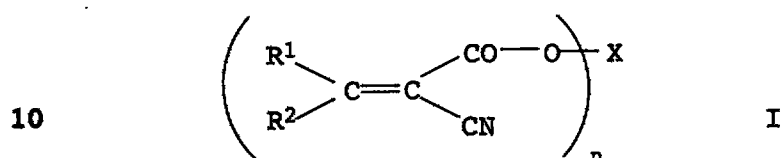


2-Cyanoacrylic esters

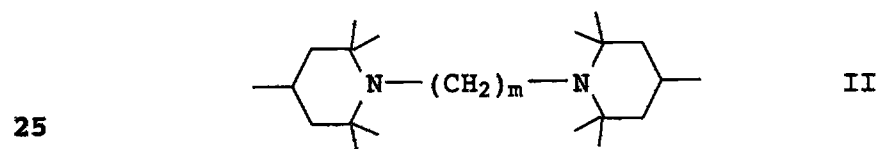
The present invention relates to novel 2-cyanoacrylic esters of
5 the formula I



where R^1 and R^2 are each hydrogen or a radical having an iso- or
heterocyclic ring system with at least one iso- or heteroaromatic
15 nucleus, and at least one of the radicals R^1 or R^2 must be dif-
ferent from hydrogen,

n is from 2 to 10, and

20 X is, when $n = 2$, a radical of the formula II



where m is from 2 to 8, and

30 X is, when $n > 2$, the radical of an n -hydric aliphatic or cyclo-
aliphatic polyol having 3-20 carbon atoms, it also being possible
for a cycloaliphatic radical to contain 1 or 2 hetero atoms, and
for an aliphatic radical to be interrupted by up to 8 non-
adjacent oxygen atoms, sulfur atoms, imino or $\text{C}_1\text{-C}_4$ -alkylimino
35 groups.

The invention furthermore relates to a process for preparing the
compounds I, to the use thereof as stabilizers, in particular
against the action of light, for organic materials, in particular
40 for cosmetic or dermatological preparations, plastics or paints,
and to organic materials which contain the compounds I.

US-A 3 215 725 and DE-A 41 22 475 disclose 2-cyanoacrylic esters
of monohydric and dihydric alcohols as light stabilizers for
45 plastics and paints.

However, these compounds have the technical disadvantage of a relatively high volatility. Since, moreover, they are only conditionally compatible with many organic materials, especially with polyolefins, they are prone, especially on storage at
 5 elevated temperature, to migrate and consequently display exudation.

It is an object of the present invention to remedy these disadvantages by novel stabilizers of the 2-cyanoacrylic ester
 10 type.

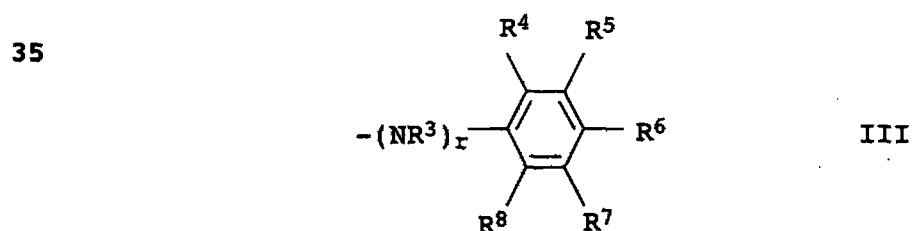
We have found that this object is achieved by the 2-cyanoacrylic esters of the general formula I defined at the outset.

15 We have furthermore found a process for preparing these compounds, their use as light protection factors or stabilizers for organic materials, and organic formulations which contain these compounds as stabilizers.

20 If the radicals R^1 and R^2 are different, the 2-cyanoacrylic ester groups of I may be either in the cis or the trans form. The preparation of the compounds usually results in mixtures of these isomers. It is possible to separate these isomers, but this is unnecessary for most industrial applications.

25 Suitable organic radicals for R^1 and R^2 are, in general, cyclic structures which contain at least one iso- or heteroaromatic nucleus, which is preferably linked directly to the 3-C atom of the acrylic group but can also be linked to this carbon atom via
 30 aliphatic or cycloaliphatic groups and via a linker $-NR^3-$.

R^1 or R^2 is preferably a radical of the formula III



where R^3 is hydrogen or C_1 - C_{10} -alkyl, r is 0 or 1, and R^4 to R^8 are each, independently of one another, hydrogen, C_1 - C_8 -alkyl, chlorine, bromine, cyano, nitro, amino, mono(C_1 - C_4 -alkyl)amino,
 45 di(C_1 - C_4 -alkyl)amino, hydroxyl, C_1 - C_8 -acyl, C_1 - C_8 -acyloxy,

C₁-C₁₈-alkoxy, C₁-C₁₂-alkoxycarbonyl, C₃-C₆-cycloalkyl or C₃-C₆-cycloalkoxycarbonyl.

Suitable radicals R³ besides hydrogen are C₁-C₁₀-alkyl radicals
 5 such as methyl, ethyl, n-propyl, isopropyl, isopropyl [sic],
 n-propyl [sic], n-butyl, isobutyl, sec-butyl, tert-butyl,
 n-pentyl, isopentyl, sec-pentyl, tert-pentyl, neopentyl, n-hexyl,
 n-heptyl, n-octyl, isooctyl, 2-ethylhexyl, n-nonyl, isononyl,
 n-decyl and isodecyl.

10

If one or more of the radicals R⁴ to R⁸ are C₁-C₈-alkyl,
 C₁-C₈-acyl, C₁-C₁₈-alkoxy or C₁-C₁₂-alkoxycarbonyl, the alkyl radi-
 cals therein can be, for example, methyl, ethyl, n-propyl, iso-
 propyl, n-propyl [sic], n-butyl, isobutyl, sec-butyl, tert-butyl,
 15 n-pentyl, isopentyl, sec-pentyl, tert-pentyl, neopentyl, n-hexyl,
 n-heptyl, n-octyl or 2-ethylhexyl.

Examples of suitable longer-chain alkyl radicals in C₁-C₁₈-alkoxy
 and C₁-C₁₂-alkoxycarbonyl groups are nonyl, 2-methylnonyl, iso-
 20 nonyl, 2-methyloctyl, decyl, isodecyl, 2-methylnonyl [sic],
 undecyl, isoundecyl, dodecyl, isododecyl, tridecyl, isotridecyl,
 tetradecyl, pentadecyl, hexadecyl, heptadecyl and octadecyl. (The
 terms isooctyl, isononyl, isodecyl and isotridecyl are trivial
 names derived from the carbonyl compounds obtained by the oxo
 25 synthesis; compare in this connection Ullmann's Encyclopedia of
 Industrial Chemistry, 5th edition, Vol. A1, pages 290-293, and
 Vol. A10, pages 284 and 285).

Examples of suitable C₃-C₆-cycloalkyl radicals are cyclopropyl,
 30 cyclobutyl, cyclopentyl, methylcyclopentyl or cyclohexyl. These
 cycloalkyl groups are also suitable radicals in C₃-C₆-cycloalkyl-
 carbonyl groups.

Preferred 2-cyanoacrylic esters I are those where R³ is hydrogen,
 35 methyl or ethyl.

Further preferred 2-cyanoacrylic esters I are those where up to
 three, particularly preferably one, of the radicals R⁴ to R⁸ are
 hydrogen, C₁-C₄-alkyl, chlorine, cyano, hydroxyl, acetyl,
 40 C₁-C₅-alkoxy, C₁-C₈-alkoxycarbonyl or cyclohexoxycarbonyl, and the
 remainder of these radicals are hydrogen.

Particularly preferred 2-cyanoacrylic esters I are those where R⁶
 is hydroxyl, methoxy, ethoxy, propoxy, isopropoxy, butoxy, iso-
 45 butoxy, sec-butoxy or tert-butoxy, because such 4-substituted
 phenyl groups contribute to the stabilizing effect of the com-
 pounds. For the same reason, 2-cyanoacrylic esters where R⁵ and/or

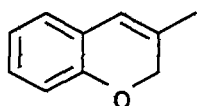
R^7 are hydrogen, methyl or tert-butyl, in particular when R^6 is hydroxyl, are also particularly preferred.

Preferred compounds I according to the invention are those where r is 0.

Further preferred compounds according to the invention are those where R^1 or R^2 is hydrogen, those where R^1 and R^2 are identical radicals, and those where one of the radicals R^1 or R^2 is phenyl-
 10 amino, p-tolylamino, p-methoxy- or p-ethoxycarbonylphenylamino and the other is hydrogen.

Another preferred radical for R^1 or R^2 is the chroman residue Ib

15



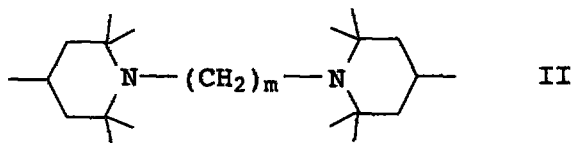
Ib

20 or its substituted derivatives, because these also enhance the stabilizing effect of the compounds I.

Further suitable radicals R^1 and R^2 are heterocyclic groups such as substituted or unsubstituted thiophenyl [sic], furfuryl and
 25 pyridyl radicals.

If $n = 2$, X is a radical of the formula II

30

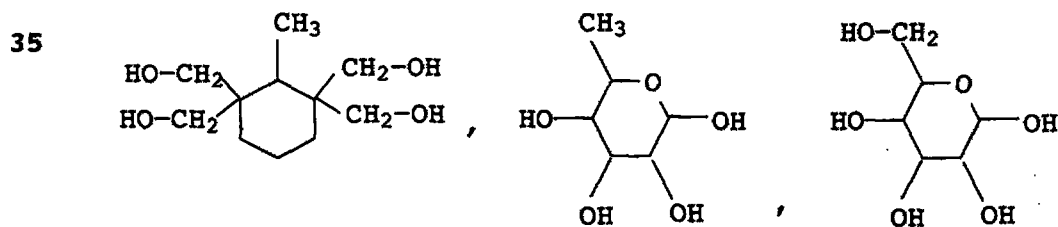
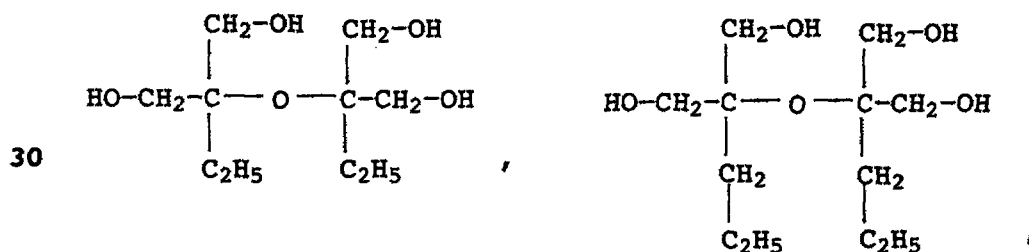
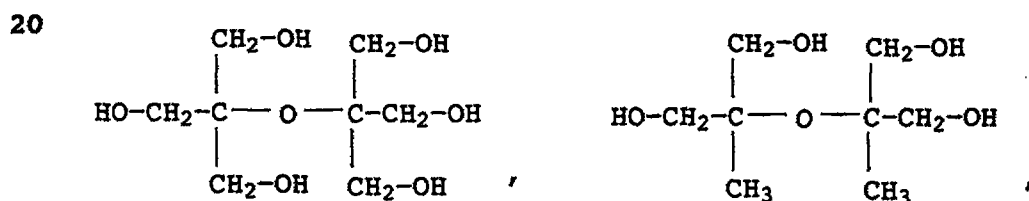
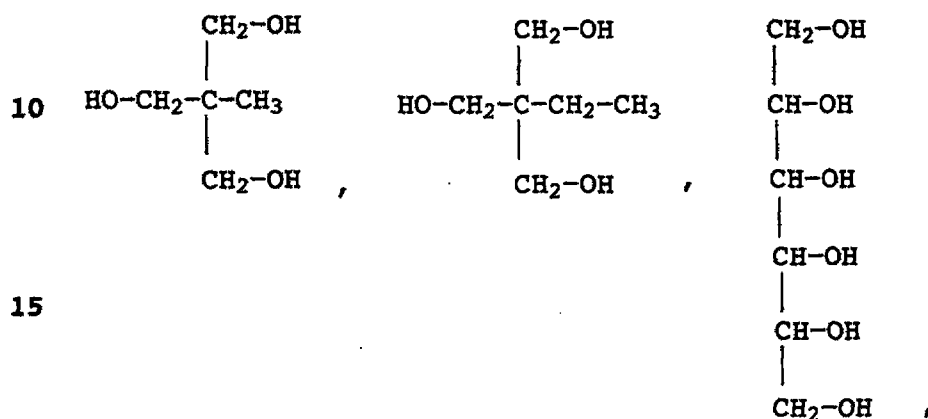
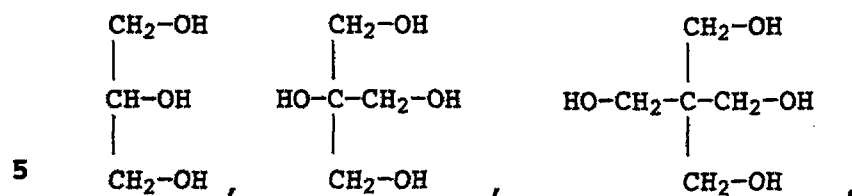


II

35 where m is from 2 to 8, preferably 2 to 6, but particularly preferably 2.

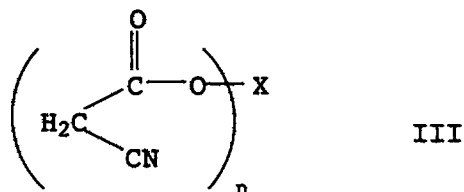
If $n > 2$, X is the radical of an n -hydric aliphatic or cycloaliphatic alcohol. These alcohols may be linear or branched,
 40 and their carbon chains can be interrupted by one or more oxygen or sulfur atoms, by imino groups ($-NH-$) or C_1-C_4 -alkylimino groups.

The group X is preferably derived from the following known
 45 polyols:



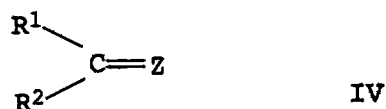
The 2-cyanoacrylic esters of the formula I where R¹ and R² are not linked via a nitrogen atom to the β-C atom are preferably obtainable by reacting cyanoacetic esters of the formula III

5



with n mol of a compound (IV)

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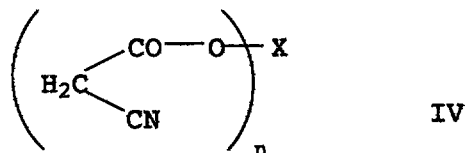


15 under the conditions of the Knoevenagel condensation. The reaction can, for example, be carried out in aromatic solvents such as toluene or xylene (see, for example, Organikum, 1976 edition, page 572). However, polar organic solvents such as dimethylformamide, dimethylacetamide, N-methylpyrrolidone, trialkyl orthoformate or alcohols such as n-propanol, n-butanol, ethylene glycol, diethylene glycol, ethylene glycol monomethyl ether, cyclohexanol or similar compounds are preferably used. If the starting compounds themselves form a liquid mixture, it is possible to dispense with an additional solvent. The reaction is preferably carried out at from 20 to 180°C, particularly preferably from 40 to 150°C. The pressure is preferably atmospheric pressure. The use of a catalyst or catalyst mixture may be advantageous depending on the reactivity of the compound IV employed. Examples of suitable catalysts are ammonium acetate, piperidine and β-alanine and acetates thereof.

Catalysts which can additionally be used for the reaction if the reaction times are very long are Lewis acids such as AlCl₃, ZrCl₄, TiCl₄ or, in particular, ZnCl₂ in the amounts customary for this purpose.

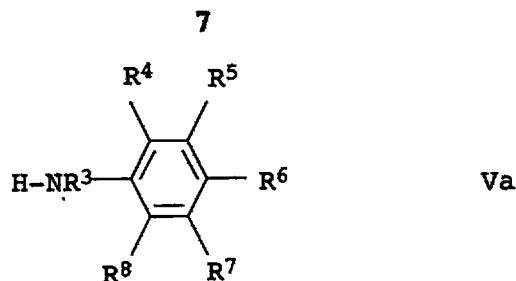
The 2-cyanoacrylic esters of the formula I where r is 1, ie. where one radical R¹ or R² is linked via a nitrogen atom to the β-C atom, can advantageously be prepared by reacting a cyanoacetic ester of the formula IV

45



with an aromatic amine of the formula Va

5



in the presence of trialkyl orthoformate. Examples of trialkyl
 10 orthoformates which have proven suitable are trimethyl orthoformate and triethyl orthoformate.

The cyanoacetic esters II can be prepared, for example, by reacting cyanoacetic acid or esters thereof with the appropriate
 15 polyols $X(OH)_n$ in the presence of a catalyst such as boric acid, Na_2CO_3 or K_2CO_3 or tetrabutyl orthotitanate, preferably in toluene or xylene.

The compounds according to the invention are outstandingly suitable for stabilizing organic materials against the action of
 20 light, oxygen and heat.

Examples of plastics which can be stabilized by the compounds I according to the invention are:

25

polymers of mono- and diolefins, eg. low and high density polyethylene, polypropylene, linear poly-1-butene, polyisoprene, polybutadiene, and copolymers of mono- or diolefins or mixtures of said polymers;

30

copolymers of mono- or diolefins with other vinyl monomers, eg. ethylene/alkyl acrylate copolymers, ethylene/alkyl methacrylate copolymers, ethylene/vinyl acetate copolymers or ethylene/acrylic acid copolymers;

35

polystyrene and copolymers of styrene or α -methylstyrene with dienes and/or acrylic derivatives, eg. styrene/butadiene, styrene/acrylonitrile (SAN), styrene/ethyl methacrylate, styrene/butadiene/ethyl acrylate, styrene/acrylonitrile/methacrylate,
 40 acrylonitrile/butadiene/styrene (ABS) or methyl methacrylate/butadiene/styrene (MBS);

halogenated polymers, eg. polyvinyl chloride, polyvinyl fluoride, polyvinylidene fluoride and copolymers thereof;

45

polymers derived from α,β unsaturated acids and derivatives thereof, such as polyacrylates, polymethacrylates, polyacrylamides and polyacrylonitriles;

- 5 polymers derived from unsaturated alcohols and amines or their acrylic derivatives or acetals, eg. polyvinyl alcohol and polyvinyl acetate;

- polyurethanes, polyamides, polyureas, polyphenylene ethers, poly-
10 esters, polycarbonates, polyoxymethylenes, polysulfones, polyether sulfones and polyether ketones.

It is furthermore possible to use the compounds I according to the invention to stabilize surface coatings, eg. industrial coat-
15 ings. Among these, particular attention is drawn to stoved coatings, and among these in turn to automotive coatings, preferably two-layer coatings.

The compounds I according to the invention can be added in solid
20 or dissolved form to the coating material. Their good solubility in coating systems is a particular advantage in this context.

The compounds I according to the invention are preferably used for stabilizing polyolefins, especially polyethylene, poly-
25 carbonates, polyamides, polyesters, polystyrene, ABS and polyurethanes. It is also possible, in particular, to stabilize sheets of said plastics.

For these applications, the compounds are employed in concentra-
30 tions of from 0.01 to 5% of the weight of the plastic, preferably in a concentration of from 0.02 to 2% by weight. Combination with other stabilizers, for example antioxidants, metal deactivators or other light stabilizers, and with antistatic agents or flame retardants, is often advantageous. Examples of particularly
35 important costabilizers are sterically hindered phenols, and phosphites, phosphonites, amines and sulfur compounds.

Examples of suitable costabilizers are:

- 40 phenolic antioxidants such as
2,6-di-tert-butyl-4-methylphenol,
n-octadecyl β -(3,5-di-tert-butyl-4-hydroxyphenyl)propionate
[sic],
1,1,3-tris(2-methyl-4-hydroxy-5-tert-butylphenyl)butane,
45 1,3,5-trimethyl-2,4,6-tris(3,5-di-tert-butyl-4-hydroxybenzyl)-
benzene,
1,3,5-tris(3,5-di-tert-butyl-4-hydroxybenzyl) isocyanurate,

- 1,3,5-tris[β -(3,5-di-tert-butyl-4-hydroxyphenyl) propionylethyl] isocyanurate,
 1,3,5-tris(2,6-dimethyl-3-hydroxy-4-tert-butylbenzyl) isocyanurate and pentaerythritol
 5 tetrakis[β -(3,5-di-tert-butyl-4-hydroxy)propionate] [sic],

 phosphorous-containing antioxidants such as tris(nonylphenyl) phosphite, distearyl pentaerythritol phosphite [sic],
 tris(2,4-di-tert-butylphenyl) phosphite,
 10 tris(2-tert-butyl-4-methylphenyl) phosphite,
 bis(2,4-di-tert-butylphenyl) pentaerythritol diphosphite and
 tetrakis(2,4-di-tert-butylphenyl) 4,4'-biphenylene diphosphite,

 sulfur-containing antioxidants such as
 15 dilauryl thiodipropionate,
 dimyristyl thiodipropionate,
 distearyl thiodipropionate,
 pentaerythritol tetrakis(β -laurylthiopropionate) and
 pentaerythritol tetrakis(β -hexylthiopropionate),
 20
 sterically hindered amines such as
 bis(2,2,6,6-tetramethylpiperidyl) sebacate,
 bis(1,2,2,6,6-pentamethylpiperidyl) sebacate,
 bis(1,2,2,6,6-pentamethylpiperidyl) esters,
 25 N,N'-bis(formyl)-bis(2,2,6,6-tetramethyl-4-piperidyl)-1,6-hexane-
 diamine,
 the condensate of
 1-hydroxy-2,2,6,6-tetramethyl-4-hydroxypiperidine and succinic
 acid,
 30 the condensate of
 N,N'-(2,2,6,6-tetramethylpiperidyl)hexamethylenediamine and
 4-tert-octylamino-2,6-dichloro-1,3,5-s-triazine,
 poly[3-(eicosyl/tetracosyl)-1-(2,2,6,6-tetramethyl-4-piperidinyl)-
 -2,5-pyrrolidinedione],
 35 tris(2,2,6,6-tetramethylpiperidyl) nitrilotriacetate,
 tetrakis(2,2,6,6-tetramethyl-4-piperidyl)-
 1,2,3,4-butanetetracarboxylic acid [sic],
 1,1'-(1,2-ethanediyl)bis(3,3,5,5-tetramethylpiperazinone),
 the condensates of
 40 4-amino-2,2,6,6-tetramethylpiperidines and
 tetramethylolacetylenediureas, and

 2-(2'-hydroxyphenyl)benzotriazoles,
 2-hydroxybenzophenones,
 45 aryl esters of hydroxybenzoic acids,
 α -cyanocinnamic acid derivatives,

nickel compounds or
oxanilides.

The compounds I according to the invention can be mixed, in
5 particular with plastics, using all known apparatus and methods
for mixing stabilizers or other additives into polymers.

The 2-cyanoacrylic esters I according to the invention are dis-
tinguished by high compatibility with conventional types of
10 plastic and by good solubility and excellent compatibility in
conventional coating systems. As a rule, they have very little or
no intrinsic color, are stable and nonvolatile at conventional
plastic- and surface coating-processing temperatures and afford
long-lasting protection to the materials treated with them. Above
15 all, however, they show virtually no tendency to migrate in
plastics.

UV radiation is divided into three regions: the UV-A region
(320-400 nm), the UV-B region (290-320 nm) and the UV-C region
20 (200-290 nm). The high-energy UV-C region is predominantly
absorbed by the ozone layer. Radiation in the UV-B region is
responsible in particular for the development of sunburn and skin
cancer. UV-A radiation produces on lengthy exposure tanning of
the skin but is also partly responsible for aging of the skin.

25 Because of the favorable solubility properties and the good
absorption properties, especially in the UV-A region, the
compounds according to the invention are particularly suitable
for applications in cosmetics and dermatological products. The
30 compounds can also be used advantageously for protecting cosmetic
products such as perfumes, creams and lotions. Combinations with
sunscreen agents which absorb in the UV-B region are particularly
preferred. The 2-cyanoacrylic esters I are used for cosmetic
formulations in concentrations of from 0.05 to 15%, preferably
35 from 0.1 to 10%, of the total weight of the cosmetic formulation.

Other organic materials to which the compounds according to the
invention can advantageously be added are pharmaceutical
formulations such as pills and suppositories, photographic
40 recording materials, especially photographic emulsions, and
precursors for plastics and paints.

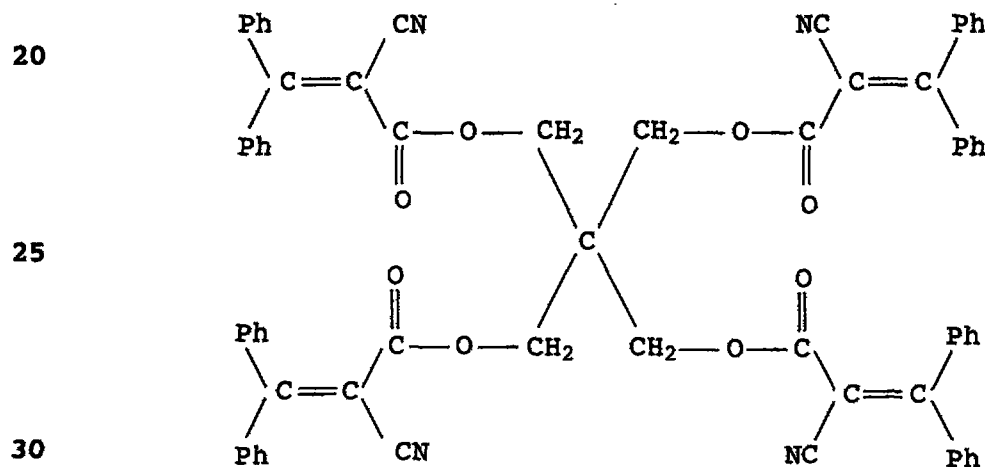
Examples

Preparation Examples

5 Example 1

16.2 g (0.04 mol) of 2,2-bis(hydroxymethyl)-1,3-propanediol tetracyanoacetate were dissolved in 100 ml of N,N-dimethylformamide (DMF) and heated to 80°C. To this were added dropwise under a gentle stream of nitrogen 29.6 g (0.16 mol) of benzophenone imine (97% pure) dissolved in 25 ml of DMF, over the course of 2 h. The mixture was heated at about 100°C until ammonia evolution ceased. It was then cooled and 300 ml of ethanol were added. The product was initially oily and became solid after lengthy stirring. It was filtered off with suction and washed with ethanol.

37.5 g (88.4%) of theory of the compound of the formula



were obtained with melting point 123–126°C (glassy); UV (CH₂Cl₂): λ_{max} = 310 nm, ε = 50,000.

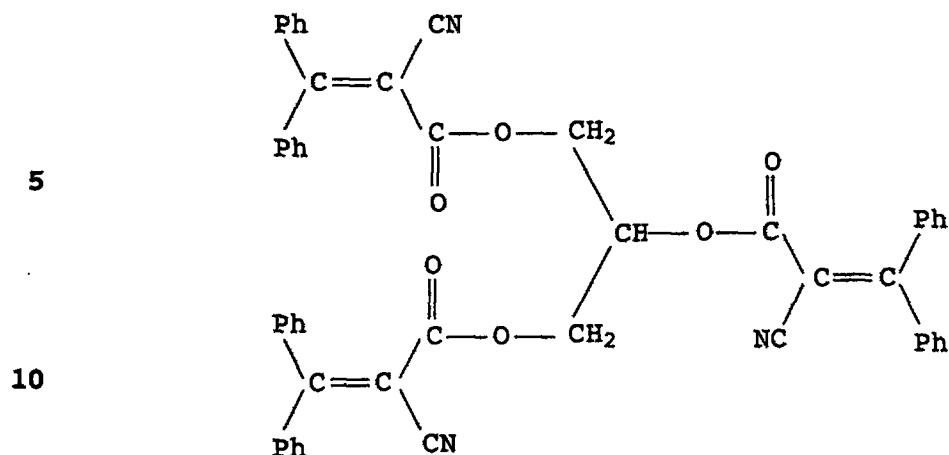
35 Example 2

The compound of the formula

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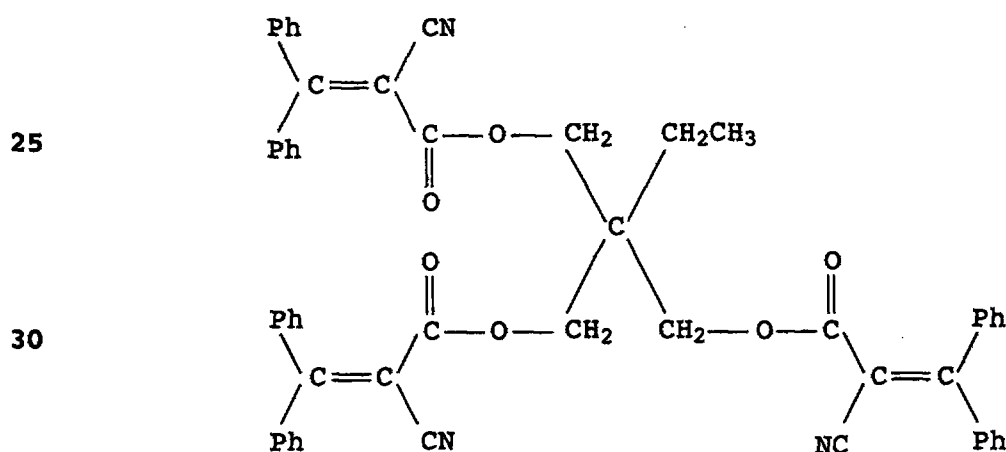
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was prepared in a similar way to Example 1 from the appropriate
 15 cyanoacetic ester and benzophenone imine; melting point:
 100–104°C; UV (CH₂Cl₂): λ_{max} = 310 nm, ϵ = 36,400.

Example 3

20 The compound of the formula



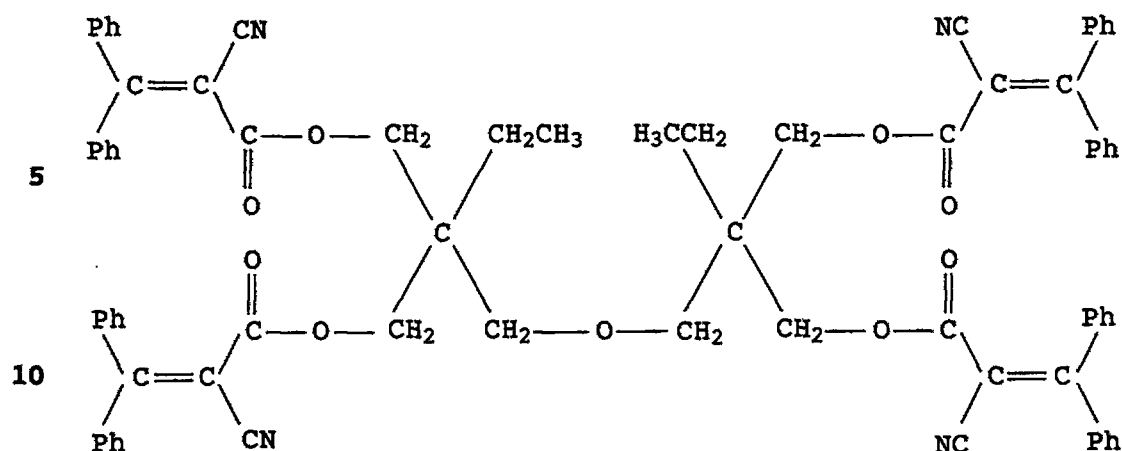
35 was prepared in a similar way to Example 1 from the appropriate
 cyano [sic] ester and benzophenone imine; melting point: 92°C;
 UV (CH₂Cl₂): λ_{max} = 308 nm, ϵ = 36,700.

Example 4

40

The compound of the formula

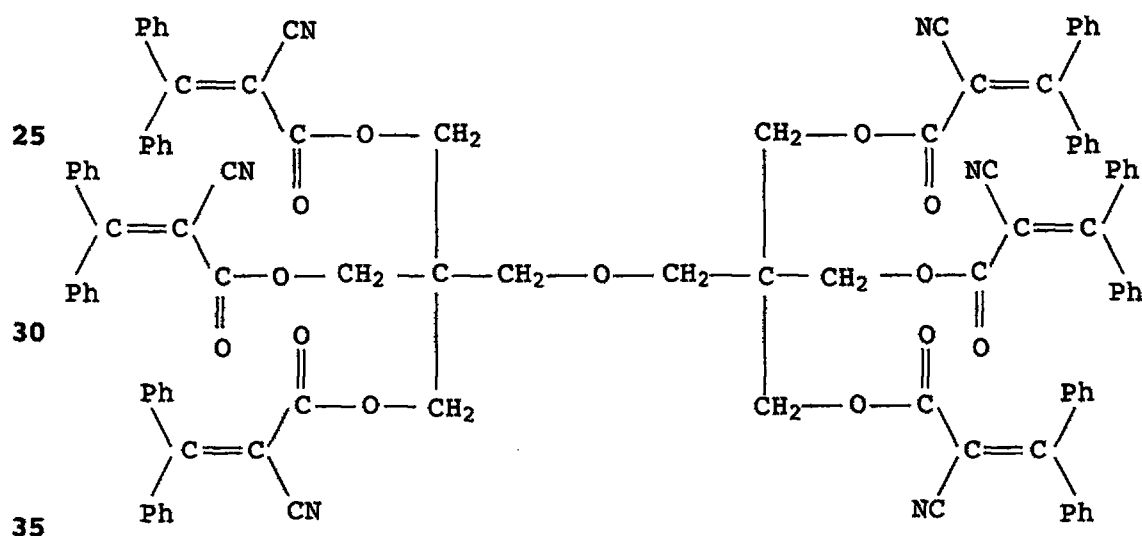
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was prepared in a similar way to Example 1 from the appropriate
 15 cyanoacetic ester and benzophenone imine; melting point: 83–95°C;
 UV (CH₂Cl₂): λ_{max} = 308 nm, ϵ = 51,700.

Example 5

20 The compound of the formula



was prepared in a similar way to Example 1 from the appropriate
 cyanoacetic ester and benzophenone imine; melting point 124–128°C;
 UV (CH₂Cl₂): λ_{max} = 308 nm, ϵ = 76,000.

40

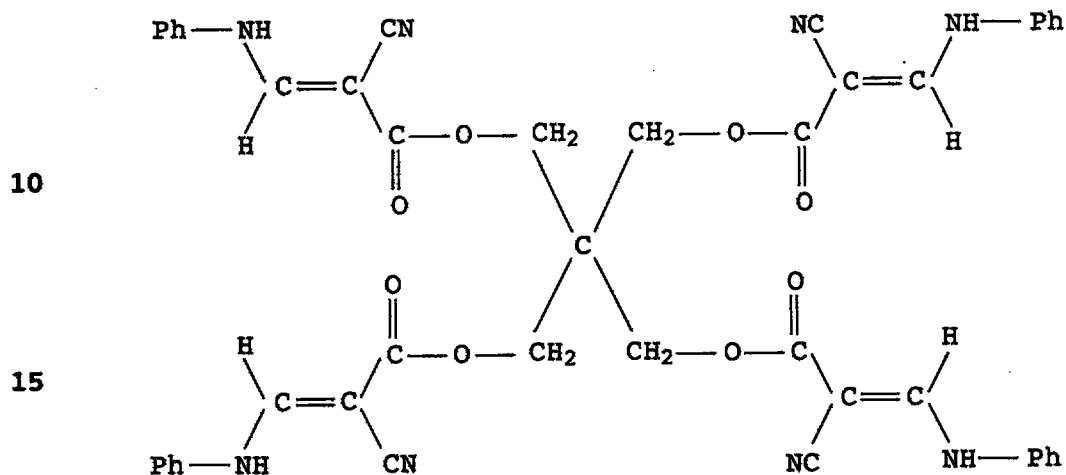
Example 6

30.3 g (0.075 mol) of 2,2-bis(hydroxymethyl)-1,3-propanediol
 tetracyanoacetate were refluxed with 29.8 g (0.32 mol) of aniline
 45 and 52 g (0.35 mol) of trimethyl orthoformate for 6 h. Then 80 ml
 of ethanol were added, and the suspension was refluxed for 1 h.

It was then filtered while hot under suction and the residue was thoroughly washed with ethanol.

55 g (90% of theory) of a yellowish compound of the formula

5

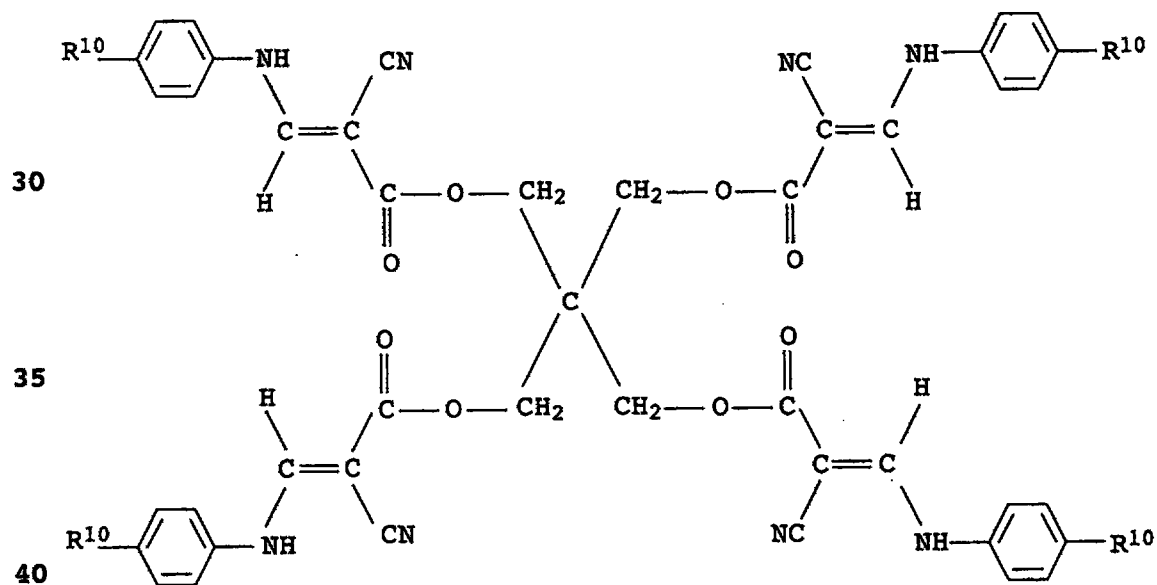


were obtained with melting point 298–300°C; UV (DMSO):
 20 $\lambda_{\text{max}} = 322 \text{ nm}$, $\epsilon = 98,000$ (DMSO = dimethyl sulfoxide).

Examples 7 and 8

The compound of the formula

25



$R^{10} = \text{CH}_3$ (Example 7) or $\text{COOCH}_2\text{CH}_3$ (Example 8)

45 were [sic] prepared in a similar way to Example 6 from the appropriate cyanoacetic ester, the appropriate aromatic amine and trimethyl orthoformate; melting points: 321–323°C (Example 7) and

15

269–273°C (Example 8); UV (DMSO): λ_{max} = 326 nm (Example 7) and 334 nm (Example 8), ϵ = 99,000 (Example 7) and 150,000 (Example 8).

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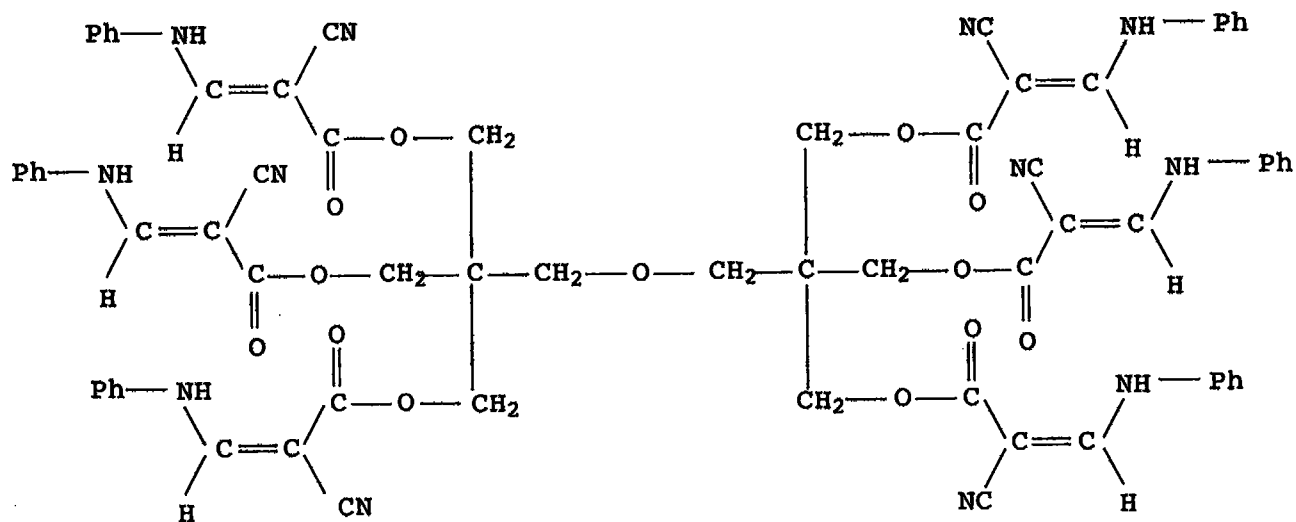
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Example 9

The compound of the formula



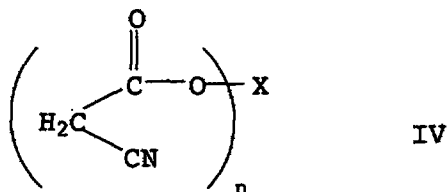
was prepared in a similar way to Example 6 from the appropriate cyanoacetic ester, aniline and tri-methyl orthoformate; melting point 240-248°C; UV (CH₂Cl₂): λ_{max} = 320 nm; ε = 145,000.

Examples 10-36

General preparation method for the reaction of cyanoacetic esters
5 IV with aldehydes (R^1 or R^2 = hydrogen)

0.1 mol of an n-functional cyanoacetic ester IV,

10

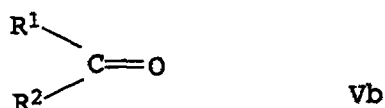


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which has been obtained by reacting cyanoacetic acid with the
appropriate n-hydric alcohol in a conventional way,

were [sic] reacted with 0.12 n mol of an aldehyde Vb

20



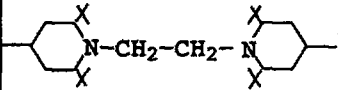
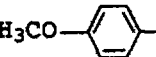
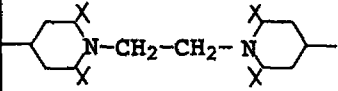
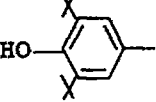
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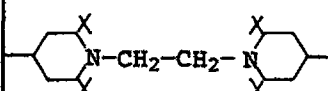
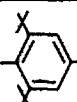
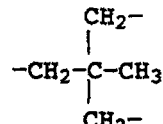
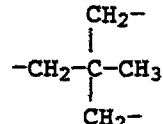
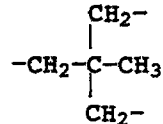
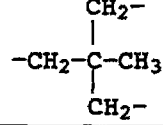
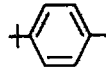
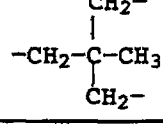
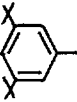
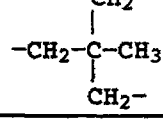
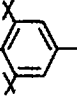
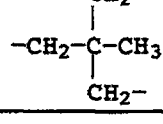
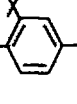
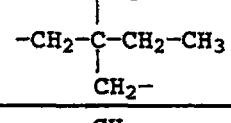
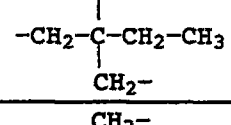
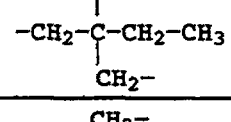
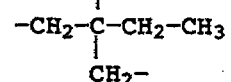
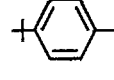
in 100 ml of N,N-dimethylacetamide [sic] in the presence of
0.5 ml of piperidine and 0.3 ml of glacial acetic acid. After
3 hours at 70°C, the precipitate was separated off, washed with
methanol and water and dried.

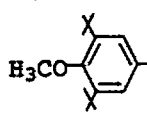
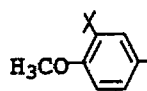
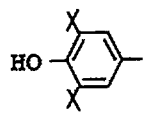
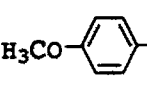
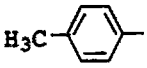
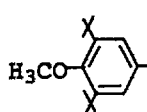
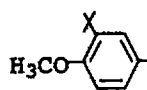
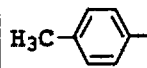
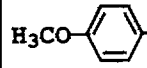
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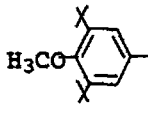
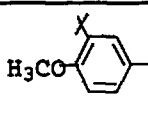
Details of these experiments and the properties of the com-
pounds I obtained are to be found in the following table.

35

No.	X	R^1 or R^2	* λ_{max} [nm]	Molar extinction coefficient ϵ [l·cm ⁻¹ ·mol ⁻¹]	Melting point [°C]	Yield [%]
40 10			342	57 000	>265	95
45 11			350	59 000	>265	70

No.	X	R ¹ or R ²	* λ _{max} [nm]	Molar extinction coefficient ε [l·cm ⁻¹ ·mol ⁻¹]	Melting point [°C]	Yield [%]
5		H ₃ CO- 	336	47 000	>265	92
10			306	59 188	110-112	70
15		H ₃ C- 	322	66 678	115-120	77
20		H ₃ CO- 	346	76 912	75-80	90
25			324	73 332	90-95	84
30		H ₃ CO- 	340	72 000	179-181	70
35		HO- 	353	72 000	170-174	77
40		H ₃ CO- 	354	72 100	95-100	88
45			306	58 256	114-116	63
		H ₃ C- 	322	67 090	95-102	74
		H ₃ CO- 	346	75 519	30-35	73
			322	57 601	168-170	67

No.	X	R ¹ or R ²	* $\lambda_{\max}$ [nm]	Molar extinction coefficient ϵ [l·cm ⁻¹ ·mol ⁻¹]	Melting point [°C]	Yield [%]
5						
24	$\begin{array}{c} \text{CH}_2^- \\ \\ -\text{CH}_2-\text{C}-\text{CH}_2-\text{CH}_3 \\ \\ \text{CH}_2^- \end{array}$		338	68 000	103-105	74
10						
25	$\begin{array}{c} \text{CH}_2^- \\ \\ -\text{CH}_2-\text{C}-\text{CH}_2-\text{CH}_3 \\ \\ \text{CH}_2^- \end{array}$		354	72 000	85-87	74
15						
26	$\begin{array}{c} \text{CH}_2^- \\ \\ -\text{CH}_2-\text{C}-\text{CH}_2- \\ \\ \text{CH}_2^- \end{array}$		358	106 480	275-276	66
20						
27	$\begin{array}{c} \text{CH}_2^- \\ \\ -\text{CH}_2-\text{C}-\text{CH}_2- \\ \\ \text{CH}_2^- \end{array}$		346	102 298	215-216	90
20						
28	$\begin{array}{c} \text{CH}_2^- \\ \\ -\text{CH}_2-\text{C}-\text{CH}_2- \\ \\ \text{CH}_2^- \end{array}$		308	63 909	148-155	79
25						
29	$\begin{array}{c} \text{CH}_2^- \\ \\ -\text{CH}_2-\text{C}-\text{CH}_2- \\ \\ \text{CH}_2^- \end{array}$		324	102 273	250	79
30						
30	$\begin{array}{c} \text{CH}_2^- \\ \\ -\text{CH}_2-\text{C}-\text{CH}_2- \\ \\ \text{CH}_2^- \end{array}$		324	101 131	130-131	67
30						
31	$\begin{array}{c} \text{CH}_2^- \\ \\ -\text{CH}_2-\text{C}-\text{CH}_2- \\ \\ \text{CH}_2^- \end{array}$		342	51 000	98-100	60
35						
32	$\begin{array}{c} \text{CH}_2^- \\ \\ -\text{CH}_2-\text{C}-\text{CH}_2- \\ \\ \text{CH}_2^- \end{array}$		356	110 500	115-118	87
40						
33	$\begin{array}{c} \text{CH}_2^- \quad \text{CH}_2^- \\ \quad \quad \\ -\text{CH}_2-\text{C}-\text{CH}_2-\text{O}-\text{CH}_2-\text{C}-\text{CH}_2- \\ \quad \quad \\ \text{CH}_2^- \quad \text{CH}_2^- \end{array}$		320	120 582	128-132	65
34	$\begin{array}{c} \text{CH}_2^- \quad \text{CH}_2^- \\ \quad \quad \\ -\text{CH}_2-\text{C}-\text{CH}_2-\text{O}-\text{CH}_2-\text{C}-\text{CH}_2- \\ \quad \quad \\ \text{CH}_2^- \quad \text{CH}_2^- \end{array}$		342	145 000	105-108	88

No.	X	R ¹ or R ²	* $\lambda_{\max}$ [nm]	Molar extinction coefficient ϵ [l·cm ⁻¹ ·mol ⁻¹]	Melting point [°C]	Yield [%]
5 35	$\begin{array}{c} \text{CH}_2- \\ \\ -\text{CH}_2-\text{C}-\text{CH}_2-\text{O}-\text{CH}_2-\text{C}-\text{CH}_2- \\ \\ \text{CH}_2- \end{array}$		338	149 300	150-151	58
10 36	$\begin{array}{c} \text{CH}_2- \\ \\ -\text{CH}_2-\text{C}-\text{CH}_2-\text{O}-\text{CH}_2-\text{C}-\text{CH}_2- \\ \\ \text{CH}_2- \end{array}$		352	145 000	135-140	51

* UV measurements in CH₂Cl₂

15 Example 37

Use Example: Migration test in polyethylene

0.3% by weight of the UV stabilizer indicated below was dissolved
20 in polyethylene by extrusion twice at a polymer temperature of
180°C, and then the polymer was granulated and blown to films
100 µm thick.

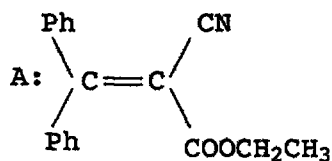
After storage at room temperature (20°C) or in an oven (50°C) for
25 ten days, the surface of the film was assessed visually according
to the following criteria:

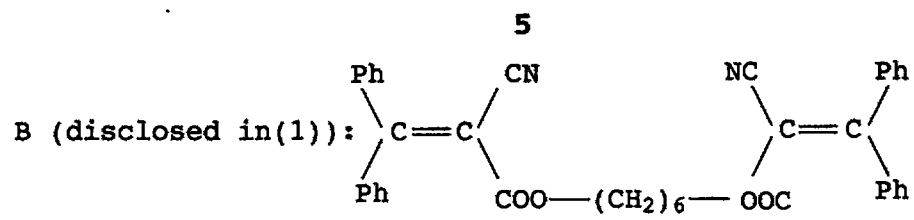
- 30
- + no deposit
 - o slight deposit
 - heavy deposit

The following table shows the UV stabilizers used and the results
of the tests:

35	UV stabilizer	Storage at 20°C	Storage at 50°C
	Compound from Example No. 1	+	+
	Compound A (for comparison)	o	-
	Compound B (for comparison)	-	-

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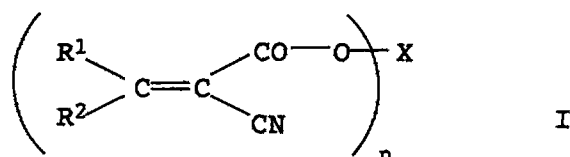
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We claim:

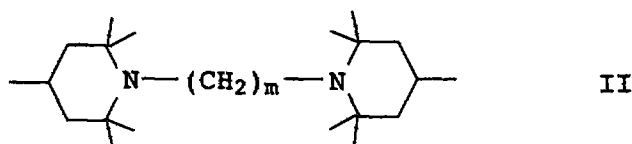
1. A 2-cyanoacrylic ester of the formula I



where R^1 and R^2 are each hydrogen or a radical having an iso- or heterocyclic ring system with at least one iso- or hetero- aromatic nucleus, and at least one of the radicals R^1 or R^2 must be different from hydrogen,

n is from 2 to 10, and

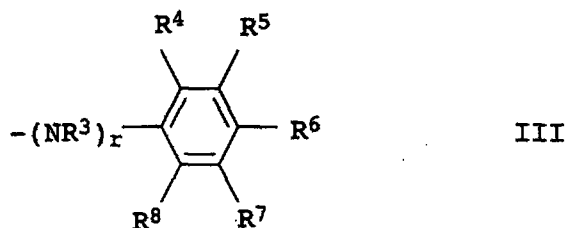
X is, when $n = 2$, a radical of the formula II



where m is from 2 to 8, and

X is, when $n > 2$, the radical of an n -hydric aliphatic or cycloaliphatic polyol having 3-20 carbon atoms, it also being possible for a cycloaliphatic radical to contain 1 or 2 hetero atoms, and for an aliphatic radical to be interrupted by up to 8 non-adjacent oxygen atoms, sulfur atoms, imino or C_1 - C_4 -alkylimino groups.

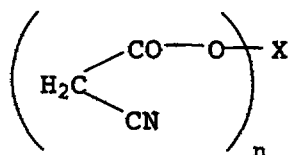
2. A 2-cyanoacrylic ester as claimed in claim 1, wherein one of the radicals R^1 and R^2 is a radical of the formula III



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where R³ is hydrogen or C₁-C₁₀-alkyl, r is 0 or 1, and R⁴ to R⁸ are each, independently of one another, hydrogen, C₁-C₈-alkyl, chlorine, bromine, cyano, nitro, amino, mono-(C₁-C₄-alkyl)amino, di(C₁-C₄-alkyl)amino, hydroxyl, C₁-C₈-acyl, C₁-C₈-acyloxy, C₁-C₁₈-alkoxy, C₁-C₁₂-alkoxycarbonyl, C₃-C₆-cycloalkyl or C₃-C₆-cycloalkoxycarbonyl.

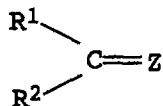
3. A 2-cyanoacrylic ester as claimed in claim 1 or 2, wherein R³ is hydrogen, methyl or ethyl.
4. A 2-cyanoacrylic ester as claimed in claims 1 and 2, wherein up to three of the radicals R⁴ to R⁸ are hydrogen, C₁-C₄-alkyl, chlorine, cyano, hydroxyl, acetyl, C₁-C₅-alkoxy, C₁-C₈-alkoxycarbonyl or cyclohexoxycarbonyl, and the remainder of these radicals are hydrogen.
5. A 2-cyanoacrylic ester as claimed in claims 1 to 4, wherein R⁶ is a hydroxyl group or a C₁-C₄-alkoxy group.
6. A 2-cyanoacrylic ester as claimed in claims 1 to 5, wherein R⁵ and/or R⁷ are hydrogen, methyl or tert-butyl.
7. A 2-cyanoacrylic ester as claimed in claims 1 to 6, wherein r is 0.
8. A 2-cyanoacrylic ester as claimed in claims 1 to 7, wherein X is the radical of an n-hydric polyol having 3 to 12 carbon atoms, which may be interrupted in its linear or branched carbon skeleton by up to 3 non-adjacent oxygen atoms, and n is from 3 to 6.
9. A process for preparing 2-cyanoacrylic esters as claimed in claims 1 to 8, where r is 0, which comprises reacting a cyanoacetic ester of the formula IV



IV



with n mol of a compound of the formula V



V

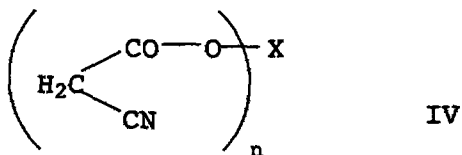
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where Z is oxygen or NH, under the conditions of the Knoevenagel condensation in a polar solvent and in the presence of a catalyst.

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10. A process for preparing 2-cyanoacrylic esters as claimed in claims 2 to 6 and 8, where r is 1, which comprises reacting a cyanoacetic ester of the formula IV

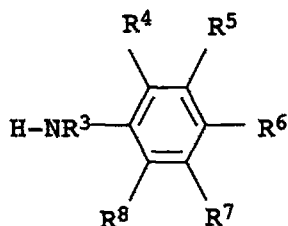
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with an aromatic amine of the formula Va

25



Va



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in the presence of trialkyl orthoformate.

11. The use of the 2-cyanoacrylic esters as claimed in claims 1 to 8 as stabilizers, in particular against the action of light, for organic materials.

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12. The use of the 2-cyanoacrylic esters as claimed in claims 1 to 8 as stabilizers, in particular against the action of light, in cosmetic or dermatological preparations.

- 40 13. The use of the 2-cyanoacrylic esters as claimed in claims 1 to 8 as stabilizers, in particular against the action of light, in plastics or paints.

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14. Organic material stabilized against the action of light, oxygen and heat, which comprises from 0.01 to 10% by weight, based on the amount of organic material, of one or more 2-cyanoacrylic esters as claimed in claims 1 to 8.

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15. A cosmetic or dermatological preparation stabilized against the action of light, oxygen and heat, which comprises from 0.01 to 15% by weight, based on the amount of this preparation, of one or more 2-cyanoacrylic esters as claimed in claims 1 to 8.

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16. A plastic or paint stabilized against the action of light, oxygen and heat, which comprises from 0.01 to 10% by weight, based on the amount of the plastic or paint, of one or more 2-cyanoacrylic esters as claimed in claims 1 to 8.

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