

Coated wire

The invention relates to a coated wire comprising a silver-based wire core and a coating layer superimposed on the surface of the wire core. The invention further relates to a process for manufacturing such coated wire.

The use of bonding wires in electronics and microelectronics applications is well-known state of the art. While bonding wires were made from gold in the beginning, nowadays less expensive materials are used such as copper, copper alloys, silver and silver alloys. Such wires may have a metal coating.

With respect to wire geometry, most common are bonding wires of circular cross-section and bonding ribbons which have a more or less rectangular cross-section. Both types of wire geometries have their advantages making them useful for specific applications.

It is an object of the invention to provide a coated silver-based wire suitable for use in wire bonding applications, the wire apart from satisfying fundamental requirements like wire softness, feasibility of FAB (free air ball) formation under air atmosphere and corrosion as well oxidation resistance, also being outstanding in particular in prevention of flowery bonded ball, in prevention of wire twist on unwinding from spool as well as in stable looping behaviour with regard to wire sway.

A contribution to the solution of said object is provided by the subject-matter of the category-forming claims. The dependent sub-claims of the category-forming claims represent preferred embodiments of the invention, the subject-matter of which also makes a contribution to solving the objects mentioned above.

The invention relates to a wire comprising a wire core (hereinafter also called “core” for short) with a surface, the wire core having a coating layer superimposed on its surface, wherein the wire core itself is a silver-based wire core, wherein the coating layer is a double-layer comprised of a 1 to 100 nm thick inner layer of palladium or nickel and an adjacent 1 to 250 nm thick outer layer of gold, wherein the wire exhibits at least one of the following intrinsic properties A1) to A3) (see “Test methods A and B” as described below):

A1) the average grain size of the crystal grains in the wire core, measured in longitudinal direction, is in the range of from 0.7 to 1.1 μm ;

A2) the fraction of twin boundaries, measured in longitudinal direction of the wire, is in the range of from 5 to 40 %;

A3) 20 to 70 % of the crystal grains of the wire core are oriented in $\langle 100 \rangle$ direction, and 3 to 40 % of the crystal grains of the wire core are oriented in $\langle 111 \rangle$ direction, each % with respect to the total number of crystal grains with orientation parallel to the drawing direction of the wire.

The term “intrinsic property” is used herein with regard to the wire of the invention. Intrinsic property means a property which a wire has of itself (independently of other factors), while an extrinsic property depends on the wire’s relationship or interaction with other external factors.

The wire of the invention is preferably a bonding wire for bonding in microelectronics. It is preferably a one-piece object. For the invention, the term “bonding wire” comprises bonding wires with circular cross-section and thin diameters. The average cross-section is in the range of, for example, from 50 to 5024 μm^2 or preferably 110 to 2400 μm^2 ; accordingly, the average diameter is in the range of, for example, from 8 to 80 μm or preferably 12 to 55 μm .

The average diameter or, simply stated, the diameter of a wire or wire core can be obtained by the “sizing method”. According to this method the physical weight of the wire for a defined length is determined. Based on this weight, the diameter of a wire or wire core is calculated using the density of the wire material. The diameter is calculated as arithmetic mean of five measurements on five cuts of a particular wire.

The wire core is a silver-based wire core ; i.e. the wire core consists of a silver-based material in the form of (a) doped silver, (b) a silver alloy or (c) a doped silver alloy.

The term “doped silver” used herein means a silver-based material consisting of (a1) silver in an amount in the range of from > 99.49 to 99.997 wt.-%, (a2) at least one doping element other than silver in a total amount of from 30 to < 5000 wt.-ppm and (a3) further components (components other than silver and the at least one doping element) in a total amount of from 0 to 100 wt.-ppm. In a preferred embodiment, the term “doped silver” used herein means doped silver consisting of (a1) silver in an amount in the range of from > 99.49 to 99.997 wt.-%, (a2) at least one doping element selected from the group consisting of calcium, nickel, platinum, palladium, gold, copper, rhodium and ruthenium in a total amount of from 30 to < 5000 wt.-ppm and (a3) further components (components other than silver, calcium, nickel,

platinum, palladium, gold, copper, rhodium and ruthenium) in a total amount of from 0 to 100 wt.-ppm.

The term “silver alloy” used herein means a silver-based material consisting of (b1) silver in an amount in the range of from 89.99 to 99.5 wt.-%, preferably 97.99 to 99.5 wt.-%, (b2) at least one alloying element in a total amount in the range of from 0.5 to 10 wt.-%, preferably 0.5 to 2 wt.-% and (b3) further components (components other than silver and the at least one alloying element) in a total amount of from 0 to 100 wt.-ppm. In a preferred embodiment, the term “silver alloy” used herein means a silver alloy consisting of (b1) silver in an amount in the range of from 89.99 to 99.5 wt.-%, preferably 97.99 to 99.5 wt.-%, (b2) at least one alloying element selected from the group consisting of nickel, platinum, palladium, gold, copper, rhodium and ruthenium in a total amount in the range of from 0.5 to 10 wt.-%, preferably 0.5 to 2 wt.-% and (b3) further components (components other than silver, nickel, platinum, palladium, gold, copper, rhodium and ruthenium) in a total amount of from 0 to 100 wt.-ppm.

The term “doped silver alloy” used herein means a silver-based material consisting of (c1) silver in an amount in the range of from > 89.49 to 99.497 wt.-%, preferably 97.49 to 99.497 wt.-%, (c2) at least one doping element in a total amount of from 30 to < 5000 wt.-ppm, (c3) at least one alloying element in a total amount in the range of from 0.5 to 10 wt.-%, preferably 0.5 to 2 wt.-% and (c4) further components (components other than silver, the at least one doping element and the at least one alloying element) in a total amount of from 0 to 100 wt.-ppm, wherein the at least one doping element (c2) is other than the at least one alloying element (c3). In a preferred embodiment, the term “doped silver alloy” used herein means a doped silver alloy consisting of (c1) silver in an amount in the range of from > 89.49 to 99.497 wt.-%, preferably 97.49 to 99.497 wt.-%, (c2) at least one doping element selected from the group consisting of calcium, nickel, platinum, palladium, gold, copper, rhodium and ruthenium in a total amount of from 30 to < 5000 wt.-ppm, (c3) at least one alloying element selected from the group consisting of nickel, platinum, palladium, gold, copper, rhodium and ruthenium in a total amount in the range of from 0.5 to 10 wt.-%, preferably 0.5 to 2 wt.-% and (c4) further components (components other than silver, calcium, nickel, platinum, palladium, gold, copper, rhodium and ruthenium) in a total amount of from 0 to 100 wt.-ppm, wherein the at least one doping element (c2) is other than the at least one alloying element (c3).

This disclosure mentions “further components” and “doping elements”. The individual amount of any further component is less than 30 wt.-ppm. The individual amount of any

doping element is at least 30 wt.-ppm. All amounts in wt.-% and wt.-ppm are based on the total weight of the core or its precursor item or elongated precursor item.

The core of the wire of the invention may comprise so-called further components in a total amount in the range of from 0 to 100 wt.-ppm, for example, 10 to 100 wt.-ppm. In the present context, the further components, often also referred as “inevitable impurities”, are minor amounts of chemical elements and/or compounds which originate from impurities present in the raw materials used or from the wire core manufacturing process. The low total amount of 0 to 100 wt.-ppm of the further components ensures a good reproducibility of the wire properties. Further components present in the core are usually not added separately. Each individual further component is comprised in an amount of less than 30 wt.-ppm, based on the total weight of the wire core.

The core of the wire is a homogeneous region of bulk material. Since any bulk material always has a surface region which might exhibit different properties to some extent, the properties of the core of the wire are understood as properties of the homogeneous region of bulk material. The surface of the bulk material region can differ in terms of morphology, composition (e.g. sulfur, chlorine and/or oxygen content) and other features. The surface is an interface region between the wire core and the coating layer superimposed on the wire core. Typically, the coating layer is completely superimposed on the wire core's surface. In the region of the wire between its core and the coating layer superimposed thereon a combination of materials of both, the core and the coating layer, can be present.

The coating layer superimposed on the surface of the wire core is a double-layer comprised of a 1 to 100 nm thick, preferably 1 to 30 nm thick inner layer of palladium or nickel and an adjacent 1 to 250 nm thick, preferably 20 to 200 nm thick outer layer of gold. In this context the term “thick” or “coating layer thickness” means the size of the coating layer in perpendicular direction to the longitudinal axis of the core.

The wire of the invention exhibits at least one of the following intrinsic properties A1) to A3):

A1) the average grain size of the crystal grains in the wire core, measured in longitudinal direction, is in the range of from 0.7 to 1.1 μm ;

A2) the fraction of twin boundaries, measured in longitudinal direction of the wire, is in the range of from 5 to 40 %, preferably 10 to 40 %;

A3) 20 to 70 %, preferably 30 to 70 %, most preferably 30 to 60 % of the crystal grains of the wire core are oriented in $\langle 100 \rangle$ direction, and 3 to 40 %, preferably 3 to 30 %, most preferably 5 to 20 % of the crystal grains of the wire core are oriented in $\langle 111 \rangle$ direction,

each % with respect to the total number of crystal grains with orientation parallel to the drawing direction of the wire.

It is preferred that the wire of the invention exhibits at least two of the intrinsic properties A1) to A3). It is more preferred that the wire of the invention exhibits all of the intrinsic properties A1) to A3).

In an advantageous embodiment, the gold layer comprises at least one member selected from the group consisting of antimony, bismuth, arsenic and tellurium in a total proportion in the range of from 10 to 300 wt.-ppm, preferably 10 to 150 wt.-ppm, based on the weight of the wire of the invention. At the same time, in an embodiment, the total proportion of the at least one member selected from the group consisting of antimony, bismuth, arsenic and tellurium may be in the range of from 300 to 9500 wt.-ppm, preferably 300 to 5000 wt.-ppm, most preferably 600 to 3000 wt.-ppm, based on the weight of the gold of the gold layer.

Within the advantageous embodiment, it is preferred that antimony is present within the gold layer and it is even more preferred that antimony is alone present within the gold layer, i.e. without the simultaneous presence of bismuth, arsenic and tellurium. In other words, in a most preferred variant of the advantageous embodiment, the gold layer comprises antimony in a proportion in the range of from 10 to 300 wt.-ppm, preferably 10 to 150 wt.-ppm, based on the weight of the wire (wire core plus coating layer), without bismuth, arsenic and tellurium being present within the gold layer; at the same time, in an embodiment, the proportion of the antimony may be in the range of from 300 to 9500 wt.-ppm, preferably 300 to 5000 wt.-ppm, most preferably 600 to 3000 wt.-ppm, based on the weight of the gold of the gold layer.

In an embodiment, the at least one member selected from the group consisting of antimony, bismuth, arsenic and tellurium may exhibit a concentration gradient within the gold layer, said gradient increasing in the direction towards the wire core, i.e. in perpendicular direction to the longitudinal axis of the wire core.

In another aspect, the invention relates also to a process for the manufacture of the coated wire of the invention in any of its embodiments disclosed above. The process comprises at least the steps (1) to (5):

- (1) providing a silver-based precursor item,
- (2) elongating the precursor item to form an elongated precursor item, until an intermediate diameter in the range of from 30 to 200 μm is obtained,

(3) applying a double-layer coating of an inner layer of palladium or nickel and an adjacent outer layer of gold on the surface of the elongated precursor item obtained after completion of process step (2),

(4) further elongating the coated precursor item obtained after completion of process step (3) until a desired final diameter and a double-layer comprised of an inner layer of palladium or nickel having a desired final thickness in the range of from 1 to 100 nm and an adjacent outer layer of gold having a desired final thickness in the range of from 1 to 250 nm is obtained, and

(5) finally strand annealing the coated precursor obtained after completion of process step (4) at an oven set temperature in the range of from >400 to $460\text{ }^{\circ}\text{C}$ for an exposure time in the range of from ≥ 0.8 to 10 seconds, preferably ≥ 0.8 to 2 seconds to form the coated wire, wherein step (2) may include one or more sub-steps of intermediate batch annealing of the precursor item at an oven set temperature of from 200 to $650\text{ }^{\circ}\text{C}$ for an exposure time in the range of from 30 to 300 minutes, preferably at an oven set temperature of from 300 to $500\text{ }^{\circ}\text{C}$ for an exposure time in the range of from 60 to 180 minutes.

The term “strand annealing” is used herein. It is a continuous process allowing for a fast production of a wire with high reproducibility. In the context of the invention, strand annealing means that the annealing is done dynamically while the coated precursor to be annealed is pulled or moved through a conventional annealing oven and spooled onto a reel after having left the annealing oven. Here, the annealing oven is typically in the form of a cylindrical tube of a given length. With its defined temperature profile at a given annealing speed which may be chosen in the range of, for example, from 10 to 60 meters/minute the annealing time/oven temperature parameters can be defined and set.

The term “oven set temperature” is used herein. It means the temperature fixed in the temperature controller of the annealing oven. The annealing oven may be a chamber furnace type oven (in case of batch annealing) or a tubular annealing oven (in case of strand annealing).

This disclosure distinguishes between precursor item, elongated precursor item, coated precursor item, coated precursor and coated wire. The term “precursor item” is used for those wire pre-stages which have not reached the desired final diameter of the wire core, while the term “precursor” is used for a wire pre-stage at the desired final diameter. After completion of process step (5), i.e. after the final strand annealing of the coated precursor at the desired final diameter a coated wire in the sense of the invention is obtained.

The precursor item as provided in process step (1) is a silver-based precursor item; i.e. the precursor item consists of (a) doped silver, (b) a silver alloy or (c) a doped silver alloy. As regards the meaning of the terms “doped silver”, “silver alloy” and “doped silver alloy” reference is made to the afore made disclosure.

In the embodiment of a silver-based precursor item the latter can be obtained by alloying, doping or alloying and doping silver with the desired amount of the required components. The doped silver or silver alloy or doped silver alloy can be prepared by conventional processes known to the person skilled in the art of metal alloys, for example, by melting together the components in the desired proportional ratio. In doing so, it is possible to make use of one or more conventional master alloys. The melting process can for example be performed making use of an induction furnace and it is expedient to work under vacuum or under an inert gas atmosphere. The materials used can have a purity grade of, for example, 99.99 wt.-% and above. The melt so-produced can be cooled to form a homogeneous piece of silver-based precursor item. Typically, such precursor item is in the form of a rod having a diameter of, for example, 2 to 25 mm and a length of, for example, 2 to 100 m. Such rod can be made by continuous casting the silver-based melt using an appropriate mold, followed by cooling and solidifying.

In process step (2) the silver-based precursor item is elongated to form an elongated precursor item, until an intermediate diameter in the range of from 30 to 200 μm is obtained. Techniques to elongate a precursor item are known and appear useful in the context of the invention. Preferred techniques are rolling, swaging, die drawing or the like, of which die drawing is particularly preferred. In the latter case the precursor item is drawn in several process steps until the desired intermediate cross-section or the desired intermediate diameter is reached. Such wire die drawing process is well known to the person skilled in the art. Conventional tungsten carbide and diamond drawing dies may be employed and conventional drawing lubricants may be employed to support the drawing.

Step (2) of the process of the invention may include one or more sub-steps of intermediate batch annealing of the elongated precursor item at an oven set temperature in the range of from 200 to 650 $^{\circ}\text{C}$ for an exposure time in the range of from 30 to 300 minutes, preferably at an oven set temperature of from 300 to 500 $^{\circ}\text{C}$ for an exposure time in the range of from 60 to 180 minutes. Said optional intermediate batch annealing may be performed, for example, with a rod drawn to a diameter of 2 mm and coiled on a drum.

The optional intermediate batch annealing of process step (2) may be performed under an inert or reducing atmosphere. Numerous types of inert atmospheres as well as reducing atmospheres are known in the art and are used for purging the annealing oven. Of the known inert atmospheres, nitrogen or argon is preferred. Of the known reducing atmospheres, hydrogen is preferred. Another preferred reducing atmosphere is a mixture of hydrogen and nitrogen. Preferred mixtures of hydrogen and nitrogen are 90 to 98 vol.-% nitrogen and, accordingly, 2 to 10 vol.-% hydrogen, wherein the vol.-% total 100 vol.-%. Preferred mixtures of nitrogen/hydrogen are equal to 93/7, 95/5 and 97/3 vol.-%/vol.-%, each based on the total volume of the mixture.

In process step (3) a coating in the form of a double-layer coating of an inner layer of palladium or nickel and an adjacent outer layer of gold is applied on the surface of the elongated precursor item obtained after completion of process step (2) so as to superimpose the coating over said surface.

The skilled person knows how to calculate the thickness of such coating on an elongated precursor item to finally obtain the coating in the layer thickness disclosed for the embodiments of the wire, i.e. after finally elongating the coated precursor item. The skilled person knows numerous techniques for forming a coating layer of a material according to the embodiments on a silver-based surface. Preferred techniques are plating, such as electroplating and electroless plating, deposition of the material from the gas phase such as sputtering, ion plating, vacuum evaporation and physical vapor deposition, and deposition of the material from the melt. In case of applying said double-layer comprised of inner palladium or nickel layer and outer gold layer, it is preferred to apply the palladium or nickel layer by electroplating.

The gold layer is also preferably applied by electroplating. Gold electroplating is performed making use of a gold electroplating bath, i.e. an electroplating bath that allows for a palladium or a nickel cathode surface to be electroplated with gold. In other words, the gold electroplating bath is a composition allowing for direct application of gold in elemental, metallic form onto a palladium or a nickel surface wired as cathode.

Electroplating application of the gold layer is performed by guiding the palladium- or nickel-coated elongated precursor item wired as a cathode through the gold electroplating bath. The so obtained gold-coated precursor item exiting the gold electroplating bath may be rinsed and dried before process step (4) is performed. The use of water as a rinsing medium is expedient, with alcohol and alcohol/water mixtures being further examples of rinsing

media. The gold electroplating of the palladium- or nickel-coated elongated precursor item passing through the gold electroplating bath can take place at a direct voltage in the range of, for example, 0.2 to 20 V at a current in the range of, for example, 0.001 to 5 A, in particular 0.001 to 1 A or 0.001 to 0.2 A. Typical contact times may be in the range of, for example, 0.1 to 30 seconds, preferably 2 to 8 seconds. The current densities used in this context can be in the range of, for example, 0.01 to 150 A/dm². The gold electroplating bath may have a temperature in the range of, for example, 45 to 75°C, preferably 55 to 65°C.

The thickness of the gold coating layer can be adjusted as desired essentially via the following parameters: chemical composition of the gold electroplating bath, contact time of the elongated precursor item with the gold electroplating bath, current density. In this context, the thickness of the gold layer can generally be increased by increasing the concentration of the gold in the gold electroplating bath, by increasing the contact time of the elongated precursor item wired as cathode and the gold electroplating bath, and by increasing the current density.

In an embodiment, the process of the invention is a process for the manufacture of the coated wire of the invention in its afore disclosed advantageous embodiment. Here, the application of the gold layer in step (3) is performed by electroplating it from a gold electroplating bath comprising gold and at least one member selected from the group consisting of antimony, bismuth, arsenic and tellurium. Hence, in said embodiment, the gold electroplating bath is a composition allowing for the deposition of not only the elemental gold but also allowing for depositing said at least one member selected from the group consisting of antimony, bismuth, arsenic and tellurium within the gold layer. It is unknown what chemical species said at least one member is, i.e. whether it is present in the gold layer in elemental form or as a chemical compound. In said embodiment, the gold electroplating bath can be made by adding said at least one member in a suitable chemical form (e.g. compounds like Sb₂O₃, BiPO₄, As₂O₃ or TeO₂) to an aqueous composition containing gold as dissolved salt or dissolved salts. Examples of such aqueous compositions into which the at least one member can be added are Aurocor® K 24 HF made by Atotech and Auruna® 558 and Auruna® 559 made by Umicore. Alternatively, one can use a gold electroplating bath which already comprises at least one member selected from the group consisting of antimony, bismuth, arsenic and tellurium, like, for example, a gold electroplating bath based on MetGold Pure ATF made by Metalor. The concentration of the gold in the gold electroplating bath can be in the range of, for example, 8 to 40 g/l (grams per liter), preferably 10 to 20 g/l. The concentration of the at least one member selected from the

group consisting of antimony, bismuth, arsenic and tellurium in the gold electroplating bath can be in the range of, for example, 15 to 50 wt.-ppm, preferably 15 to 35 wt.-ppm.

In process step (4) the coated precursor item obtained after completion of process step (3) is further elongated until (4) a desired final diameter of the wire having a double-layer comprised of an inner layer of palladium or nickel having a desired final thickness in the range of from 1 to 100 nm, preferably 1 to 30 nm, and an adjacent outer layer of gold having a desired final thickness in the range of from 1 to 250 nm, preferably 20 to 200 nm, is obtained. Techniques to elongate the coated precursor item are the same elongation techniques like those mentioned above in the disclosure of process step (2).

In process step (5) the coated precursor obtained after completion of process step (4) is finally strand annealed at an oven set temperature in the range of from > 400 to $460\text{ }^{\circ}\text{C}$ for an exposure time in the range of from ≥ 0.8 to 10 seconds, preferably ≥ 0.8 to 2 seconds to form the coated wire.

In a preferred embodiment, the finally strand annealed coated precursor, i.e. the still hot coated wire is quenched in water which, in an embodiment, may contain one or more additives, for example, 0.01 to 0.2 volume-% of additive(s). The quenching in water means immediately or rapidly, i.e. within 0.2 to 0.6 seconds, cooling the finally strand annealed coated precursor from the temperature it experienced in process step (5) down to room temperature, for example by dipping or dripping.

After completion of process step (5) and the optional quenching the coated wire of the invention is finished. In order to fully benefit from its properties, it is expedient to either use it immediately for wire bonding applications, i.e. without delay, for example, within no longer than 25 to 70, preferably 60 days after completion of process step (5). Alternatively, in order to keep the wire's wide wire bonding process window property and in order to prevent it from oxidative/sulfidation or other chemical attack, the finished wire is typically spooled and vacuum sealed immediately after completion of process step (5), i.e. without delay, for example, within < 1 to 5 hours after completion of process step (5) and then stored for further use as bonding wire. Storage in vacuum sealed condition should not exceed 12 months. After opening the vacuum seal, the wire should be used for wire bonding within no longer than 25 to 70, preferably 60 days.

It is preferred that all process steps (1) to (5) as well as spooling and vacuum sealing are carried out under clean room conditions (US FED STD 209E cleanroom standards, 1k standard).

A third aspect of the invention is a coated wire obtainable by the afore disclosed process according to any embodiment thereof. It has been found that the coated wire of the invention is well suited for use as a bonding wire in wire bonding applications. Wire bonding technique is well known to the skilled person. In the course of wire bonding it is typical that a ball bond (1st bond) and a stitch bond (2nd bond, wedge bond) are formed. During bond forming a certain force (typically measured in grams) is applied, supported by application of ultrasonic energy (typically measured in mA). The wire of the invention exhibits a considerably wide wire bonding process window.

The following non-limiting examples illustrate the invention. These examples serve for exemplary elucidation of the invention and are not intended to limit the scope of the invention or the claims in any way.

Test methods

All tests and measurements were conducted at T = 20 °C and a relative humidity RH = 50 %.

A. Electron Backscattered Diffraction (EBSD) Pattern Analysis for determination of the crystallographic orientation and twin boundaries of the crystal grains of the wire core:

The main steps adopted to measure wire texture were sample preparation, getting good Kikuchi pattern and component calculation:

The wires were first potted using epoxy resin and polished as per standard metallographic technique. Ion milling was applied in the final sample preparation step to remove any mechanical deformation of the wire surface, contamination and oxidation layer. The ion-milled cross-sectioned sample surface was sputtered with gold. Then ion milling and gold sputtering were carried out for two further rounds. No chemical etching or ion-etching was carried out.

The sample was loaded in a FESEM (field emission scanning electron microscope) with a 70° angled holder to the normal FESEM sample holding table surface. The FESEM was further equipped with an EBSD detector. The electron back-scattering patterns (EBSP) containing the wire crystallographic information were obtained.

These patterns were further analyzed for crystal grain orientation fraction, average crystal grain size, etc. (using a software called EBSD program developed by Oxford Instruments). Points of similar orientation were grouped together to form the texture component.

To distinguish different texture components, a maximum tolerance angle of 10° was used. The wire drawing direction was set as a reference orientation. The $\langle 100 \rangle$ and $\langle 111 \rangle$ texture percentages were calculated by measurement of the percentage of crystals with $\langle 100 \rangle$ and $\langle 111 \rangle$ plane of orientation parallel to the reference orientation.

Twin boundaries (also called $\Sigma 3$ CSL twin boundaries) were excluded in the average crystal grain size calculation. The twin boundary was described by a 60° rotation about $\langle 111 \rangle$ plane of orientation between the neighboring crystallographic domains. The number of scanning points of the area of interest depends on the step size, which was less than 1/5 of the observed finest-crystal grain size (about 100nm).

The EBSD pattern analysis has been performed at five different locations per sample. Reported are the average values of the five different locations.

B. Linear Intercept Method for determination of the crystal grain sizes:

The wires were first potted using cold-mounting epoxy resin and then polished (cross-sectioned) by standard metallographic technique. A multi-prep semi-automatic polisher was used with low force and optimal speed to grind and polish the sample with minimum deformation strain on the sample surface. Finally, the polished sample was chemically etched using ferric chloride to reveal the crystal grain boundary. The crystal grain size was measured using linear intercept method under optical microscopy with a magnification of 1000, according to the ASTM E112-12 Standard.

C. Evaluation of Flowery Bonded Ball:

C.1) Preparation of FAB:

It was worked according to the procedures described in the KNS Process User Guide for FAB (Kulicke & Soffa Industries Inc, Fort Washington, PA, USA, 2002, 31 May 2009) in ambient atmosphere. FAB was prepared by performing conventional electric flame-off (EFO) firing by standard firing (single step, 17.5 μm wire, EFO current of 50 mA, EFO time 125 μs).

C.2) Ball bonding:

The formed FAB descended to an Al-0.5wt.-%Cu bond pad from a predefined height (tip of 203.2 μm) and speed (contact velocity of 6.4 $\mu\text{m}/\text{sec}$). Upon touching the bond pad, a set of

defined bonding parameters (bond force of 100 g, ultrasonic energy of 95 mA and bond time of 15 ms) took into effect to deform the FAB and formed the bonded ball. After forming the ball, the capillary rose to a predefined height (kink height of 152.4 μm and loop height of 254 μm) to form the loop. After forming the loop, the capillary descended to the lead to form the stitch. After forming the stitch, the capillary rose and the wire clamp closed to cut the wire to make the predefined tail length (tail length extension of 254 μm). For each sample a meaningful number of 2500 bonded wires were optically inspected using a microscope with a magnification of 1000. The percentage of defects was determined.

C.3) Evaluation of bonded ball with respect to flowery bonded ball:

- + Poor: $\geq 15\%$ of the bonded balls are not round but deformed
- ++ Good: $\geq 10\%$ to $< 15\%$ of the bonded balls are not round but deformed
- +++ Very good: $< 10\%$ of the bonded balls are not round but deformed

C.4) Evaluation of Wire Sway:

- ++ Good: $< 5\%$ of the wires deflect towards neighbor wire in the loop
- +++ Excellent: wires show no loop deflections

D. Evaluation of Wire Twist:

- ++ Good: Wire turns into coil on free fall from spool on unwinding with < 5 turns
- +++ Very good: Wire does not turn into coil on free fall from spool on unwinding

Wire examples

A quantity of 98.5 wt% silver (Ag) and 1.5 wt% palladium (Pd) of at least 99.99 % purity ("4N") for each metal were melted in a crucible. Then wire core precursor items in the form of 8 mm rods were continuously cast from the melt. The rods were then drawn in several drawing steps to form wire core precursors having circular cross-sections with diameters of 2 mm. The wire core precursors were intermediate batch annealed at an oven set temperature of 500 °C for an exposure time of 60 minutes. The rods were further drawn in several drawing steps to form wire core precursors having diameters of 46 μm .

The wire core precursors were electroplated with a double-layer coating of an inner layer of nickel and an adjacent outer layer of gold. To this end, the wire core precursor while being wired as cathode was moved through a 60°C warm nickel electroplating bath and, subsequently, through a 61°C warm gold electroplating bath. The nickel electroplating bath comprised 90 g/l (grams per liter) $\text{Ni}(\text{SO}_3\text{NH}_2)_2$, 6g/l NiCl_2 and 35 g/l H_3BO_3 , whereas the gold electroplating bath (based on MetGold Pure ATF from Metalor) had a gold content of 13.2 g/l and an antimony content of 20 wt.-ppm (based on MetGold Pure ATF from Metalor).

Thereafter the coated wire precursors were further drawn to final diameters of 20 μm , followed by a final strand annealing at an oven set temperature indicated in table 1 for an exposure time of 0.9 seconds, immediately followed by quenching the so-obtained coated wires in water containing 0.07 vol.-% of surfactant. The 20 μm thick wires had an 9 nm thick inner layer of nickel and an adjacent outer 90 nm thick layer of gold.

Table 1: overview on inventive samples S1 to S3 and comparative samples C1 and C2

wire	Final annealing temperature (°C)	Crystallographic orientation wire core		twin boundaries (%)	crystal grain size (μm)	wire twist	flowery bonded ball	wire sway
		<100> (%)	<111> (%)					
S 1	447	37.7	18.7	32.3	1.0	+++	+++	+++
S 2	430	47.9	8.2	21.3	0.8	+++	+++	+++
S 3	405	50.5	11.7	13.9	0.8	+++	+++	+++
C 1	500	14.8	1.7	42.6	1.2	+++	+	++
C 2	300	51.5	1.3	3.1	0.6	++	+	+++

CLAIMS

1. A wire comprising a wire core with a surface, the wire core having a coating layer superimposed on its surface, wherein the wire core itself is a silver-based wire core, wherein the
5 coating layer is a double-layer comprised of a 1 to 100 nm thick inner layer of palladium or nickel and an adjacent 1 to 250 nm thick outer layer of gold, wherein the wire exhibits at least one of the following intrinsic properties A1) to A3):
- A1) the average grain size of the crystal grains in the wire core, measured in longitudinal direction, is in the range of from 0.7 to 1.1 μm ;
- 10 A2) the fraction of twin boundaries, measured in longitudinal direction of the wire, is in the range of from 5 to 40 %;
- A3) 20 to 70 % of the crystal grains of the wire core are oriented in $\langle 100 \rangle$ direction, and 3 to 40 % of the crystal grains of the wire core are oriented in $\langle 111 \rangle$ direction, each % with respect to the total number of crystal grains with orientation parallel to the drawing direction of the wire.
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2. The wire of claim 1 having an average cross-section in the range of from 50 to 5024 μm^2 .
3. The wire of claim 1 having an average diameter in the range of from 8 to 80 μm .
- 20 4. The wire of any one of the preceding claims, wherein the double-layer is comprised of a 1 to 30 nm thick inner layer of palladium or nickel and an adjacent 20 to 200 nm thick outer layer of gold.
5. The wire of any one of the preceding claims exhibiting two or all of the intrinsic properties
25 A1) to A3).
6. The wire of any one of the preceding claims, wherein the gold layer comprises at least one member selected from the group consisting of antimony, bismuth, arsenic and tellurium in a total proportion in the range of from 10 to 300 wt.-ppm, based on the weight of the wire.
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7. The wire of claim 6, wherein the total proportion of the at least one member selected from the group consisting of antimony, bismuth, arsenic and tellurium is in the range of from 300 to 9500 wt.-ppm, based on the weight of the gold of the gold layer.

8. The wire of claim 6 or 7, wherein the at least one member selected from the group consisting of antimony, bismuth, arsenic and tellurium exhibits a concentration gradient within the gold layer, said gradient increasing in perpendicular direction to the longitudinal axis of the wire core.

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9. The wire of claim 6, 7 or 8, wherein antimony is present within the gold layer.

10. The wire of claim 9, wherein antimony is alone present within the gold layer without the simultaneous presence of bismuth, arsenic and tellurium.

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11. A process for the manufacture of the coated wire of any one of the preceding claims, wherein the process comprises at least the steps (1) to (5):

(1) providing a silver-based precursor item,

(2) elongating the precursor item to form an elongated precursor item, until an intermediate diameter in the range of from 30 to 200 μm is obtained,

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(3) applying a double-layer coating of an inner layer of palladium or nickel and an adjacent outer layer of gold on the surface of the elongated precursor item obtained after completion of process step (2),

(4) further elongating the coated precursor item obtained after completion of process step (3)

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until a desired final diameter and a double-layer comprised of an inner layer of palladium or nickel having a desired final thickness in the range of from 1 to 100 nm and an adjacent outer layer of gold having a desired final thickness in the range of from 1 to 250 nm is obtained, and

(5) finally strand annealing the coated precursor obtained after completion of process step (4) at an oven set temperature in the range of from >400 to $460\text{ }^{\circ}\text{C}$ for an exposure time in the range

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of from ≥ 0.8 to 10 seconds to form the coated wire,

wherein step (2) may include one or more sub-steps of intermediate batch annealing of the precursor item at an oven set temperature of from 200 to $650\text{ }^{\circ}\text{C}$ for an exposure time in the range of from 30 to 300 minutes.

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12. The process of claim 11, wherein the palladium or nickel layer is applied by electroplating.

13. The process of claim 11 or 12, wherein the gold layer is applied by electroplating.

14. The process of claim 13, wherein the application of the gold layer in step (3) is performed by electroplating it from a gold electroplating bath comprising gold and at least one member selected from the group consisting of antimony, bismuth, arsenic and tellurium.