

1 **BLENDING OF LOW VISCOSITY**
2 **FISCHER-TROPSCH BASE OILS AND**
3 **FISCHER-TROPSCH DERIVED BOTTOMS OR BRIGHT STOCK**

4
5 CROSS REFERENCE TO RELATED APPLICATIONS
6

7 This application is related to co-pending U.S. Patent Application Serial
8 No. 10/235,150 filed September 4, 2002, titled "Blending of Low Viscosity
9 Fischer-Tropsch Base Oils to Produce High Quality Lubricating Base Oils"
10 and U.S. Patent Application Serial No. 10/301,391 filed November 20, 2002,
11 titled "Blending of Low Viscosity Fischer-Tropsch Base Oils With Conventional
12 Base Oils to Produce High Quality Lubricating Base Oils".

13
14 FIELD OF THE INVENTION
15

16 The invention relates to the blending of a low viscosity Fischer-Tropsch
17 derived base oil fraction with a higher viscosity Fischer-Tropsch derived
18 bottoms fraction to produce a high quality lubricating base oil that is useful for
19 preparing commercial finished lubricants such as in crankcase engine oils.

20
21 BACKGROUND OF THE INVENTION
22

23 Finished lubricants and greases used for various applications, including
24 automobiles, diesel engines, natural gas engines, axles, transmissions, and
25 industrial applications consist of two general components, a lubricating base
26 oil and additives. Lubricating base oil is the major constituent in these finished
27 lubricants and contributes significantly to the properties of the finished
28 lubricant. In general, a few lubricating base oils are used to manufacture a
29 wide variety of finished lubricants by varying the mixtures of individual
30 lubricating base oils and individual additives.

31
32 Numerous governing organizations, including original equipment
33 manufacturers (OEM's), the American Petroleum Institute (API),

1 Association des Constructeurs d' Automobiles (ACEA),
2 the American Society of Testing and Materials (ASTM),
3 the Society of Automotive Engineers (SAE), and
4 National Lubricating Grease Institute (NLGI) among others, define the
5 specifications for lubricating base oils and finished lubricants. Increasingly, the
6 specifications for finished lubricants are calling for products with excellent
7 low temperature properties, high oxidation stability, and low volatility.
8 Currently only a small fraction of the base oils manufactured today are able to
9 meet the demanding specifications of premium lubricant products.

10

11 Syncrudes prepared from the Fischer-Tropsch process comprise a mixture of
12 various solid, liquid, and gaseous hydrocarbons. Those Fischer-Tropsch
13 products which boil within the range of lubricating base oil contain a
14 high proportion of wax which makes them ideal candidates for processing into
15 lubricating base oil stocks. Accordingly, the hydrocarbon products recovered
16 from the Fischer-Tropsch process have been proposed as feedstocks for
17 preparing high quality lubricating base oils. When the Fischer-Tropsch waxes
18 are converted into Fischer-Tropsch base oils by various processes, such as
19 hydroprocessing and distillation, the base oils produced usually fall into
20 different narrow-cut viscosity ranges. Typically, the kinematic viscosity of the
21 various cuts will range between 2.1 cSt and 12 cSt at 100 degrees C. Since
22 the kinematic viscosity of lubricating base oils typically will fall within the range
23 of from 3 to 32 cSt at 100 degrees C, the base oils that fall outside of this
24 viscosity range have limited use and, consequently, have less market value
25 for engine oils.

26

27 The Fischer-Tropsch process typically produces a syncrude mixture
28 containing a wide range of products having varying molecular weights but with
29 a relatively high proportion of the products characterized by a low molecular
30 weight and viscosity. Used by itself, this low viscosity product is not suitable
31 for many lubricant applications, especially high volume applications, such as
32 for engine oil. Currently, those Fischer-Tropsch derived base oils having
33 kinematic viscosities below 3 cSt at 100 degrees C have a limited market and

1 are usually cracked into lower molecular weight material, such as diesel and
2 naphtha. However, diesel and naphtha have a lower market value than
3 lubricating base oil. It would be desirable to be able to upgrade these low
4 viscosity base oils into products suitable for use as a lubricating base oil.

5
6 Conventional base oils prepared from petroleum derived feedstocks having a
7 kinematic viscosity below 3 cSt at 100 degrees C have a low viscosity
8 index (VI) and high volatility. Consequently, low viscosity conventional base
9 oils are unsuitable for blending with higher viscosity conventional base oils
10 because the blend will fail to meet the VI and volatility specifications for many
11 finished lubricants. Surprisingly, it has been found that Fischer-Tropsch
12 derived base oils having a kinematic viscosity above 2 and below 3 cSt at
13 100 degrees C display unusually high VI's, resulting in excellent low
14 temperature properties and volatilities similar to those seen in conventional
15 Group I and Group II light neutral base oils which have a kinematic viscosity
16 generally falling in the range of between 3.8 and 4.7 cSt at 100 degrees C.
17 Even more surprising was that when the low viscosity Fischer-Tropsch
18 derived base oils were blended with certain Fischer-Tropsch derived bottom
19 fractions or bright stock, a VI premium was observed, i.e., the VI of the blend
20 was significantly higher than would have been expected from a mere
21 averaging of the VI's for the two fractions. Consequently, it is has been
22 discovered that the low viscosity Fischer-Tropsch base oils fractions may be
23 advantageously employed as blending stock to prepare premium lubricants.

24
25 While Fischer-Tropsch derived lubricating base oil blends have been
26 described in the prior art, the method used to prepare them and the properties
27 of the prior art blends differ from the present invention. See, for example,
28 U.S. Patent Nos. 6,332,974; 6,096,940; 4,812,246; and 4,906,350. It has not
29 been previously taught that Fischer-Tropsch fractions having a viscosity of
30 less than 3 cSt at 100 degrees C could be used to prepare lubricating
31 base oils suitable for blending finished lubricants meeting the specifications
32 for SAE Grade 10W, and 15W multigrade engine oils; monograde engine oils,

1 automatic transmission fluids; and ISO Viscosity Grade 22, 32, and
2 46 industrial oils. With the present invention, this becomes possible.

3

4 When referring to conventional lubricating base oils this disclosure is referring
5 to conventional petroleum derived lubricating base oils produced using
6 petroleum refining processes well documented in the literature and known to
7 those skilled in the art.

8

9 As used in this disclosure the word "comprises" or "comprising" is intended as
10 an open-ended transition meaning the inclusion of the named elements, but
11 not necessarily excluding other unnamed elements. The phrase "consists
12 essentially of" or "consisting essentially of" is intended to mean the exclusion
13 of other elements of any essential significance to the composition. The phrase
14 "consisting of" or "consists of" are intended as a transition meaning the
15 exclusion of all but the recited elements with the exception of only minor
16 traces of impurities.

17

18 SUMMARY OF THE INVENTION

19

20 The present invention is directed to a process for producing a
21 Fischer-Tropsch derived lubricating base oil blend which comprises blending
22 a Fischer-Tropsch distillate fraction with a Fischer-Tropsch derived bottoms
23 fraction in the proper proportion to produce a Fischer-Tropsch derived
24 lubricating base oil characterized as having a kinematic viscosity of between
25 about 3 and about 10 cSt at 100 degrees C and a TGA Noack volatility of less
26 than about 45 weight percent wherein said distillate fraction is characterized
27 by a kinematic viscosity of about 2 cSt or greater but less than 3 cSt at
28 100 degrees C.

29

30 The Fischer-Tropsch derived bottoms fraction will generally have a kinematic
31 viscosity at 100 degrees C of not less than about 7 cSt. The Fischer-Tropsch
32 derived bottoms fraction may constitute that residual fraction remaining at the
33 bottom of the vacuum column following the fractionation of the waxy material

1 recovered directly from Fischer-Tropsch syncrude, or it may be prepared from
2 the condensate fraction by the oligomerization of the olefins present. Except
3 for bright stock, most Fischer-Tropsch bottoms fractions will have a kinematic
4 viscosity within the range of from about 9 cSt to about 20 cSt at
5 100 degrees C, preferably, between about 10 cSt and 16 cSt. However, in the
6 case of Fischer-Tropsch derived bright stock the kinematic viscosity may be
7 considerably higher. The invention makes it possible to upgrade the
8 low viscosity Fischer-Tropsch derived base oils into more valuable premium
9 lubricants which otherwise would be cracked or blended into lower value
10 transportation fuels.

11

12 Bright stock constitutes a bottoms fraction which has been highly refined and
13 dewaxed. Bright stock is a high viscosity base oil. Conventional petroleum
14 derived bright stock is named for the SUS viscosity at 210 degrees F, having
15 viscosities above 180 cSt at 40 degrees C, preferably above 250 cSt at
16 40 degrees C, and more preferably ranging from 500 to 1100 cSt at
17 40 degrees C. Fischer-Tropsch derived bright stock has a kinematic viscosity
18 between about 15 cSt and about 40 cSt at 100 degrees C. Bright stock used
19 to carry out the invention may be produced from Fischer-Tropsch derived
20 residual stocks recovered from the bottom of the vacuum column following the
21 fractionation of the waxy product separated from the syncrude from the
22 Fischer-Tropsch plant. However, Fischer-Tropsch derived bright stock may
23 also be prepared from the oligomerization of the olefins present in the
24 Fischer-Tropsch condensate recovered from the Fischer-Tropsch reactor.
25 Blending Fischer-Tropsch derived bright stock with the Fischer-Tropsch
26 derived distillate fraction produces a lubricating base oil having especially low
27 volatility, good cold flow properties, and improved oxidation stability as
28 compared to many conventional base oils.

29

30 Lubricating base oils falling within the scope of the invention are blends of at
31 least two different fractions. One fraction is a light distillate fraction and the
32 other fraction is a bottoms fraction. Accordingly, lubricating base oils of the
33 invention are distinguished from those homogeneous lubricating base oils

1 prepared from a single distillate fraction or from only a bottoms fraction.
2 Consequently, the Fischer-Tropsch lubricating base oil blends prepared using
3 the process of the present invention are unique, and will display certain
4 properties which may be used to distinguish the blends from both
5 conventional and from other Fischer-Tropsch derived lubricating base oils
6 disclosed in the prior art. For example, lubricating base oil blends prepared
7 according to the invention will have a TGA Noack volatility of greater than
8 about 12 and more generally will have a TGA Noack volatility in excess of
9 about 20. However, if the blends are intended for use as crankcase lubricating
10 oils they will preferably have a Noack volatility of less than about
11 30 weight percent. The blends also typically will display a VI of between about
12 130 and about 185 and will have very low total sulfur, usually less than about
13 5 ppm. Most noticeably, the lubricating base oils compositions of the invention
14 display unique boiling range distributions.

15

16 The boiling range distributions characteristic of the lubricating base oils
17 prepared according to the invention will depend to some extent on the
18 properties of the Fischer-Tropsch derived bottoms fraction used in preparing
19 the blend. In general, the lubricating base oils of the invention will have an
20 initial boiling point within the range of between about 550 degrees F
21 (288 degrees C) and about 625 degrees F (330 degrees C) and an end
22 boiling point between about 1000 degrees F (538 degrees C) and about
23 1400 degrees F (760 degrees C). In addition, lubricating base oils of the
24 invention will typically display a bi-modal boiling range distribution which may
25 be described as a lubricating base oil blend in which less than
26 20 weight percent of the blend will boil within the region defined by the
27 50 percent boiling points, plus or minus 30 degrees F.

28

29 All boiling range distributions in this disclosure are measured using the
30 standard analytical method D-6352 or its equivalent unless stated otherwise.
31 As used herein, a equivalent analytical method to D-6352 refers to any
32 analytical method which gives substantially the same results as the standard
33 method.

1 The Fischer-Tropsch derived lubricating base oil blends prepared according to
2 the present invention also may be blended with conventionally derived
3 lubricating base oils, such as conventional neutral Group I and Group II
4 lubricating base oils. When the Fischer-Tropsch derived lubricating base oil is
5 blended with a conventional neutral Group I or Group II base oil, the
6 conventional base oil will typically comprise between about 40 weight percent
7 and about 90 weight percent of the total blend, with from about
8 40 weight percent to about 70 weight percent being preferred.

9
10 The Fischer-Tropsch derived lubricating base oil blends of the invention may
11 also be blended with synthetic lubricants, such as esters (mono-, di-, dimer-,
12 polyol-, and aromatic), polyalphaolefins. Polyphenyl ethers and polyglycols.

13
14 Lubricating base oil blends of the invention represent premium lubricants
15 which may be used to prepare finished lubricants. A finished lubricant,
16 such as, for example, a commercial multi-grade crankcase lubricating oil
17 meeting SAE J300, June 2001 specifications, may be prepared from the
18 lubricating base oil blends of the invention by the addition of the
19 proper additives. Typical additives added to a lubricating base oil blend when
20 preparing a finished lubricant include anti-wear additives, detergents,
21 dispersants, antioxidants, pour point depressants, VI improvers, friction
22 modifiers, demulsifiers, antifoaming agents, corrosion inhibitors,
23 seal swell agents, and the like. In addition, commercial products meeting
24 SAE standards for gear lubricants, NLGI Mark GC and LB for greases, and
25 ISO Viscosity Grade standards for industrial oils may be prepared from the
26 Fischer-Tropsch derived lubricating base oils of the invention.

27

28 DETAILED DESCRIPTION OF THE INVENTION

29

30 Noack volatility of engine oil, as measured by TGA Noack and similar
31 methods, has been found to correlate with oil consumption in passenger car
32 engines. Strict requirements for low volatility are important aspects of several
33 recent engine oil specifications, such as, for example, ACEA A-3 and B-3 in

1 Europe and ILSAC GF-3 in North America. Due to the high volatility of
2 conventional low viscosity oils with kinematic viscosities below 3 cSt at
3 100 degrees C, they have limited their use in passenger car engine oils. Any
4 new lubricating base oil stocks developed for use in automotive engine oils
5 should have a volatility no greater than current conventional Group I or
6 Group II light neutral oils.

7
8 Fischer-Tropsch wax processing typically produces a relatively high
9 proportion of products of low molecular weight and low viscosity that are
10 processed into light products such as naphtha, gasoline, diesel, fuel oil, or
11 kerosene. A proportion of products have kinematic viscosities above 3.0 cSt
12 which are useful directly as lubricating base oils for many different products,
13 including engine oils. Those base oils with kinematic viscosities between
14 2.1 and 2.8 cSt typically are further processed into lighter products
15 (e.g., gasoline or diesel) in order to be of much economic value. Alternatively,
16 these low viscosity Fischer-Tropsch derived base oils may be used in light
17 industrial oils, such as, for example, utility oils, transformer oils, pump oils, or
18 hydraulic oils; many of which have less stringent volatility requirements, and
19 all of which are in much lower demand than engine oils.

20
21 Lubricating base oils for use in engine oils are in higher demand than those
22 for use in light products. The ability to use a higher proportion of the products
23 from Fischer-Tropsch processes in lubricating base oil blends for engine oils
24 is highly desirable. By virtue of the present invention, Fischer-Tropsch derived
25 lubricating base oils characterized by low viscosity are blended with a
26 Fischer-Tropsch derived bottoms fraction to produce compositions which are
27 useful as a lubricating base oil for preparing engine oil. The lubricating
28 base oil stocks of this invention are comparable in volatility and viscosity to
29 conventional Group I and Group II neutral oils. In addition, lubricating
30 base oils of the invention also have other improved properties, such as very
31 low sulfur and exceptional oxidation stability.

1 Fischer-Tropsch Synthesis

2

3 During Fischer-Tropsch synthesis liquid and gaseous hydrocarbons are
4 formed by contacting a synthesis gas (syngas) comprising a mixture of
5 hydrogen and carbon monoxide with a Fischer-Tropsch catalyst under
6 suitable temperature and pressure reactive conditions. The Fischer-Tropsch
7 reaction is typically conducted at temperatures of from about 300 degrees to
8 about 700 degrees F (about 150 degrees to about 370 degrees C) preferably
9 from about 400 degrees to about 550 degrees F (about 205 degrees to about
10 290 degrees C); pressures of from about 10 to about 600 psia,
11 (0.7 to 41 bars) preferably 30 to 300 psia, (2 to 21 bars) and catalyst space
12 velocities of from about 100 to about 10,000 cc/g/hr., preferably
13 300 to 3,000 cc/g/hr.

14

15 The products from the Fischer-Tropsch synthesis may range from C₁ to
16 C₂₀₀ plus hydrocarbons with a majority in the C₅-C₁₀₀ plus range. The reaction
17 can be conducted in a variety of reactor types, such as, for example,
18 fixed bed reactors containing one or more catalyst beds, slurry reactors,
19 fluidized bed reactors, or a combination of different types of reactors. Such
20 reaction processes and reactors are well known and documented in the
21 literature. The slurry Fischer-Tropsch process, which is preferred in the
22 practice of the invention, utilizes superior heat (and mass) transfer
23 characteristics for the strongly exothermic synthesis reaction and is able to
24 produce relatively high molecular weight paraffinic hydrocarbons when using
25 a cobalt catalyst. In the slurry process, a syngas comprising a mixture of
26 hydrogen and carbon monoxide is bubbled up as a third phase through a
27 slurry which comprises a particulate Fischer-Tropsch type hydrocarbon
28 synthesis catalyst dispersed and suspended in a slurry liquid comprising
29 hydrocarbon products of the synthesis reaction which are liquid under the
30 reaction conditions. The mole ratio of the hydrogen to the carbon monoxide
31 may broadly range from about 0.5 to about 4, but is more typically within the
32 range of from about 0.7 to about 2.75 and preferably from about 0.7 to about
33 2.5. A particularly preferred Fischer-Tropsch process is taught in

1 European Patent Application No. 0609079, also completely incorporated
2 herein by reference for all purposes.

3

4 Suitable Fischer-Tropsch catalysts comprise one or more Group VIII catalytic
5 metals such as Fe, Ni, Co, Ru and Re, with cobalt being preferred.

6 Additionally, a suitable catalyst may contain a promoter. Thus, a preferred
7 Fischer-Tropsch catalyst comprises effective amounts of cobalt and one or
8 more of Re, Ru, Pt, Fe, Ni, Th, Zr, Hf, U, Mg and La on a suitable inorganic
9 support material, preferably one which comprises one or more refractory
10 metal oxides. In general, the amount of cobalt present in the catalyst is
11 between about 1 and about 50 weight percent of the total catalyst
12 composition. The catalysts can also contain basic oxide promoters such as
13 ThO₂, La₂O₃, MgO, and TiO₂, promoters such as ZrO₂, noble metals (Pt, Pd,
14 Ru, Rh, Os, Ir), coinage metals (Cu, Ag, Au), and other transition metals such
15 as Fe, Mn, Ni, and Re. Suitable support materials include alumina, silica,
16 magnesia and titania or mixtures thereof. Preferred supports for cobalt
17 containing catalysts comprise titania. Useful catalysts and their preparation
18 are known and illustrated in U.S. Patent No. 4,568,663, which is intended to
19 be illustrative but non-limiting relative to catalyst selection.

20

21 The products as they are recovered from the Fischer-Tropsch operation
22 usually may be divided into three fractions, a gaseous fraction consisting of
23 very light products, a condensate fraction generally boiling in the range of
24 naphtha and diesel, and a high boiling Fischer-Tropsch wax fraction which is
25 normally solid at ambient temperatures. The Fischer-Tropsch derived
26 products used to prepare base oils are usually prepared from the waxy
27 fractions of the Fischer-Tropsch syncrude by hydrotreating and/or
28 hydroisomerization. Other methods which may be used in preparing the base
29 oils include oligomerization, solvent dewaxing, atmospheric and vacuum
30 distillation, hydrocracking, hydrofinishing, and other forms of hydroprocessing.

1 **Hydroisomerization and Solvent Dewaxing**

2
3 Hydroisomerization, or for the purposes of this disclosure simply
4 "isomerization", is intended to improve the cold flow properties of the
5 Fischer-Tropsch derived product by the selective addition of branching into
6 the molecular structure. Isomerization ideally will achieve high conversion
7 levels of the Fischer-Tropsch wax to non-waxy iso-paraffins while at the same
8 time minimizing the conversion by cracking. Since wax conversion can be
9 complete, or at least very high, this process typically does not need to be
10 combined with additional dewaxing processes to produce a lubricating oil
11 base stock with an acceptable pour point. Isomerization operations suitable
12 for use with the present invention typically uses a catalyst comprising an
13 acidic component and may optionally contain an active metal component
14 having hydrogenation activity. The acidic component of the catalysts
15 preferably includes an intermediate pore SAPO, such as SAPO-11, SAPO-31,
16 and SAPO-41, with SAPO-11 being particularly preferred. Intermediate pore
17 zeolites, such as ZSM-22, ZSM-23, SSZ-32, ZSM-35, and ZSM-48, also may
18 be used in carrying out the isomerization. Typical active metals include
19 molybdenum, nickel, vanadium, cobalt, tungsten, zinc, platinum, and
20 palladium. The metals platinum and palladium are especially preferred as the
21 active metals, with platinum most commonly used.

22
23 The phrase "intermediate pore size", when used herein, refers to an effective
24 pore aperture in the range of from about 4.0 to about 7.1 Angstrom when the
25 porous inorganic oxide is in the calcined form. Molecular sieves having pore
26 apertures in this range tend to have unique molecular sieving characteristics.
27 Unlike small pore zeolites such as erionite and chabazite, they will allow
28 hydrocarbons having some branching into the molecular sieve void spaces.
29 Unlike larger pore zeolites such as faujasites and mordenites, they are able to
30 differentiate between n-alkanes and slightly branched alkenes, and larger
31 alkanes having, for example, quaternary carbon atoms. See U.S. Patent
32 No. 5,413,695. The term "SAPO" refers to a silicoaluminophosphate

1 molecular sieve such as described in U.S. Patent Nos. 4,440,871 and
2 5,208,005.

3
4 In preparing those catalysts containing a non-zeolitic molecular sieve and
5 having an hydrogenation component, it is usually preferred that the metal be
6 deposited on the catalyst using a non-aqueous method. Non-zeolitic
7 molecular sieves include tetrahedrally-coordinated [AlO₂ and PO₂] oxide units
8 which may optionally include silica. See U.S. Patent No. 5,514,362. Catalysts
9 containing non-zeolitic molecular sieves, particularly catalysts containing
10 SAPO's, on which the metal has been deposited using a non-aqueous method
11 have shown greater selectivity and activity than those catalysts which have
12 used an aqueous method to deposit the active metal. The non-aqueous
13 deposition of active metals on non-zeolitic molecular sieves is taught in
14 U.S. Patent No. 5,939,349. In general, the process involves dissolving a
15 compound of the active metal in a non-aqueous, non-reactive solvent and
16 depositing it on the molecular sieve by ion exchange or impregnation.

17
18 Solvent dewaxing attempts to remove the waxy molecules from the product by
19 dissolving them in a solvent, such as methyl ethyl ketone, methyl iso-butyl
20 ketone, or toluene, and precipitating the wax molecules and then removing
21 them by filtration as discussed in Chemical Technology of Petroleum,
22 3rd Edition, William Gruse and Donald Stevens,
23 McGraw-Hill Book Company, Inc., New York, 1960, pages 566-570. See also
24 U.S. Patent Nos. 4,477,333; 3,773,650; and 3,775,288. In general, with the
25 present invention isomerization is usually preferred over solvent dewaxing,
26 since it results in higher viscosity index products with improved low
27 temperature properties, and in higher yields of the products boiling within the
28 range of the light distillate fraction and the heavy fraction. However solvent
29 dewaxing may be advantageously used in combination with isomerization to
30 recover unconverted wax following isomerization.

1 Hydrotreating, Hydrocracking, and Hydrofinishing

2

3 Hydrotreating refers to a catalytic process, usually carried out in the presence
4 of free hydrogen, in which the primary purpose is the removal of various metal
5 contaminants, such as arsenic; heteroatoms, such as sulfur and nitrogen; or
6 aromatics from the feed stock. Generally, in hydrotreating operations cracking
7 of the hydrocarbon molecules, i.e., breaking the larger hydrocarbon molecules
8 into smaller hydrocarbon molecules, is minimized, and the unsaturated
9 hydrocarbons are either fully or partially hydrogenated.

10

11 Hydrocracking refers to a catalytic process, usually carried out in the
12 presence of free hydrogen, in which the cracking of the larger hydrocarbon
13 molecules is the primary purpose of the operation. Desulfurization and/or
14 denitrification of the feedstock also usually will occur. Although typically
15 hydrocracking operations will usually be limited to the cracking of the heaviest
16 bottoms material, in the present invention it is one method that may be used
17 to increase amount of olefins present in the Fischer-Tropsch condensate
18 recovered from the Fischer-Tropsch synthesis operation. The olefin enriched
19 distillate fraction produced from the condensate may be oligomerized to
20 prepare bright stock which is blended with the light fraction to prepare
21 lubricating base oils within the scope of the invention.

22

23 Catalysts used in carrying out hydrotreating and hydrocracking operations are
24 well known in the art. See for example U.S. Patent Nos. 4,347,121 and
25 4,810,357, the contents of which are hereby incorporated by reference in their
26 entirety, for general descriptions of hydrotreating, hydrocracking, and of
27 typical catalysts used in each of the processes. Suitable catalysts include
28 noble metals from Group VIIIA (according to the 1975 rules of the
29 International Union of Pure and Applied Chemistry), such as platinum or
30 palladium on an alumina or siliceous matrix, and Group VIII and Group VIB,
31 such as nickel-molybdenum or nickel-tin on an alumina or siliceous matrix.
32 U.S. Patent No. 3,852,207 describes a suitable noble metal catalyst and mild
33 conditions. Other suitable catalysts are described, for example, in U.S. Patent

1 Nos. 4,157,294 and 3,904,513. The non-noble hydrogenation metals, such as
2 nickel-molybdenum, are usually present in the final catalyst composition as
3 oxides, but are usually employed in their reduced or sulfided forms when such
4 sulfide compounds are readily formed from the particular metal involved.
5 Preferred non-noble metal catalyst compositions contain in excess of about
6 5 weight percent, preferably about 5 to about 40 weight percent molybdenum
7 and/or tungsten, and at least about 0.5, and generally about 1 to about
8 15 weight percent of nickel and/or cobalt determined as the corresponding
9 oxides. Catalysts containing noble metals, such as platinum, contain in
10 excess of 0.01 percent metal, preferably between 0.1 and 1.0 percent metal.
11 Combinations of noble metals may also be used, such as mixtures of platinum
12 and palladium.

13

14 The hydrogenation components can be incorporated into the overall catalyst
15 composition by any one of numerous procedures. The hydrogenation
16 components can be added to matrix component by co-mulling, impregnation,
17 or ion exchange and the Group VI components, i.e.; molybdenum and
18 tungsten can be combined with the refractory oxide by impregnation,
19 co-mulling or co-precipitation.

20

21 The matrix component can be of many types including some that have acidic
22 catalytic activity. Ones that have activity include amorphous silica-alumina or
23 zeolitic or non-zeolitic crystalline molecular sieves. Examples of suitable
24 matrix molecular sieves include zeolite Y, zeolite X and the so called ultra
25 stable zeolite Y and high structural silica-alumina ratio zeolite Y such as that
26 described in U.S. Patent Nos. 4,401,556; 4,820,402; and 5,059,567. Small
27 crystal size zeolite Y, such as that described in U.S. Patent No. 5,073,530 can
28 also be used. Non-zeolitic molecular sieves which can be used include, for
29 example, silicoaluminophosphates (SAPO), ferroaluminophosphate, titanium
30 aluminophosphate and the various ELAPO molecular sieves described in
31 U.S. Patent No. 4,913,799 and the references cited therein. Details regarding
32 the preparation of various non-zeolite molecular sieves can be found in
33 U.S. Patent Nos. 5,114,563 (SAPO) and 4,913,799 and the various

1 references cited in U.S. Patent No. 4,913,799. Mesoporous molecular sieves
2 can also be used, for example the M41S family of materials as described in
3 J. Am. Chem. Soc., 114:10834-10843(1992)), MCM-41;
4 U.S. Patent Nos. 5,246,689; 5,198,203; and 5,334,368; and
5 MCM-48 (Kresge et al., Nature 359:710 (1992)). Suitable matrix materials
6 may also include synthetic or natural substances as well as inorganic
7 materials such as clay, silica and/or metal oxides such as silica-alumina,
8 silica-magnesia, silica-zirconia, silica-thoria, silica-berylia, silica-titania as well
9 as ternary compositions, such as silica-alumina-thoria, silica-alumina-zirconia,
10 silica-alumina-magnesia, and silica-magnesia zirconia. The latter may be
11 either naturally occurring or in the form of gelatinous precipitates or gels
12 including mixtures of silica and metal oxides. Naturally occurring clays which
13 can be composited with the catalyst include those of the montmorillonite and
14 kaolin families. These clays can be used in the raw state as originally mined
15 or initially subjected to dealumination, acid treatment or chemical modification.

16

17 In performing the hydrocracking and/or hydrotreating operation, more than
18 one catalyst type may be used in the reactor. The different catalyst types can
19 be separated into layers or mixed.

20

21 Hydrocracking conditions have been well documented in the literature. In
22 general, the overall LHSV is about 0.1 hr^{-1} to about 15.0 hr^{-1} (v/v), preferably
23 from about 0.25 hr^{-1} to about 2.5 hr^{-1} . The reaction pressure generally ranges
24 from about 500 psig to about 3500 psig (about 10.4 MPa to about 24.2 MPa,
25 preferably from about 1500 psig to about 5000 psig (about 3.5 MPa to about
26 34.5 MPa). Hydrogen consumption is typically from about 500 to about
27 2500 SCF per barrel of feed (89.1 to $445 \text{ m}^3 \text{ H}_2/\text{m}^3 \text{ feed}$). Temperatures in the
28 reactor will range from about 400 degrees F to about 950 degrees F (about
29 204 degrees C to about 510 degrees C), preferably ranging from about
30 650 degrees F to about 850 degrees F (about 343 degrees C to about
31 454 degrees C).

1 Typical hydrotreating conditions vary over a wide range. In general, the
2 overall LHSV is about 0.25 to 2.0, preferably about 0.5 to 1.0. The hydrogen
3 partial pressure is greater than 200 psia, preferably ranging from about
4 500 psia to about 2000 psia. Hydrogen recirculation rates are typically greater
5 than 50 SCF/Bbl, and are preferably between 1000 and 5000 SCF/Bbl.
6 Temperatures in the reactor will range from about 300 degrees F to about
7 750 degrees F (about 150 degrees C to about 400 degrees C), preferably
8 ranging from 450 degrees F to 600 degrees F (230 degrees C to about
9 315 degrees C).

10

11 Hydrotreating may also be used as a final step in the lube base oil
12 manufacturing process. This final step, commonly called hydrofinishing, is
13 intended to improve the UV stability and appearance of the product by
14 removing traces of aromatics, olefins, color bodies, and solvents. As used in
15 this disclosure, the term UV stability refers to the stability of the lubricating
16 base oil or the finished lubricant when exposed to UV light and oxygen.
17 Instability is indicated when a visible precipitate forms, usually seen as floc or
18 cloudiness, or a darker color develops upon exposure to ultraviolet light and
19 air. A general description of hydrofinishing may be found in U.S. Patent
20 Nos. 3,852,207 and 4,673,487. Clay treating to remove these impurities is an
21 alternative final process step.

22

23 **Thermal Cracking**

24

25 Thermal cracking may also be employed to crack the paraffin molecules into
26 lower molecular weight olefins in order to olefin enrich the Fischer-Tropsch
27 condensate. As already noted, all Fischer-Tropsch syncrude as initially
28 recovered from the Fischer-Tropsch synthesis will contain olefins. By thermal
29 cracking the paraffin molecules present in the condensate fraction the amount
30 of olefins present may be significantly increased. Following the thermal
31 cracking operation the condensate fraction should have an olefinicity of at
32 least 20 percent by weight, preferably at least 40 percent by weight, and most
33 preferably at least 50 percent by weight.

1 Although batch pyrolysis reactors, such as employed in delayed coking or in
2 cyclic batch operations, could be used to carry out this operation, generally a
3 continuous flow-through operation is preferred in which the feed is first
4 preheated to a temperature sufficient to vaporize most or all of the feed after
5 which the vapor is passed through a tube or tubes. A desirable option is to
6 bleed any remaining nonvaporized hydrocarbons prior to entering the tubes in
7 the cracking furnace. Preferably, the thermal cracking is conducted in the
8 presence of steam which serves as a heat source and also helps suppress
9 coking in the reactor. Details of a typical steam thermal cracking process may
10 be found in U.S. Patent No. 4,042,488, hereby incorporated by reference in its
11 entirety. Although catalyst is generally not used in carrying out the thermal
12 cracking operation, it is possible to conduct the operation in a fluidized bed in
13 which the vaporized feed is contacted with hot fluidized inert particles, such as
14 fluidized particles of coke.

15

16 In performing the thermal cracking operation, it is preferable that the feed be
17 maintained in the vapor phase during the cracking operation to maximize the
18 production of olefins. Liquid phase cracking results in the formation of
19 significant amounts of paraffins which are unreactive in the oligomerization
20 operation and, therefore, are not desired. In the pyrolysis zone, the cracking
21 conditions should be sufficient to provide a cracking conversion of greater
22 than about 30 percent by weight of the paraffins present. Preferably, the
23 cracking conversion will be at least 50 percent by weight and most preferably
24 at least 70 percent by weight. The optimal temperature and other conditions in
25 the pyrolysis zone for the cracking operation will vary somewhat depending on
26 the feed. In general, the temperature must be high enough to maintain the
27 feed in the vapor phase but not so high that the feed is overcracked, i.e., the
28 temperature and conditions should not be so severe that excessive C₄ minus
29 hydrocarbons are generated. The temperature in the pyrolysis zone normally
30 will be maintained at a temperature of between about 950 degrees F
31 (510 degrees C) and about 1600 degrees F (870 degrees C). The optimal
32 temperature range for the pyrolysis zone in order to maximize the production
33 of olefins from the Fischer–Tropsch wax will depend upon the endpoint of the

1 feed. In general, the higher the carbon number, the higher the temperature
2 required to achieve maximum conversion. Accordingly, some routine
3 experimentation may be necessary to identify the optimal cracking conditions
4 for a specific feed. The pyrolysis zone usually will employ pressures
5 maintained between about 0 atmospheres and about 5 atmospheres, with
6 pressures in the range of from about 0 to about 2 generally being preferred.
7 Although the optimal residence time of the wax fraction in the reactor will vary
8 depending on the temperature and pressure in the pyrolysis zone, typical
9 residence times are generally in the range of from about 1.5 seconds to about
10 500 seconds, with the preferred range being between about 5 seconds and
11 about 300 seconds.

12

13 **Oligomerization**

14

15 Depending upon how the Fischer-Tropsch synthesis is carried out, the
16 Fischer-Tropsch condensate will contain varying amounts of olefins. In
17 addition, most Fischer-Tropsch condensate will contain some alcohols which
18 may be readily converted into olefins by dehydration. As already noted, the
19 condensate may also be olefin enriched through a cracking operation, either
20 by means of hydrocracking or more preferably by thermal cracking. In one
21 embodiment of the present invention these olefins may be oligomerized to
22 produce a Fischer-Tropsch derived bright stock. During oligomerization the
23 lighter olefins are not only converted into heavier molecules, but the carbon
24 backbone of the oligomers will also display branching at the points of
25 molecular addition. Due to the introduction of branching into the molecule, the
26 pour point of the products is reduced.

27

28 The oligomerization of olefins has been well reported in the literature, and a
29 number of commercial processes are available. See, for example, U.S. Patent
30 Nos. 4,417,088; 4,434,308; 4,827,064; 4,827,073; and 4,990,709. Various
31 types of reactor configurations may be employed, with the fixed catalyst bed
32 reactor being used commercially. More recently, performing the
33 oligomerization in an ionic liquid media has been proposed, since these

1 catalysts are very active, and the contact between the catalyst and the
2 reactants is efficient and the separation of the catalyst from the
3 oligomerization products is facilitated. The oligomerization reaction will
4 proceed over a wide range of conditions. Typical temperatures for carrying out
5 the reaction are between about 32 degrees F (0 degrees C) and about
6 800 degrees F (425 degrees C). Other conditions include a space velocity
7 from 0.1 to 3 LHSV and a pressure from 0 to 2000 psig. Catalysts for the
8 oligomerization reaction can be virtually any acidic material, such as, for
9 example, zeolites, clays, resins, BF_3 complexes, HF, H_2SO_4 , AlCl_3 , ionic
10 liquids (preferably ionic liquids containing a Bronsted or Lewis acidic
11 component or a combination of Bronsted and Lewis acid components),
12 transition metal-based catalysts (such as Cr/SiO_2), superacids, and the like. In
13 addition, non-acidic oligomerization catalysts including certain organometallic
14 or transition metal oligomerization catalysts may be used, such as, for
15 example, zirconocenes.

16

17 Distillation

18

19 The separation of the Fischer-Tropsch derived products into the various
20 fractions used in the process of the invention is generally conducted by either
21 atmospheric or vacuum distillation or by a combination of atmospheric and
22 vacuum distillation. Atmospheric distillation is typically used to separate the
23 lighter distillate fractions, such as naphtha and middle distillates, from a
24 bottoms fraction having an initial boiling point above about 700 degrees F to
25 about 750 degrees F (about 370 degrees C to about 400 degrees C). At
26 higher temperatures thermal cracking of the hydrocarbons may take place
27 leading to fouling of the equipment and to lower yields of the heavier cuts.
28 Vacuum distillation is typically used to separate the higher boiling material,
29 such as the lubricating base oil fractions.

30

31 As used in this disclosure, the term "distillate fraction" or "distillate" refers to a
32 side stream product recovered either from an atmospheric fractionation
33 column or from a vacuum column as opposed to the "bottom fraction" which

1 represents the residual higher boiling fraction recovered from the bottom of
2 the column. In this disclosure, the term "bottoms" also includes those bottoms
3 fractions and bright stock derived from the oligomerization of olefins present in
4 the Fischer-Tropsch condensate.

5

6 **The Distillate Fraction**

7

8 The distillate fraction used to prepare the lubricating base oil product of the
9 invention represents a distillate fraction of the Fischer-Tropsch derived
10 product as defined above. Distillate fractions used in carrying out the invention
11 and the Fischer-Tropsch derived lubricating base oil blends of the invention
12 may be characterized by their true boiling point (TBP) and by their boiling
13 range distribution. For the purposes of this disclosure, unless stated
14 otherwise, TBP and boiling range distributions for a distillate fraction are
15 measured by gas chromatography according to ASTM D-6352 or its
16 equivalent.

17

18 A critical property of the distillate fractions of the invention is viscosity. The
19 distillate fraction must have a kinematic viscosity of about 2 or greater but less
20 than 3 cSt at 100 degrees C, more preferably between about 2.1 and 2.8 cSt
21 at 100 degrees C, and most preferably between about 2.2 and 2.7 cSt at
22 100 degrees C.

23

24 Another critical property of the distillate fractions and the lubricating base oil
25 products of the invention is volatility which is expressed as Noack volatility,
26 Noack volatility is defined as the mass of oil, expressed in weight percent,
27 which is lost when the oil is heated at 250 degrees C and 20 mmHg
28 (2.67 kPa; 26.7 mbar) below atmospheric in a test crucible through which a
29 constant flow of air is drawn for 60 minutes (ASTM D-5800). A more
30 convenient method for calculating Noack volatility and one which correlates
31 well with ASTM D-5800 is by using a thermo gravimetric analyzer test (TGA)
32 by ASTM D-6375. TGA Noack volatility is used throughout this disclosure
33 unless otherwise stated. As already noted above, the first distillate fraction of

1 the invention while having a viscosity below 3 cSt at 100 degrees C displays a
2 significantly lower TGA Noack volatility than would be expected when
3 compared to conventional petroleum-derived distillates having a comparable
4 viscosity. This makes it possible to blend the low viscosity first distillate
5 fraction with the higher viscosity second distillate fraction and still meet the
6 volatility specifications for the lube base oil and the finished lubricant.

7

8 **The Bottoms Fraction**

9

10 The Fischer-Tropsch derived bottoms fraction represents a high viscosity high
11 boiling fraction. Typically, the bottom fraction will have a kinematic viscosity of
12 at least 9 cSt at 100 degrees C. Fischer-Tropsch derived bottoms other than
13 bright stock will usually have a kinematic viscosity between about 9 cSt and
14 about 20 cSt at 100 degrees C, with a kinematic viscosity of between about
15 10 cSt and about 16 cSt being preferred. The bottom fraction will contain a
16 large percent of Fischer-Tropsch wax and usually will be solid at room
17 temperature. In order to improve its properties prior to being blended with the
18 distillate fraction, it may be advantageous to further process the bottom
19 fraction. For example, the bottom fraction may be hydrotreated to saturate the
20 double bonds and remove any impurities, such as any oxygenates, that may
21 be present. The bottoms fraction may also be isomerized to improve its cold
22 flow properties.

23

24 The Fischer-Tropsch derived bright stock may be prepared by highly refining
25 the waxy bottom fraction recovered directly from a Fischer-Tropsch plant.
26 However, since the Fischer-Tropsch syncrude usually does not comprise a
27 large proportion of heavy products, it may be desirable to prepare at least part
28 of the bright stock through the oligomerization of the olefins present in the
29 Fischer-Tropsch condensate. The enrichment of the Fischer-Tropsch
30 condensate with olefins and the oligomerization of the olefins to produce
31 larger molecules has already been discussed. Typically, the processing of
32 Fischer-Tropsch derived materials to yield bright stock will include dewaxing,
33 hydrofinishing, and fractionation. As already noted, Fischer-Tropsch bright

1 stock is a high viscosity material having a kinematic viscosity within the range
2 of from about 15 cSt to about 40 cSt at 100 degrees F.

3

4 **Lubricating Base Oil Blends**

5

6 Lubricating base oils are generally materials having a viscosity greater than
7 3 cSt at 100 degrees C; a pour point below 20 degrees C, preferably below
8 0 degrees C; and a VI of greater than 70, preferably greater than 90. As
9 explained below and illustrated in the examples, the lubricating base oil
10 blends prepared according to the process of the present invention meet these
11 criteria. In addition, the lubricating base oils of the invention display a unique
12 combination of properties which could not have been predicted from a review
13 of the prior art relating to both conventional and Fischer-Tropsch materials.
14 The invention takes advantage of the high VI of the light distillate fraction
15 which when blended with the heavier fraction will result in a final blend having
16 a viscosity which is within acceptable limits for use as a lubricating base oil.

17

18 Lubricating base oils within the scope of the invention will generally have a
19 kinematic viscosity between about 3 and about 10 cSt at 100 degrees C.
20 Generally the lubricating base oil will be blended to a pre-selected target
21 viscosity which is suitable for preparing a finished lubricant intended for a
22 particular application. Obviously the proportions of the various distillate and
23 heavy fractions in the blend will need to be adjusted to meet this desired
24 target viscosity in the lubricating base oil blend. The exact ratio of each of the
25 fractions in the final blend will depend on the exact viscosity of each fraction
26 and the target viscosity desired for the lubricating base oil, as well as other
27 desired properties such as, for example, VI, volatility, pour point, cloud point
28 and the like.

29

30 The lubricating base oil formed by the blending of the distillate fraction and the
31 heavy fraction is characterized as having a viscosity between about 3 and
32 about 10 cSt at 100 degrees C and a TGA Noack volatility of less than about
33 45 weight percent. Generally, the lubricating base oil will have a viscosity

1 between about 4 cSt and about 8 cSt at 100 degrees C and a Noack volatility
2 greater than about 12 weight percent. Commonly the Noack volatility will be
3 greater than about 20 weight percent. However, if the lubricating base oil
4 blend is intended for use in formulating a crankcase lubricating oil, the Noack
5 volatility preferably will be less than about 30 weight percent. Volatility of the
6 Fischer-Tropsch derived lubricating base oil blends of the invention are
7 acceptable and are comparable to conventional petroleum derived lubricating
8 base oils which is surprising given the low viscosity of the distillate fraction.
9 The use of a comparable petroleum derived base oil in a lubricating base oil
10 blend would result in an unacceptably high Noack volatility. Generally, the
11 viscosity index (VI) of the Fischer-Tropsch derived lubricating base oil blend
12 will be between about 130 and about 185. VI is an expression of the effect of
13 temperature on viscosity, and it is surprising that a lubricating base oil blend
14 prepared using a base oil having a viscosity of less than 3 cSt at
15 100 degrees C will be characterized by such a favorable VI. As noted
16 previously, it is even more surprising that the blends of the invention often
17 realize a VI premium, i.e., the VI of the lubricating base oil blend is higher than
18 would be expected from an averaging of VI of the light distillate fraction with
19 that of the heavy fraction. Since Fischer-Tropsch derived hydrocarbons are
20 typically very low in total sulfur, the total sulfur content of the lubricating base
21 oil usually will be less than about 5 ppm. Conventionally- derived, solvent
22 processed lubricating base oils will generally display much higher sulfur
23 levels, usually in excess of 2000 ppm.

24

25 Lubricating base oil blends within the scope of the invention will generally
26 have a boiling range distribution of at least 450 degrees F (about
27 232 degrees C). Typically, the Fischer-Tropsch derived lubricating base oil
28 blend will have an initial boiling point within the range of between about
29 550 degrees F (288 degrees C) and about 625 degrees F (330 degrees C), an
30 end boiling point between about 1000 degrees F (538 degrees C) and about
31 1400 degrees F (760 degrees C), and wherein less than 20 weight percent of
32 the blend boils within the region defined by the 50 percent boiling point, plus
33 or minus 30 degrees F (16.7 degrees C). The boiling range distribution of the

1 lubricating base oils of the invention are significantly broader than those
2 observed for conventional lubricating base oils. The boiling range for
3 conventionally derived lubricating base oils typically will not exceed about
4 250 degrees F (about 139 degrees C). In this disclosure when referring to
5 boiling range distribution, the boiling range between the 5 percent and
6 95 percent boiling points is what is referred to.

7
8 Pour point is the temperature at which a sample of the lubricating base oil will
9 begin to flow under carefully controlled conditions. In this disclosure, where
10 pour point is given, unless stated otherwise, it has been determined by
11 standard analytical method ASTM D-5950. Lubricating base oils prepared
12 according to the present invention have excellent pour points which are
13 comparable or even below the pour points observed for conventionally
14 derived lubricating base oils. In addition, the blends containing the distillate
15 fraction and bright stock have been observed to display a cloud point
16 premium, i.e., the cloud point is significantly lower than would have been
17 predicted from the mere averaging of the cloud points of the two components
18 making up the blend. In some cases the cloud point will actually be
19 significantly lower than the cloud point of either component. Preferably the
20 cloud point of such blends will be -15 degrees C or less. Finally, due to the
21 extremely low aromatics and multi-ring naphthene levels of blends of
22 Fischer-Tropsch derived lubricating base oils; their oxidation stability far
23 exceeds that of conventional lubricating base oil blends.

24
25 A useful property of lubricating base oils and finished lubricants intended for
26 use in automobile engine oils is measured by cold-cranking simulator (CCS)
27 apparent viscosity which correlates with low temperature cranking. It is
28 measured by ASTM D5293 at a set temperature between -10 and -
29 35 degrees C. Engine oil specifications, e.g., SAE J300, include maximum
30 limits for CCS Viscosity for multi-grade engine oils. For a finished lubricant
31 within the scope of the invention the cold-cranking simulator (CCS) apparent
32 viscosity should be less than 7000 cP at -25 degrees C and preferably of

1 6500 cP or less at -25 degrees C if the lubricant is intended for use as a
2 multi-grade engine oil in an automobile engine.

3

4 **Finished Lubricants**

5

6 Finished lubricants generally comprise a lubricating base oil and at least
7 one additive. Finished lubricants are used in automobiles, diesel engines, gas
8 engines, axles, transmissions, and industrial applications. As noted above,
9 finished lubricants must meet the specifications for their intended application
10 as defined by the concerned governing organization. Lubricating base oils of
11 the present invention have been found to be suitable for formulating finished
12 lubricants intended for many of these applications. For example, lubricating
13 base oils of the present invention may be utilized in formulations to meet
14 SAE J300, June 2001 specifications for 10W-XX, and 15W-XX multi-grade
15 crankcase lubricating oils. Although some multi-grade crankcase oils meeting
16 may be formulated using only Fischer-Tropsch lubricating base oils prepared
17 according to the present invention, in order to meet the specifications for
18 some 10W-XX and most 15W-XX, it may be desirable that the
19 Fischer-Tropsch derived lubricating base oil be blended with a conventional
20 petroleum derived lubricating base oil, such as a conventional neutral Group I
21 or Group II base oil to meet the specifications. Typically, when present the
22 conventional neutral Group I or Group II base oil will comprise from about
23 40 to about 90 weight percent of the lubricating base oil blend, more
24 preferably from about 40 to about 70 weight percent. Also Fischer-Tropsch
25 derived lubricating oils of the invention may be used to formulate mono-grade
26 engine oils, such as SAE 20 or SAE 30, which are heavily used in many parts
27 of the world where low temperature performance is not critical. In addition,
28 Fischer-Tropsch derived lubricating base oils of the invention may be used to
29 formulate finished lubricants meeting the specifications for automatic
30 transmission fluids, NLGI Mark GC and LB greases, and ISO Viscosity
31 Grade 22, 32, and 46 industrial oils.

1 The lubricating base oil compositions of the invention may also be used as a
2 blending component with other oils. For example, the Fischer-Tropsch derived
3 lubricating base oils may be used as a blending component with synthetic
4 base oils, such as esters (mono-, di-, dimer, polyol-, and aromatic),
5 polyalphaolefins, polyphenyl ethers, and polyglycols to improve the viscosity
6 and viscosity index properties of those oils. The Fischer-Tropsch derived base
7 oils may be combined with isomerized petroleum wax. They may also be used
8 as workover fluids, packer fluids, coring fluids, completion fluids, and in other
9 oil field and well-servicing applications. For example, they can be used as
10 spotting fluids to release a drill pipe which has become stuck, or they can be
11 used to replace part or all of the expensive polyalphaolefin lubricating
12 additives in downhole applications. Additionally, Fischer-Tropsch derived
13 lubricating base oils may be used in drilling fluid formulations where
14 shale-swelling inhibition is important, such as described in U.S. Patent
15 No. 4,941,981.

16

17 Additives which may be blended with the lubricating base oil to form the
18 finished lubricant composition include those which are intended to improve
19 certain properties of the finished lubricant. Typical additives include, for
20 example, anti-wear additives, detergents, dispersants, antioxidants, pour point
21 depressants, VI improvers, friction modifiers, demulsifiers, antifoaming
22 agents, corrosion inhibitors; seal swell agents, and the like. Other
23 hydrocarbons, such as those described in U.S. Patent Nos. 5,096,883 and
24 5,189,012, may be blended with the lubricating base oil provided that the
25 finished lubricant has the necessary pour point, kinematic viscosity, flash
26 point, and toxicity properties. Typically, the total amount of additives in the
27 finished lubricant will fall within the range of from about 1 to about
28 30 weight percent. However due to the excellent properties of the
29 Fischer-Tropsch derived lubricating base oils of the invention, less additives
30 than required with conventional petroleum derived base oils may be required
31 to meet the specifications for the finished lubricant. The use of additives in
32 formulating finished lubricants is well documented in the literature and well

1 within the ability of one skilled in the art. Therefore, additional explanation
2 should not be necessary in this disclosure.

3

4

EXAMPLES

5

6 The following examples are included to further clarify the invention but are not
7 to be construed as limitations on the scope of the invention.

8

Example 1

10

11 Two Fisher-Tropsch distillate fractions (designated FT-2.2 and FT-2.4,
12 respectively) having kinematic viscosities between 2 and 3 cSt at
13 100 degrees C were recovered from a Fischer-Tropsch syncrude prepared
14 using a cobalt-based catalyst. Each fraction was analyzed and their properties
15 were compared to two commercially available conventional petroleum derived
16 oils (Nexbase 3020 and Pennzoil 75HC) having viscosities within the same
17 general range. A comparison between the properties of the four samples is
18 shown below:

19

20		FT-2.2	FT-2.4	Nexbase 3020	Pennzoil 75HC
21					
22	Vis. @ 100° C (cSt)	2.18	2.399	2.055	2.885
23	Viscosity Index (VI)	123	125	96	80
24	Pour Point, C	-37	-33	-51	-38
25	Noack (wt. %)	52.3	56.64	75.1	59.1

27 It should be noted that, although the viscosity at 100 degrees C of the
28 two Fischer-Tropsch derived materials were comparable to those of the
29 conventional oils, the VI is surprisingly high, which results in a much
30 lower volatility for a given viscosity.

31

Example 2

33

34 A Fisher-Tropsch bottom fraction, designated FT-14, was recovered from a
35 Fischer-Tropsch syncrude prepared using an iron-based catalyst. The bottom

1 fraction was subsequently hydrotreated. The properties of FT-14 were as
2 follows:

3
4 Viscosity at 100 degrees C (cSt) 14.62
5 Viscosity Index (VI) 160
6 Pour Point, C -1
7

8 Example 3

9
10 Two different Fischer-Tropsch derived lubricating base oil blends were
11 prepared by blending different proportions of the FT-2.4 from example 1 and
12 FT-14 from example 2. The proportions of FT-2.4 and FT-14 in each blend are
13 shown in Table 1 below:

14
15 Table 1

	Wt % FT-2.2	Wt % FT-14
Lubricating Base Oil A	60	40
Lubricating Base Oil B	35	65

16
17 The properties for each of the lubricating base oil blends are summarized in
18 Table 2 below:

19
20 Table 2

	Lubricating Base Oil A	Lubricating Base Oil B
D-2887 Simulated TBP (WT%), °F		
TBP @0.5 (Initial Boiling Point)	593	596
TBP @5	616	634
TBP @10	630	659
TBP @20	656	708
TBP @30	680	765
TBP @40	705	1015
TBP @50	730	1032
TBP @60	760	1049
TBP @70	996	1065
TBP @80	1027	1089
TBP @90	1057	1136
TBP @95	1079	1182
TBP @99.5	1132	1251
Boiling Range Distribution (5-95)	463	548
Viscosity at 40°C	21.00	38.62
Viscosity at 100°C	4.969	7.718
Viscosity Index	174	174
Pour Point, °C	-29	-19
CCS at -25°C, cP*		2293
CCS at -35°C, cP	1058	8570
TGA Noack	37.85	22.47

21
22
23 *This property represents cold-cranking simulator (CCS) apparent viscosity which is a measure of low temperature cold-cranking in automobile engines determined by ASTM D-5293.

1 It should be noted that both Fischer-Tropsch blends had volatilities, as
 2 measured by TGA Noack, which was suitable for blending engine oils. It
 3 should also be noted that the VI of each of the blends was higher than the VI
 4 of either FT-2.4 or FT-14 indicating that the blends were realizing a VI
 5 premium.

7 Example 4

8
 9 The properties of the Fischer-Tropsch derived lubricating base oils as shown
 10 in Table 2 above may be compared to the properties of commercially
 11 available petroleum derived conventional Group I and Group II light neutral
 12 base oils as summarized in Table 3 below.

13
 14 Table 3

	ChevronTexaco 100R	Generic Gulf Coast Solvent 100	Gulf Coast H.P. 100	Exxon Americas Core 100
API Base Oil Category (API 1509 E.1.3)	II	I	II	I
D-6352 Simulated TBP (WT%), °F				
TBP @5	659			647
TBP @10	677			672
TBP @20	703			703
TBP @30	723			725
TBP @50	756			761
TBP @70	786			796
TBP @90	825			839
TBP @95	842			858
TBP @99.5	878			907
Boiling Range Distribution (5-95)	219			211
Viscosity at 40°C	20.0	20.4	20.7	20.2
Viscosity at 100°C	4.1	4.1	4.1	4.04
Viscosity Index	102	97	97	95
Pour Point, °C	-14	-18	-15	-19
CCS at -25°C, cP	1450	1430	1550	1513
CCS at -35°C, cP	>3000	>3000	>3000	>3000
Noack Volatility, wt%	26	29	25.5	29.3

15
 16 A comparison of Table 2 and 3 illustrate that the Fischer-Tropsch derived
 17 lubricating base oils have a similar Noack volatility to conventional Group I
 18 and Group II light neutral oils. The kinematic viscosity of
 19 Lubricating Base Oil A was comparable to that of the Group I and Group II
 20 light neutral oils while that of Lubricating Base Oil B was significantly higher.
 21 Lubricating Base Oil A displays a lower pour point than the conventional light

1 neutral oils. The Fischer-Tropsch derived lubricating base oils of the invention
2 also display significantly better VI values.

3
4 **Example 5**

5
6 A lubricating base oil blend was prepared which contained 25 weight percent
7 of FT-2.2 and 75 weight percent of FT-14. The kinematic viscosity at
8 100 degrees C was found to be 9.007 and the VI was 173. Once again the
9 blend displays a VI premium over the VI's of both FT-2.2 and FT-14.

10
11 **Example 6**

12
13 A Fisher-Tropsch distillate fraction designated FT-2.5 was blended with a
14 Fischer-Tropsch derived bright stock designated FT-BS that was prepared by
15 oligomerizing the olefins in a Fischer-Tropsch derived feed. The properties of
16 the two Fischer-Tropsch derived feeds were as follows:

17
18

	FT-2.5	FT-BS
19 Vis. @ 100° C (cSt)	2.583	30.12
20 Viscosity Index (VI)	133	132
21 Pour Point, C	-30	-46
22 Cloud Point, C	-16	-10
23 Noack (wt. %)	48.94	

24
25

26 Two different Fischer-Tropsch derived lubricating base oil blends were
27 prepared and the respective proportions of FT-2.5 and FT-BS in each blend
28 are shown in Table 4 below:

29
30 Table 4

	Wt % FT-2.5	Wt % FT-BS
Lubricating Base Oil C	70	30
Lubricating Base Oil D	30	70

1 The properties for each of the lubricating base oil blends are summarized in
2 Table 5 below:

Table 5

	Lubricating Base Oil C	Lubricating Base Oil D
D-2887 Simulated TBP (WT%), °F		
TBP @0.5 (Initial Boiling Point)	599	604
TBP @5	617	634
TBP @10	630	664
TBP @20	656	728
TBP @30	684	809
TBP @40	711	1048
TBP @50	739	1114
TBP @60	773	1165
TBP @70	809	1210
TBP @80	1111	1261
TBP @90	1226	1312
TBP @95	1288	1335
TBP @99.5	1349	1373
Boiling Range Distribution (5-95)	671	701
Viscosity at 40°C	20.70	84.14
Viscosity at 100°C	4.799	12.58
Viscosity Index	162	147
Pour Point, °C	-23	-33
Cloud Point, °C	-16	-23
CCS at -20°C, cP*		4,017
CCS at -25°C, cP*		6,665
CCS at -30°C, cP	1,186	11,911
TGA Noack	34.55	14.52

*This property represents cold-cranking simulator (CCS) apparent viscosity which is a measure of low temperature cold-cranking in automobile engines determined by ASTM D-5293.

8 It should be noted that both Fischer-Tropsch blends had excellent VI,
9 low pour points, and low cloud points. Note particularly the VI of the blends
10 which demonstrate a VI premium when compared to the VI's for FT-2.5 and
11 FT-BS. Also note the significant improvement in cloud point as compared with
12 FT-BS. Note that the cloud point for lubricating base oil D displays a cloud
13 point premium, i.e., the cloud point is significantly lower than that for either
14 FT-2.5 or FT-BS. Base oils with a premium cloud point have utility in products
15 which require cold filtration, such as, for example, refrigeration oils.
16 Base oil C, although too high in volatility to be used in engine oils alone, can
17 be further blended as a minority component for engine oils, or used as a
18 majority component in many other lubricant applications, such as, for
19 example, transmission fluids, industrial oils, diluent oils, spray oils, process
20 oils, hydraulic oils, and the like. Base oil D can be used to make
21 15W 40 engine oil with no added viscosity modifier.

1 WHAT IS CLAIMED IS:
2
3

- 4 1. A process for producing a Fischer-Tropsch derived lubricating base oil
5 blend which comprises blending a Fischer-Tropsch distillate fraction
6 with a Fischer-Tropsch derived bottoms fraction in the proper
7 proportion to produce a Fischer-Tropsch derived lubricating base oil
8 characterized as having a kinematic viscosity of between about 3 and
9 about 10 cSt at 100 degrees C and a TGA Noack volatility of less than
10 about 45 weight percent wherein said distillate fraction is characterized
11 by a kinematic viscosity of about 2 cSt or greater but less than 3 cSt at
12 100 degrees C.
13
- 14 2. The process of claim 1 wherein the distillate fraction has a viscosity
15 between about 2.1 and 2.8 cSt at 100 degrees C.
16
- 17 3. The process of claim 2 wherein the distillate fraction has a viscosity
18 between about 2.2 and 2.7 cSt at 100 degrees C.
19
- 20 4. The process of claim 1 wherein the Fischer-Tropsch derived bottoms
21 fraction has a kinematic viscosity between about 9 cSt and about
22 20 cSt at 100 degrees C.
23
- 24 5. The process of claim 4 wherein the bottoms fraction has a viscosity of
25 between about 10 and about 16 cSt at 100 degrees C.
26
- 27 6. The process of claim 1 wherein the Fischer-Tropsch derived bottoms
28 fraction is bright stock.
29
- 30 7. The process of claim 6 wherein the bright stock is produced by
31 oligomerizing the olefins present in an olefin-containing
32 Fischer-Tropsch derived condensate.

- 1
- 2 8. The process of claim 7 including the preliminary step of enriching the
- 3 olefins in the Fischer-Tropsch condensate by the dehydration of the
- 4 alcohols present in the condensate.
- 5
- 6 9. The process of claim 7 including the preliminary step of enriching the
- 7 olefins in the Fischer-Tropsch condensate by the thermal cracking of
- 8 the hydrocarbons present in the condensate.
- 9
- 10 10. The process of claim 1 wherein the lubricating base oil blend has a
- 11 Noack volatility of less than 30 weight percent.
- 12
- 13 11. The process of claim 1 including the additional step of blending the
- 14 Fischer-Tropsch lubricating base oil blend with at least one additive to
- 15 produce a finished lubricant.
- 16
- 17 12. A Fischer-Tropsch derived lubricating base oil blend characterized by a
- 18 viscosity of between about 3 and about 10 cSt at 100 degrees C; a
- 19 TGA Noack volatility of less than 45 weight percent; an initial boiling
- 20 point within the range of between about 550 degrees F and about
- 21 625 degrees F; an end boiling point between about 1000 degrees F
- 22 and about 1400 degrees F; and wherein less than 30 weight percent of
- 23 the blend boils within the region defined by the 50 percent boiling
- 24 points, plus or minus 25 degrees F.
- 25
- 26 13. The Fischer-Tropsch derived lubricating base oil blend of claim 12
- 27 having a boiling range distribution of at least 450 degrees F between
- 28 the 5 percent and 95 percent points as measured by analytical method
- 29 D-6352 or its equivalent.
- 30

- 1 14. The Fischer-Tropsch lubricating base oil blends of claim 12 wherein
2 the TGA Noack volatility is 12 weight percent or greater.
3
- 4 15. The Fischer-Tropsch lubricating base oil blend of claim 14 wherein the
5 TGA volatility is greater than about 20 weight percent.
6
- 7 16. The Fischer-Tropsch lubricating base oil blend of claim 12 wherein the
8 TGA volatility is less than about 30 weight percent.
9
- 10 17. The Fischer-Tropsch lubricating base oil blend of claim 12 wherein the
11 VI is between about 130 and about 185.
12
- 13 18. The Fischer-Tropsch lubricating base oil blend of claim 12 wherein the
14 total sulfur content is less than about 5 ppm.
15
- 16 19. A finished lubricant comprising a Fischer-Tropsch derived lubricating
17 base oil blend and at least one additive wherein the Fischer-Tropsch
18 derived lubricating base oil blend is characterized by a viscosity of
19 between about 3 and about 10 cSt at 100 degrees C and comprises a
20 Fischer-Tropsch distillate fraction and a Fischer-Tropsch derived
21 bottoms fraction wherein:
22
- 23 (a) said distillate fraction is characterized by a viscosity of about
24 2 cSt or greater but less than 3 cSt at 100 degrees C and
25
- 26 (b) said Fischer-Tropsch derived bottoms fraction is characterized
27 by a viscosity of not less than about 9 cSt at 100 degrees C.
28
- 29 20. The finished lubricant of claim 19 which is suitable for use as an
30 engine oil crankcase lubricant.
31