

ANTIBIOTICS

2173649

This invention relates to new antitumor antibiotic substances and to their production and isolation.

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The antitumor compounds of the present invention have not yet been identified in terms of structure. In view of their unique physical, chemical and biological properties, however, applicant believes that the BBM-1675C and BBM-1675D antibiotics
10 are novel substances.

United Kingdom Patent Application No. 2,141,425, published December 19, 1984, discloses fermentation of Actinomadura verrucosospora strain H964-92 (ATCC 39334) or
15 Actinomadura verrucosospora strain A1327Y (ATCC 39638) to produce a new antitumor antibiotic complex designated as BBM-1675. Two major bioactive components of the BBM-1675 complex described therein were designated as BBM-1675A₁ and BBM-1675A₂. The structures of the BBM-1675A₁ and BBM-1675A₂ antibiotics, also
20 known as esperamicin A₁ and esperamicin A₂, respectively, have not yet been elucidated, but both components exhibit excellent antimicrobial and antitumor activity.

United States Patent No. 4,530,835, issued
25 July 23, 1985 to Bunge et al., discloses fermentation of an unidentified Actinomycete isolate WP-444 (ATCC 39363) to produce antitumor antibiotics designated CL-1577A and CL-1577B. The structures of the CL-1577 antibiotics have not yet been elucidated, but the characterizing properties given for the
30 antibiotics indicate that CL-1577A and CL-1577B are similar in structure to the BBM-1675 antibiotics, and especially BBM-1675A₁ and A₂ mentioned above in United Kingdom Patent Application No. 2,141,425.

There is disclosed by R. H. Bunge et al., in J. Antibiotics, 37(12), 1566-1571 (1984) the fermentation of Actinomadura sp. (ATCC 39363) to produce a bioactive complex from which two major components, PD 114,759 and PD 115,028, were isolated. In J. Chem. Soc. Chem. Commun., 919-920 (1985), J. H. Wilton et al. described the partial structural elucidation of the antibiotics PD 114,759 and PD 115,028. The production, isolation and characterization of the PD 114,759 and PD 115,028 antibiotics appear to be identical to the above-mentioned CL-1577A and CL-1577B antibiotics, respectively.

European Patent Application No. 95,154, published November 30, 1983, discloses fermentation of Actinomadura pulveraceus sp. nov. No. 6049 (ATCC 39100) to produce antitumor antibiotics designated WS 6049-A and WS 6049-B. The structures of the WS 6049 antibiotics have not yet been elucidated, but the characterizing properties given for the antibiotics indicate that WS 6049-A and WS 6049-B are related in structure to the BBM-1675 antibiotics of United Kingdom Patent Application No. 2,141,425 and to the CL-1577 antibiotics of United States Patent No. 4,530,835. Spectral data show, however, that neither WS 6049-A nor WS 6049-B is identical to any of the BBM-1675 components. Moreover, the producing organism described in European Patent Application No. 95,154 may be clearly differentiated from Actinomadura verrucosospora employed in United Kingdom Patent Application No. 2,141,425 in the color of its aerial mycelium on ISP Medium Nos. 2, 3 and 4, in its positive milk peptonization and in its positive utilization of D-fructose, D-mannitol, trehalose and cellulose.

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There is provided by the present invention new antitumor antibiotic substances designated herein as BBM-1675C and BBM-1675D, also known as BMY-27305 and BMY-27307, respectively, said substances being produced by selective chemical hydrolysis of the bioactive components BBM-1675A₁ (esperamicin A₁) or BBM-1675A₂ (esperamicin A₂), which are themselves produced by cultivating a

BBM-1675-producing strain of Actinomadura verrucosospora. The bioactive substances BBM-1675C and BBM-1675D may be separated and purified by conventional chromatographic procedures, and both substances exhibit excellent antimicrobial and antitumor activity.

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According to the invention there is provided the antitumor antibiotic BBM-1675C which in substantially pure form :

- (a) appears as an amorphous solid;
- 10 (b) is soluble in methanol, ethanol, ethyl acetate, acetone, tetrahydrofuran and chloroform;
- (c) exhibits in silica gel thin layer chromatography an R_f value
15 of 0.28 with the solvent system toluene:acetone (3:2, v/v);
- (d) has an apparent molecular weight of 855 as determined by high resolution FAB mass spectroscopy;
- 20 (e) has an ultraviolet absorption spectrum in methanol solution substantially as shown in FIG. 1 exhibiting ultraviolet absorption maxima and absorptivities at 210 nm ($a = 21,770$), 274 nm ($a = 9,340$) and 313 nm (shoulder) ($a = 4,190$) with no significant change upon addition of acid or base;
- 25 (f) has an infrared absorption spectrum (KBr, film) substantially as shown in FIG. 3 exhibiting principal absorption peaks at
30 540, 740, 955, 990, 1017, 1065, 1080, 1118, 1150, 1250, 1305, 1325, 1340, 1370, 1385, 1440, 1690, 1705, 1735, 2900, 2920, 2930, 2970, and 3450 reciprocal centimeters;
- 35 (g) has a low resolution mass spectrum substantially as shown in FIG. 5 exhibiting a molecular ion $[M+H]^+$ of 856;
- (h) has a 360 MHz proton magnetic resonance spectrum in $CDCl_3$ substantially as shown in FIG. 7 exhibiting signals at

6.54 (1H, dd, J=7.7, 7.0); 6.21 (1H, brs); 5.87 (1H, d, J=9.6); 5.78 (1H, dd, J=9.6, 1.5); 5.66 (1H, brd, J=2.9); 4.94 (1H, dd, J=10.3, 1.8); 4.61 (1H, d, J=7.7); 4.25 (1H, s); 4.09 (1H, q, J=2.6); 3.97 (1H, t, J=9.6); 3.92-3.53 (10H), 3.45 (1H, dt, J=10.3, 4.0); 3.37 (3H, s); 2.77 (1H, m); 2.69 (1H, dt, J=9.9, 5.2); 2.49 (1H, dd, J=10.3, 2.6); 2.48 (3H, s); 2.30 (2H, m); 2.13 (1H, m); 2.09 (3H, s); 1.90 (2H, m); 1.37 (3H, d, J=5.9); 1.32 (3H, d, J=6.3); and 1.08 (6H) parts per million downfield from tetramethylsilane;

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(i) has a 90.6 MHz carbon-13 magnetic resonance spectrum in CDCl₃ substantially as shown in FIG. 9 exhibiting signals at 13.7, 17.5, 19.8, 22.3, 22.7, 23.5, 34.2, 35.2, 39.5, 47.7, 52.7, 55.8, 56.1, 57.7, 62.4, 64.7, 67.4, 69.3, 69.8, 71.9, 76.1, 77.1, 77.7, 79.7, 83.2, 88.4, 97.3, 99.7, 123.4, 124.6, 130.1, and 193.1 parts per million downfield from tetramethylsilane.

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There is also provided the antitumor antibiotic BBM-1675D which in substantially pure form:

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- (a) appears as an amorphous solid;
- (b) is soluble in methanol, ethanol, acetone and tetrahydrofuran, and slightly soluble in chloroform;
- (c) exhibits in silica gel thin-layer chromatography an R_f value of 0.22 with the solvent system chloroform:methanol (5:0.5, v/v) and exhibits in reverse phase silica gel thin layer chromatography an R_f value of 0.37 with the solvent system methanol:water (70:30, v/v);
- (d) has an apparent molecular weight of 695 as determined by high resolution FAB mass spectroscopy;
- (e) has an ultraviolet absorption spectrum in methanol solution substantially as shown in FIG. 2 exhibiting ultraviolet absorption maxima and absorptivities at 214 nm ($a = 27,000$),

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274 nm ($a = 12,800$), and 325 nm ($a = 5,400$) with no significant change upon addition of acid or base;

- (f) has an infrared absorption spectrum (KBr, film) substantially as shown in FIG. 4 exhibiting principal absorption peaks at

735, 755, 910, 960, 1000, 1020, 1085, 1150, 1195, 1250, 1310, 1335, 1365, 1385, 1445, 1510, 1685, 1720, 1735, 2880, 2930, 2960, and 3400 reciprocal centimeters;

- (g) has a low resolution mass spectrum substantially as shown in FIG. 6 exhibiting a molecular ion $[M+H]^+$ of 696;

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- (h) has a 360 MHz proton magnetic resonance spectrum in $CDCl_3 + 10\% CD_3OD$ substantially as shown in FIG. 8 exhibiting signals at
- 6.43 (1H, dd, $J=4.4$, 10.3); 6.13 (1H, s); 5.81 (1H, d, $J=8.8$); 5.70 (1H, d, $J=8.8$); 5.48 (1H, 6 brs); 4.48 (1H, d, $J=8.1$); 4.02 (1H, d, $J=2.0$); 3.95-3.80 (solvent background); 3.77 (1H, t, $J=9.0$); 3.70-3.40 (11H, brm); 3.35 (1H, m); 3.28 (3H, s); 3.22 (3H, brs); 2.66-2.55 (2H, m); 2.38 (3H, s); 2.23-2.12 (2H, m); 1.42 (1H, brdt); 1.22 (3H, d, $J=5.9$); 0.94 (3H, d, $J=6.6$); and 0.87 (3H, d, $J=5.9$) parts per million downfield from tetramethylsilane;

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- (i) has a 90.6 MHz carbon-13 magnetic resonance spectrum in $CDCl_3 + 10\% CD_3OD$ substantially as shown in FIG. 10 (FIG. 10A + 10B) exhibiting signals at

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17.5, 21.6, 22.2, 23.0, 33.4, 39.2, 46.4, 52.3, 55.8, 62.1, 67.8, 69.8, 70.1, 71.3, 75.8, 77.1, 78.1, 82.4, 83.3, 88.2, 97.4, 99.6, 122.6, 124.8, 130.1, 130.8, 134.3, 148.7, and 192.8 parts per million downfield from tetramethylsilane.

DESCRIPTION OF THE DRAWINGS

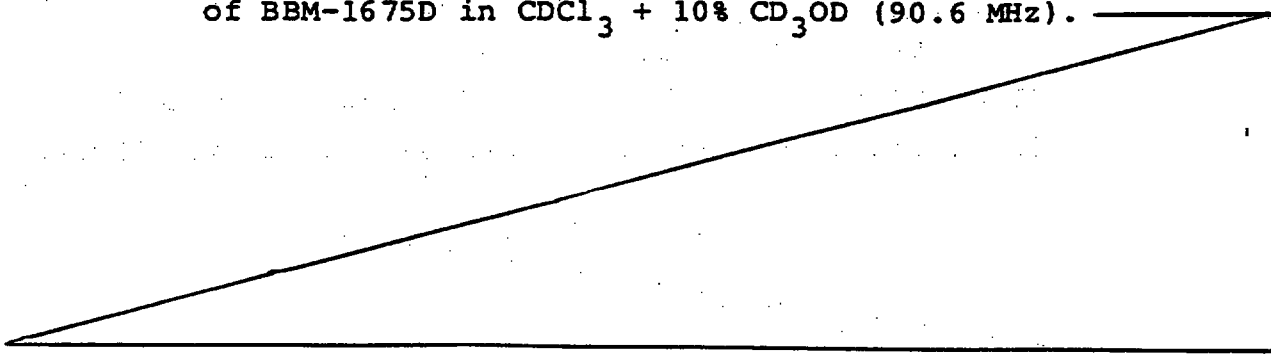
- FIG. 1 shows the ultraviolet absorption spectrum of BBM-1675C.
- FIG. 2 shows the ultraviolet absorption spectrum of BBM-1675D.
- 05 FIG. 3 shows the infrared absorption spectrum of BBM-1675C (KBr, film).
- FIG. 4 shows the infrared absorption spectrum of BBM-1675D (KBr, film).
- 10 FIG. 5 shows the relative abundance mass spectrum of BBM-1675C.
- FIG. 6 shows the relative abundance mass spectrum of
- 15 BBM-1675D.
- FIG. 7 shows the proton magnetic resonance spectrum of BBM-1675C in CDCl_3 (360 MHz).
- 20 FIG. 8 shows the proton magnetic resonance spectrum of BBM-1675D in $\text{CDCl}_3 + 10\% \text{CD}_3\text{OD}$ (360 MHz).
- FIG. 9 shows the ^{13}C magnetic resonance spectrum of BBM-1675C in CDCl_3 (90.6 MHz).
- 25 FIG. 10A shows the ^{13}C magnetic resonance spectrum (110-200 ppm) of BBM-1675D in $\text{CDCl}_3 + 10\% \text{CD}_3\text{OD}$ (90.6 MHz).
- FIG. 10B shows the ^{13}C magnetic resonance spectrum (0-110 ppm)
- 30 of BBM-1675D in $\text{CDCl}_3 + 10\% \text{CD}_3\text{OD}$ (90.6 MHz).
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FIG. 11A shows the proton magnetic resonance spectrum of compound 3A (α -anomer) in CDCl_3 (360 MHz).

FIG. 11B shows the proton magnetic resonance spectrum of compound 3B (β -anomer) in CDCl_3 (360 MHz).

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DETAILED DESCRIPTION OF THE INVENTION

This invention relates to two novel antitumor antibiotic substances designated herein as BBM-1675C and BBM-1675D, also known as BMY-27305 and BMY-27307, respectively, said substances being produced by selective chemical hydrolysis of the bioactive components BBM-1675A₁ (esperamicin A₁) or BBM-1675A₂ (esperamicin A₂), which are themselves produced by cultivating a BBM-1675-producing strain of Actinomadura verrucosospora, most preferably Actinomadura verrucosospora strain H964-92 (ATCC 39334) or Actinomadura verrucosospora strain A1327Y (ATCC 39638), or a mutant thereof. In another aspect, the present invention provides a process for producing the BBM-1675C substance by selective hydrolysis of the bioactive components BBM-1675A₁ or BBM-1675A₂. In a further aspect, the present invention provides a process for the preparation of BBM-1675D by selective hydrolysis of the BBM-1675C substance or, more preferably, from the bioactive components BBM-1675A₁ or BBM-1675A₂. The isolation and purification of BBM-1675C and BBM-1675D from the reaction mixture may be accomplished by conventional chromatographic procedures.

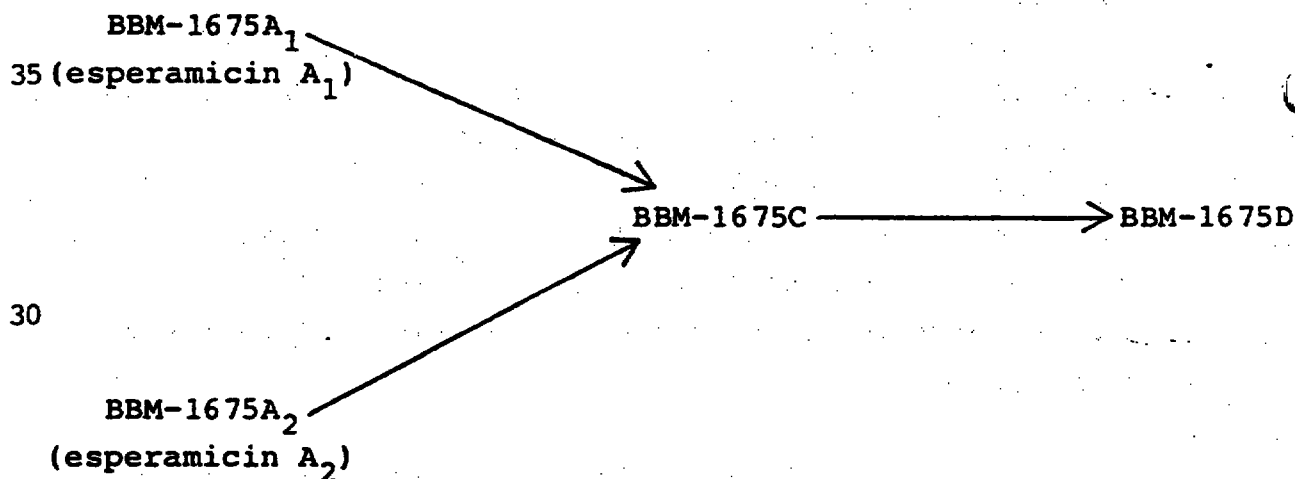
The bioactive substances BBM-1675C and BBM-1675D exhibit antimicrobial activity against a broad spectrum of microorganisms and have also been shown to exhibit inhibitory activity against various mouse tumor systems, such as P-388 leukemia and B16 melanoma. The newly described substances of the

present invention, therefore, may be used as antimicrobial agents or as antitumor agents for inhibiting mammalian tumors.

During the course of degradation studies to elucidate the structure of the antitumor antibiotics BBM-1675A₁ (esperamicin A₁) and BBM-1675A₂ (esperamicin A₂), a mixture of components were produced which lead to the isolation and identification of two inactive fragments, compounds of the Formulas 1 and 2, respectively. However, it was surprisingly found that the chemical degradation lead to the stepwise liberation of two bioactive fragments BBM-1675C and BBM-1675D. Even more surprising, it was found that the two different antibiotics BBM-1675A₁ and A₂ produced the same bioactive fragments as illustrated in Scheme 1. Still more surprising, the smaller molecular weight fragments BBM-1675C and D (having approximately 70% and 55% of the molecular weight of the parent antibiotics BBM-1675A₁ and A₂, respectively) were found to be more effective than BBM-1675A₂ and comparable to BBM-1675A₁ as antitumor and antimicrobial agents.

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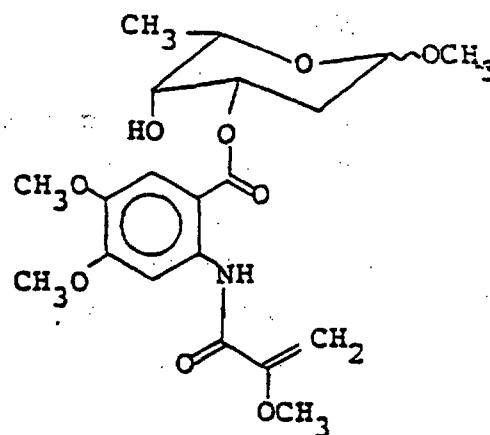
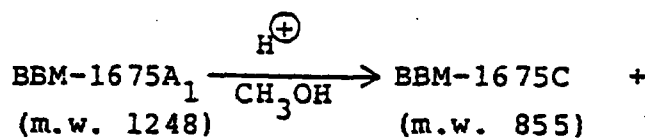
Scheme 1



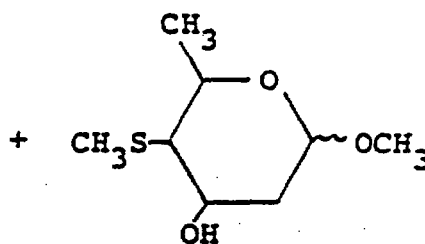
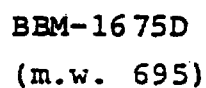
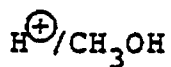
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The BBM-1675C and BBM-1675D substances may be prepared by selective chemical hydrolysis of the antibiotic BBM-1675A₁ as outlined in Scheme 2.

Scheme 2

Formula 1: Compound of
m.w. 425 which is a mixture
of α and β anomers.



Formula 3: Compound of
m.w. 192 which is a mixture
of α and β anomers.

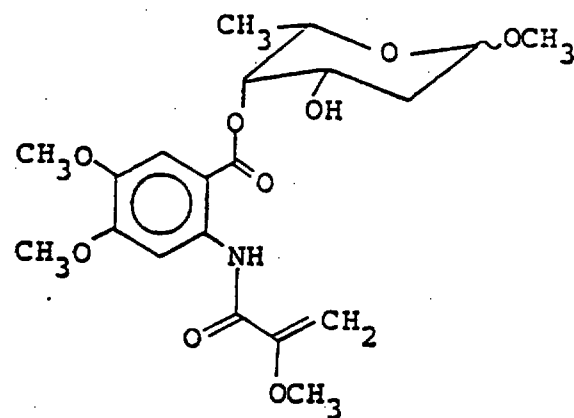
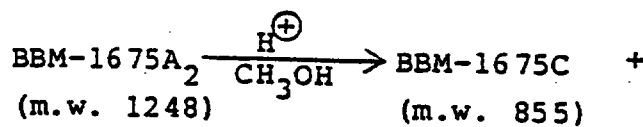
The starting BBM-1675A₁ compound is prepared according to the procedure described in United Kingdom Patent Application No. 2,141,425, published December 19, 1984. The purified BBM-1675A₁ component is hydrolyzed with a mineral or organic acid such as hydrogen chloride, sulfuric acid, p-toluenesulfonic acid, benzenesulfonic acid or the like, in an organic or mixed aqueous-organic inert solvent at a temperature of about 0°C to the refluxing temperature of the solvent until a substantial amount of the desired BBM-1675C or BBM-1675D is produced. Preferably, the hydrolysis is carried out in C₁-C₆ alcohol solvents, and most preferably, the alcoholysis is carried out in methanol. The temperature of the reaction is not critical, but it is preferred to conduct the reaction at about ambient temperature to 60°C, and most preferably from about 40° to 60°C.

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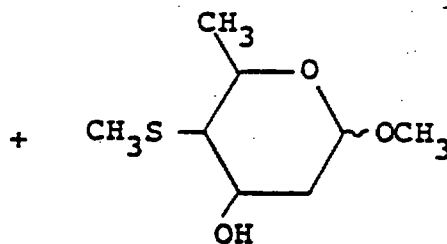
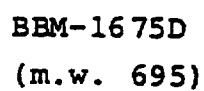
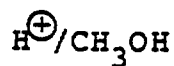
The selective hydrolysis of BBM-1675A₁ proceeds in a stepwise manner with the initial production of the BBM-1675C antibiotic and the inactive fragment of Formula 1. Subsequent or continued treatment under hydrolyzing conditions leads to the liberation of a mixture of α and β anomers of the thiosugar of Formula 3 and the production of the antibiotic BBM-1675D. It should be appreciated by those skilled in the art that altering the reaction conditions such as time, temperature and concentration of acid will produce varying relative amounts of the antibiotics BBM-1675C and D. Thus, it is desirable to monitor the progress of reaction by thin layer chromatography as described in the examples herein.

When it is desired to prepare only the BBM-1675D antibiotic, the selective hydrolysis is preferably carried out with an organic acid such as p-toluenesulfonic acid as described herein to yield a quantitative amount of BBM-1675D.

The BBM-1675C and BBM-1675D substances may also be prepared by selective chemical hydrolysis of the antibiotic BBM-1675A₂ as outlined in Scheme 3.

Scheme 3

Formula 2: Compound of m.w. 425 which is a mixture of α and β anomers.



Formula 3: Compound of m.w. 192 which is a mixture of α and β anomers.

The starting BBM-1675A₂ compound is prepared according to the procedure described in United Kingdom Patent Application No. 2,141,425, published December 19, 1984. The selective hydrolysis of purified BBM-1675A₂ likewise proceeds in a stepwise manner with the initial production of the BBM-1675C antibiotic and the inactive fragment of Formula 2. Continued treatment under hydrolyzing conditions leads to the liberation of a mixture of α and β anomers of the thiosugar of Formula 3 and the production of the antibiotic BBM-1675D.

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The reaction conditions utilized for the selective chemical hydrolysis of BBM-1675A₂ are substantially the same as those utilized for the hydrolysis of BBM-1675A₁ described above. In a manner similar to the production of BBM-1675D from BBM-1675A₁, when it is preferred to produce only the BBM-1675D antibiotic, the hydrolysis of BBM-1675A₂ is carried out until substantially all of BBM-1675A₂ and BBM-1675C is converted to BBM-1675D. Most preferably, the hydrolysis is carried out with an organic acid such as p-toluenesulfonic acid.

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The discovery, as described herein, that the same BBM-1675C and D antibiotics are produced from two different antibiotics BBM-1675A₁ and BBM-1675A₂ with the concurrent loss of two inactive fragments of Formulas 1 and 2, respectively, and the thiosugar of Formula 3, provides an additional advantage for the present invention. Accordingly, in a further aspect of the present invention, there is provided a process for the selective hydrolysis of a mixture of BBM-1675A₁ and A₂ to produce BBM-1675C and D as illustrated in Scheme 4.

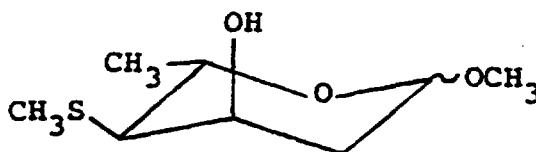
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Scheme 4



This advantage becomes apparent when one considers that the relative amounts of BBM-1675A₁ and A₂ produced in the fermentation process is subject to variability. The production of BBM-1675C and D is therefore independent of the relative amounts of BBM-1675A₁ and A₂ utilized as starting material in the present invention.

As described herein, the hydrolysis of the BBM-1675A₁, A₂ and C antibiotics results in the release of an inactive thiosugar fragment. The said thiosugar was isolated to provide further information into the chemical structure of the BBM-1675C antibiotic and hence, for the BBM-1675A₁ and A₂ antibiotics. The compound of Formula 3 was identified as a mixture of α and β anomers of a thiosugar which has the structure illustrated in Schemes 2 and 3. Further characterization was made possible when the products of the alcoholysis, the α and β anomers, were separated. The proton magnetic resonance spectra (360 MHz) of the compound 3A (α -anomer) and compound 3B (β -anomer) are shown in FIGS. 11A and 11B, respectively. From an analysis of the spectral data, the thiosugar methyl glycosides of Formula 3 were tentatively assigned the relative stereochemistry of the formula



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At the present time, the absolute stereochemistry, i.e. D or L, has not yet been determined. Accordingly, based on the present interpretation of the spectral data, it is concluded that the thiosugar of Formula 3 (less the CH₃ group from the anomeric methoxy which is incorporated during the methanolysis) is a component in the structure of the antibiotic BBM-1675C and furthermore, is a component in the structure of the starting BBM-1675A₁ and A₂ antibiotics.

Physico-chemical Properties of BBM-1675C

Description: amorphous solid

- 5 Ultraviolet absorption spectrum: See FIG. 1
 Instrument : Hewlett-Packard 8458
 Solvent : methanol
 Concentration : 0.0155 g/l

10	<u>λ_{max} (nm)</u>	<u>absorptivities</u>
	210	21,770
	274	9,340
	313 sh (shoulder)	4,190

15 No significant change is observed with acid or base.

- Infrared absorption spectrum : See FIG. 3
 Instrument : Nicolet 5DX FT-IR
 20 Major absorption bands (KBr, film):
 540, 740, 955, 990, 1017, 1065, 1080, 1118,
 1150, 1250, 1305, 1325, 1340, 1370, 1385,
 1440, 1690, 1705, 1735, 2900, 2920, 2930,
 2970, 3450 cm^{-1} .

- 25 Mass spectrum : See FIG. 5
 Instrument : Finigan 4500 TSQ
 Method : fast atom bombardment (FAB)
 ionization

30	<u>Matrix</u>	<u>m/z</u>	<u>Molecular Ion</u>	<u>Relative Abundance</u>
	glycerol	856	$[\text{M}+\text{H}]^+$	100%
35	glycerol + NaCl	876	$[\text{M}+\text{Na}]^+$	100%
	dithiothreitol:dithioerythritol (3:1) (w:w)	856	$[\text{M}+\text{H}]^+$	100%

Instrument : Kratos MS-50
High resolution FAB (m/z): $[M+H]^+ = 856.3362$

Molecular weight: apparent MW = 855

5 (based on above-described mass spectral data)

Elemental composition: $C_{36}H_{61}N_3O_{14}S_3$
(based on above-described high resolution data)

10 Proton Magnetic Resonance Spectrum: See FIG. 7

Instrument: WM 360 Bruker

Solvent : $CDCl_3$

1H NMR 360 MHz δ (ppm):

6.54 (1H, dd, $J=7.7, 7.0$); 6.21 (1H, brs); 5.87 (1H, d,
15 $J=9.6$); 5.78 (1H, dd, $J=9.6, 1.5$); 5.66 (1H, brd,
 $J=2.9$); 4.94 (1H, dd, $J=10.3, 1.8$); 4.61 (1H, d,
 $J=7.7$); 4.25 (1H, s); 4.09 (1H, q, $J=2.6$); 3.97 (1H, t,
 $J=9.6$); 3.92-3.53 (10H), 3.45 (1H, dt, $J=10.3, 4.0$);
3.37 (3H, s); 2.77 (1H, m); 2.69 (1H, dt, $J=9.9, 5.2$);
20 2.49 (1H, dd, $J=10.3, 2.6$); 2.48 (3H, s); 2.30 (2H, m);
2.13 (1H, m); 2.09 (3H, s); 1.50 (2H, m); 1.37 (3H, d,
 $J=5.9$); 1.32 (3H, d, $J=6.3$); 1.08 (6H).

13C Magnetic Resonance Spectrum: See FIG. 9

25 Instrument: WM 360 Bruker

Solvent : $CDCl_3$

^{13}C NMR 90.6 MHz δ (ppm):

13.7, 17.5, 19.8, 22.3, 22.7, 23.5, 34.2, 35.2, 39.5,
47.7, 52.7, 55.8, 56.1, 57.7, 62.4, 64.7, 67.4, 69.3,
30 69.8, 71.9, 76.1, 77.1, 77.7, 79.7, 83.2, 88.4, 97.3,
99.7, 123.4, 124.6, 130.1, 193.1.

Physico-chemical Properties of BBM-1675D

Description: amorphous solid

5 Ultraviolet absorption spectrum: See FIG. 2

Instrument : Hewlett-Packard 8458

Solvent : methanol

Concentration : 0.01 g/l

10	<u>λ_{max} (nm)</u>	<u>absorptivities</u>
	214	27,000
	274	12,800
	325	5,400

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No significant change is observed with acid or base.

Infrared absorption spectrum: See FIG. 4

Instrument : Nicolet 5DX FT-IR

20

Major absorption bands (KBr, film):

735, 755, 910, 960, 1000, 1020, 1085, 1150, 1195,
1250, 1310, 1335, 1365, 1385, 1445, 1510, 1685,
1720, 1735, 2880, 2930, 2960, 3400 cm^{-1} .

25 Mass spectrum

: See FIG. 6

Instrument : Finigan 4500 TSQ

Method : fast atom bombardment (FAB)
ionization

Matrix : thioglycerol

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Molecular ion (m/z): $[\text{M}+\text{H}]^+ = 696$

Relative abundance : 100%

Instrument : Kratos MS-50

High resolution FAB (m/z): $[\text{M}+\text{H}]^+ = 696.2794$

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Molecular weight: apparent MW = 695

(based on above-described mass spectral data)

Elemental composition: $C_{29}H_{49}N_3O_{12}S_2$
 (based on above-described high resolution data)

5 Correlation of $[M+H]^+$ and $[(M+H)+2]^+$ relative
 abundances to their calculated values
 confirms the elemental composition derived
 from high resolution-FAB measurements.

Proton magnetic resonance spectrum: See FIG. 8

10 Instrument : WM 360 Bruker
 Solvent : $CDCl_3$ + 10% CD_3OD
 1H NMR 360 MHz δ (ppm):

6.43 (1H, dd, $J=4.4$, 10.3); 6.13 (1H, s); 5.81 (1H, d,
 $J=8.8$); 5.70 (1H, d, $J=8.8$); 5.48 (1H, 6 brs); 4.48
 15 (1H, d, $J=8.1$); 4.02 (1H, d, $J=2.0$); 3.95-3.80 (solvent
 background); 3.77 (1H, t, $J=9.0$); 3.70-3.40 (11H, brm);
 3.35 (1H, m); 3.28 (3H, s); 3.22 (3H, brs); 2.66-2.55
 (2H, m); 2.38 (3H, s); 2.23-2.12 (2H, m); 1.42 (1H,
 brdt); 1.22 (3H, d, $J=5.9$); 0.94 (3H, d, $J=6.6$); 0.87
 20 (3H, d, $J=5.9$).

^{13}C Magnetic resonance spectrum: See FIG. 10A and 10B

Instrument : WM 360 Bruker
 Solvent : $CDCl_3$ + 10% CD_3OD
 25 ^{13}C NMR 90.6 MHz δ (ppm):

17.5, 21.6, 22.2, 23.0, 33.4, 39.2, 46.4, 52.3, 55.8,
 62.1, 67.8, 69.8, 70.1, 71.3, 75.8, 77.1, 78.1, 82.4,
 83.3, 88.2, 97.4, 99.6, 122.6, 124.8, 130.1, 130.8,
 134.3, 148.7, 192.8.

Biological Properties of BBM-1675 Substances

Antimicrobial activity of the BBM-1675 substances was determined for a variety of gram-positive and gram-negative 5 microorganisms. Table 1 below provides data in the form of results of an antimicrobial screening procedure involving the parent BBM-1675A₁ component and the BBM-1675C and BBM-1675D substances of the present invention. In the screening procedure, each test compound at a uniform concentration of 10 µg/ml of 10 solution impregnated on a paper strip was placed on the growth culture, and the measure of antibiotic activity is the resulting zone of inhibition from the paper strip. As shown in Table 1, the BBM-1675C and D substances showed a broad spectrum of antimicrobial activity which were at least as effective as the 15 BBM-1675A₁ component; and in particular, the BBM-1675C and D substances were more effective as inhibitors of gram-negative organisms.

TABLE I

ANTIMICROBIAL ACTIVITY OF BBM-1675 SUBSTANCES

5	<u>Test Microorganism</u>	<u>Zone of Inhibition, mm</u>		
		<u>BBM-1675A₁</u>	<u>BBM-1675C</u>	<u>BBM-1675D</u>
	<u>Escherichia coli</u> AS 19	22	52	51
	<u>Escherichia coli</u> K 12	13	36	35
10	<u>Escherichia coli</u> P 1373	12	34	33
	<u>Escherichia coli</u> R Azaserine	14	35	34
	<u>Escherichia coli</u> R Netropsin	11	32	32
	<u>Escherichia coli</u> R Mitomycin C	12	35	34
	<u>Escherichia coli</u> R Bleomycin	16	38	36
15	<u>Escherichia coli</u> R Daunomycin	19	45	44
	<u>Escherichia coli</u> R Neomycin	24	53	52
	<u>Escherichia coli</u> R Sibiromycin	14	32	30
	<u>Escherichia coli</u> R Hedamycin	14	30	25
	<u>Escherichia coli</u> R Aclacinomycin	15	41	40
20	<u>Bacillus subtilis</u> ATCC 6633	34	43	41
	<u>Klebsiella pneumoniae</u>	17	35	35
	<u>Staphylococcus</u> 209 P	32	47	44
	<u>Staphylococcus</u> R Actinoleukin	33	35	33
	<u>Staphylococcus</u> R Streptonigrin	37	50	48
25	<u>Staphylococcus faecalis</u> P1377	30	39	38
	<u>Streptococcus aureus</u> Smith P	36	47	45
	<u>Staphylococcus aureus</u> Smith R	40	55	53
	Actinomycin D			
	<u>Staphylococcus aureus</u> Smith R	17	32	31
30	Aureolic acid			
	<u>Acinetobacter</u>	16	33	32
	<u>Micrococcus luteus</u>	35	57	55
	<u>Saccharomyces cerevisiae</u> petite	22	42	43

35 R = resistant to named antibiotic

Activity Against P-388 Leukemia

Tables II and III contain the results of laboratory tests with CDF₁ mice implanted intraperitoneally with a tumor inoculum of 10⁶ ascites cells of P-388 leukemia and treated with various doses of BBM-1675A₁, C or D. The substances were administered by intraperitoneal injection. Groups of six mice were used for each dosage amount, and they were treated with a single dose of the substance on the day after inoculation. A group of ten saline treated control mice was included in each series of experiments. The BBM-1675A₁ treated group in Table III was included as a direct comparison. A 30-day protocol was employed with the mean survival time in days being determined for each group of mice and the number of survivors at the end of the 5-day period being noted. The mice were weighed before treatment and again on day four. The change in weight was taken as a measure of drug toxicity. Mice weighing 20 grams each were employed, and a loss in weight of up to approximately 2 grams was not considered excessive. The vehicle treated control animals usually died within nine days. The results were determined in terms of a % T/C which is the ratio of the mean survival time of the treated group to the mean survival time of the vehicle treated control group times 100. An effect in terms of % T/C equal to or greater than 125 indicates that a significant antitumor effect was achieved. The screening results in Table II show the initially unexpected level of antitumor activity of the BBM-1675C substance. In Table III, the results of a direct comparison of BBM-1675A₁ (esperamicin A₁) and the BBM-1675C and BBM-1675D substances are reported. The data suggest that BBM-1675C is about comparable to BBM-1675A₁ in potency and antitumor effectiveness and that it is not schedule dependent, while BBM-1675D is only slightly less effective.

Additionally, it is reported in the present invention that the same substances BBM-1675C and BBM-1675D can also be obtained from the BBM-1675A₂ (esperamicin A₂) component. In comparison of the data reported herein for BBM-1675C and BBM-1675D and the data reported in published U.K. Patent

Application No. 2,141,425 for the BBM-1675A₂ component, it is surprisingly found that the substances BBM-1675C and D are more effective as antitumor agents than the parent BBM-1675A₂ component from which they were derived.

5

TABLE II

10

EFFECT OF BBM-1675C ON P-388 LEUKEMIA

(Day 1 Treatment)

15	<u>Compound</u>	<u>Dose, IP</u> <u>mg/kg/inj.</u>	<u>MST</u> <u>days</u>	<u>Effect</u>	<u>AWC</u>	<u>Survivors</u> <u>Day 5</u>
				<u>MST</u> <u>% T/C</u>	<u>gm</u> <u>Day 4</u>	
20	BBM-1675C	3.2	TOX	TOX	-	0/6
		0.8	TOX	TOX	-	0/6
		0.2	TOX	TOX	-	0/6
		0.05	TOX	TOX	-1.8	1/6
		0.0125	11.0	122	-2.5	5/6
		0.003125	13.5	150	-2.5	6/6
25	Vehicle	-	9.0	100	0.4	10/10

Tumor inoculum: 10^6 ascites cells implanted i.p.

Host: CDF₁ male mice

Evaluation: MST = median survival time

30 Effect: % T/C = (MST treated/MST control) x 100

Criteria: % T/C $\geq$ 125 considered significant antitumor activity

AWC: average weight change (treated-control) in grams (on day 4)

TABLE III

EFFECT OF BBM-1675 SUBSTANCES ON P-388 LEUKEMIA

5	Compound	Treatment Schedule	Dose, IP mg/kg/inj.	MST days	Effect MST & T/C	AWC gm Day 4	Survivors Day 5
10	BBM-1675A ₁	d. 1	0.0512	TOX	TOX	-	0/6
			0.0256	TOX	TOX	-	0/6
			0.0128	TOX	TOX	-1.8	3/6
			0.0064	15.5	172	-0.3	6/6
			0.0032	15.0	167	-0.6	6/6
			0.0016	15.5	172	0.6	6/6
			0.0008	12.5	139	0.3	6/6
			0.0004	12.0	133	1.4	6/6
			0.0002	11.0	122	0.8	6/6
			0.0001	11.5	128	1.4	6/6
20	BBM-1675C	d. 1	0.0256	TOX	TOX	-	0/6
			0.0128	TOX	TOX	-0.8	3/6
			0.0064	11.5	128	-0.3	6/6
			0.0032	14.5	161	-0.1	6/6
			0.0016	10.5	117	0.0	6/6
			0.0008	12.0	133	0.3	6/6
			0.0004	11.5	128	0.8	6/6
			0.0002	11.0	122	1.4	6/6
			0.0001	11.0	122	0.8	6/6
			0.00005	10.5	117	1.3	6/6
25	BBM-1675C	d. 1	0.0256	TOX	TOX	-	0/6
			0.0128	TOX	TOX	-0.8	3/6
			0.0064	11.5	128	-0.3	6/6
			0.0032	14.5	161	-0.1	6/6
			0.0016	10.5	117	0.0	6/6
			0.0008	12.0	133	0.3	6/6
			0.0004	11.5	128	0.8	6/6
			0.0002	11.0	122	1.4	6/6
			0.0001	11.0	122	0.8	6/6
			0.00005	10.5	117	1.3	6/6

TABLE III (cont.)

<u>Compound</u>	<u>Treatment Schedule</u>	<u>Dose, IP mg/kg/inj.</u>	<u>MST days</u>	<u>MST % T/C</u>	<u>gm Day 4</u>	<u>Survivors Day 5</u>
5						
BBM-1675D	d. 1	0.0256	9.0	100	0.1	6/6
		0.0128	11.5	128	0.3	6/6
		0.0064	12.5	139	0.3	6/6
		0.0032	12.0	133	0.1	6/6
		0.0016	11.5	128	0.8	6/6
10						
		0.0008	10.0	111	0.2	6/6
		0.0004	10.0	111	0.5	6/6
		0.0002	9.5	106	1.7	6/6
		0.0001	9.5	106	1.7	6/6
		0.00005	9.0	100	2.0	6/6
15						
BBM-1675C	d. 1→5	0.0032	16.0	178	-1.3	6/6
		0.0016	13.5	150	-1.0	6/6
		0.0008	13.5	150	-0.3	6/6
		0.0004	12.0	133	-0.4	6/6
		0.0002	12.0	133	-0.4	6/6
20						
		0.0001	11.0	122	-0.4	5/6
		0.00005	11.0	122	0.9	6/6
		0.000025	8.5	94	2.2	6/6
		0.0000125	8.0	89	2.4	6/6
		0.00000625	8.0	89	2.4	6/6
25						
Vehicle		-	9.0	100	2.4	10/10

30 Tumor inoculum: 10^6 ascites cells implanted i.p.

Host: CDF₁ female mice

Evaluation: MST = median survival time

Effect: % T/C = (MST treated/MST control) x 100

Criteria: % T/C $\geq$ 125 considered significant antitumor activity

35 AWC: average weight change (treated-control) in grams (on day 4)

Activity Against B16 Melanoma

Table IV contains results of antitumor tests using the B16 melanoma grown in mice. BDF₁ mice were employed and inoculated subcutaneously with the tumor implant. A 60-day protocol was used. Groups of ten mice were used for each dosage amount tested, and the mean survival time for each group was determined. Control animals inoculated in the same way as the test animals and treated with the injection vehicle and no drug exhibited a mean survival time of 22.5 days. For each dosage level, the test animals were treated with the test compound on days 1, 5 and 9 by intraperitoneal injection. An effect in terms of % T/C equal to or greater than 125 indicates that a significant antitumor effect was achieved. The results in Table IV show that in a direct comparison BBM-1675C was also effective in treatment of mice bearing B16 melanoma and was about comparable to BBM-1675A₁ in potency.

TABLE IVEFFECT OF BBM-1675 SUBSTANCES ON B16 MELANOMA

5

(Day 1, 5 and 9 Treatments)

<u>Compound</u>	<u>Dose, IP</u> <u>mg/kg/inj.</u>	<u>MST</u> <u>days</u>	<u>Effect</u>	<u>AWC</u>	<u>Survivors</u> <u>Day 10</u>
			<u>MST</u> <u>% T/C</u>	<u>gm</u> <u>Day 12</u>	
10					
BBM-1675A ₁	0.0032	37.5	167	0.3	10/10
	0.0016	37.5	167	0.3	10/10
	0.0008	38.5	171	1.4	10/10
	0.0004	37.0	164	1.8	10/10
15					
	0.0002	34.5	153	2.0	10/10
	0.0001	32.0	142	1.9	10/10
BBM-1675C	0.0008	31.5	140	0.6	10/10
	0.0004	37.0	164	1.2	10/10
20					
	0.0002	31.0	138	0.6	10/10
	0.0001	31.5	140	1.0	10/10
	0.00005	27.5	122	0.8	10/10
	0.000025	25.0	111	0.5	10/10
25Vehicle	-	22.5	100	0.3	10/10

Tumor inoculum: 0.5 ml of a 10% brei, IP

Host: BDF₁ female mice

Evaluation: MST = median survival time

30 Effect: % T/C = (MST treated/MST control) x 100

Criteria: % T/C $\geq$ 125 considered significant antitumor activity

AWC: average weight change (treated-control) in grams (on day 12)

As indicated by the antimicrobial and mouse tumor data provided above, BBM-1675C and BBM-1675D are thus useful as antibiotics in the therapeutic treatment of mammals and other animals for infectious diseases and also as antitumor agents for 5 therapeutically inhibiting the growth of mammalian tumors.

The present invention, therefore, provides a method for therapeutically treating an animal host affected by a microbial infection or by a malignant tumor which comprises administering 10 to said host an effective antimicrobial or tumor-inhibiting dose of BBM-1675C or BBM-1675D, or a pharmaceutical composition thereof.

The invention includes within its scope pharmaceutical 15 compositions containing an effective antimicrobial or tumor-inhibiting amount of BBM-1675C or BBM-1675D in combination with an inert pharmaceutically acceptable carrier or diluent. Such compositions may also contain other active antimicrobial or antitumor agents and may be made up in any pharmaceutical form 20 appropriate for the desired route of administration. Examples of such compositions include solid compositions for oral administration such as tablets, capsules, pills, powders and granules, liquid compositions for oral administration such as solutions, suspensions, syrups or elixirs and preparations for parenteral 25 administration such as sterile aqueous or non-aqueous solutions suspensions or emulsions. They may also be manufactured in the form of sterile solid compositions which can be dissolved in sterile water, physiological saline or some other sterile injectable medium immediately before use.

30

For use as an antimicrobial agent, the BBM-1675C or BBM-1675D, or a pharmaceutical composition thereof is administered so that the concentration of active ingredient is greater than the minimum inhibitory concentration for the particular 35 organism being treated. For use as an antitumor agent, optimal dosages and regimens of BBM-1675C or BBM-1675D for a given mammalian host can be readily ascertained by those skilled in the art. It will, of course, be appreciated that the actual dose of

BBM-1675C or BBM-1675D used will vary according to the particular composition formulated, the mode of application and the particular situs, host and disease being treated. Many factors that modify the action of the drug will be taken into account including age, weight, sex, diet, time of administration, route of administration, rate of excretion, condition of the patient, drug combinations, reaction sensitivities and severity of the disease. Administration can be carried out continuously or periodically within the maximum tolerated dose. Optimal application rates for a given set of conditions can be ascertained by those skilled in the art using conventional dosage determination tests in view of the above guidelines.

The following examples are provided for illustrative purposes only and are not intended to limit the scope of the invention.

Chemical Preparation and Isolation of BBM-1675C and BBM-1675D

20

Example 1

A sample of BBM-1675A₁ (50 mg) was dissolved in 2.5 ml of methanol and treated with 2.5 ml of 0.1 molar solution of hydrogen chloride in methanol. The reaction was allowed to proceed at a temperature of about 50°C, and the disappearance of the starting material (approximately 30 minutes) was monitored every 5 to 10 minutes by thin layer chromatography (TLC) on silica gel plates (Analtech, 250 micron, GF) with toluene:acetone (3:2, v/v) as the eluting solvent. After the starting material has been consumed, the reaction mixture was neutralized with a saturated solution of NaHCO₃ in methanol, then evaporated under reduced pressure to yield a dry residue containing the bioactive fragments. The BBM-1675C substance was isolated from the residue by flash column chromatography on a 2 cm i.d. x 10 cm column packed with Woelm silica gel (52-65 micron particle size). The column was eluted with toluene:acetone (3:2, v/v) collecting 3 ml fractions. Each fraction was analyzed by TLC [silica gel with toluene:acetone (3:2, v/v) as eluent], and the TLC spots were

visualized with a UV 254 nm light source and a ceric sulfate spray (1% ceric sulfate and 2.5% molybdic acid in 10% sulfuric acid). Fractions 6-12 (R_f value for BBM-1675C is 0.28) were pooled and evaporated to dryness to yield 12 mg (35%) of substantially pure BBM-1675C.

The physico-chemical properties of BBM-1675C appear in the specification and the ultraviolet, infrared, mass, ^1H NMR and ^{13}C NMR spectra of the compound appear as Figures 1, 3, 5, 7 and 9, respectively.

Example 2

When the reaction time of the procedure in Example 1 is extended, the amount of BBM-1675C decreases, and two new products denoted as compound 3 ($R_f = 0.65$) and BBM-1675D (R_f remains at baseline) [TLC: silica, toluene:acetone (3:2, v/v)] appear and become more prominent with time.

Compound BBM-1675D which usually accompanies the production of BBM-1675C was isolated from the chromatographic column described in Example 1 by eluting the column with chloroform:methanol (5:1, v/v). The appropriate fractions were pooled and evaporated to dryness to yield 18 mg of substantially pure BBM-1675D from the reaction described in Example 1.

The BBM-1675D substance exhibits one major spot at $R_f = 0.37$ in reverse phase TLC (Whatman MKC₁₈F, 200 micron (Whatman is a Registered Trade Mark)) using 30% water in methanol as the eluent and $R_f = 0.22$ in normal phase silica gel TLC using chloroform:methanol (5:0.5, v/v) as the eluent.

Example 3

Substantial improvement in the yield of BBM-1675D can be achieved by using p-toluenesulfonic acid in place of hydrogen chloride in the chemical hydrolysis of BBM-1675A₂ or BBM-1675A₁.

as illustrated by the procedures of Examples 3 and 5, respectively.

A sample of BBM-1675A₂ (15.2 mg) was hydrolyzed with 0.03 molar solution of p-toluenesulfonic acid in methanol (1 ml) at a temperature of about 63°C for about one hour. The reaction mixture was then evaporated to dryness under reduced pressure at about 30°C. The BBM-1675D substance was isolated from the dry residue by flash column chromatography on a column packed with Woelm silica gel (32-63 micron particle size). The column was eluted with chloroform:methanol (5:0.5, v/v), and the collected fractions were analyzed by TLC [silica gel with chloroform:methanol (5:0.5, v/v) as eluent]. The applied chromatography conditions permitted the separation of the mixture of inactive compounds 2 and 3 (7 mg) from the bioactive BBM-1675D substance which has an R_f value of 0.22. The appropriate fractions were pooled and evaporated to dryness to yield 8 mg of substantially pure BBM-1675D in near quantitative yield.

The physico-chemical properties of BBM-1675D appear in the specification and the ultraviolet, infrared, mass, ¹H NMR and ¹³C NMR spectra of the compound appear as Figures 2, 4, 6, 8 and combined 10A and 10B, respectively.

25

Example 4

A sample of BBM-1675A₂ (40 mg) was treated with 5 ml of an 0.5 molar solution of hydrogen chloride in methanol at about 50°C for about 2 hours according to the general procedure and isolation method described in Example 1. After neutralization with NaHCO₃ and evaporation to dryness, the BBM-1675C substance was isolated from the residue by flash column chromatography on a column packed with Woelm silica gel (32-63 micron particle size) using toluene:acetone (3:2, v/v) as the eluent. The appropriate fractions were combined and evaporated to dryness to yield 8.4 mg of substantially pure BBM-1675C which is identical to the product isolated in Example 1.

The chromatographic column of above was then eluted with chloroform:methanol (5:0.25, v/v) and the fractions collected were pooled and evaporated to dryness to yield BBM-1675D. The BBM-1675D substance was further purified by an additional flash chromatography column with silica gel utilizing chloroform:methanol (5:0.5, v/v) as the eluent. The appropriate fractions were combined and evaporated to dryness to yield 6.3 mg of substantially pure BBM-1675D which is identical to the product isolated in Example 3.

10

Example 5

A sample of BBM-1675A₁ (49.3 mg) was hydrolyzed with 0.037 M solution of p-toluenesulfonic acid in methanol (1.5 ml) at a temperature of about 60°C for about 1.5 hours. The reaction mixture was evaporated to dryness under reduced pressure at about 30°C to give a residue which contains BBM-1675D and the inactive compounds 1 and 3. The BBM-1675D bioactive substance was isolated from the residue by flash column chromatography on a column packed with Woelm silica gel (32-63 micron particle size) utilizing chloroform:methanol (5:0.25, v/v) as the eluent. The appropriate fractions were combined and evaporated to dryness to yield 27 mg of substantially pure BBM-1675D which is identical to the product isolated in Example 3.

25

Example 6

A sample of BBM-1675C (5.1 mg) was hydrolyzed with 0.5 molar solution of hydrogen chloride in methanol (1 ml) at about 40-50°C overnight. After neutralization with NaHCO₃ and evaporation to dryness, the BBM-1675D bioactive substance was isolated from the residue by flash column chromatography on a column packed with Woelm silica gel (32-63 micron particle size) utilizing chloroform:methanol (5:0.25, v/v) as the eluent. The appropriate fractions yielded substantially pure BBM-1675D which is identical to the product isolated in Example 3.

35

Example 7

When the general procedure of Examples 1 and 2 are repeated, except that the starting material BBM-1675A₁ is re-
5 placed by an equimolar amount of a mixture containing BBM-1675A₁ and BBM-1675A₂, there is thereby produced the BBM-1675C and BBM-1675D substances.

Example 8

10

When the general procedure of Example 5 is repeated, except that the starting material BBM-1675A₁ is replaced by an equimolar amount of a mixture containing BBM-1675A₁ and BBM-1675A₂, there is thereby produced the BBM-1675D substance.

C L A I M S

1. The antitumor antibiotic BBM-1675C which in substantially pure form:
- (a) appears as an amorphous solid;
 - 5 (b) is soluble in methanol, ethanol, ethyl acetate, acetone, tetrahydrofuran and chloroform;
 - (c) exhibits in silica gel thin layer chromatography an R_f value
10 of 0.28 with the solvent system toluene:acetone (3:2, v/v);
 - (d) has an apparent molecular weight of 855 as determined by high resolution FAB mass spectroscopy;
 - 15 (e) has an ultraviolet absorption spectrum in methanol solution substantially as shown in FIG. 1 exhibiting ultraviolet absorption maxima and absorptivities at 210 nm ($a = 21,770$), 274 nm ($a = 9,340$) and 313 nm (shoulder) ($a = 4,190$) with no significant change upon addition of acid or base;
 - 20 (f) has an infrared absorption spectrum (KBr, film) substantially as shown in FIG. 3 exhibiting principal absorption peaks at
25 540, 740, 955, 990, 1017, 1065, 1080, 1118, 1150, 1250, 1305, 1325, 1340, 1370, 1385, 1440, 1690, 1705, 1735, 2900, 2920, 2930, 2970, and 3450 reciprocal centimeters;
 - 30 (g) has a low resolution mass spectrum substantially as shown in FIG. 5 exhibiting a molecular ion $[M+H]^+$ of 856;
 - (h) has a 360 MHz proton magnetic resonance spectrum in $CDCl_3$ substantially as shown in FIG. 7 exhibiting signals at



6.54 (1H, dd, J=7.7, 7.0); 6.21 (1H, brs); 5.87 (1H, d, J=9.6); 5.78 (1H, dd, J=9.6, 1.5); 5.66 (1H, brd, J=2.9); 4.94 (1H, dd, J=10.3, 1.8); 4.61 (1H, d, J=7.7); 4.25 (1H, s); 4.09 (1H, q, J=2.6); 3.97 (1H, t, J=9.6); 3.92-3.53 (10H), 3.45 (1H, dt, J=10.3, 4.0); 3.37 (3H, s); 2.77 (1H, m); 2.69 (1H, dt, J=9.9, 5.2); 2.49 (1H, dd, J=10.3, 2.6); 2.48 (3H, s); 2.30 (2H, m); 2.13 (1H, m); 2.09 (3H, s); 1.50 (2H, m); 1.37 (3H, d, J=5.9); 1.32 (3H, d, J=6.3); and 1.08 (6H) parts per million downfield from tetramethylsilane;

10

(i) has a 90.6 MHz carbon-13 magnetic resonance spectrum in CDCl₃ substantially as shown in FIG. 9 exhibiting signals at 13.7, 17.5, 19.8, 22.3, 22.7, 23.5, 34.2, 35.2, 39.5, 47.7, 52.7, 55.8, 56.1, 57.7, 62.4, 64.7, 67.4, 69.3, 69.8, 71.9, 76.1, 77.1, 77.7, 79.7, 83.2, 88.4, 97.3, 99.7, 123.4, 124.6, 130.1, and 193.1 parts per million downfield from tetramethylsilane.

15

2. The antitumor antibiotic BBM-1675D which in substantially pure form:

20

(a) appears as an amorphous solid;

25

(b) is soluble in methanol, ethanol, acetone and tetrahydrofuran, and slightly soluble in chloroform;

30

(c) exhibits in silica gel thin-layer chromatography an R_f value of 0.22 with the solvent system chloroform:methanol (5:0.5, v/v) and exhibits in reverse phase silica gel thin layer chromatography an R_f value of 0.37 with the solvent system methanol:water (70:30, v/v);

35

(d) has an apparent molecular weight of 695 as determined by high resolution FAB mass spectroscopy;

(e) has an ultraviolet absorption spectrum in methanol solution substantially as shown in FIG. 2 exhibiting ultraviolet absorption maxima and absorptivities at 214 nm (a = 27,000),



274 nm ($a = 12,800$), and 325 nm ($a = 5,400$) with no significant change upon addition of acid or base;

- (f) has an infrared absorption spectrum (KBr, film) substantially as shown in FIG. 4 exhibiting principal absorption peaks at

735, 755, 910, 960, 1000, 1020, 1085, 1150, 1195, 1250, 1310, 1335, 1365, 1385, 1445, 1510, 1685, 1720, 1735, 2880, 2930, 2960, and 3400 reciprocal centimeters;

- (g) has a low resolution mass spectrum substantially as shown in FIG. 6 exhibiting a molecular ion $[M+H]^+$ of 696;

- (h) has a 360 MHz proton magnetic resonance spectrum in $CDCl_3 + 10\% CD_3OD$ substantially as shown in FIG. 8 exhibiting signals at

6.43 (1H, dd, $J=4.4, 10.3$); 6.13 (1H, s); 5.81 (1H, d, $J=8.8$); 5.70 (1H, d, $J=8.8$); 5.48 (1H, 6 brs); 4.48 (1H, d, $J=8.1$); 4.02 (1H, d, $J=2.0$); 3.95-3.80 (solvent background); 3.77 (1H, t, $J=9.0$); 3.70-3.40 (11H, brm); 3.35 (1H, m); 3.28 (3H, s); 3.22 (3H, brs); 2.66-2.55 (2H, m); 2.38 (3H, s); 2.23-2.12 (2H, m); 1.42 (1H, brdt); 1.22 (3H, d, $J=5.9$); 0.94 (3H, d, $J=6.6$); and 0.87 (3H, d, $J=5.9$) parts per million downfield from tetramethylsilane;

- (i) has a 90.6 MHz carbon-13 magnetic resonance spectrum in $CDCl_3 + 10\% CD_3OD$ substantially as shown in FIG. 10 (FIG. 10A + 10B) exhibiting signals at 17.5, 21.6, 22.2, 23.0, 33.4, 39.2, 46.4, 52.3, 55.8, 62.1, 67.8, 69.8, 70.1, 71.3, 75.8, 77.1, 78.1, 82.4, 83.3, 88.2, 97.4, 99.6, 122.6, 124.8, 130.1, 130.8, 134.3, 148.7, and 192.8 parts per million downfield from tetramethylsilane.

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3. The process for the production of the antitumor antibiotic BBM-1675C, which comprises hydrolyzing BBM-1675A₁ or BBM-1675A₂ with a mineral or organic acid until a substantial



amount of BBM-1675C is produced and then recovering BBM-1675C from the reaction medium.

4. The process for the production of the antitumor anti-biotic BBM-1675D, which comprises hydrolyzing BBM-1675A₁ or BBM-1675A₂ with a mineral or organic acid until an amount of BBM-1675D is produced which is at least sufficient for recovery and then recovering BBM-1675D from the reaction medium.

5. The process for the production of the antitumor anti-biotic BBM-1675D, which comprises hydrolyzing BBM-1675C with a mineral or organic acid until an amount of BBM-1675D is produced which is at least sufficient for recovery and then recovering BBM-1675D from the reaction medium.

6. The process for the production of the antitumor anti-biotic BBM-1675C, which comprises hydrolyzing a mixture of BBM-1675A₁ and BBM-1675A₂ with a mineral or organic acid until an amount of BBM-1675C is produced which is at least sufficient for recovery and then recovering BBM-1675C from the reaction medium.

7. The process for the production of the antitumor anti-biotic BBM-1675D, which comprises hydrolyzing a mixture of BBM-1675A₁ and BBM-1675A₂ with a mineral or organic acid until an amount of BBM-1675D is produced which is at least sufficient for recovery and then recovering BBM-1675D from the reaction medium.

8. The antitumor antibiotic BBM-1675C, substantially as hereinbefore specifically described with particular reference to the examples.

9. The antitumor antibiotic BBM-1679D, substantially as hereinbefore specifically described with particular reference to the examples.

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10. A process for the production of BBM-1675C, substantially as hereinbefore specifically described with particular reference to the examples.



11. A process for the production of BBM-1675D, substantially as hereinbefore specifically described with particular reference to the examples.
- 5 12. A pharmaceutical composition comprising an effective antimicrobial amount of BBM-1675C or BBM 1675D in combination with a pharmaceutical carrier or diluent.
13. A pharmaceutical composition comprising an effective tumor-
10 inhibiting amount of BBM-1675C or BBM-1675D in combination with a pharmaceutical carrier or diluent.
14. A pharmaceutical compound as defined in claim 1 for use in a method of treatment of the human or animal body.
- 15 15. A pharmaceutical compound as defined in claim 2 for use in a method of treatment of the human or animal body.
16. A compound according to claims 1 or 2 for use as an antitumor
20 antibiotic.
17. A compound according to either of claims 1 or 2 substantially as hereinbefore specifically described in each of the examples for the use hereinbefore specifically described.
- 25 18. A pharmaceutical composition according to either of claims 12 or 13 substantially as hereinbefore specifically described in examples.
19. A compound when produced by a process as claimed in any one of
30 claims 3 to 7, 10 or 11.
20. The use of a substance according to either of claims 1 or 2 for the manufacture of a medicament which is antitumor antibiotic.

