

COMPOSITION BASED ON ORGANOTIN MALEATES WHICH CAN BE
USED TO STABILIZE AND LUBRICATE THERMOPLASTIC POLYMERS.

PROCESS FOR PRODUCING THE SAID COMPOSITION.

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5 The subject of the present invention is a
composition composed of organotin maleates and a
process for the production thereof.

 The invention also relates to thermoplastic
polymer compositions stabilized and lubricated using
10 the said composition based on organotin maleates. More
particularly, the invention relates to vinyl chloride
polymer compositions of controlled melting which are
stabilized and lubricated using the said composition
based on organotin maleates.

15 A further subject of the invention relates to
extruded rigid articles obtained from vinyl chloride
polymer compositions of controlled melting which are
stabilized and lubricated using the said composition
based on organotin maleates.

20 A number of organotin maleates forming good
heat-stabilization and photostabilization agents for
vinyl chloride polymers are known.

 Thus, compositions containing organotin
maleates, such as dibutyltin bis(isooctyl maleate) or
25 dibutyltin bis(isooctyl maleate) oxide, are described
in the literature. The use of these compositions as
heat-stabilizers for halogenated organic compounds, in

particular halogenated polymers such as poly(vinyl chloride) (PVC) and chlorinated poly(vinyl chloride) (CPVC), is also well known.

5 The effectiveness of stabilizers based on organotin maleate for PVC, in particular articles intended for external use, is well known.

 Thus, in the article entitled "Worldwide Weathering of Polyvinyl Chloride", by Emery Szabo and Robert Lally; Polymer and Engineering Science, April 10 1975, Vol. 15, No. 4, the results of studies on long-term exposure to weather are presented. The stabilizing compositions used were compositions containing barium/cadmium soaps, compositions containing organotin maleates and compositions containing organotin 15 mercaptoacetates. It is noted, in the conclusions, that DBTM (dibutyltin maleate) gives the best results as UV stabilizer and that organotin mercaptoacetates give the worst results.

 Although the superiority of compositions 20 containing organotin maleates in the stabilization of PVC formulations subjected to weathering has been recognized for a long time, these compositions have not succeeded in fully satisfying market requirements because of many disadvantages.

25 Thus, British Patent 787,930 explains, from page 1, line 65 to page 2, line 3, that dibutyltin maleate is difficult to disperse and in addition generates, during the conversion of PVC, a volatile

maleic anhydride fraction which has lacrimatory and irritating effects on the people handling it. In order to overcome these faults, GB 787,930 provides compositions containing organotin maleate hemiesters which are liquid, such as dibutyltin bis(monobutyl maleate). The presence of the alcohol group of the ester has the effect of decreasing the tin content, causing reduced heat stabilization, but does not remove the volatile fraction generated during the stabilization mechanism, that is to say the formation of the hemiester of maleic acid by the reaction of organotin bis(monoalkyl maleate) with the gaseous HCl generated during the processing of PVC.

United States Patent 3,296,289 describes stabilizing compositions containing diorganotin maleate hemiester which are improved insofar as these compositions, composed of diorganotin bis(maleate hemiesters), are solid at room temperature. A typical composition from Patent US 3,296,289 was dibutyltin bis(cyclohexyl maleate), having a melting point of 71°C to 73°C, as described in Example 1. Although the solid compositions containing organotin maleate hemiester described in US 3,296,289 make it possible to solve some handling problems, they do not for all that remove the irritating volatile compounds generated when the stabilizing compositions are subjected to the heat conditions applied during the processing of PVC, which is carried out at approximately 200°C.

Patent US 3,555,060 relates to organic tin compounds, such as, in particular, dibutyltin bis(isooctyl maleate).

These compounds are complex chelates
5 exhibiting an $\equiv\text{Sn}-\text{O}-\text{Sn}\equiv$ bond with 2 tin atoms per molecule.

However, these products are viscous, coloured liquids which are difficult to use.

Another problem which is raised by the use of
10 organotin maleates as heat-stabilizer for vinyl chloride polymers is the difficulty of processing these formulations. Indeed, it is well known, in particular during the extrusion of rigid PVC, that the molten polymer containing the said stabilizers exhibits a
15 strong tendency to adhere to the conversion equipment.

To overcome this problem, use is made of sometimes complex combinations of lubricants and attempts are more rarely made to modify the structure of the stabilizer itself.

20 Thus, Patent GB 1,378,851 provides for the addition of a mixture of paraffin wax and of a lubricating acrylic polymer to a PVC formulation containing dibutyltin di(methyl maleate), so as to reduce the adhesion to the equipment. The improvement
25 thus obtained is characterized by a greater ease in detaching a film converted on a roll mill but, however, remains unsatisfactory.

However, the significant, indeed excessive,

addition of lubricants to PVC formulations can have disadvantages. Thus, an excessively high content of internal lubricant will have, inter alia, consequences of increasing the plasticization of the PVC and thus of decreasing its Vicat point. In the case of rigid formulations, this can be a factor which prohibits the use of organotin maleates.

In the same way, an excess of external lubricant, intended to limit problems of adhesion during extrusion, will have a tendency to promote phenomena of exudation and thus of deposits on the equipment.

Thus, despite more than thirty years of research, heat-stabilizers for PVC based on organotin maleates continue to have a limited use although they exhibit excellent stabilizing properties and very good resistance to weathering. Their properties cannot be regarded as entirely satisfactory:

- either because, during processing, they decompose to give volatile products which are irritating and lacrimatory,

- or because they are difficult to process.

A composition containing organotin maleates has now been found which can be obtained by reacting, optionally in solvent medium and/or in the presence of water, a mixture composed of at least one component RA and at least one component R¹A with maleic anhydride or maleic acid and by then bringing the reaction mixture

thus obtained into contact with at least one dialkyltin oxide $(R^2)_2Sn=O$ or at least one alkyltin chloride $(R^2)_xSnCl_{4-x}$, given that:

RA represents

5 - either an alcohol ROH in which R represents a linear or branched aliphatic hydrocarbon radical having a number of carbon atoms ranging from 1 to 10 and preferably of between 5 and 8 or a mixture of alcohols with a weight-average molecular mass $\overline{Mw}$ greater than 32 and not more than 158,

 - or an epoxyalkane $C_nH_{2n}O$ in which n ranges from 1 to 10 or a mixture of epoxyalkanes with a weight-average molecular mass $\overline{Mw}$ greater than 30 and not more than 156;

15 R^1A represents

 - either an alcohol R^1OH in which R^1 represents a linear or branched aliphatic hydrocarbon radical having a number of carbon atoms ranging from 20 to 50 or a mixture of saturated primary alcohols with a weight-average molecular mass $\overline{Mw}$ ranging from 298 to 718,

 - or an epoxyalkane $C_pH_{2p}O$ in which p ranges from 20 to 50 or a mixture of epoxyalkanes with a weight-average molecular mass $\overline{Mw}$ ranging from 296 to 716;

R^2 represents a linear or branched aliphatic hydrocarbon radical having a number of carbon atoms ranging from 1 to 12 and preferably equal to 1, 4 or 8,

x is an integer equal to 1 or 2.

The thermoplastic polymer in which the composition based on organotin maleates can be incorporated for the purpose of improving, in particular, the heat stability and the light stability thereof can in particular consist of one or more addition polymers chosen from the group formed by vinyl chloride homopolymers which can optionally be overchlorinated and the copolymers, optionally grafted, which result from the copolymerization of vinyl chloride with one or more ethylenically unsaturated comonomers. The following are in particular suitable as comonomers for the preparation of such copolymers: vinylidene halides, such as vinylidene chloride or fluoride, vinyl carboxylates, such as vinyl acetate, vinyl propionate or vinyl butyrate, acrylic and methacrylic acids and the nitriles, amides and alkyl esters which derive therefrom, in particular acrylonitrile, acrylamide, methacrylamide, methyl methacrylate, methyl acrylate, butyl acrylate, ethyl acrylate or 2-ethylhexyl acrylate, vinylaromatic derivatives, such as styrene or vinyl naphthalene, or olefins, such as bicyclo[2.2.1]hept-2-ene, bicyclo[2.2.1]hepta-2,5-diene, ethylene, propylene or 1-butene.

The invention very particularly relates, among these polymers, to homo- and copolymers of vinyl chloride which are optionally overchlorinated.

The composition based on organotin maleates according to the present invention can be prepared according to the preferred method below, which consists first in producing a homogeneous and liquid mixture of the components RA and R¹A in solvent medium while heating. The mixture obtained is brought to a temperature of at least 50°C and preferably of between 70°C and 100°C, maleic anhydride is then first added, continuously or portionwise, over a period of time which can range from 15 minutes to approximately 1 hour and then a dialkyltin oxide (R²)₂Sn=O is subsequently added, continuously or portionwise, over a period of time of at least 15 minutes and preferably ranging from 30 minutes to 90 minutes.

The reaction mixture is then maintained at a temperature ranging from 50°C to 120°C and preferably of between 75°C and 100°C for a period of time of at least 15 minutes and preferably of between 30 minutes and 90 minutes.

The water formed during the reaction and the solvent can be removed by distillation under reduced pressure at a temperature ranging from 70°C to 120°C and preferably of between 80°C and 100°C.

The solvent used must be inert with respect to the reactants and the products formed.

Mention will be made, by way of illustration of solvents which can be used according to the present invention, of toluene, xylenes, heptane or THF.

According to the present invention, the preparation is carried out with stirring and preferably while bubbling through an inert gas, such as nitrogen.

The preparation is generally carried out at
5 atmospheric pressure (10^5 Pa) but it would not be departing from the scope of the invention if the preparation were carried out at a different pressure.

Should epoxyalkanes be used, the preparation will be carried out in the presence of water. The
10 latter can advantageously be introduced before the addition of the maleic anhydride.

According to the present invention, the components RA and R^1A , and the maleic anhydride, will be used in a maleic anhydride/(RA + R^1A) molar ratio of at
15 least 1.5 and preferably of between 2 and 2.5.

According to the present invention, the ratio of the molar percentages RA/ R^1A can vary within wide limits. This ratio can range from 1/99 to 99/1 and preferably of between 20/80 and 80/20.

20 Mention will be made, by way of illustration of components RA which can be used according to the present invention, of alcohols ROH, such as methanol, ethanol, n-propanol, n-butanol, n-pentanol, n-hexanol, 2-ethylhexanol, n-octanol, n-decanol or a mixture of at
25 least two of the abovementioned alcohols; 1,2-epoxyalkanes $C_nH_{2n}O$, such as 1,2-epoxybutane, 1,2-epoxyhexane, 1,2-epoxyoctane, 1,2-epoxydecane or a mixture of at least 2 of the abovementioned

epoxyalkanes.

2-Ethylhexanol will preferably be used.

Mention will be made, by way of illustration of components R^1A which can be used according to the present invention, of alcohols R^1OH , such as 1-nonacosanol (montanyl alcohol), hentriacontanol, n-triacontanol (myricyl alcohol) or mixtures of saturated primary alcohols having a weight-average molecular mass $\overline{M}_w$ ranging from 375 to 700, a polydispersity $\overline{M}_w/\overline{M}_n$ in the region of 1 ($\overline{M}_n$ representing the number-average molecular mass) and a regular distribution of the linear aliphatic hydrocarbon radicals as a function of the length of the hydrocarbon chains, or mixtures of 1,2-epoxyalkanes having a weight-average molecular mass $\overline{M}_w$ ranging from 313 to 632.

Use will preferably be made of a mixture of saturated primary alcohols having a mass $\overline{M}_w$ of 460 or of a mixture of epoxyalkanes having a mass $\overline{M}_w$ of 632.

According to the present invention, at least 0.50 mol of dialkyltin oxide $(R^2)_2Sn=O$ and preferably an amount ranging from 0.75 mol to 0.85 mol will be used per 1 mol of maleic anhydride.

Mention will be made, by way of illustration of dialkyltin oxides which can be used according to the present invention, of dimethyltin oxide, dibutyltin oxide or dioctyltin oxide.

The use, among these compounds, of dibutyltin oxide is very particularly preferred.

It is also possible to use, during the preparation of the composition based on organotin maleates, at least one alkyltin chloride $(R^2)_xSnCl_{4-x}$ instead of a dialkyltin oxide $(R^2)_2Sn=O$.

5 Use will preferably be made of a mixture of monoalkyltin trichloride R^2SnCl_3 and of dialkyltin dichloride $(R^2)_2SnCl_2$, it being possible for the radicals R^2 of the trichloride and of the dichloride to be identical or different.

10 In this case, the chlorides formed during the preparation of the composition can be neutralized with an alkaline hydroxide.

 The composition based on organotin maleates according to the present invention has a tin content by weight of at least 10% and preferably of between 15% and 25%.

 In the general case where the compositions obtained are solid at room temperature, the latter can be isolated by means known to a person skilled in the art, such as, in particular, by pouring the hot liquid reaction mixture onto a cooled surface and then flaking the solidified product.

 It would not be departing from the scope of the invention if the composition containing organotin maleates was obtained by separately preparing two compositions containing organotin maleates which respectively contain low molecular weight components RA and high molecular weight components R^1A and by then

subsequently mixing them.

The tin content of the composition obtained can be determined by elemental analysis. The infrared spectrum of the composition according to the invention
5 is characterized by:

- an absorption band in the region of 1730 cm^{-1} characteristic of ester functional groups,
- an absorption band in the region of 1580 cm^{-1} characteristic of tin carboxylates,
- 10 - an absorption band in the region of 680 cm^{-1} characteristic of =Sn-O-Sn= bonds.

The composition based on organotin maleates can be obtained according to an alternative form which consists in introducing, after the removal of the water
15 formed and optionally of the solvent, an amount of costabilizer of at least 10% by weight and preferably of between 15% and 30%, with respect to the reactants employed (excluding water).

Having carried out this addition, the
20 reaction mixture is kept stirred and heated until a homogeneous mixture is obtained.

Mention will be made, by way of illustration of costabilizers which can be used according to the present invention, of zeolites, hydrotalcites or
25 calcium and zinc salts of fatty acids.

Another subject of the invention is a process for producing organotin maleates, characterized in that a mixture composed of at least one component RA and at

least one component R^1A is reacted with maleic anhydride and that the reaction mixture thus obtained is then brought into contact with at least one dialkyltin oxide $(R^2)_2Sn=O$ or with at least one alkyltin chloride

5 $(R^2)_xSnCl_{4-x}$.

RA , R^1A , $(R^2)_2Sn=O$ and $(R^2)_xSnCl_{4-x}$ have the same meaning as given above. The operating conditions and parameters, the reactants used and their proportions are as defined above.

10 The invention also relates to compositions comprising the thermoplastic polymer and the composition based on organotin maleates as they have been defined above.

The composition based on organotin maleates
15 can be used in amounts ranging from 0.5 to 5 parts by weight, preferably from 1 to 4 parts, per 100 parts by weight of thermoplastic polymer.

Such compositions can additionally contain, as a function in particular of the processing or
20 conversion conditions and/or of the applications for which they are intended, the usual additives, such as pigments, fillers, lubricants, processing aids, impact modifiers, antioxidants, plasticizers and blowing agents.

25 The composition based on tin as obtained according to the present invention can be incorporated at the same time [lacuna] or before the additives mentioned above, when they are made use of.

According to a particularly recommended form, this operation is carried out in a fast mixer and the thermoplastic polymer, the composition based on organotin maleates and then the additives and the
5 fillers are successively introduced.

As a general rule, this operation can be carried out at room temperature, it being possible for the operation itself to cause a temperature rise up to 70°C, indeed more.

10 The stabilizing action of the composition based on organotin maleates as obtained according to the present invention can be demonstrated by carrying out various tests which evaluate the dynamic and static thermal stability of thermoplastic polymer compositions
15 containing it.

Thermoplastic polymer compositions, in particular PVC compositions, comprising the composition based on organotin maleates as obtained according to the present invention can be moulded by injection,
20 calendered and then thermoformed, extruded or coextruded as rigid articles, such as interior covering for buildings, door frames, window sections, sheets and pipes. The converted articles can be compact or expanded.

25 The heat- and light-stabilizing action of the composition based on organotin [lacuna] can be demonstrated for the converted articles by measurement of the trichromaticity coordinates L^* , a^* and b^* , by

measurement of the residual stability time by a Congo Red test and by subjecting these articles to UV radiation.

The composition containing organotin maleates
5 as obtained according to the present invention exhibits the advantages, in addition to heat stabilizing thermoplastic polymer compositions containing it, of controlling the gelling and the lubrication thereof.

The composition containing organotin maleates
10 of the present invention makes it possible in particular to obtain controlled melting of thermoplastic polymer compositions containing it. This being obtained by varying the nature, the ratio and the total amount of the various reactants and more
15 particularly of the components RA and R¹A.

It has been observed that the high molecular weight components R¹A contribute a lubricating effect to thermoplastic polymer compositions, whereas the low molecular weight components RA make it possible to
20 promote gelling and to control the melting time. These characteristics can be demonstrated by studying the rheology of thermoplastic polymer compositions using a torsional couple rheometer or by reading the conversion parameters during processing. It has thus been found
25 that the lubricating effect contributed by the presence of high molecular weight component R¹A in the composition containing organotin maleates makes it possible to significantly reduce the addition of

lubricant to thermoplastic polymer compositions.

The composition containing organotin maleates as obtained according to the present invention also has the advantage of not detrimentally affecting the Vicat
5 point of converted thermoplastic polymer compositions.

Moreover, during the conversion of thermoplastic polymer compositions containing the said composition containing organotin maleates, and more particularly by extrusion, no release of irritating
10 products is observed.

It has also been observed that the use of the components RA and R¹A during the preparation of the composition containing organotin maleates according to the present invention made it possible to reduce,
15 indeed eliminate, the formation of deposits on the equipment used for the conversion of thermoplastic polymer compositions containing the said composition containing organotin maleates.

The following examples illustrate the
20 invention.

I Preparation of compositions containing organotin maleates in accordance with the invention

The compositions containing organotin maleates were prepared by using the following
25 reactants:

- Unilin® 425 (sold by the Company Petrolite): mixture of saturated primary alcohols

having a weight-average molecular mass $\overline{M}_w$ of 460, a polydispersity $\overline{M}_w/\overline{M}_n$ of 1, a melting point of 91°C, a hydroxyl number of 105 mg KOH/g of sample determined according to ASTM Standard D 222 and a mean number of
5 carbon atoms of 30;

- Unilin® 550 (sold by the Company Petrolite):

mixture of saturated primary alcohols having a weight-average molecular mass $\overline{M}_w$ of 550, a
10 polydispersity $\overline{M}_w/\overline{M}_n$ of 1, a melting point of 99°C, a hydroxyl number of 83 mg KOH/g of sample determined according to ATSM Standard D 222 and a mean number of carbon atoms of 40;

- Vikolox® C30+ (sold by the company Elf Atochem North America Inc.): mixture of 1,2-
15 epoxyalkanes having a weight [lacuna] molecular mass $\overline{M}_w$ of 632 and exhibiting a melting point of 72.7°C and a mean number of carbon atoms of 45;

- Valfor® 100 (sold by the company The PQ
20 Corporation): Zeolite A of sodium aluminosilicate hydrate type;

- 2-ethylhexanol,
- maleic anhydride,
- dibutyltin oxide.

EXAMPLE 1:**PREPARATION OF A COMPOSITION CONTAINING
ORGANOTIN MALEATES ACCORDING TO THE INVENTION IN
SOLVENT MEDIUM.**

5 The following are introduced into a 500 ml
jacketed reactor equipped with a stirrer, a thermometer
pocket, a reflux condenser equipped with a Dean and
Stark apparatus, and an inert gas inlet:

- 100 ml of heptane (solvent),
- 10 - 84.2 g (0.183 mol) of Unilin® 425,
- 5.9 g (0.045 mol) of 2-ethylhexanol,
- i.e. a ratio of the 2-ethylhexanol/Unilin®
425 molar percentages of 20/80.

 The mixture is heated to approximately 85°C
15 and stirred so as to obtain a homogeneous, liquid
reaction mixture. 53.6 g (0.546 mol) of maleic
anhydride are introduced in four portions over 30
minutes under a stream of nitrogen, causing a slight
exotherm to 95°C. When the temperature again stabilizes
20 at approximately 85°C, 109.5 g (0.440 mol) of
dibutyltin oxide are added in five portions over 1
hour. The reaction mixture is maintained with stirring
under a stream of nitrogen for 1 hour at 85°C. The
temperature is raised by 5°C to 10°C before applying a
25 pressure of 6.66×10^3 Pa in order to distil off the
heptane and the water of reaction (approximately 2 g of
water). After distilling for 1 hour, the reactor is
returned to atmospheric pressure. Heating and stirring

are halted and the reaction mixture is poured onto a cooled metal plate, so as to solidify it.

Approximately 248 g of the composition are obtained, i.e. a yield by weight in the region of 99%.

5 The percentage by weight of tin in the composition thus obtained is 20.8%. The infrared spectrum exhibits an absorption band at 1729 cm^{-1} corresponding to the ester functional groups, an absorption band at 1579 cm^{-1} corresponding to the tin carboxylate functional groups
10 $-\text{C}(\text{O})-\text{O}-\text{Sn}\equiv$ and an absorption band at 680 cm^{-1} characteristic of $\equiv\text{Sn}-\text{O}-\text{Sn}\equiv$ bonds.

EXAMPLE 2:

PREPARATION OF A COMPOSITION CONTAINING
ORGANOTIN MALEATES ACCORDING TO THE INVENTION USING A
15 ZEOLITE AS COSTABILIZER.

The preparation is carried out as in Example 1, using the same reactants but according to different amounts. The following are thus introduced:

- 100 ml of heptane (solvent),
- 20 - 79.7 g (0.173 mol) of Unilin[®] 425,
- 10.4 g (0.079 mol) of 2-ethylhexanol,
i.e. a ratio of the 2-ethylhexanol/Unilin[®]
425 molar percentages of approximately 30/70,
- 53.6 g (0.546 mol) of maleic anhydride,
- 25 - 109.5 g (0.440 mol) of dibutyltin oxide.

After the removal of the heptane and the water of reaction by distillation under reduced pressure, the reactor is returned to atmospheric

pressure and 64 g of Valfor® 100 are introduced in four portions over 30 minutes with stirring while maintaining the temperature at 85°C. After homogenizing the reaction mixture, the product is isolated as in

5 Example 1.

Approximately 315 g of the composition are obtained, i.e. a yield by weight in the region of 100%. The percentage by weight of tin in the composition thus obtained is 16.6%. The infrared spectrum exhibits bands
10 similar to those of the composition of Example 1.

EXAMPLE 3

PREPARATION OF A COMPOSITION ACCORDING TO THE INVENTION USING A MIXTURE OF 1,2-EPOXYALKANES IN THE PRESENCE OF WATER AND IN THE ABSENCE OF SOLVENT.

15 The preparation is carried out according to the operating conditions of Example 1, in an identical reactor, with the amounts of following reactants:

- 25 g of water,
- 79.7 g (0.126 mol) of Vikolox® C30+,
- 20 - 10.4 g (0.079 mol) of 2-ethylhexanol,
i.e. a ratio of the 2-ethylhexanol/Vikolox® C30+ molar percentages of approximately 40/60,
- 53.6 g (0.546 mol) of maleic anhydride,
- 109.5 g (0.440 mol) of dibutyltin oxide.

25 Distillation under reduced pressure is carried out similarly to Example 1 and 22 g of water are collected after 1 hour. 64 g of Valfor® 100 are then introduced into the reaction mixture as described

in Example 2.

The product is collected as in Examples 1 and 2. Approximately 319 g of the composition are obtained, i.e. a yield by weight in the region of 100%. The percentage by weight of tin in the composition thus obtained is 16.4%.

The infrared spectrum exhibits an absorption band at 1736 cm^{-1} corresponding to the ester functional groups, an absorption band at 1581 cm^{-1} corresponding to the tin carboxylate functional groups $-\text{C}(\text{O})-\text{O}-\text{Sn}\equiv$ and an absorption band at 681 cm^{-1} characteristic of $\equiv\text{Sn}-\text{O}-\text{Sn}\equiv$ bonds.

EXAMPLE 4:

PREPARATION OF A COMPOSITION ACCORDING TO THE INVENTION USING A MIXTURE OF 1,2-EPOXYALKANES IN THE PRESENCE OF WATER AND IN THE ABSENCE OF SOLVENT.

The preparation is carried out as in Example 3, using the same reactants but according to different amounts. The following are thus introduced:

- 25 g of water,
- 74.6 g (0.118 mol) of Vikolox® C30+,
- 15.4 g (0.118 mol) of 2-ethylhexanol,
- i.e a ratio of the 2-ethylhexanol/Vikolox® C30+ molar percentages of 50/50,
- 53.6 g (0.546 mol) of maleic anhydride,
- 109.5 g (0.440 mol) of dibutyltin oxide.

Distillation under reduced pressure is carried out similarly to Example 1 and 23 g of water

are collected after 1 hour. 64 g of Valfor® 100 are then introduced into the reaction mixture as described in Example 2.

The product is collected as in Examples 1, 2 and 3. Approximately 319 g of the composition are obtained, i.e. a yield by weight in the region of 100%. The percentage by weight of tin in the composition thus obtained is 16.4%. The infrared spectrum exhibits absorption bands similar to those of Example 3.

10

EXAMPLE 5:

PREPARATION OF A COMPOSITION ACCORDING TO THE INVENTION SIMILAR TO EXAMPLE 2 WITH VARIATION IN THE RATIO OF THE 2-ETHYLHEXANOL/UNILIN® 425 MOLAR PERCENTAGES.

15

The preparation is carried out as in Example 2, using the same reactants but according to different amounts. The following are thus introduced:

20

- 100 ml of heptane (solvent),
- 87.3 g (0.190 mol) of Unilin® 425,
- 2.8 g (0.022 mol) of 2-ethylhexanol,
- i.e. a ratio of the 2-ethylhexanol/Unilin® 425 molar percentages of 10/90,
- 53.6 g (0.546 mol) of maleic anhydride,
- 109.5 g (0.440 mol) of dibutyltin oxide.

25

After the distillation under reduced pressure of the heptane and the water formed, 64 g of Valfor® 100 are introduced, as in Example 2.

Approximately 315 g of the composition are

obtained, i.e. a yield by weight in the region of 100%.
The percentage by weight of tin in the composition thus
obtained is 16.6%. The infrared spectrum exhibits
absorption bands similar to those of the composition of
5 Example 2.

EXAMPLE 6:

**PREPARATION OF A COMPOSITION ACCORDING TO THE
INVENTION SIMILAR TO EXAMPLE 2 WITH VARIATION IN THE
RATIO OF THE ETHYLHEXANOL/UNILIN® 425 MOLAR
10 PERCENTAGES.**

The preparation is carried out as in Example
2, using the same reactants but according to different
amounts. The following are thus introduced:

- 100 ml of hexane (solvent),
- 15 - 70.2 g (0.153 mol) of Unilin® 425,
- 19.9 g (0.153 mol) of 2-ethylhexanol,
- i.e. a ratio of the 2-ethylhexanol/Unilin®
425 molar percentages of 50/50,
- 53.6 g (0.546 mol) of maleic anhydride,
- 20 - 109.5 g (0.440 mol) of dibutyltin oxide.

After the distillation under reduced pressure
of the heptane and the water formed, 64 g of Valfor®
100 are introduced, as in Example 2.

Approximately 314 g of the composition are
25 obtained, i.e. a yield by weight in the region of 100%.
The percentage by weight of tin in the composition thus
obtained is 16.6%. The infrared spectrum exhibits
absorption bands similar to those of the composition of

Example 2.

EXAMPLE 7:

**PREPARATION OF A COMPOSITION ACCORDING TO THE
INVENTION USING UNILIN® 550 AS REPLACEMENT FOR
5 UNILIN® 425.**

The preparation is carried out as in Example 2, using the reactants according to the following amounts:

- 100 ml of heptane (solvent),
- 10 - 82.5 g (0.150 mol) of Unilin® 550,
- 8.2 g (0.063 mol) of 2-ethylhexanol,
- i.e. a ratio of the 2-ethylhexanol/Unilin®
550 molar percentages of 30/70,
- 53.6 g (0.546 mol) of maleic anhydride,
- 15 - 109.5 g (0.440 mol) of dibutyltin oxide.

After the distillation under reduced pressure of the heptane and the water formed, 64 g of Valfor® 100 are introduced, as in Example 2.

Approximately 315 g of the composition are
20 obtained, i.e. a yield by weight in the region of 100%.
The percentage by weight of tin in the composition thus
obtained is 16.6%. The infrared spectrum exhibits
absorption bands similar to those of the composition of
Example 2.

25 **II Preparation of PVC compositions, herein-
after denoted by PVC formulations, containing the
compositions containing organotin maleates prepared
above and evaluation of the stabilizing and rheological**

properties of the said compositions containing organotin maleates.

The trade names, natures, suppliers and functions of the various materials used in the preparation of the PVC formulations are given below:

- Lacovyl S110P, PVC resin with a K value of 67, Elf Atochem S.A.,

- Durastrength 300, acrylic polymer, Ceca S.A., impact modifier,

10 - Stavinor CAPSE, calcium stearate, Ceca S.A., lubricant,

- Metablen P551, acrylic polymer, Metablen Company B.V., processing aid,

15 - Kronos S2220, titanium oxide, Kronos, pigment,

- Hydrocarb 95T, calcium carbonate, Omya S.A., filler,

- AC 316, oxidized polyethylene wax, Allied Signal, lubricant.

20 THE PVC FORMULATIONS ARE PREPARED ACCORDING TO THE FOLLOWING OPERATING CONDITIONS (THE AMOUNTS OF THE MATERIALS USED ARE EXPRESSED BY WEIGHT):

The following is introduced into a Henschel jacketed fast mixer with a stirring speed of 25 3800 revolutions/minute:

- 100 parts of PVC resin (Lacovyl S 110 P).

A rise in temperature is observed.

3.5 parts of a composition containing

organotin maleates as obtained according to the present invention are introduced at 60°C and then 0.2 part of oxidized polyethylene wax (AC 316) is introduced at 65°C.

5 The following are added at 80°C:

- 0.6 part of calcium stearate (Stavinor CA PSE),
- 7.5 parts of impact modifier (Durastrength 300) and,

10 - 1 part of processing aid (Metablen P 551).

The following are then added at 85°C:

- 4 parts of titanium oxide (Kronos S2220),
- 5 parts of calcium carbonate (Hydrocarb 95 T).

15 The PVC formulation is stirred at 3800 revolutions/minute until a temperature of 110°C is reached and then cooled by reducing the stirring to 1500 revolutions/minute and by circulating a stream of cold water in the jacket of the mixer.

20 The formulation obtained is collected when the temperature reaches approximately 45°C.

Two PVC formulations as prepared above are evaluated on a Haake-Rheocord 90 rheometer at 185°C at a speed of 45 revolutions/minute and with a load of 25 65 g.

The results are reported in Table 1.

The following designations are used in this Table 1:

- PVC formulation 1, a formulation containing the composition containing organotin maleates of Example 3;

5 - PVC formulation 2, a formulation containing the composition containing organotin maleates of Example 4;

- Degradation time, the time at the end of which a large rise in the torque is observed, related to crosslinking and thus to complete degradation of the
10 PVC.

	PVC formulation 1	PVC formulation 2
Melting time (min)	2.00	1.33
Torque at melting (N.m)	18.0	21.5
15 Torque at equilibrium (N.m)	17.0	20.0
Degradation time (min)	28.0	28.0

TABLE 1

The influence of the molar ratio of the low molecular weight alcohols or epoxides to the high
20 molecular weight alcohols or epoxides on the melting time and the viscosity (expressed by the torque) of the PVC formulation in the molten state is thus demonstrated.

On increasing the amount of low molecular

weight alcohol (2-ethylhexanol) with respect to the amount of high molecular weight 1,2-epoxyalkane (Vikolox® C30+), everything otherwise being equal, the melting time is significantly decreased.

- 5 Conversely, the increase in the amount of high molecular weight 1,2-epoxyalkane results in a significant decrease in the torque, demonstrating the lubricating effect contributed by the long chains.

10 The same formulations were converted on a Krauss-Maffei KMDL25 extruder equipped with a 30 x 3 mm flat die. The parameters recorded during the extrusion are presented in Table 2:

	PVC formulation 1	PVC formulation 2
Temperature profile (°C)	130 - 180 - 190 - 200	
Bulk temperature (°C)	190	191
15 Motor speed (rev/min)	15	15
Torque (%)	24	50
Bulk head pressure (bar)	120	134
Throughput (kg/h)	4.0	4.0

TABLE 2

- 20 The following evaluations were carried out on the strips extruded:

- Colorimetry (measurement of the trichromaticity coordinates L*, a* and b* and

calculation of the yellowing indices (YI) and of the whiteness indices (WI) according to ASTM Standard E313) using a Minolta Chromameter CR200 spectrophotometer,

- Vicat point or softening temperature,
5 measured using an HDT Vicat Tester AFS,
- Residual stability expressed in minutes,
measured using a Liebisch Congo Red bench.

The results are presented in Table 3:

	PVC formulation 1	PVC formulation 2
10 L*	94.3	94.0
a*	-0.74	-0.71
b*	2.50	2.37
YI	3.7	3.6
WI	73.0	73.1
Vicat Point (°C)	79.4	79.4
15 Congo Red (min)	44	40

TABLE 3

The influence of the ratio of the molar percentages of the low molecular weight alcohols or-1,2-epoxyalkanes to the high molecular weight
20 alcohols or 1,2-epoxyalkanes on the lubrication of the PVC formulation is again demonstrated, as shown by the torque and bulk pressure values obtained (Table 2).

The colorations of the two PVC formulations

extruded testify to the effectiveness of the compositions containing organotin maleates obtained according to the present invention in preventing the heat degradation of PVC during the processing thereof.

5 It is also shown that these compositions containing organotin maleates make it possible to retain a high Vicat point level, in accordance with the applicational requirements.

A dynamic thermal stability test is carried
10 out on three other PVC formulations prepared like the PVC formulations 1 and 2.

The following designations are used:

- PVC formulation 3, a formulation containing the composition containing organotin maleates of
15 Example 5;

- PVC formulation 4, a formulation containing the composition containing organotin maleates of Example 2;

- PVC formulation 5, a formulation containing
20 the composition containing organotin maleates of Example 6.

This dynamic thermal stability test makes it possible to demonstrate the remarkable stabilizing effect contributed by the compositions described,
25 varying via the molar ratio of the species RA and R¹A of low and high molecular weight.

150 g of the PVC formulations 3, 4 and 5 are evaluated using a Collin roll mill, the rollers of

which are brought to 200°C. The rotational speeds of the two cylinders are respectively adjusted to 20 rev/min and 24 rev/min, so as to gel and then squeeze the material between the cylinders while contributing frictional mechanical work. The separation between the cylinders is adjusted to 0.7 mm.

Samples are withdrawn from the cylinders at regular time intervals, their coloration being recorded, until complete degradation.

The yellowing indices (YI, ASTM Standard E313) measured on each sample withdrawn and the degradation time corresponding to complete darkening are presented in Table 4.

	PVC for- mulation 3	PVC for- mulation 4	PVC for- mulation 5
YI (2 min)	5.7	5.4	4.8
YI (4 min)	8.4	7.4	7.3
YI (6 min)	10.0	8.9	9.1
YI (8 min)	11.8	10.0	10.2
YI (10 min)	13.4	11.5	11.9
YI (12 min)	15.2	13.0	12.7
YI (15 min)	18.5	15.7	15.3
YI (20 min)	27.3	23.4	21.3
YI (25 min)	33.9	32.3	30.5
YI (30 min)	38.6	37.0	36.1
Complete degradation (min)	45	45	50

TABLE 4

III EVALUATION OF THE FINAL PROPERTIES OF
RIGID PVC SECTIONS FORMED FROM PVC FORMULATIONS
CONTAINING THE COMPOSITIONS CONTAINING ORGANOTIN
MALEATES.

5 In order to demonstrate the final properties
of rigid sections heat-stabilized by the compositions
containing organotin maleates described in the
invention, a rigid PVC formulation stabilized using
lead salts has been extruded in the same way as the PVC
10 formulation 4 and the PVC formulation 6 (PVC
formulation containing the composition containing
organotin maleates of Example 7). Lead salts form a
stabilizing system, well known to a person skilled in
the art, which represents a reference, the
15 effectiveness of which is proved by its extensive use
by converters of rigid sections.

To do this, a PVC formulation 7 is prepared
according to the procedure used to prepare the PVC
formulations 1 to 6, except that the 3.5 parts of
20 composition containing organotin maleates are replaced
with a combination of 3.5 parts of dibasic lead
phosphite ($2\text{PbO} \cdot \text{PbHPO}_3 \cdot 1/2\text{H}_2\text{O}$) and 0.5 part of dibasic
lead stearate ($2\text{PbO} \cdot \text{Pb}(\text{OOC}_{17}\text{H}_{35})_2$).

A dynamic thermal stability test makes it
25 possible to demonstrate the remarkable stabilizing
effect contributed by the compositions described:

150 g of the PVC formulations 4, 6 and 7 are
evaluated using a Collin roll mill according to the

procedure described above (evaluation of the PVC formulations 3, 4 and 5, Table 4). The yellowing indices (YI, ASTM Standard E313) measured on each sample withdrawn are presented in Table 5.

		PVC for- mulation 4	PVC for- mulation 6	PVC for- mulation 7 (reference)
5	YI (2 min)	5.1	5.6	5.4
	YI (4 min)	7.5	8.6	7.7
	YI (6 min)	9.3	10.8	8.7
	YI (8 min)	10.7	12.2	9.2
	YI (10 min)	12.0	13.3	9.6
10	YI (15 min)	15.0	17.0	11.0

TABLE 5

The three PVC formulations 4, 6 and 7 have also been converted on a Reifenhäuser BT65 industrial extruder equipped with a T-shaped die for window sections.

The parameters recorded during the extrusion are presented in Table 6.

		PVC for- mulation 4	PVC for- mulation 6	PVC for- mulation 7 (reference)	
5	Temperature profile (°C)				
	In the barrel	zone 1	180	180	170
		zone 2	180	185	175
		zone 3	180	180	170
		zone 4	180	180	175
	in the die	zone 8	185	185	185
zone 9		190	190	190	
10	Screw temperature (°C)		140	140	140
	Screw speed (rev/min)		25	25	25
	Covering of the screws		yes	yes	yes
	Bulk temperature (°C)		190	181	187
	Amperage		7.3	5.7	7.5
	Throughput (kg/h)		62.8	58.6	59.8

TABLE 6

15 The parameters recorded during extrusion show that the behaviour of the PVC formulation 4 (containing the composition of Example 2 obtained with Unilin® 425) is very similar to that of the PVC formulation 7 (containing the stabilization system with lead).

20 On the other hand, the PVC formulation 6 (containing the composition of Example 7 obtained with

Unilin® 550) has a significantly different rheological behaviour, once again demonstrating the lubricating effect of high molecular weight alcohols. This formulation would require the lubrication to be optimized in order to produce rheological behaviour similar to that of the PVC formulation 7 containing lead.

The sections extruded from the PVC formulations 4 and 7 were evaluated and the results are presented in Table 7.

	PVC formulation 4	PVC formulation 7 (reference)
Coloration of sections		
L*	93.8	94.8
a*	-0.43	-0.36
b*	2.85	2.26
60° Gloss	24.6	15.7
Residual stability (min) (Congo Red)	38	60
Softening temperature (°C) (Vicat Point)	79.7	80.3

TABLE 7

The mechanical properties of the sections were evaluated by measuring the Charpy impact

resilience according to DIN Standard 53753, used in Germany for the evaluation of rigid sections for external joinery. The results are presented in Table 8.

	PVC formulation 4	PVC formulation 7 (reference)
Resilience <R> in kJ/m ²	60.5	62.6

5

TABLE 8

The effectiveness of the compositions containing organotin maleates described in the invention as heat-stabilizer during the processing on industrial equipment of opaque rigid PVC formulations intended for the manufacture of extruded sections is thus demonstrated.

Indeed, they make it possible to ensure easy processing of the formulations, as is shown by the extrusion parameters of the PVC formulation 4, which are very similar to those obtained for a reference formulation stabilized with lead salts, in particular as regards the amperage of the motor for driving the screws and the throughput of the extruder.

The colorations of the sections obtained with the PVC formulation 4, which are similar to those observed with the reference formulation containing lead, show the effectiveness of the compounds based on organotin maleates in heat-stabilizing the formulations and thus reducing the heat degradation of PVC during

the processing thereof.

Finally, it is noted that the Charpy impact resilience value obtained for the sections stabilized with the composition based on organotin maleates (PVC
5 formulation 4) is greater than the value of 50 kJ/m² recommended by DIN Standard 53753.

CLAIMS

1. Composition containing organotin
maleates as obtained by the reaction of a mixture
composed of at least one component RA and at least one
5 component R^1A with maleic anhydride and then bringing
the reaction mixture thus obtained into contact with at
least one dialkyltin oxide $(R^2)_2Sn=O$ or with at least
one alkyltin chloride $(R^2)_xSnCl_{4-x}$, given that:

RA represents

10 - either an alcohol ROH in which R represents
a linear or branched aliphatic hydrocarbon radical
having a number of carbon atoms ranging from 1 to 10
and preferably of between 5 and 8 or a mixture of
alcohols with a weight-average molecular mass $\overline{Mw}$
15 greater than 32 and not more than 158,

- or an epoxyalkane $C_nH_{2n}O$ in which n ranges
from 1 to 10 or a mixture of epoxyalkanes with a
weight-average molecular mass $\overline{Mw}$ greater than 30 and
not more than 156;

20 R^1A represents

- either an alcohol R^1OH in which R^1
represents a linear or branched aliphatic hydrocarbon
radical having a number of carbon [lacuna] ranging from
20 to 50 or a mixture of saturated primary alcohols
25 with a weight-average molecular mass $\overline{Mw}$ ranging from
298 to 718,

- or an epoxyalkane $C_pH_{2p}O$ in which p ranges



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from 20 to 50 or a mixture of epoxyalkanes with a weight-average molecular mass $\overline{M}_w$ ranging from 296 to 716;

R^2 represents a linear or branched aliphatic hydrocarbon radical having a number of carbon atoms ranging from 1 to 12 and preferably equal to 1, 4 or 8, x is an integer equal to 1 or 2.

2. Composition according to Claim 1, characterized in that the reaction takes place in the presence of a solvent.

3. Composition according to Claim 2, characterized in that it is prepared according to the following stages, which consist in producing a mixture of the components RA and R^1A in solvent medium while heating, in then adding to the mixture thus obtained, which has been brought to a temperature of at least 50°C and preferably of between 70°C and 100°C, maleic anhydride over a period of time ranging from 15 minutes to 1 hour, and in then subsequently adding a dialkyltin oxide $(R^2)_2Sn=O$ over a period of time ranging from 15 minutes to 90 minutes; in heating the reaction mixture thus obtained at a temperature ranging from 50°C to 120°C and preferably of between 75°C and 100°C for a period of time ranging from 15 minutes to 90 minutes and in removing the water formed and the solvent under reduced pressure at a temperature ranging from 70°C to 120°C and preferably of between 80°C and 100°C.

4. Composition according to one of Claims 1



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to 3, characterized in that the maleic anhydride/(RA + R¹A) molar ratio is at least 1.5 and preferably between 2 and 2.5.

5. Composition according to one of Claims 1 to 3, characterized in that the ratio of the molar percentages RA/R¹A ranges from 1/99 to 99/1 and preferably of between 20/80 and 80/20.

6. Composition according to one of Claims 1 to 3, characterized in that at least 0.50 mol of dialkyltin oxide (R²)₂Sn=O and preferably an amount ranging from 0.75 mol to 0.85 mol is used per 1 mol of maleic anhydride.

7. Composition according to one of Claims 1 to 5, characterized in that the compound RA is an alcohol ROH.

8. Composition according to Claim 7, characterized in that the alcohol ROH is 2-ethylhexanol.

9. Composition according to one of Claims 1 to 5, characterized in that the compound R¹A is a mixture of saturated primary alcohols having a weight-average molecular mass $\overline{M_w}$ ranging from 375 to 700 and a polydispersity $\overline{M_w}/\overline{M_n}$ in the region of 1 and preferably a mixture of alcohols having a mass $\overline{M_w}$ of 460 or 550.

10. Composition according to one of Claims 1 to 5, characterized in that the compound R¹A is a mixture of epoxyalkanes C_pH_{2p}O having a weight [lacuna]



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molecular mass $\overline{M_w}$ ranging from 313 to 632 and preferably a mixture of epoxyalkanes having a mass $\overline{M_w}$ of 632.

11. Composition according to one of Claims
5 according to 1 to 3 and 6, characterized in that the dialkyltin oxide is dibutyltin oxide.

12. Composition according to Claim 1,
characterized in that a mixture of monoalkyltin trichloride R^2SnCl_3 , and of dialkyltin dichloride
10 $(R^2)_2SnCl_2$ is used.

13. Composition according to one of Claims 1 to 12, characterized in that the reaction takes place in the presence of water.

14. Composition according to one of Claims 1
15 to 13, characterized in that a costabilizer is additionally used.

15. Composition according to Claim 14, characterized in that the costabilizer is a zeolite.

16. Process for producing a composition
20 containing organotin maleates, characterized in that a mixture composed of at least one component RA and at least one component R^1A is reacted with maleic anhydride and that the reaction mixture thus obtained is then brought into contact with at least one dialkyltin oxide
25 $(R^2)_2Sn=O$ or with at least one alkyltin chloride $(R^2)_xSnCl_{4-x}$, given that:

RA represents

- either an alcohol ROH in which R represents



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a linear or branched aliphatic hydrocarbon radical having a number of carbon atoms ranging from 1 to 10 and preferably of between 5 and 8 or a mixture of alcohols ROH with a weight-average molecular mass $\overline{Mw}$ greater than 32 and not more than 158,

- or an epoxyalkane $C_nH_{2n}O$ in which n ranges from 1 to 10 or a mixture of epoxyalkanes with a weight-average molecular mass $\overline{Mw}$ greater than 30 and not more than 156;

10 R^1A represents

- either an alcohol R^1OH in which R^1 represents a linear or branched aliphatic hydrocarbon radical having a number of carbon [lacuna] ranging from 20 to 50 or a mixture of saturated primary alcohols with a weight-average molecular mass $\overline{Mw}$ ranging from 298 to 718,

- or an epoxyalkane $C_pH_{2p}O$ in which p ranges from 20 to 50 or a mixture of epoxyalkanes with a weight-average molecular mass $\overline{Mw}$ ranging from 296 to 716;

R^2 represents a linear or branched aliphatic hydrocarbon radical having a number of carbon atoms ranging from 1 to 12 and preferably equal to 1, 4 or 8,

x is an integer equal to 1 or 2.

25 17. Process according to Claim 16, characterized in that the operation is carried out in solvent medium.

18. Process according to Claim 17,





characterized in that a mixture of the components RA and R¹A is prepared in solvent medium while heating, that maleic anhydride is then added over a period of time ranging from 15 minutes to 1 hour to the mixture thus obtained, which has been brought to a temperature of at least 50°C and preferably of between 70°C and 100°C, and that a dialkyltin oxide (R²)₂Sn=O is then subsequently added over a period of time ranging from 15 minutes to 90 minutes; that the reaction mixture thus obtained is heated at a temperature ranging from 50°C to 120°C and preferably of between 75°C and 100°C for a period of time ranging from 15 minutes to 90 minutes and that the water formed and the solvent are removed under reduced pressure at a temperature ranging from 70°C to 120°C and preferably of between 80°C and 100°C.

19. Process according to one of Claims 16 to 18, characterized in that the maleic anhydride/(RA + R¹A) molar ratio is at least 1.5 and preferably between 2 and 2.5.

20. Process according to one of Claims 16 to 18, characterized in that the ratio of the molar percentages RA/R¹A ranges from 1/99 to 99/1 and preferably of between 20/80 and 80/20.

21. Process according to one of Claims 16 to 18, characterized in that at least 0.50 mol of dialkyltin oxide (R²)₂Sn=O and preferably an amount ranging from 0.75 mol to 0.85 mol is used per 1 mol of



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maleic anhydride.

22. Process according to one of Claims 16 to 20, characterized in that the compound RA is an alcohol ROH.

5 23. Process according to Claim 22, characterized in that the alcohol ROH is 2-ethylhexanol.

24. Process according to one of Claims 16 to 20, characterized in that the compound R¹A is a mixture
10 of saturated primary alcohols having a weight-average molecular mass $\overline{M}_w$ ranging from 375 to 700 and a polydispersity $\overline{M}_w/\overline{M}_n$ in the region of 1 and preferably a mixture of alcohols having a mass $\overline{M}_w$ of 460 or 550.

25. Process according to one of Claims 16 to
15 20, characterized in that the compound R¹A is a mixture of epoxyalkanes C_pH_{2p}O having a weight [lacuna] molecular mass $\overline{M}_w$ ranging from 313 to 632 and preferably a mixture of epoxyalkanes having a mass $\overline{M}_w$ of 632.

20 26. Process according to one of Claims 16 to 18 and 21, characterized in that the dialkyltin oxide is dibutyltin oxide.

27. Process according to Claim 16, characterized in that the operation is carried out with
25 a mixture of monoalkyltin trichloride R²SnCl₃ and of dialkyltin dichloride (R²)₂SnCl₂.

28. Process according to one of Claims 16 to 27, characterized in that the operation is carried out



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in the presence of water.

29. Process according to one of Claims 16 to 28, characterized in that the operation is carried out in the presence of a costabilizer.

5 30. Process according to Claim 29, characterized in that the costabilizer is a zeolite.

31. Stabilized and lubricated thermoplastic polymer composition including a composition containing organotin maleates according to one of Claims 1 to 15.

10 32. Composition according to Claim 31, characterized in that the thermoplastic polymer is a homo- or copolymer of vinyl chloride.

33. Compact extruded rigid article, characterized in that it is formed from a thermoplastic
15 polymer composition according to either of Claims 31 and 32.



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