

TITLE OF INVENTION**REFRIGERANT COMPOSITIONS COMPRISING FUNCTIONALIZED
ORGANIC COMPOUNDS AND USES THEREOF****5 CROSS REFERENCE(S) TO RELATED APPLICATION(S)**

This application claims the priority benefit of U.S. Provisional Application 60/584,785, filed June 29, 2004.

BACKGROUND OF THE INVENTION**10 1. Field of the Invention.**

The present invention relates to compositions for use in refrigeration and air-conditioning systems comprising at least one chlorocarbon, alcohol, ketone, ether, ester, N-(difluoromethyl)-N,N-dimethylamine, 1,1,1,2,2-pentafluoro-2-[(pentafluoroethyl)thio]ethane or
15 combinations or mixtures thereof. Further, the present invention relates to compositions for use in refrigeration and air-conditioning systems employing a centrifugal compressor comprising at least one chlorocarbon, alcohol, ketone, ether, ester, N-(difluoromethyl)-N,N-dimethylamine, 1,1,1,2,2-pentafluoro-2-[(pentafluoroethyl)thio]ethane or combinations or
20 mixtures thereof. The compositions of the present invention are useful in processes for producing refrigeration or heat or as heat transfer fluids.

2. Description of Related Art.

The refrigeration industry has been working for the past few
25 decades to find replacement refrigerants for the ozone depleting chlorofluorocarbons (CFCs) and hydrochlorofluorocarbons (HCFCs) being phased out as a result of the Montreal Protocol. The solution for most refrigerant producers has been the commercialization of hydrofluorocarbon (HFC) refrigerants. The new HFC refrigerants, HFC-
30 134a being the most widely used at this time, have zero ozone depletion potential and thus are not affected by the current regulatory phase out as a result of the Montreal Protocol.

Further environmental regulations may ultimately cause global phase out of certain HFC refrigerants. Currently, the automobile industry
35 is facing regulations relating to global warming potential for refrigerants used in mobile air-conditioning. Therefore, there is a great current need to identify new refrigerants with reduced global warming potential for the automobile air-conditioning market. Should the regulations be more

broadly applied in the future, an even greater need will be felt for refrigerants that can be used in all areas of the refrigeration and air-conditioning industry.

5 Currently proposed replacement refrigerants for HFC-134a include HFC-152a, pure hydrocarbons such as butane or propane, or "natural" refrigerants such as CO₂ or ammonia. Many of these suggested replacements are toxic, flammable, and/or have low energy efficiency. Therefore, new alternatives are constantly being sought.

10 The object of the present invention is to provide novel refrigerant compositions and heat transfer fluids that provide unique characteristics to meet the demands of low or zero ozone depletion potential and lower global warming potential as compared to current refrigerants.

15 **BRIEF SUMMARY OF THE INVENTION**

 The present invention relates to refrigerant and heat transfer fluid compositions comprising:

 nonafluoro-tert-butanol;
 1,1-dichloroethane;
20 trans-1,2-dichloroethylene;
 acetone;
 tert-butyl methyl ether;
 diisopropyl ether;
 1,2-dimethoxyethane;
25 dimethoxymethane;
 1,1-dimethoxyethane;
 methanol;
 ethanol;
 n-propanol;
30 isopropanol;
 ethyl acetate;
 ethyl formate;
 methyl acetate;
 methyl formate;
35 N-(difluoromethyl)-N,N-dimethylamine;
 1,1,1,2,2-pentafluoro-2-[(pentafluoroethyl)thio]ethane;
 or combinations thereof.

The present invention further relates to the above listed compositions specifically for use in refrigeration or air-conditioning systems employing a centrifugal compressor; a multi-stage or a 2-stage centrifugal compressor; or a single pass/single slab heat exchanger.

- 5 The present invention further relates to processes for producing cooling, heat, and transfer of heat from a heat source to a heat sink using the present inventive compositions.

DETAILED DESCRIPTION OF THE INVENTION

- 10 The refrigerant compositions of the present invention comprise at least one chlorocarbon, alcohol, ketone, ether, ester, N-(difluoromethyl)-N,N-dimethylamine, 1,1,1,2,2-pentafluoro-2-[(pentafluoroethyl)thio]ethane or mixtures thereof. The refrigerant compositions of the present invention may comprise a single compound or may be a mixture comprising more
15 than one compound.

- The refrigerant compositions of the present invention comprise compounds selected from chlorocarbons, alcohols, ethers, ketones, esters, N-(difluoromethyl)-N,N-dimethylamine, or 1,1,1,2,2-pentafluoro-2-[(pentafluoroethyl)thio]ethane. Representative chlorocarbons, alcohols,
20 ethers, ketones, and esters are listed in Table 1.

TABLE 1

<u>Compound</u>	<u>Chemical Formula</u>	<u>CAS Reg. No.</u>
Chlorocarbons		
1,1-dichloroethane	CH ₃ CHCl ₂	75-34-3
Trans-1,2-dichloroethylene	CHCl=CHCl	156-60-5
Ketones		
acetone	(CH ₃) ₂ C=O	67-64-1
Ethers		
tert-butyl methyl ether	CH ₃ OC(CH ₃) ₃	1634-04-4
diisopropyl ether	(CH ₃) ₂ CHOCH(CH ₃) ₂	108-20-3
1,2-dimethoxyethane	CH ₃ OCH ₂ CH ₂ OCH ₃	110-71-9
dimethoxymethane	CH ₃ OCH ₂ OCH ₃	109-87-5
1,1-dimethoxyethane	(CH ₃ O) ₂ CHCH ₃	534-15-6
Alcohols		
methanol	CH ₃ OH	67-56-1
ethanol	CH ₃ CH ₂ OH	64-17-5
n-propanol	CH ₃ CH ₂ CH ₂ OH	71-23-8
isopropanol	CH ₃ CH ₂ OHCH ₃	67-63-0
nonafluoro-tert-butanol	(CF ₃) ₃ COH	2378-02-1
Esters		
ethyl acetate	CH ₃ COOCH ₂ CH ₃	141-78-6
ethyl formate	HCOOCH ₂ CH ₃	109-94-4
methyl acetate	CH ₃ COOCH ₃	79-20-9
methyl formate	HCOOCH ₃	107-31-3

Other		
N-(difluoromethyl)-N,N-dimethylamine	$N(CH_3)_2(CHF_2)$	683-81-8
1,1,1,2,2-pentafluoro-2-[(pentafluoroethyl)thio]ethane	$CF_3CF_2SCF_2CF_3$	155953-22-3

The compounds listed in Table 1 are available commercially or may be prepared by processes known in the art.

The compositions of the present invention that are mixtures
 5 may be prepared by any convenient method to combine the desired amounts of the individual components. A preferred method is to weigh the desired component amounts and thereafter combine the components in an appropriate vessel. Agitation may be used, if desired.

Compositions of the present invention have low or zero ozone
 10 depletion potential and low global warming potential. For example, chlorocarbons, alcohols, ketones, ethers, esters, N-(difluoromethyl)-N,N-dimethylamine, and 1,1,1,2,2-pentafluoro-2-[(pentafluoroethyl)thio]ethane, alone or in mixtures, will have global warming potentials lower than many HFC refrigerants currently in use.

15 The refrigerant or heat transfer compositions of the present invention include:

nonafluoro-tert-butanol;
 1,1-dichloroethane;
 trans-1,2-dichloroethylene;
 20 acetone;
 tert-butyl methyl ether;
 diisopropyl ether;
 1,2-dimethoxyethane;
 dimethoxymethane;
 25 1,1-dimethoxyethane;
 methanol;
 ethanol;
 n-propanol;
 isopropanol;
 30 ethyl acetate;
 ethyl formate;
 methyl acetate;
 methyl formate;
 N-(difluoromethyl)-N,N-dimethylamine;
 35 1,1,1,2,2-pentafluoro-2-[(pentafluoroethyl)thio]ethane; and
 combinations thereof.

The compositions of the present invention may further comprise about 0.01 weight percent to about 5 weight percent of a stabilizer, free radical scavenger or antioxidant. Such additives include
5 but are not limited to, nitromethane, hindered phenols, hydroxylamines, thiols, phosphites, or lactones. Single additives or combinations may be used.

The compositions of the present invention may further comprise about 0.01 weight percent to about 5 weight percent of a water
10 scavenger (drying compound). Such water scavengers may comprise ortho esters such as trimethyl-, triethyl-, or tripropylorthoformate.

The compositions of the present invention may further comprise an ultra-violet (UV) dye and optionally a solubilizing agent. The UV dye is a useful component for detecting leaks of the refrigerant and
15 heat transfer fluid compositions by permitting one to observe the fluorescence of the dye in the refrigerant or heat transfer fluid compositions at a leak point or in the vicinity of refrigeration or air-conditioning apparatus. One may observe the fluorescence of the dye under an ultra-violet light. Solubilizing agents may be needed to increase
20 solubility of such UV dyes in some refrigerants and heat transfer fluids.

By "ultra-violet" dye is meant a UV fluorescent composition that absorbs light in the ultra-violet or "near" ultra-violet region of the electromagnetic spectrum. The fluorescence produced by the UV fluorescent dye under illumination by a UV light that emits radiation with
25 wavelength anywhere from 10 nanometer to 750 nanometer may be detected. Therefore, if refrigerant or heat transfer fluid containing such a UV fluorescent dye is leaking from a given point in a refrigeration or air-conditioning apparatus, the fluorescence can be detected at the leak point. Such UV fluorescent dyes include but are not limited to naphthalimides,
30 perylenes, coumarins, anthracenes, phenanthracenes, xanthenes, thioxanthenes, naphthoxanthenes, fluoresceins, and derivatives or combinations thereof. Solubilizing agents of the present invention comprise at least one compound selected from the group consisting of hydrocarbons, hydrocarbon ethers, polyoxyalkylene glycol ethers, amides,

nitrides, ketones, chlorocarbons, esters, lactones, aryl ethers, fluoroethers and 1,1,1-trifluoroalkanes.

Hydrocarbon solubilizing agents of the present invention comprise hydrocarbons including straight chained, branched chain or
 5 cyclic alkanes or alkenes containing 16 or fewer carbon atoms and only hydrogen with no other functional groups. Representative hydrocarbon solubilizing agents comprise propane, propylene, cyclopropane, n-butane, isobutane, n-pentane, octane, decane, and hexadecane.

Hydrocarbon ether solubilizing agents of the present invention
 10 comprise ethers containing only carbon, hydrogen and oxygen, such as dimethyl ether (DME).

Polyoxyalkylene glycol ether solubilizing agents of the present invention are represented by the formula $R^1[(OR^2)_xOR^3]_y$, wherein x is an integer from 1-3; y is an integer from 1-4; R^1 is selected from hydrogen
 15 and aliphatic hydrocarbon radicals having 1 to 6 carbon atoms and y bonding sites; R^2 is selected from aliphatic hydrocarbylene radicals having from 2 to 4 carbon atoms; R^3 is selected from hydrogen, and aliphatic and alicyclic hydrocarbon radicals having from 1 to 6 carbon atoms; at least one of R^1 and R^3 is selected from said hydrocarbon radical; and wherein
 20 said polyoxyalkylene glycol ethers have a molecular weight of from about 100 to about 300 atomic mass units. As used herein, bonding sites mean radical sites available to form covalent bonds with other radicals.

Hydrocarbylene radicals mean divalent hydrocarbon radicals.

In the present invention, preferable polyoxyalkylene glycol
 25 ether solubilizing agents are represented by $R^1[(OR^2)_xOR^3]_y$ wherein x is preferably 1-2; y is preferably 1; R^1 and R^3 are preferably independently selected from hydrogen and aliphatic hydrocarbon radicals having 1 to 4 carbon atoms; R^2 is preferably selected from aliphatic hydrocarbylene radicals having from 2 or 3 carbon atoms, most preferably 3 carbon atoms;
 30 the polyoxyalkylene glycol ether molecular weight is preferably from about 100 to about 250 atomic mass units, most preferably from about 125 to about 250 atomic mass units. The R^1 and R^3 hydrocarbon radicals having 1 to 6 carbon atoms may be linear, branched or cyclic. Representative R^1 and R^3 hydrocarbon radicals include methyl, ethyl, propyl, isopropyl, butyl,

isobutyl, sec-butyl, ~~tert-butyl~~, pentyl, isopentyl, neopentyl, ~~tert~~-pentyl, cyclopentyl, and cyclohexyl. Where free hydroxyl radicals on the present polyoxyalkylene glycol ether solubilizing agents may be incompatible with certain compression refrigeration apparatus materials of construction (e.g. Mylar®), R¹ and R³ are preferably aliphatic hydrocarbon radicals having 1 to 4 carbon atoms, most preferably 1 carbon atom. The R² aliphatic hydrocarbylene radicals having from 2 to 4 carbon atoms form repeating oxyalkylene radicals - (OR²)_x - that include oxyethylene radicals, oxypropylene radicals, and oxybutylene radicals. The oxyalkylene radical comprising R² in one polyoxyalkylene glycol ether solubilizing agent molecule may be the same, or one molecule may contain different R² oxyalkylene groups. The present polyoxyalkylene glycol ether solubilizing agents preferably comprise at least one oxypropylene radical. Where R¹ is an aliphatic or alicyclic hydrocarbon radical having 1 to 6 carbon atoms and y bonding sites, the radical may be linear, branched or cyclic. Representative R¹ aliphatic hydrocarbon radicals having two bonding sites include, for example, an ethylene radical, a propylene radical, a butylene radical, a pentylene radical, a hexylene radical, a cyclopentylene radical and a cyclohexylene radical. Representative R¹ aliphatic hydrocarbon radicals having three or four bonding sites include residues derived from polyalcohols, such as trimethylolpropane, glycerin, pentaerythritol, 1,2,3-trihydroxycyclohexane and 1,3,5-trihydroxycyclohexane, by removing their hydroxyl radicals.

Representative polyoxyalkylene glycol ether solubilizing agents include but are not limited to: CH₃OCH₂CH(CH₃)O(H or CH₃) (propylene glycol methyl (or dimethyl) ether), CH₃O[CH₂CH(CH₃)O]₂(H or CH₃) (dipropylene glycol methyl (or dimethyl) ether), CH₃O[CH₂CH(CH₃)O]₃(H or CH₃) (tripropylene glycol methyl (or dimethyl) ether), C₂H₅OCH₂CH(CH₃)O(H or C₂H₅) (propylene glycol ethyl (or diethyl) ether), C₂H₅O[CH₂CH(CH₃)O]₂(H or C₂H₅) (dipropylene glycol ethyl (or diethyl) ether), C₂H₅O[CH₂CH(CH₃)O]₃(H or C₂H₅) (tripropylene glycol ethyl (or diethyl) ether), C₃H₇OCH₂CH(CH₃)O(H or C₃H₇) (propylene glycol n-propyl (or di-n-propyl) ether), C₃H₇O[CH₂CH(CH₃)O]₂(H or C₃H₇) (dipropylene glycol n-propyl (or di-n-propyl) ether), C₃H₇O[CH₂CH(CH₃)O]₃(H or C₃H₇)

- (tripropylene glycol n-propyl (or di-n-propyl) ether), $C_4H_9OCH_2CH(CH_3)OH$ (propylene glycol n-butyl ether), $C_4H_9O[CH_2CH(CH_3)O]_2(H \text{ or } C_4H_9)$ (dipropylene glycol n-butyl (or di-n-butyl) ether), $C_4H_9O[CH_2CH(CH_3)O]_3(H \text{ or } C_4H_9)$ (tripropylene glycol n-butyl (or di-n-butyl) ether),
- 5 (CH₃)₃COCH₂CH(CH₃)OH (propylene glycol t-butyl ether), (CH₃)₃CO[CH₂CH(CH₃)O]₂(H or (CH₃)₃) (dipropylene glycol t-butyl (or di-t-butyl) ether), (CH₃)₃CO[CH₂CH(CH₃)O]₃(H or (CH₃)₃) (tripropylene glycol t-butyl (or di-t-butyl) ether), C₅H₁₁OCH₂CH(CH₃)OH (propylene glycol n-pentyl ether), C₄H₉OCH₂CH(C₂H₅)OH (butylene glycol n-butyl ether),
- 10 C₄H₉O[CH₂CH(C₂H₅)O]₂H (dibutylene glycol n-butyl ether), trimethylolpropane tri-n-butyl ether (C₂H₅C(CH₂O(CH₂)₃CH₃)₃) and trimethylolpropane di-n-butyl ether (C₂H₅C(CH₂OC(CH₂)₃CH₃)₂CH₂OH).

Amide solubilizing agents of the present invention comprise those represented by the formulae R¹C(O)NR²R³ and cyclo-[R⁴C(O)N(R⁵)-

15], wherein R¹, R², R³ and R⁵ are independently selected from aliphatic and alicyclic hydrocarbon radicals having from 1 to 12 carbon atoms; R⁴ is selected from aliphatic hydrocarbylene radicals having from 3 to 12 carbon atoms; and wherein said amides have a molecular weight of from about 100 to about 300 atomic mass units. The molecular weight of said amides

20 is preferably from about 160 to about 250 atomic mass units. R¹, R², R³ and R⁵ may optionally include substituted hydrocarbon radicals, that is, radicals containing non-hydrocarbon substituents selected from halogens (e.g., fluorine, chlorine) and alkoxides (e.g. methoxy). R¹, R², R³ and R⁵ may optionally include heteroatom-substituted hydrocarbon radicals, that

25 is, radicals, which contain the atoms nitrogen (aza-), oxygen (oxa-) or sulfur (thia-) in a radical chain otherwise composed of carbon atoms. In general, no more than three non-hydrocarbon substituents and heteroatoms, and preferably no more than one, will be present for each 10 carbon atoms in R¹⁻³, and the presence of any such non-hydrocarbon

30 substituents and heteroatoms must be considered in applying the aforementioned molecular weight limitations. Preferred amide solubilizing agents consist of carbon, hydrogen, nitrogen and oxygen. Representative R¹, R², R³ and R⁵ aliphatic and alicyclic hydrocarbon radicals include methyl, ethyl, propyl, isopropyl, butyl, isobutyl, sec-butyl, *tert*-butyl, pentyl,

isopentyl, neopentyl, ~~tert-pentyl~~, cyclopentyl, cyclohexyl, heptyl, octyl, nonyl, decyl, undecyl, dodecyl and their configurational isomers. A preferred embodiment of amide solubilizing agents are those wherein R^4 in the aforementioned formula $\text{cyclo}[\text{R}^4\text{C}(\text{O})\text{N}(\text{R}^5)-]$ may be represented

5 by the hydrocarbylene radical $(\text{CR}^6\text{R}^7)_n$, in other words, the formula $\text{cyclo}[(\text{CR}^6\text{R}^7)_n\text{C}(\text{O})\text{N}(\text{R}^5)-]$ wherein the previously-stated values for molecular weight apply; n is an integer from 3 to 5; R^5 is a saturated hydrocarbon radical containing 1 to 12 carbon atoms; R^6 and R^7 are independently selected (for each n) by the rules previously offered defining R^{1-3} . In the

10 lactams represented by the formula: $\text{cyclo}[(\text{CR}^6\text{R}^7)_n\text{C}(\text{O})\text{N}(\text{R}^5)-]$, all R^6 and R^7 are preferably hydrogen, or contain a single saturated hydrocarbon radical among the n methylene units, and R^5 is a saturated hydrocarbon radical containing 3 to 12 carbon atoms. For example, 1-(saturated hydrocarbon radical)-5-methylpyrrolidin-2-ones.

15 Representative amide solubilizing agents include but are not limited to: 1-octylpyrrolidin-2-one, 1-decylpyrrolidin-2-one, 1-octyl-5-methylpyrrolidin-2-one, 1-butylcaprolactam, 1-cyclohexylpyrrolidin-2-one, 1-butyl-5-methylpiperid-2-one, 1-pentyl-5-methylpiperid-2-one, 1-hexylcaprolactam, 1-hexyl-5-methylpyrrolidin-2-one, 5-methyl-1-

20 pentylpiperid-2-one, 1,3-dimethylpiperid-2-one, 1-methylcaprolactam, 1-butyl-pyrrolidin-2-one, 1,5-dimethylpiperid-2-one, 1-decyl-5-methylpyrrolidin-2-one, 1-dodecylpyrrolid-2-one, N,N-dibutylformamide and N,N-diisopropylacetamide.

Ketone solubilizing agents of the present invention comprise

25 ketones represented by the formula $\text{R}^1\text{C}(\text{O})\text{R}^2$, wherein R^1 and R^2 are independently selected from aliphatic, alicyclic and aryl hydrocarbon radicals having from 1 to 12 carbon atoms, and wherein said ketones have a molecular weight of from about 70 to about 300 atomic mass units. R^1 and R^2 in said ketones are preferably independently selected from

30 aliphatic and alicyclic hydrocarbon radicals having 1 to 9 carbon atoms. The molecular weight of said ketones is preferably from about 100 to 200 atomic mass units. R^1 and R^2 may together form a hydrocarbylene radical connected and forming a five, six, or seven-membered ring cyclic ketone, for example, cyclopentanone, cyclohexanone, and cycloheptanone. R^1

and R^2 may optionally include substituted hydrocarbon radicals, that is, radicals containing non-hydrocarbon substituents selected from halogens (e.g., fluorine, chlorine) and alkoxides (e.g. methoxy). R^1 and R^2 may optionally include heteroatom-substituted hydrocarbon radicals, that is, radicals, which contain the atoms nitrogen (aza-), oxygen (keto-, oxa-) or sulfur (thia-) in a radical chain otherwise composed of carbon atoms. In general, no more than three non-hydrocarbon substituents and heteroatoms, and preferably no more than one, will be present for each 10 carbon atoms in R^1 and R^2 , and the presence of any such non-hydrocarbon substituents and heteroatoms must be considered in applying the aforementioned molecular weight limitations. Representative R^1 and R^2 aliphatic, alicyclic and aryl hydrocarbon radicals in the general formula $R^1C(O)R^2$ include methyl, ethyl, propyl, isopropyl, butyl, isobutyl, sec-butyl, *tert*-butyl, pentyl, isopentyl, neopentyl, *tert*-pentyl, cyclopentyl, cyclohexyl, heptyl, octyl, nonyl, decyl, undecyl, dodecyl and their configurational isomers, as well as phenyl, benzyl, cumenyl, mesityl, tolyl, xylyl and phenethyl.

Representative ketone solubilizing agents include but are not limited to: 2-butanone, 2-pentanone, acetophenone, butyrophenone, hexanophenone, cyclohexanone, cycloheptanone, 2-heptanone, 3-heptanone, 5-methyl-2-hexanone, 2-octanone, 3-octanone, diisobutyl ketone, 4-ethylcyclohexanone, 2-nonanone, 5-nonanone, 2-decanone, 4-decanone, 2-decalone, 2-tridecanone, dihexyl ketone and dicyclohexyl ketone.

Nitrile solubilizing agents of the present invention comprise nitriles represented by the formula R^1CN , wherein R^1 is selected from aliphatic, alicyclic or aryl hydrocarbon radicals having from 5 to 12 carbon atoms, and wherein said nitriles have a molecular weight of from about 90 to about 200 atomic mass units. R^1 in said nitrile solubilizing agents is preferably selected from aliphatic and alicyclic hydrocarbon radicals having 8 to 10 carbon atoms. The molecular weight of said nitrile solubilizing agents is preferably from about 120 to about 140 atomic mass units. R^1 may optionally include substituted hydrocarbon radicals, that is, radicals containing non-hydrocarbon substituents selected from halogens

(e.g., ~~fluorine, chlorine~~) and alkoxides (e.g. methoxy). R^1 may optionally include heteroatom-substituted hydrocarbon radicals, that is, radicals, which contain the atoms nitrogen (aza-), oxygen (keto-, oxa-) or sulfur (thia-) in a radical chain otherwise composed of carbon atoms. In general,

5 no more than three non-hydrocarbon substituents and heteroatoms, and preferably no more than one, will be present for each 10 carbon atoms in R^1 , and the presence of any such non-hydrocarbon substituents and heteroatoms must be considered in applying the aforementioned molecular weight limitations. Representative R^1 aliphatic, alicyclic and aryl

10 hydrocarbon radicals in the general formula R^1CN include pentyl, isopentyl, neopentyl, *tert*-pentyl, cyclopentyl, cyclohexyl, heptyl, octyl, nonyl, decyl, undecyl, dodecyl and their configurational isomers, as well as phenyl, benzyl, cumenyl, mesityl, tolyl, xylyl and phenethyl.

Representative nitrile solubilizing agents include but are not limited to: 1-

15 cyanopentane, 2,2-dimethyl-4-cyanopentane, 1-cyanoheptane, 1-cyanoheptane, 1-cyanooctane, 2-cyanooctane, 1-cyanononane, 1-cyanodecane, 2-cyanodecane, 1-cyanoundecane and 1-cyanododecane.

Chlorocarbon solubilizing agents of the present invention comprise chlorocarbons represented by the formula RCI_x , wherein x is 1 or

20 2; R is selected from aliphatic and alicyclic hydrocarbon radicals having 1 to 12 carbon atoms; and wherein said chlorocarbons have a molecular weight of from about 100 to about 200 atomic mass units. The molecular weight of said chlorocarbon solubilizing agents is preferably from about 120 to 150 atomic mass units. Representative R aliphatic and alicyclic

25 hydrocarbon radicals in the general formula RCI_x include methyl, ethyl, propyl, isopropyl, butyl, isobutyl, *sec*-butyl, *tert*-butyl, pentyl, isopentyl, neopentyl, *tert*-pentyl, cyclopentyl, cyclohexyl, heptyl, octyl, nonyl, decyl, undecyl, dodecyl and their configurational isomers.

Representative chlorocarbon solubilizing agents include but are

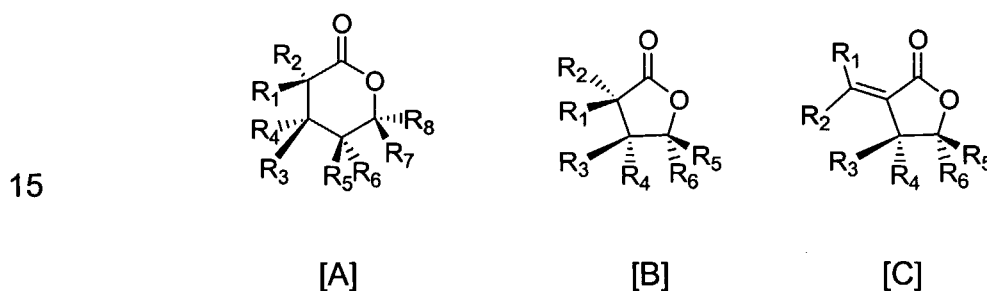
30 not limited to: 3-(chloromethyl)pentane, 3-chloro-3-methylpentane, 1-chlorohexane, 1,6-dichlorohexane, 1-chloroheptane, 1-chlorooctane, 1-chlorononane, 1-chlorodecane, and 1,1,1-trichlorodecane.

Ester solubilizing agents of the present invention comprise esters represented by the general formula $R^1C(O)OR^2$, wherein R^1 and R^2

are independently selected from linear and cyclic, saturated and unsaturated, alkyl and aryl radicals. Preferred esters consist essentially of the elements C, H and O, have a molecular weight of from about 80 to about 550 atomic mass units.

- 5 Representative esters include but are not limited to:
 $(\text{CH}_3)_2\text{CHCH}_2\text{O}(\text{O})\text{C}(\text{CH}_2)_{2-4}(\text{O})\text{COCH}_2\text{CH}(\text{CH}_3)_2$ (diisobutyl dibasic ester),
 ethyl hexanoate, ethyl heptanoate, n-butyl propionate, n-propyl propionate,
 ethyl benzoate, di-n-propyl phthalate, benzoic acid ethoxyethyl ester,
 dipropyl carbonate, "Exxate 700" (a commercial C_7 alkyl acetate), "Exxate
 10 800" (a commercial C_8 alkyl acetate), dibutyl phthalate, and tert-butyl
 acetate.

Lactone solubilizing agents of the present invention comprise lactones represented by structures [A], [B], and [C]:

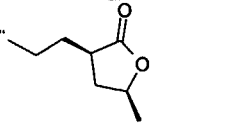
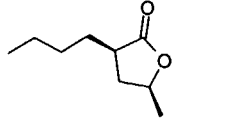
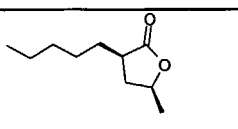
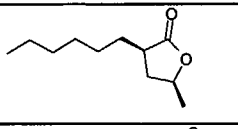
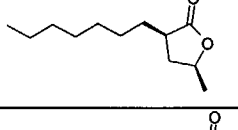
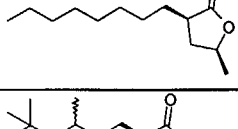
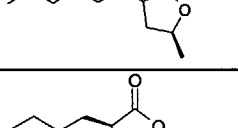
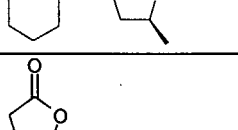
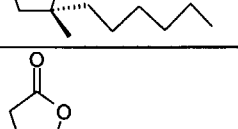
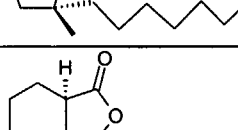
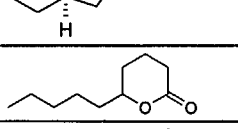
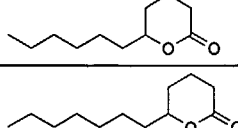
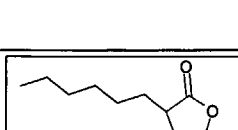
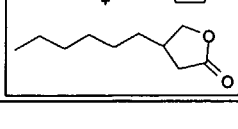
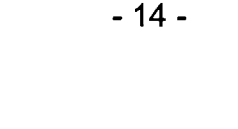


- These lactones contain the functional group $-\text{C}(\text{O})\text{O}-$ in a ring
 20 of six (A), or preferably five atoms (B), wherein for structures [A] and [B],
 R_1 through R_8 are independently selected from hydrogen or linear,
 branched, cyclic, bicyclic, saturated and unsaturated hydrocarbyl radicals.
 Each R_1 through R_8 may be connected forming a ring with another R_1
 through R_8 . The lactone may have an exocyclic alkylidene group as in
 25 structure [C], wherein R_1 through R_6 are independently selected from
 hydrogen or linear, branched, cyclic, bicyclic, saturated and unsaturated
 hydrocarbyl radicals. Each R_1 through R_6 may be connected forming a ring
 with another R_1 through R_6 . The lactone solubilizing agents have a
 molecular weight range of from about 80 to about 300 atomic mass units,
 30 preferred from about 80 to about 200 atomic mass units.

Representative lactone solubilizing agents include but are not limited to the compounds listed in Table 2.

TABLE 2

Additive	Molecular Structure	Molecular Formula	Molecular Weight (amu)
(E,Z)-3-ethylidene-5-methyl-dihydro-furan-2-one		C ₇ H ₁₀ O ₂	126
(E,Z)-3-propylidene-5-methyl-dihydro-furan-2-one		C ₈ H ₁₂ O ₂	140
(E,Z)-3-butylidene-5-methyl-dihydro-furan-2-one		C ₉ H ₁₄ O ₂	154
(E,Z)-3-pentylidene-5-methyl-dihydro-furan-2-one		C ₁₀ H ₁₆ O ₂	168
(E,Z)-3-Hexylidene-5-methyl-dihydro-furan-2-one		C ₁₁ H ₁₈ O ₂	182
(E,Z)-3-Heptylidene-5-methyl-dihydro-furan-2-one		C ₁₂ H ₂₀ O ₂	196
(E,Z)-3-octylidene-5-methyl-dihydro-furan-2-one		C ₁₃ H ₂₂ O ₂	210
(E,Z)-3-nonylidene-5-methyl-dihydro-furan-2-one		C ₁₄ H ₂₄ O ₂	224
(E,Z)-3-decylidene-5-methyl-dihydro-furan-2-one		C ₁₅ H ₂₆ O ₂	238
(E,Z)-3-(3,5,5-trimethylhexylidene)-5-methyl-dihydrofuran-2-one		C ₁₄ H ₂₄ O ₂	224
(E,Z)-3-cyclohexylmethylidene-5-methyl-dihydrofuran-2-one		C ₁₂ H ₁₈ O ₂	194
gamma-octalactone		C ₈ H ₁₄ O ₂	142
gamma-nonolactone		C ₉ H ₁₆ O ₂	156
gamma-decalactone		C ₁₀ H ₁₈ O ₂	170
gamma-undecalactone		C ₁₁ H ₂₀ O ₂	184
gamma-dodecalactone		C ₁₂ H ₂₂ O ₂	198
3-hexyldihydro-furan-2-one		C ₁₀ H ₁₈ O ₂	170
3-heptyldihydro-furan-2-one		C ₁₁ H ₂₀ O ₂	184
cis-3-ethyl-5-methyl-dihydro-furan-2-one		C ₇ H ₁₂ O ₂	128

<i>cis</i> -(3-propyl-5-methyl)-dihydro-furan-2-one		C ₈ H ₁₄ O ₂	142
<i>cis</i> -(3-butyl-5-methyl)-dihydro-furan-2-one		C ₉ H ₁₆ O ₂	156
<i>cis</i> -(3-pentyl-5-methyl)-dihydro-furan-2-one		C ₁₀ H ₁₈ O ₂	170
<i>cis</i> -(3-hexyl-5-methyl)-dihydro-furan-2-one		C ₁₁ H ₂₀ O ₂	184
<i>cis</i> -(3-heptyl-5-methyl)-dihydro-furan-2-one		C ₁₂ H ₂₂ O ₂	198
<i>cis</i> -(3-octyl-5-methyl)-dihydro-furan-2-one		C ₁₃ H ₂₄ O ₂	212
<i>cis</i> -(3-(3,5,5-trimethylhexyl)-5-methyl)-dihydro-furan-2-one		C ₁₄ H ₂₆ O ₂	226
<i>cis</i> -(3-cyclohexylmethyl-5-methyl)-dihydro-furan-2-one		C ₁₂ H ₂₀ O ₂	196
5-methyl-5-hexyl-dihydro-furan-2-one		C ₁₁ H ₂₀ O ₂	184
5-methyl-5-octyl-dihydro-furan-2-one		C ₁₃ H ₂₄ O ₂	212
Hexahydro-isobenzofuran-1-one		C ₈ H ₁₂ O ₂	140
<i>delta</i> -decalactone		C ₁₀ H ₁₈ O ₂	170
<i>delta</i> -undecalactone		C ₁₁ H ₂₀ O ₂	184
<i>delta</i> -dodecalactone		C ₁₂ H ₂₂ O ₂	198
mixture of 4-hexyl-dihydrofuran-2-one and 3-hexyl-dihydro-furan-2-one		C ₁₀ H ₁₈ O ₂	170

Lactone solubilizing agents generally have a kinematic viscosity of less than about 7 centistokes at 40°C. For instance, gamma-undecalactone has kinematic viscosity of 5.4 centistokes and cis-(3-hexyl-5-methyl)dihydrofuran-2-one has viscosity of 4.5 centistokes, both at 40°C. Lactone solubilizing agents may be available commercially or prepared by methods as described in U. S. patent application 10/910,495, filed August 3, 2004, incorporated herein by reference.

Aryl ether solubilizing agents of the present invention comprise aryl ethers represented by the formula R^1OR^2 , wherein: R^1 is selected from aryl hydrocarbon radicals having from 6 to 12 carbon atoms; R^2 is selected from aliphatic hydrocarbon radicals having from 1 to 4 carbon atoms; and wherein said aryl ethers have a molecular weight of from about 100 to about 150 atomic mass units. Representative R^1 aryl radicals in the general formula R^1OR^2 include phenyl, biphenyl, cumenyl, mesityl, tolyl, xylyl, naphthyl and pyridyl. Representative R^2 aliphatic hydrocarbon radicals in the general formula R^1OR^2 include methyl, ethyl, propyl, isopropyl, butyl, isobutyl, sec-butyl and *tert*-butyl. Representative aromatic ether solubilizing agents include but are not limited to: methyl phenyl ether (anisole), 1,3-dimethoxybenzene, ethyl phenyl ether and butyl phenyl ether.

Fluoroether solubilizing agents of the present invention comprise those represented by the general formula $R^1OCF_2CF_2H$, wherein R^1 is selected from aliphatic, alicyclic, and aromatic hydrocarbon radicals having from about 5 to about 15 carbon atoms, preferably primary, linear, saturated, alkyl radicals. Representative fluoroether solubilizing agents include but are not limited to: $C_8H_{17}OCF_2CF_2H$ and $C_6H_{13}OCF_2CF_2H$. It should be noted that if the refrigerant is a fluoroether, then the solubilizing agent may not be the same fluoroether.

Fluoroether solubilizing agents may further comprise ethers derived from fluoro-olefins and polyols. The fluoro-olefins may be of the type $CF_2=CXY$, wherein X is hydrogen, chlorine or fluorine, and Y is chlorine, fluorine, CF_3 or OR_f , wherein R_f is CF_3 , C_2F_5 , or C_3F_7 . Representative fluoro-olefins are tetrafluoroethylene,

chlorotrifluoroethylene, hexafluoropropylene, and perfluoromethylvinyl ether. The polyols may be linear or branched. Linear polyols may be of the type $\text{HOCH}_2(\text{CHOH})_x(\text{CRR}')_y\text{CH}_2\text{OH}$, wherein R and R' are hydrogen, or CH_3 , or C_2H_5 and wherein x is an integer from 0-4, and y is an integer

5 from 0-4. Branched polyols may be of the type $\text{C}(\text{OH})_t(\text{R})_u(\text{CH}_2\text{OH})_v[(\text{CH}_2)_m\text{CH}_2\text{OH}]_w$, wherein R may be hydrogen, CH_3 or C_2H_5 , m may be an integer from 0 to 3, t and u may be 0 or 1, v and w are integers from 0 to 4, and also wherein $t + u + v + w = 4$.

Representative polyols are trimethylol propane, pentaerythritol, butanediol,

10 and ethylene glycol.

1,1,1-Trifluoroalkane solubilizing agents of the present invention comprise 1,1,1-trifluoroalkanes represented by the general formula CF_3R^1 , wherein R^1 is selected from aliphatic and alicyclic hydrocarbon radicals having from about 5 to about 15 carbon atoms,

15 preferably primary, linear, saturated alkyl radicals. Representative 1,1,1-trifluoroalkane solubilizing agents include but are not limited to: 1,1,1-trifluorohexane and 1,1,1-trifluorododecane.

Solubilizing agents of the present invention may be present as a single compound, or may be present as a mixture of more than one

20 solubilizing agent. Mixtures of solubilizing agents may contain two solubilizing agents from the same class of compounds, say two lactones, or two solubilizing agents from two different classes, such as a lactone and a polyoxyalkylene glycol ether.

In the present compositions comprising refrigerant and UV

25 fluorescent dye, or comprising heat transfer fluid and UV fluorescent dye, from about 0.001 weight percent to about 1.0 weight percent of the compositions is UV dye, preferably from about 0.005 weight percent to about 0.5 weight percent, and most preferably from 0.01 weight percent to about 0.25 weight percent.

30 Solubility of these UV fluorescent dyes in refrigerants and heat transfer fluids may be poor. Therefore, methods for introducing these dyes into the refrigeration or air-conditioning apparatus have been awkward, costly and time consuming. US patent no. RE 36,951 describes a method, which utilizes a dye powder, solid pellet or slurry of dye that

may be inserted into a component of the refrigeration or air-conditioning apparatus. As refrigerant and lubricant are circulated through the apparatus, the dye is dissolved or dispersed and carried throughout the apparatus. Numerous other methods for introducing dye into a
5 refrigeration or air-conditioning apparatus are described in the literature.

Ideally, the UV fluorescent dye could be dissolved in the refrigerant itself thereby not requiring any specialized method for introduction to the refrigeration or air-conditioning apparatus. The present invention relates to compositions including UV fluorescent dye, which may
10 be introduced into the system in the refrigerant. The inventive compositions will allow the storage and transport of dye-containing refrigerant and heat transfer fluid even at low temperatures while maintaining the dye in solution.

In the present compositions comprising refrigerant, UV
15 fluorescent dye and solubilizing agent, or comprising heat transfer fluid, UV fluorescent dye and solubilizing agent, from about 1 to about 50 weight percent, preferably from about 2 to about 25 weight percent, and most preferably from about 5 to about 15 weight percent of the combined composition is solubilizing agent in the refrigerant or heat transfer fluid. In
20 the compositions of the present invention the UV fluorescent dye is present in a concentration from about 0.001 weight percent to about 1.0 weight percent in the refrigerant or heat transfer fluid, preferably from 0.005 weight percent to about 0.5 weight percent, and most preferably from 0.01 weight percent to about 0.25 weight percent.

25 Optionally, commonly used refrigeration or air-conditioning system additives may be added, as desired, to compositions of the present invention in order to enhance performance and system stability. These additives are known in the field of refrigeration and air-cinditioning, and include, but are not limited to, anti wear agents, extreme pressure
30 lubricants, corrosion and oxidation inhibitors, metal surface deactivators, free radical scavengers, and foam control agents. In general, these additives are present in the inventive compositions in small amounts relative to the overall composition. Typically concentrations of from less than about 0.1 weight percent to as much as about 3 weight percent of

each additive are used. These additives are selected on the basis of the individual system requirements. These additives include members of the triaryl phosphate family of EP (extreme pressure) lubricity additives, such as butylated triphenyl phosphates (BTPP), or other alkylated triaryl phosphate esters, e.g. Syn-0-Ad 8478 from Akzo Chemicals, tricresyl phosphates and related compounds. Additionally, the metal dialkyl dithiophosphates (e.g. zinc dialkyl dithiophosphate (or ZDDP), Lubrizol 1375 and other members of this family of chemicals may be used in compositions of the present invention. Other antiwear additives include natural product oils and asymmetrical polyhydroxyl lubrication additives, such as Synergol TMS (International Lubricants). Similarly, stabilizers such as anti oxidants, free radical scavengers, and water scavengers may be employed. Compounds in this category can include, but are not limited to, butylated hydroxy toluene (BHT) and epoxides.

Solubilizing agents such as ketones may have an objectionable odor, which can be masked by addition of an odor masking agent or fragrance. Typical examples of odor masking agents or fragrances may include Evergreen, Fresh Lemon, Cherry, Cinnamon, Peppermint, Floral or Orange Peel, all commercially available, as well as d-limonene and pinene. Such odor masking agents may be used at concentrations of from about 0.001% to as much as about 15% by weight based on the combined weight of odor masking agent and solubilizing agent.

The present invention further relates to a method of using the refrigerant or heat transfer fluid compositions further comprising ultraviolet fluorescent dye, and optionally, solubilizing agent, in refrigeration or air-conditioning apparatus. The method comprises introducing the refrigerant or heat transfer fluid composition into the refrigeration or air-conditioning apparatus. This may be done by dissolving the UV fluorescent dye in the refrigerant or heat transfer fluid composition in the presence of a solubilizing agent and introducing the combination into the apparatus. Alternatively, this may be done by combining solubilizing agent and UV fluorescent dye and introducing said combination into refrigeration or air-conditioning apparatus containing refrigerant and/or heat transfer fluid. The resulting composition may be used in the refrigeration or air-conditioning apparatus.

The present invention further relates to a method of using the refrigerant or heat transfer fluid compositions comprising ultraviolet fluorescent dye to detect leaks. The presence of the dye in the compositions allows for detection of leaking refrigerant in the refrigeration or air-conditioning apparatus. Leak detection helps to address, resolve or prevent inefficient operation of the apparatus or system or equipment failure. Leak detection also helps one contain chemicals used in the operation of the apparatus.

The method comprises providing the composition comprising refrigerant and ultra-violet fluorescent dye, or comprising heat transfer fluid and ultra-violet fluorescent dye as described herein, and optionally, a solubilizing agent as described herein, to refrigeration and air-conditioning apparatus and employing a suitable means for detecting the UV fluorescent dye-containing refrigerant. Suitable means for detecting the dye include, but are not limited to, ultra-violet lamp, often referred to as a "black light" or "blue light". Such ultra-violet lamps are commercially available from numerous sources specifically designed for this purpose. Once the ultra-violet fluorescent dye containing composition has been introduced to the refrigeration or air-conditioning apparatus and has been allowed to circulate throughout the system, a leak can be found by shining said ultra-violet lamp on the apparatus and observing the fluorescence of the dye in the vicinity of any leak point.

The present invention further relates to a method of using the compositions of the present invention for producing refrigeration or heat, wherein the method comprises producing refrigeration by evaporating said composition in the vicinity of a body to be cooled and thereafter condensing said composition; or producing heat by condensing said composition in the vicinity of the body to be heated and thereafter evaporating said composition. Where the composition of the present invention includes refrigeration or heat transfer fluid composition with an ultra-violet fluorescent dye, and/or a solubilizing agent, the refrigerant or heat transfer fluid component of the composition is evaporated and thereafter condensed to produce refrigeration, or condensed and thereafter evaporated to produce heat.

Mechanical refrigeration is primarily an application of thermodynamics wherein a cooling medium, such as a refrigerant, goes through a cycle so that it can be recovered for reuse. Commonly used

cycles include vapor-compression, absorption, steam-jet or steam-ejector, and air.

Vapor-compression refrigeration systems include an evaporator, a compressor, a condenser, and an expansion device. A vapor-compression cycle re-uses refrigerant in multiple steps producing a cooling effect in one step and a heating effect in a different step. The cycle can be described simply as follows. Liquid refrigerant enters an evaporator through an expansion device, and the liquid refrigerant boils in the evaporator at a low temperature to form a gas and produce cooling. The low-pressure gas enters a compressor where the gas is compressed to raise its pressure and temperature. The higher-pressure (compressed) gaseous refrigerant then enters the condenser in which the refrigerant condenses and discharges its heat to the environment. The refrigerant returns to the expansion device through which the liquid expands from the higher-pressure level in the condenser to the low-pressure level in the evaporator, thus repeating the cycle.

The present invention further relates to a process for producing cooling comprising evaporating the compositions of the present invention in the vicinity of a body to be cooled, and thereafter condensing said compositions.

The present invention further relates to a process for producing heat comprising condensing the compositions of the present invention in the vicinity of a body to be heated, and thereafter evaporating said compositions.

There are various types of compressors that may be used in refrigeration applications. Compressors can be generally classified as reciprocating, rotary, jet, centrifugal, scroll, screw or axial-flow, depending on the mechanical means to compress the fluid, or as positive-displacement (e.g., reciprocating, scroll or screw) or dynamic (e.g., centrifugal or jet), depending on how the mechanical elements act on the fluid to be compressed.

Either positive displacement or dynamic compressors may be used in the present inventive process. A centrifugal type compressor is the preferred equipment for the present refrigerant compositions.

A centrifugal compressor uses rotating elements to accelerate the refrigerant radially, and typically includes an impeller and diffuser housed in a casing. Centrifugal compressors usually take fluid in at an impeller eye, or central inlet of a circulating impeller, and accelerate it radially outward. Some static pressure rise occurs in the impeller, but most of the pressure rise occurs in the diffuser section of the casing, where velocity is converted to static pressure. Each impeller-diffuser set is a stage of the compressor. Centrifugal compressors are built with from 1 to 12 or more stages, depending on the final pressure desired and the volume of refrigerant to be handled.

The pressure ratio, or compression ratio, of a compressor is the ratio of absolute discharge pressure to the absolute inlet pressure.

Pressure delivered by a centrifugal compressor is practically constant over a relatively wide range of capacities.

Positive displacement compressors draw vapor into a chamber, and the chamber decreases in volume to compress the vapor. After being compressed, the vapor is forced from the chamber by further decreasing the volume of the chamber to zero or nearly zero. A positive displacement compressor can build up a pressure, which is limited only by the volumetric efficiency and the strength of the parts to withstand the pressure.

Unlike a positive displacement compressor, a centrifugal compressor depends entirely on the centrifugal force of the high-speed impeller to compress the vapor passing through the impeller. There is no positive displacement, but rather what is called dynamic-compression.

The pressure a centrifugal compressor can develop depends on the tip speed of the impeller. Tip speed is the speed of the impeller measured at its tip and is related to the diameter of the impeller and its revolutions per minute. The capacity of the centrifugal compressor is determined by the size of the passages through the impeller. This makes the size of the compressor more dependent on the pressure required than the capacity.

Because of its high-speed operation, a centrifugal compressor is fundamentally a high volume, low-pressure machine. A centrifugal

compressor works best with a low-pressure refrigerant, such as trichlorofluoromethane (CFC-11) or 1,2,2-trichlorotrifluoroethane (CFC-113).

5 Large centrifugal compressors typically operate at 3000 to 7000 revolutions per minute (rpm). Small turbine centrifugal compressors are designed for high speeds, from about 40,000 to about 70,000 (rpm), and have small impeller sizes, typically less than 0.15 meters (about 6 inches).

10 A multi-stage impeller may be used in a centrifugal compressor to improve compressor efficiency thus requiring less power in use. For a two-stage system, in operation, the discharge of the first stage impeller goes to the suction intake of a second impeller. Both impellers may operate by use of a single shaft (or axle). Each stage can build up a compression ratio of about 4 to 1; that is, the absolute discharge pressure
15 can be four times the absolute suction pressure. Several examples of two-stage centrifugal compressor systems, particularly for automotive applications, are described in US 5,065,990 and US 5,363,674, both incorporated herein by reference.

20 The compositions of the present invention suitable for use in a refrigeration or air-conditioning systems employing a centrifugal compressor comprise at least one of:

nonafluoro-tert-butanol;
1,1-dichloroethane;
trans-1,2-dichloroethylene;
25 acetone;
tert-butyl methyl ether;
diisopropyl ether;
1,2-dimethoxyethane;
dimethoxymethane;
30 1,1-dimethoxyethane;
methanol;
ethanol;
n-propanol;
isopropanol;
35 ethyl acetate;
ethyl formate;
methyl acetate;

~~methyl formate;~~

N-(difluoromethyl)-N,N-dimethylamine; or

1,1,1,2,2-pentafluoro-2-[(pentafluoroethyl)thio]ethane.

These above-listed compositions are also suitable for use in a
5 multi-stage centrifugal compressor, preferably a two-stage centrifugal
compressor apparatus.

The compositions of the present invention may be used in
stationary air-conditioning, heat pumps or mobile air-conditioning and
refrigeration systems. Stationary air-conditioning and heat pump
10 applications include window, ductless, ducted, packaged terminal, chillers
and commercial, including packaged rooftop. Refrigeration applications
include domestic or home refrigerators and freezers, ice machines, self-
contained coolers and freezers, walk-in coolers and freezers and transport
refrigeration systems.

15 The compositions of the present invention may additionally be
used in air-conditioning, heating and refrigeration systems that employ fin
and tube heat exchangers, microchannel heat exchangers and vertical or
horizontal single pass tube or plate type heat exchangers.

Conventional microchannel heat exchangers may not be ideal
20 for the low pressure refrigerant compositions of the present invention. The
low operating pressure and density result in high flow velocities and high
frictional losses in all components. In these cases, the evaporator design
may be modified. Rather than several microchannel slabs connected in
series (with respect to the refrigerant path) a single slab/single pass heat
25 exchanger arrangement may be used. Therefore, a preferred heat
exchanger for the low pressure refrigerants of the present invention is a
single slab/single pass heat exchanger.

In addition to two-stage or other multi-stage centrifugal
compressor apparatus, the following compositions of the present invention
30 are suitable for use in refrigeration or air-conditioning apparatus employing
a single slab/single pass heat exchanger:

nonafluoro-tert-butanol;

1,1-dichloroethane;

trans-1,2-dichloroethylene;

35 acetone;

tert-butyl methyl ether;

diisopropyl ether;

1,2-dimethoxyethane;

dimethoxymethane;
1,1-dimethoxyethane;
methanol;
ethanol;
5 n-propanol;
isopropanol;
ethyl acetate;
ethyl formate;
methyl acetate;
10 methyl formate;
N-(difluoromethyl)-N,N-dimethylamine;
1,1,1,2,2-pentafluoro-2-[(pentafluoroethyl)thio]ethane;
and combinations thereof.

The compositions of the present invention are particularly
15 useful in small turbine centrifugal compressors (mini-centrifugal
compressors), which can be used in auto and window air-conditioning,
heat pumps, or transport refrigeration, as well as other applications.
These high efficiency mini-centrifugal compressors may be driven by an
electric motor and can therefore be operated independently of the engine
20 speed. A constant compressor speed allows the system to provide a
relatively constant cooling capacity at all engine speeds. This provides an
opportunity for efficiency improvements especially at higher engine speeds
as compared to a conventional R-134a automobile air-conditioning
system. When the cycling operation of conventional systems at high
25 driving speeds is taken into account, the advantage of these low pressure
systems becomes even greater.

Alternatively, rather than use electrical power, the mini-
centrifugal compressor may be powered by an engine exhaust gas driven
turbine or a ratioed gear drive assembly with ratioed belt drive. The
30 electrical power available in current automobile design is about 14 volts,
but the new mini-centrifugal compressor requires electrical power of about
50 volts. Therefore, use of an alternative power source would be
advantageous. A refrigeration or air-conditioning apparatus powered by
an engine exhaust gas driven turbine is described in detail in U.S.
35 provisional patent application no. 60/658,915, filed March 4, 2005,
incorporated herein by reference. A refrigeration or air-conditioning
apparatus powered by a ratioed gear drive assembly is described in detail

in U.S. provisional patent application no. 60/663924, filed March 21, 2005, incorporated herein by reference.

5 The present invention further relates to a process to produce cooling comprising compressing a composition of the present invention, in a mini-centrifugal compressor powered by an engine exhaust gas driven turbine; condensing said composition; and thereafter evaporating said composition in the vicinity of a body to be cooled.

10 The present invention further relates to a process to produce cooling comprising compressing a composition of the present invention, in a mini-centrifugal compressor powered by a ratioed gear drive assembly with a ratioed belt drive; condensing said composition; and thereafter evaporating said composition in the vicinity of a body to be cooled.

15 Some of the low pressure refrigerant fluids of the present invention may be suitable as drop-in replacements for CFC-113 in existing centrifugal equipment.

20 The present invention further relates to a method for replacing CFC-113 in existing refrigeration apparatus or air-conditioning apparatus, said method comprising providing a composition of the present invention as the replacement.

The present invention further relates to a process for transfer of heat from a heat source to a heat sink wherein the compositions of the present invention serve as heat transfer fluids. Said process for heat transfer comprises transferring the compositions of the present invention from a heat source to a heat sink.

25 Heat transfer fluids are utilized to transfer, move or remove heat from one space, location, object or body to a different space, location, object or body by radiation, conduction, or convection. A heat transfer fluid may function as a secondary coolant by providing means of transfer for cooling (or heating) from a remote refrigeration (or heating) system. In some systems, the heat transfer fluid may remain in a constant state throughout the transfer process (i.e., not evaporate or condense). Alternatively, evaporative cooling processes may utilize heat transfer fluids as well.

35 A heat source may be defined as any space, location, object or body from which it is desirable to transfer, move or remove heat. Examples of heat sources may be spaces (open or enclosed) requiring

refrigeration or cooling, such as refrigerator or freezer cases in a supermarket, building spaces requiring air-conditioning, or the passenger compartment of an automobile requiring air-conditioning. A heat sink may be defined as any space, location, object or body capable of absorbing
5 heat. A vapor compression refrigeration system is one example of such a heat sink.

EXAMPLES**EXAMPLE 1****Tip Speed to Develop Pressure**

- 5 Tip speed can be estimated by making some fundamental relationships for refrigeration equipment that use centrifugal compressors. The torque an impeller ideally imparts to a gas is defined as

$$T = m(v_2 r_2 - v_1 r_1) \quad \text{Equation 1}$$

- 10 where

T = torque, Newton-meters

m = mass rate of flow, kg/sec

v_2 = tangential velocity of refrigerant leaving impeller (tip speed), meters/sec

- 15 r_2 = radius of exit impeller, meters

v_1 = tangential velocity of refrigerant entering impeller, meters/sec

r_1 = radius of inlet of impeller, meters

- 20 Assuming the refrigerant enters the impeller in an essentially axial direction, the tangential component of the velocity $v_1 = 0$, therefore

$$T = m v_2 r_2 \quad \text{Equation 2}$$

- 25 The power required at the shaft is the product of the torque and the rotative speed

$$P = T \omega \quad \text{Equation 3}$$

where

P = power, W

ω = angular velocity, radians/s

- 30 therefore,

$$P = T \omega = m v_2 r_2 \omega \quad \text{Equation 4}$$

At low refrigerant flow rates, the tip speed of the impeller and the tangential velocity of the refrigerant are nearly identical; therefore

$$\dot{m}_2 v_2 = \dot{W}_2$$

Equation 5

and

$$P = \dot{m}_2 v_2$$

Equation 6

5 Another expression for ideal power is the product of the mass rate of flow and the isentropic work of compression,

$$P = \dot{m} H_i (1000 \text{ J/kJ})$$

Equation 7

where

10 H_i = Difference in enthalpy of the refrigerant from a saturated vapor at the evaporating conditions to saturated condensing conditions, kJ/kg.

Combining the two expressions Equation 6 and 7 produces,

$$v_2 v_2 = 1000 H_i$$

Equation 8

15

Although Equation 8 is based on some fundamental assumptions, it provides a good estimate of the tip speed of the impeller and provides an important way to compare tip speeds of refrigerants.

20 The table below shows theoretical tip speeds that are calculated for 1,2,2-trichlorotrifluoroethane (CFC-113) and 3-ethylpentane of the present invention. The conditions assumed for this comparison are:

Evaporator temperature 40.0°F (4.4°C)
 Condenser temperature 110.0°F (43.3°C)
 Liquid subcool temperature 10.0°F (5.5°C)
 25 Return gas temperature 75.0°F (23.8°C)
 Compressor efficiency is 70%

These are typical conditions under which small turbine centrifugal compressors perform.

TABLE 3

Composition	Hi Btu/lb	Hi*0.7 Btu/lb	Hi*0.7 KJ/Kg	V2 m/s	V2 rel to CFC-113
CFC-113	10.92	7.6	17.8	133.3	na
1,1-dichloroethane	12.33	8.6	20.1	141.7	106%
trans-1,2-dichloroethylene	10.67	7.5	17.4	131.8	99%

methyl acetate	17.65	12.4	28.7	169.5	127%
methyl formate	17.68	12.4	28.8	169.7	127%
N(CH ₃) ₂ (CHF ₂)	17.87	12.5	29.1	170.6	128%
CF ₃ CF ₂ SCF ₂ CF ₃	10.85	7.6	17.7	132.9	100%

EXAMPLE 2

Performance Data

- The following table shows the performance of various refrigerants compared to CFC-113, by showing evaporator and condenser pressures, compressor discharge temperature, COP (energy efficiency) and capacity (Cap). The data are based on the following conditions.

Evaporator temperature	40.0°F (4.4°C)
Condenser temperature	110.0°F (43.3°C)
Subcool temperature	10.0°F (5.5°C)
Return gas temperature	75.0°F (23.8°C)
Compressor efficiency is	70%

TABLE 4

Compound	Evap Pres (Psia)	Evap Pres (kPa)	Cond Pres (Psia)	Cond Pres (kPa)	Comp Disch Temp (F)	Comp Disch Temp (C)	COP	Cap (Btu/min)	Cap (kW)
CFC-113	2.7	19	12.8	88	156.3	69.1	4.18	14.8	0.26
1,1-dichloroethane	1.7	12	8.8	61	214.0	101.1	4.37	12.4	0.22
trans-1,2-dichloroethylene	2.4	17	12.2	84	239.8	115.4	4.35	17.4	0.30
acetone	1.6	11	9.0	62	211.5	99.7	4.37	12.2	0.21
tert-butyl methyl ether	2.0	14	9.7	67	147.0	63.9	4.17	12.7	0.22
diisopropylether	1.1	7	6.0	42	143.8	62.1	4.15	7.4	0.13
dimethoxyethane	0.5	3	3.3	22	171.1	77.3	4.31	3.8	0.07
dimethoxymethane	2.5	17	13.3	91	176.3	80.2	4.29	17.6	0.31
methanol	0.6	4	5.5	38	347.8	175.4	4.20	6.1	0.11
ethanol	0.3	2	3.1	22	265.6	129.8	4.32	3.3	0.06
n-propanol	0.1	1	1.4	10	229.1	109.5	4.36	1.3	0.02
isopropanol	0.3	2	2.8	19	219.8	104.3	4.36	2.9	0.05
ethyl acetate	0.7	4	4.2	29	187.6	86.4	4.35	5.3	0.09
ethyl formate	1.7	11	9.8	68	199.0	92.8	4.34	12.7	0.22
methyl acetate	1.5	11	8.6	59	209.3	98.5	4.36	11.6	0.20
N(CH ₃) ₂ (CHF ₂)	2.4	17	11.9	82	170.9	77.2	4.28	16.1	0.28
CF ₃ CF ₂ SCF ₂ CF ₃	4.0	27	18.6	128	129.5	54.2	3.83	22.1	0.39

Data show that the compositions of the present invention have evaporator and condenser pressures similar to CFC-113. Additionally, some compositions have higher capacity (Cap) or energy efficiency (COP) than CFC-113.

CLAIM(S)**What is claimed is :**

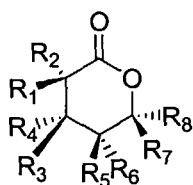
- 5 1. A refrigerant or heat transfer fluid composition comprising at least one compound selected from the group consisting of:
- nonafluoro-tert-butanol;
1,1-dichloroethane;
trans-1,2-dichloroethylene;
10 acetone;
tert-butyl methyl ether;
diisopropyl ether;
1,2-dimethoxyethane;
dimethoxymethane;
15 1,1-dimethoxyethane;
methanol;
ethanol;
n-propanol;
isopropanol;
20 ethyl acetate;
ethyl formate;
methyl acetate;
methyl formate;
N-(difluoromethyl)-N,N-dimethylamine; and
25 1,1,1,2,2-pentafluoro-2-[(pentafluoroethyl)thio]ethane.
2. A refrigerant or heat transfer fluid composition suitable for use in refrigeration or air-conditioning apparatus employing (i) centrifugal compressor or (ii) multi-stage centrifugal compressor or (iii) single
30 slab/single pass heat exchanger, said composition comprising at least one compound selected from the group consisting of:
- nonafluoro-tert-butanol;
1,1-dichloroethane;
trans-1,2-dichloroethylene;
35 acetone;
tert-butyl methyl ether;
diisopropyl ether;

- 1,2-dimethoxyethane;
dimethoxymethane;
1,1-dimethoxyethane;
methanol;
5 ethanol;
n-propanol;
isopropanol;
ethyl acetate;
ethyl formate;
10 methyl acetate;
methyl formate;
N-(difluoromethyl)-N,N-dimethylamine; and
1,1,1,2,2-pentafluoro-2-[(pentafluoroethyl)thio]ethane.
- 15 3. A process for producing cooling, said process comprising evaporating the composition of claim 1 or 2 in the vicinity of a body to be cooled, and thereafter condensing said composition.
- 20 4. A process for producing heat, said process comprising condensing the composition of claim 1 or 2 in the vicinity of a body to be heated, and thereafter evaporating said composition.
- 25 5. A method of using the compositions of claim 1 or 2, for heat transfer, said method comprising transferring said composition from a heat source to a heat sink.
- 30 6. The composition of claim 1, further comprising at least one ultra-violet fluorescent dye selected from the group consisting of naphthalimides, perylenes, coumarins, anthracenes, phenanthracenes, xanthenes, thioxanthenes, naphthoxanthenes, fluoresceins, and derivatives thereof.
- 35 7. The composition of claim 6, further comprising at least one solubilizing agent selected from the group consisting of hydrocarbons, dimethylether, polyoxyalkylene glycol ethers, amides, ketones, nitriles, chlorocarbons, esters, lactones, aryl ethers, fluoroethers, and 1,1,1-trifluoroalkanes; and wherein the refrigerant and solubilizing agent may not be the same compound.

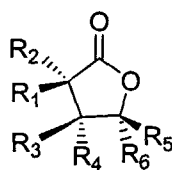
8. The composition of claim 7, wherein said solubilizing agent is selected from the group consisting of:

- 5 a) polyoxyalkylene glycol ethers represented by the formula $R^1[(OR^2)_xOR^3]_y$, wherein: x is an integer from 1 to 3; y is an integer from 1 to 4; R^1 is selected from hydrogen and aliphatic hydrocarbon radicals having 1 to 6 carbon atoms and y bonding sites; R^2 is selected from aliphatic hydrocarbylene radicals having from 2 to 4 carbon atoms; R^3 is selected from
10 hydrogen, and aliphatic and alicyclic hydrocarbon radicals having from 1 to 6 carbon atoms; at least one of R^1 and R^3 is selected from said hydrocarbon radicals; and wherein said polyoxyalkylene glycol ethers have a molecular weight of from about 100 to about 300 atomic mass units;
- 15 b) amides represented by the formulae $R^1CONR^2R^3$ and cyclo- $[R^4CON(R^5)-]$, wherein R^1 , R^2 , R^3 and R^5 are independently selected from aliphatic and alicyclic hydrocarbon radicals having from 1 to 12 carbon atoms, and at most one aromatic radical having from 6 to 12 carbon atoms; R^4 is selected from
20 aliphatic hydrocarbylene radicals having from 3 to 12 carbon atoms; and wherein said amides have a molecular weight of from about 100 to about 300 atomic mass units;
- 25 c) ketones represented by the formula R^1COR^2 , wherein R^1 and R^2 are independently selected from aliphatic, alicyclic and aryl hydrocarbon radicals having from 1 to 12 carbon atoms, and wherein said ketones have a molecular weight of from about 70 to about 300 atomic mass units;
- 30 d) nitriles represented by the formula R^1CN , wherein R^1 is selected from aliphatic, alicyclic or aryl hydrocarbon radicals having from 5 to 12 carbon atoms, and wherein said nitriles have a molecular weight of from about 90 to about 200 atomic mass units;
- 35 e) chlorocarbons represented by the formula RCI_x , wherein; x is selected from the integers 1 or 2; R is selected from aliphatic and alicyclic hydrocarbon radicals having from 1 to 12 carbon atoms; and wherein said chlorocarbons have a molecular weight of from about 100 to about 200 atomic mass units;

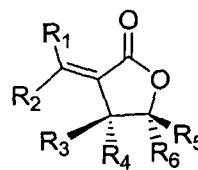
- 5 f) any ethers represented by the formula R^1OR^2 , wherein: R^1 is selected from aryl hydrocarbon radicals having from 6 to 12 carbon atoms; R^2 is selected from aliphatic hydrocarbon radicals having from 1 to 4 carbon atoms; and wherein said aryl ethers have a molecular weight of from about 100 to about 150 atomic mass units;
- g) 1,1,1-trifluoroalkanes represented by the formula CF_3R^1 , wherein R^1 is selected from aliphatic and alicyclic hydrocarbon radicals having from about 5 to about 15 carbon atoms;
- 10 h) fluoroethers represented by the formula $R^1OCF_2CF_2H$, wherein R^1 is selected from aliphatic and alicyclic hydrocarbon radicals having from about 5 to about 15 carbon atoms; or wherein said fluoroethers are derived from fluoro-olefins and polyols, wherein said fluoro-olefins are of the type $CF_2=CXY$, wherein X is hydrogen, chlorine or fluorine, and Y is chlorine, fluorine, CF_3 or OR_f , wherein R_f is CF_3 , C_2F_5 , or C_3F_7 ; and said polyols are linear or branched, wherein said linear polyols are of the type $HOCH_2(CHOH)_x(CRR')_yCH_2OH$, wherein R and R' are hydrogen, CH_3 or C_2H_5 , x is an integer from 0-4, y is an integer from 0-3 and z is either zero or 1, and said branched polyols are of the type $C(OH)_t(R)_u(CH_2OH)_v[(CH_2)_mCH_2OH]_w$, wherein R may be hydrogen, CH_3 or C_2H_5 , m is an integer from 0 to 3, t and u are 0 or 1, v and w are integers from 0 to 4, and also wherein $t + u + v + w = 4$;
- 25 i) lactones represented by structures [B], [C], and [D]:



[B]



[C]



[D]

30 wherein, R_1 through R_8 are independently selected from hydrogen, linear, branched, cyclic, bicyclic, saturated and unsaturated hydrocarbyl radicals; and the molecular weight is from about 100 to about 300 atomic mass units; and

5 ~~h) Esters represented by the general formula $R^1CO_2R^2$, wherein R^1 and R^2 are independently selected from linear and cyclic, saturated and unsaturated, alkyl and aryl radicals; and wherein said esters have a molecular weight of from about 80 to about 550 atomic mass units.~~

9. A method for introducing an ultraviolet fluorescent dye into a compression refrigeration or air-conditioning apparatus, said method comprising dissolving the ultraviolet fluorescent dye in the composition of
10 claim 1 or 2 in the presence of a solubilizing agent, and introducing the combination into said compression refrigeration or air-conditioning apparatus.
10. A method for solubilizing ultraviolet fluorescent dye in the
15 composition of claim 1 or said method comprising contacting the ultraviolet fluorescent dye with said composition, in the presence of a solubilizing agent.
11. A method for detecting leaks, said method comprising using the
20 composition of claim 6 in a compression refrigeration or air-conditioning apparatus, providing said apparatus, and providing a suitable means for detecting said composition in the vicinity of said apparatus.
12. A method of producing cooling, said method comprising:
25 evaporating the refrigerant or heat transfer fluid component of the composition of claim 6 in the vicinity of a body to be cooled and thereafter condensing said refrigerant or heat transfer fluid component.
13. A method of producing heat, said method comprising: condensing
30 the refrigerant or heat transfer fluid component of the composition of claim 6 in the vicinity of the body to be heated and thereafter evaporating said refrigerant or heat transfer fluid component.
14. The composition of claim 1 or 6 further comprising a stabilizer,
35 water scavenger, or odor masking agent.

15. ~~The composition of claim 14 wherein said stabilizer is selected from the group consisting of nitromethane, hindered phenols, hydroxylamines, thiols, phosphites and lactones.~~
- 5 16. A method of using the composition of claim 1 or 2, wherein said method comprises producing heat or refrigeration in a refrigeration or air-conditioning apparatus employing a multi-stage centrifugal compressor.
- 10 17. The method of claim 16 wherein said multi-stage centrifugal compressor is a two-stage centrifugal compressor.
18. The composition of claim 14 wherein said water scavenger is an ortho ester.
- 15 19. A process to produce cooling comprising compressing a composition of claim 1 or 2, in a mini-centrifugal compressor powered by an engine exhaust gas driven turbine; condensing said composition; and thereafter evaporating said composition in the vicinity of a body to be cooled.
- 20 20. A process to produce cooling comprising compressing a composition of claim 1 or 2 in a mini-centrifugal compressor powered by a ratioed gear drive assembly with a ratioed belt drive; condensing said composition; and thereafter evaporating said composition in the vicinity of
- 25 a body to be cooled.
21. A method for replacing CFC-113 in existing refrigeration apparatus or air-conditioning apparatus, said method comprising providing a composition of claim 1 as the replacement.

30