

Method for producing a tridimensional structure using two pre-supporting materials

Description

5 The present invention relates to a process for producing a three-dimensional structure by three-dimensional printing using a support material which allows exact contours and a smooth surface to be obtained and allows complete and simple removal of the support material.

10 Three-dimensional (3D) printing processes are usually employed for rapid prototyping. This is a one-piece production of structures as a result of which later joining of individual parts in order to produce complex structures can be dispensed with.

The first 3D printing process was developed in 1984 by Chuck Hull who referred to his
15 method as stereolithography, SL for short. In the SL method, a light-curing polymer applied as layer over an area in one plane is cured positionally selectively by a laser. The procedure is carried out in a bath which is filled with a liquid or paste-like base monomer of the light-sensitive polymer. The structural regions which are formed positionally selectively as a result of initiation by means of laser light in the working
20 plane are, in a next step, moved downward into the bath by one layer thickness, so that another polymer layer can be formed in the working plane above the cured structural regions.

For example, DE 100 24 618 A1 discloses such a stereolithographic process for
25 producing three-dimensional structures, in which liquid to gel-like silicone rubbers are irradiated with IR lasers. US 2009/0224438 describes the layer-wise processing of 3D objects by means of SL processes using materials which can be photocrosslinked by means of UV or Vis light.

30 SL processes have, inter alia, the disadvantage that only a single photocrosslinkable material can be used for building up a three-dimensional structure. Limits in respect of the elastic structural properties, for example, are also imposed thereby.

A further 3D printing process was developed at the Massachusetts Institute of
35 Technology. Here, for example, pulverulent polymers are applied in layers to a support plate. Binders are squirted by means of an ink jet printer onto the places which are to be solidified in each layer. Such a process is described, for example, in EP 0 431 924 B1 and US 5,387,380.

Furthermore, the photopolymer jetting process (also known as PolyJet process) can be used for producing three-dimensional structures. Here, a printing block having one or more printing heads moves back and forth in the manner of a line printer along an x axis and leaves behind thin photopolymer layers on a building platform. Each layer is
5 cured immediately after application by means of UV lamps which are installed directly on the printing block.

Photopolymer jetting allows fabrication of three-dimensional parts with a high degree of geometric freedom and variable materials properties, for example elasticity, using,
10 optionally, a plurality of different materials.

Thus, US 6,658,314 B1 describes a process for producing a three-dimensional structure, in which two photopolymers are mixed in different ratios in order to influence the elasticity of the three-dimensional structure in a targeted manner.

15 However, the production of overhanging structures and hollow spaces is not readily possible since, owing to the layer-wise construction, overhanging constituents and structures above hollow spaces would not be joined to the layer underneath. However, such structures can be stabilized by application of not only the object material but also
20 a support material which can usually be removed in an aqueous medium after manufacture. (See 55148)

Thus, for example, US 6,863,859 B2 describes a composition which comprises a heat-sensitive polymer and is suitable as support material and a process for producing
25 three-dimensional structures using the support material composition, in which the support material is removed in an aqueous medium after production.

US 2010/0256255 A1 describes support materials which comprise at least one dendritic oligomer, at least one monofunctional monomer and a reactive amine.

30 WO 2012/116047 A1 describes support materials which comprise at least one ethoxylated fatty alcohol and the use thereof in 3D printing processes.

WO 01/68375 A2 describes support materials which comprise at least one reactive component such as acrylates or vinyl ethers and a photoinitiator, and the use thereof in
35 3D printing processes.

It is usual for both the object material and the support material to be free-radically crosslinkable or polymerizable. As a result, mixed products can be formed in regions in which the object material and support material bound on one another. This can result in

unsharp boundaries between object material and support material and the component has to be, for example, mechanically after-treated in order to remove support material residues and smooth the surface of the three-dimensional structure.

5 It is therefore an object of the present invention to provide a process for producing a three-dimensional structure by three-dimensional printing using a support material which allows exact contours and a smooth surface to be obtained and allows complete and simple removal of the support material.

10 The object is achieved by a process for producing a three-dimensional structure by means of three-dimensional printing, which comprises the following steps:

a) ejection of an object material from a first printing head;

15 b) ejection of a first pre-support material from a second printing head; and

b') ejection of a second pre-support material from a third printing head,

20 where the pre-support materials comprise liquid components which on meeting form a solid support material;

c) radiation curing of the object material;

25 where the steps a) to c) are repeated a plurality of times in order to form the three-dimensional structure in a layerwise manner and the object material and the support material have at least one common interface; and

d) removal of the solidified support material by treatment with an aqueous medium.

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The support material is produced from two liquid components which become solid when they meet. The formation of the support structure does not interfere with formation of the framework structure. A reaction between support material and object material occurs to at least a greatly reduced extent and is preferably prevented completely. As a result, subsequent treatment of the surface of the three-dimensional structure to remove residues of the support material is unnecessary or necessary only to a greatly reduced extent.

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It is also possible to eject the support material in the form of a separating layer. The actual support is then effected by a secondary support material. In one embodiment, the process therefore comprises the following steps:

5 b) ejection of a first pre-support material from a second printing head; and

 b') ejection of a second pre-support material from a third printing head,

 where the pre-support materials comprise liquid components which on meeting
10 become solid and form a support material in the form of a separating layer; and the
 process additionally comprises:

 b'') ejection of a secondary support material from a fourth printing head;

15 where the steps a), b), b'), b'') and c) are repeated a plurality of times in order to form
 the three-dimensional structure in a layerwise manner and the support material and the
 secondary support material have at least one common interface;

 d) removal of the solidified support material and the cured secondary support material
20 by treatment with an aqueous medium.

 The viscosity of the object material, the pre-support materials and, if present, the
 secondary support material is preferably less than 20 mPas at 70°C. The viscosity is
 usually from 8 to < 20 mPas at 70°C, particularly preferably from 8 to 15 mPas at 70°C.
25 Since droplet formation in the printing head is possible only up to a particular viscosity,
 a higher viscosity would be disadvantageous.

Object material

30 In one embodiment, the object material comprises at least one free-radically curable
 compound. The free-radically curable compound has at least one ethylenically
 unsaturated double bond. Suitable free-radically curable compounds comprise
 monofunctional compounds (compounds having one ethylenically unsaturated double
 bond), polyfunctional compounds (compounds having two or more ethylenically
35 unsaturated double bonds), including ethylenically unsaturated prepolymers. The free-
 radically curable compound preferably comprises at least one polyfunctional
 compound. Owing to their low viscosity, monofunctional compounds can be
 concomitantly used, e.g. as reactive diluents.

Examples of monomeric monofunctional compounds comprise (meth)acrylic compounds and vinyl compounds.

(Meth)acrylic compounds include:

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- C₁-C₁₈-Alkyl (meth)acrylates, such as methyl acrylate, ethyl acrylate, butyl acrylate, 2-ethylhexyl acrylate, isobornyl acrylate, methyl methacrylate, ethyl methacrylate, butyl methacrylate; C₂-C₁₈-hydroxyalkyl (meth)acrylates, such as 2-hydroxyethyl (meth)acrylate, 2-hydroxyethyl methacrylate, 4-hydroxybutyl acrylate; C₁-C₁₀-alkoxy-C₂-C₁₈-alkyl (meth)acrylates, such as 2-methoxyethyl (meth)acrylate, 2-ethoxyethyl (meth)acrylate; aryloxyalkyl (meth)acrylates, such as phenoxyethyl methacrylate, p-cumylphenoxyethyl methacrylate; C₇-C₂₀-aralkyl (meth)acrylates, such as benzyl (meth)acrylate; C₂-C₇-heterocyclyl (meth)acrylates, such as tetrahydrofurfuryl (meth)acrylate; C₂-C₇-heterocyclyl-C₂-C₁₀-alkyl (meth)acrylates such as 2-N-morpholinoethyl (meth)acrylate, 2-(2-oxo-1-imidazolidinyl)ethyl (meth)acrylate; C₂-C₁₀-aminoalkyl (meth)acrylates such as 2-aminoethyl (meth)acrylate; mono- or di-C₁-C₁₀-alkyl-C₂-C₁₀-aminoalkyl (meth)acrylates such as 2-(dimethylamino)ethyl (meth)acrylate, 2-(diethylamino)ethyl (meth)acrylate, 2-(diisopropylamino)ethyl (meth)acrylate, 2-(tert-butylamino)ethyl (meth)acrylate; (meth)acrylonitrile; (meth)acrylamide; C₂-C₃₀-alkyl(meth)acrylamides such as N-n-butyl(meth)acrylamide, N-tert-butyl(meth)acrylamide, N-octyl(meth)acrylamide, N-lauryl(meth)acrylamide, N-1-methylundecyl(meth)acrylamide, N-2-ethylhexyl(meth)acrylamide and N-tert-octyl(meth)acrylamide; C₂-C₁₈-hydroxyalkyl(meth)acrylamides such as N-hydroxyethyl(meth)acrylamide, N-[tris(hydroxymethyl)methyl](meth)acrylamide; C₁-C₁₀-alkoxy-C₁-C₁₈-alkyl(meth)acrylamides such as N-(3-methoxypropyl)(meth)acrylamide, N-(butoxymethyl)(meth)acrylamide, N-(isobutoxymethyl)(meth)acrylamide; C₂-C₇-heterocyclyl(meth)acrylamides such as N-tetrahydrofurfuryl(meth)acrylamide; C₂-C₇-heterocyclyl-C₂-C₁₀-alkyl(meth)acrylamides such as N-(2-N-morpholinoethyl)(meth)acrylamide, N-(2-(2-oxo-1-imidazolidinyl)ethyl)(meth)acrylamide; C₂-C₁₀-aminoalkyl(meth)acrylamides such as N-(3-aminopropyl)(meth)acrylamide; and mono- or di-C₁-C₁₀-alkyl-C₂-C₁₀-aminoalkyl(meth)acrylamides such as N-[2-(dimethylamino)ethyl](meth)acrylamide, N-[3-(dimethylamino)propyl](meth)acrylamide.

Among these, N-branched alkyl(meth)acrylamides such as N-tert-butyl(meth)acrylamide, N-tert-octyl(meth)acrylamide and N-2-ethylhexyl(meth)acrylamide are particularly preferred.

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The vinyl compounds include vinyl esters such as vinyl acetate; N-vinylamides such as N-vinylpyrrolidone; vinylaromatics such as styrene, alkylstyrenes and halostyrenes; and vinyl halides such as vinyl chloride and vinylidene chloride.

- 5 Examples of polyfunctional compounds comprise, for example, esters of polyols with ethylenically unsaturated carboxylic acids such as acrylic acid, methacrylic acid, crotonic acid, itaconic acid, cinnamic acid, maleic acid, fumaric acid and unsaturated fatty acids such as linoleic acid, linolenic acid or oleic acid. Preference is given to acrylic acid and methacrylic acid.

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Suitable polyols comprise aromatic and in particular aliphatic and cycloaliphatic polyols. Examples of aromatic polyols comprise hydroquinone, 4,4'-dihydroxybiphenyl, 2,2-di(4-hydroxyphenyl)propane and also novolacs and resols. Examples of aliphatic and cycloaliphatic polyols comprise alkylenediols which preferably have from 2 to 12 carbon
15 atoms, e.g. ethylene glycol, 1,2- or 1,3-propanediol, 1,2-, 1,3- or 1,4-butanediol, pentanediol, hexanediol, octanediol, dodecanediol, diethylene glycol, triethylene glycol, polyethylene glycols having preferred molar masses of from 200 to 1500 g/mol, 1,3-cyclopentanediol, 1,2-, 1,3- or 1,4-cyclohexanediol, 1,4-dihydroxymethylcyclohexane, glycerol, tris(β -hydroxyethyl)amine, trimethylolethane, trimethylolpropane,
20 pentaerythritol, dipentaerythritol and sorbitol. Further suitable polyols comprise, for example, polymers and copolymers which comprise hydroxyl groups in the polymer chain or in side chains, e.g. polyvinyl alcohol and copolymers thereof, and also polyhydroxyalkyl methacrylates and copolymers thereof. Oligoesters comprising hydroxyl end groups are likewise suitable polyols.

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The polyols can be partly or fully esterified with one or more different unsaturated carboxylic acids, with the free hydroxyl groups in partial esters possibly having been modified, e.g. etherified or esterified with other carboxylic acids.

- 30 Examples of esters comprise trimethylolpropane triacrylate, trimethylolethane triacrylate, trimethylolpropane trimethacrylate, trimethylolethane trimethacrylate, tetramethylene glycol dimethacrylate, triethylene glycol dimethacrylate, tetraethylene glycol diacrylate, pentaerythritol diacrylate, pentaerythritol triacrylate, pentaerythritol tetraacrylate, dipentaerythritol diacrylate, dipentaerythritol triacrylate, dipentaerythritol
35 tetraacrylate, dipentaerythritol pentaacrylate, dipentaerythritol hexaacrylate, tripentaerythritol octaacrylate, pentaerythritol dimethacrylate, pentaerythritol trimethacrylate, dipentaerythritol dimethacrylate, dipentaerythritol tetramethacrylate, tripentaerythritol octamethacrylate, pentaerythritol diitaconate, dipentaerythritol trisitaconate, dipentaerythritol pentaitaconate, dipentaerythritol hexaitaconate, ethylene

- glycol diacrylate, hexamethylene glycol diacrylate, 1,3-butanediol diacrylate, 1,3-butanediol dimethacrylate, 1,4-butanediol diitaconate, sorbitol triacrylate, sorbitol tetraacrylate, modified pentaerythritol triacrylate, sorbitol tetramethacrylate, sorbitol pentaacrylate, sorbitol hexaacrylate, oligoester acrylates and methacrylates, bisphenol
- 5 A diacrylates, glycerol diacrylates and triacrylates, 1,4-cyclohexanediacylate, 4,4'-bis(2-acryloyloxyethoxy)diphenylpropane, bisacrylates and bismethacrylates of polyethylene glycol having a molar mass in the range from 200 to 1500 g/mol, or mixtures thereof.
- 10 Further suitable esters comprise dipropylene glycol diacrylate, tripropylene glycol diacrylate, 1,6-hexanediol diacrylate, ethoxylated glycerol triacrylate, propoxylated glycerol triacrylate, ethoxylated trimethylolpropane triacrylate, propoxylated trimethylolpropane triacrylate, ethoxylated pentaerythritol tetraacrylate, propoxylated pentaerythritol triacrylate, propoxylated pentaerythritol tetraacrylate, ethoxylated
- 15 neopentyl glycol diacrylate, propoxylated neopentyl glycol diacrylate.

- Further examples of polyfunctional compounds comprise amides of the above or other unsaturated carboxylic acids with aromatic, cycloaliphatic and aliphatic polyamines which preferably have from 2 to 6, particularly preferably from 2 to 4, amino groups.
- 20 Examples of such polyamines comprise ethylenediamine, 1,2- or 1,3-propylenediamine, 1,2-, 1,3- or 1,4-butylenediamine, 1,5-pentylenediamine, 1,6-hexylenediamine, octylenediamine, dodecylenediamine, 1,4-diaminocyclohexane, isophoronediamine, phenylenediamine, bisphenylenediamine, di- β -aminoethyl ether, diethylenetriamine, triethylenetetramine, di(β -aminoethoxy)ethane or di(β -
- 25 aminopropoxy)ethane. Further suitable polyamines are polymers and copolymers which may have further amino groups in the side chains, and oligoamides having terminal amino groups. Examples of such unsaturated amines comprise: methylenebisacrylamide, 1,6-hexamethylenebisacrylamide, diethylenetriaminetrismethacrylamide, bis(methacrylamidopropoxy)ethane,
- 30 β -methacrylamidoethyl methacrylate and N-[(β -hydroxyethoxy)ethyl]acrylamide.

- Further examples of polyfunctional compounds comprise vinyl acrylate, divinylbenzene, divinyl succinate, diallyl phthalate, triallyl phosphate, triallyl isocyanurate, tris(2-acryloyl-ethyl) isocyanurate and dicyclopentadienyl acrylate.
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Suitable prepolymers are polymers which comprise ethylenically unsaturated groups in the main chain or as side groups or are terminated therewith, e.g. unsaturated polyesters, polyamides and polyurethanes and copolymers thereof, polybutadiene and butadiene copolymers, polyisoprene and isoprene copolymers, polymers and

copolymers which comprise (meth)acrylate groups in the side chains, and also mixtures of one or more polymers of this type.

- 5 Suitable unsaturated polyesters and polyamides are prepared, for example, from maleic acid and diols or diamines. The polyesters and polyamides can also be prepared from dicarboxylic acids and ethylenically unsaturated diols or diamines, in particular relatively long-chain diols or diamines having, for example, from 6 to 20 carbon atoms. Further examples of polyesters comprise unsaturated polyester resins which are generally prepared from maleic acid, phthalic acid and one or more diols.
- 10 Suitable polyesters also comprise alkyd resins. Suitable polyamides are, for example, condensates of polyamines and unsaturated dimeric fatty acids.

- Examples of unsaturated polyurethanes comprise those which are prepared from saturated diisocyanates and unsaturated diols or from unsaturated diisocyanates and
- 15 saturated diols.

- Butadiene, which serves as monomer for polybutadiene or butadiene copolymers, generally polymerizes in such a way that an ethylenically unsaturated group remains as part of the main chain or side chain. The same applies to polyisoprene, which serves
- 20 as monomer for polyisoprene and isoprene copolymers. Examples of suitable comonomers comprise, in each case, olefins such as ethene, propene, butene, hexene, (meth)acrylates, acrylonitrile, styrene and vinyl chloride.

- Polymers which comprise (meth)acrylate groups in the side chain are known by those skilled in the art. They comprise, for example, reaction products of novolac-based epoxy resins and (meth)acrylic acid, homopolymers or copolymers of vinyl alcohol or the hydroxyalkyl derivatives thereof which have been esterified with (meth)acrylic acid, and homopolymers and copolymers of (meth)acrylates which have been esterified with hydroxyalkyl (meth)acrylates.

- 30 Copolymers which comprise (meth)acrylate groups as side groups can be obtained, for example, by functionalization of copolymers by means of (meth)acrylic acid. The functionalization of copolymers is preferably carried out using (meth)acrylic acid. In these compounds, the ethylenically unsaturated double bonds are preferably present in the form of (meth)acryloyl groups. Preference is given to at least two polymerizable
- 35 double bonds being present in the form of (meth)acryloyl groups in the molecule. The average molar mass of these compounds can, for example, be in the range from 300 to 10 000 g/mol, preferably in the range from 800 to 10 000 g/mol.

Prepolymers can also be terminated with ethylenically unsaturated compounds. For example, maleate-terminated oligomers having polyester, polyurethane, polyether and polyvinyl ether main chains are used. Further examples of ethylenically unsaturatedly terminated prepolymers comprise urethane (meth)acrylates, epoxy(meth)acrylates and acrylated epoxy resins.

Further particularly suitable compounds are urethane (meth)acrylates which can be obtained by reaction of polyisocyanates with hydroxyalkyl (meth)acrylates and optionally chain extenders such as diols, polyols, diamines, polyamines or dithiols or polythiols. These include urethane oligomers which bear terminal and/or lateral (meth)acrylic groups. Urethane oligomers are conventionally prepared by reaction of an aliphatic or aromatic diisocyanate with a divalent polyether or polyester, particularly typically a polyoxyalkylene glycol such as polyethylene glycol. Such oligomers typically have from 4 to 10 urethane groups. The isocyanate-terminated polyurethane polymer resulting from this reaction is then reacted with (meth)acrylic acid, a (meth)acrylamide or a (meth)acrylic ester having a hydroxyl group, in particular with a hydroxyalkyl (meth)acrylate such as hydroxypropyl acrylate (HPA), hydroxypropyl methacrylate (HPMA), hydroxybutyl acrylate (HBA) or hydroxybutyl methacrylate (HBMA), preferably with hydroxyethyl acrylate (HEA) or hydroxyethyl methacrylate (HEMA), or with a monohydroxy poly(meth)acrylate of a polyol, preferably of glycerol or trimethylolpropane, to give a polyurethane (meth)acrylate.

A suitable urethane (meth)acrylate is, for example, UDMA (an addition product of 2-hydroxyethyl methacrylate and 2,2,4-trimethylhexamethylene diisocyanate). Polyether urethane acrylate oligomers or polyester urethane acrylate oligomers are, for example, Ebecryl 284 and CN 982, CN 982B10 and CN 988 B88 from Sartomer.

Further particularly suitable compounds are epoxy (meth)acrylates which can be obtained by reaction of epoxides with (meth)acrylic acid. Possible epoxides are, for example, epoxidized olefins, aromatic glycidol ethers or aliphatic glycidol ethers, preferably those of aromatic or aliphatic glycidol ethers. A suitable epoxy (meth)acrylate is, for example, bis-GMA (an addition product of methacrylic acid and bisphenol A diglycidyl ether).

The free-radically curable compound is preferably present in the object material in an amount of at least 50% by weight, particularly preferably at least 70% by weight, based on the weight of the object material.

If the object material comprises at least one free-radically curable compound, it preferably also comprises a photoinitiator. Photoinitiators are photoactive substances which on illumination with UV light form free radicals and can thus initiate free-radical crosslinking or polymerization.

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As photoinitiators, it is possible to use photoinitiators known to those skilled in the art, e.g. those mentioned in "Advances in Polymer Science", Volume 14, Springer Berlin 1974 or K. K. Dietliker, Chemistry and Technology of UV and EB-Formulation for Coatings, Inks and Paints, Volume 3; Photoinitiators for Free Radical and Cationic Polymerization, P. K. T. Oldring (Eds), SITA Technology Ltd, London.

Possibilities are, for example, monoacylphosphine or bisacylphosphine oxides as described, for example, in EP-A 7 508, EP-A 57 474, DE-A 196 18 720, EP-A 495 751 or EP-A 615 980, for example 2,4,6-trimethylbenzoyldiphenylphosphine oxide (Lucirin TPO), bis(2,4,6-trimethylbenzoyl)phenylphosphine oxide (Irgacure 819), ethyl 2,4,6-trimethylbenzoylphenylphosphinate, benzophenones, hydroxyacetophenones, phenylglyoxylic acid and derivatives thereof or mixtures of these photoinitiators. As examples, mention may be made of benzophenone, acetophenone, acetonaphthoquinone, methyl ethyl ketone, valerophenone, hexanophenone, [alpha]-phenylbutyrophenone, p-morpholinopropiophenone, dibenzosuberone, 4-morpholinobenzophenone, 4-morpholinodeoxybenzoin, p-diacetylbenzene, 4-aminobenzophenone, 4'-methoxyacetophenone, beta-methylantraquinone, tert-butyl-antraquinone, anthraquinonecarboxylic esters, benzaldehyde, [alpha]-tetralone, 9-acetylphenanthrene, 2-acetylphenanthrene, 10-thioxanthene, 3-acetyl-phenanthrene, 3-acetylimidazole, 9-fluorenone, 1-indanone, 1,3,4-triacetylbenzene, thioxanthene-9-one, xanthene-9-one, 2,4-dimethylthioxanthone, 2,4-diethylthioxanthone, 2,4-diisopropylthioxanthone, 2,4-dichlorothioxanthone, benzoin, benzoin isobutyl ether, chloroxanthene, benzoin tetrahydropyranyl ether, benzoin methyl ether, benzoin ethyl ether, benzoin butyl ether, benzoin isopropyl ether, 7-H-benzoin methyl ether, benz[de]anthracene-7-one, 1-naphthaldehyde, 4,4'-bis(dimethylamino)benzophenone, 4-phenylbenzophenone, 2-chlorobenzophenone, 4-chlorobenzophenone, Michlers ketone, 1-acetonaphthone, 2-acetonaphthone, 1-benzoylcyclohexan-1-ol, 2-hydroxy-2,2-dimethylacetophenone, 2,2-dimethoxy-2-phenylacetophenone, 2,2-diethoxy-2-phenylacetophenone, 1,1-dichloroacetophenone, 1-hydroxyacetophenone, acetophenone dimethyl ketal, o-methoxybenzophenone, triphenylphosphine, tri-o-tolylphosphine, benz[a]anthracene-7,12-dione, 2,2-diethoxyacetophenone, benzil ketals such as benzil dimethyl ketal, 2-methyl-1-[4-(methylthio)phenyl]-2-morpholinopropan-1-one, anthraquinones such as 2-methylantraquinone, 2-

ethylanthraquinone, 2-tert-butylanthraquinone, 1-chloroanthraquinone, 2-amylanthraquinone and 2,3-butanedione.

5 Among the Photoinitiators mentioned, preference is given to phosphine oxides, α -hydroxyketones and benzophenones. It is also possible to use mixtures of various photoinitiators.

10 In general, the photoinitiator is present in the object material in an amount of from 0.001 to 15% by weight, preferably from 0.01 to 10% by weight, based on the total weight of the object material.

15 If the object material comprises a photoinitiator, it generally also comprises a sensitizer by means of which the photoinitiator is excited. Suitable sensitizers are normally used in combination with at least one of the above photoinitiators. A preferred combination comprises a sensitizer selected from among thioxanthone, benzophenone, coumarin and derivatives thereof. Sensitizers are preferably used in an amount in the range from 0.001 to 15% by weight, preferably from 0.01 to 10% by weight, based on the total weight of the object material.

20 If the object material comprises at least one free-radically curable compound, it preferably also comprises a stabilizer which suppresses spontaneous or thermally uncontrolled polymerization of the object material. Suitable stabilizers are, for example, hydroquinones or monomethylhydroquinones. Stabilizers are preferably concomitantly used in an amount of less than 500 ppm, more preferably less than 200 ppm, more
25 preferably less than 100 ppm.

The object material can appropriately comprise a thickener for setting a suitable viscosity. This suitable thickness are pyrogenic silica and laminar silicates.

30 The object material can further comprise other customary constituents such as antifoams, fluidizers, plasticizers, surface-active substances, pigments, dispersants and the like.

35 Free-radically curable object materials which are suitable for the purposes of the invention are commercially available. They comprise, for example, the Vero materials, Durus materials, Tango materials and FullCure materials, e.g. FullCure 720, produced by Stratasys.

In a further embodiment, the object material comprises a cationically polymerizable compound. For the present purposes, cationically polymerizable compounds are compounds in which the tendency for cationic polymerization is much greater than that for free-radical polymerization.

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In the cationic polymerization, a reactive cation, i.e. a Lewis or Brönsted acid, which reacts in an initiating reaction with the double bond of a reactive unit is used as initiator. This is followed by growth reactions in which the cation resulting from the initiating reaction adds onto a further monomer to once again form a cation.

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The cationically polymerizable compounds preferably do not comprise any ethylenically unsaturated radicals.

As cationically polymerizable compounds, it is possible to use cationically polymerizable monomers and macromonomers such as epoxides, oxetanes, oxazolines, lactones, lactams, vinyl ethers, furans, cyclic ketene acetals, spiroorthocarbonates or bicyclic ortho esters. The object material preferably comprises a cationically polymerizable compound selected from among epoxides, vinyl ethers, lactones, lactams, oxetanes and oxazolines.

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Suitable epoxides are, in particular, glycidyl ethers or cycloaliphatic epoxides such as bisphenol A diglycidyl ether, 3,4-epoxycyclohexylmethyl 3,4-epoxycyclohexanecarboxylate and bis(3,4-epoxycyclohexylmethyl) adipate.

25 Suitable oxetanes comprise 2-ethylhexyloxetane and 1,10-decanediylbis(oxy-methylene)bis(3-ethyloxetane).

Suitable oxazolines are selected from among 2-phenyloxazoline and 2-decyloxazoline.

30 Suitable lactones are selected from among β -propiolactone, γ -butyrolactone, δ -valerolactone and ϵ -caprolactone.

Suitable lactams are selected from among β -propiolactam, γ -butyrolactam, δ -valerolactam, ϵ -caprolactam and N-methyl-2-pyrrolidone.

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Suitable vinyl ethers comprise C_{1-8} -alkyl vinyl ethers.

Suitable furans comprise furan, 3-(C_{1-8} -alkyl)furans and 4-(C_{1-8} -alkyl)furans.

Suitable cyclic ketene acetals comprise 2-methylene-1,3-dioxepane, 2-phenyl-4-methylene-1,3-dioxolane.

5 Suitable spiroorthocarbonates comprise 2-methylene-1,4,6-trioxaspiro[2.2]nonane and 3,9-dimethylene-1,5,7,11-tetraoxaspiro[5.5]undecane.

10 In a preferred embodiment, the object material comprises at least 80% by weight of the cationically polymerizable compound, particularly preferably at least 90% by weight, based on the total weight of the object material.

15 If the object material comprises at least one cationically polymerizable compound, it preferably also comprises a photoacid generator. For the purposes of the present invention, a photoacid generator is a compound which on irradiation with short-wavelength light, for example UV irradiation, liberates a reactive cation (i.e. a Lewis or Brönsted acid).

Suitable photoacid generators comprise ionic and in particular nonionic photoacid generators.

20 Ionic photoacid generators are derived from stable organic onium salts, in particular those having nitrogen, phosphorus, oxygen, sulfur, selenium or iodine as central atom of the cation. Preference is given to aromatic sulfonium and iodonium salts with complex anions, phenacylsulfonium salts, hydroxyphenylsulfonium salts and sulfoxonium salts.

25 Such ionic photoacid generators comprise, for example, the commercial products having the names Irgacure 250, Irgacure PAG 290 and GSID26-1 from BASF SE; Cyracure UVI-6990 and Cyracure UVI-6974 from Union Carbide; Degacure KI 85 from Degussa; Optomer SP-55, Optomer SP-150 and Optomer SP-170 from Adeka;
30 GE UVE 1014 from General Electric, SarCat CD 1012; and SarCat KI-85, SarCat CD 1010 and SarCat® CD 1011 from Sartomer.

35 Nonionic photoacid generators comprise compounds which on photolysis liberate carboxylic acids, sulfonic acids, phosphoric acids or hydrogen halides, for example nitrobenzyl esters, sulfonic acid derivatives, phosphate esters, phenolsulfonate esters, diazonaphthoquinone and N-hydroxyimidosulfonate. These can be used either alone or in combination. Preference is given to sulfonic acid derivatives. Compared to ionic photoacid generators, nonionic photoacid generators are soluble in a wide range of solvents.

Such nonionic photoacid generators comprise, for example, N-hydroxy-5-norbornene-2,3-dicarboximide perfluoro-1-butanesulfonate, N-hydroxynaphthalimide triflate and 2-(4-methoxystyryl)-4,6-bis(trichloromethyl)-1,3,5-triazine, and also the commercial
5 products having the names Irgacure PAG 103, Irgacure PAG 121, Irgacure PAG 203, CGI 725 and CGI 1907 from BASF SE.

It is also possible to use organic silicon compounds which on UV irradiation in the presence of an aluminum-comprising organic compound liberate a silanol.
10

Further suitable photoacid generators are those which are excited by means of a sensitizer. Sensitizers are preferably used in an amount in the range from 0.001 to 15% by weight, preferably from 0.01 to 10% by weight, based on the total weight of the object material.
15

Suitable sensitizers are usually used in combination with at least one of the above photoacid generators. Preferred sensitizers are polycyclic aromatic compounds such as anthracene, naphthalene and derivatives thereof (see also US 6,313,188, EP 0 927 726, WO 2006/073021, US 4,997,717, US 6,593,388 and WO 03/076491). A
20 preferred combination comprises a sensitizer selected from among polycyclic aromatic compounds and a nonionic photoacid generator.

In a preferred embodiment of the present invention, the photoacid generator is present in the object material in an amount of from 0.001 to 15% by weight, particularly
25 preferably from 0.01 to 10% by weight.

In a preferred embodiment, the object material is free of nonaqueous solvents.

The object material can appropriately comprise a thickener for setting a suitable
30 viscosity. Suitable thickeners are pyrogenic silica and laminar silicates.

The object material can further comprise other customary constituents such as antifoams, fluidizers, plasticizers, surface-active substances, pigments, dispersants and the like.
35

Pre-support materials

The process of the invention utilizes two liquid components which on meeting form a solid support material. The formation of a solid support material can be effected by gelling and/or precipitation of a precipitate.

Suitable pre-support materials are liquid components, preferably having a viscosity of less than 20 mPas at 70°C, which can be ejected via a printing head. These properties are fulfilled by low molecular weight polymers and polymer solutions.

The solidification of the pre-support materials when they meet and the associated stabilization of the object material is based on a mechanism which differs from the free-radical mechanism of the curing of the object material. This decreases or suppresses a reaction between pre-support materials and object material and thus gives smoother surfaces after removal of the support materials by means of an aqueous solution.

The solidification of the combined pre-support materials is appropriately based on noncovalent intermolecular interactions between the pre-support materials.

In one embodiment, the first pre-support material comprises a first polymer which has a plurality of positively charged groups and/or latently positively charged groups or has a plurality of proton acceptors and the second pre-support material comprises a second polymer which has a plurality of negatively charged groups and/or latently negatively charged groups or has a plurality of proton donors.

A latently charged group is a functional group which can be converted into a charged group by release of a proton into or acceptance of a proton from the surroundings during use. Proton donors comprise a plurality of electronegative atoms, e.g. nitrogen or oxygen, to which a hydrogen atom is covalently bound. Proton acceptors comprise a plurality of atoms having free electron pairs.

On combination, the Coulomb interaction and/or hydrogen bonds between the first polymer and the second polymer, for example, lead to solidification, for example by means of gelling and/or precipitation. As a result of the Coulomb interaction and/or hydrogen bonds, the polymers are noncovalently crosslinked and solvent molecules present are optionally enclosed in the form of a gel.

Suitable first polymers are polyacids. Possible organic polyacids are polymers which have a plurality of free carboxylic acid groups. These can be homopolymers or

- copolymers. Suitable carboxyl-comprising monomers or monomers comprising carboxylic acid groups are, in particular, monoethylenically unsaturated monocarboxylic or dicarboxylic acids having from 3 to 6 carbon atoms or corresponding anhydrides thereof, for example acrylic acid, methacrylic acid, ethylacrylic acid, allylacetic acid, 5 crotonic acid, vinylacetic acid, maleic acid, itaconic acid, mesaconic acid, fumaric acid, citraconic acid, methylenemalononic acid and also the partial esters thereof, e.g. monoalkyl maleates, and mixtures thereof. In the case of monoalkyl dicarboxylates, the number of carbon atoms indicated relates to the dicarboxylic acid skeleton; the alkyl group in the ester part can have, independently thereof, from 1 to 20 carbon atoms, in 10 particular from 1 to 8 carbon atoms. Possible corresponding monoethylenically unsaturated dicarboxylic anhydrides are maleic anhydride, itaconic anhydride, citric anhydride and mixtures thereof. Preference is given to using acrylic acid, methacrylic acid, maleic acid, itaconic acid and maleic anhydride.
- 15 Further suitable first polymers are polystyrenesulfonic acid and copolymers thereof with abovementioned acid monomers and also the salts thereof and the polymers and copolymers with abovementioned acid monomers of 2-acrylamido-2-methylpropane-sulfonic acid and the salts thereof.
- 20 Particular preference is given to using poly(meth)acrylic acid as first polymer.

- Suitable second polymers are polyamines and polyamides. These include poly-2-oxazolines, polyalkylenepolyamines, polyalkylenimines, polyallylamines, 25 polyamidoamines, polyvinylamines and polyvinylpyridine.
- Specific examples are poly-2-methyloxazoline, poly-2-ethyloxazoline, tetraethylenepentamine, pentaethylenehexamine, polyethylenimine, polyallylamine hydrochloride, poly(diallyldimethylammonium chloride) and the reaction product of 30 diethylenetriamine and adipic acid.

- The molecular weight of the first or second polymer is preferably in the range from 500 g/mol to 1 000 000 g/mol, in particular from 1000 g/mol to 20 000 g/mol, particularly preferably from 2000 g/mol to 8000 g/mol.
- 35 In a preferred embodiment, the ratio of the molar mass of the first polymer to the molar mass of the second polymer is from 1.5:1 to 1:1.5, particularly preferably from 1.3:1 to 1:1.3, very particularly preferably from 1.1:1 to 1:1.1. In a further preferred embodiment, the first and second pre-support materials are ejected in such amounts that the weight ratio of the first polymer to the second polymer is from > 1:1 to 20:1 or

from 1:20 to $< 1:1$, particularly preferably from $> 1.3:1$ to 20:1 or from 1:20 to $< 1:1.3$, very particularly preferably from $> 1.5:1$ to 20:1 or from 1:20 to $< 1:1.5$.

5 The polymers are preferably ejected as solutions in a suitable solvent. Possible solvents are alcohols, esters, ketones but preferably water. In a preferred embodiment, the pre-support materials are free of nonaqueous solvents. The concentration of the polymers in solution is, for example, in the range from 2 to 70% by weight, preferably from 5 to 50% by weight.

10 Secondary support material

It is also possible to eject the support material in the form of a separating layer. The actual support is then provided by a secondary support material, with the support material and the secondary support material having at least one common interface.

15 The object material and the secondary support material preferably do not have any common interface. In this case, object material and secondary support material can be crosslinked or polymerized by the same mechanism, for example both object material and secondary support material are free-radically crosslinkable or polymerizable.

20 Possible secondary support materials are all support materials which are known for 3D printing and after radiation curing are soluble in an aqueous medium.

After radiation curing of the object material and the support material, the secondary support material can be removed together with the cured support material by treatment
25 with an aqueous medium, preferably an alkaline aqueous medium.

The secondary support material is advantageously a conventional support material for 3D printing processes. In these cases, the secondary support material is generally cheaper than the support material. Application of the support material as separating
30 layer then allows, firstly, exact contours and a smooth surface to be obtained and at the same time allows an inexpensive process.

The secondary support material can also be a wax-like material which after ejection solidifies by cooling, e.g. polyethylene glycols or ethoxylated fatty alcohols having
35 suitable melting points.

The printing materials of the present invention can also comprise suitable auxiliaries such as accelerators, absorbers, mechanical stabilizers, pigments, dyes, viscosity modifiers, agents for reducing the surface tension and wetting agents and antioxidants.

Radiation curing

5 The radiation curing of the object material and optionally the secondary support material (hereinafter referred to as the printing materials) is effected by means of high-energy light, e.g. UV light or electron beams, preferably UV light. Radiation curing can be carried out at elevated temperatures. However, the temperature is preferably below the glass transition temperature T_g of the printing materials.

10 For the present purposes, radiation curing refers to the crosslinking or polymerization of the free-radically crosslinkable or cationically polymerizable compounds of the printing materials as a result of electromagnetic and/or particle radiation, preferably UV light in the wavelength range of $\lambda = 200$ to 700 nm and/or electron beams in the range from 150 to 300 keV and particularly preferably at a radiation dose of at least
15 80 mJ/cm^2 , preferably from 80 to 3000 mJ/cm^2 .

Suitable radiation sources for radiation curing are, for example, low-pressure, intermediate-pressure and high-pressure mercury lamps and also fluorescence tubes, pulsed lamps, metal halide lamps, electronic flash devices, by means of which radiation
20 curing without photoinitiator is possible, or excimer lamps. Radiation curing is effected by action of high-energy radiation, i.e. UV radiation or daylight, preferably light in the wavelength range of $\lambda = 200$ to 700 nm, particularly preferably $\lambda = 200$ to 500 nm and very particularly preferably $\lambda = 250$ to 400 nm, or by irradiation with high-energy electrons (electron beam; 150 to 300 keV). For example, high-pressure mercury vapor
25 lamps, lasers, pulsed lamps (flash), halogen lamps, LED lamps or excimer lamps serve as radiation sources. The radiation dose which is usually sufficient for crosslinking in UV curing is in the range from 80 to 3000 mJ/cm^2 .

Of course, it is also possible to use a plurality of radiation sources, e.g. from two to
30 four, for curing. These can also radiate in different wavelength ranges.

Irradiation can also be carried out in the absence of oxygen, e.g. under an inert gas atmosphere. Suitable inert gases are preferably nitrogen, noble gases, carbon dioxide or combustion gases.

35

When curing is effected by means of UV irradiation instead of electron beams, it goes without saying that the printing materials each comprise at least one photoinitiator or photoacid generator which is activable in a wavelength range of the radiation used for irradiation.

3D printing

The process of the invention is appropriately carried out in a photopolymer jet printing apparatus. Here, a printing block having at least two printing heads moves back and forth over a building platform and leaves thin layers of photopolymerizable printing materials behind on the building platform. The amount of printing materials ejected and thus the thickness of the layers is set via a regulator which is coupled to a computer-aided construction (CAD) system. Each layer is cured immediately after application by means of UV lamps which are installed directly on the printing block. The building platform is appropriately lowered with increasing height of the printed structure, so that the printing block moves exclusively along the x axis during printing.

In a preferred embodiment of the printing head arrangement, this has a plurality of printing nozzles which are arranged along a line and through which the in each case photopolymerizable printing material can be ejected in a uniformly distributed manner. The printing heads preferably have at least 20, particularly preferably from 50 to 500, printing nozzles. During discharge of the material, the printing head arrangement is preferably moved orthogonally to the linear arrangement of the individual printing nozzles relative to the working plane. A printing head arrangement configured in this way makes it possible to dispense with stocking of a liquid photocrosslinkable material within a bath, as is customary, for example, in the SL process.

In an embodiment of the present invention, object material and support material are applied to one substrate. A stiff or flexible substrate is preferably used as substrate; in particular, the substrate can be made of a polymer material. In one embodiment, the substrate can be a plastic sheet, plastic film, membrane, glass, metal, semimetal, nonwoven or paper, preferably of biocompatible, in particular biodegradable, material.

In one embodiment of the present invention, the substrate is, after conclusion of the repeated execution of the process sequences a), b), b') and c) or a), b), b'), b'') and c), particularly preferably after step d), separated off from the resulting three-dimensional structure, in particular by means of chemical, physical or biological degradation.

In a further embodiment of the present invention, the substrate remains part of the structure produced after conclusion of the repeated execution of the process sequences a), b), b') and c) or a), b), b'), b'') and c), particularly preferably after step d), and thus becomes an integral constituent of the three-dimensional structure.

Removal of the support material

The cured support material and optionally the cured secondary support material are removed by treatment with an aqueous medium. Here, the solidification mechanism of the support material or the secondary support material is reversed and the support material or the secondary support material is dissolved. The aqueous medium is appropriately an aqueous alkali medium, e.g. aqueous sodium hydroxide solution having, for example, a concentration of from 0.1 to 2 M. As an alternative, an aqueous acidic medium is suitable. The structure obtained can be freed of the cured support material by dipping into or leaching with the aqueous medium. As an alternative, the structure can be blasted with the aqueous medium.

The pre-support materials are preferably water-based. The pre-support materials are particularly preferably free of nonaqueous solvents. In these cases, the pH of the aqueous medium by means of which the cured pre-support materials or the cured secondary support material are removed appropriately differs by at least 1, preferably at least 2, very particularly preferably at least 3, from the pH of the combined first and second pre-support material.

The invention will be illustrated with the aid of the accompanying drawing and the following examples.

Fig. 1 schematically shows an apparatus suitable for carrying out the process of the invention.

In fig. 1, the apparatus comprises a printing block 1 which comprises three or more printing heads 2, which are individually designated as 2A, 2B and 2C, and at least three storage vessels or dispensers 3 which comprise different printing materials and are individually designated as 3A (object material), 3B (first pre-support material) and 3C (second pre-support material). The dispensers 3 can be in each case be charged via lines to external reservoirs (not shown in fig. 1). Other printing materials and other combinations of printing materials can be used. The pressure heads 2 each have a plurality of nozzles as are used, for example, in inkjet processes through which the printing materials 3A, 3B and 3C are ejected.

In one embodiment of the present invention, the first dispenser comprising the object material 3A is connected to a first set of nozzles, designated as 4A, the second dispenser comprising first pre-support material 3B is connected to a second set of nozzles, designated as 4B, and the third dispenser comprising second pre-support

material 3C is connected to a third set of nozzles, designated as 4C. Accordingly, object material 3A is ejected through the nozzles 4A, first pre-support material 3B is ejected through the nozzles 4B and second pre-support material 3C is ejected through the nozzles 4C. In some embodiments, the three-dimensional printing system
5 optionally comprises (not shown) more than three printing heads, with each printing head being connected to a dispenser comprising object material or (pre-)support material and being able to be controlled in order to eject the material in the respective dispenser by means of the nozzles of the printing head. Use is optionally made of more than one object material, in which case each object material is ejected using a different
10 dispenser and printing head.

The printing apparatus additionally comprises a regulator 5, a computer-aided construction (CAD) system 6, a UV curing unit 7 and optionally a positioning device 8. The regulator 5 is coupled to the CAD system 6, the UV curing unit 7, the optional
15 positioning device 8, the printing heads 2 and each of the dispensers 3 which comprise the printing materials. Regulation can be effected by units different from those shown, e.g. one or more separate units.

The three-dimensional structure 9 to be produced is produced in layers using at least
20 one of the object materials 3A on a printing platform 10 having an adjustable height, with the height of each layer typically being able to be regulated by the discharge of the individual inkjet nozzles 4A being set selectively.

Example 1

25 In this example, the suitability of a mixture of a first pre-support material and a second pre-support material as support material was examined. An aqueous solution of randomly copolymerized 2-ethyloxazoline and 2-methyloxazoline (Mw: about 5000 g/mol) solution and an aqueous polymethacrylic acid solution (Mw: about
30 5000 g/mol) were combined in the weight ratios 1:1, 2:1 and 10:1 at in each case different concentrations (from 5 to 50% by weight). A solid always precipitated immediately. The solid could be redissolved at an increased pH (e.g. pH from 8 to 14).

Example 2a

35 Helios 1 Full Cure 525 photopolymer jet ink (comprising glycerol propoxylate (1PO/OH) triacrylate, CAS No. 52408-84-1, and a photoinitiator, object ink) from Stratasys was applied in a silicone mold as bead having a diameter of about 1 cm alongside an aqueous solution of randomly copolymerized 2-ethyloxazoline and 2-methyloxazoline

(Mw: about 5000 g/mol, 50% by weight) and aqueous polymethacrylic acid solution (Mw: about 5000 g/mol, 25% by weight) applied simultaneously as bead having a diameter of about 1 cm in such a way that the two beads were in contact at an interface.

5

The materials were then illuminated by means of an Hg lamp (365 nm) at 30 mW/cm² for 5 minutes. The polyoxazoline-polymethacrylic acid complex was removed in an aqueous-alkaline medium (1 M NaOH). For this purpose, the aqueous-alkali medium was introduced into the mold, whereupon the cured support material dissolved within a few minutes. The aqueous-alkali medium was taken out and the object which remained was washed with water. The object had no visible residues of support material.

10

It can be seen that the use of the support material used here makes it possible to produce an object surface free of residues.

15

Example 2b

Helios 1 Full Cure 525 photopolymer jet ink (comprising glycerol propoxylate (1PO/OH) triacrylate, CAS No. 52408-84-1, and a photoinitiator, object ink) from Stratasys was applied in a silicone mold as bead having a diameter of about 1 cm alongside an aqueous solution of randomly copolymerized 2-ethyloxazoline and 2-methyloxazoline (Mw: about 5000 g/mol, 30% by weight) and aqueous polymethacrylic acid solution (Mw: about 5000 g/mol, 15% by weight) applied simultaneously as bead having a diameter of about 1 cm in such a way that the two beads were in contact at an interface.

25

The materials were then illuminated by means of an Hg lamp (365 nm) at 30 mW/cm² for 5 minutes. The polyoxazoline-polymethacrylic acid complex was removed in an aqueous-alkaline medium (1 M NaOH). For this purpose, the aqueous-alkali medium was introduced into the mold, whereupon the cured support material dissolved within a few minutes. The aqueous-alkali medium was taken out and the object which remained was washed with water. The object had no visible residues of support material.

30

It can be seen that the use of the support material used here makes it possible to produce an object surface free of residues.

35

Example 3a

Helios 1 Full Cure 525 photopolymer jet ink (comprising glycerol propoxylate (1PO/OH) triacrylate, CAS No. 52408-84-1, and a photoinitiator, object ink) from Stratasys was applied in a silicone mold as bead having a diameter of about 1 cm alongside an aqueous solution of randomly copolymerized 2-ethyloxazoline and 2-methyloxazoline (Mw: about 5000 g/mol, 50% by weight) and aqueous polymethacrylic acid solution (Mw: about 5000 g/mol, 25% by weight) applied simultaneously as bead having a diameter of about 1 cm in such a way that the two beads were in contact at an interface. Next to the polyoxazoline-polymethacrylic acid complex, Support Full Cure 705 (comprising glycerol propoxylate(1PO/OH) triacrylate, CAS No. 52408-84-1, and bis(2,4,6-trimethylbenzoyl)phenylphosphine oxide (Irgacure 819), CAS No. 162881-26-7, secondary support material) from Stratasys was applied in such a way that the polyoxazoline-polymethacrylic acid complex and Support Full Cure 705 were in contact at a second interface. The beads had a thickness of from about 0.5 to 1 mm.

The materials were then illuminated by means of an Hg lamp (365 nm) at 30 mW/cm² for 5 minutes. The cured polyoxazoline-polymethacrylic acid complex or Support Full Cure 705 was removed in an aqueous-alkaline medium (1 M NaOH). For this purpose, the aqueous-alkali medium was introduced into the mold, whereupon the cured support material and the cured secondary support material dissolved within a few minutes. The aqueous-alkali medium was taken out and the object which remained was washed with water. The object had no visible residues of support material.

It can be seen that the support material ink according to the invention can also serve as layer between the object and a conventional support material.

Example 3b

Helios 1 Full Cure 525 photopolymer jet ink (comprising glycerol propoxylate (1PO/OH) triacrylate, CAS No. 52408-84-1, and a photoinitiator, object ink) from Stratasys was applied in a silicone mold as bead having a diameter of about 1 cm alongside an aqueous solution of randomly copolymerized 2-ethyloxazoline and 2-methyloxazoline (Mw: about 5000 g/mol, 30% by weight) and aqueous polymethacrylic acid solution (Mw: about 5000 g/mol, 15% by weight) applied simultaneously as bead having a diameter of about 1 cm in such a way that the two beads were in contact at an interface. Next to the polyoxazoline-polymethacrylic acid complex, Support Full Cure 705 (comprising glycerol propoxylate(1PO/OH) triacrylate, CAS No. 52408-84-1, and bis(2,4,6-trimethylbenzoyl)phenylphosphine oxide (Irgacure 819), CAS No.

162881-26-7, secondary support material) from Stratasys was applied in such a way that the polyoxazoline-polymethacrylic acid complex and Support Full Cure 705 were in contact at a second interface. The beads had a thickness of from about 0.5 to 1 mm.

- 5 The materials were then illuminated by means of an Hg lamp (365 nm) at 30 mW/cm² for 5 minutes. The cured polyoxazoline-polymethacrylic acid complex or Support Full Cure 705 was removed in an aqueous-alkaline medium (1 M NaOH). For this purpose, the aqueous-alkali medium was introduced into the mold, whereupon the cured support material and the cured secondary support material dissolved within a few minutes.
- 10 The aqueous-alkali medium was taken out and the object which remained was washed with water. The object had no visible residues of support material.

It can be seen that the support material ink according to the invention can also serve as layer between the object and a conventional support material.

Claims

1. A process for producing a three-dimensional structure, which comprises the following steps:
- 5
- a) ejection of an object material from a first printing head;
- b) ejection of a first pre-support material from a second printing head; and
- 10 b') ejection of a second pre-support material from a third printing head,
- where the pre-support materials comprise liquid components which on meeting form a solid support material;
- 15 c) radiation curing of the object material;
- where the steps a) to c) are repeated a plurality of times in order to form the three-dimensional structure in a layerwise manner and the object material and the support material have at least one common interface; and
- 20 d) removal of the solidified support material by treatment with an aqueous medium.
- 25 2. The process according to claim 1,
- wherein the pre-support materials comprise liquid components which on meeting become solid and form a support material in the form of a separating layer; and the process additionally comprises:
- 30 b'') ejection of a secondary support material from a fourth printing head;
- wherein the steps a), b), b'), b'') and c) are repeated a plurality of times in order to form the three-dimensional structure in a layerwise manner and the support material and the secondary support material have at least one common interface;
- 35 d) removal of the solidified support material and the cured secondary support material by treatment with an aqueous medium.

3. The process according to either of the preceding claims, wherein the viscosity of the object material and of the pre-support materials is in each case less than 20 MPas at 70°C.
- 5 4. The process according to any of the preceding claims, wherein the first pre-support material comprises a first polymer which has a plurality of positively charged groups and/or latently positively charged groups or has a plurality of proton acceptors and the second pre-support material comprises a second polymer which has a plurality of negatively charged groups and/or latently negatively charged groups or has a plurality of proton donors.
10
5. The process according to claim 4, wherein the ratio of the molar mass of the first polymer to the molar mass of the second polymer is from 1.5:1 to 1:1.5.
- 15 6. The process according to claim 5, wherein the ratio of the molar mass of the first polymer to the molar mass of the second polymer is from 1.3:1 to 1:1.3.
7. The process according to claim 5 or 6, wherein the first pre-support material and the second pre-support material are ejected in such amounts that the weight ratio of the first polymer to the second polymer is from 1:20 to < 1:1 or from > 1:1 to 20:1.
20
8. The process according to any of the preceding claims, wherein the pre-support materials are water-based.
25
9. The process according to claim 8, wherein the pre-support materials are free of nonaqueous solvents.
10. The process according to claim 8 or 9, wherein the aqueous medium in step d) has a pH which differs by at least 1 from the pH of the combined first and second pre-support materials.
30