

Integrated Epitaxial Metal ElectrodesCross Reference to Related Applications

[0001] This application is a continuation-in-part of and claims priority to co-pending and commonly-assigned U.S. nonprovisional application no. 16/178,495, filed November 1, 2018, which is a continuation of U.S. nonprovisional application no. 15/712,002 (issued as U.S. Patent No. 10,128,350 on November 13, 2018), which in turn claims the benefit under 35 U.S.C. § 119(e) of U.S. Provisional Application No. 62/398,416, filed on September 22, 2016.

Field of Use

[0002] This application relates to semiconductor designs, and more specifically, to a layered structure for integrated epitaxial metal electrodes, in which an epitaxial metal is introduced between a lower epitaxial oxide and upper epitaxial semiconductor.

Background

[0003] Epitaxy, epitaxial growth, and epitaxial deposition refer to growth or deposition of a crystalline layer on a crystalline substrate. The crystalline layer is referred to as an epitaxial layer. The crystalline substrate acts as a template and determines the orientation and lattice spacing of the crystalline layer. The crystalline layer can be, in some examples, lattice matched or lattice coincident. A lattice matched crystalline layer can have the same or a very similar lattice spacing as the top surface of the crystalline substrate. A lattice coincident crystalline layer can have a lattice spacing that is an integer multiple of the lattice spacing of the crystalline substrate. The quality of the epitaxy is based in part on the degree of crystallinity of the crystalline layer. Practically, a high quality epitaxial layer will be a single crystal with minimal defects and few or no grain boundaries. Traditionally, metal contact layers are applied to an epitaxial structure at some point in the upstream processing. With today's complex epitaxial structures often incorporating more than one device functionality, this can require extensive etching and deposition of metals on wafers with a large amount of topography.

[0004] Interactions between metals and semiconductors are often critical to device operation. One example of such an interaction between a metal and a semiconductor occurs in a thin film resonator such as an RF filter where the overall acoustic performance is defined by the product of the acoustic impedance of the electrode and the acoustic impedance of the

piezoelectric material. In fact, to access high resonant frequencies it is essential to make both the electrode and the piezoelectric material quite thin. This is summarized in FIG. 17, which shows resonant frequencies as a function of AlN thickness for different thickness metal electrodes (from S. Tanifuji et al, Proceedings 2009 IEEE International Ultrasonic Symposium, p. 2170, the entirety of which is incorporated by reference). Here, crystal quality is also important because without it resistivity would increase as thickness decrease due to an increasing effect of defects and grain boundaries in polycrystalline metal layers.

[0005] Growth of InP has also been attempted on metal over silicon engineered substrates, as described in Zheng et al, Journal of Applied Physics, vol. 111 p. 123112 (2012), the entirety of which is incorporated by reference. However, Zheng describes films that are polycrystalline, not epitaxial.

[0006] Epitaxial growth of metals on yttria stabilized zirconia (YSZ) is described in Gsell et al, Journal of Crystal Growth, vol. 311, p. 3731 (2009), the entirety of which is incorporated by reference. Gsell describes separating the metal from the underlying silicon substrate by using YSZ as this prevents the unwanted siliciding of any epitaxial metal. YSZ is a sputtered material (or deposited with pulsed laser deposition) using zirconia and yttria targets. It is not a single crystal material, has grain boundaries, and can be of mixed crystallinity (cubic and tetragonal). Thus, it is a suboptimal template for epitaxial growth of metals. In addition, control of the YSZ/silicon interface is technically challenging.

[0007] Accordingly, epitaxially growing metal of good crystal quality over semiconductor materials has proven to be difficult.

Summary

[0008] Systems and methods are described herein for the use of integrated epitaxial metal electrodes in layered structures over which semiconductor layers may be grown. Systems and methods described herein may include a layered structure, comprising a substrate, a first rare earth oxide layer epitaxially grown over the substrate, a first metal layer epitaxially grown over the rare earth oxide (REO) layer, and a first semiconductor layer epitaxially grown over the first metal layer. In some embodiments, the substrate includes one or more group IV element including but not limited to silicon (Si), germanium (Ge), silicon on insulator (SOI), SiGe. In some embodiments, the substrate has a crystal orientation of either $\langle 100 \rangle$ or $\langle 111 \rangle$ with a miscut of upto 10 degrees. In some embodiments, the substrate includes elements from group III and group V including but not limited to GaAs, InP, GaN.

In some embodiments, the substrate is another metal oxide, including but not limited to Ga_2O_3 , Al_2O_3 .

[0009] In some embodiments, rare earth oxide layer includes a rare earth metal element selected from a lanthanide group of a periodic table, scandium (Sc) and yttrium (Y). In some embodiments, the REO layer is composed of an REO having an oxygen-to-metal ratio between 1 and 2. In some embodiments, the first metal layer includes a metal element selected from a transition metal group of a periodic table. In some embodiments, the first semiconductor layer includes an element selected from group III, group IV, group V. In some embodiments, the substrate is composed of silicon, the REO layer is composed of erbium oxide having an oxygen-to-metal ratio of 1.5 ($\text{ErO}_{1.5}$), and the first metal layer is composed of molybdenum (Mo). In some embodiments, the first semiconductor layer is composed of $\text{Al}_x\text{Sc}_{1-x}\text{N}$ ($0 \leq x < 1$). In some embodiments, the substrate, when composed of Si, has a crystal orientation of $\langle 100 \rangle$, the REO layer, when composed of $\text{ErO}_{1.5}$, has a crystal orientation of $\langle 110 \rangle$, and the first metal layer, when composed of Mo, has a crystal orientation of $\langle 211 \rangle$. For example, the substrate that is composed of silicon may have an orientation of $\langle 111 \rangle$, the REO layer, when composed of $\text{ErO}_{1.5}$ has a crystal orientation of $\langle 110 \rangle$. In some embodiments, the REO layer is composed of multiple rare metal oxide components, and the multiple rare metal oxide components have different metal elements or different oxygen-to-metal ratios.

[0010] In some embodiments, the REO layer includes a first sublayer composed of a first REO and a second sublayer composed of a second REO. In some embodiments, the REO layer includes a first region composed of a first REO and a second region composed of a second REO, and wherein the first region transits to the second region in a graded pattern. In some embodiments, the REO layer includes a first sublayer composed of a first REO and a second sublayer composed of a second REO, and wherein the first sublayer and the second sublayer are repeated in a super lattice structure. In some embodiments, the second metal oxide further comprises a group III element. In some embodiments, the first metal layer includes a first sublayer composed of a first metal and a second sublayer composed of a second metal. In some embodiments, the first metal layer includes a first region composed of a first metal and a second region composed of a second metal, and wherein the first region transits to the second region in a graded pattern. In some embodiments, metal layer includes a first sublayer composed of a first metal and a second sublayer composed of a second metal, and wherein the first sublayer and the second sublayer are repeated in a super lattice

structure. In some embodiments, layered structure further comprises a second metal layer epitaxially grown over the semiconductor layer.

[0011] In some embodiments, the layered structure further comprises a second semiconductor layer epitaxially grown over the second metal layer. In some embodiments, the layered structure further comprises up to 20 repetitions of a combination of a metal layer and a semiconductor layer. In some embodiments, the layered structure further comprises repetitions of a combination of a metal layer and a REO layer. In some embodiments, the layered structure further comprises a second REO layer grown over the semiconductor layer. In some embodiments, a second metal layer epitaxially grown over the second REO layer. In some embodiments, the layered structure of claim 1, further comprising an epitaxial layer grown from the first metal layer, wherein the epitaxial layer includes a component selected from a group of a two-dimensional (2D) material, a cap layer, and an insulator. In some embodiments, the 2D material is selected from a group of graphene and transition metal disulfide. In some embodiments, the cap layer is composed of a material selected from a group of metal oxides and metal silicides. In some embodiments, the insulator is composed of a REO. In some embodiments, an interlayer that transits from the first metal layer to the first semiconductor layer. In some embodiments, the interlayer is composed of one or more components selected from a group of metal nitride, metal pnictide and a template 2D electrode.

[0012] In some embodiments, an interlayer that transits from the first REO layer to the first metal layer. In some embodiments, the interlayer is grown with a metal component from the first metal layer and oxygen. In some embodiments, the first metal layer has a non-continuous pattern with a first gap space between a first part of the first metal layer and a second part of the first metal layer, and the first semiconductor layer is grown over both the gap and the metal region.

Brief Description of Drawings

[0013] Further features of the disclosure, its nature and various advantages will be apparent upon consideration of the following detailed description, taken in conjunction with the accompanying drawings, in which:

[0014] FIG. 1 depicts a layered structure to grow an epitaxial metal layer between a substrate and a semiconductor layer, according to an illustrative embodiment;

[0015] FIG. 2 depicts a flow diagram to grow the layered structure shown in FIG. 1, according to an illustrative embodiment;

[0016] FIGS. 3-16 depict various examples of layered structures, each of which is a specific example of the layered structure shown in FIG. 1, according to an illustrative embodiment;

[0017] FIG. 17 depicts resonant frequencies as a function of AlN thickness for different thickness metal electrodes, in accordance with an embodiment of the prior art;

[0018] FIG. 18, depicts calculated Distributed Bragg Reflector (DBR) reflectivity with and without an epitaxial metal electrode, where the addition of a crystalline REO and a metal beneath a standard III-nitride DBR increases peak reflectivity by 2%;

[0019] FIG. 19 depicts a graph showing how DBR is constructed from 11 periods of AlN and GaN, according to an illustrative embodiment;

[0020] FIG. 20 depicts a calculated reflectivity at 450nm that is plotted for both for AlN-GaN DBR and AlN-GaN DBR that is constructed on single pair of AlN over Mo, according to an illustrative embodiment;

[0021] FIG. 21 depicts a flowchart of a method for growing the layered structure shown in FIG. 1, according to an illustrative embodiment;

[0022] FIGS. 22A-C provide example diagrams illustrating layered structures of metal electrodes build upon a silicon substrate containing a porous silicon portion, according to an embodiment described herein;

[0023] FIGS. 23A-B provide example diagrams showing layered structures built upon the layered structures shown in FIGS. 22A-C, with the rare earth oxide layer to support the metal layer, according to an embodiment described herein;

[0024] FIGS. 24A-B and 25A-B provide various example diagrams showing layered structures using non-continuous rare earth oxide regions to define the location of the porous portion in the substrate, according to an embodiment described herein;

[0025] FIG. 26 provides an example diagram illustrating a layered structure having the metal layer defining the boundary of the porous portion of the substrate with a continuous rare earth oxide layer in between, according to an embodiment described herein; and

[0026] FIG. 27 provides an example diagram illustrating a layered structure having multiple semiconductors, according to an embodiment described herein. For example, multiple semiconductors may be used in any of the layered structures described herein.

Detailed Description

[0027] Structures and methods described herein provide an integrated epitaxial metal electrode that incorporates an epitaxial metal within an epitaxial stack thereby

incorporating a buried contact layer. The structures and methods disclosed herein include high quality epitaxial metal layers and the ability to continue growth of a semiconductor material above the epitaxial metal layer. In one example, a crystalline REO layer may be epitaxially grown over a substrate or semiconductor, and a metal layer may be epitaxially grown over the crystalline REO layer. A semiconductor layer may be grown over the epitaxial metal layer. An REO layer is a layer that contains one or more rare earth (RE) species and oxygen. The rare earth species include Lanthanum (La), Cerium (Ce), Praseodymium (Pr), Neodymium (Nd), Promethium (Pm), Samarium (Sm), Europium (Eu), Gadolinium (Gd), Terbium (Tb), Dysprosium (Dy), Holmium (Ho), Erbium (Er), Thulium (Tm), Ytterbium (Yb), Lutetium (Lu), Scandium (Sc) and Yttrium (Y).

[0028] REOs are known to exhibit fluorite-type structures. These structures exhibit morphology differences as a function of the atomic weight of the rare-earth cation present in the oxide, among any other factors.

[0029] In particular, oxides comprising lighter rare-earths form cubic CaF_2 -type crystal structure as a result of possible ionization states of +2 and/or +3 and/or +4. Oxides having this crystal structure exhibit significant net charge defect due to a multiplicity of possible oxidation states (for rare-earth oxides).

[0030] On the other hand, oxides formed from heavier rare-earths (e.g., RE_2O_3 , etc.), exhibit a distorted CaF_2 -type crystal structure which includes anion vacancies due to an ionization state of RE^{3+} . The crystal structure associated with rare-earth oxides of heavier rare earths is also known as “Bixbyite.”

[0031] An illustrative example of a rare-earth oxide having the formula RE_2O_3 , is Er_2O_3 . The crystal structure of a unit cell of Er_2O_3 is an oxygen-vacancy-derived fluorite derivative (i.e., Bixbyite structure). REO dielectric layers may comprise an assemblage of these unit cells.

[0032] The number and position of the anion vacancies determines the crystal shape of the RE_2O_3 unit cell. The crystal shape of this cell may be engineered to provide a suitable match to the lattice constant of the underlying semiconductor substrate. Oxygen vacancies along the body diagonal and/or the face diagonal lead to a C-type cubic structure. For example, two anion vacancies per fluorite unit cell causes the unit cell of Er_2O_3 to increase to nearly twice the unit cell size of Si. This, in turn, enables low-strain, single-phase Er_2O_3 to be epitaxially grown directly on a silicon substrate.

[0033] Furthermore, the number and position of the anion vacancies may be engineered to induce a desired strain (tensile or compressive) in the dielectric layer and/or overgrown

layers. For example, in some embodiments, strain in the semiconductor layer is desired in order to affect carrier mobility.

[0034] Each fluorite unit cell has two oxygen vacancies, which lie along the body diagonal. The presence of these two oxygen vacancies causes the Er_2O_3 unit cell to double in size, thereby doubling its lattice constant, which provides a suitable match to the lattice constant of $\langle 100 \rangle$ silicon.

[0035] In some examples, oxygen vacancies lie at the ends of the face diagonal. In some other examples, oxygen vacancies are distributed between the ends of the face diagonal and the body diagonal.

[0036] A buried metal contact layer may be grown using epitaxial deposition of metal over a semiconductor layer. The epitaxial metal layer may be grown directly on the semiconductor layer and/or directly on a substrate. In some examples, an optional transitional layer may be between the epitaxial metal layer and the underlying semiconductor layer, and/or between the epitaxial metal layer and the underlying substrate. As well as the electrical advantages a buried contact layer would bring, there are often interactions between a metal and the overlying semiconductor that may be utilized. These interactions, such as in RF filters, are more useful when the interface between the metal and semiconductor (and any intervening interface) is high quality with few defects. In addition, epitaxial metals may be made thinner than sputtered metals while preserving high film quality. This is in part because epitaxial interfaces are higher quality, and as layers are thinned, the interfaces become a larger proportion of the overall material. Thus, while a thick film is less affected by poor quality interfaces and its properties are dominated by the bulk material properties, the properties of a thin film are more dominated by the interfacial properties. Thus, high quality interfaces important when depositing thin films.

[0037] In addition, an epitaxial metal layer may be used to modify the reflectivity of an epitaxial stack of layers. For devices where light emission is from the top surface, light that is emitted towards the substrate is generally considered to be lost to the overall output power. In vertical cavity surface emitting lasers (VCSELs), for example, the back mirror has to have a reflectivity $> 99.8\%$. This is difficult to achieve solely through semiconductor materials.

[0038] FIG. 18, shows a graph depicting calculated DBR reflectivity with and without an epitaxial metal electrode, where the addition of a crystalline rare earth oxide and a metal beneath a standard III-nitride DBR increases peak reflectivity by 2%. A III-nitride material is a material comprising a Group III species and nitrogen. The Group III species may include one or more elements in Group III of the Periodic Table, including B, Al, Ga, In, and Tl. The

III-nitride layer may be a compound that includes multiple Group III elements. The III-nitride layer may include binary compounds such as GaN, ternary compounds such as $\text{Al}_x\text{Ga}_{1-x}\text{N}$ ($0 \leq x \leq 1$) and $\text{In}_x\text{Ga}_{1-x}\text{N}$ ($0 \leq x \leq 1$), quaternary compounds such as $\text{In}_x\text{Al}_y\text{Ga}_{1-x-y}\text{N}$ ($0 \leq x, y \leq 1$), and quinary compounds such as $\text{Ga}_x\text{In}_y\text{Al}_z\text{Sb}_{1-x-y-z}\text{N}$ ($0 \leq x, y, z \leq 1$). The III-nitride layer may be undoped, unintentionally doped, or doped with donor or acceptor dopants.

[0039] A crystalline rare earth oxide (REO) epitaxial layer may be used as a template for epitaxial metal on semiconducting substrates such as silicon. Substrates other than silicon may be used, and examples include germanium, Si-Ge alloys, sapphire, silicon dioxide, silicon-on-insulator (SOI), and silicon-on-semiconductor (SOS), substrates with a top layer of one of the above, and any semiconducting substrate. For the purpose of metal epitaxy, crystalline REO is a superior material compared to YSZ. For a start, the interface between the crystalline REO and the substrate is set as part of the epitaxial process. With the appropriate choice of rare earth oxide, crystalline REO templates may be epitaxially grown that are 100% (or nearly 100%) cubic with no secondary phase. Other parameters and process characteristics of crystalline REO that are beneficial to the overall epitaxial stacks are an oxide-silicon interface that is free of any parasitic charge, a higher density than YSZ (8.6 to 6.1 g/cm³), and a 5x better thermal conductivity than YSZ. In addition to serving as a template for epitaxial metal growth, the crystalline REO layer may also prevent interdiffusion between the epitaxial metal layer and any substrate below. This prevents the formation of, for example, unwanted metal silicides (where the substrate is silicon).

[0040] FIG. 1 shows an example diagram illustrating a layered structure 400 according to an illustrative embodiment. Structure 100 includes a substrate 102, an REO layer 104, grown over the substrate 102, a metal layer 106 grown epitaxially over the REO layer 104, and a semiconductor layer 108 grown epitaxially over the metal layer 106. The thickness of the REO layer 104 is defined as t_{ox} where typically the thickness of the oxide may be defined as $0 \leq t_{ox} \leq 500\text{nm}$. The layer structure 100 depicted in FIG. 1 may be manufactured in a single epitaxial process, either molecular beam epitaxy (MBE), metalorganic vapor phase epitaxy (MOCVD) or any of the other well-known epitaxial deposition techniques. As required by the process, a deposition tool to deposit the material may either be single chamber, or use any of the well-known cluster tool formats where specific parts of the process are done in different interconnected chambers, or multiple deposition tools may be used. The crystalline REO layer 104 is a template for the epitaxial metal layer 106, which may comprise one or more constituent epitaxial metal layers. The semiconductor layer 108

may comprise one or more of a III-nitride material, a III-V material, and a Group IV material. III-V materials include one or more species from Group III of the Periodic Table (such as B, Al, Ga, In, and Tl) and one or more species from Group V of the Periodic Table (such as N, P, As, Sb, and Bi). III-nitrides are III-V materials and include a species from Group III and nitrogen. Examples of III-nitride materials include GaN, $\text{In}_x\text{Al}_y\text{Ga}_{1-x-y}\text{N}$ ($0 \leq x, y \leq 1$), and/or AlN. Examples of other III-V materials include one or more of GaAs, InP, InAs, InSb, InGaAs, GaAsP, InGaAsP, and the like. In some embodiments, the oxygen-to-metal ratio for the REO layer 104, ranges from 1 to 2. In some embodiments, the oxygen to metal ratio for the REO layer 104 may be between 1.4 to 1.6.

[0041] The layer structure 100 of FIG. 1 may be grown over a substrate 102 such as a silicon substrate. If the semiconductor material in a filter is epitaxial, then it lends itself to integration of additional semiconductor elements (not necessarily directly electrically connected to the filter) that may be grown above the filter. For example, a transistor (examples of which include a field effect transistor, a high electron mobility transistor, and a heterojunction bipolar transistor) may be grown over the filter, thus reducing the chip area required for a given system.

[0042] FIG. 2 depicts a process schematic 200 that illustrates a single epitaxial process for fabricating the structure depicted in FIG. 1. A crystalline REO layer 104 is epitaxially grown over a substrate 102. A metal layer 106 is epitaxially grown over the crystalline REO layer 104. A semiconductor layer 108 is epitaxially grown over the metal layer 106. In some embodiments, an additional metal layer 210 may be epitaxially grown over the semiconductor layer 108. Each of the layers depicted in FIG. 2 may include one or more sub-layers. The composition of each layer is described in further detail in FIGS. 3 – 13.

[0043] The epitaxial metal used could be a rare earth metal or a metal such as ruthenium or molybdenum, or other representative metals listed in Table 1 below. Attributes to consider for selecting the metal element for the epitaxial metal layer 210 include resistivity, and also density, Young's modulus and refractive index which determine optical and acoustic properties of the layer. Other metals not listed in Table 1 may also be used.

Table 1. Example Metals for the Metal Layer

Metal	Crystal Structure	Resistivity (nΩ-m)	Density (g/cm ³)	Young's modulus (GPa)	Refractive index @ 635 nm
Ruthenium	hcp	71	12.5	447	
Molybdenum	bcc	53	10.3	329	3.71

Platinum	fcc	105	21.5	168	2.33
Copper	fcc	17	8.9	119	0.23
Aluminum	fcc	28	2.7	70	1.39
Neodymium	fcc	643	7.0	41	
Gadolinium	dhcp	1310	7.9	55	
Erbium	hcp	860	9.1	70	
Ytterbium	fcc	250	6.9	24	
Scandium	hcp	562	2.9	74	

Table 1: Representative metals and selected properties

(hcp – hexagonal close packed, fcc – face centered cubic, bcc – body centered cubic, dhcp – double hexagonal close packed)

[0044] For example, the substrate 102 may be composed of silicon; the REO layer 104 may be composed of erbium oxide having an oxygen-to-metal ratio of 1.5 ($\text{ErO}_{1.5}$), and the first metal layer 106 may be composed of molybdenum (Mo). The first semiconductor layer may be composed of $\text{Al}_x\text{Sc}_{1-x}\text{N}$ ($0 \leq x < 1$). The substrate 102, when composed of Si, may have a crystal orientation of $\langle 111 \rangle$, and the first metal layer, when composed of Mo, has a crystal orientation of $\langle 110 \rangle$. In this example, the oxygen-to-metal ratio in the REO may have a range between 1.4 and 1.6.

[0045] For another example, the substrate 102 that is composed of silicon may have an orientation of $\langle 100 \rangle$, the REO layer 104, when composed of $\text{ErO}_{1.5}$ may have a crystal orientation of $\langle 110 \rangle$, and the first metal layer 106, when composed of Mo, may have a crystal orientation of $\langle 211 \rangle$. In this example, the oxygen-to-metal ratio in the REO may have a range between 1.4 and 1.6, and the semiconductor layer 108 may have a variety of compositions.

[0046] FIGS. 3 – 5 depict structures including multiple epitaxial metal layers over a crystalline REO layer 104. The epitaxial metal layers in FIGS. 3 – 5 may include multiple metal layers. Multiple metal layers may be grown either as a stacked as in FIGS. 3-4, e.g., a step change from one type of metal to another type of metal, or as graded change as shown in FIG. 5. For illustrative purpose only, two types of metal layers are depicted in FIGS. 3-5, but more than two types of metal layers may be used in the structure in a similar manner as shown in FIGS. 3-5.

[0047] FIG. 3 shows an example diagram illustrating a layered structure 300 according to an illustrative embodiment. Structure 300 includes a first metal layer 302 epitaxially grown

over REO layer 104, a second metal layer 304 epitaxially grown over the first metal layer 302 in a stepwise type configuration. In some embodiments, the metal in the first metal layer 302 and the second metal layer 304 may be same. In some embodiments, the metals in the first layer 302 and the second metal layer 304 may be different. In some embodiments, the first metal layer 302 and the second metal layer 304 may have the same thickness. In some embodiments, the first metal layer 302 may have a thickness different from the second metal layer 304.

[0048] FIG. 4 shows an example diagram illustrating a layered structure 400 according to an illustrative embodiment. Structure 400 includes multiple alternating layers of a first metal (layers 402, 406) and a second metal (layers 404, 408) epitaxially grown over REO layer 104 in a superlattice type configuration. In some embodiments, the metals in the first metal layers 402, 406 and the second metal layers 404, 408 may be the same. In some embodiments, the metal in the first layers 402, 406 and the metal in the second layers 404 and 406 may be different. In some embodiments, the first metal layers 402, 406 and the second metal layers 404, 408 may have the same thickness. In some embodiments, the first metal layers 402, 406 may have a thickness different from the second metal layers 404 and 408. It is to be noted that two repetitions of two different types of metal layers are shown in FIG. 4 for illustrative purpose only, and a different number of repetitions (e.g., three, four, five, etc.) may be used in the structure.

[0049] FIG. 5 shows an example diagram illustrating a layered structure 500 according to an illustrative embodiment. Structure 500 includes a metal layer 106 epitaxially grown over REO layer 104, where the metal layer 106 has a first region 502 and a second region 504 where a first concentration of a first metal and a second concentration of a second metal is modified in a graded configuration. The grading of the first and second concentrations of the first and second metal in FIG. 5 may be linear (e.g., a linear change in composition from the first metal to the second metal), superlinear (e.g., a higher order polynomial), sublinear, or stepwise (e.g., discrete changes in material composition). In some embodiments, the first concentration may have a first value in the first region 502 and a second value in a second region 504. The concentration of the first metal may vary across the thickness of layer 106. Similarly, the second concentration of the second metal may have a third value in the first region 502 and a fourth value in the second region 504. The concentration of the second metal may vary across the thickness of layer 106.

[0050] The layer structures 300 – 500 depicted in FIGS. 3 – 5 may be included in a radio-frequency (RF) filter. The semiconductor layer 108 grown over the epitaxial metal layer 106

may be a piezoelectric material that acts as a coupled electro-mechanical resonator. The first epitaxial metal layer (302, 402) may be a first electrode for the RF filter, and the second metal layer (304, 404) may be a second electrode for the RF filter. Epitaxial metal layers are particularly useful for electrodes in RF filters because they provide the high conductivity of metal with single-crystal structures that serve as templates for the subsequent growth of single-crystal layers (e.g., the semiconductor layer) over the metal layers. Single-crystal semiconductor layers are useful as the semiconductor material in RF filters because they provide higher piezoelectric coefficients, narrower bandwidths, and lower losses. In part the increased performance is due to quality and crystalline registry of the epitaxial metal electrodes, which results in higher quality of subsequent films.

[0051] Any of the structures depicted in FIGS. 3-5 may be included in optical devices. One such application would be in a distributed Bragg reflector (DBR). For a DBR, one critical consideration is the index of refraction between the constituent layers. The more dissimilar the index of refraction, the fewer periods required and the wider the stop band is. This offers a route to reducing total layer stack thickness and thereby reduction in manufacturing cost / complexity. For example in a DBR employing the semiconductors AlN and GaN the delta in refractive index at 420nm is 0.34. If the two materials were changed to AlN over epitaxial Mo this difference would increase to 0.85.

[0052] In many photonic devices all the light does not travel normal to the epitaxial surface. When the DBRs is constructed from semiconductors offering only a small difference in refractive index there is a strong dependence of reflectivity on the angle of incidence. Examples of performance of reflectivity of layered structure 100 in comparison to a layered structure 100 without the epitaxial metal layer 106 are shown in FIGS. 18-20.

[0053] FIGS. 6 – 8 depict structures including multiple crystalline REO layer layers over a substrate 102. The REO layer layer 106 in FIGS. 6 and 7 may include multiple REO layer layers. Multiple rare earth oxide layers may be grown either as a stacked as in FIGS. 6-7, e.g., a step change from one type of REO to another type of REO, or a graded change as shown in FIG. 8. In some embodiments, there may be instances where there is a first optimum REO to be placed adjacent to the substrate 102 and a second optimum REO to be placed to support epitaxy of metal layer 106. For illustrative purpose only, two types of REO layers are depicted in FIGS. 6-8, but more than two types of REO layers may be used in the structure in a similar manner as shown in FIGS. 6-8.

[0054] FIG. 6 depicts a layer structure 600 that includes a first REO layer 602 epitaxially grown over substrate 102, a rare earth oxide layer 604 epitaxially grown over the first rare

earth oxide layer 602 in a stepwise type configuration. In some embodiments, the rare earth metal in the first rare earth oxide layer 602 and the second rare earth oxide layer 604 may be same. In some embodiments, the rare earth metals in the first rare earth oxide layer 602 and the second rare earth oxide layer 604 may be different. In some embodiments, the first REO layer 602 and the second REO layer 604 may have a same thickness. In some embodiments, the first REO layer 602 may have a thickness different from the second REO layer 604. In some embodiments, the first rare earth metal may have a first concentration in the first layer 602 and a second concentration in the second layer 604. Similarly, the second rare earth metal may have a third concentration in the first layer 602 and a fourth concentration in the second layer 604. In some embodiments, the concentration of oxygen may be different in the first layer 602 and the second layer 604.

[0055] FIG. 7 depicts a layer structure 700 that includes multiple alternating layers of a first REO (layers 702, 706) and a second REO (layers 704, 708) epitaxially grown over substrate 102 in a superlattice type configuration. In some embodiments, the rare earth metals in the first REO layers 702, 706 and the second rare earth metal oxide layers 704, 708 may be same. In some embodiments, the rare earth metals in the first layers 702, 706 and the rare earth metal in the second layers 704 and 706 may be different. In some embodiments, the first REO layers 702, 706 and the second REO layers 704, 708 may have the same thickness. In some embodiments, the first REO layers 702, 706 may have a thickness different from the second REO layers 704 and 708. In some embodiments, the first rare earth metal may have a first concentration in the layer 702 and a second concentration in the layer 704. Similarly, the second rare earth metal may have a third concentration in the layer 702 and a fourth concentration in the layer 704. In some embodiments, the concentration of oxygen may be different in layer 702 and the layer 704. It is to be noted that two repetitions of two different types of REO layers are shown in FIG. 7 for illustrative purpose only, and a different number of repetitions (e.g., three, four, five, etc.) may be used in the structure.

[0056] FIG. 8 shows an example diagram illustrating a layered structure 800 according to an illustrative embodiment. Structure 800 includes a REO layer 104 epitaxially grown over substrate 102, where the REO layer 106 has a first region 802 and a second region 804 where a first concentration of a first rare earth metal and a second concentration of a second rare earth metal is modified in a graded configuration. The grading of the first and second concentrations of the first and second rare earth metal in FIG. 8 may be linear (e.g., a linear change in composition from the first metal to the second metal), superlinear (e.g., a higher order polynomial), sublinear, or stepwise (e.g., discrete changes in material composition). In

some embodiments, the first concentration of the first rare earth metal may have a first value in the first region 802 and a second value in a second region 804. The concentration of the first rare earth metal may vary across the thickness of layer 106. Similarly, the second concentration of the second metal may have a third value in the first region 802 and a fourth value in the second region 804. The concentration of the second metal may vary across the thickness of layer 106.

[0057] FIG. 9 depicts shows an example diagram illustrating a layered structure 900 according to an illustrative embodiment. Structure 800 depicts an example of the structure shown in FIG. 1, where the semiconductor layer 108 is a III-nitride layer, in particular an $\text{Al}_{1-x}\text{Sc}_x\text{N}$ ($0 \leq x \leq 1$) layer, the metal layer 106 is a Mo layer, the REO layer 104 is an Er_2O_3 layer, and the substrate 102 is a Si <111> substrate. Other examples of the structure shown in FIG. 9 are possible, and each of the layers may include one or more sub-layers as described in FIGS. 3 – 8.

[0058] In some embodiments, the layered structure 100 as shown in FIG. 1 may be modified to include an interlayer either between the epitaxial metal layer 106 and semiconductor 108 or between the REO layer 104 and the epitaxial metal layer 106. The purpose of such a layer is to allow chemical or crystallographic engineering of the transition from oxide to metal or metal to semiconductor. Chemical engineering may include encouraging nucleation or migration of the semiconductor or metal atoms during initial epitaxial deposition of the semiconductor or metal layer. Crystallographic engineering may include aiding in a transition in crystal structure or lattice constant between the metal and semiconductor layers. An example of a transition in crystal structure is a transition from a hexagonal-type crystal structure to a cubic-type crystal structure.

[0059] FIG. 10 shows an example diagram illustrating a layered structure 1000 according to an illustrative embodiment. Structure 1000 depicts an epitaxial metal 106 over an interlayer 1002 which is epitaxially grown over the crystalline REO layer 104. In some embodiments, the interlayer 1002 may be a metal oxide 1004 made with a combination of the metal in the epitaxial metal layer 106 and oxygen..

[0060] FIG. 11 shows an example diagram illustrating a layered structure 1100 according to an illustrative embodiment. Structure 1100 depicts an epitaxial metal layer 106 over a REO layer 104, an epitaxial interlayer 1102 over the epitaxial metal layer, and an epitaxial semiconductor layer 106 over the interlayer 1102. In some embodiments, the interlayer may be composed of a metal silicide. In some embodiments, the interlayer may be composed of metal nitride 1104. In some embodiments, the interlayer 1102 may be composed of rare earth

pnictides 1106 that commonly include rare earth nitride, rare earth arsenide, and rare earth phosphide. In some embodiments, the interlayer 1102 may be composed of a two-dimensional (2D) electrode 1108.

[0061] In some embodiments, more semiconductors of different composition / types might be epitaxially grown over the other semiconductor layer 108. In some embodiments, a second metal may be grown over the semiconductor layer. For this embodiment any of the previously described metal epitaxy schemes may be utilized, and any of the interlayers previously described that were epitaxially grown between the metal and the semiconductor could be used for the overall epitaxial process depending on what features were required of the final epitaxial stack. The layers above the semiconductor do not have to match those below the semiconductor. For example, the layers above the semiconductor may be the same or different from layers below the semiconductor

[0062] In some embodiments, an epitaxial metal layer may be grown over a semiconductor layer 108. In some embodiments three possible epitaxial interlayers, a metal silicide, a metal nitride, and a rare earth pnictide may be grown between the semiconductor layer 108 and the epitaxial metal layer. If the choice was made to grow a n epitaxial metal layer over semiconductor 108, then any or all of the above examples may be repeated for the purpose of epitaxially growing another semiconductor layer over metal.

[0063] FIG. 12 shows an example diagram illustrating a device composed of units of layered structures 1202 and 1204 according to an illustrative embodiment. Structure 1200 depicts an example of repeated metal/semiconductor structures with optional interlayers. FIG. 12 depicts a layer stack of three units 1204. Layer stacks may contain other numbers of units, but three are shown here for illustrative purposes. Each unit may be the same, or one or more of the units in a layer stack may be different. Layered structure 1202 depicts an exemplary unit within the layer stack 1204. This exemplary unit contains a first interlayer epitaxially grown over a first epitaxial metal layer, a semiconductor layer 108 epitaxially grown over the first interlayer, a second interlayer epitaxially grown over the semiconductor layer, and a second epitaxial metal layer epitaxially grown over the second interlayer. Any of the units within a layer stack may include none, one, or both of the first and second interlayers. In addition, the second epitaxial metal layer in one unit may be the same as the first epitaxial metal layer in the unit above. One or both of the epitaxial metal layers in a unit may be a single metal, a graded metal layer, a metal layer with multiple sub-layers, and/or a superlattice with multiple metal layers. Layer stacks such as those depicted in 1204 may be

used in photonic applications. For example, a layer stack may be a metal-semiconductor mirror, such as a DBR.

[0064] FIG. 13 shows an example diagram illustrating layered structures 1302, 1304, 1306 according to an illustrative embodiment. Structures 1302, 1304, and 1306 depict examples of final epitaxial layers to match the layers below to ex-situ processing and/or device operation. These include but are not limited to the use of a metal silicide to protect the upper metal layer from oxidation as shown in 1302, the addition of graphene or other 2D structures to enhance conductivity as shown in 1304, and the addition of a crystalline REO layer either as a dielectric or an insulator to electrically isolate the underlying epitaxial stack as shown in 1306. In some embodiments, a second epitaxial metal layer may be grown over the REO layer as grown over the semiconductor 108. Note although these three uppermost layers are shown as single layer entities it is expected that provision of such layers may require additional layers not shown here.

[0065] FIG. 14 shows an example diagram illustrating modifications to layered structures 100 according to an illustrative embodiment. Structure 1402 depicts a second epitaxial metal layer 1404 over semiconductor layer 108. Structure 1406 depicts a second semiconductor layer 1408 grown over the second epitaxial metal layer 1404. In some embodiments, a combination of the second epitaxial metal layer 1404 and the second semiconductor layer 1408 may be a mirror. The layer above the second epitaxial metal layer 1404 may be used as a template for a next phase of epitaxy to deliver additional functionality. Growth of an oxide may electrically isolate portion 1406 from layer structure 100 over which portion 1406 may be grown.

[0066] FIG. 15 shows an example diagram illustrating layered structure 1500 according to an illustrative embodiment. Structure 1500 depicts a repetition pattern for a combination 1502 of REO layer 104 and epitaxial metal layer 106 multiple times to build a stack 1504 before semiconductor 108 is grown over the stack. In some embodiments, the portion 1502 may be 1, 2, 3...20...or any other number of times before growing semiconductor 108 over the stack 1504.

[0067] FIG. 16 shows an example diagram illustrating layered structure 1602 and 1606 according to an illustrative embodiment. Structure 1602 depicts an epitaxial metal layer 106 that is segmented either by incorporating a mask within the reactor, a pattern on the surface of the oxide or control of the metal chemistry such that the growth is 3D rather than 2D. In some embodiments, semiconductor layer 108 may be grown as a continuous segment over the segmented metal layer 106. In some embodiments, semiconductor layer 108 may also be

segmented either by incorporating a mask within the reactor, a pattern on the surface of the oxide or control of the metal chemistry such that the growth is 3D rather than 2D as shown in 1606. In some embodiments, a second metal layer 1604 may be grown over the fragmented semiconductor layer 108 wherein the metal layer 1404 is grown in the cavities between the various semiconductor segments in layer 108. The second metal layer may be grown over different segments of the semiconductor layer. In some embodiments, upstream processes can access the second metal layer metal and use as a template / seed for additional processing steps (e.g., electroplating of thick contacts). In some embodiments, the semiconductor layer 108 may have different functions if grown over metal or grown over oxide.

[0068] FIG. 17 shows resonant frequencies as a function of AlN thickness for different thickness metal electrodes, in accordance with an embodiment of the prior art (from S. Tanifuji et al, Proceedings 2009 IEEE International Ultrasonic Symposium, p. 2170, the entirety of which is incorporated by reference). Here, crystal quality is also important because without it resistivity would increase as thickness decrease due to an increasing effect of defects and grain boundaries in polycrystalline metal layers.

[0069] FIG. 19 depicts a graph showing how DBR is constructed from 11 periods of AlN and GaN. As the angle of incidence increase the effective layer thickness results in the stop band shifting to lower wavelengths, meaning that at some angle the design wavelength (in this example 450nm) will fall outside of the central stop band.

[0070] The addition of an AlN over a metal (in this case molybdenum) decreases considerably this sensitivity to the angle of incidence.

[0071] FIG. 20 depicts the calculated reflectivity at 450nm is plotted for both an 11 period AlN-GaN DBR and a 10 period AlN-GaN DBR that is constructed on single pair of AlN over Mo. As may be seen at 60° angle of incidence the addition of the epitaxial metal layer has increased the reflectivity from 30% to 65%.

[0072] The epitaxial metal layers result in a larger grain size and less grain boundaries which enables thinner metal layers before losses associated with grain boundaries and defects become significant. Additionally the interfaces between the metal layers and the semiconductor are clean and discrete, both of which reduce the losses of a semiconductor – metal DBR when compared to a polycrystalline / sputtered DBR construct.

[0073] FIG. 21 is flow chart of a process 2100 growing layered structure 100, according to an illustrative embodiment. The process starts at 2102, when a substrate 102 is obtained. At 2104, a first REO layer 104 is grown over the substrate 102. At 2106 a first metal layer 106 is

epitaxially grown over the first REO layer 104. At 2108, a semiconductor layer 108 is epitaxially grown over the first metal layer 106.

[0074] At 2102, a substrate (e.g., see substrate 102 in FIG. 1) is obtained. In some embodiments, substrate includes a group IV element selected from a group of silicon (Si), germanium (Ge), silicon on insulator (SOI), and silicon carbide (SiC), wherein the substrate has a crystal orientation of either $\langle 100 \rangle$ or $\langle 111 \rangle$ with a miscut of upto 10 degrees, in one example.

[0075] At 2104, a first REO layer (e.g., see REO layer 104 in FIG. 1) epitaxially grown over the substrate.

[0076] At 2106, a first metal layer (e.g., see metal layer 106 in FIG. 1) epitaxially grown over the first REO layer.

[0077] At 2108, a first semiconductor layer (e.g., see semiconductor layer 104 in FIG. 1) epitaxially grown over the first metal layer.

[0078] FIGS. 22A-C provide example diagrams illustrating layered structures of metal electrodes build upon a silicon substrate containing a porous silicon portion, according to an embodiment described herein. As shown at layered structures 2201-2203, the substrate 102 may have a porous portion 102a. For example, the substrate 102, which may be a monolithic substrate, may include one or more group IV elements such as Si, Ge, silicon on insulator (SOI), and SiGe. The porous portion 102a may take a form as a sublayer. At 2201-2203, the porous sublayer 102a may be located at an upper portion of the substrate 102 such that the upper surface of the porous sublayer 102a is in direct contact with the layer that is over the substrate 102 without any transition layer between the porous sublayer 102a and the other layer. For example, at 2201, an epitaxial metal layer is directly over the porous sublayer 102a, with the semiconductor layer 108 over the epitaxial metal layer 106. In another example shown at 2202, any additional layer may be directly over the porous sublayer 102a, e.g., the semiconductor layer 108.

[0079] As shown at 2203, the porous portion in the substrate 102 may be non-continuous, e.g., with non-continuous and non-overlapping porous portions 102a and 102b. For example, the non-continuity of the porous portion may be extended to all three dimensions, e.g., different porous portions may be distributed in the substrate 102 vertically, or horizontally in two dimensions. For another example, different portions or regions (e.g., 102a/b) of the porous portion may have different porosities.

[0080] FIGS. 23A-B provide example diagrams showing layered structures built upon the layered structures shown in FIGS. 22A-C, with the rare earth oxide layer to support the

epitaxial metal layer, according to an embodiment described herein. As shown at 2301 and 2302, the rare earth oxide layer 104 may be grown or deposited over the substrate 102, over which the epitaxial metal layer 106 may be grown or deposited. Specifically, at 2301, a transition layer 102c, e.g., in the form of a continuous sublayer of the substrate 102, is located between the porous portion 102a and any other layer grown or deposited over the substrate 102 to transition the porous silicon to the other layer (e.g., rare earth oxide layer 104 in this example). The transition layer 102c may have a thickness of 5-10 nm.

[0081] Alternatively, as shown at 2302, no transition layer is between the porous sublayer 102a and the rare earth oxide layer 104. Namely, the rare earth oxide layer 104 is grown or deposited directly over the porous sublayer 102a.

[0082] FIGS. 24A-B and 25A-B provide various example diagrams showing layered structures using non-continuous rare earth oxide regions to define the location of the porous portion in the substrate, according to an embodiment described herein. As shown at 2401, the rare earth oxide layer may have a non-continuous pattern, e.g., with a first region 104a and a second region 104b that do not overlap with each other. The non-continuous regions 104a-b of the rare earth oxide layer may take a form similar to grids, rows, columns, dots, rings, or other irregular shapes. The epitaxial metal layer 106 is located and bounded by an empty region between the non-continuous regions 104a and 104b. Or alternatively, the empty region where the epitaxial metal layer 106 is located at can be surrounded by a continuous portion of the rare earth oxide layer.

[0083] The epitaxial metal layer 106 is in direct contact with the substrate 102. A second semiconductor layer 109 may be grown or deposited over the first semiconductor layer 108, and another rare earth oxide layer 112 is grown or deposited over the second semiconductor layer 108.

[0084] As shown at 2402, the substrate 102 may have a porous portion 102a. The boundary of the porous portion 102a aligns with the boundary of the epitaxial metal layer 106. Thus, the size of the porous portion 102a is also bounded by the gap or space between the rare earth oxide regions 104a and 104b.

[0085] As shown at 2501, the non-continuous rare earth oxide regions 104a-b may further bound the first semiconductor layer 108. Namely, the first semiconductor layer 108 may be deposited or grown over the epitaxial metal layer 106, but also bounded by the empty region between the non-continuous rare earth oxide regions 104a-b.

[0086] As shown at 2502, The porous portion 102a in the substrate aligns with the epitaxial metal layer 106, e.g., bounded by the empty region between the rare earth oxide regions 104a and 104b.

[0087] It is worth noting that the porous portion 102a and the empty region between the non-continuous rare earth oxide regions 104a-b are for illustrative purpose only. In some embodiments, there can be multiple non-continuous porous portions in the substrate 102, each of which aligns with a respective empty region in the rare earth oxide layer. Each respective empty region in the rare earth oxide layer may be filled with an epitaxial metal layer 106 (as shown in FIGS. 24A-B), or a combination of a epitaxial metal layer 106 and a semiconductor layer 108 (shown in FIGS. 25A-B). When a layered structure includes two or more empty regions in the rare earth oxide layer, a subset of the empty regions may be filled with epitaxial metal layer only (as shown in FIGS. 24A-B), and another subset of the empty regions may be filled with a combination of metal and semiconductor (as shown in FIGS. 25A-B).

[0088] FIG. 26 provides an example diagram illustrating a layered structure having the epitaxial metal layer defining the boundary of the porous portion of the substrate with a continuous rare earth oxide layer in between, according to an embodiment described herein. At 2600, a continuous rare earth oxide layer 104 is grown or deposited over the substrate 102. The epitaxial metal layer 106 is grown or deposited over the rare earth oxide layer 104, but having a smaller size than the rare earth oxide layer 104. The porous portion 102a of the substrate 102 aligns with the boundary of the epitaxial metal layer 106, with the continuous rare earth oxide layer 104 in between.

[0089] In some embodiments, the first semiconductor layer 108 is grown over or deposited over the epitaxial metal layer 106. A portion of the first semiconductor layer 108 may be in contact with the rare earth oxide layer 104.

[0090] In some embodiments, the epitaxial metal layer 106 may include multiple non-continuous regions distributed over the rare earth oxide layer 104. Each of the multiple non-continuous regions may align with a respective porous portion in the substrate 102, with the rare earth oxide layer 104 in between.

[0091] FIG. 27 provides an example diagram illustrating a layered structure having multiple semiconductors, according to an embodiment described herein. For example, multiple semiconductors may be used in any of the layered structures described herein. As shown in FIG. 27, semiconductors 108, 109 and 110 may be grown or deposited one over another, while the semiconductors 108 and 109 may form a device layer, and the semiconductors 109

and 110 may form another device layer. For example, the device layer may be a High-electron-mobility transistor (HEMT), etc. Any of the semiconductors 108-110 may be provided by a safer bonding operation.

[0092] Interfaces between semiconductors may include an addition of oxide, metal or the combination thereof to interface two semiconductor layers 108 and 109, or 109 and 110. For example, as shown in FIG. 27, a rare earth oxide layer 124 may interface between semiconductor 109 and semiconductor 110. For another example, an epitaxial metal layer 126 may interface between semiconductor 109 and semiconductor 110. Yet for another example, a combination of a rare earth oxide layer 124 and a epitaxial metal layer 126 may interface between semiconductor 109 and semiconductor 110. Yet for another example, a combination of rare earth oxide layer 124a, epitaxial metal layer 126 and another rare earth oxide layer 124b may interface between semiconductor 109 and semiconductor 110. In other embodiments, a repetition of the combination of rare earth oxide layer and epitaxial metal layer may be used to interface between semiconductor 109 and semiconductor 110. Any of the above interface layers may be applied between semiconductor 108 and semiconductor 109 as well.

[0093] The growth and/or deposition described herein may be performed using one or more of chemical vapor deposition (CVD), metalorganic chemical vapor deposition (MOCVD), organometallic vapor phase epitaxy (OMVPE), atomic layer deposition (ALD), molecular beam epitaxy (MBE), halide vapor phase epitaxy (HVPE), pulsed laser deposition (PLD), and/or physical vapor deposition (PVD).

[0094] As described herein, a layer means a substantially-uniform thickness of a material covering a surface. A layer can be either continuous or discontinuous (i.e., having gaps between regions of the material). For example, a layer can completely or partially cover a surface, or be segmented into discrete regions, which collectively define the layer (i.e., regions formed using selective-area epitaxy).

[0095] Monolithically-integrated means formed on the surface of the substrate, typically by depositing layers disposed on the surface.

[0096] Disposed on means "exists on" or "over" an underlying material or layer. This layer may comprise intermediate layers, such as transitional layers, necessary to ensure a suitable surface. For example, if a material is described to be "disposed on" or "over a substrate," this can mean either (1) the material is in intimate contact with the substrate; or (2) the material is in contact with one or more transitional layers that reside on the substrate.

[0097] Single-crystal means a crystalline structure that comprises substantially only one type of unit-cell. A single-crystal layer, however, may exhibit some crystalline defects such as stacking faults, dislocations, or other commonly occurring crystalline defects.

[0098] Single-domain means a crystalline structure that comprises substantially only one structure of unit-cell and substantially only one orientation of that unit cell. In other words, a single-domain crystal exhibits no twinning or anti-phase domains.

[0099] Single-phase means a crystalline structure that is both single-crystal and single-domain.

[0100] Substrate means the material on which deposited layers are formed. Exemplary substrates include, without limitation: bulk germanium wafers, bulk silicon wafers, in which a wafer comprises a homogeneous thickness of single-crystal silicon or germanium; composite wafers, such as a silicon-on-insulator wafer that comprises a layer of silicon that is disposed on a layer of silicon dioxide that is disposed on a bulk silicon handle wafer; or the porous germanium, germanium over oxide and silicon, germanium over silicon, patterned germanium, germanium tin over germanium, and/or the like; or any other material that serves as base layer upon which, or in which, devices are formed. Examples of such other materials that are suitable, as a function of the application, for use as substrate layers and bulk substrates include, without limitation, alumina, gallium-arsenide, indium-phosphide, silica, silicon dioxide, borosilicate glass, pyrex, and sapphire. A substrate may have a single bulk wafer, or multiple sub-layers. Specifically, a substrate (e.g., silicon, germanium, etc.) may include multiple non-continuous porous portions. The multiple non-continuous porous portions may have different densities and may be horizontally distributed or vertically layered.

[0101] Miscut Substrate means a substrate which comprises a surface crystal structure that is oriented at an angle to that associated with the crystal structure of the substrate. For example, a 6° miscut <100> silicon wafer comprises a <100> silicon wafer that has been cut at an angle to the <100> crystal orientation by 6° toward another major crystalline orientation, such as <110>. Typically, but not necessarily, the miscut will be up to about 20°. Unless specifically noted, the phrase “miscut substrate” includes miscut wafers having any major crystal orientation. That is, a <111> wafer miscut toward the <011> direction, a <100> wafer miscut toward the <110> direction, and a <011> wafer miscut toward the <001> direction.

[0102] Semiconductor refers to any solid substance that has a conductivity between that of an insulator and that of most metals. An example semiconductor layer is composed of silicon. The semiconductor layer may include a single bulk wafer, or multiple sub-layers. Specifically, a silicon semiconductor layer may include multiple non-continuous porous portions. The

multiple non-continuous porous portions may have different densities and may be horizontally distributed or vertically layered.

[0103] Semiconductor-on-Insulator means a composition that comprises a single-crystal semiconductor layer, a single-phase dielectric layer, and a substrate, wherein the dielectric layer is interposed between the semiconductor layer and the substrate. This structure is reminiscent of prior-art silicon-on-insulator (“SOI”) compositions, which typically include a single-crystal silicon substrate, a non-single-phase dielectric layer (e.g., amorphous silicon dioxide, etc.) and a single-crystal silicon semiconductor layer. Several important distinctions between prior-art SOI wafers and the inventive semiconductor-on-insulator compositions are that:

[0104] Semiconductor-on-insulator compositions include a dielectric layer that has a single-phase morphology, whereas SOI wafers do not. In fact, the insulator layer of typical SOI wafers is not even single crystal.

[0105] Semiconductor-on-insulator compositions include a silicon, germanium, or silicon-germanium “active” layer, whereas prior-art SOI wafers use a silicon active layer. In other words, exemplary semiconductor-on-insulator compositions include, without limitation: silicon-on-insulator, germanium-on-insulator, and silicon-germanium-on-insulator.

[0106] A first layer described and/or depicted herein as “configured on,” “on” or “over” a second layer can be immediately adjacent to the second layer, or one or more intervening layers can be between the first and second layers. A first layer that is described and/or depicted herein as “directly on” or “directly over” a second layer or a substrate is immediately adjacent to the second layer or substrate with no intervening layer present, other than possibly an intervening alloy layer that may form due to mixing of the first layer with the second layer or substrate. In addition, a first layer that is described and/or depicted herein as being “on,” “over,” “directly on,” or “directly over” a second layer or substrate may cover the entire second layer or substrate, or a portion of the second layer or substrate.

[0107] A substrate is placed on a substrate holder during layer growth, and so a top surface or an upper surface is the surface of the substrate or layer furthest from the substrate holder, while a bottom surface or a lower surface is the surface of the substrate or layer nearest to the substrate holder. Any of the structures depicted and described herein can be part of larger structures with additional layers above and/or below those depicted. For clarity, the figures herein can omit these additional layers, although these additional layers can be part of the structures disclosed. In addition, the structures depicted can be repeated in units, even if this repetition is not depicted in the figures.

[0108] From the above description it is manifest that various techniques may be used for implementing the concepts described herein without departing from the scope of the disclosure. The described embodiments are to be considered in all respects as illustrative and not restrictive. It should also be understood that the techniques and structures described herein are not limited to the particular examples described herein, but can be implemented in other examples without departing from the scope of the disclosure. Similarly, while operations are depicted in the drawings in a particular order, this should not be understood as requiring that such operations be performed in the particular order shown or in sequential order, or that all illustrated operations be performed, to achieve desirable results.

What is claimed is:

1. A layered structure, comprising:
a monolithic substrate having a continuous portion and a porous portion;
a first epitaxial metal layer over the monolithic substrate; and
a first semiconductor layer over the first epitaxial metal layer.
2. The layered structure of claim 1, wherein the monolithic substrate includes one or more group IV elements selected from a group of silicon (Si), germanium (Ge), silicon on insulator (SOI), and SiGe.
3. The layered structure of claim 1, wherein the monolithic substrate comprises a transition layer composed of continuous material between the porous portion and a first layer that is directly over the monolithic substrate.
4. The layered structure of claim 1, wherein the porous portion borders a layer that is directly over the monolithic substrate without any transition layer in between
5. The layered structure of claim 1, wherein the first epitaxial metal layer is directly over the monolithic substrate, and wherein the first epitaxial metal layer is in direct contact with the porous portion of the monolithic substrate
6. The layered structure of claim 1, wherein the porous portion has at least a first region and a second region that are distributed at different locations horizontally, vertically or at another dimension, and wherein the first region and the second region have different porosities.
7. The layered structure of claim 1, further comprising:
a first rare earth oxide layer between the monolithic substrate and the first metal layer.
8. The layered structure of claim 7, wherein the first metal layer covers a portion of an upper surface of the first rare earth oxide layer, and
wherein the first metal layer aligns with the porous portion of the monolithic substrate with the first rare earth oxide layer in between

9. The layered structure of claim 1, further comprising:
a second semiconductor layer over the first semiconductor layer; and
a second metal layer over the second semiconductor layer.
10. The layered structure of claim 1, further comprising:
a non-continuous rare earth layer over the monolithic substrate,
wherein the non-continuous rare earth oxide layer has at least an empty region
among other portions of the non-continuous rare earth oxide layer.
11. The layered structure of claim 10, wherein the first epitaxial metal layer is
filled within the empty region.
12. The layered structure of claim 11, wherein the first semiconductor layer is
over the empty region filled with the first epitaxial metal layer and the other portions of the
non-continuous rare earth oxide layer, and the layered structure further comprises a second
semiconductor layer over the first semiconductor layer.
13. The layered structured of claim 10, wherein the first semiconductor layer is
located within the empty region and over the first metal epitaxial layer, and the layered
structure further comprises a second semiconductor layer over the empty region filled with
the first semiconductor layer on top of the first epitaxial metal layer, and the other portions of
the non-continuous rare earth oxide layer.
14. The layered structure of claim 11, wherein the porous portion of the
monolithic substrate aligns with the first metal layer.
15. The layered structure of claim 1, further comprising:
a second semiconductor layer over the first semiconductor layer, and
a third semiconductor layer over the second semiconductor layer.
16. The layered structure of claim 15, wherein an interface between the first
semiconductor and the second semiconductor, or between the second semiconductor layer
and third semiconductor layer includes
a first rare earth oxide sublayer.

17. The layered structure of claim 16, wherein the interface further includes an epitaxial metal layer over the first rare earth oxide layer.
18. The layered structure of claim 17, wherein the interface further includes a second rare earth oxide layer over the epitaxial metal layer
19. The layered structure of claim 15, wherein an interface between the first semiconductor and the second semiconductor, or between the second semiconductor layer and third semiconductor layer includes an epitaxial metal layer.
20. The layered structure of claim 15, wherein the first semiconductor layer and the second semiconductor layer jointly form a first device layer, and
wherein the second semiconductor layer and the third semiconductor layer jointly form a second device layer.