H), 5.37 (br s, 0.5H), 5.33 (s, 0.5H), 4.77 (d, J=11.6Hz, 1H), 4.56 (dd, J=11.6, 3.7Hz, 1H), 4.37 (br s, 1H), 4.18 (br s, 1H), 3.32 (br s, 1.4H), 3.24 (br s, 0.6H), 2.97 (br d, J=11.6Hz, 1H), 2.35 (dd, J=18.1, 1.7Hz, 1H), 2.1-2.3 (br, 1H), 1.81 (br s, 3H), 1.58 (s, 9H) ppm. MS (ES⁺): m/e=405 (M+H).

In a similar manner, (R)-2-(2-ethoxy-5-oxo-tetrahydrofuran-3-ylcarbamoyl)-pyrrolidine-1-carboxylic acid *tert*-butyl ester (I, R¹=Et) (0.07 g, 7% yield) was prepared.

Example 5. Preparation of 4-bromo-5-ethoxy-5H-furan-2-one (VII, R¹=Et)

This procedure was carried out in a manner similar to that described by C. Escobar, et al., *Ann. Quim.* (1971), 67, 43-57.). To a solution of 5-ethoxy-5H-furan-2-one (II, R¹=Et) (10.0 g, 78.0 mmol) in carbon tetrachloride (50 mL) at 0°C was added over 0.5h a solution of bromine (4.05 mL, 78.2 mmol) in carbon

tetrachloride (25 mL). The reaction was stirred 1h at 0°C, then 2h at room temperature. The solvents were removed under reduced pressure and the residue was short-path distilled at pump vacuum (about 0.5mm). The fraction
5 collected at 100°C-120°C provided 4-bromo-5-ethoxy-5H-furan-2-one (13.2 g, 82% yield) as a yellow oil. ¹H-NMR (500 MHz, CDCl₃) δ 6.24 (s, 1H), 5.63 (s, 1H), 3.71 (m, 1H), 3.63 (m, 1H), 1.14 (t, J=7.1Hz, 3H) ppm.

10 Example 6. Preparation of 4-azido-5-ethoxy-5H-furan-2-one (VIII, R¹=Et)

A mixture of 4-bromo-5-ethoxy-5H-furan-2-one (2.07 g, 10.0 mmol) and sodium azide (0.66 g, 10.2 mmol) in dimethylformamide (10 mL) was stirred at room
15 temperature under nitrogen for 24h. The reaction was diluted with ethyl acetate, was washed with 0°C water and with brine, was dried over sodium sulfate, and was evaporated. Purification by flash chromatography (SiO₂) eluted with 1:9 ethyl acetate:hexanes afforded 4-azido-5-
20 ethoxy-5H-furan-2-one (1.04 g, 62% yield) as a pale yellow oil. ¹H-NMR (500 MHz, CDCl₃) δ 5.83 (s, 1H), 5.63 (s, 1H), 3.99 (m, 1H), 3.88 (m, 1H), 1.35 (t, J=7.1Hz, 3H) ppm.

25 Example 7. Preparation of 4-amino-5-ethoxy-5H-furan-2-one (IX, R¹=Et)

Method 1. Via hydrogenation:

A mixture of 4-azido-5-ethoxy-5H-furan-2-one
30 (0.62 g, 3.67 mmol) and 10% palladium on charcoal (0.31 g) in de-oxygenated ethyl acetate (20 mL) was stirred at

1 atm hydrogen for 24h. The reaction was filtered (Celite) and evaporated to provide 4-amino-5-ethoxy-5H-furan-2-one (0.53 g, 100% yield) as a yellow oil. ¹H-NMR (500 MHz, CDCl₃) δ 5.60 (s, 1H), 4.85 (br s, 2H), 4.81 (s, 1H), 3.81 (m, 1H), 3.67 (m, 1H), 1.22 (t, J=7.1Hz, 3H) ppm.

Method 2. Via Staudinger reaction:

A mixture of 4-azido-5-ethoxy-5H-furan-2-one (0.10 g, 0.59 mmol) and triphenylphosphine (0.15 g, 0.59 mmol) in tetrahydrofuran (5 mL) and water (0.5 mL) was stirred for 20h at room temperature, followed by 3d at 65°C under nitrogen. The reaction was diluted with dichloromethane, washed with water, dried over sodium sulfate and evaporated to afford 4-amino-5-ethoxy-5H-furan-2-one in a mixture with triphenylphosphine oxide (0.25 g, 100% yield).

Example 8. Preparation of 4-[(triphenylphosphoranylidene)amino]-5-ethoxy-5H-furan-2-one (XI, R¹=Et)

A solution of 4-azido-5-ethoxy-5H-furan-2-one (0.17 g, 1.0 mmol) and triphenylphosphine (0.26 g, 1.0 mmol) in toluene (5mL) at room temperature under nitrogen was stirred for 1h, then heated at 60-70°C for 8h. The reaction was cooled, was diluted with ethyl acetate, was washed with sodium bisulfate, sodium bicarbonate and brine, was dried over sodium sulfate and was evaporated. Purification by flash chromatography (SiO₂) eluted with 1:1 ethyl acetate:hexanes afforded 4-[(triphenylphosphoranylidene)amino]-5-ethoxy-5H-furan-2-

one (0.18g, 45% yield) as an off-white solid. $^1\text{H-NMR}$ (500 MHz, CDCl_3) δ 7.73 (m, 6H), 7.65 (m, 3H), 7.55 (m, 6H), 5.58 (s, 1H), 4.33 (s, 1H), 3.81 (m, 1H), 3.64 (s, 1H), 1.27 (t, J 7.1Hz, 3H) ppm. MS (ES+) m/e =404 (M+H).

5

Example 9. Preparation of (R)-2-(2-ethoxy-5-oxo-2,5-dihydrofuran-3-ylcarbamoyl)-pyrrolidine-1-carboxylic acid *tert*-butyl ester (X, $R^1=\text{Et}$)

10 Method 1. Via peptide coupling conditions:

A solution of 4-amino-5-ethoxy-5H-furan-2-one (0.04 g, 0.30 mmol), (S)-pyrrolidine-1,2-dicarboxylic acid 1-*tert*-butyl ester (0.07 g, 0.30 mmol), diisopropylethylamine (0.12 mL, 0.66 mmol) and O-
15 benzotriazol-1-yl-N,N,N',N'-tetramethyluronium hexafluorophosphate (HBTU) (0.14 g, 0.38 mmol) in dichloromethane (3 mL) was stirred 24h, was evaporated, was redissolved in 1-methyl-pyrrolidinone (3 mL) and was stirred 3d. The reaction was diluted with ethyl acetate,
20 was washed with 10% sodium bisulfate, saturated sodium bicarbonate, and brine, was dried over sodium sulfate, and was evaporated. Purification by two flash chromatographies (SiO_2), eluted first with 4:6, then with 35:65 ethyl acetate:hexanes afforded (R)-2-(2-ethoxy-5-
25 oxo-2,5-dihydrofuran-3-ylcarbamoyl)-pyrrolidine-1-carboxylic acid *tert*-butyl ester (0.009g, 9% yield) as a film. $^1\text{H-NMR}$ (500 MHz, CDCl_3) δ 10.2 (br s, 0.7H), 10.1 (br s, 0.3H), 6.21 (br s, 0.7H), 6.17 (br s, 0.3H), 5.68 (s, 0.7H), 5.60 (br s, 0.3H), 4.38 (br s, 1H), 3.85 (m, 1H), 3.72 (m, 1H), 3.25-3.45 (m, 2H), 2.53 (br d, 12.5Hz,

30

0.7H), 2.1 (br, 0.3H), 1.87 (br m, 3H), 1.44 (s, 9H),
1.21 (m, 3H) ppm.

MS (ES+): m/e=341 (M+H).

5

Method 2(A). Via anion formation/acylation:

To a solution of 4-amino-5-ethoxy-5H-furan-2-one (0.08 g, 0.58 mmol) in tetrahydrofuran (10 mL) at -78°C under nitrogen was dropwise added a solution of 1M
10 lithium bis(trimethylsilyl)amide in tetrahydrofuran (0.64 mL, 0.64 mmol). The reaction was stirred 3h at 0°C. A solution of 2-fluorocarbonyl-pyrrolidine-1-carboxylic acid *tert*-butyl ester (0.20 g, 0.77 mmol) in tetrahydrofuran (3 mL) was added dropwise. The reaction
15 was stirred for 16h at room temperature. The mixture was diluted with ethyl acetate, was washed with 10% sodium bisulfate, saturated sodium bicarbonate, and brine, was dried over sodium sulfate, and was evaporated. Purification by flash chromatography (SiO₂) eluted with
20 35:65 ethyl acetate:hexanes afforded (R)-2-(2-ethoxy-5-oxo-2,5-dihydrofuran-3-ylcarbamoyl)-pyrrolidine-1-carboxylic acid *tert*-butyl ester (0.05g, 26% yield). Also isolated was starting 4-amino-5-ethoxy-5H-furan-2-one (0.03 g, 36% yield).

25

Method 2(B). Via anion formation/acylation:

To a solution of 4-amino-5-ethoxy-5H-furan-2-one (0.05 g, 0.35 mmol) and 2-fluorocarbonyl-pyrrolidine-1-carboxylic acid *tert*-butyl ester (0.09 g, 0.42 mmol) in
30 tetrahydrofuran (5 mL) at room temperature under nitrogen was added sodium *tert*-butoxide (0.05 g, 0.49 mmol). The

reaction was stirred 3h at reflux. After cooling, the mixture was diluted with ethyl acetate, was washed with 10% potassium bisulfate, saturated sodium bicarbonate, and brine, was dried over sodium sulfate, and was
5 evaporated. Purification by flash chromatography (SiO₂) eluted with 4:6 ethyl acetate:hexanes afforded (R)-2-(2-ethoxy-5-oxo-2,5-dihydrofuran-3-ylcarbamoyl)-pyrrolidine-1-carboxylic acid *tert*-butyl ester (0.075g, 63% yield). Also recovered was starting 4-amino-5-ethoxy-5H-furan-2-one
10 one (0.016 g, 32% yield).

Example 10. Preparation of (R)-2-[(2R, 3S)-2-ethoxy-5-oxo-tetrahydrofuran-3-ylcarbamoyl]-pyrrolidine-1-carboxylic acid *tert*-butyl ester (VI, R¹=Et)

15 A mixture of 4-amino-5-ethoxy-5H-furan-2-one (0.13 g, 0.75 mmol), (S)-pyrrolidine-1,2-dicarboxylic acid-1-*tert*-butyl ester (0.16 g, 0.75 mmol) and several crystals of Congo Red indicator in ethanol (5 mL) was treated with sodium cyanoborohydride (0.06 g, 0.90 mmol),
20 followed by dropwise addition of 4M HCl in dioxane to attain and maintain a blueish color (~pH3). The reaction was stirred 2h, was filtered (Celite) and was evaporated. The residue was dissolved in dichloromethane (5 mL), was treated with di-isopropylethylamine (0.52 mL, 3.0 mmol),
25 EDC (0.18 g, 0.94 mmol), and HOBT (0.13 g, 0.94 mmol) and was stirred at room temperature under nitrogen for 20h. The reaction was diluted with ethyl acetate, was washed with 10% sodium bisulfate, saturated sodium bicarbonate, and brine, was dried over sodium sulfate, and was
30 evaporated. Purification by flash chromatography (SiO₂) eluted with 1:1 ethyl acetate:hexanes provided a 4:1

mixture (by $^1\text{H-NMR}$) of (R)-2-[(2R, 3S)-2-ethoxy-5-oxo-tetrahydrofuran-3-ylcarbamoyl]-pyrrolidine-1-carboxylic acid *tert*-butyl ester and (R)-2-[(2S, 3S)-2-ethoxy-5-oxo-tetrahydrofuran-3-ylcarbamoyl]-pyrrolidine-1-carboxylic acid *tert*-butyl ester (0.06 g, 23% yield) as a colorless oil. $^1\text{H-NMR}$ (500 MHz, CDCl_3) δ 7.60 (br, 0.15H), 7.44 (br s, 0.5H), 6.51 (0.35H), 5.38 (d, $J=5.3\text{Hz}$, 0.8H), 5.29 (s, 0.2H), 4.65 (br m, 0.8H), 4.1-4.3 (br m, 1.2H), 3.84 (m, 0.8H), 3.78 (m, 0.2H), 3.59 (m, 1H), 3.25-3.45 (br m, 2H), 2.95 (dd, $J=17.6, 7.1\text{Hz}$, 0.2H), 2.78 (br m, 0.8H), 2.34 (dd, $J=17.2, 10.4\text{Hz}$, 0.8H), 1.9-2.3 (br 1.7H), 1.83 (br s, 2.5H), 1.39 (s, 9H), 1.19 (m, 3H) ppm.

Also isolated was this product, (R)-2-(2-ethoxy-5-oxo-tetrahydrofuran-3-ylcarbamoyl)-pyrrolidine-1-carboxylic acid *tert*-butyl ester, as a mixture of stereoisomers (0.030 g, 11% yield) as a colorless oil.

Example 11. Preparation of 1-[2-(4-amino-3-chloro-benzoylamino)-3,3-dimethylbutyryl]pyrrolidine-2-carboxylic acid (2-ethoxy-5-oxo-tetrahydrofuran-3-yl)amide

Step A. {1-[2-(2-Ethoxy-5-oxo-tetrahydro-furan-3-ylcarbamoyl)-pyrrolidine-1-carbonyl]-2,2-dimethyl-propyl}-carbamic acid benzyl ester

To a solution of (R)-2-(2-ethoxy-5-oxo-2,5-dihydrofuran-3-ylcarbamoyl)-pyrrolidine-1-carboxylic acid *tert*-butyl ester (X, $\text{R}^1=\text{Et}$) (0.14 g, 0.41 mmol) ($^1\text{H-NMR}$ shows ~ 8:2 syn:anti epimers) and lutidine (0.48 mL, 4.1 mmol) in dichloromethane (5 mL) at room temperature under nitrogen was dropwise added

trimethylsilyltrifluoromethane-sulfonate (0.48 mL, 2.46 mmol). The reaction was stirred for 0.5 h, then was treated with saturated sodium bicarbonate, was extracted with three portions of dichloromethane, was dried (sodium sulfate), and was evaporated. To the crude intermediate was added 2-benzyloxycarbonylamino-3,3-dimethylbutyric acid (0.12 g, 0.45 mmol) in dichloromethane (5 mL), EDC (0.10 g, 0.51 mmol) and HOBT (0.07 g, 0.51 mmol). The resulting mixture was stirred at room temperature under nitrogen for 3 days. The reaction was diluted with ethyl acetate, was washed with 10% potassium bisulfate, saturated sodium bicarbonate and brine, was dried (sodium sulfate) and was evaporated. Purification by flash chromatography (SiO₂) eluted with 1:1 ethyl acetate : hexanes provided {1-[2-(2-ethoxy-5-oxo-tetrahydrofuran-3-yl carbamoyl)-pyrrolidine-1-carbonyl]-2,2-dimethylpropyl}carbamic acid benzyl ester (0.12 g, 59 % yield) as a white foam. ¹H-NMR (500 MHz, CDCl₃) δ 7.43 (br d, J=7.7Hz, 1H), 7.28 (s, 5H), 5.40 (m, 2H), 5.02 (AB q, J=12.1, 31.0Hz, 2H), 4.55 (m, 2H), 4.29 (d, J=9.6Hz, 1H), 4.23 (m, 0.2H), 3.85 (m, 0.8H), 3.73 (m, 1H), 3.58 (m, 2H), 2.90 (m, 0.2H), 2.74 (dd, J=17.0, 8.4Hz, 0.8H), 2.30 (m, 2H), 2.05 (m, 1H), 1.90 (m, 1H), 1.80 (m, 1H), 1.20 (t, J=7.0Hz, 2.4H), 1.15 (t, J=7.0Hz, 0.6H), 0.93 (s, 9H) ppm.

¹H-NMR shows ~ 8:2 syn:anti epimers. LC/MS (ES+): m/e=490.14 (M+H)

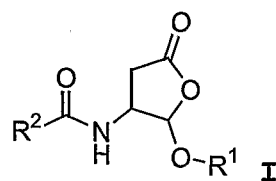
Step B. 1-[2-(4-amino-3-chloro-benzoylamino)-3,3-dimethylbutyryl]pyrrolidine-2-carboxylic acid (2-ethoxy-5-oxo-tetrahydrofuran-3-yl)amide

A solution of {1-[2-(2-ethoxy-5-oxo-tetrahydrofuran-3-yl carbamoyl)-pyrrolidine-1-carbonyl]-2,2-dimethylpropyl}carbamic acid benzyl ester (0.12 g, 0.24 mmol) in ethanol (5 mL) was treated with 10% palladium hydroxide on carbon (0.05 g), was stirred under 1 atm hydrogen pressure for 4h, was filtered through Celite and was evaporated. The crude intermediate was dissolved in dichloromethane (5 mL), and was treated with 4-amino-3-chlorobenzoic acid (0.04 g, 0.26 mmol), EDC (0.06 g, 0.29 mmol) and diisopropylethylamine (0.13 mL, 0.71 mmol) and was stirred at room temperature under nitrogen for 20h. The reaction was diluted with ethyl acetate, was washed with 10% potassium bisulfate, saturated sodium bicarbonate and brine, was dried (sodium sulfate) and was evaporated. Purification by flash chromatography (SiO₂) eluted with 7:3 ethyl acetate : hexanes provided 1-[2-(4-amino-3-chloro-benzoylamino)-3,3-dimethylbutyryl]pyrrolidine-2-carboxylic acid (2-ethoxy-5-oxo-tetrahydrofuran-3-yl)amide (0.08g, 62% yield) as a colorless film. ¹H-NMR (500 MHz, CDCl₃) δ 7.67 (d, J=2.0Hz, 1H), 7.50 (m, 0.2H), 7.44 (dd, J=8.4, 2.0Hz, 1.0H), 7.33 (d, J=8.0Hz, 0.8H), 6.69 (d, J=8.4Hz, 1H), 6.55 (d, J=9.2Hz, 1H), 5.39 (d, J=5.2Hz, 0.8H), 5.29 (s, 0.2H), 4.79 (d, J=9.4Hz, 1H), 4.62 (m, 0.8H), 4.50 (m, 1.0H), 4.25 (m, 0.2H), 3.83 (m, 0.8H), 3.77 (m, 0.2H), 3.62 (m, 0.8H), 3.55 (m, 0.2H), 2.92 (m, 0.2H), 2.76 (dd, J=17.2, 8.4Hz, 0.8H), 2.30 (m, 2H), 2.05 (m, 1H), 1.93 (m, 1.0H), 1.85 (m, 1H), 1.22 (t, J=7.1Hz, 2.4H), 1.16 (t, J=7.1Hz, 0.6H), 1.00 (s, 9H) ppm. ¹H-NMR shows ~ 8:2 syn:anti epimers. LC/MS (ES⁺): m/e= 509.08 (M+H).

While we have described a number of embodiments of this invention, it is apparent that our basic examples may be altered to provide other embodiments which utilize the compounds and methods of this invention. Therefore,
5 it will be appreciated that the scope of this invention is to be defined by the appended claims rather than by the specific embodiments which have been represented by way of example.

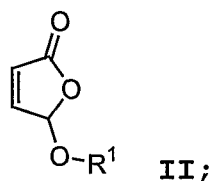
We claim:

1. A process for making a compound of formula **I**:

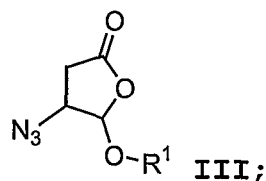


wherein R^1 is an optionally substituted group selected from an aliphatic group, aralkyl group, heterocyclalkyl group or aryl group, and R^2 is an organic radical, said process comprising the steps of:

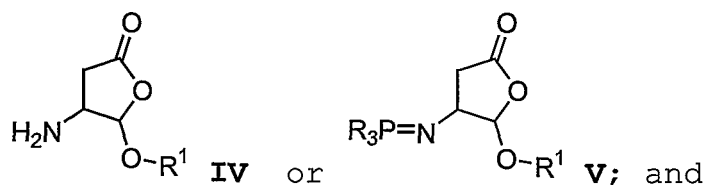
- (a) providing a butenolactone of formula **II**:



- (b) treating **II** with an azide N_3-Y , where Y is hydrogen, a silyl group, or a counterion, to form an azidolactone **III**:



- (c) converting **III** to an aminolactone **IV** or an iminophosphorane **V**:



(d) coupling either **IV** or **V** with $R^2\text{COOH}$, or a reactive equivalent thereof, to form **I**.

2. The process of claim 1 wherein R^2 is a P_4 - P_3 - P_2 moiety of a caspase inhibitor, or portion thereof.

3. The process of claim 2 wherein R^2 is a P_4 - P_3 - P_2 moiety of a caspase inhibitor, and wherein said moiety is one of the groups listed in Table 1.

4. The process of claim 3 wherein R^2 is a P_4 - P_3 - P_2 moiety of a caspase inhibitor, and wherein said moiety is one of the groups listed in Table 2.

5. The process of claim 4 wherein R^2 is a P_4 - P_3 - P_2 moiety wherein the P_4 portion thereof is selected from $R\text{-CO}$, ROC=O , RNHC=O , RC(O)C=O or RSO_2 and R is one of the groups listed in Table 3.

6. The process of claim 5 wherein R^2 is a P_4 - P_3 - P_2 moiety selected from one of the groups listed in Table 4.

7. The process of claim 2 wherein R^1 is methyl, ethyl, propyl, 2-propyl, butyl, pentyl, hexyl, 4-methylpentyl, 2-methylpropyl, cyclopentyl, cyclohexyl, cyclopentylmethyl, cyclohexylmethyl, phenylethyl, phenylpropyl, phenylbutyl, (d)-menthyl, (l)-menthyl, 1-

adamantyl, 2-adamantyl, 1-indanyl, 2-indanyl, bornyl, 3-tetrahydrofuranyl, benzyl, α -methylbenzyl, 4-chlorobenzyl, 4-fluorobenzyl, 4-methylbenzyl, 4-(2-propyl)benzyl, or 4-trifluoromethylbenzyl.

8. The process of claim 2 wherein R^1 is ethyl or an optionally substituted benzyl.

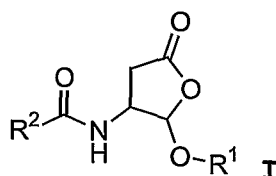
9. The process of claim 8 wherein R^1 is ethyl or benzyl.

10. The process of claim 1 or 2 wherein N_3 -Y is selected from LiN_3 , NaN_3 , $TMS-N_3$, HN_3 , or $EtAlN_3$.

11. The process of claim 1 or 2 which proceeds through the aminolactone **IV**.

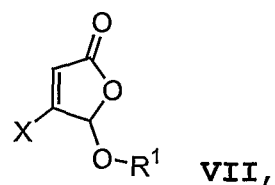
12. The process of claim 1 or 2 which proceeds through the iminophosphorane **V**.

13. A process for making a compound of formula **I**:



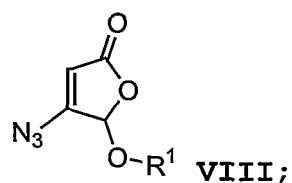
wherein R^1 is an optionally substituted group selected from an aliphatic group, aralkyl group, heterocyclylalkyl group or aryl group, and R^2 is an organic radical, said process comprising the steps of:

(a) providing a butenolactone **VII**:

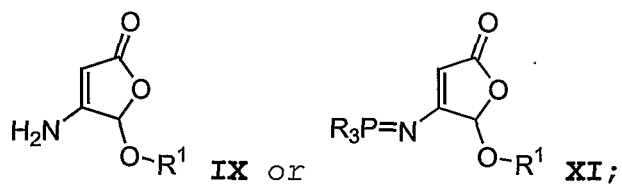


where X is chloro, bromo or iodo;

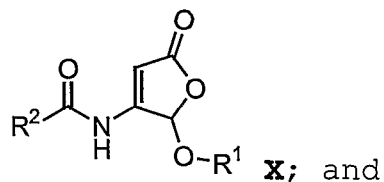
(b) treating **VII** with an azide N_3 -Y, where Y is a silyl group or a counterion, to form the azidobutenolactone **VIII**:



(c) converting **VIII** to an aminobutenolactone **IX** or an iminophosphorane **XI**:



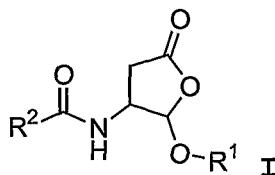
(d) coupling either **IX** or **XI** with R^2COOH , or a reactive equivalent thereof, to form **X**:



(e) reducing the furanone ring double bond in **X** to provide **I**.

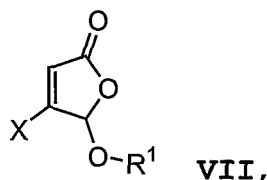
14. The process of claim 13 wherein R^2 is a P_4 - P_3 - P_2 moiety of a caspase inhibitor, or portion thereof.
15. The process of claim 14 wherein R^2 is a P_4 - P_3 - P_2 moiety of a caspase inhibitor, wherein said moiety is one of the groups listed in Table 1.
16. The process of claim 15 wherein R^2 is a P_4 - P_3 - P_2 moiety of a caspase inhibitor, wherein said moiety is one of the groups listed in Table 2.
17. The process of claim 16 wherein R^2 is a P_4 - P_3 - P_2 moiety wherein the P_4 portion thereof is selected from R -CO, $ROC=O$, $RNHC=O$, $RC(O)C=O$ or RSO_2 and R is one of the groups listed in Table 3.
18. The process of claim 17 wherein R^2 is a P_4 - P_3 - P_2 moiety selected from one of the groups listed in Table 4.
19. The process of claim 14 wherein R^1 is methyl, ethyl, propyl, 2-propyl, butyl, pentyl, hexyl, 4-methylpentyl, 2-methylpropyl, cyclopentyl, cyclohexyl, cyclopentylmethyl, cyclohexylmethyl, phenylethyl, phenylpropyl, phenylbutyl, (d)-menthyl, (l)-menthyl, 1-adamantyl, 2-adamantyl, 1-indanyl, 2-indanyl, bornyl, 3-tetrahydrofuranyl, benzyl, α -methylbenzyl, 4-chlorobenzyl, 4-fluorobenzyl, 4-methylbenzyl, 4-(2-propyl)benzyl, or 4-trifluoromethylbenzyl.
20. The process of claim 14 wherein R^1 is ethyl or an optionally substituted benzyl.

21. The process of claim 20 wherein R^1 is ethyl or benzyl.
22. The process of claim 13 or 14 wherein N_3 -Y is selected from LiN_3 , NaN_3 , $TMS-N_3$, or $EtAlN_3$.
23. The process of claim 13 or 14 which proceeds through the aminolactone **IX**.
24. The process of claim 13 or 14 which proceeds through the iminophosphorane **XI**.
25. A process for making a compound of formula **I**:



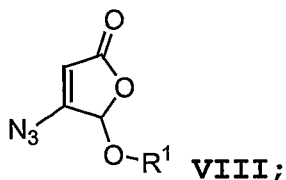
wherein R^1 is an optionally substituted group selected from an aliphatic group, aralkyl group, heterocyclylalkyl group or aryl group, and R^2 is an organic radical, said process comprising the steps of:

- (a) providing a butenolactone **VII**:

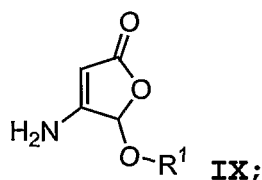


wherein X is chloro, bromo or iodo;

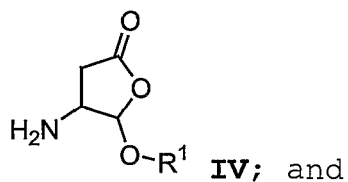
(b) treating **VII** with an azide N_3-Y , where Y is a silyl group or a counterion, to form an azidobutenolactone **VIII**:



(c) converting **VIII** to an aminobutenolactone **IX**:



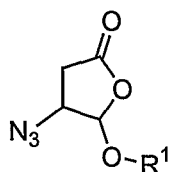
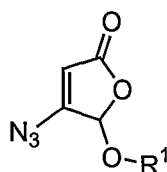
(d) reducing the ring double bond of **IX** to provide the aminolactone **IV**:



(e) coupling **IV** with R^2COOH , or a reactive equivalent thereof, to form **I**.

26. The process of claim 25 wherein R^2 is a $P_4-P_3-P_2$ moiety of a caspase inhibitor, or portion thereof.

27. A compound of formula **III** or **VIII**:

**III****VIII**

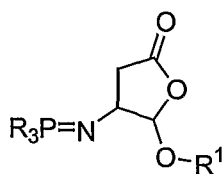
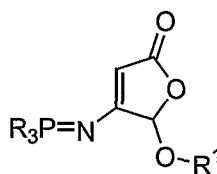
wherein R^1 is selected from an optionally substituted aliphatic group, aralkyl group or aryl group.

28. The compound of claim 27 wherein R^1 is methyl, ethyl, propyl, 2-propyl, butyl, pentyl, hexyl, 4-methylpentyl, 2-methylpropyl, cyclopentyl, cyclohexyl, cyclopentylmethyl, cyclohexylmethyl, phenylethyl, phenylpropyl, phenylbutyl, (d)-menthyl, (l)-menthyl, 1-adamantyl, 2-adamantyl, 1-indanyl, 2-indanyl, bornyl, 3-tetrahydrofuranyl, benzyl, α -methylbenzyl, 4-chlorobenzyl, 4-fluorobenzyl, 4-methylbenzyl, 4-(2-propyl)benzyl, or 4-trifluoromethylbenzyl.

29. The compound of claim 27 wherein R^1 is ethyl or an optionally substituted benzyl.

30. The compound of claim 29 wherein R^1 is ethyl or benzyl.

31. A compound of formula **V** or **IX**:

**V** or**IX**,

wherein R^1 is selected from an optionally substituted aliphatic group, aralkyl group or aryl group.

32. The compound of claim 31 wherein R¹ is methyl, ethyl, propyl, 2-propyl, butyl, pentyl, hexyl, 4-methylpentyl, 2-methylpropyl, cyclopentyl, cyclohexyl, cyclopentylmethyl, cyclohexylmethyl, phenylethyl, phenylpropyl, phenylbutyl, (d)-menthyl, (l)-menthyl, 1-adamantyl, 2-adamantyl, 1-indanyl, 2-indanyl, bornyl, 3-tetrahydrofuranyl, benzyl, α -methylbenzyl, 4-chlorobenzyl, 4-fluorobenzyl, 4-methylbenzyl, 4-(2-propyl)benzyl, or 4-trifluoromethylbenzyl.

33. The compound of claim 31 wherein R¹ is ethyl or an optionally substituted benzyl.

34. The compound of claim 33 wherein R¹ is ethyl or benzyl.