

STABLE PHARMACEUTICAL PREPARATION CONTAINING
GRANULOCYTE COLONY STIMULATING FACTOR AND
PROCESS FOR PRODUCING THE SAME

The present invention relates to a pharmaceutical
5 preparation containing a granulocyte colony stimulating
factor. In particular, the present invention relates to a
stabilized pharmaceutical preparation containing a granulo-
cyte colony stimulating factor that is protected against
loss or inactivation of the active component (i.e., granulo-
10 cyte colony stimulating factor) due to adsorption on the
wall of a container in which the preparation is put, or to
association, polymerization or oxidation of said component.

Chemotherapy has been undertaken as one method for
treating a variety of infectious diseases but it has
15 recently been found that chemotherapy causes some serious
clinical problems such as the generation of drug-resistant
organisms, change of causative organisms, and high side
effects. In order to avoid these problems associated with
chemotherapy involving the use of therapeutic agents such as
20 antibiotics and bactericides, attempts are being made to use
a substance that activates the prophylactic capabilities of
the host of an infection-causing organism and thereby pro-
viding a complete solution to the aforementioned problems of
chemotherapy. Of the various prophylactic capabilities of
25 the host, the phagocytic bactericidal action of leucocytes
is believed to cause the strongest influence in the initial
period of bacterial infection and it is therefore assumed to
be important to enhance the infection protecting capabili-
ties of the host by promoting the growth of neutrophiles and

their differentiation into the mature state. A granulocyte colony stimulating factor (G-CSF) is one of the very useful substances that exhibit such actions and the same assignee of the present invention previously filed a patent application on an infection protecting agent using G-CSF (Japanese Patent Application No. 23777/1985).

As mentioned above, chemotherapy as currently practiced involves various unavoidable problems and intensive efforts are being made to use a drug substance that is capable of activating the prophylactic functions of the host or the person who has been infected.

Needless to say, G-CSF displays by itself the ability to activate the prophylactic functions of the host and it has also been found that G-CSF exhibits greater therapeutic effects in clinical applications if it is used in combination with a substance that activates the prophylactic capabilities of the host.

G-CSF is used in a very small amount and a pharmaceutical preparation containing 0.1 - 500 μ g (preferably 5 - 50 μ g) of G-CSF is usually administered at a dose rate of 1 - 7 times a week per adult. However, G-CSF has a tendency to be adsorbed on the wall of its container such as an ampule for injection or a syringe. Therefore, if the drug is used as an injection in such a form as an aqueous solution, it will be adsorbed on the wall of its container such as an ampule or a syringe. This either results in the failure of G-CSF to fully exhibit its activity as a pharmaceutical agent or necessitates the incorporation of G-CSF

in a more-than-necessary amount making allowance for its possible loss due to adsorption.

In addition, G-CSF is liable and highly susceptible to environmental factors such as temperature, humidity, oxygen and ultraviolet rays. By the agency of such factors, G-CSF undergoes physical or chemical changes such as association, polymerization and oxidation and suffers a great loss in activity. These phenomenon make it difficult to ensure complete accomplishment of a therapeutic act by administering a very small amount of G-CSF in a very exact manner.

It is therefore necessary to develop a stable pharmaceutical preparation of G-CSF that is fully protected against a drop in the activity of its effective component. This is the principal object of the present invention which provides a stable pharmaceutical preparation of G-CSF.

The present inventors conducted intensive studies in order to enhance the stability of a G-CSF containing pharmaceutical preparation and found that this object can effectively be attained by addition of a pharmaceutically acceptable surfactant, saccharide, protein or other high-molecular weight compound.

According to the present invention, there is provided a stable granulocyte colony stimulating factor containing pharmaceutical preparation that contains, in addition to the granulocyte colony stimulating factor present as the effective ingredient, at least one substance selected from pharmaceutically acceptable surfactants, saccharides, proteins and other high-molecular weight compounds such as natural polymers or synthetic polymers; wherein the granulocyte colony stimulating factor has the following physicochemical properties:

- (i) molecular weight: about $19,000 \pm 1,000$ as measured by electrophoresis

through a sodium
dodecylsulfate-
polyacrylamide gel;

(ii) isoelectric point: having at least one of the
three isoelectric points,
 $pI = 5.5 \pm 0.1$, $pI = 5.8 \pm 0.1$, and $pI = 6.1 \pm 0.1$;

(iii) ultraviolet absorption: having a maximum
absorption at 280 nm and a minimum
absorption at 250 nm;

(iv) amino acid sequence of the 21 residues from N
terminus:

H₂N-Thr-Pro-Leu-Gly-Pro-Ala-Ser-Ser-Leu-Pro-Gln-
Ser-Phe-Leu-Leu-Lys-Cys-Leu-Glu-Gln-Val-.

The present invention also provides a process for
producing a stable granulocyte colony stimulating
factor containing pharmaceutical preparation, which
process comprises mixing granulocyte colony stimulating
factor (G-CSF) with at least one substance selected
from pharmaceutically acceptable surfactants,
saccharides, proteins and other high molecular weight
substances such as natural polymers and synthetic
polymers, dissolving the resulting mixture in a buffer
solution, and charging a container with the solution;
wherein said G-CSF has the following properties:

(i) molecular weight: about $19,000 \pm 1,000$ as
measured by electrophoresis
through a sodium
dodecylsulfate-
polyacrylamide gel;

(ii) isoelectric point: having at least one of the
three isoelectric points,
 $pI = 5.5 \pm 0.1$, $pI = 5.8 \pm 0.1$, and $pI = 6.1 \pm 0.1$;

(iii) ultraviolet absorption: having a maximum
absorption at 280 nm and a minimum

absorption at 250 nm;

(iv) amino acid sequence of the 21 residues from N terminus:

5 H₂N-Thr-Pro-Leu-Gly-Pro-Ala-Ser-Ser-Leu-Pro-Gln-Ser-Phe-Leu-Leu-Lys-Cys-Leu-Glu-Gln-Val-.

For details of the method for preparing the G-CSF to be used in the present invention, see the specifications of Japanese Patent Application Nos. 153273/1984, 269455/1985, 269456/1985, 270838/1985 and 10 270839/1985, all having been filed by the assignee of the present invention.

The human G-CSF containing solution obtained may be stored in a frozen state after being further purified and concentrated, as required, by any known 15 technique. Alternatively, the solution may be stored after being dehydrated by such means as freeze-drying.

The human G-CSF thus prepared can be processed as specified by the present invention in order to attain stable G-CSF containing pharmaceutical preparations.

20 Typical examples of the surfactant that is used to attain the stable G-CSF containing pharmaceutical preparation of the present invention are listed below: nonionic surfactants with HLB of 6 -18 such as sorbitan aliphatic acid esters (e.g. sorbitan monocaprylate, 25 sorbitan monolaurate and sorbitan monopalmitate), glycerin aliphatic acid esters (e.g. glycerin monocaprylate, glycerin monomyristate, and glycerin monostearate), polyglycerin aliphatic acid esters (e.g. decaglyceryl monostearate, decaglyceryl distearate and 30 decaglyceryl monolinoleate), polyoxyethylene sorbitan aliphatic acid esters (e.g. polyoxyethylene

sorbitan monolaurate, polyoxyethylene sorbitan monooleate, polyoxyethylene sorbitan monostearate, polyoxyethylene sorbitan monopalmitate, polyoxyethylene sorbitan trioleate, and polyoxyethylene sorbitan tristearate), polyoxyethylene sorbitol aliphatic acid esters (e.g. polyoxyethylene sorbitol tetrastearate and polyoxyethylene sorbitol tetraoleate), polyethylene glycerin aliphatic acid esters (e.g. polyoxyethylene glyceryl monostearate), polyethylene glycol aliphatic acid esters (e.g. polyethylene glycol distearate), polyoxyethylene alkyl ethers (e.g. polyoxyethylene lauryl ether), polyoxyethylene polyoxypropylene alkyl ethers (e.g. polyoxyethylene polyoxypropylene glycol ether, polyoxyethylene polyoxypropylene propyl ether, and polyoxyethylene polyoxypropylene cetyl ether), polyoxyethylene alkylphenyl ethers (e.g. polyoxyethylene nonylphenyl ether), polyoxyethylated castor oil, polyoxyethylated hardened castor oil (polyoxyethylated hydrogenated castor oil), polyoxyethylated beeswax derivatives (e.g. polyoxyethylated sorbitol beeswax), polyoxyethylene lanolin derivatives (e.g. polyoxyethylene lanolin), and polyoxyethylene aliphatic acid amides (e.g. polyethylene stearic acid amide); anionic surfactants such as alkyl sulfuric acid salts having a $C_{10} - C_{18}$ alkyl group (e.g. sodium cetyl sulfate, sodium lauryl sulfate and sodium oleyl sulfate), polyoxyethylene alkyl ether sulfuric acid salts wherein the average molar number of ethylene oxide addition is 2 - 4 and the alkyl group has 10 - 18 carbon atoms (e.g. polyoxyethylene sodium lauryl sulfate), salts of alkyl sulfosuccinate esters wherein the alkyl group has 8 -

18 carbon atoms (e.g. sodium lauryl sulfosuccinate ester) and natural surfactants such as lecithin, glycerophospholipid, sphingophospholipid (e.g. sphingomyelin), and sucrose aliphatic acid esters wherein the aliphatic acid has 12 - 18 carbon atoms. These surfactants may of course be used either independently or in admixture.

The surfactants listed above are preferably used in amounts of 1 - 10,000 parts by weight per part by weight of G-CSF.

10 The saccharide to be used in making the stable G-CSF containing pharmaceutical preparation of the present invention may be selected from among monosaccharides, oligosaccharides, and polysaccharides, as well as phosphate esters and nucleotide derivatives thereof so long as they are pharmaceutically acceptable. Typical examples are listed below: trivalent and higher sugar alcohols such as glycerin, erythritol, arabitol, xylitol, sorbitol, and mannitol; acidic sugars such as glucuronic acid, iduronic acid, neuraminic acid, galacturonic acid, gluconic acid, 20 mannuronic acid, ketoglycolic acid, ketogalactonic acid and ketogulonic acid; hyaluronic acid and salts thereof, chondroitin sulfate and salts thereof, heparin, inulin, chitin and derivatives thereof, chitosan and derivatives thereof, dextrin, dextran with an average molecular weights of 5,000 25 - 150,000, and alginic acid and salts thereof. All of these saccharides may be used with advantage either independently or in admixture.

The saccharides listed above are preferably used in amounts of 1 - 10,000 parts by weight per part by weight of G-CSF.

Typical examples of the protein to be used in making the stable G-CSF containing pharmaceutical preparation of the present invention include human serum albumin, human serum globulin, gelatin, acid-treated gelatin (average mol. wt. = 7,000 - 100,000), alkali-treated gelatin (average mol. wt. = 7,000 - 100,000), and collagen. Needless to say, these proteins may be used either independently or in admixture.

The proteins listed above are preferably used in amounts of 1 - 20,000 parts by weight per part by weight of G-CSF.

Typical examples of the high-molecular weight compound to be used in making the stable G-CSF containing pharmaceutical preparation of the present invention include: natural polymers such as hydroxypropyl cellulose, hydroxymethyl cellulose, sodium carboxymethyl cellulose, and hydroxyethyl cellulose; and synthetic polymers such as polyethylene glycol (mol. wt. = 300 - 6,000), polyvinyl alcohol (mol. wt. = 20,000 - 100,000), and polyvinylpyrrolidone (mol. wt. = 20,000 - 100,000). Needless to say, these high-molecular weight compounds may be used either alone or in combination.

The high-molecular weight compounds listed above are desirably used in amounts of 1 - 20,000 parts by weight per part by weight of G-CSF.

In addition to the surfactant, saccharide, protein or high-molecular weight compound described above, at least one member selected from the group consisting of an amino acid, a sulfureous reducing agent and an antioxidant may also be incorporated in making the G-CSF containing pharmaceutical preparation of the present invention. Illustrative amino acids include glycine, threonine, tryptophan, lysine, hydroxylysine, histidine, arginine, cysteine, cystine, and methionine. Illustrative sulfureous reducing agents include: N-acetylcysteine, N-acetylhomocysteine, thiocetic acid, thiodiglycol, thioethanolamine, thioglycerol, thio-sorbitol, thioglycolic acid and salts thereof, sodium thio-sulfate, sodium hydrogensulfite, sodium pyrosulfite, sodium sulfite, thiolactic acid, dithiothreitol, glutathione, and a mild sulfureous reducing agent having a sulfhydryl group such as a $C_1 - C_7$ thioalkanoic acid. Illustrative antioxidants include erythorbic acid, dibutylhydroxytoluene, butylhydroxyanisole, d α -tocopherol, tocopherol acetate, L-ascorbic acid and salts thereof, L-ascorbic acid palmitate, L-ascorbic acid stearate, triamyl gallate, propyl gallate and chelating agents such as disodium ethylenediaminetetraacetate (EDTA), sodium pyrophosphate and sodium metaphosphate.

The above-listed amino acids, sulfureous reducing agents and antioxidants or mixtures thereof are preferably used in amounts of 1 - 10,000 parts by weight per part by weight of G-CSF.

For the purpose of formulating the stable G-CSF containing preparation of the present invention in a suitable dosage form, one or more the following agents may be incorporated: a diluent, a solubilizing aid, an isotonic agent, an excipient, a pH modifier, a soothing agent, and a buffer.

The stabilized G-CSF pharmaceutical preparation of the present invention may be formulated either for oral administration or for parenteral administration such as by injection applied in various ways, and a variety of dosage forms may be employed depending upon the specific mode of administration. Typical dosage forms include: those intended for oral administration such as tablets, pills, capsules, granules and suspensions; solutions, suspensions and freeze-dried preparations principally intended for intravenous injection, intramuscular injection, subcutaneous injection and intracutaneous injection; and those intended for transmucosal administration such as rectal suppositories, nasal drugs, and vaginal suppositories.

According to the present invention, at least one substance selected from the group consisting of a surfactant, a saccharide, a protein or a high-molecular weight compound is added to a G-CSF containing pharmaceutical preparation so that it is prevented from being adsorbed on the wall of its container or a syringe while at the same time, it remains stable over a prolonged period of time.

The detailed mechanism by which the substances mentioned above stabilized G-CSF or prevent it from being

adsorbed is yet to be clarified. In the presence of a surfactant, the surface of G-CSF which is a hydrophobic protein would be covered with the surfactant to become solubilized so that the G-CSF present in a trace amount is effectively prevented from being adsorbed on the wall of its container or a syringe. A saccharide or hydrophilic high-molecular weight compound would form a hydrated layer between G-CSF and the adsorptive surface of the wall of its container or a syringe, thereby preventing adsorption of G-CSF in an effective manner. A protein would compete with G-CSF for adsorption on the wall of its container or a syringe, thereby effectively inhibiting adsorption of G-CSF.

Besides the prevention of G-CSF adsorption, the substances mentioned above would also contribute to the prevention of association or polymerization of the molecules of G-CSF. In the presence of a surfactant, saccharide, protein or high-molecular weight compound, the individual molecules of G-CSF are dispersed in these substances and the interaction between the G-CSF molecules is sufficiently reduced to cause a significant decrease in the probability of their association or polymerization. In addition, these substances would retard the autoxidation of G-CSF that is accelerated under high temperature or humidity or prevent G-CSF from being associated or polymerized as a result of its autoxidation. These effects of retarding autoxidation of G-CSF or preventing it from being associated or polymerized would be further enhanced by addition of an amino

acid, a sulfurous reducing agent or an antioxidant.

The problems described above are particularly noticeable in solutions for injecting and in suspensions but they also occur during the process of
5 formulating G-CSF in other dosage forms such as tablets. The addition of surfactants, saccharides, proteins or high-molecular weight compounds is also effective in this latter case.

Through the addition of at least one substance
10 selected from the group consisting of a surfactant, saccharide, protein and a high-molecular weight compound, G-CSF is highly stabilized and maintains its activity for a prolonged period of time, as will be demonstrated in the examples that follow. To attain
15 these results, the following ranges of the amount of each of these substances are desirable: 1 - 10,000 parts by weight of surfactant, 1 - 10,000 parts by weight of saccharide, 1-20,000 parts by weight of protein, and 1 - 20,000 parts by weight of high-
20 molecular weight compound, per 1 part by weight of G-CSF.

According to the present invention, a surfactant, a saccharide, a protein and/or a high-molecular weight compound is used in a specified concentration and this
25 is effective not only in preventing G-CSF from being adsorbed on the wall of its container or a syringe but also in enhancing the stability of a G-CSF containing pharmaceutical preparation. As a result, it becomes possible to ensure the administration of a small but
30 highly precise dose of G-CSF to patients; since G-CSF is costly, its efficient

utilization will lead to lower costs for the production of G-CSF containing pharmaceutical preparations.

The following examples are provided for the purpose of further illustrating the present invention but are in no sense to be taken as limiting. In these examples, the residual activity of G-CSF was determined by one of the following methods.

(a) Soft agar method using mouse bone marrow cells:

A horse serum (0.4 ml), 0.1 ml of the sample, 0.1 ml of a C3H/He (female) mouse bone marrow cell suspension ($0.5 - 1 \times 10^5$ nuclear cells), and 0.4 ml of a modified McCoy's 5A culture solution containing 0.75% of agar were mixed, poured into a plastic dish for tissue culture (35 mm^Ø), coagulated, and cultured for 5 days at 37°C in 5% CO₂/95% air, and at 100% humidity. The number of colonies formed was counted (one colony consisting of at least 50 cells) and the activity was determined with one unit being the activity for forming one colony.

The modified McCoy's 5A culture solution used in the method (a) was prepared by the following procedures.

Modified McCoy's 5A culture solution (double concentration)

Twelve grams of McCoy's 5A culture solution (Gibco), 2.55 g of MEM amino acid-vitamin medium (Nissui Seiyaku Co., Ltd.), 2.18 g of sodium bicarbonate and 50,000 units of potassium penicillin G were dissolved twice in 500 ml of distilled water and the solution was aseptically filtered through a Millipore filter (0.22 µm).

(b) Reverse-phase high-performance liquid chromatography:

Using a reverse-phase C8 column (4.6 mm x 300 mm; 5 µm) and an n-propanol/trifluoroacetic acid mixture as a mobile phase, the residual activity of G-CSF (injected in an amount equivalent to 1 µg) was determined under the following gradient conditions:

	<u>Time (sec)</u>	<u>Solvent (A)</u>	<u>Solvent (B)</u>	<u>Gradient</u>
	0	100%	0%	} linear
	15	0%	100%	
10	25	100%	0%	} linear

Solvent (A): 30% n-propanol and 0.1% trifluoroacetic acid

Solvent (B): 60% n-propanol and 0.1% trifluoroacetic acid

Detection was conducted at a wavelength of 210 nm and the percentage of the residual G-CSF activity was calculated by the following formula:

$$\text{Residual G-CSF activity (\%)} = \frac{\text{the residual amount of G-CSF after the lapse of a given time}}{\text{the initial amount of G-CSF}} \times 100$$

The residual amount of G-CSF as determined by this method correlated very well with the result attained in measurement by the soft agar method (a) using mouse bone marrow cells.

20 Example 1

To 5 µg of G-CSF, one of the stabilizing agents listed in Table 1 was added and the mixture was aseptically dissolved in a 20 mM buffer solution (containing 100 mM sodium chloride; pH 7.4) to make a pharmaceutical preparation containing 5 µg of G-CSF per ml, which was then freezer-dried. The time-dependent change in G-CSF activity

was measured by method (a) and the results are shown in Table 1. The term "activity (%)" in the table represents the residual activity of G-CSF relative to the initial unit and is defined by the following formula:

$$\text{Activity (\%)} = \frac{\text{activity unit after the lapse of a given time}}{\text{initial activity unit}} \times 100$$

5 Freeze-drying was conducted by the following procedures:

 The G-CSF solution containing a stabilizing agent was put into a sterile sulfa-treated glass vial, frozen at -40°C or below for 4 hours, subjected to primary drying by heating
10 from -40°C to 0°C over a period of 48 hours with the pressure increased from 0.03 to 0.1 torr, then to secondary during by heating from 0°C to 20°C for a period of 12 hours with the pressure increased from 0.03 to 0.08 torr; there-
15 after, the interior of the vial was filled with a sterile dry nitrogen gas to attain an atmospheric pressure and the vial was plugged with a freeze-drying rubber stopper, then sealed with an aluminum cap.

Table 1

Stabilizing agent	Amount (parts by weight)	Activity (%)	
		After storage at 4°C for 6 months	After storage at 37°C for 1 month
xylitol	10,000	92	86
mannitol	10,000	91	85
glucuronic acid	10,000	86	82
hyaluronic acid	2,000	92	89
dextran (m.w. 40,000)	2,000	95	90
heparin	5,000	85	80
chitosan	2,000	93	91
alginic acid	2,000	90	90
human serum albumin	1,000	98	99
human serum globulin	1,000	98	95
acid-treated gelatin	2,000	97	95
alkali-treated gelatin	1,000	99	96
collagen	2,000	95	90
polyethylene glycol (m.w. 4,000)	10,000	94	90
hydroxypropyl cellulose	1,000	98	94
sodium carboxymethyl cellulose	1,000	88	80
hydroxymethyl cellulose	5,000	92	90
polyvinyl alcohol (m.w. 50,000)	2,000	96	95
polyvinylpyrrolidone (m.w. 50,000)	2,000	95	94
human serum albumin	2,000	100	97
mannitol	2,000		
cysteine	100		

Table 1 (cont'd)

Stabilizing agent	Amount (parts by weight)	Activity (%)	
		After storage at 4°C for 6 months	After storage at 37°C for 1 month
human serum albumin	2,000		
polyoxyethylene sorbitan monolaurate	100	99	96
mannitol	2,000		
human serum albumin	2,000		
hydroxypropyl cellulose	500	98	92
dextran (m.w. 40,000)	2,000		
polyoxyethylene sorbitan monolaurate	100		
sorbitol	2,000	98	96
polyoxyethylated hardened castor oil	100		
dextran (m.w. 40,000)	2,000	94	92
not added	-	74	58

Example 2

To 10 µg of G-CSF, one of the stabilizing agents listed in Table 2 was added and the mixture was aseptically dissolved in a 20 mM phosphate buffer solution (containing 5 100 mM sodium chloride; pH, 7.4) to make a pharmaceutical preparation containing 10 µg of G-CSF per ml. The preparation was aseptically charged into a sulfa-treated glass vial and sealed to make a G-CSF solution. The time-dependent change in the activity of G-CSF in this solution was 10 measured by the same method as used in Example 1 and the results are shown in Table 2.

Table 2

Stabilizing agent	Amount (parts by weight)	Activity (%)		
		After storage at 4°C for 7 days	After storage at 4°C for 2 months	After storage at RT for 1 month
mannitol	5,000	91	87	82
hyaluronic acid	2,000	93	87	70
dextran (m.w. 40,000)	2,000	96	95	85
glycerin	10,000	90	90	88
neuraminic acid	5,000	93	91	84
chitin	2,000	95	92	86
dextrin	2,000	90	92	87
human serum albumin	1,000	99	95	92
human serum globulin	1,000	98	94	90
acid-treated gelatin	2,000	97	96	87
alkali-treated gelatin	500	99	95	92
collagen	2,000	99	94	88
polyethylene glycol (m.w. 4,000)	10,000	94	89	90
hydroxypropyl cellulose	2,000	98	95	92
sodium carboxymethyl cellulose	2,000	92	91	80
hydroxyethyl cellulose	4,000	92	94	90
polyvinyl alcohol (m.w. 50,000)	4,000	97	93	90
polyvinylpyrrolidone (m.w. 50,000)	4,000	95	95	92
sorbitan monolaurate	400	97	96	95
polyoxyethylene sorbitan monolaurate	400	100	96	94

Table 2 (cont'd)

Stabilizing agent	Amount (parts by weight)	Activity (%)		
		After storage at 4°C for 7 days	After storage at 4°C for 2 months	After storage at RT for 1 month
polyoxyethylene sorbitan monostearate	400	98	97	94
polyoxyethylene polyoxypropylene glycol ether	400	100	94	93
polyoxyethylated hardened castor oil	400	99	98	90
sodium lauryl sulfate	2,000	97	93	87
lecithin	2,000	97	94	90
human serum albumin	2,000	100	99	97
mannitol	2,000			
cysteine	100			
human serum albumin	2,000	99	97	95
polyoxyethylene sorbitan monolaurate	100			
mannitol	2,000			
human serum albumin	1,000	99	97	95
hydroxypropyl cellulose	500			
dextran (m.w. 40,000)	2,000			
polyoxyethylene sorbitan monopalmitate	100	96	96	93
sorbitol	2,000			
polyoxyethylated hardened castor oil	100	95	92	92
dextran (m.w. 40,000)	2,000			
not added	-	72	61	47

Example 3

To 10 µg of G-CSF, one of the stabilizing agents listed in Table 3 was added and the mixture was aseptically dissolved in a 20 mM phosphate buffer solution (containing 5 100 mM sodium chloride; pH, 7.4) to make a pharmaceutical preparation containing 10 µg of G-CSF per ml. One milliliter of the preparation was charged into a sulfa-treated silicone-coated glass vial and left at 4°C. The effectiveness of each stabilizing agent in preventing G-CSF adsorption was evaluated by measuring the residual activity of 10 G-CSF in the solution after 0.5, 2 and 24 hours. The measurement was conducted by method (b) using reverse-phase high-performance liquid chromatography. The results are shown in Table 3.

Table 3

Stabilizing agent	Amount (parts by weight)	Residual activity (%)			
		initial	0.5 h	2 h	24 h
monnitrol	5,000	100	93	90	91
hyaluronic acid	2,000	100	97	92	92
dextran (m.w. 40,000)	2,000	100	98	95	96
glycerin	10,000	100	94	91	90
heparin	2,000	100	92	90	90
glucuronic acid	5,000	100	96	90	91
ketoglycolic acid	5,000	100	92	88	90
human serum albumin	1,000	100	100	101	99
human serum globulin	1,000	100	98	100	98
alkali-treated gelatin	500	100	99	98	99
acid-treated gelatin	2,000	100	99	97	97
collagen	2,000	100	100	98	99
polyethylene glycol (m.w. 4,000)	10,000	100	100	100	99
hydroxypropyl cellulose	2,000	100	100	100	99
sodium carboxymethyl cellulose	2,000	100	98	96	95
hydroxyethyl cellulose	4,000	100	96	93	92
polyvinyl alcohol (m.w. 50,000)	4,000	100	99	100	98
polyvinylpyrrolidone (m.w. 50,000)	4,000	100	98	98	96
sorbitan monocaprylate	400	100	100	100	98
polyoxyethylene sorbitan monostearate	400	100	100	98	100
polyoxyethylated hardened castor oil	400	100	99	101	99

Table 3 (cont'd)

Stabilizing agent	Amount (parts by weight)	Residual activity (%)			
		initial	0.5 h	2 h	24 h
sodium lauryl sulfate	2,000	100	100	99	97
lecithin	2,000	100	99	100	98
human serum albumin	2,000				
mannitol	2,000	100	100	100	101
cysteine	100				
human serum albumin	2,000				
polyoxyethylene	100	100	100	98	99
sorbitan monolaurate					
mannitol	2,000				
human serum albumin	1,000				
hydroxypropyl cellulose	500	100	101	99	100
dextran (m.w. 40,000)	2,000				
polyoxyethylene	100				
sorbitan monolaurate		100	100	99	99
sorbitol	2,000				
polyoxyethylated	100				
hardened castor oil		100	100	98	97
dextran (m.w. 40,000)	2,000				
not added	-	100	91	72	73



CLAIMS

1. A stable granulocyte colony stimulating factor containing pharmaceutical preparation that contains, in addition to the granulocyte colony stimulating factor present as the effective ingredient, at least one substance selected from pharmaceutically acceptable surfactants, saccharides, proteins and other high-molecular weight compounds such as natural polymers or synthetic polymers; wherein the granulocyte colony stimulating factor has the following physicochemical properties:

(i) molecular weight: about $19,000 \pm 1,000$ as measured by electrophoresis through a sodium dodecylsulfate-polyacrylamide gel;

(ii) isoelectric point: having at least one of the three isoelectric points, $pI = 5.5 \pm 0.1$, $pI = 5.8 \pm 0.1$, and $pI = 6.1 \pm 0.1$;

(iii) ultraviolet absorption: having a maximum absorption at 280 nm and a minimum absorption at 250 nm;

(iv) amino acid sequence of the 21 residues from N terminus:

H₂N-Thr-Pro-Leu-Gly-Pro-Ala-Ser-Ser-Leu-Pro-Gln-Ser-Phe-Leu-Leu-Lys-Cys-Leu-Glu-Gln-Val-.

2. A preparation according to claim 1 which contains the surfactant in an amount of 1 - 10,000 parts by weight per part by weight of the granulocyte colony stimulating factor.

3. A preparation according to claim 1 or 2 wherein said surfactant is a nonionic surfactant, an anionic surfactant or a natural surfactant; the nonionic surfactant being a sorbitan aliphatic acid ester, a glycerin aliphatic acid ester, a polyglycerin



aliphatic acid ester, a polyoxyethylene sorbitan
aliphatic acid ester, a polyoxyethylene sorbitol
aliphatic acid ester, a polyoxyethylene glycerin
aliphatic acid ester, a polyethylene glycol aliphatic
5 acid ester, a polyoxyethylene alkyl ether, a
polyoxyethylene polyoxypropylene alkyl ether, a
polyoxyethylene alkylphenyl ether, a polyoxyethylated
hardened castor oil, a polyoxyethylated beeswax
derivative, a polyoxyethylene lanolin derivative, or a
10 polyoxyethylene aliphatic acid amide; the anionic
surfactant being an alkyl sulfate salt, a
polyoxyethylene alkyl ether sulfate salt, or an alkyl
sulfosuccinate ester salt; and the natural surfactant
being lecithin, glycerophospholipid,
15 sphingophospholipid, or a sucrose aliphatic acid ester.

4. A preparation according to claim 1 which
contains the saccharide in an amount of 1 - 10,000
parts by weight per part by weight of the granulocyte
colony stimulating factor.

20 5. A preparation according to Claim 1 or 4
wherein said saccharide is at least one of glycerin,
erythritol, arabitol, xylitol, sorbitol, mannitol,
glucuronic acid, iduronic acid, galacturonic acid,
neuraminic acid, glyconic acid, mannuronic acid,
25 ketoglycolic acid, ketogalactonic acid, ketogulonic
acid, hyaluronic acid and salts thereof, chondroitin
sulfate and salts thereof, heparin, inulin, chitin and
derivatives thereof, chitosan and derivatives thereof,
dextrin, dextran with an average molecular weight of
30 5,000 - 150,000 and alginic acid and salts thereof.

6. A preparation according to claim 1 which
contains the protein in an amount of 1 - 20,000 parts
by weight per part by weight of the granulocyte colony
stimulating factor.

35 7. A preparation according to claim 1 or 6
wherein said protein is at least one of human serum



albumin, human serum globulin, gelatin, acid- or alkali-treated gelatin with an average molecular weight of 7,000 - 100,000, and collagen.

8. A preparation according to claim 1 which
5 contains the high-molecular weight compound in an amount of 1 - 20,000 parts by weight per part by weight of the granulocyte colony stimulating factor.

9. A preparation according to claim 1 or 8
10 wherein said high-molecular weight compound is at least one of hydroxypropyl cellulose, hydroxymethyl cellulose, sodium carboxymethyl cellulose, hydroxyethyl cellulose, polyethylene glycol with a molecular weight of 300 - 6,000, polyvinyl alcohol with a molecular weight of 20,000 - 100,000, and polyvinylpyrrolidone
15 with a molecular weight of 20,000 - 100,000.

10. A process for producing a stable granulocyte colony stimulating factor containing pharmaceutical preparation, which process comprises mixing granulocyte colony stimulating factor (G-CSF) with at least one
20 substance selected from pharmaceutically acceptable surfactants, saccharides, proteins and other high molecular weight substances such as natural polymers and synthetic polymers, dissolving the resulting mixture in a buffer solution; and charging a container
25 with the solution; wherein said G-CSF has the following properties:

(i) molecular weight: about $19,000 \pm 1,000$ as
measured by electrophoresis
through a sodium
30 dodecylsulfate-
polyacrylamide gel;

(ii) isoelectric point: having at least one of the
three isoelectric points,
 $pI = 5.5 \pm 0.1$, $pI = 5.8 \pm$
35 0.1 , and $pI = 6.1 \pm 0.1$;

(iii) ultraviolet absorption: having a maximum



-26-

absorption at 280 nm and a minimum
absorption at 250 nm;

(iv) amino acid sequence of the 21 residues from N
terminus:

5 H₂N-Thr-Pro-Leu-Gly-Pro-Ala-Ser-Ser-Leu-Pro-Gln-
 Ser-Phe-Leu-Leu-Lys-Cys-Leu-Glu-Gln-Val-.

11. A preparation according to claim 1,
substantially as hereinbefore described in any of the
10 Examples.

12. A process according to claim 10,
substantially as hereinbefore described in any of the
Examples.

13. A stable granulocyte colony stimulating
15 factor containing pharmaceutical preparation, when
produced by a process according to claim 10 or 12.
