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DESCRIPTION

Pb-free Soldering Alloy

Technical Field

- 5 The present invention relates to Pb-free soldering alloy that does not include any amount of Pb.

Background Art

- 10 In the past, some Japanese patent publication laid-open such as Japanese patent publication laid-open 10-13015 and Japanese patent publication laid-open 11-277290 disclosed Pb-free soldering alloy comprising Ag, Ni and Sn and not including any amount of Pb.

- 15 The above-mentioned Japanese patent publication laid-open 10-13015 disclosed Pb-free soldering alloy comprising Ag, Ni and Sn and not including any amount of Pb but did not disclose their concrete contents.

- 20 On the other hand, the above-mentioned Japanese patent publication laid-open 11-277290 disclosed Pb-free soldering alloy comprising Ni of 0.01 to 0.5 wt.%, Ag of 0.5 to 3.39 wt.% and Sn of the balance. However, Ag of 0.5 to 3.39 wt.% was comparatively low amount and Sn was comparatively high amount to the contrary. Thus, there was a problem that much amount of Cu comprised in a printed-circuit board
25 to be jointed by soldering was reduced to be diffused in the soldering alloy whereby resulting in lowering the mechanical strength of the jointed part of the printed-circuit board.

- 30 Furthermore, in case the amount of Ag to be added was less than 3.5% relative to the amount of Sn, the excessive amount of Sn precipitated as β -Sn at the soldered part and fine cracks by the shrinkage cavity occurred on the upper layer of the soldered part during solidification.

- 35 Thus, the purpose of the present invention is to propose Pb-free soldering alloy which does not include any amount of Pb, which prevents much amount of Cu comprised in the printed-circuit board from being reduced to be diffused in



the soldering alloy and fine cracks from occurring at the soldered part by combining Cu comprised in the printed-circuit board with Ni comprised in the soldering alloy and which in addition keeps high mechanical strength at the soldered part.

Disclosure of Invention

Pb-free soldering alloy according to the invention defined in claim 1 is characterized in that it comprises Ag of 3.5 to 6.0 wt.%, Ni of 0.001 to 1.0 wt.% and Sn of the balance.

Pb-free soldering alloy according to the invention defined in claim 2 is that as defined in claim 1 wherein it includes further one or more of P, Ga and Ge of 0.001 to 1.0 wt.%.

According to the invention defined in claim 1, Cu comprised in the printed-circuit board is prevented from being diffused in the soldering alloy by adding Ni to Sn-Ag soldering alloy. Thus, in case of soldering a Cu fine line such as a lead, less amount of Cu is precipitated from the Cu fine line to the soldering alloy. As the result, the prevention of the reduction of Cu of the Cu fine line and the suppression of the reduction of its diameter can keep its mechanical strength itself.

In addition, as Ni is diffused in the soldering alloy, the mechanical strength of the soldering alloy itself can be kept advantageously.

Even when an electronic part and a printed-circuit board are soldered by conventional Sn-Pb soldering alloy, Cu is precipitated from a Cu foil of the printed-circuit board and is diffused in the soldering alloy.

However, in case soldering alloy including Pb can not be used by reason that Pb causes an environmental problem and Pb-free soldering alloy including Sn principally is used instead, more amount of Cu is precipitated than in Sn-Pb soldering alloy and diffused in the soldering alloy because the more amount of Sn is included in the soldering alloy, the more easily Cu is precipitated and diffused in the soldering

alloy. Thus, the Pb-rich soldering alloy has been used inevitably to prevent much amount of Cu from being precipitated and diffused in the soldering alloy.

From the above-mentioned reason, Sn-Ag-Cu soldering alloy and Sn-Ag-Cu-Bi soldering alloy reduce the mechanical strength of a lead to be soldered because the principal constituent of them is Sn and they cause the precipitation and the diffusion of more amount of Cu than the conventional type soldering alloy.

According to the present invention, as the amount of Ag included in the soldering is 3.0 to 6.0 wt.% which is more than in the conventional type, the amount of β -Sn in the soldering alloy is less. Owing to the multiplicative effects of the presence of fine Ag_3Sn and the diffusion of Ni, the mechanical strength of the soldered joint is raised.

In addition, We found that the addition of Ni prevented Cu of the Cu foil on the printed-circuit board from being precipitated and diffused in the soldering alloy.

According to the invention defined in claim 2, as any one or more of P, Ga and Ge is or are added to the soldering alloy by 0.001 to 1 wt.%, the oxidation of the soldering alloy during soldering, the generation of dross is suppressed and good soldering capability is given.

Best Mode for Carrying Out the Invention

We explain the present invention showing some Examples below.

As shown in table 1, in Example 1, the melting point of Pb-free soldering alloy comprising Ni of 0.01 wt.%, P of 0.05 wt.%, Ag of 3.8 wt.% and the rest Sn of 96.14 wt.% was 225°C.

In Example 2, that of Pb-free soldering alloy comprising Ni of 0.05 wt.%, Ga of 0.05 wt.%, Ag of 3.8 wt.% and the rest Sn of 96.1 wt.% was 225°C.

In Example 3, that of Pb-free soldering alloy comprising Ni of 0.1 wt.%, Ge of 0.05 wt.%, Ag of 4.0 wt.% and the rest Sn

of 95.85 wt.% was 227°C.

In Example 4, that of Pb-free soldering alloy comprising Ni of 0.3 wt.%, P of 0.07 wt.%, Ga of 0.03 wt.%, Ag of 4.0 wt.% and the rest Sn of 95.6 wt.% was 290°C.

5 In Example 5, that of Pb-free soldering alloy comprising Ni of 0.5 wt.%, Ga of 0.05 wt.%, Ge of 0.05 wt.%, Ag of 4.0% and the rest Sn of 95.4 wt.% was 356°C.

10 In Example 6, that of Pb-free soldering alloy comprising Ni of 1.0 wt.%, P of 0.03 wt.%, Ge of 0.07 wt.%, Ag of 4.5 wt.% and the rest Sn of 94.4 wt.% was 420°C.

In Comparative Example 1, that of Pb-free soldering alloy comprising Ni of 0.05 wt.%, Ag of 2.5 wt.% and the rest Sn of 97.45 wt.% was 224°C.

15 In Comparative Example 2, that of Pb-free soldering alloy comprising Ni of 0.1 wt.%, Ag of 3.3 wt.% and the rest Sn of 96.6 wt.% was 223°C.

In Compavative Example 3, that of Pb-free soldering alloy comprising Ag of 3.5 wt.% and the rest Sn of 96.5 wt.% was 221°C.

20 Using these types, we soldered Cu lines for 2 seconds at 350°C and 400°C respectively and examined the reduction of Cu amount at the soldered part of Cu lines. As the result, we found that by using Ni-added soldering alloy we could suppress the precipitation of Cu of Cu lines and its diffusion
25 in the soldering alloy.

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Table 1
Cu reduction test results

	Sn wt. %	Ag wt. %	Ni wt. %	P wt. %	Ga wt. %	Ge wt. %	Solid phase line °C	liquid phase line °C	Section reduction ratio %	
									350°C × 2 seconds	400°C × 2 seconds
Example 1	96.14	3.8	0.01	0.05			221	225	91.7	84.6
Example 2	96.10	3.8	0.05		0.05		221	225	92.9	84.9
Example 3	95.85	4.0	0.10			0.05	221	227	93.3	85.2
Example 4	95.60	4.0	0.30	0.07	0.03		221	290	93.8	86.1
Example 5	95.40	4.0	0.50		0.05	0.05	221	356	94.8	90.1
Example 6	94.40	4.5	1.00	0.03		0.07	221	420	96.7	91.9
Comparative Example 1	97.45	2.5	0.05				221	224	85.5	78.2
Comparative Example 2	96.60	3.3	0.10				221	223	87.6	79.6
Comparative Example 3	96.50	3.5	0				221	221	86.1	80.2
Cu line before soldering									100 supposed	

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Furthermore, in order to examine the mechanical strength at the soldered part, we inserted a square-sectioned pin in 1.2mm × 1.2mm made of brass coated with Sn into a hole of 1.8mm in diameter of a Cu-pattered printed-circuit board whose land circumference was 3.0mm and soldered them. In this case, the soldering alloy was 30mg in weight and we prepared the test piece using liquid flux and a soldering iron for soldering. The temperatures of the iron tip for the tests were 350°C and 400°C and the soldering time was 2 seconds in both the cases.

We pulled off the pin from the test piece at 10mm/min in tension speed using a tension tester and measured the tension load required for pulling it off. We measured it five times for the same test piece and calculated the mean value of the measured ones to show it in a table. In addition, in order to measure the weight of the generated dross, we jetted soldering alloy in a jet solder tank for 8 hours and took out the generated dross to measure its weight on a balance. Table 2 shows the total weight of the dross jetted for 8 hours and its average weight per hour.

Table 2

Results of tens ion tests and dross generation tests

	tension test ($\times 10^2\text{N}$)		weight of generated dross per 100 kg	
	350°C $\times 2$ seconds	400°C $\times 2$ seconds	total weight for 8 hrs	average weight per 1 hr
Example 1	3.3	3.1	2.43 kg	0.30 kg
Example 2	3.4	3.1	2.44 kg	0.31 kg
Example 3	3.7	3.2	2.41 kg	0.30 kg
Example 4	3.9	3.2	2.35 kg	0.29 kg
Example 5	4.0	3.3	2.22 kg	0.28 kg
Example 6	4.4	3.6	2.11 kg	0.27 kg
Comparative Example 1	3.0	2.8	4.35 kg	0.54 kg
Comparative Example 2	3.2	3.0	4.25 kg	0.53 kg
Comparative Example 3	3.2	2.9	4.18 kg	0.52 kg

CLAIMS

1. Pb-free soldering alloy characterized in that it comprises
Ag of 3.5 to 6.0 wt.%, Ni of 0.001 to 1.0 wt.% and Sn of the
5 balance.
2. Pb-free soldering alloy as claimed in claim 1 wherein one
or more of P, Ga and Ge is or are added to the soldering alloy
by 0.001 to 1 wt.%.

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