

Process for the manufacture of primary diamines

The present invention relates to a process for the manufacture of a primary diamine by hydrogenation of a dinitrile compound in the presence of a catalyst.

It relates more particularly to a process for the manufacture of hexamethylenediamine by hydrogenation of adiponitrile.

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Hexamethylenediamine is a chemical intermediate of great importance, used in particular as monomer in the manufacture of polyamides. Thus, hexamethylenediamine is used in combination with adipic acid to form an amine salt, hexamethylenediamine adipate, also known as Nylon salt. This salt is employed in the manufacture of poly(hexamethylene adipamide), more usually known as PA 6,6.

20 Hexamethylenediamine is also an important chemical intermediate in the manufacture of diisocyanate compounds.

The processes for the manufacture of hexamethylene-diamine used industrially consists in hydrogenating a dinitrile compound, namely adiponitrile, in the presence of a catalyst.

Thus, several hydrogenation processes are known and made use of. They can be classified into two main groups according to the catalyst used and the temperature and pressure conditions.

In the first group of processes, the catalyst is generally a metal oxide, such as iron oxide or cobalt oxide.

The hydrogenation is carried out at high pressures and temperatures and often in the presence of ammonia.

The second group comprises processes using metal catalysts of Raney type, such as Raney nickel or Raney cobalt. The hydrogenation can be carried out at a fairly low pressure and temperature. The Raney metal catalyst is often used in combination with doping metal oxides or metal elements. In order to maintain the activity and the selectivity of this type of catalyst, it is necessary and obligatory to use a strong base.

10 In these two groups of processes, the hydrogenated products and in particular hexamethylenediamine are recovered in the pure form by a sequence of distillations intended to remove water and the light impurities, on the one hand, and the heavy impurities, 15 on the other hand.

The term "light impurities" is understood to mean, in the field of the distillation of organic compounds, the compounds exhibiting a lower boiling point than that or those of the hydrogenated compounds which have to be recovered.

Likewise, the impurities referred to as "heavy impurities" are those which exhibit a higher boiling point than that of the hydrogenated compounds.

The formation of heavy or light impurities will be more or less favoured or more or less limited according to the conditions under which the hydrogenation process is carried out.

However, impurities are necessarily formed as they are generated by the decomposition of certain compounds or by reaction between the molecules present.

35 Mention may be made, among these impurities, of hexamethyleneimine (HMI), dicyclohexane (DCH),

tetrahydroazepine (THA) or aminomethylcyclopentenamine (AMCPA).

5 The majority of these impurities are a hindrance in the use of hexamethylenediamine, in particular as monomer in the manufacture of polyamides. This is because they can generate impurities in the polyamide obtained, causing a yellow colouring of the latter and inhomogeneities in the material which bring about
10 defects and breakages, in particular during the manufacture of yarns.

The presence of these impurities or compounds is detected and measured in particular by polarographic
15 analysis, expressed quantitatively by the polarographic index (PI) or moles of THA per million moles of HMD, and by gas chromatography analysis, for the determination of the HMI, DCH and AMCPA concentrations.

20 It is therefore necessary to employ a process for the recovery of the hydrogenated compounds, such as hexamethylenediamine, present in the hydrogenation reaction medium which makes it possible to obtain these compounds with a high degree of purity.

25 A process for the recovery of hexamethylenediamine comprising a first distillation of the hydrogenation reaction medium in order to remove the water has already been provided. The impurity hexamethyleneimine
30 (HMI) is also removed in this distillation column as it is entrained by the distilling water.

The bottom fraction collected in this first distillation is subjected to a second distillation in
35 which the light impurities are removed.

The pure hydrogenated product, such as hexamethylene-diamine, is obtained in a third distillation of the bottom fraction recovered in the second distillation.

- 5 The heavy impurities are removed in the bottom fraction from this third distillation stage.

There are a number of disadvantages to this sequence of distillations. In particular, it can be difficult to
10 obtain a hexamethylenediamine exhibiting a high purity. Furthermore, the energy consumption is high.

One of the aims of the present invention is to provide a process for the manufacture of diamines by
15 hydrogenation of dinitriles comprising a process for the recovery of the diamine which makes it possible to obtain, in a controlled way, a pure diamine with minimized energy consumption.

- 20 To this end, the invention provides a process for the manufacture of primary diamine compounds, more particularly of hexamethylenediamine (HMD), by hydrogenation of a dinitrile compound in the presence of a hydrogenation catalyst which consists in,
25 successively:

- Hydrogenating the dinitrile compound using hydrogen or a hydrogen-comprising gas and in recovering or withdrawing from this stage a stream E_0 comprising
30 the hydrogenated compounds.
- Subjecting the stream E_0 to a first distillation in order to recover a top fraction E_1 comprising water and the imines present (HMI) and a bottom fraction Q_1 comprising the hydrogenated compounds.
- 35 ➤ Subjecting the bottom stream Q_1 to a second distillation in order to recover, as bottom fraction Q_2 , the compounds forming impurities with a boiling point greater than that of the hydrogenated

compounds, known as "heavy" impurities, and, as top fraction E_2 , a fraction comprising the hydrogenated compounds. These heavy impurities are generally referred to as "tars" in the field of organic chemistry.

- 5 ➤ Subjecting the top stream E_2 to a third distillation in order to recover a top fraction E_3 comprising the light compounds or impurities and a bottom fraction Q_3 comprising the hydrogenated compounds. and
- 10 ➤ Subjecting the stream Q_3 to a fourth distillation in order to recover a top fraction E_4 composed of the pure diamines or hydrogenated compounds and a bottom fraction Q_4 comprising the heavy impurities.

15 This sequence of successive distillations makes it possible, from the second distillation, to remove the greater part of the compounds or impurities with a high boiling point or heavy impurities.

20 Thus, the removal at the beginning of the process of these "heavy" impurities makes it possible to improve and to facilitate the operation of the following distillations. By way of example, the fouling of the packing components of the columns is greatly reduced.

25 In addition, the energy necessary to purify the diamine in the following distillations is minimized.

It is also advantageous, in order to limit the losses of diamines (hexamethylenediamine), to treat the

30 fractions Q_2 and Q_4 comprising the "heavy" impurities. This treatment can be carried out in a conventional distillation column with distillation of the diamine or in columns of thin film evaporation type. The diamine recovered can advantageously be recycled in one of the

35 preceding columns, such as the final distillation column or the column which makes it possible to separate the heavy impurities and to recover the fraction Q_2 .

This process applies in particular when the hydrogenation stage is carried out in the presence of a catalyst based on a Raney metal, such as a Raney nickel or Raney cobalt, in combination with a strong inorganic
5 base, such as sodium hydroxide or potassium hydroxide.

This is because the presence of a strong base can result in the formation of products of high molecular weight, in particular when the medium is heated to
10 relatively high temperatures, such as those reached in the reboilers of the distillation columns.

The process of the invention applies in particular to the manufacture of hexamethylenediamine (HMD) by
15 hydrogenation of adiponitrile in the presence of a catalyst based on Raney nickel.

The hydrogenation reaction is carried out in conventional devices for the implementation of this
20 reaction and at standard temperature conditions.

Mention may be made, as example of the illustration of the implementation of this hydrogenation reaction, of the patents: FR 913 997, FR 1 463 409, BE 700 877,
25 US 3 821 305, US 3 056 837, WO 00/37424 and WO 00/03972.

The various distillation stages are carried out in conventional and standard distillation devices, such as
30 perforated plate columns, valve tray columns, packed columns, structured packed columns or plate columns.

The operating conditions for these columns will be shown in the detailed description of an embodiment of
35 the invention.

The invention makes it possible to recover a hexamethylenediamine exhibiting a high purity

corresponding to the specifications required in particular for the manufacture of a polyamide.

Other advantages and details of the invention will
5 become more clearly apparent in the light of the detailed description of an embodiment of the process of the invention, with reference to the single appended Figure 1, which represents a block diagram of this embodiment of the process of the invention.

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A reaction medium E_0 originating from a stage of hydrogenation of adiponitrile, not represented, in the presence of a catalyst based on Raney nickel and of potassium hydroxide is fed to a distillation column 1.
15 This column 1 operates under a pressure of 50 to 220 mmHg (6.6 kPa to 29.3 kPa) with a number of theoretical plates of between 5 and 20.

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This column is a structured packed column.

This column 1, known as the dehydration column, makes it possible to recover at the top E_1 the water present in the reaction medium and also hexamethyleneimine. The bottom fraction Q_1 comprises less than 50 ppm of water.

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This fraction Q_1 is fed to a second distillation column 2 operating under a pressure of 50 to 250 mmHg (6.6 kPa to 33.3 kPa) with a number of theoretical plates of between 1 and 5.

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The bottom fraction Q_2 is composed of compounds with a higher boiling point than that of hexamethylenediamine. It represents from 1 to 10% by weight of the heavy compounds present in the reaction medium E_0 .

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The top fraction E_2 is fed to a third distillation column 3 operating under a pressure of 10 to 80 mmHg (1.3 kPa to 10.6 kPa) with a number of theoretical

plates of between 30 and 80. This column 3 is advantageously a packed column.

5 The bottom fraction Q_3 recovered exhibits a concentration of diaminocyclohexane (DCH) of less than 10 ppm.

The top fraction E_3 is composed of light compounds with a lower boiling point than that of hexamethylenediamine.

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The bottom fraction Q_3 is fed to a fourth and final distillation column 4 operating under a pressure of 10 to 50 mmHg (1.3 kPa to 6.6 kPa) and with a number of theoretical plates of between 30 and 70. This column 4
15 is a plate column.

The hexamethylenediamine is recovered at the column top in the form of a stream E_4 comprising less than 2 ppm of tetrahydroazepine and with a PI of less than 15.

20 The bottom fraction Q_4 comprises heavy compounds.

Advantageously, the bottom fractions Q_2 and Q_4 can be subjected to a treatment in order to extract the hexamethylenediamine present, for example by
25 distillation in a distillation column or in a column with evaporation in the form of thin films (not represented).

30 The process of the invention, by removing a portion of the heavy compounds before the removal of the light compounds and the distillation of hexamethylenediamine, makes it possible to carry out a separation in the third and fourth columns with a minimum of energy and an excellent yield.

CLAIMS

1. Process for the manufacture of primary diamines by hydrogenation of a dinitrile compound in the presence
5 of a hydrogenation catalyst, characterized in that it consists in

- Hydrogenating the dinitrile compound using hydrogen or a hydrogen-comprising gas.
- 10 ➤ Subjecting the stream E_0 resulting from the hydrogenation medium to a first distillation in order to recover a top fraction E_1 comprising the water and the imines present and a bottom fraction Q_1 comprising the hydrogenated compounds.
- 15 ➤ Subjecting the stream Q_1 to a second distillation in order to recover a top fraction E_2 comprising the hydrogenated compounds and a bottom fraction Q_2 comprising compounds with a higher boiling point than the primary diamine.
- 20 ➤ Subjecting the top fraction E_2 to a third distillation in order to recover a bottom fraction Q_3 comprising the hydrogenated compounds and a top fraction E_3 comprising the compounds with a lower boiling point than that of the diamine compound
25 formed. and
- Subjecting the bottom fraction Q_3 to a fourth distillation in order to recover a top fraction E_4 comprising the pure diamine.

30 2. Process according to Claim 1, characterized in that the catalyst is a Raney metal chosen from the group consisting of Raney nickel and Raney cobalt.

3. Process according to Claim 2, characterized in
35 that the hydrogenation is carried out in the presence of a strong inorganic base chosen from the group consisting of sodium hydroxide and potassium hydroxide.

4. Process according to one of the preceding claims,
characterized in that the distillation stages are
carried out in distillation columns chosen from the
group consisting of perforated plate columns, plate
5 columns, packed columns, valve tray columns and
structured packed columns.

5. Process according to one of the preceding claims,
characterized in that the fractions comprising heavy
10 impurities, Q_2 and Q_4 , are treated in a distillation
column in order to recover the diamine and to recycle
this recovered diamine in the second or fourth
distillation.

15 6. Process according to one of the preceding claims,
characterized in that the diamine is hexamethylene-
diamine and the dinitrile compound is adiponitrile.