

OVERBASED ALKALINE EARTH METAL ALKYLHYDROXYBENZOATES HAVING LOW CRUDE SEDIMENT

The present invention relates to a process for the preparation of novel detergent-dispersant additives for lubricating oil applications for internal combustion engines. In particular, the process of the present invention provides middle to high TBN detergent-dispersant additives having very low crude sediment that when used in lubricating oil compositions is highly effective for the lubrication of mechanical components in land and marine engines.

BACKGROUND OF THE INVENTION

10 Overbased detergents are well described to provide lubricating properties. Often such detergent additives are proportioned with other lubricating additives to provide lubricating oil compositions that exhibit certain desired lubricating properties.

15 Alkaline-earth metal hydroxybenzoates are also known as additives for engine lubricating oils.

U.S. Patent No. 5,895,777 describes lubricating oil additives comprising the alkaline-earth metal salts of aromatic carboxylic hydroxy acids containing
20 carboxylic acids having 16 to 36 carbon atoms.

European Patent Application No. 1,154,012 describes lubricating compositions comprising an oil, an anti-wear additive and a sole oil-soluble overbased detergent comprising an aromatic carboxylate, such as a calcium salicylate
25 substituted by a hydrocarbon remainder.

British Patent No. 1,146,925 describes lubricating compositions comprising, as lubricating agents, polyvalent metal salts, in particular calcium, and alkylsalicylic acids comprising more than 12, preferably 14 to 18 carbon atoms in the alkyl

group. These salts can be prepared from the corresponding sodium salts, as synthesis intermediates.

British Patent No. 786,167 describes polyvalent metal salts of oil-soluble organic acids, such as sulfonic hydrocarbons, naphthenic acids or alkylhydroxybenzoic acids, in particular alkylsalicylic acids having an alkyl radical of up to 22 carbon atoms. The alkylsalicylic acids can be prepared from sodium alkylsalicylic acids according to the processes described in British Patents Nos. 734,598; 734,622 and 738,359. The sodium alkylsalicylates described in these British patents are useful as synthetic intermediates for the preparation of alkaline-earth alkylsalicylates, which are also useful as additives for lubricating oil.

In general, the above references describe processes for aromatic hydroxy carboxylic acids and their salts which are derived from alkaline salts of phenol derivatives, such as phenol itself, cresols, mono- and dialkylphenols, the alkyl group having from about 8 to 18 carbon atoms, halogenated phenols, aminophenols, nitrophenols, 1-naphthol, 2-naphthol, halogenated naphthols, and the like. The processes described above, however, lead to products having high sediment content at high TBN that decrease product yield and create added disposal expense. Thus, it is desirable to have a process that improves product yield by minimizing the sediment resulting from such processes.

SUMMARY OF THE INVENTION

The present invention provides middle to high overbased detergent-dispersant additives as lubricating oil additives employable in lubricating oil compositions for the lubrication of mechanical components in land and marine engines, such as, for example, hydraulic systems, transmissions, two-stroke and four-stroke vehicular engines, trunk piston and two stroke crosshead marine engines.

Accordingly, the present invention relates to a process for the preparation of novel detergent-dispersant additives having low crude sediment. More particularly, the present invention relates to a process for the preparation of

overbased alkaline earth metal alkylhydroxybenzoates, characterized in that the crude sediment is less than 3 volume %, preferably less than 2 volume % and more preferably less than 1 volume %.

- 5 In one embodiment, the present invention relates to a process for preparing an overbased alkaline earth metal alkylhydroxybenzoate comprising overbasing an alkaline earth metal alkylhydroxybenzoate or a mixture of alkaline earth metal alkylhydroxybenzoate and up to 50 mole % of alkylphenol, based on the total mixture of alkylhydroxybenzoate and alkylphenol, with a molar excess of
10 alkaline earth metal base and at least one acidic overbasing material in the presence of at least one carboxylic acid having from one to four carbon atoms and a solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof.

In another embodiment, the present invention relates to a process for
15 preparing an overbased alkaline earth metal alkylhydroxybenzoate obtained by the process comprising:

- a) Reacting alkylphenol with an alkali metal base to produce an alkali metal alkylphenate;
- 20 b) Carboxylating the alkali metal alkylphenate obtained in step a) with carbon dioxide so that at least 50 mole % of the starting alkylphenol has been converted to an alkali metal alkylhydroxybenzoate;
- c) Acidifying the alkali metal alkylhydroxybenzoate obtained in step b)
25 with an aqueous solution of a strong acid to produce an alkylhydroxybenzoic acid;
- d) Contacting the alkylhydroxybenzoic acid in step c) with at least one carboxylic acid having from about one to four carbon atoms;

5 e) Neutralizing the mixture of alkylhydroxybenzoic acid and the at least one carboxylic acid from step d) with an alkaline earth metal base and at least one solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof, to form an alkaline earth metal alkylhydroxybenzoate and at least one alkaline earth metal carboxylic acid salt; and

10 f) Overbasing the alkaline earth metal alkylhydroxybenzoate from step e) with a molar excess of alkaline earth metal base and at least one acidic overbasing material in the presence of the at least one alkaline earth metal carboxylic acid salt from step e) and a solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof.

15 In yet another embodiment, the present invention relates to a process for preparing an overbased alkaline earth metal alkylhydroxybenzoate obtained by the process comprising:

- 20 a) Reacting alkylphenol with an alkali metal base to produce an alkali metal alkylphenate;
- 25 b) Carboxylating the alkali metal alkylphenate obtained in step a) with carbon dioxide so that at least 50 mole % of the starting alkylphenol has been converted to an alkali metal alkylhydroxybenzoate;
- c) Acidifying the alkali metal alkylhydroxybenzoate obtained in step b) with an aqueous solution of a strong acid to produce an alkylhydroxybenzoic acid;
- d) Neutralizing the alkylhydroxybenzoic acid from step c) with a molar excess of an alkaline earth metal base and at least one solvent selected from the group consisting of aromatic hydrocarbons,

aliphatic hydrocarbons; monoalcohols, and mixtures thereof to form an alkaline earth metal alkylhydroxybenzoate;

- 5 e) Contacting the alkaline earth metal alkylhydroxybenzoate and alkaline earth metal base from step d) with at least one carboxylic acid having from about one to four carbon atoms in the presence of a solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof to form a mixture of alkaline earth metal alkylhydroxybenzoate and at least one alkaline earth metal carboxylic acid salt; and
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- 15 f) Overbasing the alkaline earth metal alkylhydroxybenzoate from step e) with a molar excess of alkaline earth metal base and at least one acidic overbasing material in the presence of the at least one alkaline earth metal carboxylic acid salt from step e) and a solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof.

The present invention is also directed to overbased alkaline earth metal alkylhydroxybenzoates produced by the processes of the present invention described above.

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Further, the present invention also relates to a lubricating oil composition comprising a major amount of a base oil of lubricating viscosity and a minor amount of the overbased alkali earth metal alkylhydroxybenzoate prepared by the processes described above.

- 25 Among other factors, the present invention is based on the surprising discovery that middle to high overbased alkaline earth metal alkylhydroxybenzoates obtained by overbasing an alkaline earth metal alkylhydroxybenzoate or a mixture of alkaline earth metal alkylhydroxybenzoate and up to 50 mole % of alkylphenol in the presence of least one carboxylic acid having from about one

to four carbon atoms and certain solvents lead to a very low amounts of crude sediments compared to a process without the use of the carboxylic acid. The detergent-dispersant additives prepared by the process of the present invention have improved low temperature viscosity and are effective for the

5 lubrication of mechanical components in land and marine engines, such as for example, hydraulic systems, transmissions, two-stroke and four-stroke vehicular engines, trunk piston and two-stroke crosshead marine engines. In particular, the detergent-dispersant additives of the present invention are useful in improving pumpability at low temperatures in automotive formulations. The

10 process of the present invention also significantly decreases the level of waste since lower crude sediments are produced which effectively lowers the cost of production.

DETAILED DESCRIPTION OF THE INVENTION

Prior to discussing the present invention in detail, the following terms will have

15 the following meanings unless expressly stated to the contrary.

Definitions

The term "alkali metal" or "alkaline metal" refers to lithium, sodium or potassium.

The term "alkaline earth metal" refers to calcium, barium, magnesium and

20 strontium.

The term "alkyl" refers to both straight- and branched-chain alkyl groups.

The term "alkylphenate" means a metal salt of an alkylphenol.

The term "alkylphenol" means a phenol having one or more alkyl substituents, wherein at least one of the alkyl substituents has a sufficient number of carbon

25 atoms to impart oil solubility to the phenol.

The term "aryl group" is a substituted or non-substituted aromatic group, such as the phenyl, tolyl, xylyl, ethylphenyl and cumenyl groups.

The term "calcium base" refers to a calcium hydroxide, calcium oxide, calcium alkoxides, and the like, and mixtures thereof.

- 5 The term "hydrocarbyl" means an alkyl or alkenyl group.

The term "hydrocarbyl phenol" refers to a phenol having one or more hydrocarbyl substituent; at least one of which has sufficient number of carbon atoms to impart oil solubility to the phenol.

- 10 The term "lime" refers to calcium hydroxide, also known as slaked lime or hydrated lime.

The term "metal" means alkali metals, alkaline earth metals, or mixtures thereof.

- 15 The term "metal base" refers to a metal hydroxide, metal oxide, metal alkoxides and the like and mixtures thereof, wherein the metal is selected from the group consisting of lithium, sodium, potassium, magnesium, calcium, strontium, barium or mixtures thereof.

- 20 The term "overbased" refers to a class of metal salts or complexes. These materials have also been referred to as "basic", "superbased", "hyperbased", "complexes", "metal complexes", "high-metal containing salts", and the like. Overbased products are metal salts or complexes characterized by a metal content in excess of that which would be present according to the stoichiometry of the metal and the particular acidic organic compound reacted with the metal, e.g., a carboxylic acid.

The term "phenate" means a metal salt of a phenol.

- 25 The term "Total Base Number" or "TBN" refers to the equivalent number of

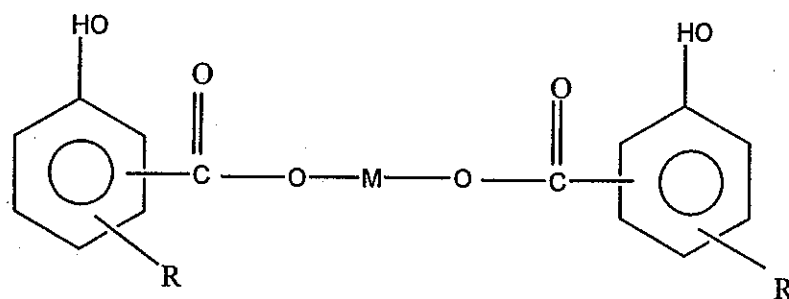
milligrams of KOH needed to neutralize 1 gram of a product. Therefore, a high TBN reflects strongly overbased products and, as a result, a higher base reserve for neutralizing acids. The TBN of a product can be determined by ASTM Standard No. D2896 or equivalent procedure.

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Overbased Alkaline Earth Metal Alkylhydroxybenzoate

The overbased alkaline earth metal alkylhydroxybenzoate of the present invention will typically have a structure as shown below as Formula (I).

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Formula (I)

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wherein R is a linear aliphatic group, branched aliphatic group or a mixture of linear and branched aliphatic groups. Preferably, R is an alkyl or alkenyl group. More preferably, R is an alkyl group.

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M is an alkaline earth metal selected of the group consisting of calcium, barium, magnesium, strontium. Calcium and magnesium are the preferred alkaline earth metal. Calcium is more preferred.

When R is a linear aliphatic group, the linear alkyl group typically comprises
25 from about 12 to 40 carbon atoms, more preferably from about 18 to 30 carbon atoms.

When R is a branched aliphatic group, the branched alkyl group typically comprises at least 9 carbon atoms, preferably from about 9 to 40 carbon atoms, more preferably from about 9 to 24 carbon atoms and most preferably from about 10 to 18 carbon atoms. Such branched aliphatic groups are
5 preferably derived from an oligomer of propylene or butene.

R can also represent a mixture of linear or branched aliphatic groups. Preferably, R represents a mixture of linear alkyl containing from about 20 to 30 carbon atoms and branched alkyl containing about 12 carbon atoms.

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When R represents a mixture of aliphatic groups, the alkaline-earth metal alkylhydroxybenzoic acid employed in the present invention may contain a mixture of linear groups, a mixture of branched groups, or a mixture of linear and branched groups. Thus, R can be a mixture of linear aliphatic groups,
15 preferably alkyl; for example, an alkyl group selected from the group consisting of C₁₄-C₁₆, C₁₆-C₁₈, C₁₈-C₂₀, C₂₀-C₂₂, C₂₀-C₂₄ and C₂₀-C₂₈ alkyl and mixtures thereof and derived from normal alpha olefins. Advantageously, these mixtures include at least 95 mole %, preferably 98 mole % of alkyl groups and originating from the polymerization of ethylene.

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The alkaline earth metal alkylhydroxybenzoates of the present invention wherein R represents a mixture of alkyl groups, can be prepared from linear alpha olefin cuts, such as those marketed by Chevron Phillips Chemical Company under the names Normal Alpha Olefin C₂₆-C₂₈ or Normal Alpha Olefin
25 C₂₀-C₂₄, by British Petroleum under the name C₂₀-C₂₆ Olefin, by Shell Chimie under the name SHOP C₂₀-C₂₂, or mixtures of these cuts or olefins from these companies having from about 20 to 28 carbon atoms.

The -COOM group of Formula (I) can be in the ortho, meta or para position
30 with respect to the hydroxyl group.

The alkaline earth metal alkylhydroxybenzoates of the present invention can be any mixture of alkaline-earth metal alkylhydroxybenzoic acid having the -

COOM group in the ortho, meta or para position.

The alkaline earth metal alkylhydroxybenzoates of the present invention are generally soluble in oil as characterized by the following test.

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A mixture of a 600 Neutral diluent oil and the alkylhydroxybenzoate at a content of 10 wt % with respect to the total weight of the mixture is centrifuged at a temperature of 60°C and for 30 minutes, the centrifugation being carried out under the conditions stipulated by the standard ASTM D2273 (it should be noted that centrifugation is carried out without dilution, i.e. without adding solvent); immediately after centrifugation, the volume of the deposit which forms is determined; if the deposit is less than 0.05 % v/v (volume of the deposit with respect to the volume of the mixture), the product is considered as soluble in oil.

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Advantageously, the TBN of the high overbased alkaline earth metal alkylhydroxybenzoate of the present invention is greater than 250, preferably from about 250 to 450 and more preferably from about 300 to 400 and will generally have less than 3 volume %, preferably less than 2 volume % and more preferably less than 1 volume % crude sediment. For the middle overbased alkaline earth metal alkylhydroxybenzoate of the present invention, the TBN is from about 100 to 250, preferably from about 140 to 230 and will generally have less than 1 volume %, preferably less than 0.5 volume % crude sediment.

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Process

In the first embodiment of the present invention, the process for preparing the overbased alkaline earth metal alkylhydroxybenzoate involves overbasing the alkaline earth metal alkylhydroxylbenzoate or a mixture of alkaline earth metal alkylhydroxylbenzoate and up to 50 mole % of alkylphenol, based on the total mixture of alkylhydroxybenzoate and alkylphenol, with a molar excess of alkaline earth metal base and at least one acidic overbasing material in

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presence of at least one carboxylic acid having from one to four carbon atoms and a solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof.

5 Overbasing of the alkaline earth metal alkylhydroxybenzoate or mixture of alkaline earth metal alkylhydroxybenzoate and alkylphenol may be carried out by any method known by a person skilled in the art to produce overbased alkaline earth metal alkylhydroxybenzoates. However, it has been surprisingly discovered that the addition of a small quantity of C₁-C₄ carboxylic acid at this step decreases the crude sediment obtained at the end of overbasing step by a
10 factor of at least 3.

The C₁-C₄ carboxylic acids used in the neutralization step include formic acid, acetic acid, propionic acid, and butyric acid, which may be used alone or in mixture. It is preferable to use mixtures of such acids as, for example, formic acid:acetic acid, in a molar ratio of formic acid:acetic acid of from about 0.1:1 to
15 100:1, preferably from about 0.5:1 to 4:1, more preferably from about 0.5:1 to 2:1, and most preferably about 1:1.

Generally, the overbasing reaction is carried out in a reactor in the presence of alkylhydroxybenzoic acid from about 10 wt % to 70 wt %, alkylphenol from about 1 wt % to 30 wt %, diluent oil from about 0 wt % to 40 wt %, an aromatic
20 solvent from about 20 wt % to 60 wt %. The reaction mixture is agitated. The alkaline earth metal associated with an aromatic solvent, a monoalcohol and carbon dioxide are added to the reaction while maintaining the temperature between about 20°C and 80°C.

25 The degree of overbasing may be controlled by the quantity of the alkaline earth metal, carbon dioxide and the reactants added to the reaction mixture and the reaction conditions used during the carbonation process.

The weight ratios of reagents used (methanol, xylene, slaked lime and CO₂) will correspond to the following weight ratios:

- Xylene:slaked lime from about 1.5:1 to 7:1, preferably from about 2:1 to 4:1.
- Methanol:slaked lime from about 0.25:1 to 4:1, preferably from about 0.4:1 to 1.2:1.
- Carbon dioxide:slaked lime from a molar ratio about 0.5:1 to 1.3:1, preferably from about 0.7:1 to 1.0:1.
- C₁-C₄ carboxylic acid:alkylhydroxybenzoic acid a molar ratio from about 0.02:1 to 1.5:1, preferably from about 0.1:1 to 0.7:1.

10 Lime is added as a slurry. i.e., as a pre-mixture of lime, methanol, xylene, and CO₂ is introduced over a period of 1 hour to 4 hours, at a temperature between about 20°C and 65°C.

The quantity of lime and CO₂ are adjusted in order to obtain a high overbased material (TBN > 250) and crude sediment in the range of 0.4 to 3 volume %, preferably in the range of 0.6 to 1.8 volume %, without any deterioration of the performance. With the omission of C₁-C₄ carboxylic acid, it is not able to reach this low level of crude sediment. Typically, crude sediment without a C₁-C₄ carboxylic acid will range from about 4 to 8 volume %.

20 For a middle overbased material (TBN from about 100 to 250), the quantity of lime and CO₂ are adjusted in order to obtain a crude sediment in the range of 0.2 to 1 volume %. The crude sediment without the use of C₁-C₄ carboxylic acid will range from about 0.8 to 3 volume %.

In a second embodiment of the present invention, the overbased alkaline earth metal alkylhydroxybenzoate may be prepared by the following steps:

25 A. Formation of the Alkali Metal Base Alkylphenate:

In the first step, alkylphenols are neutralized using an alkali metal base preferably in the presence of a light solvent, such as toluene, xylene isomers, light alkylbenzene or the like, to form the alkali metal base alkylphenate. In one

embodiment, the solvent forms an azeotrope with water. In another embodiment, the solvent may also be a mono-alcohol such as 2-ethylhexanol. In this case, the 2-ethylhexanol is eliminated by distillation before carboxylation. The objective with the solvent is to facilitate the elimination of water.

The hydrocarbyl phenols may contain up to 100 wt % linear hydrocarbyl groups, up to 100 wt % branched hydrocarbyl groups, or both linear and branched hydrocarbyl groups. Preferably, the linear hydrocarbyl group, if present, is alkyl, and the linear alkyl group contains from about 12 to 40 carbon atoms, more preferably from about 18 to 30 carbon atoms. The branched hydrocarbyl group, if present, is preferably alkyl and contains at least 9 carbon atoms, preferably from about 9 to 40 carbon atoms, more preferably from about 9 to 24 carbon atoms and most preferably from about 10 to 18 carbon atoms. In one embodiment, the hydrocarbyl phenols contain up to 85 wt % of linear hydrocarbyl phenol (preferably at least 35 wt % linear hydrocarbyl phenol) in mixture with at least 15 wt % of branched hydrocarbyl phenol. In one embodiment, the hydrocarbyl phenols are 100 % linear alkylphenols.

The use of an alkylphenol containing up to at least 35 wt % of long linear alkylphenol (from about 18 to 30 carbon atoms) is particularly attractive because a long linear alkyl chain promotes the compatibility and solubility of the additives in lubricating oils.

Branched alkylphenols can be obtained by reaction of phenol with a branched olefin, generally originating from propylene.

They consist of a mixture of monosubstituted isomers, the great majority of the substituents being in the para position, very few being in the ortho position, and hardly any in the meta position.

On the other hand, linear alkylphenols can be obtained by reaction of phenol with a linear olefin, generally originating from ethylene. They consist of a

- mixture of monosubstituted isomers in which the proportion of linear alkyl substituents in the ortho, meta, and para positions is much more uniformly distributed. Of course, linear alkylphenols may contain alkyl substituents with some branching which increases the amount of para substituents and,
- 5 resultantly may increase the relative reactivity towards alkali metal bases.

The alkali metal bases that can be used for carrying out this step include the oxides or hydroxides of lithium, sodium or potassium. In a preferred embodiment, potassium hydroxide is preferred. In another preferred embodiment, sodium hydroxide is preferred.

- 10 An objective of this step is to have an alkylphenate having less than 2000 ppm, preferably less than 1000 ppm and more preferably less than 500 ppm of water.

- In this regard, the first step is carried out at a temperature high enough to eliminate water. In one embodiment, the product is put under a slight vacuum
- 15 in order to require a lower reaction temperature.

In one embodiment, xylene is used as a solvent and the reaction conducted at a temperature between 130°C and 155°C, under an absolute pressure of 800 mbar (8×10^4 Pa).

- In another embodiment, 2-ethylhexanol is used as solvent. As the boiling point of 2-ethylhexanol (184°C) is significantly higher than xylene (140°C), the
- 20 reaction is conducted at a temperature of at least 150°C.

The pressure is reduced gradually below atmospheric in order to complete the distillation of water reaction. Preferably, the pressure is reduced to no more than 70 mbar (7×10^3 Pa).

- 25 By providing that operations are carried out at a sufficiently high temperature and that the pressure in the reactor is reduced gradually below atmospheric, the formation of the alkali metal base alkylphenate is carried out without the

need to add a solvent and forms an azeotrope with the water formed during this reaction. For instance, temperature is heated up to 200°C and then the pressure is reduced gradually below atmospheric. Preferably the pressure is reduced to no more than 70 mbar (7×10^3 Pa).

- 5 Elimination of water is done over a period of at least 1 hour, preferably at least 3 hours.

The quantities of reagents used should correspond to the following molar ratios:

- 10
- alkali metal base:alkylphenol from about 0.5:1 to 1.2:1, preferably from about: 0.9:1 to 1.05:1
 - solvent:alkylphenol (wt:wt) from about 0.1:1 to 5:1, preferably from about 0.3:1 to 3:1

B. Carboxylation:

- 15 This carboxylation step is conducted by simply bubbling carbon dioxide (CO_2) into the reaction medium originating from the preceding neutralization step and is continued until at least 50 mole % of the starting alkylphenol has been converted to alkylhydroxybenzoic acid (measured as hydroxybenzoic acid by potentiometric determination).
- 20 At least 50 mole %, preferably 75 mole %, and more preferably 85 mole %, of the starting alkylphenol is converted to alkylhydroxybenzoate using carbon dioxide at a temperature between about 110°C and 200°C under a pressure within the range of from about atmospheric to 15 bar (15×10^5 Pa), preferably from 1 bar (1×10^5 Pa) to 5 bar (5×10^5 Pa), for a period between about 1 and
- 25 8 hours.

In one variant with potassium salt, temperature is preferably between about 125°C and 165°C and more preferably between 130°C and 155°C, and the

pressure is from about atmospheric to 15 bar (15×10^5 Pa), preferably from about atmospheric to 4 bar (4×10^5 Pa).

In another variant with sodium salt, temperature is directionally lower preferably between from about 110°C and 155°C. More preferably from about 120°C and 140°C and the pressure from about 1 bar to 20 bar (1×10^5 to 20×10^5 Pa), preferably from 3 bar to 15 bar (3×10^5 to 15×10^5 Pa).

The carboxylation is usually carried out, diluted in a solvent such as hydrocarbons or alkylate, e.g., benzene, toluene, xylene and the like. In this case, the weight ratio of solvent:hydroxybenzoate is from about 0.1:1 to 5:1, preferably from about 0.3:1 to 3:1.

In another variant, no solvent is used. In this case, carboxylation is conducted in the presence of diluent oil in order to avoid a too viscous material.

The weight ratio of diluent oil:alkylhydroxybenzoate is from about 0.1:1 to 2:1, preferably from about 0.2:1 to 1:1, and more preferably from about 0.2:1 to 0.5:1.

C. Acidification:

The objective of this step is to acidify the alkylhydroxybenzoate salt diluted in the solvent to give an alkylhydroxybenzoic acid. Any acid stronger than alkylhydroxybenzoic acid could be utilized. Usually hydrochloric acid or aqueous sulfuric acid is utilized.

Acidification step is conducted with an H^+ equivalent excess of acid versus potassium hydroxide of at least 5 H^+ equivalent %, preferably 10 H^+ equivalent % and more preferably 20 H^+ equivalent %, the acidification is complete.

In one embodiment, sulfuric acid is used. It is diluted to about 5 volume % to 50 volume %, preferably 10 volume % to 30 volume %. The quantity of sulfuric acid used versus hydroxybenzoate (salicylate), on a per mole of

hydroxybenzoate basis, is at least 0.525 mole, preferably 0.55 mole and more preferably 0.6 mole of sulfuric acid.

5 The acidification reaction is carried out under agitation or with any suitable mixing system at a temperature from about room temperature to 95°C, preferably from about 50°C to 70°C, over a period linked with the efficiency of the mixing. For example, when a stirred reactor is utilized and the period is from about 15 minutes to 300 minutes, preferably from about 60 minutes to 180 minutes. When a static mixer is utilized, the period may be shorter.

10 At the end of this period time, the agitation is stopped in order to allow good phase separation before the aqueous phase was separated. After phase separation is complete, the organic phase is then neutralized, overbased, centrifugated to eliminate impurities and distilled to eliminate solvent. The water phase is treated as a waste material. In one embodiment, the organic phase is sent through a coalescer to decrease the level of residual water and
15 water-soluble impurities such as sulfuric acid and potassium sulfate as a consequence.

D. Contact with Carboxylic Acid:

The alkylhydroxybenzoic acid in step C is contacted with at least one carboxylic acid having from about one to four carbon atoms.

20 E. Neutralization:

The mixture of alkylhydroxybenzoic acid and the at least one carboxylic acid from step D is neutralized with an alkaline earth metal base and at least one solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons monoalcohols, and mixtures thereof to form an alkaline earth
25 metal alkylhydroxylbenzoate and at least one alkaline earth metal carboxylic acid salt.

F. Overbasing:

Overbasing of the mixture of alkylhydroxybenzoic acid and alkylphenol may be carried out by any method known by a person skilled in the art to produce alkylhydroxybenzoates. However, it has been surprisingly discovered that the
5 addition of a small quantity of C₁-C₄ carboxylic acid at this step decreases the crude sediment obtained at the end of overbasing step by a factor of at least 3.

The C₁-C₄ carboxylic acids used in the neutralization step include formic acid, acetic acid, propionic acid, and butyric acid, which may be used alone or in mixture. It is preferable to use mixtures of such acids as, for example, formic
10 acid:acetic acid, in a molar ratio of formic acid:acetic acid of from about 0.1:1 to 100:1, preferably from about 0.5:1 to 4:1, and more preferably from about 0.5:1 to 2:1.

Generally, the overbasing reaction is carried out in a reactor in the presence of alkylhydroxybenzoic acid from about 10 wt % to 70 wt %, alkylphenol from
15 about 1 wt % to 30 wt %, diluent oil from about 0 wt % to 40 wt %, an aromatic solvent from about 20 wt % to 60 wt %. The reaction mixture is agitated. The alkaline earth metal associated with an aromatic solvent, a monoalcohol and carbon dioxide are added to the reaction while maintaining the temperature between about 20°C and 80°C.

20 The degree of overbasing may be controlled by the quantity of the alkaline earth metal, carbon dioxide and the reactants added to the reaction mixture and the reaction conditions used during the carbonation process.

The weight ratios of reagents used (methanol, xylene, slaked lime and CO₂) will correspond to the following weight ratios:

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- Xylene:slaked lime from about 1.5:1 to 7:1, preferably from about 2:1 to 4:1.

- Methanol:slaked lime from about 0.25:1 to 4:1, preferably from about 0.4:1 to 1.2:1.
 - Carbon dioxide:slaked lime from a molar ratio about 0.5:1 to 1.3:1, preferably from about 0.7:1 to 1.0:1.
- 5 ▪ C₁-C₄ carboxylic acid:alkylhydroxybenzoic acid a molar ratio from about 0.02:1 to 1.5:1, preferably from about 0.1:1 to 0.7:1.

Lime is added as a slurry, i.e., as a pre-mixture of lime, methanol, xylene, and CO₂ is introduced over a period of 1 hour to 4 hours, at a temperature between about 20°C and 65°C.

- 10 The quantity of lime and CO₂ are adjusted in order to obtain a high overbased material (TBN > 250) and crude sediment in the range of 0.4 to 3 volume %, preferably in the range of 0.6 to 1.8 volume %, without any deterioration of the performance. With the omission of C₁-C₄ carboxylic acid, it is not able to reach this low level of crude sediment. Typically, crude sediment without a C₁-C₄
- 15 carboxylic acid will range from about 4 to 8 volume %.

For a middle overbased material (TBN from about 100 to 250), the quantity of lime and CO₂ are adjusted in order to obtain a crude sediment in the range of 0.2 to 1 volume %. The crude sediment without the use of C₁-C₄ carboxylic acid will range from about 0.8 to 3 volume %.

- 20 In the third embodiment of the present invention, the overbased alkaline earth metal alkylhydroxybenzoate may be obtained by a process having steps A through C above followed by:

D. Neutralization:

- 25 The mixture of alkylhydroxybenzoic acid from step C is neutralized with a molar excess of an alkaline earth metal base and at least one solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof to form an alkaline earth metal alkylhydroxybenzoate.

E. Contact with Carboxylic Acid:

The alkaline earth metal alkylhydroxybenzoate and alkaline earth metal base formed in step D is contacted with at least one carboxylic acid having from about one to four carbon atoms to form a mixture of alkaline earth metal alkylhydroxybenzoate and at least one alkaline earth metal carboxylate.

F. Overbasing:

The alkaline earth metal alkylhydroxybenzoate is then overbased according to the description provided above.

Optionally, predistillation, centrifugation and distillation may also be utilized to remove solvent and crude sediment. Water, methanol and a portion of the xylene may be eliminated by heating between about 110°C to 134°C. This may be followed by centrifugation to eliminate unreacted lime. Finally, xylene may be eliminated by heating under vacuum in order to reach a flash point of at least about 160°C as determined with the Pensky-Martens Closed Cup (PMCC) Tester described in ASTM D93.

Lubricating Oil Composition

The present invention also relates to lubricating oil compositions containing an overbased alkaline earth metal alkylhydroxybenzoate prepared by the process of the present invention. Such lubricating oil compositions will comprise a major amount of a base oil of lubricating viscosity and a minor amount of an overbased alkaline earth metal alkylhydroxybenzoate prepared by the process of the present invention having a TBN is from about 250 to 450, preferably from about 300 to 400, and a crude sediment of less than about 3 volume %, preferably less than about 2 volume %, more preferably less than about 1 volume %, in the case of a high overbased alkaline earth metal alkylhydroxybenzoate and in the case of a middle overbased alkaline earth metal alkylhydroxybenzoate having a TBN from about 100 to 250, preferably

from about 140 to 230, with a crude sediment of less than 1 volume %, and preferably less than 0.5 volume %.

Base oil as used herein is defined as a base stock or blend of base stocks
5 which is a lubricant component that is produced by a single manufacturer to the same specifications (independent of feed source or manufacturer's location); that meets the same manufacturer's specification; and that is identified by a unique formula, product identification number, or both. Base stocks may be manufactured using a variety of different processes including but not limited to
10 distillation, solvent refining, hydrogen processing, oligomerization, esterification, and rerefining. Rerefined stock shall be substantially free from materials introduced through manufacturing, contamination, or previous use. The base oil of this invention may be any natural or synthetic lubricating base oil fraction particularly those having a kinematic viscosity at 100° Centigrade
15 (C) and about 4 centistokes (cSt) to about 20 cSt. Hydrocarbon synthetic oils may include, for example, oils prepared from the polymerization of ethylene, polyalphaolefin or PAO, or from hydrocarbon synthesis procedures using carbon monoxide and hydrogen gases such as in a Fisher-Tropsch process. A preferred base oil is one that comprises little, if any, heavy fraction; e.g., little, if
20 any, lube oil fraction of viscosity about 20 cSt or higher at about 100°C. Oils used as the base oil will be selected or blended depending on the desired end use and the additives in the finished oil to give the desired grade of engine oil, e.g. a lubricating oil composition having an SAE Viscosity Grade of 0W, 0W-20, 0W-30, 0W-40, 0W-50, 0W-60, 5W, 5W-20, 5W-30, 5W-40, 5W-50, 5W-
25 60, 10W, 10W-20, 10W-30, 10W-40, 10W-50, 15W, 15W-20, 15W-30, or 15W-40.

The base oil may be derived from natural lubricating oils, synthetic lubricating oils or mixtures thereof. Suitable base oil includes base stocks obtained by
30 isomerization of synthetic wax and slack wax, as well as hydrocrackate base stocks produced by hydrocracking (rather than solvent extracting) the aromatic and polar components of the crude. Suitable base oils include those in all API

- categories I, II, III, IV and V as defined in API Publication 1509, 14th Edition, Addendum I, December 1998. Saturates levels and viscosity indices for Group I, II and III base oils are listed in Table I. Group IV base oils are polyalphaolefins (PAO). Group V base oils include all other base oils not
- 5 included in Group I, II, III, or IV. Group III base oils are preferred.

Table I.

SATURATES, SULFUR AND VISCOSITY INDEX OF GROUP I, II, III, IV
AND V BASE STOCKS

Group	Saturates (As determined by ASTM D2007) Sulfur (As determined by ASTM D2270)	Viscosity Index (As determined by ASTM D4294, ASTM D4297 or ASTM D3120)
I	Less than 90 % saturates and/or Greater than to 0.03 % sulfur	Greater than or equal to 80 and less than 120
II	Greater than or equal to 90 % saturates and less than or equal to 0.03 % sulfur	Greater than or equal to 80 and less than 120
III	Greater than or equal to 90 % saturates and less than or equal to 0.03% sulfur	Greater than or equal to 120
IV	All Polyalphaolefins (PAOs)	
V	All others not included in Groups I, II, III, or IV	

Natural lubricating oils may include animal oils, vegetable oils (e.g., rapeseed oils, castor oils and lard oil), petroleum oils, mineral oils, and oils derived from coal or shale.

Synthetic oils may include hydrocarbon oils and halo-substituted hydrocarbon oils such as polymerized and inter-polymerized olefins, alkylbenzenes, polyphenyls, alkylated diphenyl ethers, alkylated diphenyl sulfides, as well as their derivatives, analogues and homologues thereof, and the like. Synthetic
5 lubricating oils also include alkylene oxide polymers, interpolymers, copolymers and derivatives thereof wherein the terminal hydroxyl groups have been modified by esterification, etherification, etc. Another suitable class of synthetic lubricating oils comprises the esters of dicarboxylic acids with a variety of alcohols. Esters useful as synthetic oils also include those made from C₅ to C₁₂
10 monocarboxylic acids and polyols and polyol ethers. Tri-alkyl phosphate ester oils such as those exemplified by tri-n-butyl phosphate and tri-iso-butyl phosphate are also suitable for use as base oils.

Silicon-based oils (such as the polyakyl-, polyaryl-, polyalkoxy-, or polyaryloxy-
15 siloxane oils and silicate oils) comprise another useful class of synthetic lubricating oils. Other synthetic lubricating oils include liquid esters of phosphorus-containing acids, polymeric tetrahydrofurans, polyalphaolefins, and the like.

20 The base oil may be derived from unrefined, refined, rerefined oils, or mixtures thereof. Unrefined oils are obtained directly from a natural source or synthetic source (e.g., coal, shale, or tar sand bitumen) without further purification or treatment. Examples of unrefined oils include a shale oil obtained directly from a retorting operation, a petroleum oil obtained directly from distillation, or an
25 ester oil obtained directly from an esterification process, each of which may then be used without further treatment. Refined oils are similar to the unrefined oils except that refined oils have been treated in one or more purification steps to improve one or more properties. Suitable purification techniques include distillation, hydrocracking, hydrotreating, dewaxing, solvent extraction, acid or
30 base extraction, filtration, and percolation, all of which are known to those skilled in the art. Rerefined oils are obtained by treating used oils in processes similar to those used to obtain the refined oils. These rerefined oils are also

known as reclaimed or reprocessed oils and often are additionally processed by techniques for removal of spent additives and oil breakdown products.

5 Base oil derived from the hydroisomerization of wax may also be used, either alone or in combination with the aforesaid natural and/or synthetic base oil. Such wax isomerase oil is produced by the hydroisomerization of natural or synthetic waxes or mixtures thereof over a hydroisomerization catalyst.

10 It is preferred to use a major amount of base oil in the lubricating oil composition of the present invention. A major amount of base oil as defined herein comprises 40 wt % or more. Preferred amounts of base oil comprise from about 40 wt % to 97 wt %, preferably greater than from about 50 wt % to 97 wt %, more preferably from about 60 wt % to 97 wt % and most preferably from about 80 wt % to 95 wt % of the lubricating oil composition. (When weight
15 percent is used herein, it is referring to weight percent of the lubricating oil unless otherwise specified.)

The overbased alkaline earth metal alkylhydroxybenzoate produced by the process of the present invention in the lubricating oil composition will be in a minor amount compared to the base oil of lubricating viscosity. Generally, it will
20 be in an amount from about 1 to 15 wt %, preferably from about 2 to 12 wt % and more preferably from about 3 to 8 wt % based on the total weight of the lubricating oil composition.

Other Additive Components

25 The following additive components are examples of components that can be favorably employed in combination with the lubricating additive of the present invention. These examples of additives are provided to illustrate the present invention, but they are not intended to limit it.

(A) Ashless Dispersants: alkenyl succinimides, alkenyl succinimides modified with other organic compounds, and alkenyl succinimides modified with boric acid, alkenyl succinic ester.

(B) Oxidation Inhibitors:

- 5 1) Phenol type phenolic) oxidation inhibitors: 4,4'-methylenebis (2,6-di-tert-butylphenol), 4,4'-bis(2,6-di-tert-butylphenol), 4,4'-bis(2-methyl-6-tert-butylphenol), 2,2'-(methylenebis(4-methyl-6-tert-butylphenol), 4,4'-butylidenebis(3-methyl-6-tert-butylphenol), 4,4'-isopropylidenebis(2,6-di-tert-butylphenol), 2,2'-methylenebis(4-methyl-6-nonylphenol), 2,2'-isobutylidene-
10 bis(4,6-dimethylphenol), 2,2'-methylenebis(4-methyl-6-cyclohexylphenol), 2,6-di-tert-butyl-4-methylphenol, 2,6-di-tert-butyl-4-ethylphenol, 2,4-dimethyl-6-tert-butylphenol, 2,6-di-tert- α -dimethylamino-p-cresol, 2,6-di-tert-4(N.N'
dimethylaminomethylphenol), 4,4'-thiobis(2-methyl-6-tert-butylphenol), 2,2'-thiobis(4-methyl-6-tert-butylphenol), bis(3-methyl-4-hydroxy-5-tert-butylbenzyl)-
15 sulfide, and bis (3,5-di-tert-butyl-4-hydroxybenzyl).
- 2) Diphenylamine type oxidation inhibitor: alkylated diphenylamine, phenyl- α -naphthylamine, and alkylated α -naphthylamine.
- 3) Other types: metal dithiocarbamate (e.g., zinc dithiocarbamate), and methylenebis (dibutyldithiocarbamate).

20 (C) Rust Inhibitors (Anti-rust agents):

- 1) Nonionic polyoxyethylene surface active agents: polyoxyethylene lauryl ether, polyoxyethylene higher alcohol ether, polyoxyethylene nonylphenyl ether, polyoxyethylene octylphenyl ether, polyoxyethylene octyl stearyl ether, polyoxyethylene oleyl ether, polyoxyethylene sorbitol monostearate,
25 polyoxyethylene sorbitol mono-oleate, and polyethylene glycol monooleate.

- 2) Other compounds: stearic acid and other fatty acids, dicarboxylic acids, metal soaps, fatty acid amine salts, metal salts of heavy sulfonic acid, partial carboxylic acid ester of polyhydric alcohol, and phosphoric ester.
- (D) Demulsifiers: addition product of alkylphenol and ethyleneoxide,
5 polyoxyethylene alkyl ether, and polyoxyethylene sorbitane ester.
- (E) Extreme Pressure Agents (EP agents): zinc dialkyldithiophosphate (Zn-DTP, primary alkyl type & secondary alkyl type), sulfurized oils, diphenyl sulfide, methyl trichlorostearate, chlorinated naphthalene, benzyl iodide, fluoroalkylpolysiloxane, and lead naphthenate.
- 10 (F) Friction Modifiers: fatty alcohol, fatty acid, amine, borated ester, and other esters
- (G) Multifunctional Additives: sulfurized oxymolybdenum dithiocarbamate, sulfurized oxymolybdenum organo phosphorodithioate, oxymolybdenum monoglyceride, oxymolybdenum diethylate amide, amine-molybdenum
15 complex compound, and sulfur-containing molybdenum complex compound
- (H) Viscosity Index Improvers: polymethacrylate type polymers, ethylene-propylene copolymers, styrene-isoprene copolymers, hydrated styrene-isoprene copolymers, polyisobutylene, and dispersant type viscosity index improvers.
- 20 (I) Pour-point Depressants: polymethyl methacrylate.
- (K) Foam Inhibitors: alkyl methacrylate polymers and dimethyl silicone polymers.
- (L) Metal Detergents: sulfurized or unsulfurized alkyl or alkenyl phenates, alkyl or alkenyl aromatic sulfonates, calcium sulfonates, sulfurized or unsulfurized
25 metal salts of multi-hydroxy alkyl or alkenyl aromatic compounds, alkyl or alkenyl hydroxy aromatic sulfonates, sulfurized or unsulfurized alkyl or alkenyl

naphthenates, metal salts of alkanoic acids, metal salts of an alkyl or alkenyl multi-acid, and chemical and physical mixtures thereof.

EXAMPLES

The invention will be further illustrated by the following examples, which set forth particularly advantageous method embodiments. While the Examples are provided to illustrate the present invention, they are not intended to limit it.

Example 1

Overbased Alkaline Earth Metal Alkylhydroxybenzoate Having 350 TBN

A) Formation of the Alkali Metal Base Alkylphenate:

- 10 Alkylphenols (1000 g) prepared from mixtures of linear normal alpha olefins (C₂₀-C₂₈ alpha olefins from Chevron Phillips Chemical Company), xylene (500 g) was placed in a reactor and heated to 60°C over a period of 15 minutes then 290 g of an aqueous solution at 45 % KOH (2.325 mole) and 0.2 g of a defoamer called Rhodorsil 47V300 (commercialized by Rhodia) were added.
- 15 The reactor was then heated further to 145°C over a period of 2 hours while gradually decreasing the pressure from atmospheric pressure (1013 mbar absolute – 1×10^5 Pa) to 800 mbar absolute (8×10^4 Pa). Under these conditions, reflux begins and was maintained for 3 hours. During this period, approximately 179 ml of water was removed.

20 B) Carboxylation:

- The reactor containing the alkali metal alkylphenate from step A) was allowed to cool to 140°C. The reactor was then pressurized with CO₂ at 4 bar (4×10^5 Pa) (absolute pressure) and maintained under these conditions for 4 hours. At the end of this period, CO₂ was vacated to allow the reactor to reach
- 25 atmospheric pressure. At this step, 82 g of CO₂ was incorporated.

C) Acidification:

The alkali metal alkylhydroxybenzoate was reacted with a 20 molar % excess of a 10 % aqueous solution of sulfuric acid to convert it to an alkylhydroxybenzoic acid as follows:

- 5 A mixture of 140 g of sulfuric acid at 98 % and 1257 g of water in order to obtain 1397 g of a solution of sulfuric acid diluted at 10 % was placed in a 6 liter reactor and heated to 50°C under agitation at 250 rpm; the alkylhydroxybenzoate from step B) and xylene (1500 g) were loaded over a period of 30 minutes. Xylene assisted in phase separation. The reactor was
10 heated to and maintained at 60°C to 65°C for 2 hours with continued agitation at 250 rpm. At the end of this period, agitation was stopped, but the reactor was maintained at 60°C to 65°C for 2 hours to allow the phase separation to occur. Upon phase separation, the lower aqueous phase which contains water and potassium sulfate was decanted. The upper organic phase containing the
15 alkylhydroxybenzoic acid and xylene were collected for the following step. The concentration of alkylhydroxybenzoic acid was determined as an equivalent number of mg of KOH/g – V1, also known as the Total Acid Number (T.A.N.), as described in ASTM D664.

D) Neutralization:

- 20 The upper organic phase (3045 g) containing the alkylhydroxybenzoic acid (having a T.A.N. of 35 mg KOH/g) was loaded under agitation into a reactor over a 10 minute period. Then a slurry of methanol (573 g), lime (573 g) and xylene (735 g) was introduced. Due to the exothermic reaction, temperature increased from about 20°C to 28°C. Once the slurry was added, the reactor
25 was heated to 40°C over a period of 30 minutes and a mixture of formic acid (14.65 g):acetic acid (14.65 g) was added and allowed to react with the contents in the reactor. After a period of 5 minutes, the reactor was cooled to 30°C over a period of 30 minutes.

E) Overbasing:

Once the temperature of the reactor had cooled to 30°C, CO₂ (70.3 g) was introduced into the reactor at a flow rate of 1.37 g/minute over a period of 15 minutes then 171 g of CO₂ was introduced at a flow rate of 1.62 g/minute over a period of 105 minutes. A slurry of methanol (109 g), lime (109 g) and xylene (145 g) was added. Then additional CO₂ (128.4 g) was added over a period of 79 minutes at a flow rate of 1.62 g/minute. The reaction yielded an overbased alkaline earth metal alkylhydroxybenzoate. The percentage of crude sediment 1.2 volume % was determined at this step following the ASTM D2273 method.

10 F) Predistillation, Centrifugation, and Final Distillation:

The mixture contained within the reactor was taken in stages to a temperature between 65°C to 128°C over a period of 110 minutes. This procedure removed methanol, water and a portion of the xylene. Once 128°C was reached, diluent oil (775 g) was added. Crude sediment was then measured. The amount of 15 crude sediment in the overbased alkaline earth metal alkylhydroxybenzoate was very low (1.2 volume %).

The reaction mixture was centrifuged to remove crude sediment and then distilled at 204°C for 10 minutes under vacuum at 50 mbar absolute (50×10^2 Pa) to remove the remaining xylene.

20 Loads are in Table II and analyses in Table III.

Example 2

Overbased Alkaline Earth Metal Alkylhydroxybenzoate Having 200 TBN

The overbased alkaline earth metal alkylhydroxybenzoate having a 200 TBN was made following Example 1 except for the following changes to steps C to F of Example 1.

C) Acidification

At acidification step, a larger quantity of xylene is added: 2500 g instead of 1500 g.

D) Neutralization:

4045 g of the upper organic phase containing the alkylhydroxybenzoic acid (having a T.A.N. of 26.3 g KOH/g) was loaded under agitation into a reactor over a 10 minutes period. Then a slurry of methanol (267.0 g), lime (267.0 g) and xylene (649 g) was introduced. Due to the exothermic reaction, temperature increased from about 20°C to 28°C. Once the slurry was added, the reactor was heated to 40°C over a period of 30 minutes and a mixture of formic acid (11.8 g)/acetic acid (11.8 g) was added and allowed to react with the contents in the reactor. After a period of 5 minutes, the reactor was cooled to 25°C over a period of 30 minutes.

E) Overbasing:

Once the temperature of the reactor had cooled to 25°C, CO₂ (30.6 g) was introduced into the reactor at a flow rate of 0.74 g/minute as the temperature was increased from about 25°C to 40°C over a period of 95 minutes. The reaction yielded an overbased alkaline earth metal alkylhydroxybenzoate.

F) Predistillation, Centrifugation, and Final Distillation:

The mixture contained within the reactor was taken in stages to a temperature between 65°C to 128°C over a period of 110 minutes. This procedure removed methanol, water and a portion of the xylene. Once 128°C was reached, diluent
5 oil (573 g) was added. Crude sediment was then measured. The amount of crude sediment in the overbased alkaline earth metal alkylhydroxybenzoate was very low (0.2 volume %). The reaction mixture centrifuged to remove crude sediment and then distilled at 204°C for 10 minutes under vacuum at 50 mbar absolute (50×10^2 Pa) to remove the remaining xylene.

10 Loads are in Table II and analyses in Table III.

Example 3

Same process as Example 2 but a lower TBN (150) was attained.

See loads in Table II and analyses in Table III.

Example 4

15 Same process as Example 1 but at step A), KOH was replaced by NaOH on an equal molar basis and a higher quantity of lime is added at the overbasing step (step E).

See loads in Table II and analyses in Table III.

Example 5

20 A repeat of Example 1 except the loads in the neutralization, overbasing, and predistillation steps are different due mainly to a higher quantity of xylene at the acidification step (step C).

See loads in Table II and analyses in Table III.

Comparative Example A

Comparative Example A was prepared according to the procedure described for Example 1 except the mixture of formic acid acetic acid was not added. As
5 crude sediment is higher, a larger quantity of lime is added in order to reach the same TBN. The overbased alkaline earth metal alkylhydroxybenzoate prepared in the absence of the mixture of formic acid:acetic acid contained 6 volume % crude sediments.

Loads are in Table II and analyses in Table III.

10

Comparative Example B

Comparative Example B was prepared according to the procedure described for Example 5 except the mixture of formic acid acetic acid was not added. As
crude sediment is higher a larger quantity of lime is added in order to reach the same TBN. The overbased alkaline earth metal alkylhydroxybenzoate prepared
15 in the absence of the mixture of formic acid:acetic acid contained 6 volume % crude sediments.

Loads are in Table II and analyses in Table III.

Table II.

LOADS	Examples					Comparative Examples	
	1	2	3	4	5	A	B
A. Neutralization Step Linear Alkylphenol from CPC C20-C28 olefin - (g) - (mole) KOH/Alkylphenol (Molar Ratio) Xylene (g) KOH (diluted at 45 % water) (g) NaOH (diluted at 45 % water) (g) KOH (diluted at 45 % water) (mole) NaOH (diluted at 45 % water) (mole) Water eliminated (g)	1000 2.325 1 500 290 2.32 199	1000 2.325 1 500 290 2.32 199	1000 2.325 1 500 290 2.32 199	1000 2.325 500 207 2.32 154	1000 2.325 1 500 290 2.32 199	1000 2.325 1 500 290 2.32 199	1000 2.325 1 500 290 2.32 199
	82	82	82	82	82	82	82
B. Carboxylation CO ₂ (g)							
C. Acidification							
Xylene	1500	2500	1500	1500	2667	1500	2667
Sulfuric acid at 98 % (g)	140	140	140	140	140	140	140
Water (g)	1257	1257	1257	1257	1257	1257	1257
D. Neutralization/Overbasing carboxylic acid in xylene ^a (mg KOH/g) <u>First Slurry</u> Xylene Methanol Lime Formic acid Acetic acid CO ₂ <u>Second Slurry</u> Xylene Methanol Lime CO ₂ Diluent oil	3045 35 735 573 573 14.65 14.65 192	4045 26.3 649 267 267 11.8 11.8 70.6	4045 26.3 649 187 187 11.8 11.8 39.6	3045 35 735 573 573 14.65 14.65 192	1544 25.3 324 210 210 5.4 5.4 95	3045 35 735 573 573 0 0 192	1544 25.3 324 210 210 0 0 95
	145 109 109 128.4 775	0 0 0 0 573	0 0 0 0 501	145 163 163 173.5 835	100 40 40 21.8 384	145 163 163 154.7 857	100 40 40 21.8 384

^aNot the totality of Step 3, only one part.

Table III.

ANALYSIS ¹	Examples					Comparative Examples		
	1	2	3	4	5	A	B	
A. Neutralization step Conversion % alkylphenol (dialysis)	97	97	97	90	97	97	97	
B. Carboxylation CO ₂ (g) Acid hydroxybenzoic (mg KOH/g) V1 Alkylphenol + alkylphenate (mg KOH/g) V2	68.0 78.7 16.8	68.0 78.7 16.8	68.0 78.7 16.8	65.0 69.9 20.1	68.0 78.7 16.8	68.0 78.7 16.8	68.0 78.7 16.8	
C. Acidification Upper phase T.A.N. (mg KOH/g)	35.0	26.3	26.3	35.0	25.3	35.0	25.3	

Table III.

ANALYSIS ¹	Examples					Comparative Examples		
	1	2	3	4	5	A	B	
D. Overbasing								
Pre-distillation								
% Crude Sediment (128°C) (ASTM D2273)	1.2	0.2	0.2	1.6	1.2	6.0	6.0	
Final product								
Sediment after filtration (vol %) (ASTM D2273)	0.02	0.01	0.01	0.02	0.02	0.02	0.02	
Calcium (wt %)	12.5	7.23	5.52	13.03	12.42	12.19	11.6	
BN (ASTM D2896) (mg KOH/g)	360	202	155	365	348	341	326	
Viscosity at 100°C (mm ² /s) (ASTM D445)	260	101	90.2	462	151	173	111	
	200	193	194	192	193	195	196	
Flash point (PMCC) °C (ASTM D93)	28.0	40.6	41.4	26.4	28.6	29.0	28.9	

Table III.

ANALYSIS ¹	Examples					Comparative Examples		
	1	2	3	4	5	A	B	
<u>Composition thru dialysis</u> Hydroxybenzoates [expressed as hydroxybenzoic acid (wt %)] Alkylphenates [expressed as alkylphenol (wt %)] Unreacted alkylphenol (wt %) Calcium carbonate (wt %) Diluent oil (wt %) Calcium formiate + calcium acetate (wt%)	5.1	6.2	6.1	5.8	6.4	6.9	6.3	
	3.0	6.7	10.8	4.3	3.4	1.9	2.4	
	28.3	13.6	9.6	29.4	27.7	27.2	25.4	
	34.5	31.7	30.8	33.0	33.9	35.0	37	
	1.1	1.2	1.3	1.1	1.1	0	0	

¹ANALYTICAL DETERMINATION

A- NEUTRALIZATION OF ALKYLPHENOL

Conversion % alkylphenols

5 In a first step, the product obtained at the end of step A is dialyzed through a membrane: the phenate salt stays inside the membrane and after elimination of the solvent, it is weighted (M1).

Xylene and the unreacted alkylphenol move through the membrane xylene and the solvents utilized are eliminated by vaporization, a weight M2 is obtained.

$$\% \text{ Conversion} = \frac{M1}{M1 + M2} \times 100$$

10 B. CARBOXYLATION:

The product obtained at the end of step B is acidified by hydrochloric acid, it is titrated by tetra-n-butylammonium hydroxide.

Three inflexions points are observed:

15 - The first two inflexion points (V1, V2) correspond to the hydroxybenzoic acid, dicarboxylic acids and sulfurized benzoic acids.

- Third one V3 corresponds to alkylphenols + alkylphenate

V1, V2, V3 are expressed in mg KOH/g of product.

C. ACIDIFICATION STEP UP PHASE:

20 The level of hydroxybenzoic acid is determined through the method as above except no acidification by hydrochloric acid because the product has already been acidified by sulfuric acid.

Composition through dialysis

The method is the following:

1°) Dialysis of the final material

25• A "residue" (calcified part) stays inside the membrane

• Dialysate: non calcified part (unreacted alkylphenol and diluent oil) moves through the membrane

2°) Analysis of residue

30 It contained calcium carbonate, Ca phenate, Ca sulfurized phenate, Ca hydroxybenzoate and sulfurized Ca hydroxybenzoate. After elimination of solvent, the residue is weighted. After acidification, the quantity of phenate and hydroxybenzoate are determined through a potentiometric method.

35 Determination of calcium carbonate. A known quantity of final product is acidified, the organic phase contains hydroxybenzoic acid, alkylphenol and sulfurized derivatives thereof. After elimination of solvent (of this organic phase), the quantity of calcium carbonate is obtained by difference: weight of starting sample minus weight of this organic phase after elimination of solvent and correction.

3°) Analysis of dialysate

40 Diluent oil and alkylphenols go through a silica column to separate alkylphenols and diluent oil. Quantity of alkylphenols is determined by difference of weight.

PERFORMANCE

- Lubricating oil formulations (I and II) for automotive engine oil (AEO) applications were prepared with Example 5 and Comparative Example B as shown in Table IV. The additive composition from Example 5 and Comparative
- 5 Example B were added according to the wt % indicated in Table IV. Each formulation was examined in the ASTM D4684 MRV Test (Mini Rotary Viscometer Test) grade 5W30 at – 35°C.

- The ASTM D4684 MRV test is used to determine the viscosity of an oil after a 45-hour soak and cooling to test temperature by measuring the yield stress.
- 10 The test is used to evaluate pumpability and viscosity of engine oils at low temperatures. The test covers the measurement of the yield stress ($0 < \gamma < 35$ max) and viscosity (60,000 cp max) of the engine oils after cooling at controlled rates over a period not exceeding 45 hours to a final test temperature between -10°C and -40°C. In the MRV test, an engine oil sample is held at 80°C and
- 15 then cooled at a programmed cooling rate to a final test temperature. A low torque is applied to the rotor shaft to measure the yield stress. A higher torque is then applied to determine the apparent viscosity of the sample. The viscosity measurement is made at shear stress of 525 Pa over a shear rate of 0.4 to 15 s⁻¹.

Table IV

Component	Lubricating Oil Formulation	
	I	II
Polybutene bissuccinimide (wt %)	8	8
Zinc dithiophosphate (wt %)	1.08	1.08
Calcium sulfonate (wt %)	1.36	1.36
Oxidation inhibitor (wt %)	1.40	1.40
Product of the invention ^a		
Example 5 (wt %)	1.80	
Comparative Example B (wt %)		1.93
Corrosion inhibitor (wt %)	0.40	0.40
Antifoam agent (wt %)	0.0030	0.0030
Viscosity index improver (wt %)	6.00	6.00
Base oil/Group III (wt %) (Fortum) ^b	QSP100	QSP100
Performances analyses (ASTM 4684)		
MRV yield stress	0<y<35 pass	35<y<70 fail
MRV viscosity (cP)	30855	33780

^aQSP – Quantity sufficient to provide 100 wt %

^bQuantity of overbased calcium hydroxybenzoate load is such as it provides 56 millimole calcium per kg of the Formulations I or II.

5 For Example 5, the calculation is the following:

$$\frac{40.08 \times 56}{10000} = 0.224 \text{ quantity of calcium required per liter}$$

$$\frac{0.224 \times 100}{12.42 \%} = 1.80\% \text{ of overbased calcium hydroxybenzoate of Example 5.}$$

The results of Table IV indicate that the additive composition of the present invention comprising a calcium overbased alkylhydroxybenzoate detergent which contains at least one carboxylate salt, having from one to four carbon atoms, improves viscosity at low temperature versus a calcium overbased

5 alkylhydroxybenzoate detergent which does not contain a least one carboxylate salt having from one to four carbon atoms

The calcium overbased alkylhydroxybenzoate is utilized in the AEO formulation at a level from about 15 to 200 millimoles calcium per kg of the formulation.

CLAIMS

1. A process for preparing an overbased alkaline earth metal alkylhydroxybenzoate, said process comprising overbasing an alkaline earth metal alkylhydroxybenzoate or a mixture of alkaline earth metal alkylhydroxybenzoate and up to 50 mole % of alkylphenol, based on the total mixture of alkylhydroxybenzoate and alkylphenol, with a molar excess of alkaline earth metal base and at least one acidic overbasing material in the presence of at least one carboxylic acid having from one to four carbon atoms and a solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof.
2. The process according to Claim 1, wherein the process is carried out in the absence of sulfur.
3. The process according to claim 1 or 2, wherein the process results in an overbased alkaline earth metal alkylhydroxybenzoate having less than 3 volume % crude sediment.
4. The process according to any one of the preceding claims, wherein the process results in an overbased alkaline earth metal alkylhydroxybenzoate having less than 2 volume % crude sediment.
5. The process according to any one of the preceding claims, wherein the process results in an overbased alkaline earth metal alkylhydroxybenzoate having less than 1 volume % crude sediment.

6. The process according to any one of the preceding claims, wherein the overbased alkaline earth metal alkylhydroxybenzoate has a TBN from about 250 to 450.
7. The process according to any one of the preceding claims, wherein the overbased alkaline earth metal alkylhydroxybenzoate has a TBN from about 300 to 400.
8. The process according to any one of the preceding claims, wherein the overbased alkaline earth metal alkylhydroxybenzoate has a TBN from about 100 to 250.
9. The process according to any one of the preceding claims, wherein the overbased alkaline earth metal alkylhydroxybenzoate has a TBN from about 140 to 230.
10. The process according to any one of the preceding claims, wherein the alkaline earth metal is calcium or magnesium.
11. The process according to any one of the preceding claims, wherein the alkaline earth metal is calcium.
12. The process according to any one of the preceding claims, wherein the alkyl group of the alkylphenol is a linear or branched alkyl or a mixture of linear and branched alkyl groups.
13. The process according to any one of the preceding claims, wherein the alkyl group of the alkylphenol is a linear alkyl group having from about 12 to 40 carbon atoms.

14. The process according to any one of the preceding claims, wherein the alkyl group of the alkylphenol is a linear alkyl group having from about 12 to 40 carbon atoms derived from the polymerization of ethylene.
15. The process according to any one of the preceding claims, wherein the alkyl group of the alkylphenol is a linear alkyl group having from about 18 to 30 carbon atoms.
16. The process according to any one of claims 1 to 12, wherein the alkyl group of the alkylphenol is a branched alkyl group having at least 9 carbon atoms.
17. The process according to any one of claims 1 to 12 and 16 wherein the alkyl group of the alkylphenol is a branched alkyl group having from about 9 to 40 carbon atoms.
18. The process according to any one of claims 1 to 12, 16 and 17, wherein the alkyl group of the alkylphenol is a branched alkyl group having from about 9 to 24 carbon atoms.
19. The process according to any one of claims 1 to 14, wherein the alkyl group of the alkylphenol is selected from the group consisting of C₁₄-C₁₆, C₁₆-C₁₈, C₁₈-C₃₀, C₂₀-C₂₂, C₂₀-C₂₄, C₂₀-C₂₆ and C₂₀-C₂₈ alkyl and mixtures thereof, and is derived from normal alpha olefins.
20. The process according to any one of claims 1 to 12, wherein the alkyl group of the alkylphenol is a mixture of linear alkyl having from about 20 to 30 carbon atoms and branched alkyl containing about 12 carbon atoms.
21. The process according to any one of the preceding claims, wherein the alkylphenol is up to 25 mole % of the total mixture of alkaline earth metal alkylhydroxybenzoate and alkylphenol.

22. The process according to any one of the preceding claims, wherein the alkylphenol is up to 15 mole % of the total mixture of alkaline earth metal alkylhydroxybenzoate and alkylphenol.
23. The process according to any one of the preceding claims, wherein the acidic overbasing material is carbon dioxide.
24. The process according to any one of the preceding claims, wherein the carboxylic acid is selected from the group consisting of acetic acid, formic acid, propionic acid, butyric acid, and mixtures thereof.
25. The process according to any one of the preceding claims, wherein the carboxylic acid is formic acid, acetic acid or mixtures thereof.
26. The process according to any one of the preceding claims, wherein the carboxylic acid is a mixture of formic acid and acetic acid.
27. The process according to claim 26, wherein the mixture of formic acid and acetic acid is in a weight ratio from about 0.1:1 to 100:1, formic to acetic acid.
28. The process according to claim 26 or 27, wherein the mixture of formic acid and acetic acid is in a weight ratio from about 0.5:1 to 4:1, formic to acetic acid.
29. The process according to any one of claims 26 to 28, wherein the mixture of formic acid and acetic acid is in a weight ratio of from about 0.5:1 to 2:1, formic to acetic acid.
30. The process according to any one of claims 26 to 29, wherein the mixture of formic acid and acetic acid is in a weight ratio of about 1:1.

31. The process according to any one of the preceding claims, wherein the solvent is selected from the group consisting of xylene, methanol, toluene, cyclohexane, 2-ethylhexanol, and mixtures thereof.
32. The process according to any one of the preceding claims, wherein the solvent is xylene, methanol, 2-ethylhexanol or mixtures thereof.
33. The process according to any one of the preceding claims, wherein the solvent is a mixture of xylene and methanol.
34. A product produced by the process of any one of the preceding claims.
35. A process for preparing an overbased alkaline earth metal alkylhydroxybenzoate, said process comprising:
 - a) Reacting alkylphenol with an alkali metal base to produce an alkali metal alkylphenate;
 - b) Carboxylating the alkali metal alkylphenate obtained in step a) with carbon dioxide so that at least 50 mole % of the starting alkylphenol has been converted to an alkali metal alkylhydroxybenzoate;
 - c) Acidifying the alkali metal alkylhydroxybenzoate obtained in step b) with an aqueous solution of a strong acid to produce an alkylhydroxybenzoic acid;
 - d) Contacting the alkylhydroxybenzoic acid in step c) with at least one carboxylic acid having from about one to four carbon atoms;
 - e) Neutralizing the mixture of alkylhydroxybenzoic acid and the at least one carboxylic acid from step d) with an alkaline earth metal base and at least one solvent selected from the group consisting of aromatic

hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof, to form an alkaline earth metal alkylhydroxybenzoate and at least one alkaline earth metal carboxylic acid salt; and

- f) Overbasing the alkaline earth metal alkylhydroxybenzoate from step (e) with a molar excess of alkaline earth metal base and at least one acidic overbasing material in the presence of the at least one alkaline earth metal carboxylic acid salt from step e) and a solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof.
- 36. The process of claim 35, further comprising eliminating the excess solvents and crude sediments.
 - 37. The process according to claim 35 or 36, wherein the process is carried out in the absence of sulfur.
 - 38. The process according to any one of claims 35 to 37, wherein the process results in an overbased alkaline earth metal alkylhydroxybenzoate having less than 3 volume % crude sediment.
 - 39. The process according to any one of claims 35 to 38, wherein the process results in an overbased alkaline earth metal alkylhydroxybenzoate having less than 2 volume % crude sediment.
 - 40. The process according to any one of claims 35 to 39, wherein the process results in an overbased alkaline earth metal alkylhydroxybenzoate having less than 1 volume % crude sediment.
 - 41. The process according to any one of claims 35 to 40, wherein the overbased alkaline earth metal alkylhydroxybenzoate has a TBN from about 250 to 450.

42. The process according to any one of claims 35 to 41, wherein the overbased alkaline earth metal alkylhydroxybenzoate has a TBN from about 300 to 400.
43. The process according to any one of claims 35 to 42, wherein the overbased alkaline earth metal alkylhydroxybenzoate has a TBN from about 100 to 250.
44. The process according to any one of claims 35 to 43, wherein the overbased alkaline earth metal alkylhydroxybenzoate has a TBN from about 140 to 230.
45. The process according to any one of claims 35 to 44, wherein the alkaline earth metal is calcium or magnesium.
46. The process according to any one of claims 35 to 45, wherein the alkaline earth metal is calcium.
47. The process according to any one of claims 35 to 46, wherein the alkyl group of the alkylphenol is a linear or branched alkyl or a mixture of linear and branched alkyl groups.
48. The process according to any one of claims 35 to 47, wherein the alkyl group of the alkylphenol is a linear alkyl group having from about 12 to 40 carbon atoms.
49. The process according to any one of claims 35 to 48, wherein the alkyl group of the alkylphenol is a linear alkyl group having from about 12 to 40 carbon atoms derived from the polymerization of ethylene.

50. The process according to any one of claims 35 to 49, wherein the alkyl group of the alkylphenol is a linear alkyl group having from about 18 to 30 carbon atoms.
51. The process according to any one of claims 35 to 47, wherein the alkyl group of the alkylphenol is a branched alkyl group having at least 9 carbon atoms.
52. The process according to any one of claims 35 to 47 and 51 wherein the alkyl group of the alkylphenol is a branched alkyl group having from about 9 to 40 carbon atoms.
53. The process according to any one of claims 35 to 47, 51 and 52, wherein the alkyl group of the alkylphenol is a branched alkyl group having from about 9 to 24 carbon atoms.
54. The process according to any one of claims 35 to 49, wherein the alkyl group of the alkylphenol is selected from the group consisting of C₁₄-C₁₆, C₁₆-C₁₈, C₁₈-C₃₀, C₂₀-C₂₂, C₂₀-C₂₄, C₂₀-C₂₆ and C₂₀-C₂₈ alkyl and mixtures thereof, and is derived from normal alpha olefins.
55. The process according to any one of claims 35 to 47, wherein the alkyl group of the alkylphenol is a mixture of linear alkyl having from about 20 to 30 carbon atoms and branched alkyl containing about 12 carbon atoms.
56. The process according to any one of claims 35 to 55, wherein the alkylphenol is up to 25 mole % of total mixture of alkaline earth metal alkylhydroxybenzoate and alkylphenol.
57. The process according to any one of claims 35 to 56, wherein the alkylphenol is up to 15 mole % of total mixture of alkaline earth metal alkylhydroxybenzoate and alkylphenol.

58. The process according to any one of claims 35 to 57, wherein the acidic overbasing material is carbon dioxide.
59. The process according to any one of claims 35 to 58, wherein the carboxylic acid is selected from the group consisting of acetic acid, formic acid, propionic acid, butyric acid, and mixtures thereof.
60. The process according to any one of claims 35 to 59, wherein the carboxylic acid is formic acid, acetic acid or mixtures thereof.
61. The process according to any one of claims 35 to 60, wherein the carboxylic acid is a mixture of formic acid and acetic acid.
62. The process according to claim 61, wherein the mixture of formic acid and acetic acid is in a weight ratio from about 0.1:1 to 100:1, formic to acidic acid.
63. The process according to claim 61 or 62, wherein the mixture of formic acid and acetic acid is in a weight ratio from about 0.5:1 to 4:1, formic to acidic acid.
64. The process according to claim 61, 62 or 63 wherein the mixture of formic acid and acetic acid is in a weight ratio of from about 0.5:1 to 2:1, formic to acetic acid.
65. The process according to any one of claims 61 to 64, wherein the mixture of formic acid and acetic acid is in a weight ratio of about 1:1.
66. The process according to any one of claims 35 to 65, wherein the solvent is selected from the group consisting of xylene, methanol, toluene, cyclohexane, 2-ethylhexanol, and mixtures thereof.

67. The process according to any one of claims 35 to 66, wherein the solvent is xylene, methanol, 2-ethylhexanol, or mixtures thereof.
68. The process according to any one of claims 35 to 67, wherein the solvent is a mixture of xylene and methanol.
69. A product produced by the process of any one of claims 35 to 68.
70. A process for preparing an overbased alkaline earth metal alkylhydroxybenzoate, said process comprising:
- a) Reacting alkylphenol with an alkali metal base to produce an alkali metal alkylphenate;
 - b) Carboxylating the alkali metal alkylphenate obtained in step a) with carbon dioxide so that at least 50 mole % of the starting alkylphenol has been converted to an alkali metal alkylhydroxybenzoate;
 - c) Acidifying the alkali metal alkylhydroxybenzoate obtained in step b) with an aqueous solution of a strong acid to produce an alkylhydroxybenzoic acid;
 - d) Neutralizing the alkylhydroxybenzoic acid from step c) with a molar excess of an alkaline earth metal base and at least one solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols and mixtures thereof to form an alkaline earth metal alkylhydroxybenzoate;
 - e) Contacting the alkaline earth metal alkylhydroxybenzoate and alkaline earth metal base from step d) with at least one carboxylic acid having from about one to four carbon atoms in the presence of a solvent selected from the group consisting of aromatic hydrocarbons, aliphatic

hydrocarbons, monoalcohols, and mixtures thereof, to form a mixture of alkaline earth metal alkylhydroxybenzoate and at least one alkaline earth metal carboxylic acid salt; and

- f) Overbasing the alkaline earth metal alkylhydroxybenzoate from step e) with a molar excess of alkaline earth metal base and at least one acidic overbasing material in the presence of the at least one alkaline earth metal carboxylic acid salt from step e) and a solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof.
- 71. The process of claim 70, further comprising eliminating the excess solvents and crude sediments.
 - 72. The process according to any one of claims 70 to 71, wherein the process is carried out in the absence of sulfur.
 - 73. The process according to any one of claims 70 to 72, wherein the process results in an overbased alkaline earth metal alkylhydroxybenzoate having less than 3 volume % crude sediment.
 - 74. The process according to any one of claims 70 to 73, wherein the process results in an overbased alkaline earth metal alkylhydroxybenzoate having less than 2 volume % crude sediment.
 - 75. The process according to any one of claims 70 to 74, wherein the process results in an overbased alkaline earth metal alkylhydroxybenzoate having less than 1 volume % crude sediment.
 - 76. The process according to any one of claims 70 to 75, wherein the overbased alkaline earth metal alkylhydroxybenzoate has a TBN from about 250 to 450.

77. The process according to any one of claims 70 to 76, wherein the overbased alkaline earth metal alkylhydroxybenzoate has a TBN from about 300 to 400.
78. The process according to any one of claims 70 to 77, wherein the overbased alkaline earth metal alkylhydroxybenzoate has a TBN from about 100 to 250.
79. The process according to any one of claims 70 to 78, wherein the overbased alkaline earth metal alkylhydroxybenzoate has a TBN from about 140 to 230.
80. The process according to any one of claims 70 to 79, wherein the alkaline earth metal is calcium or magnesium.
81. The process according to any one of claims 70 to 80, wherein the alkaline earth metal is calcium.
82. The process according to any one of claims 70 to 81, wherein the alkyl group of the alkylphenol is a linear or branched alkyl or a mixture of linear and branched alkyl groups.
83. The process according to any one of claims 70 to 82, wherein the alkyl group of the alkylphenol is a linear alkyl group having from about 12 to 40 carbon atoms.
84. The process according to any one of claims 70 to 83, wherein the alkyl group of the alkylphenol is a linear alkyl group having from about 12 to 40 carbon atoms derived from the polymerization of ethylene.

85. The process according to any one of claims 70 to 84, wherein the alkyl group of the alkylphenol is a linear alkyl group having from about 18 to 30 carbon atoms.
86. The process according to any one of claims 70 to 82, wherein the alkyl group of the alkylphenol is a branched alkyl group having at least 9 carbon atoms.
87. The process according to any one of claims 70 to 82 and 86 wherein the alkyl group of the alkylphenol is a branched alkyl group having from about 9 to 40 carbon atoms.
88. The process according to any one of claims 70 to 82, 86 and 87, wherein the alkyl group of the alkylphenol is a branched alkyl group having from about 9 to 24 carbon atoms.
89. The process according to any one of claims 70 to 84, wherein the alkyl group of the alkylphenol is selected from the group consisting of C₁₄-C₁₆, C₁₆-C₁₈, C₁₈-C₃₀, C₂₀-C₂₂, C₂₀-C₂₄, C₂₀-C₂₆ and C₂₀-C₂₈ alkyl and mixtures thereof, and is derived from normal alpha olefins.
90. The process according to any one of claims 70 to 82, wherein the alkyl group of the alkylphenol is a mixture of linear alkyl having from about 20 to 30 carbon atoms and branched alkyl containing about 12 carbon atoms.
91. The process according to any one of claims 70 to 90, wherein the alkylphenol is up to 25 mole % of total mixture of alkaline earth metal alkylhydroxybenzoate and alkylphenol.
92. The process according to any one of claims 70 to 91, wherein the alkylphenol is up to 15 mole % of total mixture of alkaline earth metal alkylhydroxybenzoate and alkylphenol.

93. The process according to any one of claims 70 to 92, wherein the acidic overbasing material is carbon dioxide.
94. The process according to any one of claims 70 to 93, wherein the carboxylic acid is selected from the group consisting of acetic acid, formic acid, propionic acid, butyric acid, and mixtures thereof.
95. The process according to any one of claims 70 to 94, wherein the carboxylic acid is formic acid, acetic acid or mixtures thereof.
96. The process according to any one of claims 70 to 95, wherein the carboxylic acid is a mixture of formic acid and acetic acid.
97. The process according to claim 96, wherein the mixture of formic acid and acetic acid is in a weight ratio from about 0.1:1 to 100:1, formic to acetic acid.
98. The process according to any one of claim 96 or 97, wherein the mixture of formic acid and acetic acid is in a weight ratio from about 0.5:1 to 4:1, formic to acetic acid.
99. The process according to any one of claims 96 to 98, wherein the mixture of formic acid and acetic acid is in a weight ratio of from about 0.5:1 to 2:1, formic to acetic acid.
100. The process according to any one of claims 96 to 99, wherein the mixture of formic acid and acetic acid is in a weight ratio of about 1:1.
101. The process according to any one of claims 70 to 100, wherein the solvent is selected from the group consisting of xylene, methanol, toluene, cyclohexane, 2-ethylhexanol, and mixtures thereof.

102. The process according to any one of claims 70 to 101, wherein the solvent is xylene, methanol, 2-ethylhexanol, or mixtures thereof.
103. The process according to any one of claims 70 to 102, wherein the solvent is a mixture of xylene and methanol.
104. A product produced by the process of any one of claims 70 to 103.
105. A lubricating oil composition comprising a major amount of a base oil of lubricating viscosity and a minor amount of an overbased alkaline earth metal alkylhydroxybenzoate prepared by the process comprising overbasing an alkaline earth metal alkylhydroxybenzoate or a mixture of alkaline earth metal alkylhydroxybenzoate and up to 50 mole % of alkylphenol, based on the total mixture of alkylhydroxybenzoate and alkylphenol, with a molar excess of alkaline earth metal base and carbon dioxide in presence of at least one carboxylic acid having from one to four carbon atoms and a solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof.
106. A lubricating oil composition comprising a major amount of a base oil of lubricating viscosity and a minor amount of an overbased alkaline earth metal alkylhydroxybenzoate prepared by the process comprising:
- a) Reacting alkylphenol with an alkali metal base to produce an alkali metal alkylphenate;
 - b) Carboxylating the alkali metal alkylphenate obtained in step a) with carbon dioxide so that at least 50 mole % of the starting alkylphenol has been converted to an alkali metal alkylhydroxybenzoate;

- c) Acidifying the alkali metal alkylhydroxybenzoate obtained in step b) with an aqueous solution of a strong acid to produce an alkylhydroxybenzoic acid;
 - d) Contacting the alkylhydroxybenzoic acid in step c) with at least one carboxylic acid having from about one to four carbon atoms;
 - e) Neutralizing the mixture of alkylhydroxybenzoic acid and the at least one carboxylic acid from step d) with an alkaline earth metal base and at least one solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof, to form an alkaline earth metal alkylhydroxybenzoate and at least one alkaline earth metal carboxylic acid salt; and
 - f) Overbasing the alkaline earth metal alkylhydroxybenzoate from step e) with a molar excess of alkaline earth metal base and at least one acidic overbasing material in the presence of the at least one alkaline earth metal carboxylic acid salt from step e) and a solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof.
107. A lubricating oil composition comprising a major amount of a base oil of lubricating viscosity and a minor amount of an overbased alkaline earth metal alkylhydroxybenzoate prepared by the process comprising:
- a) Reacting alkylphenol with an alkali metal base to produce an alkali metal alkylphenate;
 - b) Carboxylating the alkali metal alkylphenate obtained in step a) with carbon dioxide so that at least 50 mole % of the starting alkylphenol has been converted to an alkali metal alkylhydroxybenzoate;

- c) Acidifying the alkali metal alkylhydroxybenzoate obtained in step b) with an aqueous solution of a strong acid to produce an alkylhydroxybenzoic acid;
- d) Neutralizing the alkylhydroxybenzoic acid from step c) with a molar excess of an alkaline earth metal base and at least one solvent selected from the group consisting of aromatic hydrocarbons; aliphatic hydrocarbons, monoalcohols, and mixtures thereof, to form an alkaline earth metal alkylhydroxybenzoate;
- e) Contacting the alkaline earth metal alkylhydroxybenzoate and alkaline earth metal base from step d) with at least one carboxylic acid having from about one to four carbon atoms in the presence of a solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof, to form a mixture of alkaline earth metal alkylhydroxybenzoate and at least one alkaline earth metal carboxylic acid salt; and
- f) Overbasing the alkaline earth metal alkylhydroxybenzoate from step e) with a molar excess of alkaline earth metal base and at least one acidic overbasing material in the presence of the at least one alkaline earth metal carboxylic acid salt from step e) and a solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof.

108. A method of improving the low temperature pumpability of a lubricating oil composition in an internal combustion engine, said method comprising operating the internal combustion engine with a lubricating oil composition comprising a major amount of a base oil of lubricating viscosity and a minor amount of an overbased alkaline earth metal

alkylhydroxybenzoate prepared by the process comprising overbasing an alkaline earth metal alkylhydroxybenzoate or a mixture of alkaline earth metal alkylhydroxybenzoate and up to 50 mole % of alkylphenol, based on the total mixture of alkylhydroxybenzoate and alkylphenol, with a molar excess of alkaline earth metal base and carbon dioxide in presence of at least one carboxylic acid having from one to four carbon atoms and a solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols and mixtures thereof.

109. A method of improving the low temperature pumpability of a lubricating oil composition in an internal combustion engine, said method comprising operating the internal combustion engine with a lubricating oil composition comprising comprising a major amount of a base oil of lubricating viscosity and a minor amount of an overbased alkaline earth metal alkylhydroxybenzoate prepared by the process comprising:
- a) Reacting alkylphenol with an alkali metal base to produce an alkali metal alkylphenate;
 - b) Carboxylating the alkali metal alkylphenate obtained in step a) with carbon dioxide so that at least 50 mole % of the starting alkylphenol has been converted to an alkali metal alkylhydroxybenzoate;
 - c) Acidifying the alkali metal alkylhydroxybenzoate obtained in step b) with an aqueous solution of a strong acid to produce an alkylhydroxybenzoic acid;
 - d) Contacting the alkylhydroxybenzoic acid in step c) with at least one carboxylic acid having from about one to four carbon atoms;

- e) Neutralizing the mixture of alkylhydroxybenzoic acid and the at least one carboxylic acid from step d) with an alkaline earth metal base and at least one solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof, to form an alkaline earth metal alkylhydroxybenzoate and at least one alkaline earth metal carboxylic acid salt; and
- f) Overbasing the alkaline earth metal alkylhydroxybenzoate from step e) with a molar excess of alkaline earth metal base and at least one acidic overbasing material in the presence of the at least one alkaline earth metal carboxylic acid salt from step e) and a solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols and mixtures thereof.

110. A method of improving the low temperature pumpability of a lubricating oil composition in an internal combustion engine, said method comprising operating the internal combustion engine with a lubricating oil composition comprising a major amount of a base oil of lubricating viscosity and a minor amount of an overbased alkaline earth metal alkylhydroxybenzoate prepared by the process comprising:

- a) Reacting alkylphenol with an alkali metal base to produce an alkali metal alkylphenate;
- b) Carboxylating the alkali metal alkylphenate obtained in step a) with carbon dioxide so that at least 50 mole % of the starting alkylphenol has been converted to an alkali metal alkylhydroxybenzoate;
- c) Acidifying the alkali metal alkylhydroxybenzoate obtained in step b) with an aqueous solution of a strong acid to produce an alkylhydroxybenzoic acid;

- d) Neutralizing the alkylhydroxybenzoic acid from step c) with a molar excess of an alkaline earth metal base and at least one solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof, to form an alkaline earth metal alkylhydroxybenzoate;
- e) Contacting the alkaline earth metal alkylhydroxybenzoate and alkaline earth metal base from step d) with at least one carboxylic acid having from about one to four carbon atoms in the presence of a solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols, and mixtures thereof, to form a mixture of alkaline earth metal alkylhydroxybenzoate and at least one alkaline earth metal carboxylic acid salt; and
- f) Overbasing the alkaline earth metal alkylhydroxybenzoate from step e) with a molar excess of alkaline earth metal base and at least one acidic overbasing material in the presence of the at least one alkaline earth metal carboxylic acid salt from step e) and a solvent selected from the group consisting of aromatic hydrocarbons, aliphatic hydrocarbons, monoalcohols and mixtures thereof.