NOVEL PYRAZOLOPYRIMIDINES AS CYCLIN DEPENDENT KINASE INHIBITORS

Field of the Invention

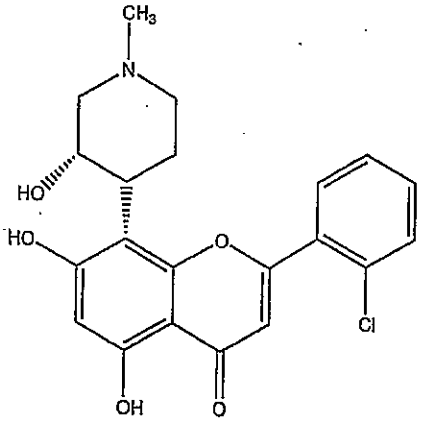
The present invention relates to pyrazolo[1,5-a]pyrimidine compounds useful as protein kinase inhibitors (such as for example, the inhibitors of the cyclin-dependent kinases, mitogen-activated protein kinase (MAPK/ERK), glycogen synthase kinase 3(GSK3beta) and the like), pharmaceutical compositions containing the compounds, and methods of treatment using the compounds and compositions to treat diseases such as, for example, cancer, inflammation, arthritis, viral diseases, neurodegenerative diseases such as Alzheimer's disease, cardiovascular diseases, and fungal diseases. This application claims benefit of priority from U.S. provisional patent applications, Serial No. 60/408,027 filed September 4, 2002, and Serial No. 60/421,959 filed October 29, 2002.

Background of the Invention

Protein kinase inhibitors include kinases such as, for example, the inhibitors of the cyclin-dependent kinases (CDKs), mitogen activated protein kinase (MAPK/ERK), glycogen synthase kinase 3 (GSK3beta), and the like. Protein kinase inhibitors are described, for example, by M. Hale *et* al in WO02/22610 A1 and by Y. Mettey *et* al in *J. Med. Chem.,* (2003) 46 222-236. The cyclin-dependent kinases are serine/threonine protein kinases, which are the driving force behind the cell cycle and cell proliferation. Individual CDK's, such as, CDK1, CDK2, CDK3, CDK4, CDK5, CDK6 and CDK7, CDK8, CDK9 and the like, perform distinct roles in cell cycle progression and can be classified as either

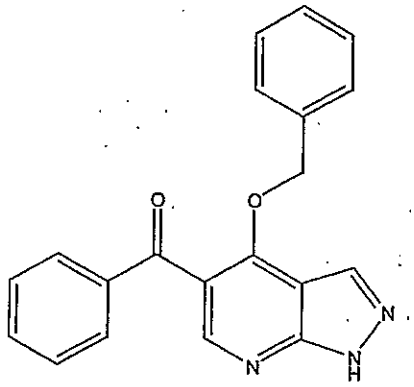
G1, S, or G2M phase enzymes. Uncontrolled proliferation is a hallmark of cancer cells, and misregulation of CDK function occurs with high frequency in many important solid tumors. CDK2 and CDK4 are of particular interest because their activities are frequently misregulated in a wide variety of human cancers. CDK2 activity is required for progression through G1 to the S phase of the cell cycle, and CDK2 is one of the key components of the G1 checkpoint. Checkpoints serve to maintain the proper sequence of cell cycle events and allow the cell to respond to insults or to proliferative signals, while the loss of proper checkpoint control in cancer cells contributes to tumorgenesis. The CDK2 pathway influences tumorgenesis at the level of tumor suppressor function (e.g. p52, RB, and p27) and oncogene activation (cyclin E). Many reports have demonstrated that both the coactivator, cyclin E, and the inhibitor, p27, of CDK2 are either over – or underexpressed, respectively, in breast, colon, nonsmall cell lung, gastric, prostate, bladder, non-Hodgkin's lymphoma, ovarian, and other cancers. Their altered expression has been shown to correlate with increased CDK2 activity levels and poor overall survival. This observation makes CDK2 and its regulatory pathways compelling targets for the development years, a number of adenosine 5'-triphosphate (ATP) competitive small organic molecules as well as peptides have been reported in the literature as CDK inhibitors for the potential treatment of cancers. U.S. 6,413,974, col. 1, line 23- col. 15, line 10 offers a good description of the various CDKs and their relationship to various types of cancer.

CDK inhibitors are known. For example, flavopiridol (Formula I) is a nonselective CDK inhibitor that is currently undergoing human clinical trials, A. M. Sanderowicz *et al, J. Clin. Oncol.* (1998) 16, 2986-2999.



Formula I

Other known inhibitors of the CDKs include, for example, olomoucine (J. Vesely *et al, Eur. J. Biochem.,* (1994) 224, 771-786) and roscovitine (l. Meijer *et al, Eur. J. Biochem.,* (1997) 243, 527-536). U.S. 6,107,305 describes certain pyrazolo[3,4-b] pyridine compounds as CDK inhibitors. An illustrative compound from the '305 patent has the Formula II:



Formula II

K. S. Kim *et al, J. Med. Chem.* 45 (2002) 3905-3927 and WO 02/10162 disclose certain aminothiazole compounds as CDK inhibitors.

Pyrazolopyrimidines are known. For Example, WO92/18504, WO02/50079, WO95/35298, WO02/40485, EP94304104.6, EP0628559 (equivalent to US Patents 5,602,136, 5,602,137 and 5,571,813), U.S. 6,383,790, *Chem. Pharm. Bull.,* (1999) 47 928, *J. Med. Chem.,* (1977) 20, 296, *J. Med. Chem.,* (1976) 19 517 and *Chem. Pharm. Bull., (1962)* 10 620 disclose various pyrazolopyrimidines. Other publications of interest are: WO 03/101993

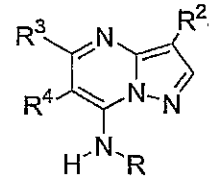
(published December 11, 2003), WO 03/091256 (published November 6, 2003), and DE 10223917 (published December 11, 2003).

There is a need for new compounds, formulations, treatments and therapies to treat diseases and disorders associated with CDKs. It is, therefore, an object of this invention to provide compounds useful in the treatment or prevention or amelioration of such diseases and disorders.

Summary of the Invention

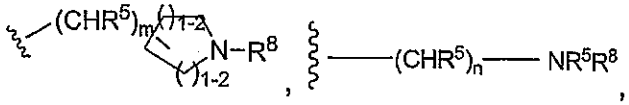
In its many embodiments, the present invention provides a novel class of pyrazolo[1,5-a]pyrimidine compounds as inhibitors of cyclin dependent kinases, methods of preparing such compounds, pharmaceutical compositions comprising one or more such compounds, methods of preparing pharmaceutical formulations comprising one or more such compounds, and methods of treatment, prevention, inhibition or amelioration of one or more diseases associated with the CDKs using such compounds or pharmaceutical compositions.

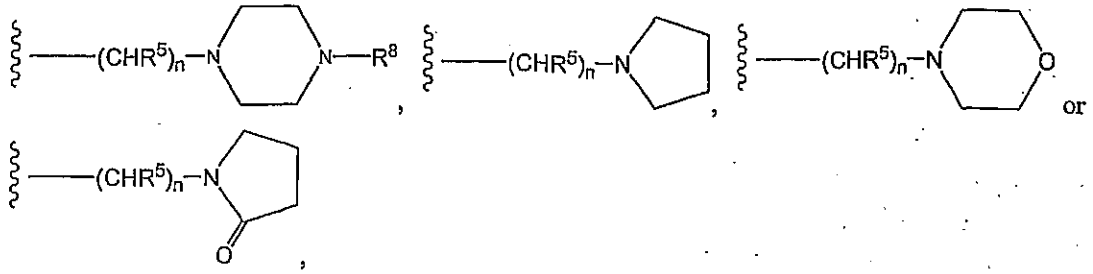
In one aspect, the present application discloses a compound, or pharmaceutically acceptable salts or solvates of said compound, said compound having the general structure shown in Formula III:



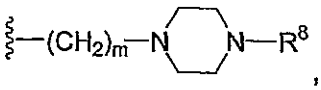
Formula III

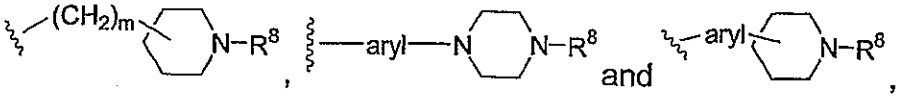
wherein:

R is H, alkyl, alkenyl, alkynyl, arylalkyl, arylalkenyl, cycloalkyl, cycloalkylalkyl, alkenylalkyl, alkynylalkyl, heterocyclyl, heterocyclylalkyl, heteroarylalkyl (including N-oxide of said heteroaryl), -$(CHR^5)_n$-aryl, -$(CHR^5)_n$-heteroaryl, 



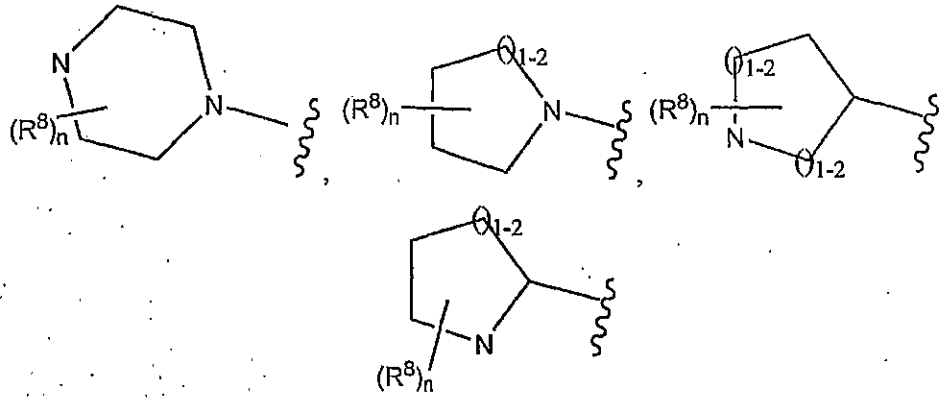
wherein each of said alkyl, alkenyl, alkynyl, aryl, cycloalkyl, heterocyclyl, and heteroaryl can be unsubstituted or optionally substituted with one or more moieties which can be the same or different, each moiety being independently selected from the group consisting of halogen, alkyl, aryl, cycloalkyl, heterocyclylalkyl, CF_3, OCF_3, CN, $-OR^5$, $-NR^5R^{10}$, $-C(R^4R^5)_p-R^9$, $-N(R^5)Boc$, $-(CR^4R^5)_pOR^5$, $-C(O_2)R^5$, $-C(O)R^5$, $-C(O)NR^5R^{10}$, $-SO_3H$, $-SR^{10}$, $-S(O_2)R^7$, $-S(O_2)NR^5R^{10}$, $-N(R^5)S(O_2)R^7$, $-N(R^5)C(O)R^7$ and $-N(R^5)C(O)NR^5R^{10}$;

R^2 is selected from the group consisting of H, R^9, alkyl, alkenyl, alkynyl, CF_3, heterocyclyl, heterocyclylalkyl, halogen, haloalkyl, aryl, arylalkyl, heteroarylalkyl, alkynylalkyl, cycloalkyl, heteroaryl, alkyl substituted with 1-6 R^9 groups which can be the same or different and are independently selected from the list of R^9 shown below, aryl substituted with 1-3 aryl or heteroaryl groups which can be the same or different and are independently selected from phenyl, pyridyl, thiophenyl, furanyl and thiazolo groups, aryl fused with an aryl or heteroaryl group, heteroaryl substituted with 1-3 aryl or heteroaryl groups which can be the same or different and are independently selected from phenyl, pyridyl, thiophenyl, furanyl and thiazolo groups, heteroaryl fused with an aryl or heteroaryl group,  ,



wherein one or more of the aryl and/or one or more of the heteroaryl in the above-noted definitions for R^2 can be unsubstituted or optionally substituted with one or more moieties which can be the same or different, each moiety being independently selected from the group consisting of halogen, -CN, $-OR^5$, $-SR^5$, $-S(O_2)R^6$, $-S(O_2)NR^5R^6$, $-NR^5R^6$, $-C(O)NR^5R^6$, CF_3, alkyl, aryl and OCF_3;

R^3 is selected from the group consisting of H, halogen, $-NR^5R^6$, $-OR^6$, $-SR^6$, $-C(O)N(R^5R^6)$, alkyl, alkynyl, cycloalkyl, aryl, arylalkyl, heterocyclyl, heterocyclylalkyl, heteroaryl and heteroarylalkyl,



wherein each of said alkyl, cycloalkyl, aryl, arylalkyl, heterocyclyl, heterocyclylalkyl, heteroaryl and heteroarylalkyl for R^3 and the heterocyclyl moieties whose structures are shown immediately above for R^3 can be unsubstituted or optionally independently substituted with one or more moieties which can be the same or different, each moiety being independently selected from the group consisting of halogen, alkyl, aryl, cycloalkyl, CF_3, CN, $-OCF_3$, $-(CR^4R^5)_pOR^5$, $-OR^5$, $-NR^5R^6$, $-(CR^4R^5)_pNR^5R^6$, $-C(O_2)R^5$, $-C(O)R^5$, $-C(O)NR^5R^6$, $-SR^6$, $-S(O_2)R^6$, $-S(O_2)NR^5R^6$, $-N(R^5)S(O_2)R^7$, $-N(R^5)C(O)R^7$, $-N(R^5)C(R^4R^5)_nN(R^5R^6)$ and $-N(R^5)C(O)NR^5R^6$, with the proviso that no carbon adjacent to a nitrogen atom on a heterocyclyl ring carries a $-OR^5$ moiety;

R^4 is H, halo or alkyl;

R^5 is H, alkyl, aryl, heteroaryl or cycloalkyl;

R^6 is selected from the group consisting of H, Boc, alkyl, alkenyl, aryl, arylalkyl, arylalkenyl, cycloalkyl, heterocyclyl, heterocyclylalkyl, heteroaryl, and heteroarylalkyl, wherein each of said alkyl, aryl, arylalkyl, cycloalkyl, heterocyclyl, heterocyclylalkyl, heteroaryl, and heteroarylalkyl can be unsubstituted or optionally substituted with one or more moieties which can be the same or different, each moiety being independently selected from the group consisting of halogen, alkyl, aryl, cycloalkyl, heterocyclylalkyl, CF_3, OCF_3, CN, $-OR^5$, $-NR^5R^{10}$, $-C(R^4R^5)_p-R^9$, $-N(R^5)Boc$, $-(CR^4R^5)_pOR^5$, $-C(O_2)R^5$, $-C(O)R^5$, $-C(O)NR^5R^{10}$, $-SO_3H$, $-SR^{10}$, $-S(O_2)R^7$, $-S(O_2)NR^5R^{10}$, $-N(R^5)S(O_2)R^7$, $-N(R^5)C(O)R^7$ and $-N(R^5)C(O)NR^5R^{10}$;

R^{10} is selected from the group consisting of H, alkyl, aryl, arylalkyl, cycloalkyl, heterocyclyl, heterocyclylalkyl, heteroaryl, and heteroarylalkyl, wherein each of said alkyl, aryl, arylalkyl, cycloalkyl, heterocyclyl, heterocyclylalkyl, heteroaryl, and heteroarylalkyl can be unsubstituted or optionally substituted with one or more moieties which can be the same or different, each moiety being independently selected from the group consisting of halogen, alkyl, aryl, cycloalkyl, heterocyclylalkyl, CF_3, OCF_3, CN, $-OR^5$, $-NR^4R^5$, $-C(R^4R^5)_p-R^9$, $-N(R^5)Boc$, $-(CR^4R^5)_pOR^5$, $-C(O_2)R^5$, $-C(O)NR^4R^5$, $-C(O)R^5$, $-SO_3H$, $-SR^5$, $-S(O_2)R^7$, $-S(O_2)NR^4R^5$, $-N(R^5)S(O_2)R^7$, $-N(R^5)C(O)R^7$ and $-N(R^5)C(O)NR^4R^5$;

or optionally (i) R^5 and R^{10} in the moiety $-NR^5R^{10}$, or (ii) R^5 and R^6 in the moiety $-NR^5R^6$, may be joined together to form a cycloalkyl or heterocyclyl moiety, with each of said cycloalkyl or heterocyclyl moiety being unsubstituted or optionally independently being substituted with one or more R^9 groups;

R^7 is selected from the group consisting of alkyl, cycloalkyl, aryl, arylalkenyl, heteroaryl, arylalkyl, heteroarylalkyl, heteroarylalkenyl, and heterocyclyl, wherein each of said alkyl, cycloalkyl, heteroarylalkyl, aryl, heteroaryl and arylalkyl can be unsubstituted or optionally independently substituted with one or more moieties which can be the same or different, each moiety being independently selected from the group consisting of halogen, alkyl, aryl, cycloalkyl, CF_3, OCF_3, CN, $-OR^5$, $-NR^5R^{10}$, $-CH_2OR^5$, $-C(O_2)R^5$, $-C(O)NR^5R^{10}$, $-C(O)R^5$, $-SR^{10}$, $-S(O_2)R^{10}$, $-S(O_2)NR^5R^{10}$, $-N(R^5)S(O_2)R^{10}$, $-N(R^5)C(O)R^{10}$ and $-N(R^5)C(O)NR^5R^{10}$;

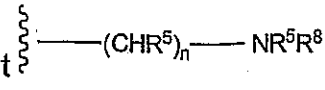
R^8 is selected from the group consisting of R^6, $-OR^6$, $-C(O)NR^5R^{10}$, $-S(O_2)NR^5R^{10}$, $-C(O)R^7$, $-C(=N-CN)-NH_2$, $-C(=NH)-NHR^5$, heterocyclyl, and $-S(O_2)R^7$;

R^9 is selected from the group consisting of halogen, $-CN$, $-NR^5R^{10}$, $-SCN$, $-NO_2$, $-C(O)R^5$, $-C(O_2)R^6$, $-C(O)NR^5R^{10}$, $-OR^6$, $-SR^6$, $-S(O_2)R^7$, $-S(O_2)NR^5R^{10}$, $-N(R^5)S(O_2)R^7$, $-N(R^5)C(O)R^7$and $-N(R^5)C(O)NR^5R^{10}$;

m is 0 to 4;

n is 1 to 4; and

p is 1 to 4,

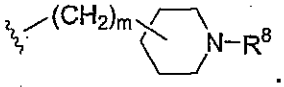
with the proviso that when R^2 is phenyl, R^3 is not alkyl, alkynyl or halogen, and that when R^2 is aryl, R is not  , and with the further proviso that when R is arylalkyl, then any heteroaryl substituent on the aryl of said arylalkyl contains at least three heteroatoms.

The compounds of Formula III can be useful as protein kinase inhibitors and can be useful in the treatment and prevention of proliferative diseases, for example, cancer, inflammation and arthritis. They may also be useful in the treatment of neurodegenerative diseases such Alzheimer's disease, cardiovascular diseases, viral diseases and fungal diseases.

Detailed Description

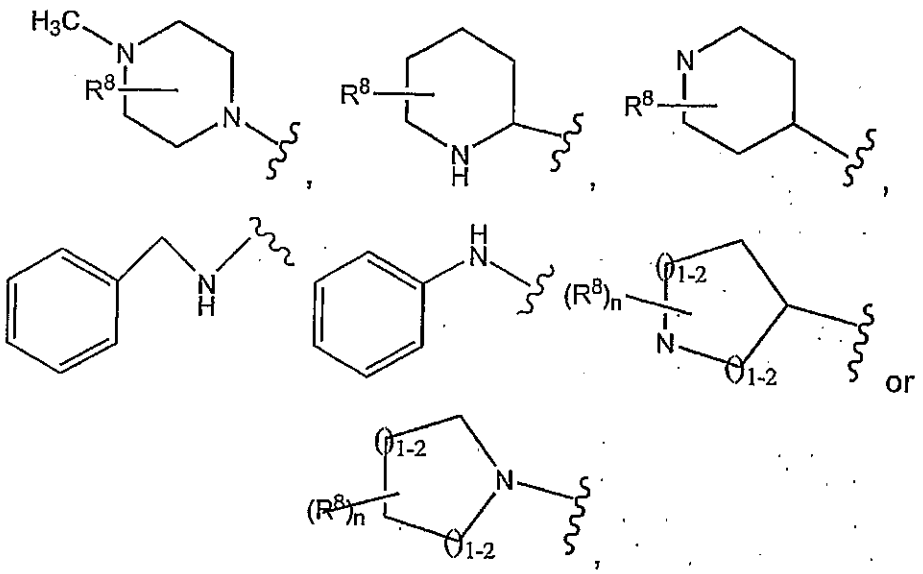
In one embodiment, the present invention discloses pyrazolo[1,5-a]pyrimidine compounds which are represented by structural Formula III, or a pharmaceutically acceptable salt or solvate thereof, wherein the various moieties are as described above.

In another embodiment, R is -$(CHR^5)_n$-aryl, –$(CHR^5)_n$-heteroaryl, –$(CHR^5)_n$-heteroaryl (with said heteroaryl being substituted with an additional, same or different, heteroaryl), –$(CHR^5)_n$-heterocyclyl (with said heterocyclyl being substituted with an additional, same or different, heterocyclyl), or



In another embodiment, R^2 is halogen, CF_3, CN, lower alkyl, alkyl substituted with -OR^6, alkynyl, aryl, heteroaryl or heterocyclyl.

In another embodiment, R^3 is H, lower alkyl, aryl, heteroaryl, cycloalkyl, -NR^5R^6,



wherein said alkyl, aryl, heteroaryl, cycloalkyl and the heterocyclyl structures shown immediately above for R^3 are optionally substituted with one or more moieties which can be the same or different, each moiety being independently selected from the group consisting of halogen, CF_3, OCF_3, lower alkyl, CN, $-C(O)R^5$, $-S(O_2)R^5$, $-C(=NH)-NH_2$, $-C(=CN)-NH_2$, hydroxyalkyl, alkoxycarbonyl, $-SR^5$, and OR^5, with the proviso that no carbon adjacent to a nitrogen atom on a heterocyclyl ring carries a $-OR^5$ moiety.

In another embodiment, R^4 is H or lower alkyl.

In another embodiment, R^5 is H, lower alkyl or cycloalkyl.

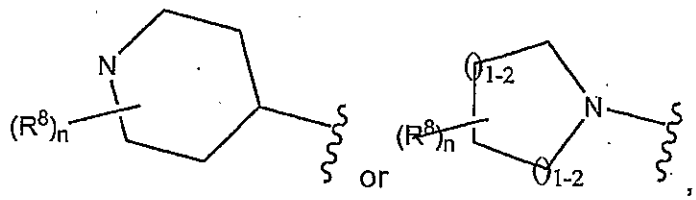
In another embodiment, n is 1 to 2.

In an additional embodiment, R is $-(CHR^5)_n$-aryl, $-(CHR^5)_n$-heteroaryl.

In an additional embodiment, R^2 is halogen, CF_3, CN, lower alkyl, alkynyl, or alkyl substituted with $-OR^6$.

In an additional embodiment, R^2 is lower alkyl, alkynyl or Br.

In an additional embodiment, R^3 is H, lower alkyl, aryl,



wherein said alkyl, aryl and the heterocyclyl moieties shown immediately above for R^3 are optionally substituted with one or more moieties which can be the same

or different, each moiety being independently selected from the group consisting of halogen, CF_3, lower alkyl, hydroxyalkyl, alkoxy, $-S(O_2)R^5$, and CN.

In an additional embodiment, R^4 is H.

In an additional embodiment, R^5 is H, ethyl, cyclopropyl, cyclobutyl, cyclopentyl or cyclohexyl.

In an additional embodiment, R^8 is alkyl or hydroxyalkyl.

In an additional embodiment, n is 1.

In an additional embodiment, p is 1 or 2.

In another embodiment, R is H.

In another embodiment, R^2 is halogen.

In another embodiment, R^2 is thiophene, furan, pyridine, pyrazole, alkylthio or arylthio, wherein each of said alkyl and aryl of R^2 can independently be unsubstituted or substituted as defined above.

In another embodiment, R^2 is thiophene, furan, pyridine or pyrazole.

In another embodiment, R^2 is an amide which can be unsubstituted or substituted as defined above.

In another embodiment, R^2 is an urea, which may be unsubstituted or substituted as defined above.

In another embodiment, R^2 is an alkenyl.

In another embodiment, R^2 is an alkynyl.

In another embodiment, R3 is $-NR^5R^6$.

In another embodiment, R^4 is H.

In another embodiment, R^4 is unsubstituted lower alkyl.

In another embodiment, R^4 is alkyl substituted with $-OR^6$.

In another embodiment, R^4 is halo.

In another embodiment, R^4 is -F, -Cl or -Br.

In another embodiment, both R^2 and R^4 are halogen.

In another embodiment, both R^2 and R^4 are halo and R is H.

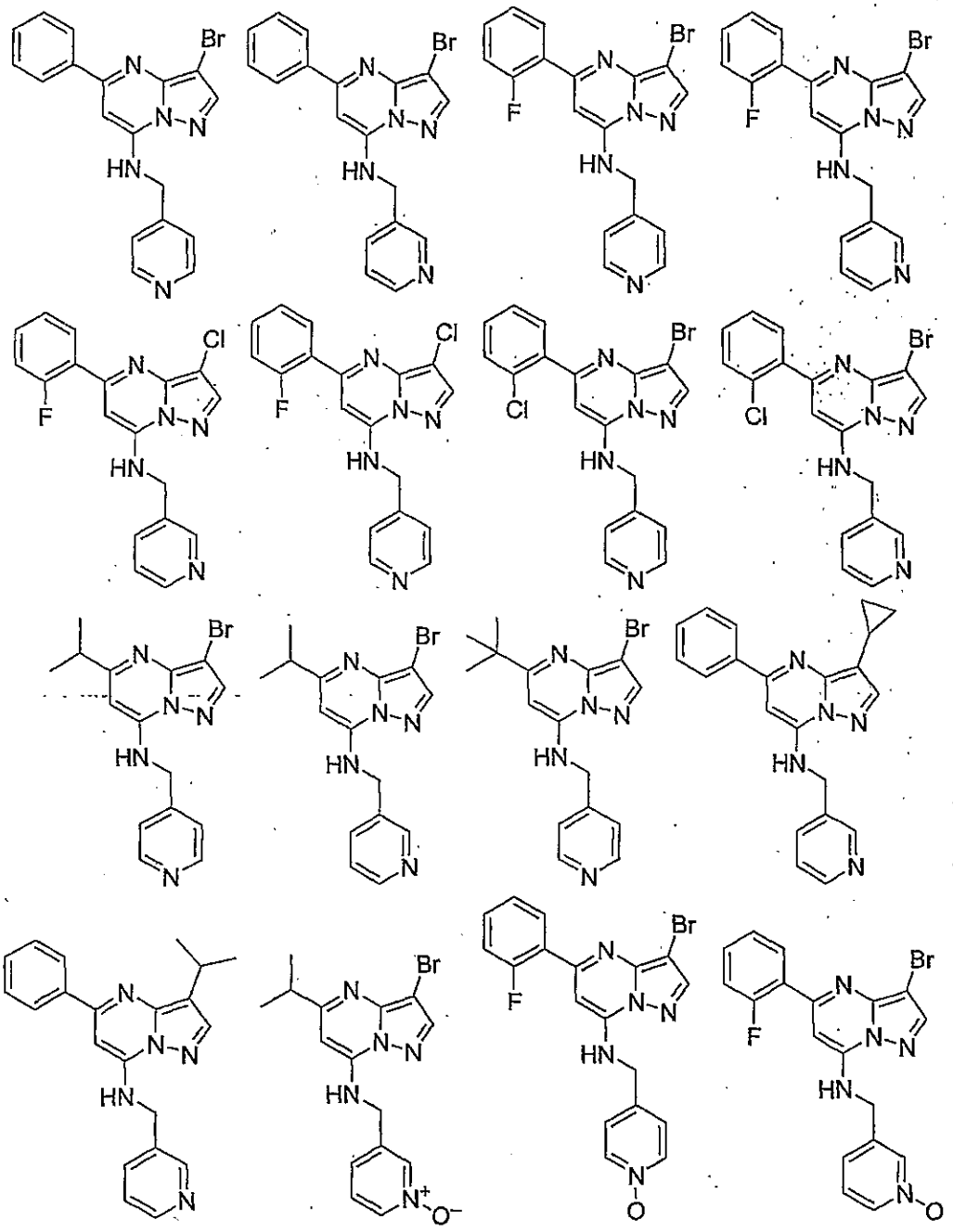
In another embodiment, R is H, R^2 is halo and R^3 is heteroaryl.

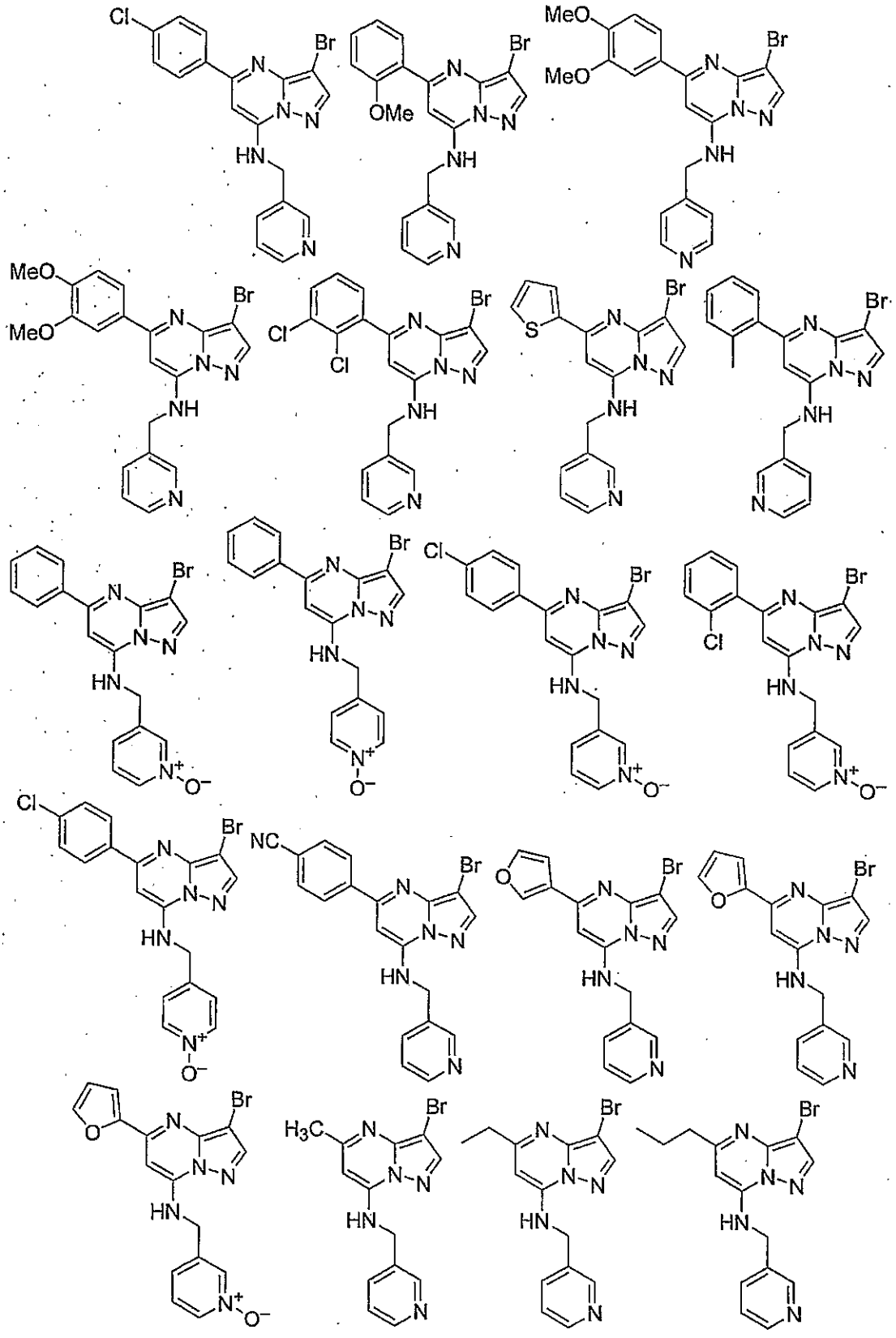
In another embodiment, R is H, R^2 is halo and R^3 is aryl.

In another embodiment, R is H, R^2 is halo and R^3 is heterocyclyl.

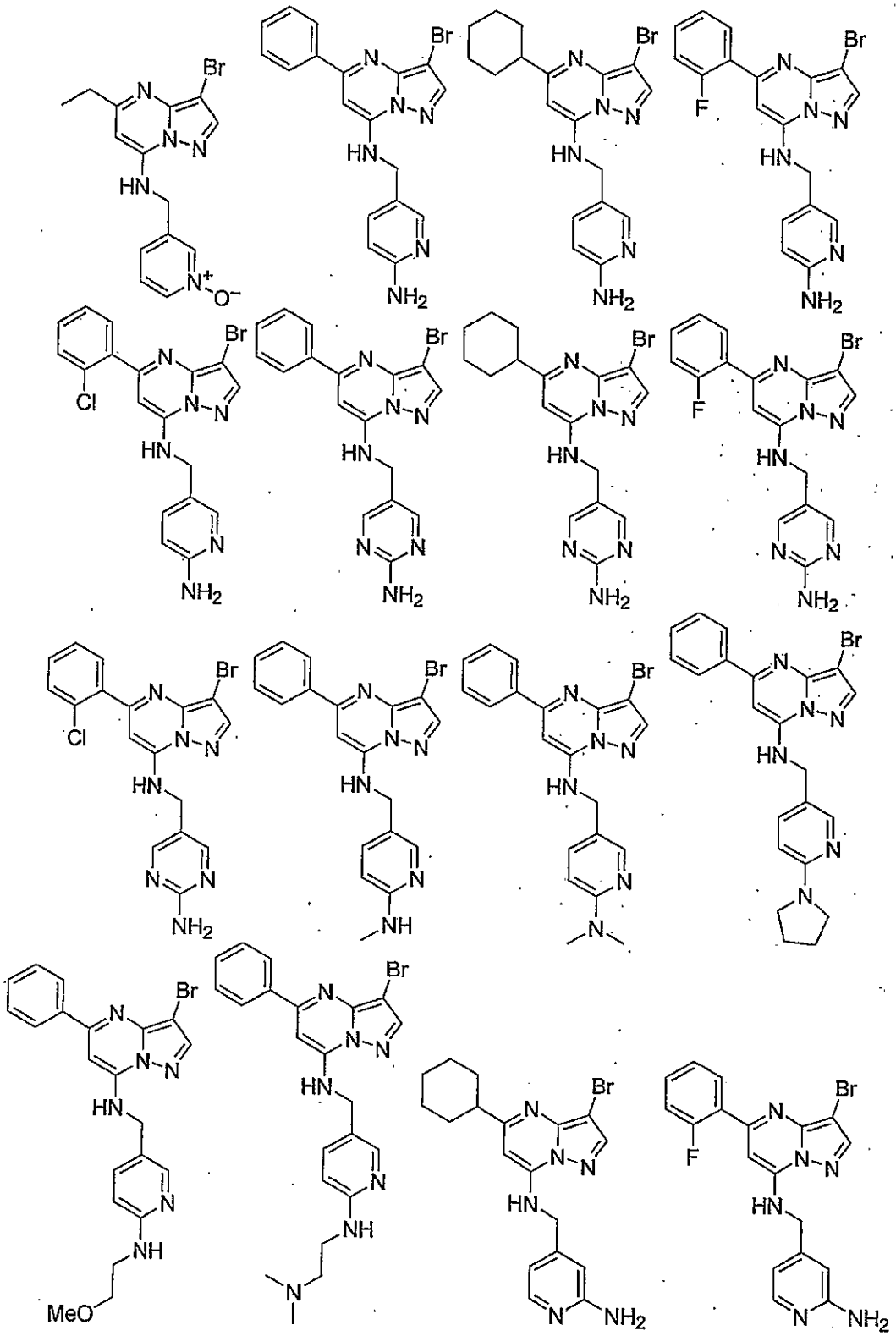
Another embodiment discloses the inventive compounds shown in **Table 1**, which exhibited CDK2 inhibitory activity of about 0.0001μM to > about 5 μM. The assay methods are described later (from page 333 onwards).

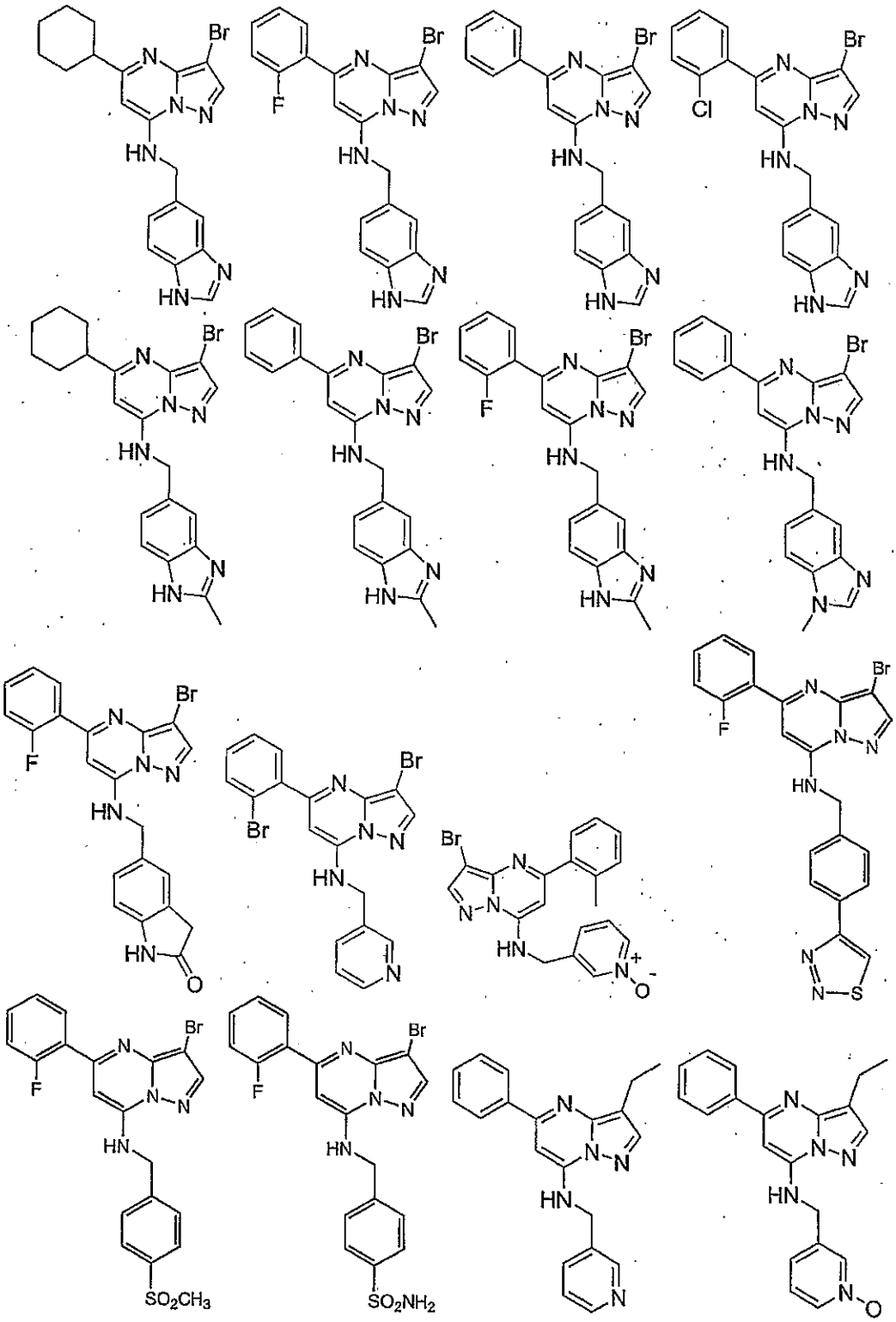
Table 1

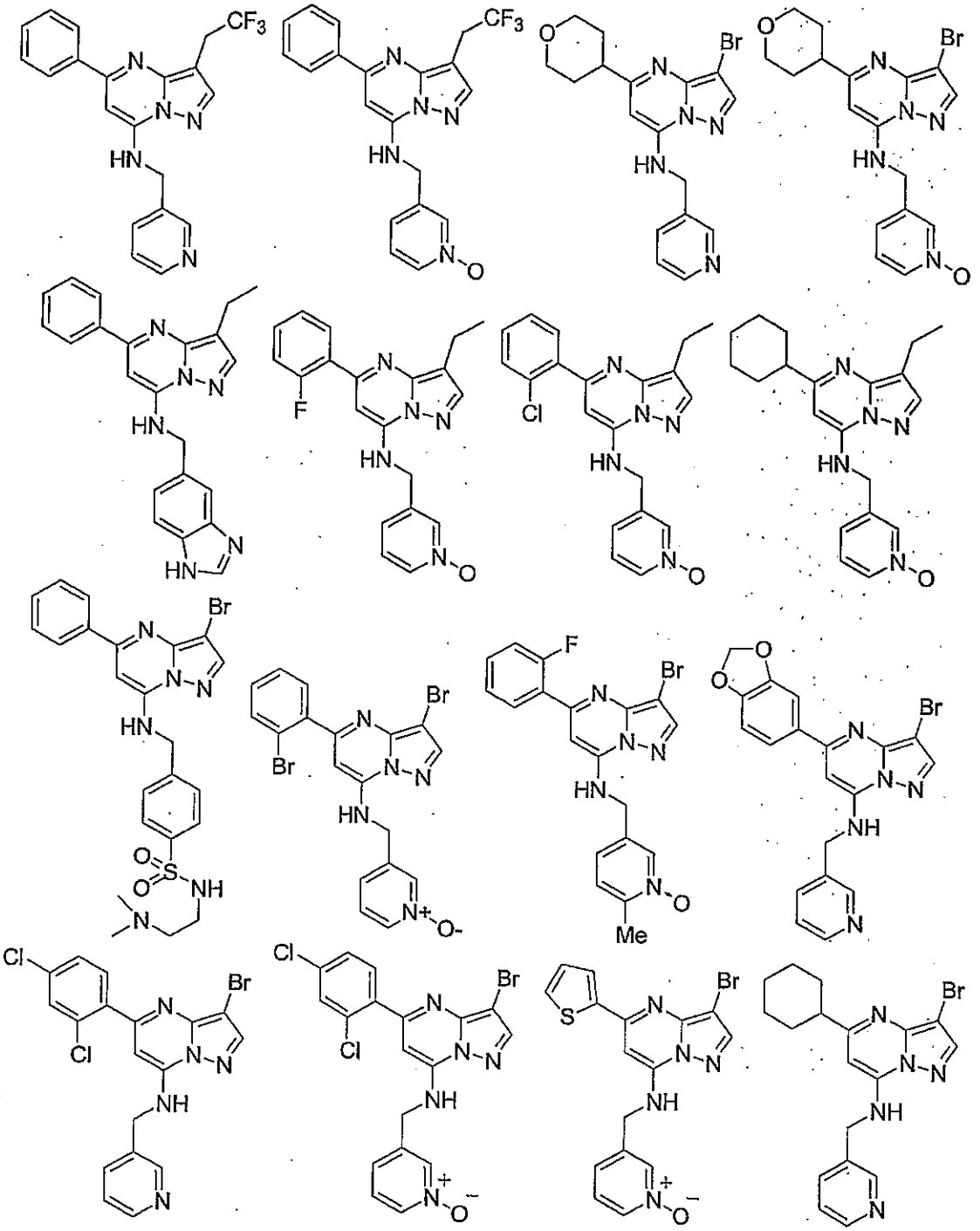


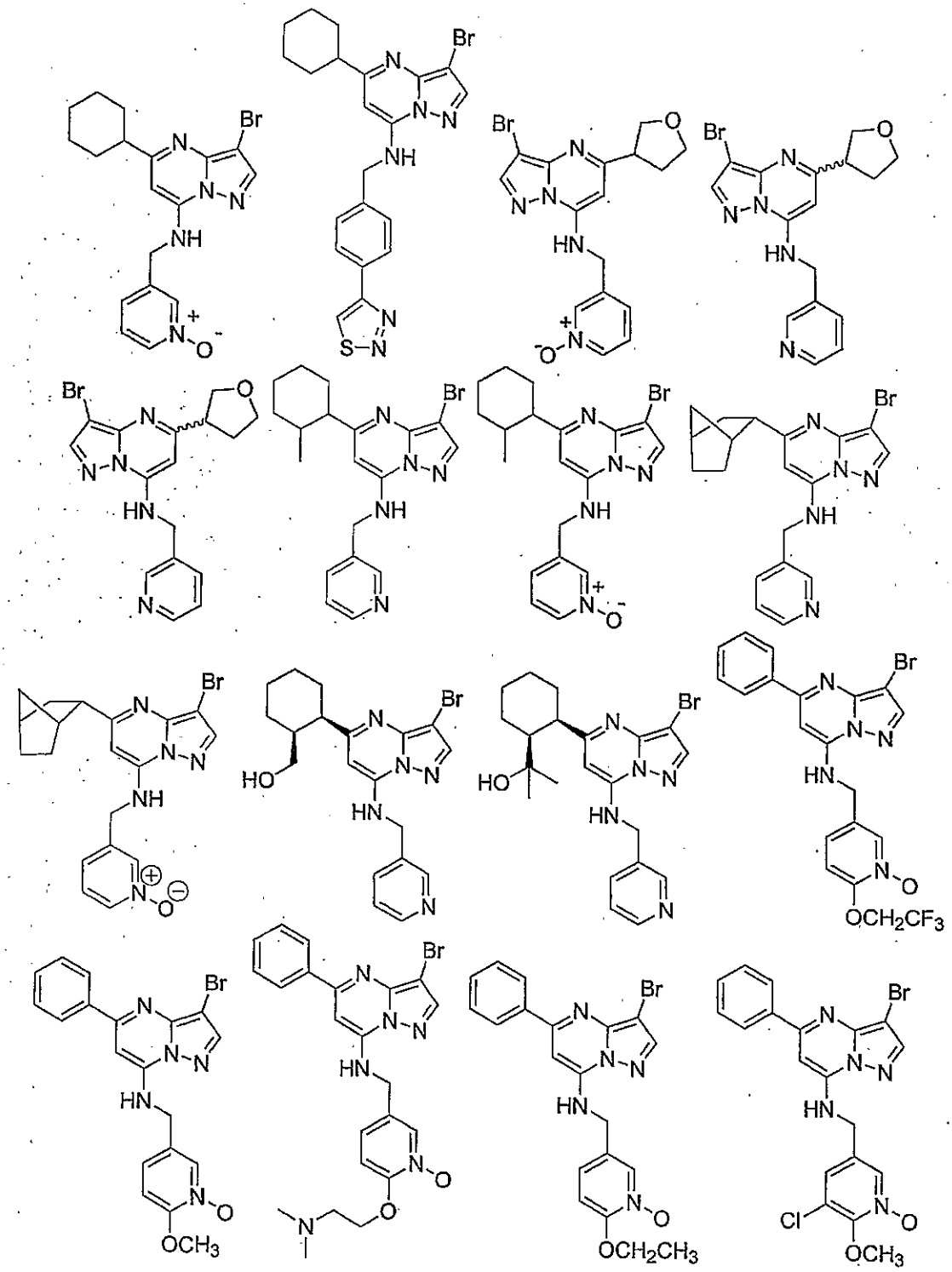


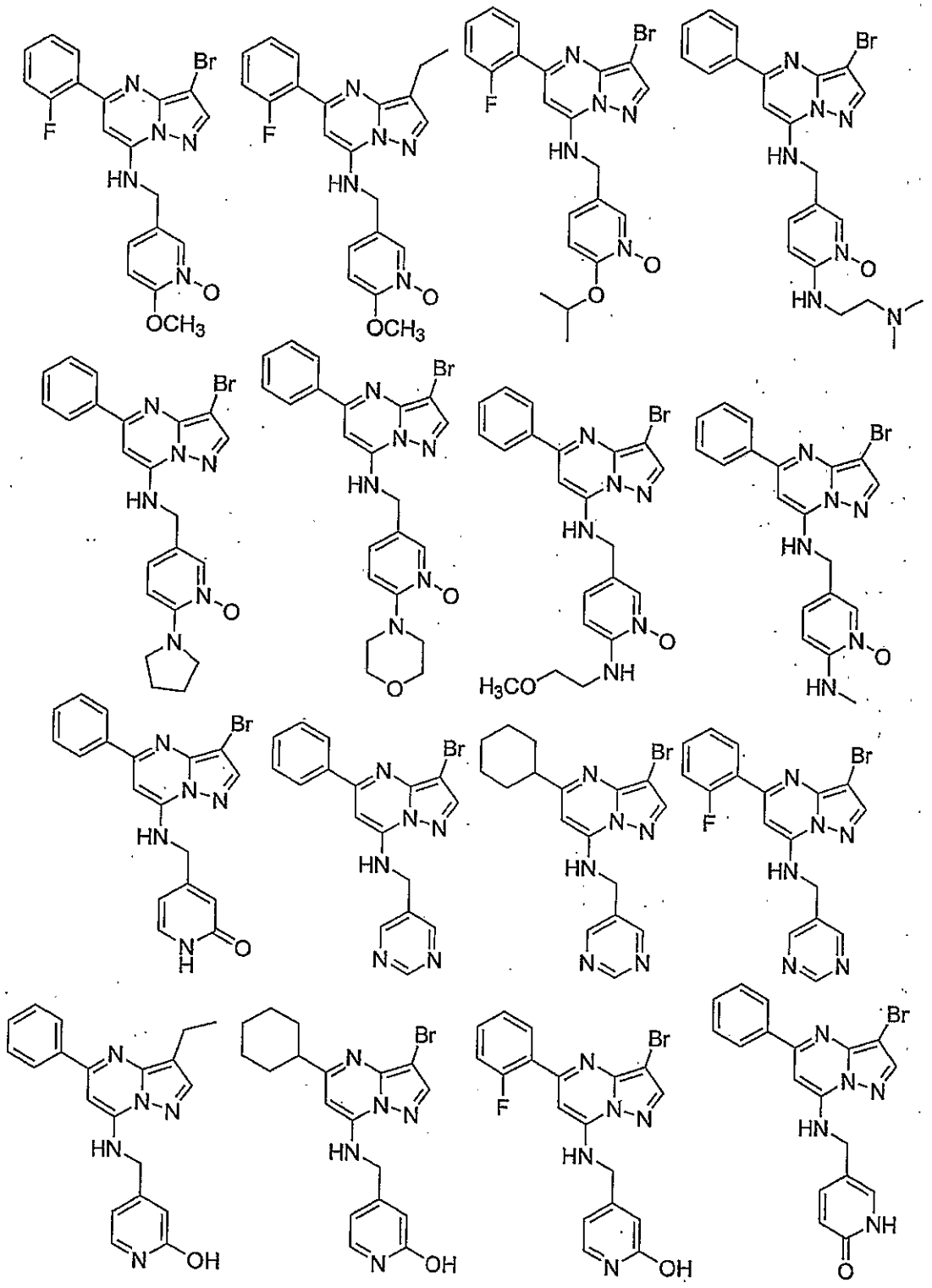
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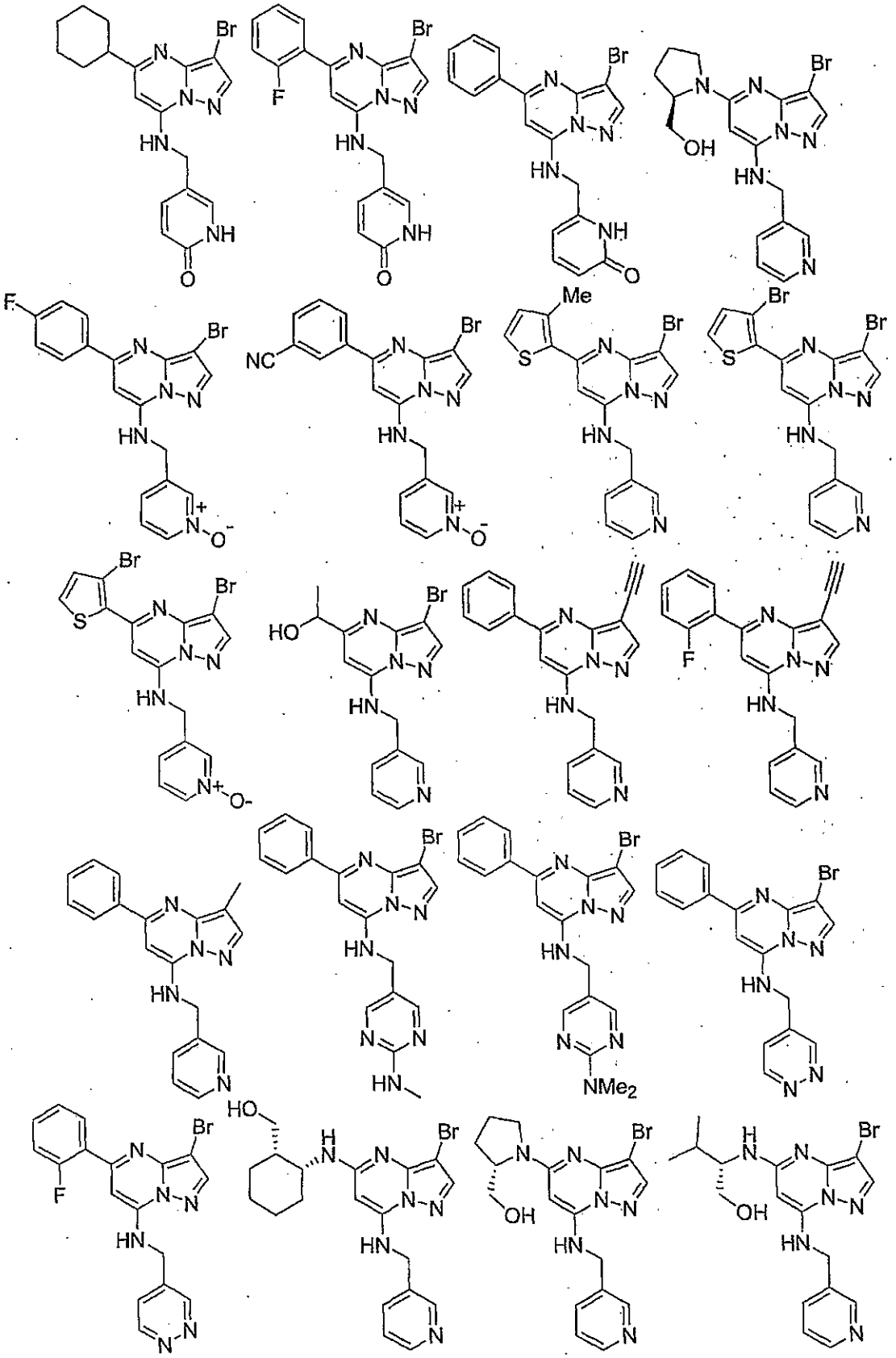


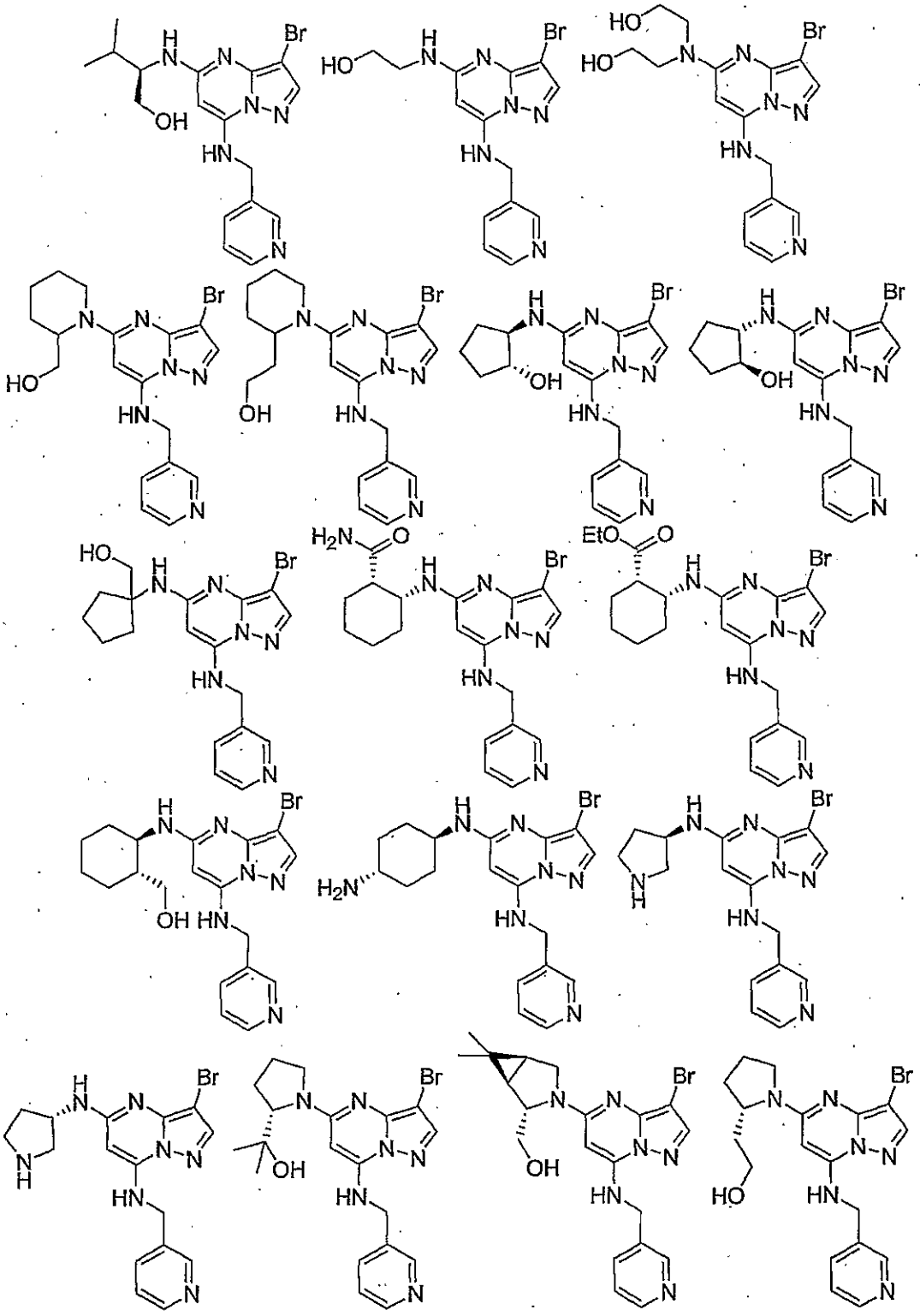




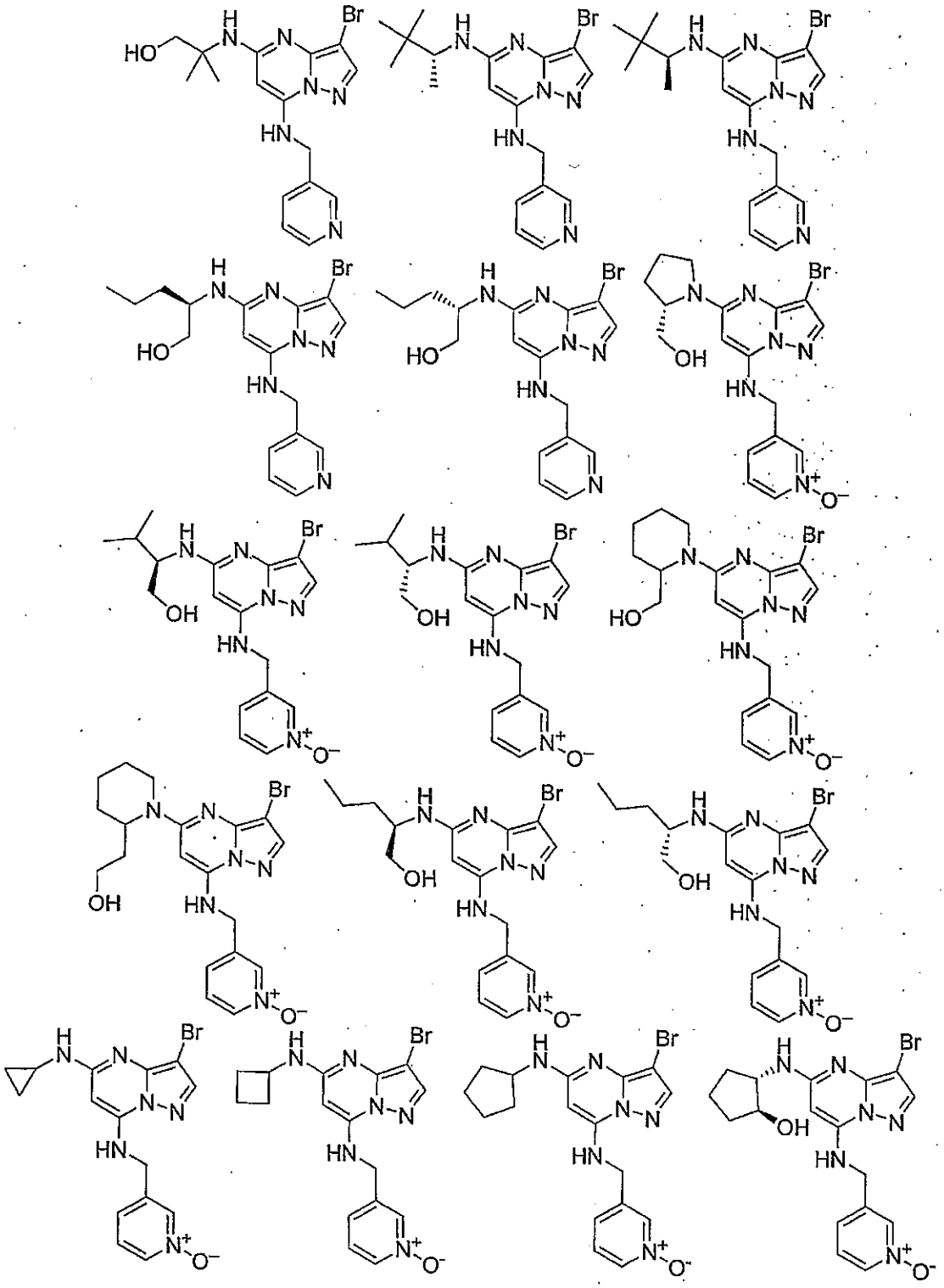




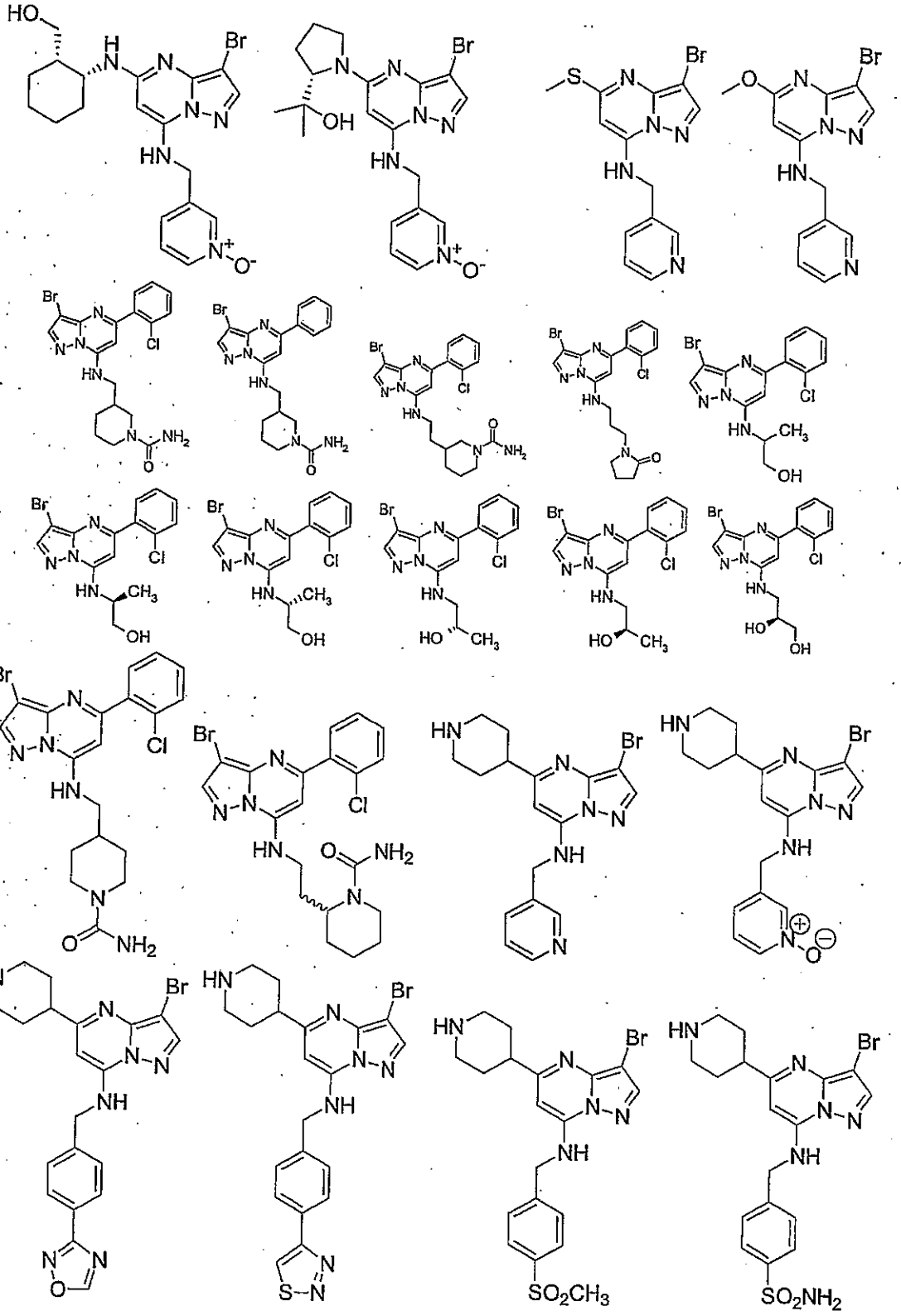




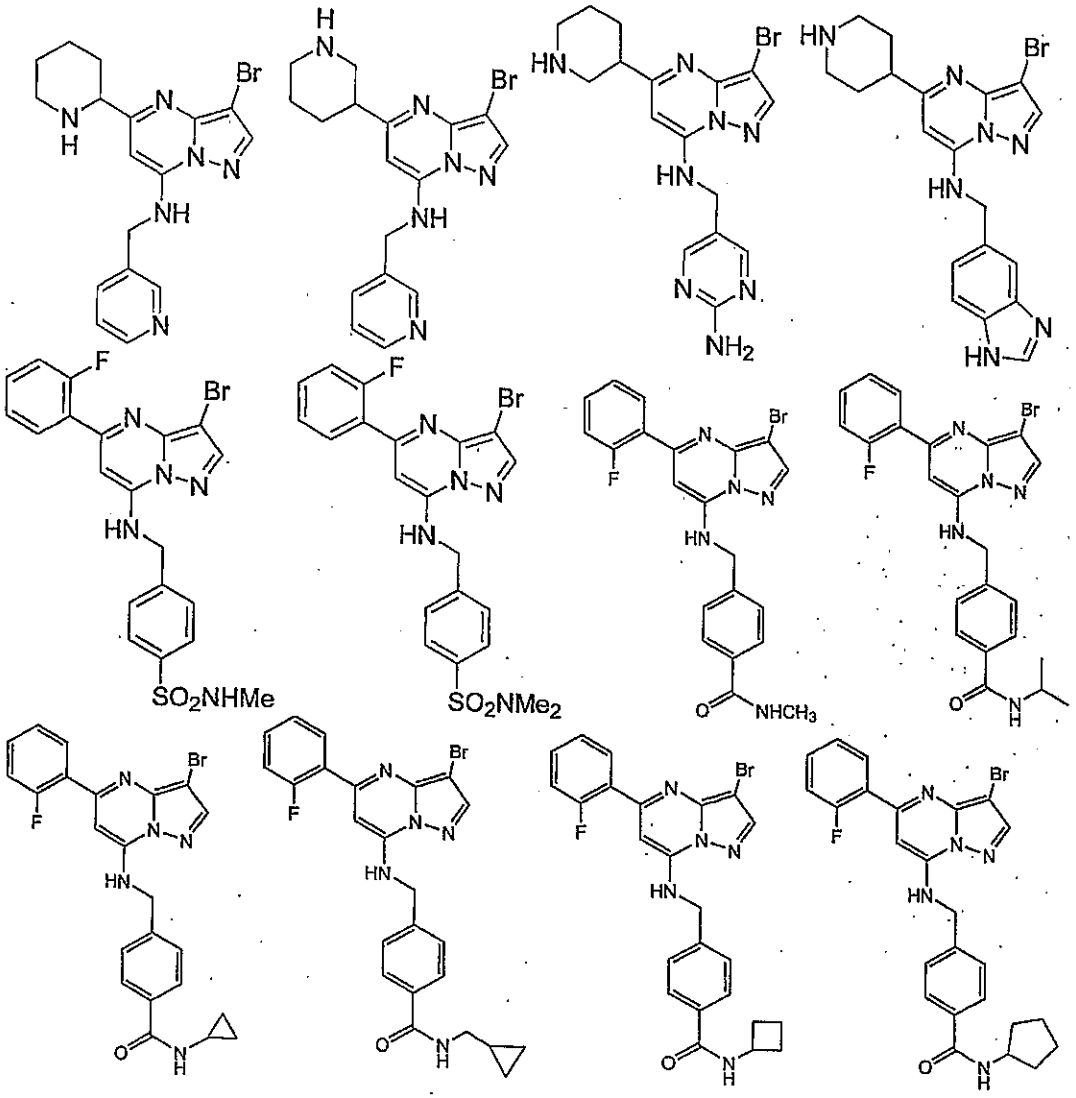
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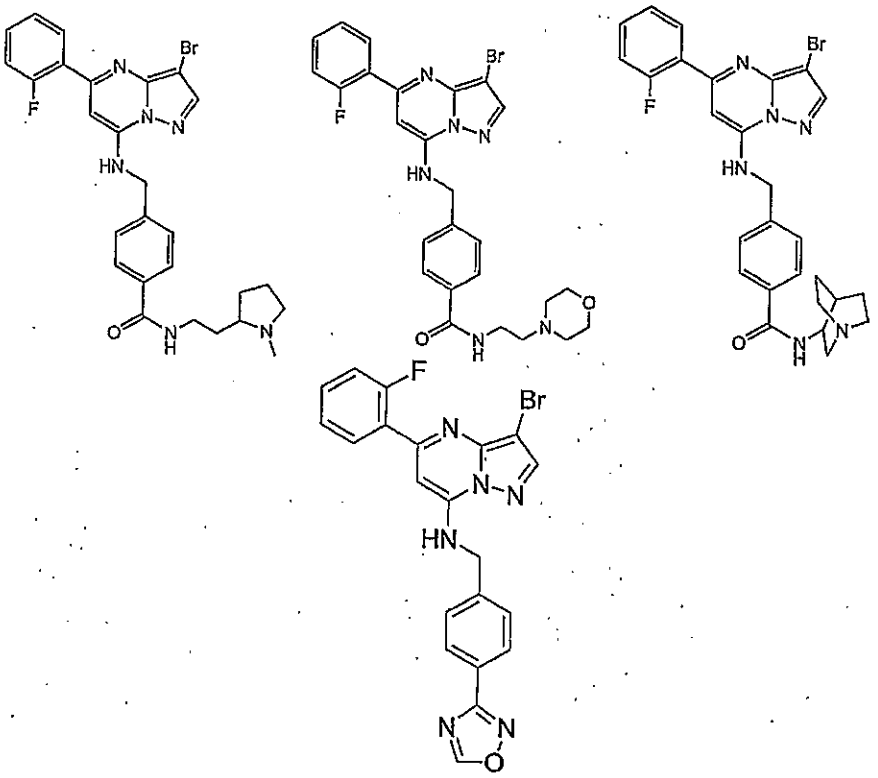


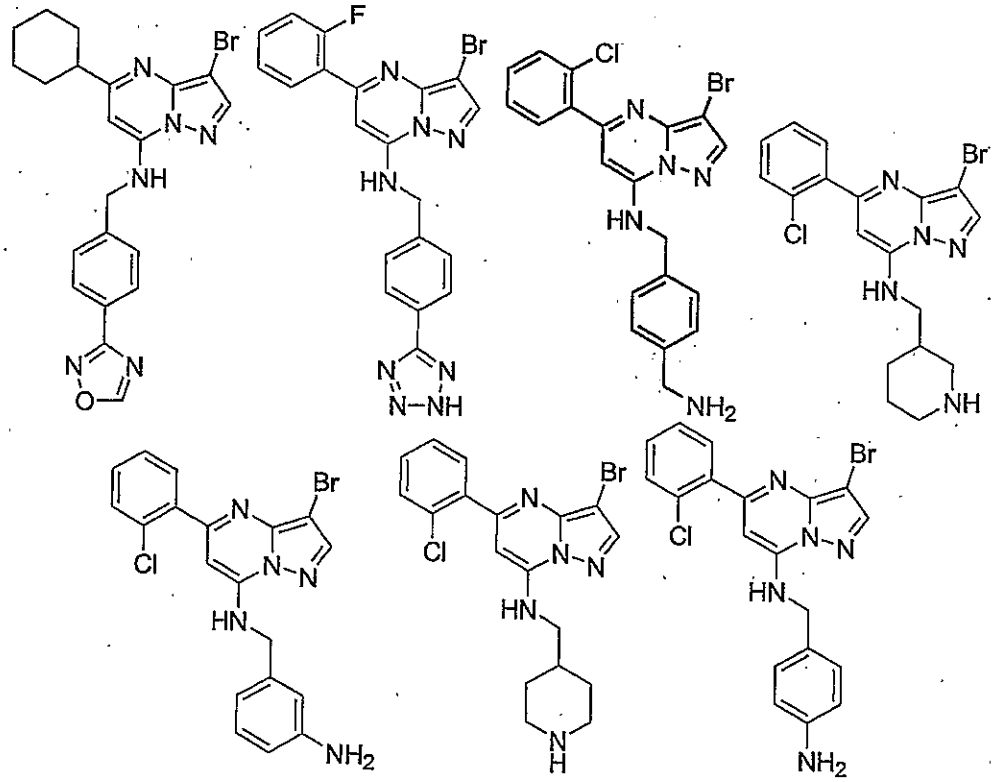
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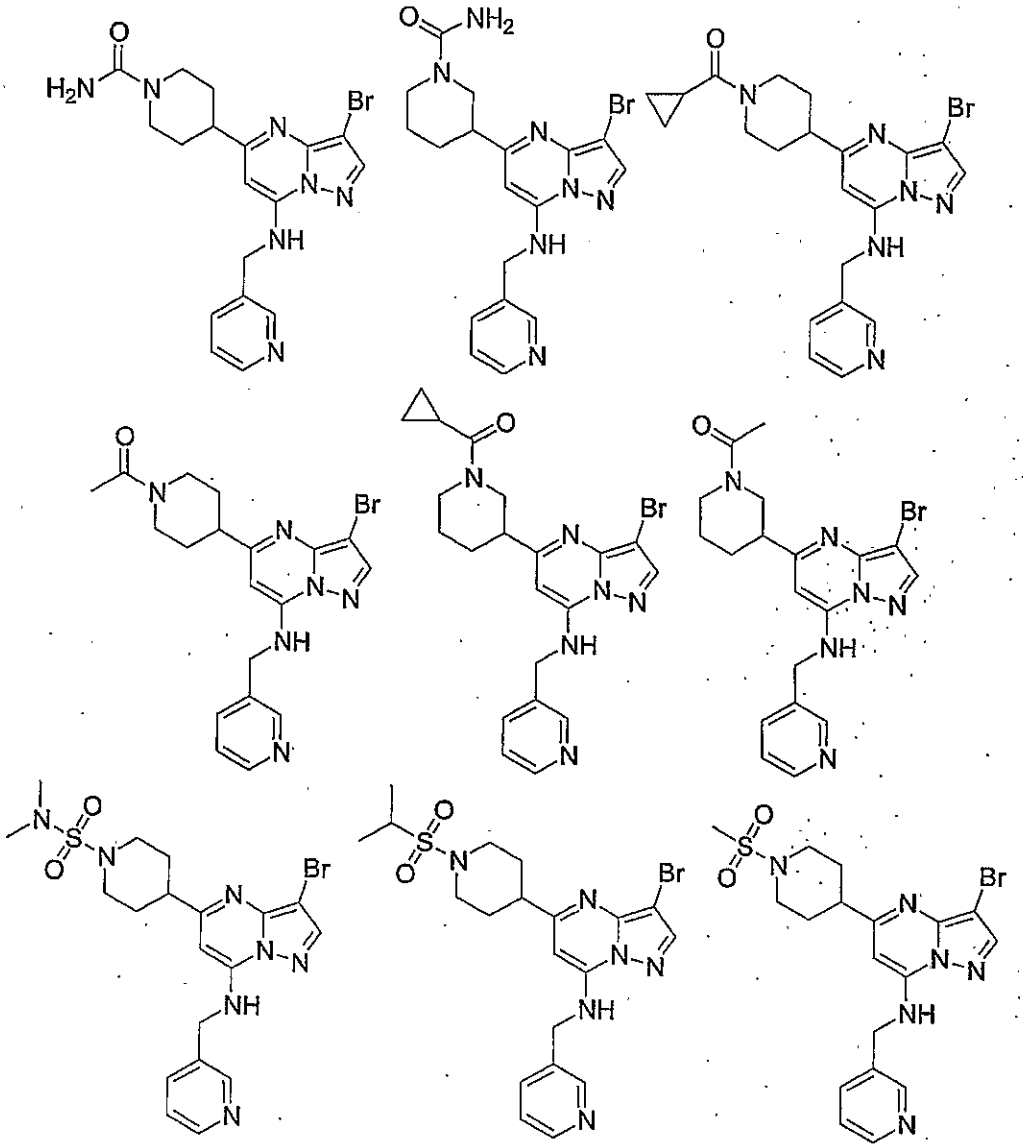
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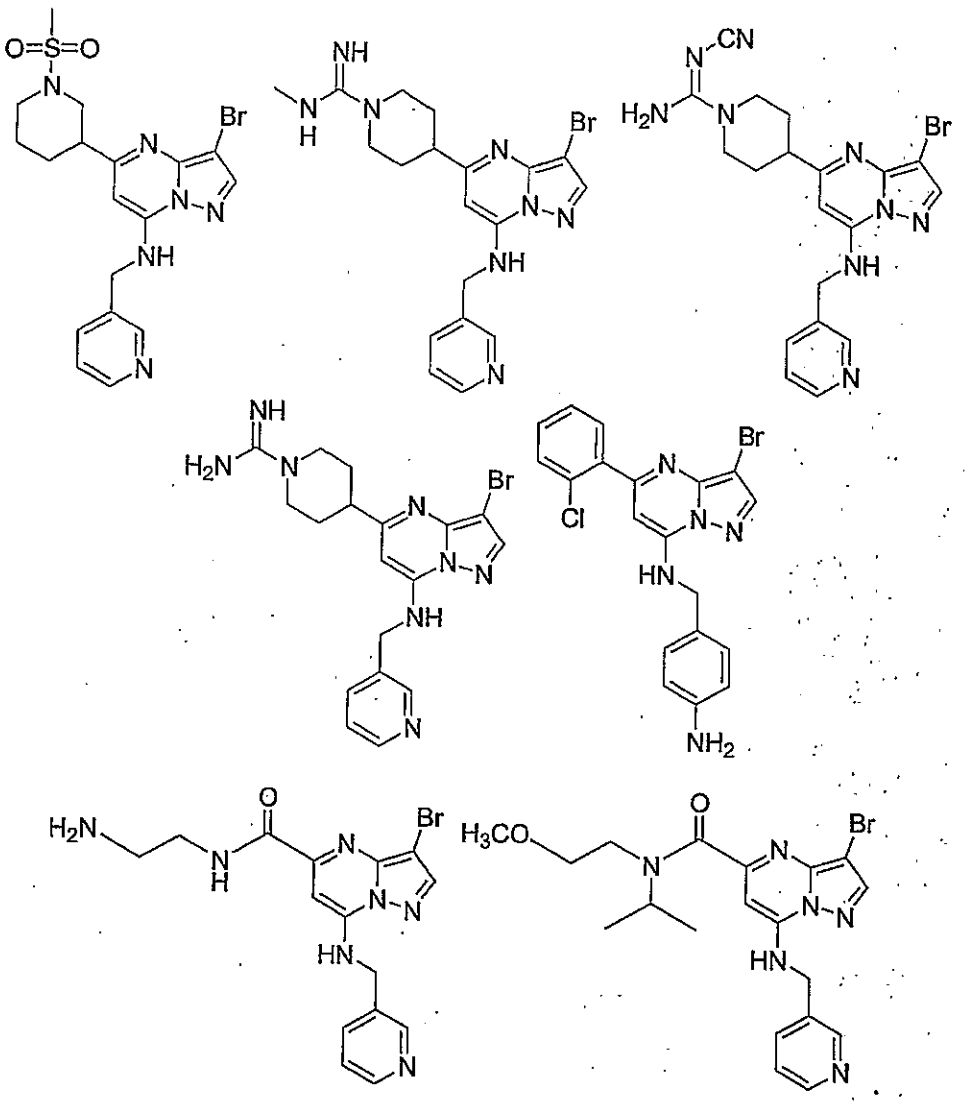




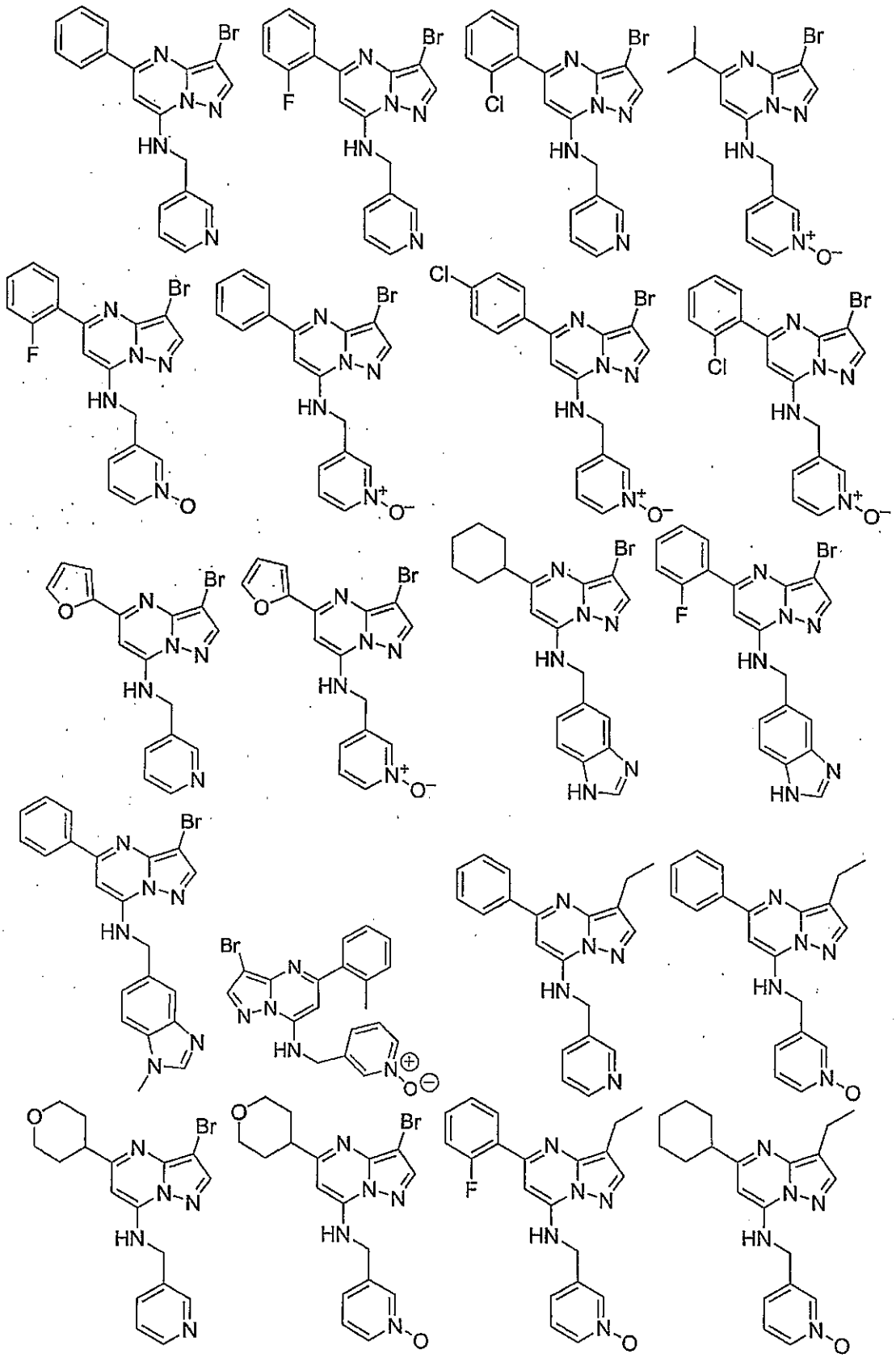


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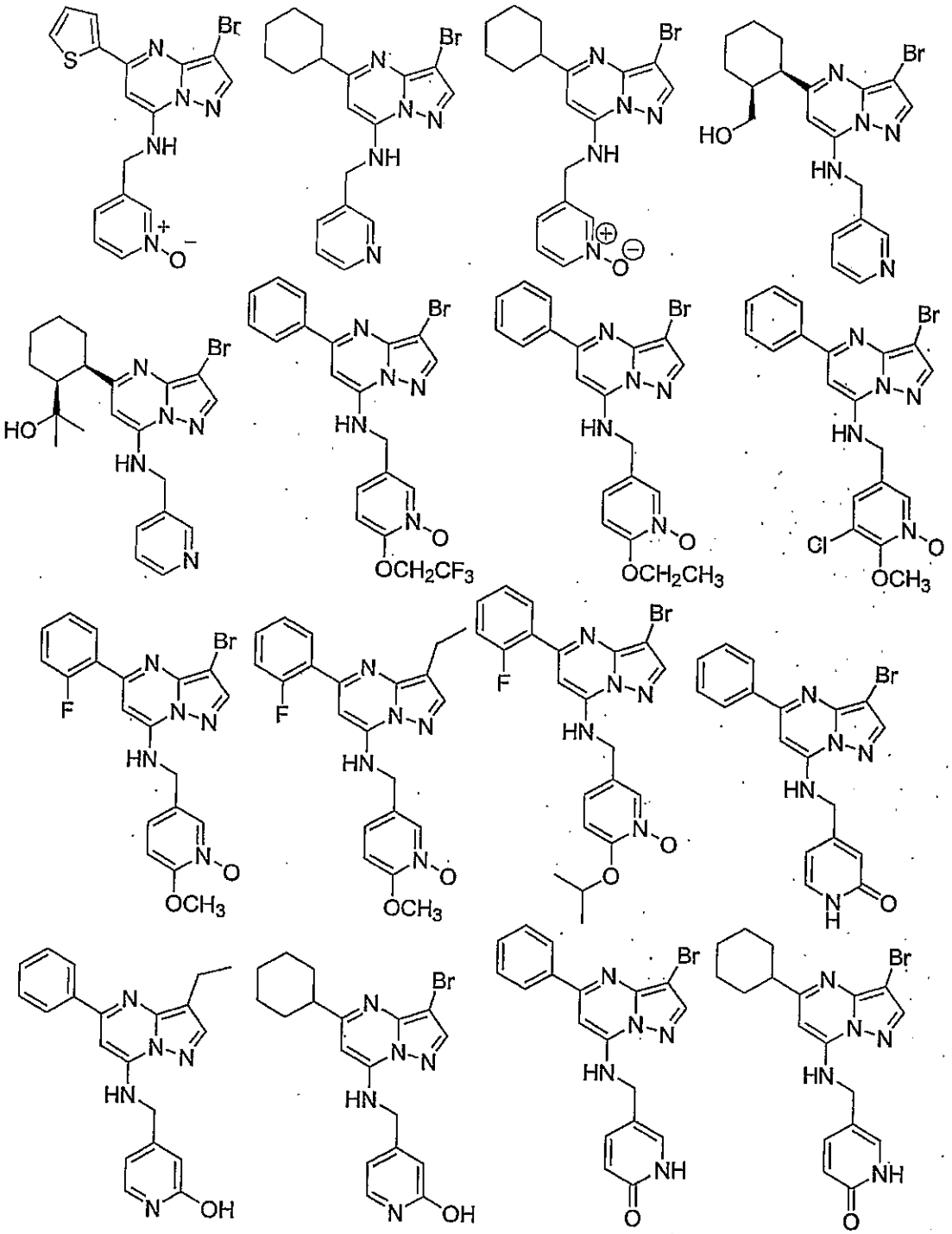


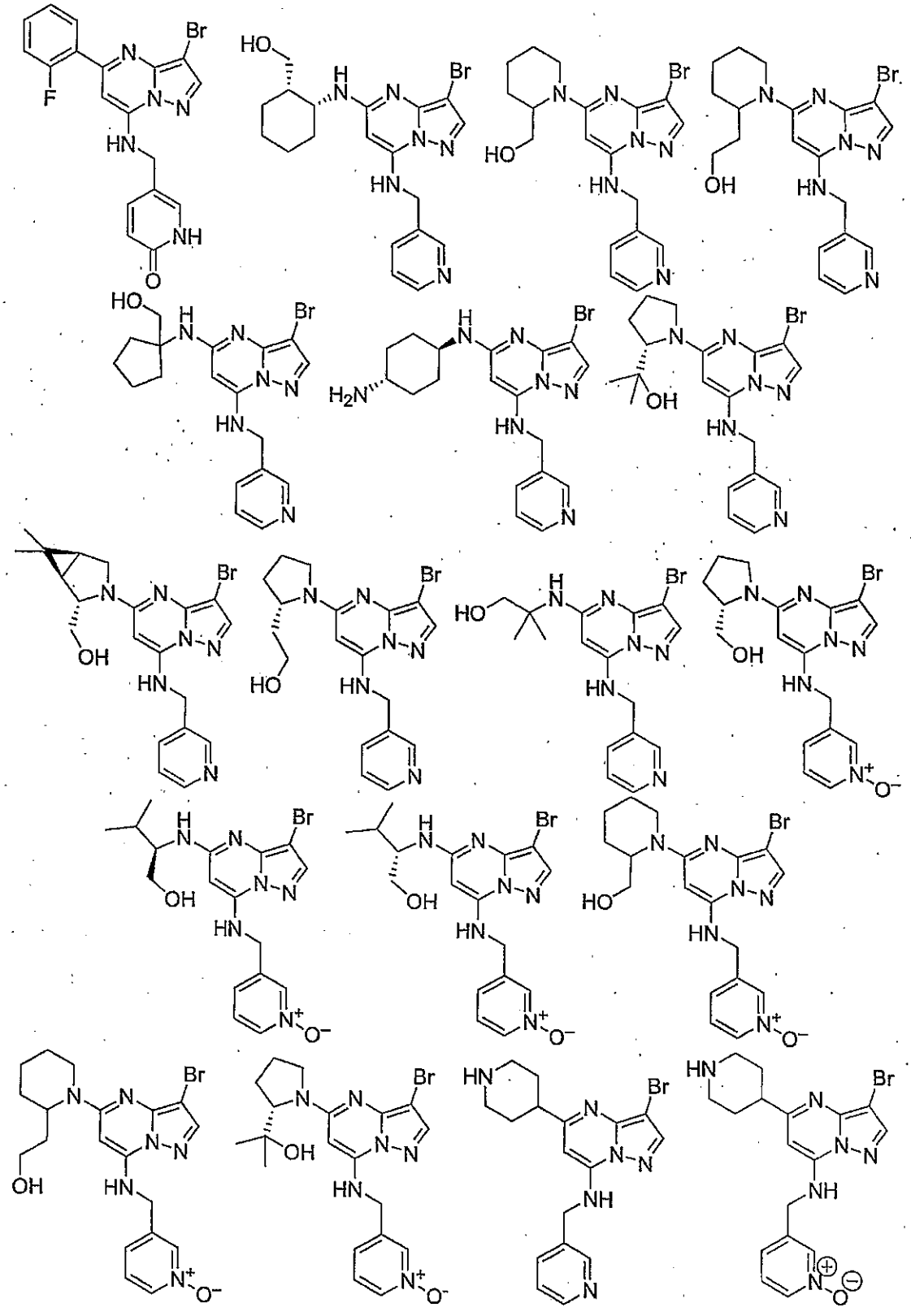


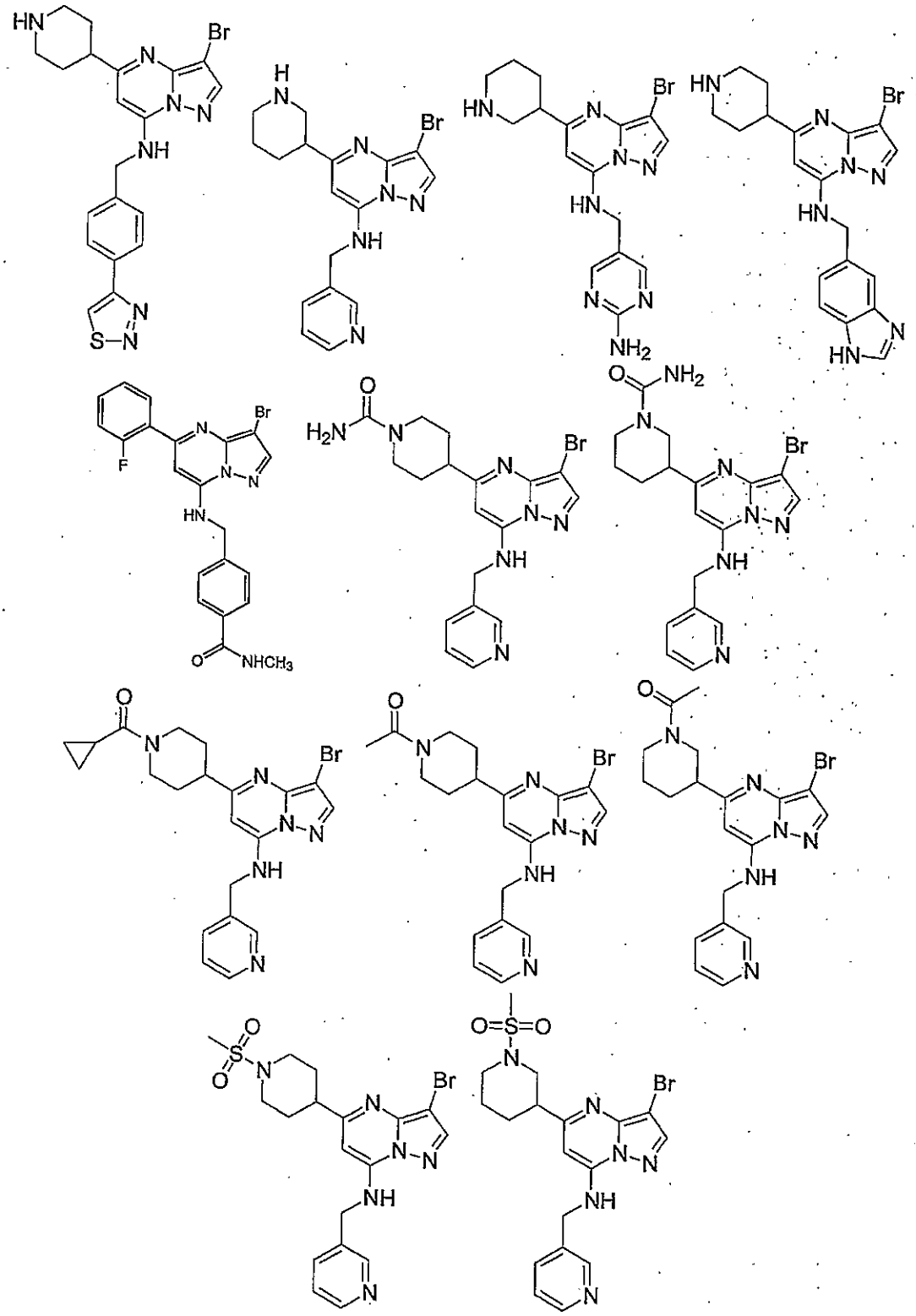
Another embodiment of the invention discloses the following compounds, which exhibited CDK2 inhibitory activity of about 0.0001μM to about 0.5μM:



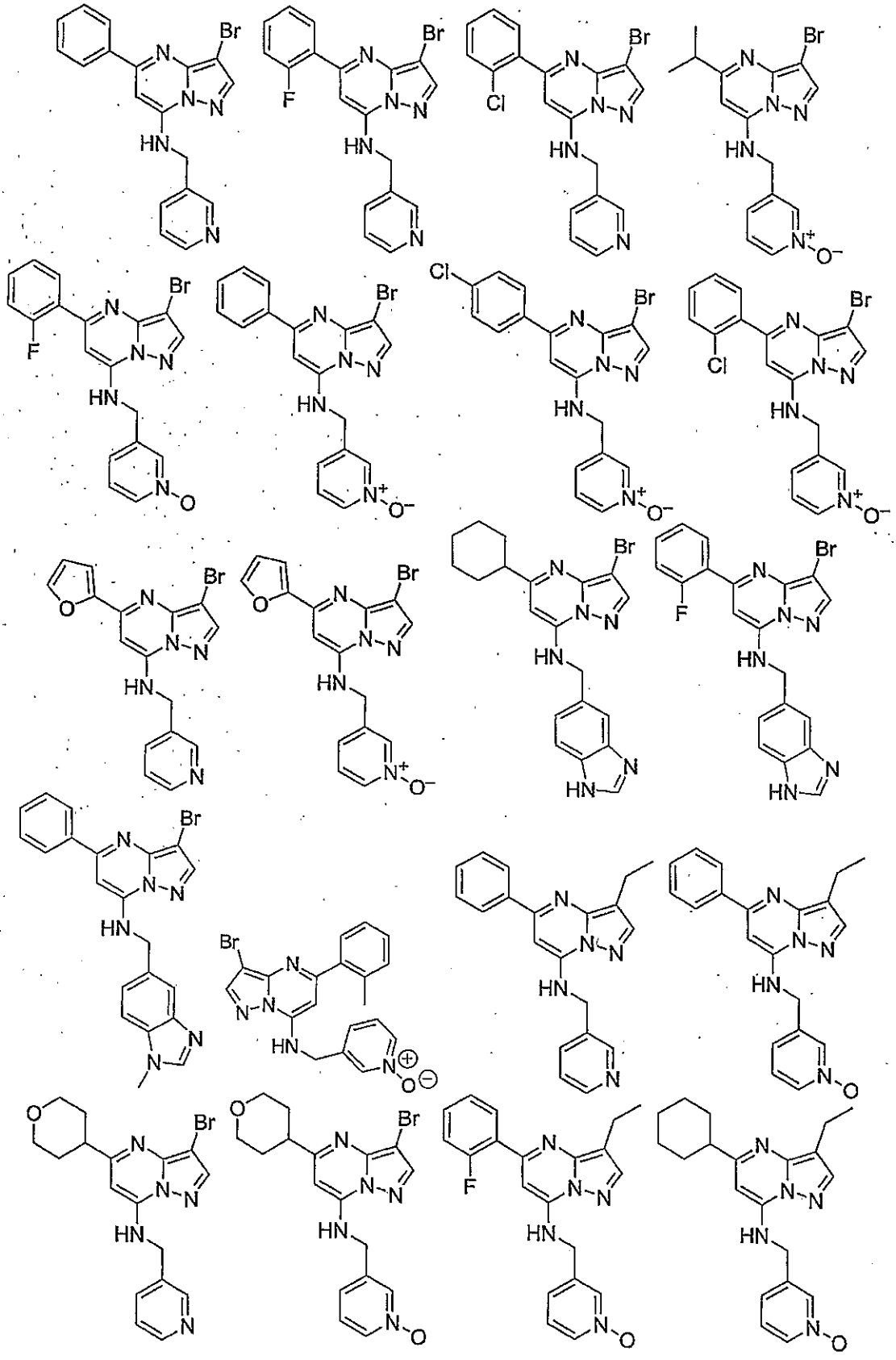
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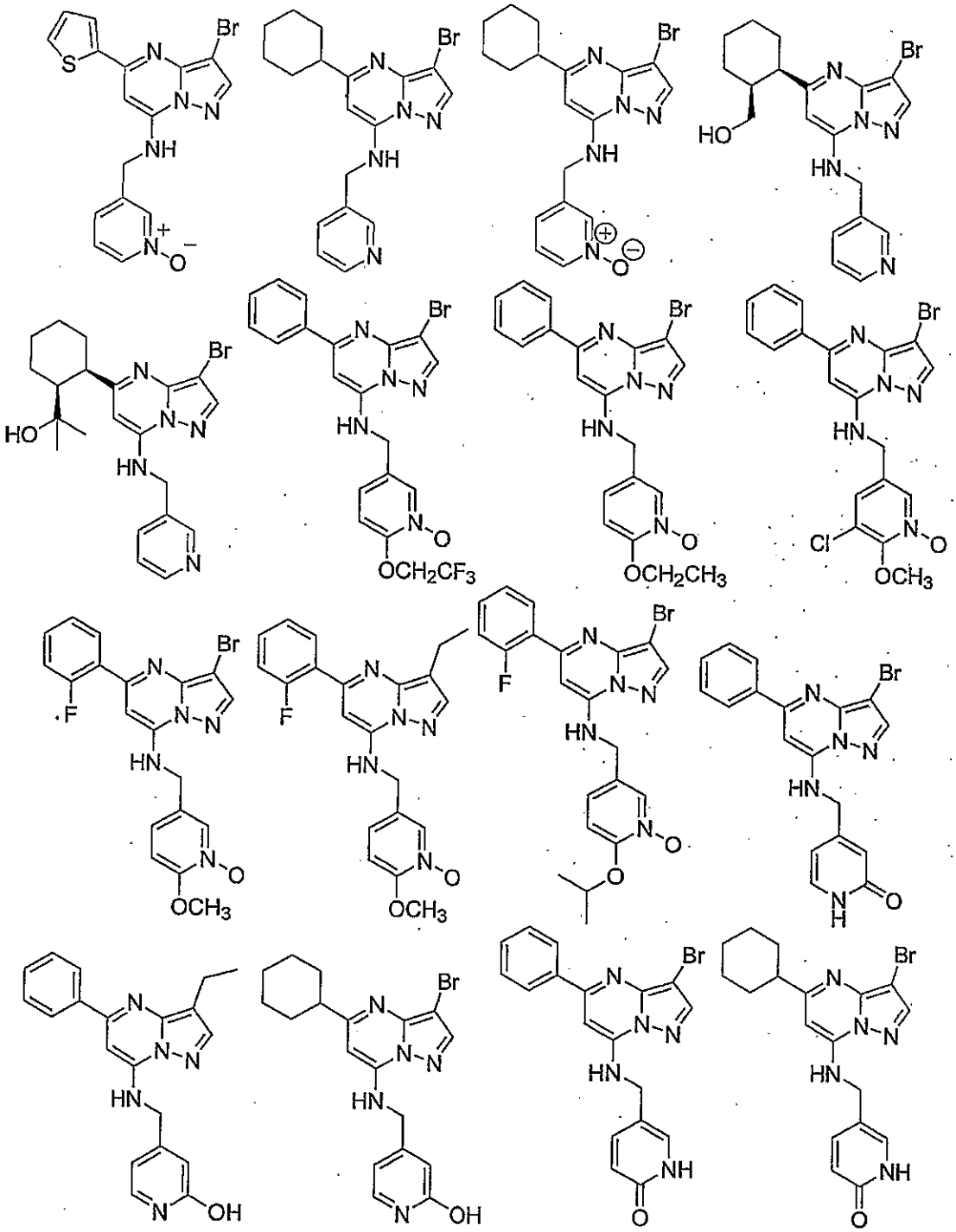


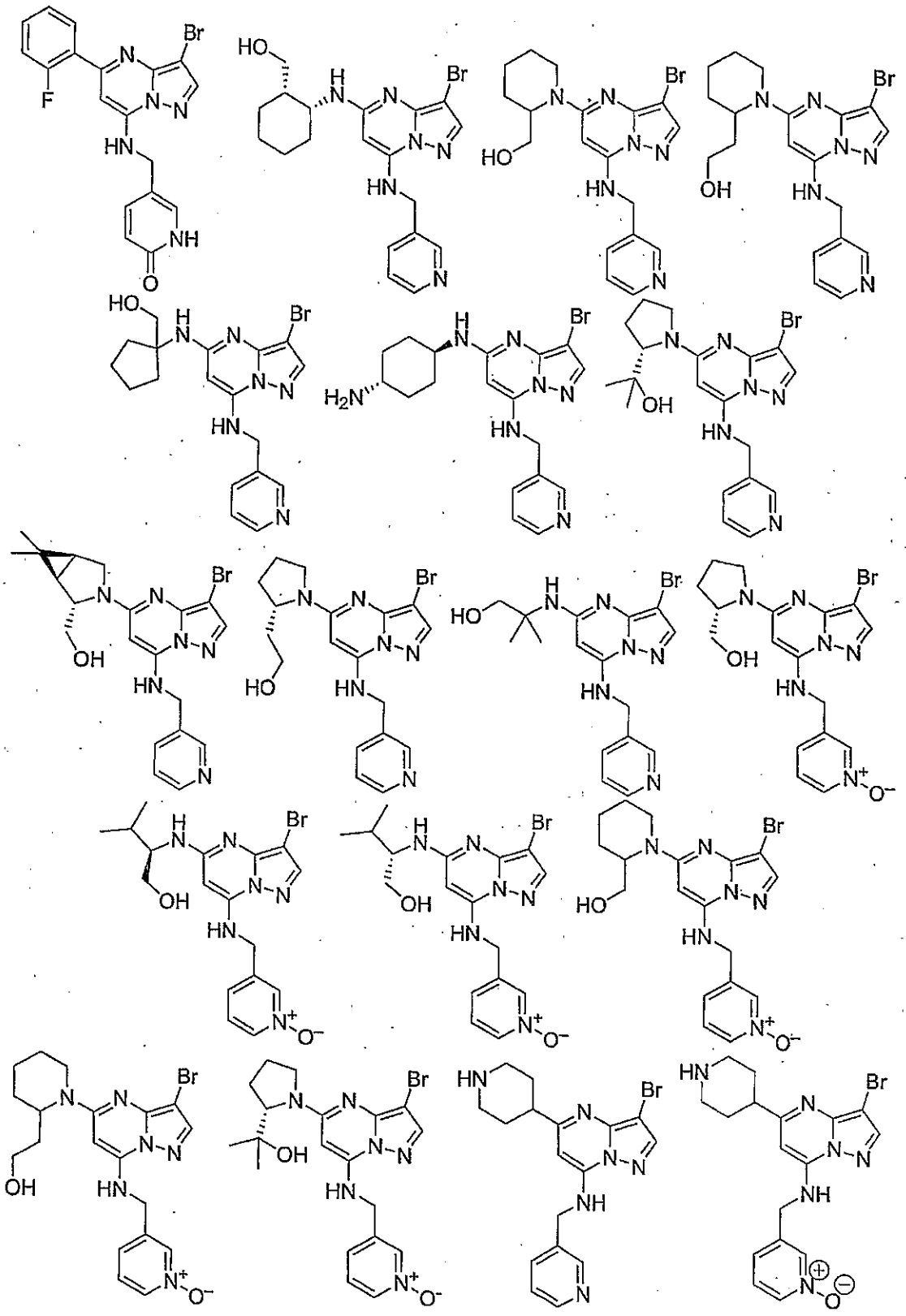


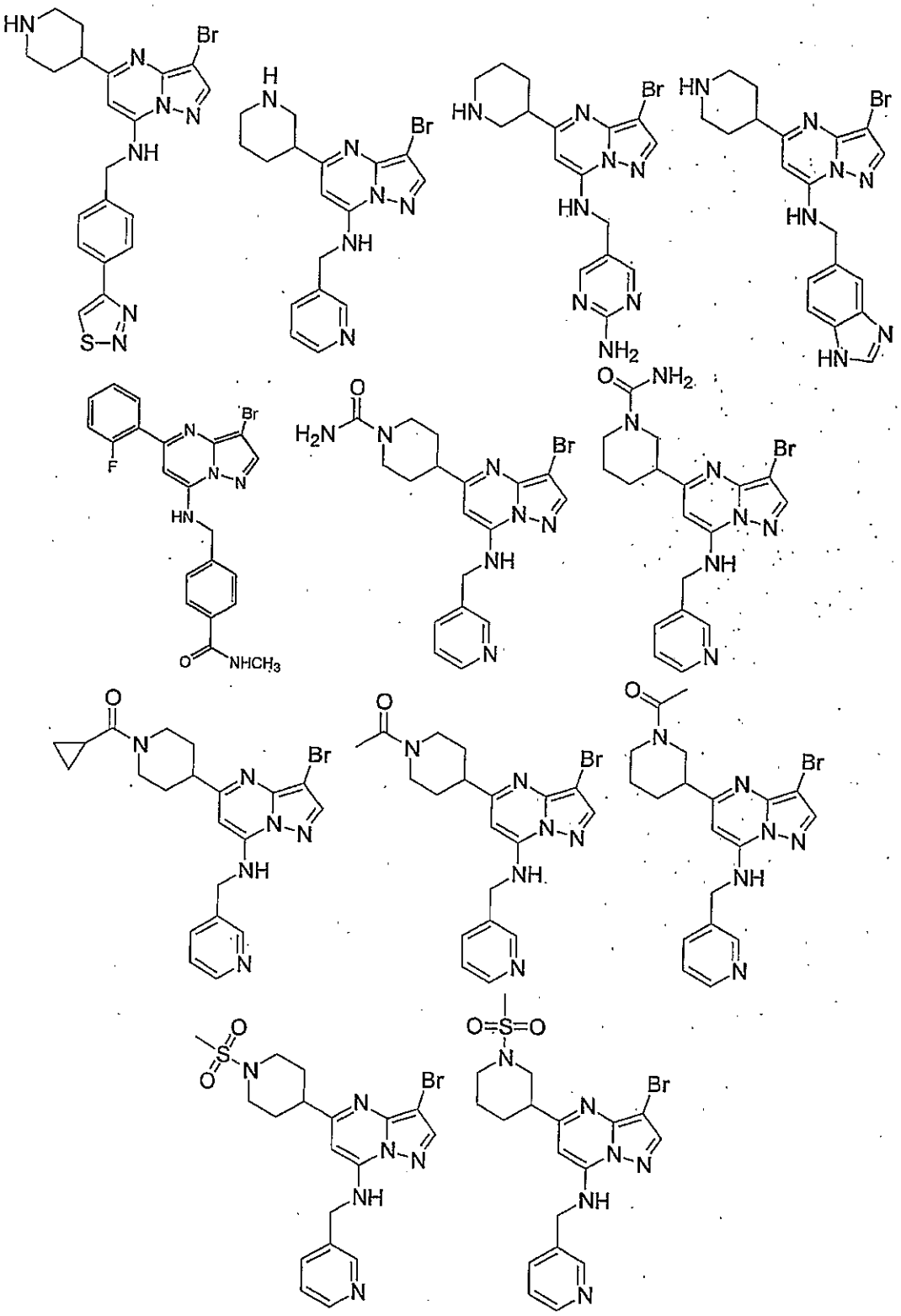


Another embodiment of the invention discloses the following compounds, which exhibited CDK2 inhibitory activity of about 0.0001μM to about 0.1μM:

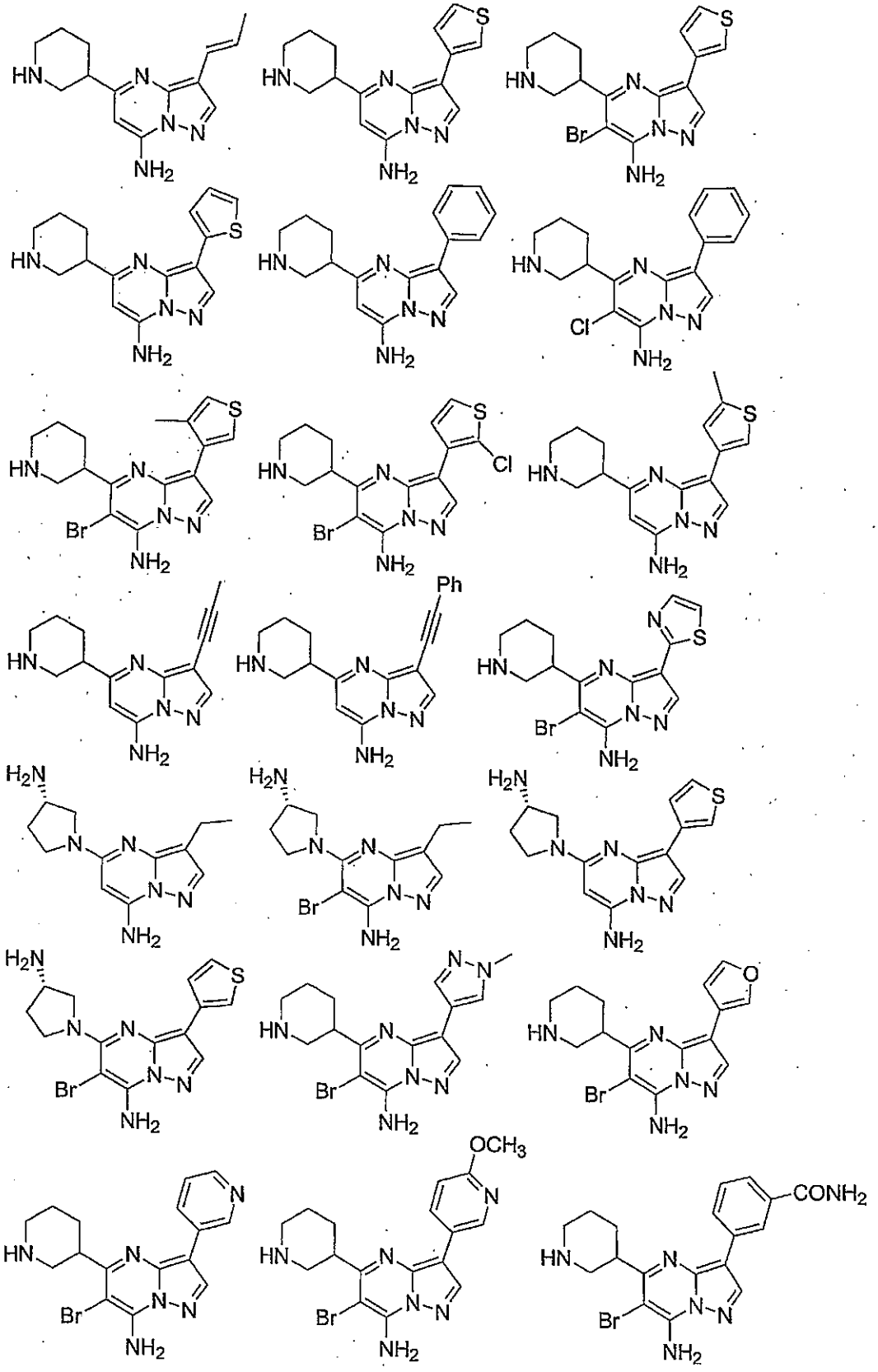


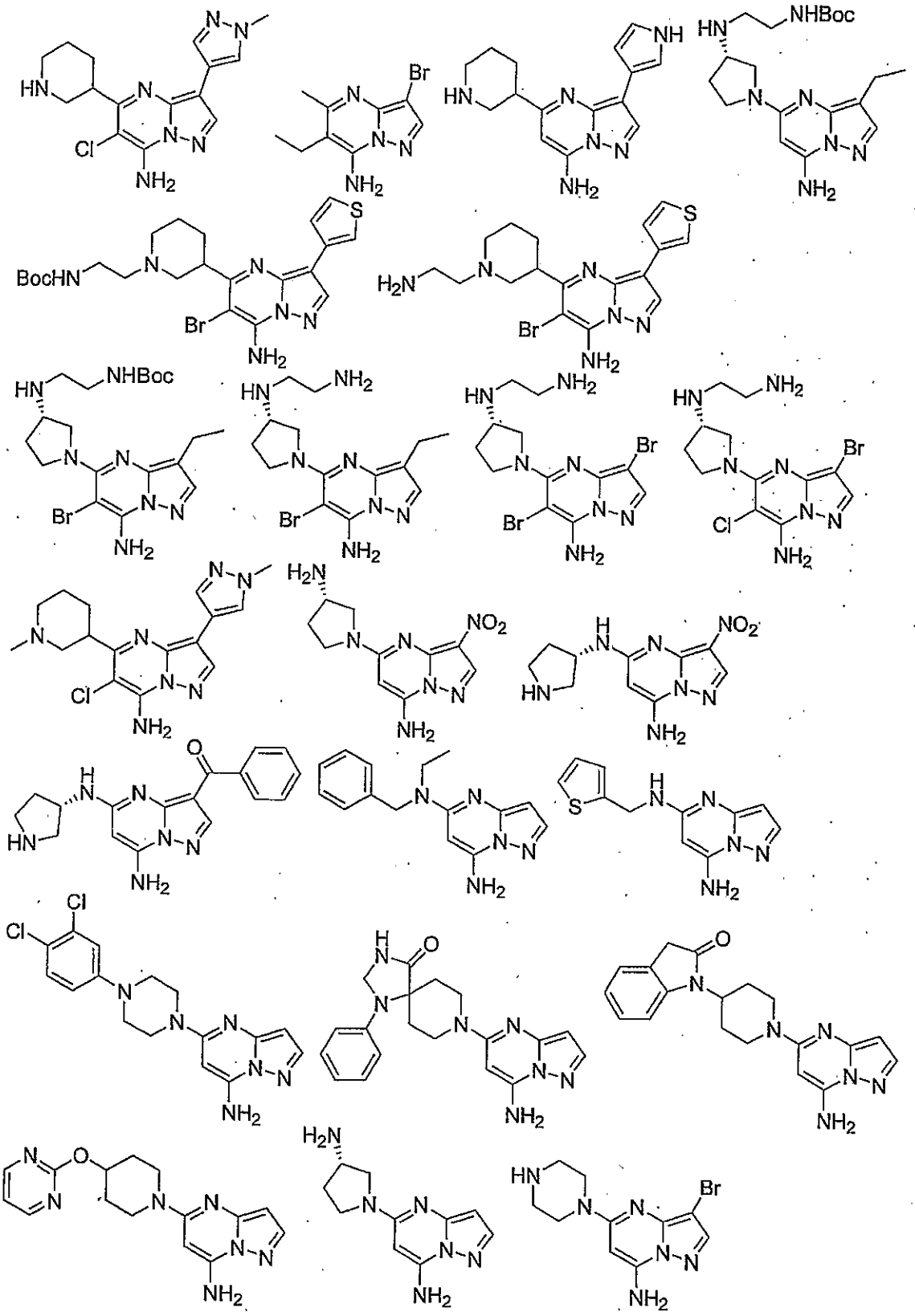


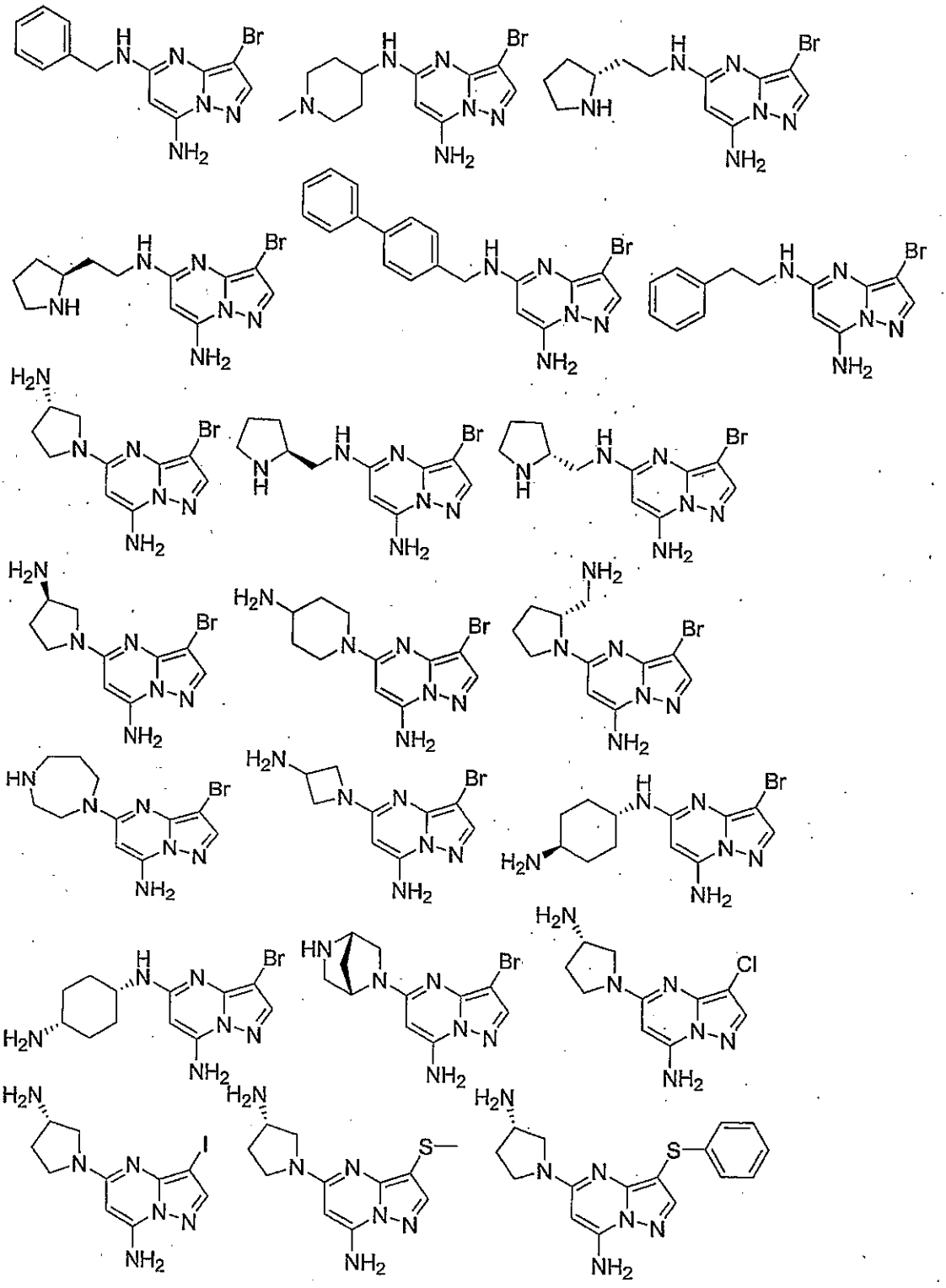


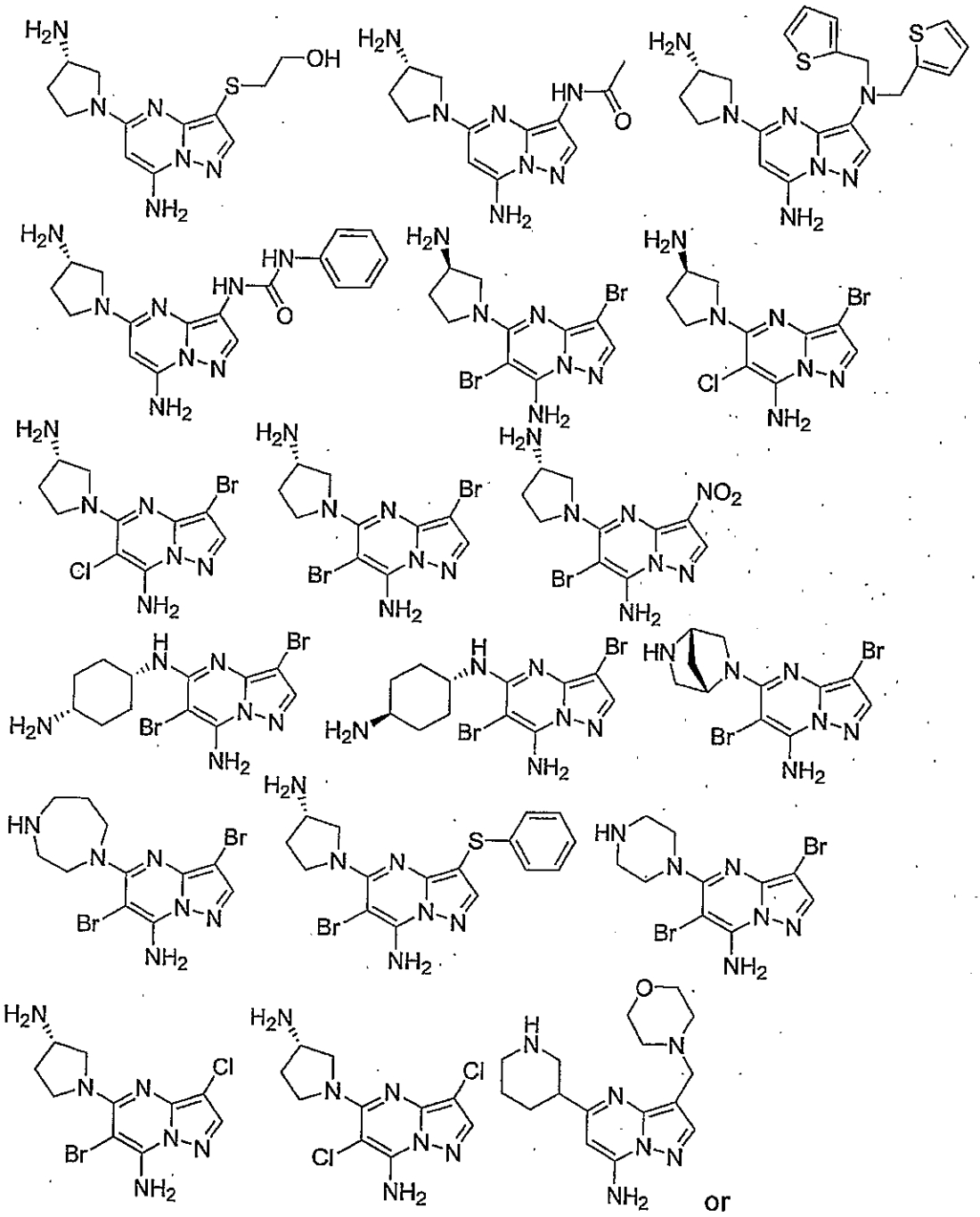


Still additionally disclosed are the following compounds:

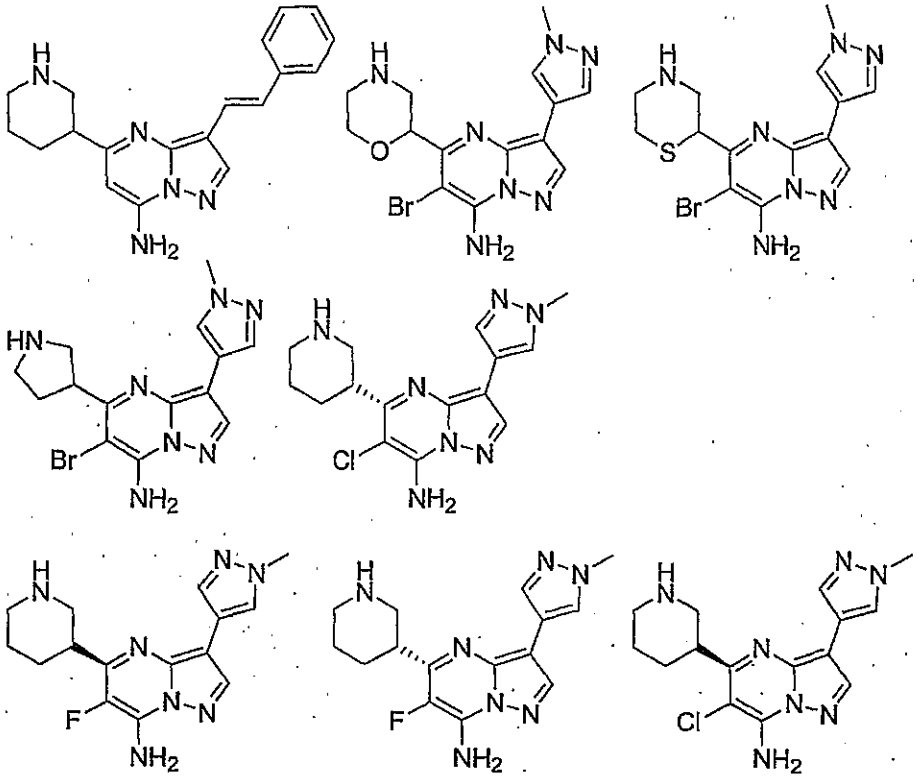








or



or a pharmaceutically acceptable salt or solvate thereof.

As used above, and throughout this disclosure, the following terms, unless otherwise indicated, shall be understood to have the following meanings:

"Patient" includes both human and animals.

"Mammal" means humans and other mammalian animals.

"Alkyl" means an aliphatic hydrocarbon group which may be straight or branched and comprising about 1 to about 20 carbon atoms in the chain. Preferred alkyl groups contain about 1 to about 12 carbon atoms in the chain. More preferred alkyl groups contain about 1 to about 6 carbon atoms in the chain. Branched means that one or more lower alkyl groups such as methyl, ethyl or propyl, are attached to a linear alkyl chain. "Lower alkyl" means a group having about 1 to about 6 carbon atoms in the chain which may be straight or branched. "Alkyl" may be unsubstituted or optionally substituted by one or more substituents which may be the same or different, each substituent being independently selected from the group consisting of halo, alkyl, aryl, cycloalkyl, cyano, hydroxy, alkoxy, alkylthio, amino, -NH(alkyl), -NH(cycloalkyl), -N(alkyl)$_2$,

carboxy and –C(O)O-alkyl. Non-limiting examples of suitable alkyl groups include methyl, ethyl, n-propyl, isopropyl and t-butyl.

"Alkenyl" means an aliphatic hydrocarbon group containing at least one carbon-carbon double bond and which may be straight or branched and comprising about 2 to about 15 carbon atoms in the chain. Preferred alkenyl groups have about 2 to about 12 carbon atoms in the chain; and more preferably about 2 to about 6 carbon atoms in the chain. Branched means that one or more lower alkyl groups such as methyl, ethyl or propyl, are attached to a linear alkenyl chain. "Lower alkenyl" means about 2 to about 6 carbon atoms in the chain which may be straight or branched. "Alkenyl" may be unsubstituted or optionally substituted by one or more substituents which may be the same or different, each substituent being independently selected from the group consisting of halo, alkyl. aryl, cycloalkyl, cyano, alkoxy and –S(alkyl). Non-limiting examples of suitable alkenyl groups include ethenyl, propenyl, n-butenyl, 3-methylbut-2-enyl, n-pentenyl, octenyl and decenyl.

"Alkylene" means a difunctional group obtained by removal of a hydrogen atom from an alkyl group that is defined above. Non-limiting examples of alkylene include methylene, ethylene and propylene.

"Alkynyl" means an aliphatic hydrocarbon group containing at least one carbon-carbon triple bond and which may be straight or branched and comprising about 2 to about 15 carbon atoms in the chain. Preferred alkynyl groups have about 2 to about 12 carbon atoms in the chain; and more preferably about 2 to about 4 carbon atoms in the chain. Branched means that one or more lower alkyl groups such as methyl, ethyl or propyl, are attached to a linear alkynyl chain. "Lower alkynyl" means about 2 to about 6 carbon atoms in the chain which may be straight or branched. Non-limiting examples of suitable alkynyl groups include ethynyl, propynyl, 2-butynyl and 3-methylbutynyl. "Alkynyl" may be unsubstituted or optionally substituted by one or more substituents which may be the same or different, each substituent being independently selected from the group consisting of alkyl, aryl and cycloalkyl.

"Aryl" means an aromatic monocyclic or multicyclic ring system comprising about 6 to about 14 carbon atoms, preferably about 6 to about 10

carbon atoms. The aryl group can be optionally substituted with one or more "ring system substituents" which may be the same or different, and are as defined herein. Non-limiting examples of suitable aryl groups include phenyl and naphthyl.

"Heteroaryl" means an aromatic monocyclic or multicyclic ring system comprising about 5 to about 14 ring atoms, preferably about 5 to about 10 ring atoms, in which one or more of the ring atoms is an element other than carbon, for example nitrogen, oxygen or sulfur, alone or in combination. Preferred heteroaryls contain about 5 to about 6 ring atoms. The "heteroaryl" can be optionally substituted by one or more "ring system substituents" which may be the same or different, and are as defined herein. The prefix aza, oxa or thia before the heteroaryl root name means that at least a nitrogen, oxygen or sulfur atom respectively, is present as a ring atom. A nitrogen atom of a heteroaryl can be optionally oxidized to the corresponding N-oxide. Non-limiting examples of suitable heteroaryls include pyridyl, pyrazinyl, furanyl, thienyl, pyrimidinyl, pyridone (including N-substituted pyridones), isoxazolyl, isothiazolyl, oxazolyl, thiazolyl, pyrazolyl, furazanyl, pyrrolyl, pyrazolyl, triazolyl, 1,2,4-thiadiazolyl, pyrazinyl, pyridazinyl, quinoxalinyl, phthalazinyl, oxindolyl, imidazo[1,2-a]pyridinyl, imidazo[2,1-b]thiazolyl, benzofurazanyl, indolyl, azaindolyl, benzimidazolyl, benzothienyl, quinolinyl, imidazolyl, thienopyridyl, quinazolinyl, thienopyrimidyl, pyrrolopyridyl, imidazopyridyl, isoquinolinyl, benzoazaindolyl, 1,2,4-triazinyl, benzothiazolyl and the like. The term "heteroaryl" also refers to partially saturated heteroaryl moieties such as, for example, tetrahydroisoquinolyl, tetrahydroquinolyl and the like.

"Aralkyl" or "arylalkyl" means an aryl-alkyl- group in which the aryl and alkyl are as previously described. Preferred aralkyls comprise a lower alkyl group. Non-limiting examples of suitable aralkyl groups include benzyl, 2-phenethyl and naphthalenylmethyl. The bond to the parent moiety is through the alkyl.

"Alkylaryl" means an alkyl-aryl- group in which the alkyl and aryl are as previously described. Preferred alkylaryls comprise a lower alkyl group. Non-

limiting example of a suitable alkylaryl group is tolyl. The bond to the parent moiety is through the aryl.

"Cycloalkyl" means a non-aromatic mono- or multicyclic ring system comprising about 3 to about 10 carbon atoms, preferably about 5 to about 10 carbon atoms. Preferred cycloalkyl rings contain about 5 to about 7 ring atoms. The cycloalkyl can be optionally substituted with one or more "ring system substituents" which may be the same or different, and are as defined above. Non-limiting examples of suitable monocyclic cycloalkyls include cyclopropyl, cyclopentyl, cyclohexyl, cycloheptyl and the like. Non-limiting examples of suitable multicyclic cycloalkyls include 1-decalinyl, norbornyl, adamantyl and the like.

"Cycloalkylalkyl" means a cycloalkyl moiety as defined above linked via an alkyl moiety (defined above) to a parent core. Non-limiting examples of suitable cycloalkylalkyls include cyclohexylmethyl, adamantylmethyl and the like.

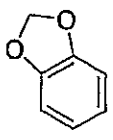
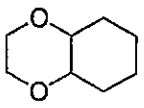
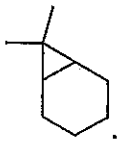
"Cycloalkenyl" means a non-aromatic mono or multicyclic ring system comprising about 3 to about 10 carbon atoms, preferably about 5 to about 10 carbon atoms which contains at least one carbon-carbon double bond. Preferred cycloalkenyl rings contain about 5 to about 7 ring atoms. The cycloalkenyl can be optionally substituted with one or more "ring system substituents" which may be the same or different, and are as defined above. Non-limiting examples of suitable monocyclic cycloalkenyls include cyclopentenyl, cyclohexenyl, cyclohepta-1,3-dienyl, and the like. Non-limiting example of a suitable multicyclic cycloalkenyl is norbornylenyl.

"Cycloalkenylalkyl" means a cycloalkenyl moiety as defined above linked via an alkyl moiety (defined above) to a parent core. Non-limiting examples of suitable cycloalkenylalkyls include cyclopentenylmethyl, cyclohexenylmethyl and the like.

"Halogen" means fluorine, chlorine, bromine, or iodine. Preferred are fluorine, chlorine and bromine.

"Ring system substituent" means a substituent attached to an aromatic or non-aromatic ring system which, for example, replaces an available hydrogen on the ring system. Ring system substituents may be the same or different, each

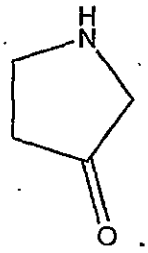
being independently selected from the group consisting of alkyl, alkenyl, alkynyl, aryl, heteroaryl, aralkyl, alkylaryl, heteroaralkyl, heteroarylalkenyl, heteroarylalkynyl, alkylheteroaryl, hydroxy, hydroxyalkyl, alkoxy, aryloxy, aralkoxy, acyl, aroyl, halo, nitro, cyano, carboxy, alkoxycarbonyl, aryloxycarbonyl, aralkoxycarbonyl, alkylsulfonyl, arylsulfonyl, heteroarylsulfonyl, alkylthio, arylthio, heteroarylthio, aralkylthio, heteroaralkylthio, cycloalkyl, heterocyclyl, -C(=N-CN)-NH_2, -C(=NH)-NH_2, -C(=NH)-NH(alkyl), Y_1Y_2N-, Y_1Y_2N-alkyl-, $Y_1Y_2NC(O)$-, $Y_1Y_2NSO_2$- and -$SO_2NY_1Y_2$, wherein Y_1 and Y_2 can be the same or different and are independently selected from the group consisting of hydrogen, alkyl, aryl, cycloalkyl, and aralkyl. "Ring system substituent" may also mean a single moiety which simultaneously replaces two available hydrogens on two adjacent carbon atoms (one H on each carbon) on a ring system. Examples of such moiety are methylene dioxy, ethylenedioxy, -$C(CH_3)_2$- and the like which form moieties such as, for example:



and


"Heteroarylalkyl" means a heteroaryl moiety as defined above linked via an alkyl moiety (defined above) to a parent core. Non-limiting examples of suitable heteroaryls include 2-pyridinylmethyl, quinolinylmethyl and the like.

"Heterocyclyl" means a non-aromatic saturated monocyclic or multicyclic ring system comprising about 3 to about 10 ring atoms, preferably about 5 to about 10 ring atoms, in which one or more of the atoms in the ring system is an element other than carbon, for example nitrogen, oxygen or sulfur, alone or in combination. There are no adjacent oxygen and/or sulfur atoms present in the ring system. Preferred heterocyclyls contain about 5 to about 6 ring atoms. The prefix aza, oxa or thia before the heterocyclyl root name means that at least a nitrogen, oxygen or sulfur atom respectively is present as a ring atom. Any –NH in a heterocyclyl ring may exist protected such as, for example, as an -N(Boc), -N(CBz), -N(Tos) group and the like; such protections are also considered part of this invention. The heterocyclyl can be optionally substituted by one or more "ring system substituents" which may be the same or different, and are as

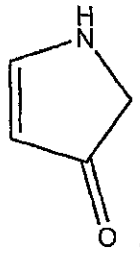
defined herein. The nitrogen or sulfur atom of the heterocyclyl can be optionally oxidized to the corresponding N-oxide, S-oxide or S,S-dioxide. Non-limiting examples of suitable monocyclic heterocyclyl rings include piperidyl, pyrrolidinyl, piperazinyl, morpholinyl, thiomorpholinyl, thiazolidinyl, 1,4-dioxanyl, tetrahydrofuranyl, tetrahydrothiophenyl, lactam, lactone, and the like. "Heterocyclyl" may also mean a single moiety (e.g., carbonyl) which simultaneously replaces two available hydrogens on the same carbon atom on a ring system. Example of such moiety is pyrrolidone:



"Heterocyclylalkyl" means a heterocyclyl moiety as defined above linked via an alkyl moiety (defined above) to a parent core. Non-limiting examples of suitable heterocyclylalkyls include piperidinylmethyl, piperazinylmethyl and the like.

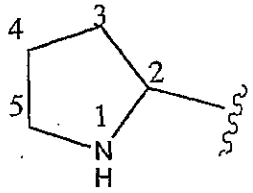
"Heterocyclenyl" means a non-aromatic monocyclic or multicyclic ring system comprising about 3 to about 10 ring atoms, preferably about 5 to about 10 ring atoms, in which one or more of the atoms in the ring system is an element other than carbon, for example nitrogen, oxygen or sulfur atom, alone or in combination, and which contains at least one carbon-carbon double bond or carbon-nitrogen double bond. There are no adjacent oxygen and/or sulfur atoms present in the ring system. Preferred heterocyclenyl rings contain about 5 to about 6 ring atoms. The prefix aza, oxa or thia before the heterocyclenyl root name means that at least a nitrogen, oxygen or sulfur atom respectively is present as a ring atom. The heterocyclenyl can be optionally substituted by one or more ring system substituents, wherein "ring system substituent" is as defined above. The nitrogen or sulfur atom of the heterocyclenyl can be optionally oxidized to the corresponding N-oxide, S-oxide or S,S-dioxide. Non-limiting examples of suitable heterocyclenyl groups include 1,2,3,4- tetrahydropyridinyl, 1,2-dihydropyridinyl, 1,4-dihydropyridinyl, 1,2,3,6-tetrahydropyridinyl, 1,4,5,6-

tetrahydropyrimidinyl, 2-pyrrolinyl, 3-pyrrolinyl, 2-imidazolinyl, 2-pyrazolinyl, dihydroimidazolyl, dihydrooxazolyl, dihydrooxadiazolyl, dihydrothiazolyl, 3,4-dihydro-2H-pyranyl, dihydrofuranyl, fluorodihydrofuranyl, 7-oxabicyclo[2.2.1]heptenyl, dihydrothiophenyl, dihydrothiopyranyl, and the like. "Heterocyclenyl" may also mean a single moiety (e.g., carbonyl) which simultaneously replaces two available hydrogens on the same carbon atom on a ring system. Example of such moiety is pyrrolidinone:



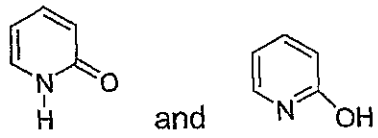
"Heterocyclenylalkyl" means a heterocyclenyl moiety as defined above linked via an alkyl moiety (defined above) to a parent core.

It should be noted that in hetero-atom containing ring systems of this invention, there are no hydroxyl groups on carbon atoms adjacent to a N, O or S, as well as there are no N or S groups on carbon adjacent to another heteroatom. Thus, for example, in the ring:



there is no -OH attached directly to carbons marked 2 and 5.

It should also be noted that tautomeric forms such as, for example, the moieties:



are considered equivalent in certain embodiments of this invention.

"Alkynylalkyl" means an alkynyl-alkyl- group in which the alkynyl and alkyl are as previously described. Preferred alkynylalkyls contain a lower alkynyl and

a lower alkyl group. The bond to the parent moiety is through the alkyl. Non-limiting examples of suitable alkynylalkyl groups include propargylmethyl.

"Heteroaralkyl" means a heteroaryl-alkyl- group in which the heteroaryl and alkyl are as previously described. Preferred heteroaralkyls contain a lower alkyl group. Non-limiting examples of suitable aralkyl groups include pyridylmethyl, and quinolin-3-ylmethyl. The bond to the parent moiety is through the alkyl.

"Hydroxyalkyl" means a HO-alkyl- group in which alkyl is as previously defined. Preferred hydroxyalkyls contain lower alkyl. Non-limiting examples of suitable hydroxyalkyl groups include hydroxymethyl and 2-hydroxyethyl.

"Acyl" means an H-C(O)-, alkyl-C(O)- or cycloalkyl-C(O)-, group in which the various groups are as previously described. The bond to the parent moiety is through the carbonyl. Preferred acyls contain a lower alkyl. Non-limiting examples of suitable acyl groups include formyl, acetyl and propanoyl.

"Aroyl" means an aryl-C(O)- group in which the aryl group is as previously described. The bond to the parent moiety is through the carbonyl. Non-limiting examples of suitable groups include benzoyl and 1- naphthoyl.

"Alkoxy" means an alkyl-O- group in which the alkyl group is as previously described. Non-limiting examples of suitable alkoxy groups include methoxy, ethoxy, n-propoxy, isopropoxy and n-butoxy. The bond to the parent moiety is through the ether oxygen.

"Aryloxy" means an aryl-O- group in which the aryl group is as previously described. Non-limiting examples of suitable aryloxy groups include phenoxy and naphthoxy. The bond to the parent moiety is through the ether oxygen.

"Aralkyloxy" means an aralkyl-O- group in which the aralkyl group is as previously described. Non-limiting examples of suitable aralkyloxy groups include benzyloxy and 1- or 2-naphthalenemethoxy. The bond to the parent moiety is through the ether oxygen.

"Alkylthio" means an alkyl-S- group in which the alkyl group is as previously described. Non-limiting examples of suitable alkylthio groups include methylthio and ethylthio. The bond to the parent moiety is through the sulfur.

"Arylthio" means an aryl-S- group in which the aryl group is as previously described. Non-limiting examples of suitable arylthio groups include phenylthio and naphthylthio. The bond to the parent moiety is through the sulfur.

"Aralkylthio" means an aralkyl-S- group in which the aralkyl group is as previously described. Non-limiting example of a suitable aralkylthio group is benzylthio. The bond to the parent moiety is through the sulfur.

"Alkoxycarbonyl" means an alkyl-O-CO- group. Non-limiting examples of suitable alkoxycarbonyl groups include methoxycarbonyl and ethoxycarbonyl. The bond to the parent moiety is through the carbonyl.

"Aryloxycarbonyl" means an aryl-O-C(O)- group. Non-limiting examples of suitable aryloxycarbonyl groups include phenoxycarbonyl and naphthoxycarbonyl. The bond to the parent moiety is through the carbonyl.

"Aralkoxycarbonyl" means an aralkyl-O-C(O)- group. Non-limiting example of a suitable aralkoxycarbonyl group is benzyloxycarbonyl. The bond to the parent moiety is through the carbonyl.

"Alkylsulfonyl" means an alkyl-S(O_2)- group. Preferred groups are those in which the alkyl group is lower alkyl. The bond to the parent moiety is through the sulfonyl.

"Arylsulfonyl" means an aryl-S(O_2)- group. The bond to the parent moiety is through the sulfonyl.

The term "substituted" means that one or more hydrogens on the designated atom is replaced with a selection from the indicated group, provided that the designated atom's normal valency under the existing circumstances is not exceeded, and that the substitution results in a stable compound. Combinations of substituents and/or variables are permissible only if such combinations result in stable compounds. By "stable compound' or "stable structure" is meant a compound that is sufficiently robust to survive isolation to a useful degree of purity from a reaction mixture, and formulation into an efficacious therapeutic agent.

The term "optionally substituted" means optional substitution with the specified groups, radicals or moieties.

The term "purified", "in purified form" or "in isolated and purified form" for a compound refers to the physical state of said compound after being isolated from a synthetic process or natural source or combination thereof. Thus, the term "purified", "in purified form" or "in isolated and purified form" for a compound refers to the physical state of said compound after being obtained from a purification process or processes described herein or well known to the skilled artisan, in sufficient purity to be characterizable by standard analytical techniques described herein or well known to the skilled artisan.

It should also be noted that any carbon as well as heteroatom with unsatisfied valences in the text, schemes, examples and Tables herein is assumed to have the sufficient number of hydrogen atom(s) to satisfy the valences.

When a functional group in a compound is termed "protected", this means that the group is in modified form to preclude undesired side reactions at the protected site when the compound is subjected to a reaction. Suitable protecting groups will be recognized by those with ordinary skill in the art as well as by reference to standard textbooks such as, for example, T. W. Greene *et al*, *Protective Groups in organic Synthesis* (1991), Wiley, New York.

When any variable (e.g., aryl, heterocycle, R^2, etc.) occurs more than one time in any constituent or in Formula III, its definition on each occurrence is independent of its definition at every other occurrence.

As used herein, the term "composition" is intended to encompass a product comprising the specified ingredients in the specified amounts, as well as any product which results, directly or indirectly, from combination of the specified ingredients in the specified amounts.

Prodrugs and solvates of the compounds of the invention are also contemplated herein. A discussion of prodrugs is provided in T. Higuchi and V. Stella, *Pro-drugs as Novel Delivery Systems* (1987) 14 of the A.C.S. Symposium Series, and in *Bioreversible Carriers in Drug Design*, (1987) Edward B. Roche, ed., American Pharmaceutical Association and Pergamon Press. The term "prodrug" means a compound (e.g, a drug precursor) that is transformed *in vivo* to yield a compound of Formula (III) or a pharmaceutically acceptable salt,

hydrate or solvate of the compound. The transformation may occur by various mechanisms (e.g., by metabolic or chemical processes), such as, for example, through hydrolysis in blood. A discussion of the use of prodrugs is provided by T. Higuchi and W. Stella, "Pro-drugs as Novel Delivery Systems," Vol. 14 of the A.C.S. Symposium Series, and in Bioreversible Carriers in Drug Design, ed. Edward B. Roche, American Pharmaceutical Association and Pergamon Press, 1987.

For example, if a compound of Formula (III) or a pharmaceutically acceptable salt, hydrate or solvate of the compound contains a carboxylic acid functional group, a prodrug can comprise an ester formed by the replacement of the hydrogen atom of the acid group with a group such as, for example, (C_1–C_8)alkyl, (C_2-C_{12})alkanoyloxymethyl, 1-(alkanoyloxy)ethyl having from 4 to 9 carbon atoms, 1-methyl-1-(alkanoyloxy)-ethyl having from 5 to 10 carbon atoms, alkoxycarbonyloxymethyl having from 3 to 6 carbon atoms, 1-(alkoxycarbonyloxy)ethyl having from 4 to 7 carbon atoms, 1-methyl-1-(alkoxycarbonyloxy)ethyl having from 5 to 8 carbon atoms, N-(alkoxycarbonyl)aminomethyl having from 3 to 9 carbon atoms, 1-(N-(alkoxycarbonyl)amino)ethyl having from 4 to 10 carbon atoms, 3-phthalidyl, 4-crotonolactonyl, gamma-butyrolacton-4-yl, di-N,N-(C_1-C_2)alkylamino(C_2-C_3)alkyl (such as β-dimethylaminoethyl), carbamoyl-(C_1-C_2)alkyl, N,N-di (C_1-C_2)alkylcarbamoyl-(C1-C2)alkyl and piperidino-, pyrrolidino- or morpholino(C_2-C_3)alkyl, and the like.

Similarly, if a compound of Formula (III) contains an alcohol functional group, a prodrug can be formed by the replacement of the hydrogen atom of the alcohol group with a group such as, for example, (C_1-C_6)alkanoyloxymethyl, 1-((C_1-C_6)alkanoyloxy)ethyl, 1-methyl-1-((C_1-C_6)alkanoyloxy)ethyl, (C_1-C_6)alkoxycarbonyloxymethyl, N-(C_1-C_6)alkoxycarbonylaminomethyl, succinoyl, (C_1-C_6)alkanoyl, α-amino(C_1-C_4)alkanyl, arylacyl and α-aminoacyl, or α-aminoacyl-α-aminoacyl, where each α-aminoacyl group is independently selected from the naturally occurring L-amino acids, $P(O)(OH)_2$, -$P(O)(O(C_1$-$C_6)alkyl)_2$ or glycosyl (the radical resulting from the removal of a hydroxyl group of the hemiacetal form of a carbohydrate), and the like.

If a compound of Formula (III) incorporates an amine functional group, a prodrug can be formed by the replacement of a hydrogen atom in the amine group with a group such as, for example, R-carbonyl, RO-carbonyl, NRR'-carbonyl where R and R' are each independently (C_1-C_{10})alkyl, (C_3-C_7) cycloalkyl, benzyl, or R-carbonyl is a natural α-aminoacyl or natural α-aminoacyl, —C(OH)C(O)OY^1 wherein Y^1 is H, (C_1-C_6)alkyl or benzyl, —$C(OY^2)Y^3$ wherein Y^2 is (C_1-C_4) alkyl and Y^3 is (C_1-C_6)alkyl, carboxy (C_1-C_6)alkyl, amino(C_1-C_4)alkyl or mono-N—or di-N,N-(C_1-C_6)alkylaminoalkyl, —$C(Y^4)Y^5$ wherein Y^4 is H or methyl and Y^5 is mono-N— or di-N,N-(C_1-C_6)alkylamino morpholino, piperidin-1-yl or pyrrolidin-1-yl, and the like.

One or more compounds of the invention may exist in unsolvated as well as solvated forms with pharmaceutically acceptable solvents such as water, ethanol, and the like, and it is intended that the invention embrace both solvated and unsolvated forms. "Solvate" means a physical association of a compound of this invention with one or more solvent molecules. This physical association involves varying degrees of ionic and covalent bonding, including hydrogen bonding. In certain instances the solvate will be capable of isolation, for example when one or more solvent molecules are incorporated in the crystal lattice of the crystalline solid. "Solvate" encompasses both solution-phase and isolatable solvates. Non-limiting examples of suitable solvates include ethanolates, methanolates, and the like. "Hydrate" is a solvate wherein the solvent molecule is H_2O.

One or more compounds of the invention may optionally be converted to a solvate. Preparation of solvates is generally known. Thus, for example, M. Caira *et al, J. Pharmaceutical Sci.*, 93(3), 601-611 (2004) describe the preparation of the solvates of the antifungal fluconazole in ethyl acetate as well as from water. Similar preparations of solvates, hemisolvate, hydrates and the like are described by E. C. van Tonder *et al, AAPS PharmSciTech.*, 5(1), article 12 (2004); and A. L. Bingham *et al, Chem. Commun.*, 603-604 (2001). A typical, non-limiting, process involves dissolving the inventive compound in desired amounts of the desired solvent (organic or water or mixtures thereof) at a higher than ambient temperature, and cooling the solution at a rate sufficient to form

crystals which are then isolated by standard methods. Analytical techniques such as, for example I. R. spectroscopy, show the presence of the solvent (or water) in the crystals as a solvate (or hydrate).

"Effective amount" or "therapeutically effective amount" is meant to describe an amount of compound or a composition of the present invention effective in inhibiting the above-noted diseases and thus producing the desired therapeutic, ameliorative, inhibitory or preventative effect.

The compounds of Formula III can form salts which are also within the scope of this invention. Reference to a compound of Formula III herein is understood to include reference to salts thereof, unless otherwise indicated. The term "salt(s)", as employed herein, denotes acidic salts formed with inorganic and/or organic acids, as well as basic salts formed with inorganic and/or organic bases. In addition, when a compound of Formula III contains both a basic moiety, such as, but not limited to a pyridine or imidazole, and an acidic moiety, such as, but not limited to a carboxylic acid, zwitterions ("inner salts") may be formed and are included within the term "salt(s)" as used herein. Pharmaceutically acceptable (i.e., non-toxic, physiologically acceptable) salts are preferred, although other salts are also useful. Salts of the compounds of the Formula III may be formed, for example, by reacting a compound of Formula III with an amount of acid or base, such as an equivalent amount, in a medium such as one in which the salt precipitates or in an aqueous medium followed by lyophilization.

Exemplary acid addition salts include acetates, ascorbates, benzoates, benzenesulfonates, bisulfates, borates, butyrates, citrates, camphorates, camphorsulfonates, fumarates, hydrochlorides, hydrobromides, hydroiodides, lactates, maleates, methanesulfonates, naphthalenesulfonates, nitrates, oxalates, phosphates, propionates, salicylates, succinates, sulfates, tartarates, thiocyanates, toluenesulfonates (also known as tosylates,) and the like. Additionally, acids which are generally considered suitable for the formation of pharmaceutically useful salts from basic pharmaceutical compounds are discussed, for example, by P. Stahl *et al*, Camille G. (eds.) *Handbook of Pharmaceutical Salts. Properties, Selection and Use.* (2002) Zurich: Wiley-VCH;

S. Berge *et al*, *Journal of Pharmaceutical Sciences* (1977) 66(1) 1-19; P. Gould, *International J. of Pharmaceutics* (1986) 33 201-217; Anderson *et al*, *The Practice of Medicinal Chemistry* (1996), Academic Press, New York; and in *The Orange Book* (Food & Drug Administration, Washington, D.C. on their website). These disclosures are incorporated herein by reference thereto.

Exemplary basic salts include ammonium salts, alkali metal salts such as sodium, lithium, and potassium salts, alkaline earth metal salts such as calcium and magnesium salts, salts with organic bases (for example, organic amines) such as dicyclohexylamines, t-butyl amines, and salts with amino acids such as arginine, lysine and the like. Basic nitrogen-containing groups may be quarternized with agents such as lower alkyl halides (e.g. methyl, ethyl, and butyl chlorides, bromides and iodides), dialkyl sulfates (e.g. dimethyl, diethyl, and dibutyl sulfates), long chain halides (e.g. decyl, lauryl, and stearyl chlorides, bromides and iodides), aralkyl halides (e.g. benzyl and phenethyl bromides), and others.

All such acid salts and base salts are intended to be pharmaceutically acceptable salts within the scope of the invention and all acid and base salts are considered equivalent to the free forms of the corresponding compounds for purposes of the invention.

Pharmaceutically acceptable esters of the present compounds include the following groups: (1) carboxylic acid esters obtained by esterification of the hydroxy groups, in which the non-carbonyl moiety of the carboxylic acid portion of the ester grouping is selected from straight or branched chain alkyl (for example, acetyl, n-propyl, t-butyl, or n-butyl), alkoxyalkyl (for example, methoxymethyl), aralkyl (for example, benzyl), aryloxyalkyl (for example, phenoxymethyl), aryl (for example, phenyl optionally substituted with, for example, halogen, C_{1-4}alkyl, or C_{1-4}alkoxy or amino); (2) sulfonate esters, such as alkyl- or aralkylsulfonyl (for example, methanesulfonyl); (3) amino acid esters (for example, L-valyl or L-isoleucyl); (4) phosphonate esters and (5) mono-, di- or triphosphate esters. The phosphate esters may be further esterified by, for example, a C_{1-20} alcohol or reactive derivative thereof, or by a 2,3-di (C_{6-24})acyl glycerol.

Compounds of Formula III, and salts, solvates, esters and prodrugs thereof, may exist in their tautomeric form (for example, as an amide or imino ether). All such tautomeric forms are contemplated herein as part of the present invention.

The compounds of Formula (III) may contain asymmetric or chiral centers, and, therefore, exist in different stereoisomeric forms. It is intended that all stereoisomeric forms of the compounds of Formula (III) as well as mixtures thereof, including racemic mixtures, form part of the present invention. In addition, the present invention embraces all geometric and positional isomers. For example, if a compound of Formula (III) incorporates a double bond or a fused ring, both the cis- and trans-forms, as well as mixtures, are embraced within the scope of the invention.

Diastereomeric mixtures can be separated into their individual diastereomers on the basis of their physical chemical differences by methods well known to those skilled in the art, such as, for example, by chromatography and/or fractional crystallization. Enantiomers can be separated by converting the enantiomeric mixture into a diastereomeric mixture by reaction with an appropriate optically active compound (e.g., chiral auxiliary such as a chiral alcohol or Mosher's acid chloride), separating the diastereomers and converting (e.g., hydrolyzing) the individual diastereomers to the corresponding pure enantiomers. Also, some of the compounds of Formula (III) may be atropisomers (e.g., substituted biaryls) and are considered as part of this invention. Enantiomers can also be separated by use of chiral HPLC column.

It is also possible that the compounds of Formula (III) may exist in different tautomeric forms, and all such forms are embraced within the scope of the invention. Also, for example, all keto-enol and imine-enamine forms of the compounds are included in the invention.

All stereoisomers (for example, geometric isomers, optical isomers and the like) of the present compounds (including those of the salts, solvates, esters and prodrugs of the compounds as well as the salts, solvates and esters of the prodrugs), such as those which may exist due to asymmetric carbons on various substituents, including enantiomeric forms (which may exist even in the absence

of asymmetric carbons), rotameric forms, atropisomers, and diastereomeric forms, are contemplated within the scope of this invention, as are positional isomers (such as, for example, 4-pyridyl and 3-pyridyl). (For example, if a compound of Formula (III) incorporates a double bond or a fused ring, both the cis- and trans-forms, as well as mixtures, are embraced within the scope of the invention. Also, for example, all keto-enol and imine-enamine forms of the compounds are included in the invention.) Individual stereoisomers of the compounds of the invention may, for example, be substantially free of other isomers, or may be admixed, for example, as racemates or with all other, or other selected, stereoisomers. The chiral centers of the present invention can have the S or R configuration as defined by the *IUPAC* 1974 Recommendations. The use of the terms "salt", "solvate", "ester", "prodrug" and the like, is intended to equally apply to the salt, solvate, ester and prodrug of enantiomers, stereoisomers, rotamers, tautomers, positional isomers, racemates or prodrugs of the inventive compounds.

The present invention also embraces isotopically-labelled compounds of the present invention which are identical to those recited herein, but for the fact that one or more atoms are replaced by an atom having an atomic mass or mass number different from the atomic mass or mass number usually found in nature. Examples of isotopes that can be incorporated into compounds of the invention include isotopes of hydrogen, carbon, nitrogen, oxygen, phosphorus, fluorine and chlorine, such as ^{2}H, ^{3}H, ^{13}C, ^{14}C, ^{15}N, ^{18}O, ^{17}O, ^{31}P, ^{32}P, ^{35}S, ^{18}F, and ^{36}Cl, respectively.

Certain isotopically-labelled compounds of Formula (III) (e.g., those labeled with ^{3}H and ^{14}C) are useful in compound and/or substrate tissue distribution assays. Tritiated (i.e., ^{3}H) and carbon-14 (i.e., ^{14}C) isotopes are particularly preferred for their ease of preparation and detectability. Further, substitution with heavier isotopes such as deuterium (i.e., ^{2}H) may afford certain therapeutic advantages resulting from greater metabolic stability (e.g., increased in vivo half-life or reduced dosage requirements) and hence may be preferred in some circumstances. Isotopically labelled compounds of Formula (III) can generally be prepared by following procedures analogous to those disclosed in

the Schemes and/or in the Examples hereinbelow, by substituting an appropriate isotopically labelled reagent for a non-isotopically labelled reagent.

Polymorphic forms of the compounds of Formula III, and of the salts, solvates, esters and prodrugs of the compounds of Formula III, are intended to be included in the present invention.

The term "pharmaceutical composition" is also intended to encompass both the bulk composition and individual dosage units comprised of more than one (e.g., two) pharmaceutically active agents such as, for example, a compound of the present invention and an additional agent selected from the lists of the additional agents described herein, along with any pharmaceutically inactive excipients. The bulk composition and each individual dosage unit can contain fixed amounts of the afore-said "more than one pharmaceutically active agents". The bulk composition is material that has not yet been formed into individual dosage units. An illustrative dosage unit is an oral dosage unit such as tablets, pills and the like. Similarly, the herein-described method of treating a patient by administering a pharmaceutical composition of the present invention is also intended to encompass the administration of the afore-said bulk composition and individual dosage units.

The compounds according to the invention have pharmacological properties; in particular, the compounds of Formula III can be inhibitors of protein kinases such as, for example, the inhibitors of the cyclin-dependent kinases, mitogen-activated protein kinase (MAPK/ERK), glycogen synthase kinase 3(GSK3beta) and the like. The cyclin dependent kinases (CDKs) include, for example, CDC2 (CDK1), CDK2, CDK4, CDK5, CDK6, CDK7 CDK8 and CDK9. The novel compounds of Formula III are expected to be useful in the therapy of proliferative diseases such as cancer, autoimmune diseases, viral diseases, fungal diseases, neurological/neurodegenerative disorders, arthritis, inflammation, anti-proliferative (e.g., ocular retinopathy), neuronal, alopecia and cardiovascular disease. Many of these diseases and disorders are listed in U.S. 6,413,974 cited earlier, the disclosure of which is incorporated herein.

More specifically, the compounds of Formula III can be useful in the treatment of a variety of cancers, including (but not limited to) the following:

carcinoma, including that of the bladder, breast, colon, kidney, liver, lung, including small cell lung cancer, non-small cell lung cancer, head and neck, esophagus, gall bladder, ovary, pancreas, stomach, cervix, thyroid, prostate, and skin, including squamous cell carcinoma;

hematopoietic tumors of lymphoid lineage, including leukemia, acute lymphocytic leukemia, acute lymphoblastic leukemia, B-cell lymphoma, T- cell lymphoma, Hodgkins lymphoma, non-Hodgkins lymphoma, hairy cell lymphoma, mantle cell lymphoma, myeloma, and Burkett's lymphoma;

hematopoietic tumors of myeloid lineage, including acute and chronic myelogenous leukemias, myelodysplastic syndrome and promyelocytic leukemia;

tumors of mesenchymal origin, including fibrosarcoma and rhabdomyosarcoma;

tumors of the central and peripheral nervous system, including astrocytoma, neuroblastoma, glioma and schwannomas; and

other tumors, including melanoma, seminoma, teratocarcinoma, osteosarcoma, xenoderoma pigmentosum, keratoctanthoma, thyroid follicular cancer and Kaposi's sarcoma.

Due to the key role of CDKs in the regulation of cellular proliferation in general, inhibitors could act as reversible cytostatic agents which may be useful in the treatment of any disease process which features abnormal cellular proliferation, e.g., benign prostate hyperplasia, familial adenomatosis polyposis, neuro-fibromatosis, atherosclerosis, pulmonary fibrosis, arthritis, psoriasis, glomerulonephritis, restenosis following angioplasty or vascular surgery, hypertrophic scar formation, inflammatory bowel disease, transplantation rejection, endotoxic shock, and fungal infections.

Compounds of Formula III may also be useful in the treatment of Alzheimer's disease, as suggested by the recent finding that CDK5 is involved in the phosphorylation of tau protein (*J. Biochem,* (1995) 117, 741-749).

Compounds of Formula III may induce or inhibit apoptosis. The apoptotic response is aberrant in a variety of human diseases. Compounds of Formula III, as modulators of apoptosis, will be useful in the treatment of cancer (including

but not limited to those types mentioned hereinabove), viral infections (including but not limited to herpevirus, poxvirus, Epstein- Barr virus, Sindbis virus and adenovirus), prevention of AIDS development in HIV-infected individuals, autoimmune diseases (including but not limited to systemic lupus, erythematosus, autoimmune mediated glomerulonephritis, rheumatoid arthritis, psoriasis, inflammatory bowel disease, and autoimmune diabetes mellitus), neurodegenerative disorders (including but not limited to Alzheimer's disease, AIDS-related dementia, Parkinson's disease, amyotrophic lateral sclerosis, retinitis pigmentosa, spinal muscular atrophy and cerebellar degeneration), myelodysplastic syndromes, aplastic anemia, ischemic injury associated with myocardial infarctions, stroke and reperfusion injury, arrhythmia, atherosclerosis, toxin-induced or alcohol related liver diseases, hematological diseases (including but not limited to chronic anemia and aplastic anemia), degenerative diseases of the musculoskeletal system (including but not limited to osteoporosis and arthritis) aspirin-sensitive rhinosinusitis, cystic fibrosis, multiple sclerosis, kidney diseases and cancer pain.

Compounds of Formula III, as inhibitors of the CDKs, can modulate the level of cellular RNA and DNA synthesis. These agents would therefore be useful in the treatment of viral infections (including but not limited to HIV, human papilloma virus, herpesvirus, poxvirus, Epstein-Barr virus, Sindbis virus and adenovirus).

Compounds of Formula III may also be useful in the chemoprevention of cancer. Chemoprevention is defined as inhibiting the development of invasive cancer by either blocking the initiating mutagenic event or by blocking the progression of pre-malignant cells that have already suffered an insult or inhibiting tumor relapse.

Compounds of Formula III may also be useful in inhibiting tumor angiogenesis and metastasis.

Compounds of Formula III may also act as inhibitors of other protein kinases, e.g., protein kinase C, her2, raf 1, MEK1, MAP kinase, EGF receptor, PDGF receptor, IGF receptor, PI3 kinase, wee1 kinase, Src, Abl and thus be effective in the treatment of diseases associated with other protein kinases.

Another aspect of this invention is a method of treating a mammal (*e.g.*, human) having a disease or condition associated with the CDKs by administering a therapeutically effective amount of at least one compound of Formula III, or a pharmaceutically acceptable salt or solvate of said compound to the mammal.

A preferred dosage is about 0.001 to 500 mg/kg of body weight/day of the compound of Formula III. An especially preferred dosage is about 0.01 to 25 mg/kg of body weight/day of a compound of Formula III, or a pharmaceutically acceptable salt or solvate of said compound.

The compounds of this invention may also be useful in combination (administered together or sequentially) with one or more of anti-cancer treatments such as radiation therapy, and/or one or more anti-cancer agents selected from the group consisting of cytostatic agents, cytotoxic agents (such as for example, but not limited to, DNA interactive agents (such as cisplatin or doxorubicin)); taxanes (e.g. taxotere, taxol); topoisomerase II inhibitors (such as etoposide); topoisomerase I inhibitors (such as irinotecan (or CPT-11), camptostar, or topotecan); tubulin interacting agents (such as paclitaxel, docetaxel or the epothilones); hormonal agents (such as tamoxifen); thymidilate synthase inhibitors (such as 5-fluorouracil); anti-metabolites (such as methoxtrexate); alkylating agents (such as temozolomide (TEMODAR™ from Schering-Plough Corporation, Kenilworth, New Jersey), cyclophosphamide); Farnesyl protein transferase inhibitors (such as, SARASAR™(4-[2-[4-[(11R)-3,10-dibromo-8-chloro-6,11-dihydro-5H-benzo[5,6]cyclohepta[1,2-b]pyridin-11-yl-]-1-piperidinyl]-2-oxoehtyl]-1-piperidinecarboxamide, or SCH 66336 from Schering-Plough Corporation, Kenilworth, New Jersey), tipifarnib (Zarnestra® or R115777 from Janssen Pharmaceuticals), L778,123 (a farnesyl protein transferase inhibitor from Merck & Company, Whitehouse Station, New Jersey), BMS 214662 (a farnesyl protein transferase inhibitor from Bristol-Myers Squibb Pharmaceuticals, Princeton, New Jersey); signal transduction inhibitors (such as, Iressa (from Astra Zeneca Pharmaceuticals, England), Tarceva (EGFR kinase inhibitors), antibodies to EGFR (e.g., C225), GLEEVEC™ (C-abl kinase inhibitor from Novartis Pharmaceuticals, East Hanover, New Jersey); interferons such as,

for example, intron (from Schering-Plough Corporation), Peg-Intron (from Schering-Plough Corporation); hormonal therapy combinations; aromatase combinations; ara-C, adriamycin, cytoxan, and gemcitabine.

Other anti-cancer (also known as anti-neoplastic) agents include but are not limited to Uracil mustard, Chlormethine, Ifosfamide, Melphalan, Chlorambucil, Pipobroman, Triethylenemelamine, Triethylenethiophosphoramine, Busulfan, Carmustine, Lomustine, Streptozocin, Dacarbazine, Floxuridine, Cytarabine, 6-Mercaptopurine, 6-Thioguanine, Fludarabine phosphate, oxaliplatin, leucovirin, oxaliplatin (ELOXATIN™ from Sanofi-Synthelabo Pharmaeuticals, France), Pentostatine, Vinblastine, Vincristine, Vindesine, Bleomycin, Dactinomycin, Daunorubicin, Doxorubicin, Epirubicin, Idarubicin, Mithramycin, Deoxycoformycin, Mitomycin-C, L-Asparaginase, Teniposide 17α-Ethinylestradiol, Diethylstilbestrol, Testosterone, Prednisone, Fluoxymesterone, Dromostanolone propionate, Testolactone, Megestrolacetate, Methylprednisolone, Methyltestosterone, Prednisolone, Triamcinolone, Chlorotrianisene, Hydroxyprogesterone, Aminoglutethimide, Estramustine, Medroxyprogesteroneacetate, Leuprolide, Flutamide, Toremifene, goserelin, Cisplatin, Carboplatin, Hydroxyurea, Amsacrine, Procarbazine, Mitotane, Mitoxantrone, Levamisole, Navelbene, Anastrazole, Letrazole, Capecitabine, Reloxafine, Droloxafine, Hexamethylmelamine, Avastin, herceptin, Bexxar, Velcade, Zevalin, Trisenox, Xeloda, Vinorelbine, Porfimer, Erbitux, Liposomal, Thiotepa, Altretamine, Melphalan, Trastuzumab, Lerozole, Fulvestrant, Exemestane, Fulvestrant, Ifosfomide, Rituximab, C225 (or Cetuximab from Merck KGaA, Darmstadt, Germany), and Campath.

The compounds of this invention may specifically be useful in combination (administered together, concurrently or sequentially) with temozolomide and/or radiation therapy.

If formulated as a fixed dose, such combination products employ the compounds of this invention within the dosage range described herein and the other pharmaceutically active agent or treatment within its dosage range. For example, the CDC2 inhibitor olomucine has been found to act synergistically with

known cytotoxic agents in inducing apoptosis (*J. Cell Sci.*, (1995) 108, 2897. Compounds of Formula III may also be administered sequentially with known anticancer or cytotoxic agents when a combination formulation is inappropriate. The invention is not limited in the sequence of administration; compounds of Formula III may be administered either prior to or after administration of the known anticancer or cytotoxic agent. For example, the cytotoxic activity of the cyclin-dependent kinase inhibitor flavopiridol is affected by the sequence of administration with anticancer agents. *Cancer Research*, (1997) 57, 3375. Such techniques are within the skills of persons skilled in the art as well as attending physicians.

Accordingly, in an aspect, this invention includes combinations comprising an amount of at least one compound of Formula III, or a pharmaceutically acceptable salt or solvate thereof, and an amount of one or more anti-cancer treatments and anti-cancer agents listed above wherein the amounts of the compounds/ treatments result in desired therapeutic effect.

The pharmacological properties of the compounds of this invention may be confirmed by a number of pharmacological assays. The exemplified pharmacological assays which are described later have been carried out with the compounds according to the invention and their salts.

This invention is also directed to pharmaceutical compositions which comprise at least one compound of Formula III, or a pharmaceutically acceptable salt or solvate of said compound and at least one pharmaceutically acceptable carrier.

For preparing pharmaceutical compositions from the compounds described by this invention, inert, pharmaceutically acceptable carriers can be either solid or liquid. Solid form preparations include powders, tablets, dispersible granules, capsules, cachets and suppositories. The powders and tablets may be comprised of from about 5 to about 95 percent active ingredient. Suitable solid carriers are known in the art, e.g., magnesium carbonate, magnesium stearate, talc, sugar or lactose. Tablets, powders, cachets and capsules can be used as solid dosage forms suitable for oral administration. Examples of pharmaceutically acceptable carriers and methods of manufacture

for various compositions may be found in A. Gennaro (ed.), *Remington's Pharmaceutical Sciences,* 18th Edition, (1990), Mack Publishing Co., Easton, Pennsylvania.

Liquid form preparations include solutions, suspensions and emulsions. As an example may be mentioned water or water-propylene glycol solutions for parenteral injection or addition of sweeteners and opacifiers for oral solutions, suspensions and emulsions. Liquid form preparations may also include solutions for intranasal administration.

Aerosol preparations suitable for inhalation may include solutions and solids in powder form, which may be in combination with a pharmaceutically acceptable carrier, such as an inert compressed gas, e.g. nitrogen.

Also included are solid form preparations that are intended to be converted, shortly before use, to liquid form preparations for either oral or parenteral administration. Such liquid forms include solutions, suspensions and emulsions.

The compounds of the invention may also be deliverable transdermally. The transdermal compositions can take the form of creams, lotions, aerosols and/or emulsions and can be included in a transdermal patch of the matrix or reservoir type as are conventional in the art for this purpose.

The compounds of this invention may also be delivered subcutaneously.

Preferably the compound is administered orally or intravenously.

Preferably, the pharmaceutical preparation is in a unit dosage form. In such form, the preparation is subdivided into suitably sized unit doses containing appropriate quantities of the active component, e.g., an effective amount to achieve the desired purpose.

The quantity of active compound in a unit dose of preparation may be varied or adjusted from about 1 mg to about 100 mg, preferably from about 1 mg to about 50 mg, more preferably from about 1 mg to about 25 mg, according to the particular application.

The actual dosage employed may be varied depending upon the requirements of the patient and the severity of the condition being treated. Determination of the proper dosage regimen for a particular situation is within the

skill of the art. For convenience, the total daily dosage may be divided and administered in portions during the day as required.

The amount and frequency of administration of the compounds of the invention and/or the pharmaceutically acceptable salts thereof will be regulated according to the judgment of the attending clinician considering such factors as age, condition and size of the patient as well as severity of the symptoms being treated. A typical recommended daily dosage regimen for oral administration can range from about 1 mg/day to about 500 mg/day, preferably 1 mg/day to 200 mg/day, in two to four divided doses.

Another aspect of this invention is a kit comprising a therapeutically effective amount of at least one compound of Formula III, or a pharmaceutically acceptable salt or solvate of said compound and a pharmaceutically acceptable carrier, vehicle or diluent.

Yet another aspect of this invention is a kit comprising an amount of at least one compound of Formula III, or a pharmaceutically acceptable salt or solvate of said compound and an amount of at least one anticancer therapy and/or anti-cancer agent listed above, wherein the amounts of the two or more ingredients result in desired therapeutic effect.

The invention disclosed herein is exemplified by the following preparations and examples which should not be construed to limit the scope of the disclosure. Alternative mechanistic pathways and analogous structures will be apparent to those skilled in the art.

Where NMR data are presented, ^1H spectra were obtained on either a Varian VXR-200 (200 MHz, ^1H), Varian Gemini-300 (300 MHz) or XL-400 (400 MHz) and are reported as ppm down field from Me_4Si with number of protons, multiplicities, and coupling constants in Hertz indicated parenthetically. Where LC/MS data are presented, analyses was performed using an Applied Biosystems API-100 mass spectrometer and Shimadzu SCL-10A LC column: Altech platinum C18, 3 micron, 33mm x 7mm ID; gradient flow: 0 min – 10% CH_3CN, 5 min – 95% CH_3CN, 7 min – 95% CH_3CN, 7.5 min – 10% CH_3CN, 9 min – stop. The retention time and observed parent ion are given.

The following solvents and reagents may be referred to by their abbreviations in parenthesis:

Thin layer chromatography: TLC

dichloromethane: CH_2Cl_2

ethyl acetate: AcOEt or EtOAc

methanol: MeOH

trifluoroacetate: TFA

triethylamine: Et_3N or TEA

butoxycarbonyl: n-Boc or Boc

nuclear magnetic resonance spectroscopy: NMR

liquid chromatography mass spectrometry: LCMS

high resolution mass spectrometry: HRMS

milliliters: mL

millimoles: mmol

microliters: µl

grams: g

milligrams: mg

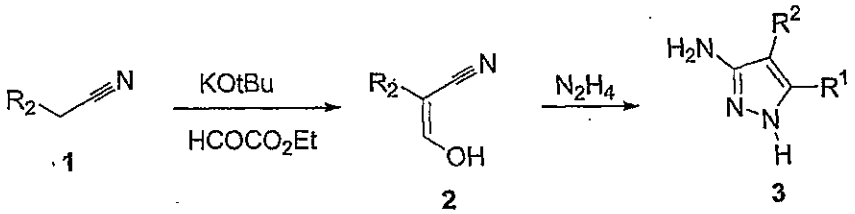
room temperature or rt (ambient): about 25°C.

dimethoxyethane: DME

EXAMPLES

In general, the compounds described in this invention can be prepared through the general routes described below in Scheme 1. Treatment of the

Scheme 1

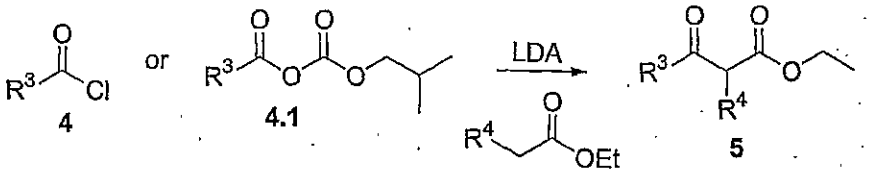


1 2 3

starting nitrile with potassium t-butoxide and ethyl formate gives rise to the intermediate enol 2 which upon treatment with hydrazine gives the desired substituted 3-aminopyrazole. Condensation of compounds of type 3 with the appropriately functionalized keto ester of type 5 gives rise to the pyridones 6 as

shown in Scheme 3. The keto esters used in this general route are either commercially available or can be made as illustrated in Scheme 2.

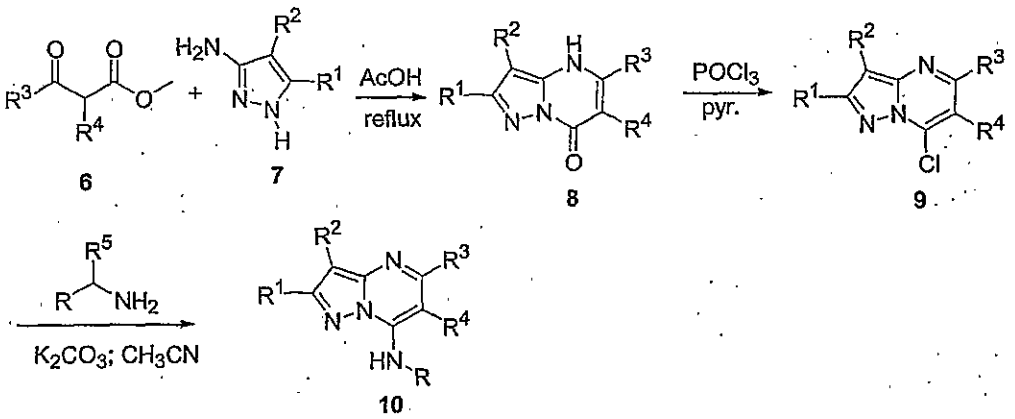
Scheme 2



The chlorides of type 9 can be prepared by treatment of the pyridones 8 with $POCl_3$. When R^2 is equal to H, substitution in this position is possible on the compounds of type 9 by electrophilic halogenation, acylation, and various other electrophilic aromatic substitutions.

Introduction of the N7-amino functionality can be accomplished through displacement of the chloride of compounds of type 9 by reaction with the appropriate amine as shown in Scheme 3.

Scheme 3



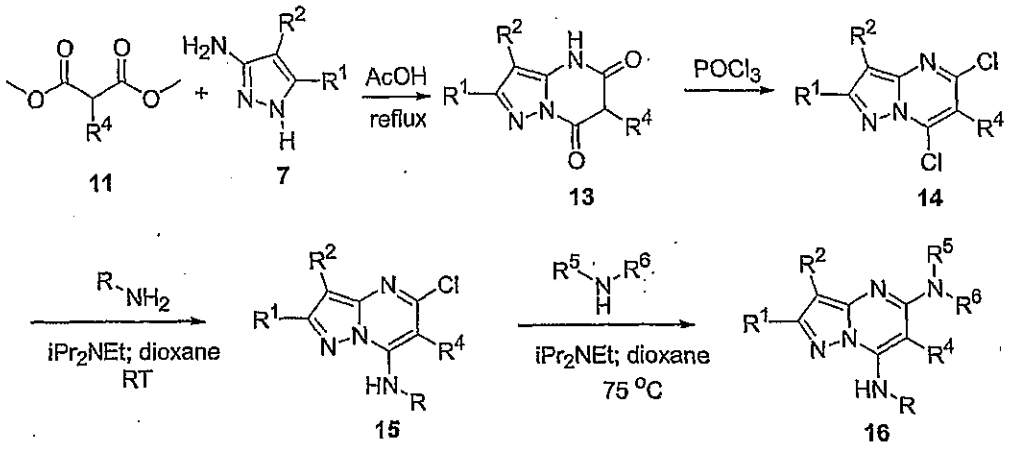
Condensation of compounds of type 7 with the appropriately functionalized malonate ester of type 11 gives rise to the pyridones 13 as shown in Scheme 4.

The chlorides of type 14 can be prepared by treatment of the pyridones 13 with $POCl_3$. When R^2 is H, substitution in this position is possible on compounds of type 9 by electrophilic halogenation, acylation, and various other electrophilic aromatic substitutions.

Incorporation of the N7-amino functionality can be accomplished through regioselective displacement of the chloride of compounds of type 14.

Incorporation of the N5-amino functionality by addition of an appropriate amine at higher temperature.

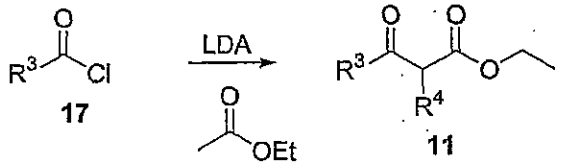
Scheme 4



11 7 13 14 15 16

Alternatively, condensations of the aminopyrazoles of type 7 with an appropriately functionalize keto ester as prepared in Scheme 5, leads to compounds of type 13 as shown in Scheme 4.

Scheme 5



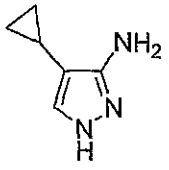
17 11

The chlorides of type 14 can be prepared by treatment of the pyridones 13 with $POCl_3$. When R^2 is equal to H, substitution in this position is possible on compounds of type 14 by electrophilic halogenation, acylation, and various other electrophilic aromatic substitutions.

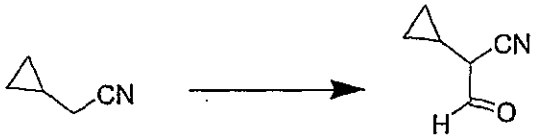
Incorporation of the N7-amino functionality can be accomplished through displacement of the chloride of compounds of type 15.

Preparative Examples:

PREPARATIVE EXAMPLE 1:

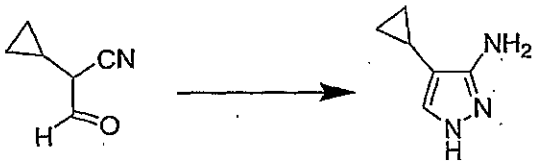


Step A:



A procedure in German patent DE 19834047 A1, p 19 was followed. To a solution of KOtBu (6.17 g, 0.055 mol) in anhydrous THF (40 mL) was added, dropwise, a solution of cyclopropylacetonitrile (2.0 g, 0.025 mol) and ethyl formate (4.07 g, 0.055 mol) in anhydrous THF (4 mL). A precipitate formed immediately. This mixture was stirred for 12 hr. It was concentrated under vacuum and the residue stirred with Et_2O (50 mL). The resulting residue was decanted and washed with Et_2O (2 x 50 mL) and Et_2O removed from the residue under vacuum. The residue was dissolved in cold H_2O (20 mL) and pH adjusted to 4 – 5 with 12 N HCl. The mixture was extracted with CH_2Cl_2 (2 x 50 mL). The organic layers were combined, dried over $MgSO_4$ and concentrated under vacuum to give the aldehyde as a tan liquid.

Step B:

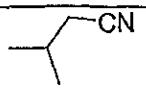
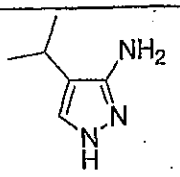
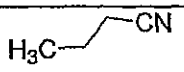
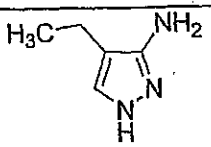
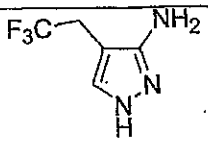


The product from Preparative Example 1, Step A (2.12 g, 0.0195 mol), $NH_2NH_2 \cdot H_2O$ (1.95 g, 0.039 mol) and 1.8 g (0.029 mole) of glacial CH_3CO_2H (1.8 g, 0.029 mol) were dissolved in EtOH (10 mL). It was refluxed for 6 hr and concentrated under vacuum. The residue was slurried in CH_2Cl_2 (150 mL) and the pH adjusted to 9 with 1N NaOH. The organic layer was washed with brine, dried over $MgSO_4$ and concentrated under vacuum to give the product as a waxy orange solid.

PREPARATIVE EXAMPLES 2-4:

By essentially the same procedure set forth in Preparative Example 1, only substituting the nitrile shown in Column 2 of Table 2, the compounds in Column 3 of Table 2 were prepared:

TABLE 2

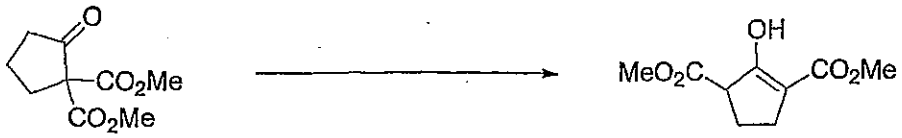
Prep. Ex.	Column 2	Column 3
2		
3		
3.10		

PREPARATIVE EXAMPLE 4



2-Carbomethoxycyclopentanone (6.6 ml, 0.05 mol) in THF (15 ml) was added dropwise to a vigorously stirred suspension of NaH (60% in mineral oil, 4 g, 0.1 mol) in THF (100 ml) at 0 – 10 °C. When bubbling ceased, the reaction mixture was treated at the same temperature with ClCOOMe (7.8 ml, 0.1 mol) in THF (15 ml). The resulted off-white suspension was stirred for 30 minutes at room temperature and 30 minutes under reflux. The reaction was monitored by TLC for disappearance of starting material. The reaction mixture was quenched with water carefully and partitioned between ethyl acetate and saturated solution of ammonium chloride in a funnel. Shaken and separated, the organic layer was washed with brine and dried over anhydrous sodium sulfate. Solvents were removed, and the residue was purified by flash chromatography, eluted with 5% and then 10% ethyl acetate in hexane. 9.4 g colorless oil was obtained with 94% yield. ^{1}H NMR ($CDCl_3$) δ 3.90(s, 3H), 3.73(s, 3H), 2.65(m, 4H), 1.98(m, 2H).

PREPARATIVE EXAMPLE 5



To lithium diisopropylamide solution in THF (2.0 N, 0.04 mol) at –65 °C, was added dropwise 2,2-dicarbomethoxycyclopentanone (4 g, 0.02 mol) in THF (60 ml). The resulted reaction mixture was stirred at the same temperature before adding methyl chloroformate (1.54 ml, 0.02 mol). Reaction mixture stirred for an hour and poured into saturated ammonium chloride solution with some ice. This solution was extracted three times with ether, and the combined ethearal layers were dried over sodium sulfate. Solvents were removed in vacuo, and the residue was purified by flash chromatography, eluted with 30% increased to 50% ethyl acetate in hexane. 2.3 g yellowish oil was obtained with 58% yield. ^{1}H NMR ($CDCl_3$) δ 3.77(s, 6H), 3.32(t, 1H), 3.60-3.10(m, 4H).

PREPARATIVE EXAMPLE 6:

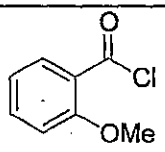
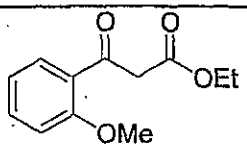
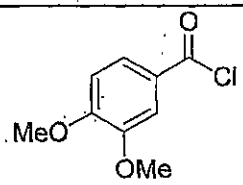
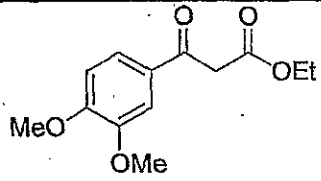
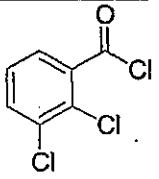
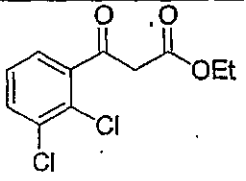
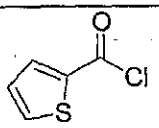
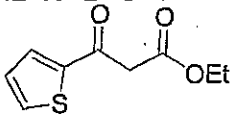
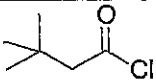
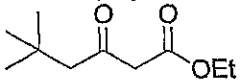
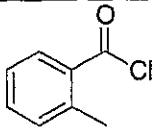
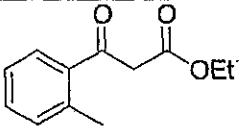
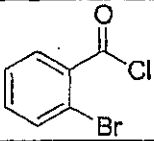
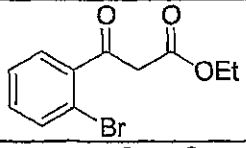
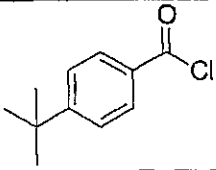
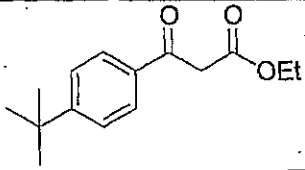


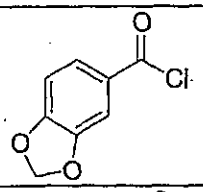
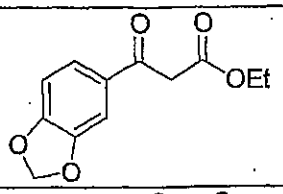
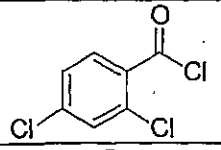
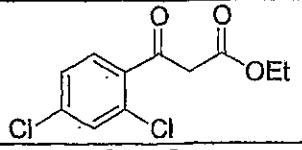
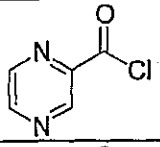
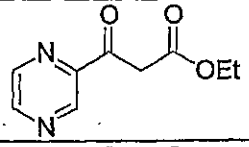
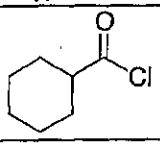
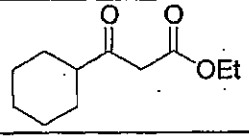
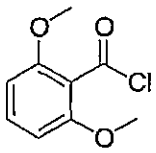
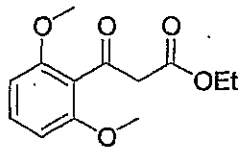
The reactions were done as outlined in (K. O. Olsen, *J. Org. Chem.*, (1987) 52, 4531 – 4536). Thus, to a stirred solution of lithium diisopropylamide in THF at -65 to –70 C was added freshly distilled ethyl acetate, dropwise. The resulting solution was stirred for 30 min and the acid chloride was added as a solution in THF. The reaction mixture was stirred at –65 to –70° C for 30 min and then terminated by the addition of 1 N HCl solution. The resulting two-phased mixture was allowed to warm to ambient temperature. The resulting mixture was diluted with EtOAc (100 mL) the organic layer was collected. The aqueous layer was extracted with EtOAc (100 mL). The organic layers were combined, washed with brine, dried (Na_2SO_4), and concentrated *in vacuo* to give the crude β-keto esters, which were used in the subsequent condensations.

PREPARATIVE EXAMPLES 7-19:

By essentially the same procedure set forth in Preparative Example 6 only substituting the acid chlorides shown in Column 2 of Table 3, the β-keto esters shown in Column 3 of Table 3 were prepared:

TABLE 3

Prep. Ex.	Column 2	Column 3	DATA
7			LCMS: MH^+ = 223
8			LCMS: MH^+ = 253
9			LCMS: MH^+ = 261
10			MH^+ = 199
11			
12			
13			LCMS: MH^+ = 271
14			Yield = quant MH^+ =249

15			Yield = quant MH^+ =237
16			Yield = quant MH^+ = 262
17			Yield = 48 MH^+ = 195
18			Yield = 99 MH^+ =199
19			Yield=77% ^{1}H NMR ($CDCl_3$) δ 7.42(t, 1H), 6.68(d, 2H), 4.29(q, 2H), 3.97(d, 2H), 3.95(s, 3H), 1.38(t, 3H).

PREPARATIVE EXAMPLE 20:

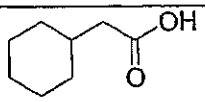
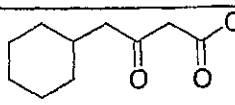
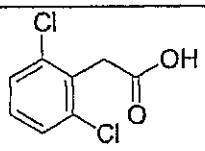
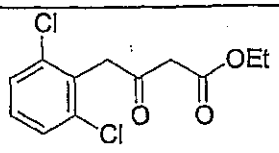
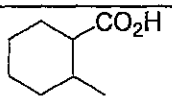
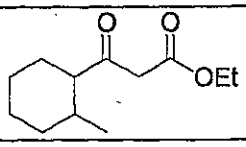
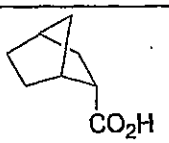
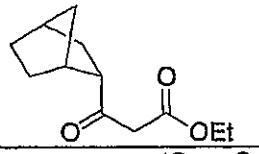
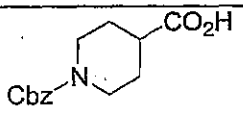
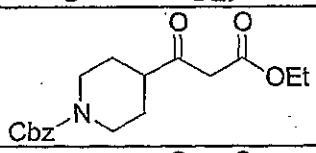
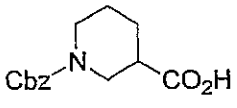
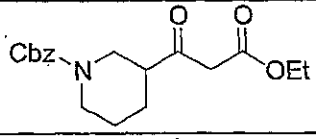
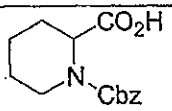
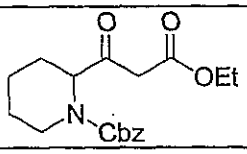
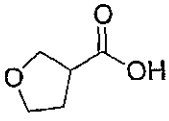
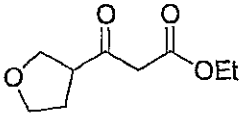


To a solution of the acid in THF was added Et_3N, followed by isobutyl chloroformate at –20 to –30°C. After the mixture was stirred for 30 min at –20 to –30°C, triethylamine hydrochloride was filtered off under argon, and the filtrate was added to the LDA-EtOAc reaction mixture (prepared as outlined in Method A) at –65 to –70°C. After addition of 1 N HCl, followed by routine workup of the reaction mixture and evaporation of the solvents, the crude β-keto esters were isolated. The crude material was used in the subsequent condensations.

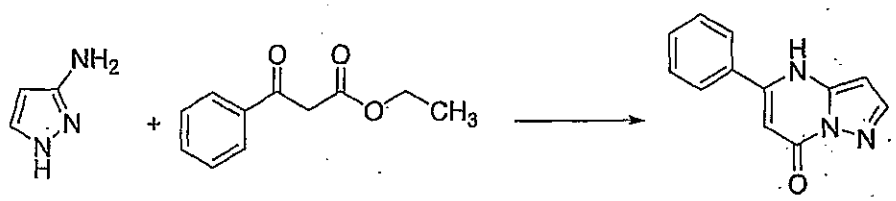
PREPARATIVE EXAMPLES 21 - 28:

By essentially the same conditions set forth in Preparative Example 20 only substituting the carboxylic acid shown in Column 2 of Table 4, the compounds shown in Column 3 of Table 4 were prepared:

TABLE 4

Prep. Ex.	Column 2	Column 3	CMPD
21			Yield = 99% MH^+ = 213
22			Yield = 70% MH^+ = 275
23			Yield = quant MH^+ =213
24			Yield = quant MH^+ =211
25			Yield = 99 MH^+ = 334
26			Yield = 99 MH^+ = 334
27			Yield = 99 MH^+ = 334
28			Yield=77% ^1H NMR ($CDCl_3$) δ 4.21(q, 2H), 3.95(d, 2H), 3.93-3.79(m, 4H), 3.52(s, 2H), 2.65(m, 1H), 1.25(t, 3H), 1.23-1.2(m, 2H).

PREPARATIVE EXAMPLE 29:

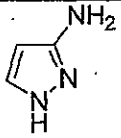
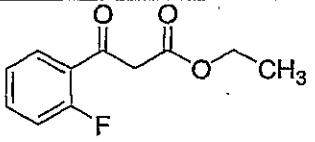
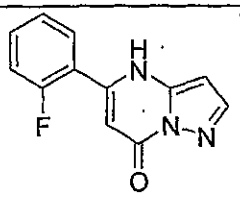
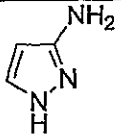
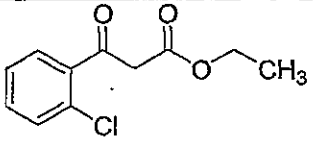
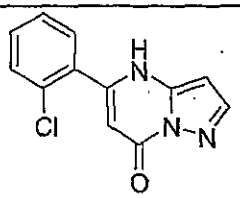
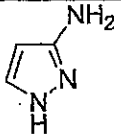
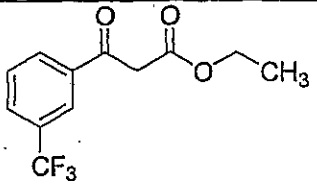
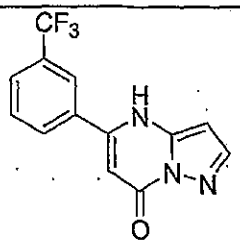


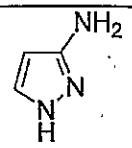
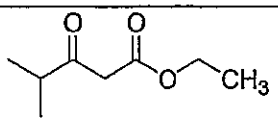
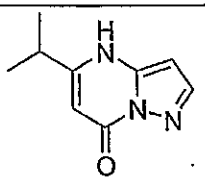
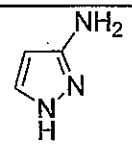
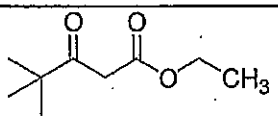
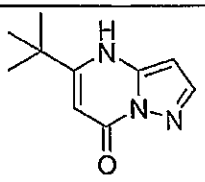
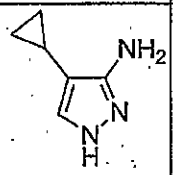
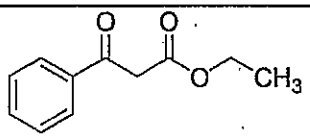
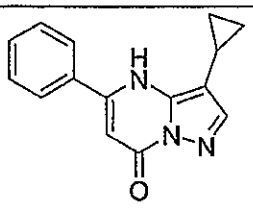
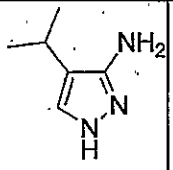
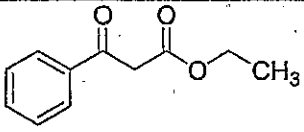
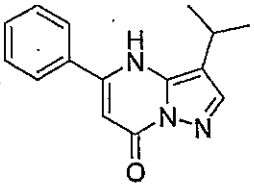
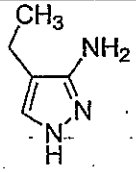
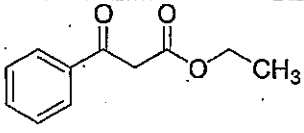
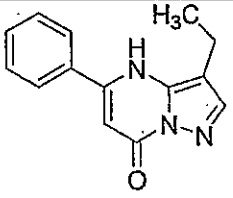
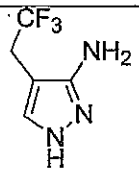
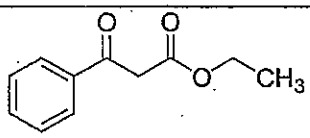
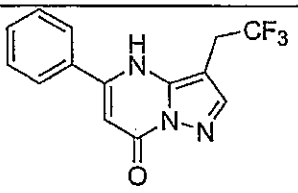
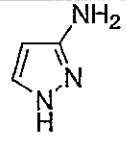
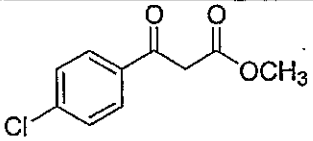
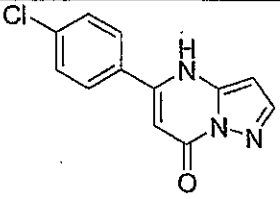
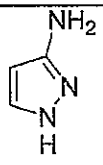
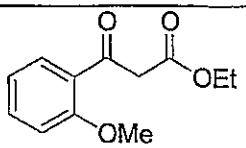
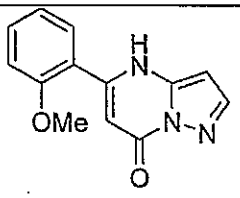
A solution of 3-aminopyrazole (2.0g, 24.07 mmol) and ethyl benzoylacetate (4.58 mL, 1.1 eq.) in AcOH (15 mL) was heated at reflux for 3 hours. The reaction mixture was cooled to room temperature and concentrated *in vacuo*. The resulting solid was diluted with EtOAc and filtered to give a white solid (2.04 g, 40% yield).

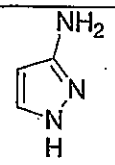
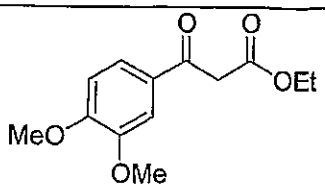
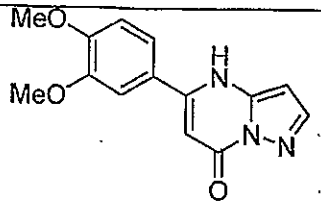
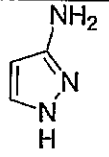
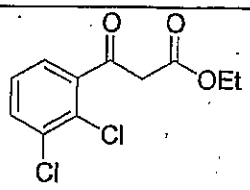
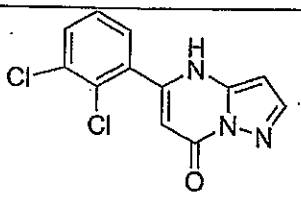
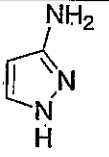
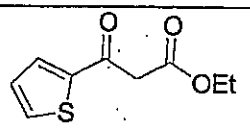
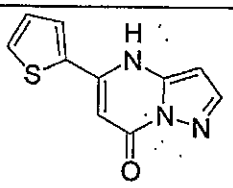
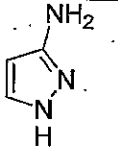
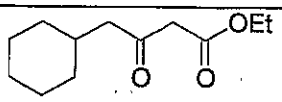
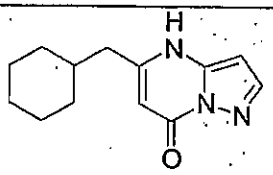
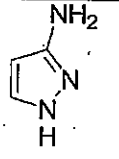
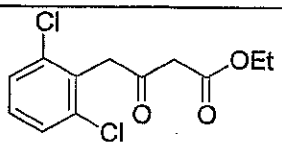
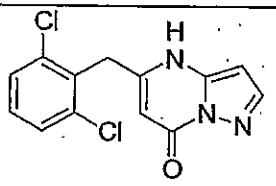
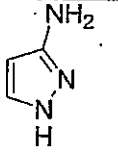
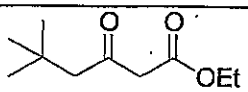
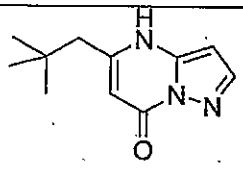
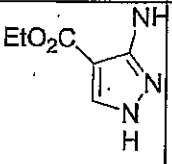
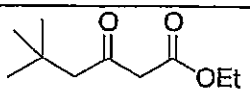
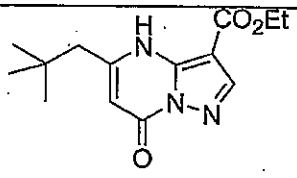
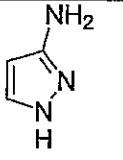
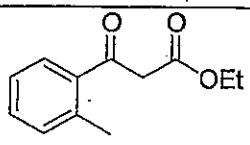
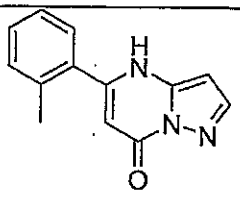
PREPARATIVE EXAMPLES 30-73:

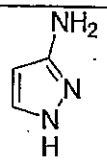
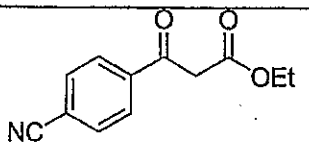
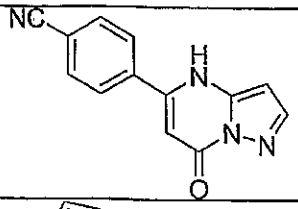
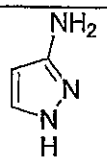
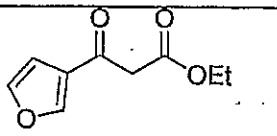
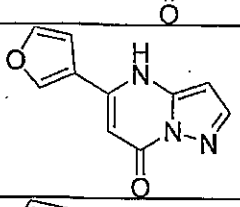
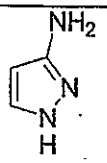
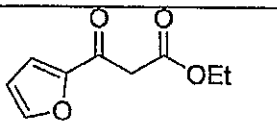
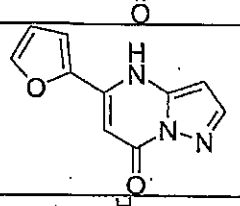
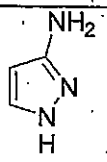
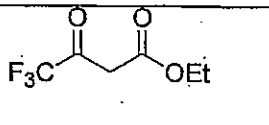
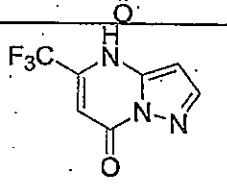
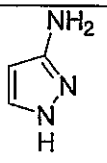
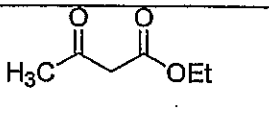
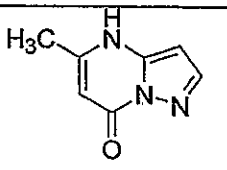
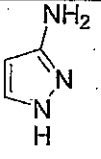
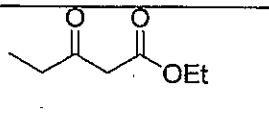
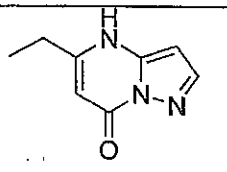
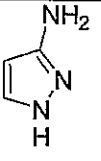
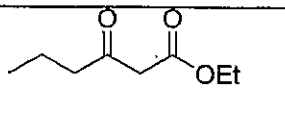
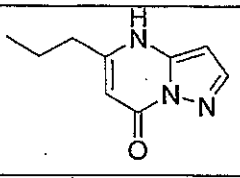
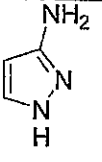
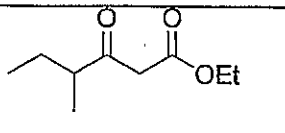
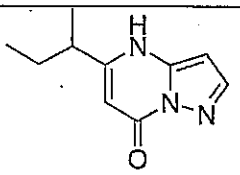
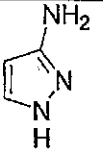
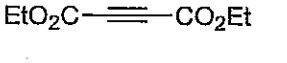
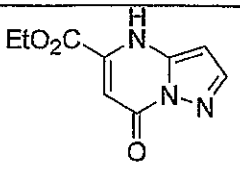
By essentially the same procedure set forth in Preparative Example 29 only substituting the aminopyrazole shown in Column 2 of Table 5 and the ester shown in Column 3 of Table 5, the compounds shown in Column 4 of Table 5 were prepared:

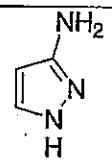
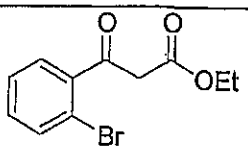
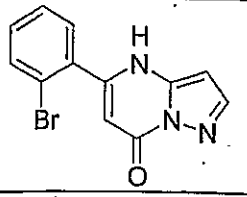
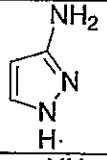
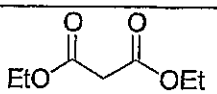
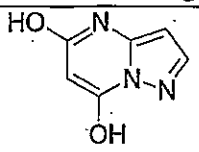
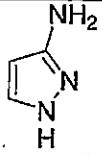
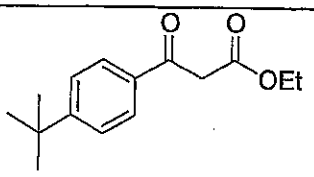
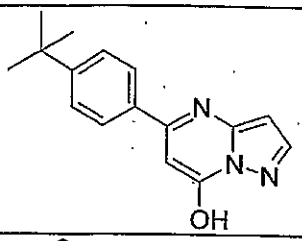
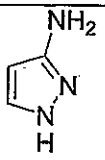
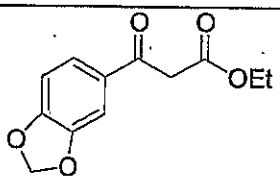
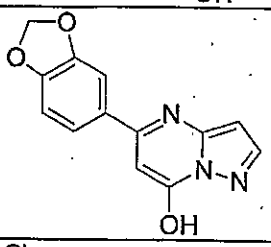
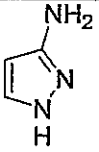
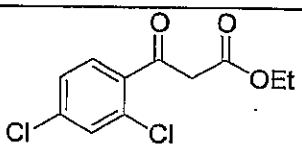
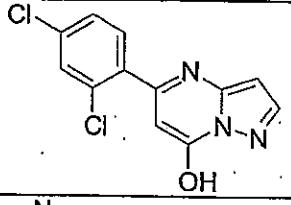
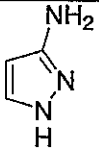
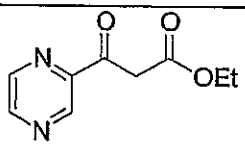
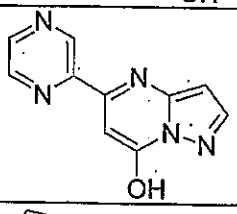
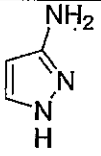
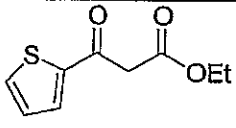
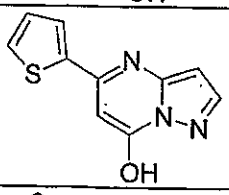
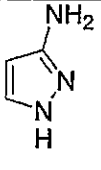
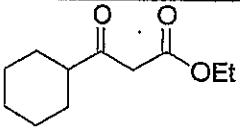
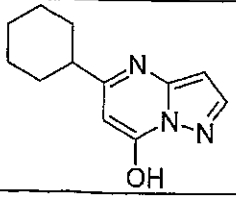
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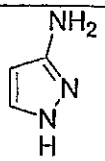
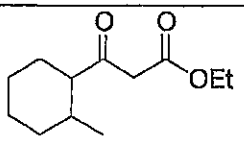
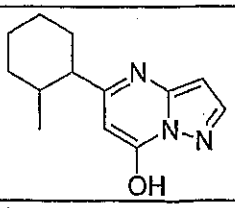
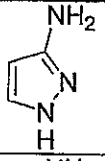
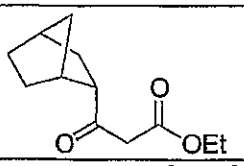
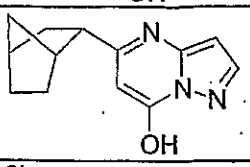
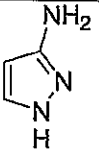
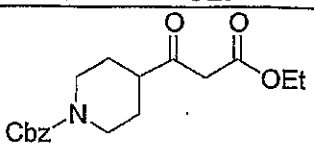
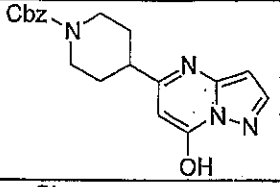
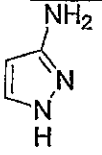
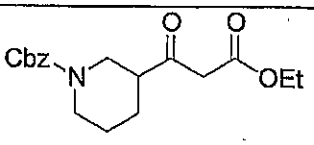
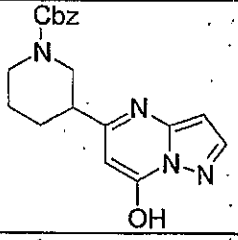
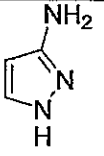
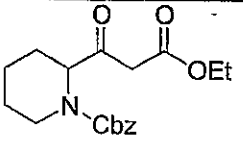
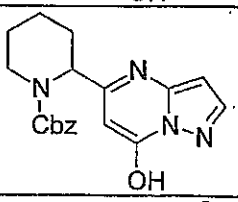
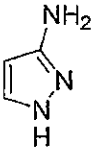
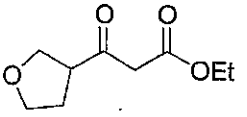
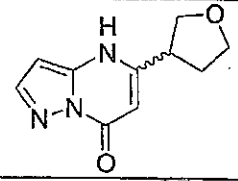
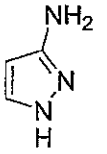
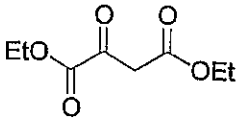
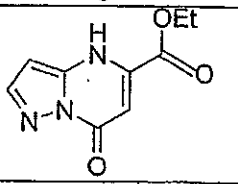
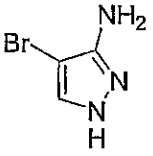
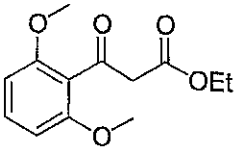
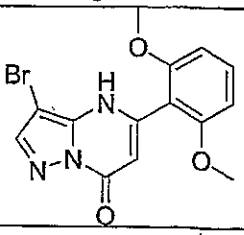
Prep. Ex.	Column 2	Column 3	Column 4	Column 5
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32				

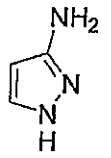
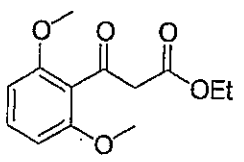
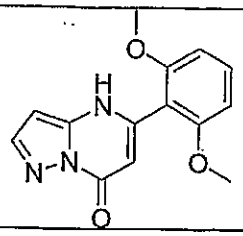
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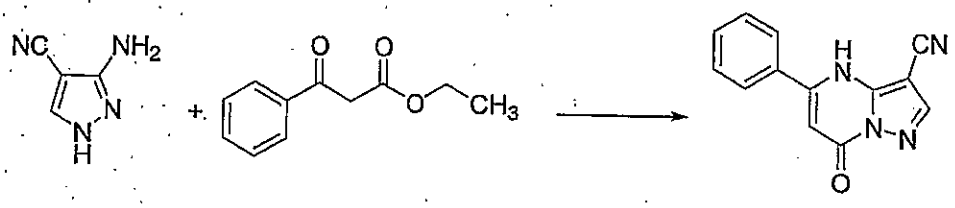
48				
49				
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56				

57				
58				Yield = 68 MH^+ = 152
59				Yield = 46 MH^+ = 268
60				Yield = 63 MH^+ = 255
61				Yield = 80 MH^+ = 280
62				Yield =72 MH^+ = 214
63				Yield = 51 MH^+ = 218
64				Yield = 82 MH^+ =218

65				Yield = 39 MH^+ =232
66				Yield = 30 MH^+ =230
67				Yield = 80 MH^+ =353
68				Yield = 49 MH^+ =353
69				Yield = 42 MH^+ =353
70				
71				
72				

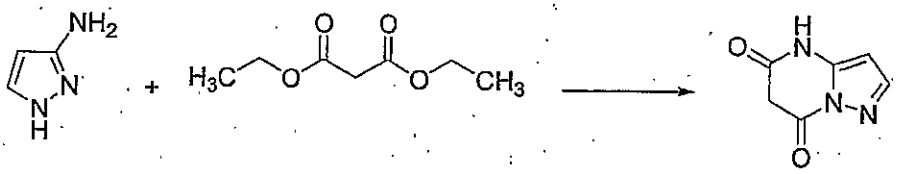
73				

PREPARATIVE EXAMPLE 74:



Ethyl benzoylacetate (1.76 mL, 1.1 eq.) and 3-amino-4-cyanopyrazole (1.0 g, 9.25 mmol) in AcOH (5.0 mL) and H_2O (10 mL) was heated at reflux 72 hours. The resulting solution was cooled to room temperature, concentrated *in vacuo*, and diluted with EtOAc. The resulting precipitate was filtered, washed with EtOAc, and dried *in vacuo* (0.47 g, 21% yield).

PREPARATIVE EXAMPLE 75

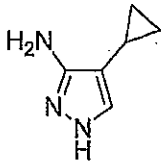
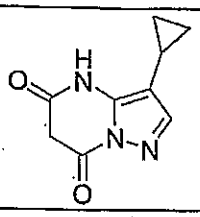
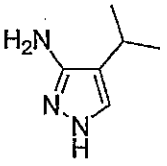
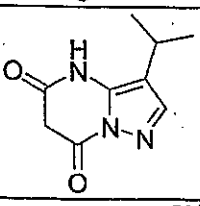
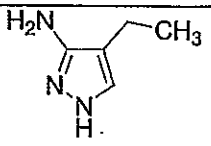
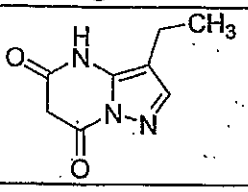


A procedure in US patent 3,907,799 was followed. Sodium (2.3 g, 2 eq.) was added to EtOH (150 mL) portionwise. When the sodium was completely dissolved, 3-aminopyrazole (4.2 g, 0.05 mol) and diethyl malonate (8.7 g, 1.1 eq.) were added and the resulting solution heated to reflux for 3 hours. The resulting suspension was cooled to room temperature and filtered. The filter cake was washed with EtOH (100 mL) and dissolved in water (250 mL). The resulting solution was cooled in an ice bath and the pH adjusted to 1-2 with concentrated HCl. The resulting suspension was filtered, washed with water (100 mL) and dried under vacuum to give a white solid (4.75 g, 63% yield).

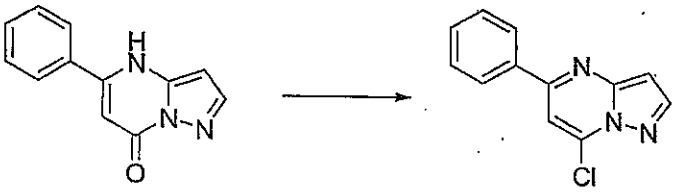
PREPARATIVE EXAMPLES 76-78:

By essentially the same procedure set forth in Preparative Example 75 only substituting the compound shown in Column 2 of Table 6, the compounds shown in Column 3 of Table 6 are prepared:

TABLE 6

Prep. Ex.	Column 2	Column 3
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77		
78		

PREPARATIVE EXAMPLE 79:

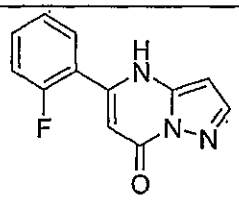
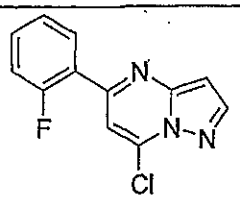
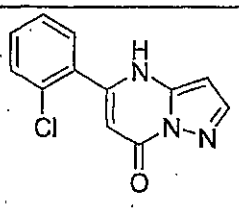
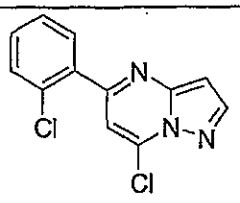
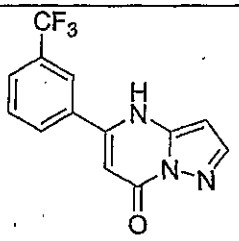
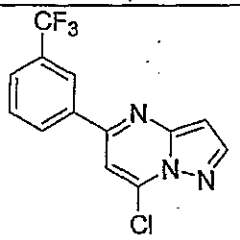
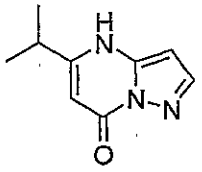
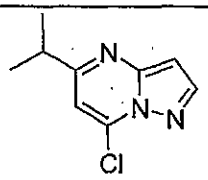
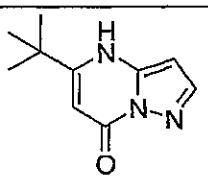
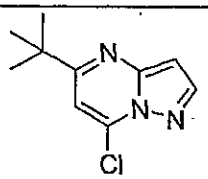
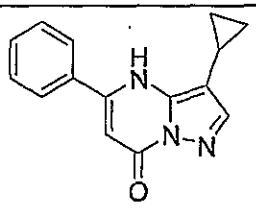
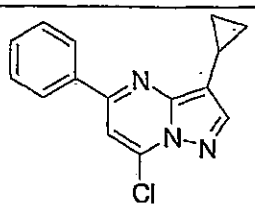
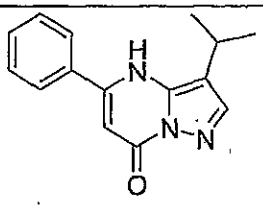
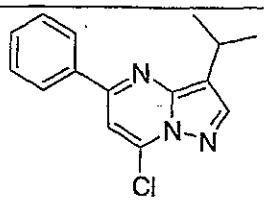


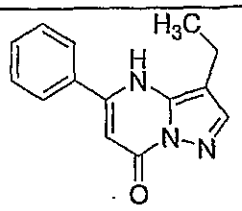
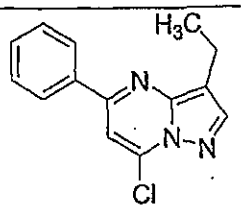
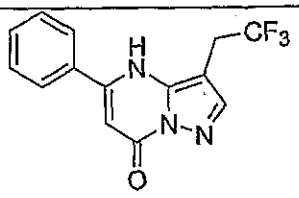
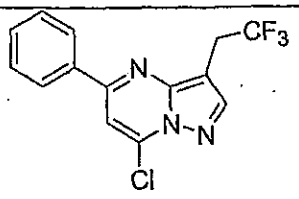
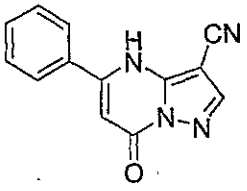
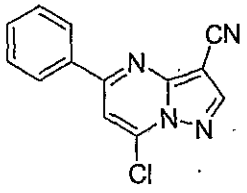
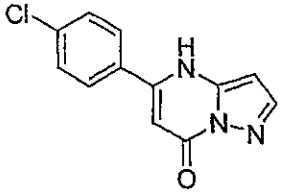
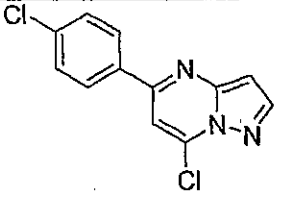
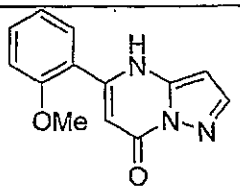
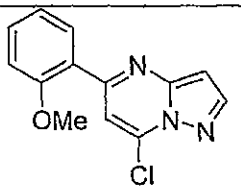
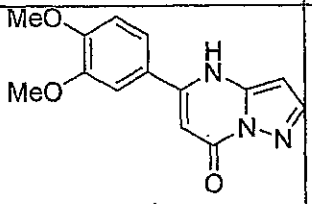
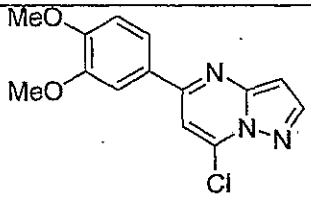
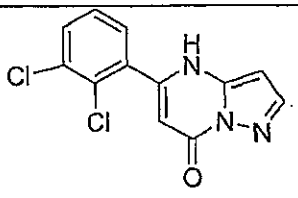
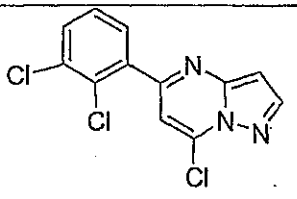
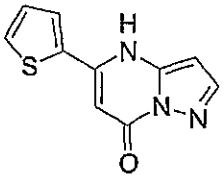
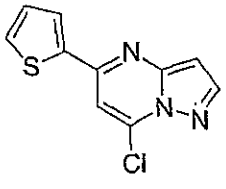
A solution of the compound prepared in Preparative Example 29 (1.0 g, 4.73 mmol) in $POCl_3$ (5 mL) and pyridine (0.25 mL) was stirred at room temperature 3 days. The resulting slurry was diluted with Et_2O, filtered, and the solid residue washed with Et_2O. The combined Et_2O washings were cooled to 0°C and treated with ice. When the vigorous reaction ceased, the resulting mixture was diluted with H_2O, separated, and the aqueous layer extracted with Et_2O. The combined organics were washed with H_2O and saturated NaCl, dried over Na_2SO_4, filtered, and concentrated to give a pale yellow solid (0.86 g, 79% yield). LCMS: MH^+=230.

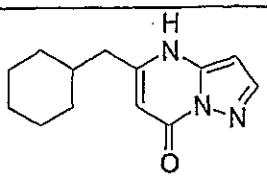
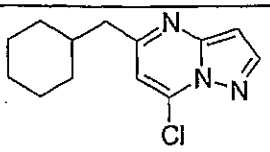
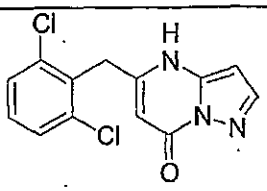
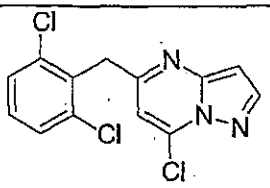
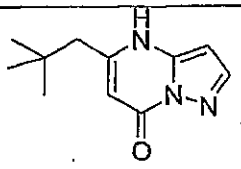
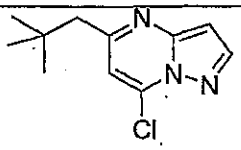
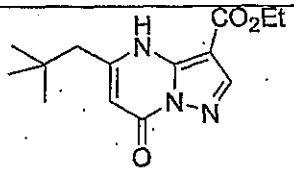
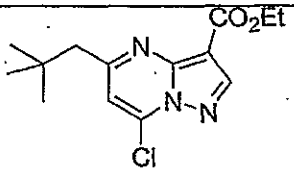
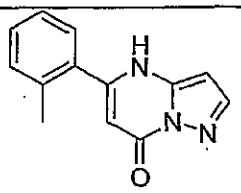
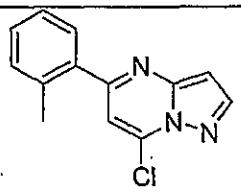
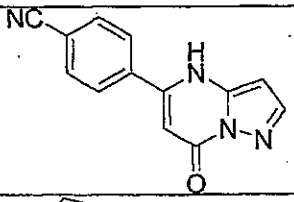
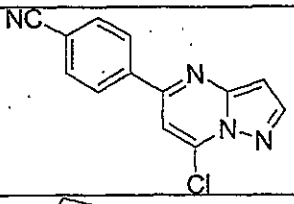
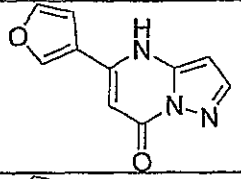
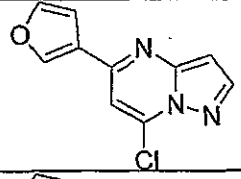
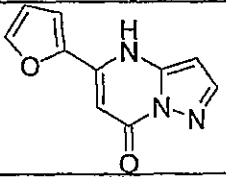
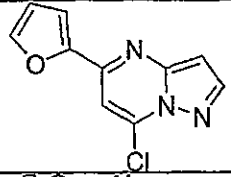
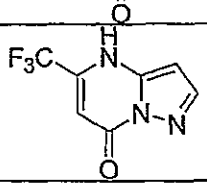
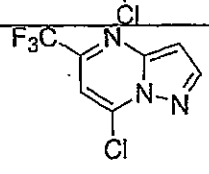
PREPARATIVE EXAMPLE 80-122:

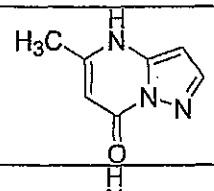
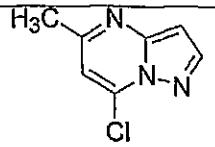
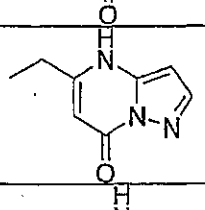
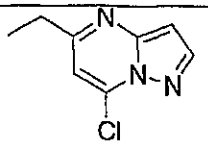
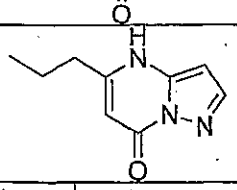
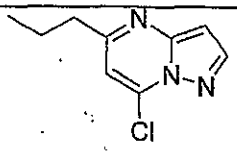
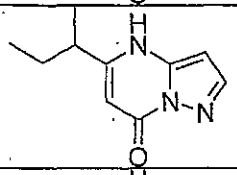
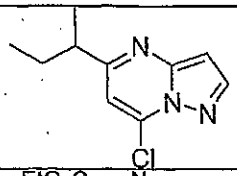
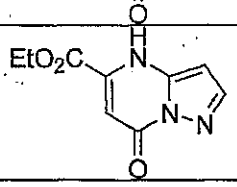
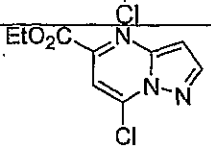
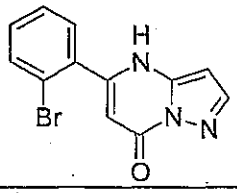
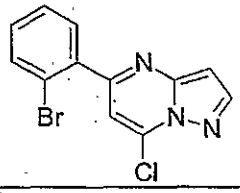
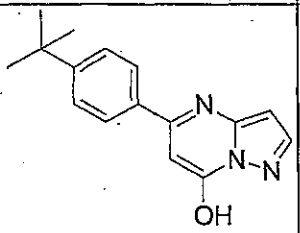
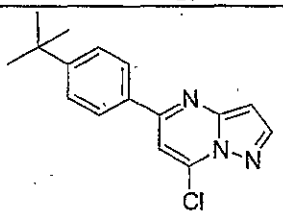
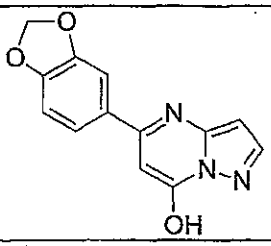
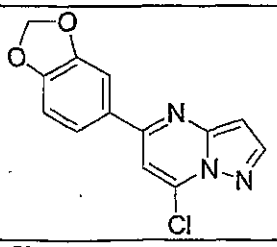
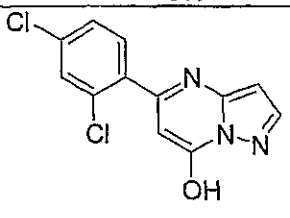
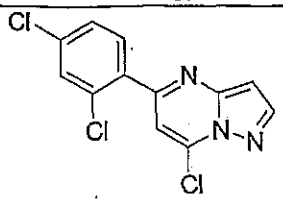
By essentially the same procedure set forth in Preparative Example 79, only substituting the compound shown in Column 2 of Table 7, the compounds shown in Column 3 of Table 7 were prepared:

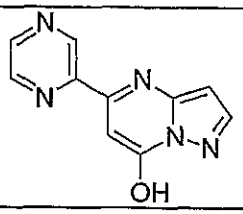
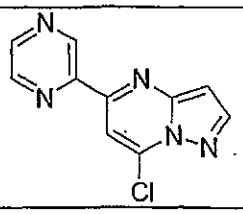
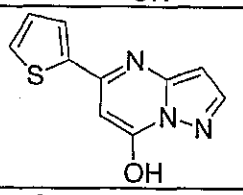
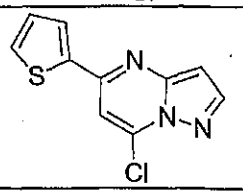
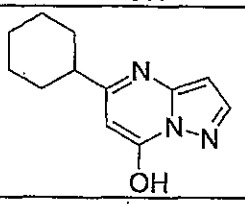
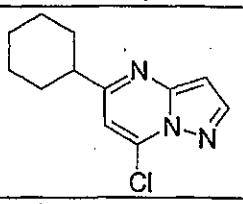
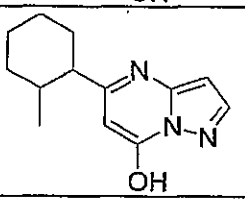
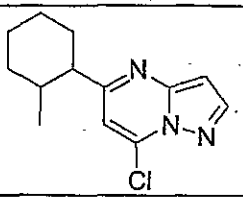
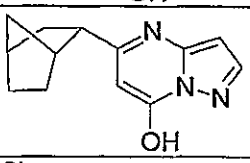
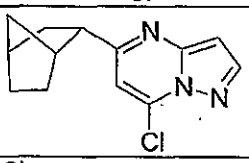
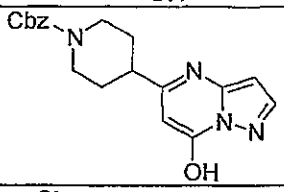
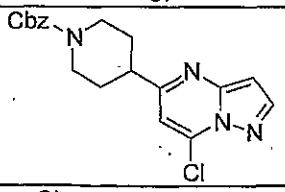
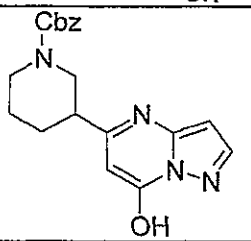
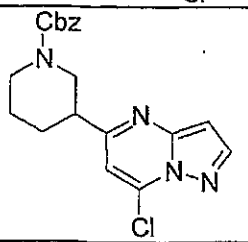
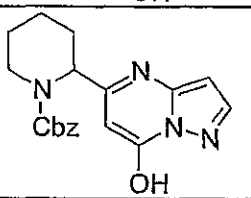
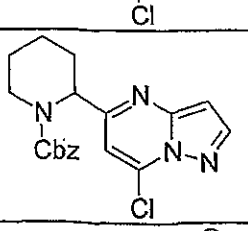
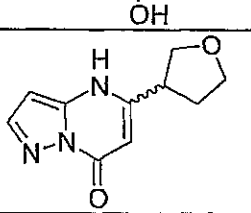
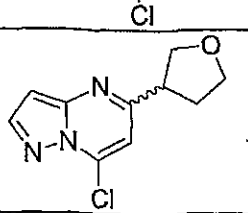
TABLE 7

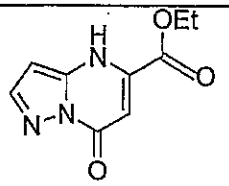
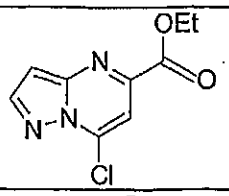
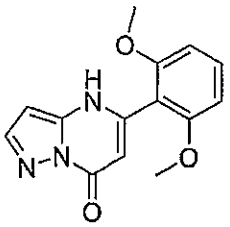
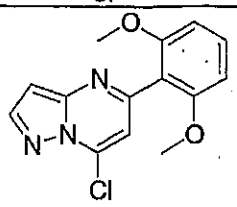
Prep. Ex.	Column 2	Column 3	CMPD
80			MS: MH^+=248
81			
82			MS: MH^+=298
83			MS: MH^+=196
84			MS: MH^+=210
85			
86			MS: MH^+=272

87			
87.10			
88			MS: MH^+=255
89			
90			Yield = 65% MS: MH^+ = 260
91			Yield = 35% MS: MH^+ = 290
92			Yield = 32%. MS: MH^+ = 298
93			Yield = 45% MS: MH^+ = 236

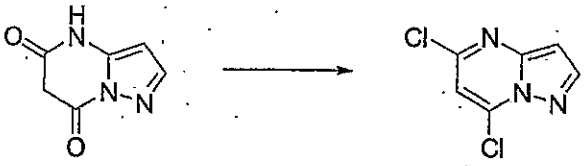
94			Yield = 100% LCMS: MH^+ = 250
95			Yield = 88% MS: MH^+ = 314
96			Yield=43% MS: MH^+=223
97			Yield=30% MS: MH^+=295
98			Yield=98% MS: MH^+=244
99			
100			
101			
102			

103			
104			
105			
106			
107			45% yield; MS: MH^+=226
108			MS: MH^+ = 308
109			Yield = quant MH^+ = 286
110			Yield = 50 MH^+ = 272
111			Yield = 85 MH^+ = 299

112			Yield = 97 MH^+ = 231
113			Yield = 45 MH^+ = 236
114			Yield = quant. MH^+ =236
115			Yield = 57 MH^+ =250
116			Yield = 89 MH^+ =248
117			Yield = 96 MH^+ =371
118			Yield = 99 MH^+ =371
119			Yield = 50 MH^+ =371
120			Yield=57% LCMS: MH^+=224

121			Yield=34% LCMS: MH^+=226
122			Yield=100% ^{1}H NMR ($CDCl_3$) δ 8.53(d, 1H), 7.66(t, 1H), 7.51(s, 1H), 7.45(d, 1H), 6.84(d, 2H).

PREPARATIVE EXAMPLE 123

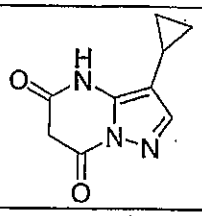
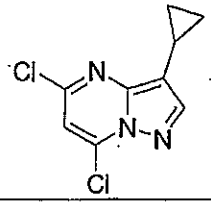
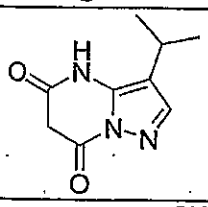
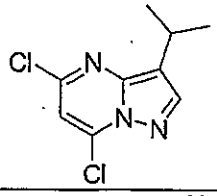
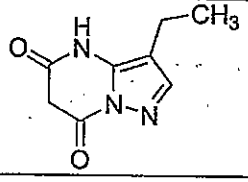
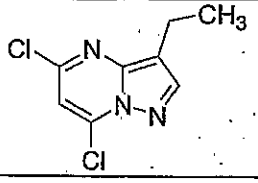


$POCl_3$ (62 mL) was cooled to 5 °C under nitrogen and dimethylaniline (11.4 g, 2.8 eq.) and the compound prepared in Preparative Example 75 (4.75 g, 0.032 mol). The reaction mixture was warmed to 60 °C and stirred overnight. The reaction mixture was cooled to 30 °C and the $POCl_3$ was distilled off under reduced pressure. The residue was dissolved in CH_2Cl_2 (300 mL) and poured onto ice. After stirring 15 minutes, the pH of the mixture was adjusted to 7-8 with solid $NaHCO_3$. The layers were separated and the organic layer was washed with H_2O (3 x 200 mL), dried over $MgSO_4$, filtered, and concentrated. The crude product was purified by flash chromatography using a 50 : 50 CH_2Cl_2 : hexanes solution as eluent to elute the dimethyl aniline. The eluent was then changed to 75 : 25 CH_2Cl_2 : hexanes to elute the desired product (4.58 g, 77% yield). MS: MH^+=188.

PREPARATIVE EXAMPLES 124-126

By essentially the same procedure set forth in Preparative Example 123 only substituting the compound in Column 2 of Table 8, the compounds shown in Column 3 of Table 8 are prepared:

TABLE 8

Prep. Ex.	Column 2	Column 3
124		
125		
126		

PREPARATIVE EXAMPLE 127:

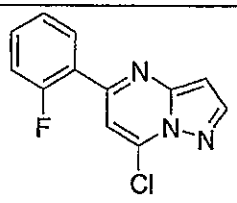
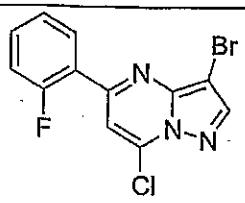
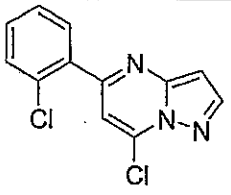
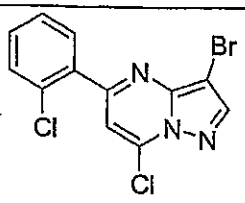
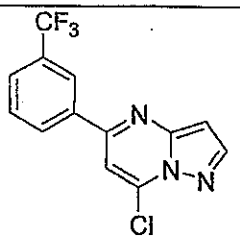
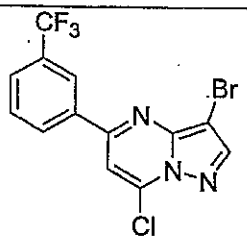
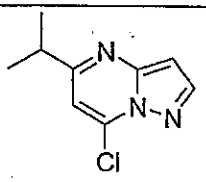
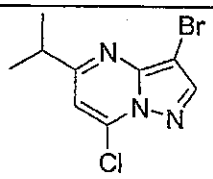
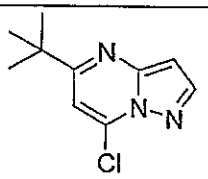
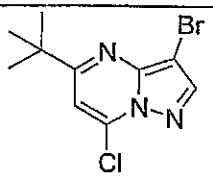
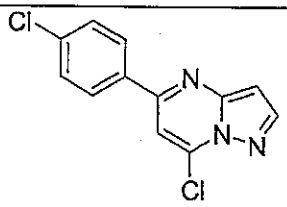
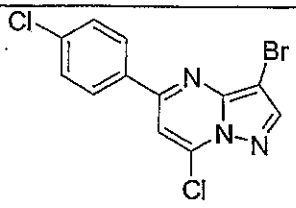
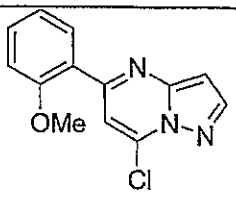
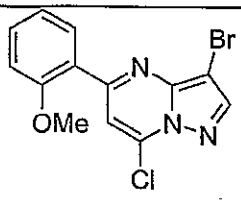


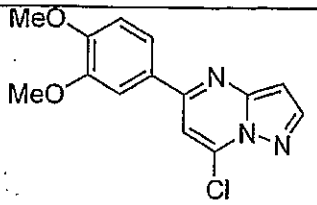
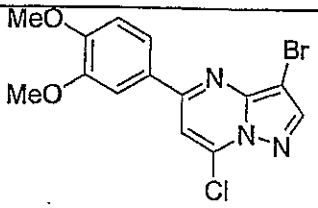
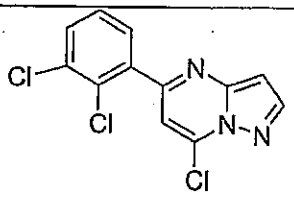
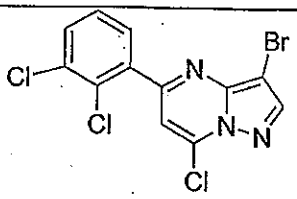
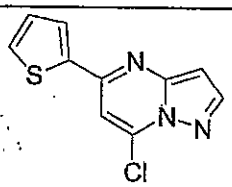
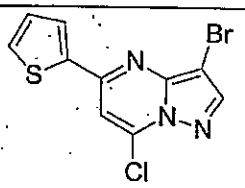
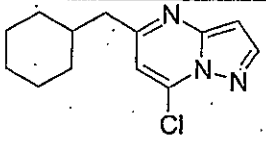
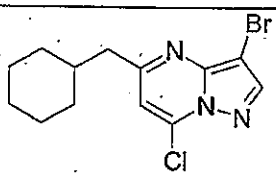
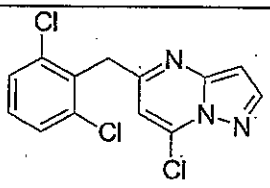
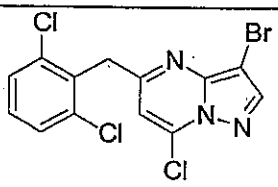
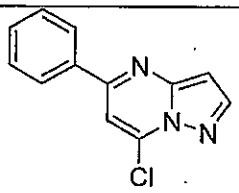
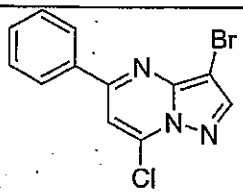
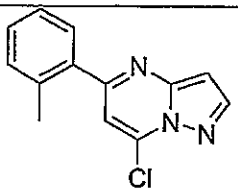
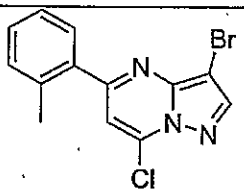
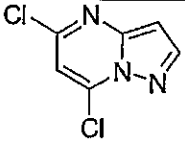
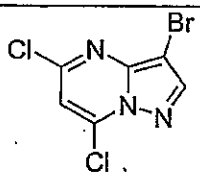
A solution of the compound prepared in Preparative Example 79 (0.10 g, 0.435 mmol) in CH_3CN (3 mL) was treated with NBS (0.085 g, 1.1 eq.). The reaction mixture was stirred at room temperature 1 hour and concentrated under reduced pressure. The crude product was purified by flash chromatography using a 20% EtOAc-in-hexanes solution as eluent (0.13 g, 100% yield). LCMS:MH^+=308.

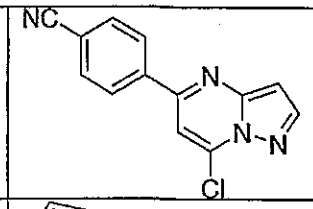
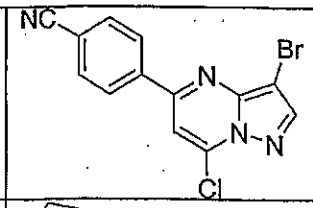
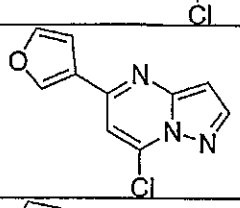
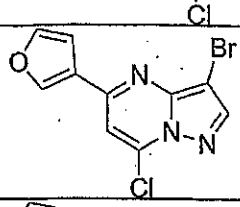
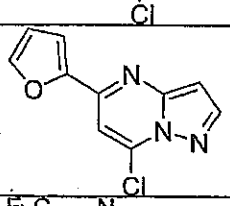
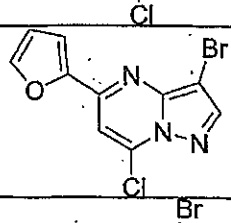
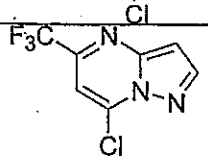
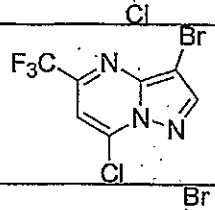
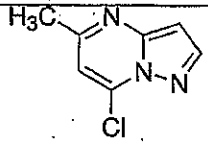
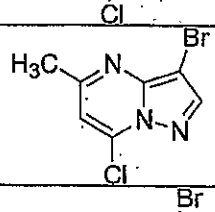
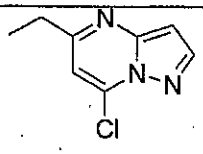
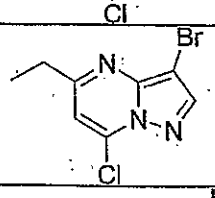
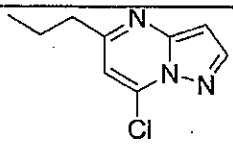
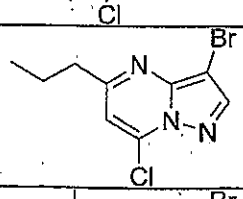
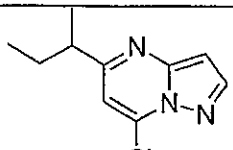
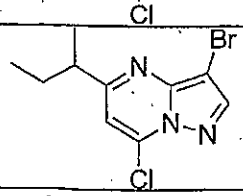
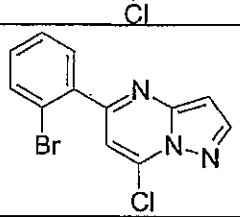
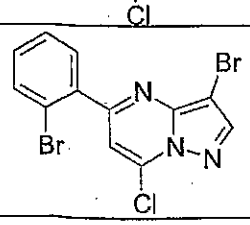
PREPARATIVE EXAMPLES 128-164:

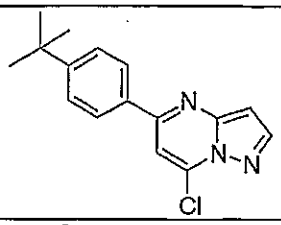
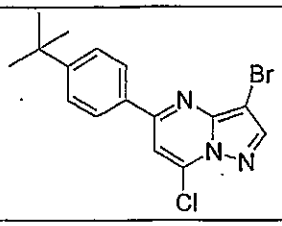
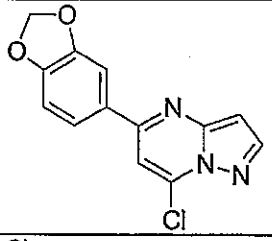
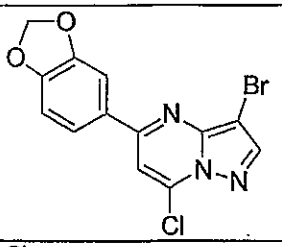
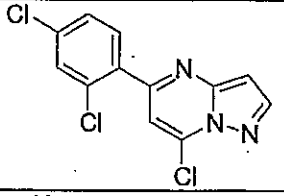
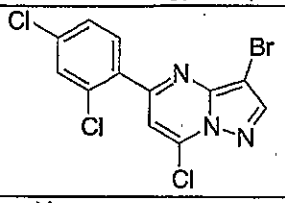
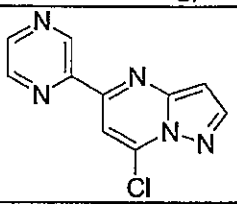
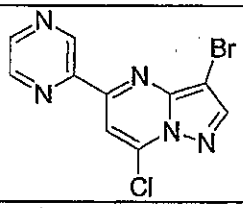
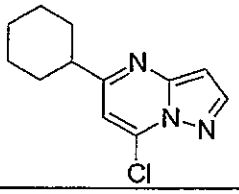
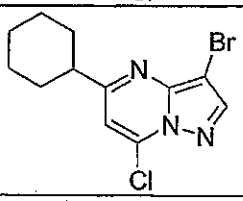
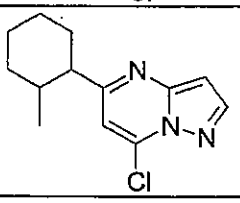
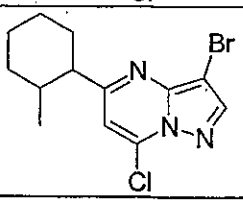
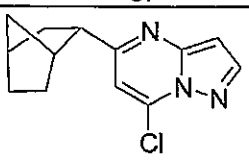
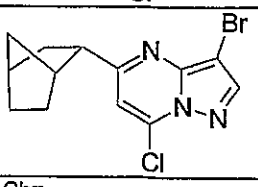
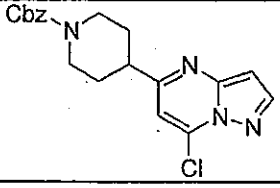
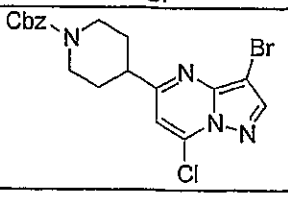
By essentially the same procedure set forth in Preparative Example 127 only substituting the compounds shown in Column 2 of Table 9, the compounds shown in Column 3 of Table 9 were prepared:

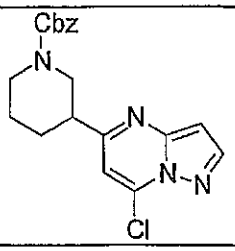
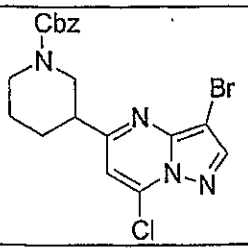
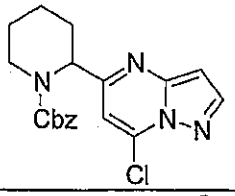
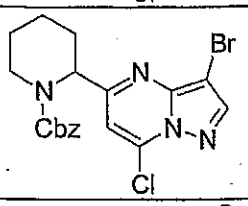
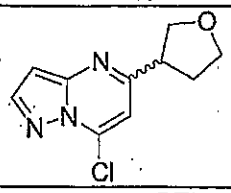
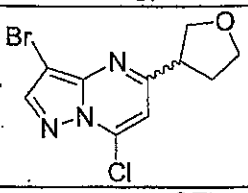
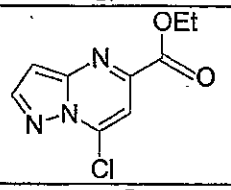
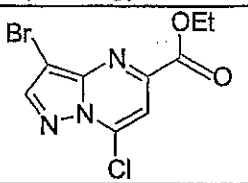
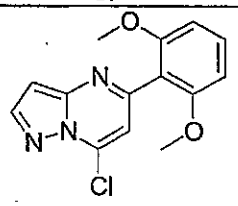
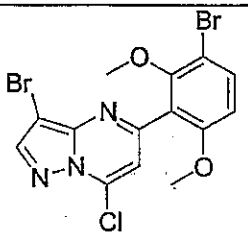
TABLE 9

Prep. Ex.	Column 2	Column 3	CMPD
128			MS: MH^+= 326
129			MS: MH^+= 342
130			MS: MH^+= 376
131			MS: MH^+=274
132			MS: MH^+=288
133			
134			Yield = 75% MS: MH^+ = 338

135			Yield = 52% MS: MH^+ = 368
136			Yield = 87% MS: MH^+ = 376
137			Yield = 100% MS: MH^+ = 316
138			Yield = 92% MS: MH^+ = 330
139			Yield = 82% MS: MH^+ = 395
140			Yield=88% MS: MH^+=308
141			Yield=100% MS: MH^+=322
142			MH^+=266

143			
144			
145			
146			
147			
148			
149			
150			
151			LCMS: $MH^+ = 386$

152			Yield = quant MH^+ = 364
153			Yield = quant MH^+ = 353
154			Yield = 95 MH^+ = 378
155			Yield = 77 MH^+ = 311
156			Yield =quant. MH^+ =314
157			Yield = 99 MH^+ =328
158			Yield = 98 MH^+ =326
159			Yield = 99 MH^+ =449

160			Yield = 95 MH^+ =449
161			Yield = 72 MH^+ =449
162			Yield=98% LCMS: MH^+=302
163			Yield=95% LCMS: MH^+=305
164			Yield=50% ^1H NMR ($CDCl_3$) δ 8.36(s, 1H), 7.72(d, 1H), 7.20(s, 1H), 6.82(d, 1H), 3.99(s, 3H), 3.90(s, 3H);

PREPARATIVE EXAMPLE 165:



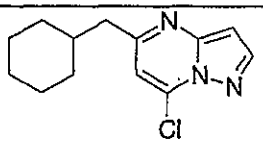
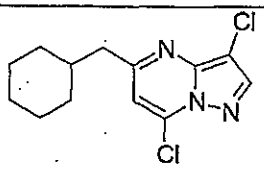
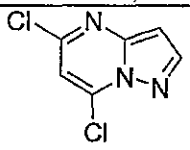
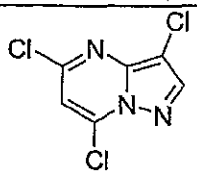
A solution of the compound prepared in Preparative Example 80 (0.3 g, 1.2 mmol) in CH_3CN (15 mL) was treated with NCS (0.18 g, 1.1 eq.) and the resulting solution heated to reflux 4 hours. Additional NCS (0.032 g, 0.2 eq.) added and the resulting solution was stirred at reflux overnight. The reaction mixture was cooled to room temperature, concentrated *in vacuo* and the residue

purified by flash chromatography using a 20% EtOAc in hexanes solution as eluent (0.28 g, 83% yield). LCMS: MH^+= 282.

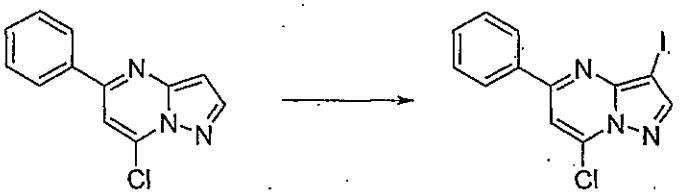
PREPARATIVE EXAMPLE 166-167:

By essentially the same procedure set forth in Preparative Example 165 only substituting the compound shown in Column 2 of Table 10, the compound shown in Column 3 of Table 10 was prepared:

TABLE 10

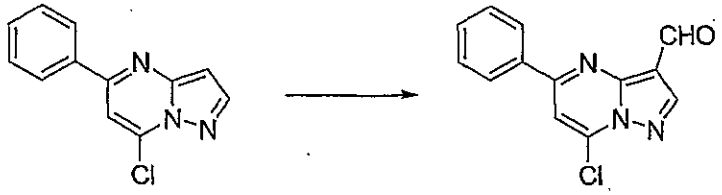
Prep. Ex.	Column 2	Column 3	CMPD
166			Yield = 82% LCMS: MH^+ = 286
167			

PREPARATIVE EXAMPLE 167.10:



By essentially the same procedure set forth in Preparative Example 165 only substituting N-iodosuccinimide, the above compound was prepared.

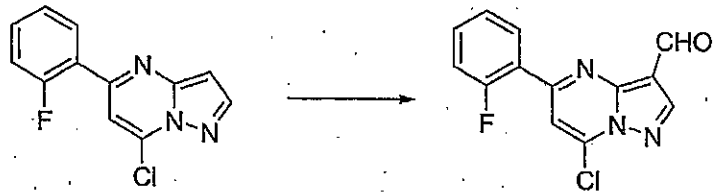
PREPARATIVE EXAMPLE 168:



To a solution of the compound from Preparative Example 79 (1.0 g, 4.35 mmol) in DMF (6 mL) was added $POCl_3$ (1.24 mL, 3.05 eq.) and the resulting mixture was stirred at room temperature overnight. The reaction mixture was cooled to 0°C and the excess $POCl_3$ was quenched by the addition of ice. The

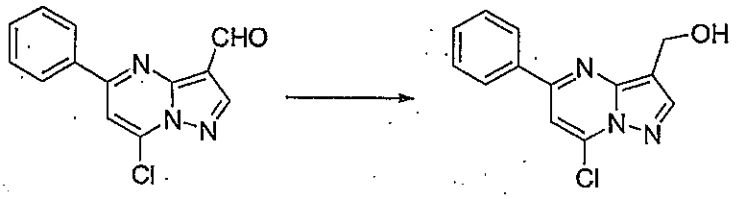
resulting solution was neutralized with 1N NaOH, diluted with H_2O, and extracted with CH_2Cl_2. The combined organics were dried over Na_2SO_4, filtered and concentrated *in vacuo*. The crude product was purified by flash chromatography using a 5% MeOH in CH_2Cl_2 solution as eluent (0.95 g, 85% yield). LCMS: MH^+=258.

PREPARATIVE EXAMPLE 169:



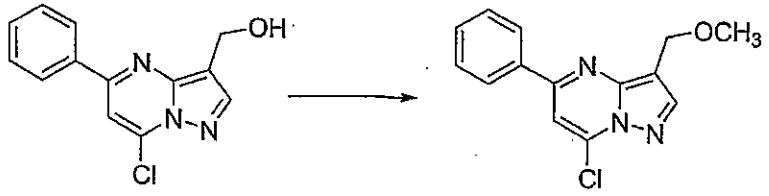
By essentially the same procedure set forth in Preparative Example 168 only substituting the compound prepared in Preparative Example 80, the above compound was prepared (0.45 g, 40% yield).

PREPARATIVE EXAMPLE 170:



To a solution of the product of Preparative Example 169 (0.25 g, 0.97 mmol) in THF was added $NaBH_4$ (0.041 g, 1.1 eq.) and the resulting solution was stirred at room temperature overnight. The reaction mixture was quenched by the addition of H_2O and extracted with CH_2Cl_2. The combined organics were dried over Na_2SO_4, filtered, and concentrated under reduced pressure. The crude product was purified by flash chromatography using a 60: 40 hexanes : EtOAc mix as eluent (0.17 g, 69% yield). MS: MH^+=260.

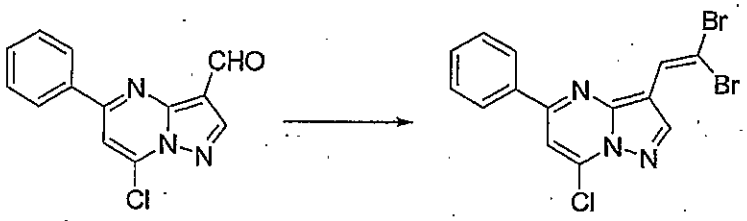
PREPARATIVE EXAMPLE 171:



A solution of the compound prepared in Preparative Example 170 (0.12 g, 0.462 mmol), dimethyl sulfate (0.088 mL, 2.0 eq), 50% NaOH (0.26 mL) and

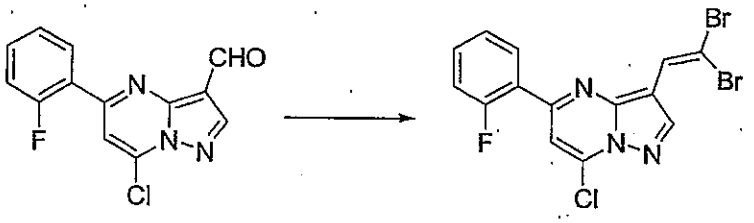
catalytic Bu_4NBr in CH_2Cl_2 (4 mL) was stirred at room temperature overnight. The reaction mixture was diluted with H_2O and extracted with CH_2Cl_2. The combined organics were dried over Na_2SO_4, filtered, and concentrated under reduced pressure. The crude product was purified by flash chromatography using a 30% EtOAc-in-hexanes solution as eluent (0.062 g, 48% yield).

PREPARATIVE EXAMPLE 172



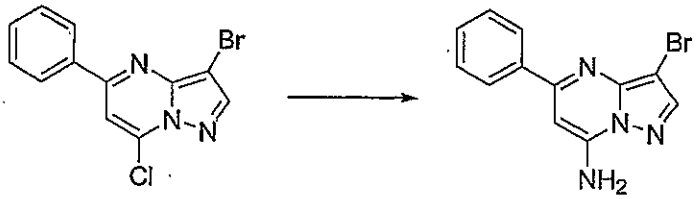
To a solution of PPh_3 (4.07 g, 4.0 eq.) and CBr_4 (2.57 g, 2.0 eq.) in CH_2Cl_2 (75 mL) at 0 °C was added the compound prepared in Preparative Example 168 (1.0 g, 3.88 mmol). The resulting solution was stirred at 0 °C for 1 hour and concentrated under reduced pressure. The residue was purified by flash chromatography using a 20% EtOAc in hexanes solution as eluent (1.07 g, 67% yield).

PREPARATIVE EXAMPLE 173:



By essentially the same procedure set forth in Preparative Example 172 only substituting the compound prepared in Preparative Example 169 the above compound was prepared (0.5 g, 70% yield).

PREPARATIVE EXAMPLE 174:



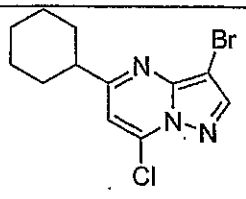
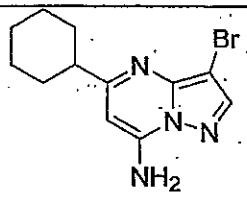
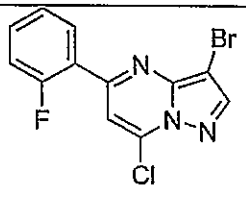
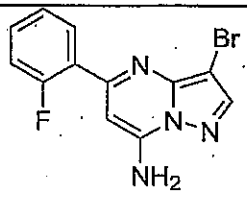
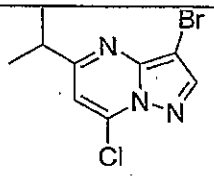
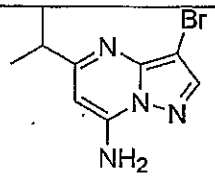
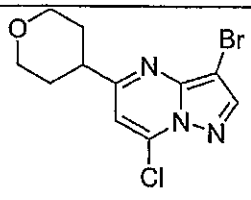
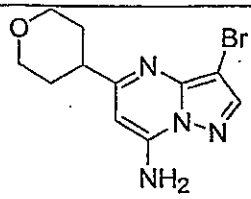
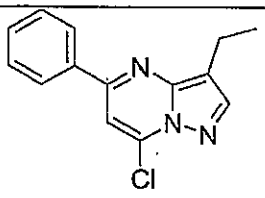
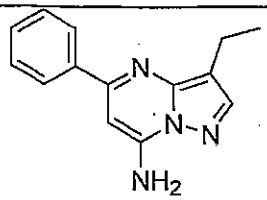
The compound prepared in Preparative Example 127 (3.08 g, 10.0 mmol), 2.0 M NH_3 in 2-propanol (50 mL, 100.0 mmol), and 37 % aqueous NH_3 (10.0 mL) were stirred in a closed pressure vessel at 50°C for 1 day. The solvent was

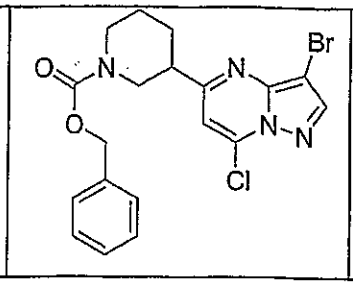
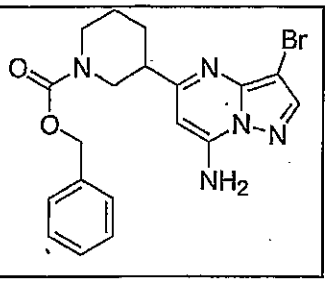
evaporated and the crude product was purified by flash chromatography using 3:1 CH_2Cl_2:EtOAc as eluent. Pale yellow solid (2.30 g, 80%) was obtained. LCMS: M^+=289.

PREPARATIVE EXAMPLES 175-180:

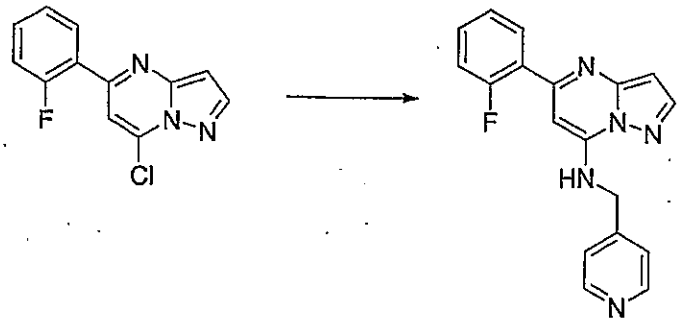
By essentially the same procedure set forth in Preparative Example 174 only substituting the compound shown in Column 2 of Table 11, the compounds shown in Column 3 of Table 11 were prepared.

TABLE 11

Prep. Ex.	Column 2	Column 3
175		
176		
177		
178		
179		

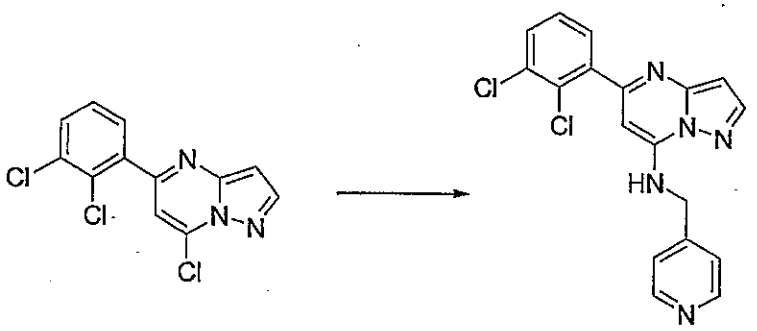
180		

PREPARATIVE EXAMPLE 181:



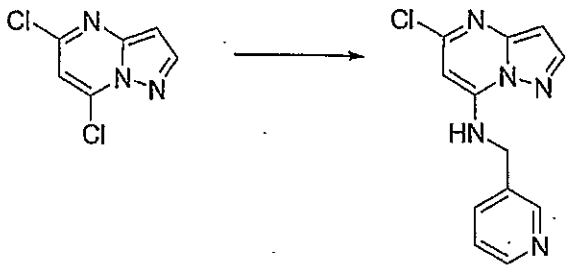
The compound prepared in Preparative Example 80 (0.3 g, 1.2 mmol), K_2CO_3 (0.33 g, 2 eq.), and 4-aminomethylpyridine (0.13 mL, 1.1 eq.) was heated to reflux overnight. The reaction mixture was cooled to room temperature and concentrated under reduced pressure. The residue was diluted with H_2O and extracted with CH_2Cl_2. The combined organics were dried over Na_2SO_4, filtered and, concentrated. The crude product was purified by flash chromatography using a 5% (10% NH_4OH in MeOH) solution in CH_2Cl_2 as eluent (0.051 g, 40% yield). LCMS: MH^+=320.

PREPARATIVE EXAMPLE 182:



By essentially the same procedure set forth in Preparative Example 181 only substituting the compound described in Preparative Example 92, the above compound was prepared. LCMS: MH^+=370.

PREPARATIVE EXAMPLE 183:

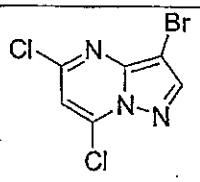
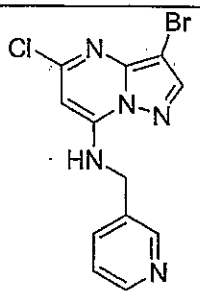


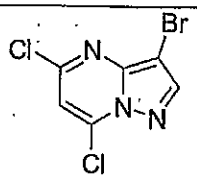
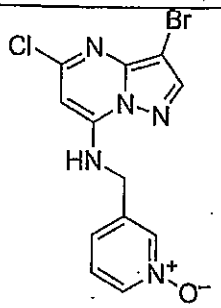
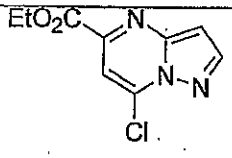
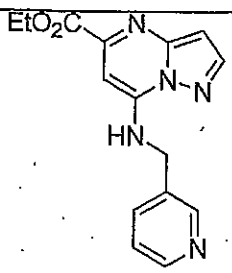
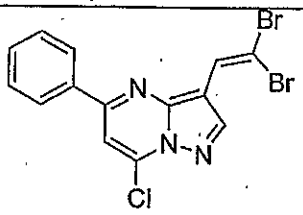
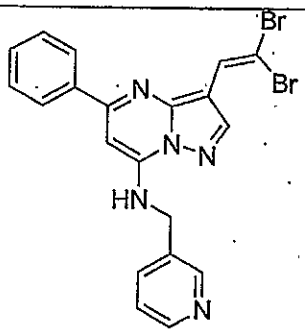
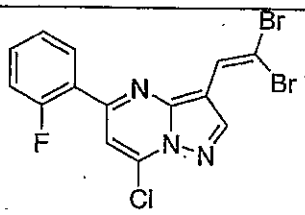
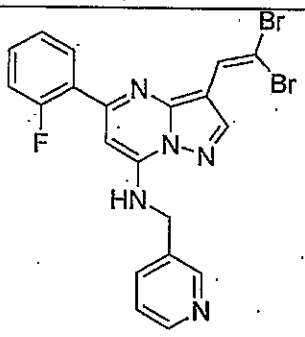
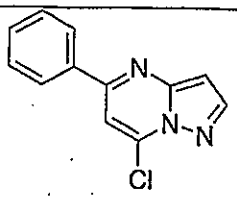
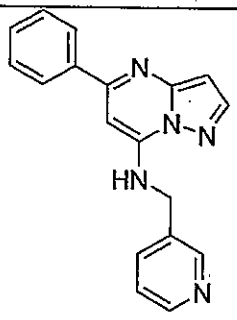
To a solution of the compound prepared in Preparative Example 123 (0.25 g, 1.3 mmol) in dioxane (5 mL) was added iPr_2NEt (0.47 mL, 2.0 eq.) and 3-aminomethylpyridine (0.15 ml, 1.1 eq.). The resulting solution was stirred at room temperature 72 hours. The reaction mixture was diluted with H_2O and extracted with EtOAc. The combined organics were washed with H_2O and saturated NaCl, dried over Na_2SO_4, filtered, and concentrated in vacuo. The crude product was purified by flash chromatography using a 5% MeOH in CH_2Cl_2 solution as eluent (0.29 g, 83% yield). MS: MH^+=260.

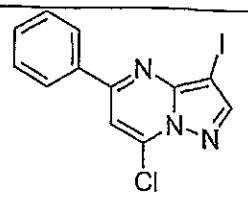
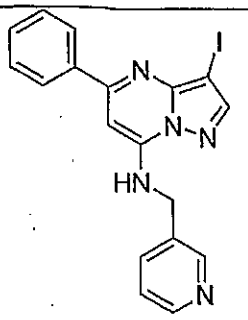
PREPARATIVE EXAMPLES 184-187:

By essentially the same procedure set forth in Preparative Example 183 only substituting the compound shown in Column 2 of Table 12, the compounds shown in Column 3 of Table 12 are prepared.

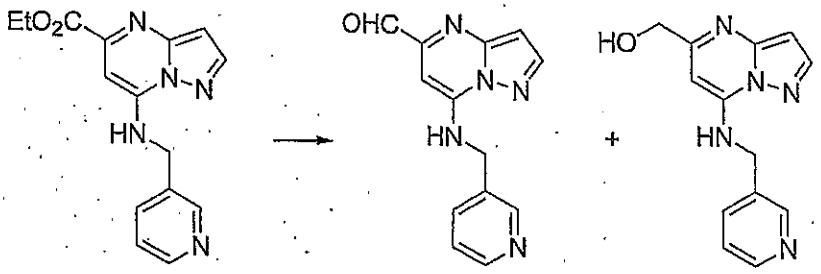
TABLE 12

Prep. Ex.	Column 2	Column 3
184		

184.1		
185		
186		
187		
187.1		

187.11		

PREPARATIVE EXAMPLE 188 and PREPARATIVE EXAMPLE 189:

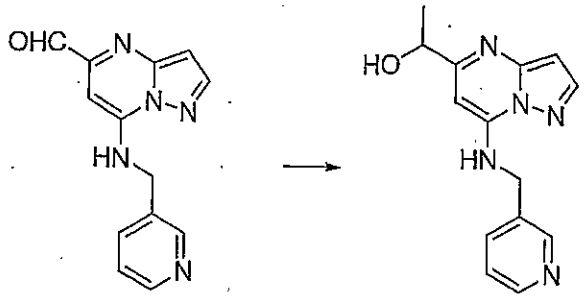


To a solution of the compound prepared in Preparative Example 185 (1.18 g, 3.98 mmol) in THF (35 mL) at -78 °C was added LAH (4.78 mL, 1M in Et_2O, 1.0 eq.) dropwise. The reaction mixture was stirred at -78 °C for 3 hours at which time additional LAH (2.0 mL, 1M in Et_2O, 0.42 eq.) was added dropwise. The reaction mixture was stirred an additional 1.25 hours and quenched by the addition of saturated Na_2SO_4 (8.5 mL). The reaction mixture was diluted with EtOAC (23 mL), H_2O (2 mL), and CH_3OH (50 mL). The resulting slurry was filtered through a plug of Celite. The Celite was washed with CH_3OH and the filtrate dried with Na_2SO_4, filtered, and concentrated. The product was purified by flash chromatography using a CH_2Cl_2 : CH_3OH (93 : 7) solution as eluent to yield aldehyde as the first eluting product and alcohol as the second eluting product.

Preparative Example 188: (aldehyde): 0.4 g, 39% yield. MS: MH^+ = 254.

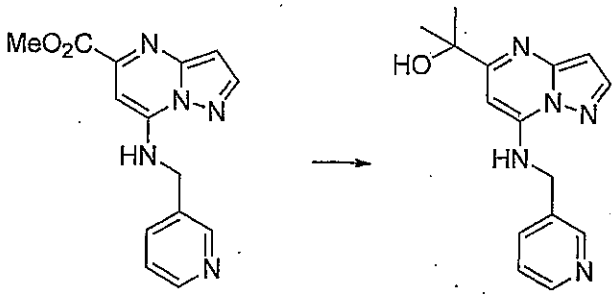
Preparative Example 189: (alcohol): 0.25 g, 24% yield. MS: MH^+ = 256.

PREPARATIVE EXAMPLE 190:



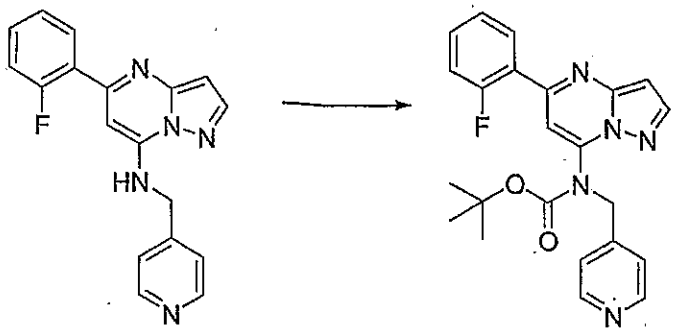
To a solution of the compound prepared in Preparative Example 188 (0.075 g, 0.30 mmol) in THF (2.0 mL) at 0 °C was added CH_3MgBr (0.3 mL, 3.0M solution in Et_2O, 3.0 eq.) dropwise. The resulting solution was stirred at 0 °C an additional 1.5 hours, warmed to room temperature, and stirred overnight. Additional CH_3MgBr (0.15 mL, 3.0M in Et_2O, 1. eq.) was added and the resulting solution stirred an additional 1.5 hours. The reaction mixture was cooled to 0 °C and quenched by the addition of saturated NH_4Cl. The resulting solution was diluted with CH_2Cl_2 and H_2O and extracted with CH_2Cl_2. The combined organics were washed with saturated NaCl and dried over Na_2SO_4, filtered, and concentrated. The crude product was purified by flash chromatography using a CH_2Cl_2 : CH_3OH (90 : 10) solution as eluent (0.048 g, 60% yield). MS: MH^+ = 270.

PREPARATIVE EXAMPLE 191:



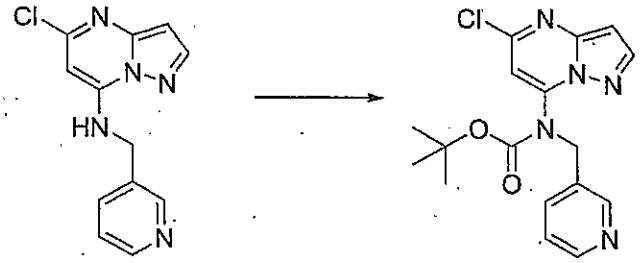
By essentially the same procedure set forth in Preparative Example 190 only substituting the compound prepared in Preparative Example 185 and using excess MeMgBr (5 eq.), the above compound was prepared.

PREPARATIVE EXAMPLE 192:



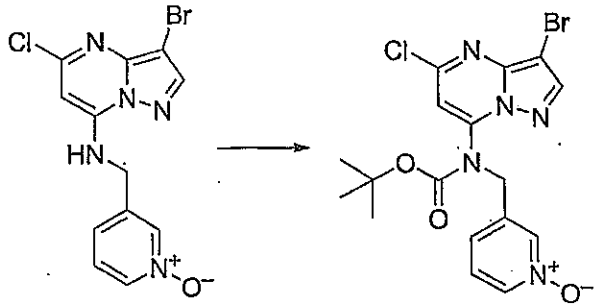
The compound prepared in Preparative Example 181 (0.29 g, 0.91 mmol), BOC_2O (0.22 g, 1.1 eq), and DMAP (0.13 g, 1.1 eq.) in dioxane (10 mL) was stirred at room temperature 3 days. Additional BOC_2O (0.10g, 0.5 eq.) was added and the reaction mixture was stirred 4 hours. The reaction mixture was concentrated *in vacuo*, diluted with saturated $NaHCO_3$ (15 mL), and extracted with CH_2Cl_2 (2 x 100 mL). The combined organics were dried over Na_2SO_4, filtered , and concentrated under reduce pressure. The crude product was purified by flash chromatography using a 5% (10% NH_4OH in MeOH) solution in CH_2Cl_2 as eluent (0.35 g, 91% yield). LCMS: MH^+= 420.

PREPARATIVE EXAMPLE 193:



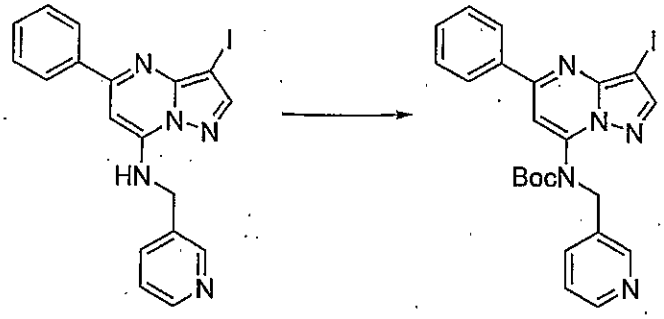
By essentially the same procedure set forth in Preparative Example 192 only substituting the compound prepared in Preparative Example 183, the above compound was prepared. MS: MH^+ = 360.

PREPARATIVE EXAMPLE 193.10:



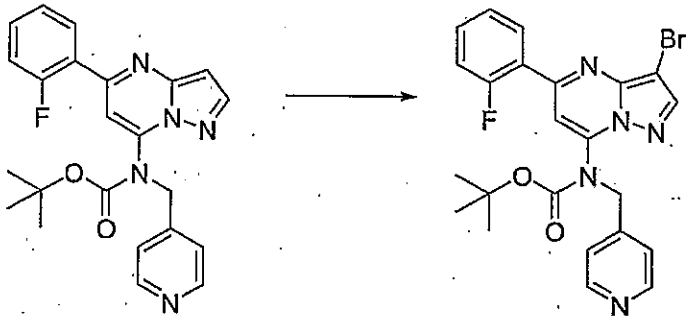
By essentially the same procedure set forth in Preparative Example 192 only substituting the compound prepared in Preparative Example 184.1, the above compound was prepared. MS: MH^+ = 454.

PREPARATIVE EXAMPLE 194:



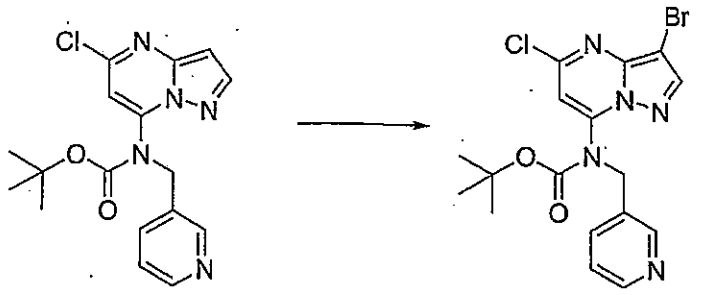
By essentially the same procedure set forth in Preparative Example 192 only substituting the above compound prepared in Preparative Example 187.11, the above compound was prepared (0.223 g, 88% yield). MS: MH^+ = 528.

PREPARATIVE EXAMPLE 195:



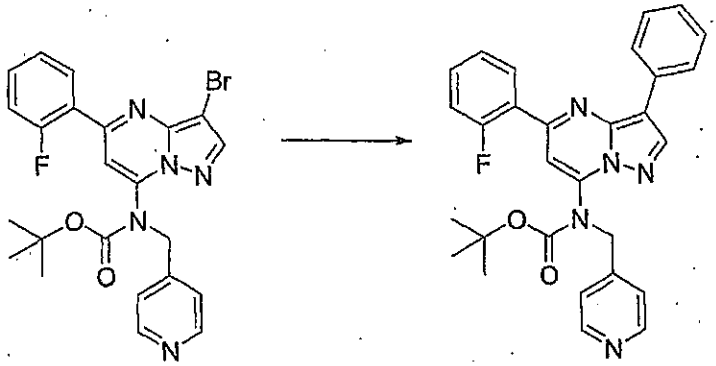
By essentially the same procedure set forth in Preparative Example 127 only substituting the compound prepared in Preparative Example 192, the above compound was prepared (0.38 g, 95% yield). LCMS: MH^+= 498.

PREPARATIVE EXAMPLE 196:



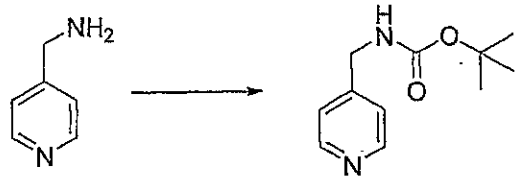
By essentially the same procedure set forth in Preparative Example 195, only substituting the compound prepared in Preparative Example 193, the above compound was prepared (0.3 g, 83% yield). MS: MH^+ = 438.

PREPARATIVE EXAMPLE 197:



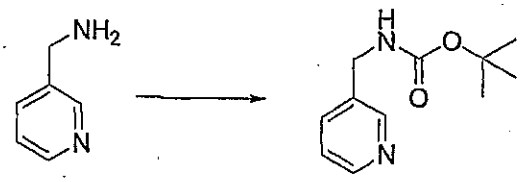
A solution of the compound prepared in Preparative Example 195 (0.15 g, 0.3 mmol), phenylboronic acid (0.073 g, 2.0 eq.), K_3PO_4 (0.19 g, 3.0 eq.), and $Pd(PPh_3)_4$ (0.017 g, 5 mol %) was heated at reflux in DME (16 mL) and H_2O (4 mL) 7 hours. The resulting solution was cooled to room temperature, diluted with H_2O (10 mL), and extracted with CH_2Cl_2 (3 x 50 mL). The combined organics were dried over Na_2SO_4, filtered, and concentrated. The crude product was purified by flash chromatography using a 2.5% (10% NH_4OH in MeOH) in CH_2Cl_2 solution as eluent (0.16 g, 100% yield).

PREPARATIVE EXAMPLE 198:



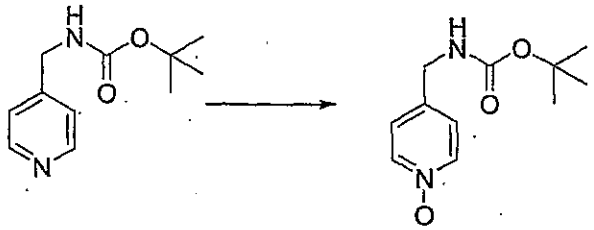
To a solution of 4-aminomethylpyridine (1.41 mL, 13.87 mmol) in CH_2Cl_2 (50 mL) was added BOC_2O (3.3 g, 1.1 eq.) and TEA and the resulting solution was stirred a room temperature 2 hours. The reaction mixture was diluted with H_2O (50 mL) and extracted with CH_2Cl_2. The combined organics were dried over Na_2SO_4, filtered and concentrated under reduced pressure. The crude product was purified by flash chromatography using a 5% (10% NH_4OH in MeOH) solution in CH_2Cl_2 as eluent to give a yellow solid (2.62 g, 91% yield). LCMS: MH^+= 209.

PREPARATIVE EXAMPLE 199:



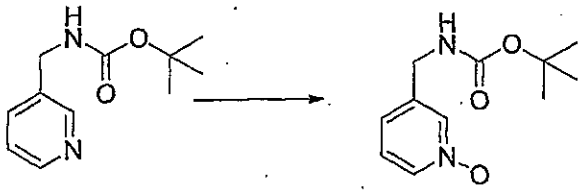
By essentially the same procedure set forth in Preparative Example 198 only substituting 3-aminomethylpyridine, the above compound was prepared as a yellow oil (2.66 g, 92% yield). LCMS: MH^+= 209.

PREPARATIVE EXAMPLE 200:



To a solution of the compound prepared in Preparative Example 198 (0.20 g, 0.96 mmol) in CH_2Cl_2 (5 mL) at 0°C was added *m*-CPBA (0.17 g, 1.0 eq) and the resulting solution stirred at 0°C 2 hours and stored at 4°C overnight at which time the reaction mixture was warmed to room temperature and stirred 3 hours. The reaction mixture was diluted with H_2O and extracted with CH_2Cl_2. The combined organics were dried over Na_2SO_4, filtered, and concentrated. The crude product was purified by flash chromatography using a 10% (10% NH_4OH in MeOH) solution as eluent: LCMS: MH^+= 255.

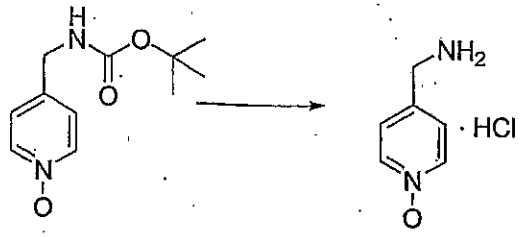
PREPARATIVE EXAMPLE 201:



A solution of oxone (58.6 g) in H_2O (250 mL) was added dropwise to the compound prepared in Preparative Example 199 (27 g, 0.13 mol) and $NaHCO_3$ (21.8 g, 2.0 eq.) in MeOH (200 mL) and H_2O (250 mL). The resulting solution was stirred at room temperature overnight. The reaction mixture was diluted with CH2Cl2 (500 mL) and filtered. The layers were separated and the aqueous layer extracted with CH_2Cl_2. The combined organics were dried over Na_2SO_4,

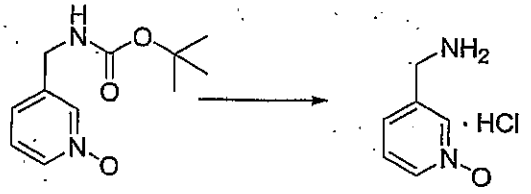
filtered, and concentrated under reduced pressure to give a white solid (21.0 g, 72% yield). MS: MH^+= 255.

PREPARATIVE EXAMPLE 202:



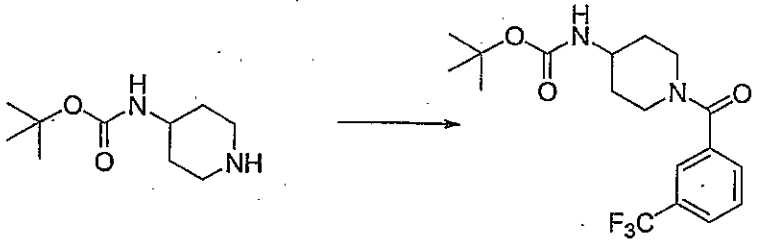
The compound prepared in Preparative Example 200 (0.29 g, 1.29 mmol) was stirred at room temperature in 4M HCl in dioxane (0.97 mL) 2 hours. The reaction mixture was concentrated *in vacuo* and used without further purification. LCMS: MH^+= 125.

PREPARATIVE EXAMPLE 203:



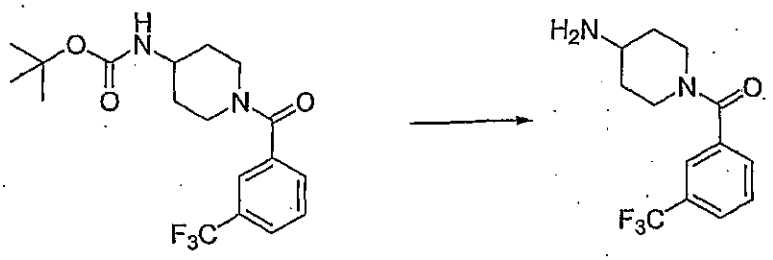
By essentially the same procedure set forth in Preparative Example 202 only substituting the compound prepared in Preparative Example 201, the compound shown above was prepared. LCMS: MH^+= 125.

PREPARATIVE EXAMPLE 204:



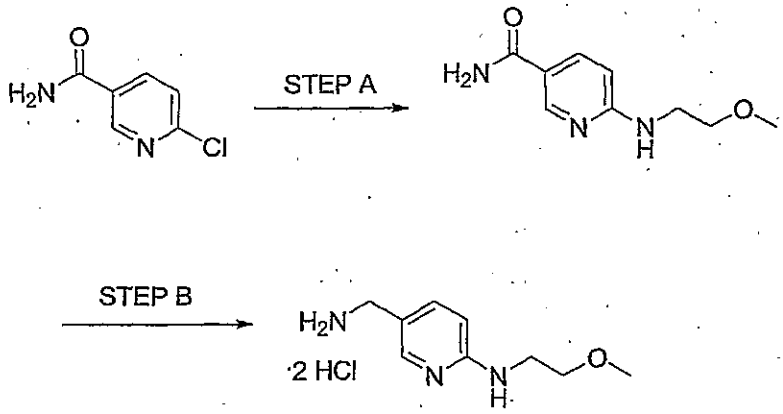
To 4-N-t-Butoxycarbonylaminopiperidine (0.8 g, 4.0 mmol) in CH_2Cl_2 (10 mL) at 0°C was added TEA (1.40 mL, 2.5 eq.) and 3-trifluoromethyl benzoyl chloride (1.05 g, 1.25 eq.). The resulting solution was stirred 15 minutes and warmed to room temperature and stirred 3 hours. The reaction mixture was diluted with CH_2Cl_2 and washed with 5% Na_2CO_3 (2 x 100 mL). The organic layer was dried over Na_2SO_4, filtered and concentrated to yield a pale yellow solid (quantitative crude yield).

PREPARATIVE EXAMPLE 205:



To a solution of the compound prepared in Preparative Example 204 (1.0 g, 2.76 mmol) in CH_2Cl_2 (15 mL) at 0°C was added TFA (8 mL) and the resulting solution was stirred at 0°C for 30 minutes and room temperature 1 hour. The reaction mixture was poured onto Na_2CO_3 (40 g) and H_2O (400 mL) added and the resulting mixture was extracted with CH_2Cl_2. The combined organics were dried over Na_2SO_4, filtered, and concentrated under reduced pressure. The crude product was purified by flash chromatography using a 20% (7N NH_3 in MeOH) solution in CH_2Cl_2 as eluent (0.6 g, 82% yield).

PREPARATIVE EXAMPLES 206:



STEP A:

To a solution of 6-chloronicotinamide (1g, 6.39 mmol) in isoamyl alcohol (15 mL) at rt was added Na_2CO_3 (0.81g, 7.67 mmol) followed by methoxyethylamine (0.67 mL, 7.67 mmol). The mixture was heat at 130 °C for 16h, cooled to rt, and was filtered thru a medium glass-fritted filter. The resulting filtrate was concentrated under reduced pressure and the resultant solid was triturated with Et_2O (2 x 10 mL). The crude solid was placed under high vacuum to afford 1.2 g (96%) of a light yellow solid. M+H = 196.

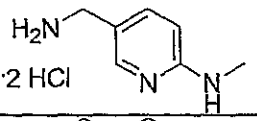
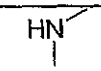
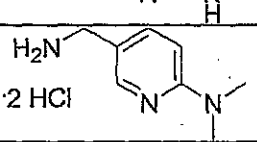
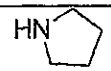
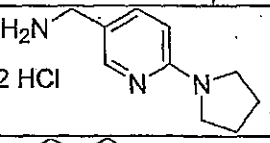
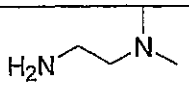
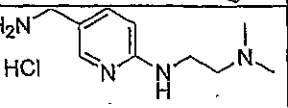
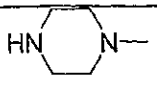
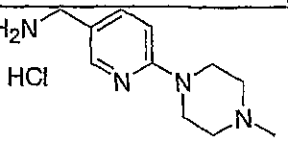
STEP B:

To a solution of amide (1.2 g, 6.12 mmol) from Preparative Example 206, Step A in THF (5 mL) at 0 °C was added a solution of BH_3-THF (43 mL; 43 mmol) dropwise over 10 min. The resultant solution was warmed to rt and stirred for 14 h. The mixture was cooled to 0 °C and was sequentially treated with 6M HCl (35 mL), water (30 mL), and MeOH (150 mL). The mixture was stirred for 8 h and was concentrated under reduced pressure. The crude residue was triturated with MeOH, concentrated under reduced pressure, and placed under high vacuum to afford 1.6 g (82%) of a white solid as the dihydrochloride salt. M+H (free base) = 182.0. This material was used crude in the coupling with 7-Cl adducts.

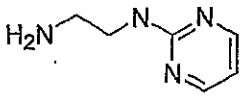
PREPARATIVE EXAMPLES 207-211:

By essentially the same known procedure set forth in Preparative Example 206 only by utilizing the amines shown in Column 2 of Table 13 and the amines shown in Column 3 of Table 13 were prepared:

TABLE 13

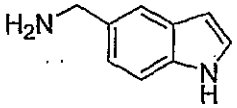
Prep.Ex.	Column 2 (Amine)	Column 3 (Amine)	CMPD M+H (free base)
207			M+H = 138
208			M+H = 152
209			M+H = 178
210			M+H = 195
211			M+H = 207

PREPARATIVE EXAMPLE 212:



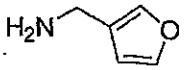
The above compound was prepared accordingly to the methods described in WO 91/18904.

PREPARATIVE EXAMPLE 213:



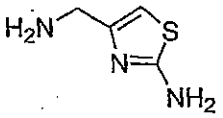
The above compound was prepared accordingly to the methods described in US 6,180,627 B1.

PREPARATIVE EXAMPLE 214:



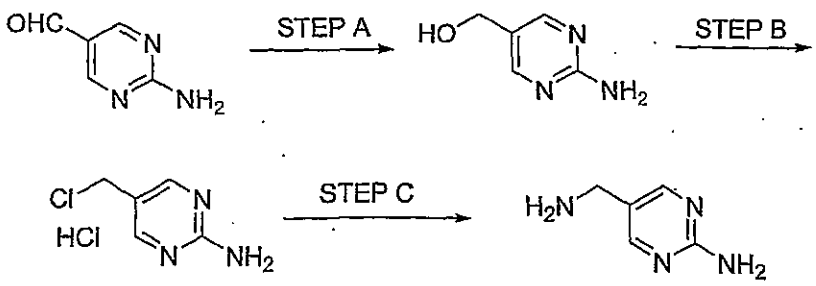
The known amine was prepared as described in *J. Med. Chem.* (2001), *44*, 4505-4508.

PREPARATIVE EXAMPLE 215:



The known amine was prepared as described in *J. Med. Chem.* (1997), *40*, 3726-3733.

PREPARATIVE EXAMPLE 216:



STEP A:

A solution of aldehyde (50 g, 0.41 mol) [WO 0232893] in MeOH (300 mL) was cooled to 0 °C and carefully treated with $NaBH_4$ (20g, 0.53 mol in 6 batches) over 20 minutes. The reaction was then allowed to warm to 20 °C and was stirred for 4 hours. The mixture was again cooled to 0 °C, carefully quenched with saturated aqueous NH_4Cl, and concentrated. Flash chromatography (5–10% 7N NH_3-MeOH/CH_2Cl_2) provided the primary alcohol (31g, 62%) as a light yellow solid.

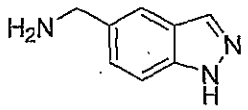
STEP B:

A slurry of alcohol (31 g, 0.25 mol) from Preparative Example 216, Step A in CH_2Cl_2 (500 mL) was cooled to 0 °C and slowly treated with $SOCl_2$ (55mL, 0.74 mol over 30 minutes). The reaction was then stirred overnight at 20 °C. The material was concentrated, slurried in acetone, and then filtered. The resulting beige solid was dried overnight in vacuo (38.4g, 52%, HCl salt).

STEP C:

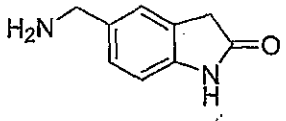
To a 15 mL pressure tube charged with a stir bar was added chloride (150 mg, 0.83 mmol) from Preparative Example 216, Step B followed by 7 M NH_3/MeOH (10 mL). The resulting solution was stirred for 48 h at rt where upon the mixture was concentrated under reduced pressure to afford a light yellow solid (0.146 g, 83%). M+H (free base) = 140.

PREPARATIVE EXAMPLE 217:



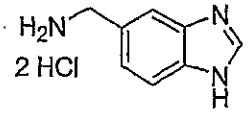
The above compound was prepared accordingly to methods described in WO 00/26210.

PREPARATIVE EXAMPLE 218:



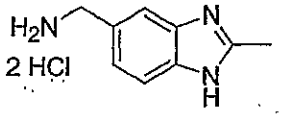
The above compound was prepared accordingly to methods described in WO 99/10325.

PREPARATIVE EXAMPLE 219:



The known amine dihydrochloride was prepared according to methods described in WO 02/64211.

PREPARATIVE EXAMPLE 220:



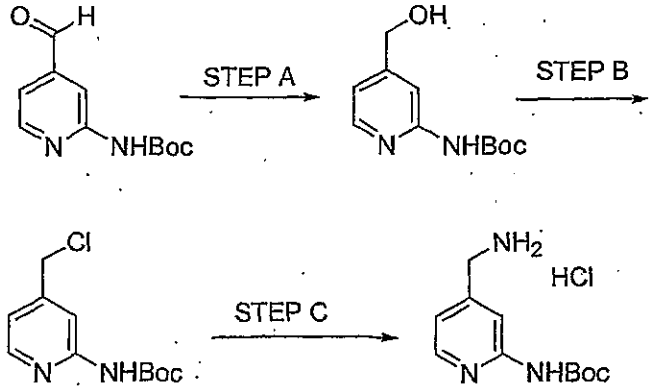
The above compound was prepared according to methods described in WO 02/64211.

PREPARATIVE EXAMPLE 221:



The known primary alcohol was prepared according to WO 00/37473 and was converted to the desired amine dihydrochloride in analogous fashion as Preparative Example 220 according to WO 02/064211.

PREPARATIVE EXAMPLE 222:



STEP A:

To a solution of aldehyde (WO 02/32893) (0.46 g, 2.07 mmol) in MeOH/THF (2 mL/2 mL) at 0 °C was added $NaBH_4$ (94 mg, 2.48 mmol) in one portion. The resulting mixture was stirred for 12 h at rt and was diluted with sat.

aq. NH_4Cl (3 mL). The mixture was concentrated under reduced pressure and the resultant aqueous layer was extracted with CH_2Cl_2 (3 x 5 mL). The organic layers were combined, washed with brine (1 x 5 mL), dried (Na_2SO_4), and filtered. The organic layer was concentrated under reduced pressure to afford 417 mg (90% yield) of a white solid. M+H = 225.

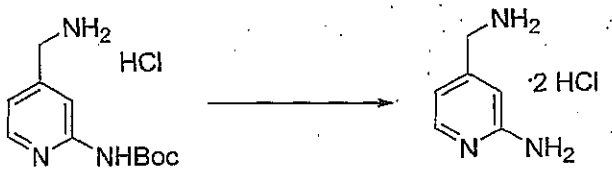
STEP B:

The crude alcohol from Preparative Example 222, step A (0.4 g, 1.78 mmol) in CH_2Cl_2 (4 mL) was added $SOCl_2$ (0.65 mL, 8.91 mmol) and the mixture was stirred for 2 h at rt. The mixture was concentrated under reduced pressure to afford 407 mg (94%) of a light yellow solid. M+H = 243. The crude product was taken on without further purification.

STEP C:

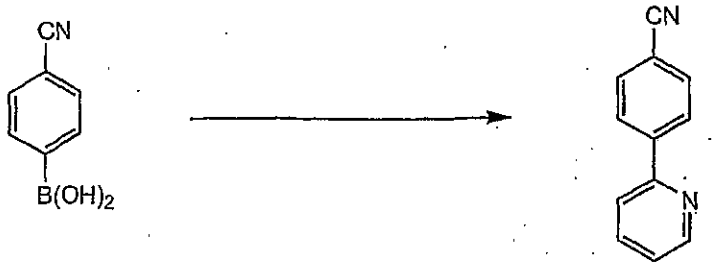
To a solution of crude chloride from Preparative Example 222, Step B (0.33 g, 1.36 mmol) in a pressure tube charged with 7M NH_3/MeOH (35 mL) and the mixture was stirred for 72 h. The mixture was concentrated under reduced pressure to afford 257 mg (85%) of a yellow semisolid. M+H (free base) = 224.

PREPARATIVE EXAMPLE 223:



To a round bottom flask charged with amine hydrochloride (0.24 g, 1.1 mmol) from Preparative Example 222 and a stir bar was added 4N HCl/dioxane (10 mL). The resulting solution was stirred for 12h at rt, concentrated under reduced pressure, and triturated with CH_2Cl_2 (3 x 5 mL). The crude product was filtered, washed with Et2O (2 x 5mL), and dried under high vacuum to afford 0.19g (91%) as the dihydrochloride salt. M+H (free base) = 124.

PREPARATIVE EXAMPLE 224:

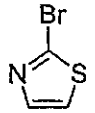
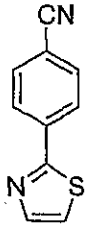
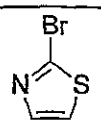
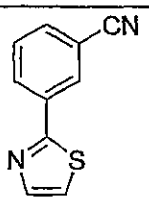


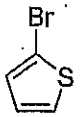
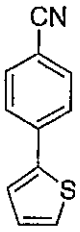
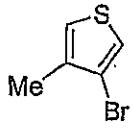
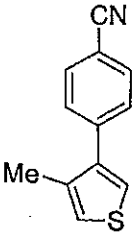
$Pd(PPh_3)_4$ (0.404 gm, 0.35 mmol) was added to a degassed solution of 4-cyanobenzene boronic acid (1.029 g, 7 mmol) and 2-bromopyridine (1.11 g, 7 mmol) in 75 mL acetonitrile. 0.4 M sodium carbonate solution (35 mL) was added to the reaction mixture and the resulting solution was refluxed at 90°C under Ar for 24 hours (progress of reaction was monitored by TLC). The reaction mixture was cooled and aqueous layer was separated. The organic layer containing the product and spent catalyst was mixed with silica gel (15 g) and concentrated to dryness. The 4-(2-pyridyl)-benzonitrile was isolated by column chromatography (0.850 g, 68%). LCMS: MH^+ = 181; ^1H NMR ($CDCl_3$) δ 8.85 (d, 1H), 8.7 (dd, 1H), 7.9 (dd, 1H), 7.75 (d, 2H), 7.7 (d, 2H), 7.4 (dd, 1H).

PREPARATIVE EXAMPLES 225-228:

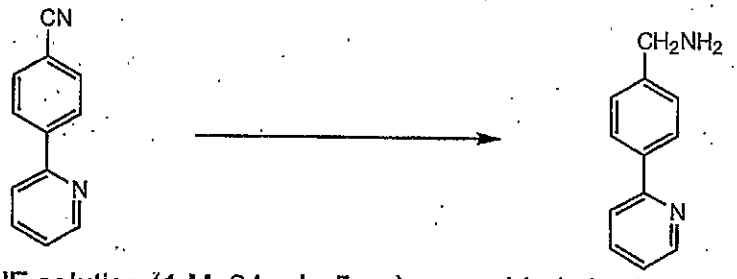
By following essentially same procedure described in Preparative Example 224, only substituting the bromides in column 2 of Table 14, compounds in column 3 of Table 14 were prepared.

Table 14

Prep. Ex.	Column 2	Column 3	Column 4
225			Yield = 70% LCMS: MH^+ = 187
226			Yield = 60% LCMS: MH^+ = 187

227			Yield = 70% LCMS: MH^+ = 186
228			Yield = 70% LCMS: MH^+ = 200

PREPARATIVE EXAMPLE 229:

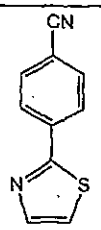
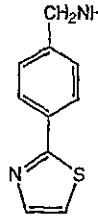
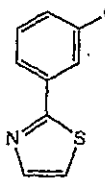
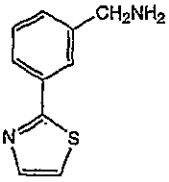
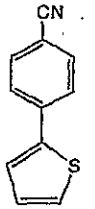
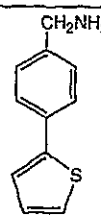
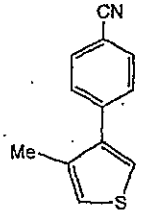
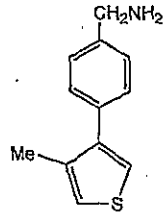


BH_3-THF solution (1 M, 24 mL, 5 eq) was added slowly to a stirring solution of 4-(2-pyridyl)-benzonitrile (0.85 g, 4.72 mmol) in anhydrous THF (25 mL) under Ar, and the resulting solution was refluxed for about 12 hr. The solution was cooled to 0°C using ice-water. Methanol (15 mL) was added drop-wise to the cold reaction mixture and stirred for 1 h to destroy excess BH_3. Added HCl – methanol (1M, 10 mL) slowly to the reaction mixture and refluxed for 5 h. Concentrated the solution to dryness and the residue was dissolved in 25 mL water and extracted with ether to remove any un-reacted material. The aqueous solution was neutralized with solid potassium carbonate to pH 10-11. The free amine, thus formed was extracted with ether, dried over potassium carbonate (0.45 g, 50%). LCMS: MH+ = 185; ^1H NMR ($CDCl_3$) δ 8.85 (d, 1H), 8.7 (dd, 1H), 7.9 (dd, 1H), 7.75 (d, 2H), 7.7 (d, 2H), 7.4 (dd, 1H), 3.7 (t, 2H), 1.7 (t, 2H).

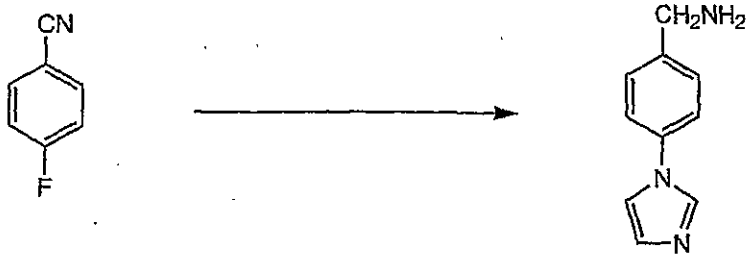
PREPARATIVE EXAMPLES 230-233:

By following essentially the same procedure set forth in Preparative Example 229, compounds in column 3 of Table 15 were prepared.

Table 15

Prep. Ex.	Column 2	Column 3	Column 4
230			Yield = 60% LCMS: MH^+ = 191
231			Yield = 60% LCMS: MH^+ = 191
232			Yield = 70% LCMS: MH^+ = 190
233			Yield = 70% LCMS: MH^+ = 204

PREPARATIVE EXAMPLE 234:



Step A:

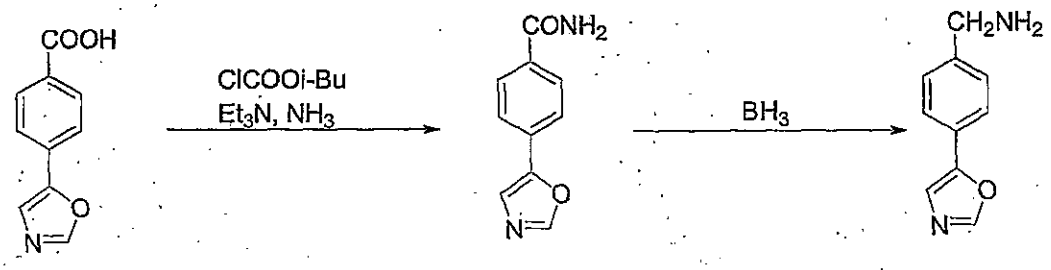
A mixture 4-fluorobenzonitrile (3 g, 25 mmol) and imidazolyl sodium (2.48 g, 27.5 mmol) in DMF (50 mL) was stirred at 80°C under Ar for 12 h. Progress of reaction was monitored by TLC. The reaction mixture was concentrated *in vacuo* and the residue was diluted with 50 mL water and stirred. The aqueous mixture was extracted with EtOAc (2 x 50 mL). Combined EtOAc extracts was dried over

anhydrous MgSO4, concentrated, and the 4-(1-imidazolyl)-benzonitrile was isolated by column chromatography (3.6 g, 78%). LCMS: MH^+ = 170; ^1H NMR ($CDCl_3$) δ 8.0 (s, 1H), 7.5 (d, 2H), 7.4 (m, 3H), 7.3 (d, 1H)

Step B:

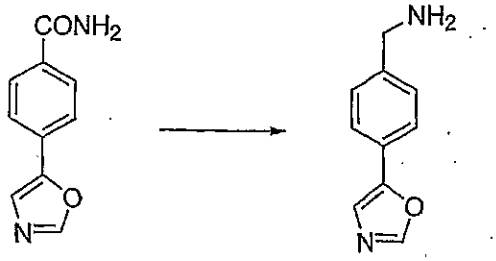
4-(1-imidazolyl)-benzonitrile (1g, 5.92 mmol) was dissolved in anhydrous THF (10 mL) and added drop-wise to a stirring solution of LAH –THF (1 M in THF, 18 mL) at room temperature. The reaction mixture was refluxed under Ar for 2 h and the progress was monitored by TLC. The mixture was cooled to 0°C and quenched by drop-wise addition of a saturated $Na_2SO_4 – H_2O$ solution. The mixture was stirred for 1 h and filtered to remove lithium salts. The filtrate was dried over anhydrous $MgSO_4$ and concentrated to obtain 4-(1-imidazolyl)-benzylamine (0.8 g, 80%). LCMS: MH^+ = 174.

PREPARATIVE EXAMPLE 235:



A mixture of 4-(5-oxazolyl)benzoic acid (1.0 g, 5.46 mmol) and Et_3N (552 mg, 5.46 mmol) in 25 mL of THF was cooled to 0 °C and ClCOO*i*-Bu (745 mg, 5.46 mmol) was added dropwise. After the addition was over, the reaction mixture was stirred for additional 5 min and then aq NH_4OH (0.63 mL of 28% solution, 10.46 mmol) was added. After overnight stirring, the solvent was evaporated, the residue was taken up in water and basified to pH 9. The precipitated solid was filtered, washed with water and dried over P_2O_5 in a vacuum desiccator to provide 500 mg (48%) of the 4-(5-oxazolyl)-benzamide: ^1H NMR (DMSO-d6) δ 8.50 (s, 1H), 8.20-7.80 (m, 5H).

PREPARATIVE EXAMPLE 236:

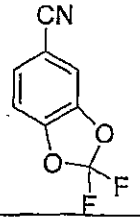
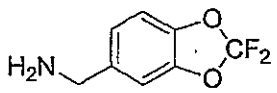
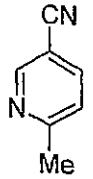
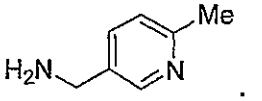


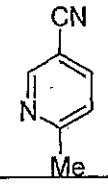
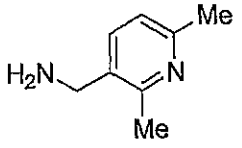
A suspension of 4-(5-oxazolyl)benzamide (500 mg, 2.657 mmol) in 10 mL of dry THF was cooled to 0 °C and 10 mL of 1 M BH_3.THF (10.00 mmol) was added. The contents were refluxed overnight and the excess borane was destroyed by dropwise addition of methanol. The solvent was evaporated and the residue was treated with methanolic HCl to decompose the amine-borane complex. After evaporation of the methanol, the residue was taken in water, basified to pH 10 and the product was extracted in to DCM. The DCM layer was dried (K_2CO_3) and the solvent was removed to provide 150 mg (32%) of 4-(5-oxazolyl)benzylamine: ^{1}H NMR ($CDCl_3$) δ 7.90 (s, 1H), 7.60 (d, 2H), 7.40 (d, 2H), 7.30 (s, 1H), 3.90 (s, 2H).

PREPARATIVE EXAMPLES 237-239:

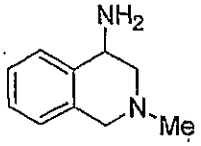
By essentially the same procedures set forth above, the compounds in Column 2 of Table 16 were reduced using the method indicated in Column 3 of Table 16 to give the amine indicated in Column 4 of Table 16.

Table 16

Prep. Ex.	Column 2	Column 3	Column 4	CMPD
237		BH_3		^{1}H NMR ($CDCl_3$) δ 7.15-6.90 (m, 3H), 3.85 (s, 2H), 1.45 (s, 2H)
238		H_2		^{1}H NMR ($CDCl_3$) δ 8.40(s, 1H), 7.55 (dd, 1H), 7.10 (d, 1H), 3.85 (s, 2H), 2.50 (s, 3H), 1.70 (bs, 2 H)

239		BH_3		

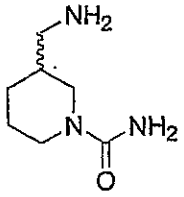
PREPARATIVE EXAMPLE 240



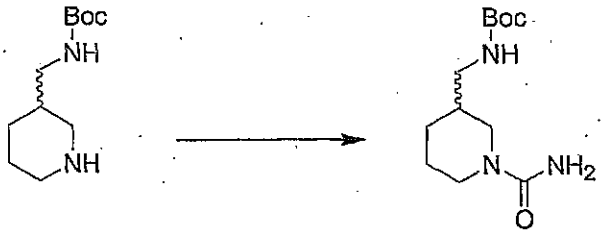
Prepared by the literature procedure (PCT Int. Appl, WO 0105783): ^{1}H NMR ($CDCl_3$) δ 7.35 (d, 1H), 7.24-7.10 (m, 2 H), 7.02 (d, 1H), 3.95 (t, 1H), 3.70 (d, 1H), 3.37 (d, 1H), 2.65 (m, 2H), 2.45 (s, 3H), 1.90 (bs, 2H)

PREPARATIVE EXAMPLE 241:

3-(AMINOMETHYL)PIPERIDINE-1-CARBOXAMIDE



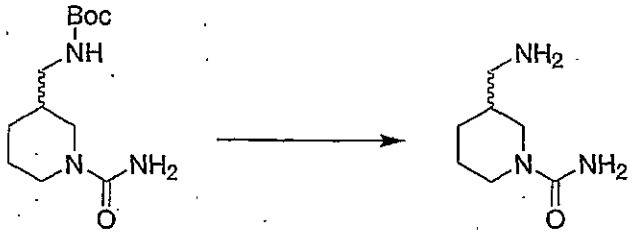
A. 3-(*tert*-BUTOXYCARBONYLAMINOMETHYL)PIPERIDINE-1-CARBOXAMIDE



3(R/S)-(*tert*-Butoxycarbonylaminomethyl)piperidine (3g, 14.0mmoles) was dissolved in anhydrous dichloromethane (50mL) and trimethylsilylisocyanate (9.68g, 11.4mL, 84.0mmoles) was added. The mixture was stirred under argon at 25°C for 68h. Additional trimethylsilylisocyanate (4.84g, 5.7mL, 42.0mmoles) was added and the mixture was stirred at 25°C for a total of 90h. The mixture was evaporated to dryness and chromatographed on a silica gel column (30x5cm) using 2% (10% conc. ammonium hydroxide in methanol)-dichloromethane as the eluant to give 3-(*tert*-

butoxycarbonylaminomethyl)piperidine-1-carboxamide (3.05g, 85%): FABMS: m/z 258.1 (MH^+); HRFABMS: m/z 258.1816 (MH^+). Calcd. for $C_{12}H_{24}O_3N_3$: m/z 258.1818; δ_H ($CDCl_3$) 1.22 91H, m, CH_2), 1.42 (9H, s, -COOC(CH_3)$_3$), 1.48 (1H, m, CH_2), 1.67 (2H, m, CH_2), 1.78 (1H, m, CH), 2.80 (1H, m, CH_2), 2.99, 3H, m, CH_2), 3.59 (1H, m, $CH_2$0 3.69 (1H, m, CH_2), 4.76 (2H, bm, $CONH_2$) and 4.98ppm (1H, bm, NH); δ_C ($CDCl_3$) CH_3: 28.5, 28.5, 28.5; CH_2: 24.0, 28.3, 43.2, 45.1, 47.8; CH: 36.5; C: 79.4, 156.3, 158.5.

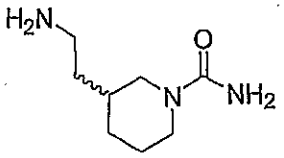
B. 3-(AMINOMETHYL)PIPERIDINE-1-CARBOXAMIDE



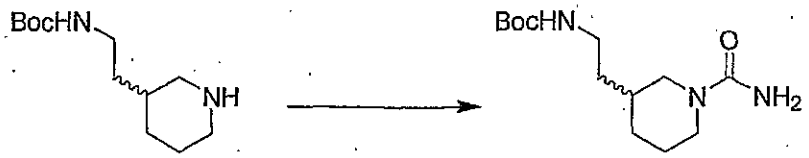
3-(*tert*-Butoxycarbonylaminomethyl)piperidine-1-carboxamide (150mg, 0.583mmoles) (prepared as described in Preparative Example 241, Step A above) was dissolved in methanol (3mL). 10% conc. sulfuric acid in 1,4-dioxane (7.9mL) was added and the mixture was stirred at 25°C for 1h. The mixture was diluted with methanol and BioRad AG1-X8 resin (OH^- form) was added until the pH was basic. The resin was filtered off, washed with methanol, evaporated to dryness and chromatographed on a silica gel column (15x2cm) using dichloromethane followed by 15% (10% conc, ammonium hydroxide in methanol)-dichloromethane as the eluant to give the 3-(aminomethyl)piperidine-1-carboxamide (80mg, 87%): FABMS: m/z 158.1 (MH^+); HRFABMS: m/z 158.1294 (MH^+). Calcd. for $C_7H_{16}N_3O$: m/z 158.1293; δ_H ($CDCl_3$ + drop CD_3OD) 1.20 (1H, m, CH_2), 1.48 (1H, m, CH_2), 1.60 (1H, m, CH), 1.68 (1H, m, CH_2), 1.83 (1H, m, CH_2), 2.64 (bm, 2H, -CH_2NH_2), 2.82 (1H, m, CH_2), 3.02 (1H, m, CH_2), 2.98 (2H, m, CH_2), 3.70 (1H, m, -CH_2NH_2), 3.78 (1H, m, -CH_2NH_2) and 5.24 ppm (1H, bs, NH); δ_C ($CDCl_3$ + drop CD_3OD) CH_2: 24.1, 28.6, 44.0, 44.8, 47.9; CH: 38.3; C: 159.0.

PREPARATIVE EXAMPLE 242:

3-(2-AMINOETHYL)PIPERIDINE-1-CARBOXAMIDE

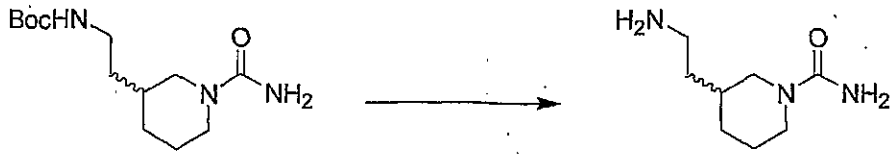


A. 3-(2-tert-BUTOXYCARBONYLAMINOETHYL)PIPERIDINE-1-CARBOXAMIDE



3-(2-*tert*-Butoxycarbonylaminoethyl)piperidine (500mg, 2.19mmoles) was dissolved in anhydrous dichloromethane (10mL) and trimethylsilylisocyanate (2.96mL, 21.9mmoles) was added. The mixture was stirred under argon at 25°C for 3.35h. The mixture was diluted with dichloromethane and washed with saturated aqueous sodium bicarbonate. The organic layer was dried ($MgSO_4$), filtered, evaporated to dryness and chromatographed on a silica gel column (15x5cm) using 5% (10% conc. ammonium hydroxide in methanol)-dichloromethane as the eluant to give 3-(2-*tert*-butoxycarbonylaminoethyl)piperidine-1-carboxamide (417.7mg, 70%): FABMS: m/z 272.0 (MH^+); HRFABMS: m/z 272.1979 (MH^+). Calcd. for $C_{13}H_{26}O_3$: m/z 272.1974; δ_H ($CDCl_3$) 1.16 (1H, m, CH_2), 1-30-1.60 (5H, m, CH/CH_2), 1.46 (9H, s, -COOC($C\underline{H}_3$)$_3$), 1.68 (1H, m, CH_2), 1 84 (1H, m, CH_2), 2.54 (1H, dd, CH_2), 2.73 (1H, m, CH_2), 3.08 (1H, m, CH_2), 3.42 (1H, m, CH_2), 4.02 (1H, m, CH_2), 4.10 (1H, m, CH_2), 4.84 (1H, m, NH) and 4.96 ppm (2H, bm, $CONH_2$); δ_C ($CDCl_3$) CH_3: 28.5, 28.5, 28.5; CH_2: 25.2, 31.7, 34.9, 37.3, 44.6, 50.3; CH: 32.9; C: 79.5, 156.4, 158.2.

B. 3-(2-AMINOETHYL)PIPERIDINE-1-CARBOXAMIDE

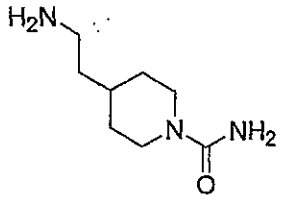


3-(2-*tert*-Butoxycarbonylaminoethyl)piperidine-1-carboxamide (392.7mg, 1.45mmoles) (prepared as described in Preparative Example 242, Step A above) was dissolved in methanol (7.5mL) and 10% conc. sulfuric acid in 1,4-dioxane (19.5mL) was added. The mixture was stirred at 25°C for 1.25h. The mixture was

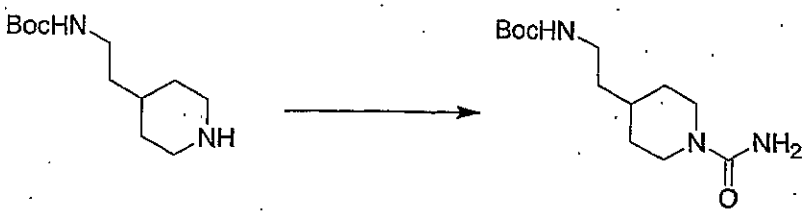
diluted with methanol and BioRad AG1-X8 resin (OH^- form) was added until the pH was basic. The resin was filtered off, washed with methanol, evaporated to dryness and chromatographed on a silica gel column (30x2.5cm) using 15% (10% conc, ammonium hydroxide in methanol)-dichloromethane as the eluant to give 3-(2-aminoethyl)piperidine-1-carboxamide (233mg, 94%): FABMS: m/z 172.1 (MH^+); HRFABMS: m/z 172.1444(MH^+). Calcd for $C_8H_{18}N_3O$ requires: m/z 172.1450; δ_H ($CDCl_3$ + 3% CD_3OD) 1.14 (1H, m, CH_2), 1.40 (2H, m, CH_2), 1.49 (1H, m, CH), 1.58 (1H, m, CH_2), 1.69 (1H, m, CH_2), 1.85 (1H, m, CH_2), 2.55 (1H, m, CH_2), 2.67 (5H, m, CH_2/NH_2), 2.76 (1H, bm, CH_2), 2.84 (1H, m, CH_2) and 3.82 ppm (2H, m, $CONH_2$); δ_C ($CDCl_3$ + 3% CD_3OD) CH_2: 24.8, 30.9, 36.6, 38.9, 44.9, 50.0; CH: 33.4.

PREPARATIVE EXAMPLE 243:

4-(2-AMINOETHYL)PIPERIDINE-1-CARBOXAMIDE



A. 4-(2-tert-BUTOXYCARBONYLAMINOETHYL)PIPERIDINE-1-CARBOXAMIDE



4-(2-*tert*-Butoxycarbonylaminoethyl)piperidine (500mg, 2.19mmoles) was dissolved in anhydrous dichloromethane (10mL) and trimethylsilylisocyanate (2.96mL, 21.9mmoles) was added. The mixture was stirred under argon at 25°C for 3.25h. The mixture was diluted with dichloromethane and washed with saturated aqueous sodium bicarbonate. The organic layer was dried ($MgSO_4$), filtered, evaporated to dryness and chromatographed on a silica gel column (15x5cm) using 5% (10% conc. ammonium hydroxide in methanol)-dichloromethane as the eluant to give 4-(2-*tert*-butoxycarbonylaminoethyl)piperidine-1-carboxamide (308.2mg, 52%): FABMS:

m/z 272.0 (MH^+); HRFABMS: m/z 272.1965 (MH^+). Calcd. for $C_{13}H_{26}O_3N_3$: m/z 272.1974; δ_H ($CDCl_3$) 1.20 (2H, m, CH_2), 1.47 (9H, s, -COOC(CH_3)$_3$), 1.45-1.55 (3H, m, CH/CH_2), 1.75 (2H, m, CH_2), 2.82 (2H, m, CH_2), 3.19 (2H, m, CH_2), 3.96 (2H, m, CH_2), 4.64 (2H, m, CH_2) and 4.70 ppm (1H, bm, NH); δ_C ($CDCl_3$) CH_3: 28.5, 28.5, 28.5; CH_2: 31.8, 31.8, 36.7, 38.0, 44.5, 44.5; CH: 33.4; C: 79.2, 156.7, 158.1.

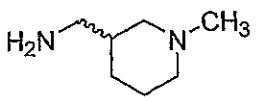
A. 3-(2-AMINOETHYL)PIPERIDINE-1-CARBOXAMIDE



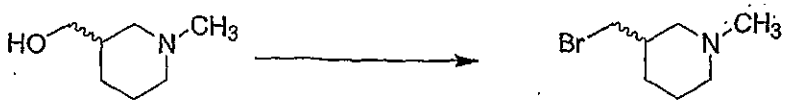
4-(2-*tert*-Butoxycarbonylaminoethyl)piperidine-1-carboxamide (283.3mg, 1.04mmoles) (prepared as described in Preparative Example 243, Step A above) was dissolved in methanol (5.4mL) and 10% conc. sulfuric acid in 1,4-dioxane (14.2mL) was added and the mixture was stirred at 25°C for 1.25h. The mixture was diluted with methanol and BioRad AG1-X8 resin (OH^- form) was added until the pH was basic. The resin was filtered off, washed with methanol, evaporated to dryness and chromatographed on a silica gel column (30x2.5cm) using 15% (10% conc, ammonium hydroxide in methanol)-dichloromethane as the eluant to give the 3-(2-aminoethyl)piperidine-1-carboxamide (170mg, 95%): FABMS: m/z 172.1 (MH^+); HRFABMS: m/z 172.1442. Calcd for $C_8H_{18}N_3O$ requires: m/z 172.1450; δ_H ($CDCl_3$ + 3% CD_3OD) 1.16 (2H, m, CH_2), 1.43 (2H, m, CH_2), 1.52 (1H, m, CH), 1.70 (2H, m, CH_2), 2.70-2.85 (8H, m, CH_2) and 3.92 ppm (2H, m, $CONH_2$); δ_C ($CDCl_3$ + 3% CD_3OD) CH_2: 31.9, 31.9, 39.0, 39.7, 44.4, 44.4; CH: 33.5; C: 158.7.

PREPARATIVE EXAMPLE 244:

3-(AMINOMETHYL)-1-METHYLPIPERIDINE



A. 3-(BROMOMETHYL)-1-METHYLPIPERIDINE



3-(Hydroxymethyl)-1-methylpiperidine (2g, 15.5mmoles) was dissolved in anhydrous acetonitrile (32mL) and anhydrous pyridine (2.02mL, 24.8mmoles) was added and the solution was cooled to 0°C. Dibromotriphenylphosphorane (8.49g, 20.2mmoles) was added at 0°C and the mixture was allowed to warm up to 25°C and was stirred for 94h. The mixture was evaporated to dryness and the residue was chromatographed on a silica gel column (30x5cm) using gradient elution with dichloromethane, 35% diethyl ether in dichloromethane and 5-10% methanol in dichloromethane as the eluant to give 3-(bromomethyl)-1-methylpiperidine (3.13g, 100%): FABMS: m/z 192.1 (MH^+); δ_H ($CDCl_3$) 1.52 (1H, m, CH_2), 1.99 (2H, m, CH_2), 2.43 (1H, m, CH_2), 2.75 (2H, m, CH_2), 2.82 (1H, m, CH), 2.86/2.88 (3H, s, NCH_3), 3.42/3.49 (2H, dd, -$C\underline{H}_2$Br) and 3.56 ppm (2H, m, CH_2); δ_C ($CDCl_3$) CH_3: 44.3; CH_2: 22.1, 26.6, 35.4, 54.8, 58.2; CH: 34.6.

A. 3-(*Di-tert*-BUTOXYCARBONYLAMINOMETHYL)-1-METHYLPIPERIDINE



3-(Bromomethyl)-1-methylpiperidine (1.5g, 7.81mmoles) (from Preparative Example 244, Step A above) and di-*tert*-butyliminodicarboxylate (1.697g, 7.81mmoles) were dissolved in anhydrous acetonitrile (25mL). Cesium carbonate (5.1g, 15.6mmoles) and lithium iodide (52mg, 0.391mmoles) were added and the mixture was stirred at 70°C for 20h. The mixture was evaporated to dryness and the residue was partitioned between dichloromethane and saturated aqueous sodium bicarbonate. The organic layer was dried ($MgSO_4$), filtered and evaporated to dryness. the residue was chromatographed on a silica gel column (30x5cm) using 3% methanol in dichloromethane as the eluant to

give 3-(di-*tert*-butoxycarbonylamino)-1-methylpiperidine (1.331g, 52%): FABMS: m/z 329.2 (MH^+); HRFABMS: m/z 329.2438 (MH^+). Calcd. for $C_{17}H_{33}N_2O_4$: m/z 329.2440; δ_H ($CDCl_3$) 1.10 (1H, m, CH_2), 1.54 (18H, s, -$COOC(CH_3)_3$), 1.86 (2H, m, CH_2), 2.01 (1H, m, CH_2), 2.19 (1H m, CH), 2.34 (2H, bm, CH_2), 2.59 (3H, -NCH_3), 3.19 (2H, m, CH_2) and 3.52/3.52 ppm (2H, -CH_2N-); δ_C ($CDCl_3$) CH_3: 28.5, 28.5, 28.5, 28.5, 28.5, 28.5, 47.2; CH_2: 25.4, 28.3, 50.4, 56.8, 60.8; CH: 37.2; C: 83.0, 83.0,153.5, 153.5.

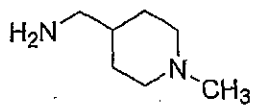
A. 3-(AMINOMETHYL)-1-METHYLPIPERIDINE



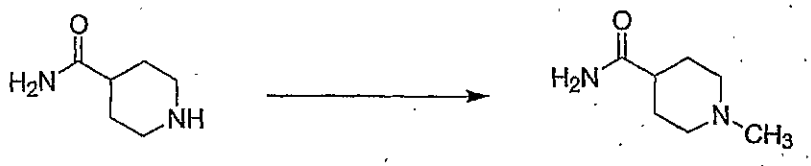
3-(Di-*tert*-butoxycarbonylamino)-1-methylpiperidine (500mg, 1.52mmoles) (from Preparative Example 244, Step B above) was dissolved in methanol (7.5mL) and 10% (v/v) conc. sulfuric acid in 1,4-dioxane (19.75mL) was added. The solution was stirred at 25°C for 0.5h. Methanol (300mL) was added, followed by BioRad AG1-X8 resin (OH^- form) until the pH was ~10. The resin was filtered off and washed with methanol (2x200mL). The combined eluates were evaporated to dryness and the residue was chromatographed on a silica gel column (30x2.5cm) using 10% (10% conc. ammonium hydroxide in methanol)-dichloromethane as the eluant to give 3-(aminomethyl)-1-methylpiperidine (69.2mg, 35%): FABMS: m/z 129.1 (MH^+); HRFABMS: m/z 129.1392 (MH^+). Calcd. for $C_7H_{17}N_2$: m/z 129.1392; δ_H ($CDCl_3$) 0.90 (2H, m, CH_2), 1.65 (2H, m, CH_2), 1.72 (1H, m, CH), 1.79 (1H, m, CH_2), 1.91 (1H, m, CH_2), 2.30 (3H, s, -NCH_3), 2.64 (2H, m, CH_2), 2.82 (1H, m, -$C\underline{H}_2NH_2$) and 2.92 ppm (1H, m, -$C\underline{H}_2NH_2$); δ_C ($CDCl_3$) CH_3: 46.7; CH_2: 25.2, 28.0, 46.3, 56.4, 60.3; CH: 39.9.

PREPARATIVE EXAMPLE 245:

4-(AMINOMETHYL)-1-METHYLPIPERIDINE

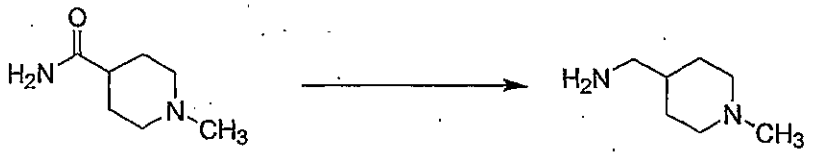


A. 1-METHYLISONIPECOTAMIDE



Isonipecotamide (10g, 78.0mmoles) was dissolved in distilled water (100mL) and 37% aqueous formaldehyde (7.6mL, equivalent to 2.81g HCHO, 93.6mmoles) was added. Wet 10% Pd-C (8 spoon spatulas) was added under argon and the mixture was hydrogenated at 25°C and 50psi for 43h. The catalyst was filtered off through Celite and the latter was washed with water and methanol. The combined filtrates were evaporated to dryness and the residue was chromatographed on a silica gel column (60x5cm) using 8%-10%-20% (10% conc. ammonium hydroxide in methanol)-dichloromethane as the eluant to give 1-methylisonipecotamide (7.15g, 64%): FABMS: m/z 143.1 (MH^+); HRFABMS: m/z 143.1184 (MH^+). Calcd. for $C_7H_{15}N_2O$: m/z 143.1184; δ_H (d_6-DMSO) 1.50/1.57 (4H, m, CH_2), 1.76/1.94 (4H, m, CH_2), 2.10 (3H, s, $-NCH_3$), 2.72 (1H, m, CH) and 6.68/7.18 ppm (2H, m, $CONH_2$); δ_C (d_6-DMSO) CH_3: 41.2; CH_2: 28.5, 28.5, 54.9, 54.9; CH: 46.2; C: 176.7.

B. 4-(AMINOMETHYL)-1-METHYLPIPERIDINE

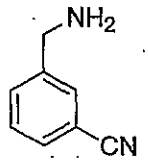


1-Methylisonipecotamide (6.75g, 47.5mmoles) (prepared as described in Preparative Example 245, Step A above) was dissolved in anhydrous THF (350mL) and the resulting mixture was added in portions to a stirred slurry of lithium aluminum hydride (1.8g, 47.5mmoles) in anhydrous THF (100mL) at 0°C under nitrogen. The mixture was stirred at 0°C for 30min and then heated at 66°C for 25h under nitrogen. Distilled water (1.88mL) was added dropwise to the

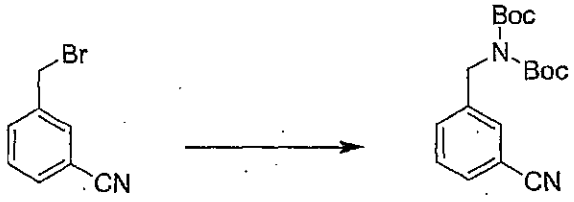
stirred mixture at 0°C, followed by 20% aqueous sodium hydroxide (1.42mL) and then distilled water (6.75mL) and the mixture was stirred for 15min. The mixture was filtered and the solids were washed with THF and dichloromethane. The combined filtrates were evaporated to dryness and chromatographed on a silica gel column (30x5cm) using 15%-20% (10% conc. ammonium hydroxide in methanol)-dichloromethane as the eluant to give 4-(aminomethyl)-1-methylpiperidine (0.678g, 11%): FABMS: m/z 129.1 (MH^+); HRFABMS: m/z 129.1389 (MH^+). Calcd. for $C_7H_{17}N_2$: m/z 129.1392; δ_H (d_6-DMSO): 2.08ppm (3H, s, $-NCH_3$); δ_C (d_6-DMSO): CH_3: under DMSO peaks; CH_2: 29.6, 29.6, 46.7, 55.2, 55.2; CH: 46.2.

PREPARATIVE EXAMPLE 246:

3-(AMINOMETHYL)BENZONITRILE



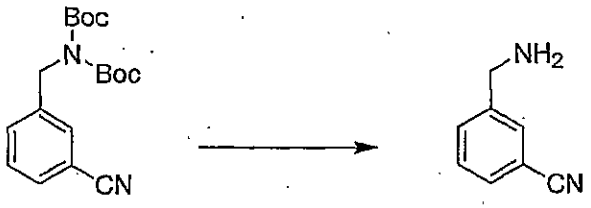
A. 3-(Di-*tert*-BUTOXYCARBONYLAMINO)BENZONITRILE



3-(Bromomethyl)benzonitrile (5g, 25.5mmoles) and di-tert-butyliminodicarboxylate (5.54g, 25.5mmoles) were dissolved in anhydrous THF (50mL) and cesium carbonate (16.62g, 25.5mmoles) and lithium iodide (170.5mg, 1.275mmoles) were added. The mixture was stirred at 70°C for 22h and the reaction was worked up as described in Preparative Example 89, Step B above. The residue was chromatographed on a silica gel column (60x5cm) using 5% ethyl acetate in hexane as the eluant to give 3-(di-*tert*-butoxycarbonylamino)benzonitrile (7.39g, 87%): FABMS: m/z 333.2 (MH^+);

HRFABMS: m/z 333.1815 (MH^+); Calcd. for $C_{18}H_{25}N_2O_4$: m/z 333.1814; δ_H ($CDCl_3$) 1.52 (18H, s, -COOC(CH_3)$_3$), 4.84 (2H, s, CH_2), 7.48 (1H, m, Ar-H), 7.60 (2H, m, Ar-H) and 7.65 ppm (1H, m, Ar-H); δ_C ($CDCl_3$) CH_3: 28.1, 28.1, 28.1, 28.1, 28.1, 28.1; CH_2: 48.4; CH: 129.2, 131.0, 131.0, 131.9; C: 83.2, 83.2, 112.5, 118.8, 140.1, 152.5, 152.5.

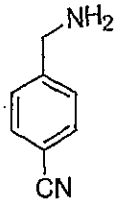
B. 3-(AMINOMETHYL)BENZONITRILE



3-(Di-*tert*-butoxycarbonylamino)benzonitrile (2g, 6.0mmoles) (prepared as described in Preparative Example 246, Step A above) was dissolved in methanol (30mL) and 10%(v/v) (10% conc. sulfuric acid in 1,4-dioxane) (79mL) was added. The solution was stirred at 25°C for 0.25h and worked up as described in Preparative Example 89, Step C above). The residue was chromatographed on a silica gel column (15x5cm) using 3% (10% conc. ammonium hydroxide in methanol)-dichloromethane as the eluant to give the title compound (651.4mg, 82%): FABMS: m/z 133.1 (MH^+); HRFABMS: m/z 133.0762 (MH^+). Calcd. for $C_8H_9N_2$: m/z 133.0766 ; δ_H ($CDCl_3$) 2.57 (2H, s, -$CH_2\underline{NH_2}$), 3.92 (2H, s, -$C\underline{H_2}NH_2$), 7.46 (1H, m, Ar-H), 7.57 (2H, m, Ar-H) and 7.64 ppm (1H, m, Ar-H); δ_C ($CDCl_3$) CH_2: 45.2; CH: 129.4, 130.7, 130.7, 131.8; C: 112.4, 118.8, 143.8.

PREPARATIVE EXAMPLE 247:

4-(AMINOMETHYL)BENZONITRILE

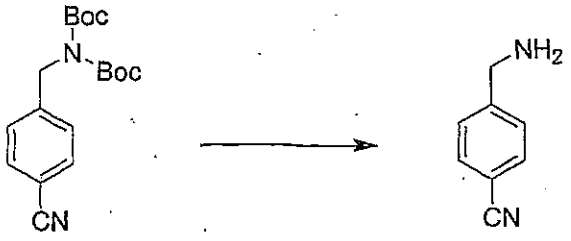


A. 3-(Di-*tert*-BUTOXYCARBONYLAMINOMETHYL)BENZONITRILE



4-(Bromomethyl)benzonitrile (5g, 25.5mmoles) and di-tert-butyliminodicarboxylate (5.54g, 25.5mmoles) were dissolved in anhydrous THF (50mL) and cesium carbonate (16.62g, 25.5mmoles) and lithium iodide (170.5mg, 1.275mmoles) were added. The mixture was stirred at 70°C for 23h and the reaction was worked up as described in Preparative Example 244, Step B above. The residue was chromatographed on a silica gel column (50x5cm) using 5% ethyl acetate in hexane as the eluant to give 4-(di-*tert*-butoxycarbonylaminomethyl)benzonitrile (7.07g, 83%): FABMS: m/z 333.2 (MH^+); HRFABMS: m/z 333.1816 (MH^+). Calcd. for $C_{18}H_{25}N_2O_4$: m/z 333.1814; δ_H ($CDCl_3$) 1.45 (18H, s, -COOC(CH_3)$_3$), 4.81 (2H, s, CH_2), 7.37 (2H, d, Ar-H) and 7.62 ppm (2H, d, Ar-H); δ_C ($CDCl_3$) CH_3: 28.1, 28.1, 28.1, 28.1, 28.1, 28.1; CH_2: 49.2 ; CH: 127.8, 127.8, 132.3, 132.3; C: 83.2, 83.2, 111.1, 118.9, 144.1, 152.4, 152.4.

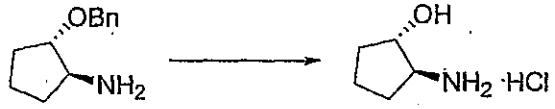
B. 4-(AMINOMETHYL)BENZONITRILE



4-(Di-*tert*-butoxycarbonylaminomethyl)benzonitrile (2g, 6.0mmoles) (prepared as described in Preparative Example 247, Step A above) was dissolved in TFA (4mL) and the solution was stirred at 25°C for 0.25h. The

reaction mixture was diluted with dichloromethane and extracted with 1N sodium hydroxide. The organic layer was dried ($MgSO_4$), filtered and evaporated to dryness. The residue was chromatographed on a silica gel column (15x5cm) using 3% (10% conc. ammonium hydroxide in methanol)-dichloromethane as the eluant to give 4-(aminomethyl)benzonitrile (108mg, 68%): FABMS: m/z 133.1 (MH^+); HRFABMS: m/z133.0764 (MH^+). Calcd. for $C_8H_9N_2$: m/z 133.0766; δ_H ($CDCl_3$) 2.04 (2H, s, -$CH_2N\underline{H_2}$), 3.89 (2H, s, -$C\underline{H_2}NH_2$), 7.40 (2H , d, Ar-H) and 7.59 ppm (2H, d, Ar-H); δ_C ($CDCl_3$) CH_2: 45.7; CH: 127.8, 127.8, 132.4, 132.4; C: 110.6, 118.9, 148.0.

PREPARATIVE EXAMPLE 248



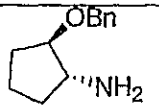
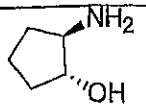
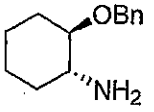
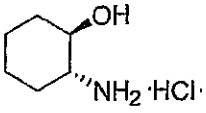
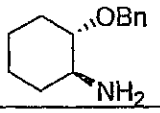
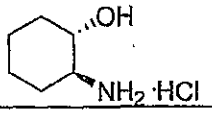
To a solution of (1*S*,2*S*)-2-benzyloxycyclopentyl amine (1.5 g, 7.84 mmol) in MeOH (50 mL) at rt was added 10 % Pd/C (50% wet, 1.0 g) followed by dropwise addition of conc. HCl (0.7 mL). The mixture was stirred under a balloon of H_2 for 14 h and the catalyst was filtered off thru a pad of Celite. The pad of Celite was washed with MeOH (2 x 10 mL) and the resulting filtrate was concentrated under reduced pressure to afford 0.97 g (90%) of a yellow semisolid; M+H (free base) = 102

PREPARATIVE EXAMPLES 249-251

In an analogous fashion to Preparative Example 248, the benzyl protected cycloalkyl amines (Column 2) were converted to the desired aminocycloalkanol hydrochloride derivatives (Column 3) as listed in Table 17.

TABLE 17

Ex.	Column 2 (Amine)	Column 3 (Cleavage method)	CMPD M+H

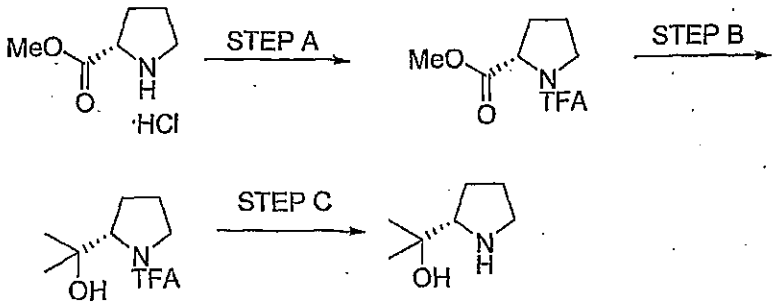
249			M+H = 102 (free base)
250			M+H = 116 (free base)
251			M+H = 116 (free base)

PREPARATIVE EXAMPLE 252



To a solution of ester (prepared according to *J. Org. Chem.* (1999), *64*, 330) (0.5 g, 2.43 mmol) in THF (8 mL) at 0 °C was added $LiAlH_4$ (0.37 g, 9.74 mmol) in one portion. The resulting mixture was heated at reflux for 12h and was cooled to 0 °C. The mixture was treated sequentially with H_2O (1 mL), 1 M NaOH (1 mL), and H_2O (3 mL). CH_2Cl_2 (10 ml) was added to the mixture which was stirred vigorously for 30 min. The mixture was filtered thru a pad of Celite which was washed generously with CH_2Cl_2 (3 x 5 mL). The resulting filtrate was concentrated under reduced pressure to afford 0.41 g (85%) of a yellow/orange solid. M+H = 142.

PREPARATIVE EXAMPLE 253



STEP A:

To a solution of *L*-proline methyl ester hydrochloride (0.50 g, 3.0 mmol) in CH_2Cl_2 (15 mL) at 0 °C was added Et_3N (1.1 mL, 7.55 mmol) followed by TFAA (0.56 mL, 3.92 mmol). The mixture was stirred for 12 h at rt and 1N HCl (25 mL) was added. The layers were separated and the organic layer was washed sequentially with sat. aq. $NaHCO_3$ (1 x 25 mL), and brine (1 x 25 mL). The organic layer was dried (Na_2SO_4), filtered, and concentrated under reduced pressure to afford 0.72 g (100%) of a yellow oil. M+H = 226. The crude material was taken onto Step B without further purification.

STEP B:

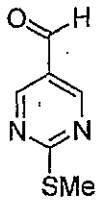
To a solution of the compound prepared in Preparative Example 253, Step A (0.68 g, 3.0 mmol) in THF (20 mL) at 0 °C was added MeMgI (5.1 mL, 3.0M in Et_2O) dropwise over 10 min. The resulting solution was stirred for 16 h at rt whereupon the mixture was quenched by addition of sat. aq. NH_4Cl. The mixture was concentrated to dryness and the resultant residue was stirred with EtOAc (100 mL) for 45 min and filtered. The filtrate was concentrated under reduced pressure to afford 0.68g (100%) of a yellow/orange oil. M+H = 226. The crude material was taken onto Step C without further purification.

STEP C:

To a solution of the compound prepared in Preparative Example 253, Step B (0.68 g, 3.0 mmol) in MeOH (5 mL) was added a solution of KOH (0.68 g, 12.1 mmol) in MeOH (5 mL). The mixture was stirred at reflux for 12h and rt for 72h whereupon the mixture was concentrated to dryness. The crude residue was suspended in EtOAc (50 mL) and was stirred vigorously for 30 min and was filtered. This procedure was repeated 2X more and the resultant filtrate was

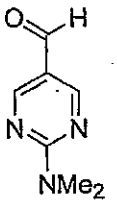
concentrated under reduced pressure to afford 128 mg (33%) of a maroon/orange oil. M+H = 130. This material was used without purification in the subsequent coupling step.

PREPARATIVE EXAMPLE 254:



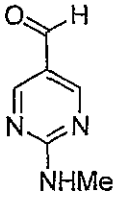
The aldehyde was prepared according to the procedure of Gupton (*J. Heterocyclic Chem.* (1991), *28*, 1281).

PREPARATIVE EXAMPLE 255



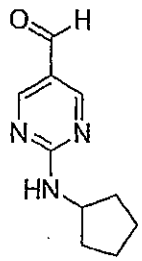
Using the aldehyde from Preparative Example 254, the procedure of Gupton (*J. Heterocyclic Chem.* (1991), *28*, 1281) was employed to prepare the title aldehyde.

PREPARATIVE EXAMPLE 256



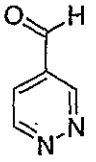
The title aldehyde was prepared according to the procedure of Ragan et. al *Synlett* (2000), *8*, 1172-1174.

PREPARATIVE EXAMPLE 257



The reaction of known cyclopentyl guanidine hydrochloride (*Org. Lett.* (2003), *5*, 1369-1372) under the conditions of Ragan (*Synlett* (2000), *8*, 1172-1174) afforded the title aldehyde.

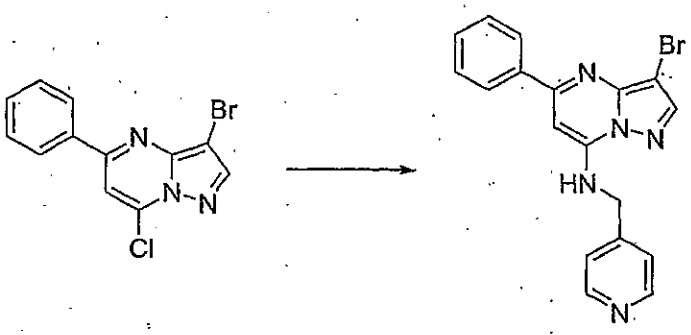
PREPARATIVE EXAMPLE 258



The title compound was prepared according to known literature *Monatshefte fur Chemie* (1973), *104*, 1372-1382.

EXAMPLES

EXAMPLE 1:

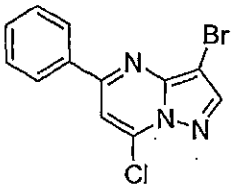
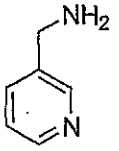
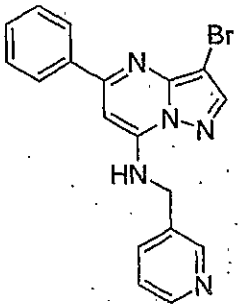
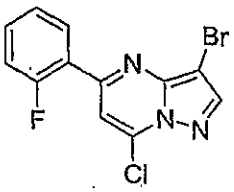
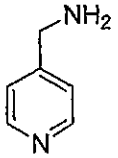
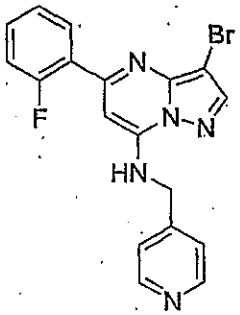
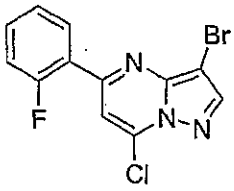
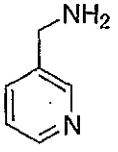
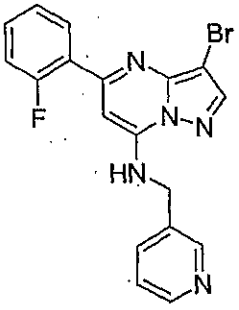


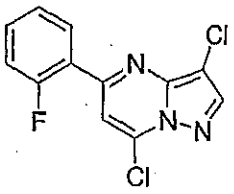
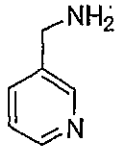
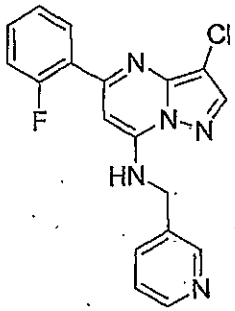
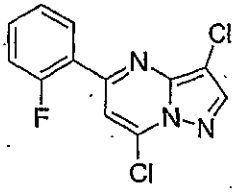
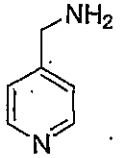
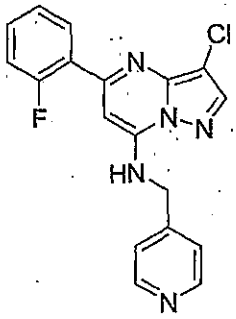
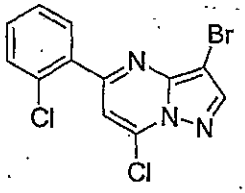
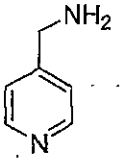
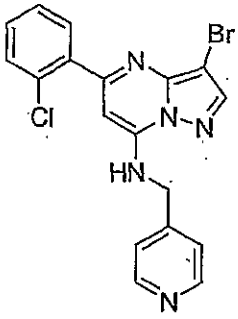
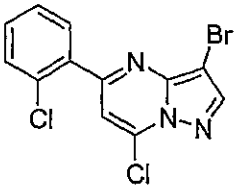
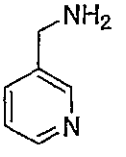
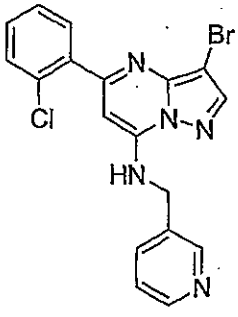
A solution of the product from Preparative Example 127 (0.27 g, 0.875 mmol), 4-aminomethylpyridine (0.12 g, 1.3 eq.), and K_2CO_3 (0.24 g, 2 eq.) in CH_3CN (5 mL) was stirred at room temperature 48 hours. The reaction mixture was diluted with H_2O and extracted with CH_2Cl_2. The combined organics were dried over Na_2SO_4, filtered and concentrated. The crude product was purified by flash chromatography using a 4% MeOH in CH_2Cl_2 solution as eluent (0.28 g, 93% yield). LCMS: MH^+=380; mp= >205°C (dec).

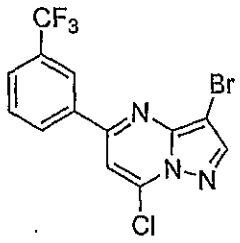
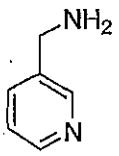
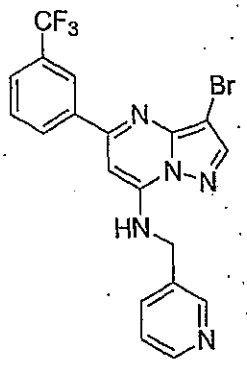
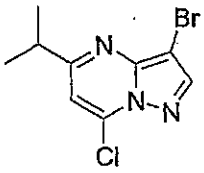
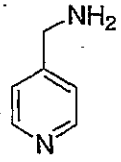
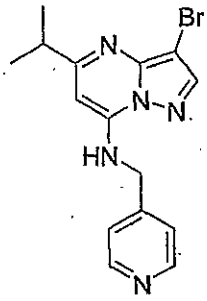
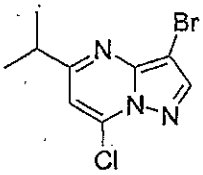
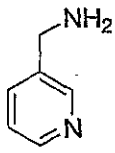
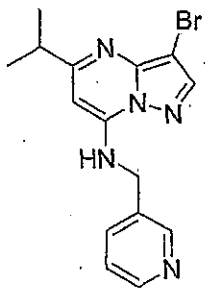
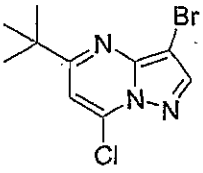
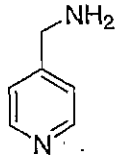
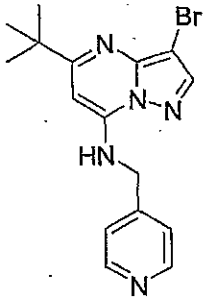
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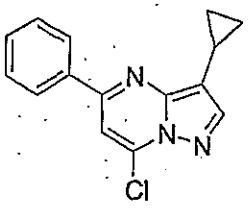
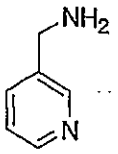
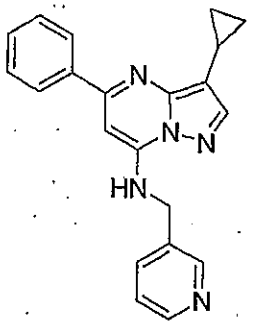
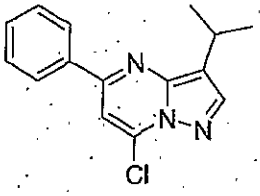
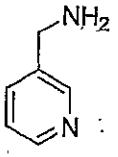
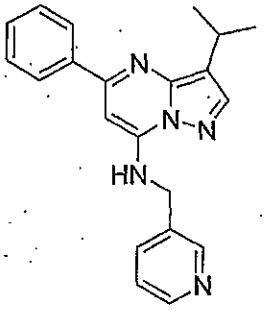
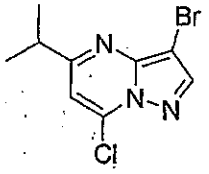
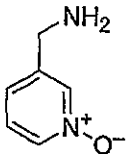
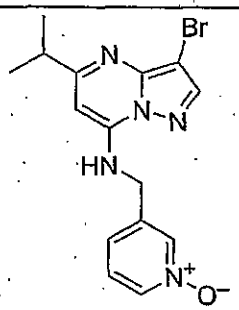
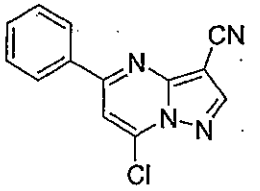
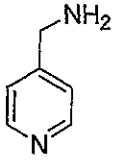
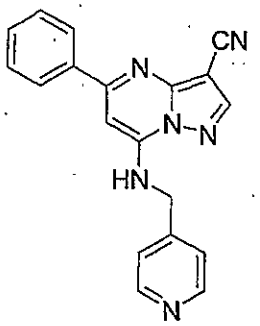
By following essentially the same procedure set forth in Example 1 only substituting the chlorides shown in Column 2 of Table 18 and the amines shown in Column 3 of Table 18, the compounds in Column 4 of Table 18 were prepared:

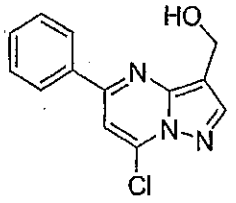
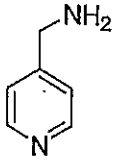
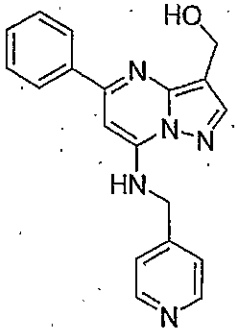
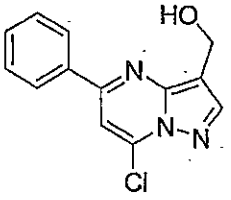
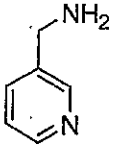
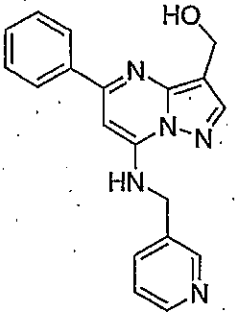
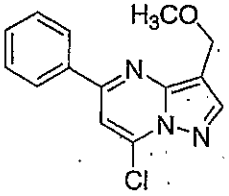
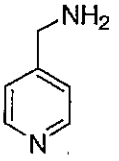
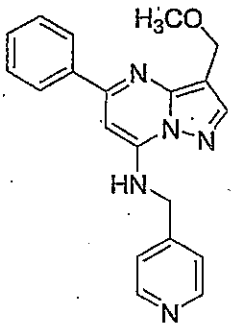
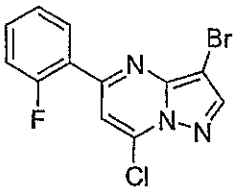
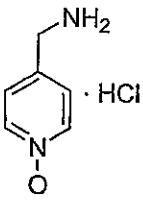
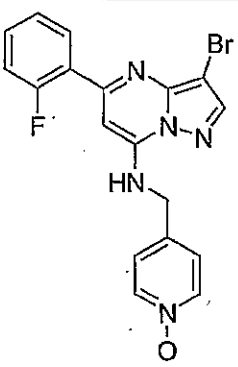
TABLE 18

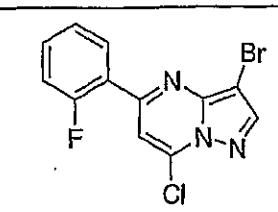
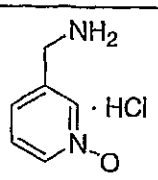
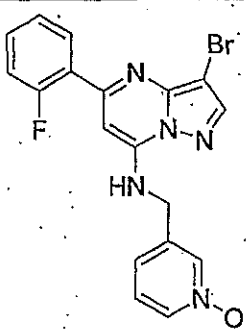
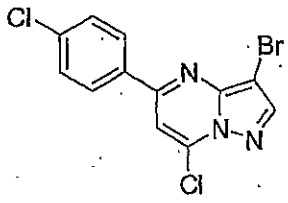
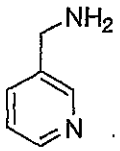
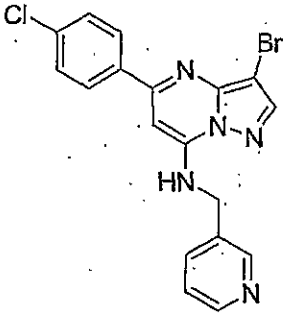
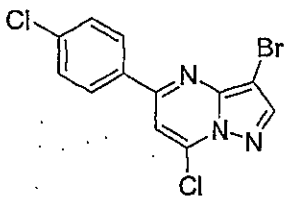
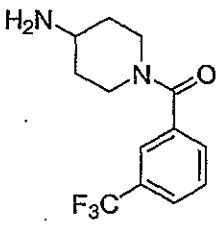
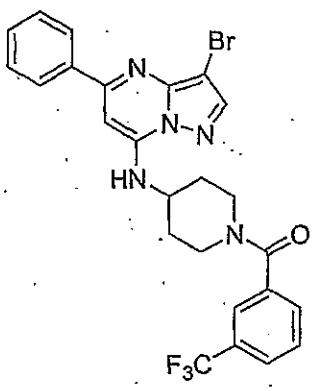
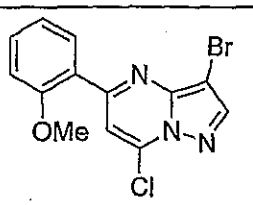
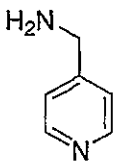
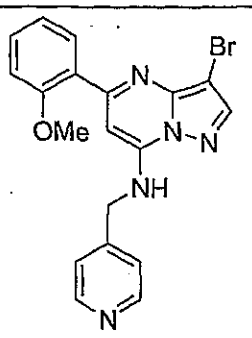
Ex.	Column 2	Column 3	Column 4	Data
2				LCMS: MH^+= 380; mp=175-176°C
3				LCMS: MH^+= 398; mp= 156-157°C
4				LCMS: MH^+= 398; mp= 45-49°C

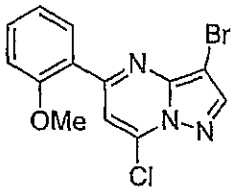
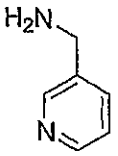
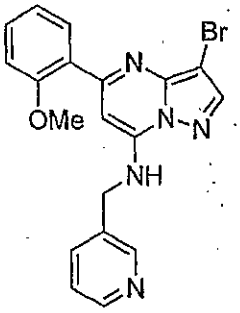
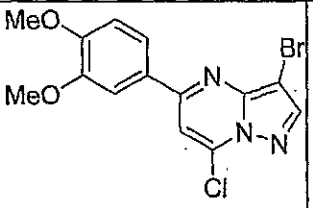
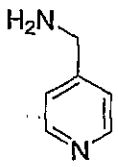
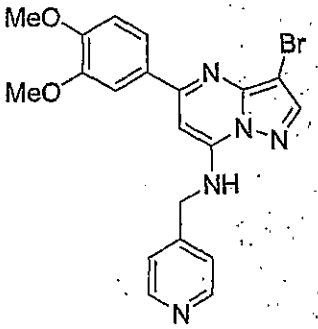
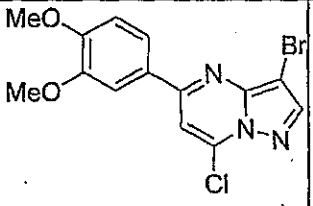
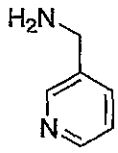
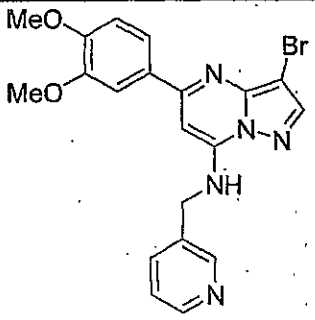
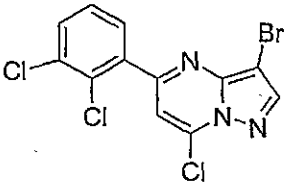
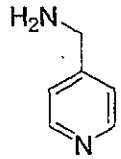
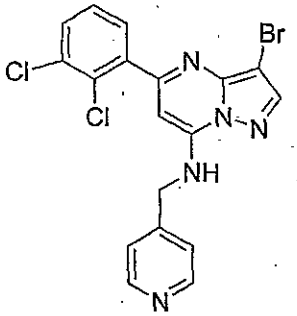
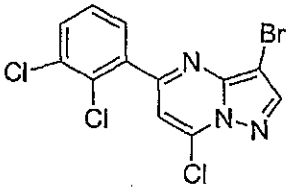
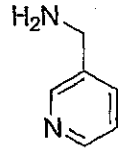
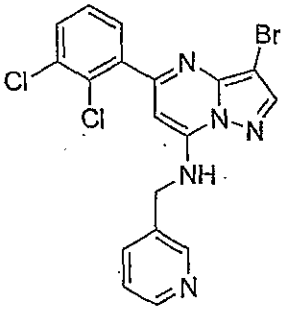
5				LCMS: MH^+= 354; mp= 43-46°C
6				LCMS: MH^+= 354; mp= 149-150°C
7				LCMS: MH^+= 414; mp= 86-92°C
8				LCMS: MH^+= 414; mp= 185-186°C

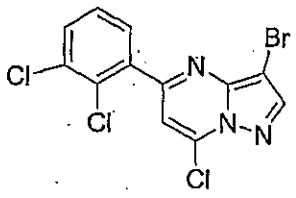
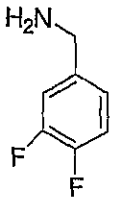
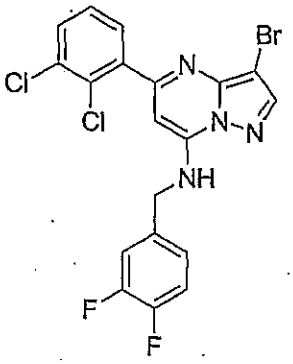
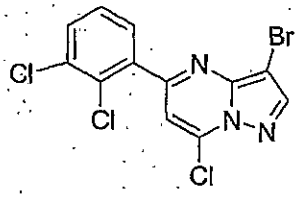
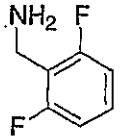
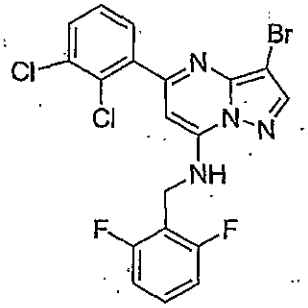
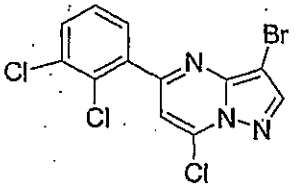
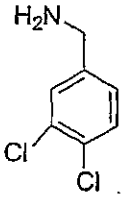
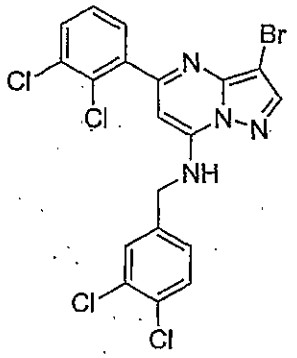
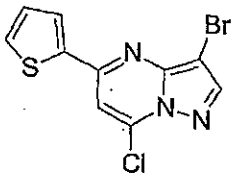
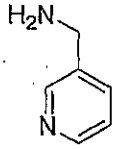
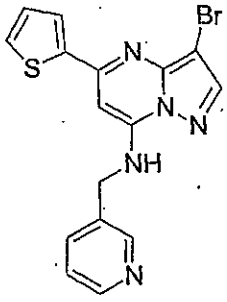
9				LCMS: MH^+= 448; mp= 167-168°C
10				LCMS: MH^+= 346; mp= 57-58°C
11				LCMS: MH^+= 347; mp=122.9-125.3 °C
12				LCMS: MH^+= 360; mp= 127-128°C

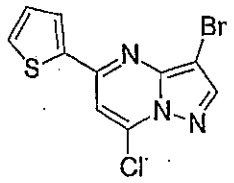
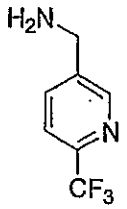
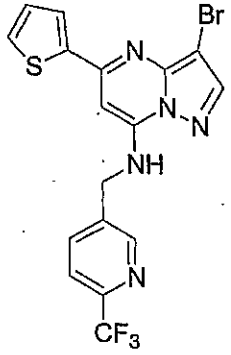
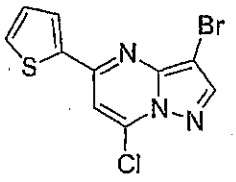
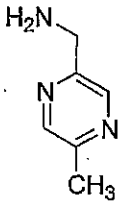
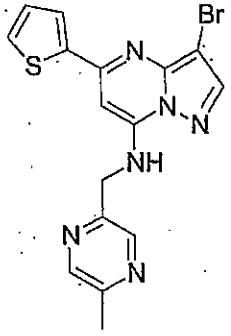
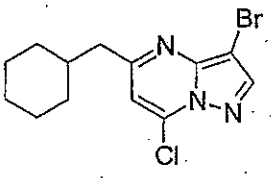
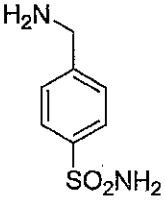
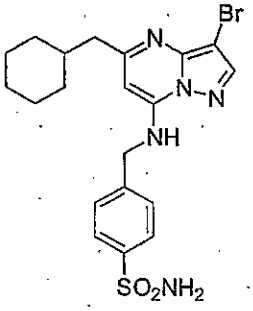
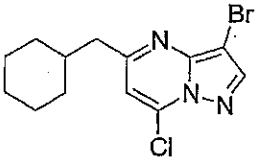
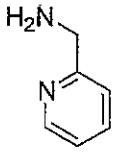
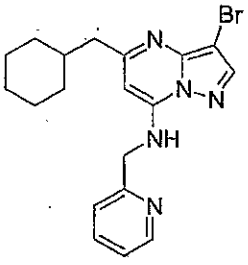
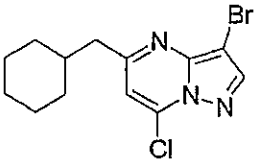
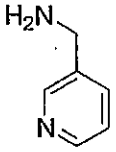
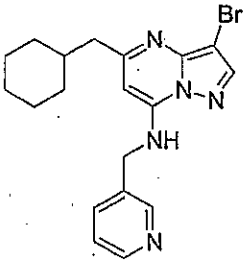
13				LCMS: MH^+= 342; mp= 133-135°C
14				LCMS: MH^+= 344; mp= 152-155°C
15				LCMS: MH^+= 362; mp= 164-167°C
16				LCMS: MH^+= 327; mp= 146-155°C

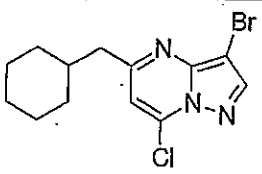
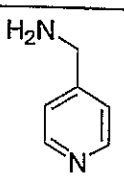
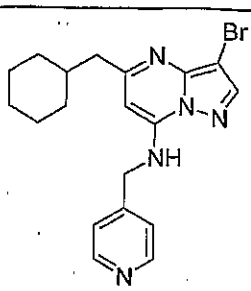
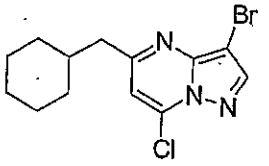
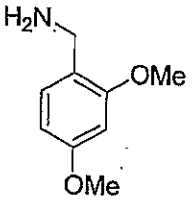
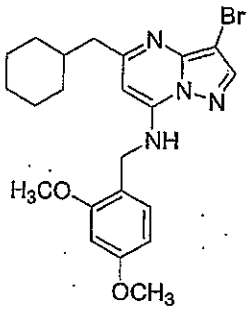
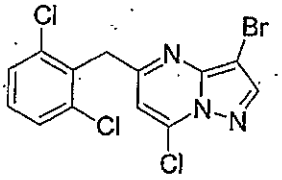
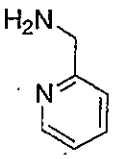
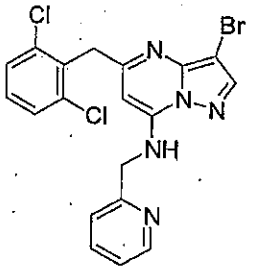
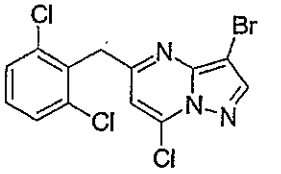
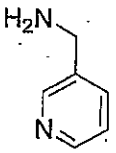
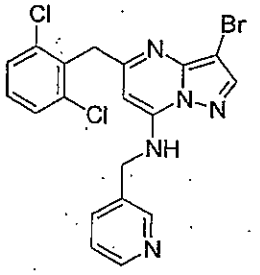
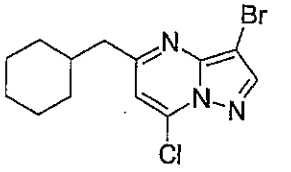
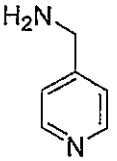
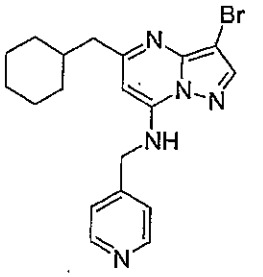
17				LCMS: MH^+= 332; mp= 71-82°C
17.1				MS: MH^+ = 332.
18				LCMS: MH^+= 346; mp= 58-65°C
19				LCMS: MH^+= 414; mp= 211-213°C

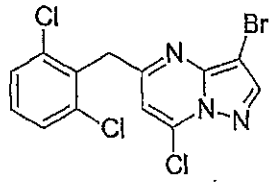
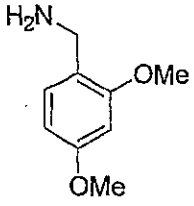
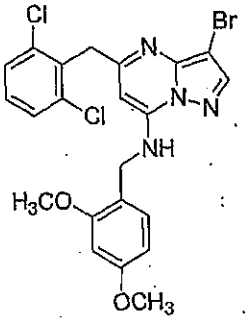
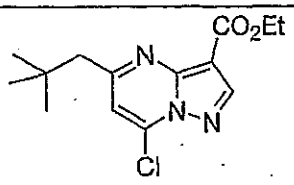
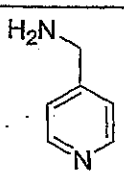
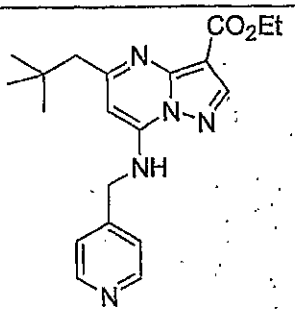
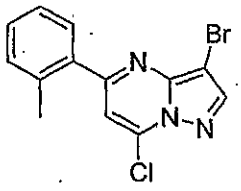
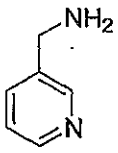
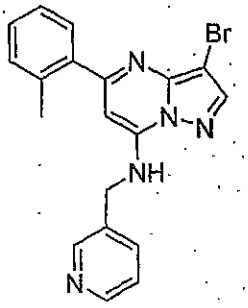
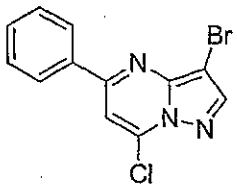
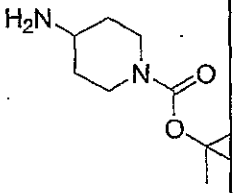
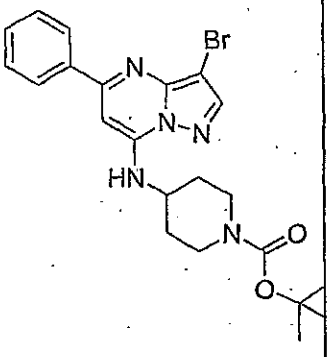
20				LCMS: MH^+= 414; mp= 194-197°C
21				MS: MH^+ = 414 m.p. 211 – 216°C
22				LCMS: MH^+= 544; mp= 104-107°C
23				Yield = 83% LCMS: MH^+ = 410.

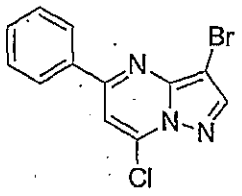
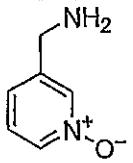
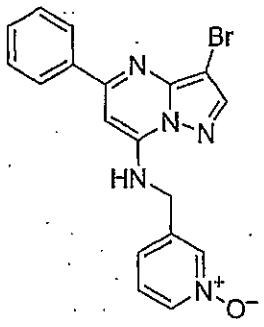
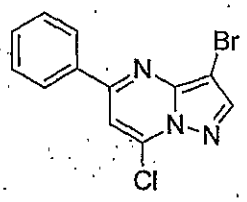
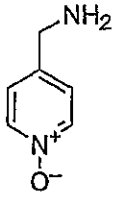
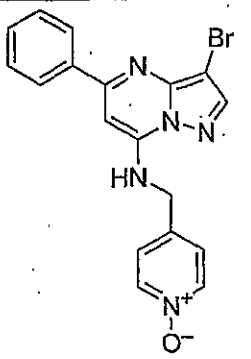
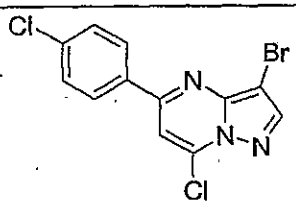
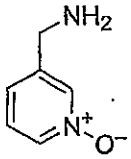
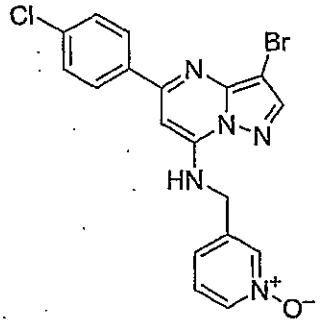
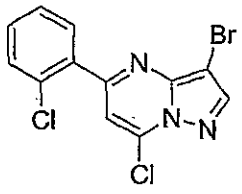
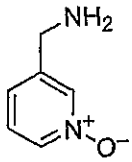
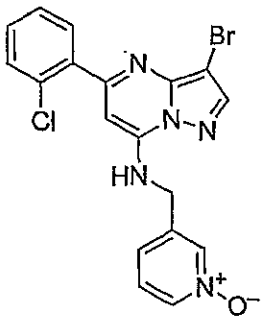
24				Yield = 84% LCMS: MH^+ = 410.
25				Yield = 96% LCMS: MH^+ = 440.
26				Yield = 99% LCMS: MH^+ = 440.
27				Yield = 89% LCMS: MH^+ = 448.
28				Yield = 78% LCMS: MH^+ = 448.

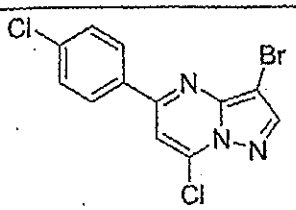
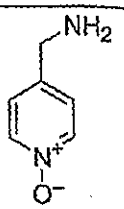
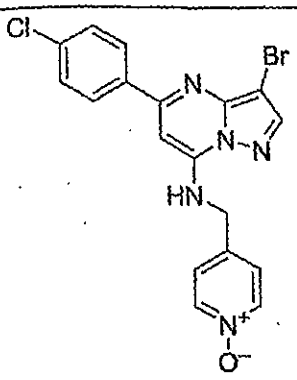
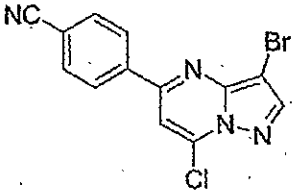
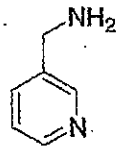
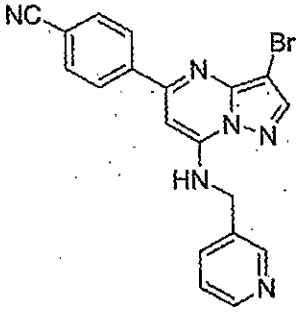
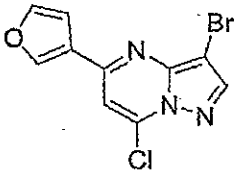
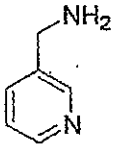
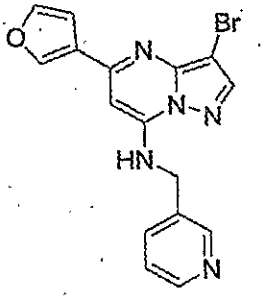
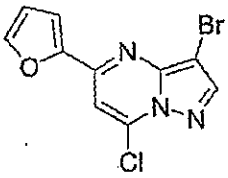
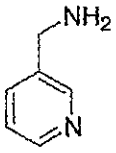
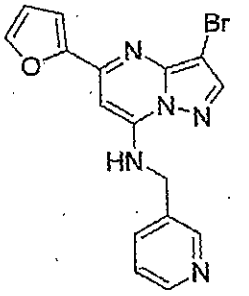
30				Yield = 96% LCMS: MH^+ = 483.
31				Yield = 35% LCMS: MH^+ = 483.
32				Yield = 77% LCMS: MH^+ = 515.
33				Yield = 100% m.p. 179°C LCMS: MH^+ = 388

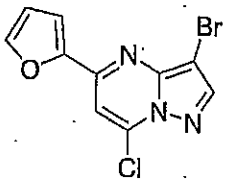
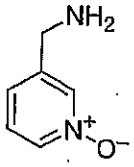
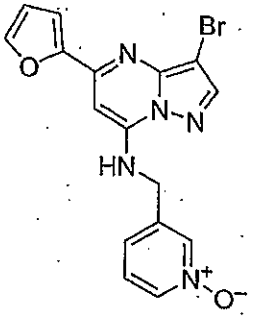
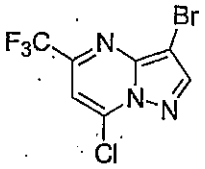
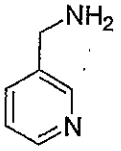
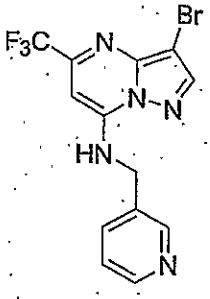
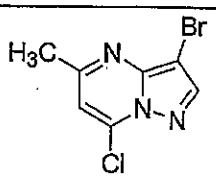
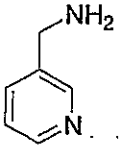
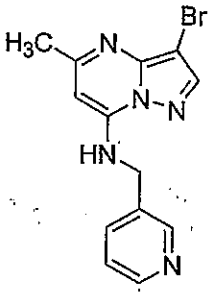
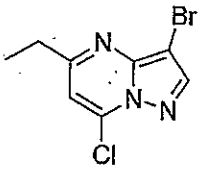
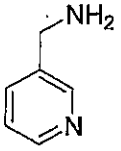
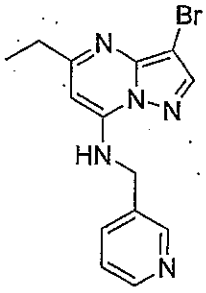
34				Yield = 99% m.p. 186°C LCMS: MH^+ = 456
35				Yield = 98% m.p. 181°C LCMS: MH^+ = 401
36				Yield = 63% m.p. 192°C LCMS: MH^+ = 480
37				Yield = 75% m.p. 126 – 127°C LCMS: MH^+ = 400
38				Yield = 94% m.p. 132 -133°C LCMS: MH^+ = 400

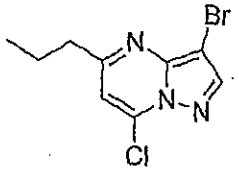
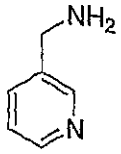
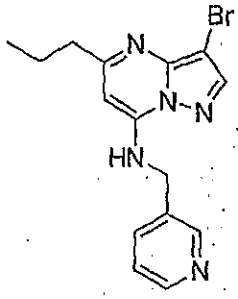
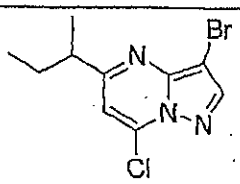
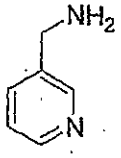
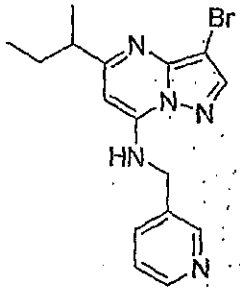
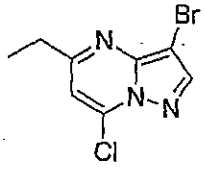
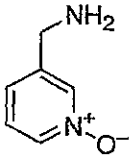
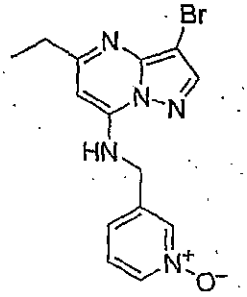
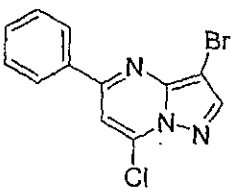
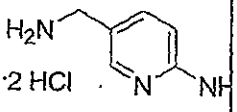
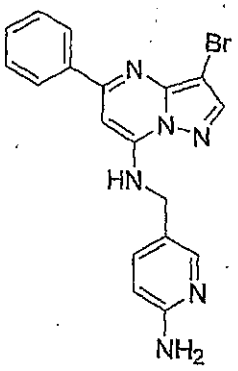
39				Yield = 95% m.p. 121 – 122°C LCMS: MH^+ = 400
40				Yield = 98% LCMS: MH^+ = 460
41				Yield = 87% m.p. 170 - 171°C LCMS: MH^+ = 464
42				Yield = 84% m.p. 216 -217°C LCMS: MH^+ = 464
43				Yield = 96% m.p. 214°C LCMS: MH^+ = 464

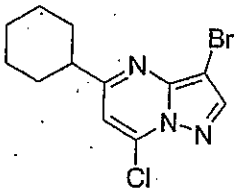
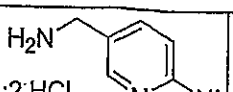
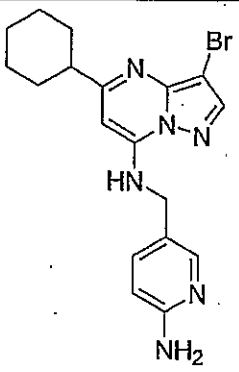
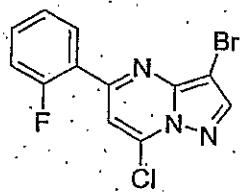
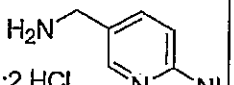
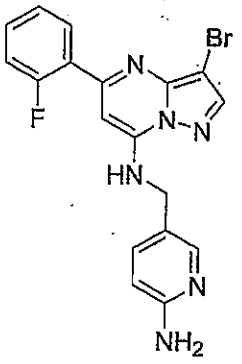
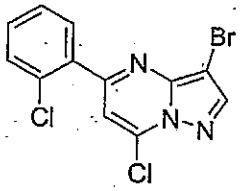
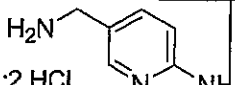
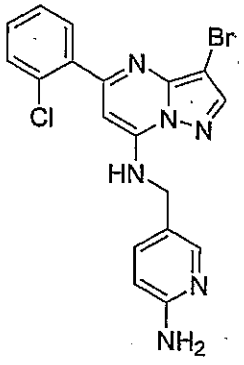
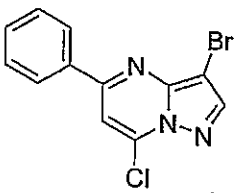
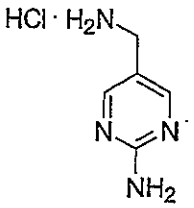
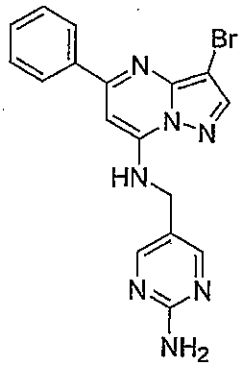
44				Yield = 95% m.p. 158°C LCMS: MH^+ = 522
45				Yield=90 % LCMS: MH^+= 278
46				Yield=100%; LCMS: MH^+=394
47				LCMS: MH+ = 473 m.p. 84 - 87°C

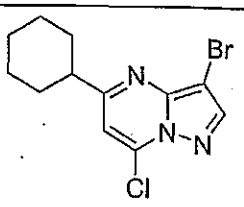
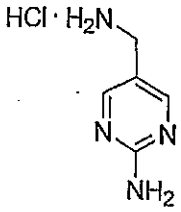
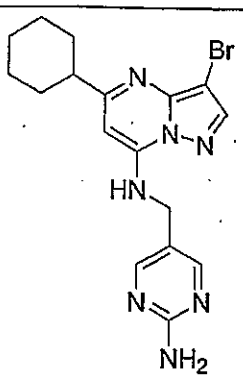
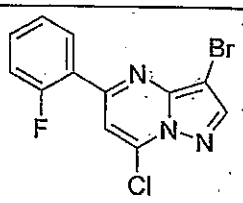
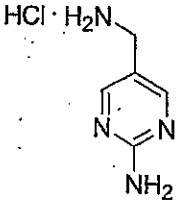
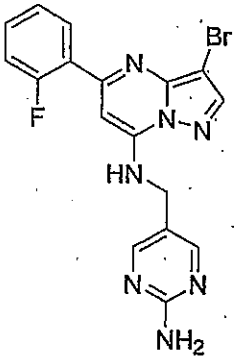
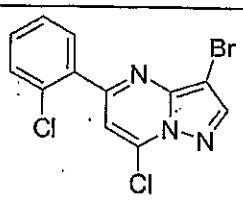
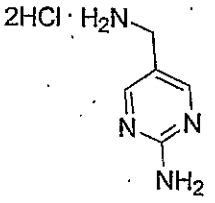
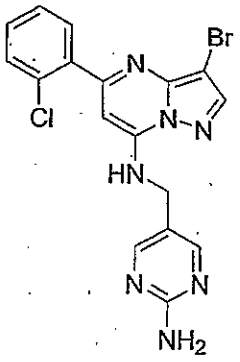
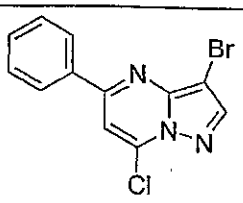
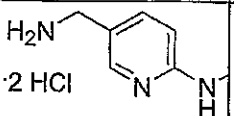
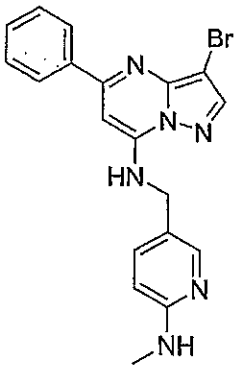
48				MS: MH^+ = 396 m.p. 91.5 – 93.3°C
49				MS: MH^+ = 396 m.p. 196 – 199°C
50				MS: MH^+ = 430 m.p. 242 – 244°C
51				MS: MH^+ = 430 m.p. 218°C

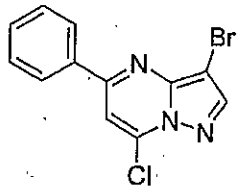
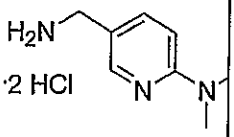
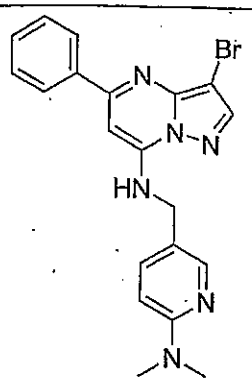
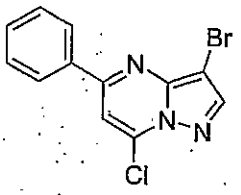
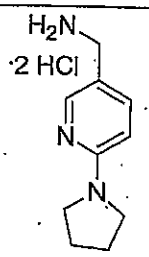
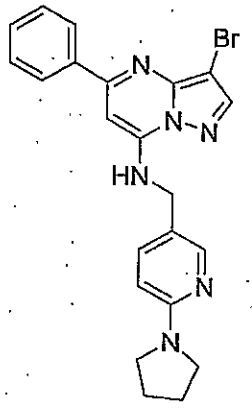
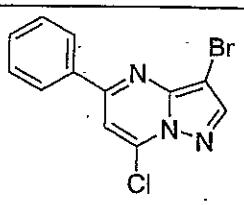
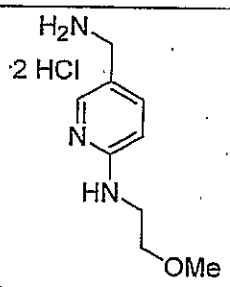
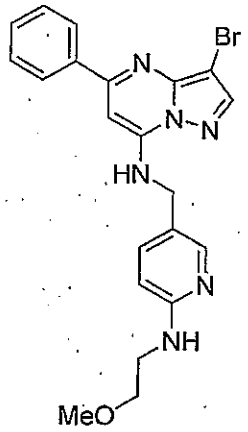
52				MS: MH^+ = 430 m.p. 230 – 233°C
54				MS: MH^+ = 405 m.p. 185 – 188°C
55				MS: MH^+ = 370 m.p. 229 – 232°C
56				MS: MH^+ = 370 m.p. 85 – 90°C

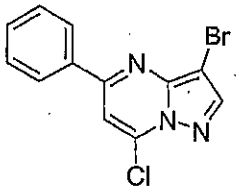
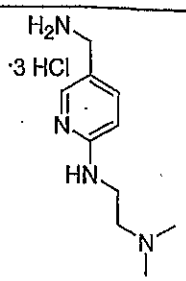
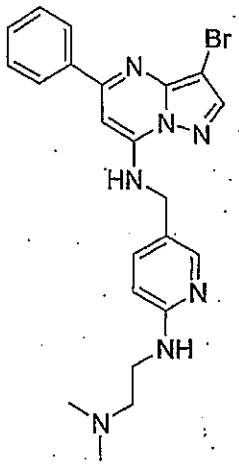
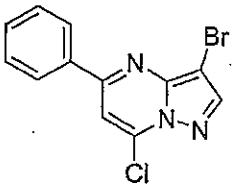
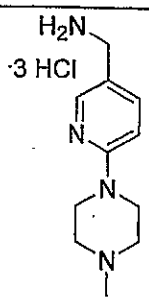
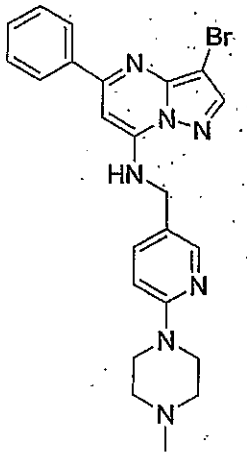
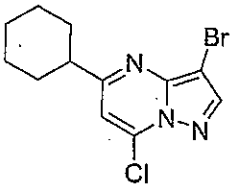
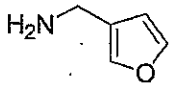
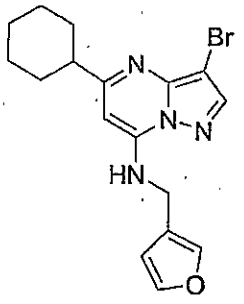
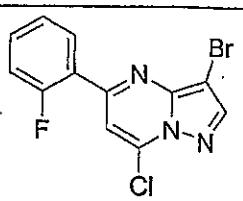
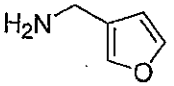
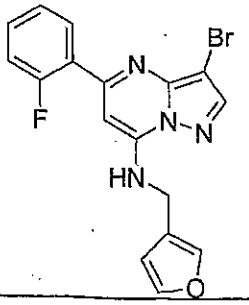
57				MS: MH^+ = 386 m.p. 227 – 230°C
58				MS: MH^+ = 372 m.p. 212 – 215°C
59				MS: MH^+ = 318 m.p. 169 – 171°C
60				MS: MH^+ = 332 m.p. 170 – 173°C

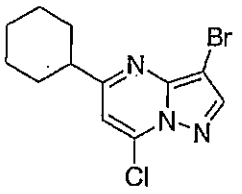
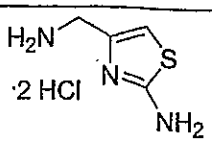
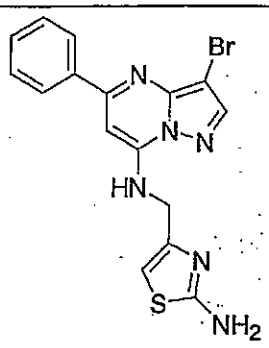
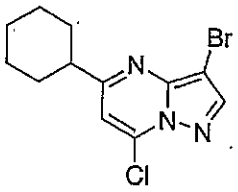
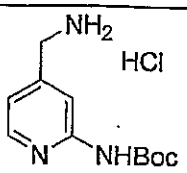
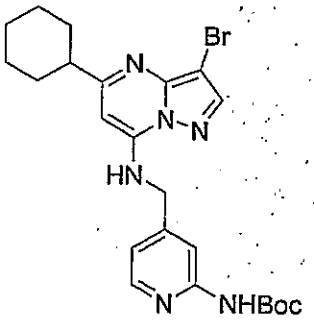
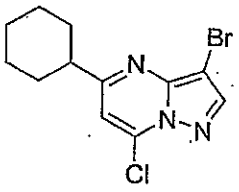
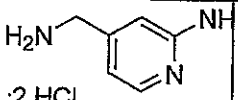
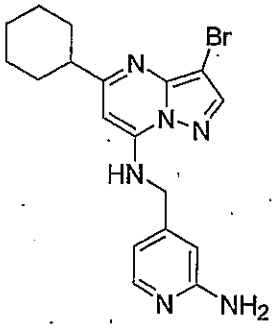
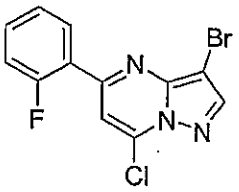
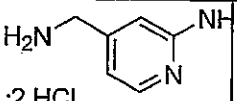
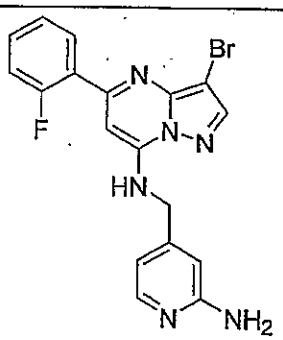
61				MS: MH^+ = 346 m.p. 156 – 159°C
62				MS: MH^+ = 360 m.p. 114 – 116°C
63				MS: MH^+ = 348 m.p. 197 – 200°C
64				1. mp = 230-232 2. M+H = 396

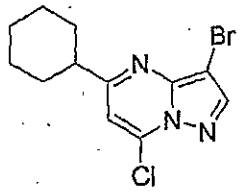
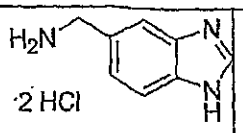
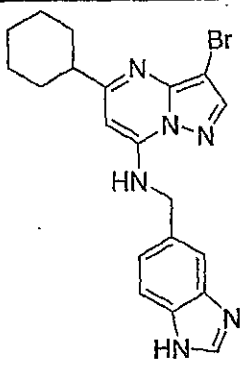
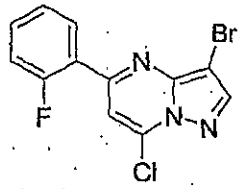
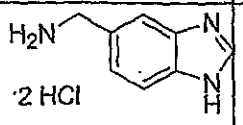
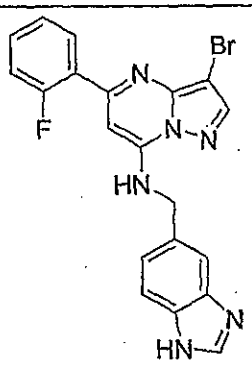
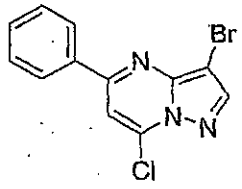
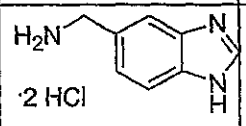
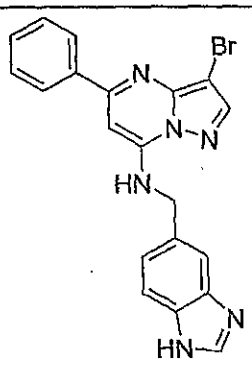
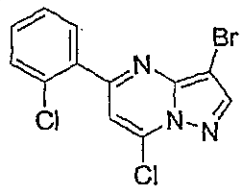
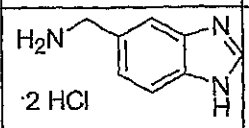
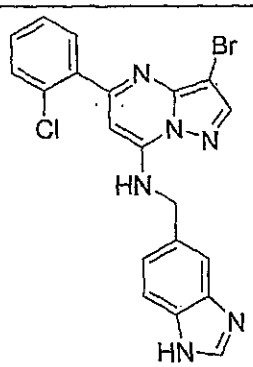
65				1. mp = 205-207 2. M+H = 402
66				1. mp = 220-223 2. M+H = 414
67				1. mp = 191-193 2. M+H = 431
68				1. mp = 235-237 2. M+H = 397

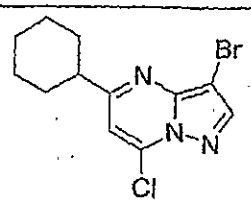
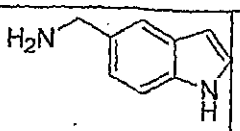
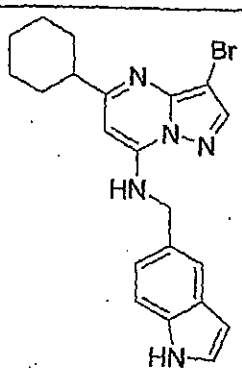
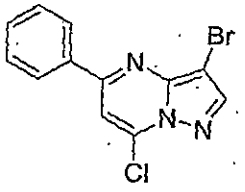
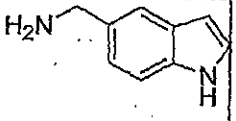
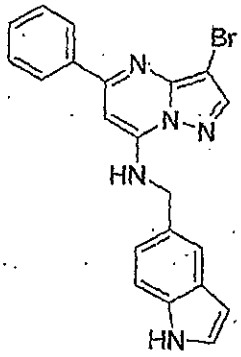
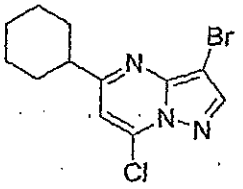
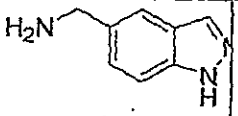
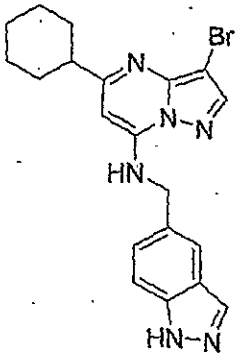
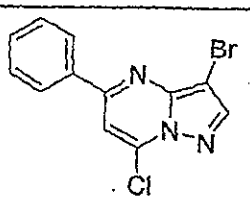
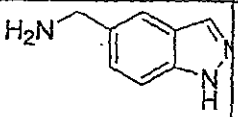
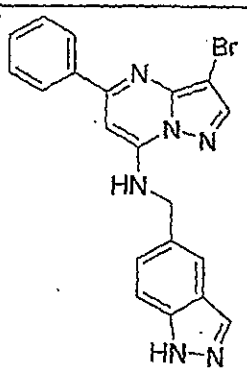
69				1. mp = >250 2. M+H = 403
70				1. mp = 230-232 2. M+H = 415
71				1. mp = 235-238 2. M+H = 431
72				1. mp = 186-188 2. M+H = 410

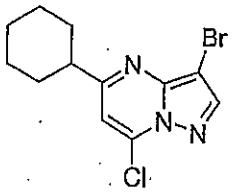
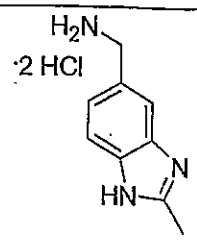
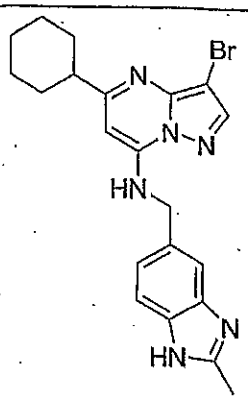
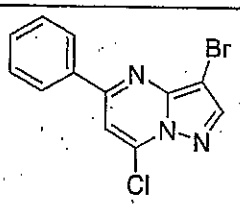
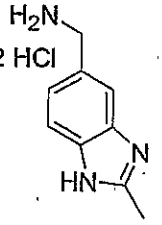
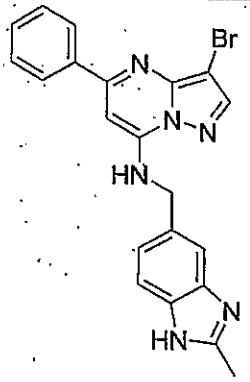
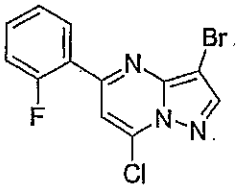
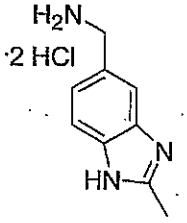
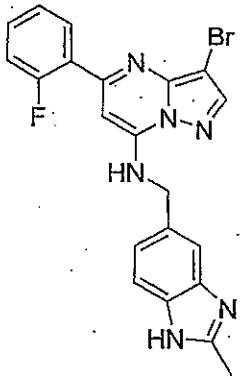
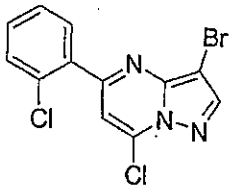
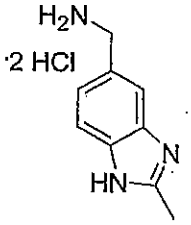
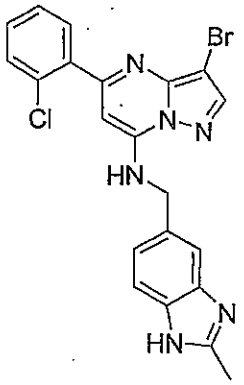
73				1. mp = 136-138 2. M+H = 424
74				1. mp = 192-195 2. M+H = 450
75				1. mp = 88-90 2. M+H = 454

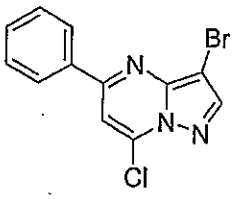
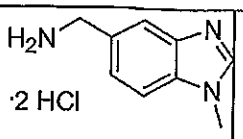
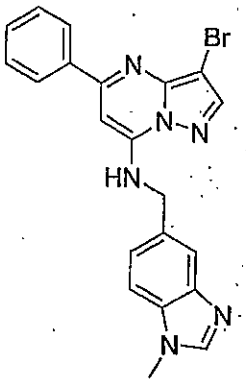
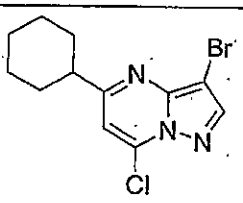
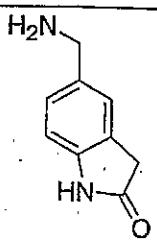
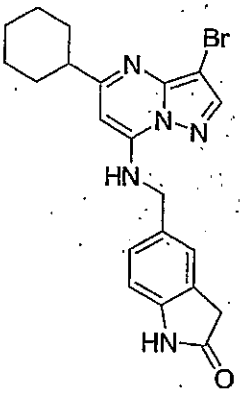
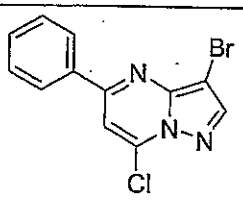
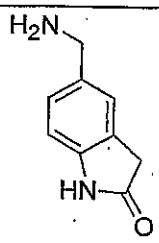
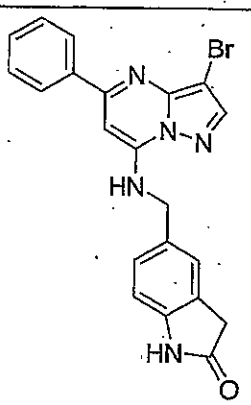
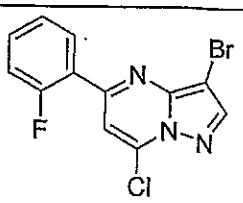
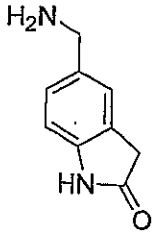
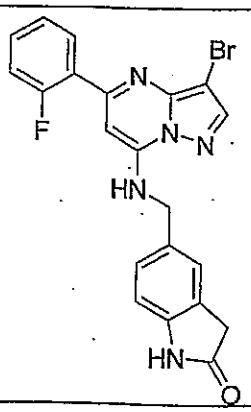
76				1. mp = 230-232 2. M+H = 467
77				1. mp = 131-133 2. M+H = 479
78				1. mp = 85-88 2. M+H = 376
79				1. mp = 131-133 2. M+H = 388

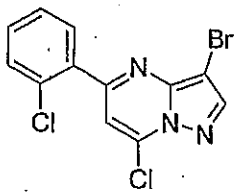
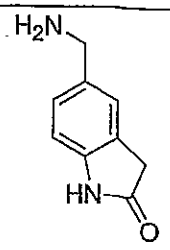
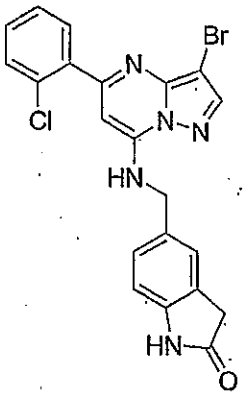
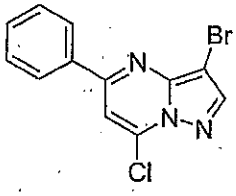
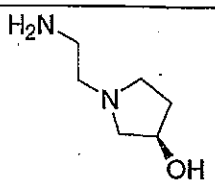
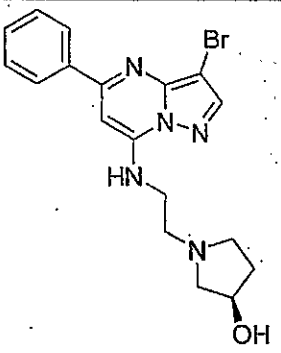
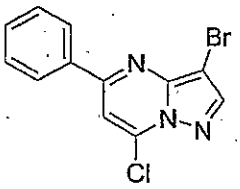
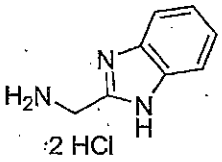
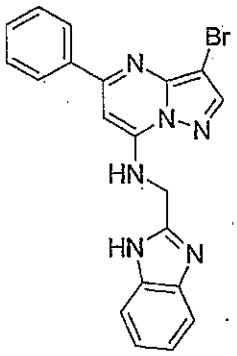
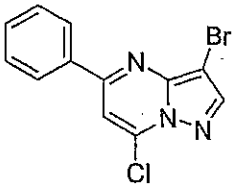
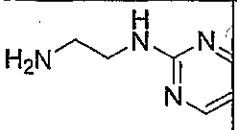
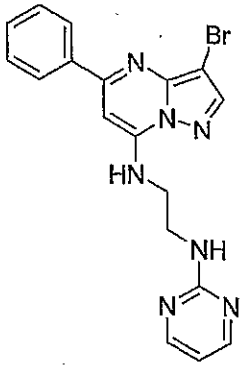
80				1. mp = 206-208 2. M+H = 408
81				1. mp = 108-110 2. M+H = 502
82				1. mp = 83-85 2. M+H = 402
83				1. mp = 220 2. M+H = 414

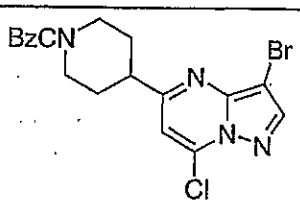
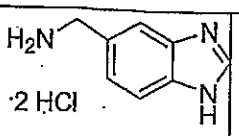
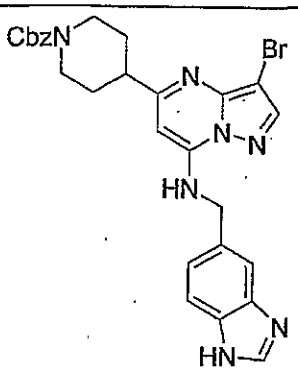
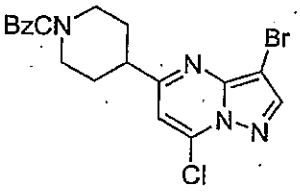
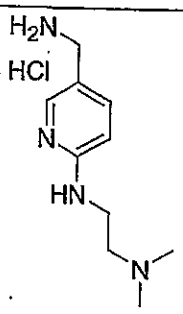
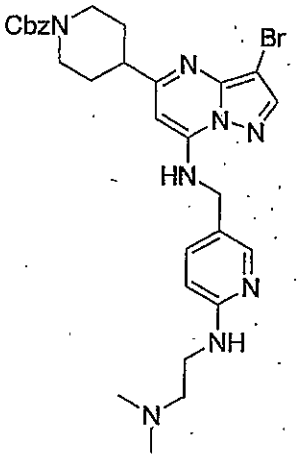
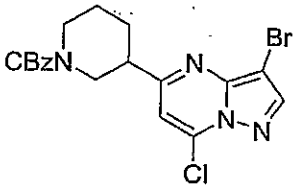
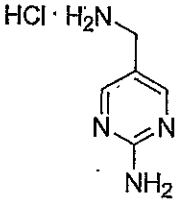
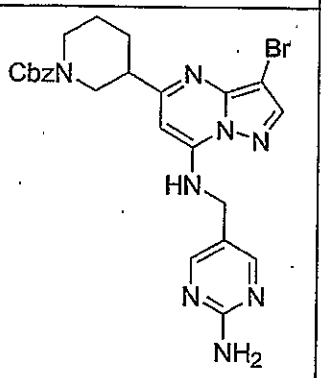
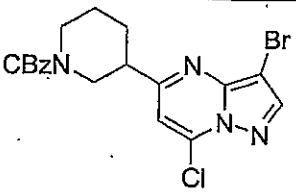
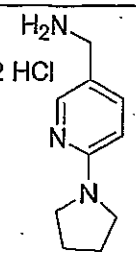
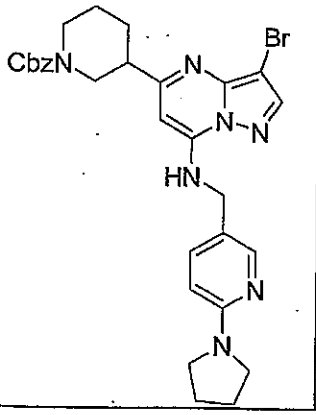
84				1. mp = 154-156 2. M+H = 426
85				1. mp = 152-153 2. M+H = 438
86				1. mp = 159-161 2. M+H = 420
87				1. mp = >220 2. M+H = 455

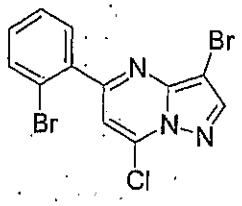
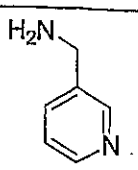
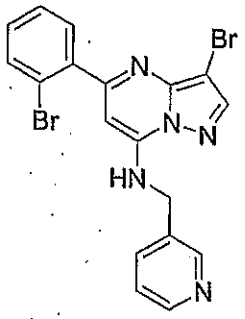
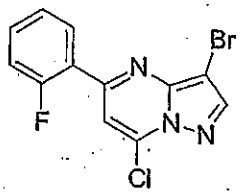
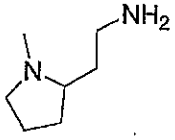
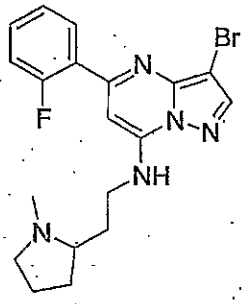
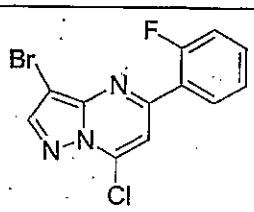
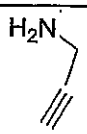
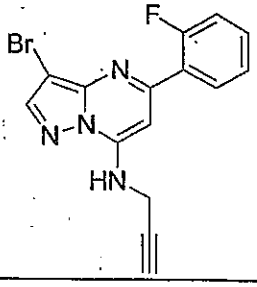
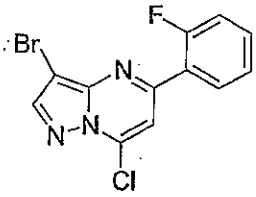
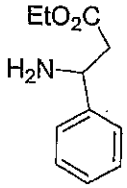
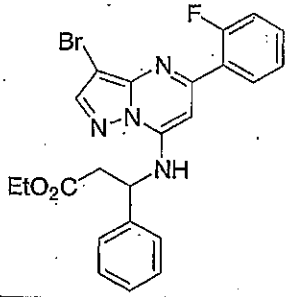
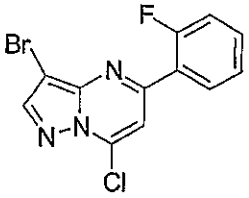
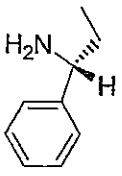
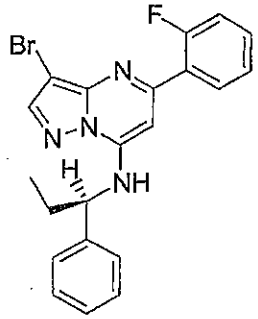
88				1. mp = 223-225 2. M+H = 425
89				1. mp = 199-201 2. M+H = 419
90				1. mp = 184-186 2. M+H = 426
91				1. mp = 196-198 2. M+H = 420

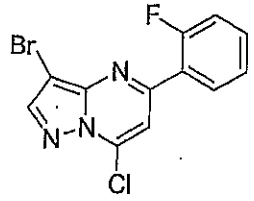
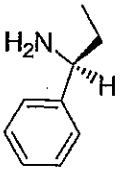
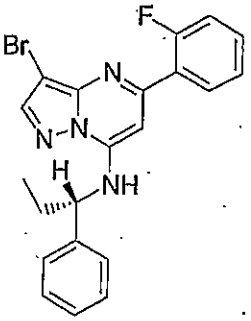
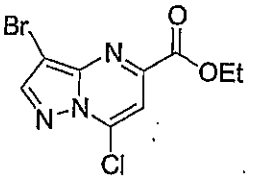
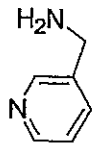
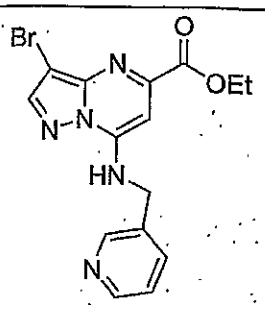
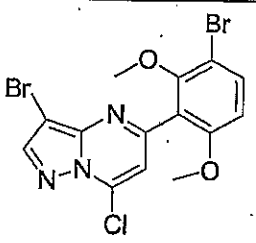
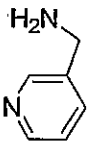
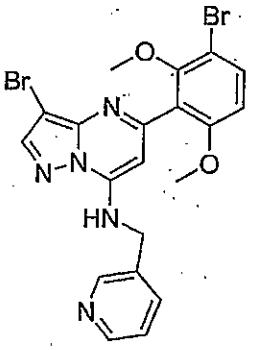
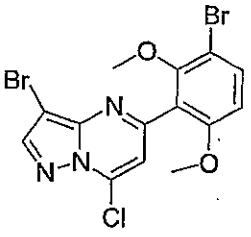
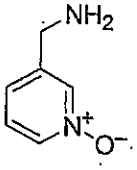
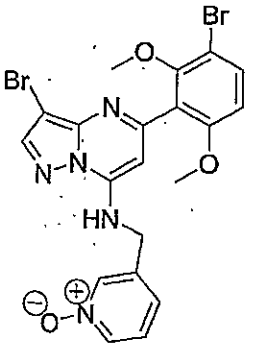
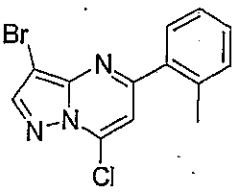
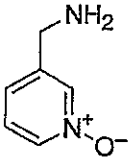
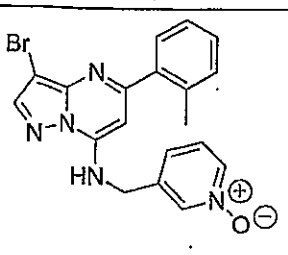
92				1. mp = 156-159 2. M+H = 440
93				1. mp = 173-176 2. M+H = 434
94				1. mp = 173-175 2. M+H = 452
95				1. mp = 174-176 2. M+H = 469

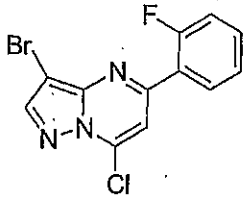
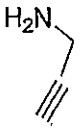
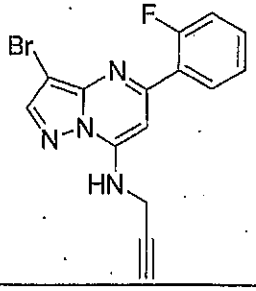
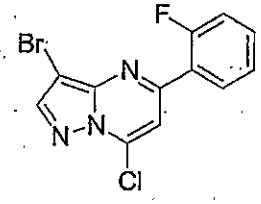
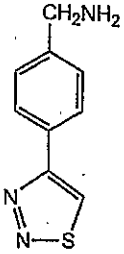
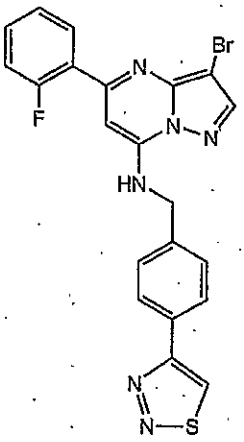
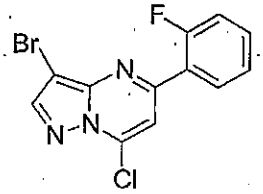
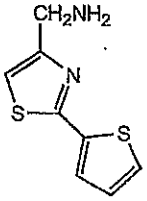
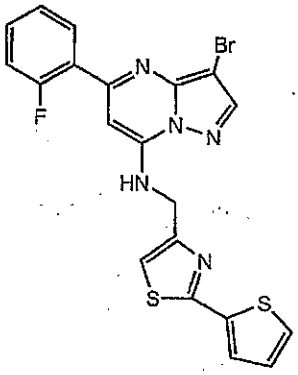
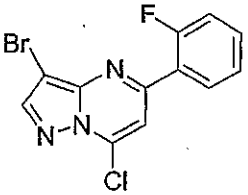
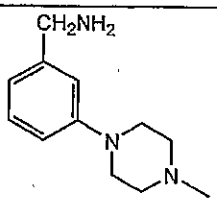
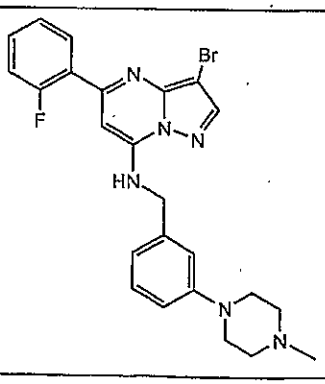
96		 ·2 HCl		1. mp = 230-234 2. M+H = 434
97				1. mp = 191-193 2. M+H = 441
98				1. mp = 202-205 2. M+H = 434
99				1. mp = 209-212 2. M+H = 453

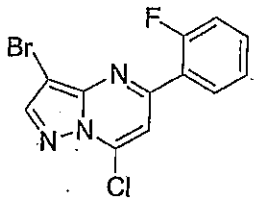
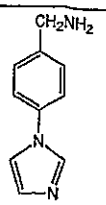
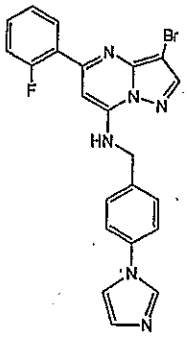
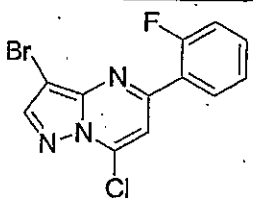
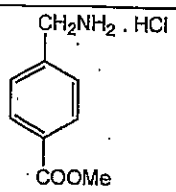
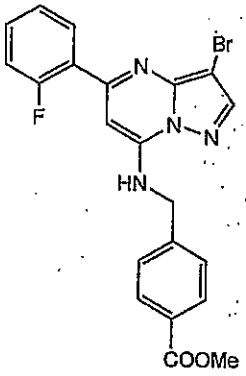
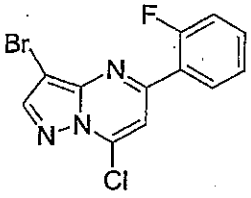
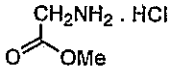
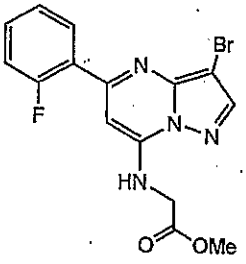
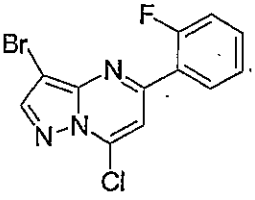
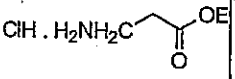
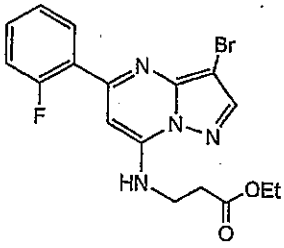
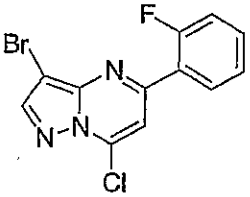
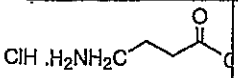
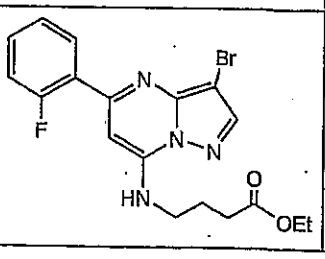
100				1. mp = 219-221 2. M+H = 469
101				1. mp = 64-66 2. M+H = 403
102				1. mp = 168-170 2. M+H = 420
103				1. mp = 213-216 2. M+H = 411

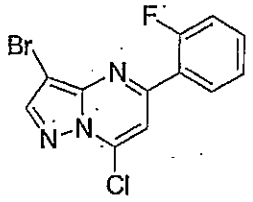
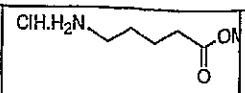
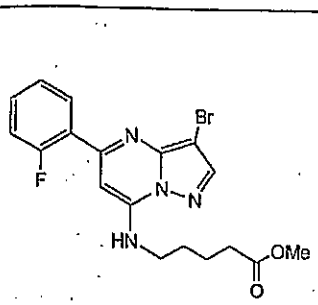
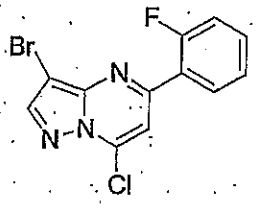
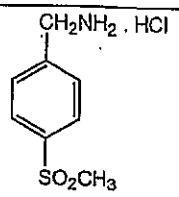
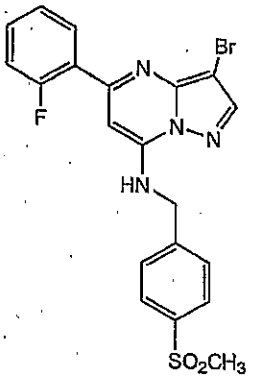
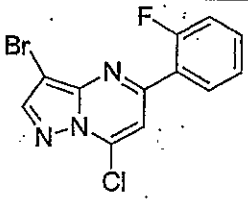
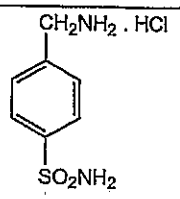
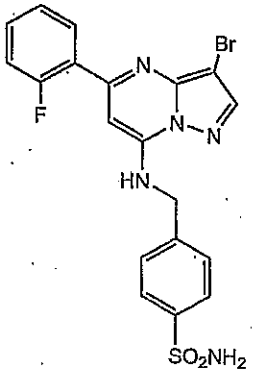
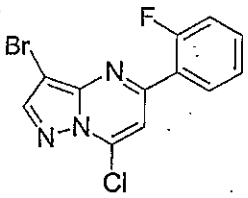
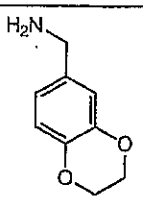
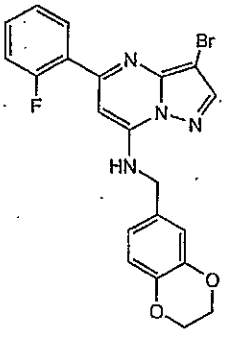
104				1. mp = 98-100 2. M+H = 561
105				1. mp 70-72 2. M+H = 608
106				1. mp 168-170 2. M+H = 538
107				1. mp 189-191 2. M+H = 592

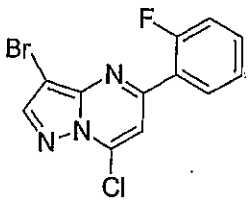
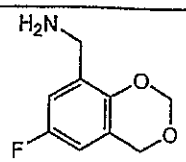
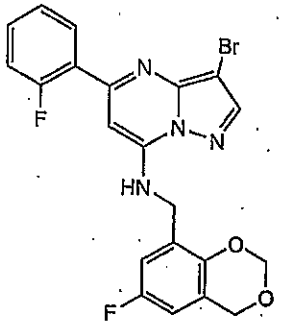
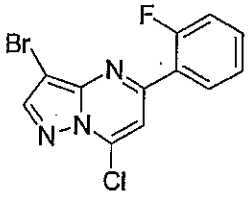
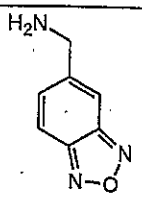
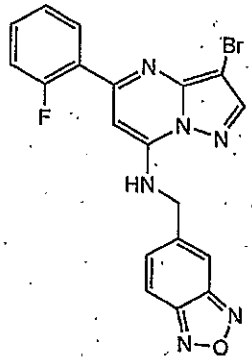
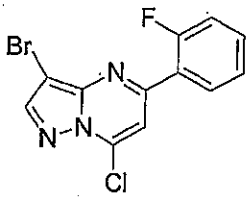
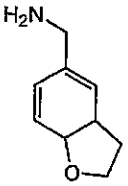
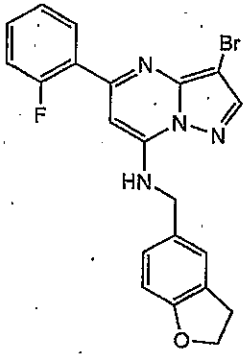
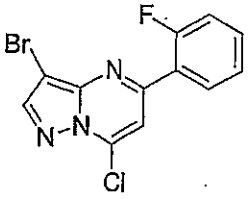
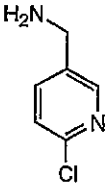
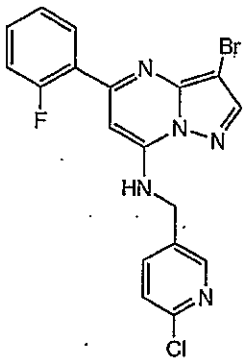
108				LCMS: MH^+ = 458;
109				Yield = 89 LCMS: MH^+ = 418 m. p. = 131-132 °C
110				Yield=95% LCMS: MH^+= 347
111				Yield=91% 3H); LCMS: MH^+= 484
112				Yield=87% LCMS: MH^+= 427

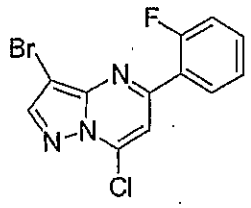
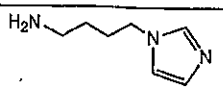
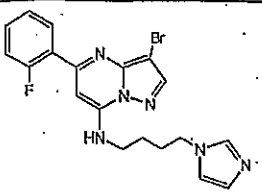
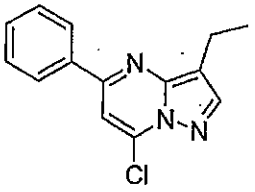
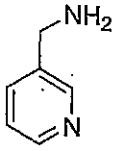
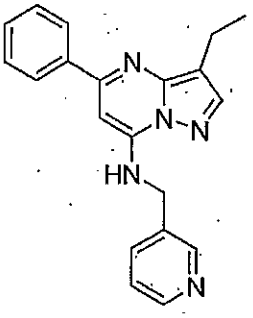
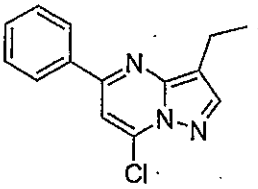
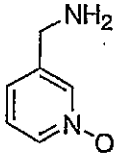
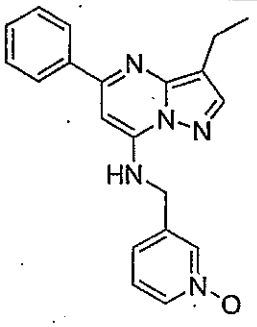
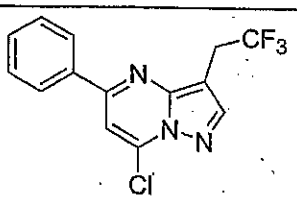
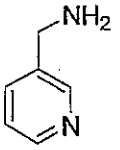
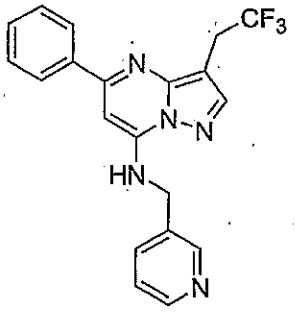
113				Yield=80% LCMS: MH^+= 427
114				Yield=91% LCMS: MH^+= 378
115				Yield=92% , 3H); LCMS: MH^+=520
116				Yield=98% LCMS: MH^+=536
117				Yield=82% LCMS: MH^+=410

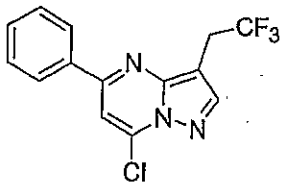
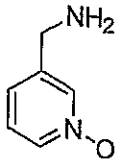
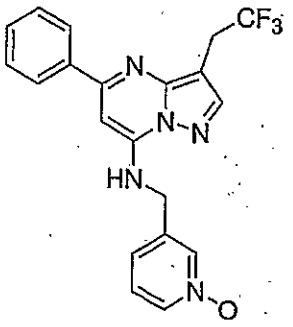
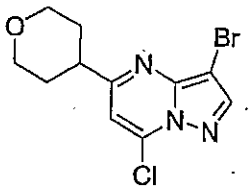
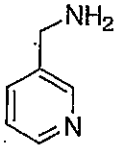
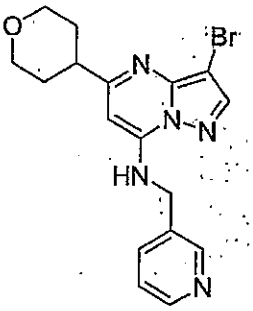
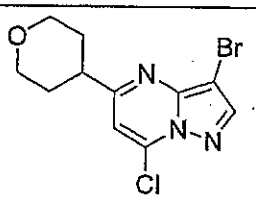
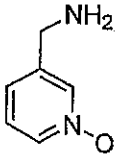
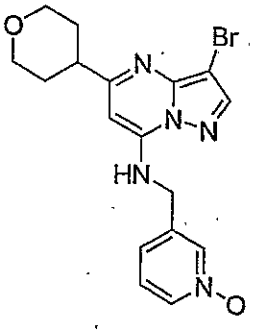
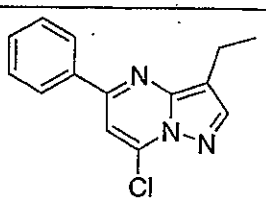
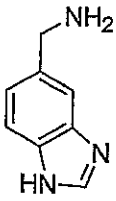
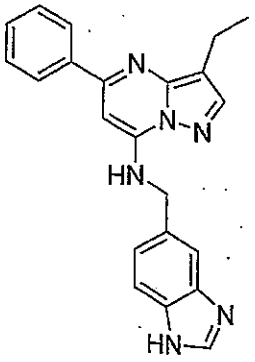
118				Yield=95% LCMS: MH^+= 347
121				Yield = 65% LCMS: MH^+ = 481.02
126				Yield=71% MH^+ = 486
127				Yield=71% MH^+ = 495.1

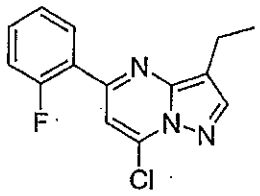
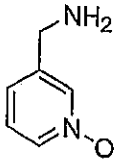
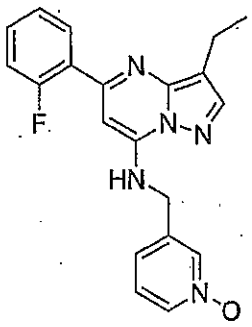
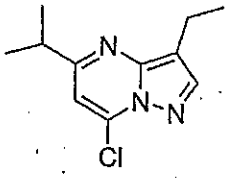
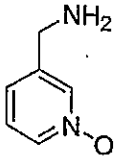
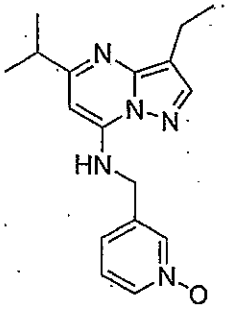
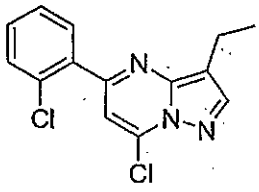
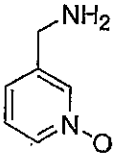
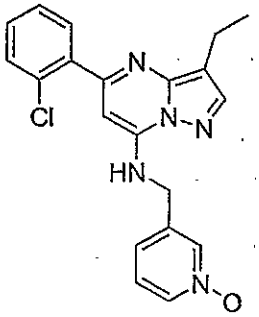
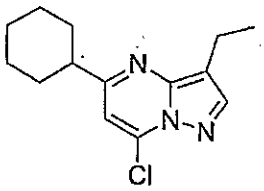
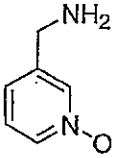
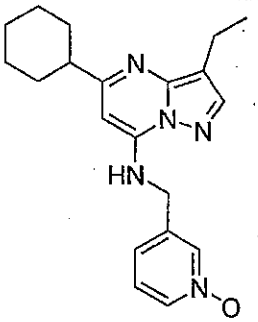
128				Yield=55% MH^+ = 463
129				Yield = 77% LCMS: MH^+ = 455
130				^1H NMR (Yield = 75% LCMS: MH^+ = 379
131				Yield = 75% LCMS: MH^+ = 407
132				Yield = 75% LCMS: MH^+ = 421

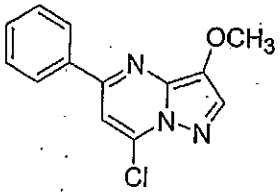
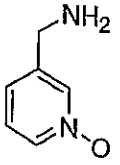
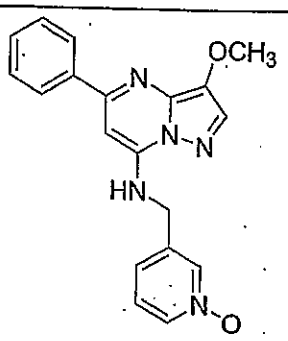
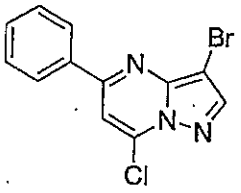
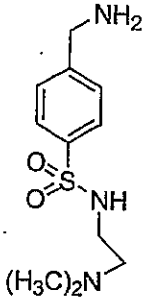
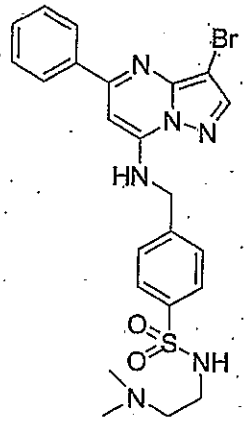
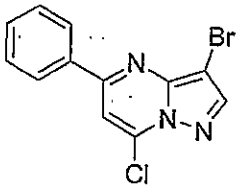
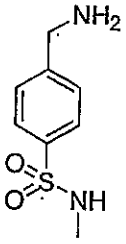
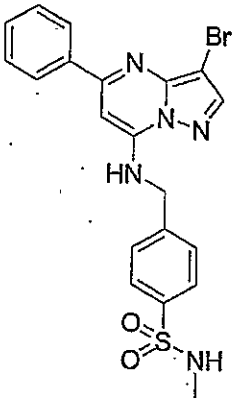
133				Yield = 70% LCMS: MH^+ = 421
134				Yield = 78% LCMS: MH^+ = 475
135				Yield = 75% LCMS: MH^+ = 476
136				Yield = 65% LCMS: MH^+ = 455

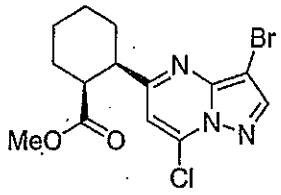
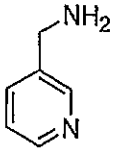
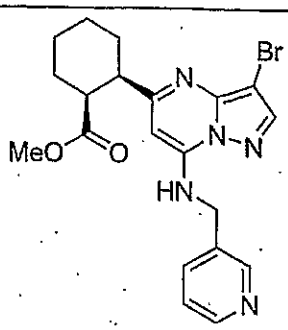
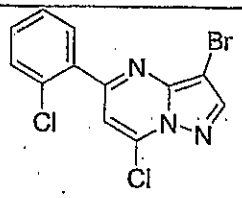
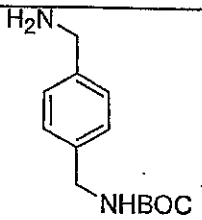
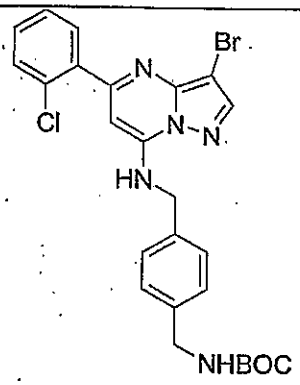
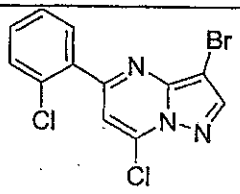
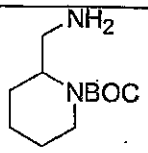
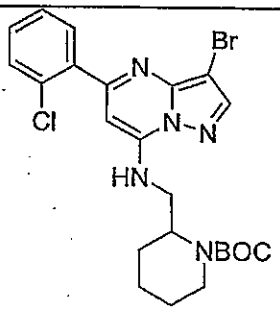
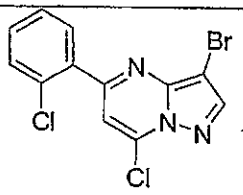
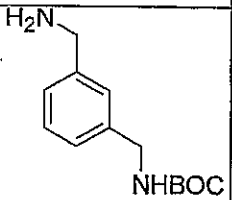
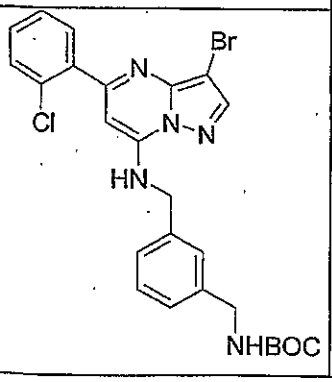
137				Yield = 55% LCMS: MH^+ = 473)
138				Yield = 60% LCMS: MH^+ = 439
139				Yield = 65% LCMS: MH^+ = 441
140				Yield = 80% LCMS: MH^+ = 432

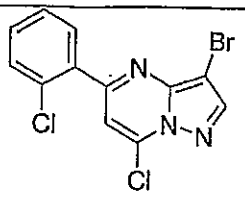
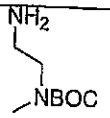
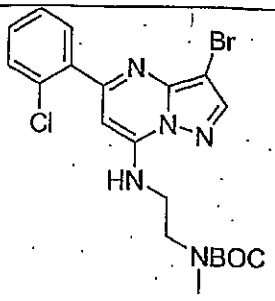
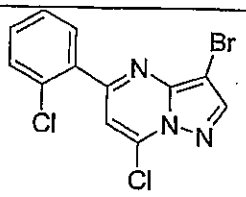
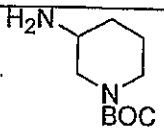
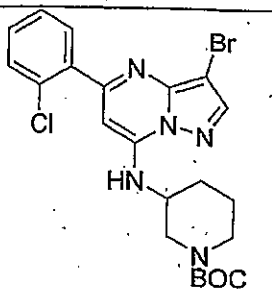
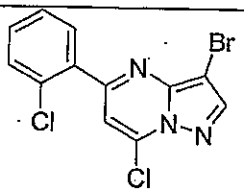
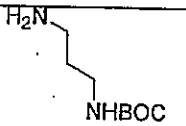
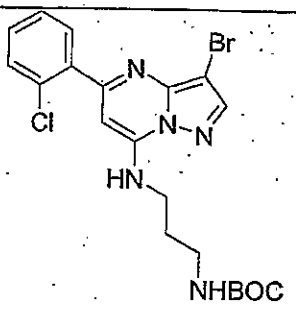
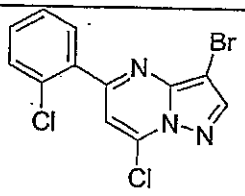
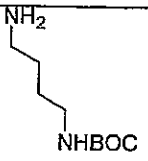
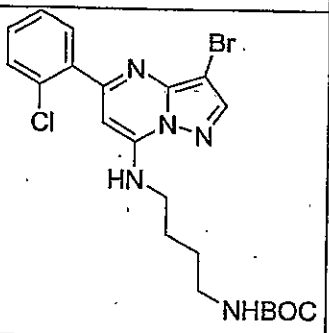
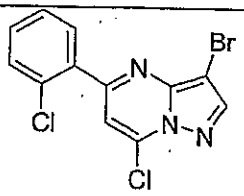
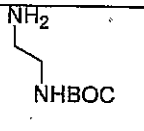
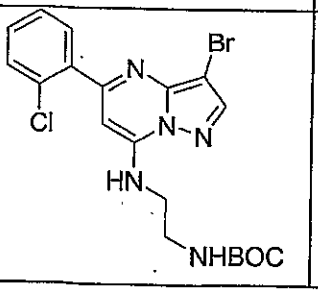
141				Yield = 60% LCMS: MH^+ = 429
142				LCMS: MH^+=330 ; mp=109-111°C
143				LCMS: MH^+=346 ; mp=186-188°C
144				LCMS: MH^+=384 ; mp=148-150°C

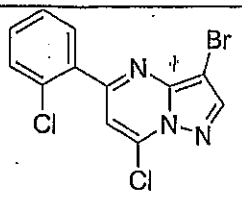
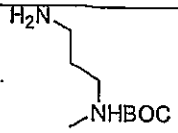
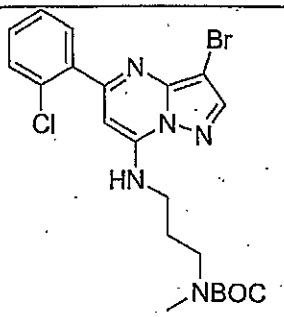
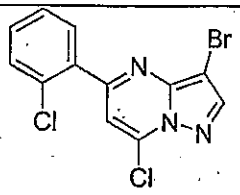
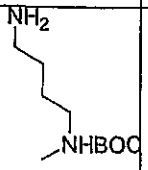
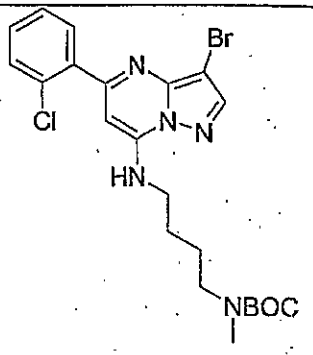
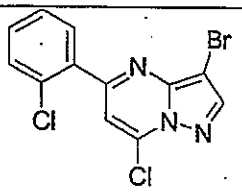
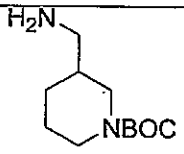
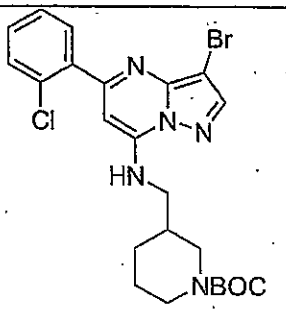
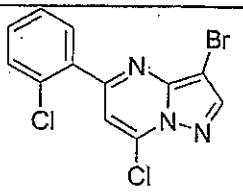
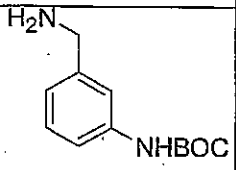
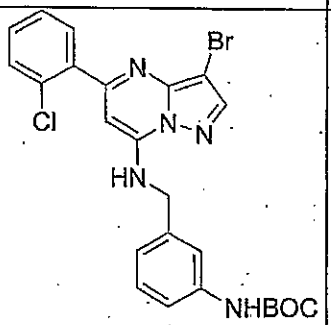
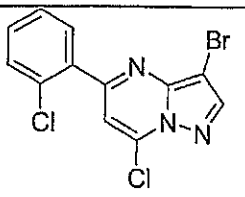
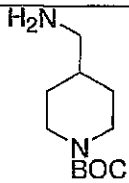
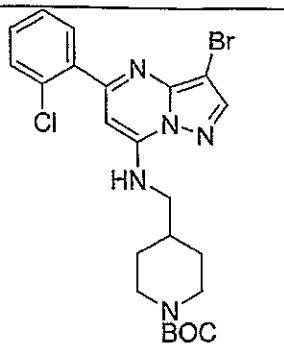
145				LCMS: MH^+=400; mp=186-188°C
146				LCMS: $M2H^+$=390; mp=192-194°C
147				LCMS: M^+=404; mp=220-222°C
148				LCMS: MH^+=369; mp>230° C

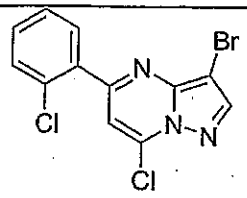
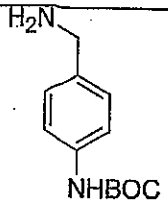
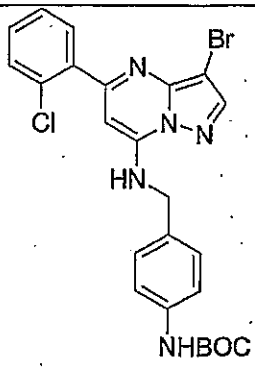
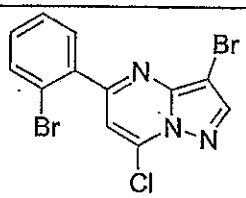
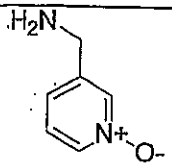
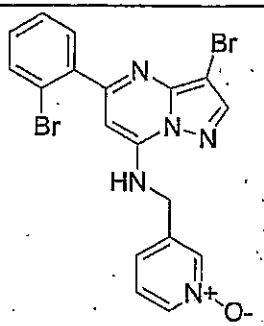
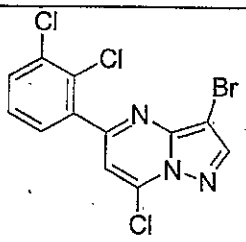
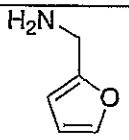
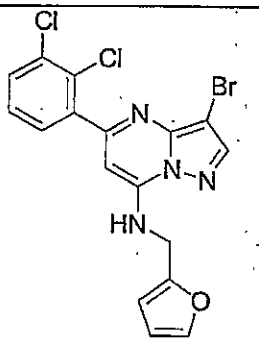
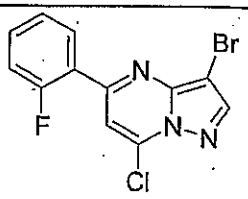
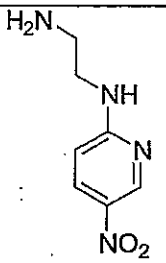
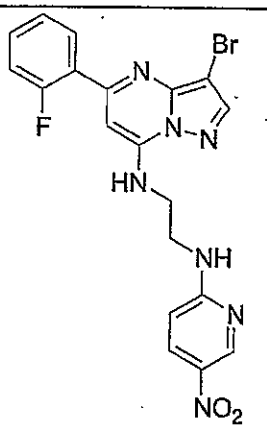
149				LCMS: MH^+=364 ; mp=186-188°C
150				LCMS: MH^+=312 ; mp=138-140°C
151				LCMS: M^+=380; mp=172-174°C
152				LCMS: MH^+=352 ; mp=201-203°C

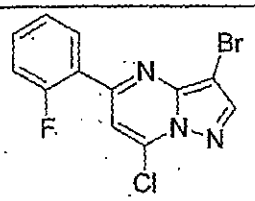
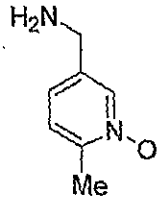
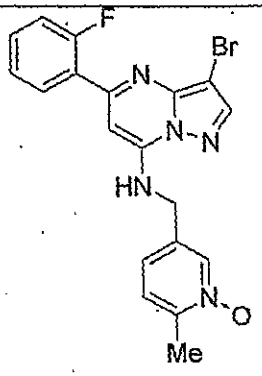
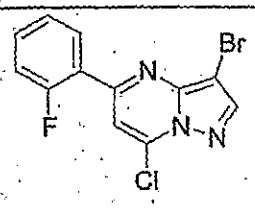
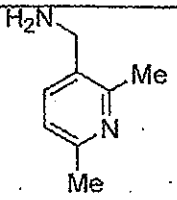
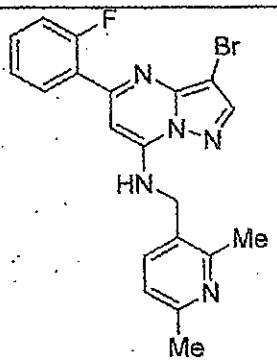
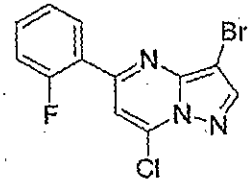
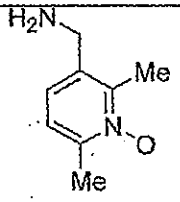
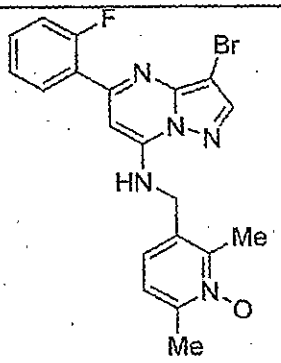
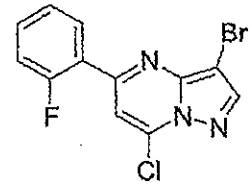
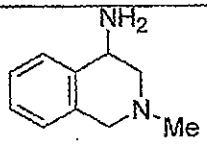
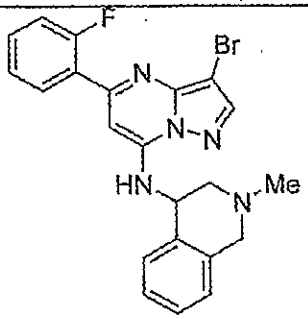
153				LCMS: MH^+=348; mp=166-168°C
154				LCMS: $M2H^+$=531; mp=78-80°C
155				LCMS: $M2H^+$=474; mp=161-163°C

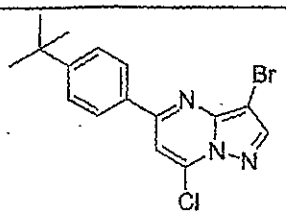
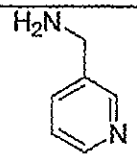
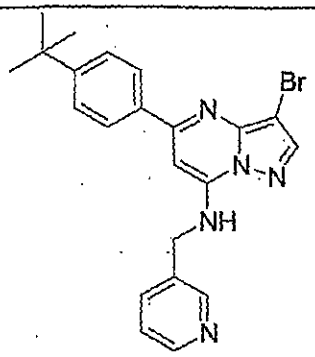
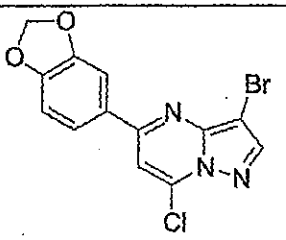
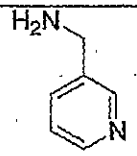
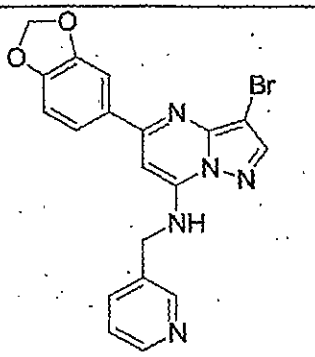
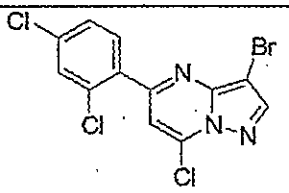
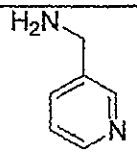
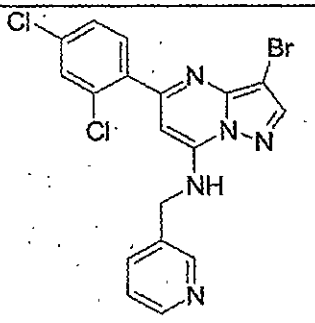
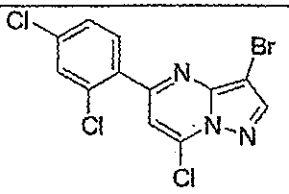
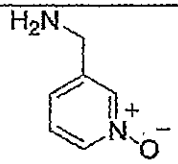
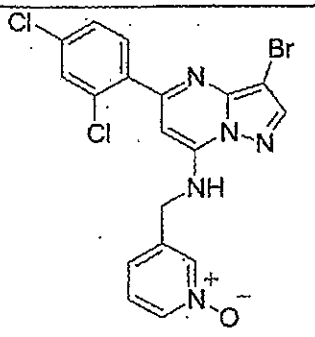
156				LCMS: M^+=444; mp=48-51°C
157				MH^+ = 542.1
158				MH^+ = 520.1
159				MH^+ = 542.1

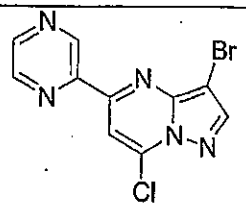
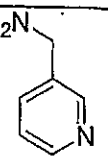
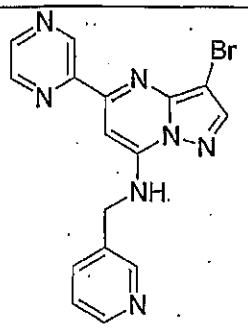
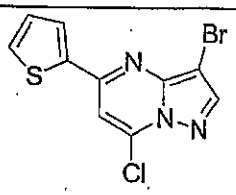
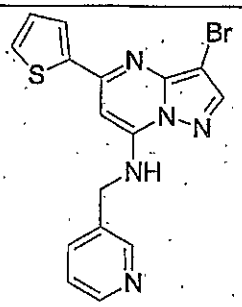
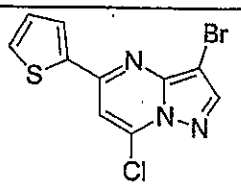
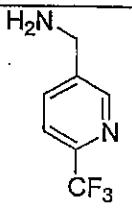
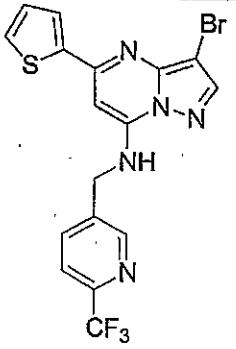
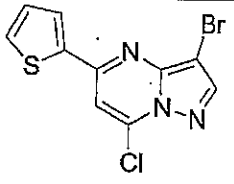
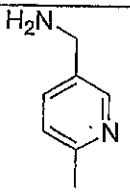
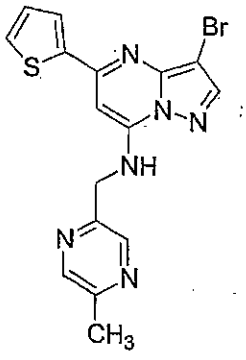
160				MH^+ = 480.1
161				MH^+ = 506.1
162				MH^+ = 480.1
163				MH^+ = 494.1
164				MH^+ = 466.1

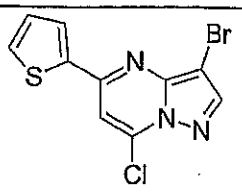
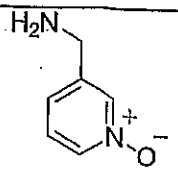
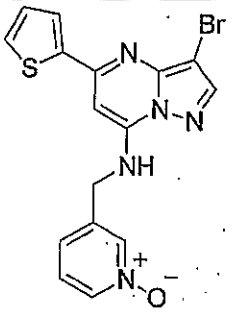
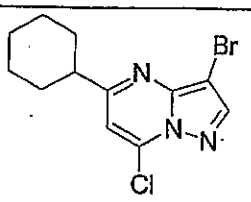
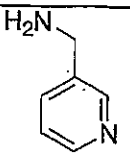
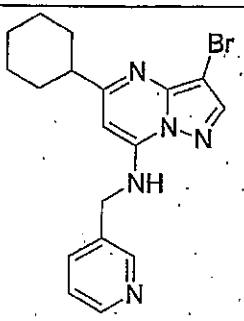
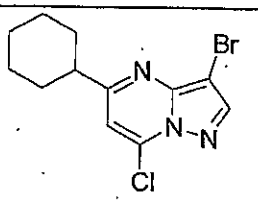
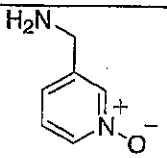
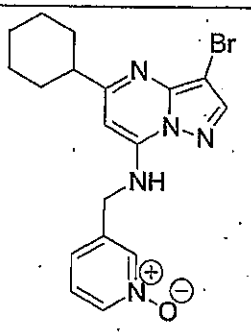
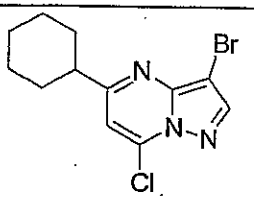
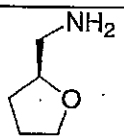
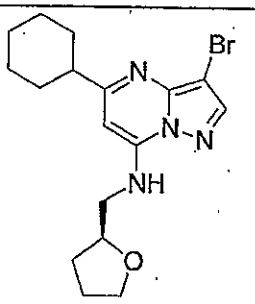
165				MH^+ = 494.1
166				MH^+ = 508.1
167				MH^+ = 520.1
168				MH^+ = 528.1
169				MH^+ = 520.1

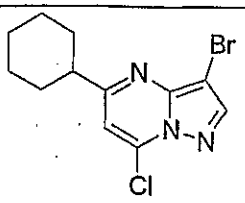
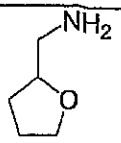
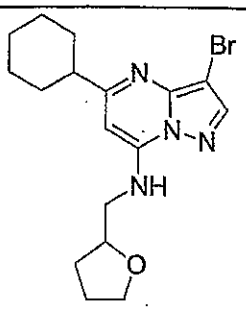
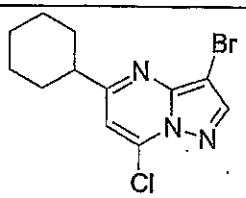
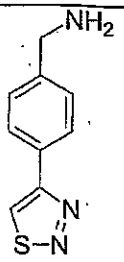
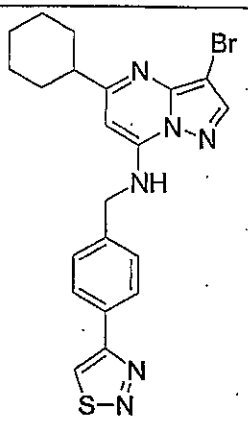
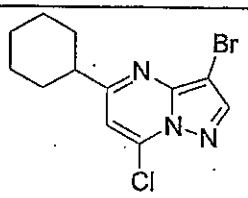
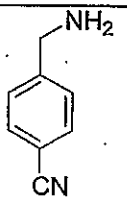
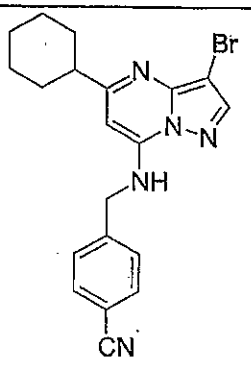
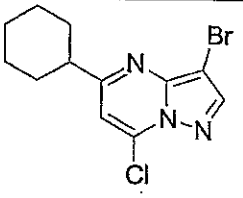
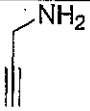
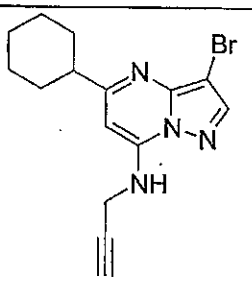
170				MH^+ = 528.1
171				LCMS: MH^+ = 474;
172				LCMS: MH^+ = 437;
173				LCMS: MH^+ = 472;

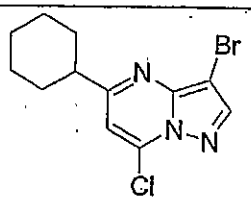
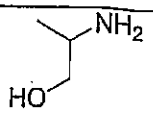
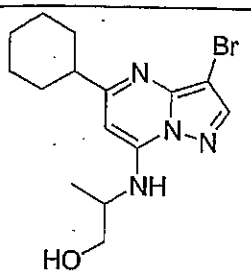
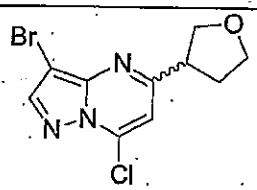
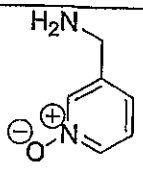
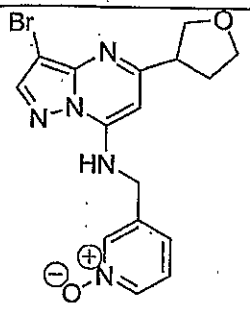
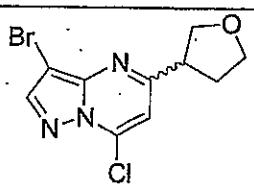
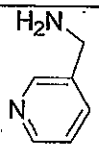
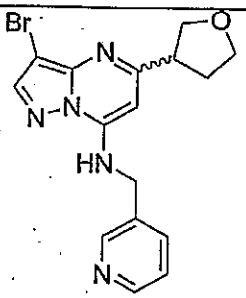
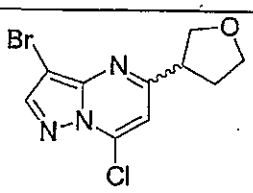
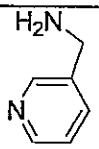
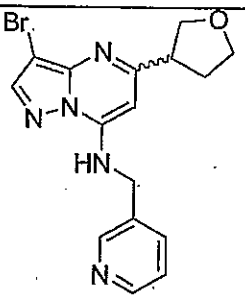
174				LCMS: MH^+ = 428.1
175				LCMS: MH^+ = 426.2
176				LCMS: MH^+ = 442.0
177				LCMS: MH^+ = 452.0

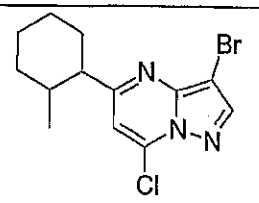
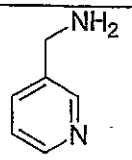
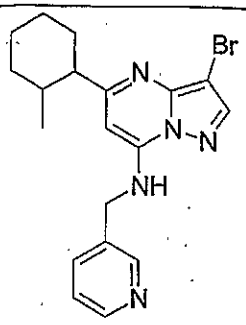
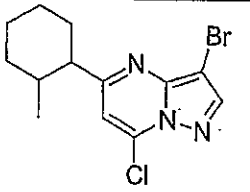
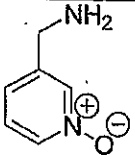
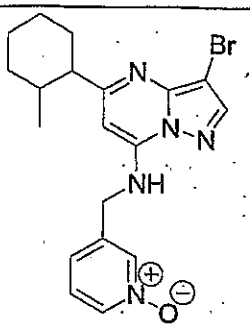
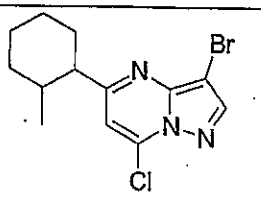
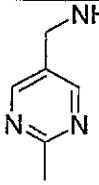
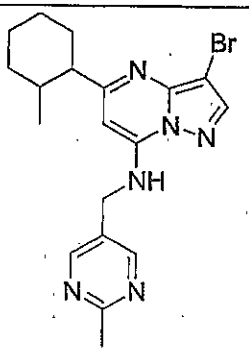
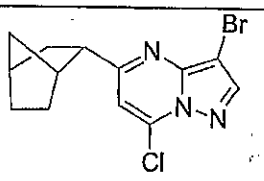
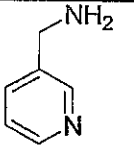
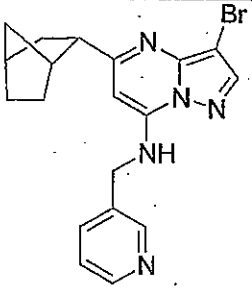
178				Yield = 90 MH^+ =436 m. pt. = 89.1 °C
179				MH^+ =424 m. pt. = 188.2 °C
180				MH^+ =448 m. pt. = 211.3 °C
181				Yield = quant. MH^+ = 464

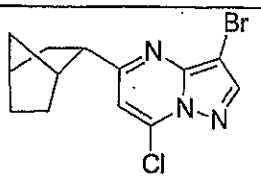
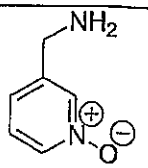
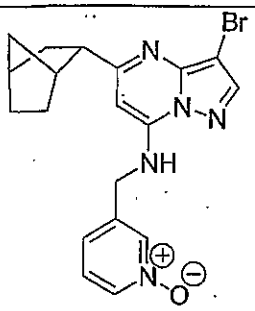
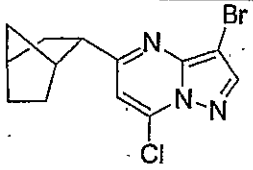
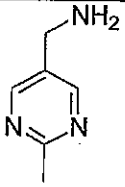
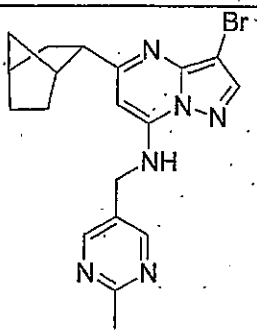
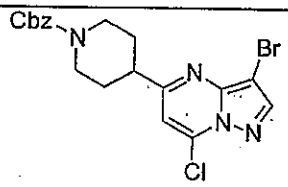
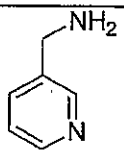
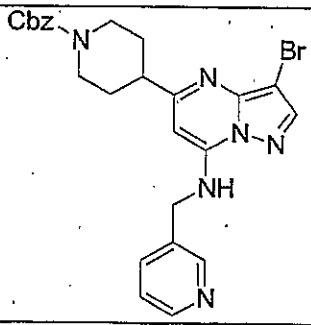
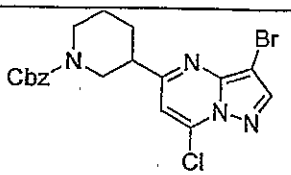
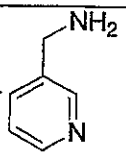
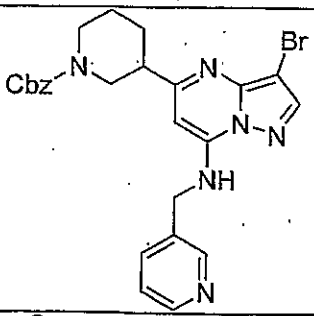
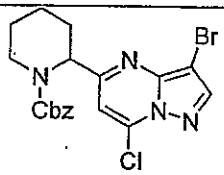
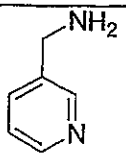
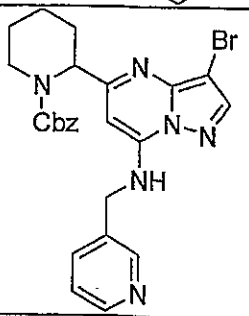
182				MH^+ = 382 m. pt. = 185.8 °C
183				MH^+ = 387 m. pt. = 181 – 182 °C
184				MH^+ = 453
185				MH^+ = 401 m. pt. = 178.3 °C

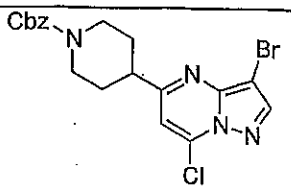
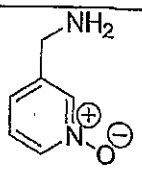
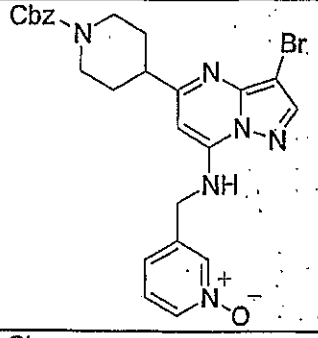
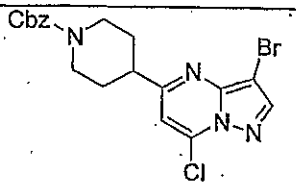
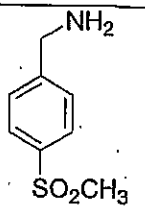
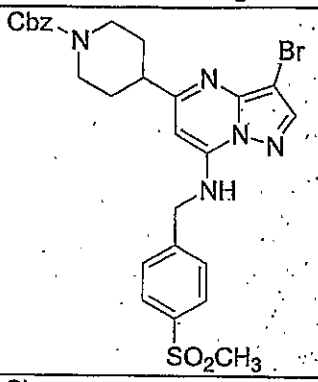
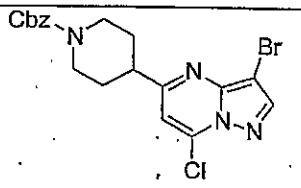
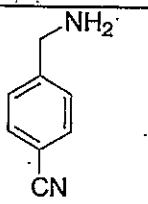
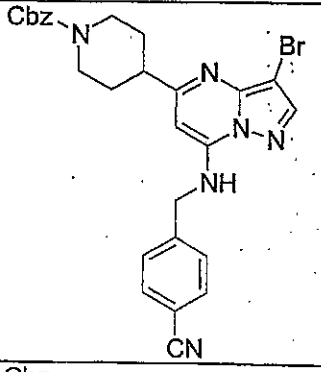
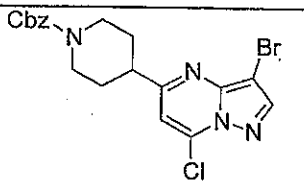
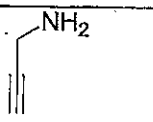
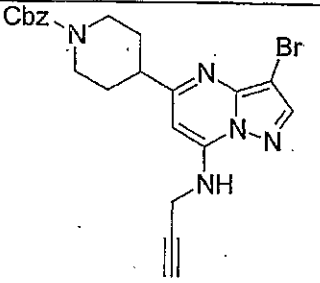
186				MH^+ = 402
187				Yield = 91 MH^+ = 386 m. pt. = 148.3 °C
188				Yield = 65 MH^+ = 402 m. pt. = 174.5 °C
189				MH^+ = 379 m. pt. = 82 – 83 °C

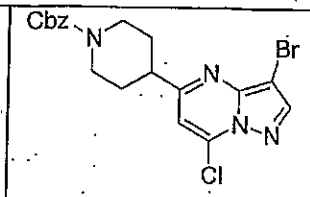
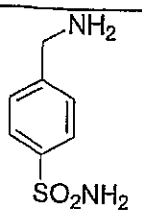
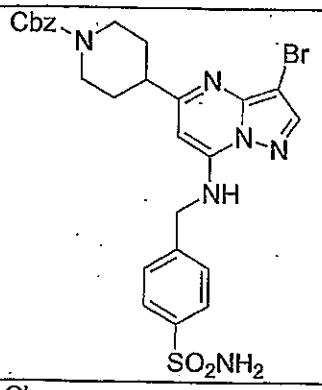
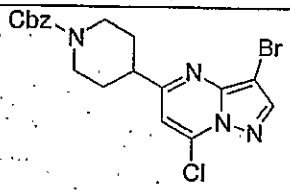
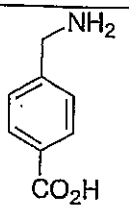
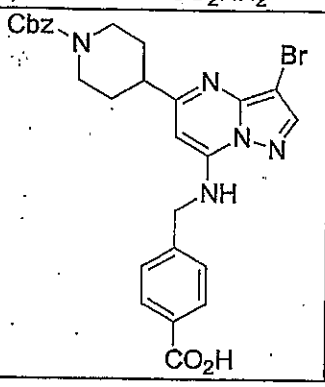
190				MH^+ = 379 m. pt. = 50.7 °C
191				Yield = 89 MH^+ = 469 m. pt. = 186.7°C
192				Yield = 93 MH^+ = 410 m. pt. = 86.7°C
193				Yield = 76 MH^+ = 333 m. pt. = 120.3°C

194				Yield = 86 MH^+ = 353 m. pt. = 188.9 °C
195				Yield=11% LCMS:374 MH^+= 390
196				Yield=88% LCMS:374 MH^+= 346
197				Yield=88% LCMS:374 MH^+= 346

198				Yield = MH^+ = 400 m. pt. = 111.5 – 112.2 °C
199				MH^+ = 416
200				MH^+ = 415
201				MH^+ =398 m.p. = 156.5°C

202				MH^+ = 414 m.p. = 89.5 °C
203				MH^+ = 413
204				Yield = 86 MH^+ = 521 m.p. = 79.9 °C
204 .10				
204 .11				Yield = 87 MH^+ = 521 m.p. = 128.6 °C

205				Yield = 99 MH^+ = 537 m.p. = 83.5 °C
206				Yield = 94 MH^+ = 598 m.p. = 110.8 °C
207				Yield = quant. MH^+ = 545
208				Yield = 96 MH^+ = 468 m.p. = 69.2 °C

209				MH^+ = 498 m.p. = 226.5 °C
210				MH^+ = 564 m.p. = 174.2 °C

Additional data for select examples given below.

Example 23: ^{1}H NMR (CD_3OD) δ 8.63 (d, J = 5.7 Hz, 2H), 8.18 (s, 1H), 7.81 (dd, J = 8.1 Hz, 2.1 Hz, 1H), 7.58 (d, J = 6.0 Hz, 2H), 7.48 (m, 1H), 7.15-7.10 (m, 2H), 6.50 (s, 1H), 4.86 (s, 2H), 3.70 (s, 3H)

Example 24: ^{1}H NMR ($CDCl_3$) δ 8.82 (s, 1H), 8.73 (d, J = 4.2 Hz, 1H), 8.11 (s, 1H), 8.06 (dd, J = 7.8 Hz, 1.8 Hz, 1H), 7.91 (d, J = 8.1 Hz, 1H), 7.53-7.47 (m, 2H), 7.20 (m, 1H), 7.08 (d, J = 8.1 Hz, 1H), 6.75 (s, 1H), 4.81 (d, J = 4.5 Hz, 2H), 3.86 (s, 3H)

Example 25: ^{1}H NMR ($CDCl_3$) δ 8.75 (d, J = 5.7 Hz, 2H), 8.12 (s, 1H), 7.81 (d, J = 2.1 Hz, 1H), 7.53 (dd, J = 8.4, 2.1 Hz, 1H), 7.45 (d, J = 6.0 Hz, 2H), 6.96 (t, J = 6.0 Hz, 2H), 6.33 (s, 1H), 4.85 (d, J = 6.0 Hz, 2H), 4.09 (s, 3H), 4.03 (s, 3H)

Example 26: ^{1}H NMR ($CDCl_3$) δ 8.82 (s, 1H), 8.72 (s, 1H), 8.09 (m, 1H), 7.87-7.83 (m, 2H), 7.60 (m, 1H), 7.45 (m, 1H), 7.03 (d, J = 8.4 Hz, 1H), 6.87 (s, 1H), 6.43 (s, 1H), 4.83 (d, J = 4.5 Hz, 2H), 4.11 (s, 3H), 4.04 (s, 3H)

Example 27: ^{1}H NMR ($CDCl_3$) δ 8.75 (d, J = 4.5 Hz, 2H), 8.19 (s, 1H), 7.63 (d, J = 7.8 Hz, 2H), 7.44-7.40 (m, 3H), 7.07 (m, 1H), 6.26 (s, 1H), 4.83 (d, J = 5.1 Hz, 2H)

Example 28: ^{1}H NMR ($CDCl_3$) δ 8.86 (s, 1H), 8.74 (m, 1H), 8.17 (s, 1H), 7.97 (m, 1H), 7.66-7.63 (m, 2H), 7.62 (m, 1H), 7.41 (m, 1H), 7.07 (m, 1H), 6.35 (s, 1H), 4.87 (d, J = 6.0 Hz, 2H)

Example 30: ^{1}H NMR ($CDCl_3$) δ 8.16 (s, 1H), 7.66-7.62 (m, 2H), 7.41 (m, 1H), 7.33-7.22 (m, 3H), 6.96 (t, J = 6.0 Hz, 1H), 6.33 (s, 1H), 4.73 (d, J = 6.0 Hz, 2H)

Example 31: ^{1}H NMR ($CDCl_3$) δ 8.13 (s, 1H), 7.66 (d, J = 7.8 Hz, 2H), 7.45-7.40 (m, 2H), 7.10-7.04 (m, 2H), 6.93 (t, J = 6.6 Hz, 1H), 6.60 (s, 1H), 4.84 (d, J = 6.6 Hz, 2H)

Example 32: ^{1}H NMR ($CDCl_3$) δ 8.16 (s, 1H), 7.66-7.62 (m, 2H), 7.57-7.55 (m, 2H), 7.41 (t, J = 7.8 Hz, 1H), 7.31 (dd, J = 7.8, 1.8 Hz, 1H), 6.99 (t, J = 6.0 Hz, 1H), 6.32 (s, 1H), 4.73 (d, J = 6.0 Hz, 2H)

Example 40: ^{1}H NMR ($CDCl_3$) δ 8.01 (s, 1H), 7.31 – 7.24 (d, J = 8.2 Hz, 1H), 6.72 – 6.64 (br t, J = 5.4 Hz, 1H), 6.62 – 6.52 (m, 2H), 6.05 – 6.01 (s, 1H), 5.56 – 4.64 (d, J = 6.0 Hz, 2H), 4.03 – 3.93 (s, 3H), 3.94 – 3.86 (s, 3H), 2.79 – 2.70 (d, J = 8.1 Hz, 2H), 2.02 – 1.66 (m, 6H), 1.43 –1.22 (m, 3H), 1.20 – 1.02 (m, 2H)

Example 45: ^{1}H NMR ($CDCl_3$) δ 8.73(d, 2H), 8.54(s, 1H), 7.41(d, 2H), 7.02(br, 1H), 5.90(s, 1H), 4.80(s, 2H), 4.48(q, 2H), 2.75(s, 2H), 1.50(t, 2H), 1.06(s, 9H);

Example 46: ^{1}H NMR ($CDCl_3$) δ 8.79(s, 1H), 8.72(d, 1H), 8.14(s, 1H), 7.84(d, 1H), 7.54-7.33(m, 4H), 6.97(t, 1H), 6.18(s, 1H), 4.79(d, 2H), 2.47(s, 3H)

Example 108: ^{1}H NMR ($CDCl_3$) δ 8.79 (s, 1H), 8.72 (d, J = 3.0 Hz, 1H), 8.16 (s, 1H), 7.84 (d, J = 7.8 Hz, 1H), 7.74 (d, J = 7.5 Hz, 2H), 7.55-7.35 (m, 3H), 6.92 (t, J = 6.3 Hz, 1H), 6.42 (s, 1H), 4.81 (d, J = 6.3 Hz, 2H)

Example 110: ^{1}H NMR ($CDCl_3$) δ 8.18(t, 1H), 8.03(s, 1H), 7.44(m, 1H), 7.30(t, 1H), 7.17(q, 1H), 6.66(s, 1H), 6.56(br, 1H), 4.28(d, 2H), 2.38(s, 1H)

Example 111: ^{1}H NMR ($CDCl_3$) δ 8.72(br, 1H), 8.59(d, 1H), 8.11(t, 1H), 8.06(s, 1H), 7.73(d, 1H), 7.44(d, 1H), 7.42-7.21(m, 3H), 7.07(q, 1H), 6.39(d, 1H), 5.21(q, 1H), 4.16(q, 2H), 3.08(d, 2H), 1.22(t, 3H)

Example 112: ^{1}H NMR ($CDCl_3$) δ 8.22(t, 1H), 8.15(s, 1H), 7.51-7.33(m, 7H), 7.21(q, 1H), 6.82(d, 1H), 6.51(s, 1H), 4.68(q, 1H), 2.18(m, 2H), 1.17(t, 3H)

Example 113: ^{1}H NMR ($CDCl_3$) δ 8.22(t, 1H), 8.14(s, 1H), 7.51-7.33(m, 7H), 7.21(q, 1H), 6.82(d, 1H), 6.51(s, 1H), 4.68(q, 1H), 2.18(m, 2H), 1.17(t, 3H)

Example 114: ^{1}H NMR ($CDCl_3$) δ 8.81(s, 1H), 8.75(d, 1H), 8.21(s, 1H), 7.84(d, 1H), 7.47(q, 1H), 6.96(s, 1H), 6.94(t, 1H), 4.85(d, 2H), 4.60(q, 2H), 1.58(t, 3H)

Example 115: ^{1}H NMR ($CDCl_3$) δ 8.77(s, 1H), 8.72(d, 1H), 8.14(s, 1H), 7.83(d, 1H), 7.65(d, 1H), 7.44(q, 1H), 7.80(t, 1H), 7.6(d, 1H), 6.18(s, 1H), 4.75(d, 2H), 3.91(s, 3H), 3.81(s, 3H)

Example 116: ^{1}H NMR ($CDCl_3$) δ 8.67(s, 1H), 8.55(d, 1H), 8.50(s, 1H), 7.92(d, 1H), 7.90(d, 1H), 7.78(t, 1H), 7.10(d, 1H), 6.97(s, 1H), 5.11(s, 2H), 3.77(s, 6H)

Example 117: ^{1}H NMR ($CDCl_3$) δ 8.38(s, 1H), 8.30(d, 1H), 8.17(s, 1H), 7.52-7.37(m, 6H), 6.97(t, 1H), 6.13(s, 1H), 4.77(d, 2H), 2.50(s, 3H)

Example 118: ^{1}H NMR ($CDCl_3$) δ 8.18(t, 1H), 8.03(s, 1H), 7.44(m, 1H), 7.30(t, 1H), 7.17(q, 1H), 6.66(s, 1H), 6.56(br, 1H), 4.28(d, 2H), 2.38(s, 1H);

Example 121: ^{1}H NMR ($CDCl_3$) δ 8.6 (S, 1H), 8.15 (dt,1H), 8.1 (s, 1H), 8.0 (d, 2H), 7.5 (d, 2H), 7.4 (dd, 1H), 7.2 (d, 1H), 7.15 (dd, 1H), 6.8 (t, 1H), 6.6 (s, 1H), 4.75 (d, 2H).

Example 126: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.5 (d, 1H), 7.42 – 7.35 (m, 2H), 7.3 –7.2 (m, 2H), 7.15 (dd, 1H), 7.1 (dd, 1H), 7.0 (t, 1H), 6.6 (s, 1H), 4.8 (d, 2H).

Example 127: ^{1}H NMR ($CDCl_3$) δ 8.2 (dt, 1H), 8.0 (s,1H), 7.4 (dd, 1H), 7.3-7.25 (m, 3H), 7.1 (dd, 1H), 6.9 - 6.85 (m, 2H), 6.7 (t, 1H), 6.6 (s, 1H), 4.6 (d, 2H), 3.2 (m, 4H), 2.6 (m, 4H), 2.3 (s, 3H)

Example 128: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt,1H), 8.1 (s, 1H), 8.0 (d, 2H), 7.5 (d, 2H), 7.4 (m, 2H), 7.25 (d, 1H), 7.2 (s,1H), 7.15 (dd, 1H), 7.0 (s, 1H), 6.8 (t, 1H), 6.6 (s, 1H), 4.75 (d, 2H).

Example 129: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.05 (s, 1H), 8.0 (d, 2H), 7.5 (d, 2H), 7.4 (m, 1H), 7.3 (dd, 1H), 7.15 (dd, 1H), 6.9 (t, 1H), 6.5 (s, 1H), 4.75 (d, 2H), 3.85 (s, 3H)

Example 130: ^{1}H NMR ($CDCl_3$) δ 8.2 (dt,1H), 8.0 (s, 1H), 7.4 (dd, 1H), 7.3(dd, 1H), 7.15 (dd, 1H), 6.8 (t, 1H), 6.4 (s,1H), 4.2 (d, 2H), 3.8 (s, 3H).

Example 131: ^{1}H NMR ($CDCl_3$) δ 8.2 (dt, 1H), 8.0 (s,1H), 7.4 – 7.15 (m, 3H), 6.7 (t, 1H), 4.2 (q, 2H), 3.8 (dt, 2H), 2.8 (t, 2H), 1.2 (t, 3H)

Example 132: ^{1}H NMR ($CDCl_3$) δ 8.2 (dt, 1H), 8.0 (s,1H), 7.4 – 7.15 (m, 3H), 6.7 (t, 1H), 4.2 (q, 2H), 3.8 (dt, 2H), 2.8 (t, 2H), 2.05 (m, 2H) 1.2 (t, 3H)

Example 133: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s,1H), 7.4 (m, 1H), 7.3 (dd 1H), 7.2 (dd, 1H), 6.5 (s, 1H), 6.4 (t,1H), 3.7 (s, 3H), 3.5 (dd,2H),2.4 (t, 2H), 1.8 (m, 4H)

Example 134: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.95 (d, 2H), 7.6 (d, 2H), 7.4 (m, 1H), 7.25 (dd, 1H), 7.1 (dd, 1H), 6.9 (t, 1H), 6.5 (s, 1H), 4.8 (d, 2H), 3.0 (s, 3H)

Example 135: ^{1}H NMR (DMSO d6) δ 9.1 (bs, 2H), 8.4 (s,1H), 8.0 (t, 1H), 7.85 (d, 2H), 7.7 (d, 2H), 7.6 (m, 1H), 7.4 (m, 2H), 6.6 (s, 1H), 4.8 (bs, 2H)

Example 136: ^{1}H NMR ($CDCl_3$) δ 8.2 (dt, 1H), 8.0 (s,1H), 7.4 (m, 1H), 7.25 (dd, 1H), 7.15 (dd, 1H), 6.9 (m, 3H), 6.7 (t, 1H), 6.5 (s, 1H), 4.5 (d, 2H), 4.2 (s, 4H)

Example 137: ^{1}H NMR ($CDCl_3$) δ 8.2 (dt, 1H), 8.0 (s, 1H), 7.4 (m, 1H), 7.3 (dd, 1H), 7.2 (dd, 1H), 6.9 (dd, 1H), 6.8 (t, 1H), 6.7 (m, 1H), 6.6 (s, 1H), 5.3 (s, 2H), 4.85 (s, 2H), 4.6 (d, 2H).

Example 138: ^{1}H NMR ($CDCl_3$) δ 8.2 (dt, 1H), 8.0 (s, 1H), 7.9 (d,1H), 7.8 (d, 1H), 7.4 (m, 2H), 7.3 (dd, 1H), 7.1 (dd, 1H), 6.9 (t, 1H), 6.6 (s, 1H), 4.8 (d, 2H)

Example 139: ^{1}H NMR ($CDCl_3$) δ 8.2 (dt, 1H), 8.0 (s, 1H), 7.4 (m, 1H), 7.3 (m, 2H), 7.2 (dd, 1H), 7.1 (dd, 1H), 6.8 (d, 1H), 6.7 (t, 1H), 6.6(s, 1H), 4.6 (m, 4H), 3.2 (t, 2H)

Example 140: ^{1}H NMR ($CDCl_3$) δ 8.45 (s, 1H), 8.2 (dt, 1H), 8.0 (s, 1H), 7.7 (dd, 1H), 7.4 – 7.3 (m, 3H), 7.15 (dd, 1H), 6.8 (t, 1H), 6.6 (s,1H), 4.7 (d, 2H)

Example 141: ^{1}H NMR ($CDCl_3$) δ 8.2 (dt, 1H), 8.0 (s, 1H), 7.45 - 7.1 (m, 7H), 6.6 (s, 1H), 4.4 (dt, 2H), 2.6 (t, 2H), 1.8 (m, 2H), 1.4 (m, 2H)

Example 171: ^{1}H NMR (CD_3OD) δ 8.41 (s, 1H), 8.25 (d, J = 6.3 Hz, 1H), 8.15 (s, 1H), 7.67 (d, J = 7.8 Hz, 2H), 7.55-7.48 (m, 2H), 7.45 (dd, J = 7.5, 1.2 Hz, 1H), 7.34 (dd, J = 7.5, 1.8 Hz, 1H), 6.28 (s, 1H), 4.79 (s, 2H).

Example 172: ^{1}H NMR ($CDCl_3$) δ 8.64 (s, 1H), 7.68-7.64 (m, 2H), 7.52 (m, 1H), 7.43 (t, J = 7.8 Hz, 1H), 6.89 (t, J = 6.0 Hz, 1H), 6.51 (s, 1H), 6.48 (m, 2H), 4.74 (d, J = 6.0 Hz, 2H).

Example 173: ^{1}H NMR (DMSO-d_6) δ 8.86 (s, 1H), 8.46 (s, 1H), 8.32-8.28 (m, 2H), 7.97 (m, 1H), 7.87 (m, 1H), 7.52 (m, 1H), 7.35-7.24 (m, 2H), 6.57 (s, 1H), 6.46 (m, 1H), 3.65 (m, 4H).

Example 174: ^{1}H NMR ($CDCl_3$) d 8.37 (s, 1H), 8.16 (t, J = 7.5 Hz, 1H), 7.45-7.35 (m, 1H), 7.32-7.20 (m, 3H), 7.17-7.07 (m, 1H), 6.92 (t, J = 6 Hz, 1H), 6.48 (s, 1H), 4.65 (d, 2H), 2.50 (s, 3H).

Example 175: ^{1}H NMR ($CDCl_3$) d 8.16 (t, J = 9 Hz, 1H), 8.00 (s, 1H), 7.49 (d, J= 9 Hz, 1H), 7.46-7.36 (m, 1H), 7.18-7.08 (m, 1H), 7.00 (d, J = 9 Hz, 1H), 6.62-6.50 (m, 2H), 2.60 (s, 3H), 2.55 (s, 3H).

Example 176: ^{1}H NMR ($CDCl_3$) d 8.15 (t, J = 9 Hz, 1H), 8.00 (s, 1H), 7.45-7.35 (m, 1H), 7.32-7.20 (m, 1H), 7.20-7.05 (m, 3H), 6.80 (t, 1H), 6.50 (s, 1H), 4.65 (d, 2H), 2.65 (s, 3H), 2.50 (s, 3H).

Example 177: ^{1}H NMR ($CDCl_3$) d 8.20 (t, 1H), 7.90 (s, 1H), 7.50-7.05 (m, 8H), 6.80 (s, 1H), 5.05-4.90 (m, 2H), 3.80 (d, 1H), 3.45 (d, 1H), 3.00 (dd, 1H), 2.90 (dd, 1H), 2.50 (s, 3H).

Example 181: ^{1}H NMR (300MHz, $CDCl_3$) □ 8.41 (s, 1H), 8.28 – 8.23 (d, 1H), 8.15 (s, 1H), 7.69 –7.60 (d, 1H), 7.62 – 7.50 (m, 3H), 7.50 – 7.47 (dd, 1H), 6.35 (s, 1H), 5.36 (s, 1H), 4.80 (s, 2H).

Example 184: ^{1}H NMR (300MHz, $CDCl_3$) □8.96 - 8.90 (s, 1H), 8.08 (s, 1H), 8.04 (d, 1H), 7.72 (d, 1H), 7.70 – 7.61 (dd, 1H), 7.24 – 7.20 (dd, 1H), 6.92 – 6.84 (t, 1H), 6.36 (s, 1H), 4.96 – 4.89 (d, 2H).

Example 186: ^{1}H NMR (300MHz, $CDCl_3$) □8.96 - 8.90 (s, 1H), 8.08 (s, 1H), 8.44 (s, 1H), 8.27 – 8.24 (d, 1H), 8.02 (s, 1H), 7.78 – 7.76 (d, 1H), 7.73 – 7.70 (d, 1H), 7.58 – 7.51 (m, 2H), 7.13 –7.08 (dd, 1H), 5.51 (s, 2H).

Example 195: ^{1}H NMR (CD_3OD) δ 8.40(s, 1H), 8.27(d, 1H), 8.03(s, 1H), 7.75-7.50(m, 2H), 6.10(s, 1H), 4.76(s, 2H), 4.05(m, 2H), 3.88(m, 2H), 3.52(m, 1H), 2.33(m, 1H), 2.20(m, 1H).

Example 196: ^{1}H NMR (CD_3OD) δ 8.73(d, 1H), 8.58(q, 1H), 8.12(s, 1H), 8.00(d, 1H), 7.54(q, 1H), 6.19(s, 1H), 4.86(s, 2H), 4.22-4.08(m, 2H), 4.03-3.93(m, 2H), 3.63(m, 1H), 2.50-2.39(m, 1H), 2.32-2.21(m, 1H).

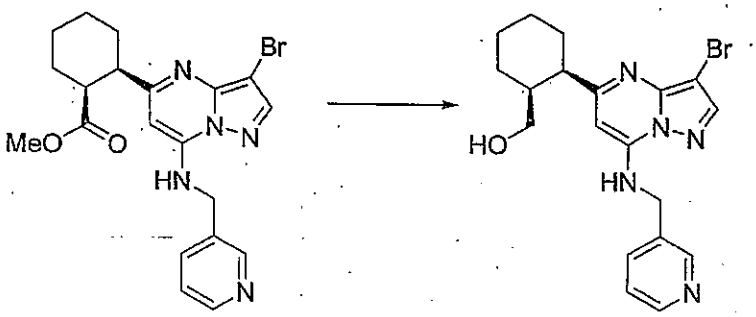
Example 197: ^{1}H NMR (CD_3OD) δ 8.73(d, 1H), 8.58(q, 1H), 8.12(s, 1H), 8.00(d, 1H), 7.54(q, 1H), 6.19(s, 1H), 4.86(s, 2H), 4.22-4.08(m, 2H), 4.03-3.93(m, 2H), 3.63(m, 1H), 2.50-2.39(m, 1H), 2.32-2.21(m, 1H).

Example 199: ^{1}H NMR (300MHz, $CDCl_3$) □8.29 (s, 1H), 8.15 (br s, 1H), 7.95 (s, 1H), 7.28 (d, 1H), 7.05 - 6.95 (appt t, 1H), 5.70 (s, 1H), 4.62 (d, 2H), 2.90 (m, 1H), 2.30 (m, 1H), 1.9 – 1.2 (m, 8H), 0.65 (d, 3H).

Example 200: ^{1}H NMR (300MHz, $CDCl_3$) □ 8.71 (s, 2H), 8.00 (s, 1H), 6.13 (s, 1H), 3.59 (s, 2H), 3.01 – 2.58 (m, 1H), 2.51 – 2.45 (m, 1H), 2.44 –2.30 (m,1H), 2.20 (s, 3H), 2.09 – 1.95 (m, 2H), 1.85 –1.70 (m, 2H), 0.80 – 0.76 (d, 3H).

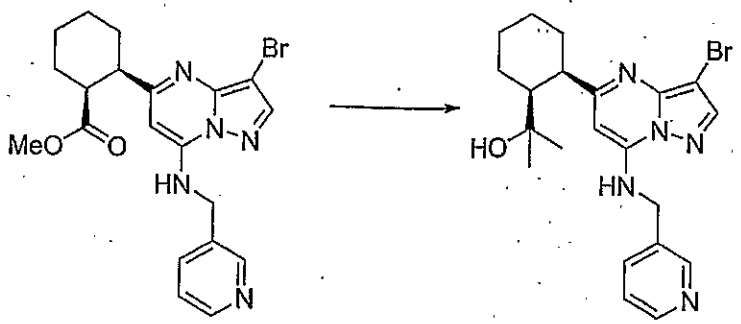
Example 203: ^{1}H NMR (300MHz, $CDCl_3$) □8.10 (s, 1H), 8.08 (s, 1H), 6.27 (s, 2H), 4.95 (s, 2H), 3.00 – 2.90 (dd, 2H), 2.60 (m, 2H), 2.48 (br s, 1H), 2.39 (s, 3h), 2.25 m, 1H), 1.95 – 1.70 (m, 3H).

EXAMPLE 211:



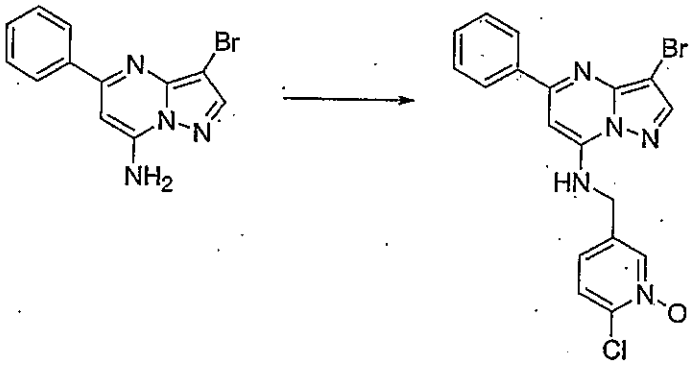
To a solution of the compound prepared in Example 156 (100 mg, 0.23 mmol) in dry THF (4 mL) was added $LiAlH_4$ (1.0 M in THF, 0.110 mL, 0.110 mmol) at 0°C under N_2. The mixture was stirred at 0°C for 1hr, warmed to 25°C, then additional $LiAlH_4$ (1.0 M in THF, 0.400 mL) was added, the mixture was stirred for 20 min and then quenched with MeOH (2.0 mL). The solvent was evaporated and the crude product was purified by flash chromatography using 10:1 CH_2Cl_2:MeOH as eluent. White solid (46 mg, 49%) was obtained. LCMS: M^+= 416. Mp=71-72 °C.

EXAMPLE 212:



To a solution of the compound prepared in Example 156 (70 mg, 0.16 mmol) in dry THF (3 mL) was added MeMgBr (3.0 M in Et_2O, 1.10 mL, 3.20 mmol) under N_2. The mixture was stirred at 25°C for 45 min and then quenched with saturated aqueous NH_4Cl (5.0 mL). The mixture was poured into saturated aqueous NH_4Cl (30 mL) and extracted with CH_2Cl_2 (3x20 mL). The extracts were dried over Na_2SO_4 and filtered. The solvent was evaporated and the crude product was purified by flash chromatography using 20:1 CH_2Cl_2:MeOH as eluent. White solid (25 mg, 36%) was obtained. LCMS: M^+= 444. Mp=76-80 °C.

EXAMPLE 213:



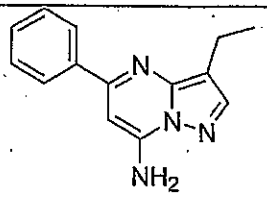
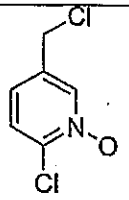
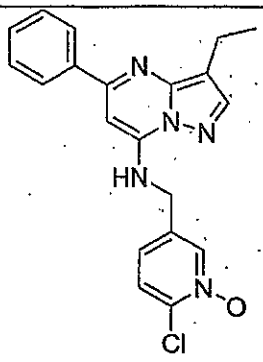
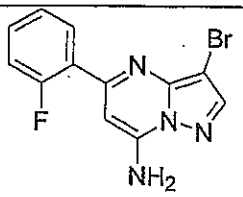
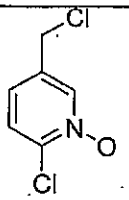
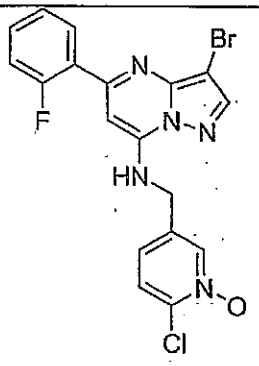
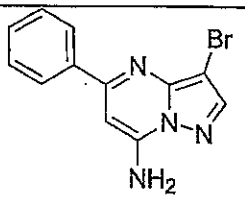
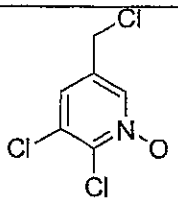
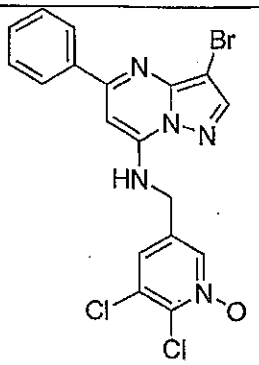
Anhydrous DMF (40 mL) was added under N_2 to the compound prepared in Preparative Example 174 (2.50 g, 8.65 mmol) and 60 % NaH in mineral oil (346 mg, 8.65 mmol). The mixture was stirred at 25°C for 1 hr, then 2-chloro-5-chloromethylpyridine N-oxide (1.54 g, 8.65 mmol) in anhydrous DMF (20 mL) was added slowly. The mixture was stirred at 25°C for 18 hr, the solvent was evaporated and the crude product was purified by flash chromatography using

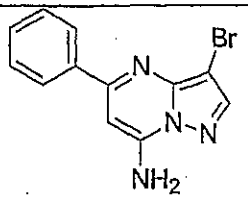
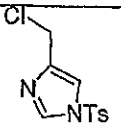
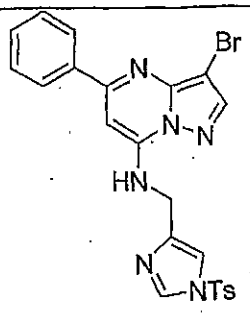
30:1 CH_2Cl_2:MeOH as eluent. So obtained solid was triturated by 50 mL of 1:1 EtOAc: hexane. Pale yellow solid (1.25 g, 34%) was obtained. LCMS: MH^+=432. Mp=224-226 °C.

EXAMPLES 214-217:

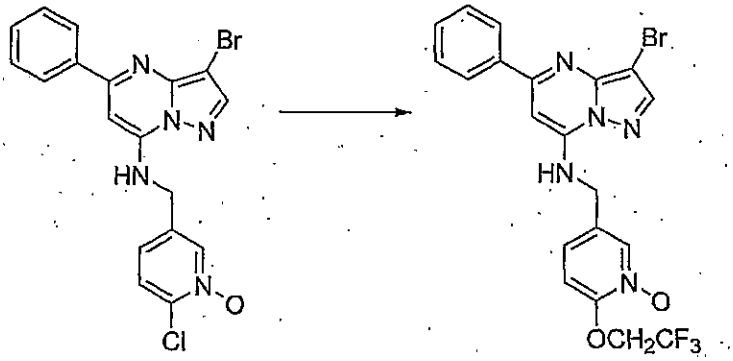
By essentially the same procedure set forth in Example 213 combining the compounds shown in Column 2 of Table 19 with compounds in Column 3 of Table 19, the compounds shown in Column 3 of Table 19 were prepared.

TABLE 19

Ex.	Column 2	Column 3	Column 4	CMPD
214				LCMS: MH^+=380; mp=°C
215				LCMS: MH^+=450; mp=218-222°C
216				LCMS: MH^+=466; mp=126-128°C

217				LCMS: M^+=523

EXAMPLE 218:



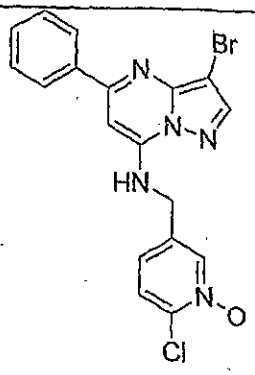
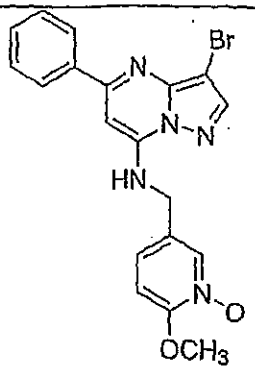
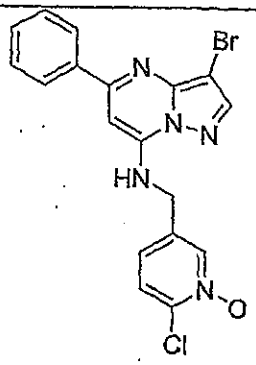
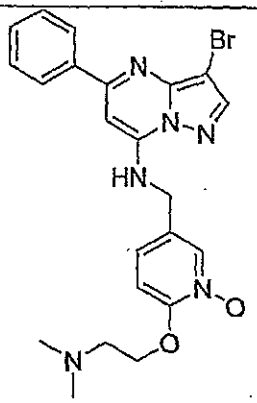
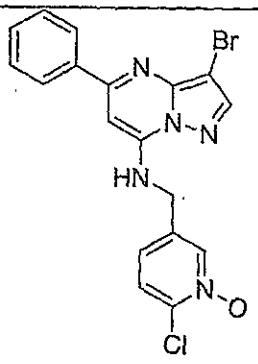
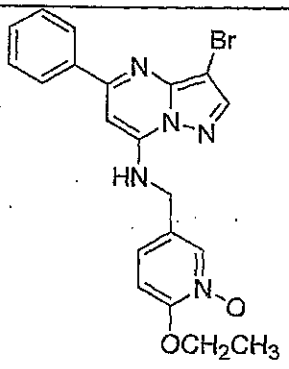
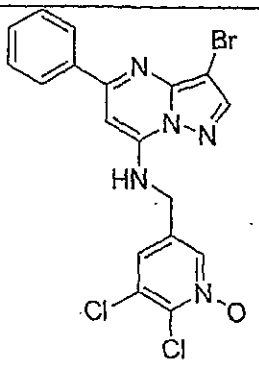
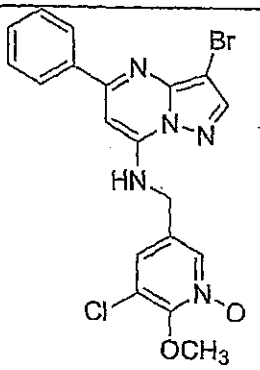
CF_3CH_2OH (3.0 mL) was added under N_2 to 60% NaH in mineral oil (40 mg, 1.0 mmol), the mixture was stirred for 20 min, then the product prepared in Example 213 (50 mg, 0.12 mmol) was added. The mixture was refluxed for 20 hr, the solvent was evaporated, and the residue was purified by flash chromatography using 20:1 CH_2Cl_2:MeOH as eluent to yield pale yellow solid (35 mg, 61%). LCMS: $M2H^+$=496. Mp=208-210 °C.

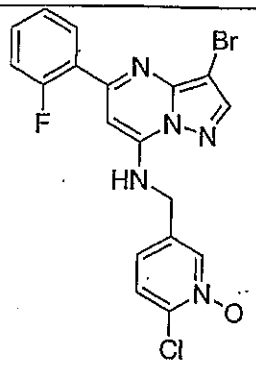
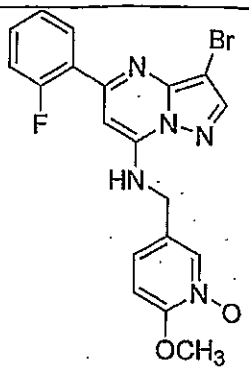
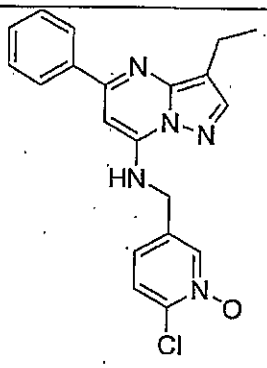
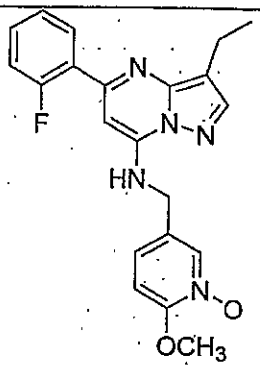
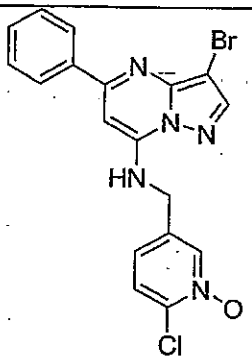
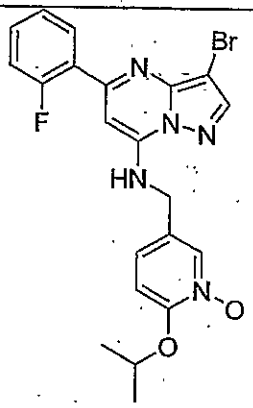
EXAMPLES 219-225:

By essentially the same procedure set forth in Example 218 combining the compounds shown in Column 1 of Table 20 with the appropriate alcohol , the compounds shown in Column 2 of Table 20 were prepared.

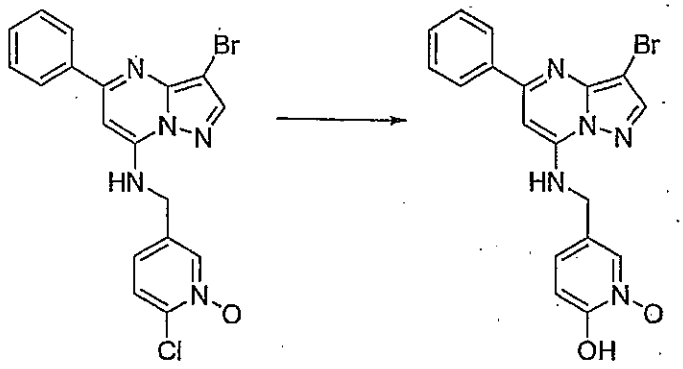
TABLE 20

Ex.	Column 1	Column 2	Data

219			LCMS: M^+=426; mp=126-128°C
220			LCMS: M^+=483; mp=89-91°C
221			LCMS: $M2H^+$=442; mp=112-114°C
222			LCMS: MH^+=462; mp=121-123°C

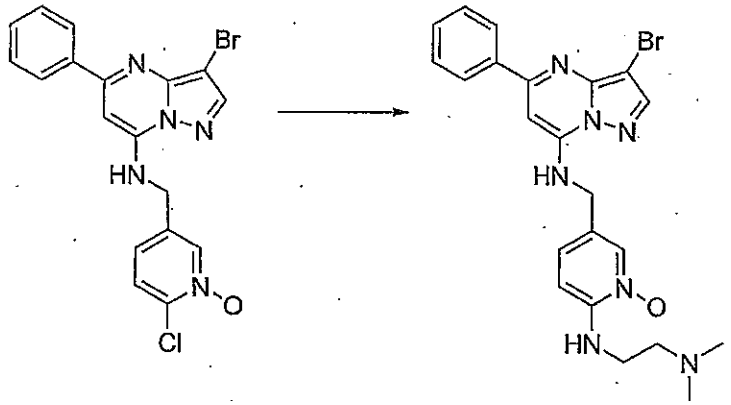
223			LCMS: MH^+=444; mp=112-114°C
224			LCMS: M^+=376; mp=°C
225			LCMS: MH^+=; mp=°C

EXAMPLE 226:



A mixture of the product prepared in Example 213 (100 mg, 0.23 mmol) and KOH (95 mg, 1.70 mmol) in 1,2-dimethoxyethane (3mL) and H_2O (1.5 mL) was refluxed under N_2 for 20 hr, quenched with acetic acid (0.30 mL), and the solvent was evaporated. The residue was suspended in H_2O (15 mL), filtered and the solid was washed with H_2O (15 mL) and Et_2O (10 mL). Then it was mixed with CH_2Cl_2 (2 mL) and Et_2O (2 mL) and filtered. Et_2O (5 mL) was added to the filtrate and the mixture was allowed to stand overnight. The solid was removed by filtration, washed with Et_2O and then dissolved in MeOH (5 mL). The solution was filtered and the solvent from the filtrate was evaporated. Off-white solid (5 mg, 5%) was obtained. LCMS: M^+=412. Mp= 206-208°C.

EXAMPLE 227:



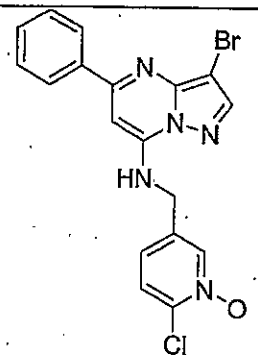
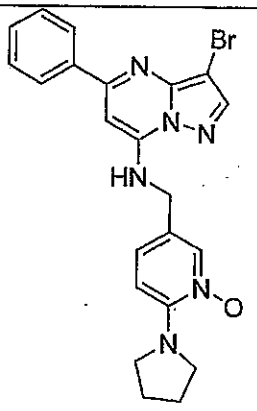
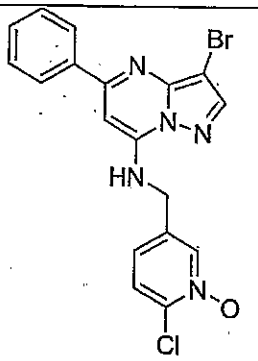
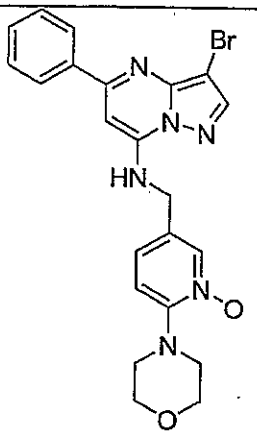
A mixture of the product prepared in Example 213 (129 mg, 0.30 mmol), N,N-dimethylethylenediamine (0.165 mL, 1.50 mmol), and diisopropylethylamine (0.10 mL) in anhydrous N-methylpyrrolidinone (1.0 mL) was stirred at 100°C for

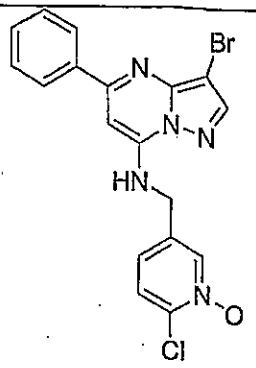
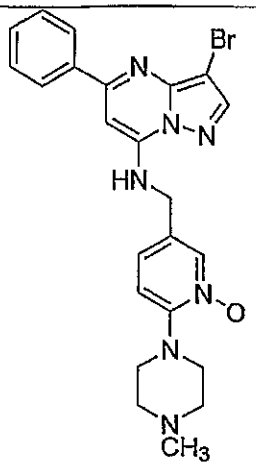
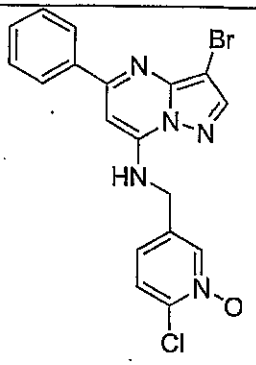
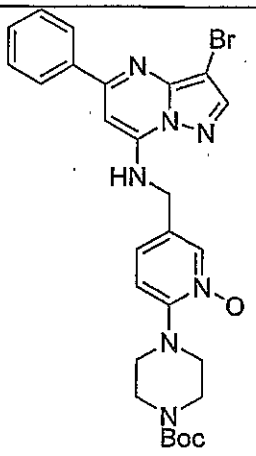
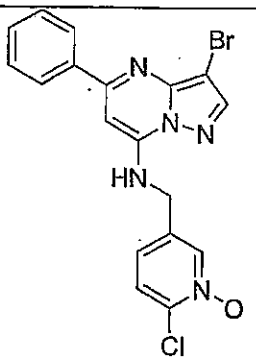
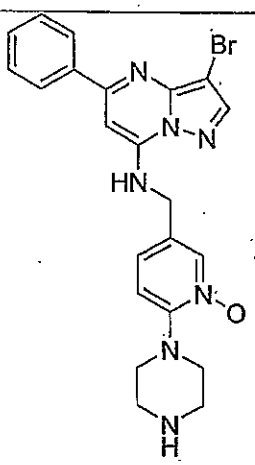
24 hr. The solvent was evaporated, and the residue was purified by flash chromatography using 20:1 CH_2Cl_2: 7N NH_3 in MeOH as eluent to yield pale yellow solid (110 mg, 76%). LCMS: M^+=482. Mp=76-78 °C.

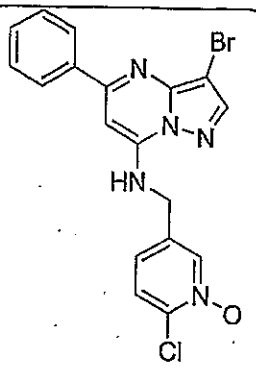
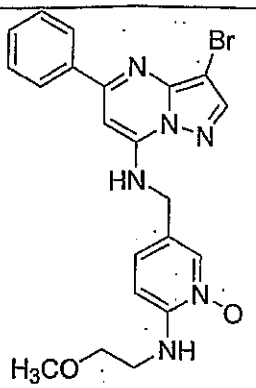
EXAMPLES 228-233:

By essentially the same procedure set forth in Example 227 combining the compounds shown in Column 1 of Table 21 with the appropriate amine, the compounds shown in Column 2 of Table 21 were prepared.

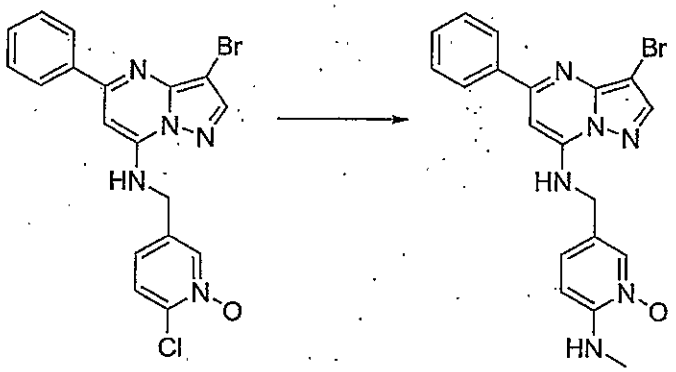
TABLE 21

Ex.	Column 1	Column 2	Data
228			LCMS: $M2H^+$=467 ; mp126-128=°C
229			LCMS: M^+=481; mp=128-130°C

230			LCMS: M^+=494; mp=108-110°C
231			LCMS: $M2H^+$=482 ; mp=129-133°C
232			LCMS: $M2H^+$=482 ; mp=124-126°C

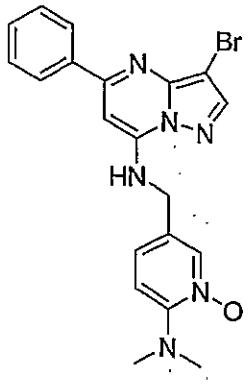
233			LCMS: $M2H^+$=471 ; mp=88-90°C

EXAMPLE 234:



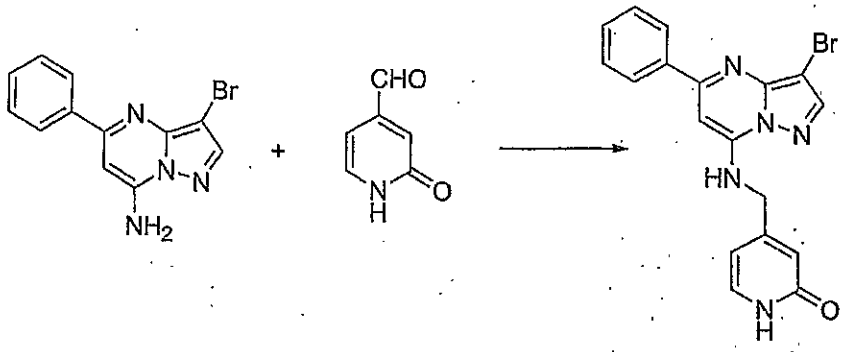
A mixture of the product prepared in Example 213 (80 mg, 0.19 mmol) and 2.0 M methylamine in THF was stirred in a closed pressure vessel at 50°C for 72 hr. The solvent was evaporated, and the residue was purified by flash chromatography using 10:1 CH_2Cl_2: MeOH as eluent to yield pale yellow solid (40 mg, 51%). LCMS: $M2H^+$=427. Mp=217-219 °C.

EXAMPLE 235:



By essentially the same procedure set forth in Example 234, the compound shown above was prepared. LCMS: $M2H^+$=441. Mp=98-101 °C.

EXAMPLE 236:



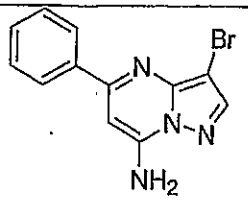
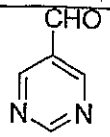
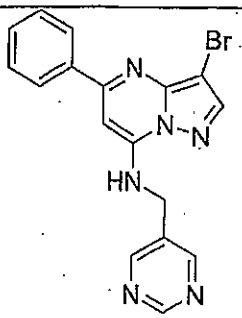
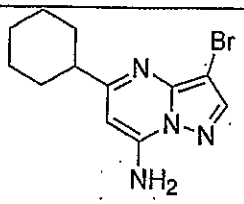
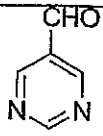
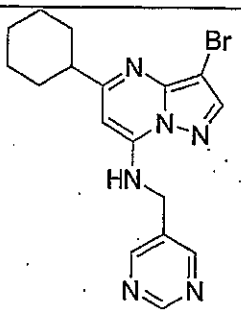
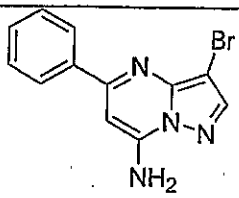
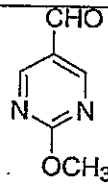
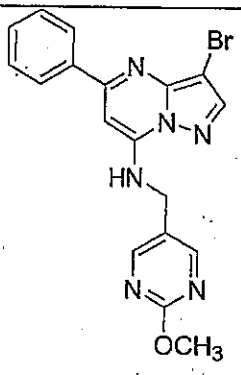
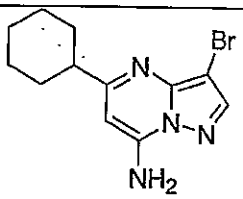
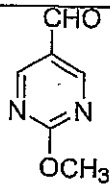
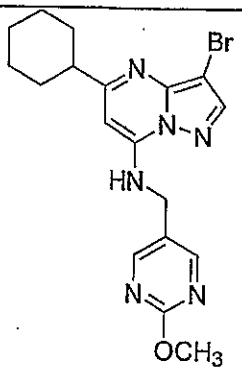
The compound prepared in Preparative Example 174 (140 mg, 0.48 mmol) and the aldehyde (71 mg, 0.58 mmol) were stirred in anhydrous THF (4 mL) at 50°C under N_2. $Ti(OiPr)_4$ (0.574 mL, 1.92 mmol) was added, the mixture was stirred at 50°C 3 hr, and cooled to 25°C. $NaBH_3CN$ (181 mg, 2.88 mmol) was added, the mixture was stirred for 2 more hr, then poured into 10 % aqueous Na_2CO_3 (100 mL), and extracted with CH_2Cl_2 (3 x 50 mL). Combined extracts were dried over Na_2SO_4, filtered, and the solvent was evaporated. The residue was purified by flash chromatography using 15:1 CH_2Cl_2:MeOH as eluent to yield pale yellow solid (40 mg, 21%). LCMS: MH^+=398. Mp>230 °C.

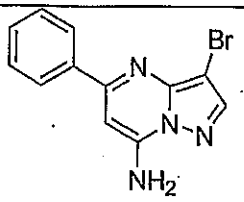
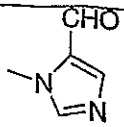
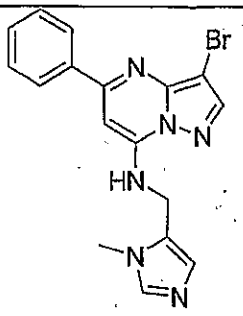
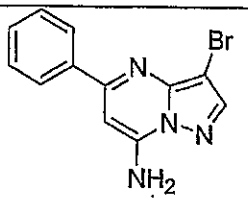
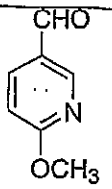
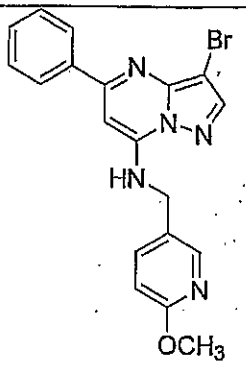
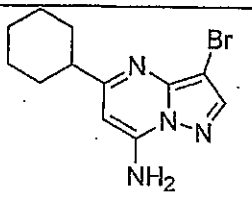
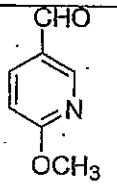
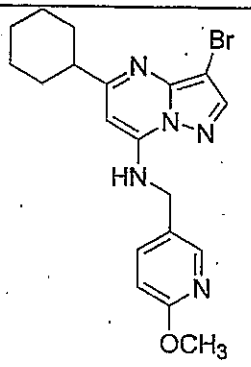
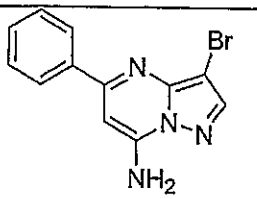
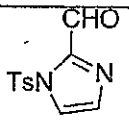
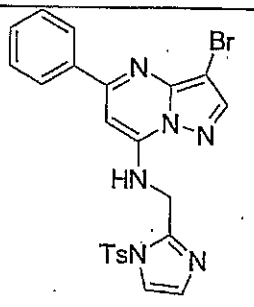
EXAMPLES 237-256:

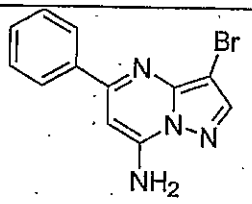
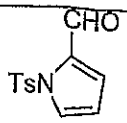
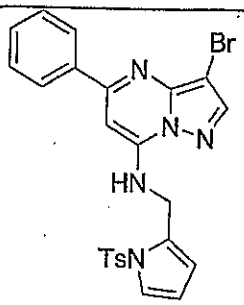
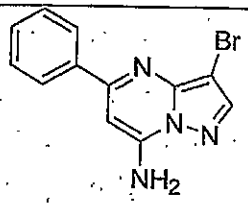
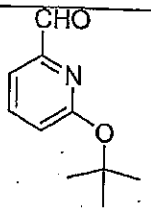
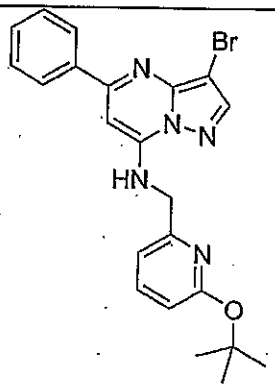
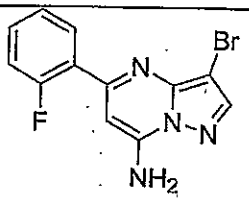
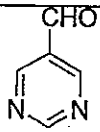
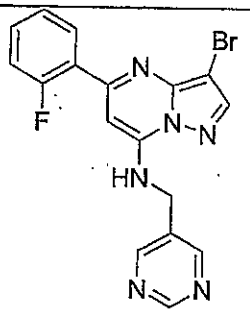
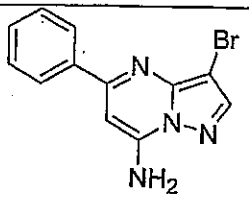
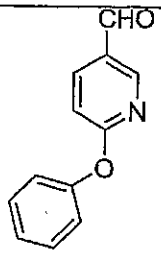
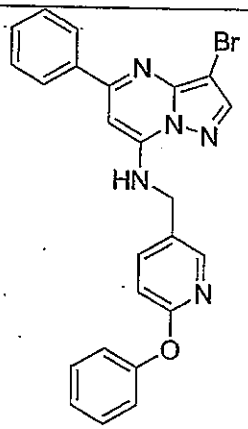
By essentially the same procedure set forth in Example 236 combining the compounds shown in Column 2 and 3 of Table 22, the compounds shown in Column 4 of Table 22 were prepared.

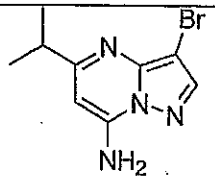
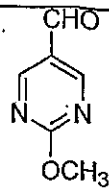
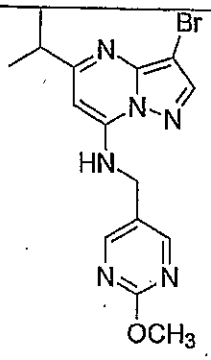
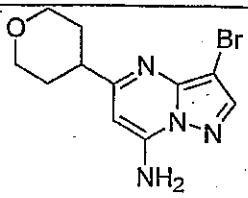
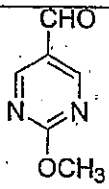
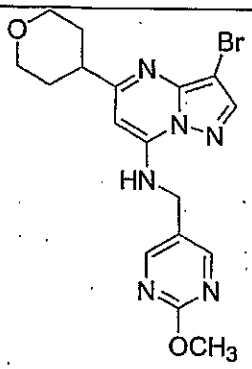
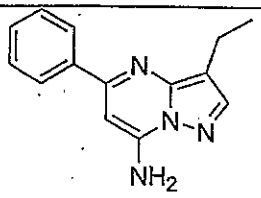
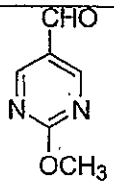
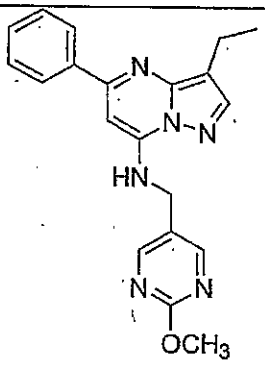
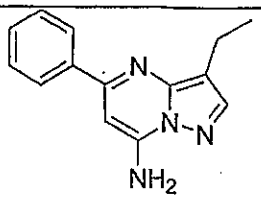
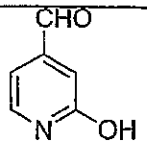
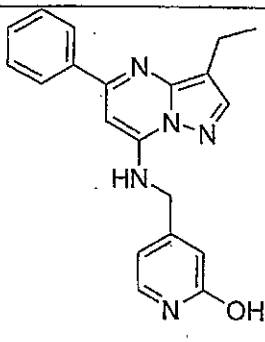
TABLE 22

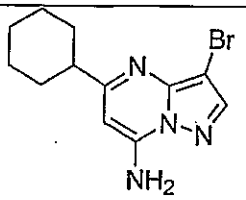
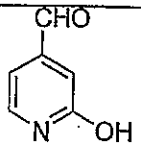
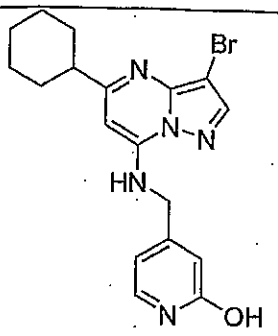
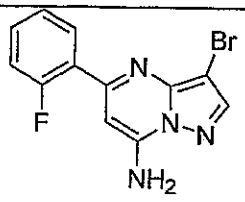
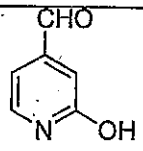
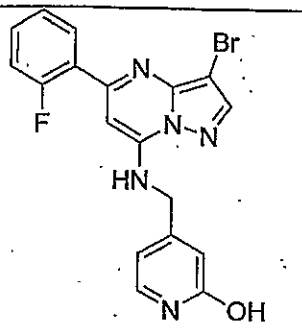
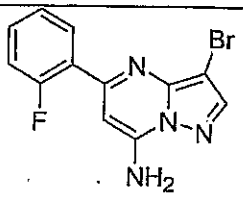
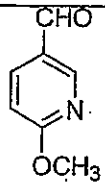
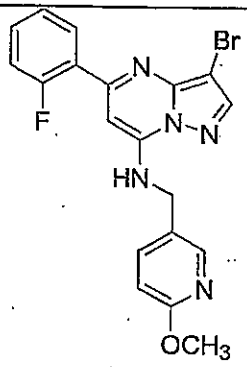
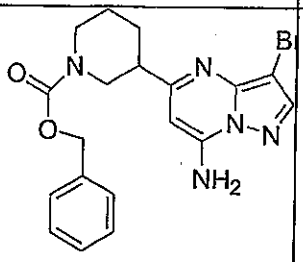
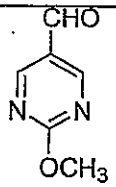
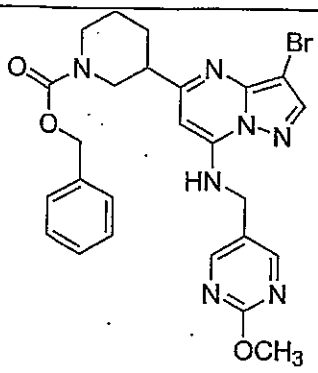
Ex.	Column 2	Column 3	Column 4	Data

237				LCMS: M^+=381; mp>200°C
238				LCMS: M^+=387; mp=°C
239				LCMS: MH^+=413; mp=157-159°C
240				LCMS: $M2H^+$=419 ; mp=77-79°C

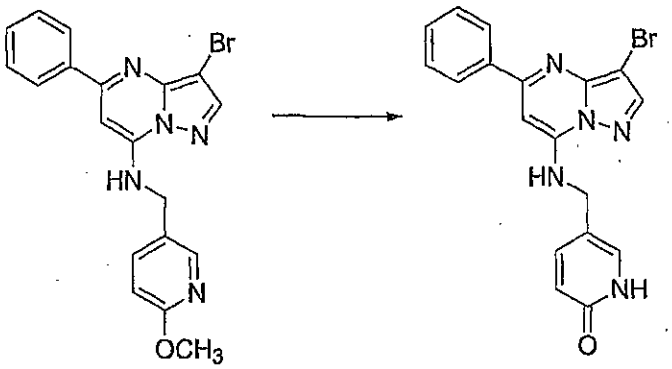
241				LCMS: $M2H^+$=385 ;.mp=214-216°C
242				LCMS: MH^+=; mp=°C
243				LCMS: M^+=416; mp=80-82°C
244				

245				
246				LCMS: M^+=452; mp=54-56°C
247				LCMS: MH^+= 401; mp>200°C
248				LCMS: $M2H^+$= 474; mp>200.0. °C dec.

249				LCMS: MH^+= 377; mp= 65-67°C
250				LCMS: $M2H^+$=421; mp=87-93°C
251				LCMS: MH^+=361; mp>225°C
252				LCMS: MH^+=346; mp=270-271°C

253				LCMS: MH^+=402; mp=250-255°C
254				LCMS: MH^+=416; mp=210-215°C
255				LCMS: MH^+=428; mp=145°C
256				LCMS: MH^+=; mp=°C

EXAMPLE 257:

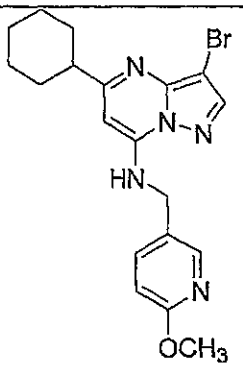
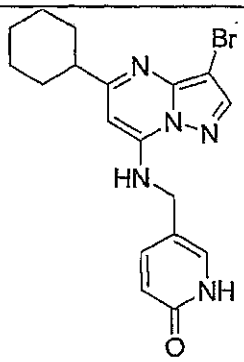


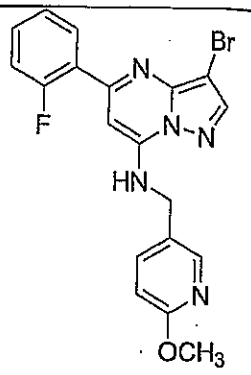
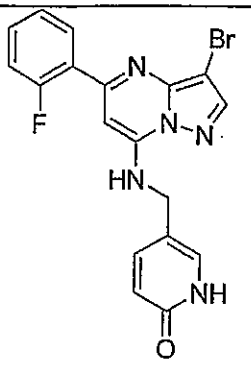
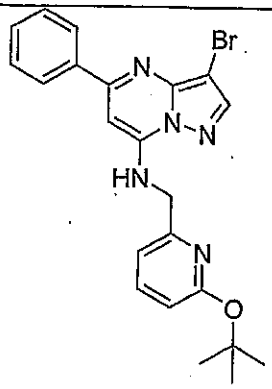
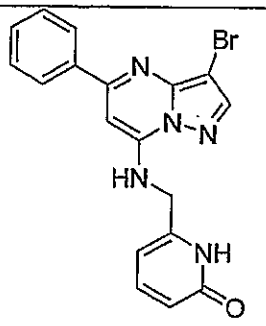
A mixture of the compound prepared in Example 242 (100 mg, 0.24 mmol), conc. aqueous HCl (1.0 mL) and acetic acid (2.0 mL) were stirred at 100°C under $N_2$for 2 hr, then poured onto Na_2CO_3 (15 g), and extracted with 1:1 acetone:CH_2Cl_2 (3 x 30 mL). Combined extracts were filtered, and the solvent was evaporated. The residue was purified by flash chromatography using 10:1 CH_2Cl_2:MeOH as eluent to yield pale yellow solid (36 mg, 37%). LCMS: $M2H^+$=398.

EXAMPLES 258-260:

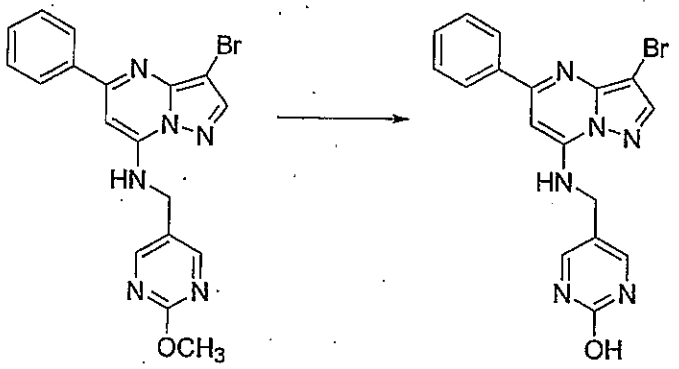
By essentially the same procedure set forth in Example 257 starting from the compounds shown in Column 1 of Table 23, the compounds shown in Column 2 of Table 23 were prepared.

TABLE 23

Ex.	Column 1	Column 2	Data
258			LCMS: M^+=402; mp=229-231°C

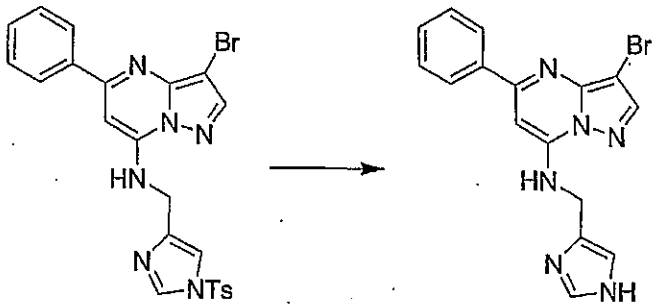
259			LCMS: MH^+=416; mp=215-218°C
260			LCMS: $M2H^+$=398 ; mp>230°C

EXAMPLE 261:



To a stirred solution of the compound prepared in Example 239 (41 mg, 0.10 mmol) in CH_2Cl_2 was added 1.0 M BBr_3 (0.30 mL, 0.30 mmol) in CH_2Cl_2 at -78°C. The mixture was stirred at -78°C for 5 min, then at 24°C for 3 hr, then MeOH (2.0 mL) was added and the mixture was stirred for 10 min. The solvent was evaporated and the residue was purified by flash chromatography using 5:1:0.1 CH_2Cl_2:MeOH:conc. NH_4OH as eluent to yield white solid (39 mg, 99%). LCMS: M^+=397. Mp>230 °C.

EXAMPLE 262:

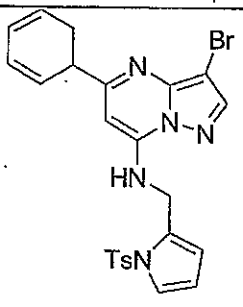
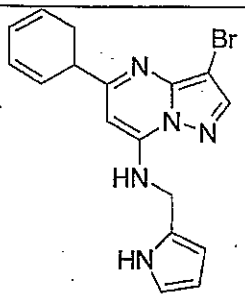


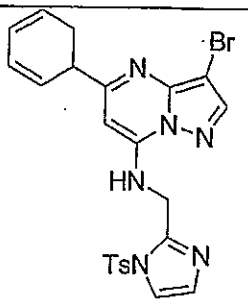
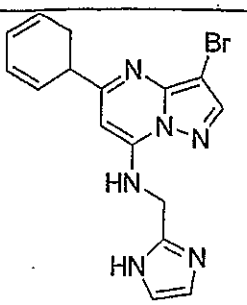
A mixture of the product prepared in Example 217 (40 mg, 0.077 mmol) and 5.0 M aqueous NaOH (0.8 mL) in MeOH (3.0 mL) was refluxed under N_2 for 1 hr. $NaHCO_3$ (700 mg) was added, the solvent evaporated, and the residue was purified by flash chromatography using 10:1:0.1 CH_2Cl_2: MeOH: conc. NH_4OH as eluent to yield white solid (10 mg, 35%). LCMS: $M2H^+$=371. Mp=237-239 °C.

EXAMPLES 263-264:

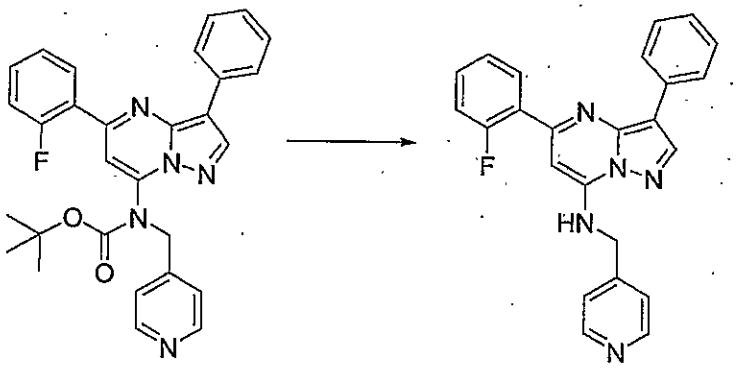
By essentially the same procedure set forth in Example 262 starting from the compounds shown in Column 1 of Table 24, the compounds shown in Column 2 of Table 24 were prepared.

TABLE 24

Ex.	Column 1	Column 2	Data
263			LCMS: $M2H^+$=370 ; mp=166-168°C

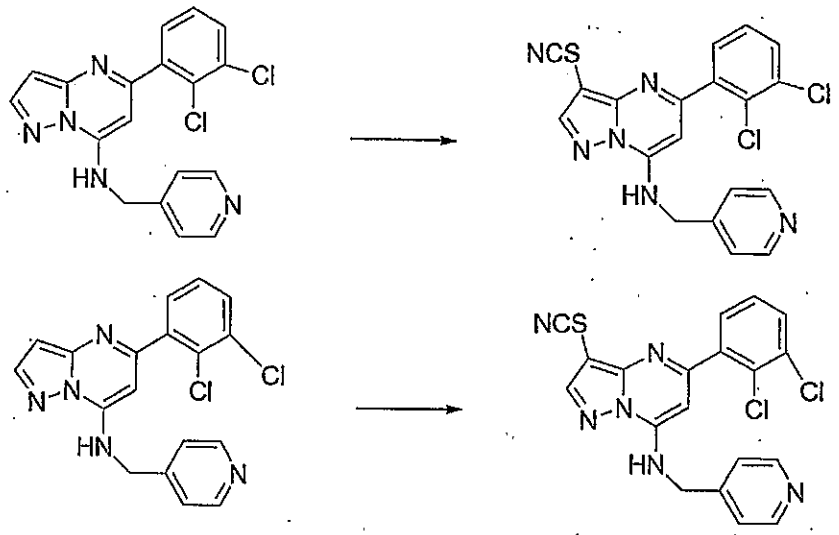
264			LCMS: $M2H^+$=371 ; mp=180-182°C

EXAMPLE 265:



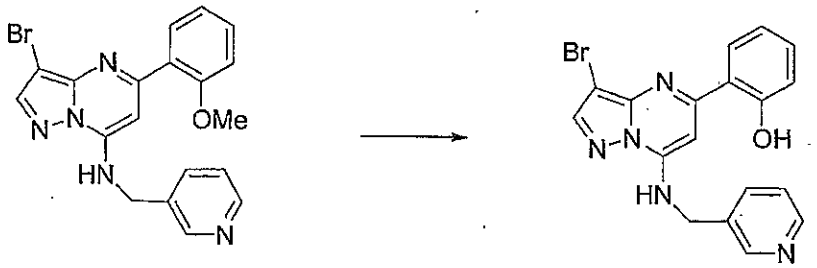
TFA (0.5 mL) was added to a solution of the compound prepared in Preparative Example 197 (0.08 g, 0.16 mmol) in CH_2Cl_2 (2.0 mL) at 0°C and the resulting solution stirred 2.5 hours and stored at 4°C overnight at which time additional TFA (0.5 mL) was added. The resulting solution was stirred 4 hours and concentrated *in vacuo*. The residue was neutralized with 1N NaOH and extracted with CH_2Cl_2. The combined organics were dried over Na_2SO_4, filtered, and concentrated under reduced pressure. The crude product was purified by flash chromatography using a 2.5% (10% NH_4OH in MeOH) in CH_2Cl_2 solution as eluent (0.009 g, 15% yield). LCMS: MH^+=396; mp= 53-54°C.

EXAMPLE 266:



A solution of the compound prepared in Preparative Example 182 (26 mg, 0.070 mmol) and potassium thiocyanate (13 mg, 0.14 mmol) in MeOH (1 mL) was cooled in a cold water bath. To it was added a solution of bromine (22 mg, 0.14mmol) in MeOH (0.7 mL) dropwise. The resulting reaction mixture was stirred for 4 h at room temperature and the volatiles were removed under reduced pressure. The residue obtained was suspended in a small amount of CH_2Cl_2. The potassium bromide was filtered off and pH of the filtrate was adjusted to about 7 by the addition of aqueous ammonia. It was concentrated under reduced pressure and the residual oil was purified by preparative thin-layer chromatography using 15% MeOH in CH_2Cl_2 as eluent (26 mg, 87% yield). ^1H NMR ($CDCl_3$) δ 8.75 (d, J = 4.2 Hz, 2H), 8.38 (s, 1H), 7.68-7.64 (m, 2H), 7.46-7.39 (m, 3H), 7.22 (t, J = 6.3 Hz, 1H), 6.43 (s, 1H), 4.84 (d, J = 6.3 Hz, 2H); LCMS: MH^+ = 427.

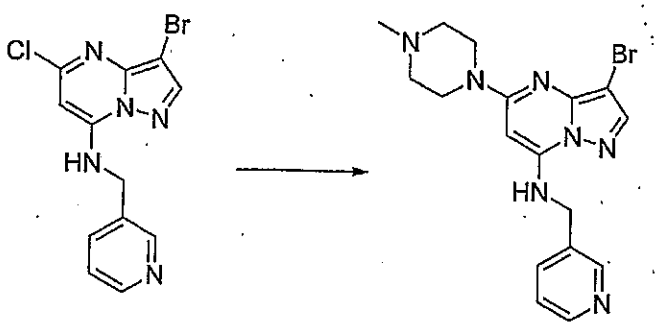
EXAMPLE 267:



Boron tribromide (1 M in CH_2Cl_2, 0.60 mL, 0.60 mmol) was added dropwise to an ice-cold stirred solution of the compound prepared in Example 24 (50 mg, 0.12 mmol) in CH_2Cl_2 (1.5 mL) under an argon atmosphere. The

resulting reaction mixture was stirred at 0°C for 30 minutes, allowed to warm up to room temperature, and stirred overnight. The mixture was quenched by the addition of a small amount of water and extracted with CH_2Cl_2. The organic layer was dried over magnesium sulfate and concentrated *in vacuo* (45 mg, 94% yield). ^1H NMR (CD_3OD) δ 9.16 (s, 1H), 8.95 (s, 1H), 8.88 (d, J = 8.1 Hz, 1H), 8.24 (t, J = 6.9 Hz, 1H), 8.18 (s, 1H), 7.95 (d, J = 7.8 Hz, 1H), 7.40 (t, J = 7.8 Hz, 1H), 7.00-6.96 (m, 2H), 6.86 (s, 1H), 5.28 (s, 2H); LCMS: MH^+ = 396.

EXAMPLE 268:



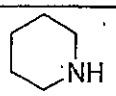
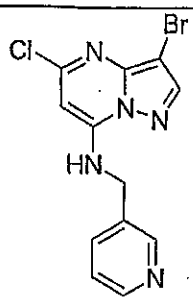
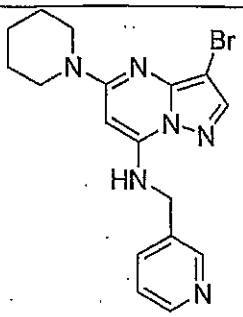
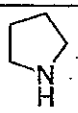
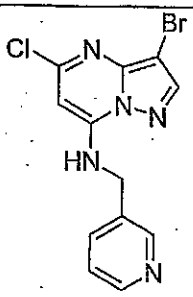
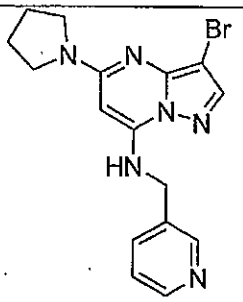
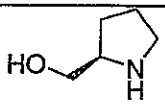
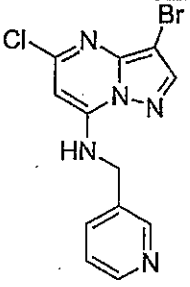
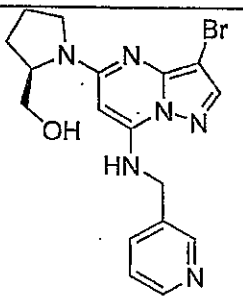
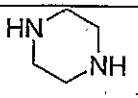
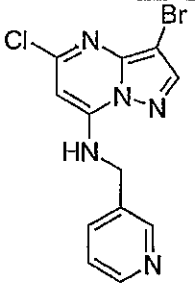
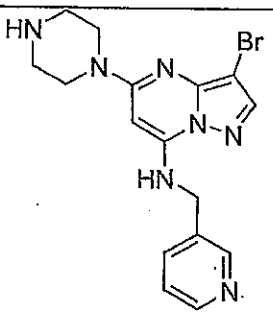
A solution of the compound from Preparative Example 184 (0.05 g, 0.15 mmol), N-methylpiperazine (20 μL, 1.2 eq.) and iPr_2Et (52 μL, 2.0 eq.) in dioxane (1 mL) was heated to 70 °C overnight. The reaction mixture was cooled to room temperature and diluted with H_2O and saturated $NaHCO_3$. The resulting mixture was extracted with CH_2Cl_2, the combined organics dried over Na_2SO_4, filtered, and concentrated under reduced pressure. The crude product was purified by Preparative TLC using a 5% (10% NH_4OH in MeOH) in CH_2Cl_2 solution as eluent (0.028 g, 47% yield). MS: MH^+= 402. mp = 210 °C (dec.)

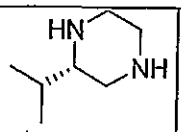
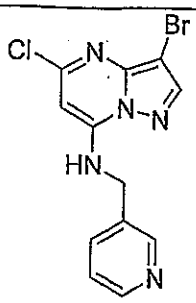
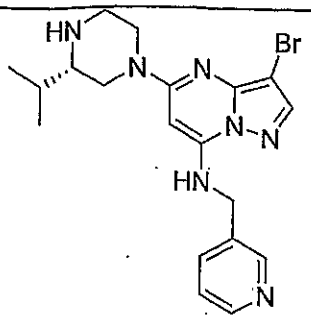
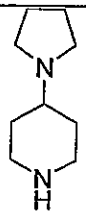
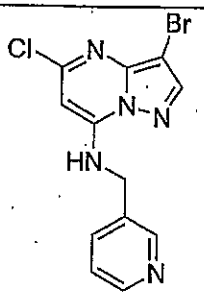
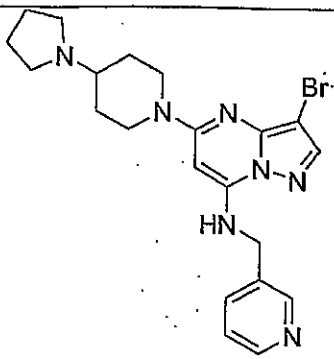
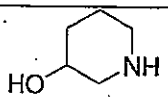
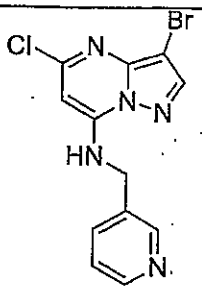
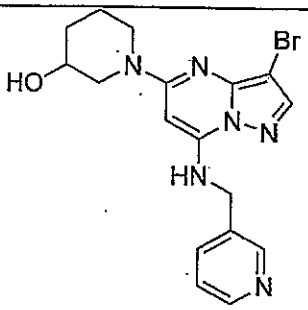
EXAMPLES 269-275:

By essentially the same procedure set forth in Example 268 only substituting the amine in Column 2 of Table 25 and the chlorides in Column 3 of Table 25, the compounds shown in Column 4 of Table 25 are prepared:

TABLE 25

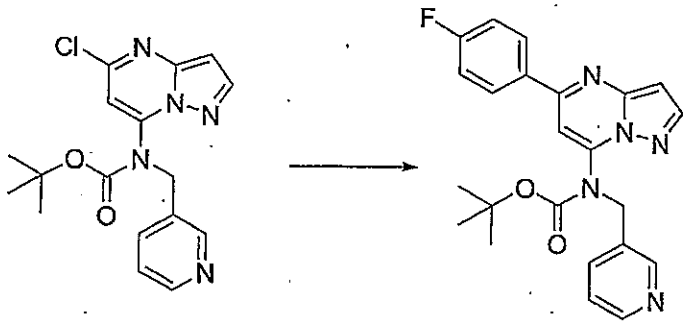
Ex.	Column 2	Column 3	Column 4	CMPD

269				MS: MH^+ = 387 m.p. 182 – 183°C
270				MS: MH^+ = 373 m.p. 190 – 191 °C
271				MS: MH^+ = 403 m.p. 227 – 230 °C
272				MS: MH^+ = 388 m.p. 198 – 201 °C

273				MS: MH^+ = 430 m.p. 100-103 °C
274				MS: MH^+ = 456 m.p. 175 – 178 °C
275				MS: MH^+ = 403 m.p. 218 °C

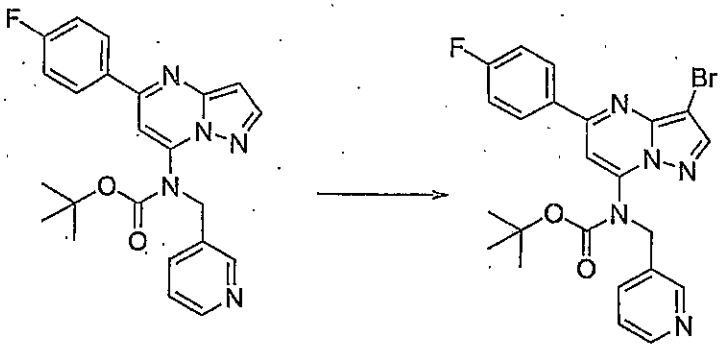
EXAMPLE 276:

Step A:



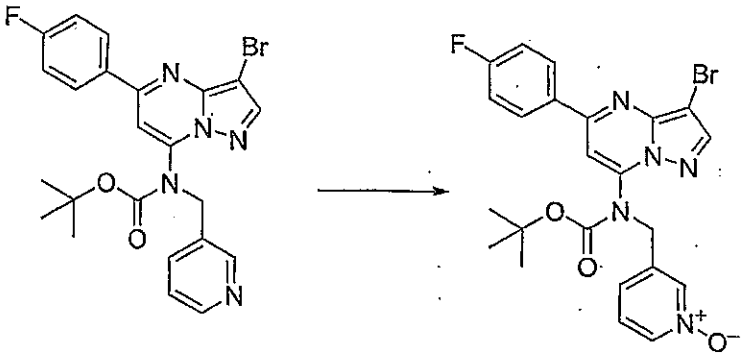
4-Fluorophenyl magnesium bromide (0.68 mL, 1.2 eq.) was added to the compound prepared in Preparative Example 193 (0.20 g, 0.55 mmol) and $PdCl_2(dppf)_2$ (0.037 g, 10 mol%) in THF and the resulting solution was stirred at room temperature 72 hours. The reaction mixture was dilute with saturated NH_4Cl and extracted with EtOAc. The combined organics were washed with saturated NaCl, dried over Na_2SO_4, filtered and concentrated. The crude product was purified by flash chromatography using neat EtOAc as eluent (0.15 g, 65% yield). MS: $MH^+ = 420$.

Step B:



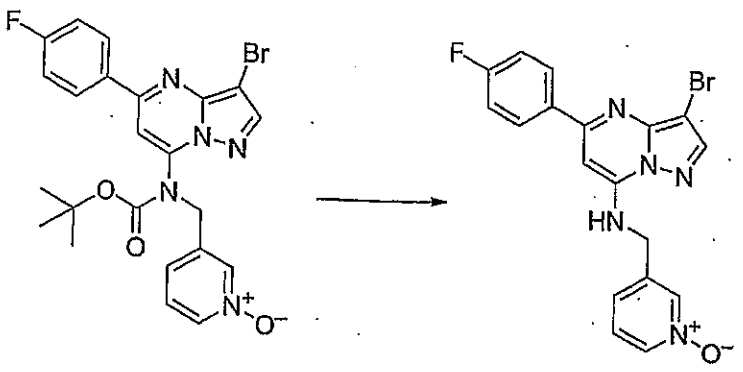
By essentially the same procedure set forth in Preparative Example 127 only substituting the compound prepared in Example 276, Step A, the above compound was prepared (0.17 g, 94% yield).

Step C:



By essentially the same procedure set forth in Preparative Example 200 only substituting the compound prepared in Example 276, Step B, the above compound was prepared (0.1g, 100% yield).

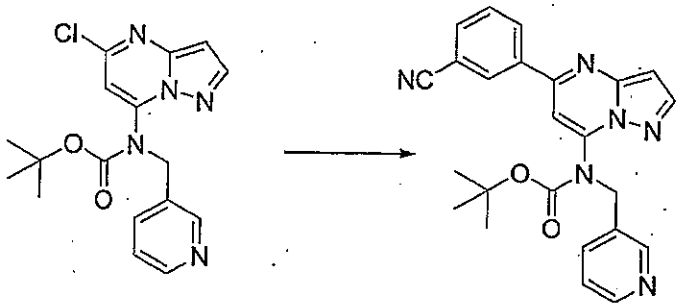
Step D:



By essentially the same procedure set forth in Example 265 only substituting the compound prepared in Example 276, Step C, the above compound was prepared (0.049 g, 62% yield). MS: MH^+ = 414; mp = 110-115 °C.

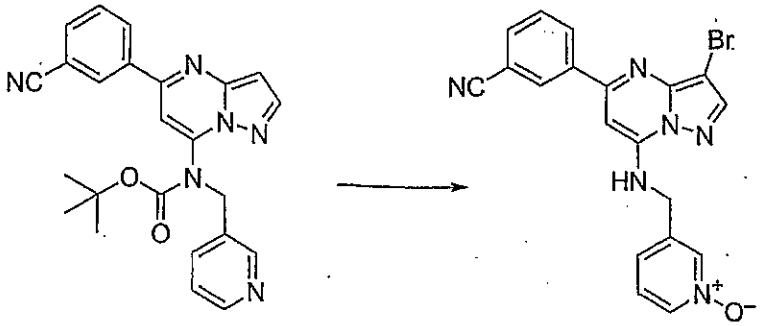
EXAMPLE 277:

Step A:



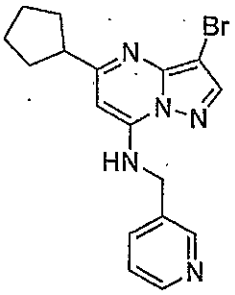
$Pd(PPh_3)_4$ (0.065 g, 10 mol%) was added to 3-cyanophenyl zinc iodide (2.2 mL, 0.5 M solution in THF, 2 eq.) and the compound prepared in Preparative Example 193 (0.2 g, 0.56 mmol) in DMF (2.0 mL) and the resulting solution heated to 80 °C g for 144 hours. The reaction mixture was cooled to room temperature, diluted with saturated NH_4Cl and extracted with EtOAc. The combined organics were washed with H_2O and brine, dried over Na_2SO_4, filtered, and concentrated under reduced pressure. The crude product was purified by flash chromatography using a neat EtOAC solution as eluent (0.07 g, 29% yield). MS: MH^+ = 427.

Step B through Step D:



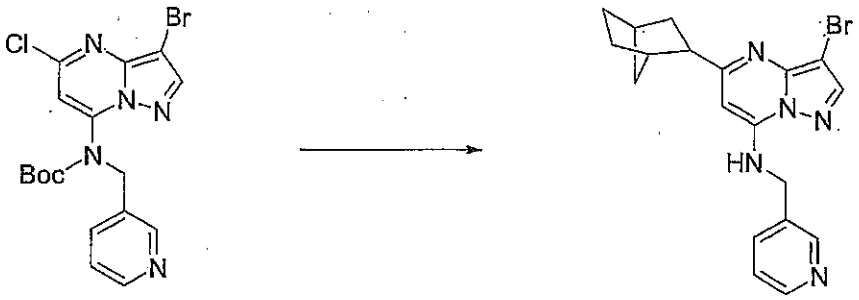
By essentially the same procedures set forth in Example 276, Step B through Step D, the above compound was prepared (0.023 g, 53% yield). MS: MH^+ = 421; mp = 230 °C (dec.)

EXAMPLE 278:



By essentially the same procedure set forth in Example 276 only substituting the appropriate cyclopropylmagnesium bromide in Step A, the compound was prepared. MS: MH^+ = 372; m. p. = 96-98 °C.

EXAMPLE 279:



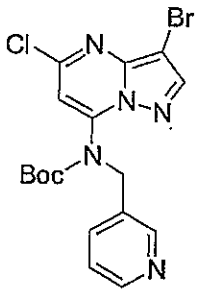
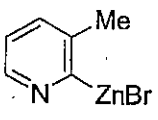
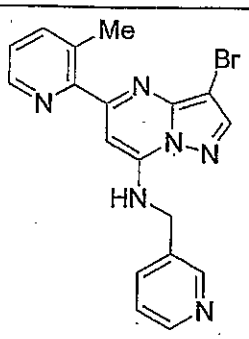
The palladium-catalyzed zinc cross-coupling reaction was carried out in a manner similar to the procedure described in *J. Org. Chem.* (1999), 453. A solution of the chloropyrazolopyrimidine (200 mg, 0.458 mmol), $Pd(PPh_3)_4$ (53

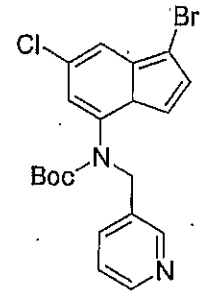
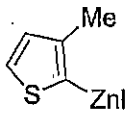
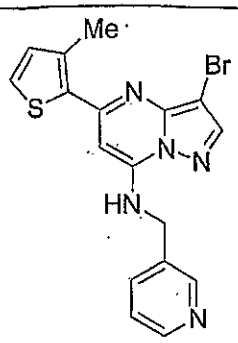
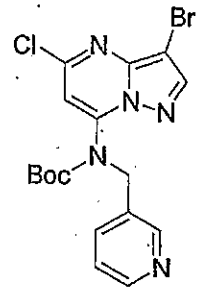
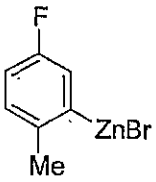
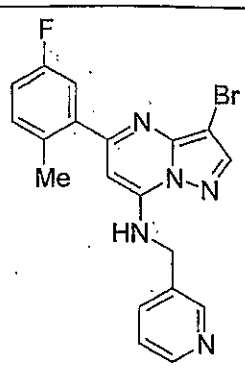
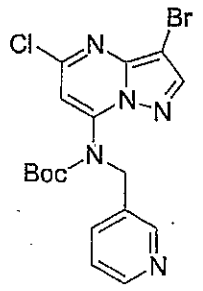
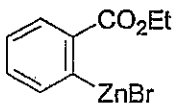
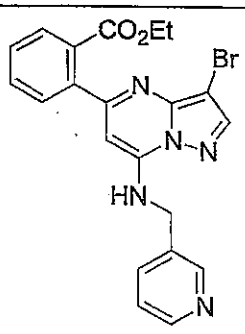
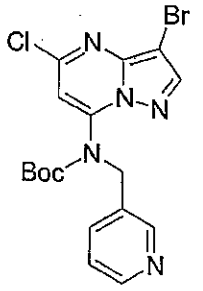
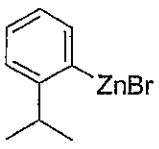
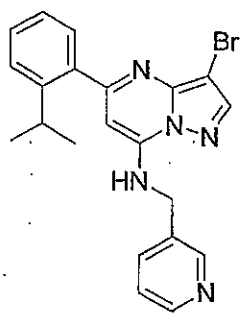
mg, 0.046 mmol), and *exo*-2-norbonylzinc bromide (0.5 M in THF, 0.95 mL, 0.47 mmol) in DMF (2 mL) was refluxed at 100°C (oil bath temp.) overnight. The reaction mixture was quenched with half-saturated NH_4Cl and extracted with CH_2Cl_2. The organic phase was dried over $MgSO_4$ and concentrated under reduced pressure. The residue was purified by flash chromatography using a 50% EtOAc in hexanes solution as eluent. A solution of the obtained *N*-Boc-protected product (121 mg, 53% yield, LCMS: MH^+ = 498) and TFA (1 mL) in CH_2Cl_2 (2 mL) was stirred at room temperature for 2 hr. The volatiles were removed under reduced pressure. The residue was dissolved in CH_2Cl_2, neutralized with saturated $NaHCO_3$, and extracted with CH_2Cl_2. The organic phase was dried over $MgSO_4$ and concentrated *in vacuo* (96 mg, 99% yield). LCMS: MH^+ = 398; ^1H NMR ($CDCl_3$) δ 8.78 (s, 1H), 8.71 (d, J = 4.2 Hz, 1H), 8.04 (d, J = 3.9 Hz, 1H), 7.80 (d, J = 7.8 Hz, 1H), 7.44 (m, 1H), 6.73 (m, 1H), 5.98 (d, J = 7.5 Hz, 1H), 4.74 (d, J = 5.4 Hz, 2H), 3.40-1.00 (m, 11H).

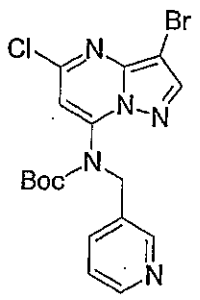
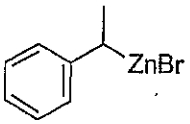
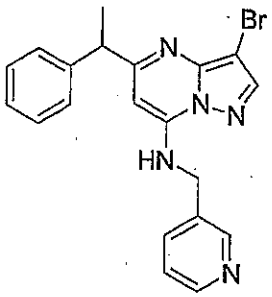
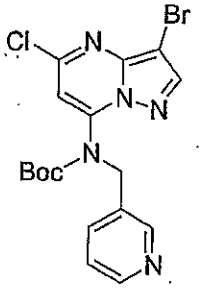
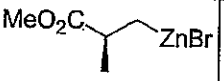
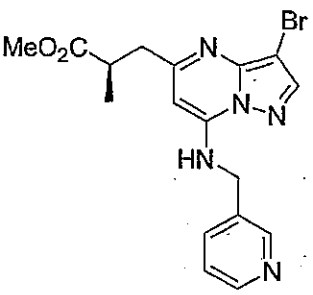
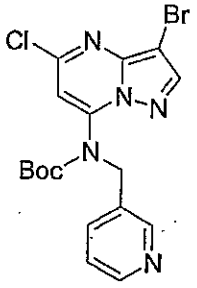
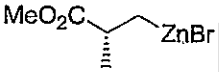
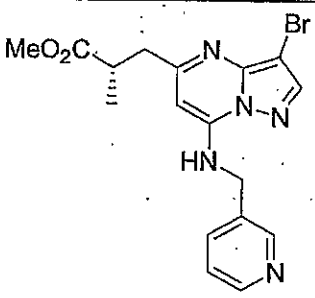
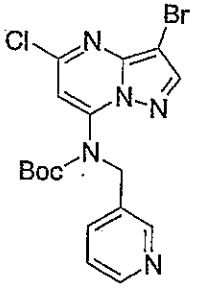
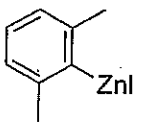
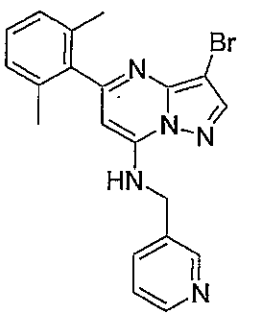
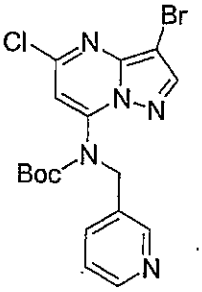
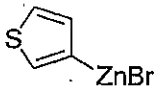
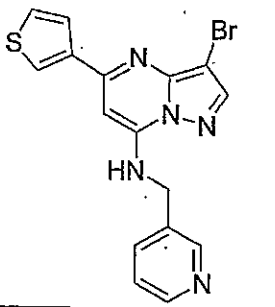
EXAMPLES 280-294:

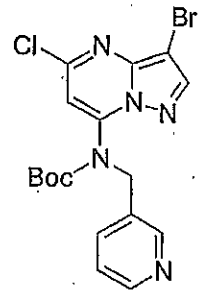
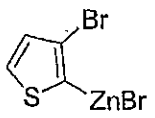
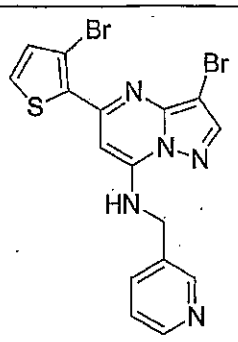
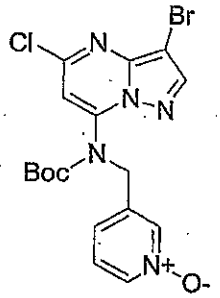
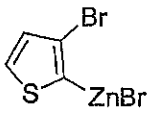
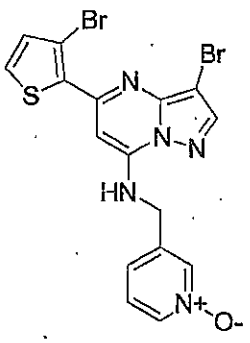
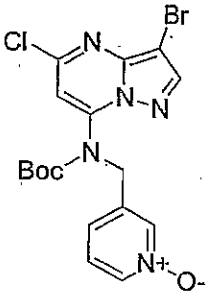
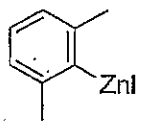
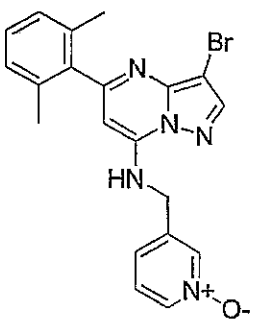
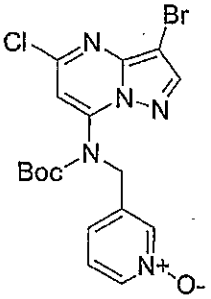
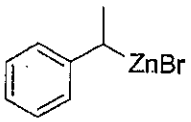
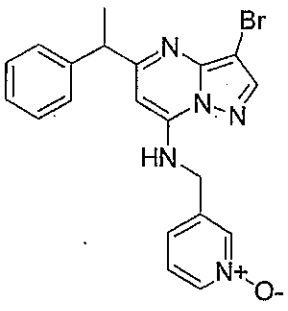
By following essentially the same procedure set forth in Example 279 only substituting the chlorides shown in Column 2 of Table 26 and the organozinc reagents shown in Column 3 of Table 26, the compounds in Column 4 of Table 26 were prepared:

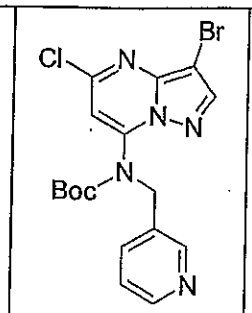
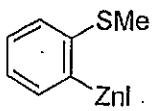
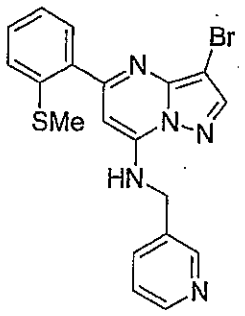
Table 26

Ex.	Column 2	Column 3	Column 4	Data
280				LCMS: MH^+ = 395

281				LCMS: MH^+ = 400
282				LCMS: MH^+ = 412
283				LCMS: MH^+ = 452
284				LCMS: MH^+ = 422

285				LCMS: MH^+ = 408
286				LCMS: MH^+ = 404
287				LCMS: MH^+ = 404
288				LCMS: MH^+ = 408
289				LCMS: MH^+ = 386

290				LCMS: MH^+ = 464
291				LCMS: MH^+ = 480
292				LCMS: MH^+ = 424
293				LCMS: MH^+ = 424

294				LCMS: MH^+ = 426

Additional data for select compounds is shown below.

EXAMPLE 280: ^{1}H NMR ($CDCl_3$) δ 8.65 (s, 1H), 8.57 (d, J = 4.2 Hz, 1H), 8.50 (d, J = 4.5 Hz, 1H), 8.01 (s, 1H), 7.69 (d, J = 7.5 Hz, 1H), 7.61 (d, J = 7.8 Hz, 1H), 7.31-7.22 (m, 2H), 6.77 (m, 2H), 4.71 (d, J = 5.4 Hz, 2H), 2.68 (s, 3H).

EXAMPLE 281: ^{1}H NMR ($CDCl_3$) δ 8.80 (s, 1H), 8.72 (d, J = 4.8 Hz, 1H), 8.08 (s, 1H), 7.85-7.40 (m, 3H), 7.02 (d, J = 5.1 Hz, 1H), 6.90 (t, J = 6.0 Hz, 1H), 6.29 (s, 1H), 4.79 (d, J = 6.0 Hz, 2H), 2.61 (s, 3H).

EXAMPLE 282: ^{1}H NMR ($CDCl_3$) δ 8.67 (s, 1H), 8.61 (d, J = 3.9 Hz, 1H), 8.03 (s, 1H), 7.72-7.31 (m, 3H), 7.22-7.00 (m, 2H), 6.81 (t, J = 6.0 Hz, 1H), 6.03 (s, 1H), 4.68 (d, J = 6.0 Hz, 2H), 2.28 (s, 3H).

EXAMPLE 283: ^{1}H NMR ($CDCl_3$) δ 8.68 (s, 1H), 8.63 (d, J = 4.0 Hz, 1H), 8.00 (s, 1H), 7.80-7.72 (m, 2H), 7.54-7.47 (m, 3H), 7.35 (m, 1H), 6.74 (t, J = 6.0 Hz, 1H), 6.19 (s, 1H), 4.67 (d, J = 6.0 Hz, 2H), 4.21 (q, J = 7.2 Hz, 2H), 1.13 (t, J = 7.2 Hz, 3H).

EXAMPLE 284: ^{1}H NMR ($CDCl_3$) δ 7.97 (s, 1H), 7.65 (d, J = 7.2 Hz, 1H), 7.33-7.15 (m, 5H), 6.73 (t, J = 5.4 Hz, 1H), 5.99 (s, 1H), 4.61 (d, J = 5.4 Hz, 2H), 3.09 (sept, J = 6.9 Hz, 1H), 1.11 (d, J = 6.9 Hz, 6H).

EXAMPLE 285: ^{1}H NMR ($CDCl_3$) δ 8.56-8.55 (m, 2H), 7.94 (s, 1H), 7.54 (m, 1H), 7.30-7.22 (m, 6H), 6.59 (t, J = 5.7 Hz, 1H), 5.66 (s, 1H), 4.47 (d, J = 5.7 Hz, 2H), 4.26 (q, J = 7.2 Hz, 1H), 1.68 (d, J = 7.2 Hz, 3H).

EXAMPLE 286: ^{1}H NMR ($CDCl_3$) δ 8.67 (m, 2H), 7.94 (s, 1H), 7.69 (d, J = 7.8 Hz, 1H), 7.34 (m, 1H), 6.63 (t, J = 5.7 Hz, 1H), 5.87 (s, 1H), 4.62 (d, J = 5.7 Hz, 2H), 3.64 (s, 3H), 3.13 (m, 2H), 2.82 (m, 1H), 1.22 (m, 3H).

EXAMPLE 287: ^{1}H NMR ($CDCl_3$) δ 8.66 (m, 2H), 7.94 (s, 1H), 7.68 (d, J = 7.8 Hz, 1H), 7.34 (m, 1H), 6.62 (t, J = 6.0 Hz, 1H), 5.87 (s, 1H), 4.62 (d, J = 6.0 Hz, 2H), 3.64 (s, 3H), 3.13 (m, 2H), 2.81 (m, 1H), 1.22 (m, 3H).

EXAMPLE 288: ^{1}H NMR ($CDCl_3$) δ 8.64 (s, 1H), 8.60 (d, J = 3.6 Hz, 1H), 8.04 (s, 1H), 7.68 (m, 1H), 7.31 (m, 1H), 7.16 (m, 1H), 7.07-7.05 (m, 2H), 6.80 (t, J = 6.3 Hz, 1H), 5.93 (s, 1H), 4.64 (d, J = 6.3 Hz, 2H), 2.08 (s, 6H).

EXAMPLE 289: ^{1}H NMR ($CDCl_3$) δ 8.72 (s, 1H), 8.62 (d, J = 4.8 Hz, 1H), 7.99-7.97 (m, 2H), 7.73-7.69 (m, 2H), 7.40-7.33 (m, 2H), 6.67 (t, J = 6.0 Hz, 1H), 6.29 (s, 1H), 4.71 (d, J = 6.0 Hz, 2H).

EXAMPLE 290: ^{1}H NMR ($CDCl_3$) δ 8.73 (s, 1H), 8.62 (d, J = 4.5 Hz, 1H), 8.01 (s, 1H), 7.76 (m, 1H), 7.41 (d, J = 5.1 Hz, 1H), 7.34 (dd, J = 8.1, 5.1 Hz, 1H), 7.05 (d, J = 5.1 Hz, 1H), 7.01 (s, 1H), 6.79 (t, J = 6.0 Hz, 1H), 4.74 (d, J = 6.0 Hz, 2H).

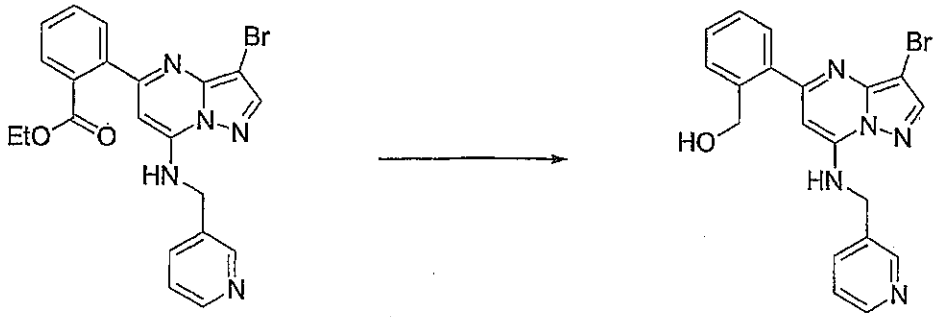
EXAMPLE 291: ^{1}H NMR (DMSO-d_6) δ 9.12 (s, 1H), 8.40 (s, 1H), 8.33 (s, 1H), 8.13 (m, 1H), 7.82 (d, J = 5.1 Hz, 1H), 7.40-7.39 (m, 2H), 7.22 (d, J = 5.1 Hz, 1H), 6.86 (s, 1H), 4.86 (s, 2H).

EXAMPLE 292: ^{1}H NMR ($CDCl_3$) δ 8.23 (s, 1H), 8.16 (d, J = 6.0 Hz, 1H), 8.06 (s, 1H), 7.31-7.05 (m, 5H), 6.86 (m, 1H), 5.87 (s, 1H), 4.62 (d, J = 6.3 Hz, 2H), 2.09 (s, 6H).

EXAMPLE 293: ^{1}H NMR ($CDCl_3$) δ 8.14 (s, 1H), 8.12 (d, J = 6.3 Hz, 1H), 7.94 (s, 1H), 7.29-7.16 (m, 6H), 7.07 (m, 1H), 6.78 (t, J = 6.0 Hz, 1H), 5.54 (s, 1H), 4.44 (d, J = 6.0 Hz, 2H), 4.24 (t, J = 7.2 Hz, 1H), 1.68 (d, J = 7.2 Hz, 3H).

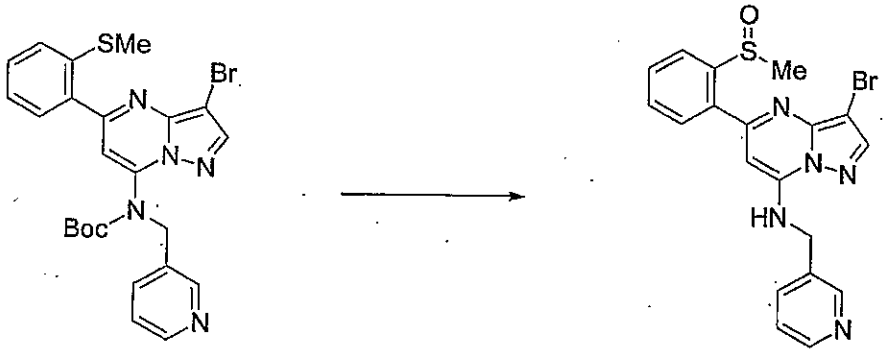
EXAMPLE 294: ^{1}H NMR ($CDCl_3$) δ 8.67 (s, 1H), 8.59 (d, J = 4.8 Hz, 1H), 8.01 (s, 1H), 7.71 (m, 1H), 7.52 (dd, J = 7.8, 1.8 Hz, 1H), 7.40-7.19 (m, 4H), 6.78 (t, J = 6.0 Hz, 1H), 6.32 (s, 1H), 4.67 (d, J = 6.0 Hz, 2H), 2.38 (s, 3H).

EXAMPLE 295:



To a suspension of lithium aluminum hydride (10 mg, 0.26 mmol) in anhydrous THF (2 mL) at 0°C was added dropwise a solution of the compound prepared in Example 283 (20 mg, 0.044 mmol) in anhydrous THF (2 mL). The resulting mixture was refluxed for 1 hr and stirred at room temperature overnight, neutralized with dilute sulfuric acid, and extracted with EtOAc. The organic phase was dried over $MgSO_4$ and concentrated under reduced pressure. The crude product was purified by preparative thin-layer chromatography using a 5% MeOH in EtOAc solution as eluent (15 mg, 83% yield). LCMS: $MH^+ = 410$; ^1H NMR ($CDCl_3$) δ 8.69 (s, 1H), 8.61 (d, J = 3.9 Hz, 1H), 8.05 (d, J = 2.1 Hz, 1H), 7.74 (d, J = 7.8 Hz, 1H), 7.52-7.31 (m, 5H), 6.97 (t, J = 6.3 Hz, 1H), 6.55 (d, J = 2.7 Hz, 1H), 6.20 (s, 1H), 4.71 (d, J = 6.3 Hz, 2H), 4.52 (s, 2H).

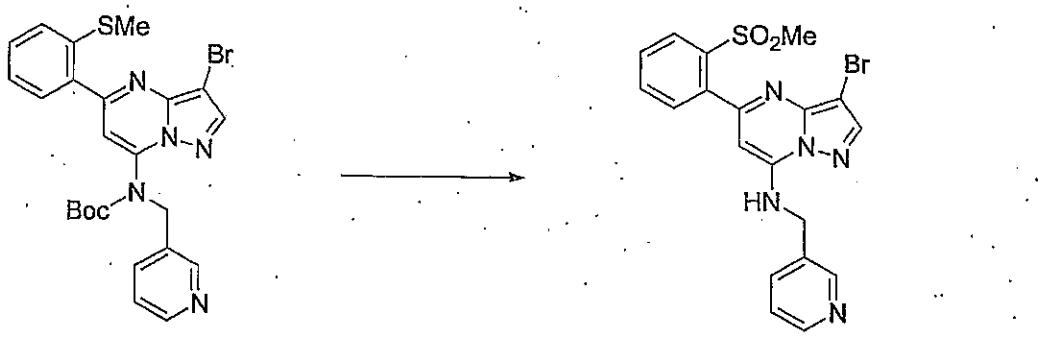
EXAMPLE 296:



To a solution of the *N*-Boc-protected compound prepared in Example 294 (45 mg, 0.085 mmol) in CH_2Cl_2 (4 mL) at –50°C was added *m*-CPBA (18 mg, 0.10 mmol). After stirring for I hr at –50°C more *m*-CPBA (4 mg, 0.02 mmol) was added. The mixture was stirred for a further 2 hr, diluted with CH_2Cl_2 (20 mL), and washed with saturated $NaHCO_3$ (20 mL). The organic phase was dried over $MgSO_4$ and concentrated under reduced pressure. The residue was purified by preparative thin-layer chromatography using a 2.5% MeOH in CH_2Cl_2 solution as eluent. A solution of the obtained *N*-Boc-protected product (37 mg, 80% yield, LCMS: $MH^+ = 542$) and TFA (1 mL) in CH_2Cl_2 (2 mL) was stirred at room temperature for 2 hr. The volatiles were removed under reduced pressure. The residue was dissolved in CH_2Cl_2, neutralized with saturated $NaHCO_3$, and

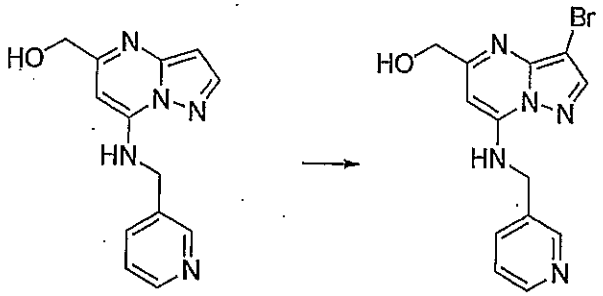
extracted with CH_2Cl_2. The organic phase was dried over $MgSO_4$ and concentrated under reduced pressure. The crude product was purified by preparative thin-layer chromatography using a 5% MeOH in EtOAc solution as eluent (26 mg, 89% yield). LCMS: $MH^+ = 442$; ^1H NMR ($CDCl_3$) δ 8.71 (s, 1H), 8.64 (d, J = 3.9 Hz, 1H), 8.41 (m, 1H), 8.03 (s, 1H), 7.75-7.54 (m, 4H), 7.36 (dd, J = 8.1, 5.1 Hz, 1H), 6.81 (t, J = 6.0 Hz, 1H), 6.34 (s, 1H), 4.74 (d, J = 6.0 Hz, 2H), 3.25 (s, 3H).

EXAMPLE 297:



To a solution of the *N*-Boc-protected compound prepared in Example 294 (56 mg, 0.11 mmol) in CH_2Cl_2 (4 mL) at 0°C was added *m*-CPBA (42 mg, 0.24 mmol). After stirring for 2 hr at room temperature more *m*-CPBA (13 mg, 0.075 mmol) was added. The mixture was stirred at room temperature overnight, diluted with CH_2Cl_2 (20 mL), and washed with saturated $NaHCO_3$ (20 mL). The organic phase was dried over $MgSO_4$ and concentrated under reduced pressure. The residue was purified by preparative thin-layer chromatography using a 2.5% MeOH in EtOAc solution as eluent. A solution of the obtained *N*-Boc-protected product (29 mg, 49% yield, LCMS: $MH^+ = 558$) and TFA (1 mL) in CH_2Cl_2 (2 mL) was stirred at room temperature for 2 hr. The volatiles were removed under reduced pressure. The residue was dissolved in CH_2Cl_2, neutralized with saturated $NaHCO_3$, and extracted with CH_2Cl_2. The organic phase was dried over $MgSO_4$ and concentrated under reduced pressure. The crude product was purified by preparative thin-layer chromatography using a 2.5% MeOH in EtOAc solution as eluent (21 mg, 90% yield). LCMS: $MH^+ = 458$; ^1H NMR ($CDCl_3$) δ 8.64 (s, 2H), 8.20 (m, 1H), 8.01 (s, 1H), 7.73-7.60 (m, 3H), 7.46 (m, 1H), 7.35 (s, 1H), 6.82 (t, J = 5.9 Hz, 1H), 6.17 (s, 1H), 4.65 (d, J = 5.7 Hz, 2H), 3.60 (s, 3H).

EXAMPLE 298

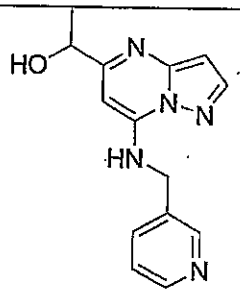
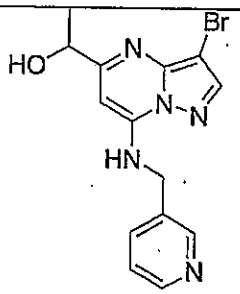
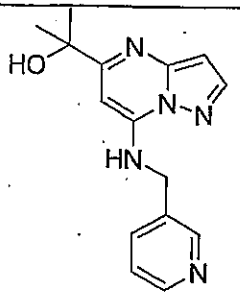
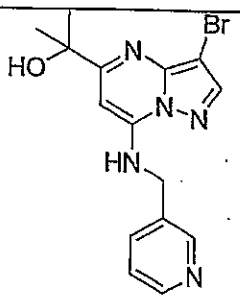


By essentially the same procedure set forth in Preparative Example 127 only substituting the compound prepared in Preparative Example 189, the above compound was prepared. MS: MH^+ = 334; mp = 170-173 °C.

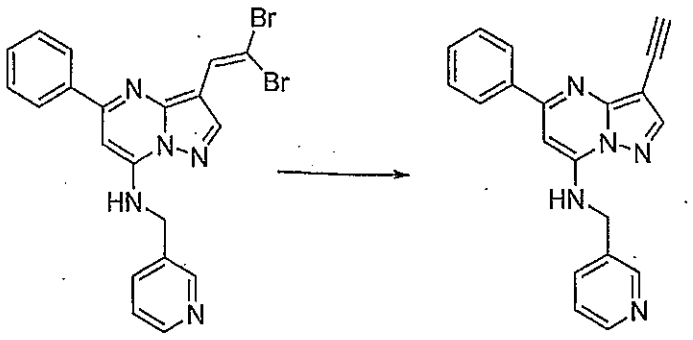
Examples 299-300:

By essentially the same procedure set forth in Example 298 only substituting the compound shown in Table 27, Column 2, the compounds shown in Table 27, Column 3 were prepared:

Table 27

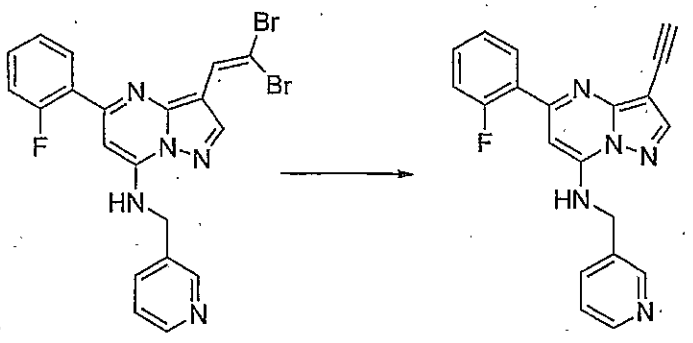
Ex.	Column 2	Column 3	CMPD
299			MS: MH^+ = 348 m.p. = 73 – 83 °C
300			MS: MH^+ = 362 m.p. = 165 – 175 °C

EXAMPLE 301:



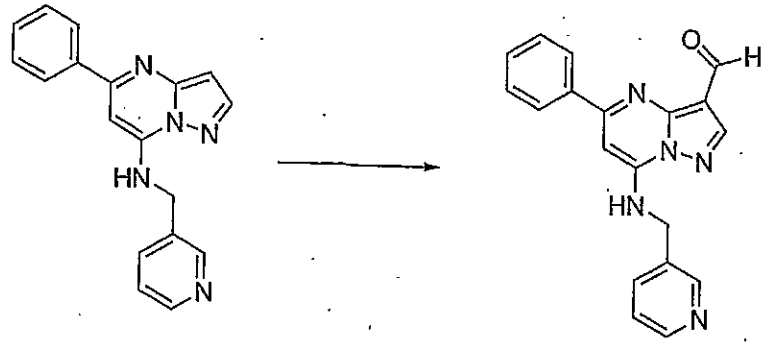
To a solution of the compound prepared in Preparative Example 186 (0.1 g, 0.21 mmol) in THF (4.0 mL) at -78 °C was added nBuLi (0.57 mL, 2.16M in hexanes, 5.0 eq.) at -78 °C. The reaction mixture was stirred 2 hours at -78 °C, quenched with H_2O, warmed to room temperature, and extracted with EtOAc. The combined organics were dried over Na_2SO_4, filtered, and concentrated under reduced pressure. The crude product was purified by Preparative TLC using a 2.5% (10%NH_4OH in CH_3OH) solution in CH_2Cl_2 as eluent (0.013 g, 20% yield). MS: MH^+ = 326; mp = 71-72 °C.

EXAMPLE 302:



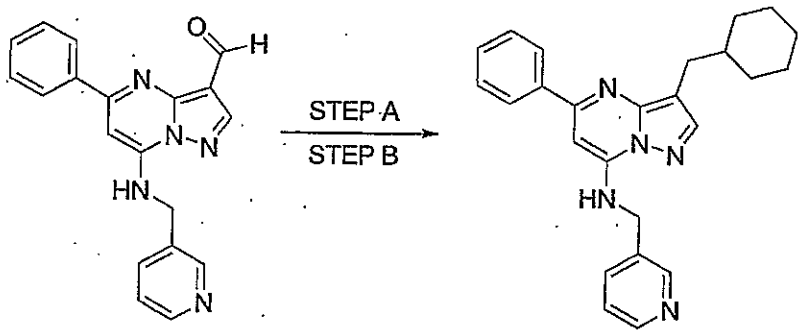
By essentially the same procedure set forth in Example 301 only substituting the compound from Preparative Example 187, the above compound was prepared (0.049 g, 68% yield).MS: MH^+ = 344; mp = 69-71 °C.

EXAMPLE 303:



To a solution of 3-H adduct from Preparative Example 187.1 (0.70 g, 2.32 mmol) in DMF (4.2 mL) at 0 °C was added $POCl_3$ (0.67 mL, 7.2 mmol) dropwise. The mixture was stirred for 14h at rt, cooled to 0 °C, and was quenched by addition of ice. 1N NaOH was carefully added to adjust pH to 8 and the mixture was extracted with CH_2Cl_2 (3 x 25 mL). The organic layers were combined, dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was recrystallized from EtOAc to afford 0.43 g (56%) of a yellow solid. mp 181-183 °C; M+H = 330.

EXAMPLE 304:



STEP A:

To a solution of aldehyde (100 mg, 0.30 mmol) from Example 303 in THF (1 mL) at 0 °C was added cyclohexyl magnesium bromide (0.46 mL, 2.0M in Et_2O) dropwise over 5 min. The resulting mixture was stirred at 0 °C for 2h and at rt for 12h. The mixture was cooled to 0 °C and was treated with sat. aq. NH_4Cl (3 mL) and CH_2Cl_2 (5 mL). The layers were separated and the aqueous layer was extracted with CH_2Cl_2 (2 x 5 mL). The organic layers were combined, washed with brine (1 x 5 mL), dried (Na_2SO4), filtered and concentrated under

reduced pressure to afford 110 mg (89%) of a light yellow semisolid. M+H = 414. This material was carried on crude to Step B without further purification.

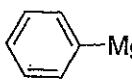
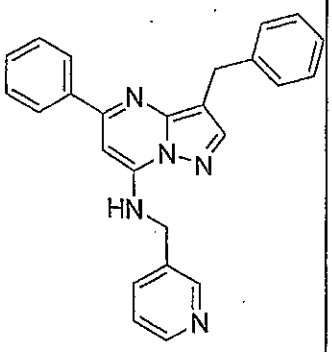
STEP B:

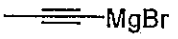
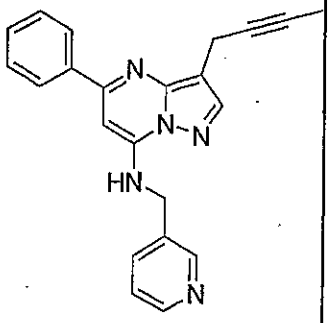
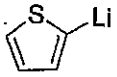
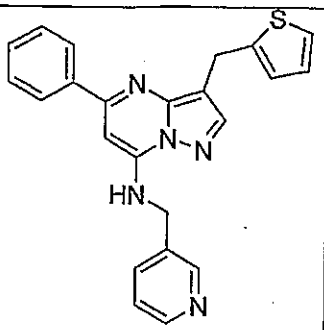
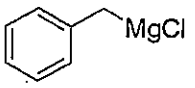
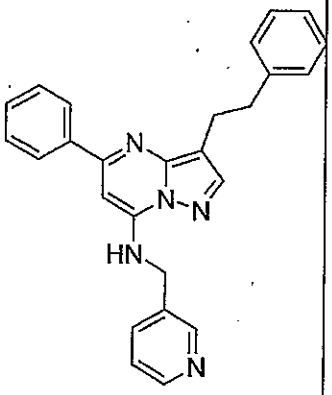
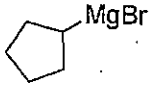
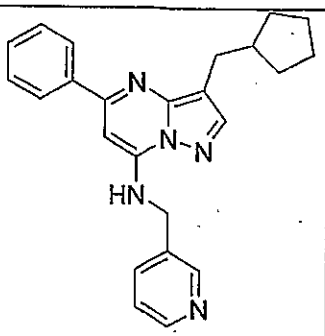
To a solution of alcohol (53 mg, 0.13 mmol) in CH_2Cl_2 (0.5 mL) at 0 °C was added Et_3SiH (24 μL, 0.15 mmol) followed by TFA (24 μL, 0.30 mmol). The mixture was stirred for 2 h at 0 °C and rt for 2 h whereupon additional portions of Et_3SiH (24 μL, 0.15 mmol) and TFA (24 μL, 0.30 mmol) were added and the mixture was stirred for 3 h at rt (until complete by TLC). The mixture was concentrated under reduced pressure and the crude residue was partitioned between CH_2Cl_2 (5 mL) and sat. aq. $NaHCO_3$ (2.5 mL). The layers were separated and the aqueous layer was extracted with CH_2Cl_2 (2 x 5 mL). The organic layers were combined, washed with brine (1 x 5 mL), dried (Na_2SO4), filtered and concentrated under reduced pressure. The crude product was purified by prep TLC (8 x 1000 mM) eluting with CH_2Cl_2/MeOH (22:1) to afford 29 mg (56%) of a yellow semisolid. M+H = 398.

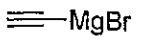
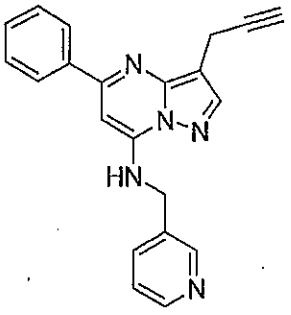
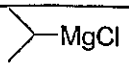
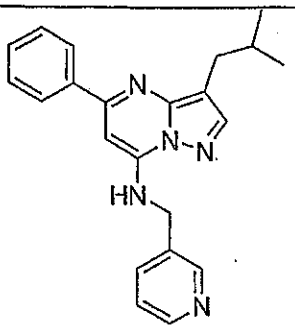
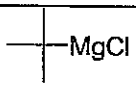
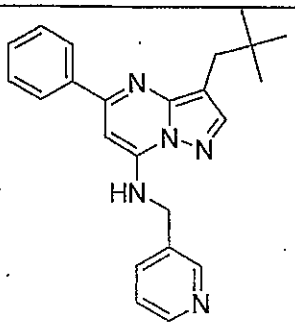
EXAMPLES 305-312:

By essentially the same procedure set forth in Example 304, utilizing the aldehyde from Example 303 and substituting the Grignard or organolithium reagents shown in Column 2 of Table 28, the compounds in Column 3 of Table 28 were prepared:

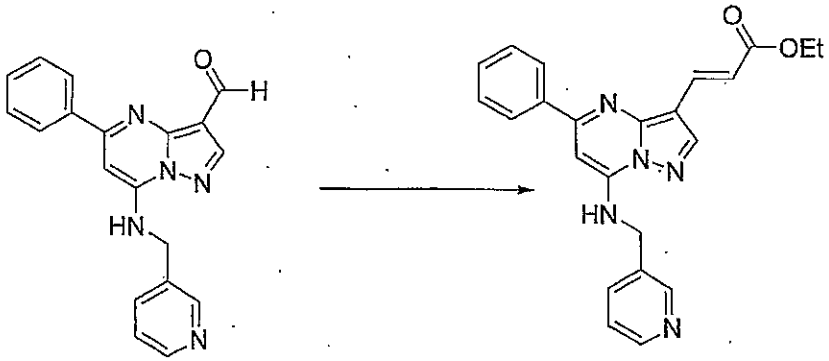
TABLE 28

Ex.	Column 2 (Organometallic)	Column 3 (Final Structure)	CMPD 1. mp (°C) 2. M+H
305			1. yellow oil 2. M+H = 392

306			1. red oil 2. M+H = 353
307			1. red oil 2. M+H = 398
308			1. yellow oil 2. M+H = 406
309			1. yellow semisolid 2. M+H = 384

310			1. semisolid 2. M+H = 340
311			1. mp = 141-143 2. M+H = 358
312			1. mp = 148-150 2. M+H = 372

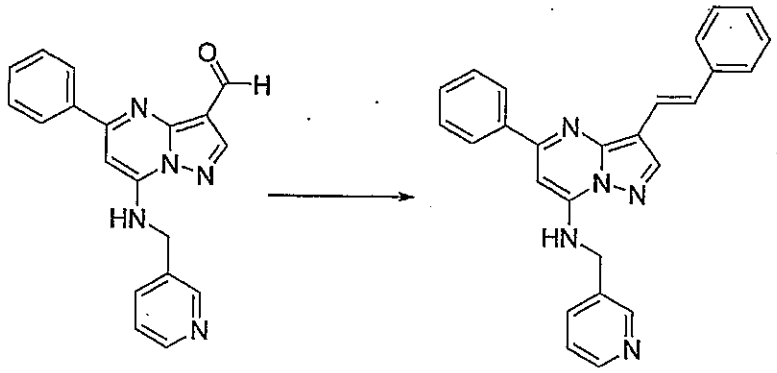
EXAMPLE 313:



To solution of aldehyde (81 mg, 0.25 mmol) from Example 303 in benzene (2.5 mL) was added carboethoxymethylene triphenyl phosphorane

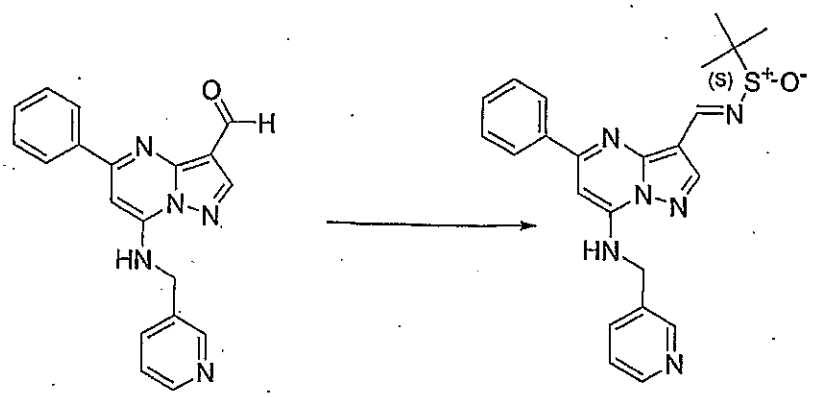
(0.12 g, 0.33 mmol) in one portion. The mixture was heated at reflux for 24h, cooled to rt, and concentrated under reduced pressure. The mixture was diluted CH_2Cl_2 (5 mL), brine (2 mL) was added, and the layers were separated. The aqueous layer was extracted with CH_2Cl_2 (2 x 4 mL). The organic layers were combined, dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by preparative TLC (8 x 1000 μM) eluting with CH_2Cl_2/MeOH (20:1) to afford 98 mg (100%) of white solid. mp 151-153 °C; M+H = 400.

EXAMPLE 314:



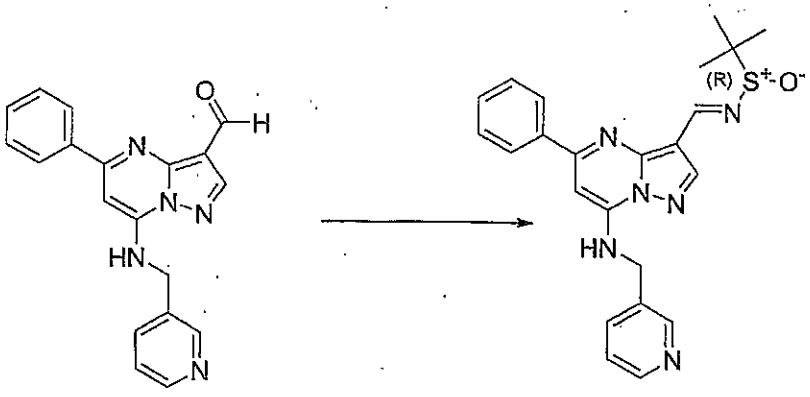
To a mixture of benzyltriphenylphosphonium bromide (0.59 g, 1.37 mmol) in THF (3 mL) was added NaH (55 mg, 1.37 mmol) and the mixture was stirred for 30 min. The aldehyde (0.15 g, 0.46 mmol) from Example 303 was added in a single portion and the mixture was heated at reflux for 36h. The mixture was cooled to rt and was concentrated under reduced pressure. The mixture was diluted CH_2Cl_2 (5 mL), brine (2 mL) was added, and the layers were separated. The aqueous layer was extracted with CH_2Cl_2 (2 x 4 mL). The organic layers were combined, dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by preparative TLC (8 x 1000 μM) eluting with CH_2Cl_2/MeOH (20:1) to afford 58 mg (32%) of yellow solid. mp 138-141 °C; M+H = 404.

EXAMPLE 315:



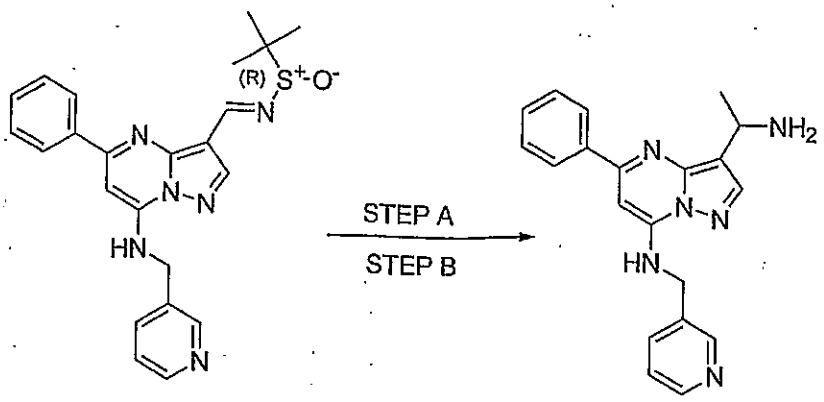
To a solution of aldehyde (0.20 g, 0.60 mmol) from Example 303 in THF (3 mL) was added Ti (i-OPr)$_4$ (0.36 mL, 1.21 mmol) dropwise followed by addition of (*S*)-(-)-2-methyl-2-propanesulfinamide (74 mg, 0.61 mmol). The resulting mixture was stirred for 18h at reflux, cooled to rt, and quenched with brine (2 mL). The mixture was filtered thru a pad of Celite which was washed with EtOAc (2 x 2 mL). The layers were separated and the aqueous layer was extracted with EtOAc (2 x 4 mL). The organic layers were combined, dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by preparative TLC (8 x 1000 μM) eluting with CH_2Cl_2/MeOH (20:1) to afford 0.21 g (80%) of yellow solid. mp 108-110 °C; M+H = 433.

EXAMPLE 316:



Prepared in the same fashion as Example 315 except substituting (*R*)-(-)-2-methyl-2-propanesulfinamide to afford 0.25 g (94%) as a yellow solid. mp 107-109 °C; M+H = 433.

EXAMPLE 317:



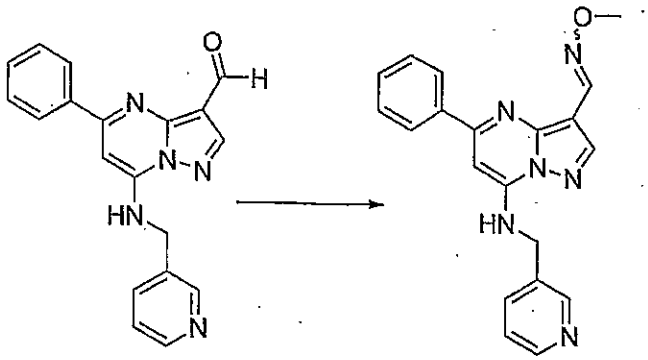
STEP A:

To a solution of sulfinimine (50 mg, 0.12 mmol) from Example 316 in CH_2Cl_2 (2.5 mL) at -40 °C was added MeMgBr (96 mL, 0.29 mmol) dropwise. The mixture was stirred for 5h at -40 °C and was stirred at rt for 12h. An additional portion of MeMgBr (96 mL, 0.29 mmol) and the mixture was stirred for 12 h. Sat. aq. NH_4Cl (2 mL) was added and the mixture was extracted with EtOAc (3 x 4 mL). The organic layers were combined, dried (Na_2SO_4), filtered, and concentrated under reduced pressure to afford 30 mg (58%) of crude residue. This material was taken onto the next step without purification.

STEP B:

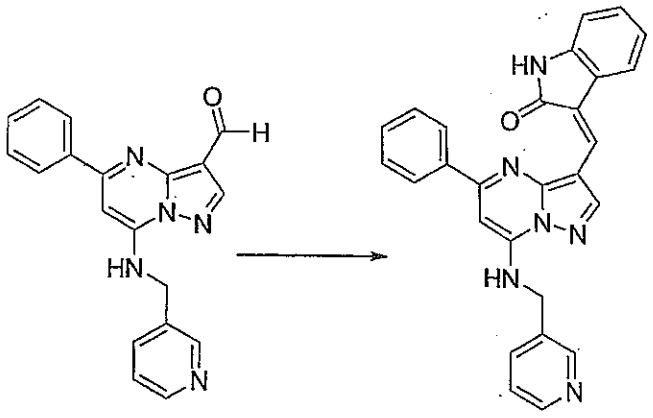
The crude material from Step A (30 mg, 0.067 mmol) in MeOH (2 mL) was added conc. HCl (2 mL). The mixture was stirred at rt for 12h and the mixture was concentrated to dryness. The crude material was partitioned between CH_2Cl_2 (3 mL) and sat. aq. $NaHCO_3$ (2 mL) and the layers were separated. The aqueous layer was extracted with CH_2Cl_2 (2 x 3 mL) and the organic layers were combined. The organic layer was dried (Na_2SO_4), filtered, and concentrated under reduced pressure to afford 6 mg (24%) of the title compound as a light yellow solid. mp 100-102 °C; M + H = 345.

EXAMPLE 318:



To a solution of aldehyde (75 mg, 0.23 mmol) from Example 300 in THF/CH_2Cl_2 (5 mL/1mL) at rt was added $MeONH_2$·HCl (38 mg, 0.46 mmol) followed by dropwise addition of pyridine (46 μL, 0.57 mmol). The mixture was stirred for 72h at rt whereupon the mixture was concentrated to dryness. The crude material was partitioned between CH_2Cl_2 (3 mL) and sat. aq. $NaHCO_3$ (2 mL) and the layers were separated. The aqueous layer was extracted with CH_2Cl_2 (2 x 3 mL) and the organic layers were combined. The organic layer was dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by preparative TLC (3 x 1000 μM) eluting with CH_2Cl_2/MeOH (22:1) to afford 90 mg (100%) of light yellow solid. mp 173-175 °C ; M + H = 359.

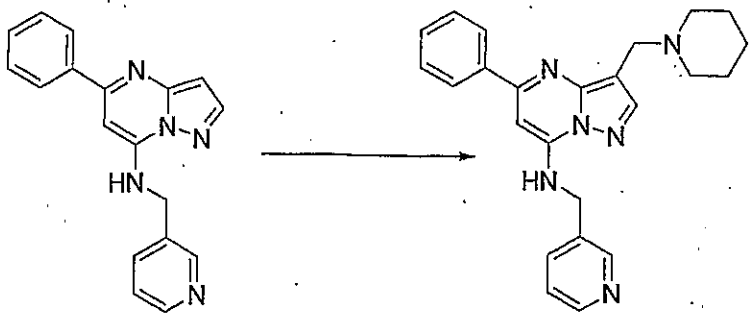
EXAMPLE 319:



To solution of aldehyde (60 mg, 0.18 mmol) from Example 303 at EtOH (2.5 mL) was added oxindole (48 mg, 0.37 mmol) followed by piperidine (3 drops). The mixture was heated at reflux for 14h and the mixture was cooled to

rt. The resultant precipitate was filtered and washed with cold EtOH (2 x 2 mL). The product was dried under high vacuum to afford 81 mg (100%) of the title compound as an orange/brown solid. mp 182-185 °C; M+H = 445.

EXAMPLE 320:

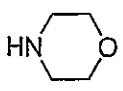


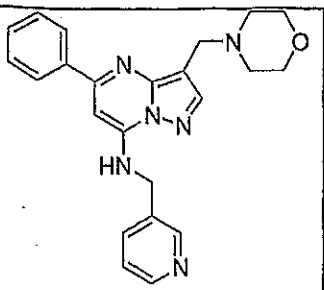
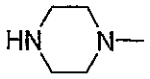
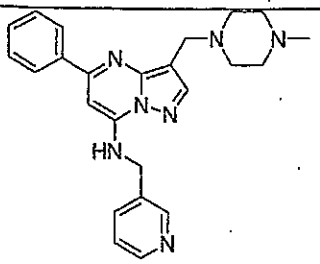
To a solution of 3-H analog (106 mg, 0.35 mmol) from Preparative Example 187.10 in AcOH (2 mL) was added 37% aqueous formaldehyde (1.5 ml; 1.40 mmol) followed by piperidine (100 μL; 0.37 mmol). The resulting mixture was stirred at rt for 24h and the AcOH was removed under reduced pressure. The mixture was diluted with water (2 mL) and neutralized with 2M NaOH until pH = 8. The aqueous layer was extracted with CH_2Cl_2 (3 x 7 mL) and the organic layers were combined. The organic layer was washed with brine (1 x 4 mL), dried (Na_2SO_4), filtered, and concentrated under reduced pressure to afford 96 mg (69%) of an off-white solid. mp 88-90 °C; M+H 399.

EXAMPLES 321-322:

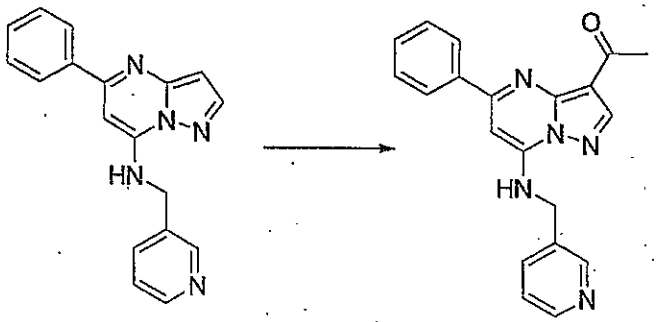
By essentially the same procedure set forth in Example 320 only substituting the amines in Column 2 of Table 29 and employing the 3-H adduct from Preparative Example 187.10, the compounds in Column 3 of Table 29 were prepared:

TABLE 29

Ex.	Column 2 (Amine)	Column 3 (Final Structure)	CMPD 1. mp (°C) 2. M+H
321			1. mp = 178-180 2. M+H = 401

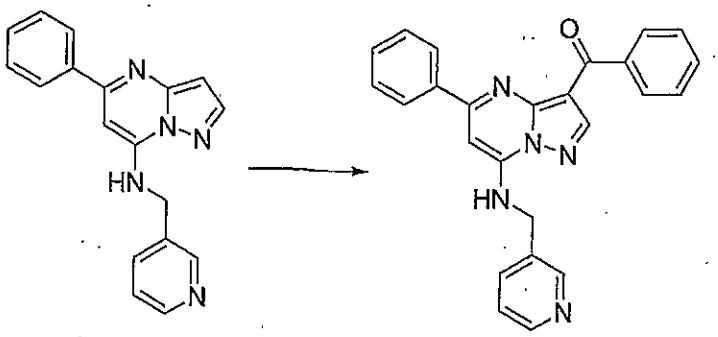
			
322			1. mp = 102-104 2. M+H = 414

EXAMPLE 323:



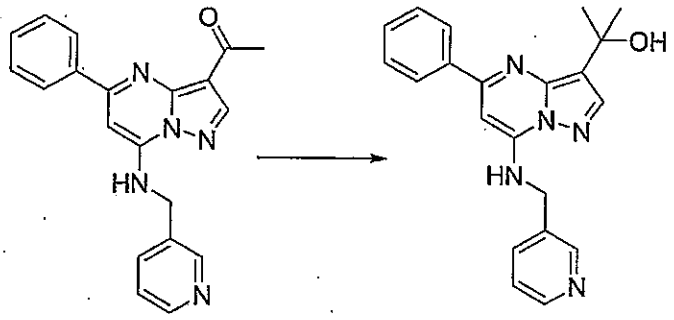
To a solution of 3-H analog (113 mg, 0.38 mmol) from Preparative Example 187.10 in CH_2Cl_2 (5 mL) at rt was added $AlCl_3$ (215 mg, 1.61 mmol) followed by AcCl (100 mL, 1.40 mmol). The mixture was heated at reflux for 12h and was cooled to rt. The mixture was treated sequentially with 3M HCl (3 mL) followed by sat. aq. $NaHCO_3$ (until pH = 8). The layers were separated and the aqueous layer was extracted with CH_2Cl_2 (2 x 5 mL). The organic layers were combined, dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by preparative TLC (8 x 1000 mM) eluting with CH_2Cl_2/MeOH (20:1) to afford 68 mg (52%) of white solid. mp 220-221 °C; M+H = 344.

EXAMPLE 324:



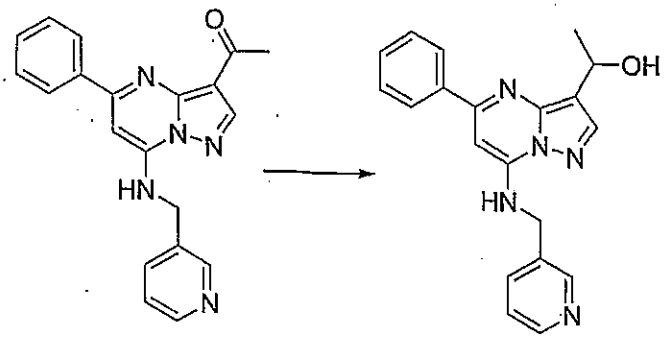
Utilizing the method described in Example 323, except employing benzoyl chloride, the title compound was prepared in 61% yield as a white solid. mp 172-175 °C; M+H = 406.

EXAMPLE 325:



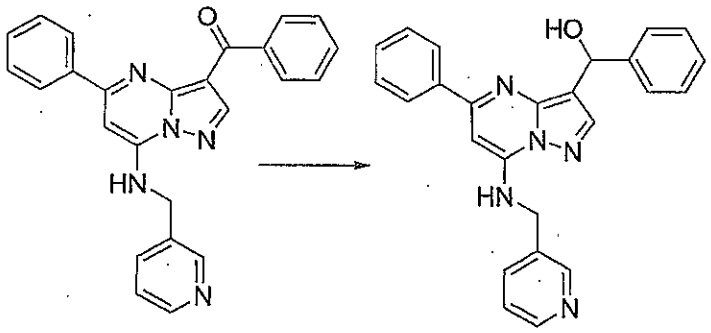
To a solution of ketone (100 mg, 0.29 mmol) from Example 323 in CH_2Cl_2 (2.5 mL) at 0 °C was added MeMgBr (0.35 mL, 3.0M in Et_2O) dropwise. The resulting mixture was stirred for 18h at rt and was carefully quenched by addition of sat. aq. NH_4Cl (2 mL) and CH_2Cl_2 (2 mL) were added. The layers were separated and the aqueous layer was extracted with CH_2Cl_2 (2 x 4 mL). The organic layers were combined, dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by preparative TLC (8 x 1000 μM) eluting with CH_2Cl_2/MeOH (10:1) to afford 68 mg (52%) of a yellow solid. mp 160-162 °C; M+H = 360.

EXAMPLE 326:



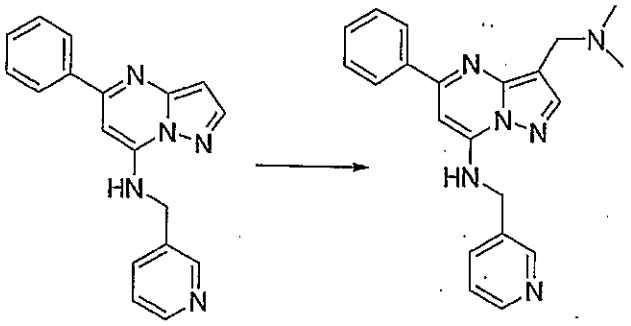
To a solution of ketone (84 mg, 0.24 mmol) from Example 323 in MeOH/THF (1:1; 2 mL total) at 0 °C was added $NaBH_4$ (12 mg, 0.30 mmol) in one portion. The resulting mixture was stirred for 18h at rt whereupon and additional portion of $NaBH_4$ (12 mg, 0.30 mmol) was added. The mixture was stirred for 12h whereupon the mixture was quenched with ice followed by addition of 1M NaOH to adjust the pH = 9. The mixture was diluted with CH_2Cl_2 (5 mL). The layers were separated and the aqueous layer was extracted with CH_2Cl_2 (2 x 4 mL). The organic layers were combined, dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by preparative TLC (8 x 1000 μM) eluting with CH_2Cl_2/MeOH (10:1) to afford 25 mg (30%) of a yellow solid. mp 148-150 °C; M+H = 346.

EXAMPLE 327:



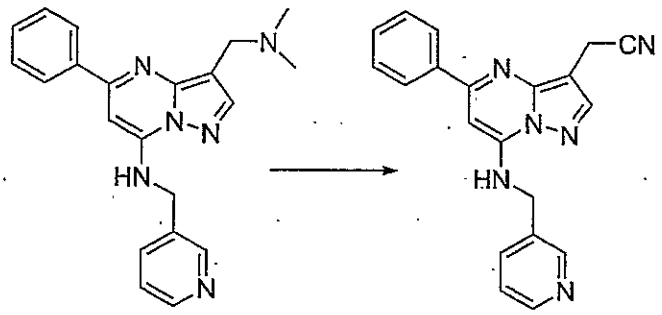
Using the same procedure as outlined in Example 326, the ketone (84 mg, 0.21 mmol) was converted to 53 mg (62%) as a light yellow solid. mp 78-80 °C; M+H = 408.

EXAMPLE 328:



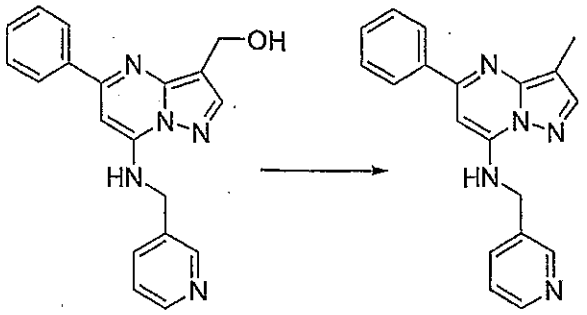
To a solution of 3-H adduct (1.3 g, 4.31 mmol) from Preparative Example 187.10 in CH_2Cl_2 (50 mL) was added Eschenmoser's salt (0.79 g, 4.31 mmol) followed by dropwise addition of TFA (0.56 mL, 7.33 mmol). The mixture was stirred at rt for 48 h and was diluted with CH_2Cl_2 (250 mL). The organic layer was washed with sat. aq. $NaHCO_3$ (2 x 125 mL) to afford 1.41 h (92%) of a yellow solid. mp 231-233 °C; M+H = 359.

EXAMPLE 329:



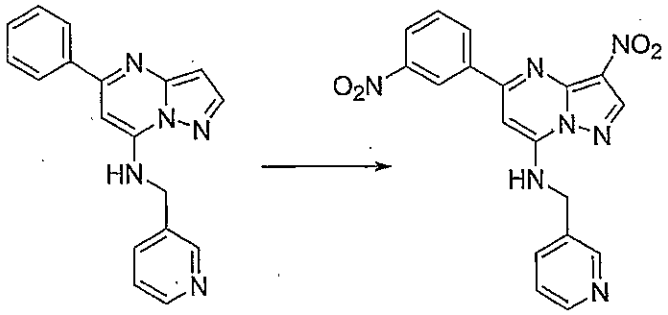
To a solution of tertiary amine adduct (100 mg, 0.28 mmol) from Example 328 in 50% aq. DMF (5 mL) in a pressure tube was added KCN (0.15 g, 2.32 mmol). The tube was capped and heated at 100 °C for 96h. The mixture was cooled to rt and was diluted with EtOAc (25 mL). The organic layer was washed with brine (1 x 5 mL) and water (1 x 5 mL). The organic layers was dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by preparative TLC (4 x 1000 μM) eluting with EtOAc to afford 21 mg (30%) of brown solid. mp 152-155 °C; M+H = 341.

EXAMPLE 330:



To a solution of alcohol (45 mg, 0.14 mmol) from Example 17.10 in CH_2Cl_2 (0.7 mL) at 0 °C was added Et_3SiH (26 μL, 0.16 mmol) followed by TFA (25 μL, 0.33 mmol). The mixture was stirred for 2 h at 0 °C and rt for 2 h whereupon additional portions of Et_3SiH (26 μL, 0.16 mmol) and TFA (25 μL, 0.33 mmol) were added and the mixture was stirred for 4 h at rt (until complete by TLC). The mixture was concentrated under reduced pressure and the crude residue was partitioned between CH_2Cl_2 (3 mL) and sat. aq. $NaHCO_3$ (1.5 mL). The layers were separated and the aqueous layer was extracted with CH_2Cl_2 (2 x 4 mL). The organic layers were combined, washed with brine (1 x 5 mL), dried (Na_2SO_4), filtered and concentrated under reduced pressure. The crude product was purified by prep TLC (4 x 1000 mM) eluting with CH_2Cl_2/MeOH (20:1) to afford 21 mg (48%) of a yellow solid. mp 146-148 °C; M+H = 316.

EXAMPLE 331:

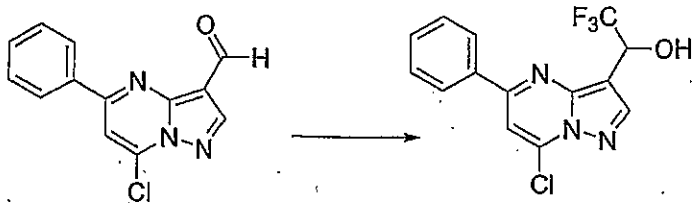


To a solution of 3-H adduct (90 mg, 0.30 mmol) from Preparative Example 187.10 in conc. H_2SO_4 (2 mL) at 0 °C was added fuming HNO_3 (30 μL, 0.72 mmol) dropwise. The resulting mixture was stirred for 1 h at 0 °C whereupon ice

(~1g) was added to the mixture. The resulting precipitate was collected and was washed with water (2 x 2 mL) and CH_2Cl_2 (2 x 2 mL). The crude product was dried under high vacuum to afford 67 mg (60%) of the monosulfate salt as a yellow/orange solid. mp 250 °C; M+H (free base) = 392.

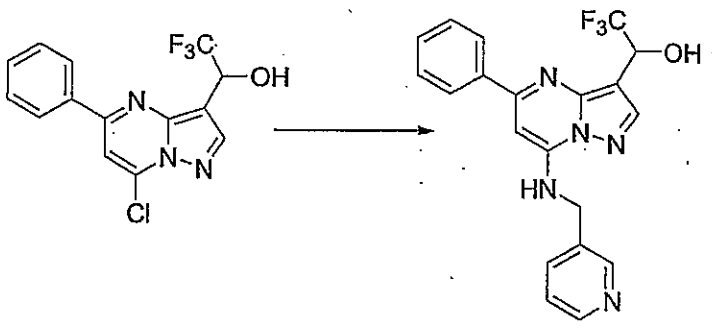
EXAMPLE 332:

Step A:



To a solution of aldehyde (0.10 g, 0.39 mmol) from Preparative Example 168 in THF (2.5 mL) at 0 °C was added CF_3TMS (64 mL, 0.43 mmol) followed by CsF (10 mg). The resulting mixture was stirred for 2 h at 0 °C and 2h at rt. 1M HCl (5 mL) was added and the mixture was diluted with CH_2Cl_2 (10 mL). The layers were separated, the aqueous layer was extracted with CH_2Cl_2 (2 x 10 mL), and the organic layers were combined. The organic layer was washed with brine (1 x 10 mL), dried (Na_2SO_4), filtered, and concentrated under reduced pressure to afford 127 mg (99%) of a yellow semisolid. M+H =328. The crude product was carried on without further purification.

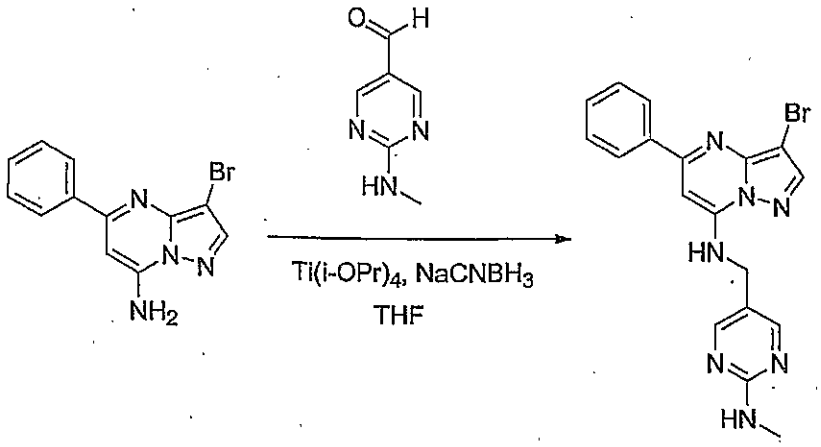
Step B:



By utilizing the general procedure set forth in Example 1, the 7-Cl adduct (127 mg, 0.39 mmol) from Example 332, Step A was reacted with 3-

(aminomethyl)pyridine (73 μL, 0.43 mmol) to afford 80 mg (51%) of the title compound as a light yellow solid. mp 68-72 °C; M+H = 400.

EXAMPLE 333:



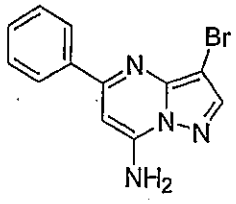
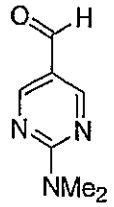
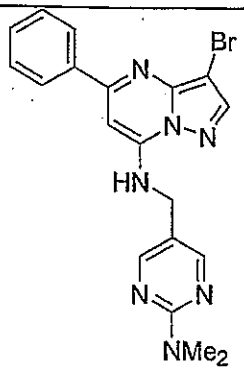
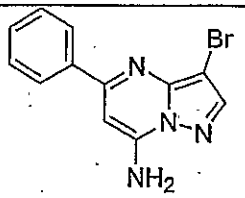
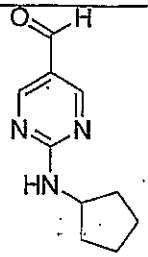
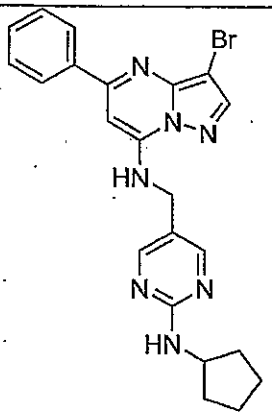
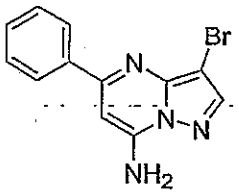
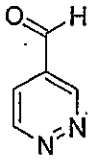
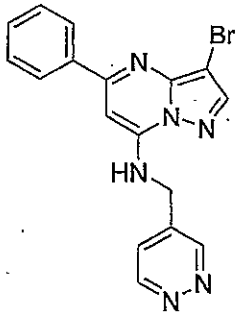
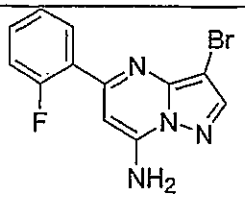
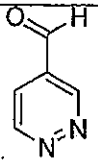
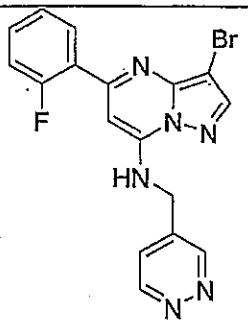
To a solution of aniline (200 mg, 0.69 mmol) from Preparative Example 174 in THF (6 mL) at rt was added aldehyde (114 mg, 0.83 mmol) from Preparative Example 256 followed by dropwise addition of Ti(*i*-OPr)$_4$ (0.82 mL, 2.77 mmol). The mixture was stirred at reflux for 4 h and was cooled to rt. $NaCNBH_3$ (347 mg, 5.53 mmol) was added and the mixture was stirred for 2 h at rt. The mixture was cooled to 0 oC, treated with 1M NaOH (4 mL) and brine (1 mL) and stirred for 30 min. The mixture was extracted with CH_2Cl_2 (3 x 10 mL) and the organic layers were combined. The organic layer was washed with brine (1 x 7 mL), dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by preparative thin-layer chromatography (8 x 1000 υM plates) eluting with CH_2Cl_2/MeOH (25:1) to afford 89 mg (31%) of the title compound as a yellow solid. mp 210-213 °C ; M+H = 411.

EXAMPLES 334-337:

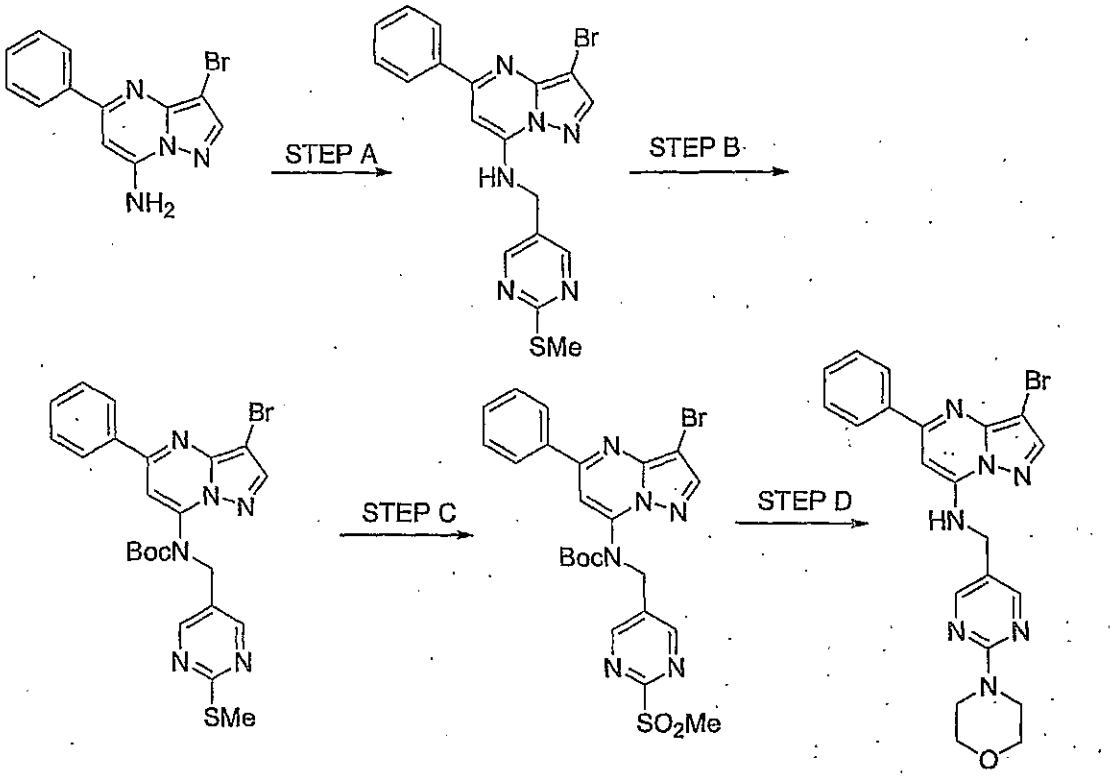
By essentially the same procedure set forth in Example 333 only by utilizing the anilines shown in Column 2 of Table 30 and the aldehydes shown in Column 3 of Table 30, the compounds in Column 4 of Table 30 were prepared:

TABLE 30

Ex.	Column 2 (Aniline)	Column 3 (Aldehyde)	Column 4 (Final Structure)	CMPD 1. mp (°C) 2. M+H

334				1. mp = 85-87 2. M+H = 425
335				1. mp = 160-162 2. M+H = 451
336				1. mp = 117-119 2. M+H = 382
337				1. mp = 171-175 2. M+H = 400

EXAMPLE 338:



STEP A:

Reaction of aniline (0.20 g, 0.69 mmol) with aldehyde (0.13 g, 0.83 mmol) under the reaction conditions described in Example 333 afforded 70 mg (23%) of thiomethyl derivative as a yellow solid. M+H = 428.

STEP B:

To a solution of thiomethyl derivative (60 mg, 0.14 mmol) from Example 338, Step A in dioxane (2 mL) was added Boc_2O (61 mg, 0.28 mmol) followed by DMAP (21 mg, 0.17 mmol). The mixture was stirred for 14h at rt and was concentrated under reduced pressure. The crude product was purified by preparative thin-layer chromatography (6 x 1000 μM plates) eluting with hexanes/EtOAc (4:1) to afford 61 mg (83%) of the title compound as a yellow solid. M+H = 528.

STEP C:

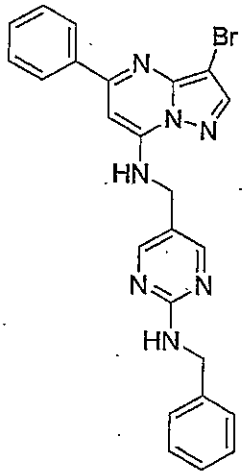
To a solution of thiomethyl derivative from Example 338, Step B (41 mg, 0.078 mmol) in CH_2Cl_2 (2 mL) was added MCPBA (33 mg, 0.19 mmol) in one portion. The resulting mixture was stirred for 3h at rt and the mixture was diluted

with CH_2Cl_2 (5 mL) and sat. aq. $NaHCO_3$ (2.5 mL). The layers were separated, the aqueous layer was extracted with CH_2Cl_2 (2 x 5 mL), and the organic layers were combined. The organic layer was dried (Na_2SO_4), filtered, and concentrated under reduced pressure to afford 40 mg (92%) of the sulfone adduct as a light yellow solid. M+H = 560.

STEP D:

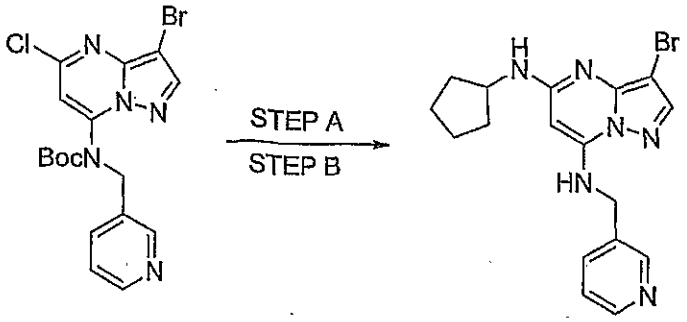
To a flask charged with sulfone from Example 338, Step C (75 mg, 0.13 mmol) and a stir bar was added morpholine (2 ml; 22 mmol). The mixture was heated at reflux for 12h, cooled to rt, and concentrated to dryness under high vacuum. The crude product was purified by preparative thin-layer chromatography (6 x 1000 μM plates) eluting with CH_2Cl_2/MeOH (40:1) to afford 41 mg (68%) of the title compound as a yellow solid. mp 209-210 °C; M+H = 466.

EXAMPLE 339:



The title compound was prepared according to the procedure outlined in Example 338 except using benzyl amine to afford 12 mg (70%) of a white solid. mp 194-196; M+H = 487.

EXAMPLE 340:



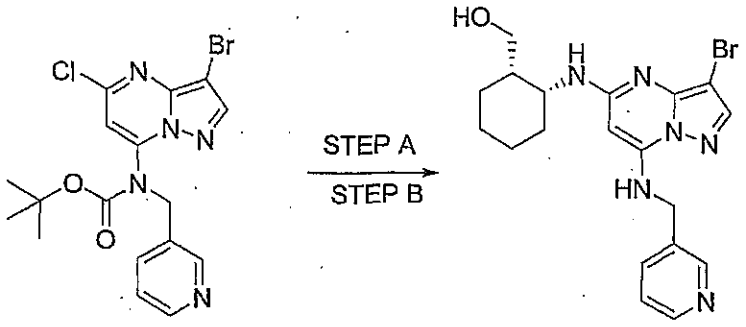
STEP A:

To a solution of 5-chloro adduct (0.15 g, 0.34 mmol) in dioxane/DIPEA (2.5mL/1.0mL) at rt was added cyclopentylamine (0.041 μL, 0.41 mmol) dropwise. The resulting solution was stirred at reflux for 16h, cooled to rt, and concentrated under reduced pressure. The crude material was purified by preparative thin-layer chromatography (8 x 1000 μM) eluting with CH_2Cl_2/MeOH (25:1) to afford 148 mg (89%) of a yellow oil. M+H = 489.

STEP B: Removal of the t-butoxycarbonyl protecting group with TFA

To a solution of the compound prepared in Example 340, Step A (135 mg, 0.28 mmol) in CH_2Cl_2 (2 mL) at rt was added TFA (0.54 mL, 7.0 mmol) dropwise. The resulting solution was stirred for 18 h at rt and was concentrated under reduced pressure. The crude material was redissolved in CH_2Cl_2 (5 mL) and the organic layer was sequentially washed with sat. aq. $NaHCO_3$ (2 x 2 mL) and brine (1 x 2 mL). The organic layer was dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude material was purified by preparative thin-layer chromatography (8 x 1000 μM) eluting with CH_2Cl_2/MeOH (20:1) to afford 105 mg (97%) of white solid. mp 120-122 °C; M+H = 389.

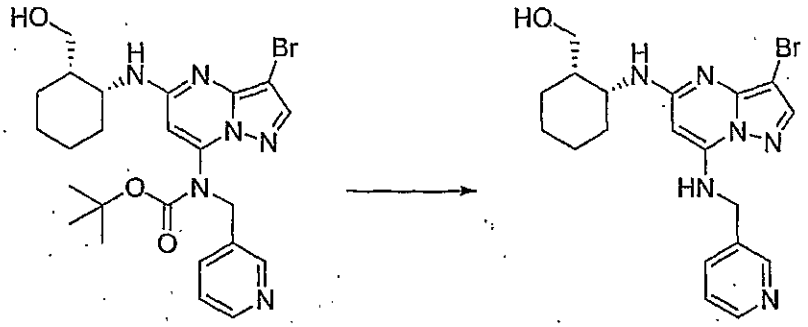
EXAMPLE 341:



Step A:

By essentially the same procedure set forth in Example 340 only substituting the appropriate amine, the above compound was prepared. MS: MH^+= 431..

Step B: Removal to t-butoxycarbonyl protecting group with KOH.

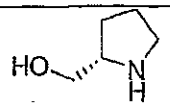
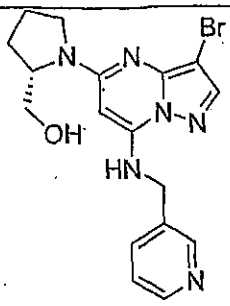
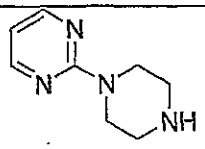
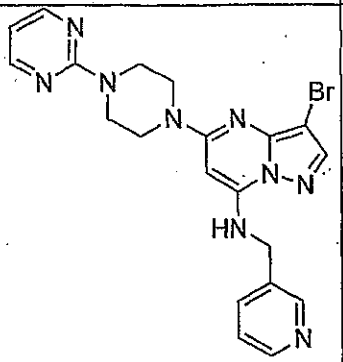
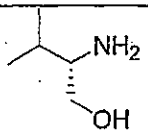
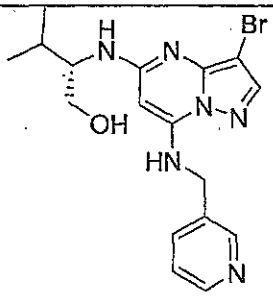
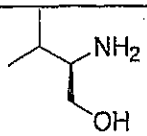
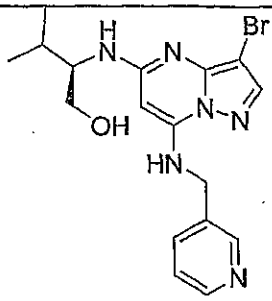


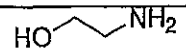
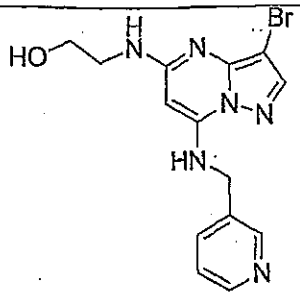
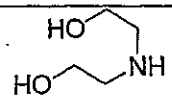
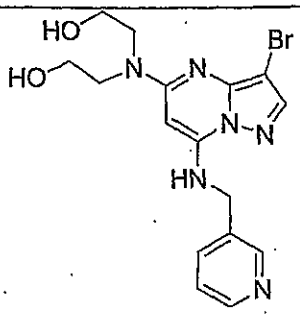
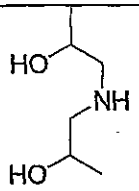
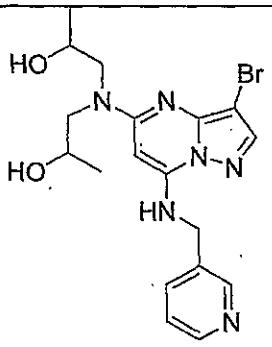
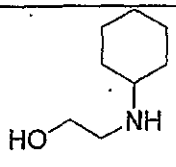
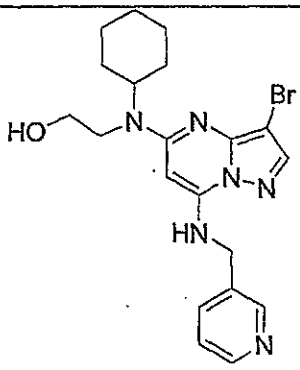
To a mixture of the compound prepared in Example 341, Step A (0.14 g, 0.26 mmol) in EtOH : H_2O (3 mL, 2 : 1) was added KOH (0.29 g, 20 eq.) in one portion. The resulting solution was stirred at reflux 14 hours, cooled to room temperature, and concentrated under reduced pressure. The residue was taken up in CH_2Cl_2 (5 mL) and diluted with saturated $NaHCO_3$ (2 mL). The layers were separated and the aqueous layer extracted with CH_2Cl_2 (2 x 4 mL). The combined organics were washed with brine, dried over Na_2SO_4, filtered, and concentrated under reduced pressure. The crude product was purified by preparative TLC (8 x 1000 μM) eluting with 5% MeOH in CH_2Cl_2 solution (0.066 g, 59% yield). MS: MH^+ = 432; mp = 219-221°C.

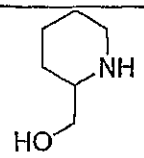
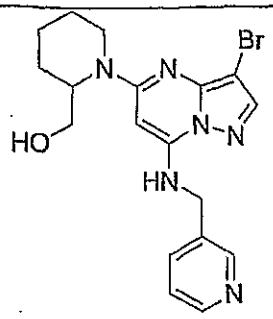
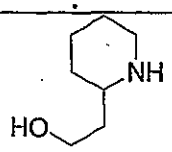
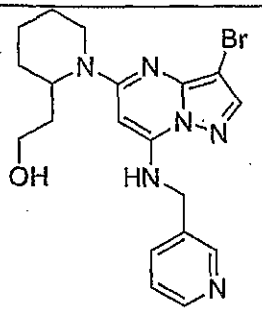
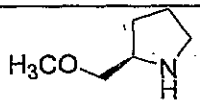
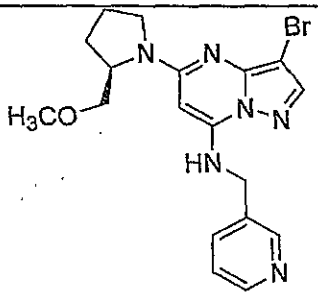
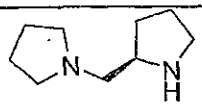
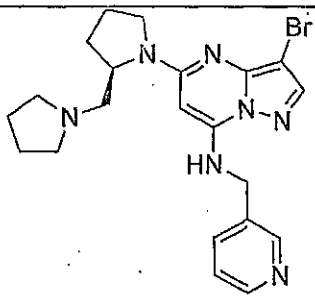
EXAMPLES 342-397:

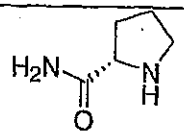
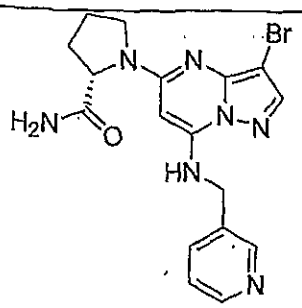
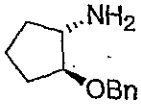
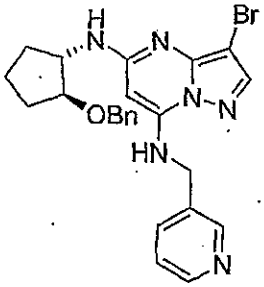
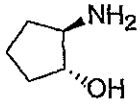
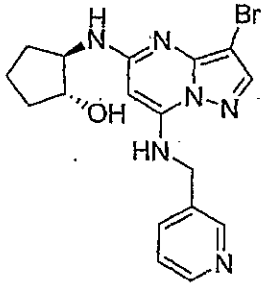
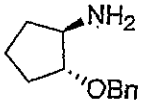
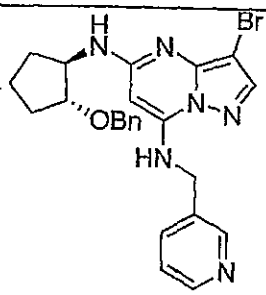
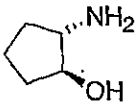
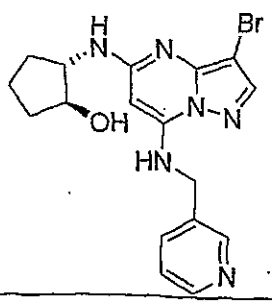
By essentially the same procedure set forth in Example 340 only substituting the chlorides in Column 2 of Table 31 and removing the t-butoxycarbonyl protecting group by the method shown in Column 3 of Table 31, the compounds shown in Column 4 of Table 31 were prepared.

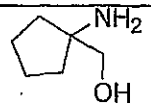
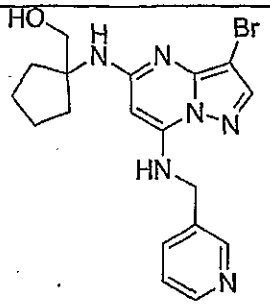
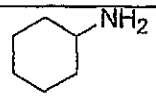
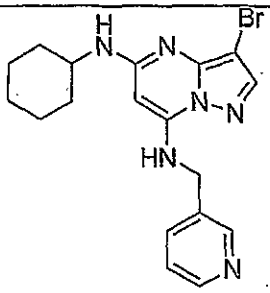
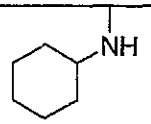
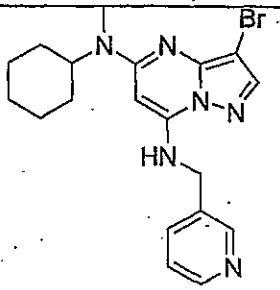
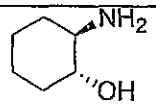
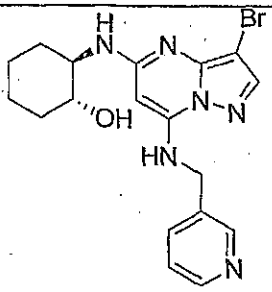
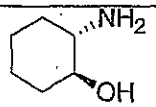
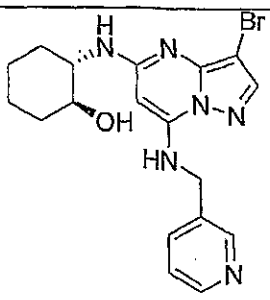
Table 31

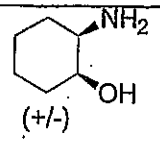
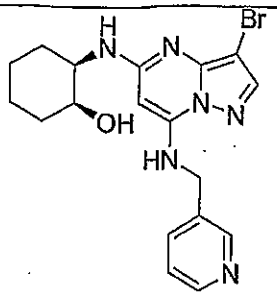
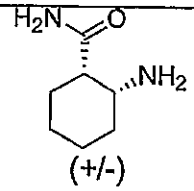
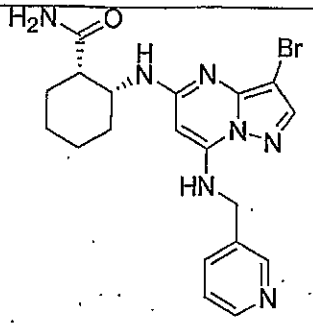
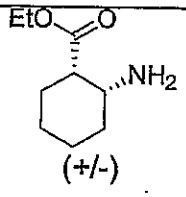
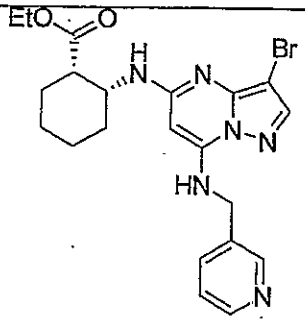
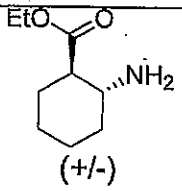
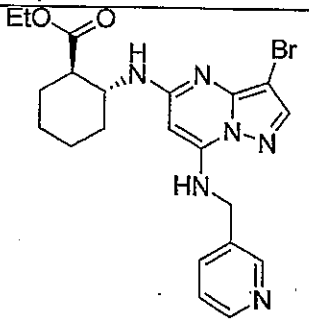
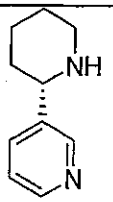
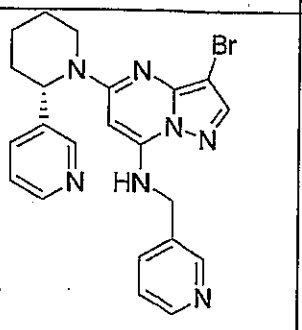
Ex.	Column 2	Column 3	Column 4	CMPD
342		HCl		MS: MH^+ = 403 m.p. 151 – 157 °C
343		HCl		MS: MH^+ = 466 m.p. 212 – 217 °C
344		HCl		MS: MH^+ = 405 m.p. 53 – 58 °C
345		HCl		MS: MH^+ = 405 m.p. 63 – 69 °C

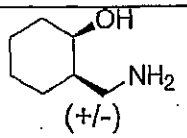
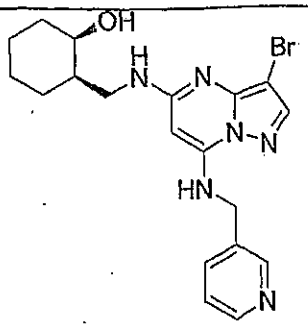
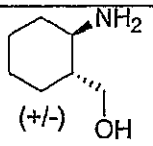
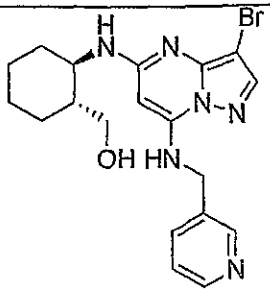
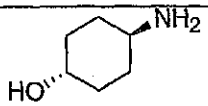
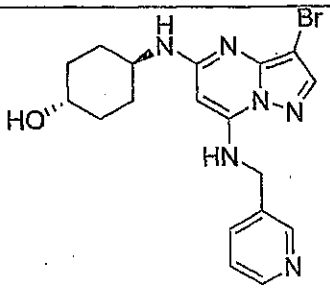
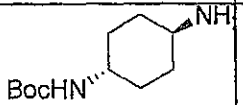
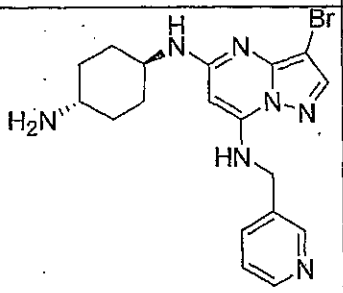
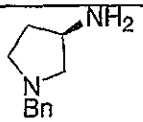
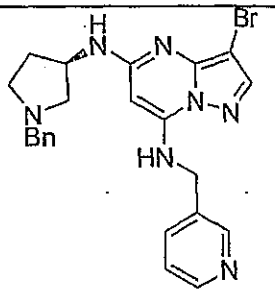
346		HCl		MS: MH^+ = 363 m.p. 170 – 171 °C
347		HCl		MS: MH^+ = 407 m.p. 148 – 151 °C
348		HCl		MS: MH^+ = 435 m.p. 56 – 59 °C
349		HCl		MS: MH^+ = 445 m.p. 66 – 68 °C

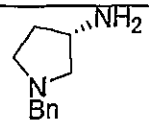
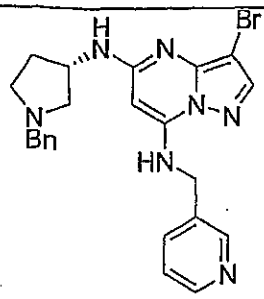
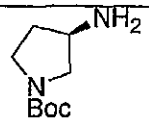
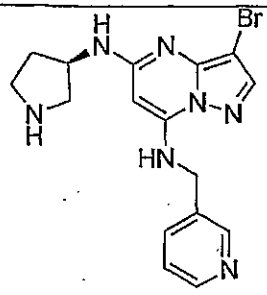
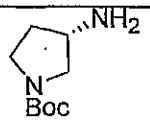
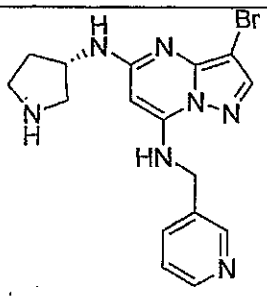
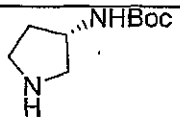
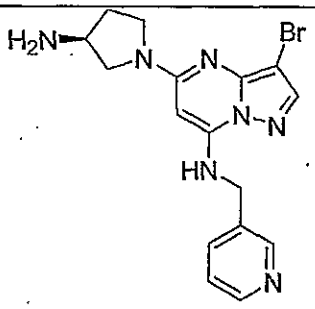
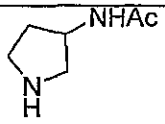
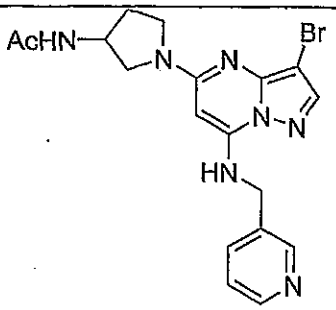
350		KOH		MS: MH^+ = 417 m.p. 149 – 151 °C
351		KOH		MS: MH^+ = 431 m.p. 111 – 114 °C
352		KOH		MS: MH^+ = 417 m.p. 53 – 58 °C
353		KOH		MS: MH^+ = 456 m.p. 186 – 189 °C

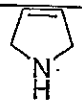
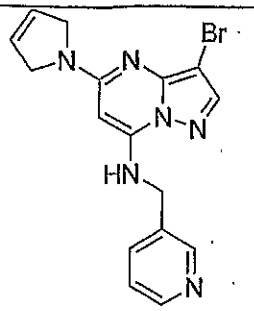
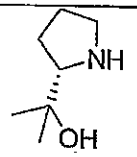
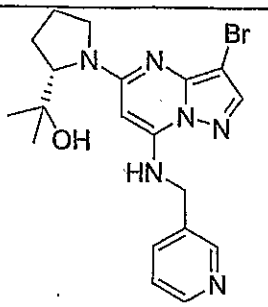
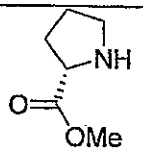
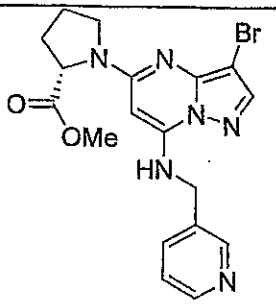
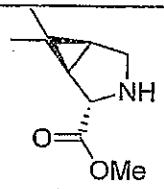
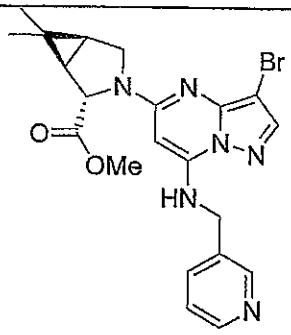
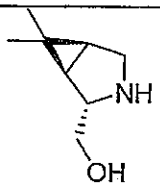
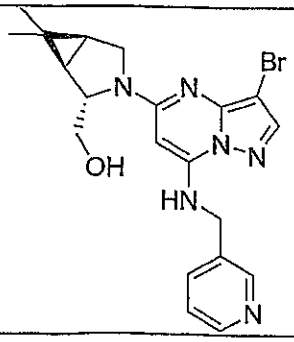
354		KOH		MS: MH^+ = 416 m.p. 210 – 213 °C
355		TFA		1. mp = 68-70 2. M+H = 494
356		KOH		1. mp = 181-183 2. M+H = 404
357		TFA		1. mp = 69-71 2. M+H = 494
358		KOH		1. mp = 182-184 2. M+H = 404

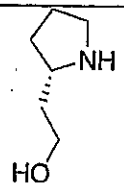
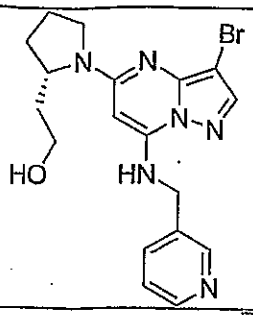
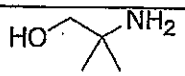
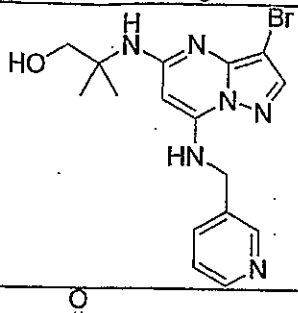
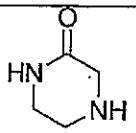
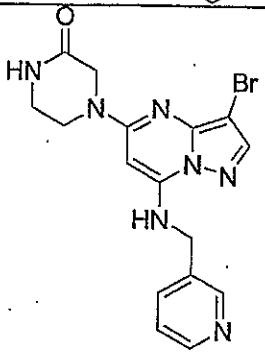
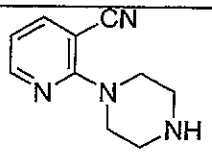
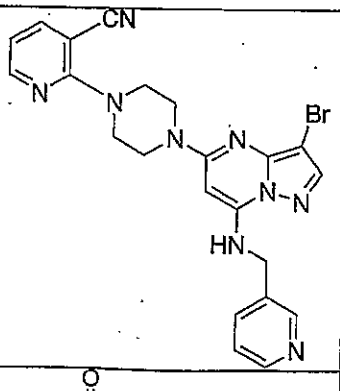
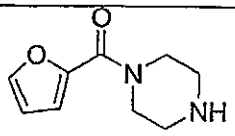
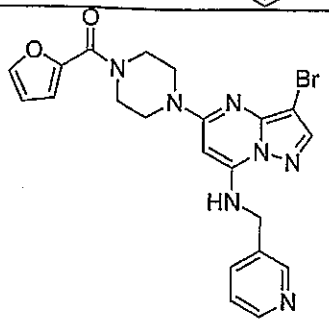
359		KOH		1. mp = 202-204 2. M+H = 418
360		TFA		1. mp = 160-162 2. M+H = 402
361		TFA		1. mp = 151-153 2. M+H = 416
362		KOH		1. mp = 140-143 2. M+H = 418
363		KOH		1. mp = 139-142 2. M+H = 418

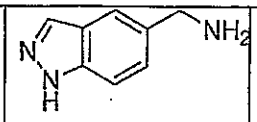
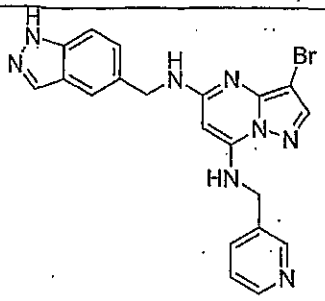
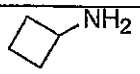
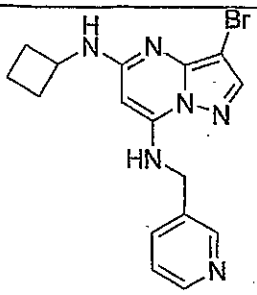
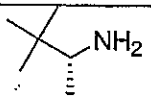
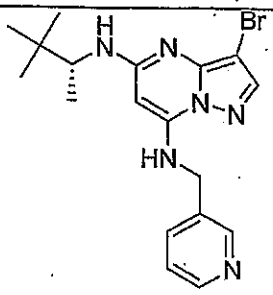
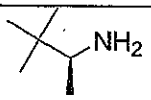
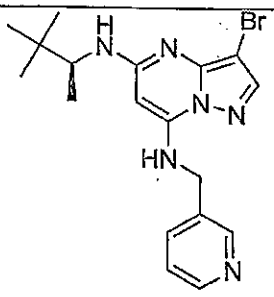
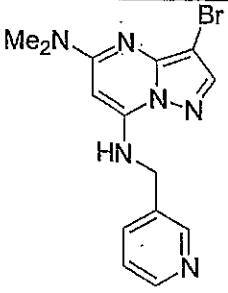
364		KOH		1. mp = 115-117 2. M+H = 418
366		TFA		1. mp = 102-104 2. M+H = 445
367		TFA		1. mp = 118-120 2. M+H = 474
368		TFA		1. mp = 106-108 2. M+H = 474
369		TFA		1. mp = 160-161 2. M+H = 464

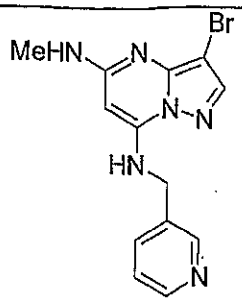
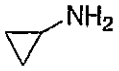
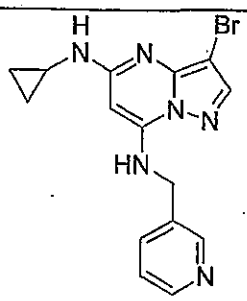
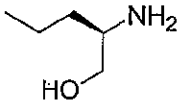
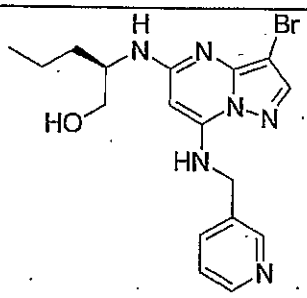
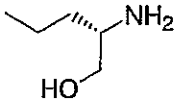
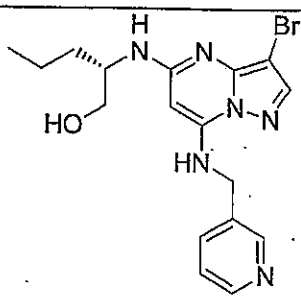
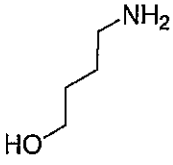
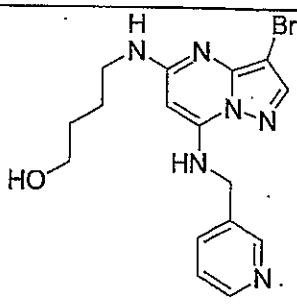
370		TFA		1. mp = 93-95 2. M+H = 432
371		KOH		1. mp = 108-110 2. M+H = 432
372		KOH		1. mp = 180-182 2. M+H = 418
373		TFA		1. mp = 169-170 2. M+H = 417
374		TFA		1. mp = 77-79 2. M+H = 479

375		TFA		1. mp = 76-79 2. M+H = 479
376		TFA		1. mp = 105-107 2. M+H = 389
377		TFA		1. mp = 105-107 2. M+H = 389
378		TFA		1. mp = 130-133 2. M+H = 389
379		TFA		1. mp = 132-135 2. M+H = 431

380		TFA		1. mp = 135-137 2. M+H = 372
381		KOH		1. mp = 78-82 2. M+H = 432
382		TFA		1. mp = 101-103 2. M+H = 432
383		TFA		1. mp = 92-95 2. M+H = 472
384		TFA		1. mp = 107-111 2. M+H = 444

384.10		TFA		1. mp = 2. M+H = 417
384.11		TFA		1. mp = 210-212 2. M+H = 391
385		TFA		1. mp = 122-124 2. M+H = 403
386		TFA		1. mp = 186-188 2. M+H = 491
387		TFA		1. mp = 173-175 2. M+H = 483

388		TFA		1. mp = 167-169 2. M+H = 450
389		TFA		1. mp = 90-92 2. M+H = 374
390		TFA		1. mp = 113-115 2. M+H = 404
391		TFA		1. mp = 114-116 2. M+H = 404
392	$HNMe_2$	TFA		LCMS: MH^+ = 347;

393	H_2NMe	TFA		LCMS: MH^+ = 333;
394		TFA		LCMS: MH^+ = 359;
395		TFA		LCMS: MH^+ = 405;
396		TFA		LCMS: MH^+ = 405;
397		TFA		LCMS: MH^+ = 391;

Additional data for select example shown below:

Example 392: ^{1}H NMR (DMSO-d_6) δ 8.65 (s, 1H), 8.46 (d, J = 3.3 Hz, 1H), 8.21 (t, J = 6.6 Hz, 1H), 7.90 (s, 1H), 7.80 (d, J = 7.8 Hz, 1H), 7.35 (dd, J = 7.8, 4.8 Hz, 1H), 5.46 (s, 1H), 4.61 (d, J = 6.9 Hz, 2H), 3.01 (s, 6H).

Example 393: ^{1}H NMR ($CDCl_3$) δ 8.65 (s, 1H), 8.60 (d, J = 4.8 Hz, 1H), 7.76 (s, 1H), 7.70 (m, 1H), 7.32 (dd, J = 8.1, 4.8 Hz, 1H), 6.43 (t, J = 6.0 Hz, 1H), 5.08 (s, 1H), 4.80 (m, 1H), 4.56 (d, J = 6.0 Hz, 2H), 2.96 (d, J = 5.1 Hz, 3H).

Example 394: ^{1}H NMR ($CDCl_3$) δ 8.68 (s, 1H), 8.60 (d, J = 4.8 Hz, 1H), 7.76 (s, 1H), 7.72 (m, 1H), 7.32 (dd, J = 7.8, 5.4 Hz, 1H), 6.55 (t, J = 5.7 Hz, 1H), 5.53 (s, 1H), 5.35 (s, 1H), 4.62 (d, J = 5.7 Hz, 2H), 2.49 (m, 1H), 0.75 (m, 2H), 0.51 (m, 2H).

Example 395: ^{1}H NMR ($CDCl_3$) δ 8.65 (s, 1H), 8.60 (d, J = 4.0 Hz, 1H), 7.75 (s, 1H), 7.69 (m, 1H), 7.33 (dd, J = 8.1, 5.1 Hz, 1H), 6.45 (t, J = 6.0 Hz, 1H), 5.07 (s, 1H), 4.69 (m, 1H), 4.54 (d, J = 6.0 Hz, 2H), 3.98 (m, 1H), 3.79 (dd, J = 10.8, 2.4 Hz, 1H), 3.59 (dd, J = 11.1, 7.2 Hz, 1H), 1.59-1.36 (m, 4H), 0.94 (t, J = 6.9 Hz, 3H).

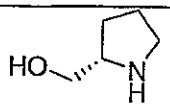
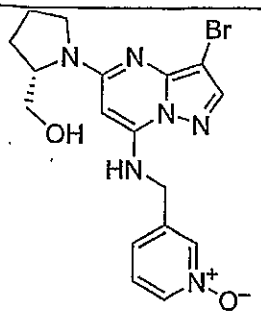
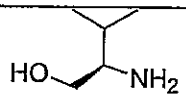
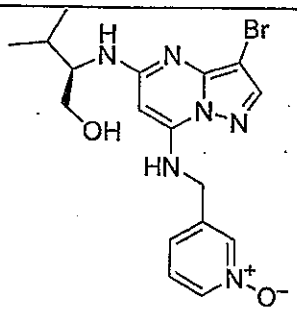
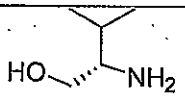
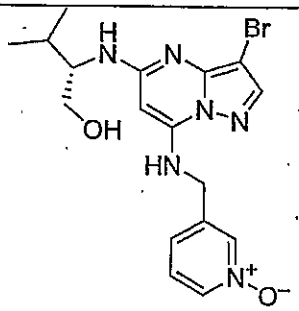
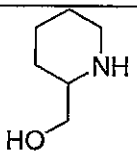
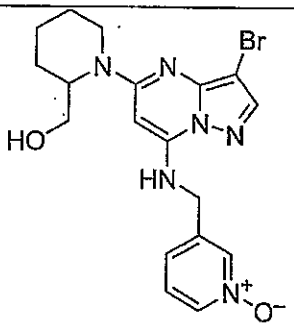
Example 396: ^{1}H NMR ($CDCl_3$) δ 8.60 (s, 1H), 8.56 (d, J = 4.2 Hz, 1H), 7.73 (s, 1H), 7.66 (m, 1H), 7.31 (dd, J = 7.8, 4.8 Hz, 1H), 6.51 (t, J = 6.0 Hz, 1H), 5.05 (s, 1H), 4.86 (d, J = 6.6 Hz, 1H), 4.50 (d, J = 6.0 Hz, 2H), 3.94 (m, 1H), 3.78 (dd, J = 11.1, 2.4 Hz, 1H), 3.57 (dd, J = 11.1, 7.2 Hz, 1H), 1.57-1.34 (m, 4H), 0.91 (t, J = 7.2 Hz, 3H).

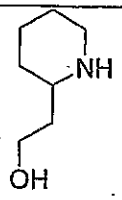
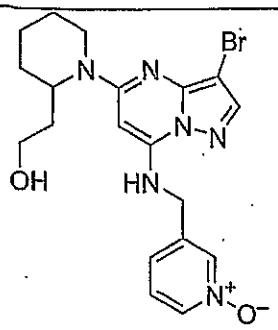
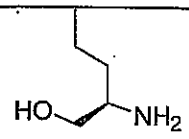
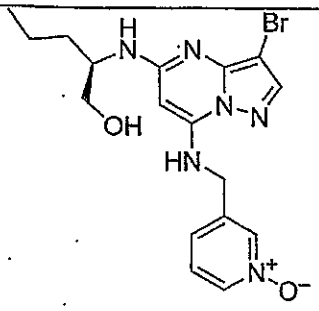
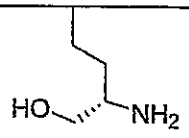
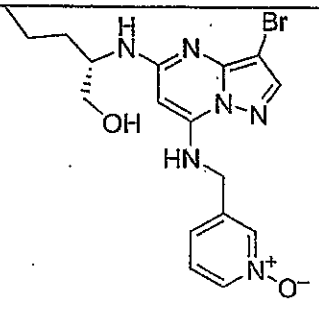
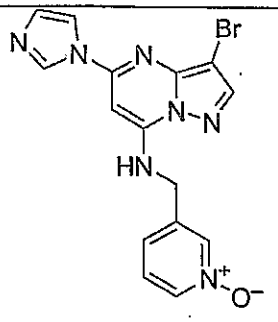
Example 397: ^{1}H NMR ($CDCl_3$) δ 8.65 (s, 1H), 8.59 (d, J = 4.5 Hz, 1H), 7.75 (s, 1H), 7.69 (m, 1H), 7.31 (m, 1H), 6.43 (t, J = 6.0 Hz, 1H), 5.06 (s, 1H), 4.88 (m, 1H), 4.55 (d, J = 6.0 Hz, 2H), 3.70 (m, 2H), 3.38 (m, 2H), 1.79-1.61 (m, 4H).

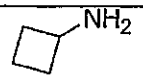
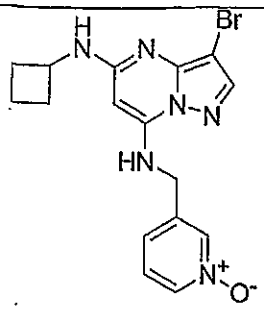
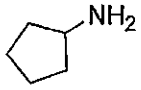
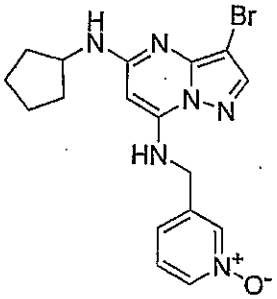
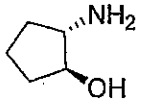
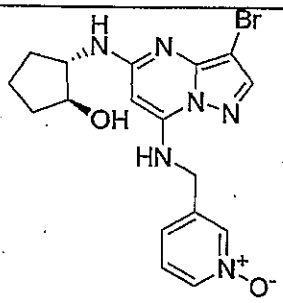
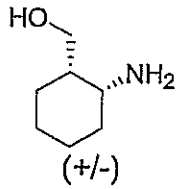
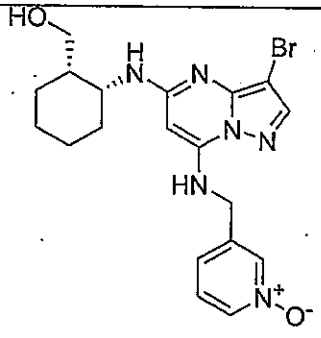
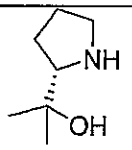
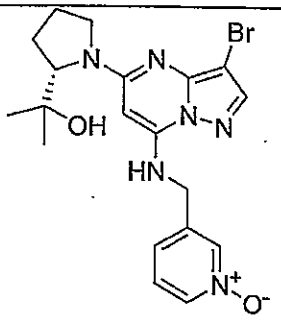
EXAMPLES 398-416:

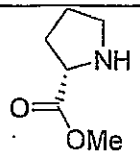
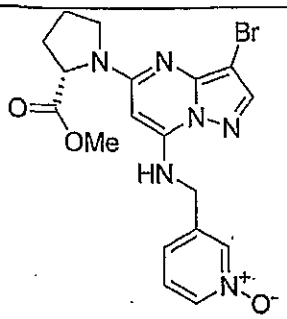
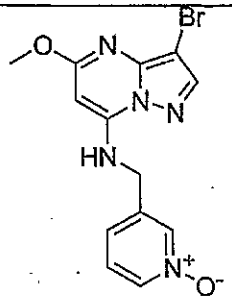
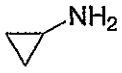
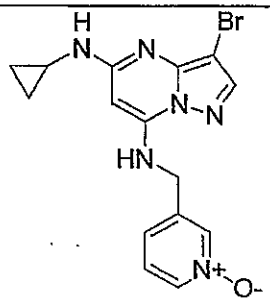
By essentially the same conditions set forth in Example 341, Steps A and B only substituting the compound prepared in Preparative Example 193.10, the compounds in Column 4 of Table 32 were prepared.

Table 32

Ex.	Column 2	Column 3	Column 4	CMPD
398				MS: MH^+ = 419 m.p. 102 – 105 °C
399				MS: MH^+ = 421 m.p. 79 – 81 °C
400				MS: MH^+ = 421 m.p. 78 – 79 °C
401				MS: MH^+ = 433 m.p. 228-231 °C

402				MS: MH^+ = 447 m.p. 97-102 °C
403				MS: MH^+ = 421 m.p. °C
404				MS: MH^+ = 421 m.p. °C
405				MS: MH^+ = 386 m.p. °C

407		KOH		1. mp = 98-100 2. M+H = 390
408		TFA		1. mp = 170-173 2. M+H = 404
409		KOH		1. mp = 219-221 2. M+H = 420
410		KOH		1. mp = 110-112 2. M+H = 448
411		TFA		1. mp = 81-83 2. M+H = 448

412		TFA		1. mp = 136-138 2. M+H = 448
413	NaOMe	KOH		1. mp = 107-110 2. M+H = 351
414				LCMS: MH^+ = 375;

Additional data for select examples shown below:

Example 414: ^{1}H NMR (DMSO-d_6) δ 8.26 (s, 1H), 8.23 (m, 1H), 8.13 (m, 1H), 7.90 (s, 1H), 7.40-7.27 (m, 3H), 5.34 (s, 1H), 4.49 (d, J = 6.3 Hz, 2H), 2.56 (m, 1H), 0.67 (m, 2H), 0.35 (m, 2H).

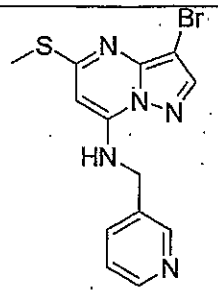
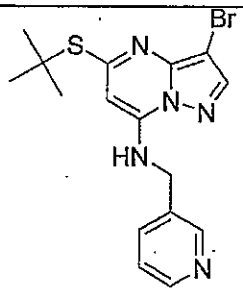
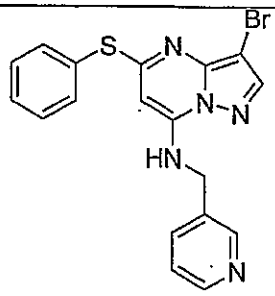
Example 403: ^{1}H NMR (DMSO-d_6+$CDCl_3$) δ 8.08 (s, 1H), 7.90 (d, J = 6.3 Hz, 1H), 7.49 (s, 1H), 7.34 (t, J = 6.3 Hz, 1H), 7.16-7.09 (m, 2H), 5.65 (d, J = 6.6 Hz, 1H), 4.97 (s, 1H), 4.90 (s, 1H), 4.29 (d, J = 6.3 Hz, 2H), 3.70 (m, 1H), 3.46 (m, 1H), 3.34 (m, 1H), 1.35-1.17 (m, 4H), 0.71 (t, J = 7.2 Hz, 3H).

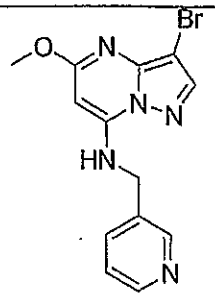
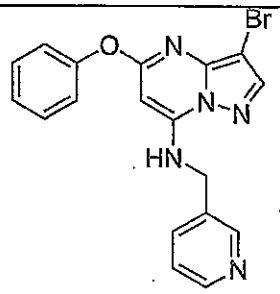
Example 404: ^{1}H NMR (DMSO-d_6) δ 8.21 (s, 1H), 8.12 (d, J = 6.6 Hz, 1H), 8.06 (m, 1H), 7.86 (s, 1H), 7.38 (t, J = 7.8 Hz, 1H), 7.30 (d, J = 7.5 Hz, 1H), 6.73 (d, J = 8.7 Hz, 1H), 5.28 (s, 1H), 4.70 (t, J = 5.1 Hz, 1H), 4.41 (d, J = 6.6 Hz, 2H), 4.00 (s, 1H), 3.39 (m, 1H), 1.53 (m, 1H), 1.36-1.25 (m, 3H), 0.86 (t, J = 7.0 Hz, 3H).

EXAMPLES 417—421:

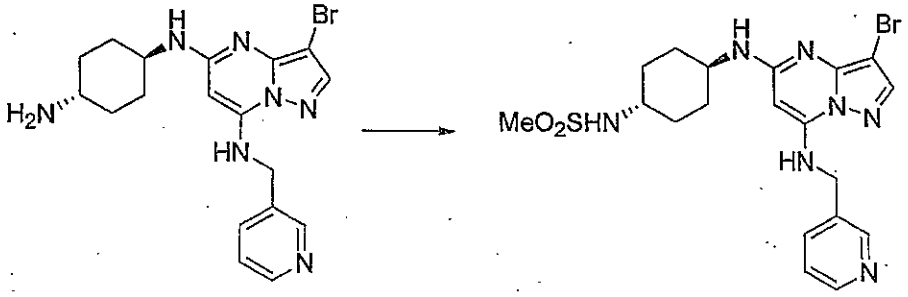
By the procedure set forth in *Chem. Pharm. Bull.* **1999**, *47*, 928-938. utilizing the oxygen or sulfur nucleophiles shown in Column 2 as described of Table 33 and by employing the cleavage method listed in Column 3 of Table 33, the compounds in Column 4 of Table 33 were prepared:

TABLE 33

Ex.	Column 2 (Nucleophile)	Column 3 (Cleavage method)	Column 4 (Final Structure)	CMPD 1. mp. 2. M+H
417	NaSMe	TFA		1. mp = 172-175 2. M+H = 351
418	NaSt-Bu	TFA		1. mp = 165-168 2. M+H = 392
419	NaSPh	TFA		1. mp = 154-156 2. M+H = 412
420	NaOMe	TFA		1. mp = 161-163 2. M+H = 335

				
421	NaOPh	TFA		1. mp = 64-66 2. M+H = 397

EXAMPLE 422:

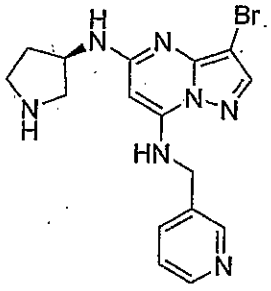
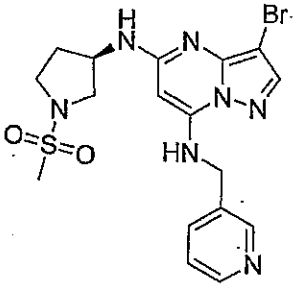
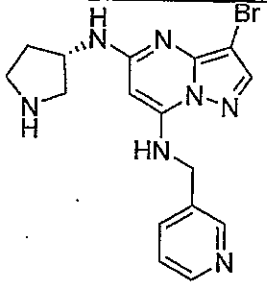
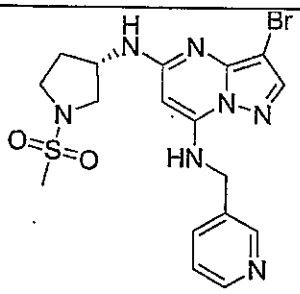


To a solution of amino compound (18 mg, 0.043 mmol) from Example 373 in CH_2Cl_2 (1 mL) at rt was added DIPEA (10 μL, 0.056 mmol) followed by $MeSO_2Cl$ (4 mL, 0.052 mmol). The mixture was stirred at rt for 12 h and was diluted with CH_2Cl_2 (2 mL) and sat. aq. $NaHCO_3$ (2 mL). The layers were separated and the organic layer was extracted with brine (1 x 2 mL). The organic layer was dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude material was purified by preparative thin-layer chromatography (4 x 1000 μM) eluting with CH_2Cl_2/MeOH (20:1) to afford 16 mg (75%) of white solid. mp 152-154 °C; M+H = 495.

EXAMPLES 423-424:

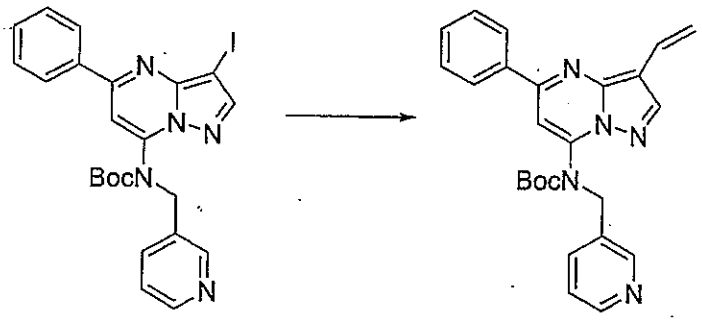
Utilizing the procedure outlined in Example 422, the amino compounds (Column 2) were converted to the corresponding methylsulfonamides (Column 3) in Table 34.

TABLE 34

Ex.	Column 2 (Amine)	Column 3 (Final Structure)	CMPD 1. mp. 2. M+H
423			1. mp = 166-168 2. M+H = 467
424			1. mp = 165-168 2. M+H = 467

EXAMPLE 425:

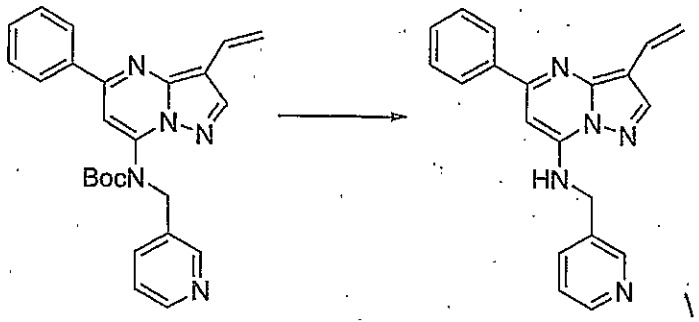
STEP A:



A mixture of the compound prepared in Preparative Example 194 (132 mg, 0.25 mmol), tributylvinyltin (95 mg, 0.30 mmol) and

tetrakis(triphenylphospine) palladium (29 mg, 0.025 mmol) in anhydrous dioxane (5 mL) was refluxed under N_2 for 24 hr. The solvent was evaporated and the residue was purified by flash chromatography using 2:1 CH_2Cl_2:EtOAc as eluent to yield yellow waxy solid (53 mg, 50%). LCMS: MH^+=428.

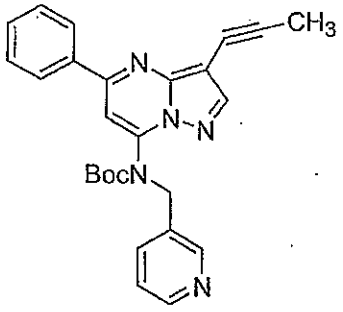
STEP B:



A mixture of the compound prepared in Example 425, Step A (50 mg, 0.12 mmol) and KOH (100 mg, 1.80 mmol) in ethanol (3 mL) and H_2O (0.6 mL) was stirred at 70°C under N_2 for 24 hr. $NaHCO_3$ (1.0 g), Na_2SO_4 (2.0g), and CH_2Cl_2 (20 mL) were added, the mixture was shaken and then filtered. The solvent was evaporated and the residue was purified by flash chromatography using 20:1:0.1 CH_2Cl_2:MeOH:conc.NH_4OH as eluent to yield yellow waxy solid (17 mg, 45%). LCMS: MH^+=328. Mp=48-51 °C.

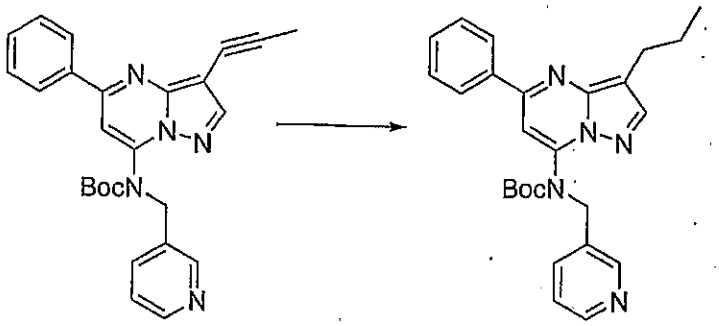
EXAMPLE 426:

STEP A:



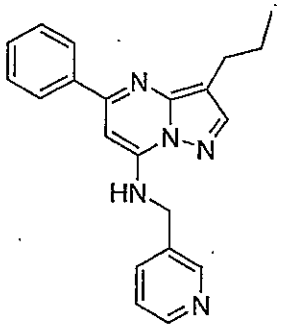
By essentially the same procedure set forth in Example 425, Step A only using tributylmethylethynyltin, the compound shown above was prepared.

STEP B:



A mixture of the compound prepared in Example 426, Step A (150 mg, 0.34 mmol) and PtO2 (30 mg, 0.13 mmol) in glacial acetic acid (5 mL) was stirred under 1 atmosphere of H_2 for 20 hr. The mixture was filtered, fresh PtO2 (30 mg, 0.13 mmol) was added and the mixture was stirred under 1 atmosphere of H_2 for 2.5 hr. The mixture was poured onto Na_2CO_3 (20 g) and H_2O (200 mL) and it was extracted with CH_2Cl_2 (4x20 mL). The combined extracts were dried over Na_2SO_4 and filtered. The solvent was evaporated and the residue was purified by flash chromatography using 1:1 CH_2Cl_2:EtOAc as eluent to yield yellow waxy solid (68 mg, 45%).

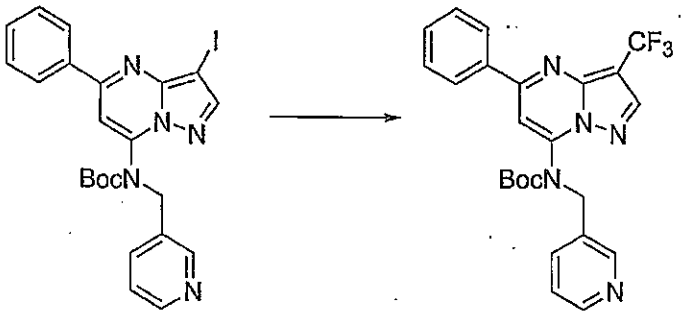
STEP C:



By essentially the same procedure set forth in Example 425, Step B only substituting the compound prepared in Example 426, Step B, the compound shown above was prepared, MS: MH^+=344. Mp=110-112 °C.

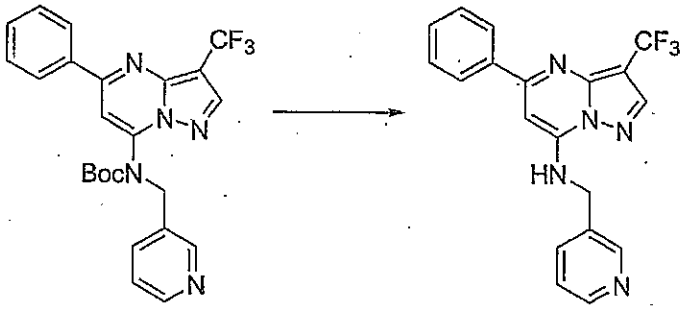
EXAMPLE 427:

STEP A:



A mixture of the compound prepared in Preparative Example 194 (527 mg, 1.00 mmol), triethyl(trifluoromethyl)silane (666 mg, 3.60 mmol), potassium fluoride (210 mg, 3.60 mmol), and CuI (850 mg, 4.46 mmol) in anhydrous DMF (4 mL) was stirred in a closed pressure vessel at 80°C for 72 hr. CH2Cl2 (80 mL) was added and the mixture was filtered through Celite. The solvent was evaporated and the residue was purified by flash chromatography using 2:1 CH_2Cl_2: EtOAc as eluent to yield pale orange waxy solid (70 mg, 15%). LCMS: M^+=470.

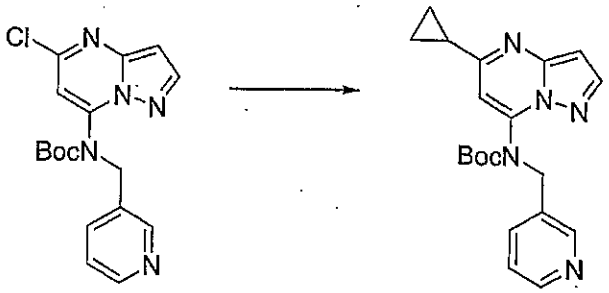
STEP B:



TFA (0.70 mL) was added at 0°C under N_2 to a stirred solution of the compound prepared in Example 427, Step A (70 mg, 0.15 mmol), in anhydrous CH_2Cl_2 (3 mL). The mixture was stirred at 0°C for 10 min, then at 25°C for 2 hr. It was poured into 10 % aqueous Na_2CO_3 (50 mL), extracted with CH_2Cl_2 (3x15 mL), dried over Na_2SO_4, and filtered. The solvent was evaporated and the residue was purified by flash chromatography using EtOAc as eluent to yield off-white solid (40 mg, 73%). LCMS: M^+=370. Mp=156-158 °C.

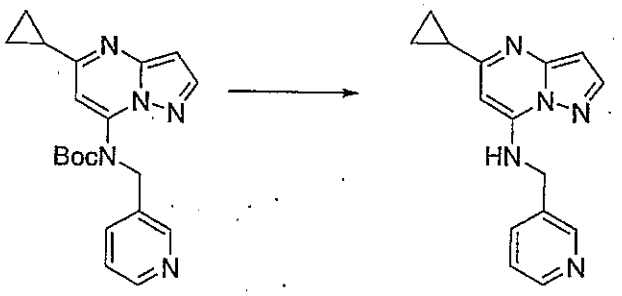
EXAMPLE 428:

STEP A:



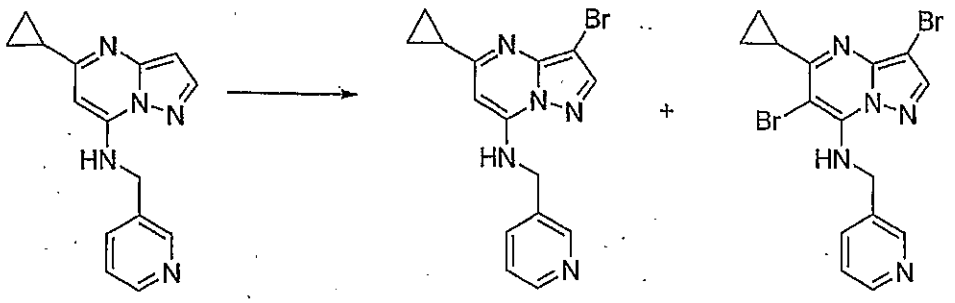
A mixture of the compound prepared in Preparative Example 193 (100 mg, 0.28 mmol), tetracyclopropylltin (91 mg, 0.32 mmol), Pd_2dba_3 (8.0 mg, 0.009 mmol) and $Pd(Pt\text{-}Bu_3)_2$ (9.0 mg, 0.017 mmol) in anhydrous dioxane (3 mL) was refluxed under N_2 for 27 hr. The solvent was evaporated and the residue was purified by flash chromatography using 1:1 CH_2Cl_2:EtOAc as eluent to yield colorless waxy solid (38 mg, 38%). LCMS: MH^+=366.

STEP B:



A mixture of the compound prepared in Example 428, Step A (36 mg, 0.10 mmol) and KOH (300 mg, 5.40 mmol) in ethanol (3 mL), 1,2-dimethoxyethane (3.0 mL0 and H_2O (0.8 mL) was refluxed under N_2 for 4 hr. It was poured into saturated aqueous $NaHCO_3$ (100 mL), extracted with CH_2Cl_2 (5x10 mL), dried over Na_2SO_4, and filtered. The solvent was evaporated and the residue was purified by flash chromatography using 30:1 EtOAc:MeOH as eluent to yield colorless waxy (18 mg, 69%). LCMS: MH^+=266.

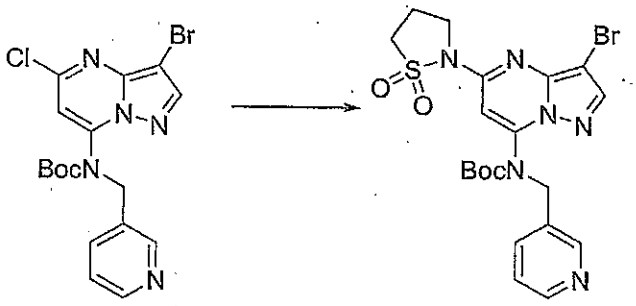
STEP C:



N-bromosuccinimide (12 mg, 0.068 mmol) in anhydrous CH_3CN (2 mL) was added under N_2 to a stirred solution of the compound prepared in Example 428, Step B (18 mg, 0.068 mmol), in anhydrous CH_3CN (2 mL). The mixture was stirred at 25°C for 2 hr. The solvent was evaporated and the residue was purified by flash chromatography using EtOAc as eluent to yield 5 mg (17%) of the dibromo compound (white solid, LCMS: MH^+=370, mp= 150-152°C) and 8 mg (34%) of the monobromo compound (colorless solid, LCMS: M^+=344, mp= 196-198°C).

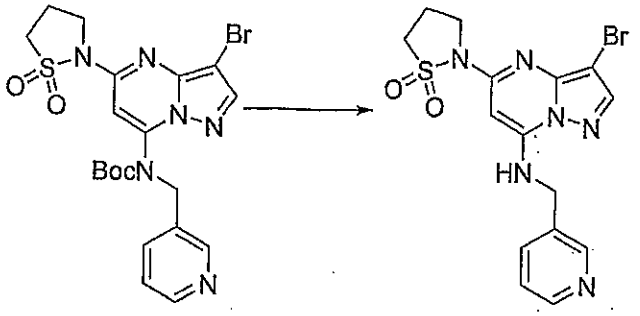
EXAMPLE 429:

STEP A:



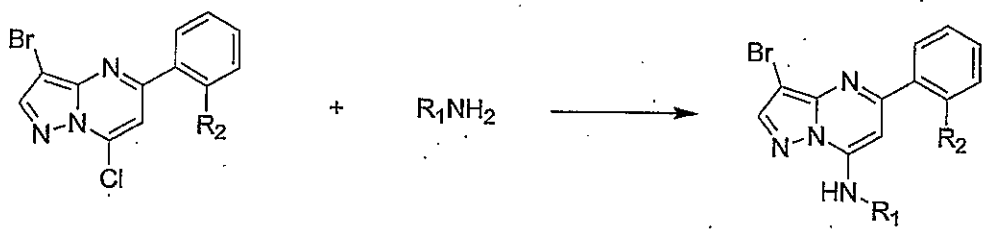
1,3-propanesultam (72 mg, 0.60 mmol) in anhydrous DMF (3 mL) was added under N_2 to 60 % NaH in mineral oil (36 mg, 0.90 mmol). The mixture was stirred for 20 min, then the compound prepared in Preparative Example 196 (200 mg, 0.46 mmol) was added. The mixture was stirred at 100°C for 30 min, the solvent was evaporated and the residue was purified by flash chromatography using EtOAc as eluent to yield colorless solid (150 mg, 63%). LCMS: M^+=523.

STEP B:



TFA (1.5 mL) was added at 0°C under N_2 to a stirred solution of the compound prepared in Preparative Example 196 (140 mg, 0.27 mmol), in anhydrous CH_2Cl_2 (5 mL). The mixture was stirred at 0°C for 10 min, then at 25°C for 2 hr. It was poured onto Na_2CO_3 (10 g), extracted with CH_2Cl_2 (3x50 mL), and filtered. The solvent was evaporated and the residue was purified by flash chromatography using 40:1 EtOAc:MeOH as eluent to yield white solid (32 mg, 28%). LCMS: M^+=423. Mp=218-220 °C.

EXAMPLE 430:

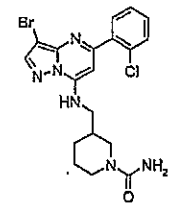
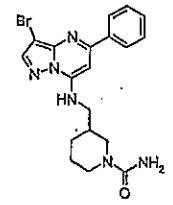
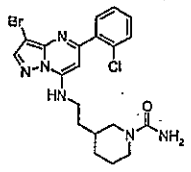
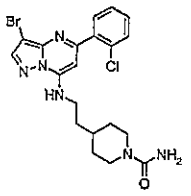
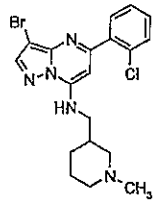


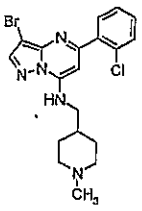
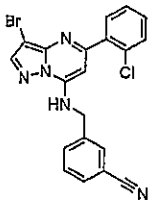
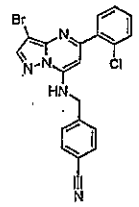
Where: R_2=H, or Cl

3-Bromo-7-chloro-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidine (1 equivalent) (prepared as described in Preparative Example 129), or 3-Bromo-7-chloro-5-phenylpyrazolo[1,5-a]pyrimidine (1 equivalent) (prepared as described in Preparative Example 127), R_1NH_2 (1.2 equivalents) and diisopropyl ethylamine (2 equivalents) were dissolved in anhydrous 1,4-dioxane and the mixture was heated at 75°C for the time given in Table 97. The solution was evaporated to dryness and the residue was chromatographed on a silica gel column as described in Table 97, to give the title compound.

Using the appropriate reactants and essentially the same procedure as described above, the products of Examples 431 to 438 were prepared. Variations in the reaction conditions are noted in Table 35.

TABLE 35

Ex.	Structure	MW	FABMS MH^+	Reaction Conditions	Yield	Chromatographic Data
431		463.8	463.0	75°C / 26h	52%	15x2.5cm 0.5-2% (10% Conc. ammonium hydroxide in methanol)-dichloromethane
432		429.3	429.2	75°C / 26h 25°C / 39h	53%	15x5cm Dichloromethane; 1.5% (10% Conc. ammonium hydroxide in methanol)-dichloromethane
433		477.8	477.1	75°C / 26h	48%	15x5cm Dichloromethane; 3.5-15% (10% Conc. ammonium hydroxide in methanol)-dichloromethane
434		477.8	477.0	75°C / 26h	50%	15x5cm Dichloromethane; 3.5-15% (10% Conc. ammonium hydroxide in methanol)-dichloromethane
435		434.8	434.1	75°C / 24h 25°C / 65h	53%	15x2.5cm 3% (10% Conc. ammonium hydroxide in methanol)-dichloromethane
436		434.8	434.2	75°C / 27h	31%	15x2.5cm 3% (10% Conc. ammonium hydroxide in methanol)-dichloromethane

						
437		438.7	438.1	75°C / 21h 25°C / 46h	97%	15x2.5cm 0.25% (10% Conc. ammonium hydroxide in methanol)-dichloromethane
438		438.7	438.1	75°C / 28h -20°C / 72h	95%	60x2.5cm 20% Ethyl acetate in hexane

Additional physical data for the compounds are given below:

EXAMPLE 431: Reactants: 3-Bromo-7-chloro-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidine (110mg, 0.318mmoles) (prepared as described in Preparative Example 129); 3-(aminomethyl)piperidine-1-carboxamide (60mg, 0.382mmoles) (prepared as described in Preparative Example 241 above); diisopropyl ethylamine (0.111mL, 0.636mmoles); anhydrous 1,4-dioxane (2.5mL). Physical properties: HRFABMS: m/z 463.0628 (MH^+). Calcd. for $C_{19}H_{21}N_6OBrCl$: m/z 463.0649: δ_H ($CDCl_3$) 1.38 (1H, m, CH_2), 1.52 (1H, m, CH_2), 1.73 (1H, m, CH), 1.93 (1H, m, CH_2), 2.02 (1H, m, CH_2), 2.98 (1H, m, CH_2), 3.06 (1H, m, CH_2), 3.37 (2H, m, CH_2), 3.58 (1H, m, CH_2), 3.82 (1H, m, CH_2), 4.87 (2H, bm, $CONH_2$), 6.28 (1H, s, H_6), 7.02 (1H, m, NH), 7.36 (2H, m, Ar-H), 7.45 (1H, m, Ar-H), 7.68 (1H, m, Ar-H) and 8.00 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_2: 23.7, 28.1, 44.6, 45.5, 47.2; CH: 35.2, 87.4, 127.2, 130.1, 130.3, 131.6, 143.9: C: 83.1, 132.1, 138.6, 145.5, 146.5, 158.0, 158.4.

EXAMPLE 432: Reactants: 3-Bromo-7-chloro-5-phenylpyrazolo[1,5-a]pyrimidine (500mg, 1.62mmoles) (prepared as described in Preparative Example 127); 3-(aminomethyl)piperidine-1-carboxamide (306mg, 1.944mmoles) (prepared as described in Preparative Example 241 above); diisopropyl ethylamine (0.566mL, 3.24mmoles); anhydrous 1,4-dioxane (13mL). Physical properties: HRFABMS: m/z 429.1031 (MH^+). Calcd. for $C_{19}H_{22}N_6OBr$: m/z 429.1038; δ_H ($CDCl_3$) 1.44 (1H, m, CH_2), 1.59 (1H, m, CH_2), 1.79 (1H, m, CH), 2.01 (1H, m, CH_2), 2.08 (1H, m, CH_2), 3.03 (1H, m, CH_2), 3.13 (1H, m, CH_2), 3.39 (1H, m, CH_2), 3.47 (1H, m, CH_2), 3.63 (1H, m, CH_2), 3.90 (1H, m, CH_2), 4.88 (2H, bm, $CONH_2$), 6.40 (1H, s, H_6), 6.90 (1H, m, NH), 7.53 (2H, m, Ar-H), 8.02 (1H, s, H_2) and 8.12 (1H, m, Ar-H); δ_C ($CDCl_3$) CH_2: 23.7, 28.2, 44.7, 45.5, 47.3; CH: 35.2, 82.9, 127.5, 127.5, 128.7, 128.7, 130.0, 143.9; C: 83.0, 138.5, 145.8, 147.1, 158.3, 158.5.

EXAMPLE 433: Reactants: 3-Bromo-7-chloro-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidine (347mg, 1.01mmoles) (prepared as described in Preparative Example 129); 3-(aminoethyl)piperidine-1-carboxamide (208mg, 1.21mmoles) (prepared as described in Preparative Example 242 above); diisopropyl ethylamine (0.393mL, 2.02mmoles); anhydrous 1,4-dioxane (9mL). Physical properties: δ_H ($CDCl_3$) 1.24 (1H, m, CH_2), 1.55 (1H, m, CH), 1.72 (4H, m, CH_2), 1.93 (1H, m, CH_2), 2.69 (1H, m, CH_2), 2.94 (1H, m, CH_2), 3.55 (2H, m, CH_2), 3.73 (1H, m, CH_2), 3.98 (1H, m, CH_2), 4.83 (2H, bm, $CONH_2$), 6.55 (1H, s, H_6), 6.78 (1H, m, NH), 7.41 (2H, m, Ar-H), 7.50 (1H, m, Ar-H), 7.75 (1H, m, Ar-H) and 8.04 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_2: 24.6, 30.7, 32.6, 39.9, 45.3, 49.3; CH: 33.3, 87.5, 127.4, 130.1, 130.2, 131.6, 143.8; C: 83.2, 132.1, 138.8, 145.7, 146.2, 158.1, 158.1.

EXAMPLE 434: Reactants: 3-Bromo-7-chloro-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidine (275mg, 0.803mmoles) (prepared as described in Preparative Example 129); 4-(aminoethyl)piperidine-1-carboxamide (165mg, 0.963mmoles) (prepared as described in Preparative Example 243 above); diisopropyl ethylamine (0.311mL, 0.963mmoles); anhydrous 1,4-dioxane (7.2mL). Physical properties: δ_H (d_6-DMSO) 1.00 (2H, m, CH_2), 1.50 (1H, m, CH), 1.59 (2H, m,

CH_2), 1.67 (2H, m, CH_2), 2.60 (2H, m, CH_2), 3.48 (2H, m, CH_2), 3.70 (2H, m, CH_2), 5.84 (2H, bs, $CONH_2$), 6.43 (1H, s, H_6), 7.50 (2H, m, Ar-H), 7.62 (2H, m, Ar-H), 8.30 (1H, s, H_2) and 8.36 ppm (1H, m, NH); δ_C (d_6-DMSO) CH_2: 31.5, 31.5, 34.8, 43.5, 43.5, 43.5; CH: 32.8, 86.8, 127.1, 129.7, 130.3, 131.0; 143.3; CH: 81.3, 131.0, 138.7, 145.1, 146.4, 157.3, 157.8.

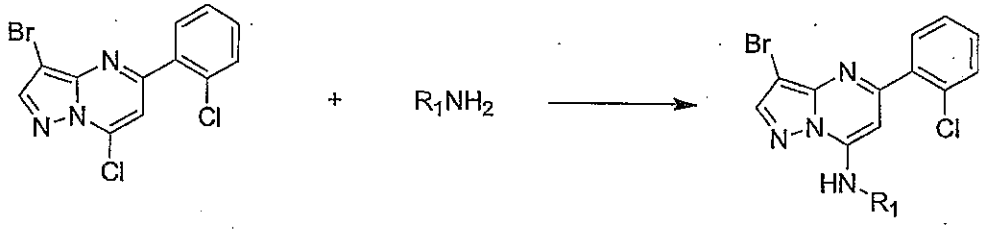
EXAMPLE 435: Reactants: 3-Bromo-7-chloro-5-phenylpyrazolo[1,5-a]pyrimidine (174mg, 0.507mmoles) (prepared as described in Preparative Example 129) and 3-(aminomethyl)-1-methylpiperidine (65mg, 0.507mmoles) (prepared as described in Preparative Example 244 above); diisopropyl ethylamine (0.178mL, 1.014mmoles); anhydrous 1,4-dioxane (2.5mL). Physical properties: HRFABMS: m/z 434.0742 (MH^+). Calcd. for $C_{19}H_{22}N_5BrCl$: m/z 434.0747; δ_H ($CDCl_3$) 1.18 (1H, m, CH_2), 1.68 (1H, m, CH_2), 1.80 (1H, m, CH_2), 1.87 (1H, m, CH_2), 1.96 (1H, m, CH), 2.14 (2H, m, CH_2), 2.32 (3H, s, NCH_3), 2.75 (1H, m, CH_2), 2.29 (1H, m, CH_2), 3.42 (2H, m, -NHC$\underline{H_2}$CH), 6.36 (1H, s, H_6), 6.64 (1H, bm, NH), 7.41 (2H, m, Ar-H), 7.51 (1H, m, Ar-H), 7.74 (1H, m, Ar-H) and 8.06 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_3: 46.6; CH_2: 24.4, 27.9, 46.1, 56.1, 59.6; CH: 36.0, 87.4, 127.1, 130.1, 130.2, 131.6, 143.8; C: 83.2, 132.1, 138.9, 145.6, 146.4, 158.2.

EXAMPLE 436: Reactants: 3-Bromo-7-chloro-5-phenylpyrazolo[1,5-a]pyrimidine (111.4mg, 0.325mmoles) (prepared as described in Preparative Example 129); 4-(aminomethyl)-1-methylpiperidine (50mg, 0.39mmoles) (prepared as described in Preparative Example 245 above); diisopropyl ethylamine (0.1135mL, 0.65mmoles); anhydrous 1,4-dioxane (1.5mL). Physical data: HRFABMS: m/z 434.0735 (MH^+). Calcd. for $C_{19}H_{22}N_5BrCl$: m/z 434.0747; δ_H ($CDCl_3$) 1.42 (2H, m, CH_2), 1.72 (1H, m, CH), 1.82 (2H, m, CH_2), 1.93 (2H, m, CH_2), 2.20 (3H, s, NCH_3), 2.89 (2H, m, CH_2), 3.34 (2H, m, -NHC$\underline{H_2}$CH), 6.31 (1H, s, H_6), 6.46 (1H, m, NH), 7.36 (2H, m, Ar-H), 7.46 (1H, m, Ar-H), 7.70 (1H, m, Ar-H) and 8.00 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_3: 46.4; CH_2: 30.2, 30.2, 48.0, 55.3, 55.3; CH: 35.4, 87.5, 127.2, 130.2, 130.2, 131.6, 143.8; C: 83.3, 132.2, 138.9, 145.7, 146.4, 158.1.

EXAMPLE 437: Reactants: 3-Bromo-7-chloro-5-phenylpyrazolo[1,5-a]pyrimidine (191mg, 0.557mmoles) (prepared as described in Preparative Example 129); 3-(aminomethyl)benzonitrile (88.3mg, 0.668mmoles) (prepared as described in Preparative Example 246 above); diisopropyl ethylamine (0.192mL, 1.114mmoles); anhydrous 1,4-dioxane (4.5mL). Physical data: HRFABMS: m/z 438.0125 (MH^+). Calcd. for $C_{19}H_{12}N_5BrCl$: m/z 438.0121; δ_H ($CDCl_3$) 4.76 (2H, d, -$C\underline{H}_2$NH-), 6.32 (1H, s, H_6), 7.00 (1H, m, -$CH_2N\underline{H}$-), 7.40 (2H, m, Ar-H), 7.46 (1H, m, Ar-H), 7.55 (1H, m, Ar-H), 7.67 (2H, m, Ar-H), 7.71 (1H, m, Ar-H), 7.75 (1H, m Ar-H) and 8.10 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_2: 45.5; CH: 88.2, 127.2, 130.0, 130.2, 130.4, 130.6, 131.4, 131.6, 131.9, 144.1; C: 83.8, 113.4, 118.3, 132.0, 137.8, 138.3, 145.6, 145.9, 158.0.

EXAMPLE 438: Reactants: 3-Bromo-7-chloro-5-phenylpyrazolo[1,5-a]pyrimidine (233.5mg, 0.681mmoles) (prepared as described in Preparative Example 129); 4-(aminomethyl)benzonitrile (108mg, 0.817mmoles) (prepared as described in Preparative Example 247 above); diisopropyl ethylamine (0.235mL, 1.362mmoles); anhydrous 1,4-dioxane (5.3mL). Physical data: HRFABMS: m/z 438.0117 (MH^+) Calcd. for $C_{20}H_{14}N_5BrCl$: m/z 438.0121; δ_H ($CDCl_3$) 4.80 (2H, d, CH_2), 6.30 (1H, s, H_6), 7.01 (1H, m, NH), 7.40 (2H, m, Ar-H), 7.47 (1H, m, Ar-H), 7.70 (2H, m, Ar-H), 7.72 (2H, m, Ar-H), 7.80 (1H, m, Ar-H) and 8.10 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_2: 45.8; CH: 88.2, 127.2, 127.7, 127.7, 130.2, 130.4, 131.6, 132.9, 132.9, 144.1; C: 83.8, 112.2, 118.4, 132.0, 138.2, 141.5, 145.5, 146.0, 158.0.

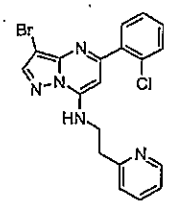
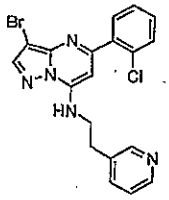
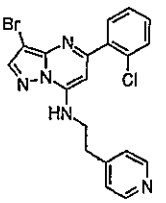
EXAMPLE 439:

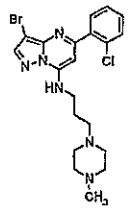
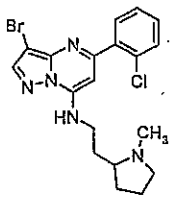
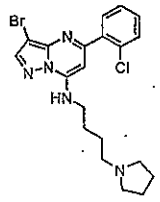
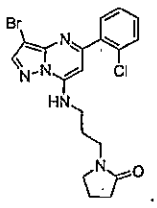
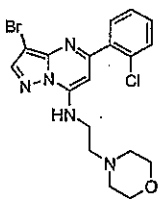


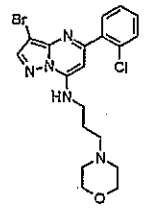
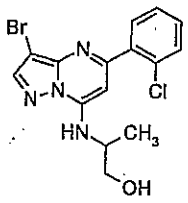
3-Bromo-7-chloro-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidine (50mg, 0.146mmoles) (prepared as described in Preparative Example 129) was dissolved in anhydrous 1,4-dioxane (5mL) in a GeneVac Technologies carousel reaction tube. PS-diisopropyl ethylamine resin (161mg, 0.5828mmoles) was

added to each tube. A freshly prepared 1M solution of the appropriate amine R_1NH_2 in anhydrous 1,4-dioxane (0.2185mL, 0.2185mmoles) was added to each tube and the tubes were sealed and heated at 70°C for 78h with magnetic stirring in the reaction block. Each tube was filtered and the resin was washed with anhydrous 1,4-dioxane and then dichloromethane. The combined individual filtrates from each tube were evaporated to dryness and the residues were each re-dissolved in anhydrous 1,4-dioxane (5mL) and placed in GeneVac reaction tubes. To each tube was added PS-isocyanate resin (594mg, 0.8742mmoles) and PS-trisamine resin (129mg, 0.4371mmoles) and the tubes were stirred at 25°C for 20h in the reaction block. The resins were filtered off and washed with anhydrous 1,4-dioxane and dichloromethane. The filtrates from each tube were evaporated to dryness and the residues were each chromatographed on a silica gel column using the column size and the eluant shown in Table 36, to give the title compounds.

TABLE 36

Ex.	Structure	MW	FABMS MH^+	Yield	Chromatographic Data
440		428.7	428.0	81%	15x2.5cm Dichloromethane; 0.5% Methanol in dichloromethane
441		428.7	428.0	48%	20x2cm Dichloromethane; 1.5% Methanol in dichloromethane
442		428.7	428.0	24%	15x2.5cm Dichloromethane; 1.5% Methanol in dichloromethane

443		463.8	463.0	44%	15x2.2cm Dichloromethane; 5% Methanol in dichloromethane
444		434.8	434.1	63%	15x2.5cm 5% Methanol in dichloromethane
445		448.8	448.2	65%	15x2.5cm 5% Methanol in dichloromethane
446		448.8	448.1	40%	15x2.5cm Dichloromethane; 0.5% Methanol in dichloromethane
447		436.7	436.1	72%	15x2.5cm 0.5% Methanol in dichloromethane

448		450.8	450.0	53%	20x2cm Dichloromethane; 0.5% Methanol in dichloromethane
449		381.7	381.0	44%	20x2cm 1.5% Methanol in dichloromethane

Additional physical data for the compounds are given below:

EXAMPLE 440: Physical properties: HRFABMS: m/z 428.0272 (MH^+). Calcd. for $C_{19}H_{16}N_5BrCl$: m/z 428.0278; δ_H ($CDCl_3$) 3.28 (2H, dd, $C_5H_4NC\underline{H}_2CH_2NH$-), 3.94 (2H, ddd, $C_5H_4NCH_2C\underline{H}_2NH$-), 6.40 (1H, s, H_6), 7.22-7.29 (3H, m, Ar-H), 7.38-7.44 (2H, m, Ar-H), 7.51 (1H, m, Ar-H), 7.68 (1H, ddd, Ar-H), 7.73 (1H, Ar-H), 8.18 (1H, s, H_2) and 8.68ppm (1H, NH); δ_C ($CDCl_3$) CH_2: 36.4, 41.5; CH: 87.3, 122.1, 123.6, 127.1, 130.1, 130.1, 131.6, 137.0, 143.8, 149.5; C: 83.1, 132.1, 138.9, 145.7, 146.3, 158.0, 158.1.

EXAMPLE 441: Physical properties: HRFABMS: m/z 428.0272 (MH^+). Calcd. for $C_{19}H_{16}N_5BrCl$: m/z 428.0278; δ_H ($CDCl_3$) 3.12 (2H, dd, $C_5H_4NC\underline{H}_2CH_2NH$-), 3.77 (2H, ddd, $C_5H_4NCH_2C\underline{H}_2NH$-), 6.40 (1H, s, H_6), 6.59 (1H, m, Ar-H), 7.34 (1H, bm, Ar-H), 7.39-7.45 (2H, m, Ar-H), 7.52 (1H, m, Ar-H), 7.62 (1H, m, Ar-H), 7.75 (1H, m, Ar-H), 8.05 (1H, s, H_2) and 8.63ppm (1H, m, NH); δ_C ($CDCl_3$) CH_2: 32.7, 43.1; CH: 87.5, 127.2, 130.2, 130.3, 131.6, 136.4, 142.9, 148.3, 149.8; C: 83.5, 132.0, 138.6, 145.6, 145.9, 158.1.

EXAMPLE 442: Physical properties: HRFABMS: m/z 428.0275 (MH^+). Calcd. for $C_{19}H_{16}N_5BrCl$: m/z 428.0278; δ_H ($CDCl_3$) 3.13 (2H, dd, $C_5H_4NC\underline{H}_2CH_2NH$-), 3.80 (2H, ddd, $C_5H_4NCH_2C\underline{H}_2NH$-), 6.42 (1H, s, H_6), 6.53 (1H, m, Ar-H), 7.23 (2H, m, Ar-H), 7.40-7.46 (2H, m, Ar-H), 7.62 (1H, m, Ar-H), 7.76 (1H, m, Ar-H), 8.07 (1H,

s, H_2) and 8.63ppm (1H, m, NH); δ_C ($CDCl_3$) CH_2: 34.7, 42.5; CH: 87.4, 124.5, 124.5, 127.2, 130.2, 130.3, 131.6, 144.0, 150.2, 150.2; C: 83.5, 132.0, 138.6, 145.6, 145.9, 146.6, 158.1.

EXAMPLE 443: Physical properties: HRFABMS: m/z 463.1003 (MH^+). Calcd. for $C_{20}H_{25}N_6BrCl$: m/z 463.1013; δ_H ($CDCl_3$) 1.98 (2H, m, $=NCH_2C\underline{H_2}CH_2NH$-), 2.43 (3H, s, NCH_3), 2.67 (2H, m, $=NC\underline{H_2}CH_2CH_2NH$-), 2.70 (8H, piperazine CH_2), 3.58 (2H, m, $=NCH_2CH_2C\underline{H_2}NH$-), 6.32 (1H, s, H_6), 7.37-7.43 (2H, m, Ar-H), 7.50 (1H, m, Ar-H), 7.73 (1H, m, Ar-H), 8.06 (1H, s, H_2) and 8.60ppm (1H, m, NH); δ_C ($CDCl_3$) CH_3: 46.1; CH_2: 24.1, 42.8, 53.3, 54.6, 54.6, 57.5, 57.5; CH: 87.1, 127.0, 130.0, 130.1, 131.5, 143.4; C: 82.7, 132.1, 139.2, 145.7, 146.7, 158.0.

EXAMPLE 444: Physical properties: HRFABMS: m/z 434.0742 (MH^+). Calcd. for $C_{19}H_{22}N_5BrCl$: m/z 434.0747; δ_H ($CDCl_3$) 1.72 (1H, m, CH/CH_2), 1.78-1.90 (2H, m, CH/CH_2), 2.02 (3H, m, CH/CH_2), 2.50 (1H, m, CH/CH_2), 2.45 (3H, s, NCH_3), 2.51 (1H, m, CH/CH_2), 3.23 (1H, m, CH/CH_2), 3.54 (1H, m, CH/CH_2), 3.60 (1H, m, CH/CH_2), 6.32 (1H, s, H_6), 7.38-7.44 (2H, m, Ar-H), 7.51 (1H, m, Ar-H), 7.75 (1H, m, Ar-H), 7.96 (1H, bm, NH) and 8.05 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_3: 40.7; CH_2: 22.7, 29.3, 30.1, 39.4, 57.0; CH: 64.2, 87.1, 127.1, 130.0, 130.1, 131.6, 143.8; C: 82.8, 132.1, 139.1, 145.7, 146.4, 158.0.

EXAMPLE 445: Physical properties: HRFABMS: m/z 448.0910 (MH^+). Calcd. for $C_{20}H_{24}N_5BrCl$: m/z 448.0904; δ_H ($CDCl_3$) 1.90 (4H, m, CH_2), 2.00 (4H, m, CH_2), 2.84 (2H, m, CH_2), 2.95 (4H, m, CH_2), 3.51 (2H, m, CH_2), 6.32 (1H, s, H_6), 7.05 (1H, bm, NH), 7.37-7.43 (2H, m, Ar-H), 7.50 (1H, m, Ar-H), 7.73 (1H, m, Ar-H) and 8.04 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_2: 23.4, 23.4, 24.8, 26.4, 41.8, 53.9, 53.9, 55.2; CH: 87.3, 127.1, 130.1, 130.2, 131.6, 143.7; C: 83.0, 132.0, 138.9, 145.7, 146.3, 158.1.

EXAMPLE 446: Physical properties: HRFABMS: m/z 448.0548 (MH^+). Calcd. for $C_{19}H_{20}N_5OBrCl$: m/z 448.0540; δ_H ($CDCl_3$) 1.94 (2H, m, CH_2), 2.09 (2H, m, CH_2), 2.49 (2H, m, CH_2), 3.45 (2H, m, CH_2), 3.51 (4H, m, CH_2), 6.32 (1H, s, H_6), 7.37-7.44 (3H, m, Ar-H/NH), 7.51 (1H, m, Ar-H), 7.75 (1H, m, Ar-H) and 8.10 ppm

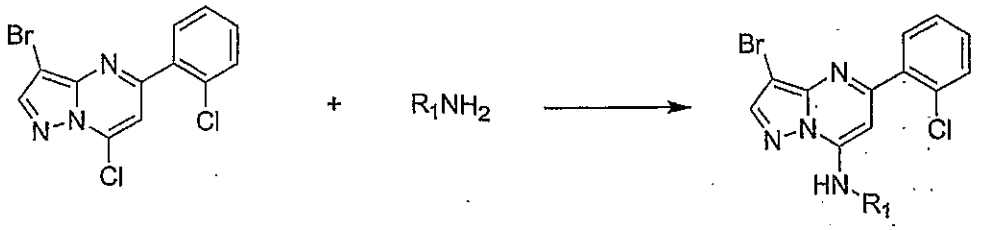
(1H, s, H_2); δ_C ($CDCl_3$) CH_2: 18.0, 26.3, 30.8, 39.2, 39.9, 47.5; CH: 87.0, 127.1, 130.1, 130.1, 131.6, 144.1; C: 82.9, 132.1, 138.9, 145.6, 146.2, 157.9, 176.2.

EXAMPLE 447: Physical properties: HRFABMS: m/z 436.0532 (MH^+). Calcd. for $C_{18}H_{20}N_5OBrCl$: m/z 436.0540; δ_H ($CDCl_3$) 2.60 (4H, bm, -N($C\underline{H}_2CH_2)_2O$), 2.83 (2H, m, =$NC\underline{H}_2CH_2NH$-), 3.57 (2H, m, =$NCH_2C\underline{H}_2NH$-), 3.83 (4H, m, -$N(CH_2C\underline{H}_2)_2O$), 6.37 (1H, s, H_6), 6.99 (1H, bm, NH), 7.38-7.45 (2H, m, Ar-H), 7.51 (1H, m, Ar-H), 7.75 (1H, m, Ar-H) and 8.09 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_2: 38.2, 53.3, 53.3, 56.2, 66.9, 66.9; CH: 87.6, 127.1, 130.1, 130.2, 131.6, 143.9; C; 83.1, 132.1, 138.9, 145.7, 146.2, 158.1.

EXAMPLE 448: Physical properties: HRFABMS: m/z 450.0688 (MH^+). Calcd. for $C_{19}H_{22}N_5OBrCl$: m/z 450.0696; δ_H ($CDCl_3$) 1.98 (2H, m, =$NCH_2C\underline{H}_2CH_2NH$-), 2.58 (4H, m, -$N(C\underline{H}_2CH_2)_2O$), 2.67 (2H,m, =$NC\underline{H}_2CH_2CH_2NH$-), 3.59 (2H, m, =$NCH_2CH_2C\underline{H}_2NH$-), 3.94 (4H, m, -$N(CH_2C\underline{H}_2)_2O$), 6.31 (1H, s, H_6), 7.37-7.44 (2H, Ar-H), 7.51 (1H, m, Ar-H), 7,78 (1H, m, Ar-H), 8.08 (1H, s, H_2) and 8.60 ppm (1H, bm, NH); δ_C ($CDCl_3$) CH_2: 23.7, 42.7, 52.9, 52.9, 58.0, 66.6, 66.6; CH: 87.0, 127.1, 130.0, 130.1, 131.5, 143.6; C: 82.8, 132.1, 139.1, 145.7, 146.7, 158.0.

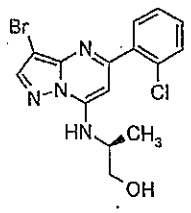
EXAMPLE 449: Physical properties: HRFABMS: m/z 381.0114 (MH^+). Calcd. for $C_{15}H_{15}N_4OBrCl$: m/z 381.0118; δ_H ($CDCl_3$) 1.39 (3H, d, $CHC\underline{H}_3$), 2.76 (1H, bm, -OH), 3.71 (1H, m, =$CHC\underline{H}_2OH$), 3.81 (1H, m, =$CHC\underline{H}_2OH$), 3.88 (1H, m, =$C\underline{H}CH_2OH$), 6.38 (1H, s, H_6), 7.38 (2H, m, Ar-H), 7.48 (1H, m, Ar-H), 7.68 (1H, m, Ar-H) and 8.02 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_3: 16.9; CH_2: 65.0; CH: 50.0, 88.0, 127.1, 130.1, 130.3, 131.4, 143.8; C: 83.0, 132.0, 138.5, 145.6, 146.0, 158.2.

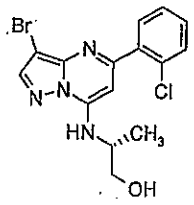
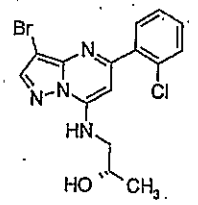
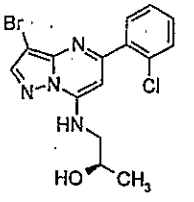
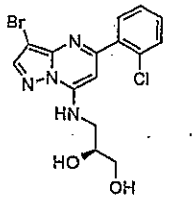
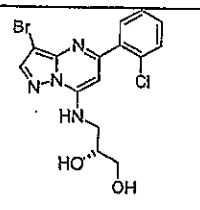
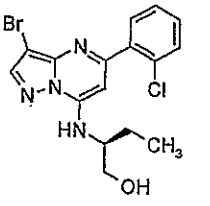
EXAMPLE 450:

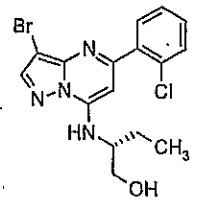
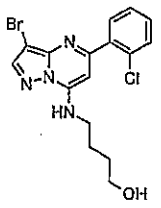


3-Bromo-7-chloro-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidine (50mg, 0.146mmoles) (prepared as described in Preparative Example 129) was dissolved in anhydrous 1,4-dioxane (5mL) in a GeneVac Technologies carousel reaction tube. PS-diisopropyl ethylamine resin (161mg, 0.5828mmoles) was added to each tube. A freshly prepared solution of the appropriate amine R_1NH_2 (0.219mmoles) in anhydrous 1,4-dioxane (0.3mL) was added to each tube, with the exception of Example 99-5 in which the amine was dissolved in 10% MeOH in 1,4-dioxane (0.3mL), and the tubes were sealed and heated at 70°C for 74h with magnetic stirring in the reaction block. Each tube was filtered and the resin was washed with anhydrous 1,4-dioxane and then dichloromethane. The combined individual filtrates from each tube were evaporated to dryness and the residues were each re-dissolved in anhydrous 1,4-dioxane (5mL) and placed in GeneVac reaction tubes. To each tube was added PS-isocyanate resin (594mg, 0.8742mmoles) and PS-trisamine resin (129mg, 0.4371mmoles) and the tubes were stirred at 25°C for 20h in the reaction block. The resins were filtered off and washed with anhydrous 1,4-dioxane and dichloromethane. The filtrates from each tube were evaporated to dryness and the residues were each chromatographed on a silica gel column using the column size and the eluant shown in Table 37, to give the title compounds.

TABLE 37

Ex.	Structure	MW	FABMS MH^+	Yield	Chromatographic Data
451		381.7	380.9	66%	15x2.5cm; 0.5% Methanol in dichloromethane

452		381.7	380.9	60%	20x2cm; 0.5% Methanol in dichloromethane
453		381.7	380.9	69%	15x2.5cm; 0.35% Methanol in dichloromethane
454		381.7	380.9	75%	15x2.5cm; 0.35% Methanol in dichloromethane
455		397.7	397.2	84%	15x2.5cm; 1.5% Methanol in dichloromethane
456		397.7			
457		395.7	395.0	60%	15x2.5cm; 0.35% Methanol in dichloromethane

458		395.7	396.3	50%	15x2.5cm; 0.35% Methanol in dichloromethane
459		395.7	396.0	76%	15x2.5cm; 0.35% Methanol in dichloromethane

Additional physical data for the compounds are given below:

EXAMPLE 451: Physical properties: HRFABMS: m/z 381.0115 (MH^+). Calcd. for $C_{15}H_{15}N_4OBrCl$: m/z 381.0118; $[\alpha]_D^{25°C}$ +1.4° (c=0.25 , MeOH); δ_H ($CDCl_3$) 1.44 (3H, d, -CHC<u>H</u>$_3$), 3.77 3.89 (1H, dd, CHC<u>H</u>$_2$OH), (1H, dd, CHC<u>H</u>$_2$OH), 3.94 (1H, m, C<u>H</u>CH$_2$OH), 6.41 (1H, s, H_6), 6.58 (1H, d, NH), 7.41 (2H, m, Ar-H), 7.51 (1H, m, Ar-H), 7.74 (1H, m, Ar-H) and 8.04 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_3: 17.1; CH_2: 65.5; CH: 49.9, 88.0, 127.1, 130.1, 130.2, 131.6, 143.8; C: 83.2, 132.1, 138.7, 145.6, 145.8, 158.1.

EXAMPLE 452: Physical properties: HRFABMS: m/z 381.0115 (MH^+). Calcd. for $C_{15}H_{15}N_4OBrCl$: m/z 381.0118; $[\alpha]_D^{25°C}$ +6.5° (c=0.32 , MeOH); δ_H ($CDCl_3$) 1.44 (3H, d, -CHC<u>H</u>$_3$), 3.78 (1H, dd, CHC<u>H</u>$_2$OH), 3.89 (1H, dd, CHC<u>H</u>$_2$OH), 3.96 (1H, m, C<u>H</u>CH$_2$OH), 6.41 (1H, s, H_6), 6.58 (1H, d, NH), 7.41 (2H, m, Ar-H), 7.51 (1H, m, Ar-H), 7.75 (1H, m, Ar-H) and 8.04 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_3: 17.1; CH_2: 65.5; CH: 49.9, 88.0, 127.1, 130.1, 130.3, 131.6, 143.8; C: 83.2, 132.1, 138.6, 145.6, 145.8, 158.1.

EXAMPLE 453: Physical properties: HRFABMS: m/z 381.0115 (MH^+). Calcd. for $C_{15}H_{15}N_4OBrCl$: m/z 381.0118; $[\alpha]_D^{25°C}$ +9.4° (c=0.27, MeOH); δ_H ($CDCl_3$) 1.33 (3H, d, CH_3), 2.25 (1H, bs, OH), 3.37 (1H, dd, CH_2), 3.51 (1H, m, CH_2), 4.16 (1H, m, C<u>H</u>OH), 6.35 (1H, s, H_6), 6.93 (1H, m, NH), 7.40 (2H, m, Ar-H), 7.50 (1H, m,

Ar-H), 7.70 (1H, m, Ar-H) and 8.04 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_3: 20.8; CH_2: 49.2; CH: 65.7, 87.8, 127.1, 130.1, 130.2, 131.2, 143.9; C: 83.1, 132.1, 138.5, 145.6, 146.6, 158.3.

EXAMPLE 454: Physical properties: HRFABMS: m/z 381.0112 (MH^+). Calcd. for $C_{15}H_{15}N_4OBrCl$: m/z 381.0118; $[\alpha]_D^{25°C}$ -3.2° (c=0.29 ; MeOH); δ_H ($CDCl_3$) 1.32 (3H, d, CH_3), 2.48 (1H, bs, OH), 3.35 (1H, dd, CH_2), 3.49 (1H, m, CH_2), 4.15 (1H, m, C<u>H</u>OH), 6.34 (1H, s, H_6), 6.93 (1H, m, NH), 7.39 (2H, m, Ar-H), 7.49 (1H, m, Ar-H), 7.68 (1H, m, Ar-H) and 8.03 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_3: 20.8; CH_2: 49.2; CH: 65.7, 87.7, 127.1, 130.1, 130.3, 131.4, 143.9; C: 83.0, 132.0, 138.6, 145.6, 146.6, 158.3.

EXAMPLE 455: Physical properties: HRFABMS: m/z 397.0054 (MH^+). Calcd. for $C_{15}H_{15}N_4O_2BrCl$: m/z 397.0067; $[\alpha]_D^{25°C}$ -9.5° (c= 0.28, MeOH); δ_H ($CDCl_3$) 3.18 (2H, bs, OH), 3.47 (1H, dd, CH_2), 3.58 (1H, dd, CH_2), 3.63 (1H, dd, C<u>H</u>$_2$OH), 3.70 (1H, dd, C<u>H</u>$_2$OH), 3.98 (1H, m, CH), 6.35 (1H, s, H_6), 7.10 (1H, m, NH), 7.37 (2H, m, Ar-H), 7.46 (1H, m, Ar-H), 7.64 (1H, m, Ar-H) and 8.01 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_2: 44.7, 64.0; CH: 69.7, 87.7, 127.0, 130.1, 130.3, 131.3, 143.9; C: 82.9, 132.0, 138.4, 145.4, 146.7, 158.3.

EXAMPLE 456: This enantiomer may be prepared by essentially the same manner as described above.

EXAMPLE 457: Physical properties: HRFABMS: m/z 395.0260 (MH^+). Calcd. for $C_{16}H_{17}N_4OBrCl$: m/z 395.0274; $[\alpha]_D^{25°C}$ -34.3° (c= 0.28, MeOH); δ_H ($CDCl_3$) 1.08 (3H, dd, CH_3), 1.78 (1H, m, CH_2), 1.86 (1H, m, CH_2), 2.35 (1H, bs, CH_2O<u>H</u>), 3.71 (1H, m, C<u>H</u>NH), 3.81 (1H, dd, C<u>H</u>$_2$OH), 3.90 (1H, dd, C<u>H</u>$_2$OH), 6.42 (1H, s, H_6), 6.53 (1H, m, NH), 7.41 (2H, m, Ar-H), 7.51 (1H, Ar-H), 7.75 (1H, m, Ar-H) and 8.04 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_3: 10.5; CH_2: 24.5, 63.7; CH: 55.9, 88.0, 127.1, 130.1, 130.2, 131.6, 143.8; C: 83.2, 132.1, 138.6, 145.6, 146.3, 158.1.

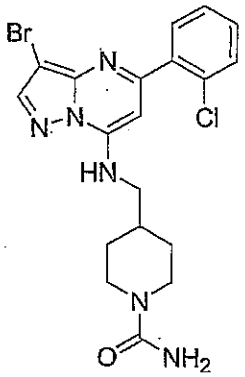
EXAMPLE 458: Physical properties: HRFABMS: m/z 395.0274 (MH^+). Calcd. for $C_{16}H_{17}N_4OBrCl$: m/z 395.0274; $[\alpha]_D^{25°C}$ +27.5° (c= 0.25, MeOH); δ_H ($CDCl_3$) 1.05

(3H, dd, CH_3), 1.76 (1H, m, CH_2), 1.85 (1H, m, CH_2), 2.28 (1H, bs, $CH_2O\underline{H}$), 3.67 (1H, m, $C\underline{H}NH$), 3.77 (1H, dd, $C\underline{H_2}OH$), 3.84 (1H, dd, $C\underline{H_2}OH$), 6.49 (1H, s, H_6), 6.66 (1H, m, NH), 7.39 (2H, m, Ar-H), 7.49 (1H, Ar-H), 7.71 (1H, m, Ar-H) and 8.04 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_3: 10.5; CH_2: 24.3, 63.3; CH: 56.1, 88.0, 127.1, 130.1, 130.3, 131.5, 143.8; C: 83.0, 132.1, 138.6, 145.6, 146.3, 158.2.

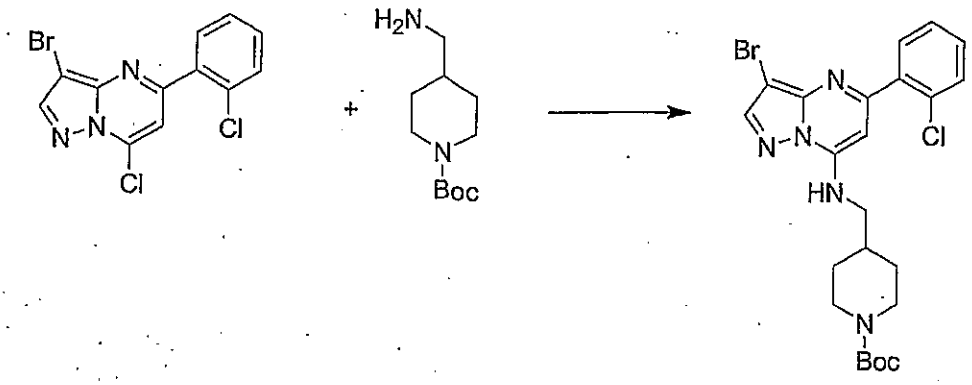
EXAMPLE 459: Physical properties: HRFABMS: m/z 395.0264 (MH^+). Calcd. for $C_{16}H_{17}N_4OBrCl$: m/z 395.0274; δ_H ($CDCl_3$) 1.77 (2H, m, -$NHCH_2CH_2C\underline{H_2}CH_2OH$), 1.90 (1H, bm, -$NHCH_2CH_2CH_2CH_2O\underline{H}$), 1.93 (2H, m, -$NHCH_2C\underline{H_2}CH_2CH_2OH$), 3.54 (2H, m, -$NHC\underline{H_2}CH_2CH_2CH_2OH$), 3.77 (2H, m, -$NHCH_2CH_2CH_2C\underline{H_2}OH$), 6.37 (1H, s, H_6), 6.72 (1H, m, -$N\underline{H}CH_2CH_2CH_2CH_2OH$), 7.41 (2H, m, Ar-H), 7.51 (1H, m, Ar-H), 7.75 (1H, m, Ar-H) and 8.06 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_2: 25.7, 29.7, 42.2, 62.2; CH: 87.4, 127.1, 130.1, 130.2, 131.6, 143.8; C: 83.1, 132.1, 138.8, 145.6, 146.3, 158.1.

EXAMPLE 460:

4-{[3-BROMO-5-(2-CHLOROPHENYL)PYRAZOLO[1,5-a]PYRIMIDIN-7-YLAMINO]METHYL}PIPERIDINE-1-CARBOXYLIC ACID AMIDE:

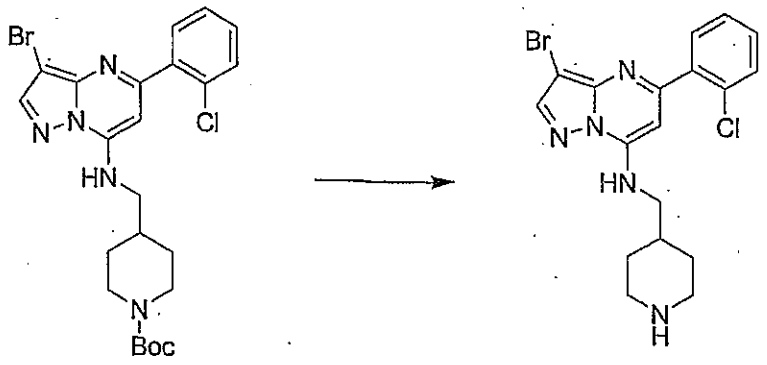


A. 4-{[3-BROMO-5-(2-CHLOROPHENYL)PYRAZOLO[1,5-a]PYRIMIDIN-7-YLAMINO]METHYL}PIPERIDINE-1-CARBOXYLIC ACID *tert*-BUTYL ESTER:



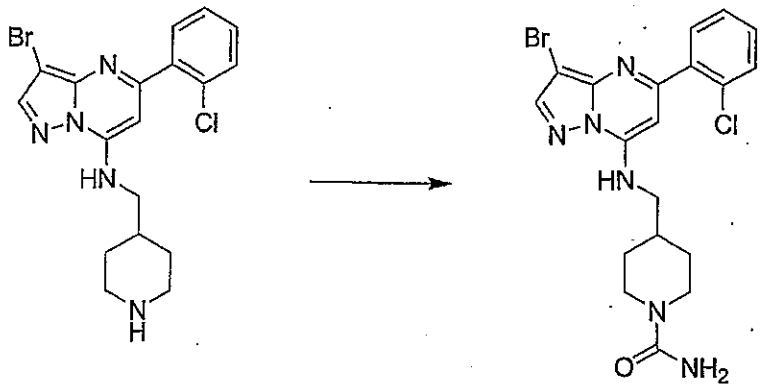
3-Bromo-7-chloro-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidine (300mg, 0.875mmoles) (prepared as described in Preparative Example 129) was dissolved in anhydrous 1,4-dioxane (6.8mL). 4-(aminomethyl)piperidine-1-carboxylic acid *tert*-butyl ester (225mg, 1.05mmoles) and diisopropyl ethylamine (0.3055mL, 1.75mmoles) were added and the mixture was heated at 75°C for 24h. The solution was evaporated to dryness and the residue was chromatographed on a silica gel column (15x5cm) using dichloromethane as the eluant to give 4-{[3-bromo-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidin-7-ylamino]methyl}piperidine-1-carboxylic acid *tert*-butyl ester (461.2mg, 100%): FABMS: m/z 520.1 (MH^+); HRFABMS: m/z 520.1111 (MH^+). Calcd. for $C_{23}H_{28}N_5O_2BrCl$: m/z 520.1115; δ_H ($CDCl_3$) 1.30 (2H, m, CH_2), 1.51 (9H, s, -$COOC(CH_3)_3$), 1.85 (2H, d, CH_2), 1.95 (1H, m , CH), 2.76 (2H, m, CH_2), 3.40 (2H, m, CH_2), 6.37 (1H, s, H_6), 6.55 (1H, m, NH), 7.42 (2H, m, Ar-H), 7.52 (1H, m, Ar-H), 7.76 (1H, m, Ar-H) and 8.07 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_3: 28.5, 28.5, 28.5; CH_2: 29.1, 29.1, 43.5, 43.5, 47.9; CH: 36.3, 87.5, 127.2, 130.2, 130.3, 131.6, 143.9; C: 79.7, 83.3, 132.1, 138.6, 145.4, 146.3, 154.7, 158.1.

B. [3-BROMO-5-(2-CHLOROPHENYL)PYRAZOLO[1,5-a]PYRIMIDIN-7-YL]PIPERIDIN-4-YLMETHYLAMINE:



4-{[3-Bromo-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidin-7-ylamino]methyl}piperidine-1-carboxylic acid *tert*-butyl ester (441mg, 0.847mmoles) (prepared as described in Example 460, Step A above) was dissolved in methanol (4.5mL) and 10% (v/v) conc. sulfuric acid in 1,4-dioxane (11.46mL) was added. The mixture was stirred at 25°C for 0.5h. The product was worked up as described in Preparative Example 241, step B and chromatographed on a silica gel column (15x5cm) using 8% (10% conc. ammonium hydroxide in methanol)-dichloromethane as the eluant to give [3-bromo-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidin-7-yl]piperidin-4-ylmethylamine (314.4mg, 88%): FABMS: m/z 420.0 (MH^+); HRFABMS: m/z 420.0585 (MH^+). Calcd. for $C_{18}H_{20}N_5BrCl$: m/z 420.0591; δ_H ($CDCl_3$) 1.34 (2H, m, CH_2), 1.86 (2H, m, CH_2), 1.91 (1H, m, CH), 2.10 (1H, bm, piperidine-NH), 2.67 (2H, m, CH_2), 3.18 (2H, m, CH_2), 3.38 (2H, m, CH_2), 6.37 (1H, s, H_6), 6.53 (1H, m, NH), 7.42 (2H, m, Ar-H), 7.52 (1H, m, Ar-H), 7.76 (1H, m, Ar-H) and 8.06 ppm (1H, s Ar-H); δ_C ($CDCl_3$) CH_2: 31.2, 31.2, 46.2, 46.2, 48.4; CH: 36.4, 89.5, 127.1, 130.1, 130.5, 131.6, 143.8; C: 83.2, 132.1, 138.9, 145.6, 146.4, 158.1.

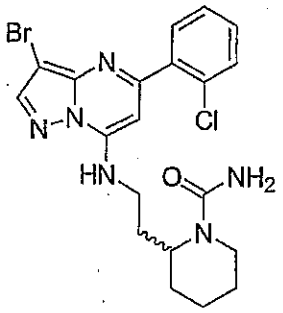
C. 4-{[3-BROMO-5-(2-CHLOROPHENYL)PYRAZOLO[1,5-a]PYRIMIDIN-7-YLAMINO]METHYL}PIPERIDINE-1-CARBOXYLIC ACID AMIDE:



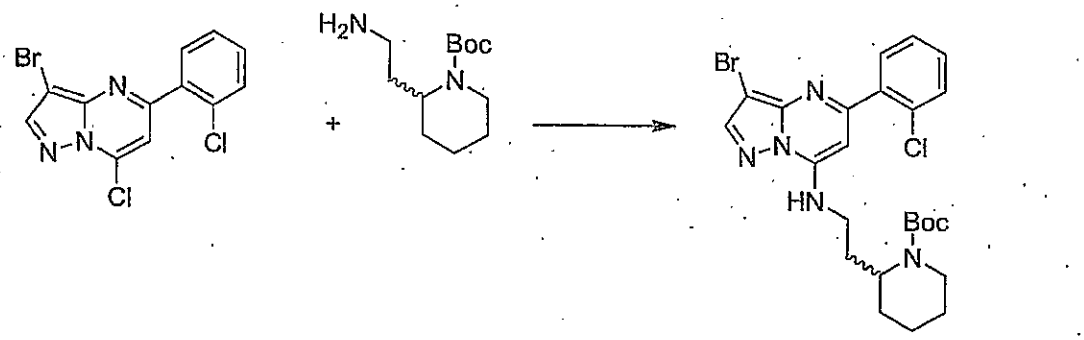
[3-Bromo-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidin-7-yl]piperidin-4-ylmethylamine (57mg, 0.136mmoles) (prepared as described in Example 460, Step B above) was dissolved in anhydrous dichloromethane (1.2mL) and trimethylsilylisocyanate (0.091mL, 0.679mmoles) was added. The mixture was stirred at 25°C for 2.5h. The mixture was diluted with dichloromethane and washed with saturated aqueous sodium bicarbonate. The organic layer was dried ($MgSO_4$), filtered and evaporated to dryness. The residue was chromatographed on a silica gel column (30x2.5cm) using 3% (10% conc. ammonium hydroxide in methanol)-dichloromethane as the eluant to give 4-{[3-bromo-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidin-7-ylamino]methyl}piperidine-1-carboxylic acid amide (53.7mg, 86%): FABMS: m/z 463.1 (MH^+); HRFABMS: m/z 463.0647 (MH^+). Calcd. for $C_{19}H_{21}N_6OBrCl$: m/z 463.0649; δ_H (d_6-DMSO) 1.09 (2H, m, CH_2), 1.63 (2H, m, CH_2), 1.87 (1H, m, CH), 2.60 (2H, m, CH_2), 3.53 (2H, bm, $CONH_2$), 3.91 (2H, d, CH_2), 6.52 (1H, s, H_6), 7.50 (2H, m, Ar-H), 7.62 (2H, m, Ar-H), 8.33 (1H, s, H_2) and 8.52 ppm (1H, m, NH); δ_C (d_6-DMSO) CH_2: 30.1, 30.1, 44.2, 44.2, 47.7; CH: 36.4, 88.2, 128.1, 130.7, 131.4, 132.1, 147.9; C: 82.1, 132.1, 139.4, 145.7, 147.9, 158.1, 158.8.

EXAMPLE 461:

2-{2-[3-BROMO-5-(2-CHLOROPHENYL)PYRAZOLO[1,5-a]PYRIMIDIN-7-YLAMINO]ETHYL}PIPERIDINE-1-CARBOXYLIC ACID AMIDE:



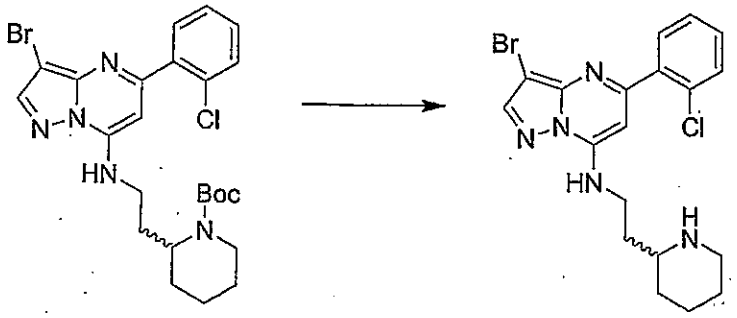
A. 2-{2-[3-BROMO-5-(2-CHLOROPHENYL)PYRAZOLO[1,5-a]PYRIMIDIN-7-YLAMINO]ETHYL}PIPERIDINE-1-CARBOXYLIC ACID *tert*-BUTYL ESTER:



3-Bromo-7-chloro-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidine (400mg, 1.166mmoles) (prepared as described in Preparative Example 129) was dissolved in anhydrous 1,4-dioxane (5.7mL). 2- Aminoethylpiperidine-1-carboxylic acid *tert*-butyl ester (266mg, 1.166mmoles) and diisopropyl ethylamine (0.409mL, 2.33mmoles) were added and the mixture was heated at 75°C for 48h. Additional diisopropyl ethylamine (0.204mL, 1.166mmoles) was added and the heating was continued for a total of 58h. The solution was evaporated to dryness and the residue was chromatographed on a silica gel column (15x5cm) using dichloromethane followed by 0.3% (10% conc. ammonium hydroxide in methanol)-dichloromethane as the eluant to give 2-{[3-bromo-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidin-7-ylamino]ethyl}piperidine-1-carboxylic acid *tert*-butyl ester (491.1mg, 79%): FABMS: m/z 534.1 (MH^+); HRESIMS: m/z 534.12797 (MH^+). Calcd. for $C_{24}H_{30}N_5O_2BrCl$: m/z 534.12714; δ_H ($CDCl_3$) 1.50 (1H, m, CH_2), 1.51 (9H, s, $COOC(CH_3)_3$), 1.57 (2H, m, CH_2), 1.68 (2H, m, CH_2), 1.76 (2H, m, CH_2), 2.24 (1H, bm, CH_2), 2.82/3.40/3.54/4.08/4.51 (5H, m, CH/CH_2), 6.34 (1H, s, H_6), 7.41 (2H, m, Ar-H),

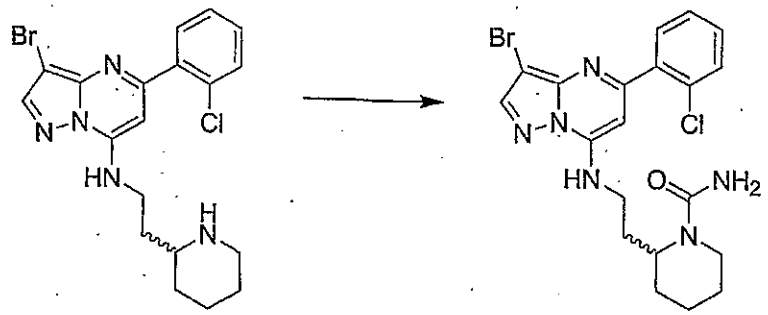
7.51 (1H, m, Ar-H), 7.76 (1H, m, Ar-H) and 8.08 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_3: 28.5, 28.5, 28.5; CH_2: 19.2, 25.5, 29.2, 29.2, 39.2, 67.1; CH: ~47.4, 87.1, 127.1, 130.1, 130.1, 131.6, 143.9; C: 80.0, 83.0, 132.1, 138.9, 145.7, 146.2, 158.0.

B. [3-BROMO-5-(2-CHLOROPHENYL)PYRAZOLO[1,5-a]PYRIMIDIN-7-YL]-(2-PIPERIDIN-2-YLETHYL)AMINE:



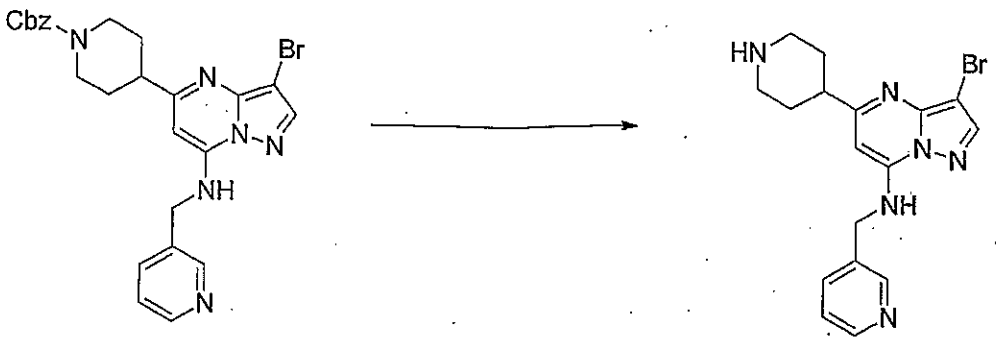
2-{[3-Bromo-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidin-7-ylamino]ethyl}piperidine-1-carboxylic acid *tert*-butyl ester (465mg, 0.869mmoles) (prepared as described in Example 461, Step A above) was dissolved in methanol (4.5mL) and 10% (v/v) conc. sulfuric acid in 1,4-dioxane (11.76mL) was added. The mixture was stirred at 25°C for 1.5h. The product was worked up as described in Preparative Example 241, step B and chromatographed on a silica gel column (15x5cm) using 3.5% (10% conc. ammonium hydroxide in methanol)-dichloromethane as the eluant to give [3-bromo-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidin-7-yl]piperidin-2-ylethyl)amine (365.6mg, 97%): FABMS: m/z 434.1 (MH^+); HRFABMS: m/z 434.0726 (MH^+). Calcd. for $C_{19}H_{22}N_5BrCl$: m/z 434.0747; δ_H ($CDCl_3$) 1.24 (1H, m, CH_2), 1.41 (1H, m, CH_2), 1.49 (1H, m, CH_2), 1.66 (1H, m, CH_2), 1.73 (1H, m, CH_2), 1.81 (1H, m, CH_2), 1.88 (2H, m, CH_2), 2.68 (1H, m, CH_2), 2.78 (1H, m, CH_2), 3.20 (1H,m, CH), 3.55 (1H, m, CH_2), 3.60 (1H, m, CH_2), 6.32 (1H, s, H_6), 7.41 (2H, m, Ar-H), 7.51 (1H, m, Ar-H), 7.74 (1H, m, Ar-H), 7.78 (1H, m, NH) and 8.05 ppm (1H, s, H_2); δ_C ($CDCl_3$) CH_2: 24.7, 26.8, 33.1, 35.2, 40.3, 47.0; CH: 55.7, 87.2, 127.1, 130.0, 130.1, 131.5, 143.8; C: 82.9, 132.1, 139.0, 145.7, 146.5, 158.1.

C. 2-{2-[3-BROMO-5-(2-CHLOROPHENYL)PYRAZOLO[1,5-a]PYRIMIDIN-7-YLAMINO]ETHYL}PIPERIDINE-1-CARBOXYLIC ACID AMIDE:



[3-Bromo-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidin-7-yl]piperidin-2-ylethyl)amine (200mg, 0.46mmoles) (prepared as described in Example 461, Step B above) was dissolved in anhydrous dichloromethane (2mL) and trimethylsilylisocyanate (0.31mL, 2.3mmoles) was added. The mixture was stirred at 25°C for 1.25h. Additional trimethylsilylisocyanate (0.155mL, 1.15mmoles) was added and the stirring was continued for a total of 3h. The mixture was diluted with dichloromethane and washed with saturated aqueous sodium bicarbonate. The organic layer was dried ($MgSO_4$), filtered and evaporated to dryness. The residue was chromatographed on a silica gel column (30x2.5cm) using 2% (10% conc. ammonium hydroxide in methanol)-dichloromethane as the eluant to give 2-{2-[3-bromo-5-(2-chlorophenyl)pyrazolo[1,5-a]pyrimidin-7-ylamino]ethyl}piperidine-1-carboxylic acid amide (106.3mg; 48%): FABMS: m/z 477.0 (MH^+); HRFABMS: m/z 477.0804 (MH^+). Calcd. for $C_{20}H_{23}N_6OBrCl$: m/z 477.0805; δ_H (d_6-DMSO) 1.29 (1H, m, CH_2), 1.52 (5H, m, CH_2), 1.72 (1H, m, CH_2), 2.05 (1H, m, CH_2), 2.51 (2H, s, $CONH_2$), 2.79 (1H, dd, CH), 3.31 (1H, m, CH_2), 3.34 (1H, m, CH_2), 3.76 (1H, m, CH_2), 4.30 (1H, bm, CH_2), 6.42 (1H, s, H_6), 7.50 (2H, m, Ar-H), 7.60 (1H, m, Ar-H), 7.63 (1H, m, Ar-H), 8.29 (1H, s, H_2) and 8.38 ppm (1H, dd, NH); δ_C (d_6-DMSO) CH_2: 18.6, 25.2, 28.2, 38.4, 38.6, 54.8; CH: 46.7, 86.6, 127.1, 129.7, 130.3, 131.0, 143.4; C: 81.2, 131.0, 138.7, 145.1, 146.4, 158.2.

EXAMPLE 462:

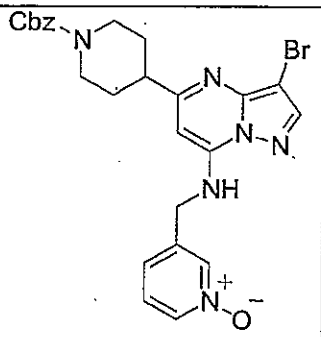
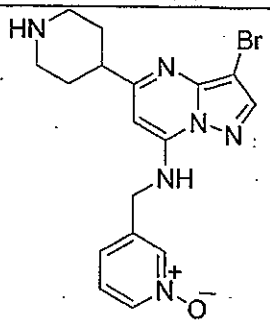


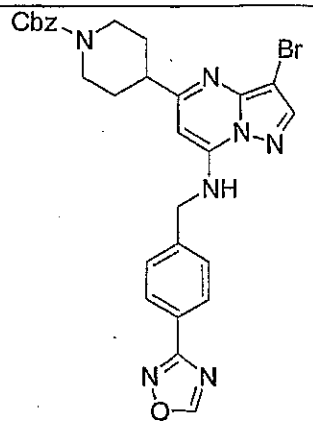
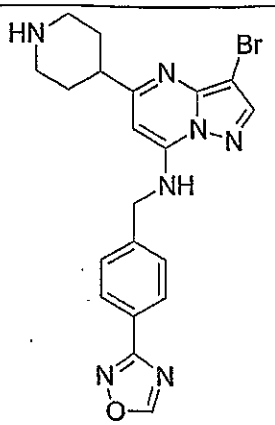
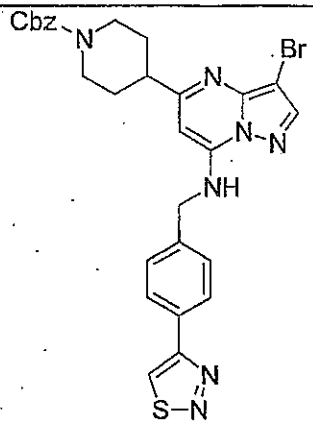
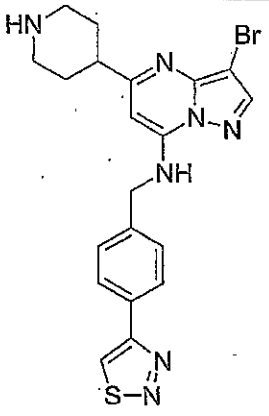
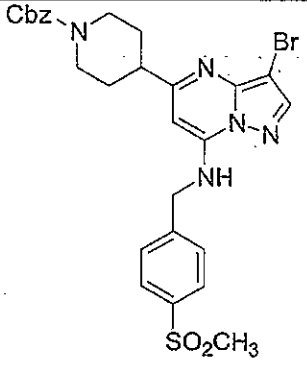
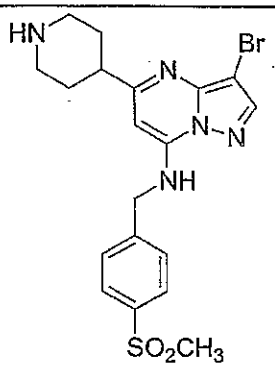
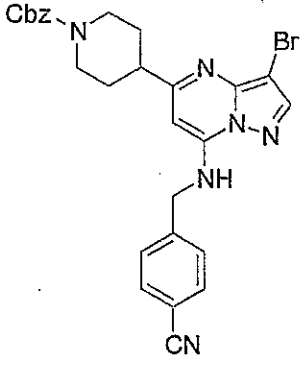
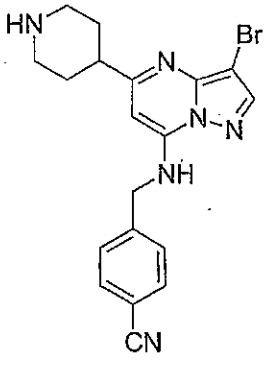
To a solution of the compound prepared in Example 204 (1.11 g, 2.12 mmol) in anhydrous acetonitrile (20 mL) was added TMSI (1.70 g, 8.52 mmol), dropwise at ambient temperature. After 10 minutes the acetonitrile was removed *in vacuo*. The resulting yellow foam was treated with 2 N HCl solution (7 mL) and then washed immediately with Et_2O (5X). The pH of the aqueous was adjusted to 10 with 50 % NaOH (aq) and the product was isolated by saturation of the solution with NaCl (s) followed by extraction with CH_2Cl_2 (5X) to give the crystalline product (733 mg, 89% yield). MH^+ = 387; m. p. = 207.5 °C

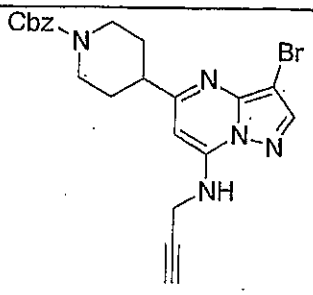
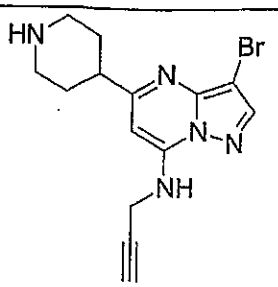
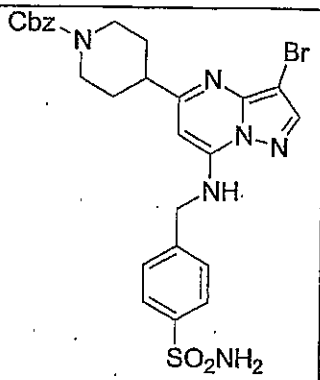
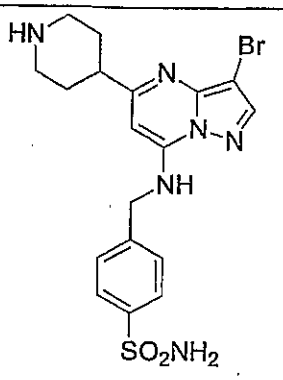
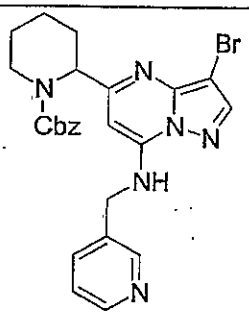
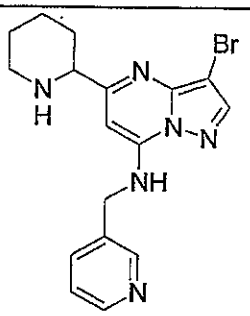
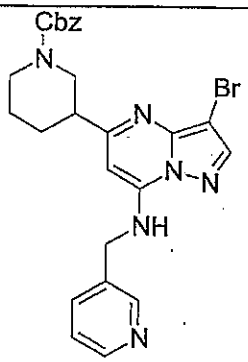
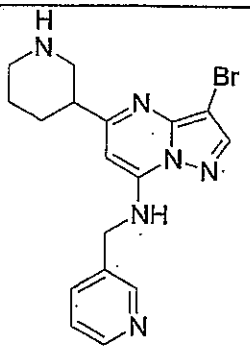
EXAMPLES 463-472:

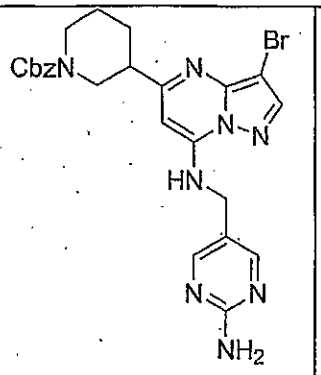
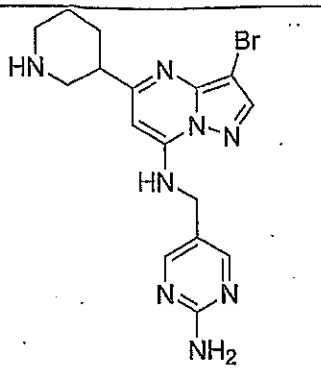
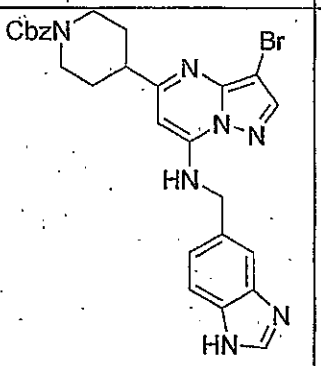
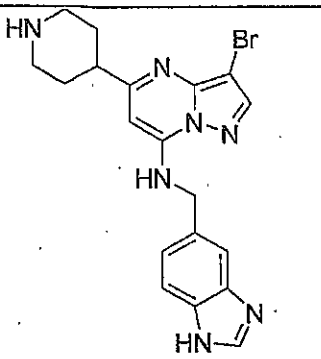
By essentially the same procedure set forth in Example 462 only substituting the compounds shown in Column 2 of Table 38, the compounds shown in Column 3 of Table 38 were prepared.

TABLE 38

Ex.	Column 2	Column 3	CMPD
463			MH^+ = 403 ^1H NMR (300MHz, $CDCl_3$) δ 8.52 (s, 1H), 8.38 (d, 1H), 8.04 (s, 1H), 7.78 (d, 1H), 7.65 (t, 1H), 6.18 (s, 1H), 4.89 (s, 2H), 3.26 –3.21 (d, 2H), 2.96 – 2.70 (m, 3H), 2.05 –1.78 (m, 4H).

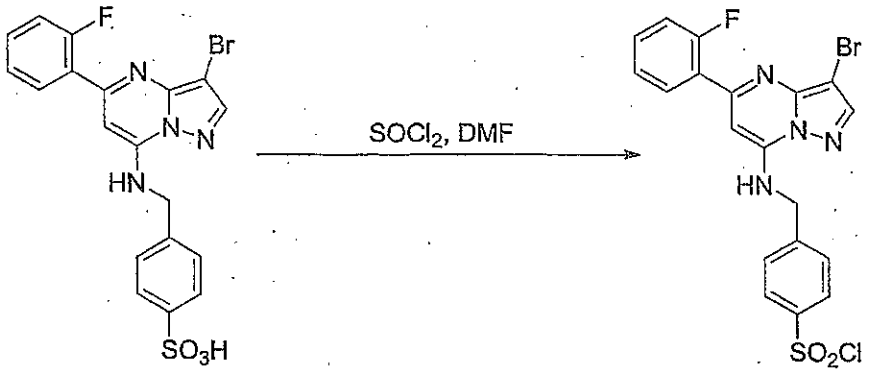
464			MH^+ = 454 m. p. = 175.4 °C
465			Yield = 87 MH^+ = 470 m. p. = 220 °C m. pt (hydrochloride salt) = 164.3 °C
466			MH^+ = 464 m. p. = 206 °C
467			MH^+ = 411 m. p. = 169.5 °C

468			MH^+ = 334 m. p. = 176.2 °C
469			MH^+ = 465 m. p. = 250.4 °C
470			MH^+ = 387 m. p. = 68.5 °C
471			MH^+ = 387 m. p. = 59.4 °C

472			1. mp = 230-232 2. M+H = 396
472.10			1. mp = 157-160 2. M+H = 427

EXAMPLE 473:

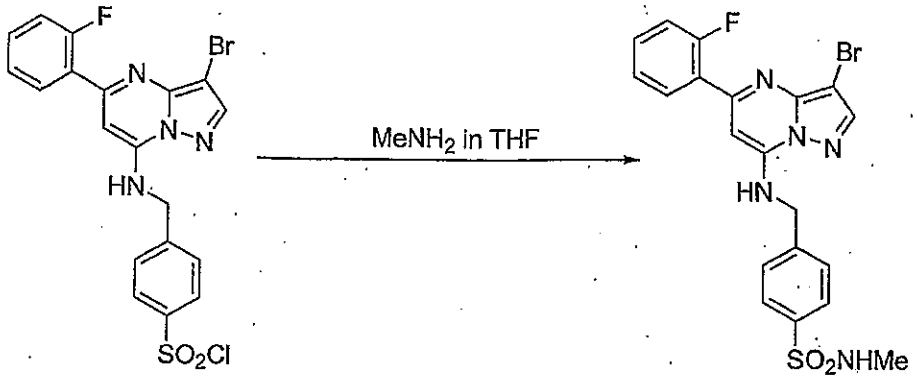
Step A:



A solution of the sulfonic acid (560 mg, 1.17 mmol) in 5 mL of dry DMF was cooled to 0 °C and $SOCl_2$ (278 mg, 2.34 mmol) was added. The reaction mixture was brought to RT and stirred overnight. The next day the contents were poured on ice and the pH was carefully adjusted to 8. The product was extracted in to EtOAc and the solvent was removed after drying (Na_2SO_4) to provide 240 mg (41%) of the crude sulfonyl chloride which was used for the next

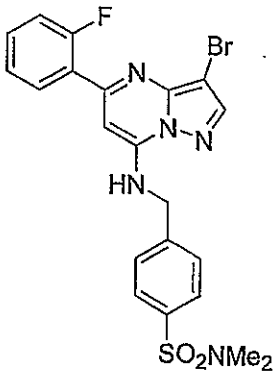
step without further purification. ^{1}H NMR ($CDCl_3$) δ 8.20-8.10 (m,1H), 8.10-7.95 (m, 3H), 7.65 (d, 2H), 7.45-7.35 (m, 1H), 7.35-7.20 (m, 1H), 7.15-7.05 (m, 1H), 6.95 (t, 1H), 4.85 (d, 2H).

Step B:



A solution of compound prepared in Example 473, Step A (120 mg, 0.24 mmol) in 10 mL of THF was treated with 2 mL of 1 M $MeNH_2$ (2.00 mmol) in THF at RT overnight. The solvent was removed and the residue was purified by chromatography (silica, hexane:EtOAc (4:1→1:1)) to provide 56 mg (48%) of the sulfonamide. ^{1}H NMR (DMSO-d6) δ 9.05 (t, J = 9 Hz, 1H), 8.35 (s, 1H), 7.90 (t, J = 7.5 Hz, 1H), 7.75 (d, J = 9 Hz, 2H), 7.62 (d, J = 9 Hz, 2H), 7.55-7.46 (m, 1H), 7.45-7.38 (m, 1H), 7.38-7.25 (m, 1H), 6.50 (s, 1H), 4.80 (d, 2H), 3.30 (s, 3H) LCMS: MH^+ = 492.1

EXAMPLE 474:

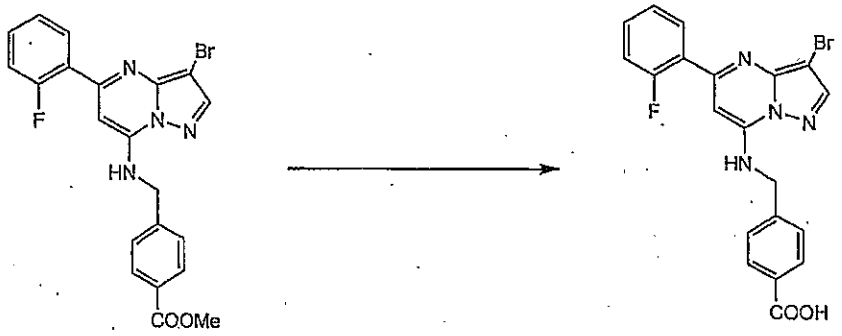


By essentially the same procedure set forth in Example 473, only substituting dimethylamine, the above compound was prepared. ^{1}H NMR ($CDCl_3$)

δ 8.14 (t, J = 9 Hz, 1H), 8.00 (s, 1H), 7.76 (d, J = 9 Hz, 2H), 7.54 (d, J = 9 Hz, 2H), 7.34-7.44 (m, 1H), 7.26 (t, J = 9 Hz, 1H), 7.14-7.04 (m, 1H), 6.93 (t, J= 6 Hz, 1H), 6.45 (s, 1H), 4.75 (d, 2H), 2.70 (s, 6H)

LCMS: MH^+ = 504.2

EXAMPLE 475:

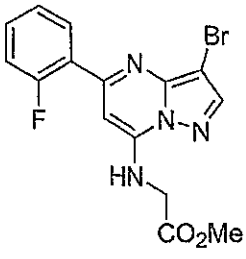
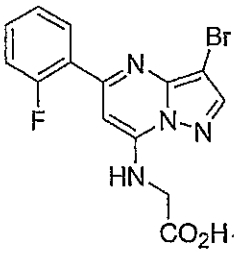


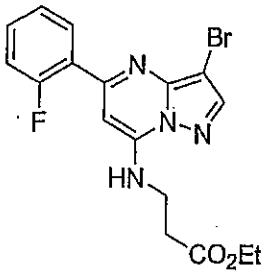
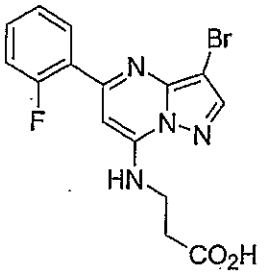
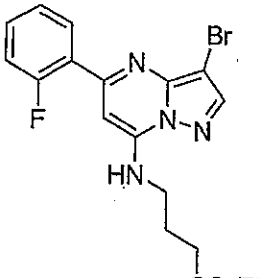
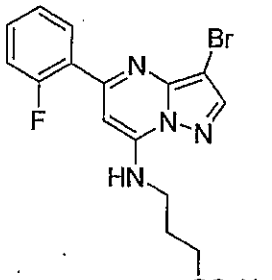
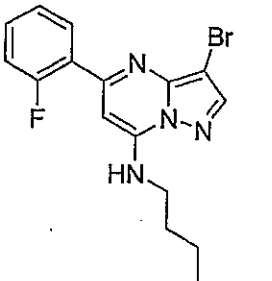
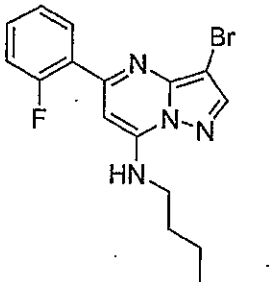
A mixture of the compound prepared in Example 129 (300 mg, 0.66 mmol), NaOH (5 g), CH_3OH –H_2O (100 mL, 90:10) was stirred at 25 C for about 15 h. Progress of hydrolysis was checked by TLC. Reaction mixture was concentrated to remove methanol. The concentrate was diluted with 50 mL water, and extracted with ether to remove any un-reacted ester. Aqueous solution, thus obtained, was neutralized with 3 N HCl to pH 4 to obtain free acid, filtered and washed repeatedly with water. The acid was dried under vacuum (270 mg, 93%) and used without further purification.

Example 476-479:

By essentially the same procedure set forth in Example 475 only substituting the compounds in Column 2 of Table 39, the compounds in Column 3 of Table 39 were prepared.

Table 39

Ex.	Column 2	Column 3	CMPD
476			Yield = 82% LCMS: MH^+ = 365

477			Yield = 82% LCMS: MH^+ = 379
478			Yield = 72% LCMS: MH^+ = 393
479			Yield = 70% LCMS: MH^+ = 407

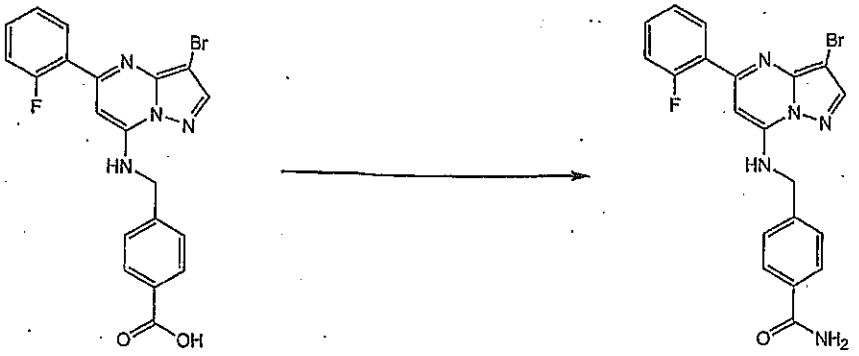
Additional data for select examples shown below:

Example 476: ^{1}H NMR ($CDCl_3$) δ 8.15 (m, 2H), 8.0 (m, 1H), 7.6 (m, 1H), 7.3 (m, 2H), 6.6 (s, 1H), 4.2 (d, 2H).

Example 477: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.4 (m, 1H), 7.25 (dd, 1H), 7.15 (dd, 1H), 7.0 (t, 1H), 6.5 (s, 1H), 3.8 (dt, 2H), 2.6 (t, 2H).

Example 479: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.4 (m, 1H), 7.25 (dd, 1H), 7.15 (dd, 1H), 6.8 (t, 1H), 3.5 (dt, 2H), 2.4 (t, 2H), 1.8 (m, 4H).

EXAMPLE 480:

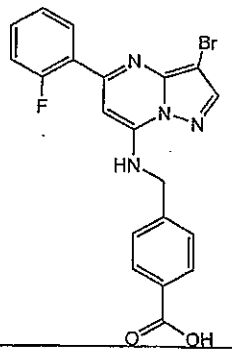
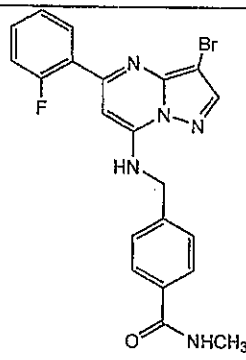


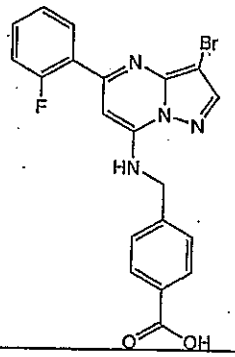
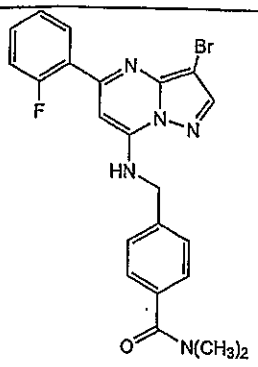
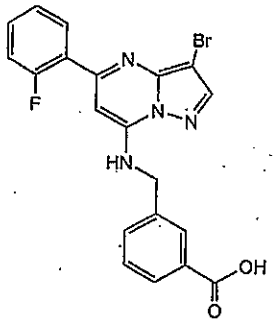
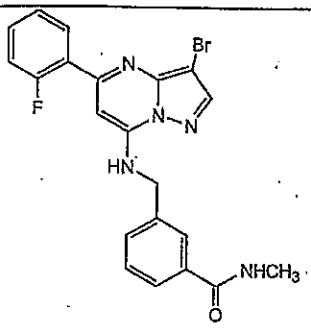
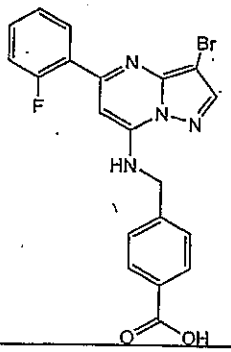
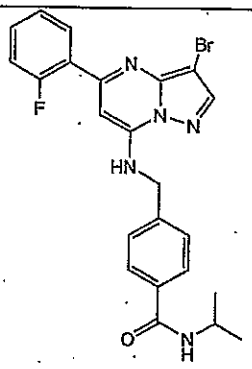
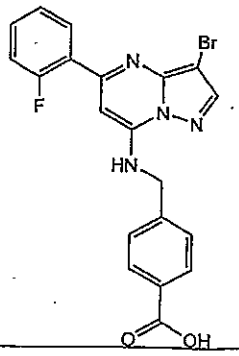
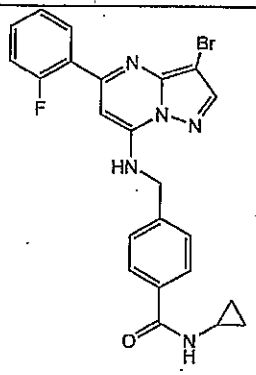
A mixture of the acid from Example 475 (85mg, 0.193 mmol) and Et_3N (20 mg, 0.193 mmol) in THF (20 mL) was stirred at 25 C for 15 min. Isobutyryl chloroformate (28mg, 0.205 mmol) was added to the reaction mixture and stirred for 10 min followed by addition of NH4OH solution (0.5 mL). The reaction mixture was stirred for 1 hr and concentrated to dryness. The dry mass was purified by column chromatography.

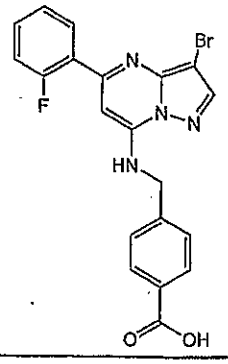
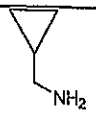
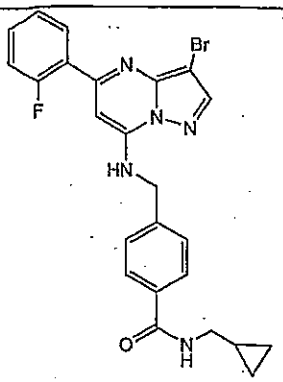
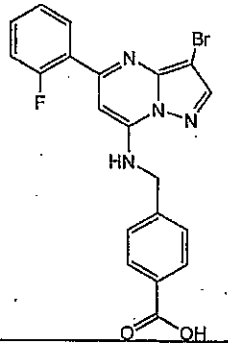
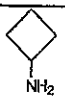
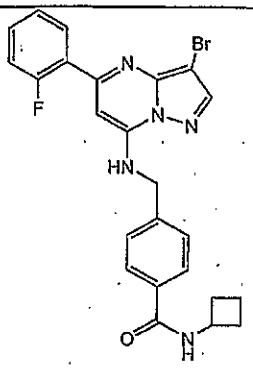
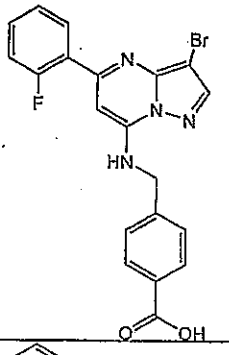
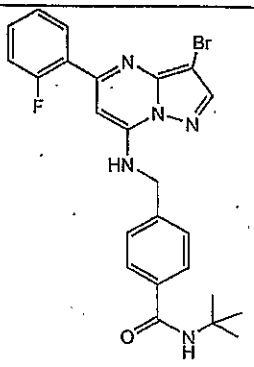
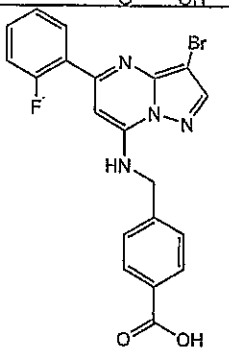
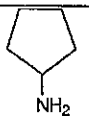
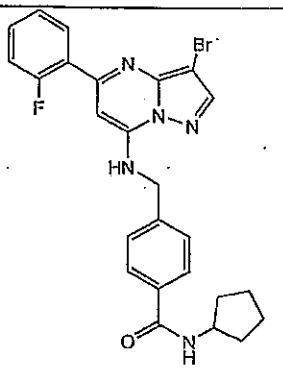
EXAMPLES 481-509:

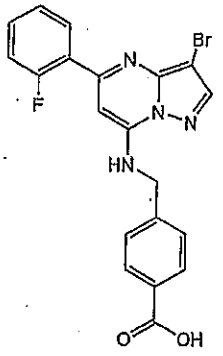
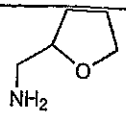
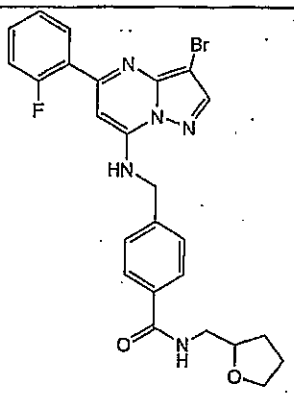
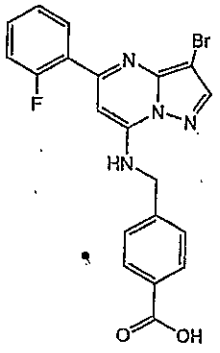
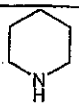
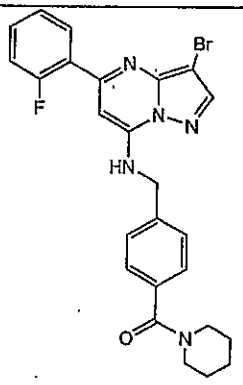
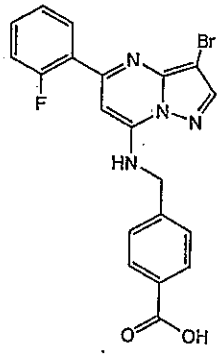
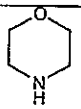
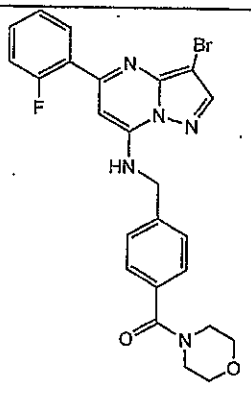
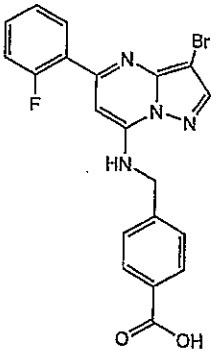
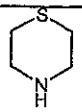
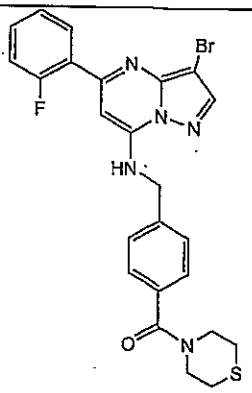
By essentially the same procedure set forth in Example 480 only substituting the carboxylic acid shown in Column 2 of Table 40 and the amine shown in Column 3 of Table 40, the compounds shown in Column 4 of Table 40 were prepared.

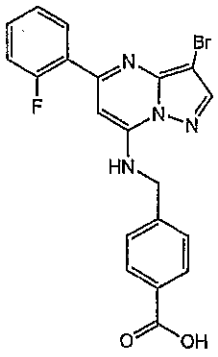
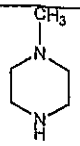
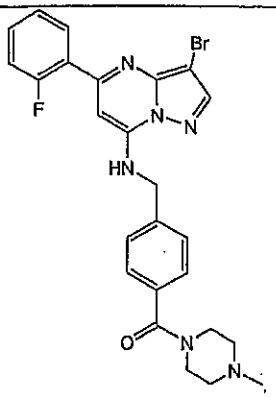
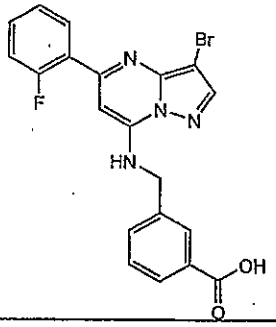
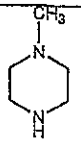
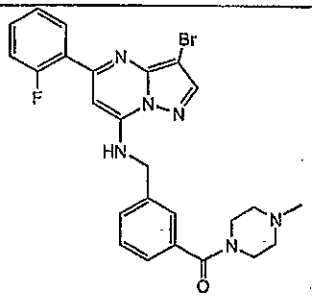
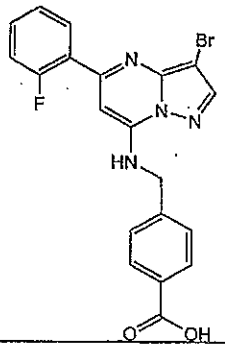
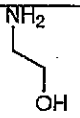
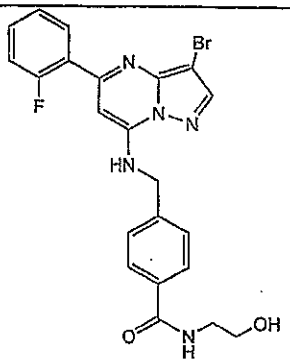
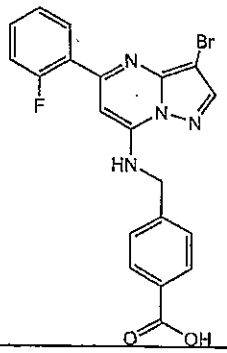
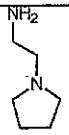
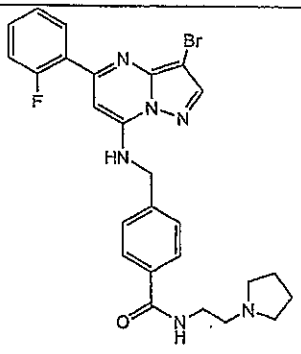
Table 40

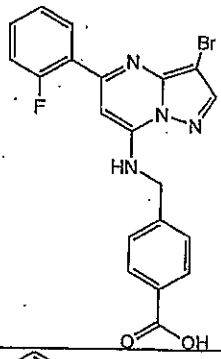
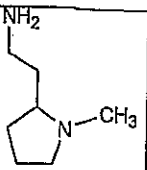
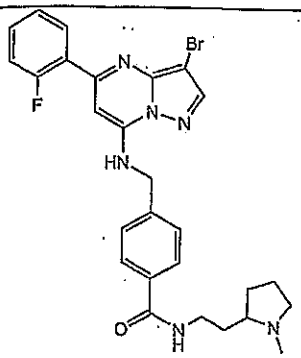
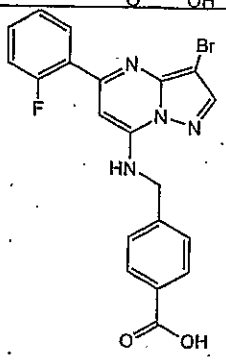
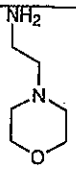
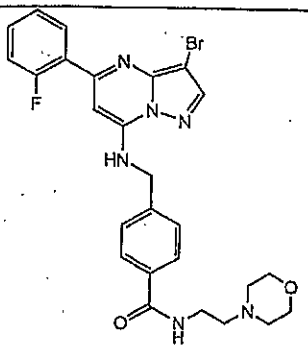
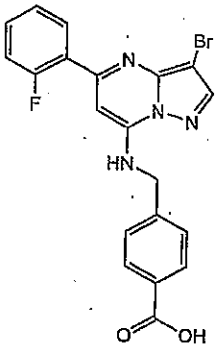
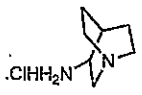
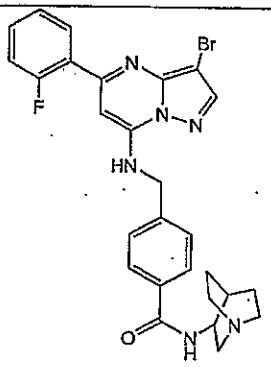
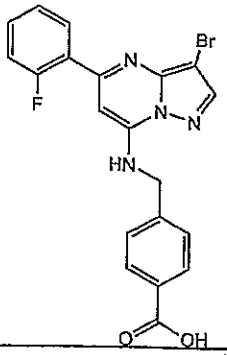
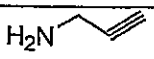
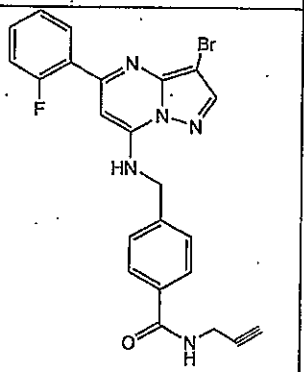
Ex.	Column 2	Column 3	Column 4	CMPD
481		CH_3NH_2		Yield = 88% LCMS: MH^+ = 454

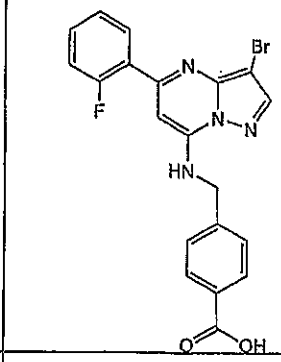
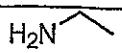
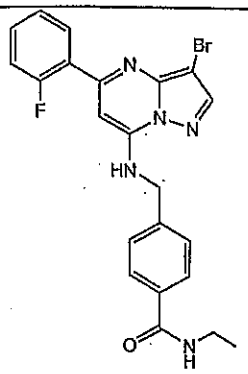
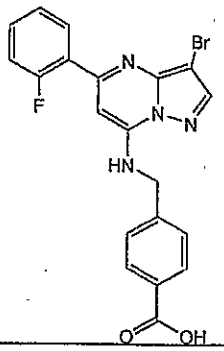
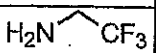
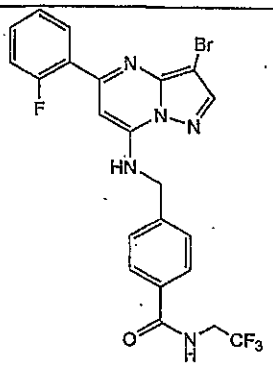
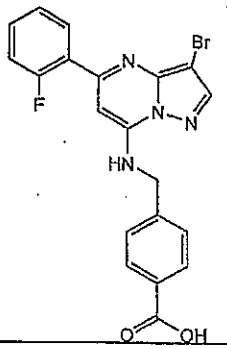
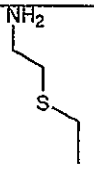
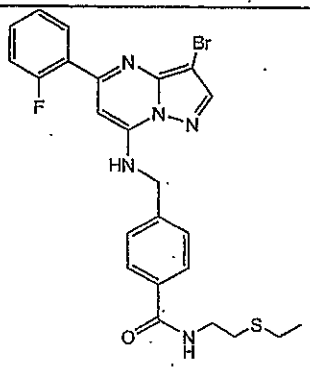
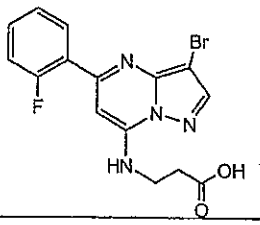
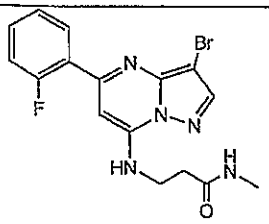
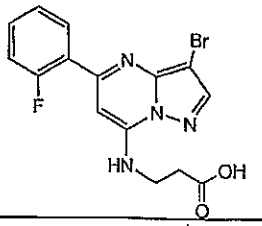
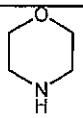
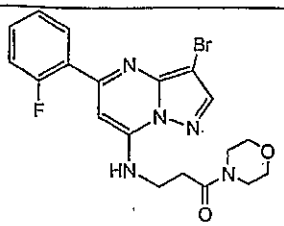
482		$(CH_3)_2NH$		Yield=80 % LCMS MH^+ = 468
483		CH_3NH_2		Yield=70 % LCMS MH^+ = 454.
484				Yield=75 % LCMS MH^+ = 482.1
485				Yield=71 % LCMS MH^+ = 480.1

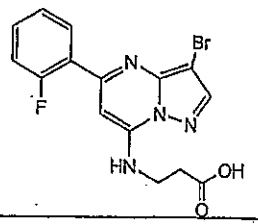
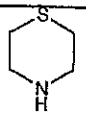
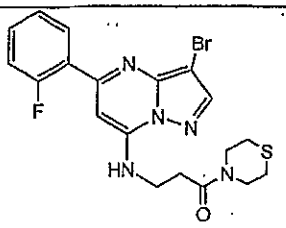
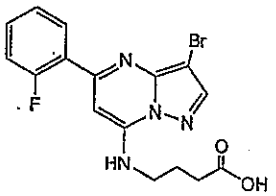
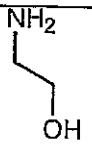
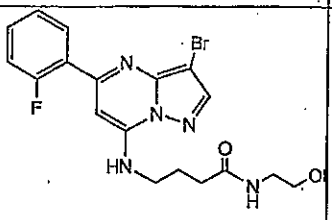
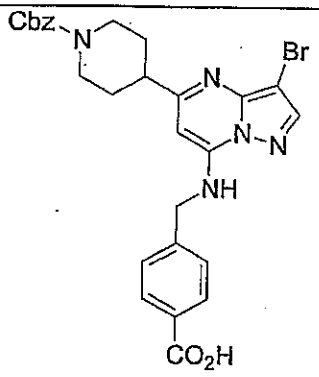
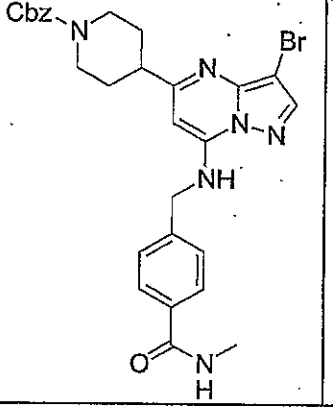
486				Yield=75 % LCMS MH^+ = 494.1
487				Yield=75 % MH^+ = 494.1
488				Yield=75 % MH^+ = 496.1
489				Yield=75 % LCMS MH^+ = 508.1

490				Yield=78 % LCMS MH^+ = 524.1
491				Yield=73 % LCMS MH^+ = 508.1
492				Yield=73 % LCMS MH^+ = 510.1
493				Yield=76 % LCMS MH^+ = 526.1

494				Yield=76 % MH^+ = 523.1
495				Yield=76 % MH^+ = 523.1
496				Yield=51 % LCMS MH^+ = 484.1
497				Yield=66 % MH^+ = 537.1

498				Yield=76 % LCMS MH^+ = 551.2
499				Yield=79 % LCMS MH^+ = 552.1
500				Yield=80 % MH^+ = 549.1
501				Yield=80 % LCMS MH^+ = 478.1

502				Yield=80 % LCMH$^+$ = 468.1
503				Yield=80 % MH$^+$ = 522.1
504				Yield=82 % LCMS MH$^+$ = 528.1
505		CH_3NH_2		Yield=60 % MH$^+$ = 392
506				Yield=60 % LCMH$^+$ = 448.1

507				Yield=70 % MH^+ = 464.1
508				Yield=50 % LCMS MH^+ = 436.1
508.10		CH_3NH_2		Yield = 92 MH^+ = 577

Additional data for select examples given below:

Example 481: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.7 (d, 2H), 7.4 (s, 1H), 7.35 (d, 2H), 7.25 (dd, 1H), 7.1 (dd, 1H), 6.95 (t, 1H), 6.5 (s, 1H), 6.25 (bs, 1H), 4.7 (d, 2H), 3.0 (d, 3H).

Example 482: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H),8.0 (s, 1H), 7.45 – 7.35 (m, 4H), 7.25 (d, 2H), 7.15 (dd, 1H), 6.7 (t, 1H), 6.5 (s, 1H), 4.7 (d, 2H), 3.1 (s, 3H), 3.0 (s, 3H).

Example 483: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.8 (bs, 1H), 7.7 (d, 1H), 7.5 – 7.3 (m, 3H), 7.25 (d, 1H), 7.15 (dd, 1H), 6.75 (t, 1H), 6.5 (s, 1H), 6.2 (bs, 1H), 4.7 (d, 2H), 3.0 (d, 3H).

Example 484: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.7 (d, 2H), 7.4 (d, 2H), 7.35 (m, 1H), 7.25 (dd, 1H), 7.15 (dd, 1H), 6.8 (t, 1H), 6.5 (s, 1H), 6.0 bs, 1H), 4.7 (d, 2H), 4.25 (m, 1H), 1.2 (d, 6H).

Example 485: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.7 (d, 2H), 7.4 (d, 2H), 7.35 (s, 1H), 7.25 (dd, 1H), 7.1 (dd, 1H), 6.9 (t, 1H), 6.5 (s, 1H), 6.3 (t, 1H), 4.7 (d, 2H), 2.9 (m, 1H), 0.8 (bt, 2H), 0.6 (bt, 2H).

Example 486: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.8 (d, 2H), 7.4 (d, 2H), 7.35 (d, 1H), 7.25 (dd, 1H), 7.1 (dd, 1H), 6.9 (t, 1H), 6.5 (s, 1H), 6.2 (t, 1H), 4.7 (d, 2H), 3.3 (dd, 2H), 1.05 (m, 1H), 0.5 (m, 2H), 0.25 (m, 2H).

Example 487: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.7 (d, 2H), 7.4 (d, 2H), 7.35 (m, 1H), 7.25 (dd, 1H), 7.15 (dd, 1H), 6.85 (t, 1H), 6.5 (s, 1H), 6.2 (bs, 1H), 4.7 (d, 2H), 4.6 (m, 1H), 2.4 (m, 2H), 1.95 (m, 1H), 1.75 (m, 2H).

Example 488: ^{1}H NMR ($CDCl_3$) δ 8.5 (t, 1H), 8.15 (dt, 1H), 8.0 (s, 1H), 7.7 (d, 2H), 7.4 (d, 2H), 7.35 (m, 1H), 7.25 (dd, 1H), 7.15 (dd, 1H), 6.8 (t, 1H), 6.5 (s, 1H), 5.9 (bs, 1H), 4.7 (d, 2H), 1.4 (s, 9H).

Example 489: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.7 (d, 2H), 7.4 (d, 2H), 7.35 (m, 1H), 7.25 (dd, 1H), 7.15 (dd, 1H), 6.8 (t, 1H), 6.5 (s, 1H), 6.0 bs, 1H), 4.7 (d, 2H), 4.4 (m, 1H), 2.05 (m, 2H), 1.7 (m, 4H), 1.4 (m, 2H).

Example 490: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.7 (d, 2H), 7.4 (d, 2H), 7.35 (m, 1H), 7.25 (dd, 1H), 7.15 (dd, 1H), 6.8 (t, 1H), 6.5 (s, 1H), 6.5 (bs, 2H), 4.7 (d, 2H), 4.1 (m, 1H), 3.9 – 3.7 (m, 3H), 3.3 (m, 1H), 2.0 – 1.9 (m, 4H).

Example 491: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.45 – 7.35 (m, 5H), 7.25 (dd, 1H), 7.1 (dd, 1H), 6.8 (t, 1H), 6.5 (s, 1H), 4.7 (d, 2H), 3.7 (bs, 2H),3.3 (bs, 2H), 1.7 (bs, 4H), 1.5 (bs, 2H).

Example 492: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.45 – 7.35 (m, 5H), 7.25 (dd, 1H), 7.1 (dd, 1H), 6.85 (t, 1H), 6.5 (s, 1H), 4.7 (d, 2H), 3.8 – 3.4 (bm, 8H).

Example 493: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.45 – 7.35 (m, 5H), 7.25 (dd, 1H), 7.1 (dd, 1H), 6.80 (t, 1H), 6.5 (s, 1H), 4.7 (d, 2H), 4.0 (m, 2H), 3.6 (m, 2H), 2.8 – 2.45 (m, 4H).

Example 494: ^{1}H NMR (CH3OD) δ 8.15 (s, 1H), 8.0 (dt, 1H), 7.45 – 7.35 (m, 5H), 7.25 (dd, 1H), 7.1 (dd, 1H), 6.80 (t, 1H), 6.5 (s, 1H), 4.7 (d, 2H), 3.7 (bs, 2H), 3.4 (bs, 2H), 2.5 – 2.4 (m, 4H), 2.2 (s, 3H).

Example 495: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.45 – 7.35 (m, 5H), 7.25 (dd, 1H), 7.1 (dd, 1H), 6.80 (t, 1H), 6.5 (s, 1H), 4.7 (d, 2H), 3.75 (bs, 2H), 3.35 (bs, 2H), 2.4 (bs, 2H), 2.3 (s, 3H), 2.2 (bs, 2H).

Example 496: ^{1}H NMR ($CDCl_3$) δ 7.95 (s, 1H), 7.9 (dt, 1H), 7.8 (t, 1H), 7.7 (d, 2H), 7.15 (m, 4H), 7.05 (dd,1H), 6.9 (dd, 1H), 6.2 (s,1H), 4.5 (d, 2H), 3.6 (t, 2H), 3.3 (dt, 2H).

Example 497: ^{1}H NMR (CH3OD) δ 8.1 (s, 1H), 7.9 (dt, 1H), 7.8 (d, 2H), 7.5 (d, 2H), 7.4 (m, 1H), 7.3 (dd, 1H), 7.2 (dd, 1H), 6.4 (s, 1H), 4.7 (d, 2H), 3.5 (t, 2H), 2.7 (m, 2H), 2.6 (bs, 4H), 1.8 (bs, 4H).

Example 498: ^{1}H NMR ($CDCl_3$) δ 8.5 (t, 1H), 8.15 (dt, 1H), 8.0 (s, 1H), 7.8 (d, 2H), 7.4 (d, 2H), 7.35 (m, 1H), 7.25 (dd, 1H), 7.15 (dd, 1H), 6.8 (t, 1H), 6.5 (s, 1H), 4.7 (d, 2H), 3.7 – 2.5 (m, 4H), 2.35 (s, 3H), 2.2 (m, 1H), 1.9 – 1.6 (m, 6H).

Example 499: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.8 (d, 2H), 7.4 (d, 2H), 7.35 (m, 1H), 7.25 (dd, 1H), 7.15 (dd, 1H), 6.8 (t, 1H), 6.5 (s, 1H), 4.7 (d, 2H), 3.7 (m, 4H), 3.5 (dt, 2H), 2.6 (t, 2H), 2.5 (m, 4H).

Example 500: ^{1}H NMR (CH3OD) δ 8.15 (s, 1H), 7.9 (dt, 1H), 7.8 (d, 2H), 7.45 (d, 2H), 7.4 (m, 1H),7.25 (dd, 1H), 7.15 (dd, 1H), 6.4 (s, 1H), 4.75 (d, 2H), 4.2 (m, 1H), 3.4 – 2.8 (m, 7H), 1.9 – 1.6 (m, 4H).

Example 501: ^{1}H NMR ($CDCl_3$) δ 8.05 (dt, 1H), 8.0 (s, 1H), 7.6 (d, 2H), 7.4 (s, 1H), 7.35 (d, 2H), 7.25 (dd, 1H), 7.1 (dd, 1H), 6.9 (t, 1H), 6.5 (s, 1H), 6.4 (t,1H), 4.7 (d, 2H), 4.2 (d, 2H), 2.3 (bs, 1H).

Example 502: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.75 (d, 2H), 7.45 (s, 1H), 7.4 (d, 2H), 7.3 (dd, 1H), 7.1(dd, 1H), 6.8 (t, 1H), 6.5 (s, 1H), 6.1(bs, 1H), 4.7 (d, 2H), 3.5 (dq, 2H), 1.2 (t, 3H).

Example 503: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.8 (d, 2H), 7.4 (d, 2H), 7.35 (m, 1H), 7.25 (dd, 1H), 7.15 (dd, 1H), 6.9 (t, 1H), 6.5 (s, 1H), 6.4 (t, 1H), 4.75 (d, 2H), 4.1 (m, 2H).

Example 504: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.8 (d, 2H), 7.45 (d, 2H), 7.4 (m, 1H), 7.25 (dd, 1H), 7.1 (dd, 1H), 6.8 (t, 1H), 6.6 (t, 1H), 6.5 (s, 1H), 4.7 (d, 1H), 3.6 (m, 2H), 2.8 (t, 2H), 2.6 (q, 2H), 1.3 (t, 3H).

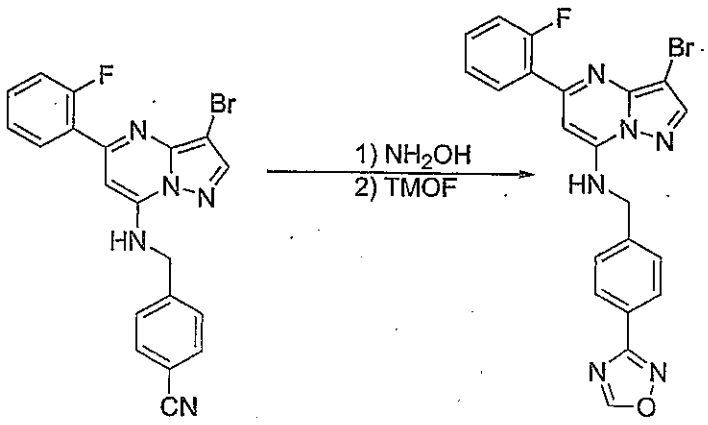
Example 505: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.4 (m, 1H), 7.25 (dd, 1H), 7.15 (dd, 1H), 7.0 (t, 1H), 6.5 (s, 1H), 3.8 (m, 2H), 2.7 (t, 2H), 3.0 (d, 3H).

Example 506: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.4 (m, 1H), 7.25 (dd, 1H), 7.15 (dd, 1H), 7.0 (t, 1H), 6.5 (s, 1H), 3.8 (m, 2H), 3.6 (m, 6H), 3.4 (m, 2H), 2.7 (t, 2H).

Example 507: ^{1}H NMR ($CDCl_3$) δ 8.15 (dt, 1H), 8.0 (s, 1H), 7.4 (m, 1H), 7.25 (dd, 1H), 7.15 (dd, 1H), 7.0 (t, 1H), 6.5 (s, 1H), 3.9 (t, 2H), 3.8 (dt, 2H), 3.7 (t, 2H), 2.7 (t, 2H), 2.6 (m, 4H).

Example 508: ^{1}H NMR (CH_3OD) δ 8.1 (s, 1H), 7.95 (dt, 1H), 7.5 (m, 1H), 7.35 – 7.2 (m, 2H), 6.5 (s, 1H),3.6 (m, 4H), 3.25 (m, 4H), 2.4 (t, 2H), 2.05 (dt, 2H).

EXAMPLE 509:

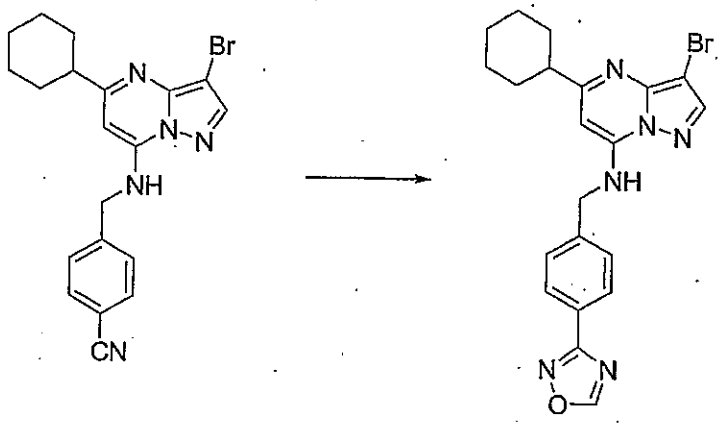


A solution of NaOH (59 mg, 1.47 mmol) in 1 mL of water was added to a suspension of $NH_2OH.HCl$ (102 mg, 1.47 mmol) in 10 mL of methanol at 0 °C. After 5 min, the compound prepared in Example 210.10 (208 mg, 0.49 mmol) was added and the reaction mixture was refluxed overnight. The solvent was removed *in vacuo* and the residue was partitioned between water and EtOAc. The EtOAc layer was dried (Na_2SO_4) and the solvent was evaporated. The resulting crude amidoxime was suspended in trimethyl orthoformate containing catalytic amount of PTS acid and refluxed overnight. The solvent was removed and the residue was taken up in EtOAc. The EtOAc layer was washed with aq $NaHCO_3$ followed by water and brine. The solvent was evaporated and the residue was purified by chromatography (silica, hexane:EtOAc (1:1)) to provide

80 mg (35%) of the oxadiazole. ^{1}H NMR ($CDCl_3$) δ 8.75 (s, 1H), 8.20-8.10 (m, 3H), 8.03 (s, 1H), 7.53 (d, J = 9 Hz, 2H), 7.45-7.36 (m, 1H), 7.30-7.22 (m, 2H), 7.16-7.08 (m, 1H), 6.80 (t, J = 5 Hz, 1H), 6.56 (s, 1H).

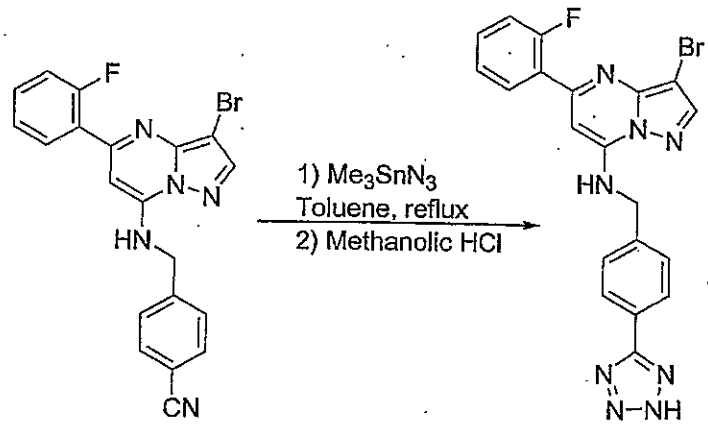
LCMS: MH^+ = 465.2

Example 510:



By essentially the same procedure set forth in Example 509 only substituting the compound prepared in Preparative Example 192, the above compound was prepared. yield = 75; MH^+ = 453; m. p. = 79.3°C.

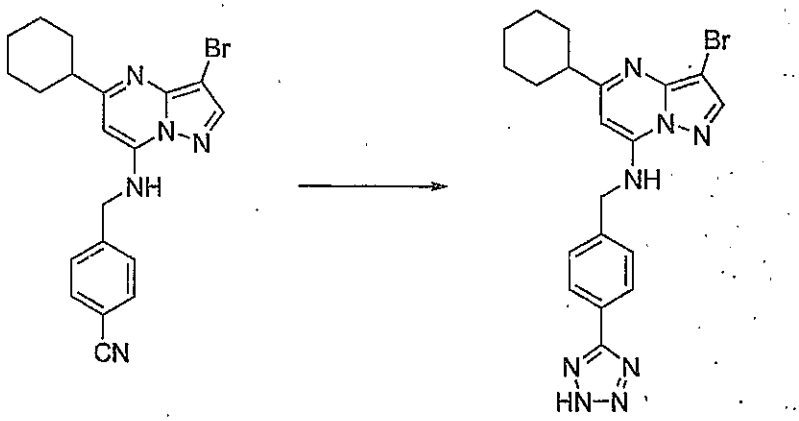
EXAMPLE 511:



A mixture of the nitrile (235 mg, 0.56 mmol) and Me_3SnN_3 (343 mg, 1.67 mmol) in 20 mL of dry toluene was refluxed for 2 days under Ar. The solvent was removed *in vacuo* and the residue was dissolved in dry methanol. HCl gas was bubbled through the solution for 15 min and the reaction mixture allowed to stand at overnight at RT. The next day, the solvent was removed, the residue was taken in water and the pH was adjusted to 5. The precipitated product was

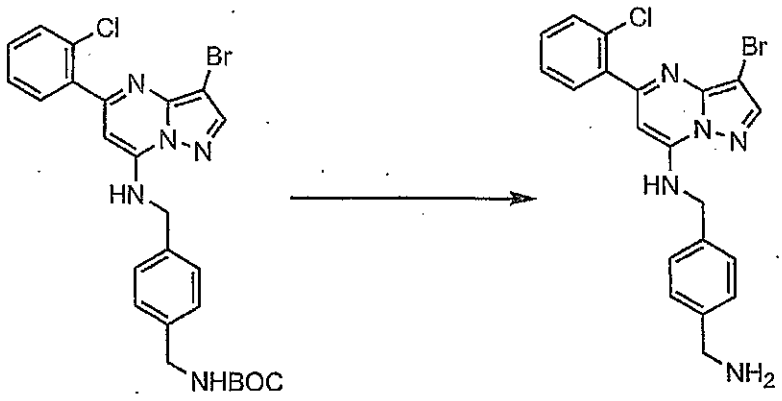
extracted into EtOAc. Evaporation of the EtOAc layer after drying (Na_2SO_4) provided the residue which was purified by chromatography (silica, DCM:MeOH (98:2→95:5)) to yield 50 mg (19%) of the pure tetrazole. ^1H NMR (CD_3OD) δ 8.10 (s, 1H), 8.00 (d, J = 9 Hz, 2H), 7.90 (t, J = 7 Hz, 1H), 7.65 (d, J = 9 Hz, 2H), 7.50-7.40 (m, 1H), 7.30-7.10 (m, 2H), 6.45 (s, 1H), 4.80 (s, 2H); LCMS: MH^+ = 465.0

EXAMPLE 512:



By essentially the same procedure set forth in Example 511 only substituting the compound prepared in Example 192, the above compound was prepared. Yield = 64; MH^+ = 453; m. p. = 238.9°C.

EXAMPLE 513:



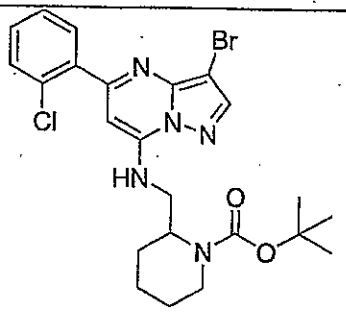
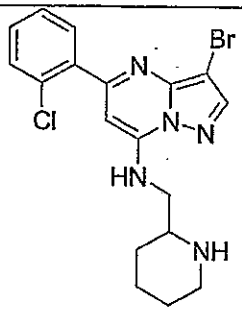
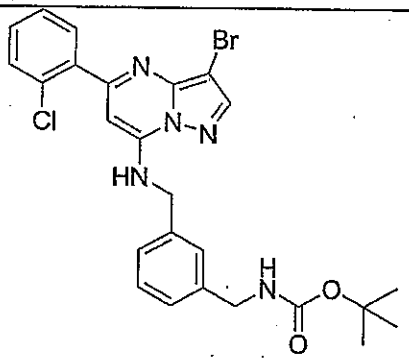
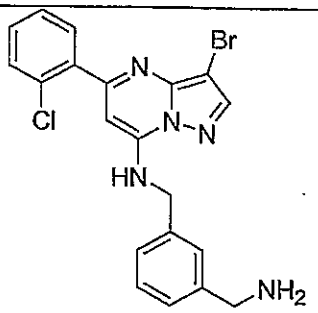
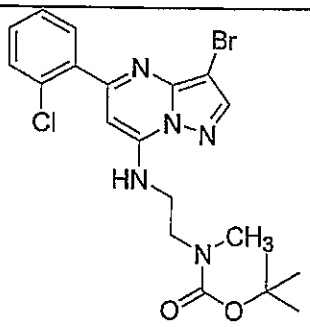
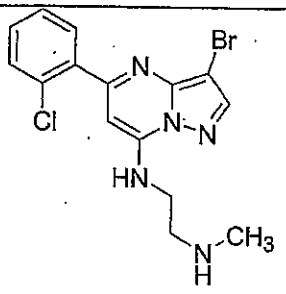
The compound prepared in Example 157 was dissolved in dioxane (30 mL) and a HCl-dioxane solution (4 M, 30 mL) was added. The reaction mixture was stirred at room temperature for 4 h. The reaction mixture was evaporated under reduced pressure and ethyl acetate (200 mL) was added. The organic

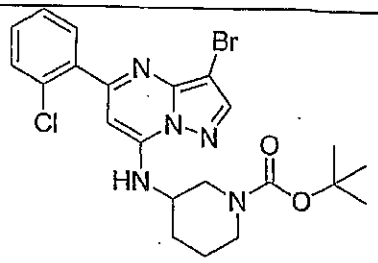
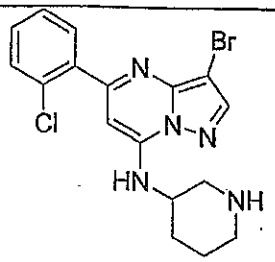
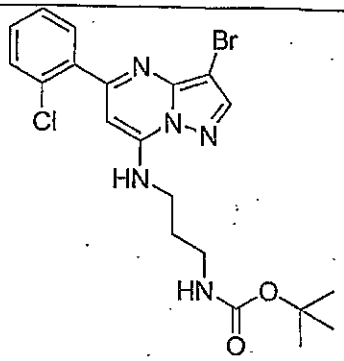
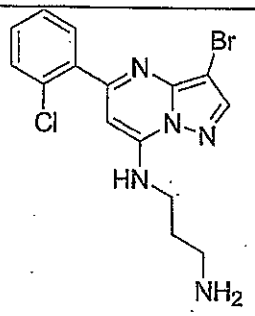
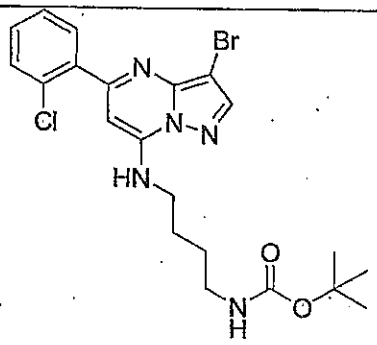
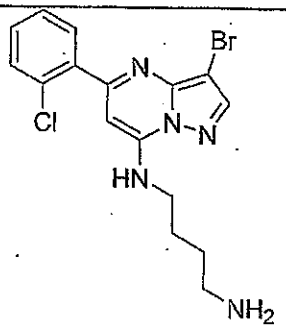
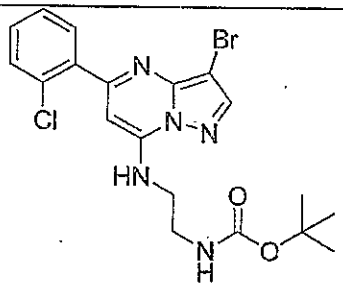
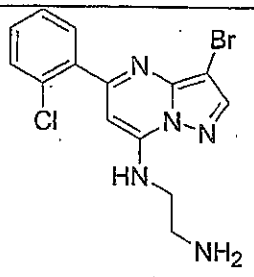
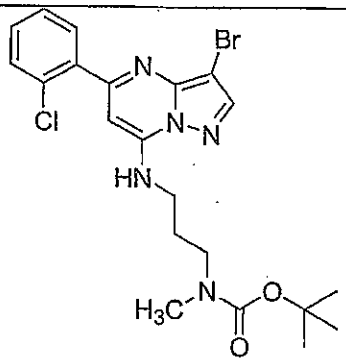
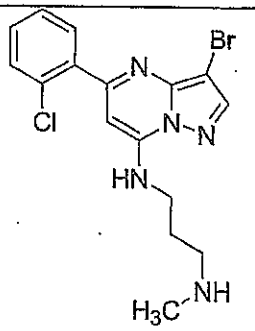
solution was washed with 1 N sodium hydroxide followed by saturated brine. The organic layer was dried over anhydrous sodium sulfate and evaporated under reduced pressure. MH^+ = 442.1

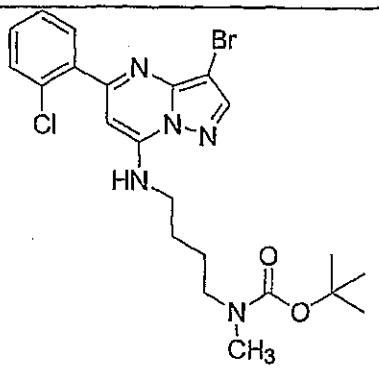
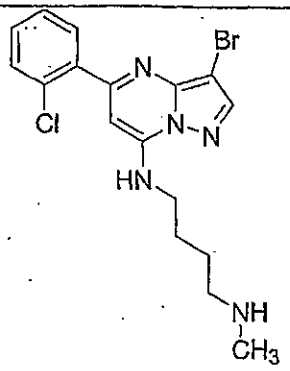
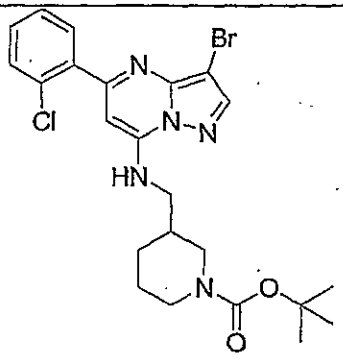
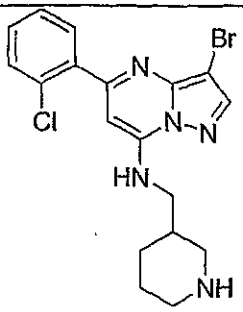
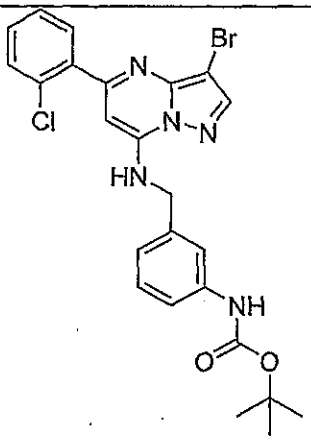
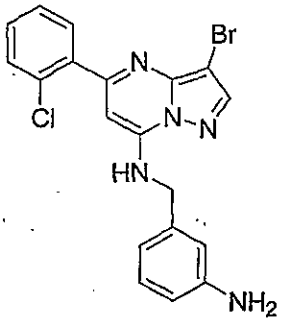
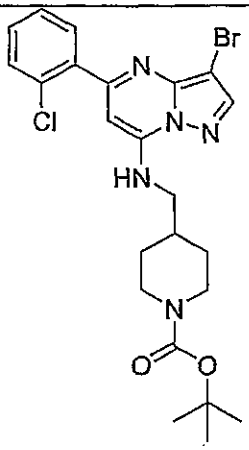
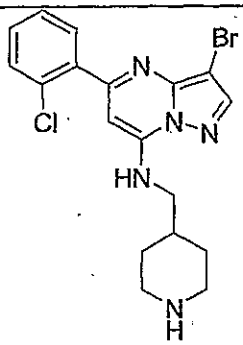
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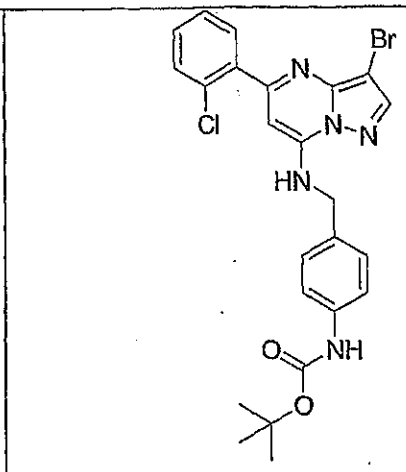
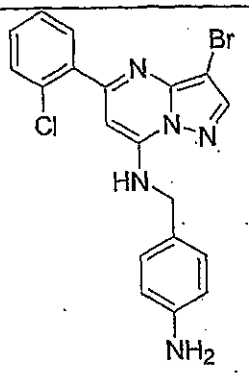
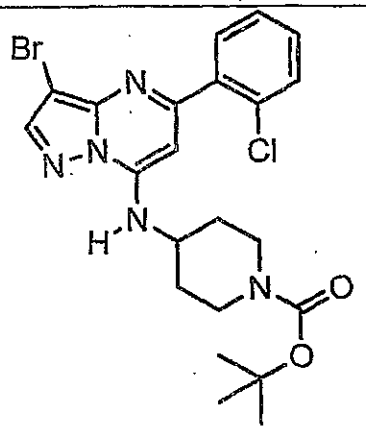
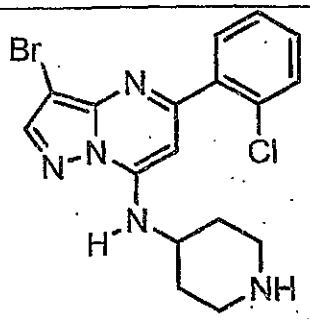
By essentially the same procedure set forth in Example 513, only substituting the compounds shown in Column 2 of Table 41, the compounds shown in Column 3 of Table 41 were prepared.

TABLE 41

Ex.	Column 2	Column 3	CMPD
514			MH^+ = 420.1
515			MH^+ = 442.1
516			MH^+ = 380.1

517			MH^+ = 406.1
518			MH^+ = 380.1
519			MH^+ = 394.1
520			MH^+ = 366
521			MH^+ = 394

522			MH^+ = 408.1
523			MH^+ = 420.1
524			
525			MH^+ = 420.1

526			MH^+ = 428.1
526.10			

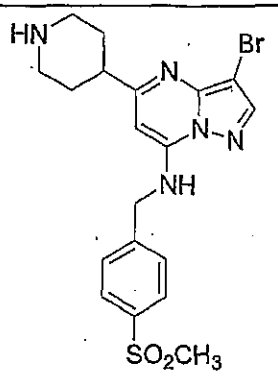
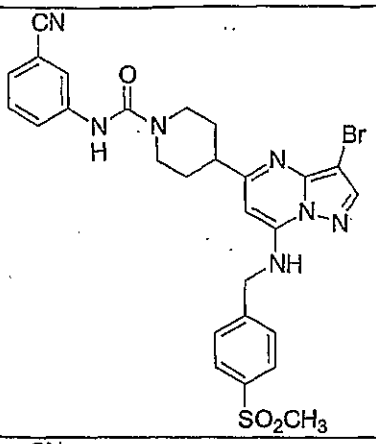
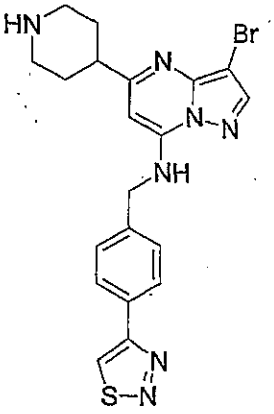
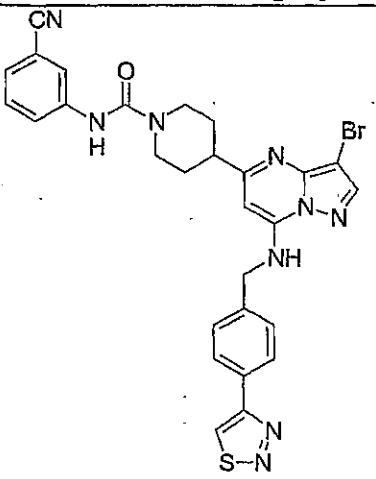
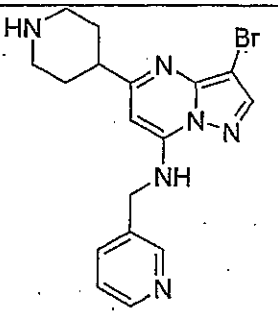
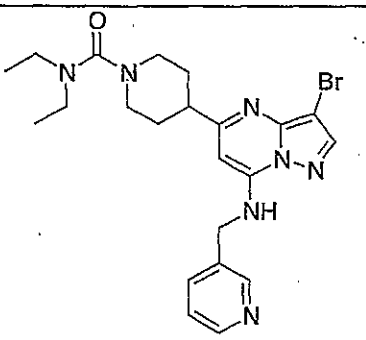
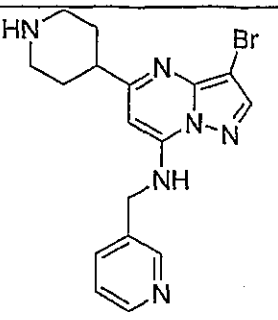
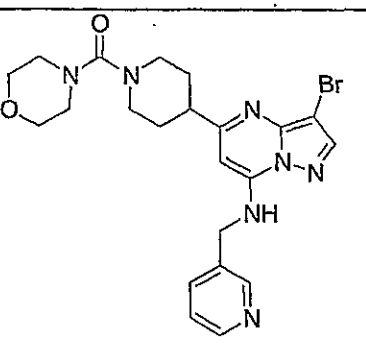
EXAMPLES 528-564:

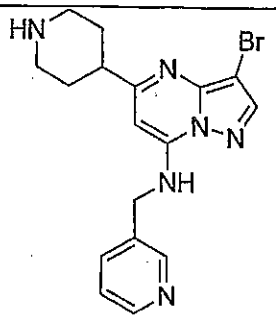
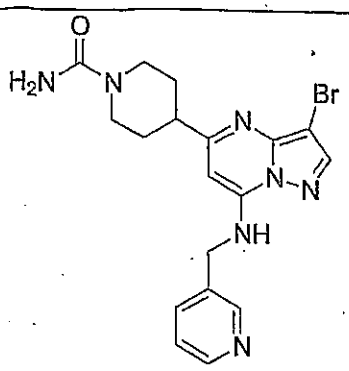
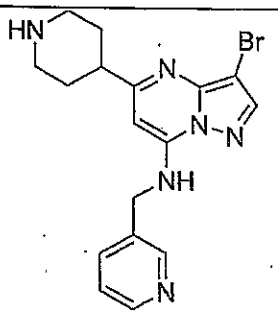
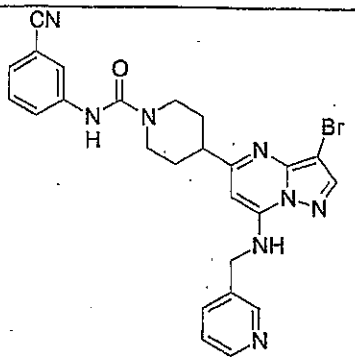
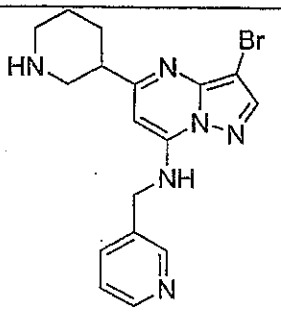
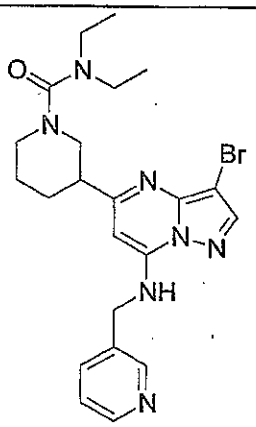
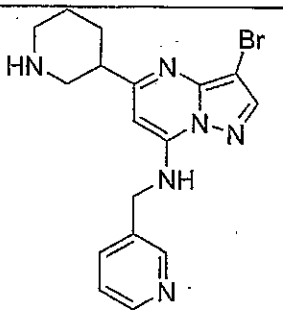
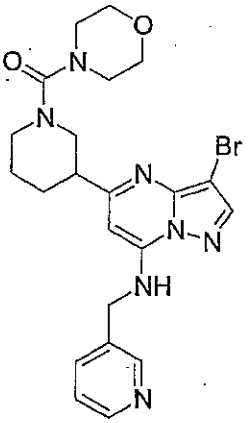
General procedure for 5-piperidinyl parallel library formation:

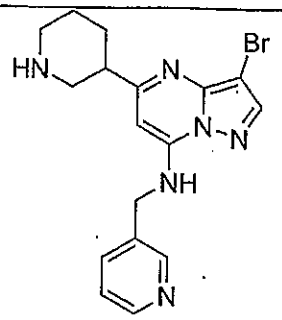
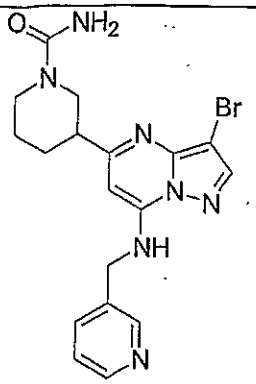
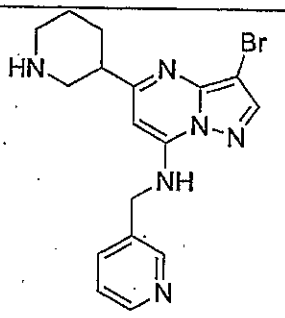
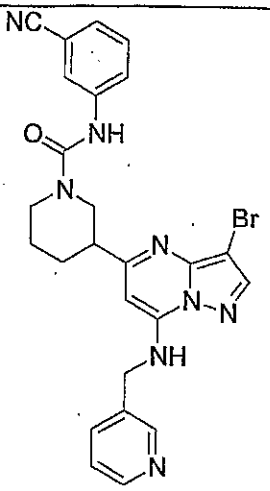
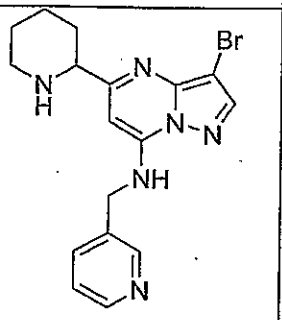
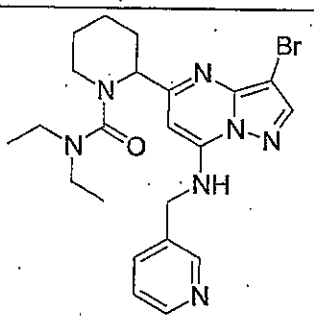
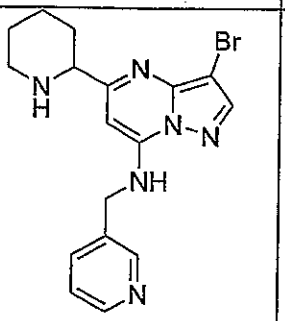
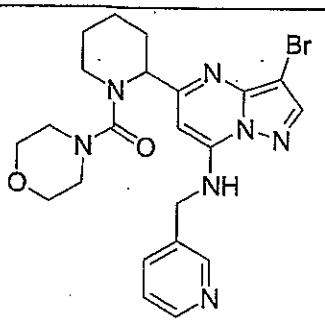
To a mixture of the starting material (80 mg, 0.21 mmol) shown in Column 2 of Table 42 in anhydrous CH_2Cl_2 (1.5 mL) was added DIPEA (75 μL, 0.42 mmol) and the appropriate capping reagent (1.1 equiv., 0.23 mmol). After 1 to 2 h, the reaction mixture was applied to 1000 micron preparatory TLC plate and was subsequently developed using a 8 – 10 % EtOH – CH_2Cl_2 as eluent to afford the compounds shown in Column 3 of Table 42.

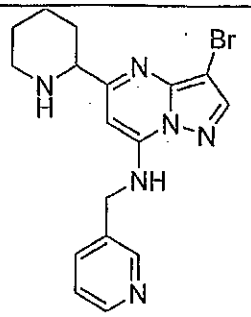
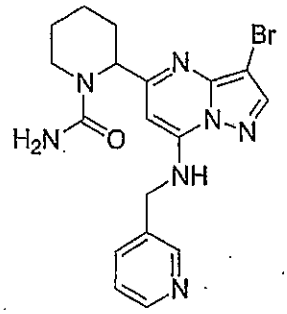
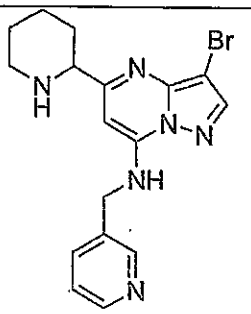
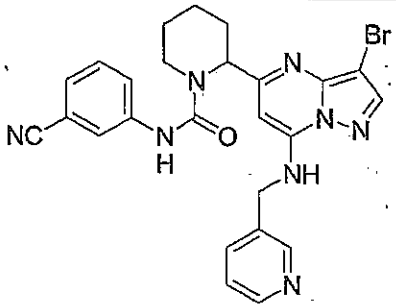
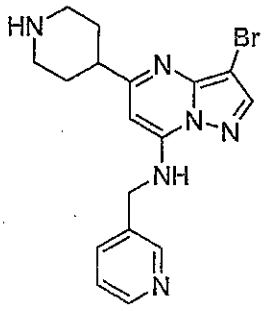
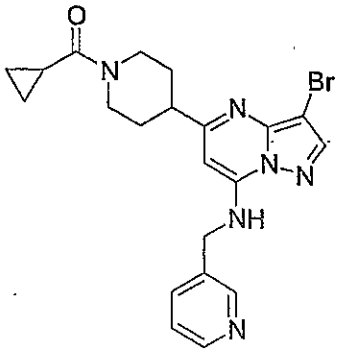
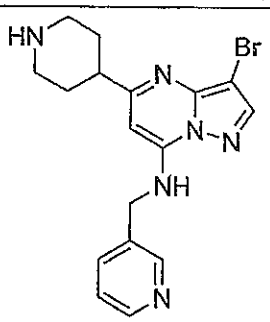
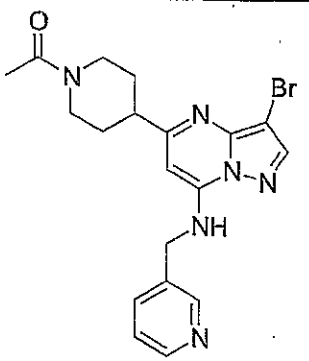
TABLE 42

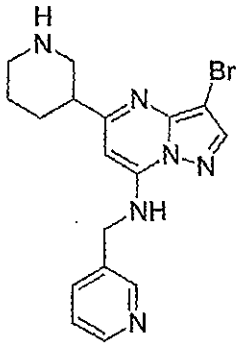
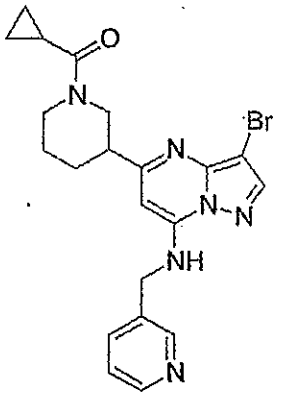
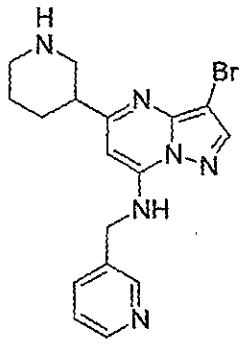
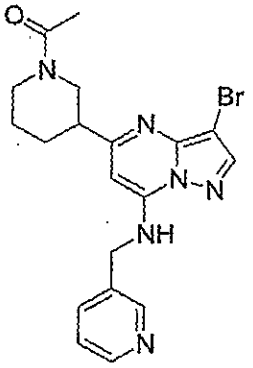
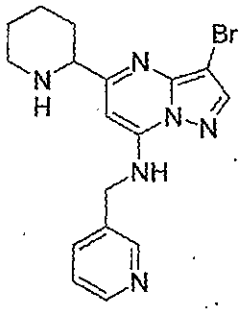
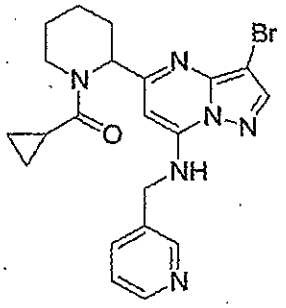
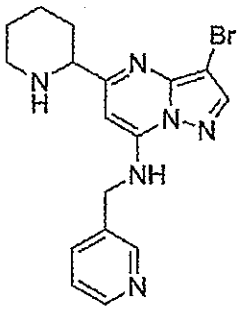
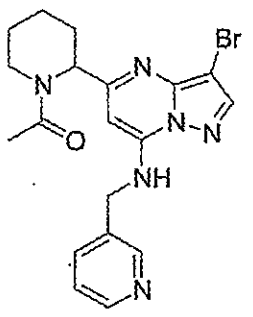
Ex.	Column 2	Column 3	CMPD

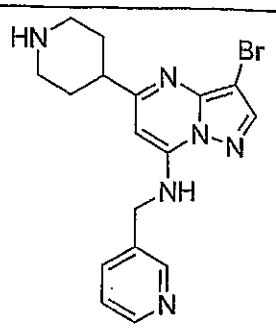
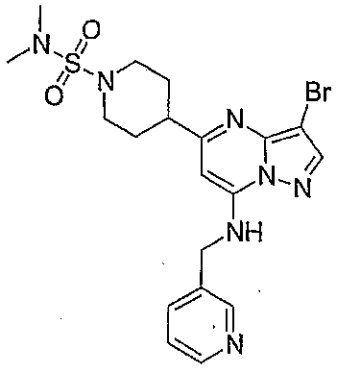
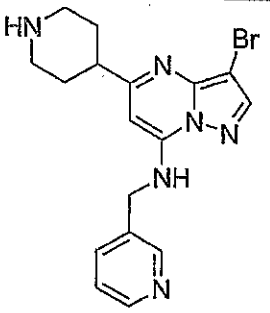
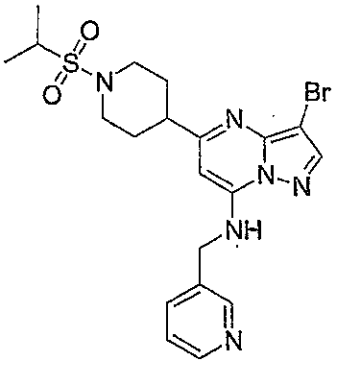
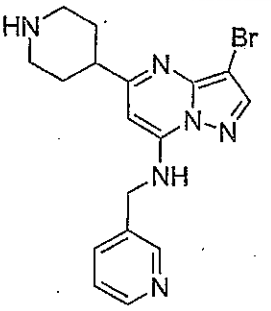
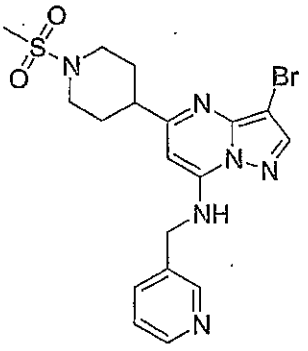
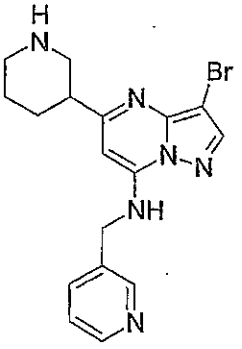
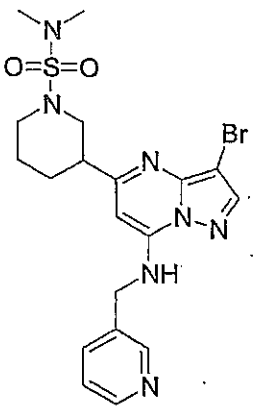
528			MH^+ = 608 m. p. = 230.1 °C
529			Yield = 82 MH^+ = 614 m. p. = 235.4 °C
530			MH^+ = 486 m. p. = 60.5 °C
531			MH^+ = 500 m. p. = 113.6 °C

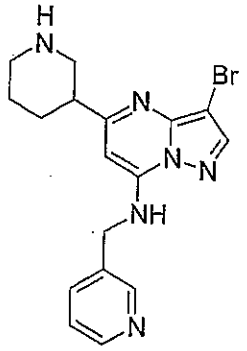
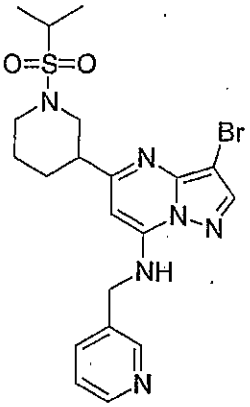
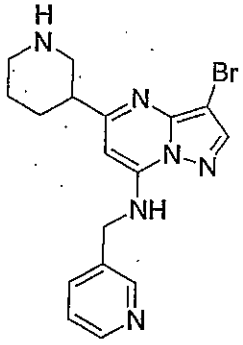
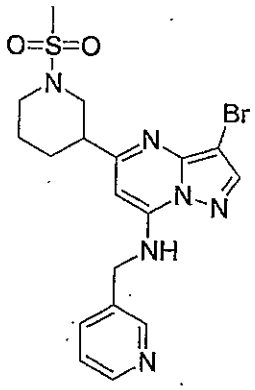
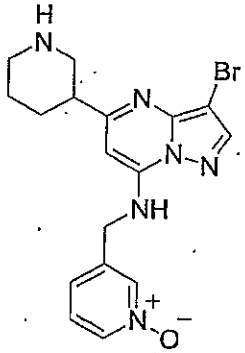
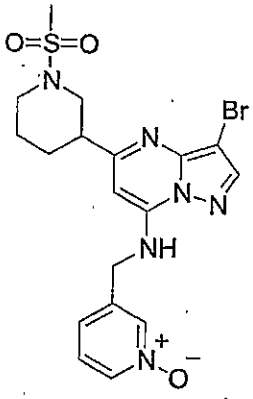
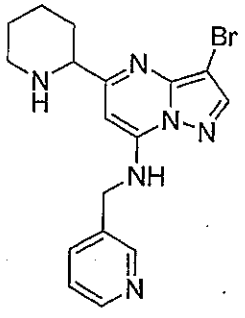
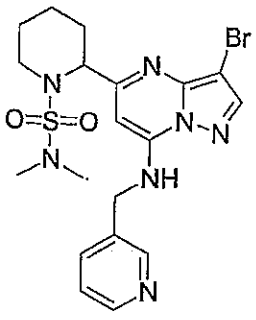
532			MH^+ = 430 m. p. = 158.3 – 159.2 °C
533			MH^+ = 531 m. p. =105.9 °C
534			MH^+ = 486
535.			MH^+ = 500

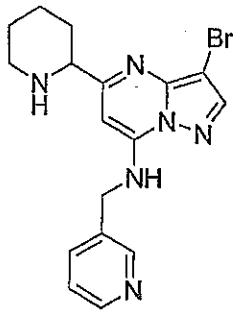
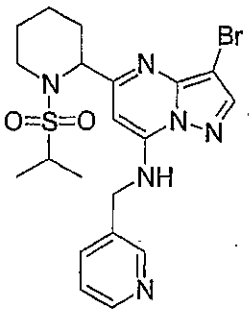
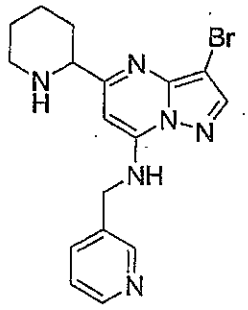
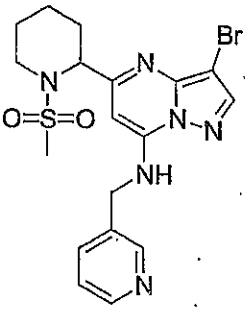
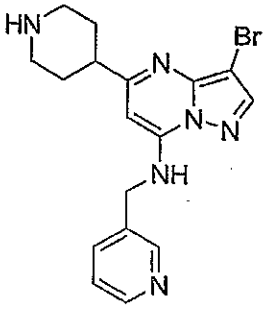
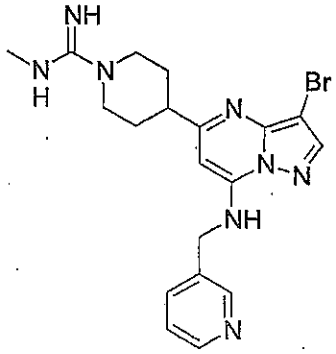
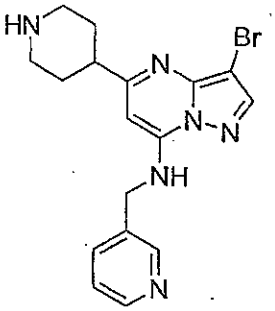
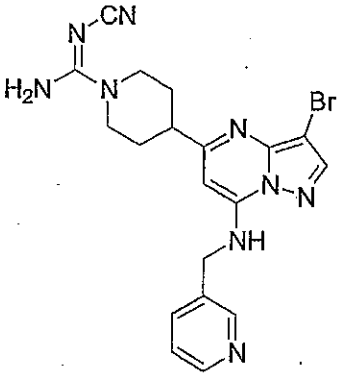
536			MH^+ = 430
537			MH^+ = 531
538			MH^+ = 486 m. p. = 69.6 °C
539			MH^+ = 500 m. p. = 82.3 °C

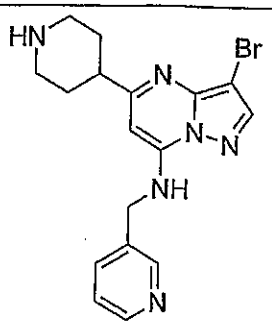
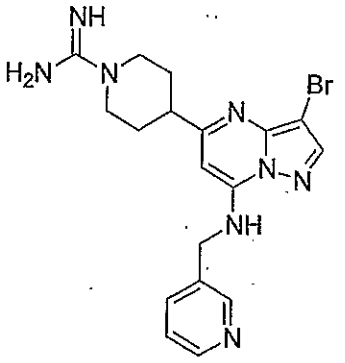
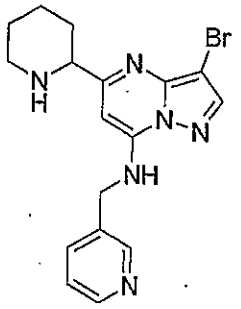
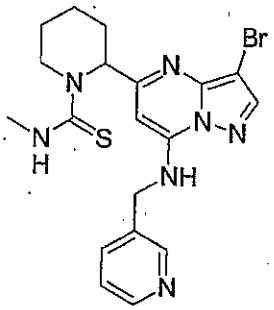
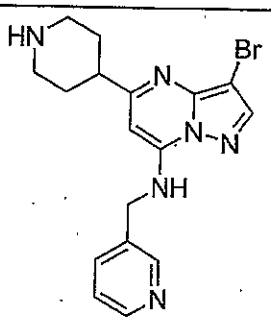
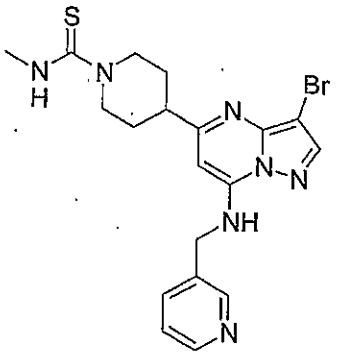
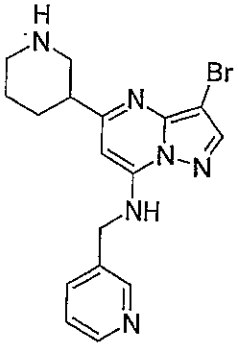
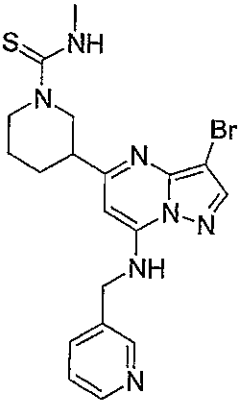
540			MH^+ = 430 m. p. = 223.6 °C
541			MH^+ = 531 m. p. = 118.1 °C
542			MH^+ = 455 m. p. = 109 – 110 °C
543			MH^+ = 429 m. p. = 111.5 °C

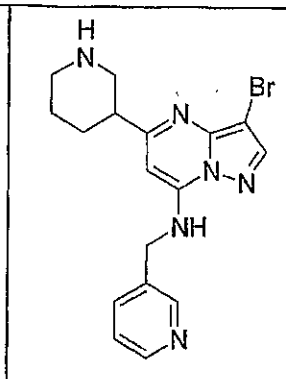
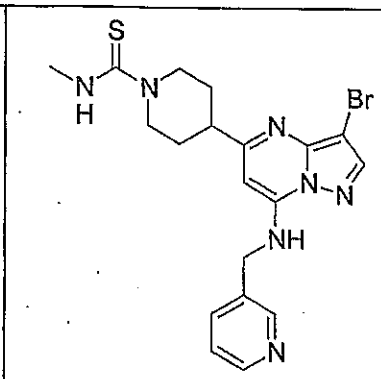
544			MH^+ = 455
545			MH^+ = 429
546			MH^+ = 455 m. p. =80.1 °C
547			MH^+ = 429 m. p. = 64.7 °C

548			MH^+ = 494 m. p. = 76.5 °C
549			MH^+ = 493 m. p. = 83.6 °C
550			MH^+ = 465 m. p. =207.5 °C
551			MH^+ = 494

552			MH^+ = 493
553			MH^+ = 465
554			MH^+ = 481 m. p. = 102.7 °C
555			MH^+ = 494 m. p. = 85.3 °C

556			MH^+ = 493 m. p. = 89.1 °C
557			MH^+ = 465 m. p. = 83.8 °C
558			Yield = quant. MH^+ = 443 m. p. = 98.3 °C (HCl salt)
559			MH^+ = 454

560			Yield = quant. MH^+ = 429 m. p. = 111.5 – 112. 6 °C
561			MH^+ = 460 m. p. = 122.7 °C
562			MH^+ = 460 m. p. = 95.4 °C
563			MH^+ = 460

564			MH^+ = 460 m. p. = 95.4 °C

Additional data for select examples given below.

Example 534: ^{1}H NMR (300MHz, $CDCl_3$) δ 8.66 – 8.62 (s, 1H), 8.62 – 8.58 (d, 1H), 7.95 (s, 1H), 7.72 – 7.68 (d, 1H), 7.36 – 7.31 (dd, 1H), 6.66 – 6.62 (t, 1H), 5.93 (s, 1H), 4.65 – 4.62 (d, 2H), 3.86 – 3.82 (d, 1H), 3.65 – 3.58 (m, 1H), 3.26 – 3.12 (dd, 4H), 3.02 – 2.80 (m, 3H), 2.10 – 2.00 (m, 1H), 1.67 – 1.57 (m, 3H).

Example 535: ^{1}H NMR (300MHz, $CDCl_3$) δ 8.66 – 8.62 (s, 1H), 8.62 – 8.58 (d, 1H), 7.95 (s, 1H), 7.72 – 7.67 (d, 1H), 7.36 – 7.30 (dd, 1H), 6.70 – 6.64 (t, 1H), 5.90 (s, 1H), 4.63 – 4.61 (d, 2H), 3.93 – 3.86 (m, 1H), 3.69 – 3.61 (m, 4H), 3.27 – 3.23 (m, 4H), 3.10 – 3.01 (dd, 1H), 2.93 – 2.84 (m, 2H), 2.08 – 2.03 (m, 1H), 1.90 – 1.57 (m, 4H).

Example 536: ^{1}H NMR (300MHz, $CDCl_3$) δ 8.67 (s, 1H), 8.62 – 8.58 (d, 1H), 7.96 (s, 1H), 7.72 – 7.68 (d, 1H), 7.36 – 7.30 (dd, 1H), 6.79 – 6.72 (t, 1H), 5.96 (s, 1H), 4.86 (br s, 2H), 4.66 – 4.63 (d, 2H), 3.89 – 3.73 (m, 2H), 3.55 – 3.32 (m, 2H), 3.00 – 2.89 (m, 1H), 2.10 – 1.97 (m, 2H), 1.70 – 1.53 (m, 2H).

Example 537: ^{1}H NMR (300MHz, $CDCl_3$) δ 8.66 (s, 1H), 8.62 – 8.58 (d, 1H), 7.98 (s, 1H), 7.77 – 7.76 (t, 1H), 7.72 – 7.69 (d, 1H), 7.63 – 7.59 (m, 1H), 7.56 (s, 1H), 7.36 – 7.29 (dd, 1H), 6.83 – 6.79 (t, 1H), 5.96 (s, 1H), 4.67 – 4.64 (d, 2H), 3.98 – 3.93 (dd, 1H), 3.79 – 3.68 (m, 2H), 3.37 – 3.28 (m, 1H), 3.03 – 2.94 (m, 1H), 2.12 – 1.99 (m, 1H), 1.76 – 1.56 (m, 3H).

Example 544: ^{1}H NMR (300MHz, $CDCl_3$) δ 8.66 – 8.62 (d, 1H), 8.61 – 8.58 (dd, 1H), 7.95 (s, 1H), 7.72 – 7.67 (d, 1H), 7.36 – 7.30 (dd, 1H), 6.80 – 6.62 (br s, 1H), 5.88 (s, 1H), 4.63 (s, 2H), 3.08 – 2.95 (m, 2H), 2.87 – 2.80 (m, 2H), 2.04 (m,

1H), 1.85 – 1.78 (m, 4H), 1.52 – 1.44 (m, 1H), 0.87 – 0.82 (m, 2H), 0.72 – 0.66 (m, 2H).

Example 545: ^{1}H NMR (300MHz, $CDCl_3$) δ 8.66 (s, 1H), 8.62 – 8.58 (br t, 1H), 7.97 (s, 1H), 7.73 – 7.68 (d, 1H), 7.36 – 7.30 (br t, 1H), 6.79 – 6.72 (br t, 1H), 5.96 (s, 1H), 4.64 (br s, 2H), 4.59 – 4.46 (br d, 1H), 3.95 – 3.74 (br m, 1H), 3.57 – 3.49 (dd, 1H), 3.10 – 3.01 (dd, 1H), 2.86 – 2.70 (m, 2H), 2.13 (s, 3H), 2.06 – 2.00 (m, 2H), 1.65 – 1.48 (m, 2H).

Example 551: ^{1}H NMR (300MHz, $CDCl_3$) δ 8.67 (s, 1H), 8.63 – 8.59 (d, 1H), 7.96 (s, 1H), 7.74 – 7.69 (d, 1H), 7.36 – 7.30 (dd, 1H), 6.69 – 6.64 (t, 1H), 5.95 (s, 1H), 4.67 – 4.63 (d, 2H), 3.85 3.65 (m,1H), 3.75 – 3.65 (m, 1H), 3.25 – 3.18 (dd, 1H), 3.03 – 2.90 (m, 2H), 2.81 (s, 6H), 2.03 – 1.95 (m, 1H), 1.89 – 1.68 (m, 3H).

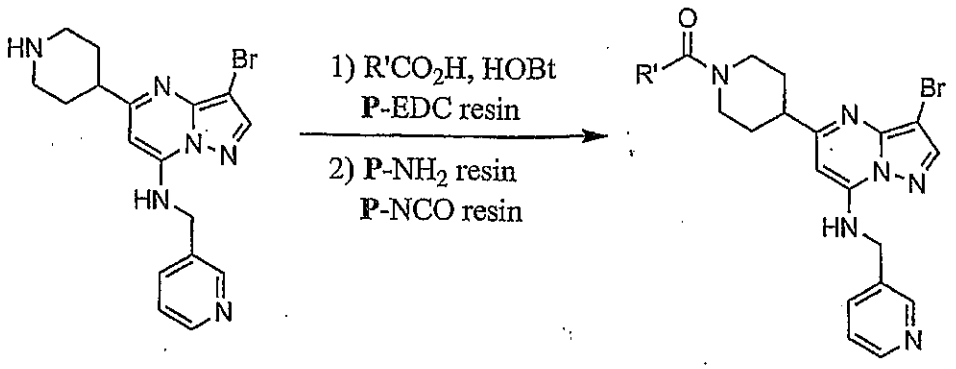
Example 552: ^{1}H NMR (300MHz, $CDCl_3$) δ 8.67 (s, 1H), 8.62 – 8.59 (d, 1H), 7.95 (s, 1H), 7.74 – 7.69 (d, 1H), 7.36 – 7.31 (dd, 1H), 6.67 – 6.60 (t, 1H), 5.98 (s, 1H), 4.67 – 4.63 (d, 2H), 3.92 – 3.86 (m, 1H), 3.85 – 3.75 (m, 1H), 3.40 – 3.30 (dd, 1H), 3.27 – 3.16 (m, 1H), 3.10 – 2.86 (m, 2H), 2.10 – 1.78(m, 3H), 1.40– 1.30 (d, 6H).

Example 553: ^{1}H NMR (300MHz, $CDCl_3$) δ 8.67 (s, 1H), 8.62 (br s, 1H), 7.96 (s, 1H), 7.74 – 7.69 (d, 1H), 7.36 – 7.31 (dd, 1H), 6.70 – 6.66 (t, 1H), 5.98 (s, 1H), 4.67 – 4.63 (d, 2H), 3.88 – 3.81 (m, 1H), 3.71 – 3.65 (m, 1H), 3.20 – 3.11 (dd, 1H), 3.02 – 2.91 (m, 1H), 2.90 – 2.80 (m, 4H), 2.01 – 1.80 (m, 3H).

Example 559: ^{1}H NMR (300MHz, $CDCl_3$) δ 8.66 –8.60 (d, 1H), 8.50 – 8.44 (dd, 1H), 8.01 (s, 1H), 7.93 (m, 1H), 7.48 – 7.40 (dd, 1H), 6.08 (s, 1H), 4.80 – 7.74 (s, 2H), 4.32 –4.19 (br d, 2H), 3.10 – 2.86 (m, 2H), 1.95 – 1.68 (m, 4H).

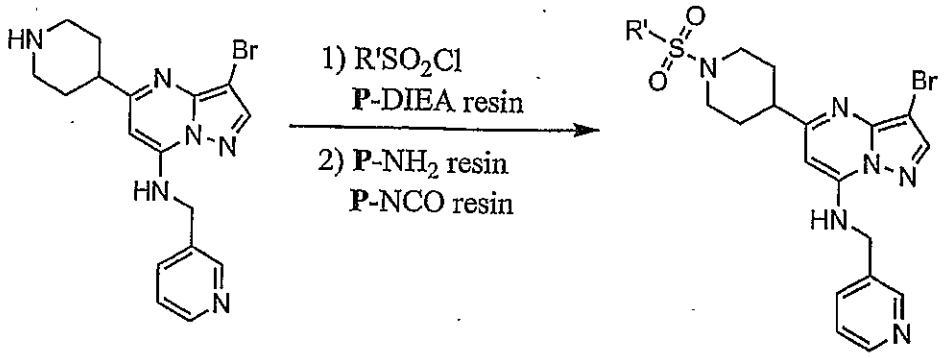
Example 563: ^{1}H NMR (300MHz, $CDCl_3$) δ 8.66 (s, 1H), 8.62 – 8.58 (d, 1H), 7.96 (s, 1H), 7.73 – 7.68 (d, 1H), 7.36 – 7.30 (dd, 1H), 6.96 – 6.86 (br s, 1H), 6.79 – 6.74 (t, 1H), 6.00 (s, 1H), 4.67 – 4.64 (d, 2H), 4.37 – 4.30 (dd, 1H), 4.22 – 4.13 (m, 1H), 3.97 – 3.86 (dd, 1H), 3.73 – 3.64 (m, 1H), 3.17 – 3.14 (d, 3H), 3.07 – 2.99 (m, 1H), 2.20 – 1.97 (m, 2H), 1.68 – 1.48 (m, 2H).

GENERAL PROCEDURE 1: Procedure for the amide formation parallel synthesis:



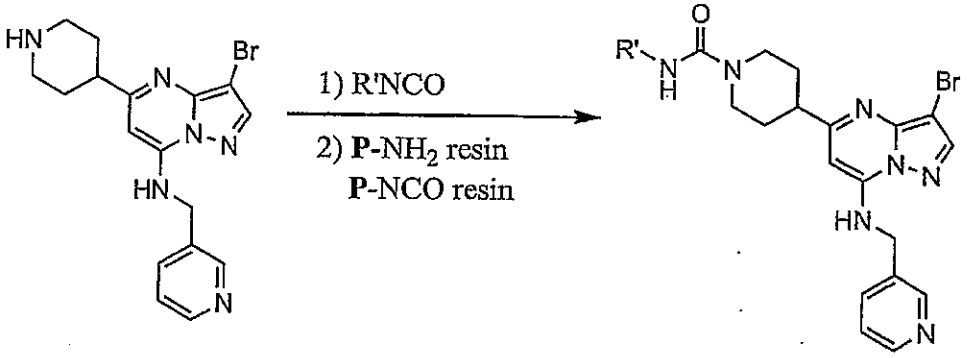
Parallel synthesis was conducted in polypropylene 96-well reaction blocks with removable top seal and fixed bottom seal. Each reaction well was fitted with a 20 micron polypropylene bottom frit and the maximum volume was 3 mL. Collection block was not fitted with bottom frit. To each reaction well was added a solution of an amine (0.021 mmol) dissolved in a DMF-THF-MeCN mixture (4:3:3 v/v, 0.95 mL), EDC resin (**P**-EDC, Polymer Laboratories Ltd., 43 mg, 0.063 mmol), 1-hydroxybenzotriazole (HOBt, 5.67 mg, 0.042 mmol) and a solution of a carboxylic acid in dimethylformamide (1 M, 0.0315 mL, 0.0315 mmol). The reaction mixture was agitated at room temperature for 16 h. The crude product solution was filtered into a reaction well loaded with trisamine resin (P-NH2, Argonaut Tech. Inc., 30 mg, 0.126 mmol) and isocyanate resin (P-NCO, Argonaut Tech. Inc., 35 mg, 0.063 mmol). The reaction mixture was agitated at room temperature for 16 h and filtered into the collection block. The product solution was evaporated under reduced pressure to afford the desired amide product.

GENERAL PROCEDURE 2: Procedure for the sulfonamide formation parallel synthesis



Parallel synthesis was conducted in polypropylene 96-well reaction blocks with removable top seal and fixed bottom seal. Each reaction well was fitted with a 20 micron polypropylene bottom frit and the maximum volume was 3 mL. Collection block was not fitted with bottom frit. To each reaction well was added a solution of an amine (0.021 mmol) dissolved in a DMF-THF-MeCN mixture (3:2:2 v/v, 0.95 mL), DIEA resin (**P**-DIEA, Argonaut Tech. Inc., 18 mg, 0.063 mmol) and a solution of a sulfonyl chloride in dimethylformamide (1 M, 0.0315 mL, 0.0315 mmol). The reaction mixture was agitated at room temperature for 16 h. The crude product solution was filtered into a reaction well loaded with trisamine resin (**P**-NH2, Argonaut Tech. Inc., 30 mg, 0.126 mmol) and isocyanate resin (**P**-NCO, Argonaut Tech. Inc., 35 mg, 0.063 mmol). The reaction mixture was agitated at room temperature for 16 h and filtered into the collection block. The product solution was evaporated under reduced pressure to afford the desired sulfonamide product.

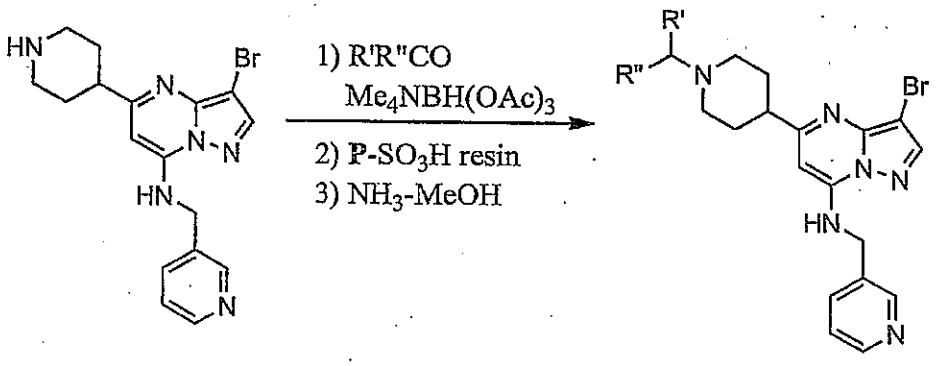
GENERAL PROCEDURE 3: Procedure for the urea formation parallel synthesis



Parallel synthesis was conducted in polypropylene 96-well reaction blocks with removable top seal and fixed bottom seal. Each reaction well was fitted with

a 20 micron polypropylene bottom frit and the maximum volume was 3 mL. Collection block was not fitted with bottom frit. To each reaction well was added a solution of an amine (0.021 mmol) dissolved in a DMF-MeCN mixture (1:1 v/v, 0.95 mL) and a solution of an isocyanate in dichloromethane (0.33 M, 0.126 mL, 0.042 mmol). The reaction mixture was agitated at room temperature for 16 h. The crude product solution was filtered into a reaction well loaded with trisamine resin (**P**-NH2, Argonaut Tech. Inc., 30 mg, 0.126 mmol) and isocyanate resin (**P**-NCO, Argonaut Tech. Inc., 35 mg, 0.063 mmol). The reaction mixture was agitated at room temperature for 16 h and filtered into the collection block. The product solution was evaporated under reduced pressure to afford the desired urea product.

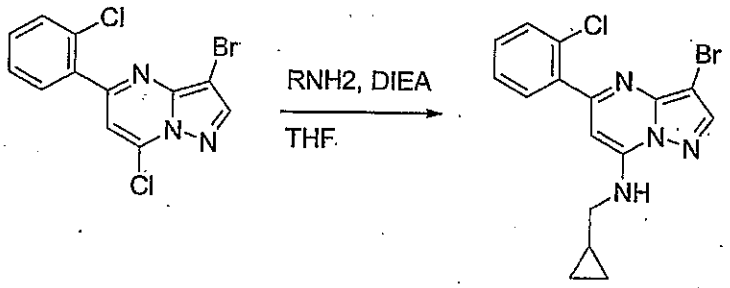
GENERAL PROCEDURE 4: Procedure for the reductive alkylation parallel synthesis



Parallel synthesis was conducted in polypropylene 96-well reaction blocks with removable top seal and fixed bottom seal. Each reaction well was fitted with a 20 micron polypropylene bottom frit and the maximum volume was 3 mL. Collection block was not fitted with bottom frit. To each reaction well was added a solution of an amine (0.021 mmol) dissolved in AcOH-DCE mixture (1:99 v/v, 0.5 mL), a solution of an aldehyde or ketone in dichloroethane (1 M, 0.147 mL, 0.147 mmol), and a solution of tetramethylammonium triacetoxyborohydride (11 mg, 0.042 mmol) dissolved in AcOH-DCE mixture 1:99 v/v, 0.5 mL). The reaction mixture was agitated at room temperature for 3 days. The crude product solution was filtered into a reaction well loaded with sulfonic acid resin Lanterns (**P**-SO_3H, MimotopesPty Ltd., 0.3 mmol). The reaction mixture was agitated at room temperature for 2 h and decanted. The product resin Lanterns

were washed with methanol (1 mL) for three times. A solution of ammonia in methanol (2 M, 1.2 mL) was added. The reaction mixture was agitated at room temperature for 30 min. and filtered into the collection block. The product solution was evaporated under reduced pressure to afford the desired tertiary amine product.

GENERAL PROCEDURE 5: Procedure for the parallel synthesis of 7,N-substituted pyrazolo[1,5a]pyrimidines

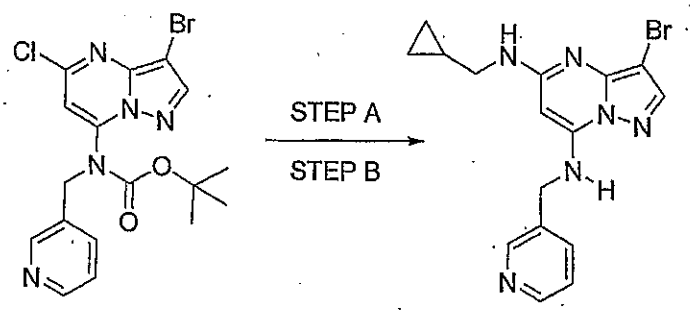


To 3-bromo-7-chloro-5-(2-chloro-phenyl)-pyrazolo[1,5-a]pyrimidine (9.0 mg, 0.03 mmol) in tetrahydrofuran were added di-*iso*-propylethylamine (12 μL, 0.07), followed by cyclopropylmethylamine (70 μL, .07 mmol; 1M solution in DMF). The reaction mixture was heated to 70 °C for 36 h and then cooled to rt. The mixture was treated with (**P**-NCO, Argonaut Tech. Inc 70 mg, 0.12 mmol), and **P**-CO_3^- (Argonaut Tech. Inc 70 mg, 0.24 mmol) and shaken at rt for 12-18 h. The solution was filtered and evaporated to dryness to provide the product. observed m/z 375.21.

GENERAL PROCEDURE 6: Procedure for the parallel synthesis of 5,N-substituted pyrazolo[1,5a]pyrimidines

General protocols:

Parallel synthesis was performed in a 96 well polypropylene blocks as described elsewhere. In the instance that heating was required, reactions were conducted in 2.5 mL glass tubes individually sealed with a polypropylene mat and heating achieved by a 96 well heat transfer block.



STEP A:

To the 3-bromo-5-chloro-7-N-Boc-alkylamino-pyrazolo[1,5-a]pyrimidine (17 mg, 0.04 mmol) in *p*-dioxane were added DIEA (9 μL, 0.05), followed by cyclopropyl-methylamine (80 μL, .08 mmol; 1M solution in isopropanol). The reaction mixture was heated to 90 °C for 36 h and then cooled to rt. The mixture was treated with P-NCO (Argonaut Tech. Inc. 70 mg, 0.12 mmol) and P-CO_3^- (Argonaut Tech. Inc. 70 mg, 0.24 mmol) and shaken at rt for 12-18 h. The solution was filtered and evaporated to dryness to provide the product.

STEP B(acidic):

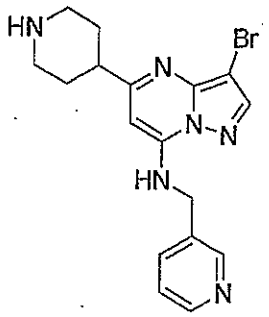
The product from STEP A was taken up in 35% TFA/DCM and agitated for 4 h followed by concentration under high vacuum. The residue was treated with 10% HCl(aq) in MeOH agitated for 2 h and then concentrated to give the desired product. . observed m/z 375.21.

STEP B(basic):

The product from step A was taken up in EtOH and treated with Ambersep® 900-OH ion exchange resin (Acros, 100mg), heated at reflux for 48 h with gently stirring. The reaction mixture was cooled to rt, filtered and concentrated to provide the desired product.

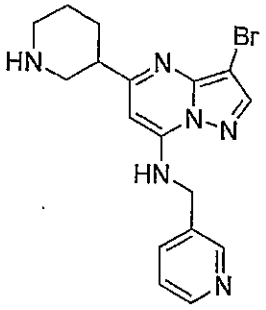
EXAMPLE 565:

By utilizing the procedure set forth in General Procedure 1 and the compound from Example 462 shown below, the compounds with the observed m/z shown in Table 43 were prepared.



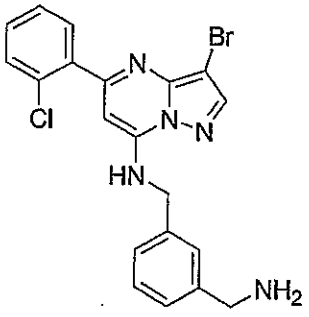
EXAMPLE 566:

By utilizing the procedure set forth in General Procedure 1 and the compound from Example 471 shown below, the compounds shown in Table 44 with the observed m/z were prepared.



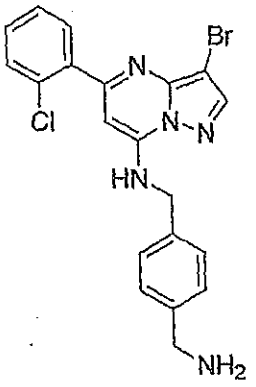
EXAMPLE 567:

By utilizing the procedure set forth in General Procedure 1 and the compound from Example 515 shown below, the compounds shown in Table 45 with the observed m/z were prepared.



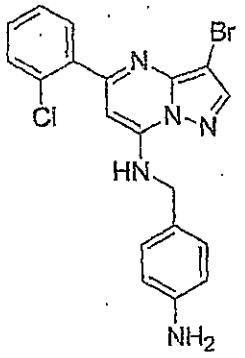
EXAMPLE 568:

By utilizing the procedure set forth in General Procedure 1 and the compound from Example 513 shown below, the compounds shown in Table 46 with the observed m/z were prepared.



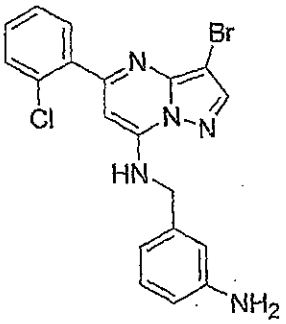
EXAMPLE 569:

By utilizing the procedure set forth in General Procedure 1 and the compound from Example 526 shown below, the compounds shown in Table 47 with the observed m/z were prepared.



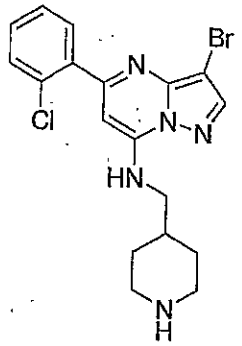
EXAMPLE 570:

By utilizing the procedure set forth in General Procedure 1 and the compound from Example 524 shown below, the compounds shown in Table 48 with the observed m/z were prepared.



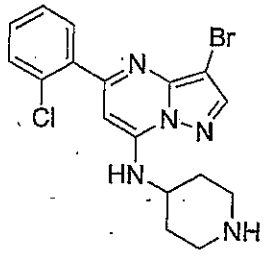
EXAMPLE 571:

By utilizing the procedure set forth in General Procedure 1 and the compound from Example 525 shown below, the compounds shown in Table 49 with the observed m/z were prepared.



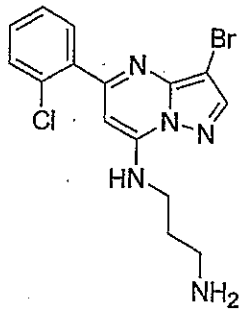
EXAMPLE 572:

By utilizing the procedure set forth in General Procedure 1 and the compound from Example 526.10 shown below, the compounds shown in Table 50 with the observed m/z were prepared.



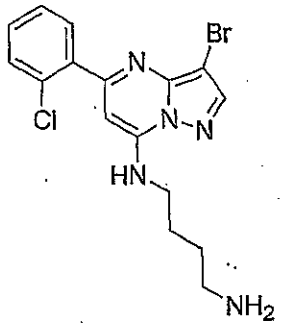
EXAMPLE 573:

By utilizing the procedure set forth in General Procedure 1 and the compound from Example 518 shown below, the compounds shown in Table 51 with the observed m/z were prepared.



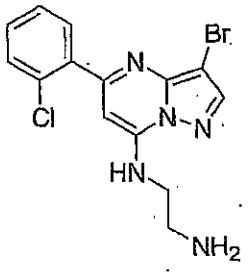
EXAMPLE 574:

By utilizing the procedure set forth in General Procedure 1 and the compound from Example 519 shown below, the compounds shown in Table 52 with the observed m/z were prepared.



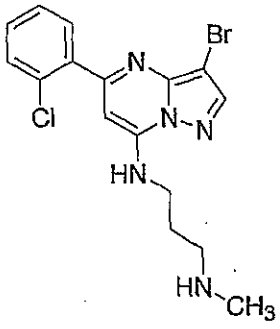
EXAMPLE 575:

By utilizing the procedure set forth in General Procedure 1 and the compound from Example 520 shown below, the compounds shown in Table 53 with the observed m/z were prepared.



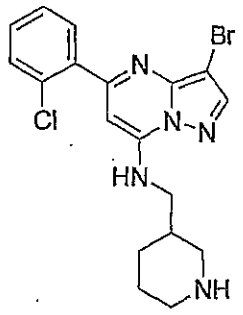
EXAMPLE 576:

By utilizing the procedure set forth in General Procedure 1 and the compound from Example 522 shown below, the compounds shown in Table 54 with the observed m/z were prepared.



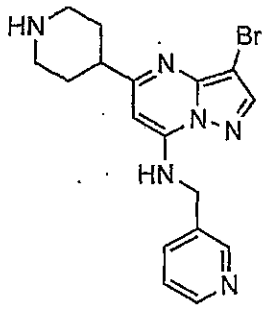
EXAMPLE 577:

By utilizing the procedure set forth in General Procedure 1 and the compound from Example 523 shown below, the compounds shown in Table 55 with the observed m/z were prepared.



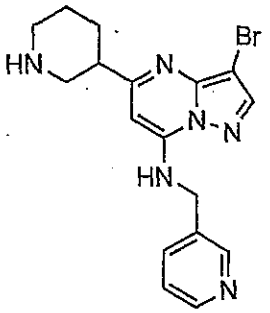
EXAMPLE 578:

By utilizing the procedure set forth in General Procedure 2 and the compound from Example 462 shown below, the compounds shown in Table 56 with the observed m/z were prepared.



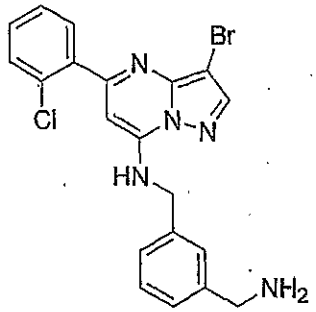
EXAMPLE 579:

By utilizing the procedure set forth in General Procedure 2 and the compound from Example 471 shown below, the compounds shown in Table 57 with the observed m/z were prepared.



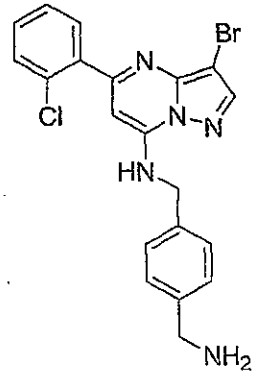
EXAMPLE 580:

By utilizing the procedure set forth in General Procedure 2 and the compound from Example 515 shown below, the compounds shown in Table 58 with the observed m/z were prepared.



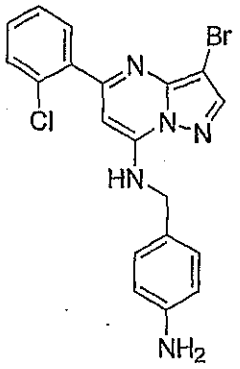
EXAMPLE 581:

By utilizing the procedure set forth in General Procedure 2 and the compound from Example 513 shown below, the compounds shown in Table 59 with the observed m/z were prepared.



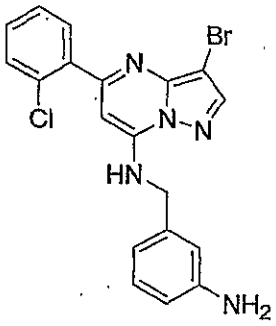
EXAMPLE 582:

By utilizing the procedure set forth in General Procedure 2 and the compound from Example 513 shown below, the compounds shown in Table 60 with the observed m/z were prepared.



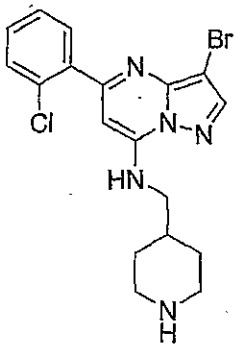
EXAMPLE 583:

By utilizing the procedure set forth in General Procedure 2 and the compound from Example 524 shown below, the compounds shown in Table 61 with the observed m/z were prepared.



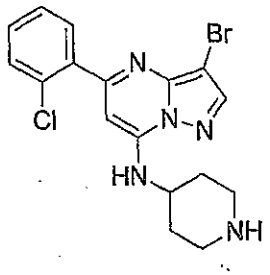
EXAMPLE 584:

By utilizing the procedure set forth in General Procedure 2 and the compound from Example 525 shown below, the compounds shown in Table 62 with the observed m/z were prepared.



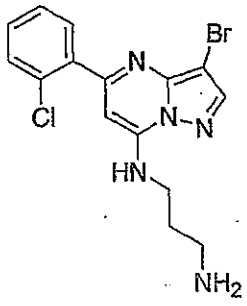
EXAMPLE 585:

By utilizing the procedure set forth in General Procedure 2 and the compound from Example 526.10 shown below, the compounds shown in Table 63 with the observed m/z were prepared.



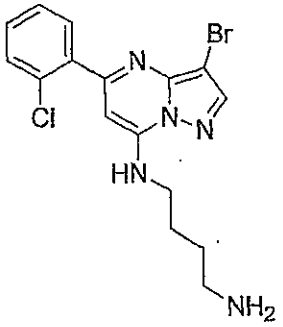
EXAMPLE 586:

By utilizing the procedure set forth in General Procedure 2 and the compound from Example 518 shown below, the compounds shown in Table 64 with the observed m/z were prepared.



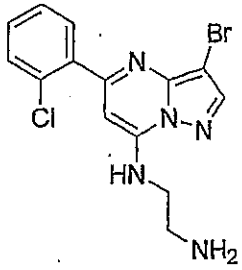
EXAMPLE 587:

By utilizing the procedure set forth in General Procedure 2 and the compound from Example 519 shown below, the compounds shown in Table 65 with the observed m/z were prepared.



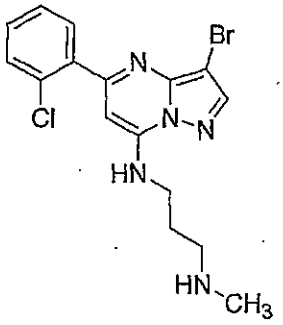
EXAMPLE 588:

By utilizing the procedure set forth in General Procedure 2 and the compound from Example 520 shown below, the compounds shown in Table 67 with the observed m/z were prepared.



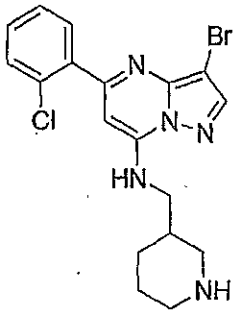
EXAMPLE 589:

By utilizing the procedure set forth in General Procedure 2 and the compound from Example 521 shown below, the compounds shown in Table 68 with the observed m/z were prepared.



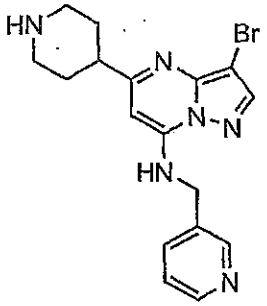
EXAMPLE 590:

By utilizing the procedure set forth in General Procedure 2 and the compound from Example 523 shown below, the compounds shown in Table 69 with the observed m/z were prepared.



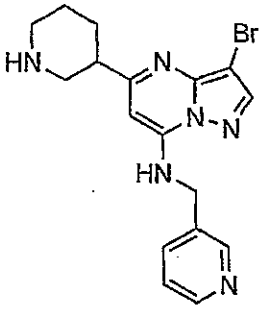
EXAMPLE 591:

By utilizing the procedure set forth in General Procedure 3 and the compound from Example 462 shown below, the compounds shown in Table 70 with the observed m/z were prepared.



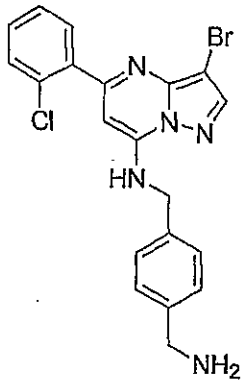
EXAMPLE 592:

By utilizing the procedure set forth in General Procedure 3 and the compound from Example 471 shown below, the compounds shown in Table 71 with the observed m/z were prepared.



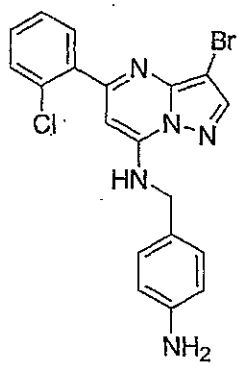
EXAMPLE 593:

By utilizing the procedure set forth in General Procedure 3 and the compound from Example 513 shown below, the compounds shown in Table 72 with the observed m/z were prepared.



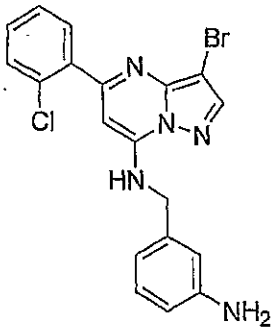
EXAMPLE 594:

By utilizing the procedure set forth in General Procedure 3 and the compound from Example 524 shown below, the compounds shown in Table 73 with the observed m/z were prepared.



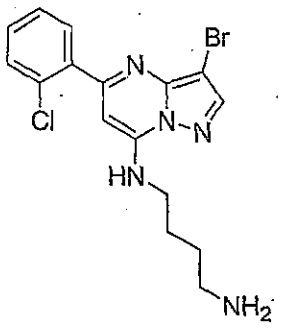
EXAMPLE 595:

By utilizing the procedure set forth in General Procedure 3 and the compound from Example 524 shown below, the compounds shown in Table 74 with the observed m/z were prepared.



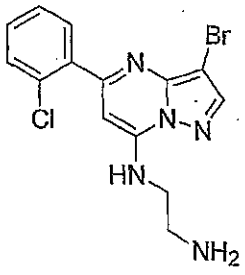
EXAMPLE 596:

By utilizing the procedure set forth in General Procedure 3 and the compound from Example 519 shown below, the compounds shown in Table 75 with the observed m/z were prepared.



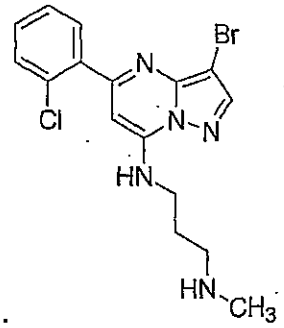
EXAMPLE 597:

By utilizing the procedure set forth in General Procedure 3 and the compound from Example 520 shown below, the compounds shown in Table 76 with the observed m/z were prepared.



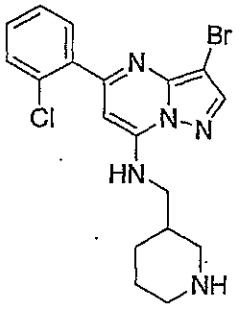
EXAMPLE 598:

By utilizing the procedure set forth in General Procedure 3 and the compound from Example 521 shown below, the compounds shown in Table 77 with the observed m/z were prepared.



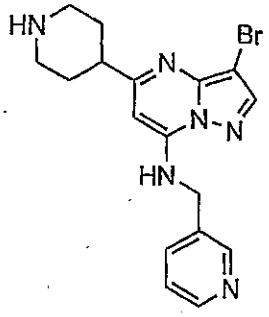
EXAMPLE 599:

By utilizing the procedure set forth in General Procedure 3 and the compound from Example 523 shown below, the compounds shown in Table 78 with the observed m/z were prepared.



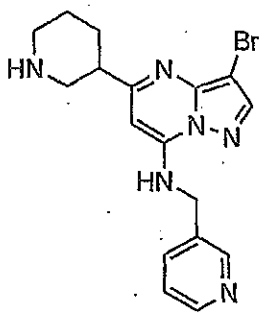
EXAMPLE 600:

By utilizing the procedure set forth in General Procedure 4 and the compound from Example 462 shown below, the compounds shown in Table 79 with the observed m/z were prepared.



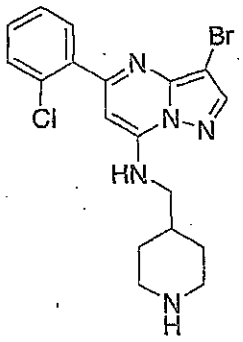
EXAMPLE 601:

By utilizing the procedure set forth in General Procedure 4 and the compound from Example 471 shown below, the compounds shown in Table 80 with the observed m/z were prepared.



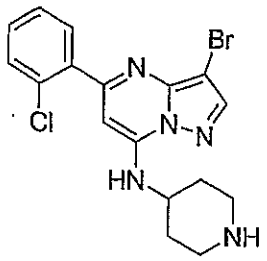
EXAMPLE 602:

By utilizing the procedure set forth in General Procedure 4 and the compound from Example 525 shown below, the compounds shown in Table 81 with the observed m/z were prepared.



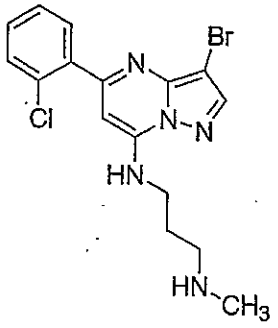
EXAMPLE 603:

By utilizing the procedure set forth in General Procedure 4 and the compound from Example 526.10 shown below, the compounds shown in Table 82 with the observed m/z were prepared.



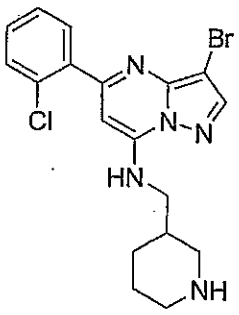
EXAMPLE 604:

By utilizing the procedure set forth in General Procedure 4 and the compound from Example 521 shown below, the compounds shown in Table 83 with the observed m/z were prepared.



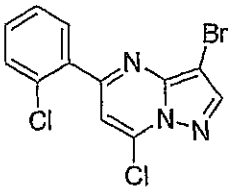
EXAMPLE 605:

By utilizing the procedure set forth in General Procedure 4 and the compound from Example 523 shown below, the compounds shown in Table 84 with the observed m/z were prepared.



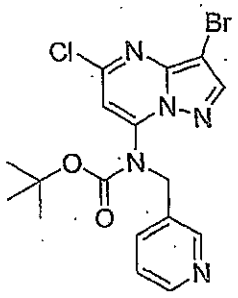
EXAMPLE 606:

By utilizing the procedure set forth in General Procedure 5 and the compound from Preparative Example 81 shown below, the compounds shown in Table 85 with the observed m/z were prepared.

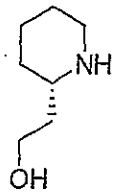


EXAMPLE 607:

By utilizing the procedure set forth in General Procedure 6 and the compound from Preparative Example 196, the compounds shown in Table 86 with the observed m/z were prepared.

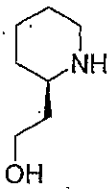


PREPARATIVE EXAMPLE 500



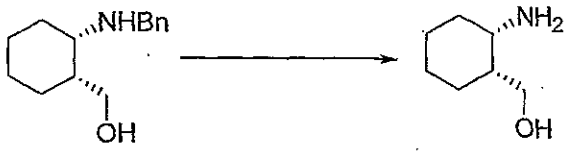
Piperidine-2-ethanol (127 g, 980 mmol) in 95% EtOH (260 mL) was added to *(S)*-(+)-camphorsulfonic acid (228.7 g, 1.0 eq.) in 95% EtOH (150 mL) and the resulting solution was warmed to reflux. To the warm solution was added Et_2O (600 mL) and the solution cooled to room temperature and let stand 3 days. The resulting crystals were filtered and dried *in vacuo* (25 g): mp 173-173 °C (lit. 168 °C). The salt was then dissolved in NaOH (3M, 100 mL) and stirred 2 hours and the resulting solution was extracted with CH_2Cl_2 (5 x 100 mL). The combined organics were dried over Na_2SO_4, filtered, filtered and concentrated under reduced pressure to give *(S)*-piperidine-2-ethanol (7.8 g) a portion of which was recrystallized from Et_2O: mp= 69-70 °C (lit. 68-69 °C); $[\alpha]_D$ = 14.09° ($CHCl_3$, c=0.2).

PREPARATIVE EXAMPLE 501



Bye essentially the same procedure set forth in Preparative Example 500 only substituting *(R)*-(-)-camphorsulfonic acid, *(R)*-piperidine-2-ethanol was prepared. (1.27 g): $[\alpha]_D$ = 11.3° ($CHCl_3$, c=0.2).

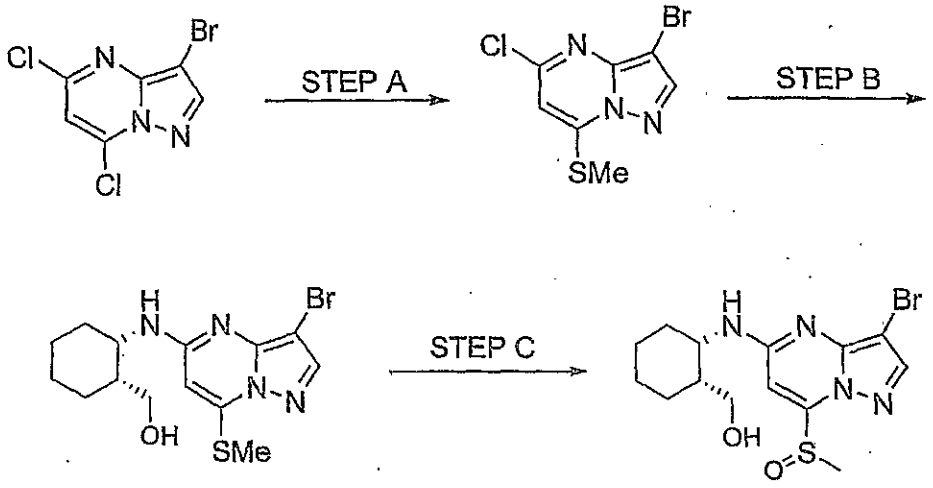
PREPARATIVE EXAMPLE 502



To pressure bottle charged with a solution of *cis*-(1*R*,2*S*)-(+)-2-(Benzylamino) cyclohexanemethanol (1g, 4.57 mmol) in MeOH (35 mL) was added 20% wt $Pd(OH)_2$ (0.3g, >50% wet) in one portion. The mixture was shaken under 50 psi of H_2 in a Parr hydrogenation apparatus for 12 h. The

mixture was purged to N_2 and was filtered through a pad of Celite. The pad was generously washed with MeOH (2 x 25 mL) and the resulting filtrate was concentrated under reduces pressure to afford 0.57g (97%) of a white solid. M+H = 130.

PREPARATIVE EXAMPLE 503



Step A:

To a solution of 3-Br adduct (1.1 g, 4.1 mmol) from Preparative Example 142 in THF (40 mL) at 0 °C was added CH_3SNa (0.32 g, 4.53 mmol) in one portion. The heterogenous mixture was stirred for 72 h at rt and the mixture was concentrated under reduced pressure. The crude product was partitioned between water (10 mL) and EtOAc (30 mL) and the layers were separated. The organic layer was washed with brine (1 x 10 mL) and dried (Na_2SO_4). The organic layer was filtered and concentrated under reduced pressure to afford 1.0 g (88%) of a yellow solid. mp 150-152 °C; M+H = 280. This material was taken onto Step B without further purification.

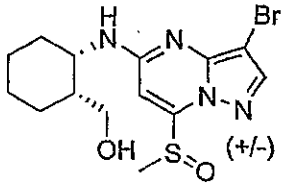
Step B:

To a solution of thiomethyl derivative (1.5 g, 5.37 mmol) from Step A in dioxane/DIPEA (15 mL/4 mL) at rt was added amino alcohol (1.3 g, 8.06 mmol) from Preparative Example 10. The mixture was heated at reflux for 48 h, cooled to rt, and concentrated under reduced pressure. The crude product was purified by flash chromatography using CH_2Cl_2/MeOH (30:1) as eluent to afford 1.8 g of product (90%) as a yellow crystalline solid. mp 167-169 °C; M+H = 373.

Step C:

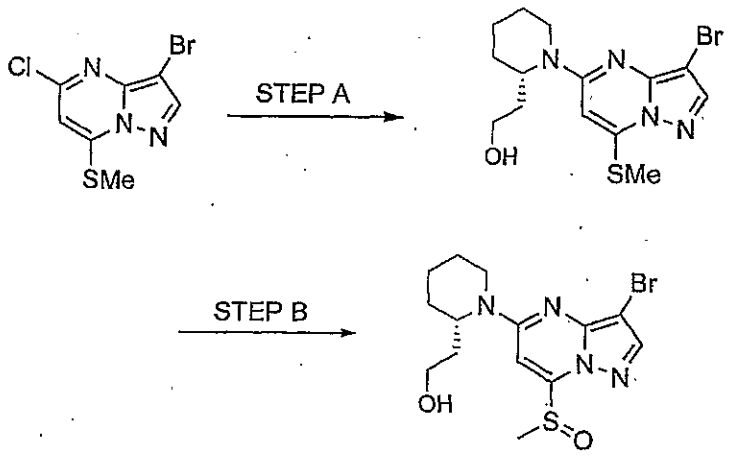
To a solution of thiomethyl derivative (2.2 g, 5.92 mmol) from Step B in CH_2Cl_2 (20 mL) at 0 °C was added MCPBA (1.53 g, 8.9 mmol) in one portion. The resulting mixture was stirred for 2h at 0 °C whereupon the mixture was diluted with CH_2Cl_2 (20 mL) and sat. aq. $NaHCO_3$ (15 mL). The layers were separated and the organic layer was washed with sat. aq. $NaHCO_3$ (15 mL) and brine (1 x 15 mL). The organic layer was dried (Na_2SO_4), filtered, and concentrated under reduced pressure to afford 2.0 g of a brown solid (87%). mp 181-183 °C; M+H = 388.

PREPARATIVE EXAMPLE 504



The title compound (racemic) was prepared according to the procedure set forth in Preparative Example 503 except substituting the commercially available *cis*-hydroxymethyl-1-cyclohexylamine hydrochloride in Step B.

PREPARATIVE EXAMPLE 505



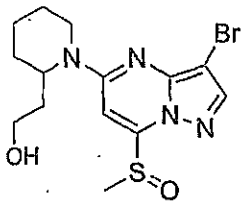
Step A:

Treatment of thiomethyl derivative (2.0 g, 7.2 mmol) from Step A of Preparative Example 503 with (*S*)-piperidine-2-ethanol (1.2 g, 9.3 mmol) from Preparative Example 500 under the identical conditions as described in Step B of Preparative Example 503, 0.90 g (34%) of the title compound was prepared semisolid. mp 173-175 °C. M+H = 372.

Step B:

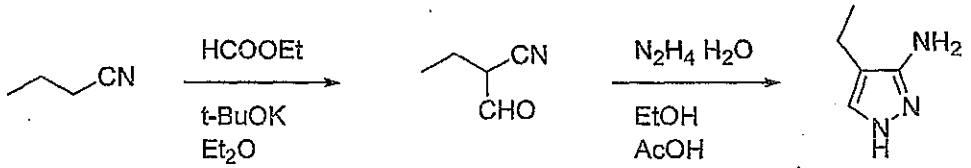
Following the procedure from Step C in Preparative Example 503, the thiomethyl derivative (0.30 g, 0.81 mmol) was treated with MCPBA (0.21 g, 1.2 mmol) to afford 0.31 g (99%) the title compound as a yellow viscous oil. M+H = 388.

PREPARATIVE EXAMPLE 506



The title compound (racemic) was prepared according to the procedure set forth in Preparative Example 505 except substituting the commercially available piperidine-2-ethanol. M+H = 388.

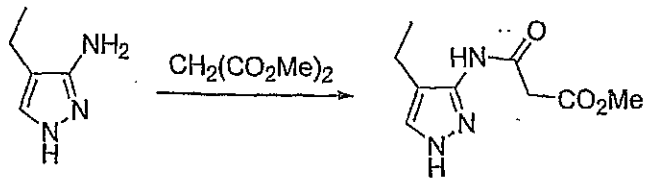
PREPARATIVE EXAMPLE 507



t-BuOK (112.0g, 1.00 mol) was stirred under N_2 in dry Et_2O (3.0 L) in a 5 L flask equipped with an addition funnel. A mixture of butyronitrile (69.0 g, 1.00 mol) and ethylformate (77.7 g, 1.05 mol) was added dropwise during 3 hrs, the reaction mixture was then stirred overnight at room temperature. The mixture was cooled to 0°C, AcOH (57 mL) was added, the mixture was filtered, and the solid was washed with Et_2O (500 mL). The combined filtrates were evaporated at room temperature on a rotovap to give pale yellow oil (95.1g).

The oil was dissolved in dry EtOH (100 mL), 99% hydrazine monohydrate (48 mL) was added, then AcOH (14 mL) was added, and the mixture was refluxed under N_2 overnight. The solvents were evaporated and the resulting oil was chromatographed on silicagel with CH_2Cl_2:7N NH_3 in MeOH. 22.4 g (20%) of 3-amino-4-ethylpyrazole was obtained as clear oil that solidified upon standing.

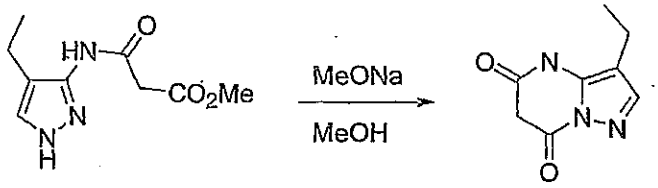
PREPARATIVE EXAMPLE 508



Step A:

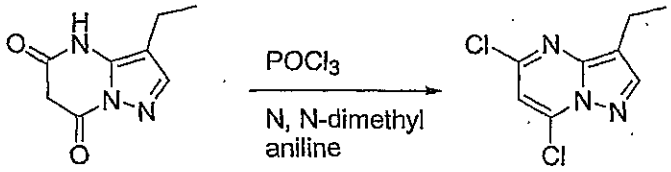
The pyrazole from Preparative Example 507 (9.80g) and dimethylmalonate (45 mL) were stirred and refluxed under N_2 for 3 hrs. The excess of dimethylmalonate was evaporated in a vacuum and the residue was chromatographed with 15:1 CH_2Cl_2:MeOH to yield pale yellow solid (10.6 g, 57%). LCMS: $MH^+ = 212$.

Step B:



Dry MeOH (200 mL) was added under N_2 to a mixture of the amide from Setp A (11.9g, 56.4 mmol) and sodium methoxide (4.57g, 84.6 mmol). The mixture was stirred and refluxed under N_2 for 5 hrs, cooled to rt, and conc. HCl (20 mL) was added. The solvents were evaporated and the residue was suspended in H_2O (300 mL). The solid was filtered off, washed on filter with 2x300 mL of H_2O, and dried in a vacuum at 100°C. 7.40g (73%) of cream-colored solid was obtained. LCMS: $MH^+ = 180$.

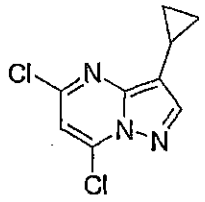
Step C:



$POCl_3$ (100 mL) and N, N-dimethylaniline (20 mL) were added under N_2 to the diketone from Step B (7.70 g), and the mixture was stirred and refluxed for 20 hrs under N_2. Then it was cooled to rt, carefully poured onto 1 L of crushed ice, and extracted with EtOAc (2x500 mL). The extracts were washed with H_2O (500 mL), dried over Na_2SO_4, filtered, and the solvent was evaporated. The

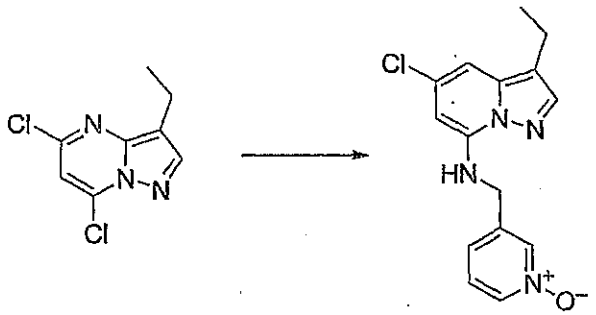
residue was chromatographed with CH_2Cl_2 to yield pale yellow solid (8.20 g, 90%). LCMS: MH^+ = 216.

PREPARATIVE EXAMPLE 508.10



By essentially the same procedure set forth in Preparative Example 508, only substituting the compound from Preparative Example 1, the above compound was prepared. LCMS: MH^+ = 228.

PREPARATIVE EXAMPLE 509

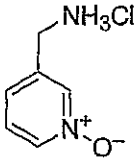
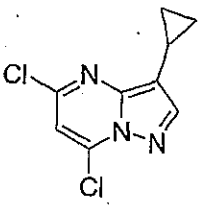
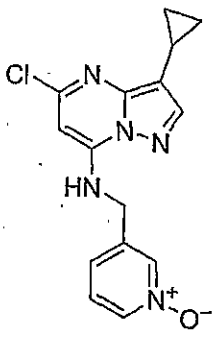
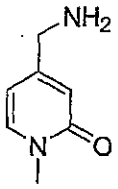
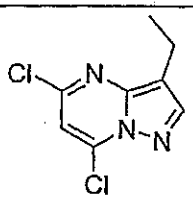
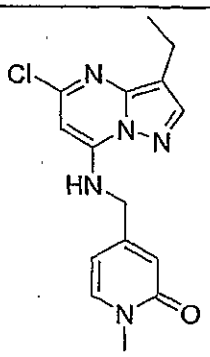
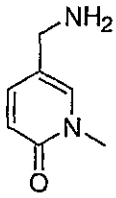
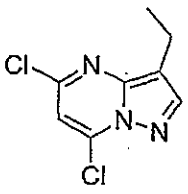
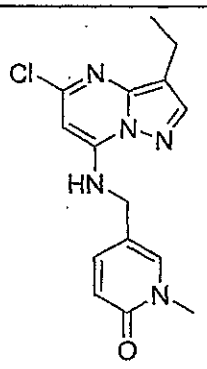
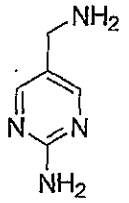
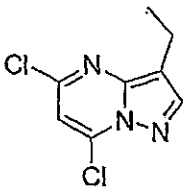
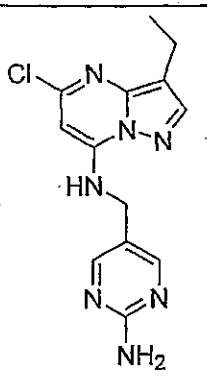


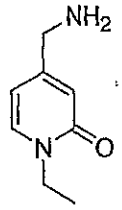
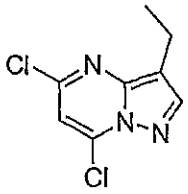
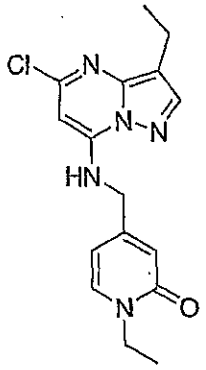
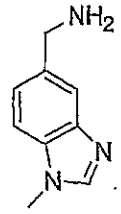
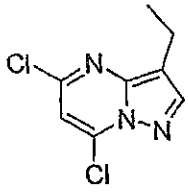
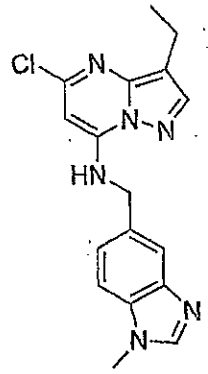
A mixture of the dichloride from Preparative Example 508 (3.13g, 14.5 mmol), the amine.HCl from Preparative Example (3.00g, 18.9 mmol), DIPEA (7.5 mL), and dry NMP (40 mL) plus dry dioxane (40 mL) was stirred at 60°C for 4 days under N_2. The solvents were then distilled off in a vacuum and the residue was chromatographed with 6:1 EtOAc: MeOH and then rechromatographed with 12:1 CH_2Cl_2:MeOH. So obtained solid was suspended in H_2O (100 mL), filtered, washed on filter with H_2O (2x100 mL), and dried in a vacuum. Pale rose solid (2.37g, 54%) was obtained. M+H = 304.

Preparative Examples 510-516

By essentially the same procedure set forth in Preparative Example 509 only substituting the amines in Column 2 of Table 500 and the chlorides shown in Column 3 of Table 500, the compounds shown in Column 4 of Table 500 were prepared.

TABLE 500

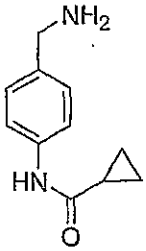
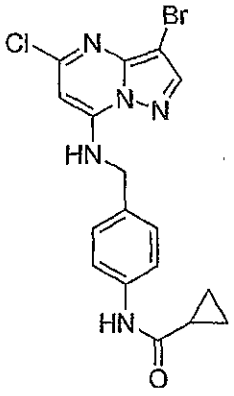
Prep. Ex.	Column 2	Column 3	Column 4	CMPD
510				M+H = 316
512				M+H = 318
513				M+H = 318
514				

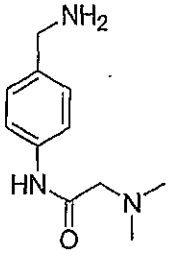
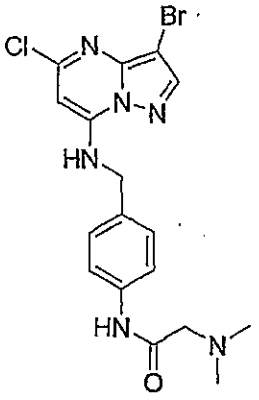
515				M+H = 332
516				

PREPARATIVE EXAMPLE 517:

By essentially the same procedure set forth in Preparative Example 184 only substituting the amines in Column 2 of Table 501, the compounds shown in Column 3 of Table 501 were prepared.

TABLE 501

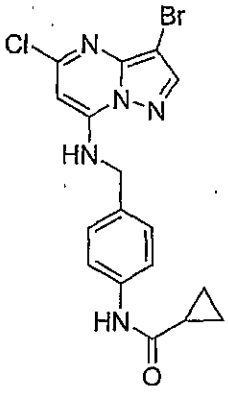
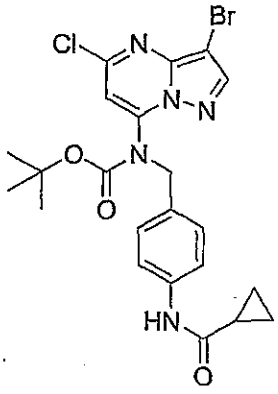
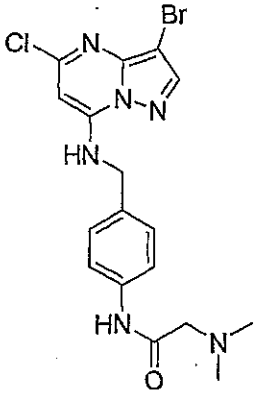
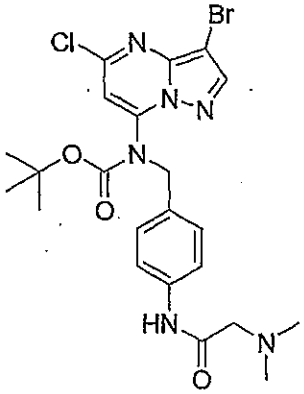
Prep. Ex.	Column 2	Column 3	CMPD
518			M+H= 422.1

519			

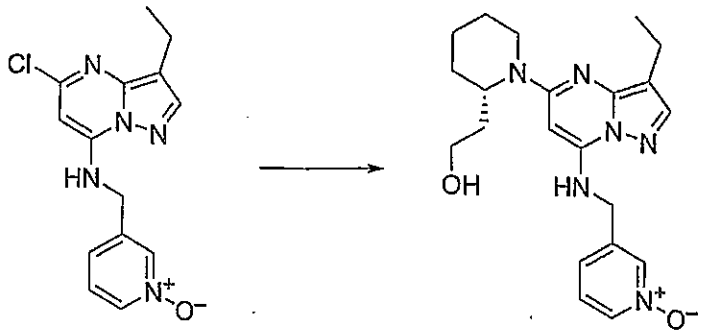
PREPARATIVE EXAMPLE 520-521:

By essentially the same procedure set forth in Preparative Example 192 only substituting the compounds in Column 2 of Table 502, the compounds shown in Column 3 of Table 502 were prepared.

TABLE 502

Prep. Ex.	Column 2	Column 3	CMPD
520			M+H= 522.1
521			M+H= 539.1

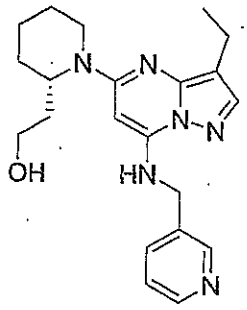
EXAMPLE 1000:



A mixture of the compound prepared in Preparative Example 509 (1.50 g, 4.94 mmol) with the aminoalcohol from Preparative Example 500 (1.91 g, 14.8 mmol) in dry NMP (3 mL) was stirred under N_2 at 160°C for 48 hr. The NMP was distilled off in a vacuum and the residue was chromatographed first with 5:1 EtOAc:MeOH, then the crude product was rechromatographed with 10:1 CH_2Cl_2:MeOH. White solid (460 mg, 24%) was obtained. LCSM: MH^+ = 397; mp = 113-115 °C.

EXAMPLE 1001:

Major side product isolated (540 mg, 29%) was deoxygenated product (LCMS: MH^+ = 381; mp = 49-52 °C:

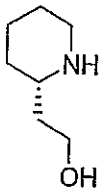
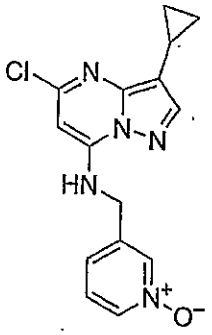
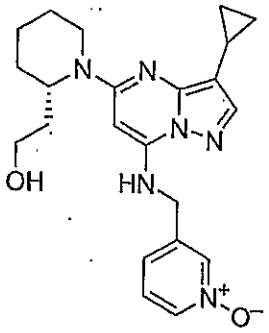
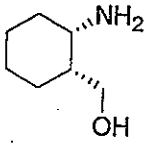
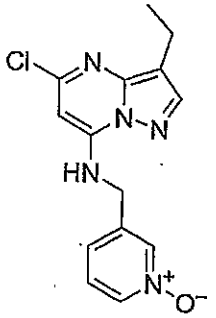
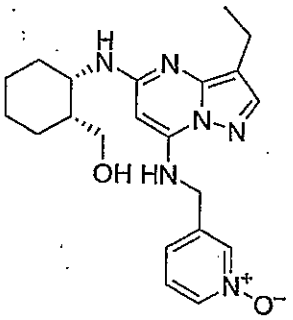
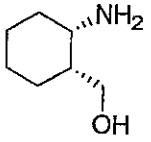
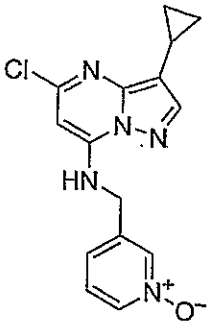
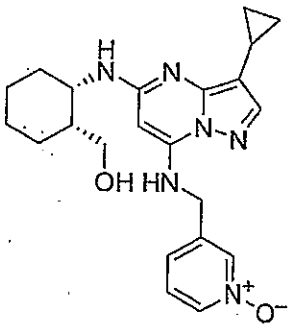
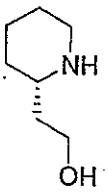
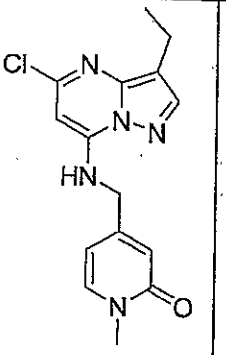
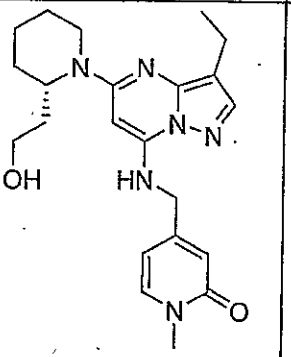


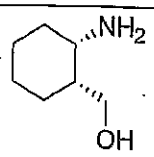
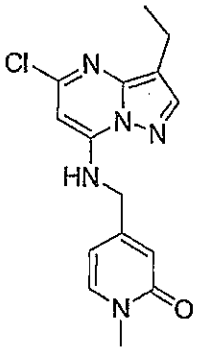
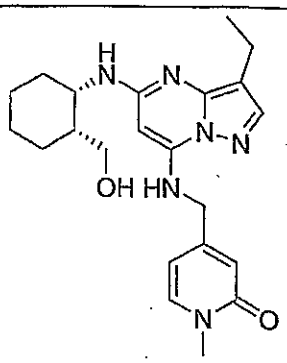
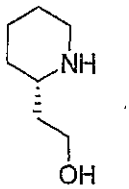
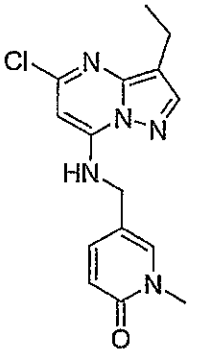
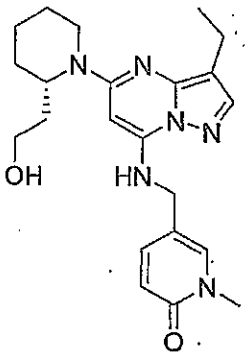
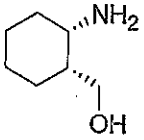
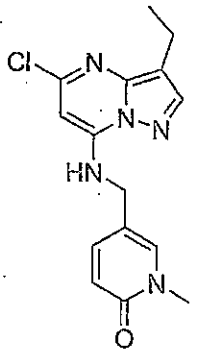
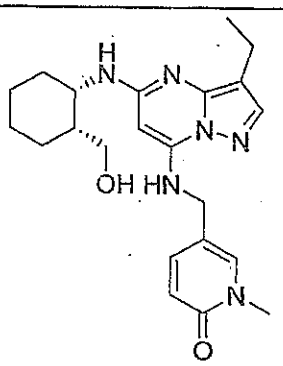
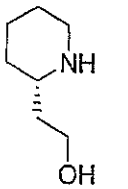
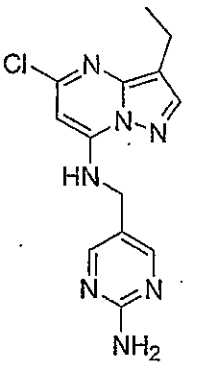
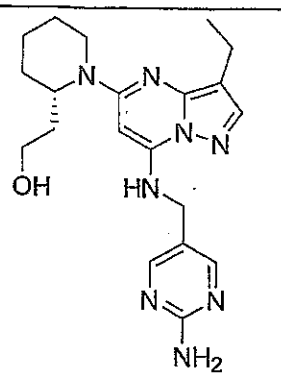
EXAMPLES 1002 – 1014:

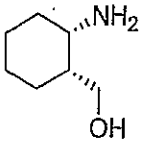
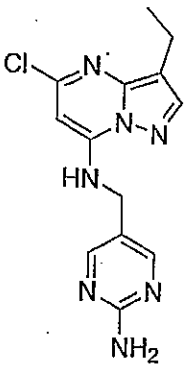
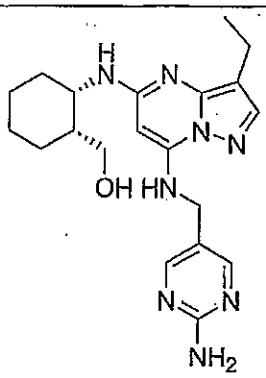
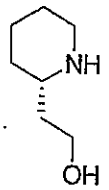
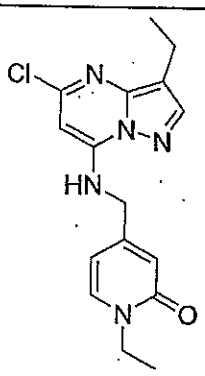
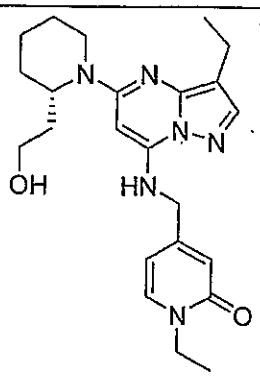
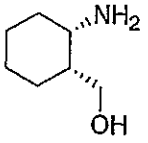
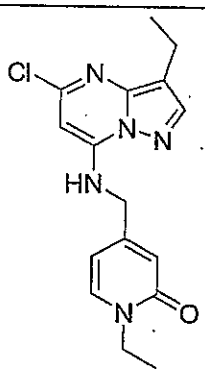
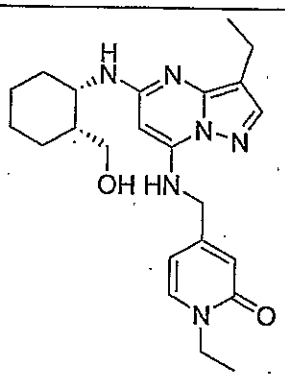
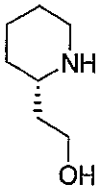
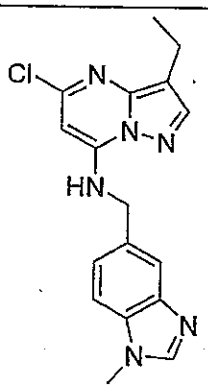
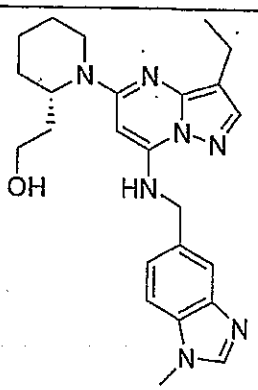
By essentially the same procedure set forth in Example 1000 only substituting the amines in Column 2 of Table 1000 and the chlorides in Column 3 of Table 1000 the compounds in column 4 of Table 1000 were prepared.

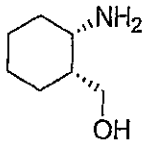
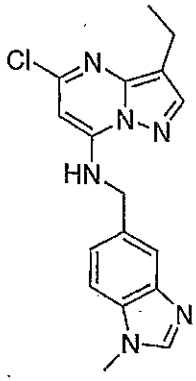
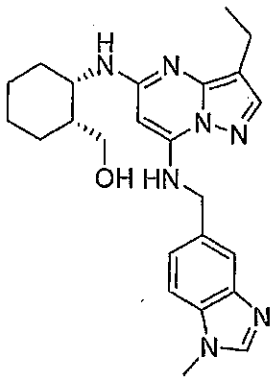
TABLE 1000

Ex.	Column 2	Column 3	Column 4	CMPD

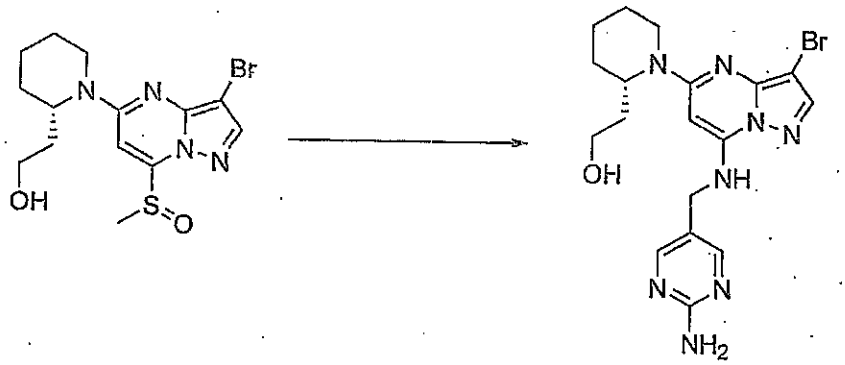
1002				MH^+ = 409; mp = 165-171 °C
1003				MH^+ = 397; mp = 219-221 °C
1004				MH^+ = 409; mp = 138-142 °C
1005				MH^+ = 411; mp = 194-196 °C

1006				MH^+ = 411; mp = 118-120 °C
1007				MH^+ = 411; mp = 85-87 °C
1008				MH^+ = 411; mp = 105-108 °C
1009				MH^+ = 397; mp = 173-177 °C

1010				MH^+ = 397; mp = 169-173 °C
1011				MH^+ = 425
1012				MH^+ = 425; mp = 232-234 °C
1013				

1014				

EXAMPLE 1015:



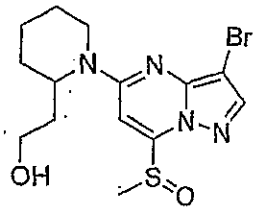
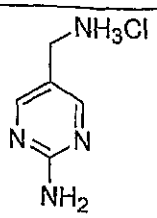
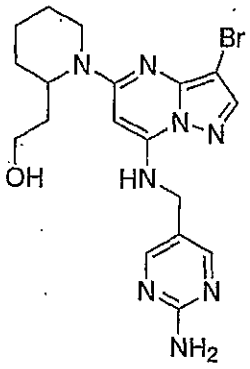
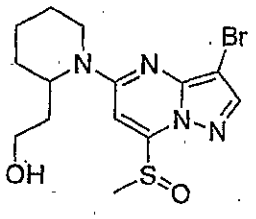
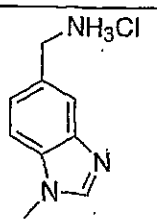
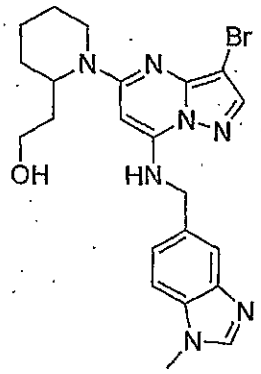
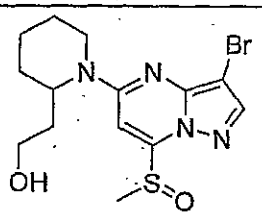
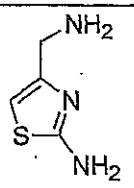
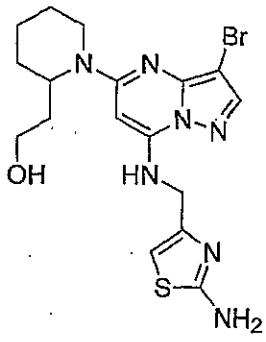
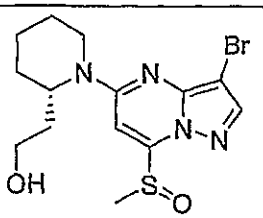
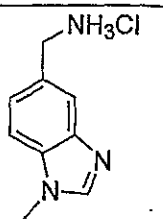
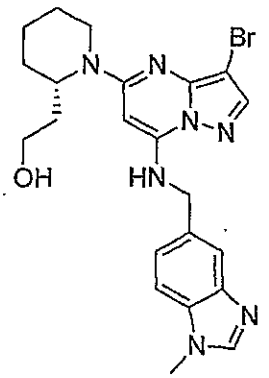
To a solution of sulfoxide from Preparative Example 505 (0.10 g, 0.28 mmol) in *n*-BuOH in a sealed tube was added Et_3N (0.13 mL, 1.0 mmol) followed by the amine dihydrochloride (0.13 g, 0.65 mmol) from Preparative Example 216. The tube was sealed and was heated to 100 °C, cooled to room temperature, and was concentrated under reduced pressure. The crude residue was purified by preparative TLC (6 x 1000 μM) eluting with CH_2Cl_2/MeOH (20:1) to afford 50 mg (40 %) of a pale white solid. mp 182-185 °C; M+H = 446.

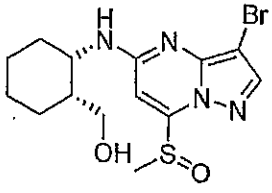
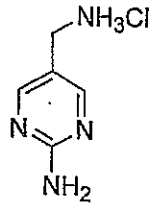
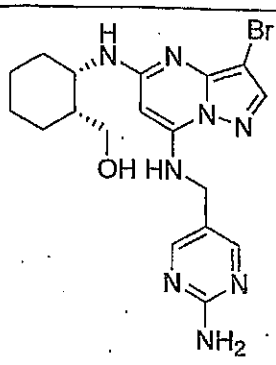
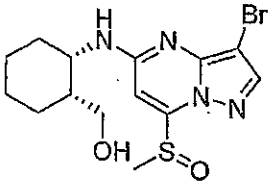
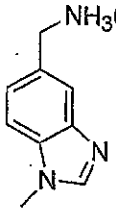
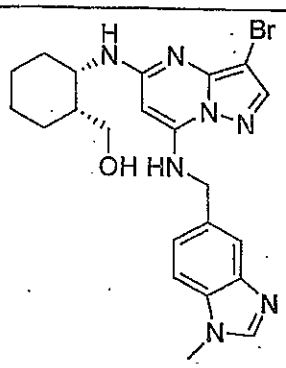
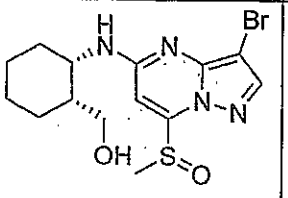
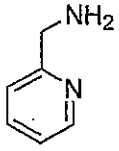
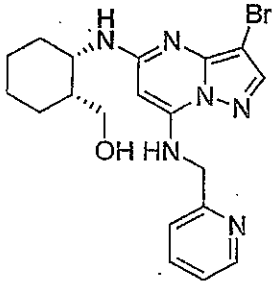
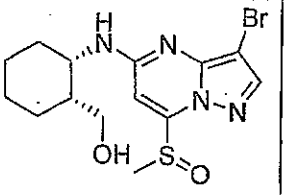
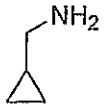
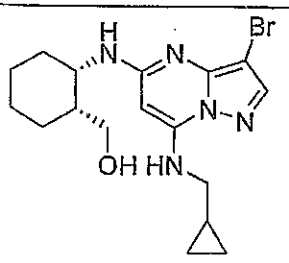
EXAMPLES 1016-1026:

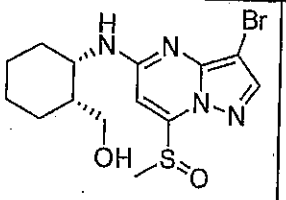
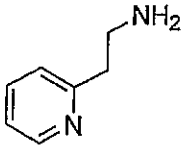
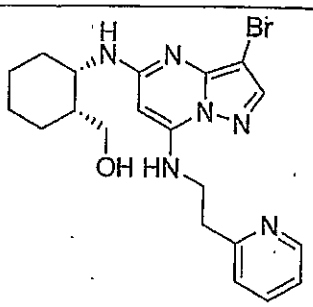
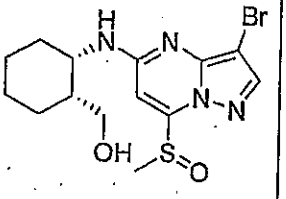
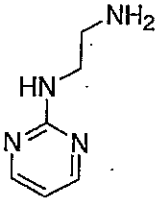
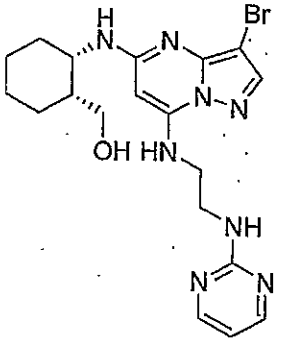
By essentially the same procedure set forth in Example 1015 only substituting the sulfoxide shown in Column 2 of Table 1001 and the amine in Column 3 of Table 1001, the compounds shown in Column 4 of Table 1001 were prepared.

TABLE 1001

Ex.	Column 2	Column 3	Column 4	CMPD

1016				mp = 182-185 °C; M+H = 448
1017				mp = 187-189 °C; M+H = 445
1018				mp = 139-143 °C; M+H = 453
1020				mp = 186-189 °C; M+H = 485

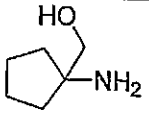
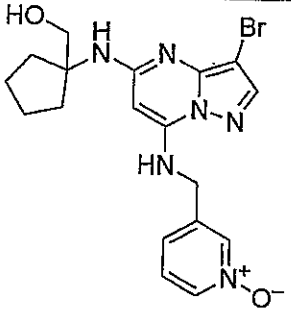
1021				mp = 154-157 °C; M+H = 448
1022				mp = 103-105 °C; M+H = 485
1023	 (±)		 (±)	mp = 203-205 °C; M+H = 432
1024	 (±)		 (±)	mp = 210-212 °C; M+H = 395

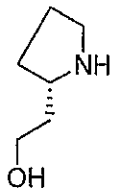
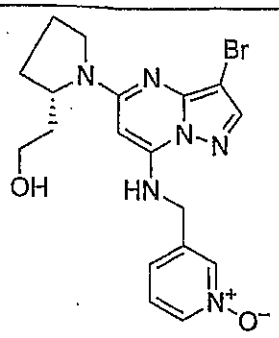
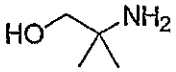
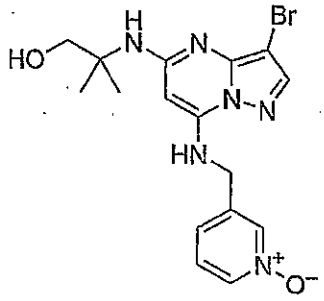
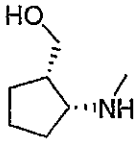
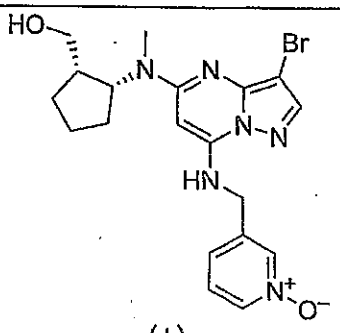
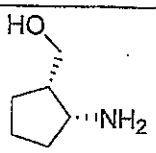
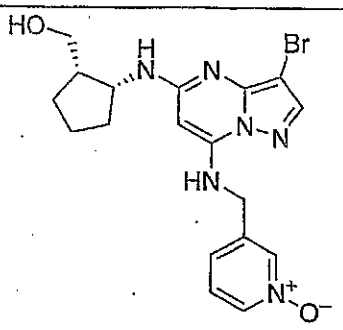
1025	 (±)		 (±)	mp = 82-84 °C; M+H = 446
1026	 (±)		 (±)	mp = 86-90 °C; M+H = 462

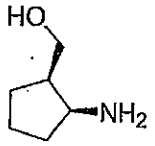
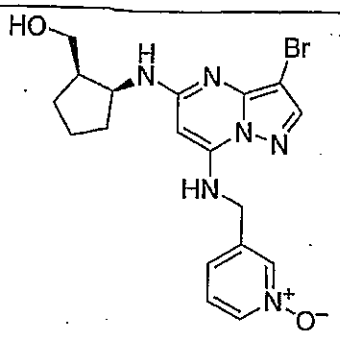
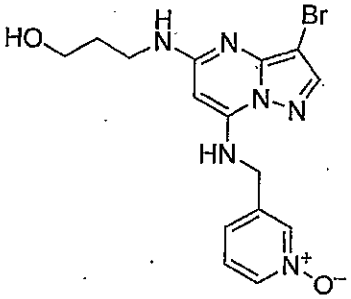
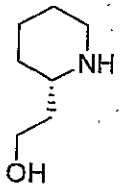
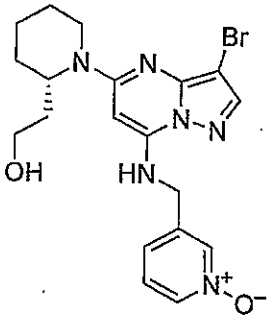
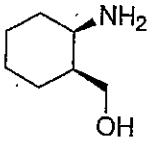
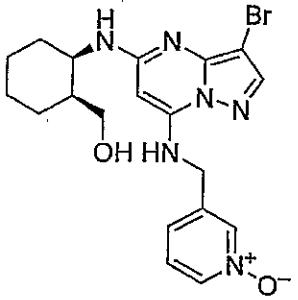
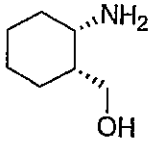
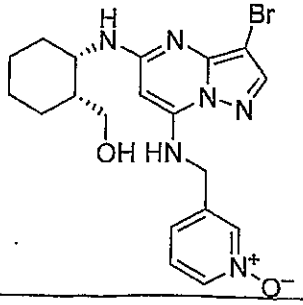
EXAMPLES 1027-1038:

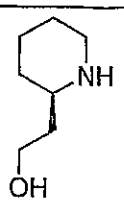
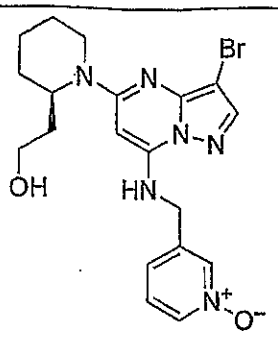
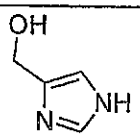
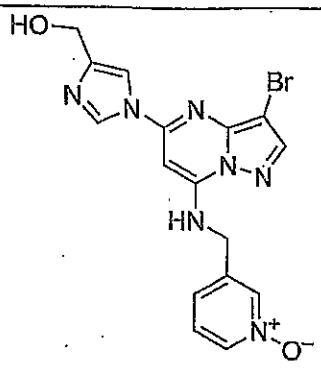
By essentially the same conditions set forth in Example 341, Steps A and B only substituting the amines in Column 2 of Table 1002 and the compound prepared in Preparative Example 193.10, the compounds in Column 4 of Table 1002 were prepared.

TABLE 1002

Ex.	Column 2	Column 4	CMPD
1027			mp = 160-163 °C; M+H = 434

1028			mp = 122-124 °C; M+H = 434
1029			mp = 153-156 °C; M+H = 408
1030	 (±)	 (±)	mp = 170-174 °C; M+H = 448
1031			mp = 166-169 °C; M+H = 434

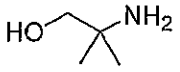
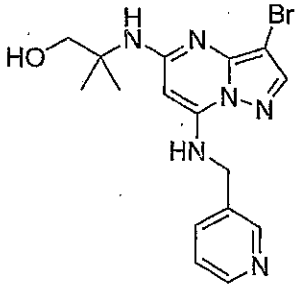
1032			mp = 167-168 °C; M+H = 434
1033			MH^+ = 393
1034			mp = 157-160 °C; M+H = 447
1035			mp = 164-168 °C; M+H = 448
1036			mp = 165-168 °C; M+H = 448

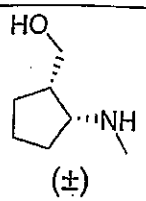
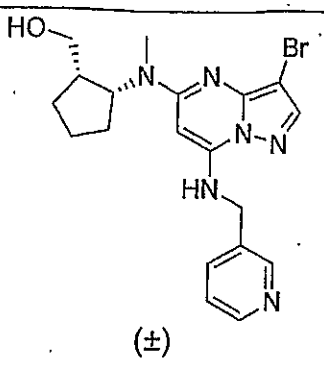
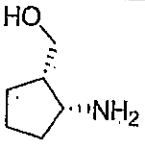
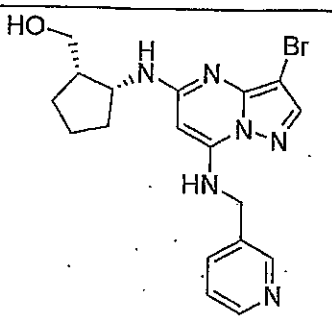
1037			mp = 131-135 °C; M+H = 447
1038			

EXAMPLES 1039-1041:

By essentially the same procedure set forth in Example 340 only substituting the amines in Column 2 of Table 1003, the compounds shown in Column 4 of Table 1003 were prepared.

TABLE 1003

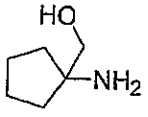
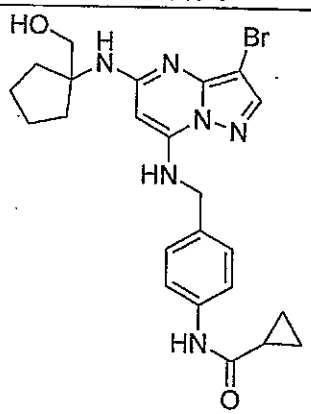
Ex.	Column 2	Column 4	CMPD
1039			mp = 210-212 °C; M+H = 392

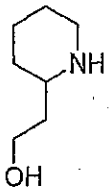
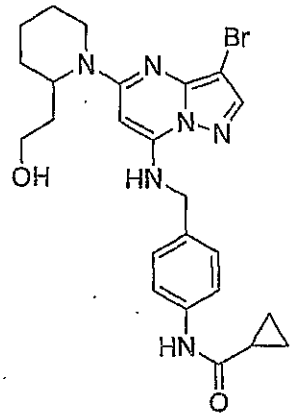
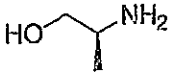
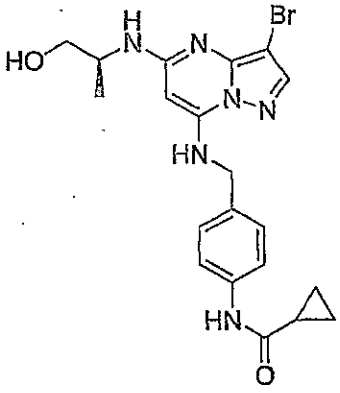
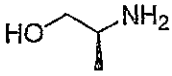
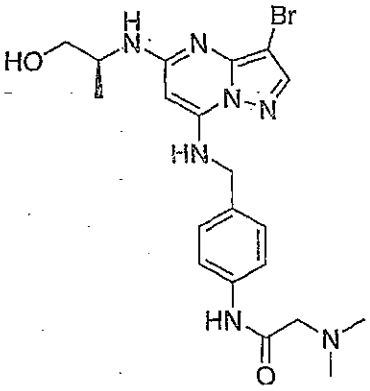
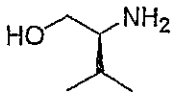
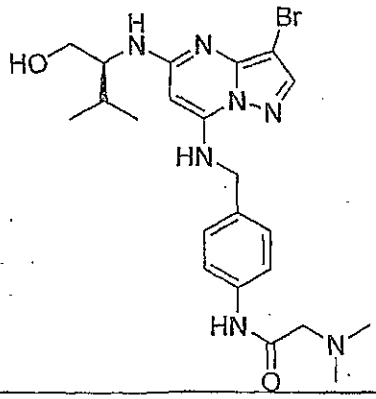
1040	 (±)	 (±)	mp = 128-130 °C; M+H = 432
1041			mp = 148-151 °C; M+H = 18

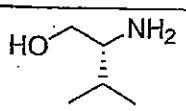
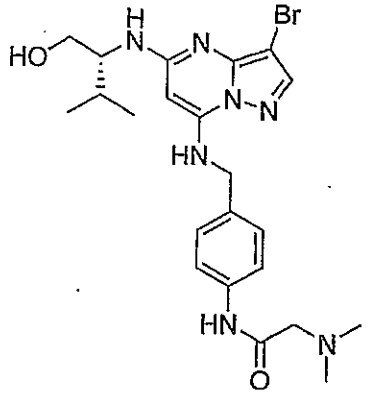
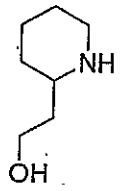
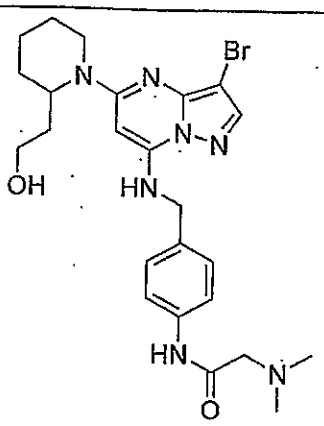
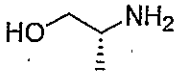
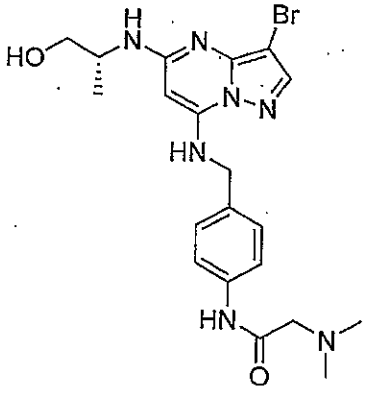
EXAMPLES 1042-1057:

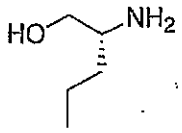
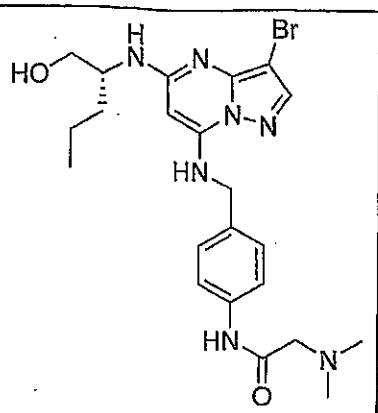
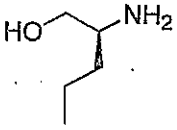
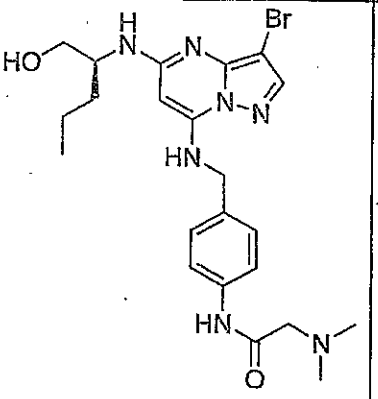
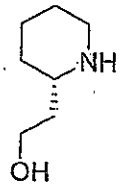
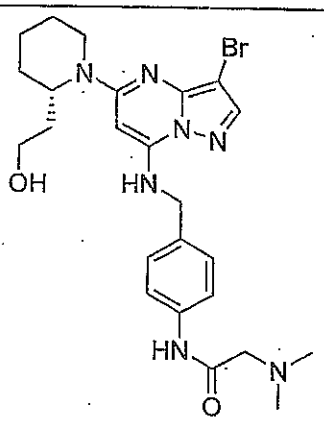
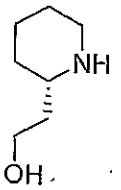
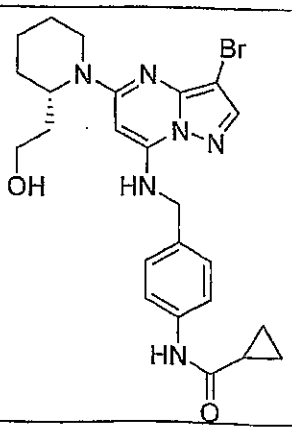
By essentially the same procedure set forth in Example 340 only using the appropriate 5-chloroderivative and substituting the amines in Column 2 of Table 1004, the compounds shown in Column 4 of Table 1004 were prepared.

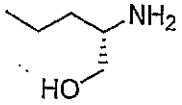
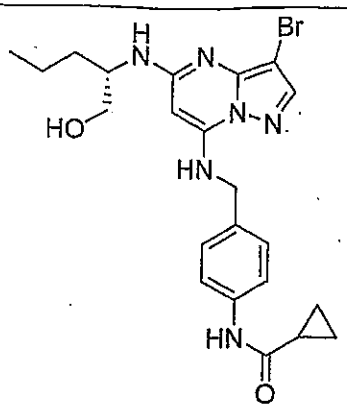
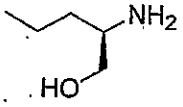
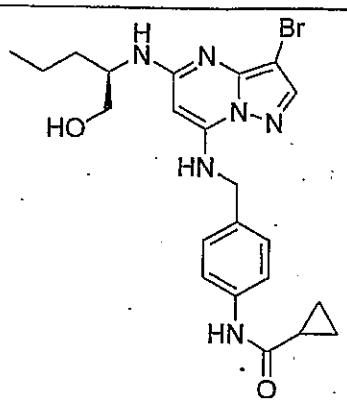
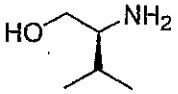
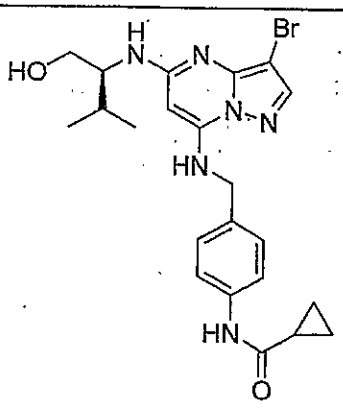
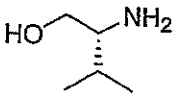
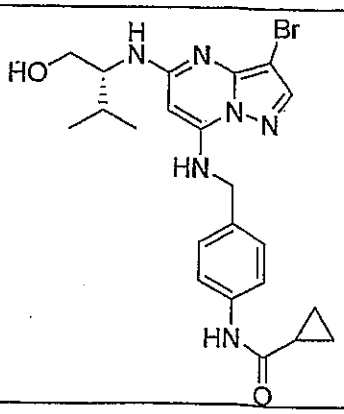
TABLE 1004

Ex.	Column 2	Column 4	CMPD
1042			M+H= 500.3

1043			M+H= 514.1
1044			M+H= 460.3
1045			M+H= 477.1
1046			M+H= 505.1

1047			M+H= 505.1
1048			M+H=531.1
1049			M+H= 477.1

1050			M+H= 505.1
1051			M+H= 505.1
1052			M+H= 531.1
1053			M+H= 514.1

1054			M+H= 488.3
1055			M+H= 488.3
1056			M+H= 488.1
1057			M+H= 488.1

PREPARATIVE EXAMPLE 10-K:

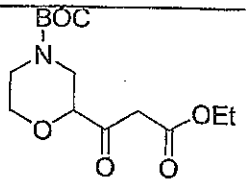
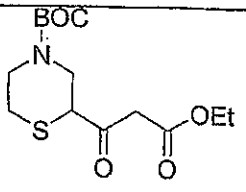
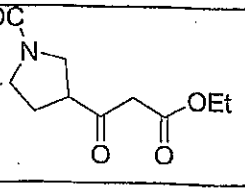


$SOCl_2$ (18.5 mL) was added slowly under N_2 to a stirred mixture of the acid (50.0 g, 218 mmol) and pyridine (44.0 mL) in anhydrous CH_2Cl_2 (60 mL). The mixture was stirred at 25°C for 20 min, then Meldrum's acid (35.0 g, 243 mmol) and DMAP (66.6 g, 546 mmol) were added and the mixture was stirred under N_2 for 1 hr. Then Et_2O (2 L) was added, the mixture was washed with 1 M HCl (3x500 mL), brine (500 mL), and the organic layer was dried over Na_2SO_4, filtered, and the solvent was evaporated. The residue was dissolved in MeOH (580 mL), and the mixture was refluxed for 4 hr. The solvent was evaporated and the residue was purified by column chromatography on silica gel with 10:1 CH_2Cl_2/EtOAc as eluent. Pale yellow oil (26.5 g, 43 %) was obtained.

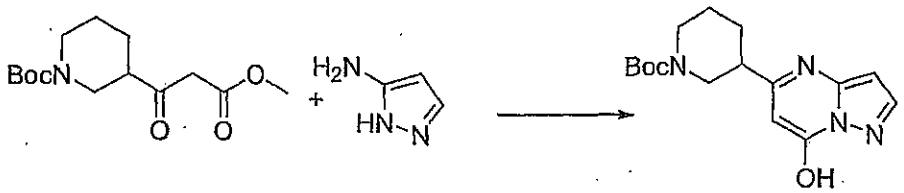
PREPARATIVE EXAMPLES 10-K-40-K:

By essentially same procedure set forth in Preparative Example 10-K, the compounds given in Column 2 of Table 10-K were prepared.

TABLE 10-K

Prep. Ex.	Column 2
20-K	
30-K	
40-K	

PREPARATIVE EXAMPLE 100-K:

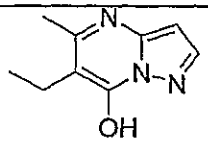
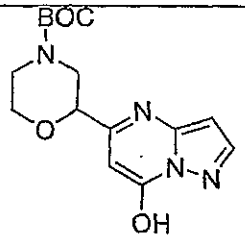
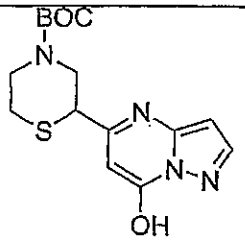


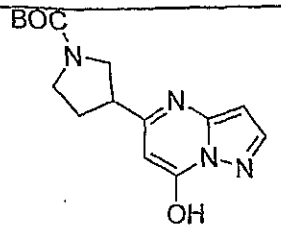
A mixture of the b-ketoester from Preparative Example 10-K (20.0 g, 70.1 mmol) and 3-aminopyrazole (5.40 g, 65.0 mmol) in anhydrous toluene (60 mL) was stirred and refluxed under N_2 for 24 hr. The solvent was evaporated and the residue was purified by column chromatography on silica gel with 20:1 CH_2Cl_2/MeOH as eluent. White solid (15.0 g, 73 %) was obtained. LC-MS: 319 [M+H].

PREPARATIVE EXAMPLE 101-K – 104-K:

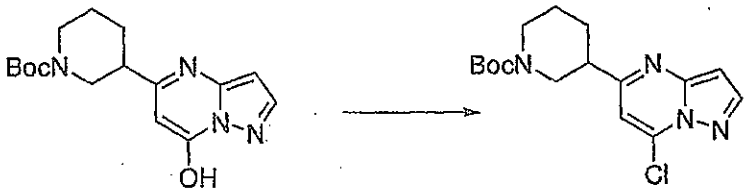
By essentially same procedure set forth in Preparative Example 100-K, combining 3-aminopyrazole with the corresponding β-ketoesters, compounds given in Column 2 of Table 100-K were prepared.

TABLE 100-K

Prep. Ex.	Column 2
101-K	
102-K	
103-K	

104-K	

PREPARATIVE EXAMPLE 200-K:

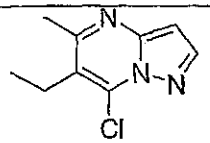
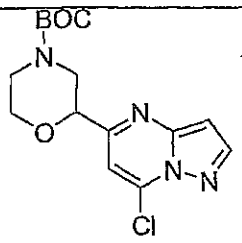


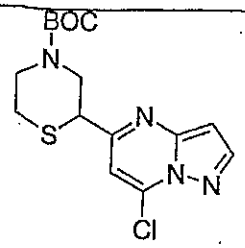
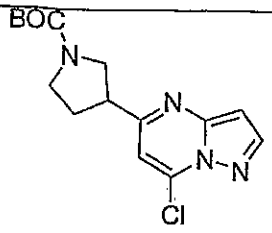
A mixture of the product from Preparative Example 100-K (12.50 g, 39.3 mmol), N,N-dimethylaniline (15.5 mL), and $POCl_3$ (125 mL) was stirred at 25°C for 4 days. Excess of $POCl_3$ was evaporated and the residue was poured into saturated aqueous $NaHCO_3$ (600 mL). The mixture was extracted with CH_2Cl_2 (3x200 mL), the combined extracts were dried over Na_2SO_4, filtered, and the solvent was evaporated. The residue was purified by column chromatography on silica gel with 8:1 CH_2Cl_2/EtOAc as eluent. Pale yellow wax (9.41 g, 71 %) was obtained. LC-MS: 337 [M+].

PREPARATIVE EXAMPLE 201-K – 204-K:

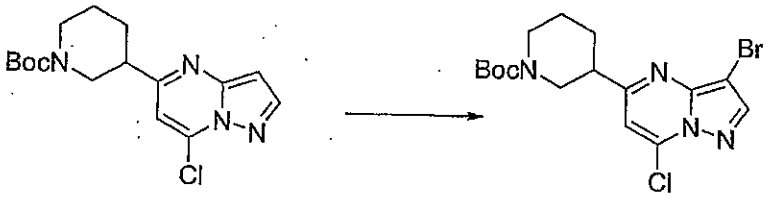
By essentially same procedure set forth in Preparative Example 200-K, compounds given in Column 2 of Table 200-K were prepared.

TABLE 200-K

Prep. Ex.	Column 2
201-K	
202-K	

203-K	
204-K	

PREPARATIVE EXAMPLE 300-K:



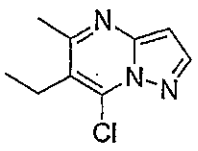
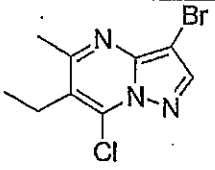
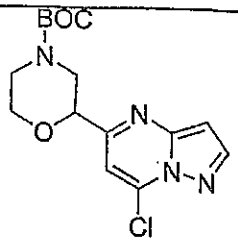
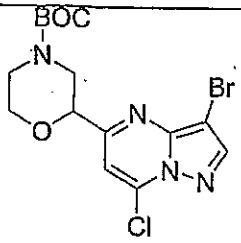
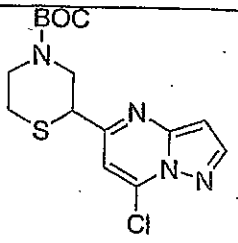
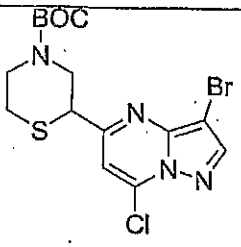
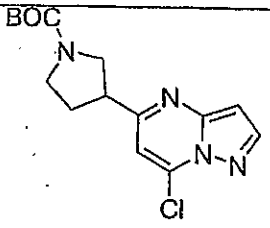
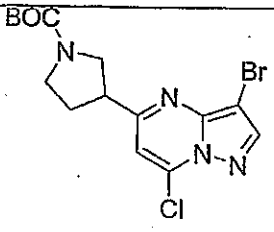
A solution of NBS (4.03 g, 22.7 mmol) in anhydrous CH_3CN (40 mL) was added under N_2 to a stirred solution of the product from Preparative Example 200-K (7.63 g, 22.7 mmol) in anhydrous CH_3CN (60 mL) and CH_2Cl_2 (20 mL). The mixture was stirred for 2 hr, the solvents were evaporated, and the residue was purified by column chromatography on silica gel with 20:1 CH_2Cl_2/EtOAc as eluent. Pale yellow solid foam (9.20 g, 97 %) was obtained. LC-MS: 417 [M+H].

PREPARATIVE EXAMPLE 301-K – 304-K:

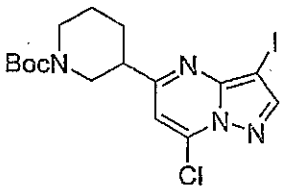
By essentially same procedure set forth in Preparative Example 300-K only substituting the compounds shown in Column 2 of Table 300-K, the compounds given in Column 3 of Table 300-K were prepared.

TABLE 300-K

Prep. Ex:	Column 2	Column 3

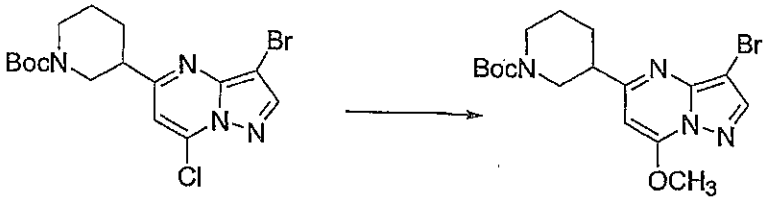
301-K		
302-K		
303-K		
304-K		

PREPARATIVE EXAMPLE 303-K:



By essentially the same procedure set forth in Preparative Example 300-K only substituting N-iodosuccinimide the above compound was prepared.

PREPARATIVE EXAMPLE 400-K:



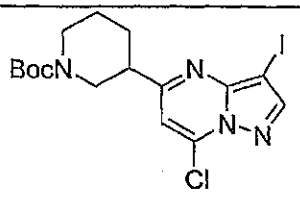
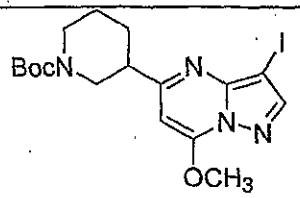
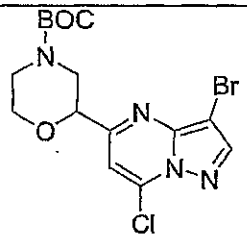
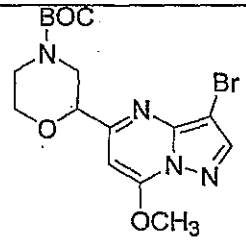
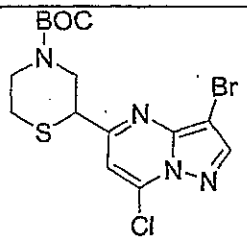
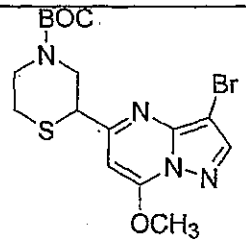
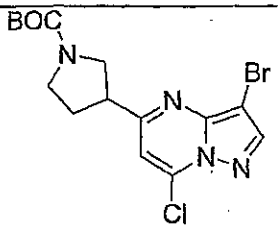
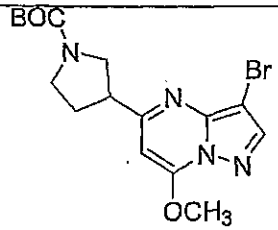
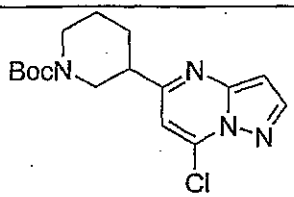
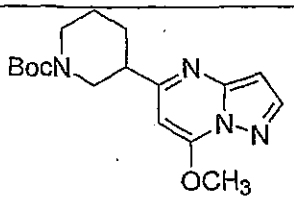
A mixture of the product from Preparative Example 300-K (8.00 g, 19.3 mmol) and MeONa (2.16 g, 40.0 mmol) in anhydrous MeOH (100mL) was stirred for 20 hr. CH_2Cl_2 (200mL) was then added, the mixture was filtered through

Celite, the solvent was evaporated, and the residue was purified by column chromatography on silica gel with 2:1 CH_2Cl_2/EtOAc as eluent. White solid (7.75 g, 98 %) was obtained.

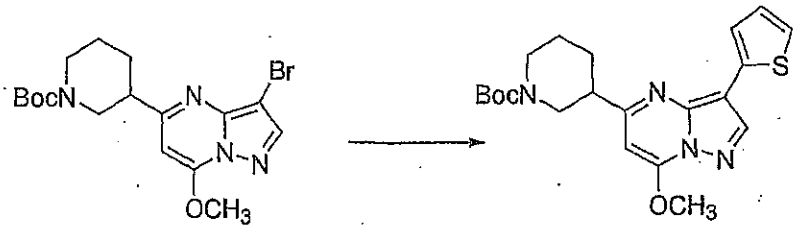
PREPARATIVE EXAMPLES 401-K – 405-K:

By essentially the same procedure set forth in Preparative Example 400-K, only substituting the compound shown in Column 2 of Table 400-K, the compounds shown in Column 3 of Table 400-K were prepared.

TABLE 400-K

Prep. Ex.	Column 2	Column 3
401-K		
402-K		
403-K		
404-K		
405-K		

PREPARATIVE EXAMPLE 500-K:

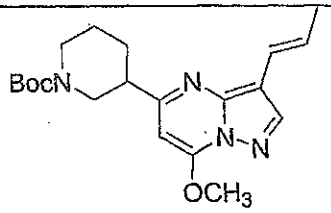
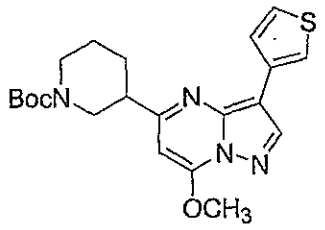


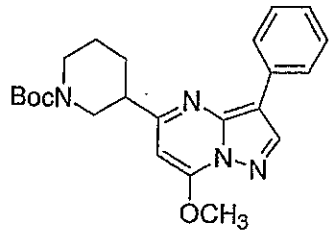
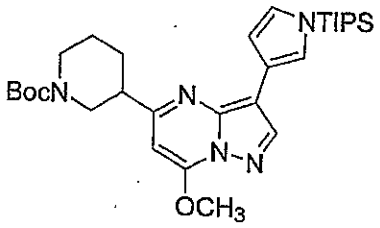
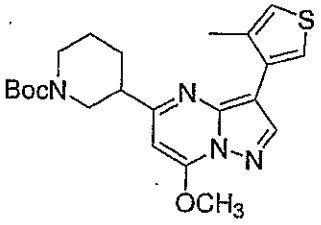
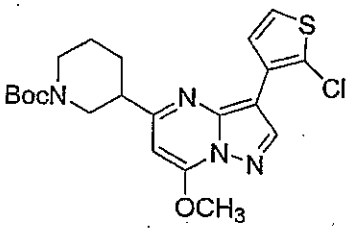
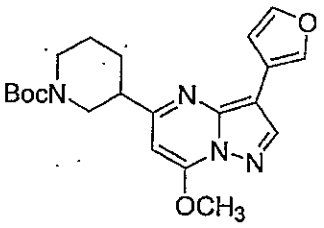
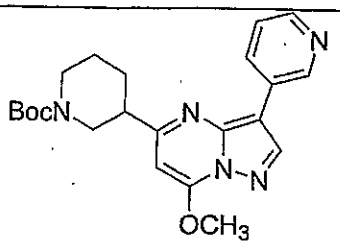
A mixture of the product from Preparative Example 400-K (600 mg, 1.46 mmol), 2-thienylboronic acid (230 mg, 1.80 mmol), $Pd[PPh_3]_4$ (160 mg, 0.14 mmol), and Na_2CO_3 (466 mg, 4.40 mmol) in 1,2-dimethoxyethane (20 mL) and H_2O (4 mL) was stirred and refluxed under N_2 for 20 hr. The solvents were evaporated and the residue was purified by column chromatography on silica gel with 2:1 CH_2Cl_2/EtOAc as eluent. Pale yellow solid (355 mg, 59 %) was obtained. LC-MS: 415 [M+].

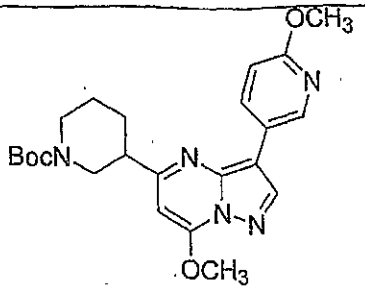
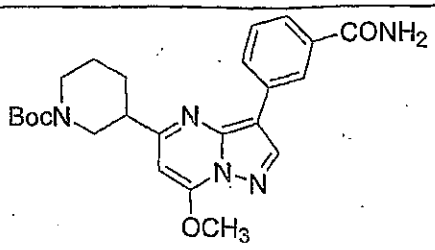
PREPARATIVE EXAMPLES 501-511-K:

By essentially same procedure set forth in Preparative Example 500-K only substituting the appropriate boronic acid, the compounds given in Column 2 of Table 500-K were prepared.

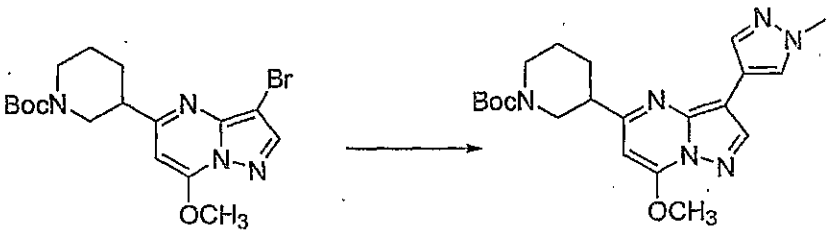
TABLE 500-K

Prep. Ex.	Column 2	Data
501-K		LCMS: MH^+=373
502-K		LCMS: MH^+=415

503-K		LCMS: MH^+=409
504-K		LCMS: MH^+=555
505-K		LCMS: MH^+=429
506-K		
507-K		LCMS: MH^+=399
508-K		LCMS: MH^+=410

510-K		LCMS: MH^+=425
511-K		LCMS: MH^+=452

PREPARATIVE EXAMPLE 520-K:



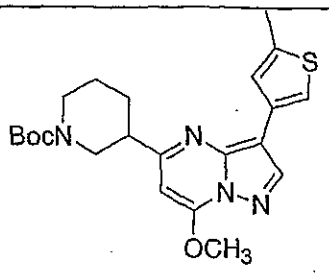
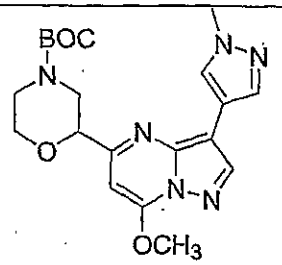
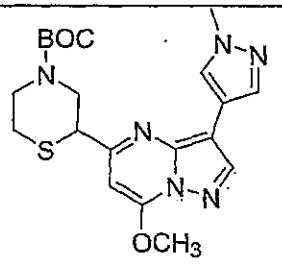
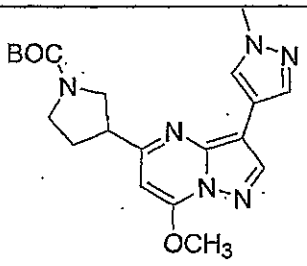
A mixture of the product from Preparative Example 400-K (1.23 g, 3.00 mmol), 1-methyl-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2yl)1-H-pyrazole (850 mg, 4.00 mmol), $Pd[PPh_3]_4$ (348 mg, 0.30 mmol), and Na_2CO_3 (1.27 g, 12.0 mmol) in 1,2-dimethoxyethane (36 mL) and H_2O (7.5 mL) was stirred and refluxed under N_2 for 20 hr. The solvents were evaporated and the residue was purified by column chromatography on silica gel with 20:1 EtOAc/MeOH as eluent. Pale orange solid (740 mg, 60 %) was obtained. LC-MS: 413 [M+H].

PREPARATIVE EXAMPLE 530-K – 533-K:

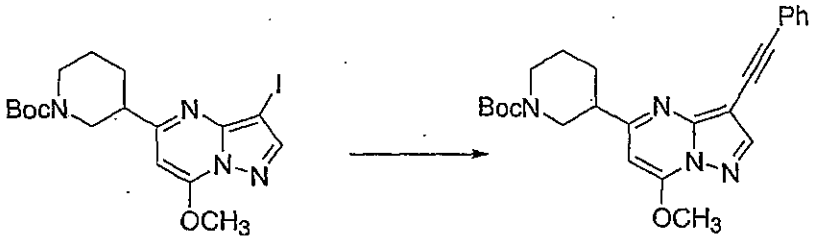
By essentially same procedure set forth in Preparative Example 520-K, the compounds shown in Column 2 of Table 530-K were prepared.

TABLE 530-K

Prep. Ex.	Column 2	Data

530-K		LC-MS: 429 [M+H].
531-K		
532-K		
533-K		

PREPARATIVE EXAMPLE 550-K:



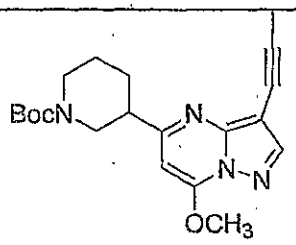
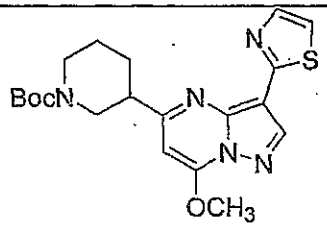
A mixture of the compound from Preparative Example 512-K (230 mg, 0.50 mmol), tributylphenyltin (235 mg, 0.60 mmol), and Pd[PPh3]4 (58 mg, 0.05 mmol) in anhydrous dioxane (5 mL) was stirred under N2 at 90oC for 3 hr. The solvent was evaporated and the residue was purified by column chromatography

on silica gel with 20:1 CH_2Cl_2/EtOAc as eluent. Pale yellow wax (46 mg, 21 %) was obtained. LC-MS: 433 [M+].

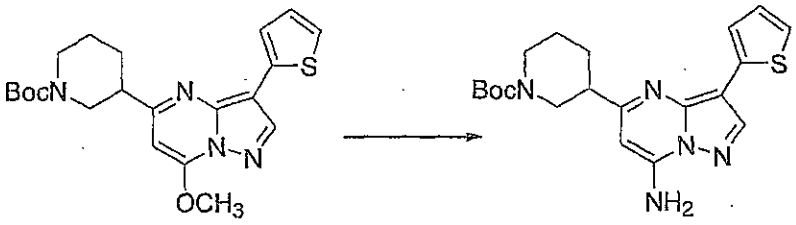
PREPARATIVE EXAMPLES 551-K -552-K:

By essentially same procedure set forth in Preparative Example 550-K, only substituting the appropriate tributyltin reagent, the compounds given in Column 2 of Table 550-K were prepared.

TABLE 550-K

Prep. Ex.	Column 2	Data
551-K		LCMS: MH^+=371
552-K		LCMS: MH^+=416

PREPARATIVE EXAMPLE 600-K:

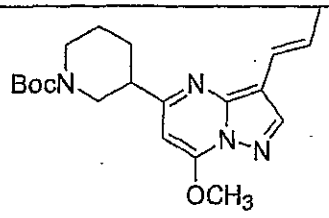
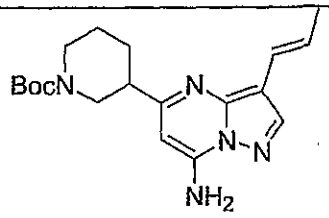
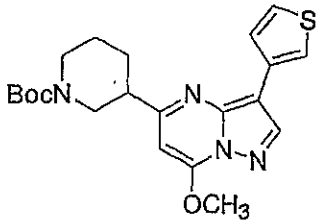
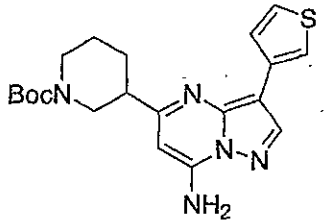
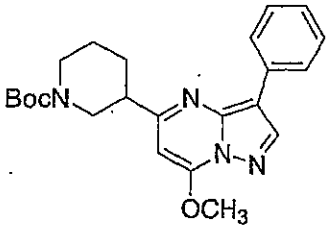
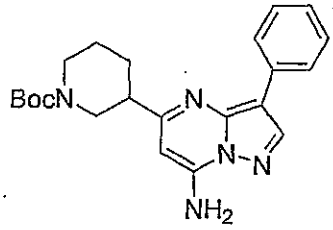
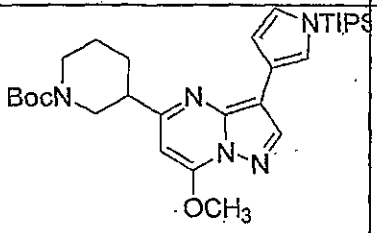
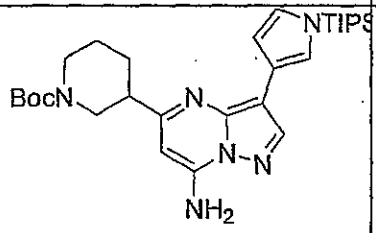
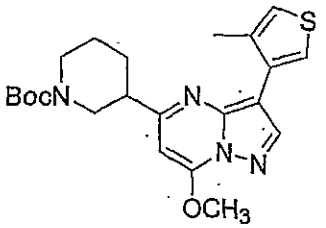
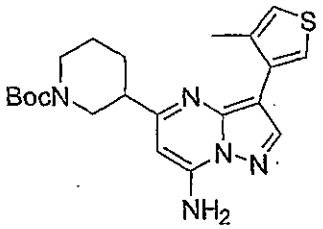


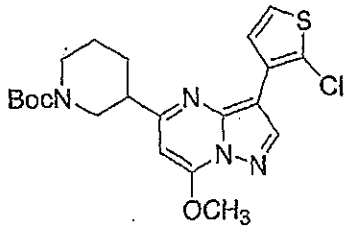
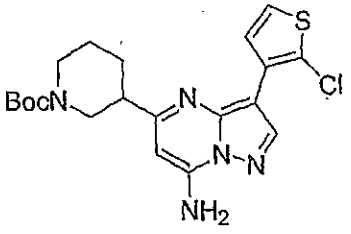
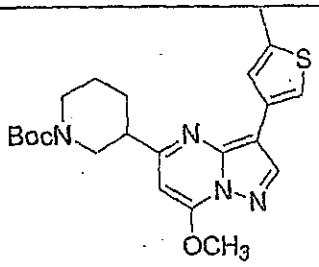
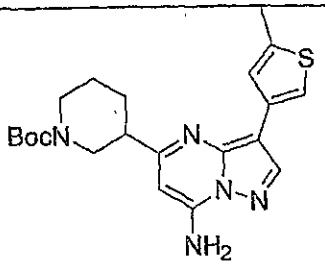
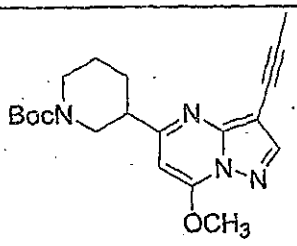
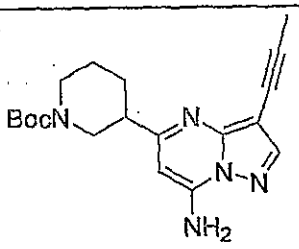
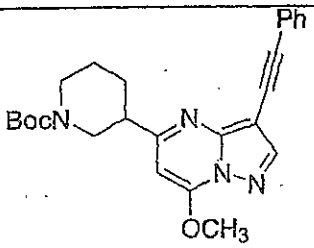
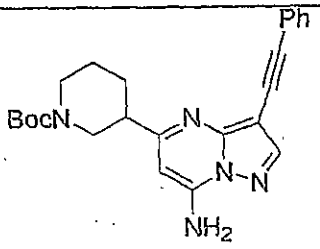
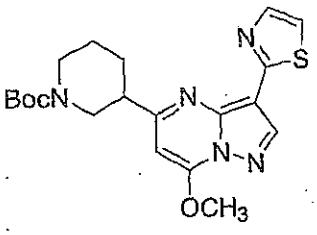
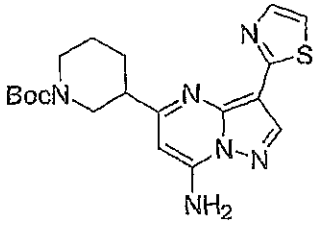
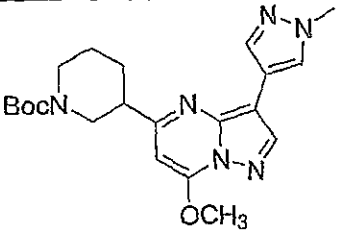
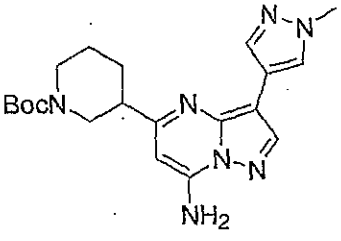
A mixture of the product from Preparative Example 500-K (350 mg, 0.85 mmol), 2.0 M NH_3 in 2-propanol (7.5 mL), and conc. aqueous NH_4OH (0.75 mL) was stirred in a closed pressure vessel at 75°C for 3 days. The solvents were evaporated and the residue was purified by column chromatography on silica gel with 50:1 CH_2Cl_2/MeOH as eluent. Pale yellow solid (143 mg, 42 %) was obtained. LC-MS: 400 [M+].

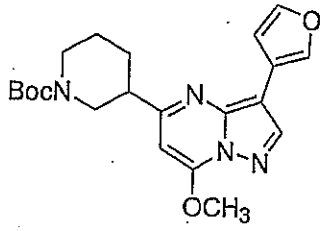
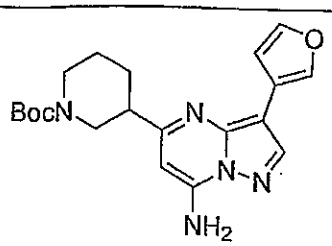
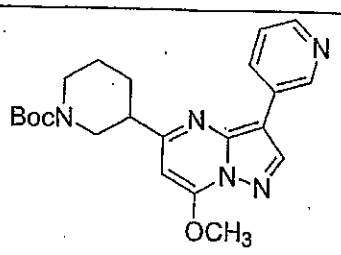
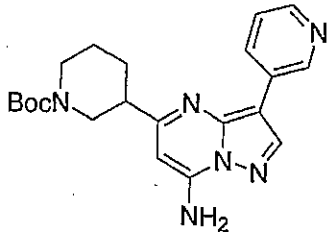
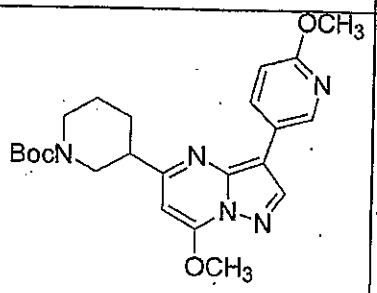
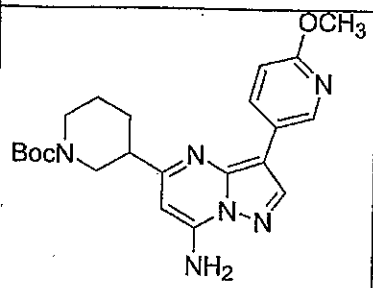
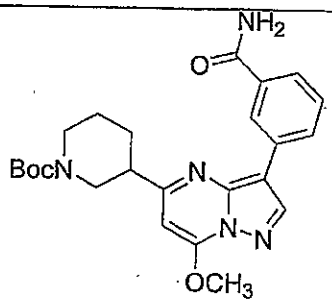
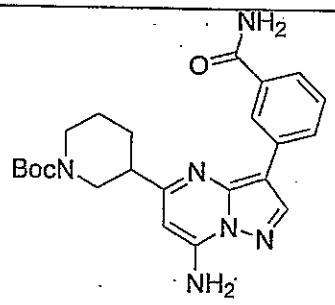
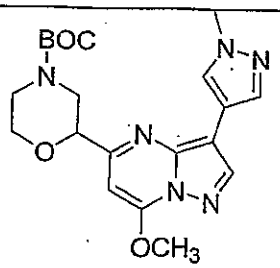
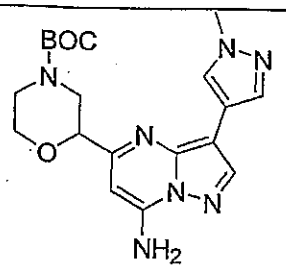
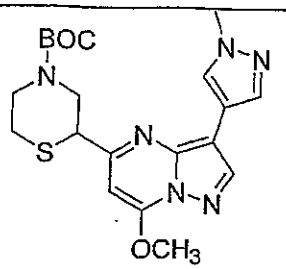
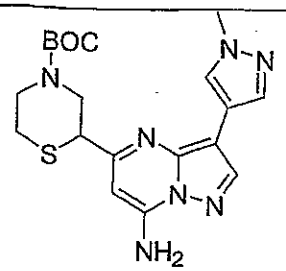
PREPARATIVE EXAMPLE 601-K -615-K:

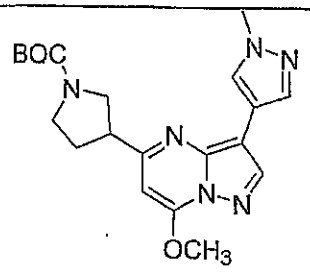
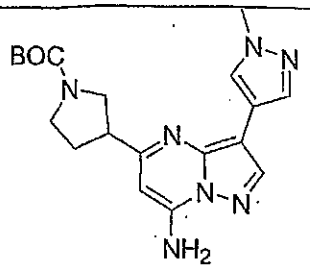
By essentially same procedure set forth in Preparative Example 600-K only substituting the compounds in Column 2 of Table 600-K, the compounds given in Column 3 of Table 600-K were prepared.

TABLE 600-K

Prep. Ex.	Column 2	Column 3	Data
601-K			LCMS: MH^+=359
602-K			LCMS: MH^+=400
603-K			
604-K			LCMS: MH^+=540
605-K			LCMS: MH^+=414

606-K			LCMS: M^+=434
607-K			LCMS: MH^+=414
608-K			LCMS: MH^+=356
609-K			LCMS: MH^+=418
610-K			LCMS: MH^+=401
611-K			LCMS: MH^+=398

612-K			LCMS: MH^+=384
613-K			LCMS: MH^+=395
614-K			LCMS: MH^+=425
615-K			LCMS: MH^+=437
616-K			
617-K			

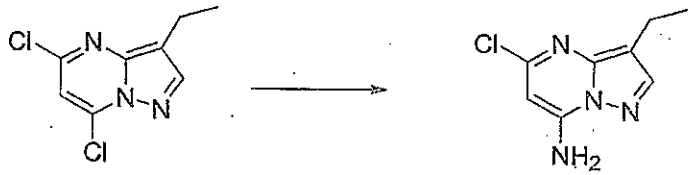
618-K		

PREPARATIVE EXAMPLE 650-K:



A mixture of the product from Preparative Example 201-K (100 mg, 0.37 mmol), 2.0 M NH_3 in 2-propanol (2.0 mL), and conc. aqueous NH_4OH (0.2 mL) was stirred in a closed pressure vessel at 50°C for 2 days. The solvents were evaporated and the residue was purified by column chromatography on silica gel with 30:1 CH_2Cl_2/MeOH as eluent. White solid (38 mg, 41 %) was obtained. LC-MS: 257 [M+2H].

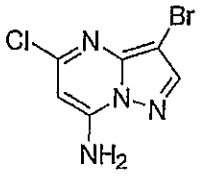
PREPARATIVE EXAMPLE 660-K:



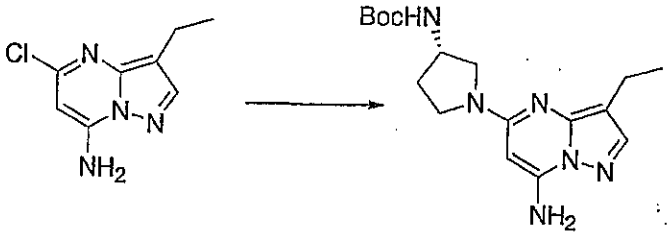
A mixture of the starting material (2.16 g, 10.0 mmol), 2.0 M NH_3 in 2-propanol (50.0 mL), and conc. aqueous NH_4OH (10.0 mL) was stirred in a closed pressure vessel at 50°C for 2 days. The solvents were evaporated, the solid was suspended in H_2O (100 mL), filtered, washed on filter with H_2O (100 mL), and dried in a vacuum at 100°C. Slightly beige solid (1.80 g, 91 %) was obtained.

PREPARATIVE EXAMPLE 661-K:

By essentially same procedure set forth in Preparative Example 660-K, compound given below was prepared.



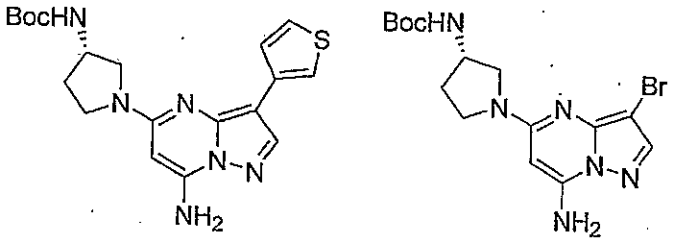
PREPARATIVE EXAMPLE 662-K:



A mixture of the product from Preparative Example 660-K (600 mg, 3.05 mmol), the amine (837 mg,4.50 mmol), and $NaHCO_3$ (756 mg, 9.00 mmol) in anhydrous N-methylpyrrolidone (4 mL) was stirred under N_2 at 140°C for 18 hr. The mixture was cooled to 25oC, CH_2Cl_2 (15 mL) was added and the mixture was filtered. The solvents from the filtrate were distilled off and the residue was purified by column chromatography on silica gel with 2:1 CH_2Cl_2/EtOAc as eluent. White solid (1.01g, 95 %) was obtained.

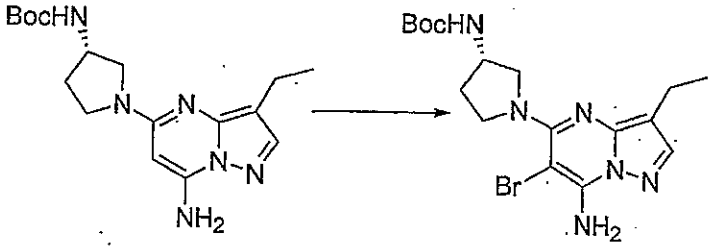
PREPARATIVE EXAMPLE 663-K:

By essentially same procedure set forth in Preparative Example 662-K, compounds given below were prepared.



LC-MS: 401 [M+H].

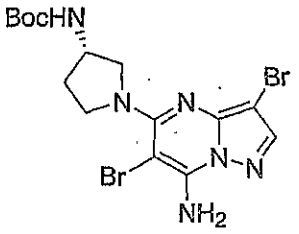
PREPARATIVE EXAMPLE 664-K:



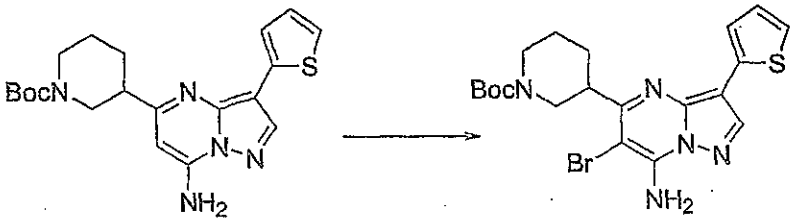
A solution of NBS (142 mg, 0.80 mmol) in anhydrous CH_3CN (5 mL) was added under N_2 to a stirred solution of the product from Preparative Example 662-K (400 mg, 0.90 mmol) in anhydrous CH_3CN (5 mL) and CH_2Cl_2 (5 mL). The mixture was stirred for 18 hr, the solvents were evaporated, and the residue was purified by column chromatography on silica gel with 3:1 CH_2Cl_2/EtOAc as eluent. White solid (330 mg, 67 %) was obtained.

PREPARATIVE EXAMPLE 665-K:

By essentially same procedure set forth in Preparative Example 664-K, compound given below was prepared.



PREPARATIVE EXAMPLE 700-K:



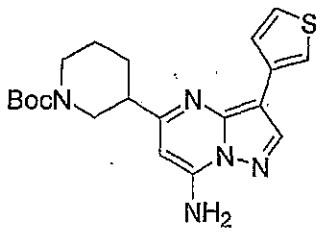
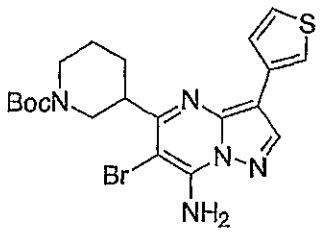
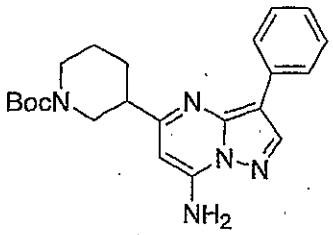
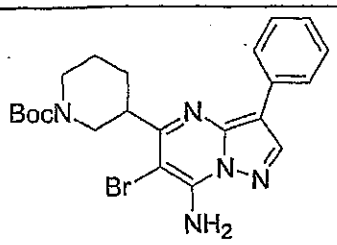
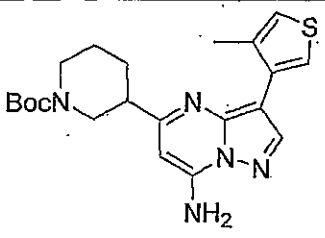
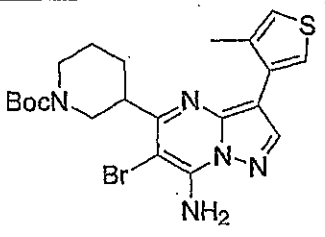
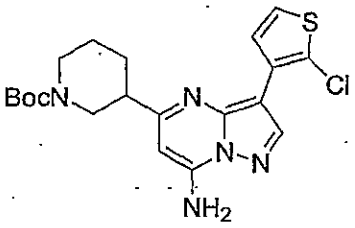
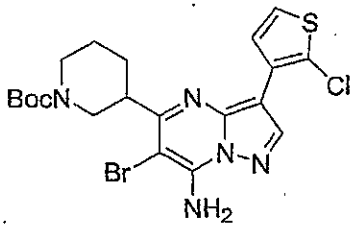
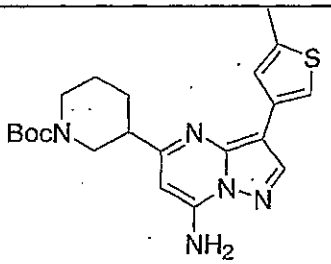
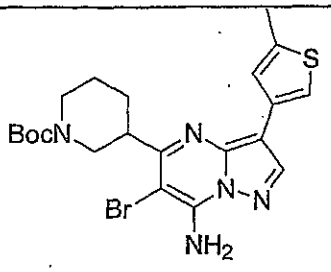
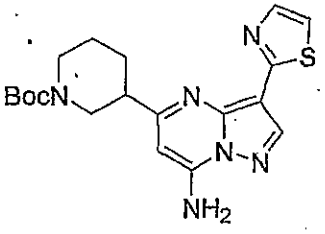
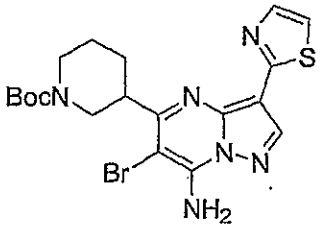
A solution of Br_2 (16 mg) in CH_2Cl_2 (0.5 mL) was added under N_2 to a stirred solution of the product from Preparative Example 600-K (40 mg, 0.10 mmol) in t-$BuNH_2$ (2.0 mL). The mixture was stirred for 1 hr, the solvents were evaporated, and the residue was purified by column chromatography on silica gel with 10:1 CH_2Cl_2/EtOAc as eluent. Off-white solid (18 mg, 38 %) was obtained. LC-MS: 480 [M+H].

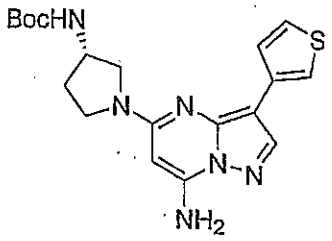
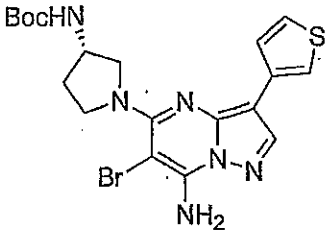
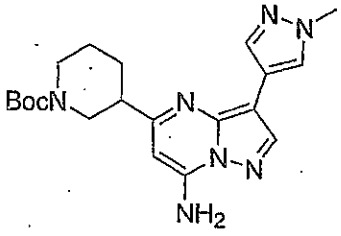
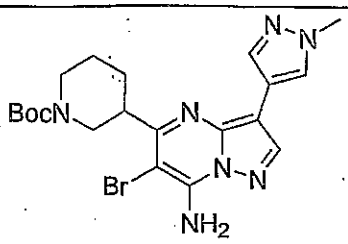
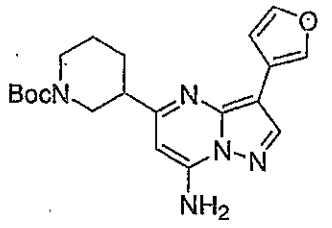
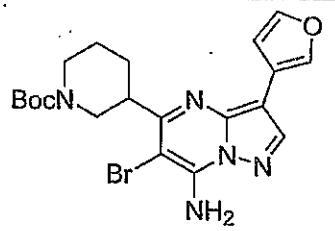
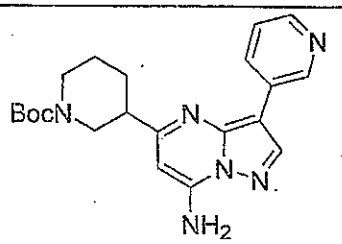
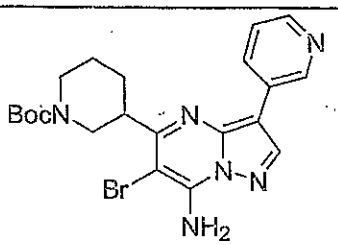
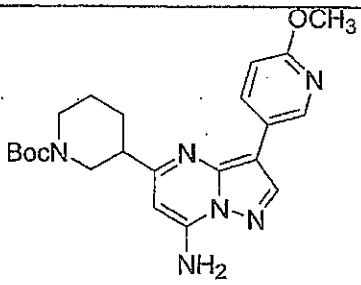
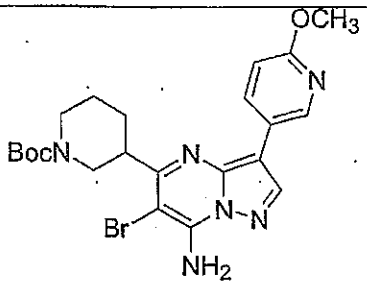
PREPARATIVE EXAMPLE 701-712-K:

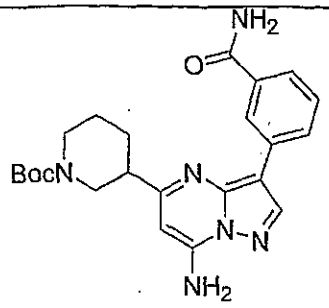
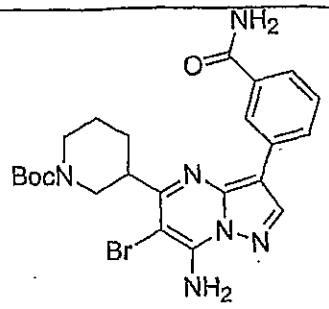
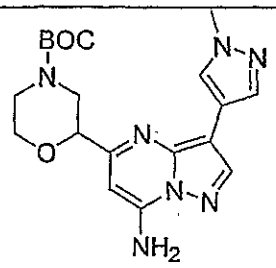
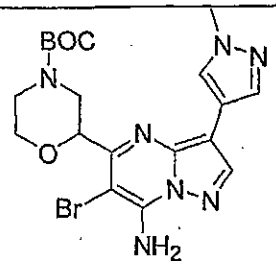
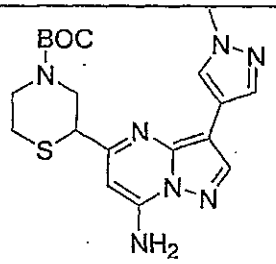
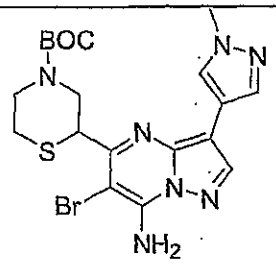
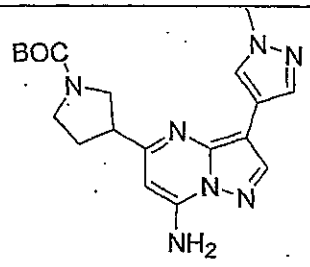
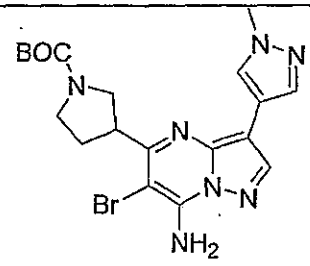
By essentially the same procedure set forth in Preparative Example 700-K only substituting the compounds in Column 2 of Table 700-K, the compounds in Column 3 of Table 700-K were prepared.

TABLE 700-K

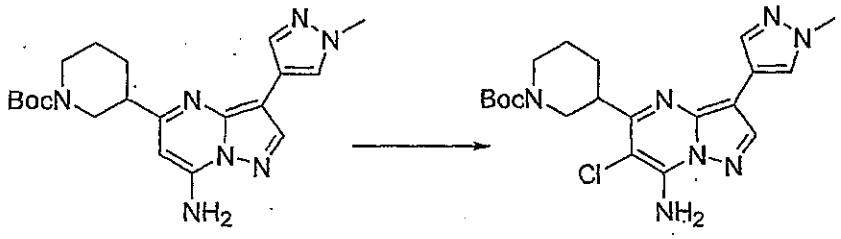
Prep. Ex.	Column 2	Column 3	Data

701-K			LCMS: MH^+=
702-K			
703-K			LCMS: MH^+=
704-K			LCMS: M^+=
705-K			LCMS: MH^+=
706-K			LCMS: MH^+=

707-K			LCMS: M^+=479
708-K			LCMS: M^+=476
709-K			LCMS: M^+=462
710-K			LCMS: M^+=473
711-K			LCMS: M^+ = 503

712-K			LCMS: M^+H = 517
713-K			
714-K			
715-K			

PREPARATIVE EXAMPLE 750-K:



A solution of N-chlorosuccinimide (23 mg, 0.17 mmol) in anhydrous CH_3CN (2 mL) was added under N_2 to a stirred solution of the product from Preparative Example 611-K (73 mg, 0.17 mmol) in anhydrous CH_3CN (2 mL). The mixture was stirred for 24 hr, the solvents were evaporated, and the residue

was purified by column chromatography on silica gel with 1:1 CH_2Cl_2/EtOAc as eluent. White solid (54 mg, 68.%) was obtained. LC-MS: 432 [M+].

PREPARATIVE EXAMPLE 800-K:



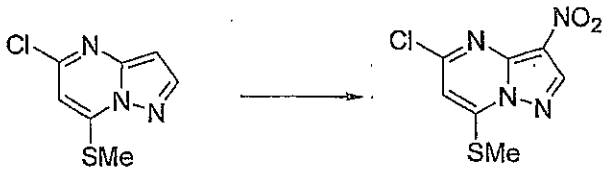
To a solution of dichloride (3.0 g, 16.0 mmol) prepared as described earlier in THF (25 mL) was added NaSMe (1.1 g, 16.0 mmol) in one portion. The resulting mixture was stirred for 12 h at rt and was concentrated under reduced pressure. The crude product was partitioned between EtOAc (150 mL) and H_2O (30 mL) and the layers were separated. The organic layer was washed sequentially with H_2O (2 x 30 mL) and brine (1 x 30 mL). The organic layer was dried (Na_2SO_4), filtered, and concentrated under reduced pressure to afford 3.0 g (94 % yield) of a tan solid. LC-MS: 200.1 [M+H], purity 99%.

PREPARATIVE EXAMPLE 900-K:



Prepared the procedure outlined in Preparative Example 800-K except starting with the dichloride shown above (7.4 g, 27.7 mmol) prepared as described earlier and NaSMe (2.1 g, 30.5 mmol) afforded 7.4 g (96% yield) of the title compound as a light orange solid. LC-MS: 278.1 [M+H], purity 95%.

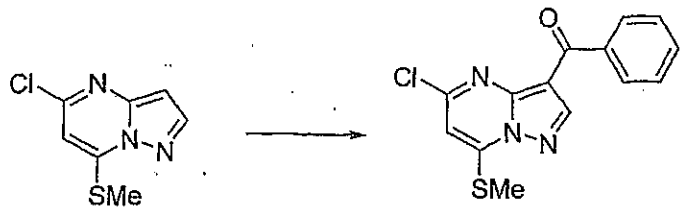
PREPARATIVE EXAMPLE 1000-K:



To a solution of thiomethyl derivative from Preparative Example 800-K (1.5 g, 7.5 mmol) in H_2SO_4 (9 mL) at 0 °C was added 60% HNO_3 (4.5 mL) dropwise. The resulting solution was stirred for 2h at 0 °C and 2h at rt. The reaction mixture was poured into beaker containing ice (~15 g) and the heterogeneous mixture was stirred for 45 min. The resulting ppt was collected

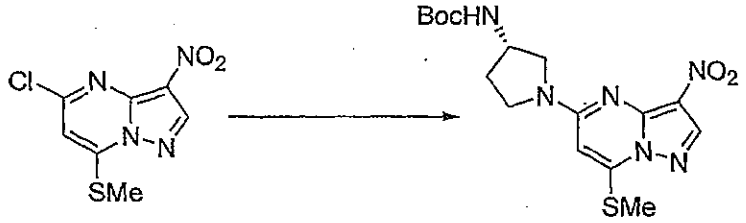
by filtration and was washed sequentially with H_2O (2 x 3 mL) and Et_2O (2 x 3 mL). The ppt was placed under high vacuum to remove trace volatiles to afford 1.1 g (60%) of an orange solid. LC-MS: 245.0 [M+H], purity 90%.

PREPARATIVE EXAMPLE 1100-K:



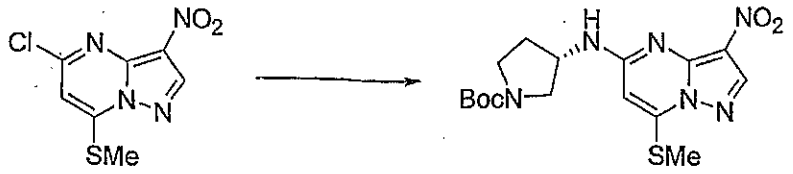
To a solution of thiomethyl derivative (0.20 g, 1.0 mmol) from Preparative Example 800-K in CH_2Cl_2 (5 mL) at 0 °C was added $AlCl_3$ (0.70 g, 5.3 mmol) followed by benzoyl chloride (0.40 mL, 3.5 mmol). The mixture was heated at reflux for 12 h at rt whereupon the mixture was carefully quenched at 0 °C with sat. aq. $NaHCO_3$ (3 mL). CH_2Cl_2 (5 mL) was added, the layers were separated, and the aqueous layer was extracted with CH_2Cl_2 (2 x 5 mL). The organic layers were combined, washed with brine (1 x 5 mL), dried (Na_2SO_4), filtered, and concentrated under reduced pressure to afford 0.27 g (88% crude yield) of a brown solid. LC-MS: 304.0 [M+H], purity 85%.

PREPARATIVE EXAMPLE 1200-K:



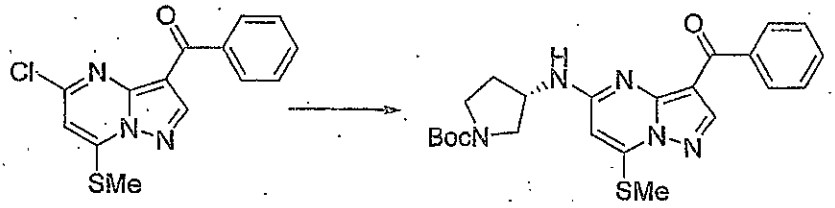
To a solution of thiomethyl derivative (0.4 g, 1.63 mmol) from Preparative Example 1000-K in dioxane/DIPEA (7 ml/2 mL) at room temperature was added (*S*)-(-)-3-(Boc-amino)pyrrolidine (0.36 g, 1.96 mmol) in a single portion. The mixture was stirred at reflux for 12 h and was cooled to rt. The mixture was filtered and washed with CH_2Cl_2 (2 x 2 mL) and Et_2O (2 x 2 mL). The resultant ppt was dried under vacuum to afford 0.57 g (90% yield) of a yellow solid. LC-MS: 395.1 [M+H], purity 90%.

PREPARATIVE EXAMPLE 1300-K:



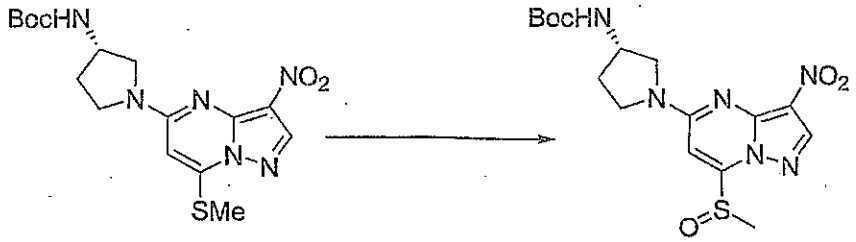
The above compound was prepared according to the procedure outlined in Preparative Example 1200-K utilizing thiomethyl derivative from Preparative Example 1000 (66 mg, 0.27 mmol) and (*S*)-(-)-1-(Boc-amino)pyrrolidine (65 mg, 0.35 mmol) to afford 83 mg (77% yield) of a yellow solid. LC-MS: 395.1 [M+H], purity 99%.

PREPARATIVE EXAMPLE 1400-K:



The title compound was prepared according to the procedure outlined in Preparative Example 1200-K utilizing thiomethyl derivative (0.30 g, 1 mmol) from Preparative Example 1100-K and (*S*)-(-)-1-(Boc-amino)pyrrolidine (0.24 g, 1.3 mmol) to afford 160 mg (35% yield) of a yellow solid. LC-MS: 454.1 [M+H], purity 60%.

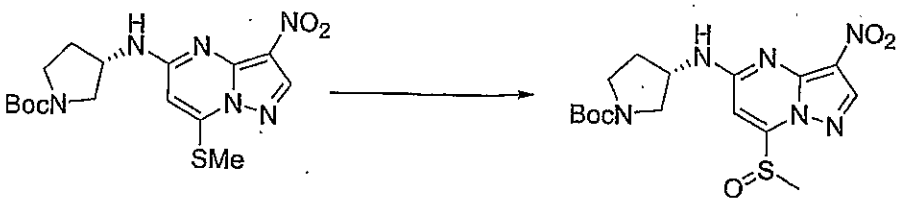
PREPARATIVE EXAMPLE 1500-K:



To a solution of thiomethyl adduct (0.17 g, 0.42 mmol) from Preparative Example 1200-K in CH_2Cl_2 (2 mL) at 0 °C was added MCPBA (0.11 g, 0.63 mmol) in one portion. The mixture was stirred for 2 h at 0 °C and 1 h at rt whereupon an additional portion of MCPBA (54 mg, 0.32 mmol) was added. The mixture was stirred for 2 h at rt where upon CH_2Cl_2 (3 mL) and sat. aq. $NaHCO_3$ (3 mL) were added. The layers were separated and the aqueous layer was extracted with CH_2Cl_2 (2 x 5 mL). The organic layers were combined and washed with brine (1 x 3mL). The organic layer was dried (Na_2SO_4), filtered, and

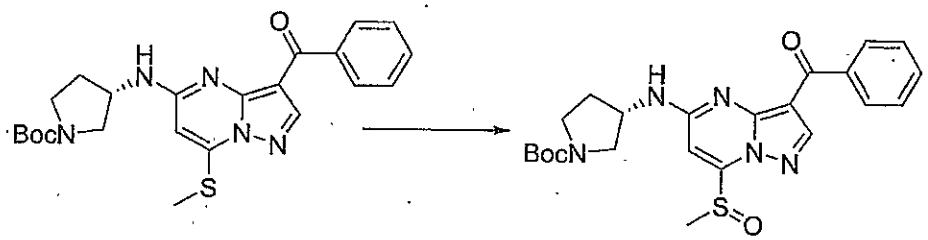
concentrated under reduced pressure to afford 0.17 (98% crude yield) of a yellow solid. This material was used directly without further purification. LC-MS: 411.1 [M+H], purity 85%.

PREPARATIVE EXAMPLE 1600-K:



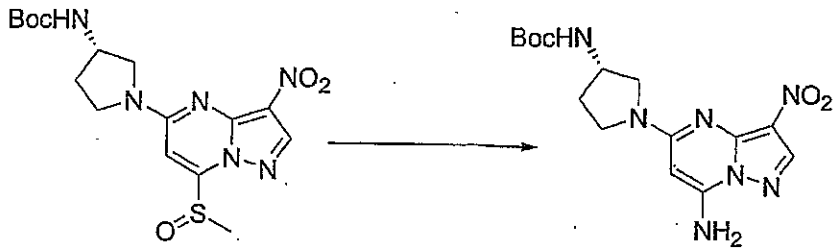
The title compound was prepared according to the procedure outlined in Preparative Example 1500-K utilizing thiomethyl derivative (82 mg, 0.21 mmol) from Preparative Example 1300-K 0 to afford 84 mg (98% yield) of a yellow solid. This material was used directly without further purification. LC-MS: 411.1 [M+H], purity 89%.

PREPARATIVE EXAMPLE 1700:



The title compound was prepared according to the procedure outlined in Preparative Example 1500-K utilizing thiomethyl derivative (0.15 g, 0.33 mmol) from Preparative Example 1400-K to afford 152 mg (98% yield) of a yellow solid. This material was used directly without further purification. LC-MS: 470.1 [M+H], purity 40%.

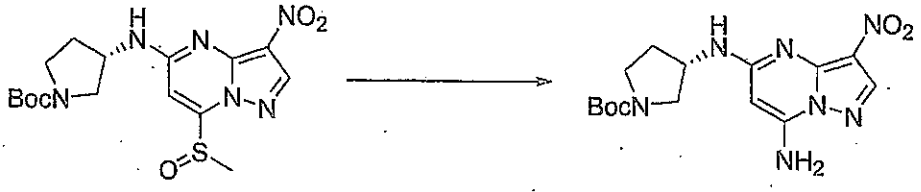
PREPARATIVE EXAMPLE 1800-K:



To a pressure tube charged with a stir bar was added sulfoxide (0.15 g, 0.36 mmol) from Preparative Example 1500-K followed by 2M NH_3 in IPA (4 mL) and conc. NH_4OH (1 mL). The tube was capped and was heated to 85 °C and

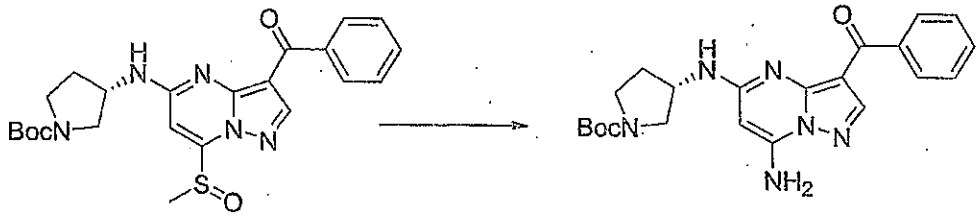
stirred for 14h. The mixture was cooled to room temperature, concentrated under reduced pressure, and placed under high vacuum to remove trace volatiles. The crude product was purified by preparative thin-layer chromatography (4 x 1000 μM) plates using a 20:1 mixture of CH_2Cl_2/MeOH as eluent to afford 105 mg (80% yield) of a yellow/orange solid. LC-MS: 364.2 [M+H], purity 99%.

PREPARATIVE EXAMPLE 1900-K:



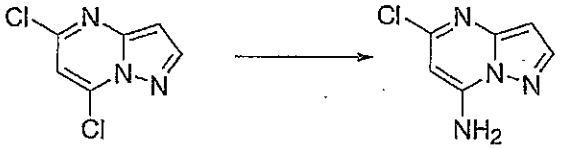
The title compound was prepared according to the procedure outlined in Preparative Example 1800-K utilizing thiomethyl derivative (60 mg, 0.15 mmol) from Preparative Example 1600-K to afford 21 mg (39% yield) of an off-white solid. LC-MS: 364.1 [M+H], purity 90%.

PREPARATIVE EXAMPLE 2000-K:



The title compound was prepared according to the procedure outlined in Preparative Example 1800-K utilizing thiomethyl derivative (140 mg, 0.30 mmol) from Preparative Example 1700-K to afford 30 mg (24% yield) of yellow crystalline solid. LC-MS: 423.1 [M+H], purity 93%.

PREPARATIVE EXAMPLE 2100-K:



To a pressure tube charged with dichloride (5 g, 26.7 mmol) and a stir bar was added conc. NH_4OH (110 mL). The tube was sealed and heated to 70 °C. The mixture was stirred for 5 h, cooled to rt, and concentrated to dryness under reduced pressure. The resultant solid was suspended in H_2O (70 mL), filtered,

and the solid was washed with Et_2O (100 mL). The crude material was dried under reduced pressure to afford 4.2 g (93% yield) of a yellow solid that was used without purification. LC-MS: 169.0 [M+H], purity 99%.

PREPARATIVE EXAMPLE 2200-K:



The title compound was prepared according to the procedure outlined in Preparative Example 2100-K except utilizing the 3-bromo derivative (2.0 g, 7.5 mmol) to afford 1.7 g (92% yield) of a white solid. mp >215 °C; LC-MS: 249.0 [M+H], purity 99%.

PREPARATIVE EXAMPLE 2300-K:



To a solution of 7-amino adduct (4.2 g, 24.8 mmol) from Preparative Example 2100-K in CH_2Cl_2 (50 mL) at rt was added Boc_2O (5.9 g, 27.3 mmol) and DMAP (3.3 g, 27.3 mmol). The resulting solution was stirred for 12 h at rt and was diluted with CH_2Cl_2 (50 mL) and sat. aq. $NaHCO_3$ (25 mL). The layers were separated and the aqueous layer was extracted with CH_2Cl_2 (2 x 30 mL). The combined organic layers were washed with brine (1 x 30 mL), dried (Na_2SO_4), filtered and concentrated under reduced pressure. The crude product was purified by 40M Flash Elute using a gradient (1L of CH_2Cl_2 to 2.5L of 1% MeOH (7M NH_3):CH_2Cl_2 to afford 6.3 g (95% yield) of a white solid. LC-MS: 269.0 [M+H], purity 99%.

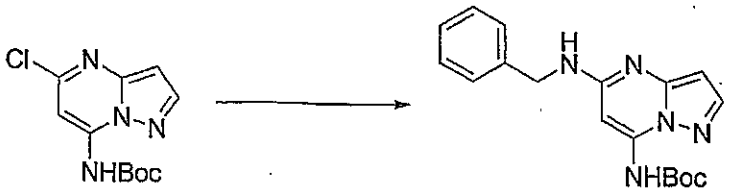
PREPARATIVE EXAMPLE 2400-K:



The title compound was prepared according to the procedure outlined in Preparative Example 2300-K except utilizing the 3-bromo derivative (2.0 g, 8.1

mmol) from Preparative Example 2200-K to afford 2.5 g (89% yield) of a white solid; LC-MS: 349.1 [M+H], purity 95%.

PREPARATIVE EXAMPLE 2500-K:

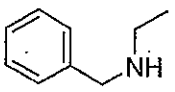
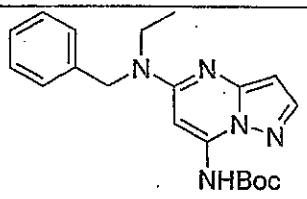
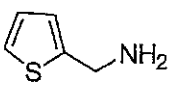
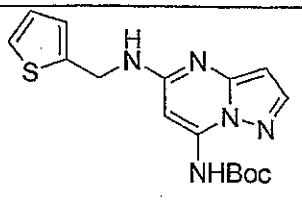


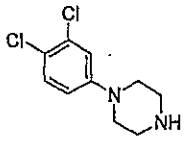
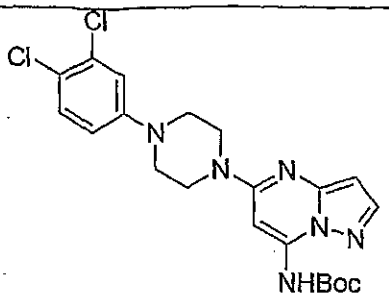
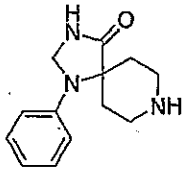
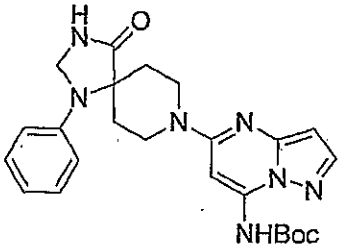
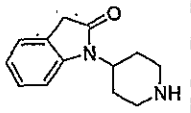
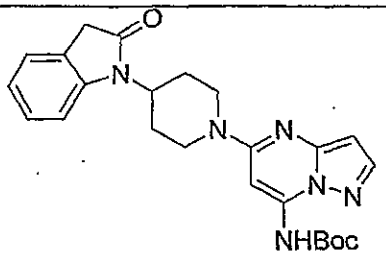
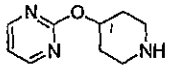
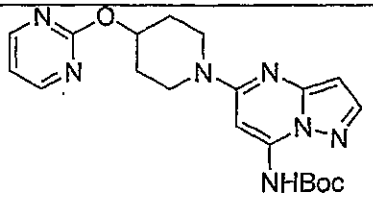
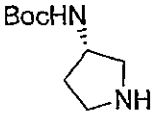
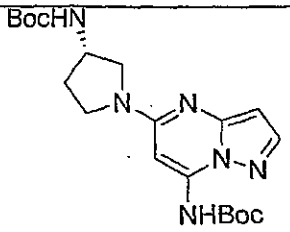
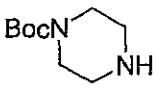
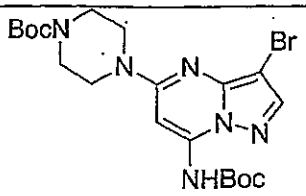
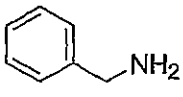
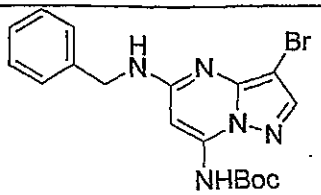
To a solution of Boc derivative (0.10 g, 0.37 mmol) from Preparative Example 2300-K in dioxane/DIPEA (3 ml/0.5 mL) at rt was added benzylamine (61 μL ,0.56 mmol) dropwise. The mixture was stirred at reflux for 12 h, cooled to rt, and concentrated under reduced pressure. The crude product was purified by preparative thin-layer chromatography (4 x 1000 μM plates) using a 30:1 mixture of CH_2Cl_2/MeOH as eluent to afford 65 mg (52% yield) of a yellow solid. LC-MS: 340.1 [M+H], purity 99%.

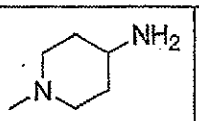
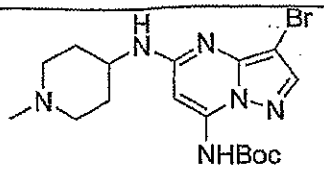
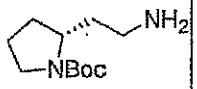
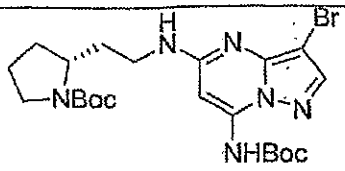
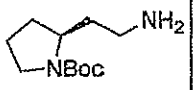
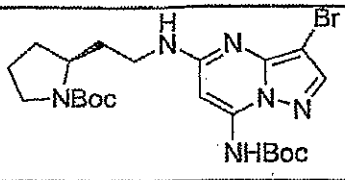
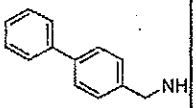
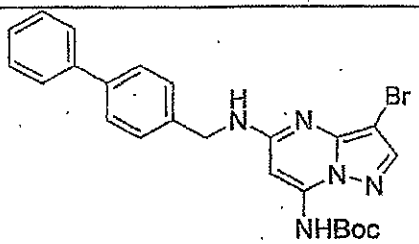
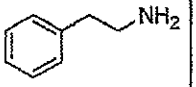
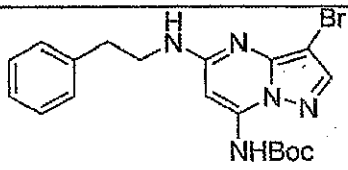
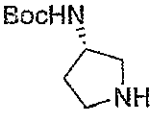
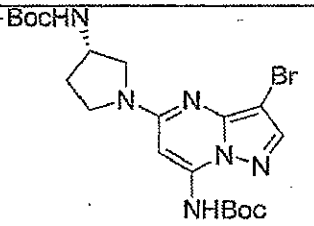
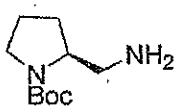
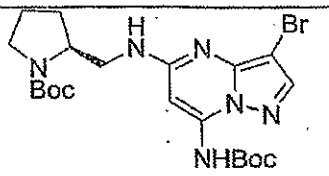
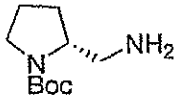
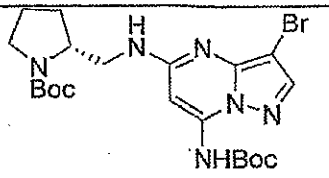
PREPARATIVE EXAMPLES 2501-K -2525-K:

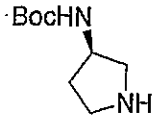
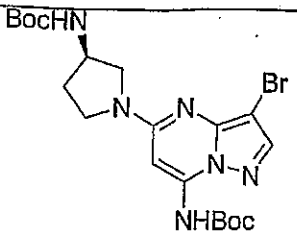
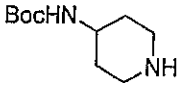
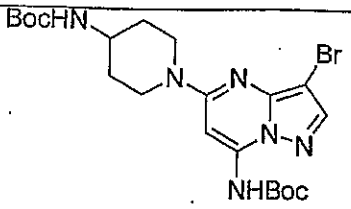
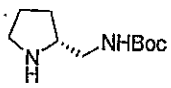
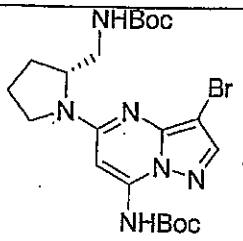
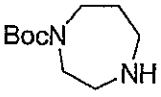
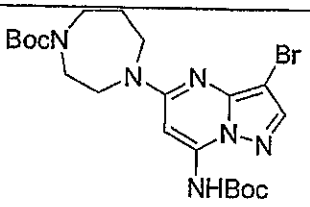
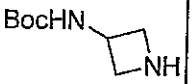
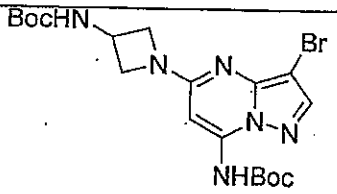
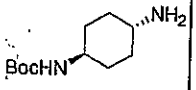
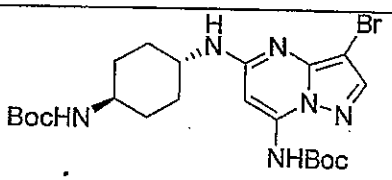
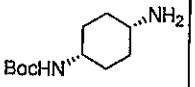
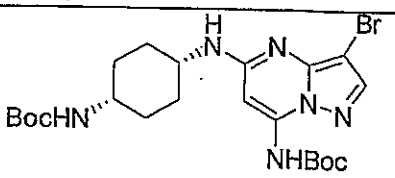
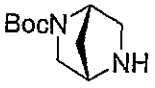
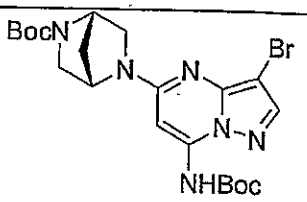
Following the procedure set forth in Preparative Example 2500-K but substituting commercially available amines, the compounds in Column 3 of Table 2500-K.

TABLE 2500-K

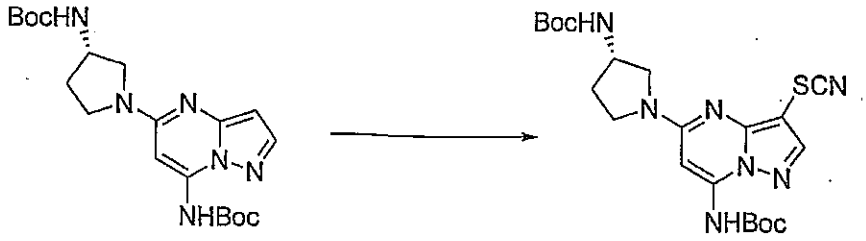
Prep. Ex.	*Column 2*	*Column 3*	1.Yield (%) 2. LC-MS
2501-K			1. 12 2. 368.1
2502-K			1. 30 2. 346.1

2503-K			1. 69 2. 463.1
2504-K			1. 98 2. 464.1
2505-K			1. 46 2. 449.1
2506-K			1. 72 2. 412.1
2507-K			1. 94 2. 419.2
2508-K			1. 90 2. 499.1
2509-K			1. 47 2. 420.1

2510-K			1. 57 2. 427.1
2511-K			1. 85 2. 525.1
2512-K			1. 80 2. 525.1
2513-K			1. 53 2. 494.1
2514-K			1. 49 2. 434.1
2515-K			1. 95 2. 499.1
2516-K			1. 54 2. 513.1
2517-K			1. 84 2. 513.1

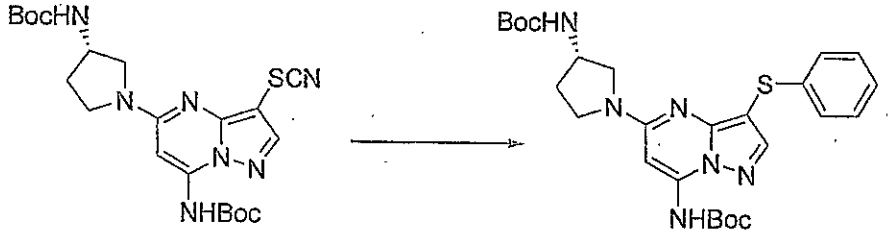
2618-K			1. 79 2. 499.1
2519-K			1. 78 2. 513.1
2520-K			1. 82 2. 513.1
2521-K			1. 92 2. 513.1
2522-K			1. 12 2. 485.3
2523-K			1. 28 2. 525.1
2524-K			1. 30 2. 525.1
2525-K			1. 89 2. 511.3

PREPARATIVE EXAMPLE 2600-K:



To a solution of Boc derivative (0.53 g, 1.27 mmol) from Preparative Example 2507-K in MeOH (4 mL) at 0 °C was added KSCN (0.15 g, 1.52 mmol) followed by Br_2 (75 µL, 1.52 mmol). The mixture was allowed to warm to rt , stir for 12h, and was concentrated under reduced pressure. The crude residue was suspended in EtOAc (7 mL) and sat. aq. $NaHCO_3$ (2 mL) and the layers were separated. The aqueous layer was extracted with EtOAc (2 x 7 mL) and the organic layers were combined. The organic layer was washed with brine (1 x 5 mL), dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by column chromatography using a 20:1 mixture of CH_2Cl_2/MeOH as eluent to afford 0.37g (65% yield) as a light yellow solid. LCMS: 476.3 [M+H} 99% purity.

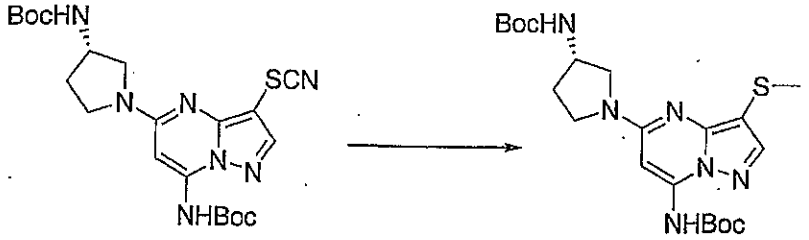
PREPARATIVE EXAMPLE 2700-K:



To a solution of thiocyanate (100 mg, 0.21 mmol) from Preparative Example 2600-K in THF (2 mL) at 0 °C was added $Pd(PPh_3)_4$ (24 mg, 0.021 mmol) followed by dropwise addition of PhMgBr (1.0 M in THF, 1.26 mL). The mixture was stirred for 12 h at rt and was treated with sat. aq NH_4Cl (2 mL) and CH_2Cl_2 (5 mL). The layers were separated and the aqueous layer was extracted with CH_2Cl_2 (2 x 5 mL). The organic layers were combined; washed with brine (1 x 3 mL), dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by preparative thin-layer chromatography (4 x 1000

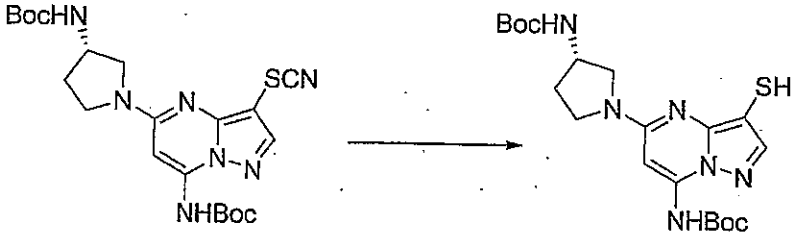
μM plates) using a 3:1 mixture of hexanes/EtOAc as eluent to afford 66 mg (60% yield) of a yellow semisolid. LC-MS: 527.1 [M+H], purity 99%.

PREPARATIVE EXAMPLE 2800-K:



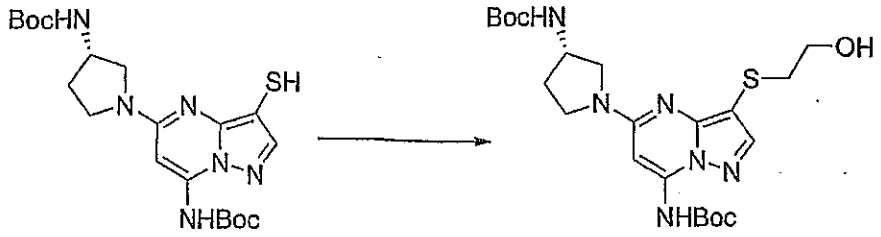
To a solution of thiocyanate (100 mg, 0.21 mmol) from Preparative Example 2600-K in MeOH (1.5 mL) at rt was added PPh_3 (55 mg, 0.21 mmol) in a single portion. The mixture was heated to reflux, stirred for 12 h, cooled and concentrated under reduced pressure. The crude product was purified by preparative thin-layer chromatography (4 x 1000 μM plates) using a 3:1 mixture of hexanes/EtOAc as eluent to afford 66 mg (68% yield) of a white solid. LC-MS: 465.3 [M+H], purity 99%.

PREPARATIVE EXAMPLE 2900-K:



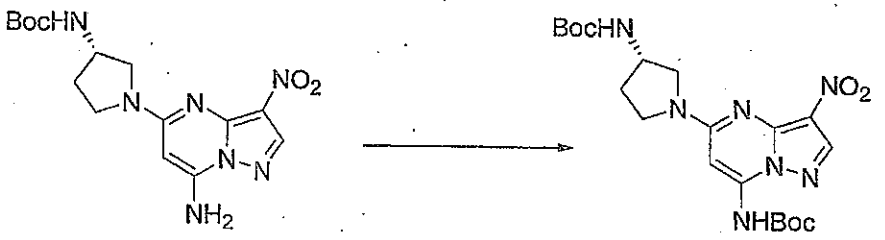
To a solution of thiocyanate (0.15 g, 0.32 mmol) from Preparative Example 2600-K in EtOH (2.5 mL) at rt was added a 0.1 M soln. of KH_2PO_4 (0.1 mL) followed by DTT (0.19 g, 1.26 mmol). The resulting mixture was stirred for 12 h at rt and was concentrated under reduced pressure. The resulting semisolid was dissolved in CH_2Cl_2 (5 mL) and was washed sequentially with H_2O (3 x 2 mL) and brine (3 x 2 mL). The organic layer was dried (Na_2SO_4), filtered, and concentrated under reduced pressure to afford 0.13 g (90% crude yield) of a light yellow semisolid. This material was taken on crude without purification.

PREPARATIVE EXAMPLE 3000-K:



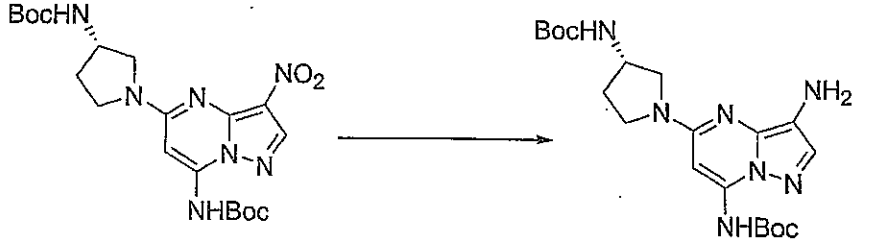
To a solution of the crude sulfhydryl derivative (80 mg, 0.18 mmol) from Preparative Example 2900-K in DMF (1.5 mL) at 0 °C was added K_2CO_3 (73 mg, 0.53 mmol) followed by bromoethanol (37 μL, 0.53 mmol). The resulting mixture was stirred for 72 h at rt where upon the mixture was diluted with EtOAc (5 mL) and water (2 mL). The layers were separated and the organic layer was washed with H_2O (3 x 2 mL) and brine (3 x 2 mL). The organic layer was dried (Na_2SO_4), filtered, and concentrated under reduced pressure to afford a yellow semisolid. The crude product was purified by preparative thin-layer chromatography (4 x 1000 μM plates) using a 1:1 mixture of hexanes/EtOAc as eluent to afford 37 mg (41% yield) of a light brown semisolid. LC-MS: 495.1 [M+H], purity 99%.

PREPARATIVE EXAMPLE 3100-K:



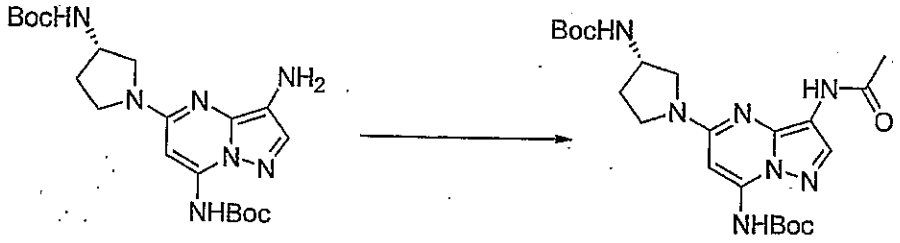
The title compound was prepared according to the procedure outlined in Preparative Example 2300-K except utilizing the 3-nitro derivative (0.20 g, 0.55 mmol) from Preparative Example 1800-K to afford 0.25 g (97% yield) of a yellow semisolid. LC-MS: 464.3 [M+H], purity 99%.

PREPARATIVE EXAMPLE 3200-K:



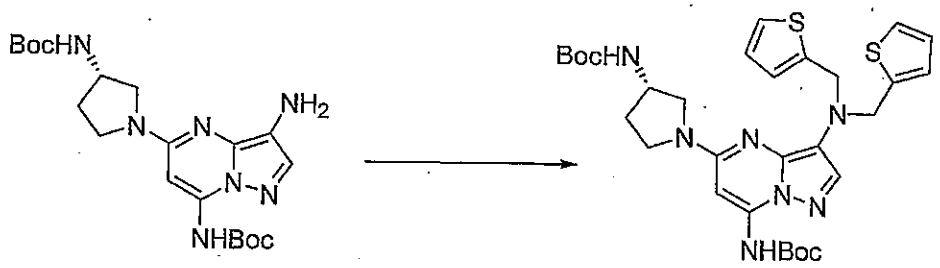
To a solution of 3-nitro adduct (70 mg, 0.15 mmol) from Preparative Example 3100-K in MeOH (2 mL) at rt was added Pd/C (10%, 25 mg dry). The mixture was stirred under H_2 (1 atm) for 12h whereupon the mixture was filtered thru a pad of Celite. The pad was washed generously with MeOH (2 x 4 mL) and the filtrate was concentrated under reduced pressure to afford 64 mg (98% yield) of a pink solid. LC-MS: 434.1 [M+H], purity 90%.

PREPARATIVE EXAMPLE 3300-K:



To a solution of 3-amino adduct (60 mg, 0.14 mmol) from Preparative Example 3200-K in CH_2Cl_2 (3 mL) at 0 °C was added Et_3N (29 μL, 0.21 mmol) followed by AcCl (12 μL, 0.17 mmol). The resulting mixture was stirred at rt for 12 h and was diluted with brine (3 mL) and CH_2Cl_2 (3 mL). The layers were separated and the aqueous layer was extracted with CH_2Cl_2 (2 x 3 mL). The organic layers were combined, dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by preparative thin-layer chromatography (2 x 1000 μM plates) using a 20:1 mixture of CH_2Cl_2/MeOH as eluent to afford 65 mg (68% yield) of a pink semisolid. LC-MS: 476.1 [M+H], purity 99%.

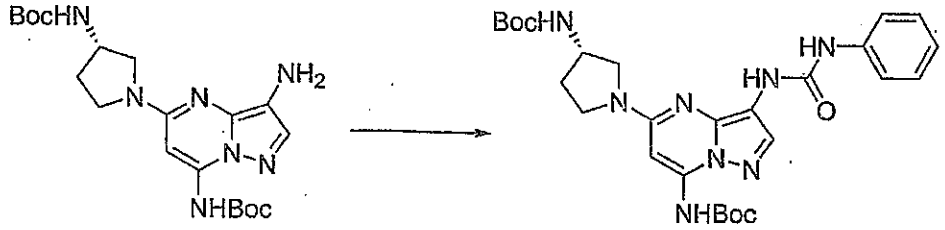
PREPARATIVE EXAMPLE 3400-K:



To a solution of 3-amino adduct (69 mg, 0.16 mmol) from Preparative Example 3200-K in CH_2Cl_2 (1 mL)/AcOH (0.1 mL) at 0 °C was added 2-thiophenecarboxaldehyde (18 μL, 0.19 mmol) followed by $NaB(OAc)_3H$ (41 mg, 0.19 mmol). The resulting mixture was stirred at rt for 12 h and was diluted with

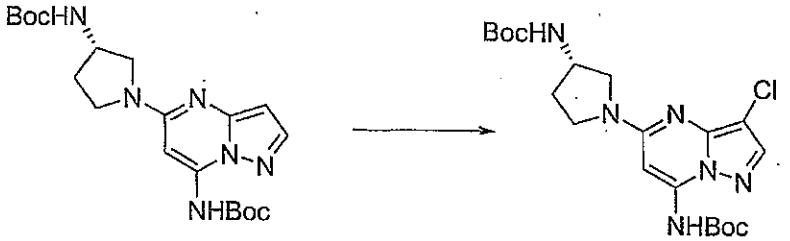
1N NaOH (1 mL) and CH_2Cl_2 (3 mL). The layers were separated and the aqueous layer was extracted with CH_2Cl_2 (2 x 3 mL). The organic layers were combined, dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by preparative thin-layer chromatography (2 x 1000 μM plates) using a 30:1 mixture of CH_2Cl_2/MeOH as eluent to afford 97 mg (96% yield) of a brown/orange oil. LC-MS: 626.1 [M+H], purity 96%.

PREPARATIVE EXAMPLE 3500-K:



To a solution of 3-amino adduct (90 mg, 0.21 mmol) from Preparative Example 3200-K in CH_2Cl_2 (1.5 mL) at 0 °C was added Et_3N (32 μL, 0.31 mmol) followed by PhNCO (27 μL, 0.25 mmol). The resulting mixture was stirred at rt for 12 h and was diluted with sat. aq. $NaHCO_3$ (2 mL) and CH_2Cl_2 (5 mL). The layers were separated and the aqueous layer was extracted with CH_2Cl_2 (2 x 3 mL). The organic layers were combined, washed with brine (1 x 3 mL), dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by preparative thin-layer chromatography (2 x 1000 μM plates) using a 20:1 mixture of CH_2Cl_2/MeOH as eluent to afford 110 mg (95% yield) of a white solid. LC-MS: 553.1 [M+H], purity 99%.

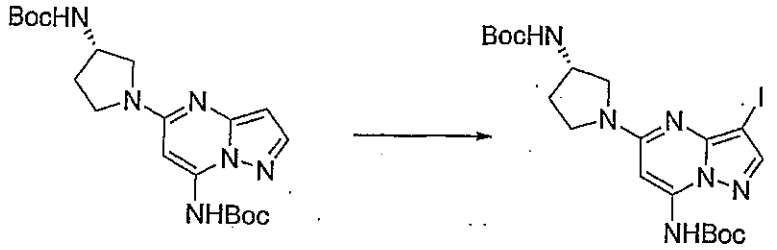
PREPARATIVE EXAMPLE 3600-K:



To a solution of 3-H adduct (0.15 g, 0.36 mmol) from Preparative Example 2507-K in CH_3CN (2 mL) at 0 °C was added NCS (48 mg, 0.36 mmol) in a single portion. The mixture was stirred for 3h at 0 °C whereupon the mixture was concentrated under reduced pressure. The crude product was purified by preparative thin-layer chromatography (4 x 1000 μM plates) using a 2:1 mixture

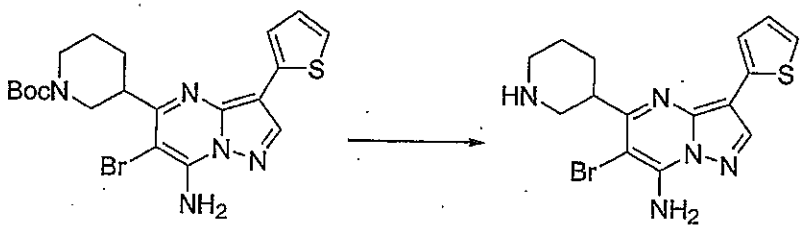
of hexanes/EtOAc as eluent to afford 0.16 g (98% yield) of a light yellow semisolid. LC-MS: 453.1 [M+H], purity 83%.

PREPARATIVE EXAMPLE 3700-K:



The title compound was prepared according to the procedure outlined in Preparative Example 3600-K utilizing the 3-H derivative (0.20 g, 0.55 mmol) from Preparative Example 2507-K and NIS (0.13 g, 0.60 mmol) to afford 0.25 g (97% yield) of a yellow semisolid. LC-MS: 464.3 [M+H], purity 99%.

EXAMPLE 100-K:



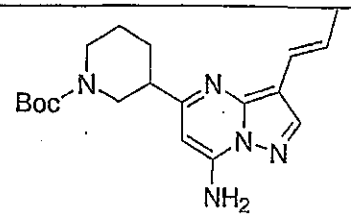
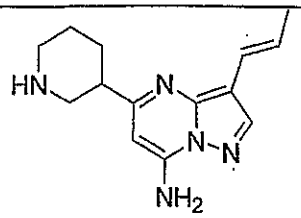
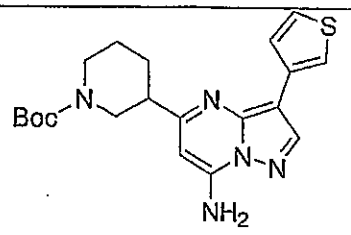
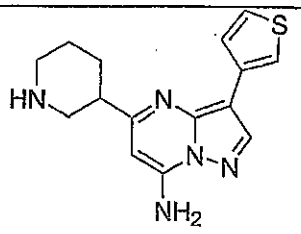
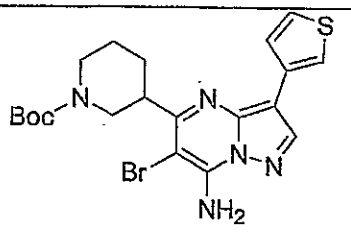
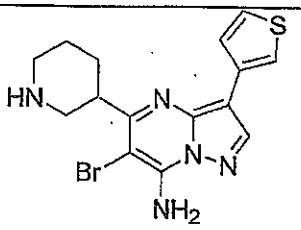
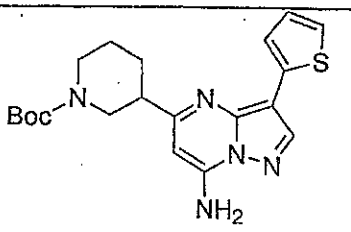
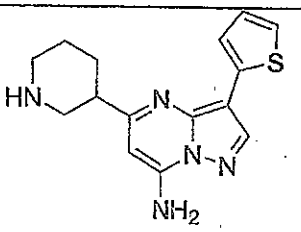
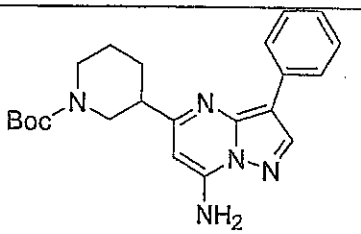
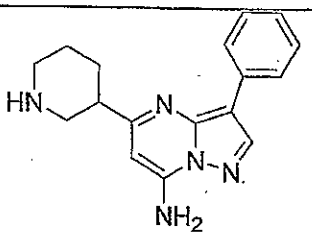
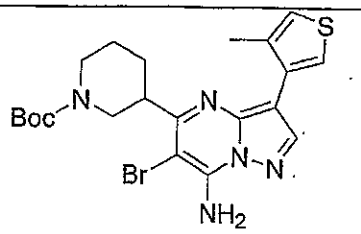
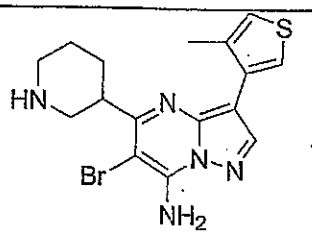
TFA (0.3 mL) was added under N_2 to a stirred solution of the product from Preparative Example 700-K (18 mg, 0.04 mmol) in CH_2Cl_2 (1.5 mL). The mixture was stirred for 1 hr, the solvent was evaporated, and the residue was mixed with solid Na_2CO_3 (100 mg) and 10:1 CH_2Cl_2/MeOH (2.0 mL). The mixture was stirred under N_2 for 10 min, the solution was withdrawn and loaded onto a silica gel preparative TLC plate, which was then developed with 10:1 CH_2Cl_2/7N NH_3 in MeOH. Pale yellow solid (5 mg, 35%) was obtained. LC-MS: 380 [M+H]. Mp = 106-109°C. SCH 773943

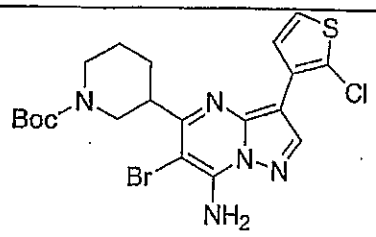
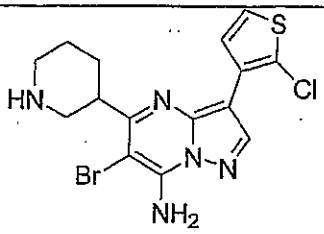
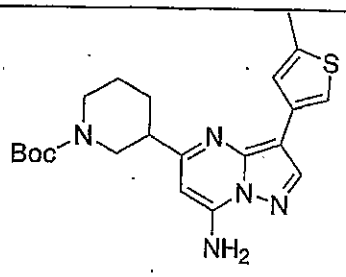
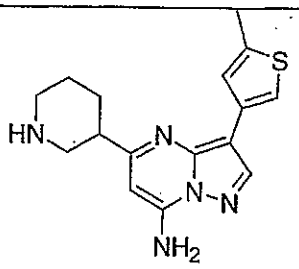
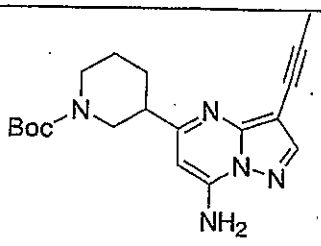
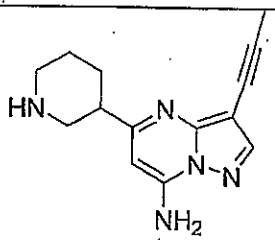
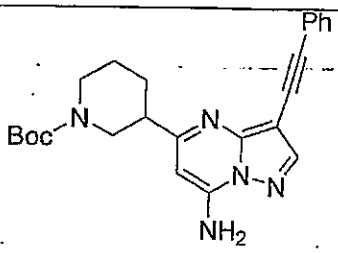
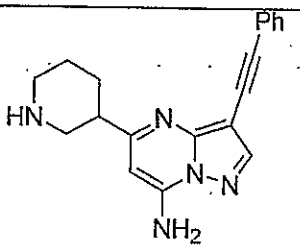
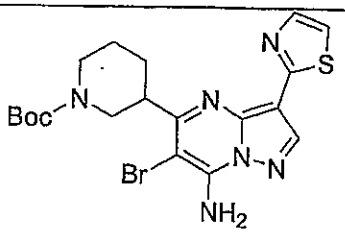
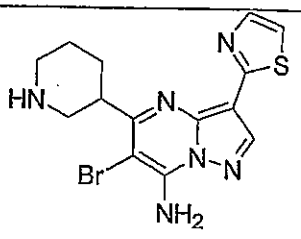
EXAMPLES 101-K -123-K:

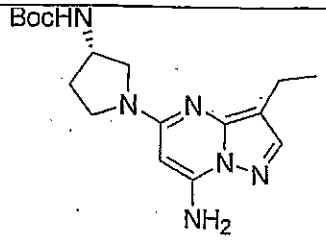
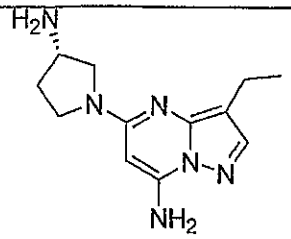
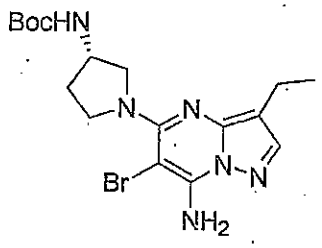
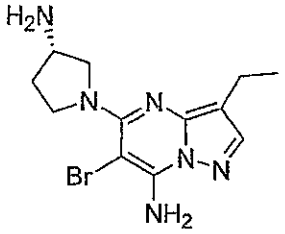
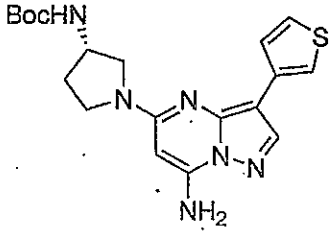
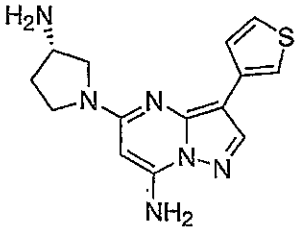
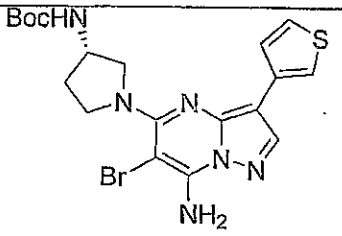
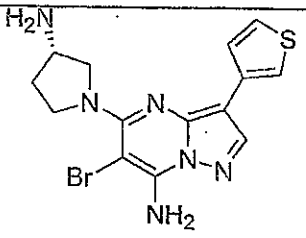
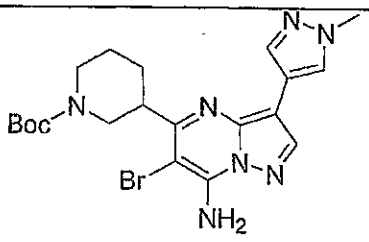
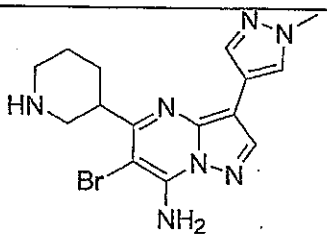
By essentially the same procedure set forth in Example 100-K only substituting the compounds shown in Column 2 of Table 800-K, the compounds shown in Column 3 of Table 800-K were prepared.

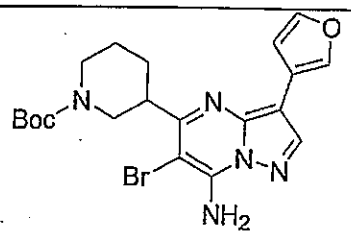
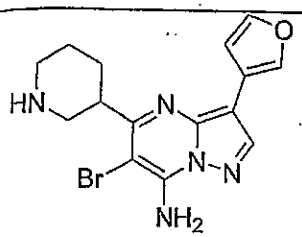
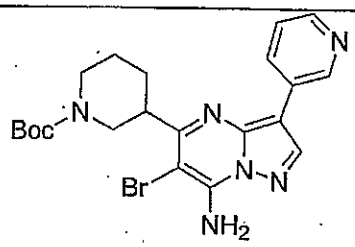
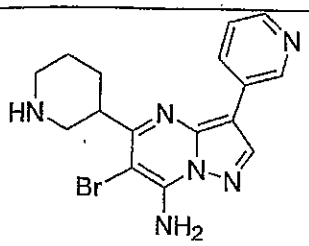
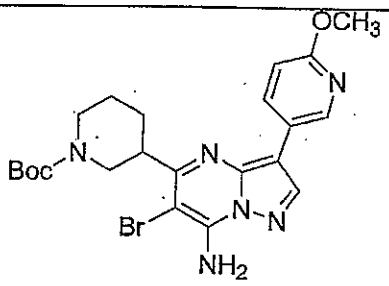
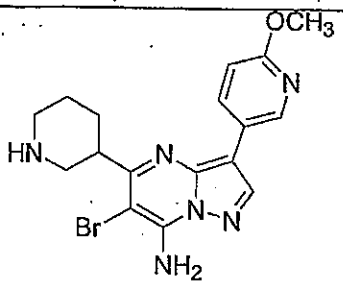
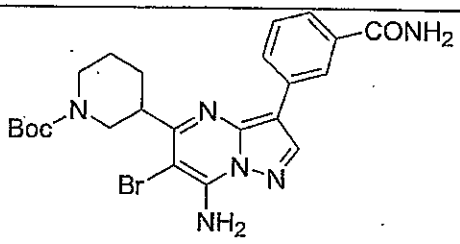
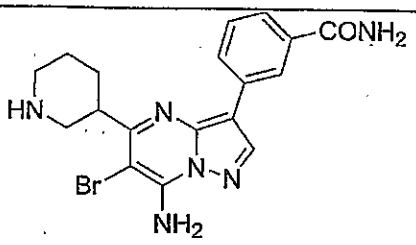
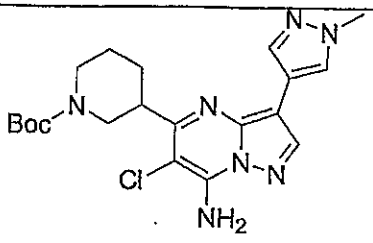
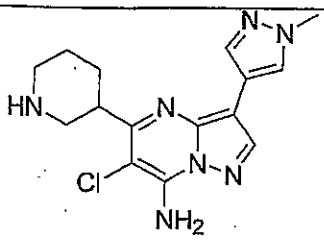
TABLE 800-K

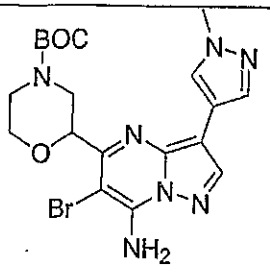
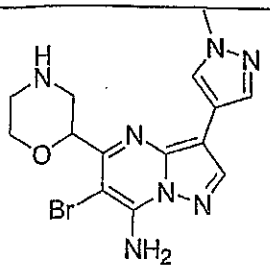
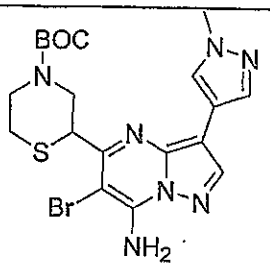
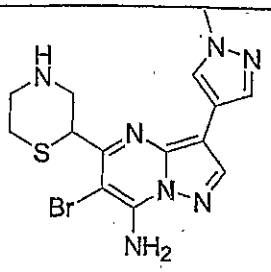
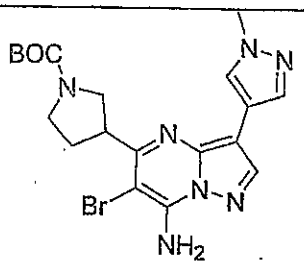
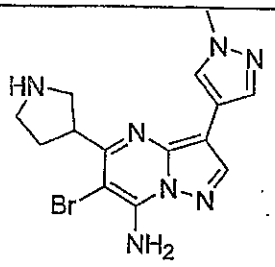
Ex.	Column 2	Column 2	Data

101 -K			LCMS: MH^+=258 mp = 200-202°C
102 -K			LCMS: MH^+=300 mp = 104-106°C
103 -K			LCMS: MH^+=380
104 -K			LCMS: MH^+=300 mp = 191-193°C
105 -K			
106 -K			mp = 157-159°C

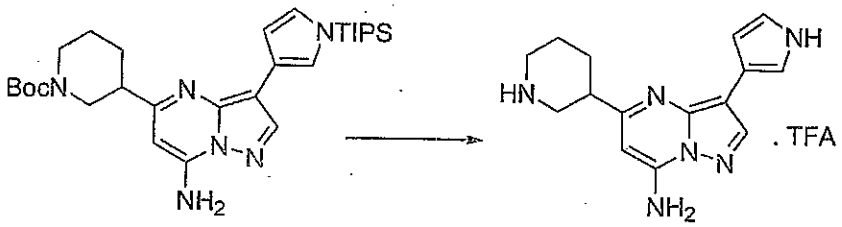
107 -K			LCMS: MH^+=414 mp = 96-99°C
108 -K			LCMS: $M2H^+$=394
109 -K			LCMS: MH^+=256 wax
110 -K			LCMS: MH^+=318 wax
111 -K			LCMS: M^+=379 mp = 247-249°C

112 -K			LCMS: MH^+=247 mp = 148-150°C
113 -K			LCMS: MH^+=327 mp = 114-116°C
114 -K			LCMS: MH^+=301 mp = 227-229°C
115 -K			LCMS: MH^+=379 mp = 156-158°C
118 -K			LCMS: M^+=376 mp = 95-98°C

119 -K			LCMS: M^+=362 mp = 161-163°C.
120 -K			LCMS: M^+=373 mp = 220-222°C
121 -K			LCMS: M^+=403 mp = 95-98°C
122 -K			LCMS: M^+=415 mp = 90-93°C
123 -K			LCMS: M^+=332 mp = 200-203°C

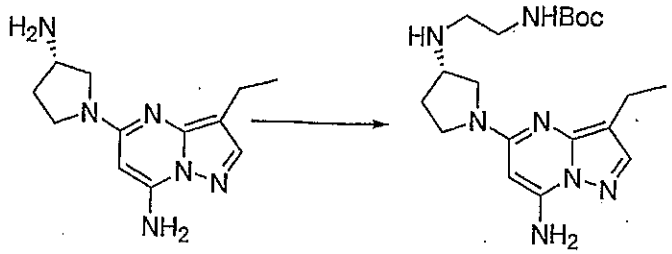
124 -K			
125 -K			
126 -K			

EXAMPLE 200-K:



TFA (0.5 mL) was added under N_2 to a stirred solution of the product from Preparative Example 604-K (30 mg, 0.056 mmol) in CH_2Cl_2 (1.0 mL). The mixture was stirred for 4 hr, the solvent was evaporated, and the residue was mixed with solid NaHCO3 (200 mg) and 4:1 CH_2Cl_2/MeOH (5 mL). The mixture was stirred under N_2 for 30 min, additional CH2Cl2 (5 mL) was then added, the mixture was filtered through Celite, and the solvent was evaporated. Pale brown wax (31 mg) was obtained. LC-MS: 284 [M+2H].

EXAMPLE 300-K:

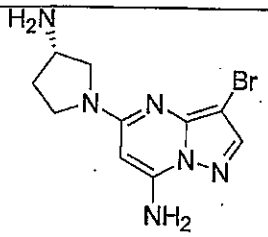
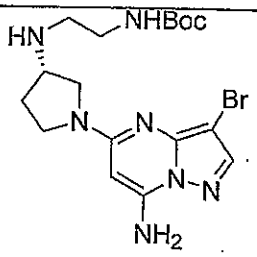
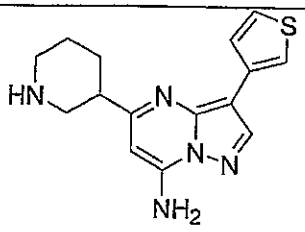
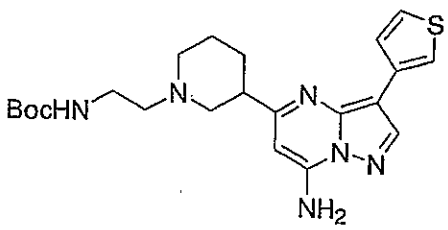


A mixture of the product from Example 112-K (70 mg, 0.29 mmol) and tert-butyl N-(2-oxoethyl)carbamate (45 mg,0.29 mmol) in anhydrous CH_2Cl_2 (2 mL) was stirred under N_2 at 25°C for 1 hr. Then $NaBH(OAc)_3$ (106 mg, 0.50 mmol) was added, the mixture was stirred for 1 hr, and then quenched with MeOH (2 mL). The solvents were evaporated and the residue was purified by column chromatography on silica gel with 20:1 CH_2Cl_2/ 7 N NH_3 in MeOH as eluent. White solid (84 mg, 76 %) was obtained. LC-MS: 390 [M+H]. Mp = 75-78°C.

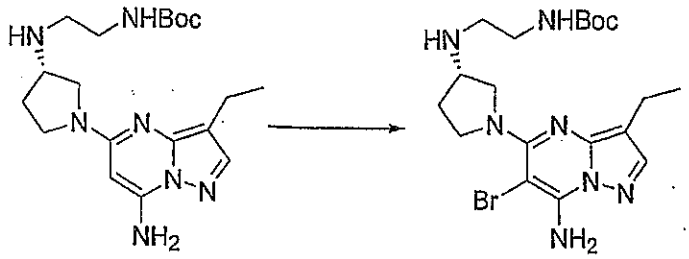
EXAMPLE 401-K - 402-K:

By essentially same procedure set forth in Example 300-K only substituting the compounds shown in Column 2 of Table 900-K, the compounds shown in Column of Table 900-K were prepared.

TABLE 900-K

Ex.	Column 2	Column 3	Data
401-K			white solid LC-MS: 440 [M+].
402-K			pale yellow wax . LC-MS: 443 [M+H].

EXAMPLE 500-K:

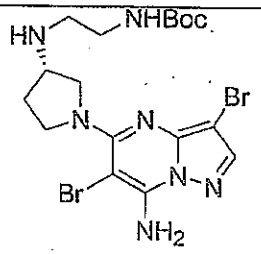
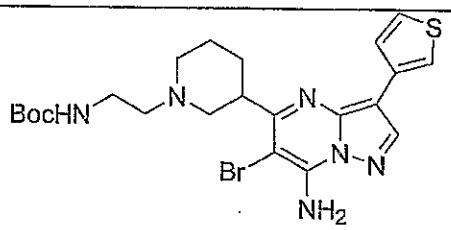


A solution of NBS (23 mg, 0.13 mmol) in anhydrous CH_3CN (1.0 mL) was added under N_2 to a stirred solution of the product from Example 300-K (50 mg, 0.13 mmol) in anhydrous CH_3CN (2.0 mL) and CH_2Cl_2 (1.0 mL). The mixture was stirred for 20 hr, the solvents were evaporated, and the residue was purified by column chromatography on silica gel with 20:1 CH_2Cl_2/MeOH as eluent. White solid (38 mg, 63 %) was obtained. LC-MS: 468 [M+]. Mp = 150-152°C.

EXAMPLE 500-K - 502-K:

By essentially same procedure set forth in Example 500-K, compounds shown in Column 2 of Table 930-K were prepared.

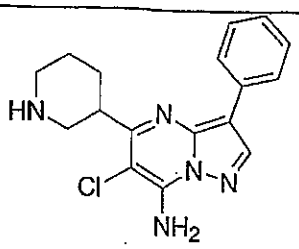
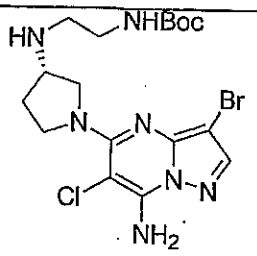
TABLE 930-K

Ex.	Column 2	Data
501-K		colorless solid. LC-MS: 520 [M+H].
502-K		pale yellow wax LC-MS: 523 [M+H]. SCH 785438

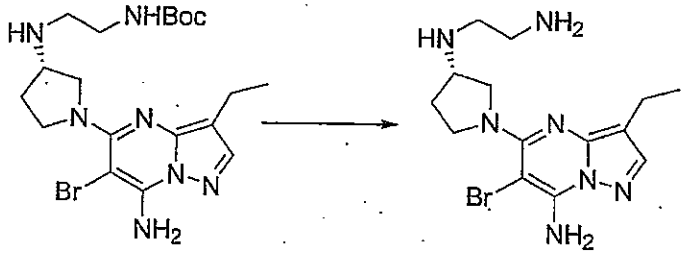
EXAMPLE 701-K - 702-K:

By essentially same procedure set forth in Example 500-K only substituting N-chlorosuccinimide, compounds shown in Column 2 of Table 940-K were prepared.

TABLE 940-K

Ex.	Column 2	Data
701 -K		LCMS: M^+= mp =
702 -K		white solid. LC-MS: 476 [M+H].

EXAMPLE 800-K:

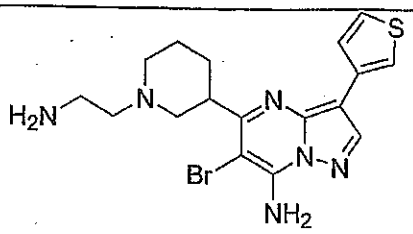
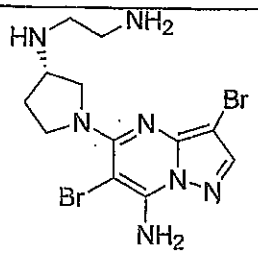
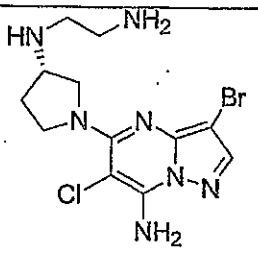


TFA (0.4 mL) was added under N_2 to a stirred solution of the product from Example 500-K (30 mg, 0.064 mmol) in CH_2Cl_2 (2.0 mL). The mixture was stirred for 1 hr, the solvent was evaporated, and the residue was mixed with solid Na_2CO_3 (200 mg) and 10:1 CH_2Cl_2/MeOH (2.0 mL). The mixture was stirred under N_2 for 10 min, the solution was withdrawn and loaded onto a silica gel preparative TLC plate, which was then developed with 10:1 CH_2Cl_2/7N NH_3 in MeOH. White solid (10 mg, 42%) was obtained. LC-MS: 368 [M+]. Mp = 132-135°C.

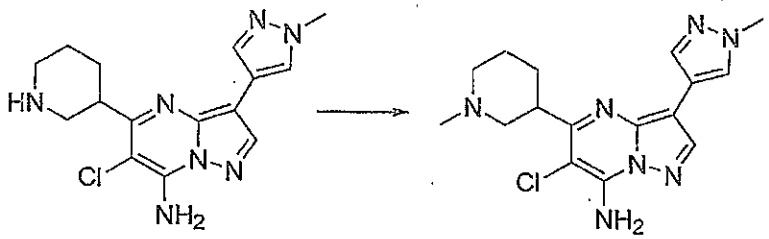
EXAMPLES 801-K - 803-K:

By essentially same procedure set forth in Example 800-K, compounds shown in Column 2 of Table 950-K were prepared:

TABLE 950-K

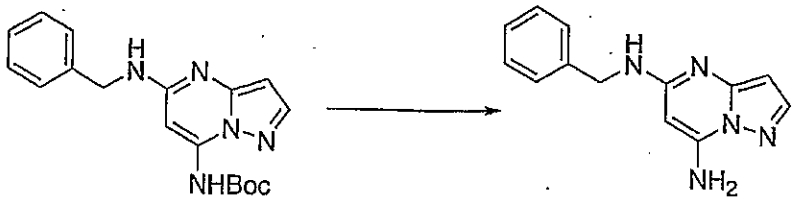
Ex.	Column 2	Data
801 -K		pale yellow solid LC-MS: 421 [M+H]. Mp = 148-150°C.
802 -K		white solid LC-MS: 420 [M+H]. Mp = 92-95°C.
803 -K		pale yellow solid. LC-MS: 376 [M+H]. Mp = 138-140°C.

EXAMPLE 900-K:



A solution of 37% HCHO (0.002 mL) in H_2O (0.02 mL) was added under N2 to a stirred solution of the product from Preparative Example 823-K (9 mg, 0.027 mmol) in anhydrous THF (1 mL) an MeOH (0.3 mL) and then the mixture was stirred under N_2 at 25°C for 5 hr. Then $NaBH(OAc)_3$ (8 mg, 0.038 mmol) was added and the mixture was stirred for 1 hr. The solvents were evaporated and the residue was purified by preparative TLC on silica gel with 10:1 CH_2Cl_2/ 7 N NH_3 in MeOH as eluent. White solid 8 mg, 85 %) was obtained. LC-MS: 346 [M+]. Mp = 92-95°C.

PREPARATIVE EXAMPLE 3800-K:

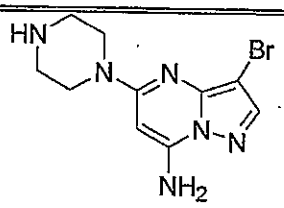


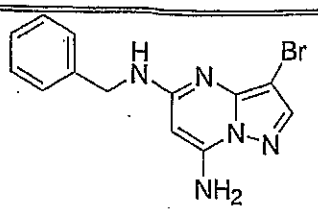
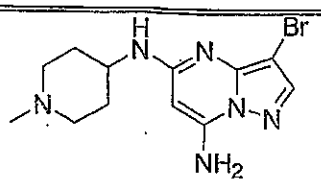
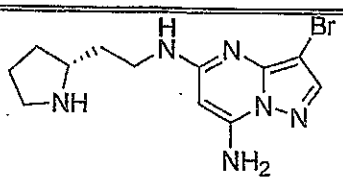
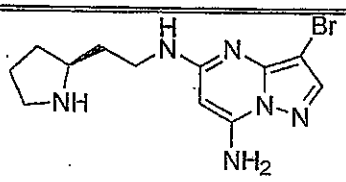
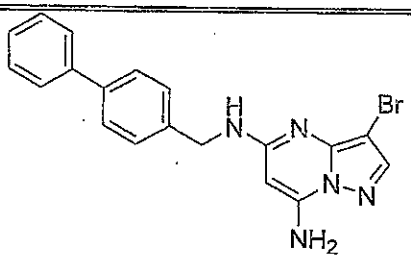
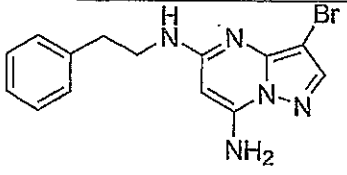
To a mixture of 7-amino adduct (65 mg, 0.19 mmol) from Preparative Example 2500-K in CH_2Cl_2 (2 mL) at 0 °C was added TFA (1 mL) dropwise. The resulting mixture was stirred for 12 h at rt and was concentrated under reduced pressure. The crude material was partitioned between EtOAc (5 mL) and sat. aq. Na_2CO_3 (2 mL) and the layers were separated. The aqueous layer was extracted with EtOAc (2 x 5 mL) and the organic layers were combined. The organic layer was washed with brine (1 x 3 mL), dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by preparative thin-layer chromatography (4 x 1000 μM plates) using a 20:1 mixture of CH_2Cl_2/MeOH (7M NH_3) as eluent to afford 41 mg (89% yield) as a brown solid. mp 111-113 °C; LC-MS: 240.0 [M+H], purity 98%.

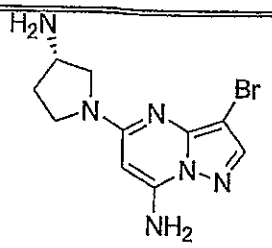
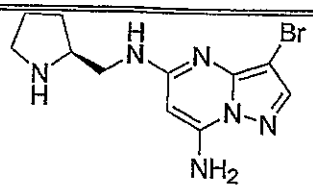
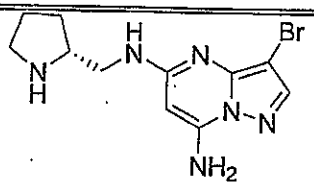
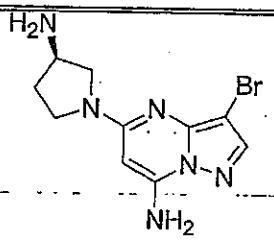
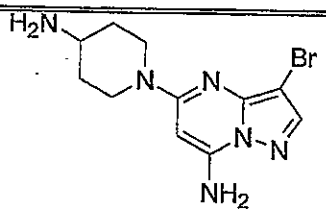
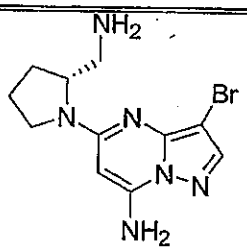
EXAMPLES 1000-1025-K:

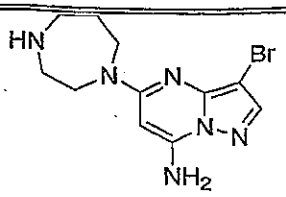
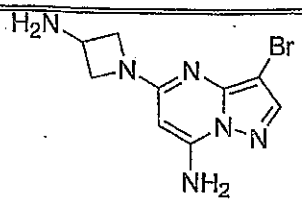
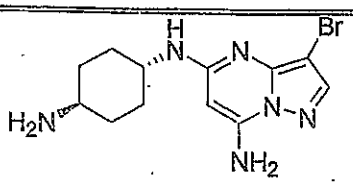
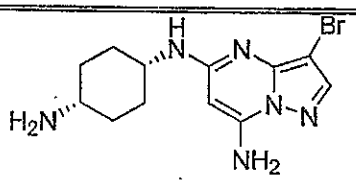
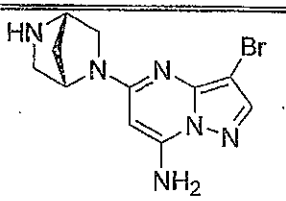
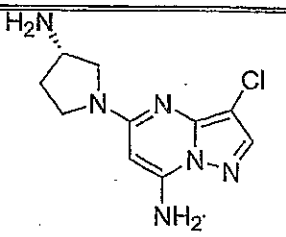
By essentially the same procedure set forth in Preparative Example 3800-K but utilizing the appropriate Boc derivatives, the compounds in Column 2 of Table 1000-K were prepared.

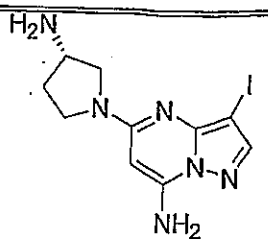
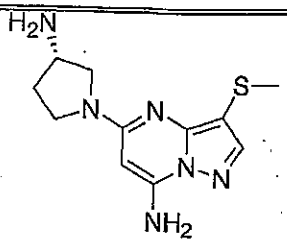
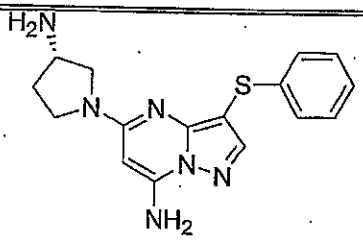
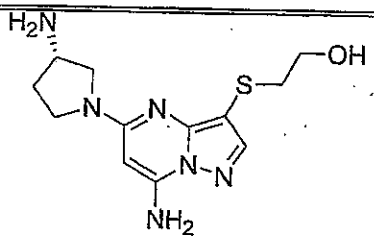
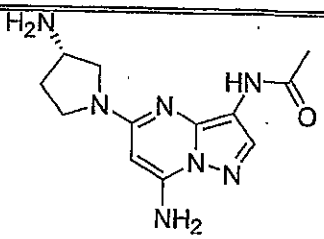
TABLE 1000-K

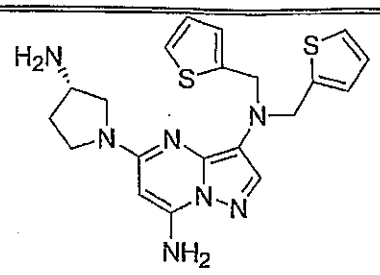
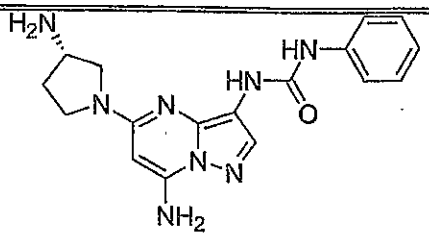
Ex.	*Column 2*	1.Yield (%) 2. LC-MS 3. mp
1000-K		1. 10 2. 297.1 3. 138-141

1001-K		1. 95 2. 320.1 3. 115-117
1002-K		1. 70 2. 327.1 3. 110-113
1003-K		1. 26 2. 325.1 3. 114-117
1004-K		1. 46 2. 325.1 3. 105-107
1005-K		1. 81 2. 394.1 3. 166-168
1006-K		1. 90 2. 334.1 3. 43-45

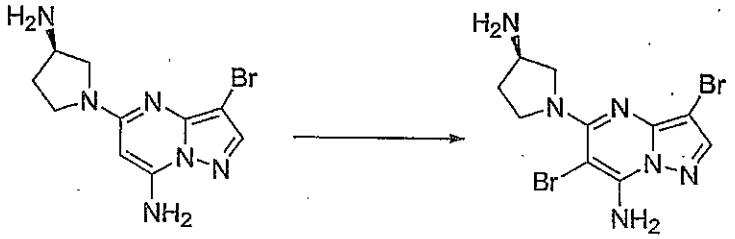
1007-K		1. 45 2. 299.1 3. 115-117
1008-K		1. 73 2. 313.1 3. 131-133
1009-K		1. 62 2. 313.1 3. 130-134
1010-K		1. 88 2. 299.1 3. 114-117
1011-K		1. 67 2. 313.1 3. 169-171
1012-K		1. 36 2. 313.1 3. 90-93

1013-K		1. 89 2. 313.1 3. 154-156
1014-K		1. 30 2. 285.3
1015-K		1. 83 2. 325.1 3. 115-117
1016-K		1. 80 2. 325.1 3. 105-107
1017-K		1. 64 2. 311.1 3. 98-102
1018-K		1. 79 2. 253.1 3. 188-190

1019-K		1. 20 2. 345.2 3. 127-129
1020-K		1. 70 2. 265.1 3. 120-122
1021-K		1. 87 2. 327.1 3. xxx-xxx
1022-K		1. 21 2. 295.1 3. xxx-xxx
1023-K		1. 17 2. 276.1

1024-K		1. 21 2. 295.1 3. xxx-xxx
1025-K		1. 20 2. 353.2 3. >200

EXAMPLE 1100-K:

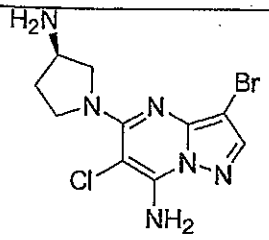
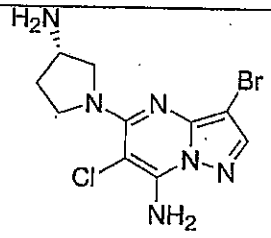
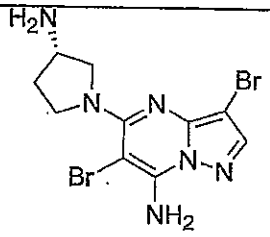
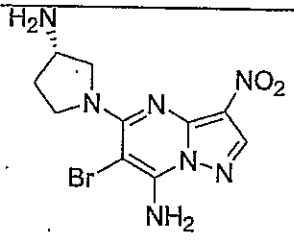
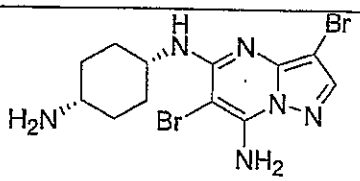


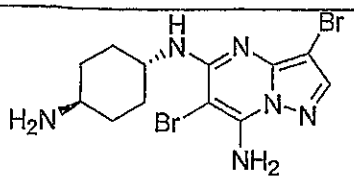
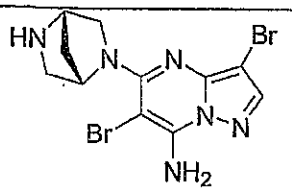
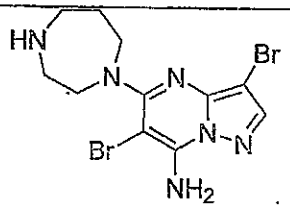
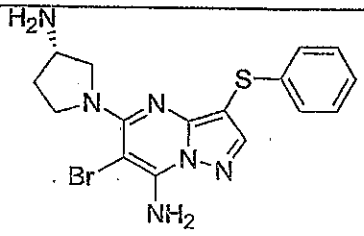
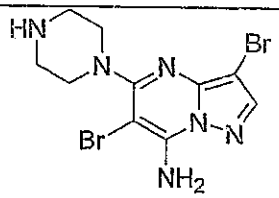
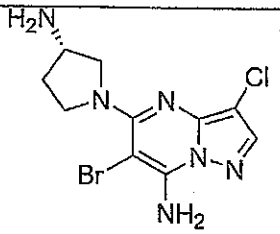
To a solution of 7-amino adduct (20 mg, 0.067 mmol) from Example 1010-K in CH_3CN at 0 °C was added NBS (12 mg, 0.067 mmol) in a single portion. The mixture was stirred for 2 hours at rt and was concentrated under reduced pressure. The crude product was purified by preparative thin-layer chromatography (2 x 1000 μM plates) using a 10:1 mixture of CH_2Cl_2/MeOH(7M NH_3) as eluent to afford 20 mg (79% yield) as a light orange solid. mp 159-161 °C. LC-MS: 377.1 [M+H], purity 99%.

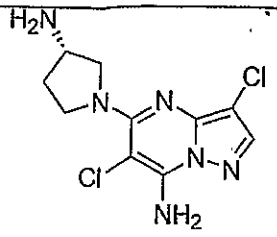
EXAMPLES 1101-K - 1112-K:

Following the procedure set forth in Example 1100-K and utilizing either NBS, NCS, or NIS the compounds in Column 3 of Table 1100-K were prepared.

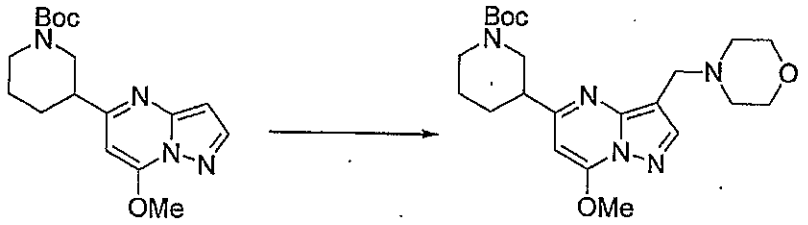
TABLE 1100-K

Prep. Ex.	*Column 2*	*Column 3*	1.Yield (%) 2. LC-MS 3. mp (°C)
1101-K	NCS		1. 41 2. 333.2 3. 123-125
1102-K	NCS		1. 64 2. 333.2 3. 122-125
1103-K	NBS		1. 66 2. 377.2 3, 159-161
1104-K	NBS		1. 38 2. 344.1 3. xxx-xxx
1105-K	NBS		1. 89 2. 403.1 3. 66-68

1106-K	NBS		1. 90 2. 403.1 3. 170-172
1107-K	NBS		1. 53 2. 389.1 3. xxx-xxx
1108-K	NBS		1. 48 2. 391.1 3. 142-144
1109-K	NBS		1. 59 2. 407.2 3. 156-158
1110-K	NBS		1. 75 2. 375.1 3. 134-136
1111-K	NBS		1. 63 2. 333.1 3. 153-155

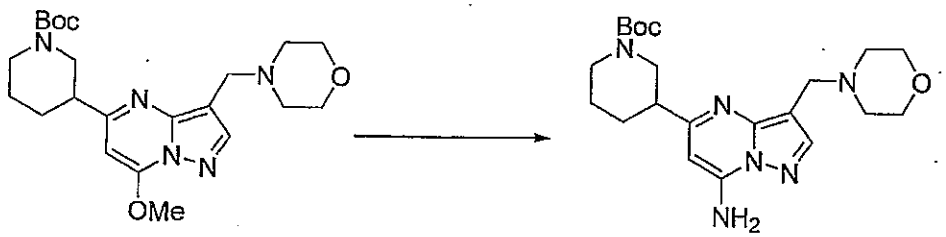
1112-K	NCS		1. 15 2. 287.1 3. >200

PREPARATIVE EXAMPLE 3900-K:



To a solution of 7-methoxy adduct (0.10 g, 0.30 mmol) from Preparative Example 402-K in AcOH at rt was added morpholine (29 μL, 0.33 mmol) followed by 40% formaldehyde in H_2O (1 mL). The resulting mixture was stirred at rt for 12h whereupon the mixture was concentrated under reduced pressure. The crude semisolid was partitioned between CH_2Cl_2 (5 mL) and sat. aq. $NaHCO_3$ (3 mL) and the layers were separated. The aqueous layer was extracted with CH_2Cl_2 (2 x 5 mL) and the organic layers were combined. The organic layer was washed with brine (1 x 3 mL), dried (Na_2SO_4), filtered, and concentrated under reduced pressure to afford 0.11 g (84% crude yield) of a light yellow semisolid. LC-MS: 432.1 [M+H], purity 94%.

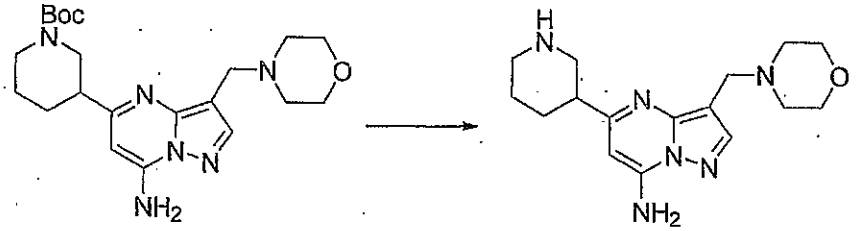
PREPARATIVE EXAMPLE 4000-K:



To a pressure tube charged with a stir bar was added 7-methoxy (96 mg, 0.22 mmol) from Preparative Example 3900-K followed by 2M NH_3 in IPA (2 mL) and conc. NH_4OH (2 mL). The tube was capped and was heated to 85 °C and stirred for 14h. The mixture was cooled to rt, concentrated under reduced pressure, and placed under high vacuum to remove trace volatiles. The crude product was purified by preparative thin-layer chromatography (4 x 1000 μM)

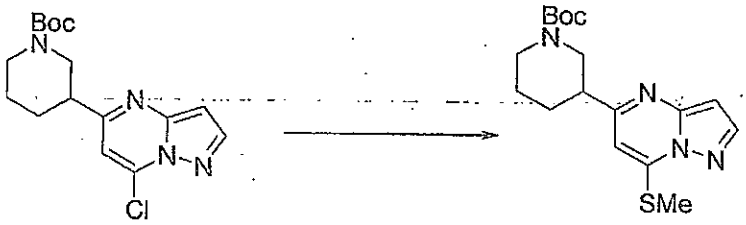
plates using a 10:1 mixture of CH_2Cl_2/MeOH as eluent to afford 40 mg (43% yield) of a yellow semisolid. LC-MS: 417.1 [M+H], purity 89%.

EXAMPLE 1200-K:



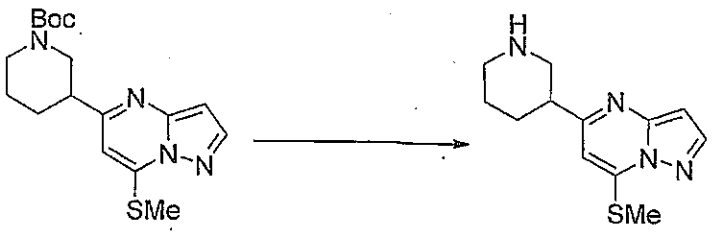
To a mixture of 7-amino adduct (40 mg, 0.096 mmol) from Preparative Example 4000-K was added 4N HCl/dioxane (3 mL) dropwise. The resulting mixture was stirred for 12 h at rt and was concentrated under reduced pressure. The crude material was taken up in CH_2Cl_2 (5 mL) and was concentrated under reduced pressure and this procedure was repeated 3x. The resulting solid was triturated with Et2O (3 x 3 m) and was concentrated under reduced pressure to afford 26 mg (75 % yield) of a light yellow solid. mp 190-193 °C; LC-MS: 317.1 [M+H], purity >90%.

PREPARATIVE EXAMPLE 4100-K:



The title compound was prepared from 7-chloro adduct (2.6 g, 7.7 mmol) from Preparative Example 200-K and NaSMe (0.65 g, 9.3 mmol) according to the procedure outlined in Preparative Example 1 to afford 2.6 g (98% yield) of a pale white solid. LC-MS: 349.1 [M+H], purity 98%.

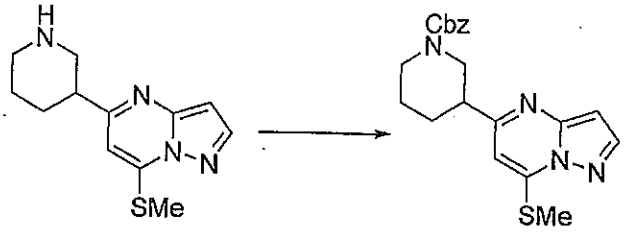
PREPARATIVE EXAMPLE 4200-K:



The above compound was prepared from 7-thiomethyl adduct (1.1 g, 3.1 mmol) from Preparative Example 4100-K according to the procedure outlined in

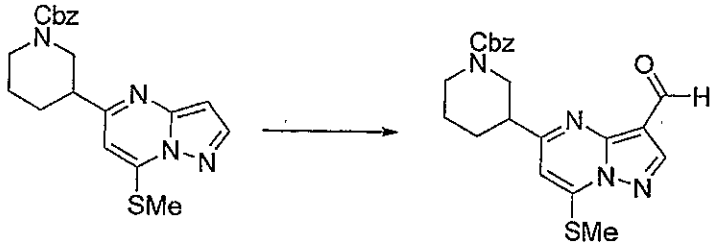
Example 1 to afford 0.70 g (90% yield) of a yellow semisolid. LC-MS: 249.1 [M+H], purity 98%.

PREPARATIVE EXAMPLE 4300-K:



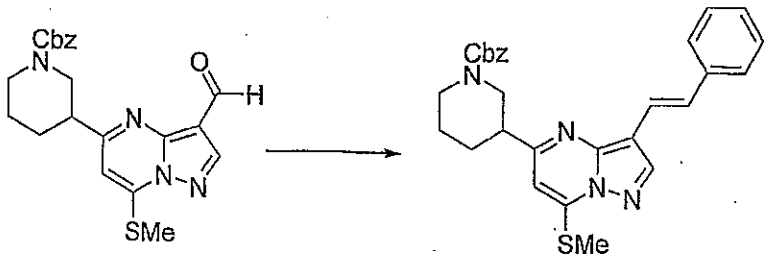
To a solution of thiomethyl adduct (1.3 g, 3.13 mmol) from Preparative Example 4200-K in CH_2Cl_2 (20 mL) at 0 °C was added Et_3N (0.65 mL, 4.7 mmol) followed by CbzCl (0.48 mL, 3.4 mmol). The resulting mixture was stirred for 12h at rt and the mixture was diluted CH_2Cl_2 (15 mL) and sat. aq. $NaHCO_3$ (5 mL) and the layers were separated. The aqueous layer was extracted with CH_2Cl_2 (2 x 10 mL) and the organic layers were combined. The organic layer was washed with brine (1 x 5 mL), dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by flash chromatography using CH_2Cl_2 as eluent to afford 0.96 g (80% yield) of a yellow oil. LC-MS: 383.1 [M+H], purity 91%.

PREPARATIVE EXAMPLE 4400-K:



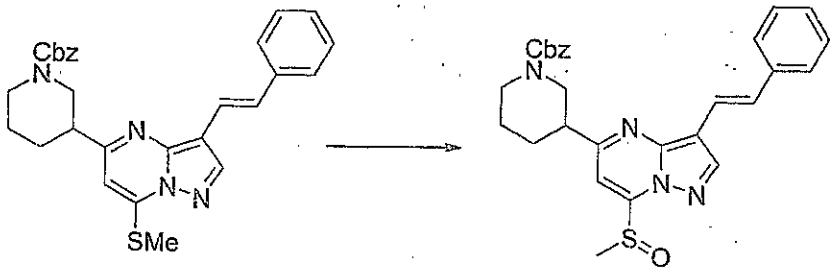
To a solution of 7-thiomethyl adduct (0.29 g, 0.75 mmol) from Preparative Example 4300-K in DMF (3 ml) at 0 °C was added $POCl_3$ (0.10 mL, 1.1 mmol). The resulting mixture was allowed to warm to rt and stir for 12h. The mixture was cooled to 0 oC and was treated with ice water (2 ml) and CH_2Cl_2 (5 mL). The layers were separated and the aqueous layer was extracted with CH_2Cl_2 (2 x 5 mL). The organic layers were combined, washed with brine (1 x 5 mL). The organic layer was dried (Na_2SO_4), filtered, and concentrated under reduced pressure to afford 0.28 g (91% yield) of a yellow solid. LC-MS: 433.2 [M+H], purity 84%.

PREPARATIVE EXAMPLE 4500-K:



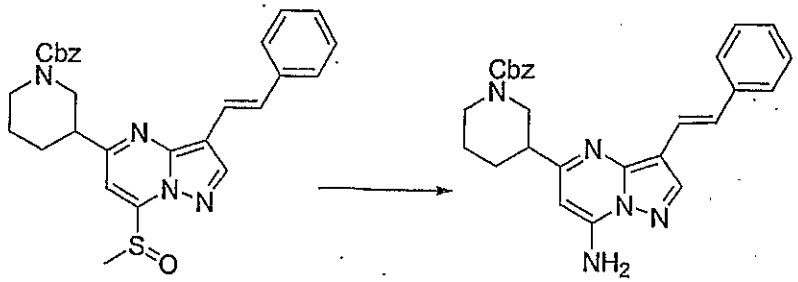
To a mixture of benzyltriphenyphosphonium bromide (0.64 g, 1.48 mmol) and NaH (59 mg, 1.48 mmol) in dry THF (3 mL) at rt was added thiomethyl adduct (0.21 g, 0.50 mmol) from Preparative Example 4400-K. The mixture was heated at reflux for 12h, cooled to rt, and diluted with CH_2Cl_2 (5 mL) and brine (2 mL). The layers were separated and the aqueous layer was extracted with CH_2Cl_2 (2 x 5 mL). The organic layers were combined, washed with brine (1 x 5 mL). The organic layer was dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by preparative thin-layer chromatography (2 x 1000 μM plates) using a 1:1 mixture of hexanes/EtOAc as eluent to afford 100 mg (42% yield) as a light orange solid.

PREPARATIVE EXAMPLE 4600-K:



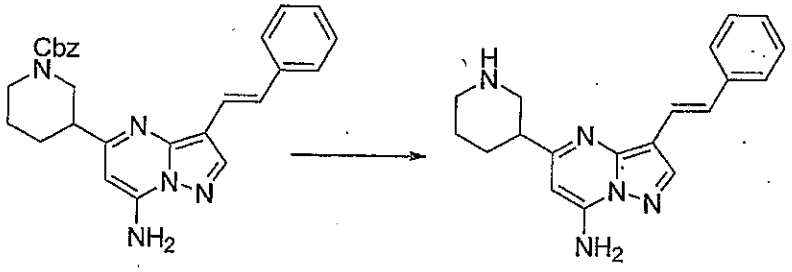
The title compound was prepared from 7-thiomethyl adduct (0.10 g, 0.21 mmol) from Preparative Example 4500-K according to the procedure outlined in Preparative Example 1500-K to afford 0.11 g (quantitative yield) of a yellow solid. LC-MS: 501.3 [M+H], purity 63%.

PREPARATIVE EXAMPLE 4700-K:



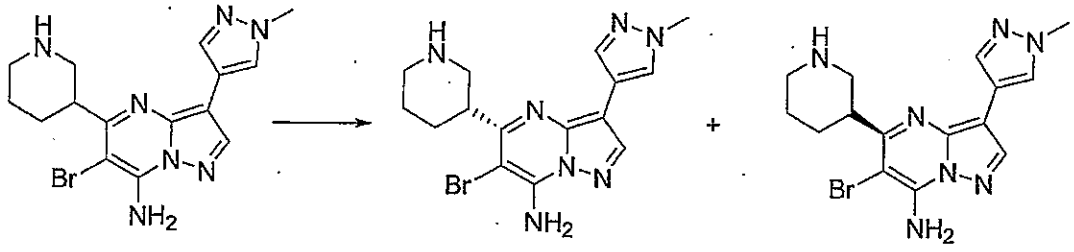
The title compound was prepared from 7-sulfoxide adduct (0.11 g, 0.21 mmol) from Preparative Example 4600-K according to the procedure outlined in Preparative Example 1800-K to afford 83 mg (40% yield) of a yellow solid. LC-MS: 454.1 [M+H], purity 90%.

EXAMPLE 1300-K:



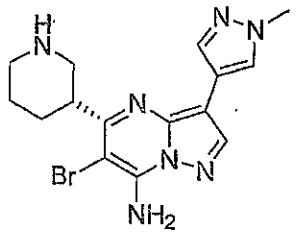
To a solution of 7-amino adduct (38 mg, 0.083 mmol) from Preparative Example 4700-K in CH_3CN (3 mL) at 0 °C was added TMSI (0.12 mL, 0.83 mmol) dropwise. The mixture was stirred for 12h at rt whereupon 1N NaOH (2 mL) was added. The mixture was extracted with EtOAc (5 x 5 mL) and the organic layers were combined. The organic layer was washed with brine (1 x 3 mL), dried (Na_2SO_4), filtered, and concentrated under reduced pressure. The crude product was purified by preparative thin-layer chromatography (2 x 1000 μM plates) using a 10:1 mixture of CH_2Cl_2/MeOH(7M NH_3) as eluent to afford 15 mg (56% yield) as a light yellow solid. mp 144-146 °C. LCMS: 319.4 [M+H} 90% purity.

EXAMPLE 1400-K – 1500-K:

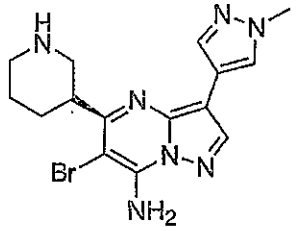


The compound prepared in Example 118-K was separated into individual enantiomers by semi-preparative HPLC using a ChiralPak OD column (flow rate = 9 mL/min) and a gradient of 85 : 15 hexanes : isopropanol with 0.2% diethylamine to 75 : 25 hexanes : isopropanol solution with 0.2% diethylamine as eluent.

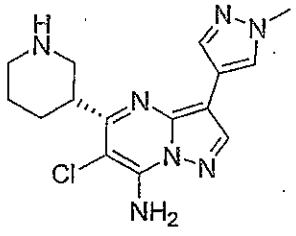
Example 1400-K (first eluting isomer, (R)): LCMS: M^+ = 376.



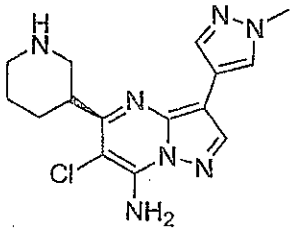
Example 1500-K (second eluting isomer, (S)): M^+ = LCMS: 376:



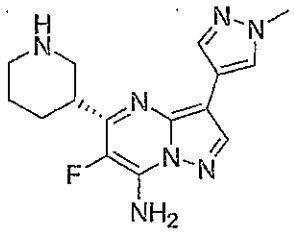
Example 1600K-, Example 1700-K, Example 1800-K and Example 1900-K: By a similar procedure, the following compounds can also be prepared:



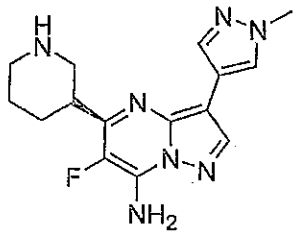
1600-K



1700-K



1800-K



1900-K

ASSAY:

BACULOVIRUS CONSTRUCTIONS: Cyclins A and E were cloned into pFASTBAC (Invitrogen) by PCR, with the addition of a GluTAG sequence

(EYMPME) at the amino-terminal end to allow purification on anti-GluTAG affinity columns. The expressed proteins were approximately 46kDa (cyclin E) and 50kDa (cyclin A) in size. CDK2 was also cloned into pFASTBAC by PCR, with the addition of a haemaglutinin epitope tag at the carboxy-terminal end (YDVPDYAS). The expressed protein was approximately 34kDa in size.

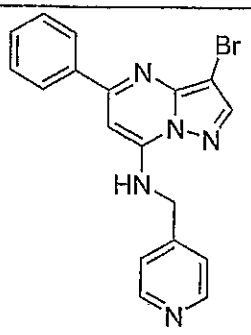
ENZYME PRODUCTION: Recombinant baculoviruses expressing cyclins A, E and CDK2 were infected into SF9 cells at a multiplicity of infection (MOI) of 5, for 48 hrs. Cells were harvested by centrifugation at 1000 RPM for 10 minutes. Cyclin-containing (E or A) pellets were combined with CDK2 containing cell pellets and lysed on ice for 30 minutes in five times the pellet volume of lysis buffer containing 50mM Tris pH 8.0, 0.5% NP40, 1mM DTT and protease/phosphatase inhibitors (Roche Diagnostics GmbH, Mannheim, Germany). Mixtures were stirred for 30-60 minutes to promote cyclin-CDK2 complex formation. Mixed lysates were then spun down at 15000 RPM for 10 minutes and the supernatant retained. 5ml of anti-GluTAG beads (for one liter of SF9 cells) were then used to capture cyclin-CDK2 complexes. Bound beads were washed three times in lysis buffer. Proteins were competitively eluted with lysis buffer containing 100-200ug/mL of the GluTAG peptide. Eluate was dialyzed overnight in 2 liters of kinase buffer containing 50mM Tris pH 8.0, 1mM DTT, 10mM MgCl2, 100uM sodium orthovanadate and 20% glycerol. Enzyme was stored in aliquots at -70°C.

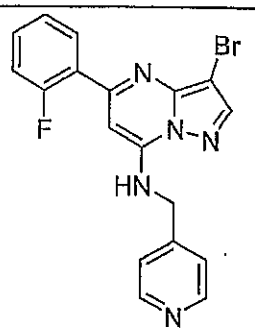
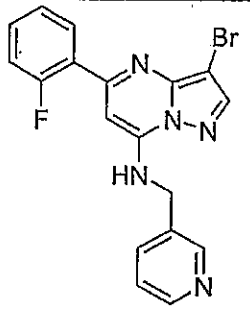
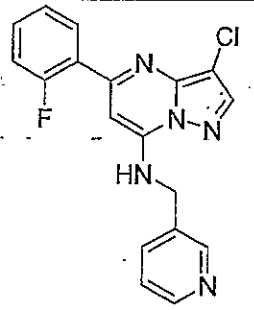
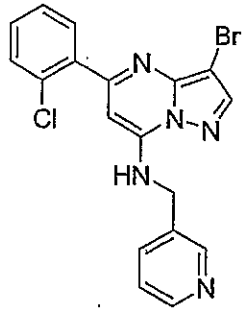
IN VITRO KINASE ASSAY: CDK2 kinase assays (either cyclin A or E-dependent) were performed in low protein binding 96-well plates (Corning Inc, Corning, New York). Enzyme was diluted to a final concentration of 50 μg/ml in kinase buffer containing 50mM Tris pH 8.0, 10mM $MgCl_2$, 1mM DTT, and 0.1mM sodium orthovanadate. The substrate used in these reactions was a biotinylated peptide derived from Histone H1 (from Amersham, UK). The substrate was thawed on ice and diluted to 2 μM in kinase buffer. Compounds were diluted in 10%DMSO to desirable concentrations. For each kinase reaction, 20 μl of the 50 μg/ml enzyme solution (1 μg of enzyme) and 20 μl of the 1 μM substrate solution

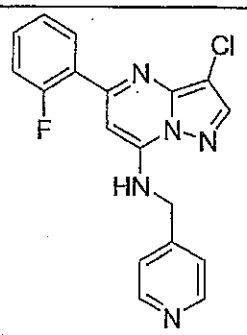
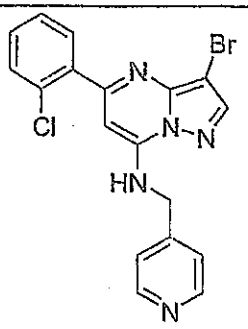
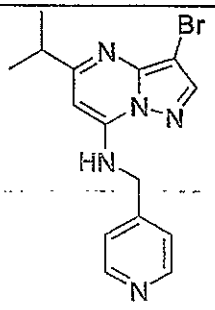
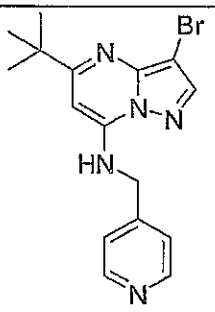
were mixed, then combined with 10 μl of diluted compound in each well for testing. The kinase reaction was started by addition of 50 μl of 4 μM ATP and 1 μCi of 33P-ATP (from Amersham, UK). The reaction was allowed to run for 1 hour at room temperature. The reaction was stopped by adding 200 μl of stop buffer containing 0.1% Triton X-100, 1mM ATP, 5mM EDTA, and 5 mg/ml streptavidine coated SPA beads (from Amersham, UK) for 15 minutes. The SPA beads were then captured onto a 96-well GF/B filter plate (Packard/Perkin Elmer Life Sciences) using a Filtermate universal harvester (Packard/Perkin Elmer Life Sciences.). Non-specific signals were eliminated by washing the beads twice with 2M NaCl then twice with 2 M NaCl with 1% phosphoric acid. The radioactive signal was then measured using a TopCount 96 well liquid scintillation counter (from Packard/Perkin Elmer Life Sciences).

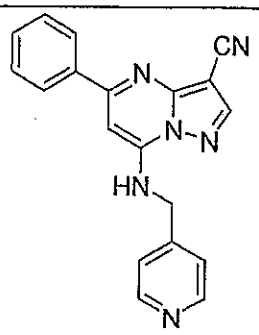
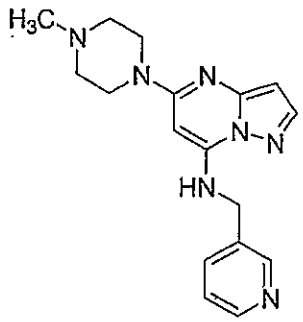
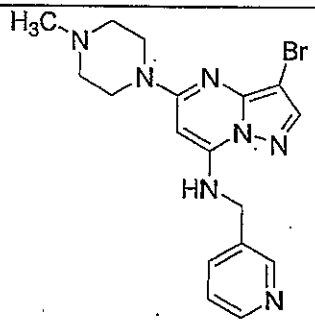
IC_{50} DETERMINATION: Dose-response curves were plotted from inhibition data generated, each in duplicate, from 8 point serial dilutions of inhibitory compounds. Concentration of compound was plotted against % kinase activity, calculated by CPM of treated samples divided by CPM of untreated samples. To generate IC_{50} values, the dose-response curves were then fitted to a standard sigmoidal curve and IC_{50} values were derived by nonlinear regression analysis. The thus-obtained IC_{50} values for the compounds of the invention are shown in **Table 87**. These kinase activities were generated by using cyclin A or cyclin E using the above-described assay.

Table 87

CMPD	Example	IC_{50} (μM)
	1	0.020 0.029

	3	0.032 0.024
	4	0.011
	5	0.021
	8	0.003

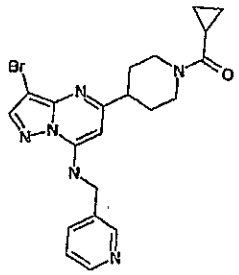
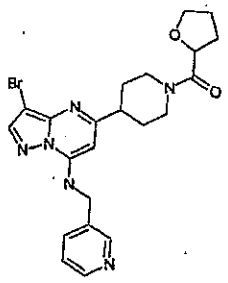
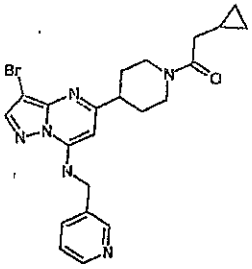
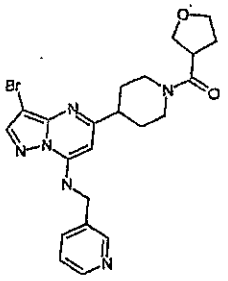
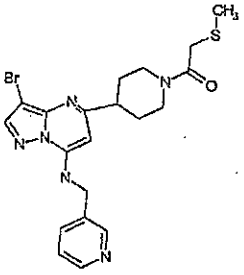
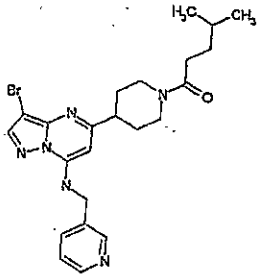
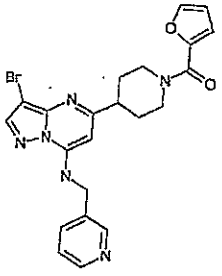
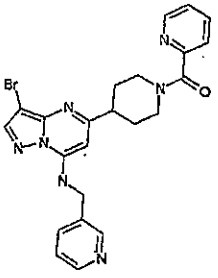
	6	0.064 0.029
	7	0.01 0.006
	10	0.042
	12	0.17

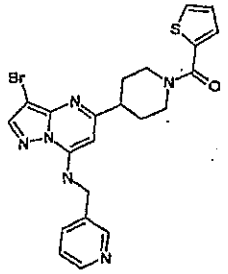
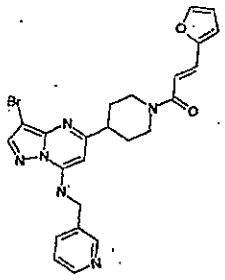
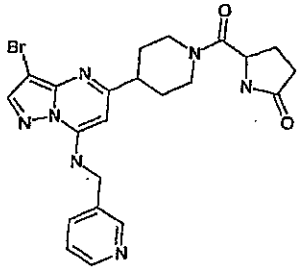
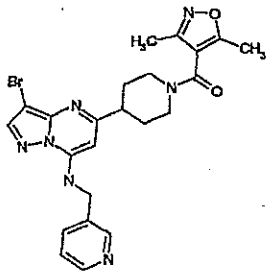
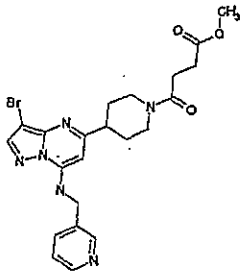
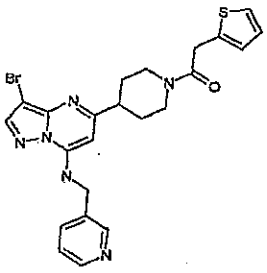
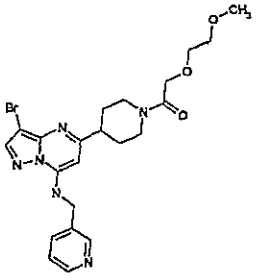
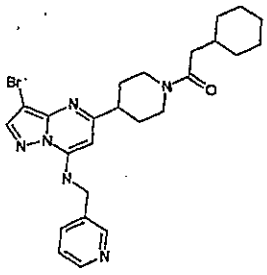
	16	0.62
	1	5.6
	3	0.14

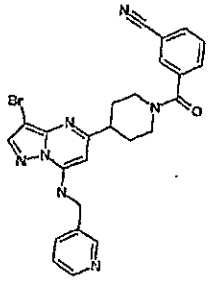
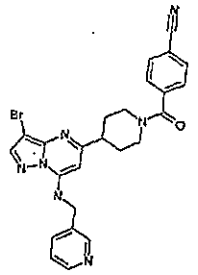
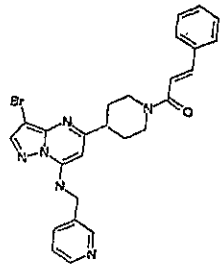
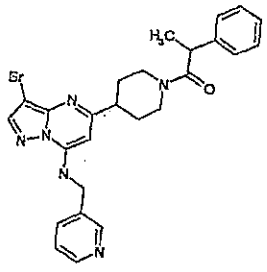
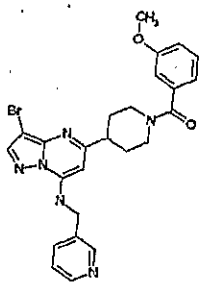
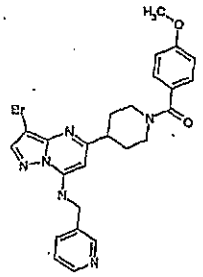
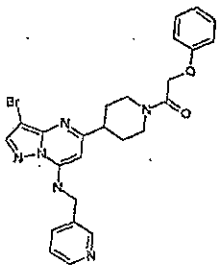
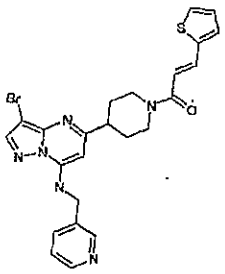
As demonstrated above by the assay values, the compounds of the present invention exhibit excellent CDK inhibitory properties.

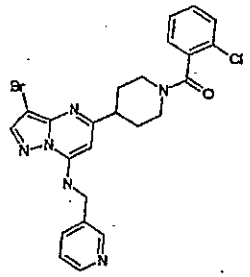
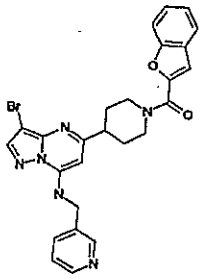
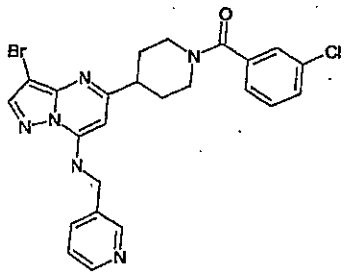
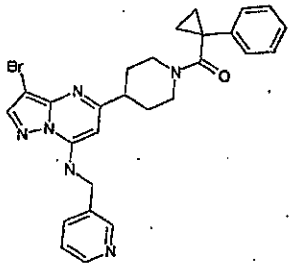
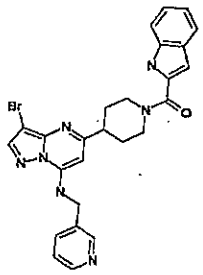
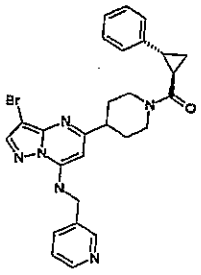
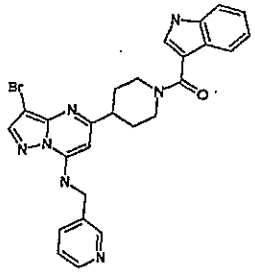
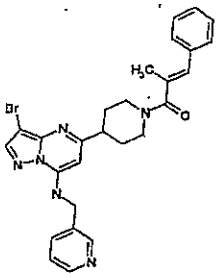
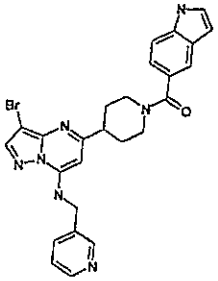
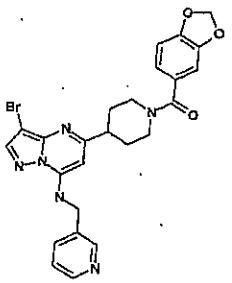
While the present invention has been described with in conjunction with the specific embodiments set forth above, many alternatives, modifications and other variations thereof will be apparent to those of ordinary skill in the art. All such alternatives, modifications and variations are intended to fall within the spirit and scope of the present invention.

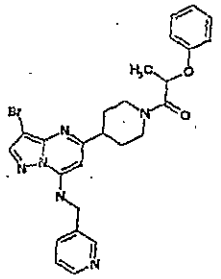
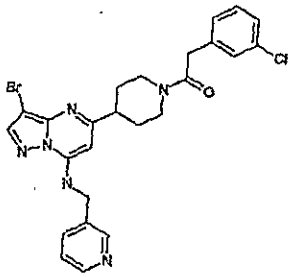
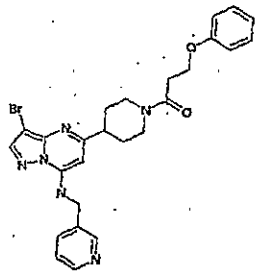
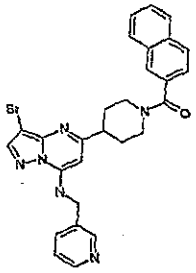
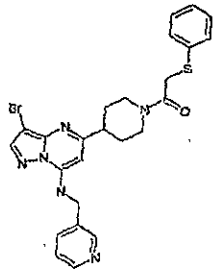
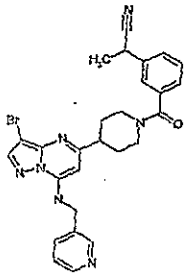
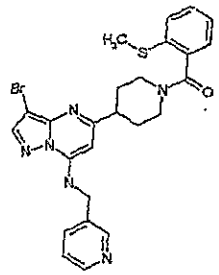
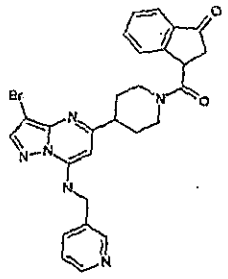
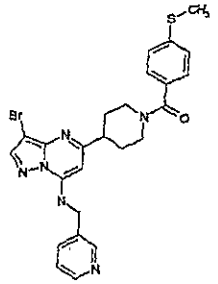
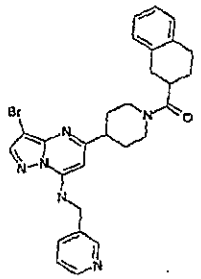
TABLE 43

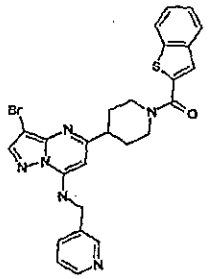
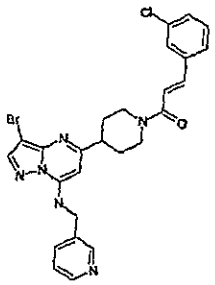
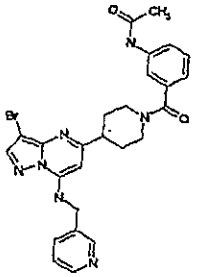
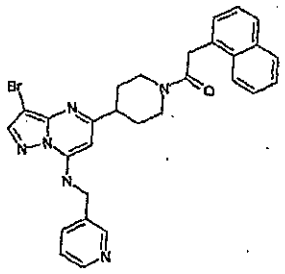
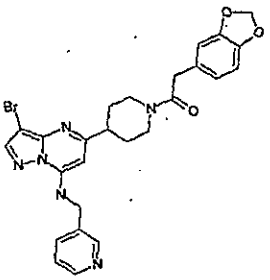
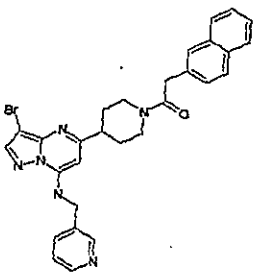
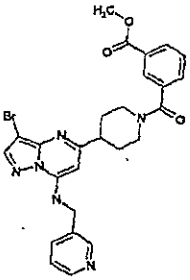
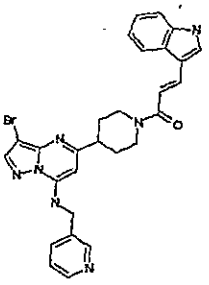
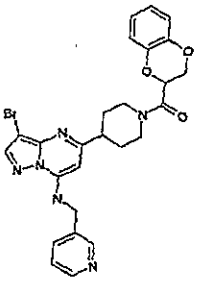
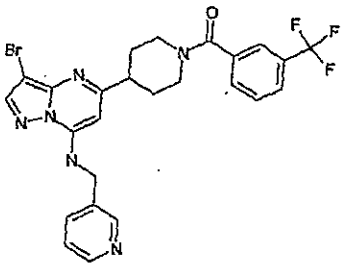
Ex.	Compound	m/z	Ex.	Compound	m/z
4301		457.25	4305		487.27
4302		471.26	4306		487.27
4303		477.26	4307		487.27
4304		483.27	4308		494.27

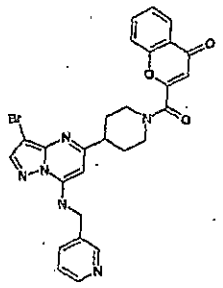
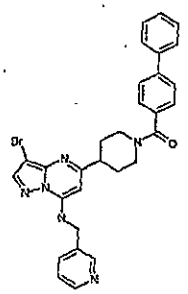
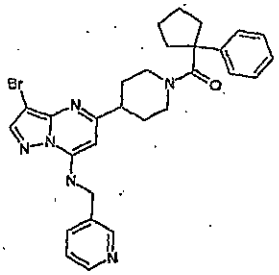
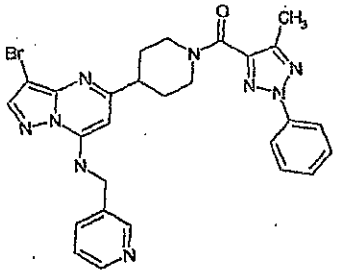
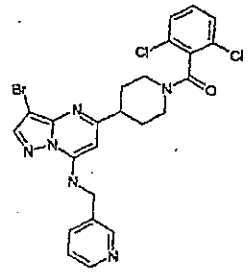
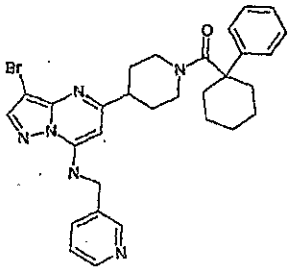
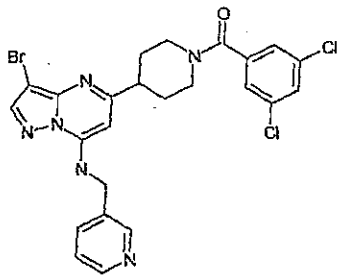
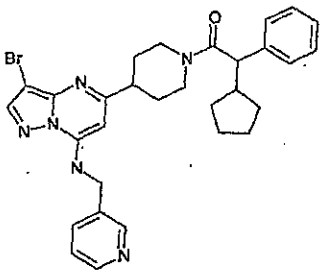
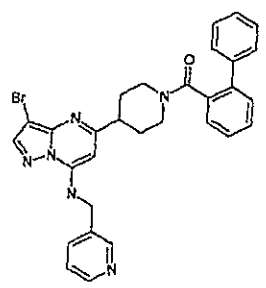
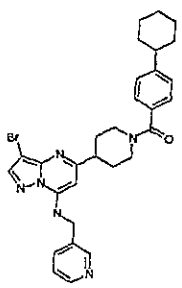
Ex.	Product	m/z	Ex.	Product	m/z
4309		499.27	4313		509.28
4310		500.27	4314		512.28
4311		503.28	4315		513.28
4312		505.28	4316		513.28

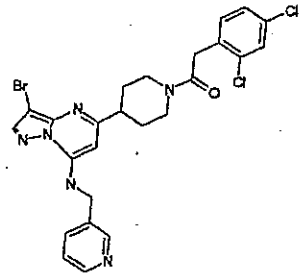
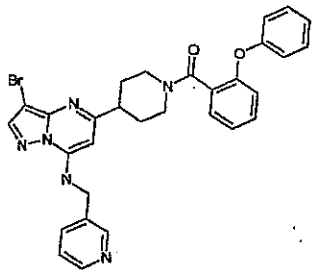
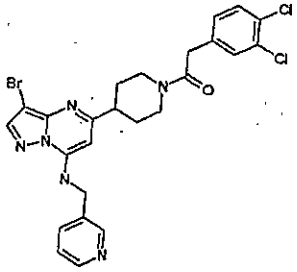
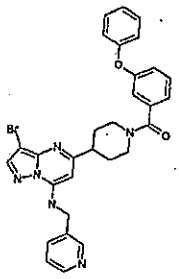
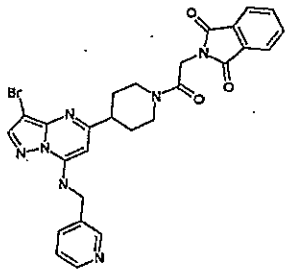
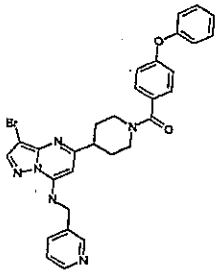
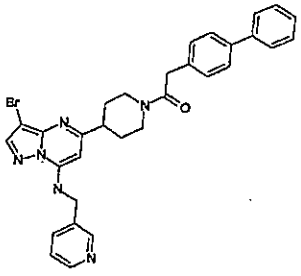
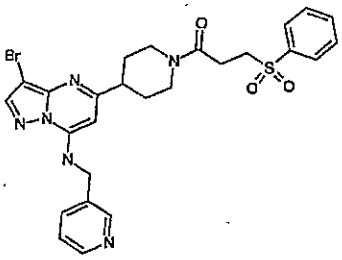
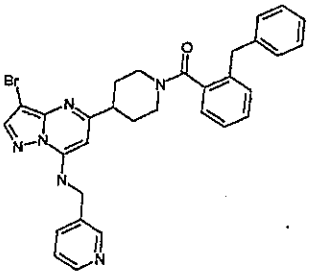
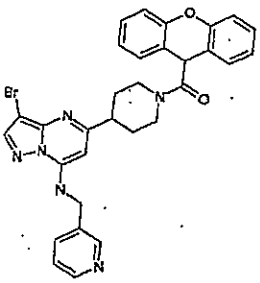
Ex.	Product	m/z
4317		518.28
4318		518.28
4319		519.29
4320		521.29
4321		523.29
4322		523.29
4323		523.29
4324		525.29

Product	1.Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4325 2. 527.29		1. 4330 2. 533.29
	1. 4326 2. 527.29		4331
	1. 4327 2. 532.29		1. 4332 2. 533.29
	1. 4328 2. 532.29		1. 4333 2. 533.29
	1. 4329 2. 532.29		1. 4334 2. 537.3

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4335 2.537.3		1. 4340 2. 541.3
	1.4336 2. 537.3		1. 4341 2. 543.3
	1. 4337 2. 539.3		1. 4342 2. 546.3
	1. 4338 2. 539.3		1. 4343 2. 547.3
	1. 4339 2. 539.3		1. 4344 2. 547.3

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4345 2. 549.3		1. 4350 2. 553.3
	1. 4346 2. 550.3		1. 4351 2. 557.31
	1. 4347 2. 551.3		1. 4352 2. 557.31
	1. 4348 2. 551.3		1. 4353 2. 558.31
	1. 4349 2. 551.3		1. 4354 2. 561.31

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4355 2. 561.31		1. 4360 2. 569.31
	1. 4356 2. 561.31		1. 4361 2. 574.32
	1. 4357 2. 561.31		1. 4362 2. 573.32
	1. 4358 2. 561.31		1. 4363 2. 573.32
	1. 4359 2. 569.31		1. 4364 2. 575.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4365 2. 575.32		1. 43700 2. 585.32
	1. 43666 2. 575.32		1. 4371 2. 583.32
	1. 4367 2. 574.32		1. 4372 2. 585.32
	1. 43688 2. 583.32		1. 4373 2. 585.32
	1. 4369 2. 583.32		1. 4374 2. 597.33

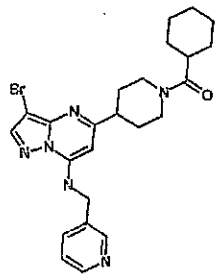
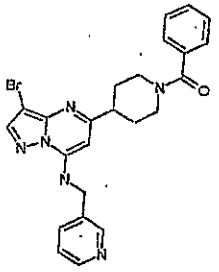
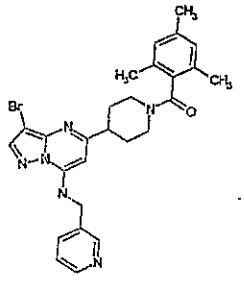
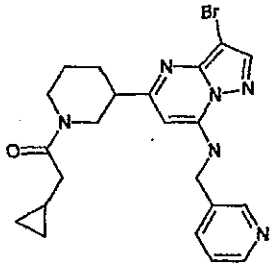
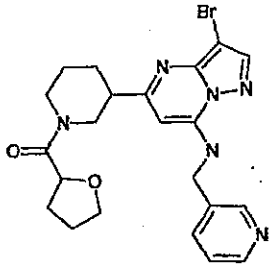
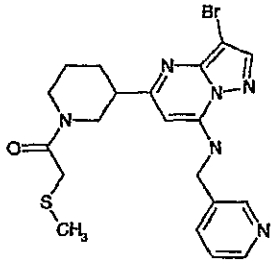
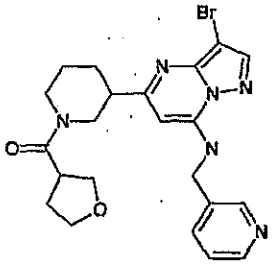
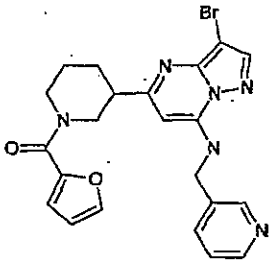
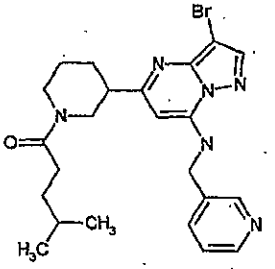
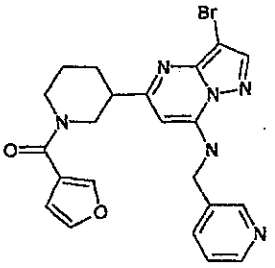
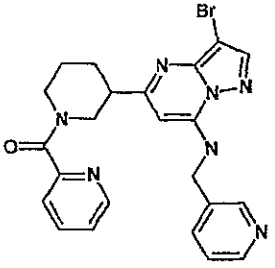
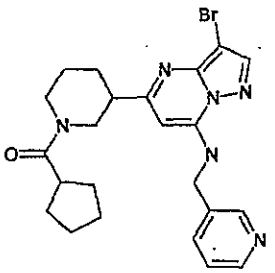
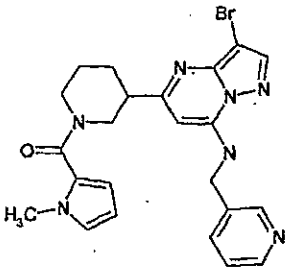
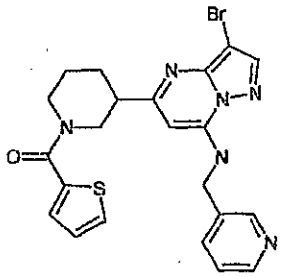
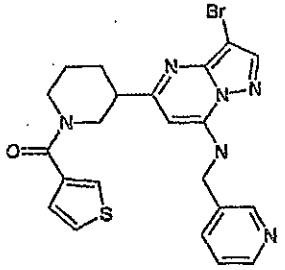
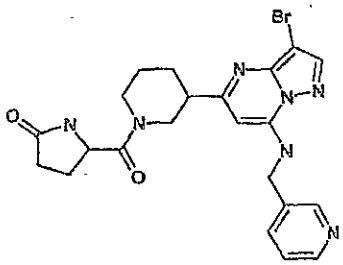
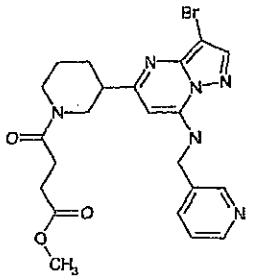
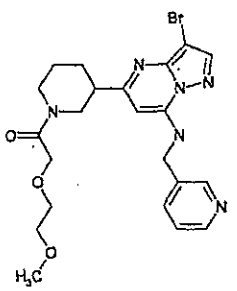
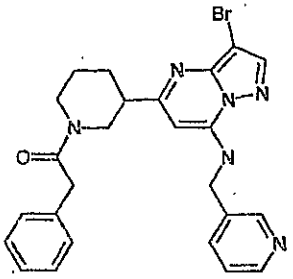
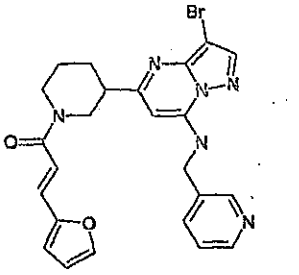
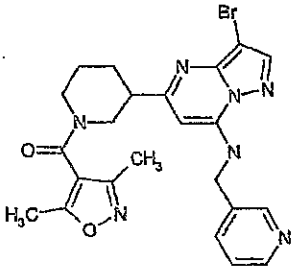
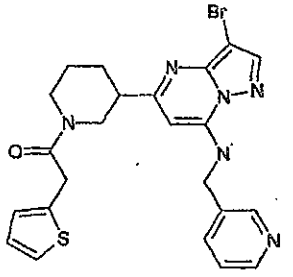
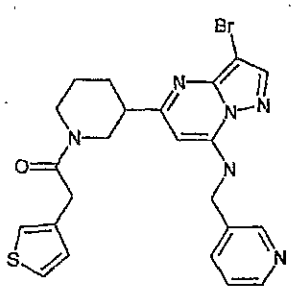
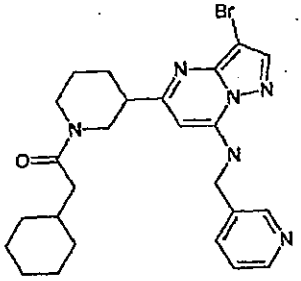
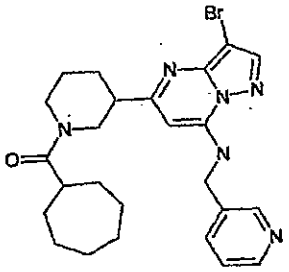
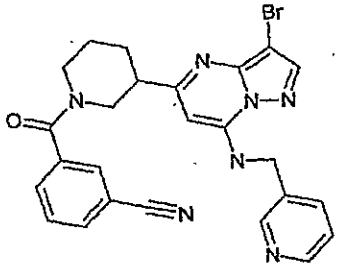
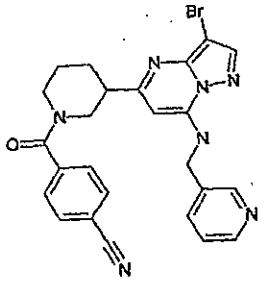
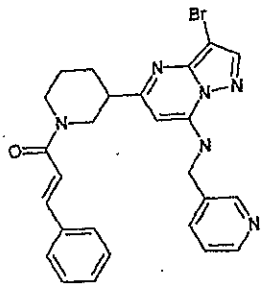
Product	1. Ex. 2. m/z
	1. 4375 2. 499.27
	1. 4376 2. 493.27
	1. 4377 2. 535.29

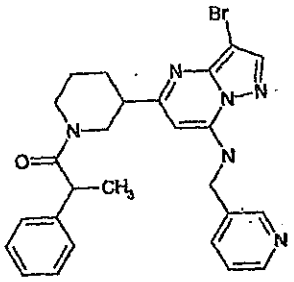
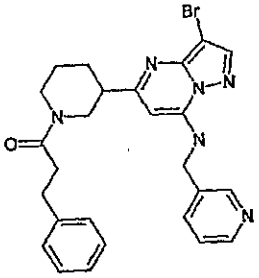
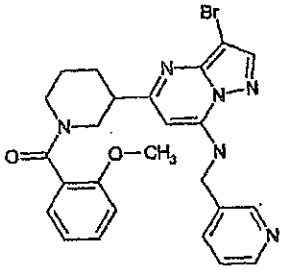
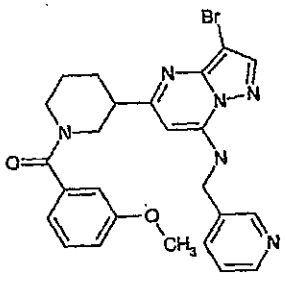
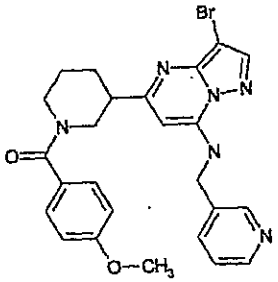
TABLE 44

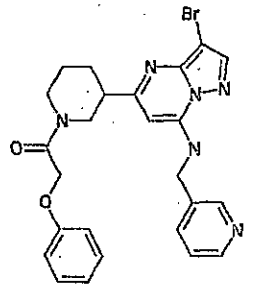
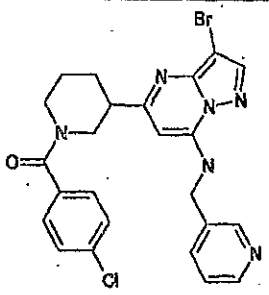
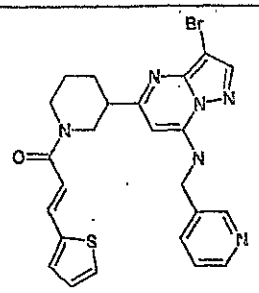
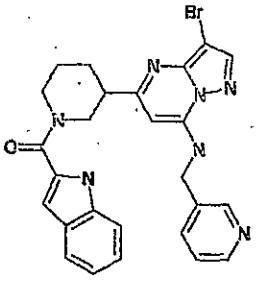
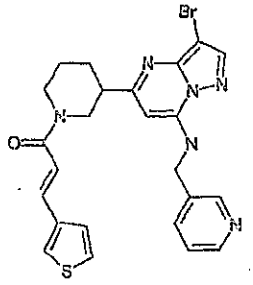
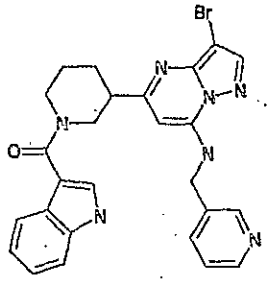
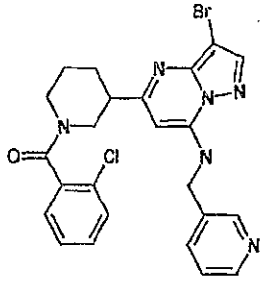
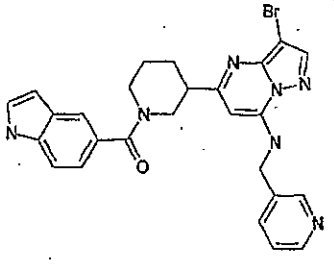
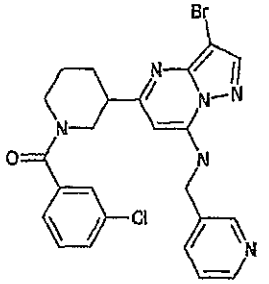
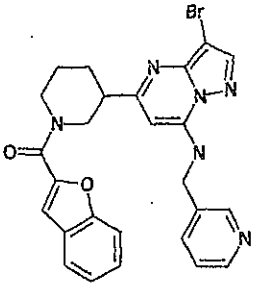
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	1. 4402 2. 475.26		1. 4407 2. 487.27
	1. 4403 2. 483.27		1. 4408 2. 487.27
	1. 4404 2. 481.26		1. 4409 2. 494.3
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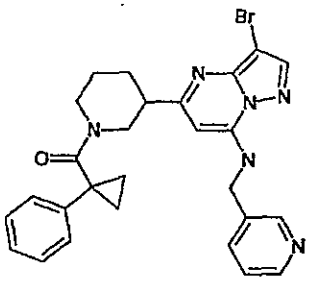
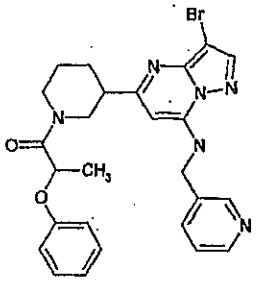
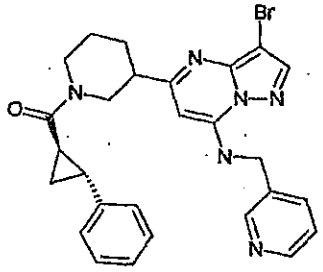
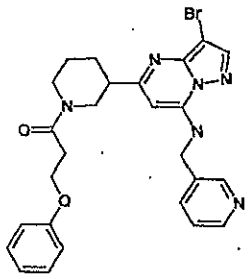
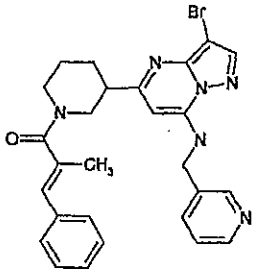
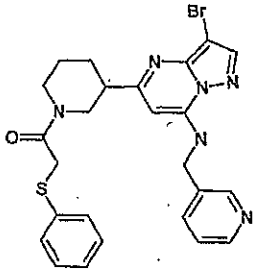
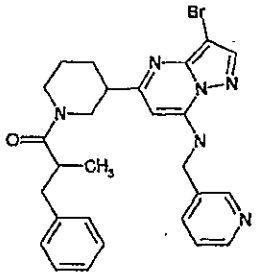
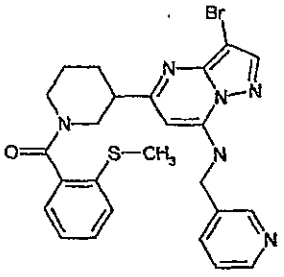
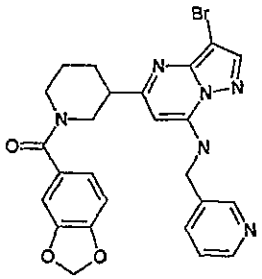
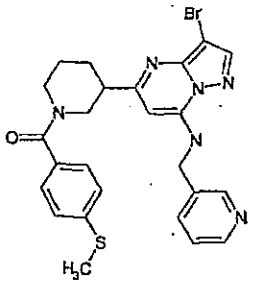
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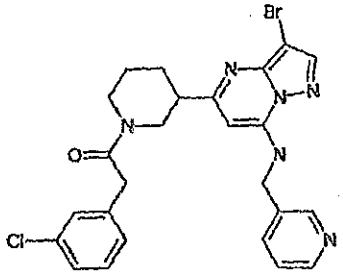
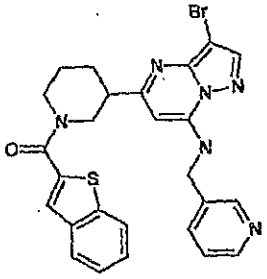
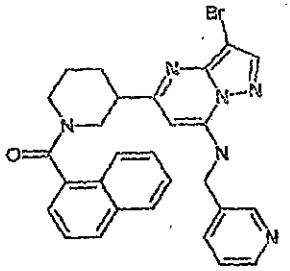
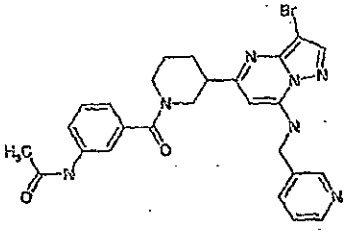
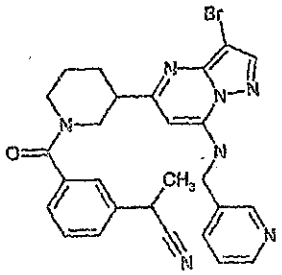
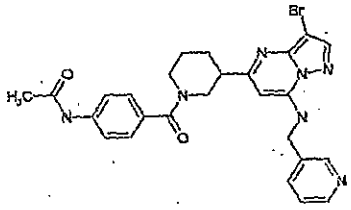
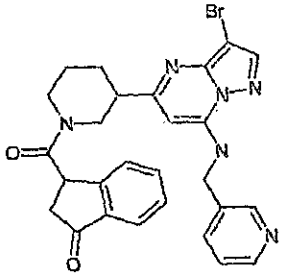
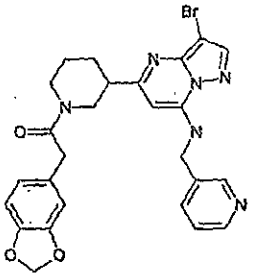
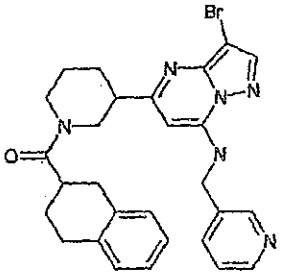
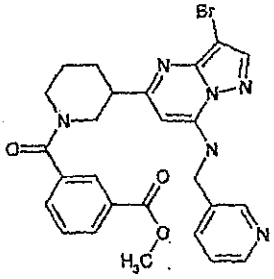
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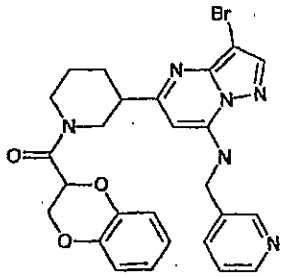
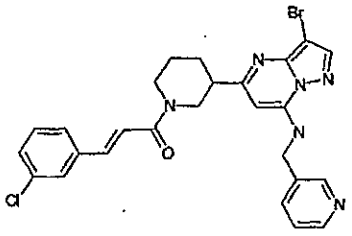
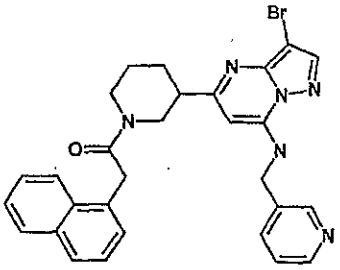
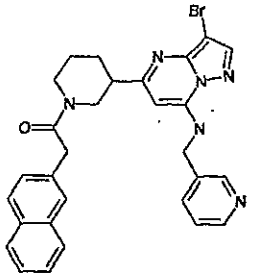
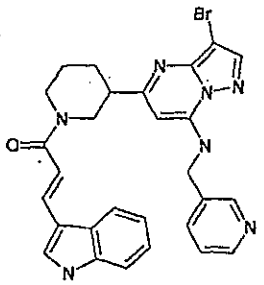
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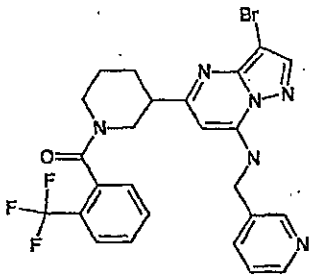
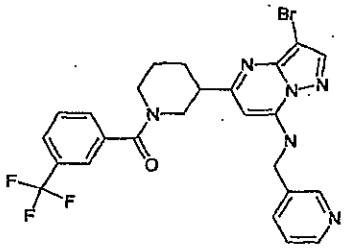
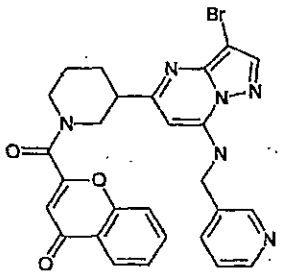
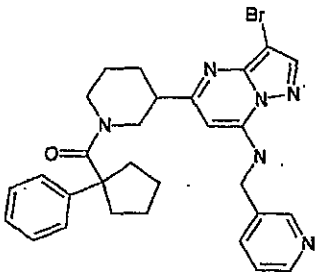
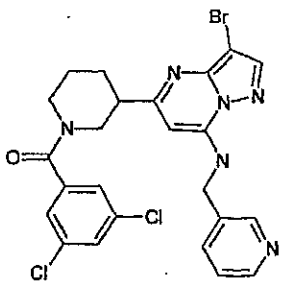
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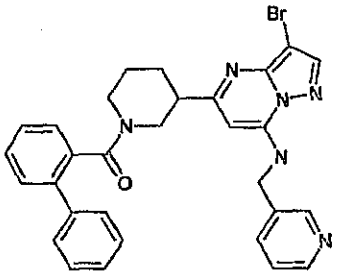
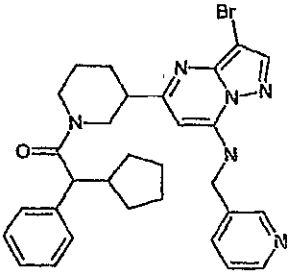
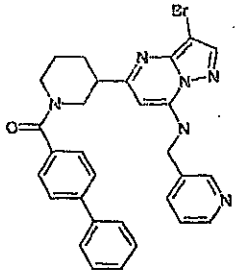
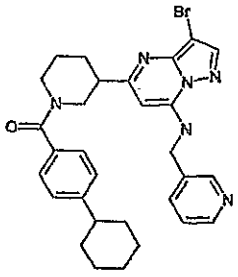
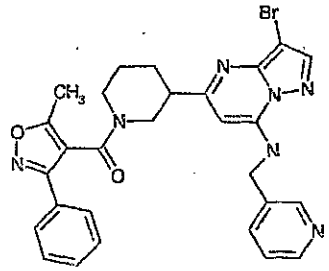
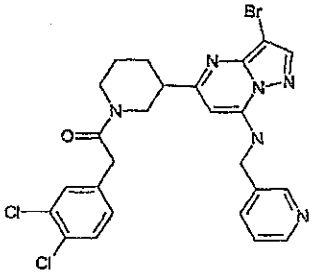
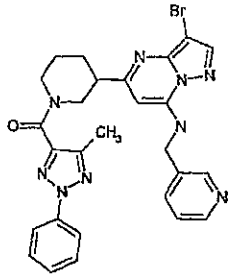
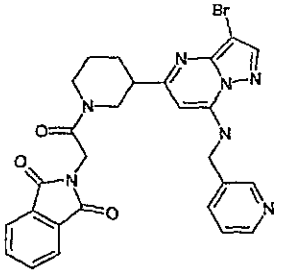
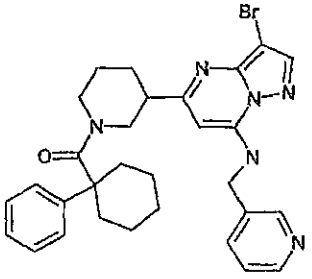
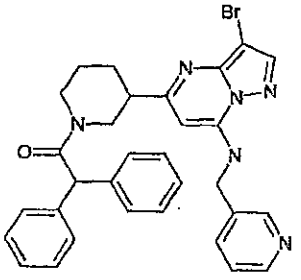
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	1. 4432 2. 525.29			1. 4437 2. 532.29
	1. 4433 2. 525.3			1. 4438 2. 532.29
	1. 4434 2. 527.29			1. 4439 2. 532.29
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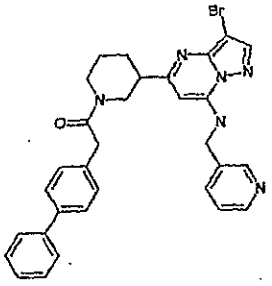
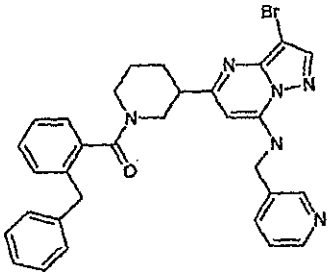
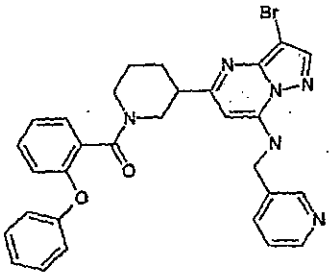
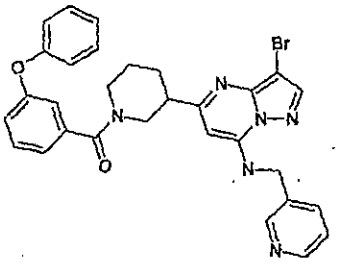
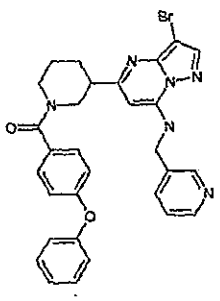
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	1. 4441 2. 533.3		1. 4446 2. 537.3
	1. 4442 2. 531.29		1. 4447 2. 537.3
	1. 4443 2. 533.29		1. 4448 2. 539.3
	1. 4444 2. 535.29		1. 4449 2.539.3
	1. 4445 2. 537.3		1. 4450 2.539.3

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4451 2. 541.3		1. 4456 2. 549.3
	1. 4451 2. 543.3		1. 4457 2. 550.3
	1. 4453 2. 546.3		1. 4458 2. 550.3
	1. 4454 2.547.3		1. 4459 2. 551.3
	1. 4455 2.547.3		1. 4460 2. 551.3

Product	1. Ex. 2. m/z
	1. 4461 2. 549.3
	1. 4462 2. 553.3
	1. 4463 2. 557.3
	1. 4464 2. 557.31
	1. 4465 2. 558.31

Product	1. Ex. 2. m/z
	1. 4466 2. 561.31
	1. 4467 2. 561.31
	1. 4468 2. 561.31
	1. 4469 2. 561.31
	1. 4470 2. 562.3

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4471 2. 569.31		1. 4476 2.572.31
	1. 4472 2. 569.31		1. 4477 2. 573.32
	1. 4473 2. 572.31		1. 4478 2. 574.32
	1. 4474 2. 572.31		1. 4479 2. 576.32
	1. 4475 2. 575.32		1. 4480 2. 583.32

Product	1. Ex. 2. m/z
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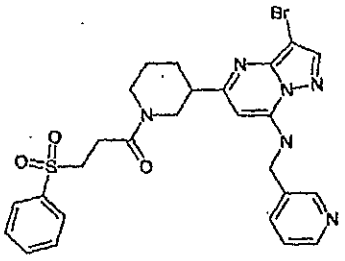
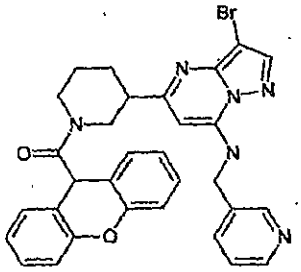
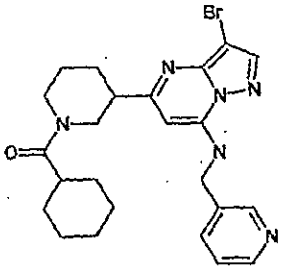
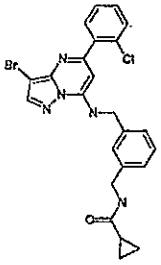
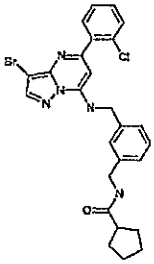
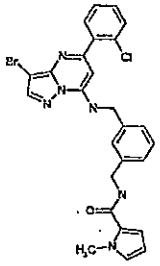
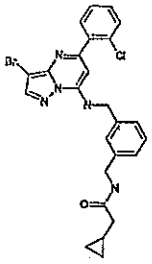
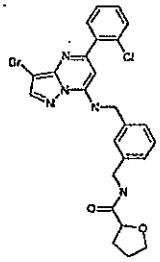
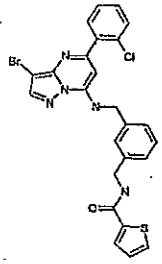
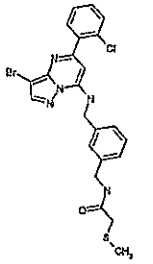
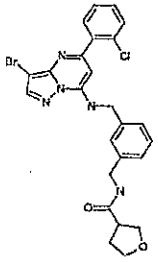
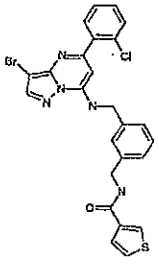
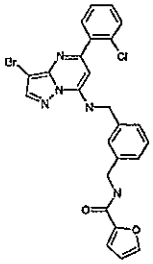
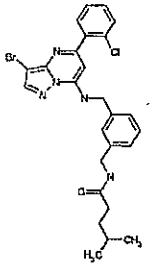
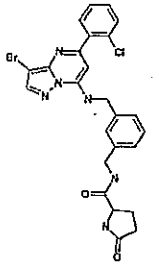
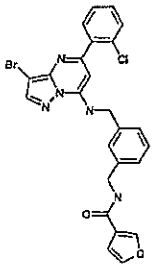
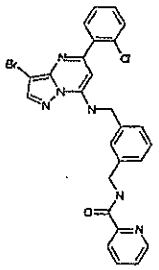
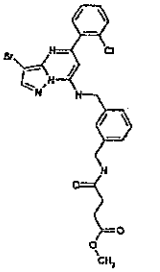
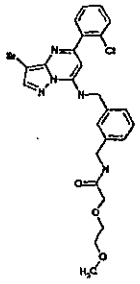
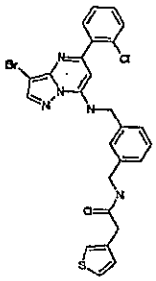
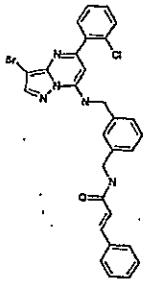
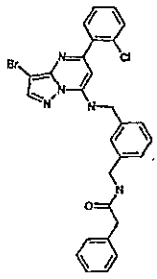
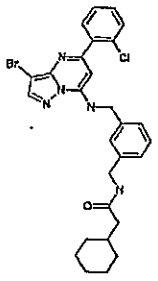
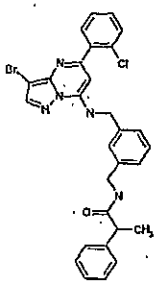
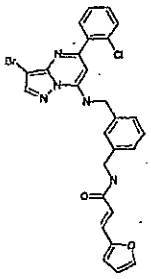
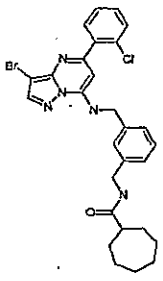
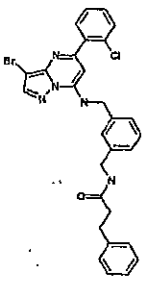
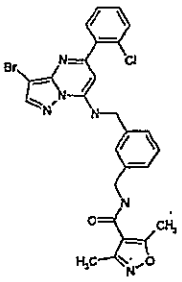
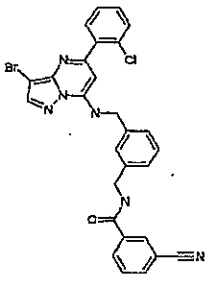
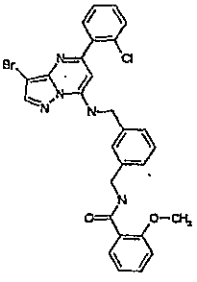
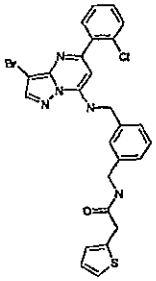
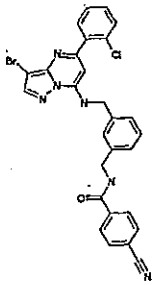
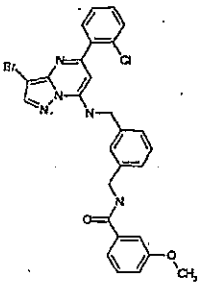
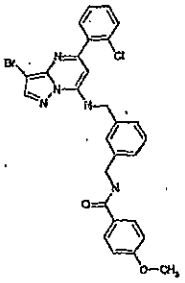
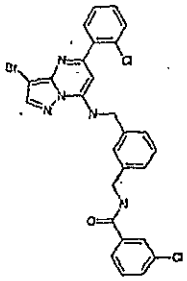
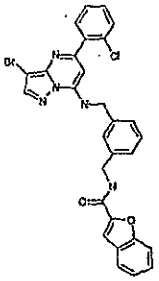
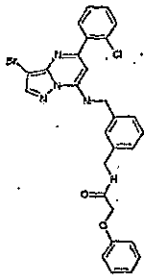
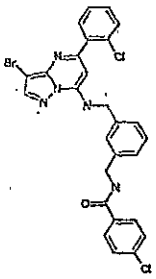
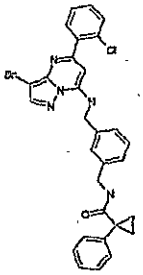
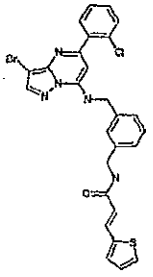
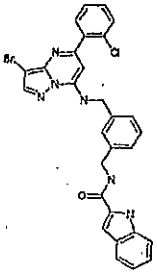
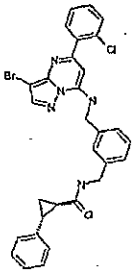
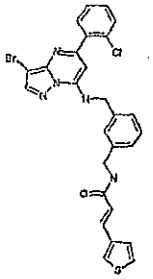
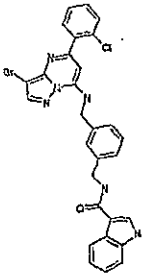
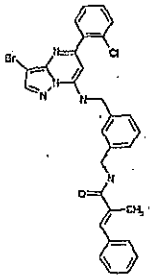
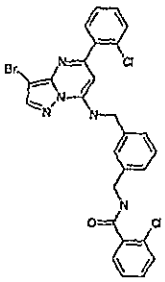
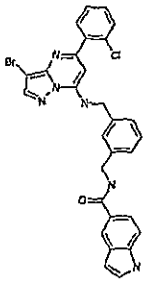
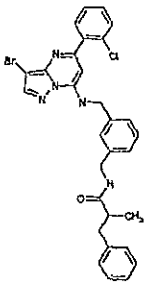
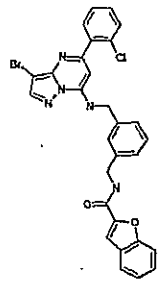
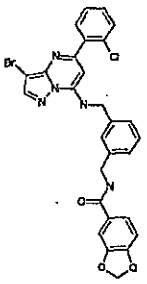
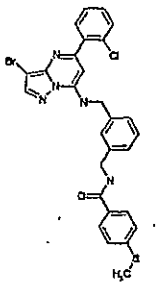
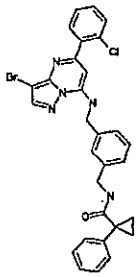
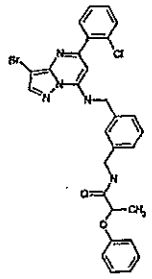
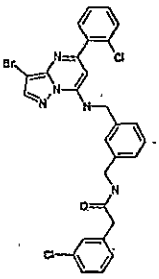
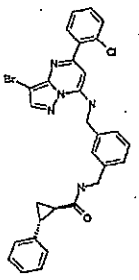
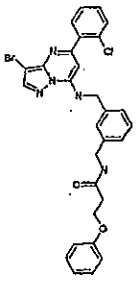
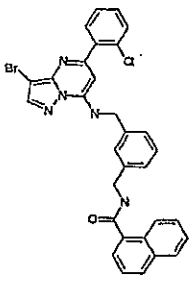
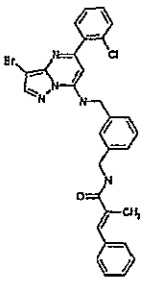
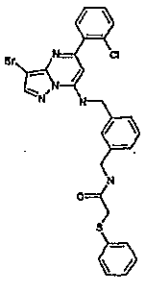
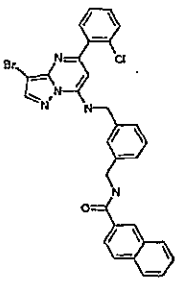
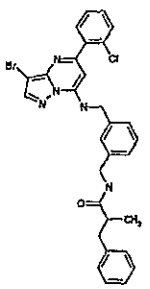
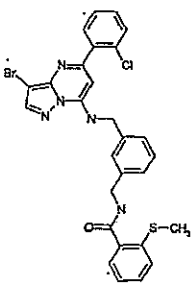
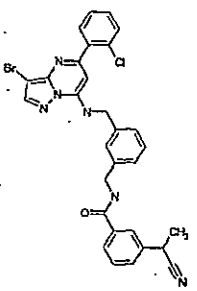
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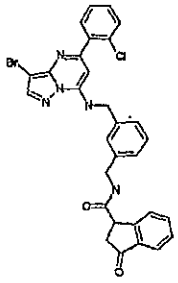
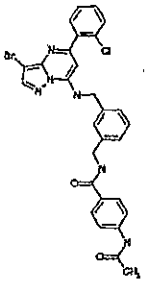
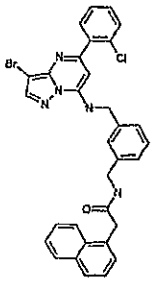
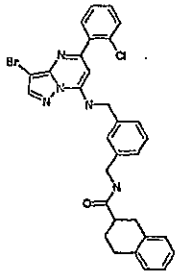
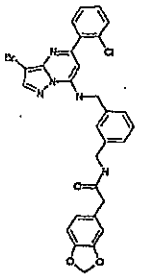
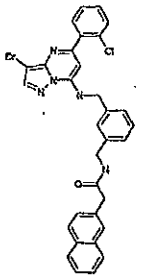
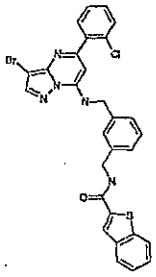
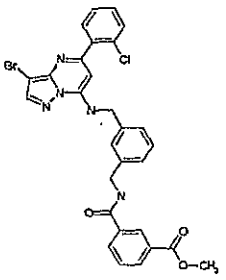
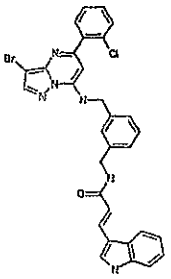
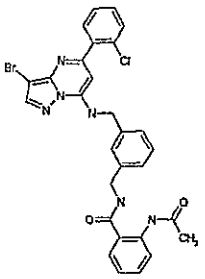
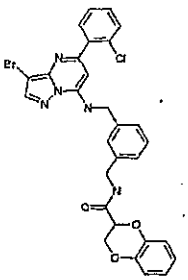
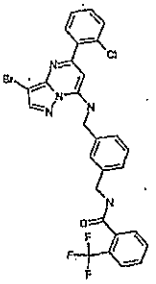
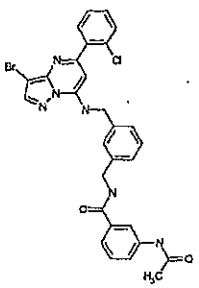
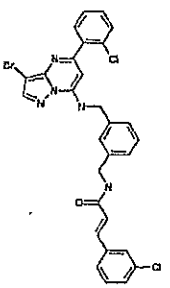
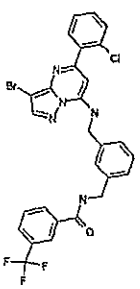
TABLE 45

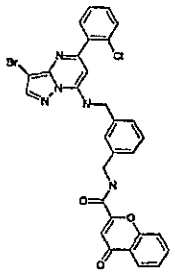
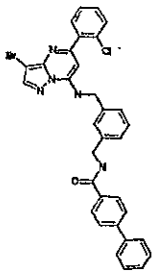
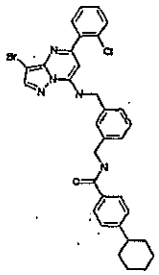
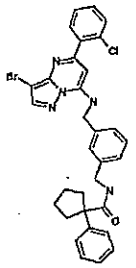
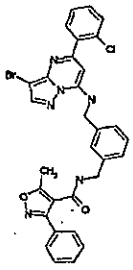
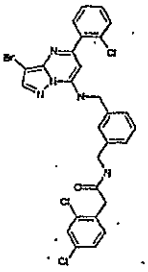
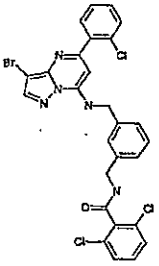
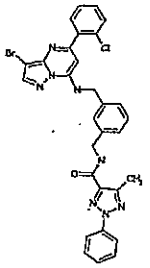
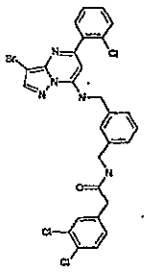
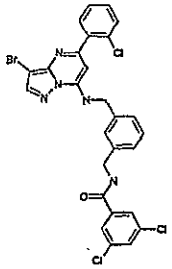
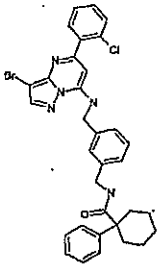
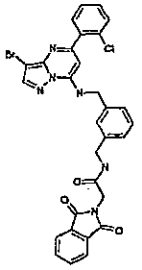
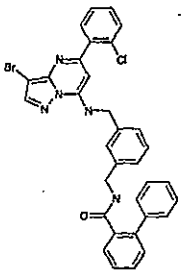
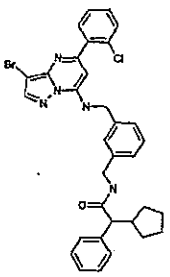
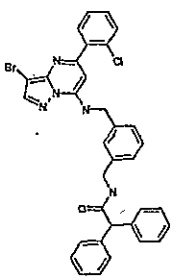
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	1. 4501 2. 512.28		1. 4506 2. 540.3		1. 4511 2. 551.3
	1. 4502 2. 526.29		1. 4507 2. 542.3		1. 4512 2. 554.3
	1. 4503 2. 532.29		1. 4508 2. 542.3		1. 4513 2. 554.3
	1. 4504 2. 538.3		1. 4509 2. 542.3		1. 4514. 2. 555.31
	1. 4505 2. 538.3		1. 4510 2. 549.3		1. 4515 2. 558.31

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4516 2. 560.31		1. 4521 2. 568.31		1. 4526 2. 574.32
	1. 4517 2. 562.31		1. 4522 2. 568.31		1. 4527 2. 576.32
	1. 4518 2. 564.31		1. 4523 2. 568.31		1. 4528 2. 576.32
	1. 4519 2. 567.31		1. 4524 2. 573.32		1. 4529 2. 578.32
	1. 4520 2. 568.31		1. 4525 2. 573.32		1. 4530 2. 578.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4531 2. 578.32		1. 4536 2. 582.32		1. 4541 2. 588.32
	1. 4532 2. 578.32		1. 4537 2. 582.32		1. 4542 2. 588.32
	1. 4533 2. 580.32		1. 4538 2. 587.32		1. 4543 2. 588.32
	1. 4534 2. 580.32		1. 4539 2. 587.32		1. 4544 2. 588.32
	1. 4535 2. 582.32		1. 4540 2. 587.32		1. 4545 2. 590.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4546 2. 588.32		1. 4551 2. 592.33		1. 4556 2. 593.33
	1. 4547 2. 588.32		1. 4552 2. 592.33		1. 4557 2. 596.33
	1. 4548 2. 588.32		1. 4553 2. 592.33		1. 4558 2. 596.33
	1. 4549 2. 588.32		1. 4554 2. 594.33		1. 4559 2. 598.33
	1. 4550 2. 590.32		1. 4555 2. 594.33		1. 4560 2. 601.33

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. E 2. m
	1. 4561 2. 602.33		1. 4665 2. 605.33		1. 457 612.3
	1. 4562 2. 602.33		1. 4567 2. 606.33		1. 4572 612.3
	1. 4563 2. 604.33		1. 4568 2. 606.33		1. 4573 613.34
	1. 4564 2. 603.3		1. 4569 2. 606.33		1. 4574 2 616.34
	1. 4565 2. 605.33		1. 4570 2. 608.33		1. 4575 2 616.34

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4576 2. 616.34		1. 4581 2. 624.34		1. 4586 2. 630.35
	1. 4577 2. 616.34		1. 4582 2. 629.35		1. 4587 2. 630.35
	1. 4578 2. 616.34		1. 4583 2. 629.35		1. 4588 2. 630.35
	1. 4579 2. 616.34		1. 4584 2. 630.35		1. 4589 2. 631.35
	1. 4580 2. 624.34		1. 4585 2. 630.35		1. 4590 2. 638.35

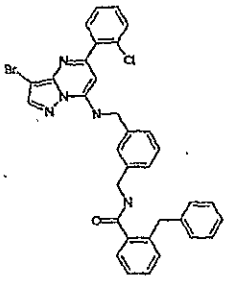
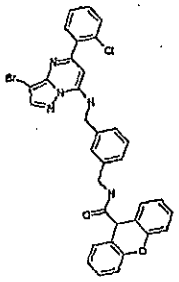
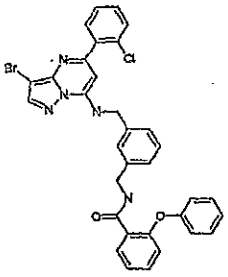
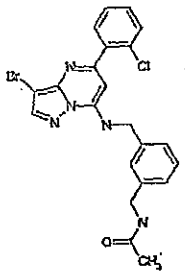
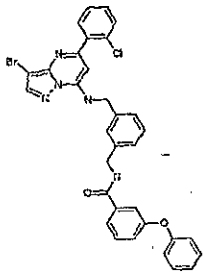
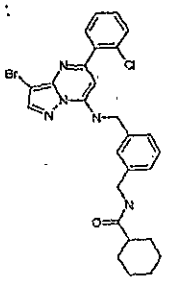
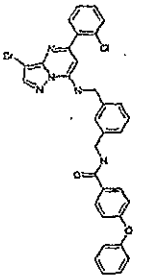
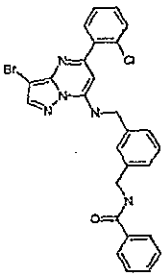
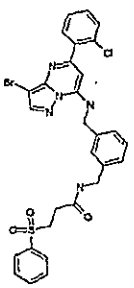
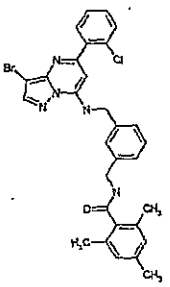
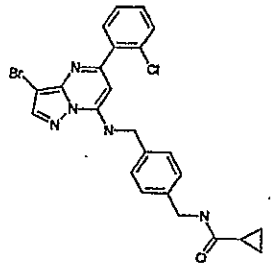
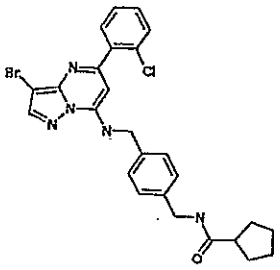
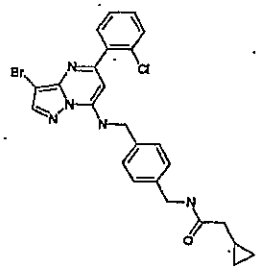
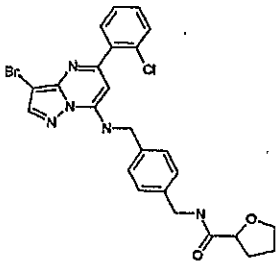
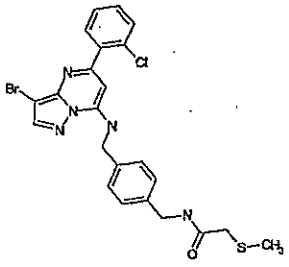
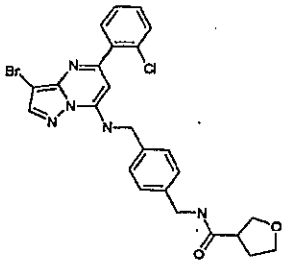
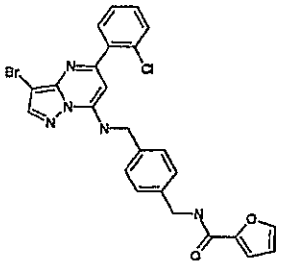
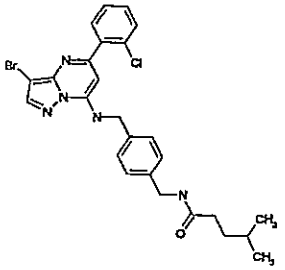
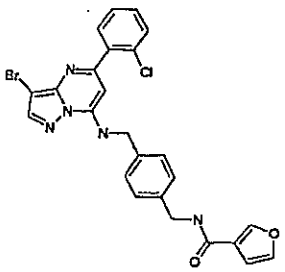
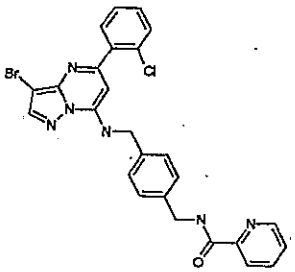
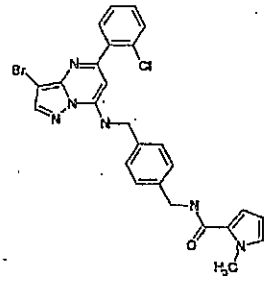
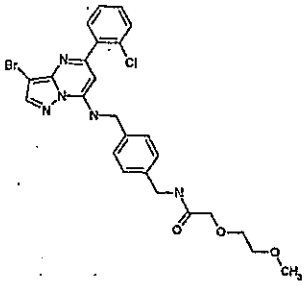
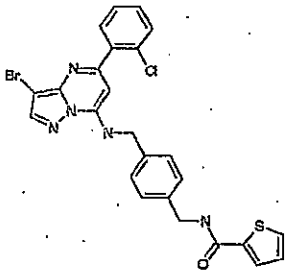
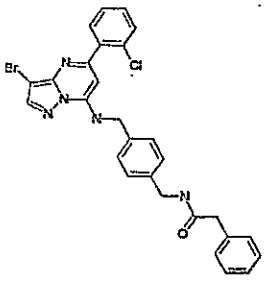
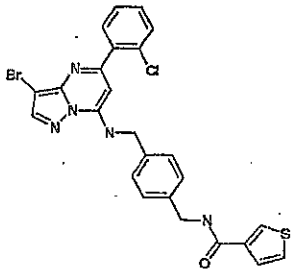
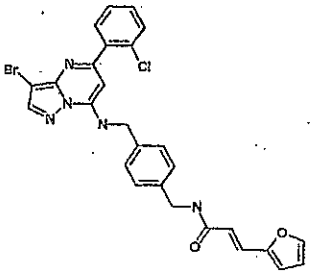
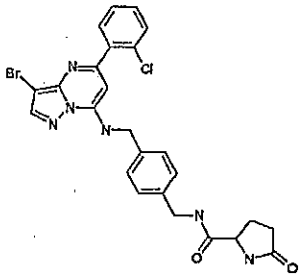
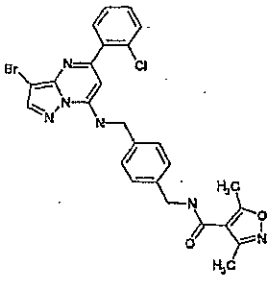
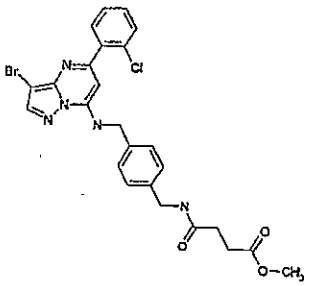
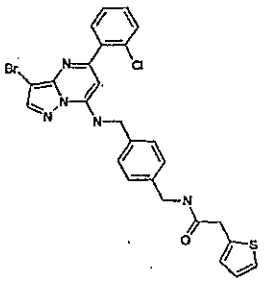
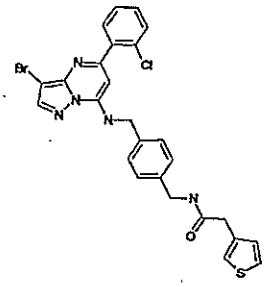
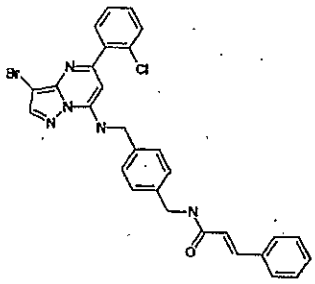
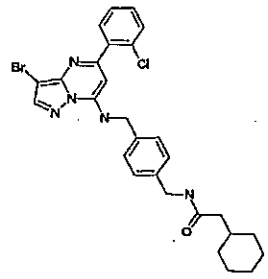
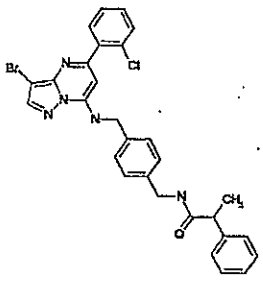
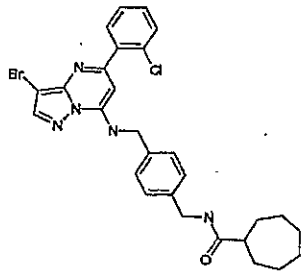
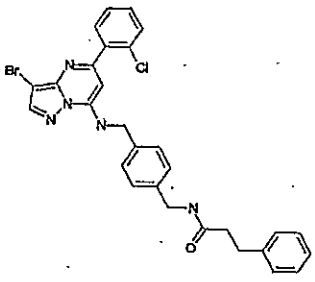
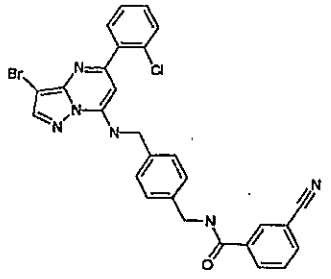
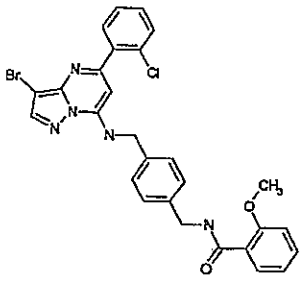
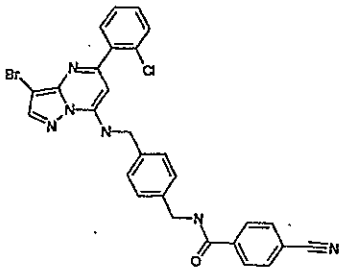
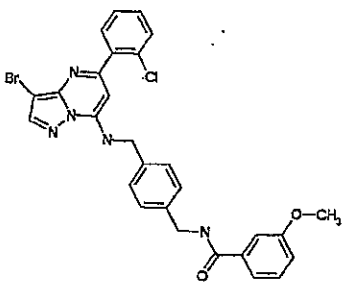
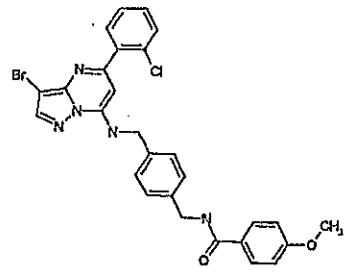
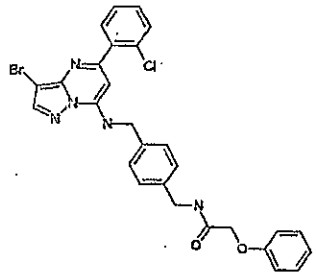
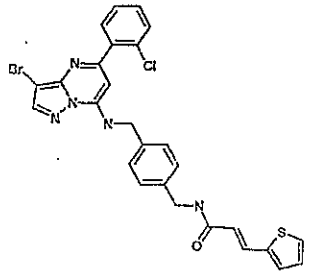
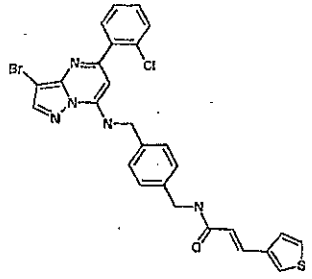
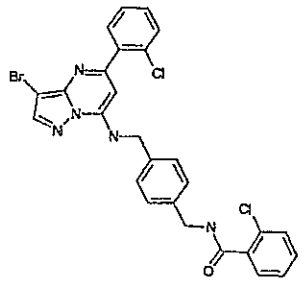
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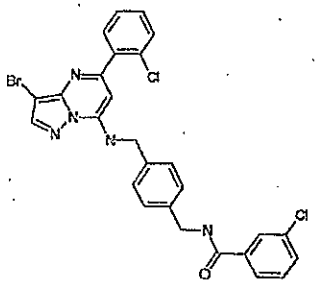
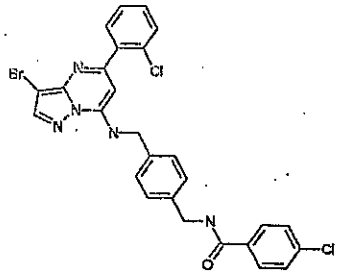
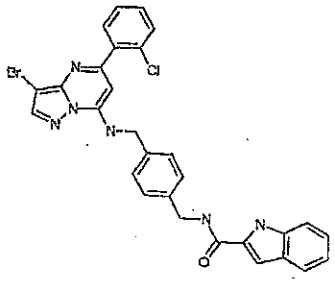
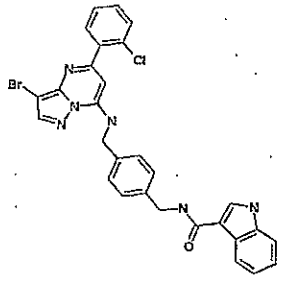
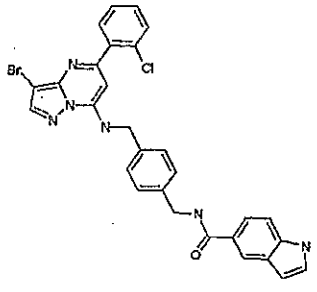
TABLE 46

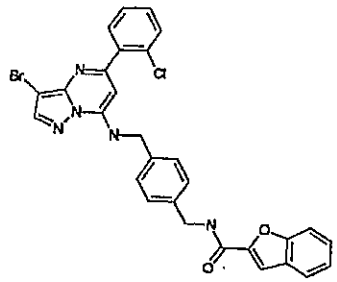
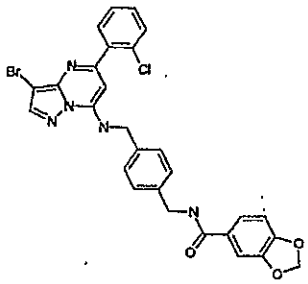
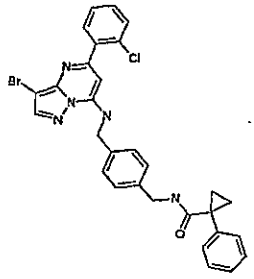
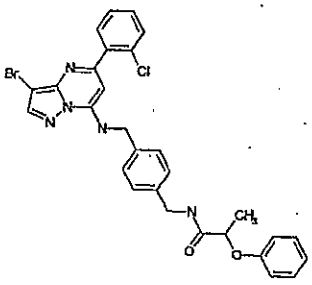
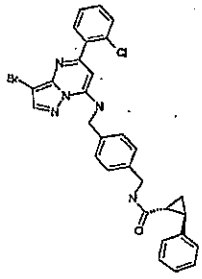
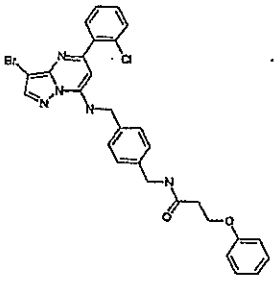
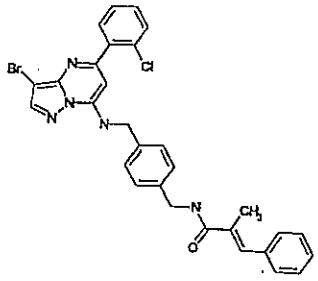
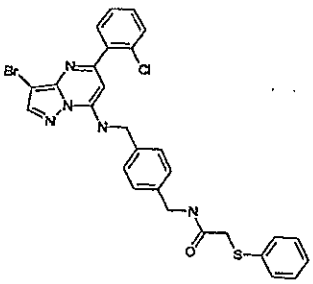
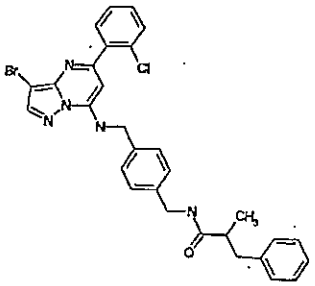
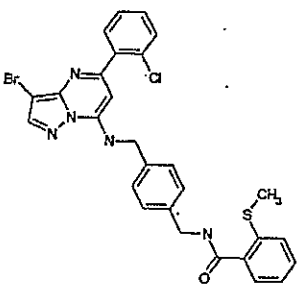
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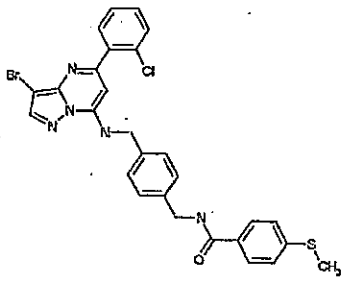
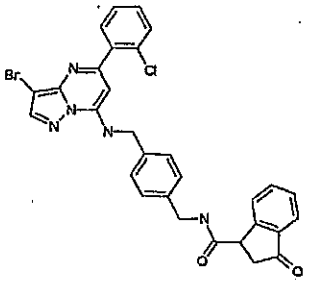
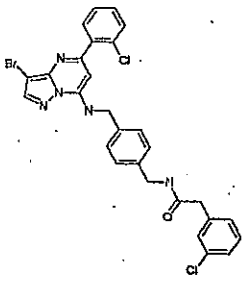
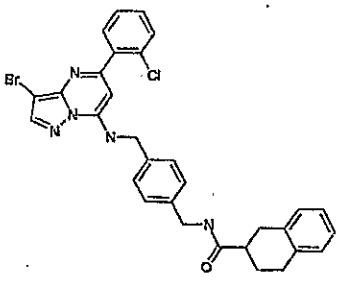
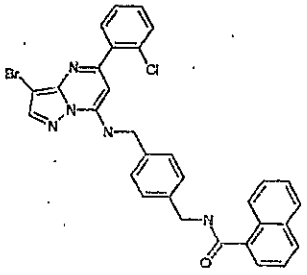
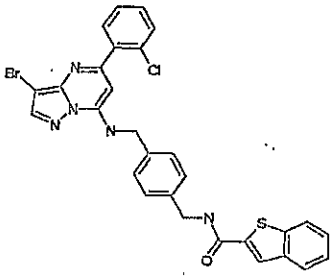
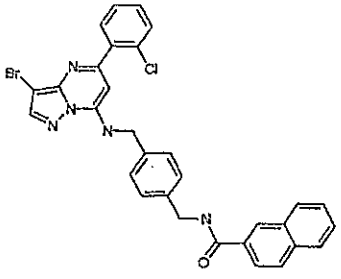
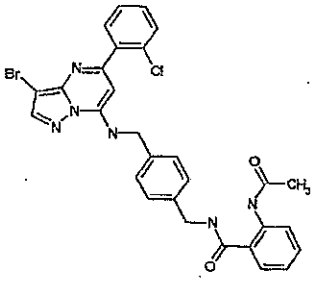
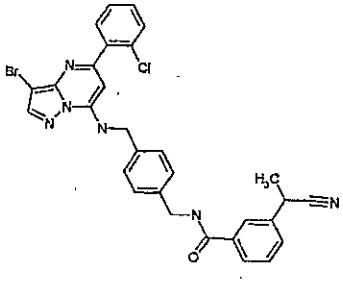
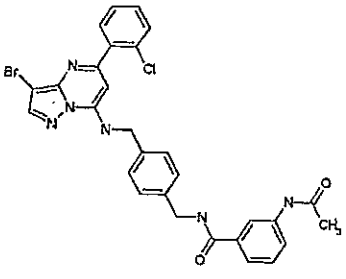
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	1. 4611 2. 551.3		1. 4616 2. 560.31
	1. 4612 2. 552.3		1. 4617 2. 562.31
	1. 4613 2. 552.3		1. 4618 2. 562.3
	1. 4614 2. 555.31		1. 4619 2. 567.31
	1. 4615 2. 558.31		1. 4620 2. 568.31

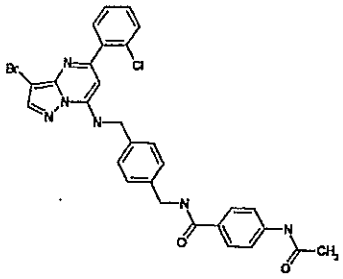
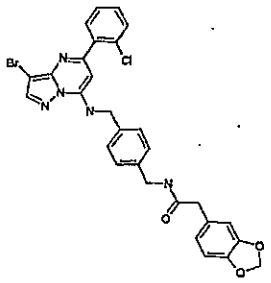
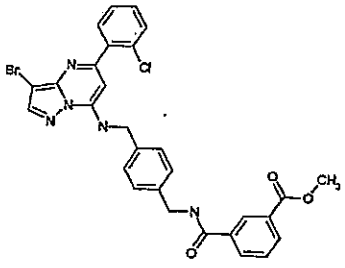
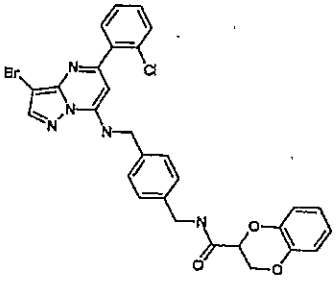
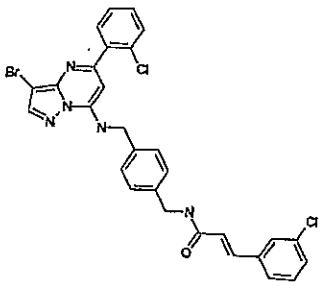
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	1. 4621 2. 568.31		1. 4626 2. 574.32
	1. 4622 2. 568.31		1. 4627 2. 576.32
	1. 4623 2. 568.31		1. 4628 2. 576.32
	1. 4624 2. 573.32		1. 4629 2. 578.32
	1. 4625 2. 573.32		1. 4630 2. 578.32

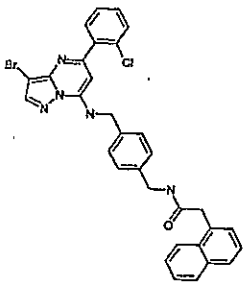
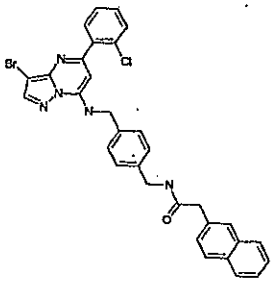
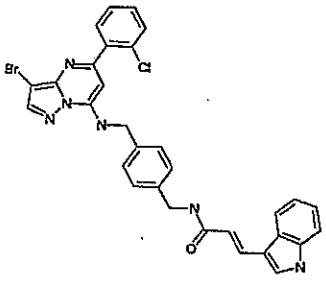
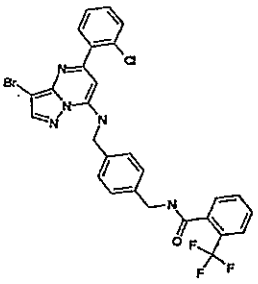
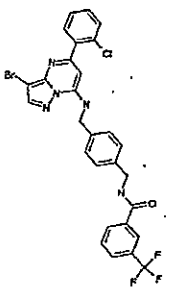
Product	1. Ex. 2. m/x
	1. 4631 2. 578.32
	1. 4632 2. 578.32
	1. 4633 2. 580.32
	1. 4634 2. 580.32
	1. 4635 2. 582.32

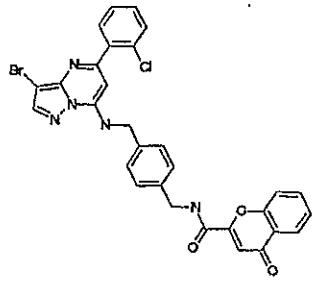
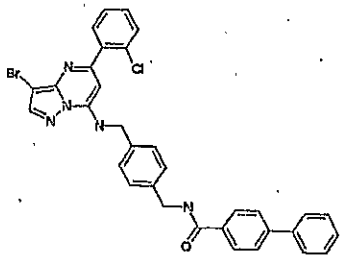
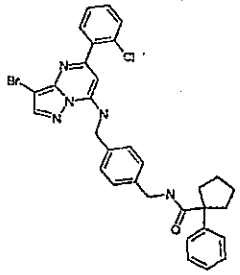
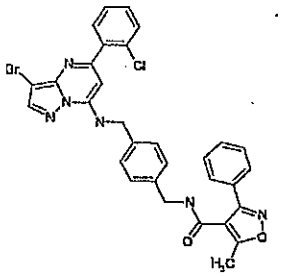
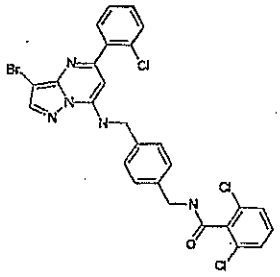
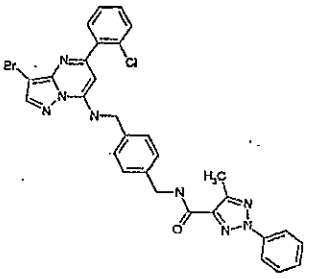
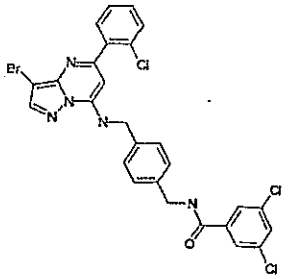
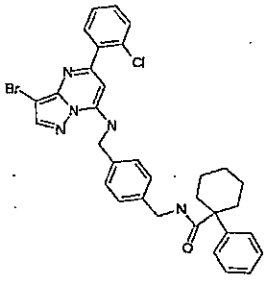
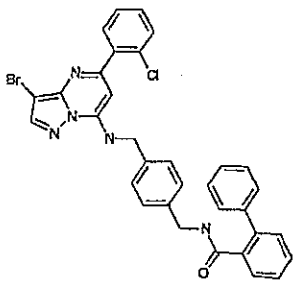
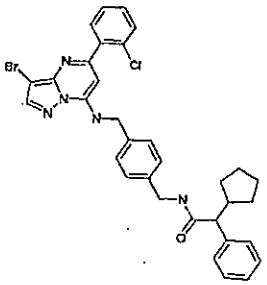
Product	1. Ex. 2. m/z
	1. 4636 2. 582.32
	1. 4637 2. 582.32
	1. 4638 2. 587.32
	1. 4639 2. 585.3
	1. 4640 2. 585.3

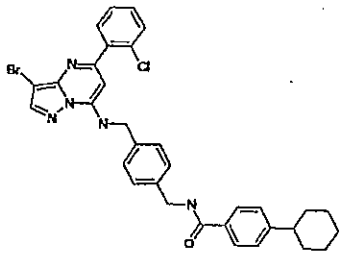
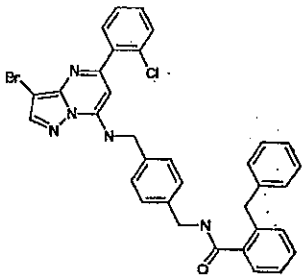
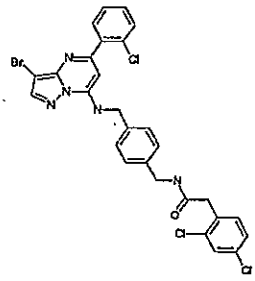
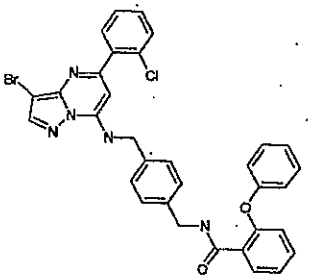
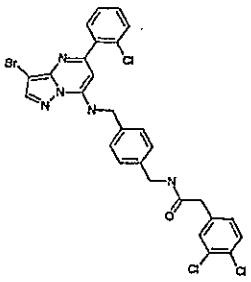
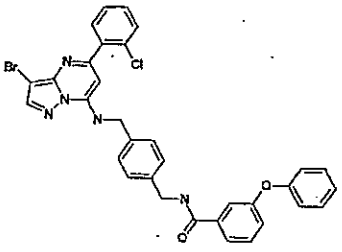
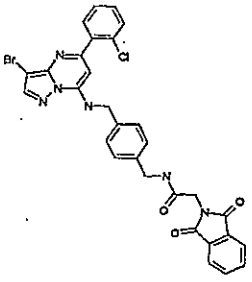
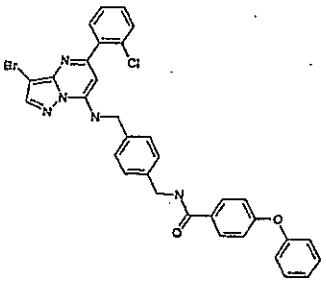
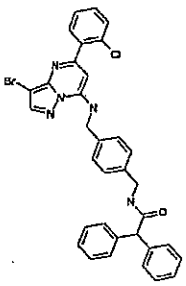
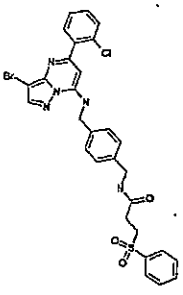
Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4641 2. 588.32		1. 4646 2. 592.33
	1. 4642 2. 588.32		1. 4647 2. 592.33
	1. 4643 2. 588.32		1. 4648 2. 592.33
	1. 4644 2,. 588.32		1. 4649 2.
	1. 4645 2. 590.32		1. 4650 2. 594.33

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4651 2. 594.33		1. 4656 2. 602.33
	1. 4652 2. 596.33		1. 4657 2. 602.33
	1. 4653 2. 596.3		1. 4658 2. 604.33
	1. 4654 2. 598.33		1. 4659 2. 603.3
	1. 4655 2. 601.33		1. 4660 2. 605.33

Product	1. Ex. 2. m/z
	1. 4661 2. 605.33
	1. 4662 2. 606.33
	1. 4663 2. 606.33
	1. 4664 2. 606.33
	1. 4665 2. 608.33

Product	1. Ex. 2. m/z
	1. 4666 2. 612.34
	1. 4667 2. 612.34
	1. 4668 2. 613.34
	1. 4669 2. 616.34
	1. 4670 2. 616.34

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4671 2. 616.34		1. 4676 2. 624.34
	1. 4672 2. 616.34		1. 4677 2. 629.35
	1. 4673 2. 616.34		1. 4678 2. 629.35
	1. 4674 2. 616.34		1. 4679 2. 630.35
	1. 4675 2. 624.34		1. 4680 2. 630.35

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4681 2. 630.35		1. 4686 2. 638.35
	1. 4682 2. 630.35		1. 4687 2. 640.35
	1. 4683 2. 630.35		1. 4688 2. 640.35
	1. 4684 2. 631.35		4689 2.
	1. 4685 2. 638.35		1. 4690 2. 640.35

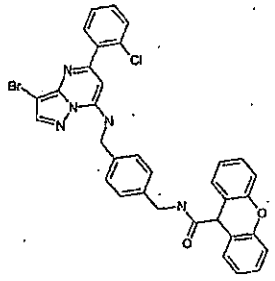
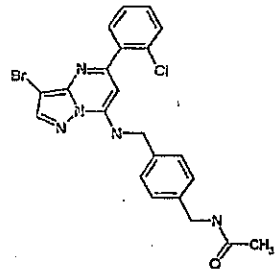
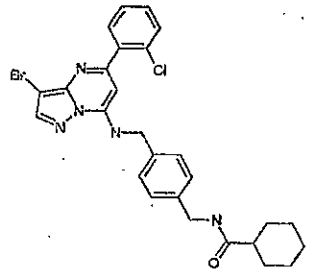
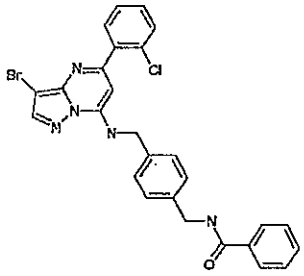
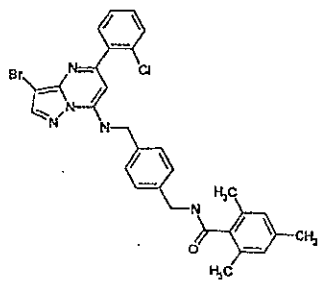
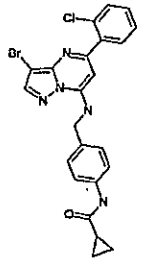
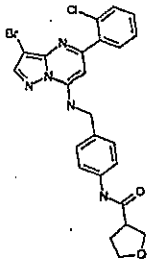
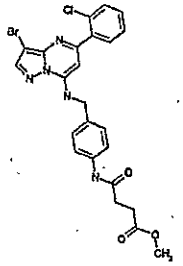
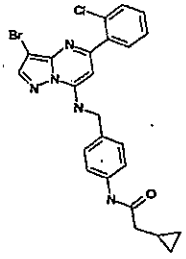
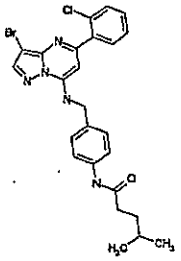
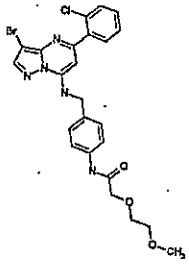
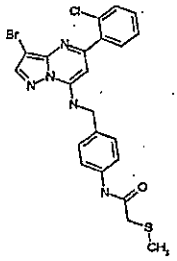
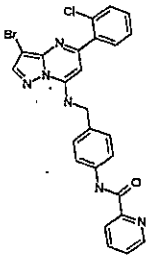
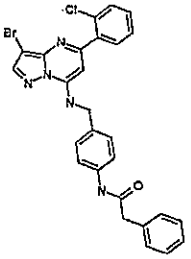
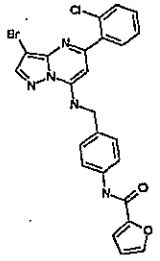
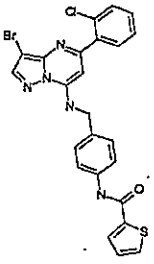
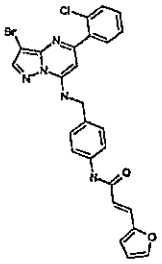
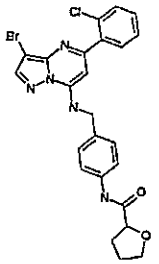
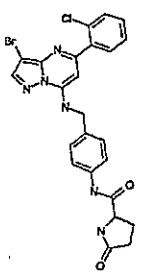
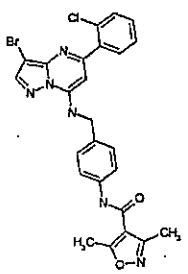
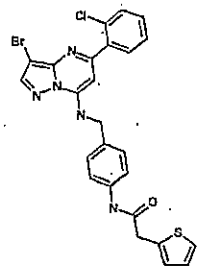
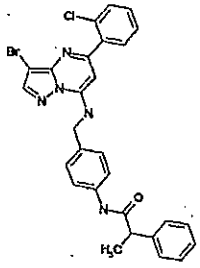
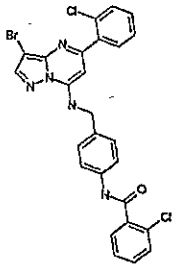
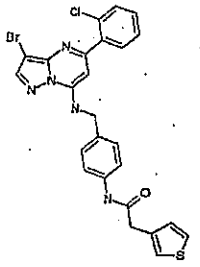
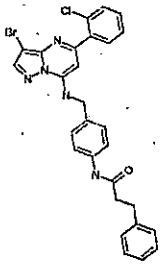
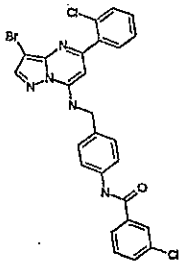
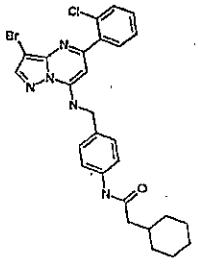
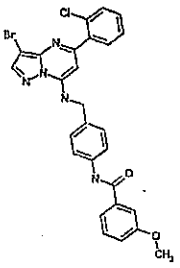
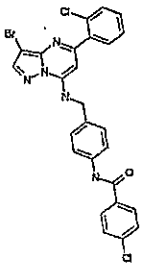
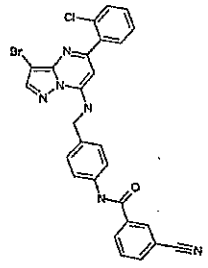
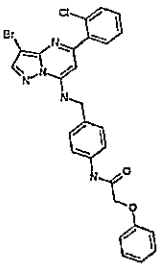
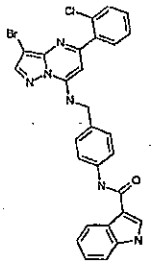
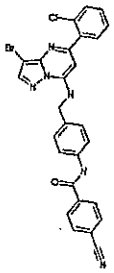
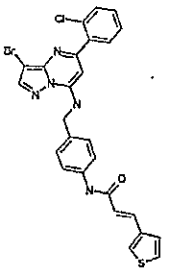
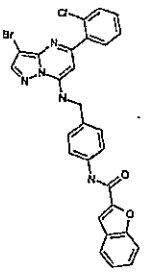
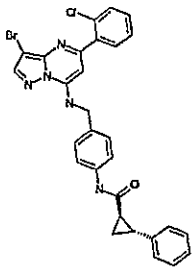
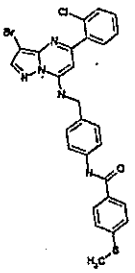
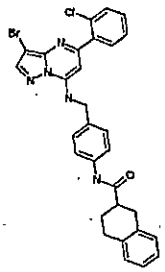
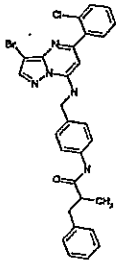
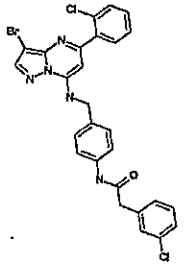
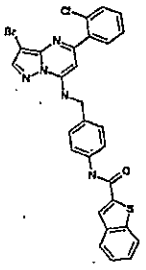
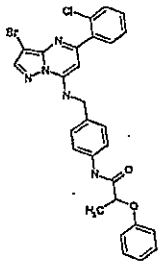
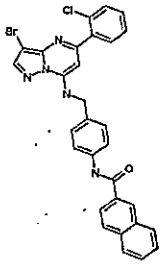
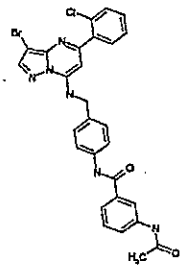
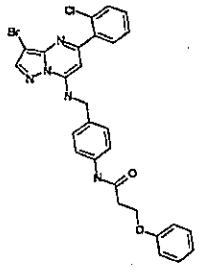
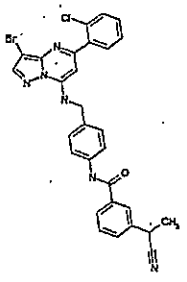
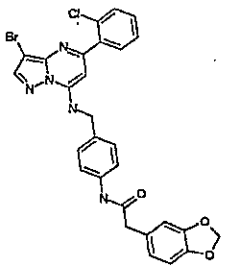
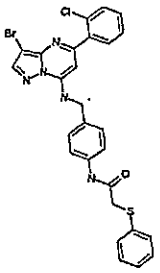
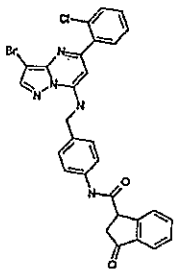
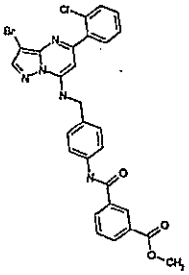
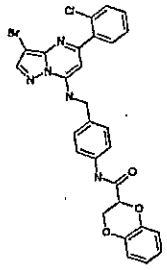
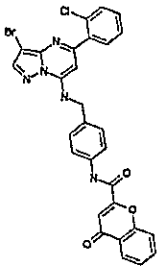
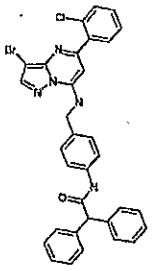
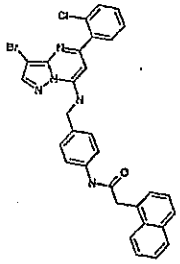
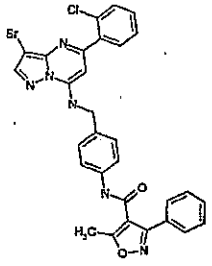
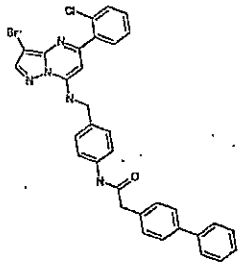
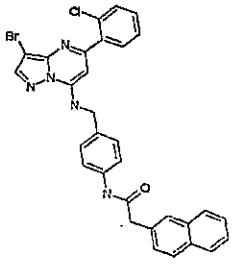
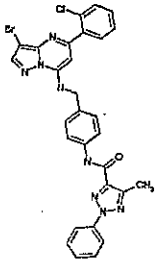
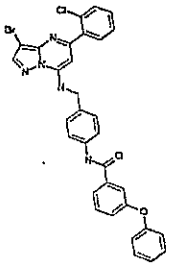
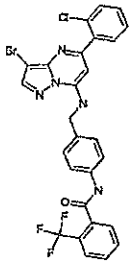
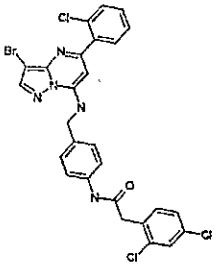
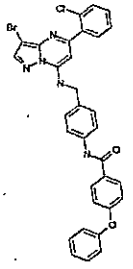
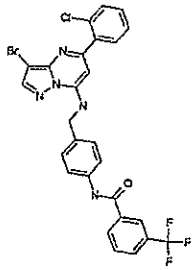
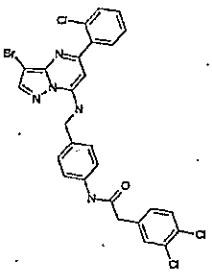
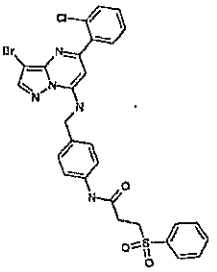
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	1. 4691 2. 652.36
	1. 4692 2. 484.3
	1. 4693 2. 554.3
	1. 4694 2. 548.3
	1. 4695 2. 590.32

TABLE 47

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4701 2. 498.27		1. 4706 2. 528.29		1. 4711 2. 544.3
	1. 4702 2. 512.28		1. 4707 2. 528.29		1. 4712 2. 546.3
	1. 4703 2. 518.28		1. 4708 2. 535.29		1. 4713 2. 548.3
	1. 4704 2. 524.29		1. 4709 2. 540.3		1. 4714 2. 550.3
	1. 4705 2. 528.29		1. 4710 2. 541.3		1. 4715 2. 553.3

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4716 2. 554.3		1. 4721 2. 562.31		1. 4726 2. 568.31
	1. 4717 2. 554.3		1. 4722 2. 562.31		1. 4727 2. 568.31
	1. 4718 2. 554.3		1. 4723. 2 564.31		1. 4728 2. 568.31
	1. 4719 2. 559.31		1. 4724 2. 564.31		1. 4729 2. 573.32
	1. 4720 2. 559.31		1. 4725 2. 566.31		1. 4730 2. 574.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4731 2. 574.32		1. 4736 2. 580.32		1. 4741 2. 588.32
	1. 4732 2. 576.32		1. 4737 2. 582.32		1. 4742 2. 590.32
	1. 4733 2. 578.32		1. 4738 2. 584.32		1. 4743 2. 591.33
	1. 4734 2. 578.32		1. 4739 2. 585.32		1. 4744 2. 592.33
	1. 4735 2. 580.32		1. 4740 2. 588.32		1. 4745 2. 592.33

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4746 2. 592.33		1. 4751 2. 602.33		1. 4756 2. 624.34
	1. 4747 2. 598.33		1. 4752 2. 615.34		1. 4757 2. 624.34
	1. 4748 2. 598.33		1. 4753 2. 615.34		1. 4758 2. 626.34
	1. 4749 2. 602.33		1. 4754 2. 616.34		1. 4759 2. 624.34
	1. 4750 2. 602.33		1. 4755 2. 616.34		1. 4760 2. 626.34

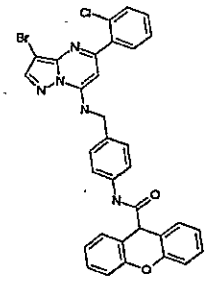
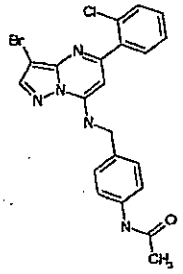
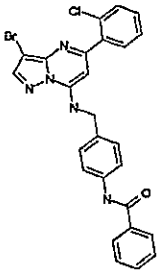
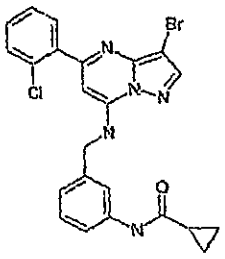
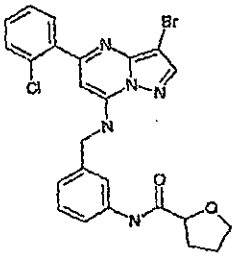
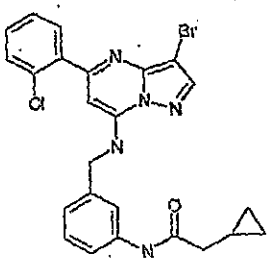
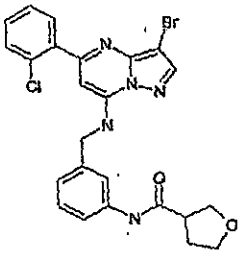
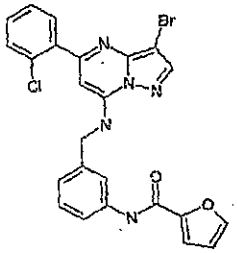
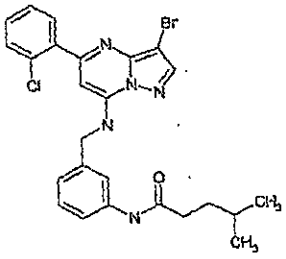
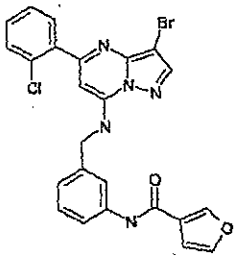
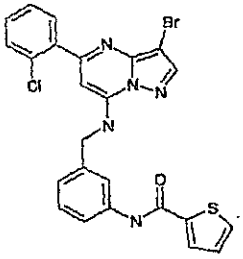
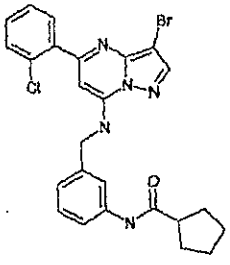
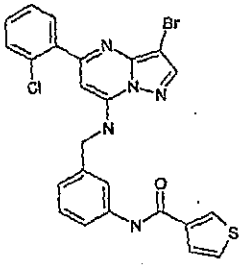
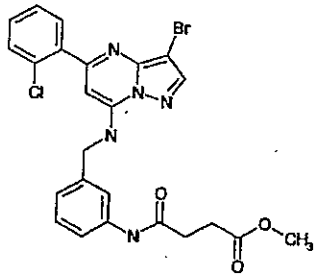
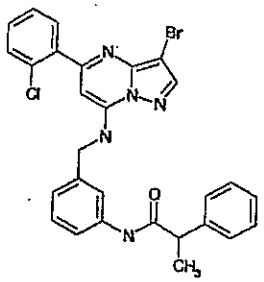
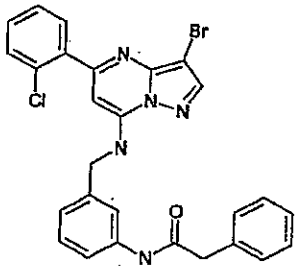
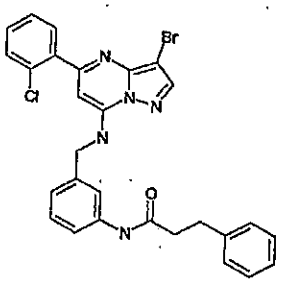
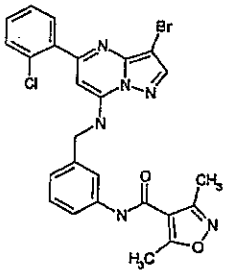
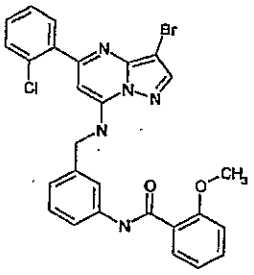
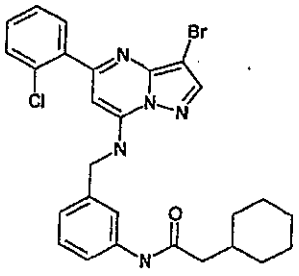
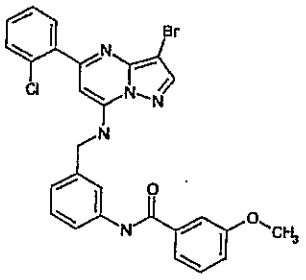
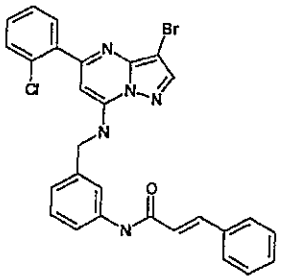
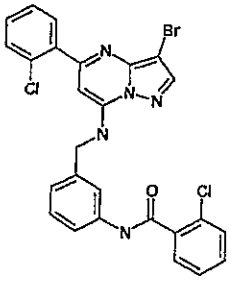
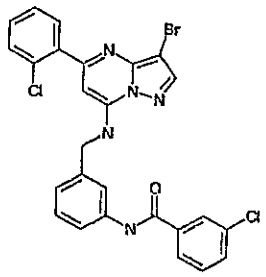
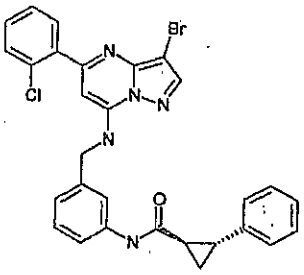
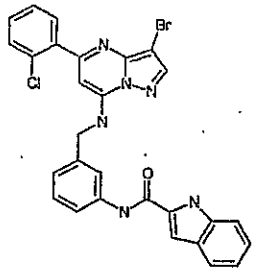
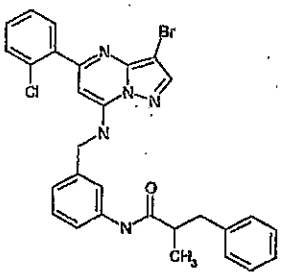
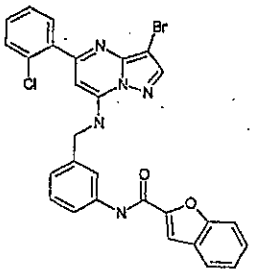
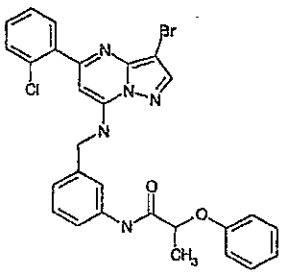
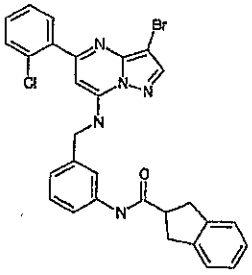
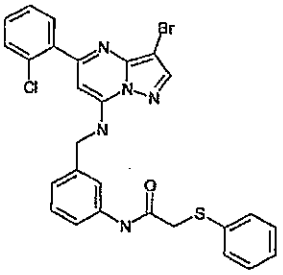
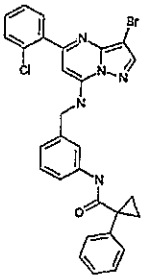
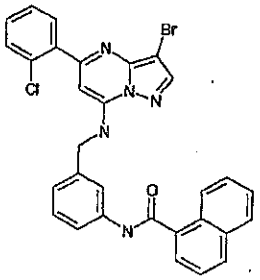
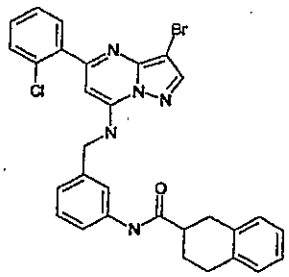
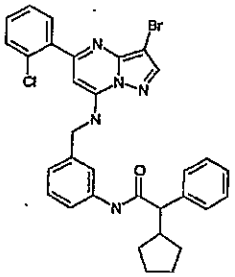
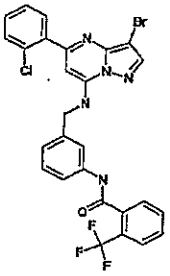
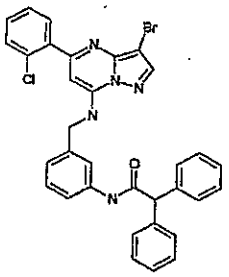
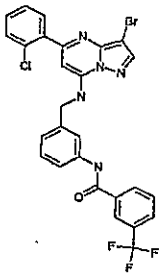
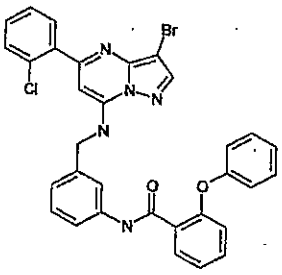
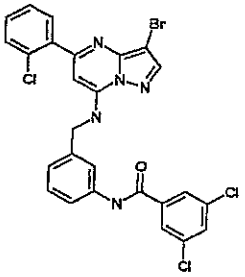
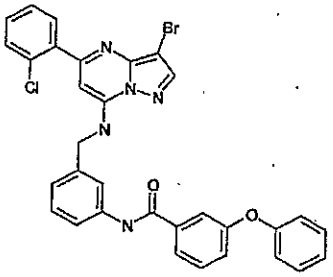
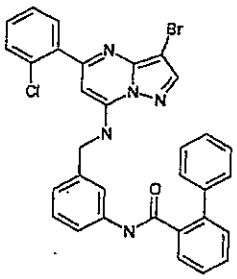
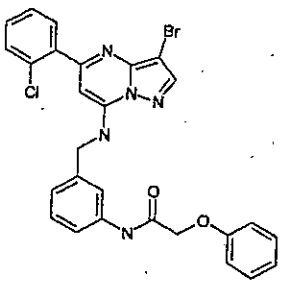
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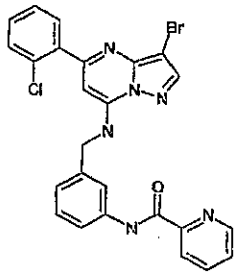
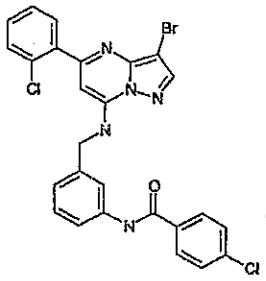
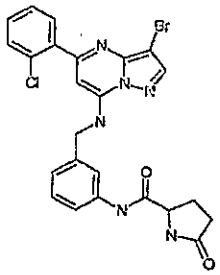
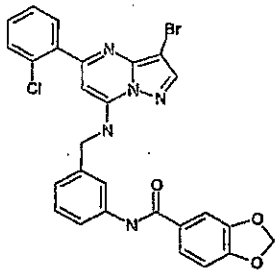
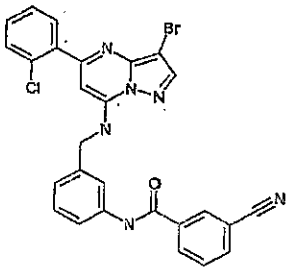
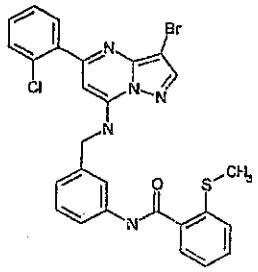
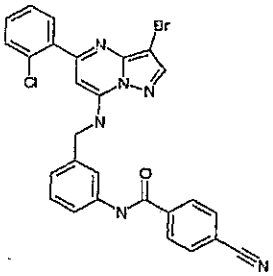
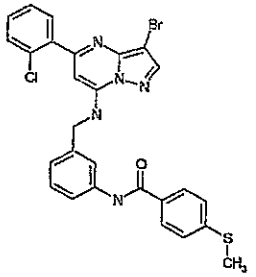
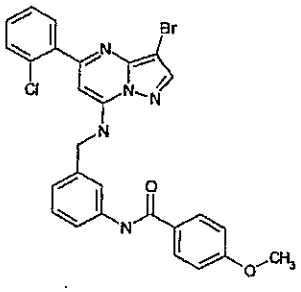
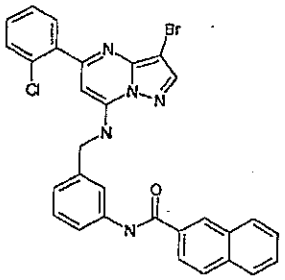
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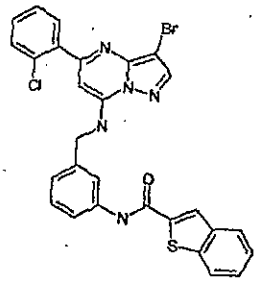
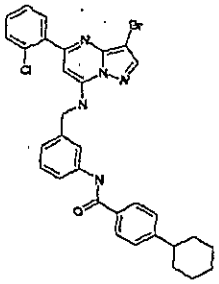
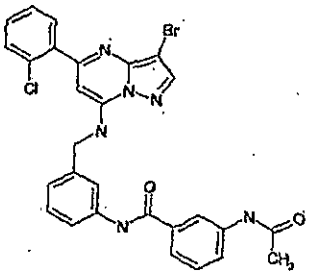
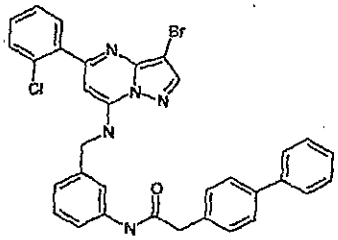
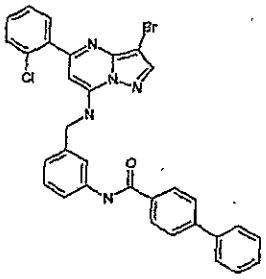
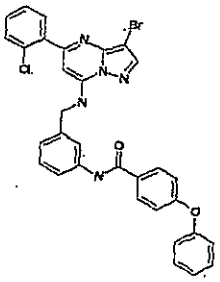
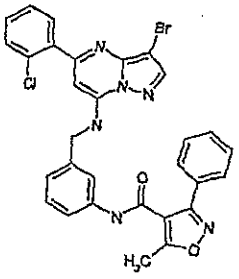
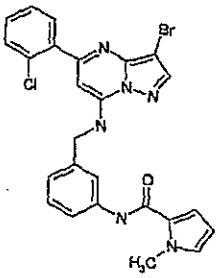
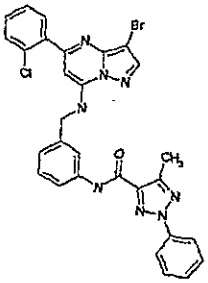
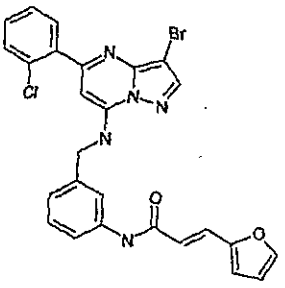
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	1. 4801 2. 498.27		1.4806 2. 528.29
	1. 4802 2. 512.28		1.. 4807 2. 528.29
	1. 4803 2. 524.29		1. 4808 2. 528.29
	1. 4804 2. 524.29		1. 4809 2. 540.3
	1. 4805 2. 526.29		1. 4810 2. 540.3

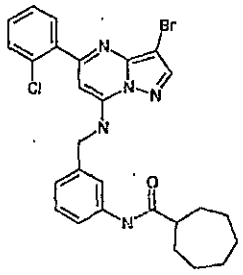
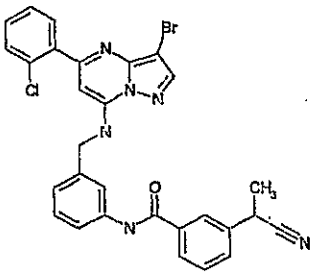
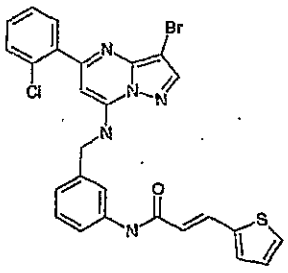
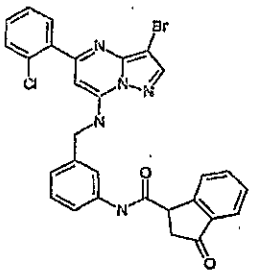
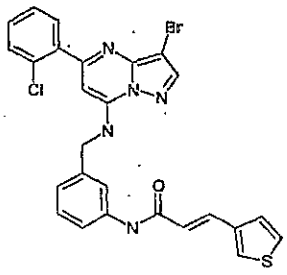
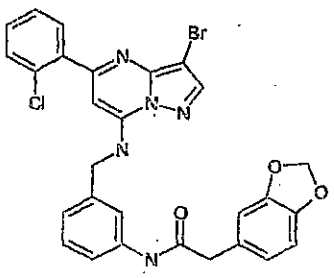
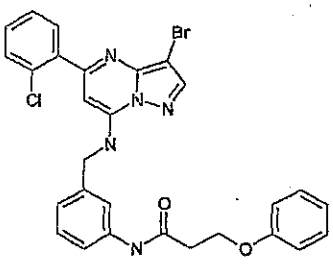
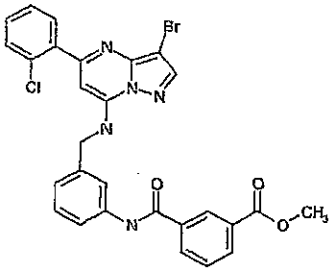
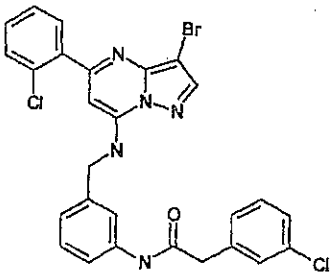
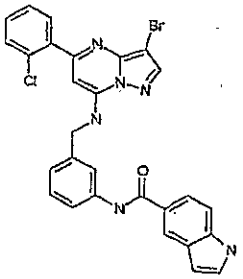
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	1. 4811 2. 544.3		1. 4816 2. 562.31
	1. 4812 2. 548.3		1. 4817 2. 562.31
	1. 4813 2. 553.3		1. 4818 2. 564.31
	1. 4814 2. 554.3		1. 4819 2. 564.31
	1. 4815 2. 560.31		1. 4820 2. 568.31

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4821 2. 568.31		1. 4826 2. 574.32
	1. 4822 2. 573.32		1. 4828 2. 576.32
	1. 4821 2. 574.32		1. 4829 2. 578.32
	1. 4824 2. 574.32		1. 4829 2. 580.32
	1. 4825 2. 574.32		1. 4830 2. 584.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4831 2. 588.32		1. 4836 2. 616.34
	1. 4832 2. 602.33		1. 4837 2. 624.34
	1. 4833 2. 602.33		1. 4838 2. 626.34
	1. 4834 2. 602.33		1. 4839 2. 626.34
	1. 4835 2. 610.34		1. 4840 2. 564.31

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4841 2. 535.29		1. 4846 2. 568.31
	1. 4842 2. 541.3		1. 4847 2. 578.32
	1. 4843 2. 559.31		1. 4848 2. 580.32
	1. 4844 2. 559.31		1. 4849 2. 580.32
	1. 4845 2. 564.31		1. 4850 2. 584.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4851 2. 590.32		1. 4856 2. 616.34
	1. 4852 2. 591.33		1. 4857 2. 624.34
	1. 4853 2. 610.34		1. 4858 2. 626.34
	1. 4854 2. 615.34		1. 4859 2. 537.3
	1. 4855 2. 615.34		1. 4860 2. 550.3

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4861 2. 554.3		1. 4866 2. 587.32
	1. 4862 2. 566.31		1. 4867 2. 588.32
	1. 4863 2. 566.31		1. 4868 2. 592.33
	1. 4864 2. 578.32		1. 4869 2. 592.33
	1. 4865 2. 582.32		1. 4870 2. 573.32

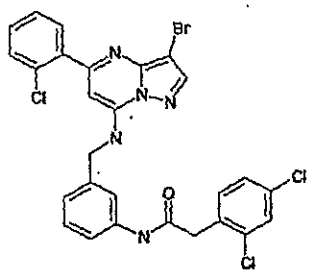
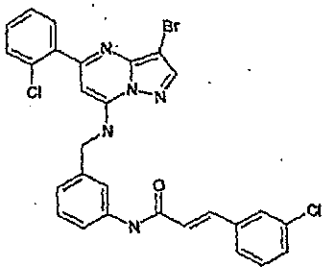
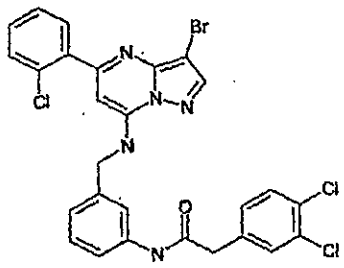
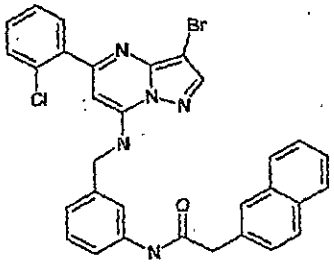
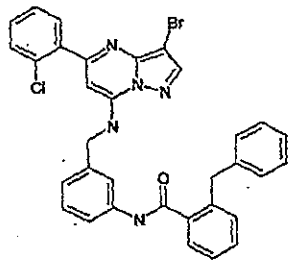
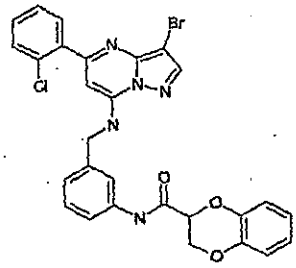
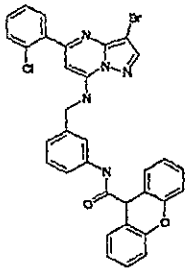
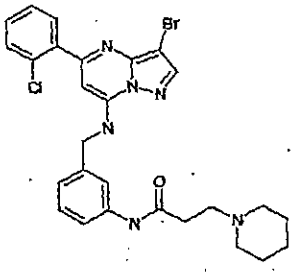
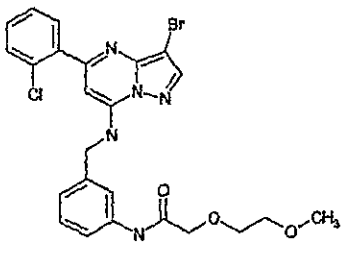
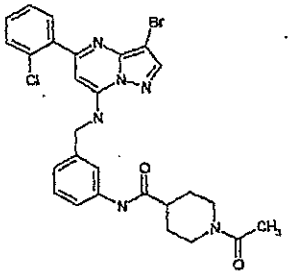
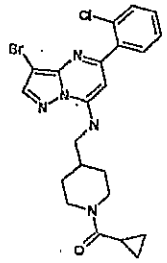
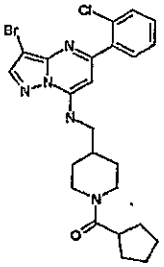
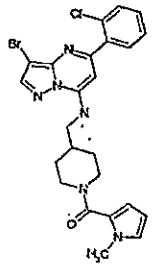
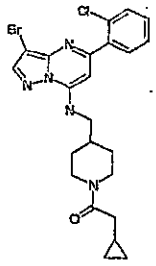
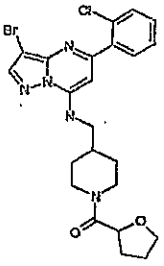
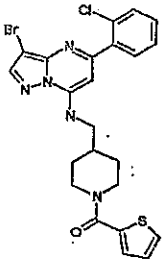
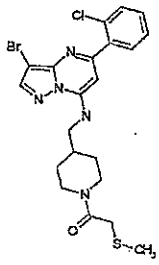
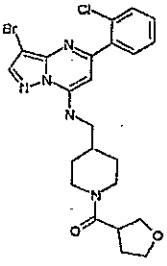
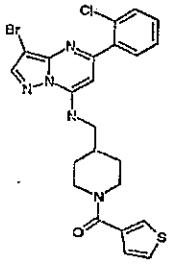
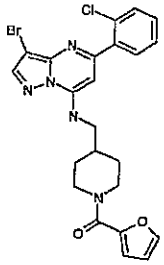
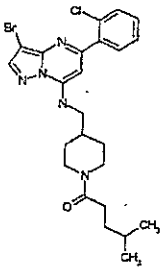
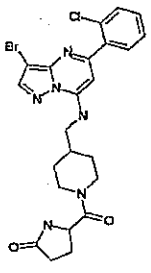
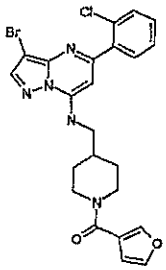
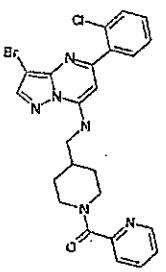
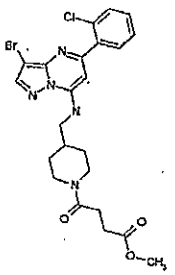
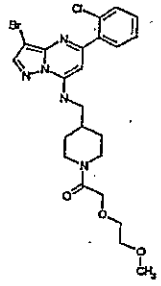
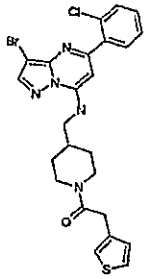
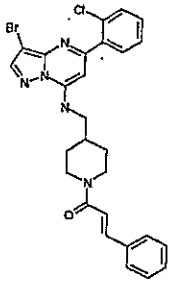
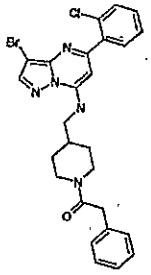
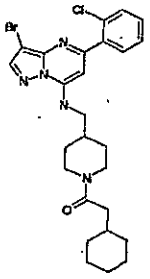
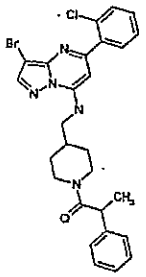
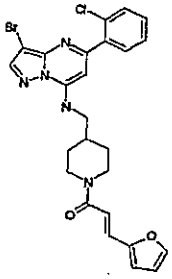
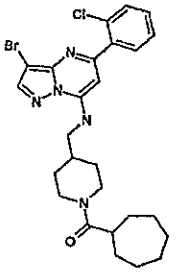
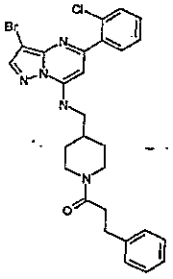
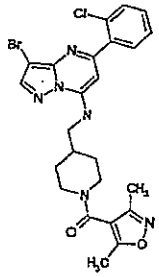
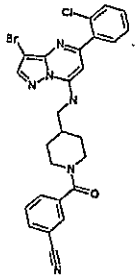
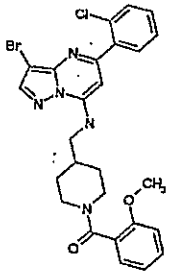
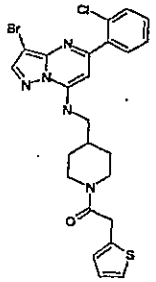
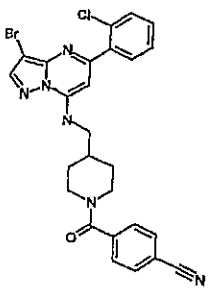
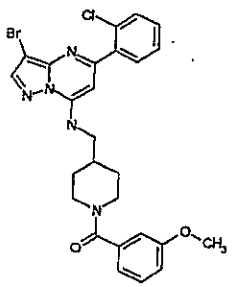
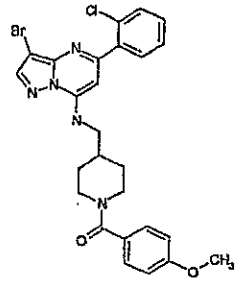
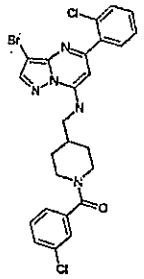
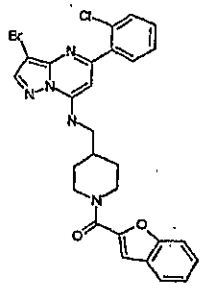
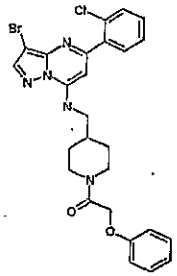
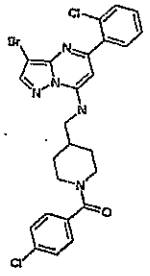
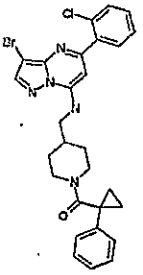
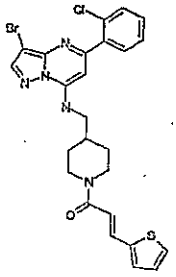
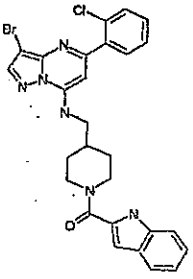
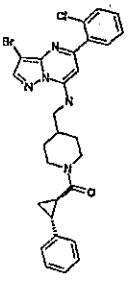
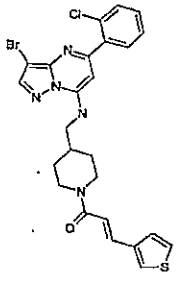
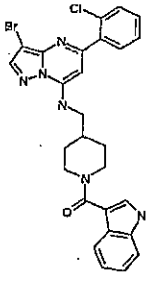
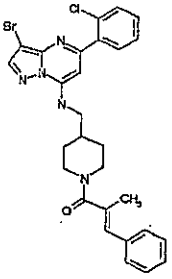
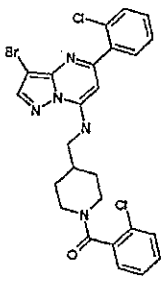
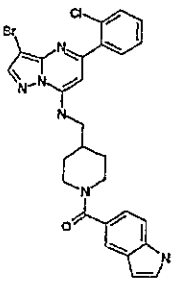
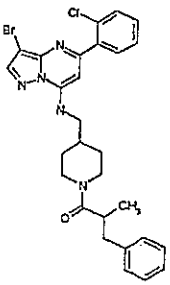
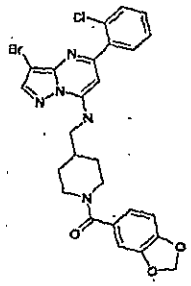
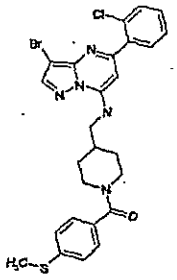
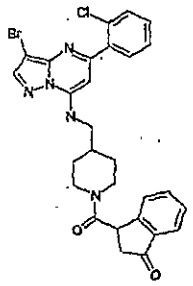
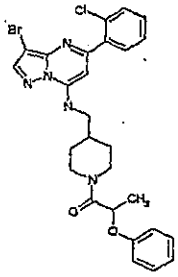
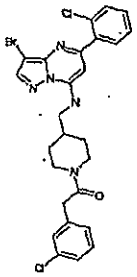
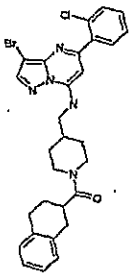
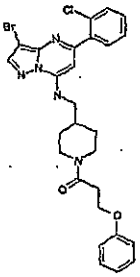
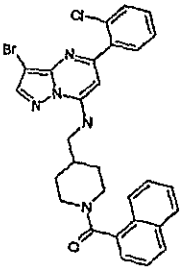
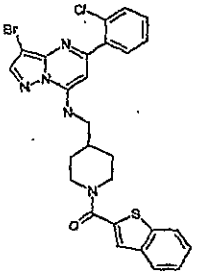
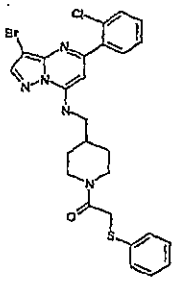
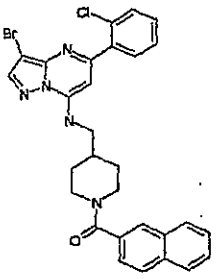
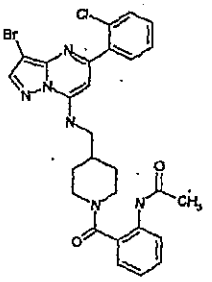
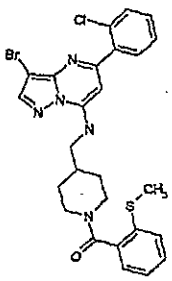
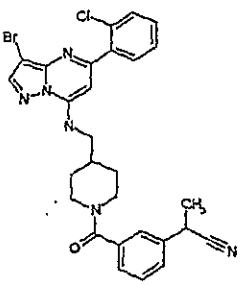
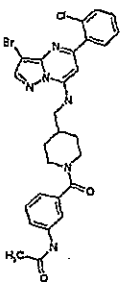
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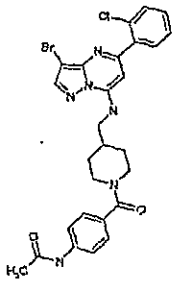
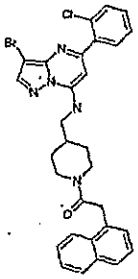
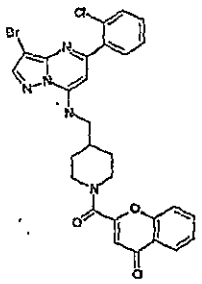
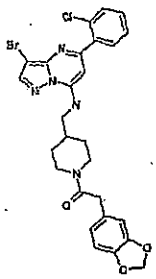
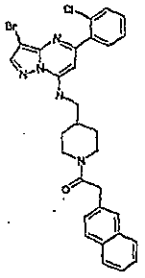
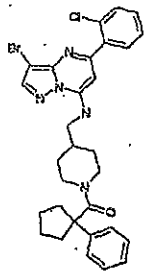
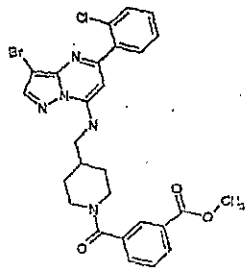
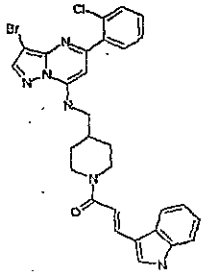
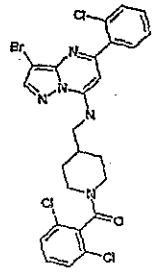
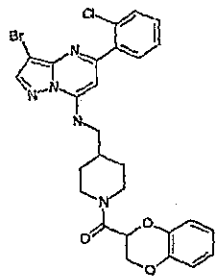
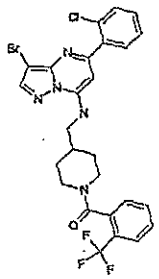
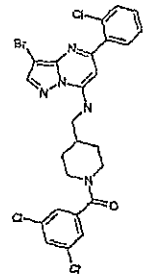
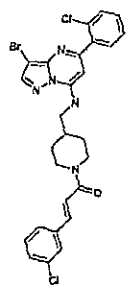
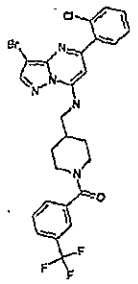
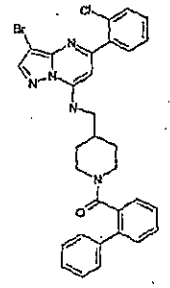
TABLE 49

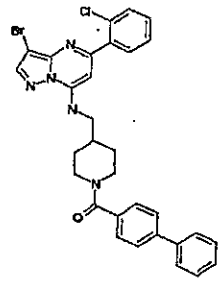
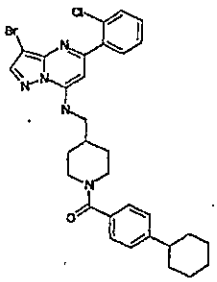
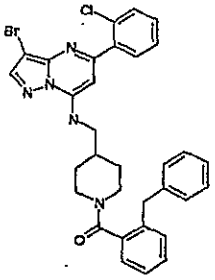
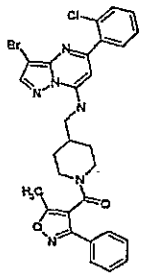
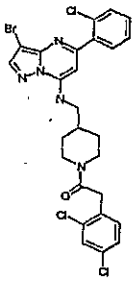
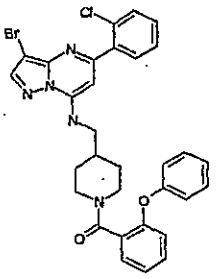
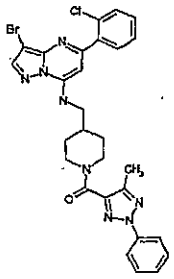
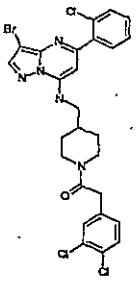
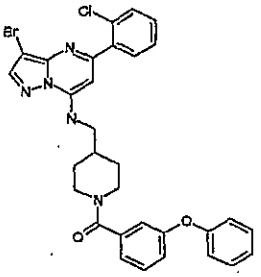
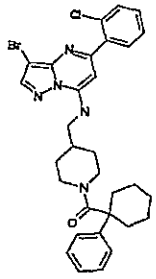
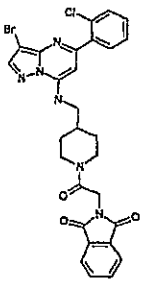
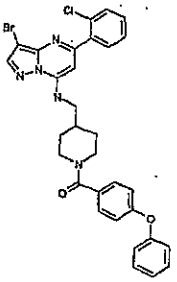
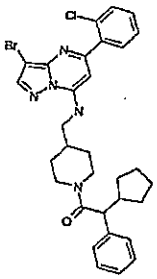
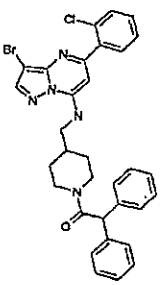
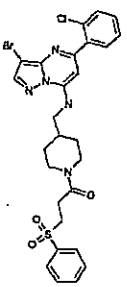
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	1. 4902 2. 504.28		1. 4907 2. 520.29		1. 4912 2. 532.29
	1. 4903 2. 510.28		1. 4908 2. 520.29		1. 4913 2. 532.29
	1. 4904 2. 516.28		1. 4909 2. 520.29		1. 4914 2. 533.29
	1. 4905 2. 516.28		1. 4910 2. 527.29		1. 4915 2. 536.29

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4916 2. 538.3		1. 4921 2. 546.3		1. 4926 2 552.3
	1. 4917 2 .540.3		1. 4922 2. 546.3		1. 4927 2. 554.3
	1. 4918 2. 542.3		1. 4923 2. 546.3		1. 4928 2. 554.3
	1. 4919 2. 545.3		1. 4924 2. 551.3		1. 4929 2. 556.31
	1. 4920 2. 546.3		1. 4925 2. 551.3		1. 4930 2. 556.31

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4931 2. 556.31		1. 4936 2. 560.31		1. 4941 2. 566.31
	1. 4932 2. 556.31		1. 4937 2. 560.31		1. 4942 2. 566.31
	1. 4933 2. 558.31		1. 4938 2. 565.31		1. 4943 2. 566.31
	1. 4934 3. 558.31		1. 4939 2. 565.3		1. 4944 2. 566.31
	1. 4935 2. 560.31		1. 4940 2. 565.31		1. 4945 2. 568.31

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. m/z
	1. 4946 2. 570.31		1. 4951 2. 571.31		1. 4956 580.3
	1. 4947 2. 570.31		1. 4952 2. 574.32		1. 4957 2 580.32
	1. 4948 2. 570.31		1. 4953 2. 576.32		1. 4958 2. 582.32
	1. 4949 2. 572.31		1. 4954 2. 576.32		1. 4959 2. 583.32
	1. 4950 2. 572.31		. 4955 2. 579.32		1. 4960 2. 583.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. E m
	1. 4961 2. 583.32		1. 4966 2. 590.32		1. 49 594
	1. 4962 2. 584.32		1. 4967 2. 590.32		1. 497 594.3
	1. 4963 2. 584.32		1. 4968 2. 591.3		1. 4973 594.33
	1. 4964 2. 584.32		1. 4969 2. 594.33		1. 4974 2 594.33
	1. 4965 2. 586.32		1. 4970 2. 594.33		1. 4975 2. 602.33

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 4976 2. 602.33		1. 4981 2. 608.33		1. 4986 2. 616.34
	1. 4977 2. 607.33		1. 4982 2. 608.33		1. 4987 2. 618.34
	1. 4978 2. 607.33		1. 4983 2. 608.33		1. 4988 2. 618.34
	1. 4979 2. 605.33		1. 4984 2. 609.33		1. 4089 2. 618.34
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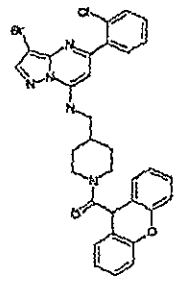
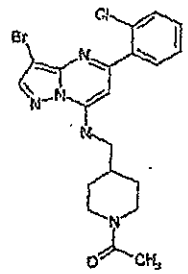
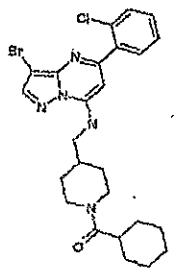
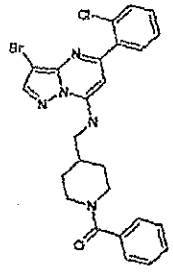
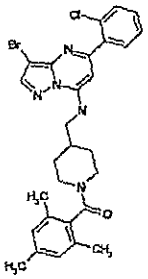
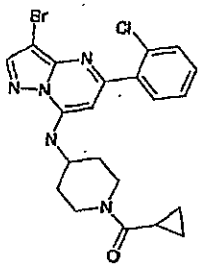
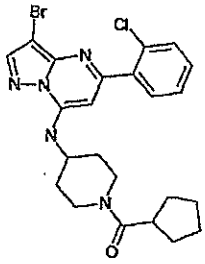
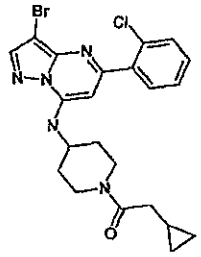
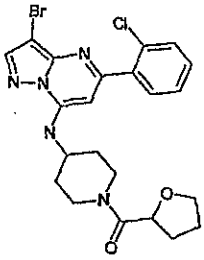
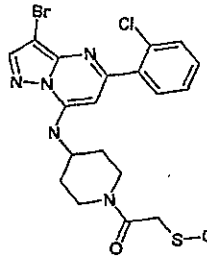
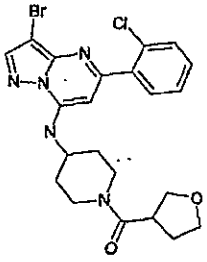
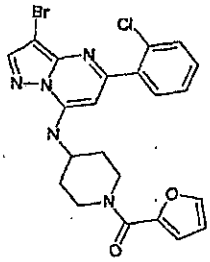
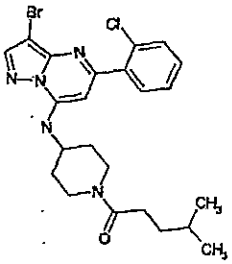
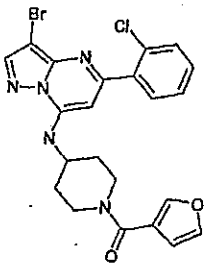
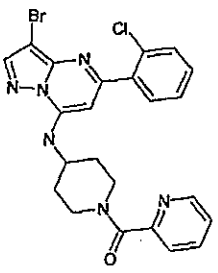
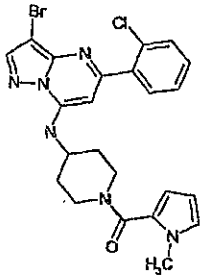
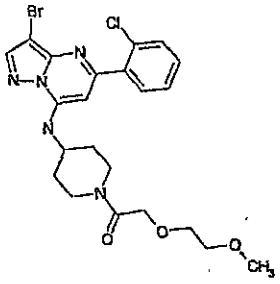
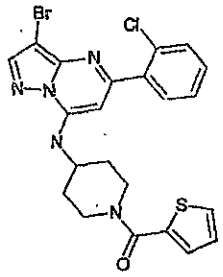
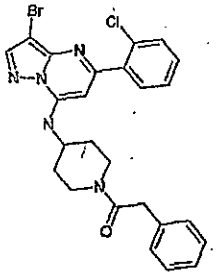
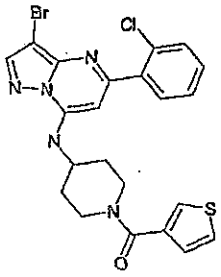
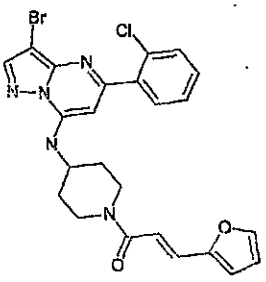
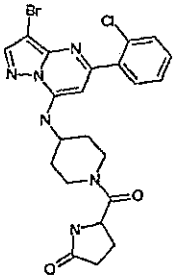
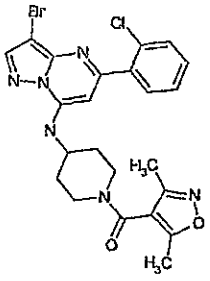
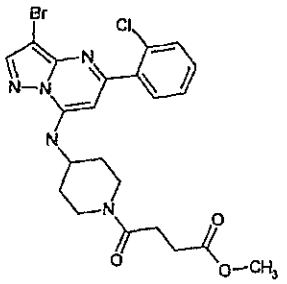
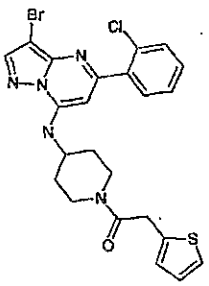
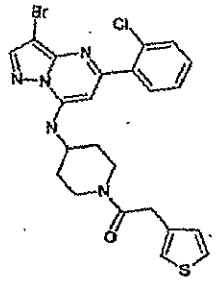
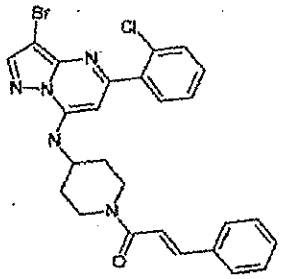
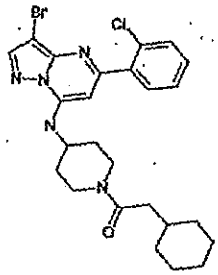
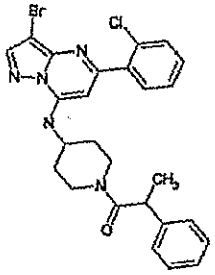
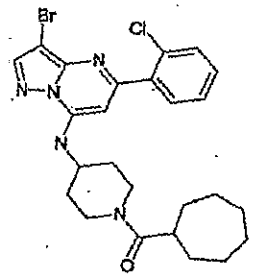
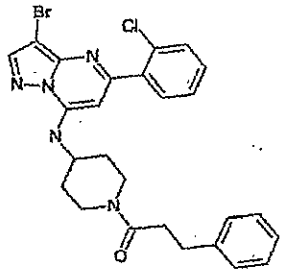
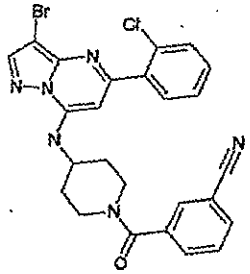
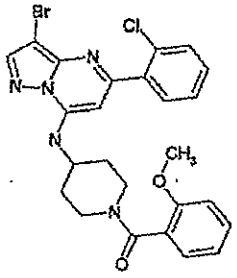
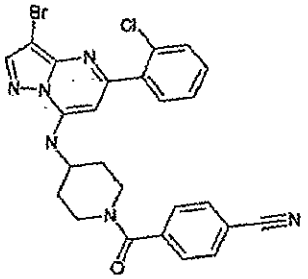
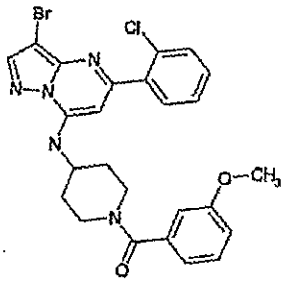
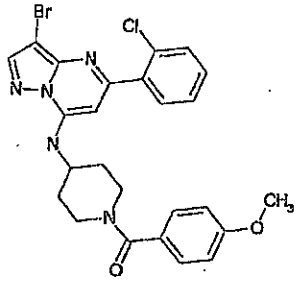
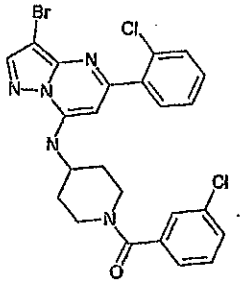
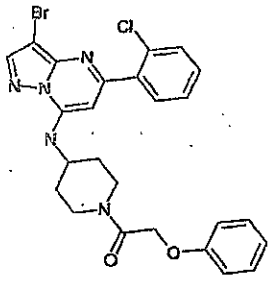
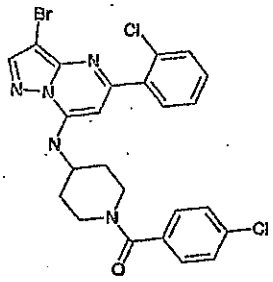
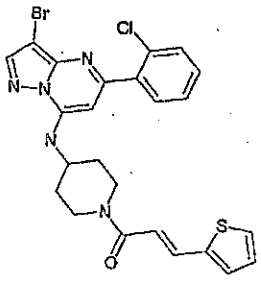
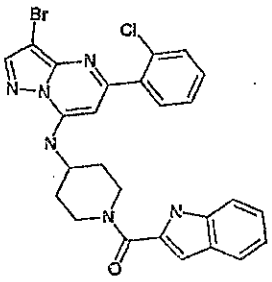
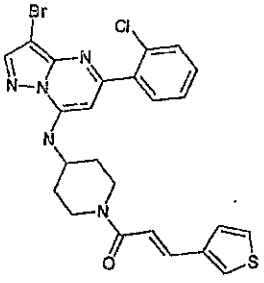
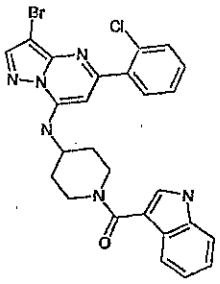
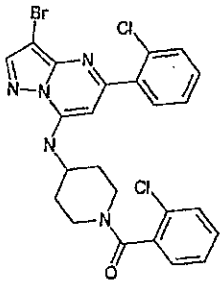
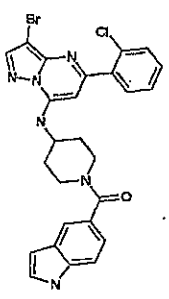
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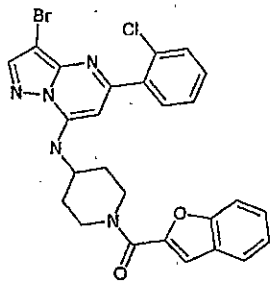
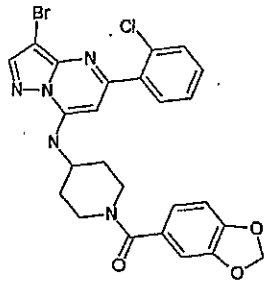
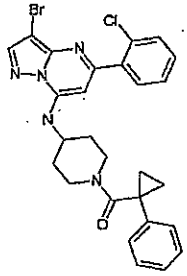
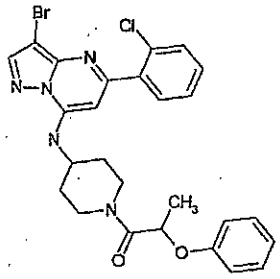
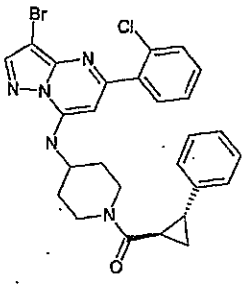
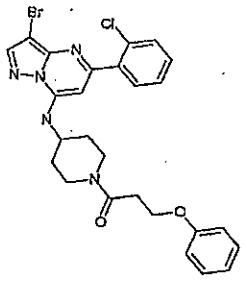
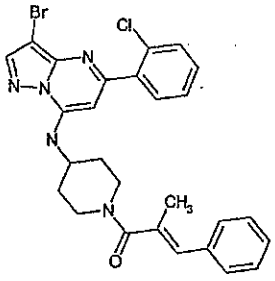
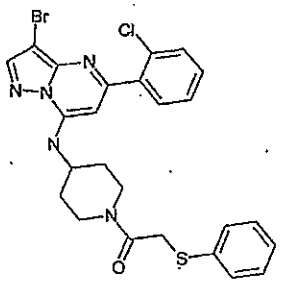
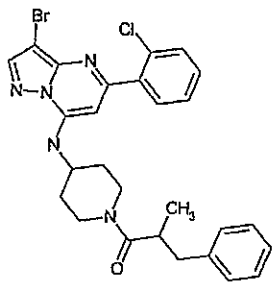
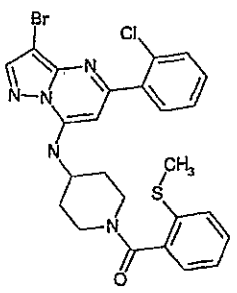
TABLE 50

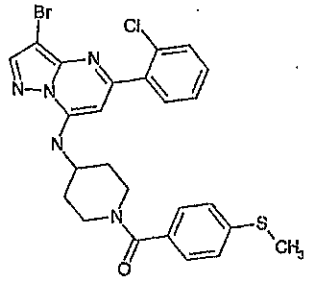
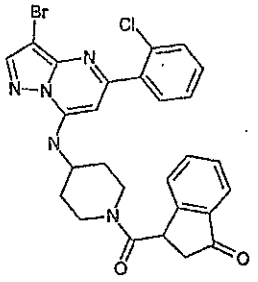
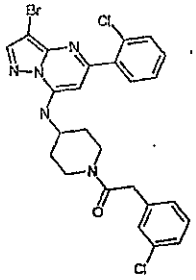
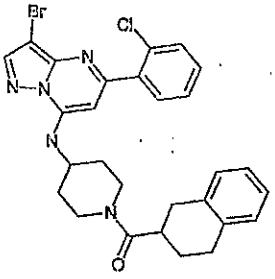
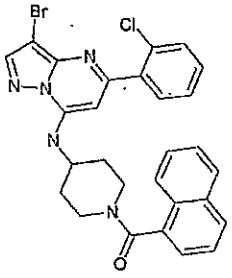
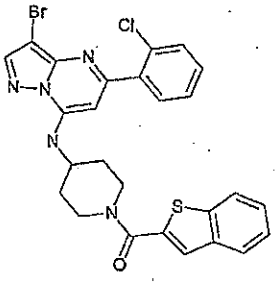
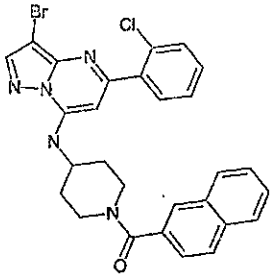
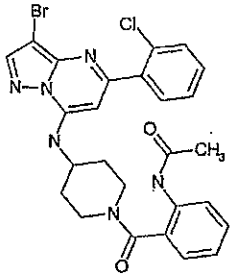
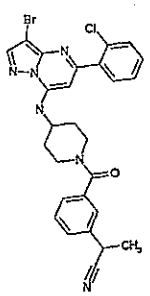
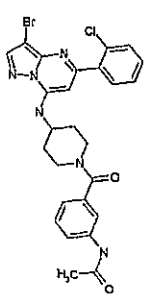
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	1. 5002 2. 490.27		1. 5007 2. 506.28
	1. 5003 2. 496.27		1. 5008 2. 506.28
	1. 5004 2. 502.28		1. 5009 2. 506.28
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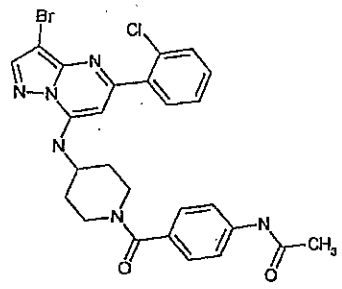
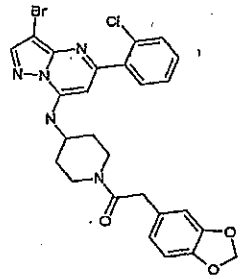
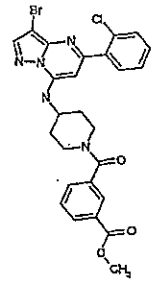
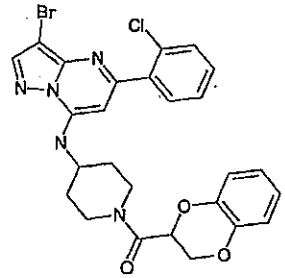
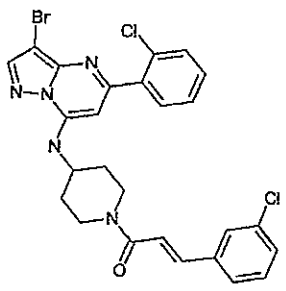
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	1. 5012 2. 518.28		1. 5017 2. 526.29
	1. 5013 2. 518.28		1. 5018 2. 528.29
	1. 5014 2. 519.29		1. 5019 2. 531.29
	1. 5015 2. 522.29		1. 5020 2. 532.29

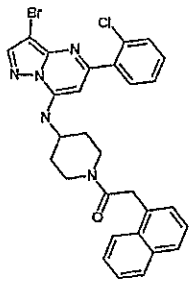
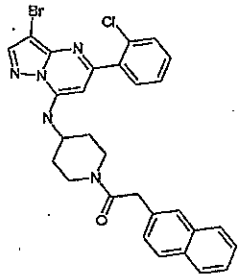
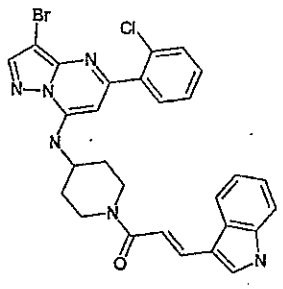
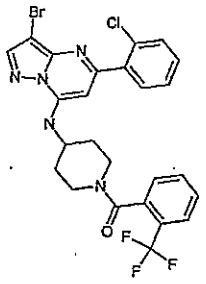
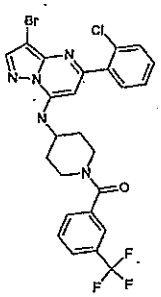
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	1. 5023 2. 532.29		1. 5028 2. 540.3
	1. 5024 2. 537.3		1. 5029 2. 542.3
	1. 5025 2. 537.3		1. 5030 2. 542.3

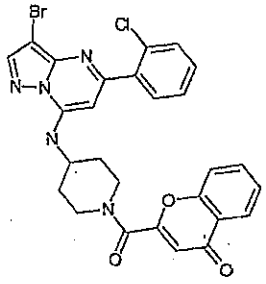
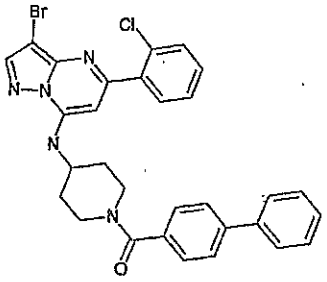
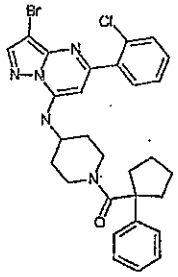
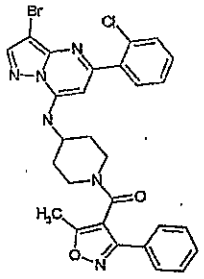
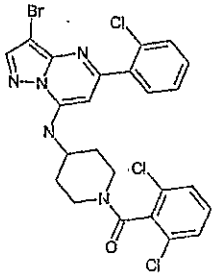
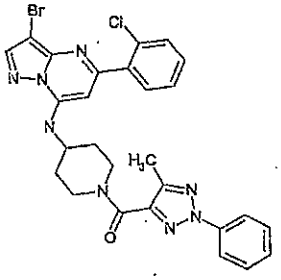
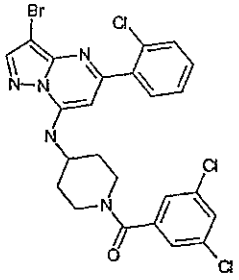
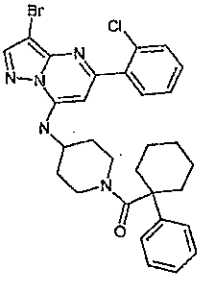
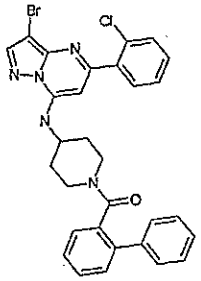
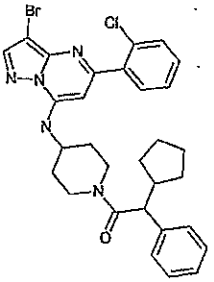
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	1. 5032 2. 542.3		1. 5037 2 .546.3
	1. 5033 2. 544.3		1. 5038 2. 551.3
	1. 5034 2. 544.3		1. 5039 2. 551.3
	1. 5035 2 .546.3		1. 5040

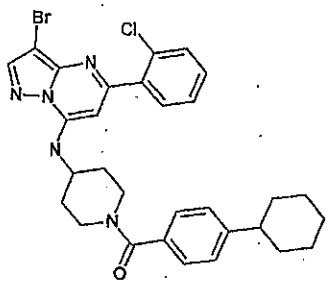
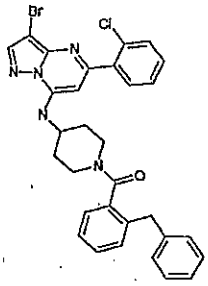
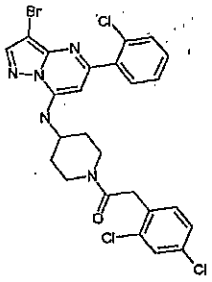
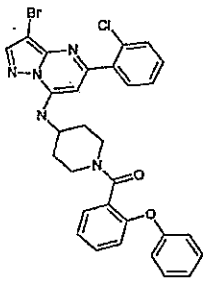
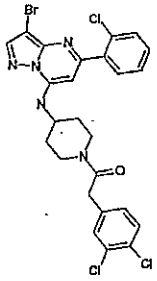
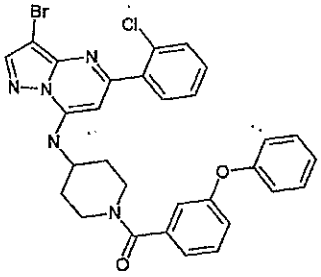
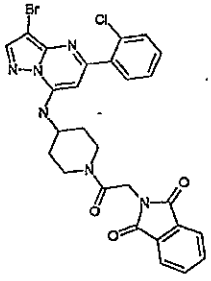
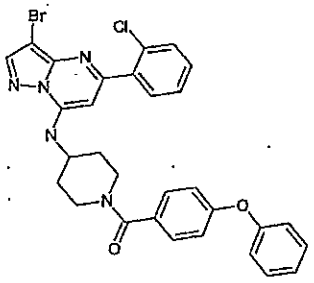
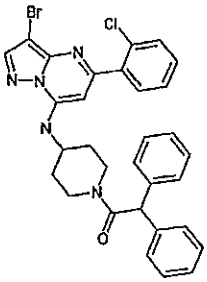
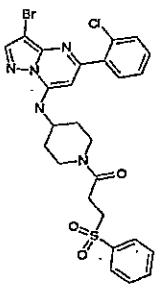
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	1. 5042 2. 552.3		1. 5047 2. 556.31
	1. 5043 2. 552.3		1. 5048 2. 556.31
	1. 5044 2. 552.3		1. 5049 2. 558.31
	1. 5045 2. 554.3		1. 5050 2. 558.31

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5051 2. 558.31		1. 5056 2. 566.31
	1. 5052 2. 560.31		1. 5057 2. 566.31
	1. 5053 2. 562.31		1. 5058 2. 568.31
	1. 5054 2. 562.31		1. 5059 2. 569.31
	1. 5055 2. 565.31		1. 5060 2. 569.31

Product	1. Ex. 2. m/z
	1. 5061
	1. 5062 2. 570.31
	1. 5063 2. 570.31
	1. 5064 2. 570.31
	1. 5065 572.31

Product	1. Ex. 2. m/z
	1. 5066 2. 576.32
	1. 5067 2. 576.32
	1. 5068 2. 577.32
	1. 5069 2. 580.32
	1. 5070 2. 580.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5071 2. 580.32		1 5076 2. 588.32
	1. 5072 2. 580.32		1. 5077 2. 593.33
	1 5073 2. 580.32		1 5078 2. 593.33
	1 5074 2. 580.32		1. 5079 2. 594.33
	1 5075 2. 588.32		1. 5080 2. 594.33

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5081 2. 594.33		1. 5086 2. 602.33
	1. 5082 2. 594.33		1.5087 2. 604.33
	1. 5083 2. 594.33		1. 5088 2. 604.33
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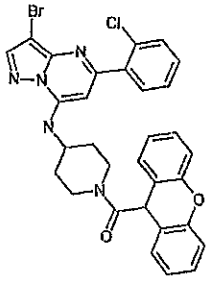
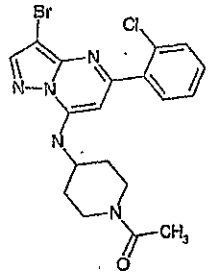
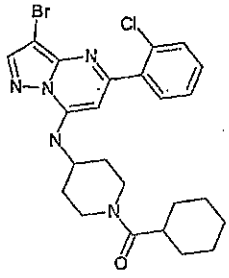
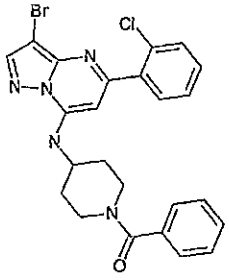
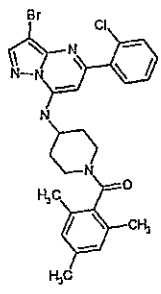
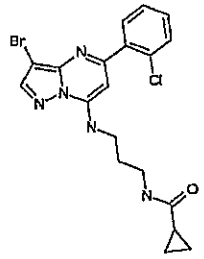
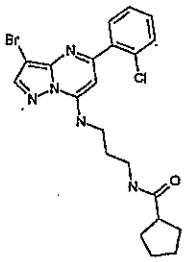
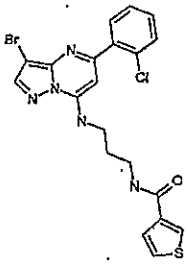
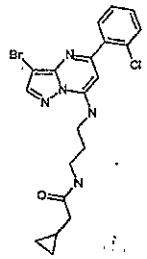
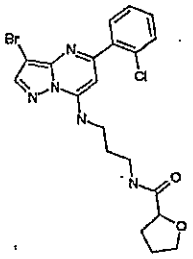
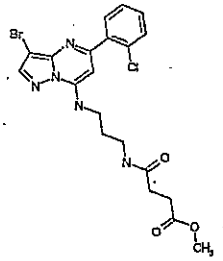
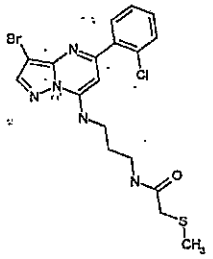
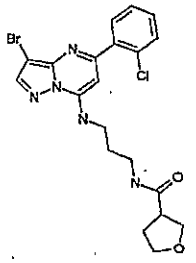
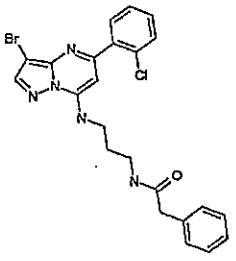
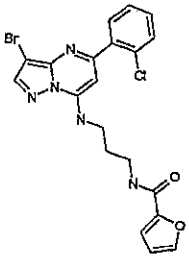
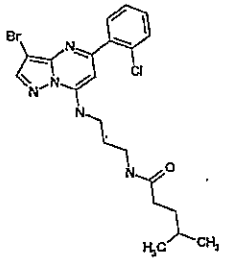
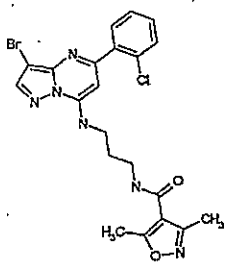
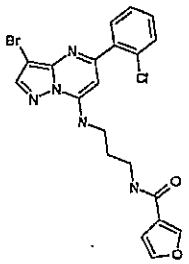
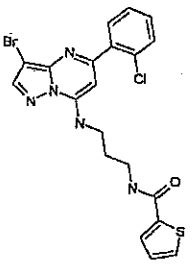
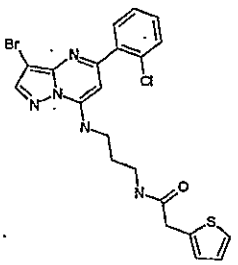
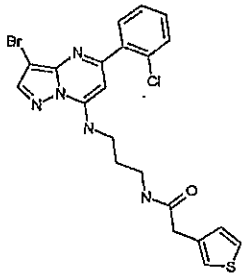
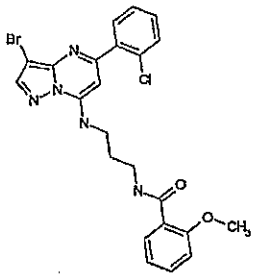
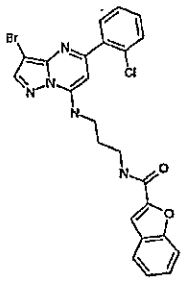
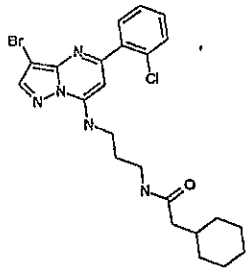
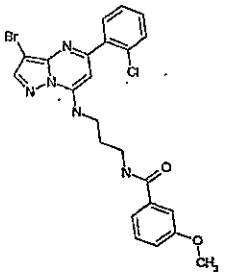
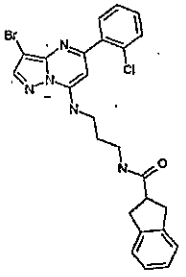
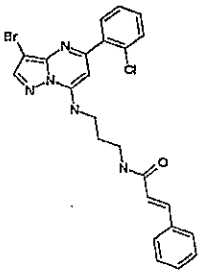
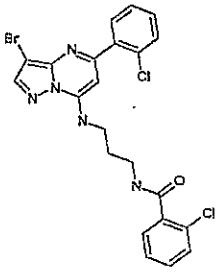
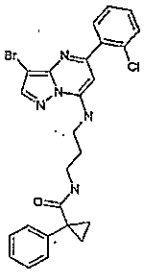
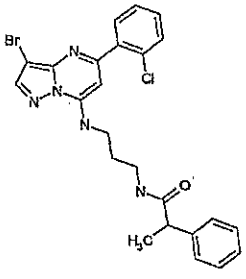
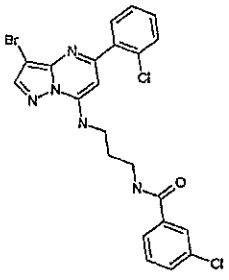
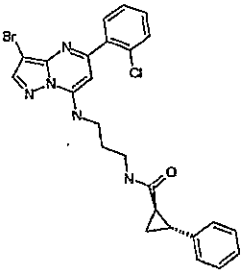
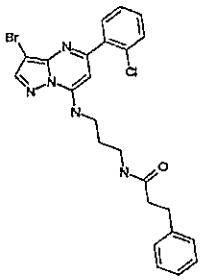
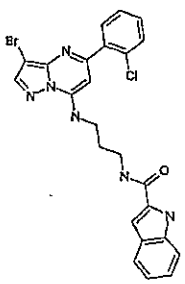
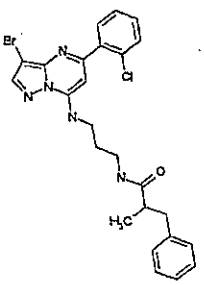
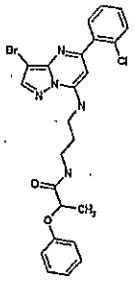
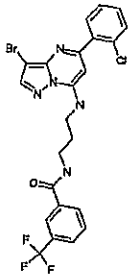
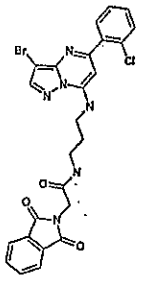
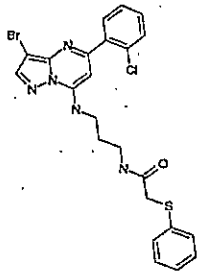
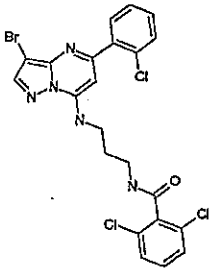
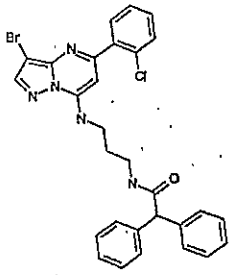
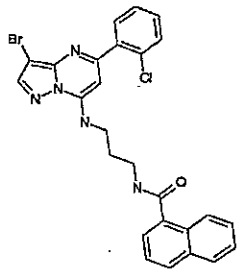
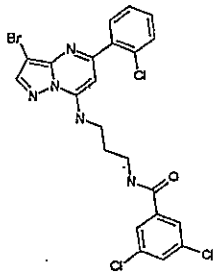
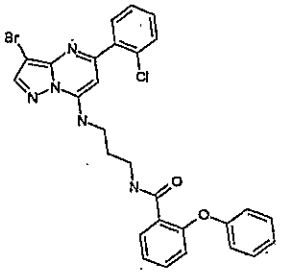
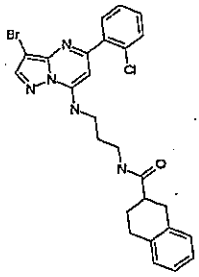
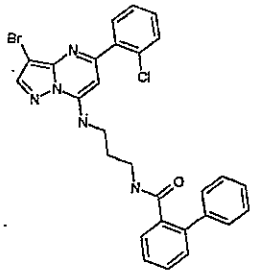
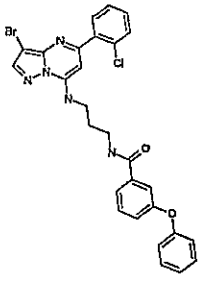
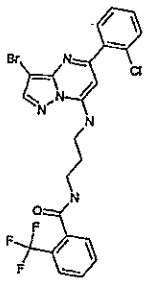
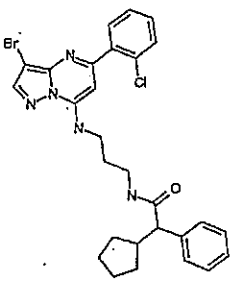
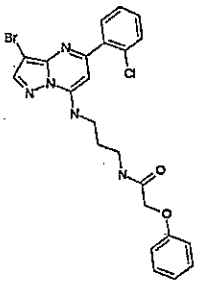
Product	1. Ex. 2. m/z
	1. 5091 2. 616.34
	1. 5092 2. 450.25
	1. 5093 2. 518.28
	1. 5094 2. 512.28
	1. 5095 2.554.3

TABLE 51

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5101 2. 448.8		1. 5106 2. 476.8		1. 5111 2. 490.8
	1. 5102 2. 462.8		1. 5107 2. 478.8		1. 5112 2. 494.8
	1: 5103 2. 468.8		1. 5108 2. 478.8		1. 5113 2. 498.8
	1. 5104 2. 474.7		1. 5109 2. 478.8		1. 5114 2. 503.8
	1. 5105 2.. 474.7		1. 5110 2. 490.8		1. 5115 2. 504.8

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5116 2. 504.8		1. 5121 2. 514.8		1. 5126 2. 524.8
	1. 5117 2. 504.9		1. 5122 2. 514.8		1. 5127 2. 524.9
	1. 5118 2. 510.8		1. 5123 2. 519.2		1. 5128 2 . 524.9
	1. 5119 2. 512.8		1. 5124 2. 519.2		1. 5129 2. 524.9
	1. 5120 2. 512.8		1. 5125 2. 523.8		1. 5130 2. 526.9

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5131 2. 528.8		1. 5136 2. 552.8		1. 5141 2. 567.8
	1. 5132 2. 530.9		1. 5137 2. 553.7		1. 5142 2. 574.9
	1. 5133 2. 534.8		1. 5138 2. 553.7		1. 5143 2. 576.9
	1. 5134 2. 538.9		1. 5139 2. 560.9		1. 5144 2. 576.9
	1. 5135 2. 552.8		1. 5140 2. 566.9		1. 5145 2. 514.8

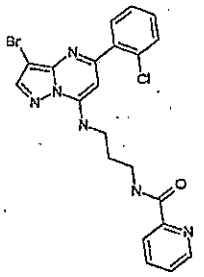
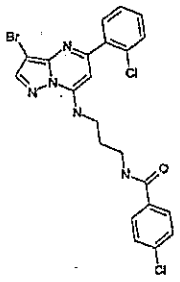
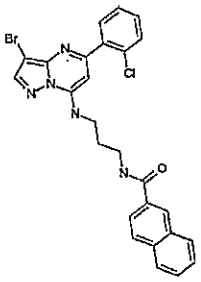
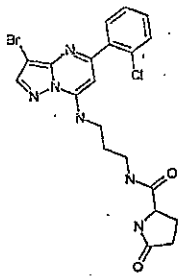
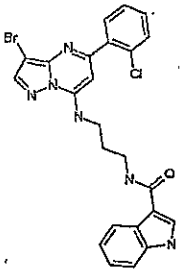
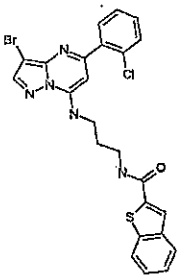
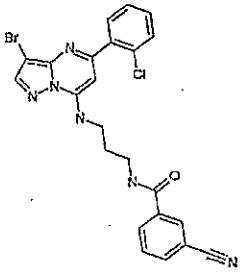
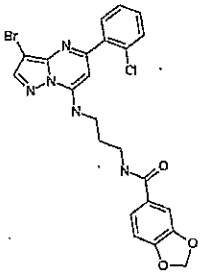
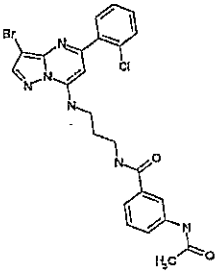
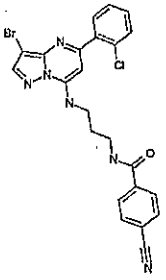
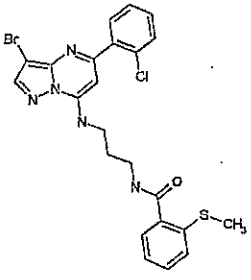
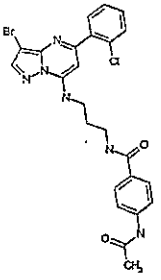
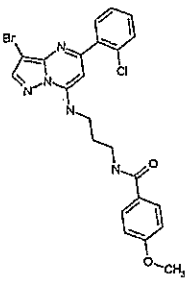
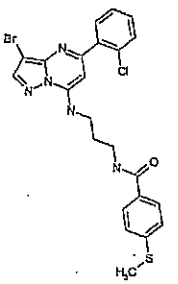
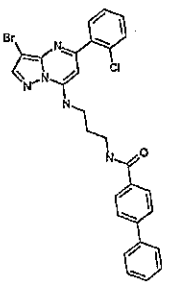
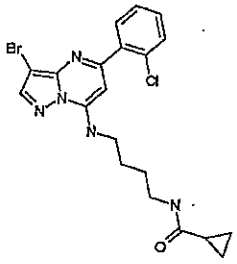
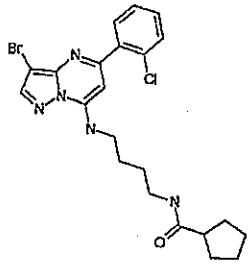
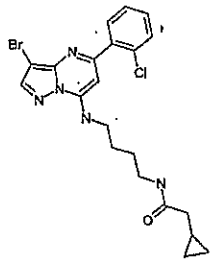
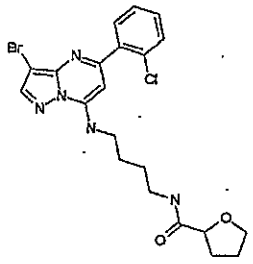
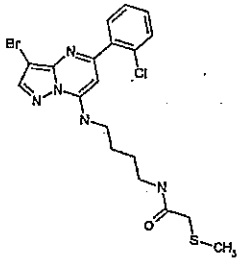
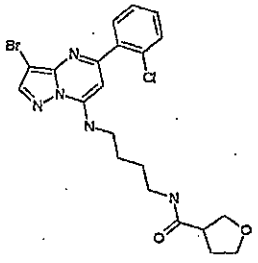
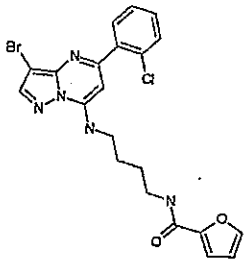
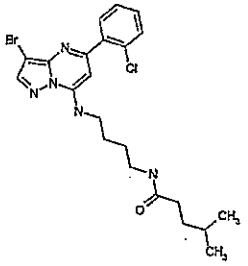
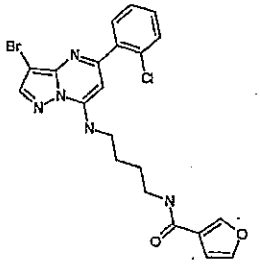
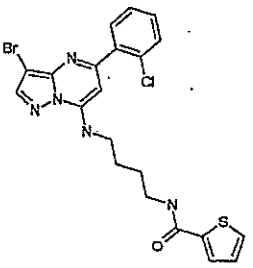
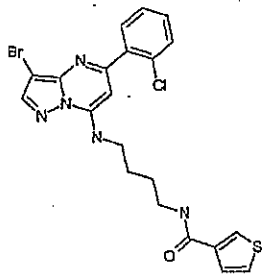
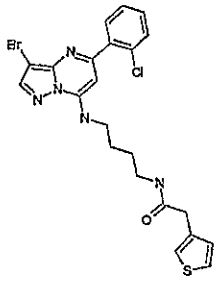
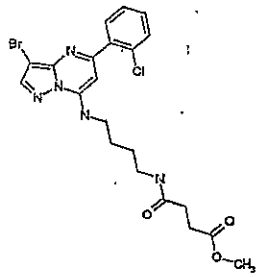
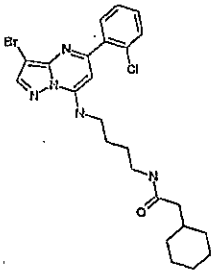
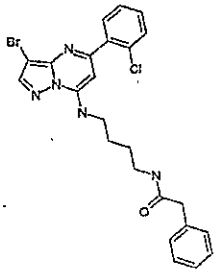
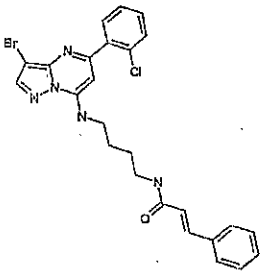
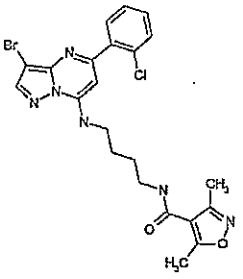
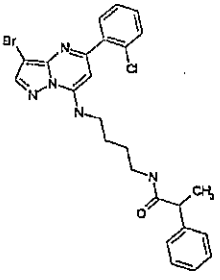
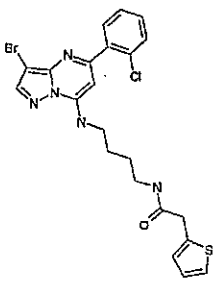
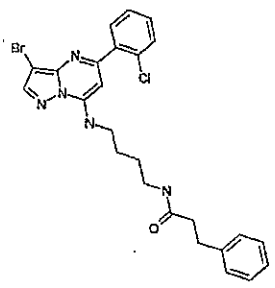
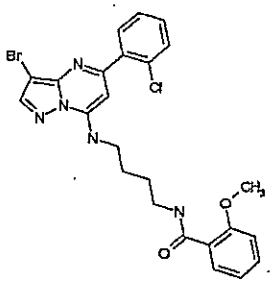
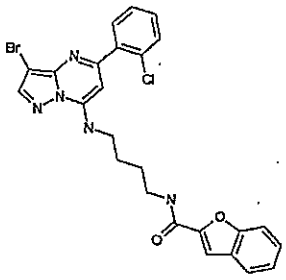
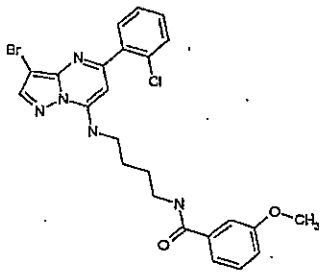
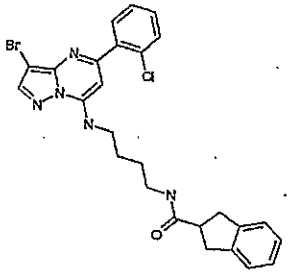
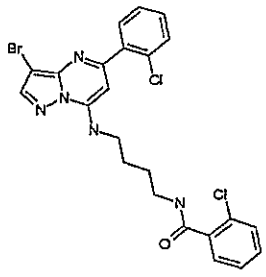
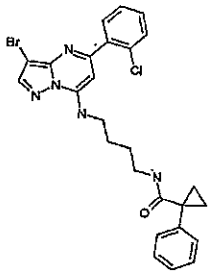
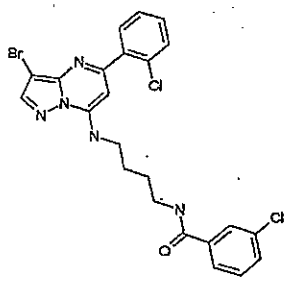
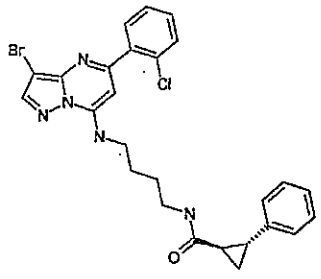
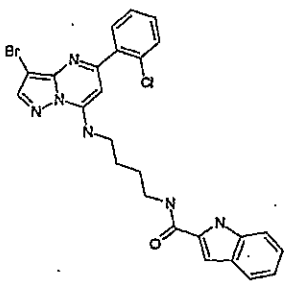
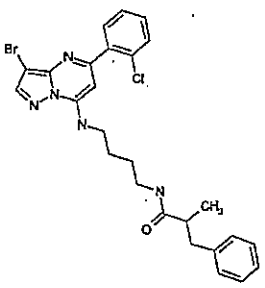
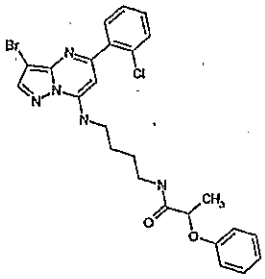
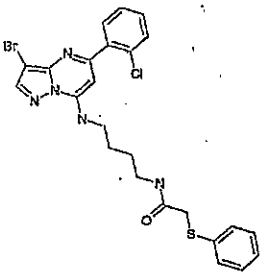
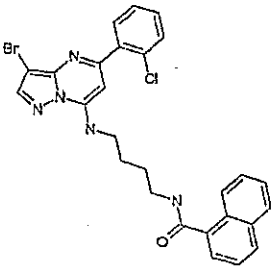
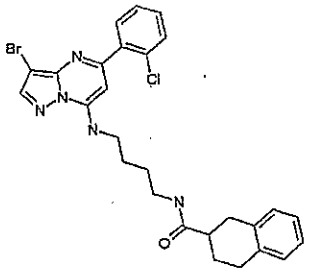
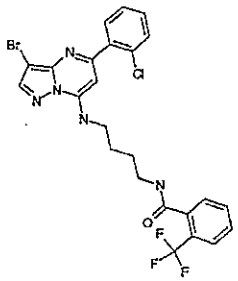
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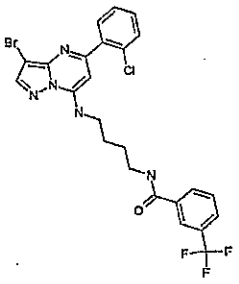
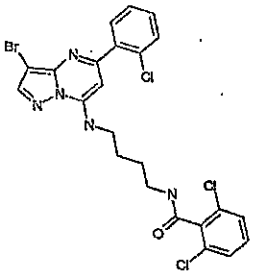
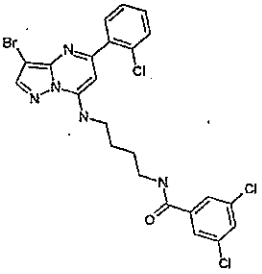
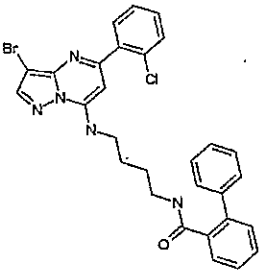
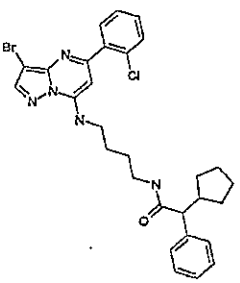
TABLE 52

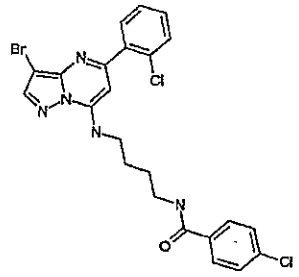
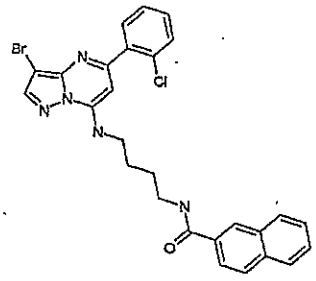
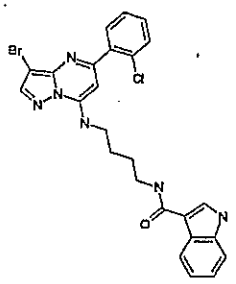
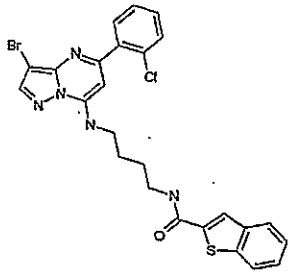
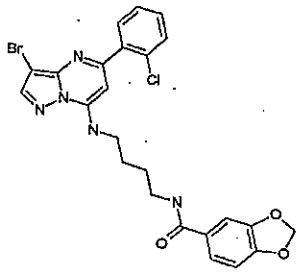
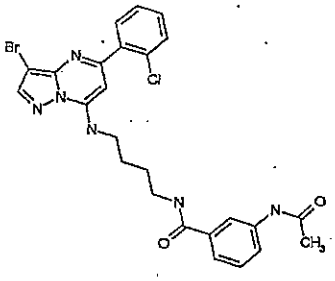
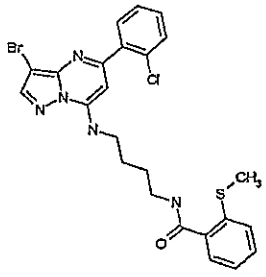
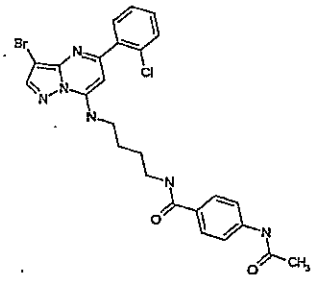
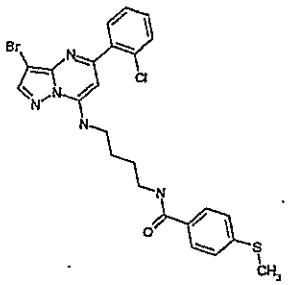
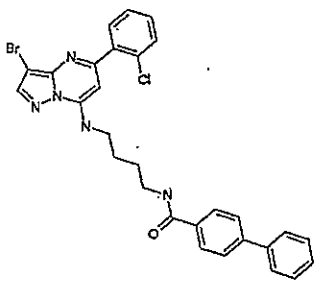
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	1. 5202 2. 476.8		1. 5207 2. 492.8
	1. 5203 2. 482.8		1. 5208 2. 492.8
	1. 5204 2. 488.8		1. 52009 2. 492.9
	1. 5205 2. 488.8		1. 5210 2. 504.8

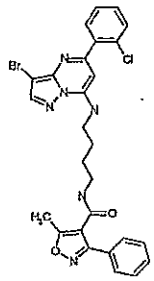
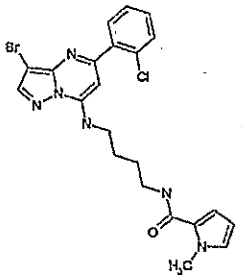
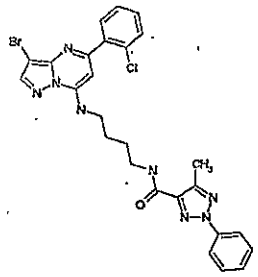
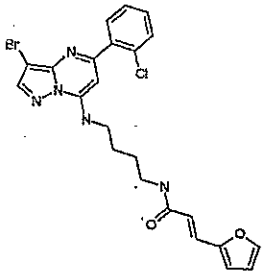
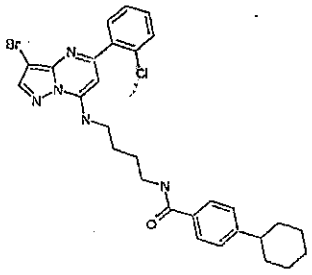
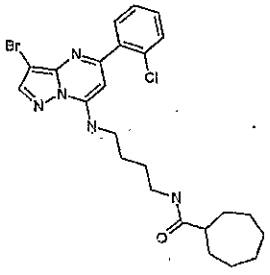
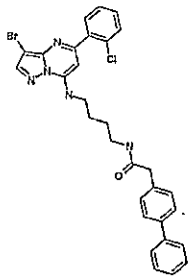
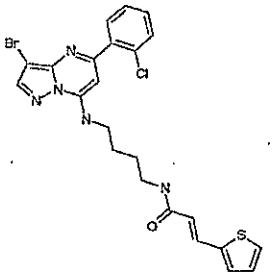
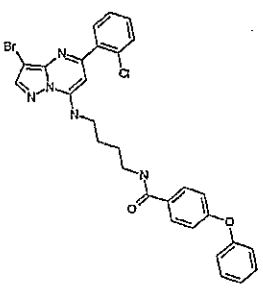
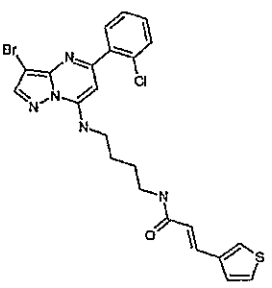
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	1. 5211 2. 504.8		1. 5216 2. 518.9
	2. 5212 2. 508.8		1. 5217 2. 518.9
	1. 5213 2. 512.8		1. 5218 2. 524.9
	1. 5214 2. 517.8		1. 5219 2. 526.9
	1. 5215 2. 518.9		1. 5220 2. 526.9

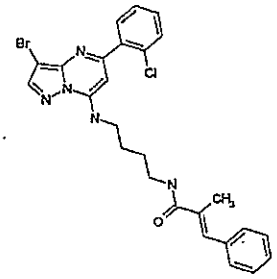
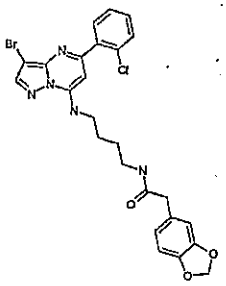
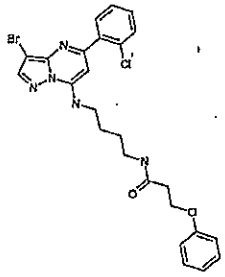
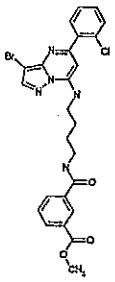
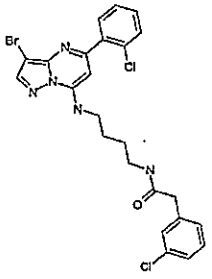
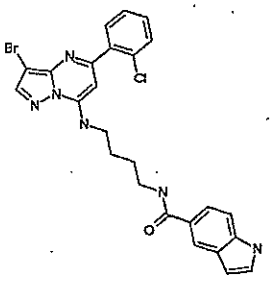
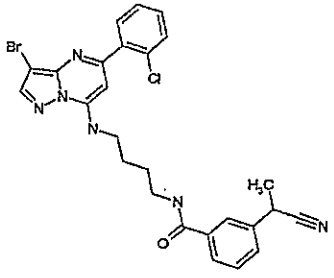
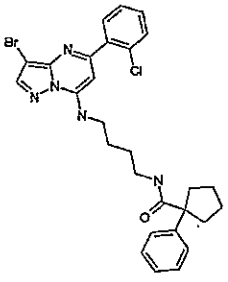
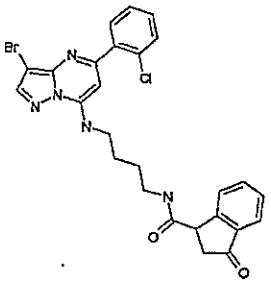
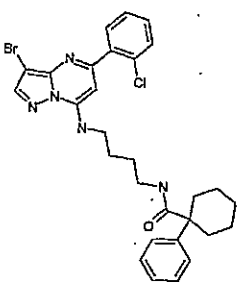
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	1. 5221 2. 528.8		1. 5226 2. 538.8
	1. 5222 2. 528.8		1. 5227 2. 538.9
	1. 5223 2. 533.3		1. 5228 2. 538.9
	1. 5224 2. 533.3		1. 5229 2. 538.9
	1. 5225 2. 537.9		1. 5230 2. 540.9

Product	1. Ex. 2. m/z
	1. 5231 2. 542.9
	1. 5232 2. 544.9
	1. 5233 2. 548.9
	1. 5234 2. 552.9
	1. 5235 2. 566.8

Product	1. Ex. 2. m/z
	1. 5236 2. 566.8
	1. 5237 2. 567.7
	1. 5238 2. 567.7
	1. 5239 2. 574.9
	1. 5240 2. 581

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5241 2. 533.3		1. 5246 2. 548.9
	1. 5242 2. 537.9		1. 5247 2. 554.9
	1. 5243 2. 542.8		1. 5248 2. 555.9
	1. 5244 2. 544.9		1. 5249 2. 555.9
	1. 5245 2. 544.9		1. 5250 2. 574.9

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5251 2. 579.9		1. 5256 2. 501.8
	1. 5252 2. 579.9		1. 5257 2. 514.8
	1. 5253 2. 581		1. 5258 2. 518.9
	1. 5254 2. 588.9		1. 5259 2. 530.9
	1. 5255 2. 590.9		1. 5260 2. 530.9

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5261 2. 538.9		1. 5266 2. 556.9
	1. 5262 2. 542.9		1. 5267 2. 556.9
	1. 5263 2. 547.3		1. 5268 2. 537.9
	1. 5264 2. 551.9		1. 5269 2. 566.9
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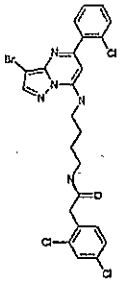
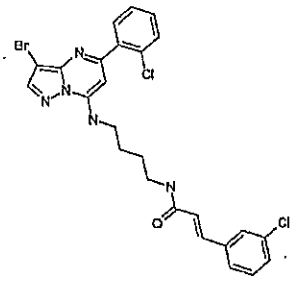
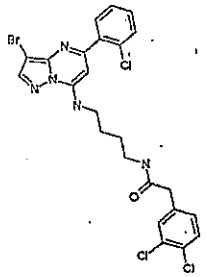
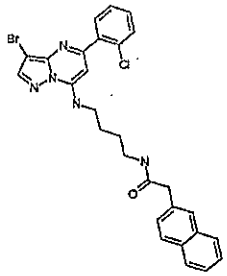
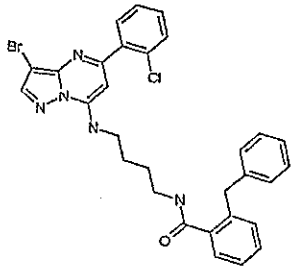
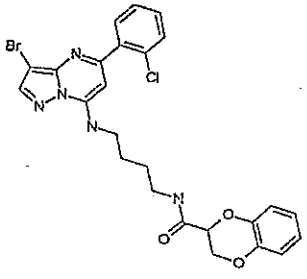
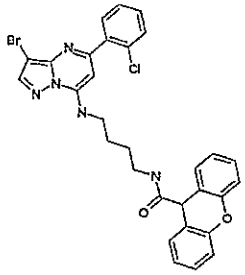
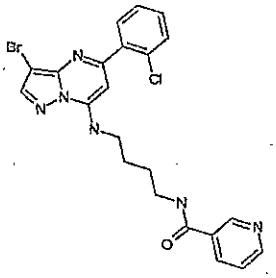
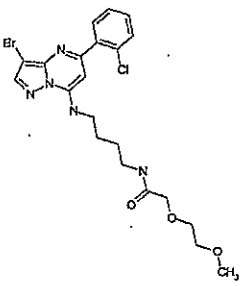
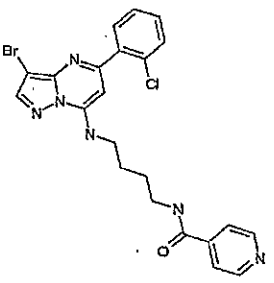
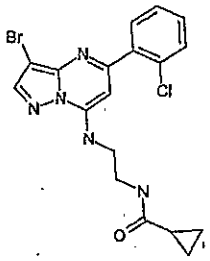
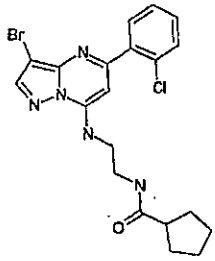
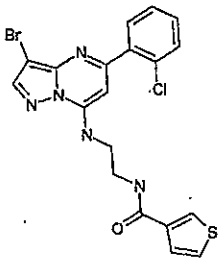
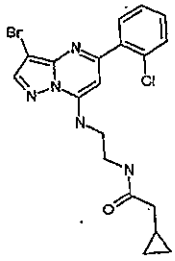
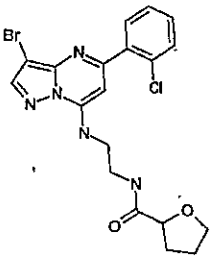
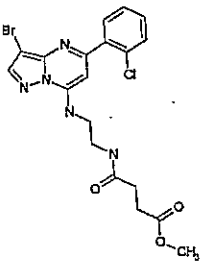
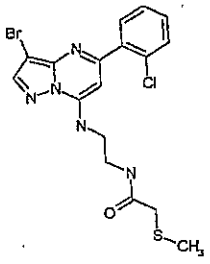
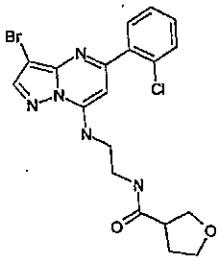
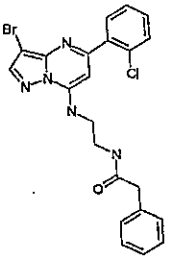
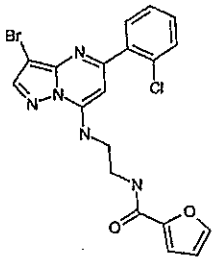
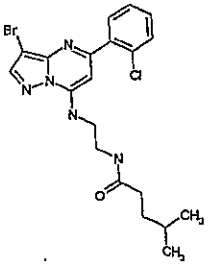
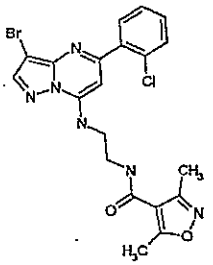
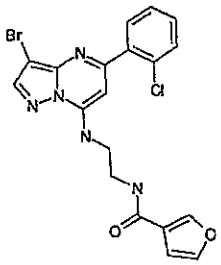
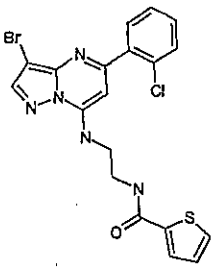
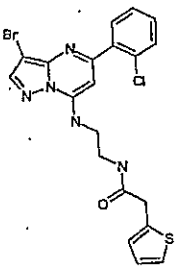
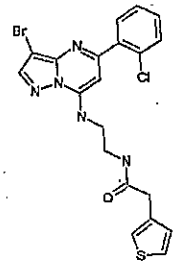
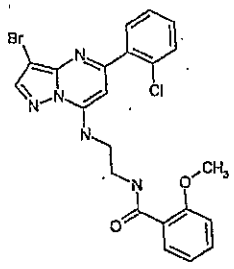
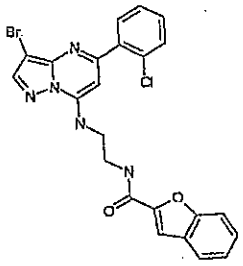
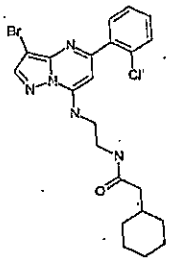
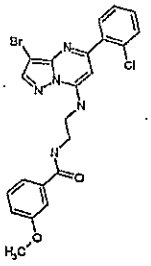
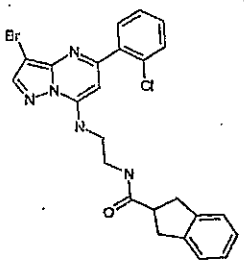
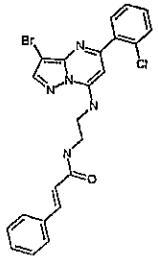
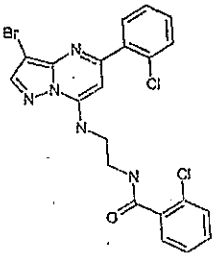
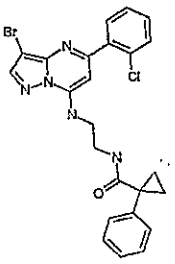
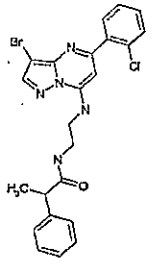
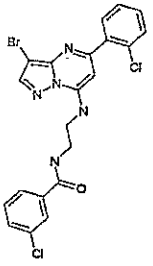
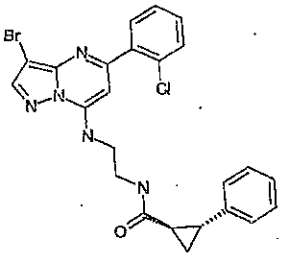
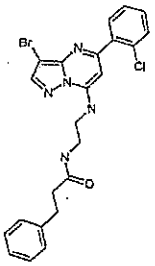
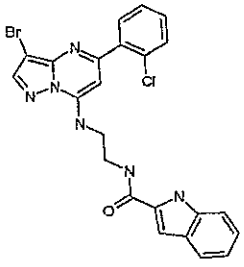
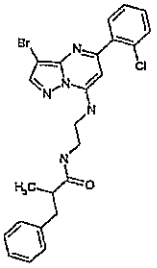
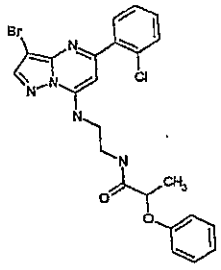
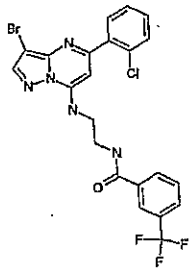
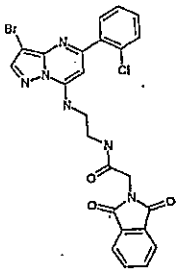
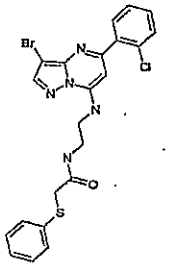
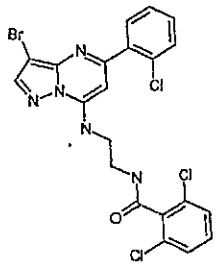
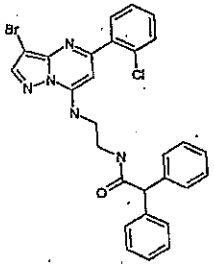
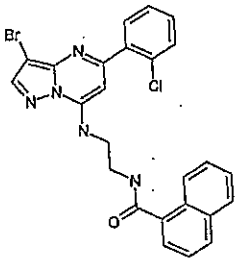
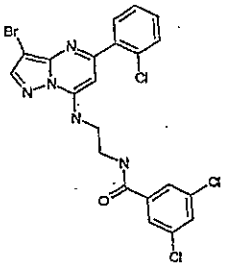
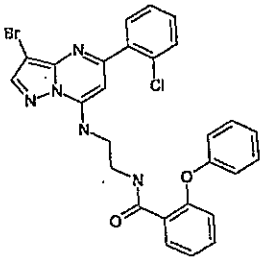
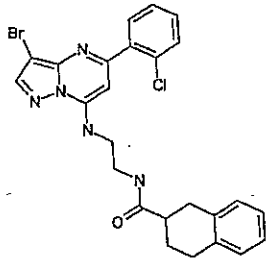
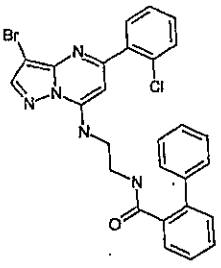
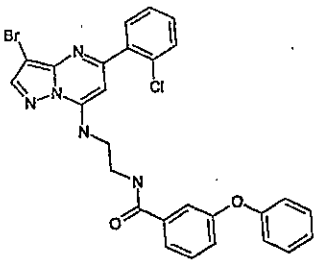
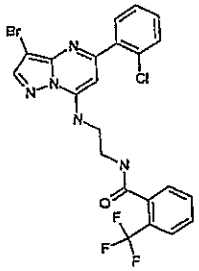
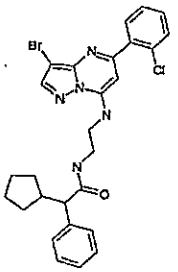
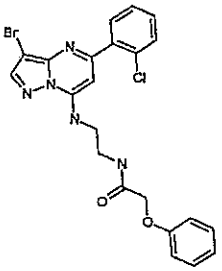
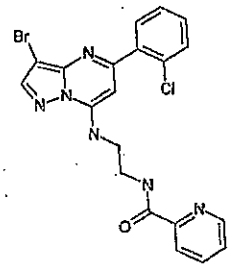
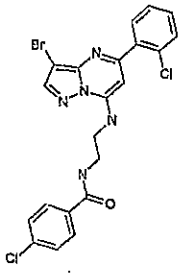
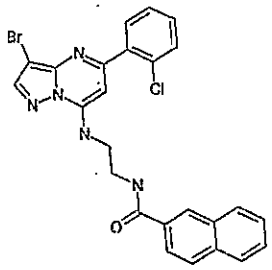
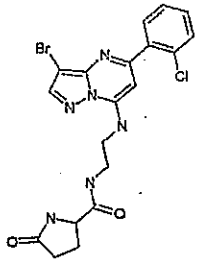
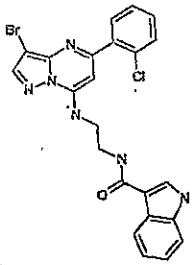
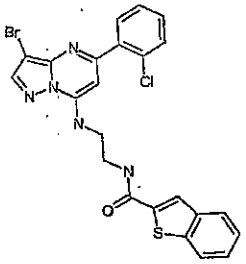
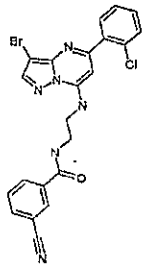
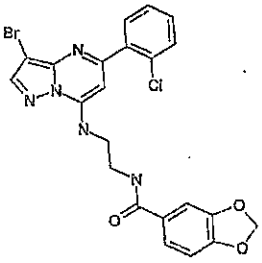
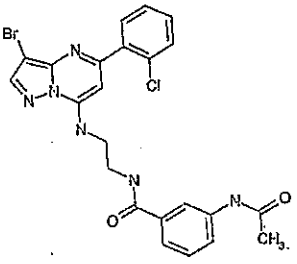
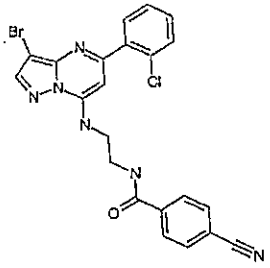
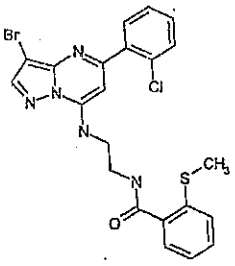
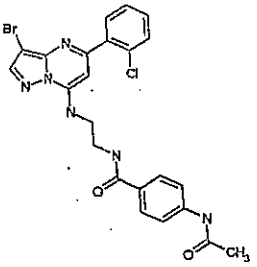
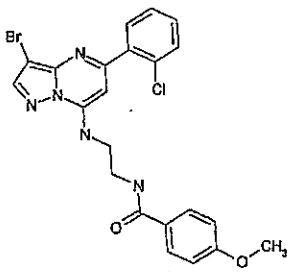
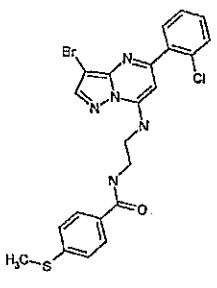
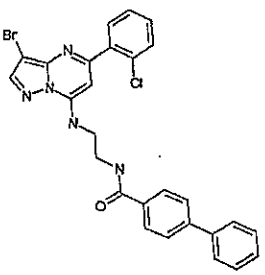
Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5271 2. 581.7		1. 5276 2. 559.3
	1. 5272 2. 581.7		1. 5277 2. 562.9
	1. 5273 2. 588.9		1. 5278 2. 556.9
	1. 5274 2. 602.9		1. 5279 2. 499.8
	1. 5275 2. 510.8		1. 5280 2. 499.8

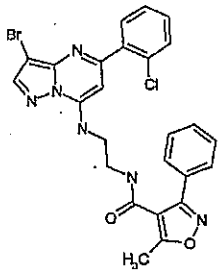
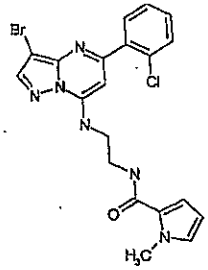
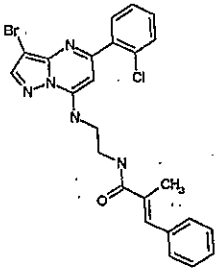
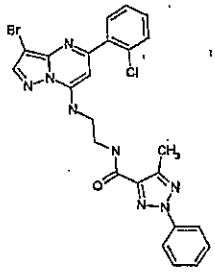
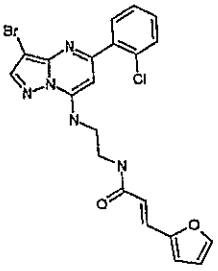
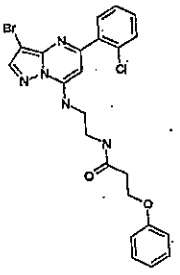
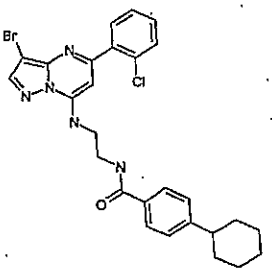
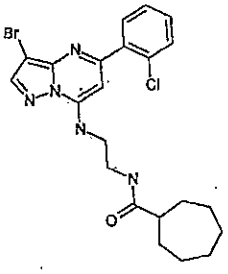
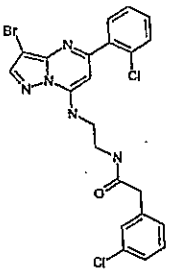
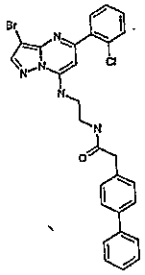
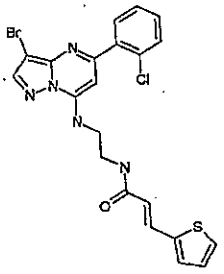
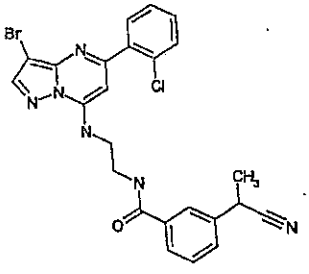
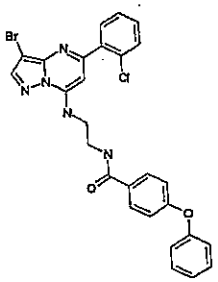
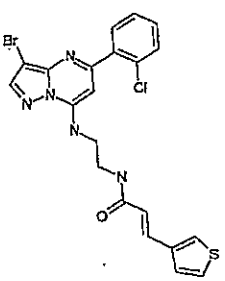
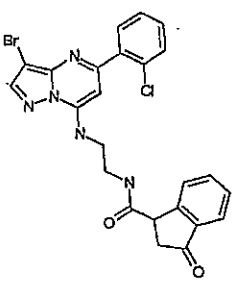
TABLE 53

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5301 2. 434.7		1. 5306 2. 462.8		1. 5311 2. 476.8
	1. 5302 2. 448.8		1. 5307 2. 464.8		1. 5312 2. 480.8
	1. 5303 2. 454.8		1. 5308 2. 464.8		1. 5313 2. 484.8
	1. 5304 2. 460.7		1. 5309 2. 464.8		1. 5314 2. 489.8
	1. 5305 2. 460.7		1. 5310 2. 476.8		1. 5315 2. 490.8

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5316 2. 490.8		1. 5321 2. 500.8		1. 5326 2. 510.8
	1. 5317 2. 490.8		1. 5322 2. 500.8		1. 5327 2. 510.8
	1. 5318 2. 496.8		1. 5323 2. 505.2		1. 5328 2. 510.8
	1. 5319 2. 498.8		1. 5324 2. 505.2		1. 5329 2. 510.8
	1. 5320 2. 498.8		1. 5325 2. 509.8		1. 5330 2. 512.8

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5331 2. 514.8		1. 5336 2. 538.8		1. 5341 2 553.8
	1. 5332 2. 516.9		1. 5337 2. 539.6		1. 5342 2 560.9
	1. 5333 2. 520.8		1. 5338 2. 539.6		1. 5343 2. 562.9
	1. 5334 2. 524.9		1. 5339 2. 546.9		1. 5344 2. 562.9
	1. 5335 2. 538.8		1.5340 2. 552.9		1. 5345 2. 500.8

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5346 2. 471.7		1. 5351 2. 505.2		1. 5356 2. 520.8
	1. 5347 2. 477.8		1. 5352 2. 509.8		1. 5357 2. 526.8
	1. 5348 2. 495.8		1. 5353 2. 514.8		1. 5358 2. 527.8
	1. 5349 2. 495.8		1. 5355 2. 516.9		1. 5359 2. 527.8
	5350 2. 500.8		1. 5356 2. 516.9		1. 5360 2. 546.9

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5361 2. 551.8		1. 5366 2. 473.8		1. 5371 2.510.8
	1. 5362 2. 551.8		1. 5367 2. 486.8		1. 5372 2 514.8
	1. 5363 2. 552.9		1. 5368 2. 490.8		1. 5373 2 519.2
	1. 5364 2. 560.9		1. 5369 2. 502.8		1. 5374 2 523.8
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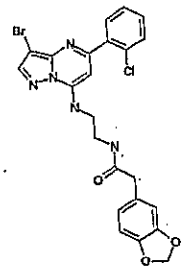
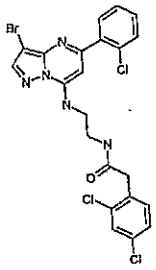
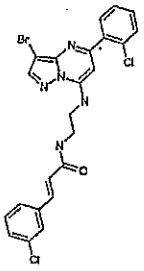
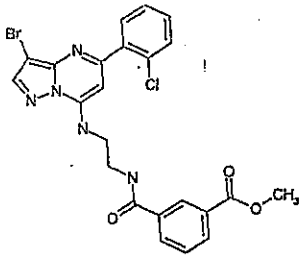
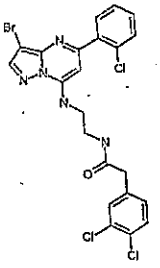
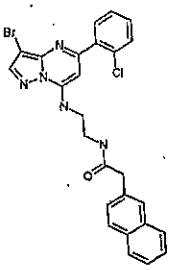
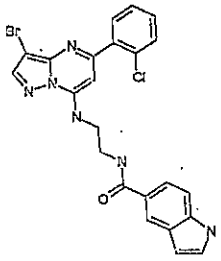
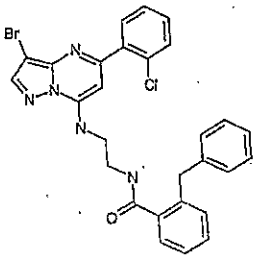
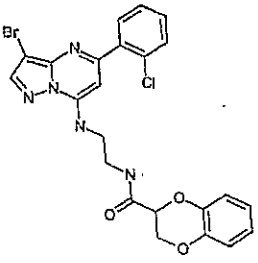
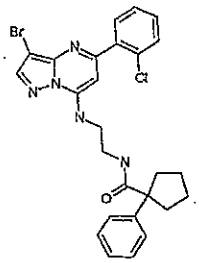
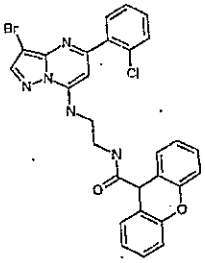
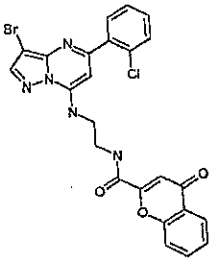
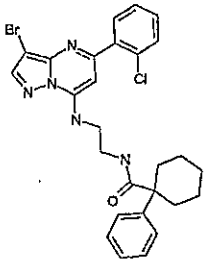
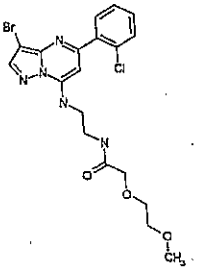
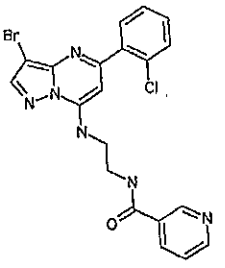
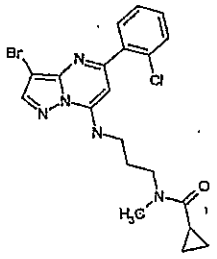
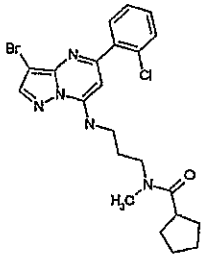
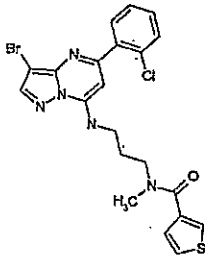
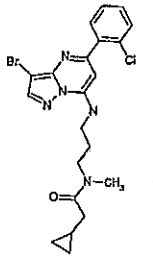
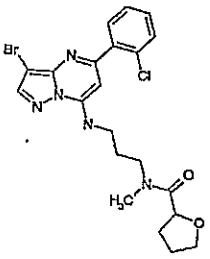
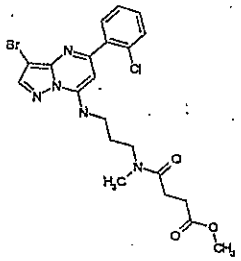
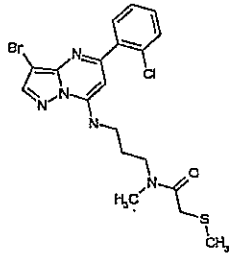
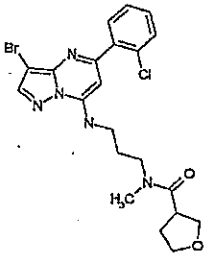
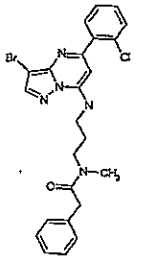
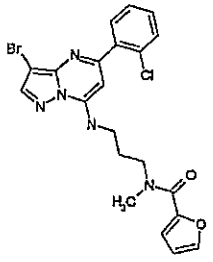
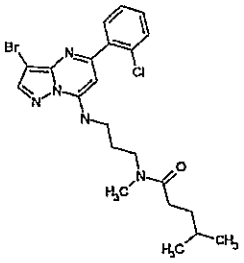
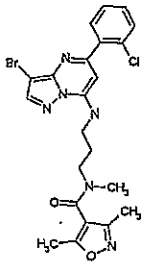
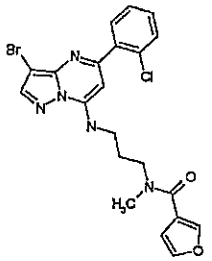
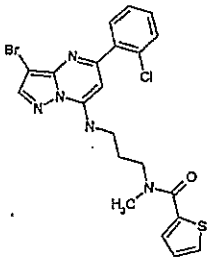
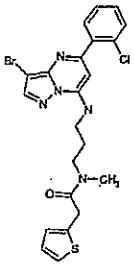
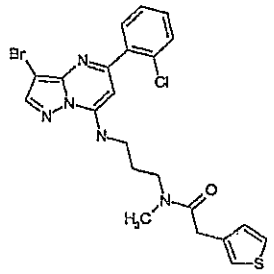
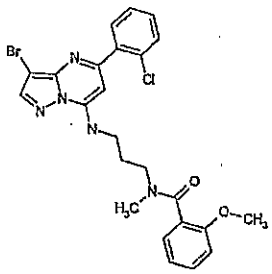
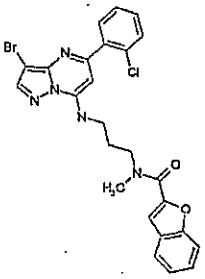
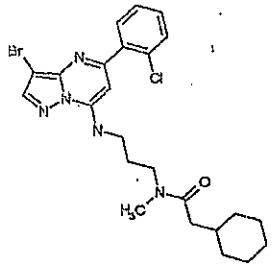
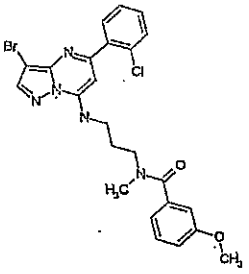
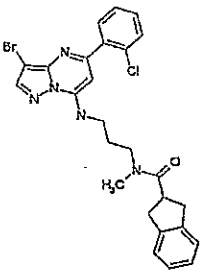
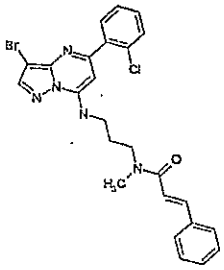
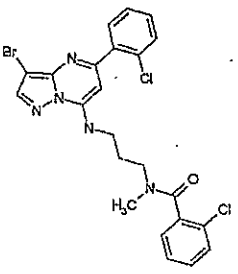
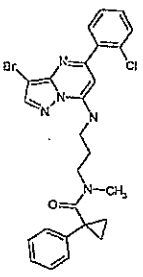
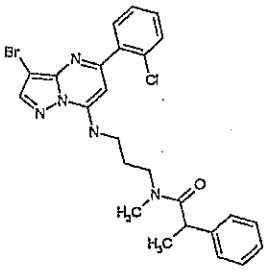
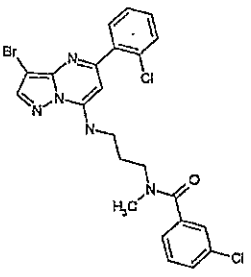
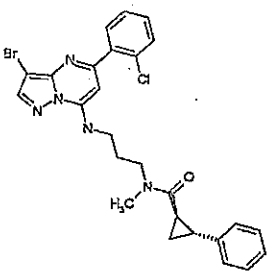
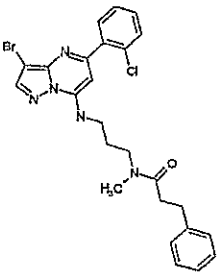
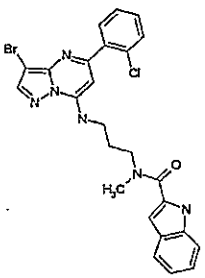
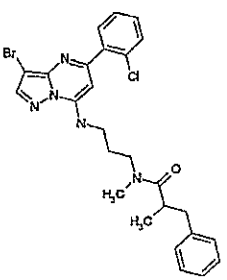
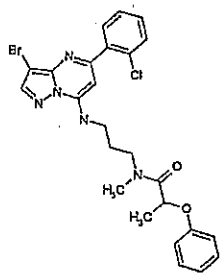
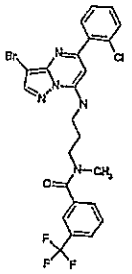
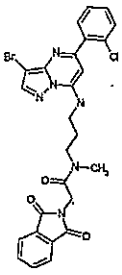
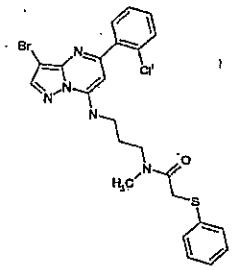
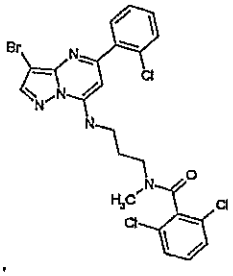
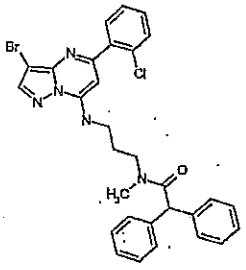
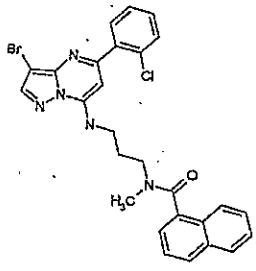
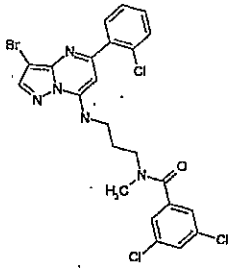
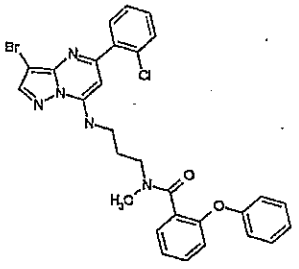
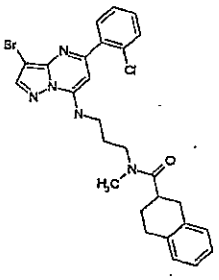
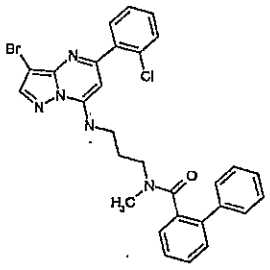
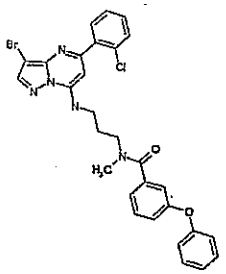
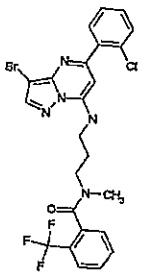
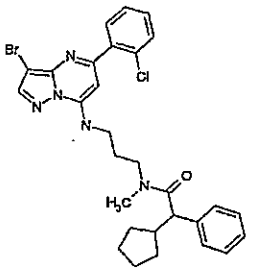
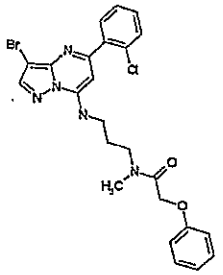
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	1. 5377 2. 528.8		1. 5382 2. 553.7		1. 5387 2 534.8
	1. 5378 2. 509.8		1. 5383 2. 560.9		1. 5388 2 528.8
	1. 5379 2. 538.9		1. 5384 2. 574.9		1. 5389 2. 538.8
	1. 5380 2. 552.9		1. 5385 2. 482.8		1. 5390 2. 471.7

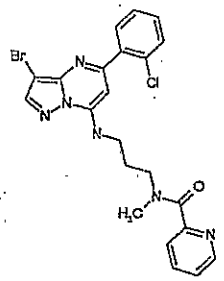
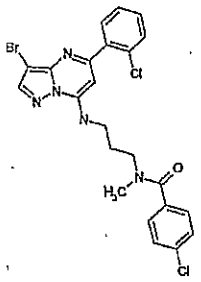
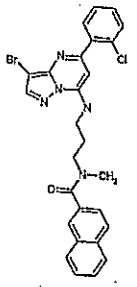
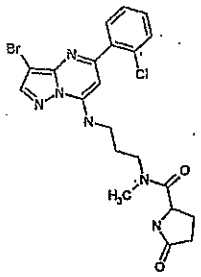
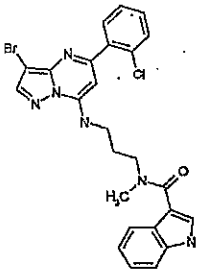
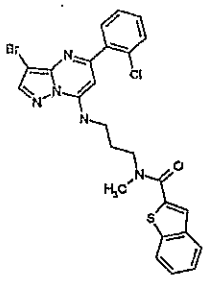
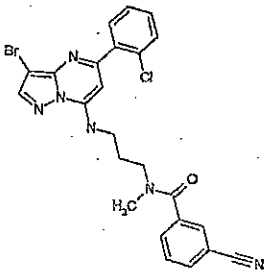
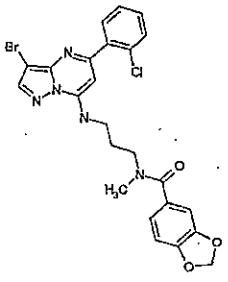
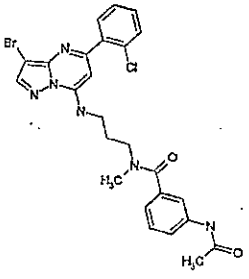
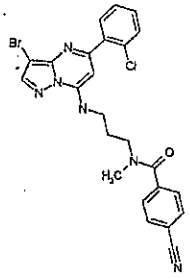
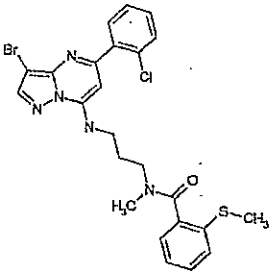
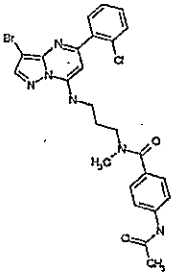
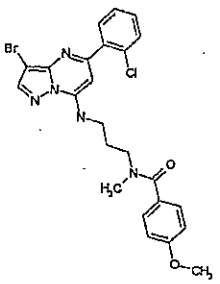
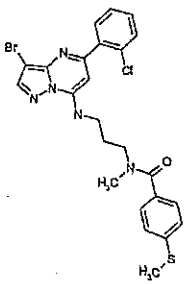
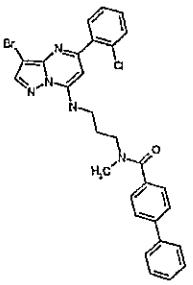
TABLE 54

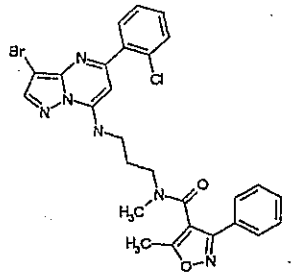
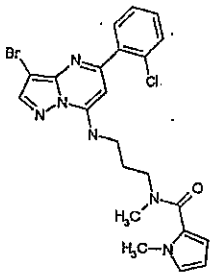
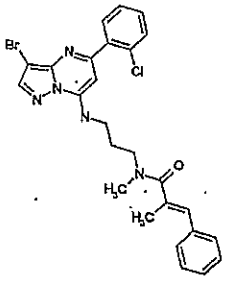
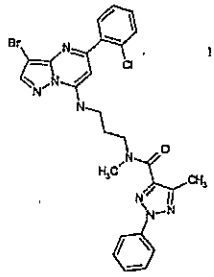
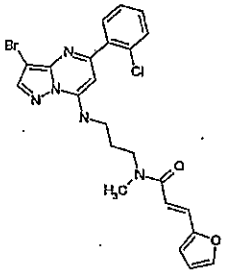
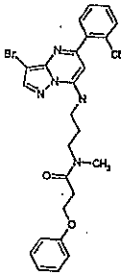
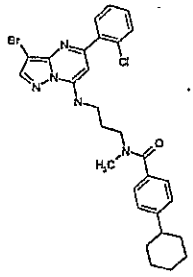
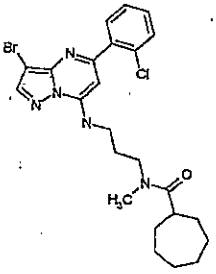
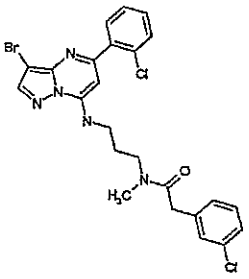
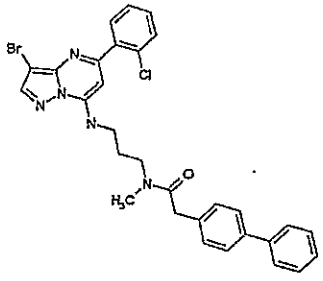
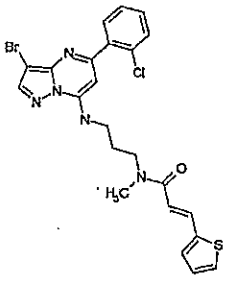
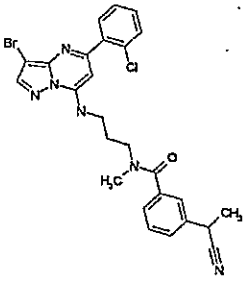
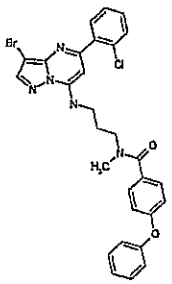
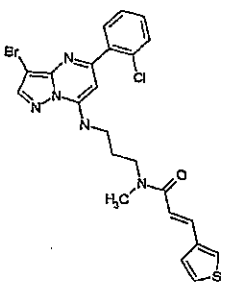
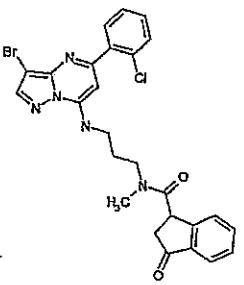
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	1. 5401 2. 463.3		1. 5406 2. 491.3		1. 5411 2. 505.3
	1. 5402 2. 477.3		1. 5407 2. 493.3		1. 5412 2. 509.3
	1. 5403 2. 483.3		1. 5408 2. 493.3		1. 5413 2. 513.3
	1. 5404 2. 489.3		1. 5409 2. 493.3		1. 5414 2. 518.3
	1. 5405 2. 489.3		1. 5410 2. 505.3		1. 5415 2. 519.3

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Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5416 2. 519.3		1. 5121 2. 529.3		1. 5426 2. 539.3
	1. 5417 2. 519.3		1. 5422 2. 529.3		1. 5427 2. 539.3
	1. 5418 2. 525.3		1. 5423 2. 533.3		1. 5428 2. 539.3
	1. 5419 2. 527.3		1. 5424 2. 533.3		1. 5429 2. 539.3
	1. 5420 2. 527.3		1. 5425 2. 538.3		1. 5430 2. 541.3

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2 m/z
	1. 5431 2. 543.3		1. 5436 2. 567.3		1. 5441 2 582.3
	1. 5432 2. 545.3		1. 5437 2. 567.3		1. 5442 2 589.3
	1. 5433 2. 549.3		1. 5438 2. 567.3		1. 5443 2 591.3
	1. 5434 2. 553.3		1. 5439 2. 575.3		1. 5444 2. 591.3
	1. 5435 2. 567.3		1. 5440 2. 581.3		1. 5445 2. 529.3

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5446 2. 500.3		1. 5451 2. 533.3		1. 5456 2. 549.3
	1. 5447 2. 830.5		1. 54525 2. 719.4		1. 5457 2. 555.3
	1. 5448 2. 524.3		1. 5453 2. 543.3		1. 5458 2. 556.3
	1. 5449 2. 524.3		1. 5454 2. 545.3		1.5459 2. 556.3
	1. 5450 2. 529.3		1. 5455 2. 545.3		1. 5460 2. 575.3

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5461 2. 580.3		1. 5466 2. 502.3		1. 5471 2 539.3
	1. 5462 2. 580.3		1. 5467 2. 515.3		1. 5472 2 543.3
	1. 5463 2. 581.3		1. 5468 2. 519.3		1. 5473 2 547.3
	1. 5464 2. 589.3		1. 5469 2. 531.3		1. 5474 2. 552.3
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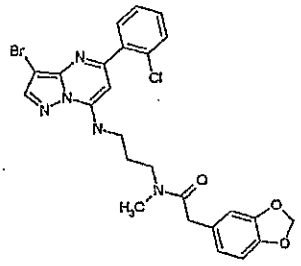
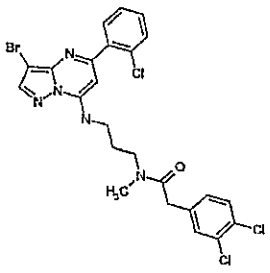
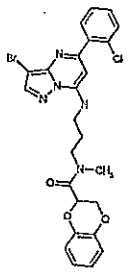
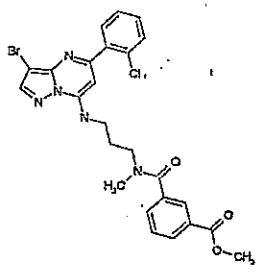
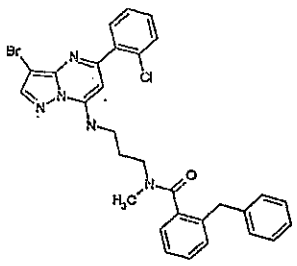
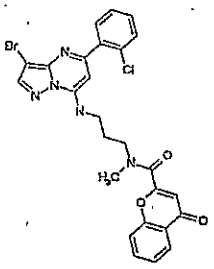
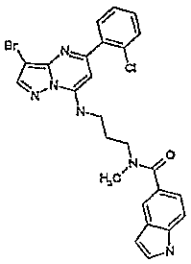
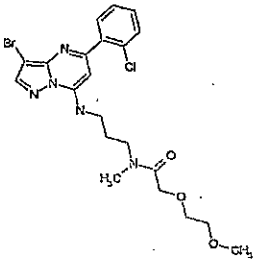
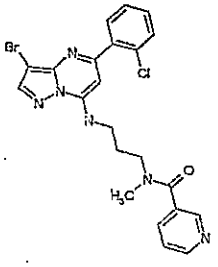
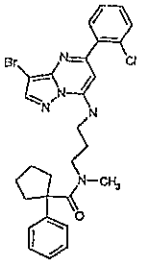
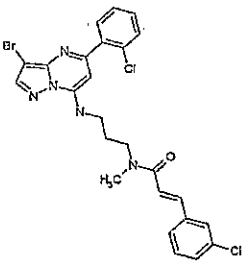
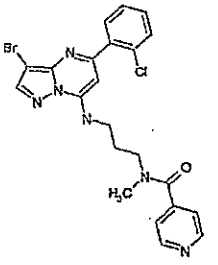
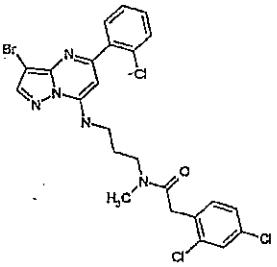
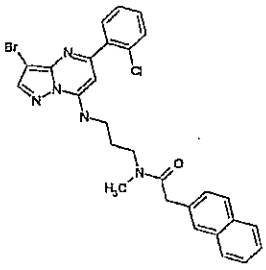
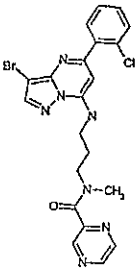
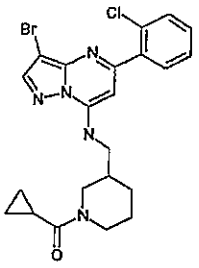
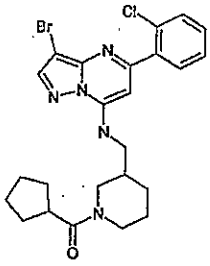
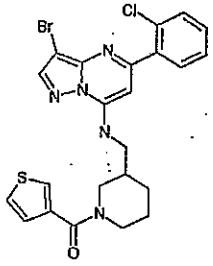
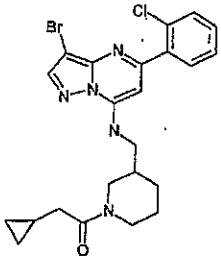
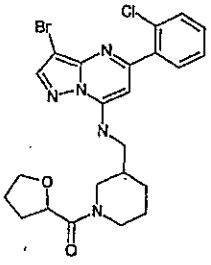
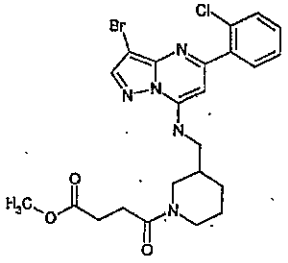
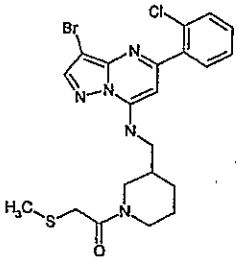
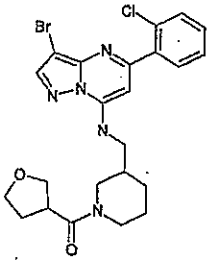
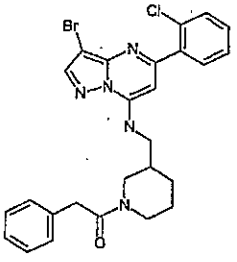
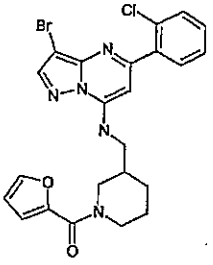
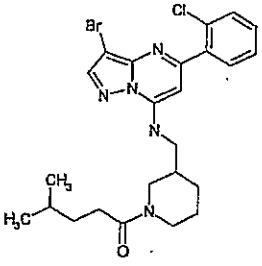
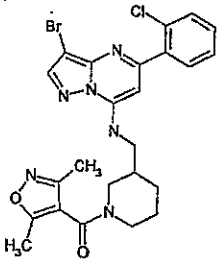
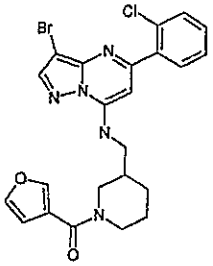
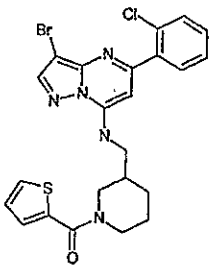
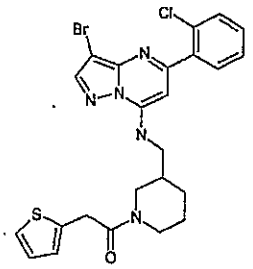
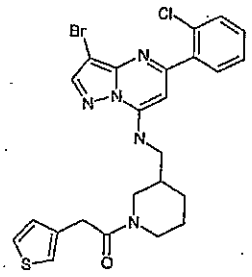
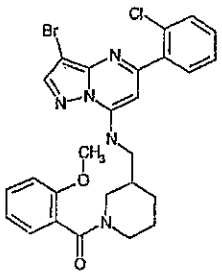
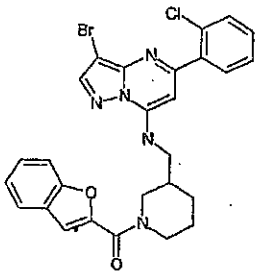
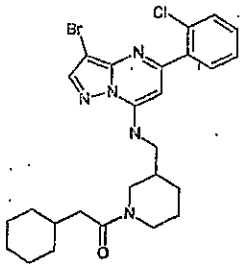
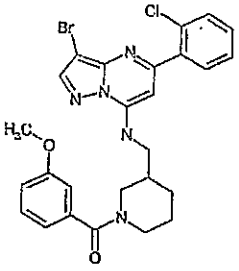
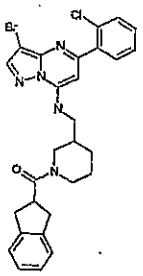
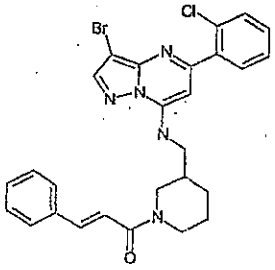
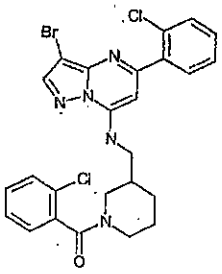
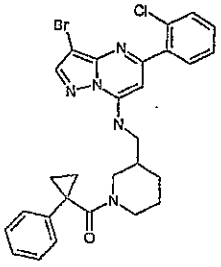
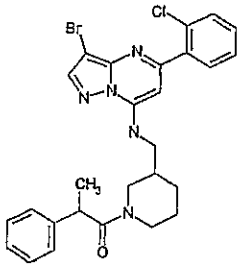
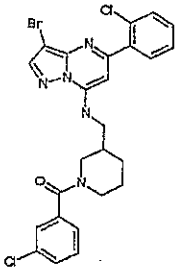
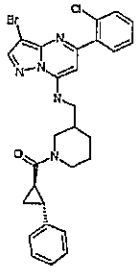
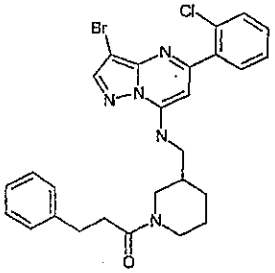
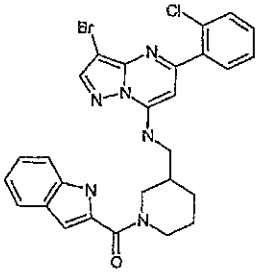
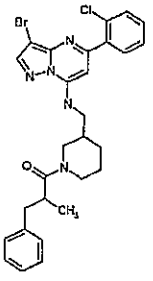
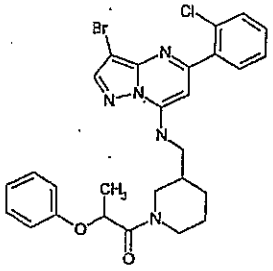
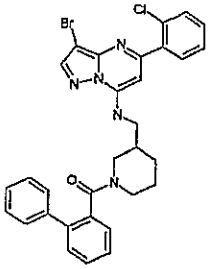
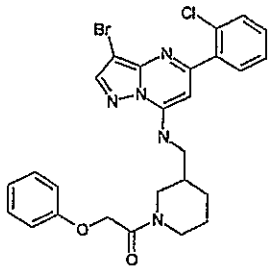
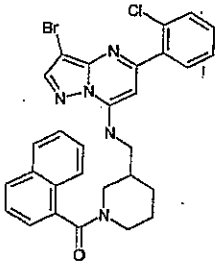
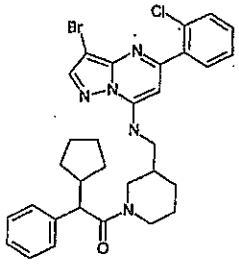
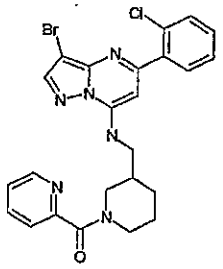
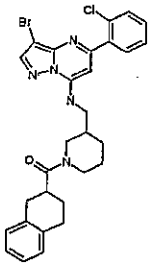
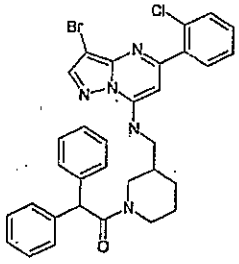
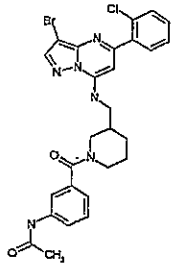
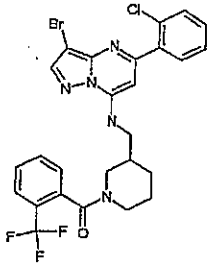
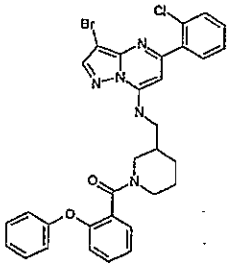
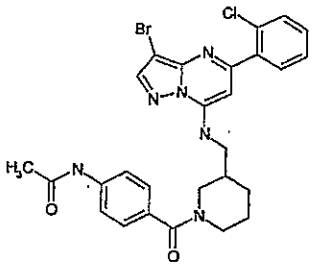
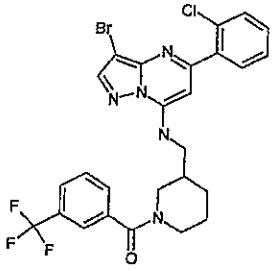
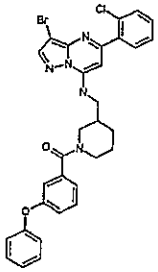
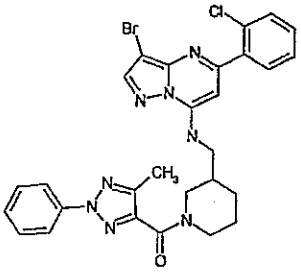
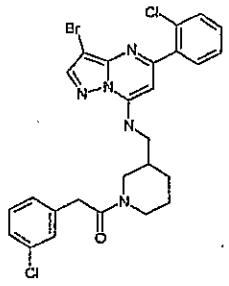
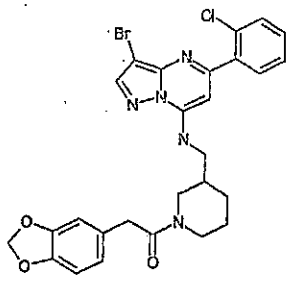
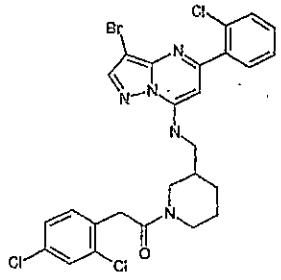
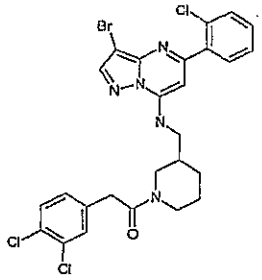
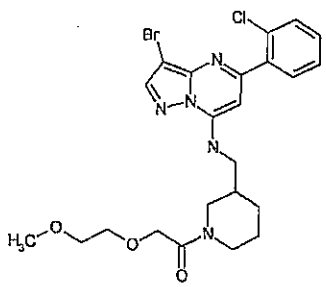
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	1. 5476 2. 557.3		1. 5481 2. 581.3		. 5486 2.
	1. 5477 2. 557.3		1. 5482 2. 589.3		1. 5487 2. 891.5
	1. 5478 2. 538.3		1. 5483 2. 511.3		1. 5488 2. 500.3
	1. 5479 2. 567.3		1. 5484 2. 559.3		1. 5489 2. 500.3
	1. 5480 2. 581.3		1. 5485 2. 563.3		1. 5490 2. 501.3

TABLE55

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5501 2. 488.81		1. 5506 2. 516.87		1. 5511 2. 530.87
	1. 5502 2. 502.84		1. 5507 2. 518.84		1. 5512 2. 534.84
	1. 5503 2. 508.87		1. 5508 2. 518.84		1. 5513 2. 538.87
	1. 5504 2. 514.81		1. 5509 2. 518.88		1. 5514 2. 543.85
	1. 5505 2. 514.81		1. 5510 2. 530.87		1. 5515 2. 544.9

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5516 2. 544.9		1. 5521 2. 554.87		1. 5526 2. 564.87
	1. 5517 2. 544.92		1. 5522 2. 554.87		1. 5527 2. 564.91
	1. 5518 2. 550.89		1. 5523 2. 559.29		1. 5528 2. 564.91
	1. 5519 2. 552.9		1. 5524 2. 559.29		1. 5529 2. 564.91
	1. 5520 2. 552.9		1. 5525 2. 563.88		1. 5530 2. 566.93

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5531 2. 568.9		1. 5536 2. 600.95		1. 5541
	1. 5532 2. 574.91		1. 5537 2. 606.99		1. 5542 2 525.83
	1. 5533 2. 578.94		1. 5538 2. 614.97		1. 5543 2. 581.9
	1. 5534 2. 592.85		1. 5539 2. 616.95		1. 5544 2. 581.9
	1. 5535 2. 592.85		1. 5540 2. 616.95		1. 5545 2. 605.92

Product	1. Ex. 2. m/z
	1. 5546 2. 573.32
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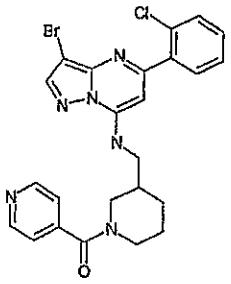
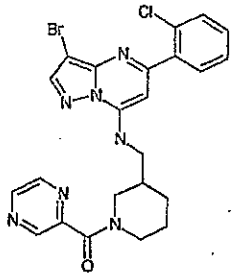
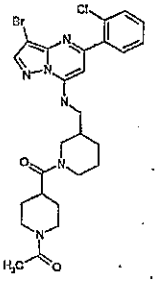
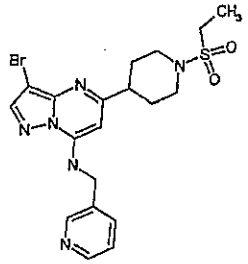
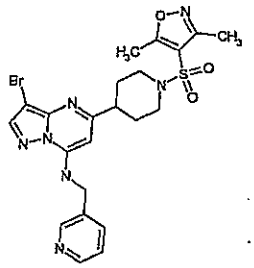
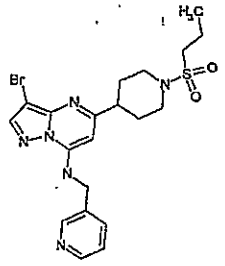
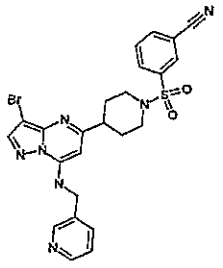
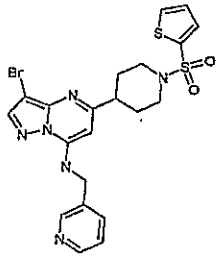
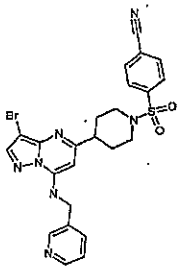
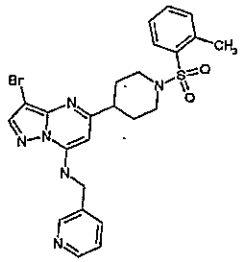
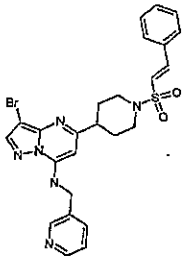
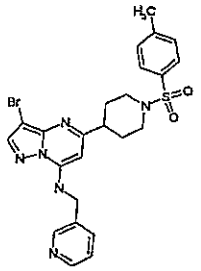
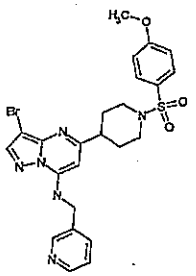
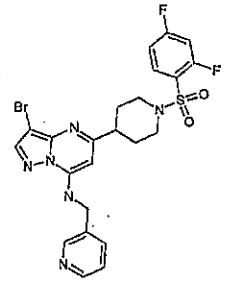
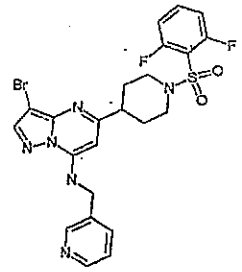
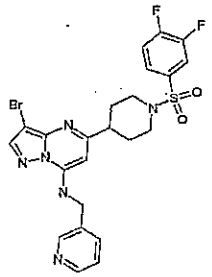
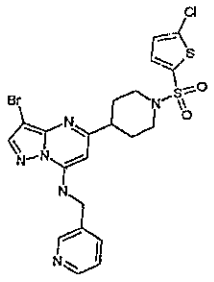
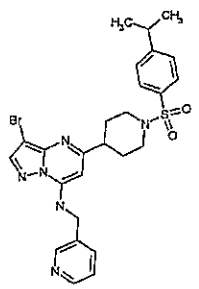
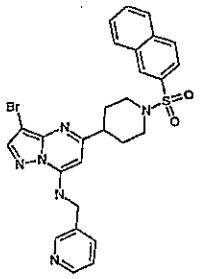
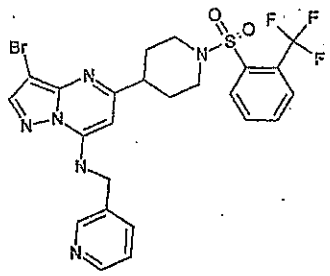
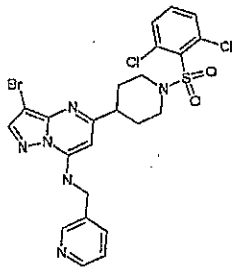
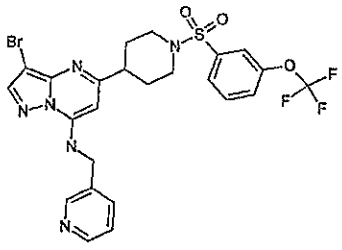
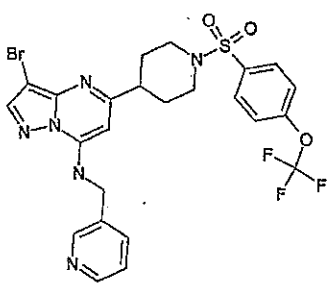
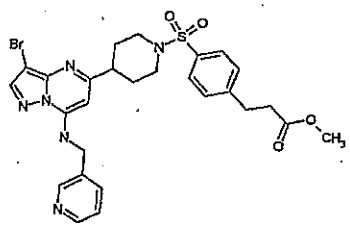
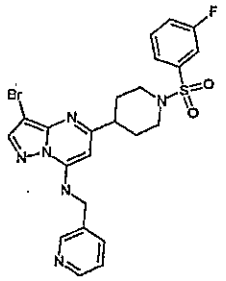
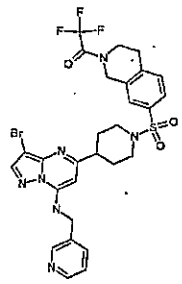
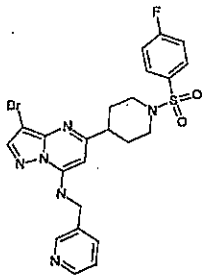
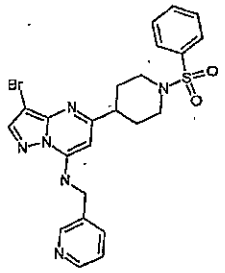
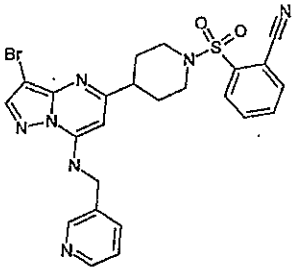
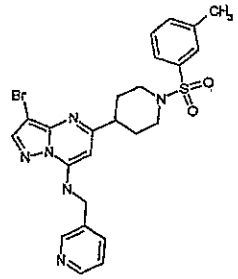
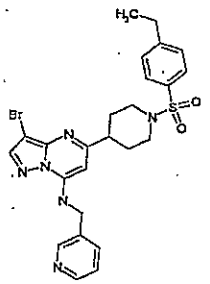
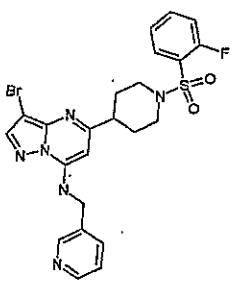
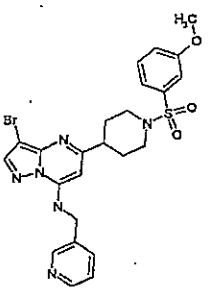
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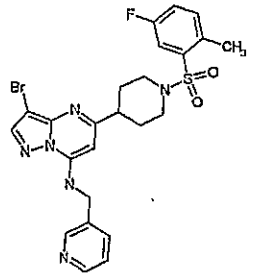
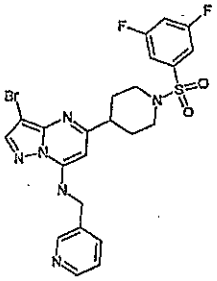
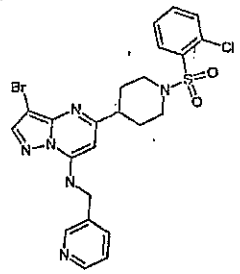
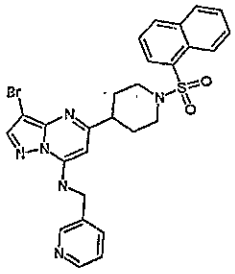
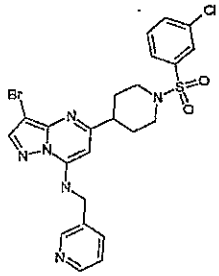
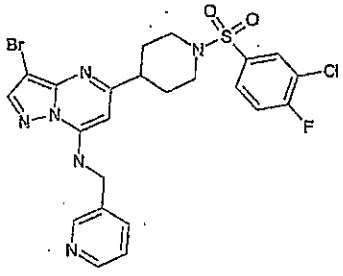
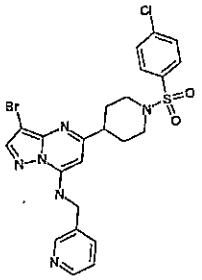
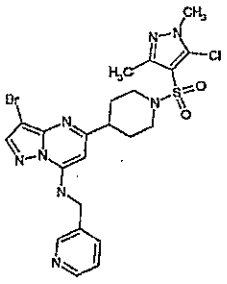
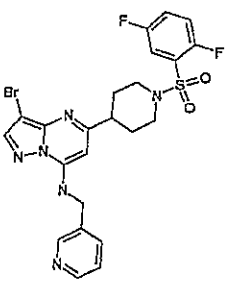
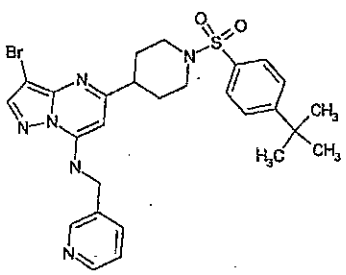
TABLE 56

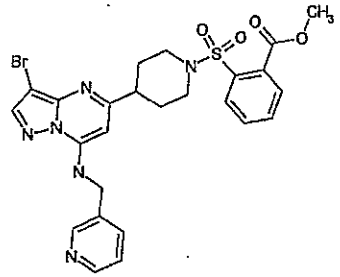
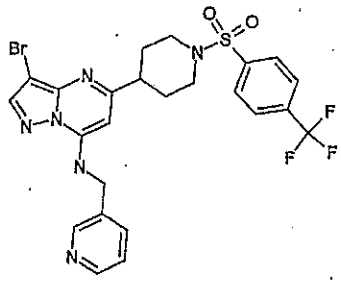
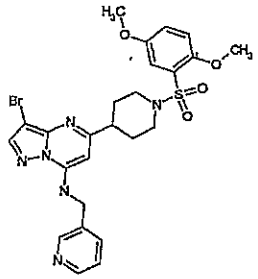
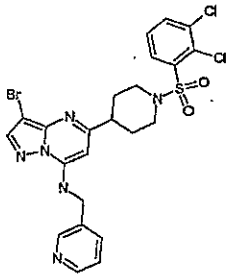
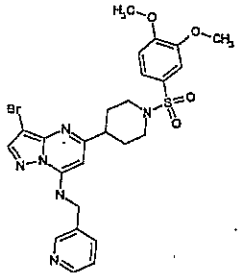
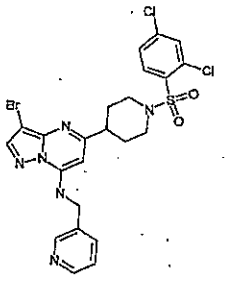
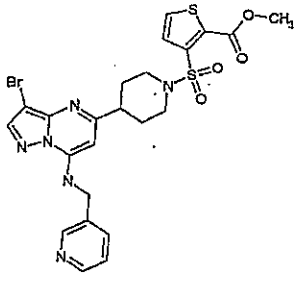
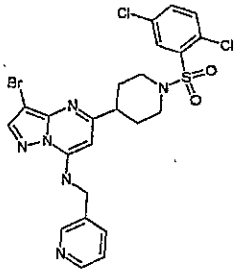
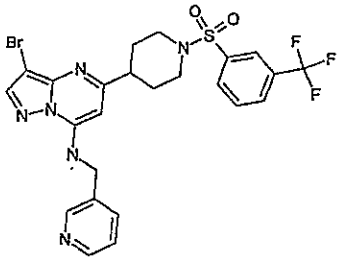
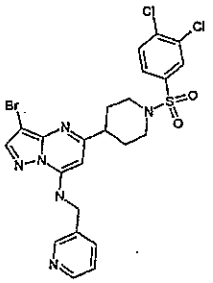
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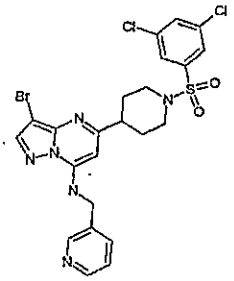
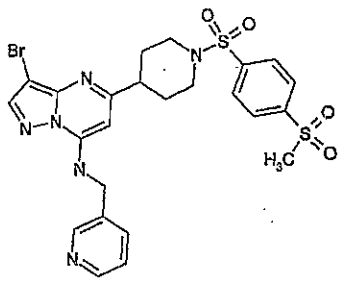
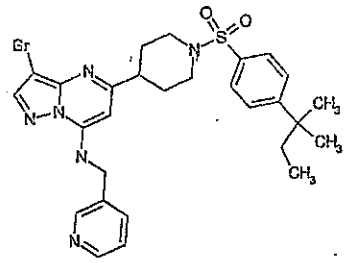
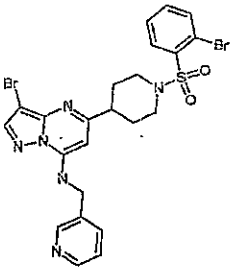
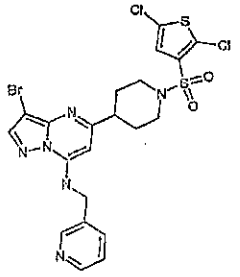
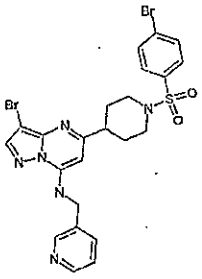
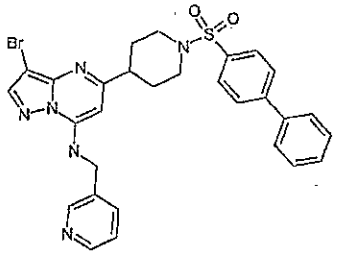
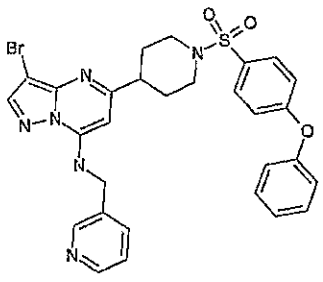
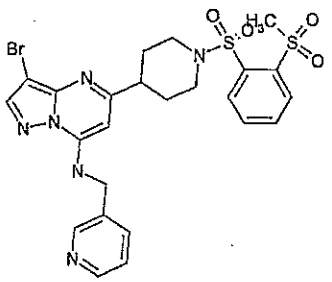
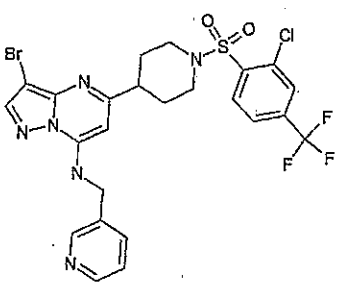
Product	1. Ex. 2. m/z
	1. 5611 2. 565.31
	1. 5612 2. 565.31
	1. 5613 2. 565.31
	1. 5614 2. 569.31
	1. 5615 2. 571.31

Product	1. Ex. 2. m/z
	1 .5616 2. 579.32
	1. 5617
	1. 5618
	1. 5619
	1. 5620

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5621 2. 615.34		1. 5626 2. 547.3
	2. 5622 2. 680.37		1. 5627 2. 547.3
	1. 5623 2. 529.29		1. 5628 2. 554.3
	1. 5624 2. 543.3		1. 5629 2. 555.31
	1. 5625 2. 547.3		1. 5630 2. 559.31

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5631 2. 561.31		1. 5636 2. 565.31
	1. 5632 2. 563.31		1. 5637 2. 579.32
	1. 5633 2. 563.31		1. 5638 2. 581.32
	1. 5634 2. 563.31		1. 5639 2. 581.32
	1. 5635 2. 563.31		1. 5640 2. 585.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5641 2. 587.32		1. 5646
	1. 5642 2. 589.32		1. 5647 2. 597.33
	1. 5643 2. 589.32		1. 5648 2. 597.33
	1. 5644 2. 593.33		1. 5649 2. 597.33
	1. 5645 2. 597.33		1. 5650 2. 597.33

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5651 2. 597.33		1. 5656 2. 607.33
	1. 5652 2. 599.33		1. 5657 2. 607.33
	1. 5653 2. 603.33		1. 5658 2. 607.33
	1. 5654 2. 605.33		1. 5659 2. 621.34
	1. 5655 2. 607.33		1. 5660 2. 631.35

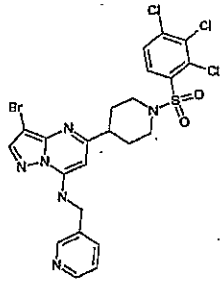
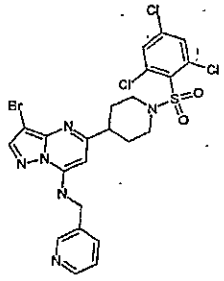
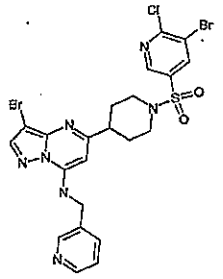
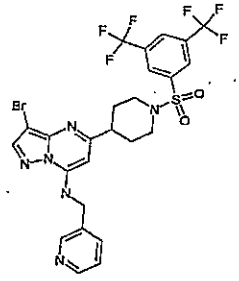
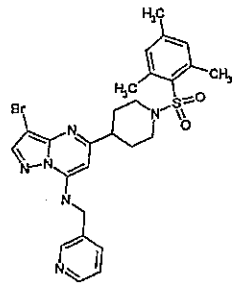
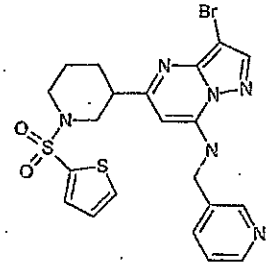
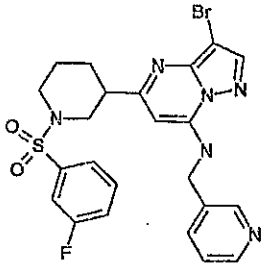
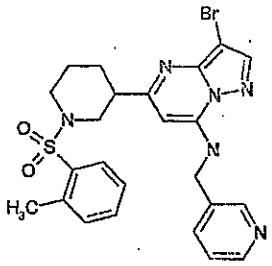
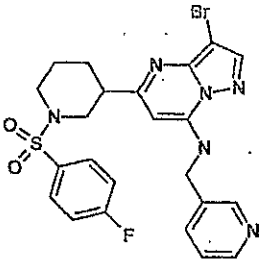
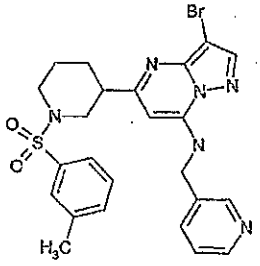
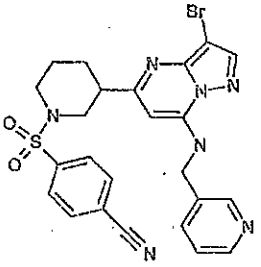
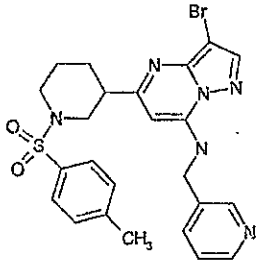
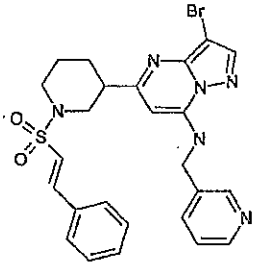
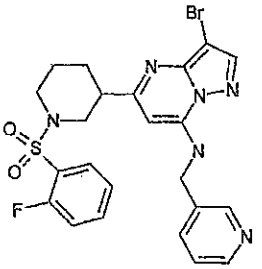
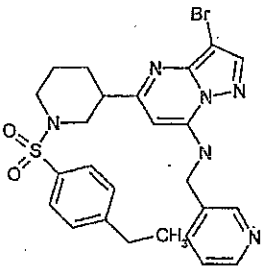
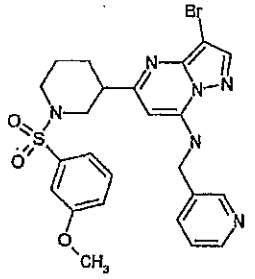
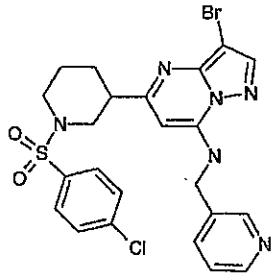
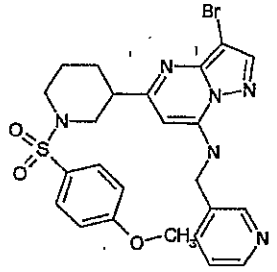
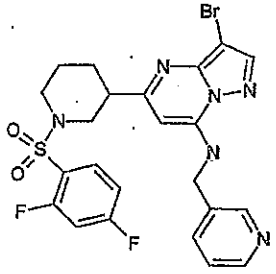
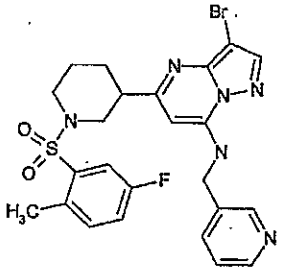
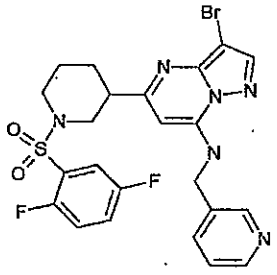
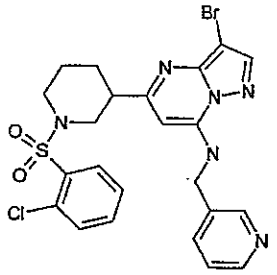
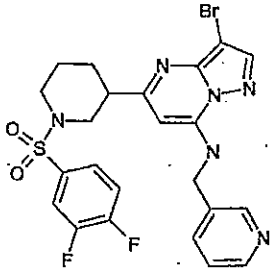
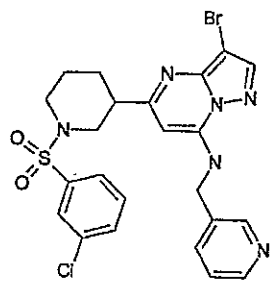
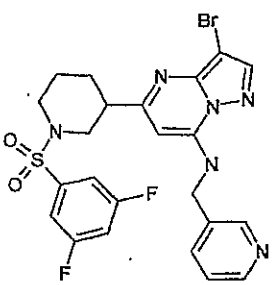
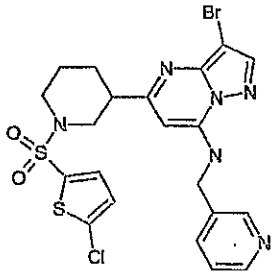
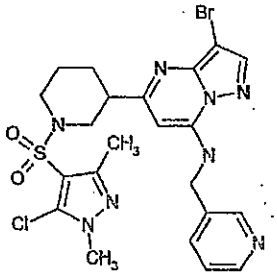
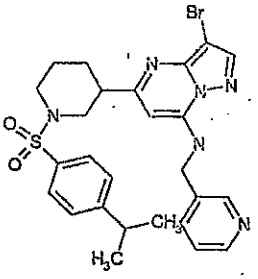
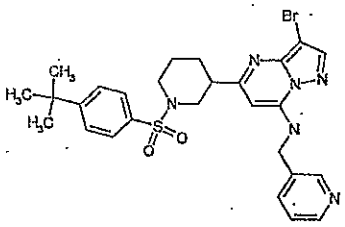
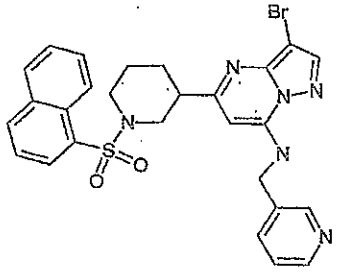
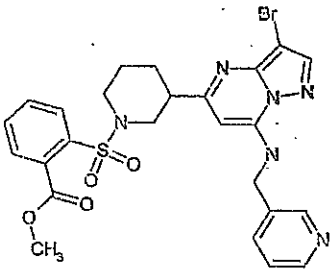
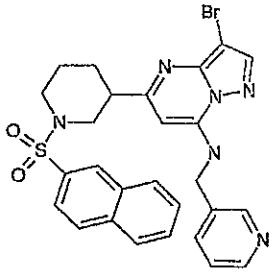
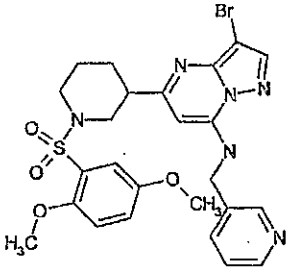
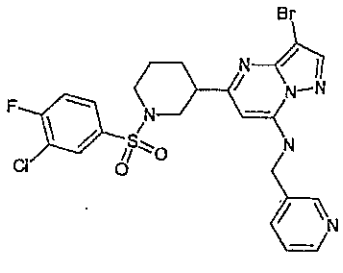
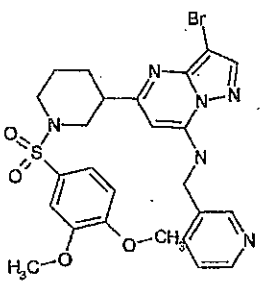
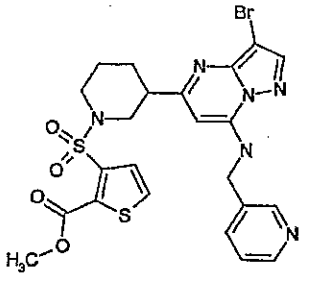
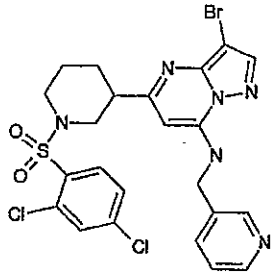
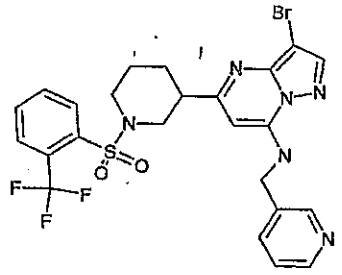
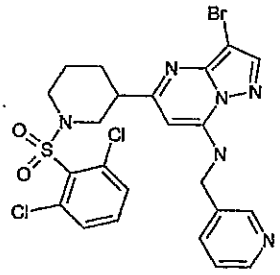
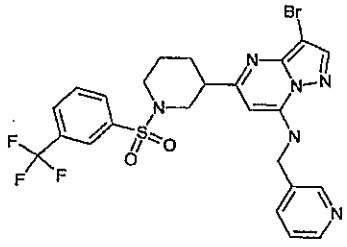
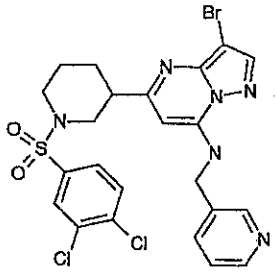
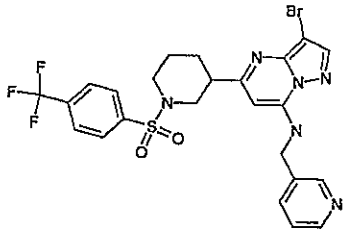
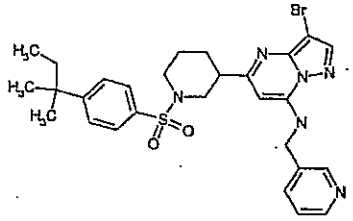
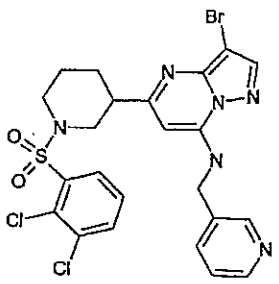
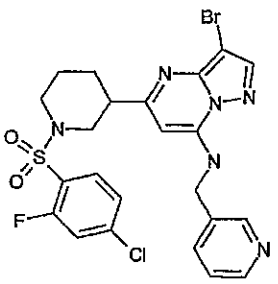
Product	1. Ex. 2. m/z
	1. 5661 2. 631.35
	1. 5662 2. 631.35
	1. 5663 2. 642.35
	1. 5664 2. 665.37
	1. 5665 2. 571.31

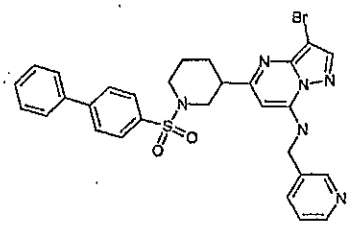
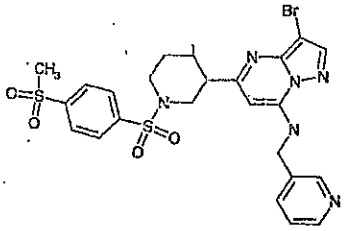
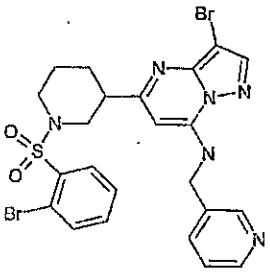
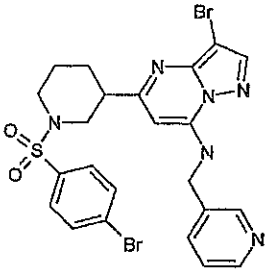
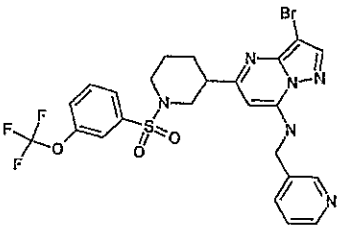
TABLE 57

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5701 2. 535.29		1. 5706 2. 547.3
	1. 5702 2. 540.3		1. 5707 2. 547.3
	1. 5703 2. 543.3		1. 5708 2. 554.3
	1. 5704 2. 543.3		1. 5709 2. 555.31
	1. 5705 2. 547.3		1. 5710 2. 555.31

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5711 2. 559.31		1. 5716 2. 563.31
	1. 5712 2. 559.31		1. 5717 2. 565.31
	1. 5713 2. 561.31		1. 5718 2. 565.31
	1. 5714 2. 562.31		1. 5719 2. 565.31
	1. 5715 2. 563.31		1. 5720 2. 565.31

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5721 2. 569.31		1. 5726 2. 581.32
	1. 5722 2. 571.31		1. 5727 2. 585.32
	1. 5723 2. 579.32		1. 5728 2. 587.32
	1. 5724 2. 579.32		1. 5729 2. 589.32
	1. 5725 2. 581.32		1. 5730 2. 589.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5731 2. 591.33		1. 5736 2. 597.33
	1. 5732 2. 597.33		1. 5737 2. 597.33
	1. 5733 2. 595.33		1. 5738 2. 596.33
	1. 5734 2. 597.33		1. 5739 2. 597.33
	1. 5735 2. 597.33		1. 5740 2. 581.32

Product	1. Ex. 2. m/z
	1. 5741 2. 605.33
	1. 5742 2. 607.33
	1. 5743 2. 607.33
	1. 5744 2. 607.33
	1. 57454 2. 613.34

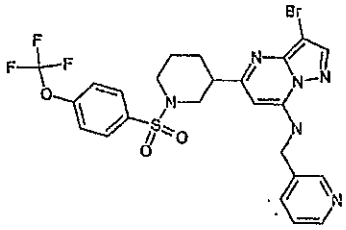
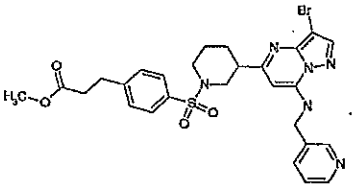
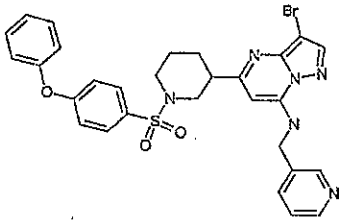
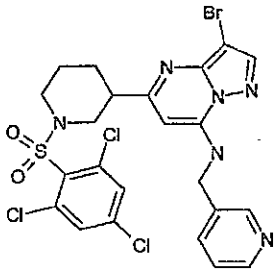
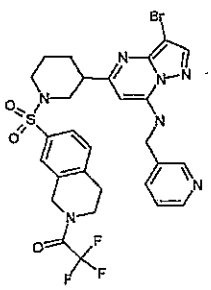
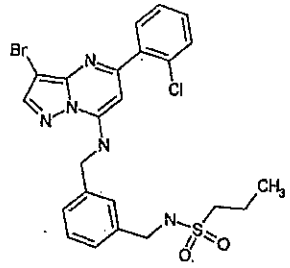
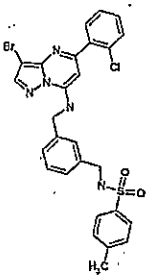
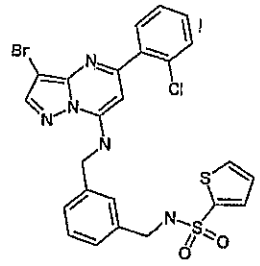
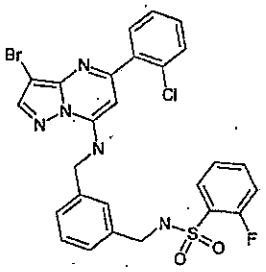
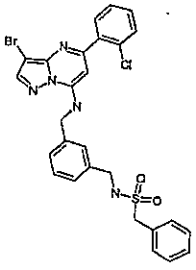
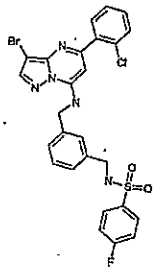
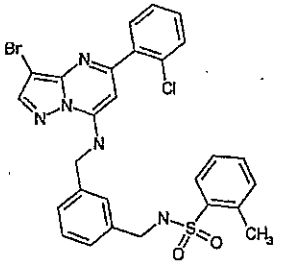
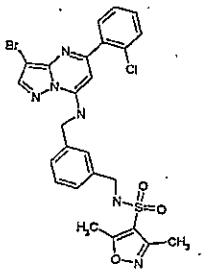
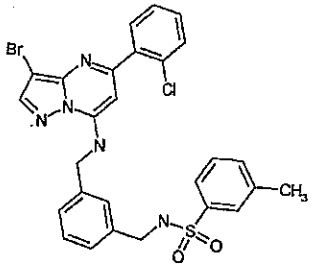
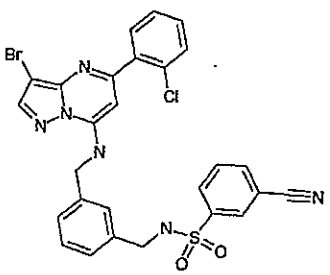
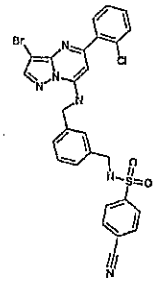
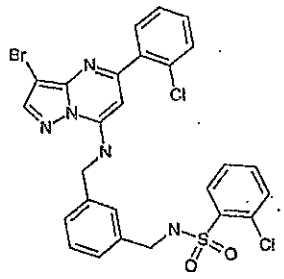
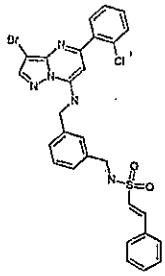
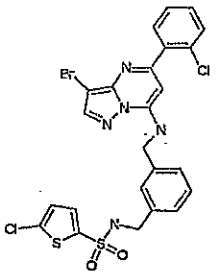
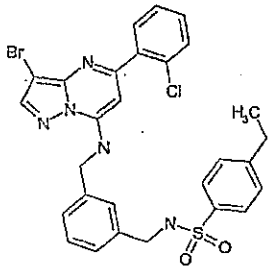
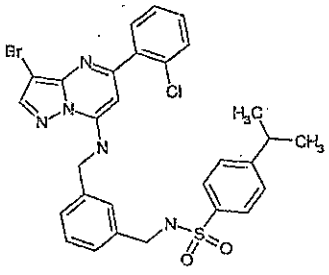
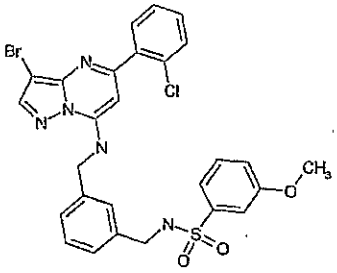
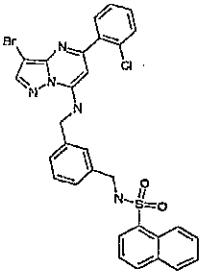
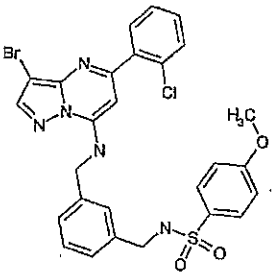
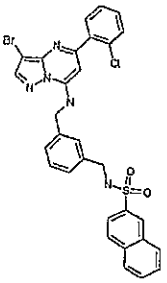
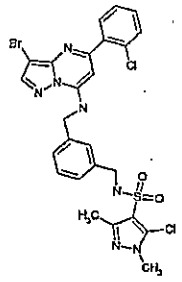
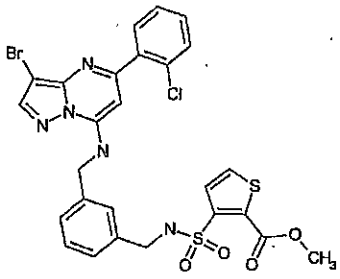
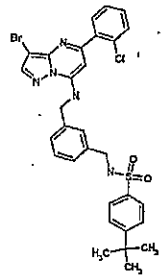
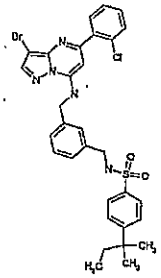
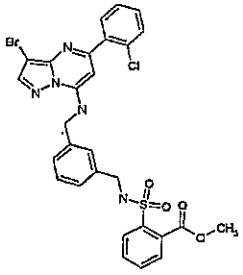
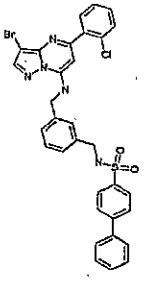
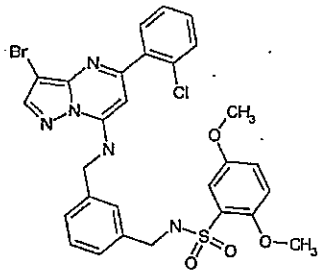
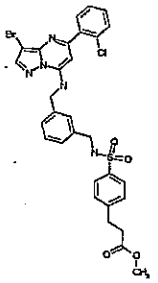
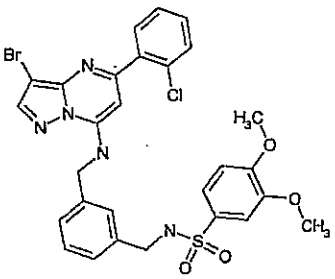
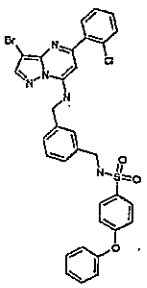
Product	1. Ex. 2. m/z
	1. 5746 2. 613.34
	1. 5747 2. 615.34
	1. 5748 2. 621.34
	1. 5749 2. 631.35
	1. 5750 2. 680.37

TABLE 58

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5801 2. 550.3		1. 5806 2. 598.33
	1. 5802 2. 590.32		10 5807 2. 602.33
	1. 5803 2. 598.33		1. 5808 2. 602.33
	1. 5804 2. 598.33		1. 5809 2. 603.33
	1. 5805 2. 598.33		1. 5810 2. 609.33

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5811 2. 609.33		1. 5816 2. 618.34
	1. 5812 2. 610.34		1. 5817 2. 624.34
	1. 5813 2. 612.34		1. 5818 2. 626.34
	1. 5814 2. 614.34		1. 5819 2. 634.35
	1. 5815 2. 614.34		1. 5820 2. 632.35

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5821 2. 636.35		1. 5826 2. 648.36
	1. 5822 2. 640.35		1. 5827 2. 654.36
	1. 5823 2. 642.35		1. 5828 2. 660.36
	1. 5824 2. 644.35		1. 5829 2. 670.37
	1. 5825 2. 644.35		1. 5830 2. 676.37

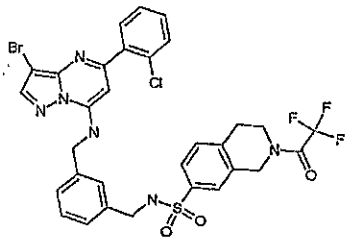
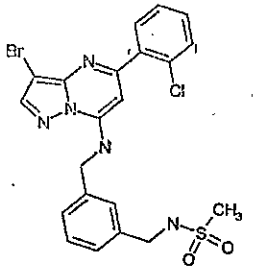
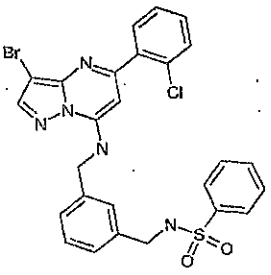
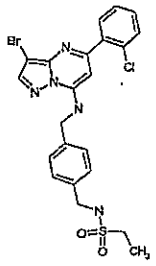
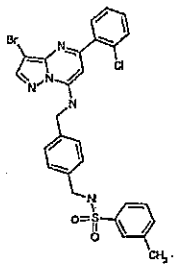
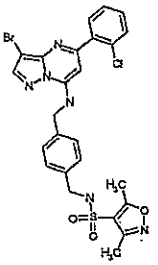
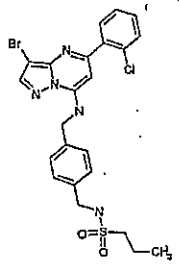
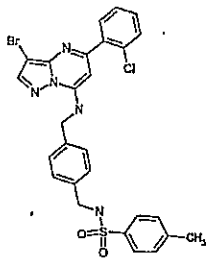
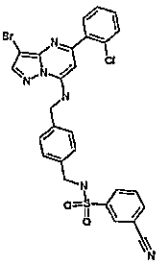
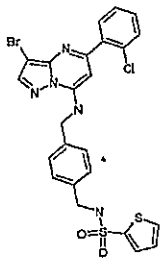
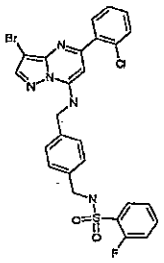
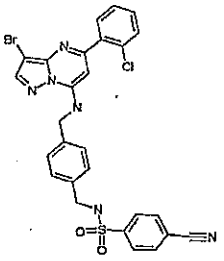
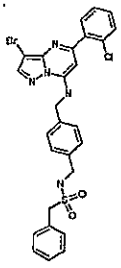
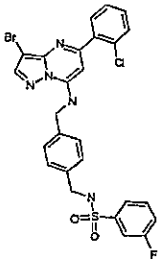
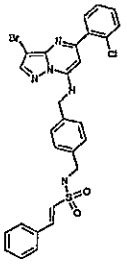
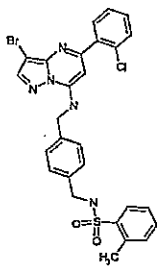
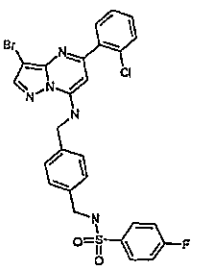
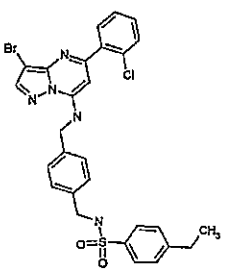
Product	1. Ex. 2. m/z
	1. 5831 2. 735.4
	1. 5832 2. 522.29
	1. 5833 2. 584.32

TABLE 59

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 5901 2. 536.29		1. 5906 2. 598.33		1. 5911 2. 603.33
	1. 5902 2. 550.3		1. 5907 2. 598.33		1. 5912 2. 609.33
	1. 5903 2. 590.32		1. 5908 2. 602.33		1. 5913 2. 609.33
	1. 5904 2. 598.33		1. 5909 2. 602.33		1. 5914 2. 610.34
	1. 5905 2. 598.33		1. 5910 2. 602.33		1. 5915 2. 612.34

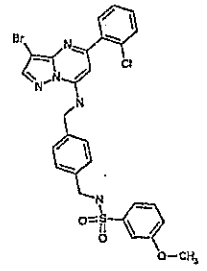
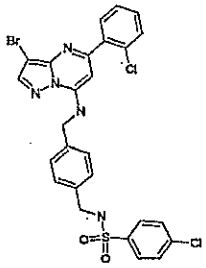
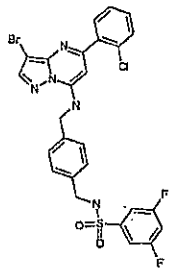
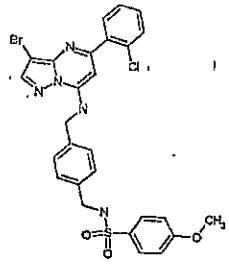
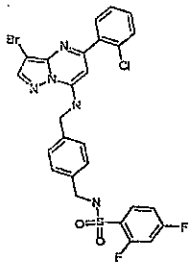
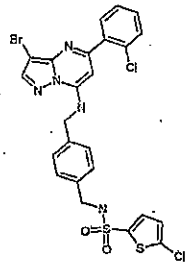
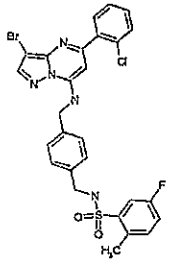
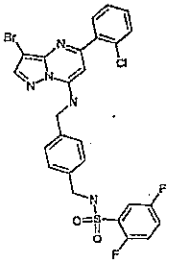
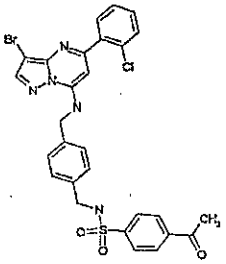
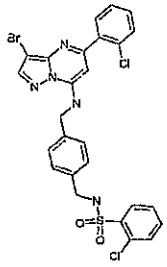
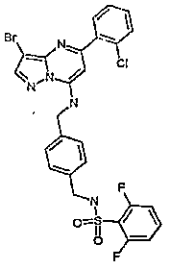
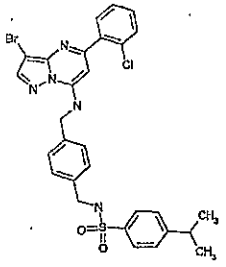
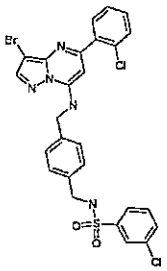
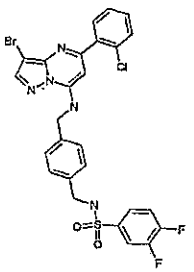
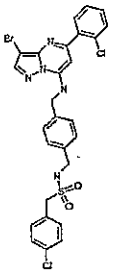
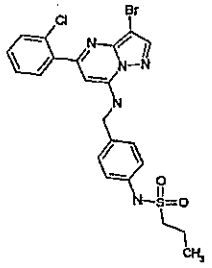
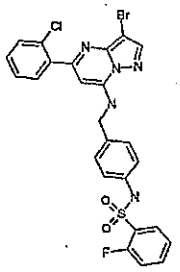
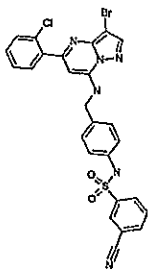
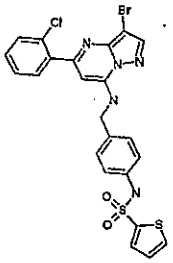
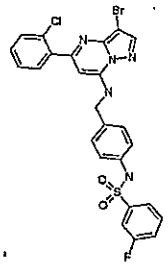
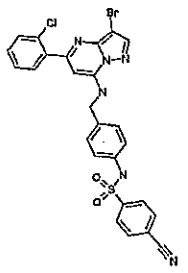
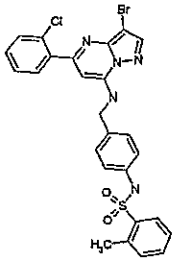
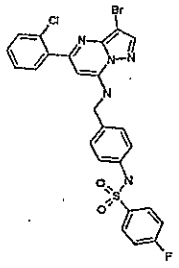
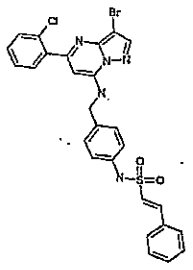
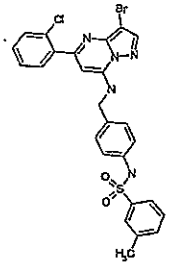
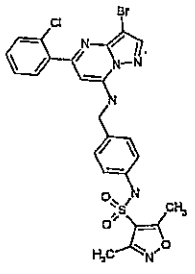
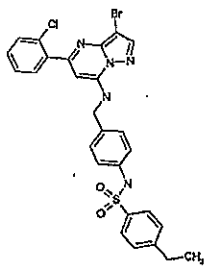
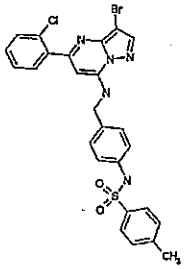
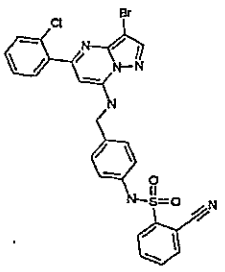
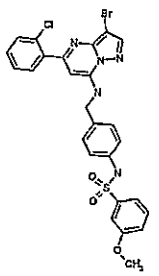
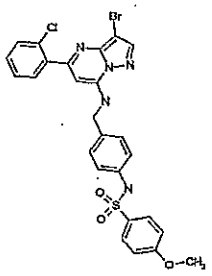
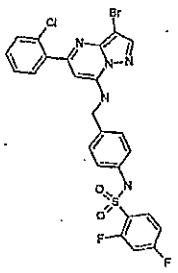
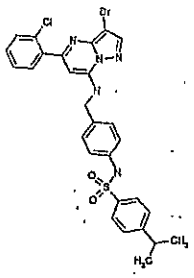
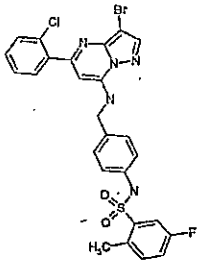
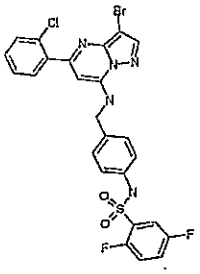
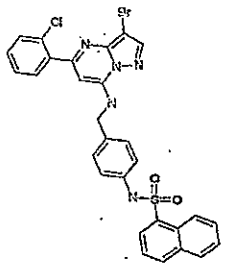
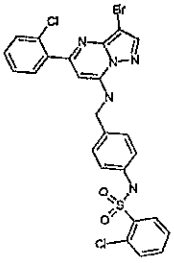
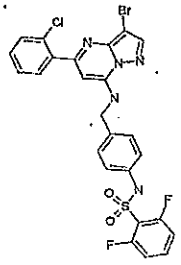
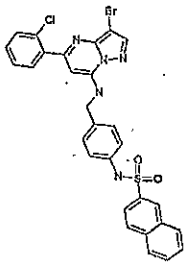
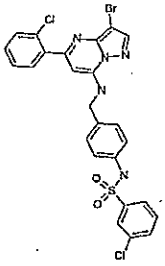
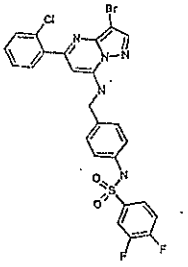
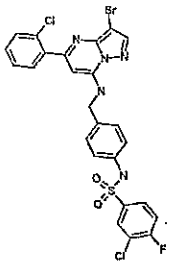
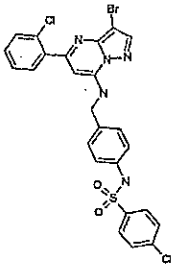
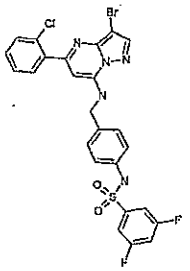
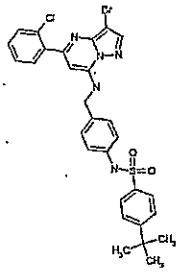
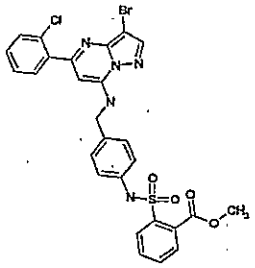
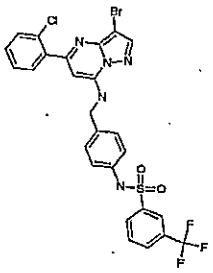
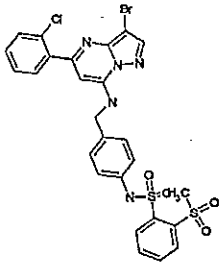
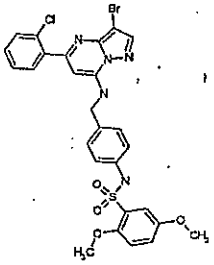
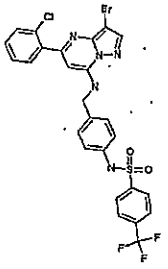
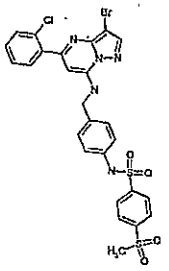
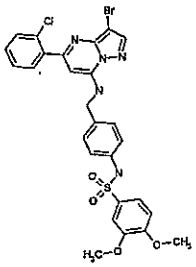
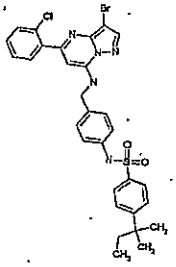
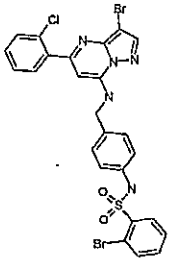
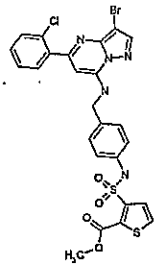
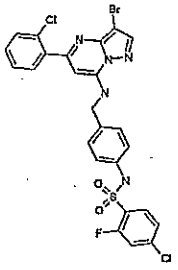
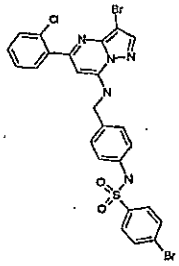
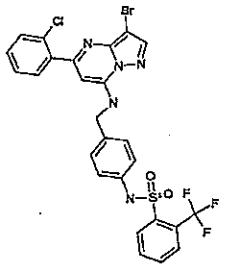
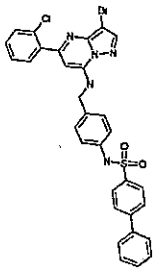
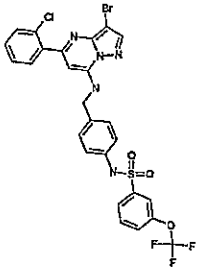
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	1. 5920 2. 618.34		1. 5925 2. 620.34		1. 5930 2. 632.35

TABLE60

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6001 2. 536.29		1. 6006 2. 588.32		1. 6011 2. 595.33
	1. 6002 2. 576.32		1. 6007 2. 588.32		1. 6012 2. 595.33
	1. 6003 2. 584.32		1. 6008 2. 588.32		1. 6013 2. 596.33
	1. 6004 2. 584.32		1. 6009 2. 589.32		1. 6014 2. 598.33
	1. 6005 2. 584.32		1. 6010 2. 595.33		1. 6015 2. 600.33

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6016 2. 600.33		1. 6021 2. 606.33		1. 6026 2. 612.34
	1. 6017 2. 602.33		1. 6022 2. 606.33		1. 6027 2. 620.34
	1. 6018 2. 604.33		1. 6023 2. 606.33		1. 6028 2. 620.34
	1. 6019 2. 604.33		1. 6024 2. 606.33		1. 6029 2. 622.34
	1. 6020 2. 604.33		1. 6025 2. 606.33		1. 6030 2. 626.34

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6031 2. 628.35		1. 6036 2. 638.35		1. 6041 2. 648.36
	1. 6032 2. 630.35		1. 6037 2. 638.35		1. 60425 2. 648.36
	1. 6033 2. 630.35		1. 6038 2. 638.35		1. 6043 2. 648.36
	1. 6034 2. 634.35		1. 6039 2. 622.34		1. 6044 2. 648.36
	1. 6035 2. 638.35		1. 6040 2. 646.36		1. 6045 2. 654.36

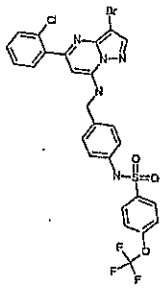
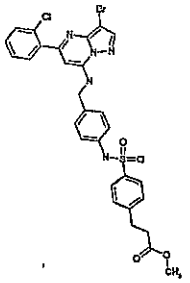
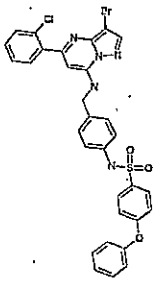
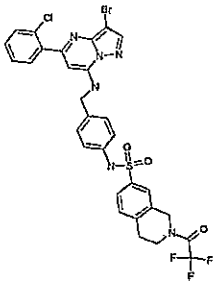
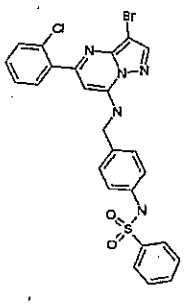
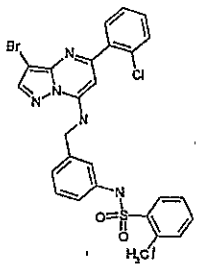
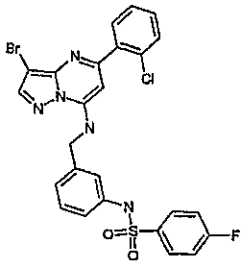
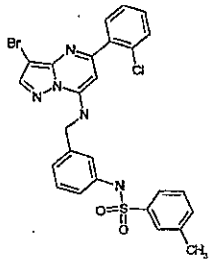
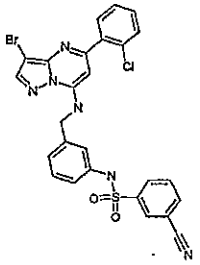
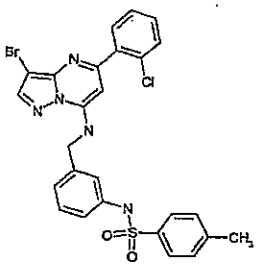
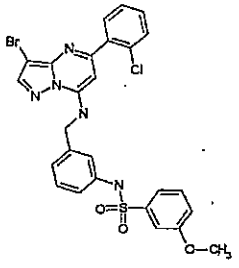
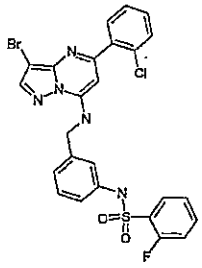
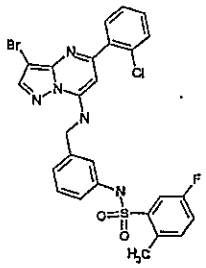
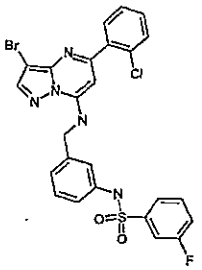
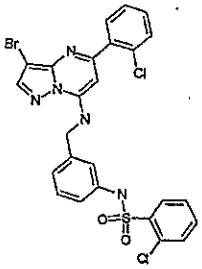
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	1. 6046 2. 654.36
	1. 6047 2. 656.36
	1. 6048 2. 662.36
	1. 6049 2. 721.4
	1. 6050 2. 570.31

TABLE 61

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6101 2. 584.32		1. 6106 2. 588.32
	1. 6102 2. 584.32		1. 6107 2. 595.33
	1. 6103 2. 584.32		1. 6108 2. 600.33
	1. 6104 2. 588.32		1. 6109 2. 602.33
	1. 6105 2. 588.32		1. 6110 2. 604.33

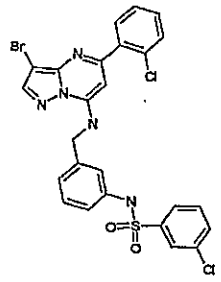
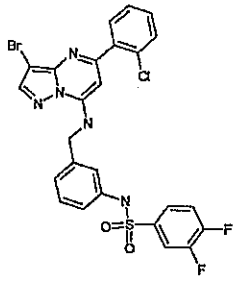
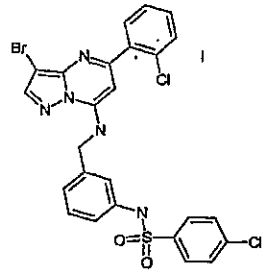
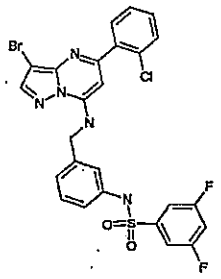
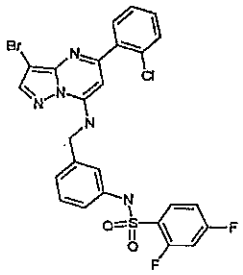
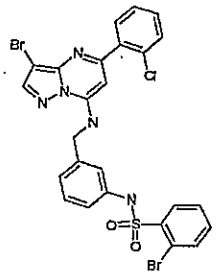
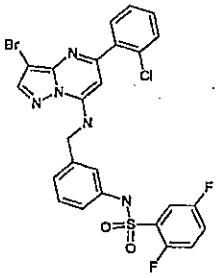
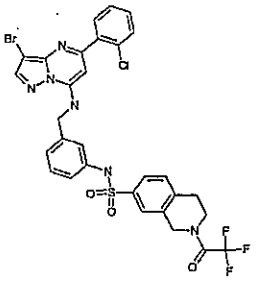
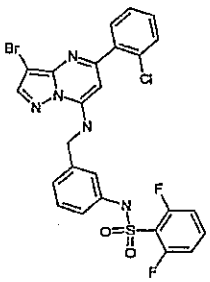
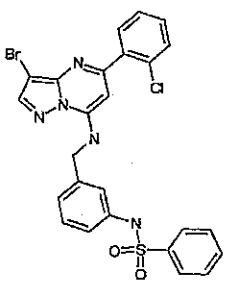
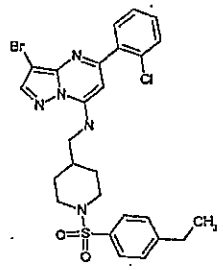
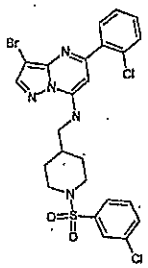
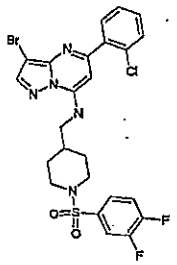
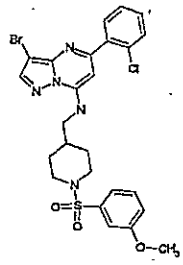
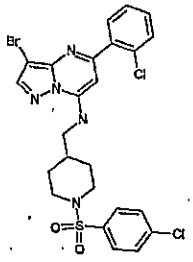
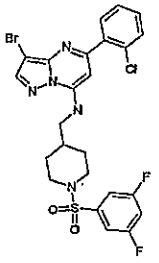
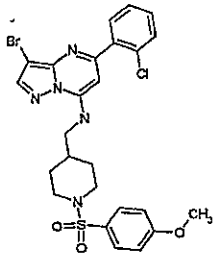
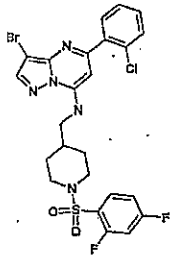
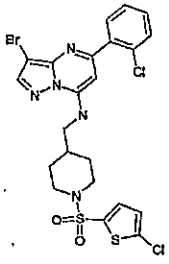
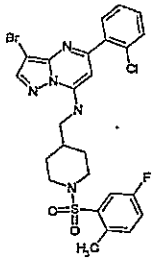
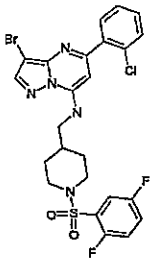
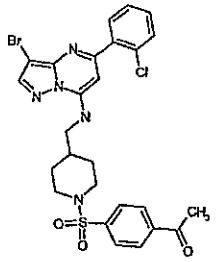
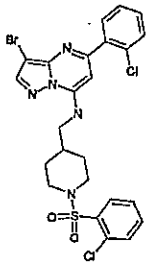
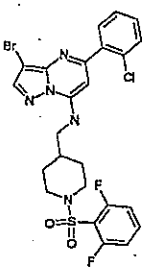
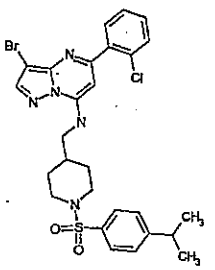
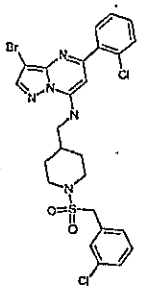
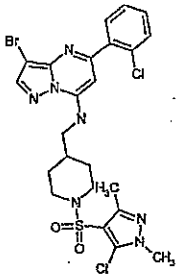
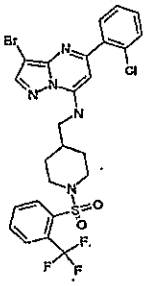
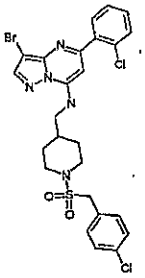
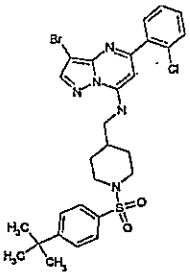
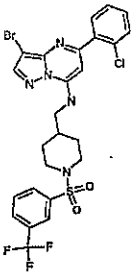
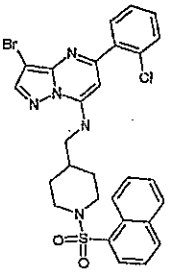
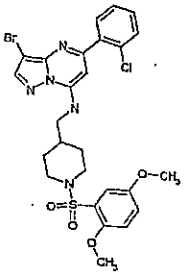
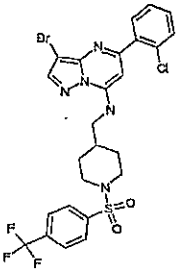
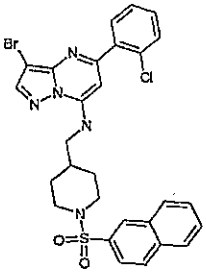
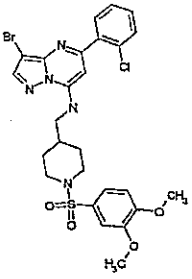
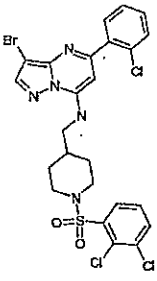
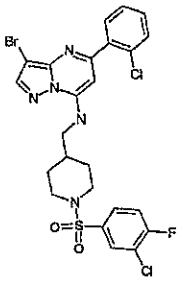
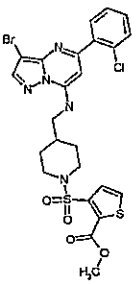
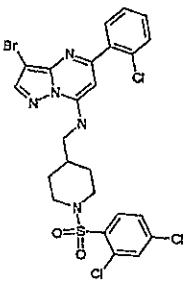
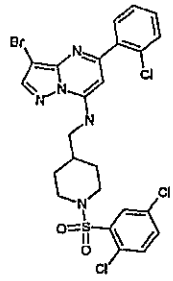
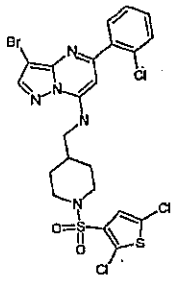
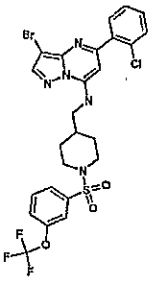
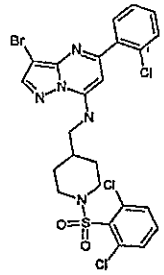
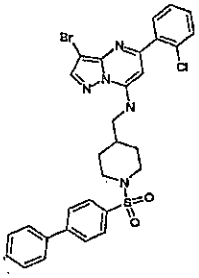
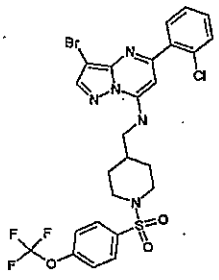
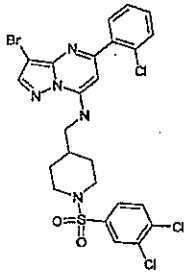
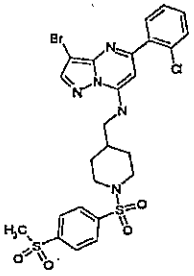
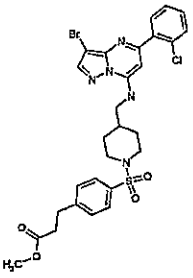
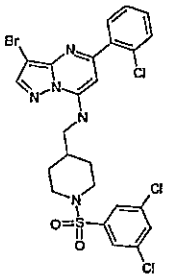
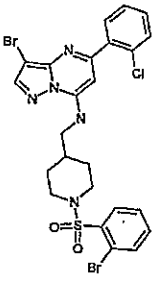
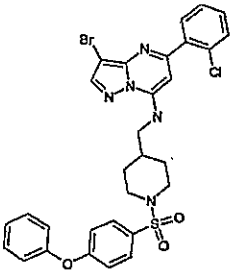
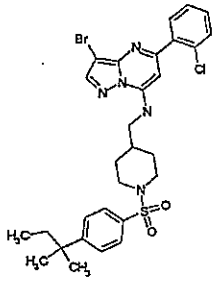
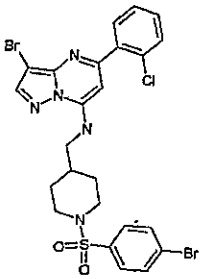
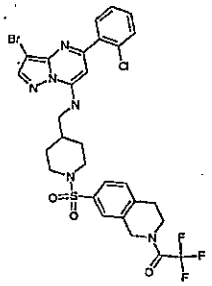
Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6111 2. 604.33		1. 6116 2. 606.33
	1. 6112 2. 604.33		1. 6117 2. 606.33
	1. 6113 2. 606.33		1. 6118 2. 648.36
	1. 6114 2. 606.33		1. 6119 2. 721.4
	1. 6115 2. 606.33		1. 6120 2. 570.31

TABLE62 **Error! Objects cannot be created from editing field codes.**

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6216 2. 590.32		1. 6221 2. 596.33		1. 6226 2. 598.33
	1. 6217 2. 592.33		1. 6222 2. 596.33		1. 6227 2. 596.33
	1. 6218 2. 592.33		1. 6223 2. 598.33		1. 6228 2. 602.33
	1. 6219 2. 592.33		1. 6224 2. 596.33		1. 6229 2. 604.33
	1. 6220 2. 594.33		1. 6225 2. 598.33		1. 6230 2. 604.33

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6231 2. 610.34		1. 6236 2. 614.34		1. 6241 2. 630.35
	1. 6232 2. 610.34		1. 6237 2. 618.34		1. 6242 2.
	1. 6233 2. 612.34		1. 6238 2. 622.34		1. 6243 2. 630.35
	1. 6234 2. 612.34		1. 6239 2. 622.34		1. 6244 2. 630.35
	1. 6235 2. 614.34		1. 6240 2. 626.34		1. 6245 2. 630.35

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1 .6246 2. 630.35		1 .6251 2. 636.35		1. 6256 2. 646.36
	1. 6247 2. 630.35		1.0 6252 2. 638.35		1. 6257 2. 646.36
	1. 6248 2. 630.35		1. 6253 2. 640.35		1. 6258 2. 648.36
	1. 6249 2. 630.35		1. 6254 2. 640.35		1. 6259 2. 654.36
	1 .6250 2. 632.35		1. 6255 2. 638.35		1. 6260 2. 713.39

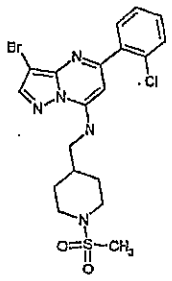
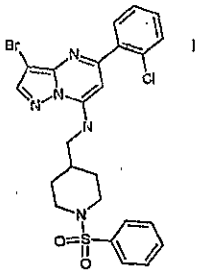
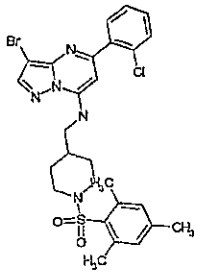
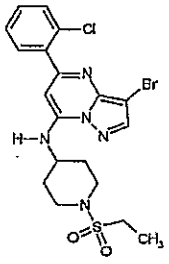
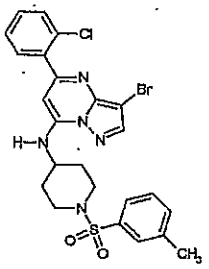
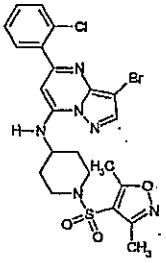
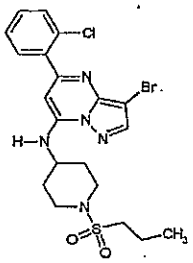
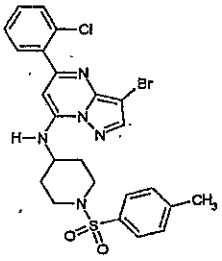
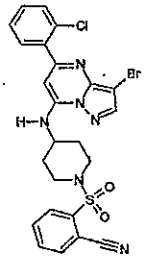
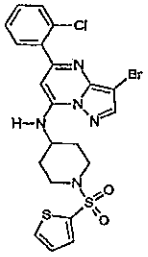
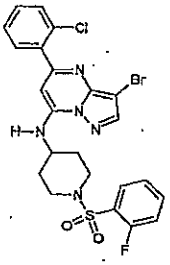
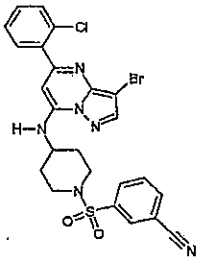
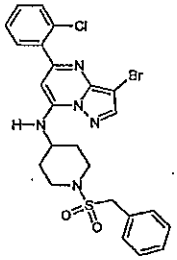
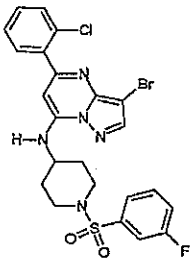
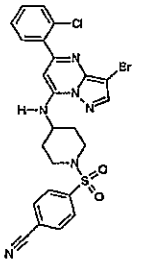
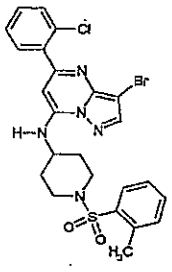
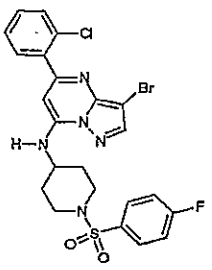
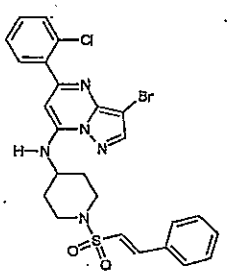
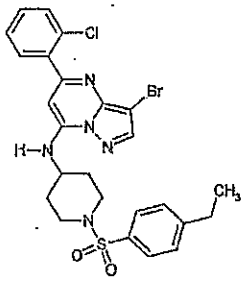
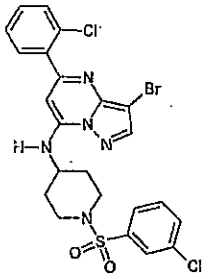
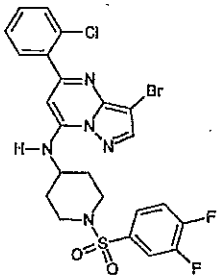
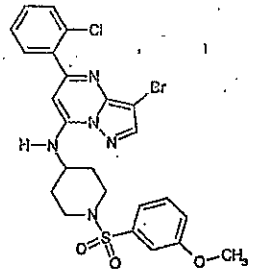
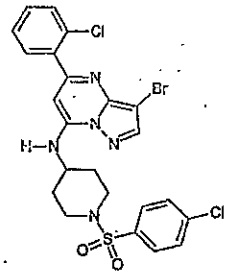
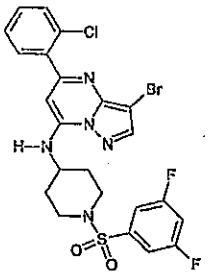
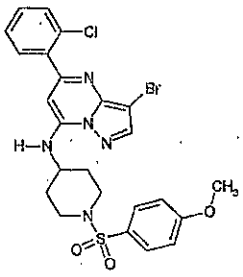
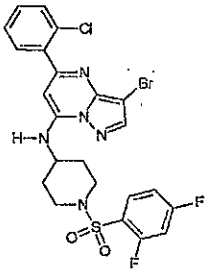
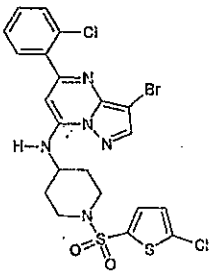
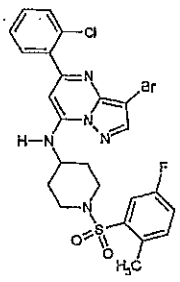
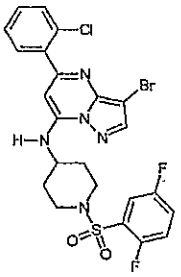
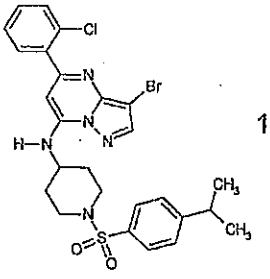
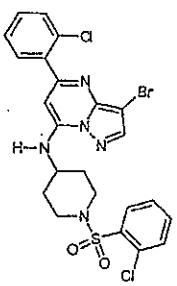
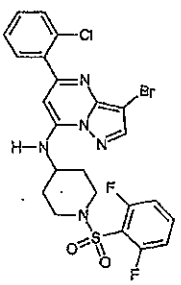
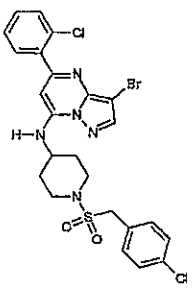
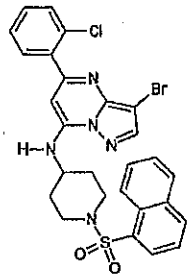
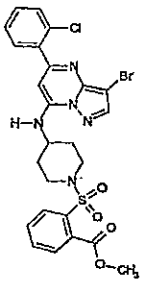
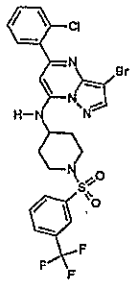
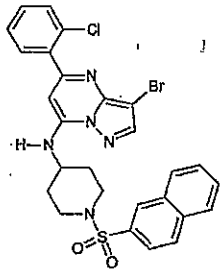
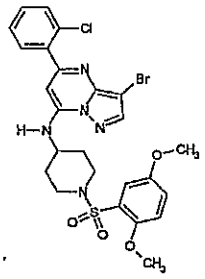
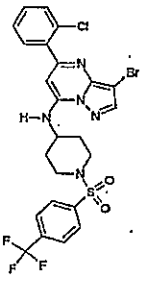
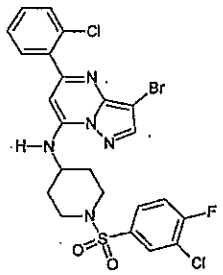
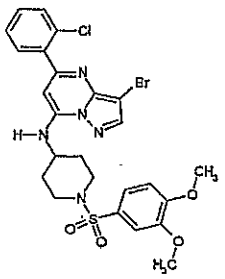
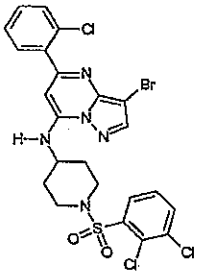
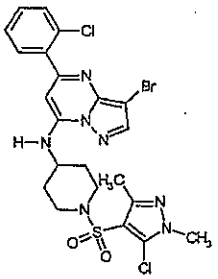
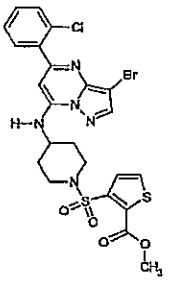
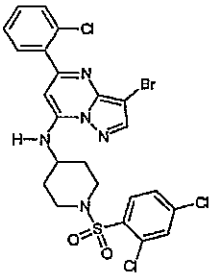
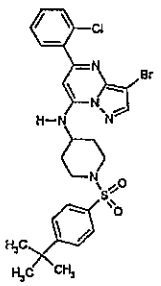
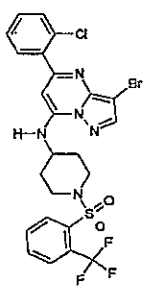
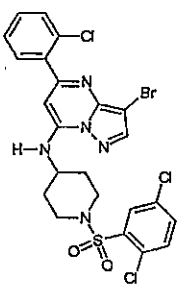
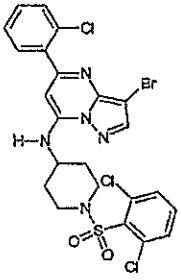
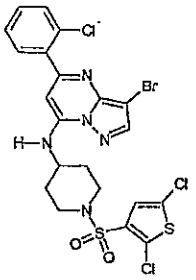
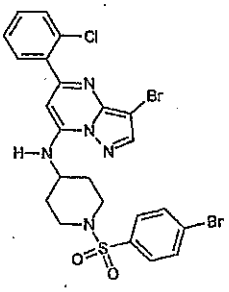
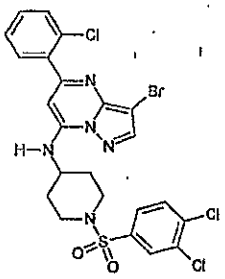
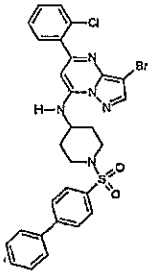
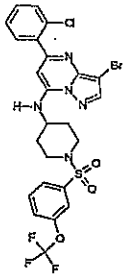
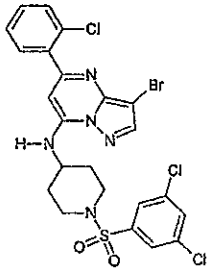
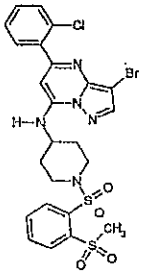
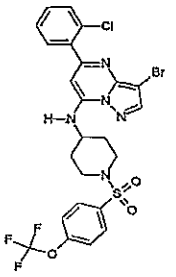
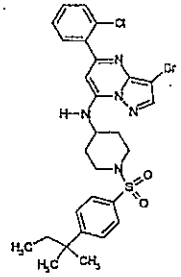
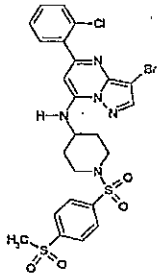
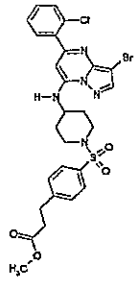
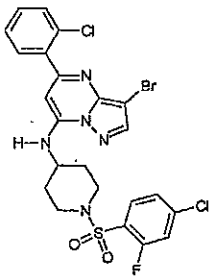
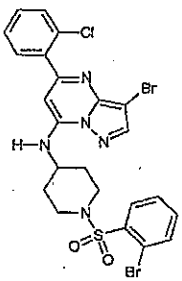
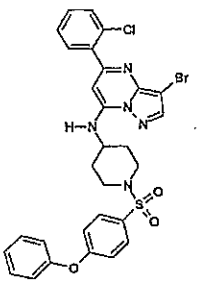
Product	1. Ex. 2. m/z
	1. 6261 2. 500.27
	1. 6262 2. 562.31
	1. 6263 2. 604.33

TABLE 63

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6301 2. 500.27		1. 6306 2. 562.31		1. 6311 2. 567.31
	1. 6302 2. 514.28		1. 6307 2. 562.31		1. 6312 2. 573.32
	1. 6303 2. 554.3		1. 6308 2. 566.31		1. 6313 2. 573.32
	1. 6304 2. 562.31		1. 6309 2. 566.31		1. 6314 2. 573.32
	1. 6305 2. 562.31		1. 6310 2. 566.31		1. 6315 2. 574.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6316 2. 576.32		1. 6321 2. 582.32		1. 6326 2. 584.32
	1. 6317 2. 578.32		1. 6322 2. 582.32		1. 6327 2. 584.32
	1. 6318 2. 578.32		1. 6323 2. 584.32		1. 6328 2. 588.32
	1. 6319 2. 580.32		1. 6324 2. 582.32		1. 6329 2. 590.32
	1. 6320 2. 582.32		1. 6325 2. 584.32		1. 6330 2. 596.33

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6331 2. 598.33		1. 6336 2. 606.33		1. 6341 2. 616.34
	1. 6332 2. 598.33		1. 6337 2. 608.33		1. 6342 2. 616.34
	1. 6333 2. 600.33		1. 6338 2. 608.33		1. 6343 2. 616.34
	1. 6334 2. 600.33		1. 6339 2. 612.34		1.6344 2. 616.34
	1. 6335 2. 604.33		1. 6340 2. 616.34		1. 6345 2. 616.34

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6346 2. 616.34		1. 6351 2. 622.34		1. 6356 2. 626.34
	1. 6347 2. 616.34		1. 6352 2. 624.34		1. 6357 2. 632.35
	1. 6348 2. 616.34		1. 6353 2. 626.34		1. 6358 2. 632.35
	1. 6349 2. 618.34		1. 6354 2. 626.34		1. 6359 2. 634.35
	1. 6350 2. 600.33		1. 6355 2. 626.34		1. 6360 2. 640.35

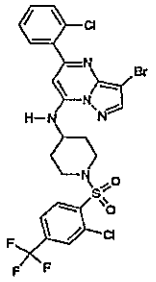
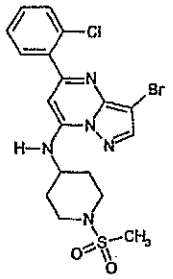
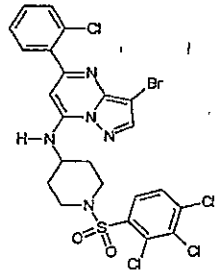
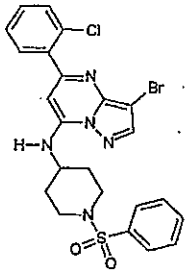
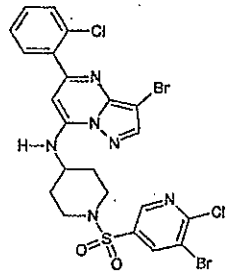
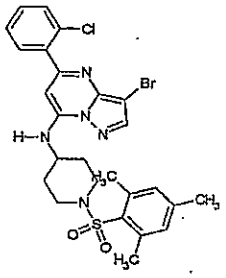
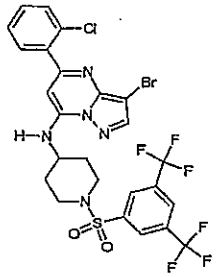
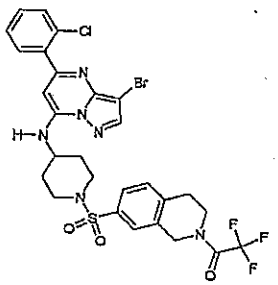
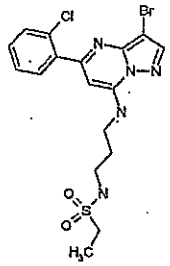
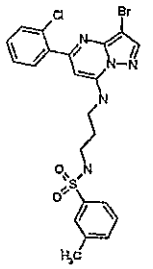
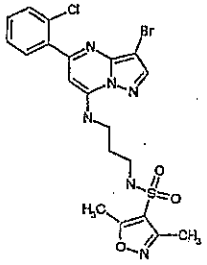
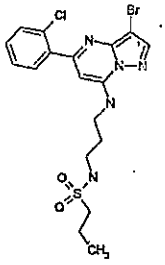
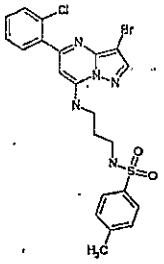
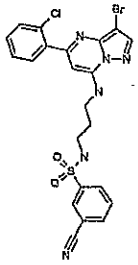
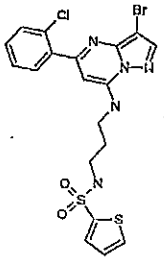
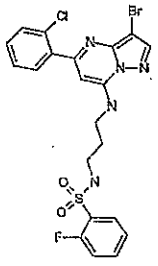
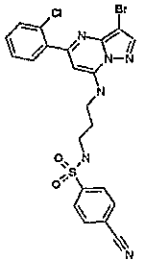
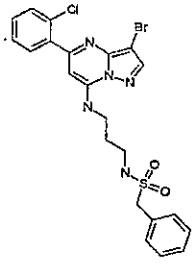
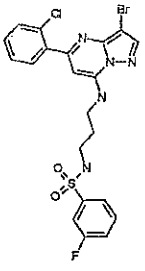
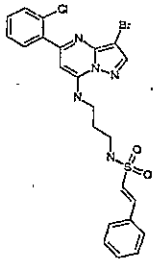
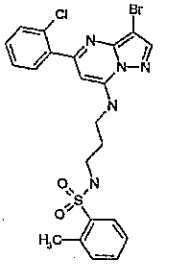
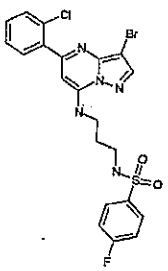
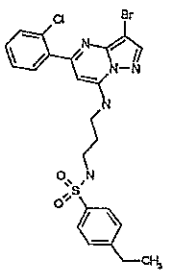
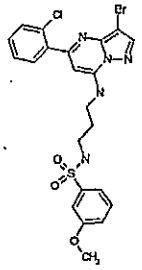
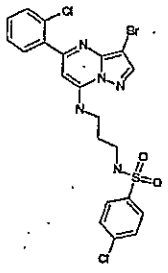
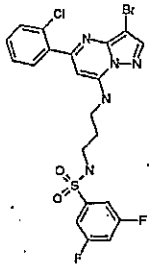
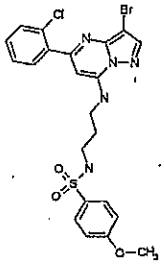
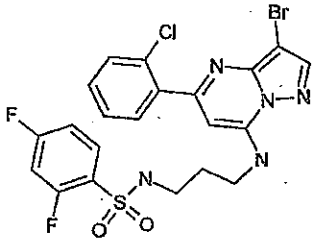
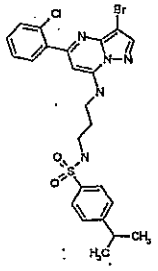
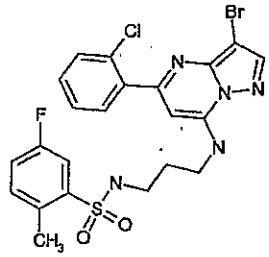
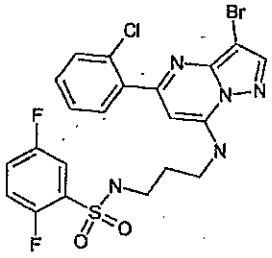
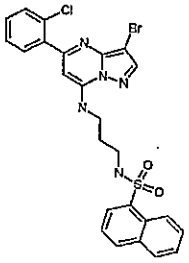
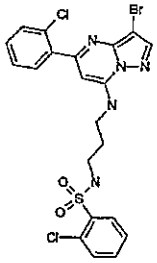
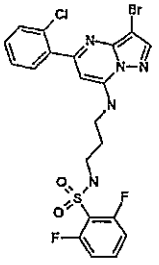
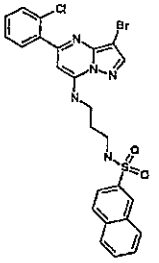
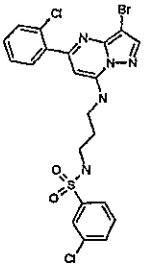
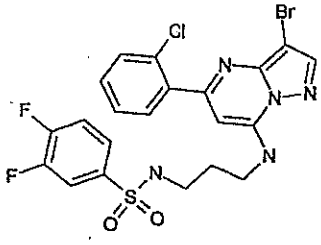
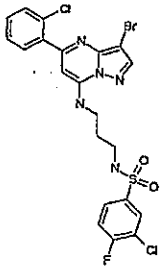
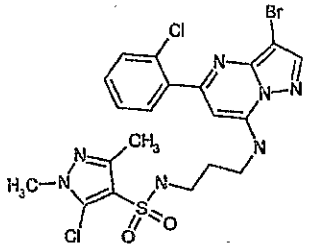
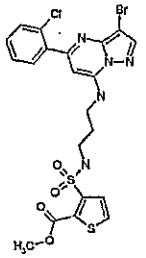
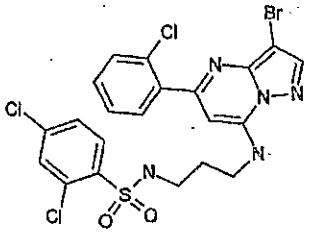
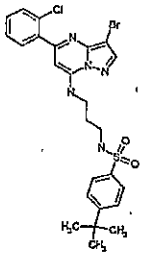
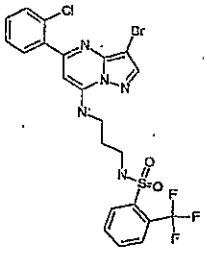
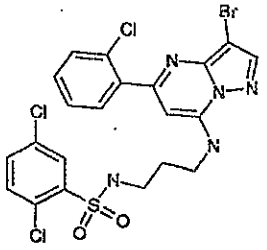
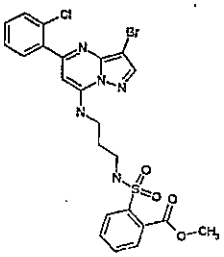
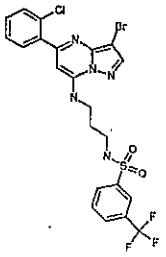
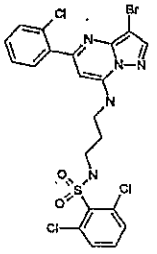
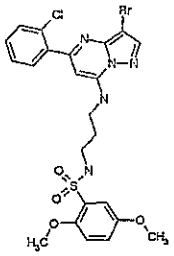
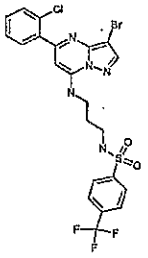
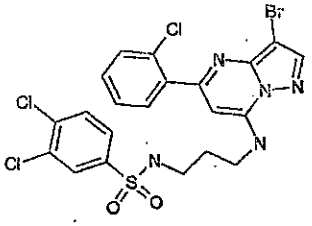
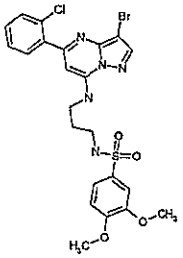
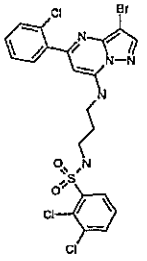
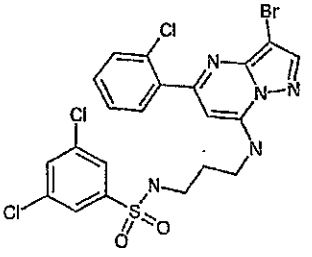
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	1. 6361 2. 650.36		1. 6366 2. 486.27
	1. 6362 2. 650.36		1. 6367 2. 548.3
	1. 6363 2. 661.36		1. 6368 2. 590.32
	1. 6364 2. 684.38		
	1. 6365 2. 699.38		

TABLE 64

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6401 2. 474.26		1. 6406 2. 536.29		1. 6411 2. 541.3
	1. 6402 2. 488.27		1. 6407 2. 536.29		1. 6412 2. 547.3
	1. 6403 2. 528.29		1. 6408 2. 540.3		1. 6413 2. 547.3
	1. 6404 2. 536.29		1. 6409 2. 540.3		1. 6414 2. 548.3
	1. 6405 2. 536.29		1. 6410 2. 540.3		1. 6415 2. 550.3

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6416 2. 552.3		1. 6421 2. 556.31		1. 6426 2 558.31
	1. 6417 2. 552.3		1. 6122 2. 558.31		1. 6427 2 564.31
	1. 6418 2. 554.3		1. 6423 2. 558.31		1. 6428 2. 572.31
	1. 6419 2. 556.31		1. 6424 2. 558.31		1. 6429 2. 572.31
	1. 6420 2. 556.31		1. 6425 2. 558.31		1. 6430 2. 574.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2 m/z
	1. 6431 2. 574.32		1. 6436 2. 586.32		1. 6441 2 590.32
	1. 6432 2. 578.32		1. 6437 2. 590.32		1. 6442 2 590.32
	1. 6433 2. 580.32		1. 6438 2. 590.32		1. 6443 2 590.32
	1. 6434 2. 582.32		1. 6439 2. 590.32		1. 6444 2 590.32
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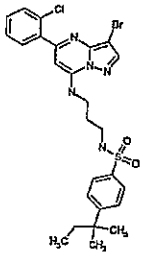
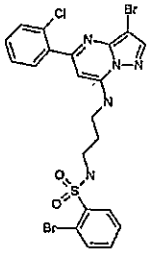
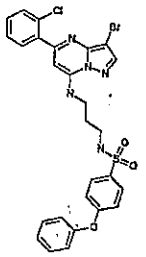
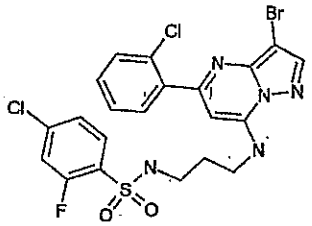
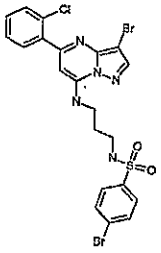
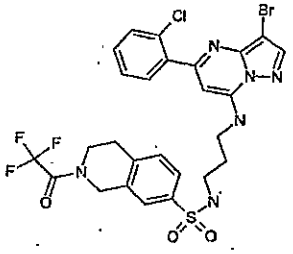
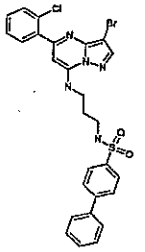
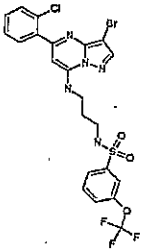
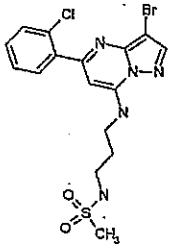
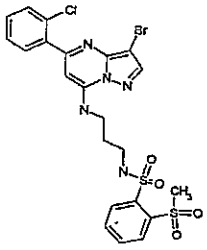
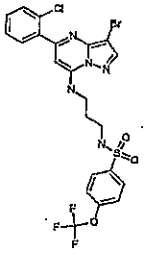
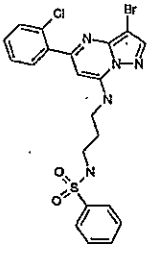
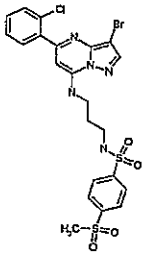
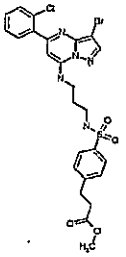
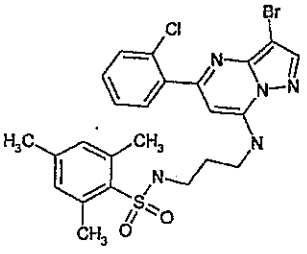
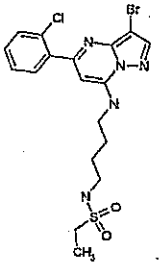
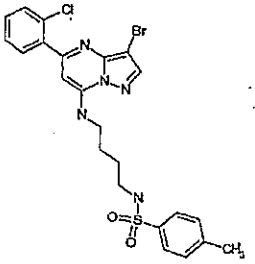
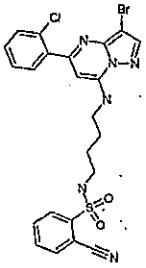
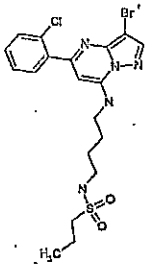
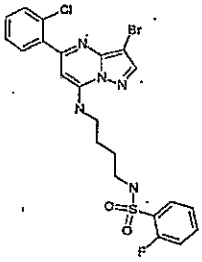
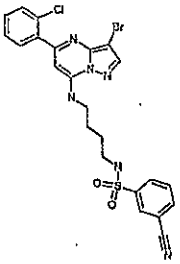
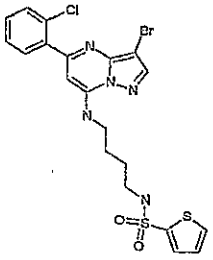
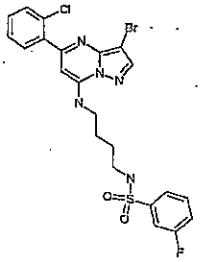
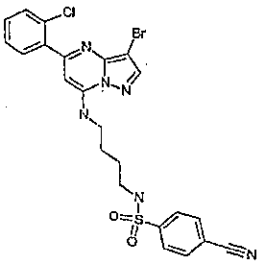
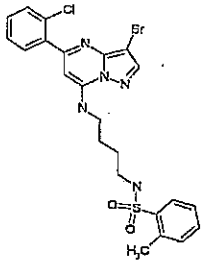
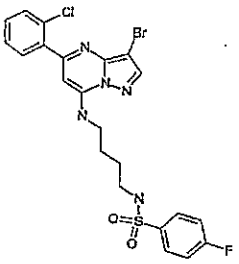
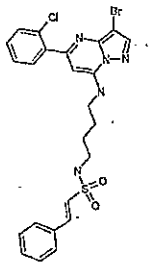
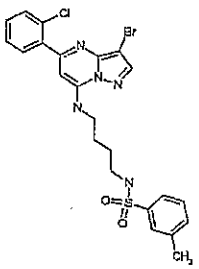
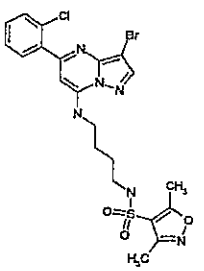
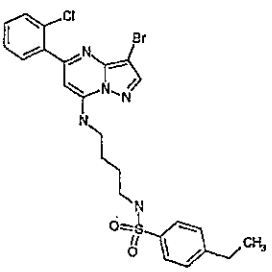
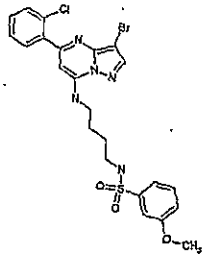
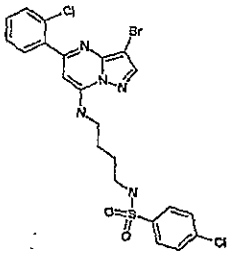
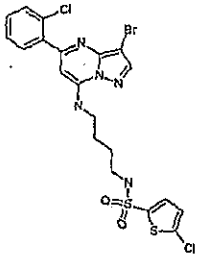
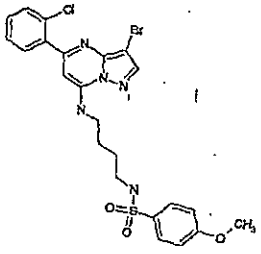
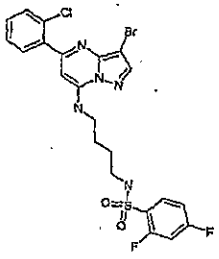
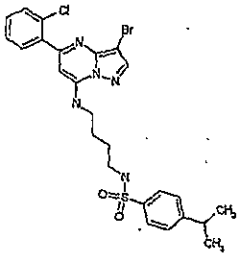
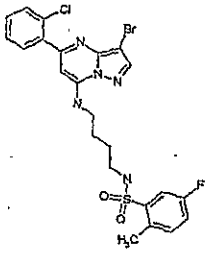
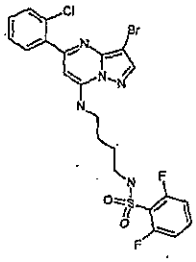
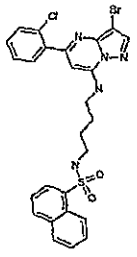
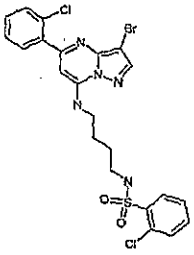
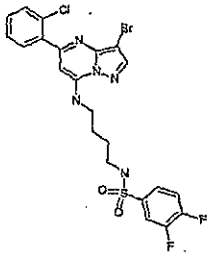
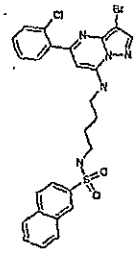
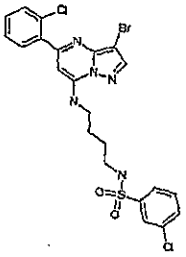
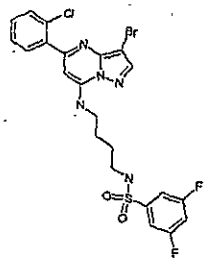
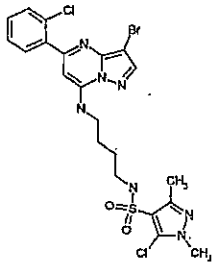
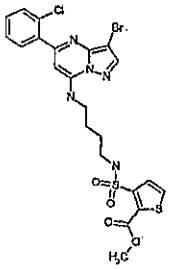
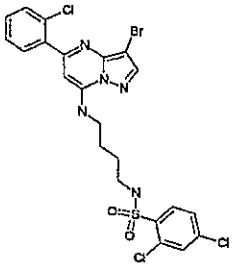
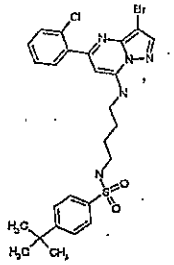
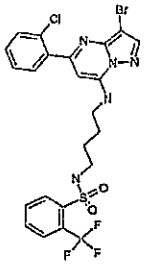
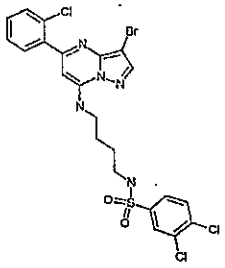
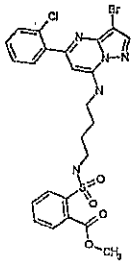
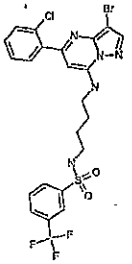
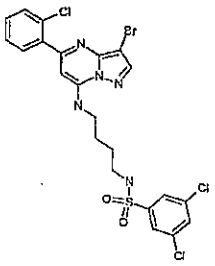
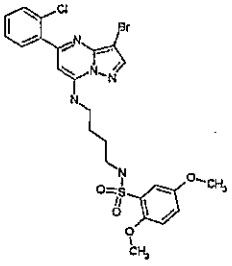
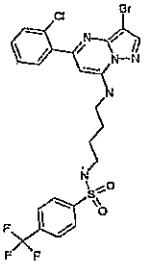
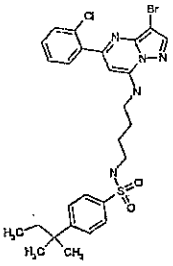
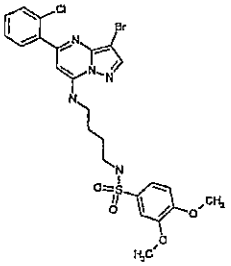
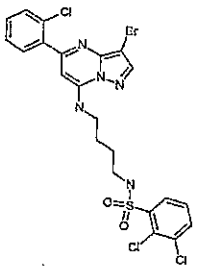
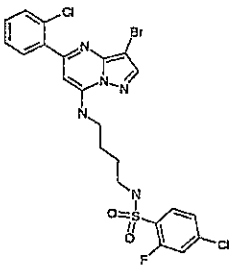
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	1. 6446 2. 592.33		1. 6451 2. 600.33		1. 6456 2 614.34
	1. 6447 2. 574.32		1. 6452 2. 600.33		1. 6447 2 673.37
	1. 6448 2. 598.33		1. 6453 2. 606.33		1. 6458 2 460.25
	1. 6449 2. 600.33		1. 6454 2. 606.33		6459 2.
	1. 6450 2. 600.33		1. 6455 2. 608.33		1. 6460 2. 564.31

TABLE 65

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6501 2. 488.27		1. 6506 2. 550.3		1. 6511 2. 561.31
	1. 6502 2. 502.28		1. 6507 2. 554.3		1. 6512 2. 561.31
	1. 6503 2. 542.3		1. 6508 2. 554.3		1. 6213 2. 561.31
	1. 6504 2. 550.3		1. 6509 2. 554.3		1. 6514 2. 562.31
	1. 6505 2. 550.3		1. 6510 2. 555.31		1. 6515 2. 564.31

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6516 2. 566.31		1. 6521 2. 570.31		1. 6526 2. 576.32
	1. 6517 2. 566.31		1. 6522 2. 572.31		1. 6527 2. 578.32
	1. 6518 2. 568.31		1. 6523 2. 572.31		1. 6528 2. 586.32
	1. 6519 2. 570.31		1. 6524 2. 572.31		1. 6529 2. 586.32
	1. 6520 2. 570.31		1. 6525 2. 572.31		1. 6530 2. 588.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6531 2. 588.32		1. 6536 2. 600.33		1. 6541 2. 604.33
	1. 6532 2. 592.33		1. 6537 2. 604.33		1. 6542 2. 604.33
	1. 6533 2. 594.33		1. 6538 2. 604.33		1. 6543 2. 604.33
	1. 6534 2. 596.33		1. 6539 2. 604.33		1. 6544 2. 606.33
	1. 6535 2. 596.33		1. 6540 2. 604.33		1. 6545 2. 588.32

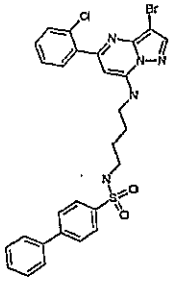
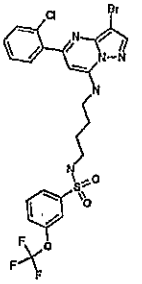
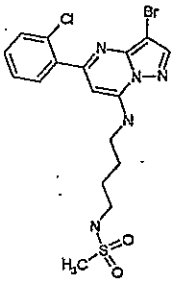
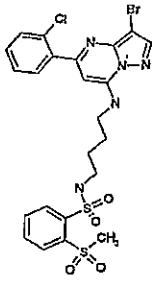
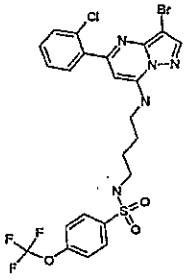
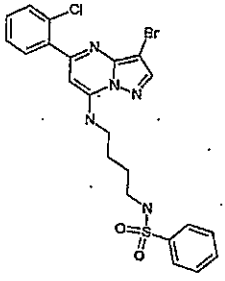
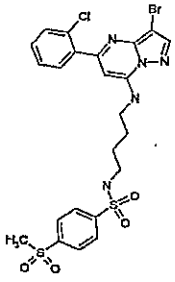
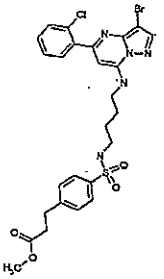
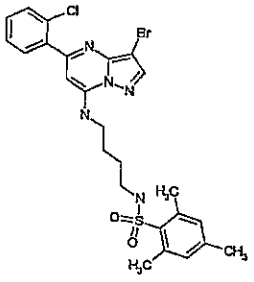
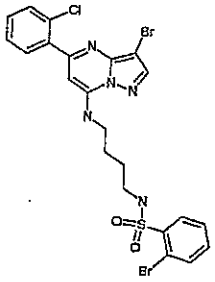
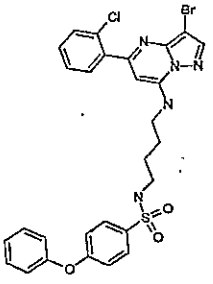
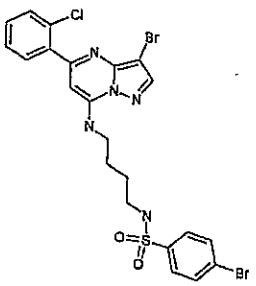
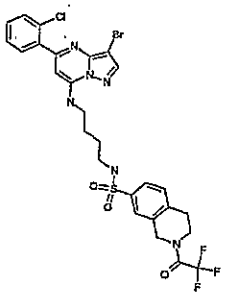
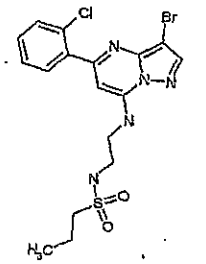
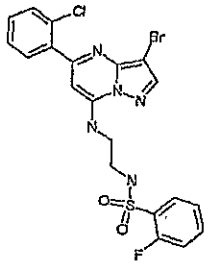
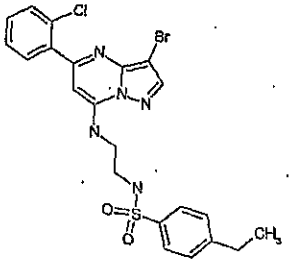
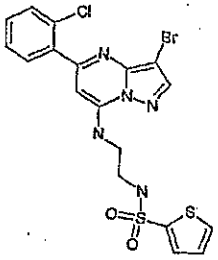
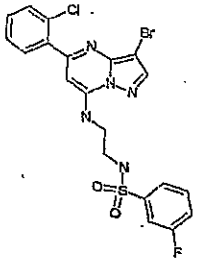
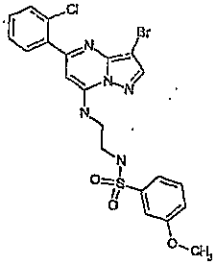
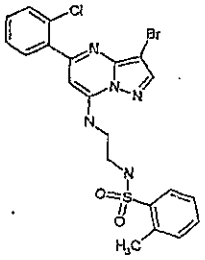
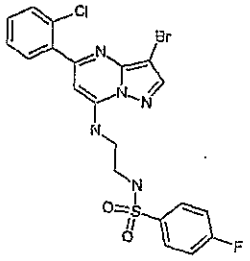
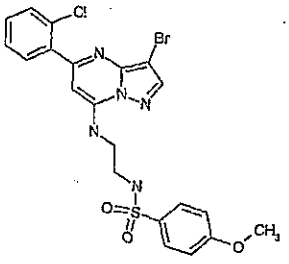
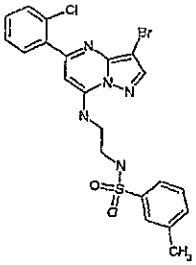
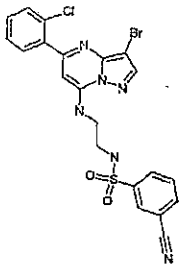
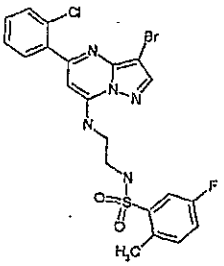
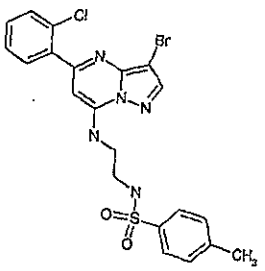
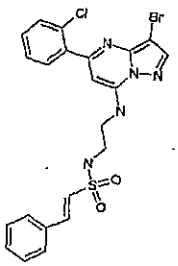
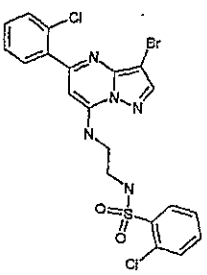
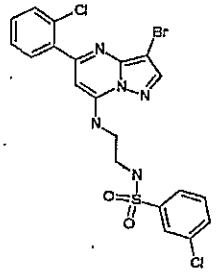
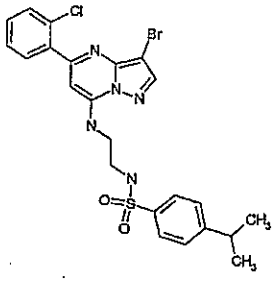
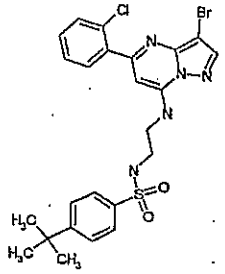
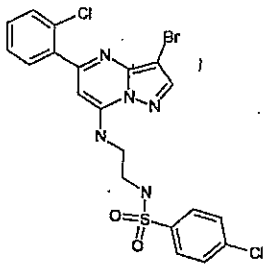
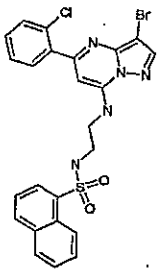
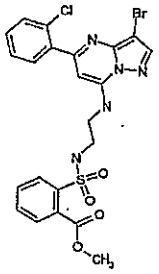
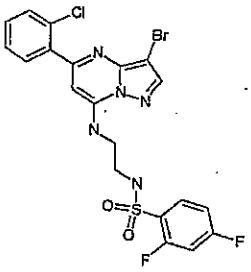
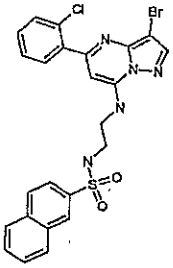
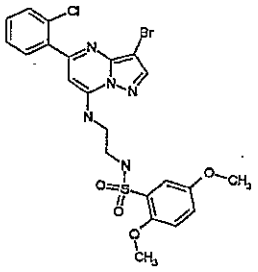
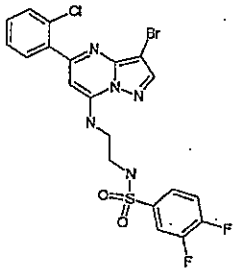
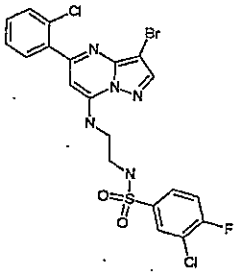
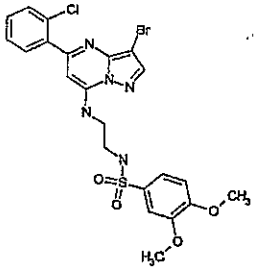
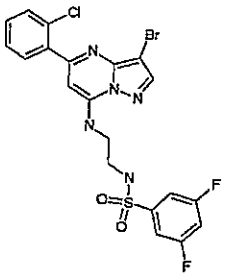
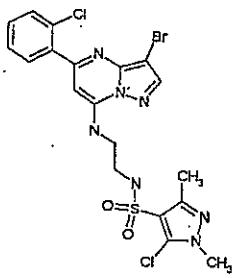
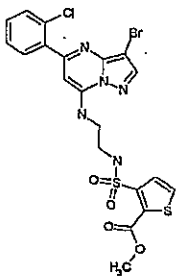
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	1. 6546 2. 612.34		1. 6551 2. 620.34		1. 6556 2. 474.26
	1. 6547 2. 614.34		1. 6552 2. 620.34		1. 6557 2. 536.29
	1. 6548 2. 614.34		1. 6553 2. 622.34		1. 6558 2. 578.32
	1. 6549 2. 614.34		1. 6554 2. 628.35		
	1. 6550 2. 614.34		1. 6555 2. 687.38		

TABLE 67

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. m/z
	1. 6701 2. 474.26		1. 6706 2. 526.29		1. 6711 536.2
	1. 6702 2. 514.28		1. 6707 2. 526.29		1. 6712. 538.3
	1. 6703 2. 522.29		1. 6708 2. 526.29		1. 6713 2 538.3
	1. 6704 2. 522.29		1. 6709 2. 533.29		1. 6714 2.
	5705 2. 22.29		1. 6710 2. 534.29		1. 6715 2. 542.3

5

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6716 2. 542.3		1. 6721 2. 550.3		1. 6726 2 564.31
	1. 6717 2. 542.3		1. 6722 2. 558.31		1. 6727 2. 566.31
	1. 6718 2. 544.3		1. 6723 2. 558.31		1. 6728 2. 568.31
	1. 6719 2. 544.3		1. 6724 2. 560.31		1. 6729 2. 568.31
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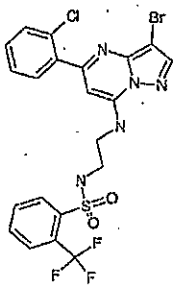
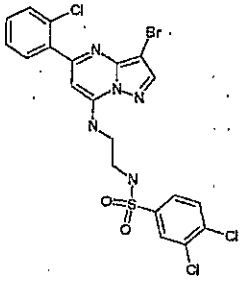
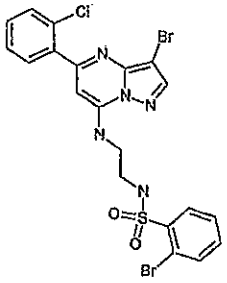
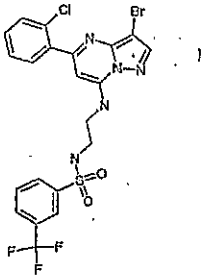
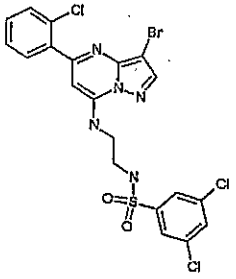
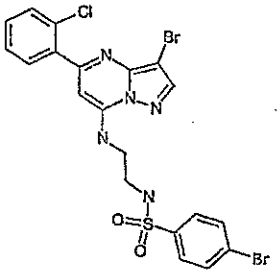
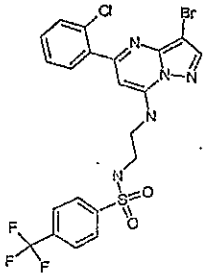
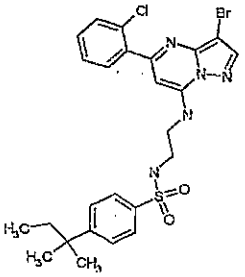
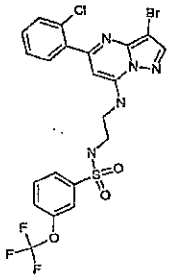
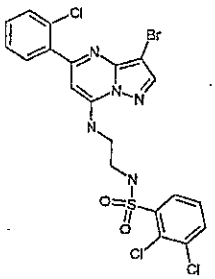
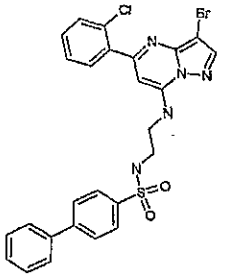
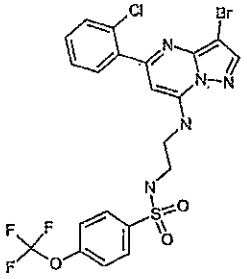
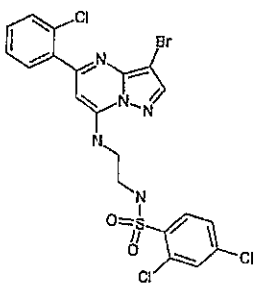
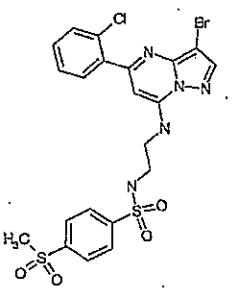
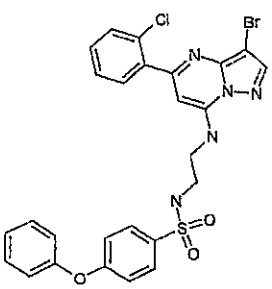
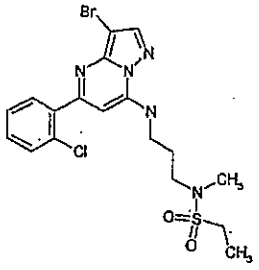
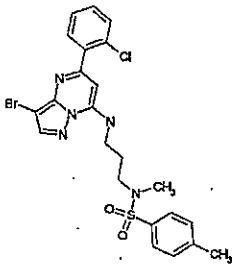
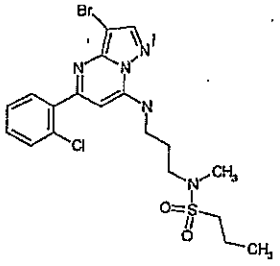
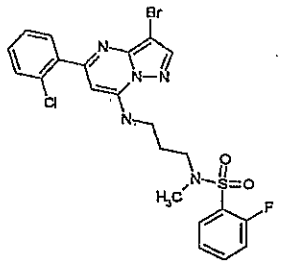
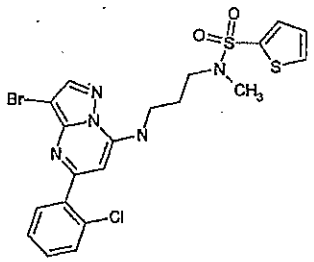
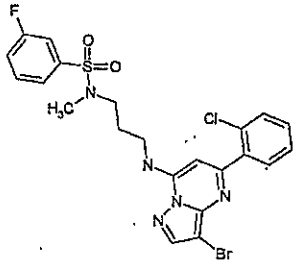
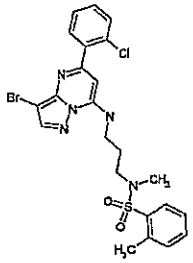
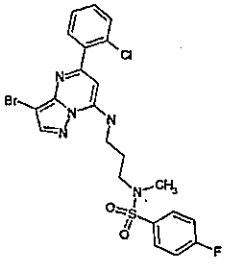
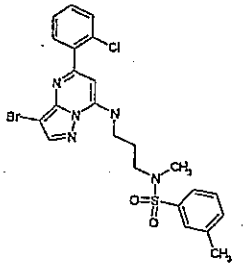
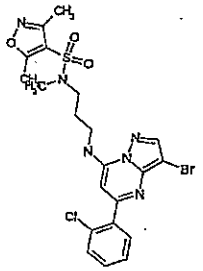
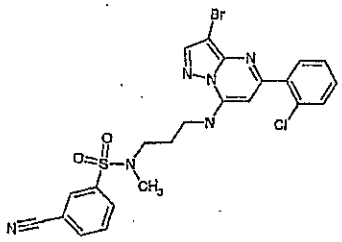
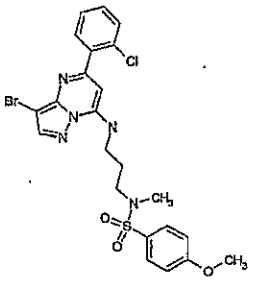
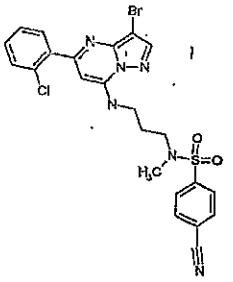
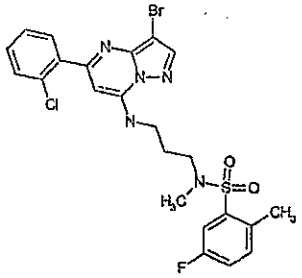
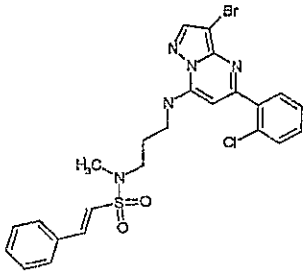
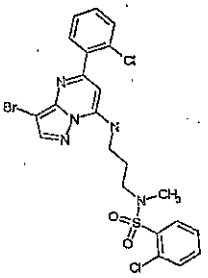
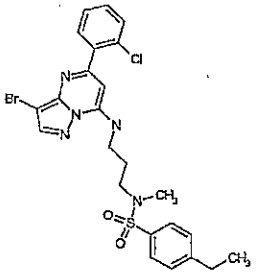
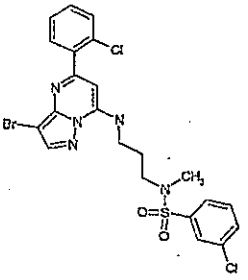
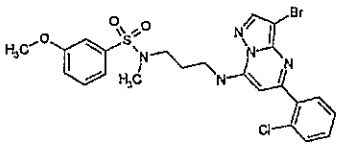
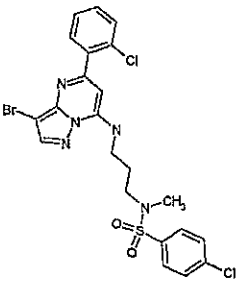
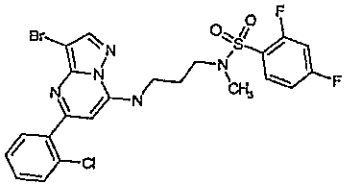
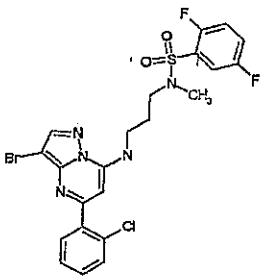
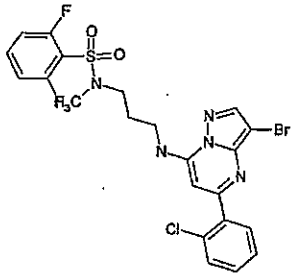
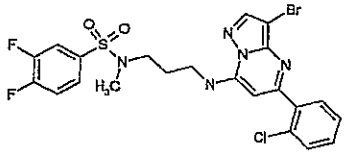
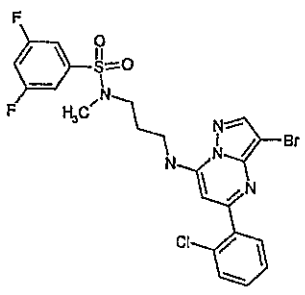
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	1. 6732 2. 576.32		1. 6737 2. 576.32		1. 6742 2 586.32
	1. 6733 2. 576.32		1. 6738 2. 576.32		1. 6743 2. 592.33
	1. 6734 2. 576.32		1. 6739 2. 584.32		1. 6744 2. 592.33
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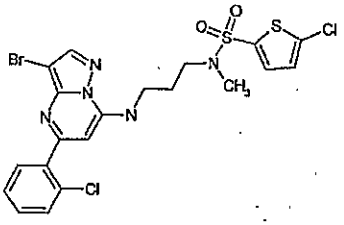
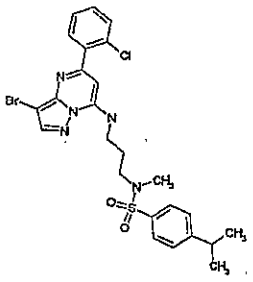
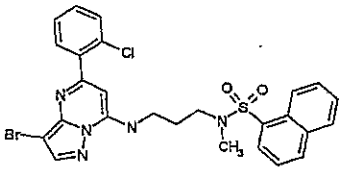
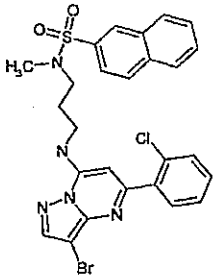
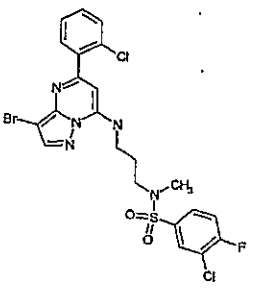
TABLE 68

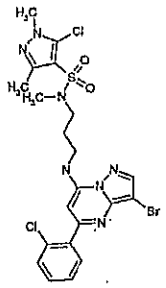
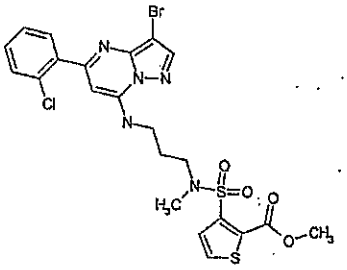
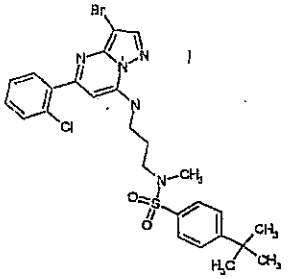
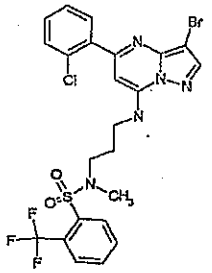
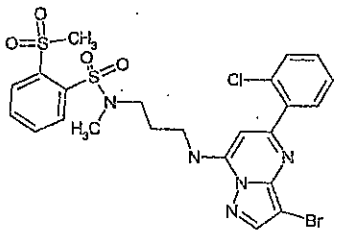
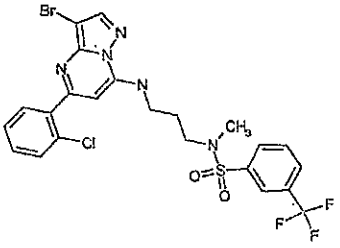
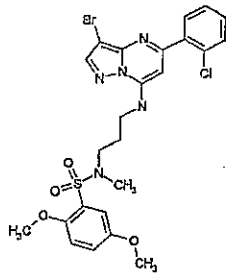
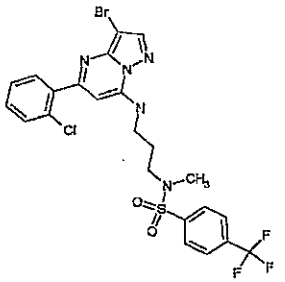
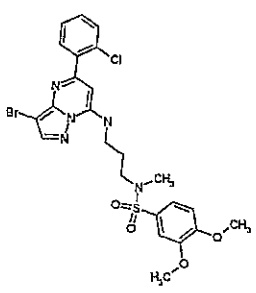
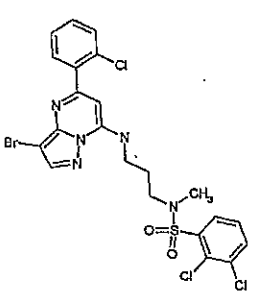
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	1. 6802 2. 502.28		1. 6807 2. 554.3
	1. 6803 2. 542.3		1. 6808 2. 554.3
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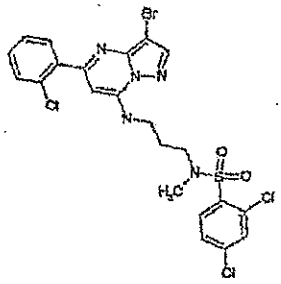
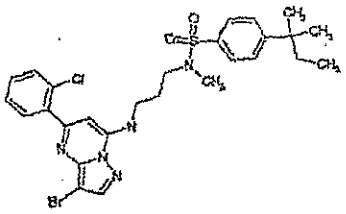
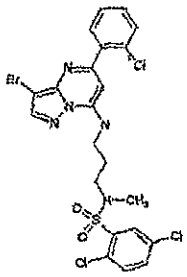
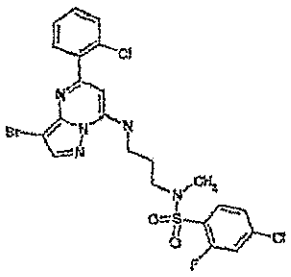
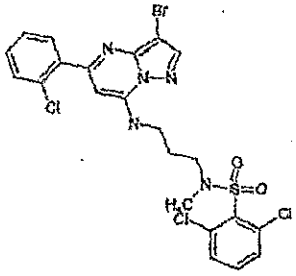
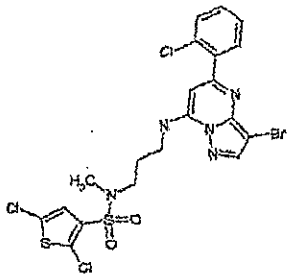
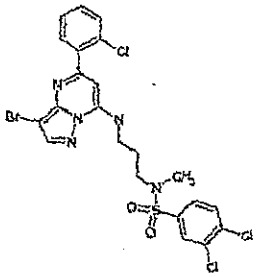
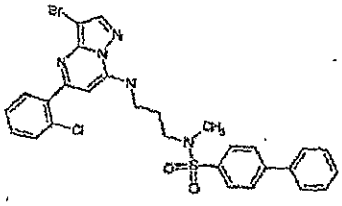
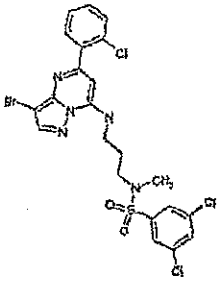
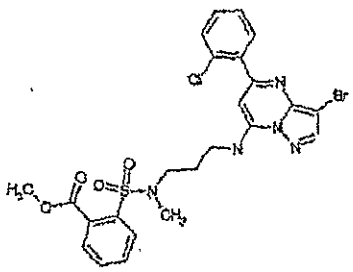
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	1. 6812 2. 561.31		1. 6817 2. 568.31
	1. 6813 2. 562.31		1. 6818 2. 570.31
	1. 6814 2. 564.31		1. 6819 2. 570.31
	1. 6815 2. 566.31		1. 6820 2. 570.31

Product	1. Ex. 2. m/z
	1. 6821 2. 571.31
	1. 6822 2. 572.31
	1. 6823 2. 572.31
	1. 6824 2. 572.31
	1. 6825 2. 572.31

Product	1. Ex. 2. m/z
	1. 6826 2. 576.32
	1. 6827 2. 578.32
	1. 6828 2. 586.32
	1. 6829 2. 586.32
	1. 6830 2. 588.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6831 2. 588.32		1. 6836 2. 600.33
	1. 6832 2. 592.33		1. 6837 2. 602.33
	1. 6833 2. 614.34		1. 6838 2. 604.33
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Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
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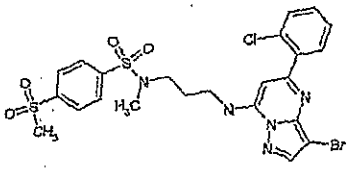
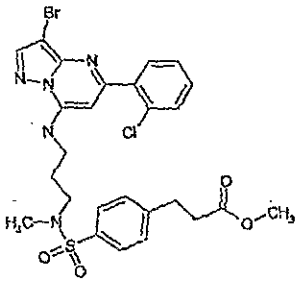
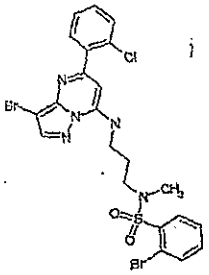
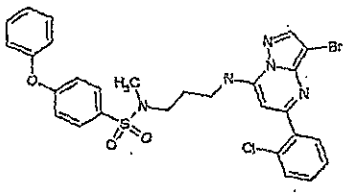
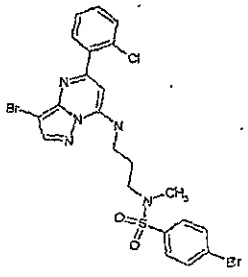
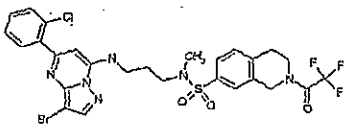
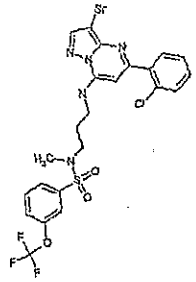
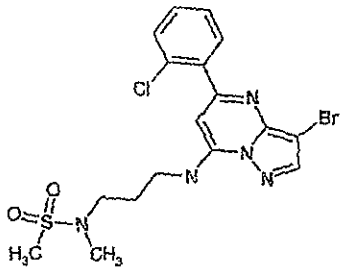
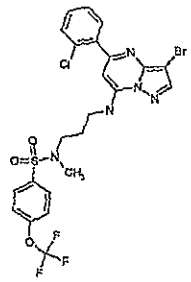
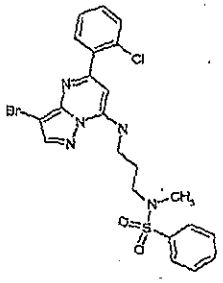
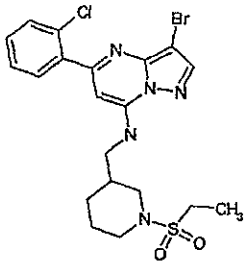
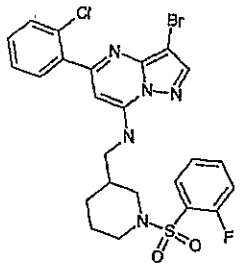
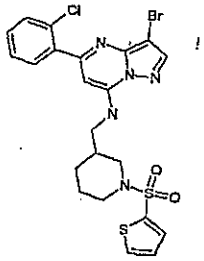
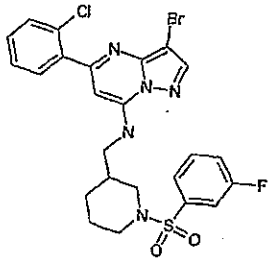
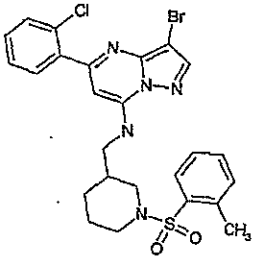
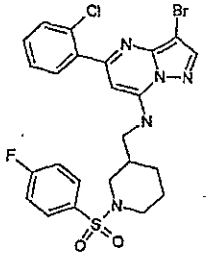
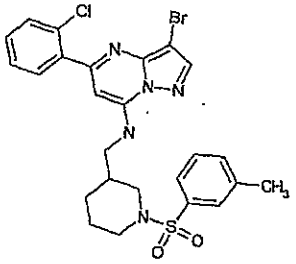
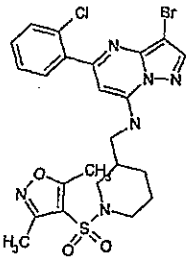
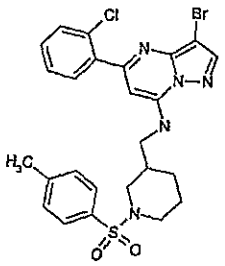
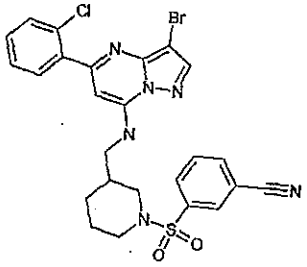
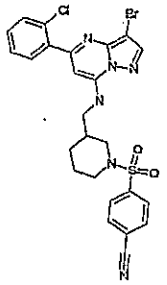
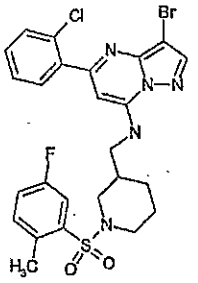
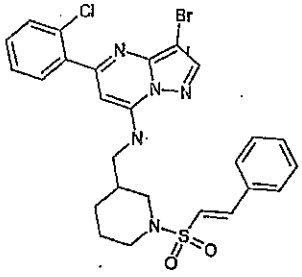
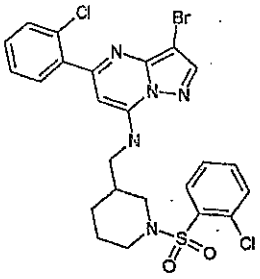
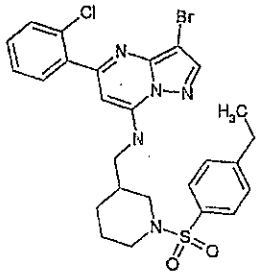
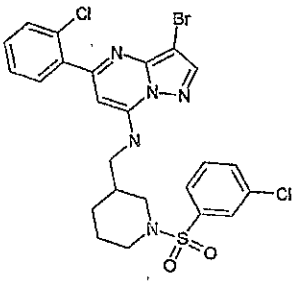
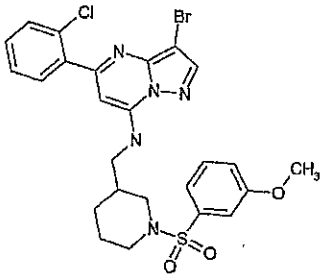
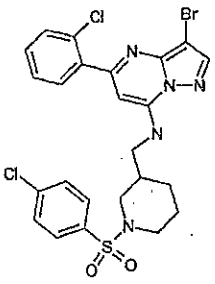
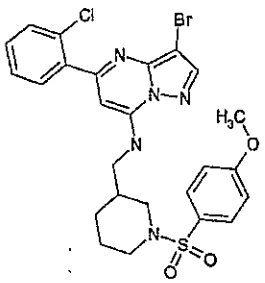
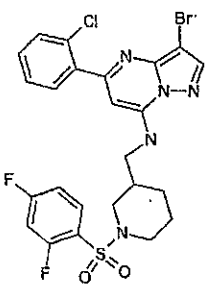
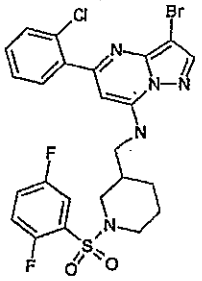
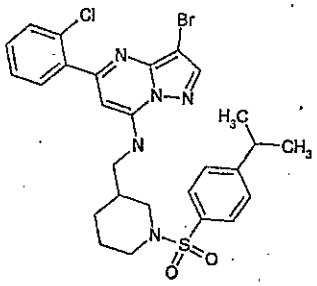
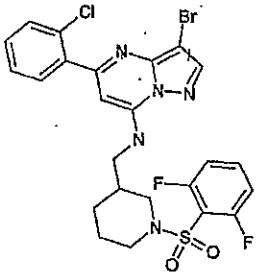
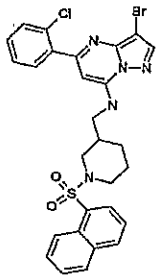
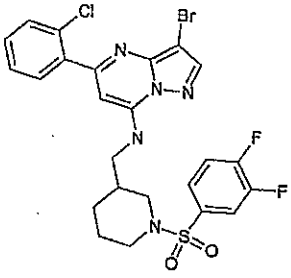
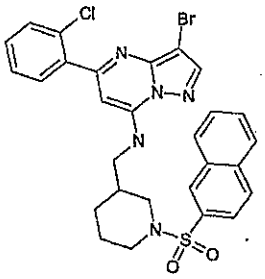
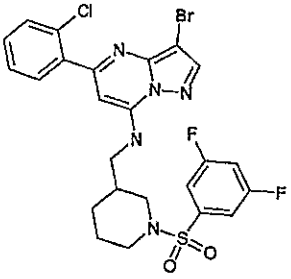
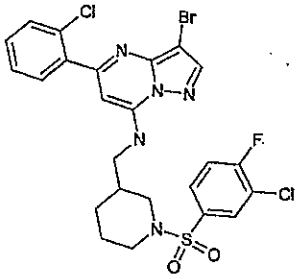
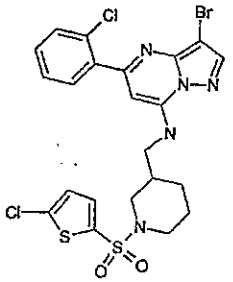
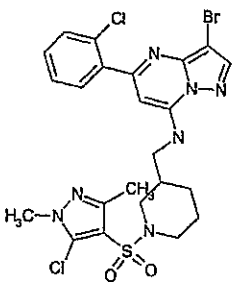
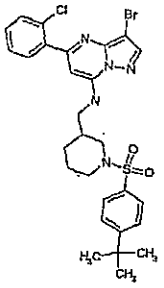
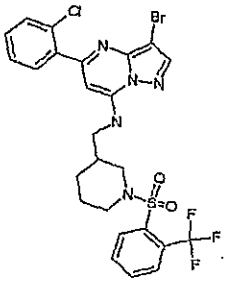
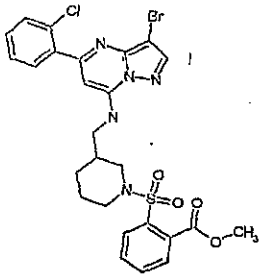
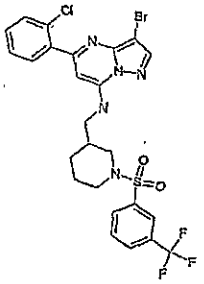
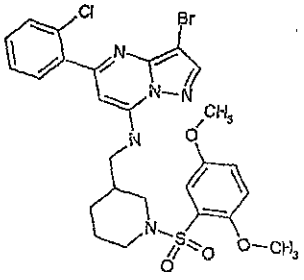
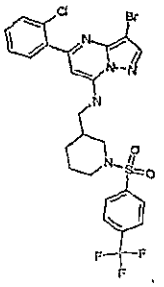
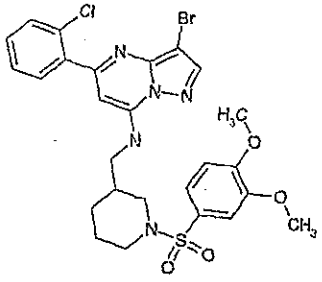
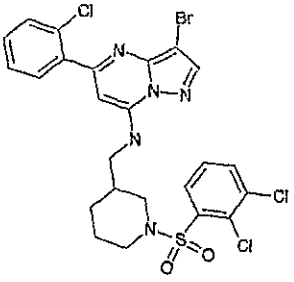
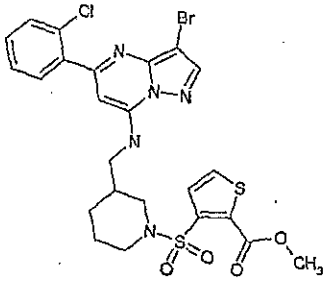
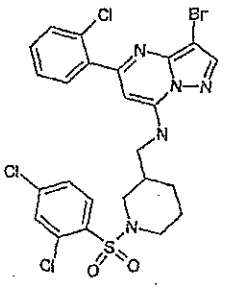
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	1. 6852 2. 614.34		1. 6856 2. 628.35
	1. 6853 2. 614.34		1. 6857 2. 687.38
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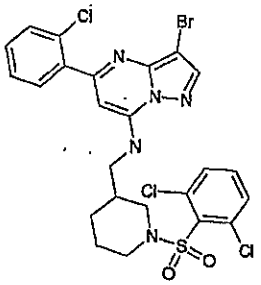
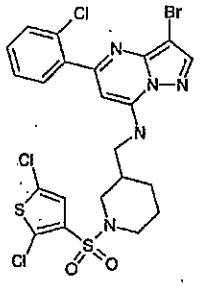
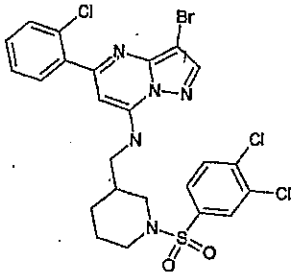
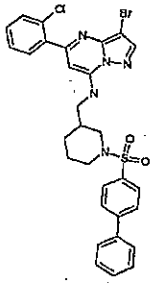
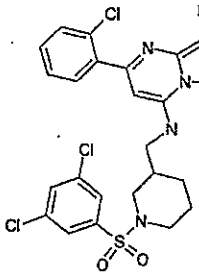
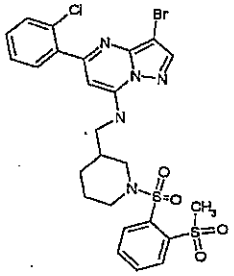
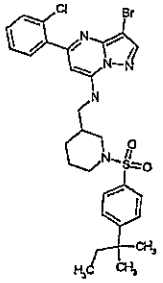
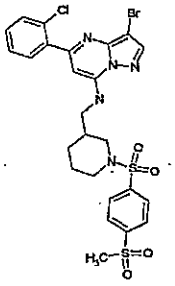
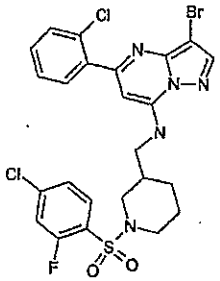
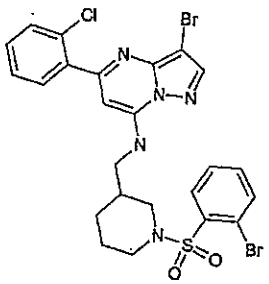
TABLE 69

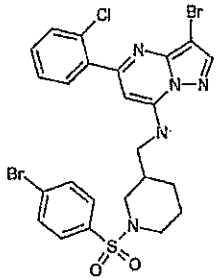
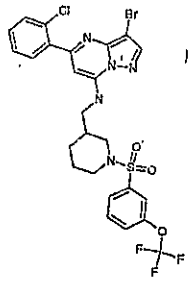
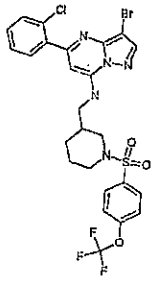
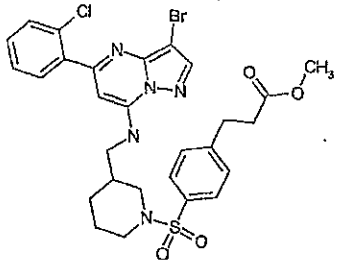
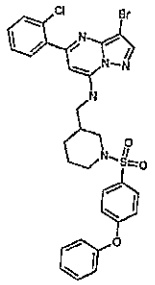
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	1. 6902 2. 568.31		1. 6907 2. 580.32
	1. 6903 2. 576.32		1. 6908 2. 580.32
	1. 6904 2. 576.32		1. 6909 2. 581.32
	1. 6905 2. 576.32		1. 6910 2. 587.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6911 2. 587.32		1. 6916 2. 594.33
	1. 6912 2. 588.32		1. 6917 2. 596.33
	1. 6913 2. 590.32		1. 6918 2. 596.33
	1. 6914 2. 592.33		1. 6919 2. 596.33
	1. 6915 2. 592.33		1. 6920 2. 598.33

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6921 2. 598.33		1. 6926 2. 604.33
	1. 6922 2. 598.33		1. 6927 2. 612.34
	1. 6923 2. 598.33		1. 6928 2. 612.34
	1. 6924 2. 598.33		1. 6929 2. 614.34
	1. 6925 2. 602.33		1. 6930 2. 614.34

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6931 2. 618.34		1. 6936 2. 630.35
	1. 6932 2. 620.34		1. 6937 2. 630.35
	1. 6933 2. 622.34		1. 6938 2. 630.35
	1. 6934 2. 622.34		1. 6939 2. 630.35
	1. 6935 2. 626.34		1. 6940 2. 630.35

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 6941 2. 630.35		1. 6946 2. 636.35
	1. 6942 2. 630.35		1. 6947 2. 638.35
	1. 6943 2. 630.35		1. 6948 2. 640.35
	1. 6944 2. 632.35		1. 6949 2. 640.35
	1. 6945 2. 614.34		1. 6950 2. 640.35

Product	1. Ex. 2. m/z
	1. 6951 2. 640.35
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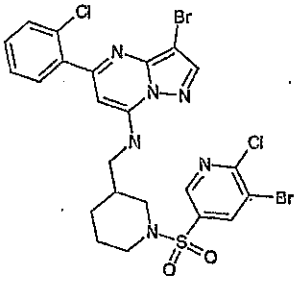
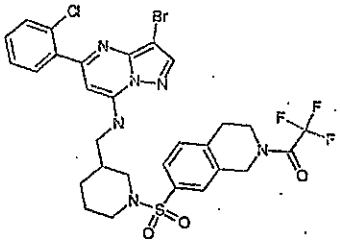
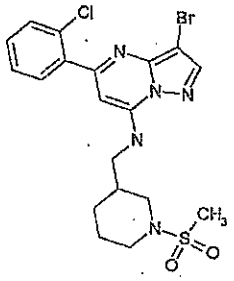
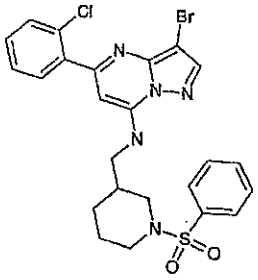
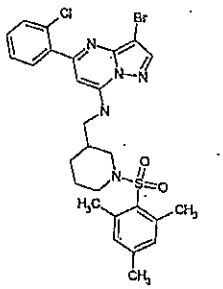
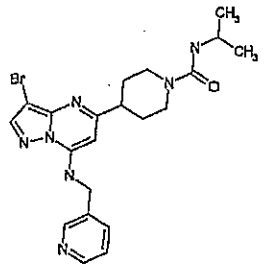
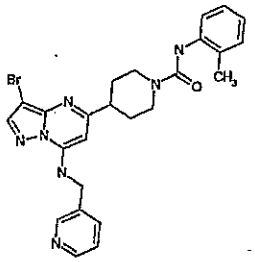
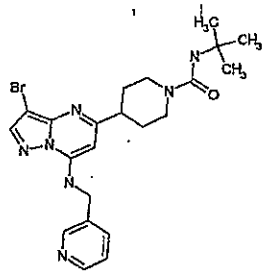
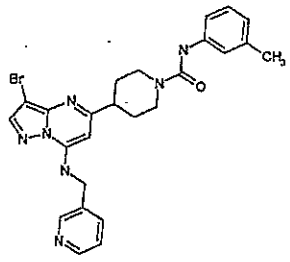
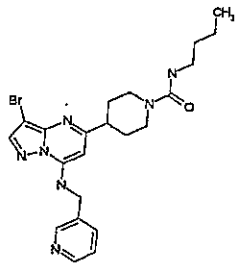
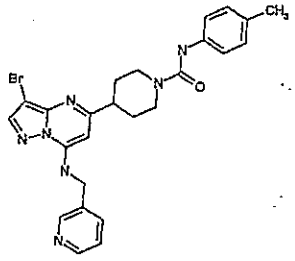
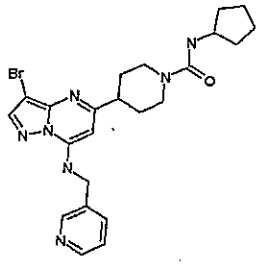
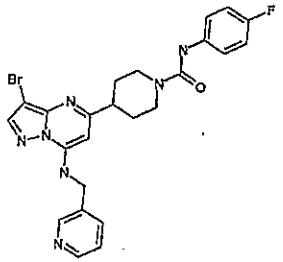
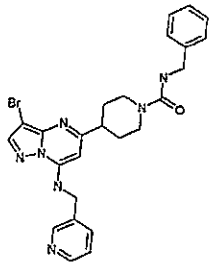
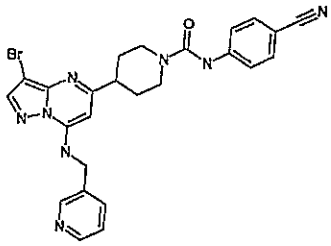
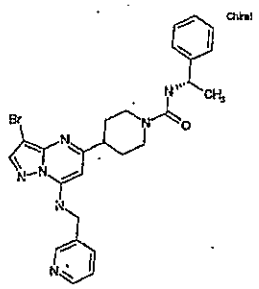
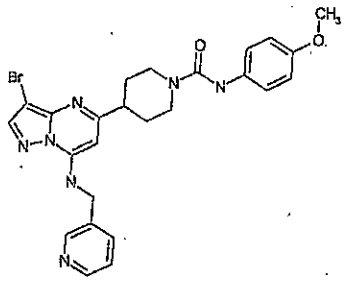
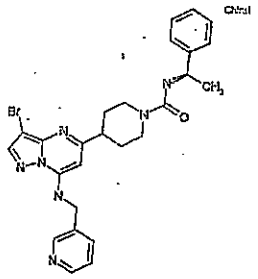
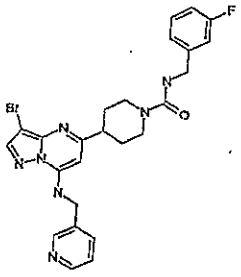
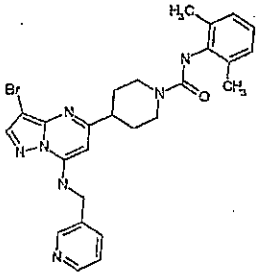
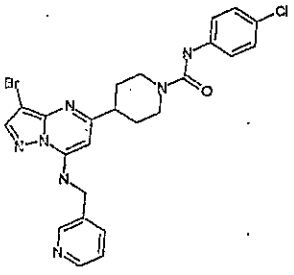
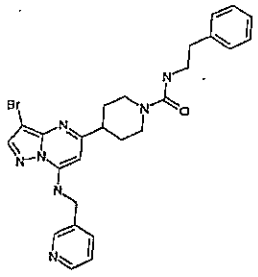
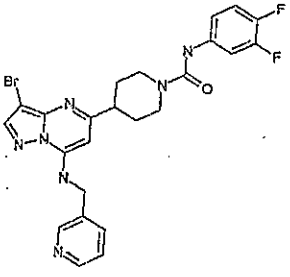
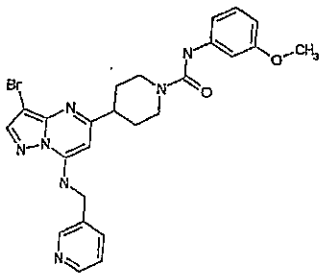
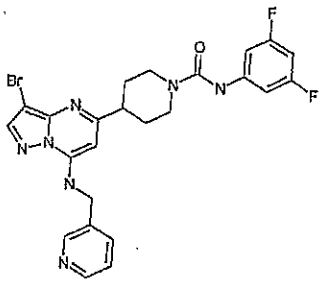
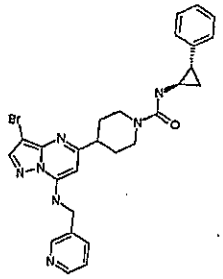
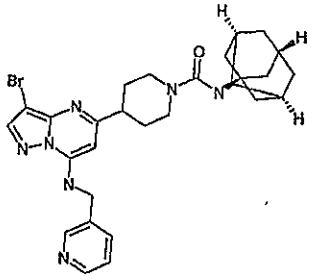
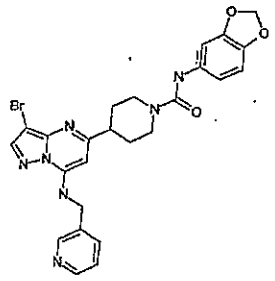
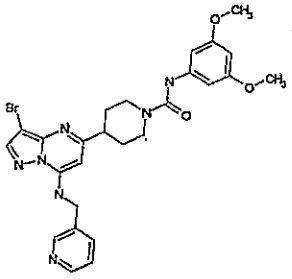
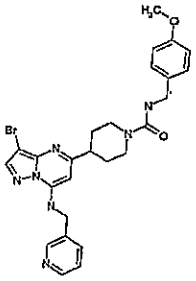
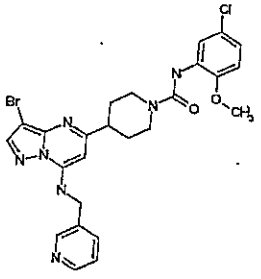
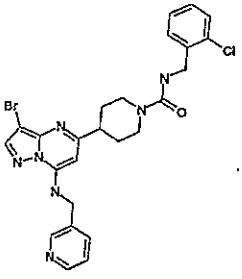
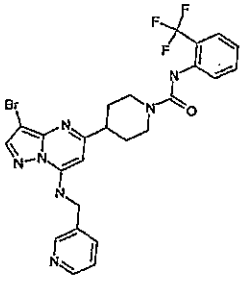
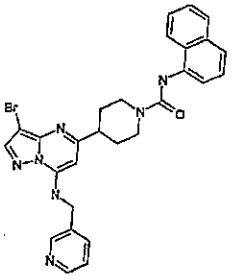
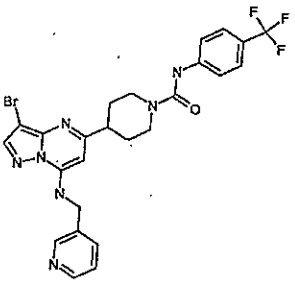
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	1. 6956 2. 675.23
	1. 6957 2. 713.39
	1. 6958 2. 500.27
	1. 6959 2. 562.31
	1. 6960 2. 604.33

TABLE 70

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7001 2. 472.26		1. 7006 2. 522.29
	1. 7002 2. 488.27		1. 7007 2. 520.29
	1. 7003 2. 488.27		1. 7008 2. 520.29
	1. 7004 2. 500.27		1. 7009 2. 526.29
	1. 7005 2. 520.29		1. 7010 2. 533.29

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7011 2. 536.29		1. 7016 2. 538.3
	1. 7012 2. 536.29		1. 7017 2. 540.3
	1. 7013 2. 536.29		1. 7018 2. 542.3
	1. 7014 2. 536.29		1. 7019 2. 544.3
	1. 7015 2. 536.29		1. 7020 2. 544.3

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7021 2. 548.3		1. 7026 2. 566.31
	1. 7022 2. 552.3		1. 7027 2. 568.31
	1. 7023 2. 552.3		1. 7028 2. 572.31
	1. 7024 2. 556.31		1. 7029 2. 576.32
	1. 7025 2. 558.31		1. 7030 2. 576.32

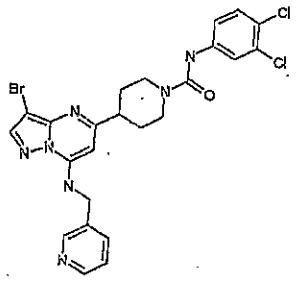
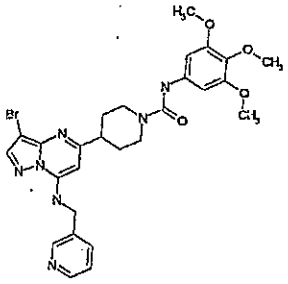
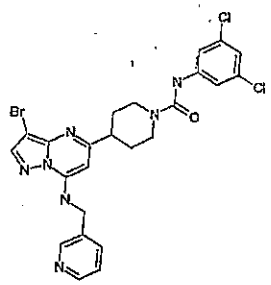
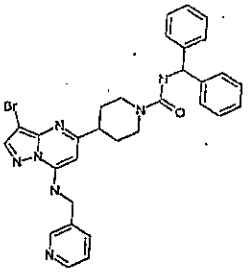
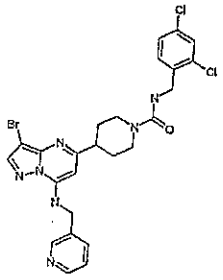
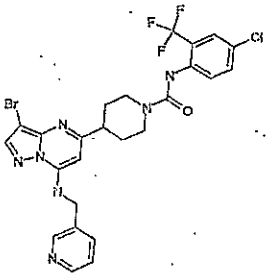
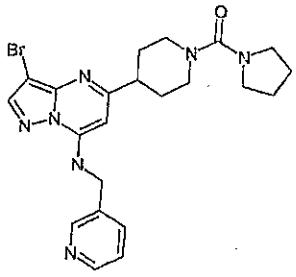
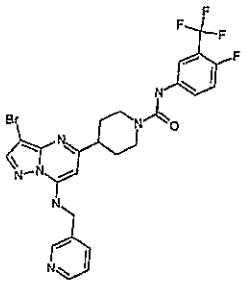
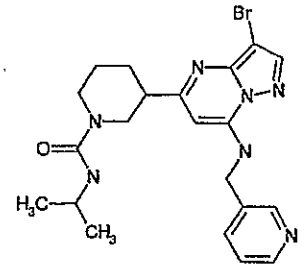
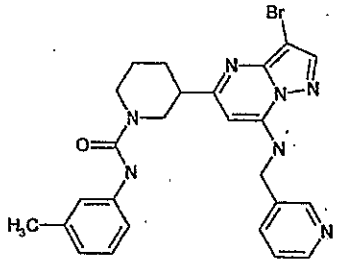
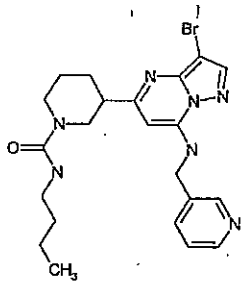
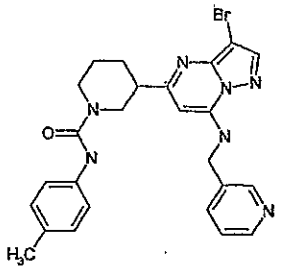
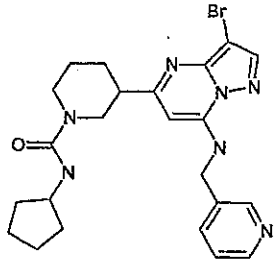
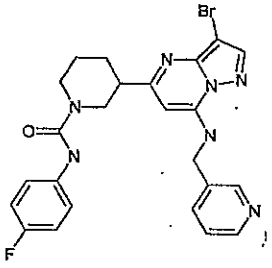
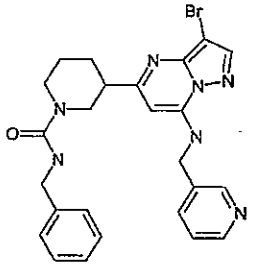
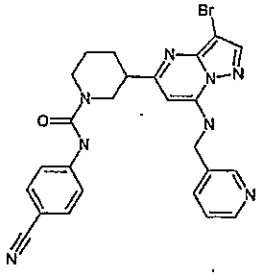
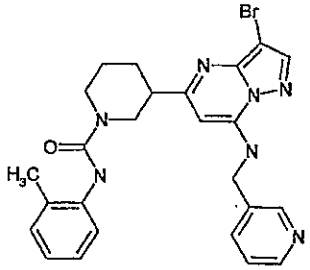
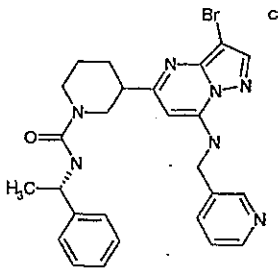
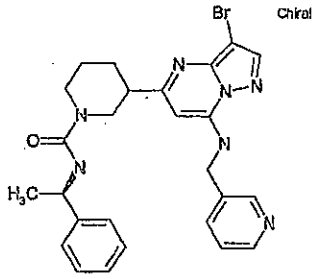
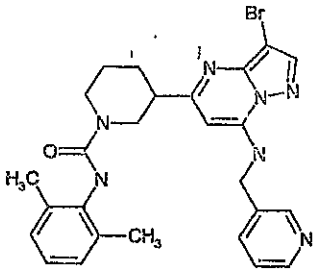
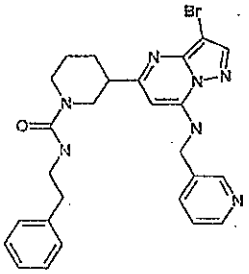
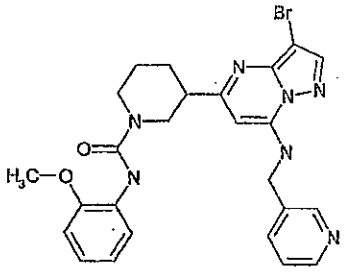
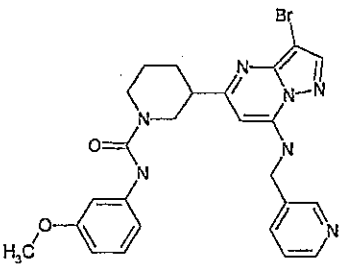
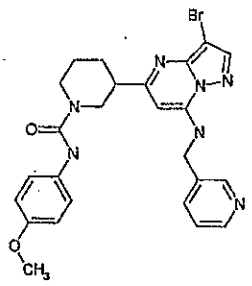
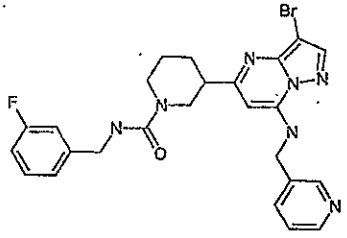
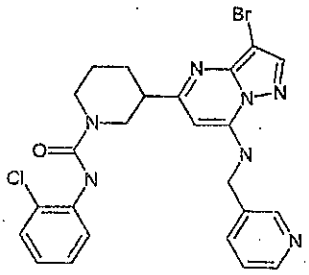
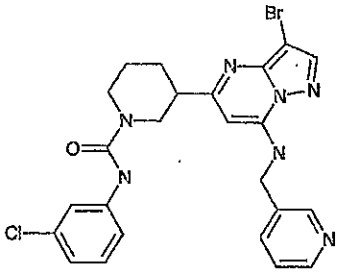
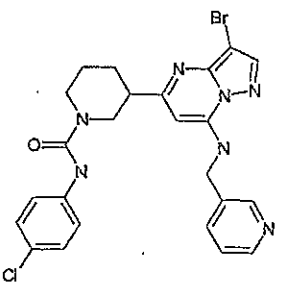
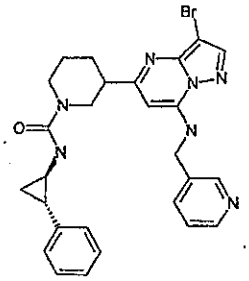
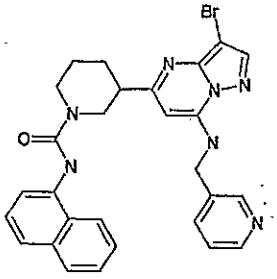
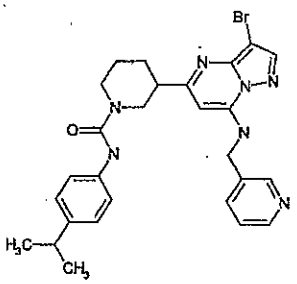
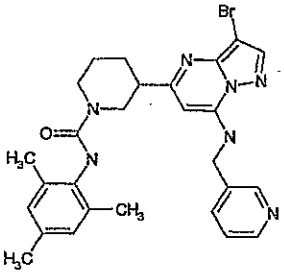
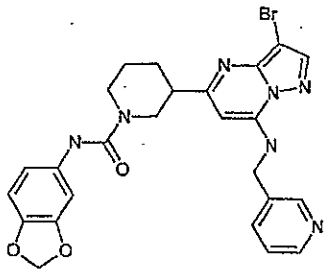
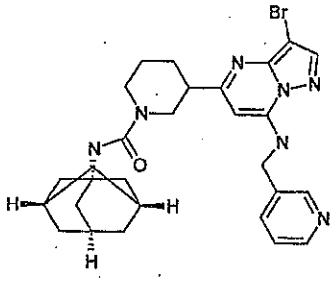
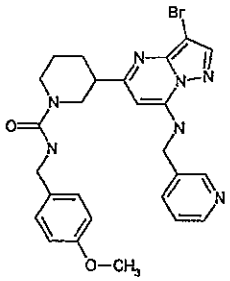
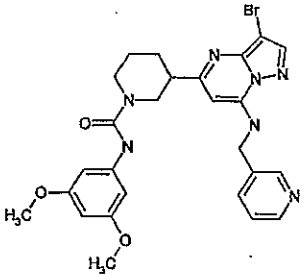
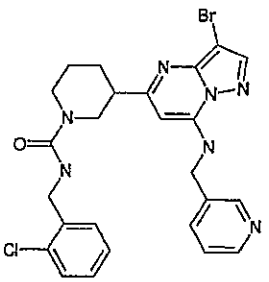
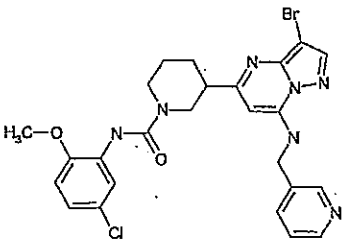
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	1. 7031 2. 576.32		1. 7036 2. 598.33
	1. 7032 2. 576.32		1. 7037 2. 598.33
	1. 7033 2. 590.32		1. 7038 2. 610.34
	1. 7034 2. 486.27		
	1. 7035 2. 594.33		

TABLE 71

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1 . 7101 2. 474.26		1. 7106 2. 522.29
	1. 7102 2. 488.27		1. 7107 2. 522.29
	1. 7103 2. 500.27		1. 7108 2. 526.3
	1. 7104 2. 522.29		1. 7109 2. 533.29
	1. 7105 2. 522.29	 Chiral	1. 7110 2. 536.29

Product	1. Ex. 2. m/z
	1. 7111 2. 536.29
	1. 7112 2. 536.29
	1. 7113 2. 536.29
	1. 7114 2. 536.29
	1. 7115 2. 536.3

Product	1. Ex. 2. m/z
	1. 7116 2. 538.3
	1. 7117 2. 540.3
	1. 7118 2. 542.3
	1. 7119 2. 542.3
	1. 7120 2. 542.3

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7121 2. 546.3		1. 7126 2. 558.31
	1. 7122 2. 550.3		1. 7127 2. 550.3
	1. 7123 2. 552.3		1. 7128 2. 566.31
	1. 7124 2. 552.3		1. 7129 2. 565.31
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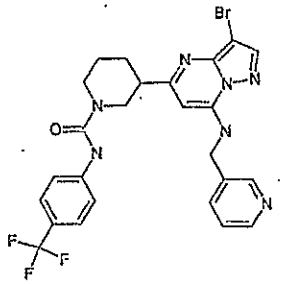
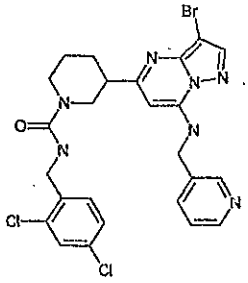
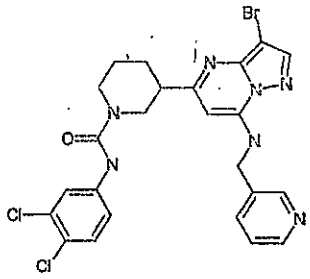
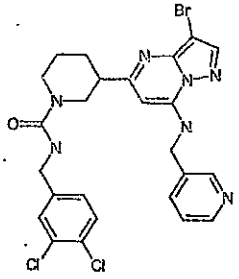
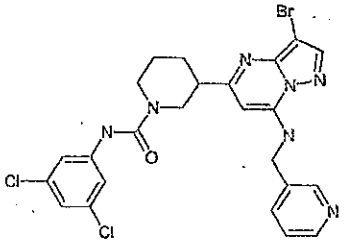
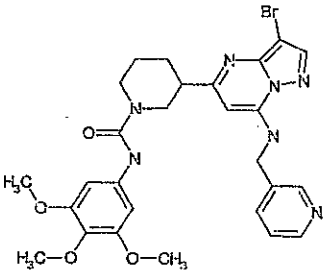
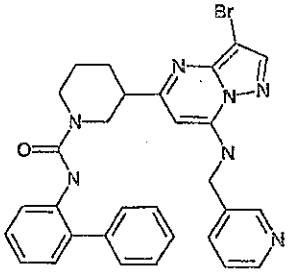
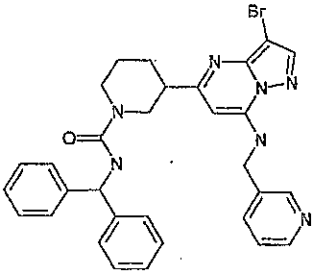
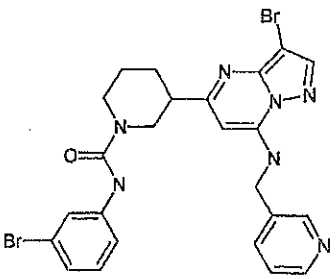
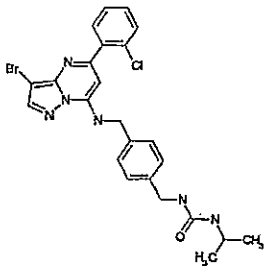
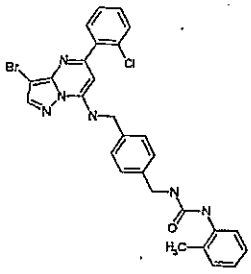
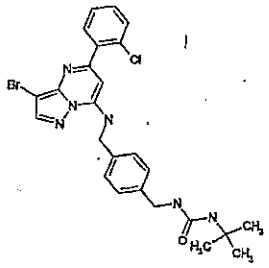
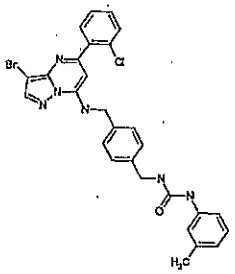
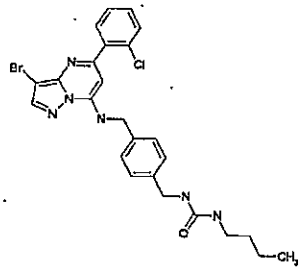
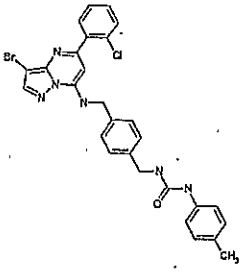
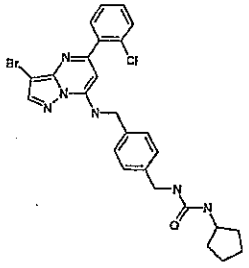
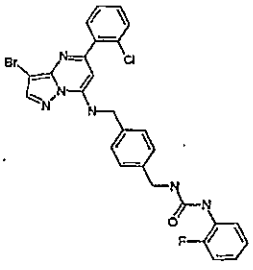
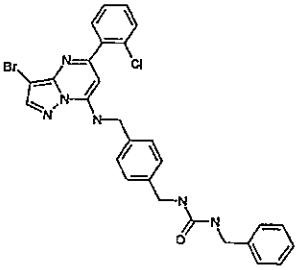
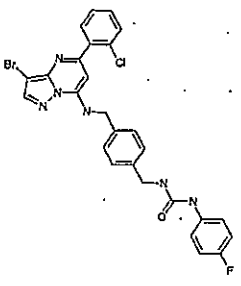
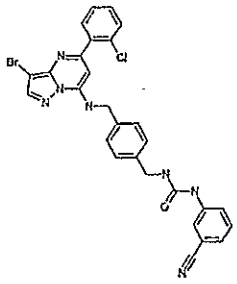
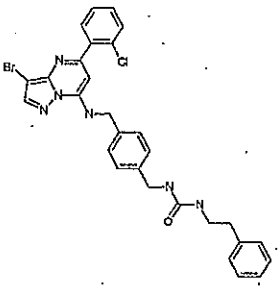
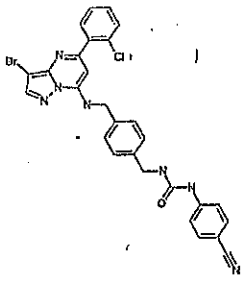
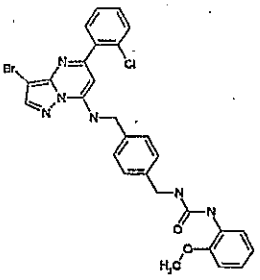
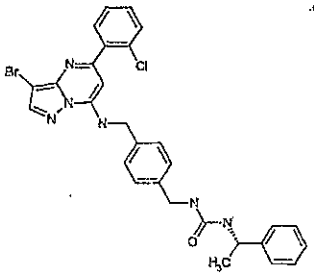
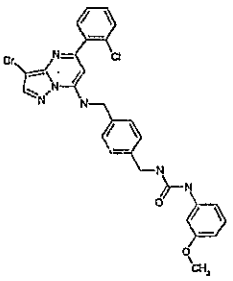
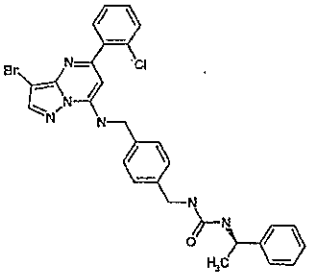
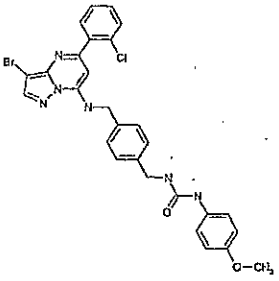
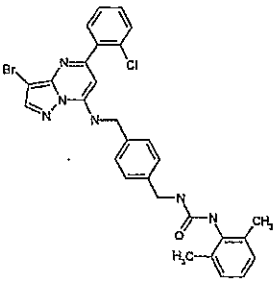
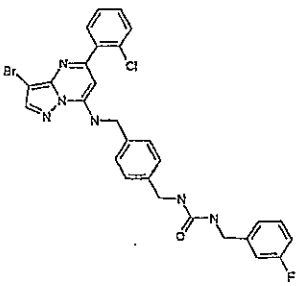
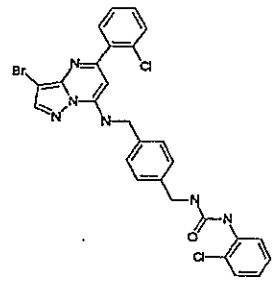
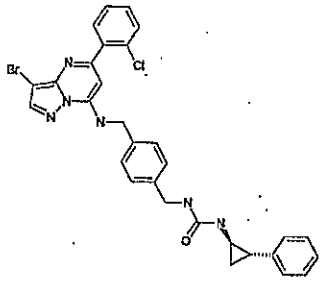
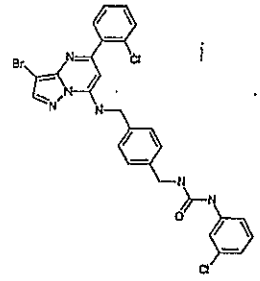
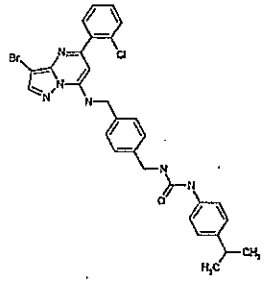
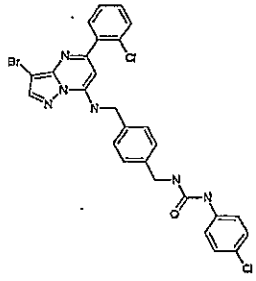
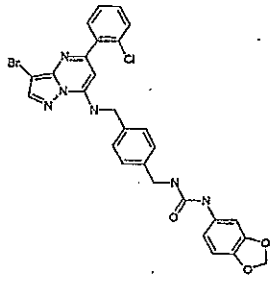
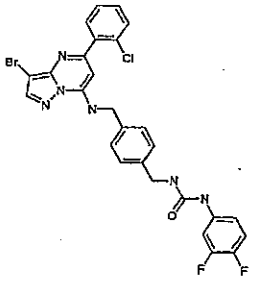
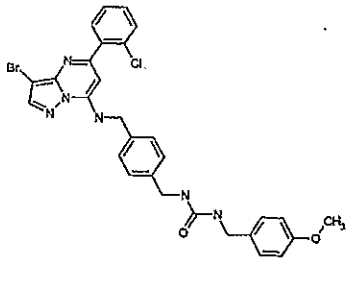
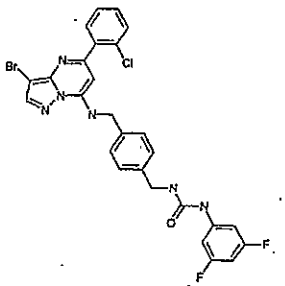
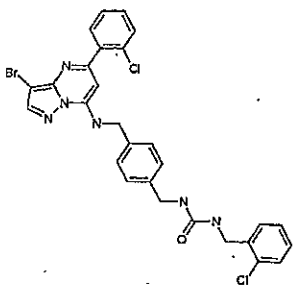
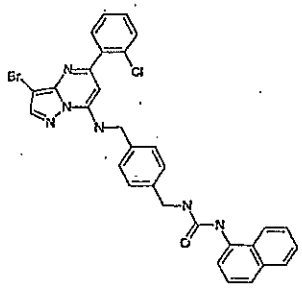
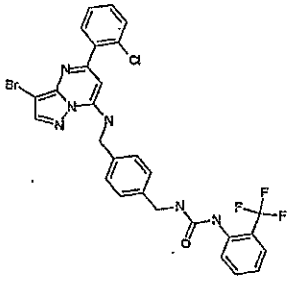
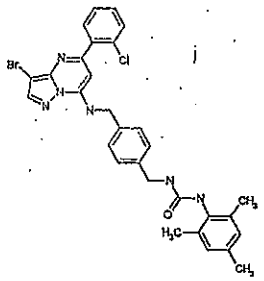
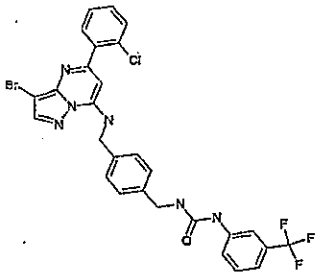
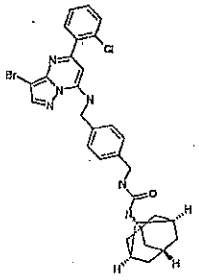
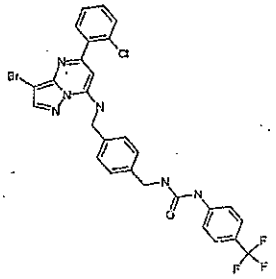
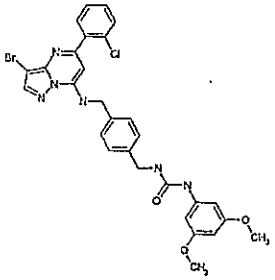
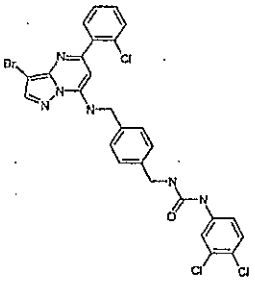
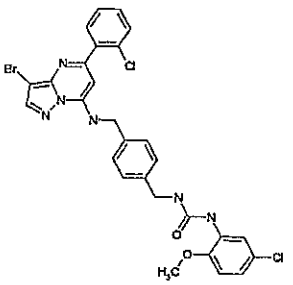
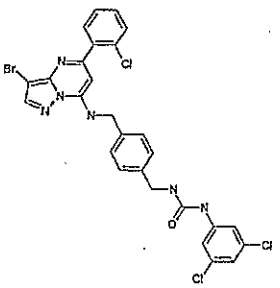
Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7131 2. 576.32		1. 7136 2. 590.32
	1. 7132 2. 576.32		1. 7137 2. 590.32
	1. 7133 2. 576.32		1. 7138 2. 598.33
	1. 7134 2. 584.32		1. 7139 2. 598.33
	1. 7135 2. 585.3		

TABLE 72

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7201 2. 529.29		1. 7206 2. 577.32
	1. 7202 2. 543.3		1. 7207 2. 577.32
	1. 7203 2. 543.3		1. 7208 2. 577.32
	1. 7204 2. 555.31		1. 7209 2. 581.32
	1. 7205 2. 577.32		1. 7210 2. 581.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7211 2. 588.32		1. 7216 2. 591.33
	1. 7212 2. 588.32		1. 7217 2. 593.33
	1. 7213 2. 591.33		1. 7218 2. 593.33
	1. 7214 2. 591.33		1. 7219 2. 593.33
	1. 7215 2. 591.33		1. 7220 2. 595.33

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7221 2. 597.33		1. 7226 2. 603.33
	1. 7222 2. 597.33		1. 7227 2. 603.33
	1. 7223 2. 597.33		1. 7228 2. 607.33
	1. 7224 2. 599.33		1. 7229 2. 607.14
	1. 7225 2. 599.33		1. 7230 2. 611.34

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7231 2. 613.34		1. 7236 2. 631.35
	1. 7232 2. 605.33		1. 7237 2. 631.35
	1. 7233 2. 621.34		1. 7238 2. 631.35
	1. 7234 2. 623.34		1. 7239 2. 631.35
	1. 7235 2. 627.34		1. 7240 2. 631.35

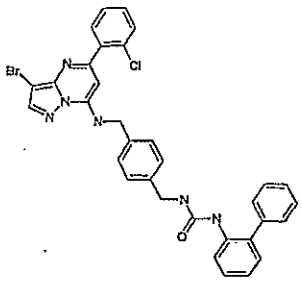
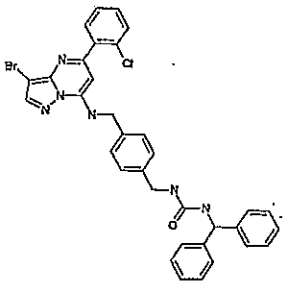
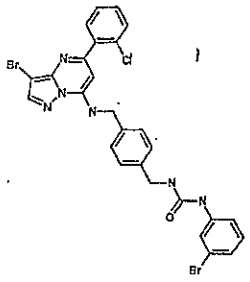
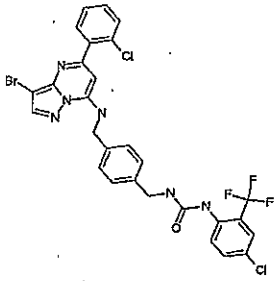
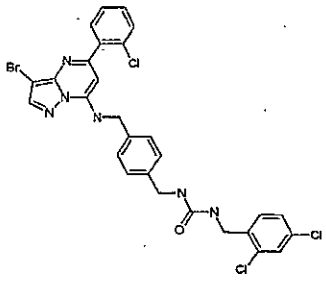
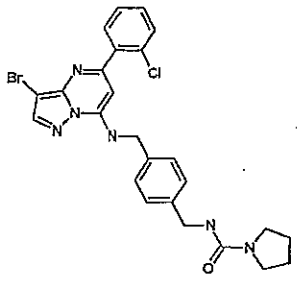
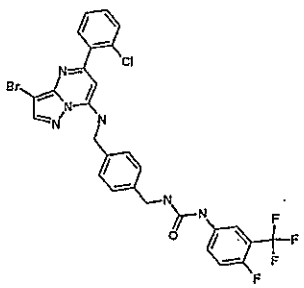
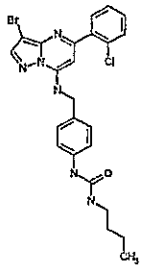
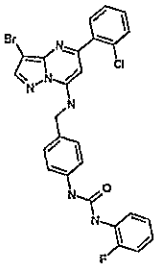
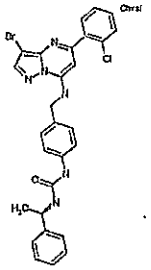
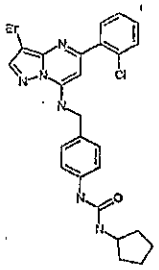
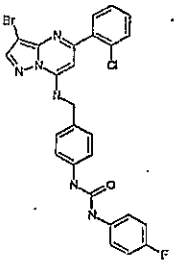
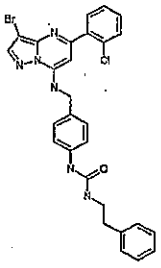
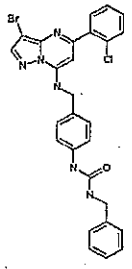
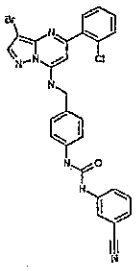
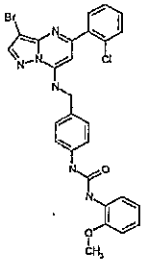
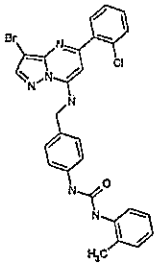
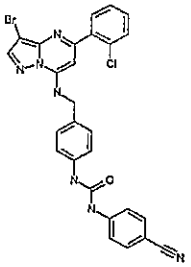
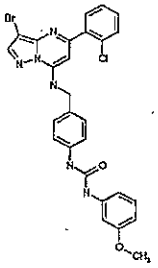
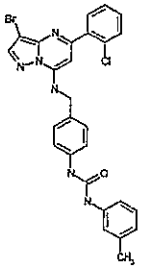
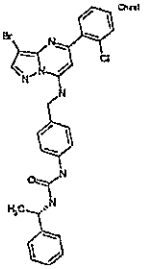
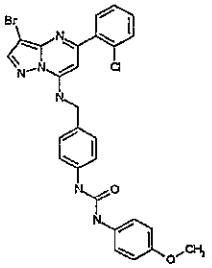
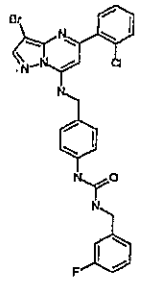
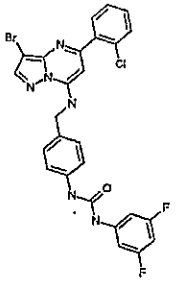
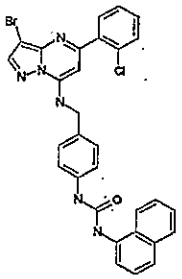
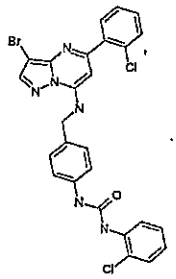
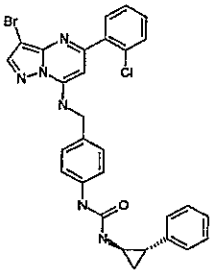
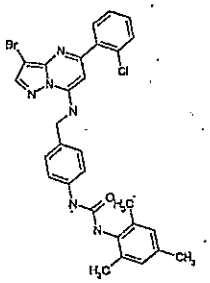
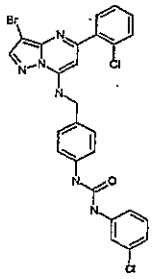
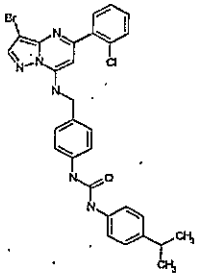
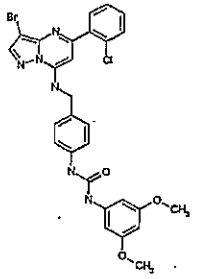
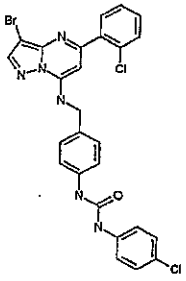
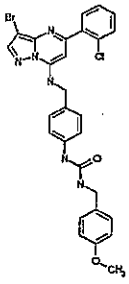
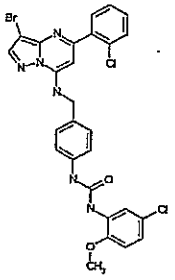
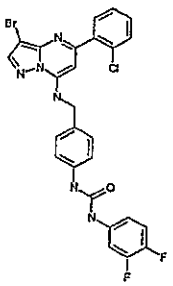
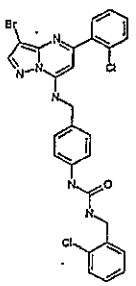
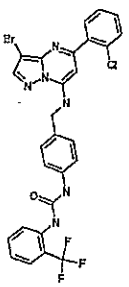
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	1. 7241 2. 639.35		1. 7246 2. 653.36
	1. 7242 2. 641.35		1. 7247 2. 665.37
	1. 7243 2. 645.35		
	1. 7244 2. 541.3		
	1. 7245 2. 649.36		

TABLE 73

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7301 2. 529.29		1. 7306 2. 567.31		1. 7311 2. 577.32
	1. 7302 2. 541.3		1. 7307 2. 567.31		1. 7312 2. 577.32
	1. 7303 2. 563.13		1. 7308 2. 574.32		1. 7313 2. 579.32
	1. 7304 2. 563.31		1. 7309 2. 574.32		1. 7314 2. 579.32
	1. 7305 2. 563.31		1. 7310 2. 577.32		1. 7315 2. 579.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7316 2. 581.32		1. 7321 2. 585.32		1. 7326 2. 599.33
	1. 7317 2. 583.32		1. 7322 2. 589.32		1. 7327 2. 591.33
	1. 7318 2. 583.32		1. 7323 2. 591.33		1. 7328
	1. 7319 2. 583.32		1. 7324 2. 593.33		1.7329 2. 613.34
	1. 7320 2. 585.32		1. 7325 2. 597.33		1. 7330 2. 617.34

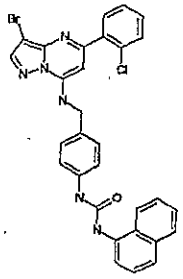
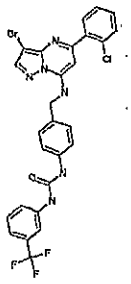
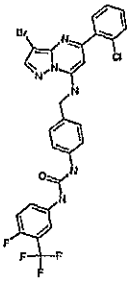
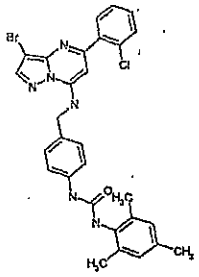
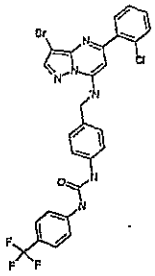
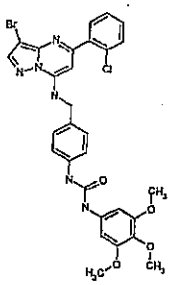
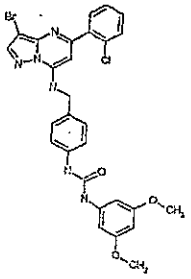
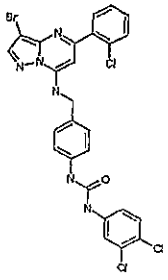
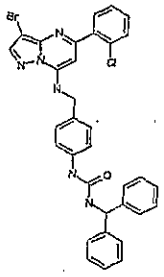
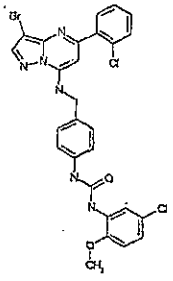
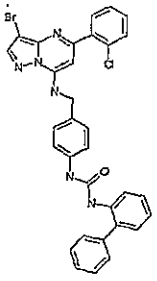
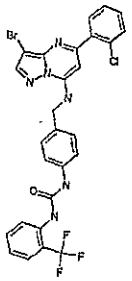
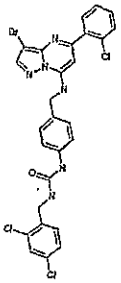
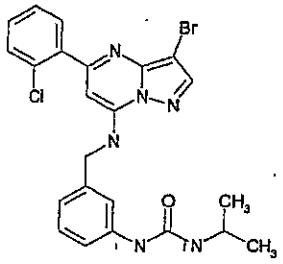
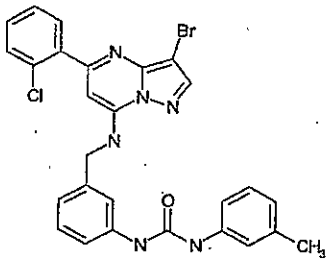
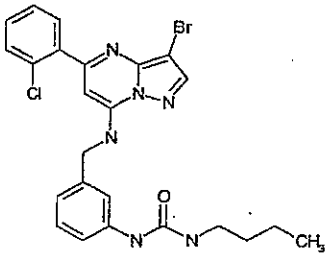
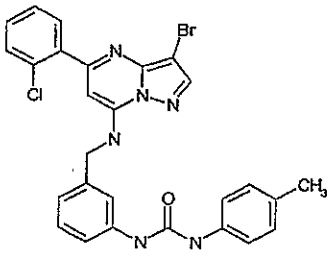
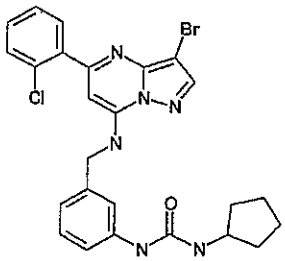
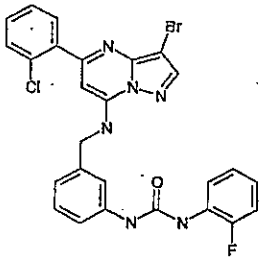
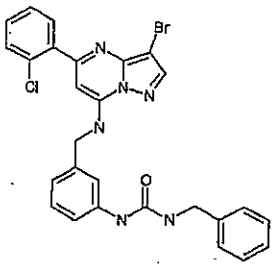
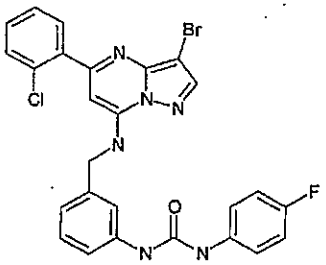
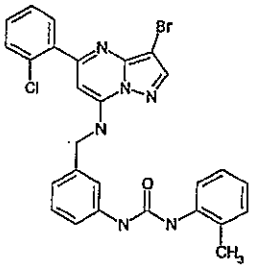
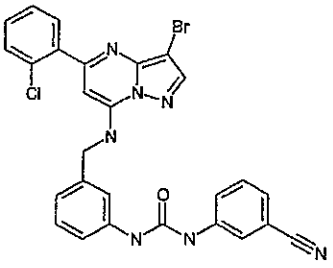
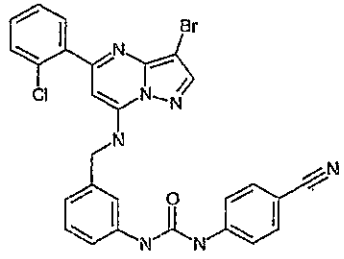
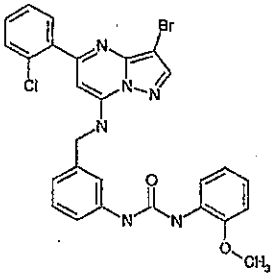
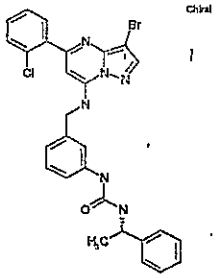
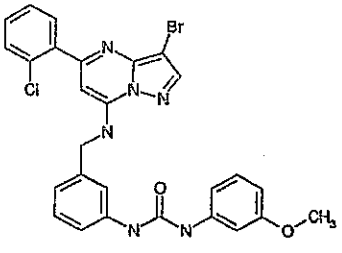
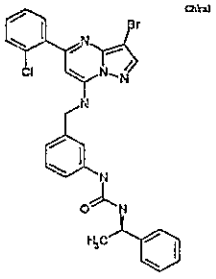
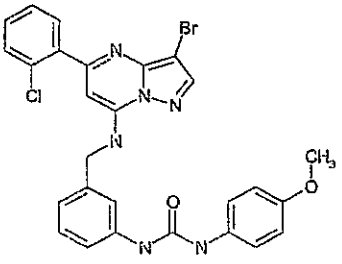
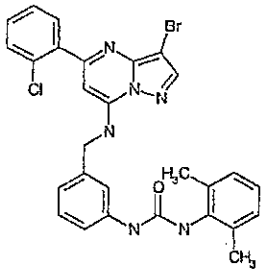
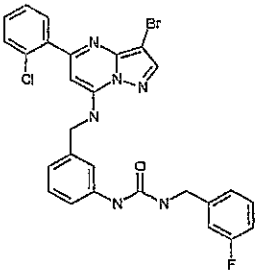
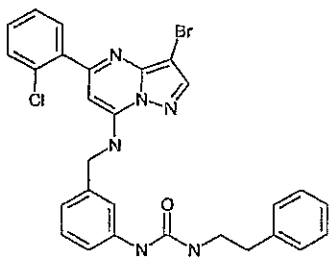
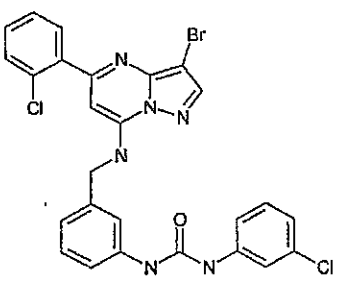
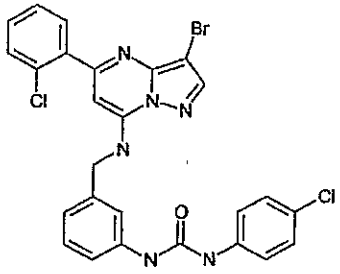
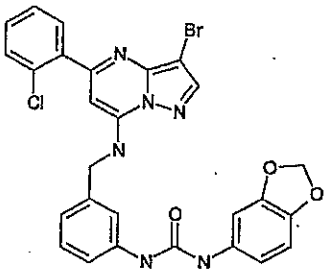
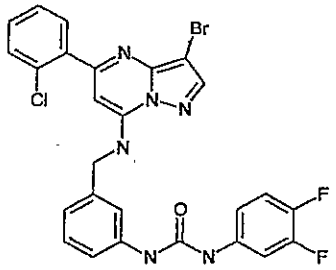
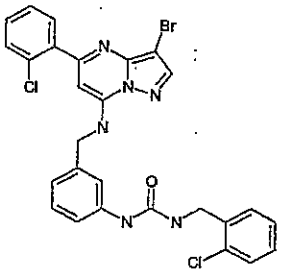
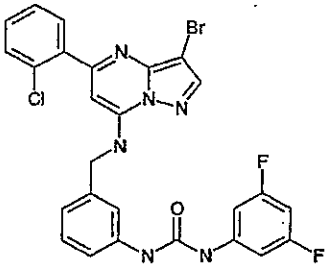
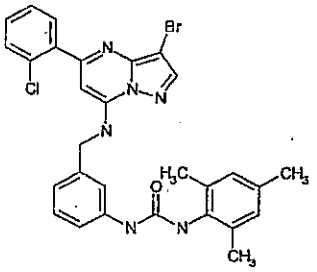
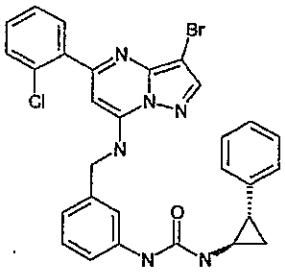
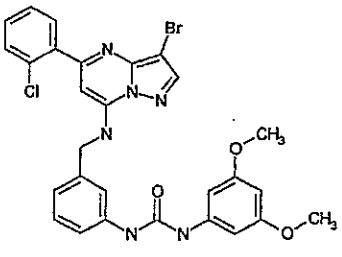
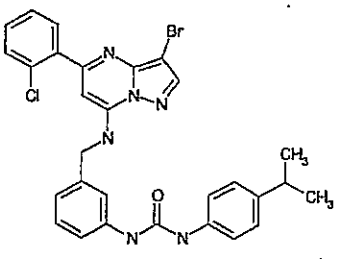
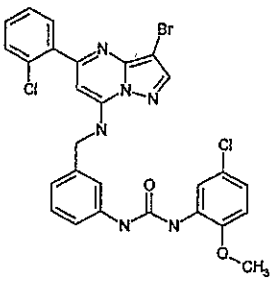
Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7331 2. 599.33		1. 7336 2. 617.34		1. 73412. 635.35
	1. 7332 2. 591.33		1. 7337 2. 617.34		1. 7342 2. 639.35
	1. 7333 2. 609.33		1. 7338 2. 617.34		1. 7343 2. 639.35
	1. 7334 2. 613.34		1. 7339 2. 625.34		
	1. 7335 2. 617.34		1. 7340 2. 631.35		

TABLE 74

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7401 2. 515.28		1. 7406 2. 563.31
	1. 7402 2. 529.29		1. 7407 2. 563.31
	1. 7403 2. 541.3		1. 7408 2. 567.31
	1. 7404 2. 563.13		1. 7409 2. 567.31
	1. 7405 2. 563.31		1. 7410 2. 574.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7411 2. 574.32		1. 7416 2. 579.32
	1. 7412 2. 577.14		1. 741 7 2. 579.32
	1. 7413 2. 577.32		1. 7418 2. 579.32
	1. 7414 2. 577.32		1. 7419 2. 581.32
	1. 7415 2. 577.32		1. 7420 2. 583.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7421 2. 583.32		1. 7426 2. 593.33
	1. 7422 2. 585.32		1. 7427 2. 597.33
	1. 7423 2. 585.32		1. 7428 2. 591.33
	1. 7424 2. 589.32		1. 7429 2. 609.33
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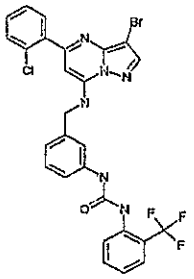
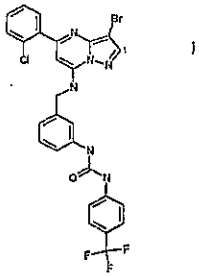
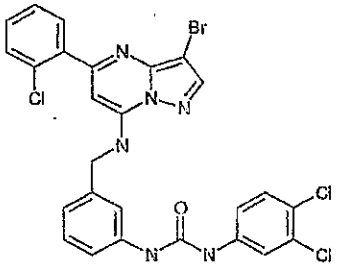
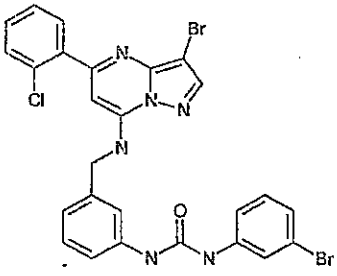
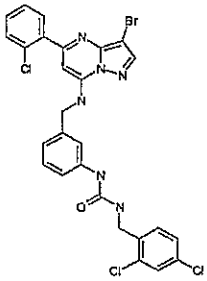
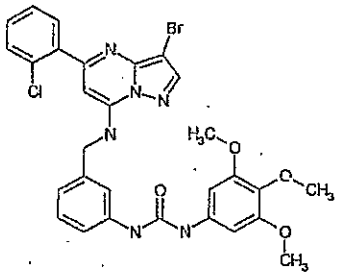
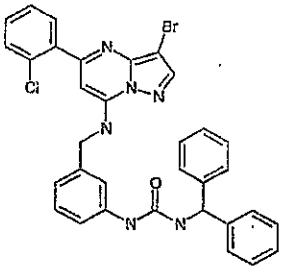
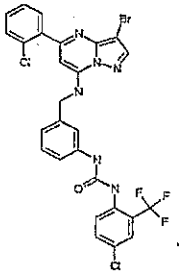
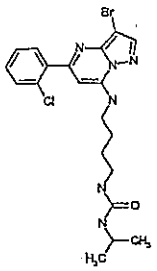
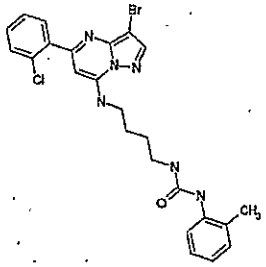
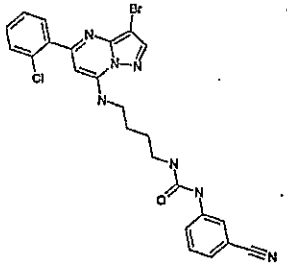
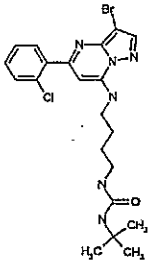
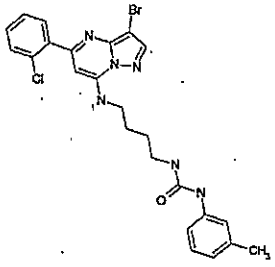
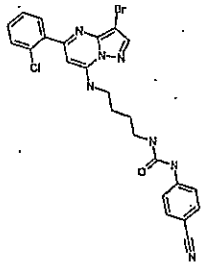
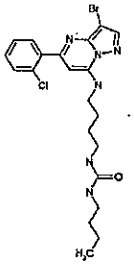
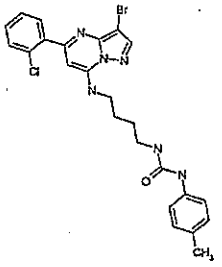
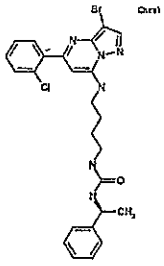
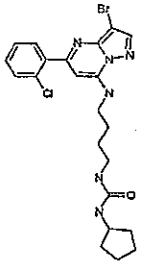
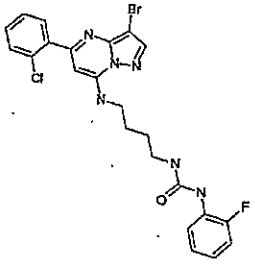
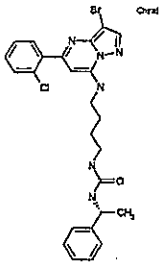
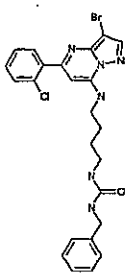
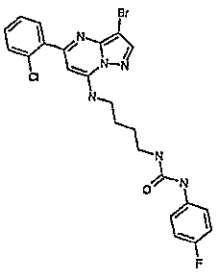
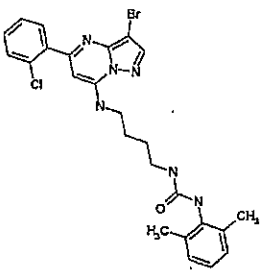
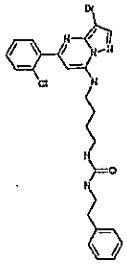
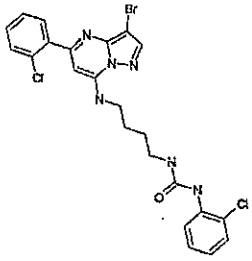
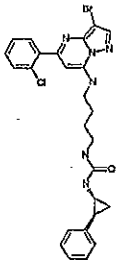
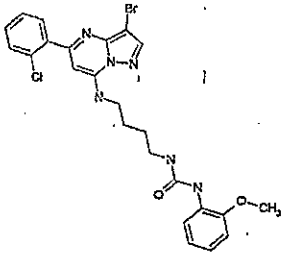
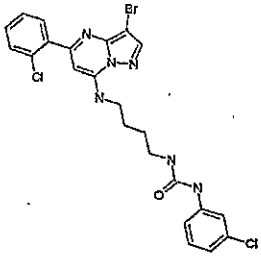
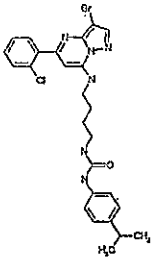
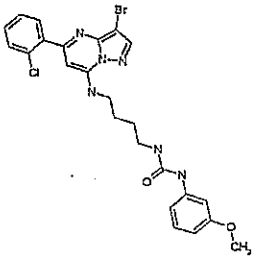
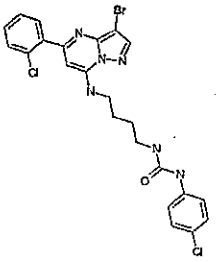
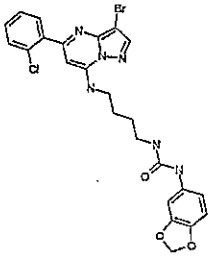
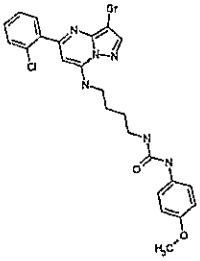
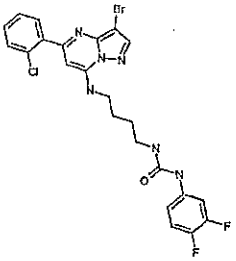
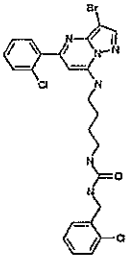
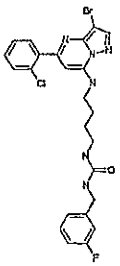
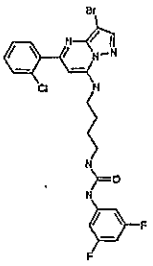
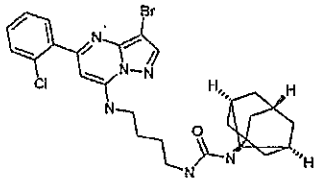
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	1. 7431 2. 617.34
	1. 7432 2. 617.34
	1. 7433 2. 617.34
	1. 7434 2. 627.34
	1. 7435 2. 631.35
	1. 7436 2. 639.35
	1. 7437 2. 639.35
	1. 7438 2. 651.36

TABLE75

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7501 2. 481.26		1. 7506 2. 529.29		1. 7511 2. 540.3
	1. 7502 2. 495.27		1. 7507 2. 529.29		1. 7512 2. 540.3
	1. 7503 2. 495.27		1. 7508 2. 529.29		1. 7513 2. 543.3
	1. 7504 2. 507.28		1. 7509 2. 533.29		1. 7514 2. 543.3
	1. 7505 2. 529.29		1. 7510 2. 533.29		1. 7515 2. 543.3

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7516 2. 543.3		1. 7521 2. 549.3		1. 7526 2 555.31
	1. 7517 2. 545.3		1. 7522 2. 549.3		1. 7527 2 557.31
	1. 7518 2. 545.3		1. 7523 2. 549.3		1. 7528 2 559.31
	1. 7519 2. 545.3		1. 7524 2. 551.3		1. 7529 2 560.31
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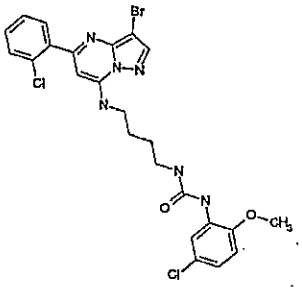
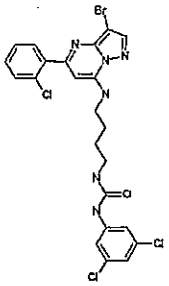
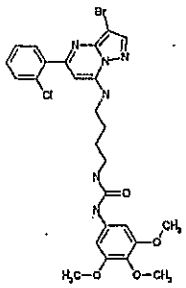
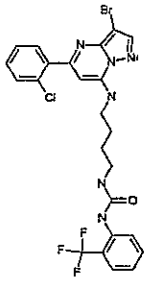
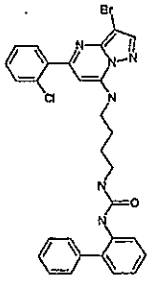
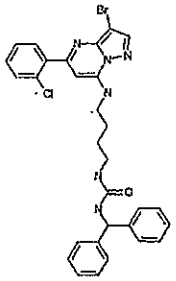
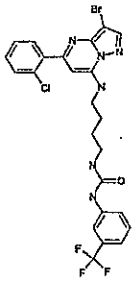
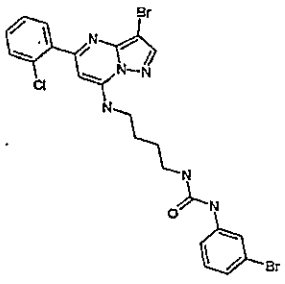
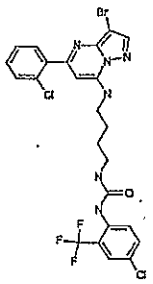
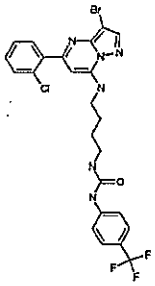
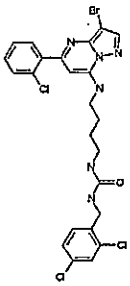
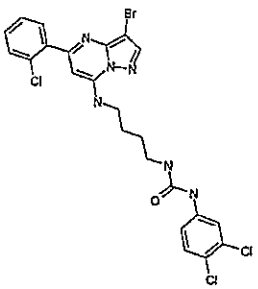
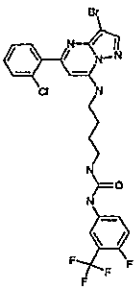
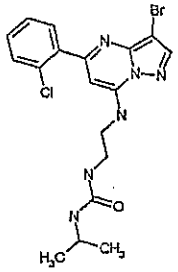
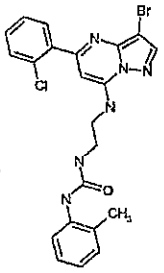
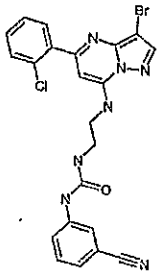
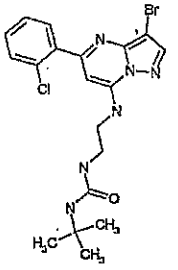
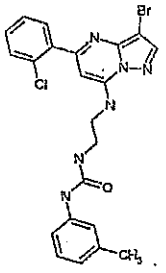
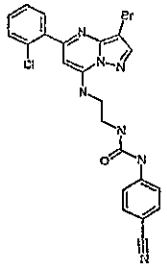
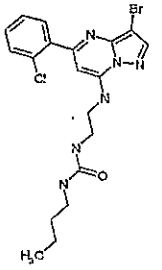
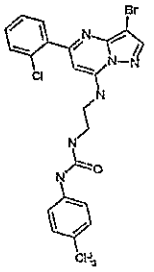
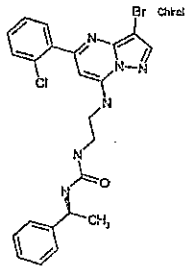
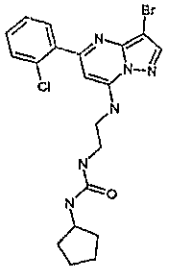
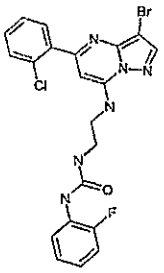
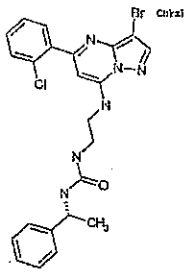
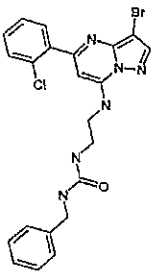
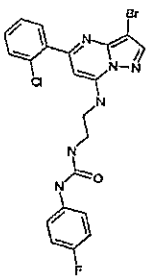
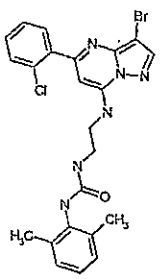
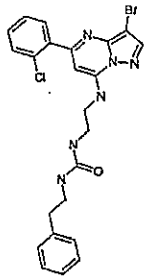
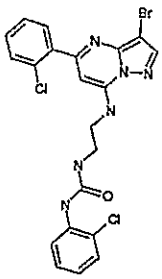
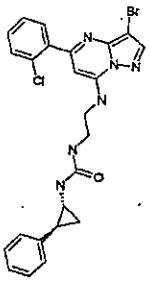
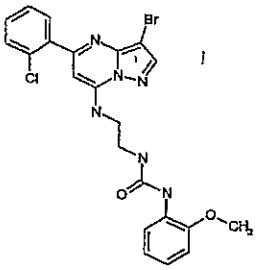
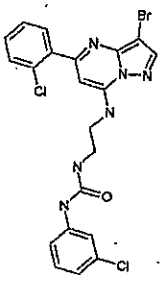
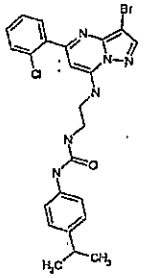
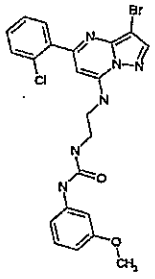
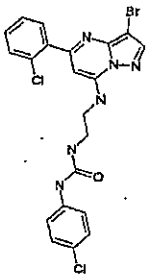
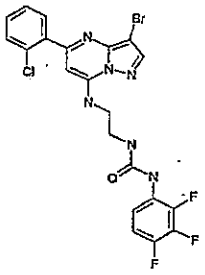
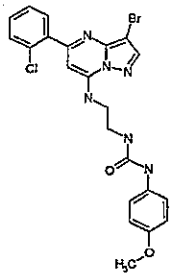
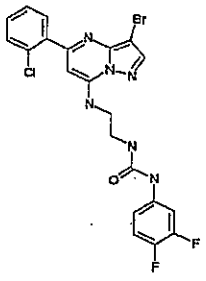
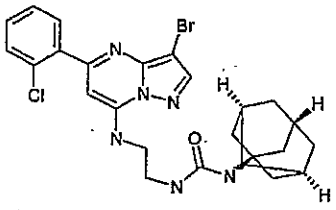
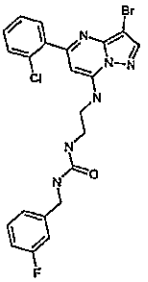
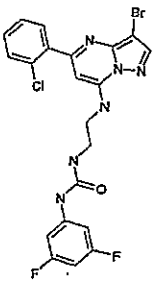
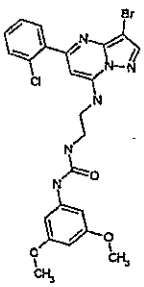
Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7531 2. 579.32		1. 7536 2. 583.32		1. 7541 2. 605.33
	1. 7532 2. 583.32		1. 7537 2. 591.33		1. 7542 2. 605.33
	1. 7533 2. 583.32		1. 7538 2. 593.33		1. 7543 2. 614.34
	1. 7534 2. 583.32		1. 7539 2. 597.33		
	1. 7535 2. 583.32		1. 7540 2. 601.33		

TABLE 76

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7601 2. 453.25		1. 7606 2. 501.28		1. 7611 2. 512.28
	1. 7602 2. 467.26		1. 7607 2. 501.28		1. 7612 2. 512.28
	1. 7603 2. 467.26		1. 7608 2.501.28		1. 7613 2. 515.28
	1. 7604 2. 479.26		1. 7609 2. 505.28		1. 7614 2. 515.28
	1. 7605 2. 501.28		1. 7610 2. 505.28		1. 7615 2. 515.28

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 761 6 2. 515.28		1. 7621 2. 521.29		1. 7626 2. 527.29
	1. 7617 2. 517.28		1. 7622 2. 521.29		1. 7627 2. 529.29
	1. 7618 2. 517.28		1. 7623 2. 521.29		1. 7628 2. 541.3
	1. 7619 2. 517.28		1. 7624 2. 523.29		1. 7629 2. 545.3
	1. 7620 2. 519.29		1. 7625 2. 523.29		1. 7630 2. 547.3

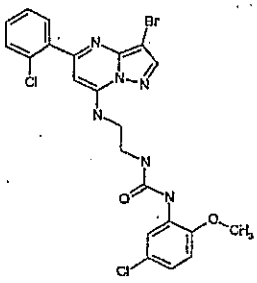
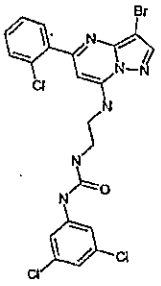
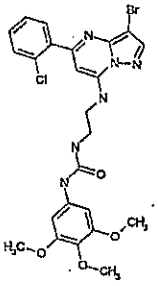
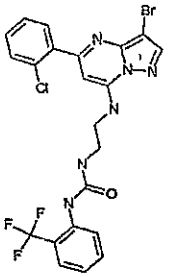
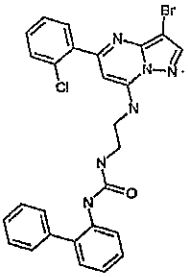
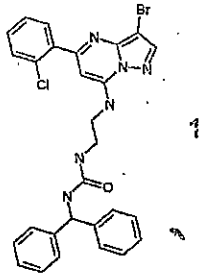
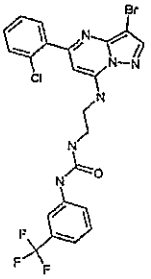
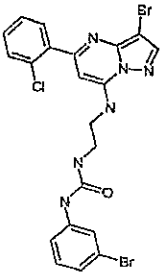
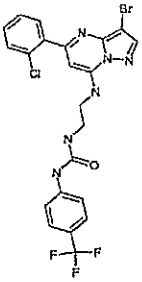
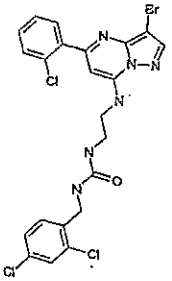
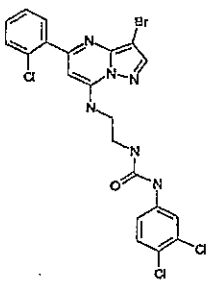
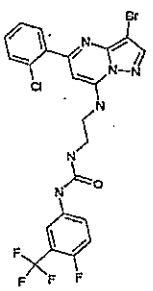
Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7631 2. 551.3		1. 7636 2. 555.31		1. 7641 2. 577.32
	1. 7632 2. 555.31		1. 7637 2. 563.31		1. 7642 2. 577.32
	1. 7633 2. 555.31		1. 7638 2. 565.31		
	1. 7634 2. 555.31		1. 7639 2. 569.31		
	1. 7635 2. 555.31		1. 7640 2. 573.32		

TABLE 77

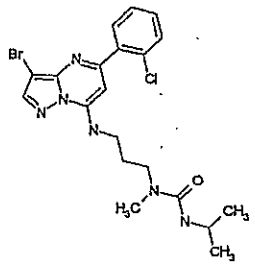
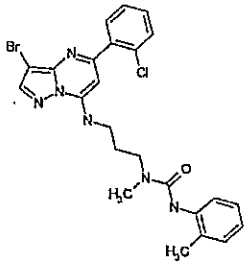
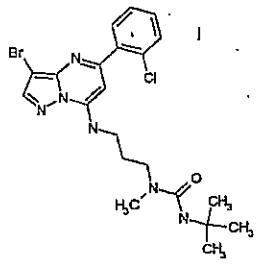
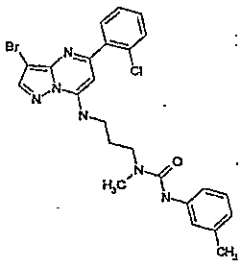
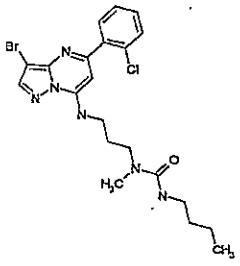
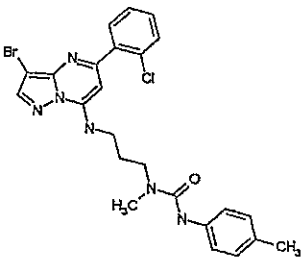
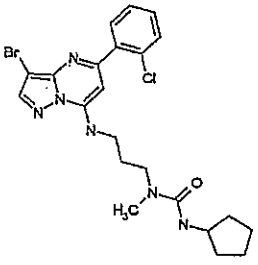
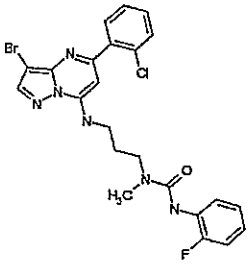
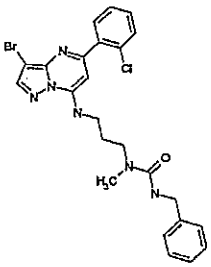
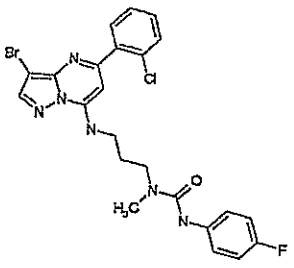
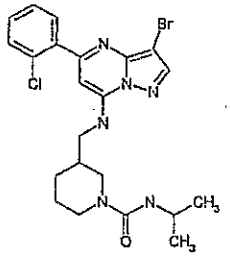
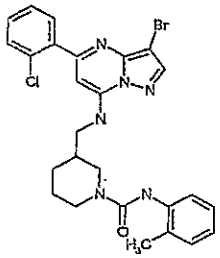
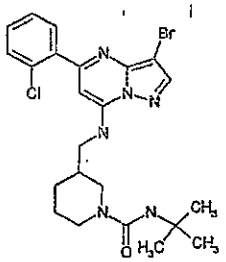
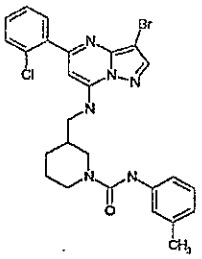
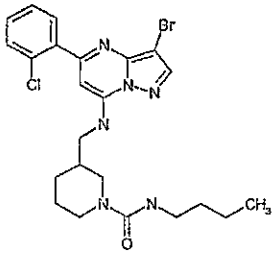
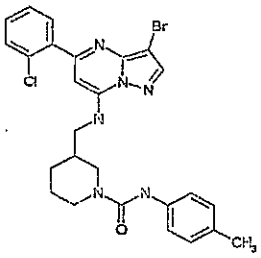
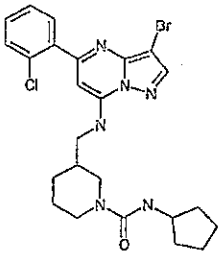
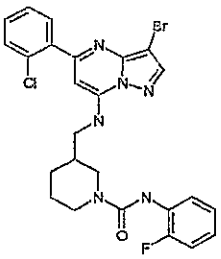
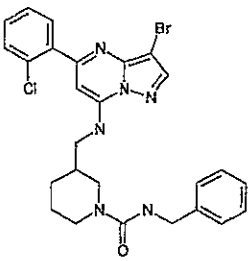
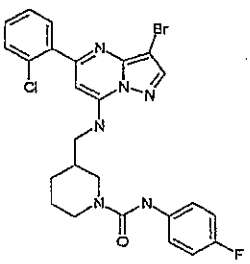
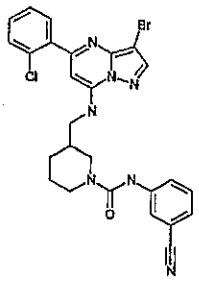
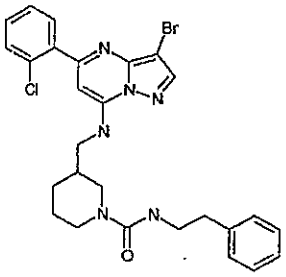
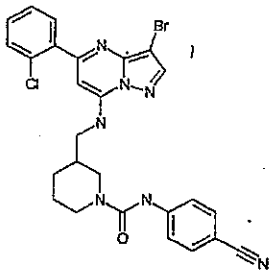
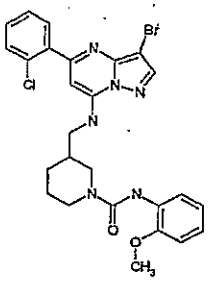
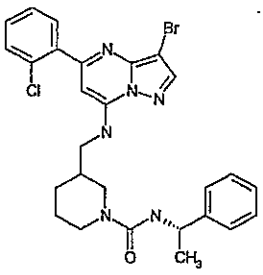
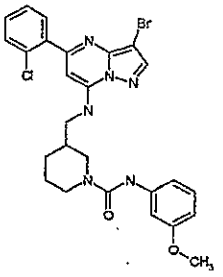
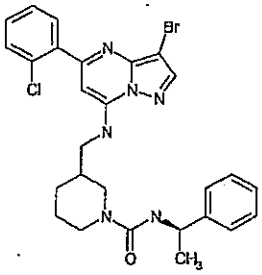
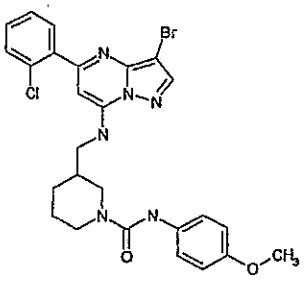
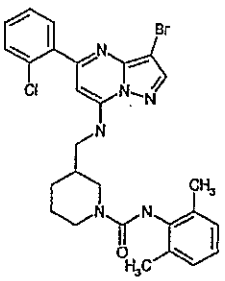
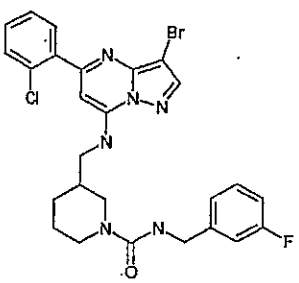
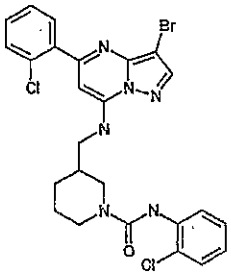
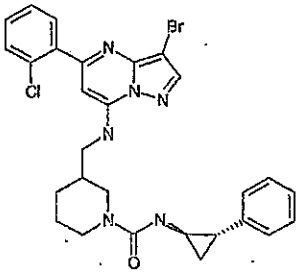
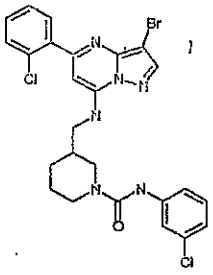
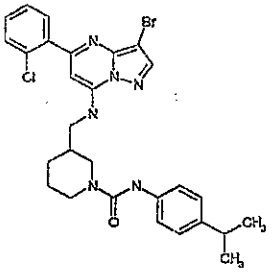
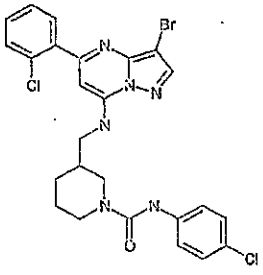
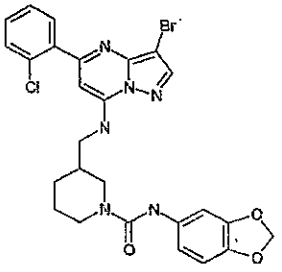
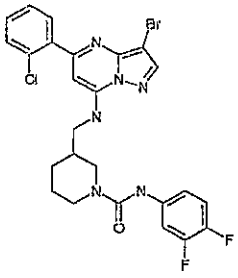
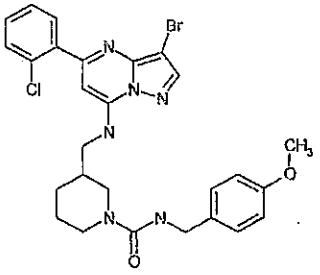
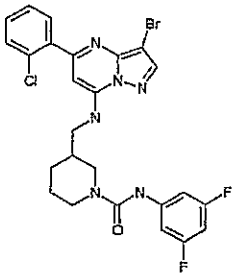
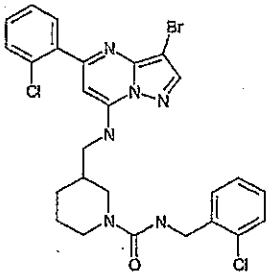
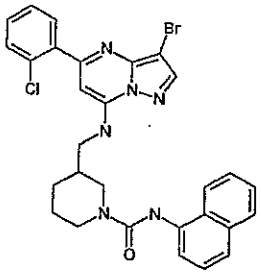
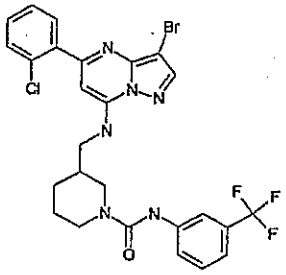
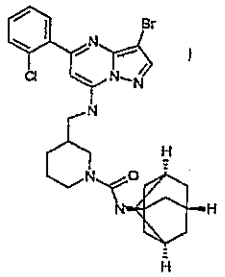
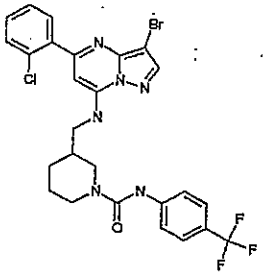
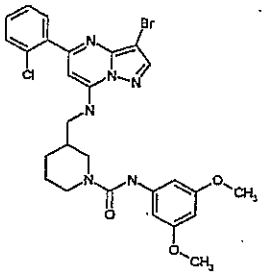
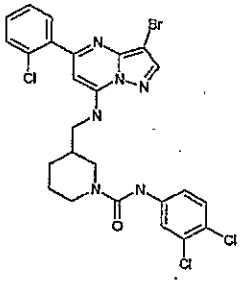
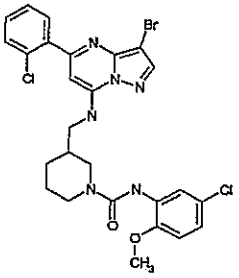
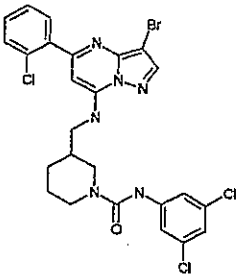
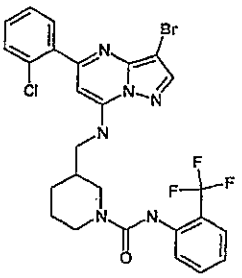
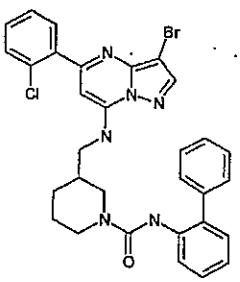
Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7701 2. 481.26		1. 7706 2. 529.29
	1. 7702 2. 495.27		1. 7707 2. 529.29
	1. 7703 2. 495.27		1. 7708 2. 529.29
	1. 7704 2. 507.28		1. 7709 2. 533.29
	1. 7705 2. 529.29		1. 7710 2. 533.29

TABLE 78

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7801 2. 507.28		1. 7806 2. 555.31
	1. 7802 2. 521.29		1. 7807 2. 555.31
	1. 7803 2. 521.29		1. 7808 2. 555.31
	1. 7804 2. 533.29		1. 7809 2. 559.31
	1. 7805 2. 555.31		1. 7810 2. 559.31

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7811 2. 566.31		1. 7816 2. 569.31
	1. 7812 2. 566.31		1. 7817 2. 571.31
Chiral 	1. 7813 2. 569.31		1. 7818 2. 571.31
Chiral 	1. 7814 2. 569.31		1. 7819 2. 571.31
	1. 7815 2. 569.31		1. 7820 2. 573.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7821 2. 575.32		1. 7826 2. 581.32
	1. 7822 2. 575.32		1. 7827 2. 583.32
	1. 7823 2. 575.32		1. 7828 2. 585.32
	1. 7824 2. 577.32		1. 7829 2. 585.32
	1. 7825 2. 577.32		1. 7830 2. 589.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7831 2. 591.33		1. 7836 2. 609.33
	1. 7832 2. 599.33		1. 7837 2. 609.33
	1. 7833 2. 601.33		1. 7838 2. 609.33
	1. 7834 2. 605.33		1. 7839 2. 609.33
	1. 7835 2. 609.33		1. 7840 2. 617.34

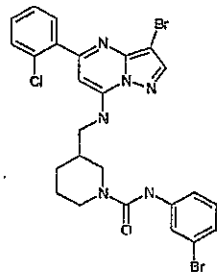
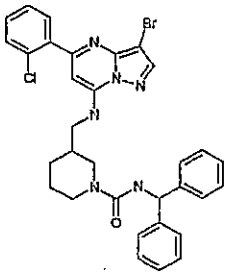
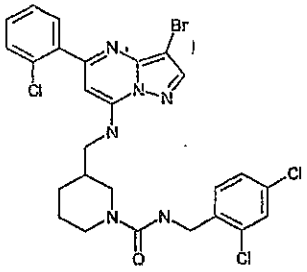
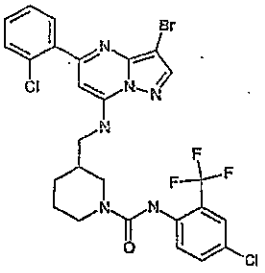
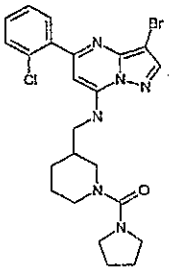
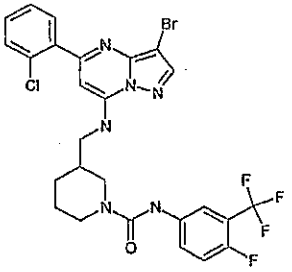
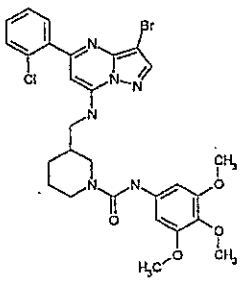
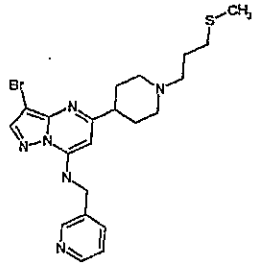
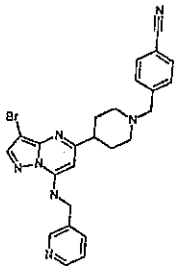
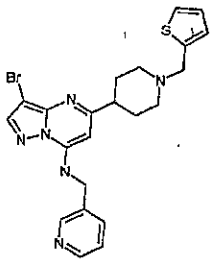
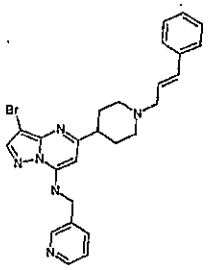
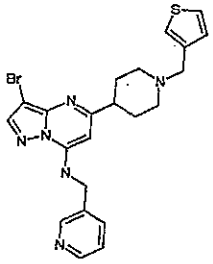
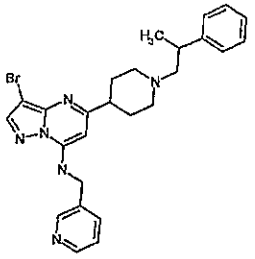
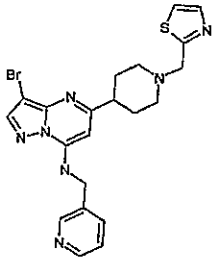
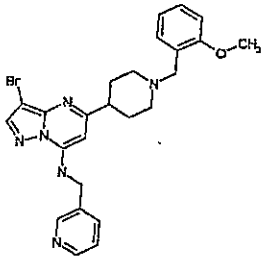
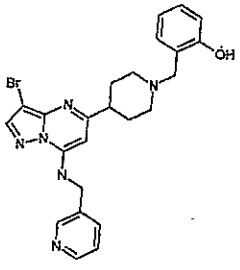
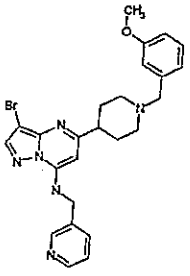
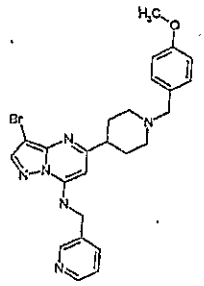
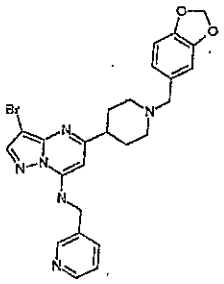
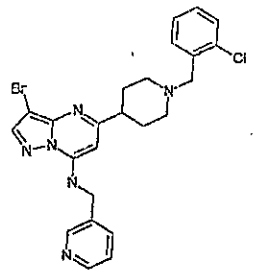
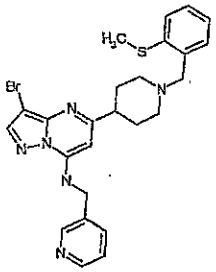
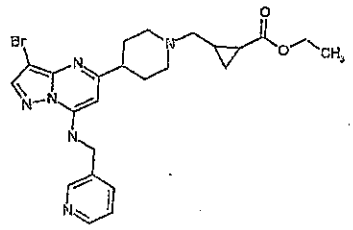
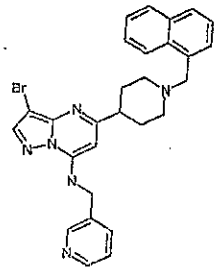
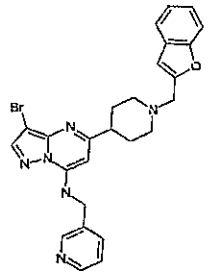
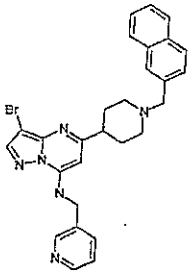
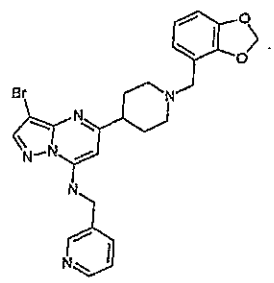
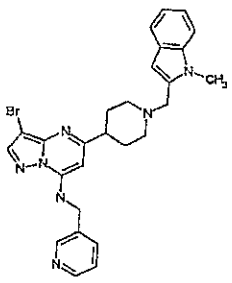
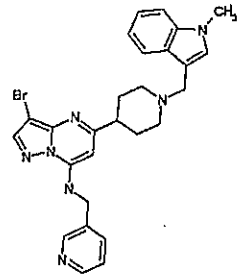
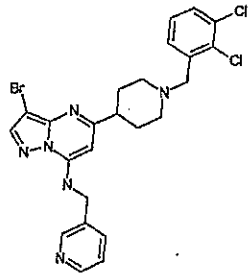
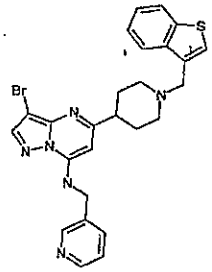
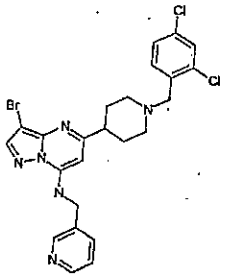
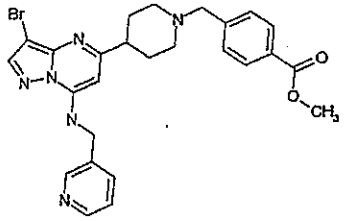
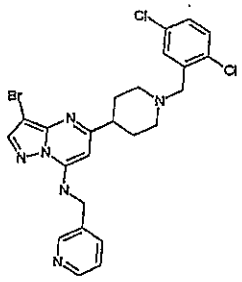
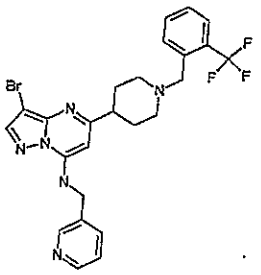
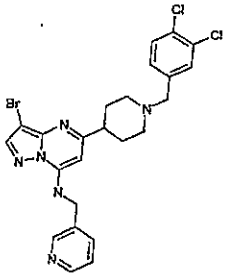
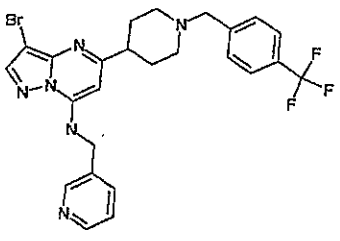
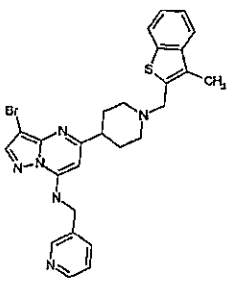
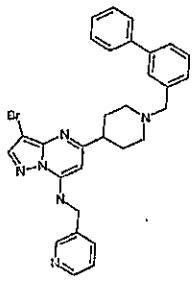
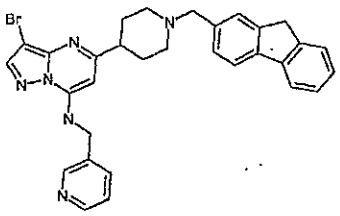
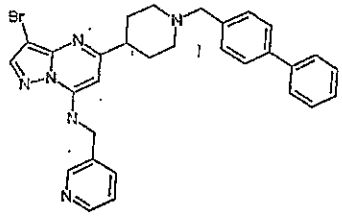
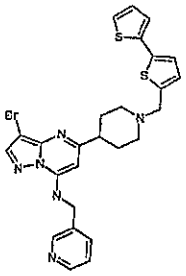
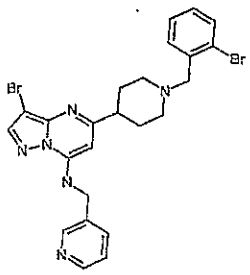
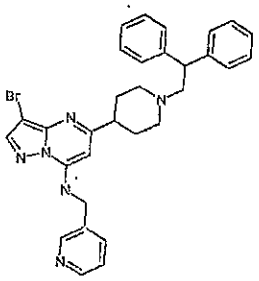
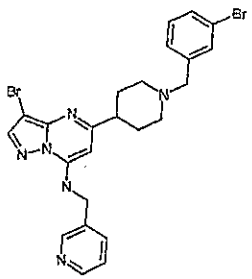
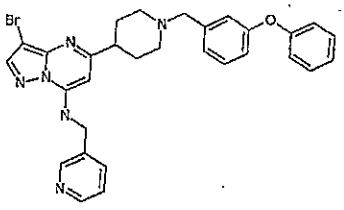
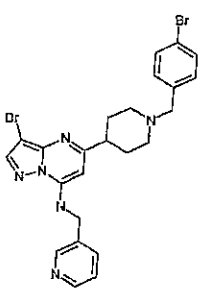
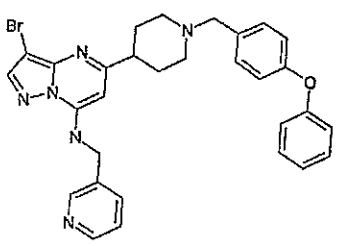
Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7841 2. 619.34		1. 7846 2. 631.35
	1. 7842 2. 623.34		1. 7847 2. 643.35
	1. 7843 2. 519.29		
	1. 7844 2. 627.34		
	1. 7845 2. 631.35		

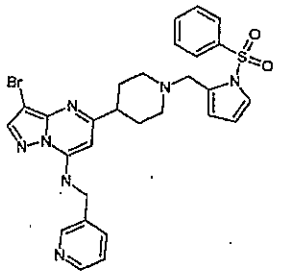
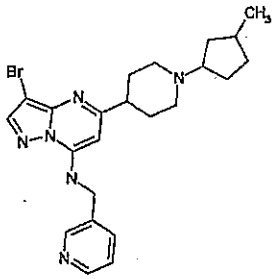
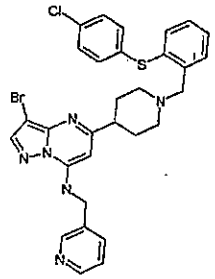
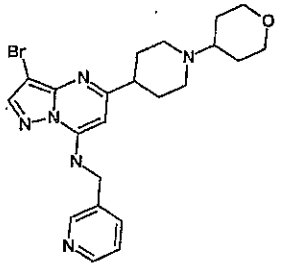
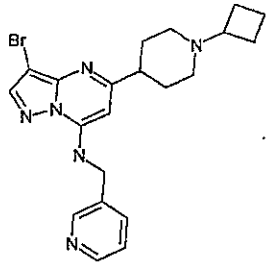
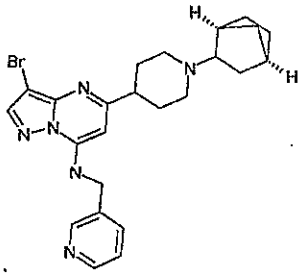
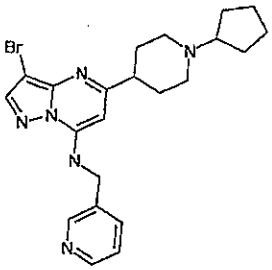
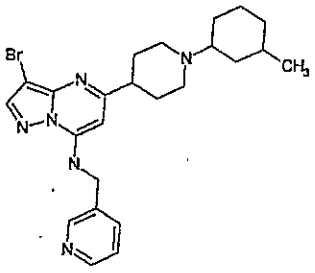
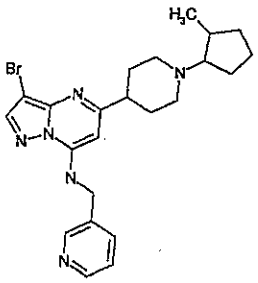
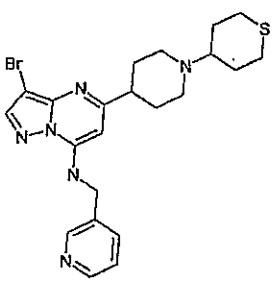
TABLE 79

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7901 2. 477.26		1. 7906 2. 504.28
	1. 7902 2. 485.27		1. 7907 2. 505.28
	1. 7903 2. 485.27		1. 7908 2. 507.28
	1. 7904 2. 486.27		1. 7909 2. 509.28
	1. 7905 2. 495.27		1. 7910 2. 509.28

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7911 2. 509.28		1. 7916 2. 523.29
	1. 7912 2. 513.28		1. 7917 2. 525.29
	1. 7913 2. 515.28		1. 7918 2. 529.29
	1. 7914 2. 519.29		1. 7919 2. 529.29
	1. 7915 2. 523.29		1. 7920 2. 530.29

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7921 2. 530.29		1. 7926 2. 547.3
	1. 7922 2. 535.29		1. 7927 2. 547.3
	1. 7923 2. 537.3		1. 7928 2. 547.3
	1. 7924 2. 547.3		1. 7929 2. 547.3
	1. 7925 2. 547.3		1. 7930 2. 549.3

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7931 2. 555.31		1. 7936 2. 567.31
	1. 7932 2. 555.31		1. 7937 2. 567.31
	1. 7933 2. 557.31		1. 7938 2. 569.31
	1. 7934 2. 557.31		1. 7939 2. 571.31
	1. 7935 2. 557.31		1. 7940 2. 571.31

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 7941 2. 608.33		1. 7946 2. 471.26
	1. 7942 2. 621.34		1. 7947 2. 473.26
	1. 7943 2. 443.24		1. 7948 2. 483.27
	1. 7944 2. 457.25		1. 7949 2. 485.27
	1. 7945 2. 471.26		1. 7950 2. 489.27

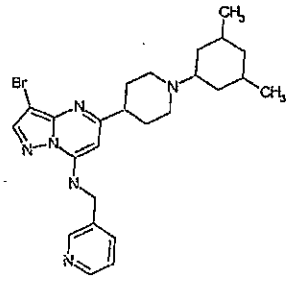
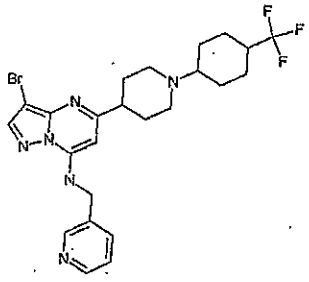
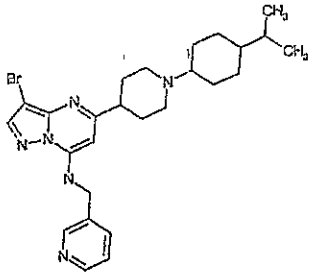
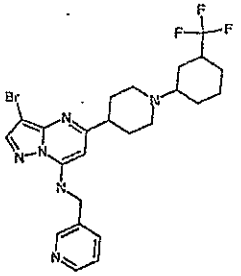
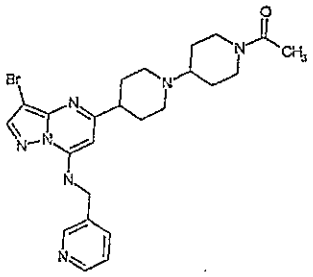
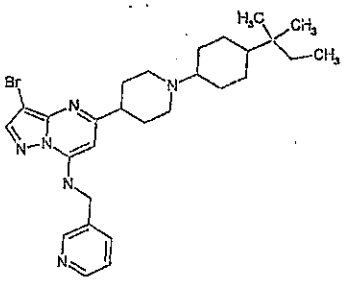
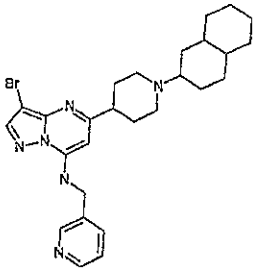
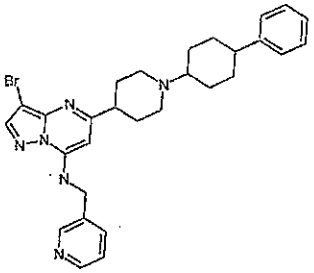
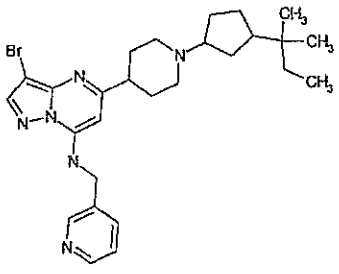
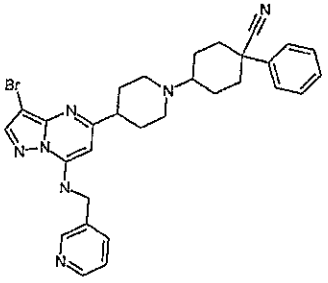
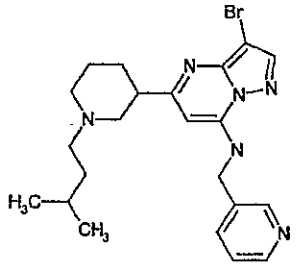
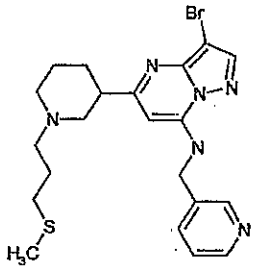
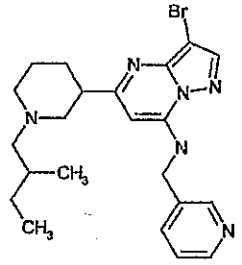
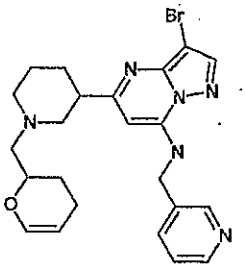
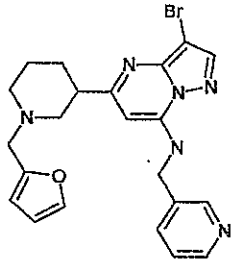
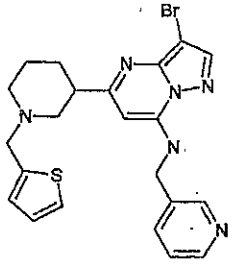
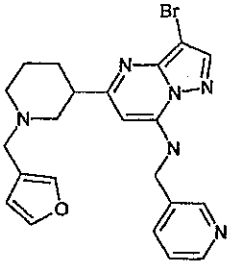
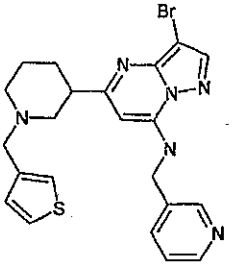
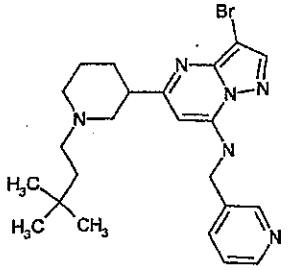
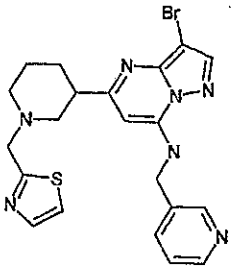
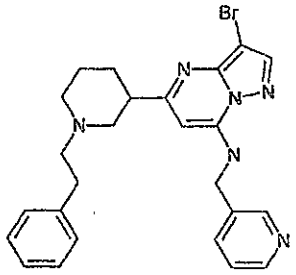
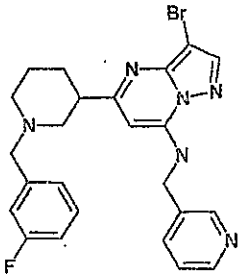
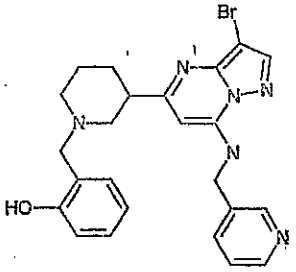
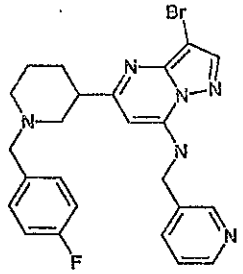
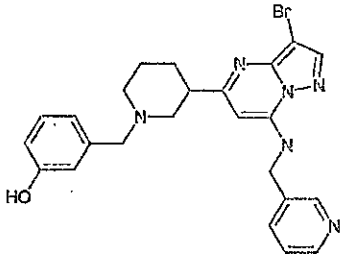
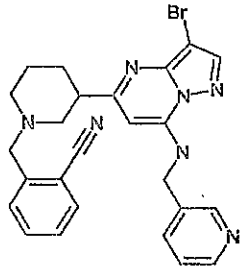
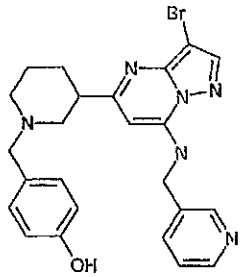
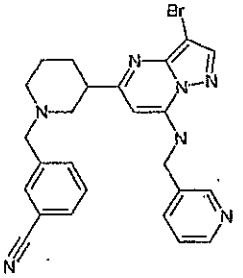
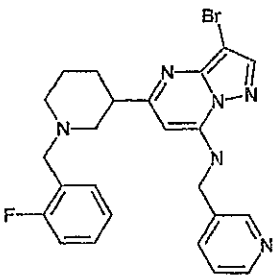
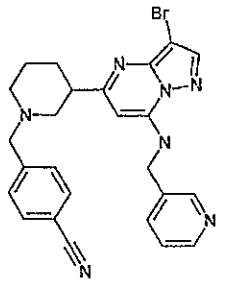
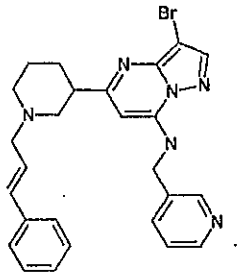
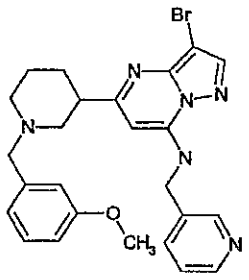
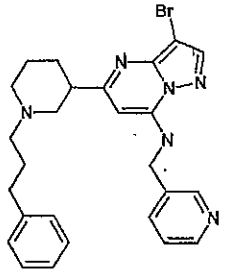
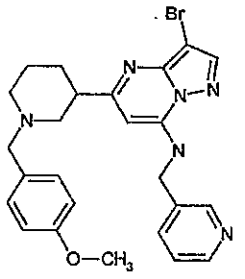
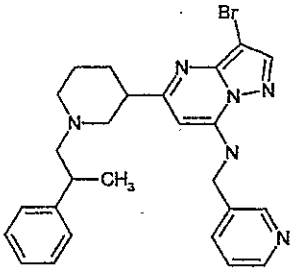
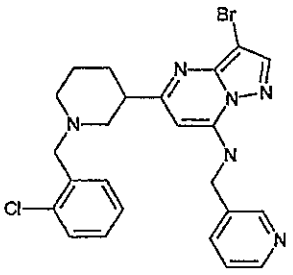
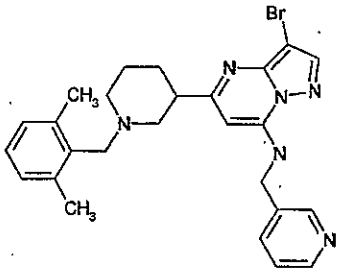
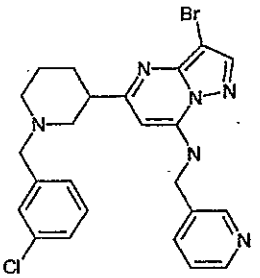
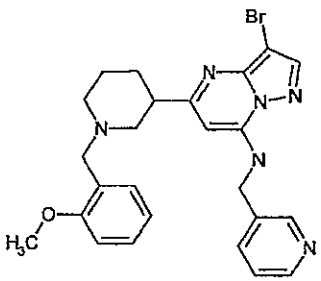
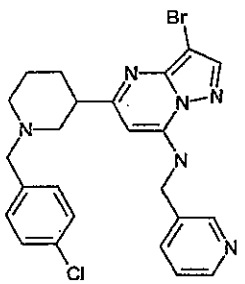
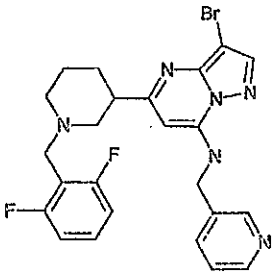
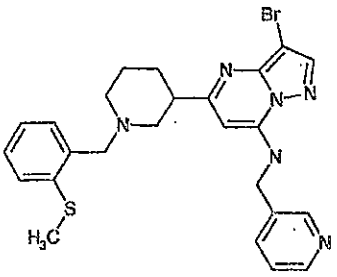
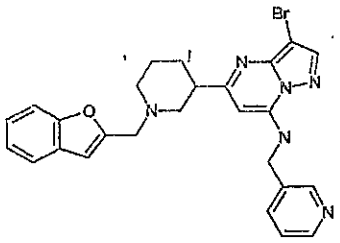
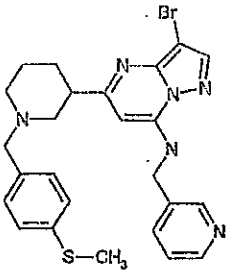
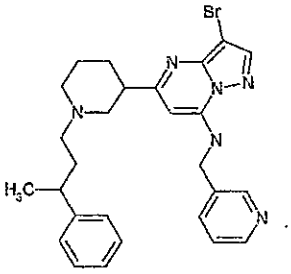
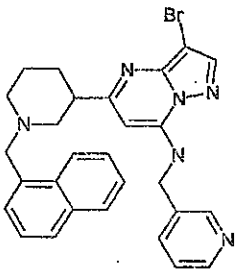
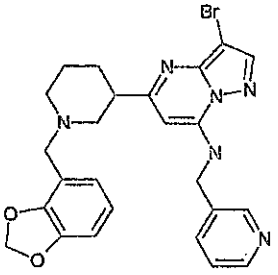
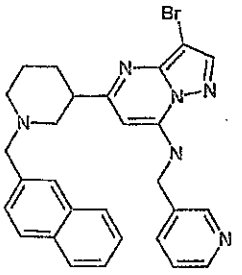
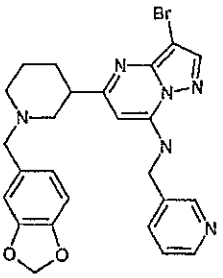
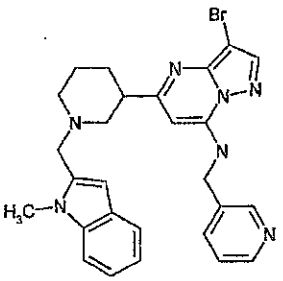
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	1. 7952 2. 513.28		1. 7957 2. 539.3
	1. 7953 2. 514.28		1. 7958 2. 541.3
	1. 7954 2. 525.29		1. 7959 2. 547.3
	1. 7955 2. 525.29		1. 7960 2. 572.31

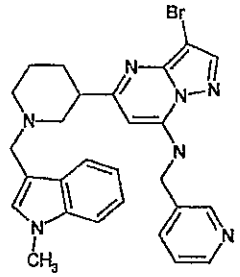
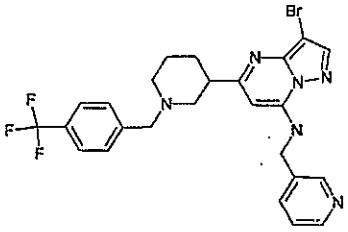
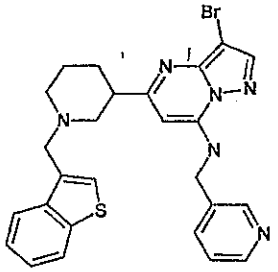
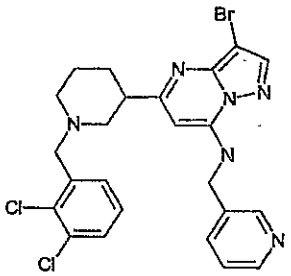
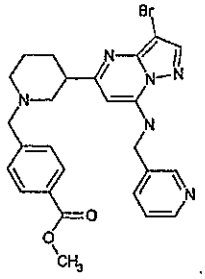
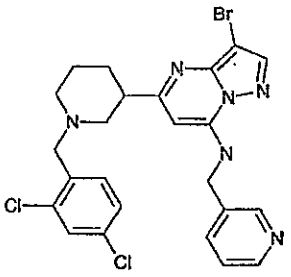
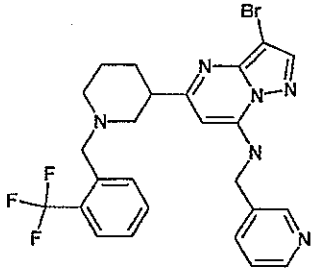
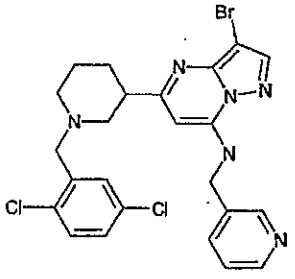
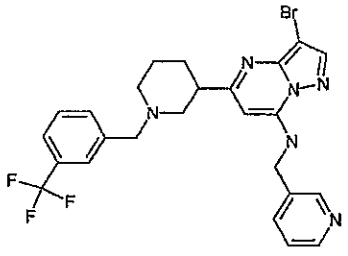
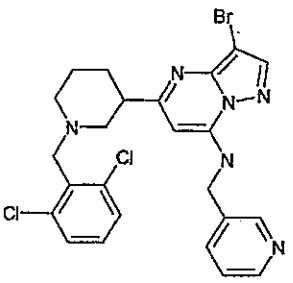
TABLE 80

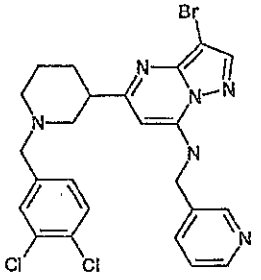
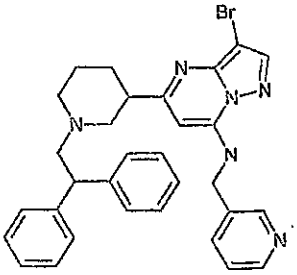
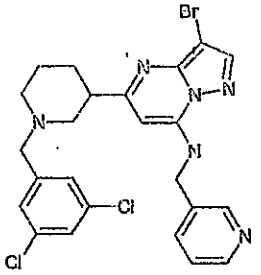
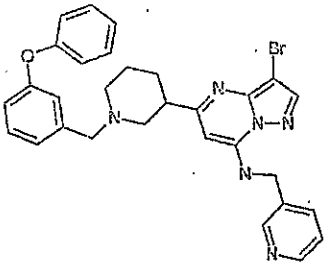
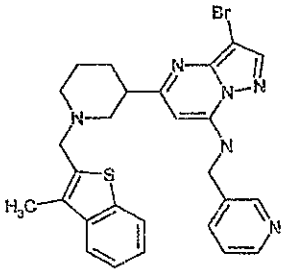
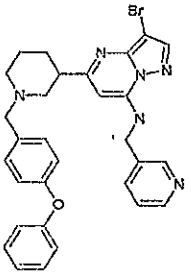
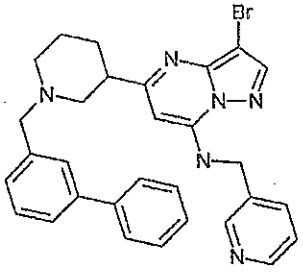
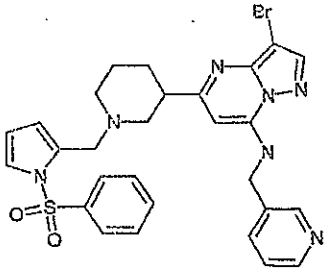
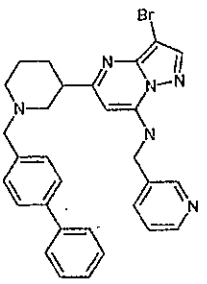
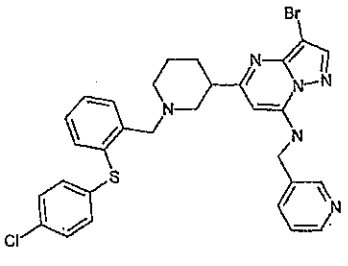
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	1.8002 2. 459.25		1. 8007 2 485.27
	1. 8003 2. 469.26		1. 8008 2. 485.27
	1. 8004 2. 469.26		1. 8009 2. 485.27
	1. 8005 2. 473.26		1. 8010 2. 486.27

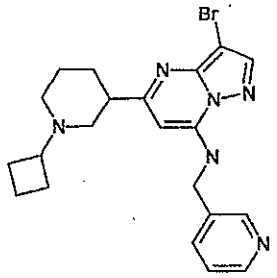
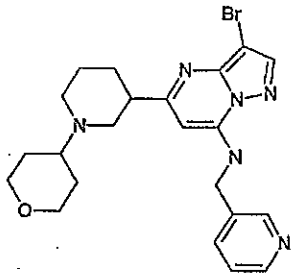
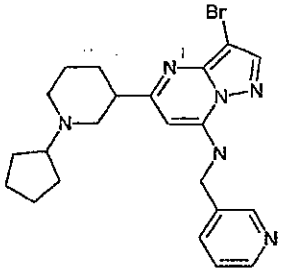
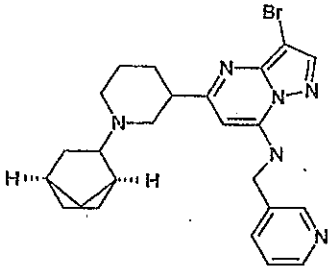
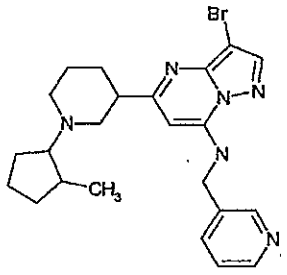
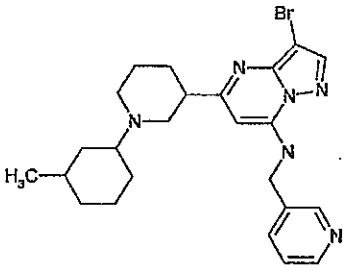
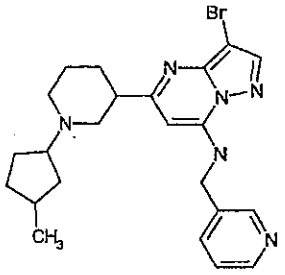
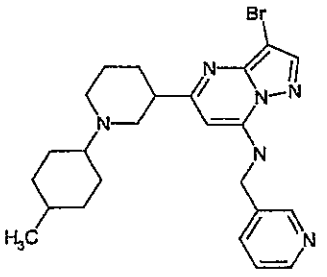
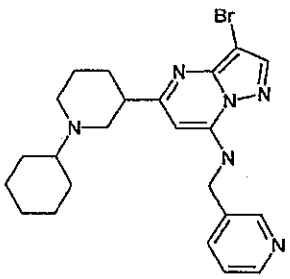
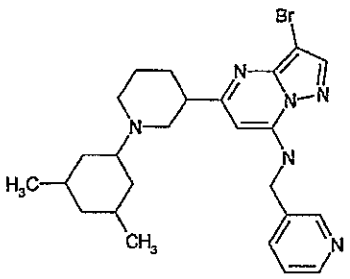
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	1. 8011 2. 493.27		1. 8016 2. 495.27
	1. 8012 2. 495.27		1. 8017 2. 495.27
	1. 8013 2. 495.27		1. 8018 2. 502.28
	1. 8014 2. 495.27		1. 8019 2. 502.28
	1. 8015 2. 495.27		1. 8020 2. 502.28

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8021 2. 505.28		1. 8026 2. 509.28
	1. 8022 2. 505.28		1. 8027 2. 509.28
	1. 8023 2. 507.28		1. 8028 2. 513.28
	1. 8024 2. 507.28		1. 8029 2. 513.28
	1. 8025 2. 507.28		1. 8030 2. 513.28

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8031 2. 515.28		1. 8036 2. 523.29
	1. 8032 2. 519.29		1. 8037 2. 525.29
	1. 8033 2. 521.29		1. 8038 2. 529.29
	1. 8034 2. 523.29		1. 8039 2. 529.29
	1. 8035 2. 523.29		1. 8040 2. 530.29

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8041 2. 530.29		1. 8046 2. 547.3
	1. 8042 2. 535.29		1. 8047 2. 544.3
	1. 8043 2. 537.3		1. 8048 2. 547.3
	1. 8044 2. 547.3		1. 8049 2. 547.3
	1. 8045 2. 547.3		1. 8050 2. 547.3

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8051 2. 548.3		1. 8056 2. 569.31
	1. 8052 2. 548.3		1. 8057 2. 569.31
	1. 8053 2. 547.3		1. 8058 2. 571.31
	1. 8054 2. 555.31		1. 8059 2. 608.33
	1. 8055 2. 553.3		1. 8060 2. 621.34

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8061 2. 443.24		1. 8066 2. 473.26
	1. 8062 2. 457.25		1. 8067 2. 483.27
	1. 8063 2. 471.26		1. 8068 2. 485.27
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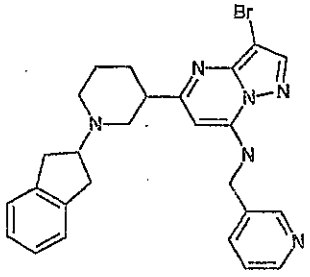
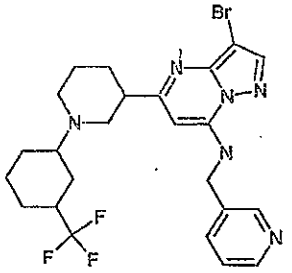
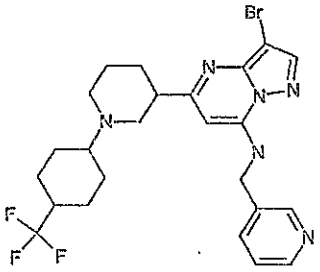
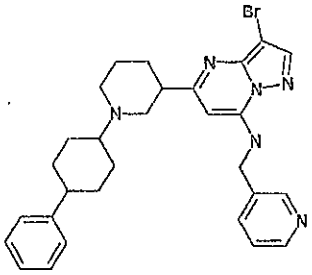
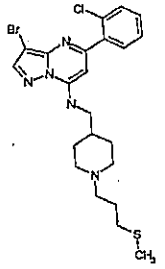
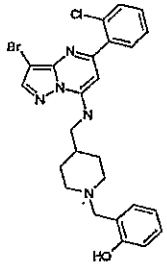
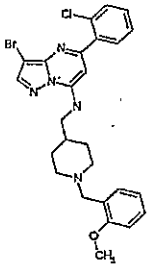
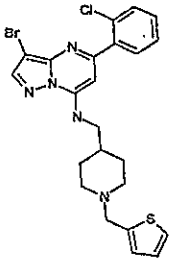
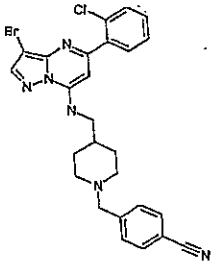
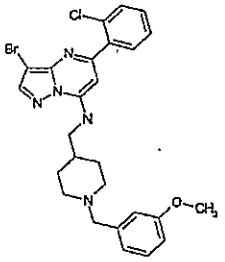
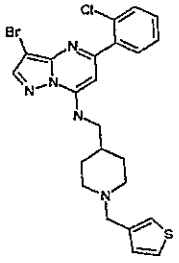
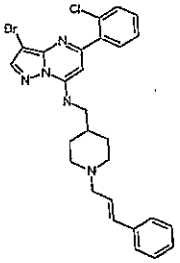
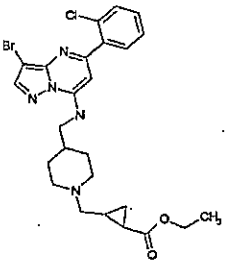
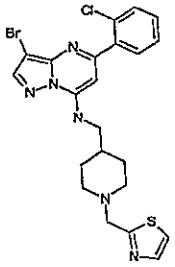
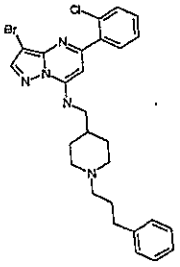
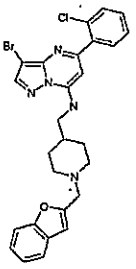
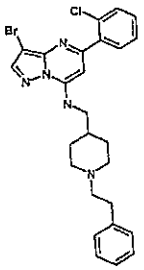
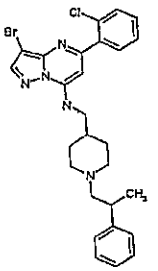
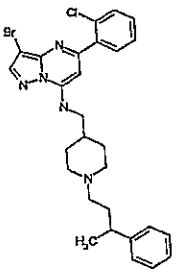
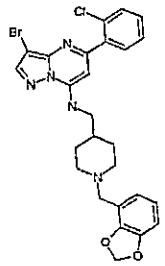
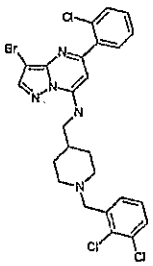
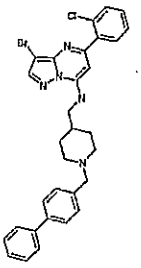
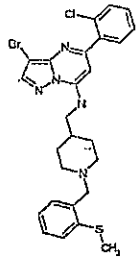
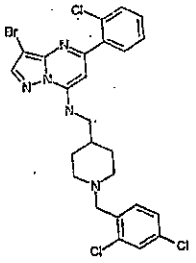
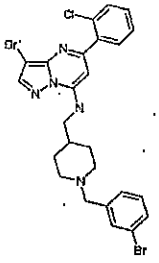
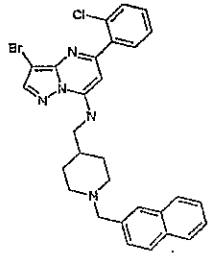
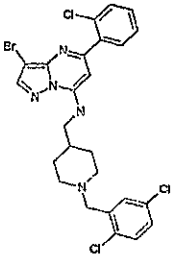
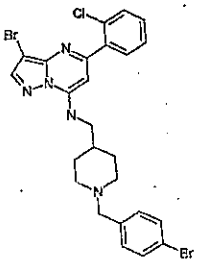
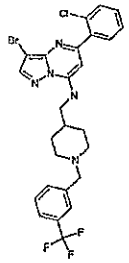
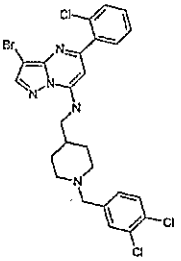
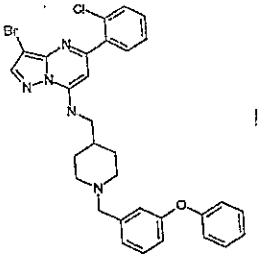
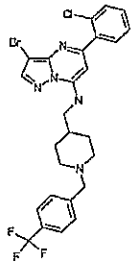
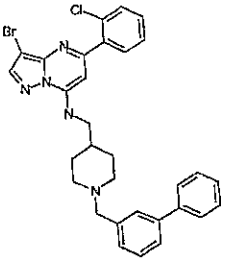
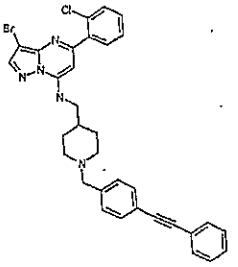
Product	1. Ex. 2. m/z
	1. 8071 2. 505.2 8
	1. 8072 2. 537.3
	1. 8073 2. 537.3
	1. 8074 2. 547.3

TABLE81

Product	1. Ex. 2. m/zz	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z.
	1. 8101 2. 510.28		1. 8106 2. 528.29		1. 8111 2. 542.3
	1. 8102 2. 518.28		1. 8107 2. 537.3		1. 8112 2. 542.3
	1. 8103 2. 518.28		1. 8108 2. 538.3		1. 8113 2. 548.3
	1. 8104 2. 519.29		1. 8109 2. 540.3		1. 8114 2. 552.3
	1. 8105 2. 526.29		1. 8110 2. 540.3		1. 8115 2. 554.3

Product	1. Ex. 2: m/zz	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8116 2. 556.31		1. 8121 2. 580.32		1. 8126 2. 588.32
	1. 8117 2. 558.31		1. 8122 2. 580.32		1. 8127 2. 590.32
	1. 8118 2. 562.31		1. 8123 2. 580.32		1. 8128 2. 590.32
	1. 8119 2. 580.32		1. 8125 2. 580.32		I. 8129 2. 604.33
	1. 8120 2. 580.32		1. 8126 2. 588.32		1. 8130 2. 612.34

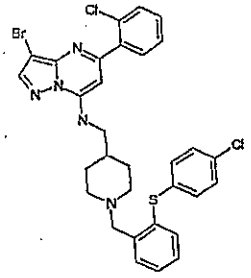
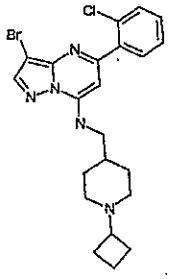
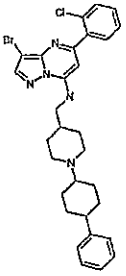
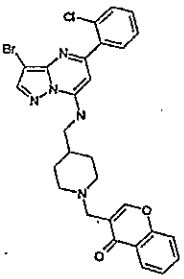
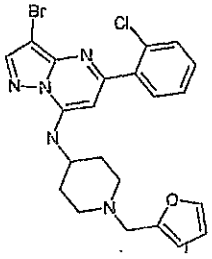
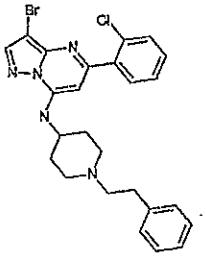
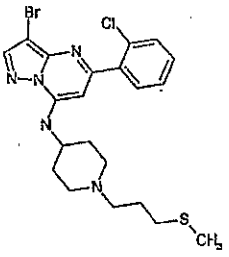
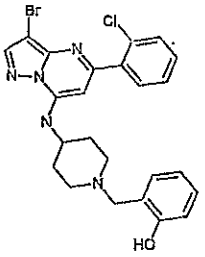
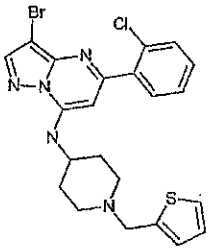
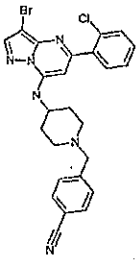
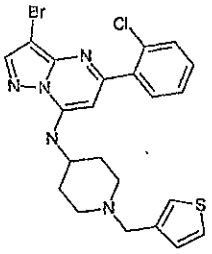
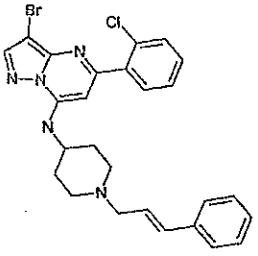
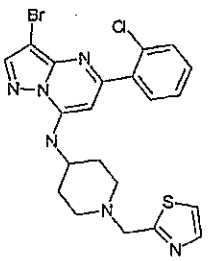
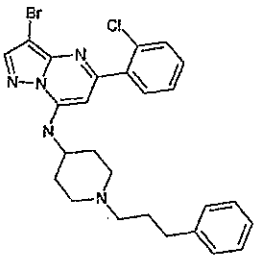
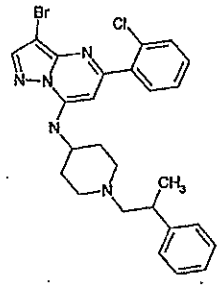
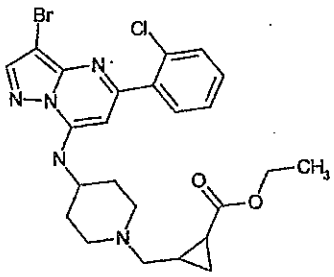
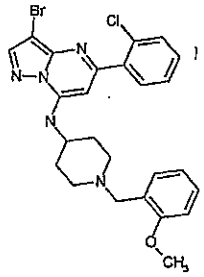
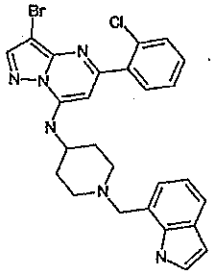
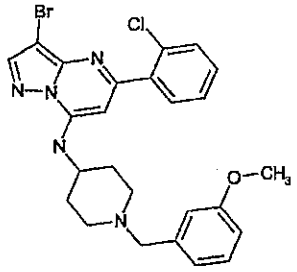
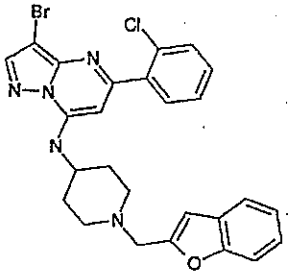
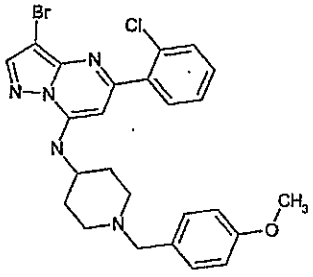
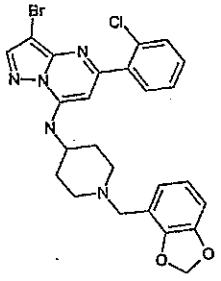
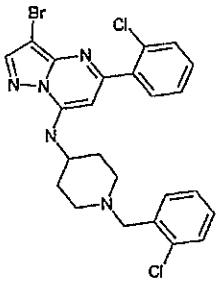
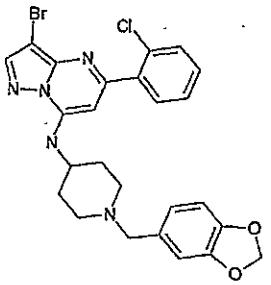
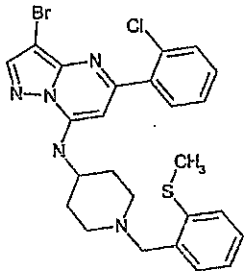
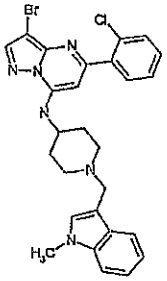
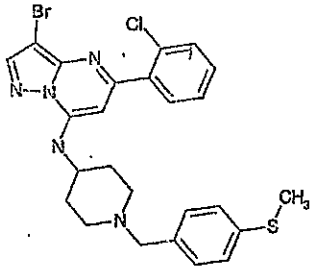
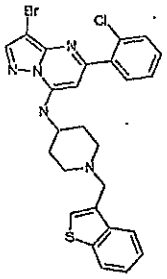
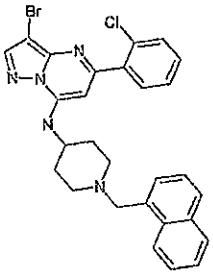
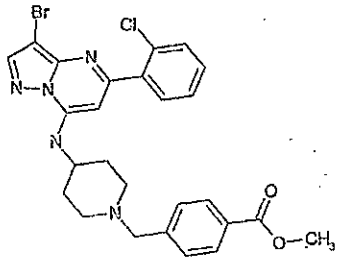
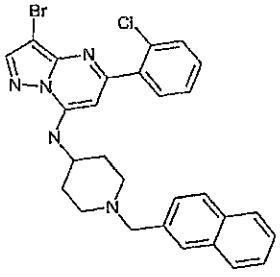
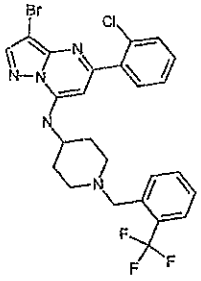
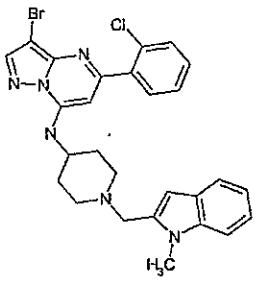
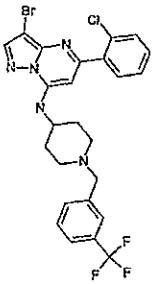
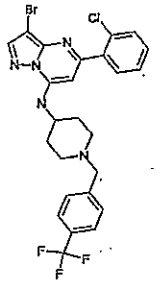
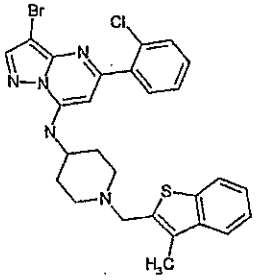
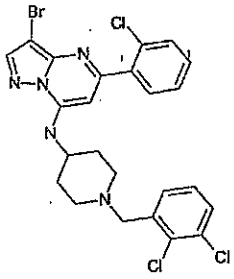
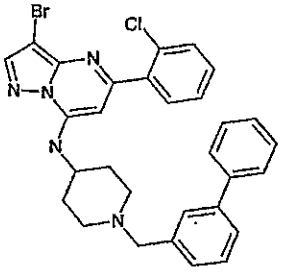
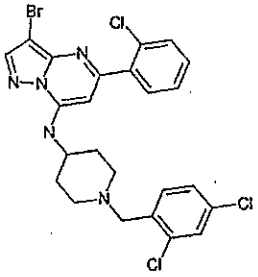
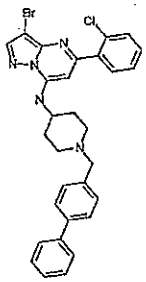
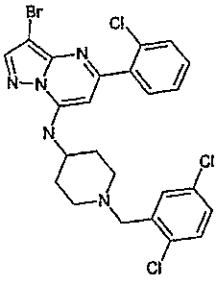
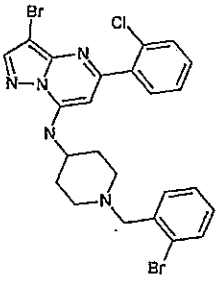
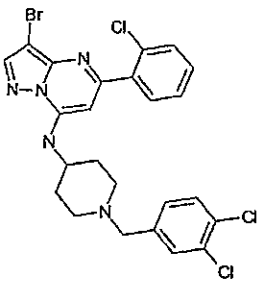
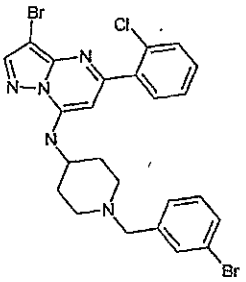
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	1. 8132 2. 476.26
	1. 8133 2. 580.32
	1. 8134 2. 580.32

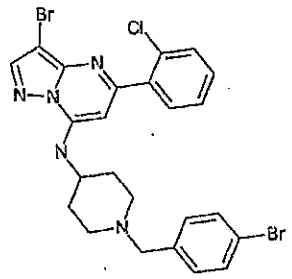
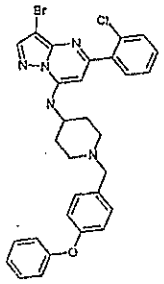
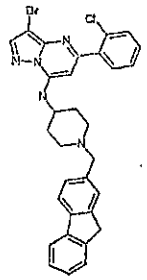
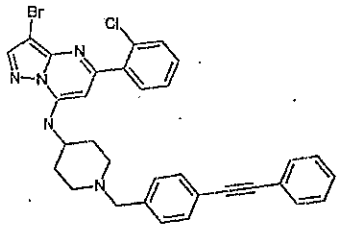
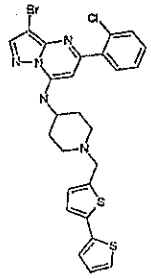
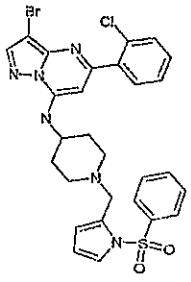
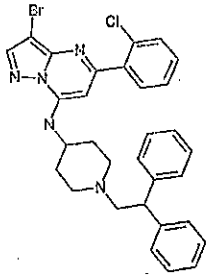
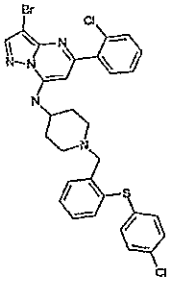
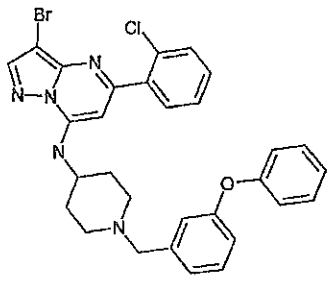
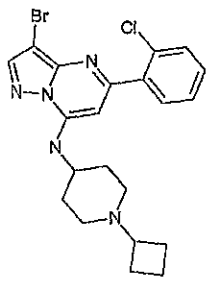
TABLE 82

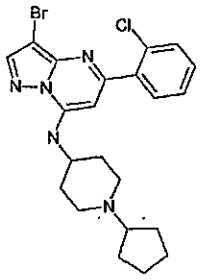
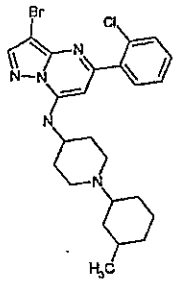
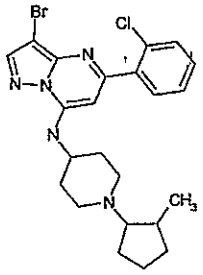
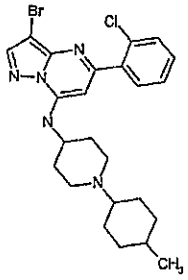
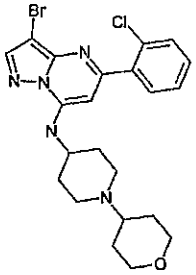
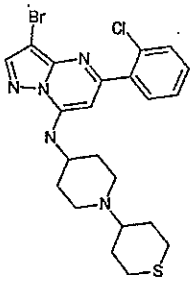
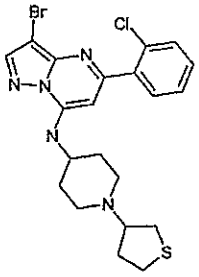
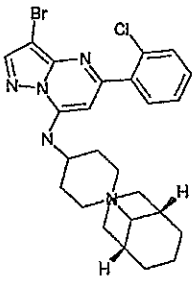
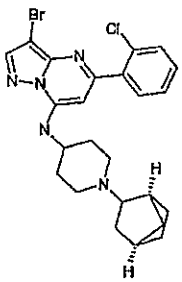
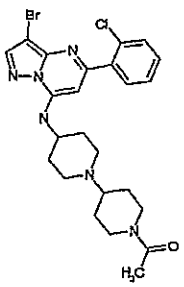
Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8201 2. 488.27		1. 8206 2. 512.28
	1. 8202 2. 496.27		1. 8207 2. 514.28
	1. 8203 2. 504.26		1. 8208 2. 523.29
	1. 8204 2. 504.28		1. 8209 2. 524.29
	1. 8205 2. 505.28		1. 8210 2. 526.29

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8211 2. 526.29		1. 8216 2. 534.29
	1. 8212 2. 528.29		1. 8217 2. 537.3
	1. 8213 2. 528.29		1. 8218 2. 538.3
	1. 8214 2. 528.29		1. 8219 2. 542.3
	1. 8215 2. 532.29		1. 8220 2. 542.3

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8221 2. 544.3		1. 8226 2. 551.3
	1. 8222 2. 544.3		1. 8227 2. 554.3
	1. 8223 2. 548.3		1. 8228 2. 556.31
	1. 8224 2. 548.3		1. 8229 2. 566.31
	1. 8225 2. 551.3		1. 8230 2. 566.31

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8231 2. 566.31		1. 8236 2. 568.31
	1. 8232 2. 566.31		1. 8237 2. 574.32
	1. 8233 2. 566.31		1. 8238 2. 574.32
	1. 8243 2. 566.31		1. 8239 2. 576.32
	1. 8235 2. 566.31		1. 8240 2. 576.32

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8241 2. 576.32		1. 8246 2. 590.32
	1. 8242 2. 586.32		1. 8247 2. 598.33
	1. 8243 2. 586.32		1. 8248 2. 627.34
	1. 8244 2. 588.32		1. 8249 2. 640.35
	1. 8245 2. 590.32		1. 8250 2. 462.25

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8251 2. 476.26		1. 8256 2. 504.28
	1. 8252 2. 490.27		1. 8257 2. 504.28
	1. 8253 2. 492.27		1. 8258 2. 508.28
	1. 8254 2. 492.27		1. 8259 2. 530.29
	1. 8255 2. 502.28		1. 8260 2. 533.29

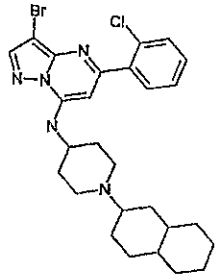
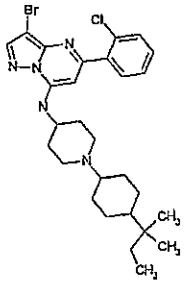
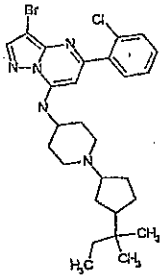
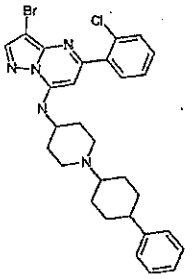
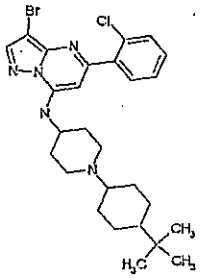
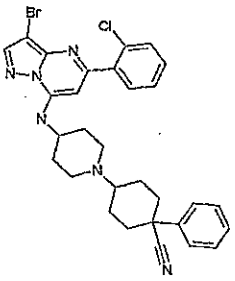
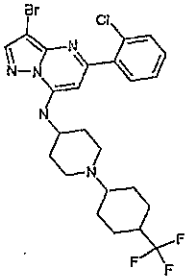
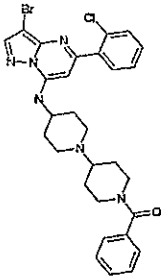
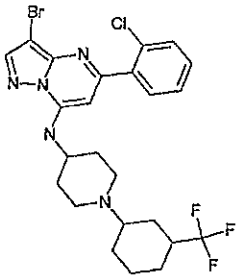
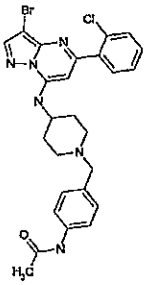
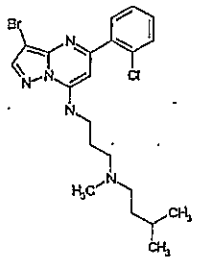
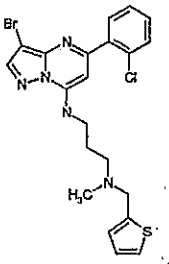
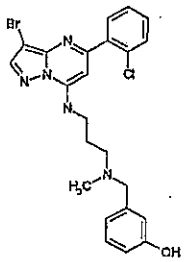
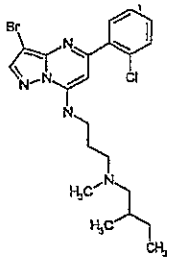
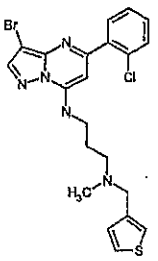
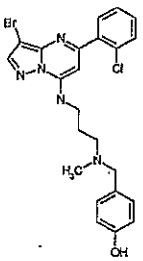
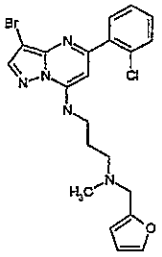
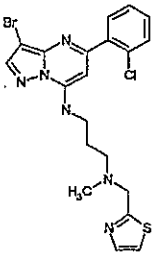
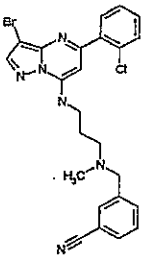
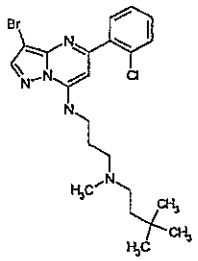
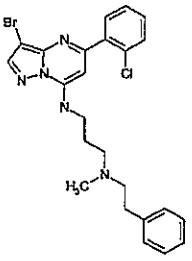
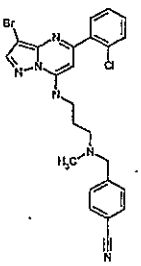
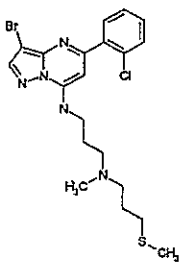
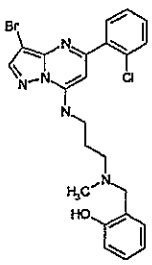
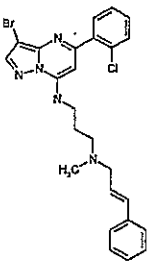
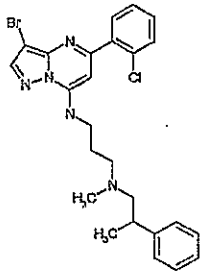
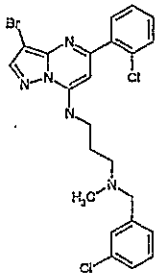
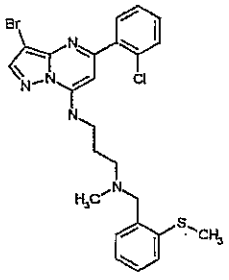
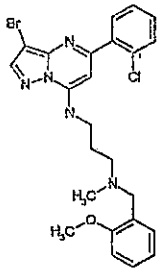
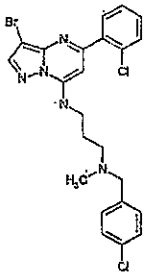
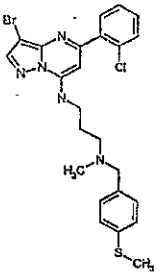
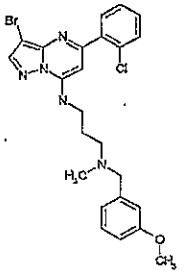
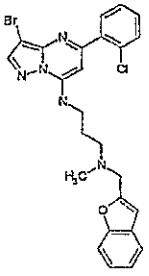
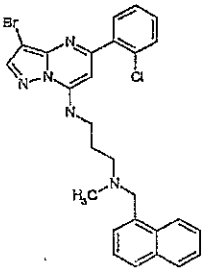
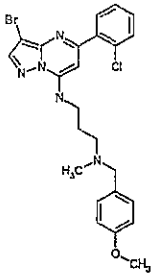
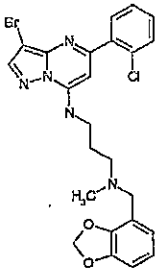
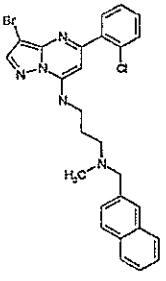
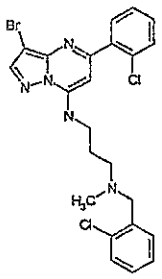
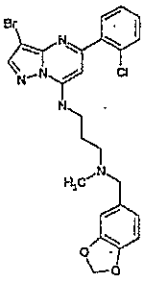
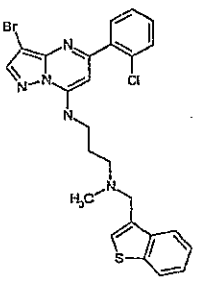
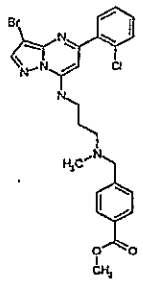
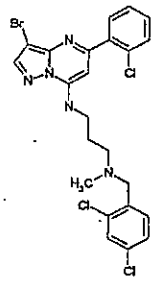
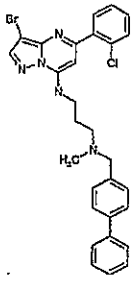
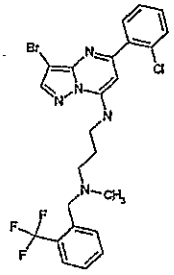
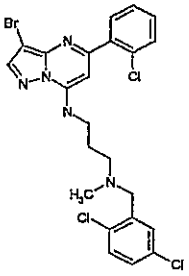
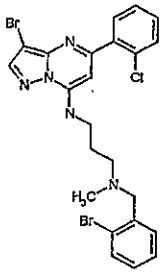
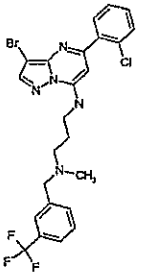
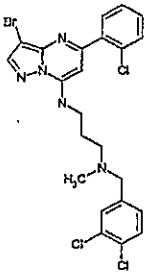
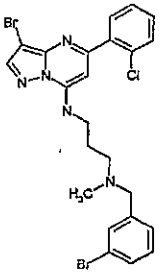
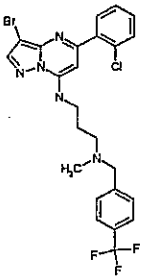
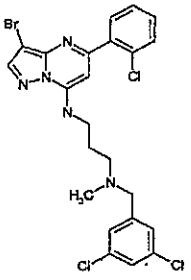
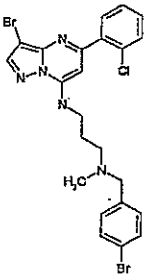
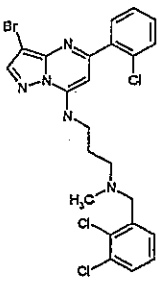
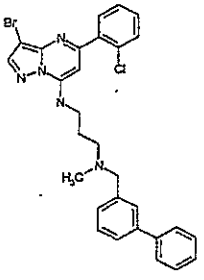
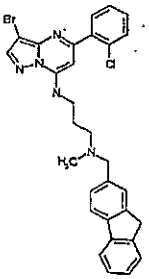
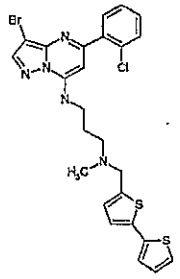
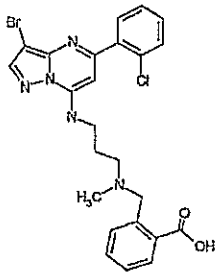
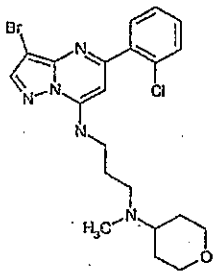
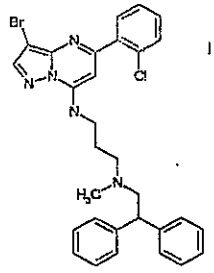
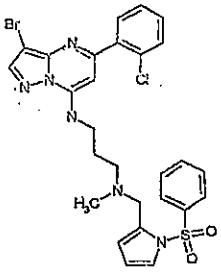
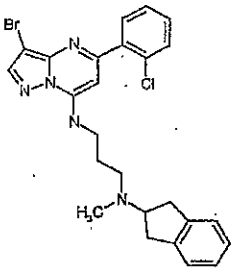
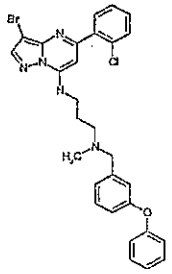
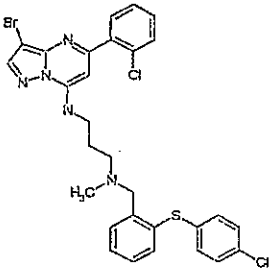
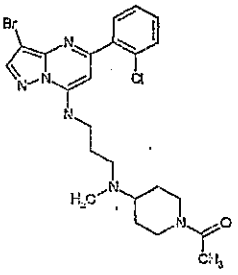
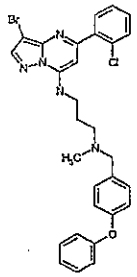
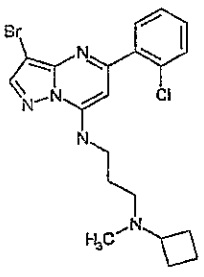
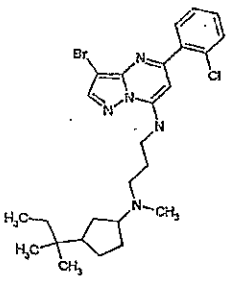
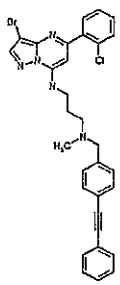
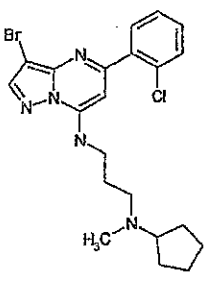
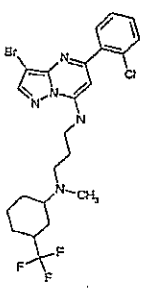
Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8261 2. 544.3		1. 8266 2. 560.31
	1. 8262 2. 546.3		1. 8267 2. 566.31
	1. 8263 2. 546.3		1. 8268 2. 591.33
	1. 8264 2. 558.31		1. 8269 2. 595.33
	1. 8265 2. 558.31		1. 8270 2. 555.31

TABLE 83

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8301 2. 465.882		1. 8306 2. 491.79		1. 8311 2. 501.825
	1. 8302 2. 465.885		1. 8307 2. 491.792		1. 8312 2. 501.833
	1. 8303 2. 475.827		1. 8308 2. 492.787		1. 8313 2. 510.822
	1. 8304 2. 479.89		1. 8309 2. 499.846		1. 8314 2. 510.821
	1. 8305 2. 483.829		1. 8310 2. 501.826		1. 8315 2. 511.841

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8316 2. 513.855		1. 8321 2. 519.783		1. 8326 2. 531.804
	1. 8317 2. 515.832		1. 8322 2. 519.781		1. 8327 2. 531.812
	1. 8318 2. 515.832		1. 8323 2. 525.813		1. 8328 2. 535.83
	1. 8319 2. 515.837		1. 8324 2. 529.806		1. 8329 2. 535.831
	1. 8320 2. 519.782		1. 8325 2. 529.81		1. 8330 2. 541.788

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8331 2. 543.821		1. 8336 2. 554.727		1. 8341 2. 561.838
	1. 8332 2. 553.797		1. 8337 2. 554.738		1. 8342 2. 564.714
	1. 8333 2. 553.796		1. 8338 2. 554.738		1. 8343 2. 564.704
	1. 8334 2. 553.795		1. 8339 2. 554.727		1. 8344 2. 564.72
	1. 8335 2. 554.73		1. 8340 2. 561.836		1. 8345 2. 573.823

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8346 2. 573.735		1. 8351 2. 529.808		1. 8356 2. 479.856
	1. 8347 2. 575.839		1. 8352 2. 614.763		1. 8357 2. 511.842
	1. 8348 2. 577.818		1. 8353 2. 628.727		1. 8358 2. 520.859
	1. 8349 2. 577.814		1. 8354 2. 449.857		1. 8359 2. 533.903
	1. 9350 2. 585.818		1. 8355 2. 463.867		1. 8360 2. 545.823

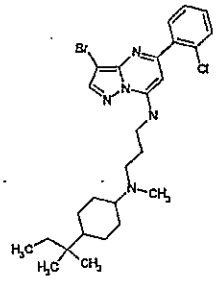
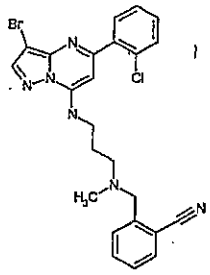
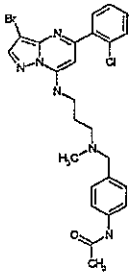
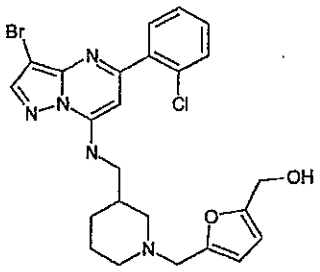
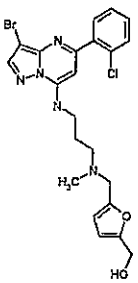
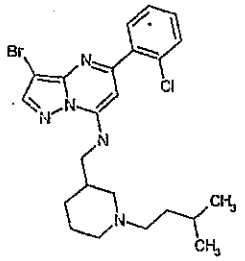
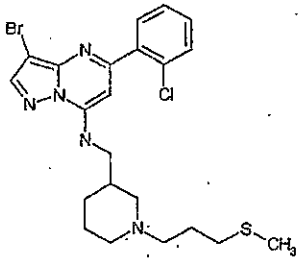
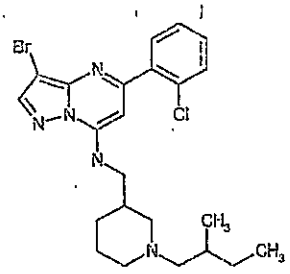
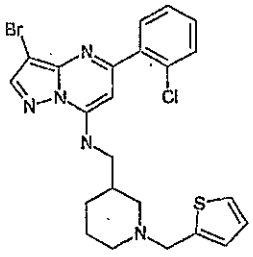
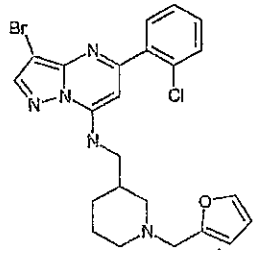
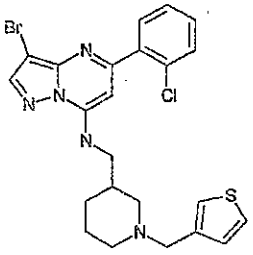
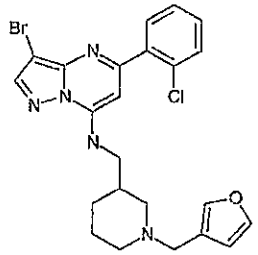
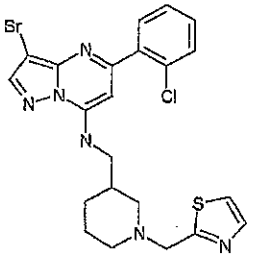
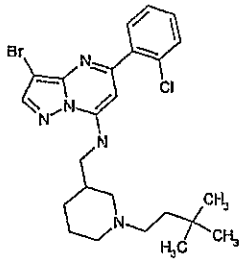
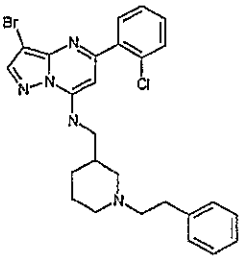
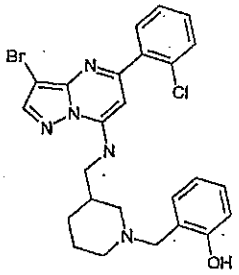
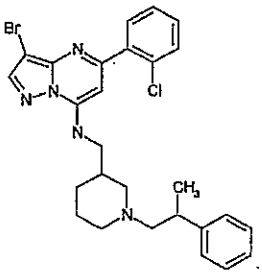
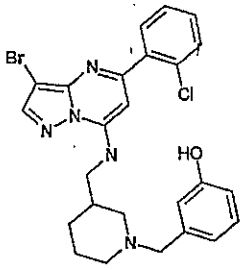
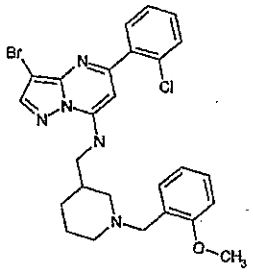
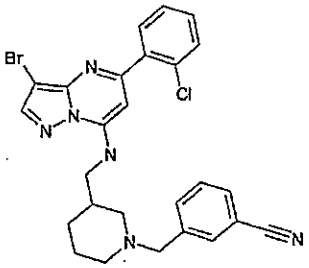
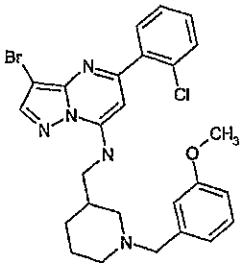
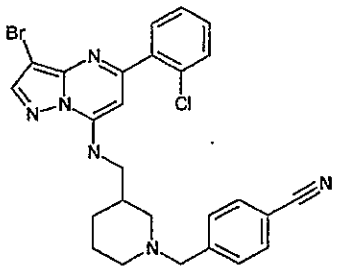
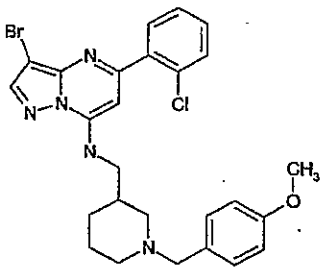
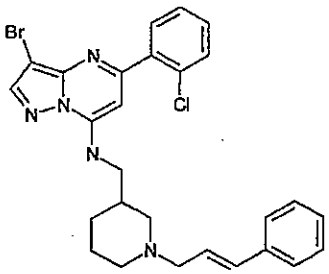
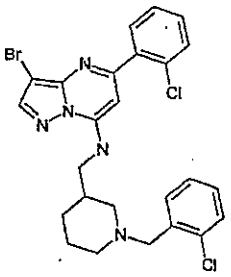
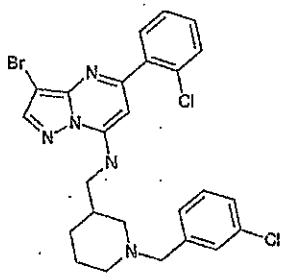
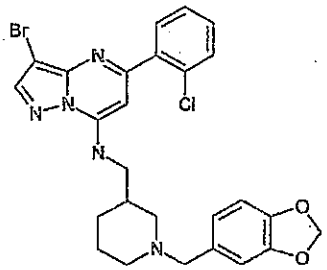
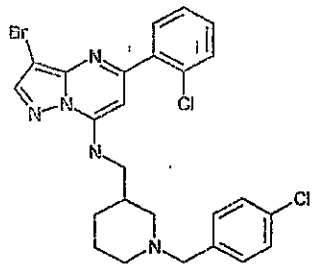
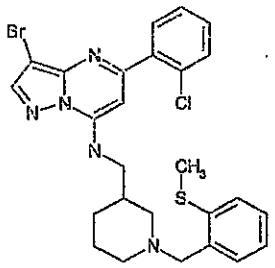
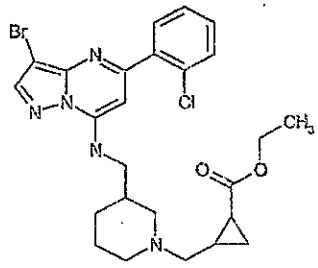
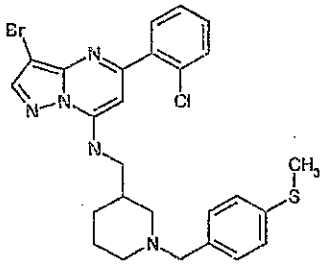
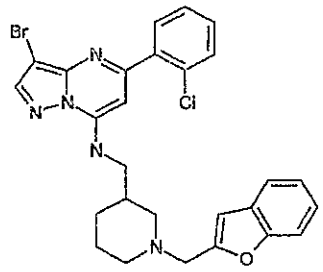
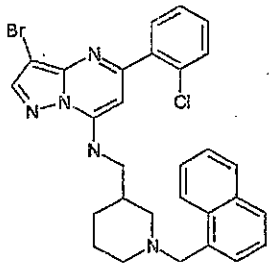
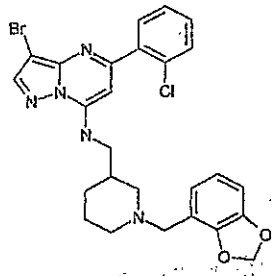
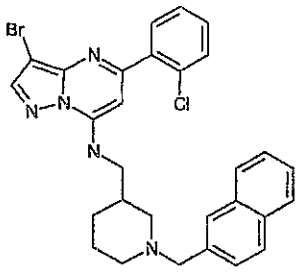
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	1. 8363 2. 542.833
	1. 8364 2. 573.815
	1. 8365 2. 547.798

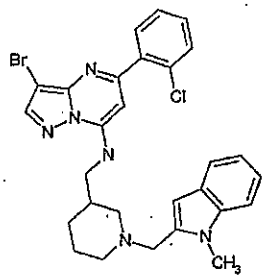
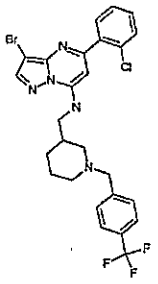
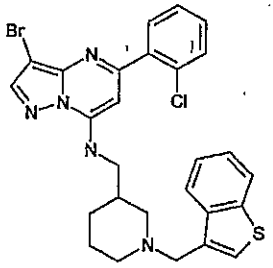
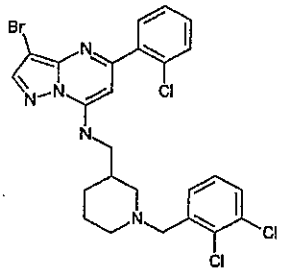
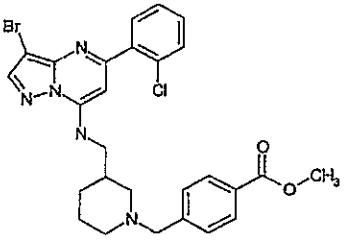
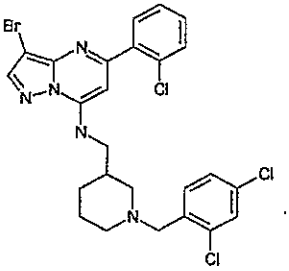
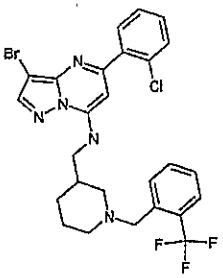
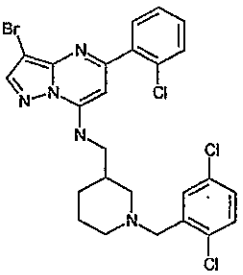
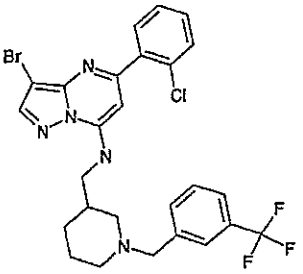
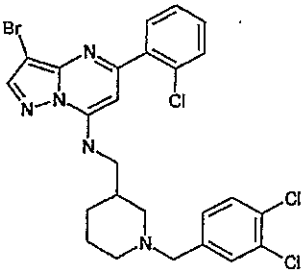
TABLE 84

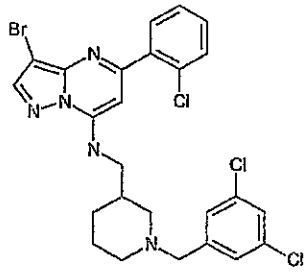
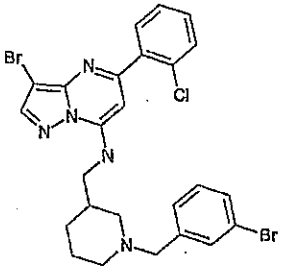
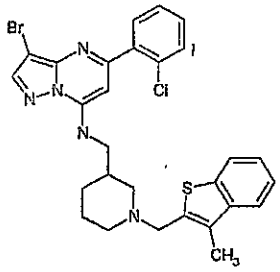
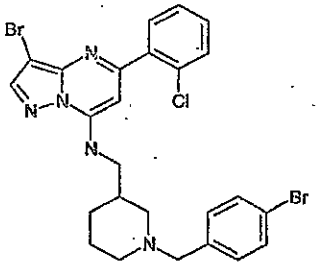
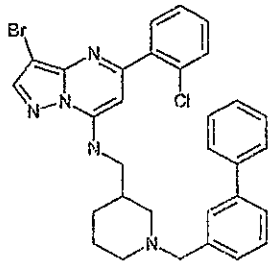
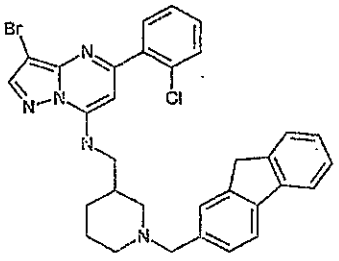
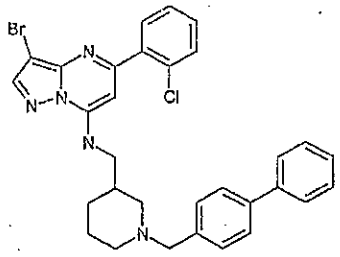
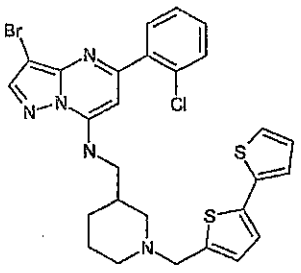
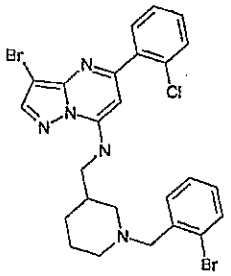
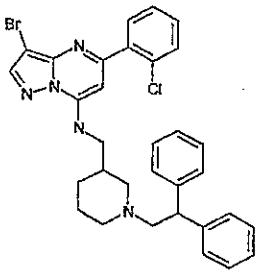
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	1. 8402 2. 491.883		1. 8407 2. 517.795
	1. 8403 2. 501.823		1. 8408 2. 517.799
	1. 8404 2. 501.827		1. 8409 2. 518.796
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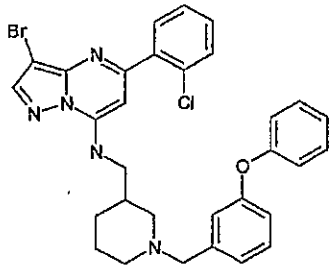
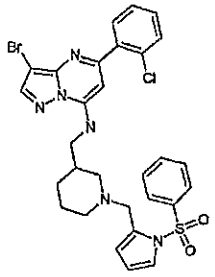
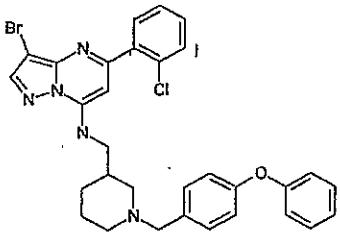
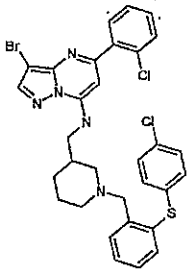
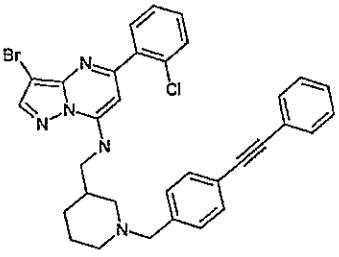
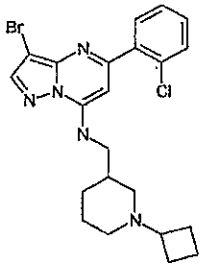
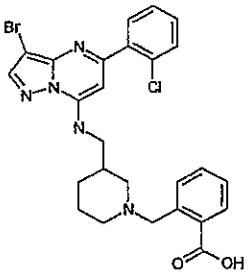
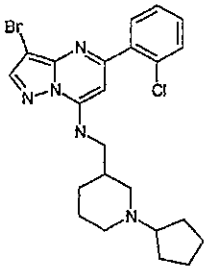
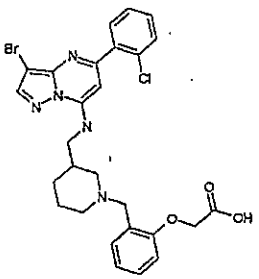
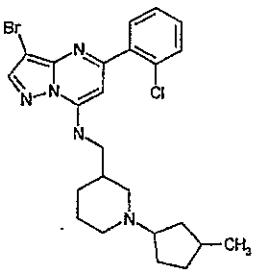
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	1. 8411 2. 527.831		1. 8416 2. 539.859
	1. 8412 2. 527.833		1. 8417 2. 541.84
	1. 8413 2. 536.833		1. 8418 2. 541.843
	1. 8414 2. 536.829		1. 8419 2. 541.843
	1. 8415 2. 537.845		1. 8420 2. 546.786

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8421 2. 546.792		1. 8426 2. 555.817
	1. 8422 2. 546.79		1. 8427 2. 557.81
	1. 8423 2. 547.854		1. 8428 2. 557.813
	1. 8424 2. 551.826		1. 8429 2. 561.839
	1. 8425 2. 555.819		1. 8430 2. 561.833

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	1. 8431 2. 564.841		1. 8436 2. 579.794
	1. 8432 2. 567.795		1. 8437 2. 580.738
	1. 8433 2. 569.823		1. 8438 2. 580.732
	1. 8434 2. 579.797		1. 8439 2. 580.733
	1. 8435 2. 579.796		1. 8440 2. 580.734

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8441 2. 580.726		1. 8446 2. 590.726
	1. 8442 2. 581.805		1. 8447 2. 590.724
	1. 8443 2. 587.844		1. 8448 2. 599.834
	1. 8444 2. 587.835		1. 8449 2. 599.749
	1. 8445 2. 590.723		1. 8450 2. 601.854

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8451 2. 603.831		1. 8456 2. 640.779
	1. 8452 2. 603.83		1. 8457 2. 654.746
	1. 8453 2. 611.83		1. 8458 2. 475.871
	1. 8454 2. 555.824		1. 8459 2. 489.879
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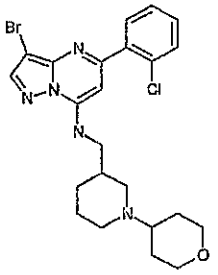
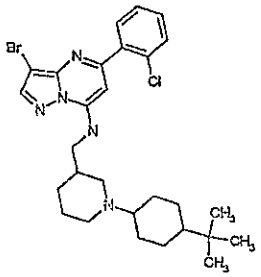
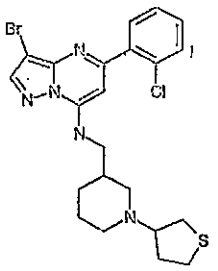
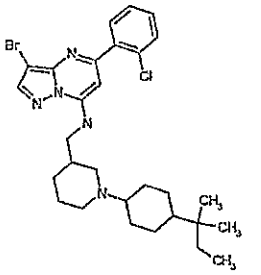
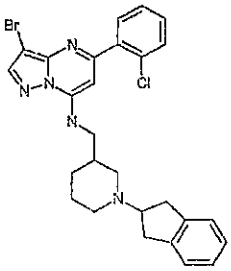
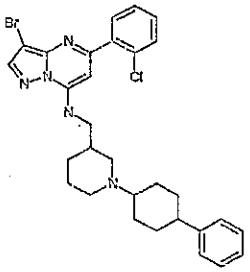
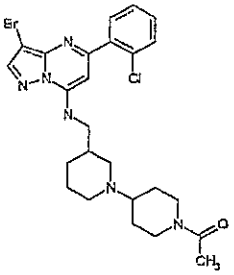
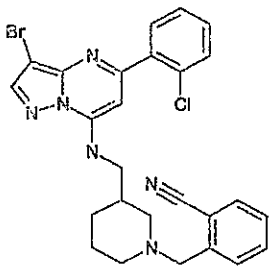
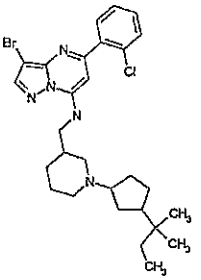
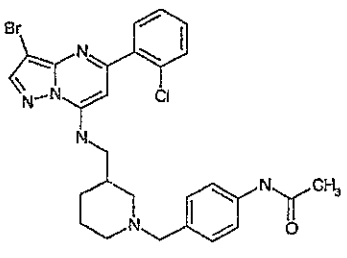
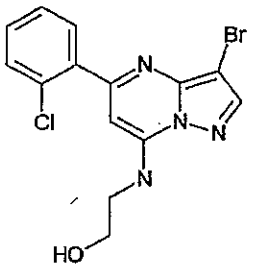
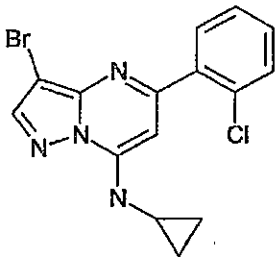
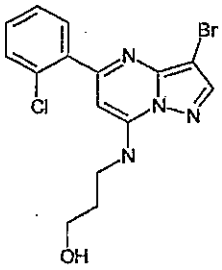
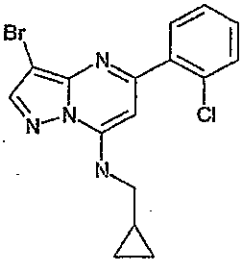
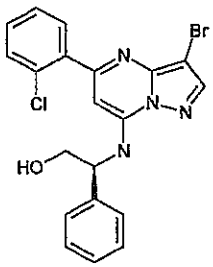
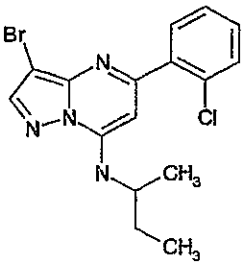
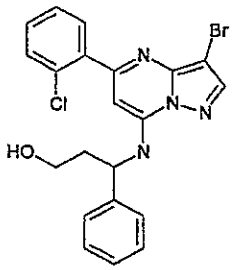
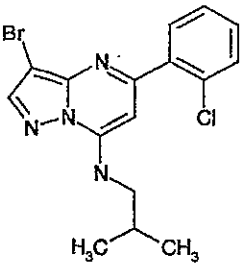
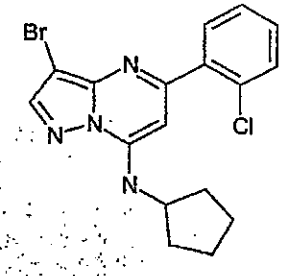
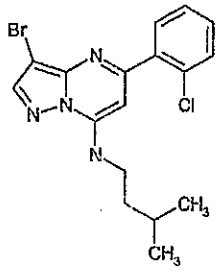
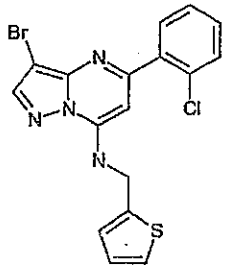
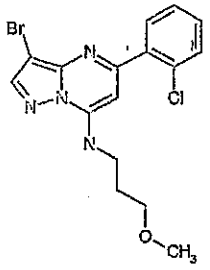
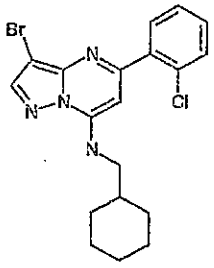
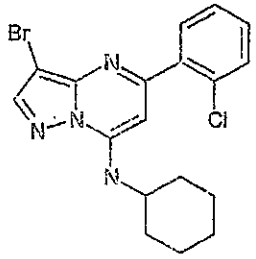
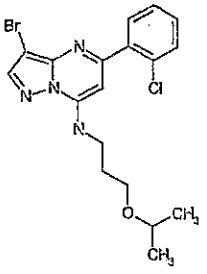
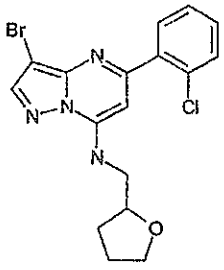
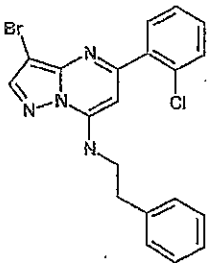
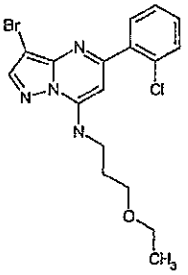
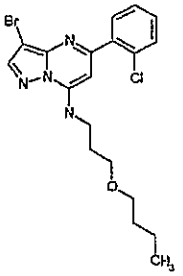
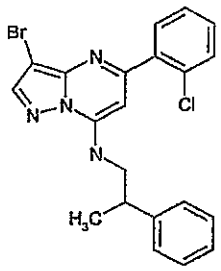
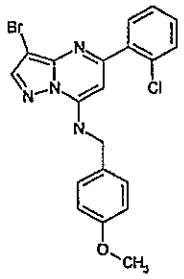
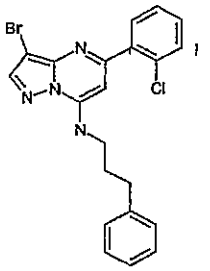
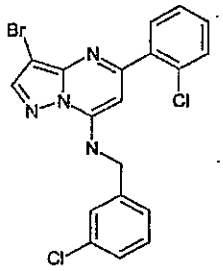
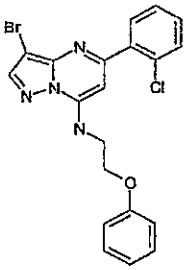
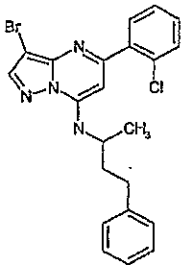
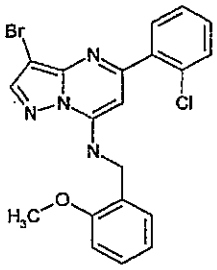
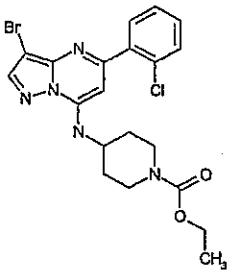
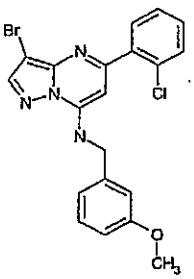
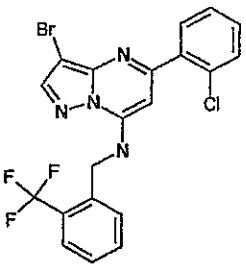
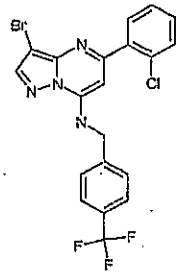
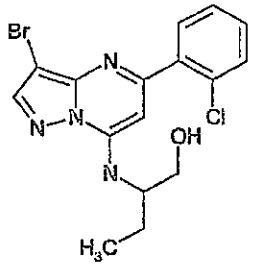
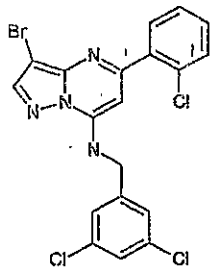
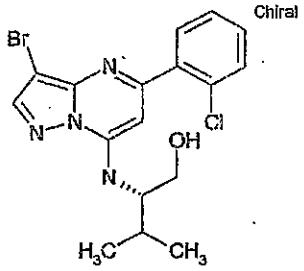
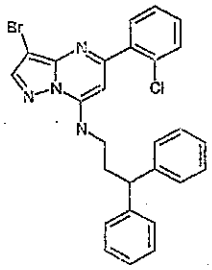
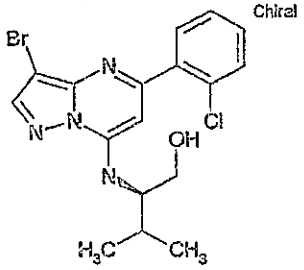
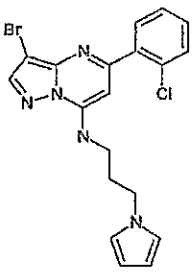
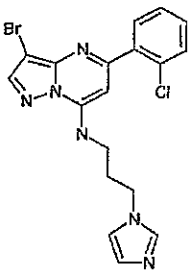
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	1. 8461 2. 505.866		1. 8466 2. 559.919
	1. 8462 2. 507.828		1. 8467 2. 573.925
	1. 8463 2. 537.859		1. 8468 2. 579.876
	1. 8464 2. 546.877		1. 8469 2. 536.831
	1. 8465 2. 559.921		1. 8470 2. 568.843

TABLE 85

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1.8501 2. 367.2		1. 8506 2. 443.1
	1. 8502 2. 381.2		1. 8507 2. 323.1
	1. 8503 2. 443.2		1. 8508 2. 365.2
	1. 8504 2. 457.3		1. 8509 2. 379.21
			1. 8510 2. 381.21

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8511 2. 381.21		1. 8516 2. 409.22
	1. 8512 2. 393.22		1. 8517 2. 411.23
	1. 8513 2. 395.22		1. 8518 2. 421.23
	1. 8514 2. 397.22		1. 8519 2. 421.23
	1. 8515 2. 407.22		1. 8520 2. 425.23

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8521 2. 429.24		1. 8526 2. 445.24
	1. 8522 2. 439.24		1. 8527 2 .444.24
	1. 8523 2. 443.24		1. 8528 2. 444.24
	1. 8524 2. 443.24		1. 8529 2. 446.25
	1. 8525 2. 445.24		1. 8530 2. 457.25

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8531 2. 480.26		1. 8536 2. 431.24
	1. 8532 2. 483.27	 Chiral	1. 8537 2. 429.24
	1. 8533 2. 483.27	 Chiral	1. 8538 2. 397.22
			1. 8539 2. 411.23
			1. 8540 2. 411.23

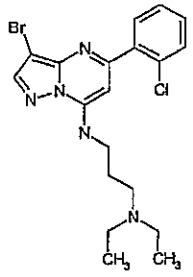
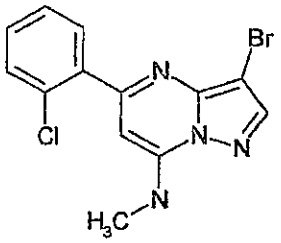
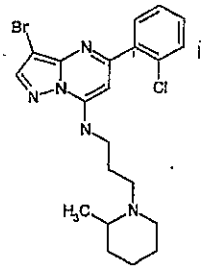
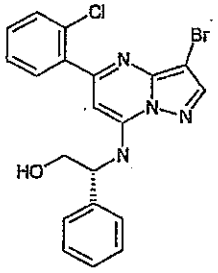
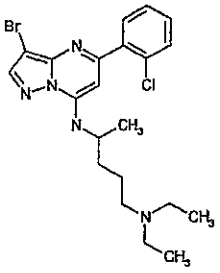
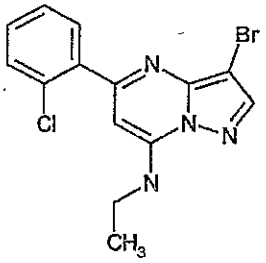
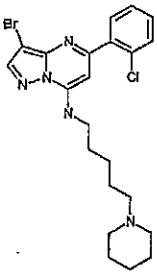
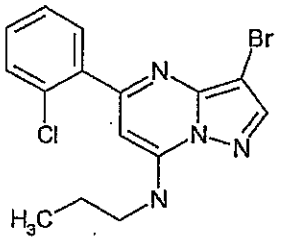
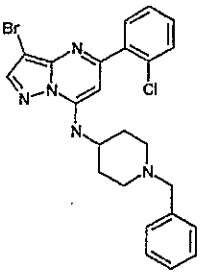
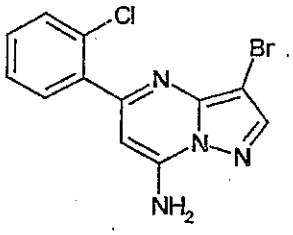
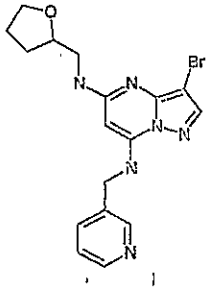
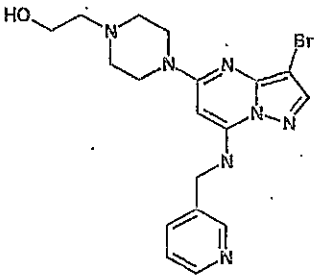
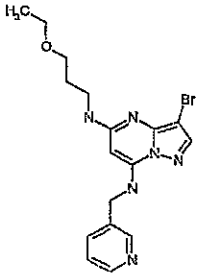
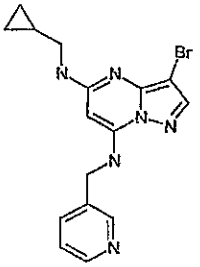
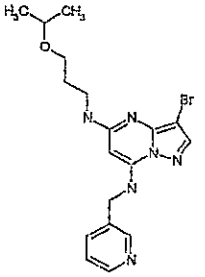
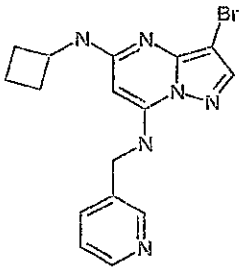
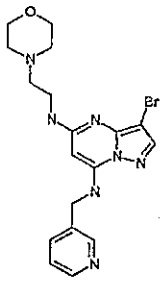
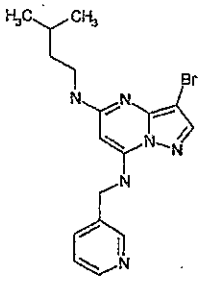
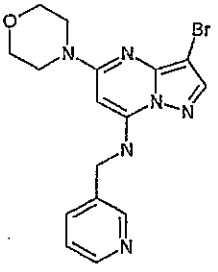
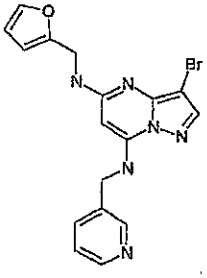
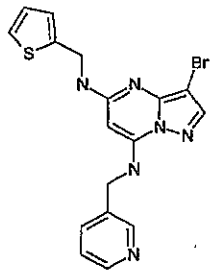
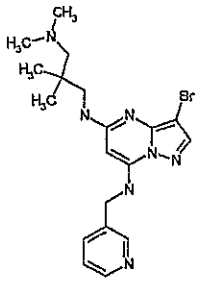
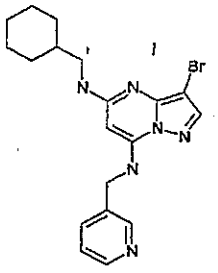
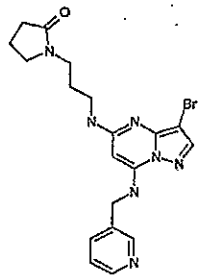
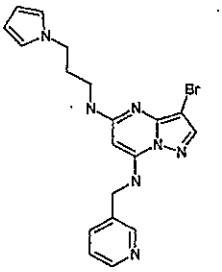
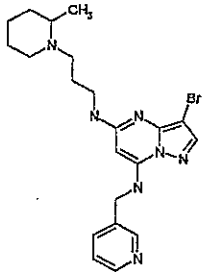
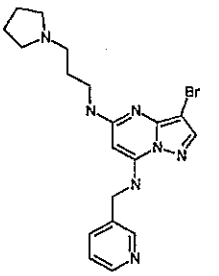
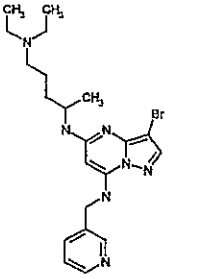
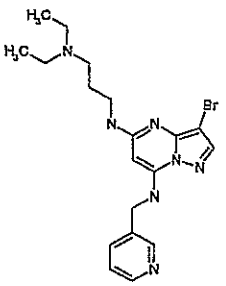
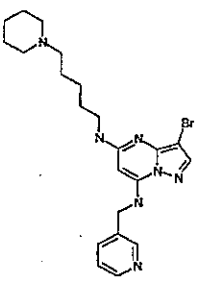
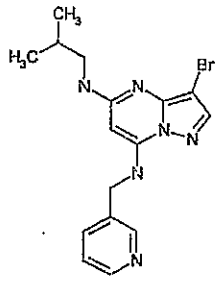
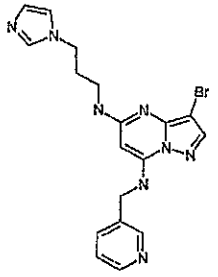
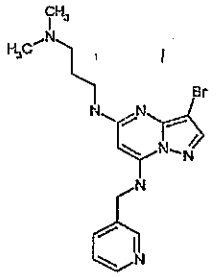
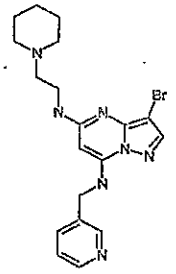
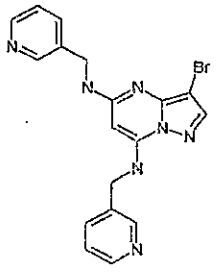
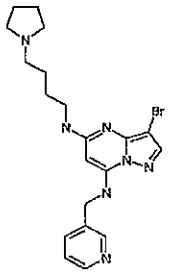
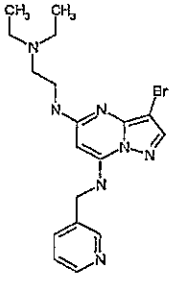
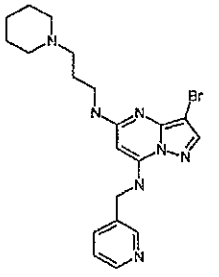
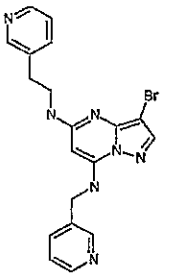
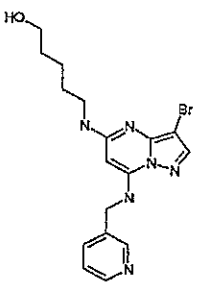
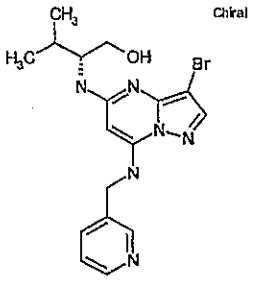
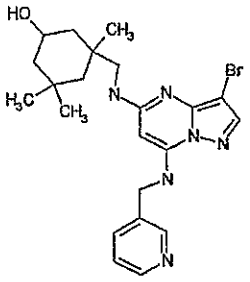
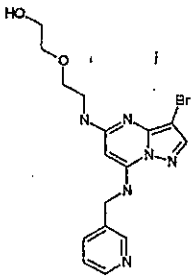
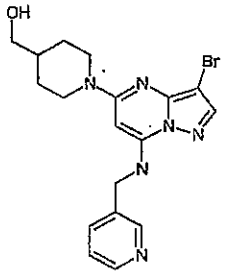
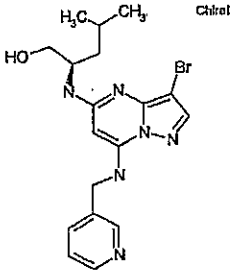
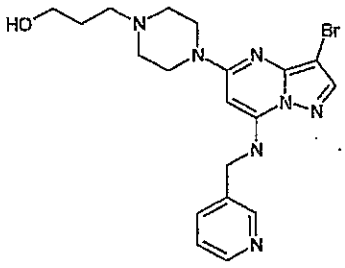
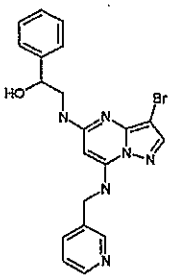
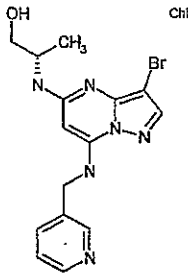
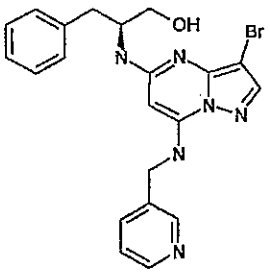
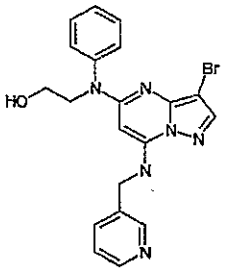
Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
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	1. 8542 2. 433.24		1. 8548 2. 498.27
	1. 8543 2. 438.24		1. 8549 2. 365.1
	1. 8545 2. 464.26		1. 8549 2. 337.1
	1. 8546 2. 466.26		1. 8550 2. 351.1

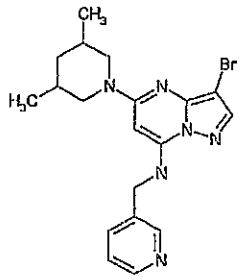
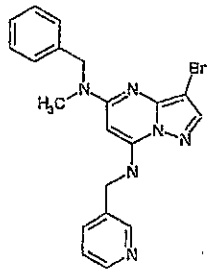
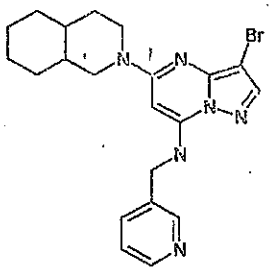
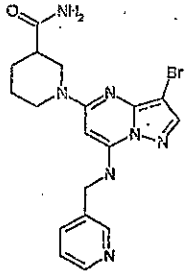
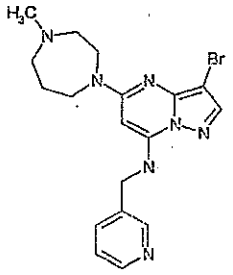
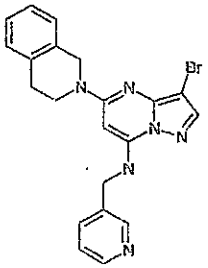
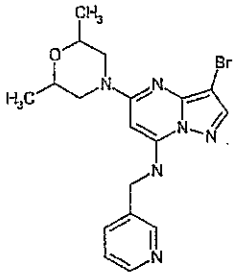
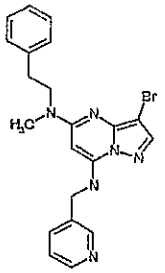
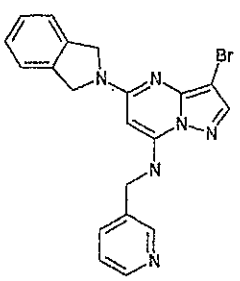
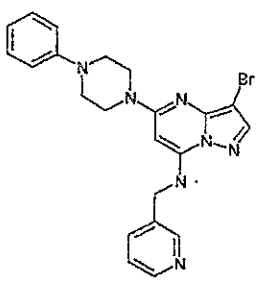
TABLE 86

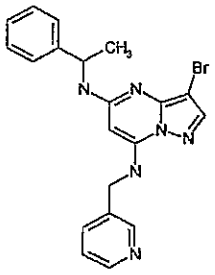
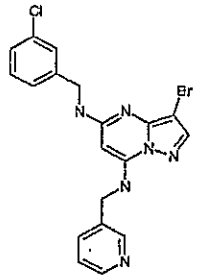
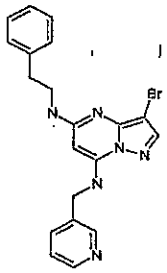
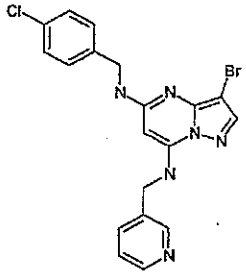
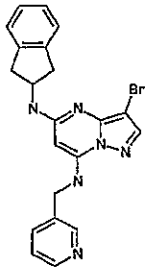
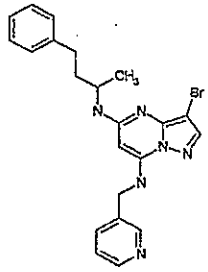
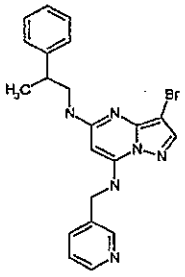
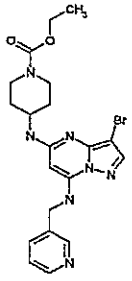
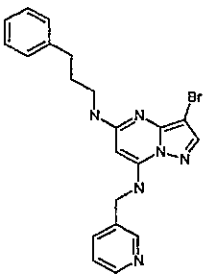
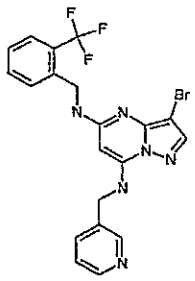
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	1. 8601 2. 403.22		1. 8606 2. 434.24
	1. 8602 2. 407.22		1. 8607 2. 375.21
	1. 8603 2. 421.23		1. 8608 2. 375.21
	1. 8604 2. 434.24		1. 8609 2. 391.22
	1. 8605 2. 391.22		1. 8610 2. 399.22

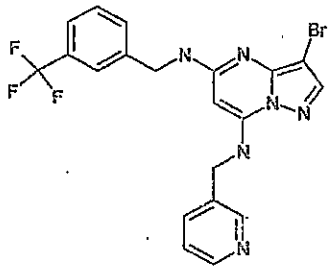
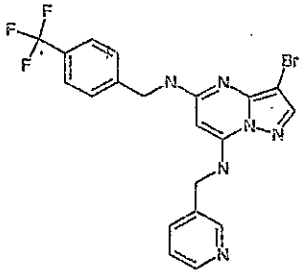
Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8611 2. 417.23		1. 8616 2. 434.24
	1. 8612 2. 415.23		1. 8617 2. 446.25
	1. 8613 2. 428.24		1. 8618 2. 460.25
	1. 8614 2. 430.24		1. 8619 2. 459.25
	1. 8615 2. 434.24		1. 8620 2. 474.26

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8621 2. 377.21		1. 8626 2. 427.23
	1. 8622 2. 406.22		1. 8627 2. 431.24
	1. 8623 2. 412.23		1. 8628 2. 446.25
	1. 8624 2. 420.23		1. 8629 2. 446.25
	1. 8625 2. 426.23		1. 8630 2. 407.2

Product	1. Ex. 2. m/z	Product	1. Ex. 2. m/z
	1. 8631 2. 407.2		1. 8636 2. 475.3
	1. 8632 2. 409.2		1. 8637 2. 419.2
	1. 8633 2. 421.2		1. 8638 2. 448.2
	1. 8634 2. 439.2		1. 8639 2. 379.2
	1. 8635 2. 455.3		1. 8640 2. 437.2

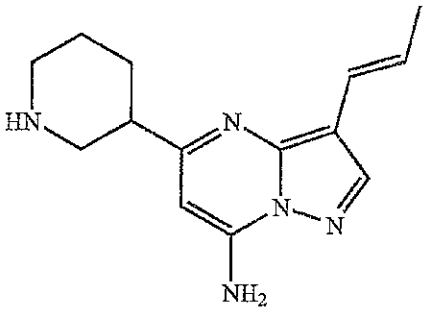
Product	1. Ex. 2. m/z	Product	1. Ex: 2. m/z
	1. 8641 2. 415.23		1. 8646 2. 423.23
	1. 8642 2. 443.24		1. 8647 2. 430.24
	1. 8643 2. 416.23		1. 8648 2. 437.24
	1. 8644 2. 417.23		1. 8649 2. 439.24
	1. 8645 2. 423.23		1. 8650 2. 466.26

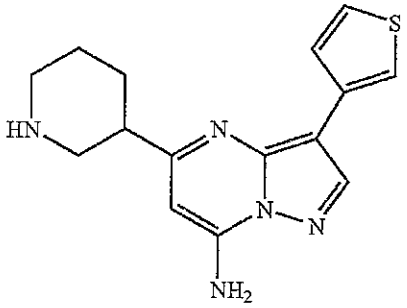
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	1. 8651 2. 423.23		1. 8656 2. 445.24
	1. 8652 2. 425.23		1. 8657 2. 445.24
	1. 8653 2. 437.24		1. 8658 2. 451.25
	1. 8654 2. 437.24		1. 8659 2. 476.26
	1. 8655 2. 437.24		1. 8660 2. 479.26

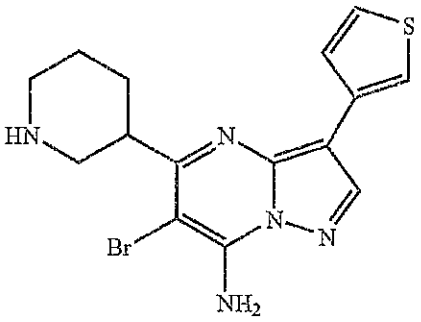
Product	1. Ex. 2. m/z
	1. 8661 2. 477.26
	1. 8662 2. 479.26

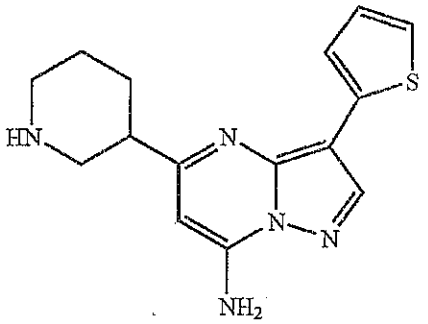
What is claimed is:

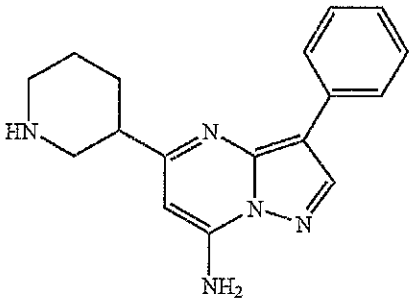
1. A compound selected from the group consisting of the compounds of the formula:

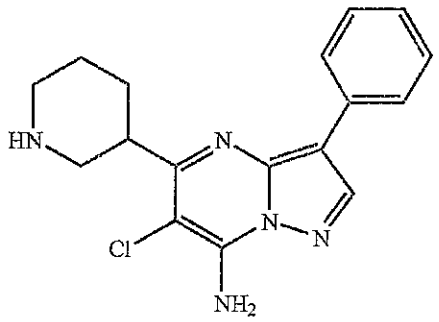


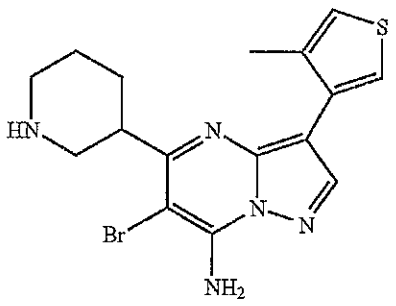


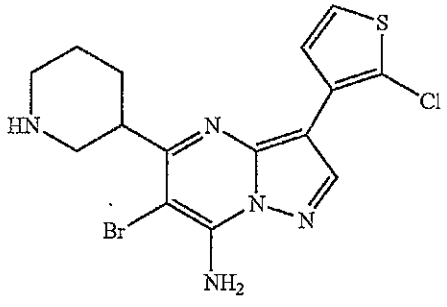


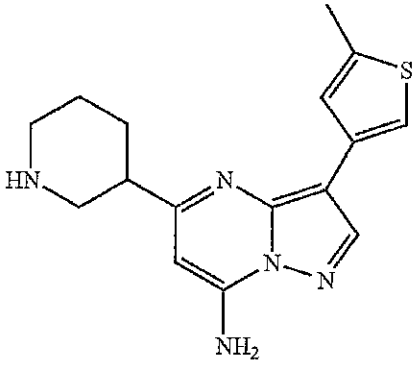


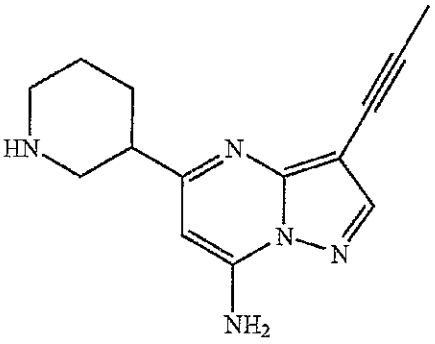


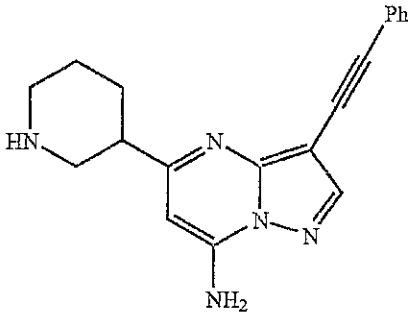


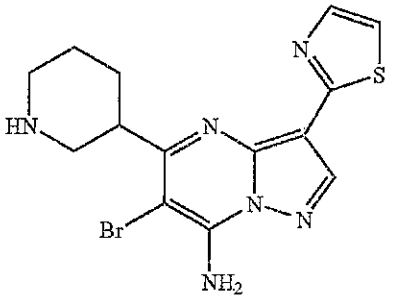


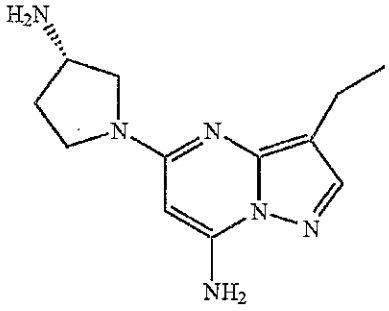


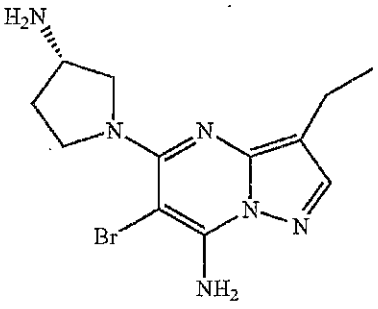


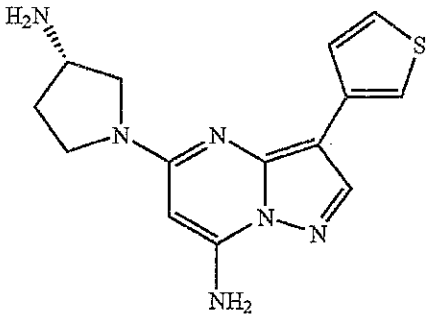


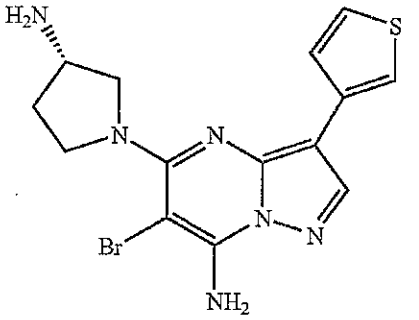


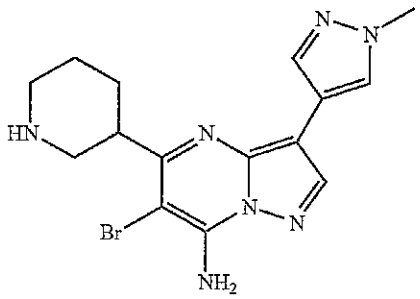


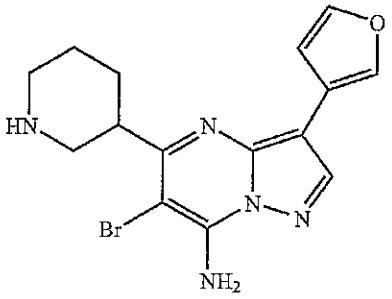


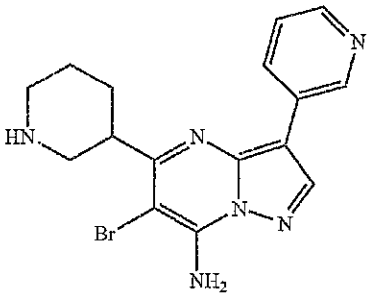


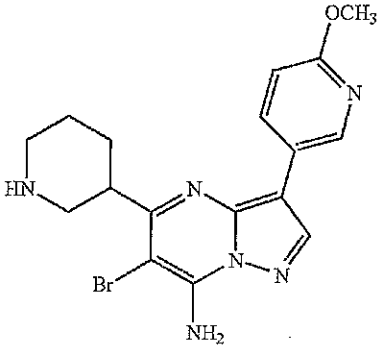


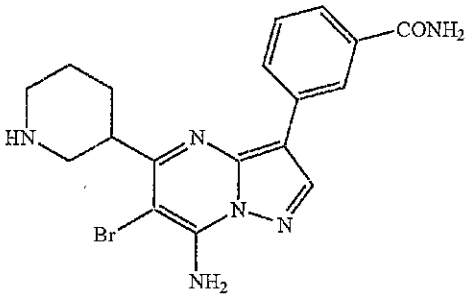


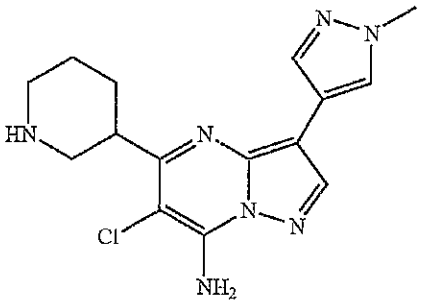


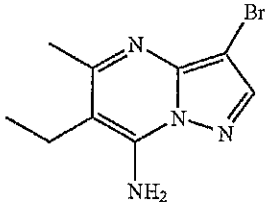


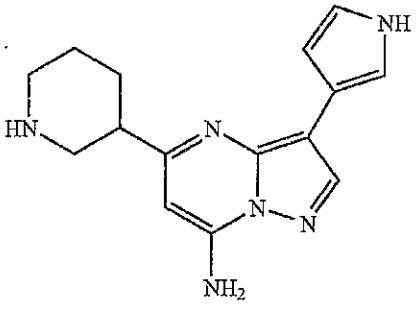


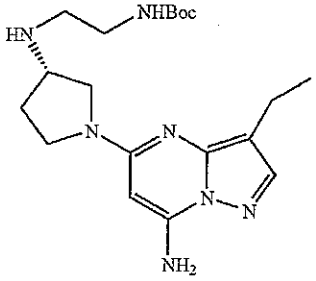


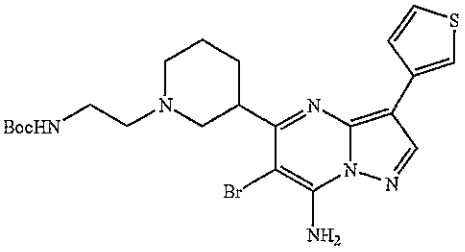


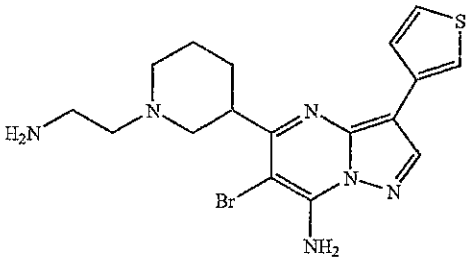


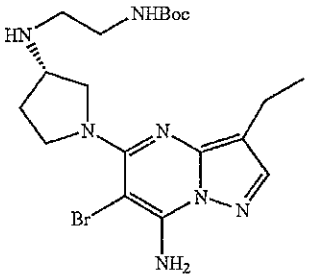


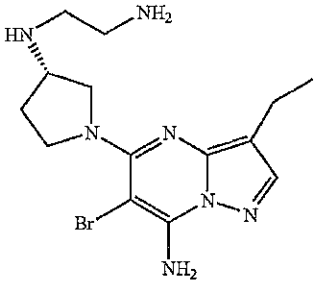


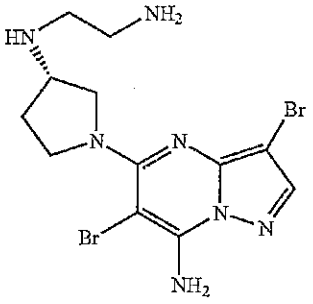


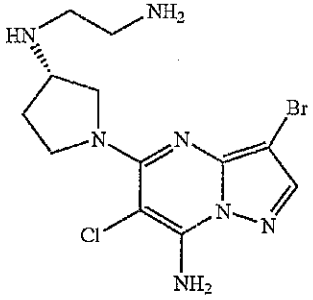


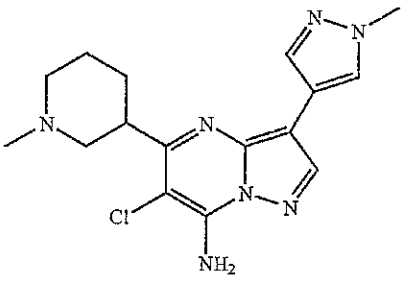


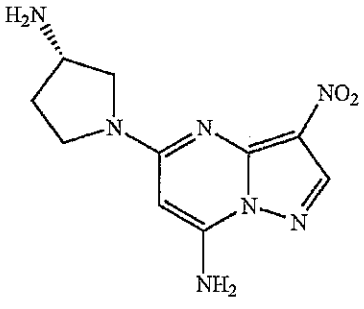


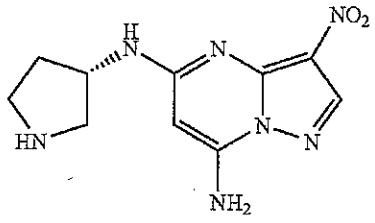


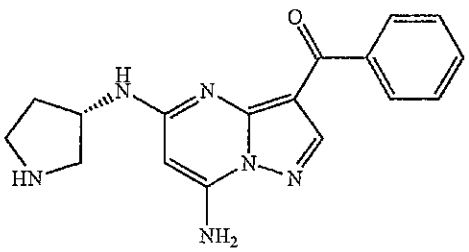


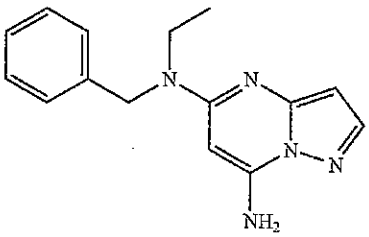


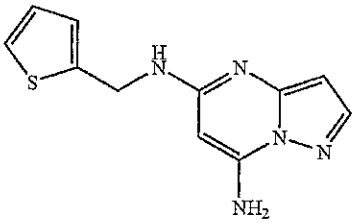


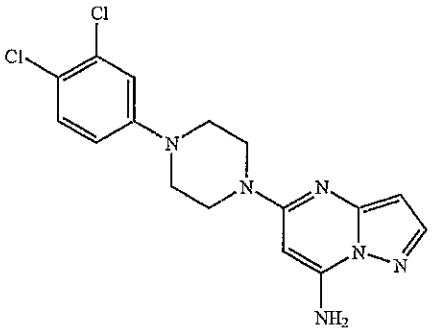


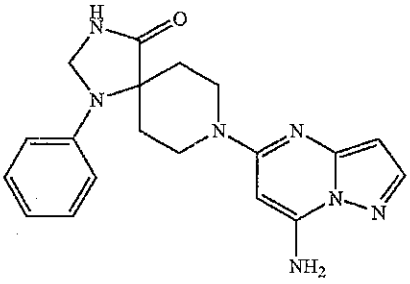


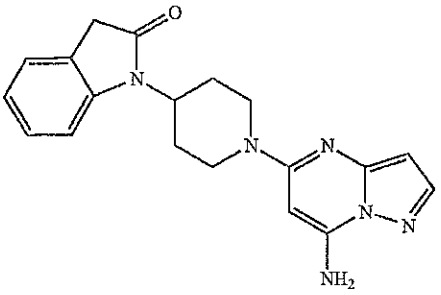


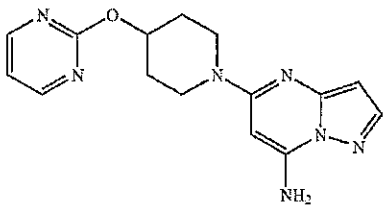


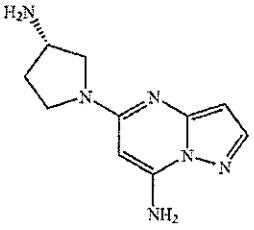


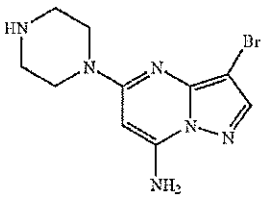


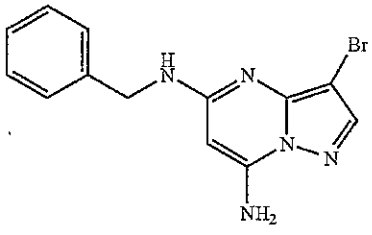


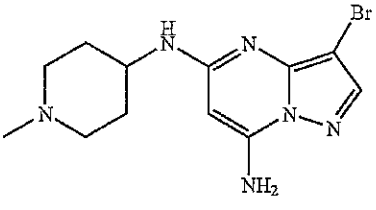


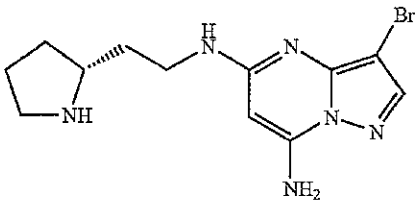


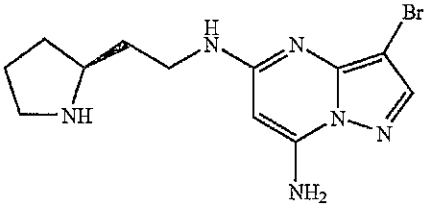


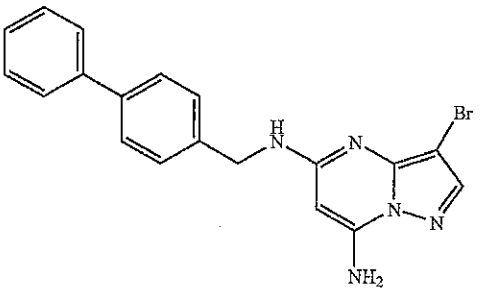


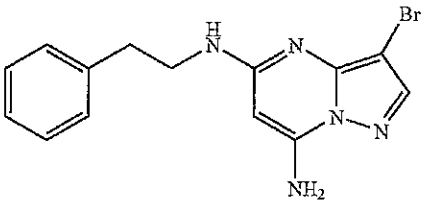


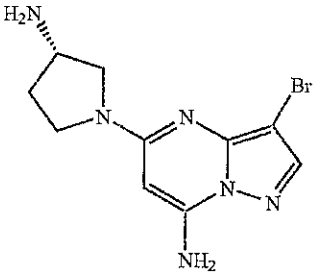


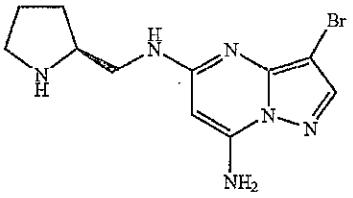


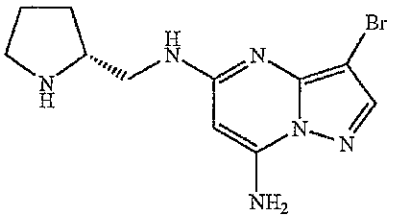


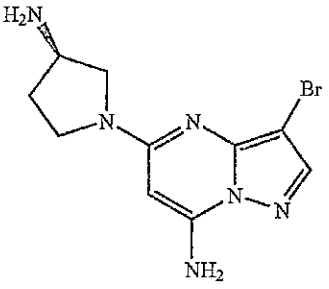


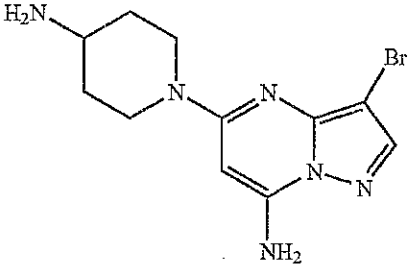


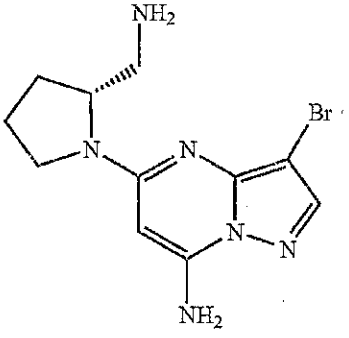


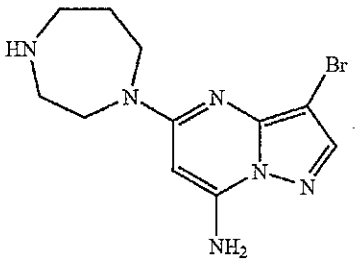


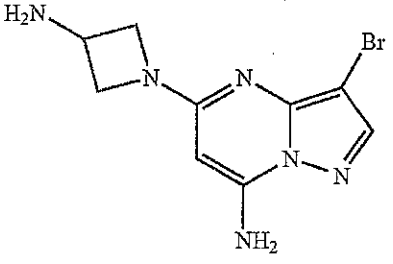


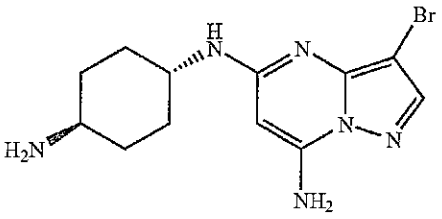


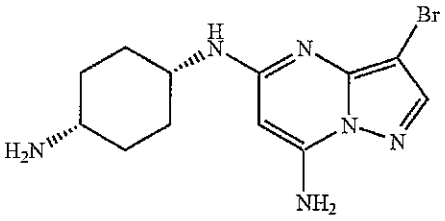


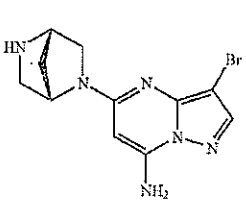


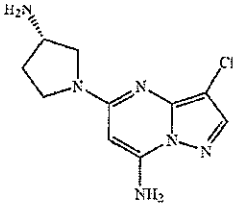


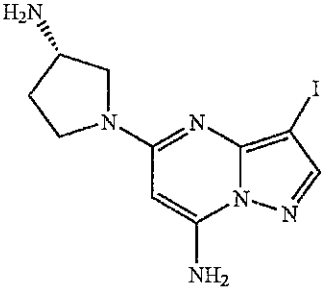


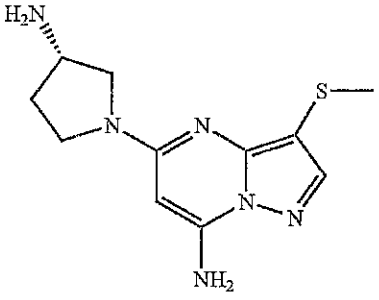


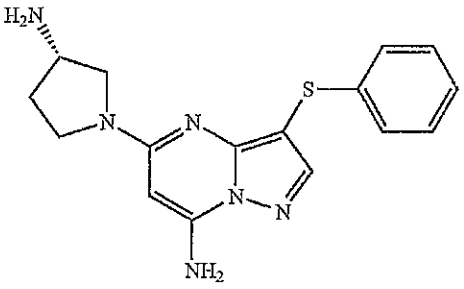


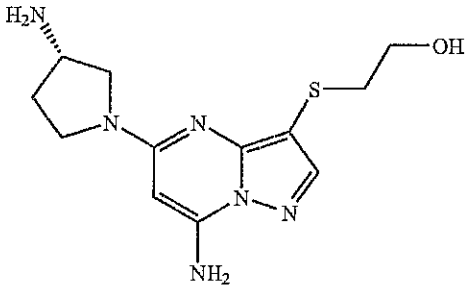


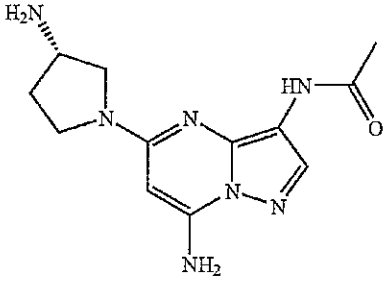


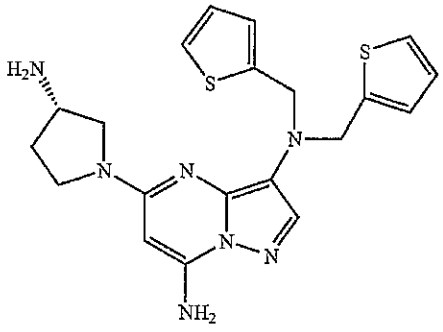


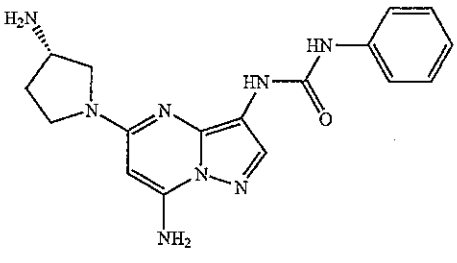


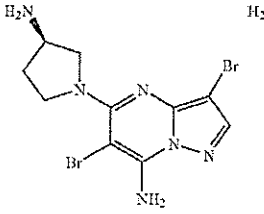
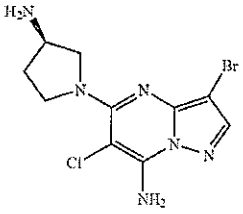
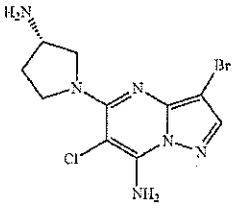
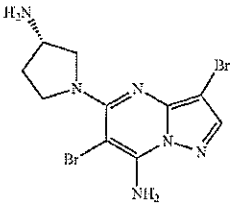
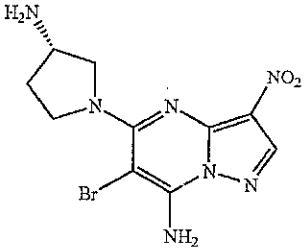
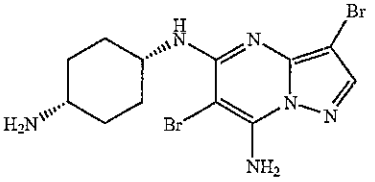
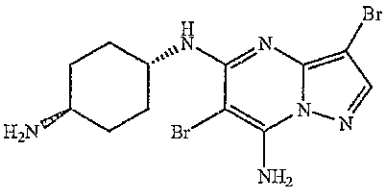
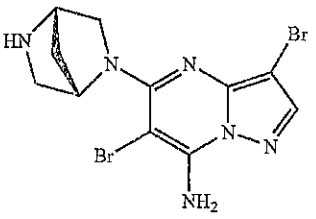
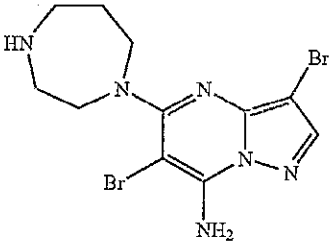
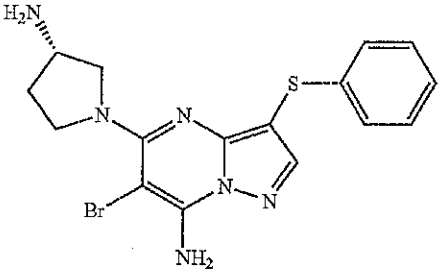
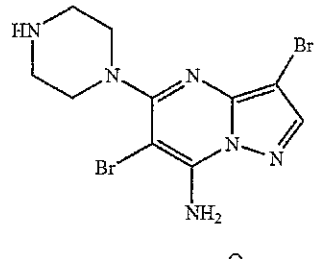
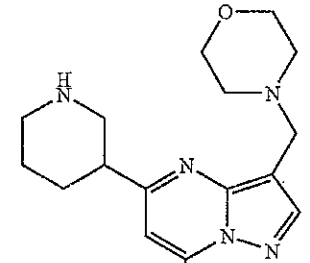
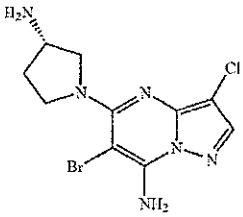
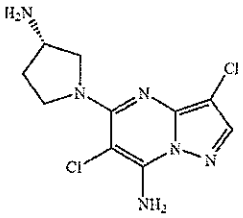


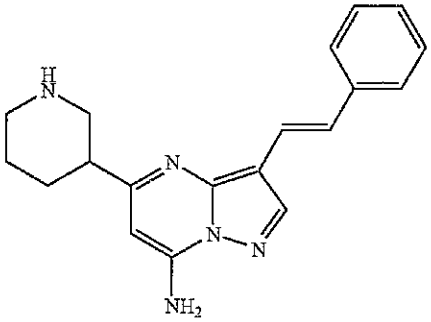


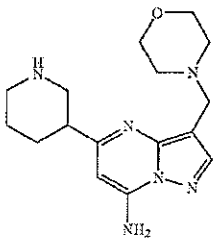


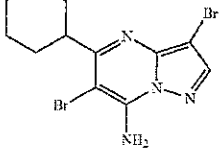


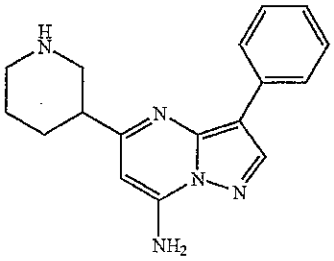


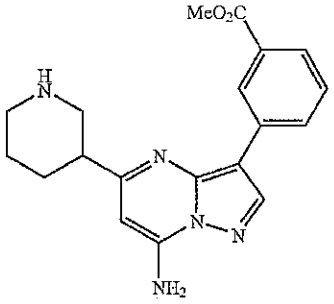















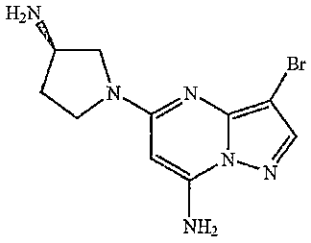


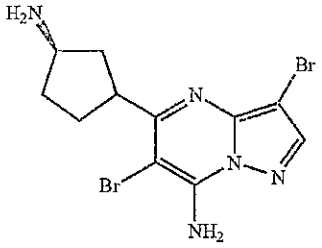


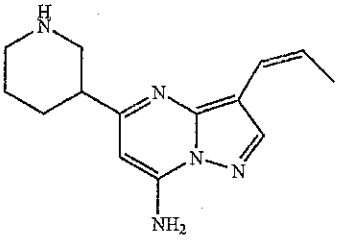


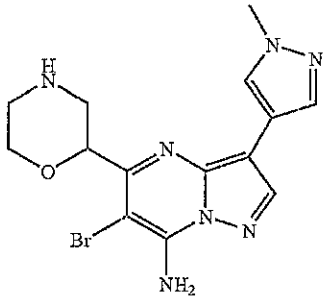


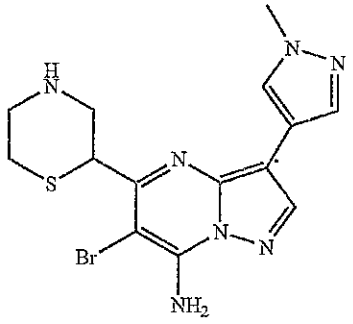


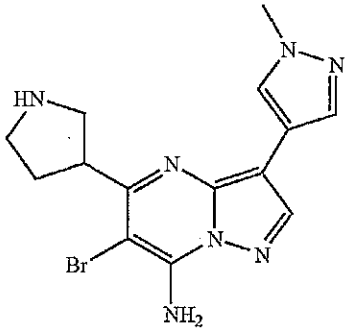


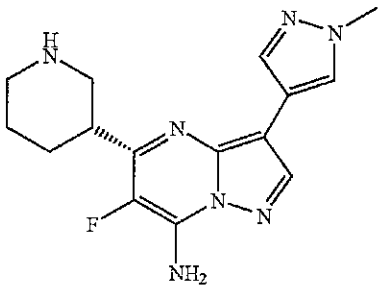




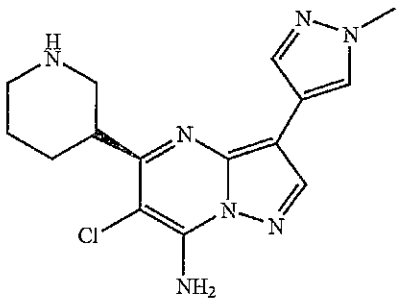


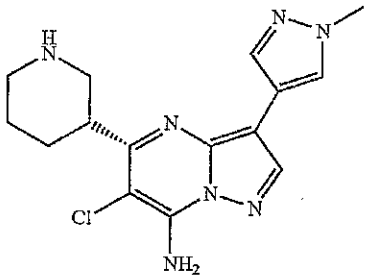






and





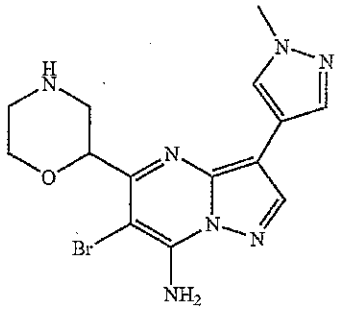
2. A pharmaceutical composition comprising a therapeutically effective amount of at least one compound of claim 1, or a pharmaceutically acceptable salt, ester or prodrug thereof, in combination with at least one pharmaceutically acceptable carrier.

3. The pharmaceutical composition of claim 2, additionally comprising one more anti-cancer agents selected from the group consisting of a cytostatic agent, taxotere, taxol, etoposide, irinotecan, camptostar, topotecan, paclitaxel, docetaxel, epothilones, tamoxifen, 5-fluorouracil, methoxtrexate, temozolomide, cyclophosphamide, SCH66336, R115777, L778,123, BMS214662, Iressa®, (gefitinib) Tarceva® (erlotinib hydrochloride), antibodies to EGFR, Gleevec® (imatinib), intron, ara-C, adriamycin, cytoxan, gemcitabine, Uracil mustard, Chlormethine, Ifosfamide, Melphalan, Chlorambucil, Pipobroman, Triethylenemelamine, Triethylenethiophosphoramine, Busulfan, Carmustine, Lomustine, Streptozocin, Dacarbazine, Floxuridine, Cytarabine, 6-Mercaptopurine, 6-Thioguanine, Fludarabine phosphate, oxaliplatin, leucovirin, ELOXATIN™, Pentostatine, Vinblastine, Vincristine, Vindesine, Bleomycin, Dactinomycin, Daunorubicin, Doxorubicin, Epirubicin, Idarubicin, Mithramycin, Deoxycoformycin, Mitomycin-C, L-Asparaginase, Teniposide 17α-Ethinylestradiol, Diethylstilbestrol, Testosterone, Prednisone, Fluoxymesterone, Dromostanolone propionate, Testolactone, Megestrolacetate, Methylprednisolone, Methyltestosterone, Prednisolone, Triamcinolone, Chlorotrianisene, Hydroxyprogesterone, Aminoglutethimide, Estramustine, Medroxyprogesteroneacetate, Leuprolide, Flutamide, Toremifene,

geserelin, Cisplatin, Carboplatin, Hydroxyurea, Amsacrine, Procarbazine, Mitotane, Mitoxantrone, Levamisole, Navelbene, Anastrazole, Letrazole, Capecitabine, Reloxafine, Droloxafine, Hexamethylmelamine, Avastin, herceptin, Bexxar, Velcade, Zevalin, Trisenox, Xeloda, Vinorelbine, Porfimer, Erbitux, Liposomal, Thiotepa, Altretamine, Melphalan, Transtuzumab, Lerozole, Fulvestrant, Extremestane, Fulvestrant, Ifosfomide, Rituximab, C225®, and Campath.

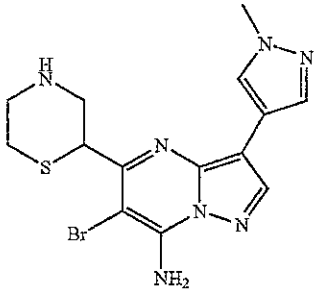
4. A pharmaceutical composition comprising (i) a therapeutically effective amount of a compound of claim 1 or a pharmaceutically acceptable salt, ester or prodrug thereof, and (ii) temozolomide.

5. A compound of the formula:



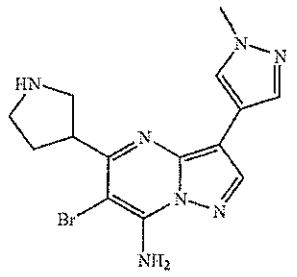
or a pharmaceutically acceptable salt, ester or prodrug thereof.

6. A compound of the formula:



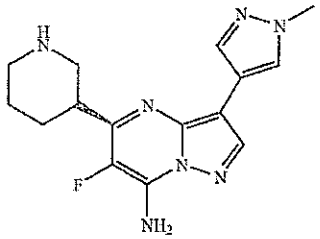
or a pharmaceutically acceptable salt, ester or prodrug thereof.

7. A compound of the formula:



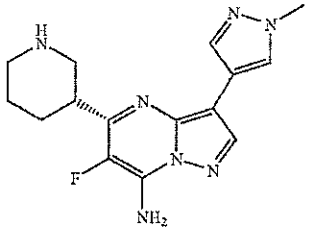
or a pharmaceutically acceptable salt, ester or prodrug thereof.

8. A compound of the formula:



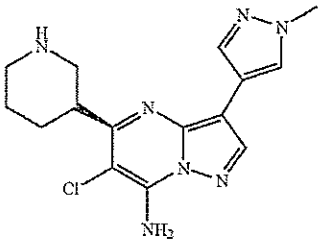
or a pharmaceutically acceptable salt, ester or prodrug thereof.

9. A compound of the formula:



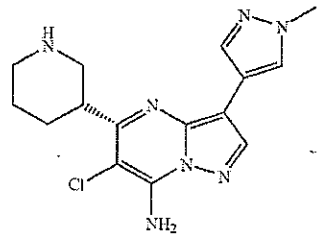
or a pharmaceutically acceptable salt, ester or prodrug thereof.

10. A compound of the formula:



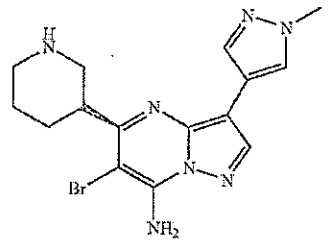
or a pharmaceutically acceptable salt, ester or prodrug thereof.

11. A compound of the formula:



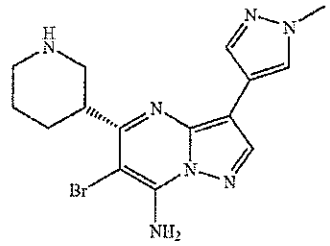
or a pharmaceutically acceptable salt, ester or prodrug thereof.

12. A compound of the formula:



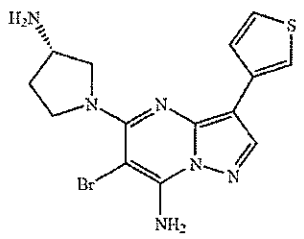
or a pharmaceutically acceptable salt, ester or prodrug thereof.

13. A compound of the formula:



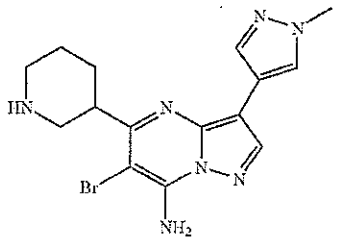
or a pharmaceutically acceptable salt, ester or prodrug thereof.

14. A compound of the formula:



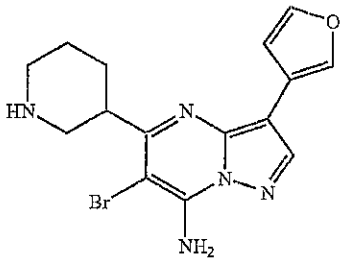
or a pharmaceutically acceptable salt, ester or prodrug thereof.

15. A compound of the formula:



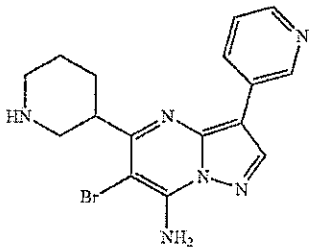
or a pharmaceutically acceptable salt, ester or prodrug thereof.

16. A compound of the formula:



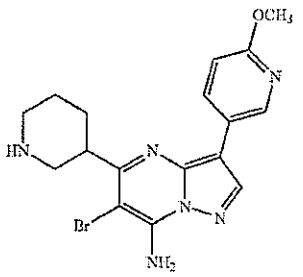
or a pharmaceutically acceptable salt, ester or prodrug thereof.

17. A compound of the formula:



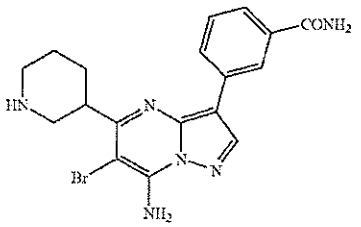
or a pharmaceutically acceptable salt, ester or prodrug thereof.

18. A compound of the formula:



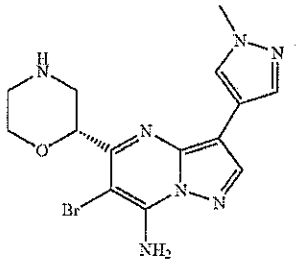
or a pharmaceutically acceptable salt, ester or prodrug thereof.

19. A compound of the formula:



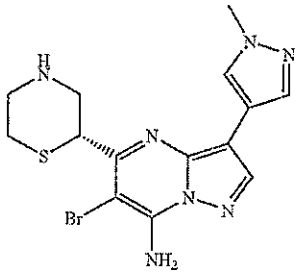
or a pharmaceutically acceptable salt, ester or prodrug thereof.

20. A compound of the formula:



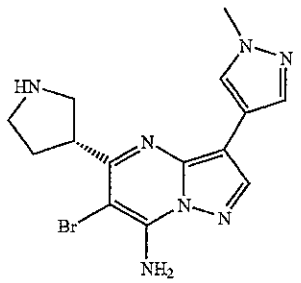
or a pharmaceutically acceptable salt, ester or prodrug thereof.

21. A compound of the formula:



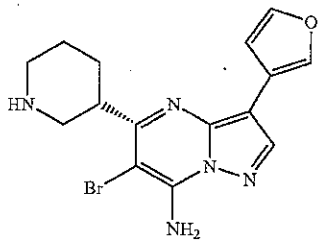
or a pharmaceutically acceptable salt, ester or prodrug thereof.

22. A compound of the formula:



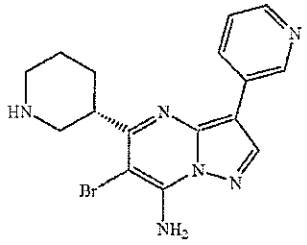
or a pharmaceutically acceptable salt, ester or prodrug thereof.

23. A compound of the formula:



or a pharmaceutically acceptable salt, ester or prodrug thereof.

24. A compound of the formula:



or a pharmaceutically acceptable salt, ester or prodrug thereof.

25. A composition comprising (i) a compound of any of claims **5-24** and (ii) temozolomide.

26. A pharmaceutical composition comprising a therapeutically effective amount of a compound of any of claims **5-24**, or a pharmaceutically acceptable salt, ester or prodrug thereof, in combination with at least one pharmaceutically acceptable carrier.