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Fragrance Precursor Compounds

The invention relates to fragrance precursor compounds. In particular, the invention relates to the use of several classes of compounds which can act as fragrance precursors in cosmetic products such as deodorants and antiperspirants and in laundry products such as detergents and fabric softeners. These compounds are normally odourless or nearly so, but upon contacting the skin as for example, in skin care compositions or in personal care compositions, produce fragrances. The compounds also produce fragrances when used in the presence of lipases, e.g. in laundry products.

A principal strategy currently employed in imparting odours to consumer products is the admixing of the fragrance directly into the product. There are however, several drawbacks to this strategy. The fragrance material can be too volatile, resulting in fragrance loss during manufacturing, storage, and use. Many fragrance materials are also unstable over time. This again results in loss during storage.

In some cases, fragrances are microencapsulated or treated with cyclodextrins to form inclusion complexes to help decrease volatility and improve stability. However, these methods are for a number of reasons often not successful. In addition, cyclodextrins can be too expensive.

In many consumer products it is desirable for the fragrance to be released slowly over time. Microencapsulation and cyclodextrins have been

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used to provide slow-release properties, however, they are subject to the same limitations as above.

The Japanese patent application 07/179 328 discloses sustained release
5 of aromatic compounds which are perfume derivatives for control of
perspiration. The compositions perfume when they are gradually
decomposed by bacteria usually residing on the body surface, resulting in a
long stable perfuming effect. Preferred derivative is one or a mixture of
phosphoric ester, amino acid and carboxylic derivatives of perfumes. The
10 active fragrancing substance, i.e. an alcohol, an aldehyde or a ketone, is
released slowly after laundering to prolong the fragrancing of the fabric after
laundering.

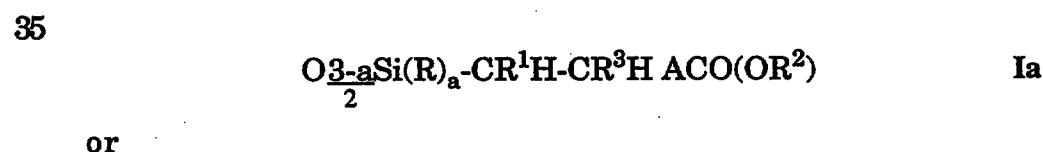
The PCT application WO 95/04809 discloses the use of esters for
15 scenting fabrics being washed in the presence of a lipase-containing
detergent.

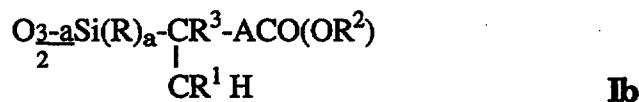
The PCT application WO 96/1999119 describes the addition of silicone
polymers to perfumes to improve surface residuality of a fragrance. The
20 fragrance is not bound to the polymer.

US 3,215,719 describes silicate esters of essential alcohols where the
oxy group RO- of an alcohol is directly bonded to silicone as Si-OR which can
be applied to textiles to impart thereto a lasting fragrance.

25 The present invention now provides compounds which show a low
level of odour, or are even odourless, prior to application to the skin, but
which release odorant molecules after application to the skin (that is, they
provide a delayed release of the fragrance), in particular to the skin in the
30 axilla. The compounds of the present invention also release odorant
molecules when used in the presence of lipase-containing products.

The compounds under consideration are organosiloxane compounds
having at least one unit of the formula





or a mixture thereof and if any other unit(s) in the said siloxanes being present this/these is/are of the formula



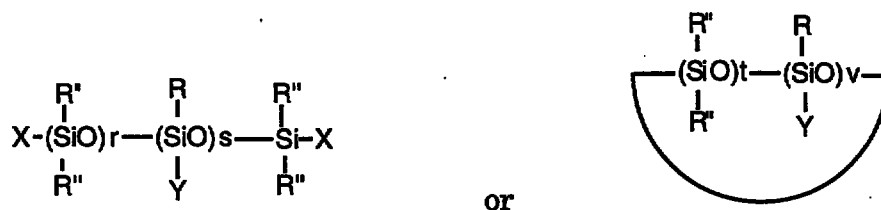
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wherein R represents a substituted or unsubstituted C₁₋₈ alkyl group or a substituted or unsubstituted aryl group, R'' represents a hydrogen atom, a monovalent C₁₋₈ hydrocarbon group or a monovalent C₁₋₈ halogenated hydrocarbon group; R¹ represents a hydrogen atom or a substituted or unsubstituted C₁₋₈ alkyl group or a substituted or unsubstituted aryl group or a bond connecting CR¹ and CR³; R³ represents a hydrogen atom or a substituted or unsubstituted C₁₋₈ alkyl group or a substituted or unsubstituted aryl group; A represents (CR⁴)_n whereby R⁴ represents a substituted or unsubstituted C₁₋₈ alkyl group or a substituted or unsubstituted aryl group or a hydrogen atom, n has a value of 0-20, preferably 1 to 10, and each R⁴ is the same or different; OR² represents the residue of an olfactive alcohol or of the enol form of an olfactive aldehyde or olfactive ketone; a has a value of 0, 1 or 2; b has a value of 0, 1, 2 or 3.

20

In the general formula of the organosiloxane compounds of the invention, R may be for example methyl, ethyl, butyl or phenyl. R'' may be for example alkyl, alkenyl, aryl, alkaryl, aralkyl and halogen substituted alkyl, ^εalkenyl, aryl, alkaryl and aralkyl. Specific examples are methyl, ethyl, vinyl, phenyl and 3,3,3-trifluoropropyl. R⁴ preferably is a hydrogen atom or methyl. It is preferred that at least 80% of all R and R'' groups are methyl groups. a is preferably 1, while b is preferably 2, making the organosiloxane compound a substantially linear or cyclic diorganosiloxane polymer. However, if the diorganosiloxane is a substantially linear polymer, at least two endblocking units must be present, thus requiring the presence of two units in which a has a value of 2, two units in which the value of b is 3 or one unit wherein a is 2 and one unit in which b is 3. Suitable preferred polymers have therefore either the general formula

30



wherein R and R'' are as defined above, X denotes a group Y or a group R'', and Y denotes a group of the formula Ia or Ib.

5 r has a value of from 0 to 1000, s has a value of from 0 to 1000, whereby at least one X denotes Y in the case that s = 0. t has the value of from 0 to 10, v has the value of from 1 to 10. Preferably v+t has a value of 3 to 6 and for v+t greater than 3, v-t having a value of at least 3 represent further interesting
10 embodiments.

The organosiloxane compounds of the present invention have at least one unit falling within the general formula Ia or Ib. Suitable organosiloxane compounds include polymeric materials, which may be homopolymers
15 consisting only of such units Ia or Ib or they may be copolymers containing in addition units having the general formula II.

The compounds of formula Ia or Ib are not limited to any particular stereoisomers, all possible stereoisomers as well as racemates are thus
20 included within the scope of formula Ia or Ib.

ε. 886 The units of formula Ia or Ib may be distributed randomly in an organosiloxane copolymer, they may be end-blocking units of these polymer or they may be located at both the end of the copolymer and in the chain of the
25 copolymer.

The organosiloxane compounds may vary from freely flowing liquids to highly viscous gum-like materials to resinous solids. Preferred, at least for cosmetic applications, are monomers or liquid, substantially linear,
30 organosiloxane homopolymers and copolymers, most preferably those having a viscosity of from 25 to 500 mm²/s as these are more easily mixed with other ingredients to make cosmetic compositions and as they will spread more easily onto the skin.

Suitable examples of alcohols R^2OH are primary or secondary alcohols or phenols such as listed in Table 1.

Table 1

5	amyl alcohol
	hexyl alcohol*
	2-hexyl alcohol*
	heptyl alcohol*
10	octyl alcohol*
	nonyl alcohol*
	decyl alcohol*
	undecyl alcohol*
	lauryl alcohol*
15	myristic alcohol
	3-methyl-but-2-en-1-ol*
	3-methyl-1-pentanol
	cis-3-hexenol**
	cis-4-hexenol*
20	3,5,5-trimethyl hexanol
	3,4,5,6,6-pentamethylheptan-2-ol
	citronellol**
	geraniol**
	oct-1-en-3-ol
25	2,5,7-trimethyl octan-3-ol
25	cis-3,7-dimethyl-2,6-octadien-1-ol
	6-ethyl-3-methyl-5-octen-1-ol*
	3,7-dimethyl-oct-3,6-dienol*
	3,7-dimethyloctanol*
30	7-methoxy-3,7-dimethyl-octan-2-ol*
	cis-6-nonenol*
	5-ethyl-2-nonanol
	6,8-dimethyl-2-nonanol*
	2,2,8-trimethyl-7 (8)-nonene-3-ol
35	nona-2,6-dien-1-ol
	4-methyl-3-decen-5-ol**
	dec-9-en-1-ol
	benzylalcohol

- 2-methyl undecanol
- 10-undecen-1-ol
- 1-phenyl ethanol*
- 2-phenyl ethanol**
- 5 2-methyl-3-phenyl-3-propenol
- 2-phenyl propanol*
- 3-phenyl propanol*
- 4-phenyl-2-butanol
- 2-methyl-5-phenyl pentanol*
- 10 2-methyl-4-phenyl-pentanol*
- 3-methyl-5-phenyl-pentanol*
- 2-(2-methylphenyl)-ethanol*
- 4-(1-methylethyl)benzene methanol
- 4-(4-hydroxyphenyl)butan-2-one*
- 15 2-phenoxy ethanol*
- 4-(1-methylethyl)-2-hydroxy-1-methyl benzene
- 2-methoxy-4-methyl phenol
- 4-methyl phenol
- anisic alcohol*
- 20 p-tolyl alcohol*
- cinnamic alcohol**
- vanillin*
- ethyl vanillin*
- eugenol**
- 25 isoeugenol**
- ε-thymol
- apethol*
- decahydro 2-naphthalenol
- borneol*
- 30 cedrenol*
- farnesol*
- fenchyl alcohol*
- menthol*
- 3,7,11-trimethyl-2,6,10-dodecatrien-1-ol
- 35 alpha ionol*
- tetrahydro ionol*
- 2-(1,1-dimethylethyl)cyclohexanol*
- 3-(1,1-dimethylethyl)cyclohexanol*



- 4-(1,1-dimethylethyl)cyclohexanol*
 4-isopropyl cyclohexanol
 6,6-dimethyl-bicyclo [3.3.1]hept-2-ene-2-ethanol
 6,6-dimethyl-bicyclo [3.1.1]hept-2-ene-methanol*
 5 p-menth-8-en-3-ol*
 3,3,5-trimethyl cyclohexanol
 2,4,6-trimethyl-3-cyclohexenyl-methanol*
 4-(1-methylethyl)cyclohexyl-methanol*
 4-(1,1-dimethylethyl)cyclohexanol
 10 2-(1,1-dimethylethyl)-cyclohexanol
 2,2,6-trimethyl-alpha-propyl cyclohexane propanol*
 5-(2,2,3-trimethyl-3-cyclopentenyl)-3-methylpentan-2-ol*
 3-methyl-5-(2,2,3-trimethylcyclopentyl-3-enyl)pent-4-en-2-ol**
 2-ethyl-4(2,2,3-trimethylcyclopentyl-3-enyl)but-2-en-1-ol**
 15 4-(5,5,6-trimethylbicyclo[2.2.1] hept-2-yl)-cyclohexanol*
 2-(2-methylpropyl)-4-hydroxy-4-methyl-tetrahydropyran
 2-cyclohexyl propanol*
 2-(1,1-dimethylethyl)-4-methyl cyclohexanol*
 1-(2-tert-butyl-cyclohexyloxy)-2-butanol*
 20 1-(4-isopropyl-cyclohexyl)-ethanol*.

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whereby one asterisk indicates the preferred alcohols and two asteriks indicate the more preferred alcohols.

- 25 Suitable examples of aldehydes are listed in Table 2.

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Table 2

- 2,6,10-trimethylundec-9-enal*
 30 undecanal
 1,2,3,4,5,6,7,8,-octahydro-8,8-dimethyl-2-napthalenecarboxaldehyde
 tridecanal
 2-[4-(1-methylethyl)phenyl]-ethanal
 1-carboxaldehyde,2,4-dimethyl-cyclohex-3-ene*
 35 4-carboxaldehyde-1,3,5-trimethyl-cyclohex-1-ene*
 1-carboxaldehyde-2,4-dimethyl-cyclohex-3-ene*
 1-carboxaldehyde-4-(4-hydroxy-4-methylpentyl)-cyclohex-3-ene*
 hex-2-enal*

- 3,5,5-trimethyl-hexanal
heptanal*
2,6-dimethyl-hept-5-eneal*
decanal**
5 dec-9-enal
dec-4-en-1-al
2-methyldecanal*
undec-10-ene-1-al**
undecanal*
10 dodecanal**
2-methylundecanal**
tridecanal
2-tridecenal
octanal**
15 nonanal*
3,5,5-trimethylhexanal
2-nonenal
undec-9-eneal**
2-phenyl-propanal*
20 4-methyl-phenyl acetaldehyde*
3,7-dimethyl-octanal*
7-hydroxy-3,7-dimethyl-octanal*
2,6-dimethyl-oct-5-ene-1-al
2-(4, -(1-methylethyl)phenyl)-ethanal*
25 3-(3-isopropylphenyl)-butanal**
2-(3,7-dimethyloct-6-en-oxy)-ethanal
1-carboxaldehyde-4-(4-methyl-3-penten-1-yl)-cyclohex-3-ene*
2,3,5,5,-tetramethyl-hexanal
longifolic aldehyde
30 2-methyl-4-(2,6,6-trimethylcyclohex-2-en-1-yl)-butanal*
2-methyl-3-(4-tert-butylphenyl)propanal**
4-(1,1-dimethylethyl)-benzenepropanal*
2-[4-(1-methylethyl)phenyl]-propanal
alpha-methyl-1,3-benzodioxole-5-propanal*
35 3,7-dimethyl-oct-6-en-1-al*
2-methyl-3-(p-isopropylphenyl)-propionaldehyde*
4-(4-hydroxy-4-methylpentyl)-cyclohex-3-en-1-carboxaldehyde**
alpha-methyl-1,3-benzodioxole-5-propanal*

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1-carboxaldehyde-4-(1,1-dimethylethyl)-cyclohexane
4-(octahydro-4,7-methano-5H-inden-5-ylidene)-butanal

whereby one asterisk indicates the preferred aldehydes and two asteriks
5 indicate the more preferred aldehydes.

Suitable examples of ketones are listed in Table 3:

Table 3

10

2-heptyl-cyclopentanone
2,2,6,10-tetramethyltricyclo[5.4.0.0(6,10)]-undecan-4-one
benzylacetone*
carvone*

15 1,2,3,5,6,7-hexahydro1,1,2,3,3,-pentamethyl-4H-inden-4-one*

methyl heptenone*
dimethyl octenone*
2-(butan-2-yl)-cyclohexanone*
2-hexyl-cyclopent-2-en-1-one*

20 2-(1-methylethyl)-5-methyl-cyclohexanone*

2-(2-methylethyl)-5-methyl-cyclohexanone*
3-methyl-cyclopentadecanone
4-tert-pentyl-cyclohexanone*

2-oxo-1-pentyl-cyclopentaneacetic acid methyl ester**

25 1-(1,2,3,4,5,6,7,8-octahydro-2,3,8,8-tetramethyl-2-naphthalenyl)-ethanone*

5 3-methyl-5-propyl-cyclohex-2-en-1-one*

whereby one asterisk indicates the preferred ketones and the two asteriks
30 indicate the more preferred ketone.

It is a matter of course that it is not possible to give complete lists of the
odoriferous alcohols or aldehydes or ketones which are liberated as a result
of the desired cleavage of the compounds of formula Ia or Ib, e.g. by bacteria,
35 in particular axilla bacteria, or lipases and which alcohols, aldehydes and
ketones are then capable of imparting agreeable odours. The skilled artisan
is, however, quite aware of those alcohols, aldehydes and ketones which
provide a positive contribution to the fragrance compositions.

The compounds of formula Ia and/or Ib are preferably used as sustained release odorants but also to mask or attenuate undesirable odours or to provide additional odours not initially present in consumer products, i.e. cosmetic products destined for application to human skin such as underarm deodorants or antiperspirants or other deodorants contacting the body, or in hand lotions, baby powders, baby lotions, ointments, foot products, facial cleansers, body wipes, facial make-up, colognes, after-shave lotions, shaving creams, etc. Additional applications include laundry detergents, fabric softeners, fabric softener sheets, automatic dishwasher detergents, and other lipase-containing consumer products.

The compounds of formula Ia and/or Ib are virtually odourless under normal temperature, i.e. at about 10 - 50 degrees Celsius, and at atmospheric conditions and about 20 to 100 % relative humidity. However, when applied to the body or when used in an application in the presence of lipases, they undergo a transformation in which the fragrant alcohol or aldehyde or ketone is released.

As said before the compounds of formula Ia and/or Ib, upon cleavage, provide alcohols, aldehydes, and ketones having organoleptic properties and therefore permit the development of methods useful in presenting or enhancing the odour of consumer products. These compounds may be used individually in an amount effective to enhance the characteristic odour of a material. More commonly, however, the compounds are mixed with other fragrance components in an amount sufficient to provide the desired odour characteristics.

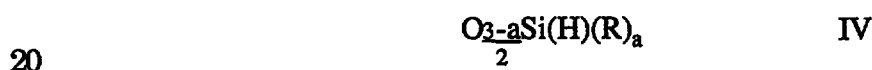
The amount required to produce the desired, overall effect varies depending upon the particular compound(s) of formula Ia and/or Ib chosen, the product in which it will be used, and the particular effect desired.

For example, depending upon the selection and concentration of the compound chosen, when a compound of formula Ia or Ib is added either singly or as a mixture, e.g. to a deodorant or laundry product composition at levels ranging from about 0.1 to about 10 % by weight, or most preferred about 0.25 to about 4 % by weight, an odorant, i.e. an odoriferous alcohol or aldehyde or ketone, in an organoleptically effective amount is released when

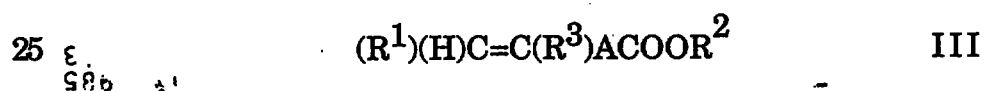
the product is used. This newly formed odorant produces or serves to enhance the odour of the fragrance.

The compounds of formula Ia and/or Ib can accordingly be used in the
5 manufacture of odorant compositions used in the preparation of cosmetic
and laundry products, e.g. deodorants, antiperspirants, laundry detergents,
fabric softeners, and as is evident from the above compilation, a broad range
of known odorants or odorant mixtures can be used. In the manufacture of
such compositions the known odorants or odorant mixtures set forth above
10 can be used according to methods known to a person skilled in the art,
normally a perfumer, or described, e.g. in W.A. Poucher, *Perfumes,
Cosmetics, Soaps*, 2, 7th Edition, Chapman and Hall, London 1974.

The compounds of formula Ia or Ib can be prepared by using standard
15 methods known to a person skilled in the art. For example the method
disclosed by US 5,550,272 can be used. The organosiloxane compounds of the
present invention can be prepared by the reaction of an organosiloxane
having at least one unit of the general formula



with an unsaturated ester or enol ester having a radical OR^2 of a fragrant
alcohol, aldehyde or ketone, a compound of the general formula



Esters of the general formula III may be prepared from the
corresponding carboxylic acids and alcohols by standard methods known per
se, see e.g. *Comprehensive Organic Chemistry*, Derek Barton and W. David
30 Ollis, eds. Vol. 2, 1979, pp.871-907. For example, esters are formed by the acid
catalyzed reaction between a carboxylic acid and an alcohol. During the
condensation, water is usually removed.

Either protic or Lewis acids may be used. Some acids which may be
35 used are p-toluensulfonic acid, sulfuric acid, and pyridinium p-
toluenesulfonate. A variety of inert solvents may be used such as toluene,
xylene, cyclohexane, and hexane.

In another example, a carboxylic acid and an alcohol react to form an ester when treated with N,N'-dicyclohexycarbodiimide and 4-pyrrolidinopyridine, see, for example, the procedure of Hassner and Alexanian in Tetrahedron Letters 4475, (1978).

5

Enol esters of the general formula III may be prepared using the procedure of J. Chem. Soc., Perkin Trans. I, 2509 (1993).

10 The preferred preparation of organosiloxane compounds of the type Ia or Ib comprises reacting together a compound of the general formula III and an organosiloxane compound having at least one unit of the general formula IV, any other units present being those represented by the general formula II.

15 The reaction is preferably carried out employing stoichiometric proportions of compounds III and IV or a slight stoichiometric excess of compound III. However, a stoichiometric deficiency of compound III can be employed if residual silicon-bonded hydrogen is desired in the product.

20 The reaction between compounds III and IV may be carried out employing known procedures for the addition of silicon-bonded hydrogen atoms to groups containing aliphatic unsaturation. Thus, such reactions are generally catalysed by a platinum group metal or a compound or complex of such a metal. Examples of catalysts which may be employed in the reaction
25 between compounds III and IV are platinum on carbon, chloroplatinic acid, platinum acetyl acetate, complexes of platinum compounds with unsaturated compounds, e.g. olefins and vinyl siloxanes, complexes of rhodium and palladium compounds and complexes of platinum compounds supported on inorganic substrates. The addition reaction may be performed
30 at reduced, atmospheric or increased pressure. It is generally preferred to employ a solvent, e.g. toluene or xylene in the reaction mixture although the presence of a solvent is not essential. It is also preferred to carry out the reaction at elevated reaction temperatures e.g. from room temperature up to the reflux temperature of the reaction mixture.

35

The invention is further described in detail by the Examples.

Example 1

Pent-4-enoic acid phenethyl ester

5 A solution of 29.3 g 4-pentenoic acid, 37.0 g phenylethyl alcohol, 1.0 g p-toluenesulfonic acid and 300 ml cyclohexane was refluxed in a flask equipped with a Dean-Stark trap for 4 hours. The reaction mixture was cooled, diluted with ether, washed with aqueous NaHCO₃ and brine. The organic phase was dried, filtered and evaporated to dryness. The resulting
10 liquid was distilled to yield 42.6 g of a colourless liquid.

NMR (CDCl₃) δ 7.39-7.20 (m, 5 H), 5.91-5.68 (m, 1H), 5.12-4.93 (m, 2H),
 4.30 (t, 2H), 2.94 (t, 2H), 2.48-2.29 (m, 4H).

15 According to the same procedure, pent-4-enoic acid 3,7-dimethyl-octyl ester was prepared from tetrahydrogeraniol and 4-pentenoic acid.

 According to the same procedure, undec-10-enoic acid phenethyl ester was prepared from phenylethyl alcohol and 10-undecenoic acid.
20

Example 2

Pent-4-enoic acid 3,7-dimethyl-octa-2,6-dienyl ester

25 A solution of 24.5 g 4-pentenoic acid, 40.1 g geraniol, 52.1 g N,N'-
ε dicyclohexylcarbodiimide, 2.8 g 4-pyrrolidinopyridine, and 400 ml
dichloromethane was stirred at room temperature for 24 hours. The reaction
was filtered, the solid washed with ether and the combined organic layers
washed with water, aqueous HCl, aqueous NaHCO₃, and brine. The organic
30 phase was dried and evaporated to dryness. The residue was distilled to yield
33.6 g of a colourless liquid.

NMR (CDCl₃) δ 5.96-5.72 (m, 1H), 5.40-5.40 (m, 1H), 5.16-4.93 (m, 3H),
 4.67-4.54 (m, 2H), 2.47-2.38 (m, 4H), 2.20-1.98 (m, 4H),
35 1.72 (s, 3H), 1.70 (s, 3H), 1.60 (s, 3H).

 According to the same procedure, pent-4-enoic acid 3,7-dimethyl-oct-6-enyl ester was prepared from citronellol and 4-pentenoic acid.

According to the same procedure, undec-10-enoic acid 3,7-dimethyl-oct-2,6-dienyl ester was prepared from geraniol and 10-undecenoic acid.

5

Example 3

5-(1,1,3,3,3-Pentamethyl-disiloxanyl)-pentanoic acid phenethyl ester

10 A solution of 5.1 g pent-4-enoic acid phenethyl ester, 1.0 g 5% Pt/C, 3.8 g pentamethyldisiloxane, and 50 ml toluene was stirred at room temperature under nitrogen for 24 hours. The solution was filtered through Celite and evaporated to dryness to yield 8.6 g of a colourless oil.

15 NMR (CDCl₃) δ 7.32-7.13 (m, 5H), 4.21 (t, 2H), 2.89 (t, 2H), 2.31-2.19 (m, 2H), 1.67-1.50 (m, 2H), 1.37-1.19 (m, 2H), 0.5-0.4 (m, 2H), 0.02 (s, 15H).

20 According to the same procedure, 5-(1,1,3,3,3-pentamethyl-disiloxanyl)-pentanoic acid 3,7-dimethyl-octyl ester was prepared from pent-4-enoic acid 3,7-dimethyl-octyl ester and pentamethyldisiloxane.

25 According to the same procedure, 11-(1,1,3,3,3-pentamethyl-disiloxanyl)-undecanoic acid phenethyl ester was prepared from undec-10-enoic acid phenethyl ester and pentamethyldisiloxane.

30 According to the same procedure, a polymer containing units of the general structures Ia and Ib and II was prepared from pent-4-enoic acid phenethyl ester and a hydrosiloxane of the general formula IV, whereby the structure type Ia is dominating.

35 According to the same procedure, a polymer containing units of the general structures Ia and Ib and II was prepared from pent-4-enoic acid 3,7-dimethyl-octyl ester and a hydrosiloxane of the general formula IV, whereby the structure type Ia is dominating.

According to the same procedure, a polymer containing units of the general structures Ia and Ib and II was prepared from undec-10-enoic

acid phenethyl ester and a hydrosiloxane of the general formula IV, whereby the structure type Ia is dominating.

According to the same procedure, 5-(2,4,6,8-tetramethyl-4,6,8-tris-(4-phenethyloxycarbonyl-butyl)-[1,3,5,7,2,4,6,8] tetroxatetrasiloxan-2-yl)-pentanoic acid phenethyl ester was prepared from pent-4-enoic acid phenethyl ester and 2,4,6,8-tetramethylcyclotetrasiloxane.

Example 4

10

5-(1,1,3,3,3-Pentamethyl-disiloxanyl)-pentanoic acid 3,7-dimethyl-oct-6-enyl ester

A solution of 6.0 g pent-4-enoic acid 3,7-dimethyl-oct-6-enyl ester, 1.0 g Pt/C, 3.8 g pentamethyldisiloxane, 0.05 g 2,2,6,6-tetramethyl-1-piperidinyloxy, free radical, and 50 ml toluene was stirred at room temperature for 24 hours. The solution was filtered through Celite and evaporated to dryness to yield 9.8 g of an oil.

20 NMR (CDCl₃) δ 5.09-4.96 (m, 1H), 4.09-3.97 (m, 2H), 2.25-2.18 (m, 2H), 2.17-1.81 (m, 2H), 1.64 (s, 3H), 1.55 (s, 3H), 1.60-1.05 (m, 9H), 0.84 (d, 3H), 0.53-0.42 (m, 2H), 0.02 (s, 15H).

According to the same procedure, 5-(1,1,3,3,3-pentamethyl-disiloxanyl)-pentanoic acid 3,7-dimethyl-octa-2,6-dienyl ester was prepared from pent-4-enoic acid 3,7-dimethyl-octa-2,6-dienyl ester and pentamethyldisiloxane.

According to the same procedure, 11-(1,1,3,3,3-pentamethyl-disiloxanyl)-undecanoic acid 3,7-dimethyl-oct-6-enyl ester was prepared from undec-10-enoic acid 3,7-dimethyl-oct-6-enyl ester and pentamethyldisiloxane.

According to the same procedure, except that the reaction was heated at 50 °C, a polymer containing the general structures Ia and Ib and II was prepared from pent-4-enoic acid 3,7-dimethyl-oct-6-enyl ester and a hydrosiloxane of the general formula IV, whereby the structure type Ia is dominating.

According to the same procedure, 11-(1,1,3,3,3-pentamethyl-disiloxanyl)-undecanoic acid 3,7-dimethyl-octa-2,6-dienyl ester was prepared from undec-10-enoic acid 3,7-dimethyl-octa-2,6-dienyl ester and pentamethyldisiloxane.

5

According to the same procedure, 5-[4,6,8-tris-[4-(3,7-dimethyl-oct-6-enyloxycarbonyl)-butyl]-2,4,6,8-tetramethyl-[1,3,5,7,2,4,6,8] tetroxatetra-siloxan-2-yl]-pentanoic acid 3,7-dimethyl-oct-6-enyl ester was prepared from pent-4-enoic acid 3,7-dimethyl-oct-6-enyl ester and 2,4,6,8-tetramethylcyclotetrasiloxane.

10

According to the same procedure, 5-[3-[4-(3,7-dimethyl-oct-6-enyloxycarbonyl)-butyl]-1,1,3,3-tetramethyl-disiloxanyl]-pentanoic acid 3,7-dimethyl-oct-6-enyl ester was prepared from pent-4-enoic acid 3,7-dimethyl-oct-6-enyl ester and tetramethyldisiloxane.

15

Example 5

E,Z-Acetic acid 3-(4-tert-butyl-phenyl)-2-methyl-propenyl ester

20

A solution of 200 g 2-methyl-3-(4-tert-butylphenyl)propanal, 280 ml triethylamine and 13.4 g sodium acetate in 800 ml of acetic anhydride was stirred at 120°C for 5.5 hours. Then the solution was cooled, water was added and the water phase was extracted with hexane. The organic phase was washed with 2N NaOH and water to neutrality, dried and evaporated to dryness. The residue was distilled to yield 185 g of a colourless liquid.

25

NMR (CDCl₃) δ 7.35-6.97 (m, 5 H), 3.43+3.21 (s, 2 H, E/Z), 2.13 (s, 3 H), 1.60 (s, 3 H), 1.30 (s, 9H).

30

According to the same procedure, acetic acid 3-(3-isopropyl-phenyl)-but-1-enyl ester was prepared from 3-(3-isopropylphenyl)-butanal and acetic anhydride.

Example 6

E,Z-Undec-10-enoic acid 3-(4-tert-butyl-phenyl)-2-methyl-propenyl ester

5 To a solution of 40.0 g acetic acid 3-(4-tert-butyl-phenyl)-2-methyl-propenyl ester in 200 ml THF a solution of 24.3 g potassium t.butoxide in 200 ml THF was dropped in at -70°C. After 90 min. at -70°C 36.5 g 10-undecenoyl-chloride were dropped in at the same temperature. After 4 hours 250 ml saturated NaHCO₃ were dropped in and the solution was diluted with ether,
10 washed neutral with water, dried and evaporated to dryness. The residue was thin-layer-distilled to yield 35.8 g colourless liquid.

NMR (CDCl₃) δ 7.39-6.96 (m, 5 H), 5.93-5.68 (m, 1 H), 5.06-4.87 (m, 2
H), 3.46+3.18 (s, 2 H, E/Z), 2.39 (t, 2 H), 2.12-1.93 (m, 2
15 H), 1.78-1.52 (m, 5 H), 1.48-1.12 (m, 19 H).

According to the same procedure, except that the reaction was stirred for 40 hours, E/Z-11-(1,1,3,3,3-Pentamethyl-disiloxanyl)-undecanoic acid 3-(4-tert-butyl-phenyl)-2-methyl-propenyl ester was prepared from undec-10-enoic
20 acid 3-(4-tert-butyl-phenyl)-2-methyl-propenyl ester and pentamethyldisiloxane.

According to the same procedure, undec-10-enoic acid 3-(3-isopropyl-phenyl)-but-1-enyl ester was prepared from acetic acid 3-(3-isopropyl-phenyl)-
25 but-1-enyl ester and 10-undecenoyl chloride.

According to the same procedure, 11-(1,1,3,3,3-pentamethyl-disiloxanyl)-undecanoic acid 3-(3-isopropyl-phenyl)-but-1-enyl ester was prepared from undec-10-enoic acid 3-(3-isopropyl-phenyl)-but-1-enyl ester and
30 pentamethyl disiloxane.

Example 7

Test cloth was washed with a lipase-containing detergent to which one
35 or more delayed release fragrances had been added. Headspace analysis of the wet and dry laundry indicated the presence of the fragrant alcohols. The alcohol level was higher than when the test cloth was washed with a lipase-containing detergent to which one or more fragrant alcohols were added.

Example 8

Test cloth was washed with a lipase-containing detergent and then a
5 fabric softener, containing one or more delayed release fragrances, was
added to the rinse cycle. Headspace analysis of the wet and dry laundry
indicated the presence of the fragrant alcohols. The alcohol level was higher
than when the test cloth was washed with a lipase-containing detergent and
then a fabric softener, containing one or more fragrant alcohols, was added
10 to the rinse cycle.

Example 9

Axilla bacteria cultures containing 0.1 % precursor I were incubated
15 for 20 hours at 30 °C. After filtration from the cells, the presence of the
parent alcohol was in each case detected by headspace-GC techniques and/or
the majority of an 18 member panel.

The same tests were carried out with inactivated cultures (85°/20 min).
20 The odour of the parent alcohols could not be detected after incubation,
excluding therefore a hydrolysis by the medium or the culture.

Example 10

25 The following set forth examples for the use of the delayed release
ε fragrances of the present invention in various products. The methods of
forming the following compositions are well known to those skilled in the
art. All formulations may contain additional ingredients known to those
skilled in the art, e.g. colorants, opacifiers, buffers, antioxidants, vitamins,
30 emulsifiers, UV absorbers, silicones and the like. All products can also be
buffered to the desired pH. All values are % w/w.

Deo-colognes					
	Delayed Release Fragrances	0.5	1.5	2.5	6.0
35	Fragrance	0.5	1.5	2.5	6.0
	Triclosan (Ciba Geigy)	1.0	-	0.75	1.0
	Alcohol to	100	100	100	100

Deo-Sticks:

Antiperspirant

	Ethylene Glycol Monostearate	7.0
5	Shea butter	3.0
	Neobee 1053 (PVO International)	12.0
	Generol 122 (Henkel)	5.0
	Kesscowax B (Akzo)	17.0
	Dimethicone Dow Corning 345	35.0
10	Aluminum Sesquichlorhydrate	20.0
	Delayed Release Fragrances	0.5
	Fragrance	0.5

Antiperspirant

15	Steary Alcohol	17.0
	Castor Wax	3.0
	Talc	5.0
	Aluminum Zirconium Tetrachlorhydrate	20.0
	Delayed Release Fragrances	1.0
20	Fragrance	1.0
	Dimethicone Dow 245	to 100.0

Clear Deodorant Stick

	Witconol APM	43.0
25	Propylene Glycol	20.0
5	Alcohol 39C	20.0
	Demin water	7.0
	Monamid 150ADD	5.0
	Millithix 925	2.0
30	Ottasept Extra	0.5
	Delayed Release Fragrances	0.75
	Fragrance	0.75

Deodorant Stick

35	Propylene Glycol	69.0
	Demin Water	21.8
	Triclosan	0.2
	Sodium Stearate	8.0

	Delayed Release Fragrances	0.5
	Fragrance	0.5
	Alcohol free Deodorant Stick	
5	PPG-3 Myristyl Ether (Witconol APM)	36.0
	Propylene Glycol	36.0
	Demin Water	19.0
	Triclosan	0.25
	Sodium Stearate	7.75
10	Delayed Release Fragrances	0.5
	Fragrance	0.5
	Antiperspirant Aerosol	
	Absolute Ethanol	15.0
15	Zirconium Aluminum	
	tetrachlorhydrate	5.0
	Bentone 38	1.5
	Delayed Release Fragrances	0.75
	Fragrance	0.75
20	S-31 Hydrocarbon propellant	to 100.0
	Antiperspirant Pump	
	Demin water	57.5
	Aluminum Sesquichlorhydrate	20.0
25	Triton X-102 (Union Carbide)	2.0
5	Dimethyl Isosorbide (ICI)	20.0
	Delayed Release Fragrances	0.25
	Fragrance	0.25
30	Roll-On	
	Dimethicone DC 354 (Dow Corning)	69.0
	Bentone 38	10.0
	Rezal 36 GP (Reheis Chem. Co.)	20.0
	Delayed Release Fragrances	0.5
35	Fragrance	0.5

In the above, the following components were used:

	Triclosan	5-chloro-2-(2,4-dichlorophenoxy) phenol
	Neobee 1053	glycerol tricaprate/caprylate
	Generol 122	soya sterol
5	Kesscowax B	cetyl alcohol and glycol polymer
	Witconol APM	polypropylene glycol-3 myristyl ether
	Monamid 150 ADD	cocoamide diethanolamine
	Millithix 925	dibenzylidene sorbitol
10	Ottasept Extra	quaternium 18 hectorite
	Bentone 38	quaternium 18 hectorite
	Triton X-102	octoxynol-13
	Dimethicone DC 354	mixture of fully methylated linear siloxanepolymers end-blocked with trimethylsiloxy units
15		Aluminium zirconium tetrachlorohydroxyglycine
	Rezal 36 GP	



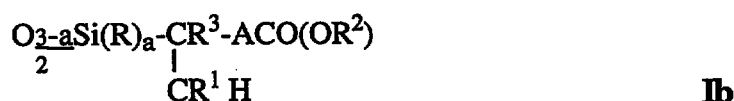
Claims

1. Organosiloxane fragrance precursor compounds having at least one organosiloxane unit of the formula

5



or



10 or a mixture thereof

and if any other siloxane unit(s) in the said organosiloxanes being present this/these is/are of formula



15

wherein R represents a substituted or unsubstituted C₁₋₈ alkyl group or a substituted or unsubstituted aryl group, R'' represents a hydrogen atom, a monovalent C₁₋₈ hydrocarbon group or a monovalent C₁₋₈ halogenated hydrocarbon group; R¹ represents a hydrogen atom or a substituted or unsubstituted C₁₋₈ alkyl group or a substituted or unsubstituted aryl group or a bond connecting CR¹ and CR³; R³ represents a hydrogen atom or a substituted or unsubstituted C₁₋₈ alkyl group or a substituted or unsubstituted aryl group; A represents (CR⁴)₂ whereby R⁴ represents a substituted or unsubstituted C₁₋₈ alkyl group or a substituted or unsubstituted aryl group or a hydrogen atom and whereby n has a value of 0-20 and each R⁴ is the same or different; OR² represents the residue of an olfactive alcohol or of the enol form of an olfactive aldehyde or olfactive ketone; a has a value of 0, 1 or 2; b has a value of 0, 1, 2 or 3.

30

2. Compounds of claim 1 wherein R is methyl, ethyl, butyl or phenyl, and/or R'' is hydrogen or a monovalent hydrocarbon or halogenated hydrocarbon group having less than 8 carbon atoms, in particular alkyl, alkenyl, aryl, alkanyl, aralkyl and halogen substituted alkyl, alkenyl, aryl, alkanyl and aralkyl whereby methyl, ethyl, vinyl, phenyl and 3,3,3-trifluoropropyl are preferred.

35

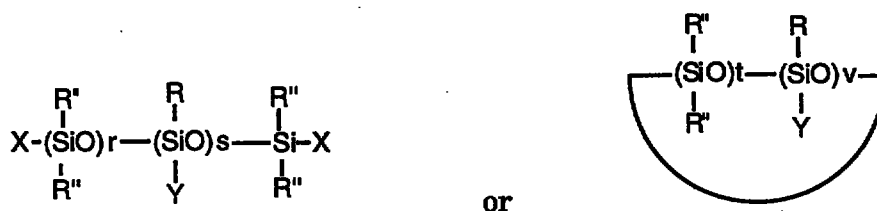


3. Compounds of claims 1 or 2 wherein at least 80% of all R and R" groups are methyl groups.

5 4. Compounds of any one of the claims 1 to 3 wherein a is 1 and b is 2 with the proviso that if the diorganosiloxane is a substantially linear polymer having at least two endblocking units, two units in which a has a value of 2, and two units in which the value of b has a value of 3 or one unit in which a has a value 2 and one unit in which b has a value of 3 are present.

10

5. Compounds of any one of the claims 1 to 4 wherein the siloxane part is either

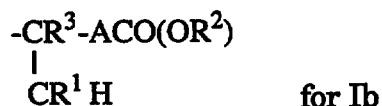


15

wherein X represents a group Y or a group R", and Y represents a group of the formula



20



ε
S P b

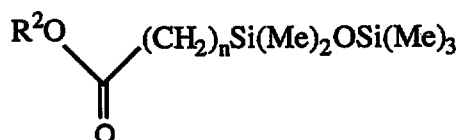
whereby r has a value of from 0 to 1000; s has a value of from 0 to 1000, and whereby at least one X denotes Y in the case that s = 0; t has the value of from 0 to 10; v has the value of from 1 to 10, preferably t+v has a value of 3 to 6.

25

6. Compounds according to claim 5 wherein if v+t is greater than 3, v-t has a value of at least 3.

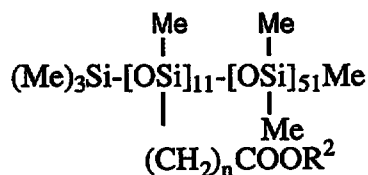
7. Compounds of claim 1 having the formula

30





8. Compounds of claim 1 having the formula



5 9. Compounds of any one of the claims 1 to 8 wherein n of A has a value of 1 to 10.

10 10. Compounds of any one of the claims 1 to 9 wherein for cosmetic applications the viscosity is from about 25 to about 500 mm²/s.

11. Compounds of any one of the claims 1 to 10 wherein OR² is a radical derived from a fragrant alcohol selected from the group listed in Table 1.

15 12. Compounds of claim 11 wherein the radical is derived from a fragrant alcohol selected from the group consisting of hexyl alcohol, 2-hexyl alcohol, heptyl alcohol, octyl alcohol, nonyl alcohol, decyl alcohol, undecyl alcohol, lauryl alcohol, 3-methyl-but-2-en-1-ol, cis-3-hexenol, cis-4-hexenol, 3,4,5,6,6-pentamethylheptan-2-ol, citronellol, geraniol, 6-ethyl-3-methyl-5-
20 octen-1-ol, 3,7-dimethyl-oct-3,6-dienol, 3,7-dimethyloctanol, 7-methoxy-3,7-dimethyl-octan-2-ol, cis-6-nonenol, 6,8-dimethyl-2-nonanol, 4-methyl-3-decen-5-ol, 1-phenyl ethanol, 2-phenyl ethanol, 2-phenyl propanol, 3-phenyl
ε propanol, 2-methyl-5-phenyl pentanol, 2-methyl-4-phenyl-pentanol, 3-methyl-
5-phenyl-pentanol, 2-(2-methylphenyl)-ethanol, 4-(4-hydroxyphenyl)butan-2-
25 one, 2-phenoxy ethanol, anisic alcohol, p-tolyl alcohol, cinnamic alcohol, vanillin, ethyl vanillin, eugenol, isoeugenol, anethol, borneol, cedrenol, farnesol, fenchyl alcohol, menthol, alpha ionol, tetrahydro ionol, 2-(1,1-dimethylethyl)cyclohexanol, 3-(1,1-dimethylethyl)cyclohexanol, 4-(1,1-dimethylethyl)cyclohexanol, 6,6-dimethyl-bicyclo [3.1.1]hept-2-ene-methanol,
30 p-menth-8-en-3-ol, 2,4,6-trimethyl-3-cyclohexenyl-methanol, 4-(1-methylethyl)cyclohexyl-methanol, 2,2,6-trimethyl-alpha-propyl cyclohexane propanol, 5-(2,2,3-trimethyl-3-cyclopentenyl)-3-methylpentan-2-ol, 3-methyl-5-(2,2,3-trimethylcyclopentyl-3-enyl)pent-4-en-2-ol, 2-ethyl-4-(2,2,3-trimethylcyclopentyl-3-enyl)but-2-en-1-ol, 4-(5,5,6-trimethylbicyclo[2.2.1]hept-
35 2-yl)-cyclohexanol, 2-(2-methylpropyl)-4-hydroxy-4-methyl-tetrahydropyran,



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2-cyclohexyl propanol, 2-(1,1-dimethylethyl)-4-methyl cyclohexanol, 1-(2-tert-butyl-cyclohexyloxy)-2-butanol, 1-(4-isopropyl-cyclohexyl)-ethanol.

13. Compounds of claim 12 wherein the radical is derived from a
5 fragrant alcohol selected from the group consisting of citronellol, geraniol, cis-3-hexenol, 4-methyl-3-decen-5-ol, 2-phenyl ethanol, iso eugenol, 3-methyl-5-(2,2,3-trimethylcyclopentyl-3-enyl)pent-4-en-2-ol, 2-ethyl-4-(2,2,3-trimethylcyclopentyl-3-enyl)but-2-en-2-ol, cinnamic alcohol and eugenol.

10 14. Compounds of any one of the claims 1 to 10 wherein OR² is a radical derived from a fragrant aldehyde selected from the group listed in Table 2.

15 15. Compounds of claim 14 wherein the radical is derived from a fragrant aldehyde selected from the group consisting of, 2,6,10-trimethylundec-9-enal, 11-carboxaldehyde, 2,4-dimethyl-cyclohex-3-ene, 4-carboxaldehyde-1,3,5-trimethyl-cyclohex-1-ene, 1-carboxaldehyde-2,4-dimethyl-cyclohex-3-ene, 1-carboxaldehyde-4-(4-hydroxy-4-methylpentyl)-cyclohex-3-ene, hex-2-enal, heptanal, 2,6-dimethyl-hept-5-eneal, decanal, 2-methyldecanal, undec-10-ene-1-al, undecanal, dodecanal, 2-methylundecanal, octanal, nonanal, undec-9-eneal, 2-phenyl-propanal, 4-methylphenyl acetaldehyde, 3,7-dimethyl-octanal, 7-hydroxy-3,7-dimethyl-octanal, 2-(4-(1-methylethyl)phenyl)-ethanal, 3-(3-isopropylphenyl)-butanal, 1-carboxaldehyde-4-(4-methyl-3-penten-1-yl)-cyclohex-3-ene, 2-methyl-4-(2,6,6-trimethylcyclohex-2-en-1-yl)-butanal, 2-methyl-3-(4-tert-
25 butylphenyl)propanal, 4-(1,1-dimethylethyl)-benzenepropanal, alpha-methyl-1,3-benzodioxole-5-propanal, 3,7-dimethyl-oct-6-en-1-al, 2-methyl-3-(p-isopropylphenyl)-propionaldehyde, 4-(4-hydroxy-4-methylpentyl)-cyclohex-3-en-1-carboxaldehyde, alpha-methyl-1,3-benzodioxole-5-propanal.

30

16. Compounds of claim 15 wherein the radical is derived from a fragrant aldehyde selected from the group consisting of undec-10-ene-1-al, undec-9-eneal, decanal, dodecanal, octanal, 2-methylundecanal, 3-(3-isopropylphenyl)-butanal, 4-(4-hydroxy-4-methylpentyl)-cyclohex-3-en-1-carboxaldehyde and 2-methyl-3-(4-tert-butylphenyl)propanal.

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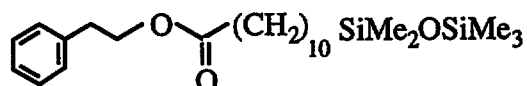
17. Compounds of any one of the claims 1 to 10 wherein OR² is a radical derived from a fragrant ketone selected from the group listed in Table 3.

5 18. Compounds of claim 17 wherein the radical is derived from a fragrant ketone selected from the group consisting of benzylacetone, carvone, 1,2,3,5,6,7-hexahydro1,1,2,3,3-pentamethyl-4H-inden-4-one, methyl heptenone, dimethyl octenone, 2-(butan-2-yl)-cyclohexanone, 2-hexyl-cyclopent-2-en-1-one, 2-(1-methylethyl)-5-methyl-cyclohexanone, 2-(2-
10 methylethyl)-5-methyl-cyclohexanone, 3-methyl-cyclopentadecanone, 4-tert-pentyl-cyclohexanone, 2-oxo-1-pentyl-cyclopentaneacetic acid, methyl ester, 1-(1,2,3,4,5,6,7,8-octahydro-2,3,8,8-tetramethyl-2-naphthalenyl)-ethanone, 3-methyl-5-propyl-cyclohex-2-en-1-one.

15 19. Compounds of claim 18 wherein the radical is derived from 2-oxo-1-pentyl-cyclopentane acetic acid methyl ester.

20 20. A compound as defined in claim 1 and as illustrated in the Examples 3, 4 or 6.

21. A compound as defined in claim 1, having the formula



25 22. A composition for cosmetic application to the human skin, laundry products, detergents or fabric softeners containing at least one of the compounds according to any one of the claims 1 to 21.

30 23. A cosmetic product, a laundry product, a detergent or a fabric softener containing at least one of the compounds according to any one of the claims 1 to 21.

35 24. A process for prolonging the effect of diffusion of the characteristic odour of an odoriferous alcohol and/or aldehyde and/or ketone on human skin or in laundry products, detergents or fabric softeners comprising applying at least one of the compounds of any one of the claims 1 to 21.



25. A method of suppressing human body malodour by means of applying a composition of claim 22 or a cosmetic product of claim 23 to the human skin.

- 5 26. The use of at least one of the compounds of any one of the claims 1 to 21 as a fragrance precursor in a cosmetic composition or cosmetic product, laundry product, detergent or fabric softener.

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