

“Ibopamine maleate, method for preparing it and pharmaceutical
compositions containing it”

* * * * *

5 The present invention relates to the salt ibopamine maleate (1:1), to
a method for preparing it and to a pharmaceutical composition for
ophthalmic use containing it.

US 4 218 470 describes ibopamine (epinine 3,4-O-diisobutyrate) as a
drug that is useful in the systemic treatment of cardiovascular
complaints.

10 EP-A-0 205 606 describes the use of ibopamine and
pharmaceutically acceptable acid-addition salts thereof as mydriatics.
The pharmaceutically acceptable acid-addition salt specifically
illustrated and tested in the said document is the hydrochloride.

15 EP-A- 0 442 958 describes an aqueous pharmaceutical solution for
ophthalmic use comprising a pharmaceutically acceptable acid-addition
salt of ibopamine, in which the said solution is buffered to pH 4.5 and
comprises from 0.1 to 0.5 parts by weight of hydroxypropylmethyl-
cellulose per one part by weight of the said ibopamine salt. In this case
also, the pharmaceutically acceptable acid-addition salt specifically
20 illustrated and tested is the hydrochloride.

It has now been found that the maleate shows better local tolerability
than the hydrochloride.

In a first aspect, the present invention thus relates to ibopamine
maleate (1:1).

25 The salt ibopamine maleate (1:1) is readily prepared via known
techniques, for instance the addition of maleic acid, dissolved in a
suitable organic solvent, to ibopamine base, also dissolved in a suitable
organic solvent, in a 1:1 molar ratio.

30 The said addition is preferably performed under an atmosphere of
an inert gas and at room temperature.

- 2 -

The salt thus formed (ibopamine maleate 1:1) is then isolated via known techniques including the precipitation and filtration of the salt or removal of the solvents by evaporation.

5 In one preferred embodiment, the abovementioned organic solvent is acetone and the salt is precipitated from the acetone solution via addition of ethyl ether.

In a second aspect, the present invention thus relates to a method for preparing ibopamine maleate (1:1), characterized in that it includes the addition of maleic acid, dissolved in a suitable organic solvent, to
10 ibopamine base, also dissolved in a suitable organic solvent, in a 1:1 molar ratio.

By virtue of its better local tolerability, ibopamine maleate is found to be particularly useful for ophthalmic use for diagnostic and therapeutic purposes.

15 In a third aspect, the present invention thus relates to a pharmaceutical composition for ophthalmic use, characterized in that it includes ibopamine maleate (1:1) together with at least one pharmaceutically acceptable vehicle.

A typical example of a pathological condition that may find benefit
20 from treatment with a pharmaceutical composition according to the present invention is ocular hypotonia.

For diagnostic purposes, the pharmaceutical composition according to the present invention is advantageously used as a mydriatic.

Preferably, the pharmaceutical composition according to the present
25 invention will be in the form of an ointment or eyedrops and may also comprise other vehicles that are suitable for ophthalmic use, for instance ethylene glycol, PEG, carboxymethylcellulose, mannitol, sorbitol, poloxamers, methylcellulose, hydroxyethylcellulose, hydroxypropylcellulose and the like.

- 3 -

This composition may also comprise other conventional ingredients, for instance: preserving agents, stabilizers, surfactants, buffers, salts for regulating osmotic pressure, emulsifiers, and the like.

5 If required by particular diagnostic or therapeutic uses, the pharmaceutical composition according to the present invention may comprise other pharmacologically active ingredients whose simultaneous administration is useful, for instance hyaluronic acid.

10 The amount of ibopamine maleate in the pharmaceutical composition of the present invention may vary within a wide range depending on known factors, for instance the particular diagnostic use or the particular type of disease to be treated, the seriousness of the disease and the number of daily administrations. However, a person skilled in the art may easily and routinely determine the optimum amount.

15 Typically, the amount of ibopamine in the pharmaceutical composition of the present invention is between 0.01% and 6% by weight and even more preferably between 0.1% and 5% by weight.

20 The dosage forms of the pharmaceutical composition of the present invention may be prepared according to techniques that are well known to pharmaceutical chemists, including mixing, dissolution, sterilization and the like.

The examples that follow are given to illustrate the present invention without, however, limiting it.

Example 1

Preparation of Ibopamine Maleate

25 Step a)

Saturated sodium carbonate solution was added to a solution of ibopamine hydrochloride (4 g) in water (10 ml) until no further precipitate was formed.

30 The precipitate was extracted with ethyl ether (50 ml). The organic phase was separated out, dried over sodium sulphate and rapidly

- 4 -

filtered through a Büchner funnel. Finally, the ether was removed by evaporation at room temperature and under reduced pressure.

The solid residue thus obtained consisted of ibopamine base (3 g).

Step b)

- 5 A solution of maleic acid (674 mg; 0.005 mol) in acetone (5 ml) was added, under an inert atmosphere and without heating, at room temperature, to a solution of ibopamine base (1.78 g; 0.005 mol) in acetone (10 ml).

The solution was left under stirring at room temperature (20 minutes).

- 10 Ethyl ether was then added dropwise to the formation of opalescence, and stirring was continued until precipitation was complete (30 minutes from the start of the opalescence).

The solid was collected by filtration and washed with ethyl ether. The desired product (1 g) was thus obtained.

- 15 m.p. = 107-108°C

Elemental Analysis

For $C_{21}H_{29}N_1O_8$	C	H	N
Calculated	59.56	6.90	3.31
Found	59.53	6.92	3.27

20

Test 1

Ocular Tolerability

Two aqueous solutions were used.

- 25 The first contained 2% by weight of ibopamine hydrochloride (corresponding to 1.79% by weight of ibopamine) buffered to pH 7.0.

The second contained 2.46% by weight of ibopamine maleate (corresponding to 1.79% by weight of ibopamine) buffered to pH 7.0.

- 30 12 male rabbits (New Zealand White) with an average weight of 2 kg and an average age of ten months were used, divided into two groups of six rabbits each. The first group was treated with 0.1 ml of the first

- 5 -

test solution three times a day for fifteen days. The second group was treated with 0.1 ml of the second test solution three times a day for fifteen days.

The tolerability was evaluated according to J. Draize et al.,
5 Pharmacol. Exp. Ther., 83, 377-390 (1944). The results are shown in Table 1 below.

Table 1

		Before the first application	After the last application
		Ibopamine hydrochloride	Ibopamine maleate
Conjunctiva	Reddening	1	1
	Swelling	1	0
	Lachrymation	2	1
Iris		1	0
Cornea	Opacity	1	0
	Area of the cornea affected by opacity	2	1
Total Score		8	3

- 6 -

CLAIMS

1. Ibopamine maleate salt (1:1).
2. Pharmaceutical composition for ophthalmic use, characterized in that it comprises ibopamine maleate (1:1) together with at least one
5 pharmaceutically acceptable vehicle.
3. Pharmaceutical composition according to Claim 2, characterized in that it is in the form of an ointment or eyedrops.
4. Pharmaceutical composition according to Claim 2 or 3, characterized in that the amount of ibopamine is between 0.01% and 6% by weight.
10
5. Pharmaceutical composition according to Claim 2 or 3, characterized in that the amount of ibopamine is between 0.1% and 5% by weight.
6. Process for preparing the ibopamine maleate salt (1:1),
15 characterized in that it includes the addition of maleic acid, dissolved in a suitable organic solvent, to ibopamine base, also dissolved in a suitable organic solvent, in a 1:1 molar ratio.
7. Process according to Claim 6, characterized in that the abovementioned addition is performed under an atmosphere of an
20 inert gas.
8. Process according to Claim 6 or 7, characterized in that the abovementioned addition is performed at room temperature.
9. Process according to any one of the preceding Claims 6 to 8, characterized in that the salt formed is isolated via precipitation and
25 filtration.
10. Process according to any one of the preceding Claims 6 to 9, characterized in that the abovementioned organic solvent is acetone.
11. Process according to Claim 10, characterized in that the salt is
30 precipitated from the acetone solution via addition of ethyl ether.