

## DESCRIPTION

## GARNET SCINTILLATOR CO-DOPED WITH MONOVALENT ION

## RELATED APPLICATIONS

The presently disclosed subject matter claims priority to and the benefit of U.S. Provisional Patent Application Serial No. 62/627,399, filed February 7, 2018 and U.S. Provisional Patent Application Serial No. 62/672,890, filed May 17, 2018; the disclosure of each of which is  
5 incorporated herein by reference in its entirety.

## TECHNICAL FIELD

The presently disclosed subject matter relates to garnet-type oxide  
10 scintillators codoped with monovalent cations, their optical (e.g., scintillation and phosphorescence) properties, and their use as scintillation materials in radiation detectors and/or in methods of detecting, for example, X-rays, gamma rays and/or neutrons. In particular, the presently disclosed subject matter relates to garnet-type oxide materials such as lutetium yttrium  
15 aluminum garnet (LuYAG)-type materials and other rare earth aluminum garnet materials, that are doped with a dopant ion, such as  $\text{Pr}^{3+}$  or another activator, and codoped with at least one type of monovalent alkali metal cation, such as  $\text{Li}^+$ .

20

## ABBREVIATIONS

|    |                |   |                       |
|----|----------------|---|-----------------------|
|    | %              | = | percentage            |
|    | °C             | = | degrees Celsius       |
|    | $\mu\text{Ci}$ | = | microcurie            |
|    | $\mu\text{s}$  | = | microseconds          |
| 25 | $\tau_d$       | = | decay time            |
|    | Al             | = | aluminum              |
|    | at             | = | atomic                |
|    | a.u.           | = | arbitrary unit        |
|    | CCD            | = | charge-coupled device |
| 30 | Ce             | = | cerium                |

|    |       |   |                                               |
|----|-------|---|-----------------------------------------------|
|    | Cs    | = | cesium                                        |
|    | CT    | = | computed tomography                           |
|    | ER    | = | energy resolution                             |
|    | Eu    | = | europium                                      |
| 5  | g     | = | grams                                         |
|    | Gd    | = | gadolinium                                    |
|    | K     | = | potassium                                     |
|    | keV   | = | kiloelectronvolts                             |
|    | La    | = | lanthanum                                     |
| 10 | Li    | = | lithium                                       |
|    | LO    | = | light output                                  |
|    | Lu    | = | lutetium                                      |
|    | LuAG  | = | lutetium aluminum garnet                      |
|    | LuYAG | = | lutetium yttrium aluminum garnet              |
| 15 | LY    | = | light yield                                   |
|    | MeV   | = | megaelectronvolt                              |
|    | mm    | = | millimeter                                    |
|    | mol   | = | mole                                          |
|    | MPa   | = | megapascals                                   |
| 20 | Na    | = | sodium                                        |
|    | nm    | = | nanometer                                     |
|    | ns    | = | nanoseconds                                   |
|    | ph    | = | photons                                       |
|    | PL    | = | photoluminescence                             |
| 25 | ppm   | = | parts per million                             |
|    | PMT   | = | photomultiplier tube                          |
|    | Pr    | = | praseodymium                                  |
|    | Rb    | = | rubidium                                      |
|    | RL    | = | radioluminescence                             |
| 30 | RT    | = | room temperature                              |
|    | Sc    | = | scandium                                      |
|    | SPECT | = | single photon emission computed<br>tomography |

|    |   |                    |
|----|---|--------------------|
| Tb | = | terbium            |
| TL | = | thermoluminescence |
| wt | = | weight             |
| Y  | = | yttrium            |

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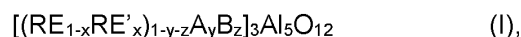
## BACKGROUND

In industrial applications for radiation detection, such as medical imaging and national security, desirable characteristics for scintillators include high light yield, good energy resolution, and fast scintillation decay time. Recently, with the development of cerium doped garnet scintillators such as lutetium aluminum garnet (LuAG) and gadolinium gallium aluminum garnet (GGAG) scintillators, researchers have reported light yields as high as 46,000 photons per megaelectronvolt (ph/MeV) and decay times reaching as fast as 30 nanoseconds (ns). However, LuAG scintillators often tend to not reach their full potential when it comes to scintillation due to intrinsic defects, such as charge carrier traps formed from the  $\text{Lu}_{\text{Al}}$  anti-site defects and oxygen vacancies.

Accordingly, there is an ongoing need for additional garnet-type scintillator materials, such as those with higher light yield, improved energy resolution, and/or more rapid scintillation decay time. There is also an ongoing need for additional methods of altering the properties of garnet-type scintillator materials.

## SUMMARY

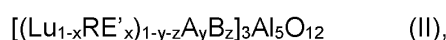
In some embodiments, the presently disclosed subject matter provides a scintillator material. In some embodiments, the scintillator material comprises a composition of Formula (I):



wherein:  $0 \leq x \leq 1.0$ ;  $0 < y \leq 0.05$ ;  $0 < z \leq 0.1$ ; RE is a first rare earth element; RE' is a second rare earth element or a combination of a second rare earth element and at least one more or more additional rare earth elements, subject to the proviso that RE' does not comprise the first rare earth element RE; A is a dopant ion selected from the group comprising Pr, Nd, Sm, Eu,

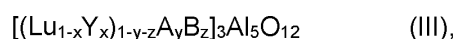
Gd, Tb, Yb, Bi, Sb, and any combination thereof, subject to the proviso that A does not comprise an ion of the same element as RE or RE'; and B is at least one type of monovalent cation, optionally a cation of an element selected from the group comprising Li, Na, K, Rb, Cs, and Fr. In some  
 5 embodiments, RE is Lu. In some embodiments, RE' is Y. In some embodiments,  $0.1 \leq x \leq 0.5$ , optionally wherein x is 0.25. In some embodiments, A is  $\text{Pr}^{3+}$ . In some embodiments,  $0.0001 \leq y \leq 0.015$ , optionally wherein y is 0.004 or 0.012. In some embodiments, B is  $\text{Li}^+$ . In some embodiments,  $0.0002 \leq z \leq 0.1$ , optionally wherein  $0.001 \leq z \leq 0.1$ , further  
 10 optionally wherein z is 0.002, 0.008, or 0.02.

In some embodiments, provided is a scintillator material comprising a composition of Formula (II):



wherein:  $0.05 \leq x \leq 1.0$ ;  $0 < y \leq 0.05$ ;  $0 < z \leq 0.1$ ; RE' is selected from the group  
 15 comprising Ce, Dy, Er, Eu, Gd, Ho, La, Nd, Pr, Pm, Sm, Sc, Tb, Tm, Yb, Y and any combination thereof; A is a dopant ion, optionally an ion of an element selected from the group comprising Ce, Pr, Nd, Sm, Eu, Gd, Tb, Yb, Bi, Sb, and any combination thereof, subject to the proviso that A does not comprise an ion of the same element as RE'; and B is at least one type of  
 20 monovalent cation, optionally a cation of an element selected from the group consisting of Li, Na, K, Rb, Cs, and Fr. In some embodiments, RE' is Y. In some embodiments,  $0.1 \leq x \leq 0.5$ , optionally wherein x is 0.25. In some embodiments, A is  $\text{Pr}^{3+}$ . In some embodiments,  $0.0001 \leq y \leq 0.015$ , optionally wherein y is 0.004 or 0.012. In some embodiments, B is  $\text{Li}^+$ . In  
 25 some embodiments,  $0.0002 \leq z \leq 0.1$ , optionally wherein  $0.001 \leq z \leq 0.1$ , further optionally wherein z is 0.002, 0.008, or 0.02.

In some embodiments, provided is a scintillator material comprising a composition of Formula (III):



30 wherein:  $0.05 \leq x \leq 0.5$ ;  $0 < y \leq 0.05$ ;  $0 < z \leq 0.1$ ; A is a dopant ion, optionally an ion of an element selected from the group comprising Ce, Pr, Nd, Sm, Eu, Gd, Tb, Yb, Bi, Sb, and any combination thereof; and B is at least one type

of monovalent cation, optionally a cation of an element selected from the group comprising Li, Na, K, Rb, Cs, and Fr. In some embodiments, x is 0.25. In some embodiments, A is  $\text{Pr}^{3+}$ . In some embodiments,  $0.0001 \leq y \leq 0.015$ , optionally wherein y is 0.004 or 0.012. In some  
 5 embodiments, B is  $\text{Li}^+$ . In some embodiments,  $0.0002 \leq z \leq 0.1$ , optionally wherein  $0.001 \leq z \leq 0.1$ , further optionally wherein z is 0.002, 0.008, or 0.02.

In some embodiments, the scintillator material comprises a material selected from the group comprising  $(\text{Lu}_{0.75}, \text{Y}_{0.25})_3\text{Al}_5\text{O}_{12}:0.4\%\text{Pr}^{3+}$  codoped with 0.2%  $\text{Li}^+$ ,  $(\text{Lu}_{0.75}, \text{Y}_{0.25})_3\text{Al}_5\text{O}_{12}:0.4\%\text{Pr}^{3+}$  codoped with 0.8%  $\text{Li}^+$ ,  
 10  $(\text{Lu}_{0.75}, \text{Y}_{0.25})_3\text{Al}_5\text{O}_{12}:0.4\%\text{Pr}^{3+}$  codoped with 2.0%  $\text{Li}^+$ ;  $(\text{Lu}_{0.75}, \text{Y}_{0.25})_3\text{Al}_5\text{O}_{12}:1.2\%\text{Pr}^{3+}$  codoped with 0.308 at%  $\text{Na}^+$ ,  $(\text{Lu}_{0.75}, \text{Y}_{0.25})_3\text{Al}_5\text{O}_{12}:1.2\%\text{Pr}^{3+}$  codoped with 0.6 at%  $\text{Li}^+$ ,  $(\text{Lu}_{0.75}, \text{Y}_{0.25})_3\text{Al}_5\text{O}_{12}:1.2\%\text{Pr}^{3+}$  codoped with 0.6 at%  $\text{K}^+$ ,  $(\text{Lu}_{0.75}, \text{Y}_{0.25})_3\text{Al}_5\text{O}_{12}:1.2\%\text{Pr}^{3+}$  codoped with 2.4 at%  $\text{Li}^+$  and  
 15  $(\text{Lu}_{0.75}, \text{Y}_{0.25})_3\text{Al}_5\text{O}_{12}:1.2\%\text{Pr}^{3+}$  codoped with 6 at%  $\text{Li}^+$ , optionally wherein the material is selected from the group comprising  $(\text{Lu}_{0.75}, \text{Y}_{0.25})_3\text{Al}_5\text{O}_{12}:0.4\%\text{Pr}^{3+}$  codoped with 0.2%  $\text{Li}^+$ ,  $(\text{Lu}_{0.75}, \text{Y}_{0.25})_3\text{Al}_5\text{O}_{12}:0.4\%\text{Pr}^{3+}$  codoped with 0.8%  $\text{Li}^+$ , and  $(\text{Lu}_{0.75}, \text{Y}_{0.25})_3\text{Al}_5\text{O}_{12}:0.4\%\text{Pr}^{3+}$  codoped with 2.0%  $\text{Li}^+$ .

In some embodiments, the scintillator material exhibits one or more  
 20 of increased light yield, improved energy resolution, and an accelerated fast decay component as compared to the scintillator material where B is absent. In some embodiments, the scintillator material is a single crystal material. In some embodiments, the scintillator material is a polycrystalline and/or a ceramic material.

25 In some embodiments, provided is a radiation detector comprising a photon detector and a scintillation material in accordance with the presently disclosed subject matter. In some embodiments, the detector is a medical diagnostic device, a device for oil exploration, or a device for container or baggage scanning.

30 In some embodiments, provided is a method of detecting gamma rays, X-rays, cosmic rays and/or particles having an energy of 1 keV or greater., the method comprising using the radiation detector comprising a

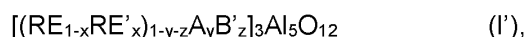
photon detector and a scintillation material in accordance with the presently disclosed subject matter.

In some embodiments, provided is a method of preparing a scintillator material in accordance with the presently disclosed subject matter, wherein  
5 the method comprises pulling a single crystal from molten raw materials.

In some embodiments, the presently disclosed subject matter provides a method of altering one or more scintillation and/or optical properties of a rare earth aluminum garnet scintillator comprising a matrix having the formula  $RE''_3Al_5O_{12}$ , wherein  $RE''$  is a mixture of at least two rare  
10 earth elements and wherein the scintillator further comprises at least one dopant selected from the group consisting of Ce, Pr, Nd, Sm, Eu, Gd, Tb, Yb, Bi, Sb, and any combination thereof, subject to the proviso that the dopant does not comprise an ion of the same element as any rare earth element of the rare earth aluminum garnet matrix, wherein the method  
15 comprises preparing the rare earth aluminum garnet scintillator in the presence of a monovalent codopant ion, thereby providing a codoped rare earth aluminum garnet scintillator material, optionally wherein the monovalent codopant ion is an alkali metal ion. In some embodiments, the presently disclosed subject matter provides a method of altering one or more  
20 scintillation and/or optical properties of a rare earth aluminum garnet scintillator comprising a matrix having the formula  $RE''_3Al_5O_{12}$ , wherein  $RE''$  is a mixture of at least two rare earth elements and wherein the scintillator further comprises at least one dopant selected from the group comprising Pr, Nd, Sm, Eu, Gd, Tb, Yb, Bi, Sb, and any combination thereof, subject to the  
25 proviso that the dopant does not comprise an ion of the same element as any rare earth element of the rare earth aluminum garnet matrix, optionally wherein the rare earth aluminum garnet scintillator is a praseodymium (Pr) doped lutetium yttrium aluminum garnet (LuYAG) scintillator, wherein the method comprises preparing the rare earth aluminum garnet scintillator in  
30 the presence of a monovalent codopant ion, thereby providing a codoped rare earth aluminum garnet scintillator material, optionally wherein the monovalent codopant ion is an alkali metal ion, further optionally wherein the monovalent codopant ion is  $Li^+$ . In some embodiments, the codoped rare

earth aluminum garnet scintillator material exhibits increased light yield, improved energy resolution, better proportionality, and/or faster decay time compared to the same rare earth aluminum garnet scintillator material without the codopant ion.

5 In some embodiments, provided is a scintillator material comprising a composition of Formula (I'):



wherein:  $0 \leq x \leq 1.0$ ;  $0 < y \leq 0.05$ ;  $0 < z \leq 0.1$ ; RE is a first rare earth element; RE' is a second rare earth element or a combination of a second rare earth element and at least one more or more additional rare earth elements, subject to the proviso that RE' does not comprise the first rare earth element RE; A is a dopant ion selected from the group comprising Pr, Nd, Sm, Eu, Gd, Tb, Yb, Bi, Sb, and any combination thereof, subject to the proviso that A does not comprise an ion of the same element as RE or RE'; and B' is a monovalent cation of a lithium isotope or a mixture thereof, optionally wherein B' is a monovalent cation of a lithium-6 isotope (i.e.,  ${}^6Li^+$ ). In some embodiments,  $0.0001 \leq z \leq 0.1$ , optionally  $0.001 \leq z \leq 0.1$ . In some embodiments, RE is Lu. In some embodiments, RE' is Y, optionally wherein x is about 0.25. In some embodiments, A is  $Pr^{3+}$ , optionally wherein  $0.0001 \leq y \leq 0.015$ , further optionally wherein y is 0.004 or 0.012.

In some embodiments, provided is a radiation detector comprising a photon detector and a scintillator material in accordance with the presently disclosed subject matter.

In some embodiments, provided is a method of detecting neutrons, wherein the method comprises using a radiation detector comprising a photon detector and a scintillation material in accordance with the presently disclosed subject matter.

Accordingly, it is an object of the presently disclosed subject matter to provide codoped rare earth aluminum garnet scintillators, radiation detectors comprising the codoped scintillators; methods of detecting gamma rays, X-rays, cosmic rays and/or particles having an energy of 1 keV or greater with the radiation detectors; methods of preparing the optical materials, and

methods of altering the scintillation and/or optical properties of the scintillators.

An object of the presently disclosed subject matter having been stated hereinabove, and which is achieved in whole or in part by the presently disclosed subject matter, other objects will become evident as the  
5 description proceeds hereinbelow.

#### BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 is a graph of the absorbance spectra (normalized intensity (in  
10 arbitrary units (a.u.)) versus wavelength (in nanometers (nm))) of a lutetium aluminum oxide single crystal where about 25 percent of the lutetium is replaced by yttrium and where the material is doped with 0.4 atomic percent (at%) praseodymium and codoped with 0.2 at% lithium (i.e.,  $\text{Lu}_{0.744}\text{Y}_{0.25}\text{Pr}_{0.004}\text{Li}_{0.002}\text{Al}_5\text{O}_{12}$ ).

Figure 2 is a graph of the X-ray excited luminescence (or  
15 radioluminescence (RL)) spectra (normalized intensity (in arbitrary units (a.u.)) versus wavelength (in nanometers (nm))) of a lutetium aluminum oxide single crystal where about 25 percent of the lutetium is replaced by yttrium and where the material is doped with 0.4 atomic percent (at%)  
20 praseodymium and codoped with 0.2 at% lithium (i.e.,  $\text{Lu}_{0.744}\text{Y}_{0.25}\text{Pr}_{0.004}\text{Li}_{0.002}\text{Al}_5\text{O}_{12}$ ).

Figure 3 is a graph of the gamma ( $\gamma$ )-ray spectra (intensity (in arbitrary  
units (a.u.)) versus channel number) of a lutetium aluminum oxide single  
25 crystal where about 25 percent of the lutetium is replaced by yttrium and where the material is doped with 0.4 atomic percent (at%) praseodymium and codoped with 0.2 at% lithium (i.e.,  $\text{Lu}_{0.744}\text{Y}_{0.25}\text{Pr}_{0.004}\text{Li}_{0.002}\text{Al}_5\text{O}_{12}$ ). Light yield is 16,000 photons per megaelectronvolt (MeV) based on a Gaussian fit of the data between about channel 250 and about channel 330. The x-ray  
30 source was 10 microcurie ( $\mu\text{Ci}$ ) of cesium-137 ( $^{137}\text{Cs}$ ; 662 kiloelectronvolts (keV)).

Figure 4 is a graph showing the scintillation time profile (normalized  
intensity (in arbitrary units (a.u.)) versus time (in nanoseconds (ns))) of a  
lutetium aluminum oxide single crystal where about 25 percent of the

lutetium is replaced by yttrium and where the material is doped with 0.4 atomic percent (at%) praseodymium and codoped with 0.2 at% lithium (i.e.,  $\text{Lu}_{0.744}\text{Y}_{0.25}\text{Pr}_{0.004}\text{Li}_{0.002}\text{Al}_5\text{O}_{12}$ ). Decay times for a three-component exponential decay fit were 22 nanoseconds (ns) (40 percent (%)), 656 ns (31%), and 119.1 ns (29%).

Figure 5 is a graph of the absorbance spectra of (normalized intensity (in arbitrary units (a.u.)) versus wavelength (in nanometers (nm))) of a lutetium aluminum oxide single crystal where about 25 percent of the lutetium is replaced by yttrium and where the material is doped with 0.4 atomic percent (at%) praseodymium (Pr) and codoped with lithium (Li). Spectra are shown for a crystal codoped with 0.2 at% Li (LuYAG:Pr, 0.2% Li, dashed line), a crystal codoped with 0.8 at% Li (LuYAG:Pr, 0.8% Li, dashed and dotted line), and a crystal codoped with 2.0 at% Li (LuYAG:Pr, 2.0% Li, heavy solid line). For comparison, the spectrum for a Pr doped material free of codopant (LuYAG:Pr, thin solid line) is also shown. In addition, the difference between the spectra for the codoped samples and the spectrum for the noncodoped sample is shown in the dashed and double dotted line. The inset shows an expanded view of the spectra between 220 nm and 300 nm.

Figure 6 is a graph showing the photoluminescence spectra (in intensity (in arbitrary units (a.u.)) versus wavelength (in nanometers (nm))) of a lutetium aluminum oxide single crystal where about 25 percent of the lutetium is replaced by yttrium and where the material is doped with 0.4 atomic percent (at%) praseodymium (Pr) and codoped with lithium (Li). Spectra are shown for a crystal codoped with 0.2 at% Li (LuYAG:Pr, 0.2% Li, dashed line), a crystal codoped with 0.8 at% Li (LuYAG:Pr, 0.8% Li, dotted line), and a crystal codoped with 2.0 at% Li (LuYAG:Pr, 2.0% Li, dashed and dotted line). For comparison, the spectra of a Pr doped material free of codopant (LuYAG:Pr, solid line) is also shown. Excitation is at 375 nm and the main emission peak at 280 nm.

Figure 7 is a graph showing the gamma ray spectra (in intensity (in arbitrary units (a.u.)) versus channel number for, from top to bottom, a lutetium aluminum oxide single crystal where about 25 percent of the

lutetium is replaced by yttrium and where the material is doped with 0.4 atomic percent (at%) praseodymium (Pr) (LuYAG:Pr); a crystal of the same material also codoped with 0.2 at% lithium (Li) (LuYAG:Pr, 0.2% Li); a crystal of the same material also codoped with 0.8 at% Li (LuYAG:Pr, 0.8% Li); and  
 5 a crystal of the same material also codoped with 2.0 at% Li (LuYAG:Pr, 20% Li). The spectra of each LuYAG crystal is shown as a solid line. For comparison, the spectra of bismuth germanate (BGO) is shown as a dashed line.

Figure 8 is a graph showing the gamma response (relative light yield  
 10 versus gamma energy (in kiloelectronvolts (keV)) for praseodymium (Pr) doped lutetium yttrium aluminum oxide (LuYAG: Pr, squares), Pr doped LuYAG codoped with 0.2 atomic percent (at%) lithium (LuYAG: Pr, 0.2% Li, circles); Pr doped LuYAG codoped with 0.2 at% lithium (LuYAG: Pr, 0.8% Li, triangles); and Pr doped LuYAG codoped with 2.0 at% lithium (LuYAG: Pr,  
 15 2.0% Li, stars). The ideal response would follow the solid line.

Figure 9 is a schematic drawing of an apparatus for detecting radiation according to an aspect of the presently disclosed subject matter. Apparatus **10** includes photon detector **12** optically coupled to scintillator material **14**. Apparatus **10** can optionally include electronics **16** for recording  
 20 and/or displaying electronic signal from photon detector **12**. Thus, optional electronics **16** can be in electronic communication with photon detector **12**.

Figure 10 is a pair of graphs showing the thermoluminescence glow curves (normalized intensity in arbitrary units (a.u.) versus temperature in Kelvin (K)) of as grown (top) and air annealed (bottom) 0.4 atomic percent  
 25 (at%) praseodymium (Pr) doped lutetium yttrium aluminum oxide (LuYAG: Pr, solid lines), 0.4 at % Pr doped LuYAG codoped with 0.2 at% lithium (LuYAG: Pr, 0.2% Li, dashed lines); 0.4 at% Pr doped LuYAG codoped with 0.8 at% lithium (LuYAG: Pr, 0.8% Li, dotted lines); and 0.4 at% Pr doped LuYAG codoped with 2.0 at% lithium (LuYAG: Pr, 2.0% Li, dashed and  
 30 dotted lines). The grey arrows emphasize a lowering in intensity of the peaks below about 250 K in the lithium codoped samples.

## DETAILED DESCRIPTION

The presently disclosed subject matter describes a method of tailoring the properties of garnet-type oxide scintillators to meet the particular needs of different applications. More particularly, in some embodiments, garnet-type oxide scintillators, such as lutetium yttrium aluminum garnet (LuYAG), with modified scintillation decay time, energy resolution, and/or light yield were prepared by codoping of at least one type of monovalent cation at a ratio of about 30,000 weight (wt) parts per million (ppm) or less with respect to all cations. These scintillators, when doped with activators such as praseodymium (Pr), are suitable for radiation detection applications such as medical imaging, homeland security, high energy physics experiments, and geophysical exploration. Codoping of these scintillators with monovalent ions can be used to modify both optical and scintillation properties.

The presently disclosed subject matter will now be described more fully. The presently disclosed subject matter can, however, be embodied in different forms and should not be construed as limited to the embodiments set forth herein below and in the accompanying Examples. Rather, these embodiments are provided so that this disclosure will be thorough and complete, and will fully convey the scope of the embodiments to those skilled in the art.

All references listed herein, including but not limited to all patents, patent applications and publications thereof, and scientific journal articles, are incorporated herein by reference in their entireties to the extent that they supplement, explain, provide a background for, or teach methodology, techniques, and/or compositions employed herein.

I. Definitions

While the following terms are believed to be well understood by one of ordinary skill in the art, the following definitions are set forth to facilitate explanation of the presently disclosed subject matter.

Unless defined otherwise, all technical and scientific terms used herein have the same meaning as commonly understood to one of ordinary skill in the art to which the presently disclosed subject matter belongs.

Following long-standing patent law convention, the terms “a”, “an”, and “the” refer to “one or more” when used in this application, including the claims.

5 The term “and/or” when used in describing two or more items or conditions, refers to situations where all named items or conditions are present or applicable, or to situations wherein only one (or less than all) of the items or conditions is present or applicable.

10 The use of the term “or” in the claims is used to mean “and/or” unless explicitly indicated to refer to alternatives only or the alternatives are mutually exclusive, although the disclosure supports a definition that refers to only alternatives and “and/or.” As used herein “another” can mean at least a second or more.

15 The term “comprising”, which is synonymous with “including,” “containing,” or “characterized by” is inclusive or open-ended and does not exclude additional, unrecited elements or method steps. “Comprising” is a term of art used in claim language which means that the named elements are essential, but other elements can be added and still form a construct within the scope of the claim.

20 As used herein, the phrase “consisting of” excludes any element, step, or ingredient not specified in the claim. When the phrase “consists of” appears in a clause of the body of a claim, rather than immediately following the preamble, it limits only the element set forth in that clause; other elements are not excluded from the claim as a whole.

25 As used herein, the phrase “consisting essentially of” limits the scope of a claim to the specified materials or steps, plus those that do not materially affect the basic and novel characteristic(s) of the claimed subject matter.

30 With respect to the terms “comprising”, “consisting of”, and “consisting essentially of”, where one of these three terms is used herein, the presently disclosed and claimed subject matter can include the use of either of the other two terms.

Unless otherwise indicated, all numbers expressing quantities of time, temperature, light output, atomic (at) or mole (mol) percentage (%), and so

forth used in the specification and claims are to be understood as being modified in all instances by the term “about”. Accordingly, unless indicated to the contrary, the numerical parameters set forth in this specification and attached claims are approximations that can vary depending upon the  
5 desired properties sought to be obtained by the presently disclosed subject matter.

As used herein, the term “about”, when referring to a value is meant to encompass variations of in one example  $\pm 20\%$  or  $\pm 10\%$ , in another example  $\pm 5\%$ , in another example  $\pm 1\%$ , and in still another example  $\pm 0.1\%$   
10 from the specified amount, as such variations are appropriate to perform the disclosed methods.

The term “scintillator” refers to a material that emits light (e.g., visible light) in response to stimulation by high energy radiation (e.g., X,  $\alpha$ ,  $\beta$ , or  $\gamma$  radiation).

15 The term “phosphor” as used herein refers to a material that emits light (e.g., visible light) in response to irradiation with electromagnetic or particle radiation.

In some embodiments, the compositional formula expression of an optical material (e.g., a scintillation material or a phosphor) can contain a  
20 colon “:”, wherein the composition of the main or base matrix material (e.g., the main rare earth aluminum garnet matrix) is indicated on the left side of the colon, and the activator (or dopant ion) or the activator and the codopant ion are indicated on the right side of the colon. In some embodiments, the dopant and codopant can replace part of the rare earth metal element(s) in a  
25 rare earth metal aluminum oxide garnet-type scintillator material. For example,  $\text{Lu}_3\text{Al}_5\text{O}_{12}:0.4\text{Pr},0.2\text{Li}$ ,  $\text{LuAG}:0.4\%\text{Pr},0.2\%\text{Li}$ , and  $\text{Lu}_3\text{Al}_5\text{O}_{12}:\text{Pr}^{3+} 0.4\%, \text{Li}^+ 0.2\%$  each represent a LuAG optical material activated by praseodymium and codoped with lithium, wherein 0.4 atomic % of the lutetium is replaced by praseodymium and 0.2 atomic % of the lutetium is  
30 replaced by lithium. Thus, in some embodiments, the atomic % of a dopant can be expressed as the atomic % relative to the total amount of dopant and rare earth metal(s) (or dopant, rare earth metal(s) and codopant) in the base material. The atomic % of the codopant ion can be expressed as the atomic

or mole % relative to the total amount of rare earth metal(s), dopant and codopant.

The term “high energy radiation” can refer to electromagnetic radiation having energy higher than that of ultraviolet radiation, including, but not limited to X radiation (i.e., X-ray radiation), alpha ( $\alpha$ ) particles, gamma ( $\gamma$ ) radiation, and beta ( $\beta$ ) radiation. In some embodiments, the high energy radiation refers to gamma rays, cosmic rays, X-rays, and/or particles having an energy of 1 keV or greater. Scintillator materials as described herein can be used as components of radiation detectors in apparatuses such as counters, image intensifiers, and computed tomography (CT) scanners.

“Optical coupling” as used herein refers to a physical coupling between a scintillator and a photosensor, e.g., via the presence of optical grease or another optical coupling compound (or index matching compound) that bridges the gap between the scintillator and the photosensor. In addition to optical grease, optical coupling compounds can include, for example, liquids, oils and gels.

“Light output” can refer to the number of light photons produced per unit energy deposited, e.g., by a gamma ray being absorbed, typically the number of light photons/MeV.

As used herein, chemical ions can be represented simply by their chemical element symbols alone (e.g., Pr for praseodymium ion(s) (e.g.,  $\text{Pr}^{3+}$ ) or Li for lithium ion(s) (e.g.,  $\text{Li}^+$ )). Similarly, the terms “alkali metal” and “rare earth element” are used herein to refer to an alkali metal ion or a combination of alkali metal ions and a rare earth element ion or a combination of rare earth element ions, respectively.

The term “rare earth element” as used herein refers to one or more elements selected from a lanthanide (e.g., lanthanum (La), cerium (Ce), Praseodymium (Pr), neodymium (Nd), promethium (Pm), samarium (Sm), europium (Eu), gadolinium (Gd), terbium (Tb), dysprosium (Dy), holmium (Ho) erbium (Er), thulium (Tm), ytterbium (Yb) and lutetium (Lu)), scandium (Sc), and yttrium (Y).

The terms "rare earth aluminum garnet" as used herein refer to compound having mainly a chemical formula  $A_3B_5O_{12}$ , wherein cations of A and B have two different types of sites, each site being surrounded by oxygen ions. A is a rare earth element cation or a mixture of rare earth element cations and B is aluminum cations. The material can also include a small amount (e.g., about 10 atomic % or less or about 5 atomic % or less relative to A) of each of one or more dopant ions (e.g., a dopant ion and a codopant ion). In some embodiments, A includes at least some Lu. In some embodiments, A is a mixture of Lu and Y.

## II. Garnet-Type Scintillators Codoped with Monovalent Cations

As described hereinabove, it is believed that some garnet-type scintillators, such as lutetium aluminum garnet (LuAG) scintillators, have not yet reached their full potential when it comes to scintillation due to intrinsic defects. According to one aspect of the presently disclosed subject matter, a method of tailoring the properties of garnet-type scintillators is provided wherein the scintillators are codoped with monovalent ions. Thus, in some embodiments, the presently disclosed subject matter provides a monovalent cation codoped rare earth aluminum garnet scintillator material. In some embodiments, the rare earth element of the rare earth aluminum garnet is Lu, Y, or a mixture thereof. In some embodiments, the rare earth component of the rare earth aluminum garnet is a mixture of Lu and Y. The rare earth aluminum garnet can be doped with any suitable dopant/activator ion. In some embodiments, the activator/dopant ion is an ion of an element selected from the group comprising Ce, Pr, Nd, Sm, Eu, Gd, Tb, Yb, Bi, Sb, or a combination thereof. In some embodiments, the activator/dopant ion is  $Ce^{3+}$  or  $Pr^{3+}$ . In some embodiments, the activator/dopant ion is an ion of an element other than Ce. In some embodiments, the activator/dopant ion is  $Pr^{3+}$ .

Generally, the activator/dopant ion and the codopant ion are each present in the material in relatively small amounts, e.g., about 10, 5.0, 1.0, 0.9, 0.8, 0.7, 0.6, 0.5, 0.4, 0.3, 0.2, 0.1 or less atomic percentage compared

to the amount of the rare earth element or elements present in the main garnet matrix. Unless otherwise indicated, when an atomic percentage of dopant or codopant ion is described, the atomic percentage is based on the amount of dopant or codopant ion present in the starting materials used to prepare the scintillator material (e.g., in the initial melt). This amount can vary in the prepared scintillator, e.g., due to segregation during melt growth. In some embodiments, the amount of dopant is about 5.0 atomic % or less compared to rare earth element in the main garnet matrix. In some embodiments, the amount of dopant is between about 1.5 atomic % and about 0.05 atomic % compared to rare earth element in the main garnet matrix. In some embodiments, the amount of dopant is about 0.4 atomic % compared to the rare earth element in the main garnet matrix. In some embodiments, the amount of dopant is about 1.2 atomic % compared to the rare earth element in the main garnet matrix.

In some embodiments, the presently disclosed subject matter provides a scintillator material comprising, consisting essentially or, or consisting of a composition of Formula (I):



wherein:

$0 \leq x \leq 1.0$ ;

$0 < y \leq 0.05$ ;

$0 < z \leq 0.1$ ;

RE is a first rare earth element (e.g., La, Ce, Pr, Nd, Pm, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu, Sc, or Y);

RE' is a second rare earth element or a combination of a second rare earth element and at least one more or more additional rare earth elements, subject to the proviso that RE' does not comprise the first rare earth element RE;

A is a dopant ion, subject to the proviso that A does not comprise an ion of the same element as RE or RE'; and

B is at least one type of monovalent cation.

In some embodiments RE' is a rare earth element other than Ce. Suitable dopant ions for A include, but are not limited to, the group

comprising Pr, Nd, Sm, Eu, Gd, Tb, Yb, Bi, Sb, and any combination thereof. In some embodiments, A is a Pr ion (e.g.,  $\text{Pr}^{3+}$ ).

Suitable codopant ions for B include, but are not limited to, cations of alkali metal elements, such as, but not limited to, Li, Na, K, Rb, Cs and Fr.

5 In some embodiments, B is Li, Na, or K. In some embodiments, B is Li. In some embodiment B is a cation of a lithium-6 isotope ( $^6\text{Li}$ ) or a mixture of lithium isotope cations enriched for  $^6\text{Li}^+$ .

The value x can describe the composition of second rare-earth element RE' in the main garnet matrix (i.e., in the scintillator material excluding dopant and/or codopant ions). In some embodiments, x is between 0.05 and about 0.5. In some embodiments, x is between 0.1 and about 0.5 (e.g., about 0.1, 0.15, 0.2, 0.25, 0.3, 0.35, 0.4, 0.45, or about 0.5). In some embodiments, x is between 0.15 and about 0.35. In some embodiments, x is between about 0.20 and about 0.30. In some  
15 embodiments, x is about 0.25.

The value y describes the composition of activator/dopant ion. If the amount of activator is too small, energy absorbed by the material is not converted as efficiently to light. If the amount of activator is too large, the distance between activator ions can become too small, resulting in  
20 quenching. In some embodiments, the activator/dopant ion is provided at between about 0.0001 and about 5 atomic % (e.g., relative to the content of the rare earth elements (e.g., Lu and Y) in the main garnet matrix). Thus, in some embodiments, y is between about 0.0001 and about 0.05. As noted above, the common practice will be used herein of stating the amount of  
25 dopant relative to the rare earth element in the starting material mixture used to prepare the scintillator (e.g., the amount present in the melt from which the material is grown). The actual content of the dopant in the as prepared material can differ from this value (e.g., due to solid-liquid segregation, etc.). In some embodiments,  $0.001 \leq y \leq 0.05$ . In some embodiments,  
30  $0.001 \leq y \leq 0.015$ . In some embodiments, the activator/dopant ion is provided at about 0.2 atomic %. Thus, in some embodiments, y is about 0.002. In some embodiments, the activator/dopant ion is provided at about 0.4 atomic %. Thus, in some embodiments, y is about 0.004. In some embodiments,

the activator/dopant ion is provided at about 1.2 atomic %. Thus, in some embodiments, y is about 0.012.

The value z can determine the composition of codopant. In some embodiments, the codopant ion is believed to change the defect structure of the scintillator material, which can result in changes in the scintillation properties and/or performance of the material as compared to a similar noncodoped material. In some embodiments, the codopant is provided at between about 0.02 and about 10 atomic % (e.g., relative to the content of the rare earth elements in the main garnet matrix). Thus, z can be between about 0.0002 and about 0.1. In some embodiments,  $0.001 \leq z \leq 0.1$ . In some embodiments,  $0.001 \leq z \leq 0.06$ . In some embodiments, z is 0.002, 0.006, 0.008, 0.020, 0.024, or 0.06. In some embodiments, z is 0.002. In some embodiments, z is 0.008. In some embodiments, z is 0.02. Alternatively, in some embodiments, a greater amount of codopant can be used (e.g., up to about 20 atomic % or up to about 15 atomic %). For instance, in some embodiments, it is believed that the presently disclosed scintillators can find use in thermal neutron detection due to the n-alpha reaction on the  ${}^6\text{Li}$  isotope. As noted above for the dopant ion, the amount of codopant is expressed herein based upon the amount of codopant present in the starting material mixture used to prepare the scintillator.

In some embodiments, the scintillator material comprises, consists essentially of, or consists of a composition of Formula (II):



wherein:

0.05  $\leq$  x  $\leq$  1.0;

0 < y  $\leq$  0.05;

0 < z  $\leq$  0.1;

RE' is selected from the group consisting of Ce, Dy, Er, Eu, Gd, Ho, La, Nd, Pr, Pm, Sm, Sc, Tb, Tm, Yb, Y and any combination thereof;

A is a dopant ion (e.g., such as, but not limited to, the group comprising Ce, Pr, Nd, Sm, Eu, Gd, Tb, Yb, Bi, Sb, and any combination thereof) subject to the proviso that A does not comprise an ion of the same element as RE'; and

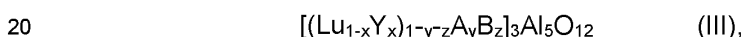
B is at least one type of monovalent cation.

In some embodiments, RE' is free of Ce. In some embodiments, RE' is Y. In some embodiments,  $0.05 \leq x \leq 0.5$ . In some embodiments,  $0.1 \leq x \leq 0.5$ . In some embodiments, x is between 0.15 and about 0.35. In some  
5 embodiments, x is between about 0.20 and about 0.30. In some embodiments, x is about 0.25.

In some embodiments, A is a dopant ion of an element other than Ce. In some embodiments, A is  $\text{Pr}^{3+}$ . In some embodiments,  $0.0001 \leq y \leq 0.05$ . In some embodiments,  $0.001 \leq y \leq 0.05$ . In some embodiments,  $0.001 \leq y \leq 0.015$ . In some embodiments, y is 0.002. In some embodiments, y is 0.004. In some embodiments, y is 0.012.

In some embodiments, B is  $\text{Li}^+$ ,  $\text{Na}^+$ , or  $\text{K}^+$ . In some embodiments, B is  $\text{Li}^+$ . In some embodiments,  $0.0002 \leq z \leq 0.1$ . In some embodiments,  $0.001 \leq z \leq 0.1$ . In some embodiments,  $0.001 \leq z \leq 0.06$ . In some embodiments, z is 0.002. In some embodiments, z is 0.006. In some embodiments, z is 0.008. In some embodiments, z is 0.02. In some embodiments, z is 0.024. In some embodiments, z is 0.06.

In some embodiments, the scintillator material comprises, consists essentially of, or consists of a composition of Formula (III):



wherein:

 $0.05 \leq x \leq 0.5$ : $0 < \gamma \leq 0.05;$  $0 < z \leq 0.1:$ 

25 A is a dopant ion (e.g., such as, but not limited to, the group comprising Ce, Pr, Nd, Sm, Eu, Gd, Tb, Yb, Bi, Sb, and any combination thereof); and

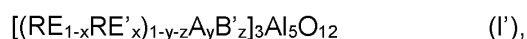
B is at least one type of monovalent cation.

In some embodiments,  $0.1 \leq x \leq 0.5$ . In some embodiments, x is  
30 between 0.15 and about 0.35. In some embodiments, x is between about  
0.20 and about 0.30. In some embodiments, x is about 0.25.

In some embodiments, A is a dopant ion of an element other than Ce. In some embodiments, A is  $\text{Pr}^{3+}$ . In some embodiments,  $0.0001 \leq y \leq 0.05$ . In some embodiments,  $0.001 \leq y \leq 0.05$ . In some embodiments,  $0.001 \leq y \leq 0.015$ . In some embodiments, y is about 0.004. In some embodiments, y is about  
 5 0.012.

In some embodiments, B is  $\text{Li}^+$ ,  $\text{Na}^+$ , or  $\text{K}^+$ . In some embodiments, B is  $\text{Li}^+$ . In some embodiments,  $0.0002 \leq z \leq 0.1$ . In some embodiments,  $0.001 \leq z \leq 0.1$ . In some embodiments,  $0.001 \leq z \leq 0.06$ . In some embodiments, z is 0.002. In some embodiments, z is 0.006. In some embodiments, z is  
 10 0.008. In some embodiments, z is 0.02. In some embodiments, z is 0.024. In some embodiments, z is 0.06.

In some embodiments, the scintillator material comprises, consists essentially of, or consists of a composition of Formula (I'):



15 wherein:

$$0 \leq x \leq .5;$$

$$0 < y \leq 0.05;$$

$$0 < z \leq 0.1;$$

RE is a first rare earth element;

20 RE' is a second rare earth element or a combination of a second rare earth element and at least one more or more additional rare earth elements, subject to the proviso that RE' does not comprise the first rare earth element RE;

A is a dopant ion selected from the group consisting of Pr, Nd, Sm, 25 Eu, Gd, Tb, Yb, Bi, Sb, and any combination thereof, subject to the proviso that A does not comprise an ion of the same element as RE or RE'; and

B' is a monovalent cation of a lithium isotope or a mixture thereof, optionally wherein B' is a monovalent cation of  ${}^6\text{Li}$  (i.e.,  ${}^6\text{Li}^+$ ) or is a mixture of lithium isotope cations enriched for  ${}^6\text{Li}^+$ .

30 In some embodiments, RE is Lu. In some embodiments, RE' is other than Ce. In some embodiments, RE' is Y. In some embodiments,  $0.1 \leq x \leq 0.5$ . In some embodiments  $0.15 \leq x \leq 0.35$ . In some embodiments,

0.20≤x≤0.3. In some embodiments x is about 0.25. In some embodiments, A is an ion of Pr (e.g., Pr<sup>3+</sup>).

In some embodiments, A is a dopant ion of an element other than Ce. In some embodiments, A is Pr<sup>3+</sup>. In some embodiments, 0.0001≤y≤0.05. In  
5 some embodiments, 0.001≤y≤0.05. In some embodiments, 0.001≤y≤0.015. In some embodiments, y is about 0.002. In some embodiments, y is about 0.004. In some embodiments, y is about 0.012.

In some embodiments, 0.0002≤z≤0.1. In some embodiments, 0.001≤z≤0.1. In some embodiments, 0.001≤z≤0.06. In some embodiments,  
10 z is 0.002. In some embodiments, z is 0.006. In some embodiments, z is 0.008. In some embodiments, z is 0.02. In some embodiments, z is 0.024. In some embodiments, z is 0.06.

In some embodiments, the scintillator materials comprise, consist essentially of, or consist of (Lu<sub>0.75</sub>,Y<sub>0.25</sub>)<sub>3</sub>Al<sub>5</sub>O<sub>12</sub>:0.4%Pr<sup>3+</sup> codoped with  
15 0.2%Li<sup>+</sup>, 0.8% Li<sup>+</sup>, or 2.0% Li<sup>+</sup>; (Lu<sub>0.75</sub>,Y<sub>0.25</sub>)<sub>3</sub>Al<sub>5</sub>O<sub>12</sub>:1.2%Pr<sup>3+</sup> codoped with 0.6 at% Li<sup>+</sup>, 2.4 at% Li<sup>+</sup> or 6 at% Li<sup>+</sup> or (Lu<sub>0.75</sub>,Y<sub>0.25</sub>)<sub>3</sub>Al<sub>5</sub>O<sub>12</sub>:1.2%Pr<sup>3+</sup> codoped with 0.6 at% K<sup>+</sup> or 0.308 at% Na<sup>+</sup>. In some embodiments, the scintillator materials comprise, consist essentially of, or consist of  
(Lu<sub>0.75</sub>,Y<sub>0.25</sub>)<sub>3</sub>Al<sub>5</sub>O<sub>12</sub>:0.4%Pr<sup>3+</sup> codoped with 0.2%Li<sup>+</sup>, 0.8% Li<sup>+</sup>, or 2.0% Li<sup>+</sup>.  
20

In some embodiments, the scintillator material exhibits one or more of increased light yield, better energy resolution, and an accelerated fast decay component as compared to the scintillator material where B is absent.

The scintillator materials of the presently disclosed subject matter (e.g., of Formulas (I), (I'), (II), or (III)) can be a single crystal, a polycrystalline  
25 material, and/or a ceramic. By “single crystal” is meant a material manufactured by a liquid phase method having few or no grain boundaries and wherein each adjoining crystal grain generally has the same orientation. In some embodiments, the material can be polycrystalline and/or ceramic and contain crystals of varying size and/or orientation.

30

### III. Radiation Detectors, Related Devices and Methods

In some embodiments, the presently disclosed subject matter provides a radiation detector comprising an optical material (e.g., a

scintillation material) comprising, consisting essentially of, or consisting of a composition of Formula (I), (I'), (II), or (III) as described hereinabove or a mixture of such materials. The radiation detector can comprise a scintillator (which absorbs radiation and emits light) and a photodetector (which detects  
5 said emitted light). The photodetector can be any suitable detector or detectors and can be or not be optically coupled to the scintillator material for producing an electrical signal in response to emission of light from the scintillator material. Thus, the photodetector can be configured to convert photons to an electrical signal. For example, a signal amplifier can be  
10 provided to convert an output signal from a photodiode into a voltage signal. The signal amplifier can also be designed to amplify the voltage signal. Electronics associated with the photodetector can be used to shape and digitize the electronic signal.

Referring now to Figure 9, in some embodiments, the presently  
15 disclosed subject matter provides an apparatus **10** for detecting radiation wherein the apparatus comprises a photon detector **12** and a scintillator material **14** (e.g., a codoped LuYAG material). Scintillator material **14** can convert radiation to light that can be collected by a charge-coupled device (CCD) or a photomultiplier tube (PMT) or other photon detector **12** efficiently  
20 and at a fast rate.

Referring again to Figure 9, photon detector **12** can be any suitable detector or detectors and can be optically coupled (e.g., via optical grease or another optical coupling compound, such as an optical coupling oil or liquid) to the scintillator (e.g., a codoped LuYAG material) for producing an  
25 electrical signal in response to emission of light from the scintillator. Thus, photon detector **12** can be configured to convert photons to an electrical signal. Electronics associated with photon detector **12** can be used to shape and digitize the electronic signal. Suitable photon detectors **12** include, but are not limited to, photomultiplier tubes, photodiodes, CCD sensors, and  
30 image intensifiers. Apparatus **10** can also include electronics **16** for recording and/or displaying the electronic signal.

In some embodiments, the radiation detector is configured for use as part of a medical or veterinary diagnostic device, a device for oil or other

geological exploration (e.g., oil well logging probes), or as a device for security and/or military-related purposes (e.g., as a device for container, vehicle, or baggage scanning or for scanning humans or other animals). In some embodiments, the medical or veterinary diagnostic device is selected from, but not limited to, a positron emission tomography (PET) device, an X-ray computed tomography (CT) device, a single photon emission computed tomography (SPECT) device, or a planar nuclear medical imaging device. For example, the radiation detector can be configured to move (e.g., via mechanical and/or electronic controls) over and/or around a sample, such as a human or animal subject, such that it can detect radiation emitted from any desired site or sites on the sample. In some embodiments, the detector can be set or mounted on a rotating body to rotate the detector around a sample.

In some embodiments, the device can also include a radiation source. For instance, an X-ray CT device of the presently disclosed subject matter can include an X-ray source for radiating X-rays and a detector for detecting said X-rays. In some embodiments, the device can comprise a plurality of radiation detectors. The plurality of radiation detectors can be arranged, for example, in a cylindrical or other desired shape, for detecting radiation emitted from various positions on the surface of a sample.

In some embodiments, the presently disclosed subject matter provides a method for detecting radiation (or the absence of radiation) using a radiation detector comprising a scintillator as described hereinabove (e.g., a codoped LuYAG scintillator material). Thus, in some embodiments, the presently disclosed subject matter provides a method of detecting gamma rays, X-rays, cosmic rays and particles having an energy of 1keV or greater, wherein the method comprises using a radiation detector comprising a material comprising a composition of one of Formulas (I), (II), or (III).

In some embodiments, the codopant comprises a lithium-6 isotope cation and the material is a composition of formula (I'). In some embodiments, the presently disclosed subject matter provides a method for thermal neutron detection using a radiation detector comprising a scintillation material of formula (I').

In some embodiments, the method can comprise providing a radiation detector comprising a photodetector and an optical (e.g., scintillator) material of the presently disclosed subject matter; positioning the detector, wherein the positioning comprises placing the detector in a location wherein the optical material is in the path of a beam of radiation (or the suspected path of a beam of radiation); and detecting light (or detecting the absence of light) emitted by the optical material with the photodetector. Detecting the light emitted by the optical material can comprise converting photons to an electrical signal. Detecting can also comprise processing the electrical signal to shape, digitize, or amplify the signal. The method can further comprise displaying the electrical signal or processed electrical signal.

In some embodiments, the presently disclosed subject matter provides a device comprising a photodetector and a scintillator material as described hereinabove, such as a material comprising a monovalent cation codoped rare earth aluminum garnet material, such as a material comprising a composition of one of Formulas (I), (I') (II), or (III), or a mixture of such materials. In some embodiments, the device comprising the photodetector and the scintillator material is adapted for use in medical imaging, geological exploration, or homeland security. In some embodiments, the presently disclosed subject matter provides a method of detecting high energy photons and particles, wherein the method comprises using the device comprising the photodetector and the optical material comprising a composition of one of Formulas (I), (I'), (II), or (III), or a mixture of such materials.

#### 25 IV. Methods of Preparation

The presently disclosed optical (e.g., scintillation) materials can be prepared via any suitable method as would be apparent to one of ordinary skill in the art upon a review of the instant disclosure. In some embodiments, the presently disclosed subject matter provides a method of preparing a codoped garnet-type scintillator material. In some embodiments, the presently disclosed subject matter provides a method for preparing a scintillator material that comprises preparing a crystal from a melt. For instance, in some embodiments, the codoped garnet-type scintillator material

can be a crystal grown by the Czochralski (pulling-up) method. However, single crystals or polycrystalline materials and/or ceramics grown or produced by other methods can also be used as a scintillator material according to the present disclosure. For example, alternative methods for producing garnet-type materials include, but are not limited to the micro-pulling down method, Bridgman method, zone melt method, Edge-defined Film-fed Growth (EFG) method, and hot isostatic press (HIP) sintering method.

In any production method of crystals, an oxide or carbonate raw material can be used as a starting material. Thus, suitable starting materials for preparing the crystals include, but are not limited to,  $\text{Gd}_2\text{O}_3$ ,  $\text{Y}_2\text{O}_3$ ,  $\alpha\text{-Al}_2\text{O}_3$ ,  $\text{CeO}_2$ ,  $\text{Pr}_6\text{O}_3$ ,  $\text{Li}_2\text{CO}_3$ ,  $\text{Lu}_2\text{O}_3$ ,  $\text{K}_2\text{CO}_3$ ,  $\text{NaHCO}_3$ , and the like. In some embodiments, the starting materials include a  $^6\text{Li}$  enriched lithium compound. When the crystal is used as a crystal for a scintillator, a high-purity raw material (e.g., having a purity of 99.99% or higher and/or not containing more than 1 ppm of an impurity) can be used. These starting materials can be weighed and mixed such that a desired composition is obtained at the time of forming a melt.

In some embodiments, the Czochralski technique (in which large single crystals are "pulled" from molten raw material) can be used to grow codoped rare-earth gallium crystal boules. Raw materials can be measured out and mixed, e.g., using a ball mill, etc., and the mixed powder placed into a crucible. Calcination can be performed at, for example, 1000 to 1700 degrees Celsius for several hours. Suitable crucible materials include platinum, iridium, rhodium, rhenium, and alloys thereof. A high frequency oscillator, a condensing heater, or a resistance heater can be used. Further, a flowing atmosphere of argon, helium, or nitrogen can be used. In some embodiments, an atmosphere of nitrogen with a small amount of oxygen (e.g., between about 0.1 to about 5 vol %) can be used.

In some embodiments, the presently disclosed materials can be provided as ceramics, for example, by using a hot press or hot isotatic press (HIP) method. In this method, the raw materials (e.g.,  $\text{Gd}_2\text{O}_3$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{Ga}_2\text{O}_3$ , cerium salt (e.g., cerium nitrate), etc.) can be measured out and mixed, e.g.,

using a ball mill, etc. Then the mixed powders can be put into a crucible (e.g., an alumina crucible) and calcination can be performed (e.g., at a temperature of 1200 to 1500 degrees Celsius (°C) for several hours. In the case of the hot press method, after the calcination, press molding can be performed to get a formed object using a die, after granulating the powder using a sieve with a suitable aperture. Then, the formed object can be set to a carbon die, and hot press sintering can be performed in an inert gas atmosphere at, for example, 1500 to 1700 °C and at a pressure of 10 megapascals (MPa) to 80 MPa. In the case of the HIP method, calcination powder is ground using a ball mill etc., and press molding can be performed to get a formed object using a die. The obtained formed object can be densified by a cold isostatic press method, put into a sagger made of alumina, and calcination carried out at a temperature of, for example, 1500 to 1700 °C, in an inactive gas atmosphere. HIP sintering can be further performed to the obtained ceramics at a pressure of 50 MPa or higher, and at a temperature of 1300 to 1700 °C.

The scintillation materials can be provided as single crystals, as a polycrystalline material, and/or as a ceramic material. In some embodiments, the material is provided as a polycrystalline and/or ceramic material. The polycrystalline and/or ceramic material can have analogous physical, optical and scintillation properties as a single crystal otherwise having the same chemical composition.

In some embodiments, the method further comprises annealing the scintillator material for a period of time (e.g., between a few hours and a few days). The annealing can be performed, for example, in air, nitrogen, or a mixture of nitrogen and hydrogen. The annealing can be done at any suitable temperature, e.g., between about 800 and about 1600 degrees Celsius (e.g., about 800, 900, 1000, 1100, 1200, 1300, 1400, 1500, and about 1600 degrees Celsius). In some embodiments, the annealing increases the light yield of the material and/or provides a material with a faster scintillation decay time. In some embodiments, the annealing is performed in air. In some embodiments, the annealing is performed at a temperature of about 1200°C and/or for a time period of about 48 hours.

## V. Methods of Altering Scintillation and/or Optical Properties

In some embodiments, the presently disclosed subject matter provides a method of altering one or more scintillation and/or optical properties of a garnet-type scintillation material, such as, but not limited to, scintillation light yield, decay time, rise time, energy resolution, proportionality, and sensitivity to light exposure. In some embodiments, the method comprises preparing the garnet-type scintillation material in the presence of a dopant ion and one or more monovalent codopant ions. In some embodiments, the garnet-type scintillation material is a rare earth aluminum garnet. In some embodiments, the garnet-type scintillation material is a mixed rare earth aluminum garnet and comprises a matrix having the formula  $RE''_3Al_5O_{12}$ , wherein  $RE''$  is a mixture of at least two rare earth elements and wherein the scintillator further comprises at least one dopant selected from the group comprising Ce, Pr, Nd, Sm, Eu, Gd, Tb, Yb, Bi, Sb, and any combination thereof, subject to the proviso that the dopant does not comprise an ion of the same element as any rare earth element of the rare earth aluminum garnet matrix. In some embodiments, the at least one dopant is selected from the group comprising Pr, Nd, Sm, Eu, Gd, Tb, Yb, Bi, Sb, and any combination thereof. In some embodiments, the matrix is doped with 5 at% or less of the dopant compared to the rare earth elements. In some embodiments, the amount of codopant is 10 at% or less compared to the rare earth element and dopant content. In some embodiments,  $RE''$  comprises Lu and at least one other type of rare earth element. In some embodiments, the at least one other type of rare earth element is other than Eu or Pr. In some embodiments, the dopant is Eu or Pr. In some embodiments, the dopant is Pr.

In some embodiments, the monovalent codopant ion is an alkali metal ion. In some embodiments, the monovalent codopant ion is selected from  $Li^+$ ,  $Na^+$ , and  $K^+$ . In some embodiments, the monovalent codopant ion is  $Li^+$ .

In some embodiments, the garnet-type scintillation material is a lutetium aluminum garnet (LuAG) or a lutetium yttrium aluminum garnet (LuYAG). In some embodiments, the garnet-type scintillation material is a

lutetium aluminum garnet (LuAG) wherein at least some of the Lu is replaced by another rare earth element. In some embodiments, the garnet-type scintillation material is a lutetium yttrium aluminum garnet (LuYAG).

In some embodiments, the codoping provides increased light yield and/or improved energy resolution. In some embodiments, the codoping provides a rare earth aluminum garnet scintillator material with energy resolution at 662 keV of about 4.8% or less. In some embodiments, the codoping provides a rare earth aluminum garnet scintillator material with energy resolution at 662 keV of about 4.4% or less. In some embodiments, the codoping provides a rare earth aluminum garnet scintillator material with energy resolution at 4.1%. In some embodiments, the codoping provides a rare earth aluminum garnet scintillator material with a faster decay time.

#### EXAMPLES

The following Examples have been included to provide guidance to one of ordinary skill in the art for practicing representative embodiments of the presently disclosed subject matter. In light of the present disclosure and the general level of skill in the art, those of skill can appreciate that the following Examples are intended to be exemplary only and that numerous changes, modifications, and alterations can be employed without departing from the scope of the presently disclosed subject matter.

#### EXAMPLE 1

##### 0.2% Li Codoped LuYAG:Pr

[Lu<sub>1-x</sub>Y<sub>x</sub>)<sub>1-y-z</sub>A<sub>y</sub>B<sub>z</sub>]<sub>3</sub>Al<sub>5</sub>O<sub>12</sub> was prepared where A is Pr<sup>3+</sup> and B is Li<sup>+</sup>. High purity raw materials were mixed and loaded directly into a 60 mm diameter iridium crucible according to the respective stoichiometric formulas. Czochralski crystal growth was carried out in a Cyberstar Oxypuller growth station (Cyberstar, Echirolles, France), using an automated system in which the derivative of the crystal weight was the process variable, to produce ~490 gram boules of the nominal composition (Lu<sub>0.748</sub>,Y<sub>0.25</sub>,Pr<sub>0.004</sub>,Li<sub>0.002</sub>)<sub>3</sub>Al<sub>5</sub>O<sub>12</sub>. Crystal growth was initiated on LuAG: Ce seed crystals oriented in the <111> direction. The atmosphere was primarily

nitrogen with a small fraction of a percent oxygen. The results were transparent single crystals that were cut into 5 x 5 x 5 mm pixels and 1 mm x 33 mm diameter slabs for measurements. The example of the type is presented below compared to previously reported doped garnet single crystals. See Table 1.

Table 1. Noncodoped LuAG and LuYAG and LuYAG:Pr Codoped with 0.2% Li.

| Composition                                                                                            | Light output (ph/MeV) | Absorbance (nm) | RL Max emission (nm) | Primary Decay time (ns) |
|--------------------------------------------------------------------------------------------------------|-----------------------|-----------------|----------------------|-------------------------|
| $(\text{Lu}_{0.744}, \text{Y}_{0.25}, \text{Pr}_{0.004}, \text{Li}_{0.002})_3\text{Al}_5\text{O}_{12}$ | 16,000                | 240, 285        | 330, 378             | 22 ns                   |
| LuAG: Pr <sup>1</sup>                                                                                  | 12,000-14,000         | 240, 285        | 325, 383             | 25 ns                   |
| LuYAG: Pr <sup>2</sup>                                                                                 | 27,000                | Not available   | 286-450 range        | 27 ns                   |

<sup>1</sup>Values obtained from Niki et al., Physical Status Solidi (a), 202(1), R4-R6 (2005). <sup>2</sup>Values obtained from Drozdowski et al., Optical Materials, 59, 107-114 (September 2016).

Absorbance measurements were completed using a Varian Cary 5000 UV-Vis-NIR Spectrophotometer (Varian Inc., Palo Alto, California, United States of America) on polished samples about 1 mm thick. The wavelengths of peaks observed in the absorbance spectrum shown in Figure 1 are attributed to characteristic absorbance of the activator, verifying the charge states that are present within the material. Radioluminescence (RL) spectra were measured at room temperature under continuous irradiation from an X-ray generator model CMX003 (32 kV and 0.1 mA). A model PI Acton Spectra Pro SP-2155 monochromator (Princeton Instruments, Acton Massachusetts, United States of America) was used to record the spectra. The single peak emission peak observed in the RL spectra shown in Figure 2 is attributed to characteristic emission of activator transitions.

Light output measurements are shown in Figure 3. A Hamamatsu  
bialkali R2059 photomultiplier tube (PMT, Hamamatsu Photonics, K.K.,  
Hamamatsu City, Japan), an Ortec 672 amplifier (Advanced Measurement  
Technology, Inc., Oak Ridge, Tennessee, United States of America), a  
5 Canberra model 2005 pre-amplifier (Canberra Industries, Inc., Meriden,  
Connecticut, United States of America), and a Tukan 8k multi-channel  
analyzer (MCA, National Center for Nuclear Research, Świerk, Poland) were  
the components of the pulse processing chain. See Guzik et al., IEEE  
Transactions on Nuclear Science, 53(1), 231-235 (2006). The sample was  
10 excited with a 10  $\mu$ Ci Cs-137 (662 keV) source and was coupled to the PMT  
with Corning optical grease. Light capture was enhanced by covering five  
sides of each sample with multiple layers of TEFLON® tape, and a reflective  
SPECTRALON® dome (Labsphere, North Sutton, New Hampshire, United  
States of America) was placed on top. The photopeaks were fitted with a  
15 Gaussian function to determine the centroid of the peak. The integral  
quantum efficiency of PMT according to the emission spectrum of the  
scintillator was used to estimate the light output in photons per unit of  
gamma-ray energy. Scintillation decay time was recorded using a  $^{137}\text{Cs}$   
source and the time-correlated single photon counting technique previously  
20 described in Bollinger and Thomas (Review of Scientific Instruments, 32, 7,  
(1961)). The decay curves shown in Figure 4 were fitted with a three-  
component exponential decay function. The decay times were 22  
nanoseconds (ns) (40 percent (%)), 656 ns (31%), and 119.1 ns (29%).

Both the light yield and decay time were modified with the addition of  
25 lithium to a LuYAG:Pr single crystal scintillator. A particular improvement  
was found in the acceleration of the primary decay component. It is believed  
that these results are not limited to this compound and can be applied to  
other garnet scintillators, such as LuGAG:Ce and LuGAG:Pr, as well as to  
related polycrystalline and/or ceramic scintillators.

30

## EXAMPLE 2

### Additional Codoped LuYAG:Pr Materials

Boules of 0.4 at% praseodymium-doped LuYAG, with a ratio of Lu to Y of 3:1, and Li concentrations of 0 at%, 0.2 at%, 0.8 at% and 2.0 at% and  
5 boules of 1.2 at% Pr-doped LuYAG with a ratio of Lu to Y of 3:1 and lithium concentrations of 0 at%, 0.6 at%, 2.4 at%, and 6 at% with respect to the rare earth element were grown via the Czochralski growth method in a Cyberstar Oxypuller growth station (Cyberstar, Echirolles, France) using an automated system in which the derivative of the crystal weight was the process variable.  
10 The  $\text{Lu}_2\text{O}_3$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{Y}_2\text{O}_3$ ,  $\text{Pr}_2\text{O}_3$ , and  $\text{Li}_2\text{CO}_3$  raw materials were added directly to a 60 mm diameter iridium crucible. In addition, boules of a 1.2 at% Pr-doped LuYAG with a ratio of Lu to Y of 3:1 and a potassium concentration of 0.6 at% or a sodium concentration of 0.308 at% with respect to the rare earth element were grown.

15 Absorbance measurements were completed using a Varian Cary 5000 UV-Vis-NIR Spectrophotometer (Varian Inc., Palo Alto, California, United States of America) on polished samples about 1 mm thick. The wavelengths of peaks observed in the absorbance spectra shown in Figure 5 are attributed to characteristic absorbance of the activator, verifying the  
20 charge states that are present within the material. The optical properties of the praseodymium activation in the noncodoped and codoped is represented by the photoluminescence (PL) excitation and emission spectra in Figure 6. PL emission and excitation spectra were acquired with a Horiba Jobin Yvon Fluorolog-3 spectrofluorometer (Horiba, Kyoto, Japan) using a 450 Watt (W)  
25 continuous Xe lamp as the excitation source. The excitation of the activator was measured at 240 and 280 nm. As shown in Figure 6, the wavelength of excitation and emission does not change with the incorporation of lithium into the matrix as a codopant.

Light output measurements of some of the 0.4 at% Pr doped Li  
30 codoped samples are shown in Figure 7. The absolute light yield was determined from pulse height spectra for each composition using a pulse processing chain consisting of a super bialkali R2059 photomultiplier tube (PMT, Hamamatsu Photonics, K.K., Hamamatsu City, Japan), an ORTEC®

672 amplifier (Advanced Measurement Technology, Inc., Oak Ridge, Tennessee, United States of America), a Canberra model 2005 pre-amplifier (Canberra Industries, Ind., Meridan, Connecticut, United States of America) and a Tukan 8k multi-channel analyzer (MCA, National Center for Nuclear Research, Świerk, Poland). Each sample was excited with a 10  $\mu$ Ci Cs-137 (662 keV) source and was coupled to the PMT with Corning optical grease. Light capture was enhanced by covering five sides of each sample with multiple layers of Teflon tape, and a reflective SPECTRALON® dome (Labsphere, North Sutton, New Hampshire, United States of America) was placed on top. The photopeaks were fitted with a Gaussian function to determine the centroid of the peak. The integral quantum efficiency of PMT according to the emission spectrum was used to estimate the light output in photons per unit of gamma-ray energy.

The simplified description of energy resolution can be described as the ratio of the full width half maximum of the photopeak divided by the centroid position of the gaussian as shown in Equation 1; however, other factors, such as nonproportionality, or the scintillators performance along a range of energies also plays a role in the energy resolution (see Knoll, Radiation Detection and Measurement, John Wiley & Sons, 2010; and Dorenbos et al., IEEE Transactions on Nuclear Science, 42(6), 2190-2202 (1995):

$$R = \frac{FWHM}{H_o} (1)$$

The relative light yield at gamma energies ranging from 32 to 1333 keV was collected on 5 x 5 x 5 mm samples taken from a similar position along the boule length. Figure 8 depicts the nonproportional response to gamma energies for some 0.4 at% Pr doped Li codoped samples with different lithium concentrations. Scintillation decay times were measured using two R2059 Hamamatsu PMTs (Hamamatsu Photonics, K.K., Hamamatsu City, Japan) and a  $^{137}\text{Cs}$  gamma source in the configuration described by Bollinger and Thomas (Review of Scientific Instruments, 32, 7 (1961).

*Discussion:* Table 2 below summarizes the data collected for the additional LuYAG samples prepared. In Table 2, the symbol \*\* means that the measurement has not yet been made. Dopant and codopant concentrations are nominal concentrations added directly to the melt inside the crucible.

Table 2. Additional Noncodoped and Codoped LuYAG Samples.

| Sample # | Dopant Conc. | Codopant Conc. | LY ph/MeV | ER (%) | Fast $\tau_d$ (ns) |
|----------|--------------|----------------|-----------|--------|--------------------|
| S-181    | 0.2 at% Pr   | none           | 30,000    | 4.7    | 62.8               |
| S-215    | 0.4 at% Pr   | 0.2 at% Li     | 16,000    | **     | 22                 |
| S-216    | 0.4 at% Pr   | none           | 16,000    | 4.8    | 41.3               |
| S-217    | 0.4 at% Pr   | 0.8 at% Li     | 25,000    | 4.1    | 48.4               |
| S-220    | 0.4 at% Pr   | 2.0 at% Li     | 22,000    | 4.4    | 49.6               |
| S-223    | 0.4 at% Pr   | 0.2at% Li      | 31,000    | 4.3    | 45.9               |
| S-226    | 1.2 at% Pr   | 2.4 at% Li     | 17,000    | 4.8    | 22.2               |
| S-227    | 1.2 at% Pr   | 6 at% Li       | 26,000    | 4.3    | 30.6               |
| S-229    | 1.2 at% Pr   | 0.6 at% Li     | **        | **     | **                 |
| S-231    | 1.2 at% Pr   | 0.6 at% K      | **        | **     | **                 |
| S-233    | 1.2 at% Pr   | 0.308 at% Na   | **        | **     | **                 |

As can be observed by comparing the results from Samples S-181 and S-216, higher concentrations of dopant (e.g., Pr) yield higher light yield but slower decay time ( $\tau_d$ ). The original 0.4 at% Pr doped, 0.2 at% Li codoped sample (S-215) described in Example 1 had a smaller crystal size

than other growths and, therefore, re-synthesized (i.e., sample S-223). It is the data from the re-synthesized S-223 sample that is shown in Figures 5-8.

By comparing the different 0.4 at% Pr doped samples (S-216, S-223, S-217, and S-220), it appears that adding a monovalent dopant (e.g., lithium) to LuYAG: Pr scintillators can improve both light yield and energy resolution when the dopant concentration is low. In particular, at 662 keV gamma energy, the energy resolution of the S-217 sample (0.8 at% Li) is 4.1%, which is a breakthrough for oxide scintillators and challenges the values obtained by that of NaI: TI (6.7%), CsI: TI (6.6%), and even LuAG: Pr reported at 4.6%. See Suzuki et al., Applied Physics Express, 5(10), 102601 (2012); and Khodyuk et al., IEEE Transactions on Nuclear Science, 57(3) 1175-1181 (2010). Higher at% of lithium codopant (e.g., 1.0 at%) do not show as great an influence on light yield and energy resolution as lower codopant amounts (e.g., 0.2 at% and 0.8 at%). The greatest increase in light yield was seen with the 0.1 at% Li codoped, 0.4 at% Pr doped sample (S-223). In general, it appears that the codopant has a lesser effect on light yield and energy resolution when the dopant amount is higher (e.g. 1.2 at%).

*Summary:* The effects of codoping on rare earth aluminum garnets, such as materials of the type  $[\text{Lu}_{1-x}\text{RE}_x]_{1-y-z}\text{A}_y\text{B}_z\text{Al}_5\text{O}_{12}$ , with a monovalent atom are described. The light yield, energy resolution, and decay time of a Pr doped LuYAG single crystal scintillator can be modified with the addition of lithium. Particular improvements were found in the increase of the light yield and the improvement in energy resolution. In some cases, there was an acceleration of the fast decay time component. It is expected that further tuning of the dopant and codopant concentrations, as well as the type of dopant (e.g, Pr or Ce) and codopant (e.g., Li, Na or K) will permit production of scintillator crystals with a desired light yield, energy resolution and decay time. It is expected that these results are not limited to the LuYAG matrix and can be applied to other garnet scintillators, such as LuGAG:Ce and LuGAG:Pr, as well as to ceramic scintillators.

### EXAMPLE 3

#### Annealing of Codoped Scintillators

Light output and scintillation decay time measurements were completed on samples described in Example 2 of 0.4 at% Pr-doped LuYAG with a ratio of Lu to Y of 3:1 and a lithium concentration of 0 at%, 0.2 at%, 0.8 at% or 2 at% with respect to the rare earth element. After preliminary measurements were completed, each sample was annealed in an oxidizing atmosphere (i.e. air) for a length of time (i.e. 48 hours) at a high temperature (i.e. 1200°C). After this annealing cycle, the light output and scintillation decay time measurements were repeated. Each sample was kept in a tin container to prevent exposure to light before each measurement.

To analyze the impact of codoping and thermal annealing on the defect structure of the LuYAG: Pr, Li single crystals, thermoluminescence studies were also completed. Samples of 5 x 5 x 5 mm dimension were cooled to a temperature of 15K before excitation from an x-ray source with power settings of 30 kV and 0.1 mA. After 15 minutes of excitation, the x-ray source was turned off, and the temperature was raised from 15K to 550K at a rate of 3 K/min. The release of electrons from deep traps within the lattice is seen as a peak on the thermoluminescence glow curve and can be observed in Figure 10, before and after each sample was annealed in air.

The effect of annealing LuYAG: Pr, Li samples in air on light yield, energy resolution, and decay times is summarized in Table 3, below. As indicated in Table 3, annealing in air for 48 h at 1200°C can improve the scintillation light yield and decay time. The energy resolution was not improved with air annealing; however, for all samples containing lithium as a codopant, the energy resolution remained below 5%. Thermoluminescence measurements were completed on all the samples before and after the thermal treatment. As shown in Figure 10, after thermal annealing the peaks below ~250 K lowered in intensity for lithium codoped LuYAG: Pr single crystals.

Table 3. Effects of Annealing on Codoped LuYAG.

| Li Conc.<br>(annealing<br>status) | Relative LY<br>(% versus<br>BGO) | ER (%) | $\tau_1$<br>(ns) | $\tau_2$<br>(ns) | $\tau_3$<br>(ns) |
|-----------------------------------|----------------------------------|--------|------------------|------------------|------------------|
| 0 at% (as<br>grown)               | 279                              | 4.8    | 41.3             | 322.3            | 1328.2           |
| 0 at%<br>(annealed)               | 343                              | 5.1    | 41.8             | 350.3            | 1481.1           |
| 0.2 at%<br>(as grown)             | 559                              | 4.3    | 46.4             | 395              | 1470.1           |
| 0.2 at%<br>(annealed)             | 591                              | 4.5    | 45.9             | 360.3            | 1410             |
| 0.8 at%<br>(as grown)             | 459                              | 4.1    | 48.4             | 379.4            | 1444.4           |
| 0.8 at%<br>(annealed)             | 551                              | 4.5    | 44.9             | 298.7            | 1177.3           |
| 2.0 at% (as<br>grown)             | 457                              | 4.4    | 49.6             | 404.3            | 1513.8           |
| 2.0 at%<br>(annealed)             | 498                              | 4.6    | 41.3             | 325              | 1356.5           |

5

It will be understood that various details of the presently disclosed subject matter may be changed without departing from the scope of the presently disclosed subject matter. Furthermore, the foregoing description is for the purpose of illustration only, and not for the purpose of limitation.

10

## CLAIMS

1. A scintillator material comprising a composition of Formula (III):



5 wherein:

$$0.05 \leq x \leq 0.5;$$

$$0.0001 \leq y \leq 0.015;$$

$$0.0002 \leq z \leq 0.1;$$

A is  $\text{Pr}^{3+}$ ; and

10 B is at least one type of monovalent cation selected from the group consisting of Li, Na, and K.

2. The scintillator material of claim 1, wherein  $0.1 \leq x \leq 0.5$ , optionally wherein x is 0.25.

15

3. The scintillator material of claim 1 or claim 2, wherein y is 0.004 or 0.012.

4. The scintillator material of any one of claims 1-3, wherein B is a monovalent cation of a lithium isotope or a mixture thereof, optionally wherein B is a monovalent cation of a lithium-6 isotope (i.e.,  ${}^6\text{Li}^+$ ).

20

5. The scintillator material of any one of claims 1-4, wherein  $0.001 \leq z \leq 0.1$ , further optionally wherein z is 0.002, 0.008, or 0.02.

25

6. The scintillator material of any one of claims 1-5, wherein the scintillator material comprises a material selected from the group consisting of  
 $(\text{Lu}_{0.75}, \text{Y}_{0.25})_3\text{Al}_5\text{O}_{12}:0.4\%\text{Pr}^{3+}$  codoped with 0.2%  $\text{Li}^+$ ,  
 $(\text{Lu}_{0.75}, \text{Y}_{0.25})_3\text{Al}_5\text{O}_{12}:0.4\%\text{Pr}^{3+}$  codoped with 0.8%  $\text{Li}^+$ ,  
 30  $(\text{Lu}_{0.75}, \text{Y}_{0.25})_3\text{Al}_5\text{O}_{12}:0.4\%\text{Pr}^{3+}$  codoped with 2.0%  $\text{Li}^+$ ;  
 $(\text{Lu}_{0.75}, \text{Y}_{0.25})_3\text{Al}_5\text{O}_{12}:1.2\%\text{Pr}^{3+}$  codoped with 0.308 at%  $\text{Na}^+$ ,  
 $(\text{Lu}_{0.75}, \text{Y}_{0.25})_3\text{Al}_5\text{O}_{12}:1.2\%\text{Pr}^{3+}$  codoped with 0.6 at%  $\text{Li}^+$ ,  
 $(\text{Lu}_{0.75}, \text{Y}_{0.25})_3\text{Al}_5\text{O}_{12}:1.2\%\text{Pr}^{3+}$  codoped with 0.6 at%  $\text{K}^+$ ,

(Lu<sub>0.75</sub>,Y<sub>0.25</sub>)<sub>3</sub>Al<sub>5</sub>O<sub>12</sub>:1.2%Pr<sup>3+</sup> codoped with 2.4 at% Li<sup>+</sup>, and (Lu<sub>0.75</sub>,Y<sub>0.25</sub>)<sub>3</sub>Al<sub>5</sub>O<sub>12</sub>:1.2%Pr<sup>3+</sup> codoped with 6 at% Li<sup>+</sup>, optionally wherein the material is selected from the group consisting of (Lu<sub>0.75</sub>,Y<sub>0.25</sub>)<sub>3</sub>Al<sub>5</sub>O<sub>12</sub>:0.4%Pr<sup>3+</sup> codoped with 0.2%Li<sup>+</sup>, (Lu<sub>0.75</sub>,Y<sub>0.25</sub>)<sub>3</sub>Al<sub>5</sub>O<sub>12</sub>:0.4%Pr<sup>3+</sup> codoped with 0.8% Li<sup>+</sup>,  
 5 and (Lu<sub>0.75</sub>,Y<sub>0.25</sub>)<sub>3</sub>Al<sub>5</sub>O<sub>12</sub>:0.4%Pr<sup>3+</sup> codoped with 2.0% Li<sup>+</sup>.

7. The scintillator material of any one of claims 1-6, wherein the scintillator material is a single crystal material or a polycrystalline or a ceramic material.

10 8. A radiation detector comprising a photon detector and a scintillation material of any one of claims 1-7.

9. The radiation detector of claim 8, wherein the detector is a medical diagnostic device, a device for oil exploration, or a device for container or  
 15 baggage scanning.

10. The use of the radiation detector of claim 8 for the detection of gamma rays, X-rays, cosmic rays and/or particles having an energy of 1 keV or greater, or for the detection of neutrons.

20

11. A method of preparing a scintillator material of any one of claims 1-7, wherein the method comprises pulling a single crystal from molten raw materials.

25 12. A method of altering one or more scintillation and/or optical properties of a rare earth aluminum garnet scintillator, wherein the rare earth aluminum garnet scintillator is a praseodymium (Pr) doped lutetium yttrium aluminum garnet (LuYAG) scintillator, wherein the method comprises preparing the rare earth aluminum garnet scintillator in the presence of a monovalent codopant ion, thereby providing a codoped rare earth aluminum garnet scintillator  
 30 material, wherein the monovalent codopant ion is an alkali metal ion, optionally wherein the monovalent codopant ion is Li<sup>+</sup>.