

PROPYLENE PRODUCTION USING A MESOPOROUS SILICA FOAM METATHESIS CATALYST

CROSS REFERENCE TO RELATED APPLICATIONS

[0001] This application claims priority to U.S. Provisional Application Serial No. 62/188,129, filed July 2, 2015, the entire contents of which are hereby incorporated by reference.

TECHNICAL FIELD

[0002] Embodiments of the present disclosure generally relate to propylene production, and more specifically relate to converting a stream comprising butene to propylene via metathesis using an amorphous mesoporous silica foam metathesis catalyst.

BACKGROUND

[0003] In recent years, there has been a dramatic increase in the demand for propylene to feed the growing markets for polypropylene, propylene oxide and acrylic acid. Currently, most of the propylene produced worldwide (74 million tons/year) is a by-product from steam cracking units (57%) which primarily produce ethylene, or a by-product from Fluid Catalytic Cracking (FCC) units (30%) which primarily produce gasoline. These processes cannot respond adequately to a rapid increase in propylene demand. However, the processing of lower cost butenes co-produced by these processes gives the refiner or the petrochemical producer an opportunity to add value depending on downstream integration and relative economics. The so-called on-purpose propylene processes such as propane dehydrogenation (PDH), metathesis of ethylene and butenes, high severity FCC, olefins cracking and methanol to olefins (MTO) contributes about 12% of total propylene production. However, propylene demand growth has exceeded ethylene and gasoline/distillate demand growth, and propylene supply has not kept pace with this increase in demand.

[0004] Olefin metathesis is considered a useful reaction to shift the composition of a pool of low-value butenes to meet market demand for propylene. In 2010, propylene production via metathesis accounted for about 5% of global propylene supply. This segment has been the

most rapidly growing on-purpose propylene production route over the past 5 years. With new capacity coming on-stream in the Middle East and Asia, propylene production via metathesis is expected to increase.

[0005] Catalytic olefin metathesis is a useful chemical reaction that is able to transform simple and cheap organic molecules into complex and valuable molecules. In olefin metathesis, two olefin molecules exchange the groups around the double bonds in the presence of a catalyst. The olefins can be of different molecules by structure and composition, or two identical molecules. In general, reaction temperatures for olefin metathesis reactions can be as low as at room temperature or can be at temperatures up to about 500 °C or greater, depending on the type of starting materials, the catalyst used, and the media in which the reaction is carried out.

[0006] However, olefin metathesis catalysts often do not have the requisite selectivity to yield propylene and other products. Additionally, olefin metathesis catalysts are subject to deactivation due to coking from aromatic products.

SUMMARY

[0007] Accordingly, ongoing needs exist for catalyst preparation for selective production of propylene using butenes metathesis. Embodiments of the present disclosure are directed to the production of propylene via metathesis using a metathesis catalyst comprising an amorphous mesoporous silica foam impregnated with metal oxides.

[0008] According to one embodiment, a metathesis process for producing propylene is provided. The process comprises providing a metathesis catalyst comprising an amorphous mesoporous silica foam impregnated with metal oxides. The metathesis catalyst has a pore size distribution of at least 3 nanometers (nm) to about 40 nm and a total pore volume of at least 0.700 centimeters³/gram (cm³/g). The metathesis process further comprises producing a product stream comprising propylene by contacting a feed stream comprising butene with the metathesis catalyst.

[0009] According to another embodiment, a metathesis catalyst for producing propylene is provided. The metathesis catalyst comprises an amorphous mesoporous silica foam with

impregnated metal oxides. The metathesis catalyst has a pore size distribution of at least 3 nm to about 40 nm and a total pore volume of at least 0.700 cm³/g. Total pore volume refers to total pore volume per unit weight.

[0010] Additional features and advantages of the present embodiments will be set forth in the detailed description which follows, and in part will be readily apparent to those skilled in the art from that description or recognized by practicing the described embodiments, including the detailed description which follows, the claims, as well as the appended drawings.

BRIEF DESCRIPTION OF THE DRAWINGS

[0011] FIG. 1 is a Transmission Electron Microscopy (TEM) image of an amorphous mesoporous silica foam impregnated with 10% by weight tungsten oxide (WO₃) in accordance with one or more embodiments of the present disclosure.

[0012] FIG. 2 is a TEM image of an SBA-15 silica support impregnated with WO₃.

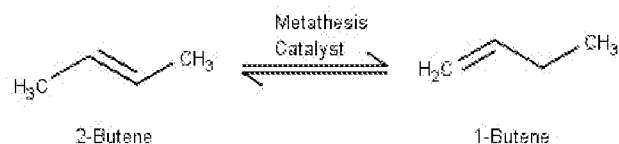
[0013] FIG. 3 is a TEM image of an MCM-41 support impregnated with WO₃.

[0014] FIG. 4 is a graph illustrating the isomerization activity and 2-butene conversion of the catalysts from FIGS. 1-3 in accordance with one or more embodiments of the present disclosure.

DETAILED DESCRIPTION

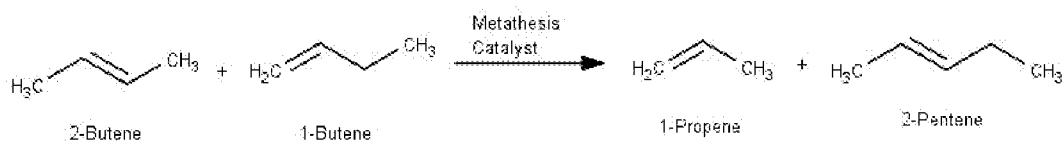
[0015] Embodiments of the present disclosure are directed to systems and methods for converting a butene stream to propylene via catalyzed metathesis. As used in the application, “metathesis” is generally a two-step process: 2-butene isomerization and then cross-metathesis using the metathesis catalyst as depicted as follows in Formulas 1 and 2.

[0016] Formula 1: 2-Butene Isomerization



[0017]

Formula 2: Cross-Metathesis



[0018]

[0019] As shown in Formulas 1 and 2, the “metathesis” reactions are not limited to these reactants and products; however, this is the basic illustration of the reaction methodology. As shown, metathesis reactions take place between two alkene molecules. The groups bonded to the carbon atoms of the double bond are exchanged between the molecules to produce two new alkenes with the swapped groups. The specific catalyst that is selected for the olefin metathesis reaction helps to determine whether a cis-isomer or trans-isomer is formed, as the coordination of the olefin molecules with the catalyst play an important role, as do the steric influences of the substituents on the double bond of the newly formed molecule.

[0020] Here, the metathesis catalyst may comprise amorphous mesoporous silica foam impregnated with metal oxides. As used in the application, “amorphous mesoporous silica foam” means a silica support with a non-ordered structure and a narrow pore size distribution. This non-ordered structure may be random and thus different than the disclosed hexagonal or cubic structures of conventional silica supports. Specifically, the amorphous mesoporous silica foam has a narrow pore size distribution of at least 3 nm to about 40 nm and a total pore volume of at least 0.700 cm³/g. Without being bound by theory, the present pore size distribution and pore volume are sized to achieve better catalytic activity and reduced blocking of pores by metal oxides, whereas smaller pore volume and pore size metathesis catalysts are susceptible to pore blocking and thereby reduced catalytic activity. Reduced blocking leads to higher dispersion of metal oxide species, such as WO₃, on the

amorphous mesoporous silica foam. Higher WO_3 dispersion leads to higher metathesis activity and thus higher propylene yield.

[0021] In one or more embodiments, the pore size distribution may range from at least 3 nm to about 40 nm, or from about 3 nm to about 20 nm, or from about 4 nm to about 10 nm, or from about 4 nm to about 8 nm, or from about 4 nm to about 6 nm. In further embodiments, the total pore volume may be from at least $0.700 \text{ cm}^3/\text{g}$ to about $2.5 \text{ cm}^3/\text{g}$, or from about $0.800 \text{ cm}^3/\text{g}$ to about $2.5 \text{ cm}^3/\text{g}$, or from about $0.800 \text{ cm}^3/\text{g}$ to about $1.5 \text{ cm}^3/\text{g}$, or from about $0.800 \text{ cm}^3/\text{g}$ to about $1.25 \text{ cm}^3/\text{g}$, or from about $0.800 \text{ cm}^3/\text{g}$ to about $1.0 \text{ cm}^3/\text{g}$, or from about $0.850 \text{ cm}^3/\text{g}$ to about $1.0 \text{ cm}^3/\text{g}$.

[0022] Moreover, the metathesis catalyst has a total acidity from about 0.125 millimole/gram (mmol/g) to about 0.500 mmol/g. Without being bound by theory, if the material exceeds 0.500 mmol/g, other detrimental side reactions may result, such as cracking and hydrogen transfer reactions. In further embodiments, the metathesis catalyst may have a total acidity from about 0.125 mmol/g to about 0.250 mmol/g, or from about 0.125 mmol/g to about 0.150 mmol/g. While various surface areas are contemplated, the metathesis catalyst may, in one or more embodiments, have a surface area of at least about $400 \text{ m}^2/\text{g}$ (m^2/g), or from about $400 \text{ m}^2/\text{g}$ about $800 \text{ m}^2/\text{g}$, or from about $400 \text{ m}^2/\text{g}$ to about $500 \text{ m}^2/\text{g}$, or from about $400 \text{ m}^2/\text{g}$ to about $450 \text{ m}^2/\text{g}$, or from about $425 \text{ m}^2/\text{g}$ to about $450 \text{ m}^2/\text{g}$.

[0023] The catalyst of the metathesis reaction is the impregnated metal oxide of the silica foam. The metal oxide may comprise one or oxides of a metal from the Periodic Table IUPAC Group Numbers 6-10. In one or more embodiments, the metal oxide may be an oxide of molybdenum, rhenium, tungsten, or combinations thereof. In a specific embodiment, the metal oxide is tungsten oxide (WO_3). It is contemplated that various amounts of metal oxide may be impregnated into the amorphous mesoporous silica foam. For example and not by way of limitation, the molar ratio of silica to metal oxide, for example, WO_3 , is about 1 to about 50, or about 1 to about 40, or about 5 to about 30, or about 5 to about 15. Moreover, the metathesis catalyst may include from about 1 to about 50% by weight, or from about 2 to about 25% by weight, or from about 5 to about 15% by weight metal oxide, for example, WO_3 .

[0024] Additionally, other optional components may be included into the impregnated mesoporous silica foam catalyst. For example, the metathesis catalyst may include a structuring agent. In one embodiment, the structuring agent is a tri-block copolymer structuring agent. In a further embodiment, the tri-block copolymer structuring agent is a poly(ethylene glycol)-block-poly(propylene glycol)-block-poly(ethylene glycol) structure, which may be also called a poloxamer structure. One suitable commercial embodiment of the surfactant tri-block copolymer structuring agent is Pluronic® P123 by BASF Corporation.

[0025] In operation, a product stream comprising propylene is produced from a stream comprising butene via metathesis conversion by contacting the metathesis catalyst. The butene stream comprises 2-butene, and optionally comprises one or more isomers, such as 1-butene, trans-2-butene, and cis-2-butene. The present discussion centers on butene based feed streams; however, it is known that other C₁-C₆ components may also be present in the feed stream. As shown previously, metathesis involves isomerization of 2-butene to 1-butene followed by cross-metathesis of the 2-butene and 1-butene into a metathesis product stream comprising propylene, and other alkenes/alkanes such as pentene.

[0026] Various systems which incorporate the metathesis catalyst are contemplated. For details regarding such systems, co-pending Saudi Aramco U.S. Application No. 62/188,052 entitled *Systems and Methods of Producing Propylene* is incorporated by reference in its entirety.

[0027] Various operating conditions are contemplated for the contacting of the butene stream with the catalyst. For example, the butene stream may contact the metathesis catalyst at a space hour velocity of about 10 to about 10,000 hr⁻¹, or about 300 to about 1200 hr⁻¹. Moreover, the butene stream may contact the metathesis catalyst at a temperature of about 200 to about 600 °C, or about 300 to about 600 °C. Furthermore, the butene stream may contact the metathesis catalyst at a pressure of about 1 to about 30 bars, or about 1 to about 10 bars. Optionally, the metathesis catalyst may be pretreated prior to metathesis. For example, the metathesis catalyst may be pretreated with N₂ for about 1 to about 5 hours before metathesis at a temperature of at least about 400 °C, or at least about 500 °C.

[0028] In one or more embodiments, the product stream may comprise at least about 35 mol.% propylene, and less than 1 mol.% aromatics. Without being bound by theory, it is desired that the aromatics yield be low as it causes coke formation, which can lead to catalyst deactivation. Without being bound by theory, the improved propylene yield and reduced side reactions is due in part to the high selectivity of the isomerization from 2-butene to 1-butene. For example, the present metathesis catalysts may yield at least 90% by weight, or at least 95% by weight conversion of 2-butene to 1-butene via isomerization. Moreover, by increasing the yield of 1-butene via isomerization, the subsequent cross-metathesis of 2-butene and 1-butene is able to yield more propylene in the product stream.

[0029] As will be demonstrated in the examples which follow, tungsten loaded on the amorphous mesoporous silica foam has higher 2-butene conversion and better propylene yield compared to tungsten loaded on ordered silica supports such as MCM-41 and SBA-15. Reiterating the prior statements, the selective isomerization of 2-butene to 1-butene and less hydrogen transfer reactions (less aromatics formation) are additional advantages of amorphous mesoporous silica foam as compared to MCM-41 and SBA-15. The formation of undesirable isobutylene is also reduced in amorphous mesoporous silica foam as compared to MCM-41 and SBA-15.

[0030] While various other metathesis catalysts are contemplated, the following method for producing a tungsten oxide impregnated mesoporous silica catalyst comprises preparing a precursor silica foam solution by mixing an aqueous triblock co-polymer surfactant solution with a sodium silicate solution, solidifying the precursor silica foam solution via a drying step, calcining the solid precursor silica foam, wet impregnating tungsten oxide into the solid precursor silica foam after calcining, and drying after wet impregnation to produce the tungsten oxide impregnated mesoporous silica catalyst. The aqueous triblock co-polymer surfactant solution includes an acid and a salt. The acid may comprise acetic acid and the salt may comprise ammonium fluoride.

[0031] Examples**[0032] Example 1: Preparation of WO₃/Silica foam (Catalyst-A)**

[0033] The mesoporous silica foam with ultra large pores was synthesized using the following method. In a typical synthesis, 3.0 grams (g) of neutral triblock co-polymer surfactant, Pluronic® P123, was dissolved in a mixture of acetic acid (3.0 g), deionized (DI) water (52 g), and ammonium fluoride (0.3 g) at 40 °C. After stirring for 2 hours, a solution of sodium silicate (2.35 g) in water (40 g) was added and the resultant mixture was reacted under vigorous stirring for 5 min. Then, the mixture was kept under static condition for 24 hours at 40 °C followed by aging at 70 °C overnight. The solid products were washed with DI water and collected by filtration and air dried. The obtained solid was then calcined at 560 °C for 6 hours to remove the template.

[0034] The silica foam was impregnated with 10% by weight of WO₃ following wet impregnation method. In a typical synthesis, 2 g of silica foam was suspended in 60 milliliters (ml) distilled water and 0.2112 g of ammonium metatungstate was added. The resulting mixture was stirred for 3 hours and dried overnight in the oven at 100 °C. The material was calcined at 550 °C for 5 hours.

[0035] Example 2: Preparation of WO₃/SBA-15 (Catalyst B)

[0036] SBA-15 material was synthesized using tri-block copolymer, poly(ethylene glycol)-block-poly(propylene glycol)-block-poly(ethylene glycol) as a structure directing agent. In a typical synthesis, 4 g of Pluronic P123 was added to 30 ml of water. After stirring for a few hours, a clear solution was obtained. About 70 g of 0.28 Molar (M) hydrochloric acid was added to it and the solution was stirred for another 2 hours. Then, 9 g of tetraethyl orthosilicate (TEOS) was added and the resulting mixture was stirred for 24 hours at 40 °C and finally heated to 100 °C for 48 hours. The solid product was recovered by filtration, washed with water for several times, and dried overnight at 100 °C. Finally, the product was calcined at 550 °C for 6 hours to remove the template.

[0037] The SBA-15 was impregnated with 10% by weight of WO₃ following a wet impregnation method. In a typical synthesis, 2 grams of silica foam was suspended in 60 ml

distilled water and 0.2112 g of ammonium metatungstate was added. The resulting mixture was stirred for 3 hours and dried overnight in the oven at 100 °C. The material was calcined at 550 °C for 5 hours.

[0038] *Example 3: Preparation of WO₃/MCM-41(Catalyst-C)*

[0039] The MCM-41 was synthesized following the procedure of Beck et al. [J.S. Beck, J.C. Vartuli, W.J. Roth, M.E. Leonowicz, C.T. Kresge, K.D. Schmitt, C.T.-U. Chu, D.H. Olsen, E.W. Sheppard, S.B. McCullen, J.B. Higgins, J.L. Schlenker, J. Am. Chem. Soc. 114, 10834 (1992)].

[0040] The MCM-41 was impregnated with 10% by weight of WO₃ following wet impregnation method. In a typical synthesis, 2 grams of silica foam was suspended in 60 ml distilled water and 0.2112 g of ammonium metatungstate was added. The resulting mixture was stirred for 3 hours and dried overnight in the oven at 100 °C. The material was calcined at 550 °C for 5 hours.

[0041] *Example 4: Catalyst Evaluation*

[0042] The prepared catalysts were tested for their activity and selectivity to butene metathesis reaction in a fixed bed continuous flow reactor (ID 0.25 in, Autoclave Engineers Ltd) at atmospheric pressure. A fixed amount of catalyst sample (2 ml) was packed in the reactor tube with silicon carbide on top and bottom of the reactor. The catalyst was pretreated under N₂ at 550 °C for 1 hour. All reactions were carried out at 550 °C, GHSV (gas hourly space velocity) of 900 h⁻¹ at atmospheric pressure using 2-butene (5 ml/min) as feed with nitrogen as diluent (25 ml/min). The quantitative analysis of the reaction products were carried out on- line using a Varian gas chromatograph with flame ionization detector (FID) (Varian 450-GC), equipped with CP-Al₂O₃/Na₂SO₄ capillary column (50 meters (m) length × 0.32 millimeters (mm) I.D. × film thickness (df) =5 micrometers (μm)).

[0043] The physico-chemical parameters observed for catalyst A-C are presented in Table 1. Catalyst A has higher pore size and pore volume compared to catalyst B and C. This shows that the impregnation of WO₃ does not block the pores of catalyst A. The total acidity of

catalyst A is also higher than catalyst B and C, indicating higher dispersion of tungsten oxides in the support.

[0044] Table 1

Catalysts	Surface Area (m ² /g)	Pore Size Distribution (nm)	Total Pore Volume (cm ³ /g)	Total Acidity (mmol./g)
10% WO ₃ /Silica Foam(A)	436	4.07	0.887	0.129
10% WO ₃ /MCM-41 (B)	419	2.1	0.453	0.118
10% WO ₃ /SBA-15 (C)	468	2.73	0.636	0.108

[0045] Transmission electron microscopy (TEM) images of the catalysts were performed on a JEOL JEM 3010 electron microscope operated at 200 kilovolts (kV). The catalysts were dispersed in methanol by sonication. Few drops of the dispersion were dropped onto a carbon-coated copper grid followed by solvent evaporation in air at room temperature. FIGS. 1-3 present TEM images of the three samples. FIG. 1 illustrates that 10W/Silica Foam exhibits that tungsten is well dispersed as compared with 10W/SBA-15 as depicted in FIG. 2 and 10W/MCM-41 as depicted in FIG. 3. The agglomeration of tungsten species was lower for 10W/Silica Foam than 10W/SBA-15 and 10W/MCM-41. This shows that the 10W/Silica Foam has more active species available for metathesis reaction compared to other catalysts.

[0046] *Example 5: Isomerization Performance*

[0047] The graph of FIG. 4 illustrates the isomerization activity and 2-butene conversion of the catalyst from FIGS. 1-3 at a GHSV (gas hourly space velocity) of 900 h⁻¹ at atmospheric pressure using 2-butene (5 ml/min) as feed with nitrogen as diluent (25 ml/min). As shown, while the 2-butene conversion (depicted on left side Y-axis) yielded similar results for the catalysts A-C; however, 2-butene isomerization to 1-butene greatly differs among catalyst A-C. Specifically, the silica foam catalyst A has a selectivity value of 95% by weight at 450 C, whereas the other catalysts B and C are at 90% or less. As stated, a higher isomerization selectivity increases the amount of propylene yielded in the cross-metathesis

step. Also, a higher isomerization selectivity also reduces side reactions which can cause coking and catalyst deactivation.

[0048] *Example 6: Metathesis Performance*

[0049] Table 2 shows the metathesis activity of catalysts A-C at 550 °C (GHSV 900 h⁻¹). Catalytic performances of catalysts A-C in the metathesis reaction of 2-butene (Reaction temperature: 550 °C, atmospheric pressure, GHSV 900 h⁻¹). Catalyst A prepared with silica foam as support shows higher 2-butene conversion and better propylene yield as compared to catalysts B and C. Higher yield of catalyst A relates to selective isomerization of 2-butene to 1-butene, which undergo further cross-metathesis reaction to produce propylene. Catalyst A has a large pore diameter, which provides higher dispersion of WO₃ species leading to high metathesis activity. The acidity of the material plays an important role in the isomerization of 2-butene to 1-butene, which enhances the activity of metathesis reaction between 2-butene and 1-butene. The yield of aromatics and higher compounds were found to be less for catalyst A compared to other catalysts thereby showing higher metathesis activity of the catalyst A.

[0050] Table 2

Yield (Mol. %)	10W/Silica Foam (A)	10W/SBA-15 (B)	10W/MCM-41 (C)
C ₂ =	10.67	8.35	8.79
C ₃ =	35.95	33.29	31.84
1-C ₄ =	8.41	10.57	9.38
2-C ₄ =	17.47	23.28	19.78
i-C ₄ =	5.78	2.02	8.48
C ₅ =	20.79	21.33	19.19
C ₆ +	0.93	1.16	2.54

[0051] CALCULATION METHODOLOGIES

[0052] The surface area of the samples was measured by nitrogen adsorption at 77 K using AUTOSORB-1 (Quanta Chrome). Before adsorption measurements, samples (ca. 0.1 g) were heated at 220 °C for 2 h under nitrogen flow. The nitrogen adsorption isotherms of catalysts were measured at liquid nitrogen temperature (77 K). The surface areas and pore

size distributions were calculated by the Brunauer Emmett-Teller (BET) method and the Barrett-Joyner-Halenda (BJH) method, respectively. The total pore volume was estimated from the amount of N₂ adsorbed at P/P₀ = 0.99. Barret EP, Joyner LJ, Halenda PH, J. Am. Chem. Soc. 73 (1951) 373-380.

[0053] The zeolite samples were characterized by X-ray powder diffraction (XRD) with a Rigaku Mini-flex II system using nickel filtered CuK α radiation ($\lambda = 1.5406 \text{ \AA}$, 30 kV and 15 mA). The XRD patterns were recorded in static scanning mode from 1.2 - 50° (2 θ) at a detector angular speed of 2° min⁻¹ with a step size of 0.02°.

[0054] It should now be understood that various aspects of the methods of making propylene with the metathesis catalysts are described in the application and that such aspects may be utilized in conjunction with various other aspects.

[0055] In a first aspect, the disclosure provides a metathesis process for producing propylene comprising providing a metathesis catalyst comprising an amorphous mesoporous silica foam impregnated with metal oxides, where the metathesis catalyst has a pore size distribution of at least 3 nm to about 40 nm and a total pore volume of at least 0.700 cm³/g. The process further comprises producing a product stream comprising propylene by contacting a feed stream comprising butene with the metathesis catalyst.

[0056] In a second aspect, the disclosure provides a process of the first aspect, in which the process further comprises tri-block copolymer structuring agent, where the tri-block copolymer structuring agent is poly(ethylene glycol)-block-poly(propylene glycol)-block-poly(ethylene glycol) structure.

[0057] In a third aspect, the disclosure provides a process of either the first or second aspects, in which the metal oxides are impregnated via wet impregnation.

[0058] In a fourth aspect, the disclosure provides a process of any one of the first through third aspects, in which the feed stream comprises 2-butene, and optionally one or more of 1-butene, trans-2-butene, and cis-2-butene.

[0059] In a fifth aspect, the disclosure provides a process of any one of the first through fourth aspects, in which the metathesis catalyst catalyzes isomerization of 2-butene to 1-butene followed by cross-metathesis of 2-butene and 1-butene into a metathesis product stream comprising propylene, where at least 90% of the 2-butene is converted to 1-butene via isomerization.

[0060] In a sixth aspect, the disclosure provides a process of any one of the first through fifth aspects, in which the pore size distribution is from at least 4 nm to about 10 nm.

[0061] In a seventh aspect, the disclosure provides a process of any one of the first through sixth aspects, in which the total pore volume is from at least $0.800 \text{ cm}^3/\text{g}$ to about $1.5 \text{ cm}^3/\text{g}$.

[0062] In an eighth aspect, the disclosure provides a process of any one of the first through seventh aspects, in which the metathesis catalyst has a total acidity from about 0.125 mmol/g to about 0.500 mmol/g .

[0063] In a ninth aspect, the disclosure provides a process of any one of the first through eighth aspects, in which the metal oxide comprises one or oxides of a metal from the Periodic Table IUPAC group numbers 6-10.

[0064] In a tenth aspect, the disclosure provides a process of any one of the first through ninth aspects, in which the metal oxide is an oxide of molybdenum, rhenium, tungsten, or combinations thereof.

[0065] In an eleventh aspect, the disclosure provides a process of any one of the first through tenth aspects, in which the metal oxide is tungsten oxide.

[0066] In a twelfth aspect, the disclosure provides a process of any one of the first through eleventh aspects, in which the metathesis catalyst comprises from about 5 to about 15% by weight metal oxide.

[0067] In a thirteenth aspect, the disclosure provides a process of any one of the first through twelfth aspects, in which the metathesis catalyst has a surface area of about 400 to about $500 \text{ m}^2/\text{g}$.

[0068] In a fourteenth aspect, the disclosure provides a process of any one of the first through thirteenth aspects, further comprising pre-treating the metathesis catalyst with N₂ about 1 to about 5 hours before metathesis at a temperature of at least about 500 °C.

[0069] In a fifteenth aspect, the disclosure provides a process of any one of the first through fourteenth aspects, in which the contact between the butene and the metathesis catalyst occurs at a space hour velocity of about 10 to about 10,000 hr⁻¹, a temperature of about 300 to about 600 °C, and a pressure of about 1 to about 10 bars.

[0070] In a sixteenth aspect, the disclosure provides a process of any one of the first through fifteenth aspects, in which the product stream comprises at least about 35 mol.% propylene.

[0071] In a seventeenth aspect, the disclosure provides a process of any one of the first through sixteenth aspects, in which the product stream comprises at least about 10 mol.% ethylene.

[0072] In an eighteenth aspect, the disclosure provides a process of any one of the first through seventeenth aspects, in which the product stream comprises less than 1 mol.% aromatics.

[0073] In a nineteenth aspect, the disclosure provides a metathesis catalyst for producing propylene suitable for use in any of the processes of the first through eighteenth aspects, where metathesis catalyst comprises an amorphous mesoporous silica foam with impregnated metal oxides, where the metathesis catalyst has a pore size distribution of at least 3 nm to about 40 nm and a total pore volume of at least 0.700 cm³/g.

[0074] In a twentieth aspect, the disclosure provides a metathesis catalyst of the nineteenth aspect, further comprising a tri-block copolymer structuring agent, where the tri-block copolymer structuring agent is poly(ethylene glycol)-block-poly(propylene glycol)-block-poly(ethylene glycol) structure.

[0075] In a twenty-first aspect, the disclosure provides a metathesis catalyst of the nineteenth or twentieth aspects, in which the pore size distribution is from at least 4 nm to about 10 nm.

[0076] In a twenty-second aspect, the disclosure provides a metathesis catalyst of any one of the nineteenth through twenty-first aspects, in which the total pore volume is from at least $0.800 \text{ cm}^3/\text{g}$ to about $1.5 \text{ cm}^3/\text{g}$.

[0077] In a twenty-third aspect, the disclosure provides a metathesis catalyst of any one of the nineteenth through twenty-second aspects, in which the metathesis catalyst has a total acidity from about 0.125 mmol/g to about 0.500 mmol/g .

[0078] In a twenty-fourth aspect, the disclosure provides a metathesis catalyst of any one of the nineteenth through twenty-third aspects, in which where the metal oxide is an oxide of molybdenum, rhenium, tungsten, or combinations thereof.

[0079] In a twenty-fifth aspect, the disclosure provides a metathesis catalyst of any one of the nineteenth through twenty-fourth aspects, in which the metal oxide is tungsten oxide.

[0080] In a twenty-sixth aspect, the disclosure provides a metathesis catalyst of any one of the nineteenth through twenty-fifth aspects, in which the amorphous mesoporous silica foam has a molar ratio for silica to tungsten oxide of about 1 to about 50.

[0081] In a twenty-seventh aspect, the disclosure provides a metathesis catalyst of any one of the nineteenth through twenty-sixth aspects, in which the metathesis catalyst comprises from about 5 to about 15% by weight metal oxide.

[0082] In a twenty-eighth aspect, the disclosure provides a metathesis catalyst of any one of the nineteenth through twenty-seventh aspects, in which the metathesis catalyst has a pore size ranging from about 2 nm to about 4 nm.

[0083] In a twenty-ninth aspect, the disclosure provides a metathesis catalyst of any one of the nineteenth through twenty-eighth aspects, in which metathesis catalyst has a surface area of about 400 to about $500 \text{ m}^2/\text{g}$.

[0084] It should be apparent to those skilled in the art that various modifications and variations can be made to the described embodiments without departing from the spirit and scope of the claimed subject matter. Thus it is intended that the specification cover the modifications and variations of the various described embodiments provided that such modification and variations come within the scope of the appended claims and their equivalents.

CLAIMS

1. A metathesis process for producing propylene comprising:
providing a metathesis catalyst comprising an amorphous mesoporous silica
5 foam impregnated with metal oxides, where the metathesis catalyst has a pore size
distribution of at least 3 nm to 40 nm and a total pore volume of at least 0.700 cm³/g;
and
producing a product stream comprising propylene by contacting a feed stream
comprising butene with the metathesis catalyst, where the metathesis catalyst has a total
10 acidity from 0.125mmol/g to 0.500 mmol/g, and a surface area of 400 to 500 m²/g.
2. The process of claim 1 further comprising tri-block copolymer
structuring agent, where the tri-block copolymer structuring agent is poly(ethylene
glycol)-block-poly(propylene glycol)-block-poly(ethylene glycol) structure.
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3. The process of claim 2 where the metathesis catalyst catalyzes
isomerization of 2-butene to 1-butene followed by cross-metathesis of 2-butene and 1-
butene into a metathesis product stream comprising propylene.
- 20 4. The process of claim 3 where at least 90% of the 2-butene is converted to
1-butene via isomerization.
5. The process of claim 1 where the pore size distribution is from at least 4
nm to 10 nm and the total pore volume is from at least 0.800 cm³/g to 1.5 cm³/g.
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6. The process of claim 1 where the metal oxide is an oxide of molybdenum,
rhenium, tungsten, or combinations thereof.
7. The process of claim 1 where the metal oxide is tungsten oxide.
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8. The process of claim 7 where the metathesis catalyst has a molar ratio for
silica to tungsten oxide of 1 to 50.

9. The process of claim 7 where the metathesis catalyst comprises from 5 to 15% by weight tungsten oxide.

5 10. A metathesis catalyst for producing propylene comprising
an amorphous mesoporous silica foam with impregnated metal oxides, where the
metathesis catalyst has a pore size distribution of at least 3 nm, to 40 nm, a total pore
volume of at least 0.700 cm³/g, a total acidity from 0.125 mmol/g to 0.500 mmol/g, and
a surface area of 400 to 500 m²/g.

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11. The metathesis catalyst of claim 10 further comprising a tri-block
copolymer structuring agent, where the tri-block copolymer structuring agent is
poly(ethylene glycol)-block-poly(propylene glycol)-block-poly(ethylene glycol)
structure.

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12. The metathesis catalyst of claim 10 where the pore size distribution is
from at least 4 nm to 10 nm, and the total pore volume is from at least 0.800 cm³/g to
1.5 cm³/g.

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13. The metathesis catalyst of claim 10 where the metal oxide is an oxide of
molybdenum, rhenium, tungsten, or combinations thereof.

14. The metathesis catalyst of claim 10 where the metal oxide is tungsten
oxide.

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15. The metathesis catalyst of claim 14 where the metathesis catalyst has a
molar ratio for silica to tungsten oxide of 1 to 50.

16. The metathesis catalyst of claim 10 where the metathesis catalyst
30 comprises from 5 to 15% by weight metal oxide.

17. The metathesis catalyst of claim 10 where the metathesis catalyst has a pore size ranging from 2 nm to 4 nm.