

Description

The present invention relates to a method of manufacturing a picture tube shadow mask according to the preamble of Claim 1. Such a method is described in FR—A—2532108.

5 A high-purity, low carbon steel plate of rimmed steel or aluminum killed steel has been used for a color picture tube shadow mask. The use of this material was determined in consideration of material supply capacity, manufacturing cost, machining properties, and mechanical strength. However, such a conventional material has a large thermal expansion coefficient (about $12 \times 10^{-6}/^{\circ}\text{C}$ for 0 to 100°C). The electron beam transmittance of a conventional shadow mask is about 15 to 20%, and many electron beams
10 impinge thereupon, so that the shadow mask itself is heated to a temperature of 30 to 80°C . As a result, the shadow mask is thermally deformed, changing the radius of curvature thereof with respect to a phosphor screen, thereby degrading color purity. Such degradation is called a purity drift (PD). In a conventional colour picture tube having a large mask aperture pitch, a wide margin (to be referred to as a guard band quantity hereinafter) for a positional error between the phosphor screen and the electron beam is
15 guaranteed. Even if the shadow mask is thermally deformed to some extent, degradation of color impurity tends not to occur. However, in a high resolution color picture tube used in a character and graphic display unit or a general commercial picture tube having a flat faceplate and a small pitch compatible with character broadcast, the above-mentioned margin is not always sufficient. More specifically, in a high resolution color picture tube, since the aperture pitch is very small, the aperture size itself is also small ($140 \mu\text{m}$ at a pitch of 0.3 mm, or $85 \mu\text{m}$ at a pitch of 0.2 mm). The guard band quantity is inevitably small. In
20 addition, in order to obtain such a small aperture size by photo-etching, the mask plate must have a small thickness, thereby decreasing the heat capacity. As compared with a thick plate, the thermal expansion quantity of such a thin mask plate is increased under identical conditions, thereby degrading the color purity.

25 In the flat faceplate color picture tube, the radius of the curvature of the mask is larger than that of a normal color picture tube. Even if the mask of the flat tube is subjected to the same thermal expansion influence as in the normal tube, the electron beams passing through the mask apertures are greatly deviated from the target positions on the phosphor screen. In addition to this disadvantage, since the pitch is small, the guard band quantity is small, and the colour purity tends to be degraded. In order to resolve the above problems, various methods have been proposed. For example, in Japanese Patent Publication No. 42—25446 and Japanese Patent Disclosure Nos. 50—58977 and 50—68650, an iron-nickel alloy having a
30 small thermal expansion coefficient, e.g., a 36 % Ni—Fe Invar alloy (having a thermal expansion coefficient of about $2.0 \times 10^{-6}/^{\circ}\text{C}$ for 0 to 100°C) or a 42% Ni—Fe alloy (having a thermal expansion coefficient of about $5.0 \times 10^{-6}/^{\circ}\text{C}$ for 0 to 100°C) is used as a material for a shadow mask. However, such a material cannot satisfy practical application conditions. This is partially because the Invar material of Fe—Ni alloy has poor
35 etching and molding properties as compared with those of the conventional low-carbon steel plate. With respect to the etching method, various proposals have been made as exemplified in Japanese Patent Publication No. 59—32859 and (EP—A—0 104 453). The shadow mask must have a surface curved with high precision. Tolerance for the radius R of curvature of 1000 mm is as strict as $\pm 5 \text{ mm}$. However, as
40 compared with the iron-based alloy, the Fe—Ni alloy has a high mechanical strength and a poor spherical formability by pressing or the like even after annealing under the same conditions. For example, as shown in Fig. 1, when an Fe—Ni mask having a thickness of 0.2 mm is formed spherically and a local recess is formed with respect to a standard radius R of curvature, a depth d of the recess which is not more than $20 \mu\text{m}$ substantially satisfies the tolerance requirement for color purity. Fig. 2 is a graph showing the recess
45 depth as a function of the yield strength in a 14 inch type shadow mask. As apparent from Fig. 2, the yield strength must be less than 196.13 N/mm^2 (20 kg/mm^2) so as to limit the depth to $20 \mu\text{m}$ or less. When a shadow mask of an Fe—Ni alloy material is annealed in an annealing furnace in a hydrogen atmosphere provided for the conventional shadow mask of aluminum killed steel material, the yield strength (a curve b) of the Fe—Ni alloy is higher than the yield strength (a curve a) of the aluminum killed low-carbon steel, as
50 shown in Fig. 3. The yield strength of the Fe—Ni alloy is decreased only to 284.4 to 294.2 N/mm^2 (29 to 30 kg/mm^2) even if it is annealed at a high temperature of 900°C . Referring to Fig. 2, the yield strength of the Fe—Ni alloy does not show a yield phenomenon inherent to carbon steel and is represented by the tensile strength when the Fe—Ni alloy is elongated by 0.2%. In this manner, the effective peripheral portion of the shadow mask of the Fe—Ni alloy material is particularly subject to deformation and recessing, thereby
55 presenting the problem of degradation in color purity due to deformation. While in order to improve anticorrosion and heat radiation properties of the shadow mask to be incorporated in a tube, a desired curved surface is obtained by pressing and a darkened oxide layer (to be referred to as a darkened layer hereinafter) is formed on the surface of the shadow mask. Although it has been assumed that the darkened layer need not be formed on the Fe—Ni shadow mask due to the presence of Ni having good anticorrosion properties, a typical difference between the electron beam mobility (i.e., PD quantity) of the Fe—Ni alloy
60 without the darkened layer caused by thermal deformation thereof and that of the aluminum killed low carbon steel cannot be observed even if the Fe—Ni has a small thermal expansion coefficient. This is because heat radiation is degraded since the darkened layer is not formed on the shadow mask, and a thermal conductivity of the Fe—Ni alloy is lower than that of the aluminum killed low-carbon steel. Thus, under identical operating conditions, the Fe—Ni shadow mask has a higher temperature than that of the
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low-carbon steel shadow mask. Therefore, unless the darkened layer having good heat radiation is formed on the shadow mask, the low thermal expansion of the Fe—Ni material cannot be effectively utilized, resulting in degradation of color purity caused by thermal deformation. However, it is very difficult to form a dense, darkened layer on the Fe—Ni alloy with good adhesion by means of a conventional method since Fe—Ni alloy has good anticorrosion properties. The darkened layer tends to be nonuniform due to impurities contained in the Fe—Ni alloy or surface contamination of the Fe—Ni mask. Consequently, a red rust is partially formed on the surface of the Fe—Ni mask. Furthermore, during formation of the darkened layer, a stress acts on the coarse inner wall surface of each shadow mask aperture due to the difference between the thermal expansion coefficients of the darkened layer and the shadow mask material. In a worst case, the darkened layer peels from the surface of the Fe—Ni alloy. Rust increases in the area of red rust formation during subsequent heat treatment to vary the aperture sizes. As compared with the darkened layer, the red rust layer more easily peels from the Fe—Ni material. The peeled, darkened and red rust layers cause a decrease in breakdown voltage, resulting in a notable disadvantage to the color picture tube.

It is an object of the present invention to provide a method of manufacturing a shadow mask so as to easily perform the pressing of a shadow mask of an Fe—Ni alloy and to form a darkened layer having a uniform density and good anticorrosion and heat radiation properties, and to provide a color picture tube having good white uniformity (WU) quality and a small purity drift.

In order to achieve the above object of the present invention, there is provided a method of manufacturing a picture tube shadow mask, having at least the steps of forming a plurality of mask apertures in a thin metal plate having iron and nickel as major constituents, annealing the metal plate with the plurality of mask apertures, and forming a darkened oxide layer on the annealed metal plate, wherein a cooling after annealing is performed in a reducing atmosphere, and the darkened oxide layer is formed in a weakly oxidizing steam atmosphere during a first half period and in a strongly oxidizing steam atmosphere during a second half period.

According to the present invention, annealing of the shadow mask of an Fe—Ni alloy is performed before the shadow mask is pressed. After the shadow mask is annealed, it is cooled in a reducing atmosphere in a furnace, thereby decreasing the yield strength of the shadow mask. Under this condition, the Fe—Ni material is pressed to control the radius of curvature. At the same time, the surface of the shadow mask is prevented from being converted to stainless steel so as to obtain a surface which easily allows a growth of an oxide film. Thus, a darkened oxide film is formed in a weakly oxidizing atmosphere in a heating furnace during the first half period and in a strongly oxidizing atmosphere during the second half period. The darkened oxide film formed in this manner has a high density, good adhesion strength, sufficient darkness and a uniform thickness.

According to the present invention, annealing, if performed in a vacuum, is performed at a temperature of 1000°C or higher, preferably within the range of 1000 to 1200°C at a pressure of 0.1333 mbar (10^{-1} torr) or higher, preferably within the range of $1.333 \cdot 10^{-1}$ to $1.333 \cdot 10^{-5}$ mbar (10^{-1} to 10^{-5} torr). The shadow mask is then cooled in a reducing atmosphere, e.g., hydrogen gas to a temperature of about 500°C.

The darkened layer is formed in a steam atmosphere obtained by supplying steam to a furnace at a rate of 20 to 50 m³/hr per unit volume of the furnace at a temperature of 500 to 700° for 10 minutes or more as the first half period. During the second half period, i.e., for another 10 minutes or more, steam is supplied at a rate of 0 to 20 m³/hr per unit volume of the furnace at a temperature of 550 to 750°C. The oxidation effect can be gradually increased without providing the first and second half periods for weak and strong oxidizing effects.

The shadow mask is preferably kept in a deoxidizing atmosphere after the annealing step until the step of forming a darkened layer is initiated.

This invention can be more fully understood from the following detailed description when taken in conjunction with the accompanying drawings, in which:

Fig. 1 is a representation for explaining deformation of a shadow mask;

Fig. 2 is a graph showing the depth of the recess as a function of the yield strength of the shadow mask;

Figs. 3 and 4 are graphs each showing the yield strength as a function of the annealing temperature; and

Fig. 5 is a graph showing the yield strength as a function of the vacuum annealing temperature.

An embodiment will be described hereinafter wherein an Invar alloy is used as a shadow mask material having an Fe—Ni alloy as a major constituent.

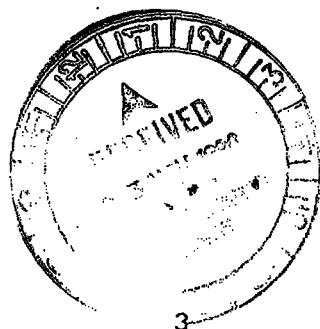


Table 1

Composition of shadow mask material (% by weight)

Composition Type	C	Mn	Si	P	S	Al	Ni(+Co)	Fe
Invar alloy	0.009	0.47	0.13	0.005	0.002	-	36.5	Bal
Al killed low-carbon steel	0.002	0.30	<0.01	0.016	0.009	0.052	-	Bal

Fig. 4 shows the yield strength of a shadow mask having a 36 Ni Invar alloy when the annealing temperature is increased while the shadow mask is placed in a hydrogen atmosphere (having a dew point of 10°C) in an annealing furnace. As is apparent from Fig. 4, even if the 36 Ni Invar alloy is annealed at a high temperature of 1200°C, the yield strength thereof is decreased only to 235,36 N/mm² (24 kg/mm²). In order to decrease the yield strength to a problem-free 196,13 N/mm² (20 kg/mm²) or less, the annealing temperature must fall within the range of 1500 to 1700°C from extrapolation with reference to Fig. 4. However, since the melting point of the Invar alloy is 1440 to 1450°C, such annealing cannot be performed.

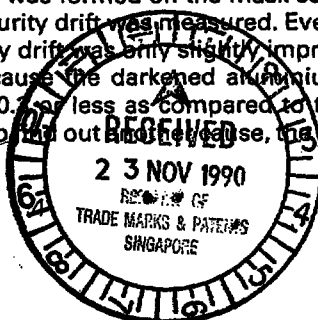
As described in EP-A-0101919, the present inventors examined the crystal structure of the annealed metal plate, and found that the inner crystal grains grew significantly upon an increase in the annealing temperature when hydrogen annealing was performed, but the crystal growth on the surface was very slight. The present inventors assumed that the insufficient surface crystal growth was associated with the yield strength and that the yield strength had to reach 196,13 N/mm² (20 kg/mm²) before the surface crystal grains would grow in the same manner as the inner crystal grains. On this account, according to the present inventors, a further assumption was made that Mn, P, S and the like having high vapor pressures among the impurities concentrated in the surface crystal interface would be evaporated to accelerate the surface crystal grain growth. Annealing was performed in a vacuum of 0,01333 mbar (10⁻² torr) at a temperature of 900 to 1200°C for 10 minutes. As shown in Fig. 5, a yield strength of 196,13 N/mm² (20 kg/mm²) or less could be obtained at a high annealing temperature of 1000°C or higher. In this case, the surface crystal growth did not differ from the inner crystal growth. As is apparent from Table 2 showing the analysis results of impurities in the surface layer having a thickness of 1/20 or less of the entire thickness, impurities such as Mn, P and S are greatly decreased.

Table 2

Composition before and after annealing (% by weight)

Annealing in vacuum	C	Mn	Si	P	S	Ni(+Co)	Fe
Before annealing	0.009	0.47	0.13	0.005	0.002	36.5	Bal
After annealing	0.007	0.052	0.12	<0.001	<0.001	36.3	Bal

The vacuum-annealed mask was pressed to obtain a smooth surface of high precision. The surface of the mask was then darkened in CO₂+O₂ gas, air and steam atmospheres at a temperature of 560°C for 15 minutes. Only a thin blackish-purple oxide film was formed on the mask surface. The resultant mask was incorporated in a colour picture tube, and the purity drift was measured. Even when the Invar material had a small thermal expansion coefficient, the purity drift was only slightly improved as compared with that of the aluminum-killed steel mask. This was because the darkened aluminum-killed steel mask had heat radiation of 0.6 and the darkened Invar mask 0.7 or less as compared to the completely darkened layer which is assumed to have a hot radiation of 1. To find out another cause, the mask temperature in operation



was measured by attaching a thermocouple to the mask surface. As a result, it was found that the temperature of the Invar mask was higher by 50 to 60°C than that of the aluminium-killed steel mask, thereby giving rise to a cause of a larger thermal deformation of the Invar mask. In order to prevent such a temperature rise, the same darkened oxide layer as in the aluminum-killed steel mask can be formed on the Invar mask to increase heat radiation. However, when the mask surface of the Invar mask after annealing was precisely examined by an ion microanalyzer (IMA), the presence of chromium was detected. It was found that the concentrated chromium was oxidized to form an inactive film having good anticorrosion properties, i.e., the surface was converted to stainless steel, and unlike the Fe—Ni, growth of the oxide film on the surface was largely prevented. The present inventors then assumed that the chromium oxide formed on the surface of the mask after annealing in a vacuum could be reduced and cooling could be performed in a reducing atmosphere to inhibit oxidation of the mask surface. After the mask was annealed at a temperature of 1100°C and a vacuum pressure of 0.01333 mbar (10^{-2} torr) for 10 minutes, the mask was cooled in the furnace while hydrogen was supplied thereto. When the mask was removed from the furnace upon cooling, it was covered with non-corrosion paper and was held in a case with a deoxidizer.

After pressing, press oil was removed from the mask by Trichrene steam cleaning, and a darkened film was formed under the same conditions as described above. The same darkened film as in the aluminum killed steel mask was obtained. Furthermore, the film obtained by steam darkening had excellent density and adhesion properties and degree of darkness. During steam darkening, when the layer is formed in a strong oxidizing atmosphere with a small amount of steam, rust dots are formed on the darkened layer which has poor adhesion properties. However, when the amount of steam is increased to obtain a weak oxidizing atmosphere, a dense layer with good adhesion strength can be obtained, but a long period of time is required to obtain a sufficient degree of darkness. Such a condition is not suitable for mass production. In order to resolve the problem, the present inventors assumed that a thin darkened layer having good adhesion properties could be formed in the weak oxidizing atmosphere during the first half period and then in the strong oxidizing atmosphere during the second half period. Based upon this assumption, the present inventors made various tests. According to the present inventors, darkening was performed at a temperature of 500 to 700°C for 10 minutes or more while steam was supplied at a rate of 20 to 50 m³/hr per unit cubic meter of the reaction chamber during the first half period. Darkening was then continued at a temperature of 550 to 750°C for 10 minutes or more while steam was supplied at a rate of 0 to 20 m³/hr per unit cubic meter of the chamber. As a result, the darkened layer had sufficient darkness and good adhesion properties. The adhesion properties of the darkened layer were evaluated such that a 90° bending test followed by a peeling test of the darkened layer by adhesion of cellophane tape to the bent portion was made. In addition, the darkened layer state was observed through a scanning electron microscope. The resultant layer was a dense film without cracks and pinholes. In forming the darkened layer, layer quality varies due to different structures of darkening furnaces even if identical conditions are established. Therefore, proper conditions must be selected for a specific darkening furnace so as to fall within the above-mentioned ranges. In addition, darkening need not be performed under different conditions during the first and second half periods. The oxidation effect can be changed in steps from a weak to a strong effect by changing the amount of steam and temperature.

In the above embodiment, the 36 Ni Invar alloy is used as the shadow mask material. However, the present invention is not limited to this material. For example, any Fe—Ni alloy containing super Invar, such as 42 Ni alloys and 32 Ni-5Co can be used. Furthermore, the darkening method of the present invention can also be used for an Fe—Ni alloy such as a mask frame and an inner shield which are incorporated in a color picture tube, in addition to a shadow mask.

The annealing process in the present invention may be carried out in an ordinary hydrogen atmosphere as disclosed in EP—A—0 124 354 by adopting molding strain-diminishing measures such as a hot-pressing or by making smooth the boundary portion between the central curved portion and outer fringe portion of the shadow mask, in which a molding strain is likely to be concentrated at the time of a press molding.

According to the present invention, the darkened oxide film has good pressing, anticorrosion and heat radiation properties. As a result, a shadow mask of an Fe—Ni alloy having good white uniformity quality and free from purity drift can be obtained.

Claims

1. A method of manufacturing a picture tube shadow mask, comprising the steps of forming a plurality of mask apertures in a thin metal plate, annealing the metal plate with the plurality of mask apertures, and after a reducing step performing an oxidation procedure in two steps in order to produce a dark oxide layer characterized in that the metal plate comprises iron and nickel as major constituents and that, after the annealing step at a temperature not lower than 1000°C, cooling is performed in a reducing atmosphere, and the darkened oxide layer is formed at first in a weak oxidizing steam atmosphere and then in a strong oxidizing steam atmosphere.

2. A method according to claim 1, characterized in that the annealing is performed at a vacuum pressure of not higher than 0.1333 mbar (10^{-1} torr), and hydrogen gas is continuously supplied to provide a reducing atmosphere until an atmosphere temperature becomes 500°C.

was measured by attaching a thermocouple to the mask surface. As a result, it was found that the temperature of the Invar mask was higher by 50 to 60°C than that of the aluminium-killed steel mask, thereby giving rise to a cause of a larger thermal deformation of the Invar mask. In order to prevent such a temperature rise, the same darkened oxide layer as in the aluminum-killed steel mask can be formed on the Invar mask to increase heat radiation. However, when the mask surface of the Invar mask after annealing was precisely examined by an ion microanalyzer (IMA), the presence of chromium was detected. It was found that the concentrated chromium was oxidized to form an inactive film having good anticorrosion properties, i.e., the surface was converted to stainless steel, and unlike the Fe—Ni, growth of the oxide film on the surface was largely prevented. The present inventors then assumed that the chromium oxide formed on the surface of the mask after annealing in a vacuum could be reduced and cooling could be performed in a reducing atmosphere to inhibit oxidation of the mask surface. After the mask was annealed at a temperature of 1100°C and a vacuum pressure of 0,01333 mbar (10^{-2} torr) for 10 minutes, the mask was cooled in the furnace while hydrogen was supplied thereto. When the mask was removed from the furnace upon cooling, it was covered with non-corrosion paper and was held in a case with a deoxidizer.

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The annealing process in the present invention may be carried out in an ordinary hydrogen atmosphere as disclosed in EP—A—0 124 354 by adopting molding strain-diminishing measures such as a hot-pressing or by making smooth the boundary portion between the central curved portion and outer fringe portion of the shadow mask, in which a molding strain is likely to be concentrated at the time of a press molding.

According to the present invention, the darkened oxide film has good pressing, anticorrosion and heat radiation properties. As a result, a shadow mask of an Fe—Ni alloy having good white uniformity quality and free from purity drift can be obtained.

Claims

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2. A method according to claim 1, characterized in that the annealing is performed at a vacuum pressure of not higher than 0,1333 mbar (10^{-1} torr), and hydrogen gas is continuously supplied to provide a reducing atmosphere until an atmosphere temperature becomes 500°C.



3. A method according to claim 1, characterized in that the darkened oxide layer is formed by supplying at first the steam at a rate of 20 to 50 m³/hr per unit volume of a treatment furnace at a temperature of 500 to 700°C for not less than 10 minutes and then at a rate of 0 to 20 m³/hr per unit volume of the treatment furnace at a temperature of 550 to 750°C for not less than 10 minutes.

4. A method according to claim 2, characterized in that the shadow mask is held in a deoxidizing atmosphere between said annealing step and said forming step.

5. A method according to claim 3, characterized in that the weak oxidizing steam atmosphere is changed in steps to be a strong oxidizing steam atmosphere.

10 Patentansprüche

1. Verfahren zur Herstellung einer Bildröhren-Schattenmaske durch Ausbilden einer Reihe von Maskenöffnungen in einer dünnen Metallplatte, Glühen der mit einer Reihe von Maskenöffnungen versehenen Metallplatte und — nach einer Reduktionsstufe — Aushführen einer zweistufigen Oxidationsbehandlung zur Ausbildung einer dunklen Oxidschicht, dadurch gekennzeichnet, daß die Metallplatte Eisen und Nickel als Hauptbestandteile enthält und daß nach der Glühstufe bei einer Temperatur nicht unter 1000°C in einer reduzierenden Atmosphäre gekühlt wird und daß die Dunkeloxidschicht zunächst in schwach oxidierender Dampfatmosfera und danach in stark oxidierender Dampfatmosfera erzeugt wird.

2. Verfahren nach Anspruch 1, dadurch gekennzeichnet, daß bei einem Vakuumdruck von nicht über 0,1333 mbar (10⁻¹ Torr) geglüht wird und zur Gewährleistung einer reduzierenden Atmosphäre kontinuierlich gasförmiger Wasserstoff zugeführt wird, bis eine Atmosphärentemperatur von 500°C erreicht ist.

3. Verfahren nach Anspruch 1, dadurch gekennzeichnet, daß die Dunkeloxidschicht gebildet wird, indem Dampf zunächst für nicht weniger als 10 min in einer Menge von 20—50 m³/h pro Einheitsvolumen eines Behandlungssofens bei einer Temperatur von 500—700°C und danach für nicht weniger als 10 min in einer Menge von 0—20 m³/h pro Einheitsvolumen des Behandlungssofens bei einer Temperatur von 550—750°C zugeführt wird.

4. Verfahren nach Anspruch 2, dadurch gekennzeichnet, daß die Schattenmaske zwischen der Glühstufe und der Formgebungsstufe in entoxidierender Atmosphäre gehalten wird.

5. Verfahren nach Anspruch 3, dadurch gekennzeichnet, daß die schwach oxidierende Dampfatmosfera stufenweise in eine stark oxidierende Dampfatmosfera geändert wird.

Revendications

1. Procédé de fabrication d'un masque de tube-image, consistant à former un ensemble de trous de masque une plaque métallique mince, à recuire la plaque métallique avec l'ensemble de trous de masque, et, après une opération de réduction, à exécuter une procédure d'oxydation en deux opérations afin de produire une couche d'oxyde foncée, caractérisé en ce que la plaque métallique comprend du fer et du nickel comme éléments constitutifs majeurs et en ce que, après l'opération de recuit à une température au moins égale à 1000°C, un refroidissement est effectué dans une atmosphère réductrice, et la couche d'oxyde foncée est formé d'abord dans une atmosphère de vapeur faiblement oxydante et ensuite dans une atmosphère de vapeur fortement oxydante.

2. Procédé selon la revendication 1, caractérisé en ce que le recuit est effectué à une pression de vide non supérieure à 0,1333 mbar (10⁻¹ torr), et un gaz d'hydrogène est fourni de façon continue pour obtenir une atmosphère réductrice jusqu'à ce que la température de l'atmosphère devienne égale à 500°C.

3. Procédé selon la revendication 1, caractérisé en ce que la couche d'oxyde foncée est formée en fournissant d'abord la vapeur à un débit compris entre 20 et 50 m³/heure par unité de volume d'un four de traitement à une température comprise entre 500 et 700°C pendant au moins 10 minutes et ensuite à un débit compris entre 0 et 20 m³/heure par unité de volume du four de traitement à une température comprise entre 550 et 750°C pendant au moins 10 minutes.

4. Procédé selon la revendication 3, caractérisé en ce que le masque est maintenu dans une atmosphère désoxydante entre l'opération de recuit et l'opération de formation.

5. Procédé selon la revendication 3, caractérisé en ce que l'atmosphère de vapeur faiblement oxydante est changée progressivement en atmosphère de vapeur fortement oxydante.

