

Polycarbonate with high extensional viscosity

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5 The present invention concerns polycarbonate with high extensional viscosity and containers containing this polycarbonate. The present invention also concerns the production and the use of these containers.

10 Containers containing polycarbonate are known in principle. These containers are produced from compositions (also known as compounds) containing polycarbonate and conventional accessory agents, for example. These compositions consisting of the polymer (polycarbonate) and accessory agents are also known as plastics. Examples of the accessory agents, or additives, include stabilisers, processing aids and others. The containers containing polycarbonate can also include other components, such as rubber seals or handles made from metal or other materials, for
15 example.

Containers containing polycarbonate have numerous advantageous properties, such as e.g. high transparency, good mechanical properties, high resistance to environmental influences and long service life, together with low weight and simple,
20 inexpensive manufacturability.

Containers containing polycarbonate can be produced by the extrusion blow moulding process or by the injection blow moulding process, for example.

25 In the extrusion blow moulding process the polycarbonate is generally melted with a single-screw extruder and moulded through a die to form a free-standing parison. The parison usually hangs down from the die. A blowing mould is then placed around the parison, squeezing together the lower end of the parison. Inside the mould the parison is then blown up so that the parison attains the desired shape.
30 After a cooling period the mould is opened and the container (hollow article) can be removed.



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The extrusion blow moulding process is disclosed for example in Brinkschröder, F. J.: "Polycarbonate" in Becker, Braun, Kunststoff-Handbuch, Volume 3/1, Polycarbonate, Polyacetale, Polyester, Celluloseester, Carl Hanser Verlag Munich, Vienna 1992, pages 257 to 264).

The injection blow moulding process is a combination of injection moulding and blow moulding.

10 Injection blow moulding proceeds in three stages:

1. Injection moulding of the parison in the plastic temperature range of the polycarbonate
2. Blowing of the parison in the thermoplastic range of the polycarbonate (the core of the injection mould also acts as the blowing mandrel)
3. Stripping of the hollow article and optional cooling of the blowing mandrel with air

20 Injection blow moulding is disclosed for example in Anders, S., Kaminski, A., Kappenstein, R., "Polycarbonate" in Becker, Braun, Kunststoff-Handbuch, Volume 3/1, Polycarbonate, Polyacetale, Polyester, Celluloseester, Carl Hanser Verlag Munich, Vienna 1992, pages 223 to 225.

25 The disadvantage of the containers containing polycarbonate known from the prior art is that they do not meet certain requirements that are important for the use of the containers in practice.

30 Thus for example the known polycarbonate containers may burst under severe mechanical loading. This can occur for example if a container filled with liquid is dropped to the ground from a great height, for example from the loading platform of a lorry on which the container is being transported.

The reason for this mechanical failure is commonly an uneven wall thickness in the container.

5 The uneven wall thickness of containers known from the prior art arises during their production, since the polycarbonate melt gives rise to uneven wall thicknesses during processing by the extrusion blow moulding process or by the injection blow moulding process.

10 The mechanical strength of containers having an uneven wall thickness can naturally be increased by using much more polycarbonate per container, such that the cross-section of the wall becomes much thicker. This has the disadvantage of increasing the material consumption, however, giving rise amongst other things to elevated costs.

15 The object of the present invention is therefore to provide a polycarbonate that allows the production of containers having as homogeneous a wall thickness as possible.

20 This object is achieved by a polycarbonate in which in the uniaxial elongation test performed at a temperature of 200°C with rising Hencky strain ϵ the extensional viscosity η_E rises more sharply than the value of the shear viscosity multiplied by three, 3η .

25 This polycarbonate is provided by the present invention.

The uniaxial elongation test and its performance are known to the person skilled in the art. The uniaxial elongation test can be performed with devices based on the Münstedt model. These are described in H. Münstedt, J. Rheol., Volume 23, page
30 421 (1979). These are also described in current textbooks such as e.g. in Ch. W. Macosko: Rheology, Verlag Chemie, 1994, in particular pages 288 to 297, and in M.

Pahl, W. Gleißle, H.-M. Laun: Praktische Rheologie der Kunststoffe und Elastomere, VDI-Verlag, 1995, in particular pages 349 to 357.

5 The methods for determining the shear viscosity as a function of time and for determining the shear viscosity as a function of time are known to the person skilled in the art.

10 Shear viscosity as a function of time is preferably determined in a rotational rheometer at low shear rates. Shear viscosity can also be determined in a rotational rheometer under oscillating deformation and converted to a time-dependent viscosity by means of conventional methods. The design and mode of use of rotational rheometers are described in current textbooks. For example in M. Pahl, W. Gleißle, H.-M. Laun: Praktische Rheologie der Kunststoffe und Elastomere, VDI-Verlag, 1995.

15 Extensional viscosity as a function of time is preferably determined using an extensional rheometer according to Münstedt. The uniaxial elongation test can also be performed with a series of other rheometers, for example with the commercially available extensional rheometer according to Meissner. This is described in J. Meissner, Rheologica Acta 8, Volume 78 (1969) and in J. S. Schulze et al., Rheol. Acta, Volume 40 (2001) pages 457-466.

20 The Hencky strain ϵ is a dimensionless quantity. The unit for extensional viscosity η_E is the Pascal multiplied by seconds. The unit for shear viscosity η is likewise the Pascal multiplied by seconds.

25 The quotient S is used as a measure for the relative increase in extensional viscosity η_E . The quotient S is dimensionless. S is the quotient of the extensional viscosity η_E and the shear viscosity multiplied by three, 3η . S depends on the measurement temperature T , the Hencky strain rate $\dot{\epsilon}$ (unit: 1 divided by seconds) and the Hencky strain ϵ and time.

The following formula applies:

$$S = \eta_E(t, \dot{\epsilon}) \text{ divided by } 3\eta(t)$$

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The total strain ϵ (unit: dimensionless) is linked to the initial specimen length L_0 (unit: metres) and the current specimen length L (unit: metres) and to the strain rate $\dot{\epsilon}$ and the time t (unit: seconds) by means of:

10 $\epsilon = \text{natural logarithm of } (L \text{ divided by } L_0) = \dot{\epsilon} \text{ multiplied by } t$

A polycarbonate is preferred for which at a temperature of 200°C, with a Hencky strain ϵ of 2.0 and a strain rate range $\dot{\epsilon}$ of between 0.1 and 0.01, the ratio S is greater than 1.1 and with a Hencky strain ϵ of 2.5 and a strain rate range $\dot{\epsilon}$ of between 0.1 and 0.01, the ratio S is greater than 1.1, wherein S is defined as $S = \eta_E$ divided by 3η .

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Particularly preferred is a polycarbonate for which at a temperature of 200°C, with a Hencky strain ϵ of 2.0 and a strain rate range $\dot{\epsilon}$ of between 0.1 and 0.01, the ratio S is greater than 1.3 and with a Hencky strain ϵ of 2.5 and a strain rate range $\dot{\epsilon}$ of between 0.1 and 0.01, the ratio S is greater than 1.5.

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The present invention also provides a container containing the polycarbonate according to the invention. This means a container which contains the polycarbonate as a wall material, for example. It does not mean a container made from completely different materials, which simply contains the polycarbonate as its contents.

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The present invention also provides a process for producing this container by extrusion blow moulding or by injection blow moulding.

In order to obtain polycarbonates having the extensional rheological properties according to the invention, the person skilled in the art can selectively adjust various parameters of the polycarbonates. He can for example influence the molecular weight and degree of crosslinking. The choice of monomers and comonomers in the case of copolycarbonates or of terminal groups also has an influence on the extensional rheological properties of the polycarbonates. The person skilled in the art can also use suitable additives to obtain the desired extensional rheological properties according to the invention. The present invention is therefore substantiated in that the person skilled in the art prepares the polycarbonates according to the invention having the extensional rheological properties according to the invention and uses them for the purpose of producing the containers according to the invention. These containers have the surprisingly good properties described in the present specification.

The present invention also provides the production of the containers according to the invention.

The present invention also provides the use of the containers according to the invention.

The advantage of the polycarbonate according to the invention lies in the fact that it makes it possible to produce the containers according to the invention with their advantageous properties.

The containers according to the invention have the advantage that with the specified amount of polycarbonate per container they have high mechanical strength.

They also have the advantage that they can be produced with a homogeneous wall thickness distribution.

The containers according to the invention have numerous other advantages. They are resistant to mechanical loading, i.e. fracture resistant, and they also have an advantageous spectrum of other mechanical properties. They have good optical properties, in particular they have high transparency. They have a high heat resistance. By virtue of their high heat resistance the containers according to the invention can be cleaned with hot water or sterilised with hot steam. They have a high resistance to the conventional cleaning agents that are used for example for cleaning water bottles for multiple use, one field of application for the containers according to the invention. They can be produced easily and inexpensively by known processes. The good processing properties of the polycarbonate are advantageously expressed here. They have low material ageing in use and hence a long service life. For a multiple usage that may optionally occur this means numerous cycles of use.

Containers within the meaning of the present invention can be used for the packaging, storage or transport of liquids, solids or gases. Containers for the packaging, storage or transport of liquids (liquid containers) are preferred, containers for the packaging, storage or transport of water (water bottles) are particularly preferred.

Containers within the meaning of the invention are preferably hollow articles having a volume of 0.1 l to 50 l, preferably 0.5 l to 50 l, whereby volumes of 1 l, 5 l, 12 l and 20 l are most particularly preferred.

3 and 5 gallon water bottles are most particularly preferred.

The containers preferably have an empty weight of preferably 0.1 g to 3000 g, by preference 50 g to 2000 g and particularly preferably 650 g to 900 g.

The wall thicknesses of the containers are preferably 0.5 mm to 5 mm, by preference 0.8 mm to 4 mm.

Containers within the meaning of the present invention preferably have a length of preferably 5 mm to 2000 mm, particularly preferably 100 mm to 1000 mm.

5 The containers preferably have a maximum circumference of preferably 10 mm to 250 mm, by preference 50 mm to 150 mm and most particularly preferably 70 to 90 mm.

10 Containers within the meaning of the invention preferably have a bottle neck of a length of preferably 1 mm to 500 mm, by preference 10 mm to 250 mm, particularly preferably 50 mm to 100 mm and most particularly preferably 70 to 80 mm.

The wall thickness of the bottle neck of the containers preferably varies between 0.5 and 10 mm, particularly preferably from 1 mm to 10 mm and most particularly preferably from 5 mm to 7 mm.

15 The diameter of the bottle neck preferably varies between 5 mm and 200 mm. 10 mm to 100 mm are particularly preferred and 45 mm to 75 mm are most particularly preferred.

20 The bottle base of the containers according to the invention has a diameter of preferably 10 mm to 250 mm, by preference 50 mm to 150 mm and most particularly preferably 70 to 90 mm.

25 Containers within the meaning of the present invention can display any geometrical shape whatsoever, they can for example be round, oval or polygonal or angular with for example 3 to 12 sides. Round, oval and hexagonal shapes are preferred.

The design of the containers can be based on any surface texture whatsoever. The surface textures are preferably smooth or ribbed. The containers according to the invention can also have several different surface textures. Ribs or beads can run
30 around the circumference of the containers. They can be any distance apart or several

mutually different distances apart. The surface textures of the containers according to the invention can have etched or integrated textures, symbols, ornaments, coats of arms, company logos, trademarks, signatures, manufacturer information, material identifications and or volume information.

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The containers according to the invention can have any number of handles, which may be located at the side, on top or underneath. The handles may protrude outwards and or may be integrated into the shape of the container. The handles can be folding or fixed. The handles can be of any shape whatsoever, e.g. oval, round or polygonal.

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The handles preferably have a length of 0.1 mm to 180 mm, by preference 20 mm to 120 mm.

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In addition to the polycarbonate according to the invention the containers according to the invention can also contain other substances to a small degree, e.g. rubber seals or handles made from other materials.

The containers according to the invention are preferably produced by the extrusion blow moulding process or by the injection blow moulding process.

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In a preferred embodiment of the to produce the containers according to the invention, the polycarbonates according to the invention are processed on extruders having a smooth or grooved, preferably a smooth feed zone.

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The drive power of the extruder is selected according to the diameter of the screw. By way of example, the drive power of the extruder is approx. 30 to 40 kW with a screw diameter of 60 mm and approx. 60 to 70 kW with a screw diameter of 90 mm.

The general purpose three-section screws conventionally used in the processing of engineering thermoplastics are suitable.

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A screw diameter of 50 to 60 mm is preferred for the production of containers having a volume of 1 l. A screw diameter of 70 to 100 mm is preferred for the production of containers having a volume of 20 l. The length of the screws is preferably 20 to 25 times the diameter of the screw.

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In the case of the blow moulding process, the blow mould is preferably heated to 50 to 90°C in order to obtain a brilliant and high-quality surface to the containers.

10 In order to ensure that the blow mould is heated evenly and efficiently, the base section and the jacket section can be heated separately.

The blow mould is preferably closed with a pinch-off force of 1000 to 1500 N per cm of pinch-off weld length.

15 Before being processed, the polycarbonate according to the invention is preferably dried so that the optical quality of the containers is not impaired by streaks or bubbles and the polycarbonate is not degraded hydrolytically during processing. The residual moisture content after drying is preferably below 0.01 wt.%. A drying temperature of 120°C is preferred. Lower temperatures do not guarantee adequate
20 drying, whilst at higher temperatures there is the risk of the pellets of polycarbonate sticking together such that they can no longer be processed. Dry air dryers are preferred.

25 The preferred melt temperature for processing of the polycarbonate according to the invention is 230°C to 300°C.

The containers according to the invention can be used for the packaging, storage or transport of liquids, solids or gases. The embodiment as containers that are used for example for the packaging, storage or transport of liquids is preferred. The
30 embodiment as a water bottle that can be used for example for the packaging, storage or transport of water is particularly preferred.

Polycarbonates within the meaning of the present invention are preferably melt processable aromatic polycarbonates. They can be both homopolycarbonates and copolycarbonates.

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Particularly preferred polycarbonates are the homopolycarbonate based on bisphenol A, the homopolycarbonate based on 1,1-bis(4-hydroxyphenyl)-3,3,5-trimethylcyclohexane and the copolycarbonates based on the two monomers bisphenol A and 1,1-bis(4-hydroxyphenyl)-3,3,5-trimethylcyclohexane.

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Polycarbonates in which up to 80 mol%, in particular from 20 mol% up to 50 mol%, of the carbonate groups are replaced by aromatic dicarboxylic acid ester groups also belong to the polycarbonates according to the invention. Such polycarbonates that contain both acid radicals of carbonic acid and acid radicals of aromatic dicarboxylic acids incorporated into the molecule chain are also classed as aromatic polyester carbonates.

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The polycarbonates according to the invention can be produced by known means from diphenols, carbonic acid derivatives, optionally chain terminators and optionally branching agents. In order to produce the polyester carbonates a part of the carbonic acid derivatives is replaced by aromatic dicarboxylic acids or derivatives of dicarboxylic acids. This is done according to the carbonate structural units in the aromatic polycarbonates that are to be replaced by aromatic dicarboxylic acid ester structural units.

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Details of the production of polycarbonates are known. Reference is made by way of example to:

1. Schnell, "Chemistry and Physics of Polycarbonates", Polymer Reviews, Volume 9, Interscience Publishers, New York, London, Sydney 1964;
- 30 2. D.C. Prevorsek, B.T. Debona and Y. Kesten, Corporate Research Center, Allied Chemical Corporation, Morristown, New Jersey 07960: "Synthesis of

Poly(ester Carbonate) Copolymers" in Journal of Polymer Science, Polymer Chemistry Edition, Vol. 19, 75-90 (1980);

3. D. Freitag, U. Grigo, P.R. Müller, N. Nouvertne', BAYER AG, "Polycarbonates" in Encyclopedia of Polymer Science and Engineering, Volume 11, Second Edition, 1988, pages 648-718;
4. U. Grigo, K. Kircher and P.R. Müller "Polycarbonate" in Becker/Braun, Kunststoff-Handbuch, Volume 3/1, Polycarbonate, Polyacetale, Polyester, Celluloseester, Carl Hanser Verlag Munich, Vienna, 1992, pages 117-299.

10 The polycarbonates including the polyester carbonates preferably have average molecular weights M_w of 12,000 to 120,000 g/mol (obtained by measuring the relative viscosity at 25°C in methylene chloride at a concentration of 0.5 g polycarbonate per 100 ml methylene chloride). 15,000 to 80,000 g/mol are preferred, whereby 15,000 to 60,000 g/mol are particularly preferred.

15 Suitable diphenols for production of the polycarbonates according to the invention are, for example, hydroquinone, resorcinol, dihydroxydiphenyl, bis(hydroxyphenyl) alkanes, bis(hydroxyphenyl) cycloalkanes, bis(hydroxyphenyl) sulfides, bis(hydroxyphenyl) ethers, bis(hydroxyphenyl) ketones, bis(hydroxyphenyl) sulfones, bis(hydroxyphenyl) sulfoxides, (α,α' -bis(hydroxyphenyl) diisopropyl benzenes, and ring-alkylated and ring-halogenated compounds thereof.

Preferred diphenols are 4,4'-dihydroxydiphenyl, 2,2-bis(4-hydroxyphenyl)-1-phenylpropane, 1,1-bis(4-hydroxyphenyl) phenyl ethane, 2,2-bis(4-hydroxyphenyl) propane, 2,4-bis(4-hydroxyphenyl)-2-methyl butane, 1,1-bis(4-hydroxyphenyl)-m/p diisopropyl benzene, 2,2-bis(3-methyl-4-hydroxyphenyl) propane, bis(3,5-dimethyl-4-hydroxyphenyl) methane, 2,2-bis(3,5-dimethyl-4-hydroxyphenyl) propane, bis(3,5-dimethyl-4-hydroxyphenyl) sulfone, 2,4-bis(3,5-dimethyl-4-hydroxyphenyl)-2-methyl butane, 1,1-bis(3,5-dimethyl-4-hydroxyphenyl)-m/p-diisopropyl benzene, 2,2- and 1,1-bis(4-hydroxyphenyl)-3,3,5-trimethylcyclohexane.

Particularly preferred diphenols are 4,4'-dihydroxydiphenyl, 1,1-bis(4-hydroxyphenyl) phenyl ethane, 2,2-bis(4-hydroxyphenyl) propane, 2,2-bis(3,5-dimethyl-4-hydroxyphenyl) propane, 1,1-bis(4-hydroxyphenyl) cyclohexane, 1,1-bis(4-hydroxyphenyl)-m/p-diisopropyl benzene and 1,1-bis(4-hydroxyphenyl)-3,3,5-trimethylcyclohexane.

These and other suitable diphenols and their production are disclosed in, for example, US-A 3 028 635, 2 999 835, 3 148 172, 2 991 273, 3 271 367, 4 982 014 and 2 999 846, in DE-A 1 570 703, 2 063 050, 2 036 052, 2 211 956 and 3 832 396, FR-A 1 561 518, in the monograph "H. Schnell, Chemistry and Physics of Polycarbonates, Interscience Publishers, New York 1964" and in JP-A 62039/1986, 62040/1986 and 105550/1986.

In the case of homopolycarbonates only one diphenol is used, in the case of copolycarbonates several diphenols are used, whereby the diphenols used (also known as bisphenols), like all other chemicals and auxiliary substances added to the synthesis, can of course be contaminated with impurities originating from their own synthesis, although it is desirable to work with the cleanest possible raw materials.

Suitable chain terminators that can be used in the production of the polycarbonates are both monophenols and monocarboxylic acids.

Suitable monophenols are for example phenol, alkyl phenols such as cresols, p-tert.-butyl phenol, p-n-octyl phenol, p-iso-octyl phenol, p-n-nonyl phenol and p-iso-nonyl phenol, halophenols such as p-chlorophenol, 2,4-dichlorophenol, p-bromophenol and 2,4,6-tribromophenol, or mixtures thereof.

Suitable monocarboxylic acids are for example benzoic acid, alkyl benzoic acids and halobenzoic acids.

Preferred chain terminators are the phenols having the formula (I)



where R^6 stands for H or a branched or unbranched C_1 - C_{18} alkyl radical.

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The quantity of chain terminator to be used is preferably 0.5 mol% to 10 mol%, relative to moles of diphenols used in each case. The chain terminators can be added before, during or after phosgenation.

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The polycarbonates can be branched. Suitable branching agents that can be used for branching the polycarbonates are the trifunctional or more than trifunctional compounds known in polycarbonate chemistry, particularly those having three or more than three phenolic OH groups.

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Suitable branching agents are, for example, phloroglucinol, 4,6-dimethyl-2,4,6-tri(4-hydroxyphenyl) heptene-2, 4,6-dimethyl-2,4,6-tri(4-hydroxyphenyl) heptane, 1,3,5-tri(4-hydroxyphenyl) benzene, 1,1,1-tri(4-hydroxyphenyl) ethane, tri(4-hydroxyphenyl) phenyl methane, 2,2-bis[4,4-bis(4-hydroxyphenyl) cyclohexyl] propane, 2,4-bis(4-hydroxyphenyl isopropyl) phenol, 2,6-bis(2-hydroxy-5'-methylbenzyl)-4-methylphenol, 2-(4-hydroxyphenyl)-2-(2,4-dihydroxyphenyl) propane, hexa(4-(4-hydroxyphenyl isopropyl) phenyl) orthoterephthalic acid ester, tetra(4-hydroxyphenyl) methane, tetra(4-(4-hydroxyphenyl isopropyl) phenoxy) methane and 1,4-bis(4',4''-dihydroxytriphenyl) methyl benzene, as well as 2,4-dihydroxybenzoic acid, trimesic acid, cyanuric chloride and 3,3-bis(3-methyl-4-hydroxyphenyl)-2-oxo-2,3-dihydroindole.

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The quantity of branching agents optionally to be used is preferably 0.05 mol% to 2.5 mol%, relative to moles of diphenols used in each case.

The branching agents can either be included with the diphenols and chain terminators in the aqueous-alkaline phase or added ahead of phosgenation dissolved in an organic solvent.

- 5 All these measures for producing polycarbonates are familiar to the person skilled in the art.

Suitable aromatic dicarboxylic acids for production of the polyester carbonates are, for example, phthalic acid, terephthalic acid, isophthalic acid, tert.-butyl isophthalic acid, 3,3'-diphenyl dicarboxylic acid, 4,4'-diphenyl dicarboxylic acid, 4,4-benzophenone dicarboxylic acid, 3,4'-benzophenone dicarboxylic acid, 4,4'-diphenyl ether dicarboxylic acid, 4,4'-diphenyl sulfone dicarboxylic acid, 2,2-bis(4-carboxyphenyl) propane, trimethyl-3-phenyl indane-4,5'-dicarboxylic acid.

- 15 Of the aromatic dicarboxylic acids, terephthalic acid and/or isophthalic acid are particularly preferably used.

Derivatives of dicarboxylic acids are for example the dicarboxylic acid dihalides and the dicarboxylic acid dialkyl esters, particularly the dicarboxylic acid dichlorides and the dicarboxylic acid dimethyl esters.

Replacement of the carbonate groups by the aromatic dicarboxylic acid ester groups is performed substantially stoichiometrically and also quantitatively, such that the molar ratio of the reaction partners is also found in the final polyester carbonate. The aromatic dicarboxylic acid ester groups can be incorporated both randomly and in blocks.

The polycarbonates to be used according to the invention are preferably produced by the interfacial polycondensation process or by the known melt interesterification process. In the first case phosgene is preferably used as carbonic acid derivative, in the second case preferably diphenyl carbonate.

In the first case phosgene is preferably used as carbonic acid derivative, in the second case preferably diphenyl carbonate.

5 Catalysts, solvents, working up, reaction conditions, etc., for polycarbonate production are known in both cases.

The melt interesterification process is described in particular in H. Schnell, "Chemistry and Physics of Polycarbonates", Polymer Reviews, Volume 9, p. 44 to
10 51, Interscience Publishers, New York, London, Sydney, 1964 and in DE-A 1 031 512, in US-A 3 022 272, in US-A 5 340 905 and in US-A 5 399 659.

The polycarbonates according to the invention can also contain the conventional additives, for example pigments, UV stabilisers, heat stabilisers, antioxidants and
15 mould release agents in the conventional quantities for polycarbonates.

Where the polycarbonates contain additives or other accessory agents, the compositions comprising polycarbonate and additives or accessory agents are also known as polycarbonate moulding compositions.

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These conventional additives can be added to polycarbonates by known means together with the components according to the invention or subsequently.

The polycarbonate moulding compositions according to the invention can be
25 processed into moulded articles on conventional processing machines by known methods under the conventional processing parameters for polycarbonate.

Raw materials and auxiliary substances having a low degree of impurities are preferably used in the production of polycarbonate. For production by the melt
30 interesterification process in particular, the bisphenols used and the carbonic acid derivatives used should be as free as possible from alkali ions and alkaline-earth

ions. Pure raw materials of this type can be obtained for example by recrystallising, washing or distilling the carbonic acid derivatives, for example carbonic acid esters, and the bisphenols.

5 Where polycarbonates are produced by the melt interesterification process, the reaction of bisphenol and the carbonic acid diester can be performed continuously or discontinuously for example in stirred-tank reactors, thin-film vaporisers, falling film vaporisers, series of stirred-tank reactors, extruders, kneaders, simple disc reactors and high-viscosity disc reactors.

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Carbonic acid diesters that can be used to produce polycarbonates are for example diaryl esters of carbonic acid, whereby the two aryl radicals preferably each have 6 to 14 C atoms. The diesters of carbonic acid based on phenol or alkyl-substituted phenols, in other words diphenyl carbonate or dicresyl carbonate for example, are preferably used. Relative to 1 mole of bisphenol, the carbonic acid diesters are
15 preferably used in a quantity of 1.01 to 1.30 mol, particularly preferably in a quantity of 1.02 to 1.15 mol.

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If phenols, alkyl phenols and/or aryl phenols are used in the production of the polycarbonates according to the invention, they have the effect of chain terminators. That means that they limit the maximum average molecular weight that can be achieved. They can either be added together with the monomers that are needed for production of the polycarbonate or in a later phase of polycarbonate synthesis. They act as monofunctional compounds within the meaning of polycarbonate synthesis
25 and thus act as chain terminators.

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The phenol, alkyl phenols and/or aryl phenols that are optionally used in the production of the polycarbonate are preferably used in a quantity of 0.25 to 10 mol%, relative to the sum of bisphenols used in each case.

Mixtures of phenol and/or one or more alkyl phenols and/or aryl phenols can also be used.

5 The alkyl phenols and/or aryl phenols optionally used in the production of the polycarbonate lead to alkyl phenyl terminal groups or to aryl phenyl terminal groups. In addition, other terminal groups can occur in the polycarbonate that is formed, such as e.g. phenolic OH terminal groups or chloroformic acid ester terminal groups, depending on the production process.

10 Exclusively phenol, alkyl phenols and/or aryl phenols, without the addition of additional substances that can act as chain terminators, are preferably used as chain terminators.

15 Suitable additional substances that can act as chain terminators are both monophenols and monocarboxylic acids. Suitable monophenols are e.g. phenol, p-chlorophenol or 2,4,6-tribromophenol. Suitable monocarboxylic acids are benzoic acid, alkyl benzoic acids and halobenzoic acids.

20 The preferred additional substances that can act as chain terminators are phenol, p-tert. butyl phenol, cumyl phenol and isooctyl phenol.

The quantity of additional substances that can act as chain terminators is preferably between 0.25 and 10 mol%, relative to the sum of bisphenols used in each case.

25 The measurement process for determining the uniaxial extensional viscosity is described below.

30 In order to measure the uniaxial extensional viscosity, a cylindrical polycarbonate specimen (dimensions: diameter between approximately 4 and 5 mm, length between approximately 20 and 25 mm) is fixed at the ends by clamps and clamped in an extensional rheometer.

The specimen is then heated by means of an oil bath, which at the measurement temperature of 200°C has approximately the same density as the polycarbonate. Once temperature constancy is reached (after approximately 10 min), the deformation is set by means of the take-off rod, which is connected to the clamps at one end of the specimen. A constant Hencky strain rate $\dot{\epsilon}$ is set. This means that the take-off rate u increases exponentially over time.

At the other end of the specimen the tensile force is measured as a function of time or total elongation. The uniaxial extensional viscosity can be determined by relating the tensile stress that is determined to the time-related cross-sectional area.

In the case of the extensional rheometer used for the measurements in the examples in the present specification, the maximum take-off length is approximately 500 mm, which corresponds to a maximum deformation of $L/L_0 = 25$ or a maximum Hencky strain of approximately $\ln(L/L_0) = 3.2$. The total elongation was not always achieved in the polycarbonates that were tested, however, since the specimens can tear or fail beforehand.

The uniaxial elongation test is evaluated as follows. The logarithm of the simple extensional viscosity value and of the shear viscosity value multiplied by three are plotted together against time in a diagram. Surprisingly it was established that the polycarbonates that are suitable for the production of containers are precisely those in which the extensional viscosities rise sharply by comparison to the shear viscosity multiplied by three (see Fig. 1). The polycarbonates in which the extensional viscosities do not rise sharply by comparison to the shear viscosity multiplied by three (see Fig. 2) are less suitable or unsuitable for the production of water bottles.

Fig. 1 and Fig. 2 are described below.

Fig. 1 shows the uniaxial extensional viscosity $\eta_E(t, \dot{\epsilon})$ and the shear viscosity multiplied by three, $3\eta(t)$, for a polycarbonate that is suitable for the production of water bottles (produced as described in the example according to the invention). The shear viscosity multiplied by three, $3\eta(t)$, is represented as a continuous line. The uniaxial extensional viscosities $\eta_E(t, \dot{\epsilon})$ for three different strain rates $\dot{\epsilon}$ of 0.3, 0.1 and 0.01 (unit: 1 divided by seconds) are represented as lines with symbols. It can be seen that for all strain rates the extensional viscosities rise sharply over time and are ultimately higher than the shear viscosity multiplied by three.

Fig. 2 shows the uniaxial extensional viscosity $\eta_E(t, \dot{\epsilon})$ and the shear viscosity multiplied by three, $3\eta(t)$, for a polycarbonate that is not suitable for the production of water bottles (produced as described in the comparative example). The shear viscosity multiplied by three, $3\eta(t)$, is represented as a continuous line. The uniaxial extensional viscosities $\eta_E(t, \dot{\epsilon})$ for three different strain rates $\dot{\epsilon}$ of 0.2, 0.1 and 0.05 (unit: 1 divided by seconds) are represented as lines with symbols. It can be seen that for all strain rates the extensional viscosities do not rise very sharply over time and ultimately come to rest in the vicinity of the shear viscosity multiplied by three.

In Fig. 1 and Fig. 2 the time axis t for a curve with a specific Hencky strain rate $\dot{\epsilon}$ can be extrapolated to the Hencky strain ϵ , since:

$$\text{Hencky strain } \epsilon = \text{Hencky strain rate } \dot{\epsilon} \text{ multiplied by time } t$$

Fig. 3 shows the bottles produced in the examples. Their dimensions are given in millimetres (mm).

Fig. 4 shows the position of the measuring points on the bottles at which the wall thickness was measured in the examples.

Fig. 5 shows the variation in wall thickness reproduced in Table 2 in graphical form. The wall thickness in mm is plotted across the measuring points 1 to 46. The bottle made from the polycarbonate according to the invention displays a regular variation (square symbols). The bottle made from the polycarbonate according to the comparative example displays an irregular variation (triangular symbols).

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Examples

5 A polycarbonate was produced with the extensional rheological properties according to the invention (according to the example). The plastic pellets were then used to make water bottles with a volume of 5 gallons and the wall thickness distribution was measured. A corresponding process was performed with a comparative product that does not have the extensional rheological properties according to the invention (according to the comparative example).

10 Water bottles having a homogeneous wall thickness distribution were obtained from the polycarbonate according to the invention but not from the polycarbonate according to the comparative example.

1. Production of the polycarbonates

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Example:

20 5515.7 g (24.16 mol) bisphenol A and 31.10 g isatin bis-cresol were dissolved in 33.40 kg of 6.5% sodium hydroxide solution under a nitrogen atmosphere whilst being stirred. A mixture of 70.6 g phenol and 36.03 kg methylene chloride was added to this solution. 2967.7 g phosgene were then introduced over 30 minutes with intensive stirring at 20 to 25°C and a pH of 13, which was maintained by the addition of further sodium hydroxide solution. This introduction was followed by the addition of 28.3 g N-ethyl piperidine with stirring for 45 minutes at a pH of 13.

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The alkaline phase was separated from the organic phase. The organic phase was adjusted to a pH of 1 with dilute phosphoric acid or hydrochloric acid. It was then washed free from electrolytes with deionised water. After replacing the methylene chloride with chlorobenzene the polycarbonate was isolated by known means using
30 an evaporation extruder.

The polycarbonate thus obtained had a relative solution viscosity, measured in a concentration of 0.5 g polycarbonate in 100 ml methylene chloride at 25°C, of 1.325.

Comparative example:

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6.91 g isatin bis-cresol and 78.4 g phenol were used as in the example above. A polycarbonate having a relative solution viscosity of 1.305 was obtained. Isatin bis-cresol is available commercially and its correct name is 3,3-bis(3-methyl-4-hydroxyphenyl)-2-oxo-2,3-dihydroindole.

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2. Description of the production of 5-gallon water bottles from polycarbonate by the extrusion blow moulding process

The bottles were produced with a KBS 2-20 extrusion blow moulding machine supplied by SIG Blowtec, which was equipped as follows. An extruder having a screw measuring 100 mm in diameter and having a length of 25 D was used, which at relatively low screw speeds introduced little frictional heat into the material. The plasticising capacity was between approx. 145 and 190 kg/h with a bottle weight of approx. 750 g net and a production rate of 130 to 144 bottles/hour. The plasticising cylinder was equipped with regulated heating zones and fans, which ensure an exact and constant temperature control. It was driven by means of a thyristor-controlled d.c. unit, which provided a uniform delivery of material and a constant torque. The parison die consisted of a fifo accumulator head (fifo = first in – first out) with a 3.5 litre accumulation volume and overlapping heart-shaped grooves. The double heart-shaped grooves, which are offset by 180°, produce an internal and external parison and convey the melt stream into the accumulator chamber. The mandrel and die in the die head were of a conical design. The mandrel was moved axially relative to the conical die by means of a wall thickness control program. This enabled the weight of the bottle to be optimised and the wall thickness to be adjusted to the corresponding bottle sections, such as e.g. in the base section.

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The extruder temperatures were 110°C in the feed zone and between 245°C and 265°C in the individual heating zones. The die head temperatures were between 245°C and 250°C and the die temperature 275°C. The composition temperature was determined as 267°C. The average cycle time was $25.8 \text{ s} \pm 0.2 \text{ s}$, with a parison delivery time of 5.3 s, which corresponds to a production rate of 138 to 140 bottles per hour. A conventional vertical wall thickness profile for 5-gallon polycarbonate bottles was used to control the wall thickness. The bottles that were produced had a net weight of 750 g to 850 g and were conditioned immediately afterwards by infrared radiation. The purpose of the conditioning was to relax the material quickly and to reduce the associated process-related internal stress. An infrared radiation oven supplied by Process Dynamics Inc., USA, with the model name Protherm 850-3, serial no.: KRK 7110, was used. The setting temperatures for the seven heating zones that were provided were selected such that a surface temperature of the bottles of $130^\circ\text{C} \pm 2^\circ\text{C}$ was ensured.

Table 1

Bottle geometry and weight of the water bottle from the example / comparative example:

Example	Average wall thickness [mm]	Surface area [cm²]	Volume [cm³]	Calculated weight [g]
Neck	2.35	129.53	30.440	36.53
Shoulder	2.01	642.44	129.130	154.96
Body	1.30	2747.82	357.217	428.66
Base	2.14	547.11	117.082	140.50
Total		4066.90	633.87	760.65

Comparative example	Average wall thickness [mm]	Surface area [cm²]	Volume [cm³]	Calculated weight [g]
Neck	2.75	129.53	35.588	42.71
Shoulder	2.30	642.44	147.681	177.22
Body	1.35	2747.82	369.696	443.64
Base	2.23	547.11	122.224	146.67
Total		4066.90	675.19	810.23

3. Description of the wall thickness measurement on the water bottles:

The wall thicknesses were measured using an ultrasonic wall thickness gauge supplied by Krautkrämer GmbH & Co, Hürth, Germany, model number CL3 DL. This device operates on the pulse-echo principle. The measurement of the transit time of the pulse through the material begins with the entry echo that is generated when a part of the ultrasonic pulse is reflected back from the interface between the delay line and the surface of the material to be measured. Depending on the thickness of the material, the CL3 DL automatically decides whether to measure from the entry echo to the first back-wall echo (interface-to-first mode) or to measure between successive back-wall echoes (multi-echo mode). An ultrasonic delay line probe for a measuring range from 0.125 mm to 3.8 mm specially designed for plastics, the ALPHA DFR-P, was used, with a nominal frequency of 22 MHz and a coupling area of 6.4 mm. The wall thickness measurements were performed at 46 measuring points (see Fig. 4) directly on the bottle using an ultrasonic couplant.

Table 2**Wall thicknesses at the measuring points**

Measuring point	Measuring zone	Wall thickness [mm] Example	Wall thickness [mm] Comp. example
1	Neck	2.27	2.57
2	Neck	2.42	2.92
3	Shoulder	2.28	2.78
4	Shoulder	2.14	2.66
5	Shoulder	1.88	2.39
6	Shoulder	1.72	1.92
7	Body	1.53	1.63
8	Body	1.36	1.36
9	Body	1.22	1.14
10	Body	1.16	1.45
11	Body	1.14	1.08
12	Body	1.16	1.32
13	Body	1.19	1.17
14	Body	1.24	1.78
15	Body	1.3	1.86
16	Body	1.38	1.96
17	Body	1.45	1.76
18	Body	1.57	1.89
19	Base	1.72	1.78
20	Base	1.94	2.28
21	Base	2.16	2.56
22	Base	2.33	2.73
23	Base	2.46	2.53
24	Base	2.45	2.39
25	Base	2.35	2.48
26	Base	2.19	2.29
27	Base	2.02	1.94
28	Base	1.76	1.36
29	Body	1.58	1.21

Measuring point	Measuring zone	Wall thickness [mm] Example	Wall thickness [mm] Comp. example
30	Body	1.45	1.09
31	Body	1.35	1.37
32	Body	1.29	1.43
33	Body	1.25	1.34
34	Body	1.19	0.94
35	Body	1.16	1.18
36	Body	1.15	0.96
37	Body	1.14	1.27
38	Body	1.22	0.94
39	Body	1.33	1.03
40	Body	1.48	1.13
41	Shoulder	1.68	1.35
42	Shoulder	1.92	2.09
43	Shoulder	2.12	2.51
44	Shoulder	2.3	2.69
45	Neck	2.45	2.86
46	Neck	2.25	2.64

In some cases the polycarbonates that are unsuitable for the production of water bottles cannot be deformed at all to high total elongation values ($\epsilon > 2.5$) since the samples constrict and/or fail.

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The measurement results for uniaxial extensional viscosity can depend very greatly on the correct experimental procedure. If the experiment is performed incorrectly, very elevated extensional viscosities can be measured that do not actually exist; in order to obtain correct measured values it is important that the experiment is performed and evaluated adequately (cf. Th. Schweizer, Rheol. Acta 39 (2000) 5, pages 428-443; J. S. Schulze et al., Rheol. Acta 40 (2001) pages 457-466; and V. C. Barroso, J. A. Covas, J. M. Maia Rheol. Acta 41 (2002) pages 154-161).

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Claims

1. A polycarbonate in which in the uniaxial elongation test performed at a temperature of 200°C with rising Hencky strain ϵ the extensional viscosity η_E rises more sharply than the value of the shear viscosity multiplied by 3, 3η .
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2. The polycarbonate according to claim 1, whereby at a temperature of 200°C, with a Hencky strain ϵ of 2.0 and a strain rate range $\dot{\epsilon}$ of between 0.1 and 0.01, the ratio S is greater than 1.1 and with a Hencky strain ϵ of 2.5 and a strain rate range $\dot{\epsilon}$ of between 0.1 and 0.01, the ratio S is greater than 1.1, whereby S is defined as $S = \eta_E$ divided by 3η .
10
3. The polycarbonate according to claim 1, whereby at a temperature of 200°C, with a Hencky strain ϵ of 2.0 and a strain rate range $\dot{\epsilon}$ of between 0.1 and 0.01, the ratio S is greater than 1.3 and with a Hencky strain ϵ of 2.5 and a strain rate range $\dot{\epsilon}$ of between 0.1 and 0.01, the ratio S is greater than 1.5, whereby S is defined as $S = \eta_E$ divided by 3η .
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4. A container containing the polycarbonate according to one of claims 1 to 3.
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5. The container according to claim 4, whereby the container is a water bottle.
6. A process for producing the container according to claim 4 or 5 by extrusion blow moulding or by injection blow moulding.
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