DESCRIPTION

High Efficiency Electrode System for Iontophoresis

TECHNICAL FIELD

The present invention relates to a novel and high efficient electrode system for iontophoresis as a system for transdermal administration of a pharmaceutical. This system combines a special electroconductive material and functional polymer so as to chemically trap ions generated on the surface of an applicator electrode at the time of operation, to act itself as a polymer electrode, to restrict the amount of coexisting competing ions generated into aqueous pharmaceutical solution in contact with the surface of skin of a subject, and to enhance the efficiency of electrical delivery of the desired pharmaceutical, that is, the delivery rate, and therefore, is expected to be highly useful in iontophoretic administration of pharmaceuticals or in cosmetics treatments.

BACKGROUND ART

Iontophoresis is a technique for enhancing the transdermal permeation of ionized substances by a weak current. In recent years, advances in biotechnology have led to the successful development of numerous peptides or proteins useful as pharmaceuticals, such as, insulins, calcitonin, growth hormones, erythropoietin, etc. These peptide pharmaceuticals exhibit high efficacy in small amounts, but generally injection rather than oral administration is used for the application due to the problems of the activity of the proteinase in the gastrointestinal tract. In order to eliminate the problems associated with the injections, that is, pain, troublesome, compliance, etc. and to cope with the fundamentals of pharmacology, the research and development of the iontophoretic transdermal drug delivery system became again active in the world. Iontophoresis devices are now available with small in

size, portable on adhesively attachable style having the nature of safe, which are free from the feeling of electricity even during operation, and flexibility for drug delivery programs. A depolarized high-frequency pulse iontophoresis device is disclosed in, for example, Sasaki M. et al, U.S. Patent No. 4,764,164.

Generally two electrodes are used in iontophoresis devices. That is, one electrode is a donor electrode which contains the ionized pharmaceutical and which causes the pharmaceutical to permeate into the body by application of a current. The other electrode is a counter electrode, which is separating placed on the skin from the other to form a circuit. When the pharmaceutical which is to be delivered has a positive charge (plus charge), the anodic electrode acts as a donor electrode and the cathode functions as a counter electrode to form a circuit. Conversely, when the pharmaceutical has a negative charge (minus charge), the cathode is a donor electrode and the anode becomes a counter electrode. In order to deliver a certain pharmaceutical by iontophoresis, a reservoir containing an ionized pharmaceutical is required. Various patents applications have been filed for applicators containing electrodes and reservoirs. In all cases, however, either there is lack in efficiency to eliminate the coexisting substance or else the construction of application has to be complicated.

During the iontophoresis operation, positively-charged ions are generated at the anodic side by the electrode oxidation reaction and negatively-charged ions are generated at the cathodic side by an electrode reduction reaction. The electrodes may be roughly divided into the following three types:

1) Inert electrodes (for example, carbon electrodes)

Anode: $2H_2O \longrightarrow 4H^+ + O_2 + 4e^-$

Cathode: $2H_2O + 4e^- \longrightarrow H_2 + 2OH^-$

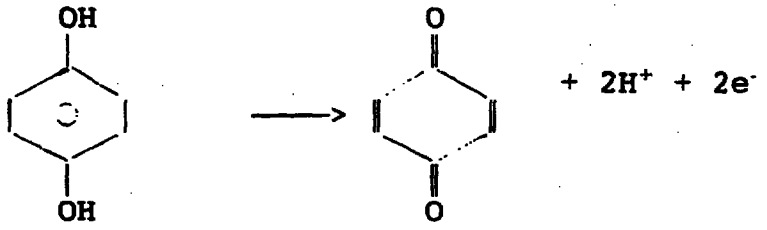
2) Active electrodes (Ag/AgCl electrodes etc.)

Anode: $Ag \longrightarrow Ag^+ + e^- \xrightarrow{+Cl^-} AgCl + e^-$

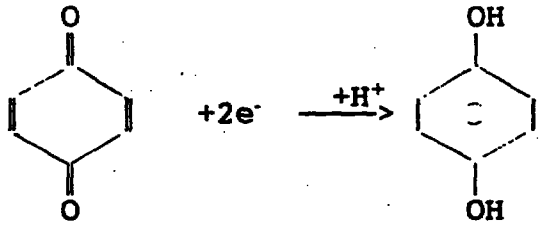
Cathode: $AgCl + e^- \longrightarrow Ag + Cl^-$

3) Organic oxidation reduction electrodes (quinone, aminal, etc.)

Anode:



Cathode:



There is a quantitative relationship between the current applied and the ions generated. This is shown by the Faraday equation:

$Q = F \times W/Me$

where, Q: amount of current

F: Faraday constant

W: Weight

Me: Molecular equivalent

During iontophoresis, the ions newly generated ions from the electrode compete with the drug ions electrically and the delivery rate of the drug (i.e., transport efficiency) is reduced inevitably. In order to solve this problem of the decrease in the drug delivery efficiency, three methods have been reported up to now.

The first method is to use an ion exchange membrane to separate the electrode space and the pharmaceutical space in the applicator structure, whereby the effects of the competing ions are eliminated (see Sanderson J.E.,

U.S. Patent No. 828,794 (1986)). The second method is to first exchange the pharmaceutical with an ion exchange resin and then exchanged with the hydrogen ion H^+ or OH^- generated during the operation, whereby the pharmaceutical is released from the ion exchange resin and the competing ions are removed (see Petelenz T.J., U.S. Patent No. 4,915,685 (1990)). The third method is to form a precipitate by a combination of a special metallic electrode and pharmaceutical or to cause neutralization, whereby the coexisting competing ions are removed (see Unterrecker, European Patent No. 0182520). These methods all make the construction of the applicator complicated, and raise the cost of manufacturing, and are unsuitable for practical use in many respects. Furthermore, in the third method combinations are limited for specific electrodes and pharmaceuticals (for example, iontophoresis of peptides by its acetates salt, etc.).

DISCLOSURE OF INVENTION

As explained above, to solve the problems and disadvantages in iontophoresis, the objects of the present invention are to eliminate the ions generated in the iontophoresis electrode applicator and to improve the efficiency of skin permeation of the drug and, further, to simplify the construction of the drug applicator.

In accordance with the present inventions, there is provided an electrode system for iontophoresis comprising a mixture of an electroconductive material and a functional polymer having a group of ammonium hydroxide or a quaternary ammonium halogen salt, sulfonic group, carboxylic group, amine group.

In accordance with the present invention, there is also provided an applicator for administering a pharmaceutical by iontophoresis using the above-mentioned electrode system.

BRIEF DESCRIPTION OF THE DRAWINGS

The present invention will be better understood from

the description set forth below with reference to the accompanying drawings, in which:

Fig. 1 is a view for explaining the efficiency of the present invention;

Figs. 2(A), 2(B) and 2(C) are views showing an embodiment of the present invention;

Fig. 3 is a graph for explaining the present invention under an applied DC in vitro;

Fig. 4 is a graph explaining the efficiency of the present invention under a pulse depolarizing current (ADIS 4030) in vitro; and

Fig. 5 is a graph for explaining the present invention using ADIS 4030 in vivo.

BEST MODE FOR CARRYING OUT THE INVENTION

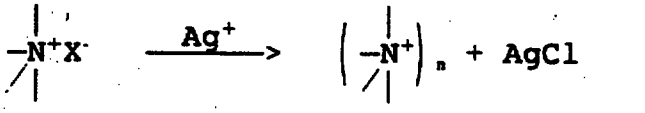
In order to solve the above-mentioned problems, the present inventors have intensively studied to thereby complete the following invention. That is, as shown in Fig. 1, a composite of an electroconductive material and a polymer was structured so that the ionized charge generated at the surface of the electroconductive material due to the operation be allowed to transfer to the electroconductive polymer and to provide a charge to the polymer, whereby the polymer per se functions as an electrode. That is, the competition is eliminated by immobilizing the charged ions competing with the pharmaceutical ions generated by electrolysis at the surface of the electrode. That is, the newly generated ions are eliminated by counter ions present in the polymer. Two approaches are used to achieve this object. One is to allow the ions to be precipitated, while the other is to allow to form a non-releasable covalent compound.

<u>First Embodiment</u>

In the case where the iontophoretic pharmaceutical-containing electrode (i.e., donor electrode) is an anodic active electrode and where a silver (Ag) electrode is selected.

Efficiency can be increased by using a polymeric quaternary ammonium salt having a chlorine ion (Cl^-) source on the surface of the silver electrode. That is, the silver ions (Ag^+) generated from the surface of the electrode due to the operation react with the chlorine ions (Cl^-) present in the polymer matrix and precipitate as AgCl and simultaneously the polymer takes on a positive charge. In this way, a difference in potential of the ion concentrations occurs and therefore a state close to one free of competing ions is approached. Therefore a high efficiency of delivery of the pharmaceutical can be obtained compared with the conventional iontophoresis electrode system.

Here, instead of the chlorine ions (Cl^-), a similar precipitate can be obtained by using iodine ions (I^-), bromine ions (Br^-), and other halogen ions and a similar effect of raising the delivery rate can be obtained.

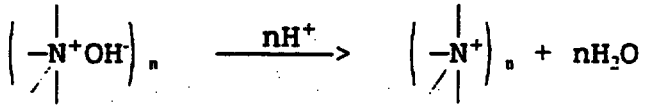


X = I, Br, Cl

Second Embodiment

In the case where the iontophoretic pharmaceutical administering electrode (i.e., donor electrode) is an anodic inert electrode.

It is possible to improve the efficiency of iontophoretic delivery of the pharmaceutical by disposing a polymer having a hydroxy ion (OH^-) source around the surface of the inert electrode. Thus, it is possible to have the ions (H^+) generated by the operation react with the hydroxy ions (OH^-) in the polymer matrix whereby the coexisting competing ions can be removed and the rate of permeation of the pharmaceutical can be increased.



Third Embodiment

In the case where the iontophoretic pharmaceutical administering electrode (i.e., donor electrode) is a cathode and where the active electrode is a silver chloride (AgCl) electrode.

Since a chlorine ion (Cl^-) is generated from the silver chloride electrode and becomes the ion competing with the negatively ionized pharmaceutical, for example, by coating a polymer (R-COOAg) having -COOAg group as a functional group, the released silver ion (Ag^+) reacts with the chlorine ion (Cl^-) released from the electrode, and therefore, the competing chlorine ion (Cl^-) can be removed and the rate of permeation of the ionic pharmaceutical (i.e., delivery rate) can be increased.

$$(R\text{—}COOAg)_n \xrightarrow{+nCl^-} (R\text{—}COO^-)_n + nAgCl$$

wherein R represents backbone of hydrocarbon polymers.

Forth Embodiment

In the case where the iontophoretic pharmaceutical administering electrode (i.e., donor electrode) is an inert electrode of carbon.

By using a polymer having simultaneously a tertiary amine and sulfonic group, the proton (H^+) or hydroxy ion (OH^-) generated at the surface of the electrode is trapped and the competing ion is removed. Therefore, the rate of permeation of the ionized pharmaceutical (i.e., rate of delivery) can be improved.

Anode: $(NR_3)_n \xrightarrow{+nH^+} (NH^+R_3)_n$

Cathode: $(R\text{—}SO_3H)_n \xrightarrow{+nOH^-} (R\text{—}SO_3^- + H_2O)_n$

EXAMPLES

The present invention will now be further illustrated by, but is by no means limited to, the following Examples, wherein all percentages are expressed on a weight basis unless otherwise noted.

Figures 2(A), (B) and (C) are a view of an embodiment of the present invention, wherein Fig. 2(A) is a central sectional view, Fig. 2(B) is a schematic view

of the electrode portion, and Fig. 2(C) is a top plan view.

In Fig. 2(A), (1) is an electrode portion. As shown in Fig. 2(B), this is a mixture of electroconductive granules (here, silver granules) (6) and polymer granules (7). (2) is a gel member, which is comprised of PVA, etc. (3) is a contact member for contact with the surface of the skin of the subject and is formed by a unwoven fabric etc. (4) is a tape, which is comprised of a non-electroconductive member. The tape (4) is disposed so as to support the electrode portion (1), gel member (2), and contact member (3) from the edges and, for example, is formed by Microform (manufactured by 3M). (5) is an adhesive member, which is provided so as to adhesively affix the entire assembly on the surface of the skin of the subject and for example is made of Bienderm (manufactured by 3M). (8) is a support member, which is comprised of a non-electroconductive substrate and supports the electrode portion (1) and tape (4). (9) is an electroconductive terminal which is connected with the electrode portion (1) and forms an input portion for electrically connecting the electrode portion (1) and the external iontophoretic electric output unit (not shown). Further, it is possible, without using the electroconductive terminal (9), to dispose the iontophoretic electric output unit inside and make it integral. The pharmaceutical to be administered may be disposed in one or both of the gel member (2) and contact member (3) or on the surface of the same, although this is not particularly limited.

Fabrication of Film Electrode

An 8.5 g amount of silver ink (DW250H-5 (manufactured by Toyobo)), 1.5 g of ion exchange resin having a group of quaternary ammonium salt and made into microgranules (1X-8 (manufactured by Dowex)), and 1.0 g of a diluent (YC-180 (manufactured by Toyobo)) were mixed well and spread to a constant thickness on a PET film

(thickness of 0.1 mm) using a doctor blade to print it. Further, this was preheated to 65°C for 30 minutes, then heated at 150°C for 30 minutes to remove the solvent and to obtain a new film electrode of a thickness of 128 μm.

On the other hand, the conventional electrode was made in the same manner by mixing 4.27 g of silver ink (DW250H-5 (manufactured by Toyobo)), 0.12 g of salt microgranules (average diameter of granules of 100 μm), and 0.5 g of a diluent (YC-180 (manufactured by Toyobo)) and coating the result on a PET film.

The above electrodes were cut in accordance with the shape of the applicator and used for the evaluation of the electrode system efficacy.

Evaluation of Efficacy in Vitro (Method of Testing Permeation Through Excised Skin of SD Rats)

The efficacy of the electrodes was evaluated in a test of permeation of TRH (thyrotropin-releasing hormone) through skin excised from rats. SD rats were anaesthetized with ether and dilapidated, the skin of the stomach portion was removed, and the fatty tissue of the skin was cleanly removed using tweezers to make the excised skin. This was allowed to stand in physiological saline at 4°C for 15 hours, then was cut into pieces of 3 × 3 cm (9 cm^2) size which were then affixed to side-by-side dispersion cells (manufactured by Advance). The donor chamber was filled with 0.1 percent TRH acetate solution (1 ml) rinsed by distilled water. The receiver chamber (3.5 ml) was filled with a 3-fold dilution of physiological saline, then was left for 60 minutes without operation, then subject to operation (electrical conditions: DC, 0.39 mA/cm^2) for 60 minutes, then samples (0.2 ml) were taken from the receiver solution over time over a period of 120 minutes and the amount of TRH permeated through the film was measured by HPLC. After samples were taken, fresh 3-fold diluted physiological saline was supplemented. The measurement conditions of

the HPLC were an HLPC apparatus of Model 6A (manufactured by Shimadzu Seisakusho), an immobilized phase of Inertsil ODS-2 4.6 × 150 mm, an eluent of 5 mM IPC-SDS aqueous solution: methanol = 40:60 (pH = 3.6, adjusted with phosphoric acid).

Figure 3 shows the relationship between the amount of permeation of the TRH acetate across exist skin under an applied direct current.

Result 1

As shown in Fig. 3, before iontophoretic operation, no permeation of TRH through the skin was observed. However, after one hour of operation (1 mA constant current, DC), the amount of permeation was 24±9 μg (mean value ± standard error) for the conventional electrode. For the new electrode, the amount of penetrated TRH was 77±13 μg (mean value ± standard error) an amount transported 3 times as much as that incorporated with conventional electrode.

Result 2

Further, as shown in Fig. 4, the new electrode exhibited an efficiency of drug delivering about six times greater than the conventional electrode (127±17 μg/20±7 μg (new/conventional), mean value ± standard error) even under conditions of operation (2 mA) with a depolarizing pulse iontophoresis system at the current density of 0.87 mA/cm^2 (ADIS; 40 kHz, 30% duty, made by Advance).

Evaluation of Efficacy of Electrode in Vivo (Method of Transdermal Drug Administration to SD Rats)

Method: SD rats were anaethetized by medicinal ether, their stomach portions were shorn by shears, and then were shaved by shavers. The stomach portions were cleaned with absorbent cotton containing 70% alcohol, then applicators having the new electrodes and/or applicators having the conventional electrodes were attached to the stomachs of the rats (n = 3). For the

counter electrode, a PVA physiological saline gel on AgCl electrode was used. The rats were placed in immobilizers (made by Natsume Seisakusho) and allowed to stand without current for one hour, then with current (ADIS, 40 kHz, 30% duty) for one hour. The blood samplings (0.5 ml) were carried out -60, 0, 30, 60, and 120 minutes, from tail vein of tail blood taken each time. The obtained blood samples were centrifuged (10,000 rpm/5 min) and the supernatents were used as the sample serums. The measurement of the TRH levels in the serums was entrusted to Mitsubishi Petrochemical BCL and was by RIA.

Result 3

As shown in Fig. 5, after the start of iontophoresis, the plasma concentration of TRH increased. After 30 minutes, a steady level of 10 ± 2.8 ng/ml was exhibited in the case of conventional electrodes, while the level was a high one of 40 ng/ml or more with the new electrodes, i.e., approximately four times the concentration in the blood was achieved.

The typical embodiments of the present invention can be embodied as follows.

1. An electrode system for iontophoresis' comprising an electroconductive material or substrate and a functional polymer having at least member selected from the group consisting of ammonium hydroxide or a quaternary ammonium halogen salt, sulfonic group, carboxylic group, amine group, or other functional groups.

2. An electrode system for iontophoresis as set forth in item 1, wherein said electroconductive material is safe to the body and for example comprises granules or foil of one or more of silver, iron, gold, platinum, titanium, and other metal materials and carbon, graphic, and other non-metallic materials.

3. An electrode system for iontophoresis as set forth in item 1, wherein said quaternary ammonium halogen salt is, for example, a chloride or a bromide or an

iodide etc.

4. An electrode system for iontophoresis as set forth in item 1, wherein said functional polymer is a mixture of one or more of a carboxymethyl cellulose system, acrylate system, vinyl alcohol system, vinyl pyrrolidone system, styrene system, and etc.

5. A film electrode system wherein the electrode system set forth in item 1 is made by mixing an ink containing an electroconductive material (for example, silver) and a polymer having quaternary ammonium chloride and printing this on a film.

6. An anode electrode system wherein the electrode system of item 1 is, for example, comprised of a combination of the following, that is, an iron electroconductive material and a polymer having ammonium hydroxide.

7. A cathode electrode system wherein the electrode system of item 1 is for example comprised of a combination of the following, that is, a silver chloride/silver (AgCl/Ag) electroconductive material and a polymer having a carboxy-silver salt group (-COOAg).

8. An anode or cathode electrode system wherein the electrode system of item 1 is, for example, comprised of a combination of the following, that is, an inert electroconductive material and a polymer having an amine group and sulfonic group or carboxylic group.

9. An anode or cathode electrode system wherein the electrode system of item 1 is, for example, comprised of a combination of the following, that is, an inert electroconductive material and a polymer having a quinone or aminal functional group and amine group and sulfonic group or carboxylic group.

10. An applicator for administering a pharmaceutical by iontophoresis using the electrode system set forth in item 1.

11. An applicator for administering a pharmaceutical by iontophoresis as set forth in claim 10,

which has a simple and safe patch structure comprised of just the electrode of item 3 and a pharmaceutical reservoir.

12. An applicator for administering a pharmaceutical by iontophoresis as set forth in claim 10, wherein the reservoir of item 11 is a gel including a pharmaceutical and a cellulose or nylon porous substrate.

INDUSTRIAL APPLICABILITY

As explained above, according to the present invention, there is provided an electrode system for administration of a pharmaceutical by iontophoresis, wherein the ions generated from the surface of the electroconductive material of the electrode portion are immediately made insoluble and captured by the functional polymer surrounding the substrate whereby the ions competing with the pharmaceutical ions can be removed from the electric field. Therefore, the delivery rate in the transdermal permeation of the pharmaceutical is improved and, as a result, an effect of the efficiency of pharmaceutical utilization is improved. Furthermore, the completed electrode system itself is simple in the construction, and therefore, it is possible to fabricate an iontophoretic device into a small patch easy to use. And its manufacturing cost can be reduced.



CLAIMS

1. An electrode system for iontophoresis comprising a mixture of (i) an electroconductive material and (ii) a functional polymer having ammonium hydroxide or a quaternary ammonium halogen salt, sulfonic group, carboxylic group, amine group.

2. An electrode system for iontophoresis as claimed in claim 1, wherein said electroconductive material is safe to the body and comprises granules or foil of one or more of silver, iron, gold, platinum, titanium, carbon, graphic.

3. An electrode system for iontophoresis as claimed in claim 1, wherein said quaternary ammonium halogen salt is a chloride, a bromide or iodide.

4. An electrode system for iontophoresis as claimed in claim 1, wherein said functional polymer is a mixture of one or more of a carboxymethyl cellulose system, acrylate system, vinyl alcohol system, vinyl pyrrolidone system, styrene system.

5. A film electrode system wherein the electrode system according to claim 1 is made by mixing an ink containing an electroconductive material and a polymer having quaternary ammonium chloride and printing this on a film.

6. An anode electrode system wherein the electrode system according to claim 1 is comprised of a combination of an iron electroconductive material and a polymer having ammonium hydroxide.

7. A cathode electrode system wherein the electrode system according to claim 1 is comprised of a combination of a silver chloride/silver (AgCl/Ag) electroconductive material and a polymer having a carboxy-silver salt group (-COOAg).

8. An anode or cathode electrode system wherein the electrode system according to claim 1 is comprised of a combination of an inert electroconductive material and a polymer having an amine group and sulfonic group or

carboxylic group.

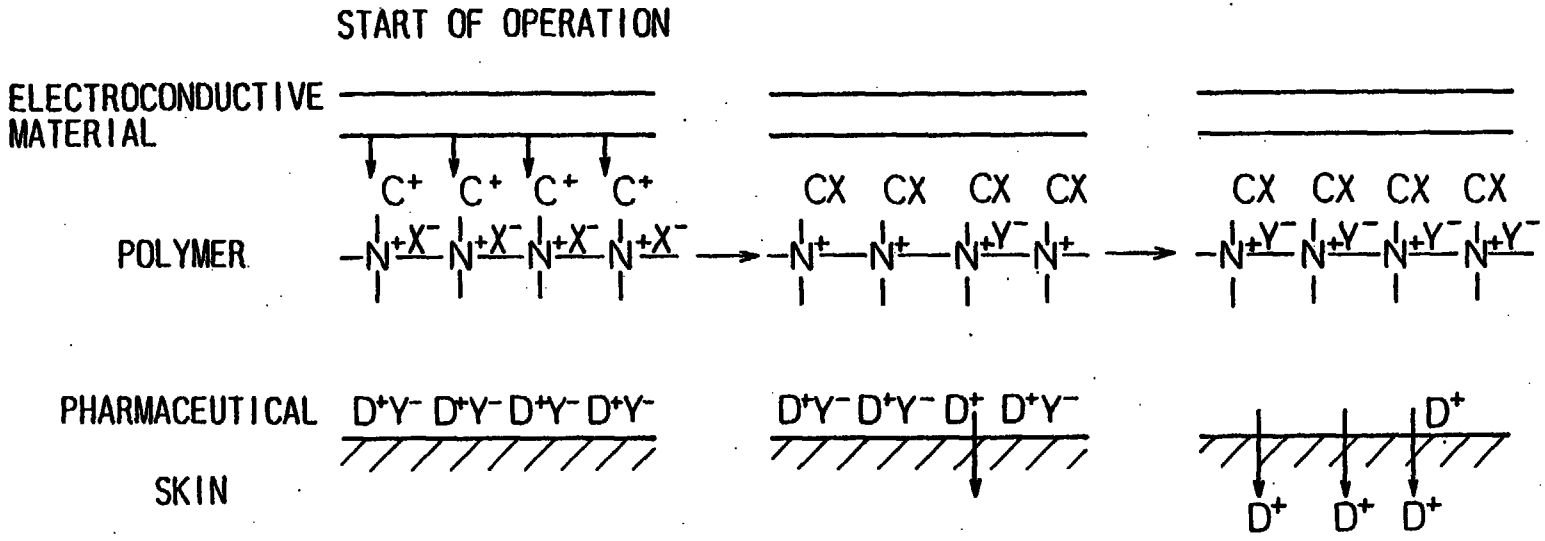
9. An anode or cathode electrode system wherein the electrode system according to claim 1 is comprised of a combination of an inert electroconductive material and a polymer having a quinone or aminal functional group and amine group and sulfonic group or carboxylic group.

10. An applicator for administering a pharmaceutical by iontophoresis using the electrode system according to claim 1.

11. An applicator for administering a pharmaceutical by iontophoresis as claimed in claim 10, which has a simple and safe patch structure comprised of just the electrode according to claim 3 and a pharmaceutical reservoir.

12. An applicator for administering a pharmaceutical iontophoresis as claimed in claim 10, wherein the reservoir of claim 11 is a gel including a pharmaceutical and a cellulose or nylon porous substrate.

Fig.1



C+: ION GENERATED FROM ELECTRODE DUE TO CURRENT
X-: COUNTER ION OF POLYMER
D+: PHARMACEUTICAL ION
Y-: PHARMACEUTICAL COUNTER ION
CX: PRECIPITATE OR NONIONIC SUBSTANCE

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Fig.2(A)

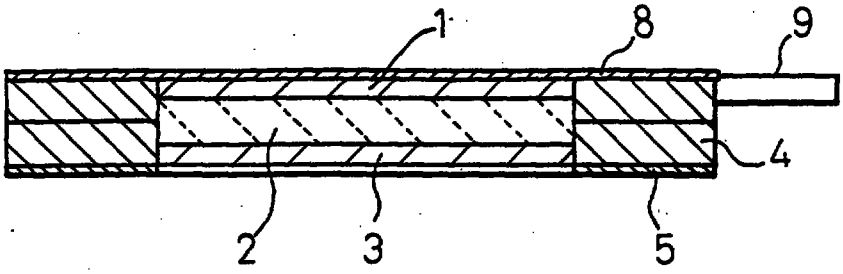


Fig.2(B)

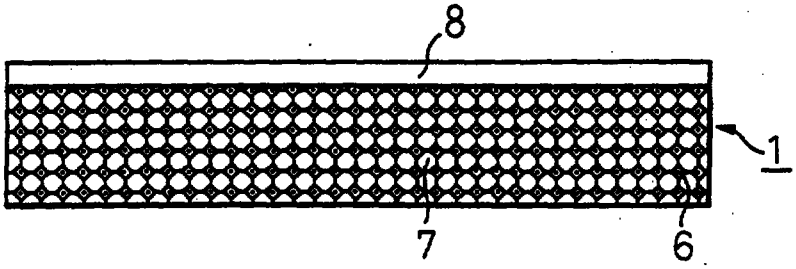


Fig.2(C)

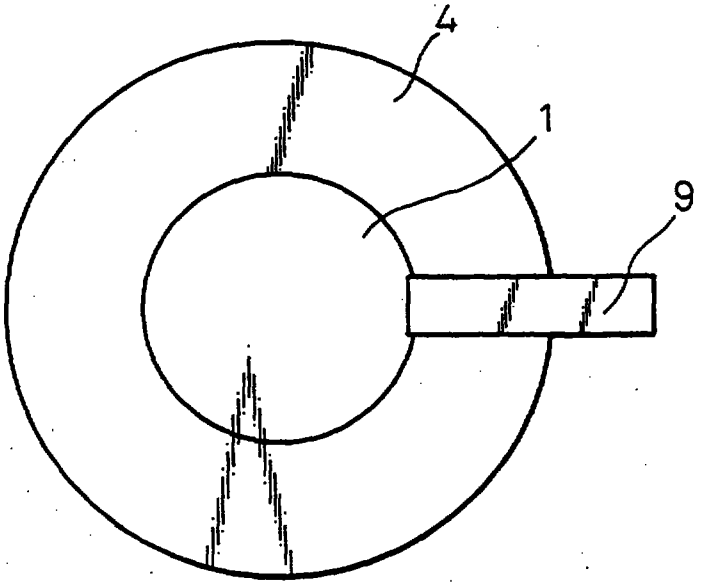


Fig.3

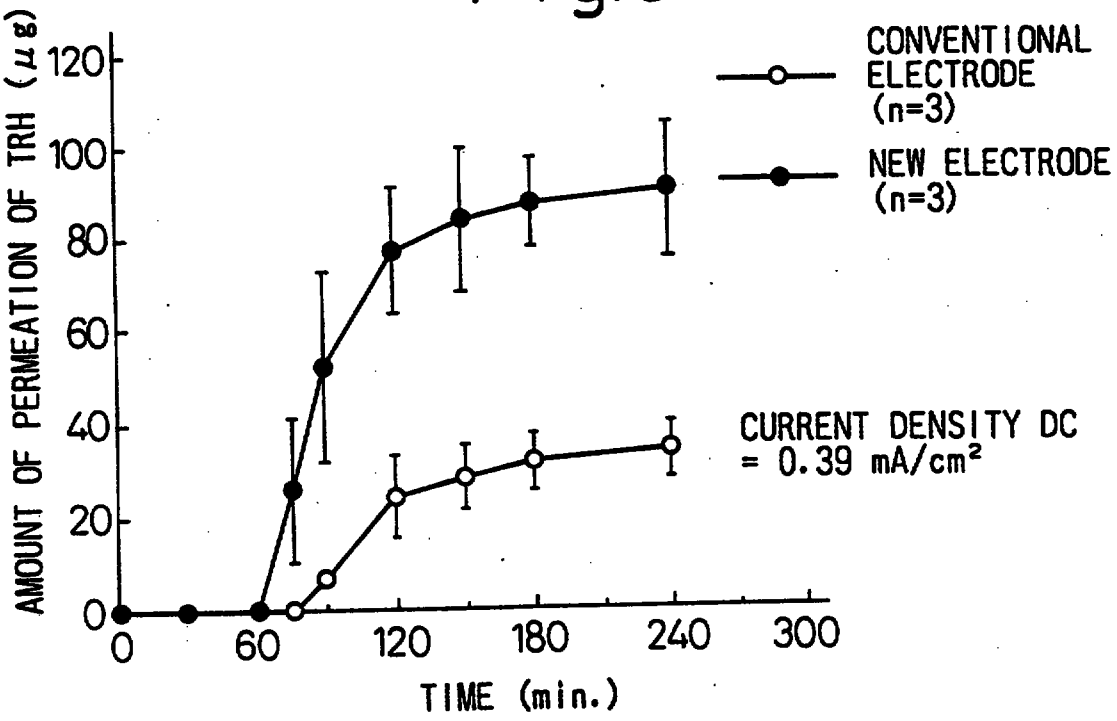

AMOUNT OF PERMEATION OF TRH (μg)
120
100
80
60
40
20
0
0
60
120
180
240
300
TIME (min.)
CONVENTIONAL ELECTRODE (n=3)
NEW ELECTRODE (n=3)
CURRENT DENSITY DC = 0.39 mA/cm²

Fig.4

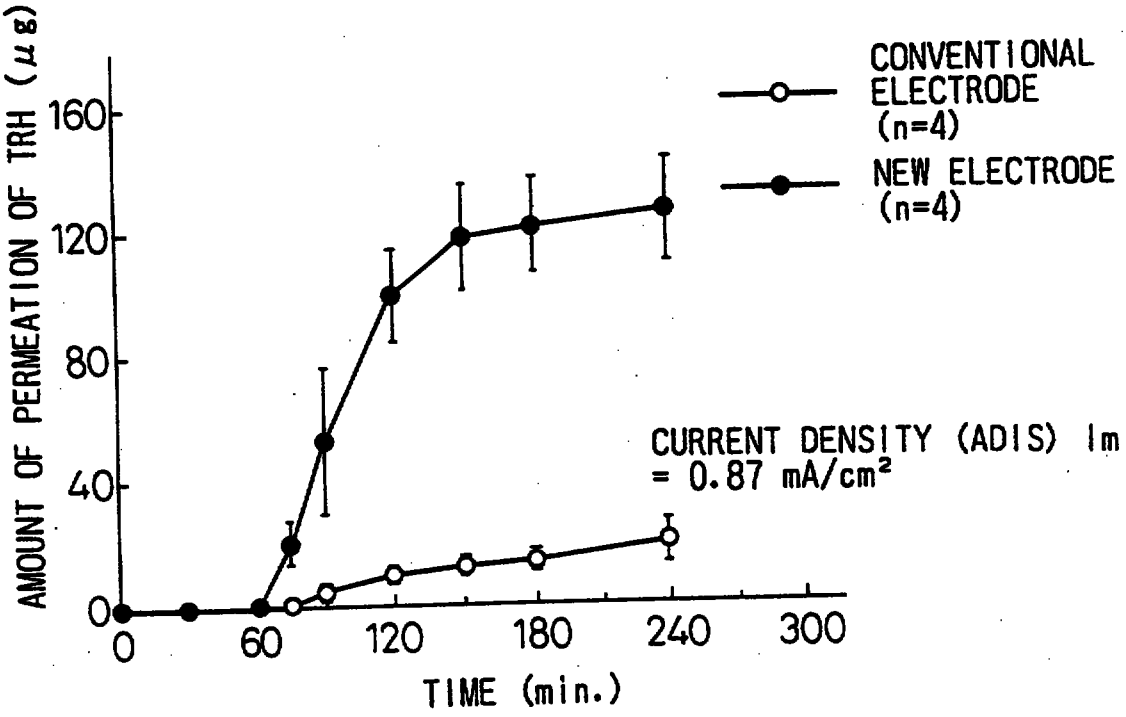
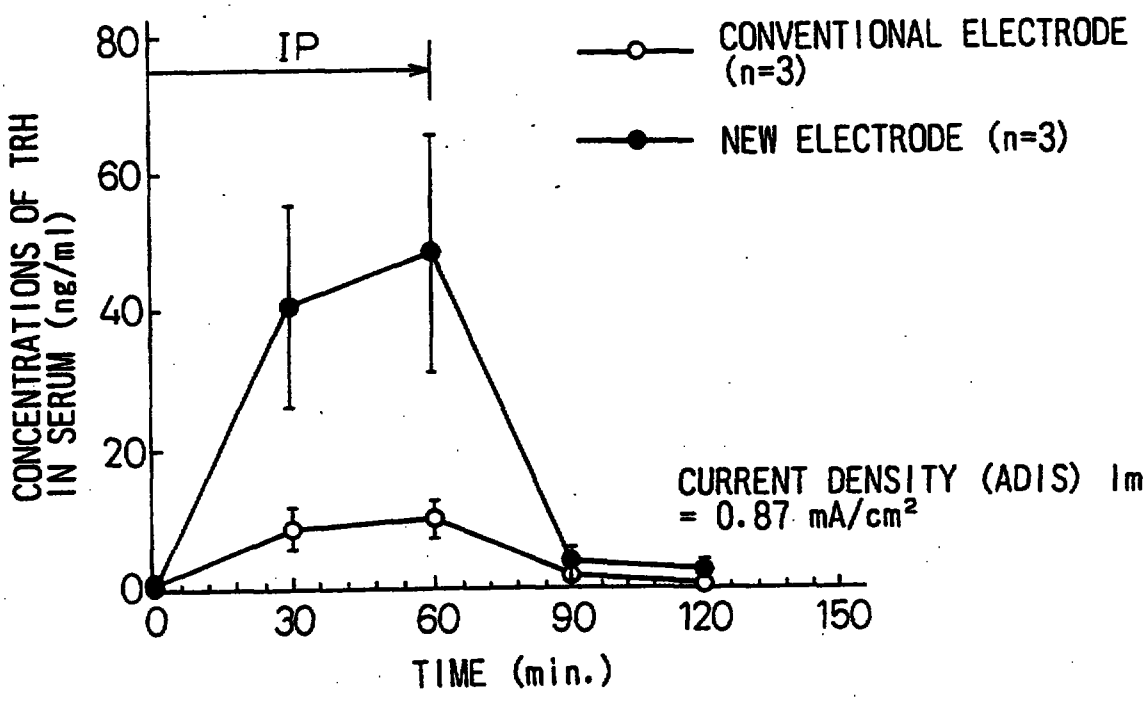

AMOUNT OF PERMEATION OF TRH (μg)
160
120
80
40
0
0
60
120
180
240
300
TIME (min.)
CONVENTIONAL ELECTRODE (n=4)
NEW ELECTRODE (n=4)
CURRENT DENSITY (ADIS) Im = 0.87 mA/cm²

Fig.5


80
60
40
20
0
IP
CONCENTRATIONS OF TRH IN SERUM (ng/ml)
CONVENTIONAL ELECTRODE (n=3)
NEW ELECTRODE (n=3)
CURRENT DENSITY (ADIS) Im = 0.87 mA/cm²
0
30
60
90
120
150
TIME (min.)

LIST OF REFERENCE NUMERAL

1. Electrode portion
2. Gel member
3. Contact member
4. Tape
5. Adhesive member
6. Silver granules
7. Polymer granules
8. Support member
9. Electroconductive terminal

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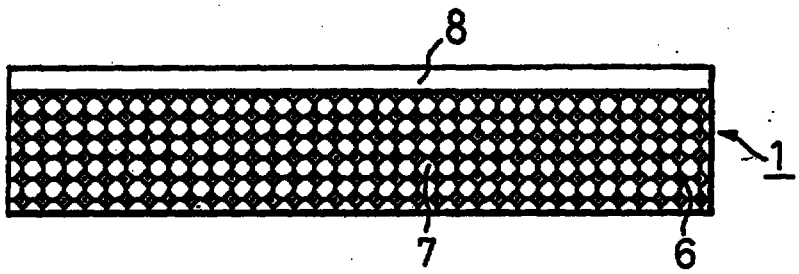
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(54) Title: HIGH EFFICIENCY ELECTRODE SYSTEM FOR IONTOPHORESIS

(57) Abstract

An electrode system for administration of a pharmaceutical by iontophoresis which is comprised of an electroconductive substrate and a functional polymer having ammonium hydroxide or a quaternary ammonium halogen salt, sulfonic group, carboxylic group, amine group.



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BB	Barbados	GN	Guinea	NE	Niger
BE	Belgium	GR	Greece	NL	Netherlands
BF	Burkina Faso	HU	Hungary	NO	Norway
BG	Bulgaria	IE	Ireland	NZ	New Zealand
BJ	Benin	IT	Italy	PL	Poland
BR	Brazil	JP	Japan	PT	Portugal
BY	Belarus	KE	Kenya	RO	Romania
CA	Canada	KG	Kyrgystan	RU	Russian Federation
CF	Central African Republic	KP	Democratic People's Republic of Korea	SD	Sudan
CG	Congo			SE	Sweden
CH	Switzerland	KR	Republic of Korea	SI	Slovenia
CI	Côte d'Ivoire	KZ	Kazakhstan	SK	Slovakia
CM	Cameroon	LI	Liechtenstein	SN	Senegal
CN	China	LK	Sri Lanka	TD	Chad
CS	Czechoslovakia	LU	Luxembourg	TG	Togo
CZ	Czech Republic	LV	Latvia	TJ	Tajikistan
DE	Germany	MC	Monaco	TT	Trinidad and Tobago
DK	Denmark	MD	Republic of Moldova	UA	Ukraine
ES	Spain	MG	Madagascar	US	United States of America
FI	Finland	ML	Mali	UZ	Uzbekistan
FR	France	MN	Mongolia	VN	Viet Nam
GA	Gabon				