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5

TECHNICAL FIELD

10 The present invention relates to oral compositions which provide more efficient fluoride uptake by the enamel of human teeth.

BACKGROUND OF THE INVENTION

The role of fluoride in the remineralization of carious lesions is well known. The use of agents to enhance this benefit is also well known. One such agent is strontium.

15 While strontium's inclusion in oral compositions containing fluoride is desirable, the agents form an insoluble precipitate unless separated before use or the strontium ions are complexed so as not to be available to form the precipitate. Strontium EDTA is such a complex.

20 The effect of strontium-EDTA complex in combination with sodium recinoleate and a fluoride source is found in the Journal of Dental Research (1982) 61 (3) 451-455. The combined effect of strontium and fluoride in reducing the acid solubility of enamel is also disclosed in the Journal of Dental Research (1983) 62 (10)
25 1049-1053. A further reference discussing the effect of strontium and fluoride Featherstone, J.D.B., "Remineralization of Artificial Carious Lesions In-vivo and In-vitro", Proceedings of Workshop, (1983) IRL Press Ltd.

30 The use of strontium in combination with fluoride in oral compositions is also disclosed in a number of patent references. Included among these references are U.S. Patent 3,888,976, June 10, 1975 to Mikvys disclosing an effervescent mouthwash tablet containing strontium ions and possibly a fluoride ion source. U.S. Patent 4,367,219, January 4, 1983 to Schole
35 discloses dentifrices containing a combination of strontium EDTA, a recinoleate salt and a fluoride ion source. U.S. Patent

4,415,549, November 15, 1983 to Shah et al. discloses toothpastes containing a glycyrrhizinate salt, strontium EDTA and a fluoride ion source. Finally European Patent Application 0,079,611, June 6, 1983, Shah, discloses oral compositions containing a strontium EDTA complex and a fluoride ion source.

The use of other materials in oral compositions to provide a variety of benefits in combination with soluble fluoride ion sources has also been disclosed in the patent literature.

Included are certain polymers and other agents. Specific agents disclosed are polyelectrolytes such as copolymers of maleic anhydride and ethylene disclosed in U.S. Patent 3,429,963, February 25, 1969 to Shedlovsky. Shedlovsky also discloses polyacrylic acid having an average molecular weight of 1500 and greater. Other references disclosing polyacrylic acids in oral compositions are South African Patent 720898, September 12, 1972 which discloses such acids having a molecular weight of from 1000 to 1,000,000; and U.S. Patent 4,304,766, December 8, 1971 to Chang discloses polyacrylic acid having a molecular weight in the range of 2,000 to 4,000,000 for use as a membrane to prevent the elution from teeth of previously applied agents. Finally, U.S. Patent 3,956,480, to Gaffar, May 11, 1976 discloses complexes of anionic polymers (e.g., acrylic acid) and a cationic therapeutic agent (e.g., chlorhexidine) as anticalculus agents.

In spite of the many disclosures of strontium, fluoride ions and polyacrylic acids in the prior art, there is lacking a suggestion to combine these agents in a single oral composition. Strontium ions form a complex with the polyacrylic acid thereby not being able to form a precipitate with fluoride.

It is therefore an object of the present invention to provide oral compositions which combine a polyacrylic acid polymer or copolymer, strontium and fluoride.

It is a further object of the present invention to provide a more effective anti-carles composition.

These and other objects will become more clear from the detailed description which follows.

All percentages and ratios used herein are by weight unless otherwise specified.

SUMMARY OF THE INVENTION

The present invention embraces oral compositions comprising:

- (a) a safe and effective amount of a polymer selected from polyacrylic acid polymers, copolymers of acrylic acid and mixtures thereof;
- (b) a safe and effective amount of a soluble strontium ion source;
- (c) a safe and effective amount of a soluble fluoride ion source;
- and
- (d) an acceptable carrier.

DETAILED DESCRIPTION OF THE INVENTION

The compositions of the present invention comprise polyacrylic acid polymers or copolymers, a strontium ion source, a fluoride ion source and an acceptable carrier.

By "oral compositions" as used herein means a product which in the ordinary course of usage is not intentionally swallowed for purposes of systemic administration of particular therapeutic agents, but is rather retained in the oral cavity for a time sufficient to contact substantially all of the dental surfaces and/or oral tissues for purposes of oral activity.

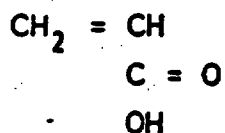
By "safe and effective amount" as used herein means sufficient amount of material to provide the desired benefit while being safe to the hard and soft tissues of the oral cavity.

By the term "comprising", as used herein, is meant that various additional components can be conjointly employed in the compositions of this invention as long as the listed materials perform their intended functions.

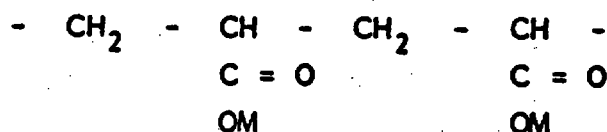
By the term "carrier", as used herein, is meant a suitable vehicle which is acceptable and can be used to apply the present compositions in the oral cavity.

Polyacrylic Acid

The polyacrylic acid polymers are staple items of commerce and are made by polymerizing acrylic acid,



to form the repeating chain



wherein M is an alkali metal ammonium or hydrogen ion. Polymers of the type useful in the present invention are available from Rohm and Haas Company.

Copolymers of acrylic acid and other monomers may also be used in the present invention. Suitable other monomers include methacrylic acid, 2-hydroxypropyl acrylate, 3-hydroxypropyl methacrylate, 3-hydroxy propyl acrylate, acrylamide and mixtures thereof. It is preferred that with methacrylic acid, the number of acrylic acid units in the polymer be at least 50% of the total units present. With other monomers it is preferred that the percentage be at least 80%. Mixtures of other monomers may also be used.

The molecular weight of the polymer may be in the range of 1000 to 1,200,000 (mass average), preferably the molecular weight is from 4000 to 10,000, most preferably from 4300 to 5200. A particularly preferred material is a polyacrylic acid polymer having a molecular weight of about 4500 which can be provided by Rohm and Haas carrying the identification LMW-45.

The amount of the polymer used in the present compositions is generally from 0.0003% to 13%, preferably from 0.03% to 5.0%, most preferably from 0.03% to 4%. For a dentifrice composition containing about 220 ppm Sr^{++} the preferred level of polymer is from 0.3% to 0.5%. For a mouthwash composition containing about 225 ppm Sr^{++} the preferred polymer level is from 0.03% to

0.10%. Mixtures of polyacrylic acid polymers or copolymers are also useful in the present invention.

Strontium Ion Source

5 The strontium ions of the present compositions can be provided by any of a wide variety of strontium salts or complexes. Included are strontium chloride, strontium acetate, strontium bromide, strontium glyconate, strontium lactate, strontium hydroxide and strontium salicylate. In another execution
10 strontium can be provided to the present compositions as a complex with the polyacrylic acid. The strontium ion source is preferably soluble enough in the composition at 25°C or when used to provide from 2 to 10,000, more preferably from 25 to 4400 ppm Sr^{++} .

Fluoride Ion Source

15 The water-soluble fluorine compound is preferably present in the compositions of this invention in an amount sufficient to give a fluoride concentration of from 0.0025% to 5.0% by weight, more preferably from 0.005% to 2.0% by weight at 25°C, in the composition or
20 when used to provide anticaries effectiveness. Preferred fluorides are sodium fluoride, stannous fluoride, indium fluoride, acidulated phosphate fluoride and sodium monofluorophosphate. U.S. Patent 2,946,735, July 26, 1960 to Norris et al. and U.S. Patent
25 3,678,154, July 18, 1972 to Widder et al. disclose such salts as well as others.

Acceptable Carrier

The carrier for the components of the present invention can be any vehicle suitable for use in the
30 oral cavity. Such carriers include the usual components of mouthwashes, toothpastes, topical dental gels, tooth-powders, prophylaxis pastes, lozenges, gums and the like and are more fully described hereinafter. Dentifrices and mouthwashes are the preferred systems with
35 toothpastes being the more preferred. Toothpastes contain as a major component an abrasive. The abrasive polishing material contemplated for use in the present invention can be any material which does not excessively abrade

dentin. These include, for example, silicas including gels and precipitates, calcium carbonate, dicalcium orthophosphate dihydrate, calcium pyrophosphate, tricalcium phosphate, calcium polymetaphosphate, insoluble sodium polymetaphosphate, hydrated alumina, and resinous abrasive materials such as particulate condensation products of urea and formaldehyde, and others such as disclosed by Cooley et al. in U.S. Patent 3,070,510, December 25, 1962. Mixtures of abrasives may also be used.

10 Silica dental abrasives, of various types, can provide the unique benefits of exceptional dental cleaning and polishing performance without unduly abrading tooth enamel or dentin. Silica abrasive materials are also exceptionally compatible with sources of soluble fluoride. For these reasons they are preferred
15 for use herein. Of course the abrasive selected should also exhibit excellent compatibility with soluble strontium ion sources.

The silica abrasive polishing materials useful herein, as well as the other abrasives, generally have an average particle size ranging between about 0.1 and 30 microns, preferably 5 and 15
20 microns. The silica abrasive can be precipitated silica or silica gels such as the silica xerogels described in Pader et al., U.S. Patent 3,538,230, issued March 2, 1970 and DiGiulio, U.S. Patent 3,862,307, June 21, 1975,

Preferred are the silica xerogels marketed under the tradename
25 "Syloid" by the W.R. Grace & Company, Davison Chemical Division. Preferred precipitated silica materials include those marketed by the J.M. Huber Corporation under the tradename, "Zeodent", particularly the silica carrying the designation "Zeodent 119". These silica abrasive are described in U.S.
30 Patent 4,340,583, July 29, 1982.

The abrasive in the dentifrice compositions described herein is present at a level of from about 6% to about 70%, preferably from about 15% to about 25% when the dentifrice is a toothpaste.
35 Higher levels, as high as 90%, may be used if the composition is a toothpowder.

Flavoring agents can also be added to the dentifrice and other compositions of the present invention. Suitable flavoring agents include oil of wintergreen, oil of peppermint, oil of spearmint, oil of sassafras, and oil of clove. Sweetening agents are also useful and include aspartame, acesulfame, saccharin, dextrose, levulose and sodium cyclamate. Flavoring and sweetening agents are generally used in the compositions herein at levels of from about 0.005% to about 2% by weight.

The compositions of this invention also contain emulsifying agents. Suitable emulsifying agents are those which are reasonably stable and foam throughout a wide pH range, including non-soap anionic, nonionic, cationic, zwitterionic and amphoteric organic synthetic detergents. Many of these suitable surfactants are disclosed by Gieske et al. in U.S. Patent 4,051,234, September 27, 1977.

Water is also present in the compositions of this invention. Water employed in the preparation of commercially suitable compositions should preferably be deionized and free of organic impurities. Water generally comprises from about 10% to 50%, preferably from about 20% to 40%, by weight of the toothpaste compositions herein. These amounts of water include the free water which is added plus that which is introduced with other materials such as with sorbitol.

In preparing toothpastes, it is necessary to add some thickening material to provide a desirable consistency. Preferred thickening agents are carboxyvinyl polymers, carrageenan, hydroxyethyl cellulose and water soluble salts of cellulose ethers such as sodium carboxymethyl cellulose and sodium carboxymethyl hydroxyethyl cellulose. Natural gums such as gum karaya, gum arabic, and gum tragacanth and polysaccharide gums such as xanthan gum can also be used. Colloidal magnesium aluminum silicate or finely divided silica can be used as part of the thickening agent to further improve texture. Thickening agents in an amount from 0.5% to 5.0% by weight of the total composition may be used.

It is also desirable to include a humectant in a toothpaste to keep it from hardening. Suitable humectants include glycerin, sorbitol, and other edible polyhydric alcohols at a level of from about 10% to about 70%.

5 Another preferred embodiment of the present invention is a mouthwash composition. Conventional mouthwash composition components can comprise the carrier for the actives of the present invention. Mouthwashes generally comprise from about 20:1 to about 2:1 of a water/ethyl alcohol solution and preferably
10 other ingredients such as flavor, sweeteners, humectants and sudsing agents such as those described above. The humectants, such as glycerin and sorbitol give a moist feel to the mouth. Generally, on a weight basis the mouthwashes of the invention comprise 5% to 60% (preferably 10% to 25%) ethyl alcohol, 0% to 20%
15 (preferably 5% to 20%) of a humectant, 0% to 2% (preferably 0.01% to 0.15%) emulsifying agent, 0% to 0.5% (preferably 0.005% to 0.06%) sweetening agent such as saccharin, 0% to 0.3% (preferably from 0.03% to 0.3%) flavoring agent, and the balance water.

Suitable lozenge and chewing gum components are disclosed
20 in U.S. Patent 4,083,955, April 11, 1978 to Grabenstetter et al.,

Suitable topical dental gels generally comprise a base of a humectant such as glycerine thickened with a suitable agent. Such gels generally do not contain an abrasive.

25 The pH of the present compositions and/or its pH in the mouth can be any pH which is safe for the mouth's hard and soft tissues. Such pH's are generally from about 3 to about 10, preferably from about 4 to about 8. The present compositions, since they are designed to deliver fluoride to tooth enamel;
30 should not contain materials which would cause significant loss of strontium and/or fluoride ions.

METHOD OF MANUFACTURE

The carrier compositions of the present invention can be made using methods which are common in the oral products area.
35 However the strontium ion source and the polyacrylic acid polymer should be added prior to the addition of the fluoride ion source.

This is to ensure that strontium and fluoride do not form an insoluble precipitate. A specific method of manufacture is set forth in the Examples.

COMPOSITION USE

5 The present invention in its method aspect involves applying to the oral cavity safe and effective amounts of the compositions described herein. These amounts (e.g. from about 0.3 to about 15g), if it is a toothpaste or mouthwash, are kept in the mouth for from about 15 to about 30 seconds.

10 The following examples further describe and demonstrate preferred embodiments within the scope of the present invention. The examples are given solely for illustration and are not to be construed as limitations of this invention as many variations thereof are possible without departing from the scope thereof.

EXAMPLE I

The composition given below is exemplary of the present invention.

	<u>Component</u>	<u>Weight %</u>
20	Sodium Fluoride	0.243
	Strontium Chloride 6H ₂ O	0.667
	Silica Dental Abrasive ¹	20.000
	Polyacrylic Acid ²	0.280
	Sodium Saccharin	0.130
25	Titanium Dioxide	0.700
	Flavor	0.600
	Carboxymethylcellulose	0.600
	Surfactant-Tween 80 ³	0.900
	Polyethyleneglycol MW 600	5.000
30	Triethanolamine	1.000
	Dye	0.050
	Water	19.000
	Sorbitol (70% Aqueous Solution)	q.s. 100%

35 ¹ Precipitated silica abrasive offered by J.M. Huber Company as Zeodent-119

² Polyacrylic acid polymer having mass average molecular weight of about 4500 offered by Rohm and Haas

³ POE (20) sorbitan monooleate offered by ICI Americas Inc.

The above composition is prepared by putting 50% of the sorbitol into a main mix tank, adding all of the polyacrylic acid and 50% of the water and mixing for a few minutes. Strontium chloride is then dissolved in 10% of the water and added to the main mix tank with the desired pH being obtained by immediately adding HCl or NaOH. Next sodium fluoride is dissolved in the remaining water and added to the main mix tank, followed by the silica abrasive, sodium saccharin, titanium dioxide and flavor. In a separate tank, the binder is mixed with the remaining sorbitol and added to the main mix tank followed by the surfactant and the dye. The final mixture is heated to 70°C, processed through a mill and deaerated if necessary.

EXAMPLE II

Given below is another example of the present invention.

	<u>Component</u>	<u>Weight %</u>
	Sodium Fluoride	0.111
20	Strontium Chloride 6H ₂ O	0.333
	Silica Dental Abrasive ¹	20.000
	Polyacrylic Acid ²	0.500
	Sodium Saccharin	0.130
	Titanium Dioxide	0.100
25	Flavor	0.600
	Carboxymethylcellulose	0.600
	Sodium Alkyl Sulfate	2.500
	Dye	0.050
	Water	19.000
30	Sorbitol (70% Aqueous Solution)	q.s. 100%

¹ Precipitated silica abrasive offered by J.M. Huber Company as Zeodent-119

² Polyacrylic acid polymer having mass average molecular weight of about 4500 offered by Rohm and Haas

EXAMPLE III

Given below is yet another example of the present invention.

	<u>Component</u>	<u>Weight %</u>
	Sodium Fluoride	0.243
5	Strontium Chloride $6H_2O$	0.100
	Silica Dental Abrasive ¹	20.000
	Polyacrylic Acid ²	2.000
	Sodium Saccharin	0.130
	Titanium Dioxide	0.700
10	Flavor	0.600
	Carboxymethylcellulose	0.600
	Sodium Lauroyl Sarcosinate	1.500
	Dye	0.050
	Water	19.000
15	Sorbitol (70% Aqueous Solution)	q.s. 100%

¹ Precipitated silica abrasive offered by J.M. Huber Company as Zeodent-119

² Polyacrylic acid polymer having mass average molecular weight of about 4500 offered by Rohm and Haas

EXAMPLE IV

Given below is another composition representative of the present invention.

	<u>Component</u>	<u>Weight %</u>
25	Sodium Fluoride	0.443
	Strontium Chloride $6H_2O$	1.200
	Silica Dental Abrasive ¹	20.000
	Polyacrylic Acid ²	3.000
	Sodium Saccharin	0.130
30	Titanium Dioxide	0.100
	Flavor	0.600
	Carboxymethylcellulose	0.600
	Cocoyl Isethionate	2.000
	Dye	0.050
35	Water	19.000
	Sorbitol (70% Aqueous Solution)	q.s. 100%

¹ Precipitated silica abrasive offered by J.M. Huber Company as Zeodent-119

² Polyacrylic acid polymer having mass average molecular weight of about 4500 offered by Rohm and Haas

5 All of the above compositions provide improved fluoride uptake by dental enamel. Similar performance is obtained if the polyacrylic acid is replaced by another acrylic acid polymer or copolymer having a molecular weight of from about 1000 to about 1,200,000. Similarly strontium chloride may be replaced with
10 another strontium salt such as strontium acetate, strontium lactate, strontium salicylate, strontium hydroxide or strontium glyconate.

Fluoride Uptake Analysis

The ability of compositions of the present invention to
15 deliver enhanced fluoride to dental enamel was measured using an in-vitro enamel disc method.

Enamel discs (4mm diameter) were mounted in lucite rods, then ground and polished--removing at least 40um of the enamel surface. Discs were decalcified for 46 hours at 37°C in 8ml of
20 0.1M lactic acid plus 1.0×10^{-4} M disodium-methanehydroxydiphosphonate (MHDP) adjusted to pH 4.5. The depth of demineralization was approximately 100um. The discs were placed in groups of four (4) per treatment group.

At the beginning of the study, each treatment group was
25 suspended for one (1) hour in 15 grams of fresh, paraffin stimulated, pooled human saliva under constant agitation. The treatment groups were then exposed to 20ml of an appropriate test solution for one (1) minute, followed by a thorough rinsing in deionized water. The one minute treatment was followed by one
30 (1) hour soaking in the saliva baths. This cyclic pattern (one minute treatment/thorough rinse/one hour saliva soak) was repeated seven (7) times a day for four (4) days for a total of twenty-eight (28) treatments. Saliva baths were changed twice daily to maintain their freshness. At the end of each day, the
35 discs were thoroughly rinsed with deionized water and placed

upright in a moist atmosphere under refrigeration in order to minimize any microbial growth.

Following the 28th treatment, specimens were thoroughly rinsed with deionized water and analyzed for fluoride content using a microdrill biopsy technique. In this technique, a carbide dental bar (diameter approximately 0.45mm) penetrates the surface of the discs and travels to the base of the demineralized area, in this case 100 μ m. The displaced enamel powder is recovered into a small polyethylene vial, where it is dissolved with 20ul of 0.5M HClO₄. To this is added 40ul of deionized water, then 40ul of a Citrate-EDTA buffer resulting in a total volume of 100ul for analysis. Fluoride analysis of this solution is done using an Orion Fluoride Ion-specific electrode (Model 96-09-00) that has been appropriately calibrated for the range of these analytical samples.

Statistical analyses were done using a standard t- test for significance.

Using the above described procedure, various strontium ion concentrations and polyacrylic acid polymer levels were used with 1100ppm F⁻ on to determine the amount of F⁻ taken up by the enamel.

Treatment Solutions*

		% Polyacrylic Acid		
	F ⁻ (ppm)	SR ⁺⁺ (ppm)	MW 4500	F ⁻ Uptake (μ g/cm ²)
25	1100	4400	6.0	20.17 $\pm$ 1.31
	1100	2200	3.0	20.07 $\pm$ 0.55
	1100	1100	1.5	18.29 $\pm$ 1.23
	1100	500	0.680	17.93 $\pm$ 0.52
30	1100	50	.068	16.70 $\pm$ 1.86
	1100	100	.136	15.84 $\pm$ 0.77
	1100	10	.014	13.96 $\pm$ 0.48
	1100	--	--	10.92 $\pm$ 1.27

*Actual levels in test solutions were at $\frac{1}{2}$ these levels. These numbers represent the equivalents in whole dentifrice before dilution.

CLAIMS

- 14 -

1. An oral composition comprising:
 - (a) a safe and effective amount of a polymer selected from polyacrylic acid polymers, copolymers of acrylic acid and mixtures thereof;
 - 5 (b) a safe and effective amount of a soluble strontium ion source;
 - (c) a safe and effective amount of a soluble fluoride ion source; and
 - (d) an acceptable carrier.
2. An oral composition according to Claim 1 wherein (a) is a copolymer of acrylic acid and a monomer selected from the group consisting of methacrylic acid, 2-hydroxypropyl acrylate, 3-hydroxypropyl acrylate, 3-hydroxypropyl methacrylate, acrylamide and mixtures thereof.
- 5 3. An oral composition according to Claim 1 wherein (a) is a polyacrylic acid polymer and is present at a level of from 0.0003% to 13% by weight.
4. An oral composition according to Claim 3 wherein the acceptable carrier is a toothpaste.
5. An oral composition according to Claim 4 wherein the soluble strontium ion source is present at a level of from 2 to 10,000 ppm Sr^{++} in the composition and/or when it is used.
6. An oral composition according to Claim 5 wherein the polyacrylic acid polymer has a molecular weight of from 1000 to 1,200,000.
7. An oral composition according to Claim 6 which also contains a silica dental abrasive.
8. An oral composition according to Claim 7 wherein the soluble fluoride ion source is sodium fluoride present at a level sufficient to give from 25 to 5000 ppm F^- in the composition and/or when it is used.
9. An oral composition according to Claim 1 wherein the acceptable carrier is a mouthwash.
10. An oral composition according to Claim 1 wherein the acceptable carrier is a topical dental gel.



11. An oral composition according to Claim 9 wherein (a) is a polyacrylic acid polymer and is present at a level of from 0.03% to 0.10% by weight..
 12. An oral composition according to Claim 11 wherein the soluble strontium ion source is present at a level from 2 to 10,000 ppm in the composition and/or when it is used.
 13. An oral composition according to Claim 12 wherein the soluble fluoride ion source is sodium fluoride present at a level sufficient to give from 25 to 5000 ppm F^- in the composition and/or when it is used.
 14. An oral composition according to Claim 1 wherein the acceptable carrier is a lozenge or chewing gum.
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