

**PROCESS AND APPARATUS FOR PREPARING ALKYL ESTERS OF
METHACRYLIC ACID**

5 The subject-matter of the present invention relates to
a process for preparing alkyl methacrylates and their
conversion products which can be used in a multitude of
chemical synthesis processes which can lead to a wide
variety of different further processing products, and
to an apparatus for performing this process.

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For the preparation of alkyl methacrylates, there
already exist processes which have been known for some
time and are widely performed. Alkyl methacrylates are
used, for example, in the preparation of oligomeric or
15 polymeric compounds, and the correspondingly obtained
oligomeric or polymeric products can be found in a
multitude of objects, especially also in a multitude of
consumer goods in daily life.

20 Owing to the wide distribution of very different
preparation processes for alkyl methacrylates and owing
to the large amounts of such compounds which are
required, the corresponding processes are typically
scaled such that the amounts of product obtained are
25 high. For this reason, the optimization of known
processes in the further development of the wide
variety of different preparation methods is at the
forefront. Even a small improvement in a known process
can contribute to considerable advantages owing to the
30 large amounts of product prepared, for example to the
reduction of the energy consumption in the preparation
process or to the protection of a resource required in
the preparation process.

35 The present invention now addresses the problem that
water is used at many points in the process described
here for preparing alkyl methacrylates, and this water
typically has to be sent to disposal after passing

through the particular use. The present invention therefore has for its object to optimize the use of water as a raw material in a process for preparing alkyl methacrylates to the effect that the water
5 consumption is minimized as far as possible, and water used in the process can be used as far as possible at different points within the process in different process elements.

10 The present invention additionally has for its aim to increase the product quality in the context of an improved water circulation.

Summary of the invention

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The invention relates to a process for preparing alkyl methacrylates as the reaction product by

- a. subjecting a reaction mixture comprising methacrylamide, water, sulphuric acid and at least
20 one alkanol to an esterification reaction in one or more reaction chambers,
- b. subjecting the crude reaction product to a separating operation in at least one rectification column,
- 25 c. condensing the reaction product which has been subjected to a separating operation in one or more heat exchangers,
- d. separating the condensate into an organic phase and an aqueous phase in at least one separating
30 apparatus,
- e. washing the organic phase with water to obtain a washed organic phase and washing water, and optionally
- f. recycling the aqueous phase removed together with
35 the washing water back into at least one reaction chamber.

The reaction chamber used may, for example, be one tank

or a battery of 2, 3, 4 or more tanks. In a further embodiment of the invention, water which comprises an alkanol and sulphuric acid is fed at least partly to the reaction chamber or to the first of two or more
5 reaction chambers.

The temperature in the reaction chamber can adjusted, for example, to be about 90 to about 120°C, and vaporous reaction product can be withdrawn from the
10 reaction mixture from one or more reaction chambers and is then subjected, for example, to a separating operation by means of a rectification column, condensed in one or more heat exchangers and subsequently washed with water as the condensate, the washing water thus
15 obtained being removed from the reaction product by phase separation and recycled back into at least one reaction chamber.

A process according to the invention may comprise, for
20 example, the following steps:

- a. a mixture comprising water, ammonium salts, reaction product, solids and sulphuric acid (spent acid) is withdrawn from at least one reaction chamber,
- 25 b. the mixture is fed to at least one flotation vessel, in which it is largely freed of solids by flotation,
- c. the flotation vessel is charged with steam to the effect that the reaction product is converted from
30 the mixture into the vapour phase,
- d. the vapour phase comprising at least water and reaction product is cooled in at least one heat exchanger so as to form a postcondensate comprising water and reaction product, and
35 e. the postcondensate is recycled into at least one reaction chamber.

It has been found to be advantageous in some cases when

the inside of at least one heat exchanger is sprayed with washing water from the washing of the condensate from the rectification column of the prepurification step and, for example, a mixture of postcondensate and washing water forms in the heat exchanger and can be recycled into at least one reaction chamber.

It often leads to good results when the reaction product is subjected to at least one prepurification and at least one main purification, the reaction product in the prepurification being freed by distillation predominantly of low-boiling impurities which have a lower boiling point than the alkyl methacrylate prepared, and also of azeotropic mixtures thereof with water, and being freed in the main purification predominantly of higher-boiling impurities which have a higher boiling point than the alkyl methacrylate prepared.

The low-boiling impurities can then be subjected, for example, to a condensation in one or more heat exchangers, in which case the condensate can be subjected to a phase separation and the aqueous phase of the condensate can be recycled, for example, into at least one reaction chamber.

It has likewise been found to be advantageous in accordance with the invention in some cases when at least one reaction chamber is charged with methacrylamide and concentrated sulphuric acid.

According to the invention, for example, a battery of 2, 3 or 4 tanks can be used as reaction chambers.

In a further embodiment of the invention, the water present in the reaction mixture stems to an extent of at least 60% by weight from one or more of the following process elements:

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- a. washing of the crude reaction product after a first separating operation with water and subsequently removing the aqueous phase by phase separation,
- 5 b. driving reaction product out of the spent acid with steam and subsequently condensing the mixture of reaction product and steam thus obtained,
- c. distillatively removing low-boiling impurities and their azeotropes with water from the crude
10 reaction product, condensing the impurities removed, admixing the condensate with water and removing the aqueous phase by phase separation.

A process according to the invention may additionally
15 comprise one or more of the following process elements:

- a. preparing acetone cyanohydrin from acetone and hydrocyanic acid;
- b. purifying the acetone cyanohydrin in a rectification column;
- 20 c. converting the acetone cyanohydrin to methacrylamide.

The invention likewise provides an apparatus for preparing alkyl methacrylate, at least comprising

- 25 a. one or more reaction chambers in which a reaction mixture comprising methacrylamide, water, sulphuric acid and an alkanol can be subjected in a to an esterification reaction,
- b. at least one rectification column in which the
30 reaction product can be subjected to a separating operation,
- c. one or more heat exchangers in which the reaction product subjected to a separating operation can be condensed,
- 35 d. at least one washing column in which the condensate can be washed with water,
- e. at least one separating apparatus in which the washing water obtained can be removed from the

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reaction product by phase separation and

- 5 f. at least one fluid-conducting connection between separating apparatus and at least one reaction chamber, through which the removed aqueous phase can be recycled back into at least one reaction chamber.

An inventive apparatus may further at least comprise:

- 10 a. at least one distillation column in which the crude reaction product can be freed of impurities having a boiling point lower than that of the alkyl methacrylate and of azeotropes thereof with water;
- 15 b. at least one heat exchanger in which the vaporous impurities removed and their azeotropes with water can be condensed to form a condensate;
- c. at least one feedline for water, in which the condensate is admixed with water;
- 20 d. at least one separating apparatus in which an aqueous phase of the condensate can be separated from an organic phase and
- e. at least one fluid-conducting connection between the aqueous phase and at least one reaction
- 25 chamber;
- f. at least one vessel (flotation vessel) in which spent acid from the esterification reaction can be freed at least partly of solids;
- g. at least one fluid-conducting connection between
- 30 flotation vessel and a reaction chamber;
- h. at least one heat exchanger for condensing vaporous substances escaping from the flotation vessel;
- i. at least one fluid-conducting connection between
- 35 separating apparatus and heat exchanger, which enables feeding of the aqueous phase obtained in the separating apparatus into the interior of the heat exchanger;

- j. at least one fluid-conducting connection between heat exchanger and at least one reaction chamber.

The invention additionally relates to a process for preparing polymers based at least partly on alkyl methacrylates, comprising the steps of:

- a. preparing an alkyl methacrylate by a process according to the invention;
- b. polymerizing the alkyl methacrylate and optionally a comonomer, for example by free-radical polymerization;
- c. working up the alkyl methacrylate.

The invention further also relates to the use of an alkyl methacrylate obtainable by a process according to the invention in the production of fibres, films, coatings, moulding compositions, mouldings, papermaking auxiliaries, leather auxiliaries, flocculants and drilling additives, and also to fibres, films, coatings, moulding compositions, mouldings, papermaking auxiliaries, leather auxiliaries, flocculants and drilling additives which are based on an alkyl methacrylate obtainable by a process according to the invention.

25 Detailed description of the invention

In the context of the present invention, a "reaction chamber" is understood to mean a chamber within which an inventive esterification reaction can take place. A reaction chamber may, for example, be a tank, a flask, a loop reactor or the like. It is unimportant whether the esterification reaction in the reaction chamber is operated continuously or batchwise.

35 In the context of the present text, an esterification reaction is understood to mean a reaction whose reaction product is an alkyl methacrylate. An

esterification reaction in the present text is preferably understood to mean a reaction between methacrylic acid and a linear or branched C₁₋₂₂-mono-alkanol, preferably an alkanol having 1, 2, 3 or 4
5 carbon atoms in the alkyl radical and one OH group.

In the context of the present invention, a "crude reaction product" is understood to mean a substance mixture as is withdrawn from the reaction chamber in
10 the course of the reaction. In a preferred embodiment of the present invention, a "crude reaction product" is understood to mean that substance mixture which can be withdrawn from one or more tanks used as the reaction chamber in the reaction before any further processing
15 or purification.

In the context of the present text, a "rectification column" is understood to mean a column having a plurality of trays or corresponding internals or
20 fillings suitable for substance separation, in which the crude reaction product is subjected to a separating operation. A separating operation is understood to mean an operation in which a crude reaction product withdrawn from the reaction chamber is purified
25 together with steam suitable as a carrier by means of one or more column trays to such an extent that essentially water, alkyl methacrylate and alkanol from compounds having a higher boiling point or from compounds which are not discharged from the column in
30 the vapour phase under the conditions existing in the column.

In the context of the present invention, a "vapour phase" is understood to mean a gaseous substance or a
35 mixture of two or more gaseous substances which may optionally also comprise a proportion of liquid phase.

In the context of the present invention, a "washing

column" is understood to mean an apparatus in which the crude reaction product, after the separating operation, can be washed with water. For example, a suitable washing column is an apparatus in which the crude
5 reaction product, after the separating operation, can be contacted with water in the countercurrent. In such an operation, for example, alkyl methacrylate formed can be processed further as an organic phase after a phase separation.

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This invention relates in particular to the hydrolysis of methacrylamide to methacrylic acid and its subsequent esterification to methacrylic esters. This reaction can take place in one or more reaction
15 chambers which can essentially be selected according to the reaction. In particular, heated, for example steam-heated, tanks have been found to be suitable. However, it has in many cases been found to be advantageous when the esterification is performed in at least two
20 successive tanks as reaction chambers, but, for example, also in three or four or more successive tanks. In this case, a solution of methacrylamide is introduced into the tank or into the first tank of a battery of tanks comprising two or more tanks.

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It is frequently preferred in accordance with the invention to perform a corresponding esterification reaction with a battery of two or more tanks as reaction chambers. Reference will therefore be made
30 hereinafter exclusively to this variant.

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In the process described here, it is possible, for example, to feed an amide solution as obtainable from the amidation reaction described in this text into a
first tank. The tank is heated, for example, with steam. The amide solution supplied generally has an elevated temperature, for example a temperature of about 100 to about 180°C, essentially corresponding to

the exit temperature of the amide solution from the amidation reaction presented above. An alkanol is also fed to the tanks, which can be used for the esterification.

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Suitable alkanols here are in principle any alkanols having 1 to about 22 carbon atoms, which may be linear or branched, saturated or unsaturated, particular preference being given to methanol. These alkanols may
10 likewise be used together with methacrylic esters, which is the case especially in transesterifications.

The tank is also charged with water, so that there is a total water concentration in the tank of about 13 to
15 about 26% by weight, in particular about 18 to about 20% by weight. It may be preferred in accordance with the invention when water which comprises, for example, an alkanol or a mixture of two or more alkanols and sulphuric acid is fed to the reaction chamber. Suitable
20 alkanols are especially those alkanols which are present in the reaction mixture in the course of the esterification reaction.

The amount of amide solution and of alkanol is
25 controlled such that a total molar ratio of amide to alkanol of about 1:1.4 to about 1:1.6 exists. The alkanol can be distributed over the tank battery such that the molar ratio in the first reactor is about 1:1.1 to about 1:1.4 and, in the subsequent reaction
30 stages, based on the total amide stream, molar ratios of about 1:0.05 to about 1:0.3 are established. The alkanol fed into the esterification may be composed of "fresh alkanol" and alkanol from recycling streams of the workup stages and, if required, also of recycling
35 streams of the downstream processes of the production system.

The first tank can be charged with water in principle

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such that water is fed to the tank from any source, provided that this water has no ingredients which might adversely affect the esterification reaction or the downstream process stages. For example, demineralized water or spring water can be fed to the tank. However, it is envisaged in the present invention to feed a mixture of water and organic compounds partly to the tank as the reaction chamber, as obtained, for example, in the purification of methacrylic acid or methacrylic esters. In a further embodiment of the process presented here, the tank is charged at least in a proportion of at least about 50% by weight, for example at least about 60% by weight or at least about 70% by weight or at least about 80% by weight or at least about 90% by weight with a mixture of water and such organic compounds.

When a battery of two or more tanks is used in the esterification reaction, the vaporous or gaseous substances formed (the two terms are used synonymously here), especially the methacrylic esters, can in principle be drawn off individually from each tank and fed to a purification. However, it has been found in some cases to be advantageous when, in a battery of two or more tanks, the gaseous products from the first tank are first fed into the second reaction vessel without the gaseous compounds from the first tank being fed directly to a purification. This procedure offers the advantage that the frequently high evolution of foam in the first tank need not be counteracted by complicated defoaming apparatus. In the case of passage of the gaseous substances from the first tank into the second tank, the foam which has been formed in the first tank and may have been entrained also enters the reaction chamber of the second tank in a simple manner. Since the foam formation there is generally significantly lower, there is no need to use defoaming apparatus.

The second tank arranged downstream of a first tank then firstly takes up the overflow of the first tank; secondly, it is fed with the gaseous substances formed in the first tank or which are present in the first tank. The second tank and any following tanks are likewise charged with methanol. It is preferred here that the amount of methanol decreases by at least 10% from tank to tank, based in each case on the preceding tank. The water concentration in the second tank and in the further tanks may differ from that of the first tank; the concentration differences are, though, often small.

The control of the water concentration is possible, for example, by means of the supply of water already present in the process circuit and by means of the discharge of water from the reaction chambers. In this context, it is advantageous in accordance with the invention that the crude reaction product can be driven out of the reaction chambers with steam as a carrier and can then be subjected to a separating operation in a rectification column, which removes a mixture of water and organic compounds, especially alkyl methacrylate, from which the water can in turn be recovered as described.

The vapours formed in the second tank are removed from the tank and introduced into the bottom of a distillation column or a rectification column.

When the esterification is performed with a battery of three or more tanks, the overflow of the second tank is transferred in each case into a third tank, and the overflow of the third tank, if appropriate, into a fourth tank. The further tanks are likewise steam-heated. The temperature in tanks 3 and, if appropriate, 4 is preferably adjusted to about 120 to about 140°C.

The vapours escaping from the tanks can be passed into a distillation column or into a rectification column, the introduction preferably being effected in the lower region of the column. The vapours comprise an azeotropic mixture of carrier steam, methacrylic esters and alkanol and, depending on the alkanol used, have a temperature of about 60 to about 120°C, for example about 70 to about 90°C, when methanol is used. In the distillation column, the methacrylic ester is separated in gaseous form from the vapour constituents which boil at higher temperatures. The high-boiling fractions (mainly methacrylic acid, hydroxyisobutyric esters and water) are recycled, for example, into the first reaction tank. The methacrylic ester formed can be drawn off at the top of the column and cooled by means of a heat exchanger or a battery of two or more heat exchangers. It has been found to be useful in some cases when the methacrylic ester is cooled by means of at least two heat exchangers, in which case a first heat exchanger with water performs the condensation and a cooling to a temperature of about 60 to about 30°C, while a second brine-cooled heat exchanger undertakes a cooling to about 5 to about 15°C. A part-stream from the water-cooled condensate can be introduced as reflux to the columns for concentration control in the column. However, it is equally possible to cool the methacrylic ester formed by means of a battery of more than two heat exchangers. In this case, it is possible, for example, first to undertake a cooling by means of two water-cooled heat exchangers connected in series and then to achieve a further cooling by means of an appropriate brine-cooled heat exchanger..

For example, in the process presented here, the methacrylic ester formed can be cooled in the gaseous state by means of a first heat exchanger with water cooling. Both condensed and uncondensed substances are then passed on into a second heat exchanger, where a

further condensation by means of water cooling takes place. At this point, for example, gaseous substances can then be transferred into a separate brine-cooled heat exchanger. The condensate in this brine-cooled
5 heat exchanger is then introduced into the distillate stream, while the remaining gaseous substances can be utilized further or sent to disposal. The methacrylic ester condensate from the second water-cooled heat exchanger is then cooled in a water-cooled or brine-
10 cooled heat exchanger to a temperature of less than 15°C, preferably about 8 to about 12°C. This cooling step can lead to the methacrylic ester formed having a significantly lower content of formic acid than would be the case without the corresponding cooling step. The
15 cooled condensate is then transferred to a phase separator. Here, the organic phase (methacrylic ester) is separated from the aqueous phase. The aqueous phase which, as well as water, may also have a content of organic compounds, especially alkanol and possibly
20 further impurities, from the distillation step may in principle be used further as desired. However, as already described above, it may be preferred to recycle this mixture of water and organic compounds back into the esterification process by feeding it into the first
25 reaction tank.

The removed organic phase is fed into a scrubber. There, the methacrylic ester is scrubbed with demineralized water. This process step offers the
30 particular advantage that it is often possible to largely remove compounds which can lead to discolorations in subsequent steps from the product. The separated aqueous phase which comprises a mixture of water and organic compounds, especially alkanol and
35 possibly further impurities, can in turn in principle be used further as desired. However, it is advantageous for economic reasons to recycle this aqueous phase back into the esterification step by feeding it, for

example, into the first tank.

Since methacrylic esters have a strong tendency to polymerize, it is in many cases advantageous when care
5 is taken in the esterification of methacrylic acid that such a polymerization is prevented.

In plants for preparing methacrylic acid or methacrylic esters, polymerization often takes place when
10 methacrylic acid or methacrylic ester firstly have a low flow rate, so that local calm zones can form, in which contact lasting over a long period between methacrylic acid or methacrylic ester and a polymerization initiator can be established, which can
15 subsequently lead to polymerization.

In order to prevent such polymerization behaviour, it may be advantageous to optimize the substance flow to the effect that, firstly, the flow rate of the
20 methacrylic ester or of the methacrylic acid is sufficiently high at substantially all points in the system that the number of calm zones is minimized. Furthermore, it may be advantageous to admix the stream of methacrylic acid or methacrylic ester with suitable
25 stabilizers such that polymerization is largely suppressed.

For this purpose, the streams in the process presented here can in principle be admixed with stabilizers such
30 that a minimum level of polymerization takes place in the system itself. To this end, the part of the plant in particular in which the methacrylic acid or the methacrylic ester is present in high concentration during or after the distillation is supplied with
35 appropriate stabilizers.

For example, it has been found to be viable to supply a stabilizer at the top of the distillation column to the

stream of methacrylic ester drawn off there. Furthermore, it has been found to be advantageous to flush those parts of the plant in which methacrylic acid or methacrylic ester is circulated with a temperature of more than about 20°C, preferably with a temperature in the range of about 20 to about 120°C, with a solution of stabilizer in methacrylic ester. For example, some of the condensate obtained in the heat exchangers, together with a suitable stabilizer, is recycled into the top of the distillation column such that the column top, on its interior, is sprayed constantly with stabilized methacrylic ester or stabilized methacrylic acid. This is preferably done in such a way that no calm zones can form in the top of the column, at which there is a risk of polymerization of methacrylic acid or methacrylic ester. The heat exchangers themselves may correspondingly likewise be charged with a stabilized solution of methacrylic acid or methacrylic ester in such a way that no calm zones can form here either.

It has also been found to be advantageous in the process presented here when, for example, the offgases comprising carbon monoxide from preceding processes, especially from the amidation step, are passed through the esterification plant together with steam. In this way, the gas mixture is once again purified to remove compounds which can be removed in solid or in liquid form. Secondly, these are collected at a central point and can be sent to further utilization or disposal.

The methacrylic ester obtained or the MMA obtained in the esterification and the subsequent prepurification, or the methacrylic acid obtained, are subsequently sent to a further treatment. The esterification results in dilute sulphuric acid as the remaining residual substance, which can likewise be sent to a further utilization.

Prepurification of the ester or of the acid

In the process presented here, the subject matter of the present invention can also be used in connection with a process for prepurifying methacrylic acid or methacrylic ester, as described in the process element which follows. For instance, in principle, crude methacrylic acid or a crude methacrylic ester can be subjected to a further purification in order to arrive at a very pure product. Such a purification which constitutes a further process element can, for example, be in one stage. However, it has been found to be advantageous in many cases when such a purification comprises at least two stages, in which case the low-boiling constituents of the product are removed in a first prepurification as described here. To this end, crude methacrylic ester or crude methacrylic acid is transferred first into a distillation column in which the low-boiling constituents and water can be removed. To this end, the crude methacrylic ester is sent to a distillation column, in which case the addition is performed, for instance, in the upper half of the column. The column bottom is heated with steam, for example, in such a way that a wall temperature of about 50 to about 120°C is achieved. The purification is performed under reduced pressure. The pressure within the column in the case of the ester is preferably about 100 to about 600 mbar. The pressure within the column in the case of the acid is preferably about 40 to about 300 mbar.

At the top of the column, the low-boiling constituents are removed. In particular, these may, for example, be ether, acetone and methyl formate. The vapours are then condensed by means of one or more heat exchangers. For example, it has been found to be useful in some cases first to perform a condensation by means of two water-

cooled heat exchangers connected in series. However, it is equally possible to use only one heat exchanger at this point. The heat exchangers are preferably operated in an upright state to increase the flow rate and in order to prevent the formation of stationary phases. Connected downstream of the water-cooled heat exchanger or the water-cooled heat exchangers may be a brine-cooled heat exchanger, but it is also possible to connect a battery of two or more brine-cooled heat exchangers downstream. In the battery of heat exchangers, the vapours are condensed, provided with stabilizer and, for example, fed to a phase separator. Since the vapours may also contain water, any aqueous phase which occurs is disposed of or sent to a further utilization. An example of a possible further utilization is recycling into an esterification reaction, for example into an esterification reaction as has been described above. In this case, the aqueous phase is preferably recycled into the first esterification tank.

The removed organic phase is fed as reflux into the top of the column. Some of the organic phase can in turn be used to spray the tops of the heat exchangers and the top of the column. Since the removed organic phase is a phase which has been admixed with stabilizer, it is thus possible firstly to effectively prevent the formation of calm zones. Secondly, the presence of the stabilizer brings about further suppression of the polymerization tendency of the vapours removed.

The condensate stream obtained from the heat exchangers is additionally preferably admixed with demineralized water in such a way that sufficient separating action can be achieved in the phase separator.

The gaseous compounds which remain after the condensation in the heat exchanger battery may,

preferably by means of steam ejectors as reduced-pressure generators, be subjected once again to a condensation by means of one or more further heat exchangers. It has been found to be advantageous for economic reasons when such a postcondensation condenses not only the gaseous substances from the prepurification. For example, it is possible to feed further gaseous substances to such a postcondensation, as obtained from the main purification of methacrylic esters. The advantage of such a procedure lies, for example, in transferring such a proportion of methacrylic ester which has not been condensed in the main purification stage once more via the phase separator into the purification column in the prepurification. It is thus ensured, for example, that a maximization of yield can take place and minimum losses of methacrylic esters occur. Moreover, the suitable selection of the design and the operation of these further heat exchangers allows the composition of the offgas leaving these heat exchangers, especially the content of low boilers, to be adjusted.

Owing to the feeding of water in the prepurification of the methacrylic ester, the water content in the esterification and the concentration of low-boiling constituents in the crude methyl methacrylate overall can rise continuously. In order to prevent this, it may be advantageous to discharge some of the water fed to the system out of the system, preferably continuously. This discharge can in principle be effected, for example, in an order of magnitude in which water is fed to the system in the prepurification. The aqueous phase separated out in the phase separator typically has a content of organic constituents. It may therefore be advantageous to feed this water to a form of disposal which utilizes this content of organic substances.

For example, it may be advantageous when water thus

contaminated with organic substances is fed to the combustion chamber in a sulphuric acid cleavage process. Owing to the oxidizable constituents, its calorific value can still be utilized at least partly.

- 5 In addition, a possibly expensive disposal of the water contaminated with organic substances is thus often avoided.

Fine purification of the methacrylic ester

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For the fine purification of the methacrylic ester, the prepurified methacrylic ester is subjected to another distillation. This frees the crude methacrylic ester of its high-boiling constituents with the aid of a
15 distillation column to obtain a pure methacrylic ester. To this end, the crude methacrylic ester is introduced into the lower half of a distillation column in a manner known to those skilled in the art.

- 20 The distillation column can in principle correspond to any design which appears to be suitable to those skilled in the art. However, it has been found to be advantageous in many cases for the purity of the resulting product when the distillation column is
25 operated with one or more packings which correspond approximately to the following requirements:

- Firstly, just like in the other lines flowed through by methacrylic ester, a minimum level of so-called "dead
30 spaces" should form in the columns. The dead spaces lead to a comparatively long residence time of the methacrylic esters, which promotes their polymerization. This in turn leads to expensive production shutdowns and cleaning of the appropriate
35 parts blocked with polymer. One way of countering the formation of dead spaces is, both by design and by a sufficient operating mode of the columns, to always load them with a sufficient amount of liquid, so that

constant flushing of the columns and especially of the column internals such as packings is achieved. For instance, the columns may have spray devices which are designed for the spraying of the column internals. In addition, the column internals may be connected to one another or to the column via interrupted adhesion seams. Such adhesion seams have at least about 2, preferably at least about 5 and more preferably at least about 10 interruptions for 1 m of adhesion seam length. The length of these interruptions may be selected such that they make up at least about 10%, preferably at least about 20% and more preferably at least about 50%, but generally not more than 95% of the adhesion seam length. Another design measure may be that, in the internal regions of the column, especially those which come into contact with the methacrylic esters, less than about 50%, preferably less than about 25% and more preferably less than about 10% of all surfaces, especially of column internals, run horizontally. For example, the stubs which open into the interior of the column may be configured conically or with oblique surfaces. Another measure may consist in keeping the amount of liquid methacrylic ester present in the column bottom as low as possible during the operation of the column, and secondly in preventing overheating of this amount in spite of moderate temperatures and large evaporation surfaces during the evaporation. It may be advantageous in this context that the amount of liquid in the column bottom makes up in the range of about 0.1 to 15% and preferably about 1 to 10% of the total amount of methacrylic ester in the column. The measures proposed in this paragraph may also find use in the distillation of methacrylic acid.

In the purification of the methacrylic ester, its high-boiling constituents are separated from the product by distillation. To this end, the column bottom is heated with steam. The bottom temperature is preferably about

50 to about 80°C, in particular about 60 to about 75°C, with wall temperature of less than about 120°C.

5 The material obtained in the column bottom is preferably removed continuously and cooled by means of a heat exchanger or a battery of several heat exchangers to a temperature in a range of about 40 to about 80°C, preferably about 40 to about 60°C and more preferably in a range of about 50 to 60°C.

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This material, which comprises predominantly methacrylic ester, hydroxyisobutyric ester, methacrylic acid and stabilizer components, is subsequently, via a storage vessel, for example, disposed of or sent to
15 another use. It has been found to be advantageous in many cases when the material obtained in the column bottom is recycled into the esterification reaction. For example, the material from the column bottom is recycled into the first tank. This gives rise to the
20 advantage that, with a view to a very economically viable method and a very high yield, relatively high-boiling compounds present in the column bottoms are recycled into the esterification reaction.

25 At the top of the column, the methacrylic ester purified by distillation is withdrawn and cooled by means of a heat exchanger or a battery of two or more heat exchangers. The heat of the vapours can be removed by means of water-cooled heat exchangers or by means of
30 brine-cooled heat exchangers or by means of a combination of the two. It has been found to be useful in some cases when the vapours from the distillation column are transferred into two or more heat exchangers connected in parallel, which are operated by means of
35 water cooling. The uncondensed fractions from the water-cooled heat exchangers can, for example, be introduced into a brine-cooled heat exchanger or a battery of two or more brine-cooled heat exchangers,

which may be arranged in series or in parallel. The condensates obtainable from the heat exchangers are introduced into a collecting vessel and sent to a buffer vessel by means of a pump via a further heat exchanger or a battery of two or more further heat exchangers. The condensate stream is cooled, for example, by means of a battery of one or two water-cooled heat exchangers and one or two brine-cooled heat exchangers down to a temperature in a range of about 0 to about 20°C, preferably about 0 to about 15°C and more preferably in a range of about 2 to 10°C.

A part-stream is withdrawn from the condensate stream and is recycled into the distillation column via the top of the column. The condensate stream can be fed into the top of the column in principle in any way, for example via distributors. However, it may be advantageous when a portion of the condensate stream is fed into the vapour line above the top of the column, for example sprayed in. It is also preferred that this feeding also introduces stabilizer into the top of the column.

A further part-stream of the condensate intended for recycling into the column can, for example, be branched off into the vapour line before introduction and be introduced directly into the top of the column. Here too, it is preferred that this feeding introduces stabilizer into the top of the column. The introduction into the top of the column can be done, for example, in such a way that the interior of the top of the column is sprayed with the condensate such that no calm zones can form in the top of the column at which the methacrylic ester can polymerize. It may additionally be advantageous to add a stabilizer for preventing polymerization to a condensate part-stream which is recycled into the column. This can be done, for example, by adding an appropriate amount of

polymerization inhibitor as stabilizer to the condensate part-stream intended for spraying of the top of the column. It has been found to be advantageous in some cases when the condensate part-stream, after the addition of the stabilizer but before entry into the top of the column, passes through a suitable mixing apparatus, preferably a static mixer, in order to achieve very uniform distribution of the stabilizer in the condensate part-stream.

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The uncondensable gaseous substances which are obtained in the purification process are, for example, sent to disposal.

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The crude product present in the buffer vessel is kept with the aid of a brine cooler at a temperature of about 0 to about 20°C, preferably about 0 to about 15°C and more preferably in a range of about 2 to 10°C.

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In order to remove any further impurities from the product and to arrive at ultrapure methacrylates, the product can also be subjected to an absorptive purification stage. It has been found to be useful, for example, when the pure product as a whole, or at least

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a portion of the pure product, is purified further with the aid of a molecular sieve. Particularly acidic impurities, especially formic acid formed in the preparation process, can thus be removed in a simple manner from the product stream. It has additionally

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been found to be useful in some cases when the product stream, after passing through the adsorptive purification stage, also passes through one or more filters in order to remove any solids present in the product.

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The streams obtained in the workup comprise predominantly polymerizable compounds. In order to, as already described more than once in this text, prevent

the formation of calm zones, it has been found to be advantageous in the case of the process described here too when the parts of the plant which come into contact with methacrylic ester are constantly flowed over with methacrylic ester. In a further embodiment of the process presented here, a part-stream of methacrylic ester is therefore withdrawn downstream of the buffer vessel but upstream of the adsorptive purification stage in order to be flushed over the top regions of those heat exchangers which take up the vapours stemming from the distillation column.

The product obtained in the purification stage is subsequently withdrawn from the purification stage with a temperature in a range of about -5 to about 20°C, preferably about 0 to about 15°C and more preferably in a range of about 2 to 10°C.

Stripping of the spent acid

In the process presented here, it may be advisable, for example, to subject the spent sulphuric acid obtained in the process to a purification in order to subsequently recycle it back into the process. In this case, for example, a stream comprising spent sulphuric acid, as can be obtained from the esterification, can be contacted with steam in a flotation vessel. As this is done, at least some of the solids present can be deposited on the surface of the liquid, and these deposited solids can be separated out. The vapours are subsequently condensed in a heat exchanger, preferably with water cooling, cooled and recycled into the esterification reaction.

It has been found to be advantageous in some cases when corrosion is prevented in the heat exchangers and the cooling action is improved further by introducing a mixture of water and organic compounds, as obtained by

scrubbing in the course of the esterification in the purification of the methacrylic ester prepared, into the heat exchangers in such a way that the tops of the heat exchangers are sprayed with this mixture. In addition to the corrosion-reducing action and the cooling of the acid in the heat exchanger, this procedure has a further advantage. Material which stems from the esterification (a mixture of water and predominantly methanol) is recycled into the esterification process together with the methacrylic acid and methacrylic ester stemming from exactly this process. In the stripper, the above-described flotation affords mixtures of acid and solids. After their removal, these are sent to any further use or to disposal. It is possible, for example, to incinerate the resulting mixture in a cleavage plant and hence to obtain sulphuric acid again to recover some of the energy used in the process.

The uncondensable gaseous compounds obtained in the stripping are sent to any further use or disposed of.

The plant described here for removing solids from the spent acid and for recycling material from the esterification process into exactly this process can also be performed, for example, twice for reasons of operational reliability. For instance, the two or more flotation vessels can be used offset in time. Since solids can settle out in these vessels, it is advantageous to remove them when the particular flotation vessel is not being used.

The present invention further relates to the use of the methacrylic acid obtainable by the process according to the invention and of the methacrylic esters obtainable by the process according to the invention in fibres, films, coatings, moulding compositions, mouldings, papermaking auxiliaries, leather auxiliaries,

flocculants and drilling additives. In addition, the present invention relates to fibres, films, coatings, moulding compositions, mouldings, papermaking auxiliaries, leather auxiliaries, flocculants and drilling additives which are based on a methacrylic acid obtainable by the process according to the invention and on a methacrylic ester obtainable by the process according to the invention.

Various process elements will be illustrated hereinafter, which can in principle be combined with the present invention individually or as an ensemble of two or more of the process elements mentioned. In some cases, it may be advantageous when the process elements presented within the present text are combined with the present invention such that they are combined overall to give a process for preparing esters of methacrylic acid or a process for preparing methacrylic acid. However, it should also be pointed out that advantageous effects can usually also be achieved when the subject matter of the present invention as such is used in another field or only combined with some of the process elements presented here.

Preparation of acetone cyanohydrin

In this process element, acetone cyanohydrin is prepared by commonly known processes (see, for example, Ullmann's Enzyklopädie der technischen Chemie, 4th Edition, Volume 7). Frequently, the reactants used are acetone and hydrocyanic acid. The reaction is an exothermic reaction. In order to counteract decomposition of the acetone cyanohydrin formed in this reaction, the heat of reaction is typically removed by a suitable apparatus. The reaction can be conducted in principle as a batch process or as a continuous process; when a continuous process is preferred, the reaction is frequently performed in a loop reactor

which is fitted out appropriately.

A main feature of a method leading to the desired product in high yields is often that the reaction product is cooled at sufficient reaction time and the reaction equilibrium is shifted in the direction of the reaction product. In addition, the reaction product is frequently admixed with an appropriate stabilizer to the advantage of the overall yield, in order to prevent decomposition in the course of the later workup to give the starting materials.

The mixing of the acetone and hydrocyanic acid reactants can in principle be effected in essentially any way. The method of mixing depends in particular on whether a batchwise mode, for example in a batch reactor, or a continuous mode, for example in a loop reactor, is selected.

In principle, it may be advantageous when the acetone is fed into the reaction via a reservoir vessel which has a scrubber tower. Venting lines which conduct waste air containing acetone and hydrocyanic acid can thus be conducted, for example, through this reservoir vessel.

In the scrubbing tower which is attached to the reservoir vessel, the waste air escaping from the reservoir vessel can be scrubbed with acetone, which removes hydrocyanic acid from the waste air and recycles it into the process. For this purpose, for example, some of the amount of acetone introduced from the reservoir vessel into the reaction is conducted in the part-stream through a cooler, preferably through a brine cooler, into the top of the scrubbing tower and the desired result is thus achieved.

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Depending on the size of the amount of end products to be produced, it may be advantageous to feed the acetone to the reaction from more than just one reservoir

vessel. In this context, it is possible for each of the two or more reservoir vessels to bear a corresponding scrubbing tower. However, it is in many cases sufficient when only one of the reservoir vessels is
5 equipped with a corresponding scrubbing tower. In this case, it is, however, often advisable for corresponding lines which conduct waste air and can transport acetone and hydrocyanic acid to be conducted through this vessel or through this scrubbing tower.

10 The temperature of the acetone in the reservoir may in principle be within an essentially arbitrary range, provided that the acetone is in the liquid state at the appropriate temperature. The temperature in the reservoir vessel is advantageously, however, about 0 to
15 about 20°C.

In the scrubbing tower, the acetone used for scrubbing is cooled by means of an appropriate cooler, for example by means of a plate cooler, with brine to a
20 temperature of about 0 to about 10°C. The temperature of the acetone on entry into the scrubbing tower is therefore preferably, for example, about 2 to about 6°C.

25 The hydrocyanic acid required in the reaction can be introduced into the reactor either in liquid or in gaseous form. It may, for example, be crude gas from the BMA process or from the Andrussow process.

30 The hydrogen cyanide can, for example, be liquefied, for example by the use of an appropriate cooling brine. Instead of liquefied hydrocyanic acid, coking oven gas can be used. For example, hydrogen cyanide-containing coking oven gases, after scrubbing with potash, are
35 scrubbed continuously in countercurrent with acetone which contains 10% water, and the reaction to give acetone cyanohydrin can be carried out in the presence

of a basic catalyst in two gas scrubbing columns connected in series.

In a further embodiment, a gas mixture comprising
5 hydrogen cyanide and inert gases, especially a crude gas from the BMA process or from the Andrussow process, can be reacted with acetone in the presence of a basic catalyst and acetone cyanohydrin in a gas-liquid reactor.

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In the process described here, preference is given to using a BMA crude gas or an Andrussow crude gas. The gas mixture resulting from the abovementioned customary processes for preparing hydrogen cyanide can be used as
15 such or after an acid scrubbing. The crude gas from the BMA process, in which essentially hydrocyanic acid and hydrogen are formed from methane and ammonia, contains typically 22.9% by volume of HCN, 71.8% by volume of H₂, 2.5% by volume of NH₃, 1.1% by volume of N₂, 1.7% by
20 volume of CH₄. In the known Andrussow process, hydrocyanic acid and water are formed from methane and ammonia and atmospheric oxygen. The crude gas of the Andrussow process, when oxygen is used as the oxygen source, contains typically 8% by volume of HCN, 22% by
25 volume of H₂, 46.5% by volume of N₂, 15% by volume of H₂O, 5% by volume of CO, 2.5% by volume of NH₃ and 0.5% by volume each of CH₄ and CO₂.

When a non-acid-scrubbed crude gas from the BMA process
30 or Andrussow process is used, the ammonia present in the crude gas frequently acts as a catalyst for the reaction. Since the ammonia present in the crude gas frequently exceeds the amount required as a catalyst and can therefore lead to high losses of sulphuric acid
35 used for stabilization, such a crude gas is often subjected to an acid scrubbing in order to eliminate ammonia therefrom. When such an acid-scrubbed crude gas is used, it is then necessary, however, to add a

suitable basic catalyst to the reactor in a catalytic amount. In principle, known inorganic or organic basic compounds may function as the catalyst.

- 5 Hydrogen cyanide in gaseous or in liquid form, or a gas mixture comprising hydrogen cyanide, and acetone are fed continually to a loop reactor in the continuous mode. In this case, the loop reactor comprises at least one means of feeding acetone or two or more such means,
10 at least one means for feeding liquid or gaseous hydrocyanic acid, or two or more such means, and at least one means for feeding a catalyst.

Suitable catalysts are in principle any alkaline
15 compounds, such as ammonia, sodium hydroxide solution or potassium hydroxide solution, which can catalyse the reaction of acetone and hydrocyanic acid to give acetone cyanohydrin. However, it has also been found to be advantageous when the catalyst used is an organic
20 catalyst, especially an amine. Suitable examples are secondary or tertiary amines such as diethylamine, dipropylamine, triethylamine, tri-n-propylamine and the like.

- 25 A loop reactor useable in the process element described further comprises at least one pump, or two or more pumps, and at least one mixing apparatus, or two or more such mixing apparatuses.

- 30 Suitable pumps are in principle all pumps which are suitable for ensuring the circulation of the reaction mixture in the loop reactor.

Suitable mixing apparatuses are both mixing apparatuses
35 with mobile elements and so-called static mixers in which immobile flow resistances are provided. In the case of use of static mixers, suitable examples are those which allow an operational transfer of at least

about 10 bar, for example at least about 15 bar or at least about 20 bar under operating conditions without significant restrictions in the functioning. Appropriate mixers may consist of plastic or metal. Suitable plastics are, for example, PVC, PP; HDPE, PVDF, PVA or PTFE. Metal mixers may consist, for example, of nickel alloys, zirconium, titanium and the like. Likewise suitable are, for example, rectangular mixers.

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The catalyst is added in the loop reactor preferably downstream of the pump and upstream of a mixing element present in the loop reactor. In the reaction described, catalysts are used, for example, in such an amount that the overall reaction is conducted at a pH of not more than 8, in particular not more than about 7.5 or about 7. It may be preferred when the pH in the reaction varies within a range of about 6.5 to about 7.5, for example about 6.8 to about 7.2.

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Alternatively to the addition of the catalyst into the loop reactor downstream of the pump and upstream of a mixing apparatus, it is also possible in the process described to feed the catalyst into the loop reactor together with the acetone. In such a case, it may be advantageous when appropriate mixing of acetone and catalyst is ensured before feeding into the loop reactor. Appropriate mixing can be effected, for example, by the use of a mixer with moving parts or by use of a static mixer.

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When a continuous process in a loop reactor is selected as the operating mode in the process described, it may be appropriate to examine the state of the reaction mixture by instantaneous or continual analyses. This offers the advantage that, where appropriate, it is also possible to react rapidly to changes in state in the reaction mixture. Furthermore, it is thus possible,

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for example, to meter in the reactants very precisely in order to minimize yield losses.

Corresponding analysis can be effected, for example, by
5 sampling in the reactor loop. Suitable analysis methods are, for example, pH measurement, measurement of the exothermicity or measurement of the composition of the reaction mixture by suitable spectroscopic processes.

10 Especially within the context of conversion monitoring, quality aspects and safety, it has been found to be useful to determine the conversion in the reaction mixture via the heat removed from the reaction mixture and to compare it with the heat which is released
15 theoretically.

In the case of suitable selection of the loop reactor, the actual reaction can in principle be effected within the tube systems arranged within the loop reactor.
20 Since the reaction, however, is exothermic, in order to avoid yield loss, sufficient cooling and sufficient removal of the heat of reaction should be ensured. It has frequently been found to be advantageous when the reaction proceeds within a heat exchanger, preferably
25 within a tube bundle heat exchanger. Depending on the amount of product to be produced, the capacity of an appropriate heat exchanger can be selected differently. For industrial scale processes, heat exchangers having a volume of about 10 to about 40 m³ in particular have
30 been found to be particularly suitable. The tube bundle heat exchangers used with preference are heat exchangers which have a tube bundle flowed through by liquid within a jacket flowed through by liquid. Depending on the tube diameter, packing density, etc.,
35 the heat transfer between the two liquids can be adjusted appropriately. It is possible in principle in the process described to conduct the reaction to the effect that the reaction mixture is conducted through

the heat exchanger in the tube bundle itself and the reaction takes place within the tube bundle, the heat being removed from the tube bundle into the jacket liquid.

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However, it has likewise been found to be practicable and in many cases to be viable to conduct the reaction mixture through the jacket of the heat exchanger, while the liquid used for cooling is circulated within the tube bundle. It has in many cases been found to be advantageous when the reaction mixture is distributed within the jacket by means of flow resistances, preferably deflecting plates, to achieve a higher residence time and better mixing.

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The ratio of jacket volume to the volume of the tube bundle may, depending on the design of the reactor, be about 10:1 to about 1:10; the volume of the jacket is preferably greater than the volume of the tube bundle (based on the contents of the tubes).

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The heat removal from the reactor is adjusted with an appropriate coolant, for example with water, such that the reaction temperature is within a corridor of about 25 to about 45°C, in particular about 30 to about 38°C, in particular about 33 to about 35°C.

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A product is removed continuously from the loop reactor. The product has a temperature within the abovementioned reaction temperatures, for example a temperature of about 35°C. The product is cooled by means of one or more heat exchangers, especially by means of one or more plate heat exchangers. For example, brine cooling is used. The temperature of the product after cooling should be about 0 to 10°C, in particular 1 to about 5°C. The product is preferably transferred into a storage vessel which has a buffer function. In addition, the product in the storage

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- 35 -

vessel can be cooled further, for example, by constantly removing a part-stream from the storage vessel to a suitable heat exchanger, for example to a plate heat exchanger, or kept at a suitable storage temperature. It is entirely possible that continued reaction can take place in the storage vessel.

The product can be recycled into the storage vessel in any way in principle. However, it has been found to be advantageous in some cases that the product is recycled by means of a system composed of one or more nozzles into the storage vessel such that corresponding mixing of the stored product takes place within the storage vessel.

Product is also removed continuously from the storage vessel into a stabilization vessel. The product is admixed there with a suitable acid, for example with H_2SO_4 . This deactivates the catalyst and adjusts the reaction mixture to a pH of about 1 to about 3, in particular about 2. A suitable acid is in particular sulphuric acid, for example sulphuric acid having a content of about 90 to about 105%, in particular of about 93 to about 98% H_2SO_4 .

The stabilized product is withdrawn from the stabilization vessel and transferred into the purification stage. A portion of the stabilized product withdrawn can be recycled, for example, into the stabilization vessel such that sufficient mixing of the vessel is ensured by means of a system composed of one or more nozzles.

ACH workup

In a further process element which can be used in connection with the present invention, acetone cyanohydrin which has been obtained in a preceding

stage, for example from the reaction of acetone with hydrocyanic acid, is subjected to a distillative workup. The stabilized crude acetone cyanohydrin is freed of low-boiling constituents by means of a
5 corresponding column. A suitable distillation process can be conducted, for example, by means of only one column. However, it is likewise possible, in an appropriate purification of crude acetone cyanohydrin, to use a combination of two or more distillation
10 columns, also combined with a falling-film evaporator. In addition, two or more falling-film evaporators or else two or more distillation columns may be combined with one another.

15 The crude acetone cyanohydrin comes from the storage to the distillation generally with a temperature of about 0 to about 15°C, for example a temperature of about 5 to about 10°C. In principle, the crude acetone cyanohydrin can be introduced directly into the column.
20 However, it has been found to be useful in some cases when the crude cool acetone cyanohydrin, by means of a heat exchanger, first takes up some of the heat of the product already purified by distillation. Therefore, in a further embodiment of the process described here, the
25 crude acetone cyanohydrin is heated by means of a heat exchanger to a temperature of about 60 to 80°C.

The acetone cyanohydrin is purified by distillation by means of a distillation column having more than 10
30 trays, or by means of a battery of two or more corresponding suitable distillation columns. The column bottom is heated preferably with steam. It has been found to be advantageous when the bottom temperature does not exceed a temperature of 140°C; good yields and
35 good purification have been achieved when the bottom temperature is not greater than about 130°C or not higher than about 110°C. The temperature data are based on the wall temperature of the column bottom.

The crude acetone cyanohydrin is fed to the column body in the upper third of the column. The distillation is performed preferably under reduced pressure, preferably at a pressure of about 50 to about 900 mbar, in particular about 50 to about 250 mbar and with good results between 50 to about 150 mbar.

At the top of the column, gaseous impurities, especially acetone and hydrocyanic acid, are removed, the removed gaseous substances are cooled by means of one heat exchanger or a battery of two or more heat exchangers. Preference is given here to using brine cooling with a temperature of about 0 to about 10°C. This gives the gaseous constituents of the vapours the opportunity to condense. The first condensation stage may take place, for example, at standard pressure. However, it is equally possible and has been found to be advantageous in some cases when this first condensation stage is effected under reduced pressure, preferably at the pressure which prevails in the distillation. The condensate is passed on into a cooled collecting vessel and collected there at a temperature of about 0 to about 15°C, in particular at about 5 to about 10°C.

The gaseous compounds which do not condense in the first condensation step are removed from the reduced pressure chamber by means of a vacuum pump. In principle, any vacuum pump can be used here. However, it has been found to be advantageous in many cases to use a vacuum pump which, owing to its design, does not lead to the introduction of liquid impurities into the gas stream. Preference is therefore given here, for example, to using dry-running vacuum pumps.

The gas stream which escapes on the pressure side of the pump is conducted through a further heat exchanger

which is preferably cooled with brine at a temperature of about 0 to about 15°C. Constituents which condense here are likewise collected in the collecting vessel which already collects the condensates obtained under vacuum conditions. The condensation performed on the pressure side of the vacuum pump can be effected, for example, by a heat exchanger, but also with a battery of two or more heat exchangers arranged in series in parallel. Gaseous substances remaining after this condensation step are removed and sent to any further utilization, for example a thermal utilization.

The collected concentrates may likewise be utilized further as desired. However, for economic reasons, it has been found to be extremely advantageous to recycle the condensates into the reaction for preparing acetone cyanohydrin. This is effected preferably at one or more points which enable access to the loop reactor. The condensates may in principle have any composition provided that they do not disrupt the preparation of the acetone cyanohydrin. In many cases, a predominant amount of the condensate will, however, consist of acetone and hydrocyanic acid, for example in a molar ratio of about 2:1 to about 1:2, frequently in a ratio of about 1:1.

The acetone cyanohydrin obtained from the bottom of the distillation column is first cooled to a temperature of about 40 to about 80°C by the cold crude acetone cyanohydrin fed in by means of a first heat exchanger. Subsequently, the acetone cyanohydrin is cooled to a temperature of about 30 to about 35°C by means of at least one further heat exchanger and optionally stored intermediately.

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Amidation

In a further process element as frequently provided in

the preparation of methacrylic acid or of esters of methacrylic acid, acetone cyanohydrin is subjected to a hydrolysis. At different temperature levels after a series of reactions, this forms methacrylamide as the product.

The reaction is brought about in a manner known to those skilled in the art by a reaction between concentrated sulphuric acid and acetone cyanohydrin. The reaction is exothermic, which means that heat of reaction is advantageously removed from the system.

The reaction here too can again be performed in a batchwise process or in continuous processes. The latter has been found to be advantageous in many cases. When the reaction is performed in a continuous process, the use of loop reactors has been found to be useful. The reaction can be effected, for example, in only one loop reactor. However, it may be advantageous when the reaction is performed in a battery of two or more loop reactors.

In the process described, a suitable loop reactor has one or more feed points for acetone cyanohydrin, one or more feed points for concentrated sulphuric acid, one or more gas separators, one or more heat exchangers and one or more mixers.

The hydrolysis of acetone cyanohydrin with sulphuric acid to give methacrylamide is, as already described, exothermic. The heat of reaction which arises in the reaction must, however, at least largely be withdrawn from the system, since the yield falls with increasing temperature in the reaction. It is possible in principle to achieve a rapid and comprehensive removal of the heat of reaction with appropriate heat exchangers. However, it may also be disadvantageous to cool the mixture too greatly, since a sufficient heat

transfer is required for appropriate exchange at the heat exchangers. Since the viscosity of the mixture rises greatly with falling temperature, the circulation in the loop reactor is firstly complicated, and
5 sufficient removal of the reaction energy from the system can secondly no longer be ensured.

In addition, excessively low temperatures in the reaction mixture can lead to a crystallization of
10 constituents of the reaction mixture at the heat exchangers. This further worsens the heat transfer, as a result of which a clear yield reduction can be detected. In addition, the loop reactor cannot be charged with the optimal amounts of reactants, such
15 that the efficiency of the process suffers overall.

In one embodiment of the process, a portion, preferably about two thirds to about three quarters, of the volume flow rate from a stream of acetone cyanohydrin is
20 introduced into a first loop reactor. A first loop reactor preferably has one or more heat exchangers, one or more pumps, one or more mixing elements and one or more gas separators. The circulation streams which pass through the first loop reactor are preferably in the
25 range of about 100 to 450 m³/h, preferably in a range of 200 to 400 m³/h and more preferably in a range of about 250 to 350 m³/h. In an at least one further loop reactor which follows the first loop reactor, the circulation streams are preferably in a range of about
30 40 to 450 m³/h, preferably in a range of 50 to 400 m³/h and more preferably in a range of about 60 to 350 m³/h. Moreover, a preferred temperature difference over the heat exchangers is about 1 to 10°C, particular preference being given to about 2 to 7°C.

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The acetone cyanohydrin can in principle be fed into the loop reactor at any point. However, it has been found to be advantageous when the feed is into a mixing

element, for example into a mixer with moving parts or a static mixer. The sulphuric acid is advantageously fed in upstream of the acetone cyanohydrin addition. Otherwise, it is, however, likewise possible to feed
5 the sulphuric acid into the loop reactor at any point.

The ratio of the reactants in the loop reactor is controlled such that an excess of sulphuric acid is present. The excess of sulphuric acid is, based on the
10 molar ratio of the constituents, about 1.8:1 to about 3:1 in the first loop reactor and about 1.3:1 to about 2:1 in the last loop reactor.

In some cases, it has been found to be advantageous to
15 perform the reaction in the loop reactor with such an excess of sulphuric acid. The sulphuric acid may serve here, for example, as a solvent and keep the viscosity of the reaction mixture low, which can ensure a higher removal of heat of reaction and a lower temperature of
20 the reaction mixture. This can entail significant yield advantages. The temperature in the reaction mixture is about 90 to about 120°C.

The heat removal is ensured by one or more heat
25 exchangers in the loop reactor. It has been found to be advantageous when the heat exchangers have a suitable sensor system for controlling the cooling performance in order to prevent excessively great cooling of the reaction mixture for the aforementioned reasons. For
30 example, it may be advantageous to measure the heat transfer in the heat exchanger or in the heat exchangers point by point or continuously and to adjust the cooling performance of the heat exchangers thereto. This can be done, for example, via the coolant itself.
35 It is equally possible to achieve appropriate heating of the reaction mixture by corresponding variation of the addition of the reactants and by the generation of more heat of reaction. A combination of the two

possibilities is also conceivable. The loop reactor should additionally have at least one gas separator. One method is to withdraw product formed continuously from the loop reactor via the gas separator. Another
5 method is thus to withdraw the gases formed in the reaction from the reaction chamber. The gas formed is mainly carbon monoxide. The product withdrawn from the loop reactor is preferably transferred into a second loop reactor. In this second loop reactor, the reaction
10 mixture comprising sulphuric acid and methacrylamide, as has been obtained by the reaction in the first loop reactor, is reacted with the remaining part-stream of acetone cyanohydrin. In this case, the excess of sulphuric acid from the first loop reactor, or at least
15 some of the excess sulphuric acid, reacts with the acetone cyanohydrin to form further methacrylamide. The performance of the reaction in two or more loop reactors has the advantage that, owing to the sulphuric acid excess in the first loop reactor, the pumpability
20 of the reaction mixture and hence the heat transfer and ultimately the yield are improved. In turn, at least one mixing element, at least one heat exchanger and at least one gas separator are arranged in the second loop reactor. The reaction temperature in the second loop
25 reactor is likewise about 90 to about 120°C.

The problem of the pumpability of the reaction mixture, of heat transfer and of a minimum reaction temperature occurs in every further loop reactor just as it does in
30 the first. Therefore, the second loop reactor too advantageously has a heat exchanger whose cooling performance can be controlled by an appropriate sensor system.

35 The acetone cyanohydrin is again fed in in a suitable mixing element, preferably into a static mixer.

The product is withdrawn from the gas separator of the

second loop reactor and heated to a temperature of about 140 to about 180°C to complete the reaction and to form the methacrylamide.

- 5 The heating is preferably performed such that the maximum temperature is attained only for a minimum period, for example for a time of about one minute to about 30 minutes, in particular for a time of about two to about eight or about three to about five minutes.
- 10 This can in principle be effected in any apparatus for achieving such a temperature for such a short period. For example, the energy can be supplied in a conventional manner by electrical energy or by steam. However, it is equally possible to supply the energy by
- 15 means of electromagnetic radiation, for example by means of microwaves.

- In various cases, it has been found to be advantageous when the heating step is effected in a heat exchanger
- 20 with two-stage or multistage arrangement of tube coils which may preferably be present in an at least double, opposite arrangement. This heats the reaction mixture rapidly to a temperature of about 140 to 180°C.

- 25 The heat exchanger can be combined, for example, with one or more gas separators. For example, it is possible to conduct the reaction mixture through a gas separator after it leaves the first tube coil in the heat exchanger. This can remove, for example, gaseous
- 30 components formed during the reaction from the reaction mixture. It is equally possible to treat the reaction mixture with a gas separator after it leaves the second coil. It may additionally be found to be advantageous to treat the reaction mixture with a gas separator both
- 35 after it leaves the first tube coil and after it leaves the second tube coil.

The amide solution thus obtainable generally has a

temperature of more than 100°C, typically a temperature of about 140 to 180°C.

The gaseous compounds obtained in the amidation can in principle be disposed of in any way or sent to further processing. However, it may be advantageous in some cases when the appropriate gases are combined in a transport line in such a way that they are optionally pressurized either continuously or as required, for example with steam pressure, and can thus be transported further.

The aforementioned will now be illustrated by way of example with reference to nonlimiting drawings. The drawings show:

- Fig. 1: a plant system for preparing and processing with methacrylic acid or methyl methacrylate,
- Fig. 2: a schematic of a plant for preparing acetone cyanohydrin,
- Fig. 3: a schematic of a workup plant for acetone cyanohydrin,
- Fig. 4: a schematic of an amidation plant,
- Fig. 5: a schematic of an esterification plant,
- Fig. 6: a schematic of a plant for prepurifying the ester,
- Fig. 7: the fine purification plant for the ester.

Fig. 1 shows the preferred elements of a plant system 1 for preparing methacrylic acid or methacrylic esters and their further processing products. The plant system 1 has various plants connected to one another usually in a fluid-conducting manner as elements of this system. This plant system includes acetone cyanohydrin preparation 20, followed by acetone cyanohydrin workup 30, followed by an amidation 40, followed by an esterification/hydrolysis 50/50a, followed by a workup for ester or methacrylic acid 60, followed in turn by a

fine purification 70, after which the ester, usually methyl methacrylate, or methacrylic acid is present. The pure ester/pure acid thus obtained can be sent to a further processing plant 80. Useful further processing plants 80 include in particular polymerization apparatus and reactors for further organic reactions. In the polymerization reactors, polymethacrylates can be prepared, and, in the reactors for organic reactions, the pure monomers obtained here can be converted to further organic compounds. The further processing plant or the further processing plants 80 is/are followed by a finishing 90. When the further processing products are polymers of methacrylic acid or methacrylic esters, especially methyl methacrylate, they are processed further to give fibres, moulding compositions, especially granules, films, slabs, automobile parts and other mouldings by suitable equipment such as extruders, blown-film extruders, injection-moulding machines, spinneret dies and the like. In addition, the plant system 1 in many cases comprises a sulphuric acid plant 100. For this plant, all sulphuric acid plants which appear to be suitable for this purpose to the person skilled in the art are useful in principle. Reference is made in this context, for example, to the in Chapter 4, page 89 ff. in "Integrated Pollution Prevention and Control - Draft Reference Document on Best Available Techniques for the Manufacture of Large Volume Inorganic Chemicals - Ammonia Acids and Fertilizers" obtainable via the European Commission. The sulphuric acid plant 10 is connected to a series of other plants. For instance, the acetone cyanohydrin preparation 20 is supplied with concentrated sulphuric acid via a sulphuric acid line 2. Moreover, a further sulphuric acid line 3 exists between the sulphuric acid plant 100 and the amidation 40. The dilute sulphuric acid also referred to as "Spent Acid" from the esterification 50 (hydrolysis 50a) is transferred to the sulphuric acid plant 100

through the lines for spent sulphuric acid 4 and 5. In the sulphuric acid plant 100, the dilute sulphuric acid can be worked up. The workup of the dilute sulphuric acid can be effected, for example, as described in
5 WO 02/23088 A1 or WO 02/23089 A1. In general, the plants are manufactured from the materials which are familiar to those skilled in the art and appear to be suitable for the particular stresses. Usually, the material is stainless steel which must in particular
10 have exceptional acid resistance. The regions of the plants which are operated with sulphuric acid and especially with concentrated sulphuric acid are additionally lined and protected with ceramic materials or plastics. In addition, the methacrylic acid obtained
15 in the methacrylic acid plant 50a can be fed via a methacrylic acid line 6 to the prepurification 60. It has also been found to be useful to add a stabilizer indicated with "S" in the acetone cyanohydrin preparation 20, the amidation 40, the esterification
20 50, the hydrolysis 50a, the prepurification 60 and also the end purification 70.

In the acetone cyanohydrin preparation 20 shown in Figure 2, the acetone is provided in an acetone vessel
25 21 and the hydrocyanic acid in a hydrocyanic acid vessel 22. The acetone vessel 21 has a scrubbing tower 23 which, in its upper region, has one or more cooling elements 24. A series of offgas lines 25 which stem from various plants in the plant system 1 open into the
30 scrubbing tower 23. The acetone is fed into a loop reactor 26 via the acetone feed 27 and the hydrocyanic acid via the hydrocyanic acid feed 28 downstream of the hydrocyanic acid feed 28 is disposed a pump 29, followed in turn by a catalyst feed 210 which is
35 followed by a static mixer 211. This is followed by a heat exchanger 212 which has a series of flow resistances 213 and at least one cooling line 214. In the loop reactor 26, the reaction mixture consisting of

acetone, hydrocyanic acid and catalyst is conducted in a circuit to a considerable degree, which is indicated by bold lines. From the heat exchanger 212, the reaction mixture is conducted via the flow resistances along the cooling lines 214, and a portion of the circulation stream is passed into a further heat exchanger 215 to which is connected a collecting vessel 216 in which a nozzle 217 is present as part of a cooling circuit 218 with a heat exchanger 219, which keeps the reaction product firstly in motion and secondly cool. Via an outlet 220 which follows the collecting vessel 216, a stabilizer vessel 221 is attached, into which a sulphuric acid feed 222 opens and from which the crude acetone cyanohydrin is conducted through the outlet 223 into the acetone cyanohydrin workup 30.

In Figure 3, coming from the cyanohydrin preparation 20, the outlet 223 opens into a heat exchanger 31 in which the stream coming from the cyan preparation 20 is heated. A vapour feed 32 is connected to the heat exchanger 31 and opens out in the upper region, preferably the top region, of a column 33. The column 33 has a multitude of packings 34 which are usually configured as trays. In the lower region of the column 33 is disposed the column bottom 35 from which a bottoms outlet 36 leads into the heat exchanger 31 and heats the streams conducted through the outlet 223 into the heat exchange 31. A pure product line 37 is connected to the heat exchanger 31, which is followed downstream by the amidation 40. In the top region of the column 33 is disposed a tops outlet 38 which opens into a heat exchanger 39 to which a vacuum pump 310 is connected and opens in turn into a heat exchanger 311. Both the heat exchanger 39 and the heat exchanger 311 are connected via lines to a cooling vessel 312 to which a recycle line 313 is connected and is connected to the loop reactor 26 in the acetone cyanohydrin

preparation 20.

The amidation 40 depicted in Fig. 4 first has an acetone cyanohydrin feed 41 and a sulphuric acid feed 42 which open into a loop reactor 43. The acetone cyanohydrin feed 41 connected to the acetone cyanohydrin workup 30 opens into the circuit of the loop reactor 43 downstream of a pump 44 and upstream of a mixer 45. Upstream of this pump 44, the sulphuric acid feed 42 opens out. The mixer 45 is followed downstream by a heat exchanger 46 which in turn opens into a gas separator 47 from which, firstly, a gas outlet 48 and a feed 49 to a further loop reactor 410 exit. The further loop reactor 410 or a third has a comparable construction to the first loop reactor 43. From the further loop reactor 410, a feed 411 enters a heat exchanger 412 which is followed by a gas separator 413, from which to a gas outlet 414 and an amide line 415 exit, the latter leading to the esterification/hydrolysis 50/MAA plant 50a.

Figure 5 shows the esterification 50, in which a solvent line 51 which conducts water and organic solvent, and an amide line 52 connected to the amidation 40 open into a tank 53 which is heatable by a tank heater 54. In addition, an alcohol line 55 shown with a broken line opens into the tank 53. The alcohol line 55 opens out both in the upper and in the lower region of the tank 53. The first tank 53 is connected to a further tank 53', which has a further tank heater 54', via an ester vapour line 56 indicated by a line of dashes and dots. This further tank 53' too is connected to the alcohol line 55 both from the bottom and from the top. The ester vapour line 56 is connected to the upper region of the tank 53' and opens into a bottom 57 of a column 58. In addition, a line for dilute sulphuric acid 59 is present in the upper region of the tank 53'. A tank unit 510 encircled in a dotted ellipse

is formed from a heatable tank 53 and 54 with alcohol line 55 and ester vapour line 56. It is possible for one, two or more of such tank units to follow in battery-like succession, each of these tank units 510
5 being connected via the ester vapour line 56 to the bottom 57 of the column 58. From the bottom 57 of the column 58, a high boiler line 511 also leads to the tank 53, in order to feed water and organic solvent back to the esterification. In the upper region,
10 preferably the top, of the column 58, a first heat exchanger 512 followed by a further phase separator 513 are connected via a suitable line. Both at the top of the column 58 and in the first heat exchanger 512, a first stabilizer feed 414 (stabilizer indicated with
15 "S") and a further stabilizer feed 515 may be provided in order to feed an inhibitor or stabilizer which prevents undesired polymerization. Connected to the further phase separator 513 is a scrubber 516 in whose lower region a solvent line 517 exits and opens out in
20 the solvent line 51 via a heat exchanger 521. From the upper region of the scrubber 516, a crude ester line exits and opens into the ester workup 60. The spent acid line 59 exiting from the upper region of the tank 53' or of the tank of the last tank unit 510 opens into
25 a flotation vessel 519 for removal of the solids and constituents insoluble in the spent acid. From the flotation vessel 519, a spent acid outlet 520 enters the sulphuric acid plant 100, and a low boiler vapour line 522 which conducts the low-boiling constituents,
30 for further workup and recycling, enters the esterification.

The ester workup shown in Figure 6 is connected to the esterification 50 via a crude ester line 61, the crude
35 ester feed 61 opening into the middle region of a vacuum distillation column 62. This column 62 has column internals 63 and a bottom heater 64 arranged in the lower region of the column 62. From the lower

region of the column 62 which constitutes the bottom of this column, an ester outlet 65 exits, opens into the ester fine purification 70 and hence feeds the crude ester freed of low boilers to the fine purification. In the upper region of the column 62, usually in the top, a first heat exchanger 66 is connected via an outlet, as are one further heat exchanger or a plurality of heat exchangers 67 which are followed by a phase separator 69. In the phase separator 69 the mixture stemming from the heat exchanger 67 is divided into organic and aqueous constituents, a recycle line 611 in the upper region being connected to the phase separator 69 and opening out in the upper region of the column 62. In the lower region of the separator, a water outlet 610 is present and opens into the esterification 50 in order to feed the water removed back to the esterification. A reduced-pressure generator 613 is connected to the heat exchangers 66 and 67 via a reduced-pressure line 612.

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In Figure 7, the ester outlet 65 stemming from the ester workup 60 opens into a distillation column 71. This comprises a plurality of column internals 71 and, in the lower region of the distillation column 71, a column bottom heater 73. From the top region of the distillation column 71, a pure ester vapour line 74 enters a first heat exchanger 75 which is followed by one (or more) further heat exchangers 76 which are connected to a reduced-pressure generator 717. The outlet of the further heat exchanger 76 has a line from which, firstly, an ester recycle line 77 opens into the upper region, or into the top, of the distillation column 71. The ester recycle line 77 has a stabilizer metering point 79 which is disposed in the ester recycle line 77 upstream of a mixer 78. Secondly, from the line of the further heat exchanger 76, a pure ester outlet 710 exits. An additional heat exchanger 711 and another heat exchanger 712 are connected to this in

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- 51 -

series connection. These are followed by a molecular sieve vessel 713 which has molecular sieve packings 714. Purified further by the molecular sieve, the ultrapure ester is transferred through the ultrapure ester outlet connected to the molecular sieve vessel into the further processing plant 80.

Reference numeral list

	1	Plant system
	2	Sulphuric acid line
5	3	Further sulphuric acid line
	4	Spent sulphuric acid line - ester
	5	Spent sulphuric acid line - acid
	6	Methacrylic acid line
	20	Acetone cyanohydrin preparation
10	30	Acetone cyanohydrin workup
	40	Amidation
	50	Esterification
	50a	Hydrolysis
	60	Prepurification
15	70	End purification
	80	Further processing plant
	90	Finishing
	100	Sulphuric acid plant
	21	Acetone vessel
20	22	Hydrocyanic acid vessel
	23	Scrubbing tower
	24	Cooling elements
	25	Offgas lines
	26	Loop reactor
25	27	Acetone feed
	28	Hydrocyanic acid feed
	29	Pump
	210	Catalyst feed
	211	Mixer
30	212	Heat exchanger
	213	Flow resistance
	214	Cooling lines
	215	Heat exchanger
	216	Collecting vessel
35	217	Nozzle
	218	Cooling circuit
	219	Heat exchanger
	220	Outlet

- 221 Stabilizing vessel
- 222 Sulphuric acid feed
- 223 Outlet
- 31 Heat exchanger
- 5 32 Vapour feed
- 33 Column
- 34 Packings
- 35 Column bottom with heat exchanger
- 36 Bottoms outlet
- 10 37 Pure product line
- 38 Tops outlet
- 39 Heat exchanger
- 310 Vacuum pump
- 311 Heat exchanger
- 15 312 Cooling vessel
- 313 Recycle line
- 41 Acetone cyanohydrin feed
- 42 Sulphuric acid feed
- 43 Loop reactor
- 20 44 Pump
- 45 Mixer
- 46 Heat exchanger
- 47 Gas separator
- 48 Gas outlet
- 25 49 Feed
- 410 Further loop reactor
- 411 Feed
- 412 Heat exchanger
- 413 Gas separator
- 30 414 Gas outlet
- 415 Amide line
- 51 Solvent line
- 52 Amide line
- 53 First tank
- 35 54 First tank heater
- 53' Further tank
- 54' Further tank heater
- 55 Alcohol line

- 56 Ester vapour line
- 57 Column bottom
- 58 Column
- 59 Spent acid line
- 5 510 Tank unit
- 511 High boiler line
- 512 Heat exchanger
- 513 Phase separator
- 514 Stabilizer feed
- 10 515 Further stabilizer feed
- 516 Extraction column
- 517 Solvent line
- 518 Crude ester line
- 519 Flotation vessel
- 15 520 Spent acid outlet
- 521 Heat exchanger
- 522 Low boiler vapour line
- 61 Crude ester line
- 62 Vacuum distillation column
- 20 63 Column internals
- 64 Bottom heater
- 65 Ester outlet
- 66 Heat exchanger
- 67 Heat exchanger
- 25 68 Water feed
- 69 Phase separator
- 610 Water outlet
- 611 Recycle line
- 612 Reduced-pressure line
- 30 613 Reduced-pressure generator
- 71 Distillation column
- 72 Column internals
- 73 Column bottom heater
- 74 Pure ester vapour line
- 35 75 First heat exchanger
- 76 Further heat exchanger
- 77 Ester recycle line
- 78 Mixer

- 55 -

- 79 Stabilizer metering point
- 710 Pure ester outlet
- 711 Additional heat exchanger
- 712 Other heat exchanger
- 5 713 Molecular sieve vessel
- 714 Molecular sieve packings
- 715 Ultrapure ester outlet
- 716 High boiler line
- 717 Low boiler draw

Claims

1. Process for preparing alkyl methacrylates as the reaction product by
 - 5 a. subjecting a reaction mixture comprising methacrylamide, water, sulphuric acid and at least one alkanol to an esterification reaction in one or more reaction chambers,
 - 10 b. subjecting the crude reaction product to a separating operation in at least one rectification column,
 - c. condensing the reaction product which has been subjected to a separating operation in one or more heat exchangers,
 - 15 d. separating the condensate into an organic phase and an aqueous phase in at least one separating apparatus,
 - e. washing the organic phase with water to obtain a washed organic phase and washing water, and optionally
 - 20 f. recycling the aqueous phase removed together with the washing water back into at least one reaction chamber.
- 25 2. Process according to Claim 1, characterized in that the reaction chamber used is one tank or a battery of 2, 3, 4 or more tanks.
3. Process according to one of Claims 1 or 2,
30 characterized in that water which comprises alkanol and sulphuric acid is fed at least partly to the reaction chamber or to the first of two or more reaction chambers.
- 35 4. Process according to one of the preceding claims, characterized in that the temperature in the reaction chamber is adjusted to about 90 to about 120°C, and vaporous reaction product is withdrawn

from the reaction mixture from one or more reaction chambers and is subjected to a separating operation by means of a rectification column, condensed in one or more heat exchangers and subsequently washed with water as the condensate, the washing water thus obtained being removed from the reaction product by phase separation and recycled back into at least one reaction chamber.

5. Process according to one of the preceding claims, characterized in that it comprises the following steps:

a. a mixture comprising water, ammonium salts, reaction product, solids and sulphuric acid (spent acid) is withdrawn from at least one reaction chamber,

b. the mixture is fed to at least one flotation vessel, in which it is largely freed of solids by flotation,

c. the flotation vessel is charged with steam to the effect that the reaction product is converted from the mixture into the vapour phase,

d. the vapour phase comprising at least water and reaction product is cooled in at least one heat exchanger so as to form a postcondensate comprising water and reaction product, and

e. the condensate is recycled into at least one reaction chamber.

6. Process according to Claim 5, characterized in that the inside of at least one heat exchanger is sprayed with washing water from the washing of the condensate from the rectification column of the prepurification step.

7. Process according to Claim 6, characterized in

that a mixture of postcondensate and washing water forms in the heat exchanger and is recycled into at least one reaction chamber.

5 8. Process according to one of the preceding claims, characterized in that the reaction product is subjected to at least one prepurification and at least one main purification, the reaction product in the prepurification being freed by distillation
10 predominantly of low-boiling impurities which have a lower boiling point than the alkyl methacrylate prepared, and also of azeotropic mixtures thereof with water, and being freed in the main
15 purification predominantly of higher-boiling impurities which have a higher boiling point than the alkyl methacrylate prepared.

9. Process according to Claim 8, characterized in that the low-boiling impurities are subjected to a
20 condensation in one or more heat exchangers, the condensate is subjected to a phase separation and the aqueous phase of the condensate is recycled into at least one reaction chamber.

25 10. Process according to one of the preceding claims, characterized in that at least one reaction chamber is charged with a mixture of methacrylamide and concentrated sulphuric acid.

30 11. Process according to one of the preceding claims, characterized in that a battery of 2, 3 or 4 tanks is used as reaction chambers.

12. Process according to one of the preceding claims,
35 characterized in that the water present in the reaction mixture stems to an extent of at least 60% by weight from one or more of the following process elements:

- a. washing of the crude reaction product after a first separating operation with water and subsequently removing the aqueous phase by phase separation,
- 5 b. driving reaction product out of the spent acid with steam and subsequently condensing the mixture of reaction product and steam thus obtained,
- 10 c. distillatively removing low-boiling impurities and their azeotropes with water from the crude reaction product, condensing the impurities removed, admixing the condensate with water and removing the aqueous phase by phase separation.
- 15 13. Process according to one of the preceding claims, characterized in that it comprises one or more of the following process elements:
- 20 a. preparing acetone cyanohydrin from acetone and hydrocyanic acid;
- b. purifying the acetone cyanohydrin in a rectification column;
- c. converting the acetone cyanohydrin to methacrylamide.
- 25 14. Apparatus for preparing alkyl methacrylate, at least comprising
- 30 a. one or more reaction chambers in which a reaction mixture comprising methacrylamide, water, sulphuric acid and an alkanol can be subjected in a to an esterification reaction,
- b. at least one rectification column in which the reaction product can be subjected to a separating operation,
- 35 c. one or more heat exchangers in which the reaction product subjected to a separating operation can be condensed,
- d. at least one separating apparatus in which

the condensate can be separated into an organic phase and an aqueous phase,

e. at least one washing column in which the organic phase can be washed with water, and

5 f. at least one fluid-conducting connection between separating apparatus and at least one reaction chamber, through which the removed aqueous phase, if appropriate together with the washing water, can be recycled back into
10 at least one reaction chamber.

15. Apparatus according to Claim 14, characterized in that it further at least comprises:

15 a. at least one distillation column in which the crude reaction product can be freed of impurities having a boiling point lower than that of the alkyl methacrylate and of azeotropes thereof with water;

20 b. at least one heat exchanger in which the vaporous impurities removed and their azeotropes with water can be condensed to form a condensate;

c. at least one feedline for water, in which the condensate is admixed with water;

25 d. at least one separating apparatus in which an aqueous phase of the condensate can be separated from an organic phase and

30 e. at least one fluid-conducting connection between the aqueous phase and at least one reaction chamber.

16. Apparatus according to Claim 14 or 15, characterized in that it further at least comprises:

35 f. at least one vessel (flotation vessel) in which spent acid from the esterification reaction can be freed at least partly of solids;

- g. at least one fluid-conducting connection between flotation vessel and a reaction chamber;
 - 5 h. at least one heat exchanger for condensing vaporous substances escaping from the flotation vessel;
 - i. at least one fluid-conducting connection between separating apparatus and heat exchanger, which enables feeding of the
10 aqueous phase obtained in the separating apparatus into the interior of the heat exchanger;
 - j. at least one fluid-conducting connection between heat exchanger and at least one
15 reaction chamber.
17. Process for preparing polymers based at least partly on alkyl methacrylates, comprising the steps of:
- 20 a. preparing an alkyl methacrylate by a process according to Claim 1;
 - b. polymerizing the alkyl methacrylate and optionally a comonomer;
 - c. working up the alkyl methacrylate.
- 25 18. Process according to Claim 17, wherein the polymerization is effected by free-radical polymerization.