

EPOXY FUNCTIONAL POLYESTER RESINS,
PROCESS FOR THEIR PREPARATION, AND OUTDOOR DURABLE
COATING COMPOSITIONS COMPRISING THEM

The present invention relates to epoxy functional polyester resins, to a process for their preparation and to outdoor durable powder coating compositions comprising them which have improved storage stability and overall coating properties.

Epoxy functional polyester resins and outdoor durable coatings comprising them, are known from e.g. European patent applications Nos. 0634434A2, 0720997A2 and International applications Nos. WO 98/23661 and WO 98/24828.

European patent application No. 0634434A2 discloses a process for the preparation of linear tertiary aliphatic carboxyl functional polyester resins, by reacting:

(a) at least one compound A' comprising one mono-functional primary- or secondary hydroxyl group and/or at least one compound A'' comprising one primary- or secondary hydroxyl group and one tertiary aliphatic carboxyl group;

(b) at least one aromatic or cycloaliphatic dicarboxylic acid compound B comprising two aromatic- or secondary aliphatic carboxyl groups or the anhydride thereof;

(c) at least one diol compound C comprising two aliphatic hydroxyl groups, which may independently be a primary or a secondary hydroxyl group; and

(d) at least one dihydroxymonocarboxylic acid compound D comprising a tertiary aliphatic carboxyl group and two aliphatic hydroxyl groups, which may each independently be primary or secondary hydroxyl,

the molar ratio of compounds A":A":B:C:D being
M:N:X+Y+1:X:Y

wherein M+N=2, X ranges from 2 to 8 and Y ranges from
2-N to 8, at a temperature of from 100 to 240 °C, until
5 essentially all the non-tertiary carboxyl groups as
initially present in the reaction mixture have been
reacted.

Moreover in this application were disclosed poly-
glycidylester resins obtainable by reacting said linear
10 tertiary aliphatic carboxyl functional polyesters with an
excess epihalohydrin in the presence of a suitable base
and optional catalyst. Preferably, the polyesters were
reacted with epichlorohydrin. Both the specified linear
polyesters and the corresponding polyglycidylesters
15 derived therefrom were used with a cross-linking agent
for powder coating compositions.

In European patent application No. 0720997A2, linear
tertiary carboxyl functional polyesters and epoxy
functional polyester resins are disclosed where these
20 polyester resins were produced by reacting:

- a) at least one aromatic and/or cycloaliphatic
carboxylic acid compound A comprising two aromatic-
and/or secondary aliphatic carboxyl groups or the
anhydride thereof,
- 25 b) at least one hydroxyl compound B comprising two
aliphatic hydroxyl groups, which groups each
independently may be primary or secondary hydroxyl
groups,
- c) at least one hydroxyl substituted carboxylic acid
30 compound C comprising at least one tertiary aliphatic
carboxyl group and two aliphatic hydroxyl groups, which
groups each independently may be primary or secondary
hydroxyl groups, and
- d) optionally one carboxylic acid compound D comprising
35 one carboxyl group,

the molar ratio of compounds A:B:C:D being

$$(X+Y-1):X:Y:Z,$$

wherein X ranges from 2 to 8, Y ranges from 2 to 8, and Z ranges from 0 to 2.

5 These polyester resins could be used together with a suitable curing agent for the production of powder coatings, or could be converted into the corresponding glycidylesters, which in combination with a suitable curing agent could be used for the production of powder
10 coatings.

From WO 98/24828 are known linear, tertiary carboxyl functional polyester resins obtainable by reaction of (a) at least one 1,4-dicarboxylcyclohexane (A) optionally mixed with a minor weight fraction of an alkane
15 dicarboxylic acid, containing in the range of from 8 to 16 carbon atoms (A'), (b) at least one dihydroxymono-carboxylic acid compound (B), comprising a tertiary aliphatic carboxyl group and two aliphatic hydroxyl groups, (c) optionally one diol compound (C) comprising
20 two aliphatic hydroxyl groups, which may independently be a primary or a secondary hydroxyl group, and optionally (d) a compound (D') comprising one monofunctional primary or secondary hydroxyl group and/or a compound (D'')
25 comprising one primary or secondary hydroxyl group and one tertiary aliphatic carboxyl group, the molar ratio of the compounds (A+A'):B:C:D':D'' being $X+Y+1:Y:X:M:N$ wherein $M+N$ is the range of from 0 to 2, wherein X ranges from 2 to 8, and Y ranges from $[2-(M+N)]$ to 8, at a
30 temperature of from 100 to 225 °C, until essentially all the non-tertiary carboxyl groups as initially present in the reaction mixture have been reacted;
polyglycidylesters derived from said polyester resins; and coating compositions comprising at least one polyester resin and/or at least one polyglycidylester
35 resin.

From WO 98/23661 are known linear, tertiary carboxyl functional polyester resins obtainable by reaction of

a) at least one compound A₁, comprising the reaction product of (i) a glycidylester of a mixture of synthetic highly branched saturated monocarboxylic acids isomers of formula (R₁)(R₂)(R₃)C-COOH (I), wherein R₁, R₂ and R₃ are alkyl groups of from 1 to 15 carbon atoms, of which at least one is methyl, each acid containing from 5 to 19 and preferably from 5 to 13 carbon atoms and preferably CARDURA glycidylesters, and (ii) a mixture of said synthetic highly branched saturated monocarboxylic acids, in a molar ratio of 1:1 (CARDURA is a trade mark); said component A₁ being optionally mixed with hydroxy pivalic acid (A₂) and/or hydrogenated diphenylolpropane (A₃); b)

at least one aromatic or cycloaliphatic dicarboxylic acid compound B, comprising two aromatic- or secondary aliphatic carboxyl groups or the anhydride thereof; optionally c) at least one dihydroxymonocarboxylic acid compound C comprising a tertiary aliphatic carboxyl group and two aliphatic hydroxyl groups, which may each independently be primary or secondary hydroxyl; and d) optionally at least one diol compound D comprising two aliphatic hydroxyl groups which may each independently be a primary or a secondary hydroxyl group; the molar ratio of compounds A₁:A₂+A₃:B:C:D being A₁:(2-A₁):X+Y+1:X:Y, wherein A₁ ranges from 0.1 to 2, wherein Y ranges from 0 to 6 and X ranges from 2 to 8, at a temperature of from 100 to 225 °C, until essentially all the non-tertiary carboxyl groups as initially present in the reaction mixture have been reacted; polyglycidylesters derived from said polyester resins and coating compositions comprising at least one polyester resin and/or at least one polyglycidyl resin.

Although the linear tertiary aliphatic carboxyl functional polyester resins and the polyglycidylesters thereof enabled a certain progress as to the requirements of excellent outdoor durability (UV stability) and resistance against hydrolysis in the cured state, with reference to their use in modern economically applied powder coatings, there is still a need for further improvement of this combination of properties.

On the other hand novel powder coating binders for the exterior durable powder coating market derived from carboxylated polyester resins, cured with epoxy functional acrylate polymers, have been proposed during the Waterborne, Higher Solids and Powder Coatings Symposium, February 5-7, 1997, New Orleans LA, USA, T Agawa and E D Dumain, p. 342-353, "New Two-component Powder Coating Binders: Polyester acrylate hybrid as TGIC Cure Alternative.

However, as indicated on page 353, further improvements have to be made to provide smoother films, lower cure temperatures and UV durability to rival that of automotive topcoating or outdoor building panel topcoating.

The epoxy functional polyester resins obtainable according to the hereinbefore discussed documents, although showing attractive combinations coating properties, such as outdoor durability, flexibility, hardness, chemical resistance comprise minor but significant amounts (≤ 25 wt%) of side products, which bear terminal hydroxy and/or hydrolyzable halogen (chlorine) and which have appeared to be formed during the incomplete glycidation of the starting carboxyl functional polyester resins.

Due to this minor contents of byproducts in the finally obtained glycidyl esters of carboxyl functional polyester resins, the latter are less reactive as

expected, which may be expressed by a lower functionality. Moreover, said side products can cause several uncontrolled side reactions when incorporated in the intended outdoor durable powder coating compositions.

5 This caused less optimal coating properties of the finally obtained cured coatings and more in particular the outdoor durability, due to the formation of ether linkages in the final coating.

Moreover, the glass transition of these glycidyl esters is too low, resulting in a deteriorated storage stability.

It will be appreciated that there is a need to improve the outdoor durable coating properties further by elimination of the uncontrolled activity of the hydroxy and/or hydrolyzable halogen in the occurring byproducts.

15 Therefore it is an object of the present invention to provide powder coating compositions, showing improved properties such as storage stability, reactivity and final coating properties such as ODD flexibility, and comprising at least a glycidylester of carboxyl functional polyester resin.

Another object of the present invention is to provide the modified compositions glycidylesters derived from carboxyl functional polyester resins.

25 As a result of extensive research and experimentation said improved powder coating compositions aimed at and the modified compositions of epoxy functional polyesters to be used therein, have been found.

Accordingly the invention relates to modified epoxy functional polyesters (I), obtainable by the reaction of epihalohydrin and preferably epichlorohydrin, in the presence of a base, with carboxyl functional polyester resins (II), obtainable by reacting:

30 (a) at least one aromatic, aliphatic or cycloaliphatic dicarboxylic acid compound A comprising two aromatic- or

secondary aliphatic carboxyl groups or the anhydride thereof;

(b) at least one diol compound B comprising two aliphatic hydroxyl groups, which may independently be a primary or a secondary hydroxyl group; and optionally

(c) compound C1 comprising one monofunctional primary- or secondary hydroxyl group and/or at least one compound C2 comprising one primary- or secondary hydroxyl group and one tertiary aliphatic carboxyl group; and optionally

(d) a dihydroxymonocarboxylic acid compound D comprising a tertiary aliphatic carboxyl group and two aliphatic hydroxyl groups, which may each independently be primary or secondary hydroxyl; and optionally

(e) a trihydroxyalkane (E1) or tetrahydroxyalkane (E2), the molar ratio of compounds A:B:C1:C2:D:E1:E2 being $X+Y+1:X:M:N:Y:Z:Q$

wherein $M+N$ is in the range of from 0 to 2, X ranges from 2 to 8 and Y ranges from 0 to 8, Z ranges from 0 to 2 and Q ranges from 0 to 2 at a temperature of from 100 to 220 °C, until a predominant part of the non-tertiary carboxyl groups as initially present in the reaction mixture, i.e. 75% or more, have been converted, and subsequent reaction of said obtained composition of epoxy functional polyester (III) having an EGC in the range of from 500 to 3000 preferably 1000-2500 and a number average molecular weight M_n in the range of from 500 to 10,000, with an isocyanate (IV), in a molar ratio of III:IV such that at least all the present terminal hydroxy groups will be converted. Preferably, when the isocyanate (IV) represents a diisocyanate the molar ratio of III:IV is in the range of from 0.9:1 to 1.2:1 and more preferably from 1.0:1 to 1.1:1.

It has surprisingly been found that such reaction of the epoxy functional polyester resins with an isocyanate leads to advanced epoxy functional polyester resins,

which show an increased Tg and improved combinations of properties of coating compositions, in which they have been incorporated.

5 It will be appreciated that as the isocyanate (IV) component can be used monoisocyanates, diisocyanates and/or polyisocyanates or mixtures thereof. Preferably diisocyanates are used, optionally mixed with monoisocyanate.

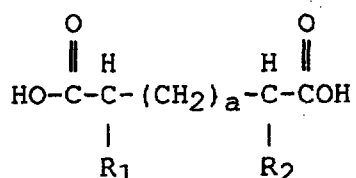
10 It will be appreciated that the present invention also relates to improved powder coating compositions, wherein the modified epoxy functional resin component is formed in situ in a premixture of the unmodified epoxy resin and a stable isocyanate or optionally a blocked isocyanate, e.g. in an extruded premix.

15 An optionally applied blocked isocyanate will deblock at higher temperatures to form a free active isocyanate.

As starting epoxy functional polyester resins (III) for said reaction with e.g. diisocyanate can be used a great variety of epoxy functional polyesters, having a straight or branched structure. Preferably epoxy
20 functional polyesters are used, having a number average molecular weight of from 700 to 1500.

According to a more preferred embodiment the starting epoxy functional polyester resins to be advanced
25 according to the present invention, have been derived from carboxyl functional polyester resins produced by reacting:

a) at least one compound of the formula



wherein $a \geq 1$

V

wherein R_1 and R_2 each may represent an alkyl group having from 1 to 4 carbon atoms, or wherein R_1 and R_2 may form together with the group

$\sim\text{CH}-(\text{CH}_2)_a-\text{CH}\sim$ a cycloalkyl group, which preferably

5 $\begin{array}{c} | \\ \text{represents 1,4-cyclohexane dicarboxylic acid (A1),} \\ | \\ \text{optionally mixed with minor amounts of a corresponding} \\ \text{compound of formula V, wherein } a=0 \text{ or anhydride thereof} \\ \text{(A2),} \end{array}$

10 b) at least one diol compound B comprising two aliphatic hydroxyl groups which may each independently be a primary or a secondary hydroxyl group;

c) optionally a dihydroxymonocarboxylic acid compound C, comprising a tertiary aliphatic carboxyl group and two
15 aliphatic hydroxyl groups, which may each independently be primary or secondary hydroxyl; and

d) optionally a trihydroxyalkane (E1) or tetrahydroxy-alkane (E2);

the molar ratio of compounds $(A_1+A_2):B:C:E_1:E_2$ being
20 $X+Y+2Z+3Q+P:X:Y:Z:Q$, wherein X ranges from 1 to 8, Y ranges from 0 to 8, Z ranges from 0 to 1 and Q ranges from 0 to 1 and wherein P ranges from 1 to 5, and preferably 1-3 and is most preferably equal to 1, at a temperature of from 100 to 220 °C, and preferably from
25 180 to 210 °C if any compound C is present, until essentially all the hydroxyl groups as initially present in the reaction mixture have been reacted.

With the term "minor amounts" as used the optional component A2 are meant amounts of from 0 to 10 mole%,
30 relative to the total molar amount of A1 and A2.

Preferably carboxyl functional polyester resins are aimed at wherein $Y > 0$ if $Z + Q = 0$, or wherein $Z + Q > 0$ if $Y = 0$.

The process for preparation of the starting carboxyl
35 functional polyesters, from which the initial epoxy

functional polyesters to be reacted subsequently with diisocyanate, can in general be carried out according to conventional esterification methods and preferably by azeotropic condensation, taking care that the terminal secondary carboxyl groups are only originating from 1,4-cyclohexane dicarboxylic acid. In particular, the condensation is carried out by charging the compounds A, B, optionally C and optionally D1 or D2, simultaneously to the reactor whereafter the temperature is increased from room temperature to a temperature in the range of from 180 to 220 °C, preferably from 180 to 210 °C in the presence of any compound B, during a period of 3 to 8 hours, thus allowing the reaction to initiate and to proceed under continuous azeotropic removal of water. Generally the azeotropic removal of water is being continued until at least 90% of the original hydroxyl groups have reached and more preferably at least 95% of the original hydroxyl groups have reacted. An esterification catalyst known in the art, such as for example dibutyltin oxide, paratoluenesulphonic acid, tin octoate, zinc octoate and lithium ricinoleate may be used in the esterification process, but is in general not required.

In order to be sure that the terminal secondary carboxyl groups have originated from the structure of formula V dicarboxylic acid wherein $a \geq 1$, and in particular 1,4-cyclohexane dicarboxylic acid, and not from the corresponding 1,2-structure ($a=0$) and in particular 1,2-dicyclohexane dicarboxylic acid, a part of the total batch of e.g. 1,4-cyclohexane di-carboxylic acid to be included, may be added during the reaction and more preferably in its last stage.

Suitable compounds A for use in the process of the present invention are for example phthalic acid (PA), tetrahydrophthalic acid, hexahydrophthalic acid (HHPA),

methylnhexahydrophthalic acid, endomethylenetetrahydrophthalic acid, methyl endomethylenetetrahydrophthalic acid, 1,4-cyclohexanedicarboxylic acid (1,4-CHCA) and 1,3-cyclohexanedicarboxylic acid or combinations thereof; whereof 1,4-cyclohexanedicarboxylic acid is particularly preferred.

Suitable compounds B for use in the preparation of the starting carboxy functional polyesters include branched aliphatic-, cycloaliphatic-, or araliphatic compounds, containing two aliphatic hydroxyl groups, each individually being either a primary or a secondary hydroxyl group, such as for example propylene glycol, neopentyl glycol, hydrogenated diphenylolpropane (HDPP), hydrogenated 4,4'-dihydroxydiphenyl, 1,4-cyclohexanedimethylol, 1,4-dihydroxycyclohexane, hydroxypivalylhydroxypivalate and 2-butyl-2-ethyl-1,3-propanediol or mixtures thereof; of which HDPP is particularly preferred.

Typical examples of a suitable compound C1 for the preparation of starting carboxy functional polyesters are aliphatic alcohols and cycloaliphatic alcohols, having primary or one secondary hydroxyl group and having from 1 to 6 carbon atoms such as neopentanol, 2-butanol, cyclohexanol, or a 1:1 adduct of a VERSATIC acid and a glycidylester of a VERSATIC acid, having from 5 to 13 carbon atoms.

Suitable compounds C2 are aliphatic and cycloaliphatic alcohols having one primary or one secondary hydroxyl group and having of from 1 to 6 carbon atoms and in addition one tertiary aliphatic carboxyl group, such as 1-methyl-4-hydroxycyclohexane-1-carboxylic acid, hydroxypivalic acid.

A typical example of a suitable compound D for use in the preparation of starting carboxy functional polyester is dimethylol propionic acid (DMPA).

A typical and preferred example of compound E, if any, is used for the preparation of the starting branched glycidyl functional polyesters, to be used for the process of the present invention, is trimethylol propane and a preferred example of compound E2 if any is used, is pentaerythritol.

It will be appreciated that the starting glycidyl functional polyester resins can be obtained by easy conversion of a precursor carboxy functional polyester resin with an excess epihalohydrin, in the presence of a suitable base and optionally a catalyst. Most conveniently epichlorohydrin is used.

It was found that those polyglycidyl ester resins to be used as starting material (III) for the preparation of epoxy functional polyester resins of the present invention, are preferred, which have been derived from carboxyl functional polyester resins (II), wherein Y ranges from 1 to 4, X simultaneously ranges from 1 to 6, Z ranges from 0 to 1 and Q ranges from 0 to 1, can provide the more preferred outdoor durable powder coating compositions. Most preferably polyglycidyl ester resins are used, wherein X ranges from 1 to 4, Y ranges from 1 to 2, Z=0, Q=0.

Suitable isocyanates which can be used for the direct in-situ conversion of epoxy functional polyester resins, can be selected from toluene diisocyanate; 1-naphthyl isocyanate, 4-methoxyphenyl isocyanate, cyclohexyl isocyanate, ethyl isocyanate, isobutylisocyanate, isopropyl isocyanate, methoxymethyl isocyanate, phenyl isocyanate, 4,4'-diphenylmethane diisocyanate; hexamethylene diisocyanate, 1,12-diisocyanatododecane; isophorone diisocyanate; 4,4'-dicyclohexylmethane diisocyanate, 1,4-cyclohexane diisocyanate.

Preferred diisocyanates are toluene diisocyanate; hexamethylene diisocyanate, 1,12-diisocyanatododecane;

isophorone diisocyanate and 4,4'-dicyclohexylmethane diisocyanate, of which 4,4'-dicyclohexyl methane diisocyanate and 1,4-cyclohexane diisocyanate are most preferred.

5 Examples of blocked diisocyanates cross-linking component include those which are based on isophorone diisocyanate blocked with ϵ -caprolactam such as described in US 3,822,240, 4,150,211 and 4,212,962.

10 The reaction between the diisocyanate and the epoxy functional polyester resin usually takes place at temperatures in the range of from ambient temperature to 180 °C and preferably from 20 °C to 160 °C.

15 The reaction can be carried out in a solvent selected from a variety of polar solvents, such as ketones, ethers, esters, of which the ketones are preferred. A more preferred solvent is methyl isobutyl ketone. However, it will be appreciated that the reaction can also be carried out in principle as a mass reaction e.g. in an extruder.

20 The curable outdoor durable powder coating compositions, forming another aspect of the present invention, may be prepared by addition of a cross-linking resin to the hereinbefore specified modified polyglycidyl ester resin (I).

25 As cross-linking resin a precursor carboxy functional polyester resin (II) from which the intermediate initially prepared polyglycidyl ester (III) can be derived by glycidation, can be used. Preferably said precursor has a polyester chain microstructure which is rather similar to that of the final polyglycidylester.

30 The amount of cross-linking compound used in the powder coating compositions of the invention will normally be such so as to provide about equal amounts of reactive groups of the cross-linking compound and the

epoxy groups present in the advanced polyglycidyl ester resin.

However, other suitable cross-linking resins can also be used in combination with the advanced polyglycidyl ester resins of the present invention, such as solid polyacids such as sebacic acid, 1,12-dodecanedioic acid; anhydrides such as polyazelaic polyanhydride; acid functional polyesters such as the reaction product of one mole of trimethylolpropane and 3 moles of hexahydro-phthalic anhydride, the reaction product of 1,6-hexanediol with a molar excess of 1,12-dodecanedioic acid, the reaction product of 4 moles 1,10-decanedicarboxylic acid, 1.49 mols hexanediol, 0.47 mols 1,1,1-tris-(hydroxymethyl)-propane and 0.27 mols pentaerythritol, the reaction product of 4 mols 1,10-decanedicarboxylic acid, 1.2 mols hexanediol, 0.45 mols trimethylolpropane, 0.29 mols pentaerythritol and 0.21 mols dimethylolpropionic acid and the reaction product of one mole of hexamethoxymethylmelamine and 3 moles of hydroxypivalic acid and amine-type curing agents.

The powder coating compositions of the present invention may further comprise a catalyst and optionally other additives, as known in the art to be suitable for use in powder coating compositions.

Suitable catalysts are for example quaternary ammonium and phosphonium salts; metal salts/compounds such as for example stannous(II)octoate; basic compounds such as for example the imidazoles; and tertiary amines such as for example diazabicycloundecene.

The amount of catalyst used will usually be somewhere in the range of from 0.01 to 2% by weight based on the weight of the total powder coating composition.

Suitable cure times and cure temperatures of the powder coating compositions of the invention are those

conventionally applied in connection with powder coating systems.

It will be appreciated that another aspect of the present invention is formed by formed substrates coated with a curable outdoor durable powder coating composition as specified hereinbefore, in cured or uncured state.

EXAMPLES

The invention is illustrated by the following examples, however without restricting its scope to these specific embodiments.

Example 1

Preparation of aliphatic carboxyl functional polyesters P1 to P6 according to invention.

Compounds A1 to D1 were charged in molar amounts as indicated in Table 1 in around bottom glass reactor, equipped with a condenser, temperature control, nitrogen inlet, stirrer and vacuum inlet.

The mixture was then heated to 150 °C in 30 minutes. The temperature of the reaction mixture was increased in two hours to 210 °C and kept at 210 °C until at least 94% of the original hydroxyl groups have reacted. The acid functional polyester thus formed was discharged and allowed to cool down to room temperature.

Table 1 1)2)

Carboxyl functional polyesters

Polyester resin code		P1	P2	P3	P4	P5	P6
1,4-CHCA	A1	5.5	9	9	4.5	3	6
DMPA	C	1.5	2.5	2	1	-	-
HDPP	B1	3	5.5	6	2	1.5	1
TMP	D1	-	-	-	-	-	1
1,4-DMCH	B2	-	-	-	-	-	2

1. Numbers given represent the amount of moles of a particular compound charged to the reactor in the process of example 1 for the preparation of the tertiary and partly secondary carboxyl functional polyester resins P1 to P6.

2. The chemical compounds represented by the abbreviations used are:

1,4-CHCA : 1,4-cyclohexanedicarboxylic acid

DMPA : dimethylolpropionic acid

HDPP : hydrogenated diphenylolpropane

TMP : trimethylol propane

1,4-DMCH : 1,4-dimethylolcyclohexane

Example 2

Preparation of corresponding glycidylester resins EP1 to EP4. An amount equalling 1 carboxyl group equivalent of linear tertiary and partly secondary aliphatic carboxyl functional polyester resins obtained as examples P1 to P5 in Table 1, was dissolved in 16 moles epichlorohydrin (ECH), 15.5 moles isopropylalcohol (IPA) and 15.5 moles of demi-water. The solution was charged to a glass-reactor equipped with a temperature control, stirrer and reflux condenser. Next, the temperature was raised to 70 °C, followed by gradual addition of an 50 %wt aqueous solution of 0.16 moles of NaOH over a period of 20 min. The temperature is increased during this caustic addition from 70 to 80 °C. At this temperature the mixture is stirred until virtually all acid groups are converted (approx. 50 min.) Subsequently, the mixture is cooled to 55 °C. At this temperature, gradual addition of an 50 %wt aqueous solution of 0.92 moles of NaOH over a period of 60 min. followed. After a post-reaction of 5 min., 15 moles of water was added and the mixture was stirred for 5 min. The reactor content was allowed to settle and subsequently the brine was separated from the organic phase. The excess of ECH,

the IPA and some remaining water was flashed off with vacuum.

The resulting crude resin EP1, EP2, EP3 and EP4 was dissolved in methyl isobutyl ketone (MIBK) (MIBK/Resin=5/1 v/v) and washed 4 times with water (water/resin=1/1 v/v) at 70-75 °C. The MIBK is removed by vacuum flash and the polyglycidylester was discharged from the reactor and allowed to cool down.

The epoxy resins obtained showed the following properties.

Table 2
Epoxy functional polyesters

Epoxy Polyester resin code	EP1	EP2	EP3	EP4
EGC [mmol/kg]	1400	1360	1214	1657
Hydroxyl content [mmol/kg], Calculated.	300-600	400-600	300-500	100-300
Melt viscosity 1) at 200 °C [Poise]	9	9	11	4
T _g 2) [°C]	32	35	39	20

1) Determined on ICI cone and plate apparatus.

2) DSC, second scan, midpoint value.

Example 3

Preparation of corresponding modified polyglycidylester resins MEP1, MEP2, MEP3 and MEP4.

Compounds EP1, EP2 and EP3 were charged in weight amounts as indicated in Table 3 in around bottom glass reactor, equipped with a condenser, temperature control, nitrogen inlet, stirrer and vacuum inlet.

The epoxy polyester was dissolved in methyl isobutyl keton (MIBK), Resin/MIBK=1/5 v/v. Subsequently a weight amount of coupling agent, as indicated in Table 3 was added. The mixture was then heated to 90 °C in 30 minutes and hold at 90 °C until virtually all OCN groups are converted (approx. 60-180 min.). Subsequently, the MIBK is removed by vacuum flash and the modified polyglycidylester was discharged from the reactor and allowed to cool down.

The modified epoxy resins obtained showed the following properties.

Table 3 1)

Modified epoxy functional polyesters

Modified Epoxy Polyester code		MEP1	MEP2	MEP3	MEP4	MEP5	MEP6
Epoxy functional polyester resin ²⁾ :	EP1	96	94	-	-	95	-
	EP2	-	-	96	-	-	-
	EP4	-	-	-	96	-	92
Coupling agent :	HMDI ³⁾	4	6	4	4	-	-
Carboxyl functional polyester ⁴⁾ :	P5	-	-	-	-	5	8
EGC [mmol/kg]		1327	1300	1308	1580	1280	1252
Hydroxyl content [mmol/kg], Calculated		<300	<150	<300	<100	400-700	300-500
Melt viscosity ⁵⁾ at 200 °C [Poise]		29	41	32	30	15	16
T _g ⁶⁾ [°C]		47	51	50	55	39	45

1. Epoxy functional polyester resin: EP1, EP2 and EP4 and coupling agents HMDI (4,4'-cyclohexylmethane Diisocyanate) and carboxyl functional polyester P5 are expressed in a w/w ratio.
2. Prepared as described in example 2, Table 2.
3. HMDI is 4,4'-cycloHexylMethane DiIsocyanate.
4. Prepared as described in example 1, Table 1.
5. Determined on ICI cone and plate apparatus.
6. DSC, second scan, midpoint value.

Table 4
White pigmented binder and coating properties

Ingredients*		BC1	BC2	BC3	BC4	BC5	BC6
Epoxy polyester resin (1)	EP1	-	-	-	885	-	-
"	EP2	-	-	-	-	895	-
"	EP4	-	-	-	-	-	823
Modified epoxy polyester (2)	MEP1	904	-	-	-	-	-
"	MEP3	-	909	-	-	-	-
"	MEP4	-	-	840	-	-	-
Carboxyl functional resin (3)	6	596	589	660	615	605	677
Titaniumdioxide		900	900	900	900	900	900
Modaflow (4)		36	36	36	36	36	36
Benzoin		12	12	12	12	12	12
Catalyst		4.5	4.5	4.5	4.5	4.5	4.5
<u>Powder characteristics</u>							
Extrusion behaviour		←-----good-----→					
Storage stability at 40 °C (5)		good	good	good	poor	poor	poor

Table 4 (Cont'd)

Ingredients*		BC1	BC2	BC3	BC4	BC5	BC6
Coatings (6)							
Gloss (7)		89	85	86	87	85	82
Reverse Impact (8)		> 60	> 60	> 40	> 40	> 60	> 20
Accelerated weather resistance (9)		> 2700	> 2100	> 2100	2400	1800	1800

* all amounts in grams

- (1) Prepared as described in example 2.
- 5 (2) Prepared as described in example 3.
- (3) Prepared as described in example 1.
- (4) MODAFLOW III is an acrylic based flow aid available from Monsanto Co.
- (5) Physical storage stability, assessment of the tendency to block. Ranking: poor, moderate and good.
- 10 (6) Coatings: Q-panel AL-36, cured at 200°C for 15 minutes and thickness 40-60 micrometres.
- (7) Measured with Gardner micro-TRI-gloss apparatus, angle 60°.
- (8) Measured in inch/pounds according to Ericksen type 304 ISO-TR-6272/1979-DIN-55669.
- 15 (9) Hours in Atlas Weather-OMeter, running SAE J1960 test method before 50% reduction in gloss.

C L A I M S

1. Modified epoxy functional polyesters, obtainable by the reaction of epihalohydrin and preferably epichlorohydrin, in the presence of a base, with carboxyl functional polyester resins II obtainable by reacting:

5 (a) at least one aromatic, aliphatic or cycloaliphatic dicarboxylic acid compound A comprising two aromatic- or secondary aliphatic carboxyl groups or the anhydride thereof;

10 (b) at least one diol compound B comprising two aliphatic hydroxyl groups, which may independently be a primary or a secondary hydroxyl group; and optionally

(c) compound C1 comprising one monofunctional primary- or secondary hydroxyl group and/or at least one compound C2 comprising one primary- or secondary hydroxyl group and one tertiary aliphatic carboxyl group; and optionally

15 (d) a dihydroxymonocarboxylic acid compound D comprising a tertiary aliphatic carboxyl group and two aliphatic hydroxyl groups, which may each independently be primary or secondary hydroxyl; and optionally

20 (e) a trihydroxyalkane (E1) or tetrahydroxyalkane (E2), the molar ratio of compounds A:B:C1:C2:D:E1:E2 being $X+Y+1:X:M:N:Y:Z:Q$

wherein M+N is in the range of from 0 to 2, X ranges from 2 to 8 and Y ranges from 0 to 8, Z ranges from 0 to 2 and Q ranges from 0 to 2 at a temperature of from 100 to 220 °C, until a predominant part of the non-tertiary carboxyl groups as initially present in the reaction mixture, i.e. 75% or more, have been converted, and subsequent reaction of said obtained composition of epoxy functional polyester (III) having an EGC in the range of from 500 to 3000 preferably 1000-2500 and a number

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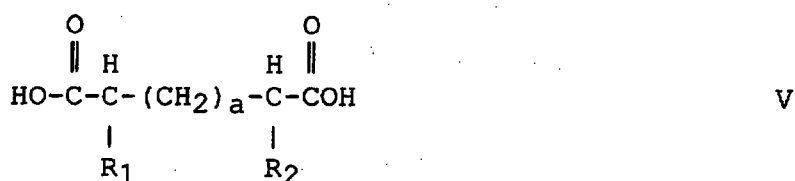
average molecular weight M_n in the range of from 500 to 10,000, with an isocyanate (IV), in a molar ratio of III:IV such that at least all the present terminal hydroxyl groups will be converted.

5 2. Modified epoxy functional polyesters according to claim 1, characterized in that the molar ratio between the starting epoxy functional polyester resin (II) and the isocyanate (IV) is in the range of from 0.9:1 to 1.2:1.

10 3. Modified epoxy functional polyesters according to claims 1-2, characterized in that the starting epoxy functional polyesters III have a number average molecular weight of from 700 to 1500.

15 4. Modified epoxy functional polyesters according to claims 1-3, characterized in that as starting epoxy functional polyester resin III are used those, which have been derived from carboxyl functional polyester resins produced by reacting:

a) at least one compound of the formula

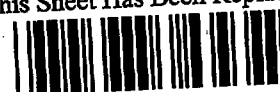


20 wherein $a \geq 1$

wherein R_1 and R_2 each may represent an alkyl group having from 1 to 4 carbon atoms, or wherein R_1 and R_2 may form together with the group

25 $\sim\text{CH}-(\text{CH}_2)_a-\text{CH}\sim$ a cycloalkyl group, which preferably represents 1,4-cyclohexane dicarboxylic acid (A1), optionally mixed with minor amounts of a corresponding compound of formula V, wherein $a=0$ or anhydride thereof (A2),

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b) at least one diol compound B comprising two aliphatic hydroxyl groups which may each independently be a primary or a secondary hydroxyl group;

5 c) optionally a dihydroxymonocarboxylic acid compound C, comprising a tertiary aliphatic carboxyl group and two aliphatic hydroxyl groups, which may each independently be primary or secondary hydroxyl; and

d) optionally a trihydroxyalkane (E1) or tetrahydroxy-alkane (E2),

10 the molar ratio of compounds $(A_1+A_2):B:C:E1:E2$ being $X+Y+2Z+3Q+P:X:Y:Z:Q$, wherein X ranges from 1 to 8, Y ranges from 0 to 8, Z ranges from 0 to 1 and Q ranges from 0 to 1 and wherein P ranges from 1 to 5, and preferably 1-3 and is most preferably equal to 1, at a
15 temperature of from 100 to 220 °C, and preferably from 180 to 210 °C if any compound C is present, until essentially all the hydroxyl groups as initially present in the reaction mixture have been reacted.

5. Modified epoxy functional polyesters according to
20 claim 4, characterized in that in the preceding carboxyl functional polyester resins is $Y>0$ if $Z+Q=0$ or $Z+Q>0$ if $Y=0$.

6. Modified epoxy functional polyesters according to
25 claims 1-5, characterized in that polyglycidyl esters III are used as starting material, which have been derived from carboxyl functional polyester resins (II), wherein Y ranges from 1 to 2, X simultaneously ranges from 1 to 4 $Z=0$ and $Q=0$.

7. Modified epoxy functional polyesters according to
30 claims 1-6, characterized in that the isocyanate is selected from toluene diisocyanate; hexamethylene diisocyanate; 1,12-diisocyanato dodecene; isophorone diisocyanate and 4,4'-dicyclohexylmethane diisocyanate.

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8. A curable outdoor durable powder coating composition comprising modified epoxy functional polyesters as claimed in claims 1-7, and a cross-linking agent.
- 5 9. A curable outdoor durable powder coating composition according to claim 8, characterized in that it comprises as cross-linking agent a carboxyl functional polyester, from which the epoxy functional polyesters have been derived.
- 10 10. A formed substrate coated with a powder coating composition according to claims 11 and 12 in uncured or cured state.



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