

- 1 -

57819J.50

WO 92/16528

PCT/EP92/00489

REPLACED SHEETS	SG
☒ Description : paged ...1... to paged ...50...	
☒ Claim(s) : paged ...51... to paged ...52...	
☐ Drawing(s) : paged to paged	
☐ Abstract : paged to paged	
DATE OF RECEIPT	

New esters of bi- and tricyclic aminoalcohols,
their preparation and their use in
pharmaceutical compositions

This Sheet Has Been Replaced



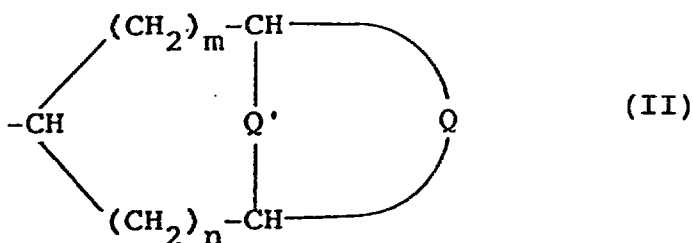
The invention relates to new esters of bi- and tricyclic amino alcohols, the preparation of these compounds and their use in pharmaceutical compositions.

The new compounds correspond to the formula



wherein

A represents the group

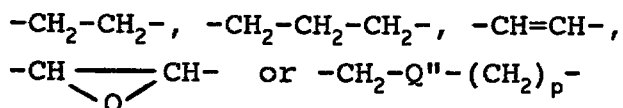


wherein

m represents 0, 1 or 2,

n represents 1 or 2 and $m + n \leq 3$,

Q represents one of the double-bonded groups



(p = 0 or 1)

and

Q' represents the group =NR, the group =NRR' or the group CH_2 , wherein R represents an optionally halo- or hydroxy-substituted C_{1-4} -alkyl group, R' represents a C_{1-4} -alkyl group and R and R' together may also form a C_{4-6} -alkylene group, and, in the case of quaternary compounds, an equivalent of an anion ($X^\ominus$) is associated with the positive charge of the N-atom,



and

Q" has the same meanings as Q', with the exception of CH₂, and with the proviso that Q represents -CH₂-Q"-(CH₂)_p- when Q' represents CH₂,

Z represents the group

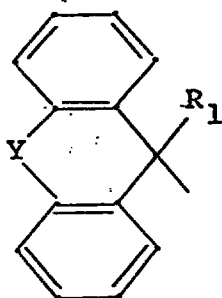


wherein

R₁ represents H, OH, CH₂OH, C₁₋₄-alkyl or C₁₋₄-alkoxy,

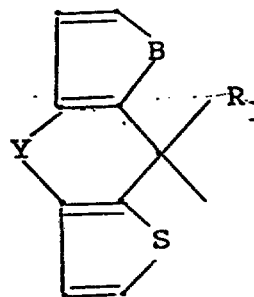
R₂ and R₃, which may be identical or different and one of which may also be H, represent

- (a) phenyl, furyl, an aromatic group which is isoelectronic with thienyl, C₅₋₇-cycloalkyl, pyridyl, C₅₋₇-cycloalkenyl or, in particular if m is equal to 0 and/or Q is equal to -CH₂-Q"-(CH₂)_p, R₂ may also represent thienyl,
- (b) an aliphatic group having up to 20 carbon atoms optionally interrupted by oxygen, or a C₁₋₆-alkyl group substituted by phenyl, phenoxy, thienyl, furyl, C₅₋₇-cycloalkyl or fluorine,
- (c) the entire group III may also represent the tricyclic group of the formula



(IV)

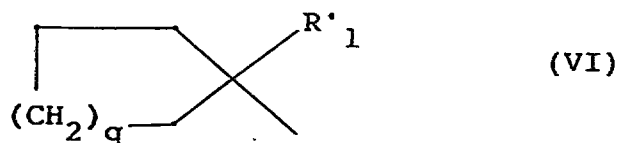
or



(V)



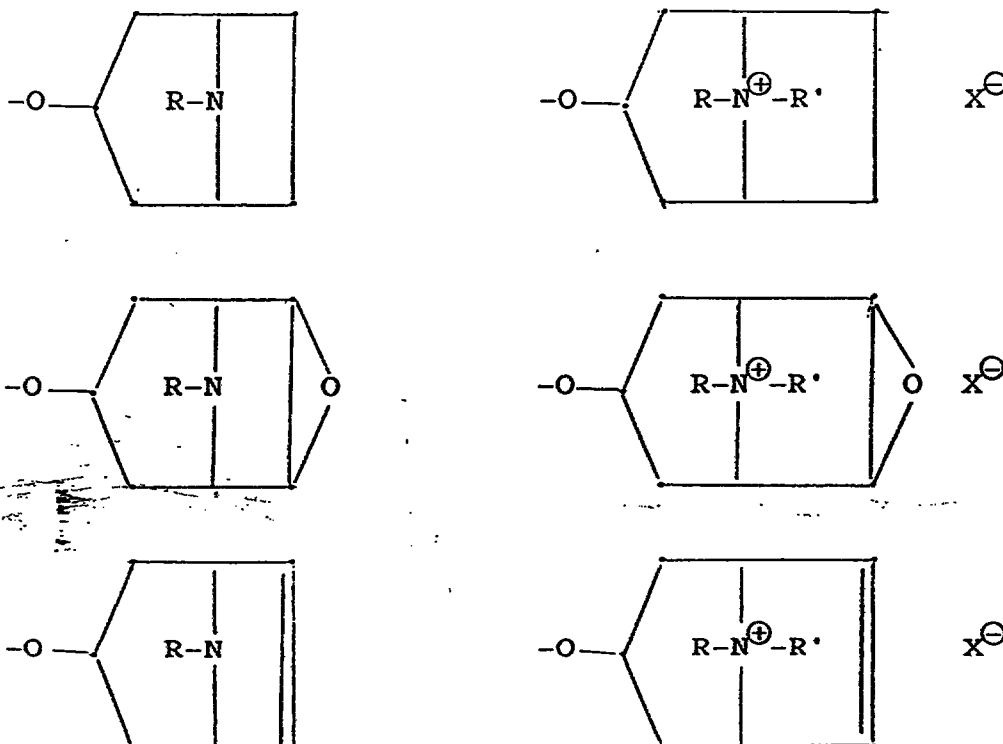
or a group of formula



wherein B may represent S or CH=CH, R'_1 has the same meaning as R_1 and may additionally represent phenyl, thienyl, furyl, thiazolyl, thiadiazolyl or methoxyphenyl, Y represents a single bond, an O- or S-atom or one of the groups $-CH_2-$, $-CH_2-CH_2-$, $-CH=CH-$, $-OCH_2-$ or $-S-CH_2-$ and q represents 1, 2 or 3.

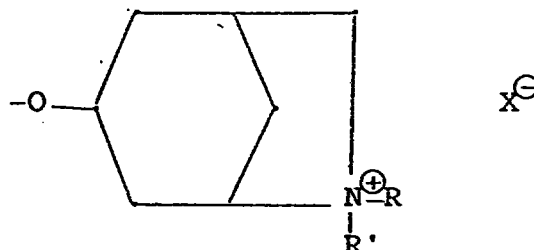
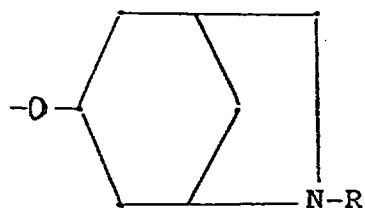
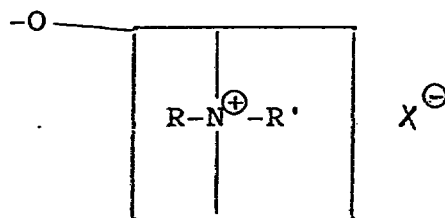
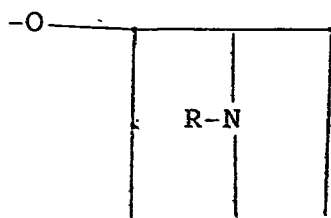
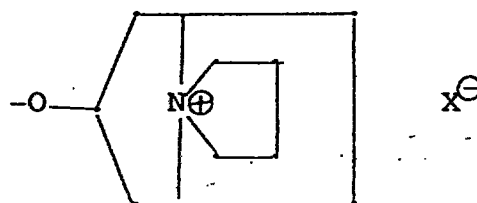
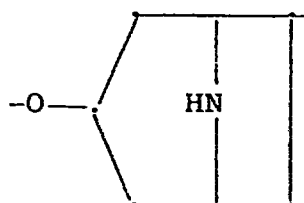
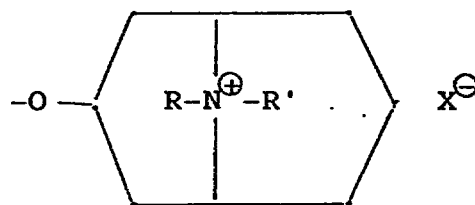
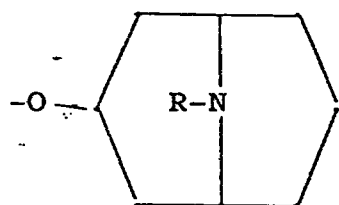
In the compounds of formula I R_2 preferably represents OH. The group -OA preferably has an α -configuration and is derived for example from scopine, tropine, granatoline or 6,7-dehydrotropine or the corresponding nor-compounds; -OA may, however, also have the β -configuration as in pseudotropine or pseudoscopine.

Examples of corresponding groups include:





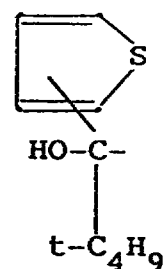
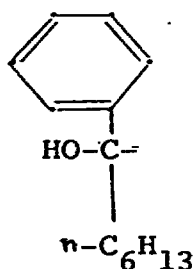
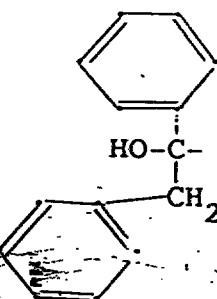
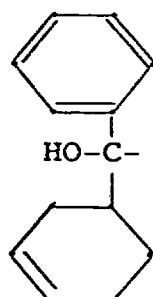
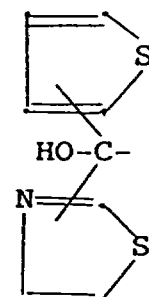
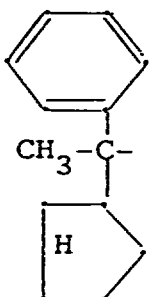
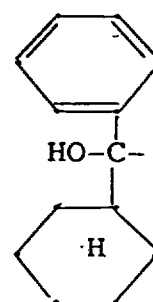
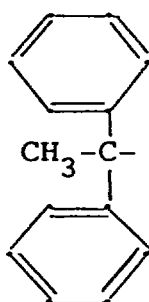
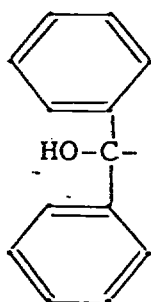
- 5 -

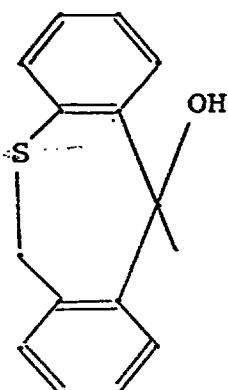
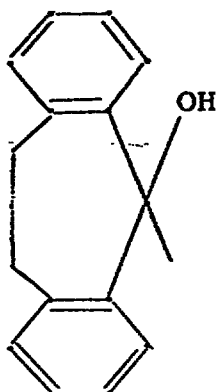
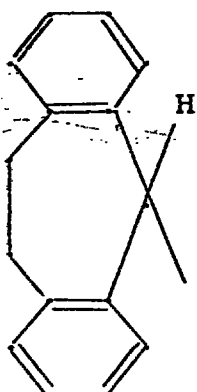
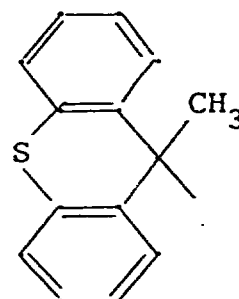
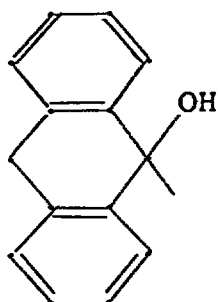
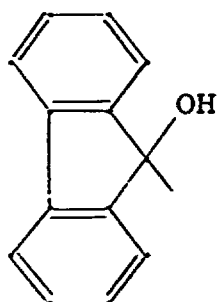
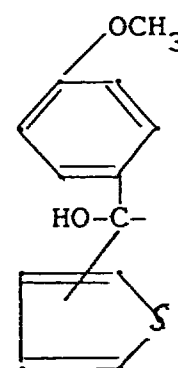
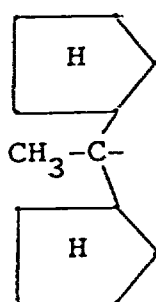
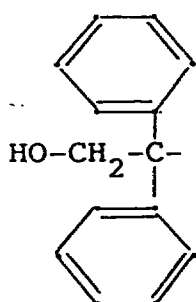
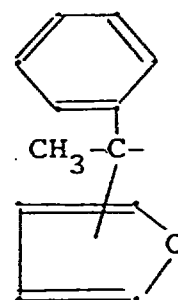
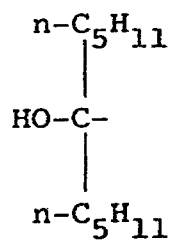
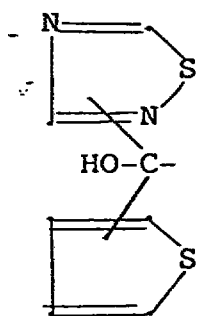


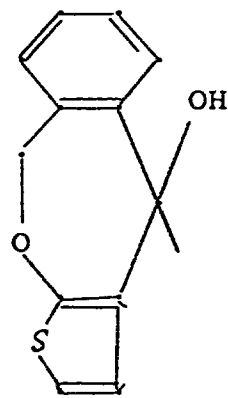
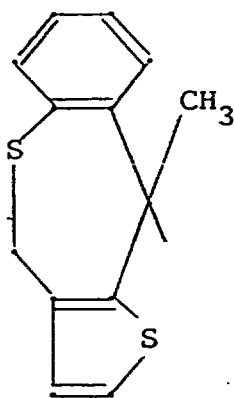
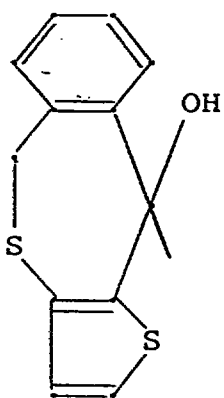
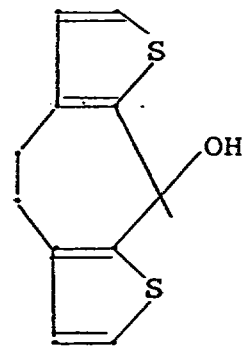
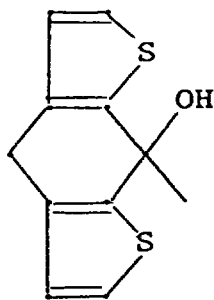
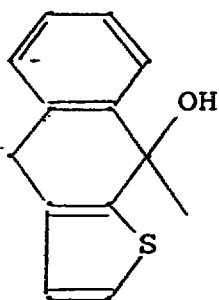
The substituent R is preferably a lower alkyl group, especially CH_3 , or may also represent, for example, C_2H_5 , $n\text{-C}_3\text{H}_7$, $i\text{-C}_3\text{H}_7$ and R' preferably represents CH_3 . R and R' together may represent, for example, $-(\text{CH}_2)_5-$. The halogen substituent for R is preferably F or, to a lesser extent, Cl. If R represents a halo- or hydroxy-substituted alkyl group it is preferably $-\text{CH}_2-\text{CH}_2\text{F}$ or $-\text{CH}_2-\text{CH}_2\text{OH}$. Accordingly, the group A represents, for example, the groups of scopolamine, N-ethylnorscopolamine, N-isopropylnorscopolamine, tropine, N-isopropylnortropine, 6,7-dehydrotropine, N- β -

fluoroethylnortropine, N-isopropyl-6,7-dehydro-nortropine, N-methylgranatoline or the corresponding quaternary compounds, the anion preferably being $\text{Br}^\ominus$ or $\text{CH}_3\text{SO}_3^\ominus$.

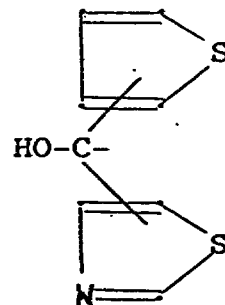
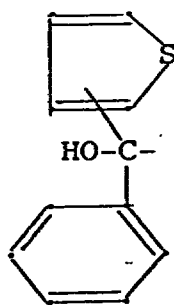
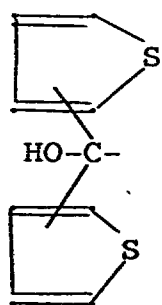
The group Z may have the following meanings, for example, whilst the aromatic groups may also be substituted, e.g. by CH_3 , OCH_3 , F or Cl:

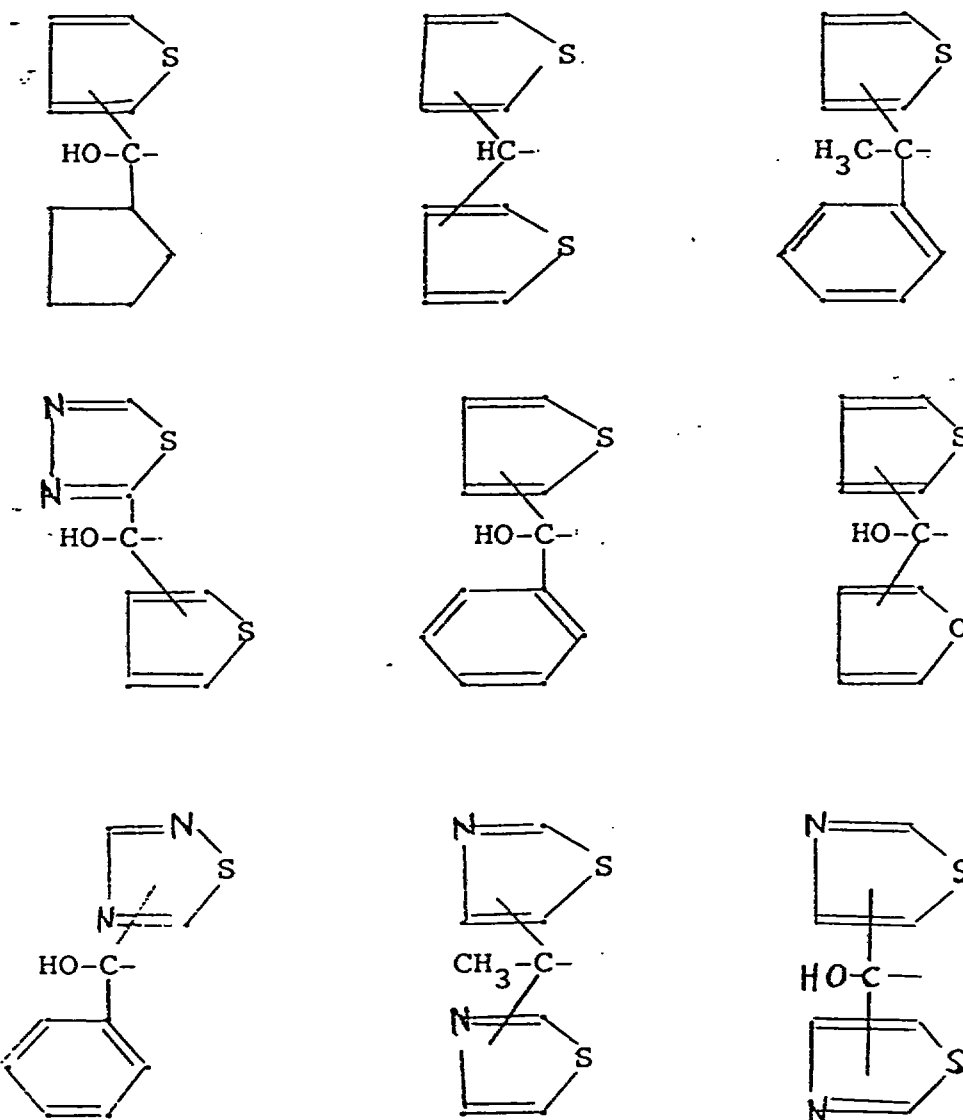






If in the group II m represents O and/or Q represents $-\text{CH}_2-\text{Q}''-(\text{CH}_2)_p-$, the group III may have the following additional meanings, in particular:





The quaternary compounds of formula I are particularly suitable for therapeutic use, while the tertiary compounds are important not only as active substances but also as intermediate products.

The compounds according to the invention are anticholinergics with a potent and long-lasting effect. At dosages in the microgram range, periods of effect of more than 24 hours are achieved after inhalation. Moreover, the toxicity is in the same range as that of the standard commercial product ipratropium bromide, whilst at the same time the therapeutic effect is in

some cases significantly greater.

- In accordance with their nature as anticholinergics, the new compounds are suitable, for example, for treating chronically obstructive bronchitis and (slight to moderate) asthma and also for treating vagally induced sinus bradycardia. Whereas in diseases of the respiratory tract it is chiefly recommended to administer the new active substances by inhalation (especially the quaternary compounds), thereby largely eliminating any side effects, the compounds are preferably administered by intravenous or oral route in the case of sinus bradycardia. It has been found to be advantageous that the new compounds have virtually no effect on gastrointestinal motility.

For use, the compounds according to the invention are processed with known excipients and/or carriers to form conventional galenic preparations, e.g. solutions for inhalation, suspensions in liquefied propellant gases, preparations containing liposomes or proliposomes, injectable solutions, plain or coated tablets, capsules, powders for inhalation for use in conventional inhalers.

This Sheet Has Been Replaced



Examples of formulations (amounts given in percent by weight):

1. Metering aerosol

Active substance according to the invention	0.005
Sorbitantriolate	0.1
Monofluorotrichloromethane and difluorodichloromethane 2 : 3	
	to 100

The suspension is poured into a conventional aerosol container with a metering valve.

Preferably, 50 μ l. of suspension are dispensed on each actuation. The active substance may also be dispensed in a higher dosage if desired (e.g. 0.02% by weight).

2. Tablets

Active substance according to the invention	0.05
Colloidal silica	0.95
Lactose	65.00
Potato starch	28.00
Polyvinylpyrrolidone	3.00
Na-cellulose glycolate	2.00
Magnesium stearate	1.00

The ingredients are processed in the usual way to form tablets weighing 200 mg.

This Sheet Has Been Replaced





of the acid Z-CO-OH, particularly an acid chloride or imidazolidine thereof, is reacted with an alcohol of formula VIII, optionally in excess or in the presence of a tertiary amine such as triethylamine, and optionally the resulting compound

- a) if Q' , represents NR' , is quaternised with a reactive monoderivative X-R of a corresponding alkane (X = leaving group) or
- b) if Q' , represents NH , is quaternised with a terminally disubstituted alkane $X-(C_{4-6}-alkylene)-X$ without intermediate isolation or
- c) if Q_a equals $-CH_2-NR'-(CH_2)_p-$, is quaternised with a reactive monoderivative X-R.

Alternatively, starting compounds may be quaternised, in which the starting compound VIII contains R representing a "halo- or hydroxy-substituted alkyl group" instead of R' at the nitrogen atom.

The transesterification according to process 1 is carried out with heating in an organic solvent, e.g. toluene, xylene or heptane, or in a melt, using strong bases such as sodium methoxide, sodium ethoxide, sodium hydride or metallic sodium as catalyst. In order to eliminate the lower alcohol released from the equilibrium, reduced pressure is used, and possibly the alcohol is distilled off azeotropically. The transesterification is carried out at temperatures generally not exceeding $95^\circ C$. Frequently, transesterification proceeds more easily in a melt. The reaction according to process 2 is carried out in an organic solvent or mixture of solvents which is sufficiently inert under the reaction conditions, e.g.

acetone or acetonitrile, at temperatures between about 0°C and the boiling temperature of the reaction mixture.

The free bases may be obtained from acid addition salts of the tertiary amines, if desired, using suitable basic compounds in a manner known per se. The quaternisation is carried out in suitable solvents, e.g. acetonitrile or acetonitrile/methylene chloride, preferably at ambient temperature; the preferred quaternising reagent is a corresponding alkyl halide, e.g. alkyl bromide, or a corresponding sulphonic acid derivative, e.g. a methane- or toluenesulphonic acid derivative. Transesterification products wherein Q' represents NH are used as starting materials for those compounds in which R and R' together represent a C₄₋₆-alkylene group. Conversion into the tertiary and then quaternary compound is carried out using suitable 1,4-, 1,5- or 1,6-dihaloalkanes without intermediate isolation.

The starting compounds, where they have not already been described, may be obtained analogously to known compounds.

Examples:

Methyl di-(2-thienyl)glycolate from dimethyloxalate and 2-thienylmagnesium bromide;

Ethyl di-(2-thienyl)glycolate from (2-thienyl)glyoxylic acid and 2-thienyllithium;

Ethyl hydroxyphenyl-(2-thienyl)acetate from methylphenylglyoxylate and 2-thienylmagnesium bromide or from methyl (2-thienyl)glyoxylate and phenylmagnesium bromide.

Similarly, methyl 2-thienylglyoxylate and cyclohexyl- or cyclopentylmagnesium bromide may be reacted.

There are also several possible methods of preparing the aminoalcohols.





Pseudoscopine can be obtained according to M. Pelonovski et al., Bull. soc. chim. 43, 79 (1928).

Pseudotropenol can be isolated from the mixture (by fractional crystallisation or distillation) which is obtained, for example, according to V. Hayakawa et al., J. Amer.Chem.Soc. 1978, 100(6), 1786 or R. Noyori et al., J.Amer.Chem.Soc. 1974, 96(10), 3336.

N-ethylnorscopine and N-isopropyl-norscopine may be prepared by hydrogenolysis from the corresponding N-alkyl-norscopolamines analogously to Banholzer DE-A P 3215933. 6-methyl-6-azabicyclo[3.2.1]octan-3- α -ol can be prepared according to F.I.Carroff et al., J. Med.Chem. 30, 805 (1987), and 7-methyl-7-azabicyclo-[2.2.1]heptan-2 α -ol may be obtained according to J.R. Pfister et al., J. Pharmac. Sciences 74, 208 (1985).

Starting from 2- or 3-furylglyoxyl-nitrile the corresponding methylesters may be prepared in conventional manner via the 2- or 3-furylglyoxylic acid obtainable from the starting material. From these methylesters, the corresponding glycolic acid esters may be obtained as described with the organometallic derivatives of 2- or 3-bromothiophene. The organometallic compounds obtainable from 2-, 3- or 4-halopyridine can be reacted with methyl 2- or 3-thienylglyoxylate to obtain the corresponding glycolic acid esters.

Thienylglycolic acid esters in which the thiophene ring in the 2- or 3-position contains fluorine may be obtained, for example, starting from 2-fluorothiophene or 3-fluorothiophene (bromination to obtain 2-bromo-3-fluoro- or 2-bromo-5-fluorothiophene and, after conversion into corresponding organometallic compounds, reaction with suitable glyoxylic acid esters to obtain the glycolic acid esters.

2-fluorothiophene and 3-fluorothiophene may be reacted analogously to Unterhalt, Arch.Pharm. 322, 839 (1989) to obtain the corresponding glyoxylic acid esters

which may then in turn be reacted with 2- or 3-thienyl derivatives, for example, to obtain glycolic acid esters, as described above. By a suitable choice of components, symmetrically substituted dithienylglycolic acid esters may be prepared analogously.

A method analogous to benzoin condensation and benzylic acid rearrangement is also possible.

The acid chlorides required may be obtained from the acids and thionyl chloride whilst the imidazolides may be obtained from the acids and carbonyldiimidazole.

This Sheet Has Been Replaced



The following Examples illustrate the invention without restricting it.

Example 1

Benzylic acid scopine ester-methobromide

- a) Benzylic acid scopine ester from α -chlorodiphenylacetic acid chloride and scopine
-

26.5 g (0.1 mol) α -chlorodiphenylacetic acid chloride are added to a solution of 31.0 g (0.2 mol) of scopine in 60 ml of anhydrous pyridine at 0°C within 50 minutes with stirring. After it has all been added the mixture is stirred for 4 hours without cooling and then left to stand for 24 hours. In order to work up the mixture the scopine hydrochloride precipitated is suction filtered. The solution separated off is evaporated down under reduced pressure, the residue is dissolved in a mixture of 600 ml of water and 15 ml of conc. hydrochloric acid and heated to about 80°C for 10 minutes. At a temperature below 20°C, sodium carbonate is added until a pH of 9 is achieved. The benzylic acid scopine ester is extracted with methylene chloride and the extracts are dried over sodium sulphate. After evaporation and treatment with acetone, white crystals are obtained, m.p. 182-3°C (decomp.), yield 31.8 g (87% of theory). Elementary analysis and spectra confirm that the title compound has been obtained which can be converted in the usual way into the hydrochloride, m.p. 256°C (decomp.; from ethanol).



b) Benzylic acid scopine ester from benzylic acid
imidazolidine and scopine

A suspension of 7.13 g (0.046 mol) of scopine and 3.2 g (0.0115 mol) of benzylic acid imidazolidine in 50 ml of acetone is heated to boiling point. After about 10 minutes a further 9.6 g (0.0345 mol) of benzylic acid imidazolidine are gradually added thereto. After the reaction has ended the mixture is cooled with ice/common salt. The crystals precipitated are suction filtered. They may be converted into the hydrochloride, m.p. 256°C (decomp.; from ethanol). Yield 8.9 g, 53% of theory).

c) Benzylic acid scopine ester methobromide

7.12 g (0.075 mol) of methylbromide dissolved in acetone are added to a suspension of 5.48 g (0.015 mol) of benzylic acid scopine ester in 120 ml of acetonitrile and 20 ml of carbon tetrachloride and the mixture is left to stand under a slight overpressure until the reaction has ended. The crystals precipitated are suction filtered, washed with cold acetonitrile then with diethylether and after drying (at 40°C under reduced pressure) recrystallised from methanol/ether, m.p. 200°C (decomp.). Elementary analysis and spectrum confirm that the desired compound has been obtained.



Example 2

1-N- β -Fluoroethylnorscopolamine-methobromide

a) 1-N- β -Fluoroethylnorscopolamine-hydrochloride

A mixture of 16.3 (0.05 mol) of 1-norscopolamine-hydrochloride, 6.3 g (0.05 mol) of 2-bromo-fluoroethane, 10.6 g (0.1 mol) of sodium carbonate and 100 ml of acetonitrile is refluxed for 6 hours. Then a further 6.3 g (0.05 mol) of 2-bromofluoroethane and 5.3 g (0.05 mol) of sodium carbonate are added and the mixture is heated for a further 24 hours. Finally, 3.2 g (0.025 mol) of 2-bromofluoroethane and 2.7 g (0.025 mol) of sodium carbonate are added and the mixture is heated for a further 48 hours. It is suction filtered and the solution is concentrated by evaporation. The residue is taken up in methylene chloride, extracted with water and the methylene chloride phase is dried over sodium sulphate. After distillation of the methylene chloride an oily residue is obtained which is reacted in the usual way to form the hydrochloride. From methanol/ether 13.1 g of white crystals are obtained (70.4% of theory), m.p. 197-8°C (decomp.).

b) Reaction to form the methobromide

7.0 g (0.021 mol) of the amine liberated in the usual way from the hydrochloride obtained in a) are reacted in 20 ml of absolute acetonitrile with 9.9 g (0.104 mol) of methylbromide for 6 days under a slight overpressure. The crystals precipitated are recrystallised from methanol/ether. 3.9 g of white crystals, m.p. 194°C (decomp.).



Elementary analysis and spectra confirm that the title compound has been obtained.

Example 3

Mandelic acid scopine ester methobromide

155.2 g (1.0 mol) of scopine are dissolved in 200 ml of absolute methylene chloride and 116.9 g (0.55 mol) of acetylmandelic acid chloride dissolved in 100 ml of absolute methylene chloride are added dropwise thereto (at 20°C within 1 hour). (The acetylmandelic acid chloride is obtained from acetylmandelic acid and thionyl chloride). After one hour the scopine hydrochloride precipitated is separated off, the methylene chloride solution is extracted with water and dried.

The combined aqueous phases are made alkaline with sodium carbonate, extracted with methylene chloride and the methylene chloride phase is dried. The solvent is distilled off from the combined methylene chloride solutions. The base which remains is converted into the hydrochloride in the usual way. After recrystallisation from acetonitrile, 124.6 g (67.7% of theory) of white crystals are obtained, m.p. 207°C (decomp.).

27.5 g (0.075 mol) of the acetyl compound thus obtained are left to stand in 110 ml of 20% hydrochloric acid for 20 hours at ambient temperature. Whilst cooling, the reaction solution is made alkaline and the mandelic acid scopine ester is extracted with methylene chloride. After drying over sodium sulphate and distillation of the solvent, the hydrochloride is prepared in the usual way. From methanol/ether, 22.5 g (92.4% of theory) of white crystals are obtained, m.p. 141-2°C.

10.7 g (0.037 mol) of the ester liberated from the hydrochloride in the usual way are left to stand for 40



hours in acetonitrile with 17.58 g (0.185 mol) of methylbromide under a slight overpressure.

The crystals precipitated are suction filtered, washed with cold acetonitrile and recrystallised from methanol/ether. 10.0 g (70.4% of theory) of white crystals are obtained, m.p. 223°C (decomp.).

Elementary analysis and spectra confirm that the title compound has been obtained.

Example 4

Xanthene-9-carboxylic acid scopine ester methobromide

a) At 20°C, a solution of 11.1 g (0.11 mol) of triethylamine in 20 ml of acetone and 26.9 g (0.11 mol) of xanthene-9-carboxylic acid chloride (obtained from xanthene-9-carboxylic acid and thionyl chloride) in 80 ml of acetone are simultaneously added dropwise to a solution of 15.5 g (0.01 mol) of scopine in 50 ml of acetone at 20°C. After 4 hours a further 1.1 g (0.01 mol) of triethylamine and 2.69 (0.011 mol) of xanthene-9-carboxylic acid chloride are added. After 4 days the mixture is suction filtered and the solvent is distilled off from the solution. Sodium carbonate solution is added to the residue and extracted with methylene chloride. The organic phase is dried over sodium sulphate and the solvent is distilled off. From the residue, the hydrochloride of the resulting xanthene-9-carboxylic acid scopine ester is obtained in the conventional manner; white crystals from acetonitrile/ether, m.p. 223°C (decomp.); yield 21.8 g.

b) The base is liberated in the usual way from a sufficient quantity of the hydrochloride obtained in a). 36.3 g (0.1 mol) thereof are reacted in a



solution of 47.5 g (0.5 mol) of methylbromide in 49 g of acetonitrile for 24 hours under slight overpressure. The crystals obtained are suction filtered, washed with acetone/ether and recrystallised from ethanol. Yield 44.0 g (95.8% of theory), white crystals, m.p. 139°C. The crystals contain 0.5 mol of ethanol. Elementary analysis and spectra confirm the presence of the title compound.

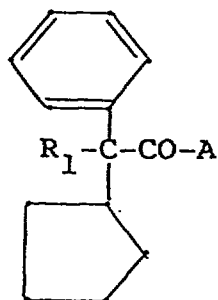
The other compounds according to the invention may also be obtained in accordance with the Examples given above. The compounds melt with decomposition.

This Sheet Has Been Replaced



T A B L E I

Compounds of formula



No.	R ₁	A	M.p. [°C]
1	OH	3α- Tropanyl-methobromide	275-6
2	OH	3α- Tropanyl-β-fluorethobromide	205-6
3	H	3α-N-Ethyl-(6β,7β-Epoxy)- nortropanyl-methobromide	228
4	H	3α-N-Propyl-(6β,7β-Epoxy)- nortropanyl-methobromide	206-7
5	H	3α-N-Isopropyl-(6β,7β-Epoxy)- nortropanyl-methobromide	218
6	H	3α-(6β,7β-Epoxy)- tropanyl-methobromide	207
7	H	3α-(6,7 -Dehydro)tropanyl- methobromide	226-8



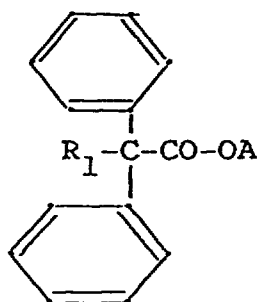
No.	R ₁	A	M.p. [°C]
8	H	3 α -Tropanyl-methobromide	275-6
9	H	3 α -N-Ethyl-nortropanyl-methobromide	256-7
10	H	(-)-3 α -N-Isopropyl-nortropanyl-methobromide	256
11	H	(+)-3 α -N-Isopropyl-nortropanyl-methobromide	256
12	H	3 α -Nortropanyl-8,1'-pyrrolidinium-bromide	267-70
13	H	(+)-3 α -Tropanyl-methobromide	278-81
14	H	(-)-3 α -Tropanyl-methobromide	278-81

This Sheet Has Been Replaced



T A B L E II

Compounds of formula



No.	R ₁	A	M.p. [°C]
1	OH	3α-N-Isopropyl-nortropanyl-methobromide	258
2	OH	3α-N-β-Chlorethyl-nortropanyl-methobromide	203
3	OH	3α-N-Ethyl-nortropanyl-methobromide	269
4	OH	3α-Tropanyl-ethobromide	258
5	OH	3α-(6β,7β-Epoxy)tropanyl-methobromide	200
6	OH	3α-N-Ethyl-(6β,7β-Epoxy)-nortropanyl-methobromide	220-1
7	OH	3α-(6β,7β-Epoxy)-N-isopropyl-nortropanyl-methobromide	234-5

This Sheet Has Been Replaced



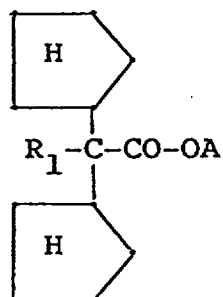
No.	R ₁	A	M.p. [°C]
8	OH	3 α -N-Methylgranatanyl-methobromide	249
9	OH	3 α -N-Isopropylgranatanyl-methobromide	219-20
10	OH	3 α -(6,7-Dehydro)tropanyl-methobromide	207-8
11	H	3 α -(6,7-Dehydro)tropanyl-methobromide	214-5
12	OH	3 α -(6,7-Dehydro)-N-isopropyl-nortropanyl-methobromide	223
13	H	3 α -Nortropanyl-8,1'-pyrrolidinium-bromide	231-2

This Sheet Has Been Replaced



T A B L E III

Compounds of formula



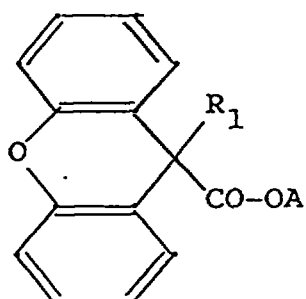
No.	R ₁	A	M.p.[°C]
1	H	3α-(6β,7β-Epoxy)-tropanyl-methobromide	213-4
2	H	3α-(6β,7β-Epoxy)-tropanyl-propargochloride	204-5
3	H	3α-N-Ethyl-(6β,7β-Epoxy)-nortropanyl-methobromide	201
4	H	3α-(6,7-Dehydro)tropanyl-methobromide	238-9
5	H	3α-Tropanyl-methobromide	203-5
6	OH	3α-(6β,7β-Epoxy)tropanyl-metho-methanesulphonate	

This Sheet Has Been Replaced



T A B L E IV

Compounds of formula



No.	R ₁	A	M.p. [°C]
1	H	3α-(6β,7β-Epoxy)-N-n-propyl nortropanyl-methobromide	213-4
2	H	3α-(6β,7β-Epoxy)-N-isopropyl- nortropanyl-methobromide	242
3	H	3α-(6β,7β-Epoxy)-N-ethyl- nortropanyl-methobromide	217
4	H	3α-(6β,7β-Epoxy)tropanyl- methobromide (with crystal ether)	139
5	H	3α-(6β,7β-Epoxy)tropanyl- ethobromide	128-31
6	H	3α-Tropanyl-ethobromide	212-3
7	OH	3α-N-Isopropylnortropanyl- metho-methanesulphonate	229-32



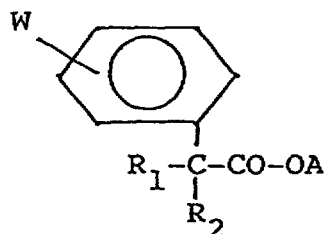
No.	R ₁	A	M.p. [°C]
8	H	3 α -N-Isopropyl-nortropanyl-methobromide	184-5
9	H	3 α -(6,7,-Dehydro)-N-isopropyl-nortropanyl-methobromide	259
10	H	3 α -(6,7 -Dehydro)-tropanyl-methobromide	237-8

This Sheet Has Been Replaced



Table V

Compounds of formula



No.	R ₁	R ₂	W	A	M.p. [°C]
1	OH	H	H	3α-(6β,7β-Epoxy)-tropanyl-methobromide	223
2	CH ₂ OH	H	H	3α-(6β,7β-Epoxy)-tropanyl-ethobromide	190
3	H	Cyclo-heptyl	H	3α-(6β,7β-Epoxy)-tropanyl-methobromide	242
4	H	Cyclo-heptyl	H	3α-(6β,7β-Epoxy)-tropanyl-propobromide	215-6
5	-C ₅ H ₈ -		H	3α-(6β,7β-Epoxy)-tropanyl-methobromide	223-4
6	CH ₂ OH	H	H	3α-(6β,7β-Epoxy)-N-(β-fluorethyl)-nortropanyl-methobromide	194
7	CH ₂ OH	H	H	3α-(6β,7β-Epoxy)-N-(β-hydroxyethyl)-nortropanyl-methobromide	211
8	OH	C ₆ H ₄	4-F	3α-Tropanyl-methobromide	220-1



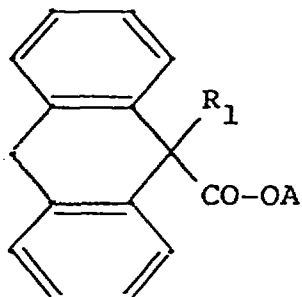
No.	R ₁	R ₂	W	A	M.p. [°C]
9	-C ₅ H ₈ -		H	3α-Tropanyl-methobromide	287-9
10	-C ₅ H ₈ -		H	3α-N-Isopropyl-nortropanyl-methobromide	263
11	-C ₆ H ₁₀ -		H	3α-N-Isopropyl-nortropanyl-methobromide	261
12	OH	C ₆ H ₁₁	H	3α-(6,7-Dehydro)-tropanyl-methobromide	233-5
13	H	C ₆ H ₁₁	3-CH ₃	3α-N-Isopropyl-nortropanyl-methobromide	252-4
14	H	C ₅ H ₉	3-CH ₃	3α-Nortropanyl-8,1'-pyrrolidinium-bromide	224-6

This Sheet Has Been Replaced



Table VI

Compounds of formula

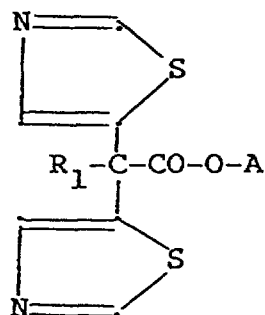


No.	A	R ₁	M.p. [°C]
1	3α-(6β,7β-Epoxy)-tropanyl-methobromide . H ₂ O	H	176
2	3α-Tropanyl-methobromide	H	
3	3α-(6,7-Dehydro)-tropanyl-methobromide	OH	
4	3α-(N-β-Fluorethyl)-nortropanyl-methobromide	H	
5	3α-Tropanyl-β-fluorethobromide	OH	
6	3α-(N-Isopropyl)-granatanyl-methobromide	H	
7	3α-(N-Isopropyl)-nortropanyl-methobromide	H	
8	3α-(6β,7β-Epoxy)-N-isopropyl-nortropanyl-methobromide	OH	
9	3α-(6β,7β-Epoxy)-N-ethylnor-tropanyl-methobromide	OH	
10	3α-(N-Ethyl)-nortropanyl-methobromide	OH	
11	3α-(N-Methyl)-granatanyl-methobromide	CH ₃	



Table VII

Compounds of formula



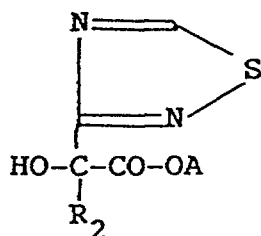
No.	A	R_1	M.p. [$^{\circ}C$]
1	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	H	
2	3 α -(6,7-Dehydro)-tropanyl-methobromide	H	
3	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	Methyl	
4	3 α -(6,7-Dehydro)-tropanyl-methobromide	Methyl	
5	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	OH	
6	3 α -(6,7-Dehydro)-tropanyl-methobromide	OH	

This Sheet Has Been Replaced



Table VIII

Compounds of formula



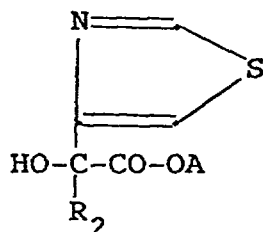
No.	A	R ₂	M.p. [°C]
1	3α-(6β,7β-Epoxy)-tropanyl-methobromide	Phenyl	
2	3α-(6,7-Dehydro)-tropanyl-methobromide	Phenyl	
3	3α-(6β,7β-Epoxy)-tropanyl-methobromide	Cyclopentyl	
4	3α-(6,7-Dehydro)-tropanyl-methobromide	3-Thienyl	
5	3α-Tropanyl-methobromide	3-Thienyl	
6	3α-(N-Methyl)-granatanyl-methobromide	3-Thienyl	

This Sheet Has Been Replaced



Table IX

Compounds of formula



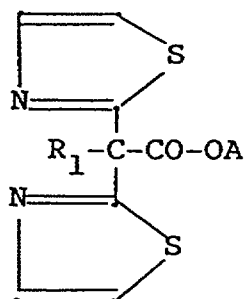
No.	A	R ₂	M.p. [°C]
1	3α-(6β,7β-Epoxy)-tropanyl-methobromide	Phenyl	
2	3α-(6,7-Dehydro)-tropanyl-methobromide	Phenyl	
3	3α-(6β,7β-Epoxy)-tropanyl-methobromide	Cyclopentyl	
4	3α-(6,7-Dehydro)-tropanyl-methobromide	3-Thienyl	
5	3α-Tropanyl-methobromide	3-Thienyl	
6	3α-(N-Methyl)-granatanyl-methobromide	3-Thienyl	

This Sheet Has Been Replaced



Table X

Compounds of formula



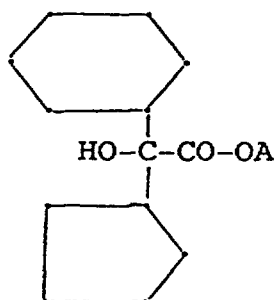
No.	A	R ₁	M.p. [°C]
1	3α-(6β,7β-Epoxy)-tropanyl-methobromide	H	
2	3α-(6,7-Dehydro)-tropanyl-methobromide	H	
3	3α-(6β,7β-Epoxy)-tropanyl-methobromide	Methyl	
4	3α-(6,7-Dehydro)-tropanyl-methobromide	Methyl	
5	3α-(6β,7β-Epoxy)-tropanyl-methobromide	OH	
6	3α-(6,7-Dehydro)-tropanyl-methobromide	OH	

This Sheet Has Been Replaced



Table XI

Compounds of formula



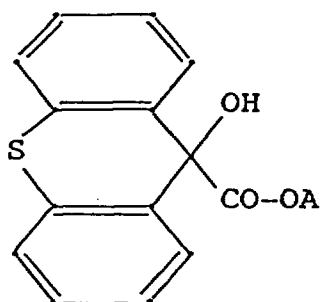
No.	A	M.p. [°C]
1	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	
2	3 α -(6,7-Dehydro)-tropanyl-methobromide	
3	3 α -(6 β ,7 β -Epoxy)-tropanyl-ethobromide	
4	3 α -(6,7-Dehydro)-tropanyl-ethobromide	
5	3 α -Tropanyl methobromide	
6	3 α -(N-Methyl)-granatanyl-methobromide	

This Sheet Has Been Replaced



Table XII

Compounds of formula



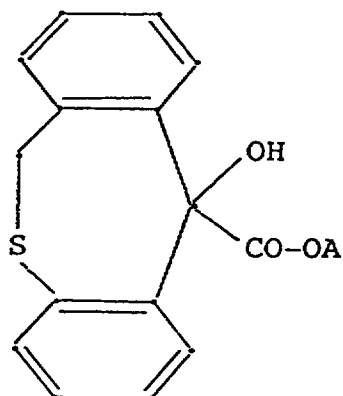
No.	A	M.p. [°C]
1	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	
2	3 α -(6,7-Dehydro)-tropanyl-methobromide	
3	3 α -(6 β ,7 β -Epoxy)-tropanyl-ethobromide	
4	3 α -(6,7-Dehydro)-tropanyl-ethobromide	
5	3 α -Tropanyl methobromide	
6	3 α -(N-Methyl)-granatanyl-methobromide	

This Sheet Has Been Replaced



Table XIII

Compounds of formula



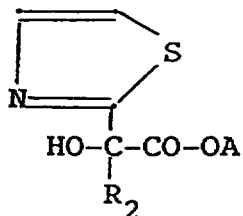
No.	A	M.p. [°C]
1	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	
2	3 α -(6,7-Dehydro)-tropanyl-methobromide	
3	3 α -(6 β ,7 β -Epoxy)-tropanyl-ethobromide	
4	3 α -(6,7-Dehydro)-tropanyl-ethobromide	
5	3 α -Tropanyl methobromide	
6	3 α -(N-Methyl)-granatanyl-methobromide	

This Sheet Has Been Replaced



Table XIV

Quarternary compounds of formula



No.	A	R ₂	M.p. [°C]
1	3α-(6β,7β-Epoxy)-tropanyl-methobromide	2-Thienyl	
2	3α-Tropanyl-methobromide	2-Thienyl	
3	3α-(6,7-Dehydro)-tropanyl-methobromide	2-Thienyl	
4	3α-(N-β-Fluorethyl)-nortropanyl-methobromide	2-Thienyl	
5	3α-Tropanyl-β-fluorethobromide	2-Thienyl	
6	3α-(N-Isopropyl)-granatanyl-methobromide	2-Thienyl	
7	3α-(N-Isopropyl)-nortropanyl-methobromide	2-Thienyl	
8	3α-(6β,7β-Epoxy)-N-isopropyl-nortropanyl-methobromide	2-Thienyl	
9	3α-(6β,7β-Epoxy)-N-ethylnor-tropanyl-methobromide	2-Thienyl	
10	3α-(N-Ethyl)-nortropanyl-methobromide	2-Thienyl	
11	3α-(N-Methyl)-granatanyl-methobromide	2-Thienyl	

This Sheet Has Been Replaced



No.	A	R ₂	M.p. [°C]
12	3 α -(6 β ,7 β -Epoxy)-N- β -fluoroethyl -nortropanyl-methobromide	2-Thienyl	
13	3 α -(6 β ,7 β -Epoxy)-N-n-propyl- nortropanyl-methobromide	2-Thienyl	
14	3 α -Tropanyl- β -hydroxyethobromide	2-Thienyl	
15	3 α -(6 β ,7 β -Epoxy)-tropanyl- methobromide	Phenyl	
16	3 α -Tropanyl-methobromide	Phenyl	
17	3 α -(N- β -Fluoroethyl)- nortropanyl-methobromide	Phenyl	
18	3 α -(6,7-Dehydro)-tropanyl- methobromide	Phenyl	
19	3 α -(N-Ethyl)-nortropanyl- methobromide	Phenyl	
20	3 α -(N-Isopropyl)-nortropanyl- methobromide	Phenyl	
21	3 α -Tropanyl-ethobromide	Phenyl	
22	3 α -(N-Ethyl)-nortropanyl- ethobromide	Phenyl	
23	3 α -(6 β ,7 β -Epoxy)-tropanyl- ethobromide	Phenyl	
24	3 α -Tropanyl- β -fluorthobromide	Phenyl	
25	3 α -Tropanyl-methobromide	Cyclohexyl	
26	3 α -(N- β -Fluorethyl)-nortropanyl- methobromide	Cyclohexyl	
27	3 α -Tropanyl- β -fluorethobromide	Cyclohexyl	
28	3 α -Tropanyl-methobromide	Cyclopentyl	
29	3 α -Tropanyl-ethobromide	Cyclopentyl	
30	3 α -(N-Ethyl)-nortropanyl- methobromide	Cyclopentyl	
31	3 α -(N-Isopropyl)-nortropanyl- methobromide	Cyclopentyl	

This Sheet Has Been Replaced



No.	A	R ₂	M.p. [°C]
32	3 α -Tropanyl- β -fluorethobromide	Cyclopentyl	
33	3 α -(N- β -Fluorethyl)-nortropanyl-methobromide	Cyclopentyl	
34	3 α (6,7-Dehydro)-tropanyl-metho-methanesulphonate	2-Thienyl	
35	3 β -(6 β ,7 β -Epoxy)-tropanyl-methobromide	2-Thienyl	
36	3 β -Tropanyl-methobromide	2-Thienyl	
37	3 β -(6,7-Dehydro)-tropanyl-methobromide	2-Thienyl	
38	3 α (6,7-Dehydro)-tropanyl-methobromide ^e	3-Thienyl	
39	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	3-Thienyl	
40	(+)-Enantiomer of No. 1		
41	(-)-Enantiomer of No. 1		
42	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	2-Furyl	
43	3 α -(6,7-Dehydro)-tropanyl-methobromide	2-Furyl	
44	3 α -Tropanyl-methobromide	2-Furyl	
45	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	2-Pyridyl	
46	3 α -(6,7-Dehydro)-tropanyl-methobromide	2-Pyridyl	
47	3 α -Tropanyl-methobromide	2-Pyridyl	
48	3 α -Tropanyl-methobromide	3-Thienyl	
49	3 α -(6,7-Dehydro)-tropanyl-methobromide	Cyclopentyl	

This Sheet Has Been Replaced



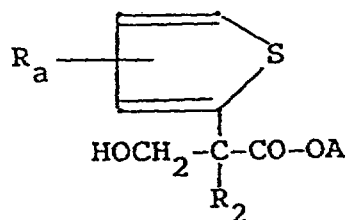
No.	A	R ₂	M.p. [°C]
50	3 α -(6 β ,7 β -Epoxy)-tropanyl methobromide	Cyclohexyl	
51	3 α -(6,7-Dehydro)-tropanyl methobromide	Cyclohexyl	
52	3 α -(6 β ,7 β -Epoxy)-tropanyl methobromide	Cyclopentyl	

This Sheet Has Been Replaced



Table XV

Compounds of formula



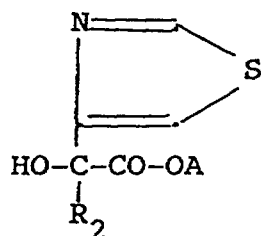
No.	A	R ₂	R _a M.p [°C]
1	3α-(6β,7β-Epoxy)-tropanyl-methobromide	2-Thienyl	5-Methyl
2	3α-(6,7-Dehydro)-tropanyl-methobromide	2-Thienyl	5-Methyl
3	3α-Tropanyl-methobromide	2-Thienyl	5-Methyl
4	3α-(6β,7β-Epoxy)-tropanyl-methobromide	2-(5-Methyl)-thienyl	5-Methyl
5	3α-(6,7-Dehydro)-tropanyl-methobromide	2-(5-Methyl)-thienyl	5-Methyl
6	3α-Tropanyl-methobromide	2-(5-Methyl)-thienyl	5-Methyl
7	3α-(6β,7β-Epoxy)-tropanyl-methobromide	2-Thienyl	5-Fluoro
8	3α-(6,7-Dehydro)-tropanyl-methobromide	2-Thienyl	5-Fluoro
9	3α-Tropanyl-methobromide	2-Thienyl	5-Fluoro
10	3α-(6β,7β-Epoxy)-tropanyl-methobromide	2-(5-Fluoro)-thienyl	5-Fluoro
11	3α-(6,7-Dehydro)-tropanyl-methobromide	2-(5-Fluoro)-thienyl	5-Fluoro
12	3α-Tropanyl-methobromide	2-(5-Fluoro)-thienyl	5-Fluoro

This Sheet Has Been Replaced



Table XVI

Compounds of formula



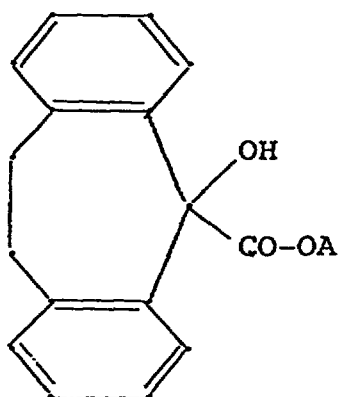
No.	A	R ₂	M.p. [°C]
1	3α-(6β,7β-Epoxy)-tropanyl methobromide	Phenyl	
2	3α-(6,7-Dehydro)-tropanyl methobromide	Phenyl	
3	3α-(6β,7β-Epoxy)-tropanyl methobromide	3-Thienyl	
4	3α-(6,7-Dehydro)-tropanyl methobromide	3-Thienyl	
5	3α-Tropanyl methobromide	3-Thienyl	
6	3α-(N-Methyl)-granatanyl methobromide	3-Thienyl	

This Sheet Has Been Replaced



Table XVII

Compounds of formula



No.	A	M.p. [°C]
1	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	
2	3 α -Tropanyl-methobromide	
3	3 α -(6,7-Dehydro)-tropanyl-methobromide	
4	3 α -(N- β -Fluorethyl)-nortropanyl-methobromide	
5	3 α -Tropanyl- β -fluorethobromide	
6	3 α -(N-Isopropyl)-granatanyl-methobromide	
7	3 α -(N-Isopropyl)-nortropanyl-methobromide	
8	3 α -(6 β ,7 β -Epoxy)-N-isopropyl-nortropanyl-methobromide	
9	3 α -(6 β ,7 β -Epoxy)-N-ethylnortropanyl-methobromide	
10	3 α -(N-Ethyl)-nortropanyl-methobromide	
11	3 α -(N-Methyl)-granatanyl-methobromide	

This Sheet Has Been Replaced



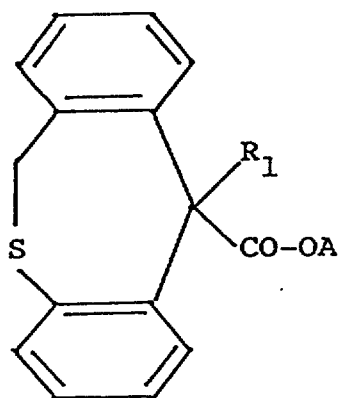
No.	A	M.p. [°C]
12	3 α -(6 β ,7 β -Epoxy)-N- β -fluoroethyl- -nortropanyl-methobromide	
13	3 α -(6 β ,7 β -Epoxy)-N-n-propyl- nortropanyl-methobromide	
14	3 α -Tropanyl- β -hydroxyethobromide	
15	3 α (6,7-Dehydro)-tropanyl- metho-methanesulphonate	
16	3 β -(6 β ,7 β -Epoxy)-tropanyl- methobromide	
17	3 β -Tropanyl-methobromide	
18	3 β -(6,7-Dehydro)-tropanyl- methobromide	

This Sheet Has Been Replaced



Table XVIII

Compounds of formula



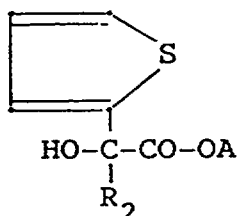
No.	A	R ₁	M.p. [°C]
1	3α-(6β,7β-Epoxy)-tropanyl methobromide	OH	
2	3α-(6,7-Dehydro)-tropanyl methobromide	OH	
3	3α-(6β,7β-Epoxy)-tropanyl methobromide	Methyl	
4	3α-(6,7-Dehydro)-tropanyl methobromide	Methyl	
5	3α-Tropanyl methobromide	H	
6	3α-(N-Ethyl)-nortropanyl methobromide	H	

This Sheet Has Been Replaced



Table A

Compounds of formula



No.	A	R ₂	Base	Mp. [°C] Hydro- chloride
1	3α-(6β,7β-Epoxy)-tropanyl	2-Thienyl	149-50	238-41
2	3α-Tropanyl	2-Thienyl	167-8	253
3	3α-(6,7-Dehydro)-tropanyl	2-Thienyl	164-5	
4	3α-(N-β-Fluorethyl)- nortropanyl	2-Thienyl		236
5	3α-(N-Isopropyl)- granatanyl	2-Thienyl		232
6	3α-(N-Isopropyl)- nortropanyl	2-Thienyl		250
7	3α-(6β,7β-Epoxy)-N-iso- propylnortropanyl	2-Thienyl		206
8	3α-(6β,7β-Epoxy)-N-ethyl- nortropanyl	2-Thienyl		212-3
9	3α-(N-Ethyl)-nortropanyl	2-Thienyl		256-7
10	3α-(N-Methyl)-granatanyl	2-Thienyl		241
11	3α-(6β,7β-Epoxy)-N-β- fluorethylnortropanyl	2-Thienyl		188-90
12	3α-(6β,7β-Epoxy)-N-n- propylnortropanyl	2-Thienyl	104-6	
13	3α-(6β,7β-Epoxy)-N-n- butylnortropanyl	2-Thienyl		225-7
14	3α-(6β,7β-Epoxy)-tropanyl	Phenyl		246-7

This Sheet Has Been Replaced



No.	A	R ₁	Base	Hydro- chloride
15	3 α -Tropanyl	Phenyl		243-4
16	3 α -(N- β -Fluoroethyl)- nortropanyl	Phenyl		219-20
17	3 α -(6,7-Dehydro)-tropanyl	Phenyl	181-3	
18	3 α -(N-Ethyl)-nortropanyl	Phenyl		231-2
19	3 α -(N-Isopropyl)- nortropanyl	Phenyl		246-7
20	3 α -Tropanyl	Cyclohexyl		260
21	3 α -(N- β -Fluoroethyl)- nortropanyl	Cyclohexyl		203-4
22	3 α -(6 β ,7 β -Epoxy)-tropanyl	Cyclopentyl		237
23	3 α -Tropanyl	Cyclopentyl		260
24	3 α -(N- β -Fluoroethyl)- nortropanyl	Cyclopentyl		182-3
25	3 α -(N-Ethyl)-nortropanyl	Cyclopentyl		227-8
26	3 α -(N-Isopropyl)- nortropanyl	Cyclopentyl		174-5
27	3 β -(6 β ,7 β -Epoxy)-tropanyl	2-Thienyl		240-2
28	3 β -Tropanyl	2-Thienyl		217-9
29	3 β -(6,7-Dehydro)-tropanyl	2-Thienyl		233-5
30	3 α -(6,7-Dehydro)-tropanyl	3-Thienyl		247-8
31	3 α -(6 β ,7 β -Epoxy)-tropanyl	3-Thienyl		242-3
32	3 α -(6 β ,7 β -Epoxy)-tropanyl	2-Furyl		
33	3 α -(6,7-Dehydro)-tropanyl	2-Furyl		
34	3 α -Tropanyl	2-Furyl		
35	3 α -Tropanyl	2-Pyridyl		
36	3 α -(6 β ,7 β -Epoxy)-tropanyl	2-Pyridyl		
37	3 α -(6,7-Dehydro)-tropanyl	2-Pyridyl		
38	3 α -Tropanyl	3-Thienyl		
39	3 α -(6,7-Dehydro)-tropanyl	Cyclopentyl		
40	3 α -(6 β ,7 β -Epoxy)-tropanyl	Cyclohexyl		
41	3 α -(6,7-Dehydro)-tropanyl	Cyclohexyl		

Note: all hydrochlorides melt with decomposition.

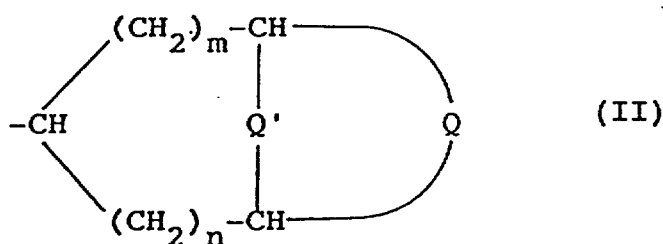
This Sheet Has Been Replaced



1. The new compounds of formula



A represents the group



wherein

m represents 0, 1 or 2,

n represents 1 or 2 and $m + n \leq 3$,

Q represents one of the double-bonded groups

$$-\text{CH}_2-\text{CH}_2-, \quad -\text{CH}_2-\text{CH}_2-\text{CH}_2-, \quad -\text{CH}=\text{CH}-,$$
$$-\text{CH} \begin{array}{c} \diagup \quad \diagdown \\ \text{O} \end{array} \text{CH}- \quad \text{or} \quad -\text{CH}_2-\text{Q}''-(\text{CH}_2)_p-$$

(p = 0 or 1)

and

Q' represents the group =NR, the group =NRR' or the group CH₂, wherein R represents an optionally halo- or hydroxy-substituted C₁₋₄-alkyl group, R' represents a C₁₋₄-alkyl group and R and R' together may also form a C₄₋₆-alkylene group, and, in the case of quaternary compounds, an equivalent of an anion (X^e) is associated with the positive charge of the N-atom.

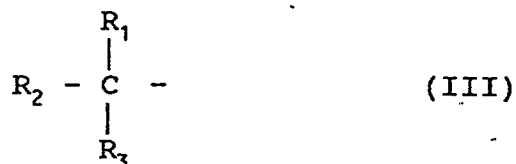




and

Q" has the same meanings as Q', with the exception of CH₂, and with the proviso that Q represents -CH₂-Q"-(CH₂)_p- when Q' represents CH₂,

Z represents the group

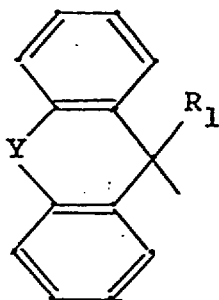


wherein

R₁ represents H, OH, CH₂OH, C₁₋₄-alkyl or C₁₋₄-alkoxy,

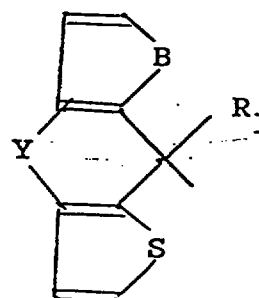
R₂ and R₃, which may be identical or different and one of which may also be H, represent

- (a) phenyl, furyl, an aromatic group which is isoelectronic with thienyl, C₅₋₇-cycloalkyl, pyridyl, C₅₋₇-cycloalkenyl or, in particular if m is equal to 0 and/or Q is equal to -CH₂-Q"-(CH₂)_p, R₂ may also represent thienyl,
- (b) an aliphatic group having up to 20 carbon atoms optionally interrupted by oxygen, or a C₁₋₆-alkyl group substituted by phenyl, phenoxy, thienyl, furyl, C₅₋₇-cycloalkyl or fluorine,
- (c) the entire group III may also represent the tricyclic group of the formula



(IV)

or

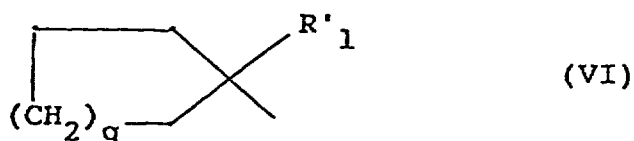


(V)





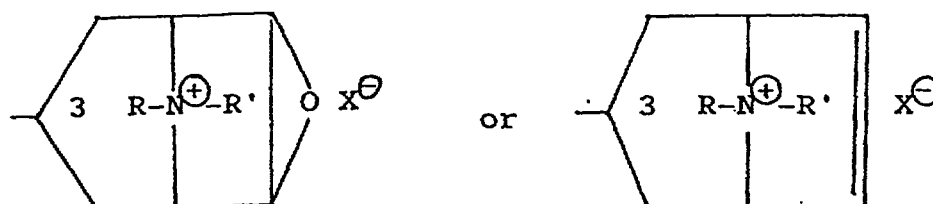
or a group of formula



wherein B may represent S or CH=CH, R₁ has the same meaning as R₁ and may additionally represent phenyl, thienyl, furyl, thiazolyl, thiadiazolyl or methoxyphenyl, Y represents a single bond, an O- or S-atom or one of the groups -CH₂-, -CH₂-CH₂-, -CH=CH-, -OCH₂- or -S-CH₂- and q represents 1, 2 or 3.

2. Compounds according to claim 1 wherein R₁ represents OH, CH₃ or CH₂OH.

3. Compounds according to claim 1 or 2, wherein A represents a group of the formula



wherein 3 represents the α -form.

4. Compounds according to claim 1, 2 or 3, wherein one of the groups R and R' represents methyl and the other group represents methyl, ethyl, n-propyl or i-propyl.

5. Compounds according to claim 1, 2, 3 or 4, wherein R₂ is a thienyl, thiazolyl or thiadiazolyl group.



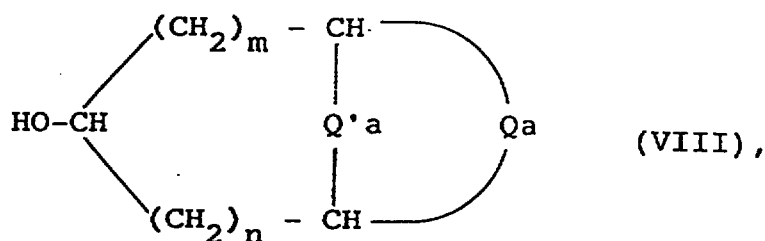


6. Compounds according to claim 1, 2, 3 or 4, wherein the group $-CR_1R_2R_3$ (III) represents one of the groups IV, V or VI.
7. Pharmaceutical compositions, characterised in that they contain a compound according to claims 1 to 6 together with conventional excipients and/or carriers.
8. Use of compounds according to claims 1 to 6 in the preparation of anticholinergic pharmaceutical compositions.
9. Use of compounds according to claims 1 to 6 for treating respiratory tract diseases and sinus bradycardia.
10. Process for preparing compounds according to claims 1 to 6, characterised in that

a) an ester of formula



wherein Z is as hereinbefore defined and R'' represents a C_{1-4} -alkyl group, preferably methyl or ethyl, is transesterified with an alcohol of formula



wherein m and n are as hereinbefore defined, Q'_a represents NR' or NH , Q_a has the same meaning as Q, except that if Q_a represents $-CH_2-Q''-(CH_2)_p-$, Q'' can





only represent NR' ,
and wherein the OH-group is in the α - or β -
position, in the presence of a conventional
transesterification catalyst or

- b) a reactive derivative (R'' being a readily
cleavable group)



of the acid Z-CO-OH , more particularly an acid
chloride or imidazolid, is reacted with an alcohol
of formula VIII, optionally in excess or in the
presence of a tertiary amine such as triethylamine,
and optionally the resulting compound

aa) if Q' , represents NR' , is quaternised with a
reactive monoderivative X-R of a corresponding
alkane (X = a leaving group) or

bb) if Q' , represents NH , is quaternised with a
terminally disubstituted alkane
 $\text{X-(C}_{4-6}\text{-alkylene)-X}$ without intermediate
isolation or

cc) if Q_a is equal to $-\text{CH}_2-\text{NR}'-(\text{CH}_2)_p-$, is
quaternised with a reactive monoderivative X-R .

This Sheet Has Been Replaced



- 1 -

REPLACEMENT SHEETS		50
☒ Doc. (you)	pages 1. to page 50	50
☒ (other)	pages 51. to page 55	55
☐ (other)	pages 1. to page 1	
☒ (other)	pages 56 to page 1	
DATE OF RECEIPT		28/8/96

New esters of bi- and tricyclic aminoalcohols,
their preparation and their use in
pharmaceutical compositions

This is the replacement sheet

Date: 28 AUG 1996

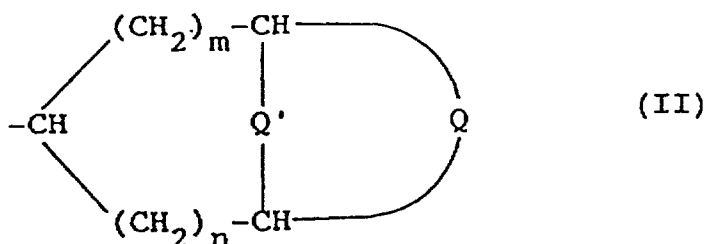
The invention relates to new esters of bi- and tricyclic amino alcohols, the preparation of these compounds and their use in pharmaceutical compositions.

The new compounds correspond to the formula



wherein

A represents the group



wherein

m represents 0, 1 or 2,

n represents 1 or 2 and $m + n \leq 3$,

Q represents one of the double-bonded groups

$-CH_2-CH_2-$, $-CH_2-CH_2-CH_2-$, $-CH=CH-$,

$-CH \begin{array}{c} \diagup \diagdown \\ O \end{array} CH-$ or $-CH_2-Q''-(CH_2)_p-$

(p = 0 or 1)

and

Q' represents the group $=NR$, the group $=NRR'$ or the group CH_2 , wherein R represents an optionally halo- or hydroxy-substituted C_{1-4} -alkyl group, R' represents a C_{1-4} -alkyl group and R and R' together may also form a C_{4-6} -alkylene group, and, in the case of quaternary compounds, an equivalent of an anion (X^e) is associated with the positive charge of the N-atom,

This is the replacement sheet

Date: 28 AUG 1996

and

Q" has the same meanings as Q', with the exception of CH₂, and with the proviso that Q represents -CH₂-Q"-(CH₂)_p- when Q' represents CH₂,

Z represents the group

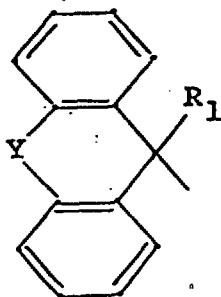


wherein

R₁ represents H, OH, CH₂OH, C₁₋₄-alkyl or C₁₋₄-alkoxy,

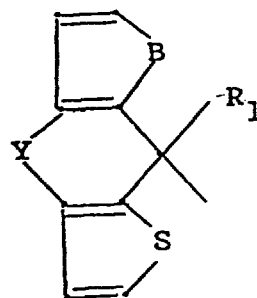
R₂ and R₃, which may be identical or different and one of which may also be H, represent

- (a) phenyl, furyl, an aromatic group which is isoelectronic with thienyl, C₅₋₇-cycloalkyl, pyridyl, C₅₋₇-cycloalkenyl or, in particular if m is equal to 0 and/or Q is equal to -CH₂-Q"-(CH₂)_p, R₂ may also represent thienyl,
- (b) an aliphatic group having up to 20 carbon atoms optionally interrupted by oxygen, or a C₁₋₆-alkyl group substituted by phenyl, phenoxy, thienyl, furyl, C₅₋₇-cycloalkyl or fluorine,
- (c) the entire group III may also represent the tricyclic group of the formula



(IV)

or

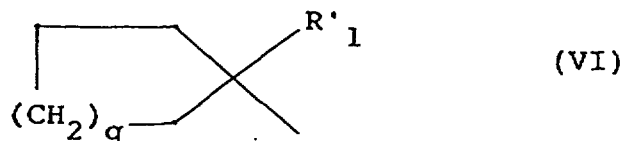


(V)

This is the replacement sheet

Date: 28 AUG 1996

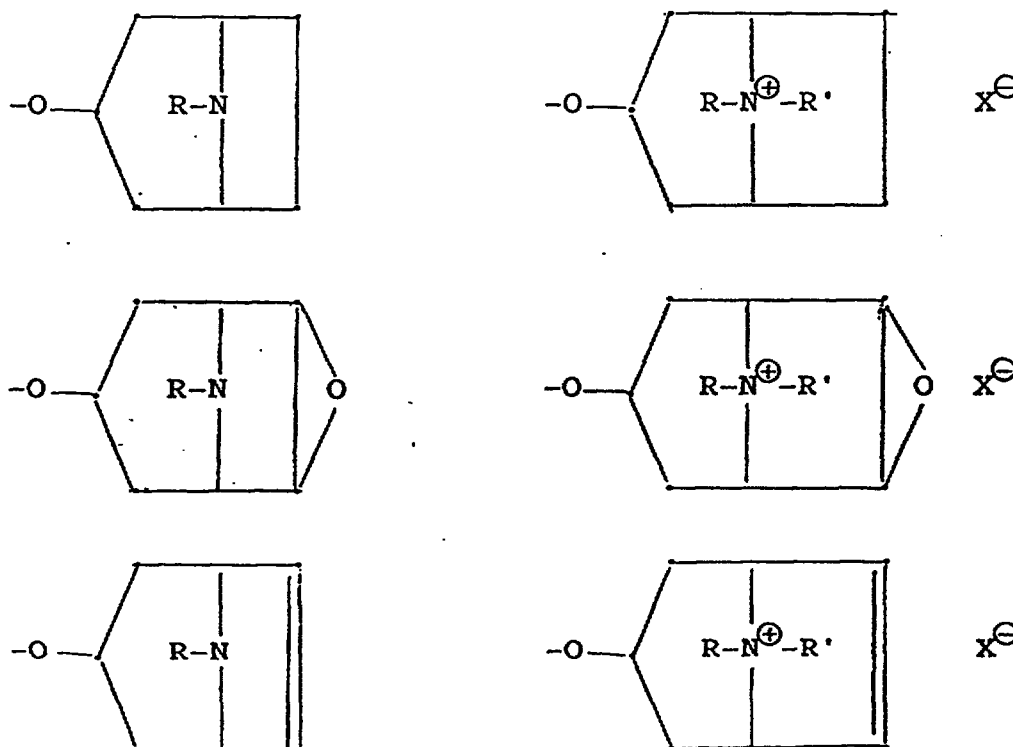
or a group of formula



wherein B may represent S or CH=CH, R'₁ has the same meaning as R₁ and may additionally represent phenyl, thienyl, furyl, thiazolyl, thiadiazolyl or methoxyphenyl, Y represents a single bond, an O- or S-atom or one of the groups -CH₂-, -CH₂-CH₂-, -CH=CH-, -OCH₂- or -S-CH₂- and q represents 1, 2 or 3.

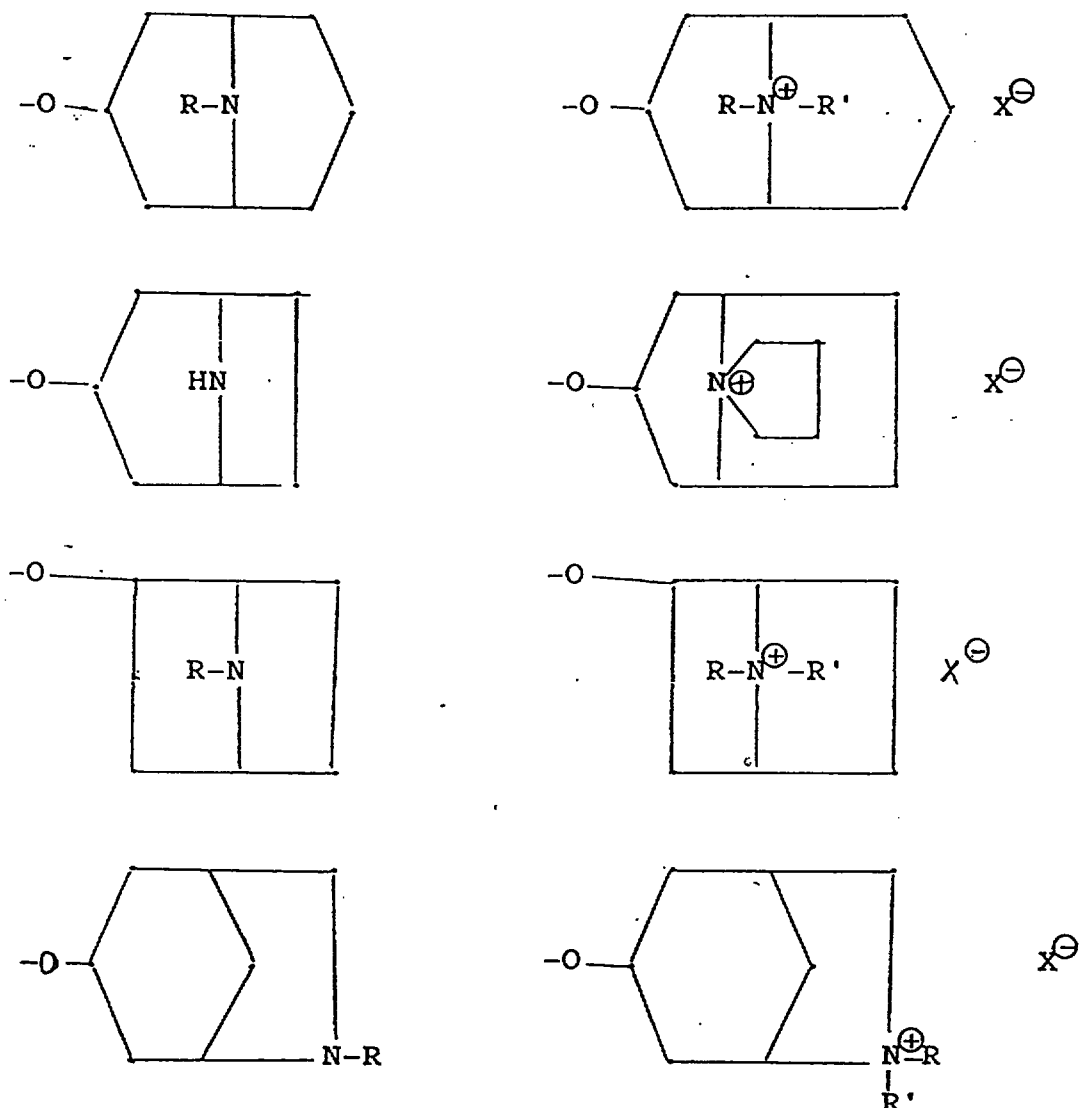
In the compounds of formula I R₂ preferably represents OH. The group -OA preferably has an α-configuration and is derived for example from scopine, tropine, granatoline or 6,7-dehydrotropine or the corresponding nor-compounds; -OA may, however, also have the β-configuration as in pseudotropine or pseudoscopine.

Examples of corresponding groups include:



This is the replacement sheet

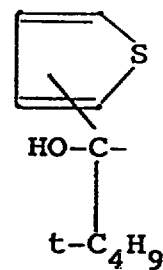
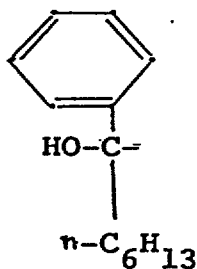
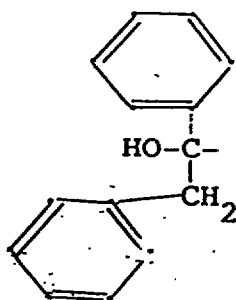
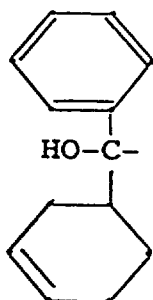
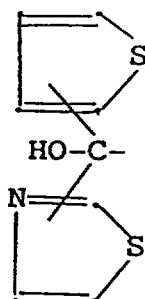
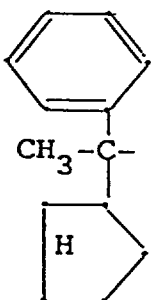
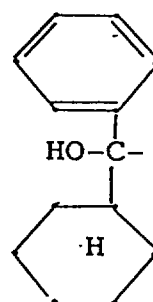
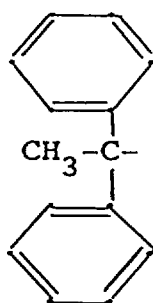
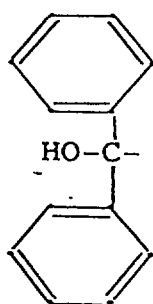
Date: 28 AUG 1996

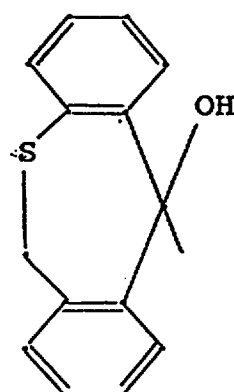
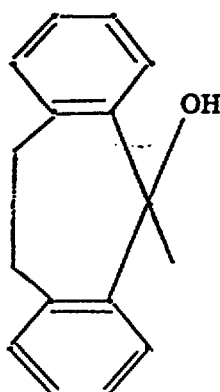
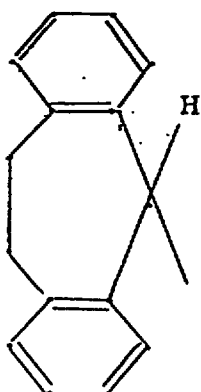
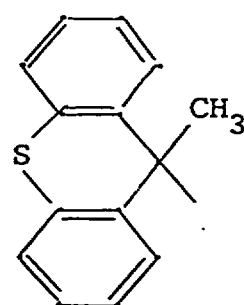
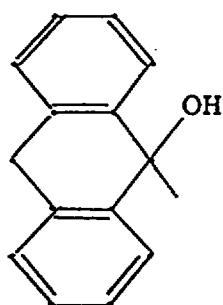
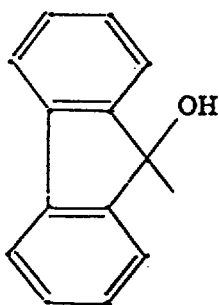
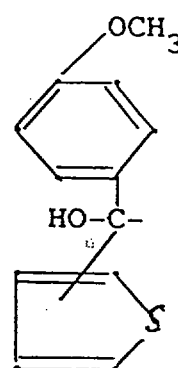
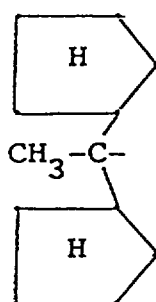
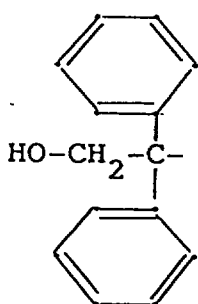
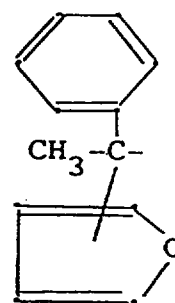
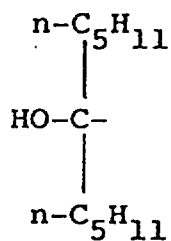
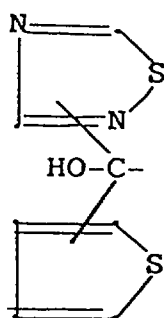


The substituent R is preferably a lower alkyl group, especially CH_3 , or may also represent, for example, C_2H_5 , $n-C_3H_7$, $i-C_3H_7$ and R' preferably represents CH_3 . R and R' together may represent, for example, $-(CH_2)_5-$. The halogen substituent for R is preferably F or, to a lesser extent, Cl . If R represents a halo- or hydroxy-substituted alkyl group it is preferably $-CH_2-CH_2F$ or $-CH_2-CH_2OH$. Accordingly, the group A represents, for example, the groups of scopolin, N-ethylnorscopin, N-isopropyl-norscopin, tropin, N-isopropyl-nortropin, 6,7-dehydrotropin, N- β -

fluoroethylnortropine, N-isopropyl-6,7-dehydronortropine, N-methylgranatoline or the corresponding quaternary compounds, the anion preferably being $\text{Br}^\ominus$ or $\text{CH}_3\text{SO}_3^\ominus$.

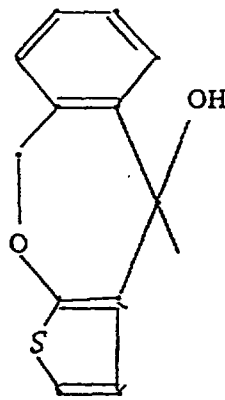
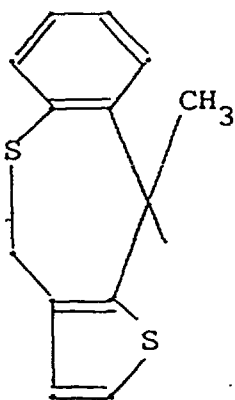
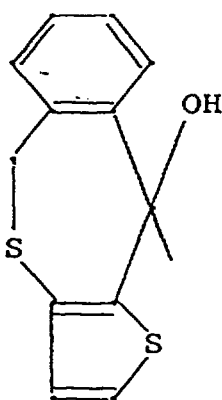
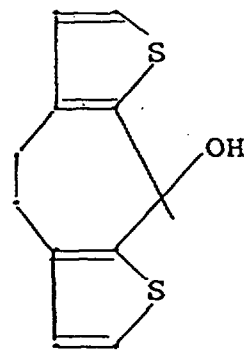
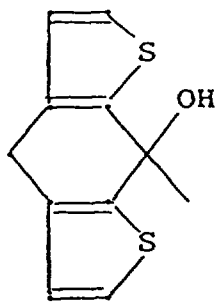
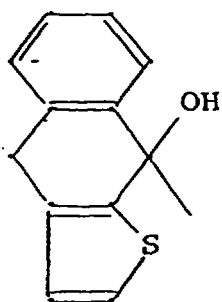
The group Z may have the following meanings, for example, whilst the aromatic groups may also be substituted, e.g. by CH_3 , OCH_3 , F or Cl:



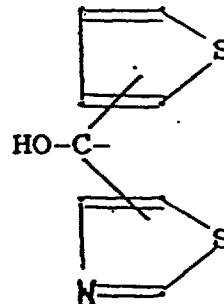
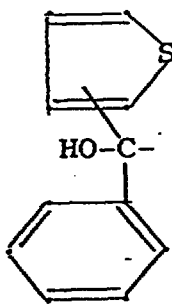
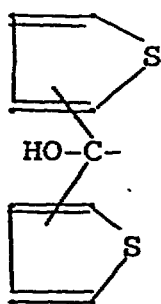


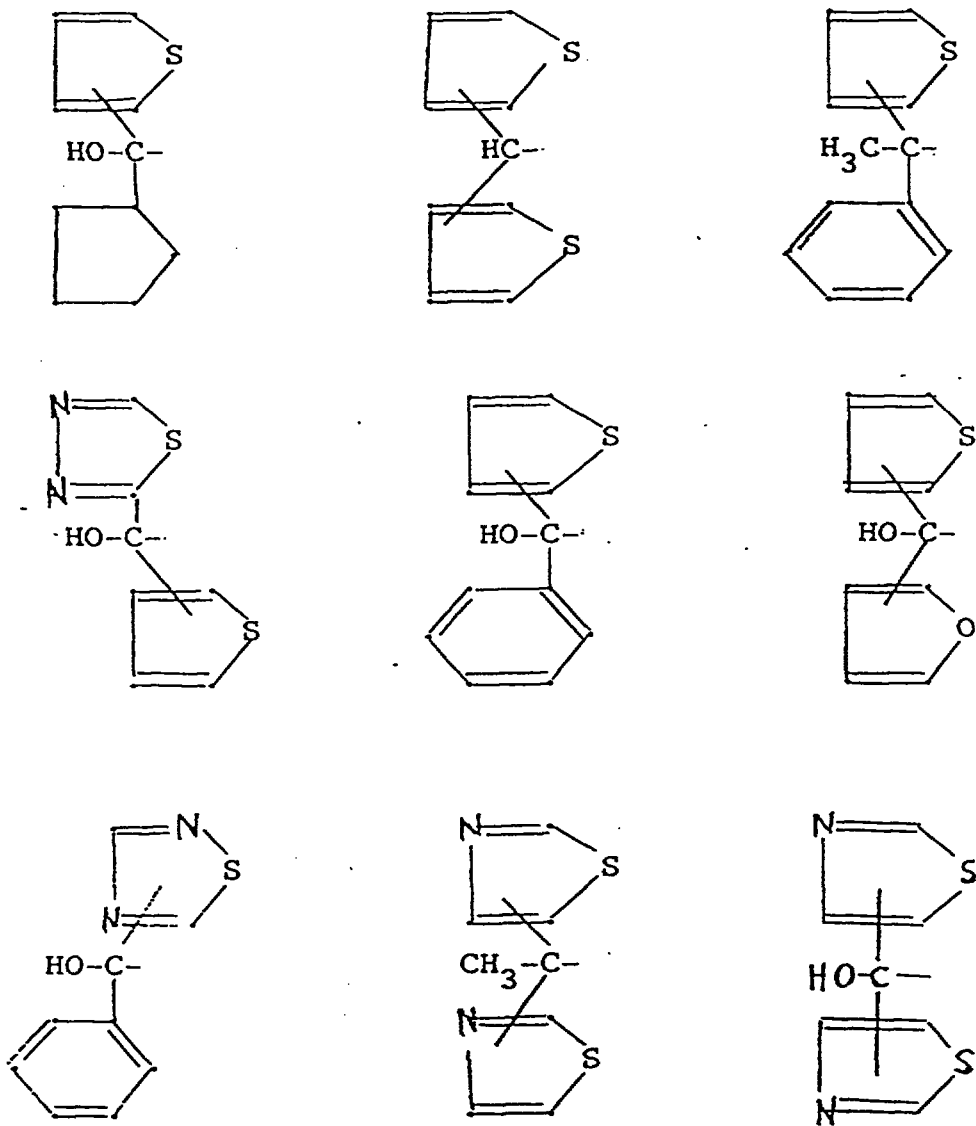
This is the replacement sheet

Date: 28 AUG 1996



If in the group II m represents O and/or Q represents $-\text{CH}_2-\text{Q}''-(\text{CH}_2)_p-$, the group III may have the following additional meanings, in particular:





The quaternary compounds of formula I are particularly suitable for therapeutic use, while the tertiary compounds are important not only as active substances but also as intermediate products.

The compounds according to the invention are anticholinergics with a potent and long-lasting effect. At dosages in the microgram range, periods of effect of more than 24 hours are achieved after inhalation. Moreover, the toxicity is in the same range as that of the standard commercial product ipratropium bromide, whilst at the same time the therapeutic effect is in

This is the replacement sheet

Date: 28 AUG 1996

some cases significantly greater.

- In accordance with their nature as anticholinergics, the new compounds are suitable, for example, for treating chronically obstructive bronchitis and (slight to moderate) asthma and also for treating vagally induced sinus bradycardia. Whereas in diseases of the respiratory tract it is chiefly recommended to administer the new active substances by inhalation (especially the quaternary compounds), thereby largely eliminating any side effects, the compounds are preferably administered by intravenous or oral route in the case of sinus bradycardia. It has been found to be advantageous that the new compounds have virtually no effect on gastrointestinal motility.

For use, the compounds according to the invention are processed with known excipients and/or carriers to form conventional galenic preparations, e.g. solutions for inhalation, suspensions in liquefied propellant gases, preparations containing liposomes or proliposomes, injectable solutions, plain or coated tablets, capsules, powders for inhalation for use in conventional inhalers.

This is the replacement sheet

Date: 28 AUG 1996

Examples of formulations (amounts given in percent by weight):

1. Metering aerosol

Active substance according to the invention	0.005
Sorbitantriolate	0.1
Monofluorotrichloromethane and difluorodichloromethane 2 : 3	
	to 100

The suspension is poured into a conventional aerosol container with a metering valve. Preferably, 50 μ l. of suspension are dispensed on each actuation. The active substance may also be dispensed in a higher dosage if desired (e.g. 0.02% by weight).

2. Tablets

Active substance according to the invention	0.05
Colloidal silica	0.95
Lactose	65.00
Potato starch	28.00
Polyvinylpyrrolidone	3.00
Na-cellulose glycolate	2.00
Magnesium stearate	1.00

The ingredients are processed in the usual way to form tablets weighing 200 mg.

This is the replacement sheet

Date: 28 AUG 1996

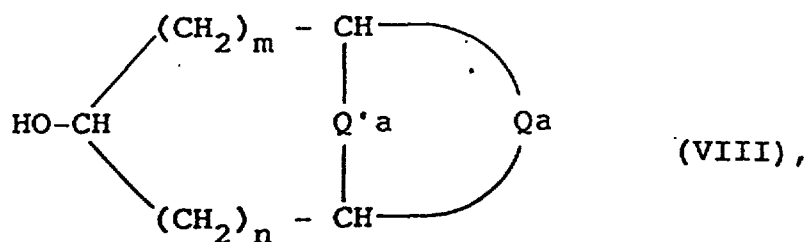
The advantageous properties of the new compounds are found for example in their inhibition of broncholysis in rabbits (acetylcholine spasm i.v.). After intravenous administration of the new active substances (dosage 3 μ g/kg i.v.) the maximum effect was obtained after 10 to 40 minutes. Even on isolated organs, e.g. on the guinea-pig ileum or rectum, numerous compounds according to the invention were found to have a long duration of activity.

The new compounds may be prepared by methods known per se.

1. Preferably an ester of formula

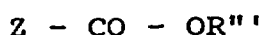


wherein Z is as hereinbefore defined and R" represents a C₁₋₄-alkyl group, preferably methyl or ethyl, is transesterified, in the presence of a conventional transesterification catalyst, with an aminoalcohol of formula



wherein m and n are as hereinbefore defined, Q' represents NR' or NH, Q_a has the same meaning as Q, with the proviso that if Q_a represents -CH₂-Q''-(CH₂)_p-, Q'' can only represent NR', and wherein the OH-group is in the α- or β-position, or

2. a reactive derivative (R' represents a readily cleavable group)



(IX)

of the acid $Z-CO-OH$, particularly an acid chloride or imidazolid thereof, is reacted with an alcohol of formula VIII, optionally in excess or in the presence of a tertiary amine such as triethylamine, and optionally the resulting compound

- a) if Q'_a represents NR' , is quaternised with a reactive monoderivative $X-R$ of a corresponding alkane (X = leaving group) or
- b) if Q'_a represents NH , is quaternised with a terminally disubstituted alkane $X-(C_{4-6}-alkylene)-X$ without intermediate isolation or
- c) if Q_a equals $-CH_2-NR'-(CH_2)_p-$, is quaternised with a reactive monoderivative $X-R$.

Alternatively, starting compounds may be quaternised, in which the starting compound VIII contains R representing a "halo- or hydroxy-substituted alkyl group" instead of R' at the nitrogen atom.

The transesterification according to process 1 is carried out with heating in an organic solvent, e.g. toluene, xylene or heptane, or in a melt, using strong bases such as sodium methoxide, sodium ethoxide, sodium hydride or metallic sodium as catalyst. In order to eliminate the lower alcohol released from the equilibrium, reduced pressure is used, and possibly the alcohol is distilled off azeotropically. The transesterification is carried out at temperatures generally not exceeding $95^\circ C$. Frequently, transesterification proceeds more easily in a melt. The reaction according to process 2 is carried out in an organic solvent or mixture of solvents which is sufficiently inert under the reaction conditions, e.g.

This is the replacement sheet

Date: 28 AUG 1996

acetone or acetonitrile, at temperatures between about 0°C and the boiling temperature of the reaction mixture.

The free bases may be obtained from acid addition salts of the tertiary amines, if desired, using suitable basic compounds in a manner known per se. The quaternisation is carried out in suitable solvents, e.g. acetonitrile or acetonitrile/methylene chloride, preferably at ambient temperature; the preferred quaternising reagent is a corresponding alkyl halide, e.g. alkyl bromide, or a corresponding sulphonic acid derivative, e.g. a methane- or toluenesulphonic acid derivative. Transesterification products wherein Q' represents NH are used as starting materials for those compounds in which R and R' together represent a C₄₋₆-alkylene group. Conversion into the tertiary and then quaternary compound is carried out using suitable 1,4-, 1,5- or 1,6-dihaloalkanes without intermediate isolation.

The starting compounds, where they have not already been described, may be obtained analogously to known compounds.

Examples:

Methyl di-(2-thienyl)glycolate from dimethyloxalate and 2-thienylmagnesium bromide;

Ethyl di-(2-thienyl)glycolate from (2-thienyl)glyoxylic acid and 2-thienyllithium;

Ethyl hydroxyphenyl-(2-thienyl)acetate from methylphenylglyoxylate and 2-thienylmagnesium bromide or from methyl (2-thienyl)glyoxylate and phenylmagnesium bromide.

Similarly, methyl 2-thienylglyoxylate and cyclohexyl- or cyclopentylmagnesium bromide may be reacted.

There are also several possible methods of preparing the aminoalcohols.

This is the replacement sheet

Date: 28 AUG 1996

Pseudoscopine can be obtained according to M. Pelonovski et al., Bull. soc. chim. 43, 79 (1928).

Pseudotropenol can be isolated from the mixture (by fractional crystallisation or distillation) which is obtained, for example, according to V. Hayakawa et al., J. Amer.Chem.Soc. 1978, 100(6), 1786 or R. Noyori et al., J.Amer.Chem.Soc. 1974, 96(10), 3336.

N-ethylnorscopine and N-isopropylnorscopine may be prepared by hydrogenolysis from the corresponding N-alkylnorscopolamines analogously to Banholzer DE-A P 3215933. 6-methyl-6-azabicyclo[3.2.1]octan-3- α -ol can be prepared according to F.I.Carroff et al., J. Med.Chem. 30, 805 (1987), and 7-methyl-7-azabicyclo-[2.2.1]heptan-2 α -ol may be obtained according to J.R. Pfister et al., J. Pharmac. Sciences 74, 208 (1985).

Starting from 2- or 3-furylglyoxyl nitrile the corresponding methylesters may be prepared in conventional manner via the 2- or 3-furylglyoxylic acid obtainable from the starting material. From these methylesters, the corresponding glycolic acid esters may be obtained as described with the organometallic derivatives of 2- or 3-bromothiophene. The organometallic compounds obtainable from 2-, 3- or 4-halopyridine can be reacted with methyl 2- or 3-thienylglyoxylate to obtain the corresponding glycolic acid esters.

Thienylglycolic acid esters in which the thiophene ring in the 2- or 3-position contains fluorine may be obtained, for example, starting from 2-fluorothiophene or 3-fluorothiophene (bromination to obtain 2-bromo-3-fluoro- or 2-bromo-5-fluorothiophene and, after conversion into corresponding organometallic compounds, reaction with suitable glyoxylic acid esters to obtain the glycolic acid esters.

2-fluorothiophene and 3-fluorothiophene may be reacted analogously to Unterhalt, Arch.Pharm. 322, 839 (1989) to obtain the corresponding glyoxylic acid esters

This is the replacement sheet

28 AUG 1986

Date:

which may then in turn be reacted with 2- or 3-thienyl derivatives, for example, to obtain glycolic acid esters, as described above. By a suitable choice of components, symmetrically substituted dithienylglycolic acid esters may be prepared analogously.

A method analogous to benzoin condensation and benzylic acid rearrangement is also possible.

The acid chlorides required may be obtained from the acids and thionyl chloride whilst the imidazolides may be obtained from the acids and carbonyldiimidazole.

The following Examples illustrate the invention without restricting it.

Example 1

Benzylic acid scopine ester-methobromide

a) Benzylic acid scopine ester from α -chlorodiphenylacetic acid chloride and scopine

26.5 g (0.1 mol) α -chlorodiphenylacetic acid chloride are added to a solution of 31.0 g (0.2 mol) of scopine in 60 ml of anhydrous pyridine at 0°C within 50 minutes with stirring. After it has all been added the mixture is stirred for 4 hours without cooling and then left to stand for 24 hours. In order to work up the mixture the scopine hydrochloride precipitated is suction filtered. The solution separated off is evaporated down under reduced pressure, the residue is dissolved in a mixture of 600 ml of water and 15 ml of conc. hydrochloric acid and heated to about 80°C for 10 minutes. At a temperature below 20°C, sodium carbonate is added until a pH of 9 is achieved. The benzylic acid scopine ester is extracted with methylene chloride and the extracts are dried over sodium sulphate. After evaporation and treatment with acetone, white crystals are obtained, m.p. 182-3°C (decomp.), yield 31.8 g (87% of theory). Elementary analysis and spectra confirm that the title compound has been obtained which can be converted in the usual way into the hydrochloride, m.p. 256°C (decomp.; from ethanol).

b) Benzylic acid scopine ester from benzylic acid
imidazolidine and scopine

A suspension of 7.13 g (0.046 mol) of scopine and 3.2 g (0.0115 mol) of benzylic acid imidazolidine in 50 ml of acetone is heated to boiling point. After about 10 minutes a further 9.6 g (0.0345 mol) of benzylic acid imidazolidine are gradually added thereto. After the reaction has ended the mixture is cooled with ice/common salt. The crystals precipitated are suction filtered. They may be converted into the hydrochloride, m.p. 256°C (decomp.; from ethanol). Yield 8.9 g, 53% of theory).

c) Benzylic acid scopine ester methobromide

7.12 g (0.075 mol) of methylbromide dissolved in acetone are added to a suspension of 5.48 g (0.015 mol) of benzylic acid scopine ester in 120 ml of acetonitrile and 20 ml of carbon tetrachloride and the mixture is left to stand under a slight overpressure until the reaction has ended. The crystals precipitated are suction filtered, washed with cold acetonitrile then with diethylether and after drying (at 40°C under reduced pressure) recrystallised from methanol/ether, m.p. 200°C (decomp.). Elementary analysis and spectrum confirm that the desired compound has been obtained.

This is the replacement sheet

Date: 28 AUG 1996

Example 2

1-N- β -Fluoroethylnorscopolamine-methobromide

a) 1-N- β -Fluoroethylnorscopolamine-hydrochloride

A mixture of 16.3 (0.05 mol) of 1-norscopolamine-hydrochloride, 6.3 g (0.05 mol) of 2-bromo-fluoroethane, 10.6 g (0.1 mol) of sodium carbonate and 100 ml of acetonitrile is refluxed for 6 hours. Then a further 6.3 g (0.05 mol) of 2-bromofluoro-ethane and 5.3 g (0.05 mol) of sodium carbonate are added and the mixture is heated for a further 24 hours. Finally, 3.2 g (0.025 mol) of 2-bromofluoroethane and 2.7 g (0.025 mol) of sodium carbonate are added and the mixture is heated for a further 48 hours. It is suction filtered and the solution is concentrated by evaporation. The residue is taken up in methylene chloride, extracted with water and the methylene chloride phase is dried over sodium sulphate. After distillation of the methylene chloride an oily residue is obtained which is reacted in the usual way to form the hydrochloride. From methanol/ether 13.1 g of white crystals are obtained (70.4% of theory), m.p. 197-8°C (decomp.).

b) Reaction to form the methobromide

7.0 g (0.021 mol) of the amine liberated in the usual way from the hydrochloride obtained in a) are reacted in 20 ml of absolute acetonitrile with 9.9 g (0.104 mol) of methylbromide for 6 days under a slight overpressure. The crystals precipitated are recrystallised from methanol/ether. 3.9 g of white crystals, m.p. 194°C (decomp.).

This is the replacement sheet

Date: 28 AUG 1996

Elementary analysis and spectra confirm that the title compound has been obtained.

Example 3

Mandelic acid scopine ester methobromide

155.2 g (1.0 mol) of scopine are dissolved in 200 ml of absolute methylene chloride and 116.9 g (0.55 mol) of acetylmandelic acid chloride dissolved in 100 ml of absolute methylene chloride are added dropwise thereto (at 20°C within 1 hour). (The acetylmandelic acid chloride is obtained from acetylmandelic acid and thionyl chloride). After one hour the scopine hydrochloride precipitated is separated off, the methylene chloride solution is extracted with water and dried.

The combined aqueous phases are made alkaline with sodium carbonate, extracted with methylene chloride and the methylene chloride phase is dried. The solvent is distilled off from the combined methylene chloride solutions. The base which remains is converted into the hydrochloride in the usual way. After recrystallisation from acetonitrile, 124.6 g (67.7% of theory) of white crystals are obtained, m.p. 207°C (decomp.).

27.5 g (0.075 mol) of the acetyl compound thus obtained are left to stand in 110 ml of 20% hydrochloric acid for 20 hours at ambient temperature. Whilst cooling, the reaction solution is made alkaline and the mandelic acid scopine ester is extracted with methylene chloride. After drying over sodium sulphate and distillation of the solvent, the hydrochloride is prepared in the usual way. From methanol/ether, 22.5 g (92.4% of theory) of white crystals are obtained, m.p. 141-2°C.

10.7 g (0.037 mol) of the ester liberated from the hydrochloride in the usual way are left to stand for 40

This is the replacement sheet

Date: 28 AUG 1996

hours in acetonitrile with 17.58 g (0.185 mol) of methylbromide under a slight overpressure.

The crystals precipitated are suction filtered, washed with cold acetonitrile and recrystallised from methanol/ether. 10.0 g (70.4% of theory) of white crystals are obtained, m.p. 223°C (decomp.).

Elementary analysis and spectra confirm that the title compound has been obtained.

Example 4

Xanthene-9-carboxylic acid scopine ester methobromide

- a) At 20°C, a solution of 11.1 g (0.11 mol) of triethylamine in 20 ml of acetone and 26.9 g (0.11 mol) of xanthene-9-carboxylic acid chloride (obtained from xanthene-9-carboxylic acid and thionyl chloride) in 80 ml of acetone are simultaneously added dropwise to a solution of 15.5 g (0.01 mol) of scopine in 50 ml of acetone at 20°C. After 4 hours a further 1.1 g (0.01 mol) of triethylamine and 2.69 (0.011 mol) of xanthene-9-carboxylic acid chloride are added. After 4 days the mixture is suction filtered and the solvent is distilled off from the solution. Sodium carbonate solution is added to the residue and extracted with methylene chloride. The organic phase is dried over sodium sulphate and the solvent is distilled off. From the residue, the hydrochloride of the resulting xanthene-9-carboxylic acid scopine ester is obtained in the conventional manner; white crystals from acetonitrile/ether, m.p. 223°C (decomp.); yield 21.8 g.
- b) The base is liberated in the usual way from a sufficient quantity of the hydrochloride obtained in a). 36.3 g (0.1 mol) thereof are reacted in a

This is the replacement sheet

Date: 28 AUG 1996

solution of 47.5 g (0.5 mol) of methylbromide in 49 g of acetonitrile for 24 hours under slight overpressure. The crystals obtained are suction filtered, washed with acetone/ether and recrystallised from ethanol. Yield 44.0 g (95.8% of theory), white crystals, m.p. 139°C. The crystals contain 0.5 mol of ethanol. Elementary analysis and spectra confirm the presence of the title compound.

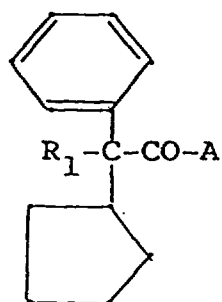
The other compounds according to the invention may also be obtained in accordance with the Examples given above. The compounds melt with decomposition.

This is the replacement sheet

Date: 28 AUG 1996

T A B L E I

Compounds of formula



No.	R_1	A	M.p. [$^{\circ}\text{C}$]
1	OH	3 α - Tropanyl-methobromide	275-6
2	OH	3 α - Tropanyl- β -fluorethobromide	205-6
3	H	3 α -N-Ethyl-(6 β ,7 β -Epoxy)- nortropanyl-methobromide	228
4	H	3 α -N-Propyl-(6 β ,7 β -Epoxy)- nortropanyl-methobromide	206-7
5	H	3 α -N-Isopropyl-(6 β ,7 β -Epoxy)- nortropanyl-methobromide	218
6	H	3 α -(6 β ,7 β -Epoxy)- tropanyl-methobromide	207
7	H	3 α -(6,7 -Dehydro)tropanyl- methobromide	226-8

This is the replacement sheet

Date: 28 AUG 1996

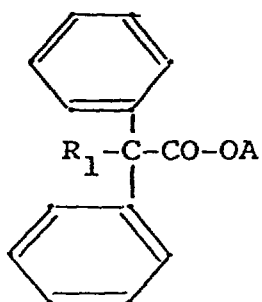
No.	R ₁	A	M.p. [°C]
8	H	3 α -Tropanyl-methobromide	275-6
9	H	3 α -N-Ethyl-nortropanyl-methobromide	256-7
10	H	(-)-3 α -N-Isopropyl-nortropanyl-methobromide	256
11	H	(+)-3 α -N-Isopropyl-nortropanyl-methobromide	256
12	H	3 α -Nortropanyl-8,1'-pyrrolidinium-bromide	267-70
13	H	(+)-3 α -Tropanyl-methobromide	278-81
14	H	(-)-3 α -Tropanyl-methobromide	278-81

This is the replacement sheet

28 AUG 1956
E14155

T A B L E II

Compounds of formula



No.	R ₁	A	M.p. [°C]
1	OH	3α-N-Isopropyl-nortropanyl-methobromide	258
2	OH	3α-N-β-Chlorethyl-nortropanyl-methobromide	203
3	OH	3α-N-Ethyl-nortropanyl-methobromide	269
4	OH	3α-Tropanyl-ethobromide	258
5	OH	3α-(6β,7β-Epoxy)tropanyl-methobromide	200
6	OH	3α-N-Ethyl-(6β,7β-Epoxy)-nortropanyl-methobromide	220-1
7	OH	3α-(6β,7β-Epoxy)-N-isopropyl-nortropanyl-methobromide	234-5

This is the replacement sheet

Date: 28 AUG 1996

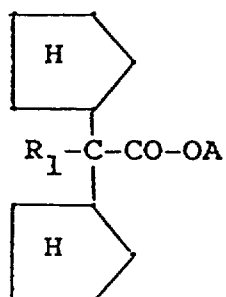
No.	R ₁	A	M.p. [°C]
8	OH	3 α -N-Methylgranatanyl-methobromide	249
9	OH	3 α -N-Isopropylgranatanyl-methobromide	219-20
10	OH	3 α -(6,7-Dehydro)tropanyl-methobromide	207-8
11	H	3 α -(6,7-Dehydro)tropanyl-methobromide	214-5
12	OH	3 α -(6,7-Dehydro)-N-isopropyl-nortropanyl-methobromide	223
13	H	3 α -Nortropanyl-8,1'-pyrrolidinium-bromide	231-2

This is the replacement sheet

Date: 28 AUG 1996

T A B L E III

Compounds of formula



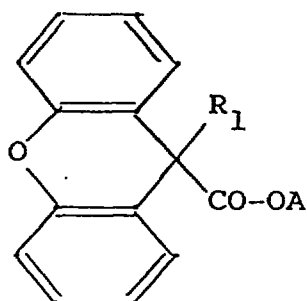
No.	R ₁	A	M.p.[°C]
1	H	3α-(6β,7β-Epoxy)-tropanyl-methobromide	213-4
2	H	3α-(6β,7β-Epoxy)-tropanyl-propargochloride	204-5
3	H	3α-N-Ethyl-(6β,7β-Epoxy)-nortropanyl-methobromide	201
4	H	3α-(6,7-Dehydro)tropanyl-methobromide	238-9
5	H	3α-Tropanyl-methobromide	203-5
6	OH	3α-(6β,7β-Epoxy)tropanyl-metho-methanesulphonate	

This is the replacement sheet

Date: 28 AUG 1996

T A B L E IV

Compounds of formula



No.	R ₁	A	M.p. [°C]
1	H	3α-(6β,7β-Epoxy)-N-n-propyl nortropanyl-methobromide	213-4
2	H	3α-(6β,7β-Epoxy)-N-isopropyl- nortropanyl-methobromide	242
3	H	3α-(6β,7β-Epoxy)-N-ethyl- nortropanyl-methobromide	217
4	H	3α-(6β,7β-Epoxy)tropanyl- methobromide (with crystal ether)	139
5	H	3α-(6β,7β-Epoxy)tropanyl- ethobromide	128-31
6	H	3α-Tropanyl-ethobromide	212-3
7	OH	3α-N-Isopropyl-nortropanyl- metho-methanesulphonate	229-32

This is the replacement sheet

Date: 28 AUG 1996

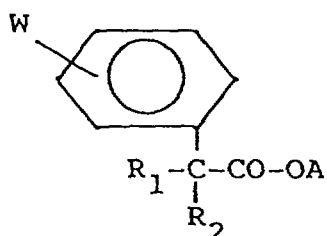
No.	R ₁	A	M.p. [°C]
8	H	3 α -N-Isopropyl-nortropanyl-methobromide	184-5
9	H	3 α -(6,7,-Dehydro)-N-isopropyl-nortropanyl-methobromide	259
10	H	3 α -(6,7 -Dehydro)-tropanyl-methobromide	237-8

This is the replacement sheet

Date: 28 AUG 1996

Table V

Compounds of formula



No.	R ₁	R ₂	W	A	M.p. [°C]
1	OH	H	H	3α-(6β,7β-Epoxy)- tropanyl-methobromide	223
2	CH ₂ OH	H	H	3α-(6β,7β-Epoxy)- tropanyl-ethobromide	190
3	H	Cyclo- heptyl	H	3α-(6β,7β-Epoxy)- tropanyl-methobromide	242
4	H	Cyclo- heptyl	H	3α-(6β,7β-Epoxy)- tropanyl-propobromide	215-6
5	-C ₅ H ₈ -		H	3α-(6β,7β-Epoxy)- tropanyl-methobromide	223-4
6	CH ₂ OH	H	H	3α-(6β,7β-Epoxy)-N- (β-fluorethyl)-nor- tropanyl-methobromide	194
7	CH ₂ OH	H	H	3α-(6β,7β-Epoxy)-N- (β-hydroxyethyl)- nortropanyl- methobromide	211
8	OH	C ₆ H ₄	4-F	3α-Tropanyl- methobromide	220-1

This is the replacement sheet

Date: 28 AUG 1996

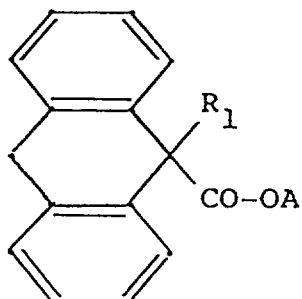
No.	R ₁	R ₂	W	A	M.p. [°C]
9	-C ₅ H ₈ -		H	3α-Tropanyl-methobromide	287-9
10	-C ₅ H ₈ -		H	3α-N-Isopropyl-nortropanyl-methobromide	263
11	-C ₆ H ₁₀ -		H	3α-N-Isopropyl-nortropanyl-methobromide	261
12	OH	C ₆ H ₁₁	H	3α-(6,7-Dehydro)-tropanyl-methobromide	233-5
13	H	C ₆ H ₁₁	3-CH ₃	3α-N-Isopropyl-nortropanyl-methobromide	252-4
14	H	C ₅ H ₉	3-CH ₃	3α-Nortropanyl-8,1'-pyrrolidinium-bromide	224-6

This is the replacement sheet

Date: 28 AUG 1996

Table VI

Compounds of formula



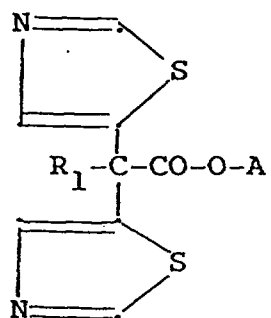
No.	A	R ₁	M.p. [°C]
1	3α-(6β,7β-Epoxy)-tropanyl-methobromide . H ₂ O	H	176
2	3α-Tropanyl-methobromide	H	
3	3α-(6,7-Dehydro)-tropanyl-methobromide	OH	
4	3α-(N-β-Fluorethyl)-nortropanyl-methobromide	H	
5	3α-Tropanyl-β-fluorethobromide	OH	
6	3α-(N-Isopropyl)-granatanyl-methobromide	H	
7	3α-(N-Isopropyl)-nortropanyl-methobromide	H	
8	3α-(6β,7β-Epoxy)-N-isopropyl-nortropanyl-methobromide	OH	
9	3α-(6β,7β-Epoxy)-N-ethylnor-tropanyl-methobromide	OH	
10	3α-(N-Ethyl)-nortropanyl-methobromide	OH	
11	3α-(N-Methyl)-granatanyl-methobromide	CH ₃	

This is the replacement sheet

Date: 28 AUG 1996

Table VII

Compounds of formula



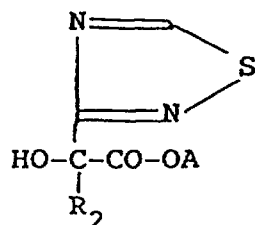
No.	A	R_1	M.p. [°C]
1	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	H	
2	3 α -(6,7-Dehydro)-tropanyl-methobromide	H	
3	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	Methyl	
4	3 α -(6,7-Dehydro)-tropanyl-methobromide	Methyl	
5	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	OH	
6	3 α -(6,7-Dehydro)-tropanyl-methobromide	OH	

This is the replacement sheet

Date: 28 AUG 1996

Table VIII

Compounds of formula



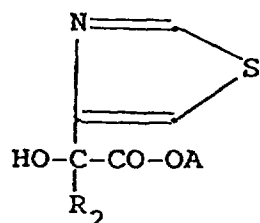
No.	A	R ₂	M.p. [°C]
1	3α-(6β,7β-Epoxy)-tropanyl-methobromide	Phenyl	
2	3α-(6,7-Dehydro)-tropanyl-methobromide	Phenyl	
3	3α-(6β,7β-Epoxy)-tropanyl-methobromide	Cyclopentyl	
4	3α-(6,7-Dehydro)-tropanyl-methobromide	3-Thienyl	
5	3α-Tropanyl-methobromide	3-Thienyl	
6	3α-(N-Methyl)-granatanyl-methobromide	3-Thienyl	

This is the replacement sheet

Date: 28 AUG 1996

Table IX

Compounds of formula



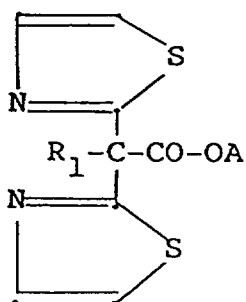
No.	A	R ₂	M.p. [°C]
1	3α-(6β,7β-Epoxy)-tropanyl-methobromide	Phenyl	
2	3α-(6,7-Dehydro)-tropanyl-methobromide	Phenyl	
3	3α-(6β,7β-Epoxy)-tropanyl-methobromide	Cyclopentyl	
4	3α-(6,7-Dehydro)-tropanyl-methobromide	3-Thienyl	
5	3α-Tropanyl-methobromide	3-Thienyl	
6	3α-(N-Methyl)-granatanyl-methobromide	3-Thienyl	

This is the replacement sheet

Date: 28 AUG 1996

Table X

Compounds of formula



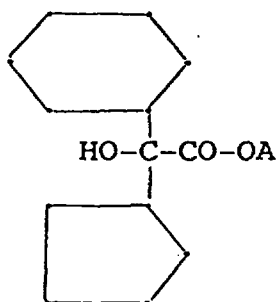
No.	A	R ₁	M.p. [°C]
1	3α-(6β,7β-Epoxy)-tropanyl-methobromide	H	
2	3α-(6,7-Dehydro)-tropanyl-methobromide	H	
3	3α-(6β,7β-Epoxy)-tropanyl-methobromide	Methyl	
4	3α-(6,7-Dehydro)-tropanyl-methobromide	Methyl	
5	3α-(6β,7β-Epoxy)-tropanyl-methobromide	OH	
6	3α-(6,7-Dehydro)-tropanyl-methobromide	OH	

This is the replacement sheet

Date: 28 AUG 1996

Table XI

Compounds of formula



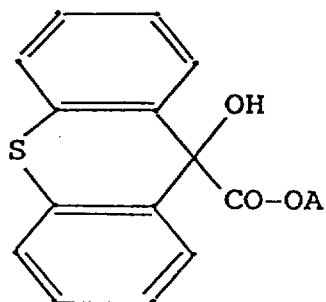
No.	A	M.p. [°C]
1	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	
2	3 α -(6,7-Dehydro)-tropanyl-methobromide	
3	3 α -(6 β ,7 β -Epoxy)-tropanyl-ethobromide	
4	3 α -(6,7-Dehydro)-tropanyl-ethobromide	
5	3 α -Tropanyl methobromide	
6	3 α -(N-Methyl)-granatanyl-methobromide	

This is the replacement sheet

Date: 2 8 AUG 1996

Table XII

Compounds of formula



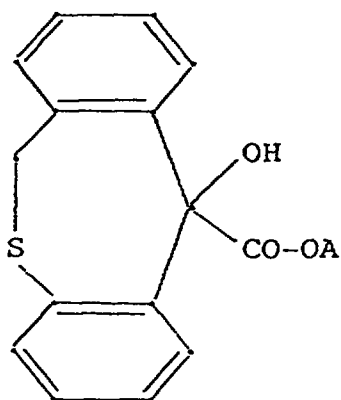
No.	A	M.p. [°C]
1	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	
2	3 α -(6,7-Dehydro)-tropanyl-methobromide	
3	3 α -(6 β ,7 β -Epoxy)-tropanyl-ethobromide	
4	3 α -(6,7-Dehydro)-tropanyl-ethobromide	
5	3 α -Tropanyl methobromide	
6	3 α -(N-Methyl)-granatanyl-methobromide	

This is the replacement sheet

Date: 28 AUG 1996

Table XIII

Compounds of formula



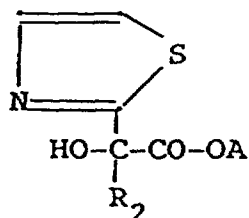
No.	A	M.p. [°C]
1	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	
2	3 α -(6,7-Dehydro)-tropanyl-methobromide	
3	3 α -(6 β ,7 β -Epoxy)-tropanyl-ethobromide	
4	3 α -(6,7-Dehydro)-tropanyl-ethobromide	
5	3 α -Tropanyl methobromide	
6	3 α -(N-Methyl)-granatanyl-methobromide	

This is the replacement sheet

Date: 2 8 AUG 1996

Table XIV

Quarternary compounds of formula



No.	A	R ₂	M.p. [°C]
1	3α-(6β,7β-Epoxy)-tropanyl-methobromide	2-Thienyl	
2	3α-Tropanyl-methobromide	2-Thienyl	
3	3α-(6,7-Dehydro)-tropanyl-methobromide	2-Thienyl	
4	3α-(N-β-Fluorethyl)-nortropanyl-methobromide	2-Thienyl	
5	3α-Tropanyl-β-fluorethobromide	2-Thienyl	
6	3α-(N-Isopropyl)-granatanyl-methobromide	2-Thienyl	
7	3α-(N-Isopropyl)-nortropanyl-methobromide	2-Thienyl	
8	3α-(6β,7β-Epoxy)-N-isopropyl-nortropanyl-methobromide	2-Thienyl	
9	3α-(6β,7β-Epoxy)-N-ethylnortropanyl-methobromide	2-Thienyl	
10	3α-(N-Ethyl)-nortropanyl-methobromide	2-Thienyl	
11	3α-(N-Methyl)-granatanyl-methobromide	2-Thienyl	

This is the replacement sheet

Date: 2 8 AUG 1996

No.	A	R ₂	M.p. [°C]
12	3 α -(6 β ,7 β -Epoxy)-N- β -fluoroethyl -nortropanyl-methobromide	2-Thienyl	
13	3 α -(6 β ,7 β -Epoxy)-N-n-propyl- nortropanyl-methobromide	2-Thienyl	
14	3 α -Tropanyl- β -hydroxyethobromide	2-Thienyl	
15	3 α -(6 β ,7 β -Epoxy)-tropanyl- methobromide	Phenyl	
16	3 α -Tropanyl-methobromide	Phenyl	
17	3 α -(N- β -Fluoroethyl)- nortropanyl-methobromide	Phenyl	
18	3 α -(6,7-Dehydro)-tropanyl- methobromide	Phenyl	
19	3 α -(N-Ethyl)-nortropanyl- methobromide	Phenyl	
20	3 α -(N-Isopropyl)-nortropanyl- methobromide	Phenyl	
21	3 α -Tropanyl-ethobromide	Phenyl	
22	3 α -(N-Ethyl)-nortropanyl- ethobromide	Phenyl	
23	3 α -(6 β ,7 β -Epoxy)-tropanyl- ethobromide	Phenyl	
24	3 α -Tropanyl- β -fluorthobromide	Phenyl	
25	3 α -Tropanyl-methobromide	Cyclohexyl	
26	3 α -(N- β -Fluorethyl)-nortropanyl- methobromide	Cyclohexyl	
27	3 α -Tropanyl- β -fluorethobromide	Cyclohexyl	
28	3 α -Tropanyl-methobromide	Cyclopentyl	
29	3 α -Tropanyl-ethobromide	Cyclopentyl	
30	3 α -(N-Ethyl)-nortropanyl- methobromide	Cyclopentyl	
31	3 α -(N-Isopropyl)-nortropanyl- methobromide	Cyclopentyl	

This is the replacement sheet

Date: 28 AUG 1996

No.	A	R ₂	M.p. [°C]
32	3 α -Tropanyl- β -fluorethobromide	Cyclopentyl	
33	3 α -(N- β -Fluorethyl)-nortropanyl-methobromide	Cyclopentyl	
34	3 α (6,7-Dehydro)-tropanyl-metho-methanesulphonate	2-Thienyl	
35	3 β -(6 β ,7 β -Epoxy)-tropanyl-methobromide	2-Thienyl	
36	3 β -Tropanyl-methobromide	2-Thienyl	
37	3 β -(6,7-Dehydro)-tropanyl-methobromide	2-Thienyl	
38	3 α (6,7-Dehydro)-tropanyl-methobromide	3-Thienyl	
39	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	3-Thienyl	
40	(+)-Enantiomer of No. 1		
41	(-)-Enantiomer of No. 1		
42	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	2-Furyl	
43	3 α -(6,7-Dehydro)-tropanyl-methobromide	2-Furyl	
44	3 α -Tropanyl-methobromide	2-Furyl	
45	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	2-Pyridyl	
46	3 α -(6,7-Dehydro)-tropanyl-methobromide	2-Pyridyl	
47	3 α -Tropanyl-methobromide	2-Pyridyl	
48	3 α -Tropanyl-methobromide	3-Thienyl	
49	3 α -(6,7-Dehydro)-tropanyl-methobromide	Cyclopentyl	

This is the replacement sheet

Date: 28 AUG 1996

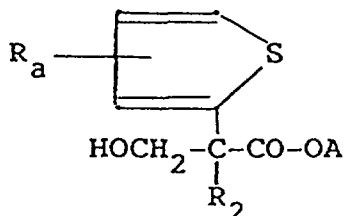
No.	A	R ₂	M.p. [°C]
50	3 α -(6 β ,7 β -Epoxy)-tropanyl methobromide	Cyclohexyl	
51	3 α -(6,7-Dehydro)-tropanyl methobromide	Cyclohexyl	
52	3 α -(6 β ,7 β -Epoxy)-tropanyl methobromide	Cyclopentyl	

This is the replacement sheet

Date: 28 AUG 1996

Table XV

Compounds of formula



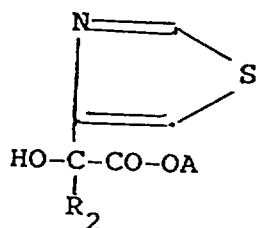
No.	A	R ₂	R _a M.p [°C]
1	3α-(6β,7β-Epoxy)-tropanyl-methobromide	2-Thienyl	5-Methyl
2	3α-(6,7-Dehydro)-tropanyl-methobromide	2-Thienyl	5-Methyl
3	3α-Tropanyl-methobromide	2-Thienyl	5-Methyl
4	3α-(6β,7β-Epoxy)-tropanyl-methobromide	2-(5-Methyl)-thienyl	5-Methyl
5	3α-(6,7-Dehydro)-tropanyl-methobromide	2-(5-Methyl)-thienyl	5-Methyl
6	3α-Tropanyl-methobromide	2-(5-Methyl)-thienyl	5-Methyl
7	3α-(6β,7β-Epoxy)-tropanyl-methobromide	2-Thienyl	5-Fluoro
8	3α-(6,7-Dehydro)-tropanyl-methobromide	2-Thienyl	5-Fluoro
9	3α-Tropanyl-methobromide	2-Thienyl	5-Fluoro
10	3α-(6β,7β-Epoxy)-tropanyl-methobromide	2-(5-Fluoro)-thienyl	5-Fluoro
11	3α-(6,7-Dehydro)-tropanyl-methobromide	2-(5-Fluoro)-thienyl	5-Fluoro
12	3α-Tropanyl-methobromide	2-(5-Fluoro)-thienyl	5-Fluoro

This is the replacement sheet

Date: 28 AUG 1996

Table XVI

Compounds of formula



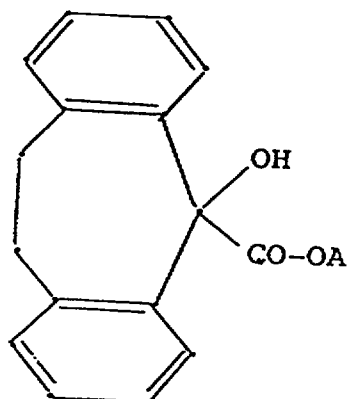
No.	A	R_2	M.p. [$^{\circ}C$]
1	3 α -(6 β ,7 β -Epoxy)-tropanyl methobromide	Phenyl	
2	3 α -(6,7-Dehydro)-tropanyl methobromide	Phenyl	
3	3 α -(6 β ,7 β -Epoxy)-tropanyl methobromide	3-Thienyl	
4	3 α -(6,7-Dehydro)-tropanyl methobromide	3-Thienyl	
5	3 α -Tropanyl methobromide	3-Thienyl	
6	3 α -(N-Methyl)-granatanyl methobromide	3-Thienyl	

This is the replacement sheet

Date: 28 AUG 1996

Table XVII

Compounds of formula



No.	A	M.p. [°C]
1	3 α -(6 β ,7 β -Epoxy)-tropanyl-methobromide	
2	3 α -Tropanyl-methobromide	
3	3 α -(6,7-Dehydro)-tropanyl-methobromide	
4	3 α -(N- β -Fluorethyl)-nortropanyl-methobromide	
5	3 α -Tropanyl- β -fluorethobromide	
6	3 α -(N-Isopropyl)-granatanyl-methobromide	
7	3 α -(N-Isopropyl)-nortropanyl-methobromide	
8	3 α -(6 β ,7 β -Epoxy)-N-isopropyl-nortropanyl-methobromide	
9	3 α -(6 β ,7 β -Epoxy)-N-ethylnor-tropanyl-methobromide	
10	3 α -(N-Ethyl)-nortropanyl-methobromide	
11	3 α -(N-Methyl)-granatanyl-methobromide	

This is the replacement sheet

Date: 28 AUG 1996

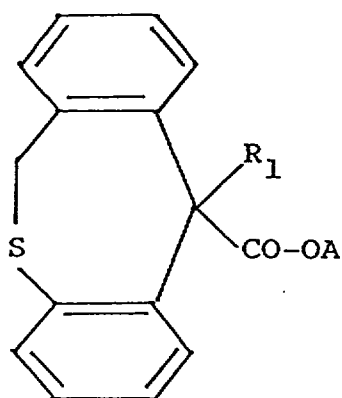
No.	A	M.p. [°C]
12	3 α -(6 β ,7 β -Epoxy)-N- β -fluoroethyl- -nortropanyl-methobromide	
13	3 α -(6 β ,7 β -Epoxy)-N-n-propyl- nortropanyl-methobromide	
14	3 α -Tropanyl- β -hydroxyethobromide	
15	3 α (6,7-Dehydro)-tropanyl- metho-methanesulphonate	
16	3 β -(6 β ,7 β -Epoxy)-tropanyl- methobromide	
17	3 β -Tropanyl-methobromide	
18	3 β -(6,7-Dehydro)-tropanyl- methobromide	

This is the replacement sheet

Date: 28 AUG 1996

Table XVIII

Compounds of formula



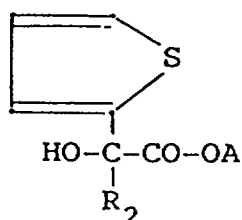
No.	A	R ₁	M.p. [°C]
1	3α-(6β,7β-Epoxy)-tropanyl methobromide	OH	
2	3α-(6,7-Dehydro)-tropanyl methobromide	OH	
3	3α-(6β,7β-Epoxy)-tropanyl methobromide	Methyl	
4	3α-(6,7-Dehydro)-tropanyl methobromide	Methyl	
5	3α-Tropanyl methobromide	H	
6	3α-(N-Ethyl)-nortropanyl methobromide	H	

This is the replacement sheet

Date: 2 8 AUG 1996

Table A

Compounds of formula



No.	A	R ₂	Base	Mp. [°C] Hydro- chloride
1	3α-(6β,7β-Epoxy)-tropanyl	2-Thienyl	149-50	238-41
2	3α-Tropanyl	2-Thienyl	167-8	253
3	3α-(6;7-Dehydro)-tropanyl	2-Thienyl	164-5	
4	3α-(N-β-Fluorethyl)- nortropanyl	2-Thienyl		236
5	3α-(N-Isopropyl)- granatanyl	2-Thienyl		232
6	3α-(N-Isopropyl)- nortropanyl	2-Thienyl		250
7	3α-(6β,7β-Epoxy)-N-iso- propylnortropanyl	2-Thienyl		206
8	3α-(6β,7β-Epoxy)-N-ethyl- nortropanyl	2-Thienyl		212-3
9	3α-(N-Ethyl)-nortropanyl	2-Thienyl		256-7
10	3α-(N-Methyl)-granatanyl	2-Thienyl		241
11	3α-(6β,7β-Epoxy)-N-β- fluorethylnortropanyl	2-Thienyl		188-90
12	3α-(6β,7β-Epoxy)-N-n- propylnortropanyl	2-Thienyl	104-6	
13	3α-(6β,7β-Epoxy)-N-n- butylnortropanyl	2-Thienyl		225-7
14	3α-(6β,7β-Epoxy)-tropanyl	Phenyl		246-7

This is the replacement sheet

Date: 28 AUG 1996

No.	A	R ₁	Base	Hydro- chloride
15	3 α -Tropanyl	Phenyl		243-4
16	3 α -(N- β -Fluoroethyl)- nortropanyl	Phenyl		219-20
17	3 α -(6,7-Dehydro)-tropanyl	Phenyl	181-3	
18	3 α -(N-Ethyl)-nortropanyl	Phenyl		231-2
19	3 α -(N-Isopropyl)- nortropanyl	Phenyl		246-7
20	3 α -Tropanyl	Cyclohexyl		260
21	3 α -(N- β -Fluoroethyl)- nortropanyl	Cyclohexyl		203-4
22	3 α -(6 β ,7 β -Epoxy)-tropanyl	Cyclopentyl		237
23	3 α -Tropanyl	Cyclopentyl		260
24	3 α -(N- β -Fluoroethyl)- nortropanyl	Cyclopentyl		182-3
25	3 α -(N-Ethyl)-nortropanyl	Cyclopentyl		227-8
26	3 α -(N-Isopropyl)- nortropanyl	Cyclopentyl		174-5
27	3 β -(6 β ,7 β -Epoxy)-tropanyl	2-Thienyl		240-2
28	3 β -Tropanyl	2-Thienyl		217-9
29	3 β -(6,7-Dehydro)-tropanyl	2-Thienyl		233-5
30	3 α -(6,7-Dehydro)-tropanyl	3-Thienyl		247-8
31	3 α -(6 β ,7 β -Epoxy)-tropanyl	3-Thienyl		242-3
32	3 α -(6 β ,7 β -Epoxy)-tropanyl	2-Furyl		
33	3 α -(6,7-Dehydro)-tropanyl	2-Furyl		
34	3 α -Tropanyl	2-Furyl		
35	3 α -Tropanyl	2-Pyridyl		
36	3 α -(6 β ,7 β -Epoxy)-tropanyl	2-Pyridyl		
37	3 α -(6,7-Dehydro)-tropanyl	2-Pyridyl		
38	3 α -Tropanyl	3-Thienyl		
39	3 α -(6,7-Dehydro)-tropanyl	Cyclopentyl		
40	3 α -(6 β ,7 β -Epoxy)-tropanyl	Cyclohexyl		
41	3 α -(6,7-Dehydro)-tropanyl	Cyclohexyl		

Note: all hydrochlorides melt with decomposition.

This is the replacement sheet

Date: 28 AUG 1996

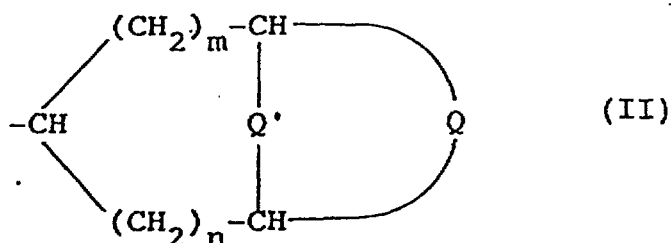
Claims

1. The new compounds of formula



wherein

A represents the group

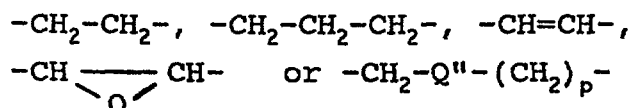


wherein

m represents 0, 1 or 2,

n represents 1 or 2 and $m + n \leq 3$,

Q represents one of the double-bonded groups



(p = 0 or 1)

and

Q' represents the group =NR, the group =NRR' or the group CH₂, wherein R represents an optionally halo- or hydroxy-substituted C₁₋₄-alkyl group, R' represents a C₁₋₄-alkyl group and R and R' together may also form a C₄₋₆-alkylene group, and, in the case of quaternary compounds, an equivalent of an anion (X^o) is associated with the positive charge of the N-atom,

This is the replacement sheet



Date:

28 AUG 1996

and

Q" has the same meanings as Q', with the exception of CH₂, and with the proviso that Q represents -CH₂-Q"-(CH₂)_p- when Q' represents CH₂,

Z represents the group

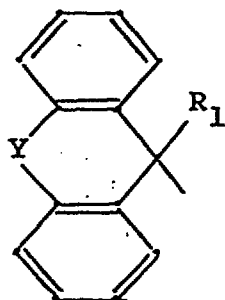


wherein

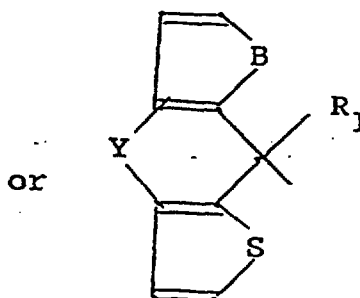
R₁ represents H, OH, CH₂OH, C₁₋₄-alkyl or C₁₋₄-alkoxy,

R₂ and R₃, which may be identical or different and one of which may also be H, represent

- (a) phenyl, furyl, an aromatic group which is isoelectronic with thienyl, C₅₋₇-cycloalkyl, pyridyl, C₅₋₇-cycloalkenyl or, in particular if m is equal to 0 and/or Q is equal to -CH₂-Q"-(CH₂)_p, R₂ may also represent thienyl,
- (b) an aliphatic group having up to 20 carbon atoms optionally interrupted by oxygen, or a C₁₋₆-alkyl group substituted by phenyl, phenoxy, thienyl, furyl, C₅₋₇-cycloalkyl or fluorine,
- (c) the entire group III may also represent the tricyclic group of the formula



(IV)



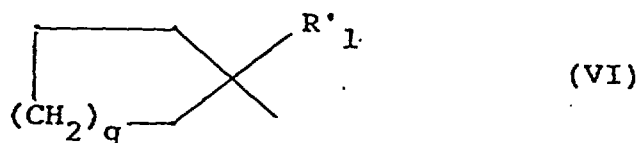
(V)



This is the replacement sheet

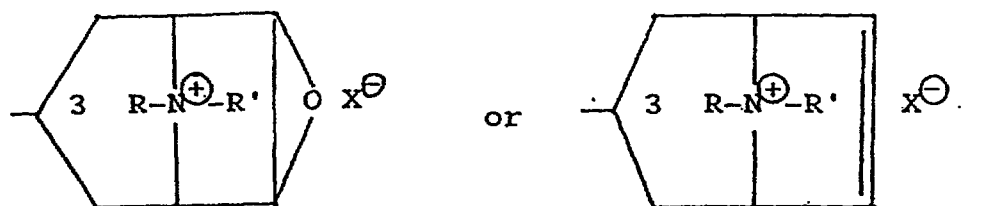
Date: 28 AUG 1996

or a group of formula



wherein B may represent S or CH=CH, R'₁ has the same meaning as R₁ and may additionally represent phenyl, thienyl, furyl, thiazolyl, thiadiazolyl or methoxyphenyl, Y represents a single bond, an O- or S-atom or one of the groups -CH₂-, -CH₂-CH₂-, -CH=CH-, -OCH₂- or -S-CH₂- and q represents 1, 2 or 3.

2. Compounds according to claim 1 wherein R₁ represents OH, CH₃ or CH₂OH.
3. Compounds according to claim 1 or 2, wherein A represents a group of the formula



wherein 3 represents the α -form.

4. Compounds according to claim 1, 2 or 3, wherein one of the groups R and R' represents methyl and the other group represents methyl, ethyl, n-propyl or i-propyl.
5. Compounds according to claim 1, 2, 3 or 4, wherein R₂ is a thienyl, thiazolyl or thiadiazolyl group.

This is the replacement sheet



Date: 28 AUG 1996

Date: 28 AUG 1996

only represent NR' ,
and wherein the OH-group is in the α - or β -
position, in the presence of a conventional
transesterification catalyst or

- b) a reactive derivative (R'' being a readily
cleavable group)



of the acid Z-CO-OH , more particularly an acid
chloride or imidazolid, is reacted with an alcohol
of formula VIII, optionally in excess or in the
presence of a tertiary amine such as triethylamine,
and optionally the resulting compound

- aa) if Q' , represents NR' , is quaternised with a
reactive monoderivative X-R of a corresponding
alkane (X = a leaving group) or
- bb) if Q' , represents NH , is quaternised with a
terminally disubstituted alkane
 $\text{X-(C}_{4-6}\text{-alkylene)-X}$ without intermediate
isolation or
- cc) if Q_2 is equal to $-\text{CH}_2-\text{NR}'-(\text{CH}_2)_p-$, is
quaternised with a reactive monoderivative X-R .



* C L H S *

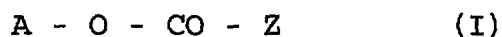
This is the replacement sheet

Date: 28 AUG 1995

New esters of bi- and tricyclic aminoalcohols,
their preparation and their use in
pharmaceutical compositions

ABSTRACT

The novel compounds of formula



(wherein A and Z are defined in the specification) may be prepared by conventional techniques; they are suitable as active ingredients for pharmaceutical compositions.

This is the replacement sheet

28 AUG 1966

Date:

4-1215000