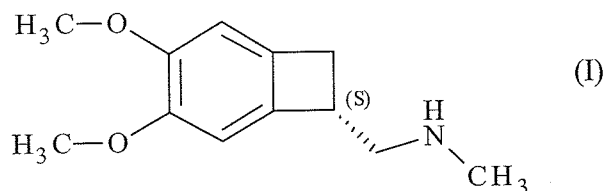
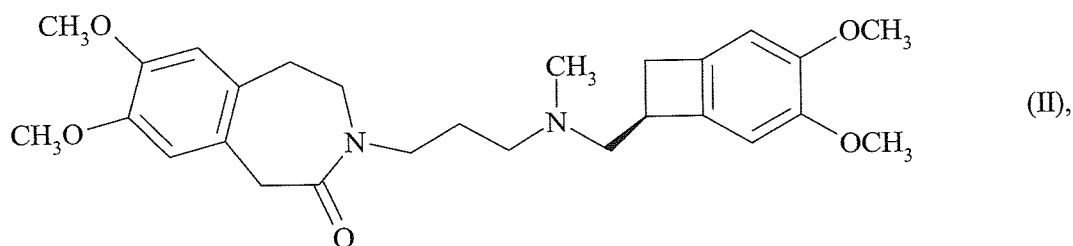


PROCESS FOR THE ENZYMATIC SYNTHESIS OF (7S)-1-(3,4-DIMETHOXYBICYCLO[4.2.0]OCTA-1,3,5-TRIEN-7-YL) N-METHYL METHANAMINE, AND APPLICATION IN THE SYNTHESIS OF IVABRADINE AND SALTS THEREOF

The present invention relates to a process for the enzymatic synthesis of the compound of formula (I), (7S)-1-(3,4-dimethoxybicyclo[4.2.0]octa-1,3,5-trien-7-yl) N-methyl methanamine:



5 and to the application thereof in the synthesis of ivabradine of formula (II):



or 3-{3-[[[(7S)-3,4-dimethoxybicyclo[4.2.0]octa-1,3,5-trien-7-yl]methyl]}(methyl)amino]-propyl}-7,8-dimethoxy-1,3,4,5-tetrahydro-2H-3-benzazepin-2-one,

its addition salts with a pharmaceutically acceptable acid and their hydrates.

10 Ivabradine, and addition salts thereof with a pharmaceutically acceptable acid, and more especially the hydrochloride thereof, have very valuable pharmacological and therapeutic properties, especially bradycardic properties, which render those compounds useful in the treatment or prevention of various clinical conditions of myocardial ischaemia, such as angina pectoris, myocardial infarction and associated rhythm disorders, as well as in various
15 pathologies involving rhythm disorders, especially supraventricular rhythm disorders, and in heart failure.

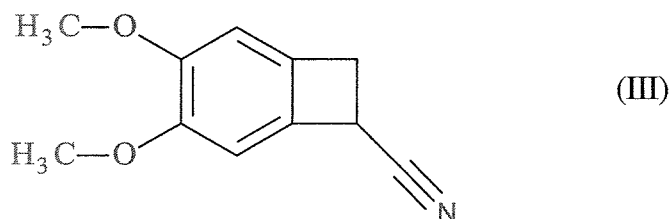
The preparation and therapeutic use of ivabradine and addition salts thereof with a pharmaceutically acceptable acid, and more especially the hydrochloride thereof, have been described in European patent specification EP 0 534 859.

That patent specification describes the synthesis of ivabradine hydrochloride starting from the compound of formula (I).

The compound of formula (I) is a key intermediate in the synthesis of ivabradine and its pharmaceutically acceptable salts.

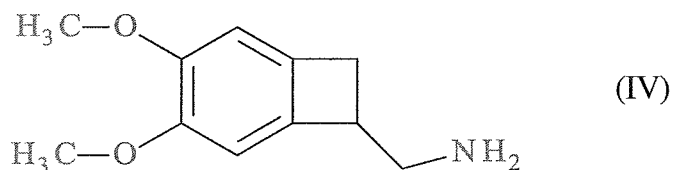
- 5 The prior art discloses several methods for obtaining the compound of formula (I).

Patent specification EP 0 534 859 describes the synthesis of the compound of formula (I) by reduction of the racemic nitrile of formula (III):

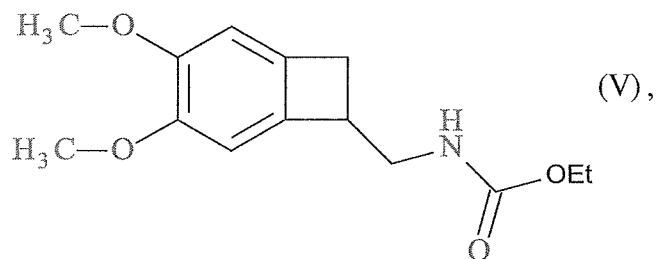


by BH_3 in tetrahydrofuran,

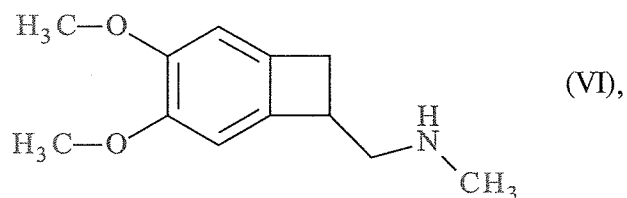
- 10 followed by addition of hydrochloric acid, to yield the hydrochloride of the racemic amine of formula (IV):



which is reacted with ethyl chloroformate to yield the carbamate of formula (V):



- 15 the reduction of which, by LiAlH_4 , yields the racemic methylated amine of formula (VI):



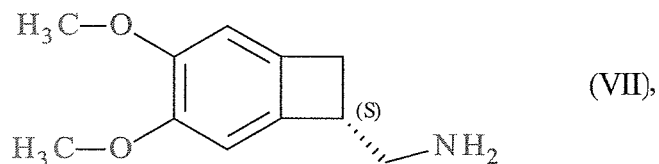
the resolution of which, using camphorsulphonic acid, yields the compound of formula (I).

That method has the disadvantage of yielding the compound of formula (I) in only a very low yield of 2 to 3 % starting from the racemic nitrile of formula (III).

- 5 That very low yield is due to the low yield (4 to 5 %) of the step of resolution of the secondary amine of formula (VI).

Patent specification EP 1 598 333 describes obtaining the compound of formula (I) by conversion of the racemic primary amine of formula (IV) into a salt using N-acetyl-L-glutamic acid, followed by recrystallisation and then returning to the base, to yield the

10 optically active primary amine of formula (VII):

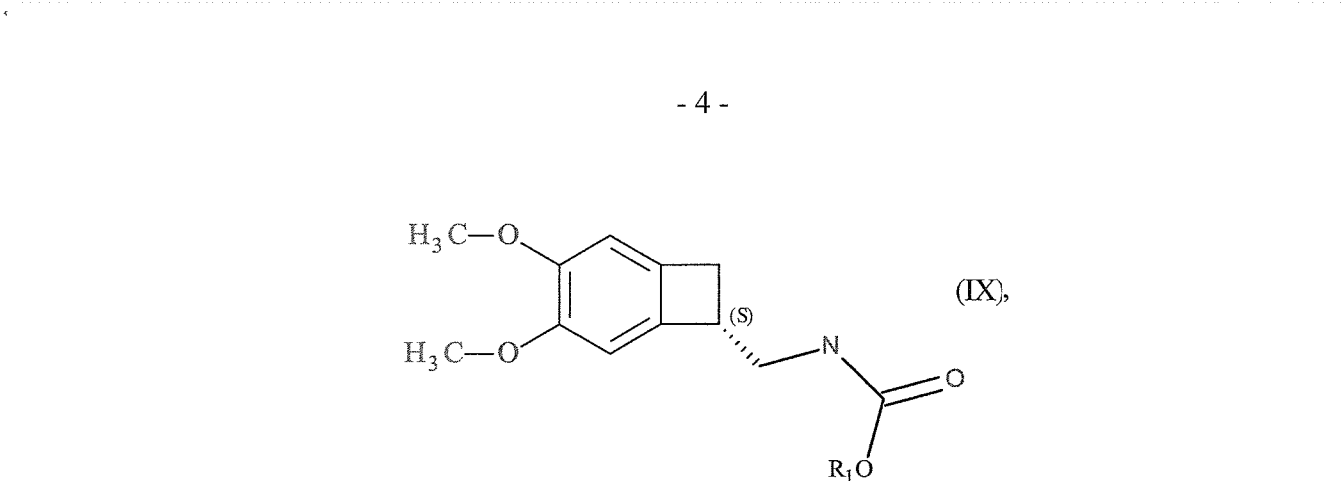


which is then methylated using the same reaction sequence as above (conversion into the carbamate, and then reduction).

- That method results in the methanamine of formula (I) in 4 steps in a yield of about 30 %
- 15 starting from the racemic primary amine of formula (IV).

The problem of the present invention was to obtain the compound of formula (I) by reducing the number of steps starting from the racemic primary amine of formula (IV) whilst maintaining a good overall yield.

- More specifically, the present invention relates to a process for the synthesis of the carbamate
- 20 of formula (IX):



wherein R₁ represents a linear or branched C₁-C₆alkyl, allyl or benzyl group,

by enantioselective enzymatic acylation of the racemic amine of formula (IV) using a lipase (EC 3.1.1.3 in the international classification of enzymes)

5 with a carbonate of formula $R_1O-(CO)-OR_1$, wherein R_1 is as defined hereinbefore,
in an amount ranging from 1 to 15 molar equivalents relative to the amine of formula (IV),
in an organic or aqueous solvent, a mixture of organic solvents or a mixture of organic and
aqueous solvents,
at a concentration from 5 to 500 g/L of compound of formula (IV) per litre of solvent or
10 mixture of solvents,
at an E/S ratio of from 10/1 to 1/100, preferably from 1/1 to 1/10,
at a temperature from 25°C to 40°C.

The carbamate of formula (IX) obtained by the process of the present invention preferably has an enantiomeric purity of more than 85 %, that is to say an enantiomeric excess of more than 70 %.

Among the lipases which may be used in the enzymatic esterification process according to the present invention there may mentioned, without implying any limitation, the lipases of *Pseudomonas fluorescens*, of *Pseudomonas cepacia*, of porcine pancreas, and the lipases PS 'Amano' SD (*Burkholderia cepacia*) and IM (immobilised on Diatomite).

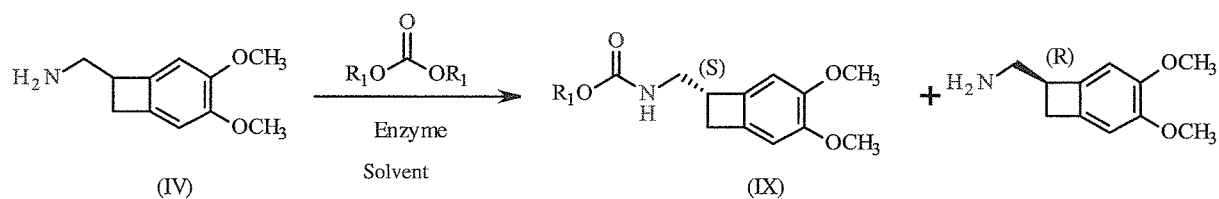
20 Lipases that are preferred according to the invention are the lipases of *Pseudomonas cepacia*
and PS 'Amano' IM.

Preferred carbonates $R_1O-(CO)-OR_1$ are those wherein R_1 represents an allyl, ethyl or benzyl group.

Among the organic solvents that may be used for the enzymatic acylation reaction according to the present invention, there may be mentioned, without implying any limitation, ethyl acetate, TBME, THF, 2-MeTHF, toluene, 1,4-dioxane, *tert*-amyl alcohol, CPME and acetonitrile.

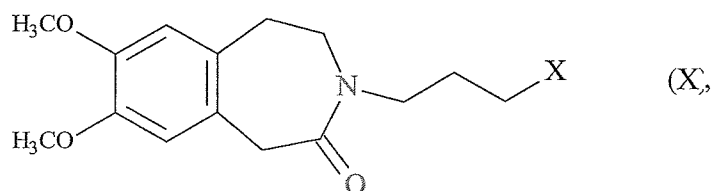
- 5 Preferred solvents are TBME, THF, 2-MeTHF and 1,4-dioxane, on their own or in admixture with a buffer of pH=7.

The enzymatic acylation scheme according to the invention is as follows:

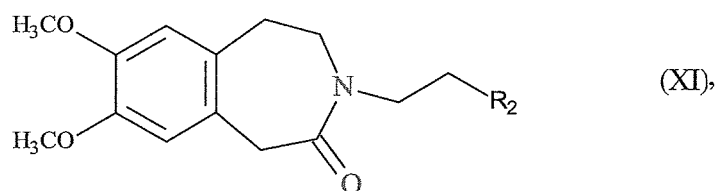


- The carbamate of formula (IX) is then isolated from the reaction mixture, and then it is reduced using an aluminium hydride such as lithium aluminium hydride ($LiAlH_4$) or sodium bis(2-methoxyethoxy)aluminium hydride (Red-Al) to yield the methylated amine of formula (I).
- 10

The latter is subsequently either coupled with a compound of formula (X):



- 15 wherein X represents a halogen atom, preferably an iodine atom, or, in the presence of a reducing agent, subjected to a reductive amination reaction with a compound of formula (XI):



wherein R_2 represents a group selected from CHO and CHR_3R_4 ,

wherein R_3 and R_4 each represent a linear or branched (C_1 - C_6)alkoxy group or form, together with the carbon atom carrying them, a 1,3-dioxane, 1,3-dioxolane or 1,3-dioxepane ring,

- 5 to yield ivabradine, which is then converted into an addition salt with a pharmaceutically acceptable acid.

The compound of formula (I) may also be used in the reductive amination reaction in the form of its addition salt with a pharmaceutically acceptable acid, preferably its hydrochloride. In that case, ivabradine is obtained directly in the form of the hydrochloride.

10 Definitions

A racemic compound is understood to be a compound in the form of a mixture of two enantiomers in a ratio of from 55:45 to 45:55.

Enantioselective acylation of an amine in the form of a mixture of two enantiomers is understood to be preferential acylation of one of the enantiomers of the mixture.

- 15 Among the pharmaceutically acceptable acids there may be mentioned, without implying any limitation, hydrochloric acid, hydrobromic acid, sulphuric acid, phosphoric acid, acetic acid, trifluoroacetic acid, lactic acid, pyruvic acid, malonic acid, succinic acid, glutaric acid, fumaric acid, tartaric acid, maleic acid, citric acid, ascorbic acid, oxalic acid, methanesulphonic acid, benzenesulphonic acid and camphoric acid.
- 20 Among the reducing agents that may be used for the reductive amination reaction between the compound of formula (I) and the compound of formula (XI) there may be mentioned, without

implying any limitation, hydride donor compounds such as sodium triacetoxyborohydride or sodium cyanoborohydride, and dihydrogen in the presence of a catalyst such as palladium, platinum, nickel, ruthenium, rhodium, or a compound thereof, especially on a support or in the form of oxides.

- 5 The preferred reducing agent for the reductive amination reaction between the compound of formula (I) and the compound of formula (XI) is dihydrogen catalysed by palladium-on-carbon.

The Examples hereinbelow illustrate the invention.

Abbreviations

10	CPME	CycloPentyl Methyl Ether
	DEA	DiEthylAmine
	E	Enantioselectivity coefficient
	E/S	Enzyme/Substrate ratio expressed in g/g
	ee	enantiomeric excess
15	eq	molar equivalent
	HPLC	High Performance Liquid Chromatography
	Red-Al	sodium bis(2-methoxyethoxy)aluminium hydride
	NMR	Nuclear Magnetic Resonance (spectroscopy)
	TBME	<i>Tert</i> -Butyl Methyl Ether
20	THF	TetraHydroFuran
	2-Me HF	2-MethylTetraHydroFuran
	rpm	revolutions per minute

EXAMPLE 1: Ethyl [(7*S*)-3,4-dimethoxybicyclo[4.2.0]octa-1,3,5-trien-7-yl]methyl]-carbamate

5 mg of 1-(3,4-dimethoxybicyclo[4.2.0]octa-1,3,5-trien-7-yl) methanamine and 12.6 mg (10 eq) of diethyl carbonate are dissolved in 2-MeTHF.

- 5 5 mg of lipase II of *Pseudomonas cepacia* (PS-CII Amano) are then added to the mixture (E/S ratio 1/1). The reaction mixture is maintained at 30°C, with rotary stirring at 250 rpm, for 24 hours to 96 hours.

The reaction is monitored by chiral-phase HPLC under conditions allowing the enantiomeric excesses of both the carbamate and the amine to be determined:

- 10 *Chiral-phase conditions:* Chiralpak® IC 250*4.6 column
50 % absolute ethanol + 0.1 % DEA + 50 % heptane + 0.1 % DEA
1 ml/min, 25°C, 288 nm

Concentration	t	Conversion c (%)	Ee (%) amine (R)	Ee (%) carbamate (S)	E
10 g/L	24 hrs	38	57.7	92.3	45
20 g/L	96 hrs	51	93.2	88.8	58
50 g/L	96 hrs	57	99.9	76.1	69

EXAMPLE 2: Ethyl [(7*S*)-3,4-dimethoxybicyclo[4.2.0]octa-1,3,5-trien-7-yl]methyl]-carbamate

15

0.5 g of 1-(3,4-dimethoxybicyclo[4.2.0]octa-1,3,5-trien-7-yl) methanamine is dissolved in 50 mL of 2-MeTHF and then diethyl carbonate (1.5 mL, 12 eq) is added. 0.5 g (E/S ratio 1/1) of lipase II of *Pseudomonas cepacia* (PS-CII Amano) is added to the mixture, which is maintained at 30°C for 48 hours with stirring at 220 rpm.

- 20 After 48 hours, the reaction mixture is filtered in order to remove the enzyme and is then evaporated. The carbamate of configuration S is obtained after separation on an SiO₂ column, eluting with cyclohexane/ethyl acetate 95/5, then 80/20 and finally 50/50, to recover the more polar amine.

The ethyl carbamate of configuration *S* (224 mg) is then obtained in a yield of 32.5 % relative to the starting amine (65 % relative to the expected amount of carbamate) and with an enantiomeric purity of 90 %.

5 The reaction is monitored by chiral-phase HPLC under conditions allowing the enantiomeric excesses of both the carbamate and the amine to be determined:

Chiral-phase conditions: Chiralpak® IC 250*4.6 column
50 % absolute ethanol + 0.1 % DEA + 50 % heptane + 0.1 % DEA
1 ml/min, 25°C, 288 nm

Time (h)	Conversion (%)	Ee (carbamate)(%)	Ee (amine) (%)	E
24	34	88	45	23
48	53	81	93	35

10 **EXAMPLE 3:** Allyl {[*(7S)*-3,4-dimethoxybicyclo[4.2.0]octa-1,3,5-trien-7-yl]methyl}-carbamate

0.87 g of 1-(3,4-dimethoxybicyclo[4.2.0]octa-1,3,5-trien-7-yl) methanamine are dissolved in 100 mL of 2-MeTHF and then diallyl carbonate (1.5 mL, ~2 eq) is added. 0.5 g (E/S ratio 1/1) of lipase II of *Pseudomonas cepacia* (PS-CII Amano) is then added to the mixture, which is
15 maintained at 30° for 42 hours with stirring at 220 rpm.

The reaction mixture is then filtered in order to remove the enzyme and then evaporated. The allyl carbamate is obtained after separation on an SiO₂ column, eluting with cyclohexane/ethyl acetate 95/5, then 80/20 and finally 50/50, to recover the more polar amine. The allyl carbamate of configuration *S* (440 mg) is then obtained in a yield of 35 % relative to
20 the starting amine (70 % relative to the expected amount of carbamate) and with an enantiomeric purity of 88 %.

Time (h)	Conversion (%)	Ee (carbamate)(%)	Ee (amine) (%)	E
42	50	89	82	26

The reaction mixture is analysed by reverse-phase HPLC, and the enantioselectivity (*ee*) of the carbamate and of the amine are monitored by chiral-phase HPLC in accordance with the
25 methods described hereinbelow:

Reverse-phase conditions: Phenomenex® LUNA HST 50*3 column C18(2) 2.5µm

0 % to 100 % of B over 8 min 0.8 ml/min 40°C
 A (1000 water + 25 ACN + 1 TFA)
 B (1000 ACN + 25 water + 1 TFA)

5 *Chiral-phase conditions:* Chiralpak® IC 250*4.6 column
 50 % isopropanol + 0.1 % EA + 50 % heptane + 0.1 % EA
 1 ml/min, 30°C, 288 nm

EXAMPLE 4: benzyl {[(7*S*)-3,4-dimethoxybicyclo[4.2.0]octa-1,3,5-trien-7-yl]methyl}-carbamate

0.5 g of 1-(3,4-dimethoxybicyclo[4.2.0]octa-1,3,5-trien-7-yl) methanamine are dissolved in
 10 50 mL of 2-MeTHF, and then dibenzyl carbonate (4.5 g, 7 eq) is added. 0.5 g (E/S ratio 1/1)
 of lipase II of *Pseudomonas cepacia* (PS-C II Amano) is added to the mixture, which is
 maintained at 30° with stirring at 220 rpm.

After 24 hours, the reaction mixture is filtered to remove the enzyme and is then evaporated.
 The carbamate of configuration *S* is obtained after separation on an SiO₂ column, eluting with
 15 cyclohexane/ethyl acetate 95/5, then 80/20 and finally 50/50, to recover the more polar amine.
 The benzyl carbamate of configuration *S* (0.26 g) is then obtained in a yield of 30 % relative
 to the starting amine (60 % relative to the expected amount of carbamate) and with an
 enantiomeric purity of 95 %.

The reaction mixture is analysed by reverse-phase HPLC, and the enantioselectivity (*ee*) of
 20 the carbamate and of the amine are monitored by chiral-phase HPLC in accordance with the
 methods described hereinbelow:

25 *Reverse-phase conditions:* Phenomenex® LUNA HST 50*3 column C18(2) 2.5 µm
 0 % to 100 % of B over 8 min 0.8 ml/min 40°C
 A (1000 water + 25 ACN + 1 TFA)
 B (1000 ACN + 25 water + 1 TFA)

Chiral-phase conditions: Chiralpak® IC 250*4.6 column
 50 % isopropanol + 0.1 % DEA + 50 % heptane + 0.1 % DEA
 1 ml/min, 25°C, 288 nm

Time (h)	Conversion (%)	Ee (carbamate)(%)	Ee (amine) (%)	E
24	58	90	87	12

EXAMPLE 5: (7S)-3,4-Dimethoxybicyclo[4.2.0]octa-1,3,5-trien-7-yl] N-methyl methanamine

Load into a reactor, under nitrogen, lithium aluminium hydride (1.41 kg) and tetrahydrofuran (32.5 l), and then pour in, at 20°C, a solution of ethyl {[(7S)-3,4-dimethoxybicyclo[4.2.0]octa-1,3,5-trien-7-yl]methyl}carbamate (5 kg) in tetrahydrofuran (50 l). Heat at reflux for 1 hour and then cool to a temperature below 15°C and hydrolyse the reaction mixture with water (1 l), 5N aqueous sodium hydroxide solution (1 l) and then water (1 l). Filter off the solid obtained. Dry the organic phase. The title product is recovered in the form of an oil, in a yield of 93 %.

¹H NMR (DMSO-d₆, ppm / TMS) = 2.60 (m; 3H); 2.85 (m; 1H); 3.15 (m; 1H); 3.25 (dd; 1H); 3.30 (m; 1H); 3.62 (m; 1H); 3.70 (s; 6H); 6.82 (s; 1H); 6.89 (s; 1H); 8.48 (sl; 1H).

EXAMPLE 6: (7S)-3,4-Dimethoxybicyclo[4.2.0]octa-1,3,5-trien-7-yl] N-methyl methanamine hydrochloride

Load (7S)-3,4-dimethoxybicyclo[4.2.0]octa-1,3,5-trien-7-yl] N-methyl methanamine (5 kg), ethyl acetate (40 l) and ethanol (10 l) into a reactor. Stir at 20°C for 30 minutes and then add, via the bottom valve of the reactor or an immersed tube, gaseous hydrogen chloride (1.012 kg). The suspension obtained is stirred at 15-20°C for 1 hour and then is filtered or is drained under suction. The precipitate is washed with a mixture of ethyl acetate/ethanol 4/1 (2 x 5 l) and is then dried to yield the title product in a yield of 92 %.

EXAMPLE 7: Ivabradine hydrochloride

Load 5.5 kg of 3-[2-(1,3-dioxolan-2-yl)ethyl]-7,8-dimethoxy-1,3-dihydro-2H-3-benzazepin-2-one, 27.5 l of ethanol and 550 g of palladium-on-carbon into an autoclave.

Purge with nitrogen and then with hydrogen, heat to 55°C, and then hydrogenate at that temperature under a pressure of 5 bars until the theoretical amount of hydrogen has been absorbed.

Then return to ambient temperature and then depressurise the autoclave.

Then add 4 kg of (7S)-3,4-dimethoxybicyclo[4.2.0]octa-1,3,5-trien-7-yl] N-methyl methanamine hydrochloride, 11 l of ethanol, 5.5 l of water and 1 kg of palladium-on-carbon. Purge with nitrogen and then with hydrogen, heat to 85°C, and then hydrogenate at that temperature under a pressure of 30 bars until the theoretical amount of hydrogen has been absorbed.

Then bring back to ambient temperature, purge the autoclave and then filter the reaction mixture; distil off the solvents and then isolate the ivabradine hydrochloride by crystallisation from a toluene/1-methyl-2-pyrrolidinone mixture.

Ivabradine hydrochloride is thereby obtained in a yield of 85 % and with a chemical purity greater than 99 %.

EXAMPLE 8: Screening of lipases for the enzymatic acylation of 1-(3,4-dimethoxy-bicyclo[4.2.0]octa-1,3,5-trien-7-yl) methanamine

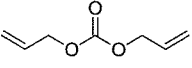
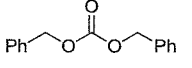
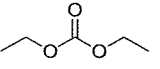
Racemic 1-(3,4-dimethoxybicyclo[4.2.0]octa-1,3,5-trien-7-yl) methanamine (5 mg; c=10 g/L) and the carbonate of formula $R_1O-(CO)-OR_1$ (10 eq) are dissolved in 0.5 mL of TBME.

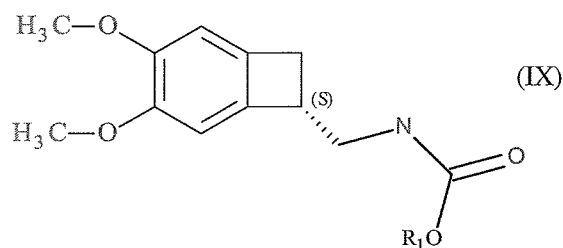
5 mg (c=10 g/L) of the lipase being studied are then added to the medium (E/S ratio = 1/1). The reaction mixture is maintained at 30°C, with rotary stirring at 250 rpm for 24 hours. The reaction mixtures are analysed by chiral-phase HPLC for checking the enantioselectivity, in accordance with the method:

*Chiralpak® IC 20 µm column, 250*4.6 acetonitrile/propan-2-ol/ DEA 90/10/0.1 % ;*

1.3 ml/min ; 30°C, 288 nm

The results are summarised in the following table:

Lipase	Carbonate	Product	Conversion c (%)	ee (%) Amine (R)	ee (%) Carbamate (S)	E
<i>Pseudomonas cepacia</i> lipase II		IXa	59	>99.9	69.8	40
<i>Pseudomonas fluorescens</i>			14	12.3	73.8	7
Lipase PS 'Amano' SD			4	3.9	83.2	11
Lipase PS 'Amano' IM			52	91.6	83.5	36
<i>Pseudomonas cepacia</i> lipase II		IXb	57	97.0	73.4	26
<i>Pseudomonas fluorescens</i>			5	3.9	78.2	8
Lipase PS 'Amano' IM			33	44.4	89.0	27
<i>Pseudomonas cepacia</i> lipase II		IXc	16	17.7	89.6	22
<i>Pseudomonas fluorescens</i>			3	2.2	66.1	5
Lipase PS 'Amano' IM			12	11.2	84.6	13



IXa: R₁ = allyl

IXb: R₁ = benzyl

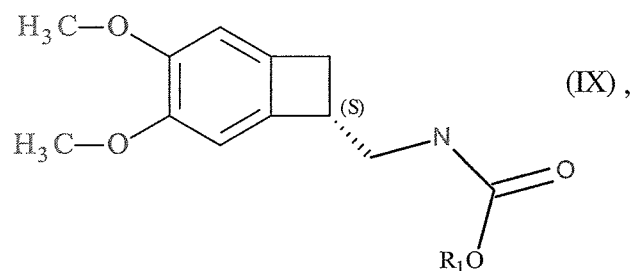
IXc: R₁ = ethyl

5 ^a Enantiomeric excess ee (en %) = % enantioE2 - % enantioE1 / % enantio E2 + % enantio E1 (enantio E2 being the predominant enantiomer)

^b Enantioselectivity coefficient E = ln[(1-c)(1-ee(S))] / ln[(1-c)(1+ee(S))] ; c = level of conversion = ee(amine) / [ee(carbamate) + ee(amine)]

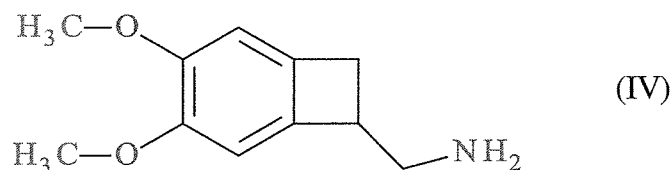
CLAIMS

1. Process for the synthesis of the compound of formula (IX):



wherein R_1 represents a linear or branched C_1 - C_6 alkyl, allyl or benzyl group,

- 5 by enantioselective enzymatic acylation of the racemic amine of formula (IV):



using a lipase (EC 3.1.1.3 in the international classification of enzymes),

with a carbonate of formula $R_1O-(CO)-OR_1$, wherein R_1 is as defined hereinbefore,

in an amount ranging from 1 to 15 molar equivalents relative to the amine of formula (IV),

in an organic or aqueous solvent, a mixture of organic solvents or a mixture of organic and aqueous solvents,

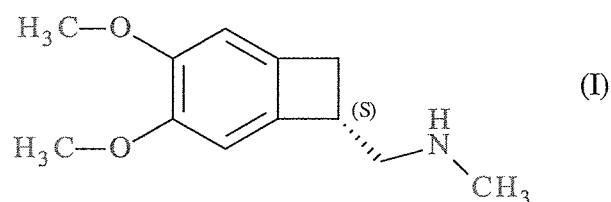
at a concentration from 5 to 500 g/L of compound of formula (IV) per litre of solvent or mixture of solvents,

at an E/S ratio of from 10/1 to 1/100, at a temperature from 25°C to 40°C.

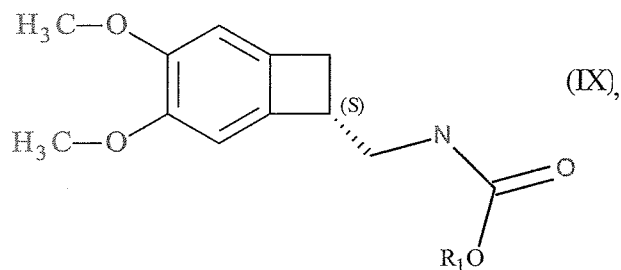
2. Synthesis process according to claim 1, wherein the lipase is selected from lipases of *Pseudomonas fluorescens*, of *Pseudomonas cepacia*, of porcine pancreas, and the lipases PS 'Amano' SD (*Burkholderia cepacia*) and IM (immobilised on Diatomite).

3. Synthesis process according to claim 2, wherein the lipase is a lipase of *Pseudomonas cepacia* or a lipase PS 'Amano' IM.

4. Synthesis process according to any one of claims 1 to 3, wherein the E/S ratio is from 1/1 to 1/10.
5. Synthesis process according to any one of claims 1 to 4, wherein the solvent is selected from TBME, THF, 2-MeTHF and 1,4-dioxane, on their own or in admixture with a buffer of pH=7.
6. Synthesis process according to any one of claims 1 to 5, wherein R₁ is an ethyl, allyl or benzyl group.
7. Process for the synthesis of the compound of formula (I):



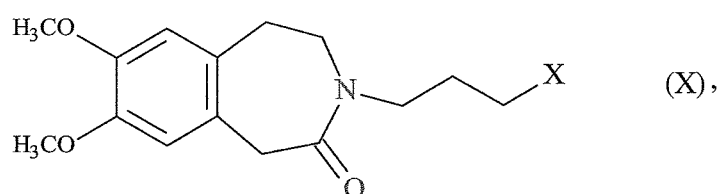
by enzymatic acylation of the racemic amine of formula (IV) in accordance with any one of claims 1 to 6 to yield the carbamate of formula (IX):



wherein R₁ represents a linear or branched C₁-C₆alkyl, allyl or benzyl group,

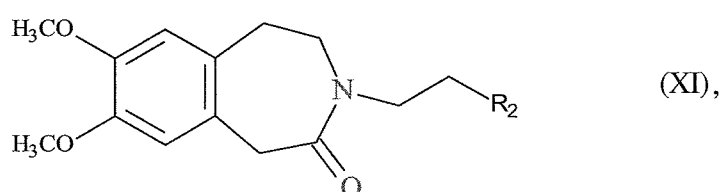
which is then reduced using a reducing agent selected from LiAlH₄ and Red-Al to yield the compound of formula (I).

8. Synthesis process according to claim 7, wherein the compound of formula (I) subsequently is either coupled with a compound of formula (X):



wherein X represents a halogen atom,

or, in the presence of a reducing agent, subjected to a reductive amination reaction with a compound of formula (XI):



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wherein R₂ represents a group selected from CHO and CHR₃R₄,

wherein R₃ and R₄ each represent a linear or branched (C₁-C₆)alkoxy group or form, together with the carbon atom carrying them, a 1,3-dioxane, 1,3-dioxolane or 1,3-dioxepane ring,

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to yield ivabradine, which is then converted into an addition salt with a pharmaceutically acceptable acid, in anhydrous or hydrate form.

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9. Synthesis process according to claim 8, wherein X is an iodine atom.
10. Synthesis process according to claim 8, wherein the compound of formula (I) used in the reductive amination reaction is in the form of its hydrochloride, to yield ivabradine in the form of the hydrochloride.
11. Synthesis process according to either claim 8 or claim 10, wherein the reductive amination reaction with a compound of formula (XI) is carried out in the presence of dihydrogen catalysed by palladium-on-carbon.