

Coated medical product

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Description

Field of invention

The present invention relates to a suspension for coating of medical devices comprising at least one tri-*O*-acylglycerol, at least one microcrystalline taxane, and at least one solvent or solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-*O*-acylglycerol is present. The present invention further relates to methods for preparing said suspension, methods for coating medical devices, and medical devices coated with at least one tri-*O*-acylglycerol and at least one microcrystalline taxane.

Background of the invention

Medical devices are used to take over functions that are missing in the body, to support the body's own functions or to transfer active substances locally with their help. Depending on the area of application, medical devices have either short- or long-term contact with an organism. The contact time can range from a few seconds to decades. If the use of a medical device becomes necessary, it is necessary to control the inevitable inflammatory processes that occur during wound healing in order to prevent overreactions of the immune system in the healing process.

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When treating vasoconstrictions (stenoses) with mechanical or thermal procedures, such as implantation of vascular stents or balloon angioplasty, restenosis occurs as a frequent complication a few weeks after treatment. To prevent restenosis, stents and catheter balloons have been coated with active agents, especially antirestenotic agents. In the past, limus active agents such as rapamycin (sirolimus) or taxanes such as paclitaxel have proven to be successful active agents. Limus active agents bind reversibly to FKBP12 and suppress cell division, whereas taxanes such as paclitaxel bind irreversibly to microtubules and also suppress cell division.

"Drug-eluting stents" (DES) are known in the prior art in which, in addition to vessel dilation and the associated injury to the vessel wall, healing at the affected site is to be controlled with the aid of suitable active agents.

Medical devices that do not remain permanently in the body, such as biodegradable stents, are also known in the art. The biodegradable stents may additionally have a drug coating to provide the benefits of a long-term medical device. However, this approach is still under development.

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In addition to stents, drug-eluting catheters, in particular balloon catheters, are also known in the prior art, which have the advantage that they only come into contact with the organism for a short period of time.

10 However, not only stents and catheters can be coated with active agents, although the requirements for coating of other medical devices naturally differ depending on the area of application.

The requirements for active agent-releasing coatings of catheters are particularly high,
15 since a long-term, well-dosed and yet as quantitative as possible application of active agents beyond the very short retention time of the medical device represents a special challenge, especially in the case of the medical devices, which are used for a very short period of time, especially in the vascular area, whereby it must be ensured that, on the one hand, the active agent is not flushed away prematurely on its way to the
20 target site or, for example, crumbles away during expansion and only an undefined or insufficient amount of active agent reaches the vessel wall. On the other hand, in the case of a coronary "drug-coated balloon" (DCB) as a medical device used for a short period of time, the very limited contact time of maximum 90 seconds must also be sufficient for the active agent to be transferred in the intended dosage from the balloon
25 catheter to or into the vessel wall. The peripheral vascular system, e.g. in the leg artery, allows longer contact times of around 120 seconds and more, depending on which vessel is being treated, with the upper limit of contact time in peripheral vessels being a maximum of 5 minutes in the *superficial femoral artery* (AFS).

30 In the prior art, some active agent coatings of catheter balloons are known that circumvent these problems by accelerating the drug release or increasing the stability of the coating by using additional excipients in the coating (see WO2010/121840A2, WO2013/007653A1, WO2012/146681A1).

35 In the case of catheter balloons coated with taxanes, another reason for the low efficacy, in addition to the low drug transfer, is the short retention time of the taxane in the vessel wall.

It is known from the prior art that taxane crystals exhibit slower dissolution behavior than amorphous taxane particles.

5 Coatings with crystalline taxanes are therefore desirable to ensure a prolonged retention time of the taxane in the vessel wall. However, the use of crystalline balloon coatings carries the risk of embolism (WO2011/147408A2), so amorphous coatings are still preferred for catheter balloons in the prior art.

10 The application of taxane crystals by means of balloon dilation to the tissue to be treated has the advantage that the crystals act as drug depots and release the drug with a delay, whereas with amorphous drugs there is an immediate release after dilation. However, it has been shown that direct coating with taxane crystals or coating with a pure crystal-containing suspension of taxane has the particular disadvantage that the taxane crystals do not adhere sufficiently to the medical device surface.
15 Another problem is that suspensions of particles larger than 1 - 5 μm tend to sediment rapidly, which subsequently makes uniform coating with microcrystalline taxane from suspensions much more difficult.

20 Active agent coatings with crystalline taxane for catheter balloons are known from the prior art, which attempt to circumvent these problems. However, the prior art methods for coating catheter balloons with crystalline taxane compounds are costly and complicated.

25 In the prior art, mainly taxane crystal suspensions are proposed in which the size of the taxane crystals is in the nanoscale of less than 1 μm . However, crystals of this size have the disadvantage that they cannot sufficiently ensure the desired extended retention time of the taxane in the vessel wall.

30 Sufficiently strong, reproducible coatings of catheter balloons with taxane crystals that exhibit adequate drug transfer upon dilation, could not be obtained by coating a catheter balloon with a solution consisting of a solvent and a taxane as well as by sputtering taxane crystals onto the balloon surface followed by optional adhesion enhancement by "solvent bonding" and by crystallization of a dissolved taxane from a solution of the taxane in a solvent and a non-solvent.

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The objective of the present invention is to provide coating formulations and coated medical devices, wherein the coating is flexible, adheres very well to the medical device surface, has an optimal size distribution of the active agent particles, and delivers the active agent as quantitatively as possible even within a very short

residence time in the body, whereby the active agent can then also diffuse from the vessel wall into the cells over a much longer period of time.

With other words, the objective of the present invention is to provide compositions for coatings of short- and long-term medical devices, which adhere to the surface of the medical device as a coating in a stable yet flexible manner and, on the other hand, ensure the most complete and controlled possible transfer of active agent to the vessel wall or tissue in order to optimally support the healing process.

In particular, the objective of the present invention is to provide a coating of catheter balloons with crystalline taxane compounds, wherein the coating adheres to the surface of the catheter balloons in a stable yet flexible manner and, on the other hand, ensures a complete as well as controlled drug transfer to the vessel wall or tissue during dilatation in order to optimally support the healing process.

This objective is solved by the technical teaching of the independent claims of the present invention. Further advantageous embodiments of the invention result from the dependent claims, the description, the figures as well as the examples.

Brief description

Surprisingly, it has been found that a suspension for coating of medical devices, in particular of catheter balloons, stents and cannulas, containing a) at least one tri-O-acylglycerol selected from trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and b) at least one taxane in the form of microcrystals, and c) a solvent or a solvent mixture in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve or in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-O-acylglycerol is present, solves the above objective.

It has now been found that a particularly advantageous crystal suspension of a microcrystalline taxane for coating of medical devices, in which the microcrystals of the taxane do not dissolve, can be provided if at least one tri-O-acylglycerol selected from trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol is present in dissolved form in said suspension.

With the tri-O-acylglycerols according to the invention, it was surprisingly possible to prepare such stable suspensions of microcrystalline taxanes that the microcrystals of the taxanes are kept "in abeyance" and thus did not sediment. This was particularly surprising since suspensions of crystals in the micrometer range, i.e. crystals larger

than 1 - 5 μm tend to sediment. The crystal suspension according to the invention is thus particularly advantageous for producing a uniform coating of microcrystals of a taxane on medical devices.

5 Another particular advantage of using at least one tri-*O*-acylglycerol selected from trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, is that the tri-*O*-acylglycerols according to the invention are capable of holding the microcrystals of the taxane on a medical device surface like "flexible adhesives". The crystal suspension according to the invention is thus particularly advantageous for
10 producing a uniform coating of microcrystals of a taxane on medical devices, in which the microcrystals of the taxane also adhere sufficiently to the medical device surface.

With further prior art tri-*O*-acylglycerols not according to the invention, the preparation of crystal suspensions according to the present invention was not possible. It could be
15 shown that tri-*O*-acylglycerols not according to the invention cause or promote dissolution of the microcrystals of the taxane in the suspension. In addition, sedimentation of the microcrystals of the taxane occurred with tri-*O*-acylglycerols not according to the invention.

20 In addition, with further prior art tri-*O*-acylglycerols not according to the invention no uniform coatings of microcrystals of a taxane could be produced on medical devices, and in particular a lack of adhesion of the microcrystals of the taxane to the medical device surface has been observed. It could be shown that tri-*O*-acylglycerols not according to the invention cannot sufficiently "stick" and hold the microcrystals of the
25 taxane on a medical device surface.

Thus, only with the tri-*O*-acylglycerols selected from trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol or also with mixtures of these tri-*O*-acylglycerols, a crystal suspension of a microcrystalline taxane
30 according to the invention could be prepared, in which the microcrystals of the taxane remain intact. A further particular advantage is that the microcrystals of the at least one taxane are floating in the suspension and are therefore uniformly distributed in the suspension, so that the taxane can be applied not only in the form of microcrystals but also uniformly on the medical device surface.

35 Medical devices that have been coated with a suspension according to the invention exhibit a coating of at least one tri-*O*-acylglycerol selected from trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and microcrystals of the at least one taxane on the medical device surface. This coating is primarily

characterized by a very good flexibility and an excellent adhesion to the medical device surface. Furthermore, this coating offers the advantageous property that even with a very short residence time in the body, the microcrystals of the at least one taxane can be quantitatively released, which can then diffuse from the vessel wall into the cells over a much longer period of time, in contrast to amorphous taxane particles.

A coating according to the invention can be provided on any medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, particularly preferred are catheter balloons. The taxane amount and the taxane delivery rate or elution rate may vary according to the required specifications at the operation site, while the at least one tri-O-acylglycerol selected from trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, on the one hand, supports optimal transfer of the microcrystalline taxane into the tissue and, at the same time, ensures high flexibility and stability of the coating, thus guaranteeing that the microcrystalline taxane actually reaches the surrounding tissue in optimal concentration without losses.

The very good flexibility and adhesion of the coating according to the invention is particularly important for medical devices that have to undergo changes in shape, e.g. stents and catheter balloons. For example, inflation, deflation, folding and crimping require special stability requirements for a coating that is also exposed to friction and body fluids as well as flows during implantation. Setting the desired elution rate of the microcrystalline taxane and the most optimal transfer amount of microcrystalline taxane into the tissue are also solved with a catheter balloon coating (DCB) according to the invention. Further requirements to be considered, which the coating according to the invention fulfills without prejudice, relate to sterilization of the medical devices, shelf life, minimum shelf life, temperature resistance and the like.

Thus, the inventive coating on medical devices solves the important tasks that are demanded for a medical device used in the body in the long term and also in the short term.

Detailed description

The present invention relates to a suspension for coating of medical devices, preferably catheter balloons, balloon catheters, stents and cannulas, the suspension containing:

- a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and

- b) at least one taxane in the form of microcrystals, and
- c) a solvent or a solvent mixture in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve.

5 Essential to the invention is the use of microcrystalline taxane and the presence of a suspension of the microcrystalline taxane wherein the suspension must contain at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol.

10 The term "**coating formulation**" or "active agent-containing composition", as used herein, refers to a mixture of at least one taxane and a solvent or a solvent mixture and at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, i.e., a solution, dispersion, suspension or emulsion. The term "formulation" is intended to indicate that
 15 it is a liquid mixture (suspension, emulsion, dispersion, solution). The term "coating formulation", as used herein, thus represents the generic term for the terms "solution" or "coating solution", "dispersion" or "coating dispersion", "suspension" or "coating suspension" and "emulsion" or "coating emulsion".

20 The term "**solution**" or "coating solution", as used herein, generally refers to a homogeneous mixture consisting of two or more chemically pure substances. Solutions are not externally recognizable as such, since by definition they form only one phase and the solutes are uniformly distributed in the solvent.

25 The term "**dispersion**" or "coating dispersion", as used herein, generally refers to a heterogeneous mixture of at least two substances that do not or hardly dissolve in each other or chemically combine with each other. In this case, one or more substances are finely dispersed as a disperse phase in another continuous substance, the so-called dispersion medium.

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The term "**emulsion**" or "coating emulsion", as used herein, generally refers to finely distributed mixture of two normally immiscible liquids without visible segregation. One liquid forms small droplets dispersed in the other liquid. The emulsion is a particular form of a dispersion.

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The term "**suspension**" or "coating suspension", as used herein, generally refers to a heterogeneous mixture of substances consisting of a finely distributed solid in a liquid. Thus, by definition, a "suspension" is not a homogeneous mixture and thus not a solution. The suspension is a specific form of a dispersion.

A "suspension" containing at least one taxane in the form of microcrystals is also referred to herein as a "crystal suspension". According to the present invention, the finely distributed solid of the suspension herein is at least one microcrystalline taxane or microcrystals of at least one taxane. The liquid of the suspension herein is a solvent or a solvent mixture, wherein at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol is present in the dissolved form in the solvent or the solvent mixture.

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Thus, a "suspension" according to the present invention relates to a heterogeneous mixture of substances of a liquid containing at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and solids finely distributed in this liquid, namely the microcrystals of the at least one taxane. Thus, according to the invention, the microcrystalline taxane is suspended in a liquid containing at least one dissolved tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol.

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The suspension according to the present invention is characterized in that neither sedimentation nor dissolution of the microcrystals of the at least one taxane occurs in the suspension. The suspension according to the present invention is also referred to herein as a "stable suspension".

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The suspension according to the invention may consist of the solvent or the solvent mixture and the microcrystalline taxane and the at least one dissolved tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol. However, the suspension may contain up to 5.0% by weight of other additives based on the taxane, i.e. for 95g of taxane, up to 5g of additives may be contained in the suspension.

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Thus, the present invention also relates to a suspension for coating of medical devices, preferably selected from a catheter balloon, a catheter, a stent or a cannula, the suspension consisting of:

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- a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and
- b) at least one taxane in the form of microcrystals, and

- c) a solvent or a solvent mixture in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve, and
- d) up to 5.0% by weight, based on the taxane, of additives. .

5 Suitable additives are the substances mentioned below, preferably antioxidants, polyvinylpyrrolidone (PVP) and flocculation inhibitors.

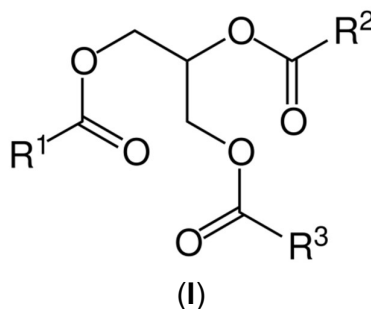
Additives, such as antioxidants, polyvinylpyrrolidone (PVP) and flocculation inhibitors can preferably be present in an amount of up to 4.0% by weight based on the
 10 at least one taxane, preferably up to 3.0 % by weight based on the at least one taxane, further preferably up to 2.5% by weight based on the at least one taxane, further preferably up to 2.0 % by weight based on the at least one taxane, further preferably up to 1.5% by weight based on the at least one taxane and further preferably up to 1.0% by weight based on the at least one taxane.

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The term "**tri-O-acylglycerol**," or short form "triacylglycerol", as used herein, refers to a chemical compound of glycerol (glycerin) esterified with three fatty acids, i.e., triple esterified glycerols (glycerins). Triglyceride or glycerol triester are synonymous terms of tri-O-acylglycerol, the term tri-O-acylglycerol is recommended by IUPAC.

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Tri-O-acylglycerols have the following general formula (I):



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wherein R^1 , R^2 and R^3 represent alkyl or alkenyl residues. The structure of tri-O-acylglycerols is diverse, since R^1 , R^2 and R^3 allow many different fatty acids and thus a high number of possible combinations. All are non-polar, i.e. lipophilic. In the case of tri-O-acylglycerols, a further distinction can be made between medium-chain and long-chain tri-O-acylglycerols. Medium-chain tri-O-acylglycerols have fatty acids with an average length of 6 to 12 carbon atoms, and long-chain tri-O-acylglycerols have fatty acids with a length of 14 to 24 carbon atoms. Thereby, two types of tri-O-acylglycerols can arise: simple and mixed tri-O-acylglycerols. In simple tri-O-acylglycerols, the fatty acid residues R^1 , R^2 and R^3 are identical; in mixed ones, at least one of the fatty acid residues R^1 , R^2 and R^3 is different from the other two.

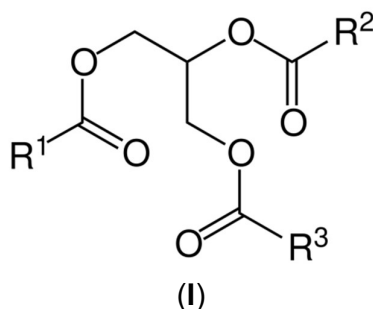
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Examples of medium-length fatty acids are caproic acid (hexanoic acid), enanthic acid (heptanoic acid), caprylic acid (octanoic acid), perlargonic acid (nonanoic acid), capric acid (decanoic acid), undecanoic acid and lauric acid (dodecanoic acid).

5 Unexpectedly, it could be shown that the chemical, physical and biological properties of tri-O-acylglycerols fully esterified with the medium-length fatty acids caprylic acid (octanoic acid), capric acid (decanoic acid), perlargonic acid (nonanoic acid) or undecanoic acid enable a uniform and sufficiently adhesive coating of medical devices with microcrystalline taxanes in the first place. Moreover, only the glycerols fully
 10 esterified with caprylic acid (octanoic acid), capric acid (decanoic acid), perlargonic acid (nonanoic acid) or undecanoic acid it was possible to prepare a crystal suspension according to the invention.

Thus, the herein preferred tri-O-acylglycerols have the following general formula (I):

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wherein R^1 , R^2 , and R^3 are independently of each other selected from $-\text{CH}_2(\text{CH}_2)_5\text{CH}_3$, $-\text{CH}_2(\text{CH}_2)_6\text{CH}_3$, $-\text{CH}_2(\text{CH}_2)_7\text{CH}_3$ and $-\text{CH}_2(\text{CH}_2)_8\text{CH}_3$.

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It could be shown that with mixed tri-O-acylglycerols in which R^1 , R^2 and R^3 are independently selected from $-\text{CH}_2(\text{CH}_2)_5\text{CH}_3$, $-\text{CH}_2(\text{CH}_2)_6\text{CH}_3$, $-\text{CH}_2(\text{CH}_2)_7\text{CH}_3$ and $-\text{CH}_2(\text{CH}_2)_8\text{CH}_3$, and wherein not all three R^1 , R^2 and R^3 are identical, stable crystal suspensions of microcrystalline taxanes can also be prepared. However, these
 25 mixed tri-O-acylglycerols are more costly to prepare and are not cost effective, so they are not preferred herein.

Therefore, according to the present invention, all three residues R^1 , R^2 and R^3 are identical, i.e. R^1 , R^2 and R^3 are $-\text{CH}_2(\text{CH}_2)_5\text{CH}_3$, or R^1 , R^2 and R^3 are $-\text{CH}_2(\text{CH}_2)_6\text{CH}_3$,
 30 or R^1 , R^2 and R^3 are $-\text{CH}_2(\text{CH}_2)_7\text{CH}_3$, or R^1 , R^2 and R^3 are $-\text{CH}_2(\text{CH}_2)_8\text{CH}_3$.

In attempts to prepare crystal suspensions containing tri-O-acylglycerols, which are completely esterified with other medium-length fatty acids not according to the invention as mentioned above, in particular the shorter medium-length fatty acids such

as caproic acid (tricaproic), but also the shorter monocarboxylic acids such as acetic acid (triacetin), it was not possible to prepare crystal suspensions according to the invention, because in the presence of these tri-O-acylglycerols not according to the invention, dissolution of the microcrystals of the taxane in the suspension occurred or
5 was promoted. Dissolution of the microcrystals of the taxane in the suspension also occurred with glycerols only partially esterified with medium-length fatty acids, such as the mono-O-acylglycerols or di-O-acylglycerols. In addition, a sedimentation of the microcrystals of the taxane occurred with the tri-O-acylglycerols not according to the invention.

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In complete absence of tri-O-acylglycerols in the suspension, sedimentation of the microcrystals of the at least one taxane occurred rapidly. Preparation of a stable suspension was not possible.

15 The presence of at least one dissolved tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol in the suspension is thus essential for the crystal suspension according to the present invention.

20 Thus, it could be shown that the glycerol fully esterified with three octanoic acid molecules, the glycerol fully esterified with three decanoic acid molecules, the glycerol fully esterified with three nonanoic acid molecules or the glycerol fully esterified with three undecanoic acid, i.e. at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and
25 triundecanoylglycerol, does not partially dissolve or does not dissolve microcrystalline taxanes in a suspension.

Thus, with the tri-O-acylglycerols according to the invention a crystal suspension can be provided as a coating formulation in which the microcrystals of the taxane remain
30 intact, suspended and uniformly distributed in the suspension, and no sedimentation of the microcrystals of the taxane or agglomeration of particles occurs, so that the taxane can be uniformly applied in microcrystalline form to a medical device surface.

Another particular advantage of using at least one tri-O-acylglycerol selected from the
35 group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol is that the tri-O-acylglycerols according to the invention are capable of holding the microcrystals of the taxane on a medical device surface like a "flexible adhesive", so that sufficient adhesion of the microcrystals of the taxane to a medical device surface can be provided.

The tri-*O*-acylglycerols according to the invention selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, or a mixture of said tri-*O*-acylglycerols have the advantage that their melting points make them safe for use in the body. It has also been found that a melting point of below 37°C is essential to ensure adequate adhesion of the microcrystals of the taxane to a medical device surface. Only the tri-*O*-acylglycerols according to the invention selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, or mixtures of said tri-*O*-acylglycerols can hold microcrystalline taxanes like a "glue", thus guaranteeing optimum flexibility and loss-free transport to the operation site. Even the next higher homolog, tridodecanoylglycerol, has the disadvantage that it melts at 45-46°C.

With tri-*O*-acylglycerols not according to the invention, which are fully esterified with medium-length fatty acids or long-chain fatty acids not according to the invention as mentioned above, such as lauric acid, myristic acid or palmitic acid, it was not possible to provide a crystal suspension according to the invention, in which the microcrystals of the taxane float in the suspension and are uniformly distributed in the suspension. When using tri-*O*-acylglycerols not according to the invention, such as tridodecanoylglycerol, uniform coating with taxane in microcrystalline form on medical device surfaces was not possible. Coated catheter balloons with a coating of taxane in microcrystalline form and tridodecanoylglycerol clearly lacked a uniform coating, the surface was uneven, and the coating easily crumbled off during inflation.

In attempts to coat medical devices with suspensions containing microcrystals of a taxane and tri-*O*-acylglycerols not according to the invention present in dissolved form in the suspension, which are fully esterified with the medium-length fatty acids or long-chain fatty acids not according to the invention as mentioned above, such as lauric acid, myristic acid or palmitic acid, it was not possible to produce coatings in which the microcrystals of the taxane are uniformly distributed on the medical device surface and adhere sufficiently to the medical device surface. For coatings with microcrystals of a taxane and a tri-*O*-acylglycerol not according to the invention, such as tridodecanoylglycerol, a higher particle release was observed in the "crumble test", compared to coatings of microcrystals of a taxane and at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol. The comparison clearly showed that for the coating according to the invention with microcrystals of a taxane and at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol the particle release

for all measured particle sizes is very far below the particle release for coatings with microcrystals of a taxane and a tri-O-acylglycerol not according to the invention, such as tridodecanoylglycerol.

5 It could be also surprisingly shown that outstanding adhesion of the microcrystals of the taxane to the medical device surface is achieved in particular when the tri-O-acylglycerol according to the invention is present in dissolved form in the crystal suspension according to the invention and is applied to the medical device surface together or simultaneously with the microcrystals of the suspended at least one taxane
10 in the suspension.

This outstanding adhesion of the microcrystals of the taxane to the medical device surface could not be reproduced when the medical device surface was first coated with a solution containing at least one tri-O-acylglycerol according to the invention and
15 when microcrystals of the taxane were subsequently applied to said tri-O-acylglycerol layer. Also, sufficient adhesion of the microcrystals of the taxane was not found when the medical device surface was first coated with crystals of the taxane according to the prior art methods and then coated with a solution containing a tri-O-acylglycerol in which the microcrystalline taxane does not dissolve.

20 In the prior art, the lack of adhesion of the microcrystals of the taxane to the medical device surface has been the major drawback for the use of microcrystalline taxanes for coating of medical devices. The present invention overcomes this drawback and provides a solution for providing coatings with microcrystalline taxanes on medical
25 device surfaces.

In the suspension according to the invention, the at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol is advantageously present in dissolved
30 form, so that when a medical device is coated with a suspension according to the invention, a coating is obtained in which not only the microcrystals of the taxane are uniformly distributed, but also the at least one tri-O-acylglycerol is uniformly distributed throughout the coating. This is particularly advantageous if a coating of microcrystals of a taxane is to be provided with a greater layer thickness and wherein the suspension
35 according to the invention is applied several times in succession. Then, the tri-O-acylglycerols according to the invention are uniformly distributed in such a coating and hold the microcrystals of the taxane like "flexible adhesive" on the medical device surface, but also among the microcrystals like "flexible adhesive".

Thus, the suspension of the present invention offers the additional advantage that the adhesion of the microcrystals of the at least one taxane is also increased among each other. If the microcrystals of the at least one taxane are first applied without a tri-*O*-acylglycerol dissolved in the suspension and, in a subsequent step, coated with a tri-*O*-acylglycerol solution, the tri-*O*-acylglycerol solution cannot sufficiently penetrate under and between the microcrystals of the taxane and thus cannot sufficiently provide the technical effect of increased adhesion on the medical device surface and between the microcrystals of the at least one taxane.

With the provision of a suspension according to the present invention containing at least one dissolved tri-*O*-acylglycerol selected from the group trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals, the major problem of poor adhesion of the microcrystals of the taxane to the medical device surface has now surprisingly been solved.

The coating of medical devices with microcrystalline taxanes in the presence of at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol also shows significantly reduced "crumbling" behavior compared to medical devices currently on the market, since they show a significantly reduced particle release compared to the medical products currently available on the market, especially in the area of drug-delivering balloon catheters, thus proving that the stability of a coating according to the present invention is equally increased. This results in particular from the advantage that the at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, firmly holds the microcrystals of the taxane on the medical device surface like "flexible adhesive", resulting in a stable, non-brittle and flexible coating.

In addition, it could be shown that tri-*O*-acylglycerols selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, or mixtures of said tri-*O*-acylglycerols have other advantages. Among other things, they enable optimal transfer of the microcrystalline taxane into a tissue. The coatings prepared with the suspension according to the invention have a smoother, uniform surface and uniform distribution of the microcrystals of the taxane on the medical surface than is the case with the known coated medical products available on the market. These advantageous properties of the at least one tri-*O*-acylglycerols selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol thus lead to optimal

active agent transfer and elution into the tissue, as well as optimal active agent distribution at the implantation site.

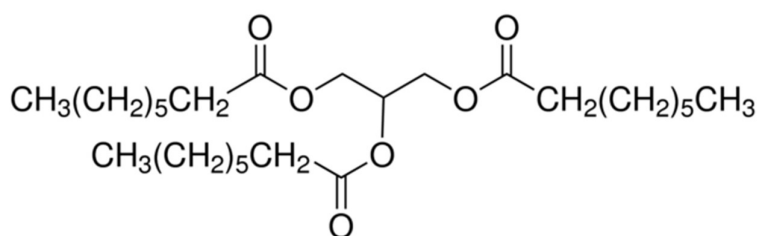
5 The tri-O-acylglycerols selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, or mixtures of these tri-O-acylglycerols thus fulfill the following important tasks, among others: They are the carriers of the microcrystalline taxane and thus influence the mechanical properties of the coating ("drug carrier") such as adhesion to the medical device surface, ensure the lowest possible loss of microcrystalline taxane during implantation ("drug transit loss"), but also influence the particle size distribution of the coating and thus the "crumbling behavior" ("particle release"), by which is meant the brittleness or pliability and adaptability of the coating before and during implantation and the associated change in shape. The uniformity of the coating ("uniformity") is considered to be another important parameter, since a uniform coating can also achieve a uniform distribution of the taxane microcrystals on the medical device surface and thus also a uniform distribution of the microcrystalline taxane into the surrounding tissue. In addition, as a "catalyst", they accelerate or facilitate the active agent transfer of the microcrystals of the taxane into the surrounding tissue without altering the microcrystalline taxane and influencing its efficacy ("drug transfer promoter").

20 The tri-O-acylglycerols according to the invention selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, or mixtures of said tri-O-acylglycerols are thus particularly advantageous for coatings of medical devices with microcrystals of a taxane, since the resulting coatings adhere to the surface of the medical device without loss, undergo changes in the shape of the substrate, e.g. elongations, without problems and without premature detachment or even dissolution, but do not prematurely "lose" the embedded or applied at least one microcrystalline taxane. Thereby, the coating does not suffer any damage even in case of severe shape changes, e.g. due to folding, expansion and deflation of such a coated balloon catheter.

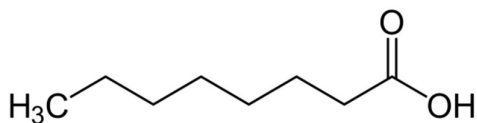
35 Thus, the objective of the present invention is solved with respect to sufficient active agent adhesion and active agent release of the microcrystalline taxane via the suspension according to the invention, which contains at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol in dissolved form.

The term "**trioctanoylglycerol**" or "tri-O-octanoylglycerol", as used herein, refers to a tri-O-acylglycerol in which the glycerol is fully esterified with caprylic acid or octanoic

acid. Synonymous names for trioctanoylglycerol known in the prior art are trioctanoylglyceride, glycerin trioctanoin, tricapryloylglycerin, octanoic acid-1,1',1''-(1,2,3-propanetriyl)ester, glycerin tricaprylate, tricaprylyl glycerin, TG(8:0/8:0/8:0), glycerol tricaprylate, caprylic acid-1,2,3-propanetriyl ester, caprylin, octanoic acid triglyceride, tricapryl glyceride, tricaprylin, trioctanoyl glycerol, octanoic acid-1,2,3-propanetriyl ester, glycerol trioctanoate, 1,2,3-propanetriol trioctanoate, caprylic acid triglyceride, glycerol tricaprylate, caprylic triglyceride, tricaprilin, tricaprylyl glycerol, 1,2,3- trioctanoyl glycerol, glycerol trioctanoate and 1,2,3-tricapryloyl glycerol. Trioctanoylglycerol has CAS number 538-23-8, a molecular weight of 470.68 g/mol, and has the following structural formula:



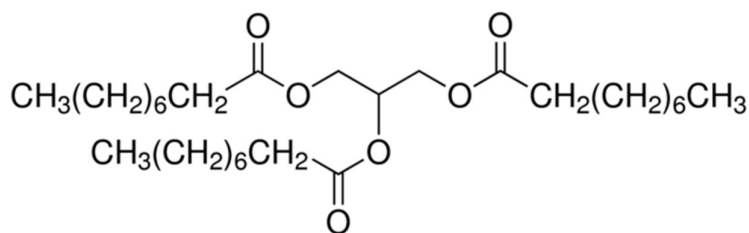
Octanoic acid is a carboxylic acid known by the trivial name caprylic acid and is a saturated fatty acid of the following structural formula:



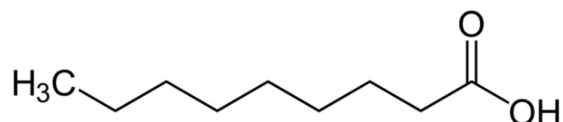
In preferred embodiments of the present invention, the suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, comprises at least trioctanoylglycerol. In preferred embodiments of the present invention, the at least one tri-O-acylglycerol is thus preferably selected from trioctanoylglycerol.

The melting point of trioctanoylglycerol is in the range of 9-10°C. Trioctanoylglycerol exists under normal conditions (20°C, 101 hPa) as an odorless clear colorless to amber liquid. Trioctanoylglycerol is practically insoluble in water.

The term "**trinonanoylglycerol**" or "tri-O-nonanoylglycerol", as used herein, refers to a tri-O-acylglycerol in which the glycerol is fully esterified with pelargonic acid or nonanoic acid. Synonymous names for tridecanoylglycerol known in the prior art are glycerol tripelargonate, trinonanoin, 1,2,3-trinonanoylglycerol, tripleargonin, 1,2,3-tripelargonoyl-glycerol. Trinonanoylglycerol has CAS number 126-53-4, a molecular weight of 512.76 g/mol and has the following structural formula:

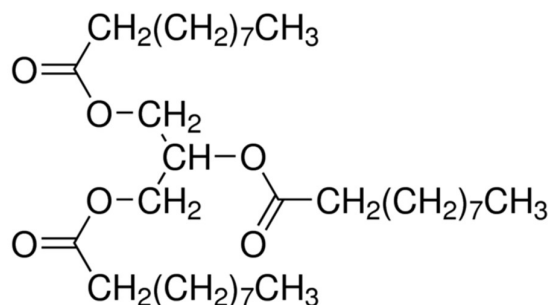


- Nonanoic acid is a carboxylic acid known by the trivial name pelargonic acid and is a saturated fatty acid of the following structural formula:

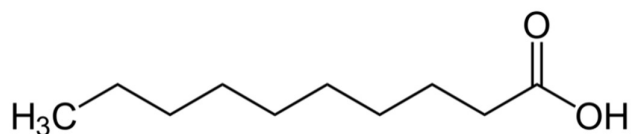


- In preferred embodiments of the present invention, the suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, comprises at least trionanoylglycerol. In preferred embodiments of the present invention, the at least one tri-O-acylglycerol is thus preferably selected from trionanoylglycerol.
- The melting point of trionanoylglycerol is in the range of 8-9°C. Trionanoylglycerol exists as a liquid under normal conditions (20°C, 101 hPa). Trionanoylglycerol is practically insoluble in water.

- The term "**tridecanoylglycerol**" or "tri-O-decanoylglycerol", as used herein, refers to a tri-O-acylglycerol in which the glycerol is fully esterified with capric acid or decanoic acid. Synonymous names for tridecanoylglycerol known in the prior art are glycerol tris-(decanoate), 1,2,3-tricaprinoyl-glycerol, tricaprin, 1,2,3-tridecanoylglycerol, glycerol tridecanoate, tridecanoin. Tridecanoylglycerol has CAS number 621-71-6, a molecular weight of 554.84 g/mol, and has the following structural formula:



Decanoic acid is a carboxylic acid known by the trivial name capric acid and is a saturated fatty acid of the following structural formula:



5

In preferred embodiments of the present invention, the suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, comprises at least tridecanoylglycerol. In preferred embodiments of the present invention, the at least one tri-*O*-acylglycerol is thus preferably selected from tridecanoylglycerol.

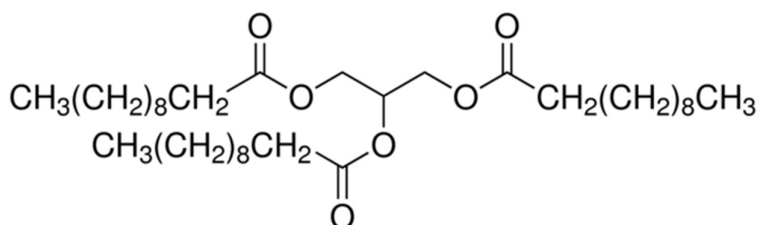
10

The melting point of tridecanoylglycerol is in the range of 31-33°C. Tridecanoylglycerol exists as a pale yellow solid under normal conditions (20°C, 101 hPa). Tridecanoylglycerol is practically insoluble in water.

15

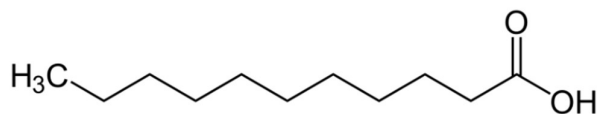
The term "**triundecanoylglycerol**" or "tri-*O*-undecanoylglycerol", as used herein, refers to a tri-*O*-acylglycerol in which the glycerol is fully esterified with undecanoic acid. Synonymous names for triundecanoylglycerol known in the prior art are glycerol triundecanoate, triundecanoin, 1,2,3-triundecanoylglycerol, triundecanin. The IUPAC name is 1,3-bis(undecanoyloxy)propan-2-yl-undecanoate. Triundecanoylglycerol has the CAS number 13552-80-2, a molecular weight of 596.9 g/mol, and has the following structural formula:

20



25

Undecanoic acid is a saturated fatty acid with the following structural formula:



In preferred embodiments of the present invention, the suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent,

30

or a cannula, comprises at least triundecanoylglycerol. In preferred embodiments of the present invention, the at least one tri-O-acylglycerol is thus preferably selected from triundecanoylglycerol.

- 5 The melting point of triundecanoylglycerol is in the range of 30-32°C. Triundecanoylglycerol exists as a solid under normal conditions (20°C, 101 hPa). Triundecanoylglycerol is practically insoluble in water.

Thus, the suspension of the present invention for coating of a medical device,
10 preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, contains at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol according to the invention.

- 15 In other words, the suspension of the present invention for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, contains according to the invention at least one tri-O-acylglycerol selected from the group consisting of tri-O-octanoylglycerol, tri-O-nonanoylglycerol, tri-O-decanoylglycerol, and tri-O-undecanoylglycerol.

20 With other words, the suspension of the present invention for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, contains according to the invention either a tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol,
25 and triundecanoylglycerol, or a mixture of at least two tri-O-acylglycerols selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol.

Still differently formulated, the suspension of the present invention for coating of a
30 medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, contains according to the invention either a tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, or a mixture of two, three or four tri-O-acylglycerols selected from the group consisting of trioctanoylglycerol,
35 trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol.

Thus, the phrase "at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol" herein also refers to mixtures of the tri-O-acylglycerol selected from the group

consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol. Thus, the phrase "at least one tri-O-acylglycerol" includes wording such as "at least two tri-O-acylglycerols" "at least three tri-O-acylglycerol" "two tri-O-acylglycerols" "three tri-O-acylglycerols" and "four tri-O-acylglycerols".

In some embodiments of the present invention, the suspension for coating of medical devices preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, contains at least two tri-O-acylglycerols selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol.

In some embodiments of the present invention, the suspension for coating of medical devices preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, contains at least three tri-O-acylglycerols selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol.

In some embodiments of the present invention, the suspension for coating medical devices preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, contains trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol.

The tri-O-acylglycerols selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol according to the invention have a melting point below 37°C, so they are present in molten form at body temperature or melt or soften at body temperature.

In some embodiments of the present invention, it is preferred that tri-O-acylglycerols present as a liquid under normal conditions (20°C, 101 hPa), such as trioctanoylglycerol or trinonanoylglycerol, especially preferably trioctanoylglycerol, are used to prepare the suspensions according to the invention. Especially with trioctanoylglycerol excellent stable, non-brittle and flexible coatings could be obtained. In some embodiments, mixtures of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol containing at least trioctanoylglycerol and/or trinonanoylglycerol, particularly preferably trioctanoylglycerol, are also preferred, since the resulting mixtures are present as a liquid under normal conditions (20°C, 101 hPa). A particularly preferred mixture of tri-O-acylglycerols herein is a mixture of trioctanoylglycerol and tridecanoylglycerol.

In some further embodiments of the present invention, it is preferred that tri-*O*-acylglycerols present as solids under normal conditions (20°C, 101 hPa), such as tridecanoylglycerol, or triundecanoylglycerol, are used to prepare the suspensions according to the invention. Especially with tridecanoylglycerol, excellent stable, non-
5 friable and flexible coatings could be obtained. Since body temperature is still higher than the melting points of these tri-*O*-acylglycerols, tridecanoylglycerol, or triundecanoylglycerol are present in molten form at body temperature during implantation, so that especially during inflation of medical devices, such as catheter balloons or stents, there are no disadvantages with regard to particle release and thus
10 friability compared to tri-*O*-acylglycerols that are present as a liquid under normal conditions (20°C, 101 hPa), such as trioctanoylglycerol or trinonanoylglycerol. In addition, if necessary or desired, the coated medical device can also be heated prior to implantation so that the tridecanoylglycerol, or triundecanoylglycerol melts or softens prior to implantation and is thus already in molten form at the beginning of implantation.

15 In some preferred embodiments of the present invention, the suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula thus contains at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and
20 triundecanoylglycerol, further preferably at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol and tridecanoylglycerol, still further preferably at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol and tridecanoylglycerol, further preferably at least one tri-*O*-acylglycerol selected from trioctanoylglycerol or at least one tri-*O*-acylglycerol selected
25 from tridecanoylglycerol.

In some further preferred embodiments of the present invention, the suspension for coating of a medical device preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula contains a mixture of tri-*O*-acylglycerols selected from
30 the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol. In some of these preferred embodiments, the suspension for coating of medical devices preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula contains a mixture of trioctanoylglycerol, trinonanoylglycerol and tridecanoylglycerol, or a mixture of trioctanoylglycerol,
35 trinonanoylglycerol and triundecanoylglycerol, or a mixture of trioctanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, or a mixture of trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, further preferably a mixture of trioctanoylglycerol, trinonanoylglycerol and tridecanoylglycerol, or a mixture of

trioctanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, still more preferably a mixture of trioctanoylglycerol, trinonanoylglycerol and tridecanoylglycerol.

5 In some preferred embodiments, the suspension for coating a of medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, contains a mixture of trioctanoylglycerol and tridecanoylglycerol, or a mixture of trioctanoylglycerol and trinonanoylglycerol, or a mixture of trioctanoylglycerol and triundecanoylglycerol, or a mixture of trinonanoylglycerol and tridecanoylglycerol, or a mixture of trinonanoylglycerol and triundecanoylglycerol, or a mixture of
10 tridecanoylglycerol, and triundecanoylglycerol.

In some preferred embodiments of the present invention, the suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, contains a mixture of tridecanoylglycerol, and trinonanoylglycerol,
15 or a mixture of tridecanoylglycerol, and triundecanoylglycerol, or a mixture of tridecanoylglycerol, and trioctanoylglycerol, further preferably a mixture of tridecanoylglycerol, and trinonanoylglycerol, or a mixture of tridecanoylglycerol, and trioctanoylglycerol, and most preferably a mixture of tridecanoylglycerol, and trioctanoylglycerol.

20 In some preferred embodiments of the present invention, the suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, contains a mixture of trioctanoylglycerol and trinonanoylglycerol, or a mixture of trioctanoylglycerol and triundecanoylglycerol, or a mixture of
25 trioctanoylglycerol and tridecanoylglycerol, further preferably a mixture of trioctanoylglycerol and trinonanoylglycerol, or a mixture of trioctanoylglycerol and tridecanoylglycerol, and most preferably a mixture of trioctanoylglycerol and tridecanoylglycerol.

30 In most preferred embodiments of the present invention, the suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, contains at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol and tridecanoylglycerol, or a mixture of trioctanoylglycerol and tridecanoylglycerol.

35 In some preferred embodiments of the present invention, the suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, contains at least one tri-O-acylglycerol selected trioctanoylglycerol, or a mixture of trioctanoylglycerol and at least one

tri-O-acylglycerol selected from the group consisting of trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, preferably trinonanoylglycerol and tridecanoylglycerol, even more preferably tridecanoylglycerol.

- 5 In some preferred embodiments of the present invention, the suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, contains trioctanoylglycerol, or a mixture of trioctanoylglycerol and at least one further tri-O-acylglycerol selected from trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, preferably trinonanoylglycerol and
10 tridecanoylglycerol, even more preferably tridecanoylglycerol.

- In some preferred embodiments of the present invention, the suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, contains at least one tri-O-acylglycerol selected from
15 tridecanoylglycerol, or a mixture of tridecanoylglycerol, and at least one further tri-O-acylglycerol selected from the group consisting of trinonanoylglycerol, trioctanoylglycerol and triundecanoylglycerol, preferably trinonanoylglycerol and trioctanoylglycerol, even more preferably trioctanoylglycerol.

- 20 In some preferred embodiments of the present invention, the suspension for coating a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, contains tridecanoylglycerol, or a mixture of tridecanoylglycerol, and at least one further tri-O-acylglycerol selected from the group consisting of trinonanoylglycerol, trioctanoylglycerol and triundecanoylglycerol, preferably
25 trinonanoylglycerol and trioctanoylglycerol, even more preferably trioctanoylglycerol.

- The at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol preferably has at least a purity of $\geq 90\%$, preferably $\geq 95\%$, and more preferably $\geq 99\%$.
30 A mixture of tri-O-acylglycerols selected from trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol preferably has at least $\geq 90\%$, preferably $\geq 95\%$ and more preferably $\geq 99\%$ of the tri-O-acylglycerols selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol.

- 35 Particularly preferably, the suspension according to the invention does not contain any other tri-O-acylglycerols in addition to the at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol.

For example, trioctanoylglycerol and tridecanoylglycerol are also present as natural components in various vegetable oils, such as soybean oil, olive oil or coconut oil, or animal oils. However, these natural vegetable oils or animal oils also contain other saturated and also unsaturated tri-*O*-acylglycerols in various proportions not according to the invention or also other substances such as mono-*O*-acylglycerols, di-*O*-acylglycerols, fatty acids and lipids, so that natural vegetable oils or animal oils are not suitable for the production of crystal suspensions according to the invention. Vegetable oils or animal oils can be used for the present invention, provided that they consist of at least $\geq 90\%$, preferably $\geq 95\%$ and particularly preferably of $\geq 99\%$ of at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol.

Mixtures of tri-*O*-acylglycerols not according to the invention are therefore herein oils such as linseed oil, hemp oil, corn oil, walnut oil, rapeseed oil, soybean oil, sunflower oil, poppy seed oil, safflower oil, wheat germ oil, safflower oil, grape seed oil, evening primrose oil, borage oil, black cumin oil, algae oil, fish oil, cod liver oil, coconut oil and/or mixtures of the aforementioned oils.

For the preparation of the suspension according to the invention containing at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, the at least one tri-*O*-acylglycerol is either used as a chemically pure substance for the preparation of the crystal suspension according to the invention or a mixture is used, that consists of at least $\geq 90\%$, preferably $\geq 95\%$ and more preferably $\geq 99\%$ of at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol. The at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol can of course also be obtained from a natural vegetable oil or animal oil, e.g. the tri-*O*-acylglycerols trioctanoylglycerol and tridecanoylglycerol, and mixtures of trioctanoylglycerol and tridecanoylglycerol, are commercially available under the trade names Captex[®] 8000 (trioctanoylglycerol), Captex[®] 1000 (tridecanoylglycerol), Captex[®] 300 (trioctanoylglycerol/tridecanoylglycerol), Captex[®] 355 (trioctanoylglycerol/tridecanoylglycerol), Miglyol[®] 810 (trioctanoylglycerol: Tridecanoylglycerol, approx. 70:30), and Miglyol[®] 812 (trioctanoylglycerol: tridecanoylglycerol, approx. 50:50). Such commercially available mixtures can be used to prepare a suspension of the present invention.

In some preferred embodiments of the present invention, the suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, thus contains a mixture of at least two tri-*O*-acylglycerols selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, wherein the mixture consists of at least $\geq 90\%$, preferably $\geq 95\%$, and more preferably $\geq 99\%$ of at least two tri-*O*-acylglycerols selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol.

The term "**weight percent**" (abbreviation: wt.%), as used herein, refers to the proportion of a substance in a mixture or solution measured in grams per 100 g of mixture. The term weight percent, as used herein, is a designation for the mass fraction of a mixture.

The term "**mass fraction**", as used herein, refers to a physicochemical quantity for the quantitative description of the composition of mixtures of substances. The mass of a considered mixture component is related to the sum of the masses of all mixture components, the mass fraction indicates the relative proportion of the mass of a considered mixture component in the total mass of the mixture.

In preferred embodiments, the mass fraction (m/m; [g]/[g]; m=mass) of tri-*O*-acylglycerol to the total mass of tri-*O*-acylglycerol and microcrystalline taxane in the suspension is preferably 5-40%, more preferably 10-30% and most preferably 20%. The mass of tri-*O*-acylglycerol here refers to the total mass of tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol and triundecanoylglycerol, i.e. in the case of mixtures of tri-*O*-acylglycerols selected from trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, the mass fraction of the mass of the mixture is calculated from the total mass of tri-*O*-acylglycerol mixture and microcrystalline taxane.

Thus, to calculate the mass fraction of tri-*O*-acylglycerol to the total mass of tri-*O*-acylglycerol and microcrystalline taxane in the suspension, the quotient of the mass of tri-*O*-acylglycerol and the total mass of tri-*O*-acylglycerol and microcrystalline taxane is formed. For example, the mass fraction of tri-*O*-acylglycerol in a mixture of 4 g of taxane and 1 g of tri-*O*-acylglycerol is 20%.

In preferred embodiments, the mass fraction (m/m; [g]/[g]; m=mass) of tri-*O*-acylglycerol to the total mass of tri-*O*-acylglycerol and microcrystalline taxane in

the suspension is preferably 0.1 - 50%, further preferably 1-40%, more preferably 10-30%, and most preferably 20%.

5 In preferred embodiments, the mass fraction (m/m ; $[g]/[g]$; $m=\text{mass}$) of taxane to the total mass of tri-*O*-acylglycerol and microcrystalline taxane in the suspension is preferably 99.9-50%, further preferably 99-60%, more preferably 90-70%, and most preferably 80%.

10 Thus, the mass fractions of tri-*O*-acylglycerol and microcrystalline taxane to the total mass of tri-*O*-acylglycerol and microcrystalline taxane are preferably 0.1-50% tri-*O*-acylglycerol and 99.9-50% taxane, further preferably 1- 40% tri-*O*-acylglycerol and 99-60% taxane, more preferably 10-30% tri-*O*-acylglycerol and 90-70% taxane and most preferably 20% tri-*O*-acylglycerol and 80% taxane.

15 If mixtures of tri-*O*-acylglycerols are used, the mass fraction of the mass of tri-*O*-acylglycerol mixture to the total mass of tri-*O*-acylglycerol mixture and microcrystalline taxane is determined. For example, the mass fraction of tri-*O*-acylglycerol mixture in a mixture of 4 g taxane and tri-*O*-acylglycerol mixture is 20%.

20 In preferred embodiments, the amount of tri-*O*-acylglycerol relative to the microcrystalline taxane is preferably 0.1 - 50 wt.%, further preferably 1 – 40 wt.%, more preferably 10 - 30 wt.%, and most preferably 20 wt.%.

25 In preferred embodiments, the amount of microcrystalline taxane relative to the tri-*O*-acylglycerol is preferably 99.9-50 wt.%, further preferably 99 - 60 wt.%, more preferably 90 - 70 wt.% and most preferably 80 wt.% in the suspension.

30 Thus, the tri-*O*-acylglycerol and the microcrystalline taxane in the suspension are preferably present at 0.1 - 50 wt.% tri-*O*-acylglycerol to 99.9 - 50 wt.% taxane, further preferably at 1 - 40 wt. % tri-*O*-acylglycerol to 99.60 wt.% taxane, particularly preferably with 10 - 30 wt.% tri-*O*-acylglycerol to 90 - 70 wt.% taxane and most preferably with 20 wt.% tri-*O*-acylglycerol and 80 wt.% taxane.

35 In preferred embodiments, the mass fraction (m/m ; $[g]/[g]$; $m=\text{mass}$) of additives in the total mass of additives and microcrystalline taxane in the suspension is preferably 0.001 - 1.0%, further preferably 0.01 - 2.5%, more preferably particularly preferably 0.05 - 5.0%.

The term "**mass ratio**", as used herein, refers to a physicochemical quantity for the quantitative description of the composition of mixtures of substances. The mass ratio indicates the ratio of the masses of two considered mixture components to each other.

- 5 The mass ratio (m/m; [g]/[g]; m=mass) of tri-O-acylglycerol to microcrystalline taxane is preferably 1:1000 - 1:1, more preferably 1:100 - 1:1.5, further preferably 1:10 - 1:3, and more preferably 1:5 - 1:4, and most preferably 1:4. The mass of tri-O-acylglycerol here refers to the total mass of tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
 10 i.e. for mixtures of tri-O-acylglycerols selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, the mass of the tri-O-acylglycerol mixture is used to calculate the mass ratio.

- Thus, to calculate the mass ratio of tri-O-acylglycerol to microcrystalline taxane, the
 15 quotient of the mass of tri-O-acylglycerol and the mass of taxane is formed. For example, the mass ratio of tri-O-acylglycerol to microcrystalline taxane in a mixture of 1 g tri-O-acylglycerol and 4 g taxane is 1:4.

- The mass ratio of tri-O-acylglycerol to microcrystalline taxane is further preferably 0.1-
 20 50%, more preferably 1-40%, more preferably 10-30%, and further preferably 20-25%, and most preferably 25%. In some preferred embodiments, the mass ratio of tri-O-acylglycerol to microcrystalline taxane is preferably 5-40%, more preferably 10-30%, and most preferably 20-25%.

- 25 Thus, to calculate the mass ratio of tri-O-acylglycerol to microcrystalline taxane, the quotient of the mass of tri-O-acylglycerol and the mass of taxane is formed. For example, the mass ratio of tri-O-acylglycerol to microcrystalline taxane in a mixture of 1 g tri-O-acylglycerol and 4 g taxane is 25%.

- 30 In preferred embodiments, the mass ratio (m/V; [g]/[100mL]; m=mass; V=volume) of microcrystalline taxane to 100 mL suspension volume is 0.5-6%, more preferably 1-5%, even more preferably 2-4%, and most preferably 3%.

- Thus, to calculate the mass ratio of microcrystalline taxane to 100 mL suspension
 35 volume, the quotient of the mass of microcrystalline taxane and 100mL suspension volume is formed. For example, the mass ratio of microcrystalline taxane to 100mL suspension volume in a suspension of 3 g taxane and 100mL suspension volume is 3%.

Based on a volume of 1 L suspension volume, the mass ratio (m/V ; $[g]/[L]$; m =mass; V =volume) of microcrystalline taxane to 1L suspension volume is 0.5-6%, more preferably 1-5%, still more preferably 2-4%, and most preferably 3%.

5 Thus, to calculate the mass ratio of microcrystalline taxane to 1 L suspension volume, the quotient of the mass of microcrystalline taxane to 1 L suspension volume is formed. For example, the mass ratio of microcrystalline taxane to 1 L suspension volume in a suspension of 30g taxane and 1 L suspension volume is 3%.

10 In preferred embodiments, the mass ratio (m/V ; $[g]/[100mL]$; m =mass; V =volume) of tri-*O*-acylglycerol to 100 mL suspension volume is 0.13-1.5%, more preferably 0.25-1.25%, even more preferably 0.5-1%, and most preferably 0.75%.

15 Thus, to calculate the mass ratio of tri-*O*-acylglycerol to 100mL suspension volume, the quotient of the mass of tri-*O*-acylglycerol and 100mL suspension volume is formed. For example, the mass ratio of tri-*O*-acylglycerol to 100mL suspension volume in a suspension of 0.75g tri-*O*-acylglycerol and 100mL suspension volume is 0.75%.

20 In other words, the mass ratio (m/V ; $[g]/[L]$; m =mass; V =volume) of tri-*O*-acylglycerol to 1L suspension volume is 0.13-1.5%, more preferably 0.25-1.25%, even more preferably 0.5-1% and most preferably 0.75%. Thus, to calculate the mass ratio of tri-*O*-acylglycerol to 1L suspension volume, the quotient of the mass of tri-*O*-acylglycerol and 1L suspension volume is formed. For example, the mass ratio of tri-*O*-acylglycerol to 1L suspension volume in a suspension of 7.5g taxane and 1L suspension volume is
25 0.75%.

The term "**mass concentration**", as used herein, refers to a physicochemical quantity for the quantitative description of the composition of substance mixtures/mixed phases. Here, the mass of a considered mixture component is related to the total volume of the
30 mixture phase.

In preferred embodiments, the mass concentration (m/V ; $[mg]/[mL]$; m =mass; V =volume) of microcrystalline taxane in the suspension is preferably 5 – 60 mg/mL, more preferably 20 – 40 mg/mL, further preferably 20 – 30 mg/mL. In some preferred
35 embodiments, the mass concentration of microcrystalline taxane in the suspension is 20 - 25 mg/mL. In some preferred embodiments, the mass concentration of microcrystalline taxane in the suspension is 25 - 30 mg/mL. In some preferred embodiments, the mass concentration of microcrystalline taxane in the suspension is 20 mg/mL. In some preferred embodiments, the mass concentration of microcrystalline

taxane in the suspension is 25 mg/mL. In some preferred embodiments, the mass concentration of microcrystalline taxane in the suspension is 30 mg/mL.

In other words, the mass concentration (m/V ; $[kg]/[m^3]$; m =mass; V =volume) of microcrystalline taxane in the suspension is preferably 5 – 60 kg/m^3 , more preferably 20 – 40 kg/m^3 , more preferably 20 – 30 kg/m^3 . In some preferred embodiments, the mass concentration of microcrystalline taxane in the suspension is 20 - 25 kg/m^3 . In some preferred embodiments, the mass concentration of microcrystalline taxane in the suspension is 25 - 30 kg/m^3 . In some preferred embodiments, the mass concentration of microcrystalline taxane in the suspension is 20 kg/m^3 . In some preferred embodiments, the mass concentration of microcrystalline taxane in the suspension is 25 kg/m^3 . In some preferred embodiments, the mass concentration of microcrystalline taxane in the suspension is 30 kg/m^3 .

In preferred embodiments, the mass concentration (m/V ; $[mg]/[mL]$; m =mass; V =volume) of tri-*O*-acylglycerol in the suspension is preferably 1.3 – 15 mg/mL, more preferably 2.5 – 12.5 mg/mL, more preferably 5 – 10 mg/mL.

In some preferred embodiments, the mass concentration of tri-*O*-acylglycerol in the suspension is 6 - 9 mg/mL. In some preferred embodiments, the mass concentration of tri-*O*-acylglycerol in the suspension is 5.5 - 9.5 mg/mL. In some preferred embodiments, the mass concentration of tri-*O*-acylglycerol in the suspension is 7.5 mg/mL. In some preferred embodiments, the mass concentration of tri-*O*-acylglycerol in the suspension is 7 mg/mL. In some preferred embodiments, the mass concentration of tri-*O*-acylglycerol in the suspension is 6.5 mg/mL.

In some preferred embodiments, the mass concentration of tri-*O*-acylglycerol in the suspension is 4 - 6 mg/mL. In some preferred embodiments, the mass concentration of tri-*O*-acylglycerol in the suspension is 4.5 - 5.5 mg/mL. In some preferred embodiments, the mass concentration of tri-*O*-acylglycerol in the suspension is 4.5 mg/mL. In some preferred embodiments, the mass concentration of tri-*O*-acylglycerol in the suspension is 5 mg/mL. In some preferred embodiments, the mass concentration of tri-*O*-acylglycerol in the suspension is 5.5 mg/mL.

In preferred embodiments, the mass concentration (m/V ; $[mg]/[mL]$; m =mass; V =volume) of microcrystalline taxane in the suspension is 5 - 60 mg/mL and of tri-*O*-acylglycerol 1.3 - 15 mg/mL, further preferred is a mass concentration of microcrystalline taxane in the suspension of 20 - 40 mg/mL and of tri-*O*-acylglycerol

2.5 - 12.5 mg/mL, further preferred is a mass concentration of microcrystalline taxane in the suspension of 20 - 30 mg/mL and of tri-O-acylglycerol 5 - 10 mg/mL .

5 The term "**amount of substance concentration**", as used herein, refers to a physicochemical quantity for the quantitative description of the composition of substance mixtures/mixed phases. Here, the amount of substance of a considered mixture component is related to the total volume of the mixed phase.

10 In preferred embodiments, the molar concentration (n/V ; [mmol]/[L]; n =amount of substance; V =volume) of microcrystalline taxane in the suspension is preferably 5 – 60 mmol/L, more preferably 20 – 40 mmol/L, more preferably 25 - 35 mmol/L.

15 In other words, the molar concentration (n/V ; [mmol]/[L]; n =amount of substance; V =volume) of microcrystalline taxane in the suspension is preferably 5 – 60 mol/m³, more preferably 20 – 40 mol/m³, more preferably 25 - 35 mol/m³.

20 In some preferred embodiments, the molar concentration (n/V ; [mmol]/[L]; n =amount of substance; V =volume) of tri-O-acylglycerol in the suspension is 2 - 35 mmol/L, more preferably 4 - 25 mmol/L, more preferably 10 - 20 mmol/L, more preferably 12 - 18 mmol/L.

25 In some preferred embodiments, the molar concentration (n/V ; [mmol]/[L]; n =amount of substance; V =volume) of tri-O-acylglycerol the suspension is 2 - 35 mol/m³ , more preferably 4 - 25 mol/m³, more preferably 10 - 20 mol/m³, more preferably 12 - 18 mol/m³ .

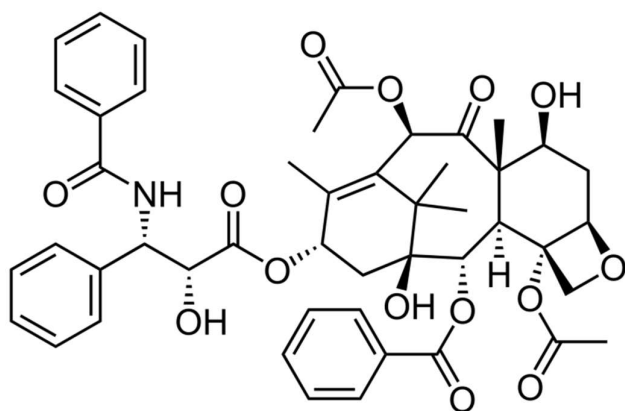
30 The term "**taxanes**", as used herein, refers to a group of active agents belonging to the cytostatic drugs and are also known as taxoids. They are used in the treatment of various cancers where they serve as chemotherapeutic agents. Taxanes such as paclitaxel are known from the prior art as suitable agents for coatings of medical devices, such as stents and catheter balloons, to prevent restenosis. Taxanes inhibit cell division and thus tumor growth by inhibiting the degradation of the spindle apparatus, rendering it inoperable for its essential function in mitosis. The microtubules that form the spindle apparatus are essential for the distribution of the duplicated genetic material to the two daughter cells in the course of cell division. The term "taxane" as used herein refers to naturally occurring taxanes, such as paclitaxel and derivatives thereof, such as docetaxel.

According to the invention, the following substances can be used as taxane:

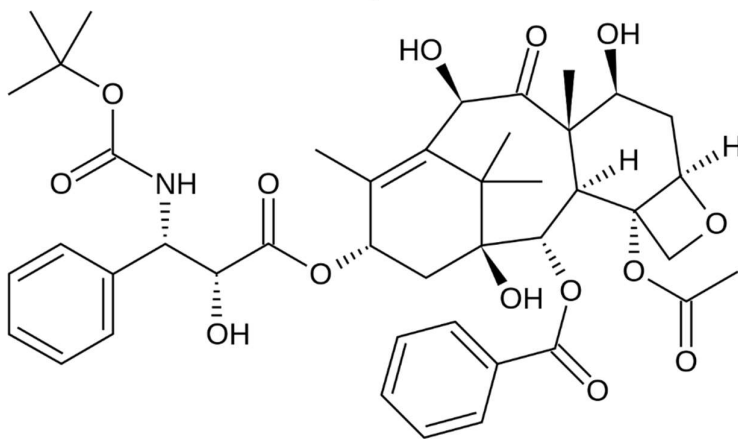
paclitaxel, taxotere, baccatin, 7-xylosyl-10-deacetyltaxol, cephalomannine, 10-deacetyl-7-epitaxol, 7-epitaxol, 10-deacetylcephalomannine, docetaxel, 7-deoxydocetaxol, 7,8-cyclopropataxanes, N-substituted 2-azetidones, 6,7-epoxy-paclitaxels, 6,7-modified paclitaxels, 10-deacetoxytaxol, 10-deacetyltaxol (from 10-deacetyl-
 5 deacetyl-*baccatin* III), phosphonoxy and carbonate derivatives of taxol, taxol 2',7-disodium 1,2-benzenedicarboxylate, 10-deacetoxy-11,12-dihydrotaxol-10,12(18)-diene derivatives, 10-deacetoxytaxol, protaxol (2'- and/or 7-O-ester derivatives), 2'- and/or 7-O-carbonate derivatives of paclitaxel, fluorotaxols, 9-deoxotaxane, 13-acetyl-9-deoxobaccatin III, 9-deoxotaxol, 7-deoxy-9-deoxotaxol, 10-deacetoxy-7-deoxy-9-
 10 deoxotaxol, sulfonated 2'-acryloyltaxol, sulfonated 2'-O-acyltaxol derivatives, succinyltaxol, 2'-gamma-aminobutyryltaxol, 2'-acetyltaxol, 7-acetyltaxol, 7-glycine-carbamate-taxol, 2'-benzoyl derivatives of paclitaxel, 2',7-dibenzoyl-taxol, 2'-acetyltaxol; 2',7-diacetyltaxol; 2'-succinyltaxol; 2'-(beta-alanyl)-taxol; ethylene glycol derivatives of 2'-succinyltaxol; 2'-glutaryltaxol; 2'-(N,N-dimethylglycyl)-taxol; 2'-(2-(N,
 15 N-dimethylamino)propionyl)taxol, 2'-orthocarboxybenzoyl-taxol, 2'-aliphatic carboxylic acid derivatives of taxol, 2'-(N,N-diethylaminopropionyl)taxol, 7-(N,N-dimethylglycyl)taxol, 2',7-di-(N,N-dimethylglycyl)taxol, 7-(N,N-diethylaminopropionyl)taxol, 2',7-di(N,N-diethylaminopropionyl)taxol, 2'-(L-glycyl)taxol, 7-(L-glycyl)taxol, 2',7-di(L-glycyl)taxol, 2'-(L-alanyl)taxol, 7-(L-alanyl)taxol,
 20 2',7-di(L-alanyl)taxol, 2'-(L-leucyl)taxol, 7-(L-leucyl)taxol, 2',7-di(L-leucyl)taxol, 2'-(L-isoleucyl)taxol, 7-(L-isoleucyl)taxol, 2',7-di(L-isoleucyl)taxol, 2'-(L-valyl)taxol, 7-(L-valyl)taxol, 2',7-di(L-valyl)taxol, 2'-(L-phenylalanyl)taxol, 7-(L-phenylalanyl)taxol, 2',7-di(L-phenylalanyl)taxol, 2'-(L-prolyl)taxol, 7-(L-prolyl)taxol, 2',7-di(L-prolyl)taxol, 2'-(L-lysyl)taxol, 7-(L-lysyl)taxol, 2',7-di(L-lysyl)taxol, 2'-(L-glutamyl)taxol, 7-(L-glutamyl)taxol, 2',7-di(L-glutamyl)taxol, 2'-(L-arginyl)taxol, 7-(L-arginyl)taxol, 2',7-di(L-arginyl)taxol, (N-debenzoyl-N-tert-(butoxycaronyl)-10-deacetyltaxol, baccatin III, 10-deacetyl-*baccatin* III, cabazitaxel, brevifoliol, yunantaxusin, taxusin, 14-beta-hydroxy-10-deacetyl-*baccatin* III, debenzoyl-2-acylpaclitaxel derivatives, benzoate paclitaxel derivatives, 18-side-chain substituted paclitaxel derivatives, chlorinated paclitaxel
 30 analogs, C4 methoxyether paclitaxel derivatives, sulfonamide taxane derivatives, brominated paclitaxel analogs, Girard's taxol derivatives, nitrophenylpaclitaxel, 10-deacetyl-*baccatin* III substituted paclitaxel derivatives, 14-beta-hydroxy-10-deacetyl-*baccatin* III derivatives, C7 taxane derivatives, 2-debenzoyl-2-acyltaxane derivatives, 2-debenzoylpaclitaxel derivatives, 2-acylpaclitaxel derivatives, 10-deacetyltaxol A, 10-deacetyltaxol B, 2-aroxy-4-acyl paclitaxel analogs, ortho-ester paclitaxel analogs, baccatin VII; baccatin VI; baccatin IV; 7-epi-baccatin III; baccatin V; baccatin I; baccatin A; epitaxol.

The at least one taxane is preferably selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel, particularly preferred is paclitaxel.

Paclitaxel is known under the brand name Taxol® and the chemical name [2aR-[2a,4,4a,6,9 (R*,S*),11,12,12a,12b]] - (benzoylamino) - hydroxybenzenepropionic acid-6,12b-bis-(acetyloxy)-12-(benzoyloxy)-2a-3, 4, 4a, 5, 6, 9, 10, 11, 12, 12a, 12b-dodecahydro-4,11-dihydroxy-4a,8,13,13-tetramethyl-5-oxo-7,11-methano-1H-cyclodeca [3,4] benz[1,2-b] oxet-9-yl ester), or (2 α ,4 α ,5 β ,7 β ,10 β ,13 α)-4,10-Bis(acetyloxy)-13-[[[(2R,3S)-3-(benzoylamino)-2-hydroxy-3-phenylpropanoyl]oxy]-1,7-dihydroxy-9-oxo-5,20-epoxytax-11-en-2-yl-benzoat (IUPAC-Name) bekannt. Paclitaxel has the following structural formula:

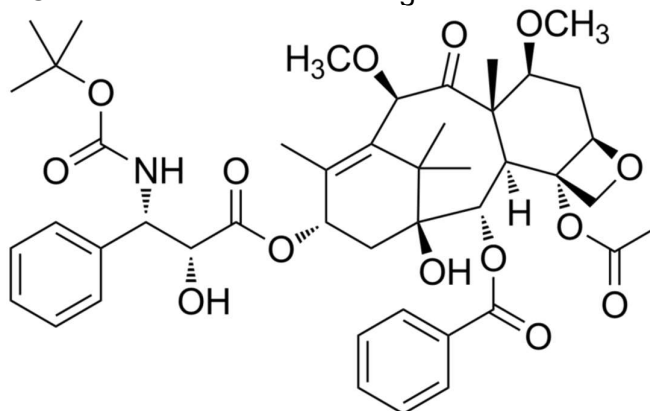


Docetaxel is known by the brand name Taxotere® and the chemical name (2R,3S)-4-acetoxy-2 α -benzyloxy-13-[3-(N-*tert*-butoxycarbonyl)amino-2-hydroxy-3-phenyl]propionyl-5 β ,20-epoxy-1,7 β ,10 β -trihydroxy-9-oxotax-11-en-13 α -ylester (IUPAC name). Docetaxel has the following structural formula:



Cabazitaxel is available under the brand name Jevtana® and known under the chemical name (1S,2S,3R,4S,7R,9S,10S,12R,15S)-4-(acetyloxy)-15-[[[(2R,3S)-3-

{[(tert-butoxy)carbonyl]amino}-2-hydroxy-3-phenylpropanoyl[oxy]-1-hydroxy-9,12-dimethoxy-10,14,17,17-tetramethyl-11-oxo-6-oxatetracyclo[11.3.1.^{03,10}.0^{4,7}]heptadec-13-en-2-ylbenzoat. Cabazitaxel has the following structural formula:



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The term "**microcrystals**", as used herein, refers to solids whose building blocks are regularly arranged in a crystal structure and have a size in the micrometer range. The term "micrometer range" as used herein corresponds to the range from 1 μm to 300 μm , where 1 μm corresponds to 10^{-6} m, 10^{-3} mm, or 1000 nm. Thus, the term

10 microcrystals, as used herein, refers to crystals having a crystal size in the range of 1 μm to 300 μm .

The term "**crystal size**," as used herein, refers to the length of the crystals along their largest dimension, i.e., along their longitudinal axis in the case of rod-shaped or needle-shaped crystals. Thus, the microcrystals, as defined herein, have a length in the range

15 of 1 μm to 300 μm along their largest dimension.

The term "**crystallinity**," as used herein, is the crystalline content of a compound, that is, the proportion of crystals of a compound to the total amount of that compound in

20 crystalline and other forms.

The term "**microcrystalline taxane**", as used herein, refers to a taxane that is in the form of microcrystals. Thus, the term "microcrystalline taxane" and "taxane in the form of microcrystals" are used interchangeably herein.

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Crystallization processes for the preparation of taxanes are known from the prior art. Generally, to crystallize taxanes, a solution of a taxane can be prepared and the solubility of the taxane in the solution can be reduced. Common methods for reducing solubility include, for example, cooling, addition of an anti-solvent, and evaporation.

30

Crystallization by cooling: The taxane can be dissolved in a solvent at room temperature or higher temperature until saturation and brought to crystallization at lower temperature e.g. at 0°C. The crystal size distribution can be influenced by a controlled cooling rate. Both polar and non-polar organic solvents, such as toluene, acetonitrile, ethyl formate, isopropyl acetate, isobutyl acetate, ethanol, dimethyl formamide, anisole, ethyl acetate, methyl ethyl ketone, methyl isopropyl ketone, tetrahydrofuran, nitromethane, propionitrile are suitable solvents for crystallization of taxanes.

- 10 Crystallization by addition of seed crystals: The taxane is dissolved to saturation in a solvent and crystallization is initiated by the addition of seed crystals to achieve a controlled reduction of supersaturation.

Crystallization by addition of anti-solvent: The taxane is dissolved in a solvent and then a non-solvent or water is added. Two-phase mixtures are also possible here. Polar organic solvents such as acetone, acetonitrile, ethyl acetate, methanol, ethanol, isopropanol, butanol, butyl methyl ether, tetrahydrofuran, dimethyl formamide or dimethyl sulfoxide can be used as solvents for dissolving the taxane. Suitable non-solvents include pentane, hexane, cyclohexane or heptane. The solvent mixture can be allowed to stand for crystallization, stirred or slowly concentrated or evaporated *in vacuo*. The crystal size and crystallinity of the taxane can be influenced by controlled addition of the nonpolar solvent. Supersaturation should be slower to produce large crystals and faster to produce small crystals. Controlling the addition rate of the anti-solvent to control the crystal size is well known.

25 For the production of microcrystals, crystallization can also be assisted by ultrasound. It is generally known that crystal size can be influenced by means of ultrasound. Ultrasound can be used at the beginning of crystallization to initiate crystallization and nucleation, with further crystal growth then proceeding unhindered so that larger crystals can grow. On the other hand, the application of continuous ultrasonic sonication of a supersaturated solution leads to smaller crystals, as many nuclei are formed in this process, resulting in the growth of numerous small crystals. Another option is to use ultrasound in pulse mode to influence crystal growth in such a way that tailored crystal sizes are achieved.

35 Herein, preferred crystallization processes for the production of microcrystalline taxanes are controlled crystallization to obtain microcrystals in native and intact state and to avoid possible damage, e.g. by milling or micronization.

Other processes known from the prior art, such as micronization, grinding or sieving, can also be used to provide the desired crystal sizes. One possibility is to grind the crystals, which can also be done during crystallization by wet grinding. Milling can be advantageous to obtain different crystal sizes i.e. a broader crystal size distribution.

5 Milling allows for all desired sizes in the crystal size range. More uniform crystal sizes can be provided by performing, for example, a special sieving process after isolation and drying. Special sieving devices known from the prior art can be used for this purpose. In the sieving process, the taxane crystals can be sieved through a stack of sieves, for example, and divided into different size ranges.

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It could be shown that particularly stable suspensions can be prepared with microcrystalline taxanes if the suspensions contain at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol. According to the invention, the at least
15 one taxane is in the form of microcrystals. Thus, the taxane is present in the form of microcrystals having a crystal size in the range of 1 μm to 300 μm . The microcrystals of the at least one taxane therefore have a crystal size between 1 μm to 300 μm .

The microcrystals of the taxane are not encapsulated and are not coated, e.g. with a
20 polymer and are not modified on the surface. In addition, the microcrystals of taxane do not contain a polymer, polymer particles, metal, metal particles, ceramic or ceramic particles. Nor do they contain any other active pharmaceutical inactive agent or proteins, amino acid or nucleotides or other biopolymers.

25 The present invention therefore relates to a suspension comprising at least one microcrystalline taxane as defined herein. In the suspension, the at least one taxane is present in the form of microcrystals. The content of taxane dissolved in the solvent or solvent mixture in the suspension is less than 10%, preferably less than 5% and further preferably less than 2%, most preferably less than 1% based on the mass of
30 taxane in the form of microcrystals used in the preparation of the suspension. It is therefore preferred that a maximum of 10%, preferably a maximum of 5% and further preferably a maximum of 2%, most preferably a maximum of 1% of the microcrystals of the at least one taxane dissolve in the suspension.

35 According to the invention, the suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, contains a solvent or a solvent mixture in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve.

The phrase "in which the microcrystals of the at least one taxane do not dissolve", as used herein, means that preferably a maximum of 10%, preferably a maximum of 9%, further preferably a maximum of 8%, further preferably a maximum of 7%, further preferably a maximum of 6%, further preferably a maximum of 5%, still more preferably a maximum of 3%, further preferably a maximum of 2%, and most preferably a maximum of 1% of the microcrystals of the at least one taxane dissolve in the suspension. Preferably, of course, 100% of the microcrystals of the at least one taxane do not dissolve in the suspension.

It is further preferred that the solubility of the microcrystals of the at least one taxane in the solvent or solvent mixture of the suspension is < 20mg/mL, more preferably < 15mg/mL, more preferably < 10mg/mL, more preferably < 9mg/mL, more preferably < 8mg/mL, further preferably < 7mg/mL, further preferably < 6mg/mL, further preferably < 5mg/mL, further preferably < 4mg/mL, further preferably < 3mg/mL, further preferably < 2mg/mL, further preferably < 1mg/mL.

Surprisingly, it was found that microcrystals of taxanes do not dissolve in a solution containing at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol.

Thus, a crystal suspension can be prepared as a coating formulation in which the microcrystals of the taxane remain intact.

The crystals of the taxane should have a size of at least 1 μm . Crystals of less than 1 μm are too small, so they dissolve relatively quickly. In preparing the suspension of the present invention, it has been found that stable suspensions can be obtained only when the at least one taxane has substantially no crystals with a crystal size of less than 1 μm . In other words, the taxane is particularly preferably not in the form of nanocrystals. The term nanocrystals, as used herein, refers to crystals having a crystal size in the range of 1 nm to less than 1000 nm.

Preferably, at least 95% - 97% of the at least one taxane, more preferably at least 95% - 99%, further preferably at least 97% - 99%, and particularly preferably at least 98% - 99.9% of the at least one taxane is present in the form of microcrystals having a crystal size of at least 1 μm . In further preferred embodiments, 100% of the at least one taxane is present in the form of microcrystals with a crystal size of at least 1 μm .

It could also be shown that it is advantageous if the taxane in the form of microcrystals has a crystal size of at least 10 μm . Therefore, it is preferred that the at least one taxane has a small proportion of microcrystals with a crystal size of 1 μm - 10 μm . It is

particularly preferred that only a few crystals, i.e. significantly less than 10% of all crystals are smaller than 10 μm . In preferred embodiments, less than 10% of all microcrystals of the taxane are present with a crystal size in the range of less than 10 μm .

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Preferably, at least 90% of the at least one taxane, preferably at least 90% - 95% of the at least one taxane, more preferably at least 93% - 98% of the at least one taxane, more preferably at least 95% - 99% of the at least one taxane, and particularly preferably at least 98% - 99.9% of the at least one taxane is present in the form of

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microcrystals having a crystal size of at least 10 μm .

It is further preferred that the microcrystals of the at least one taxane have a crystal size of at least 5 μm . It is therefore preferred that at least 90% of the at least one taxane, preferably at least 90% - 95% of the at least one taxane, more preferably at

15

least 93% - 98% of the at least one taxane, further preferably at least 95% - 99% of the at least one taxane, and particularly preferably at least 98% - 99.9% of the at least one taxane is present in the form of microcrystals with a crystal size of at least 5 μm . Microcrystals with a crystal size in the range of smaller than 5 μm can dissolve faster and are therefore less preferred.

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Still further preferred is that the microcrystals of the at least one taxane have a crystal size of at least 20 μm . It is therefore preferred that at least 90% of the at least one taxane, preferably at least 90% - 95% of the at least one taxane, more preferably at least 93% - 98% of the at least one taxane, further preferably at least 95% - 99% of

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the at least one taxane and particularly preferably at least 98% - 99.9% of the at least one taxane is present in the form of microcrystals having a crystal size of at least 20 μm .

In addition, preferably only a few microcrystals of the taxane, i.e. less than 40% and more preferably less than 30% or even less than 25%, are present with a crystal size in the range of 100 μm - 300 μm . It is therefore preferred that a maximum of 40% of the at least one taxane, preferably a maximum of 30% of the at least one taxane, further preferably a maximum of 25% of the at least one taxane is present in the form of microcrystals with a particle size of 100 μm - 300 μm . In further embodiments, it is

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preferred that a maximum of 20% of the at least one taxane, preferably a maximum of 15% of the at least one taxane, further preferably a maximum of 10% of the at least one taxane is present in the form of microcrystals having a particle size of 100 μm - 300 μm . In particularly preferred embodiments, the microcrystals of the at least one taxane are substantially present with a crystal size of at most 100 μm .

Further preferably, very few microcrystals of the taxane are present, i.e. less than 10% and more preferably less than 5% or even less than 2%, and most preferably less than 1% with a crystal size in the range of 150 μm - 300 μm . Microcrystals with a crystal
 5 size in the range of 150 μm – 300 μm could form agglomerates and coalesce into amorphous particles, which may pose the risk of vascular occlusion. It is therefore particularly preferred if the proportion of microcrystals with a particle size in the range 150 μm – 300 μm is as low as possible.

10 It is therefore preferred that at most 10% of the at least one taxane, preferably at most 5% of the at least one taxane, further preferably at most 2% of the at least one taxane, still more preferably at most 1% of the at least one taxane is present in the form of microcrystals having a particle size of 150 μm – 300 μm . In further embodiments, it is preferred that at least 99%, preferably 99.5%, further preferably at least 99.7%, still
 15 further preferably at least 99.9% and most preferably 100% of the at least one taxane is present with a particle size of $\leq 150 \mu\text{m}$. In preferred embodiments, the microcrystals of the at least one taxane are substantially present with a crystal size of at most 150 μm . In particularly preferred embodiments, the microcrystals of the at least one taxane are substantially present with a crystal size of at most 150 μm .

20 Preferably, the taxane is in the form of microcrystals with a crystal size of 1 μm to 150 μm .

It is therefore preferred that at most 10% of the at least one taxane, preferably at most
 25 5% of the at least one taxane, further preferably at most 2% of the at least one taxane, still more preferably at most 1% of the at least one taxane is present in the form of microcrystals having a particle size of 100 μm – 300 μm . In further embodiments, it is preferred that at least 99%, preferably 99.5%, further preferably at least 99.7%, still further preferably at least 99.9% and most preferably 100% of the at least one taxane
 30 is present with a particle size of $\leq 100 \mu\text{m}$. In preferred embodiments, the microcrystals of the at least one taxane are substantially present with a crystal size of at most 100 μm . In particularly preferred embodiments, the microcrystals of the at least one taxane are substantially present with a crystal size of at most 100 μm .

35 Preferably, the taxane is in the form of microcrystals with a crystal size of 1 μm to 100 μm .

It could be shown that microcrystals of the taxane with a crystal size in the range of 10 μm to 100 μm are well suited for providing a suspension according to the invention for

coating of medical devices. It is therefore preferred that at least 70% of the at least one taxane, preferably at least 70% - 80% of the at least one taxane, further preferably at least 80% - 90% of the at least one taxane, further preferably at least 90% - 95% of the at least one taxane, and particularly preferably at least 95% - 99% of the at least one taxane is present in the form of microcrystals with a crystal size of 10 μm to 100 μm .

It could be shown that microcrystals of the taxane with a crystal size in the range of 10 μm to 80 μm are well suited for providing a suspension according to the invention for coating of medical devices. It is therefore preferred that at least 70% of the at least one taxane, preferably at least 70% - 80% of the at least one taxane, further preferably at least 80% - 90% of the at least one taxane, further preferably at least 90% - 95% of the at least one taxane, and particularly preferably at least 95% - 99% of the at least one taxane is present in the form of microcrystals with a crystal size of 10 μm to 80 μm .

It could be shown that microcrystals of the taxane with a crystal size in the range of 10 μm to 70 μm are well suited for providing a suspension according to the invention for coating of medical devices. It is therefore preferred that at least 70% of the at least one taxane, preferably at least 70% - 80% of the at least one taxane, further preferably at least 80% - 90% of the at least one taxane, further preferably at least 90% - 95% of the at least one taxane, and particularly preferably at least 95% - 99% of the at least one taxane is present in the form of microcrystals with a crystal size of 10 μm to 70 μm .

It could be shown that microcrystals of the taxane with a particle size in the range of 5 μm to 50 μm are suitable for providing a stable suspension for coating medical devices. It is therefore preferred that at least 70% of the microcrystals of the taxane are present with a crystal size in the range of 5 μm to 50 μm . It is therefore preferred that at least 70% of the taxane, preferably at least 70% - 80% of the taxane, further preferably at least 80% - 90% of the taxane is present in the form of microcrystals with a particle size ranging from 5 μm to 50 μm . It is further preferred that at least 70% of the at least one taxane, preferably at least 70% - 80% of the at least one taxane, further preferably at least 80% - 90% of the at least one taxane, further preferably at least 90% - 95% of the at least one taxane, and particularly preferably at least 95% - 99% of the at least one taxane is present in the form of microcrystals having a particle size of 5 μm to 35 μm .

In particularly preferred embodiments, the microcrystals of the taxane are present with a crystal size in the range of 20 μm to 60 μm . Preferably, therefore, at least 70% of the at least one taxane, preferably at least 70% - 80% of the at least one taxane, further preferably at least 80% - 90% of the at least one taxane, further preferably at least 90% - 95% of the at least one taxane, and particularly preferably at least 95% - 99% of the at least one taxane is present in the form of microcrystals having a crystal size ranging from 20 μm to 60 μm .

Preferably, the taxane has a crystallinity of at least 80% by weight, more preferably at least 85% by weight, more preferably at least 90% by weight, more preferably at least 92.5%, more preferably at least 95%, more preferably at least 97.5% and most preferably at least 99% by weight.

The microcrystals of the at least one taxane are preferably microcrystals of paclitaxel, docetaxel, or cabazitaxel.

Preferred herein are microcrystalline paclitaxel, microcrystalline docetaxel, and microcrystalline cabazitaxel. Particularly preferred herein is microcrystalline paclitaxel.

Crystals with prismatic to acicular habit are one-dimensional elongated forms in which the length of the crystal is significantly greater than its diameter.

In preferred embodiments, the at least one taxane is paclitaxel. Paclitaxel crystallizes in the form of needles. It is therefore preferred that at least 90%, more preferably at least 92.5%, more preferably at least 95%, more preferably at least 97.5%, and most preferably at least 99% of the microcrystals of paclitaxel are in the form of needles. It is therefore preferred that at least 90%, more preferably at least 92.5%, more preferably at least 95%, more preferably at least 97.5% and most preferably at least 99% of the microcrystals of paclitaxel are needle-shaped. It is therefore preferred that at least 90%, more preferably at least 92.5%, more preferably at least 95%, more preferably at least 97.5%, and most preferably at least 99% of paclitaxel is in the form of needle-shaped microcrystals.

The term "**solvent**", as used herein, refers to a substance that exists in the liquid state of aggregation at normal temperature (20°C) and normal pressure (101hPa; 1 bar, 1atm) and that can dissolve or dilute gases, liquids or solids without chemical reactions taking place between the dissolved substance and the solvent. Liquids such as water and liquid organic substances are used as solvents to dissolve other substances.

The term "**non-solvent**," as used herein, refers to a solvent that cannot dissolve or dissolve microcrystalline taxanes, i.e. a solvent in which microcrystalline taxanes are virtually insoluble, but in which the tri-*O*-acylglycerols selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, or mixtures of said tri-*O*-acylglycerols, are soluble.

The solubility of a microcrystalline taxane in a non-solvent should be at most 1 mg/mL. Examples of solvents in which the solubility of microcrystalline taxanes is at most 1 mg/mL are water and some nonpolar organic solvents such as saturated aliphatic hydrocarbons.

Examples of solvents in which the tri-*O*-acylglycerols trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol or mixtures of said tri-*O*-acylglycerols are soluble include, but are not limited to non-polar organic solvents such as hexane, heptane, cyclohexane, toluene, but also polar organic solvents such as diethyl ether, ethyl acetate, acetone, isopropanol and ethanol.

Tri-*O*-acylglycerols are nonpolar, i.e. lipophilic, and are sparingly or virtually insoluble in very polar solvents such as water or glycerol. A suspension containing as solvent exclusively a very polar solvent such as water or glycerol is thus not according to the invention, since the tri-*O*-acylglycerols selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol or a mixture of said tri-*O*-acylglycerols do not exist in dissolved form in very polar solvents such as water or glycerol.

The term "non-solvent", as used herein, thus refers to nonpolar organic solvents, particularly saturated aliphatic hydrocarbons. A "non-solvent" may therefore also be referred to as a "nonpolar organic non-solvent." Non-polar organic solvents referred to herein as "non-solvent" therefore include saturated aliphatic hydrocarbons that are liquid at normal temperature (20°C) and normal pressure (101hPa; 1 bar, 1atm), i.e. unbranched (linear) saturated hydrocarbons with the general molecular formula $C_n H_{2n+2}$ with $n = 5$ to 16, branched saturated hydrocarbons with the general molecular formula $C_n H_{2n+2}$ with $n = 4$ to 16, or cyclic saturated hydrocarbons with the general molecular formula $C_n H_{2n}$ with $n = 5$ to 16. Examples of non-solvers include, but are not limited to, unbranched C_{5-16} alkanes such as pentane, hexane, heptane, octane, nonane, decane, petroleum ether, branched C_{5-16} -alkanes (iso-alkanes) such as isopentane, isooctane, 2-methylpentane, 3-methylpentane, 2,2-dimethylbutane, 2,3-dimethylbutane, 2,2-dimethylpentane, 2-methylhexane, 3-methylhexane, 2,3-dimethylpentane, 2,4-dimethylpentane, 3,3-dimethylpentane, 3-ethylpentane, 2,2,4-

trimethylpentane, 2,2,4-trimethylbutane, 2-methyloctane, 2-methylheptane, 3-methylheptane, 4-methylheptane, tetraethylmethane, C₅₋₁₆-cycloalkanes such as cyclopentane, cyclohexane, methylcyclopentane, tert-butylcyclohexane, methylcyclohexane, cycloheptane, cyclooctane, cyclononane, cyclodecane, 2,3-dimethylcyclobutane, 1,2-dimethylcyclobutane, decalin, pinane hexylcyclohexane, heptylcyclopentane, 1,4-dimethylcyclohexane, 1,1-dimethylcyclohexane, spiro-pentane, spirohexane, spiroheptane. Of course, mixtures of non-solvents can also be used.

Non-solvents suitable for the present invention are in the liquid state at normal temperature (20°C) and normal pressure (101hPa; 1 bar, 1atm). Preferred non-solvents have a melting point of < 20°C, more preferably < 15°C, even more preferably < 10°C. Preferred non-solvents also exhibit a boiling point of < 200°C, further preferably < 150°C, even more preferably < 100°C. Preferred non-solvents also exhibit a boiling point of > 25°C, further preferably > 30°C, even more preferably of > 40°C. Preferred non-solvents therefore have a boiling point between 25°C and 200°C, further preferably between 30°C and 150°C, still more preferably between 40°C and 100°C. The indications on melting points and boiling points refer here to normal pressure (101hPa; 1 bar, 1atm).

Preferred non-solvents also have a vapor pressure at normal temperature (20°C) of < 600hPa, more preferably < 300 hPa, even more preferably < 200 hPa. Preferred non-solvents further exhibit a vapor pressure at 20°C of > 1 hPa, more preferably > 10hPa, even more preferably > 30hPa. Preferred non-solvents therefore exhibit a vapor pressure normal temperature (20°C) of between 1 hPa to 600 hPa, more preferably between 10 hPa and 300 hPa, still more preferably between 30 hPa and 200 hPa.

Preferred non-solvents have no permanent dipole moment, i.e., have a dipole moment of 0.0 to a maximum of 0.1 D (0.0 - 0.3·10⁻³⁰ Cm).

Preferred non-solvents exhibit a dielectric constant ϵ_r of ≤ 2.5 , more preferably ≤ 2.2 , even more preferably ≤ 2.0 at normal temperature (20°C). Preferred non-solvents exhibit an n-octanol-water partition coefficient logK_{OW} of > 2.0, more preferably of ≥ 2.5 , even more preferably of ≥ 3.0 . Preferred non-solvents are therefore those that have a dielectric constant ϵ_r of ≤ 2.5 and a logK_{OW} of > 2.0 at normal temperature (20°C), more preferably a dielectric constant ϵ_r of ≤ 2.2 and a logK_{OW} of ≥ 2.5 , still more preferably a dielectric constant ϵ_r of ≤ 2.0 and a logK_{OW} of ≥ 3.0 .

Preferred non-solvents also have a density at normal temperature (20°C) of < 0.95 g/mL, more preferably < 0.9 g/mL, even more preferably < 0.8 g/mL. Preferred non-solvents also have a viscosity at normal temperature (20°C) of < 2.0 mPa s, more preferably < 1.5 mPa s, even more preferably < 1.0 mPa s.

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Thus, preferred herein are non-solvents that are in the liquid state of aggregation at normal temperature (20°C) and normal pressure (101hPa; 1 bar, 1atm) and have a melting point of < 20°C, more preferably < 15°C, still more preferably < 10°C, a boiling point of < 200°C, more preferably < 150°C, still more preferably of < 100°C, resp. a boiling point of > 25°C, more preferably > 30°C, even more preferably of > 40°C, a vapor pressure at 20°C of < 600hPa, more preferably < 300 hPa, even more preferably < 200 hPa, respectively. a vapor pressure at 20°C of >1 hPa, more preferably > 10hPa, still more preferably > 30hPa, a density of < 0.95 g/mL, more preferably < 0.9 g/mL, still more preferably < 0.8 g/mL, a dipole moment of 0.0-0.1 D (0.0-0.3·10⁻³⁰ Cm), a viscosity at 20°C of < 2.0 mPa s, still more preferably < 1.5 mPa s, even more preferably < 1.0 mPa s, and in particular have a dielectric constant ϵ_r at 20°C of ≤ 2.5 , more preferably ≤ 2.2 , even more preferably ≤ 2.0 , an n-octanol-water partition coefficient logK_{OW} of > 2.0, more preferably ≥ 2.5 , even more preferably ≥ 3.0 .

20 An overview of orientation values (the values given are rounded values) of the dielectric constant and logK_{OW} of non-solvents suitable herein is shown in table 1. An overview of other parameters of some specific non-solvents is shown in table 2 (the values given are rounded values).

25 Table 1: Overview of dielectric constants and logK_{OW} values of non-solvents (rounded values)

Solvent	Dielectric constant ϵ_r (20°C)	logK _{OW}
n-alkanes	approx. 1.80-2.00	approx. 3-8
iso-alkanes	approx. 1.80-1.95	approx. 3-8
cycloalkanes	approx. 1.90-2.20	approx. 3-8

Table 2: Overview of some physical parameters of some non-solvents (rounded values).

Solvent	Dielectric constant ϵ_r (20°C)	Boiling point (°C)	Melting point (°C)	Vapor pressure (20°C, hPa)	logK _{OW}
Pentane	1.84	36.1	-129.7	573	3.2
Hexane	1.89	69	-95	160	3.8

Heptane	1.92	98	-90.6	48	4.5
Octane	1.95	126	-57	14	5.2
Nonan	1.97	151	-54	4.8	5.7
Decan	2.00	174	-30	1.66	6.3
Undecan	2.01	196	-26	0.55	6.5
Isopentane	1.85	28	-160	761	3.2
2-Methylpentane	1.89	60	-154	227	3.6
2,2-Dimethylbutane	1.87	50	-100	348	3.8
2,2-Dimethylpentane	1.91	79	-123	111	3.8
2-Methylhexane	1.92	90	-118	53	3.7
2,3-Dimethylpentane	1.94	90	-135	72	3.8
2,2,4-Trimethylpentane	1.94	98	-107	53	4.1
Cyclopentane	1.97	49.3	-93.9	346	3.0
Cyclohexane	2.02	80.7	6.5	104	3.4
Methylcyclopentane	1.99	71	-143	18	3.4
tert-butylcyclohexane	2.08	167	-41	<20	4.0
Methylcyclohexane	2.02	101	-127	<20	3.9
Cycloheptane	2.08	119	-8	22.3	4
Cyclooctane	2.12	150	13	5.5	4.5
Cyclononan	2.14	170	11	<5	5.1
Cyclodecane	2.15	201	10	<5	5.5

Preferred non-solvents with a melting point of $<15^{\circ}\text{C}$, a dielectric constant ϵ_r at 20°C of ≤ 2.5 , a $\log K_{ow}$ of ≥ 3.0 , a boiling point of $<200^{\circ}\text{C}$, a boiling point of $>30^{\circ}\text{C}$, a vapor pressure at 20°C between 1 hPa and 600hPa include but are not limited to pentane, hexane, heptane, octane, nonane, decane, petroleum ether, isooctane, 2-methylpentane, 3-methylpentane, 2,2-dimethylbutane, 2,3-dimethylbutane, 2,2-dimethylpentane, 2-methylhexane, 3-methylhexane, 2,3-dimethylpentane, 2,4-dimethylpentane, 3,3-dimethylpentane, 3-ethylpentane, 2,2,4-trimethylpentane, 2,2,4-trimethylbutane, 2-methyloctane, 2-methylheptane, 3-methylheptane, 4-methylheptane, tetraethylmethane, cyclopentane, cyclohexane, methylcyclopentane, tert-butylcyclohexane, methylcyclohexane, cycloheptane, cyclooctane, cyclononane, cyclodecane, 2,3-dimethylcyclobutane, 1,2-dimethylcyclobutane, decalin, pinane. The melting points and boiling points given here refer to normal pressure (101hPa; 1 bar, 1atm).

Preferred non-solvents herein include pentane, cyclopentane, hexane, cyclohexane, heptane, octane, nonane and decane.

Further preferred nonsolvents having a melting point of $< 10^{\circ}\text{C}$, a dielectric constant ϵ_r at 20°C of ≤ 2.0 , a $\log K_{\text{OW}}$ of ≥ 3.0 , a boiling point of $< 150^{\circ}\text{C}$, a boiling point of $> 30^{\circ}\text{C}$, a vapor pressure at 20°C between 10 hPa and 600 hPa include but are not limited to pentane, hexane, heptane, octane, petroleum ether, isooctane, 2-methylpentane, 3-methylpentane, 2,2-dimethylbutane, 2,3-dimethylbutane, 2,2 dimethylpentane, 2-methylhexane, 3-methylhexane, 2,3-dimethylpentane, 2,4-dimethylpentane, 3,3-dimethylpentane, 3-ethylpentane, 2,2,4-trimethylpentane, 2,2,4-trimethylbutane, 2-methyloctane, 2-methylheptane, 3-methylheptane, 4-methylheptane, tetraethylmethane, cyclopentane, cyclohexane, methylcyclopentane, methylcyclohexane, cycloheptane, 2,3-dimethylcyclobutane, 1,2-dimethylcyclobutane. The melting points and boiling points given here refer to normal pressure (101hPa; 1 bar, 1atm).

Still further preferred non-solvents having a melting point of $< 10^{\circ}\text{C}$, a dielectric constant ϵ_r at 20°C of ≤ 2.0 , a $\log K_{\text{OW}}$ of ≥ 3.0 , a boiling point of $< 100^{\circ}\text{C}$, a boiling point of $> 40^{\circ}\text{C}$, a vapor pressure at 20°C between 30 hPa and 300hPa include but are not limited to hexane, heptane, 2-methylpentane, 3-methylpentane, 2,2-dimethylpentane, 2-methylhexane, 3-methylhexane, 2,3-dimethylpentane, 2,4-dimethylpentane, 3,3-dimethylpentane, 3-ethylpentane, 2,2,4-trimethylpentane, 2,2,4-trimethylbutane, cyclohexane, cycloheptane, 2,3-dimethylcyclobutane, 1,2-dimethylcyclobutane. The melting points and boiling points given here refer to normal pressure (101hPa; 1 bar, 1atm).

Herein, particularly preferred non-solvents are hexane, cyclohexane and heptane.

The term "**non-polar organic solvent**", as used herein, refers to a carbon-based solvent that is liquid at normal temperature (20°C) and pressure (101hPa; 1 bar, 1atm), i.e., has at least a melting point of $< 20^{\circ}\text{C}$. Examples of non-polar organic solvents include, but are not limited to, carbon tetrachloride, pure hydrocarbon solvents such as, for example, pentane, cyclopentane, hexane, cyclohexane, heptane, octane, nonane or decane, aromatic solvents such as toluene, benzene, xylene.

Non-polar organic solvents, as defined herein, have a dielectric constant ϵ_r at 20°C of ≤ 10 , more preferably of ≤ 5.0 , more preferably ≤ 3.0 , even more preferably of ≤ 2.0 and simultaneously an n-octanol-water partition coefficient $\log K_{\text{OW}} > 2.0$, more preferably of ≥ 2.5 , even more preferably of ≥ 3.0 . Thus, a solvent with a dielectric

constant ϵ_r at 20°C of ≤ 10 and a $\log K_{OW} \leq 2.0$, particularly ≤ 1.5 does not constitute a nonpolar organic solvent herein. For example, 1,4-dioxane has a dielectric constant ϵ_r at 20°C of about 2.3, but a $\log K_{OW}$ of about -0.4, and thus does not represent a nonpolar organic solvent herein.

5 Preferred are nonpolar organic solvents that have a dielectric constant ϵ_r of ≤ 10 at normal temperature (20°C) and a $\log K_{OW}$ of > 2.0 , more preferably a dielectric constant ϵ_r of ≤ 5 and a $\log K_{OW}$ of ≥ 2.5 , still more preferably a dielectric constant ϵ_r of ≤ 3.0 and a $\log K_{OW}$ of ≥ 3.0 . In particular, nonpolar organic solvents that have a dielectric
10 constant ϵ_r of ≤ 2.0 and a $\log K_{OW}$ of > 3.0 at normal temperature (20°C) are preferred.

Non-polar organic solvents suitable for the present invention exist in the liquid state of aggregation at normal temperature (20°C) and normal pressure (101hPa; 1 bar, 1atm). Preferred nonpolar organic solvents have a melting point of $< 20^\circ\text{C}$, more preferably $< 15^\circ\text{C}$, still more preferably $< 10^\circ\text{C}$. Preferred non-polar organic solvents also exhibit a
15 boiling point of $< 200^\circ\text{C}$, further preferably $< 150^\circ\text{C}$, even more preferably $< 100^\circ\text{C}$. Preferred nonpolar organic solvents also exhibit a boiling point of $> 25^\circ\text{C}$, more preferably $> 30^\circ\text{C}$, even more preferably of $> 40^\circ\text{C}$. Preferred nonpolar organic solvents therefore have a boiling point of between 25°C and 200°C , further preferably
20 between 30°C and 150°C , still more preferably between 40°C and 100°C . The indications on melting points and boiling points refer here to normal pressure (101hPa; 1 bar, 1atm).

Preferred non-polar organic solvents also exhibit a vapor pressure at normal
25 temperature (20°C) of < 600 hPa, more preferably < 300 hPa, even more preferably < 200 hPa. Preferred non-polar organic solvents further exhibit a vapor pressure at 20°C of > 1 hPa, more preferably > 10 hPa, even more preferably > 30 hPa. Preferred non-polar organic solvents therefore exhibit a vapor pressure normal temperature (20°C) of between 1 hPa to 600 hPa, more preferably between 10 hPa and 300 hPa, still more
30 preferably between 30 hPa and 200 hPa.

Examples of nonpolar organic solvents that have a dielectric constant ϵ_r of ≤ 10 at normal temperature (20°C) and a $\log K_{OW}$ of > 2 include, but are not limited to, unbranched C₅₋₁₆-alkanes such as pentane, hexane, heptane, octane, nonane,
35 decane, petroleum ether, branched C₅₋₁₆-alkanes such as isopentane, isooctane, 2 methylpentane, 3-methylpentane, 2,2-dimethylbutane, 2,3-dimethylbutane, 2,2-dimethylpentane, 2-methylhexane, 3-methylhexane, 2,3-dimethylpentane, 2,4-dimethylpentane, 3,3-dimethylpentane, 3-ethylpentane, 2,2,4-trimethylpentane, 2,2,4-trimethylbutane, 2-methyloctane, 2-methylheptane, 3-methylheptane, 4-

- methylheptane, tetraethylmethane, or C₅₋₁₆-cycloalkanes such as cyclopentane, cyclohexane, methylcyclopentane, tert-butylcyclohexane, methylcyclohexane, cycloheptane, cyclooctane, cyclononane, cyclodecane, 2,3-dimethylcyclobutane, 1,2-dimethylcyclobutane, decalin, pinane 1,4-dimethylcyclohexane, 1,1-dimethylcyclohexane, spiropentane, spirohexane, spiroheptane, ligroin, haloalkanes such as carbon tetrachloride, tetradecafluorohexane, aromatic hydrocarbons such as benzene, aromatic hydrocarbons with saturated aliphatic substituents such as toluene, o-xylene, m-xylene, p-xylene, mesitylene, 1-phenylbutane, 2-methyl-1-phenylpropane, 2-phenbutane, cumene, iso-butylbenzene, propylbenzene, hexylbenzene, iso-butylbenzene, halogenated aromatics such as chlorobenzene, fluorobenzene, p-dichlorobenzene, o-dichlorobenzene, hexafluorobenzene, bromobenzene, benzyl chloride, benzyl bromide or other substituted aromatics such as anisole and ethoxybenzene.
- Thus, nonpolar organic solvents are preferred herein, which are in the liquid state of aggregation at normal temperature (20°C) and normal pressure (101hPa; 1 bar, 1atm) and have a melting point of < 20°C, more preferably < 15°C, even more preferably < 10°C, a boiling point of < 200°C, more preferably < 150°C, even more preferably < 100°C, resp. a boiling point of > 25°C, more preferably > 30°C, even more preferably of > 40°C, a vapor pressure at 20°C of < 600hPa, more preferably < 300 hPa, even more preferably < 200 hPa, respectively a vapor pressure at 20°C of > 1 hPa, more preferably > 10hPa, still more preferably > 30hPa, a dipole moment of 0.0-0.5 D (0.0 - 1.7·10⁻³⁰ Cm), preferably 0.0-0.1 D (0.0 - 0.3·10⁻³⁰ Cm), and more preferably a dielectric constant ϵ_r at 20°C of ≤ 10 , more preferably ≤ 5.0 , even more preferably of ≤ 3.0 , even more preferably of ≤ 2.0 , and an n-octanol-water partition coefficient logK_{OW} of > 2.0, more preferably of ≥ 2.5 , even more preferably of ≥ 3.0 .

The term "nonpolar organic solvent" as used herein preferably refers to aprotic nonpolar solvents, which are nonpolar due to the small differences in electronegativity between carbon and hydrogen and have no permanent dipole moment, less preferred are therefore halogenated aromatics such as chlorobenzene, fluorobenzene, p-dichlorobenzene, o-dichlorobenzene, bromobenzene, benzyl chloride, benzyl bromide or further substituted aromatics such as anisole, ethoxybenzene, which have a dipole moment of at least 1.0 D (3.3·10⁻³⁰ Cm) or a dielectric constant ϵ_r at 20°C of ≥ 3 .

Preferred nonpolar organic solvents therefore have a dipole moment of 0.0 - 0.5 D (0.0 - 1.7·10⁻³⁰ Cm), and further preferred nonpolar organic solvents have no permanent dipole moment, i.e. have a dipole moment of 0.0 - 0.1 D (0.0 - 0.3·10⁻³⁰ Cm).

Aprotic nonpolar solvents are very lipophilic and very hydrophobic and are therefore preferred herein as nonpolar organic solvents. Representatives of aprotic nonpolar solvents preferred herein are alkanes, benzene and aromatics with aliphatic and aromatic substituents, perhalogenated hydrocarbons, e.g. carbon tetrachloride, hexafluorobenzene.

Other representatives of aprotic-nonpolar solvents are alkenes, alkynes, aromatics with unsaturated aliphatic substituents and other molecules with a completely symmetrical structure, such as tetramethylsilane or carbon disulfide. The aprotic nonpolar solvents such as the alkenes, alkynes, aromatics with unsaturated aliphatic substituents, and other molecules of completely symmetrical construction, such as tetramethylsilane or carbon disulfide, can be used herein as nonpolar organic solvents, but are less preferred. If such aprotic nonpolar solvents are used, it must be ensured that no chemical reactions occur between them and the microcrystalline taxane and the at least one tri-*O*-acylglycerol. A person skilled in the art is readily able to assess whether chemical reactions can occur between a particular solvent and a microcrystalline taxane or tri-*O*-acylglycerol, and whether a particular solvent is suitable for preparing a crystal suspension. The selection of a suitable solvent is therefore part of the routine work of a person skilled in the art.

Preferred non-polar organic solvents herein are therefore unbranched, branched and cyclic saturated aliphatic hydrocarbons, aromatic hydrocarbons and aromatic hydrocarbons with saturated aliphatic substituents and perhalogenated hydrocarbons.

Preferred non-polar organic solvents thus exhibit a dielectric constant ϵ_r of ≤ 3 , more preferably of ≤ 2.5 , more preferably ≤ 2.2 , even more preferably of ≤ 2.0 and an *n*-octanol-water partition coefficient $\log K_{ow} > 2.0$, more preferably of ≥ 2.5 , even more preferably of ≥ 3.0 at normal temperature (20°C) and normal pressure (101hPa; 1 bar, 1atm).

Thus, particularly preferred herein are nonpolar organic solvents that are in the liquid state of aggregation at normal temperature (20°C) and normal pressure (101hPa; 1 bar, 1atm) and have a melting point of $< 20^\circ\text{C}$, more preferably $< 15^\circ\text{C}$, still more preferably $< 10^\circ\text{C}$, a boiling point of $< 200^\circ\text{C}$, more preferably $< 150^\circ\text{C}$, still more preferably $< 100^\circ\text{C}$, respectively a boiling point of $> 25^\circ\text{C}$, more preferably $> 30^\circ\text{C}$, even more preferably of $> 40^\circ\text{C}$, a vapor pressure at 20°C of $< 600\text{hPa}$, more preferably $< 300\text{ hPa}$, even more preferably $< 200\text{ hPa}$, respectively. a vapor pressure at 20°C of $> 1\text{ hPa}$, more preferably $> 10\text{ hPa}$, still more preferably $> 30\text{hPa}$, a dipole moment of 0.0 - 0.5 D (0.0 - $1.7 \cdot 10^{-30}\text{ Cm}$), preferably 0.0 - 0.1 D (0.0 - $0.3 \cdot 10^{-30}\text{ Cm}$), and a

dielectric constant ϵ_r at 20°C of ≤ 3 , more preferably ≤ 2.5 , even more preferably of ≤ 2.2 , even more preferably of ≤ 2.0 , and an n-octanol-water partition coefficient $\log K_{ow}$ of > 2.0 , more preferably of ≥ 2.5 , even more preferably of ≥ 3.0 .

- 5 An overview of orientation values (the values given are rounded values) of the dielectric constants and $\log K_{ow}$ of nonpolar organic solvents is shown in Table 3.

Table 3: Overview of dielectric constants and $\log K_{ow}$ values for nonpolar organic solvents (rounded values).

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Solvent	Dielectric constant ϵ_r (20°C)	$\log K_{ow}$
n-alkanes	approx. 1.8-2.0	approx. 3-8
iso-alkanes	approx. 1.8-1.9	approx. 3-8
Cycloalkanes	approx. 1.9-2.2	approx. 3-8
Benzene	approx. 2.3	approx. 2.1
Aromatics with aliphatic radicals	approx. 2.2-2.5	approx. 2-6
Carbon tetrachloride	approx. 2.3	approx. 2.8
Halogenaromatics	approx. 5-10	approx. 2-4
Aromatics with other residues	approx. 4-5	approx. 2-3

An overview of some physical parameters of some specific examples of nonpolar organic solvents, in addition to those already shown in Table 2, is shown in Table 4 below (the values given are rounded values).

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Table 4: Overview of some physical parameters for some examples of nonpolar organic solvents.

Solvent	Dielectric constant ϵ_r (20°C)	Boiling point (°C)	Melting point (°C)	Steam pressure (20°C, hPa)	logK _{ow}
Benzene	2.3	80	6	100	2.1
Toluene	2.4	111	-95	29.1	2.7
o-Xylene	2.6	144	-25	<20	3.1
m-Xylene	2.4	139	-48	<20	3.2
p-Xylene	2.3	138	13	15	3.2
Mesitylene	2.4	165	-45	2.69	3.4
1-Phenylbutane	2.4	183	-88	1.3	3.4
2-Methyl-1-phenylpropane	2.4	173	-51	1.8	3.4
2-Phenylbutane	2.4	173	-75	1.3	3.4
2-Methyl-2-phenylpropane	2.4	169	-58	2.2	3.4
Cumol	2.4	152	-96	5.3	3.4
Ethylbenzene	2.4	136	-95	9.8	3.2
iso-butylbenzene	2.2	77	-23	1.8	4.1
Carbon tetrachloride	2.0	81	7	104	2.8
Tetradecafluorohexane	1.6	56	-90	300	5.8
Chlorobenzene	5.6	132	-46	18	2.9
o-Dichlorobenzene	9.9	180	-17	1.3	3.4
Fluorobenzene	6.4	85	-42	22.3	2.2
Hexafluorobenzene	2.1	81	4	77	2.6
Anisol	4.3	154	-37	3.6	2.1
Ethoxybenzene	4.2	170	-30	47	2.5

Preferred nonpolar organic solvents having a melting point of <20°C, a dielectric constant ϵ_r at 20°C of ≤ 3 , a logK_{ow} > 2.0, a boiling point of <200°C, a boiling point of > 30°C, a vapor pressure at 20°C between 1 hPa and 600hPa include but are not limited to pentane, hexane, heptane, octane, nonane, decane, petroleum ether, isooctane, 2-methylpentane, 3-methylpentane, 2,2-dimethylbutane, 2,3-dimethylbutane, 2,2-dimethylpentane, 2-methylhexane, 3-methylhexane, 2,3-dimethylpentane, 2,4-dimethylpentane, 3,3-dimethylpentane, 3-ethylpentane, 2,2,4-

trimethylpentane, 2,2,4-trimethylbutane, 2-methyloctane, 2-methylheptane, 3-methylheptane, 4-methylheptane, tetraethylmethane, cyclopentane, cyclohexane, methylcyclopentane, tert-butylcyclohexane, methylcyclohexane, 2,3-dimethylcyclobutane, cycloheptane, cyclooctane, cyclononane, cyclodecane, 1,2-dimethylcyclobutane, decalin, pinane, carbon tetrachloride, tetradecafluorohexane, hexafluorobenzene, benzene, toluene, o-xylene, m-xylene, p-xylene, mesitylene, ethylbenzene, 1 phenylbutane, 2-methyl-1-phenylpropane, 2-phenbutane, cumene, iso-butylbenzene, propylbenzene. Of course, mixtures of non-polar organic solvents can also be used.

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Still further preferred nonpolar organic solvents having a melting point of $< 10^{\circ}\text{C}$, a dielectric constant ϵ_r at 20°C of ≤ 3 , a $\log K_{OW} > 2.0$, a boiling point of $< 150^{\circ}\text{C}$, a boiling point of $> 30^{\circ}\text{C}$, a vapor pressure at 20°C between 10 hPa and 600 hPa include but are not limited to pentane, hexane, heptane, octane, petroleum ether, isooctane, 2-methylpentane, 3-methylpentane, 2,2-dimethylbutane, 2,3-dimethylbutane, 2,2-dimethylpentane, 2-methylhexane, 3-methylhexane, 2,3-dimethylpentane, 2,4-dimethylpentane, 3,3-dimethylpentane, 3-ethylpentane, 2,2,4-trimethylpentane, 2,2,4-trimethylbutane, 2-methyloctane, 2-methylheptane, 3-methylheptane, 4-methylheptane, tetraethylmethane, cyclopentane, cyclohexane, methylcyclopentane, methylcyclohexane, cycloheptane, 2,3-dimethylcyclobutane, 1,2-dimethylcyclobutane, carbon tetrachloride, tetradecafluorohexane, hexafluorobenzene, benzene, toluene, o-xylene, m-xylene, p-xylene, ethylbenzene. Of course, mixtures of non-polar organic solvents can also be used.

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Still further preferred nonpolar organic solvents having a melting point of $< 10^{\circ}\text{C}$, a dielectric constant ϵ_r at 20°C of ≤ 2.5 , a $\log K_{OW} > 2.0$, a boiling point of $< 100^{\circ}\text{C}$, a boiling point of $> 40^{\circ}\text{C}$, a vapor pressure at 20°C between 10 hPa and 300hPa include but are not limited to hexane, heptane, 2-methylpentane, 3-methylpentane, 2,2-dimethylpentane, 2-methylhexane, 3-methylhexane, 2,3-dimethylpentane, 2,4-dimethylpentane, 3,3-dimethylpentane, 3-ethylpentane, 2,2,4-trimethylpentane, 2,2,4-trimethylbutane, cyclohexane, cycloheptane, 2,3-dimethylcyclobutane, 1,2-dimethylcyclobutane, carbon tetrachloride, tetradecafluorohexane, hexafluorobenzene, benzene. Of course, mixtures of non-polar organic solvents can also be used.

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Still further preferred nonpolar organic solvents having a melting point of $< 10^{\circ}\text{C}$, a dielectric constant ϵ_r at 20°C of ≤ 2.5 , a $\log K_{OW} \geq 3.0$, a boiling point of $< 100^{\circ}\text{C}$, a boiling point of $> 40^{\circ}\text{C}$, a vapor pressure at 20°C between 10 hPa and 300 hPa include but are not limited to hexane, heptane, 2-methylpentane, 3-methylpentane, 2,2-

dimethylpentane, 2-methylhexane, 3-methylhexane, 2,3-dimethylpentane, 2,4-dimethylpentane, 3,3-dimethylpentane, 3-ethylpentane, 2,2,4-trimethylpentane, 2,2,4-trimethylbutane, cyclohexane, cycloheptane, 2,3-dimethylcyclobutane, 1,2-dimethylcyclobutane, carbon tetrachloride. Of course, mixtures of non-polar organic solvents can also be used.

Preferred nonpolar organic solvents are also anhydrous, i.e., dried nonpolar organic solvents.

Preferred nonpolar organic solvents also have a density at normal temperature (20°C) of < 0.95 g/mL, more preferably < 0.9 g/mL, even more preferably < 0.8 g/mL. Thus, particularly preferred nonpolar organic solvents represent the non-solvents defined herein. Non-polar organic solvents particularly preferred herein are therefore hexane, cyclohexane and heptane.

Oils such as coconut oil, palm oil, peanut oil, cottonseed oil, canola oil, fish oil, soybean oil, flaxseed oil, olive oil are generally nonpolar and have a dielectric constant ϵ_r at 20°C of about 2-5. Oils such as castor oil, coconut oil, palm oil, peanut oil, cottonseed oil, canola oil, fish oil, soybean oil, flaxseed oil, olive oil are less preferred herein as nonpolar organic solvents. These oils are viscous and have a viscosity at 20°C of about 30 - 160 mPa s. Non-polar organic solvents preferred herein have a viscosity at normal temperature (20°C) of < 2.0 mPa s, more preferably < 1.5 mPa s, even more preferably < 1.0 mPa s.

The term "**polar organic solvent**", as used herein, refers to a carbon-based solvent that is liquid at normal temperature (20°C) and pressure (101hPa; 1 bar, 1atm), i.e., has at least a melting point of < 20°C. Examples of common polar organic solvents include, but are not limited to, acetonitrile, dimethyl sulfoxide, ethers such as dioxane, tetrahydrofuran (THF), diethyl ether, methyl tert-butyl ether (MTDC), ketones such as acetone, butanone or pentanone, alcohols such as methanol, ethanol, propanol, isopropanol, carboxylic acids such as formic acid, acetic acid, propionic acid, amides such as dimethylformamide (DMF) or dimethylacetamide, halogenated solvents such as chloroform, methylene chloride, and carboxylic acid esters such as methyl acetate, ethyl acetate.

Polar organic solvents, as defined herein, preferably exhibit an n-octanol-water partition coefficient logK_{ow} of ≤ +2.0, preferably from -1.0 to +2.0, and more preferably from -0.5 to +1.5, further preferably from -0.4 to +1.4, even more preferably from -0.4 to +0.9. Further preferred polar organic solvents exhibit a dielectric constant ϵ_r at 20°C

of > 3 , more preferably of ≥ 5.0 . Preferred polar organic solvents also exhibit a dielectric constant ϵ_r at 20°C of < 50 , more preferably of < 40 , still more preferably of < 35 , most preferably of < 30 .

5 Organic solvents preferred herein thus have an n-octanol-water partition coefficient $\log K_{OW}$ of $\leq +2.0$, and a dielectric constant ϵ_r at 20°C of > 3 , still more preferably a $\log K_{OW}$ of -1.0 to $+2.0$ and a dielectric constant ϵ_r at 20°C of ≥ 5.0 and < 40 , more preferably a $\log K_{OW}$ of -0.5 to $+1.5$ and a dielectric constant ϵ_r at 20°C of ≥ 5.0 and < 30 , and most preferably a $\log K_{OW}$ of -0.4 to $+1.4$ and a dielectric constant ϵ_r at 20°C of ≥ 5.0 and < 30 . Preferred organic solvents thus exhibit an n-octanol-water partition
 10 coefficient $\log K_{OW}$ of -1.0 to $+2.0$ and a dielectric constant ϵ_r at 20°C of > 3.0 to 40 , even more preferably a $\log K_{OW}$ of -1.0 to $+2.0$ and a dielectric constant ϵ_r at 20°C of 5.0 to 40 , even more preferably a $\log K_{OW}$ of -1.0 to $+2.0$ and a dielectric constant ϵ_r at 20°C from 5.0 to 35 , and most preferably a $\log K_{OW}$ from -0.5 to $+1.5$ and a dielectric
 15 constant ϵ_r at 20°C from 5.0 to 30 . Thus, a solvent with a dielectric constant ϵ_r at 20°C of > 3 and a $\log K_{OW} > 2.0$ is not a polar organic solvent herein. For example, chlorobenzene has a dielectric constant ϵ_r at 20°C of about 5.6 but a $\log K_{OW}$ of about 2.9 and thus does not constitute a polar organic solvent herein.

20 Polar organic solvents suitable for the present invention exist in the liquid state of aggregate at normal temperature (20°C) and normal pressure (101hPa ; 1 bar , 1 atm). Preferred polar organic solvents have a melting point of $< 20^\circ\text{C}$, more preferably $< 15^\circ\text{C}$, still more preferably $< 10^\circ\text{C}$. Preferred polar organic solvents also exhibit a boiling point of $< 200^\circ\text{C}$, further preferably $< 150^\circ\text{C}$, even more preferably $< 100^\circ\text{C}$.
 25 Preferred polar organic solvents also exhibit a boiling point of $> 25^\circ\text{C}$, more preferably $> 30^\circ\text{C}$, even more preferably of $> 40^\circ\text{C}$. Preferred polar organic solvents thus exhibit a boiling point of between 25°C and 200°C , further preferably between 30°C and 150°C , still more preferably between 40°C and 100°C . The indications on melting points and boiling points refer here to normal pressure (101hPa ; 1 bar , 1 atm).

30 Preferred polar organic solvents also exhibit a vapor pressure at normal temperature (20°C) of $< 600\text{ hPa}$, more preferably $< 300\text{ hPa}$, even more preferably $< 200\text{ hPa}$. Preferred polar organic solvents further exhibit a vapor pressure at 20°C of $> 1\text{ hPa}$, more preferably $> 10\text{ hPa}$, even more preferably $> 30\text{hPa}$. Preferred polar organic
 35 solvents thus exhibit a vapor pressure at normal temperature (20°C) of between 1 hPa to 600 hPa , more preferably between 10 hPa and 300 hPa , still more preferably between 30 hPa and 200 hPa .

Preferred polar organic solvents exhibit a dipole moment of $\geq 1.0 \text{ D}$ ($3.3 \cdot 10^{-30} \text{ Cm}$), further preferred polar organic solvents exhibit a dipole moment of $\leq 3.0 \text{ D}$ ($9.9 \cdot 10^{-30} \text{ Cm}$), even more preferred $\leq 2.0 \text{ D}$ ($6.6 \cdot 10^{-30} \text{ Cm}$).

- 5 The term "polar organic solvent", as used herein, refers to aprotic polar solvents and protic solvents. In aprotic polar solvents, the molecule is asymmetrically substituted so that the molecule has a dipole moment. Examples of aprotic polar solvents include ethers, esters, acid anhydrides, ketones, e.g. acetone, tertiary amines, pyridine, furan, thiophene, asymmetric halogenated hydrocarbons, nitromethane, dimethylformamide
10 (DMF), dimethyl sulfoxide (DMSO), dimethyl carbonate, tetramethyl urea, tetraethyl urea, dimethyl propylene urea (DMPU), 1,3-dimethyl-2-imidazolidinone (DMEU). The most important protic solvent is water. Examples of other protic solvents are alcohols, aldehydes and carboxylic acids. Thus, protic solvents are water, methanol, ethanol and other alcohols, primary and secondary amines, carboxylic acids such as formic acid
15 and acetic acid, formamide.

Preferred polar organic solvents include, in particular, the aprotic polar solvents, as these are generally miscible with the non-polar organic solvents as defined herein, in particular the non-solvents as defined herein, in any mixing ratio.

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An overview of orientation values (the values given are rounded values) of the dielectric constants and $\log K_{ow}$ for some polar organic solvents is shown in Table 5.

25 Table 5: Overview of dielectric constants and $\log K_{ow}$ for some polar organic solvents (rounded values)

Solvent	Dielectric constant ϵ_r (20°C)	$\log K_{ow}$
Halogenalkanes	approx. 3-11	approx. +1.0 - +1.98
Alkylether	approx. 4-8	approx. -0.4 - +1.98
Alkyl ketones	approx. 15-25	approx. -0.5 - +1.5
Pyridine	approx. 9-13	approx. +0.6
Alkyl nitrile	approx. 25-38	approx. -0.3 - +0.5
Alkylamines	approx. 3-7	approx. -0.8
Alkyldiols	approx. 30-50	approx. -0.9
Higher alkyl alcohols	approx. 5-17	approx. +0.8 - $\geq$ +2.0
Lower alkyl alcohols	ca. 19-32	approx. -0.7 - +0.3
Aldehydes	approx. 9-18	approx. - 4.0

Formic acid	approx. 50	approx. -0.5
Higher carboxylic acids	approx. 2-6	approx. -0.2
Alkyl ester	approx. 3 - 9	approx. +0.0 - $\geq$ +2.0
Water	approx. 80	approx. -1.4
DMF	approx. 37	approx. -1.0
DMSO	approx. 47	approx. -1.4

An overview of some physical parameters (the values given are rounded values) of some polar organic solvents is shown in Table 6 below.

5 Table 6: Overview of some physical parameters for some examples of polar organic solvents.

Solvent	Dielectric constant ϵ_r (20°C)	Boiling point (°C)	Melting point (°C)	Steam pressure (20°C, hPa)	logK _{ow}
Dichloromethane	9.0	40	-95	475	+1.3
Chloroform	4.8	61	-64	210	+2.0
Diethyl ether	4.3	35	-117	586	+0.9
Dipropyl ether	3.3	90	-122	73	+2.0
2-Pentanone	15.4	101	-78		+0.9
Tetrahydrofuran (THF)	7.6	66	-109	173	+0.5
1,4-Dioxane	2.3	101	-12	38	-0.4
Acetone	20.7	56	-95	246	-0.2
Acetonitrile	37	82	-44	94	-0.3
Nitromethane	37	101	-29	36	-0.4
Acetic acid	6.2	118	-17	15.8	-0.2
Methanol	32.8	65	-98	129	-0.7
Ethanol	24.9	78	-114	58	-0.3
1-Propanol	20.2	97	-126	104	+0.3
iso-propanol	19.9	82	-88	44	+0.1
Ethylene glycol	37.0	197	-16	22.3	-1.4
Glycerin	47.0	290	18	77	-1.8
Methyl acetate	7.1	57	-99	228	+0.2
Ethyl acetate	6.0	77	-84	98	+0.7
n-Propyl acetate	5.6	102	-95	33	+1.4

DMF	37	153	-60	3.8	-1.0
Dimethylacetamide	38	166	-20	3.3	-0.8
N-methylpyrrolidone	32	203	-24	0.3	-0.5
DMSO	47	189	18	0.6	-1.4
Water	80	100	0	23	

Preferred polar organic solvents are tetrahydrofuran, acetone, methanol, ethanol, n-propanol, iso-propanol chloroform, methylene chloride (dichloromethane) and ethyl acetate (ethyl acetate).

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Further preferred polar organic solvents are tetrahydrofuran, acetone, ethanol, n-propanol, iso-propanol and ethyl acetate.

Most preferred are the physiologically largely harmless solvents ethanol, iso-propanol and ethyl acetate. Ethyl acetate is particularly preferred.

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Of course, mixtures of polar organic solvents can also be used.

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Water is considered a very polar organic solvent, but should be avoided because water-containing coatings are difficult to dry. In addition, the tri-O-acylglycerols according to the invention are poorly soluble or virtually insoluble in very polar solvents such as water. A suspension containing only water as solvent is not according to the invention, since the tri-O-acylglycerols selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol cannot exist in dissolved form in water. However, water-containing solvent mixtures of water-miscible polar organic solvents in which the at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol are present in dissolved form can also be provided, e. g. solvent mixtures such as water/methanol (25:75), water/isopropanol (35:65) or also water/acetonitrile (15:85). Nevertheless, anhydrous solvent mixtures are preferred herein.

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Particularly preferred are therefore anhydrous suspensions for coating a medical device, preferably catheters, stents, and cannulas, containing at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol and at least one taxane in the form of microcrystals; and a solvent or a solvent mixture, wherein the at least one tri-O-acylglycerol is dissolved in the at least one solvent or the solvent

mixture, and wherein the microcrystals of the at least one taxane do not dissolve in the solvent or the solvent mixture containing the dissolved at least one tri-O-acylglycerol.

5 An "anhydrous suspension" as used herein contains no more than 20% by volume of water, preferably less than 20% by volume, more preferably less than 10% by volume, more preferably less than 5% by volume, more preferably less than 3% by volume, more preferably less than 2% by volume, more preferably less than 1.5% by volume, still more preferably less than 1% by volume, still more preferably less than 0.5% by volume, and most preferably less than 0.1% by volume of water based on the total
10 volume of the suspension.

In some preferred embodiments, the suspension of the present invention comprises a solvent mixture wherein the at least one tri-O-acylglycerol dissolves and the microcrystals of the at least one taxane do not dissolve. Preferred solvent mixtures
15 herein are mixtures of at least one non-polar organic solvent, preferably at least one non-solvent and at least one polar organic solvent. Non-polar organic solvent or non-solvent and polar organic solvent must be miscible with each other and preferably miscible with each other in any ratio to give a homogeneous mixture.

20 The non-polar organic solvents, as defined herein, are generally immiscible with water and other highly polar organic solvents such as short chain alcohols such as methanol in any ratio. Non-polar organic solvents that are immiscible with water include, but are not limited to, benzene, carbon tetrachloride, cyclohexane, heptane, hexane, isooctane, pentane, toluene, and xylene. Thus, in particular, the non-solvents as
25 defined herein are immiscible with water. Solvent mixtures containing at least one non-solvent therefore particularly preferably do not contain water as a mixture component. Furthermore, some non-polar organic solvents such as xylene, cyclohexane, heptane, hexane, isooctane and pentane are not miscible in any ratio even with very polar organic solvents such as dimethyl sulfoxide and dimethyl formamide. In addition, the
30 non-solvents as defined herein, such as cyclohexane, heptane, hexane, isooctane and pentane are also immiscible with polar organic solvents such as acetonitrile and methanol.

Solvent mixtures of at least one non-solvent, as defined herein, and at least one polar
35 organic solvent therefore particularly preferably contain polar organic solvents having a logK_{ow} of -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of < 30, further preferably a logK_{ow} of -0.4 to +1.4 and a dielectric constant ϵ_r at 20°C of < 30.

In some embodiments, the volume ratio in the suspension between the nonpolar organic solvent and the polar organic solvent is between 25:75 to 75:25, preferably between 30:70 to 70:30, and more preferably between 35:65 and 65:35.

- 5 Preferably, the volume ratio in the suspension between the nonpolar organic solvent and the polar organic solvent is between 99:1 to 65:35, preferably between 95:5 to 70:30, and more preferably between 80:20 and 65:35, more preferably between 90:10 and 80:15 and most preferably 85:15.
- 10 Further preferred are solvent mixtures containing at least 50% by volume nonpolar organic solvent, more preferably at least 55% by volume nonpolar organic solvent, more preferably at least 60% by volume nonpolar organic solvent, more preferably at least 65% by volume nonpolar organic solvent, more preferably at least 70% by volume nonpolar organic solvent, more preferably at least 75% by volume nonpolar organic solvent, and most preferably at least 80% by volume nonpolar organic solvent.
- 15

- Further preferred are solvent mixtures containing at least 50% by volume non-solvent, more preferably at least 55% by volume non-solvent, more preferably at least 60% by volume non-solvent, more preferably at least 65% by volume non-solvent, more preferably at least 70% by volume non-solvent, more preferably at least 75% by volume non-solvent and most preferably at least 80% by volume non-solvent.
- 20

- In preferred embodiments, the solvent mixture therefore contains at least one non-polar organic solvent having a dielectric constant ϵ_r at 20°C of ≤ 10 and an n-octanol-water partition coefficient $\log K_{OW}$ of > 2.0 , more preferably having a dielectric constant ϵ_r at 20°C of ≤ 5.0 and an n-octanol-water partition coefficient $\log K_{OW}$ of ≥ 2.5 , more preferably having a dielectric constant ϵ_r at 20°C of ≤ 3.0 and an n-octanol-water partition coefficient $\log K_{OW}$ of ≥ 3.0 , and most preferably having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .
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- 30 In further preferred embodiments, the solvent mixture contains at least 50% by volume, more preferably at least 55% by volume, more preferably at least 60% by volume, more preferably at least 65% by volume, more preferably at least 70% by volume, more preferably at least 75% by volume and most preferably at least 80% by volume of non-polar organic solvent having a dielectric constant ϵ_r at 20°C of ≤ 10 and an n-octanol-water partition coefficient $\log K_{OW}$ of > 2.0 , more preferably having a dielectric constant ϵ_r at 20°C of ≤ 5.0 , and an n-octanol-water partition coefficient $\log K_{OW}$ of ≥ 2.5 , more preferably having a dielectric constant ϵ_r at 20°C of ≤ 3.0 and an n-octanol
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water partition coefficient $\log K_{OW}$ of ≥ 3.0 and most preferably having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

5 A preferred combination of polar organic solvent and non-polar organic solvent is, for example, ethanol and cyclohexane or ethyl acetate and heptane.

Preferred are solvent mixtures of at least one polar organic solvent having a $\log K_{OW}$ of -1.0 to +2.0 and a dielectric constant ϵ_r at 20°C of 3.0 to 40, still more preferably a $\log K_{OW}$ of -1.0 to +2.0 and a dielectric constant ϵ_r at 20°C of 5.0 to 40, still more
 10 preferably a $\log K_{OW}$ of -1.0 to +2.0 and a dielectric constant ϵ_r at 20°C of 5.0 to 35, more preferably a $\log K_{OW}$ of -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of 5.0 to 30, and most preferably a $\log K_{OW}$ of -0.4 to +0.9 and a dielectric constant ϵ_r at 20°C of 5.0 to 30, and a nonpolar organic solvent having a dielectric constant ϵ_r at 20°C of ≤ 10 and an n-octanol-water partition coefficient $\log K_{OW}$ of > 2.0 , more preferably
 15 having a dielectric constant ϵ_r at 20°C of ≤ 5.0 and an n-octanol-water partition coefficient $\log K_{OW}$ of ≥ 2.5 , more preferably having a dielectric constant ϵ_r at 20°C of ≤ 3.0 and an n-octanol-water partition coefficient $\log K_{OW}$ of ≥ 3.0 , and most preferably having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

20 Preferred are solvent mixtures of at least one polar organic solvent selected from the group comprising or consisting of tetrahydrofuran, acetone, methanol, ethanol, n-propanol, iso-propanol, chloroform, dichloromethane, ethyl acetate, preferably tetrahydrofuran, acetone, ethanol, n-propanol, iso-propanol, and ethyl acetate, further preferably ethanol, iso-propanol, and ethyl acetate, and a nonpolar organic solvent
 25 having a dielectric constant ϵ_r at 20°C of ≤ 10 and an n-octanol-water partition coefficient $\log K_{OW}$ of > 2.0 , more preferably having a dielectric constant ϵ_r at 20°C of ≤ 5.0 , and an n-octanol-water partition coefficient $\log K_{OW}$ of ≥ 2.5 , more preferably having a dielectric constant ϵ_r at 20°C of from ≤ 3.0 and an n-octanol-water partition coefficient $\log K_{OW}$ of ≥ 3.0 , and most preferably having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

30

A particularly preferred combination of polar organic solvent and nonpolar organic solvent is a solvent mixture of ethyl acetate and a nonpolar organic solvent as defined herein having a dielectric constant ϵ_r at 20°C of ≤ 10 and an n-octanol-water partition coefficient $\log K_{OW}$ of > 2.0 , more preferably having a dielectric constant ϵ_r at 20°C of $\leq$
 35 5.0 and an n-octanol-water partition coefficient $\log K_{OW}$ of ≥ 2.5 , more preferably having a dielectric constant ϵ_r at 20°C of ≤ 3.0 and an n-octanol-water partition coefficient $\log K_{OW}$ of ≥ 3.0 , and most preferably having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

An even more preferred combination of polar organic solvent and non-polar organic solvent is a solvent mixture of ethyl acetate and a non-solvent as defined herein having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

- 5 Preferred combinations of polar organic solvent and nonpolar organic solvent are ethyl acetate and a nonpolar organic solvent selected from the group comprising or consisting of pentane, hexane, heptane, octane, nonane, decane, petroleum ether, isooctane, 2-methylpentane, 3-methylpentane, 2,2-dimethylbutane, 2,3-
- 10 dimethylbutane, 2,2-dimethylpentane, 2-methylhexane, 3-methylhexane, 2,3-dimethylpentane, 2,4-dimethylpentane, 3,3-dimethylpentane, 3 ethylpentane, 2,2, 4-trimethylpentane, 2,2,4-trimethylbutane, 2-methyloctane, 2-methylheptane, 3-methylheptane, 4-methylheptane, tetraethylmethane, cyclopentane, cyclohexane, methylcyclopentane, tert-butylcyclohexane, methylcyclohexane, 2,3-
- 15 dimethylcyclobutane, cycloheptane, cyclooctane, cyclononane, cyclodecane, 1,2-dimethylcyclobutane, decalin, pinane, carbon tetrachloride, tetradecafluorohexane, hexafluorobenzene, benzene, toluene, o-xylene, m-xylene, p-xylene, mesitylene, ethylbenzene, 1-phenylbutane, 2-methyl-1-phenylpropane, 2-phenbutane, cumene, iso-butylbenzene, propylbenzene, preferably pentane, hexane, heptane, octane, petroleum ether, isooctane, 2-methylpentane, 3-methylpentane, 2,2-dimethylbutane,
- 20 2,3-dimethylbutane, 2,2-dimethylpentane, 2-methylhexane, 3-methylhexane, 2,3-dimethylpentane, 2,4-dimethylpentane, 3,3-dimethylpentane, 3-ethylpentane, 2,2,4-trimethylpentane, 2,2,4-trimethylbutane, 2-methyloctane, 2-methylheptane, 3-methylheptane, 4-methylheptane, tetraethylmethane, cyclopentane, cyclohexane, methylcyclopentane, methylcyclohexane, cycloheptane, 2,3-dimethylcyclobutane, 1,2-
- 25 dimethylcyclobutane, carbon tetrachloride, tetradecafluorohexane, hexafluorobenzene, benzene, toluene, o-xylene, m-xylene, p-xylene, ethylbenzene, further preferably hexane, heptane, 2-methylpentane, 3-methylpentane, 2,2-dimethylpentane, 2-methylhexane, 3-methylhexane, 2,3-dimethylpentane, 2,4-dimethylpentane, 3,3-dimethylpentane, 3-ethylpentane, 2,2,4-trimethylpentane, 2,2,4-
- 30 trimethylbutane, cyclohexane, cycloheptane, 2,3-dimethylcyclobutane, 1,2-dimethylcyclobutane, carbon tetrachloride, tetradecafluorohexane, hexafluorobenzene, benzene, still further preferably hexane, heptane, 2-methylpentane, 3-methylpentane, 2,2-dimethylpentane, 2-methylhexane, 3-methylhexane, 2,3-dimethylpentane, 2,4-dimethylpentane, 3,3-dimethylpentane, 3-
- 35 ethylpentane, 2,2,4-trimethylpentane, 2,2,4-trimethylbutane, cyclohexane, cycloheptane, 2,3-dimethylcyclobutane, 1,2-dimethylcyclobutane, carbon tetrachloride, and most preferably hexane, heptane and cyclohexane.

K_{OW} is the n-octanol-water partition coefficient. The K_{OW} is thus the partition coefficient of a substance in the two-phase system of n-octanol and water. The $\log K_{OW}$ is the decadic logarithm of the K_{OW} . The $\log K_{OW}$ is also known as $\log P$ in English-speaking countries. The K_{OW} serves as a measure of the relationship between lipophilicity and hydrophilicity of a substance. The value is greater than one if a substance is more soluble in lipophilic solvents such as n-octanol, less than one if it is more soluble in water.

n-Octanol-water partition coefficients $\log K_{OW}$ of various solvents are well known to those skilled in the art, see for example James Sangster, "Octanol-Water Partition Coefficients of Simple Organic Compounds," *J. Phys. Chem. Ref. Data* 1989, Vol. 18, No. 3, pp. 1111-1227, which is incorporated herein by reference in its entirety. Polar organic solvents are defined herein as those having a $\log K_{OW}$ of -1.0 to +2.0 and preferably of -0.5 to +1.5. Non-solvents or non-polar organic solvents are referred to herein as those having a $\log K_{OW} \geq 2.8$ preferably a $\log K_{OW} \geq 3.3$ or a $\log K_{OW}$ from +2.8 to +7.5 and preferably from +3.3 to +7.0.

Measurement methods for determining n-octanol-water partition coefficients are also well known to those skilled in the art, see for example James Sangster, "Octanol-Water Partition Coefficients of Simple Organic Compounds", *J. Phys. Chem. Ref. Data* 1989, Vol. 18, No. 3, pp. 1111-1227, in the section "Methods of Measurement". A practical determination of the $\log K_{OW}$ value can be carried out in such a way that the respective solvent with a known concentration C_{0b}^{water} is introduced into aqueous solution at a known volume overlaid with a precisely measured volume of octanol $V^{Octanol}$ and intensively mixed. Phase separation is then waited for and the octanol phase is separated. To ensure that no further volume change occurs during phase mixing, the octanol used is saturated in advance with water and the water used is saturated in advance with octanol. The $\log K_{OW}$ is positive for lipophilic and negative for hydrophilic solvents.

Table 7: Overview $\log K_{OW}$ values of some polar organic solvents

Solvent	$\log K_{OW}$	Solvent	$\log K_{OW}$
Acetonitrile	-0.34	Acetone	-0.24
Dimethyl sulfoxide	-1.35	2-Butanone	0.29
Dioxane	-0.42	2-Pentanone	0.84
Tetrahydrofuran (THF)	0.46	Methanol	-0.74
Ethyl acetate	0.73	Ethanol	-0.30
Diethyl ether	0.89	1-Propanol	0.25

Methyl tert-butyl ether (MTDC)	0.94	2-Propanol	0.05
Dimethylformamide (DMF)	-1.01	1-Butanol	0.84
Dimethylacetamide	-0.77	tert-butyl alcohol	0.35
Methylene chloride	1.25	Acetic acid	-0.17
Chloroform	1.97	Propionic acid	0.32
Formic acid	-0.54		

Table 8: Overview of logK_{OW}-values of some nonpolar organic solvents

Solvent	logK _{OW}	Solvent	logK _{OW}
Toluene	2.73	Pentane	3.45
Benzene	2.13	Cyclopentane	3.00
Cyclohexane	2.86	Hexane	4.00
Nonan	5.65	Heptane	4.50
Decan	6.25	Octane	5.15
o-Xylene	3.12	Mesitylene	3.42
m-Xylene	3.20	Petroleum ether	4.20
p-Xylene	3.15		

Between the logK_{OW} value of the polar organic solvent and the logK_{OW} value of the
 5 nonpolar organic solvent there should be at least a difference of 1.0, preferably at least 1.5, and more preferably at least 2.0.

To determine the K_{OW} of a mixture of nonpolar organic solvents, the K_{OW} values of the
 individual nonpolar organic solvents are weighted according to the volume fraction of
 10 the mixture and the mean value is determined from the weighted K_{OW} values.

The dielectric constant (relative permittivity, formula symbol ϵ_r) is a physical substance
 constant that can be used to describe certain properties of solvents. Solvents with high
 dielectric constant are good solvents for ionic and other polar compounds, those with
 15 low dielectric constant are better solvents for non-polar compounds. The term
 "dielectric constant" is also referred to in the prior art as permittivity, dielectric
 conductivity, dielectricity, or dielectric function. The relative permittivity ϵ_r of a medium,
 also called permittivity or dielectric constant, is the dimensionless ratio of the
 permittivity ϵ to the permittivity ϵ_0 of the vacuum. For gaseous, liquid and solid matter,
 20 $\epsilon_r > 1$. Measurement methods for determining the relative permittivity are well known
 to those skilled in the art. There are many methods that have been developed for

measuring the dielectric constant. For example, the open coaxial probe method is suitable for liquids. In this method, the probe is immersed in the liquids and the reflection coefficient is measured and used to determine the dielectric constant.

5 According to the present invention, the suspension according to the invention contains a solvent or a solvent mixture. According to the invention, at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol is dissolved in the solvent or in the solvent mixture. According to the invention, the suspension therefore contains a
10 solvent or a solvent mixture in which at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol is dissolved, wherein the microcrystals of the at least one taxane do not dissolve or no longer dissolve in the presence of the at least one tri-O-acylglycerol.

15 Thus, by definition, the solvent or solvent mixture forms a solution with the dissolved at least one tri-O-acylglycerol selected from trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol and thus constitutes a homogeneous mixture. The solution of at least one tri-O-acylglycerol selected from trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol in the solvent or
20 solvent mixture thus has only one phase and the dissolved at least one tri-O-acylglycerol selected from trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol is uniformly distributed in the solvent or solvent mixture.

25 Thus, the suspension according to the present invention is a combination of:
1) a solution of at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol in a solvent or a solvent mixture, and
2) at least one taxane in the form of microcrystals, wherein the microcrystals of the at
30 least one taxane do not dissolve in the solution according to 1).

The at least one taxane in the form of microcrystals is finely distributed as a solid, i.e. "suspended", in the solution of at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol in at least one solvent or a solvent mixture. Thus, according to
35 the invention, the microcrystals of the at least one taxane are "suspended" in the solution of at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol in a solvent or a solvent mixture.

Thus, the present invention also relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- 5 a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol;
 - b) at least one taxane in the form of microcrystals; and
 - c) a solvent or a mixture of solvents,
- 10 wherein the at least one tri-*O*-acylglycerol is dissolved in the solvent or solvent mixture, and
- wherein the microcrystals of the at least one taxane do not dissolve in the solvent or solvent mixture containing the dissolved at least one tri-*O*-acylglycerol.

15 With other words, the present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) a solution of at least one of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
- 20 b) at least one taxane in the form of microcrystals, the microcrystals being suspended in said solution.

With still other words, the present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- 25 a) a solution of at least one of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and
- b) microcrystals of at least one taxane suspended in said solution.

30 The present invention therefore relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and
- 35 triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals and
- c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm .

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the at least one taxane has a crystallinity of at least 80% by weight.

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel.

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the at least one taxane is selected from paclitaxel.

The present invention relates to a suspension for coating a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- 5 a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- 10 c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

- 15 wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel.

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- 20 a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- 25 c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the at least one taxane has a crystallinity of at least 80% by weight,

wherein the at least one taxane is selected from the group comprising or consisting paclitaxel, docetaxel, and cabazitaxel.

30

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- 35 a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and

- c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the microcrystals of the at least one taxane have a crystal size in the range

5 of 1 μm to 300 μm ,

wherein the at least one taxane is selected from paclitaxel.

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
 - b) at least one taxane in the form of microcrystals, and
 - 15 c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,
- wherein the at least one taxane has a crystallinity of at least 80% by weight,
- wherein the at least one taxane is selected from paclitaxel.

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The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of
 - 25 trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
 - b) at least one taxane in the form of microcrystals, and
 - c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not
 - 30 dissolve,
- wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,
- wherein the at least one taxane has a crystallinity of at least 80% by weight,
- wherein the at least one taxane is selected from paclitaxel.

35

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- 5 c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane.

10

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- 15 a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not
- 20 dissolve,

wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel,

- 25 wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane.

30 The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- 35 c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the at least one taxane has a crystallinity of at least 80% by weight,

wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel,

wherein the at least one tri-O-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-O-acylglycerol to 90% - 70% taxane.

5

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-O-acylglycerol selected from the group consisting of
 - 10 trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- c) a solvent or a solvent mixture, in which the at least one tri-O-acylglycerol
 - 15 dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

wherein the at least one taxane is selected from paclitaxel,

- 20 wherein the at least one tri-O-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-O-acylglycerol to 90% - 70% taxane.

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- 25 a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- c) a solvent or a solvent mixture, in which the at least one tri-O-acylglycerol
 - 30 dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the at least one taxane has a crystallinity of at least 80% by weight,

wherein the at least one taxane is selected from paclitaxel,

- 35 wherein the at least one tri-O-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-O-acylglycerol to 90% - 70% taxane.

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- 5 c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

- 10 wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- 15 a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- 20 b) at least one taxane in the form of microcrystals, and
- c) a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

- 25 wherein the solvent mixture is a mixture of at least one polar organic solvent having an *n*-octanol-water partition coefficient $\log K_{ow}$ of from -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of from 5.0 to 30 and at least one non-polar organic solvent having a dielectric constant ϵ_r at 20°C of from ≤ 3.0 and an *n*-octanol-water partition coefficient $\log K_{ow}$ of ≥ 3.0 .

30

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- 35 a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- c) a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel,

wherein the solvent mixture is a mixture of at least one polar organic solvent having an n-octanol-water partition coefficient $\log K_{ow}$ of from -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of from 5.0 to 30 and at least one non-polar organic solvent having a dielectric constant ϵ_r at 20°C of from ≤ 3.0 and an n-octanol-water partition coefficient $\log K_{ow}$ of ≥ 3.0 .

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- c) a solvent or a solvent mixture, in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel,

wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- c) a solvent or a solvent mixture, in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the at least one taxane is selected from paclitaxel,

wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
 - b) at least one taxane in the form of microcrystals, and
 - c) a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,
- wherein the at least one taxane is selected from paclitaxel,
- wherein the solvent mixture is a mixture of at least one polar organic solvent having an *n*-octanol-water partition coefficient $\log K_{ow}$ of -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of 5.0 to 30 and at least one non-polar organic solvent having a dielectric constant ϵ_r at 20°C of ≤ 3.0 and an *n*-octanol-water partition coefficient $\log K_{ow}$ of ≥ 3.0 .

15

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

25

wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel,

30

wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

35

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing: a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,

b) at least one taxane in the form of microcrystals, and

c) a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve, wherein the microcrystals of the at least taxane agent have a crystal size in the range of 1 μm to 300 μm ,

5 wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel,

wherein the solvent mixture is a mixture of at least one polar organic solvent having an *n*-octanol-water partition coefficient $\log K_{\text{OW}}$ of from -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of from 5.0 to 30 and at least one non-polar organic solvent having
10 a dielectric constant ϵ_r at 20°C of from ≤ 3.0 and an *n*-octanol-water partition coefficient $\log K_{\text{OW}}$ of ≥ 3.0 .

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula,
15 the suspension containing:

a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,

b) at least one taxane in the form of microcrystals, and

20 c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the at least one taxane has a crystallinity of at least 80% by weight,

wherein the at least one taxane is selected from the group comprising or consisting
25 of paclitaxel, docetaxel, and cabazitaxel,

wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

30 The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and
35 triundecanoylglycerol,

b) at least one taxane in the form of microcrystals, and

c) a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the at least one taxane has a crystallinity of at least 80% by weight,

wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel,

wherein the solvent mixture is a mixture of at least one polar organic solvent having an n-octanol-water partition coefficient $\log K_{ow}$ of from -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of from 5.0 to 30 and at least one non-polar organic solvent having a dielectric constant ϵ_r at 20°C of from ≤ 3.0 and an n-octanol-water partition coefficient $\log K_{ow}$ of ≥ 3.0 .

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- c) a solvent or a solvent mixture, in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

wherein the at least one taxane is selected from paclitaxel,

wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

25

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- c) a solvent mixture, in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

wherein the at least one taxane is selected from paclitaxel,

wherein the solvent mixture is a mixture of at least one polar organic solvent having an n-octanol-water partition coefficient $\log K_{ow}$ of from -0.5 to +1.5 and a dielectric

constant ϵ_r at 20°C of from 5.0 to 30 and at least one non-polar organic solvent having a dielectric constant ϵ_r at 20°C of from ≤ 3.0 and an n-octanol-water partition coefficient $\log K_{ow}$ of ≥ 3.0 .

- 5 The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:
- a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and
 - 10 triundecanoylglycerol,
 - b) at least one taxane in the form of microcrystals, and
 - c) a solvent mixture, in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,
- wherein the at least one taxane has a crystallinity of at least 80% by weight,
- 15 wherein the at least one taxane is selected from paclitaxel,
 - wherein the solvent mixture is selected from ethanol and cyclohexane or ethyl acetate and heptane.

- The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:
- a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and
 - 25 triundecanoylglycerol,
 - b) at least one taxane in the form of microcrystals, and
 - c) a solvent or a solvent mixture, in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,
- wherein the microcrystals of the at least one taxane have a crystal size in the range of
- 30 1 μm to 300 μm ,
 - wherein the at least one taxane has a crystallinity of at least 80% by weight,
 - wherein the at least one taxane is selected from paclitaxel,
 - wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a
 - 35 dielectric constant ϵ_r at 20°C of ≤ 2.0 .

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
 - b) at least one taxane in the form of microcrystals, and
 - 5 c) a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,
- wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,
- wherein the at least one taxane has a crystallinity of at least 980% by weight,
 - 10 wherein the at least one taxane is selected from paclitaxel,
 - wherein the solvent mixture is a mixture of at least one polar organic solvent having an *n*-octanol-water partition coefficient $\log K_{ow}$ of from -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of from 5.0 to 30 and at least one non-polar organic solvent having a dielectric constant ϵ_r at 20°C of from ≤ 3.0 and an *n*-octanol-water partition coefficient
 - 15 $\log K_{ow}$ of ≥ 3.0 .

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- 20 a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
 - b) at least one taxane in the form of microcrystals, and
 - c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol
 - 25 dissolves and in which the microcrystals of the at least one taxane do not dissolve,
- wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane,
- wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of $\leq$
 - 30 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula,

35 the suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and

c) a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve, wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane,

5 wherein the solvent mixture is a mixture of at least one polar organic solvent having an *n*-octanol-water partition coefficient $\log K_{OW}$ of from -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of from 5.0 to 30 and at least one non-polar organic solvent having a dielectric constant ϵ_r at 20°C of from ≤ 3.0 and an *n*-octanol-water partition coefficient $\log K_{OW}$ of ≥ 3.0 .

10 The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

a) at least one tri-*O*-acylglycerol selected from the group consisting of
15 trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,

b) at least one taxane in the form of microcrystals, and

c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not
20 dissolve,

wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel,

25 wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane,

wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

30 The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

a) at least one tri-*O*-acylglycerol selected from the group consisting of
35 trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,

b) at least one taxane in the form of microcrystals, and

c) a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel,

5 wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane,

wherein the solvent mixture is a mixture of at least one polar organic solvent having an *n*-octanol-water partition coefficient $\log K_{\text{OW}}$ of from -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of from 5.0 to 30 and at least one non-polar organic solvent having
10 a dielectric constant ϵ_r at 20°C of from ≤ 3.0 and an *n*-octanol-water partition coefficient $\log K_{\text{OW}}$ of ≥ 3.0 .

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula,
15 the suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- 20 c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the at least one taxane has a crystallinity of at least 80% by weight,

wherein the at least one taxane is selected from the group comprising or consisting
25 of paclitaxel, docetaxel, and cabazitaxel,

wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane,

wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a
30 dielectric constant ϵ_r at 20°C of ≤ 2.0 .

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- 35 a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and

- c) a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,
 wherein the at least one taxane has a crystallinity of at least 80% by weight,
 wherein the at least one taxane is selected from the group comprising or consisting
 5 of paclitaxel, docetaxel, and cabazitaxel,
 wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane,
 wherein the solvent mixture is a mixture of at least one polar organic solvent having an *n*-octanol-water partition coefficient $\log K_{OW}$ of from -0.5 to +1.5 and a dielectric
 10 constant ϵ_r at 20°C of from 5.0 to 30 and at least one non-polar organic solvent having a dielectric constant ϵ_r at 20°C of from ≤ 3.0 and an *n*-octanol-water partition coefficient $\log K_{OW}$ of ≥ 3.0 .

The present invention relates to a suspension for coating of a medical device,
 15 preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
 20 b) at least one taxane in the form of microcrystals, and
 c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the microcrystals of the at least one taxane have a crystal size in the range of
 25 1 μm to 300 μm ,

- wherein the at least one taxane is selected from paclitaxel,
 wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane,
 wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of $\leq$
 30 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

The present invention relates to a suspension for coating of a medical device,
 preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula,
 35 the suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
 b) at least one taxane in the form of microcrystals, and

c) a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve, wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

- 5 wherein the at least one taxane is selected from paclitaxel,
 wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane,
 wherein the solvent mixture is a mixture of at least one polar organic solvent having an *n*-octanol-water partition coefficient $\log K_{\text{OW}}$ of from -0.5 to +1.5 and a dielectric
 10 constant ϵ_r at 20°C of from 5.0 to 30 and at least one non-polar organic solvent having a dielectric constant ϵ_r at 20°C of from ≤ 3.0 and an *n*-octanol-water partition coefficient $\log K_{\text{OW}}$ of ≥ 3.0 .

The present invention relates to a suspension for coating of a medical device,
 15 preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
 20 b) at least one taxane in the form of microcrystals, and
 c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve,

wherein the at least one taxane has a crystallinity of at least 80% by weight,

- 25 wherein the at least one taxane is selected from paclitaxel,
 wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane,
 wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a
 30 dielectric constant ϵ_r at 20°C of ≤ 2.0 .

The present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- 35 a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
 b) at least one taxane in the form of microcrystals, and

- c) a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve, wherein the at least one taxane has a crystallinity of at least 80% by weight, wherein the at least one taxane is selected from paclitaxel,
- 5 wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane, wherein the solvent mixture is a mixture of at least one polar organic solvent having an *n*-octanol-water partition coefficient $\log K_{OW}$ of -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of 5.0 to 30 and at least one non-polar organic solvent having a dielectric
- 10 constant ϵ_r at 20°C of ≤ 3.0 and an *n*-octanol-water partition coefficient $\log K_{OW}$ of ≥ 3.0 .

Thus, the present invention preferably relates to a suspension for coating a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:

- 15 a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
- b) at least one taxane in the form of microcrystals, and
- c) a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol
- 20 dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-*O*-acylglycerol is present, wherein the suspension contains 1-6% taxane.

- In some further embodiments, the coating suspension may consist only of the three
- 25 ingredients a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol or a mixture of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and/or triundecanoylglycerol; b) at least one taxane in the form of microcrystals; and c) at least one solvent or a solvent mixture in which the at least one tri-*O*-acylglycerol
- 30 dissolves and in which the microcrystals of the at least one taxane do not dissolve.

Additives

- In addition to the above-mentioned at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and,
- 35 the suspension according to the invention may also contain one or more additives. In particular, the one or more additives are preferably dissolved in the suspension.

In preferred embodiments, the suspension according to the invention does not contain polymers, oligomers, metals or metal particles, organometallic compounds and salts.

In some embodiments, the medical device coating suspension according to the invention is therefore free of polymers, oligomers, metals or metal particles, organometallic compounds, and salts.

- 5 In preferred embodiments, an antioxidant may be present as an additive in the suspension according to the invention.

Suitable antioxidants include butylhydroxytoluene (BHT), butylhydroxyanisole, ascorbyl palmitate, ascorbyl stearate, tocopherol acetate, ascorbic acid, tocopherols
 10 and tocotrienols (e.g. alpha-tocopherol), carotenoids such as β -carotene, zeaxanthin, lycopene and lutein, vitamin C, nordihydroguaric acid, probucol, propyl gallate, secondary plant compounds (flavonoids) such as catechin, gallic acid, epicatechin, epigallocatechin gallate, taxifolin, isoliquiritigenin, xanthohumol, morin, quercetin (glycoside rutin and methyl ether isorhamnetin), kaempferol, myricetin, fisetin,
 15 aureusidin, luteolin, apigenin, hesperetin, naringenin, eriodictyol, genistein, daidzein, licoricidin anthocyanins, allicin, astaxanthin glutathione, resveratrol, derivatives thereof and combinations thereof.

Preferred antioxidants are therefore butylated hydroxytoluene (BHT), which prevents
 20 or delays rancidity especially in fat phases after contact with air, butylated hydroxyanisole (BHA), tocopherols, carotenoids, flavonoids, and of course also mixtures of the antioxidants.

Butylhydroxytoluene (BHT) is particularly preferred as an antioxidant.

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If one or more antioxidants are added to the suspension, their total content is calculated to be between 1.0 - 15.0 wt.% with respect to taxane, preferably 0.5- 0.005 wt.% and particularly preferably 0.1 - 0.01 wt.%.

- 30 In some embodiments, a flocculation inhibitor may be present in the suspension of the invention as an additive that may prevent sedimentation of the microcrystals of the at least one taxane in the suspension. Suitable flocculation inhibitors include, but are not limited to, polysorbates such as Tween 80. Flocculation inhibitors are preferably used in the preparation of crystal suspensions at very low active agent levels.

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For example, a 3% suspension containing taxane with uniform distribution of the crystals can be prepared without a flocculation inhibitor; from about 1.5-1% suspension (w/v) and lower, the addition of flocculation inhibitors can be advantageous, as these

additionally prevent sedimentation of the microcrystals and thus continue to allow uniform coating.

- 5 If one or more flocculation inhibitors are added to the suspension, the additional amount that can keep the microcrystals in suspension must be determined individually for the respective taxane. The total amount of microcrystalline taxane in the suspension is preferably very low between 1.0 - 0.001 wt.%, further preferred are 0.5 - 0.005 wt.%, and especially preferred are 0.1 - 0.01 wt.%.
- 10 It is also possible to add another non-polymeric excipient to the solution as a matrix. For example, contrast agents or contrast agent analogs are suitable, as are biocompatible organic substances that also improve or do not negatively change the coating properties.
- 15 In some embodiments, the suspension of the present invention may also contain one or more polymers as an additive, such as polyvinylpyrrolidone (PVP). Suitable polymer additives are polymers that can be dissolved in organic solvents, especially non-polar solvents. Very hydrophilic, water-soluble polymers that are hardly soluble or virtually insoluble in organic solvents are not preferred. In addition, care must be taken to
- 20 ensure that the microcrystals of the at least one taxane do not become attached or dissolved in the presence of a polymer dissolved in the suspension. Polymers for coatings of medical devices are known from the prior art. The person skilled in the art is thus easily able to select a suitable polymer additive.
- 25 However, polymer-free suspensions for coating of medical devices are particularly preferred herein.

Thus, the present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula,

30 the suspension consisting of:

- (a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
- b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel,

35 and

- c) a solvent or a solvent mixture in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-*O*-acylglycerol is present, and

- d) up to 5.0% by weight, based on the taxane, of additives.

Suitable additives are antioxidants, polyvinylpyrrolidone (PVP) and flocculation inhibitors.

5

Thus, the present invention preferably relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- 10 (a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
- b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel, and
- 15 c) a solvent or a solvent mixture in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-*O*-acylglycerol is present, and
- d) up to 5.0% by weight, based on the taxane, of additives,
- 20 wherein the microcrystals of the at least one taxane have a crystal size ranging from 1 μm to 300 μm .

Thus, the present invention preferably relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- 25 (a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
- b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel,
- 30 and
- c) a solvent or a solvent mixture in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-*O*-acylglycerol is present, and
- d) up to 5.0% by weight, based on the taxane, of additives,
- 35 wherein the at least one taxane has a crystallinity of at least 80% by weight.

Thus, the present invention preferably relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- (a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
- 5 b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel, and
- c) a solvent or a solvent mixture in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-O-acylglycerol is present, and
- 10 d) up to 5.0% by weight, based on the taxane, of additives, wherein the at least one taxane is selected from paclitaxel.

Thus, the present invention preferably relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- 15 (a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
- b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel, and
- 20 c) a solvent or a solvent mixture in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-O-acylglycerol is present, and
- 25 d) up to 5.0% by weight, based on the taxane, of additives, wherein the microcrystals of the at least one taxane have a crystal size ranging from 1 μm to 300 μm , wherein the at least one taxane is selected from paclitaxel.

30 Thus, the present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- 35 (a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
- b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel, and

- c) a solvent or a solvent mixture in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-O-acylglycerol is present, and
 - d) up to 5.0% by weight, based on the taxane, of additives,
- 5 wherein the at least one taxane has a crystallinity of at least 80% by weight, wherein the at least one taxane is selected from paclitaxel.

Thus, the present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- (a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
 - b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel,
- 15 and
- c) a solvent or a solvent mixture in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-O-acylglycerol is present, and
 - d) up to 5.0% by weight, based on the taxane, of additives,
- 20 wherein the microcrystals of the at least one taxane have a crystal size ranging from 1 μm to 300 μm , wherein the at least one taxane has a crystallinity of at least 80% by weight, wherein the at least one taxane is selected from paclitaxel.

25 Thus, the present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- (a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
 - b) at least one taxane in the form of microcrystals, wherein the at least one taxane taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel, and
- 30
- c) a solvent or a solvent mixture in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-O-acylglycerol is present, and
 - d) up to 5.0% by weight, based on the taxane, of additives,
- 35

wherein the at least one tri-O-acylglycerol and the at least one taxane are present at a mass fraction of 10% - 30% tri-O-acylglycerol to 90% - 70% taxane.

Thus, the present invention relates to a suspension for coating of a medical device,
5 preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula,
the suspension consisting of:

- (a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
- 10 b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel, and
- c) a solvent or a solvent mixture in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not
15 dissolve when the at least one tri-O-acylglycerol is present, and
- d) up to 5.0% by weight, based on the taxane, of additives,
wherein the microcrystals of the at least one taxane have a crystal size ranging from 1 μm to 300 μm ,
wherein the at least one taxane is selected from paclitaxel,
20 wherein the at least one tri-O-acylglycerol and the at least one taxane are present at a mass fraction of 10% - 30% tri-O-acylglycerol to 90% - 70% taxane.

Thus, the present invention relates to a suspension for coating of a medical device,
preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula,
25 the suspension consisting of:

- (a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
- 30 b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel, and
- c) a solvent or a solvent mixture in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-O-acylglycerol is present, and
- 35 d) up to 5.0% by weight, based on the taxane, of additives,
wherein the at least one taxane has a crystallinity of at least 80% by weight,
wherein the at least one taxane is selected from paclitaxel,
wherein the at least one tri-O-acylglycerol and the at least one taxane are present at a mass fraction of 10% - 30% tri-O-acylglycerol to 90% - 70% taxane.

Thus, the present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- 5 (a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
- b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel,
- 10 and
- c) a solvent or a solvent mixture in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-*O*-acylglycerol is present, and
- d) up to 5.0% by weight, based on the taxane, of additives,
- 15 wherein the microcrystals of the at least one taxane have a crystal size ranging from 1 μm to 300 μm ,
wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

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Thus, the present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- 25 (a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
- b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel, and
- 30 c) a solvent or a solvent mixture in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-*O*-acylglycerol is present, and
- d) up to 5.0% by weight, based on the taxane, of additives,
- 35 wherein the microcrystals of the at least one taxane have a crystal size ranging from 1 μm to 300 μm ,
wherein the solvent mixture is a mixture of at least one polar organic solvent having an *n*-octanol-water partition coefficient $\log K_{ow}$ of -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of 5.0 to 30 and at least one non-polar organic solvent having a dielectric constant ϵ_r at 20°C of ≤ 3.0 and an *n*-octanol-water partition coefficient $\log K_{ow}$ of ≥ 3.0 .

Thus, the present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- 5 (a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
- b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel,
- 10 and
- c) a solvent or a solvent mixture in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-*O*-acylglycerol is present, and
- d) up to 5.0% by weight, based on the taxane, of additives,
- 15 wherein the at least one taxane is selected from paclitaxel, wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

20 Thus, the present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- (a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and
- 25 triundecanoylglycerol; and
- b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel, and
- c) a solvent or a solvent mixture in which the at least one tri-*O*-acylglycerol
- 30 dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-*O*-acylglycerol is present, and
- d) up to 5.0% by weight, based on the taxane, of additives,
- wherein the at least one taxane is selected from paclitaxel,
- wherein the solvent mixture is a mixture of at least one polar organic solvent having an
- 35 *n*-octanol-water partition coefficient $\log K_{ow}$ of -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of 5.0 to 30 and at least one non-polar organic solvent having a dielectric constant ϵ_r at 20°C of ≤ 3.0 and an *n*-octanol-water partition coefficient $\log K_{ow}$ of ≥ 3.0 .

Thus, the present invention preferably relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- 5 (a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
 - b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel,, and
 - 10 c) a solvent or a solvent mixture in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-*O*-acylglycerol is present, and
 - d) up to 5.0% by weight, based on the taxane, of additives,
- wherein the microcrystals of the at least one taxane have a crystal size ranging from 1
- 15 μm to 300 μm ,
- wherein the at least one taxane is selected from paclitaxel,
- wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

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Thus, the present invention preferably relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- 25 (a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
 - b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel, and
 - 30 c) a solvent or a solvent mixture in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-*O*-acylglycerol is present, and
 - d) up to 5.0% by weight, based on the taxane, of additives,
- wherein the microcrystals of the at least one taxane have a crystal size ranging from 1
- 35 μm to 300 μm ,
- wherein the at least one taxane is selected from paclitaxel,
- wherein the solvent mixture is a mixture of at least one polar organic solvent having an *n*-octanol-water partition coefficient logK_{ow} of -0.5 to +1.5 and a dielectric constant ϵ_r

at 20°C of 5.0 to 30 and at least one non-polar organic solvent having a dielectric constant ϵ_r at 20°C of ≤ 3.0 and an n-octanol-water partition coefficient $\log K_{ow}$ of ≥ 3.0 .

Thus, the present invention preferably relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- (a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
 - b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel, and
 - c) a solvent or a solvent mixture in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-O-acylglycerol is present, and
 - d) up to 5.0% by weight, based on the taxane, of additives,
- wherein the at least one taxane has a crystallinity of at least 80% by weight, wherein the at least one taxane is selected from paclitaxel, wherein the solvent mixture is selected from ethanol and cyclohexane or ethyl acetate and heptane.

Thus, the present invention preferably relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- (a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
 - b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel, and
 - c) a solvent or a solvent mixture in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-O-acylglycerol is present, and
 - d) up to 5.0% by weight, based on the taxane, of additives,
- wherein the microcrystals of the at least one taxane have a crystal size ranging from 1 μm to 300 μm , wherein the at least one taxane has a crystallinity of at least 80% by weight, wherein the at least one taxane is selected from paclitaxel,

wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 , or the solvent mixture contains at least 50% by volume of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

- 5 Thus, the present invention relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:
- (a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
 - 10 b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel, and
 - c) a solvent or a solvent mixture in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-*O*-acylglycerol is present, and
 - 15 d) up to 5.0% by weight, based on the taxane, of additives, wherein the microcrystals of the at least one taxane have a crystal size ranging from 1 μm to 300 μm ,
 - 20 wherein the at least one taxane has a crystallinity of at least 80% by weight, wherein the at least one taxane is selected from paclitaxel, wherein the solvent mixture is a mixture of at least one polar organic solvent having an n-octanol-water partition coefficient $\log K_{ow}$ of -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of 5.0 to 30 and at least one non-polar organic solvent having a dielectric
 - 25 constant ϵ_r at 20°C of ≤ 3.0 and an n-octanol-water partition coefficient $\log K_{ow}$ of ≥ 3.0 .

Thus, the present invention preferably relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- 30 (a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
- b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel,
- 35 and
- c) a solvent or a solvent mixture in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-*O*-acylglycerol is present, and
- d) up to 5.0% by weight, based on the taxane, of additives,

wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present at a mass fraction of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane,
 wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or
 the solvent mixture contains at least 50% by volume of a non-solvent having a dielectric
 5 constant ϵ_r at 20°C of ≤ 2.0 .

Thus, the present invention preferably relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- 10 (a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
- b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel,
 15 and
- c) a solvent or a solvent mixture in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-*O*-acylglycerol is present, and
- d) up to 5.0% by weight, based on the taxane, of additives,
- 20 wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present at a mass fraction of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane,
 wherein the solvent mixture is a mixture of at least one polar organic solvent having an n-octanol-water partition coefficient logK_{OW} of -0.5 to +1.5 and a dielectric constant ϵ_r
 at 20°C of 5.0 to 30 and at least one non-polar organic solvent having a dielectric
 25 constant ϵ_r at 20°C of ≤ 3.0 and an n-octanol-water partition coefficient logK_{OW} of ≥ 3.0 .

Thus, the present invention preferably relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- 30 (a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
- b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel, and
 35 and
- c) a solvent or a solvent mixture in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-*O*-acylglycerol is present, and
- d) up to 5.0% by weight, based on the taxane, of additives,

wherein the microcrystals of the at least one taxane have a crystal size ranging from 1 μm to 300 μm ,

wherein the at least one taxane is selected from paclitaxel,

wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present at a mass fraction of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane,

wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

Thus, the present invention preferably relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

(a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and

b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel, and

c) a solvent or a solvent mixture in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-*O*-acylglycerol is present, and

d) up to 5.0% by weight, based on the taxane, of additives,

wherein the microcrystals of the at least one taxane have a crystal size ranging from 1 μm to 300 μm ,

wherein the at least one taxane is selected from paclitaxel,

wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present at a mass fraction of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane,

wherein the solvent mixture is a mixture of at least one polar organic solvent having an n-octanol-water partition coefficient $\log K_{ow}$ of -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of 5.0 to 30 and at least one non-polar organic solvent having a dielectric constant ϵ_r at 20°C of ≤ 3.0 and an n-octanol-water partition coefficient $\log K_{ow}$ of ≥ 3.0 .

Thus, the present invention preferably relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

(a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and

- b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel, and
 - c) a solvent or a solvent mixture in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-O-acylglycerol is present, and
 - d) up to 5.0% by weight, based on the taxane, of additives, wherein the at least one taxane has a crystallinity of at least 80% by weight, wherein the at least one taxane is selected from paclitaxel,
- wherein the at least one tri-O-acylglycerol and the at least one taxane are present at a mass fraction of 10% - 30% tri-O-acylglycerol to 90% - 70% taxane, wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

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Thus, the present invention preferably relates to a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, the suspension consisting of:

- (a) at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol; and
 - b) at least one taxane in the form of microcrystals, wherein the at least one taxane is selected from the group consisting of paclitaxel, docetaxel, and cabazitaxel, and
 - c) a solvent or a solvent mixture in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve when the at least one tri-O-acylglycerol is present, and
 - d) up to 5.0% by weight, based on the taxane, of additives, wherein the at least one taxane has a crystallinity of at least 80% by weight,
- wherein the at least one taxane is selected from paclitaxel, wherein the at least one tri-O-acylglycerol and the at least one taxane are present at a mass fraction of 10% - 30% tri-O-acylglycerol to 90% - 70% taxane, wherein the solvent mixture is a mixture of at least one polar organic solvent having an n-octanol-water partition coefficient $\log K_{ow}$ of -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of 5.0 to 30 and at least one non-polar organic solvent having a dielectric constant ϵ_r at 20°C of ≤ 3.0 and an n-octanol-water partition coefficient $\log K_{ow}$ of ≥ 3.0 .

Method for preparing the suspension

The present invention further relates to a method for preparing a suspension for coating a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, comprising the following steps:

- a) dissolving at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol in a solvent or solvent mixture;
 - b) addition of at least one taxane in the form of microcrystals to the solution of step a) or addition of the solution of step a) to at least one taxane in the form of microcrystals,
- wherein the microcrystals of the at least one taxane do not dissolve in the solution of step a).

In other words, the present invention further relates to a method for preparing a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, comprising the following steps:

- a) dissolving at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol in a solvent or solvent mixture;
 - b) preparing a suspension of at least one taxane in the form of microcrystals and the solution of step a),
- wherein the microcrystals of the at least one taxane do not dissolve in the solution of step a).

Preferably, the at least one taxane has a crystallinity of at least 80% by weight.

Preferably, the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm , more preferably a crystal size of at most 100 μm , more preferably a crystal size in the range of 10 μm to 100 μm . More preferably, the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel, particularly preferred is paclitaxel. Preferably, the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by vol.% of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 . Preferably, the solvent mixture is a mixture of at least one polar organic solvent with an n-octanol-water partition coefficient $\log K_{OW}$ of -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of 5.0 to 30, and at least one nonpolar organic solvent having a dielectric constant ϵ_r at 20°C of from ≤ 3.0 and an n-octanol water partition coefficient $\log K_{OW}$ of ≥ 3.0 .

With other words, the present invention relates to a method for preparing a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, comprising the following steps:

- (a) providing a solution of at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol in a solvent or a solvent mixture;
 - b) providing at least one taxane in the form of microcrystals,
 - c) preparing a suspension by combining the solution according to step a) and the at least one taxane in the form of microcrystals according to step b),
- wherein the microcrystals of the at least one taxane according to step b) do not dissolve in the solution according to step a).

Preferably, the at least one taxane has a crystallinity of at least 80% by weight. Preferably, the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm , more preferably a crystal size of at most 100 μm , more preferably a crystal size in the range of 10 μm to 100 μm . More preferably, the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel, particularly preferred is paclitaxel. Preferably, the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by vol.% of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 . Preferably, the solvent mixture is a mixture of at least one polar organic solvent with an *n*-octanol-water partition coefficient $\log K_{ow}$ of -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of 5.0 to 30, and at least one nonpolar organic solvent having a dielectric constant ϵ_r at 20°C of from ≤ 3.0 and an *n*-octanol water partition coefficient $\log K_{ow}$ of ≥ 3.0 .

It has proven essential to provide the at least one taxane in the form of microcrystals and to add the at least one taxane in the form of microcrystals together with the solution of the at least one tri-*O*-acylglycerol in the solvent or solvent mixture so that a stable crystal suspension is formed. If, on the other hand, the microcrystals of the taxane are first added to the solvent or solvent mixture and only then the at least one tri-*O*-acylglycerol is added, no suspension of the microcrystals of the taxane according to the invention is formed, from which firmly adhering coatings on medical devices can be produced. The sequence of steps for preparing the crystal suspension of the taxane is therefore essential and cannot be interchanged. Similarly, unsuitable coatings are formed when an attempt is made to produce taxane crystals from a solution of the taxane in a solvent or solvent mixture containing the at least one tri-*O*-acylglycerol after coating on the surface of the medical device.

The present invention also relates to a method for preparing a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, comprising the following steps:

- a') dissolving at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol in a solvent, preferably in a polar organic solvent,
 - a'') adding a non-polar organic solvent, preferably a non-solvent, to the solution from step a'); and optionally homogenizing and filtering,
 - b) addition of at least one taxane in the form of microcrystals to the solution from step a'') or addition of the solution from step a'') to at least one taxane in the form of microcrystals,
- wherein the microcrystals of the at least one taxane are not soluble in the solution of step a'').

The present invention also relates to a method for preparing a suspension for coating of a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, comprising the following steps:

- a') dissolving at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol in a solvent, preferably in a polar organic solvent,
 - a'') adding a non-polar organic solvent, preferably a non-solvent, to the solution from step a'); and optionally homogenizing and filtering,
 - b) providing at least one taxane in the form of microcrystals,
 - c) preparing a suspension by combining the solution according to step a'') and the at least one taxane in the form of microcrystals according to step b),
- wherein the microcrystals of the at least one taxane according to step b) do not dissolve in the solution according to step a'').

Preferably, the at least one taxane has a crystallinity of at least 80% by weight.

Preferably, the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm , more preferably a crystal size of at most 100 μm , more preferably a crystal size in the range of 10 μm to 100 μm . More preferably, the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel, particularly preferred is paclitaxel. Preferably, the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 . or a solvent mixture containing at least 50 vol.% of anon-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 . Preferably, the solvent mixture is a mixture of at least one polar organic solvent with an n-octanol-water partition coefficient $\log K_{ow}$ of -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of

5.0 to 30, and at least one nonpolar organic solvent having a dielectric constant ϵ_r at 20°C of from ≤ 3.0 and an n-octanol water partition coefficient $\log K_{ow}$ of ≥ 3.0 .

Coating method

- 5 The present invention also relates to a method of coating of a medical device, preferably a stent, or a cannula, with a suspension, comprising the following steps:
- a) providing a medical device with a medical device surface,
 - b) providing a suspension comprising at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol,
10 tridecanoylglycerol, and triundecanoylglycerol, at least one taxane in the form of microcrystals, and a solvent or a solvent mixture in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve or do not dissolve when the at least one tri-O-acylglycerol is present; and
 - 15 c) applying the suspension to the medical device surface by means of a syringe method, pipetting method, capillary method, fold spraying method, dipping method, spraying method, dragging method, thread dragging method, drop dragging method or rolling method.
- 20 The present invention also relates to a method of coating of a medical device, preferably a stent, or a cannula, with a suspension, comprising the following steps:
- a) providing a medical device with a medical device surface, optionally with a pre-treated surface (conditioning of the surface), wherein the medical device surface is uncoated or coated;
 - 25 b) providing a suspension comprising at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, at least one taxane in the form of microcrystals, and a solvent or a solvent mixture in which the at least one tri-O-acylglycerol dissolves and in which the microcrystals of the at least one
30 taxane do not dissolve or do not dissolve when the at least one tri-O-acylglycerol is present; and
 - c) applying the suspension to the surface of the medical device by means of a syringe method, pipetting method, capillary method, fold spray method, dipping method, spraying method, dragging method, thread dragging method, drop
35 dragging method or rolling method.

In preferred embodiments, the method further comprises a step d) drying the coating after step c).

Preferred herein are special coating methods for coating of medical devices in which the medical device can be coated with a defined amount of microcrystalline taxane, wherein in said coating methods a coating device with a volume measuring device for the targeted delivery of a defined amount of the coating suspension according to the invention onto the medical device surface by means of a dispensing device is preferably used.

Any device capable of providing a defined amount of coating suspension or measuring or indicating the dispensed amount of coating suspension can serve as a volume measuring device. Volume measuring devices are therefore in the simplest case scales, scaled pipettes, scaled burettes, scaled containers, scaled cavities as well as pumps, valves, syringes or other piston-shaped containers, which are capable of providing or conveying or dispensing a defined quantity of coating suspension. Thus, the volume measuring device serves to either provide or dispense a defined amount of a coating suspension or to measure and/or display a dispensed amount of coating suspension. Thus, the volume measuring device serves to determine or measure the amount of coating suspension and thus of microcrystalline taxane transferred from the dispensing device to the medical device surface.

The most important aspect of the coating device, however, is the dispensing device, which can be designed as a nozzle, a plurality of nozzles, a thread, a network of threads, a piece of textile, a leather strip, a sponge, a ball, a syringe, a needle, a cannula or a capillary. Depending on the design of the dispensing device, slightly modified coating methods result, all of which are based on the basic principle of transferring a measurable or defined amount of microcrystalline taxane to the surface of the medical device without loss. In this way, a coating with a defined active agent concentration or active agent amount of microcrystalline taxane and thus a reproducible coating is provided. Various terms are used herein to distinguish the methods, namely, syringe method, pipetting method, capillary method, fold spray method, dipping method, spraying method, dragging method, thread dragging method, droplet dragging method, or rolling method, which are the preferred embodiments of the present invention.

As a particularly preferred coating method of medical devices with crystal suspensions, the droplet dosing technique using the microdosing method such as the pipetting method or droplet dragging method is used herein. These can be used to obtain particularly uniform coatings with likewise uniform active agent concentration of the microcrystalline taxane on the medical device surface, as long as it is ensured that the microcrystals of the taxane remain uniformly distributed.

The studies on the recovery rate of active agent on balloon catheters divided into equal segments confirm the uniformity of the coating and thus the success of using a crystal suspension as well as a 100% recovery rate.

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In addition, to ensure uniform distribution of the crystals, a pneumatic swivel can be used to agitate the suspension during the coating process to prevent possible sedimentation, which can be advantageous as a precautionary measure for crystal contents of less than 2% (w/v).

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On the other hand, for any pretreatment of the medical device surface, e.g. conditioning or applying a base coat, the other common coating methods such as spraying, dipping, brushing, pipetting, drop dragging, rolling, spinning, in situ deposition, screen printing, vapor deposition or spraying can also be used. The above methods can also be

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Coated medical device

The suspension of the present invention is particularly suitable for providing coated medical devices having an active agent-releasing coating comprising at least one tri-
O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals.

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The present invention therefore relates to coated medical devices, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, having a coating comprising at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals.

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The term "**coating**" is intended to include not only a coating on the medical device surface but also a filling or coating of folds, cavities, pores, microneedles or other fillable spaces on or between or within the material as well as, in the case of expandable, folded or collapsed medical devices, deflated, partially inflated and fully
inflated or unfolded or partially unfolded medical devices.

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The term "**on the medical device surface**", as used herein, preferably means that an application is made directly to the medical device surface, i.e., directly to the material of the medical device. For example, if the medical device is made of polyamide, this

means that the application is made to the polyamide from which the medical device is made. If the medical device is made of, for example, polyamide and is subsequently coated with a polymer, then the application would not be made on the medical device surface.

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It is preferred if the entire medical device surface is uniformly coated. Furthermore, it is preferred if there is a uniform distribution of the microcrystalline taxane on the medical device surface. However, the surface can also be only partially coated or coated differently at different points (e.g., different coating thicknesses, different coatings, different active agent concentrations, only selected delimited areas, etc.).

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The term "**medical device**", as used herein, refers to articles or substances that serve to detect, prevent, monitor, treat or alleviate diseases, but achieve this purpose primarily ("intended primary effect") by physical means rather than by pharmacological/immunological means or by metabolic action. However, the physical effect of medical devices may well be supported by pharmacological, immunological or metabolic effects. Medical devices can be divided into medical devices for long-term use and medical devices for short-term use, depending on whether the medical device has short-term or long-term contact with the organism. All medical devices that are intended to remain in the body are considered to be for long-term use. Medical devices with less long-term to very short-term use are medical devices that can be removed after a certain period of time and are used for a limited period of time.

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Examples of long-term medical devices include, but are not limited to, non-biodegradable, biostable stents, implants, joint implants, vascular prostheses, brain pacemakers (such as those used in used for Parkinson's disease), artificial hearts, port catheters, vision implants, ocular lens replacements, retinal replacements, vitreous replacements, corneas, dental implants, cochlear implants, reconstructive implants, cranial reconstructions, bone replacements, penile prostheses, sphincter prostheses and the like.

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Examples of less long-term to very short-term medical devices include, but are not limited to, all forms and types of catheters, balloon catheters, angioplasty catheters, bladder catheters, breathing tubes, venous catheters, cannulas of all types, needles, winged cannulas (butterflies), drug depots, fixations, e.g., for surgical treatment of bone fractures, artificial accesses, tubes, sutures, staples, and the like.

35

The term "balloon" or "**catheter balloon**" generally refers to any expandable and re-compressible as well as temporarily implantable medical device, which is generally

used in conjunction with a catheter. The term "catheter balloon" as used herein refers to the dilatable portion, i.e., the balloon of a balloon catheter. **"Balloon catheter"** refers to a dilatation balloon catheter. Balloon catheter is a medical term for catheters that have a balloon attached to them. Examples of balloon catheters include, but are not limited to angioplasty balloon catheters used in percutaneous transluminal angioplasty to dilate and open narrowed or occluded blood vessels, bladder catheters, thrombectomy catheters used in treatment in vascular surgery, neuroradiology, and cardiology in the treatment of embolized and secondarily thrombosed peripheral arteries, but also used in neurothrombectomy in stroke therapy, embolectomy catheters used in vascular surgery for removal of fresh and soft emboli in the peripheral arterial system, Fogarty catheters, double balloon catheters, balloon catheters used in pneumology, micro-balloon catheters.

In some embodiments of the present invention, the medical device is selected from the group comprising or consisting of a catheter balloon, a balloon catheter, an angioplasty catheter, a bladder catheter, a peripheral catheter, a coronary catheter, an embolectomy catheter, a thrombectomy catheter, a neurothrombectomy catheter, a stent, an implant, a joint implant, a vascular prosthesis, a port catheter, a visual prosthesis, an ocular implant, a dental implant, a cochlear implant, a reconstructive implant, a penile prosthesis, a sphincter prosthesis, a cardiac pacemaker, a brain pacemaker, a breathing tube, a venous catheter, a cannula, a needle, a winged cannula (butterfly), an artificial access, a tube, sutures, and a medical staple.

In particularly preferred embodiments of the present invention, the medical device is selected from the group comprising or consisting of a catheter a stent, a cannula. Thus, in these embodiments, the medical device is preferably selected from the group comprising or consisting of a catheter balloon, a balloon catheter, an angioplasty catheter, a bladder catheter, a port catheter, a vein catheter, a stent, a bioresorbable stent, a cannula, a hypodermic needle, a winged (butterfly) cannula, a peripheral vein indwelling cannula, and an epidural cannula. Further preferably, the medical device is selected from the group comprising or consisting of a catheter balloon, a balloon catheter, an angioplasty catheter, and a stent. Still further preferably, the medical device is selected from the group comprising or consisting of a catheter balloon, a balloon catheter, and an angioplasty catheter. More preferably, the medical device is a catheter balloon.

Catheter balloons, regardless of their field of application, can be made of common biocompatible flexible materials, in particular polymers as described further below and in particular polyamide, such as PA 12, polyester, polyurethanes, polyacrylates,

polyethers, Pebax, etc. but also of combinations of suitable polymers, e.g. of superimposed layers of these materials, as well as of copolymers of these materials, blends and combinations of the embodiments of layers and copolymers and their blends.

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An implant can be made of common biocompatible materials such as medical stainless steel, titanium, chromium, vanadium, tungsten, molybdenum, gold, iron, nitinol, magnesium, iron, zinc, alloys of the aforementioned metals, ceramics as well as polymeric biostable or bioresorbable material such as e.g. PTFE, polysulfones, polyvinylpyrrolidone, polyamide, e.g. PA 12, polyester, polyurethane, polyacrylates, polyethers, silicone, PMMA, combinations thereof, etc. The materials are either bioinert, biostable and/or biodegradable, and the implant is expandable, compressible or non-shape-changing.

15 The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

20 wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm .

The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

25 wherein the at least one taxane has a crystallinity of at least 80% by weight.

30 The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

35 wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel.

The present invention preferably relates to a medical device selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-

acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol and at least one taxane in the form of microcrystals,

wherein the at least one taxane is selected from paclitaxel.

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The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one

10 taxane in the form of microcrystals,

wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel.

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The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one

20 taxane in the form of microcrystals,

wherein the at least one taxane has a crystallinity of at least 80% by weight,

wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel.

25 The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

30 wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

wherein the at least one taxane is selected from paclitaxel.

35 The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

wherein the at least one taxane has a crystallinity of at least 90% by weight,

wherein the at least one taxane is selected from paclitaxel.

The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

wherein the at least one taxane has a crystallinity of at least 80% by weight,

wherein the at least one taxane is selected from paclitaxel.

The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

wherein the at least one tri-O-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-O-acylglycerol to 90% - 70% taxane.

The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel,

wherein the at least one tri-O-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-O-acylglycerol to 90% - 70% taxane.

The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

wherein the at least one taxane has a crystallinity of at least 80% by weight,

wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel,

wherein the at least one tri-O-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-O-acylglycerol to 90% - 70% taxane.

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The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol and at least one

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taxane in the form of microcrystals, wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

wherein the at least one taxane is selected from paclitaxel,

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wherein the at least one tri-O-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-O-acylglycerol to 90% - 70% taxane.

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The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

wherein the at least one taxane has a crystallinity of at least 80% by weight,

wherein the at least one taxane is selected from paclitaxel,

25

wherein the at least one tri-O-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-O-acylglycerol to 90% - 70% taxane.

30

The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol and at least one taxane in the form of microcrystals,

wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

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wherein at least 70% of the taxane is in the form of microcrystals having a crystal size in the range of 10 μm to 100 μm .

The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol,

tridononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

wherein the at least one taxane has a crystallinity of at least 80% by weight,

wherein at least 70% of the taxane is in the form of microcrystals having a crystal
5 size in the range of 10 μm to 100 μm .

The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol,
10 tridononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol and at least one taxane in the form of microcrystals,

wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel,

wherein at least 70% of the taxane is in the form of microcrystals having a crystal
15 size in the range of 10 μm to 100 μm .

The present invention preferably relates to a medical device selected from a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, tridononoylglycerol, tridecanoylglycerol, and
20 triundecanoylglycerol, and at least one taxane in the form of microcrystals,

wherein the at least one taxane is selected from paclitaxel,

wherein at least 70% of the taxane is in the form of microcrystals having a crystal
size in the range of 10 μm to 100 μm .

25 The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, tridononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

30 wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel,

wherein at least 70% of the taxane is in the form of microcrystals having a crystal
35 size in the range of 10 μm to 100 μm .

The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol,

trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

wherein the at least one taxane has a crystallinity of at least 80% by weight,

wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel,

wherein at least 70% of the taxane is in the form of microcrystals having a crystal size in the range of 10 μm to 100 μm .

The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

wherein the at least one taxane is selected from paclitaxel,

wherein at least 70% of the taxane is in the form of microcrystals having a crystal size in the range of 10 μm to 100 μm .

The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

wherein the at least one taxane has a crystallinity of at least 80% by weight,

wherein the at least one taxane is selected from paclitaxel,

wherein at least 70% of the taxane is in the form of microcrystals having a crystal size in the range of 10 μm to 100 μm .

The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

wherein the at least one taxane has a crystallinity of at least 80% by weight,

wherein the at least one taxane is selected from paclitaxel,

wherein at least 70% of the taxane is in the form of microcrystals having a crystal size in the range of 10 μm to 100 μm .

5 The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

10 wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane,

wherein at least 70% of the taxane is in the form of microcrystals having a crystal size in the range of 10 μm to 100 μm .

15 The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals,

20 wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel,

wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane,

25 wherein at least 70% of the taxane is in the form of microcrystals having a crystal size in the range of 10 μm to 100 μm .

30 The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula coated with at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinononoylglycerol, tridecanoylglycerol, and triundecanoylglycerol and at least one taxane in the form of microcrystals,

wherein the at least one taxane has a crystallinity of at least 80% by weight,

35 wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel,

wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane,

wherein at least 70% of the taxane is in the form of microcrystals having a crystal size in the range of 10 μm to 100 μm .

The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula coated with at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol and at least one taxane in the form of microcrystals, wherein the microcrystals of the at least one taxane have a crystal size in the range of 1 μm to 300 μm ,

wherein the at least one taxane is selected from paclitaxel, wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane, wherein at least 70% of the taxane is in the form of microcrystals having a crystal size in the range of 10 μm to 100 μm .

The present invention preferably relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent, or a cannula coated with at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol and at least one taxane in the form of microcrystals,

wherein the at least one taxane has a crystallinity of at least 80% by weight, wherein the at least one taxane is selected from paclitaxel, wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30% tri-*O*-acylglycerol to 90% - 70% taxane, wherein at least 70% of the taxane is in the form of microcrystals having a crystal size in the range of 10 μm to 100 μm .

For the preparation of the coatings, suspensions according to the invention are used which contain the taxane in the form of microcrystals together with at least one dissolved tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol in a solvent or solvent mixture.

The present invention therefore relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula coated with a suspension containing:

- a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol;
- b) at least one taxane in the form of microcrystals; and

- c) a solvent or a solvent mixture wherein the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve or do not dissolve when at least one tri-*O*-acylglycerol is present.
- 5 The present invention therefore relates to a medical device, preferably selected from a catheter balloon, a balloon catheter, a stent or a cannula, obtainable according to a method comprising the following steps:
- a) providing the medical device selected from a catheter balloon, a balloon catheter, a stent, or a cannula, having a medical device surface;
 - 10 b) providing a suspension comprising a tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol dissolved in a solvent or a solvent mixture and at least one taxane in the form of microcrystals, wherein the microcrystals of the at least one taxane do not dissolve in the solvent or the solvent mixture or do not
 - 15 dissolve when the at least one tri-*O*-acylglycerol is present;
 - c) applying the suspension to the surface of the medical device by means of a syringe method, pipetting method, capillary method, fold spray method, dipping method, spraying method, dragging method, thread dragging method, drop dragging method, or rolling method,
 - 20 d) drying the coating

In some embodiments of the present invention, a medical device having a medical device surface may be provided that has a base coating on the medical device surface. In such embodiments, the suspension according to the invention comprising at least

25 one -*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol and at least one taxane in the form of microcrystals are applied on this base coating.

For example, the medical device surface may additionally be provided with a

30 hemocompatible, athrombogenic layer as a base coating applied by covalent immobilization of semisynthetic heparin derivatives such as desulfated, reacylated heparin or chitosan derivatives such as N-carboxymethylated, partially N-acetylated chitosan.

35 If necessary or advantageous, the implant surface can be pretreated e.g. by surface activation e.g. via plasma process, temperature treatment, wetting with suitable solvents, DLC coating ("diamond like carbon"), Teflon coating, or siliconization, etc. It could be shown that wetting with a suitable solvent has a positive effect on adhesion.

Likewise, a polymeric base coating with biodegradable and/or biostable polymers can be realized. These polymeric coatings can of course also contain additives, e.g. further active agents or mixtures of active agents, metals, salts etc. Suitable active agents or combinations of active agents include anti-inflammatory, cytostatic, cytotoxic, antiproliferative, anti-microtubule, antiangiogenic, antirestenotic (anti-restenosis), antifungal, antineoplastic, antimigrative, athrombogenic and antithrombogenic substances.

If the taxane in the form of microcrystals is not applied directly to or on the medical device surface, suitable biocompatible substances of synthetic, semisynthetic and/or native origin biostable and/or biodegradable polymers, or polysaccharides can be used as a carrier or matrix.

The following can be named as polymers that are generally biologically stable and only slowly biodegradable: polyacrylic acid and polyacrylates such as polymethyl methacrylate, polybutyl methacrylate, polyacrylamide, polyacrylonitrile, polyamides, polyetheramides, polyethyleneamine, polyimides, polycarbonates, polycarbourethanes, polyvinyl ketones, polyvinyl halides, polyvinylidene halides, polyvinyl ethers, polyvinyl aromatics, polyvinyl esters, polyvinyl pyrrolidones, polyoxymethylenes, polyethylene, polypropylene, polytetrafluoroethylene, polyurethanes, polyolefin elastomers, Polyisobutylenes, EPDM rubbers, fluorosilicones, carboxymethyl chitosan, polyethylene terephthalate, polyvalerates, carboxymethyl cellulose, cellulose, rayon, rayon triacetate, cellulose nitrates, cellulose acetates, hydroxyethyl cellulose, cellulose butyrates, cellulose acetate butyrates, ethyl vinyl acetate copolymers, polysulfones, polyether sulfones, epoxy resins, ABS resins, EPDM rubbers, silicone prepolymers, silicones such as polysiloxanes, polyvinyl halogens and copolymers, cellulose ethers, cellulose triacetates, chitosan, chitosan derivatives, polymerizable oils such as linseed oil and copolymers.e.g. linseed oil, and copolymers and/or mixtures thereof.

As generally biodegradable, biodegradable or resorbable polymers can be used, for example: Polyvalerolactones, poly- ϵ -decalactones, polylactides, polyglycolides, copolymers of polylactides and polyglycolides, poly- ϵ -caprolactone, polyhydroxybutyric acid, polyhydroxybutyrates, polyhydroxyvalerates, polyhydroxybutyrate-co-valerates, poly(1,4-dioxane-2,3-diones), poly(1,3-dioxan-2-one), poly-para-dioxanones, polyanhydrides such as polymaleic anhydrides, polyhydroxymethacrylates, fibrin, polycyanoacrylates, polycaprolactone dimethylacrylates, poly-b-maleic acid, polycaprolactone butyl acrylates, multiblock polymers such as e.g. of oligocaprolactone diols and oligodioxanone diols, polyether

ester multiblock polymers such as PEG and poly(butylene terephthalate,
 polypivotolactones, polyglycolic acid trimethyl carbonate, polycaprolactone glycolides,
 poly(g-ethylglutamate), poly(DTH-iminocarbonate), poly(DTE-co-DT-carbonate),
 Poly(bisphenol A-iminocarbonate), polyorthoester, polyglycol acid trimethyl carbonate,
 5 polytrimethyl carbonate, polyiminocarbonate, poly(N-vinyl) pyrrolidone, polyvinyl
 alcohols, polyesteramides, glycolated polyesters, polyphosphoesters,
 polyphosphazenes, poly[p-carboxyphenoxy]propane], polyhydroxypentanoic acid,
 polyethylene oxide-propylene oxide, soft polyurethanes, polyurethanes with amino
 acid residues in the backbone, polyether esters such as polyethylene oxide,
 10 polyalkenoxalates, polyorthoesters and their copolymers, carrageenans, fibrinogen,
 starch, collagen, protein-based polymers, polyamino acids, synthetic polyamino acids,
 zein, modified zein, polyhydroxyalkanoates, pectic acid, actinic acid, modified and
 unmodified fibrin and casein, carboxymethyl sulfate, albumin, hyaluronic acid, heparan
 sulfate, heparin, chondroitin sulfate, dextran, b-cyclodextrins, copolymers with PEG
 15 and polypropylene glycol, gum arabic, guar, gelatin, collagen, collagen-N-
 hydroxysuccinimide, lipids and lipoids, polymerizable oils with a low degree of
 crosslinking, modifications and copolymers and/or mixtures of the above substances.

Description of the Figures

Figure 1 shows the model formed from silicone tubing that mimics the natural course of vessels in the organism a) Simulated Peripheral Catheter; b) Simulated Femoral Artery.

5 **Figure 2** a) shows a bending test to determine the particle release of a coated catheter balloon. b) shows an edge impact test to determine the particle release of a coated catheter balloon.

10 **Figure 3** shows a prior art coated catheter balloon. a) clearly shows that a uniform coating is missing, the surface is also uneven; b) the coating crumbles off during inflation.

Figure 4 shows a prior art coated catheter balloon with large pieces of the coating peeling off along the entire length of the catheter balloon.

15 **Figure 5** shows a coating with trioctanoylglycerol/PTX during and after inflation a uniform coating with uniform surface structure, the coating does not peel off during inflation of the balloon.

Examples

Example 1

5 **Preparation of microcrystalline paclitaxel**

For the preparation of the crystal suspensions according to the present invention, microcrystals of paclitaxel was first provided. Crystallization processes for the preparation of crystalline paclitaxel are known from the prior art. Crystallization processes well known from the prior art include:

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Crystallization by cooling: Paclitaxel can be dissolved in a solvent at room or higher temperature until saturation and brought to crystallization at lower temperature e.g. at 0°C. The crystal size distribution can be influenced by a controlled cooling rate. Solvents, such as toluene, acetone, ethanol, dimethyl formamide, methanol are suitable solvents for crystallization.

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Crystallization by addition of seed crystals: Paclitaxel is dissolved to saturation in a solvent and crystallization is initiated by the addition of seed crystals to achieve a controlled reduction of supersaturation.

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Crystallization by addition of anti-solvent: Paclitaxel is dissolved in a solvent and then a non-solvent or water is added. Two-phase mixtures are also possible here. Polar organic solvents such as acetonitrile, ethanol, dimethyl formamide or dimethyl sulfoxide can be used as solvents for dissolving the paclitaxel. Suitable non-solvents include pentane, hexane, cyclohexane or heptane. The solvent mixture can be allowed to stand for crystallization, stirred or slowly concentrated or evaporated *in vacuo*. The crystal size and crystallinity of the drug can be influenced by controlled addition of the nonpolar solvent. Supersaturation should be slower to produce large crystals and faster to produce small crystals. Controlling the addition rate of the anti-solvent to control the crystal size is well known.

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For the production of microcrystals, crystallization can also be assisted by ultrasound. It is generally known that crystal size can be influenced by means of ultrasound. In this context, ultrasound can be used at the beginning of crystallization to initiate crystallization and nucleation, with further crystal growth then proceeding unhindered so that larger crystals can grow. The application of continuous sonication of a supersaturated solution with ultrasound, on the other hand, leads to smaller crystals, as many nuclei are formed in this process, resulting in the growth of numerous small

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crystals. Another option is to sonicate with ultrasound in pulse mode to influence crystal growth in such a way that tailored crystal sizes are achieved.

Other methods known from the prior art, such as micronization, grinding or sieving, can also be used to provide the desired crystal sizes. One possibility is to grind the crystals, which can also be done during crystallization by wet grinding. Milling can be advantageous to obtain different crystal sizes, i.e. a broader crystal size distribution. Milling allows for any desired sizes in the crystal size range. More uniform crystal sizes can be provided by, for example, performing a special sieving process after isolation and drying. Special sieving devices known from the prior art can be used for this purpose. In the sieving process, the paclitaxel crystals can be sieved through a stack of sieves, for example, and divided into different size ranges.

For the preparation of microcrystalline paclitaxel, crystallization procedures with controlled crystallizations were carried out. Paclitaxel could thus be obtained directly in the form of microcrystals. After crystallization, the microcrystals were isolated, washed (heptane) and dried. Optionally, further separation into different crystal sizes was then performed by sieving method to provide narrower crystal size distributions of the microcrystals.

The obtained microcrystals of paclitaxel were used to prepare crystal suspensions in the following examples.

Example 2

Preparation of crystal suspensions with tri-*O*-acylglycerols and microcrystalline paclitaxel

In the first step, solutions of tri-*O*-acylglycerols were first prepared in a solvent mixture. Subsequently, the solutions were combined with microcrystals of paclitaxel to investigate which tri-*O*-acylglycerols could be used to obtain stable crystal suspensions. The composition of the solvents and solvent mixture varied depending on the active agent used. The solutions and solvent mixtures prepared in the example apply to paclitaxel. For the preparation of the solutions, an ethyl acetate/heptane solvent mixture was used here as an example.

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1. Preparation of solutions with tri-*O*-acylglycerols

To prepare the solutions, the respective tri-*O*-acylglycerol was first dissolved in a polar organic solvent and then a non-polar solvent was added.

To prepare the solutions, 770 mg of the respective tri-*O*-acylglycerol were dissolved in 14g ethyl acetate (15.6 mL). Then, 57.4 g of n-heptane (84.4 mL) was added. Subsequently, homogenization and filtration were performed. The total volume of the solvent mixture is 100 mL.

5

Solution 1a)

Tri-*O*-acylglycerol: Trioctanoylglycerol

Solution 1b)

Tri-*O*-acylglycerol: Tridecanoylglycerol

10 Solution 1c)

Tri-*O*-acylglycerol: Trihexanoylglycerol

Solution 1d)

Tri-*O*-acylglycerol: Tributanoyleglycerol

Solution 1e)

15 Tri-*O*-acylglycerol: 770mg Triacetin

Solution 1f)

Tri-*O*-acylglycerol: Tridodecanoylglycerol

Solution 1g)

Tri-*O*-acylglycerol: Citryl/Lactyl/Linoleyl/Oleyl-*O*-Glycerols (IMWITOR®)

20 Solution 1h)

Tri-*O*-acylglycerol: Dioctanoylglycerol

Solution 1i)

Tri-*O*-acylglycerol: Monoctanoylglycerol

Solution 1j)

25 Tri-*O*-acylglycerol: Tritetradecanoylglycerol

2. Redispersion of microcrystals of paclitaxel

To a precisely weighed amount of dry microcrystals of the previously prepared paclitaxel, a defined amount of the solutions containing tri-*O*-acylglycerol and optionally antioxidant is carefully added. It was investigated whether the microcrystals of the paclitaxel are not soluble in the solutions and whether suspensions are formed.

To check whether crystal suspension can be prepared, to each 200 mg of paclitaxel in the form of microcrystals were carefully added 10 mL of one of the solutions 1a) to 1j) at room temperature. Six mixtures of 10 mL of solution for each solution were prepared. After combining, it was tested whether the microcrystals of paclitaxel dissolved directly in the solutions. For solutions where there was no immediate dissolution of the microcrystals, the suspensions were allowed to stand for a period of 100 h or heated

to 50°C to check whether the suspensions remained stable under sterilization conditions, and checked again, if the microcrystals dissolved.

Table 9: Overview of results for the preparation of crystal suspensions with solutions containing different tri-*O*-acylglycerols (+++++ = stable crystal suspension, uniform distribution of microcrystals, intact microcrystals, "floating" of microcrystals; +++ = suspension, no complete dissolution of microcrystals; + = sedimentation or partial dissolution of microcrystals, no intact crystals; --- = no suspension; -- = no suspension; - = no suspension; -/- = not studied).

Solution (10 mL)	Suspension	Suspension (100 h)	Suspension (50°C)
1a)	+++++	+++++	+++++
1b)	+++++	+++++	+++++
1c)	++	+	---
1d)	---	-/-	-/-
1e)	--	-/-	-/-
1f)	+++	+	+
1g)	-	-/-	-/-
1h)	--	-/-	-/-
1i)	--	-/-	-/-
1j)	+++	+	+

Result:

Stable crystal suspensions could be prepared with the tri-*O*-acylglycerols trioctanoylglycerol and tridecanoylglycerol, which remained stable even after 100 h and under temperature elevation to 50°C. In the solutions with trioctanoylglycerol and tridecanoylglycerol, no sedimentation of the microcrystals occurred, the microcrystals "float" in the crystal suspension and are uniformly distributed.

To evaluate the crystal size, particle size distribution (i.e., crystal size distribution), and shape of the crystals, a sample was taken with a Pasteur pipette in each case and a drop was placed on the slide of the SEM sample plate. SEM images were taken at 200x and 1000x magnification for evaluation.

In the solutions with trihexanoylglycerol, the microcrystals of paclitaxel did not dissolve directly. In these suspensions, the microcrystals were not as uniformly present after

100 h in contrast to the crystal suspensions with trioctanoylglycerol and tridecanoylglycerol. When the temperature was increased, the microcrystals of paclitaxel dissolved rapidly.

- 5 In the solutions with tridodecanoylglycerol and tritetradecanoylglycerol, the microcrystals of paclitaxel did not dissolve directly. However, these "suspensions" were also found to be unstable. In the case of tridodecanoylglycerol and tritetradecanoylglycerol, the microcrystals of paclitaxel did not dissolve completely after 100 h. However, the microcrystals of paclitaxel were still in a stable state. However, in contrast to the crystal suspensions with trioctanoylglycerol and tridecanoylglycerol, the microcrystals were not as uniformly distributed and sedimentation of the crystals occurred. An increase in temperature to 50°C accelerated this process, and checked again, if the microcrystals dissolved.
- 15 To evaluate the crystal size, particle size distribution (i.e., crystal size distribution), and shape of the crystals, one sample was taken from each using a Pasteur pipette and one drop was placed on the foil of the SEM sample plate. An additional sample was taken from the sediment and one drop was placed on the slide of the SEM sample plate. SEM images were taken at 200x and 1000x magnification for evaluation. SEM images were taken at 200 and 1000x magnification for evaluation.

SEM images showed that the microcrystals of the crystal suspensions containing tridodecanoylglycerol and tritetradecanoylglycerol did not remain intact. The crystal size distribution no longer matched the crystal size distribution of the microcrystalline paclitaxel originally used, and larger crystals were detected, especially in that of the sample taken from the sediment.

No crystal suspensions could be prepared with the other solutions of tri-O-acylglycerols. Thus, stable crystal suspensions could only be prepared with trioctanoylglycerol and tridecanoylglycerol.

Example 3

Preparation of crystal suspensions with additional tri-O-acylglycerols and microcrystalline paclitaxel

- 35 Based on the results from Example 2, further solutions of the three tri-O-acylglycerols triheptanoylglycerol, trinonanoylglycerol and triundecanoylglycerol were prepared to investigate whether stable crystal suspensions can be obtained with these. An ethyl acetate/heptane solvent mixture was also used here to prepare the solutions.

1. Preparation of solutions with tri-O-acylglycerols

To prepare the solutions, 770 mg of the respective tri-O-acylglycerol were dissolved in 14g ethyl acetate (15.6 mL). Then, 57.4 g of n-heptane (84.4 mL) was added. Subsequently, homogenization and filtration were performed. The total volume of the solvent mixture is 100 mL.

Solution 2a)

Tri-O-acylglycerol: Triheptanoylglycerol

Solution 2b)

Tri-O-acylglycerol: Trinonanoylglycerol

Solution 2c)

Tri-O-acylglycerol: Triundecanoylglycerol

2. Redispersion of microcrystals of paclitaxel

To prepare crystal suspension, 200 mg of paclitaxel each were carefully added to 10 mL of solutions 2a) to 2c) at room temperature, as in Example 2. Three mixtures of 10 mL of solution for each solution were prepared. After combining, it was tested whether the microcrystals dissolved directly in these solutions. For solutions where there was no immediate dissolution of the microcrystals, the suspensions of the solutions without antioxidant were allowed to stand for a period of 100 hours or heated to 50°C to check whether the suspensions remained stable under sterilization conditions and checked again, if the microcrystals dissolved.

For direct comparison, the results from example 2 for solutions 1a) to 1d) are included in the following table.

Table 10: Overview results for the preparation of crystal suspensions of paclitaxel with solutions containing different tri-O-acylglycerols (+++++ = excellent stable crystal suspension, uniform distribution of microcrystals, intact microcrystals, "floating" of microcrystals; ++++ = stable crystal suspension, uniform distribution of microcrystals, intact microcrystals, "floating" of microcrystals; +++ = crystal suspension,; ++ = suspension, no complete dissolution of microcrystals; + = sedimentation or partial dissolution of microcrystals, no intact crystals).

Solution (10 mL)	Suspension	Suspension (100 h)	Suspension (50°C)
1a)	+++++	+++++	+++++
1b)	+++++	+++++	+++++
2a)	+++	+	-

2b)	+++++	++++	++++
2c)	+++++	++++	++++

Results:

- 5 The tri-*O*-acylglycerols tridonanoylglycerol and triundecanoylglycerol were used to prepare stable crystal suspensions that remained stable after 24-48 hours and under temperature elevation to 50°C. In the solutions with tridonanoylglycerol and triundecanoylglycerol, no sedimentation of the microcrystals occurred; the microcrystals "floated" in the crystal suspension and were uniformly distributed.
- 10 To evaluate the crystal size, particle size distribution (i.e., crystal size distribution), and shape of the crystals, a sample was taken with a Pasteur pipette in each case and a drop was placed on the slide of the SEM sample plate. SEM images were taken at 200x and 1000x magnification for evaluation.
- 15 SEM images showed that the microcrystals of the crystal suspensions containing the tri-*O*-acylglycerols tridonanoylglycerol and triundecanoylglycerol remained intact. The microcrystals of paclitaxel were consistently in the form of needles. The crystal size distribution also continued to correspond to the crystal size distribution of the microcrystalline paclitaxel used.
- 20 In the solutions with triheptanoylglycerol, the microcrystals of paclitaxel did not dissolve directly. However, the microcrystals of paclitaxel were partially dissolved after 100 h, and under increasing the temperature, the microcrystals of paclitaxel dissolved.
- 25 To evaluate the crystal size, particle size distribution (i.e., crystal size distribution), and shape of the crystals, a sample was taken with a Pasteur pipette in each case and a drop was placed on the slide of the SEM sample plate. SEM images were taken at 200x and 1000x magnification for evaluation.
- 30 SEM images showed that the microcrystals of the crystal suspensions with triheptanoylglycerol did not remain intact. The crystal size distribution no longer matched the crystal size distribution of the microcrystalline paclitaxel originally used.
- 35 Stable crystal suspensions could thus be prepared with the further tri-*O*-acylglycerols tridonanoylglycerol and triundecanoylglycerol.

Example 4

Preparation of 3% and 1% crystal suspensions containing trioctanoylglycerol and microcrystalline paclitaxel (PTX)

I. Preparation of the solutions of trioctanoylglycerol

- 5 la)
Solvent mixture example for a 100ml batch with 3% PTX crystal content. In 14 g ethyl acetate, 770 mg trioctanoylglycerol is dissolved. To this solution, 57.4 g of n-heptane is added, homogenized and filtered.
- 10 lb) Solution mixture example for a 100 ml batch with 1% PTX crystal content. In 14 g ethyl acetate 250 mg trioctanoylglycerol, and 20 mg Tween 80 are dissolved. To this solution 57.4 g of n-heptane is added, homogenized and filtered.

II. Preparation of the crystal suspension

- 15 A defined quantity of the solvent mixture is carefully added to a precisely weighed quantity of dry active agent crystals prepared in advance. The crystals, which are insoluble in the solvent mixture, form a suspension with the solvent mixture. For solution la), 3 g of paclitaxels in the form of microcrystals were used and for solution lb), 1 g of paclitaxel in the form of microcrystals were used.

20

Example 5

Preparation of crystal suspensions containing microcrystals of paclitaxel with different proportions of tri-O-acylglycerols

- In Examples 2 and 3, it was shown that for the tri-O-acylglycerols trioctanoylglycerol, tridecanoylglycerol, trinonanoylglycerol, and triundecanoyl-glycerol, stable crystal suspensions could be obtained at a mass ratio of tri-O-acylglycerols to microcrystals of paclitaxel of 20:80.
- 25

- For this purpose, further investigation was carried out using solutions of trioctanoylglycerol or tridecanoylglycerol with different proportions of tri-O-acylglycerol to find out the optimum mass ratio of tri-O-acylglycerol to microcrystalline paclitaxel. To prepare the solutions, the appropriate amount of each tri-O-acylglycerol was dissolved in 14 g ethyl acetate (15.6 mL). Then, 57.4 g of n-heptane (84.4 mL) was added. Subsequently, homogenization and filtration were performed. The total volume of the solvent mixture is 100 mL.
- 30
- 35

Solution 3a)

Tri-O-acylglycerol: Trioctanoylglycerol

Weighing-in: 300 mg

Solution 3b)

Tri-O-acylglycerol: Trioctanoylglycerol

Weighing-in: 450 mg

Solution 3c)

5 Tri-O-acylglycerol: Trioctanoylglycerol

Weighing-in: 600 mg

Solution 3d)

Tri-O-acylglycerol: Trioctanoylglycerol

Weighing-in: 900 mg

10 Solution 3e)

Tri-O-acylglycerol: Trioctanoylglycerol

Weighing-in: 1200 mg

Solution 3f)

Tri-O-acylglycerol: Trioctanoylglycerol

15 Weighing-in: 1500 mg

Solution 4a)

Tri-O-acylglycerol: Tridecanoylglycerol

Weighing-in: 300 mg

Solution 4b)

20 Tri-O-acylglycerol: Tridecanoylglycerol

Weighing-in: 450 mg

Solution 4c)

Tri-O-acylglycerol: Tridecanoylglycerol

Weighing-in: 600 mg

25 Solution 4d)

Tri-O-acylglycerol: Tridecanoylglycerol

Weighing-in: 900 mg

Solution 4e)

Tri-O-acylglycerol: Tridecanoylglycerol

30 Weighing-in: 1200 mg

Solution 4f)

Tri-O-acylglycerol: Tridecanoylglycerol

Weighing-in: 1500 mg

35 Table 11: Overview results for the preparation of crystal suspensions of paclitaxel with solutions containing different amounts of tri-O-acylglycerol (+++++ = excellent stable crystal suspension, uniform distribution of microcrystals, intact microcrystals, "floating" of microcrystals; ++++ = stable crystal suspension, uniform distribution of microcrystals, intact microcrystals,

"floating" of microcrystals; +++ = crystal suspension; ++ = less stable crystal suspension; + = less stable crystal suspension, too viscous).

Solution (10 mL)	Suspension (100 h)
1a)	+++++
3a)	+++
3b)	+++
3c)	++++
3d)	+++
3e)	+++
3f)	++
1b)	+++++
4a)	+++
4b)	++++
4c)	++++
4d)	++++
4e)	+++
4f)	++
1a)	+++++
3a)	++
3b)	++++
3c)	+++++
3d)	+++
3e)	+++
3f)	++
1b)	+++++
4a)	+++
4b)	+++
4c)	++++
4d)	+++
4e)	+++
4f)	++

Results:

- 5 With different proportions of the tri-O-acylglycerols trioctanoylglycerol and tridecanoylglycerol, stable crystal suspensions could be prepared with microcrystalline paclitaxel. For the solutions with different proportions of trioctanoylglycerol and

tridecanoylglycerol, in particular, the crystal suspension with a ratio of 20:80 was still excellent stable after 100 h.

Example 6

5 Preparation of crystal suspensions containing microcrystalline paclitaxel in different solvent mixtures

Based on the results of the previous examples, different solvent mixtures for the preparation of crystal suspensions of microcrystalline paclitaxel were tested. The solvent mixture of ethyl acetate/heptane used in the previous examples has a ratio of
10 about 85:15 (heptane : ethyl acetate).

To investigate further solvent mixtures for the preparation of the crystal suspensions, solvent mixtures of the polar organic solvents acetone, ethanol, iso-propanol and ethyl acetate and the non-polar organic solvents hexane, heptane and cyclohexane were
15 prepared in different proportions.

To prepare the solutions, 770 mg of trioctanoylglycerol was dissolved in the polar solvent. Then the nonpolar solvent was added. It was then homogenized and filtered. The total volume of the solvent mixture is 100 mL in each case.

20

To test whether crystal suspension could be prepared, 10 mL of each of the solutions was carefully added to 200 mg of paclitaxel in the form of microcrystals at room temperature. After combining, it was tested whether stable crystal suspensions were obtained.

25

Table 12: Overview of results for the preparation of crystal suspensions of paclitaxel with various solvent mixtures (+++ = good; +/- = average, --- = poor)

Polar organic solvent	Non-polar organic solvent	Ratio polar/non-polar solvent (v/v)	Crystal suspension
Ethyl acetate	Heptane	10:90	+++
Ethyl acetate	Heptane	30:70	+++
Ethyl acetate	Heptane	40:60	+++
Ethyl acetate	Heptane	50:50	++
Ethyl acetate	Heptane	60:40	+/-
Ethyl acetate	Hexane	10:90	+++
Ethyl acetate	Hexane	30:70	+++
Ethyl acetate	Hexane	40:60	+++
Ethyl acetate	Hexane	50:50	+

Ethyl acetate	Hexane	60:40	+/-
Ethyl acetate	Cyclohexane	10:90	+++
Ethyl acetate	Cyclohexane	30:70	+++
Ethyl acetate	Cyclohexane	40:60	+++
Ethyl acetate	Cyclohexane	50:50	+
Ethyl acetate	Cyclohexane	60:40	+/-
Ethanol	Heptane	10:90	+++
Ethanol	Heptane	30:70	++
Ethanol	Heptane	40:60	++
Ethanol	Heptane	50:50	-
Ethanol	Heptane	60:40	---
Ethanol	Hexane	10:90	+++
Ethanol	Hexane	30:70	++
Ethanol	Hexane	40:60	+
Ethanol	Hexane	50:50	-
Ethanol	Hexane	60:40	---
Ethanol	Cyclohexane	10:90	+++
Ethanol	Cyclohexane	30:70	++
Ethanol	Cyclohexane	40:60	+
Ethanol	Cyclohexane	50:50	-
Ethanol	Cyclohexane	60:40	---
Acetone	Heptane	10:90	+++
Acetone	Heptane	30:70	++
Acetone	Heptane	40:60	+
Acetone	Heptane	50:50	-
Acetone	Heptane	60:40	---
Acetone	Hexane	10:90	+++
Acetone	Hexane	30:70	++
Acetone	Hexane	40:60	+
Acetone	Hexane	50:50	---
Acetone	Hexane	60:40	---
Acetone	Cyclohexane	10:90	+++
Acetone	Cyclohexane	30:70	++
Acetone	Cyclohexane	40:60	+
Acetone	Cyclohexane	50:50	---
Acetone	Cyclohexane	60:40	---
iso-propanol	Heptane	10:90	+++
iso-propanol	Heptane	30:70	++

iso-propanol	Heptane	40:60	++
iso-propanol	Heptane	50:50	---
iso-propanol	Heptane	60:40	---
iso-propanol	Hexane	10:90	+++
iso-propanol	Hexane	30:70	++
iso-propanol	Hexane	40:60	++
iso-propanol	Hexane	50:50	---
iso-propanol	Hexane	60:40	---
iso-propanol	Cyclohexane	10:90	+++
iso-propanol	Cyclohexane	30:70	++
iso-propanol	Cyclohexane	40:60	++
iso-propanol	Cyclohexane	50:50	---
iso-propanol	Cyclohexane	60:40	---

Stable crystal suspensions could be prepared with microcrystalline paclitaxel using various solvent mixtures. It has been shown that a content of at least 50% by volume of nonpolar solvent leads to very stable crystal suspensions.

5

Example 7

Coatings of balloon catheters with crystal suspensions containing trioctanoylglycerol of microcrystalline paclitaxel

- 10 Balloon catheters 4x40mm were coated with a 2 % PTX suspension containing trioctanoylglycerol (20wt% based on PTX) using a droplet dosing technique in a microdosing method such as the pipetting method, or droplet dragging method. It was possible to produce a uniform coating throughout with an equally uniform concentration of active agent on the balloon surface, where the crystals are evenly distributed. The
- 15 studies on the recovery rate of active agent on balloon catheters divided into equally sized segments confirm the uniformity of the coating and thus the success in using a crystal suspension as well as a 100% recovery rate

Example 8

Coatings of balloon catheters with crystal suspensions containing triacylglycerols not according the invention and microcrystalline paclitaxel

- In Example 2, the microcrystalline paclitaxel was found not to dissolve completely in the solutions containing tridodecanoylglycerol or tritetradecanoylglycerol. The
- 25 suspensions containing tridodecanoylglycerol or tritetradecanoylglycerol were found to

be unstable. To investigate the stability, flexibility and adhesion of coatings of balloon catheters with of microcrystalline paclitaxel, suspensions of microcrystalline paclitaxel containing tridodecanoylglycerol or tritetradecanoylglycerol (20wt% based on PTX) not according to the invention were freshly prepared and used directly for coating.

5

Balloon catheters 4x40mm were each coated with a 2% PTX suspension containing tridodecanoylglycerol or tritetradecanoylglycerol (20wt% based on PTX) using a drop dosing technique in a micro dosing method such as the pipetting method or drop dragging method. It was found that the coating was not applied sufficiently uniformly with these suspensions.

10

Subsequently, these coated balloon catheters were struck against an edge of a suitable object over a black pad (edge impact test). The particles collected on the pad were then determined microscopically and the size distribution of the detached coating was determined. The inflated balloon was then immersed in PBS solution to allow any remaining particles still loosely adhering to also fall off and be included in the evaluation. In a further investigation, additional coated balloon catheters were inflated as prescribed over a black pad and bent in different directions (bending test). Particles collected on the pad were then determined microscopically and the size distribution of the detached coating was determined. The inflated balloon was then immersed in PBS solution so that any remaining loosely adhering particles could also fall off and be included in the evaluation.

15

20

Comparison of the results of the edge impact test and bending test of the coatings according to the invention and the coatings not according to the invention

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The edge impact test and the bending test clearly showed that the total particle loss as well as the particle size distribution/balloon surface area for the coating with trioctanoylglycerol according to the invention is far below that of the coatings with tridodecanoylglycerol or tritetradecanoylglycerol not according to the invention. The particle release for the coating according to the invention with microcrystals of paclitaxel and trioctanoylglycerol shows that almost no particles are detached. The balloon catheters coated according to the invention have a determined particle count far below the prior art.

30

35 **Example 9**

Particle release ("crumble test"), determination of the loss of active agent or coating during implantation using an in vitro model, pre-wetting of implant surfaces, determination of a uniform coating

1. Particle release ("crumble test")

As a check on the mechanical adhesion of a coating on a surface, particle release is measured ("crumble test"), wherein it is determined, how many particles and of what size are released from the surface and thus lost when the coated medical device is impacted on edges and are bent (during and after inflation of the balloon). For this purpose, the coated implants are subjected to up to three mechanical tests. The weighted coated implant is weighed before and after testing.

a) Edge impact test (Fig. 2b)

The coated balloon catheter is lightly struck against a hard (sharp) edge of a suitable object over a black pad. Particles collected on the pad are then determined microscopically and the size distribution of the detached coating is determined. The inflated balloon is then immersed in PBS solution. This causes any remaining loosely adhering particles to also fall off and can be included in the evaluation.

b) Bending test (Fig. 2a)

The coated balloon catheter is inflated as prescribed and bent by hand in various directions over a black pad. Particles collected on the pad are then determined microscopically and the size distribution of the detached coating is determined. The inflated balloon is then immersed in PBS solution. This causes any remaining loosely adhering particles to also fall off and can be included in the evaluation.

c) Adhesion test

For this purpose, especially in the case of longer balloon catheters, e.g. peripheral balloons with a length of 150 mm are deflated as well as inflated, wrapped around a round vessel (e.g. test tube, standing cylinder or similar) with a suitable diameter and checked whether, on the one hand, no crumbs are formed and, on the other hand, it is checked whether the coating detaches from the balloon catheter and adheres to the surface of the vessel or not. Bend around a smooth object (preferably a glass laboratory vessel that fits from the circumference so that the catheter can be bent sufficiently and check if and if how much falls on the black pad. The hydrolysis tubes have a diameter of 12.8 mm. The balloon is bent around the hydrolysis tube in such a way that it is in contact with the wall. Smeary abrasion on the glass is tolerable, crumbling mass is not tolerable.

The "crumble tests" clearly show that the total particle loss as well as the particle size distribution/balloon surface area of all samples is far below the FDA guidelines. With trioctanoylglycerol, it can be clearly seen in Table 13 that regardless of loading with 1 μ g/mm² PTX microcrystals or 3 μ g/mm² PTX microcrystals, it releases significantly fewer particles and is well below the values obtained for commercially available and thus approved coated balloon catheters. Particle release in all measured particle size ranges for the coating with microcrystals of paclitaxel and trioctanoylglycerol according to the invention (here on different BMT catheter balloons) shows that virtually no particles can be detached. All balloon catheters coated according to the invention have a determined particle count far below the state of the art.

Thus, it can be seen that the flexibility and stability of the crystal coatings of microcrystals of paclitaxel -coated balloons according to the invention are far above the approved standard and state of the art.

15

Tab. 13: Number of particle losses/mm² balloon surface as a function of particle size with different drug loading with 1 μ g PTX/mm² and 3 μ g PTX/mm² with trioctanoylglycerol (*HTQ = Hemoteg, trioctanoylglycerol 20wt.% with respect to PTX) and comparison with state of the art catheter balloon

20

Released Particles/m² Balloon surface	HTQ* Trioctanoylglycerol	HTQ Trioctanoylglycerol	HTQ Trioctanoylglycerol	HTQ Trioctanoylglycerol	HTQ Trioctanoylglycerol	Released Particles/mm² Balloon surface
Balloon size [mm]	6.0x40	6.0x40	6.0x40	6.0x40	5.0x40	2.0x40
Loading PTX [μ g]	3	3	1	1	1	1
$\geq 10\mu$m	585.9	34.4	12.9	15.8	13.2	57.3
$\geq 25\mu$m	65.7	4.6	1.7	2.2	2.1	6.2
$\geq 65\mu$m	3.3	0.5	0.2	0.2	0.3	0.4
$\geq 100\mu$m	0.6	0.1	0.1	0.1	0.1	0.2

The crystal coating with trioctanoylglycerol/PTX during and after inflation shows a uniform coating with a uniform surface structure when inspected visually. The coating does not crumble off during inflation of the balloon.

25

In addition to excellent flexibility and virtually lossless adhesion, the crystal coating according to the invention also exhibits the required and necessary temperature stability, sterilizability (ETO sterilization is preferred) and shelf life.

2. Determination of the loss of active agent or coating during implantation using an in vitro model

To simulate the natural often curved paths through the vessels that a balloon catheter must travel to the implantation site, a silicone tubing model is formed (see Fig. 1a and Fig. 1b) that mimics the natural course of the vessels in the organism.

The catheter is inserted into the silicone tube simulating the artery and inflated. The silicone tube was previously filled with a defined volume of pyrogen-free water. After 60 sec, the balloon is deflated (pull vacuum) and carefully pulled out. Care is taken to ensure that the liquid from the tube is completely collected in a container. Subsequently, it is rinsed with a defined amount of water and also collected. Particle analysis (particle size distribution and quantification) is performed via LPC (Liquid Particle Counter).

Tab. 14: Particle release of different coatings with paclitaxel crystal suspensions containing 20% trioctanoylglycerol relative to PTX during inflation in PBS after in vitro determination by LPC and comparison with prior art catheter balloons

Sample	Total particle count			
	≥10 µm	≥25 µm	≥65 µm	≥100 µm
Paseo Lux 5.0x40	5917	946	87	0
Paseo Lux 5.0x40	7435	909	54	0
Ranger 2x80	6616	937	200	98
Eucatech 5.0x40, 1µg, 20% Monoctanoylglycerol	22123	4496	1341	784
Qualimed 6.0x40, 1µg, 20% Trioctanoylglycerol	5204	633	54	0
6.0x40, 3µg, 20% Trioctanoylglycerol	754	59	24	0
6.0x40, 1µg, 20% Trioctanoylglycerol	78	28	19	0
5.0x40, 1µg, 20% Trioctanoylglycerol	251	12	26	0
2.0x40, 1µg, 20% Trioctanoylglycerol	136	39	21	0

The optical comparison between Medtronic's In.Pact Pacific, BBraun's Paccocath and the balloon catheter according to the invention coated with trioctanoylglycerol/PTX ($1\mu\text{m}/\text{mm}^2$) clearly shows the differences in flexibility and surface adhesion after inflation (see Figs. 3 - 5). Thus, inflation alone reveals the following deficiencies:

5

Medtronic's In.Pact Pacific clearly shows a lack of uniform coating, the surface is also uneven, and the coating crumbles off during inflation (Fig. 3).

10

Upon inflation of BBraun's Paccocath, large pieces of the coating flake off along the entire length of the catheter balloon (Fig.4).

15

In contrast, the coating with trioctanoylglycerol/PTX exhibits a uniform coating with even surface texture during and after inflation, and the coating does not flake off during inflation of the balloon (Fig. 5).

In addition to excellent flexibility and virtually lossless adhesion, the coating according to the invention also exhibits the required and necessary temperature stability, sterilizability (ETO sterilization is preferred) and shelf life.

20 **3. Determination of a uniform coating**

For this purpose, the coated weighed balloon is fixed and inflated. The balloon is then cut with a scalpel into pieces of as equal size as possible, e.g. a 40 mm long balloon into 4 pieces, a 120 mm long balloon can be divided into 6 pieces. First, cut the balloon in half lengthwise and measure the layer thickness with a micrometer. Then the balloon is divided and the pieces are weighed and also the layer thicknesses are measured with the micrometer.

25

The coatings are each dissolved in a defined amount of acetone and the amount of active agent is determined via HPLC. The results are compared with each other, taking into account the balloon section areas.

30

In all cases, the presence of trioctanoylglycerol, or tridecanoylglycerol has been found to give a particularly stable and flexible coating, with the paclitaxel also adhering very well in the form of crystals, which are only released during the contact time with the target site.

35

Example 10

Comparison with crystal coatings not according to the invention

Crystal coatings of paclitaxel microcrystals and "solvent bonding"

The extent to which paclitaxel-crystals can be applied as a dry substance (powder) to balloon catheters was tested. Among other things, the adhesion of the crystals was evaluated. PTA catheters with a balloon size of 4.0x 60mm were used for the application of the coating.

First, the application of the pure crystal powder was performed. For this purpose, the powder is filled into a custom-made bowl and brought into contact with the balloon. In the process, the rotating balloon picks up crystals that adhere to the surface. A solvent was then carefully sprayed on to slightly dissolve the crystals so that they would adhere better to the balloon surface after subsequent drying. It could be shown that the microcrystals do not remain intact. The "crumble tests" clearly showed that the total particle loss and particle release is very high. The coating adheres poorly to the balloon surface.

Crystal coatings of paclitaxel microcrystals with base coat of commercial adhesive or trioctanoylglycerol solution and optional topcoat with trioctanoylglycerol solution

The extent to which paclitaxel-crystals can be applied as a dry substance (powder) to balloon catheters was tested. Among other things, the adhesion of the crystals was evaluated. PTA catheters with a balloon size of 4.0x 60mm were used for the application of the coating.

First, a base coat was applied to ensure the adhesion of the crystals to the balloon. To test the suitability of the experimental setup, a commercially available medical adhesive (Henkel) was initially used.

Base coating of commercially available adhesive (Uhu): The adhesive is applied thinly directly from the tube and evenly distributed on the rotating balloon. This is followed by the application of the crystals.

Furthermore, trioctanoylglycerol was used as a base coating. The base coating is applied by pipetting on the rotating catheter. After approx. 10min drying of the base coating, the pure crystal powder is applied. For this purpose, the powder is filled into a custom-made dish and brought into contact with the balloon. In the process, the rotating balloon picks up crystals that adhere to the surface.

Composition of the solution:

84.4% n-Heptane (volume %)

15.6% Ethyl acetate (volume %)

0.05% butylated hydroxytoluene (mass/volume)

200µl of trioctanoylglycerol is dissolved in 2ml of the solution. 2 x 50µl of the base coating solution are applied.

5

When the trioctanoylglycerol was used, it could be shown that the crystals adhered to the balloon surface, but the adhesion of the crystals to each other was not sufficient.

10 In a third coating step, a topcoat with trioctanoylglycerol was therefore applied with a pipette to increase adhesion. The adhesion was evaluated by means of a "bend test". The bend test is a method in which the coated balloon is bent 2x around a glass tube of about 14mm diameter. If many and/or larger fragments detach from the coating during this process, the adhesion is rated as insufficient.

15 Composition of the topcoat solution:

84.4% n-Heptane (volume %)

15.6% Ethyl acetate (volume %)

0.25% trioctanoylglycerol (mass/volume)

0.05% butylated hydroxytoluene (mass/volume)

20

2 x 30µl top coat-solution was pipetted repectively.

25 When the topcoat solution was used, the crystals were found to adhere better to the balloon surface, and the adhesion of the crystals to each other was also improved, so that less particle release occurred in the edge impact test. However, the bending test and inflation of the balloon showed that the adhesion of the crystals to each other was not sufficient.

30

35

Claims

- 5 1. A suspension for coating of a medical device selected from a catheter balloon, a balloon catheter, a stent, or a cannula, the suspension containing:
 - a) at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol,
 - 10 b) at least one taxane in the form of microcrystals, and
 - c) a solvent or a solvent mixture in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the at least one taxane do not dissolve.
- 15 2. The suspension according to claim 1, wherein the at least one taxane is selected from the group comprising or consisting paclitaxel, docetaxel, and cabazitaxel.
3. The suspension according to claim 1 or 2, wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% - 30%
20 tri-*O*-acylglycerol to 90% - 70% taxane.
4. The suspension according to any one of claims 1 to 3, wherein the at least one taxane is in the form of microcrystals having a crystal size in the range of 1 μm to 300 μm .
- 25 5. The suspension according to any one of claims 1 to 4, wherein at least 70% of the at least one taxane is in the form of microcrystals having a crystal size ranging from 10 μm to 100 μm .
- 30 6. The suspension according to any one of claims 1 to 5, wherein the at least one taxane has a crystallinity of at least 80% by weight.
7. The suspension according to any one of claims 1 to 6, wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture
35 contains at least 50% by volume of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .

8. The suspension according to any one of claims 1 to 6, wherein the solvent mixture is a mixture of at least one polar organic solvent having an n-octanol-water partition coefficient $\log K_{ow}$ of -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of 5.0 to 30, and at least one nonpolar organic solvent having a dielectric constant ϵ_r at 20°C of ≤ 3.0 and an n-octanol-water partition coefficient $\log K_{ow}$ of ≥ 3.0 .
9. A method of preparing the suspension according to any one of claims 1 to 8 comprising the following steps:
- a) dissolving at least one tri-O-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol in a solvent or a solvent mixture;
- b) preparing a suspension of at least one taxane in the form of microcrystals and the solution from step a), wherein the microcrystals of the at least one taxane do not dissolve in the solution of step a).
10. The method according to claim 9, wherein the at least one taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel.
11. The method according to claim 9 or 10, wherein the at least one taxane is in the form of microcrystals having a crystal size in the range of 1 μm to 300 μm .
12. The method according to any one of claims 9 to 11, wherein at least 70% of the at least one taxane is in the form of microcrystals having a crystal size in the range of 10 μm to 100 μm .
13. The method according to any one of claims 9 to 12, wherein the at least one taxane has a crystallinity of at least 80% by weight.
14. The method according to any one of claims 9 to 13, wherein the solvent is a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 or the solvent mixture contains at least 50% by volume of a non-solvent having a dielectric constant ϵ_r at 20°C of ≤ 2.0 .
15. The method according to any one of claims 9 to 14, wherein the solvent mixture is a mixture of at least one polar organic solvent having an n-octanol-water partition coefficient $\log K_{ow}$ of -0.5 to +1.5 and a dielectric constant ϵ_r at 20°C of 5.0 to 30, and at least one nonpolar organic solvent having a dielectric constant ϵ_r at 20°C of ≤ 3.0 and an n-octanol-water partition coefficient $\log K_{ow}$ of ≥ 3.0 .

16. A method for coating of a medical device selected from a catheter balloon, a balloon catheter, a stent, or a cannula, comprising the following steps:
- a) providing the medical device with a medical device surface,
 - 5 b) providing a suspension containing at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, at least one taxane in the form of microcrystals, and a solvent or a solvent mixture, in which the at least one tri-*O*-acylglycerol dissolves and in which the microcrystals of the
 - 10 c) applying the coating suspension to the surface of the medical device by means of a syringe method, pipetting method, capillary method, fold spraying method, dipping method, spraying method, dragging method, thread dragging method, drop dragging method, or rolling method.
- 15 17. The method according to claim 16, further comprising the following step d) drying the coating.
18. A medical device selected from a catheter balloon, a balloon catheter, a stent, or
- 20 a cannula, obtainable by the method according to claim 16 or 17.
19. A medical device selected from a catheter balloon, a balloon catheter, a stent, or a cannula, coated with the suspension according to any one of claims 1 to 8 and subsequent drying of the coating.
- 25 20. A medical device selected from from a catheter balloon, a balloon catheter, a stent, or a cannula coated with at least one tri-*O*-acylglycerol selected from the group consisting of trioctanoylglycerol, trinonanoylglycerol, tridecanoylglycerol, and triundecanoylglycerol, and at least one taxane in the form of microcrystals.
- 30 21. The medical device of claim 20, wherein the taxane is selected from the group comprising or consisting of paclitaxel, docetaxel, and cabazitaxel.
22. The medical device of claim 20 or 21, wherein the at least one tri-*O*-acylglycerol and the at least one taxane are present in a mass ratio of 10% – 30% tri-*O*-acylglycerol to 90% – 70% taxane.
- 35

23. The medical device according to any one of claims 20 to 22, wherein the at least one taxane is in the form of microcrystals having a crystal size in the range of 1 μm to 300 μm .
- 5 24. The medical device according to any one of claims 20 to 23, wherein at least 70% of the at least one taxane is in the form of microcrystals having a crystal size in the range of 10 μm to 100 μm .
- 10 25. The medical device according to any one of claims 20 to 24, wherein the at least one taxane has a crystallinity of at least 80% by weight.
- 15 26. The medical device according to any one of claims 20 to 25, wherein a biostable or biodegradable, bioactive or bioinert polymeric, metallic or ceramic layer is present beneath the layer of the at least one tri-O-acylglycerol and the at least one taxane.