

THIAZOLOLACTAM COMPOUND AS ERK INHIBITOR AND USE THEREOF

[0001] This application claims the priority of:

CN201911244773.3, filed on December 06, 2019;

CN201911257990.6, filed on December 10, 2019;

CN202010107001.1, filed on February 20, 2020;

CN202011138526.8, filed on October 22, 2020; and

CN202011402966.X, filed on December 02, 2020.

FIELD OF THE INVENTION

[0002] The present disclosure relates to a class of thiazololactam compounds and use thereof in the manufacture of a medicament for treating diseases related to ERK. Specially, the present disclosure relates to a compound represented by formula (III) or a pharmaceutically acceptable salt thereof.

BACKGROUND OF THE INVENTION

[0003] Ras/Raf/MEK/ERK pathway is a classical mitogen activated protein kinase (MAPK) signaling cascade pathway, is involved in the signal transduction of various growth factors, cytokines, mitogens and hormone receptors after activation, and is one of the most important signal transduction pathways for controlling cell growth, differentiation and survival.

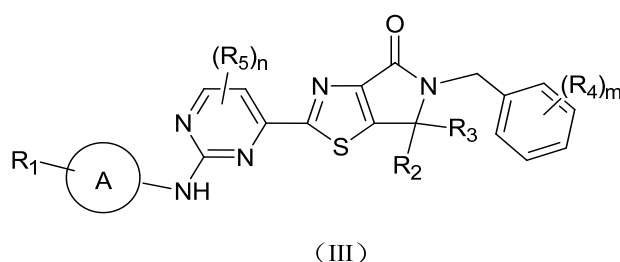
[0004] Studies have shown that abnormal activation of Ras/Raf/MEK/ERK pathway caused by mutation or amplification is a determinant of various cancers. In human tumors, the incidence of RAS mutation is about 22%, the incidence of BRAF mutation is about 7%, and the incidence of MEK mutation is about 1%. Therefore, key node proteins on this pathway have become important targets for the treatment of cancers (*Cancer Discov.* **2019**, 9, 329-341). Currently, a number of BRAF inhibitors and MEK1/2 inhibitors, as well as their combination regimens, have been approved by the US FDA for the treatment of melanoma, BRAFV600E mutant non-small cell lung cancer and other cancers. However, the use of BRAF and MEK inhibitors for these upstream nodes can rapidly lead to a problem of drug resistance due to mutation or pathway reactivation, greatly limiting their clinical application.

[0005] Extracellular regulated protein kinases (ERK) (especially ERK1 and ERK2 kinases) are major players and downstream key nodes in the Ras/Raf/MEK/ERK pathway, and their over-activation can be found in many human cancers. ERK, as the terminal signaling kinase of this pathway, has not yet been found to have mutations that lead to drug resistance. Therefore, a drug targeting ERK kinase is expected to overcome the problem of drug resistance caused by the treatment with upstream target inhibitors, and become a more potential therapeutic strategy. But so far, research on ERK inhibitors is still in the clinical phase, and no ERK inhibitors have been approved for marketing as drugs.

[0006] In summary, there is an urgent need to develop a safe and effective ERK inhibitor drug to meet the need of treatment of a tumor.

SUMMARY OF THE INVENTION

[0007] The present disclosure provides a compound represented by formula (III) or a pharmaceutically acceptable salt thereof,



[0008] wherein

[0009] R_1 is selected from H, C_{1-3} alkyl and C_{3-5} cycloalkyl, wherein the C_{1-3} alkyl and C_{3-5} cycloalkyl are optionally substituted by 1, 2 or 3 R_a ;

[0010] R_2 and R_3 are each independently C_{1-3} alkyl, wherein the C_{1-3} alkyl is optionally substituted by 1, 2 or 3 R_b ;

[0011] R_4 is selected from H, F, Cl, Br, I and C_{1-3} alkyl, wherein the C_{1-3} alkyl is optionally substituted by 1, 2 or 3 R_c ;

[0012] R_5 is selected from F, Cl, Br, I and C_{1-3} alkyl, wherein the C_{1-3} alkyl is optionally substituted by 1, 2 or 3 R_e ;

[0013] m is selected from 0, 1 and 2;

[0014] n is selected from 0, 1 and 2;

[0015] ring A is selected from pyrazolyl and tetrahydropyranyl, wherein the pyrazolyl and tetrahydropyranyl are optionally substituted by 1, 2 or 3 R_d;

[0016] R_a, R_b, R_c and R_e are each independently selected from D, F, Cl, Br, I, OH and OCH₃;

[0017] R_d is selected from F, Cl, Br, I, CH₃ and OCH₃.

[0018] In some embodiments of the present disclosure, the above-mentioned R₁ is selected from H, CH₃ and cyclopropyl, wherein the CH₃ and cyclopropyl are optionally substituted by 1, 2 or 3 R_a, and other variables are as defined in the present disclosure.

[0019] In some embodiments of the present disclosure, the above-mentioned R₁ is selected from H, CH₃, CHF₂, CD₃, CH₂CH₂OCH₃ and cyclopropyl, and other variables are as defined in the present disclosure.

[0020] In some embodiments of the present disclosure, the above-mentioned R₂ and R₃ are each independently selected from CH₃ and CH₂CH₃, wherein the CH₃ and CH₂CH₃ are optionally substituted by 1, 2 or 3 R_b, and other variables are as defined in the present disclosure.

[0021] In some embodiments of the present disclosure, the above-mentioned R₂ and R₃ are each independently CH₃, and other variables are as defined in the present disclosure.

[0022] In some embodiments of the present disclosure, the above-mentioned R₄ is selected from H, F, Cl, Br, I and CH₃, wherein the CH₃ is optionally substituted by 1, 2 or 3 R_c, and other variables are as defined in the present disclosure.

[0023] In some embodiments of the present disclosure, the above-mentioned R₄ is selected from H, F, Cl, Br, I and CH₃, and other variables are as defined in the present disclosure.

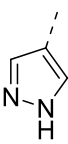
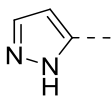
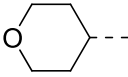
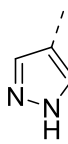
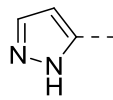
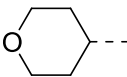
[0024] In some embodiments of the present disclosure, the above-mentioned R₅ is selected from F, Cl, Br, I and CH₃, wherein the CH₃ is optionally substituted by 1, 2 or 3 R_e, and other variables are as defined in the present disclosure.

[0025] In some embodiments of the present disclosure, the above-mentioned R₅ is selected from F, Cl, Br, I and CH₃, and other variables are as defined in the present disclosure.

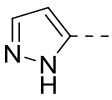
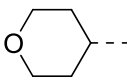
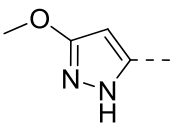
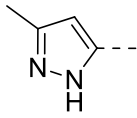
[0026] In some embodiments of the present disclosure, the above-mentioned R₅ is selected from H, F, Cl, Br, I and CH₃, wherein the CH₃ is optionally substituted by 1, 2 or 3 R_e, and other variables are as defined in the present disclosure.

[0027] In some embodiments of the present disclosure, the above-mentioned R₅ is selected from H, F, Cl, Br, I and CH₃, and other variables are as defined in the present disclosure.

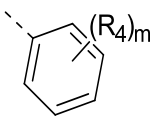
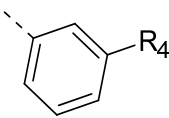
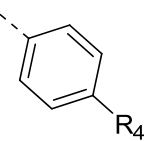
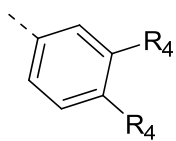
[0028] In some embodiments of the present disclosure, the above-mentioned ring A is

selected from , , and , wherein the , , and  are optionally substituted by 1, 2 or 3 R_d, and other variables are as defined in the present disclosure.

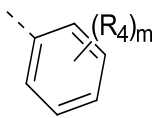
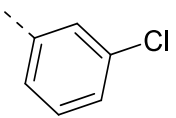
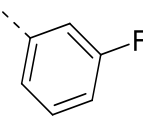
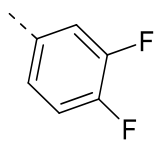
[0029] In some embodiments of the present disclosure, the above-mentioned ring A is

selected from , , , and , and other variables are as defined in the present disclosure.

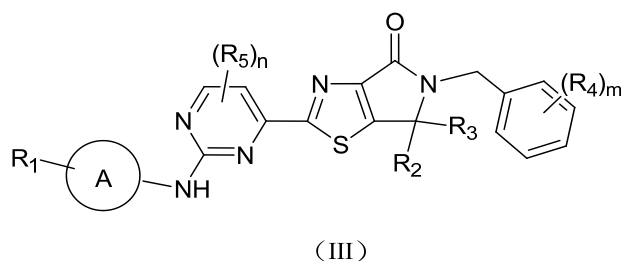
[0030] In some embodiments of the present disclosure, the above-mentioned structural

moiety  is selected from , , and , and other variables are as defined in the present disclosure.

[0031] In some embodiments of the present disclosure, the above-mentioned structural

moiety  is selected from , , and , and other variables are as defined in the present disclosure.

[0032] The present disclosure provides a compound represented by formula (III) or a pharmaceutically acceptable salt thereof,



[0033] wherein

[0034] R_1 is selected from H and C_{1-3} alkyl, wherein the C_{1-3} alkyl is optionally substituted by 1, 2 or 3 R_a ;

[0035] R_2 and R_3 are each independently C_{1-3} alkyl, wherein the C_{1-3} alkyl is optionally substituted by 1, 2 or 3 R_b ;

[0036] R_4 is selected from H, F, Cl, Br, I and C_{1-3} alkyl, wherein the C_{1-3} alkyl is optionally substituted by 1, 2 or 3 R_c ;

[0037] R_5 is selected from F, Cl, Br, I and C_{1-3} alkyl, wherein the C_{1-3} alkyl is optionally substituted by 1, 2 or 3 R_e ;

[0038] m is selected from 0, 1 and 2;

[0039] n is selected from 0, 1 and 2;

[0040] ring A is selected from pyrazolyl and tetrahydropyranyl, wherein the pyrazolyl and tetrahydropyranyl are optionally substituted by 1, 2 or 3 R_d ;

[0041] R_a , R_b , R_c and R_e are each independently selected from F, Cl, Br, I and OH;

[0042] R_d is selected from F, Cl, Br, I and CH_3 .

[0043] In some embodiments of the present disclosure, the above-mentioned R_1 is selected from H and CH_3 , wherein the CH_3 is optionally substituted by 1, 2 or 3 R_a , and other variables are as defined in the present disclosure.

[0044] In some embodiments of the present disclosure, the above-mentioned R_1 is selected from CH_3 , and other variables are as defined in the present disclosure.

[0045] In some embodiments of the present disclosure, the above-mentioned R_2 and R_3 are each independently selected from CH_3 and CH_2CH_3 , wherein the CH_3 and CH_2CH_3 are optionally substituted by 1, 2 or 3 R_b , and other variables are as defined in the present disclosure.

[0046] In some embodiments of the present disclosure, the above-mentioned R_2 and R_3 are each independently CH_3 , and other variables are as defined in the present disclosure.

[0047] In some embodiments of the present disclosure, the above-mentioned R_4 is selected from H, F, Cl, Br, I and CH_3 , wherein the CH_3 is optionally substituted by 1, 2 or 3 R_c , and other variables are as defined in the present disclosure.

[0048] In some embodiments of the present disclosure, the above-mentioned R_4 is selected

from H, F, Cl, Br, I and CH₃, and other variables are as defined in the present disclosure.

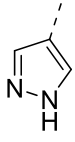
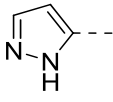
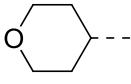
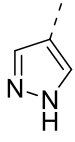
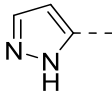
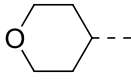
[0049] In some embodiments of the present disclosure, the above-mentioned R₅ is selected from F, Cl, Br, I and CH₃, wherein the CH₃ is optionally substituted by 1, 2 or 3 R_e, and other variables are as defined in the present disclosure.

[0050] In some embodiments of the present disclosure, the above-mentioned R₅ is selected from F, Cl, Br, I and CH₃, and other variables are as defined in the present disclosure.

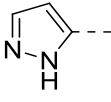
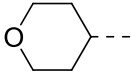
[0051] In some embodiments of the present disclosure, the above-mentioned R₅ is selected from H, F, Cl, Br, I and CH₃, wherein the CH₃ is optionally substituted by 1, 2 or 3 R_e, and other variables are as defined in the present disclosure.

[0052] In some embodiments of the present disclosure, the above-mentioned R₅ is selected from H, F, Cl, Br, I and CH₃, and other variables are as defined in the present disclosure.

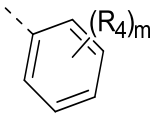
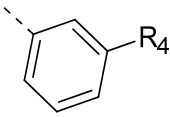
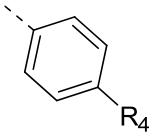
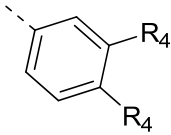
[0053] In some embodiments of the present disclosure, the above-mentioned ring A is

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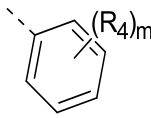
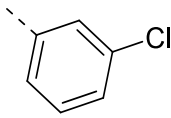
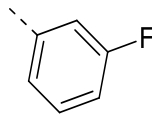
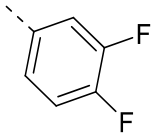
[0054] In some embodiments of the present disclosure, the above-mentioned ring A is

selected from  and , and other variables are as defined in the present disclosure.

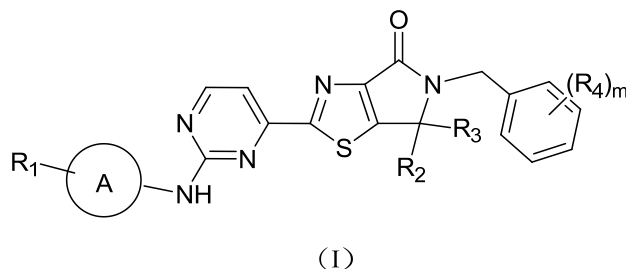
[0055] In some embodiments of the present disclosure, the above-mentioned structural

moiety  is selected from ,  and , and other variables are as defined in the present disclosure.

[0056] In some embodiments of the present disclosure, the above-mentioned structural

moiety  is selected from ,  and , and other variables are as defined in the present disclosure.

[0057] The present disclosure provides a compound represented by formula (I) or a pharmaceutically acceptable salt thereof,



[0058] wherein

[0059] R_1 is selected from H and C_{1-3} alkyl, wherein the C_{1-3} alkyl is optionally substituted by 1, 2 or 3 R_a ;

[0060] R_2 and R_3 are each independently C_{1-3} alkyl, wherein the C_{1-3} alkyl is optionally substituted by 1, 2 or 3 R_b ;

[0061] R_4 is selected from H, F, Cl, Br, I and C_{1-3} alkyl, wherein the C_{1-3} alkyl is optionally substituted by 1, 2 or 3 R_c ;

[0062] m is selected from 0, 1 and 2;

[0063] ring A is selected from pyrazolyl and tetrahydropyranyl, wherein the pyrazolyl and tetrahydropyranyl are optionally substituted by 1, 2 or 3 R_d ;

[0064] R_a , R_b and R_c are each independently selected from F, Cl, Br, I and OH;

[0065] R_d is selected from F, Cl, Br, I and CH_3 .

[0066] In some embodiments of the present disclosure, the above-mentioned R_1 is selected from H and CH_3 , wherein the CH_3 is optionally substituted by 1, 2 or 3 R_a , and other variables are as defined in the present disclosure.

[0067] In some embodiments of the present disclosure, the above-mentioned R_1 is CH_3 , and other variables are as defined in the present disclosure.

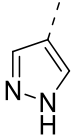
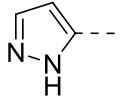
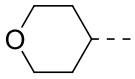
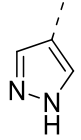
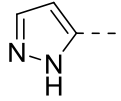
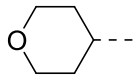
[0068] In some embodiments of the present disclosure, the above-mentioned R_2 and R_3 are each independently selected from CH_3 and CH_2CH_3 , wherein the CH_3 and CH_2CH_3 are optionally substituted by 1, 2 or 3 R_b , and other variables are as defined in the present disclosure.

[0069] In some embodiments of the present disclosure, the above-mentioned R_2 and R_3 are each independently CH_3 , and other variables are as defined in the present disclosure.

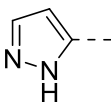
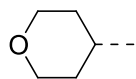
[0070] In some embodiments of the present disclosure, the above-mentioned R₄ is selected from H, F, Cl, Br, I and CH₃, wherein the CH₃ is optionally substituted by 1, 2 or 3 R_C, and other variables are as defined in the present disclosure.

[0071] In some embodiments of the present disclosure, the above-mentioned R₄ is selected from H, F, Cl, Br, I and CH₃, and other variables are as defined in the present disclosure.

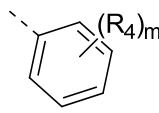
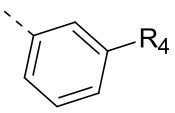
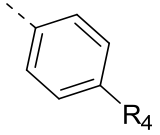
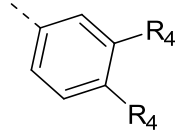
[0072] In some embodiments of the present disclosure, the above-mentioned ring A is

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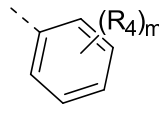
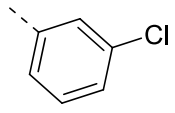
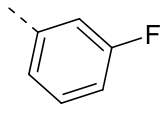
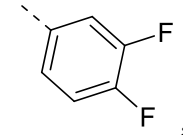
[0073] In some embodiments of the present disclosure, the above-mentioned ring A is

selected from  and , and other variables are as defined in the present disclosure.

[0074] In some embodiments of the present disclosure, the above-mentioned structural

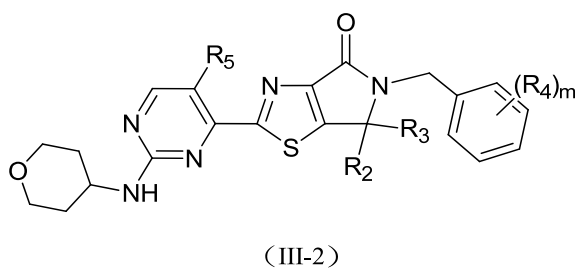
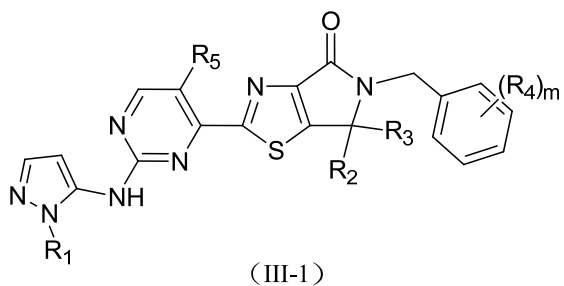
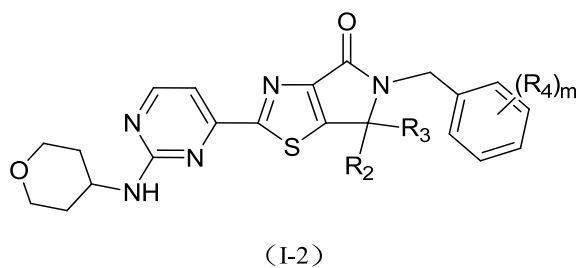
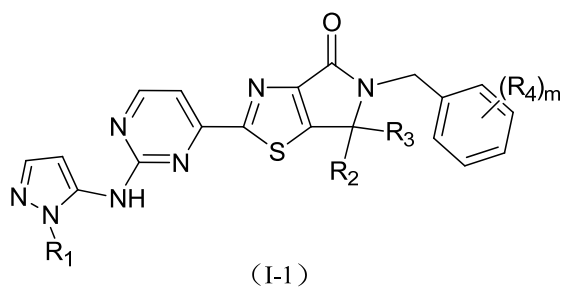
moiety  is selected from ,  and , and other variables are as defined in the present disclosure.

[0075] In some embodiments of the present disclosure, the above-mentioned structural

moiety  is selected from ,  and , and other variables are as defined in the present disclosure.

[0076] The present disclosure also includes some embodiments that are obtained by combining any of the above-mentioned variables.

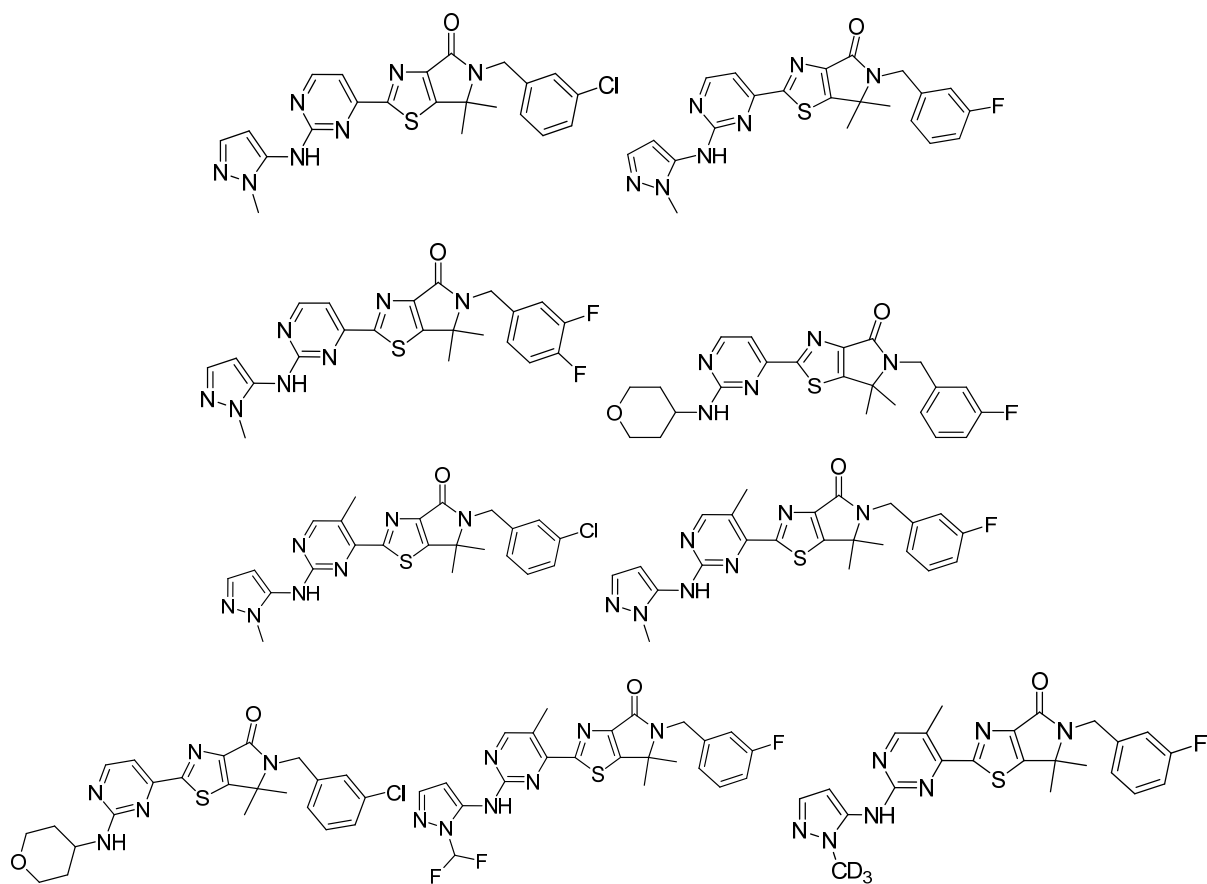
[0077] In some embodiments of the present disclosure, the above-mentioned compound or a pharmaceutically acceptable salt thereof is disclosed, wherein the compound is selected from:

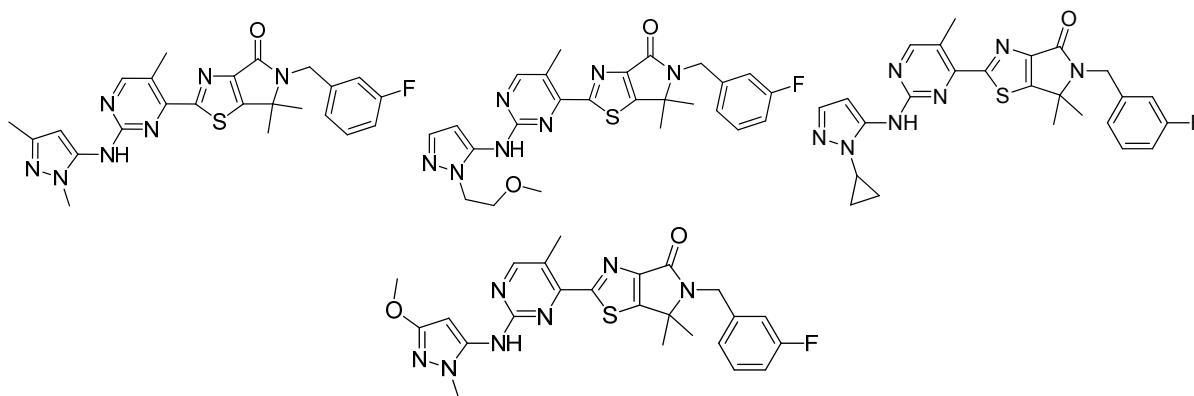


[0078] wherein

[0079] R_1 , R_2 , R_3 , R_4 , R_5 and m are as defined in the present disclosure.

[0080] The present disclosure also provides a compound represented by the following formula or a pharmaceutically acceptable salt thereof,





[0081] In some embodiments of the present disclosure, use of the above-mentioned compound or a pharmaceutically acceptable salt thereof in the manufacture of a medicament for treating diseases related to ERK is disclosed.

[0082] In some embodiments of the present disclosure, the above-mentioned use is characterized in that the medicament related to ERK inhibitor is a medicament for the treatment of solid tumor.

Technical effect

[0083] The compounds of the present disclosure exhibit excellent inhibitory activity to ERK2 kinase. Meanwhile, the compounds of the present disclosure exhibit excellent inhibitory activity to HT29 cell proliferation. The compounds of the present disclosure exhibit excellent oral exposure and bioavailability. The compounds of the present disclosure can significantly inhibit the growth of tumor. During the administration, the body weight of animals is not observed to decrease significantly, and the tolerance is good.

Definition and term

[0084] Unless otherwise specified, the following terms and phrases used herein are intended to have the following meanings. A specific term or phrase should not be considered indefinite or unclear in the absence of a particular definition, but should be understood in the conventional sense. When a trade name appears herein, it is intended to refer to its corresponding commodity or active ingredient thereof.

[0085] The term "pharmaceutically acceptable" is used herein in terms of those compounds, materials, compositions, and/or dosage forms, which are suitable for use in contact with

human and animal tissues within the scope of reliable medical judgment, with no excessive toxicity, irritation, allergic reaction or other problems or complications, commensurate with a reasonable benefit/risk ratio.

[0086] The term "pharmaceutically acceptable salt" means a salt of compounds disclosed herein that is prepared by reacting the compound having a specific substituent disclosed herein with a relatively non-toxic acid or base. When compounds disclosed herein contain a relatively acidic functional group, a base addition salt can be obtained by bringing the compound into contact with a sufficient amount of base in a pure solution or a suitable inert solvent. The pharmaceutically acceptable base addition salt includes a salt of sodium, potassium, calcium, ammonium, organic amine or magnesium or similar salts. When compounds disclosed herein contain a relatively basic functional group, an acid addition salt can be obtained by bringing the compound into contact with a sufficient amount of acid in a pure solution or a suitable inert solvent. Examples of the pharmaceutically acceptable acid addition salt include an inorganic acid salt, wherein the inorganic acid includes, for example, hydrochloric acid, hydrobromic acid, nitric acid, carbonic acid, bicarbonate, phosphoric acid, monohydrogen phosphate, dihydrogen phosphate, sulfuric acid, hydrogen sulfate, hydroiodic acid, phosphorous acid, and the like; and an organic acid salt, wherein the organic acid includes, for example, acetic acid, propionic acid, isobutyric acid, maleic acid, malonic acid, benzoic acid, succinic acid, suberic acid, fumaric acid, lactic acid, mandelic acid, phthalic acid, benzenesulfonic acid, p-toluenesulfonic acid, citric acid, tartaric acid, and methanesulfonic acid, and the like; and an salt of amino acid (such as arginine and the like), and a salt of an organic acid such as glucuronic acid and the like. Certain specific compounds disclosed herein contain both basic and acidic functional groups and can be converted to any base or acid addition salt.

[0087] The pharmaceutically acceptable salt disclosed herein can be prepared from the parent compound that contains an acidic or basic moiety by conventional chemical methods. Generally, such salt can be prepared by reacting the free acid or base form of the compound with a stoichiometric amount of an appropriate base or acid in water or an organic solvent or a mixture thereof.

[0088] Unless otherwise specified, the term "isomer" is intended to include geometric

isomers, cis- or trans- isomers, stereoisomers, enantiomers, optical isomers, diastereomers, and tautomers.

[0089] Compounds disclosed herein may be present in a specific geometric or stereoisomeric form. The present disclosure contemplates all such compounds, including cis and trans isomers, (-)- and (+)-enantiomers, (*R*)- and (*S*)-enantiomers, diastereoisomer, (*D*)-isomer, (*L*)-isomer, and a racemic mixture and other mixtures, for example, a mixture enriched in enantiomer or diastereoisomer, all of which are encompassed within the scope disclosed herein. The substituent such as alkyl may have an additional asymmetric carbon atom. All these isomers and mixtures thereof are encompassed within the scope disclosed herein.

[0090] Unless otherwise specified, the term "enantiomer" or "optical isomer" means stereoisomers that are in a mirrored relationship with each other.

[0091] Unless otherwise specified, the term "cis-trans isomer" or "geometric isomer" is produced by the inability of a double bond or a single bond between ring-forming carbon atoms to rotate freely.

[0092] Unless otherwise specified, the term "diastereomer" means a stereoisomer in which two or more chiral centers of are contained in a molecule and is in a non-mirrored relationship between molecules.

[0093] Unless otherwise specified, "(+)" means dextroisomer, "(-)" means levoisomer, and "(±)" means racemate.

[0094] Unless otherwise specified, a wedged solid bond () and a wedged dashed bond () indicate the absolute configuration of a stereocenter; a straight solid bond () and a straight dashed bond () indicate the relative configuration of a stereocenter; a wavy line () indicates a wedged solid bond () or a wedged dashed bond () or a wavy line () indicates a straight solid bond () and a straight dashed bond ()

[0095] Unless otherwise specified, the term "enriched in one isomer", "isomer enriched", "enriched in one enantiomer" or "enantiomeric enriched" means that the content of one isomer or enantiomer is less than 100%, and the content of the isomer or enantiomer is 60% or more, or 70% or more, or 80% or more, or 90% or more, or 95% or more, or 96% or more, or 97% or more, or 98% or more, or 99% or more, or 99.5% or more, or 99.6% or more, or

99.7% or more, or 99.8% or more, or 99.9% or more.

[0096] Unless otherwise specified, the term "isomer excess" or "enantiomeric excess" means the difference between the relative percentages of two isomers or two enantiomers. For example, if one isomer or enantiomer is present in an amount of 90% and the other isomer or enantiomer is present in an amount of 10%, the isomer or enantiomeric excess (ee value) is 80%.

[0097] Optically active (*R*)- and (*S*)-isomer, or *D* and *L* isomer can be prepared using chiral synthesis or chiral reagents or other conventional techniques. If one kind of enantiomer of certain compound disclosed herein is to be obtained, the pure desired enantiomer can be obtained by asymmetric synthesis or derivative action of chiral auxiliary followed by separating the resulting diastereomeric mixture and cleaving the auxiliary group. Alternatively, when the molecule contains a basic functional group (such as amino) or an acidic functional group (such as carboxyl), the compound reacts with an appropriate optically active acid or base to form a salt of the diastereomeric isomer which is then subjected to diastereomeric resolution through the conventional method in the art to afford the pure enantiomer. In addition, the enantiomer and the diastereoisomer are generally isolated through chromatography which uses a chiral stationary phase and optionally combines with a chemical derivative method (for example, carbamate generated from amine).

[0098] Compounds disclosed herein may contain an unnatural proportion of atomic isotopes at one or more of the atoms that make up the compounds. For example, a compound may be labeled with a radioisotope such as tritium (^3H), iodine-125 (^{125}I) or C-14(^{14}C). For another example, hydrogen can be replaced by heavy hydrogen to form a deuterated drug. The bond between deuterium and carbon is stronger than that between ordinary hydrogen and carbon. Compared with undeuterated drugs, deuterated drugs have advantages of reduced toxic side effects, increased drug stability, enhanced efficacy, and prolonged biological half-life of drugs. All changes in the isotopic composition of compounds disclosed herein, regardless of radioactivity, are included within the scope of the present disclosure.

[0099] The term "optional" or "optionally" means that the subsequent event or condition may occur but not requisite, that the term includes the instance in which the event or condition occurs and the instance in which the event or condition does not occur.

[00100] The term "substituted" means one or more than one hydrogen atom(s) on a specific atom are substituted by a substituent, including deuterium and hydrogen variants, as long as the valence of the specific atom is normal and the substituted compound is stable. When the substituent is oxo (i.e., =O), it means two hydrogen atoms are substituted. Positions on an aromatic ring cannot be substituted by oxo. The term "optionally substituted" means an atom can be substituted by a substituent or not, unless otherwise specified, the species and number of the substituent may be arbitrary so long as being chemically achievable.

[00101] When any variable (such as R) occurs in the constitution or structure of the compound more than once, the definition of the variable at each occurrence is independent. Thus, for example, if a group is substituted by 0-2 R, the group can be optionally substituted by up to two R, wherein the definition of R at each occurrence is independent. Moreover, a combination of the substituent and/or the variant thereof is allowed only when the combination results in a stable compound.

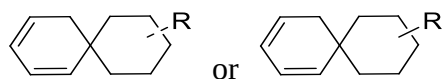
[00102] When the number of a linking group is 0, such as $-(CRR)_0-$, it means that the linking group is a single bond.

[00103] When the number of a substituent is 0, it means that the substituent does not exist. For example, $-A-(R)_0$ means that the structure is actually -A.

[00104] When a substituent is vacant, it means that the substituent does not exist. For example, when X is vacant in A-X, the structure of A-X is actually A.

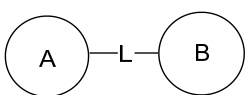
[00105] When one of variables is a single bond, it means that the two groups linked by the single bond are connected directly. For example, when L in A-L-Z represents a single bond, the structure of A-L-Z is actually A-Z.

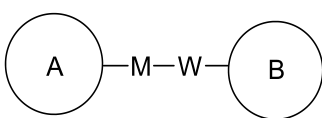
[00106] When the bond of a substituent can be cross-linked to two or more atoms on a ring, such substituent can be bonded to any atom on the ring. For example, a structural moiety

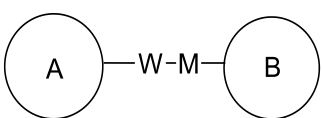


represents the substituent R thereof can be substituted at any site on cyclohexyl or cyclohexadiene. When an enumerated substituent does not indicate through which atom it is linked to the substituted group, such substituent can be bonded through any of its atoms. For example, a pyridyl group as a substituent may be linked to the substituted group through any one of carbon atoms on the pyridine ring.

[00107] When an enumerated linking group does not indicate its linking direction, its linking

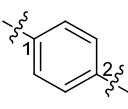
direction is arbitrary. For example, when the linking group L in  is -M-W-, the -M-W- can be linked to the ring A and the ring B in the same direction as the

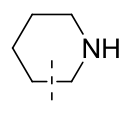
reading order from left to right to constitute , or can be linked to the ring A and the ring B in the reverse direction as the reading order from left to right to

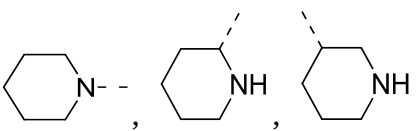
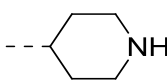
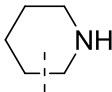
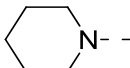
constitute . A combination of the linking groups, substituents and/or variants thereof is allowed only when such combination can result in a stable compound.

[00108] Unless otherwise specified, when a group has one or more connectable sites, any one or more sites of the group can be connected to other groups through chemical bonds. Where the connection position of the chemical bond is variable, and there is H atom(s) at a connectable site(s), when the connectable site(s) having H atom(s) is connected to the chemical bond, the number of H atom(s) at this site will correspondingly decrease as the number of the connected chemical bond increases, and the group will become a group of corresponding valence. The chemical bond between the site and other groups can be represented by a straight solid bond () , a straight dashed bond () , or a wavy line () .

For example, the straight solid bond in -OCH₃ indicates that the group is connected to other groups through the oxygen atom in the group; the straight dashed bond in  indicates that the group is connected to other groups through two ends of the nitrogen atom in the

group; the wavy line in  indicates that the group is connected to other groups

through the 1- and 2-carbon atoms in the phenyl group;  indicates that any connectable site on the piperidinyl group can be connected to other groups through one

chemical bond, including at least four connection ways, , and ; even if a H atom is drawn on -N-,  still includes the connection way of ; it's just that when one chemical bond is connected, the H at this site will be reduced by one, and the group will become the corresponding monovalent piperidinyl group.

[00109] Unless otherwise specified, the number of atoms on a ring is generally defined as the number of ring members, e.g., "5-7 membered ring" refers to a "ring" of 5-7 atoms arranged circumferentially.

[00110] Unless otherwise specified, the term " C_{1-3} alkyl" is used to indicate a linear or branched saturated hydrocarbon group consisting of 1 to 3 carbon atoms. The C_{1-3} alkyl group includes C_{1-2} and C_{2-3} alkyl groups and the like. It may be monovalent (e.g., methyl), divalent (e.g., methylene) or multivalent (e.g., methenyl). Examples of C_{1-3} alkyl groups include, but are not limited to, methyl (Me), ethyl (Et), propyl (including *n*-propyl and isopropyl), and the like.

[00111] Unless otherwise specified, " C_{3-5} cycloalkyl" means a saturated cyclic hydrocarbon group consisting of 3 to 5 carbon atoms, which is a monocyclic ring system; the C_{3-5} cycloalkyl includes C_{3-4} and C_{4-5} cycloalkyl groups, and the like; it may be monovalent, divalent or multivalent. Examples of C_{3-5} cycloalkyl include, but are not limited to, cyclopropyl, cyclobutyl, cyclopentyl, and the like.

[00112] The term "protecting group" includes, but is not limited to "amino protecting group", "hydroxy protecting group" or "thio protecting group". The term "amino protecting group" refers to a protecting group suitable for blocking the side reaction on the nitrogen of an amino. Representative amino protecting groups include, but are not limited to: formyl; acyl, such as alkanoyl (e.g. acetyl, trichloroacetyl or trifluoroacetyl); alkoxycarbonyl, such as tert-butoxycarbonyl (Boc); arylmethoxycarbonyl such as benzyloxycarbonyl (Cbz) and 9-fluorenylmethoxycarbonyl (Fmoc); arylmethyl such as benzyl (Bn), trityl (Tr), 1,1-bis-(4'-methoxyphenyl)methyl; silyl such as trimethylsilyl (TMS) and

tert-butyldimethylsilyl (TBS) and the like. The term "hydroxy protecting group" refers to a protecting group suitable for blocking the side reaction on hydroxy. Representative hydroxy protecting groups include, but are not limited to: alkyl such as methyl, ethyl and tert-butyl; acyl such as alkanoyl (e.g. acetyl); arylmethyl such as benzyl (Bn), p-methoxybenzyl (PMB), 9-fluorenylmethyl (Fm), and diphenylmethyl (benzhydryl, DPM); silyl such as trimethylsilyl (TMS) and tert-butyl dimethyl silyl (TBS) and the like.

[00113] Compounds disclosed herein can be prepared by a variety of synthetic methods well known to those skilled in the art, including the following enumerated embodiment, the embodiment formed by the following enumerated embodiment in combination with other chemical synthesis methods, and equivalent replacement well known to those skilled in the art. Alternative embodiments include, but are not limited to the embodiment disclosed herein.

[00114] The structures of compounds disclosed herein can be confirmed by conventional methods well known to those skilled in the art. If the present disclosure relates to an absolute configuration of a compound, the absolute configuration can be confirmed by conventional techniques in the art, such as single crystal X-Ray diffraction (SXRD). In the single crystal X-Ray diffraction (SXRD), the diffraction intensity data of the cultivated single crystal is collected using a Bruker D8 venture diffractometer with a light source of CuK α radiation in a scanning mode of φ/ω scan; after collecting the relevant data, the crystal structure is further analyzed by the direct method (Shelxs97) to confirm the absolute configuration.

[00115] Solvents used in the present disclosure are commercially available.

[00116] The following abbreviations are used in the present disclosure: aq represents aqueous; eq represents equivalent or equivalence; DCM represents dichloromethane; PE represents petroleum ether; DMSO represents dimethyl sulfoxide; EtOAc represents ethyl acetate; EtOH represents ethanol; MeOH represents methanol; Cbz represents benzyloxycarbonyl, which is an amine protecting group; BOC represents tert-butoxycarbonyl, which is an amine protecting group; r.t. represents room temperature; O/N represents overnight; THF represents tetrahydrofuran; Boc₂O represents di-tert-butyl dicarbonate; TFA represents trifluoroacetic acid; DIPEA represents diisopropylethylamine; iPrOH represents 2-propanol; mp represents melting point.

[00117] Compounds are named according to general naming principles in the art or by

ChemDraw® software, and commercially available compounds are named with their vendor directory names.

BRIEF DESCRIPTION OF THE DRAWINGS

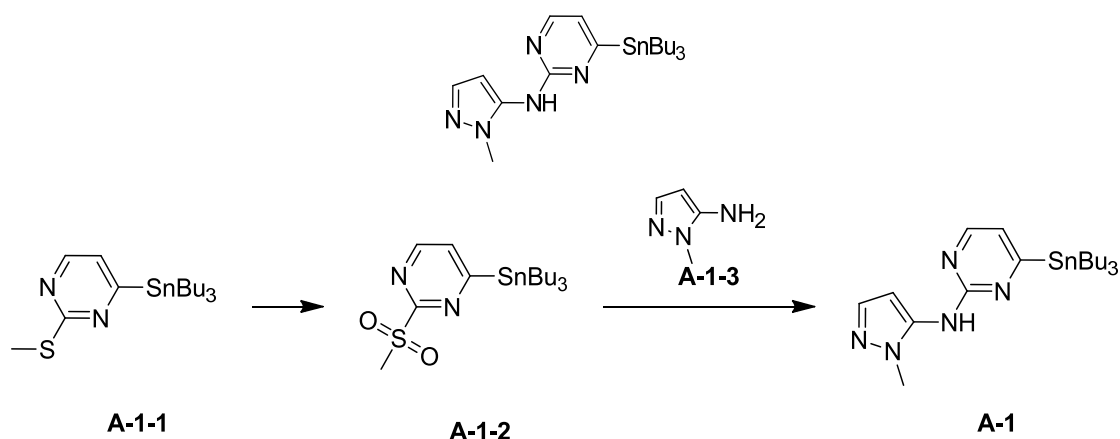
[00118] Figure 1: Tumor growth curve of human non-small cell lung cancer H358 in model animal after administration of solvent and **WX006** respectively;

[00119] Figure 2: Rate of weight change (%) in model animal of human non-small cell lung cancer H358 during the administration.

DETAILED DESCRIPTION OF THE INVENTION

[00120] The present disclosure is described in detail below by means of examples. However, it is not intended that these examples have any disadvantageous limitations to the present disclosure. The present disclosure has been described in detail herein, and embodiments are also disclosed herein. It will be apparent to those skilled in the art that various changes and modifications may be made to the embodiments disclosed herein without departing from the spirit and scope disclosed herein.

Reference example 1: Fragment A-1



[00121] Step 1: synthesis of compound **A-1-2**

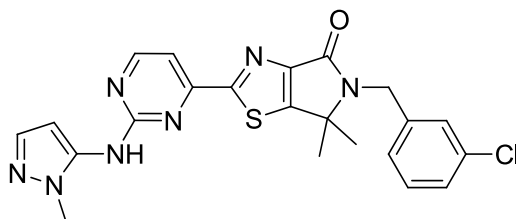
[00122] To a pre-dried single-necked flask was added a solution of sodium acetate (4.64 g, 56.60 mmol, 5 eq), potassium monopersulfate (13.92 g, 22.64 mmol, 2 eq) and water (47 mL). The mixture was cooled to 0 °C. A solution of **A-1-1** (4.7 g, 11.32 mmol, 1 eq), solvent tetrahydrofuran (47 mL) and methanol (47 mL) was added dropwise and the mixture was

stirred at 0 °C for 1 hour. Then the mixture was stirred in an oil bath at 29 °C for 15 hours. After completion of the reaction, the reaction solution was poured into water (200 mL), and the aqueous phase was extracted with ethyl acetate (50 mL*3). The organic phases were combined, and the combined organic phase was sequentially washed with saturated brine (200 mL), dried over anhydrous sodium sulfate, and filtered. The filtrate was collected and concentrated under reduced pressure to give a residue. The residue was purified by flash column chromatography to give **A-1-2**. ¹H NMR (400 MHz, CDCl₃) δ ppm 8.67 (d, *J* = 4.9 Hz, 1H), 7.64 (d, *J* = 4.9 Hz, 1H), 3.37 (s, 3H), 1.63 - 1.53 (m, 6H), 1.39 - 1.30 (m, 6H), 1.26 - 1.12 (m, 6H), 0.90 (t, *J* = 7.3 Hz, 9H).

[00123] Step 2: synthesis of compound **A-1**

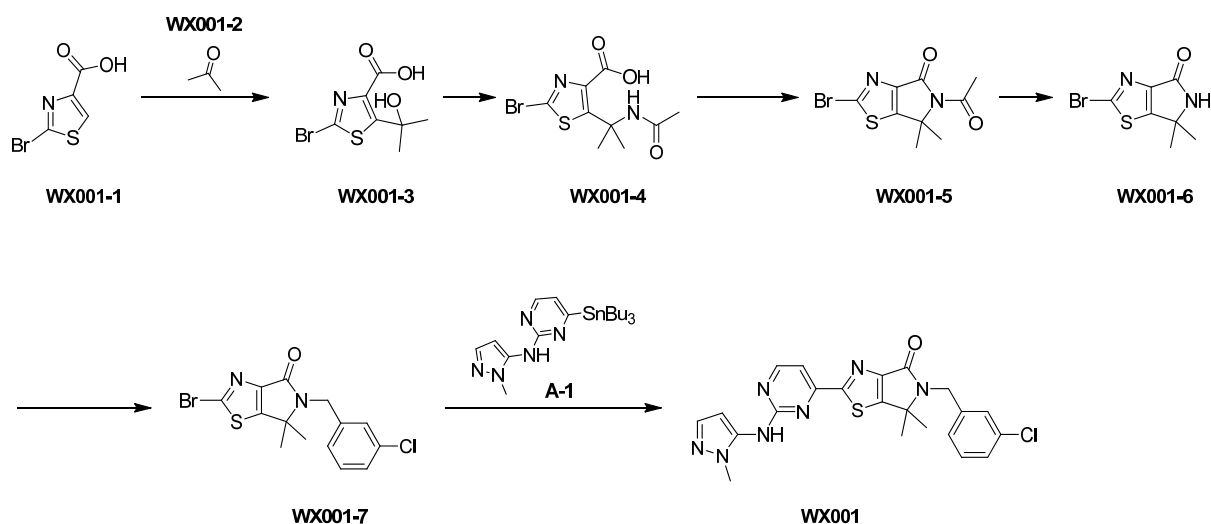
[00124] To a reaction flask were added **A-1-2** (3.9 g, 8.72 mmol, 1 eq), **A-1-3** (1.02 g, 10.46 mmol, 1.2 eq) and tetrahydrofuran (117 mL). The atmosphere was replaced with nitrogen gas, and then lithium hexamethyldisilazide (1 M, 18.31 mL, 2.1 eq) was added dropwise at -35 °C. The mixture solution was reacted at -35 °C for 10 minutes. After completion of the reaction, the reaction solution was quenched with saturated aqueous ammonium chloride solution (100 mL), and extracted with ethyl acetate (100 mL*2) and dichloromethane (100 mL). The organic phase was dried over anhydrous sodium sulfate, and filtered. The filtrate was rotary evaporated to dryness to give a crude product. The crude product was purified by column chromatography to give **A-1**. ¹H NMR (400 MHz, CDCl₃) δ ppm 8.17 (d, *J*=4.85 Hz, 1 H), 7.46 (d, *J*=1.76 Hz, 1 H), 6.91 (d, *J*=4.63 Hz, 1 H), 6.60 (s, 1 H), 6.32 (d, *J*=1.98 Hz, 1 H), 3.79 (s, 3 H), 1.52 - 1.61 (m, 6 H), 1.28 - 1.40 (m, 6 H), 1.03 - 1.20 (m, 6 H), 0.89 (t, *J*=7.28 Hz, 9 H).

Example 1



WX001

[00125] Route of synthesis:



[00126] Step 1: Synthesis of **WX001-3**

[00127] To a reaction flask was added a solution of lithium diisopropylamide (2 M, 2.88 mL, 2.4 eq) in tetrahydrofuran (5 mL) at -78 °C under nitrogen, and a solution of **WX001-1** (500 mg, 2.40 mmol, 1 eq) and tetramethylethylenediamine (418.94 mg, 3.61 mmol, 544.08 μ L, 1.5 eq) in tetrahydrofuran (1 mL) was then slowly added. The mixture was reacted at -78 °C for 0.5 hours, and **WX001-2** (279.18 mg, 4.81 mmol, 353.40 μ L, 2 eq) was then added. The mixture was reacted at -78 °C for another 2 hours. After completion of the reaction, the reaction solution was slowly poured into 30 mL of saturated aqueous ammonium chloride solution at 0 °C, adjusted to a pH of about 3-4 with hydrochloric acid (2 mol/L), and extracted with ethyl acetate (20 mL*3). The organic phases were combined, washed with saturated brine (20 mL*3), dried over anhydrous sodium sulfate, and filtered. The filtrate was concentrated under reduced pressure with a water pump at 45 °C to give **WX001-3**.

[00128] Step 2: synthesis of **WX001-4**

[00129] To a reaction flask were added **WX001-3** (250 mg, 939.45 μ mol, 1 eq) and acetonitrile (8 mL). The atmosphere was replaced with nitrogen gas, and then concentrated sulfuric acid (110.57 mg, 1.13 mmol, 60.09 μ L, 1.2 eq) was added at 0 °C. The mixture solution was reacted at 25 °C for 16 hours. After completion of the reaction, the reaction solution was diluted with 10 mL of water, and extracted with ethyl acetate (30 mL*3). The organic phases were combined, washed with saturated brine (30 mL*3), dried over anhydrous sodium sulfate, and filtered. The filtrate was concentrated under reduced pressure with a water pump at 45 °C to give **WX001-4**.

[00130] Step 3: synthesis of **WX001-5**

[00131] To a dry reaction flask were added **WX001-4** (200 mg, 651.12 μmol , 1 *eq*) and acetic anhydride (2 mL). The atmosphere was replaced with nitrogen gas, and then the mixture was reacted at 90 °C for 16 hours. After completion of the reaction, the reaction solution was concentrated under reduced pressure with an oil pump at 45 °C to give **WX001-5**.

[00132] Step 4: synthesis of **WX001-6**

[00133] To a dry reaction flask were added **WX001-5** (60 mg, 207.51 μmol , 1 *eq*), hydrochloric acid (2 M, 2 mL, 19.28 *eq*) and ethanol (2 mL). The mixture was reacted at 50 °C for 16 hours, and then reacted at 70°C for another 4 hours. After completion of the reaction, the reaction solution was concentrated under reduced pressure with a water pump at 45 °C, and extracted with ethyl acetate (30 mL*3). The organic phases were combined, washed with saturated brine (30 mL*3), dried over anhydrous sodium sulfate, and filtered. The filtrate was concentrated under reduced pressure with a water pump at 45 °C to give a crude product. The crude product was purified by thin layer preparative chromatoshheet to give **WX001-6**. ^1H NMR (400 MHz, $\text{DMSO-}d_6$) δ ppm 8.85 (br s, 1H), 1.48 (s, 6H).

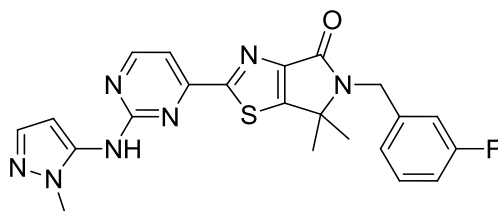
[00134] Step 5: synthesis of **WX001-7**

[00135] To a dry reaction flask were added **WX001-6** (60 mg, 242.80 μmol , 1 *eq*) and N,N-dimethylformamide (1 mL). The atmosphere was replaced with nitrogen gas, and then sodium hydride (14.57 mg, 364.21 μmol , 60% purity, 1.5 *eq*) was added at 0 °C. The mixture was reacted at 0 °C for 0.5 hours, and 3-chlorobenzyl bromide (49.89 mg, 242.80 μmol , 31.78 μL , 1 *eq*) was then added. The reaction solution was slowly warmed to 25 °C and reacted for another 2 hours. After completion of the reaction, 20 mL of water was added to the reaction solution, and the mixture was extracted with ethyl acetate (10 mL*3). The organic phases were combined, washed with saturated brine (10 mL*3), dried over anhydrous sodium sulfate, and filtered. The filtrate was concentrated under reduced pressure with a water pump at 45 °C to give a crude product. The crude product was purified by thin layer preparative chromatoshheet to give **WX001-7**. ^1H NMR (400 MHz, CDCl_3) δ ppm 7.26 (s, 1H), 7.17 (s, 3H), 4.62 (s, 2H), 1.37 (s, 6H).

[00136] Step 6: synthesis of **WX001**

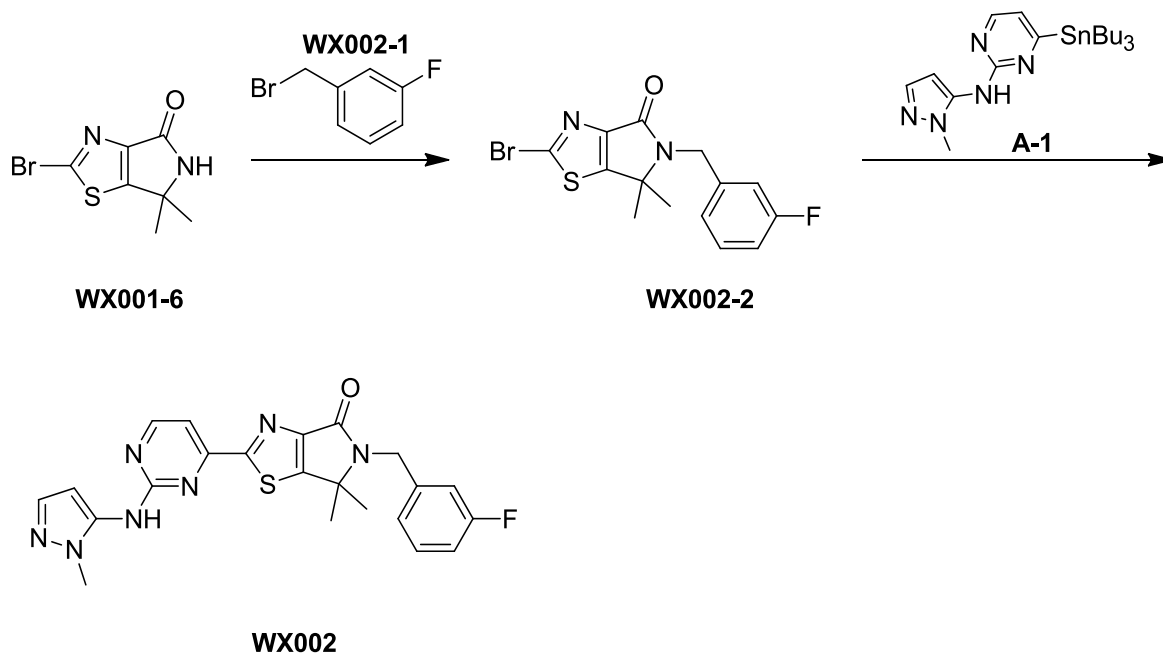
[00137] To a reaction flask were added **WX001-7** (35 mg, 94.17 μmol , 1 *eq*), **A-1** (56.28 mg, 103.58 μmol , 1.1 *eq*) and toluene (2 mL), and the atmosphere was replaced with nitrogen gas. The mixture was heated to 125 °C, and tetrakis(triphenylphosphine)palladium (21.76 mg, 18.83 μmol , 0.2 *eq*) was then slowly added. The mixture was reacted at 125 °C for 48 hours. After completion of the reaction, the reaction solution was concentrated under reduced pressure with a water pump at 45 °C to give a crude product. The crude product was purified by thin layer preparative chromatoshheet to give **WX001**.

Example 2



WX002

[00138] Route of synthesis:



[00139] Step 1: synthesis of **WX002-2**

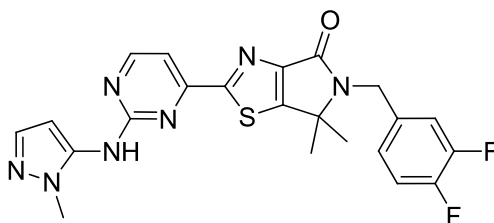
[00140] To a dry reaction flask were added **WX001-6** (50 mg, 202.34 μmol , 1 *eq*) and dimethylformamide (1 mL). The atmosphere was replaced with nitrogen gas, and then sodium hydride (12.14 mg, 303.51 μmol , 60% purity, 1.5 *eq*) was added at 0 °C. The mixture was reacted at 0 °C for 0.5 hours, and **WX002-1** (38.25 mg, 202.34 μmol , 24.84 μL , 1 *eq*)

was then added. The reaction solution was slowly warmed to 20 °C and reacted for 2 another hours. After completion of the reaction, 20 mL of water was added to the reaction solution, and the mixture was extracted with ethyl acetate (10 mL*3). The organic phases were combined, washed with saturated brine (10 mL*3), dried over anhydrous sodium sulfate, and filtered. The filtrate was concentrated under reduced pressure with a water pump at 45 °C to give a crude product. The crude product was purified by thin layer preparative chromatoshheet to give **WX002-2**. ¹H NMR (400 MHz, DMSO-*d*₆): δ (ppm) 7.32-7.40 (m, 1H), 7.15-7.23 (m, 2H), 7.03-7.11 (m, 1H), 4.67 (s, 2H), 1.48 (d, *J* = 3.8 Hz, 6H).

[00141] Step 2: synthesis of **WX002**

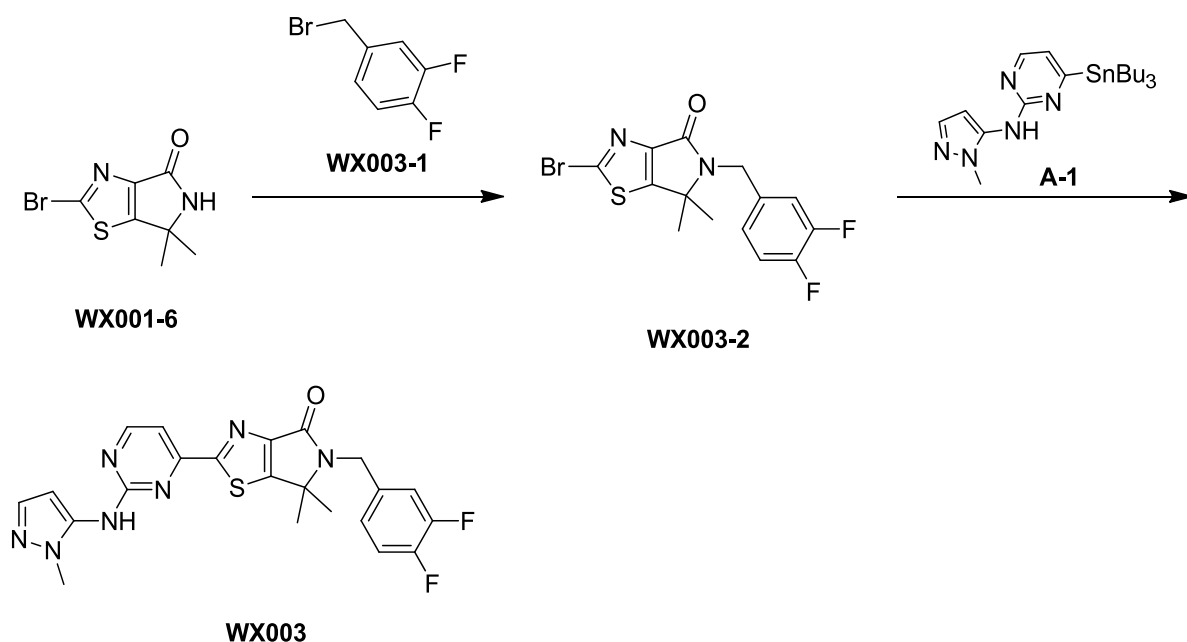
[00142] To a reaction flask were added **WX002-2** (60 mg, 168.91 μmol, 1 *eq*), **A-1** (100.94 mg, 185.80 μmol, 1.1 *eq*) and toluene (1 mL), and the atmosphere was replaced with nitrogen gas. The mixture was heated to 125 °C, and tetrakis(triphenylphosphine)palladium (39.04 mg, 33.78 μmol, 0.2 *eq*) was then slowly added. The mixture was reacted at 125 °C for 48 hours. After completion of the reaction, the reaction solution was concentrated under reduced pressure with a water pump at 45 °C to give a crude product. The crude product was purified by thin layer preparative chromatoshheet to give **WX002**.

Example 3



WX003

[00143] Route of synthesis:



[00144] Step 1: synthesis of **WX003-2**

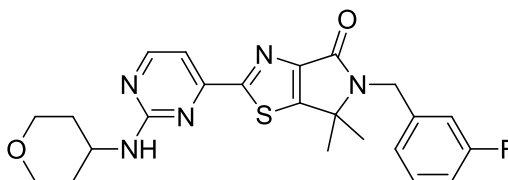
[00145] To a dry reaction flask were added **WX001-6** (80 mg, 325.04 μ mol, 1 *eq*) and dimethylformamide (4 mL). The atmosphere was replaced with nitrogen gas, and then sodium hydride (19.50 mg, 487.56 μ mol, 60% purity, 1.5 *eq*) was added at 0 °C. The mixture was reacted at 0°C for 0.5 hours, and **WX003-1** (67.29 mg, 325.04 μ mol, 41.54 μ L, 1 *eq*) was then added. The reaction solution was slowly warmed to 25 °C and reacted for another 2 hours. After completion of the reaction, 10 mL of water was added to the reaction solution, and the mixture was extracted with ethyl acetate (20 mL*3). The organic phases were combined, washed with saturated brine (20 mL*3), dried over anhydrous sodium sulfate, and filtered. The filtrate was concentrated under reduced pressure with a water pump at 45 °C to give a crude product. The crude product was purified by column chromatography to give **WX003-2**. ^1H NMR (400 MHz, DMSO- d_6): δ (ppm) 7.33-7.47 (m, 2H), 7.23 (m, 1H), 4.65 (s, 2H), 1.41-1.52 (m, 6H).

[00146] Step 2: synthesis of **WX003**

[00147] To a reaction flask were added **WX003-2** (70 mg, 187.56 μ mol, 1 *eq*), **A-1** (112.09 mg, 206.32 μ mol, 1.1 *eq*) and toluene (2 mL), and the atmosphere was replaced with nitrogen gas. The mixture was heated to 125 °C, and tetrakis(triphenylphosphine)palladium (43.35 mg, 37.51 μ mol, 0.2 *eq*) was then slowly added. The mixture was reacted at 125 °C for 48 hours. After completion of the reaction, the reaction solution was concentrated under reduced

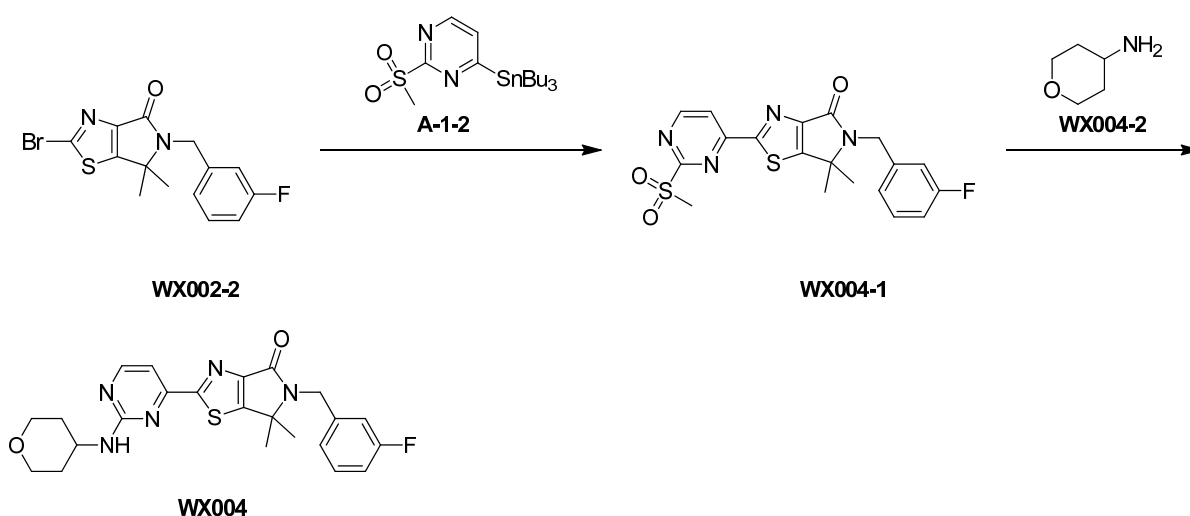
pressure with a water pump at 45 °C to give a crude product. The crude product was purified by thin layer preparative chromatoshheet to give **WX003**.

Example 4



WX004

[00148] Route of synthesis:



[00149] Step 1: synthesis of **WX004-1**

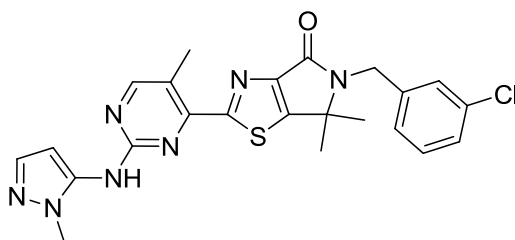
[00150] To a reaction flask were added **WX002-2** (70 mg, 197.06 μmol , 1 *eq*), **A-1-2** (113.45 mg, 216.76 μmol , 1.1 *eq*) and toluene (2 mL), and the atmosphere was replaced with nitrogen gas. The mixture was heated to 125 °C, and tetrakis(triphenylphosphine)palladium (45.54 mg, 39.41 μmol , 0.2 *eq*) was then slowly added. The mixture was reacted at 125 °C for 48 hours. After completion of the reaction, the reaction solution was concentrated under reduced pressure with a water pump at 45 °C to give a crude product. The crude product was purified by thin layer preparative chromatoshheet to give **WX004-1**. ^1H NMR (400 MHz, $\text{DMSO}-d_6$): δ (ppm) 9.26 (d, J = 5.1 Hz, 1H), 8.43 (d, J = 5.3 Hz, 1H), 7.33-7.42 (m, 1H), 7.16-7.28 (m, 2H), 7.08 (td, J = 8.6, 2.0 Hz, 1H), 4.73 (s, 2H), 3.50 (s, 3H), 1.55 (s, 6H).

[00151] Step 2: synthesis of **WX004**

[00152] To a pre-dried reaction flask were added **WX004-1** (50 mg, 115.61 μmol , 1 *eq*) and

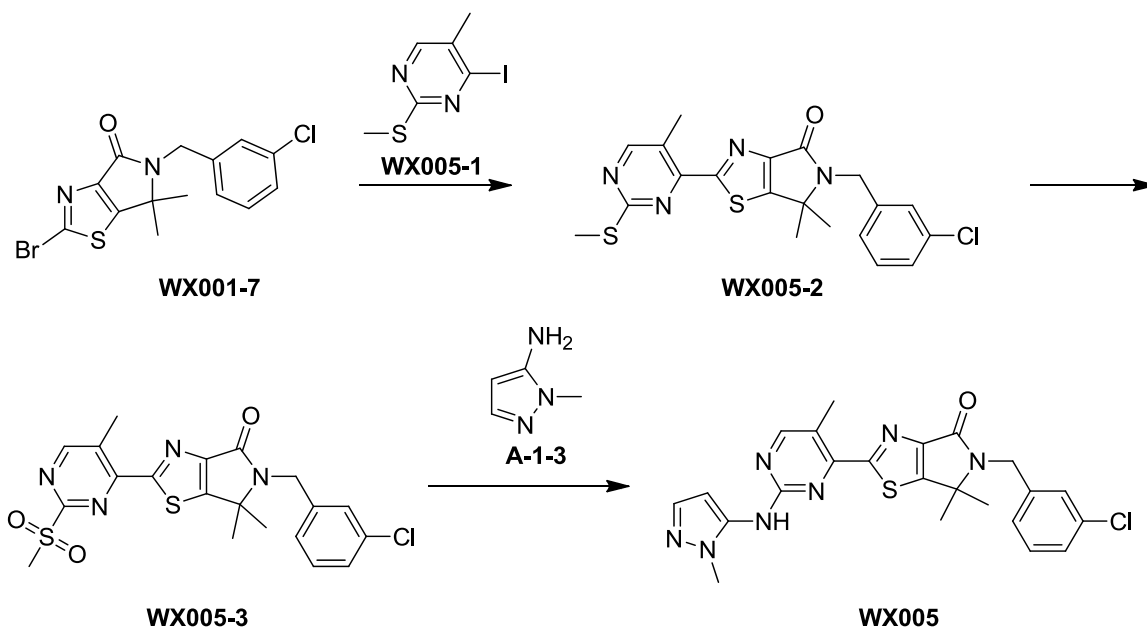
WX004-2 (11.69 mg, 115.61 μ mol, 1 *eq*), and the mixture was then dissolved with dimethyl sulfoxide (1 mL). The mixture was reacted with stirring at 100 °C for 14 hours. After completion of the reaction, the reaction solution was concentrated, and the residue was purified by thin layer preparative chromatoshheet to give **WX004**.

Example 5



WX005

[00153] Route of synthesis:



[00154] Step 1: synthesis of **WX005-2**

[00155] To a reaction flask were added **WX001-7** (530 mg, 1.43 mmol, 1 *eq*), zinc chloride (0.7 M, 1.83 mL, 0.9 *eq*) and tetrahydrofuran (3.5 mL). The atmosphere was replaced with nitrogen gas, and then *n*-butyl lithium (2.5 M, 855.58 μ L, 1.5 *eq*) was added at -25 °C. The mixture solution was reacted at 20 °C for 1 hour. A solution of **WX005-1** (379.45 mg, 1.43 mmol, 1 *eq*), tris(dibenzylideneacetone)dipalladium (65.29 mg, 71.30 μ mol, 0.05 *eq*) and tri(2-furyl)phosphine (33.11 mg, 142.60 μ mol, 0.1 *eq*) in tetrahydrofuran (0.5 mL) was added dropwise at -25 °C, and the mixture was reacted at 20 °C for 10 hours. Additional

tetrakis(triphenylphosphine)palladium (82.5 mg, 71.5 μmol , 0.05 *eq*) was added, and the mixture was reacted at 65° C for 20 hours. After completion of the reaction, the reaction solution was quenched with methanol and concentrated to give a crude product. The crude product was purified by preparative thin layer chromatography to give **WX005-2**. ^1H NMR (DMSO- d_6 , 400 MHz): δ (ppm) 8.75 (s, 1H), 7.46 (s, 1H), 7.26-7.41 (m, 3H), 4.72 (s, 2H), 2.65 (s, 3H), 2.59 (s, 3H), 1.53 (s, 6H).

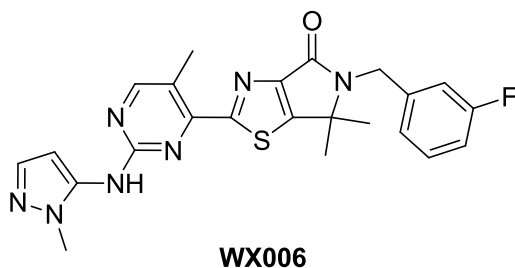
[00156] Step 2: synthesis of **WX005-3**

[00157] To a reaction flask were added **WX005-2** (140 mg, 324.85 μmol , 1 *eq*) and dichloromethane (6 mL). The atmosphere was replaced with nitrogen gas, and then m-chloroperoxybenzoic acid (210.22 mg, 974.54 μmol , 80% purity 3 *eq*) was added at 0 °C. The mixture solution was slowly warmed to 20 °C and reacted for 10 hours. After completion of the reaction, the reaction solution was quenched with saturated sodium sulfite (15 mL) and saturated aqueous sodium bicarbonate solution (15 mL) until the pH became alkaline and the color of KI test paper was not changed, and the mixture was extracted with dichloromethane (3 mL*3). The organic phase was concentrated to dryness under reduced pressure with a water pump at 45 °C to give a brown solid. Then the brown solid was slurried with ethyl acetate (3 mL) to give **WX005-3**. ^1H NMR (DMSO- d_6 , 400 MHz): δ (ppm) 9.18 (s, 1H), 7.47 (s, 1H), 7.30-7.39 (m, 3H), 4.73 (s, 2H), 3.48 (s, 3H), 2.83 (s, 3H), 1.55 (s, 6H).

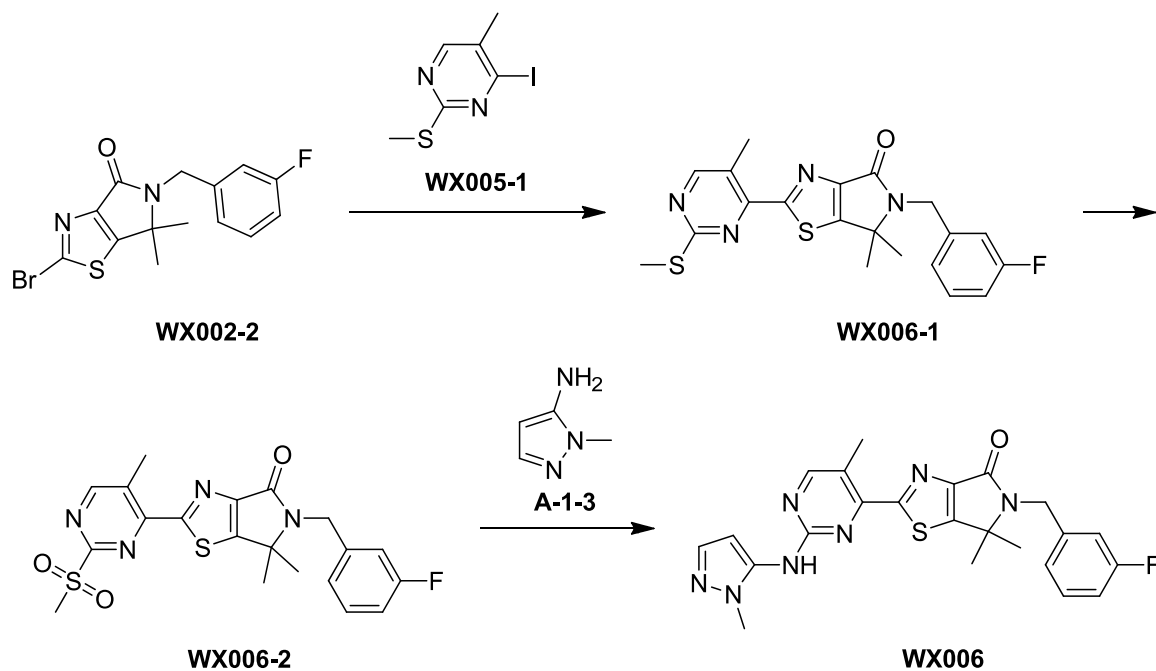
[00158] Step 3: synthesis of **WX005**

[00159] To a reaction flask were added **A-1-3** (15.73 mg, 162.00 μmol , 1.5 *eq*) and N,N-dimethylformamide (1 mL). The atmosphere was replaced with nitrogen gas, and then sodium hydride (6.48 mg, 162.00 μmol , 60% purity, 1.5 *eq*) was added at 0°C. After 0.5 hours, a solution of **WX005-3** (50 mg, 108.00 μmol , 1 *eq*) in N,N-dimethylformamide (1 mL) was added, and the mixture solution was reacted at 20 °C for 10 hours. After completion of the reaction, the reaction solution was quenched with saturated aqueous ammonium chloride solution (5 mL), and extracted with dichloromethane (1 mL*3). The organic phase was concentrated to dryness under reduced pressure with a water pump at 45 °C to give a crude product. The crude product was purified by thin layer chromatography to give **WX005**.

Example 6



[00160] Route of synthesis:



[00161] Step 1: synthesis of **WX006-1**

[00162] To a pre-dried reaction flask were added **WX002-2** (0.456 g, 1.28 mmol, 1 *eq*), zinc chloride (0.7 M, 1.65 mL, 0.9 *eq*) and tetrahydrofuran (10.5 mL). The atmosphere was replaced with nitrogen gas, and then the reaction system was cooled to -25 °C, and n-butyl lithium (2.5 M, 770.22 μ L, 1.5 *eq*) was added. The mixture solution was reacted with stirring at 20 °C for 1 hour. A solution of **WX005-1** (341.59 mg, 1.28 mmol, 1 *eq*) and tetrakis(triphenylphosphine)palladium (74.17 mg, 64.18 μ mol, 0.05 *eq*) in tetrahydrofuran (1.5 mL) was added at -25 °C. The mixture was reacted with stirring at 60 °C for 12 hours. After completion of the reaction, the reaction solution was quenched with 3 mL of methanol, and rotary evaporated to dryness to give a crude product. The crude product was purified by thin layer chromatography on silica gel plate to give **WX006-1**.

[00163] Step 2: synthesis of **WX006-2**

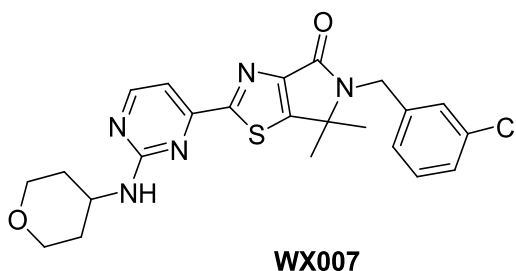
[00164] To a pre-dried reaction flask were added **WX006-1** (210 mg, 506.61 μ mol, 1 *eq*) and

dichloromethane (5 mL). The reaction system was cooled to 0 °C, and m-chloroperoxybenzoic acid (327.85 mg, 1.52 mmol, 80% purity, 3 eq) was added. The mixture was slowly warmed to room temperature (20 °C) and reacted with stirring for 12 hours. After completion of the reaction, the reaction solution was adjusted to a pH of about 8 with aqueous sodium bicarbonate solution (5 mL). Saturated sodium sulfite solution was added until the starch KI test paper did not show blue. The mixture solution was extracted with dichloromethane (10 mL*3). The organic phases were combined, washed with saturated brine (10 mL*2), dried over anhydrous sodium sulfate, and filtered. The filtrate was concentrated under reduced pressure with a water pump at 45 °C to give a crude product. The crude product was purified by thin layer chromatography on silica gel plate to give **WX006-2**.

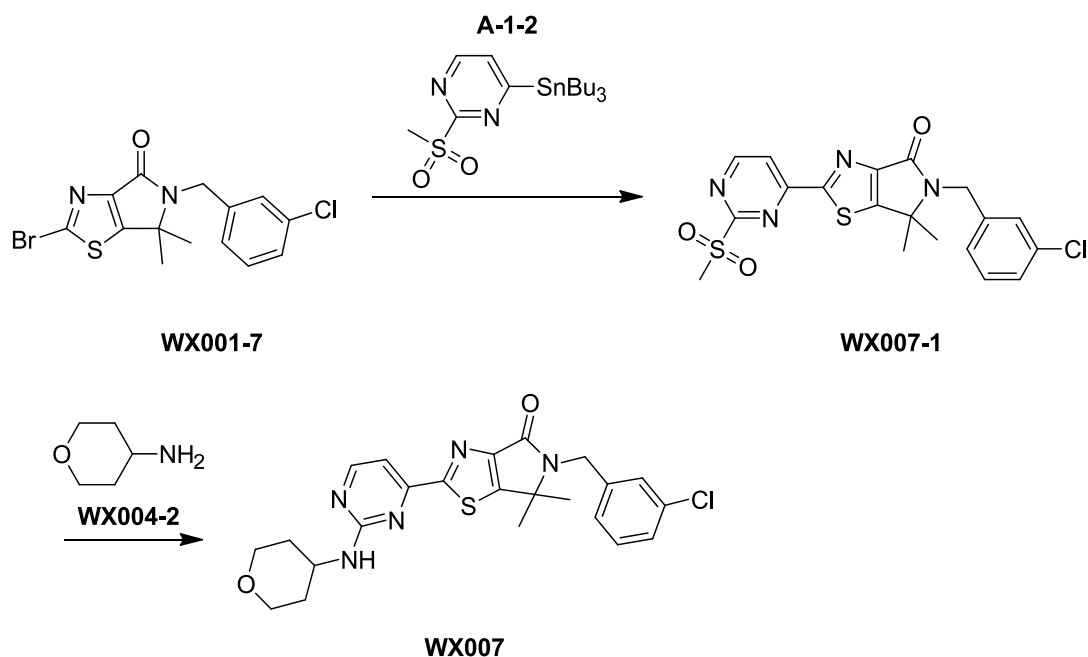
[00165] Step 3: synthesis of **WX006**

[00166] To a pre-dried reaction flask were added **A-1-3** (26.10 mg, 268.75 µmol, 1.2 eq) and N,N-dimethylformamide (3 mL). The reaction system was cooled to 0 °C. Sodium hydride (13.44 mg, 335.93 µmol, 60% purity, 1.5 eq) was then added. The mixture was reacted with stirring at 0 °C for 0.5 hours. **WX006-2** (100 mg, 223.96 µmol, 1 eq) was added, and the mixture was reacted with stirring at 0 °C for 0.5 hours. After completion of the reaction, the reaction solution was quenched with water (5 mL), and extracted with ethyl acetate (5 mL*3). The organic phases were combined, washed with saturated brine (5 mL*4), dried over anhydrous sodium sulfate, and filtered. The filtrate was concentrated under reduced pressure with a water pump at 45 °C to give a crude product. The crude product was purified by thin layer chromatography on silica gel plate to give **WX006**.

Example 7



[00167] Route of synthesis:



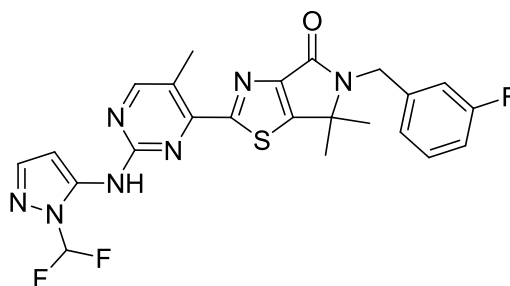
[00168] Step 1: synthesis of **WX007-1**

[00169] To a reaction flask were added **WX001-7** (50 mg, 134.52 μmol , 1 *eq*), **A-1-2** (77.45 mg, 147.98 μmol , 1.1 *eq*) and toluene (2 mL), and the atmosphere was replaced with nitrogen gas. The mixture was heated to 125 $^{\circ}\text{C}$, and tetrakis(triphenylphosphine)palladium (31.09 mg, 26.90 μmol , 0.2 *eq*) was then slowly added. The mixture was reacted at 125 $^{\circ}\text{C}$ for 48 hours. After completion of the reaction, the reaction solution was concentrated under reduced pressure with a water pump at 45 $^{\circ}\text{C}$ to give a crude product. The crude product was purified by thin layer chromatography on silica gel plate to give **WX007-1**. ^1H NMR (DMSO- d_6 , 400 MHz): δ (ppm) 9.26 (d, $J = 5.1$ Hz, 1H), 8.43 (d, $J = 5.1$ Hz, 1H), 7.47 (s, 1H), 7.28-7.39 (m, 3H), 4.73 (s, 2H), 3.50 (s, 3H), 1.55 (s, 6H).

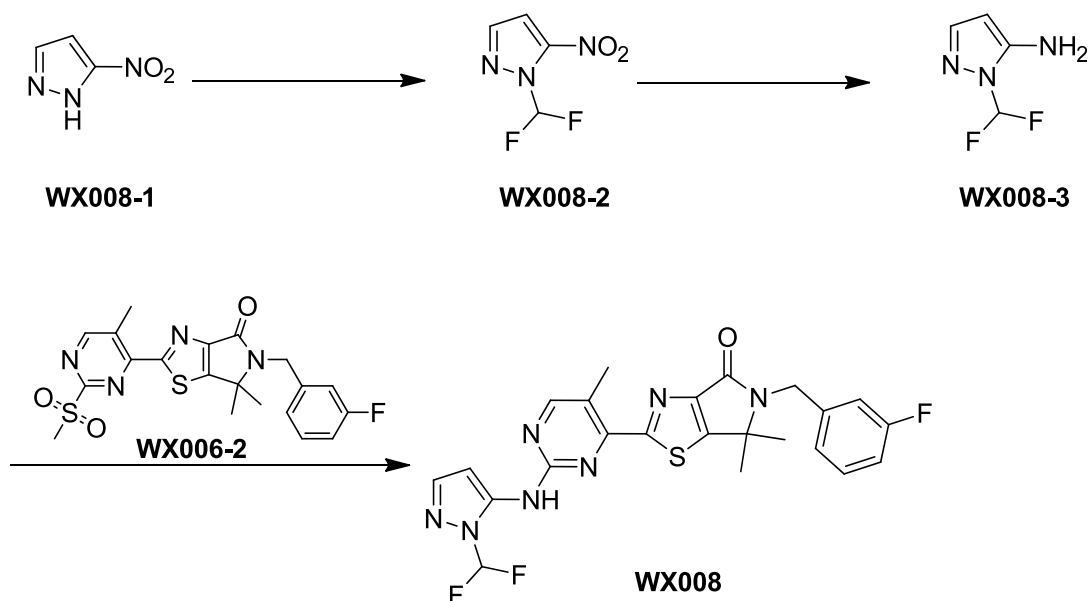
[00170] Step 2: synthesis of **WX007**

[00171] To a pre-dried reaction flask were added **WX007-1** (30 mg, 66.82 μmol , 1 *eq*) and **WX004-2** (6.76 mg, 66.82 μmol , 1 *eq*), and the mixture was then dissolved with dimethyl sulfoxide (1 mL). The mixture was reacted with stirring at 100 $^{\circ}\text{C}$ for 16 hours. After completion of the reaction, the reaction solution was concentrated, and the residue was purified by thin layer preparative chromatoshet to give **WX007**.

Example 8



[00172] Route of synthesis:



[00173] Step 1: synthesis of **WX008-2**

[00174] To a reaction flask were added **WX008-1** (9 g, 79.59 mmol, 1 eq), potassium carbonate (13.20 g, 95.51 mmol, 1.2 eq) and N,N-dimethylformamide (100 mL), and the atmosphere was replaced with nitrogen gas. The mixture was heated to 120 °C and stirred for 10 minutes. Sodium 2-chloro-2,2-difluoroacetate (24.27 g, 159.19 mmol, 2 eq) was then added in portions. The mixture solution was reacted at 120 °C for 20 minutes. After completion of the reaction, the reaction solution was diluted with water (800 mL), and extracted with ethyl acetate (200 mL*3). The organic phase was washed with saturated brine (300 mL), dried over anhydrous sodium sulfate, and filtered. The filtrate was concentrated under reduced pressure with a water pump, and the residue was purified by thin layer chromatography on silica gel plate to give **WX008-2**. ¹H NMR (DMSO-*d*₆, 400 MHz): δ (ppm) 7.96-8.28 (m, 2H), 7.50 (d, *J* = 2.0 Hz, 1H).

[00175] Step 2: synthesis of **WX008-3**

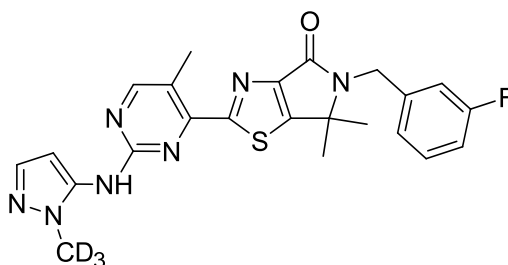
[00176] To a reaction flask were added palladium-carbon (50 mg, 367.91 μmol, 10% purity,

1 *eq*) and methanol (1 mL). The atmosphere was replaced with nitrogen gas, and then **WX008-2** (60 mg, 367.91 μ mol, 1 *eq*) was added. The atmosphere was then replaced with hydrogen gas, and the mixture solution was reacted under a pressure of 15 psi of a hydrogen (741.67 μ g, 367.91 μ mol, 1 *eq*) atmosphere at 25 °C for 1 hour. After completion of the reaction, the reaction solution was filtered through Celite, and the filtrate was concentrated to dryness under reduced pressure with a water pump to give **WX008-3**. ¹H NMR (DMSO-*d*₆, 400 MHz): δ (ppm) 7.36-7.69 (m, 1H), 7.33 (s, 1H), 5.84 (br s, 2H), 5.31 (d, *J* = 1.3 Hz, 1H).

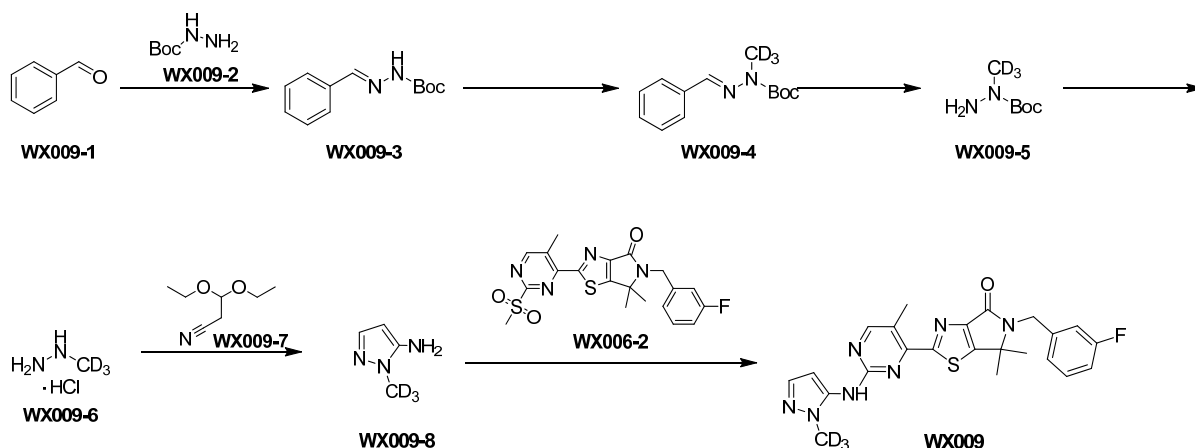
[00177] Step 3: synthesis of **WX008**

[00178] To a reaction flask were added **WX008-3** (17.88 mg, 134.37 μ mol, 1.5 *eq*), **WX006-2** (40 mg, 89.58 μ mol, 1 *eq*), dichloromethane (0.5 mL) and tetrahydrofuran (0.5 mL). The atmosphere was replaced with nitrogen gas, and the mixture was cooled to 0 °C. Lithium hexamethyldisilazide (1 M, 188.12 μ L, 2.1 *eq*) was then slowly added dropwise, and the mixture solution was reacted at 0 °C for 0.5 hours. After completion of the reaction, the reaction solution was concentrated, and the residue was purified by thin layer preparative chromatoshheet to give **WX008**.

Example 9



[00179] Route of synthesis:



[00180] Step 1: synthesis of **WX009-3**

[00181] To a reaction flask were added **WX009-1** (8.03 g, 75.67 mmol, 7.65 mL, 1 *eq*), **WX009-2** (10 g, 75.67 mmol, 1 *eq*) and tetrahydrofuran (100 mL). The atmosphere was replaced with nitrogen gas, and the mixture solution was reacted at 25 °C for 4 hours. After completion of the reaction, the reaction solution was rotary evaporated to dryness to give **WX009-3**. ¹H NMR (DMSO-*d*₆, 400 MHz): δ (ppm) 10.86 (br s, 1H), 8.00 (s, 1H), 7.59 (br d, *J* = 6.6 Hz, 2H), 7.31-7.45 (m, 3H), 1.47 (s, 9H).

[00182] Step 2: synthesis of **WX009-4**

[00183] To a reaction flask were added **WX009-3** (10 g, 45.40 mmol, 1 *eq*), potassium tert-butoxide (6.11 g, 54.48 mmol, 1.2 *eq*) and tetrahydrofuran (160 mL). The atmosphere was replaced with nitrogen gas, and then deuterated iodomethane (7.90 g, 54.48 mmol, 3.39 mL, 1.2 *eq*) was slowly added dropwise. The mixture solution was reacted at 25 °C for 16 hours. After completion of the reaction, the reaction solution was diluted with water (50 mL), and extracted with ethyl acetate (100 mL*3). The organic phase was washed with saturated brine (100 mL), dried over anhydrous sodium sulfate, and filtered. The filtrate was concentrated under reduced pressure with a water pump to give **WX009-4**. ¹H NMR (DMSO-*d*₆, 400 MHz): δ (ppm) 7.81 (s, 1H), 7.64-7.72 (m, 2H), 7.32-7.46 (m, 3H), 1.50 (s, 9H).

[00184] Step 3: synthesis of **WX009-5**

[00185] To a reaction flask were added wet palladium-carbon (2 g, 10% purity) and methanol (100 mL). The atmosphere was replaced with argon gas, and then **WX009-4** (10.5 g, 44.25 mmol, 1 *eq*) was added. The mixture solution was reacted under a hydrogen (89.19 mg, 44.25 mmol, 1 *eq*) atmosphere at a pressure of 50 psi at 50 °C for 48 hours. After completion of the reaction, the reaction solution was filtered through Celite, and washed with methanol (20 mL*2). The filtrate was concentrated to dryness under reduced pressure with a water pump to give **WX009-5**. ¹H NMR (DMSO-*d*₆, 400 MHz): δ (ppm) 4.48 (s, 2H), 1.40 (s, 9H).

[00186] Step 4: synthesis of **WX009-6**

[00187] To a reaction flask were added **WX009-5** (6.6 g, 44.23 mmol, 1 *eq*) and a solution of hydrochloric acid in ethyl acetate (4 M, 33.18 mL, 3 *eq*), and the mixture solution was reacted at 25° C for 16 hours. The reaction solution was rotary evaporated to dryness to give **WX009-6**. ¹H NMR (400 MHz, DMSO-*d*₆) δ ppm 4.95 (br s, 3 H).

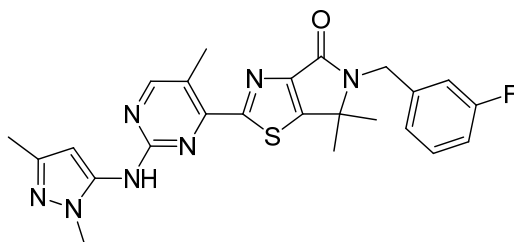
[00188] Step 5: synthesis of **WX009-8**

[00189] To a reaction flask were added **WX009-6** (1.5 g, 17.53 mmol, 1 *eq*, HCl), **WX009-7** (2.51 g, 17.53 mmol, 2.63 mL, 1 *eq*), acetic acid (2.11 g, 35.07 mmol, 2.01 mL, 2 *eq*), magnesium sulfate (5.02 g, 41.73 mmol, 2.38 *eq*) and ethanol (75 mL). The atmosphere was replaced with nitrogen gas, and then the mixture solution was reacted at 90 °C for 2 hours. After completion of the reaction, the reaction solution was diluted with saturated aqueous sodium bicarbonate solution (50 mL), and extracted with dichloromethane (50 mL*3). The organic phase was washed with saturated brine (50 mL), dried over anhydrous sodium sulfate, and filtered. The filtrate was concentrated under reduced pressure with a water pump at 45 °C to give a crude product, and the crude product was purified by column chromatography to give **WX009-8**. ¹H NMR (DMSO-*d*₆, 400 MHz): δ (ppm) 6.98 (d, *J* = 1.6 Hz, 1H), 5.24 (d, *J* = 1.8 Hz, 1H), 5.09 (br s, 2H).

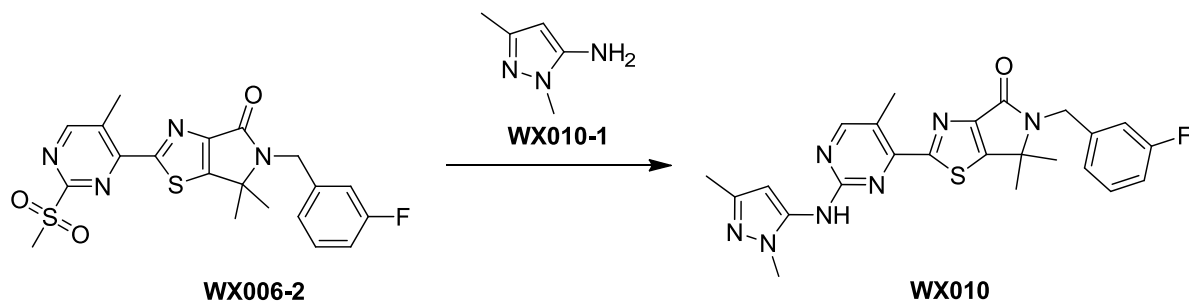
[00190] Step 6: synthesis of **WX009**

[00191] To a reaction flask were added **WX009-8** (13.46 mg, 134.37 μmol, 1.5 *eq*), **WX006-2** (40 mg, 89.58 μmol, 1 *eq*), dichloromethane (0.5 mL) and tetrahydrofuran (0.5 mL). The atmosphere was replaced with nitrogen gas, and the mixture was cooled to 0 °C. Lithium hexamethyldisilazide (1 M, 188.12 μL, 2.1 *eq*) was then slowly added dropwise, and the mixture solution was reacted at 0 °C for 0.5 hours. After completion of the reaction, the reaction solution was diluted with saturated aqueous ammonium chloride solution (5 mL), and extracted with dichloromethane (5 mL*3). The organic phase was washed with saturated brine (5 mL), dried over anhydrous sodium sulfate, and filtered. The filtrate was concentrated under reduced pressure with a water pump at 45 °C, and the residue was purified by thin layer preparative chromatoshheet to give **WX009**.

Example 10



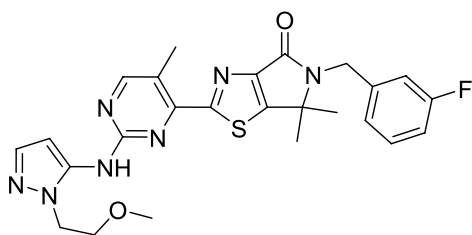
[00192] Route of synthesis:



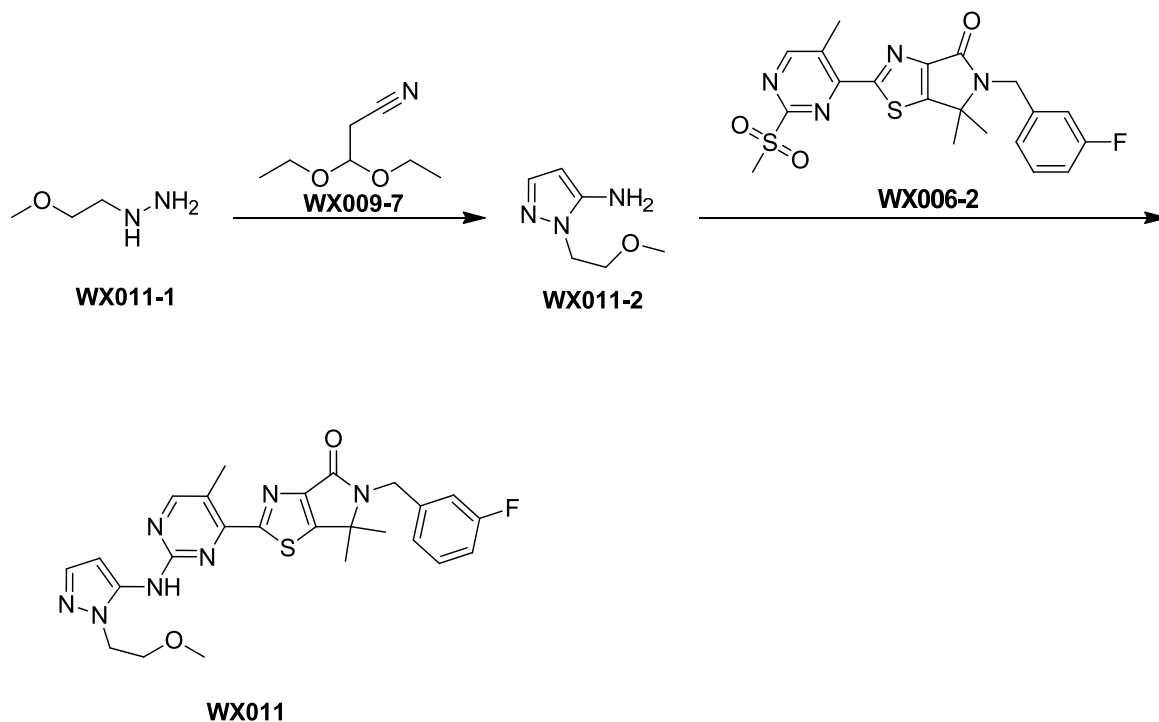
[00193] Step 1: synthesis of **WX010**

[00194] In a dry reaction flask, **WX006-2** (30 mg, 67.19 μmol , 1 *eq*) and **WX010-1** (15.68 mg, 141.09 μmol , 2.1 *eq*) were added to a mixed solution of tetrahydrofuran (1.5 mL) and dichloromethane (1.5 mL). The atmosphere was replaced with nitrogen gas, and the mixture was cooled to 0 °C. Lithium hexamethyldisilazide (1 M, 134.37 μL , 2 *eq*) was added, and the mixture was stirred at 0 °C for 0.5 hours, warmed to 25 °C, and stirred for another 1.5 hours. After completion of the reaction, the reaction solution was diluted with water (10 mL), and extracted with dichloromethane (10 mL*3). The layers were separated. The organic phase was then collected, and the collected organic phase was sequentially with saturated brine (10 mL*3), dried over anhydrous sodium sulfate, and concentrated under reduced pressure. The residue was purified by thin layer preparative chromatoshheet to give **WX010**.

Example 11



[00195] Route of synthesis:



[00196] Step 1: synthesis of **WX011-2**

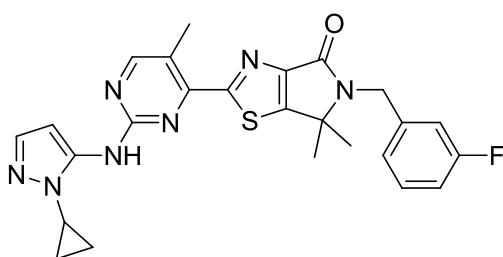
[00197] To a dry reaction flask were added **WX009-7** (226.22 mg, 1.58 mmol, 237.13 μ L, 1 *eq*), magnesium sulfate (452.63 mg, 3.76 mmol, 2.38 *eq*), acetic acid (189.75 mg, 3.16 mmol, 180.72 μ L, 2 *eq*) and ethanol (2.5 mL). The atmosphere was replaced with nitrogen gas, and the mixture was reacted at 25 °C for 1 hour. **WX011-1** (200 mg, 1.58 mmol, 1 *eq*, HCl) was then added, and the mixture was reacted at 80 °C for 2 hours and at 90 °C for another 12 hours. After completion of the reaction, the reaction solution was poured into 3 mL of saturated aqueous sodium bicarbonate solution. The mixture was extracted with dichloromethane (3 mL*2). The organic phases were combined, washed with 3 mL of saturated brine, dried over anhydrous sodium sulfate, and concentrated under reduced pressure to give a crude product. The crude product was purified by thin layer chromatography on silica gel plate to give **WX011-2**. ¹H NMR (CH₃Cl-*d*, 400 MHz): δ (ppm) 7.25 (d, *J* = 1.8 Hz, 1H), 5.50 (d, *J* = 2.0 Hz, 1H), 4.15-4.22 (m, 2H), 3.83-4.11 (m, 2H), 3.68-3.73 (m, 2H), 3.34 (s, 3H).

[00198] Step 2: synthesis of **WX011**

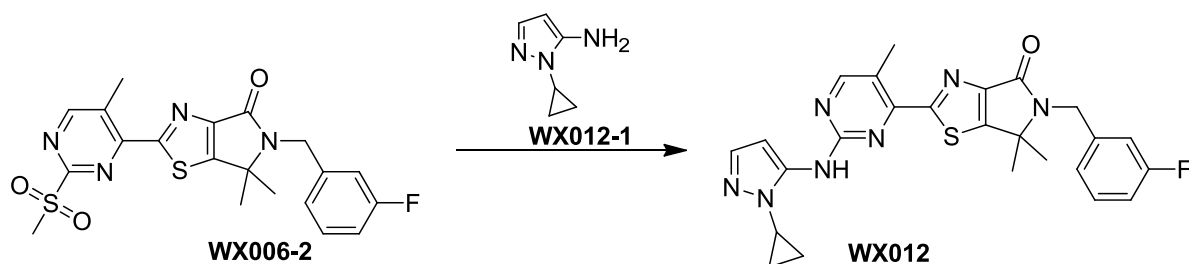
[00199] In a dry reaction flask, **WX006-2** (40 mg, 89.58 μ mol, 1 *eq*) and **WX011-2** (26.56 mg, 188.12 μ mol, 2.1 *eq*) were added to a mixed solution of tetrahydrofuran (2 mL) and dichloromethane (2 mL). The atmosphere was replaced with nitrogen gas, and the mixture

was cooled to 0 °C. Lithium hexamethyldisilazide (1 M, 179.16 μ L, 2 *eq*) was added, and the mixture was stirred at 0 °C for 0.5 hours, warmed to 25 °C, and stirred for another 1 hour. After completion of the reaction, the reaction solution was diluted with water (5 mL), and extracted with dichloromethane (5 mL*3). The layers were separated. The organic phase was then collected, and the collected organic phase was sequentially with saturated brine (5 mL*3), dried over anhydrous sodium sulfate, and concentrated under reduced pressure. The residue was purified by thin layer preparative chromatoshheet to give **WX011**.

Example 12



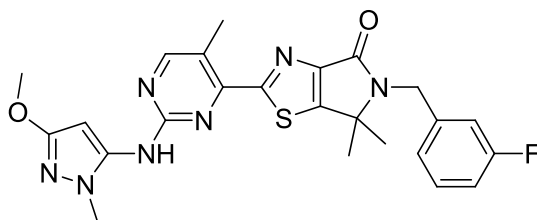
[00200] Route of synthesis:



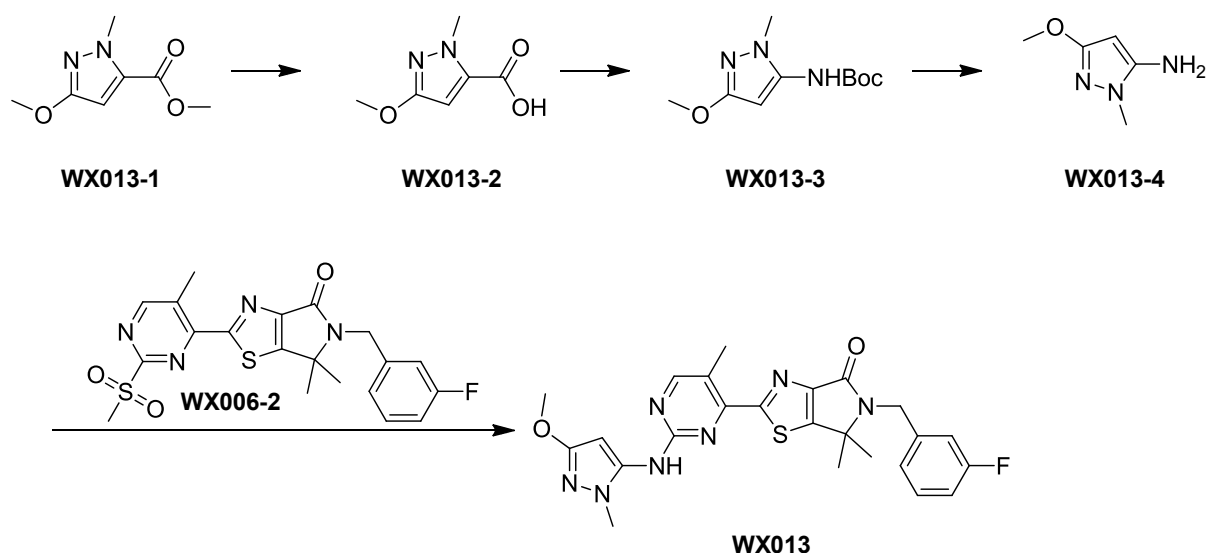
[00201] Step 1: synthesis of **WX012**

[00202] To a reaction flask were added **WX006-2** (40 mg, 89.58 μ mol, 1 *eq*), **WX012-1** (16.55 mg, 134.37 μ mol, 1.5 *eq*), dichloromethane (1 mL) and tetrahydrofuran (1 mL). The atmosphere was replaced with nitrogen gas, and the mixture was cooled to 0°C. Lithium hexamethyldisilazide (1 M, 188.12 μ L, 2.1 *eq*) was slowly added dropwise, and the mixture solution was reacted at 0 °C for 0.5 hours. After completion of the reaction, the reaction solution was concentrated and the residue was purified by thin layer preparative chromatoshheet to give **WX012**.

Example 13



[00203] Route of synthesis:



[00204] Step 1: synthesis of **WX013-2**

[00205] **WX013-1** (800 mg, 4.70 mmol, 1 *eq*) and lithium hydroxide monohydrate (986.42 mg, 23.51 mmol, 5 *eq*) were dissolved in a mixed solution of water (8 mL) and tetrahydrofuran (8 mL). The mixture was stirred at 25 °C for 2 hours. After completion of the reaction, the reaction solution was directly rotary evaporated to dryness, and then extracted with water (10 mL) and dichloromethane (10 mL*3). The organic phase was collected, washed with saturated brine, dried over anhydrous sodium sulfate, and filtered. The filtrate was rotary evaporated to dryness to give **WX013-2**.

[00206] Step 2: synthesis of **WX013-3**

[00207] **WX013-2** (500 mg, 3.20 mmol, 1 *eq*), diphenylphosphoryl azide (889.00 mg, 3.23 mmol, 700 μ L, 1.01 *eq*) and triethylamine (1.45 g, 14.37 mmol, 2 mL, 4.49 *eq*) were dissolved in tert-butanol (10 mL). The atmosphere was replaced with nitrogen for three times, and then the mixture was stirred at 85 °C for 16 hours. After completion of the reaction, the reaction solution was rotary evaporated to dryness to give a crude product. The crude product was purified by column chromatography to give **WX013-3**. ^1H NMR (400 MHz, CD_3Cl) δ

ppm 1.50 (s, 9 H) 3.58 (s, 3 H) 3.84 (s, 3 H) 5.61 (br s, 1 H).

[00208] Step 3: synthesis of **WX013-4**

[00209] To a reaction flask were added **WX013-3** (520 mg, 2.29 mmol, 1 *eq*) and a solution of hydrochloric acid in ethyl acetate (4 M, 5 mL, 8.74 *eq*), and the mixture was stirred at 25 °C for 2 hours. After completion of the reaction, the reaction solution was extracted with water (10 mL). The aqueous phase was collected and then rotary evaporated to dryness to give a crude product. The crude product was purified by column chromatography to give **WX013-4**. ¹H NMR (400 MHz, CD₃Cl) δ ppm 3.53 (s, 3 H) 3.82 (s, 3 H) 5.01 (s, 1 H).

[00210] Step 4: synthesis of **WX013**

[00211] To a dry reaction flask were added **WX006-2** (50 mg, 111.98 μmol, 1 *eq*), **WX013-4** (29.90 mg, 235.15 μmol, 2.1 *eq*), dichloromethane (1 mL) and tetrahydrofuran (1 mL). The atmosphere was replaced with nitrogen gas, and the mixture was cooled to 0 °C. Lithium hexamethyldisilazide (1 M, 223.96 μL, 2 *eq*) was added dropwise. The mixture was reacted at 0 °C for 0.5 hours and at 25 °C for another 1 hour. After completion of the reaction, the reaction solution was quenched with 10 mL of water, and extracted with 20 mL of dichloromethane. The layers were separated. The organic phase was collected, and the aqueous phase was extracted with dichloromethane (20 mL*3). The organic phases were combined, and the combined organic phase was sequentially washed with saturated brine (20 mL*3), dried over anhydrous sodium sulfate, and concentrated under reduced pressure. After completion of the concentration, the residue was purified by thin layer preparative chromatoshheet to give **WX013**.

[00212] The data of ¹H NMR spectrum and mass spectrum of each example were shown in Table 1.

[00213] Table 1

Example	Compound	NMR	MS <i>m/z</i>
1	WX001	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ ppm 9.82 (br s, 1H), 8.70 (d, <i>J</i> = 4.9 Hz, 1H), 7.56 (d, <i>J</i> = 5.0 Hz, 1H), 7.46 (s, 1H), 7.41 (d, <i>J</i> = 1.6 Hz, 1H), 7.36 (s, 3H), 6.35 (s, 1H), 4.72 (s, 2H), 3.74 (s, 3H), 1.53 (s, 6H).	466 [M+H] ⁺

2	WX002	¹ H NMR (DMSO- <i>d</i> ₆ , 400 MHz) δ (ppm) 9.78 (br s, 1H), 8.68 (d, <i>J</i> = 4.9 Hz, 1H), 7.54 (d, <i>J</i> = 5.0 Hz, 1H), 7.31-7.42 (m, 2H), 7.16-7.24 (m, 2H), 7.07 (td, <i>J</i> = 8.6, 2.2 Hz, 1H), 6.33 (d, <i>J</i> = 1.6 Hz, 1H), 4.71 (s, 2H), 3.72 (s, 3H), 1.52 (s, 6H).	450 [M+H] ⁺
3	WX003	¹ H NMR (DMSO- <i>d</i> ₆ , 400 MHz) δ (ppm) 9.80 (br s, 1H), 8.69 (d, <i>J</i> = 5.0 Hz, 1H), 7.54 (d, <i>J</i> = 5.0 Hz, 1H), 7.32-7.50 (m, 3H), 7.25 (m, 1H), 6.33 (d, <i>J</i> = 1.4 Hz, 1H), 4.68 (s, 2H), 3.72 (s, 3H), 1.52 (s, 6H).	468 [M+H] ⁺
4	WX004	¹ H NMR (400 MHz, DMSO- <i>d</i> ₆) δ (ppm) 8.52 (br d, <i>J</i> = 4.8 Hz, 1H), 7.61 (br s, 1H), 7.33-7.48 (m, 1H), 7.18-7.30 (m, 3H), 7.04-7.15 (m, 1H), 4.73 (s, 2H), 3.83-4.05 (m, 3H), 3.43 (m, 2H), 1.89 (br s, 2H), 1.49-1.64 (m, 8H).	454 [M+H] ⁺
5	WX005	¹ H NMR (DMSO- <i>d</i> ₆ , 400 MHz) δ (ppm) 9.61 (s, 1H), 8.60 (s, 1H), 7.46 (s, 1H), 7.39 (d, <i>J</i> = 1.9 Hz, 1H), 7.30-7.38 (m, 3H), 6.34 (d, <i>J</i> = 1.8 Hz, 1H), 4.71 (s, 2H), 3.72 (s, 3H), 2.59 (s, 3H), 1.51 (s, 6H).	480 [M+1] ⁺
6	WX006	¹ H NMR (400MHz, DMSO- <i>d</i> ₆) δ 9.61 (s, 1H), 8.60 (s, 1H), 7.40-7.33 (m, 2H), 7.25-7.18 (m, 2H), 7.04-7.11 (m, 1H), 6.31-6.35 (m, 1H), 4.71 (s, 2H), 3.72 (s, 3H), 2.59 (s, 3H), 1.52 (s, 6H).	464 [M+1] ⁺
7	WX007	¹ H NMR (DMSO- <i>d</i> ₆ , 400 MHz) δ (ppm) 8.51 (br d, <i>J</i> = 4.8 Hz, 1H), 7.51-7.64 (m, 1H), 7.46 (s, 1H), 7.30-7.40 (m, 3H), 7.26 (d, <i>J</i> = 4.6 Hz, 1H), 4.71 (s, 2H), 3.85-4.04 (m, 3H), 3.42 (br t, <i>J</i> = 11.2 Hz, 2H), 1.89 (br d, <i>J</i> = 4.5 Hz, 2H), 1.48-1.64 (m, 8H).	470 [M+1] ⁺
8	WX008	¹ H NMR (DMSO- <i>d</i> ₆ , 400 MHz) δ (ppm) 10.04 (br s, 1H), 8.65 (s, 1H), 7.59-7.94 (m, 2H), 7.31-7.42 (m, 1H),	500 [M+1] ⁺

		7.16-7.27 (m, 2H), 7.08 (br t, $J = 8.6$ Hz, 1H), 6.54 (s, 1H), 4.72 (s, 2H), 2.61 (s, 3H), 1.52 (s, 6H).	
9	WX009	^1H NMR (DMSO- d_6 , 400 MHz) δ (ppm) 9.62 (s, 1H), 8.60 (s, 1H), 7.32-7.41 (m, 2H), 7.17-7.26 (m, 2H), 7.07 (td, $J = 8.5, 2.4$ Hz, 1H), 6.34 (d, $J = 1.8$ Hz, 1H), 4.71 (s, 2H), 2.59 (s, 3H), 1.51 (s, 6H).	467 [M+1] ⁺
10	WX010	^1H NMR (DMSO- d_6 , 400 MHz) δ (ppm) 9.52 (s, 1H), 8.59 (s, 1H), 7.31-7.42 (m, 1H), 7.17-7.27 (m, 2H), 7.02-7.13 (m, 1H), 6.08 (s, 1H), 4.72 (s, 2H), 3.63 (s, 3H), 2.59 (s, 3H), 2.15 (s, 3H), 1.53 (s, 6H).	478 [M+1] ⁺
11	WX011	^1H NMR (400MHz, DMSO- d_6) δ 9.47 (s, 1H), 8.60 (s, 1H), 7.43 (d, $J=2.0$ Hz, 1H), 7.40 - 7.32 (m, 1H), 7.25 - 7.16 (m, 2H), 7.07 (dt, $J=2.0, 8.6$ Hz, 1H), 6.40 (d, $J=1.8$ Hz, 1H), 4.71 (s, 2H), 4.25 (t, $J=5.6$ Hz, 2H), 3.67 (t, $J=5.7$ Hz, 2H), 3.21 (s, 3H), 2.59 (s, 3H), 1.51 (s, 6H).	508 [M+1] ⁺
12	WX012	^1H NMR (400 MHz, DMSO- d_6) δ ppm 0.88 - 0.95 (m, 2 H), 0.96 - 1.03 (m, 2 H), 1.51 (s, 6 H), 2.59 (s, 3 H), 3.54 (tt, $J=7.24, 3.71$ Hz, 1 H), 4.71 (s, 2 H), 6.35 (d, $J=1.63$ Hz, 1 H), 7.02 - 7.12 (m, 1 H), 7.17 - 7.26 (m, 2 H), 7.32 - 7.41 (m, 2 H), 8.61 (s, 1 H), 9.54 (s, 1 H).	490 [M+1] ⁺
13	WX013	^1H NMR (DMSO- d_6 , 400MHz) δ (ppm) 9.66 (s, 1H), 8.61 (s, 1H), 7.31-7.43 (m, 1H), 7.15-7.27 (m, 2H), 7.03-7.12 (m, 1H), 5.81 (s, 1H), 4.72 (s, 2H), 3.75 (s, 3H), 3.57 (s, 3H), 2.59 (s, 3H), 1.52 (s, 6H).	494 [M+1] ⁺

Assay example 1. Assay of in vitro kinase activity:

[00214] 1. Purpose of the assay:

[00215] The ability of compounds to inhibit ERK2 kinase activity was measured.

[00216] 2. Assay buffer:

[00217] 20 mM Hepes (pH 7.5), 10 mM MgCl₂, 1 mM ethylenedis(oxyethylenitrilo)tetraacetic acid (EGTA), 0.02% Brij35, 0.02 mg/mL bovine serum albumin (BSA), 0.1 mM Na₃VO₄, 2 mM dithiothreitol (DTT), 1% DMSO.

[00218] 3. Processing of compound:

[00219] The assay compound was dissolved in 100% DMSO to prepare a stock solution of specific concentration. The compound was serially diluted in DMSO solution using Integra Viaflo Assist smart pipette.

[00220] 3. Method of the assay

[00221] 1) The substrate MBP was prepared in freshly prepared reaction buffer;

[00222] 2) ERK2 kinase was added to the above-mentioned MBP solution and mixed gently;

[00223] 3) The compound dissolved in 100% DMSO was added to the kinase reaction system using ultrasound technology (Echo550; nanoliter range), and the mixture was incubated at room temperature for 20 minutes;

[00224] 4) ³³P-ATP (specific concentration of 10 µCi/µL) was added to the reaction system, and the reaction was started at this time;

[00225] 5) The mixture was incubated at room temperature for 2 hours;

[00226] 6) The amount of radioactivity was detected by filter-binding method;

[00227] 7) ERK2 kinase activity was calculated as the ratio of the remaining kinase activity in the assay sample to the kinase activity of the control group (treated by DMSO). Curve was fitted using Prism (GraphPad software) and IC₅₀ values were calculated.

[00228] 4. The assay results were shown in Table 2:

[00229] Table 2: Results of kinase activity assay in vitro

Compound	ERK2
	IC ₅₀ (nM)
WX001	0.24
WX002	0.35
WX003	0.37
WX004	0.25

WX005	0.05
WX006	0.30
WX007	0.39
WX008	0.52
WX009	0.64
WX0010	0.14
WX0011	0.98
WX0012	0.48
WX0013	0.19

[00230] Conclusion: The compounds of the present disclosure exhibit excellent activity of inhibiting ERK2 kinase.

Assay example 2. Assay of in vitro cell proliferation inhibition:

[00231] 1. Purpose of the assay:

[00232] The ability of compounds to inhibit the proliferation of HT29 tumor cells was measured.

[00233] 2. Processing of compound:

[00234] The assay compound was dissolved in 100% DMSO to prepare 10 mM stock solution.

[00235] 3. Method and step of the assay

[00236] 1) UV light of a biological safety cabin was turned on, and 30 minutes were counted down;

[00237] 2) In a 37 °C water bath, RPMI1640 medium and trypsin were preheated;

[00238] 3) After completion of the UV irradiation, the biological safety cabin was opened. The preheated medium, trypsin and phosphate buffered saline (PBS), etc. were wiped with alcohol and placed in the biological safety cabin;

[00239] 4) HT29 cells were removed from the incubator, and the old medium was removed in biological safety cabin. 10 ml of PBS was added. The mixture was shaken gently, and then PBS was removed;

[00240] 5) 1.5 ml of preheated 0.25% trypsin was added. The culture vessel was shaken horizontally so that the trypsin evenly covered the cells at the bottom, and placed in an incubator for 2 minutes;

[00241] 6) Cell digestion was stopped with complete medium, and the cell suspension was pipetted to homogeneity and counted;

[00242] 7) According to the result of cell counting, the density of cell suspension was adjusted to 1500 cells per well, and the cell suspension was seeded at 50 μ l per well;

[00243] 8) The stock solution of compounds was serially diluted in DMSO solution, and compounds were added to cell plate using Tecan;

[00244] 9) The compound-added cell plate and CellTiterGlo were equilibrated at room temperature, and 25 microliters of CellTiterGlo was then added to each well. The cell plate was shaken for 1-2 minutes and then allowed to stand for 10 minutes. The signal value was then detected. The data were analyzed using XL-Fit, and the IC₅₀ of each compound was calculated.

[00245] 4. The assay results were shown in Table 3:

[00246] Table 3: Results of cell activity assay in vitro

Compound	HT29
	IC ₅₀ (nM)
WX001	14
WX002	24
WX003	27
WX004	35
WX005	8
WX006	7
WX007	26
WX008	58
WX009	6
WX010	27
WX011	26

WX012	37
WX013	18

[00247] Conclusion: The compounds of the present disclosure exhibit excellent activity of inhibiting the proliferation of HT29 cells.

Assay example 3. Assay of in vivo DMPK:

[00248] In vivo DMPK assay in mouse

[00249] 1. Purpose of the assay:

[00250] Female BALB/c mice were used as assay animals to determine the blood concentration of compounds and evaluate the pharmacokinetic behavior after a single administration.

[00251] 2. Procedure of the assay:

[00252] Eight healthy adult female BALB/c mice were selected, wherein 4 mice were in the intravenous injection group and 4 mice were in the oral group. The vehicle in the intravenous injection group was 5% DMSO+95% (20% HP- β -CD). The compound to be assayed was mixed with an appropriate amount of vehicle for intravenous injection, vortexed and sonicated to prepare a clear solution of 0.5 mg/mL. The clear solution was filtered by a microporous membrane, and then ready for use. The vehicle in the oral group was 5% DMSO+95% (20% HP- β -CD). The compound to be assayed was mixed with the vehicle, vortexed and sonicated to prepare a solution of 0.3 mg/mL. Mice were administered 1 mg/kg intravenously or 3 mg/kg orally, and then whole blood was collected for a certain period. Plasma was prepared. The drug concentration was analyzed by LC-MS/MS method, and the pharmacokinetic parameters were calculated by Phoenix WinNonlin software (Pharsight, USA).

Note: DMSO: dimethyl sulfoxide; HP- β -CD: hydroxypropyl- β -cyclodextrin.

[00253] 3. The assay results were shown in Table 4:

[00254] Table 4: Results of PK assay of the compounds

Compound	C _{max} (nM)	F%	Oral DNAUC	V _{dss} (L/kg)	Cl (mL/min/kg)	T _{1/2} (h)
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			(nM.h/mpk)			
WX001	3355	86%	2153	1.1	14.3	0.9
WX005	1029	NA	468	NA	NA	NA
WX006	1035	34%	530	1.7	23.0	1.0
WX009	1170	67%	820	1.7	28.0	1.7

Note: $C_{\max}$ is maximum concentration; F% is oral bioavailability; DNAUC is $AUC_{PO}/Dose$, AUC_{PO} is oral exposure, and Dose is drug dose; V_{dss} is distribution volume; Cl is clearance rate; $T_{1/2}$ is half-life; and NA means not assayed.

[00255] Conclusion: The compounds of the present disclosure exhibit excellent oral exposure and bioavailability.

Assay example 4. Assay of in vivo efficacy in mouse H358 model:

[00256] 1. Purpose of the assay:

[00257] The anti-tumor effect of **WX006** was evaluated using a subcutaneous xenograft tumor model of human non-small cell lung cancer H358 cells in nude mouse.

[00258] 2. Assay animal:

[00259] Species: mouse

[00260] Strain: BALB/c nude mouse

[00261] Age: 6-7 weeks old

[00262] Gender: female

[00263] Weight: 20 grams

[00264] Supplier: Shanghai Sippe-Bk Lab Animal Co., Ltd.

[00265] Animal certificate No.: 20180006017149

[00266] 3. Environment for rearing:

[00267] Animals were reared in IVC (independent air supply system, and constant temperature and humidity) cages (4 animals per cage) in SPF grade of animal room at a temperature of 20-26°C and a humidity of 40-70%;

[00268] Cage: The cage was made of polycarbonate, and had a volume of 300 mm × 180 mm × 150 mm. The bedding material was corncob, and replaced once a week;

[00269] Food: Assay animals had free access to food (sterilized by irradiation, dry pelleted food) throughout the assay period;

[00270] Drinking water: Assay animals had free access to sterilized water;

[00271] Cage identification: The animal information card for each cage should indicate the number, gender, strain, date of receipt of animals in the cage, assay numbering of administration schedule, group and start date of the assay;

[00272] Animal identification: Assay animals were identified by ear tags.

[00273] 4. Assay procedure:

[00274] 1) Assay cells and culture: Human non-small cell lung cancer H358 cells were cultured in monolayer in vitro. The culture conditions were 1640 medium plus 10% fetal bovine serum, and a 37°C 5% CO₂ incubator. Routine digestion with trypsin-EDTA was performed three times a week for passage. When the cell saturation was 80%-90% and the amount reached the requirement, the cells were harvested, counted, and seeded;

[00275] 2) Tumor tissue inoculation and grouping: 0.1 mL (5×10⁵) H358 cells were subcutaneously inoculated into the right armpit of each mouse. When the average tumor volume reached 100 mm³, the animals were randomly divided into 2 groups and the administration was started. The grouping and administration schedule of the assay were shown in Table 5.

[00276] Table 5: Grouping and administration schedule of assay animals

Group	Number of animals	Drug	Dosage (mg/kg)	Cycle of administration	Route and frequency of administration
1	6	Solvent control (Vehicle)	--	28 days	Oral administration (PO), once daily (QD)
2	6	WX006	30	28 days	Oral administration (PO), once daily (QD)

[00277] 3) Daily observation of assay animals: The development of this assay protocol and any modifications were evaluated and approved by the Institutional Animal Care and Use Committee (IACUC). The use and welfare of assay animals were carried out in accordance with the regulations of the Association for Assessment and Accreditation of Laboratory

Animal Care (AAALAC). Animals were monitored daily for health and death. Routine examinations included observation of tumor growth and the effects of drug treatment on the animals' daily behavior such as behavioral activities, food and water intake (visual inspection only), weight changes (weight measurements twice a week), appearance signs or other abnormalities. Animal deaths and side effects in each group were recorded based on the number of animals in each group.

[00278] 4) Formulation of assay compound

[00279] a) Vehicle group: 5% DMSO + 95% (20% HP- β -CD).

[00280] b) Assay compound group: A quantitative amount of the assay compound was weighed in a formulation bottle. A corresponding volume of DMSO was added and then the mixture was vortexed to obtain a clear solution. A corresponding volume of 20% HP- β -CD was added and then the mixture was vortexed to obtain a homogeneous suspension. The compound was formulated every three days.

[00281] 5) Tumor measurement and assay indicator:

[00282] a) Tumor diameter was measured twice a week with a vernier caliper. The calculation formula of tumor volume was: $TV = 1/2 \times a \times b^2$, wherein a and b represent the long and short diameters of tumor, respectively;

[00283] b) The tumor-inhibitory efficacy of the compound was evaluated by TGI (%). TGI (%) reflected the inhibition rate of tumor growth. TGI (%) was calculated as follows: $TGI (\%) = \{[1 - (\text{average tumor volume at the end of administration of a treatment group} - \text{average tumor volume at the beginning of administration of this treatment group})] / (\text{average tumor volume at the end of treatment in a solvent control group} - \text{average tumor volume at the beginning of treatment in the solvent control group})\} \times 100\%$.

[00284] 5. Assay results:

[00285] 1) As shown in Table 6 and Figure 1, in the subcutaneous xenograft tumor model of human non-small cell lung cancer H358 cells in nude mouse, when administered orally to the 28th day, **WX006** 30 mg/kg had a significant inhibitory effect on tumor growth with a TGI of 94%.

[00286] 2) The body weight of assay animals was used as a reference index for indirect determination of drug toxicity. As shown in Figure 2, when administered to the 28th day, the

body weight of all animals in the solvent control group and **WX006** group did not decrease significantly, and there was no morbidity or death.

[00287] Table 6: Results of in vivo efficacy assay in mouse H358 model

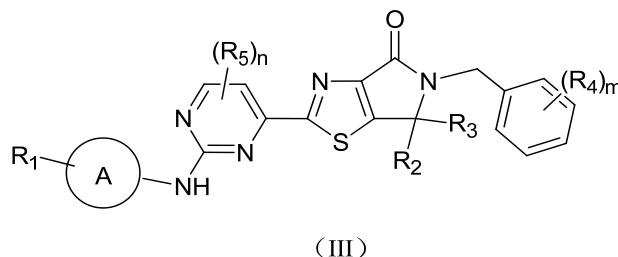
Drug	TGI
WX006 (30 mg/kg, PO, QD)	94%

[00288] 6. Assay conclusion: **WX006** can significantly inhibit the growth of tumor at the administration dose. During the administration, the body weight of animals is not observed to decrease significantly, and the tolerance is good.

CLAIMS

What is claimed is:

1. A compound represented by formula (III) or a pharmaceutically acceptable salt thereof,



wherein

R₁ is selected from H, C₁₋₃ alkyl and C₃₋₅ cycloalkyl, wherein the C₁₋₃ alkyl and C₃₋₅ cycloalkyl are optionally substituted by 1, 2 or 3 R_a;

R₂ and R₃ are each independently C₁₋₃ alkyl, wherein the C₁₋₃ alkyl is optionally substituted by 1, 2 or 3 R_b;

R₄ is selected from H, F, Cl, Br, I and C₁₋₃ alkyl, wherein the C₁₋₃ alkyl is optionally substituted by 1, 2 or 3 R_c;

R₅ is selected from F, Cl, Br, I and C₁₋₃ alkyl, wherein the C₁₋₃ alkyl is optionally substituted by 1, 2 or 3 R_e;

m is selected from 0, 1 and 2;

n is selected from 0, 1 and 2;

ring A is selected from pyrazolyl and tetrahydropyranyl, wherein the pyrazolyl and tetrahydropyranyl are optionally substituted by 1, 2 or 3 R_d;

R_a, R_b, R_c and R_e are each independently selected from D, F, Cl, Br, I, OH and OCH₃;

R_d is selected from F, Cl, Br, I, CH₃ and OCH₃.

2. The compound or pharmaceutically acceptable salt thereof according to claim 1, wherein R₁ is selected from H, CH₃, and cyclopropyl, wherein the CH₃ and cyclopropyl are optionally substituted by 1, 2, or 3 R_a.

3. The compound or pharmaceutically acceptable salt thereof according to claim 2, wherein R₁ is selected from H, CH₃, CHF₂, CD₃, CH₂CH₂OCH₃ and cyclopropyl.

4. The compound or pharmaceutically acceptable salt thereof according to claim 1, wherein R_2 and R_3 are each independently selected from CH_3 and CH_2CH_3 , wherein the CH_3 and CH_2CH_3 are optionally substituted by 1, 2 or 3 R_b .

5. The compound or pharmaceutically acceptable salt thereof according to claim 4, wherein R_2 and R_3 are each independently CH_3 .

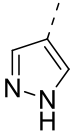
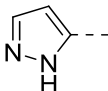
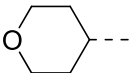
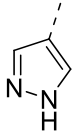
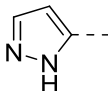
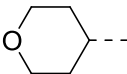
6. The compound or pharmaceutically acceptable salt thereof according to claim 1, wherein R_4 is selected from H, F, Cl, Br, I and CH_3 , wherein the CH_3 is optionally substituted by 1, 2 or 3 R_c .

7. The compound or pharmaceutically acceptable salt thereof according to claim 6, wherein R_4 is selected from H, F, Cl, Br, I and CH_3 .

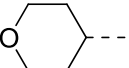
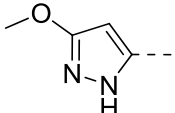
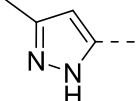
8. The compound or pharmaceutically acceptable salt thereof according to claim 1, wherein R_5 is selected from F, Cl, Br, I and CH_3 , wherein the CH_3 is optionally substituted by 1, 2 or 3 R_e .

9. The compound or pharmaceutically acceptable salt thereof according to claim 8, wherein R_5 is selected from F, Cl, Br, I and CH_3 .

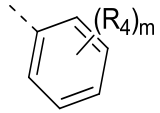
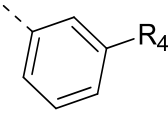
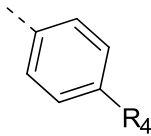
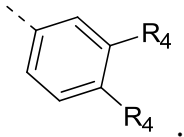
10. The compound or pharmaceutically acceptable salt thereof according to claim 1, wherein

ring A is selected from , , and , wherein the , , and  are optionally substituted by 1, 2, or 3 R_d .

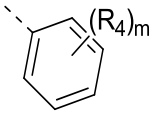
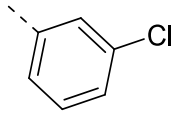
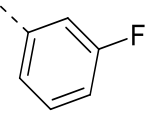
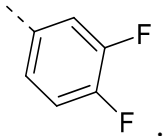
11. The compound or pharmaceutically acceptable salt thereof according to claim 10, wherein

ring A is selected from , ,  and .

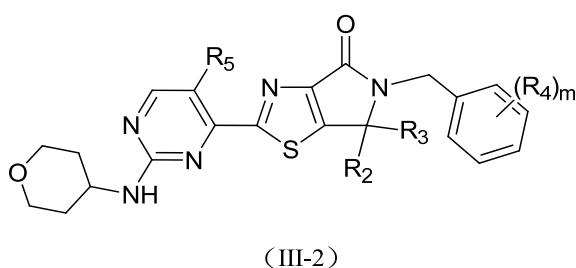
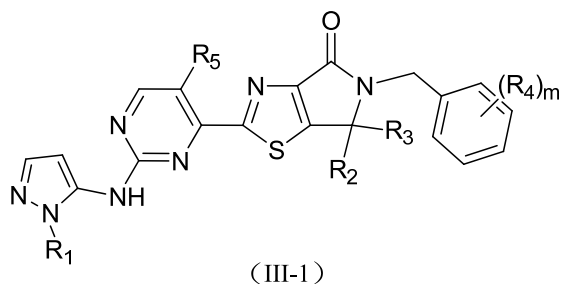
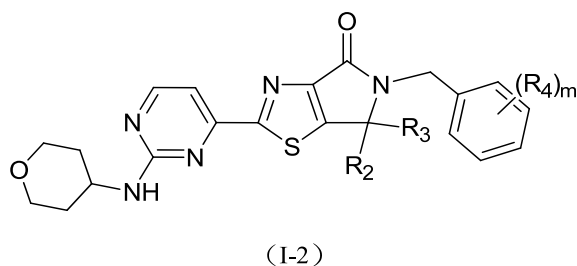
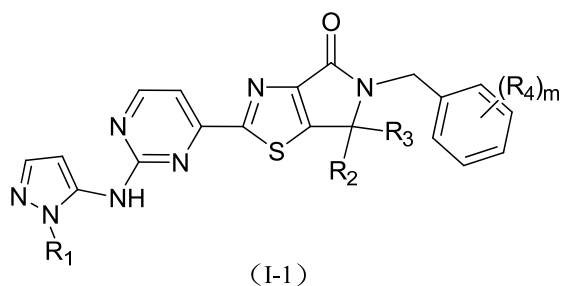
12. The compound or pharmaceutically acceptable salt thereof according to claim 1, wherein

the structural moiety  is selected from ,  and .

13. The compound or pharmaceutically acceptable salt thereof according to claim 12, wherein

the structural moiety  is selected from ,  and .

14. The compound or pharmaceutically acceptable salt thereof according to any one of claims 1-9, wherein the compound is selected from



wherein

R₁ is as defined in any one of claims 1-3;

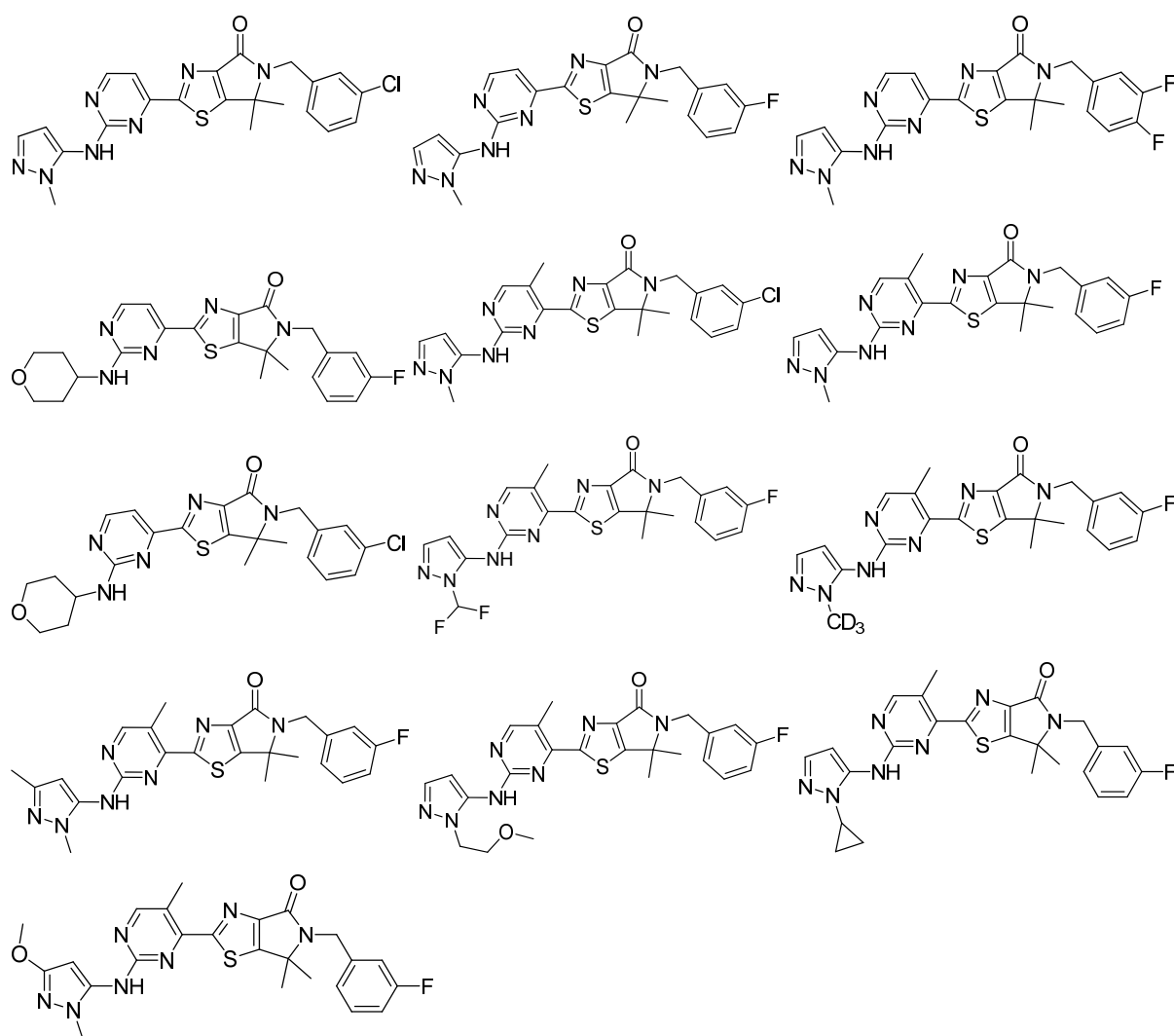
R₂ and R₃ are as defined in any one of claims 1, 4 or 5;

R₄ is as defined in any one of claims 1, 6 or 7;

R₅ is as defined in any one of claims 1, 8 or 9;

m is as defined in claim 1.

15. A compound represented by the following formula or a pharmaceutically acceptable salt thereof,



16. Use of the compound or pharmaceutically acceptable salt thereof according to any one of claims 1-15 in the manufacture of a medicament for treating diseases related to ERK.

17. The use according to claim 16, characterized in that the medicament related to ERK inhibitor is a medicament for treating solid tumor.