

FLUX CORED WIRE FOR GAS SHIELDED ARC WELDING

BACKGROUND OF THE INVENTION

Field of the Invention

[0001]

The present disclosure relates to a flux cored wire for gas shielded arc welding.

Description of Related Art

[0002]

In manufacturing procedures of large-sized structures, such as ships and bridges, fillet welding and/or butt welding is often applied. In these cases, a flux cored wire is used as welding material. Weld metals in ships, bridges, etc., are required to have excellent low-temperature toughness and mechanical properties in terms of construction work. The flux cored wire for use in welding is further desired to have excellent welding workability.

[0003]

For this reason, in the related art, for example, a flux cored wire is proposed that has improved welding workability by specifying a particle size of a Mg raw material used in a flux (see JP 61-202795 A). A flux cored wire is also proposed that has improved mechanical properties of a weld metal, in addition to improved welding workability, by specifying wire

components (see JP 7-164184 A).

[0004]

In general, hydrogen exists in a weld metal. As the hydrogen content in the weld metal increases, cold cracks are more likely to occur in the weld metal. However, the conventional flux cored wire has a problem that it absorbs moisture during storage, leading to an increase in hydrogen content in the weld metal. This is because a flux included in the flux cored wire absorbs moisture in air.

[0005]

The flux cored wires mentioned in the above JP 61-202795 A and JP 7-164184 A fail to consider hygroscopic properties. Furthermore, even other related-art techniques make it difficult to improve the hygroscopic properties of the flux itself.

SUMMARY OF THE INVENTION

[0006]

Therefore, it is a main object of the present disclosure to provide a flux cored wire for gas shielded arc welding that has excellent resistance to hygroscopicity and adequate welding workability and can produce weld metal with excellent mechanical properties.

[0007]

The flux cored wire for gas shielded arc welding according

to the present invention is a flux cored wire for gas shielded arc welding, the flux being filled into a steel outer sheath, wherein the flux contains acid-soluble Mg, a content of the acid-soluble Mg in a first flux component having a particle diameter of 75 μm or less in the flux is set at 0.1 to 5% by mass relative to the total mass of the first flux component, a content of the acid-soluble Mg in a second flux component having a particle diameter of more than 75 μm and 106 μm or less in the flux is set at 0.1 to 7% by mass relative to the total mass of the second flux component, and a content of the acid-soluble Mg in a third flux component having a particle diameter of more than 106 μm in the flux is set at 1 to 15% by mass relative to the total mass of the third flux component.

In the flux cored wire for gas shielded arc welding in the present invention, a flux filling ratio is, for example, in a range of 10 to 25% by mass relative to the total mass of the wire.

The flux cored wire for gas shielded arc welding may contain the acid-soluble Mg in a content of 0.05 to 1.0% by mass relative to the total mass of the wire.

The flux cored wire for gas shielded arc welding may include, relative to the total mass of the wire: 1 to 10% by mass of at least one of Ti and Ti compound(s) in total in terms of Ti; 0.5 to 2% by mass of at least one of Si and Si compound(s) in total in terms of Si; 1.0 to 3.0% by mass of Mn; and 0.02

to 0.15% by mass of C, wherein a P content is restricted to 0.030% by mass or less, and a S content is restricted to 0.030% by mass or less.

The flux cored wire for gas shielded arc welding can also include, relative to the total mass of the wire: 0.05 to 0.5% by mass of at least one of Na compound(s), K compound(s) and Li compound(s) in total in terms of Na, K and Li, respectively; and 0.05 to 0.3% by mass of F compound(s) in terms of F.

The flux cored wire for gas shielded arc welding may include, relative to the total mass of the wire: 0.5% by mass or less of at least one of Zr and Zr compound(s) in total in terms of Zr. The flux cored wire for gas shielded arc welding may include, relative to the total mass of the wire: 0.05 to 1.0% by mass of at least one of Mg and Mg compound(s) in total in terms of Mg.

The flux cored wire for gas shielded arc welding can also include, relative to the total mass of the wire: 0.005 to 0.050% by mass of at least one of Bi and Bi compound(s) in total in terms of Bi.

Further, the flux cored wire for gas shielded arc welding in the present invention can also include, relative to the total mass of the wire, at least one selected from the group consisting of: Cu: 0.5% by mass or less; Ni: 3.0% by mass or less; Cr: 0.5% by mass or less; Mo: 0.5% by mass or less; Nb: 0.2% by mass or less; V: 0.2% by mass or less; at least one of B and B compound(s);

0.0001 to 0.0100% by mass in total in terms of B; and rare earth metal(s) (REM(s)): 0.01 to 0.5% by mass.

On the other hand, in the flux cored wire for gas shielded arc welding in the present invention, for example, a Fe content may be 77% by mass or more, relative to the total mass of the wire.

[0008]

Accordingly, the present invention can improve the hygroscopic properties of the flux, and hence can achieve the flux cored wire for gas shielded arc welding that has excellent resistance to hygroscopicity and adequate welding workability and can produce weld metal with excellent mechanical properties.

BRIEF DESCRIPTION OF THE DRAWINGS

[0009]

Fig. 1 is schematic diagram for explaining some steps in manufacturing methods for flux cored wires in Examples and Comparative Examples.

DETAILED DESCRIPTION OF THE EMBODIMENTS

[0010]

Embodiments of the present disclosure will be described in detail below. Note that the present disclosure is not intended to be limited by the following respective embodiments.

[0011]

The flux cored wire of one embodiment is formed by filling a flux into a steel outer sheath and has an outer diameter, for example, of 1.0 to 2.0 mm. Material for the outer sheath is not specifically limited as long as it is steel, but is desirably soft steel or low-alloy steel that has excellent wire-drawing workability. Further, a flux filling ratio can be set to any value as long as the contents of the respective components in the wire satisfy the ranges defined by the present invention. However, the flux filling ratio is preferably set at 10 to 25% by mass relative to the total mass of the wire in view of the drawability of the wire and workability during welding (including feeding).

[0012]

[Acid-soluble Mg content in the flux]

In the flux cored wire of this embodiment, acid-soluble Mg is blended into the flux. Here, the term "acid-soluble Mg" as used herein means any one of Mg, Mg alloy and Mg compound(s) that can be dissolved in aqua regia (a liquid mixture of concentrated hydrochloric acid and concentrated nitric acid in a ratio of 3:1 by volume). Note that any Mg that can be dissolved in aqua regia is regarded as the acid-soluble Mg, though it has low solubility, or it is a mixture or composite of an acid-soluble Mg and an acid-insoluble Mg.

[0013]

In general, Mg is added to the flux cored wire in the form of metal or an alloy, or a compound, such as an oxide and/or an intermetallic compound. Specifically, examples of a Mg source added to the flux include metal Mg, Al-Mg, Fe-Si-Mg, Ni-Mg, MgO, MgCO₃, MgSiO₃ and MgF₂. Among them, examples of the "acid-soluble Mg" include metal Mg, Al-Mg, Fe-Si-Mg, Ni-Mg and MgO. As an example of the case of remaining without being dissolved, the acid-soluble Mg such as MgO and Mg is coated with acid-insoluble compound(s).

[0014]

In the flux cored wire of this embodiment, the acid-soluble Mg content in the flux component having each particle size is set within a specific range. Specifically, the acid-soluble Mg content in a first flux component(s) having a particle diameter of 75 μm or less in the flux is set at 0.1 to 5% by mass relative to the total mass of the first flux component; the acid-soluble Mg content in a second flux component(s) having a particle diameter of more than 75 μm and 106 μm or less in the flux is set at 0.1 to 7% by mass relative to the total mass of the second flux component(s); and the acid-soluble Mg content in a third flux component(s) having a particle diameter of more than 106 μm in the flux is set at 1 to 15% by mass relative to the total mass of the third flux component(s). Thus, the flux cored wire can improve its resistance to hygroscopicity and reduce the spatter generation

amount. To improve the resistance to hygroscopicity, the smaller absorption area is more advantageous, and hence the particle size of Mg in the flux is preferably set larger. On the other hand, to improve the arc stability and suppress the spatter generation amount, the particle size of Mg in the flux is preferably set small and dispersed in the flux. The reason for this is that Mg having its surface oxidized or Mg oxide floats at the surface of a weld metal to become a cathode spot that draws arc. However, when the Mg particle size is large and Mg is unevenly distributed, arc might be deflected to become unstable in some cases. As with the present invention, the acid-soluble Mg contents in the flux components having the respective particle sizes are set within the above-mentioned specific ranges, which can improve the resistance to hygroscopicity as well as arc stability to thereby reduce the spatter generation amount. That is, the present invention can simultaneously exhibit contradictory effects, specifically, an effect of improving the resistance to hygroscopicity and an effect of improving the arc stability to suppress the spatter generation amount. The reason why Mg can be dispersed in the flux is supposed to be that by appropriately setting the particle size distribution of Mg in the flux, Mg in the flux is well combined with other flux particles, thereby preventing Mg in the flux from unevenly distributed, and dispersing Mg in the flux.

[0015]

Here, the acid-soluble Mg content means a Mg content relative to the total mass of the flux component(s) having each particle size. The flux is sieved to be separated into a first flux component having a particle diameter of 75 μm or less in the flux, a second flux component having a particle diameter of more than 75 μm and 106 μm or less in the flux, and a third flux component having a particle diameter of more than 106 μm in the flux. The flux components having the respective particle sizes are mixed in the aqua regia. Then, the contents of the flux components dissolved in the aqua regia were analyzed by an inductively coupled plasma (ICP) method, whereby the acid-soluble Mg contents are determined.

[0016]

When the acid-soluble Mg content in the first flux component having a particle diameter of 75 μm or less in the flux is less than 0.1% by mass relative to the total mass of the first flux component, the spatter generation amount is increased, whereas when the acid-soluble Mg content in the first flux component is more than 5% by mass, the resistance to hygroscopicity is degraded. When the acid-soluble Mg content in the second flux component having a particle diameter of more than 75 μm and of 106 μm or less in the flux is less than 0.1% by mass relative to the total mass of the second flux component, the spatter generation amount is increased, whereas when the

acid-soluble Mg content in the second flux component is more than 7% by mass, the slag covering property is degraded. When the acid-soluble Mg content in the third flux component having a particle diameter of more than 106 μm in the flux is less than 1% by mass relative to the total mass of the third flux component, the toughness of the weld metal is degraded, whereas when the acid-soluble Mg content in the third flux component is more than 15% by mass, the slag covering property is degraded.

[0017]

Mg contained in the flux cored wire extends in the wire-drawing direction or is crushed into fine grains by wire-drawing. The inventors have intensively studied focusing on Mg and found that when the minor diameter of Mg after the wire-drawing is defined as a particle diameter of Mg, there is a correlation between a Mg raw material and the particle size of Mg after the wire-drawing.

[0018]

However, in the conventional flux cored wire, the Mg raw material used as the flux component does not have the appropriate particle-size distribution and thus cannot reduce the spatter generation amount, failing to improve the resistance to hygroscopicity. For this reason, the inventors have studied various Mg raw materials and found that the use of the acid-soluble Mg having a specific particle-size distribution can produce the flux that has an excellent effect

of improving the arc stability to thereby suppress the spatter generation amount as well as an excellent effect of improving the resistance to hygroscopicity.

[0019]

For example, a suitable raw material for the acid-soluble Mg contains 25 to 40% by mass of a Mg raw material having a particle diameter of 250 μm or more and less than 300 μm , 20 to 35% by mass of a Mg raw material having a particle diameter of 212 μm or more and less than 250 μm , and 15 to 30% by mass of a Mg raw material having a particle diameter of 180 μm or more and less than 212 μm . As shown from part(c) to part(d) of Fig. 1, a flux 2 containing the acid-soluble Mg is filled into a steel outer sheath 1, and then the steel outer sheath 1 is formed in such a manner as to hold the flux 2 inside the steel outer sheath 1. Then, a formed wire is drawn as illustrated from part(d) to part(e) of Fig. 1. Here, the formed wire having a wire diameter of 5.0 mm is drawn to a wire diameter of 1.2 mm, so that the particle-size distribution of the acid-soluble Mg in the flux included in the drawn flux cored wire can be set within the above-mentioned specific range.

[0020]

Note that the raw material for the acid-soluble Mg used in the flux cored wire of this embodiment can be manufactured by normal methods.

[0021]

Further, methods for causing the Mg raw material to have its particle-size distribution in the specific range are not specifically limited, but can include a method of specifying, for example, a wire diameter after wire-drawing, a flux filling ratio, thickness and width of an outer sheath, a bulk specific gravity of a flux, raw materials other than a Mg raw material used in a flux and the number of times of wire-drawing. The conditions thereof and state of the acid-soluble Mg raw material in the flux of the flux cored wire obtained by wire-drawing are shown in Table 1 below.

[0022]

[Table 1]

Item	Mg raw material after wire-drawing: rather fine	Mg raw material after wire-drawing: rather coarse
Wire diameter after wire-drawing	small	large
Flux filling ratio	high	low
Thickness of outer sheath	thin	thick
Width	narrow	wide
Bulk specific gravity of flux	small	large
Amount of raw materials other than Mg raw material in flux	large	small
Number of times of wire-drawing	rather large	rather small

[0023]

The composition of the flux cored wire in this embodiment is not specifically limited, but can contain, for example, the following components. Note that the contents of respective

components mentioned below are those relative to the total mass of the wire unless otherwise specified.

[0024]

[Mg and Mg compound(s): 0.05 to 1.0% by mass, Acid-soluble Mg: 0.05 to 1.0% by mass]

The flux cored wire in this embodiment may contain 0.05 to 1.0% by mass of at least one of Mg and Mg compound(s) in total in terms of Mg, thereby further improving the toughness of the weld metal. On the other hand, when the total content of the Mg and Mg compound(s) is less than 0.05% by mass, the toughness of the weld metal might be degraded, whereas when the total content thereof is more than 1.0% by mass, the slag covering property may be degraded.

[0025]

The content of the acid-soluble Mg of the Mg and Mg compound(s) is, for example, in a range of 0.05 to 1.0% by mass. In this way, the toughness of the weld metal and the resistance to hygroscopicity of the flux cored wire can be further improved. On the other hand, when the acid-soluble Mg content is less than 0.05% by mass, the toughness of the weld metal might be degraded. When the acid-soluble Mg content is more than 1.0% by mass, the spatter generation amount might be increased. Therefore, the acid-soluble Mg content is preferably set at 0.2 to 0.6% by mass in terms of ensuring the toughness of the weld metal and improving the resistance to hygroscopicity of the flux cored

wire.

[0026]

[Al and Al compound(s): 0.01 to 1.0% by mass]

The flux cored wire in this embodiment may contain 0.01 to 1.0% by mass of at least one of Al and Al compound(s) in total in terms of Al. In this way, the toughness of the weld metal can be further improved. On the other hand, when the total content of the Al and Al compound(s) is less than 0.01% by mass, the spatter generation amount is increased, whereas when the total content thereof is more than 1.0% by mass, the toughness of the weld metal is reduced.

[0027]

[Ti and Ti compound(s): 1 to 10% by mass]

The flux cored wire in this embodiment may contain 1 to 10% by mass of at least one of Ti and Ti compound(s) in total in terms of Ti. The total content of the Ti and Ti compound(s) is set within this range, thereby further improving the toughness of the weld metal. On the other hand, when the total amount of the Ti and Ti compound(s) is less than 1% by mass, the slag covering property might be degraded, whereas when the total amount thereof is more than 10% by mass, the weld metal tends to have excessive strength.

[0028]

Therefore, in terms of improving the slag covering property, the total content of the Ti and Ti compound(s) is

preferably 2% by mass or more, and more preferably 3% by mass or more. Further, in terms of setting the strength of weld metal in a better range, the total content thereof is preferably 8% by mass or less.

[0029]

[Si and Si compound(s): 0.5 to 2% by mass]

The flux cored wire in this embodiment may contain 0.5 to 2% by mass of at least one of Si and Si compound(s) in total in terms of Si. The total content of the Si and Si compound(s) is set within this range, thereby enhancing the viscosity of the weld metal, which can improve the wettability to a base metal and the bead shape. On the other hand, when the total content of the Si and Si compound(s) is less than 0.5% by mass, the spatter generation amount might be increased, whereas when the total content thereof is more than 2% by mass, the toughness of the weld metal might be degraded.

[0030]

[Mn: 1.0 to 3.0% by mass]

When the Mn content is less than 1.0% by mass, the toughness of the weld metal might be degraded. When the Mn content is more than 3.0% by mass, the weld metal tends to have excessive strength. Therefore, the Mn content is preferably set at 1.0 to 3.0% by mass. In this way, the toughness of the weld metal can be further improved.

[0031]

[C: 0.02 to 0.15% by mass]

When the C content is less than 0.02% by mass, the toughness of the weld metal might be degraded. When the C content is more than 0.15% by mass, hot cracking tends to occur in the weld metal. Therefore, the C content is preferably set at 0.02 to 0.15% by mass. Thus, the resistance to hot cracking of the weld metal can be improved, and the toughness of the weld metal can be further improved.

[0032]

[P: 0.030% by mass or less (including 0% by mass)]

When the P content is more than 0.030% by mass, the toughness of the weld metal might be degraded. Therefore, the P content is preferably set at 0.030% by mass or less. In this way, the toughness of the weld metal can be further improved.

[0033]

[S: 0.030% by mass or less (including 0% by mass)]

When the S content is more than 0.030% by mass, the toughness of the weld metal might be degraded. Therefore, the S content is preferably set at 0.030% by mass or less. In this way, the toughness of the weld metal can be further improved.

[0034]

[Na compound(s), K compound(s) and Li compound(s): 0.05 to 0.5% by mass]

When the total content of the Na compound(s), K compound(s) and Li compound(s) is less than 0.05% by mass, the

spatter generation amount might be increased, whereas when the total content thereof is more than 0.5% by mass, the resistance to hygroscopicity might be reduced. Thus, when adding one, or two or more of the Na compound(s), the K compound(s) and the Li compound(s), the flux cored wire in this embodiment preferably contains 0.05 to 0.5% by mass in total of the added compound(s). In this way, the spatter generation amount can be reduced, and the resistance to hygroscopicity of the flux cored wire can be further improved. Here, the contents of the Na compound(s), K compound(s) and Li compound(s) are values in terms of Na, K and Li, respectively.

[0035]

[F compound(s): 0.05 to 0.3% by mass]

When the total content of the F compound(s) is less than 0.05% by mass, the spatter generation amount might be increased, whereas when the total content thereof is more than 0.3% by mass, the resistance to hygroscopicity might be reduced. Therefore, the total content of the F compound(s) is preferably set at 0.05 to 0.3% by mass. In this way, the spatter generation amount can be reduced, and the resistance to hygroscopicity of the flux cored wire can be further improved.

[0036]

[Zr and Zr compound(s): 0.5% by mass or less (excluding 0% by mass)]

The flux cored wire in this embodiment may contain 0.5%

by mass or less of at least one of Zr and Zr compound(s) in total in terms of Zr. In this way, the spatter generation amount can be reduced. When the total content of the Zr and Zr compound(s) is more than 0.5% by mass, the spatter generation amount might be increased.

[0037]

[Bi and Bi compound(s): 0.005 to 0.050% by mass]

The flux cored wire in this embodiment may contain 0.005 to 0.050% by mass of at least one of Bi and Bi compound(s) in total in terms of Bi. The total content of the Bi and Bi compound(s) is set within this range, thereby enabling improvement of the slag removability.

[0038]

[At least one of B and B compound(s): 0.0001 to 0.0100% by mass in total in terms of B, Cu: 0.5% by mass or less (excluding 0% by mass), Ni: 3.0% by mass or less (excluding 0% by mass), Cr: 0.5% by mass or less (excluding 0% by mass), Mo: 0.5% by mass or less (excluding 0% by mass), Nb: 0.2% by mass or less (excluding 0% by mass), V: 0.2% by mass or less (excluding 0% by mass) and rare earth metal(s) (REM(s)): 0.01 to 0.5% by mass]

At least one of B and B compound(s), Cu, Ni, Cr, Mo, Nb, V and REM(s) have the effect of improving the mechanical properties (strength or toughness) of the weld metal. Thus, at least one kind of element selected from the group consisting of at least one of B and B compound(s), Cu, Ni, Cr, Mo, Nb, V

and REM(s) can be added as needed.

[0039]

Note that when the contents of these elements become large, the strength of the weld metal is excessively increased, easily causing cracks. For this reason, when adding these elements, the flux cored wire in this embodiment contains at least one of B and B compound(s): 0.0001 to 0.0100% by mass in total in terms of B, Cu: 0.5% by mass or less (excluding 0% by mass), Ni: 3.0% by mass or less (excluding 0% by mass), Cr: 0.5% by mass or less (excluding 0% by mass), Mo: 0.5% by mass or less (excluding 0% by mass), Nb: 0.2% by mass or less (excluding 0% by mass), V: 0.2% by mass or less (excluding 0% by mass) and rare earth metal(s) (REM(s)): 0.01 to 0.5% by mass. Thus, the weld metal with the improved mechanical properties can be obtained.

[0040]

[Fe: 77% by mass or more]

The flux cored wire in this embodiment preferably contains 77% by mass or more of Fe relative to the total mass of the wire to ensure the weld quantity.

[0041]

[Balance]

The balance of the composition of the flux cored wire in this embodiment includes alloying agents, such as Ca and Li, and compound(s) thereof, and inevitable impurities, such as Sb

and As. When the above-mentioned respective elements are added in the form of oxide and/or nitride, the balance of the flux cored wire in this embodiment contains O and/or N.

[0042]

[Manufacturing Method]

A manufacturing method for the flux cored wire in this embodiment is not specifically limited. However, the flux cored wire can be manufactured, for example, by the following method. First, a steel strip for configuring the outer sheath is formed by a forming roll while being fed in the longitudinal direction, into a U-shaped open pipe. Then, an oxide, metal or alloy, Fe powder, and the like are blended in predetermined quantities such that the flux has a predetermined chemical composition. Then, the prepared flux was filled into the outer sheath, the outer sheath with the flux is processed to have a circular cross-section. Thereafter, the outer sheath with the flux is drawn by a cold drawing process into a wire having a wire diameter, for example, of 1.0 to 2.0 mm. Note that the wire hardened during the cold working may be annealed to be softened.

[0043]

As mentioned in detail above, the flux cored wire of this embodiment specifies the acid-soluble Mg content in each of the flux components classified based on the particle size, and thus can have the adequate welding workability, thereby producing

the weld metal with excellent mechanical properties and resistance to hot cracking, further making the hygroscopic properties of the flux better to improve the resistance to hygroscopicity thereof. The contents of the acid-soluble Mg and any added components are set in the above-mentioned preferable ranges, further enabling improvement of the above-mentioned effects.

[0044]

As mentioned above, since the state of a Mg raw material after wire-drawing significantly exerts an influence on welding workability, it is impossible to obtain the flux cored wire which is excellent in all of hygroscopicity and mechanical properties of a weld metal by just specifying a particle size of the Mg raw material before wire-drawing, like technology mentioned in see JP 61-202795 A.

Examples

[0045]

The effects of the present invention will be specifically described below by comparison between Examples of the present invention and Comparative Examples.

In Examples, a flux was filled into a steel outer sheath having a diameter of 1.0 to 2.0 mm. In this way, flux cored wires in Examples and Comparative Examples were fabricated. Specifically, the steel outer sheath 1 having an outer sheath

composition set in a range shown in Table 2 below (% by mass) was formed as shown from part(a) to part(b) of Fig. 1, and subsequently the flux 2 containing the acid-soluble Mg was filled into the inside of the steel outer sheath 1 as shown from part(b) to part(c) of Fig. 1. Each item shown in Table 2 mentioned above was adjusted according to the filling rate of the flux 2. Then, as shown from part(c) to part(d) of Fig. 1, the steel outer sheath 1 is formed in such a manner as to hold the flux 2 within the steel outer sheath 1, followed by drawing as shown from part(d) to part(e) of Fig.1. In this way, the flux cored wires 10 in Examples of the present invention and Comparative Examples were fabricated. The particle-size distribution of the Mg raw material for the acid-soluble Mg included in the flux 2 was adjusted using a pulverization treatment.

[0046]

The following Table 2 shows the ranges (% by mass) of the outer sheath components. Note that the range (% by mass) of each outer sheath component indicates the ratio of the mass of each component to the entire mass of the outer sheath components.

[0047]

The following Tables 3 to 6 show the acid-soluble Mg content in the flux component with each particle size, the wire composition and the flux filling ratio in the flux cored wire

in each of Examples and Comparative Examples. The acid-soluble Mg content (% by mass) indicates the ratio of the acid-soluble Mg content to the total mass of the flux component having each of particle sizes (75 μm or less, more than 75 μm and 106 μm or less, and more than 106 μm). The wire composition (% by mass) indicates the ratio of the mass of each component of the wire to the total mass of the wire. The flux filling ratio (% by mass) indicates the ratio of the mass of the flux to the total mass of the wire. Note that the balance of the wire components shown in Tables 4 and 6 below corresponds to an oxygen (O) content and a nitrogen (N) content of an oxide and a nitride of each element, and contents of inevitable impurities.

[0048]

[Table 2]

	Ranges of outer sheath components(% by mass)										
	C	Si	Mn	P	S	Cu	Ni	Cr	Ti	Al	B
Outer sheath	≤ 0.10	≤ 0.10	0.10-0.90	≤ 0.03	≤ 0.03	≤ 0.20			≤ 0.05	≤ 0.05	≤ 0.005

*Balance being Fe and inevitably impurities.

[0049]

[Table 3]

	No.	Acid-soluble Mg content (% by mass)		
		75 μm or less	more than 75 μm and 106 μm or less	more than 106 μm
Examples	1	4.5	5.7	2.3
	2	2.1	5.4	7.4
	3	1.8	0.9	1.4
	4	1.0	5.2	13.5
	5	4.9	5.1	3.3
	6	1.3	2.3	7.8
	7	2.5	3.6	2.8
	8	4.5	4.2	9.8
	9	1.8	5.7	8.1
	10	4.1	7.0	3.6
	11	3.9	1.9	14.9
	12	3.0	0.2	5.5
	13	2.0	6.6	15.0
	14	2.9	3.1	1.3
	15	0.6	4.8	7.1
	16	2.2	5.0	10.0
	17	2.3	1.5	12.1
	18	4.4	1.8	11.7
	19	0.5	1.5	4.5
	20	1.3	1.0	1.2
	21	1.5	1.1	2.2
	22	0.1	2.5	7.8
	23	4.2	6.0	9.7
	24	0.9	1.2	9.5
	25	1.8	3.0	2.8
	26	1.2	3.3	5.4

[0050]

[Table 4]

	No.	Wire components (% by mass)											
		Acid-soluble Mg	Acid-insoluble Mg	Total. Mg	Acid-soluble Al	Total. Al	Total. Ti	Total. Si	Mn	C	P	S	Na+K+Li
Examples	1	0.2	0	0.20	0.4	0.6	5.9	0.9	2.6	0.07	0.03	0.03	0.20
	2	0.8	0.06	0.86	0.5	0.1	6.0	1.0	2.7	0.07	0.03	0.03	0.20
	3	0.00	0	0.04	0.4	0.7	6.7	1.4	2.3	0.07	0.014	0.027	0.08
	4	1.0	0.10	1.10	0.4	0.7	5.5	0.6	2.2	0.04	0.025	0.001	0.13
	5	0.8	0.06	0.86	0	0.8	4.3	0.8	3.0	0.10	0.003	0.026	0.47
	6	0.3	0	0.30	2.6	3.0	6.5	1.0	3.0	0.07	0.018	0.028	0.47
	7	0.1	0.09	0.19	0	0	2.5	0.5	1.5	0.06	0.020	0.010	0.50
	8	0.7	0.06	0.76	0.7	1.2	6.5	2.0	1.2	0.02	0.010	0.006	0.10
	9	0.7	0.06	0.76	0.1	0.8	0	1.0	2.5	0.09	0.003	0.022	0.26
	10	0.2	0.02	0.22	0.2	0.5	2.9	1.1	2.0	0.11	0.050	0.006	0.33
	11	0.8	0.03	0.83	0.3	0.3	10.5	1.9	2.2	0.03	0.012	0.010	0.42
	12	0.2	0.01	0.21	0.2	0.8	4.9	0	2.8	0.06	0.009	0.010	0.42
	13	0.8	0.01	0.81	0.5	0.9	3.2	0.4	1.0	0.06	0.010	0.003	0.42
	14	0.8	0.09	0.89	0.3	0.4	4.7	2.2	2.2	0.08	0.010	0.021	0.19
	15	0.6	0.06	0.66	0.2	0.3	5.1	1.1	0	0.07	0.001	0.028	0.42
	16	0.4	0.03	0.43	0.5	0.8	3.0	1.4	0.9	0.11	0.005	0.030	0.05
	17	0.6	0.06	0.66	0.2	0.8	4.9	0.8	3.1	0.10	0.004	0.008	0.26
	18	0.5	0.09	0.59	0.2	0.4	7.2	0.9	1.7	0	0.026	0.002	0.21
	19	0.7	0.01	0.71	0.2	0.6	5.7	1.5	2.4	0.17	0.005	0.028	0.39
	20	0.1	0.02	0.12	0.4	0.8	7.8	1.3	1.8	0.10	0.050	0.023	0.40
	21	0.5	0.06	0.56	0.5	0.7	7.4	0.5	1.1	0.08	0.024	0.040	0.47
	22	0.6	0.01	0.61	0.4	0.5	7.2	1.8	1.6	0.14	0.028	0.022	0
	23	0.3	0.01	0.31	0.1	0.8	6.0	1.2	2.9	0.04	0.011	0.021	0.60
	24	0.3	0.09	0.39	0.2	0.7	7.5	2.0	1.1	0.09	0.022	0.011	0.09
	25	0.4	0.03	0.43	0.3	0.7	4.8	1.2	2.1	0.05	0.024	0.013	0.34
	26	0.2	0.04	0.24	0.4	0.7	5.5	0.9	2.8	0.14	0.019	0.006	0.36

[Table 4] (continued)

	No.	Wire components (% by mass)													Flux filling ratio (% by mass)
		Total. F	Total. Zr	Total. Bi	Cu	Ni	Cr	Mo	Nb	V	REM	Total. B	Fe	Balance	
Examples	1	0.30	0	0	0	0	0	0	0	0	0	0	86.0	3.4	23
	2	0.31	0	0	0	0	0	0	0	0	0	0	88.3	1.2	25
	3	0.06	0.3	0.035	0	0	0	0	0	0	0	0.0090	87.5	0.8	15
	4	0.23	0.4	0.025	0	0	0	0	0	0	0	0.0050	89.0	1	16
	5	0.26	0	0.037	0	0	0	0	0	0	0	0.0065	86.5	3.6	22
	6	0.19	0.4	0.042	0	0	0	0	0	0	0	0.0019	83.4	1.9	11
	7	0.30	0.1	0.010	0	0	0	0	0	0	0	0.0020	90.0	4.4	17
	8	0.05	0.2	0.041	0	0	0	0	0	0	0	0.0085	85.1	3.5	22
	9	0.14	0.3	0.009	0	0	0	0	0	0	0	0.0040	92.1	2.7	14
	10	0.22	0.1	0.009	0	0	0	0	0	0	0	0.0040	88.9	3.8	21
	11	0.14	0	0.013	0	0	0	0	0	0	0	0.0010	83.5	0.9	20
	12	0.25	0.1	0.031	0	0	0	0	0	0	0	0.0065	88.7	1.9	20
	13	0.21	0.3	0.009	0	0	0	0	0	0	0	0.0067	93.5	0	12
	14	0.11	0.4	0.044	0	0	0	0	0	0	0	0.0016	89.4	0.2	22
	15	0.19	0.3	0.031	0	0	0	0	0	0	0	0.0090	90.6	1.8	19
	16	0.05	0.5	0.022	0	0	0	0	0	0	0	0.0085	89.4	3.7	22
	17	0.17	0.2	0.023	0	0	0	0	0	0	0	0.0099	88.4	1.2	12
	18	0.08	0.4	0.043	0	0	0	0	0	0	0	0.0012	85.9	3	22
	19	0.10	0.1	0.041	0	0	0	0	0	0	0	0.0062	86.8	2.2	16
	20	0.07	0.1	0.006	0	0	0	0	0	0	0	0.0084	84.6	2.9	18
	21	0.09	0.5	0.048	0	0	0	0	0	0	0	0.0040	86.2	2.8	23
	22	0.29	0.4	0.039	0	0	0	0	0	0	0	0.0009	86.2	1.8	16
	23	0.21	0.3	0.006	0	0	0	0	0	0	0	0.0087	84.7	3.2	16
	24	0	0.3	0.018	0	0	0	0	0	0	0	0.0045	83.7	4.4	14
	25	0.35	0	0.036	0	0	0	0	0	0	0	0.0089	90.1	0.3	10
	26	0.16	0.6	0.034	0	0	0	0	0	0	0	0.0025	87.9	0.8	15

[0051]

[Table 5]

	No.	Acid-soluble Mg content (% by mass)		
		75 μm or less	more than 75 μm and 106 μm or less	more than 106 μm
Examples	27	3.9	6.5	2.1
	28	4.2	0.7	4.6
	29	2.1	1.1	11.4
	30	4.4	0.7	14.9
	31	3.8	0.8	5.7
	32	3.6	6.8	11.7
	33	3.0	0.4	1.3
	34	2.7	6.5	10.3
	35	3.9	2.9	5.6
	36	2.7	3.7	14.4
	37	2.9	4.0	7.9
	38	0.3	0.8	11.7
	39	2.1	5.4	7.4
	40	2.2	5.5	7.4
	41	2.1	5.4	7.2
Comparative Examples	42	< 0.01	5.7	7.6
	43	< 0.01	< 0.01	1.2
	44	4.9	< 0.01	1.6
	45	0.4	< 0.01	< 0.01
	46	3.4	6.9	< 0.01
	47	< 0.01	< 0.01	< 0.01
	48	5.4	5.1	2.1
	49	6.0	7.5	2.2
	50	4.0	7.1	1.2
	51	3.3	7.3	15.3
	52	0.6	5.9	17.0
	53	5.5	7.6	15.5

[0052]

[Table 6]

	No.	Wire components (% by mass)											
		Acid-soluble Mg	Acid-insoluble Mg	Total Mg	Acid-soluble Al	Total Al	Total Ti	Total Si	Mn	C	P	S	Na+K+Li
Examples	27	0.3	0.05	0.35	0.4	0.5	4.4	0.5	2.1	0.09	0.001	0.021	0.36
	28	0.6	0.02	0.62	0.5	0.8	4.8	1.0	1.3	0.03	0.014	0.010	0.12
	29	0.9	0.02	0.92	0.2	0.8	5.6	1.9	2.7	0.04	0.004	0.028	0.15
	30	0.9	0.04	0.94	0.4	0.6	5.1	1.8	2.0	0.09	0.009	0.015	0.14
	31	0.9	0.05	0.95	0.4	0.6	7.1	1.5	1.3	0.08	0.014	0.023	0.28
	32	0.6	0.07	0.67	0.3	0.4	7.5	1.6	1.7	0.08	0.016	0.009	0.07
	33	0.7	0.09	0.79	0.4	0.5	5.0	1.9	1.5	0.07	0.015	0.007	0.44
	34	0.7	0.08	0.78	0.2	0.9	5.9	1.8	1.2	0.03	0.013	0.002	0.50
	35	0.4	0.09	0.49	0.5	0.9	5.1	1.1	2.8	0.03	0.020	0.003	0.41
	36	0.7	0.01	0.71	0.3	0.8	5.4	1.0	1.3	0.07	0.003	0.025	0.29
	37	0.8	0.04	0.84	0.3	0.7	5.8	1.0	2.6	0.05	0.016	0.020	0.13
	38	0.8	0.05	0.85	0.2	0.5	6.0	1.3	2.9	0.14	0.004	0.029	0.10
	39	0.8	0.06	0.86	0.5	0.1	1.1	1.0	2.7	0.07	0.029	0.026	0.20
	40	0.8	0.06	0.87	0.5	0.1	0.9	1.0	2.7	0.07	0.029	0.026	0.20
	41	0.8	0.06	0.86	0.5	0.1	2.0	1.0	2.7	0.07	0.029	0.026	0.20
Comparative Examples	42	0.7	0.06	0.76	0.5	0.6	7.6	1.2	2.9	0.04	0.024	0.010	0.20
	43	0.8	0.03	0.83	0.2	0.7	6.2	1.2	1.8	0.08	0.009	0.011	0.34
	44	0.1	0.08	0.18	0.1	0.2	6.0	1.3	2.1	0.04	0.018	0.010	0.07
	45	0.4	0.08	0.48	0.2	0.5	7.7	0.7	1.0	0.12	0.017	0.003	0.24
	46	0.4	0.06	0.46	0.2	0.9	6.2	0.7	2.2	0.11	0.013	0.001	0.40
	47	0.4	0.05	0.46	0.3	0.5	6.1	1.6	2.9	0.06	0.013	0.025	0.06
	48	0.8	0.10	0.90	0.1	0.6	5.7	1.4	2.4	0.13	0.008	0.001	0.12
	49	0.3	0.03	0.33	0.2	0.8	5.9	1.1	1.8	0.09	0.000	0.015	0.32
	50	0.1	0.04	0.14	0.3	0.5	7.5	0.5	2.2	0.10	0.004	0.024	0.09
	51	0.9	0.08	0.98	0.1	0.2	7.2	1.2	2.9	0.09	0.018	0.024	0.06
	52	0.1	0.02	0.12	0.3	0.6	7.5	0.8	1.4	0.03	0.007	0.026	0.40
	53	0.9	0.05	0.95	0.2	0.3	4.5	0.7	1.7	0.09	0.009	0.004	0.07

[Table 6] (continued)

	No.														Flux filling ratio (% by mass)
		Total. F	Total. Zr	Total. Bi	Cu	Ni	Cr	Mo	Nb	V	REM	Total. B	Fe	Balance	
Examples	27	0.22	0.3	0	0.2	2.3	0.2	0.2	0.1	0.2	0.134	0.0082	87.6	0.5	20
	28	0.24	0	0.050	0.2	2.4	0	0	0	0.1	0.021	0.0027	84.7	4.2	11
	29	0.07	0.1	0.030	0.5	2.4	0	0	0.1	0	0.132	0.0018	81.1	4.3	12
	30	0.14	0.1	0.008	0.3	3.0	0.2	0.2	0.1	0	0.251	0.0059	82.6	3.3	25
	31	0.29	0.2	0.048	0.3	0.9	0.5	0.1	0.1	0	0.300	0.0094	82.9	3.4	21
	32	0.16	0.1	0.025	0	0	0.5	0.5	0	0.1	0.288	0.0031	83.6	3.3	24
	33	0.27	0.1	0.043	0.3	2.5	0.4	0	0.2	0.2	0.311	0.0019	82.8	3.4	20
	34	0.13	0.2	0.033	0	0.2	0.2	0.1	0.1	0.2	0.402	0.0046	83.2	4.8	12
	35	0.26	0.1	0.033	0.4	0.5	0.5	0.2	0.1	0	0	0.0007	82.8	4.7	19
	36	0.12	0.4	0.049	0	0.3	0.1	0	0.1	0.2	0.500	0.0031	86.2	3.1	23
	37	0.22	0.3	0.035	0.1	0.9	0.1	0	0.1	0.1	0.229	0	83.1	4.5	24
	38	0.16	0.2	0.015	0.2	2.9	0.4	0.2	0.1	0	0.424	0.0100	80.6	3.8	18
	39	0.31	0	0	0	0	0	0	0	0	0	0	93.2	1.2	25
	40	0.31	0	0	0	0	0	0	0	0	0	0	93.4	1.2	25
	41	0.31	0	0	0	0	0	0	0	0	0	0	92.2	1.3	25
Comparative Examples	42	0.16	0.4	0.027	0	0	0	0	0	0	0	0.0033	84.1	2.7	24
	43	0.09	0.2	0.027	0	0	0	0	0	0	0	0.0007	84.9	4.4	15
	44	0.18	0.1	0.021	0	0	0	0	0	0	0	0.0074	85.3	4.6	21
	45	0.23	0.5	0.041	0	0	0	0	0	0	0	0.0087	86.4	2.5	20
	46	0.22	0.3	0.049	0	0	0	0	0	0	0	0.0068	88.6	0.2	17
	47	0.29	0.4	0.024	0	0	0	0	0	0	0	0.0099	84.8	3.2	15
	48	0.11	0.4	0.012	0	0	0	0	0	0	0	0.0002	85.1	3.9	23
	49	0.20	0.1	0.043	0	0	0	0	0	0	0	0.0093	87.9	1.7	12
	50	0.20	0.2	0.020	0	0	0	0	0	0	0	0.0052	85.7	2.9	15
	51	0.22	0.3	0.035	0	0	0	0	0	0	0	0.0054	84.4	3.3	21
	52	0.14	0.4	0.032	0	0	0	0	0	0	0	0.0041	87.6	1.0	18
	53	0.17	0.4	0.025	0	0	0	0	0	0	0	0.0067	90.7	1.3	25

[0053]

The respective flux cored wires in Examples and Comparative Examples were evaluated in the following ways.

[0054]

<Acid-soluble Mg content>

A flux was sampled from each of the flux cored wires obtained after drawing in Examples and Comparative Examples. The flux was separated by employing an analyzer RSP-105 manufactured by Seishin Enterprise Co., Ltd., and using a sieve in conformance with JIS Z 8801-1:2006, into (1) a flux component having a particle diameter of 75 μm or less, (2) a flux component having a particle diameter of more than 75 μm and 106 μm or less, and (3) a flux component having a particle diameter of more than 106 μm . At this time, conditions for sieving were as follows: an acoustic wave frequency of 80 Hz; a pulse interval of 1 sec; and a classification time of 2 minutes. The thus-obtained flux component with each particle size was dissolved in a solution (aqua regia) that included a mixture of concentrated hydrochloric acid and concentrated nitric acid in a ratio of 3:1 by volume, and Mg dissolved in this solution was defined as the acid-soluble Mg.

[0055]

<Slag covering property>

The slag covering property was evaluated by a visual inspection method. Samples in which a slag covered the entire weld bead and could be naturally separated across its entire

length from the bead were rated as being excellent (A+); samples in which a slag covered the entire weld bead and could be naturally separated from the bead, though not across the entire length of the slag, were rated as being very good (A); samples in which a slag covered the entire weld bead, though not naturally separated from the bead, were rated as being good (B); and samples in which a slag did not cover the entire weld bead were rated as being bad (C).

[0056]

<Spatter generation amount>

The spatter generation amount was determined by measuring an amount of generated spatter per unit time by a whole collection method defined by WES 2087:2000. Welding conditions were set to the same as the weld-quantity measurement conditions. Samples having a spatter generation amount of 1.0 g/min or less were rated as being excellent (A); samples having a spatter generation amount of more than 1.0 g/min and 1.5 g/min or less were rated as being good (B); and samples having a spatter generation amount of more than 1.5 g/min were rated as being bad (C).

[0057]

<Mechanical properties of weld metal>

Mechanical properties of a weld metal in each sample were evaluated by a tension test and an impact test in conformance with "Methods of tension and impact tests for deposited metal"

defined by JIS Z 3111:2005. At this time, welding conditions were set to a welding current of 290 A to 320 A (DC-EP), and an interpass temperature of $150^{\circ}\text{C} \pm 10^{\circ}\text{C}$. In the evaluation of the low-temperature toughness, samples having an impact value of 100 J or more at atmospheric temperature of -40°C were rated as being excellent (A+); samples having an impact value of 70 J or more and less than 100 J were rated as very good (A); samples having an impact value of 47 J or more and less than 70 J were rated as good (B); and samples having an impact value of less than 47 J were rated bad (C). Samples having a 0.2% yield strength of 390 MPa or more were rated as Pass. Samples having a tensile strength of 490 MPa or more and 670 MPa or less were rated as Pass.

[0058]

<Resistance to hygroscopicity>

The resistance to hygroscopicity of each sample was evaluated by cutting the flux cored wire every 3 cm to produce three specimens, exposing these specimens under an atmosphere at a temperature of 30°C and a relative humidity of 80% for 24 hours, and then measuring the amount of moisture absorbed in the flux of the flux cored wire by the Karl Fischer's method. In this measurement, to gasify the moisture of the flux in the flux cored wire, the specimen was heated at 750°C , and Ar gas was guided as a carrier gas to a measurement device. As a result, samples having a moisture content in the flux of the flux cored

wire of 500 ppm or less were rated as being very good (A) ; samples having a moisture content of more than 500 ppm and 750 ppm or less were rated as being good (B) ; and samples having a moisture content of more than 750 ppm were rated as being bad (C) , relative to the total mass of the wire.

[0059]

<Resistance to hot cracking>

The resistance to hot cracking was evaluated based on "Method of FISCO test" defined by JIS Z 3155:1993. At this time, welding conditions were set to a welding current of 270 A, an arc voltage of 30 V, and a velocity of 350 mm/min. In the evaluation, samples having the cracking rate of less than 30% except for crater cracks were rated as Pass.

[0060]

The above-mentioned results are collectively shown in Tables 7 and 8 below.

[0061]

[Table 7]

	No.	Slag covering	Spatter generation amount	Resistance to hygroscopicity	Low-temperature toughness (vE-40)
Examples	1	A	A	A	A
	2	A+	A	A	A+
	3	A	A	A	B
	4	B	A	A	A+
	5	A+	B	A	A
	6	A+	B	A	B
	7	A	B	A	B
	8	A+	B	A	A+
	9	B	B	A	A+
	10	A	A	A	A+
	11	A	A	A	A+
	12	A+	B	A	A+
	13	B	B	A	A
	14	A	A	A	B
	15	A+	A	A	B
	16	A+	A	A	A
	17	A	A	A	A+
	18	A	A	A	B
	19	A+	A	A	A+
	20	A	A	A	B
	21	A	A	A	B
	22	A+	B	A	A+
	23	A+	A	B	A+
	24	A+	B	A	A+
	25	A	A	B	A
	26	A+	B	A	A+

[0062]

[Table 8]

	No.	Slag covering	Spatter generation amount	Resistance to hygroscopicity	Low-temperature toughness (vE-40)
Examples	27	A	A	A	A
	28	A+	A	A	A+
	29	A	A	A	A
	30	A	A	A	A
	31	A+	A	A	A+
	32	A	A	A	A
	33	A	A	A	A
	34	A	A	A	A
	35	A+	A	A	A+
	36	A	A	A	A
	37	A+	A	A	A+
	38	A	A	A	A
	39	A	A	A	A+
	40	B	A	A	A+
	41	A	A	A	A+
Comparative Examples	42	A+	C	A	A+
	43	A	C	A	A
	44	A	C	A	A
	45	A	C	A	C
	46	A	A	A	C
	47	A	C	A	C
	48	A	A	C	A
	49	C	A	C	A
	50	C	A	A	A
	51	C	A	A	A
	52	C	A	A	A
	53	C	A	C	A

[0063]

As shown in Table 8 above, in the flux cored wires of samples Nos. 42 to 45 and 47, the acid-soluble Mg content(s) in each or both of the flux component with a particle diameter

of 75 μm or less and the flux component with a particle diameter of more than 75 μm and 106 μm or less did not satisfy the range(s) defined by the present invention. Thus, these wires of samples Nos. 42 to 45 and 47 increased the spatter generation amount. Further, in the flux cored wires of samples Nos. 45 and 47, the acid-soluble Mg content in the flux component with a particle diameter of more than 106 μm did not satisfy the range defined by the present invention, thus degrading the toughness of the weld metal. On the other hand, in the flux cored wire of sample No. 46, the acid-soluble Mg content in the flux component with the particle diameter of more than 106 μm did not satisfy the range defined by the present invention, resulting in excellent welding workability and resistance to hygroscopicity but degrading the toughness of the weld metal.

[0064]

In the flux cored wires of samples Nos. 48, 49 and 53, the acid-soluble Mg content in the flux component with a particle diameter of 75 μm or less exceeded the range defined by the present invention, thus degrading the resistance to hygroscopicity of the flux. In the flux cored wires of samples Nos. 49 to 53, the acid-soluble Mg content(s) in each or both of the flux component with a particle diameter of more than 75 μm and 106 μm or less, and the flux component with a particle diameter of more than 106 μm exceeded the range(s) defined by the present invention, thus degrading the slag covering

property.

[0065]

On the other hand, as shown in Tables 7 and 8, the flux cored wires of samples Nos. 1 to 41 as Examples had excellent welding workability (slag covering property, spatter generation amount), and toughness of the weld metal, and further excellent resistance to hygroscopicity of the wire, compared to the flux cored wires of samples Nos. 42 to 53 as Comparative Examples. Although not shown in Tables 7 and 8, all the flux cored wires of samples Nos. 1 to 41 as Examples were also rated "Pass" regarding the evaluation results of the 0.2% yield strength, tensile strength, and resistance to hot cracking of the weld metal. As can be seen from these results, it is confirmed that the present invention can achieve the flux cored wire for gas shielded arc welding that has excellent resistance to hygroscopicity and adequate welding workability and can produce weld metal with excellent mechanical properties.

[0066]

The embodiments according to the present disclosure can take the following arrangements.

[1] The flux cored wire for gas shielded arc welding, the flux being filled into a steel outer sheath, wherein the flux contains acid-soluble Mg, a content of the acid-soluble Mg in a first flux component having a particle diameter of 75 μm or less in the flux is set at 0.1 to 5% by mass relative to the

total mass of the first flux component, a content of the acid-soluble Mg in a second flux component having a particle diameter of more than 75 μm and 106 μm or less in the flux is set at 0.1 to 7% by mass relative to the total mass of the second flux component, and a content of the acid-soluble Mg in a third flux component having a particle diameter of more than 106 μm in the flux is set at 1 to 15% by mass relative to the total mass of the third flux component.

[2] The flux cored wire for gas shielded arc welding according to [1], a flux filling ratio is in a range of 10 to 25% by mass relative to the total mass of the wire.

[3] The flux cored wire for gas shielded arc welding according to [1] or [2], the acid-soluble Mg is contained in a content of 0.05 to 1.0% by mass relative to the total mass of the wire.

[4] The flux cored wire for gas shielded arc welding according to any one of [1] to [3], includes, relative to the total mass of the wire: 1 to 10% by mass of at least one of Ti and Ti compound(s) in total in terms of Ti; 0.5 to 2% by mass of at least one of Si and Si compound(s) in total in terms of Si; 1.0 to 3.0% by mass of Mn; and 0.02 to 0.15% by mass of C, wherein a P content is restricted to 0.030% by mass or less, and a S content is restricted to 0.030% by mass or less.

[5] The flux cored wire for gas shielded arc welding according to any one of [1] to [4], further includes, relative to the total mass of the wire: 0.05 to 0.5% by mass of at least one of Na

compound(s), K compound(s) and Li compound(s) in total in terms of Na, K and Li, respectively; and 0.05 to 0.3% by mass of F compound(s) in terms of F.

[6] The flux cored wire for gas shielded arc welding according to any one of [1] to [5] further includes, relative to the total mass of the wire: 0.5% by mass or less of at least one of Zr and Zr compound(s) in total in terms of Zr.

[7] The flux cored wire for gas shielded arc welding according to any one of [1] to [6], further includes, relative to the total mass of the wire: 0.05 to 1% by mass of at least one of Mg and Mg compound(s) in total in terms of Mg.

[8] The flux cored wire for gas shielded arc welding according to any one of [1] to [7], further comprising, relative to the total mass of the wire: 0.005 to 0.050% by mass of at least one of Bi and Bi compound(s) in total in terms of Bi. .

[9] The flux cored wire for gas shielded arc welding according to any one of [1] to [8] further includes, relative to the total mass of the wire, at least one selected from the group consisting of: Cu: 0.5% by mass or less; Ni: 3.0% by mass or less; Cr: 0.5% by mass or less; Mo: 0.5% by mass or less; Nb: 0.2% by mass or less; V: 0.2% by mass or less; at least one of B and B compound(s); 0.0001 to 0.0100% by mass in total in terms of B; and rare earth metal(s) (REM(s)): 0.01 to 0.5% by mass.

[10] In the flux cored wire for gas shielded arc welding according to any one of [1] to [9], a Fe content is 77% by mass

or more, relative to the total mass of the wire.

Description of Reference Numerals

[0067]

- 1 Steel outer sheath
- 2 Flux
- 10 Flux cored wire

What is claimed is:

1. A flux cored wire for gas shielded arc welding, the flux being filled into a steel outer sheath, wherein

the flux contains acid-soluble Mg,

a content of the acid-soluble Mg in a first flux component having a particle diameter of 75 μm or less in the flux is set at 0.1 to 5% by mass relative to the total mass of the first flux component,

a content of the acid-soluble Mg in a second flux component having a particle diameter of more than 75 μm and 106 μm or less in the flux is set at 0.1 to 7% by mass relative to the total mass of the second flux component, and

a content of the acid-soluble Mg in a third flux component having a particle diameter of more than 106 μm in the flux is set at 1 to 15% by mass relative to the total mass of the third flux component.

2. The flux cored wire for gas shielded arc welding according to claim 1, wherein a flux filling ratio is in a range of 10 to 25% by mass relative to the total mass of the wire.

3. The flux cored wire for gas shielded arc welding according to claim 1 or 2, wherein the acid-soluble Mg is contained in a content of 0.05 to 1.0% by mass relative to the total mass

of the wire.

4. The flux cored wire for gas shielded arc welding according to any one of claims 1 to 3, comprising, relative to the total mass of the wire:

1 to 10% by mass of at least one of Ti and Ti compound(s) in total in terms of Ti;

0.5 to 2% by mass of at least one of Si and Si compound(s) in total in terms of Si;

1.0 to 3.0% by mass of Mn; and

0.02 to 0.15% by mass of C, wherein

a P content is restricted to 0.030% by mass or less, and

a S content is restricted to 0.030% by mass or less.

5. The flux cored wire for gas shielded arc welding according to any one of claims 1 to 4, further comprising, relative to the total mass of the wire:

0.05 to 0.5% by mass of at least one of Na compound(s), K compound(s) and Li compound(s) in total in terms of Na, K and Li, respectively; and

0.05 to 0.3% by mass of F compound(s) in terms of F.

6. The flux cored wire for gas shielded arc welding according to any one of claims 1 to 5, further comprising, relative to the total mass of the wire:

0.5% by mass or less of at least one of Zr and Zr compound(s) in total in terms of Zr.

7. The flux cored wire for gas shielded arc welding according to any one of claims 1 to 6, further comprising, relative to the total mass of the wire:

0.05 to 1.0% by mass of at least one of Mg and Mg compound(s) in total in terms of Mg.

8. The flux cored wire for gas shielded arc welding according to any one of claims 1 to 7, further comprising, relative to the total mass of the wire:

0.005 to 0.050% by mass of at least one of Bi and Bi compound(s) in total in terms of Bi.

9. The flux cored wire for gas shielded arc welding according to any one of claims 1 to 8, further comprising, relative to the total mass of the wire, at least one selected from the group consisting of:

Cu: 0.5% by mass or less;

Ni: 3.0% by mass or less;

Cr: 0.5% by mass or less;

Mo: 0.5% by mass or less;

Nb: 0.2% by mass or less;

V: 0.2% by mass or less;

at least one of B and B compound(s): 0.0001 to 0.0100%
by mass in total in terms of B; and

rare earth metal(s) (REM(s)): 0.01 to 0.5% by mass.

10. The flux cored wire for gas shielded arc welding according
to any one of claims 1 to 9, wherein

a Fe content is 77% by mass or more, relative to the total
mass of the wire.